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**International Centre for Theoretical Physics**



**SMR.1744 - 15**

# **SCHOOL ON ION BEAM ANALYSIS AND ACCELERATOR APPLICATIONS**

**13 - 24 March 2006**

**Accelerator mass spectrometry of heavy radioisotopes**

**Walter KUTSCHERA**  
**Institute of Isotope Research, University of Vienna, Austria**

# **Accelerator Mass Spectrometry of Heavy Radioisotopes**

Walter Kutschera

*Institute for Isotope Research and Nuclear Physics  
University of Vienna*

School on Ion Beam Analysis and Accelerator Applications  
Miramare – Trieste, 17 March 2006

## Radioisotopes measured with AMS

Isotope	Half-life (year)
$^3\text{H}$	12
$^{44}\text{Ti}$	60
$^{63}\text{Ni}$	100
$^{32}\text{Si}$	140
$^{39}\text{Ar}$	269
<b><math>^{14}\text{C}</math></b>	<b>5 730</b>
$^{59}\text{Ni}$	75 000
$^{41}\text{Ca}$	104 000
$^{81}\text{Kr}$	230 000
$^{36}\text{Cl}$	301 000
$^{26}\text{Al}$	720 000
$^{10}\text{Be}$	1 520 000
$^{53}\text{Mn}$	3 600 000
$^{182}\text{Hf}$	8 900 000
$^{129}\text{I}$	17 000 000
$^{236}\text{U}$	23 000 000
$^{244}\text{Pu}$	81 000 000

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<b><math>^{129}\text{I}</math></b>	<b>17 000 000</b>
<b><math>^{236}\text{U}</math></b>	<b>23 000 000</b>
<b><math>^{244}\text{Pu}</math> (<math>^{239}, ^{240}, ^{242}\text{Pu}</math>)</b>	<b>81 000 000</b>

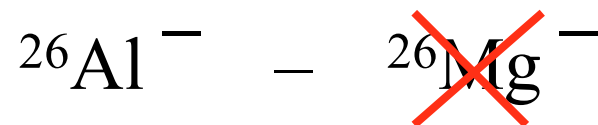
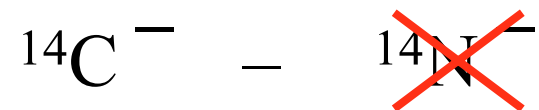
**Radioisotopes are measured with AMS through isotope ratios**

Typical range:

radioisotope/stable isotope =  $10^{-12}$  to  $10^{-16}$

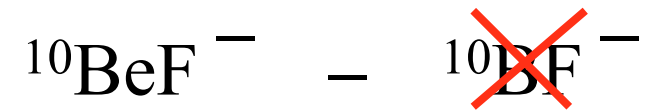
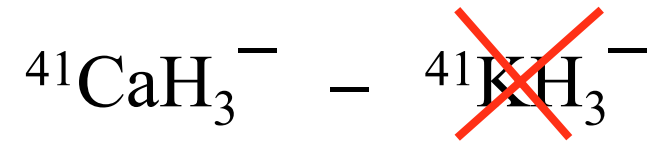
The main challenge in measuring such minute isotope ratios is the separation of the radioisotope from interfering background of stable isobars.

**There are three good cases if negative ions are used**



All other radioisotopes are more difficult and must be cleaned up from stable isobar interference.

## Sometimes special negative molecules help



# Measurement of the $^{129}\text{I}/^{131}\text{I}$ Ratio in Chernobyl Fallout

Paper presented at the Adriatico Conference on „Environmental Physics - Atmospheric Aerosol“, Trieste 22-25 July 1986

Both  $^{129}\text{I}$  and  $^{131}\text{I}$  are fission products, but with vastly different half-lives:

**$^{129}\text{I}$ : 1.6 million years**

**$^{131}\text{I}$  : 8.0 days**

W. Kutschera, D. Fink, M. Paul, G. Hollos, A. Kaufmann  
Physics Scripta 37 (1988) 310-313



## Abstract

Rainwater collected in the Munich area approximately one week after the Chernobyl reactor accident was investigated for its content of the radioisotopes  $^{129}\text{I}$  ( $T_{1/2} = 1.6 \times 10^7 \text{ yr}$ ) and  $^{131}\text{I}$  ( $T_{1/2} = 8.04 \text{ d}$ ). For the time of release, an isotopic ratio of  $^{129}\text{I}/^{131}\text{I} = 19 \pm 5$  was found. This value was obtained from a gamma-ray activity measurement of  $^{131}\text{I}$  with a Ge detector and a concentration measurement of  $^{129}\text{I}$  with accelerator mass spectrometry. From the measured ratio an operating time of the reactor prior to the accident in the vicinity of two years can be estimated, which is in fair agreement with estimates from other long-lived to short-lived radioisotope ratios in the Chernobyl fallout. Some measurements of  $^{131}\text{I}$  activity in thyroids of persons living in the Munich area are also reported.

Table I. *Activity of  $^{131}\text{I}$  in human thyroids (Munich area).*

Date of measurement	Person #	Activity <sup>a</sup> (Bq)	Main living site	Dose <sup>b</sup> (mrem)
6 May 1986	1	227	country	45
	2	98	city	20
	3	91	city	18
	4	108	city	22
11 May 1986	2	60	city	12
	5	120	country	24
	6	117	country	23
27 May 1986	2	15	city	3
	5	36	country	7

<sup>a</sup> Bq = 1 decay/sec. Uncertainty =  $\pm 20\%$ .

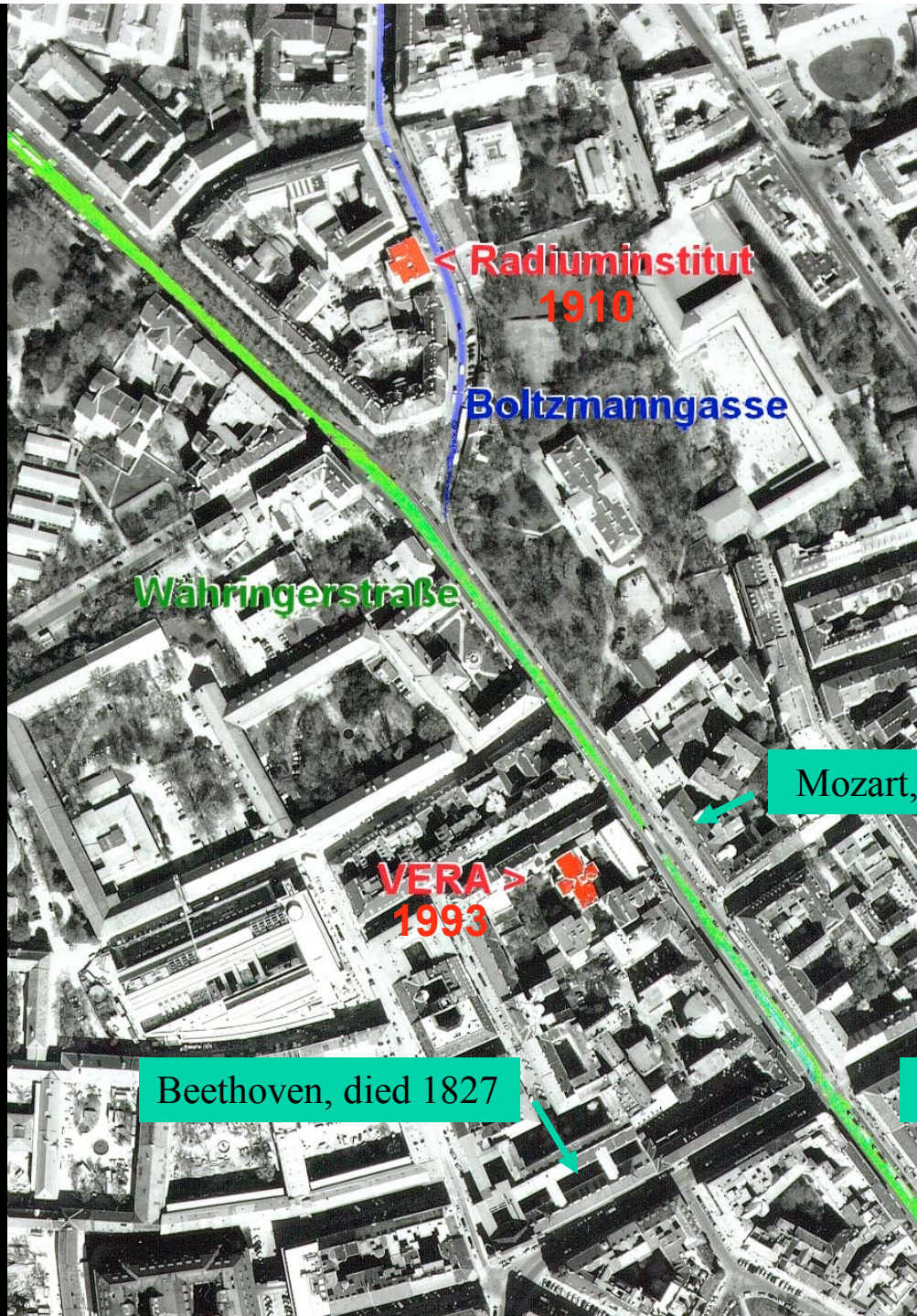
<sup>b</sup> Conversion = 0.2 mrem/Bq (=  $2 \mu\text{Sv/Bq}$ ).

# Vienna Environmental Research Accelerator V E R A

Institute für Isotope Research and Nuclear Physics  
University of Vienna  
Währinger Str. 17, A-1090 Vienna, Austria

VERA is an AMS facility for “all” isotopes based on a  
3-MV Pelletron tandem accelerator

P.Steier et al., Nucl. Instrum. & Methods 223-224 (2004) 67-71  
W. Kutschera, Int. J. Mass Spectrometry 142 (2005) 145-160



< Radiuminstitut  
1910

Boltzmannngasse

Währingerstraße

VERA >  
1993

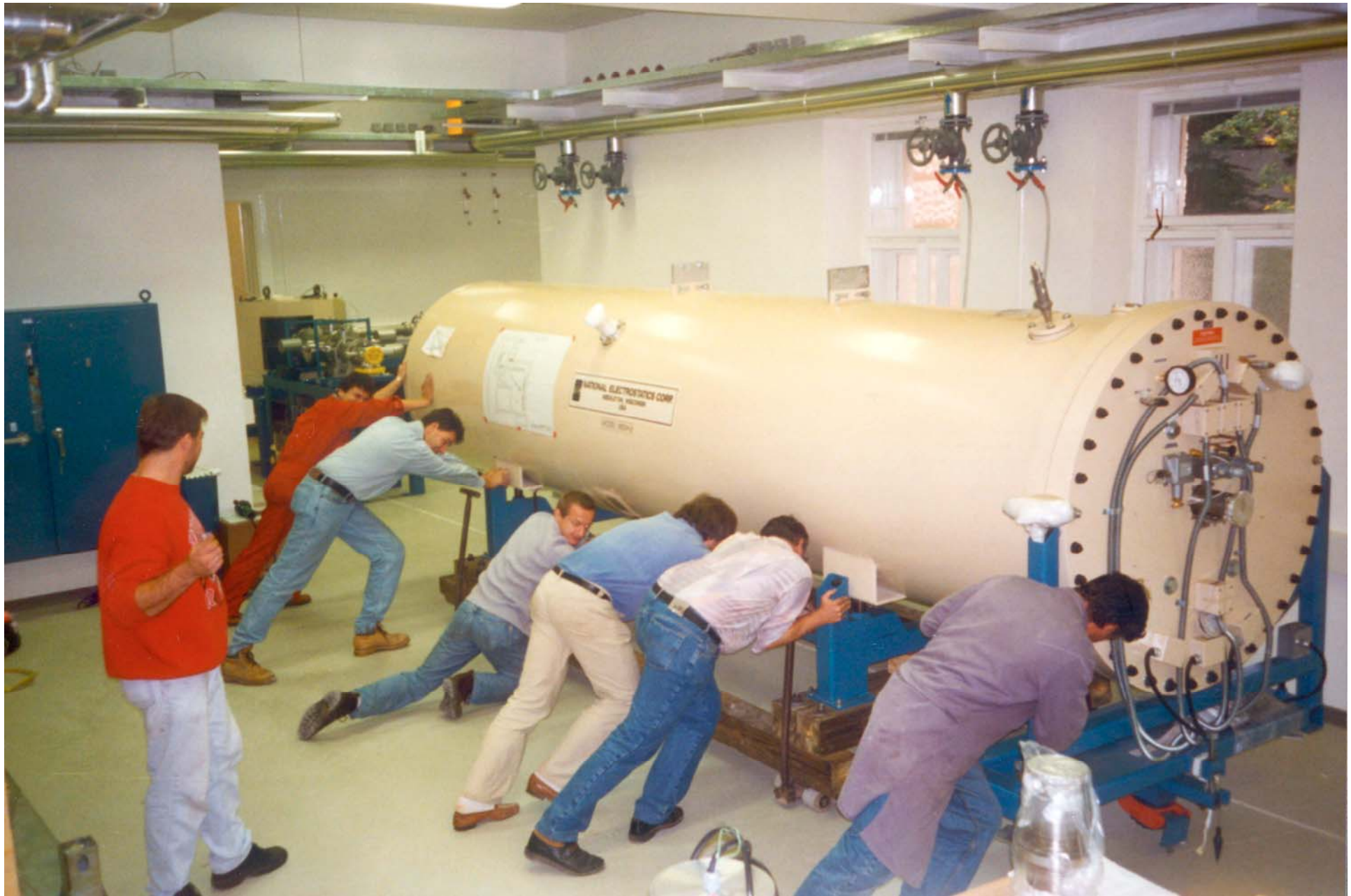
Beethoven, died 1827

Mozart, Così fan tutte

Sigmund Freud appartm.



**Positioning of the 3-MV tandem accelerator of VERA in the  
“Kavalierstrakt“, Währingerstr. 17, A-1090 Wien (1995)**



# +3 MV Tandem Accelerator

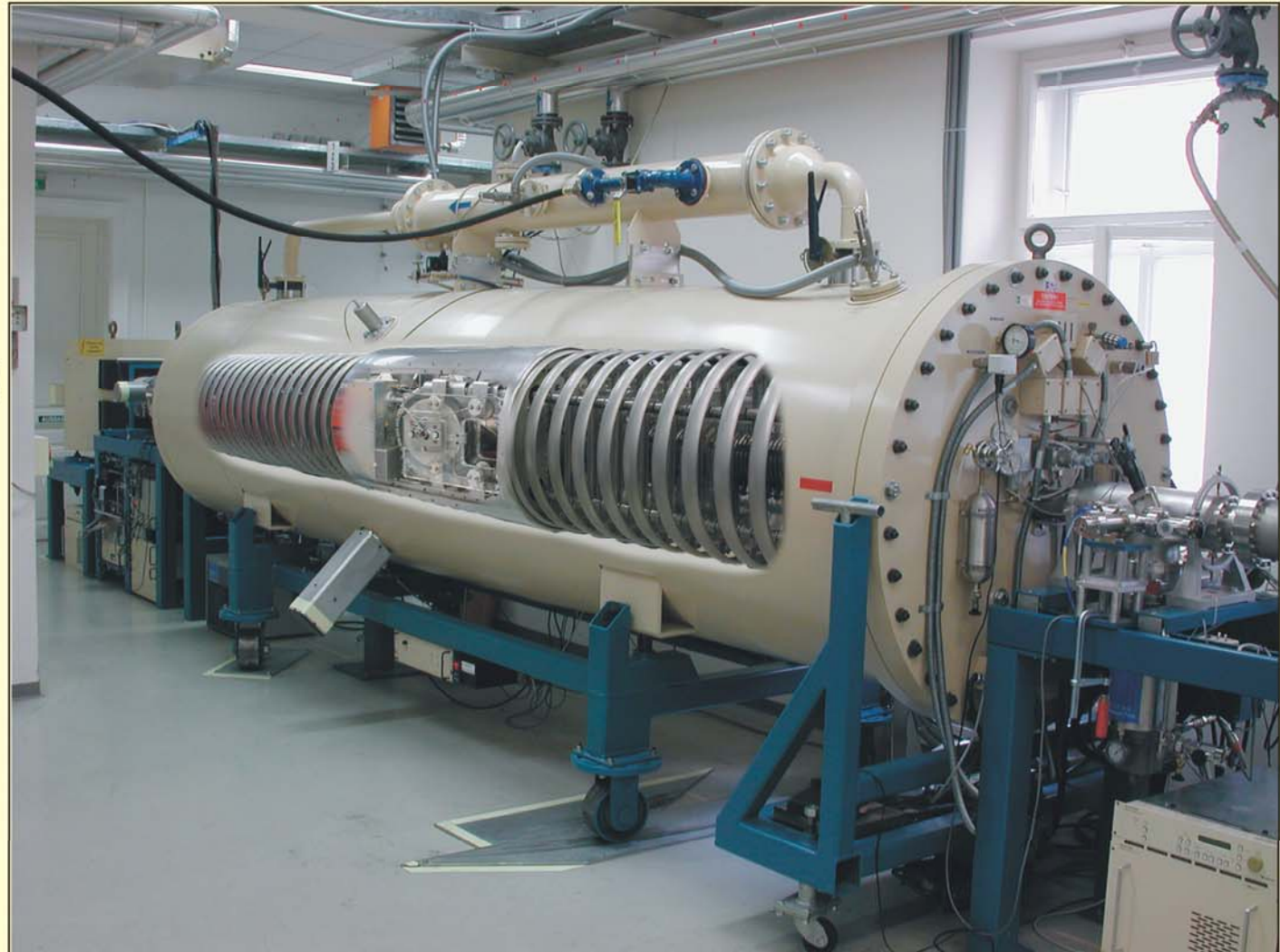
Pelletron type

2 charging  
chains

maximal charging  
current: 230  $\mu\text{A}$

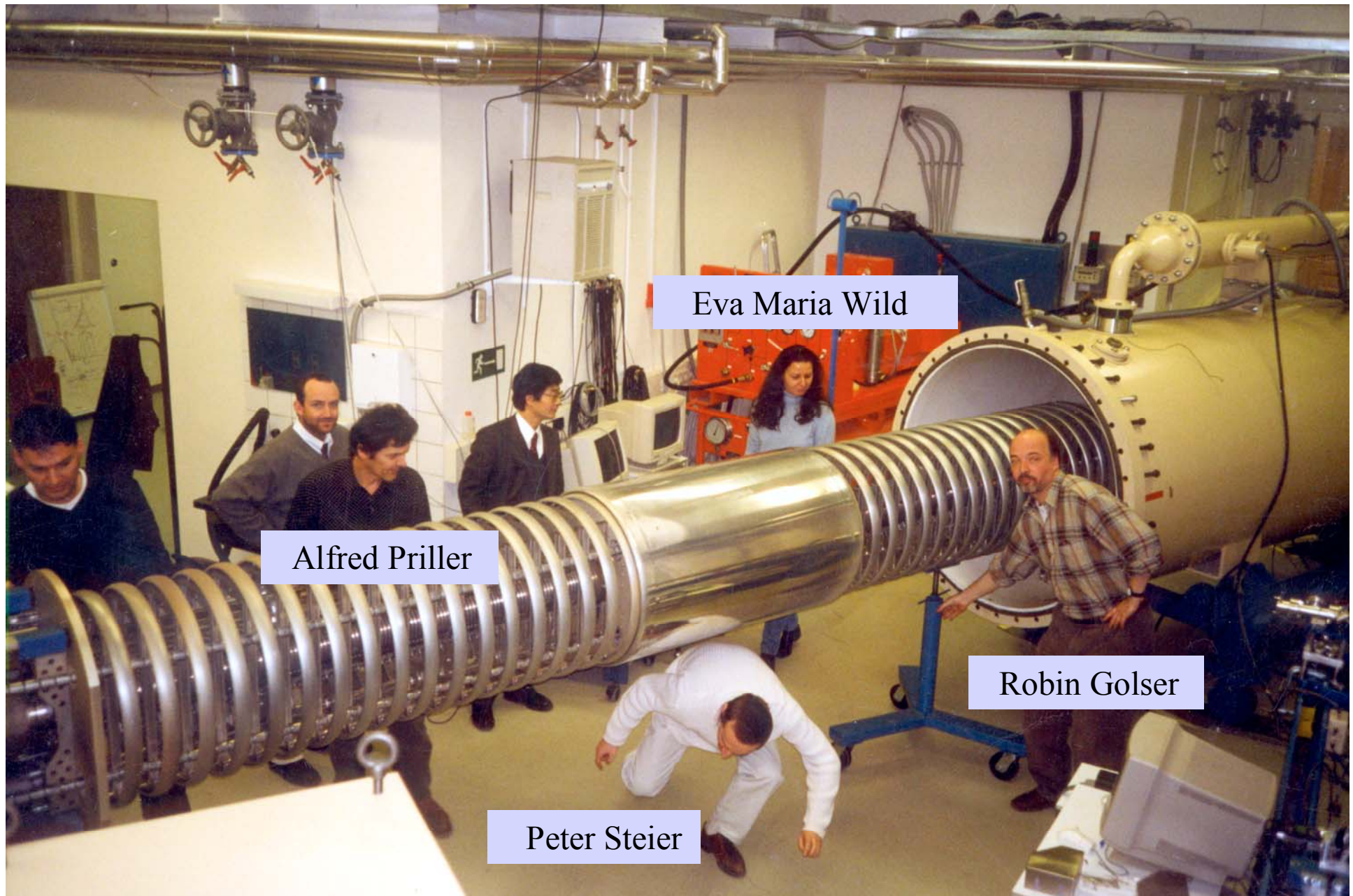
Ar gas stripper/  
foil stripper

insulating gas:  
 $\text{SF}_6$





## Column structure of the 3-MV tandem accelerator of VERA



Eva Maria Wild

Alfred Priller

Robin Golser

Peter Steier

## Staff of VERA



**Alfred Priller**  
*Technical head of VERA*  
(paleoclimate, loess)



**Peter Steier**  
*Operations manager*  
(glacier dating, DNA dating, heavy isotopes)



**Eva Maria Wild**  
*Sample preparation*  
(archaeology, paleoclimate)



**Robin Golser**  
*Atomic physics*  
(exotic atoms, PIXE)



**Anton Wallner**  
*Astrophysics*  
(supernova remnant, stellar nucleosynthesis)



# Cs-Beam Sputter Source for Negative Ions

40 Samples

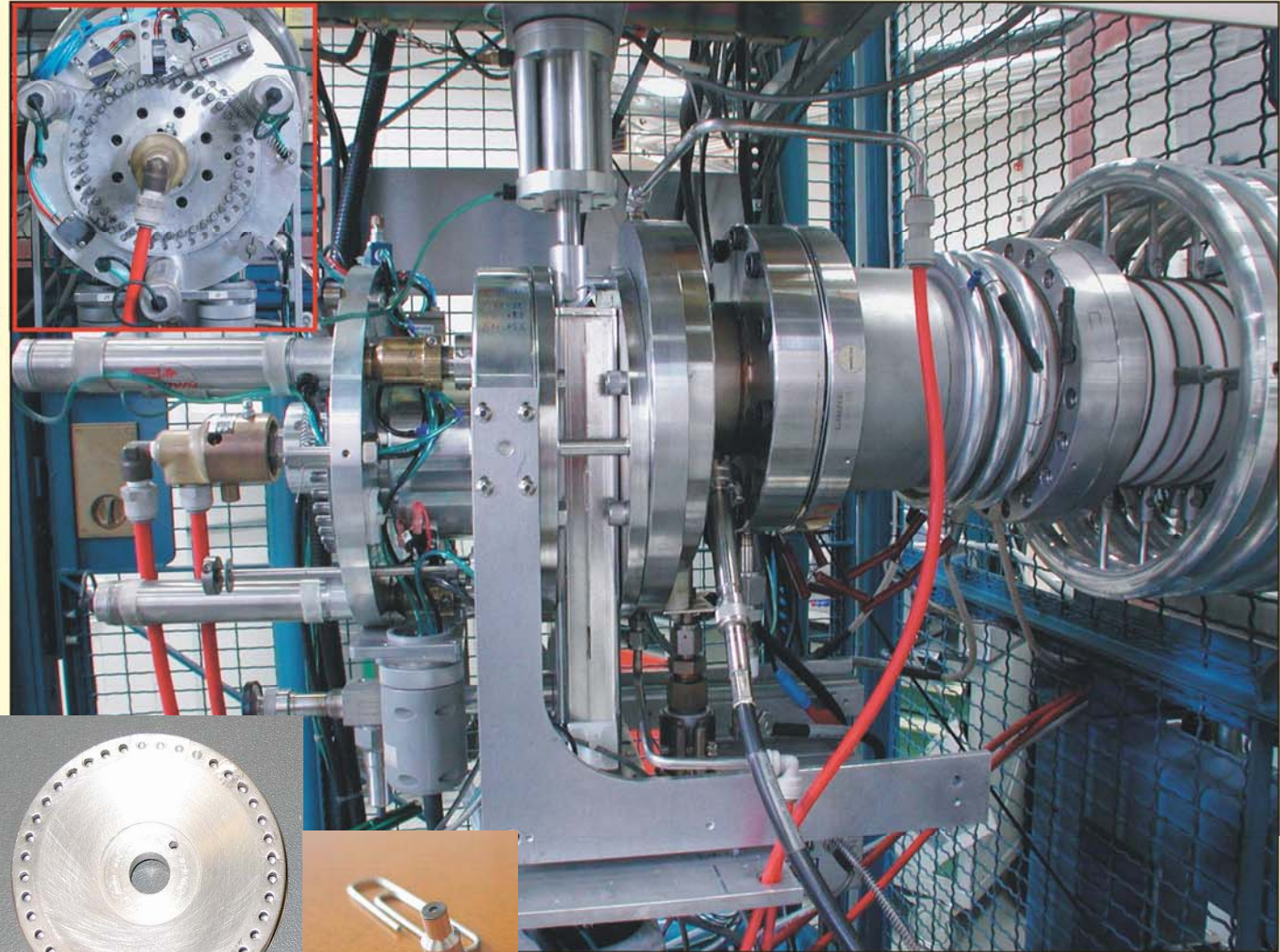
max. 75 keV  
Preacceleration

Ion Currents:

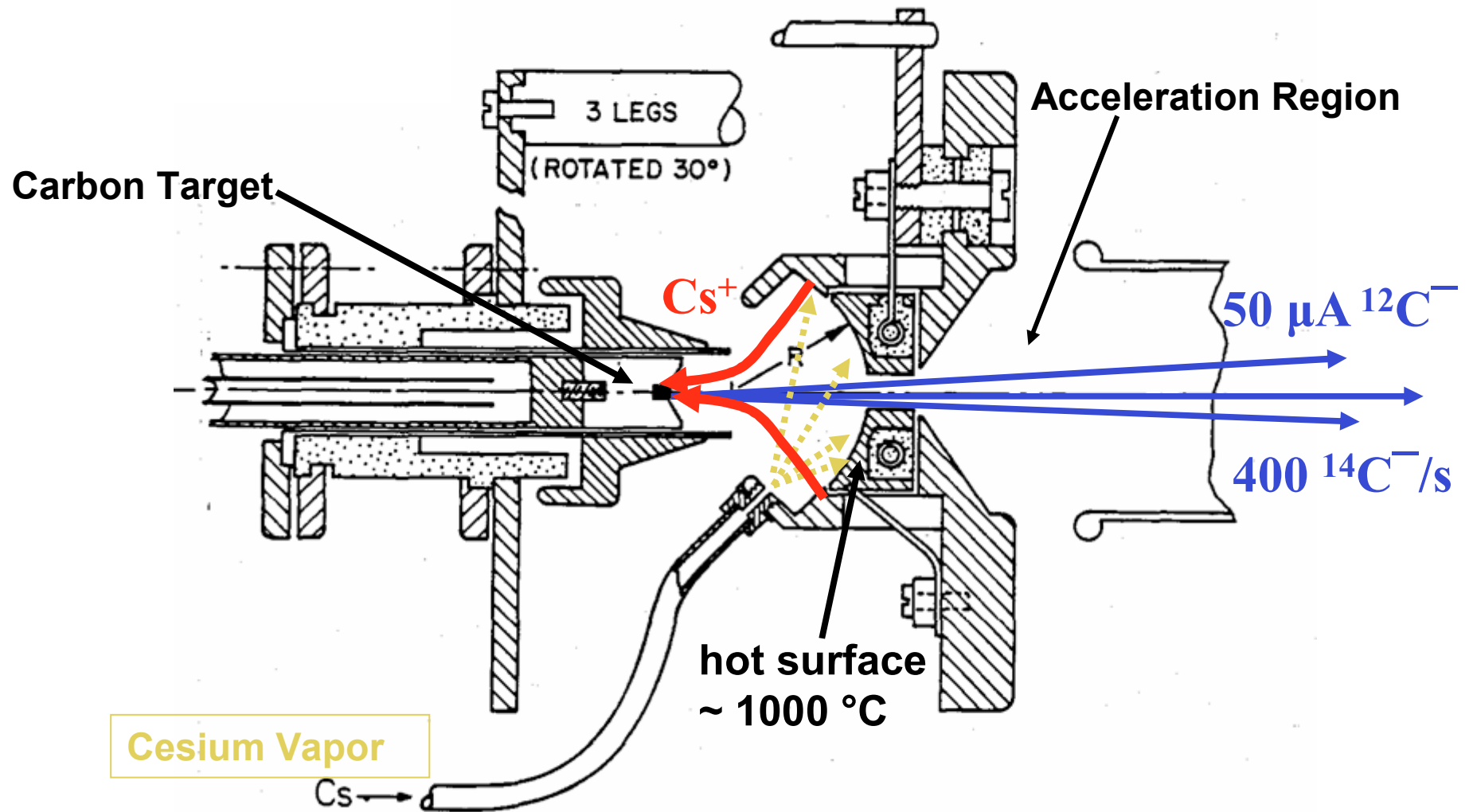
$C^-$ : 60  $\mu A$

$BeO^-$ : 3  $\mu A$

$UO^-$ : 100 nA

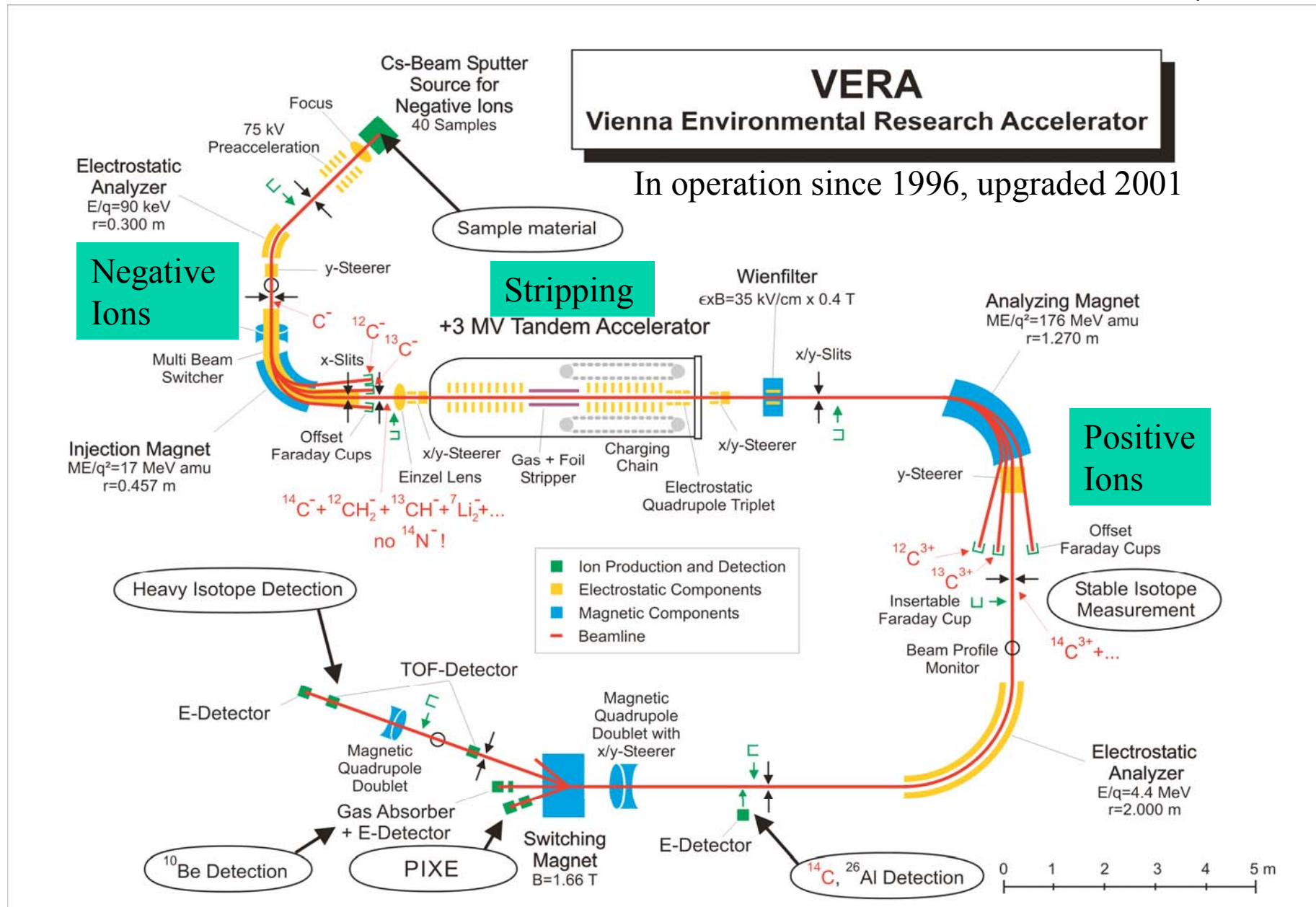


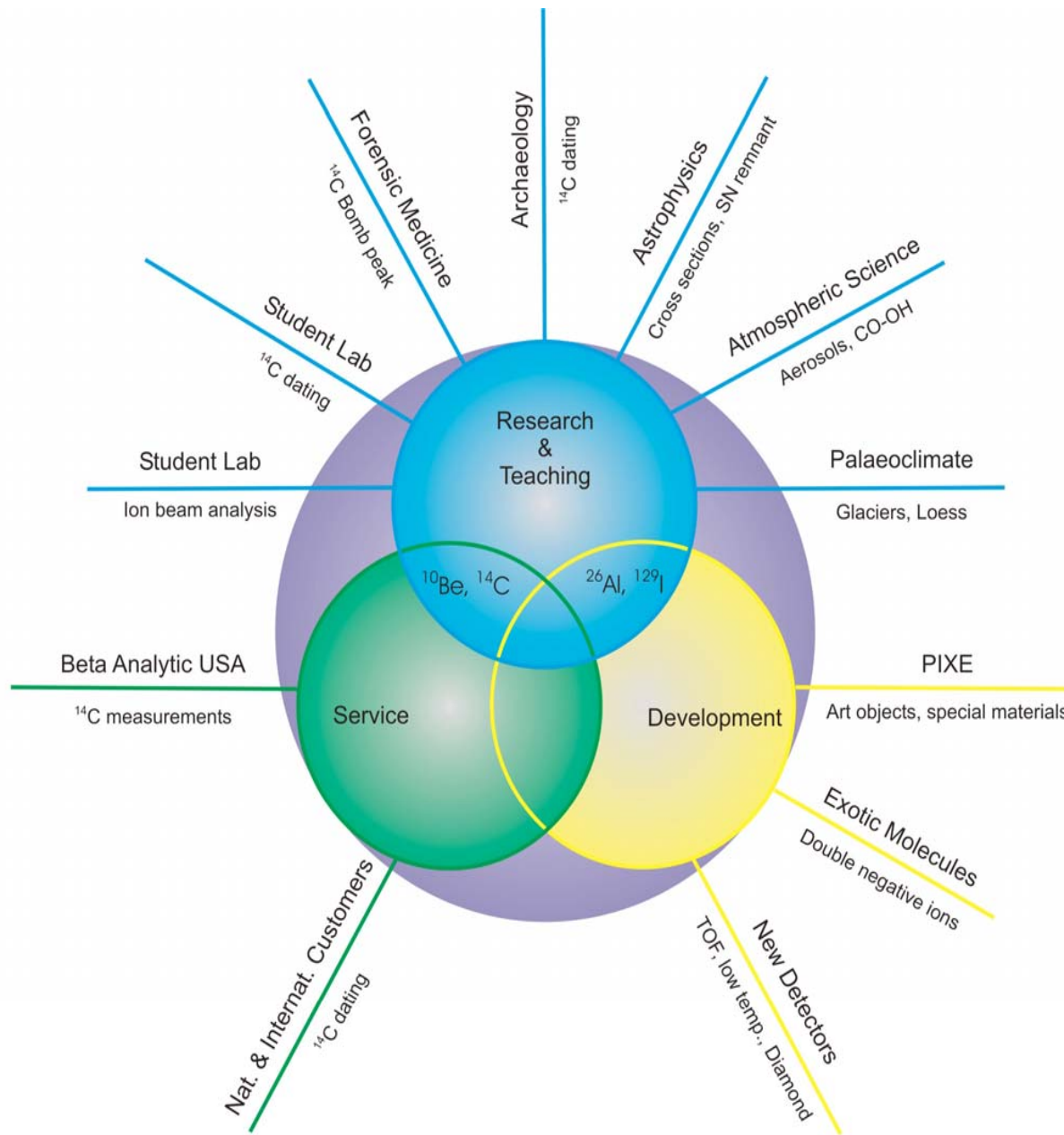
# The Cesium-Beam Sputter Source for Negative Ions





$^1\text{H}$ ,  $^{10}\text{Be}$ ,  $^{14}\text{C}$ ,  $^{26}\text{Al}$ ,  $^{36}\text{Cl}$ ,  $^{41}\text{Ca}$ ,  $^{55}\text{Fe}$ ,  $^{129}\text{I}$ ,  $^{182}\text{Hf}$ ,  $^{210}\text{Pb}$ ,  $^{236}\text{U}$ ,  $^{239-244}\text{Pu}$ ,  $(^{43}\text{Ca}^{19}\text{F}_4)^{-}$ ,  $(\text{H}_2)^{-}$

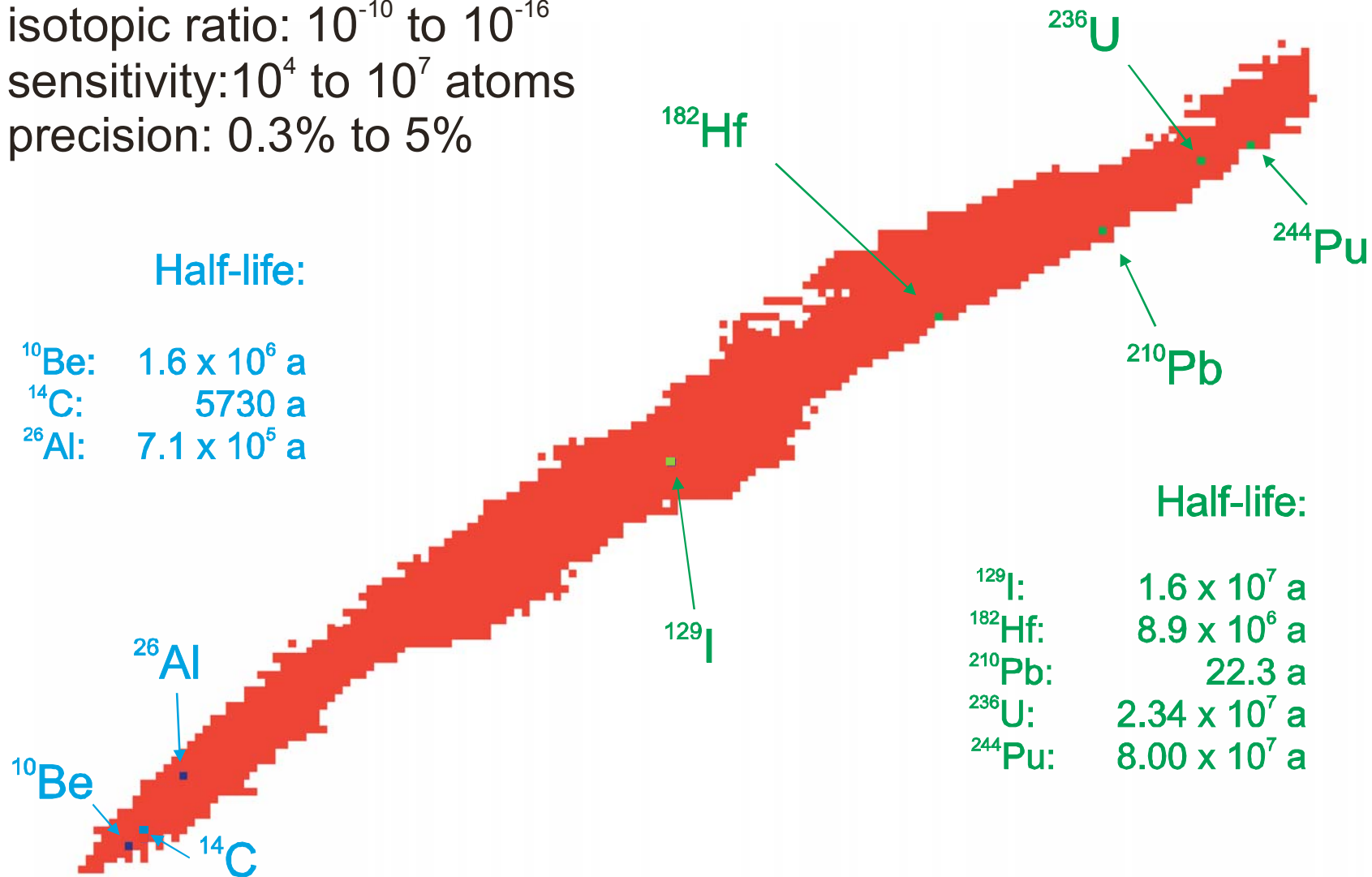




## The World of VERA

# Long-lived radionuclides measured at VERA

isotopic ratio:  $10^{-10}$  to  $10^{-16}$   
sensitivity:  $10^4$  to  $10^7$  atoms  
precision: 0.3% to 5%



# Some AMS-nuclides at VERA (3-MV)

Radio-nuclide	Half-life (Myr)	Overall Efficiency	Detection Limit	Precision
$^{10}\text{Be}$	1.5	$5 \times 10^{-5}$	$^{10}\text{Be}/^9\text{Be} < 2 \times 10^{-14}$	< 3%
$^{14}\text{C}$	5730 yr	$2 \times 10^{-2}$	$^{14}\text{C}/^{12}\text{C} < 3 \times 10^{-16}$	< 0.5 %
$^{26}\text{Al}$	0.7	$5 \times 10^{-4}$	$^{26}\text{Al}/^{27}\text{Al} < 6 \times 10^{-16}$	< 1.0 %
$^{129}\text{I}$	15.7	$1 \times 10^{-2}$	$^{129}\text{I}/^{127}\text{I} = 2 \times 10^{-14}$	2 %
$^{182}\text{Hf}$	8.9	$1 \times 10^{-4}$	$^{182}\text{Hf}/^{180}\text{Hf} = 1 \times 10^{-11}$	5 %
$^{236}\text{U}$	23.4	--	$^{236}\text{U}/^{238}\text{U} = 6 \times 10^{-12}$	5 %
$^{244}\text{Pu}$	81.0	$> 4 \times 10^{-5}$	--	5 %

# **Determination of Plutonium in environmental samples by AMS and Alpha Spectrometry**

Proceedings of the 8th International Conference in Application of Nuclear Techniques, Crete, Greece, 12-18 September 2004

E. Hrenecek, P. Steier, A. Wallner  
Applied Radiation and Isotopes 63 (2005) 633-638



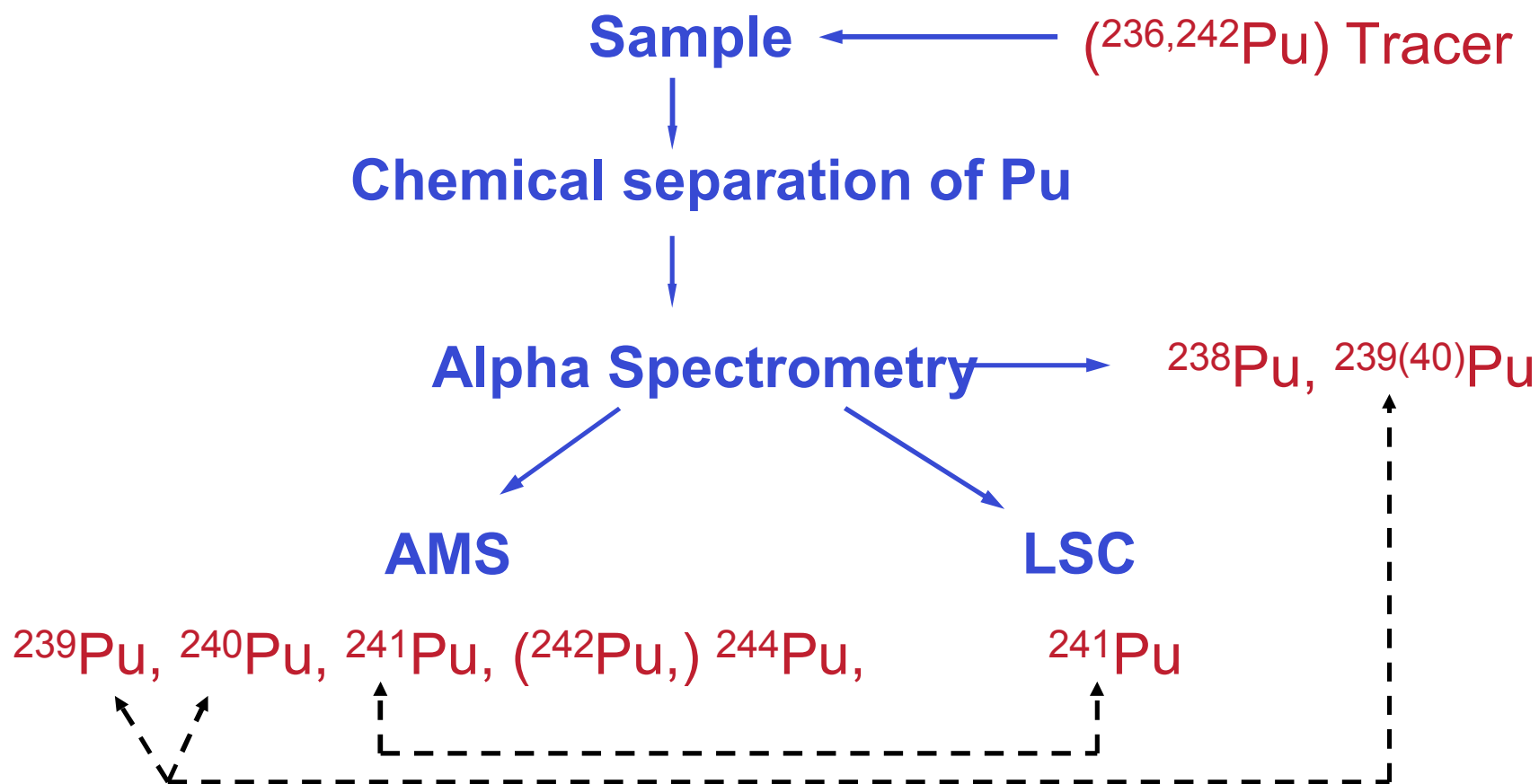
Cm	<b>Cm238</b> 2.4 h 0+	<b>Cm239</b> 2.9 h (7/2-)	<b>Cm240</b> 27 d 0+	<b>Cm241</b> 32.8 d 1/2+	<b>Cm242</b> 162.8 d 0+	<b>Cm243</b> 29.1 y 5/2+	<b>Cm244</b> 18.10 y 0+ *	<b>Cm245</b> 8500 y 7/2+	<b>Cm246</b> 4730 y 0+	<b>Cm247</b> 1.56E+7 y 9/2-
	EC, $\alpha$	EC, $\alpha$	EC, $\alpha$ ,sf,...	EC, $\alpha$	$\alpha$ ,sf	EC, $\alpha$ ,sf,...	$\alpha$ ,sf	$\alpha$ ,sf	$\alpha$ ,sf	$\alpha$
Am	<b>Am237</b> 73.0 m 5/2(-)	<b>Am238</b> 98 m 1+	<b>Am239</b> 11.9 h (5/2)-	<b>Am240</b> 50.8 h (3-)	<b>Am241</b> 432.2 y 5/2-	<b>Am242</b> 16.02 h 1- *	<b>Am243</b> 7370 y 5/2-	<b>Am244</b> 10.1 h (6-) *	<b>Am245</b> 2.05 h (5/2)+	<b>Am246</b> 39 m (7-) *
	EC, $\alpha$	EC, $\alpha$	EC, $\alpha$	EC, $\beta^-$ , $\alpha$ ,...	$\alpha$ ,sf	EC, $\beta^-$	$\alpha$ ,sf	$\beta^-$	$\beta^-$	$\beta^-$
Pu	<b>Pu236</b> 2.858 y 0+	<b>Pu237</b> 45.2 d 7/2- *	<b>Pu238</b> 87.7 y 0+	<b>Pu239</b> 24110 y 1/2+	<b>Pu240</b> 6563 y 0+	<b>Pu241</b> 14.35 y 5/2+	<b>Pu242</b> 3.733E+5 y 0+ *	<b>Pu243</b> 4.956 h 7/2+	<b>Pu244</b> 8.08E+7 y 0+	<b>Pu245</b> 10.5 h (9/2-)
	$\alpha$ ,sf	EC, $\alpha$	$\alpha$ ,sf	$\alpha$ ,sf	$\alpha$ ,sf	$\beta^-$ , $\alpha$	$\alpha$ ,sf	$\beta^-$	$\alpha$ ,sf	$\beta^-$
Np	<b>Np235</b> 396.1 d 5/2+	<b>Np236</b> 154E+3 y (6-) *	<b>Np237</b> 2.14E+6 y 5/2+	<b>Np238</b> 2.117 d 2+	<b>Np239</b> 2.3565 d 5/2+	<b>Np240</b> 61.9 m (5+) *	<b>Np241</b> 13.9 m (5/2+)	<b>Np242</b> 5.5 m (6) *	<b>Np243</b> 1.8 m (5/2-)	
	EC, $\alpha$	EC, $\beta^-$ , $\alpha$ ,...	$\alpha$ ,sf	$\beta^-$	$\beta^-$	$\beta^-$	$\beta^-$ , $\alpha$	$\beta^-$	$\beta^-$	
U	<b>U234</b> 2.455E+5 y 0+	<b>U235</b> 703.8E+6 y 7/2- *	<b>U236</b> 2.342E7 y 0+	<b>U237</b> 6.75 d 1/2+	<b>U238</b> 4.468E+9 y 0+ *	<b>U239</b> 23.45 m 5/2+	<b>U240</b> 14.1 h 0+		<b>U242</b> 16.8 m 0+	
	$\alpha$ ,n,sf,...	$\alpha$ , <sup>20</sup> Ne,sf,...	$\alpha$ ,sf	$\beta^-$	$\alpha$ ,sf	$\beta^-$	$\beta^-$		$\beta^-$	
Pa	<b>Pa233</b> 26.967 d 3/2-	<b>Pa234</b> 6.70 h 4+ *	<b>Pa235</b> 24.5 m (3/2-)	<b>Pa236</b> 9.1 m 1(-)	<b>Pa237</b> 8.7 m (1/2+)	<b>Pa238</b> 2.3 m (3-)				
	$\beta^-$	$\beta^-$	$\beta^-$	$\beta^-$	$\beta^-$	$\beta^-$				

Pu isotopes: <sup>242</sup>Pu as spike material (reference)

AMS measurement: <sup>239,240,241(244)</sup>Pu relative to <sup>242</sup>Pu



# Analytical Strategy



# Isotope ratios

source	$^{238}\text{Pu}/^{239(40)}\text{Pu}$	reference
Global weapons test fallout	0.03	Bunzl et al. 1987
Chernobyl fallout	0.33 – 0.44	IAEA 1986
Irish Sea sediment	0.05 – 0.3	Kershaw et al. 1995
Thule sediment	0.019	Aarkog 1971
Mururoa test site	0.0044	Danesi et al. 2002
Fangataufa test site	0.38	Mulsow et al. 1999

# Isotope ratios

source	$^{240}\text{Pu}/^{239}\text{Pu}$	reference
Global fallout	0.18	Buessler et al. 1987
Chernobyl fallout	0.39	MacKenzie 2000
Irish Sea sediment	0.05 – 0.25	Kershaw et al. 1995
Thule sediment	0.058	Komura et al. 1984
Mururoa average	~ 0.035	Chiappini et al. 1999
Fangataufa	0.05	Chiappini et al. 1999

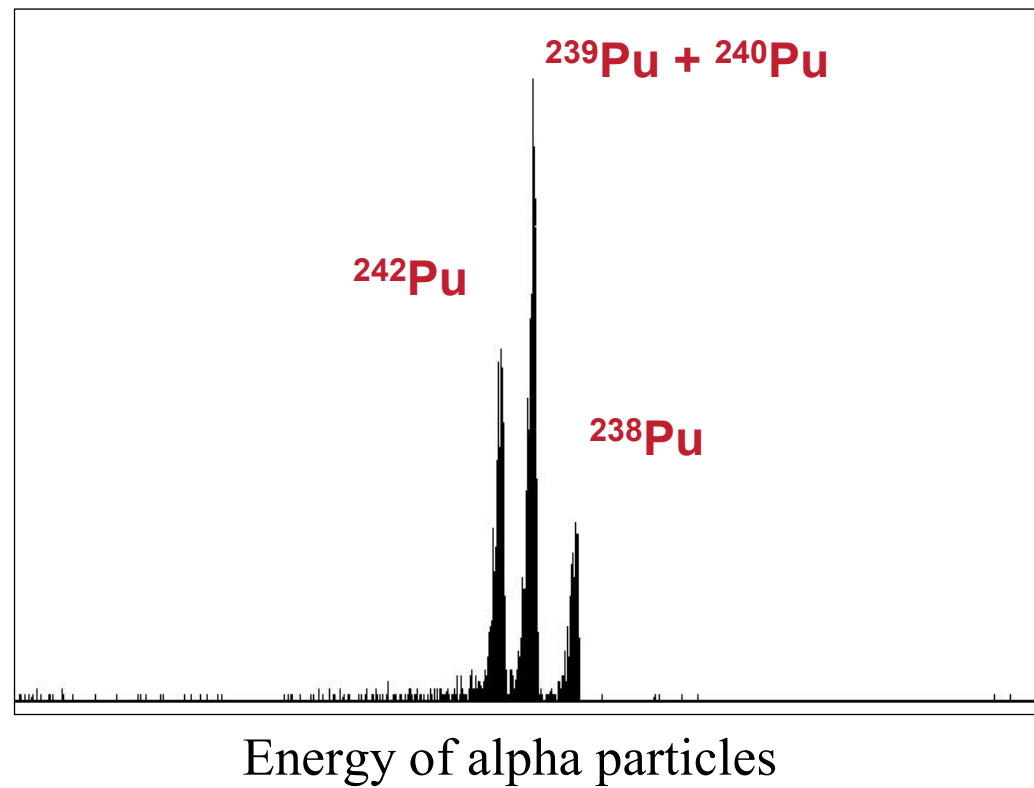
# Sample preparation for alpha counting

- Leaching 8 M HNO<sub>3</sub>
- Anion Exchange AG 1x8
- 8 M HNO<sub>3</sub>, NaNO<sub>2</sub> for Pu(IV)
- 10 M HCl for Th(IV)
- 0,1 M NH<sub>4</sub>I / 9 M HCl for Pu
- Alpha Spectrometry:  
Mikroprecipitation with NdF<sub>3</sub>,
- Cellulosenitrate filter  
0.1 μm pore size



# Alpha Spectrometry

- $^{242}\text{Pu}$  Tracer
- 2.5 to 6 g Sample size
- Detection limit:  
0.1 to 0.3 Bq / sample
- $^{238}\text{Pu}$  /  $^{239(240)}\text{Pu}$



# Sample Preparation for AMS

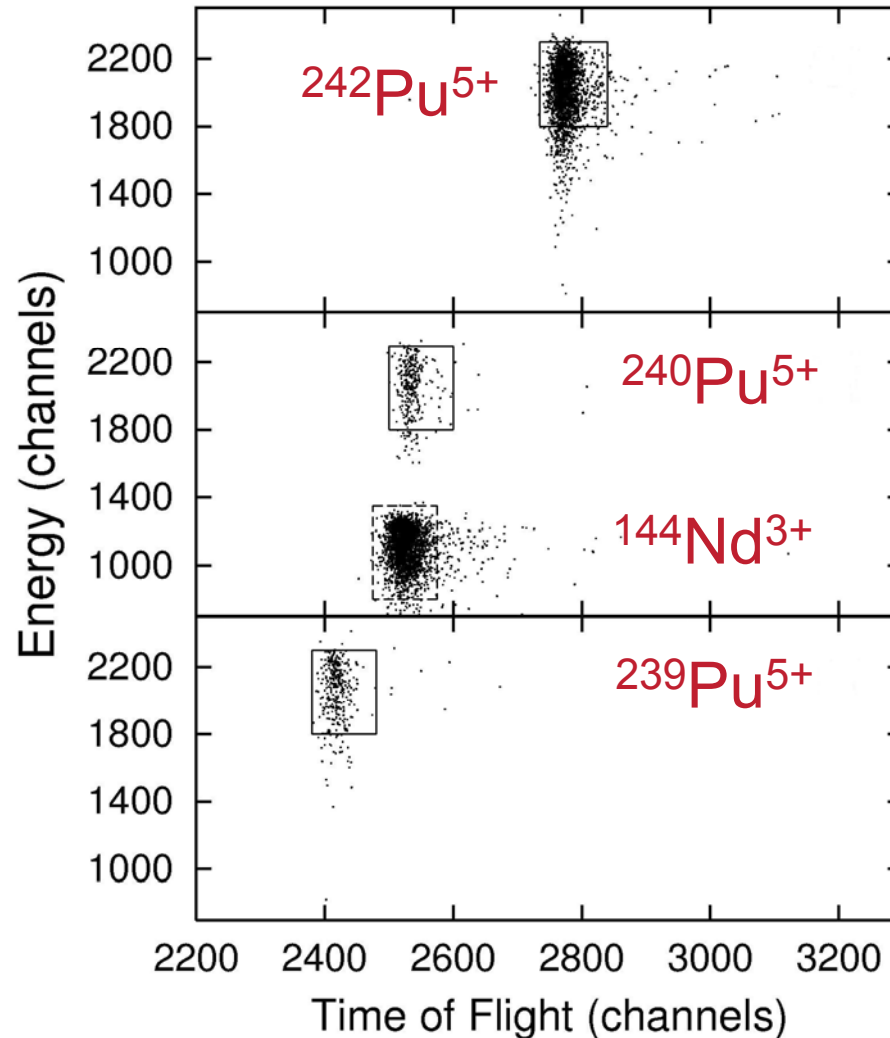
- Sample preparation:
  - combustion of filter
  - dissolution in HCl
  - 10 mg Fe carrier,  $\text{Fe}(\text{OH})_3$
  - $800^\circ\text{C}$   $\text{Fe}_2\text{O}_3$



## AMS measurement

$^{238}\text{UO}^- \rightarrow ^{238}\text{U}^{5+}$ , used for tuning (current)

$^{242}\text{PuO}^- \rightarrow ^{242}\text{Pu}^{5+}$ , scaling and ion counting



M/Q !

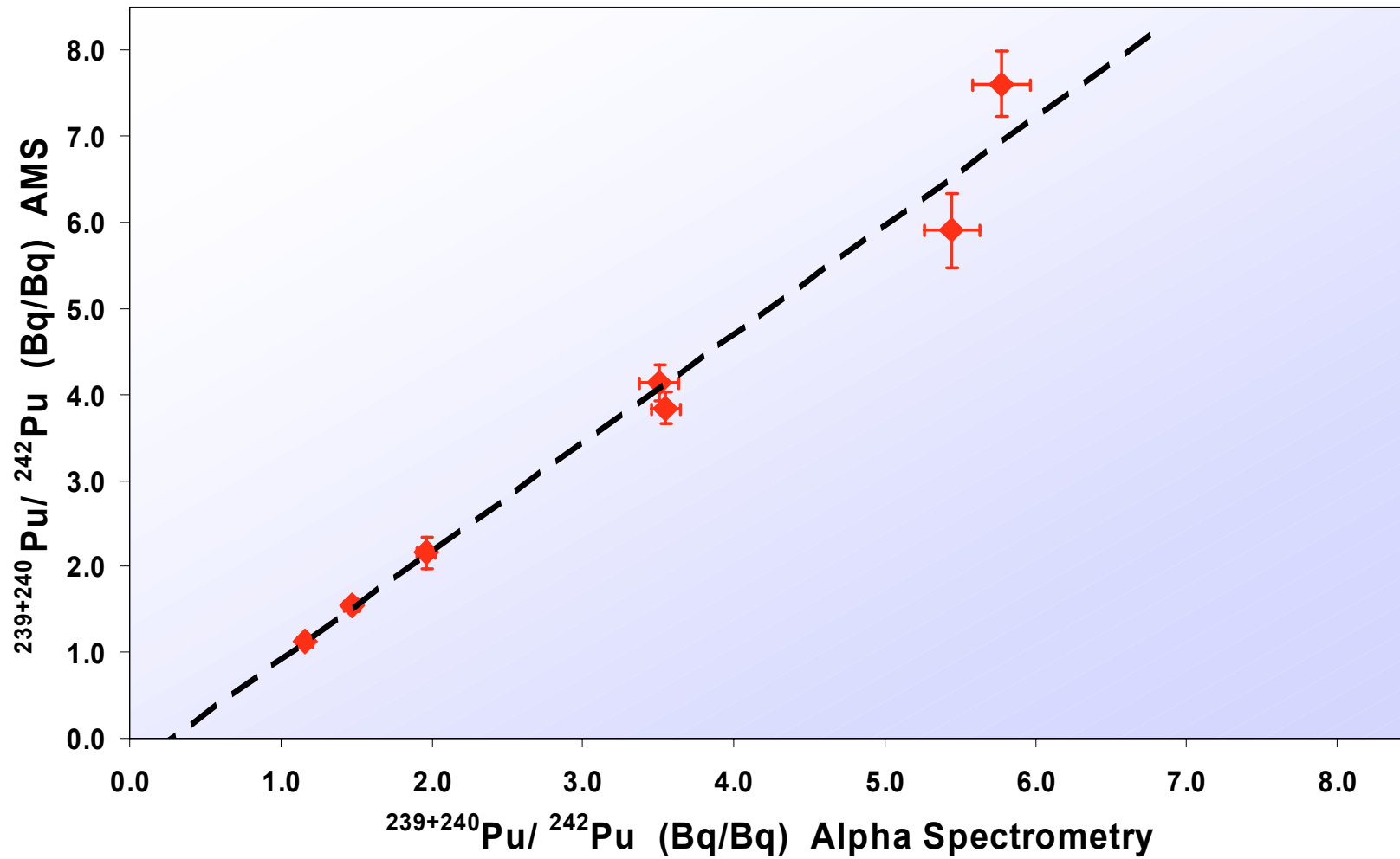
$$240/5 = 144/3$$

# Results

	AMS		Alpha Spectrometry	
	$^{240}\text{Pu}/^{239}\text{Pu}$ [at/at]	$^{239(40)}\text{Pu}$ [Bq/kg]	$^{239(40)}\text{Pu}$ [Bq/kg]	$^{238}\text{Pu}$ [Bq/kg]
11.4.1 Fangataufa (Kilo) loose coral rocks	$0.049 \pm 0.002$	$51.3 \pm 2.6$	$43.5 \pm 1.6$	$15.0 \pm 0.7$
11.4.3	$0.049 \pm 0.003$	$19.3 \pm 0.8$	$19.8 \pm 0.9$	$6.8 \pm 0.4$
11.4.4	$0.050 \pm 0.007$	$93.6 \pm 4.7$	$71.8 \pm 2.4$	$26.1 \pm 1.0$
9.3.6 Mururoa (Faucon) top soil	$0.018 \pm 0.002$	$175 \pm 15$	$159.1 \pm 4.9$	$0.77 \pm 0.18$
9.3.18	$0.017 \pm 0.002$	$115 \pm 5$	$109.9 \pm 3.7$	$0.59 \pm 0.16$
9.3.25	$0.018 \pm 0.003$	$221 \pm 11$	$204.3 \pm 5.4$	$1.21 \pm 0.19$
7.2.4.2 Mururoa (Colette)	$0.019 \pm 0.002$	$539 \pm 39$	$497 \pm 17$	$2.24 \pm 0.46$



# Comparison AMS - $\alpha$ -spectrometry



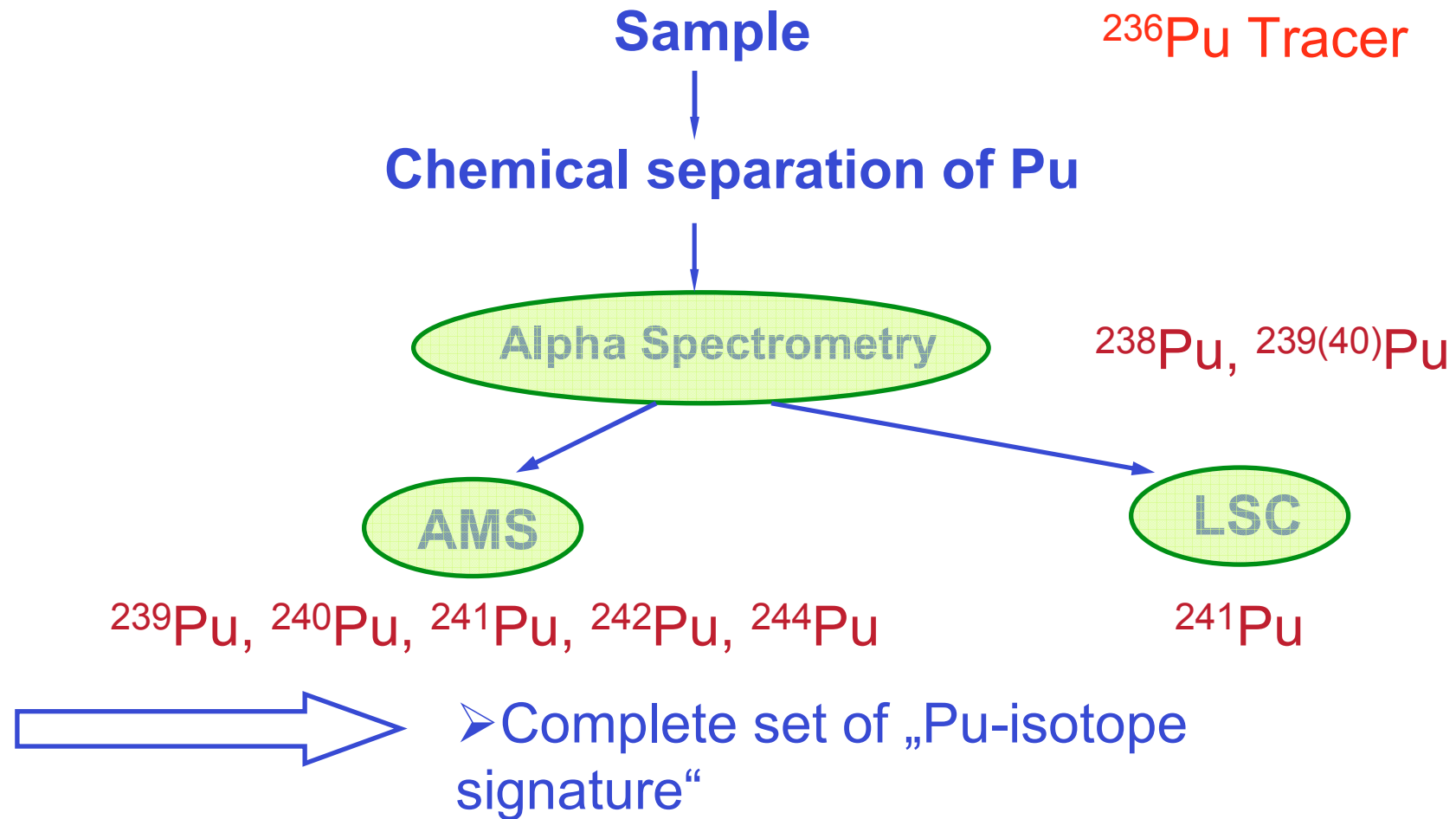
# Results

Test site	$^{238}\text{Pu}/^{239(40)}\text{Pu}$ (Alpha Spec)	
Mururoa	$0.0051 \pm 0.0006$	0.0044 Danesi et al. (2002)
Fangataufa	$0.35 \pm 0.01$	0.38 Mulsow et al. (1999)

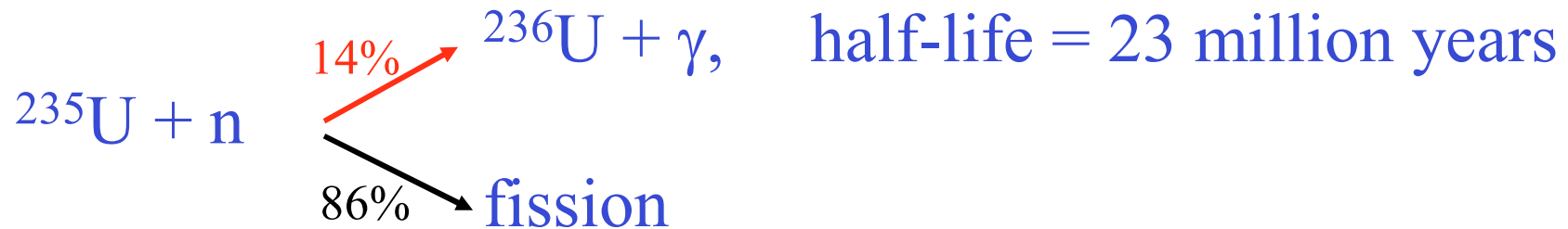
  

Test site	$^{240}\text{Pu}/^{239}\text{Pu}$ (AMS)	
Mururoa	$0.018 \pm 0.001$	< 0.03 safety tests Chiappini et al. (1999)
Fangataufa	$0.049 \pm 0.001$	0.05 Chiappini et al. (1999)

# Outlook: Analytical Strategy



## $^{236}\text{U}$ , a natural and anthropogenic neutron monitor



### Expected $^{236}\text{U}/^{238}\text{U}$ isotope ratios:

Uranium mineral:  $\sim 10^{-11}$

1 ppm U in rock:  $\sim 10^{-14}$

Power reactor:  $\sim 10^{-3}$  (after burning off 1%  $^{235}\text{U}$ )

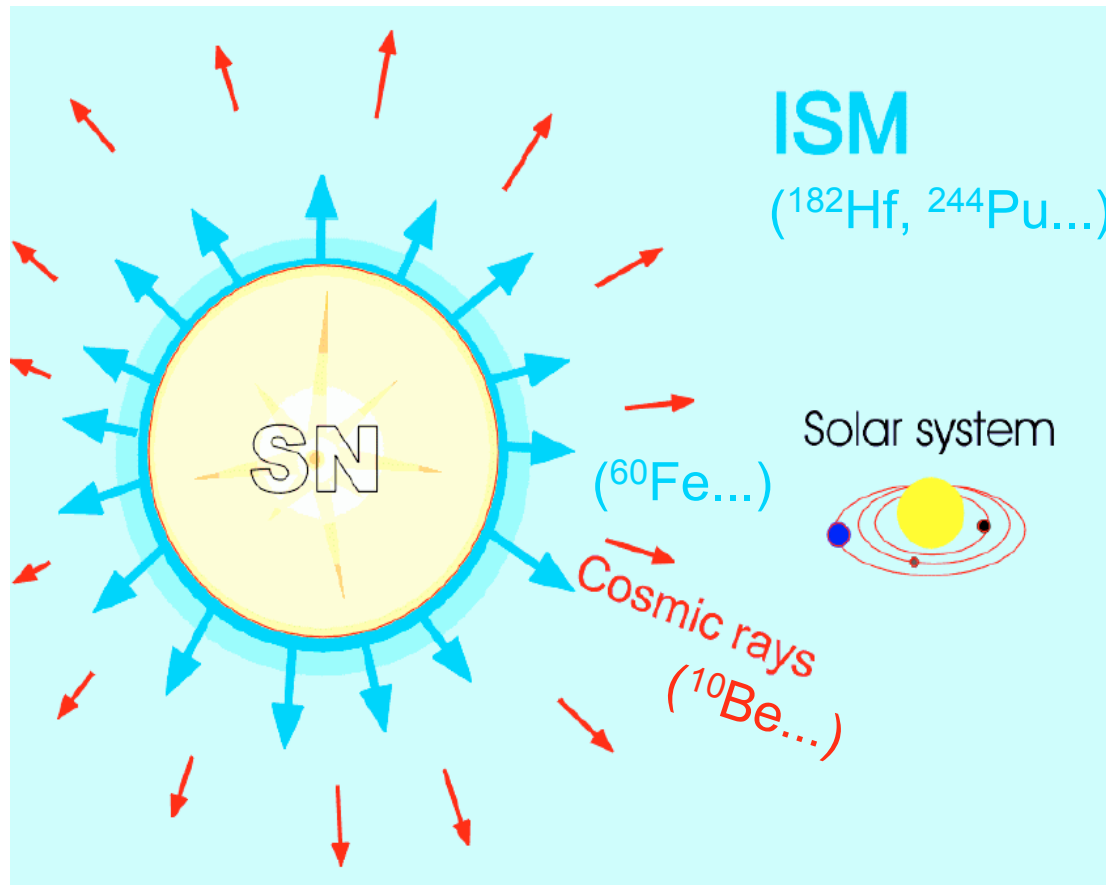
Lowest  $^{236}\text{U}/^{238}\text{U}$  ratio measured at VERA:  $6.1 \times 10^{-12}$

**Probably the most interesting frontier for  
heavy-isotope AMS:**

**Searching for live supernova remnants on Earth**

# Live long-lived radionuclides

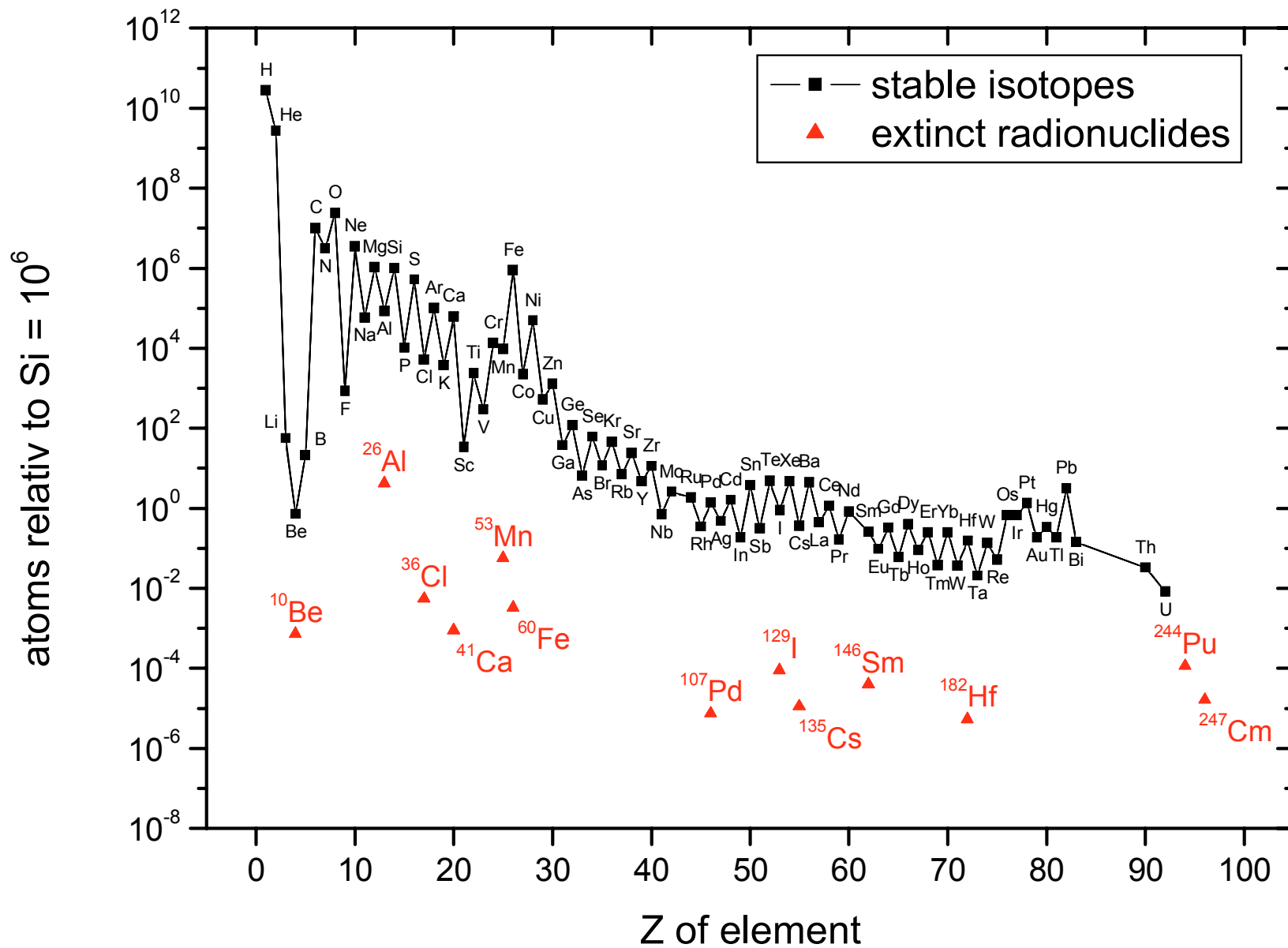
nearby supernova:  $< 100$  pc, rate  $\sim 0.3 - 10$  (Ma) $^{-1}$



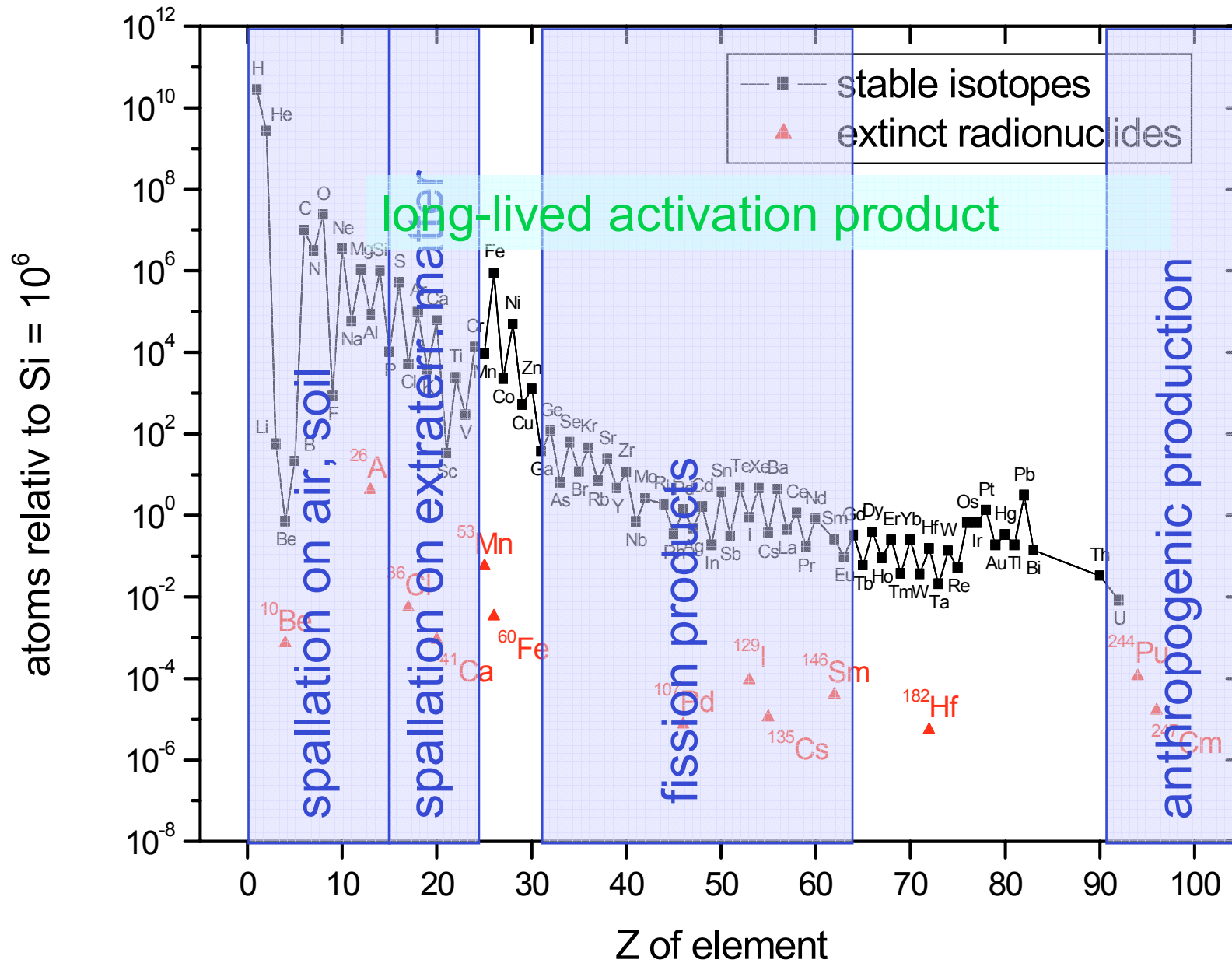
**Crab-nebula (SN from 1054 AD)  
Supernova Remnant  
at 2000 pc distance**

[http://antwrp.gsfc.nasa.gov/apod/image/9911/crab\\_vlt\\_big.jpg](http://antwrp.gsfc.nasa.gov/apod/image/9911/crab_vlt_big.jpg)

# Solar system abundance

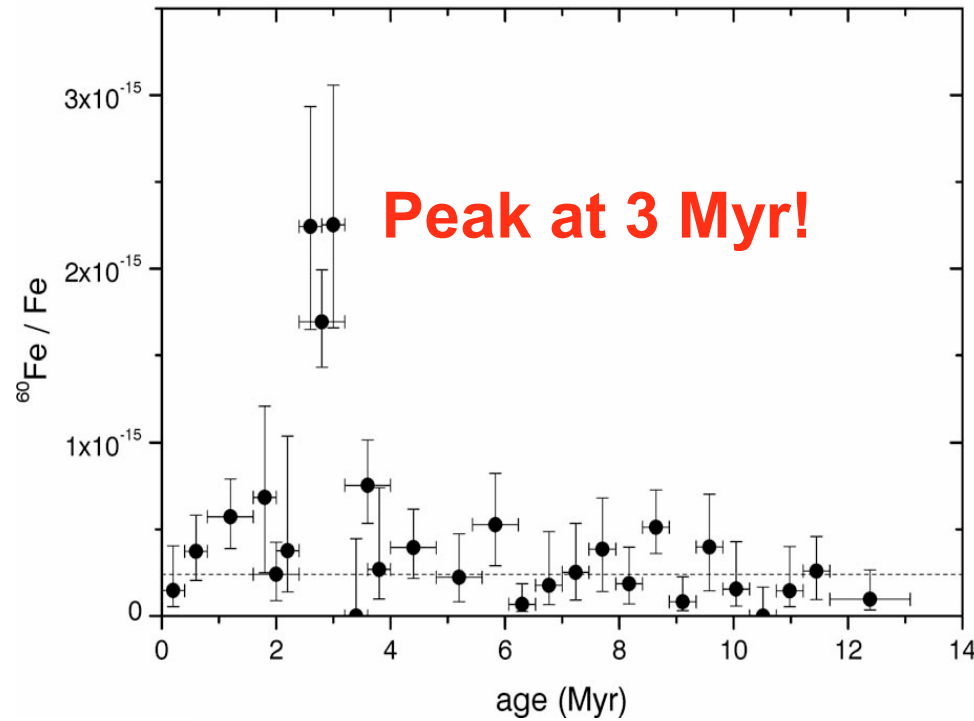


# Production of long-lived solar system radionuclides





# $^{60}\text{Fe}$ -signal in deep-sea crust



AMS measurement of  $^{60}\text{Fe}$  content of crust

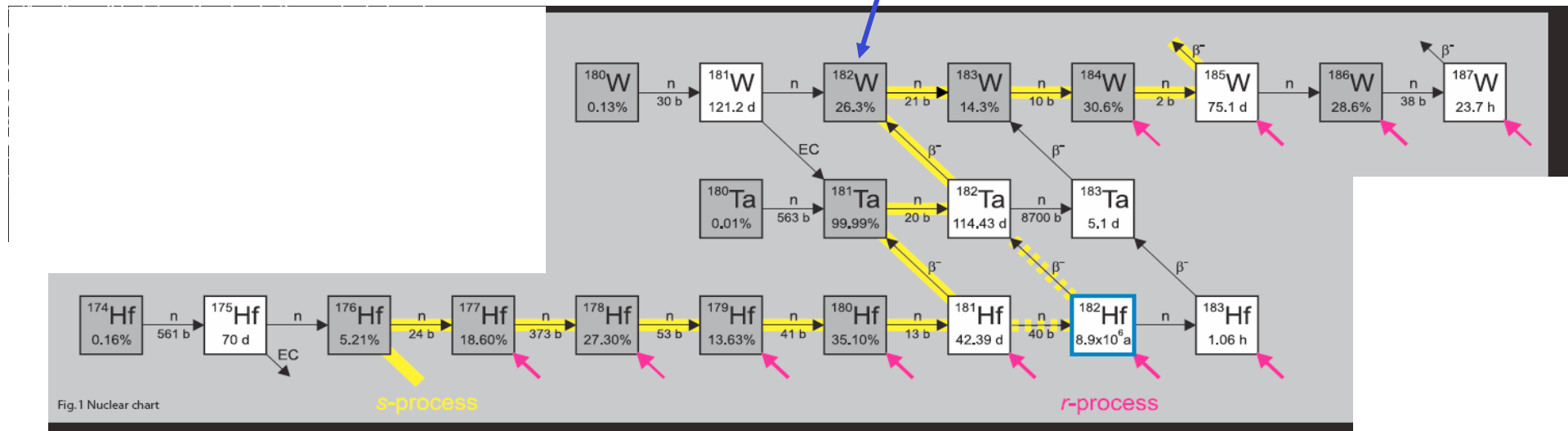
← background level

## $^{60}\text{Fe}$ Anomaly in a Deep-Sea Manganese Crust and Implications for a Nearby Supernova Source

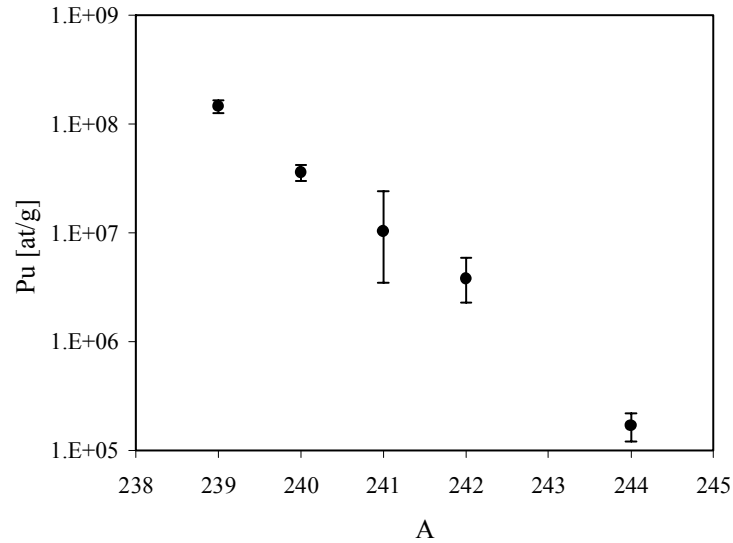
K. Knie,<sup>1</sup> G. Korschinek,<sup>1,\*</sup> T. Faestermann,<sup>1</sup> E. A. Dorfi,<sup>2</sup> G. Rugel,<sup>1,3</sup> and A. Wallner<sup>1,3</sup>

**$^{182}\text{Hf}$  would be a good candidate for a supernova remnant:**

But interference from stable isobar  $^{182}\text{W}$  limits  $^{182}\text{Hf}/^{180}\text{Hf}$  currently to  $\sim 10^{-11}$ .



# **$^{244}\text{Pu}$ : Is our Earth too much contaminated to find the faint signal of supernova-produced $^{244}\text{Pu}$ ?**



## **Manganese nodule**

**external ratios“  
(rel. to  $^{239}\text{Pu}^1$ )**

- **$^{239}\text{Pu}$ : 1**
- **$^{240}\text{Pu}$ : 0.25**
- **$^{241}\text{Pu}$ : 0.01 (0.07 d. corr.)**
- **$^{242}\text{Pu}$ : 0.026**
- **$^{244}\text{Pu}$ : 0.001**

Fig1.  
Figure 1 shows the measured concentrations of the plutonium isotopes. The value of  $^{241}\text{Pu}$  ( $T_{1/2} = 14.35$  y) is decay corrected for 1960.

	$^{239}\text{Pu}$	$^{240}\text{Pu}$	$^{241}\text{Pu}$	$^{242}\text{Pu}$	$^{244}\text{Pu}$
cts.	142	58	2	7	12
C [ $10^6$ at/g]	$146 \pm 19$	$36 \pm 6$	$1.5^{+2.0}_{-1.0}$	$3.8^{+2.1}_{-1.5}$	$0.17 \pm 0.05$

Table1.  
The measured counts of each Pu isotope are indicated. The deduced concentration with the statistical error is given in  $10^6$  at/g. In the case of  $^{241,242}\text{Pu}$  are the errors calculated according to Feldman and Cousins [ref] for low count numbers.

<sup>1</sup> after: C. Wallner et al., New Astron. Rev. (2003)

**And finally:**

**Why not searching for the unknown such as superheavy elements? An AMS facility can certainly be tuned to regions of the nuclear chart which are far away from known nuclides.**