

Materials Genome Initiative (USA) 2011-2020

Materials Informatics Initiative (Japan) 2014-2025

Materials Genome Institute (MGI-SHU, China)

MARVEL Initiative (Switzerland) 2014-2025

Materials Genome Engineering (USTB, China)

**The Importance of the Mendeleev (Periodic) Number in  
Crystal Structure Prediction - a Data-Centric Approach**

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<http://paulingfile.com>

<https://mpds.io>



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**A Data-Centric Approach  
The PAULING FILE project**



# PAULING FILE

Inorganic Materials Database and Design System

PAULING FILE was started in 1994  
as collaboration between JST and MPDS

Since January 2016 equally  
copyrighted by NIMS and MPDS

**During last 25 years:**

**500 fulltime academic  
manpower have been  
invested**

**Up to now used funds:  
30 Mio. Euro**

**Present Status of LPF**

397'000 crystal structure entries  
59'000 phase diagrams  
140'000 physical properties  
(800'000 properties counts)  
270'000 bibliographic entries

**summarizing over  
185'000 selected  
publications**

(all done by professional experts)

12 products contain LPF data:

**MPDS platform (MPDS/MD Inc.)**  
AtomWork/AtomWork-advanced (NIMS)  
Alloy Phase Diagram Center (ASM)  
LPF in SpringerMaterials (Springer)  
Pearson's Crystal Data (ASM)  
Coordinates in PDF4+ (ICDD)

...

<http://paulingfile.com>

<https://mpds.io>

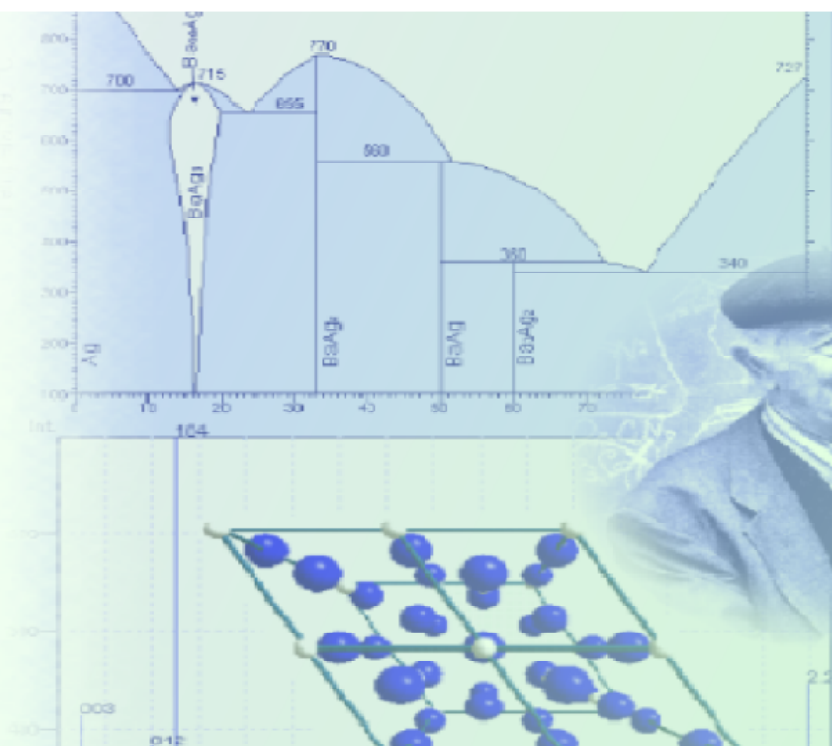
<https://crystdb.nims.go.jp>



# PAULING FILE

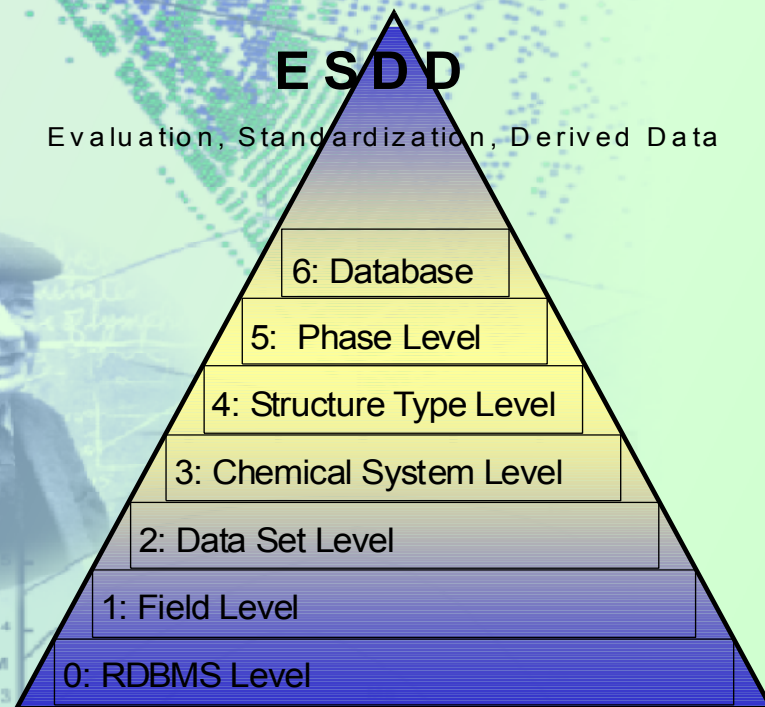
Critical Evaluation and Consequent Standardization is a **MUST**

→ All done by professional experts (not by students)



## ESDD

Evaluation, Standardization, Derived Data

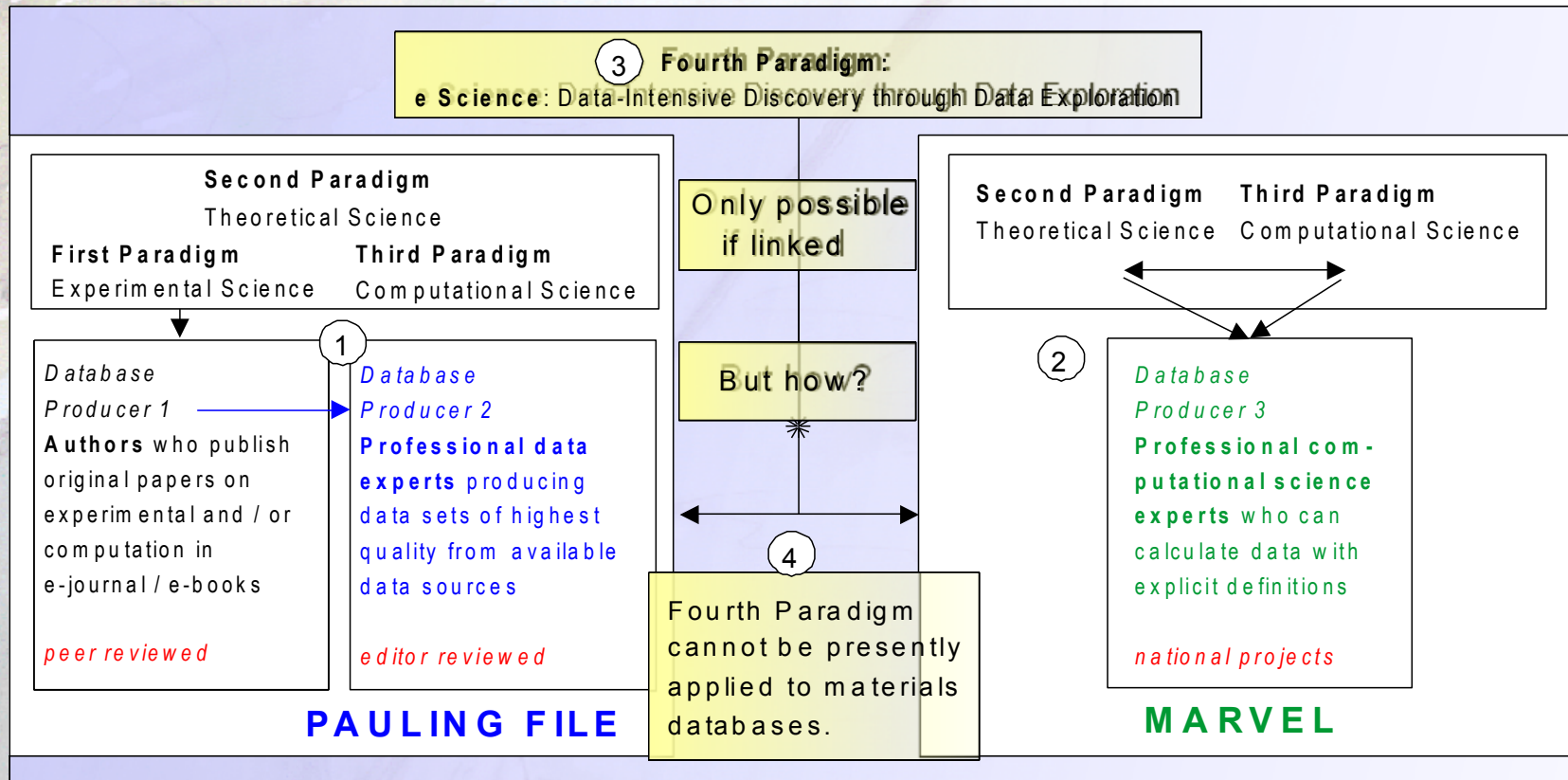


For any Data Mining Technique the following is valid:

➔ **Garbage in - Garbage out**



## The Fourth Paradigm and the present materials database situation



LINK through  
Chemical System and Chemical Formula (available in all materials databases)  
**IS NOT POSSIBLE AS**  
in about 50% of cases there exists 'Berthollide Phases' + 'Polymorphism' + Minerals



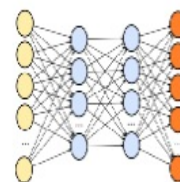
# PAULING FILE

Inorganic Materials Database and Design System

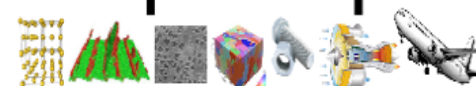
CALPHAD



Machine learning and AI



MPDS platform



Single-multi-phase bridge



QUANTUMESPRESSO



DFT



USPEX



<http://paulingfile.com>

<https://mpds.io>

## Future Development


MPDS materials platform

Materials Platform for Data Science OÜ [EE] | <https://mpds.io/#start>

Start Tutorial API Products

Pierre Villars (menu)

# MPDS





Materials Platform  
for Data Science

based on the PAULING FILE experimental inorganic database

[🔍](#) [☰](#) [μ<sub>x</sub><sup>o</sup>](#)

*e.g. silicate Cr-Sc*

peer-reviewed machine learning ab initio modeling

 <b>267 879</b> <u>scientific publications</u>	 <b>59 463</b> <u>phase diagrams</u>	 <b>397 612</b> <u>crystalline structures</u>	 <b>802 475</b> <u>property values</u>
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**New!** Visual searches for atomic environments.

About us Products Data hierarchy System status Mobile version Terms Privacy

"have a try"  
[www.mpds.io](http://www.mpds.io)

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**Data Mining by using  
Elemental Property Parameters (EPP)  
(Governing Factors)**

**+**

**the PAULING FILE**

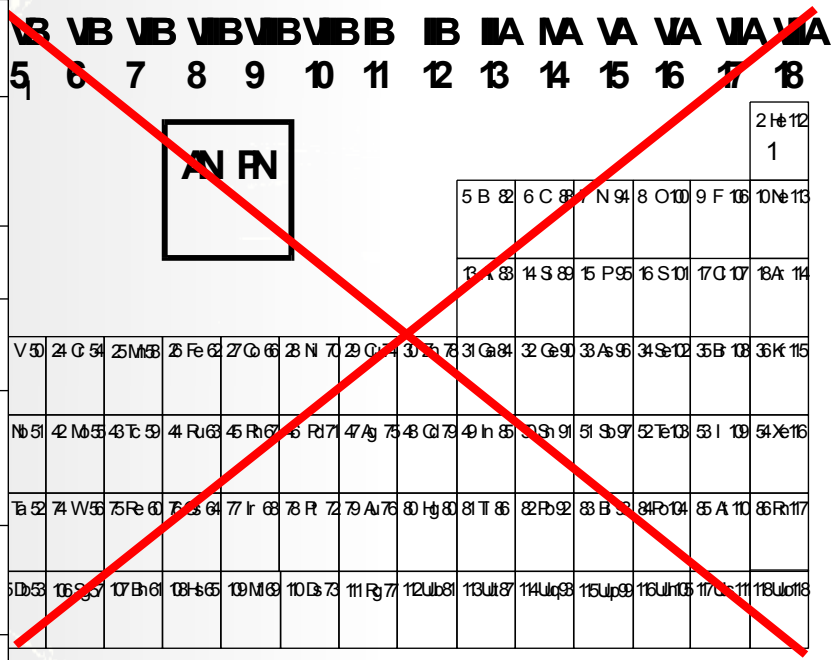
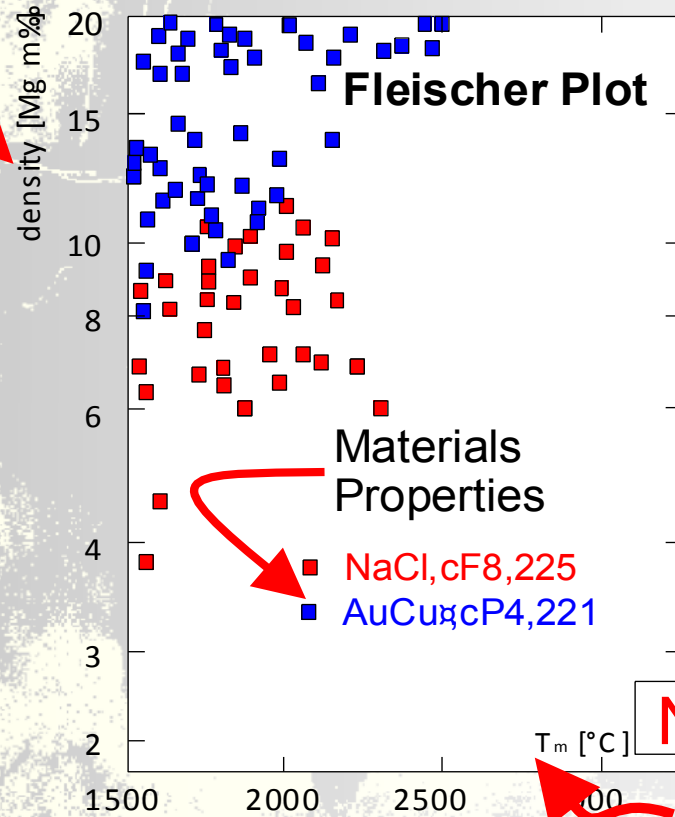
**How many distinct Governing Factors exist?**

# Data Mining PAULING FILE

Two principal different ways of Data Mining:

I) Pattern within Materials Properties

Materials Properties



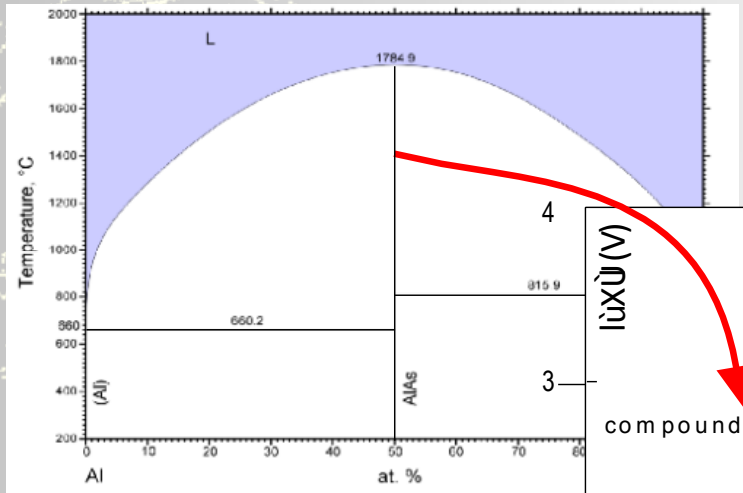
No Link to Periodic Table established

Materials Properties



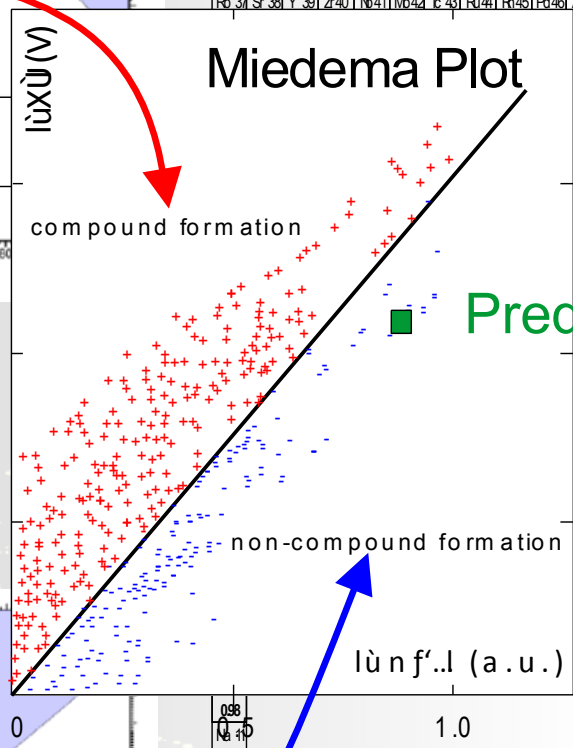
# II) Linking Elemental Property Parameters of A, B and Materials Properties of A-B ( $A_xB_y$ )

**Data Mining  
PAULING FILE**



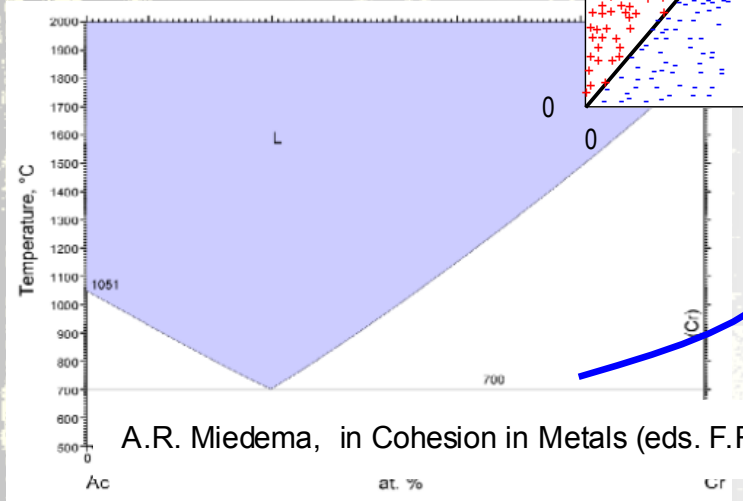
Chemical Potential after Miedema  $X_m[M]$

Li 3	285	Be 4	5.05	B 5	5.30	C 6	6.24	N 7	6.86	O 8	5.20	H 1	He 2																						
Na 11	270	Mg 12	3.45	Al 13	4.20	Si 14	4.70	P 15	5.55	S 16	4.10	F 9	Ne 10																						
K 19	225	Ca 20	2.55	Sc 21	3.25	Ti 22	3.80	V 23	4.25	Cr 24	4.65	Mn 25	4.45	Fe 26	4.93	Co 27	5.10	N 28	5.20	Cu 29	4.45	Zn 30	4.10	Ga 31	4.10	Ge 32	4.55	As 33	4.80	Se 34	4.35	Br 35	4.05	Kr 36	4.35
Rb 37	190	Sr 38	1.85	Y 39	2.40	Zr 40	2.80	Nb 41	3.20	Mo 42	3.60	Tc 43	3.90	Ru 44	4.10	Rh 45	4.30	Pd 46	4.40	Ag 47	4.35	Cd 48	4.05	In 49	3.90	Sn 50	3.90	Sb 51	4.15	Te 52	4.40	I 53	4.15	Xe 54	4.15
Cs 55	160	Ba 56	1.55	Hf 72	1.45	Ta 73	1.68	W 74	1.81	Re 75	1.85	Os 76	1.85	Ir 77	1.83	Ru 78	1.78	Au 79	1.57	Hg 80	1.24	Tl 81	1.12	Pb 82	1.12	Bi 83	1.15	Po 84	1.16	At 85	1.16	Rn 86	1.16		



Materials Property of A-B resp.  $A_xB_y$

Elemental Property Parameter of A resp. B



Interatomic electron density  $\Delta n^{0.5}$  [density units<sup>1/2</sup>]

Li 3	0.98	Be 4	1.67	B 5	1.75	C 6	1.77	N 7	1.65	O 8	1.50	H 1	He 2																						
Na 11	0.82	Mg 12	1.17	Al 13	1.39	Si 14	1.50	P 15	1.65	S 16	1.31	F 9	Ne 10																						
K 19	0.65	Ca 20	0.91	Sc 21	1.27	Ti 22	1.52	V 23	1.64	Cr 24	1.73	Mn 25	1.61	Fe 26	1.77	Co 27	1.75	N 28	1.75	Cu 29	1.47	Zn 30	1.32	Ga 31	1.31	Ge 32	1.37	As 33	1.44	Se 34	1.31	Br 35	1.31	Kr 36	1.31
Rb 37	0.55	Sr 38	0.81	Y 39	1.21	Zr 40	1.41	Nb 41	1.61	Mo 42	1.77	Tc 43	1.81	Ru 44	1.83	Rh 45	1.76	Pd 46	1.67	Ag 47	1.36	Cd 48	1.24	In 49	1.17	Sn 50	1.24	Sb 51	1.26	Te 52	1.15	I 53	1.15	Xe 54	1.15
Cs 55	0.55	Ba 56	0.81	Hf 72	1.45	Ta 73	1.68	W 74	1.81	Re 75	1.85	Os 76	1.85	Ir 77	1.83	Ru 78	1.78	Au 79	1.57	Hg 80	1.24	Tl 81	1.12	Pb 82	1.12	Bi 83	1.15	Po 84	1.16	At 85	1.16	Rn 86	1.16		

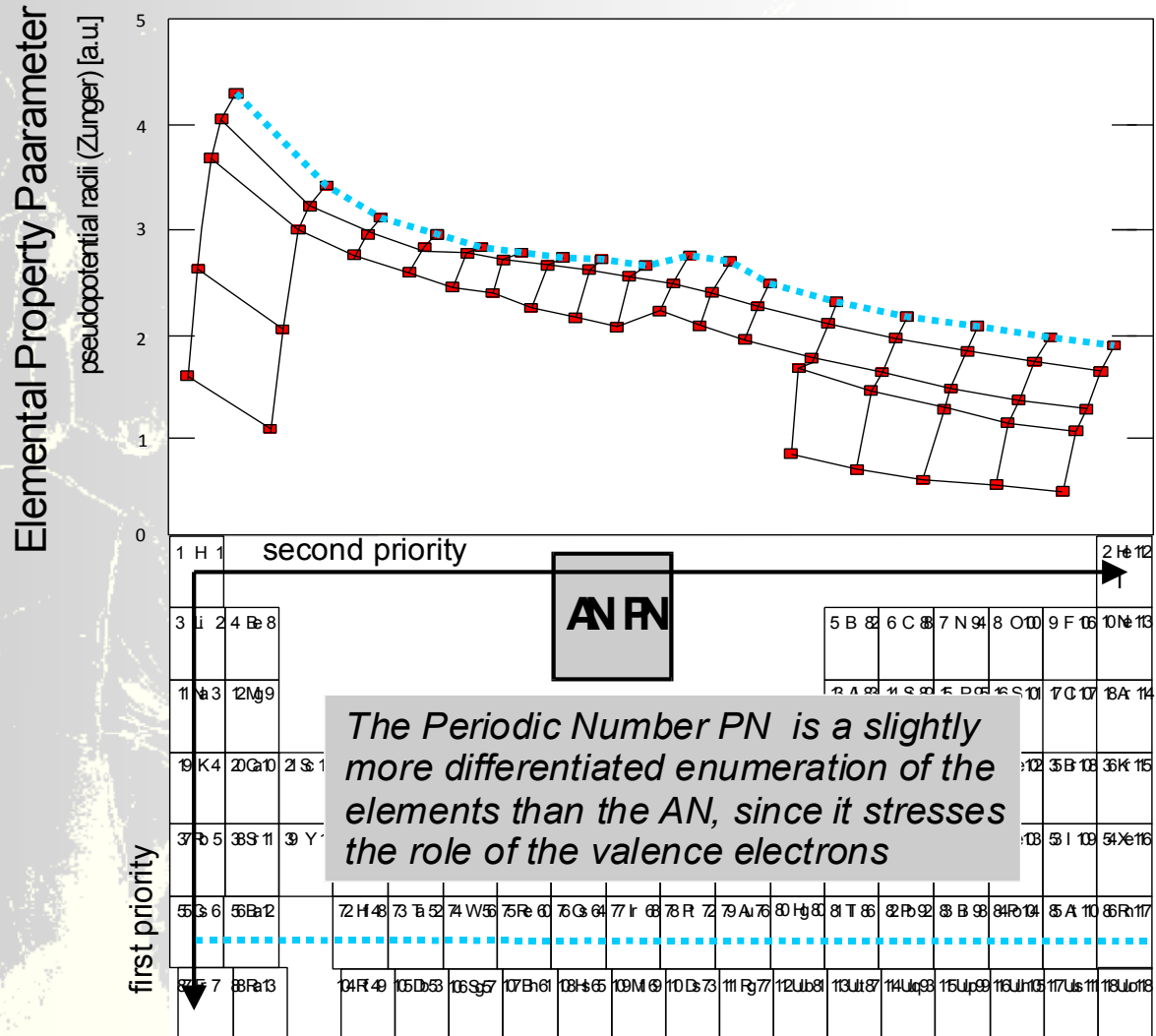
Link to Periodic Table established

A.R. Miedema, in Cohesion in Metals (eds. F.R. de Boer and D.G. Pettifor) 1988, North-Holland, Amsterdam



After Miedema: The EPP (A,B) have some Prediction Ability for A-B ( $A_xB_y$ ) Materials Properties

## Elemental Property Parameters (EPP) vs. Periodic Number (PN) Plots



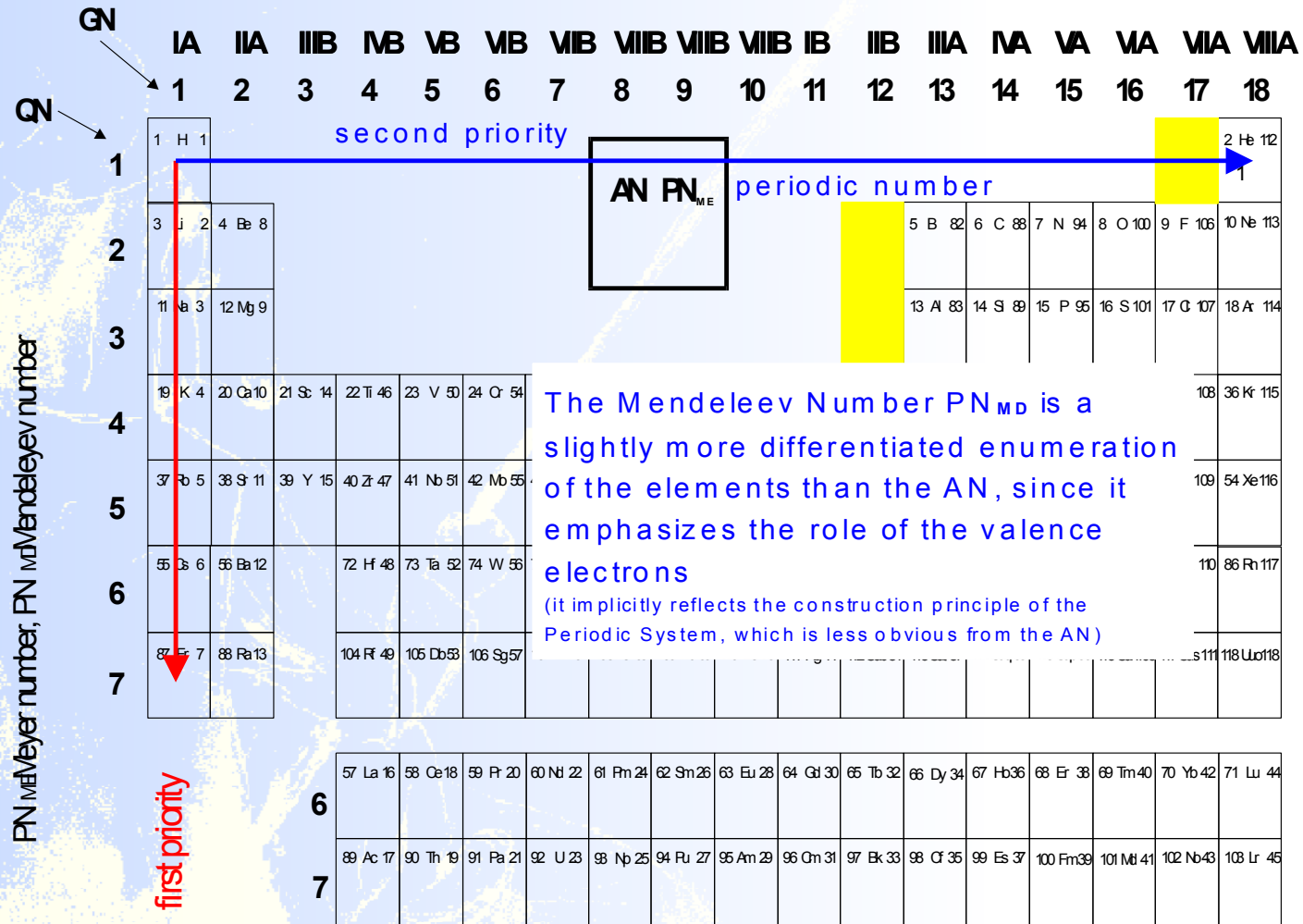
The Periodic Number PN is a slightly more differentiated enumeration of the elements than the AN, since it stresses the role of the valence electrons

About 100 different EPPs' lead to 4 different patterns:

- 1) Atomic number group
- 2) atomic size group
- 3) atomic reactivity group
- 4) atomic density (atomic affinity) group

The 4 fundamental integer Elemental-Property Parameters (EPP):

# AN, QN, PN (Mendeleev Number), GN



AN and PN (or expression of them) are optimal elemental-property parameters to be used for any kind of materials overview maps as the link to the Periodic Table is straight forward.

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$$SZ_a = k_{SZ} [\log (AN + 1)] [k_{PN} - (\log PN)^3],$$
$$RE_a = k_{RE} \{[\log (AN + 1)] [k_{PN} - (\log PN)^3]\}^{-1} = k_{SZ} k_{RE} (SZ_a)^{-1}$$

where AN is the Atomic Number

PN is the Mendeleev Number

$k_{SZ}$ ,  $k_{RE}$  are fitting parameters to fit experimental data

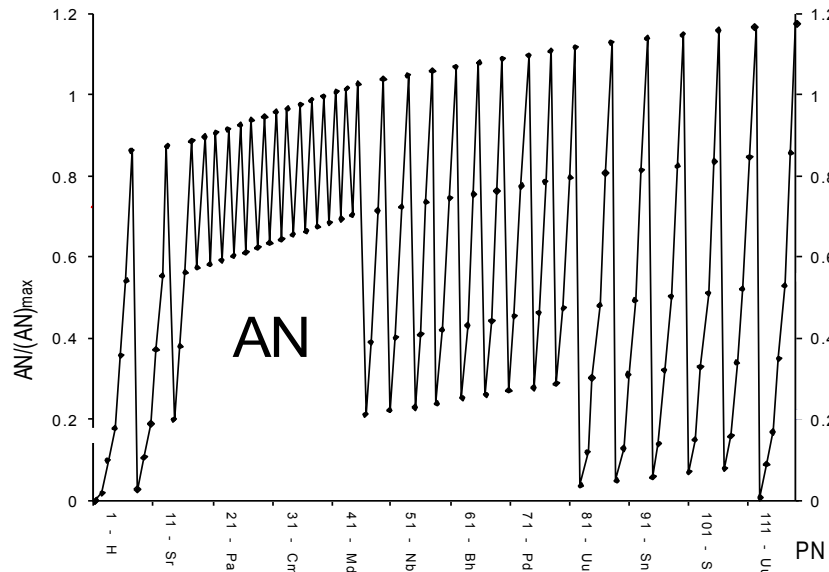
$k_{PN}$  is a scaling factor

# Functions (Patterns) within Elemental Property Parameters (EPP) versus Periodic Numbers (PN) Plots

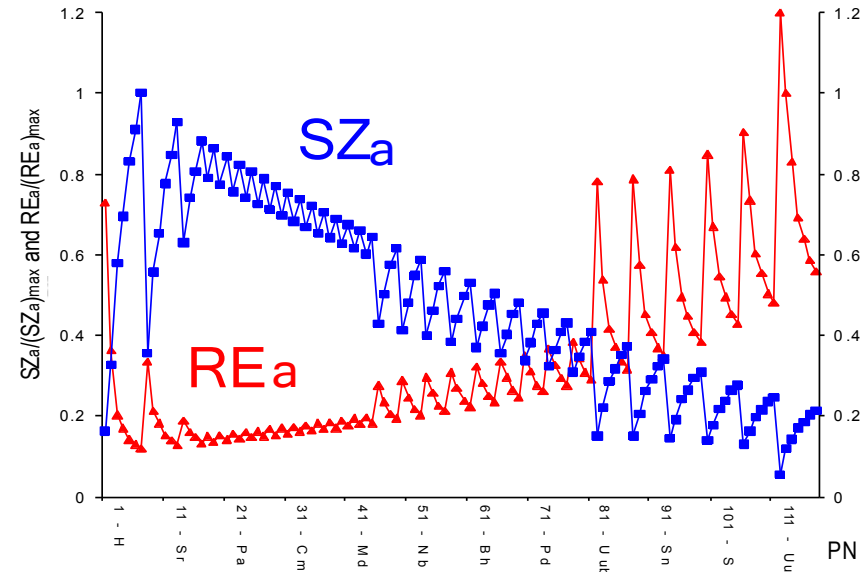
1) Atomic Number pattern

2) Atomic Size pattern

3) Atomic Reactivity pattern



$$AN = f(PN)$$



$$SZ_a = f(AN, PN)$$

$$RA_a = f(AN, PN)$$

$$DE_a = f(AN, PN)?$$

P.Villars et al., Chem. Met. Alloys 1 (2008) 1-23

Do we have reached a new qualitative and quantitative Level of Data Mining, as EPP are simple functions of AN and PN?

Fundamental Approach:

Quantum Simulations  
MARVEL

$f(AN)$

complementary

Pragmatic Approach:

Data Mining  
PAULING FILE

$f(AN, PN)$

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**Miedema's Idea to  
Link Materials Properties (e.g. Former/Nonformer)  
to its Elemental Property Parameters (EPP)**

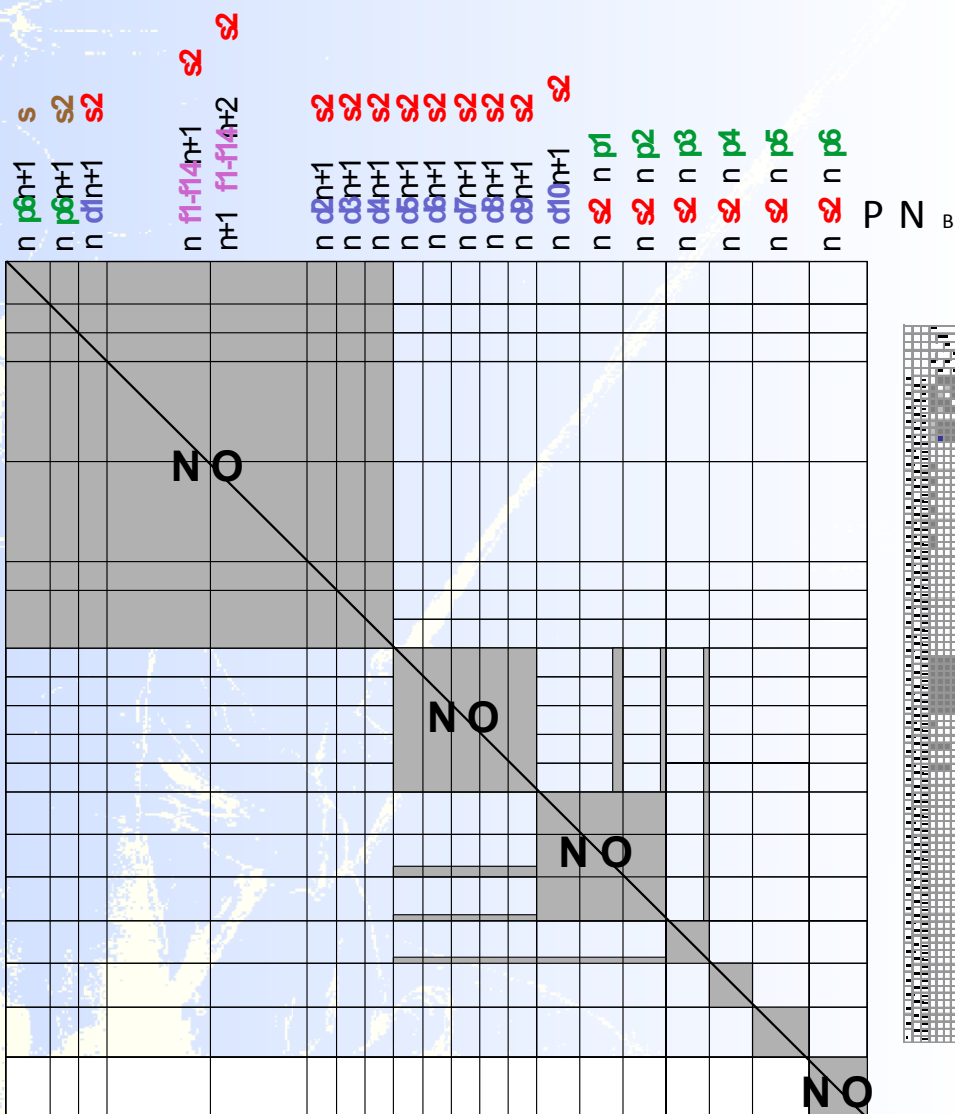
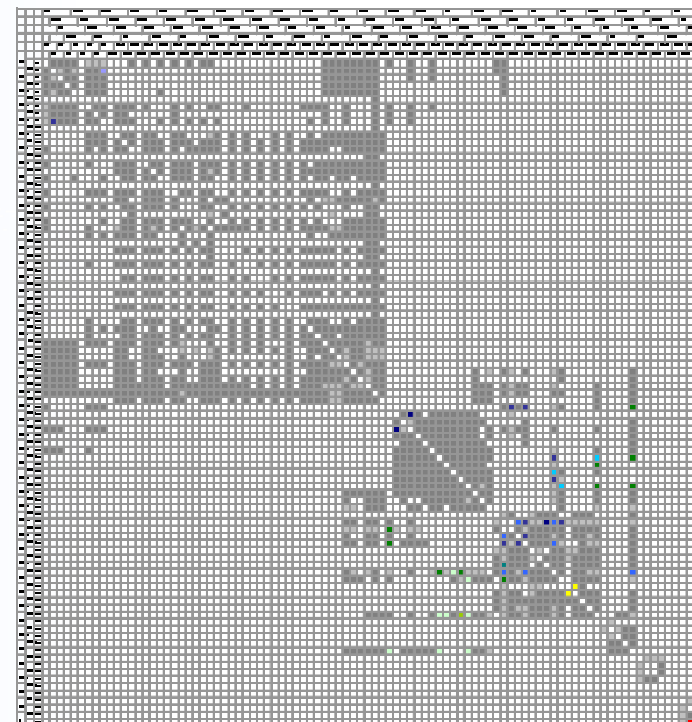
**solved with 1 Elemental Property Parameter:  
PN<sub>MD</sub> Mendeleev Number**

quinternary systems



binary systems

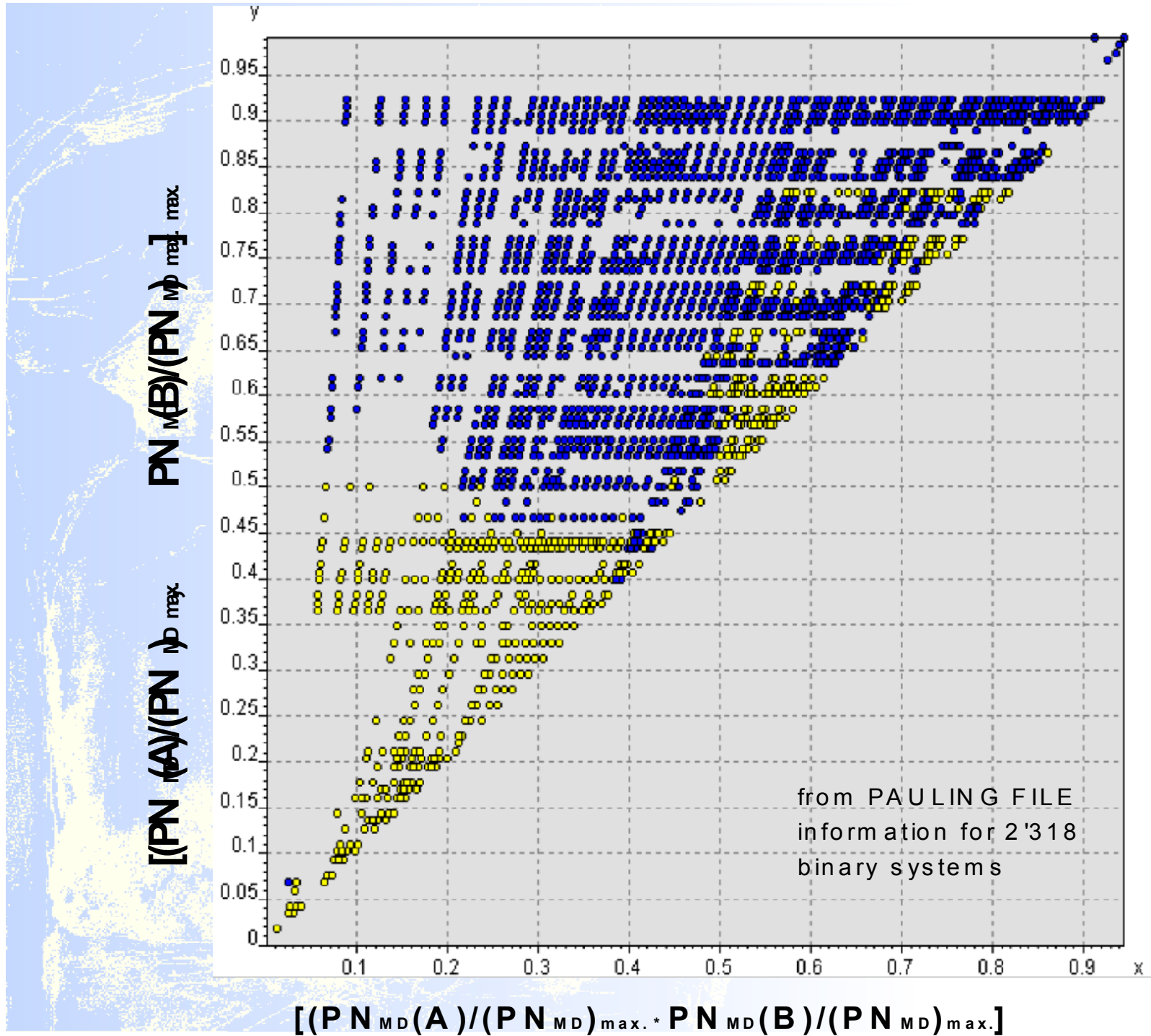
considering all data contained in LPF



$n \text{ p6 } n+1 \text{ s}$   
 $n \text{ p6 } n+1 \text{ s2}$   
 $n \text{ d1 } n+1 \text{ s2}$   
  
 $n \text{ f1-f14 } n+1 \text{ s2}$   
 $n+1 \text{ f1-f14 } n+2 \text{ s2}$   
  
 $n \text{ d2 } n+1 \text{ s2}$   
 $n \text{ d3 } n+1 \text{ s2}$   
 $n \text{ d4 } n+1 \text{ s2}$   
 $n \text{ d5 } n+1 \text{ s2}$   
 $n \text{ d6 } n+1 \text{ s2}$   
 $n \text{ d7 } n+1 \text{ s2}$   
 $n \text{ d8 } n+1 \text{ s2}$   
 $n \text{ d9 } n+1 \text{ s2}$   
 $n \text{ d10 } n+1 \text{ s2}$   
 $n \text{ s2 } n \text{ p1}$   
 $n \text{ s2 } n \text{ p2}$   
 $n \text{ s2 } n \text{ p3}$   
 $n \text{ s2 } n \text{ p4}$   
 $n \text{ s2 } n \text{ p5}$   
 $n \text{ s2 } n \text{ p6}$

About 30 % of the chemical systems have no compound





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**PAULING FILE has two  
Crystal Structure Classifications:**

- Structural Prototype Classifications**
- Atomic Environment Classification**

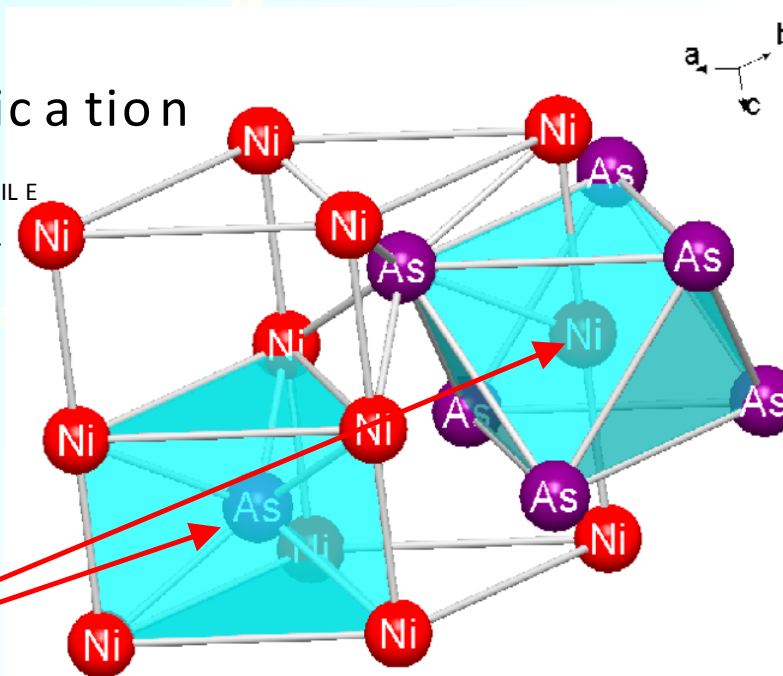
# Structural Prototyping Classification

All data sets with published coordinates are in the PAULING FILE classified in Prototypes, following the criteria defined in TYPiX.

According to this definition isotypic compounds must crystallize in:

- same space group
- have similar cell parameter ratios
- occupy same Wyckoff positions (standardized cell)
- same or similar values of the atom coordinates

If all these criteria are fulfilled, the atomic environments should be similar  
(No distinction is made between structures with fully and partly occupied atom sites)



**Struktur Bericht  
Notation (IUCr):**  
e.g.: A1, B1, B8<sub>1</sub>

Simplicity

**Pearson's HB  
Notation:**  
Cu, C1Na, AsNi

Scientifically rigorous

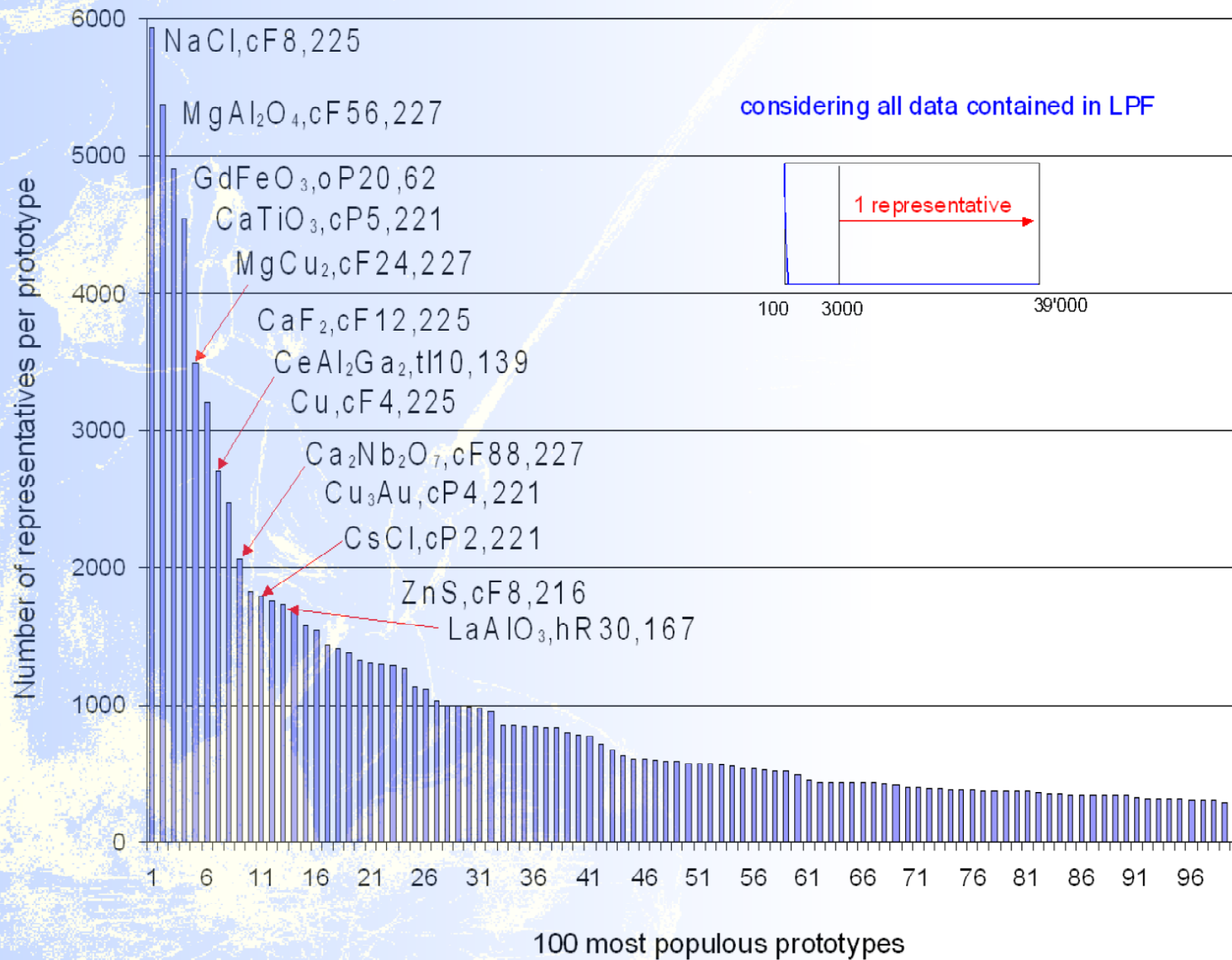
**PAULING FILE  
Notation:**  
Cu, cF4, 225  
NaCl, cF8, 225  
NiAs, hP4, 194

Prototype

Pearson Symbol

Space Group  
Number

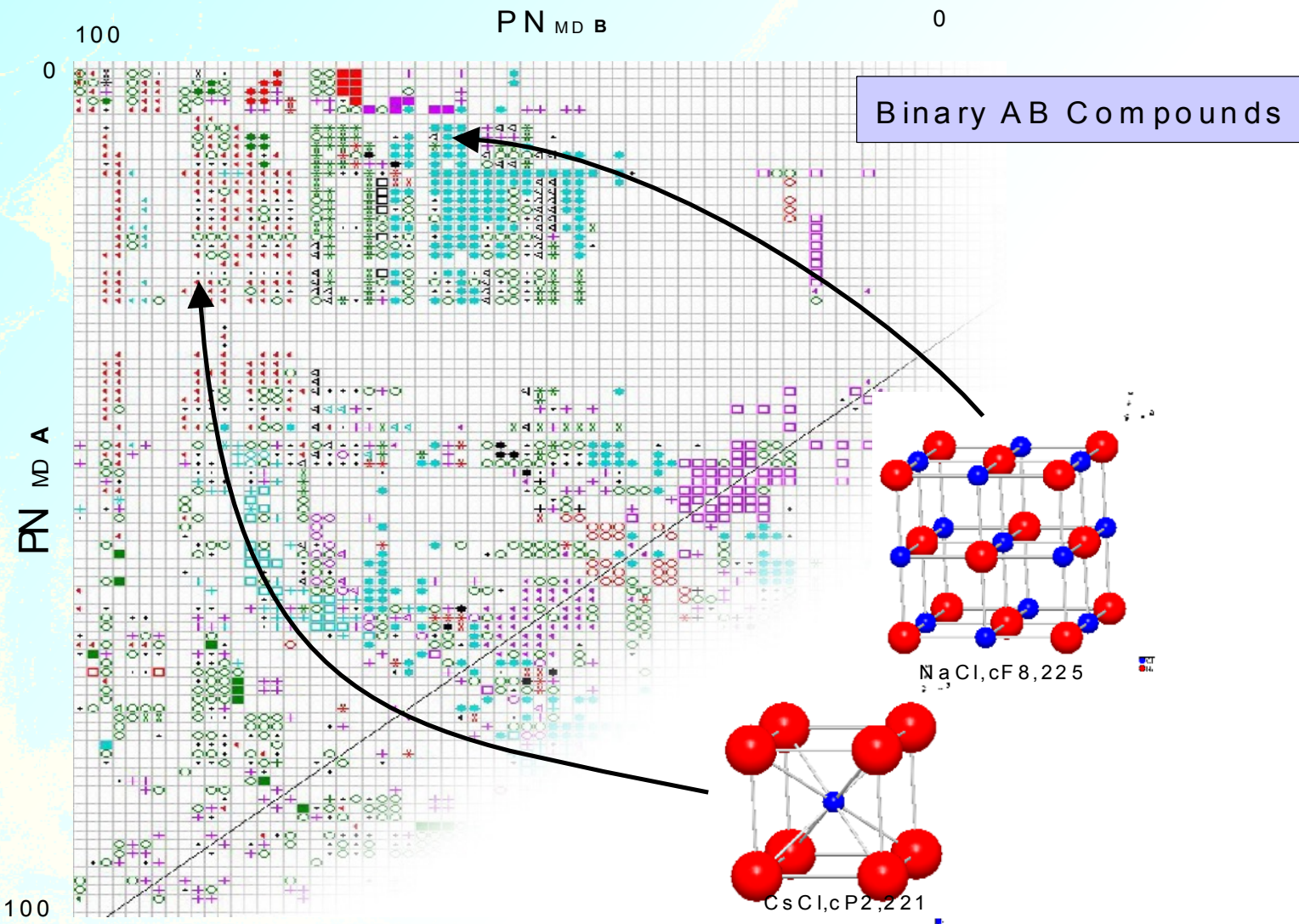
## Frequency plot of number of representatives versus 100 (0.25 %) most populous prototypes





# Classical Prototype Classification

## Structure Map



# Atomic Environment Classification

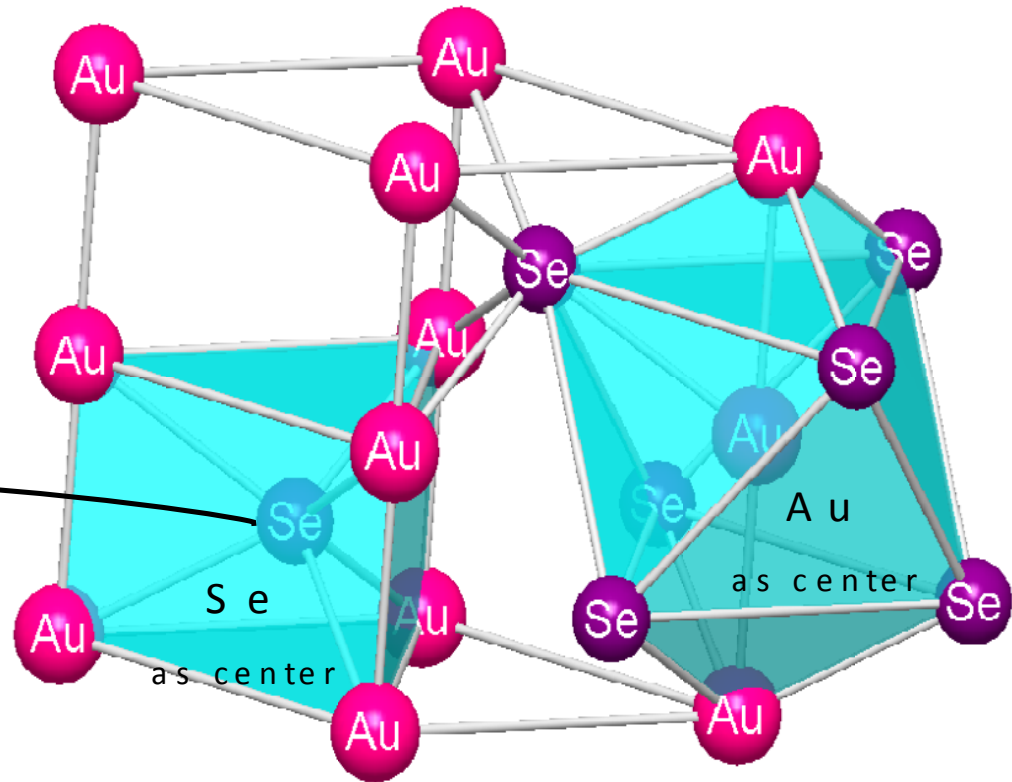
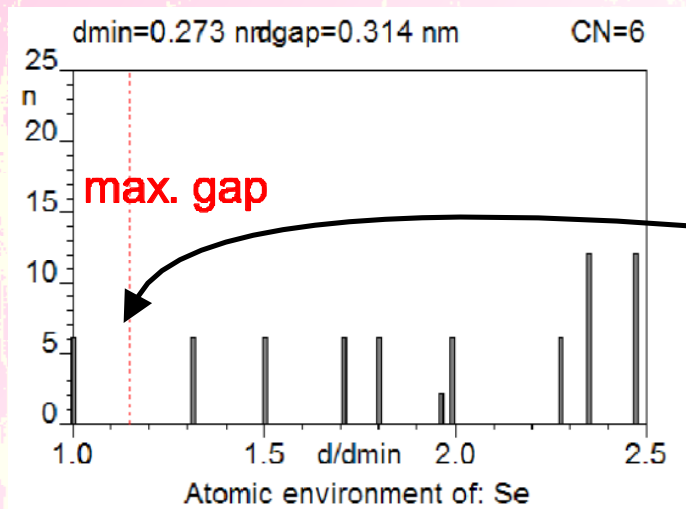
We defined an Atomic Environment Type (AET), also called coordination polyhedron, as follows:

- maximum gap rule (Brunner and Schwarzenbach)
- maximum-convex-volume rule

We kept the number of different AETs in a prototype as small as possible.

## Near Neighbor Histogram

n (no of neighbors) vs.  $d/d_{min}$

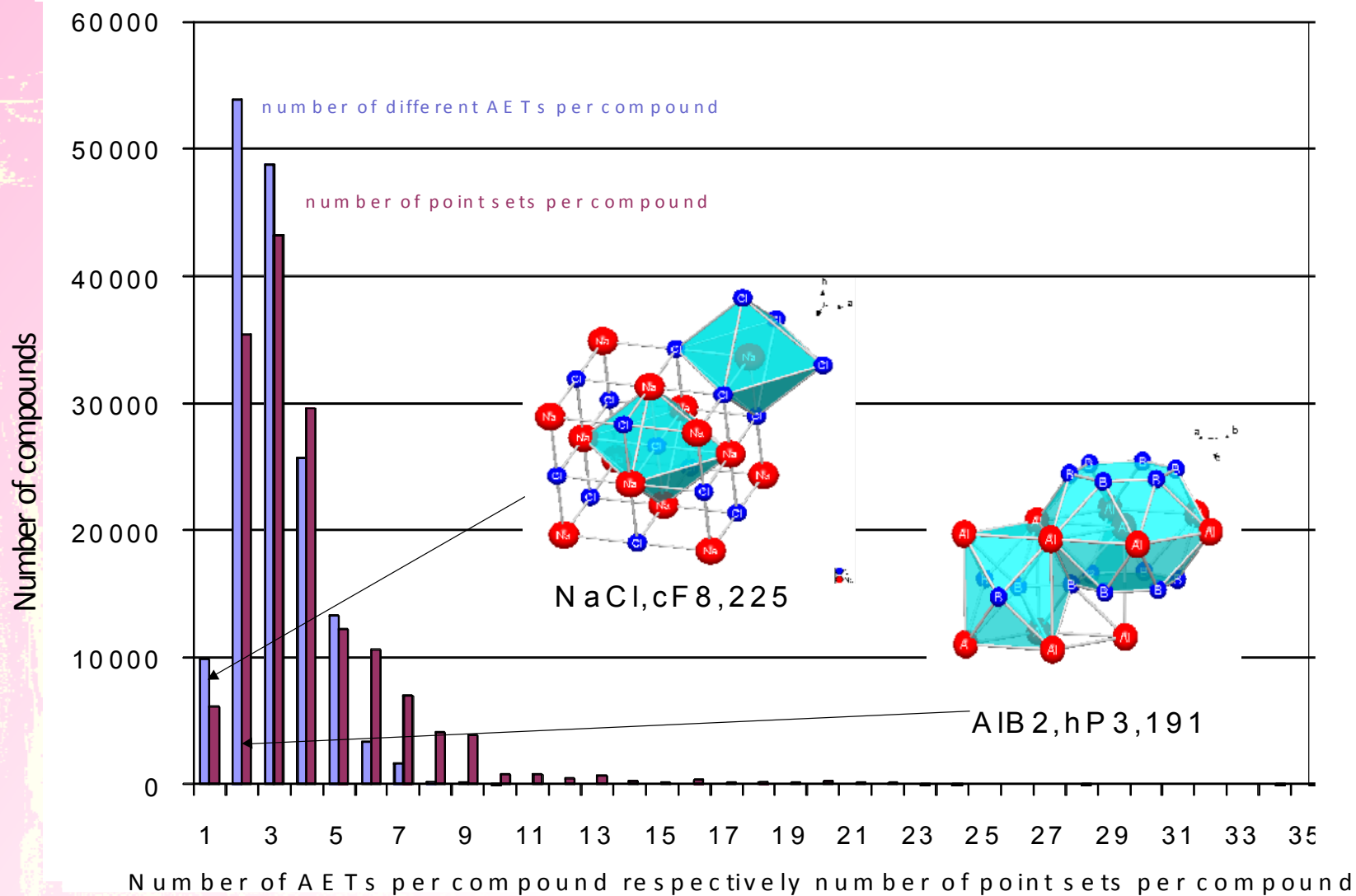


Atomic Environment Types AET are the same for all isotopic compounds



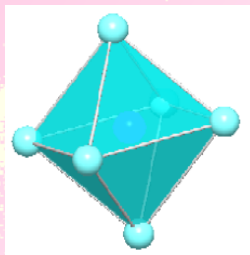
# Atomic Environment Classification

far majority less than 5 AET per compound

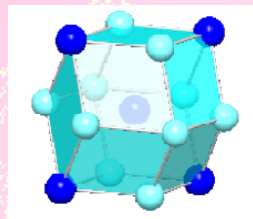


# Atomic Environment Classification

## Structure Map



octahedron



rhombic dodecahedron

### Materials Property:

AET Types (Single Environment)

Used Elemental Parameters: R, X, VE

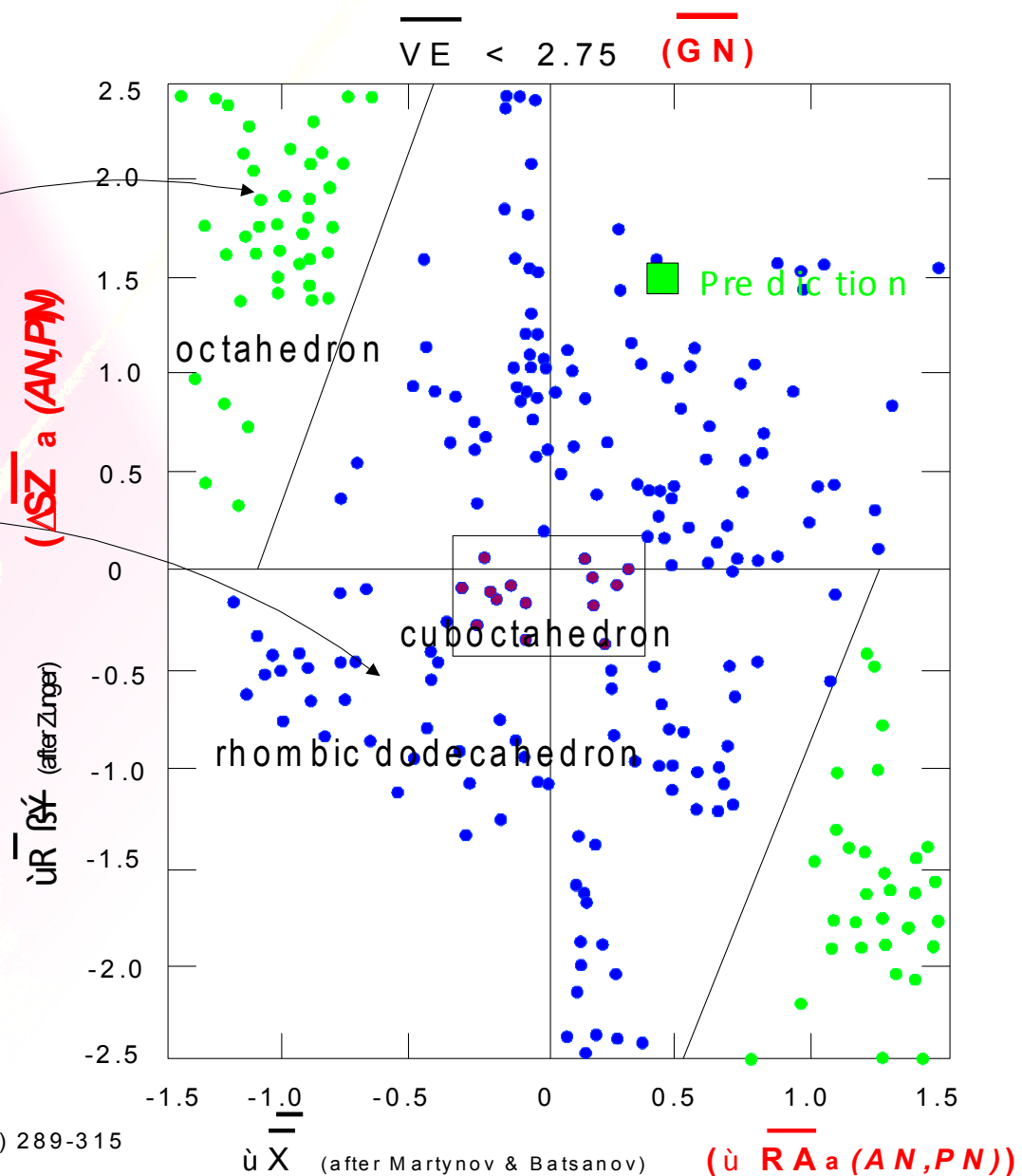
Number of used Data-Sets:

2,486 single environment compounds

**binary, ternary, quaternary** daltonide

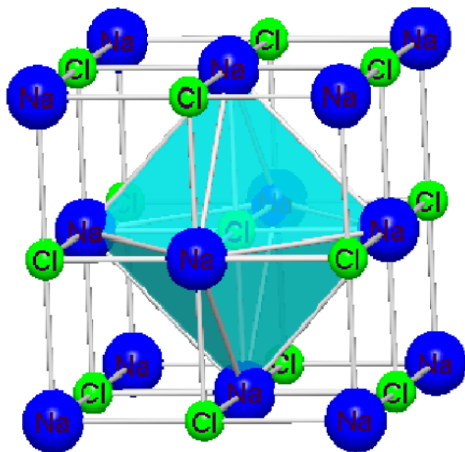
and berthollide compounds

Accuracy: 97%



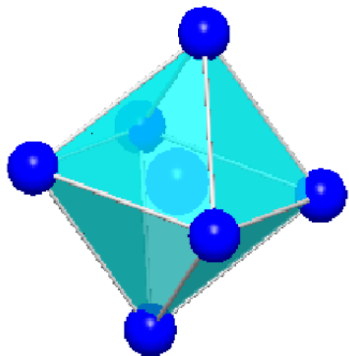
# Classical Prototype vs. Atomic Environment Classifications

prototype classification



Prototype: NaCl,  $cF8, 225$

AET classification



AET: Octahedron

## Classical View (Space Group Theory):

Isotypic compounds (same prototype) must crystallize in the same space group, having similar cell parameters occupy the same Wyckoff positions in the standardized description and have same or similar values of atom coordinates --> atomic environment should be similar.

Presently PAULING FILE contains:

over 39'000 different prototypes

with in total 397'000 entries !!

In average 10 compounds per prototype

Hopeless ?

Atomic Environment Classification:

Presently PAULING FILE contains:

over 1'000'000 fully occupied point-sets and 50 AET

In average 20'000 different pointsets per AET

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**Focusing on AB binary compounds  
using the  
Atomic Environment Classification**

# Used Database: PAULING FILE CD-ROM **Binaries Edition**



## Chemical Systems

not yet investigated  
 no compounds formed  
 no AB compounds formed  
 AB compounds (no crystal data)  
 AB compounds with crystal data  
 chemical elements

Total

Limited to  
 thermodyn.  
 defined AB  
 former

Including  
 chemical  
 elements +  
 extended s.s.

Data set 1

Data set 2

1480 (34.5%)

1481 (33.9%)

760 (17.8%)

540 (12.3%)

839 (19.6%)

794 (18.2%)

73 (1.8%)

79 (1.8%)

1126 (26.3%)

1384 (31.7%)

93 (2.1%)

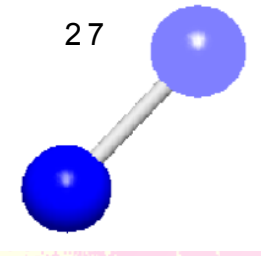
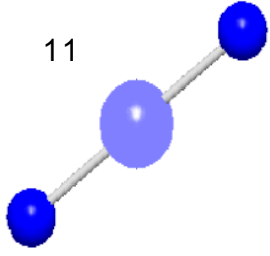
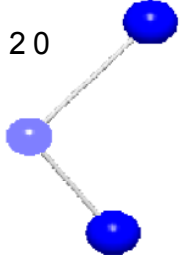
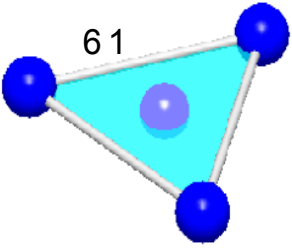
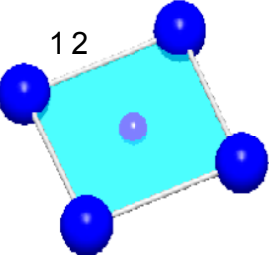
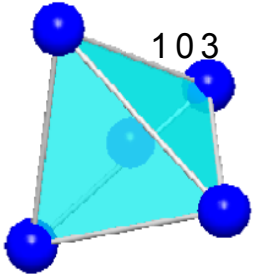
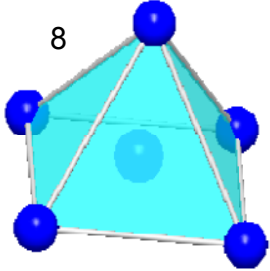
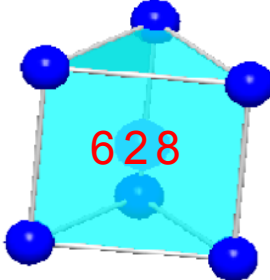
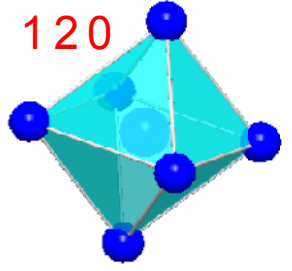
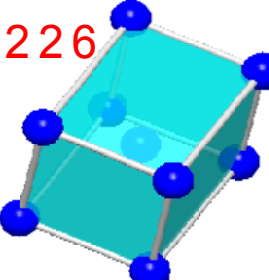
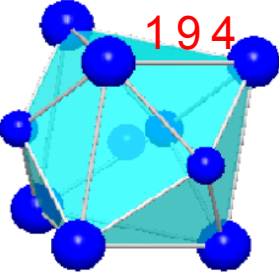
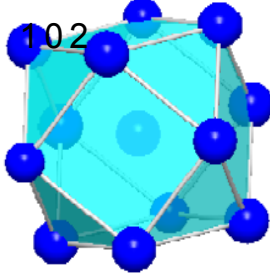
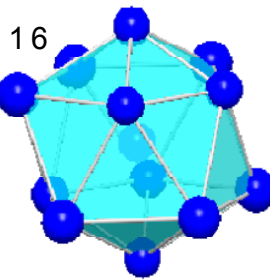
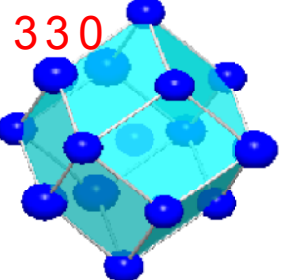
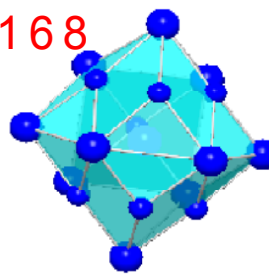
4278

4371

# 2002 AETs (15 most populous)

## AET Dataset 1

Structure Prototyping System as Door to Fundamentals

 <p>27</p>	 <p>11</p>	 <p>20</p>	 <p>61</p>	 <p>12</p>
single atom	linear	non-linear	(co-planar) triangle	(co-planar) square
 <p>103</p>	 <p>8</p>	 <p>628</p>	 <p>120</p>	 <p>226</p>
tetrahedron	square pyramid	trigonal prism	octahedron	cube
 <p>194</p>	 <p>102</p>	 <p>16</p>	 <p>330</p>	 <p>168</p>
tricapped trigonal prism	cuboctahedron	icosahedron	rhombic dodecahedron	heptacapped pentagonal prism



$PN_{MD}$  (Center A) vs.  $PN_{MD}$  (Atomic Environment B)  
for Equi-Atomic Binaries

$PN_{MD B}$

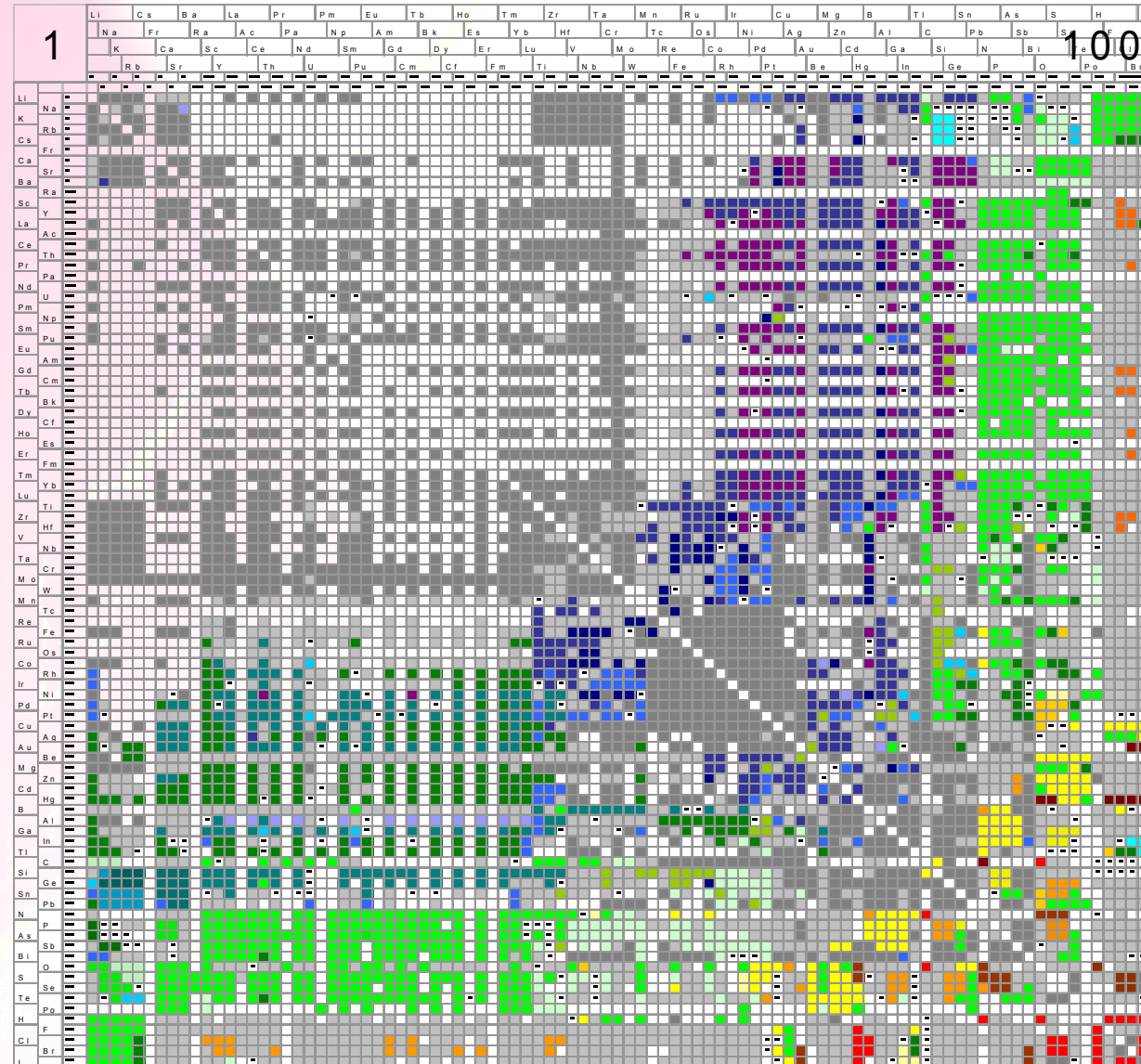
Coordinating Atoms

Data set 1

no compounds
no 1:1 compound
? unknown structure
C complex
single atom
2 linear
2 non-linear
triangle
square
tetrahedron
square pyramid
trigonal prism
octahedron
7-vertex polyhedron
cube
tricapped trigonal prism
9-11 others
(anti)cuboctahedron
icosahedron
rhombic dodecahedron
13-16 others
pentacapped pentagonal prism

1  
 $PN_{MD A}$   
100

100  
Central Atom



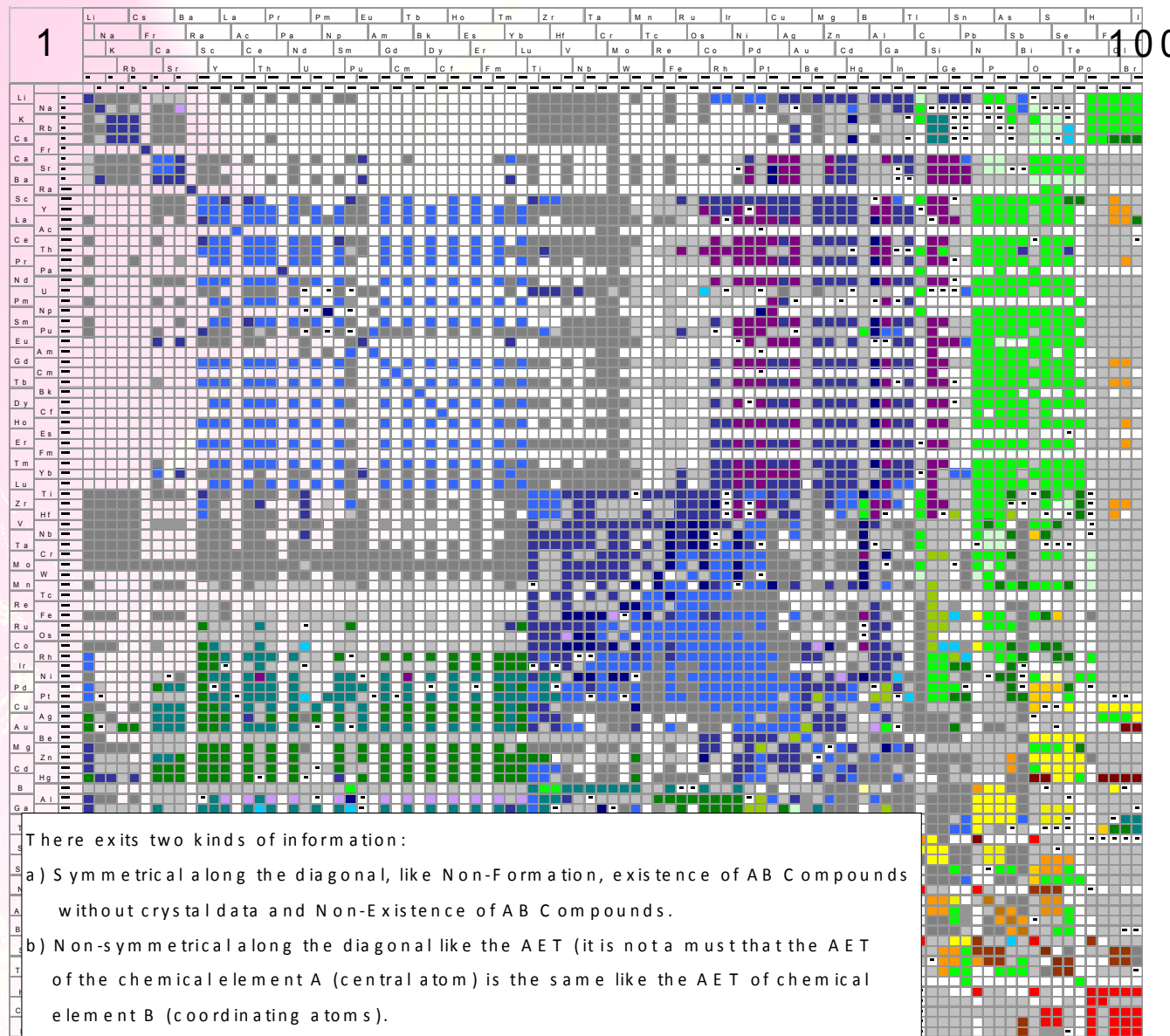
# $P N_{MD}$ (Center A) VS. $P N_{MD}$ (Atomic Environment B) for Equ-Atomic Binaries

$P N_{MD B}$   
Coordinating Atoms

Data set 2

■	no compounds
■	no 1:1 compound
?	unknown structure
C	complex
■	single atom
■	2 linear
■	2 non-linear
■	triangle
■	square
■	tetrahedron
■	square pyramid
■	trigonal prism
■	octahedron
■	7-vertex polyhedron
■	cube
■	tricapped trigonal prism
■	9-11 others
■	(anti)cuboctahedron
■	icosahedron
■	rhombic dodecahedron
■	13-16 others
■	pentacapped pentagonal prism

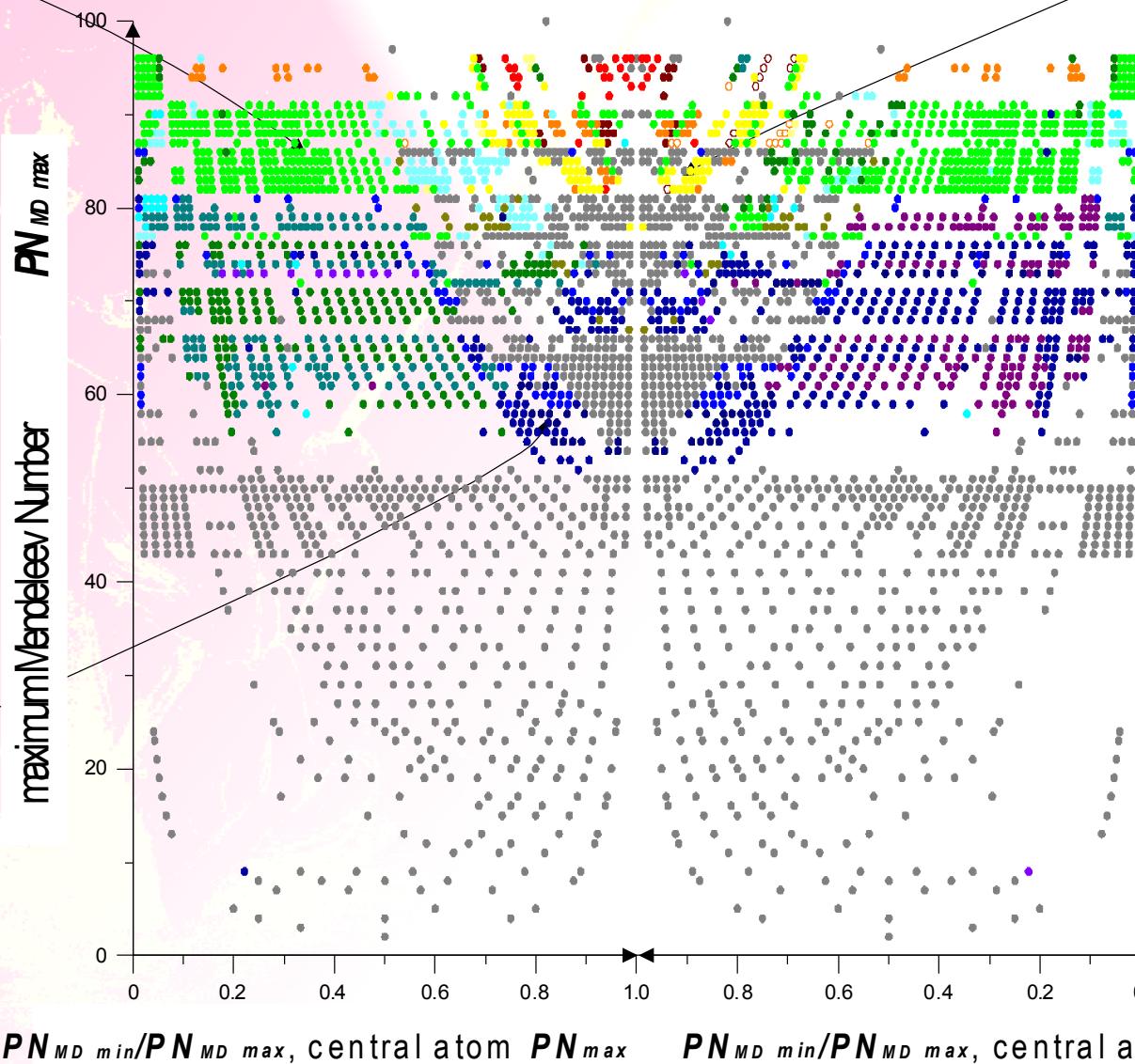
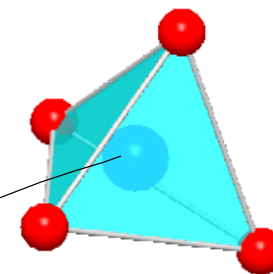
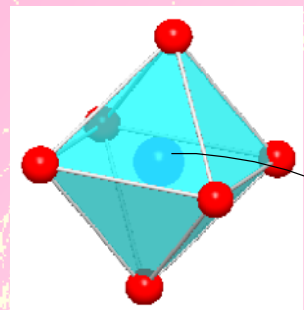
1  
 $P N_{MD A}$   
↓  
100



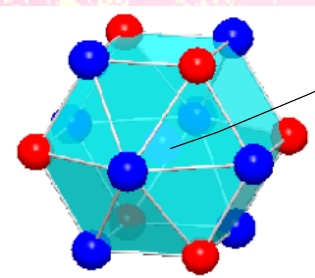
100  
Central Atom

# Atomic Environment Stability Maps for Equi-Atomic Binaries

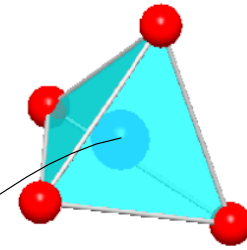
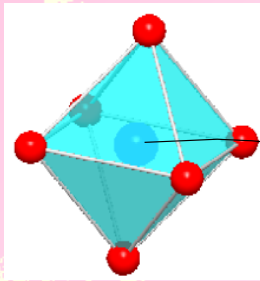
Data set 1



- no compounds
- single atom
- linear
- non-linear
- (co-planar) triangle
- (co-planar) square
- tetrahedron
- square pyramid
- trigonal prism
- octahedron
- 7-vertex polyhedron
- cube
- tricapped trigonal prism
- 9-11 vertex polyhedra
- (anti)cuboctahedron
- icosahedron
- rhombic dodecahedron
- 13-16 vertex polyhedra
- heptacapped pentagonal prism

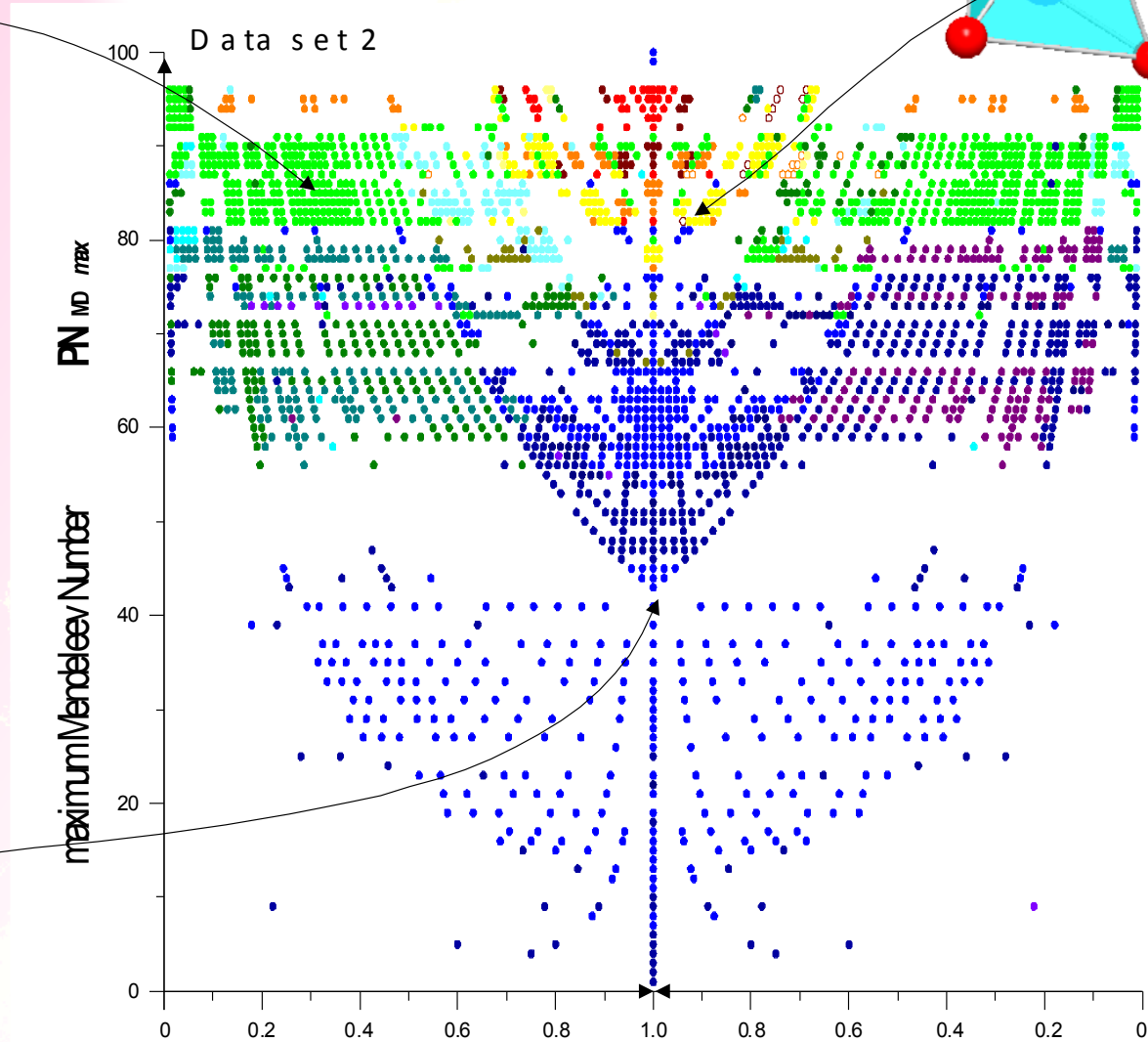
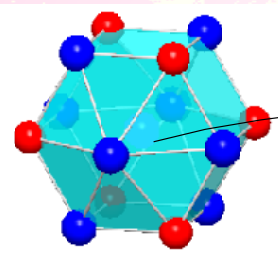


# Atomic Environment Stability Maps for Equi-Atomic Binaries



## Results:

- 1) Only 6 of 15 AET's are populous.
- 2) There exist clearly defined AET domains.
- 3) Chemical elements prefer strongly to realize just one kind of AET with another chemical element.
- 4) In extremely many cases the AET of the both chemical elements A, B within the same AB compound are the same.



$PN_{MD\ min}/PN_{MD\ max}$ , central atom  $PN_{MD\ max}$  .....  $MN_{MD\ min}/PN_{MD\ max}$ , central atom  $PN_{MD\ min}$

**Materials Genome Initiative (USA)** 2011-2020



**Materials Informatics Initiative (Japan)** 2014-2025

**Materials Genome Institute (MGI-SHU, China)**

**MARVEL Initiative (Switzerland)** 2014-2025

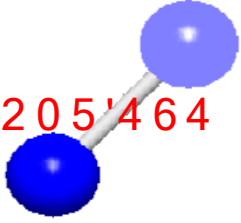
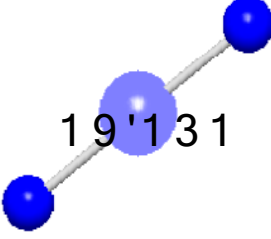
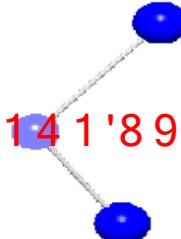
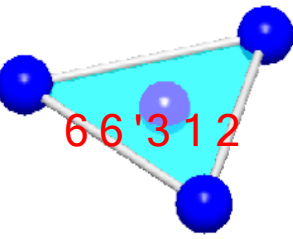
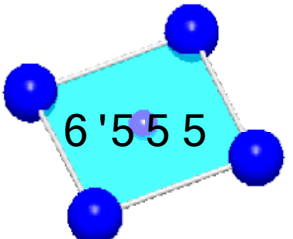
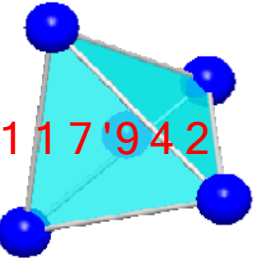
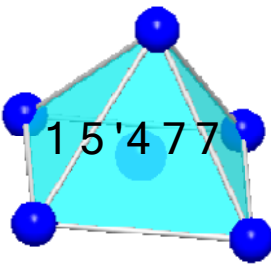
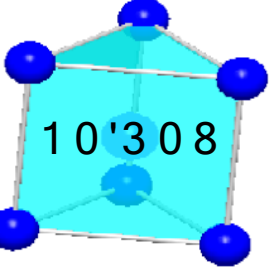
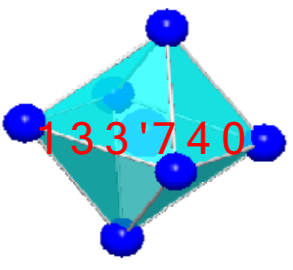
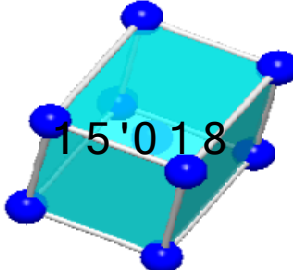
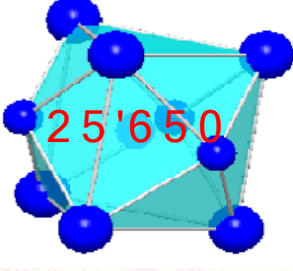
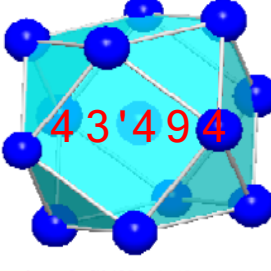
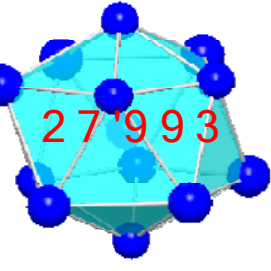
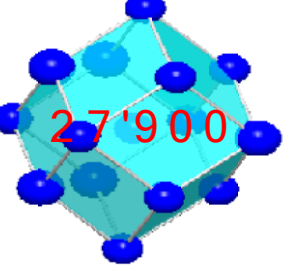
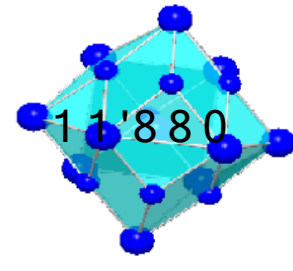
**Materials Genome Engineering (USTB, China)**

**Focusing on binary to multinary compounds  
using the  
Atomic Environment Classification**



# Atomic Environment Types Classification

36 most populous AET represent over 96 % of about 1'000'000 considered atoms (point-sets) equal to 320'000 data sets (37'000 different prototypes)

 205'464	 19'131	 141'897	 66'312	 6'555
single atom	collinear	non-collinear	coplanar triangle	coplanar square
 117'942	 15'477	 10'308	 133'740	 15'018
tetrahedron	square pyramid	trigonal prism	octahedron	cube
 25'650	 43'494	 27'993	 27'900	 11'880
tricapped trigonal prism	cuboctahedron	icosahedron	rhombic dodecahedron	Frank-Kasper 16-vertex

→ In average 28'000 different point-sets per AET

Materials Genome Initiative (USA) 2011-2020

Materials Informatics Initiative (Japan) 2014-2025

Materials Genome Institute (MGI-SHU, China)

MARVEL Initiative (Switzerland) 2014-2025

Materials Genome Engineering (USTB, China)

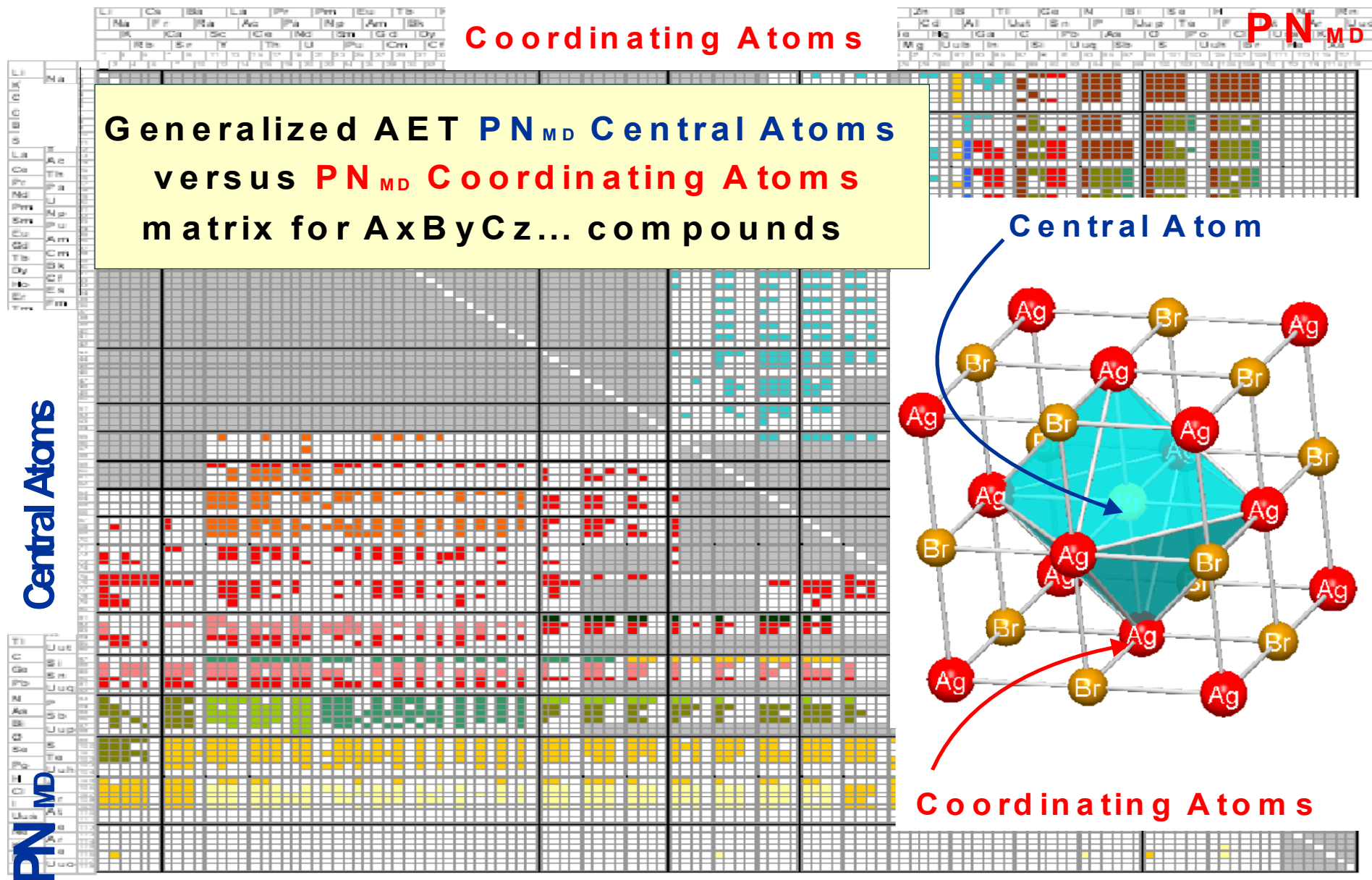
Even having in average 28'000 different point-sets per AET

**we had to simplify the Atomic Environments Types  
(AET) to its Coordination Numbers (CNs)**

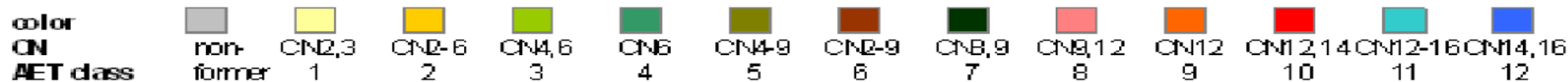
**resp. sometimes its CNs ranges**

**(bonus)**

**Independent of the number of involved chemical  
elements, as well as of its stoichiometric ratios**



P.Villars et al., Chem. Met. Alloys 1 (2008) 210-226



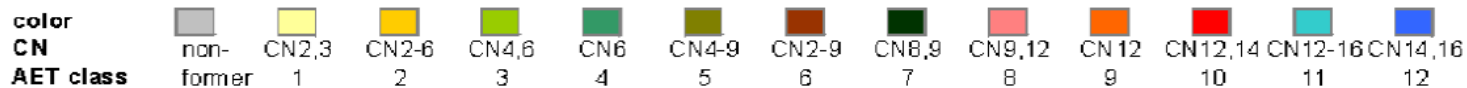
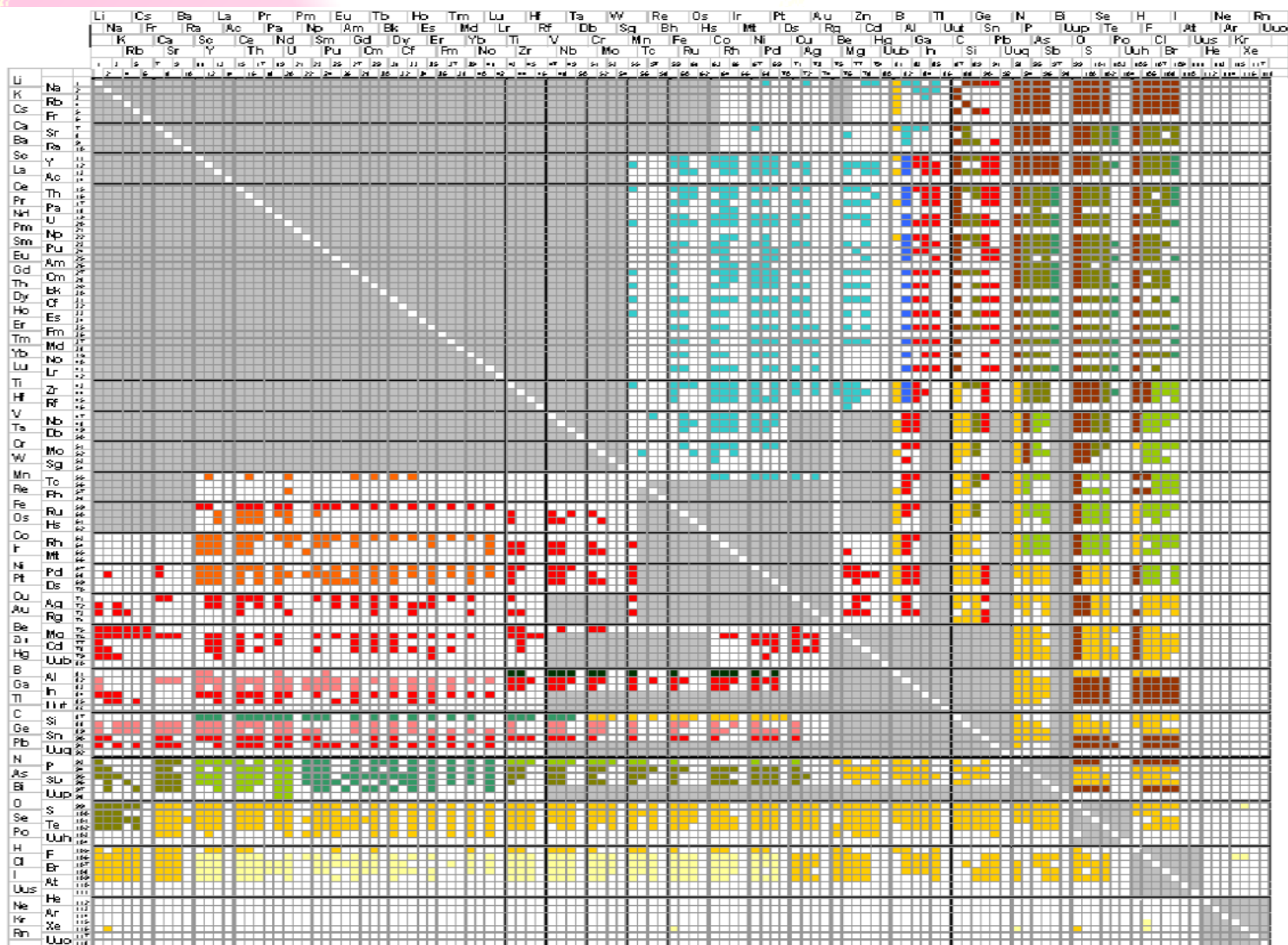
# Experimental facts

## Coordinating Atoms

P N M D

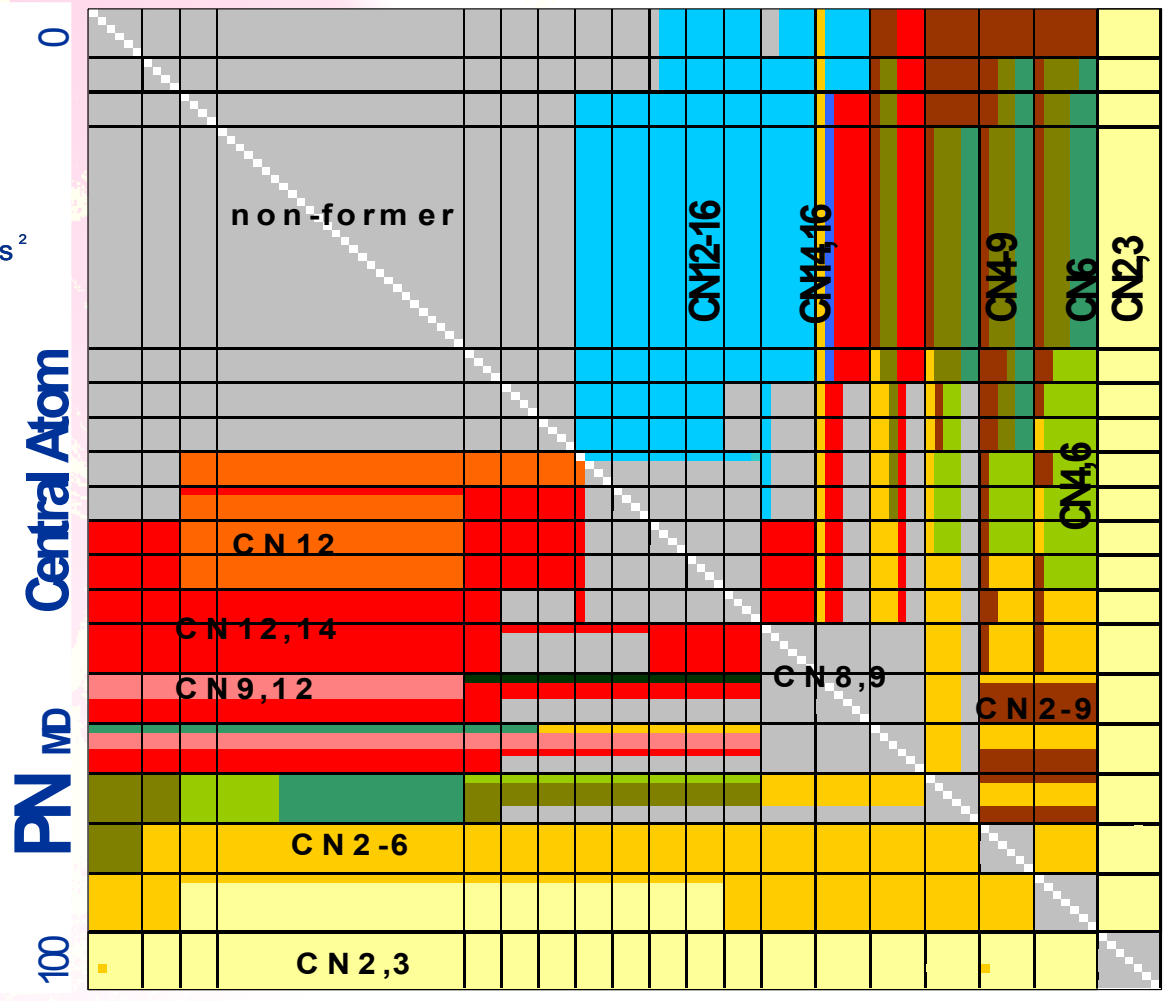
Central Atom

P N M D



$np^6(n+1)s^1$   
 $np^6(n+1)s^2$   
 $nd^1(n+1)s$   
 $n(f^1-f^{14})(n+1)s^2$   
 $(n+1)(f^1-f^{14})(n+2)s^2$   
 $nd^2(n+1)s^2$   
 $nd^3(n+1)s^2$   
 $nd^4(n+1)s^2$   
 $nd^5(n+1)s^2$   
 $nd^6(n+1)s^2$   
 $nd^7(n+1)s^2$   
 $nd^8(n+1)s^2$   
 $nd^9(n+1)s^2$   
 $nd^{10}(n+1)s^2$   
 $ns^2 np^1$   
 $ns^2 np^2$   
 $ns^2 np^3$   
 $ns^2 np^4$   
 $ns^2 np^5$   
 $ns^2 np^6$

$ns^1$   
 $ns^2$   
 $ns^3$   
 $ns^4$   
 $ns^5$   
 $ns^6$



**PN<sub>MD</sub>**





# The following observations have been made:

## General Valid:

- i) The CN are independent of the stoichiometry, as well as of the number of chemical elements in the inorganic compound: they depend only on the [Central Atom A - Coordinating Atoms B] combinations.
- ii) There exist 21 different groups of chemical elements acting as Central Atom A, which prefer particular CNs for different Coordinating Elements B, or do not form [Central Atom - Coordinating Atoms] combinations with B.
- iii) Chemical elements with  $PN_{MD} > 54$  define the CNs they are part of, regardless of whether they act as Central or as Coordinating Atoms.
- iv) The observation that the non-existence of [Central Atom - Coordinating Atoms] combinations is consistent with the non-existence of a binary equi-atomic compound formed by the same chemical elements

