



UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION  
INTERNATIONAL ATOMIC ENERGY AGENCY  
INTERNATIONAL CENTRE FOR THEORETICAL PHYSICS  
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H4.SMR/916 - 37

**SEVENTH COLLEGE ON BIOPHYSICS:**  
*Structure and Function of Biopolymers: Experimental and Theoretical  
Techniques.*  
4 - 29 March 1996

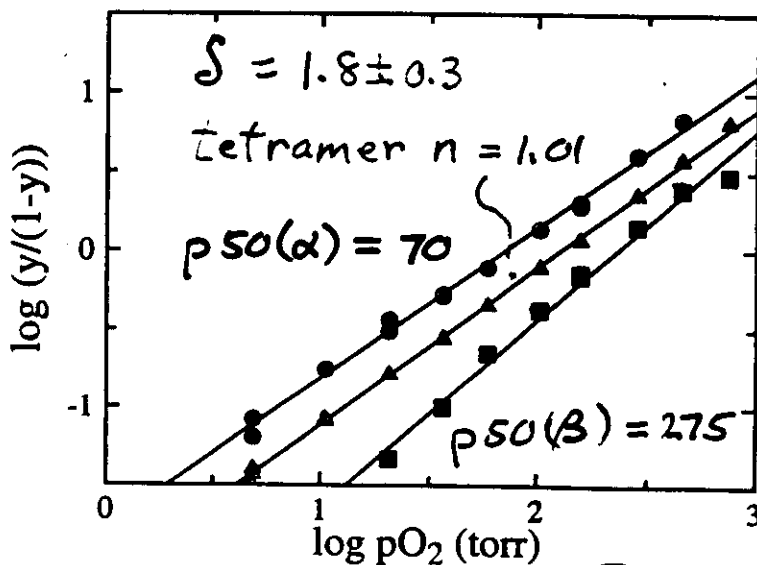
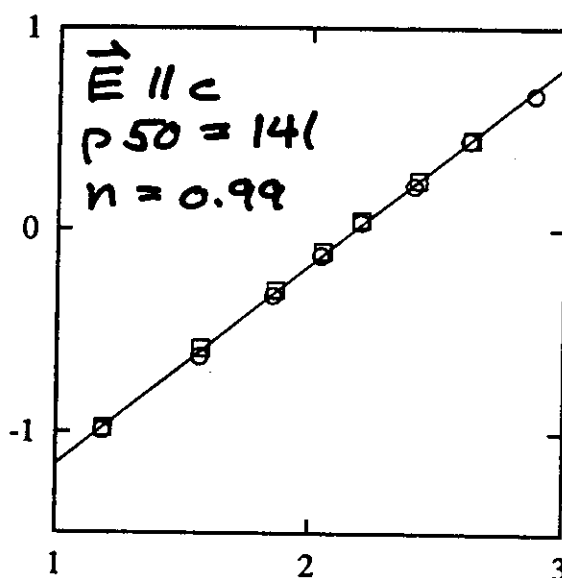
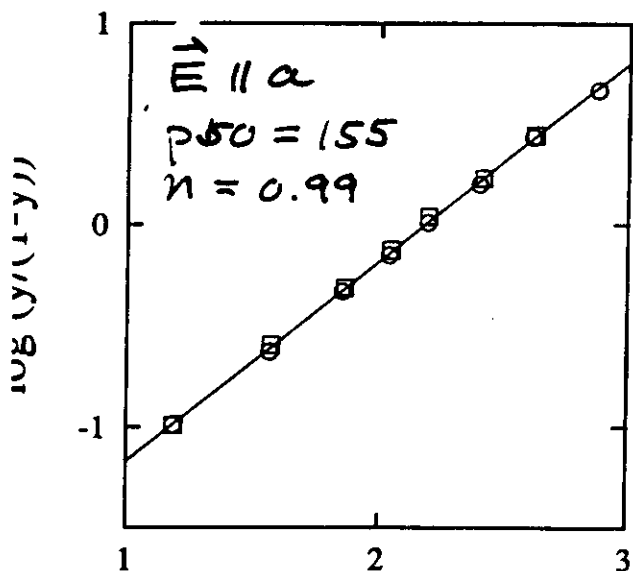
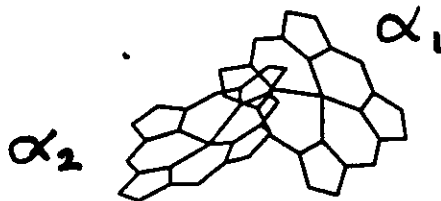
*Cooperativity in Hemoglobin -  
the Hydrogen Molecule of Biology*

*William A. Eaton*  
*Laboratory of Chemical Physics*  
*National Institutes of Health*  
*20892-0520 Bethesda*  
*U.S.A.*

- Mozzarelli (Parma)
- Rivetti (Parma)
- B. Henry (NIH)
- L. Rossi (Parma)

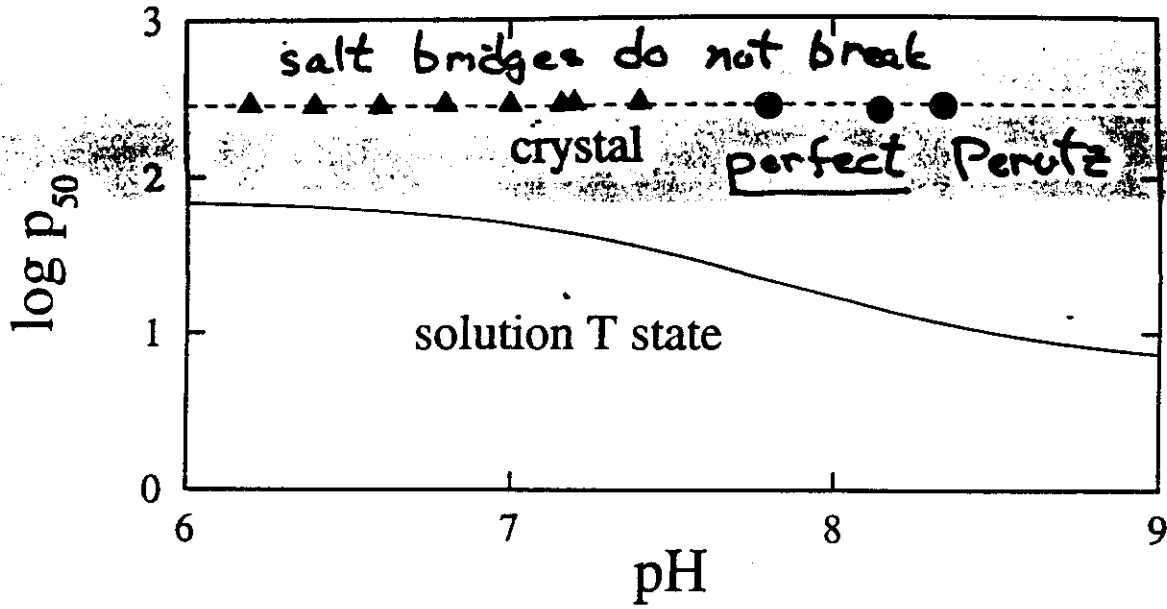


T quaternary

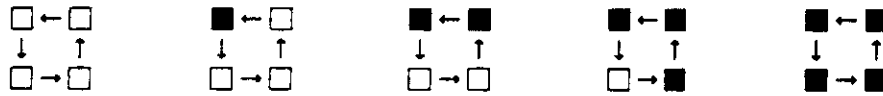


$$2 = \left[ 1 + (K_\alpha + K_\beta)p + SK_\alpha K_\beta p^2 \right]^2; \quad \text{For } n=1, q \equiv \frac{K_\alpha}{K_\beta}$$

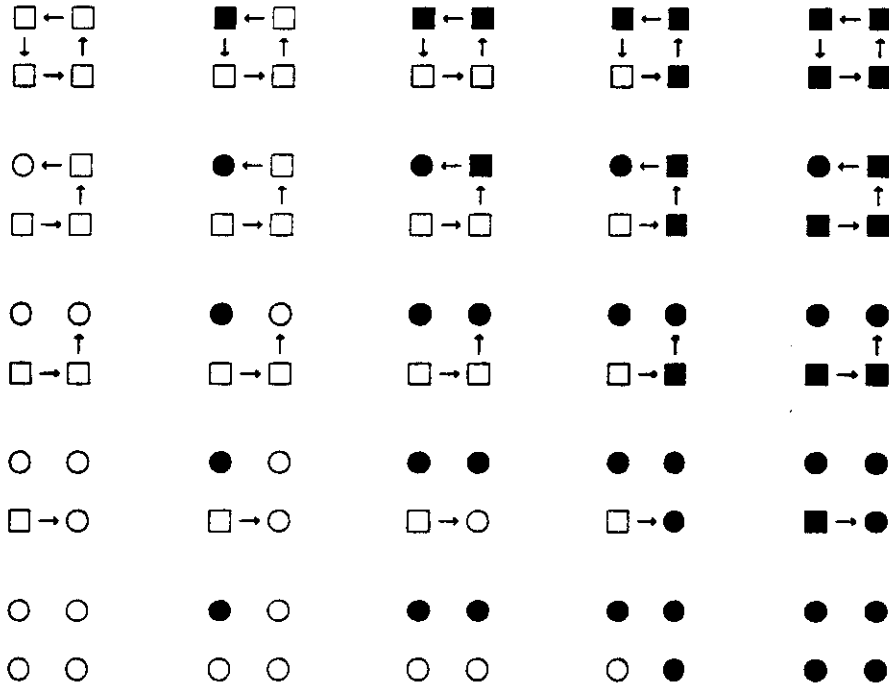
$$S = \frac{(q+1)^2}{4q}$$



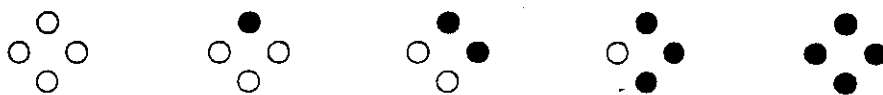
Crystal (T)

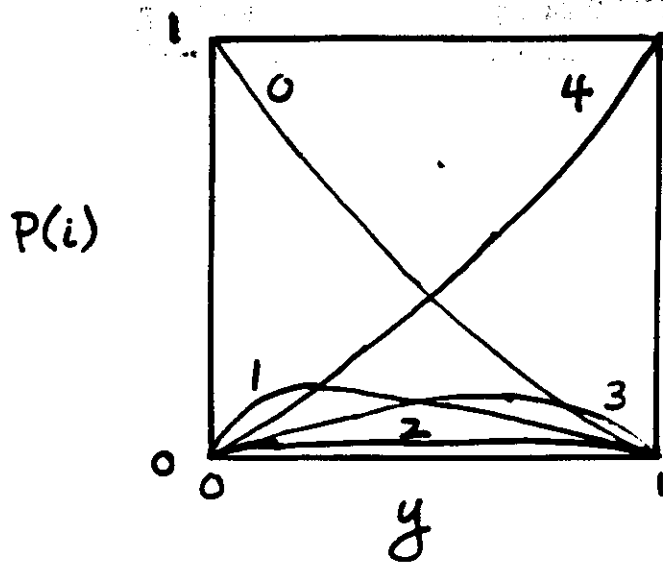


Solution (T)



Solution (R)

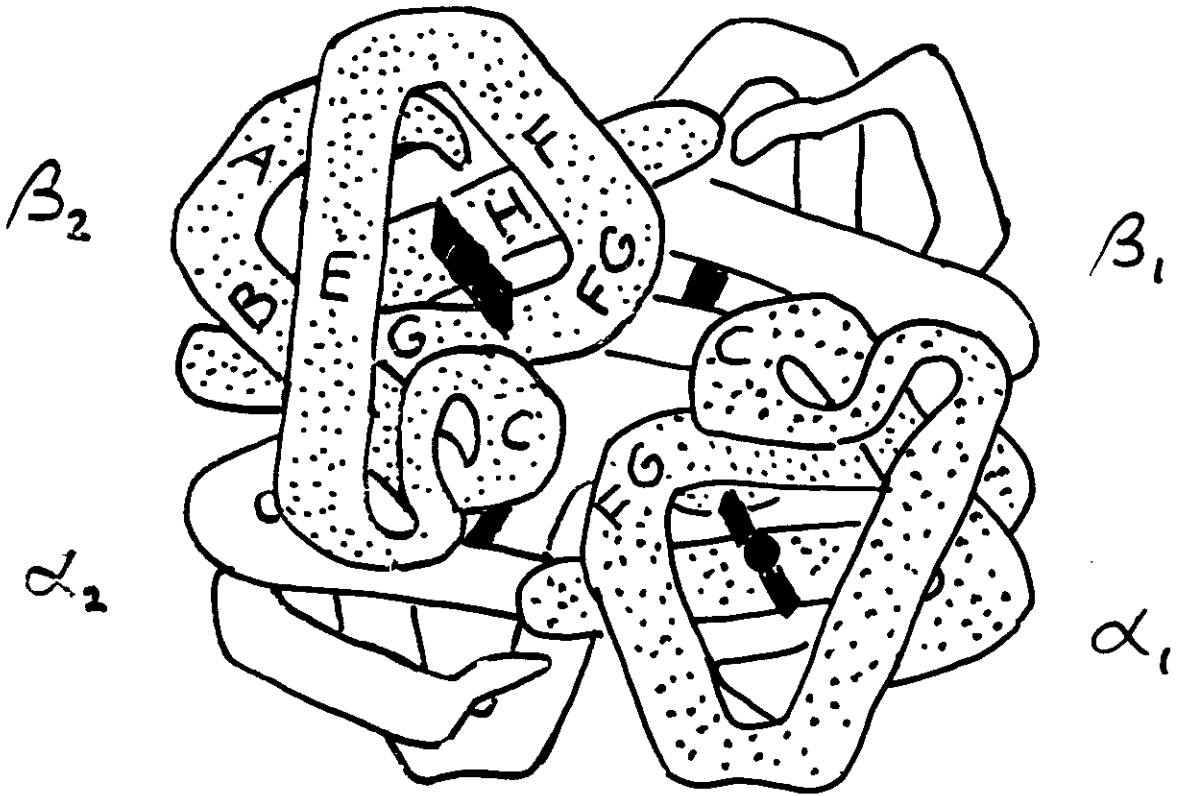




Intermediates at low population at equilibrium.

Choices:

1. Prepare artificial intermediates (can be misleading)
2. Prepare real intermediates by partial photolysis.



A

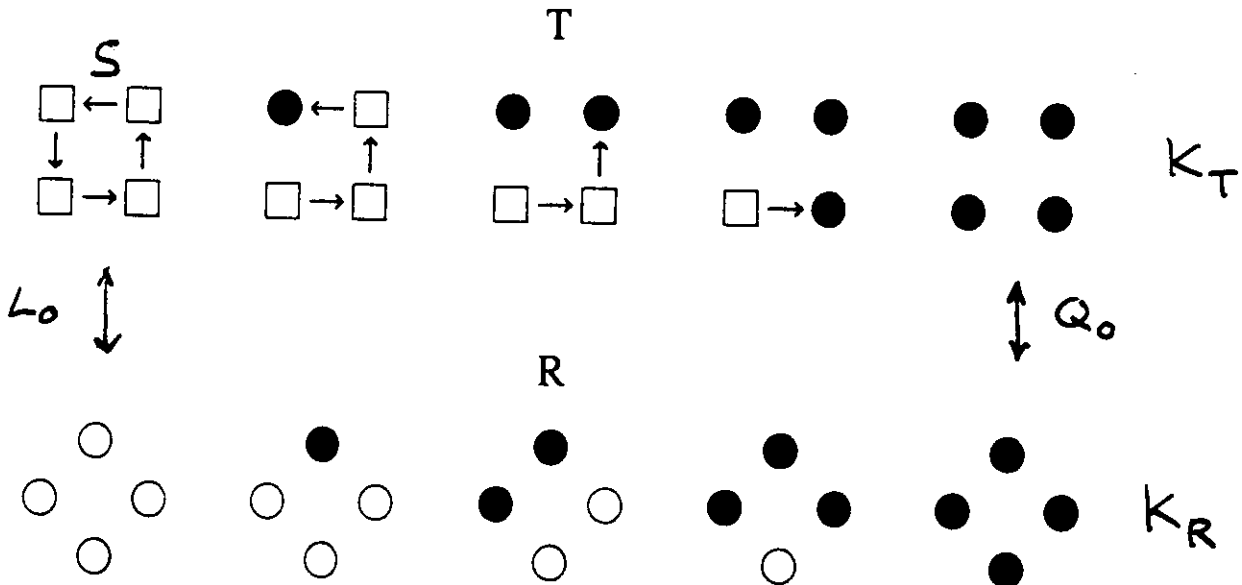
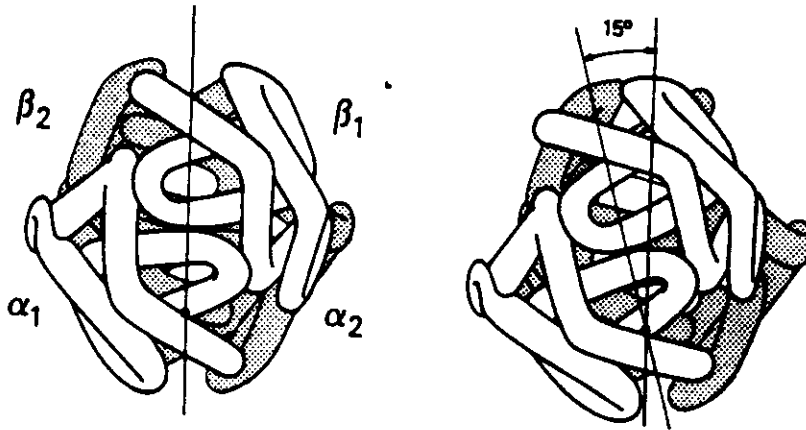
4

# MWC - PSK Model

Monod  
Wyman  
Changeux  
Perut  
Szabo  
Karplus

deoxy, T

oxy, R



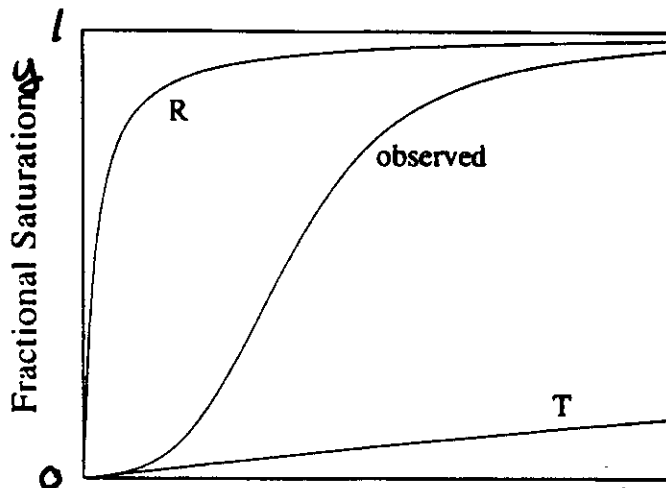
MWC

$K_R, K_T, L_0$

$L_i = L_0 c^i$

SK

$K_R, S, Q_0$

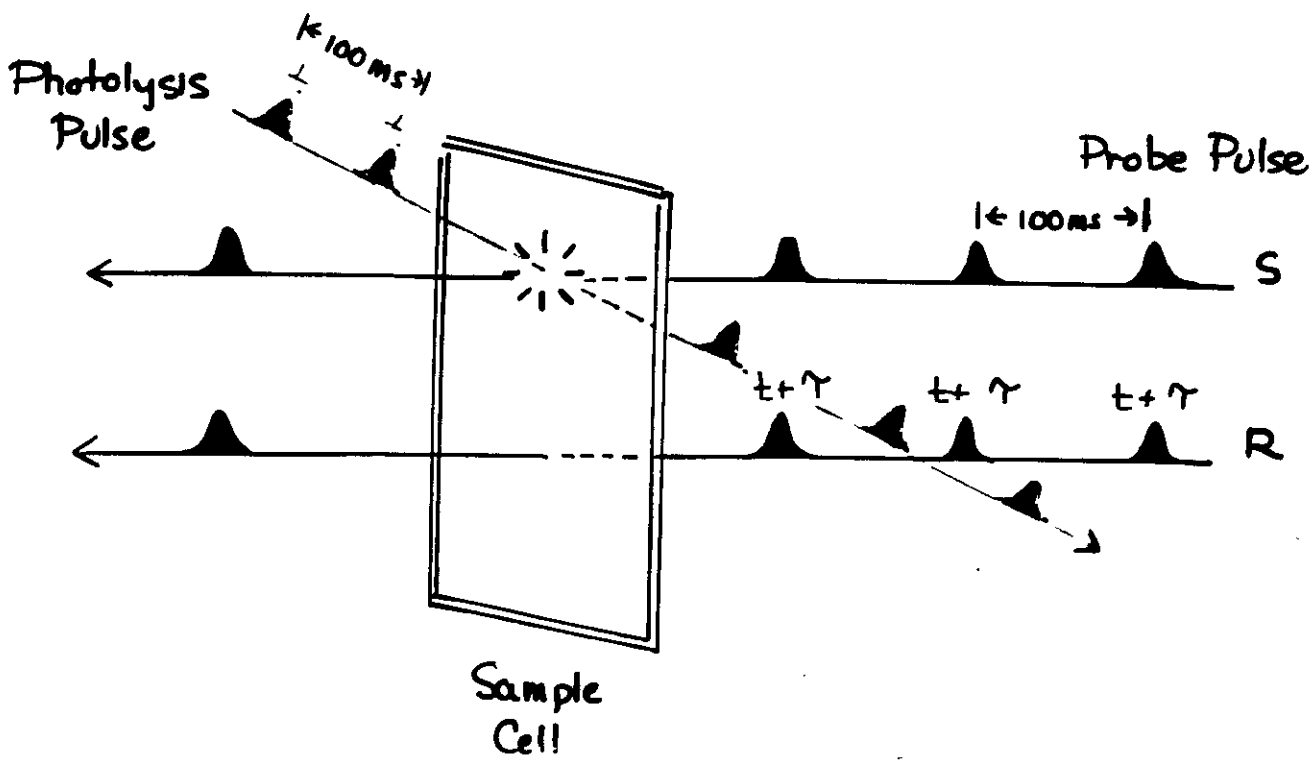
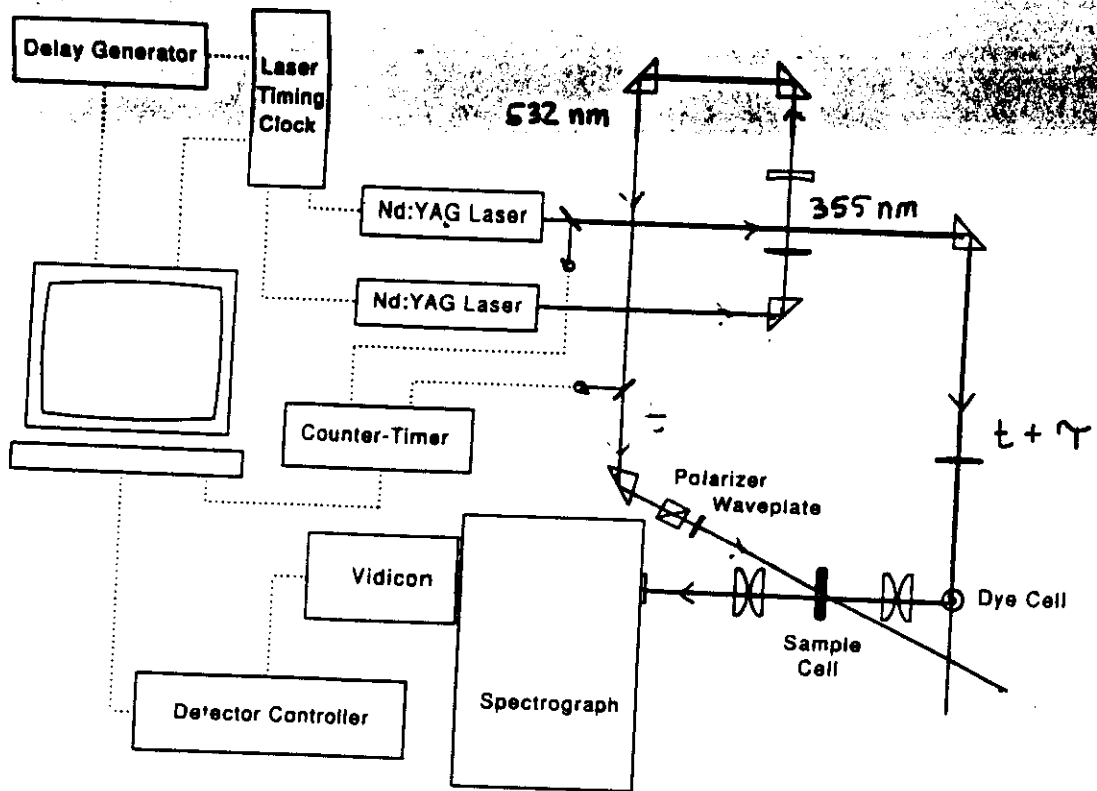


$$K_T = K_R / S$$

$$L_0 = Q_0 S^4$$

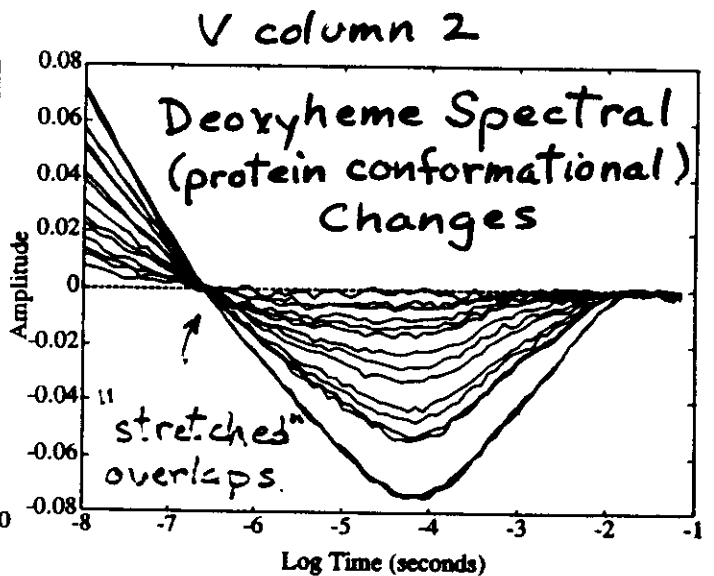
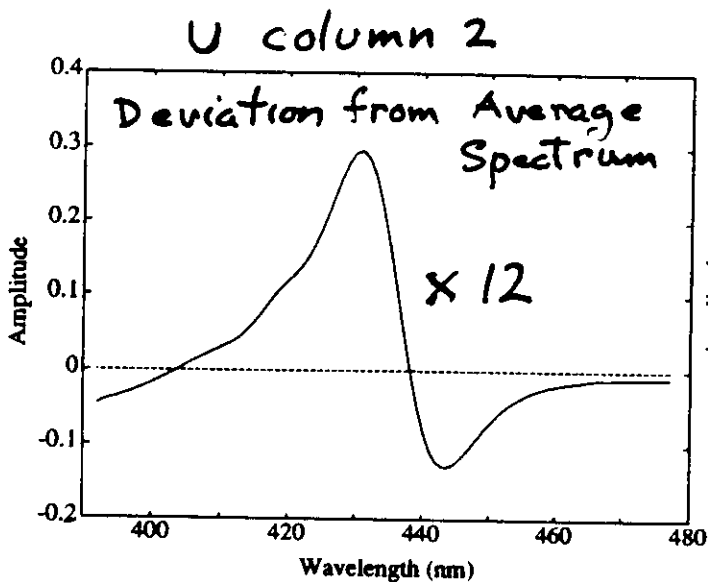
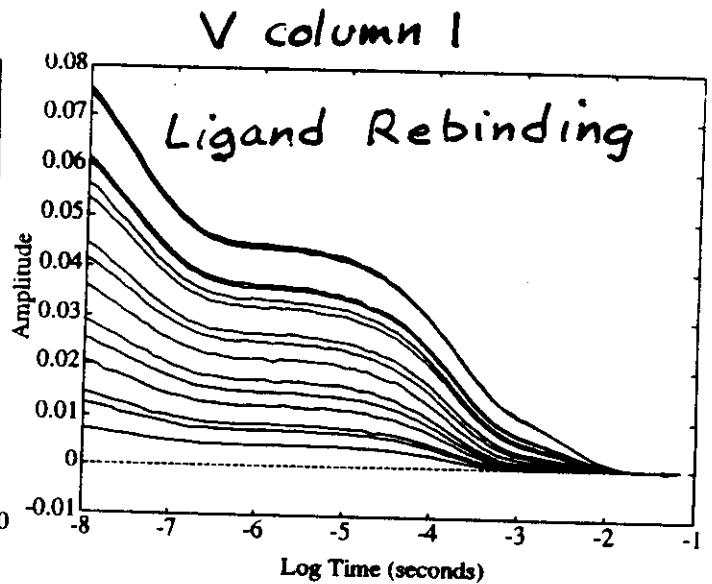
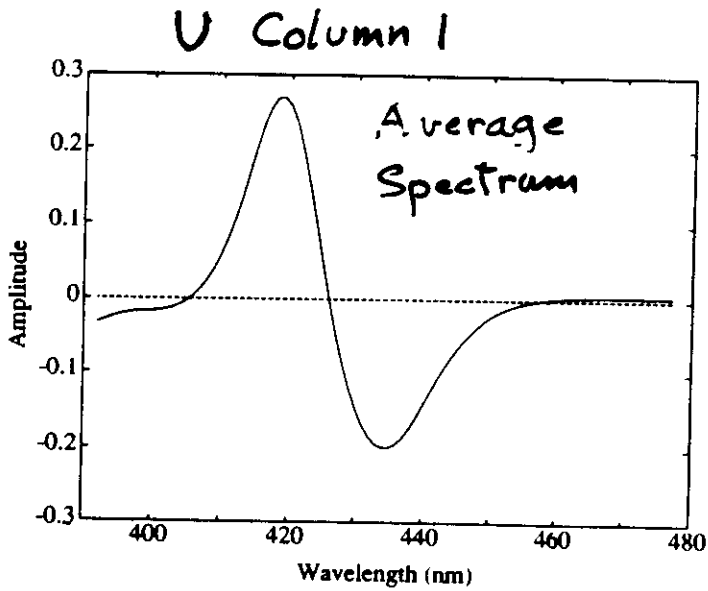
A

5



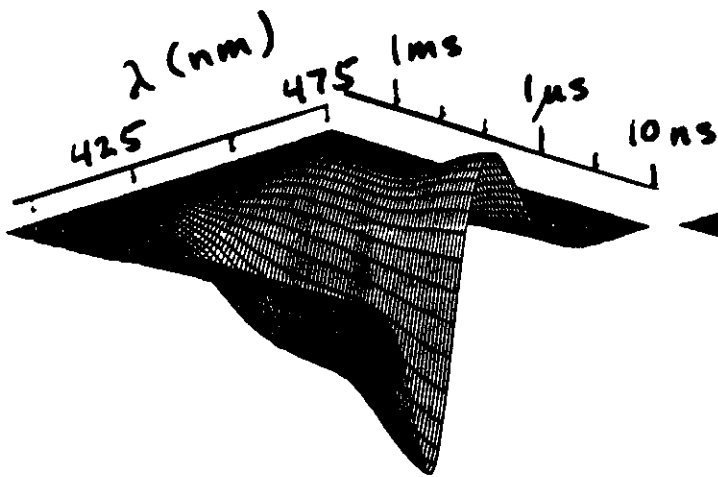
# Singular Value Decomposition (of isotropically-averaged data)

$$A = U S V^T$$



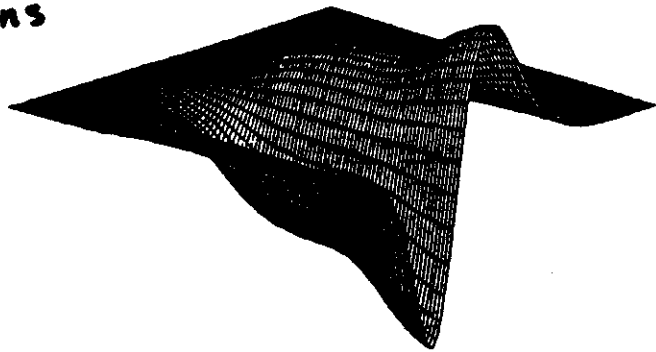


# Singular Value Decomposition (isotropic data)



data

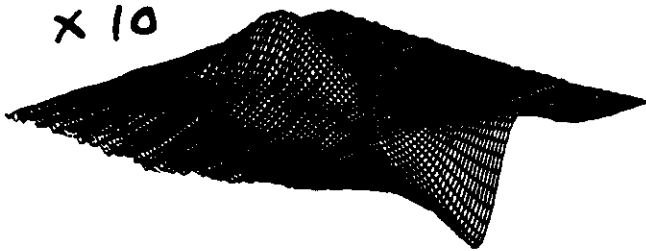
ligand rebinding



best one-component  
fit

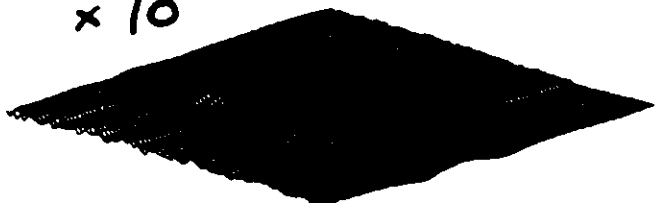
ceowheme  
(protein conformational)  
changes

x 10



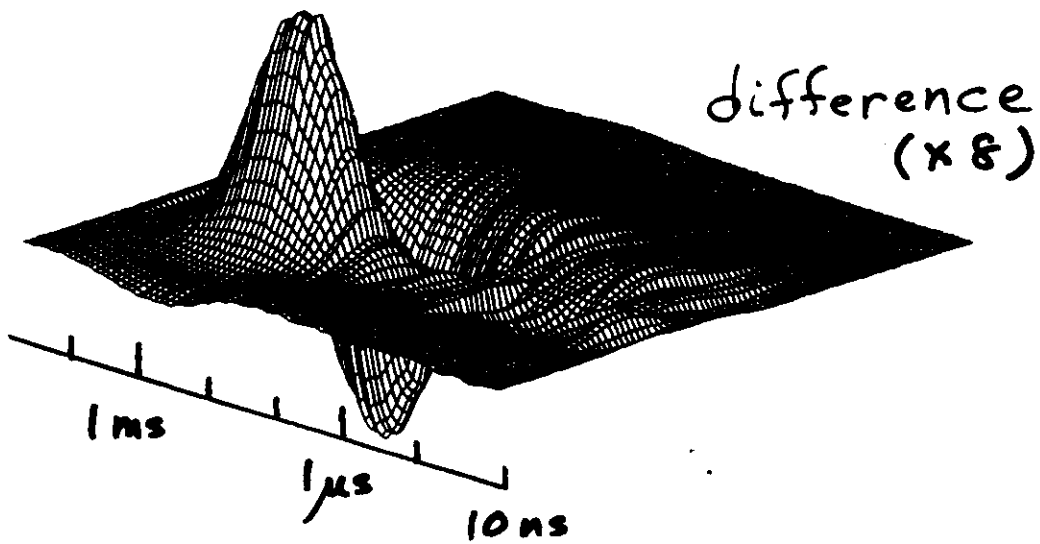
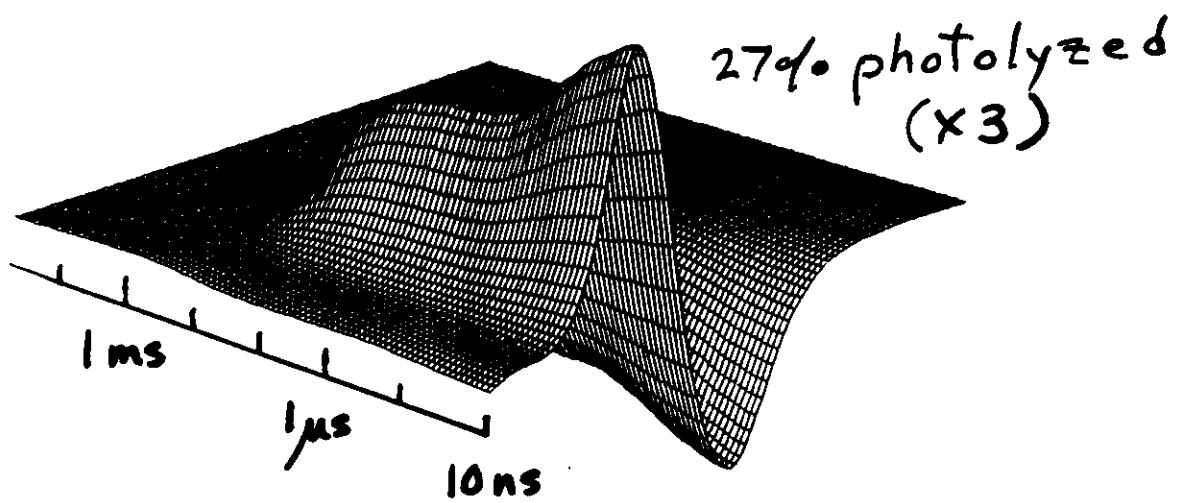
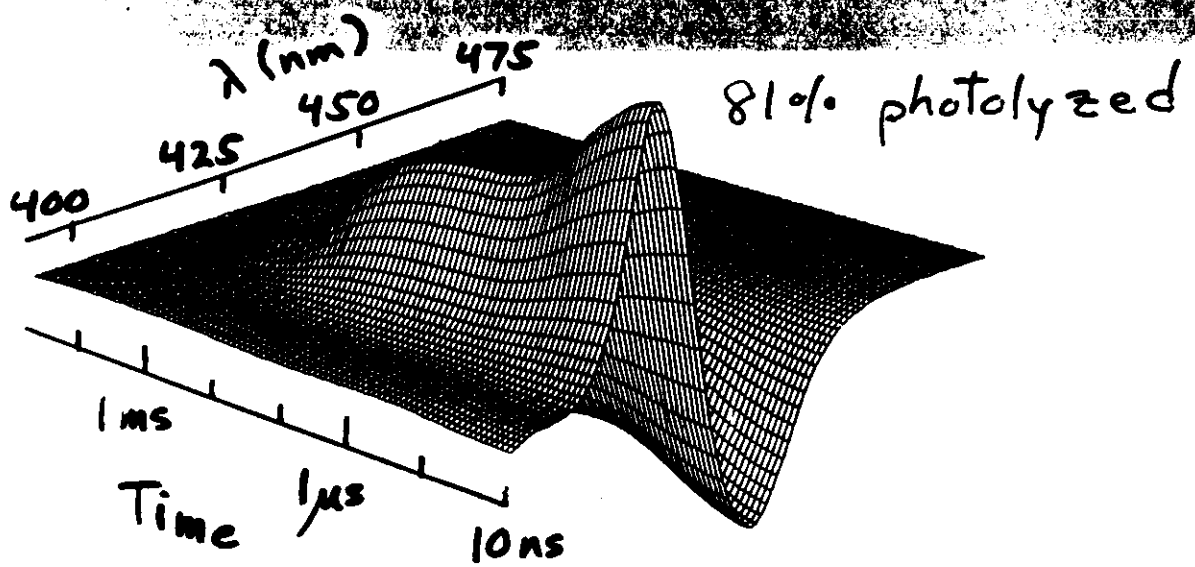
"residual"  
second-component

x 10



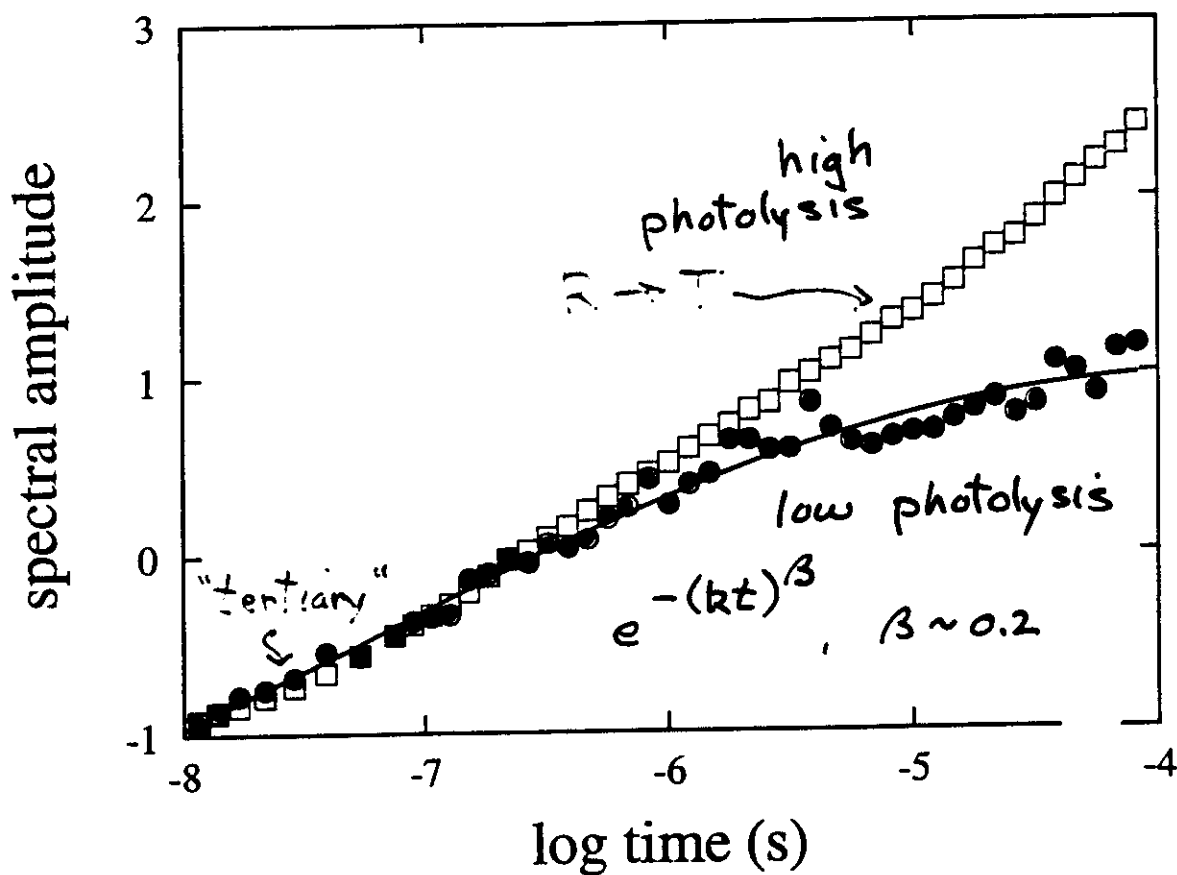
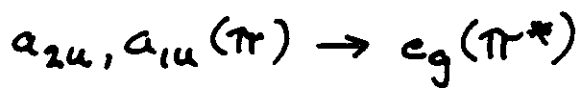
third component

# Isotropic Spectra



Colleen Jones (NIH)  
James Hofrichter (NIH)  
Eric Henry (NIH)

## Conformational Relaxation in Hemoglobin



"Stretched" exponential key to explaining kinetics in terms of a two-state allosteric model."

# Linear Free Energy Relation

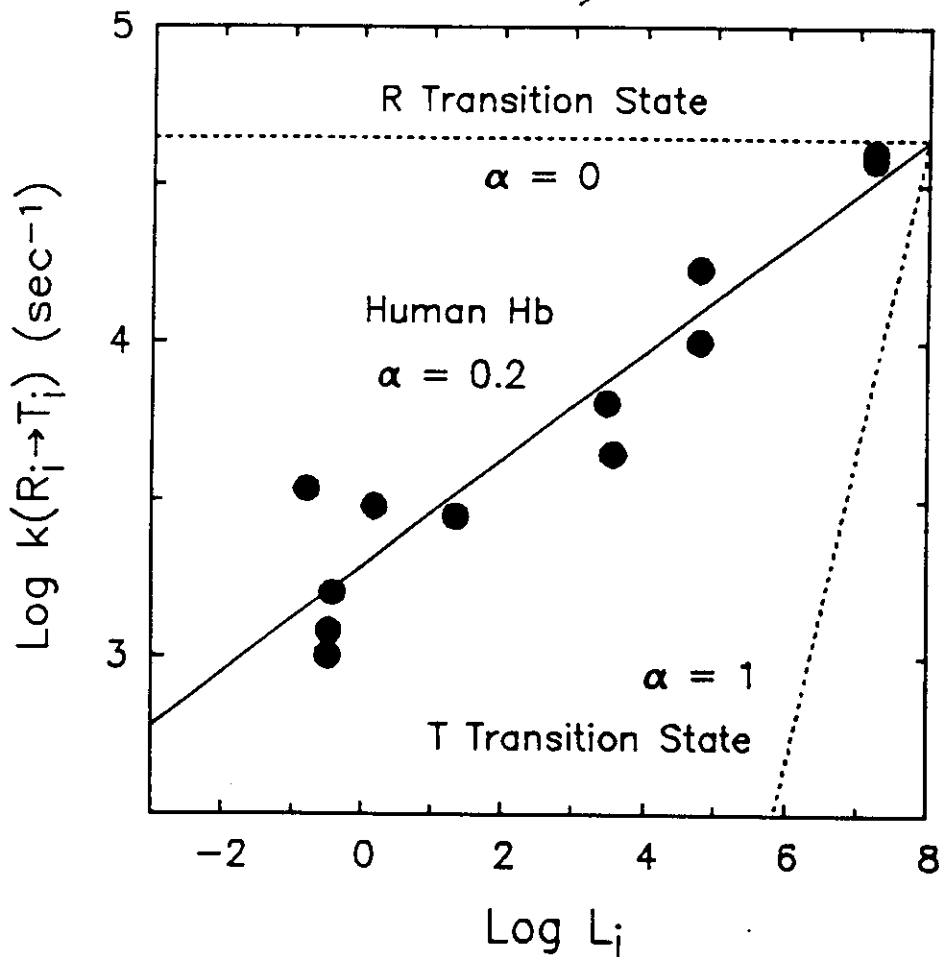
$$\delta \Delta G^\ddagger = \alpha \delta \Delta G$$

For two-state allosteric model

$$k(R_i \rightarrow T_i) = \gamma (L_0 c^i)^\alpha$$

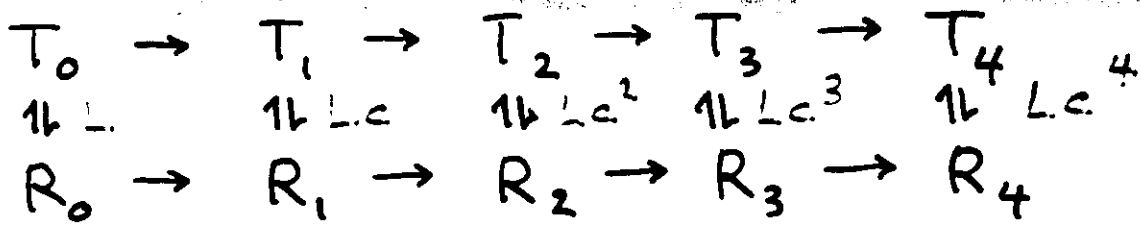
$$k(T_i \rightarrow R_i) = \gamma (L_0 c^i)^{\alpha-1}$$

$$(d = c^{-\alpha})$$

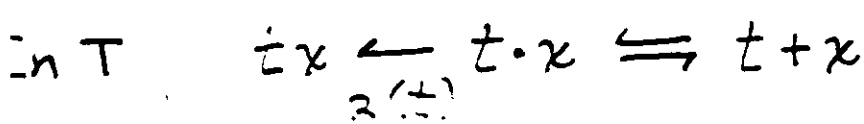
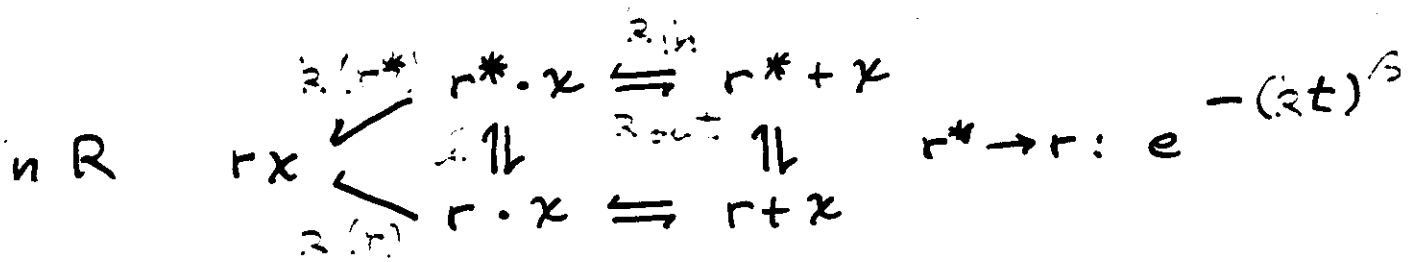


Transition state appears at 20% of the distance along the reaction path from R to T.

# MWC Model with Tertiary Transient



$$k(R_i \rightarrow T_i) = \gamma (Lc^i)^x \quad (\text{Gibson } d = c^{-x})$$



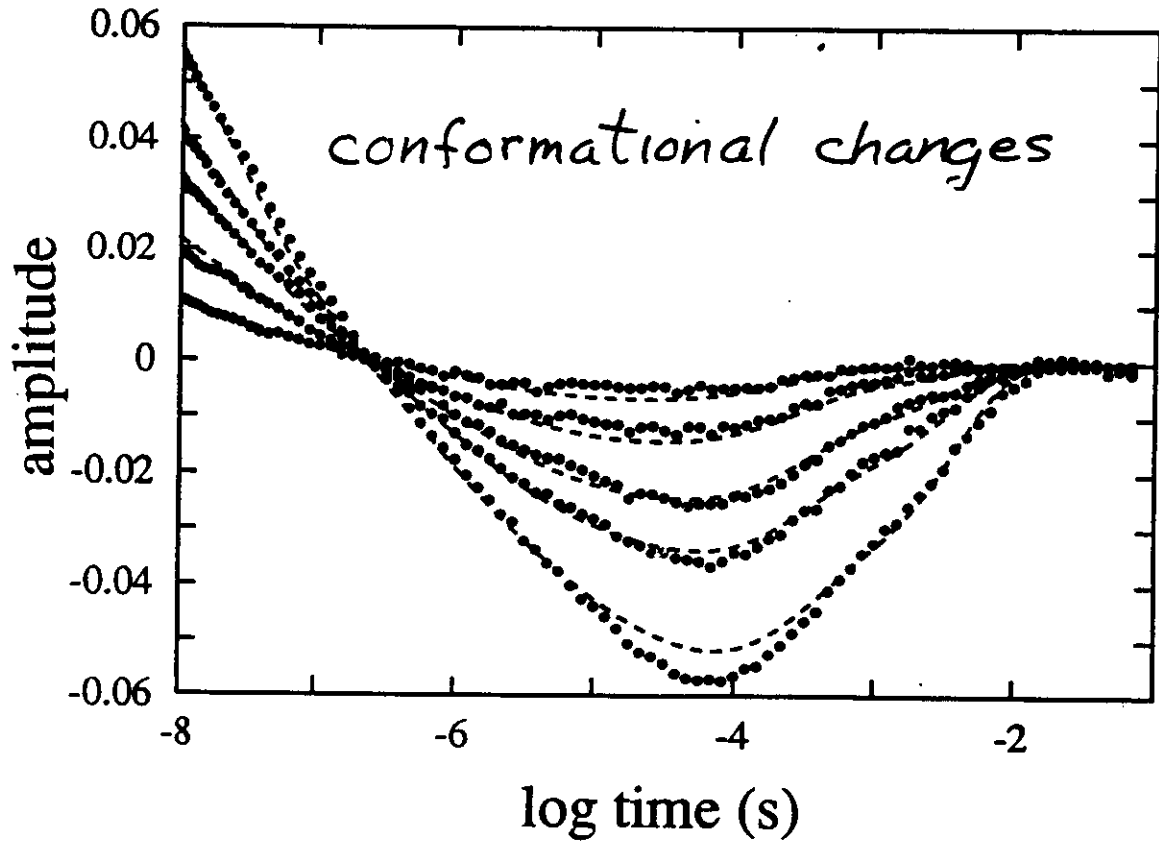
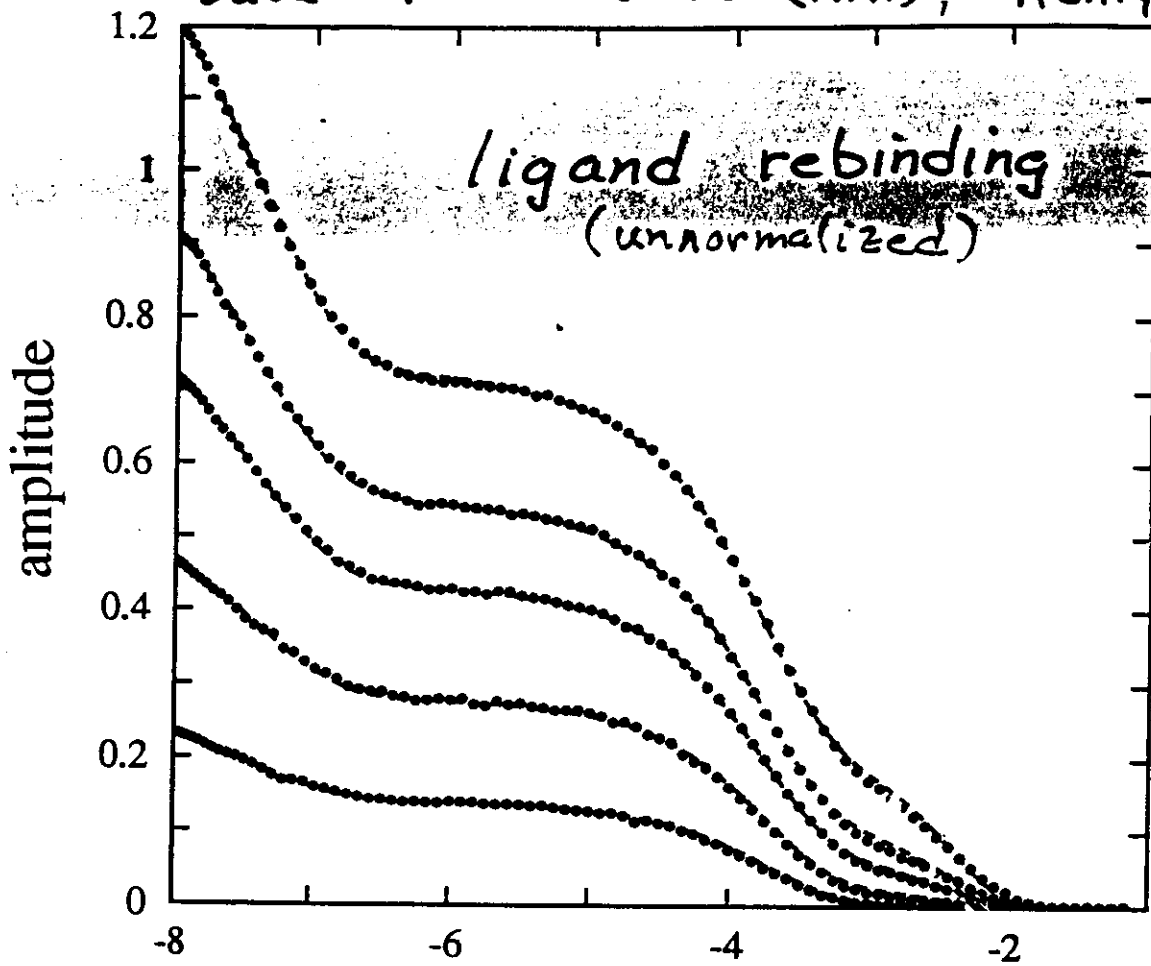
- $c, \gamma, x$  quaternary rates
- $k, \beta, \alpha$  tertiary rates
- $k(r^*), k(r), k(t)$  ligand binding rates
- $k_{in}, k_{out}$

85 distinguishable species, 3 deoxy spectra  
12 model parameters  
 to describe

37 experimental parameters, 2 deoxy spectra

Parameter space explored by a Monte-Carlo  
 "simulated annealing" ( $\chi^2$ ).

Data of C.M. Jones (NIH), Henry fit



## Take Home Lessons

1. MWC - PSK (Hopfield / Shulman / Ogawa / Edelstein) got it right!
2. Hemoglobin remains paradigm for cooperative behavior in multisubunit proteins.
3. Settled by new experiments

low-tech: single crystal oxygen binding  
medium-tech: nanosecond spectroscopy  
hi-tech: femtosecond spectroscopy

massive computing power

new ideas (stretched exponential)

