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Conference on Single Molecule Magnets and Hybrid Magnetic Nanostructures

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Bulk Transport Properties of Single Molecule Magnets and other interesting Materials

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These are preliminary lecture notes, intended only for distribution to participants

### Bulk Transport Properties of Single Molecule Magnets and other Interesting Materials

**Dr. Naresh Dalal** 





### Collaborators

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- Professor Andrew Kent
- Professor Dave Hendrickson
- Professor Jan Musfeldt
- Dr. Paul Kögerler (Ames Laboratory)
- MARTECH
- NSF/NIRT (Grant No. DMR 0103290)

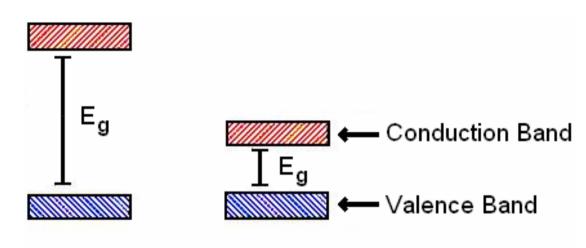
# **Transport Outline**

- Motivation
- Background and Techniques
- Compounds and Discussion
  - SMM (Mn<sub>12</sub>-Ac and Fe<sub>8</sub>Br<sub>8</sub>)
  - $-V_{12}$  and  $V_{15}$
- Conclusions
- Future Work

### Motivation

- Magnetic Semiconductors are rare
- Spintronics Applications
- Band Structure Characterization

### **Semiconductors**

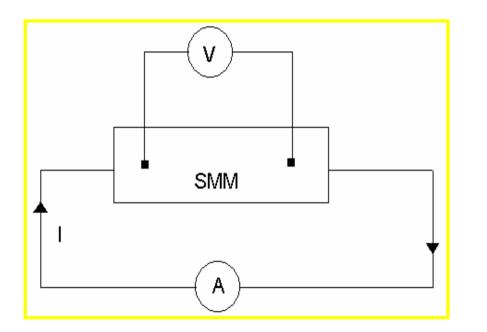


Insulator

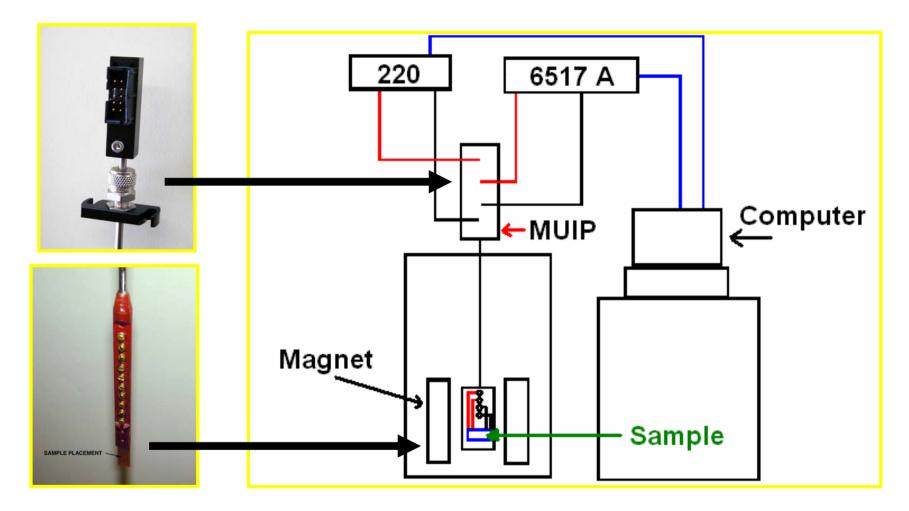
Semiconductor

- Bomprised of a filled valence band and an empty conduction band.
- C. Kiftel, Solid State Physics:
  E<sub>g</sub> represents the optical band gap
  g = 1/2E + 3/4k, T In (m,/m)
  E<sub>a</sub> represents the transport band gap
- E for a semiconductor is typically around 1eV Assuming  $m_h = m_e \rightarrow 2E_a \approx E_a$

### Four Probe Technique for *dc* Resistivity



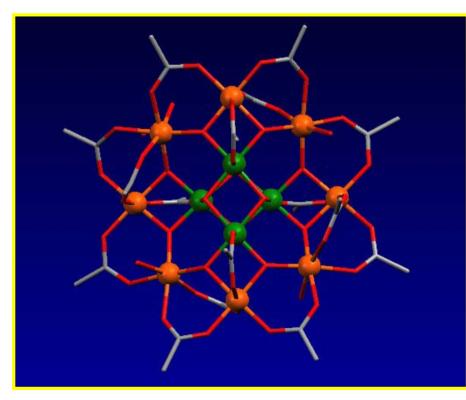
- Four probe technique for constant current measurements (FIMV)
- Two probe for constant voltage measurements (FVMI)
- Constant currents typically 0.1 nA – 10 nA
- Constant voltages typically  $\approx 100 \text{ V}$

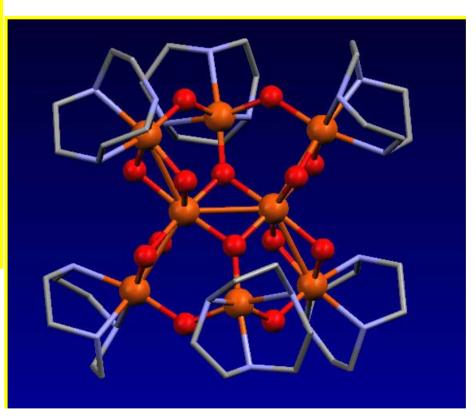


- Keithley 220 low level current source
- Keithley 6517A High resistivity electrometer
- SQUID gives precise temperature control, inert atmosphere, and the ability to apply magnetic fields

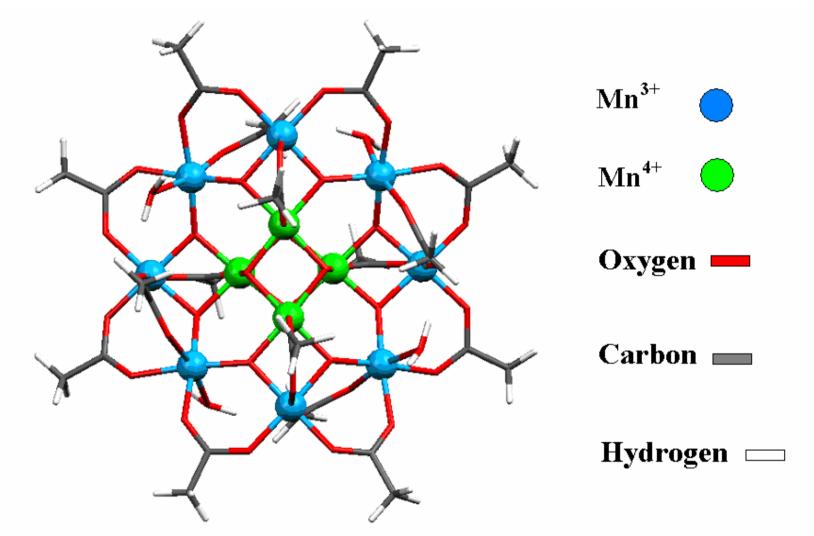


# Semiconductive Behavior of Mn<sub>12</sub> and Fe<sub>8</sub>

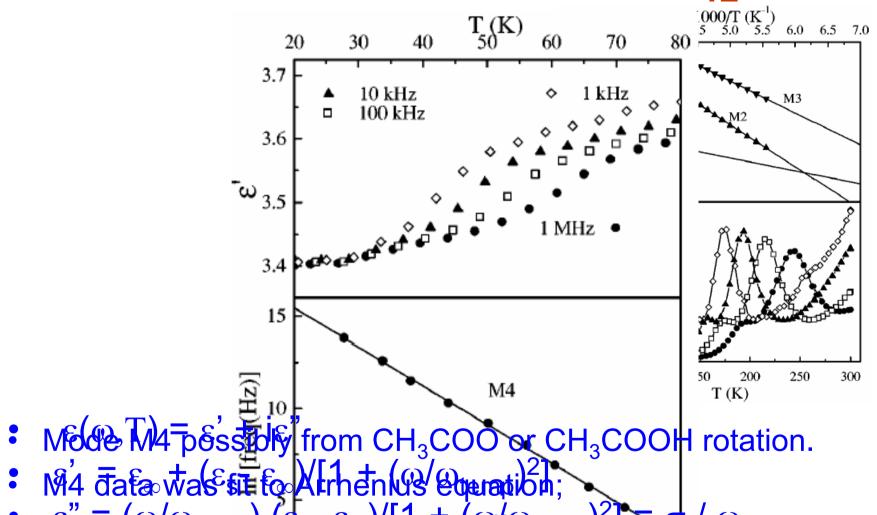






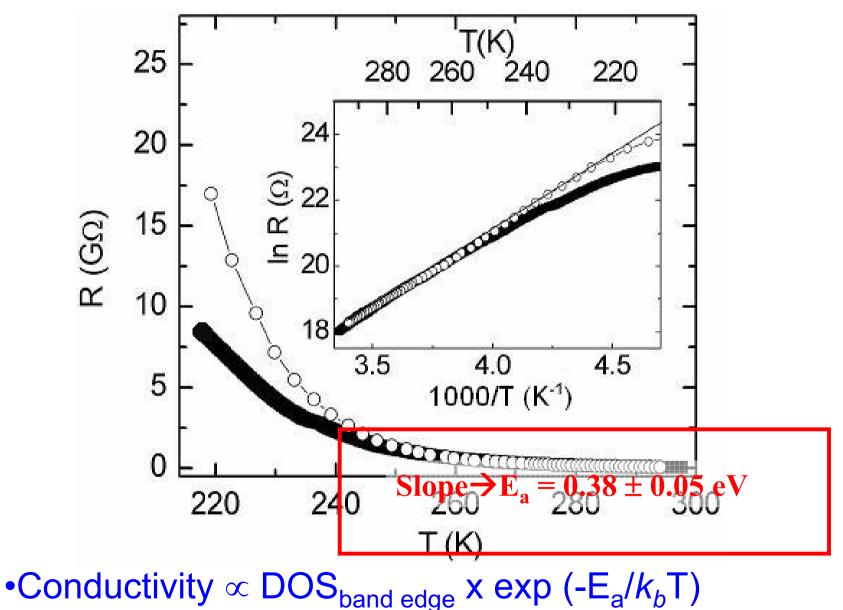


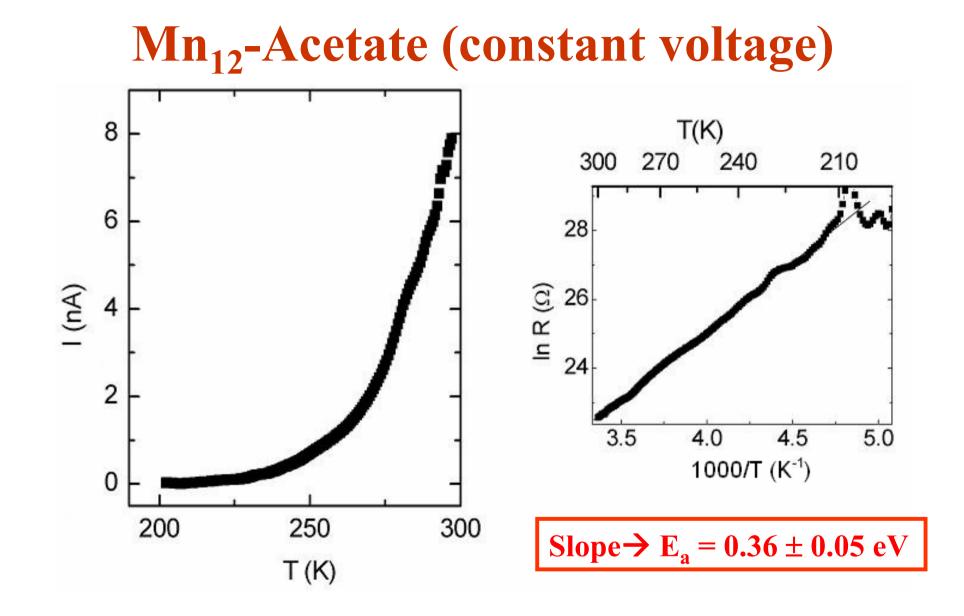
### **Dielectric Studies of Mn<sub>12</sub>-ac**



- $\mathcal{E}_{\overline{0}e^{(4)}}^{2} = \mathcal{E}_{A} / [1.0 \times (0^{4} \mathcal{O}_{E^{-1}})^{2}] = \mathcal{E}_{A} / \mathcal{O}_{C^{-1}} = \mathcal{O}_{A} / \mathcal{O}_{A} / \mathcal{O}_{A} / \mathcal{O}_{A} / \mathcal{O}_{A} = \mathcal{O}_{A} / \mathcal{O}_{A$
- Other Modes Stelten hand hand be start to the start of th

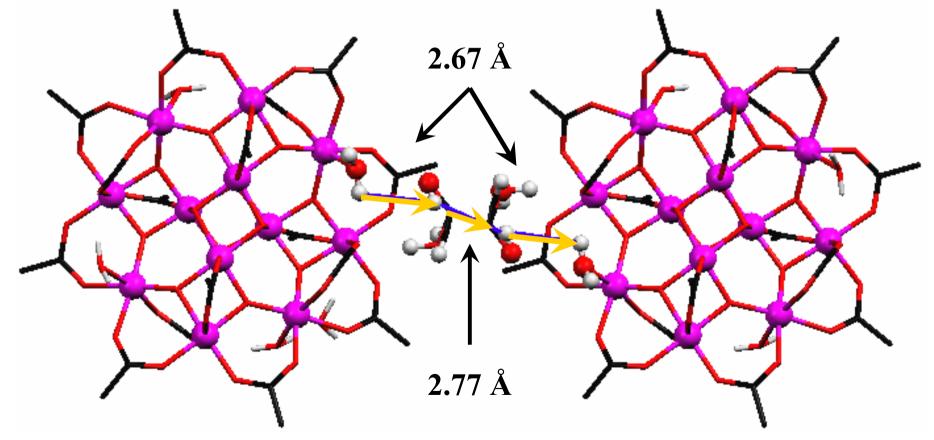
### Mn<sub>12</sub>-Acetate (constant current)





North, et al. Phys. Rev. B 67, 174407 (2003).

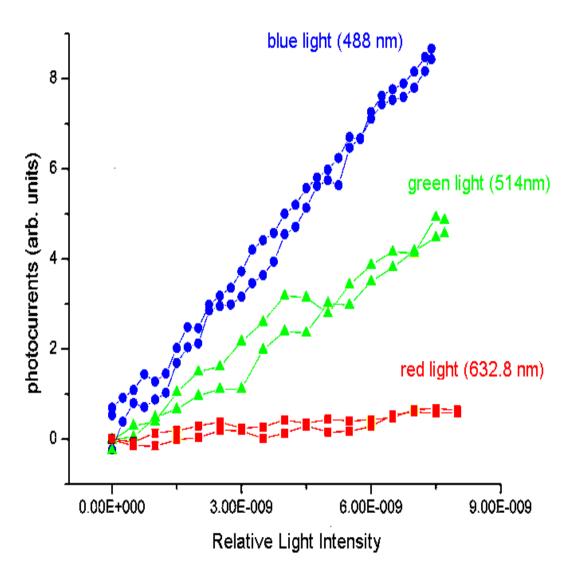
### Possible Conduction Pathway for Mn<sub>12</sub>-ac

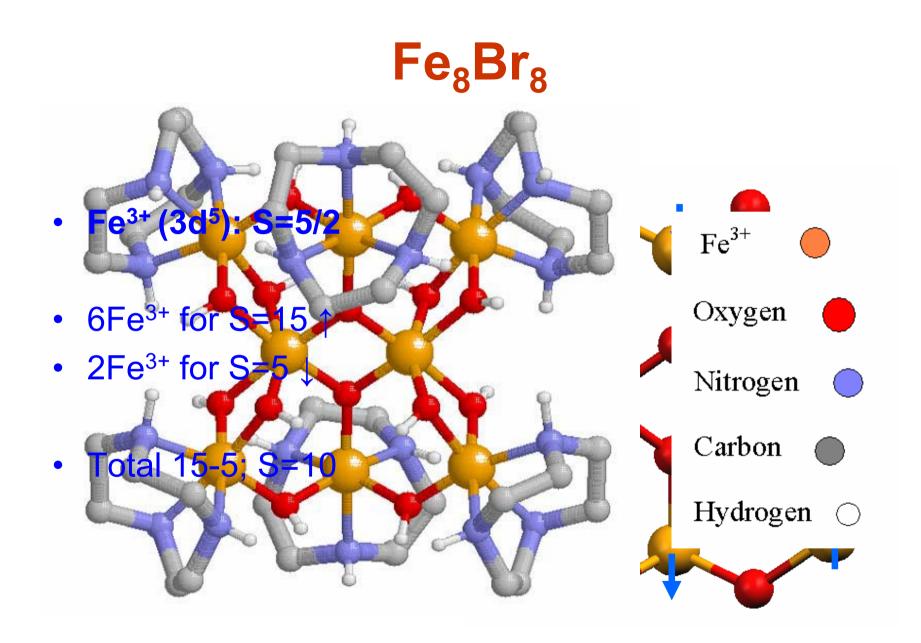


 $Mn-H_2O \rightarrow acetate \rightarrow acetate \rightarrow H_2O-Mn$ 

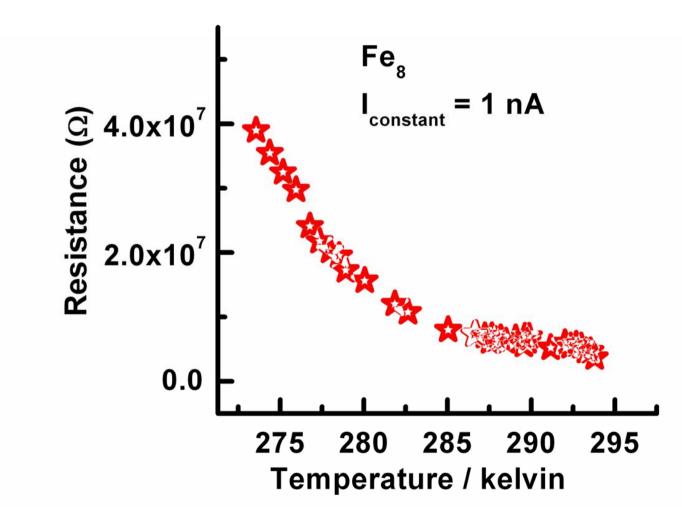
#### **Photoconductivity of Mn<sub>12</sub>-ac**

- A factor of about eight increase in the photoconductivity is observed when going from 632.8 nm to 488 nm.
- Optical data show only a factor of two increase in absorption over this range.
- This rules out a direct heating effect.



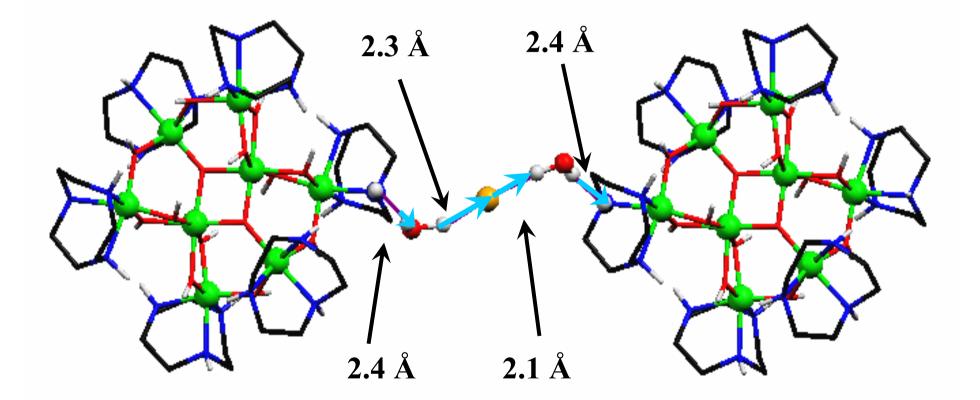


### **Resistivity of Fe<sub>8</sub>**



North, et al. Phys. Rev. B 67, 174407 (2003).

### Possible Conduction Pathway for Fe<sub>8</sub>Br<sub>8</sub>

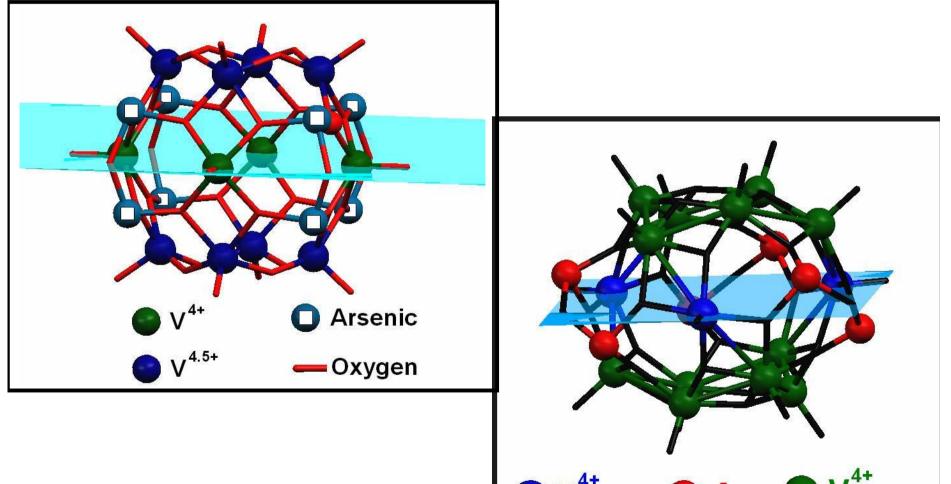


 $\mathbf{Fe} - \mathbf{N} - \mathbf{H} \rightarrow \mathbf{H}_2 \mathbf{O} \rightarrow \mathbf{Br} \rightarrow \mathbf{H}_2 \mathbf{O} \rightarrow \mathbf{H} - \mathbf{N} - \mathbf{Fe}$ 

### **SMM Conductivity Summary**

- Mn<sub>12</sub>-ac and Fe<sub>8</sub> show semiconductor behavior with band gaps of ~0.7 eV and ~1.5 eV respectively, both could be classified as intrinsic semiconductors because  $E_a \approx \frac{1}{2} E_g$
- $Mn_{12}$ -ac exhibits photoconductivity in visible region.  $Fe_8Br_8$  to be checked.
- Proposed conduction pathways are consistent with the higher activation energy found in  $Fe_8Br_8$ .
- Density of States near band edges for both materials is quite small.

# Semiconductive Behavior of $V_{12}$ and $V_{15}$

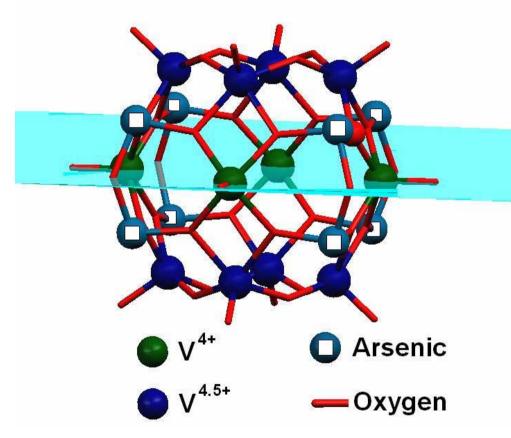


As

triangle

hexagon

# $[NH(C_{2}H_{5})_{3}]_{4}[V_{8}^{4.5+}V_{4}^{4+}As_{8}O_{40}(H_{2}O)] \bullet H_{2}O$ $(V_{12})$

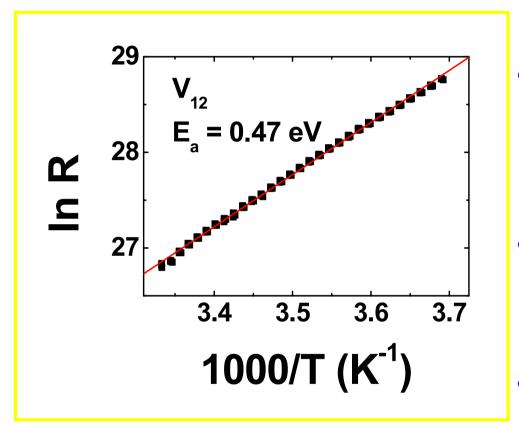


Pope *et al. Angew. Chem., Int. Ed. Engl.* **30**, 32 (1991).

• S = 0

- Two d electrons delocalized amongst four vandiums in top and bottom layers
- Four V<sup>4+</sup> ions in middle layer
- Weak antiferromagnetic interactions (~18 K) between V<sup>4+</sup> ions

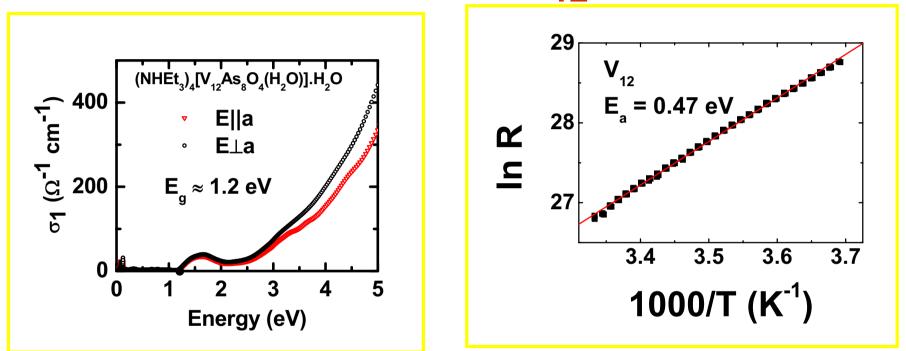
### **Resistivity of V**<sub>12</sub>



- Constant voltage configuration, V<sub>const</sub> = 250 V T = 270 - 300 K
- Constant Current configuration,

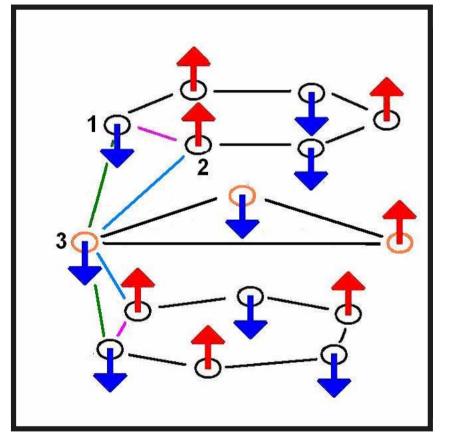
 $E_a = 0.48 \pm 0.05 \text{ eV}$ 

### Optical Band Gap vs. Transport Gap for V<sub>12</sub>



- In an intrinsic semiconductor  $E_{\alpha} \approx 2E_{a}$
- $E_g \approx 1.2 \text{ eV} \approx 2 \text{ (0.48 eV)}$

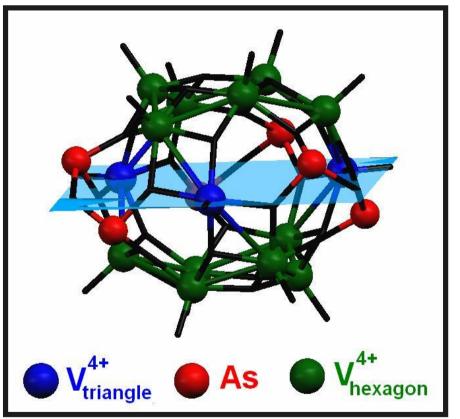
# $K_{6}[V_{15}As_{6}O_{42}(H_{2}O)] \cdot 8H_{2}O$



Müller *et al. Angew. Chem., Int. Ed. Engl.* **27,** 1721 (1988).

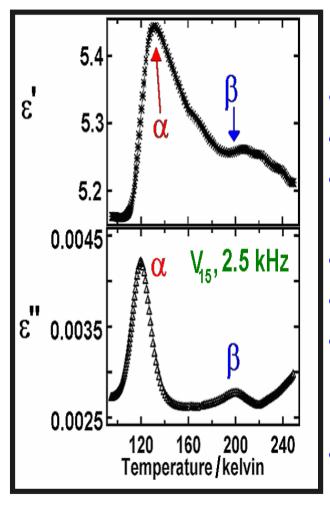
- 15 V<sup>4+</sup> ions all antiferromagnetically coupled
- $J_{1-2} >> J_{2-3} > J_{1-3}$
- Equilateral triangle of s = 1/2 ions
- Ground state doublets split (0.05 K) by Dzyaloshinskii-Moriya interactions - QTM

# **Dielectric Relaxation of V**<sub>15</sub>



- Determine the dielectric constant for theoretical work
- Dielectric relaxation studies give insight into low frequency motional dynamics
- Ac impedance bridge technique (SR830 DSP Lock-in Amplifier)
- 77 330 K; 500 Hz 250 kHz; 1 V drive amplitude

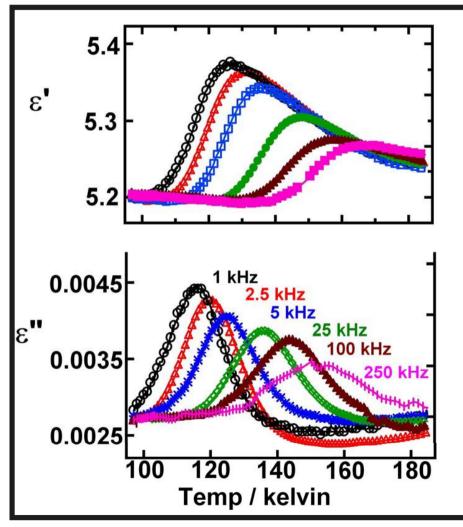
# **Dielectric Relaxation**



Modes

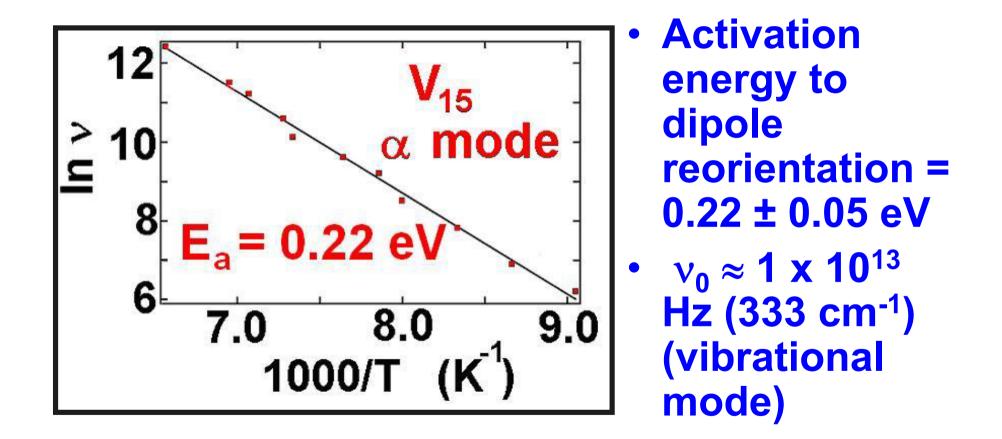
- $\varepsilon(\omega,T) = \varepsilon' + i\varepsilon''$
- $\varepsilon' = \varepsilon_{\infty} + (\varepsilon_{s} \varepsilon_{\infty}) / [1 + (\omega/\omega_{temp})^{2}]$  $\varepsilon'' = (\omega/\omega_{temp}) (\varepsilon_{s} - \varepsilon_{s}) / [1 + (\omega/\omega_{temp})^{2}]$ 
  - $\varepsilon'' = (\omega/\omega_{\text{temp}}) (\varepsilon_{s} \varepsilon_{\infty}) / [1 + (\omega/\omega_{\text{temp}})^{2}]$  $= \sigma / \omega$
- If  $(\omega/\omega_{\text{temp}}) << 1$ , then  $\varepsilon' = \varepsilon_s$ 
  - $\varepsilon_s = (C \times d) / (A \times \varepsilon_0) = 6 \pm 1$  $\alpha$  peak well fit by Gaussian
  - function for all experimental frequencies
- $\epsilon_{s}$  relatively temperature independent

# Temperature Dependence of $\alpha$ mode

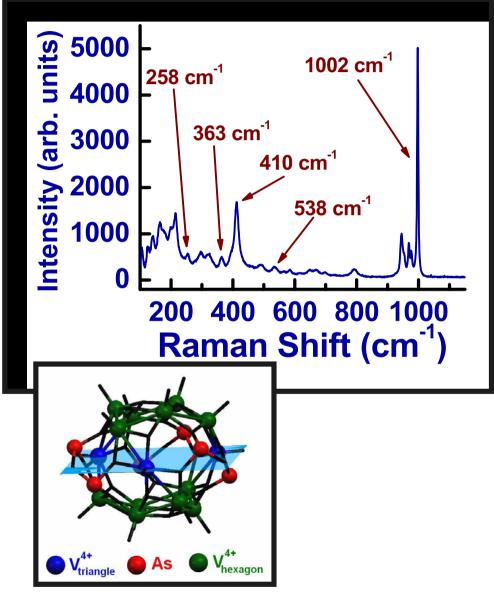


- Peak in ε" corresponds to
  - $\omega = \omega_{\text{temp}}$
- Step-like decrease of ε' and ε" indicative of a slowing of the dielectric relaxation with decreasing temperature

### **Arrhenius Analysis**

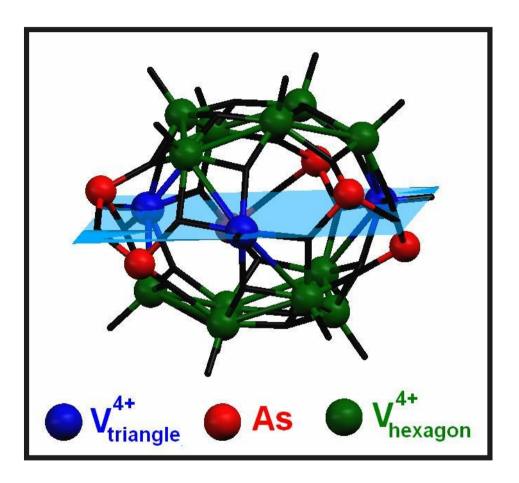


# Vibrational Modes of V<sub>15</sub>



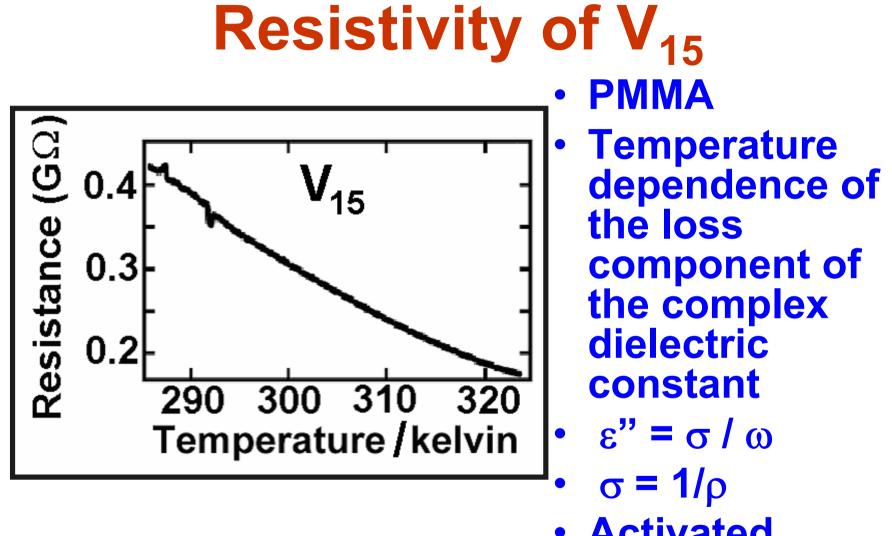
- 538 cm<sup>-1</sup> V O internal stretch
- 410 cm<sup>-1</sup> –
  O V O bend
- 363 cm<sup>-1</sup> As O As bend
- 258 cm<sup>-1</sup> –
  O As O bend
- R (120 K) ~ 1 TΩ
- Vibrational modes provide mechanism for dipole reorientation

# **Induced Polarization**



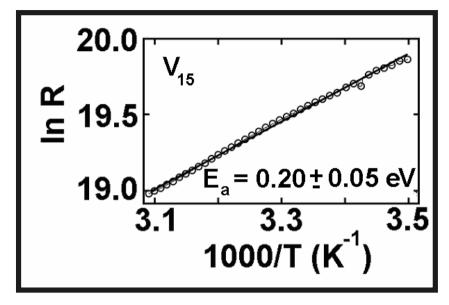
- ε<sub>s</sub> has little or no temperature dependence
- $\epsilon_{s} \propto (P + \mu^{2}/3k_{B}T)$
- No permanent dipole
- Vibrations responsible for dipole reorientation

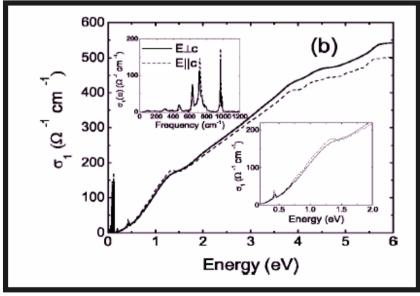
Onsager, J. Am. Chem. Soc. 58, 1486 (1936).



 Activated, Ohmic behavior above 285 K

# **Transport Gap**





• R(T) ~ exp(E<sub>a</sub>/kT)

• 
$$E_a \approx \frac{1}{2} E_g$$

•  $E_a = 0.2 \pm 0.05 \text{ eV}$ 

• 
$$E_g \approx 0.5 \text{ eV}$$

J. Choi *et al., Phys. Rev. B* **68**, 064412 (2003).

## Summary

- V<sub>15</sub> shows two independent dielectric relaxation modes
- V<sub>15</sub> has a temperature-independent static dielectric constant of 6 ± 1
- The  $\alpha$  mode is facilitated by intramolecular vibrations and has an activation energy of E<sub>a</sub> = 0.22 ± 0.05 eV

## **Band Gap Summary**

	Conductivity	Optical	Theoretical
Mn <sub>12</sub> -Ac	0.74 $\pm$ 0.1 eV $^{a}$	1.08 eV <sup>b</sup>	0.45 eV <sup>c</sup>
		1.75 eV <sup>d</sup>	2.08 eV <sup>e</sup>
			0.85 eV <sup>f</sup>
			1.10 eV <sup>g</sup>
Fe <sub>8</sub> Br <sub>8</sub>	1.46 $\pm$ 0.2 eV $^{a}$		0.9 eV <sup>h</sup>
			0.9 eV <sup>i</sup>
<b>V</b> <sub>12</sub>	0.48 ± 0.05 eV	1.2 eV	
<b>V</b> <sub>15</sub>	0.2 ± 0.05 eV	0.5 eV	
<sup>a</sup> Present work, assuming $E_g = 2E_a$ . <sup>b</sup> Oppenheimer <i>et al.</i> (Ref. 44), minority spin cluster. <sup>c</sup> Pederson <i>et al.</i> (Ref. 45), minority spin cluster. <sup>d</sup> Oppenheimer <i>et al.</i> (Ref. 44), majority spin cluster.		<sup>e</sup> Pederson <i>et al.</i> (Ref. 45), majority spin cluster. <sup>f</sup> Zeng <i>et al.</i> (Ref. 46), minority spin cluster. <sup>g</sup> Zeng <i>et al.</i> (Ref. 46), majority spin cluster. <sup>h</sup> Pederson <i>et al.</i> (Ref. 50), minority spin cluster. <sup>i</sup> Pederson <i>et al.</i> (Ref. 50), majority spin cluster.	

### Conclusions

- All of these novel materials have semiconductor behavior
- Proposed mechanism for semiconductivity involves intermolecular electron hoping

# **Future Work**

- Increase of conductivity via doping or other methods.
- Comparison with conduction from single molecules.
- Angular variation of conductivity.

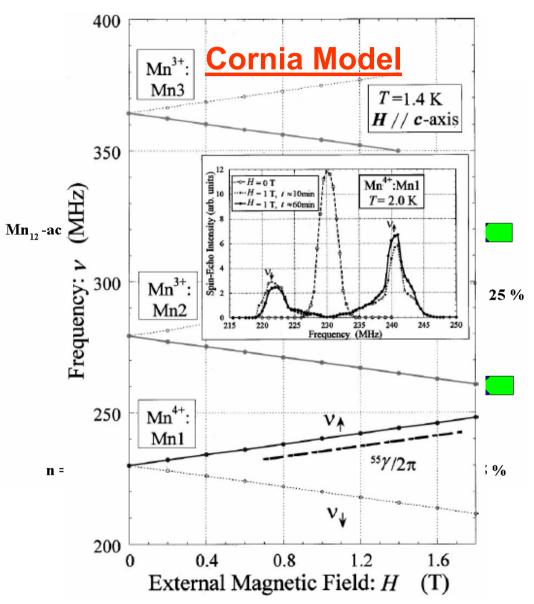
# **Recent Studies: NMR**

- Single Crystal vs. Powder NMR
- Angular dependence and field dependence studies.
- Need broad field and frequency sweeping capabilities.
- Possible because of locally developed instrumentation at the National High Magnetic Field Laboratory.



# Why NMR?

- Very sensitive to electrons and local magnetic environment
- Allows check of theoretical spin alignment models
- Compliments neutron scattering data
- Distinguishes between different isomers



# **NMR Difficulties**

- Strongly paramagnetic systems
- Small sample size (mm sized crystals)
- Spectral width up to ~30MHz
- Short T<sub>1e</sub> (nanoseconds)
- Low T<sub>b</sub> (Below 3K)
- No commercial instruments

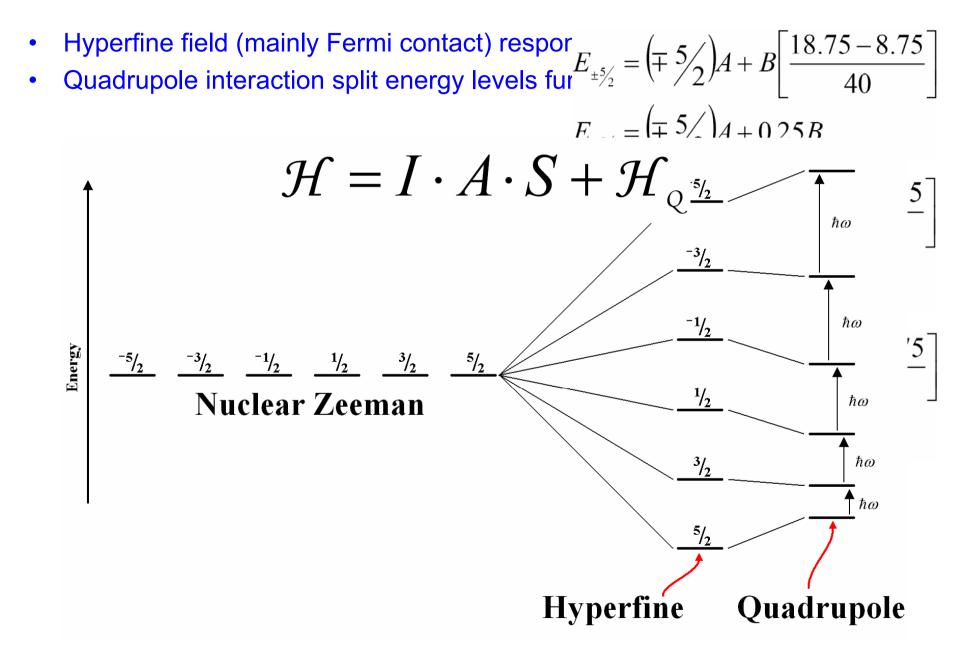
# **Aligned Powder Sample Preparation**

- Sample preparation:
  - Crush sample
  - Mix with Stycast Epoxy 1266
  - Place in a tape form
  - Set in field of 8.5 T overnight
- Possible problems:
  - Pressure and heat from crushing
  - Loss of solvent molecules

# **Single Crystal Sample Preparation**

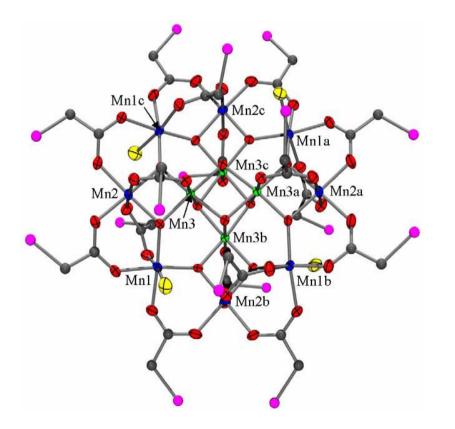
- Single crystals
  - Removed from mother liquor
  - Immediately covered in 5 minute epoxy
  - Cut shape out of epoxy
  - Wrap in Teflon tape

# **Hyperfine and Quadrupole**

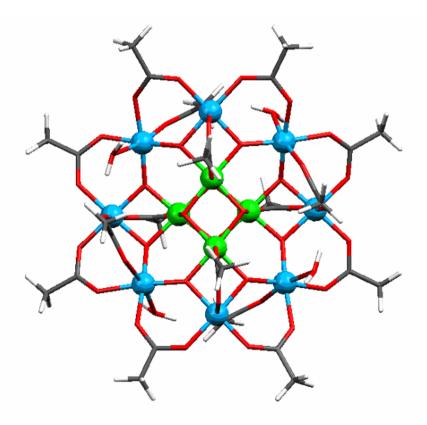


#### **SMM investigated**

• Mn<sub>12</sub>-BrAc

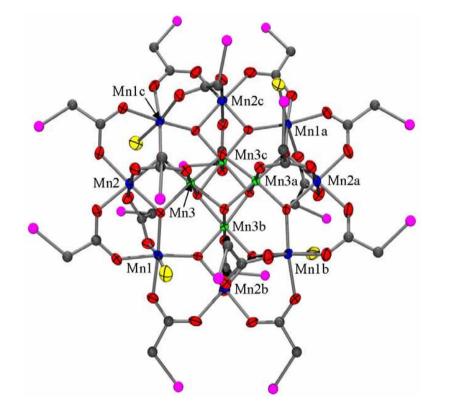






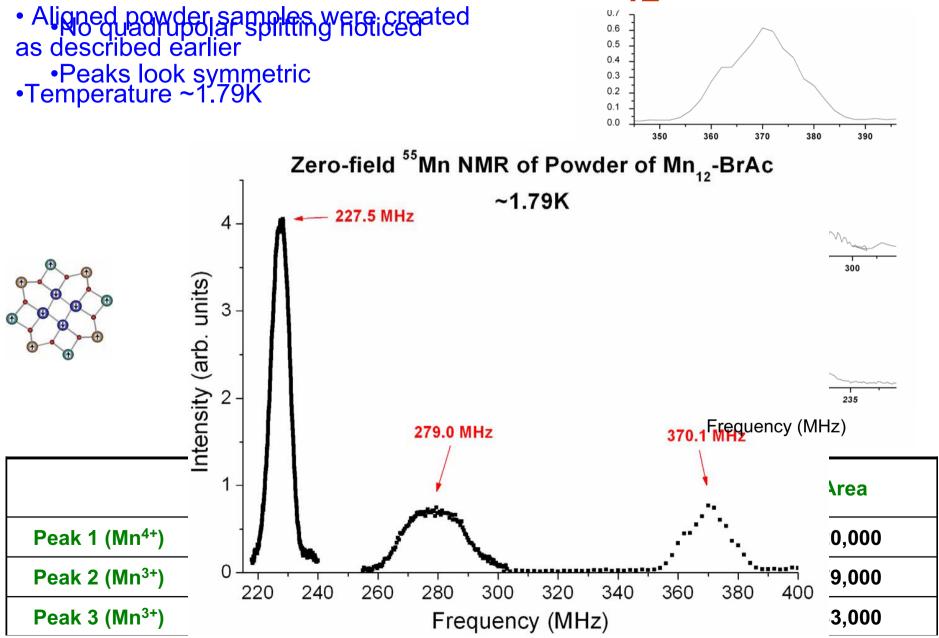
# $[Mn_{12}O_{12}(O_{2}CCH_{2}Br)_{16}(H_{2}O)_{4}] \cdot 4CH_{2}CI_{2}$

#### (Mn<sub>12</sub>-BrAc)

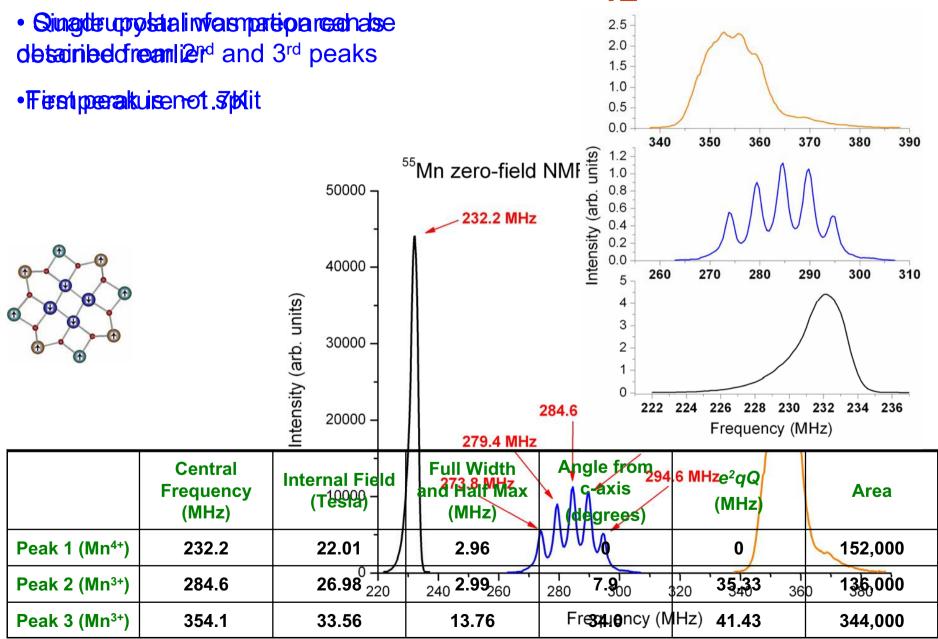


- I4<sub>1</sub>/a space group
- S = 10
- D = -0.456 cm<sup>-1</sup>
- $B_4^{0} = -2.0 \times 10^{-5} \text{ cm}^{-1}$
- Closest Mn-Mn distance is 6.07Å
- 4 CH<sub>2</sub>Cl<sub>2</sub> solvent molecules

#### **Aligned Powder Mn<sub>12</sub>-BrAc**



## Single Crystal Mn<sub>12</sub>-BrAc



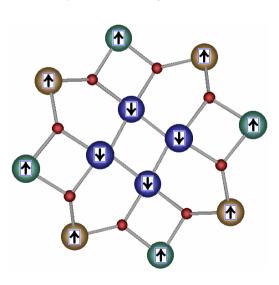
# Single Crystal vs. Powder Mn<sub>12</sub>-BrAc

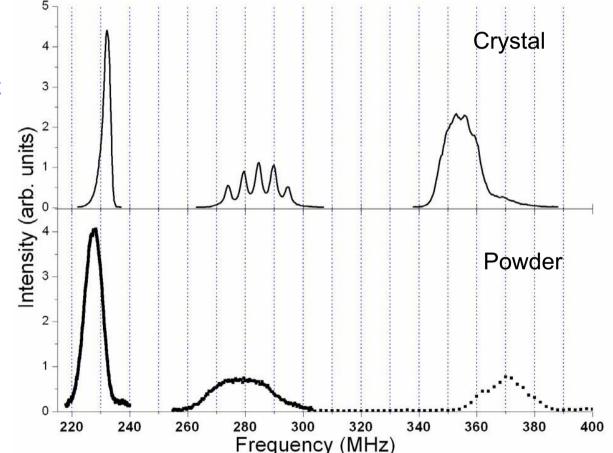
•Powder does not represent a statistical average of crystal

•Peaks shifted up to ~16 MHz

•Loss of Quadrupolar information

•Change most likely a result of powdering process (heat and pressure)

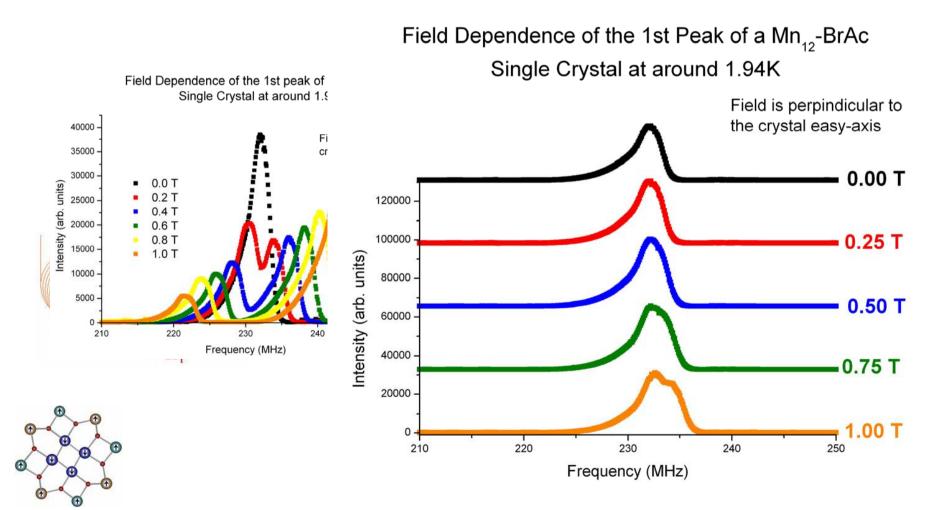




# **Field Studies of Mn<sub>12</sub>-BrAc**

•Zero-field cooled experiment

•Splitting as field is applied perpendicular to c-axis may be due to crystal misalignment

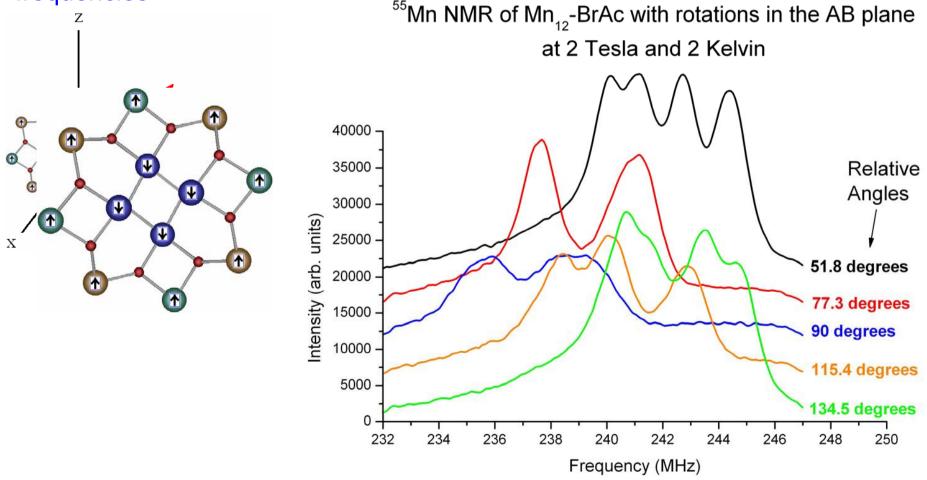


#### Angular Dependence Mn<sub>12</sub>-BrAc: ab plane

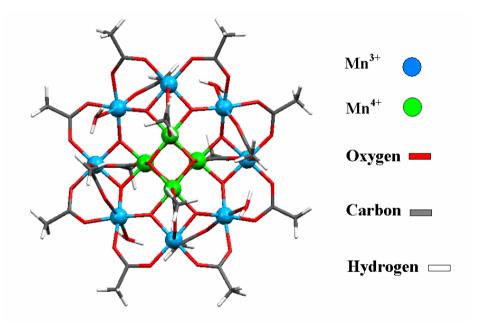
• Beyasta a veasa previously discussed

•Explerimentation/isvirasmolon/ifferced/lim/d in ra field of 5T

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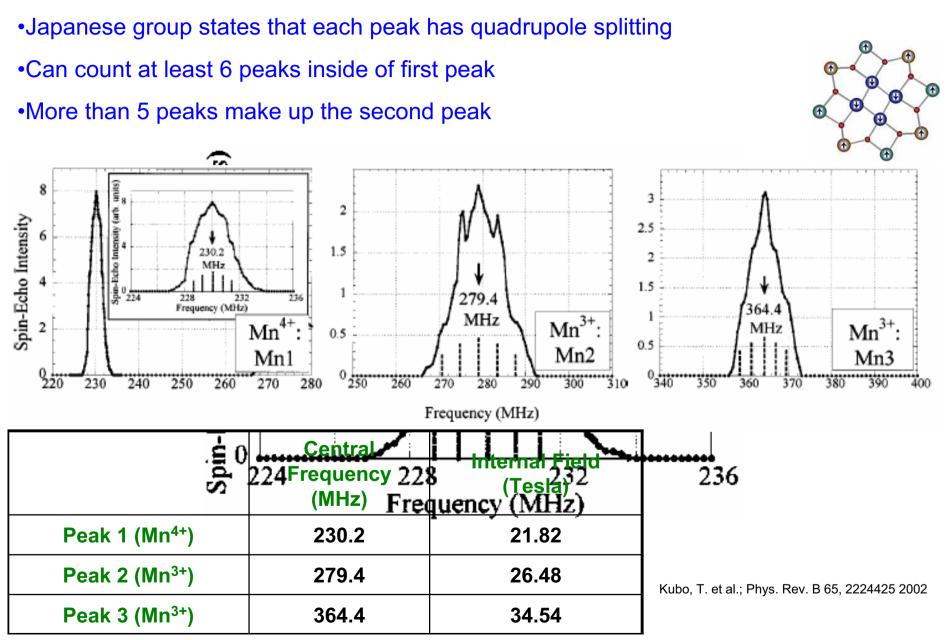
# $[Mn_{12}O_{12}(CH_{3}COO)_{16}(H_{2}O)_{4}] \cdot 2CH_{3}COOH \cdot 4H_{2}O$ $(Mn_{12}-Ac)$



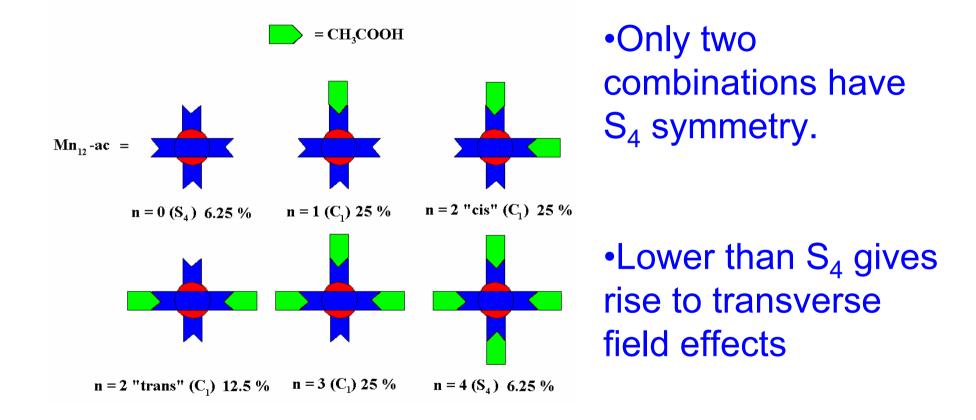
T. Lis, Acta Crysallogr. Sect. B 36, 2042 (1980).

- I4(bar)
- S = 10
- $D = -0.39 \text{ cm}^{-1}$
- $B_4^0 = -7.7 \times 10^{-4} \text{ cm}^{-1}$
- 2 HAc and 4 H<sub>2</sub>O solvent molecules

# **Aligned Powder Mn<sub>12</sub>-Ac**

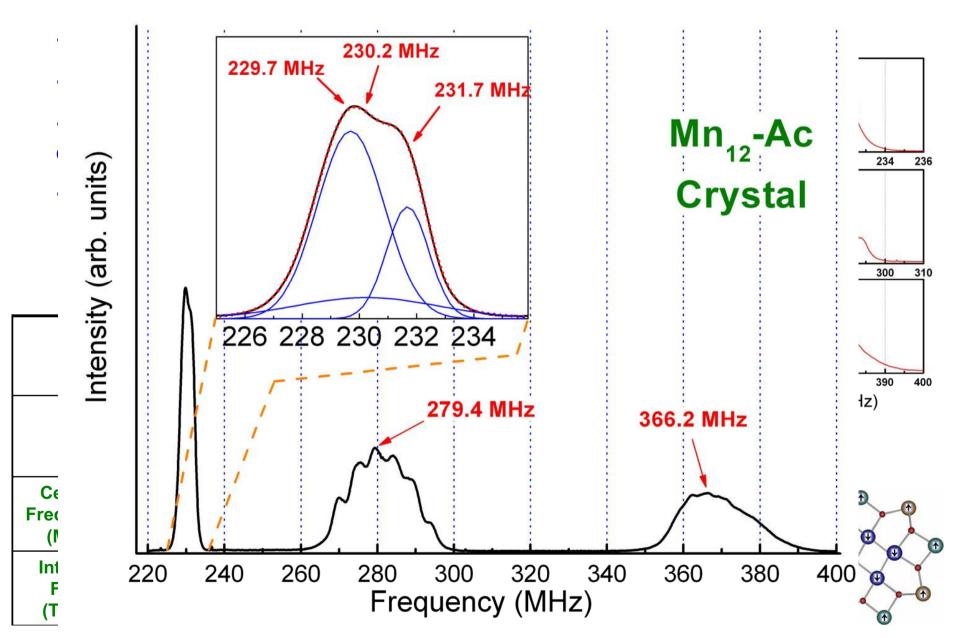


#### Cornia et al.

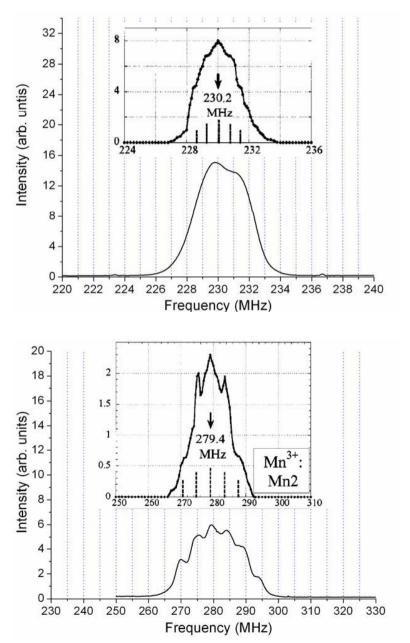


A. Cornia et al., Phys. Rev. Lett. 89, 257201-1-4 (2002).

#### Single Crystal Mn<sub>12</sub>-Ac



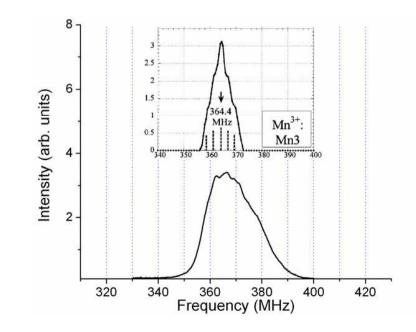
#### Single Crystal vs. Aligned Powder of Mn<sub>12</sub>-Ac



•No sign of quadrupolar splitting in first peak

•Narrower peaks from aligned powder most likely due to loss of isomers by crushing sample

•Powdering sample changes local environment at the nucleus



# Conclusion

- NMR resolution from single crystals is much better than that obtained from aligned powders.
- Powdering of crystals causes sample defects in the lattice due to loss of solvated molecules.
- Angular dependence studies show the hyperfine interaction on the Mn<sup>4+</sup> ions have an anisotropic component accounting for ~2% of the internal field. Not detectable by aligned powder NMR.
- NMR is a quicker, inexpensive technique to compliment neutron scattering for determining oxidation state and spin orientation of individual metal centers.
- NMR also compliments high field EPR data since EPR yields global (molecular) spin behavior, not that of individual ions in a complex.

## Acknowledgements

- Dr. Christou
- Andrew Harter and Dr. Achey
- Dr. Reyes and Dr. Kuhns
- Funding by NSF/NIRT-DMR Grant No. 0103290