Electronic Structure and Atomistic Computations for the interpretation of Synchrotron Experiments

Lecture Focus: Mainly Density Functional Theory (DFT) based computations

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Density Functional Theory (DFT)

Total energy:



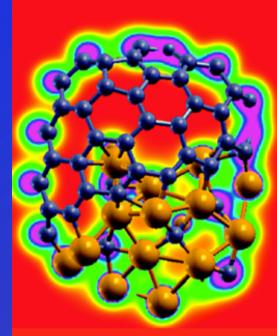
all ground-state properties determined by

Electronic density:

→ ρ obtained by solving
 One-electron Schrödinger equation
 (Kohn-Sham) :

$$H\phi_i(\vec{r}) = \varepsilon_i \phi_i(\vec{r})$$

For the N lowest energy states



Nanotube cap on $Ni_{55}C_{14}$ Nanoparticles Gomez-Ballesteros and Balbuena, *Phys. Chem. Chem. Phys.*, **17**, 15056 (2015)

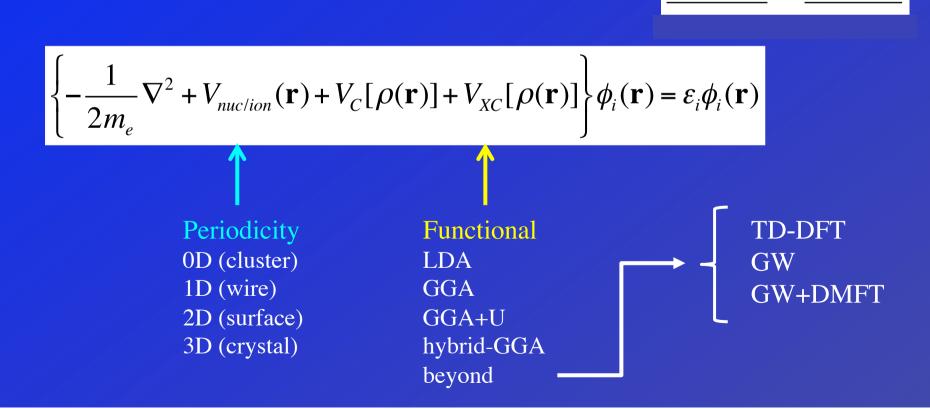
 $\rho(\vec{r}) = \sum |\phi_i(\vec{r})|^2$ $\varepsilon_i \leq E_F$

Density Functional Theory (DFT)

One-electron Kohn-Sham equation:

W. Kohn Nobel 1998

L. J. Sham



Modeling Materials with DFT

Input: Atomic Structure or Composition

Output: Total Energy & Electronic Structure

Atomic Structure

atomic forces equilibrium structure vibrational modes phonons stresses/pressure

Energetics formation energy internal energy enthalpy activation energies

Electronic properties

electron charge distribution density of states band structure magnetism

From atomic clusters to crystals

With periodic potential

$$\phi_{\vec{k}}^n(\vec{r}) = u_n(\vec{r})e^{i\vec{k}\cdot\vec{r}}$$

Bloch state

F. Bloch 1905-1983

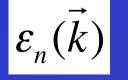


u: periodic (unit cell) functionk: wavevector in Brillouin zone

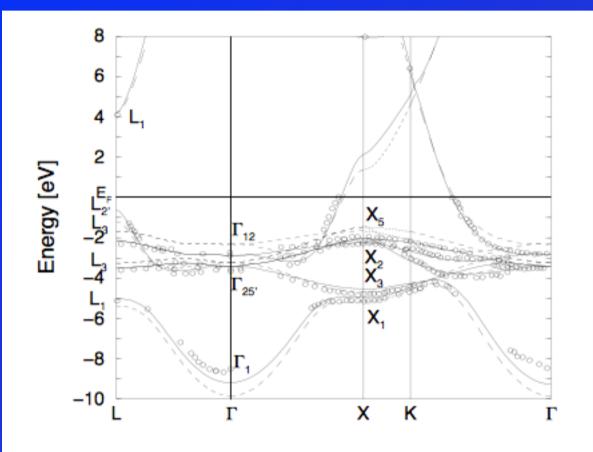


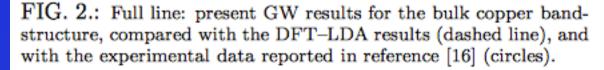


Band structure



Electronic bulk band structure: Bulk Cu - GW compared to LDA





From: Andrea Marini, Giovanni Onida, and Rodolfo Del Sole, PRL **88**, 016403 (2001)

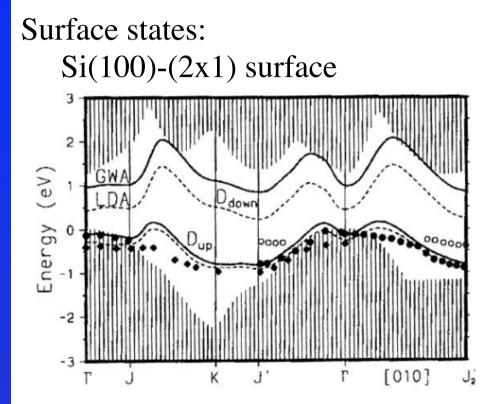
GW(solid line) best LDA (dashed line) not bad

PS: Not always so: strongly correlated materials

PE from:[16] R. Courths and S. Hufner, Phys. Rep. 112, 53 (1984)



Electronic surface band structure Si surface states – GW versus LDA



Rohlfing, Krüger, and Pollmann, PRB 52, 1905 (1995)

Figure 15. Calculated dangling-bond bands. Full curves, GWA energies; dashed curves, LDA energies. The experimental results are shown by diamonds (Uhrberg *et al* 1981) and circles (full and open) (Johansson *et al* 1990). (After Rohlfing *et al* 1995b).

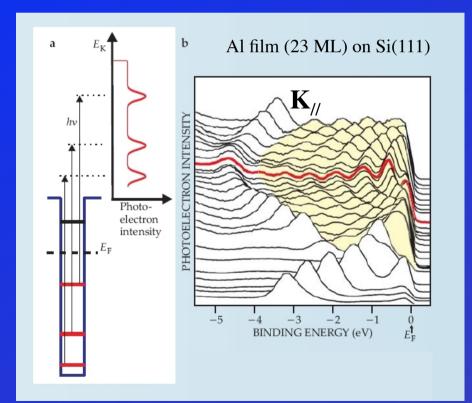
Quantum-size effects on surface reactivity

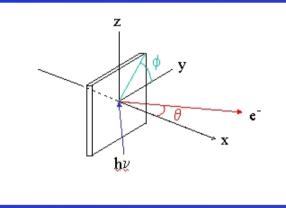
Quantum-size effects are effects related to Quantum Well States (QWS), e.g., in ultrathin epitaxial films

Observation of quantum well states *in thin metal films*

By photoemission (PE): occupied QWS

Angle-resolved photoelectron spectroscopy



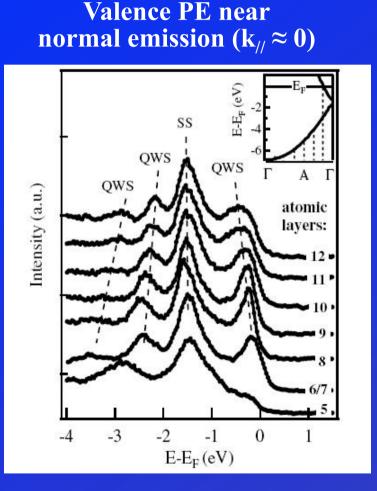


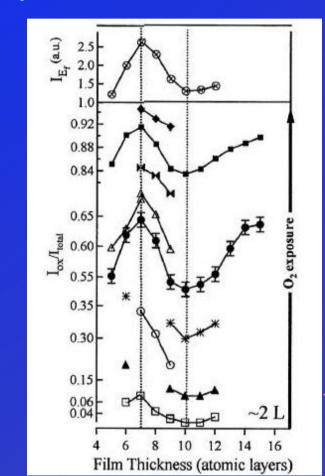
Measures the kinetic energy E_K and momentum $\mathbf{p}_{/\!/}$ of electrons emitted from the surface of the film

Tringides, Jalochowski, Bauer, Physics Today **60** (no 4), 50 (2007)

Quantum-size effects on surface reactivity: Mg films

Spectromicroscopy:Aballe et al. PRL 93, 196103(2004); J. Phys.: Cond. Matter 22, 015001 (2010)





 $I_{k_{//} \approx 0} (E_F)$

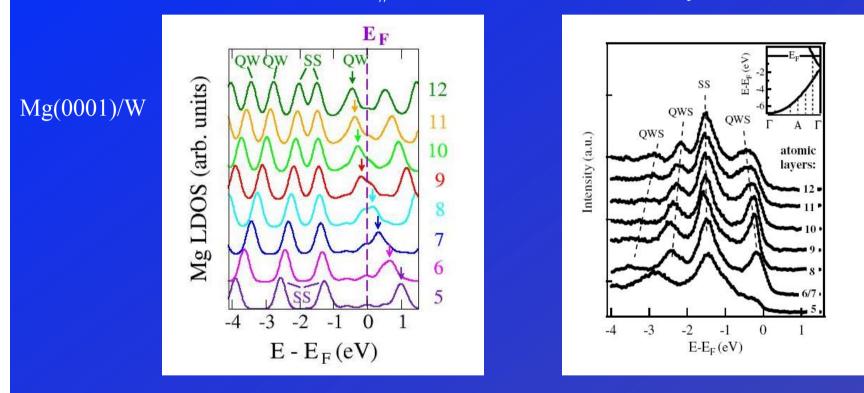
Oxidation rate (from intensity of Mg 2p oxide component)

Drastic (1-2 orders of magnitude) changes!

Origin of the correlation between surface reactivity and QWS ?

DFT calculations of Mg thin films on W(110)

Calculated DFT DOSversus measured normal-emission PE spectraPartial DOS at $\mathbf{k}_{ll} = \mathbf{0}$ PE by Aballe *et al.*



• Good general agreement in the peak positions

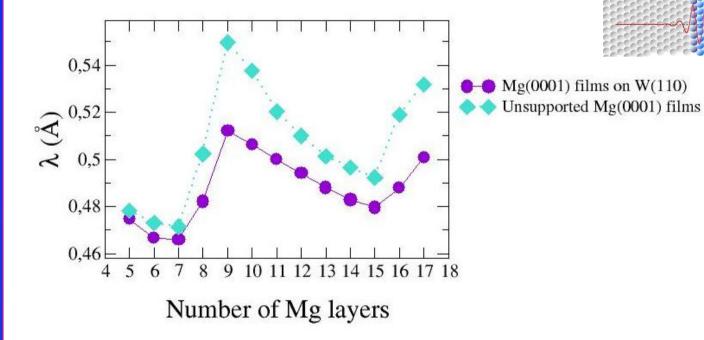
Binggeli and Altarelli, Phys. Rev. Lett. 96, 36805 (2006); Phys. Rev. B 78, 35438 (2008)

DFT calculations for Mg films on W(110)

 $\rho_{E_F}(z) \sim e^{-z/\lambda}$

 $\phi_{E_{E}}(z)$

• Decay length in vacuum λ of the metal film electronic local density of states at $E_{\rm F}$

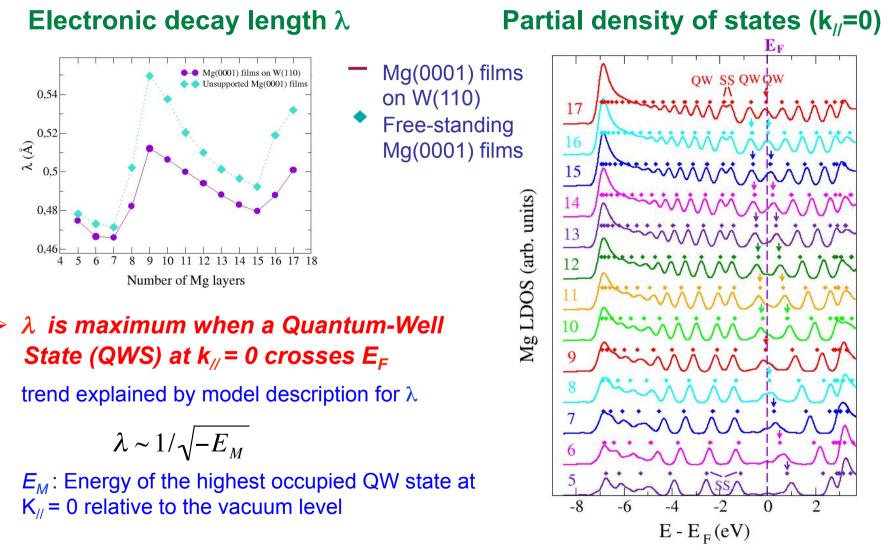


> Substantial oscillations in λ

substrate reduces the variation from 17% to 10 %

Binggeli and Altarelli, Phys. Rev. Lett. 96, 36805 (2006); Phys. Rev. B 78, 35438 (2008)

DFT calculations of Mg thin films on W(110)



Binggeli and Altarelli, Phys. Rev. Lett. 96, 36805 (2006); Phys. Rev. B 78, 35438 (2008)

Origin of the changes in the surface reactivity ?

Binggeli and Altarelli, Phys. Rev. Lett. 96, 35805 (2006); Phys. Rev. B 78, 35438 (2008)

The changes in λ are expected to influence the electron transfer process, by tunneling, which is believed to control the initial sticking of O₂ on the metal surface (via the attractive image charge potential on the ionized O₂⁻ molecule)

Hellman et al., Surface Science 532, 126 (2003).

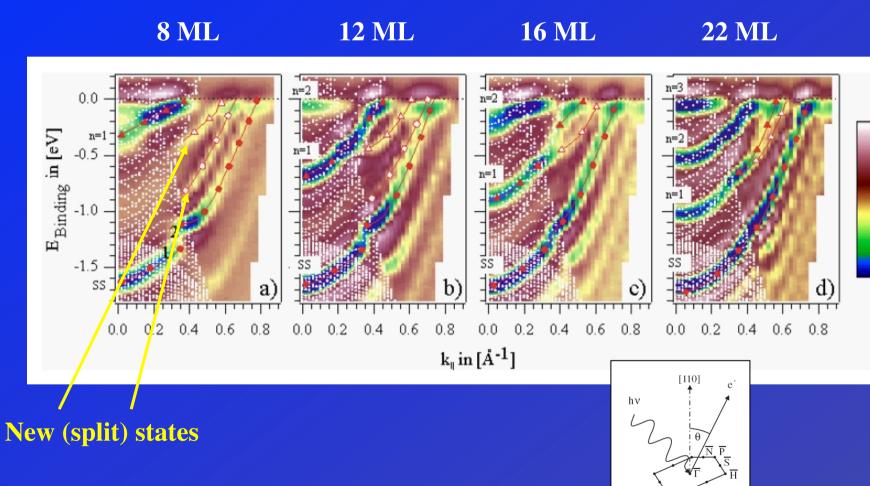
The changes in λ are expected to have an exponential impact on the electron tunneling rate, and hence on the initial sticking of O₂

A 10% change in λ will produce a 100% change in the transfer rate ~ $e^{-d/\lambda}$, at d ~3.5 Å, which is of the order of magnitude of the experimental change in the oxidation rate at low O₂ exposure

Can explain the order of magnitude change observed experimentally

Angle resolved photoemission spectra

Mg films on W(110)



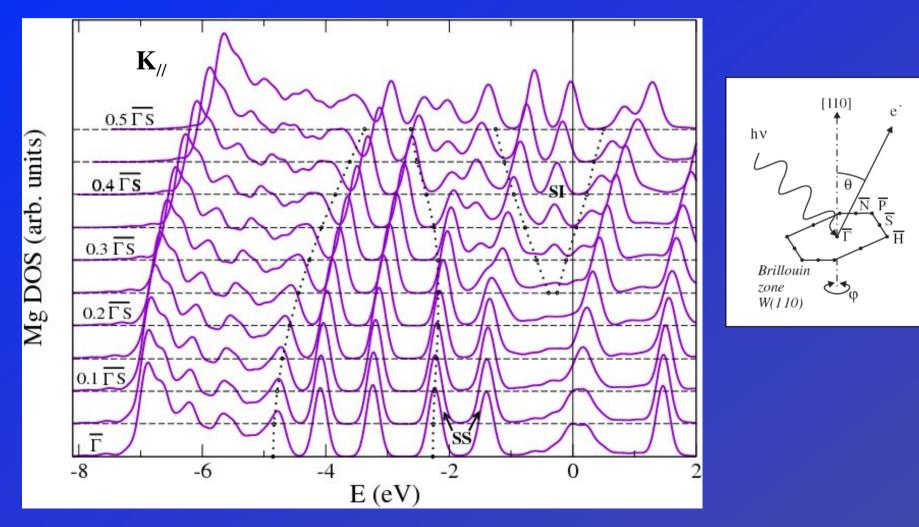
Brillouin zone W(110)

Koitzsch et al., Phys. Rev. Lett. 95, 126401 (2005)

DFT results for Mg QWS 8 Mg ML on W(110)

Binggeli and Altarelli, Phys. Rev. B **78**, 35438 (2008)

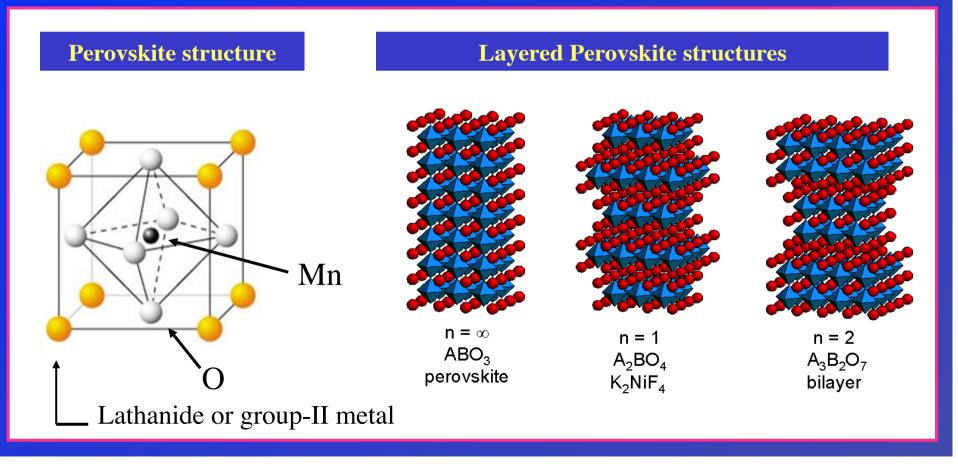
$K_{//}$ along the ΓS direction



Resonant elastic x-ray scattering to probe orbital order in complex transition-metal coumpounds

(such as manganites)

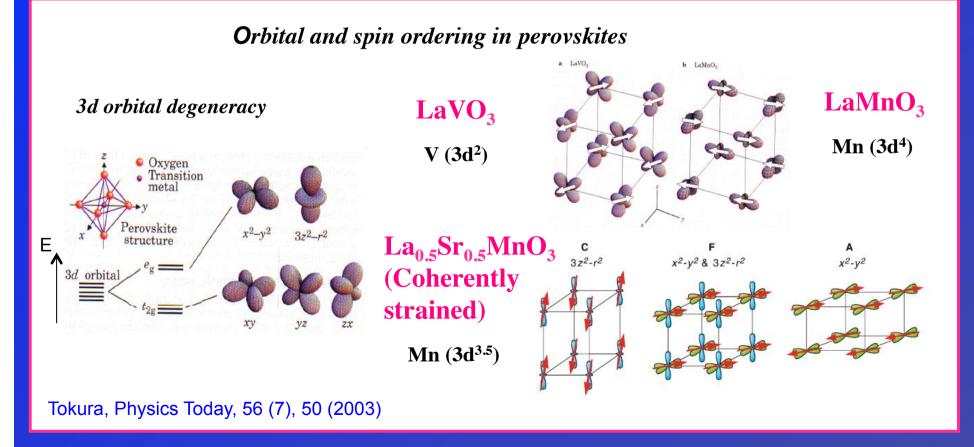
 Properties controlled by a complex interplay between structural, magnetic, and valence electronic charge/orbital degrees of freedom



Resonant elastic x-ray scattering to probe orbital order in complex transition-metal coumpounds

(such as manganites)

• Properties controlled by a *complex interplay* between *structural*, *magnetic*, and valence *electronic charge/orbital* degrees of freedom



The orbital order in complex transition-metal coumpounds

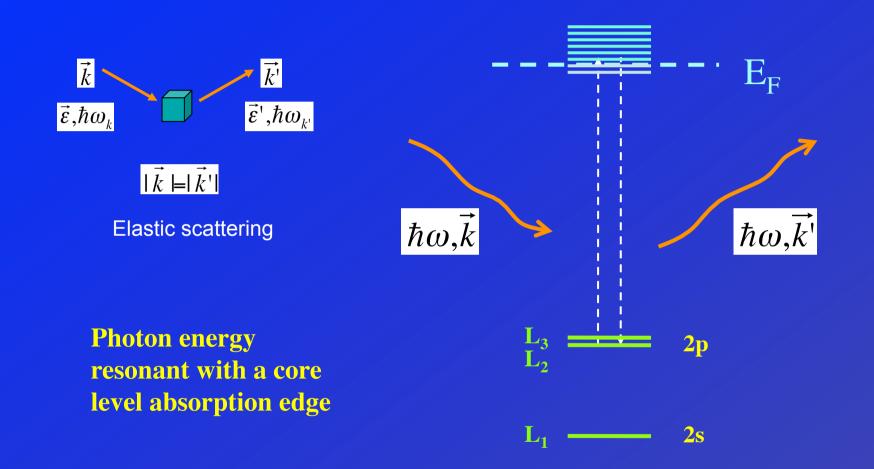
• The orbital degree of freedom of the 3d electrons plays a key role in the physics of manganites, but the orbital ordering is difficult to observe experimentally (hidden to most experimental probes)

• **Resonant elastic X-ray Scattering** (RXS) is a powerful **tool to probe orbital ordering,** but the interpretation of the experimental spectra is often controversial



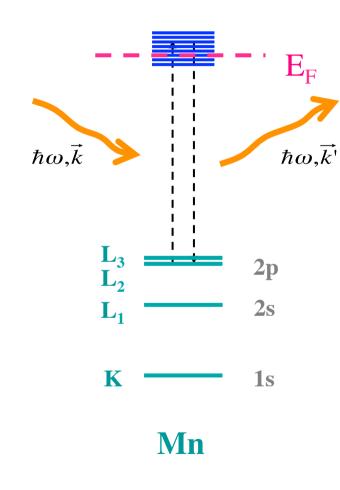
importance of theoretical predictions

Resonant elastic X-ray Scattering (RXS)



Resonant elastic x-ray scattering is a second order process in which a core electron is virtually promoted to some intermediate states above the Fermi energy, and subsequently decays to the same core level

Resonant elastic x-ray scattering



Scattering for Bragg condition:

$$\vec{k} - \vec{k}' = \vec{G}_{lmn},$$

- \vec{G}_{lmn} : reciprocal lattice (Bragg) vector
- Resonant scattering amplitude:

$$f(\vec{G},\hbar\omega) \propto \sum_{I(ions)} e^{i\vec{G}\vec{d}_I} \sum_{\alpha,\beta} F^I_{\alpha,\beta}(\hbar\omega) \varepsilon_{\alpha} \varepsilon_{\beta}$$

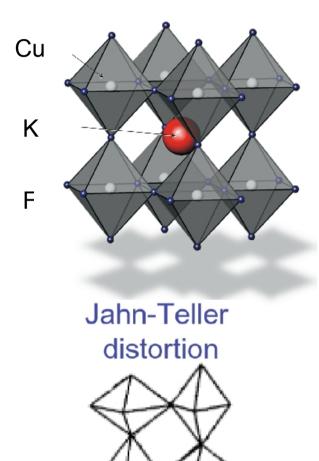
$$F_{\alpha,\beta}^{I}(\hbar\omega) \propto \sum_{n} \frac{\langle 0 | R_{\alpha}(I) | n \rangle \langle n | R_{\beta}(I) | 0 \rangle}{\hbar\omega + E_{0} - E_{n} - i\Gamma/2}$$

- > selective probe of order/periodicity
 > element sensitive
- > photon-energy dependent

 \succ At L_{2,3} edge, in principle **orbital-order sensitive** (optical selection rule)

Resonant x-ray scattering to probe orbital order in KCuF₃?

Crystal structure:



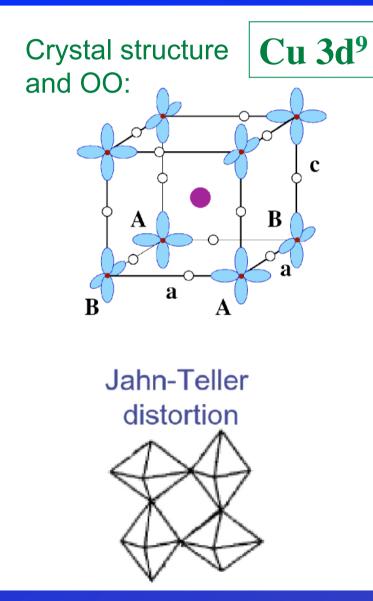
RXS at the Cu K edge in KCuF₃

Cu 1s -> 4p dipole transitions

Exp.: Paolasini et al., PRL 88, 106403 (2002); Caciuffo et al., PRB 65, 174425 (2002)

- pseudo cubic perovskite
- cooperative Jahn-Teller distortion below 1000 K
- Neel temperature ~38 K
- $d_{x^2 y^2}$ hole antiferroorbital ordering

Resonant x-ray scattering to probe orbital order in KCuF₃?



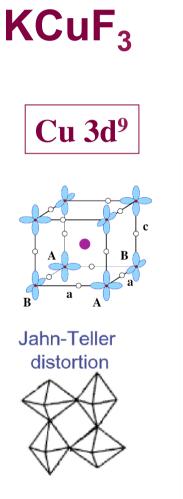
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Resonant x-ray scattering to probe orbital order in KCuF₃?



LDA+U

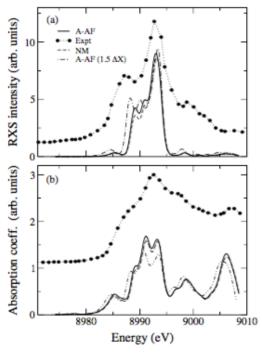


FIG. 9: Orbital RXS intensity (a) and absorption coefficient (b), as a function of photon energy, near the Cu K edge in KCuF₃. The calculated spectra of the A-AF (solid line) and NM (dashed line) structures are displayed. The effect of a 50% increase in the quadrupolar distortion, ΔX , in the A-AF structure is also shown (dashed-dotted line). The experimental data are from Ref. 7; the measured RXS spectrum corresponds to a (3,3,1) orbital Bragg reflection and a $\sigma - \pi'$ polarization.

Binggeli and Altarelli, Phys. Rev. B 70, 085117 (2004)

Cu K-edge: 1s -> 4p dipole transitions Atomic resonant scattering amplitude

$$F_{\alpha,\beta}^{j}(\hbar\omega) = \sum_{\boldsymbol{k},n} \frac{\langle \psi_{0}^{(j)} | r_{\alpha}(j) | \psi_{\boldsymbol{k},n}^{4p} \rangle \langle \psi_{\boldsymbol{k},n}^{4p} | r_{\beta}(j) | \psi_{0}^{(j)} \rangle}{\hbar\omega + E_{0} - E_{\boldsymbol{k},n}^{4p} - i\Gamma/2}$$

RXS intensity

$$I(G, \hbar\omega) \propto \left| \sum_{j} e^{iQ \cdot R_{j}} \sum_{\alpha, \beta} F^{j}_{\alpha, \beta}(\hbar\omega) \epsilon_{\alpha} \epsilon'_{\beta} \right|^{2} \text{ with } \vec{G} = \vec{Q}_{OO}$$
$$I_{\text{orb}}(\hbar\omega) \propto \left| F^{A}_{x,x}(\hbar\omega) - F^{A}_{y,y}(\hbar\omega) \right|^{2},$$

Absorption

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For Orbital Ordering

 $A(\hbar\omega) \propto Im[F^A_{x,x}(\hbar\omega) + F^A_{y,y}(\hbar\omega) + F^A_{z,z}(\hbar\omega)]$

- Good general agreement with exp.
- RXS: controlled by Jahn-Teller distortion not sensitive to OO not sensitive to spin order

Resonant elastic x-ray scattering to probe orbital order in manganites ?

RXS at the Mn L_{2,3} edges in manganites

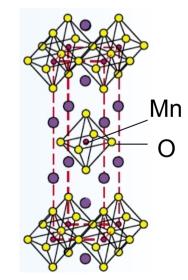
 $(L_2:2p_{1/2}, L_3:2p_{3/2})$

Mn 2p -> 3d dipole transitions

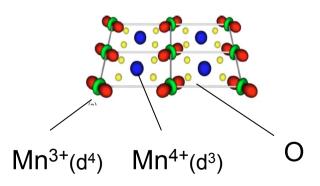
Exp.: Wilkins, Spencer, Hatton et al., PRL 91, 167205 (2003)

La_{0.5}Sr_{1.5}MnO₄

Crystal structure

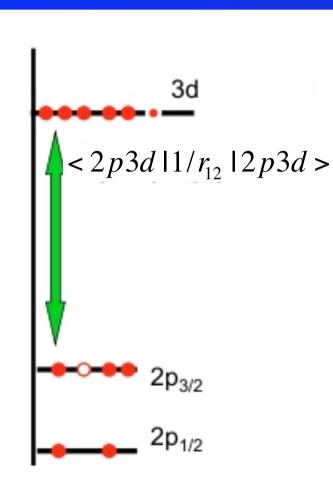


Mn-3d charge and orbital order



L_{2,3} absorption edge

Multiplet effects in XAS, RXS



Strong overlap of core and valence wave functions

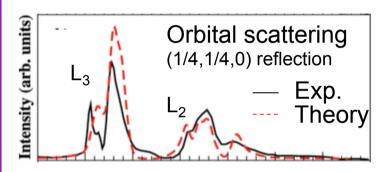
2p -> 3d dipole transitions

 $< 2p3d |1/r_{12}|2p3d > Coulomb$ interaction terms cannot be neglected

Single particle picture breaks down!

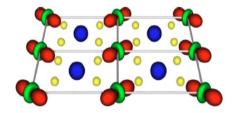
Resonant x-ray scattering at the Mn $L_{2,3}$ edges in $La_{0.5}Sr_{1.5}MnO_4$

La_{0.5}Sr_{1.5}MnO₄



Wilkins, Stojic, Beale, Binggeli *et al.*, Phys. Rev. B 71, 245102 (2005); J. Phys. Condens. Matter 18, L323 (2006).

Mn³⁺ atomic multiplet calculations in a crystal field



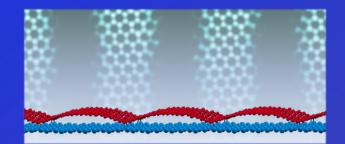
- Good general agreement with experiment
- RXS sensitive to OO and Jahn-Teller distortion

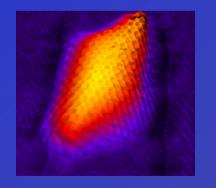
Nanostructured Graphene

Temperature-driven reversible rippling and bonding of a graphene superlattice A. Locatelli, C. Wang, C. Africh, N. Stojic, T. O. Mentes, G. Comelli, and N. Binggeli, *ACS Nano* **7**, 6955 (2013)

Nanobubbles at GPa Pressure under Graphene G. Zamborlini, M. Imam, L. L. Patera, T. O. Mentes, N. Stojic, C. Africh, A. Sala, N. Binggeli, G. Comelli, and A. Locatelli, *Nano Lett.* **15**, 6162 (2015)

on Ir(001) surface

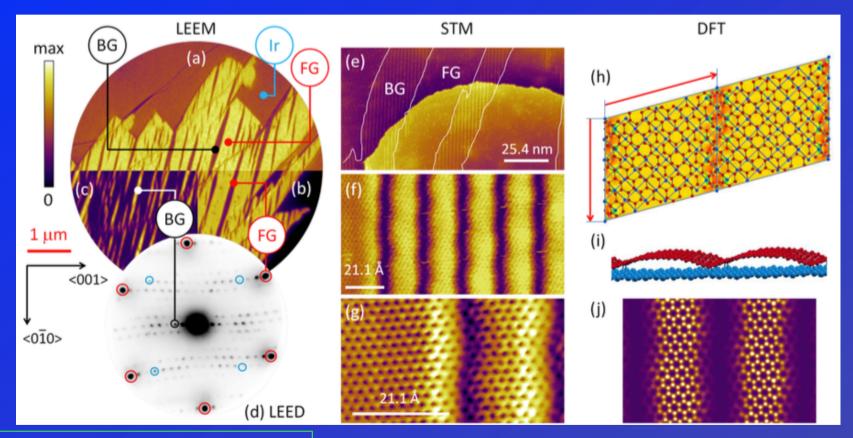




Graphene nanobuckled phase: on the Ir(001) surface

@IOM-CNR TASC

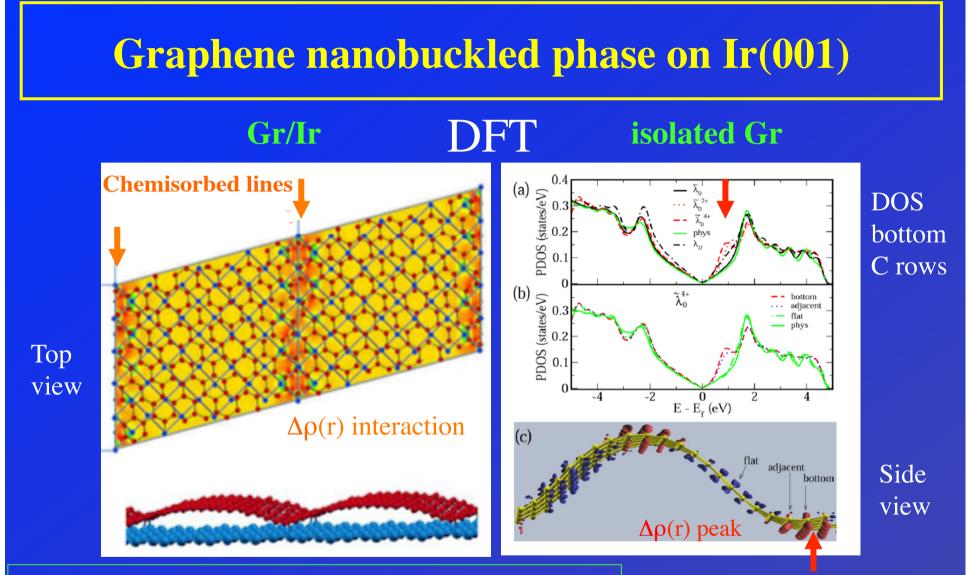
@ Elettra nanospectroscopy beamline



BG: buckled phase - large regular ripples with regular nm periodicity ($\lambda_0 = 2.1$ nm)

A. Locatelli, C. Wang, C. Africh, N. Stojic, T. O. Mentes, G. Comelli, and N. Binggeli, *ACS Nano* 7, 6955-6963(2013)

@ICTP

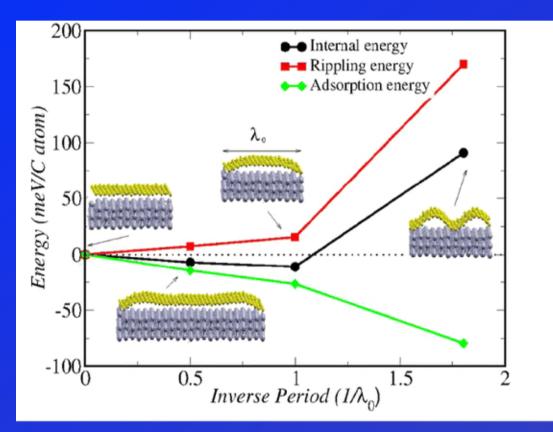


- Chemisorbed line includes ~ 3 rows of C
- BG curvature induces a C-2p_z DOS feature near E_F
- the corresponding states are found to be largely responsible for strong local chemisorption

M. Imam, N. Stojic, and N. Binggeli, *J. Phys. Chem.* C, **118**, 9514 (2014)

Graphene nanobuckled phase on Ir(001)

DFT



The nm periodicity is explained by the trends of the rippling and chemisorption energy

The nm period is related to the onset of the non-linear behavior of the rippling energy - which overcomes the chemisorption energy at short period

M. Imam, N. Stojic, and N. Binggeli, *J. Phys. Chem.* C, **118**, 9514 (2014)

Graphene nanobubbles on Ir(001)

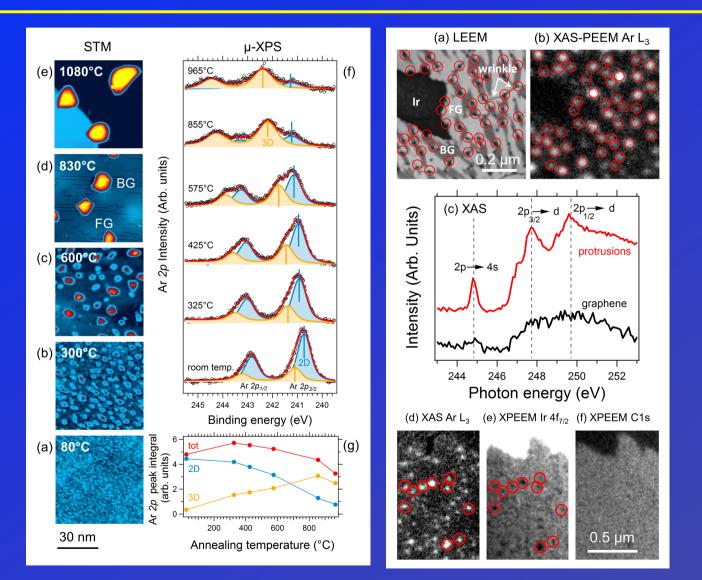
Nanospectroscopy group @ Elettra:

Upon irradiation with Ar⁺ ions (sputt. 0.1-0.4 kV) and subsequent annealing: T: 80 C, ...,1080 C

STM @ CNR-IOM TASC

Trapped Ar

Ripening with T

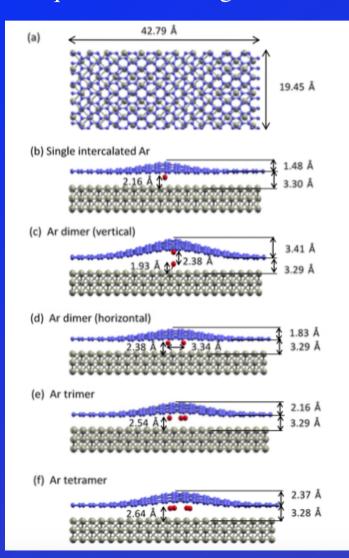


Zamborlini et al. Nano Lett. 15, 6162 (2015)

Graphene nanobubbles on Ir(001)

DFT

Equilibrium configurations



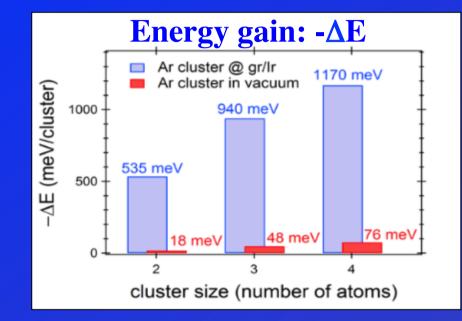
The Ar-Ar bond in intercalated cluster is strongly contracted with respect to bulk Ar (3.82 Å) or Ar dimer in vac. (3.74 Å)

The corresponding effective pressure experienced by intercalated Ar cluster is in the GPa range (8-25 GPa)

G. Zamborlini, <u>M. Imam</u> et al. *Nano Lett*. **15**, 6162 (2015)

Graphene nanobubbles on Ir(001)

Ar cluster formation energy: $\Delta E = E^{NAr} - N E^{1Ar}$



G. Zamborlini, <u>M. Imam</u> et al. *Nano Lett*. **15**, 6162 (2015)

- Drastic increase in the energy gain for Ar cluster formation at the Gr/Ir interface (compared to Ar cluster in vacuum), related to the distortion of the physisorbed graphene
- Major gain in energy when two or more bubbles merge into a single one
- The bubble formation is driven by minimization of the energy cost of membrane distortion and loss of adhesion