

Oil recovery and mitigation processes: insights from multiscale molecular simulations

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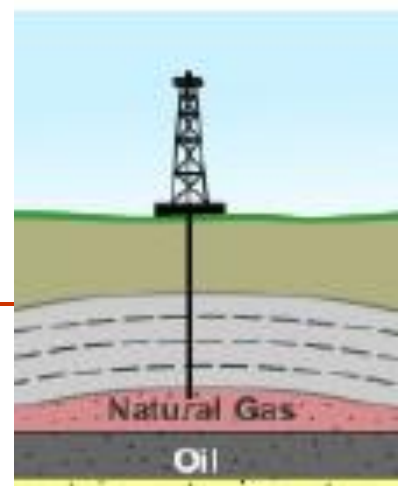
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Outline

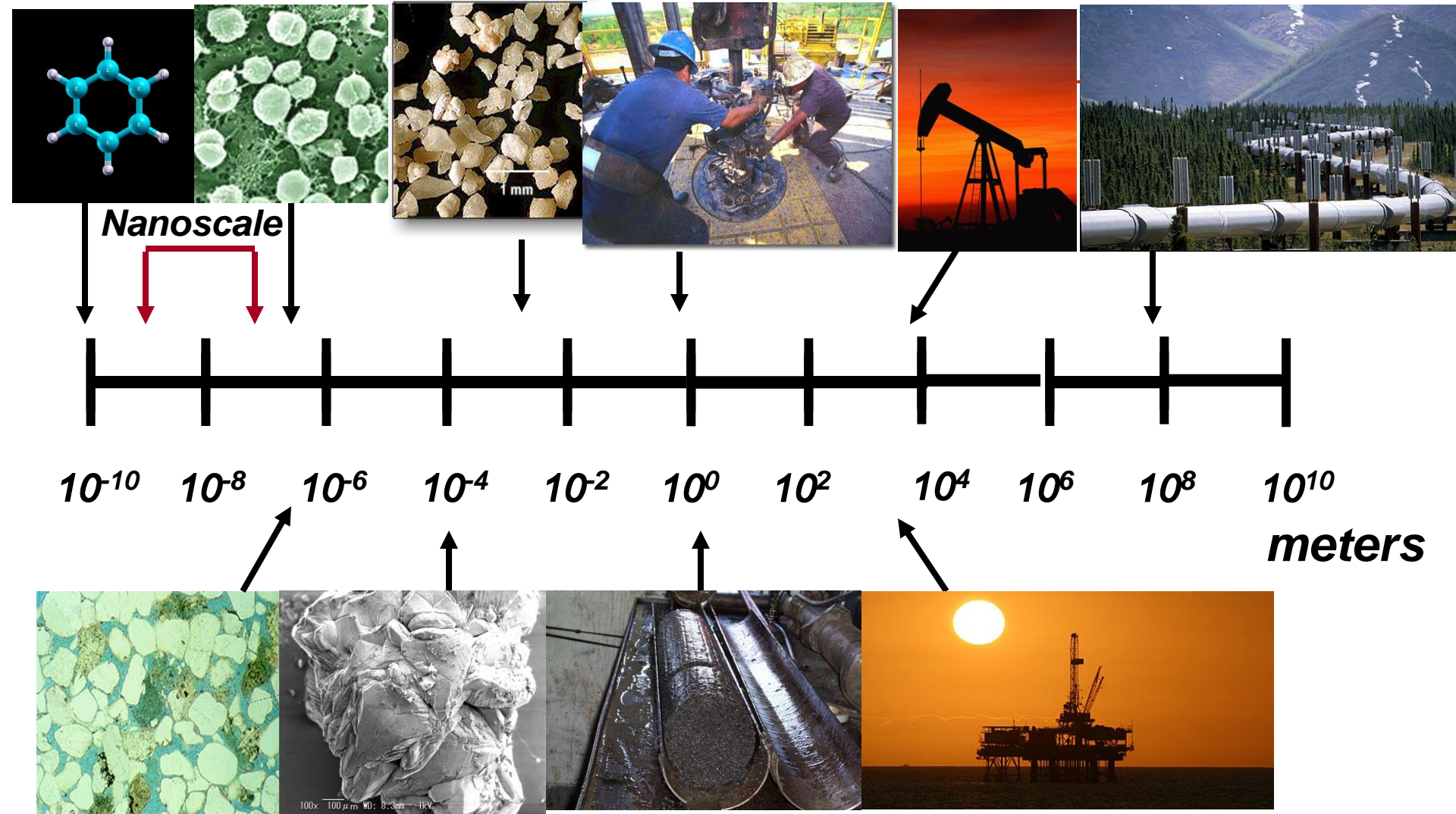
1. *Nano and O&G industry*
2. *Multiscale molecular simulations*
3. *NANO-EOR – **surface** driven flow*
4. *NANO-IOR – **pressure** driven flow*
5. ***Nanoaggregation** of complex molecules*
6. *Concluding Remarks*



Motivation

- The current average recovery factor from conventional oil reservoirs is $\sim 35\%$.
- Poor sweep effect in the reservoir & capillary forces
- Would it be possible to reach **70% recovery factor** in conventional oil fields?
- As oil recovery processes involve the interaction between mineral and hydrocarbons, the optimization of oil production requires **deep understanding** of reservoir properties at **different scales**.

Crossing scales in Oil & Gas



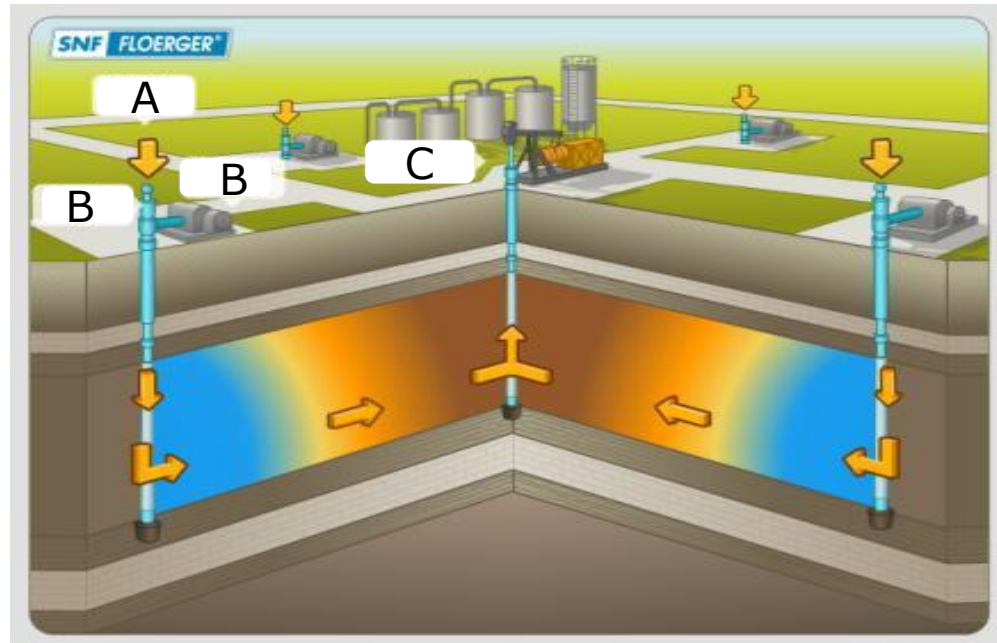
Complex physical phenomena in Materials
Oil&Gas: How the large can drive small systems ?

Temperature, Pressure,
Salinity, Heterogenous
and multiphase media

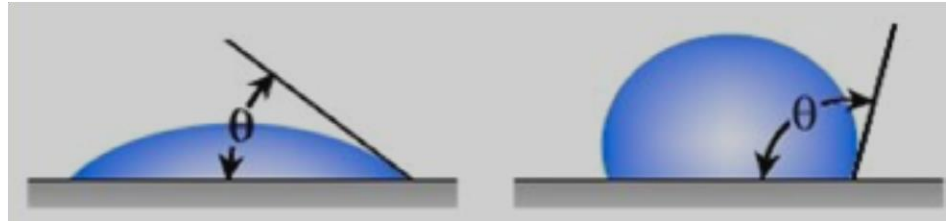
IOR & EOR

- Improved and Enhanced Oil Recovery (IOR and EOR) techniques are currently of great strategic importance to improve oil wells production.

- Water/brine
- CO₂
- N₂
- Surfactants
- Thermal
- Bio

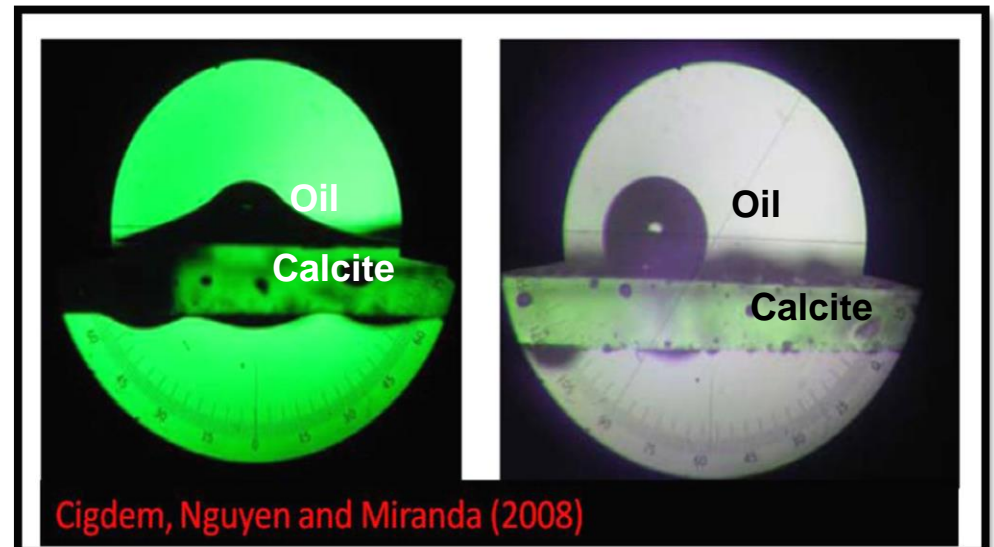
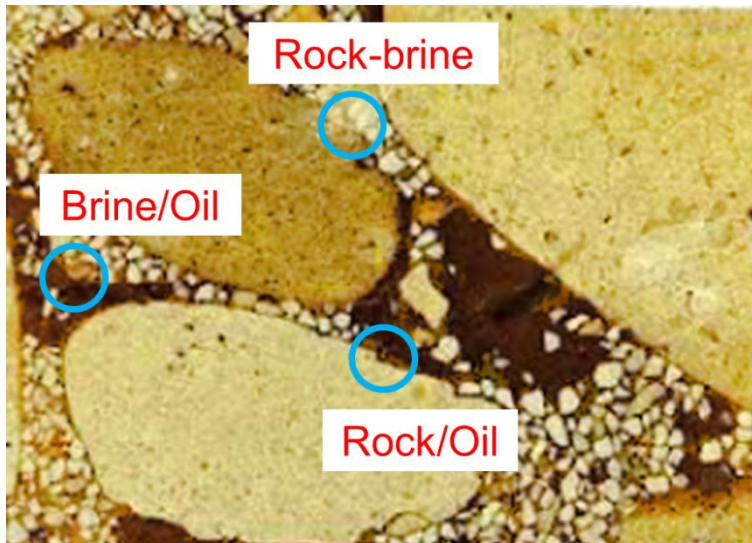


IOR & EOR

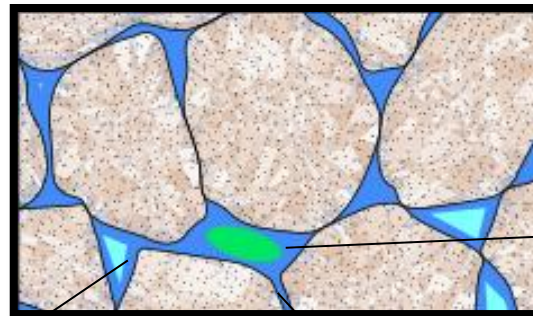
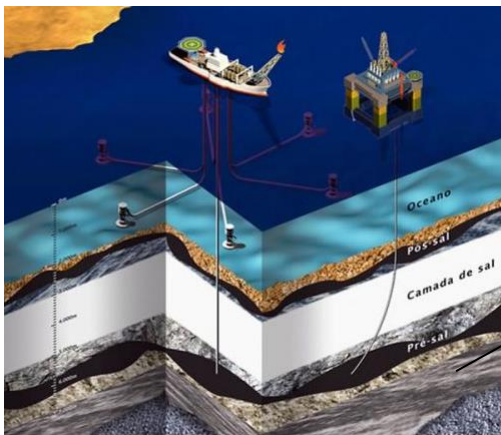


- The main strategy is to reduce the interfacial tension (IFT) or the viscosity of crude oil by molecular additives, which can be adsorbed on the oil-fluid interface, or migrate to the crude oil through the interface.

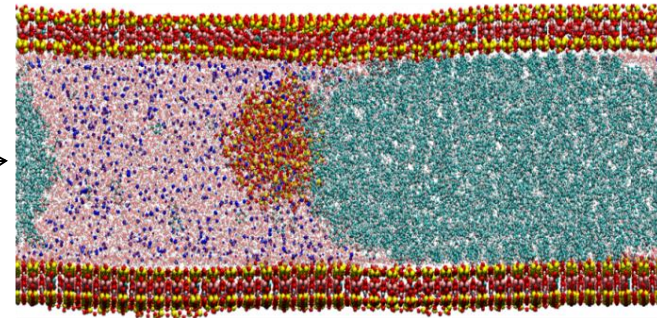
H₂O + SiO₂ nanoparticle



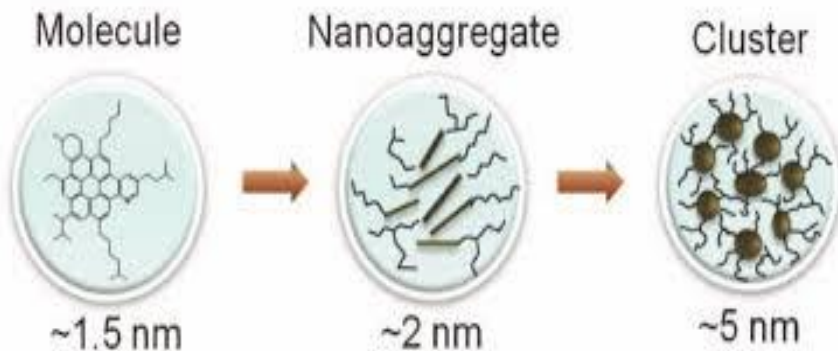
Complex physical phenomena in Materials and O&G industry from a Nanoscience perspective



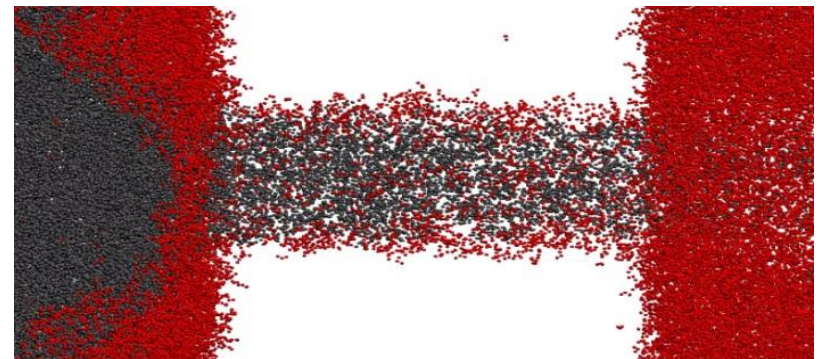
NANO-EOR - Surface driven flow: NPs at interfaces
brine-oil-rock over scales



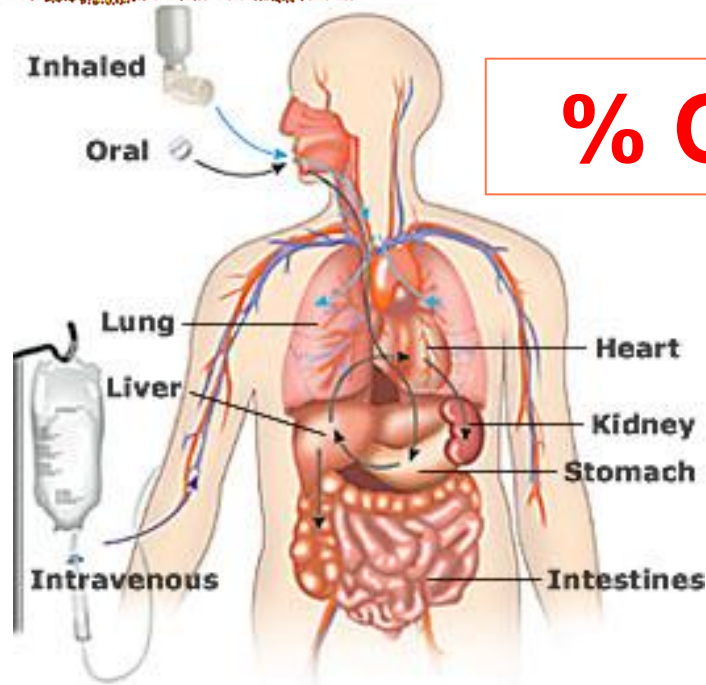
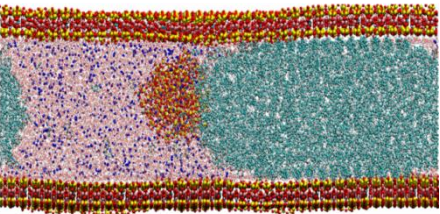
ASPHALTENES
Aggregation of complex matter



NANO-IOR – Pressure driven flow
Fluid confinement, multiphase fluids
Flow in NANO porous media

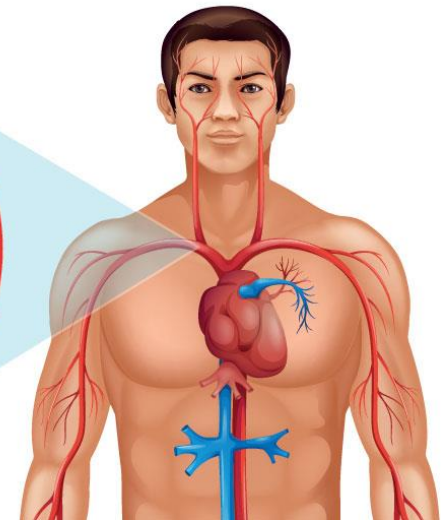
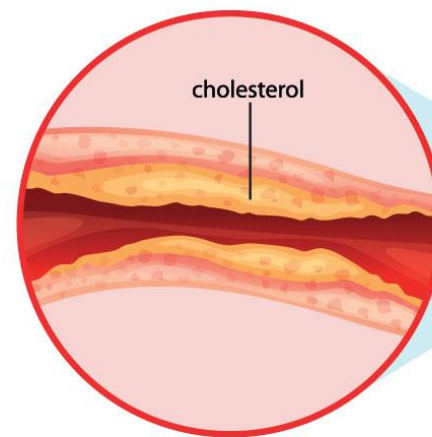


The health of oil reservoirs

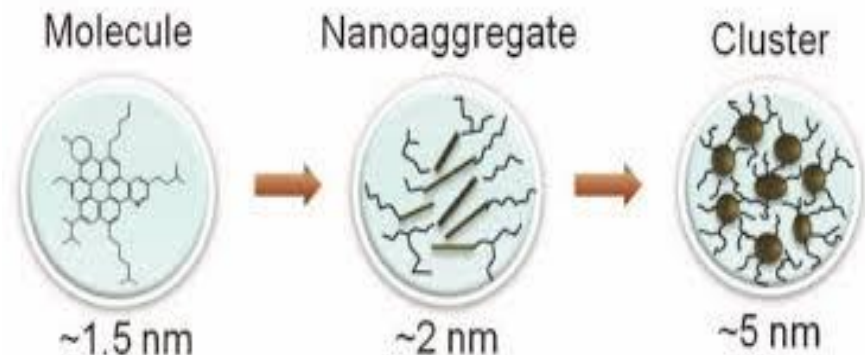
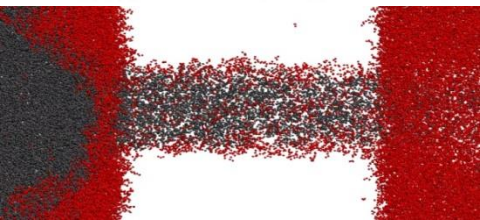


% Oil ?

Cholesterol Blocking Artery



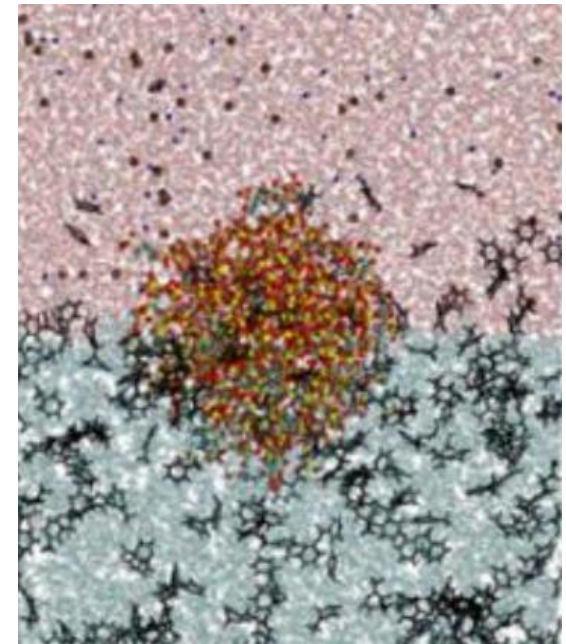
<http://nethealthbook.com/>



How to control interfaces and flow at nanoscale ?

To favor the mobilisation of hydrocarbons trapped at the pore scale through short range surface forces or capillarity:

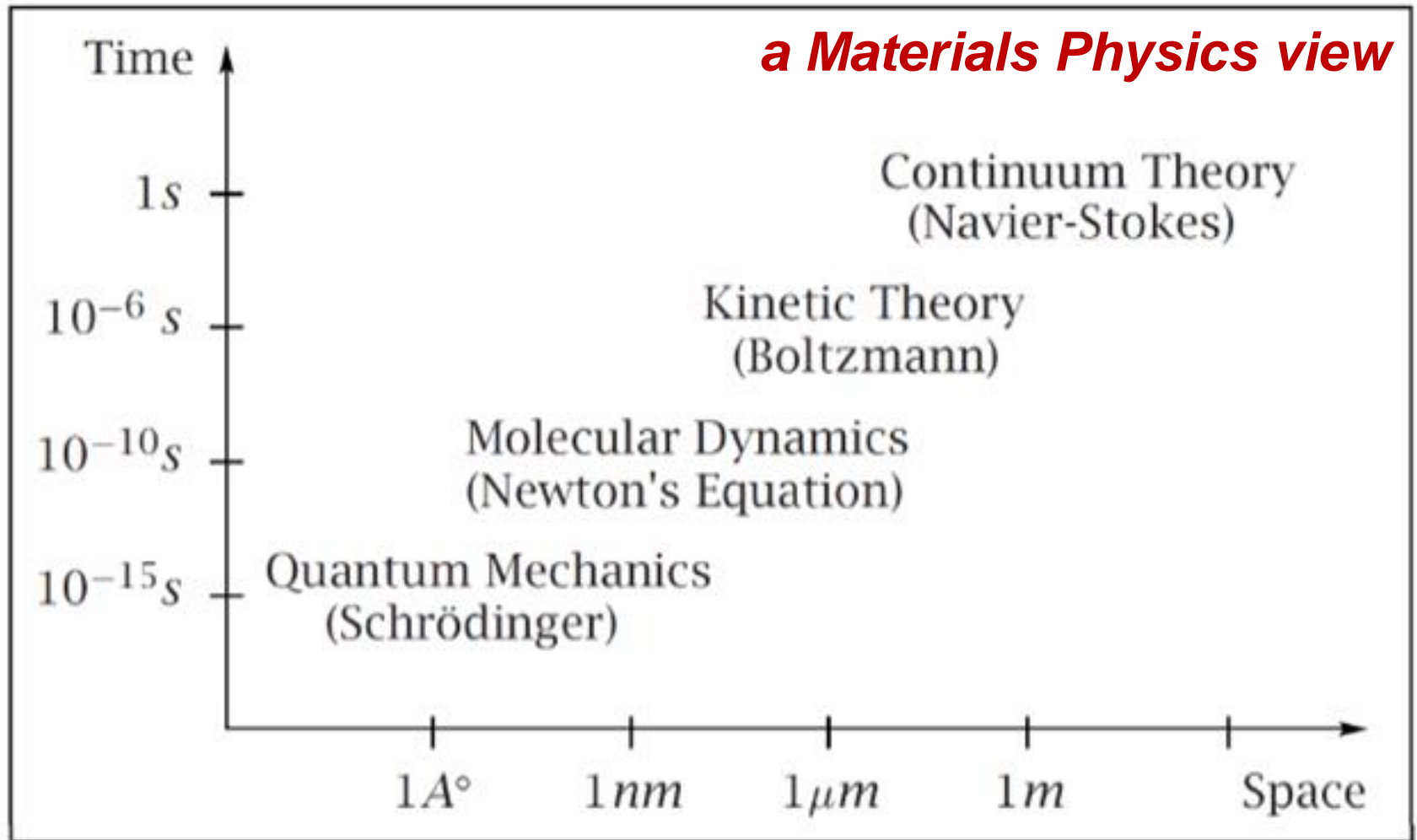
- 1) **Nano-EOR** – control the chemical environment: by functionalized NPs and surfactants “**Wetability modifiers**” - *SURFACE DRIVEN FLOW*
- 2) **Nano-IOR** – control confinement - Fluid flooding in nanoporous. *PRESSURE DRIVEN FLOW*
- 3) **Low salinity EOR** – control the electrostatic environment – “**Smart water**” *ELETROKINETIC DRIVEN FLOW*



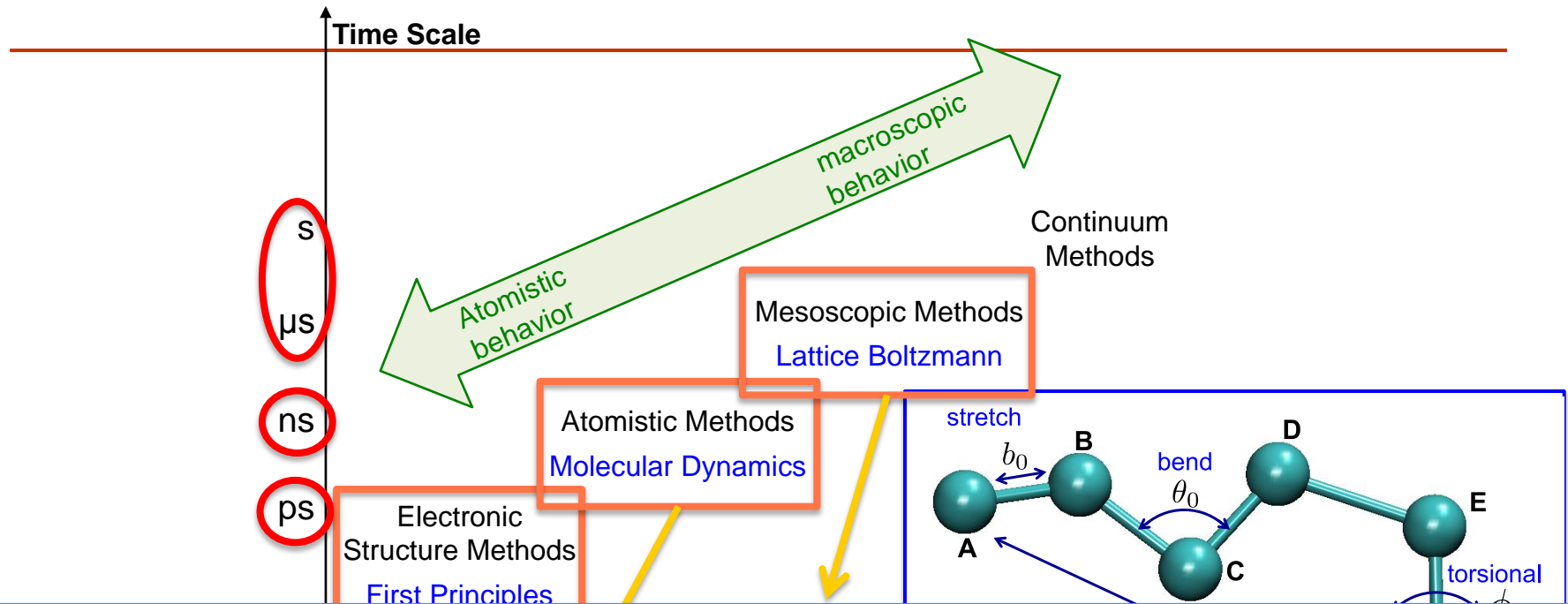
MULTISCALE MOLECULAR SIMULATIONS

(COMPLEX SYSTEMS AND CONTROLLED
CONDITIONS OVER SCALES)

Multiscale in the laws of Physics



Multiscale computational approach



Simplifies the Boltzmann equation by

re • Understanding of fluid behavior at the microscale

s • Phase separation, interface instability, bubble/droplet dynamics and wetting effects

Kinetic Theory

Lattice Boltzmann

$$\mathbf{v} \nabla_{\mathbf{x}} f + \mathbf{F} \nabla_{\mathbf{p}} f + \frac{\partial f}{\partial t} = \Omega$$

$$f_i(\mathbf{x} + \mathbf{e}_i \Delta t, t + \Delta t) = f_i(\mathbf{x}, t) + \Omega_i(\mathbf{x}, t)$$

MD ab initio and classical X LBM: a soccer perspective



Ab initio MD



Classical MD

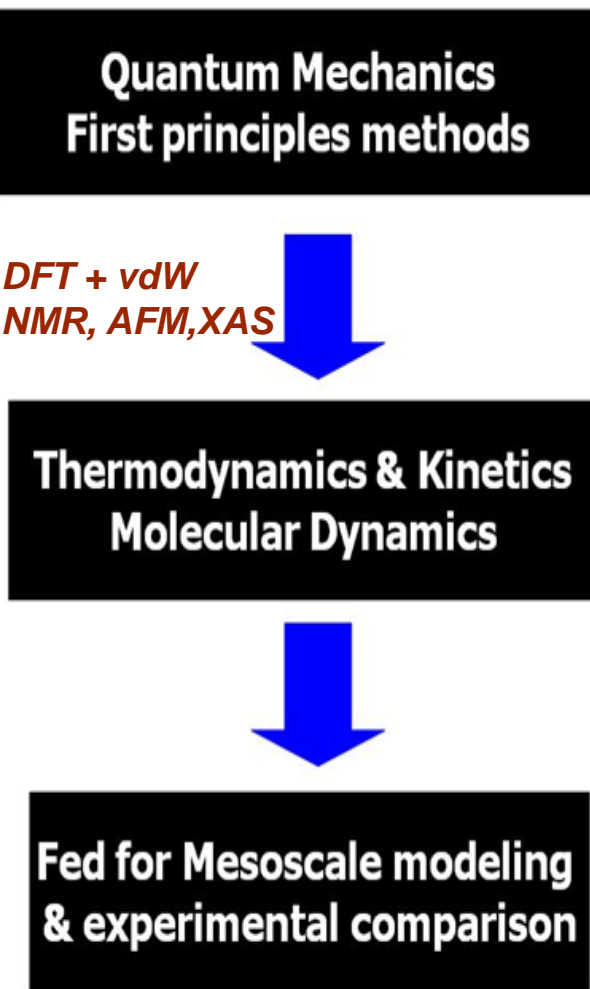


LBM

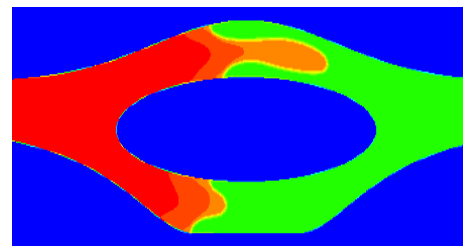
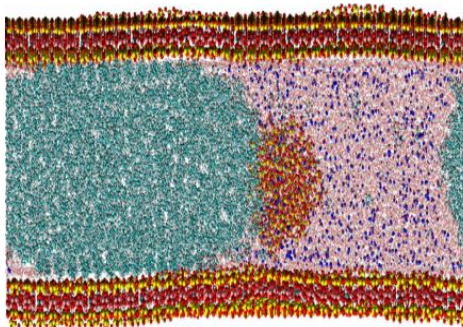
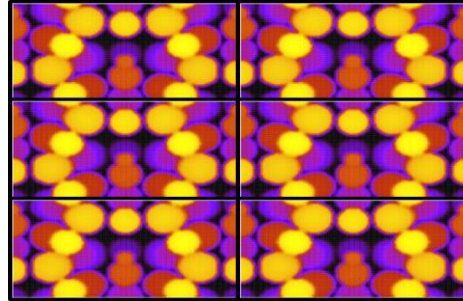
foosball

Simple way to describe
the movement of players
during a game

Multi-scale approach



Lattice Boltzmann



Recent developments (~00s ~10s):

DFT:

- *Dispersion corrections (vdW)*
- *GIPAW (Gauge Including Projector Augmented Waves)*

MD:

- *Development of polarizable dissociative ab initio based interatomic potentials.*
- *Massive parallelization*
- *New hardware (GPUs and hard-disks)*

LBM:

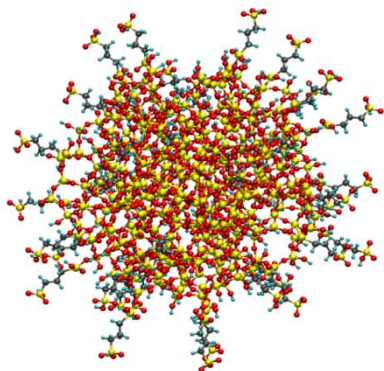
- *Free energy approach*
- *Viscosity differences*
- *Multi-phase systems*
- *Pore-scale flow in porous media*

NANO-EOR

AN INTEGRATED WAY TO APPROACH THE PROBLEM

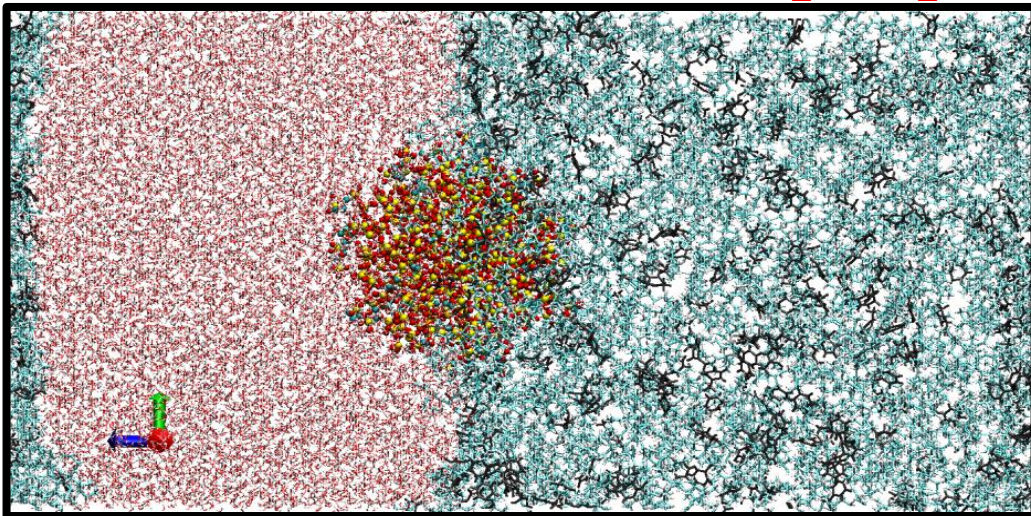
NANO-EOR

SURFACE DRIVEN FLOW



Functionalized Silica nanoparticles

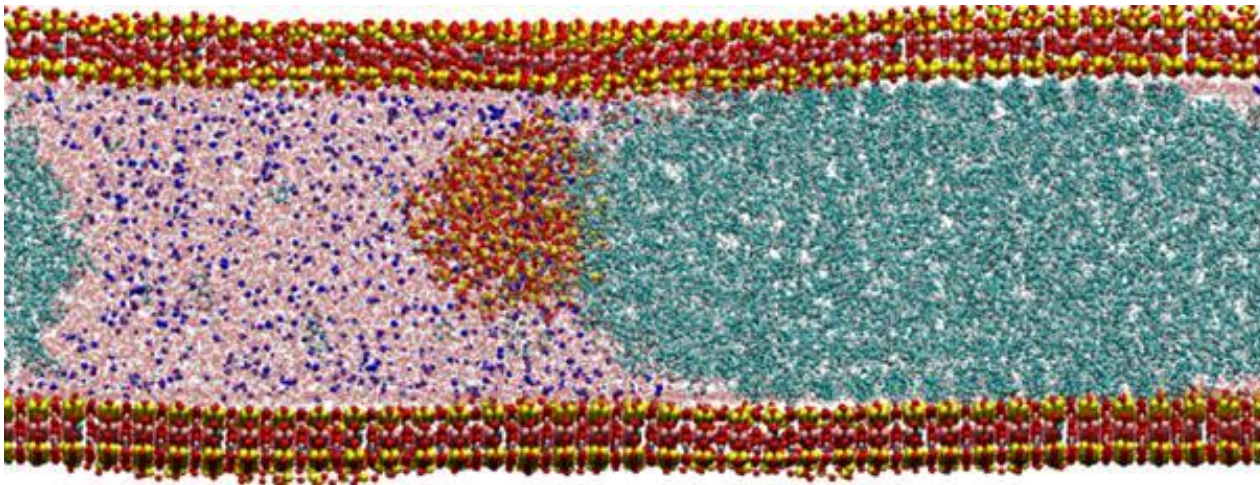
- ❖ Hydroxylated
- ❖ Polyethylene glycol (Hydrophylic)
- ❖ $-\text{CH}_2-\text{CH}_2-\text{Sulfonic acid}$ (Hydrophobic)



Temperature (300, 350, 375 and 400K)
Pressure (1 to 400 atm) [1 – 6000 psi]

NP-clay interaction

- For *montmorillonite* (MMT) and other clay systems, the effect of clay swelling occurs, which can have an impact on the wellbore instability and formation damage.
- Montmorillonite is used in the oil drilling industry as a component of *drilling mud*.

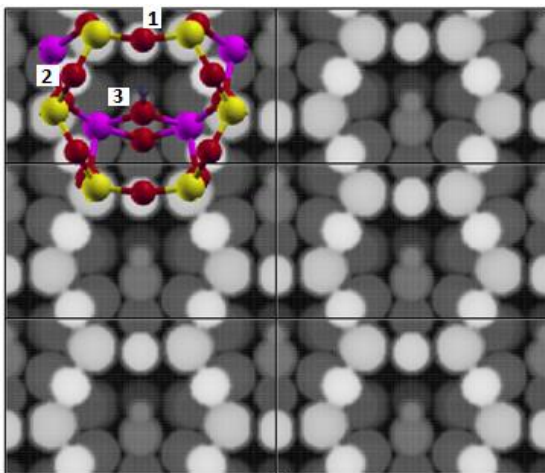


MMT is naturally hydrophilic and it has good affinity with H_2O .

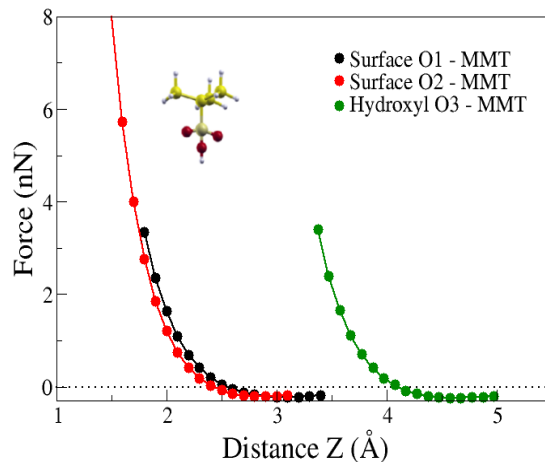
AFM Simulations through the Interaction between Functionalized Silicon Tip and the Montmorillonite (001) Surface – DFT + vdW

Alvim and Miranda, J. Phys. Chem. C, 120, 13503(2016)

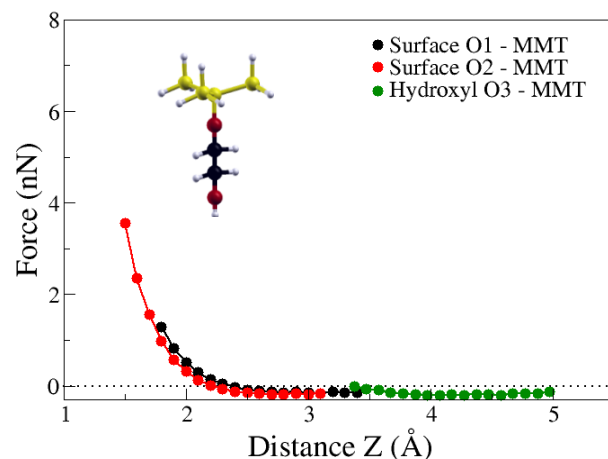
Alvim and Miranda, Phys. Chem. Chem. Phys., 17, 4952 (2015)



Sulfonic Acid (SA) Tip / Montmorillonite (MMT)



Ethylene Glycol (EG) Tip / Montmorillonite (MMT)

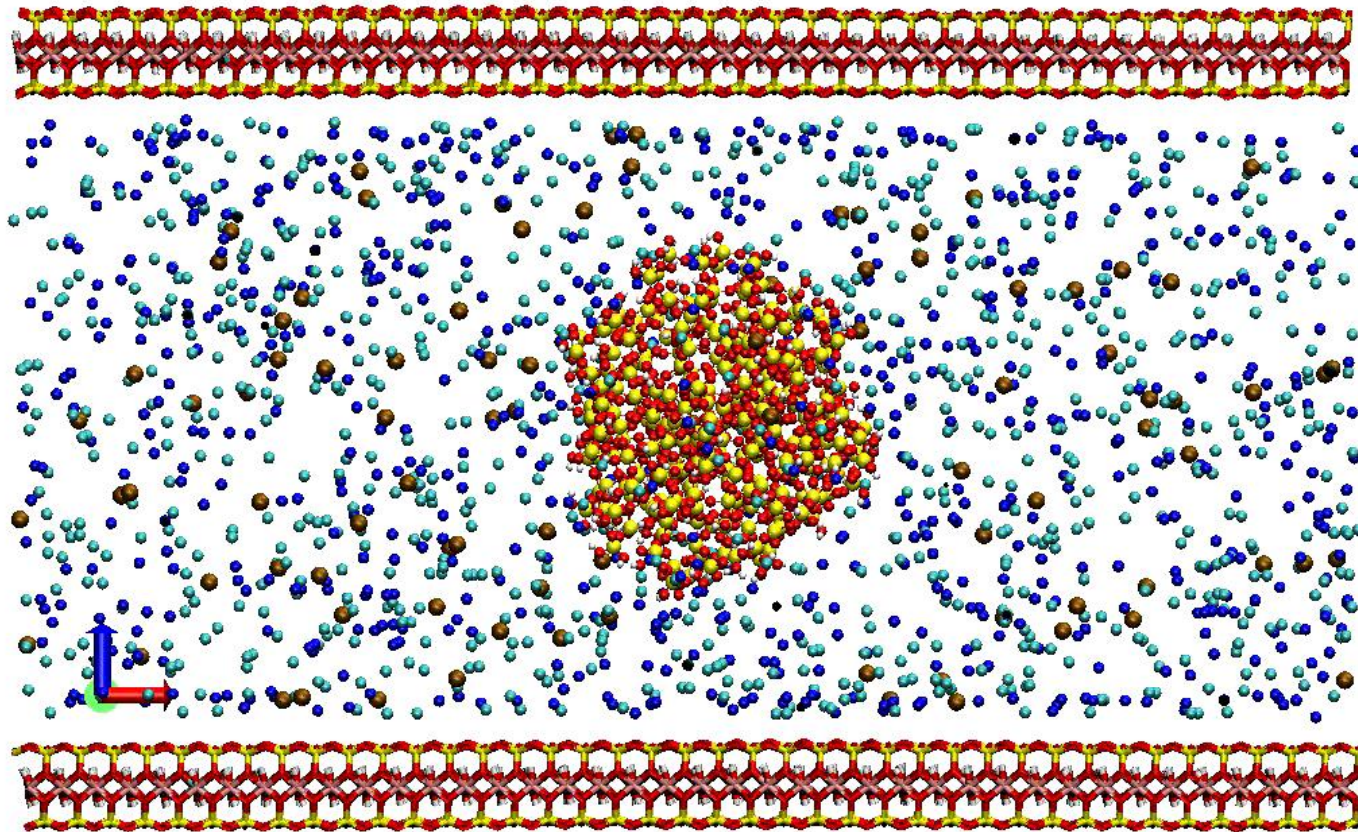


Tip Approach

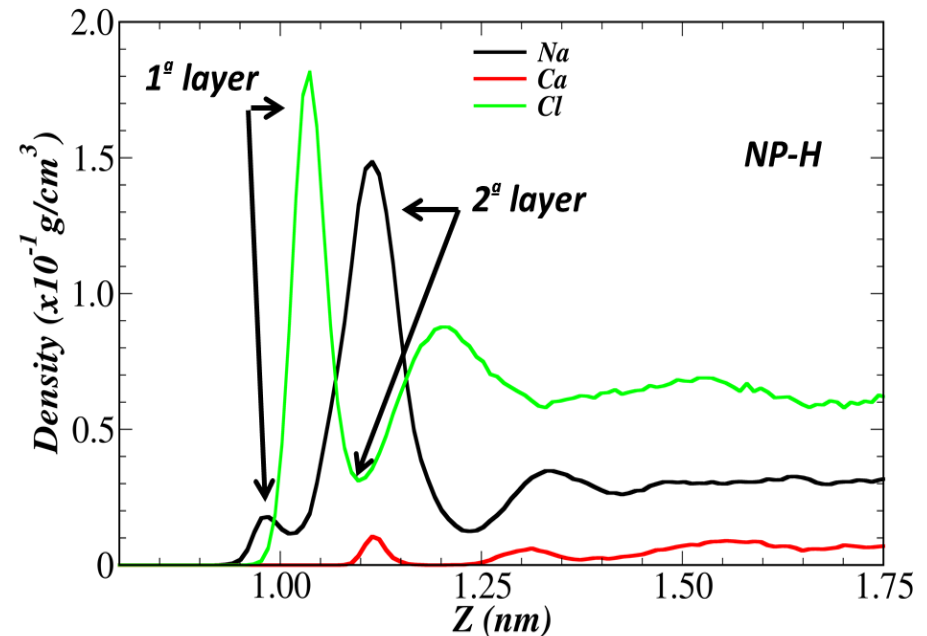
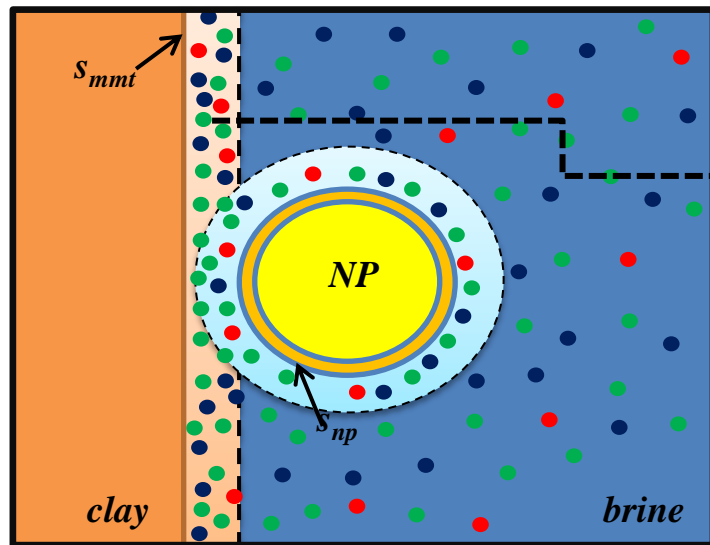
- Strength of bonding
- Surface configuration
- Adsorption density
- Interatomic Forces
- Chemical environment

**HOW TO UPSCALE THIS
VERY FUNDAMENTAL
INFORMATION ?**

Fully atomistic MD (Brine+NP/Oil/MMT)

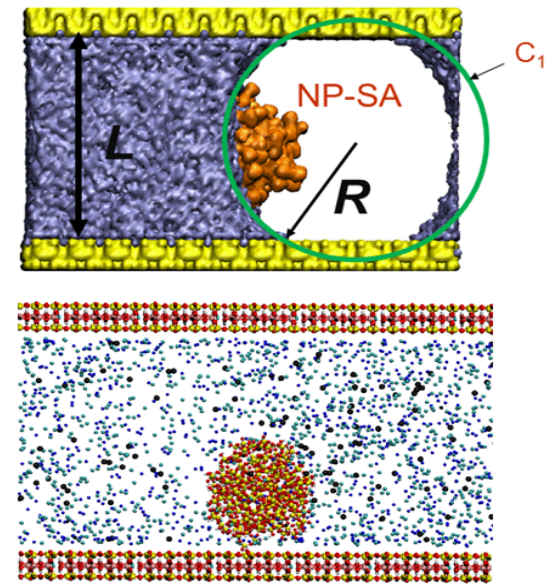
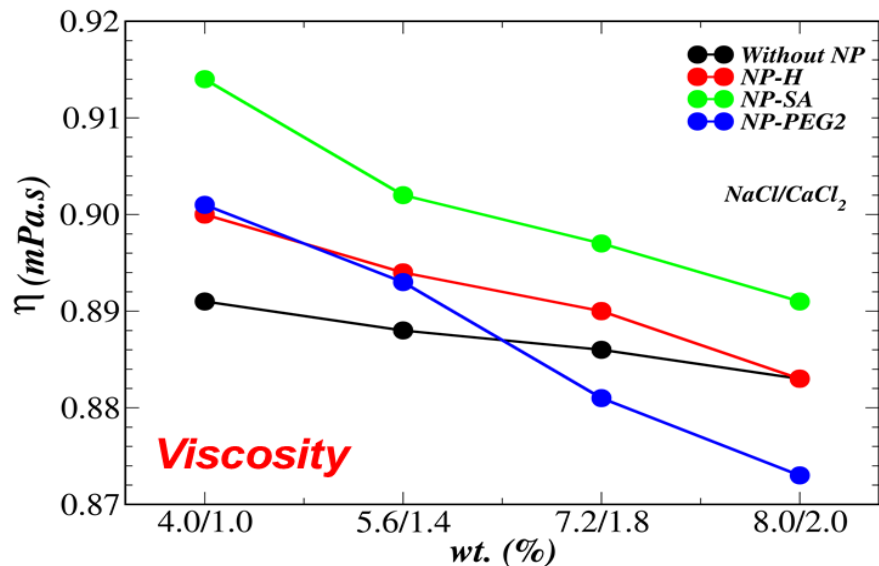
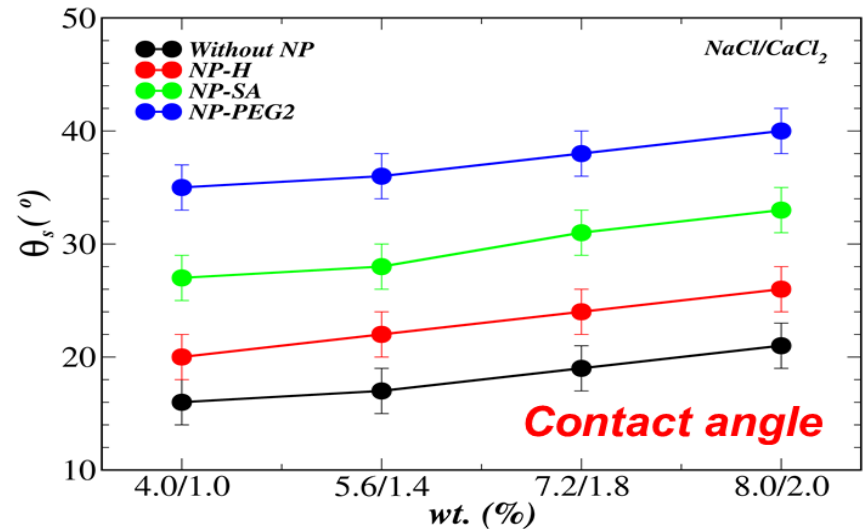
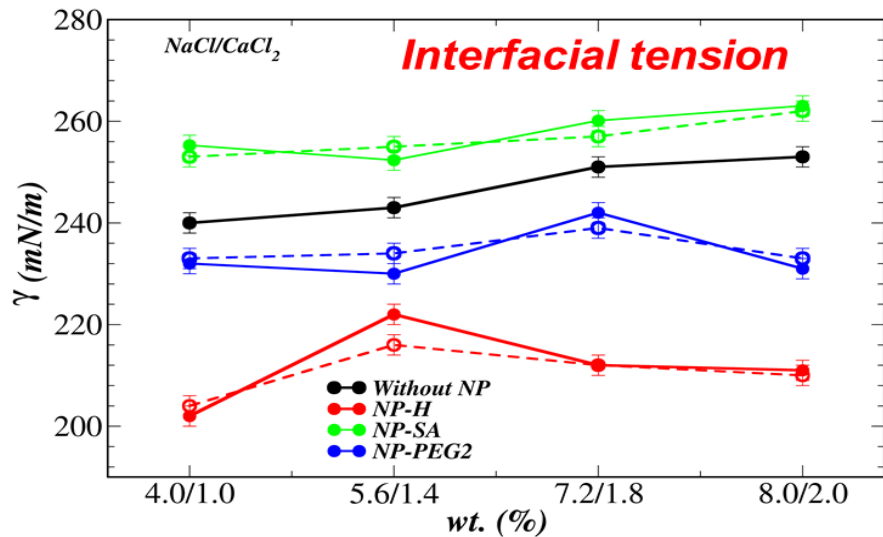


Electrical double layer



- **Electrical Double Layer (EDL) Formation.**
- **For NP adsorbed on MMT – compression of the EDL.**

Interfacial phenomena



Case study:

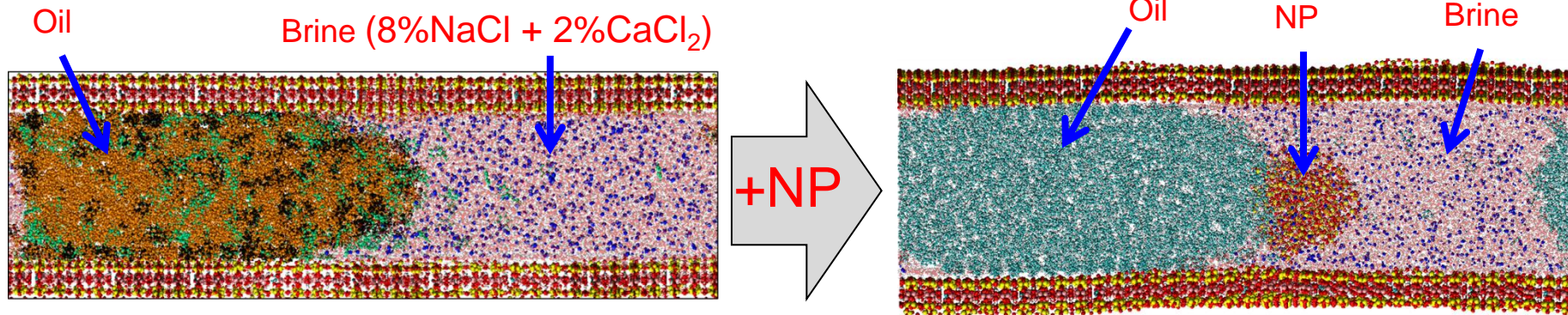
MD : nm ns

LBM : $\mu\text{m}/\text{mm}$ $\mu\text{s}/\text{ms}$

Brine+NP/Oil/MMT System

$T=300\text{K} - 400\text{K}$

$P=1$ to 200atm



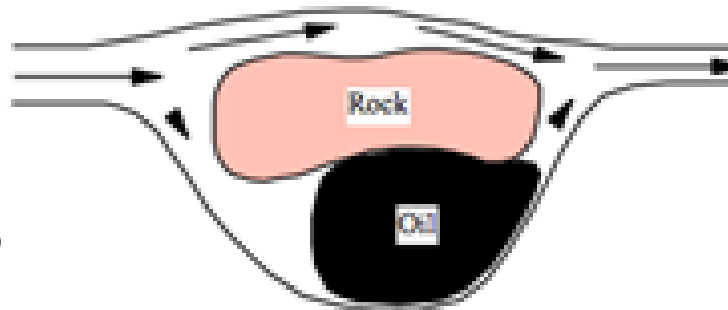
MD Physical properties

$\rho_o=0.81 \text{ g/cm}^3$; $\rho_b=0.96 \text{ g/cm}^3$;
 $\eta_o=3.62 \text{ mPa-s}$; $\eta_b=0.79 \text{ mPa-s}$;
 $\gamma_{ob}=43 \text{ mN/m}$; $\theta_w=28^\circ$

$\rho_o=0.81 \text{ g/cm}^3$; $\rho_b=0.96 \text{ g/cm}^3$;
 $\eta_o=3.60 \text{ mPa-s}$; $\eta_b=0.88 \text{ mPa-s}$;
 $\gamma_{ob}=38 \text{ mN/m}$; $\theta_w=21^\circ$

LBM parameters:

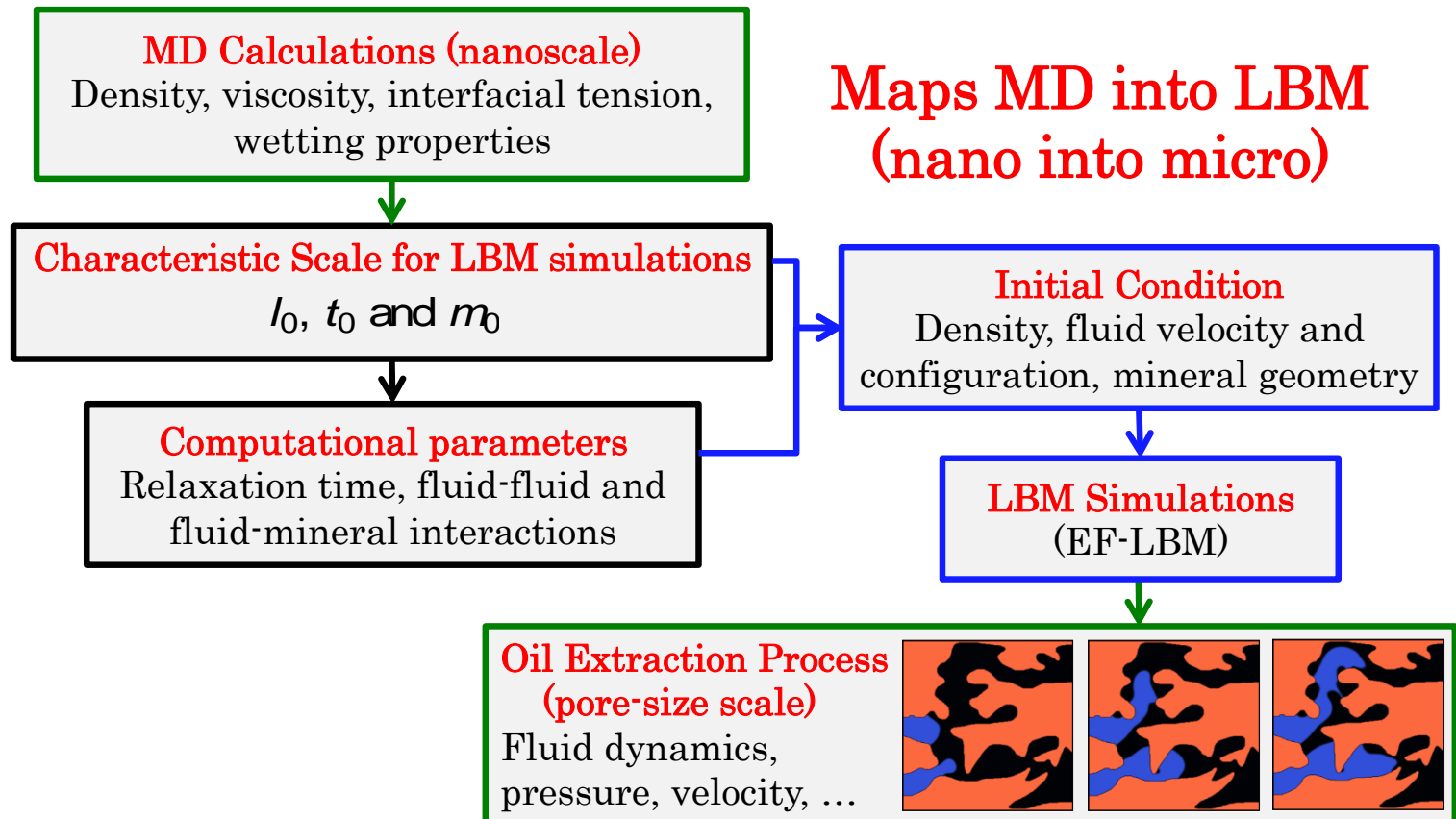
$G=0.14$; $G_w=-0.015$;
 $\tau_{oil}=1.50$; $\tau_{brine}=0.70$



$G=0.15$; $G_w=-0.02$;
 $\tau_{oil}=1.50$; $\tau_{brine}=0.75$

Hierarchical Computational Protocol: Molecular Dynamics + LBM

Versatile tool to investigate the potentialities of modified injection
fluids for EOR techniques



MMT Rock Model

<http://www.log.furg.br/morelock/clnmarsd.htm>

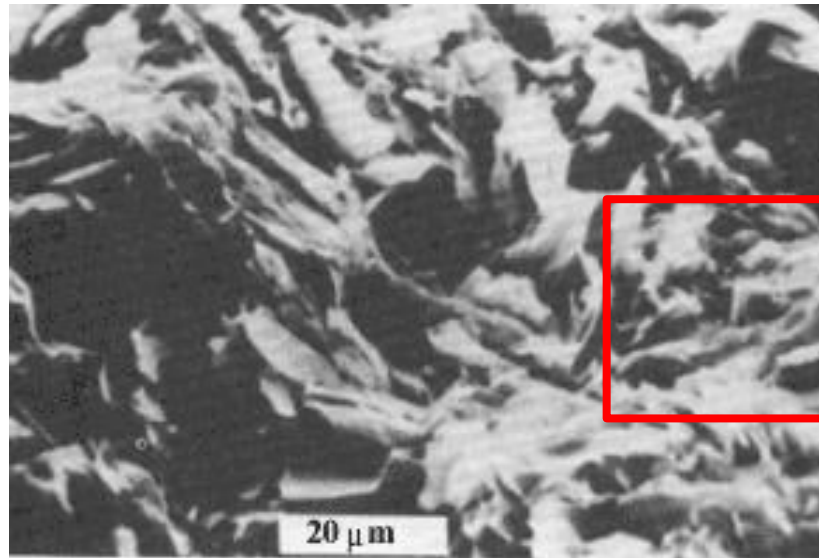
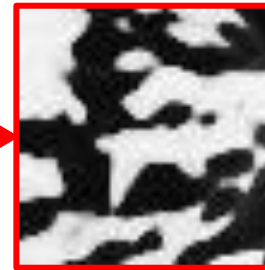


Image treatment



Computational
Rock Model



■ rock
□ pore space

LBM Parameters

	G_{12}	G_w
Without NP	0.190	0.078
NP-H	0.181	0.095
NP-SA	0.171	0.099
NP-PEG2	0.164	0.098

Characteristic Scale

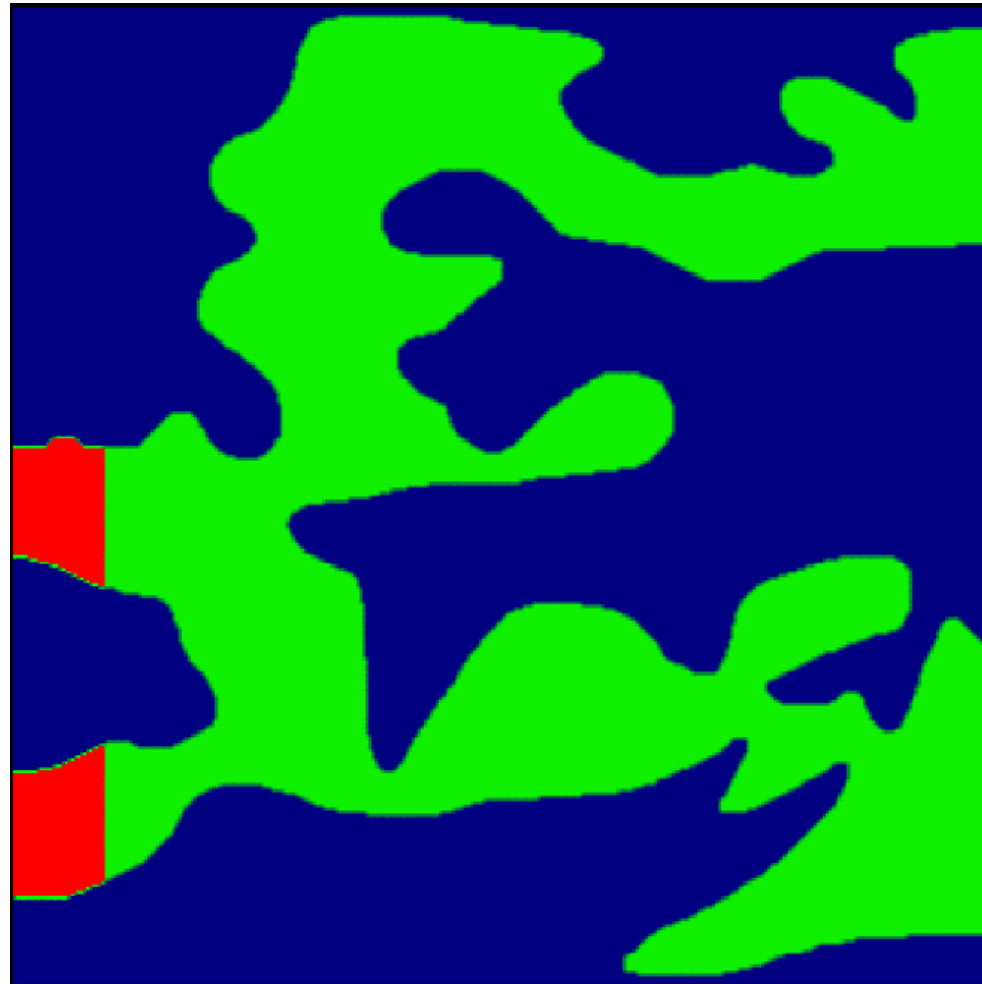
$$l_0 = 5.49 \times 10^{-5} \text{ m}$$

$$t_0 = 1.27 \times 10^{-4} \text{ s}$$

$$m_0 = 1.50 \times 10^{-10} \text{ kg}$$

Exploring Oil Extraction by Nanofluids in Clay Coated Pore Network Models

Oil displacement by Brine+NP-PEG2: First Injection



$$C_a = 1.2 \times 10^{-2}$$



Rock



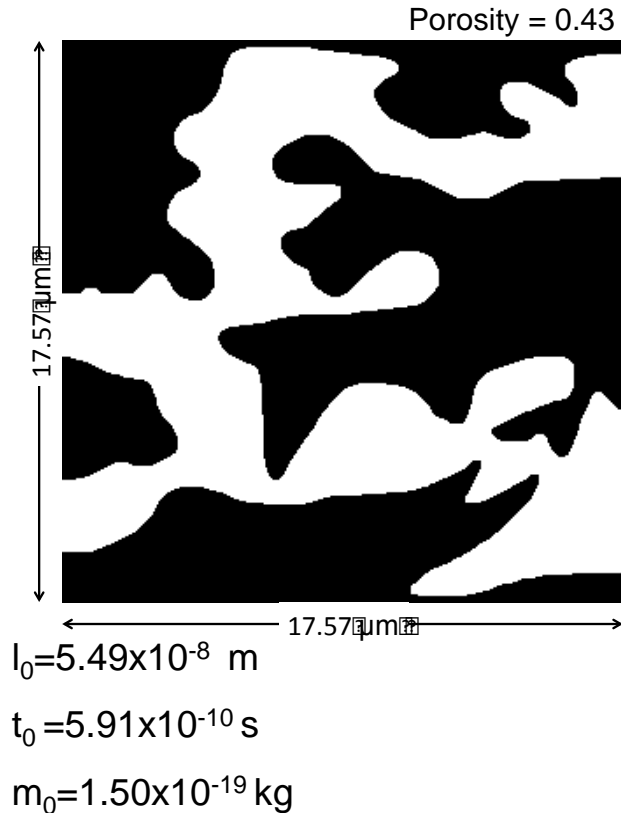
Brine



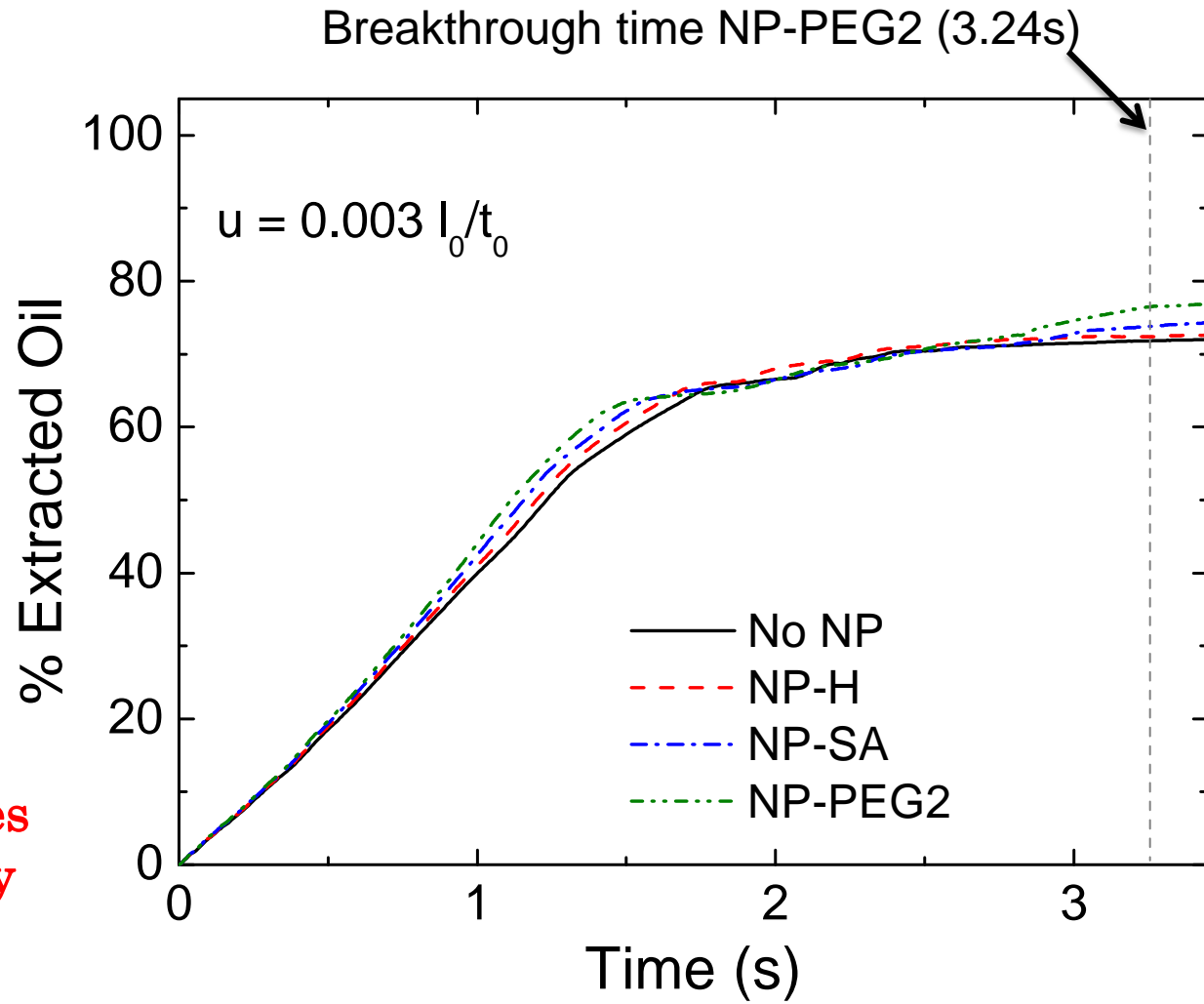
Oil

Exploring Oil Extraction by Nanofluids in Clay Coated Pore Network Models

LBM Simulations: Oil displacement at the pore-size scale



**Addition of nanoparticles
improve the oil recovery
process**



Core-shell NP combinatorial exploration for EOR applications (on going)

Highthrough computational searching for Functionalized Nanoparticles with Surfactants

- ◆ pH
- ◆ Temperature
- ◆ Salt concentration
- ◆ Ionic Strength
- ◆ Functional group size

Database over NPs and surfactants
Stability and functionality

- ✓ Anionics
- ✓ Cationics
- ✓ Nonionics
- ✓ Zwitterionics
- ✓ Mixtures



- Stability
- Adsorption
- Surface tension
- Viscosity



Optimized functionalized NP



Summary – NanoEOR for NPs/brine/oil/clay interfaces

- ✓ Surface characterization of Geological Materials by first principles (PDOS, AFM, XAS and NMR).
- ✓ Extensive MD for NP interacting with Clays/brine/oil.
- ✓ Adsorption and Swelling studies of NP on clay systems.
- ✓ Integrated FP, MD and LBM method.
- ✓ Cost effective way to search for NPs for EOR applications.

NANO-IOR

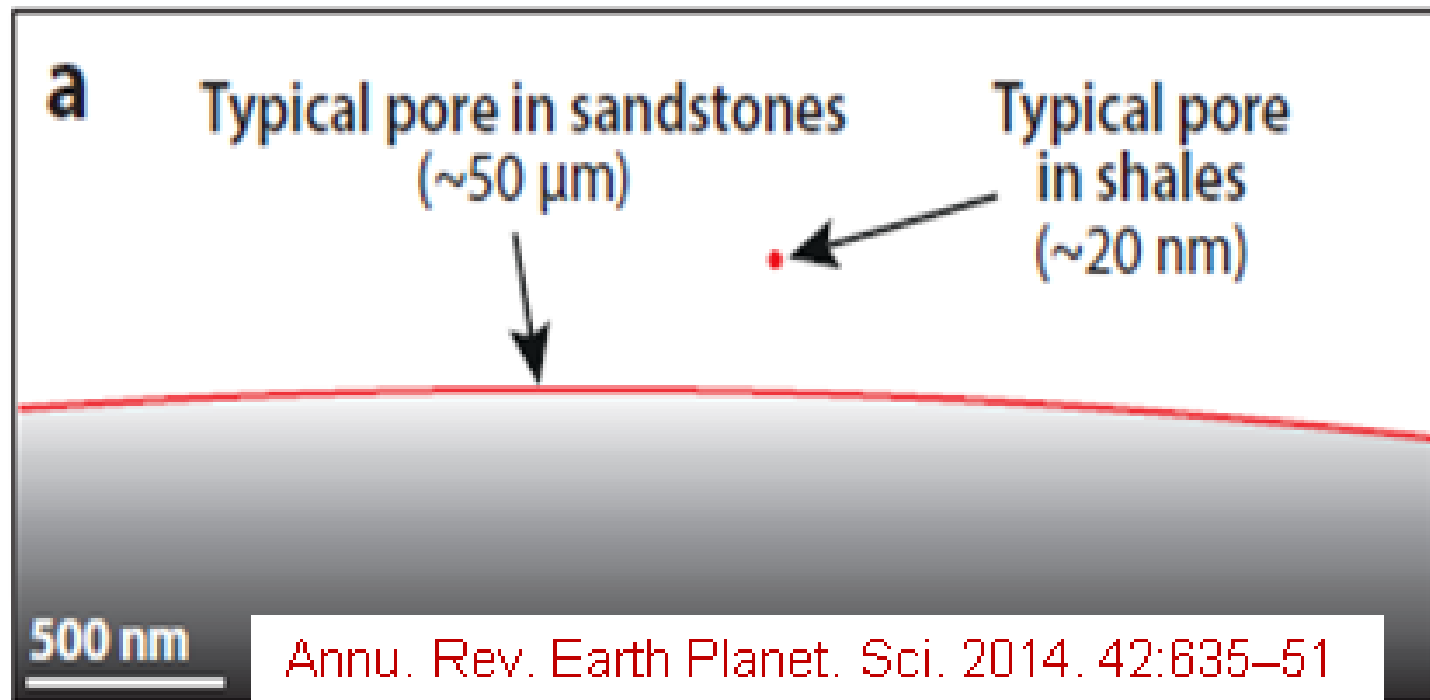
PRESSURE DRIVEN FLOW

WHERE THE CONTINUUM APPROACH MAY FAIL

NANO-IOR

PRESSURE DRIVEN FLOW

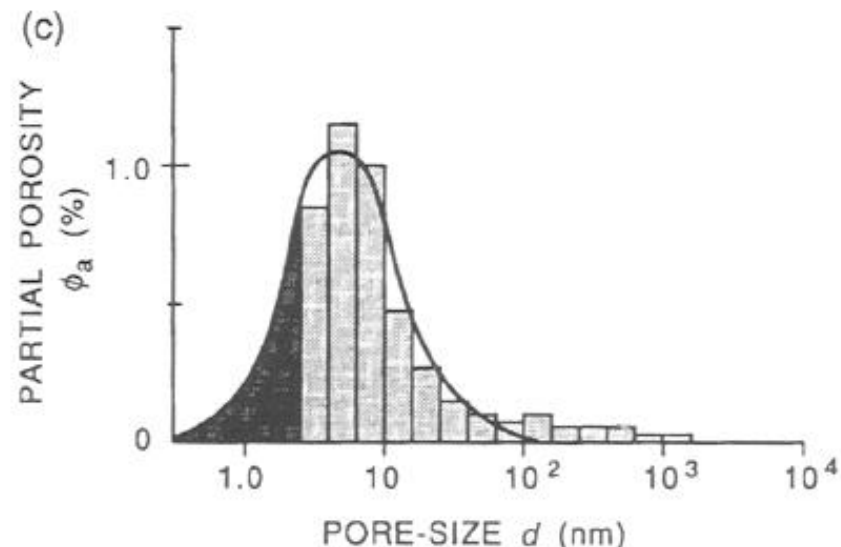
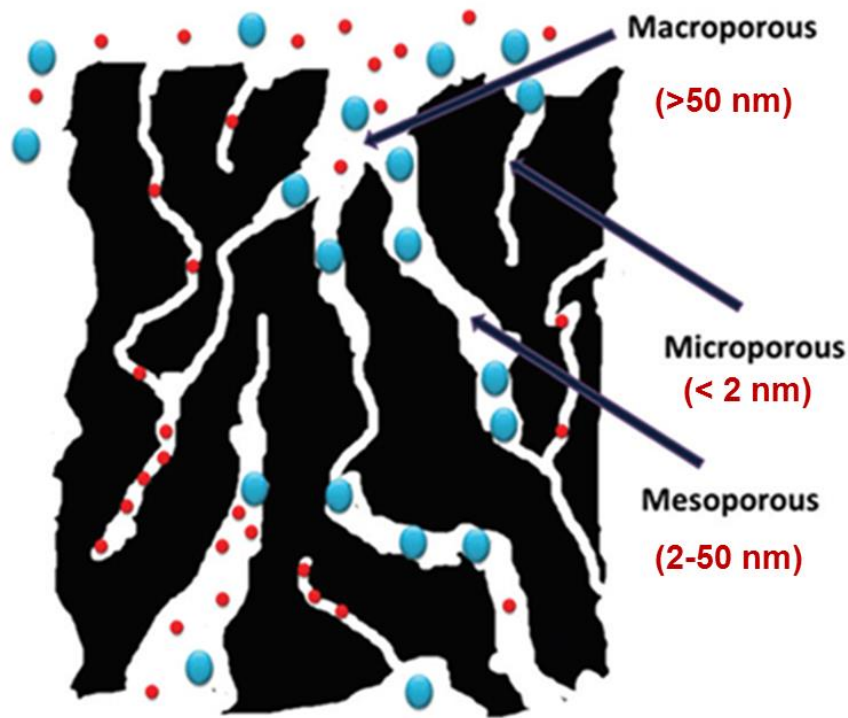
- Explore the water and oil flow through silica nanopores to:
 - a) Model the displacement of water and oil through a nanopore to mimic the fluid infiltration on geological porous media.
 - b) Simulate the process of water flooding to emulate a Nano-IOR process.



Fluid flow through nanoporous

Fluid flow through mineral porous occurs in underground aquifers, oil and shale gas reservoirs.

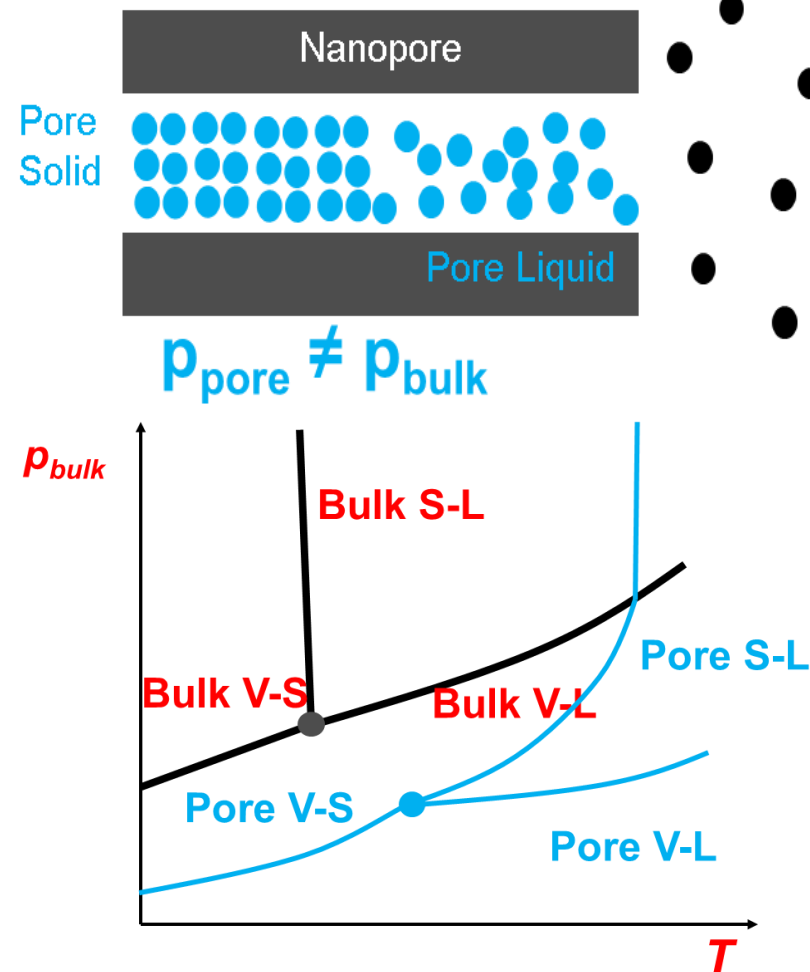
- “Invisible pores”
- Large % of porosity and surface area.
- Interconnects larger porous
- Control the permeability



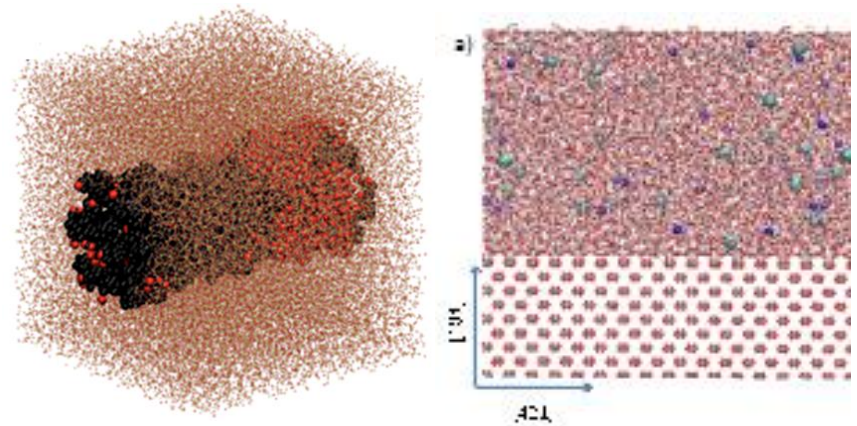
Fluids confined at nanoporous

Bulk Vapor
(unsaturated), p_{bulk}

- Under confinement, new phenomena can emerge, as **new phase transitions** and **layering near the interface**.
- At nanoscale, the **continuum models** for fluids **may not** work.
- Use of an **atomistic description is needed**.



SILICATES AND CARBONATES



Methodology

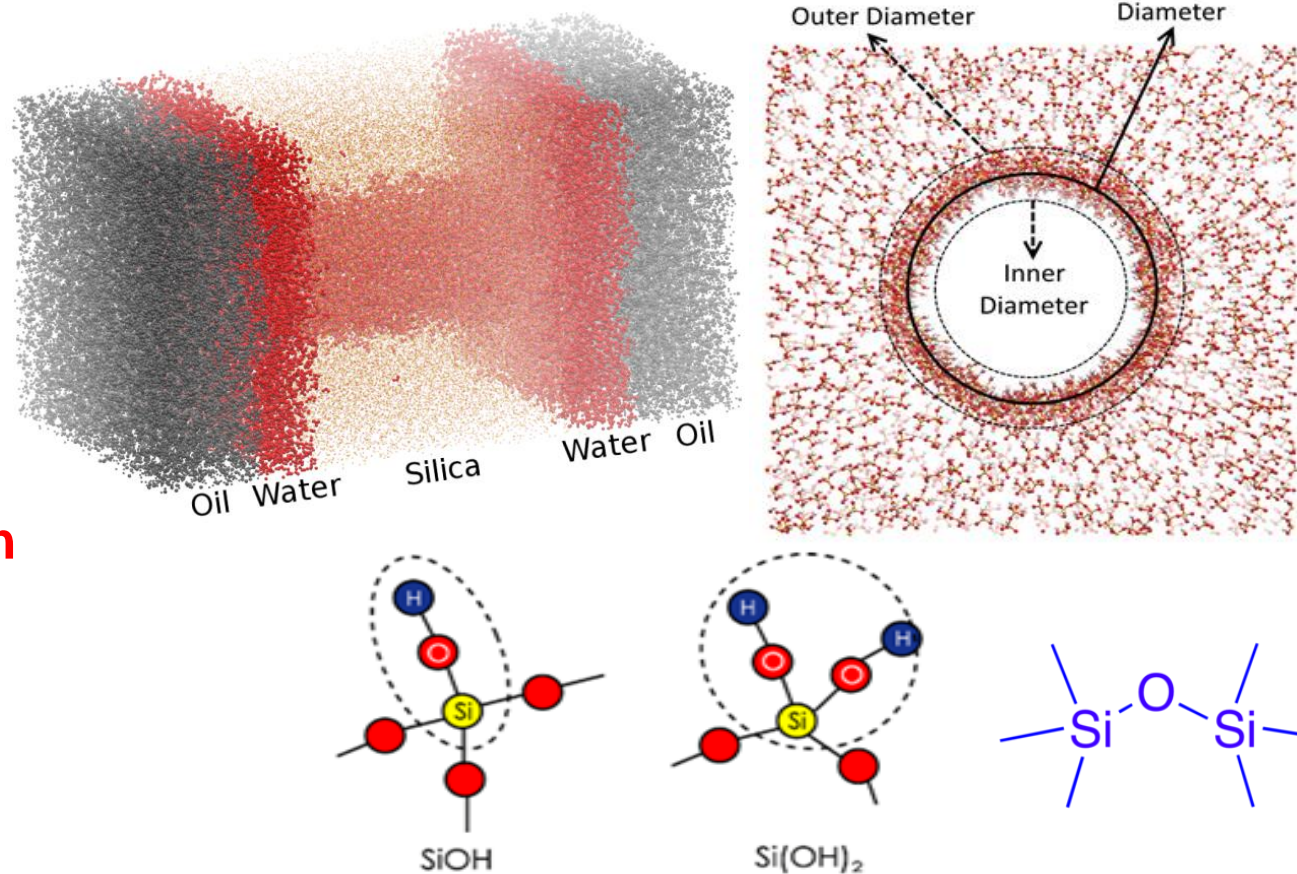
- Classical molecular dynamics (MD) simulations (LAMMPS) (over 10 ns)
- Well tested interatomic potentials : Cruz-Chu (Silica) , CHARMM (hydrocarbons) and SPCE/FH (water) with the Lorentz-Berthelot combining rules
- Realistic conditions of oil reservoirs (*300 K and 200 atm*)
- *Multicomponent oil* (light oil with alkanes and aromatic molecules)
- Induced flow process by applying an external force applied to the atoms (mimic a *pressure gradient*)

Key questions

- Would fluids infiltrate in nanoporous media (Silica) ?
- How much oil is in nanoporous ?
- At which conditions will oil infiltrate ?
- How do extract the oil ? Nano IOR

Nano IOR - Fluid infiltration in silicate nanoporous through MD

Diameters
1 nm
2 nm
3 nm
4 nm



Nanoporous filled with

- 1) Water
- 2) Oil
- 3) Oil after water
- 4) Water re-injection

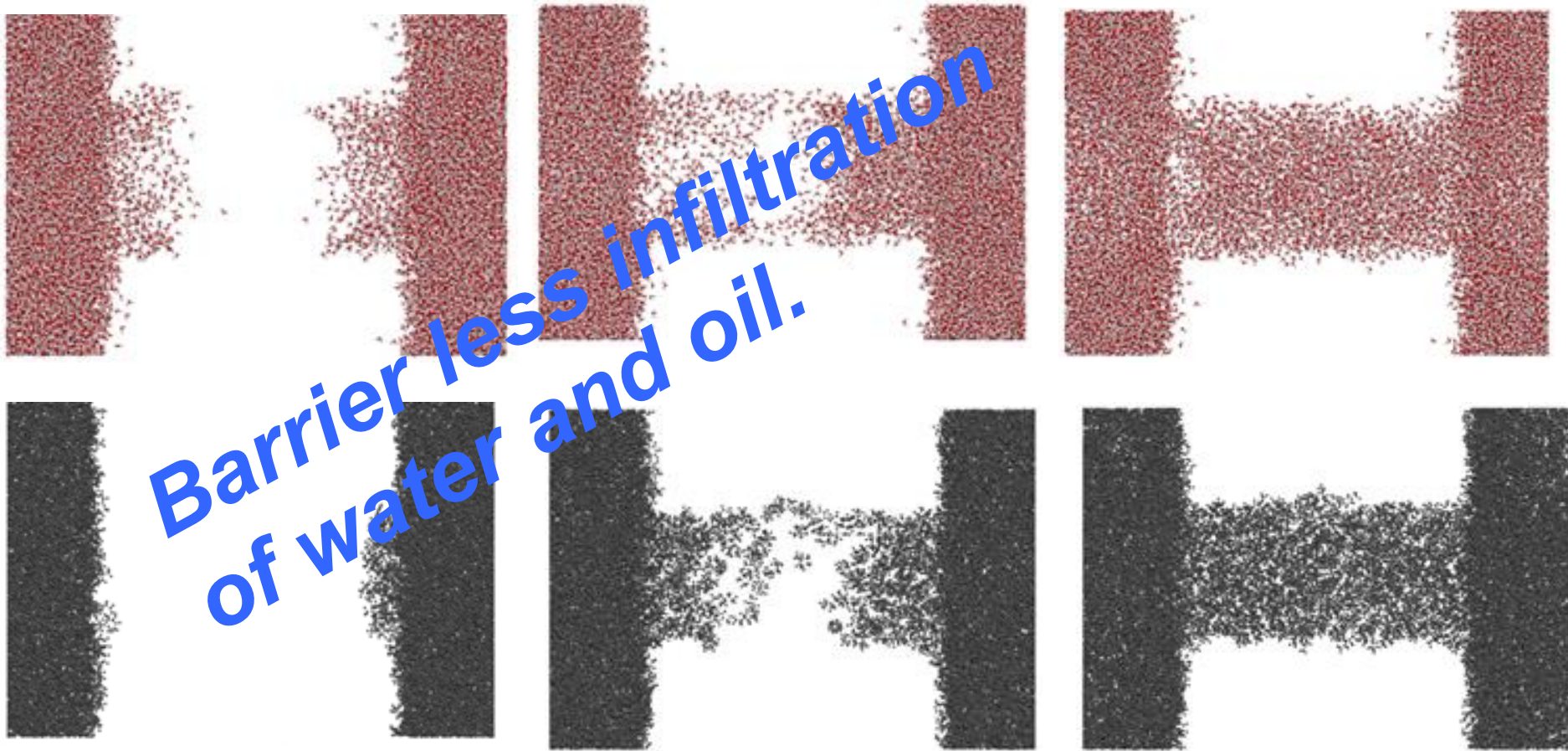
Label	Diameter (Å)	SiOH terminations	Si(OH) ₂ terminations	SiOSi terminations
SiOH-Rich	39.42	6.65%	44.50%	48.86%
SiOSi-Rich	36.45	30.76%	4.60%	65.71%

Higher hydrophilicity

Lower hydrophilicity

Do water and oil infiltrate on silica nanoporous ?

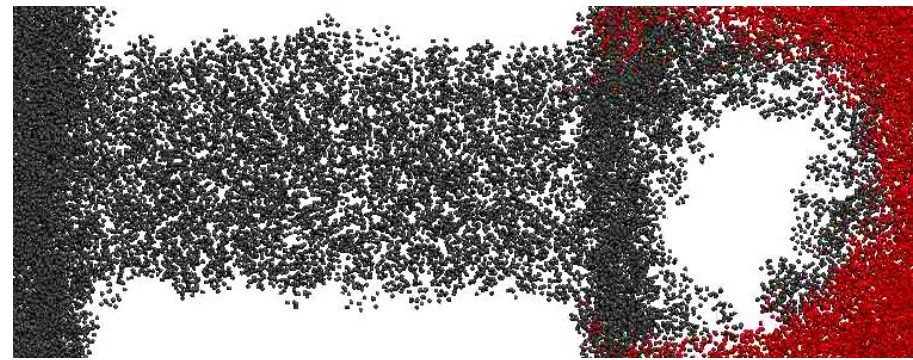
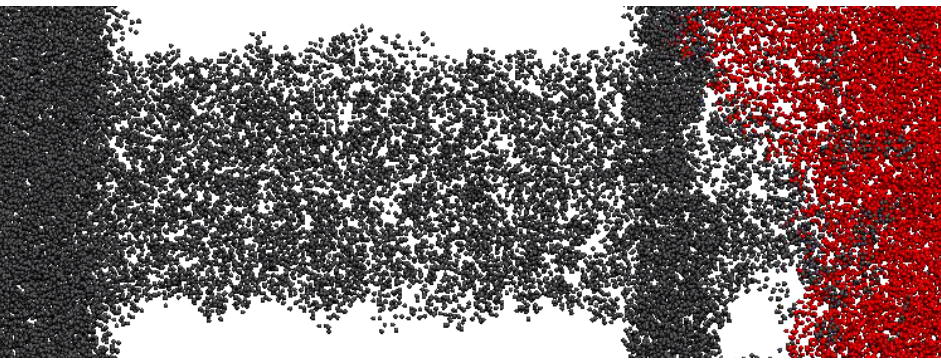
Empty nanopores (4nm) with water or oil adjacent reservoirs were simulated.



Both water and oil infiltrated quickly (less than 1 ns) on the nanopores.

Model I - Water and oil filling in nanoporous silicates – (without previous contact with water)

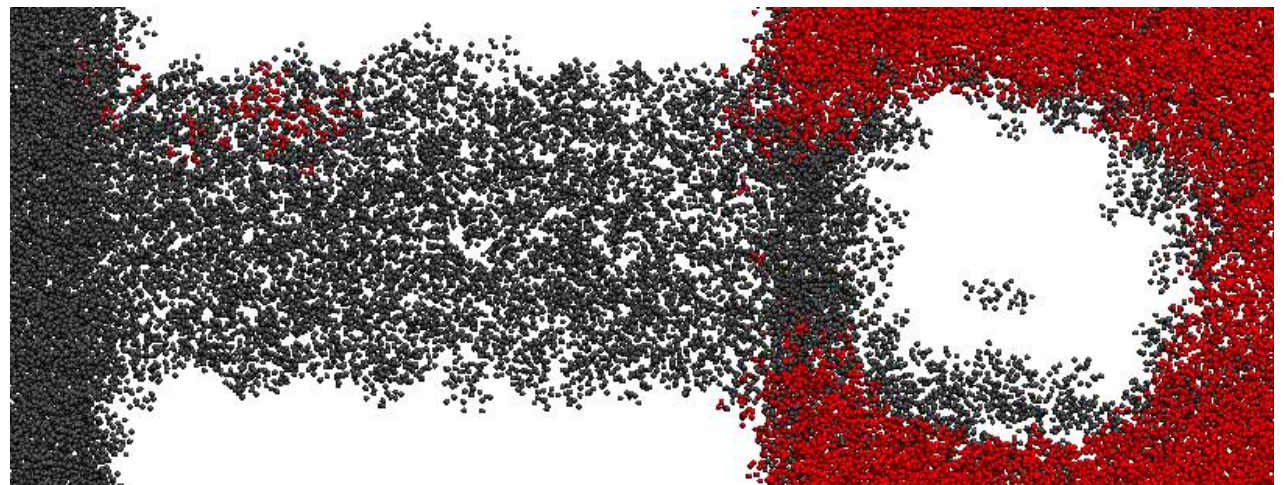
- Water flooding - 4nm (hydrophilic)
 - Oil filled pore, with no water monolayer.



Up to 2500 atm :

No water infiltration.

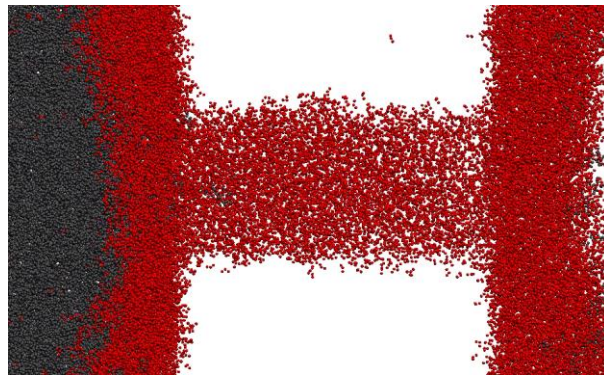
Just a few molecules enter the nanopore.



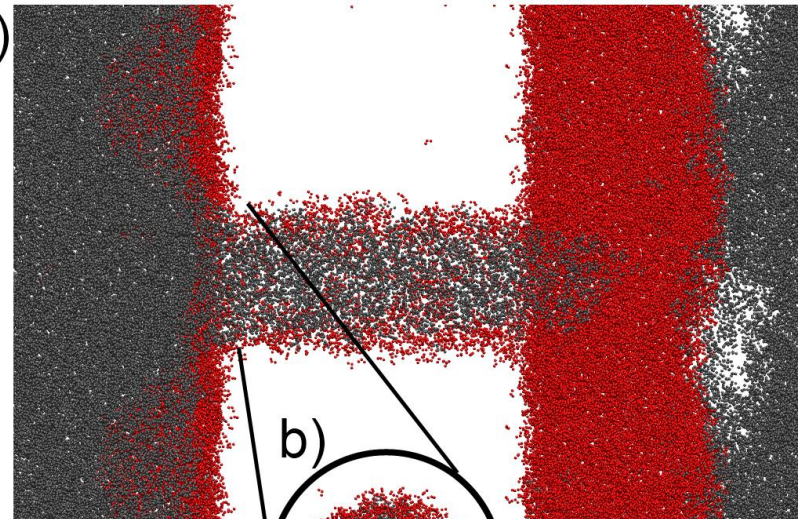
Model II (with previous contact with water)

Oil displaces water. (geological formation)

Force Applied

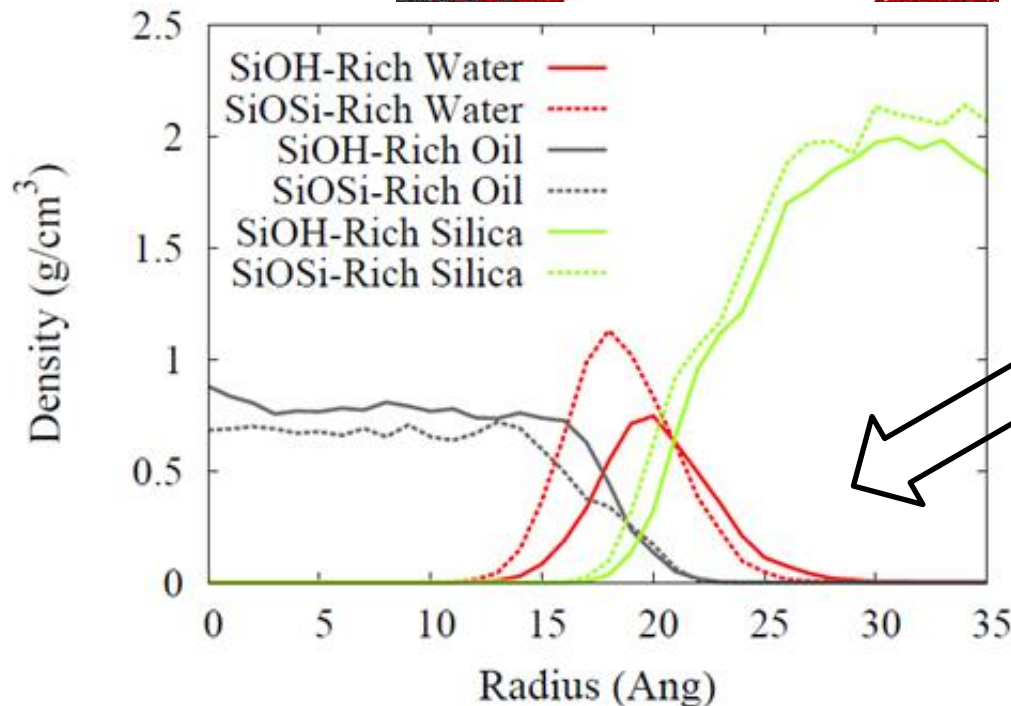
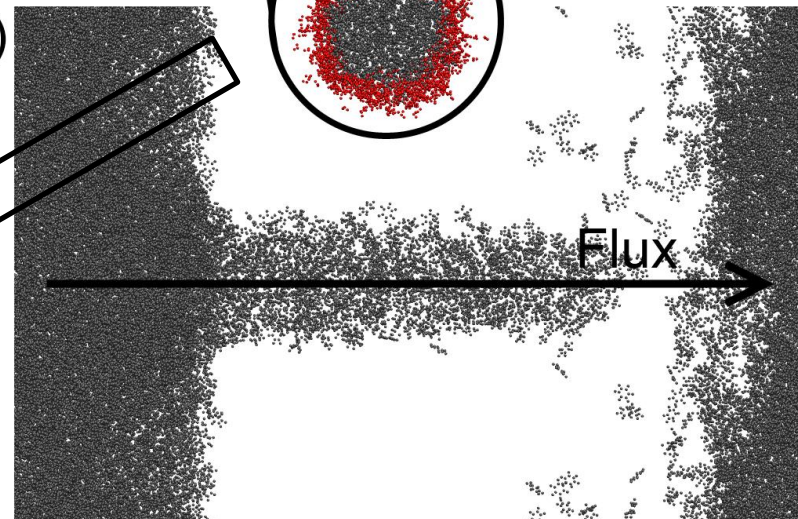


a)



b)

c)



Oil infiltration observed only for pressures above 600 atm.

NanoIOR

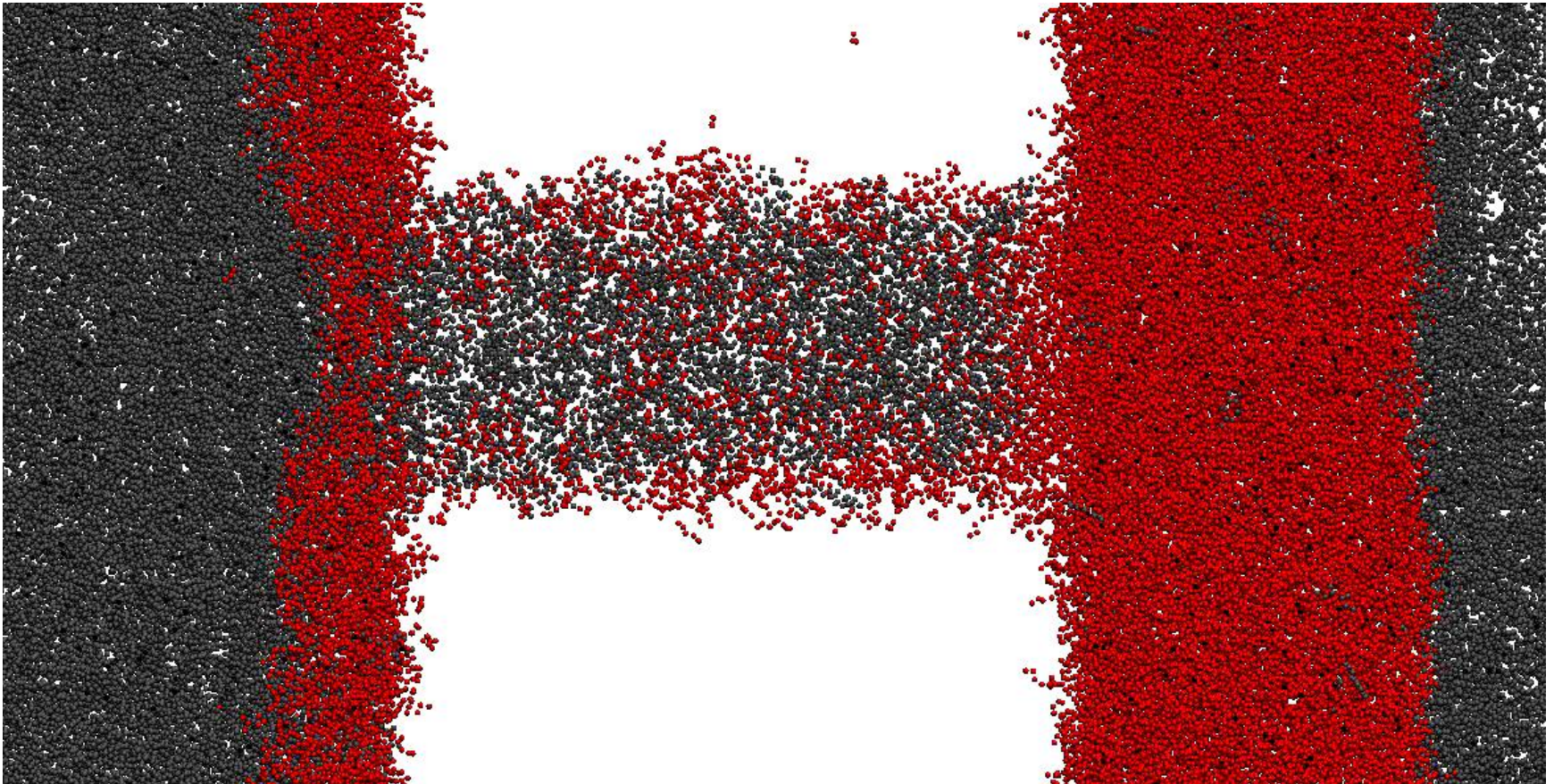
- ▣ For 4nm porous, oil infiltrates above 600 atm.
- ▣ For 1000 and 1500 atm, a dripping effect at the exit end of the nanochannel is observed.
- ▣ For 2500 atm, a steady flux with no dripping occurs.

Can we take it out ?

Nano IOR – Oil filling in nanoporous silicates

Model II (with previous contact with water)

- Water flowing back 4nm hydrophilic, 2500 atm.



Water infiltration observed for pressures as low as 10 atm.

Key questions

- Would fluids infiltrate in nanoporous media (Silica) ?

YES – no barrier

- How much oil is in nanoporous ?

Considerable, with water thin film adsorbed

- At which conditions will oil infiltrate ?

Above 600 atm

- How do extract the oil ? Nano IOR

YES, only if water is adsorbed.

Summary - Nano-IOR in nanoporous silicates

- Modeling of nanoporous silicates (hydrophobic to hydrophilic) from 1 to 4 nm.
- MD simulations used to determine the wettability and contact angle between injected fluids (brine and CO₂) with light oil (not shown).
- Plethora of dynamics in nanoporous media observed (Cavitation, bubble formation, fluid flow)



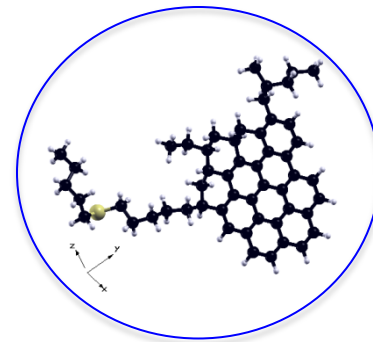
**NANOAGGREGATION OF
COMPLEX MOLECULES
ASPHALTENES
*WHERE MD IS NOT ENOUGH***

“Asphaltenes: the petroleum cholesterol”

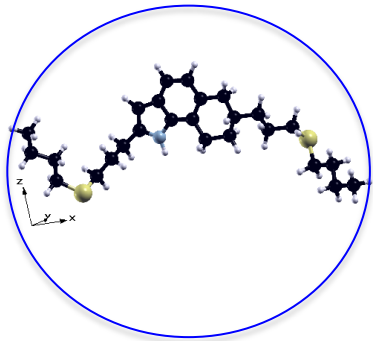
- Asphaltenes are the most polar and surface-active fraction of the oil that is insoluble in n-alkanes, but soluble in aromatic solvents.
- They can precipitate, aggregate and deposit on wells, formations, pipelines and surface facilities
- They play a key role in the Oil industry chain from oil E&P to refining processes



Crude Oil



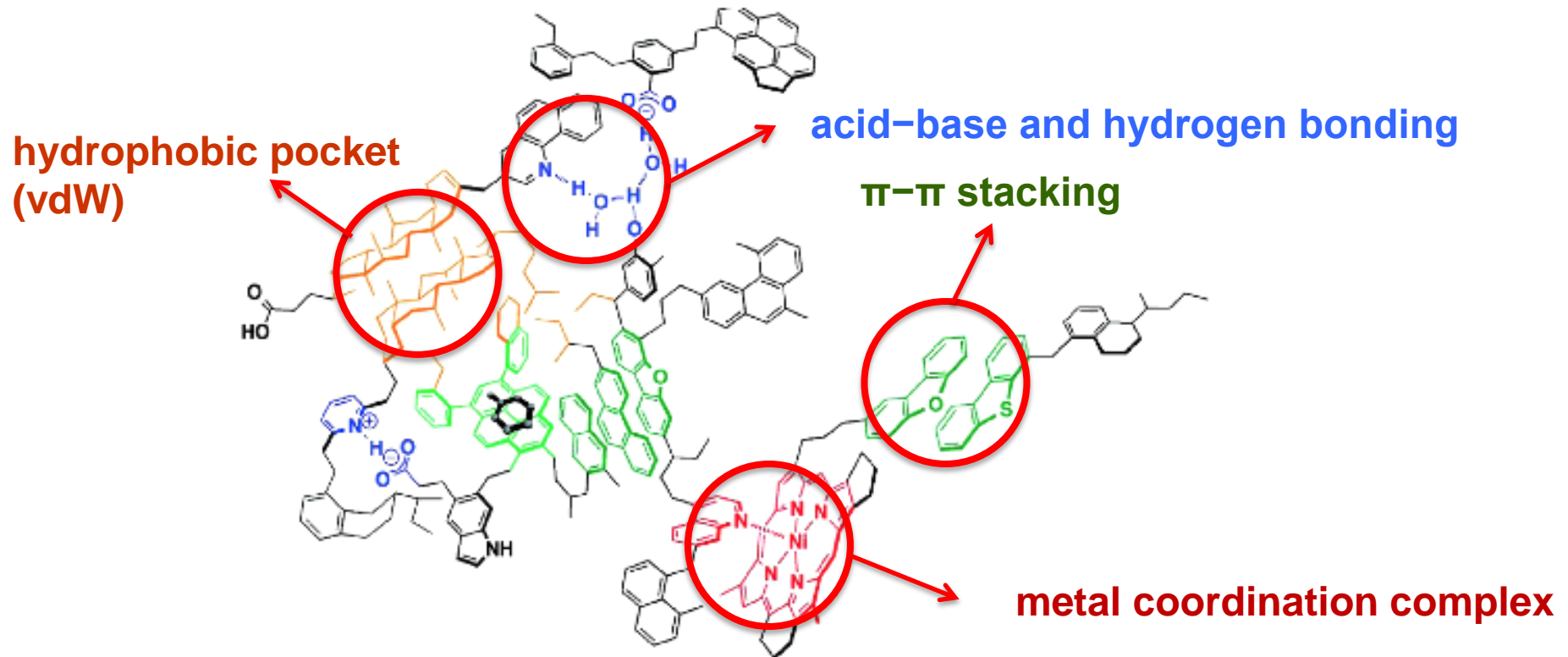
Asphaltene



Resin

Asphaltene and resin interactions

- Tendency to nanoaggregate, clustering and adsorb at solid surfaces
- Very rich and complex chemistry



Summary - nanoaggregation

- The proposed *nanoaggregation mechanism* is based on the competition of the π - orbitals, leading to degeneracy broken of the electronic states with an increasing of the displacement of the HOMO orbital towards the center of the nanoaggregate.
- The growth of the nanoaggregate is further limited given their charge rearrangement, which leads to a dipole moment decreasing.
- These findings can guide new methods for asphaltene stability control.

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