## Luminescent hydrogels: self-assembly of a tripeptide and carbon nanodots M. C. Cringoli<sup>1</sup>, S. Kralj<sup>1,2</sup>, M. Kurbasic<sup>1</sup>, M. Urban<sup>1</sup> and S. Marchesan<sup>1</sup>

<sup>1</sup> University of Trieste, Department of Chemical and Pharmaceutical Sc., 34127 – Trieste <sup>2</sup> Jožef Stefan Institute, Department for Materials Synthesis, 1000 – Ljubljana

Tripeptides are versatile minimal building blocks of supramolecular functional materials, including biocompatible and biodegradable hydrogels. These biomaterials can be used for diverse applications, including regenerative medicine, drug delivery and biocatalysis.[1] Amongst the approaches to design self-assembling motifs based on diphenylalanine [2], the introduction of D-amino acid(s) in D,L-peptides is an attractive strategy to favour self-organisation.[3-6]

The process of self-assembly involves a hierarchical mechanism, which typically proceeds from nucleation spheres, through thin fibrils, to bundled fibers. As a result, the obtained nanostructured systems may display heterogeneous morphologies. However, the addition of another component that hinders self-assembly could prove a successful strategy to avoid bundling, and to halt the assembling process to the fibril state. To this aim, carbon nanodots are added as a functional component to a self-assembling D,L-tripeptide to achieve hydrogels composed of homogeneous fibrils at physiological conditions.[7]

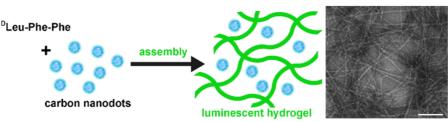


Figure 1. Assembly and TEM micrograph of luminescent hydrogel from a D,L-tripeptide and carbon nanodots. Scale bar = 200 nm.

Investigation of the two-component system revealed experimental conditions that led to stable, luminescent hydrogels, thanks to the unique properties of carbon nanodots.[8] Moreover, it is possible to modulate the rheological properties of the system. These nanocarbon composite materials could open new windows of use in the area of sensing biomaterials or even amyloidosis inhibition.

- [1] X. Du, J. Zhou, J. Shi, and B. Xu, Chem. Rev. 115, 13165 (2015).
- [2] M. Reches, E. Gazit, Science 300, 625 (2003).
- [3] S. Marchesan, K. E. Styan, C. D. Easton, L. Waddington and A. V. Vargiu, J. Mater. Chem. B 3, 8123 (2015).
- [4] A. V. Vargiu, D. Iglesias, K. E. Styan, L. J. Waddington, C. D. Easton and S. Marchesan, Chem. Commun. **52**, 5912 (2016).
- [5] A. M. Garcia, M. Kurbasic, S. Kralj, M. Melchionna and S. Marchesan, Chem. Commun. 53, 8110 (2017).
- [6] M. Melchionna, K. E. Styan, and S. Marchesan, Curr. Top. Med. Chem. 16, 2009 (2016).
- [7] M. C. Cringoli, S. Kralj, M. Kurbasic, M. Urban and S. Marchesan, Beilstein J. Nanotechnol. 8, 1553 (2017).
- F. Arcudi, L. Đorđević, M. Prato, Angew. Chem., Int. Ed. 55, 2107 (2016).