



The Abdus Salam  
International Centre for Theoretical Physics



SMR.1744 - 5

## **SCHOOL ON ION BEAM ANALYSIS AND ACCELERATOR APPLICATIONS**

**13 - 24 March 2006**

**Ion Beam Analysis - Applications in materials science**

**Gabor BATTISTIG  
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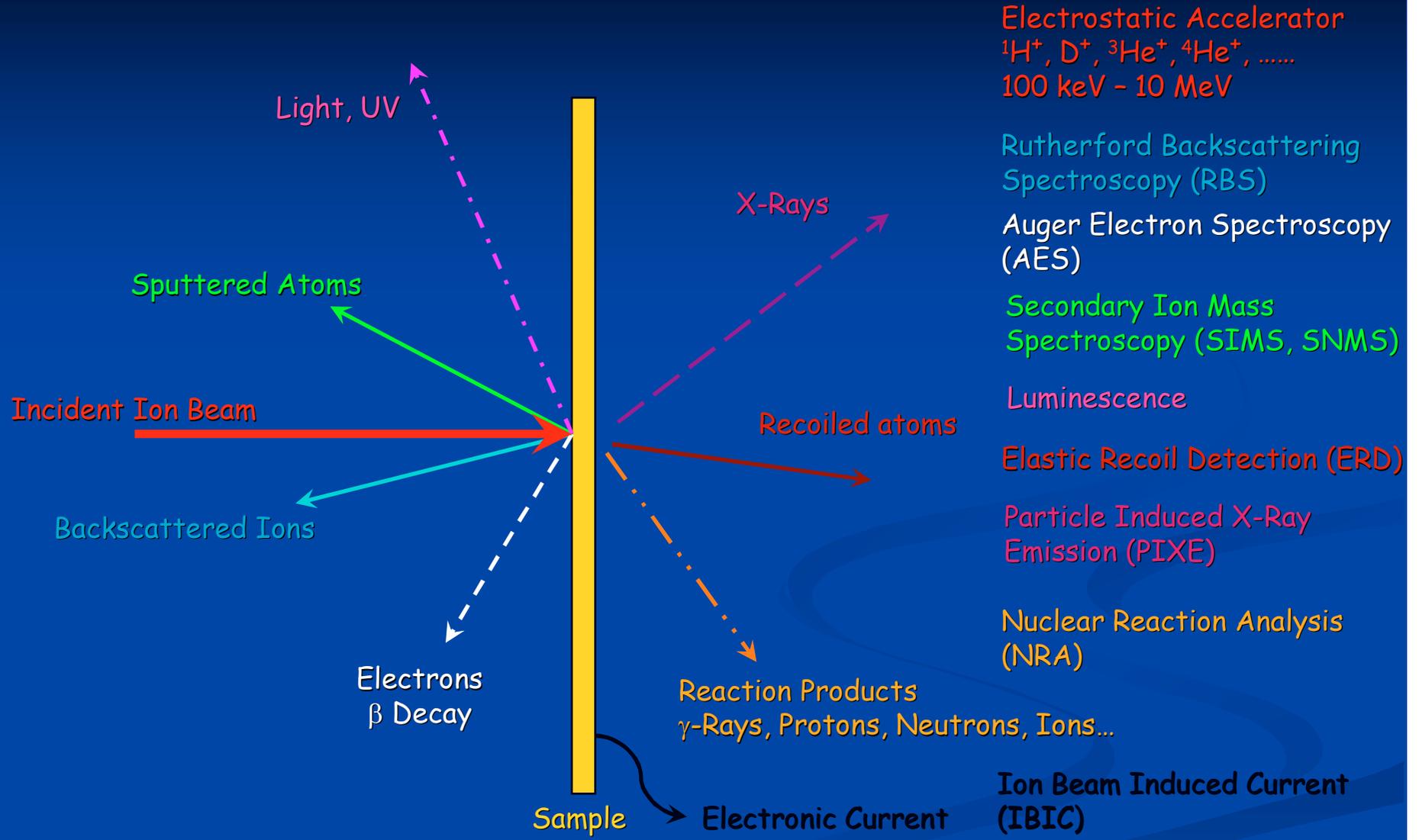
# Ion Beam Analytical methods and applications in Materials Science

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## Ion beam analytical techniques

Acronym		Interaction
<b>RBS</b>	<b>Rutherford Backscattering Spectrometry</b>	Elastic scattering at backward angles
<b>ERDA or FRS</b>	<b>Elastic Recoil Detection Analysis, Forward Recoil Spectroscopy</b>	Elastic recoil at forward angles, not necessarily Rutherford
<b>NRA</b>	<b>Nuclear Reaction Analysis</b>	Nuclear reaction between incident beam and nuclei in the target, producing a light charged particle.
<b>NRP or r-NRA</b>	<b>Nuclear Resonance Profiling, resonant Nuclear Reaction Analysis</b>	Exploitation of narrow nuclear resonances via scanning of the incident beam energy.
<b>PIXE</b>	<b>Particle-Induced X-ray Emission</b>	Characteristic X-ray emission following ionization by the primary beam.
<b>PIGE</b>	<b>Particle-Induced Gamma Emission</b>	Prompt gamma emission during ion beam irradiation

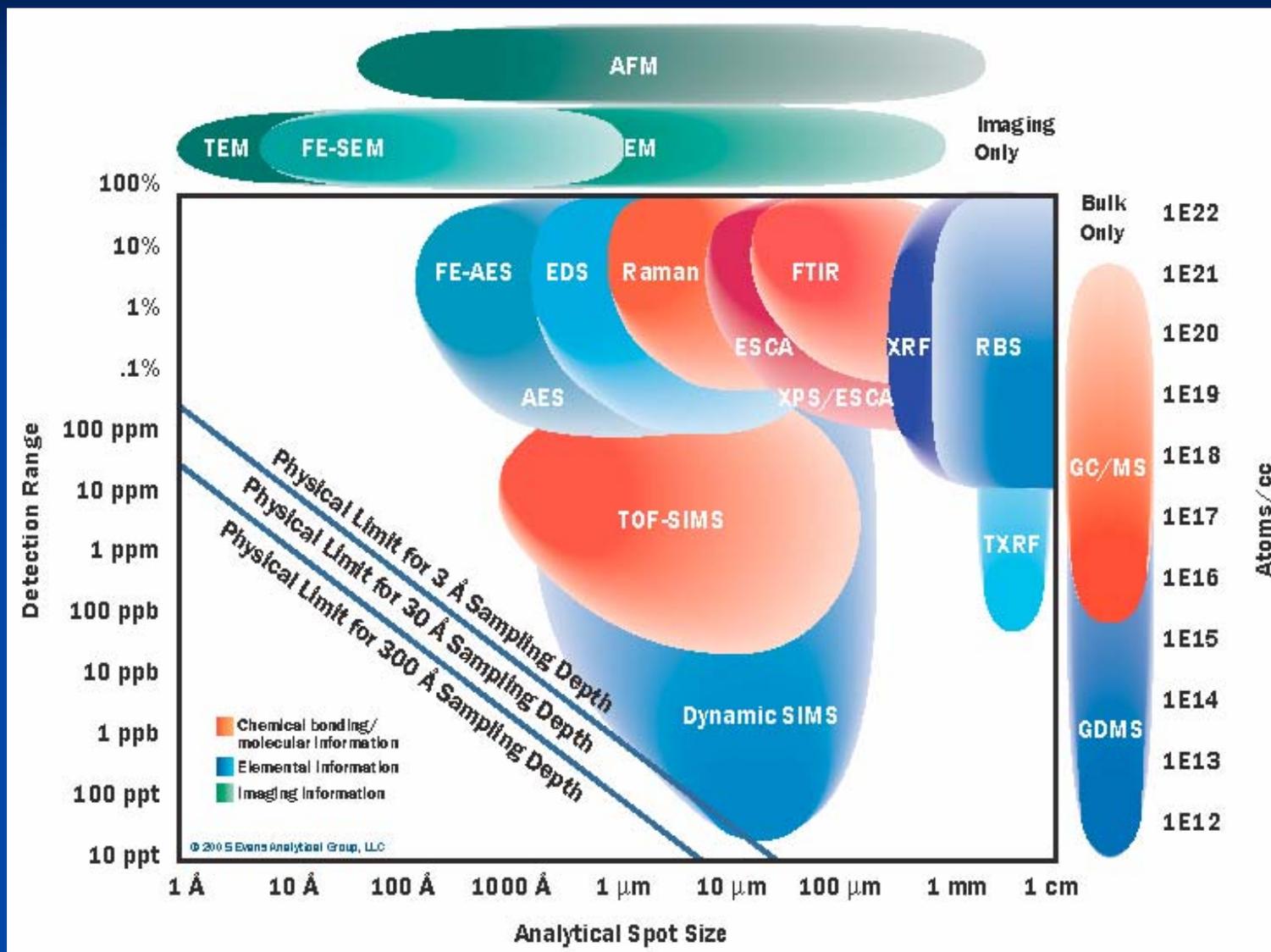


## Main questions relevant to the study of the near surface regions of materials

- Atomic composition / concentration profiles
- Atomic structure, crystallinity
- Atomic movements (transport)



## Most used analytical techniques



## SIMS, SNMS



An energetic ion beam is used to sputter atoms from the surface. Secondary ions emitted are mass analyzed.

Sensitivity  $\approx 10^{12} - 10^{14}$  atoms/cm<sup>3</sup>

Depth resolution  $\approx 10 - 200$  Å

Lateral resolution  $\geq 1$  μm (imaging)

## TXRF



X-rays incident at very shallow angles excite the surface atoms.

The atoms relax through the emission of a characteristic X-ray.

Depth resolution  $\approx 30 - 80$  Å



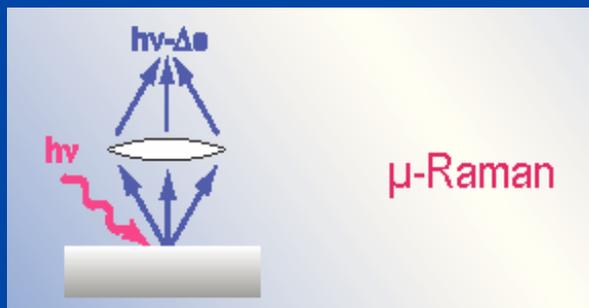
## FTIR



Infrared beam excites the sample, molecular constituents vibrate in the IR regime. The molecules, surrounding environments and concentrations of the oscillating chemical bonds may be determined.

Depth resolution  $\approx 0.1 - 1 \mu\text{m}$

## Raman



It measures the intensity of light scattered inelastically off the sample as a function of the wavelength. Chemical bonds gain or lose characteristic amount of energy, that correspond to different vibrational modes in the bonds.



## XPS



Surface of the sample is excited with X-rays, photoelectrons are emitted from the atoms near the surface. Chemical bonding information may be determined.

Depth resolution  $\approx 10 - 100 \text{ \AA}$   
Detection limit  $\approx 0.001 - 1 \text{ at\%}$

## AES



Focused electron beam produces Auger electrons with discrete energies, specific to each element. (Escape depth  $\approx 10 \text{ \AA}$ ) (Sputtering is needed.)

Depth resolution  $\approx 20 - 200 \text{ \AA}$   
Detection limit  $\approx 0.1 - 1 \text{ at\%}$

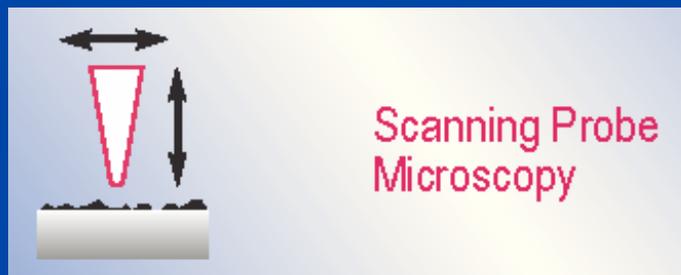


## SEM / TEM



Finely focused  $e$ -beam is rastered over the sample surface, its image is formed from secondary electrons and photon emission. Energy dispersive X-ray (EDX) spectroscopy is also possible - elemental quantification.

## STM / AFM



A very sharp tip, located a few  $\text{\AA}$  of the surface, is rastered over the sample surface. Surface morphology and physical properties are measured.



## SE



Polarized light scattered on the sample surface, complex refractive index can be determined, layer structure can be obtained. *Good depth scale!*

## Methods:

- Ion beam techniques (IBA): RBS, MEIS, ERD, NRA, resonance depth profiling, PIGE, PIXE, channeling, AMS ...  
"Non-destructive", in situ, at atmospheric pressure
- Sputtering techniques: SIMS, SNMS, AES .....  
Destructive
- Microscopy: SEM, TEM, Electron probe .....  
"Non-destructive or destructive", sample preparation artefacts
- X-ray diffraction, XRF .....  
Non-destructive

**BUT with these methods: No chemical, electrical, magnetic, optical, mechanical, etc. information.**



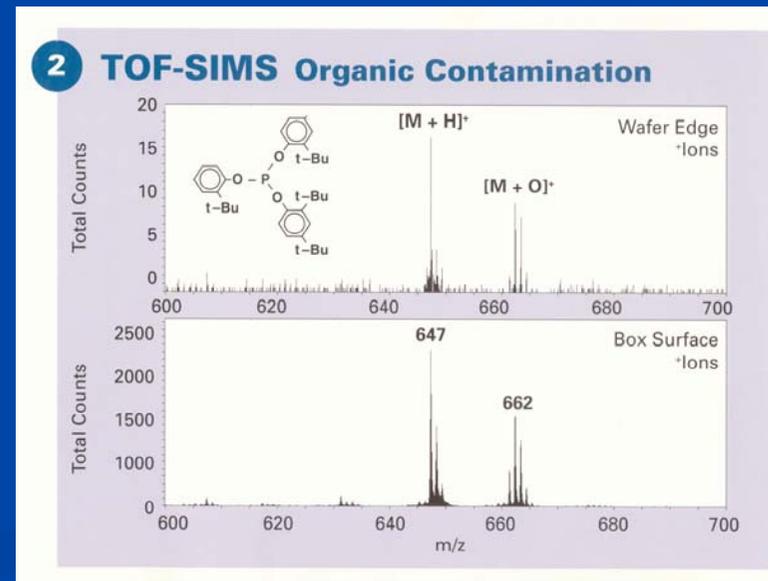
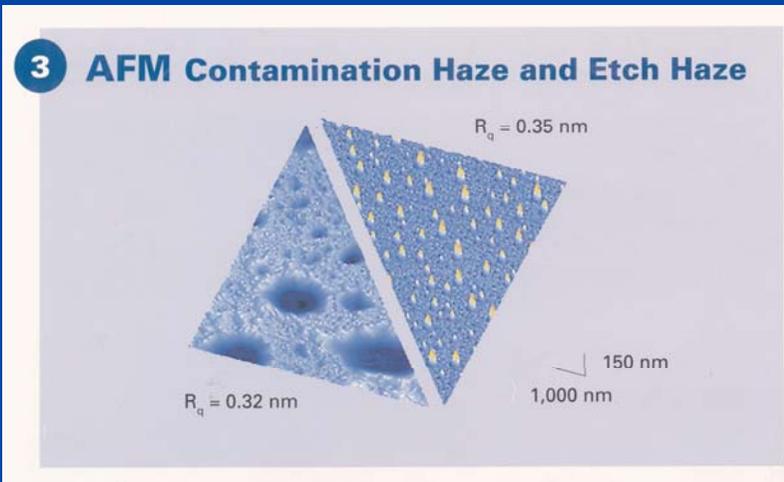
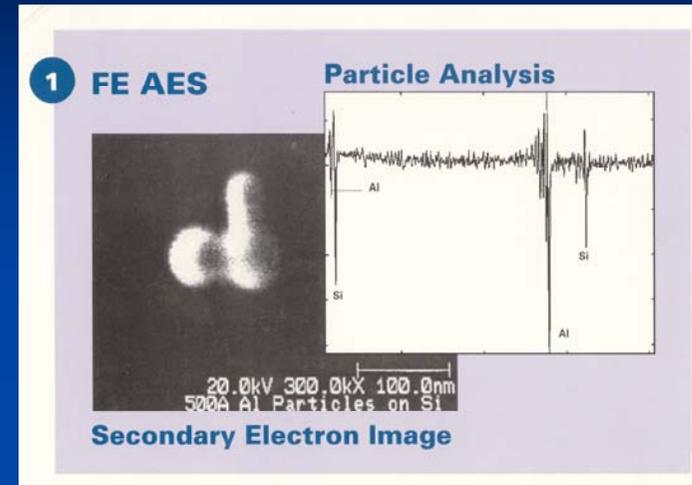
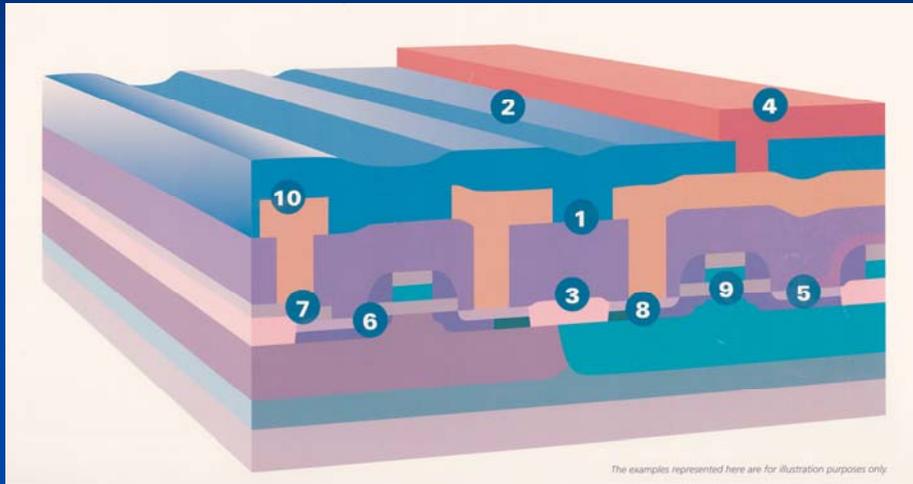
## Complementarity:

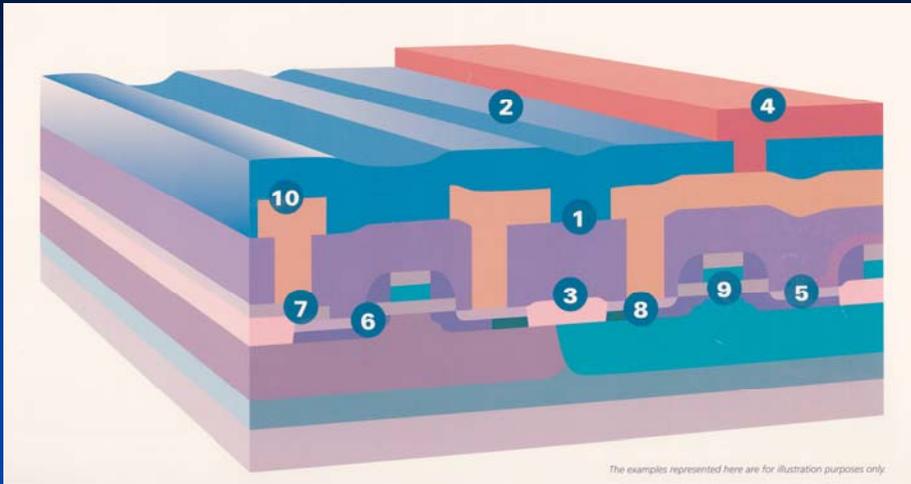
IBA  $\Leftrightarrow$  SIMS  $\Leftrightarrow$  SEM

- Near field microscopy, STM, AFM .....
- Spectroscopic Ellipsometry

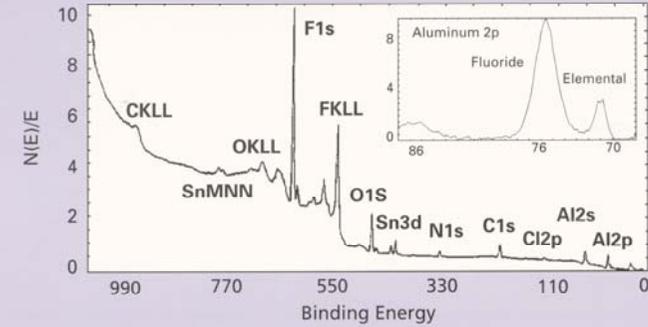


Analytical techniques used to study typical CMOS structure

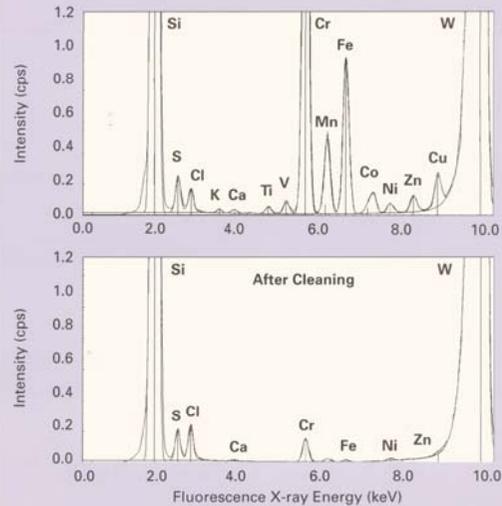




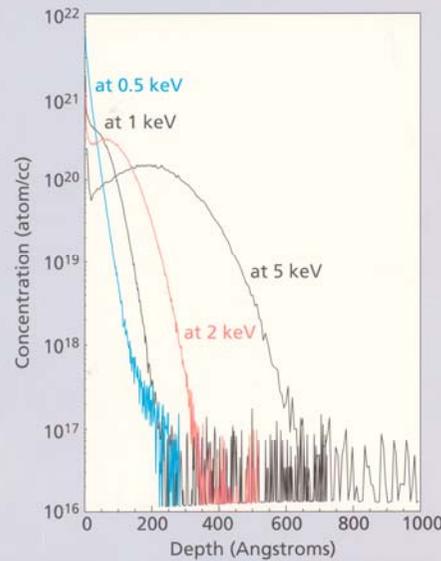
## 4 XPS/ESCA ID of Wafer Discoloration



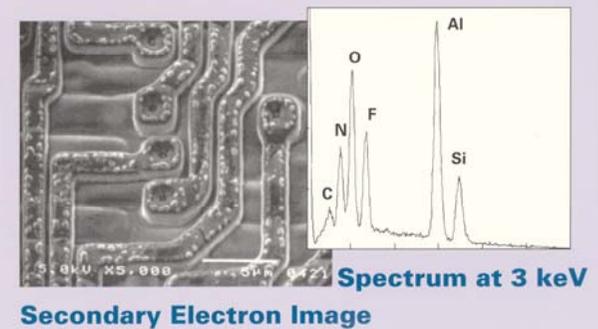
## 5 TXRF Cleaning Efficacy Measurement

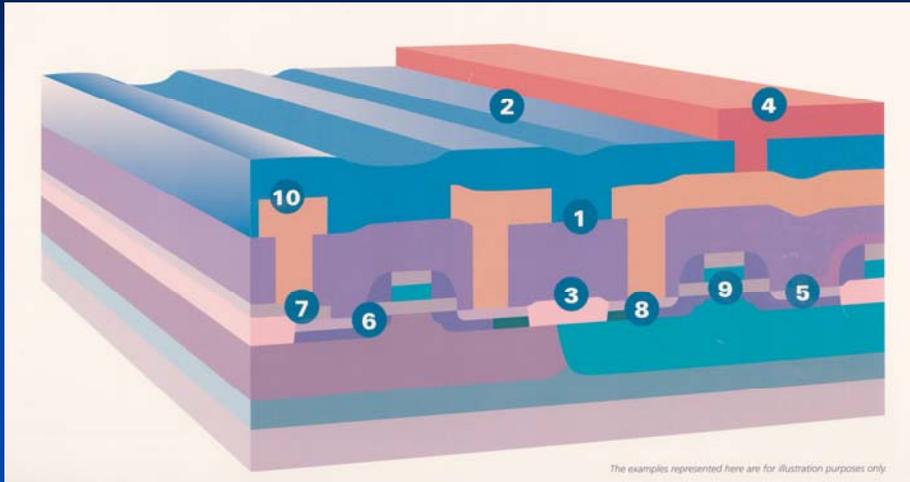


## 6 SIMS B<sup>+</sup> Implant in Pre-Amorphized Silicon, 5 x 10<sup>14</sup> at/cm<sup>2</sup>

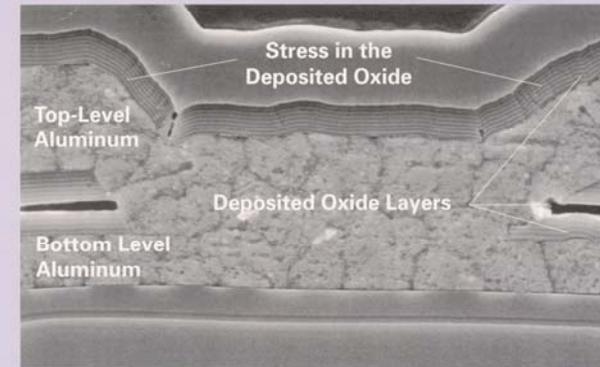


## 7 EDS Residue on Top of Metal Lines



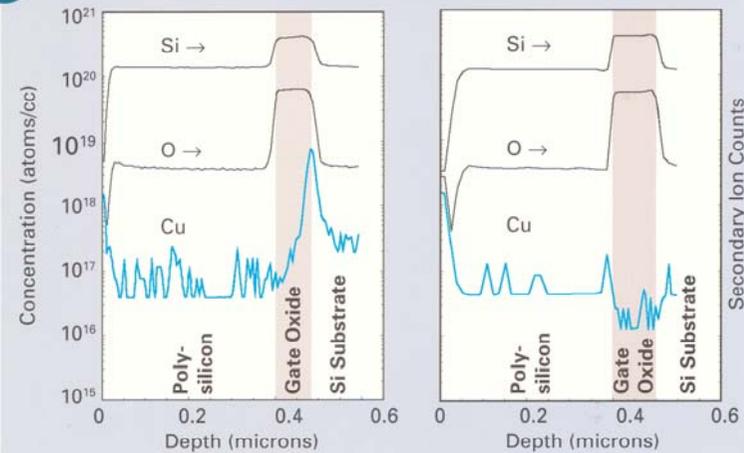


## 10 FE SEM Cross Sectional Analysis

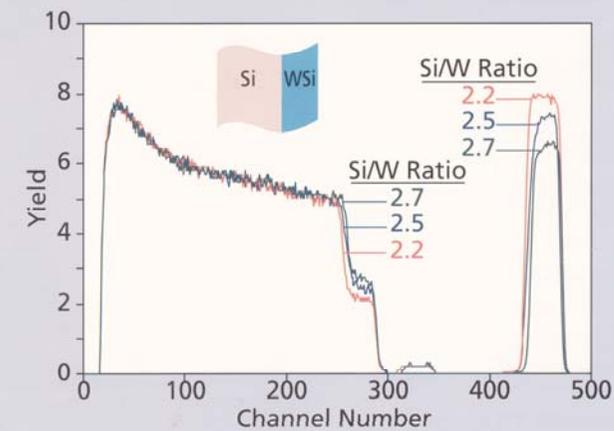


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## 9 SIMS Gate Oxide Breakdown



## 8 RBS Composition of Various $W_{Si_x}$ Films



## Atomic composition

### ➤ IBA:

#### Matrix independent

- Absolute areal density in thin films - at/cm<sup>2</sup> (↘ 10<sup>11</sup> – 10<sup>12</sup>)
- Concentration profiles - at/cm<sup>3</sup> (↘ 1-100 ppm)
- Depth scale well defined (10Å – 10μm)
- μBeam (lateral resolution ↘ ~1μm but possible beam damage)
- NRA: isotope specific

### ➤ SIMS, AES:

#### Matrix dependent yield, over ~50Å integrated yield

- Excellent sensitivity and mass discrimination, but preferential sputtering, crater formation, .....faceting effects, possible giant roughening
- Depth scale given through sputtering rate, only reliable in special cases, calibration is required
- high resolution mapping



## Atomic composition

### ➤ Microscopy:

**Only structural information, atomic resolution**

- with electron probe (X-ray detection): composition, but high Bremsstrahlung, background, integration over  $\sim 1 \mu\text{m}^3$



## Atomic Structure

➤ IBA:

**Channeling**

- information in direct space, crystal structure, defects, lattice location; amorphisation and recrystallisation studies....

➤ SIMS, AES:

- No structural information

➤ Microscopy:

- atomic resolution, defect identification (e-diffraction), but no elemental information

➤ X-ray diffraction:

- lattice identification, phases



## Atomic movements

Main questions:

diffusion, implantation, annealing, solid state reactions,  
growth mechanisms .....

➤ **IBA:**

Isotopic tracing with rare isotopes

- Resonant NRA, RBS, MEIS .....

➤ **SIMS, AES:**

- Good mass separation, high sensitivity,  
but strong sputtering effects

➤ **Microscopy:**

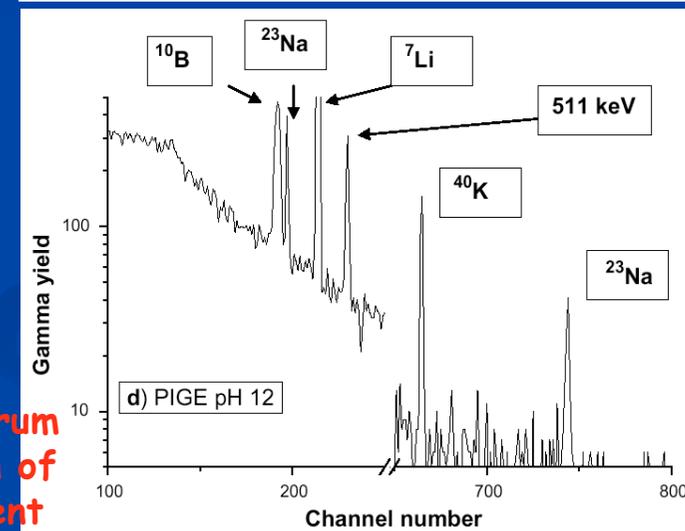
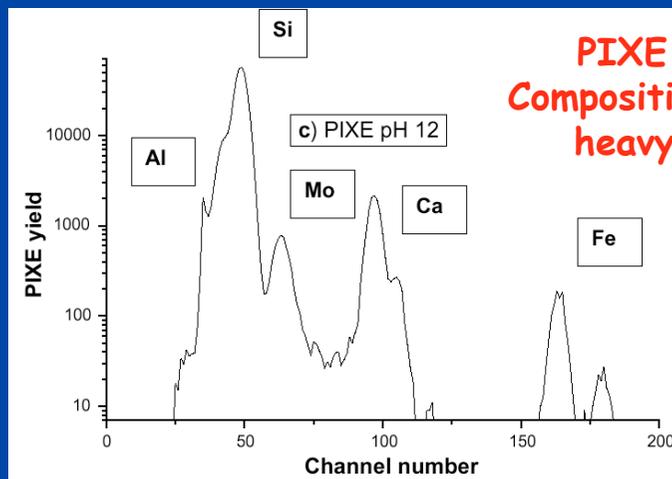
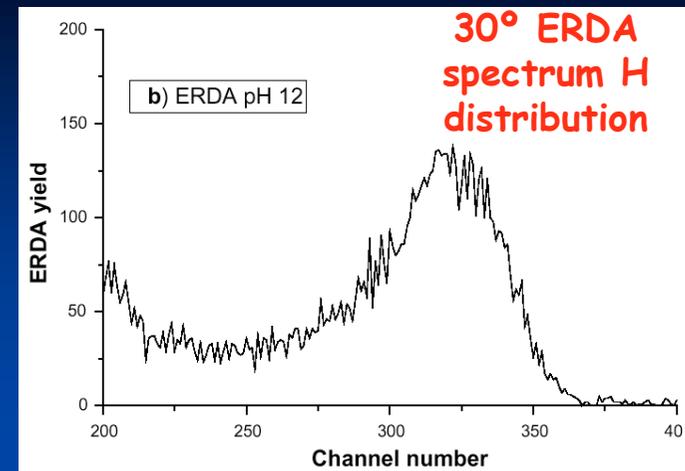
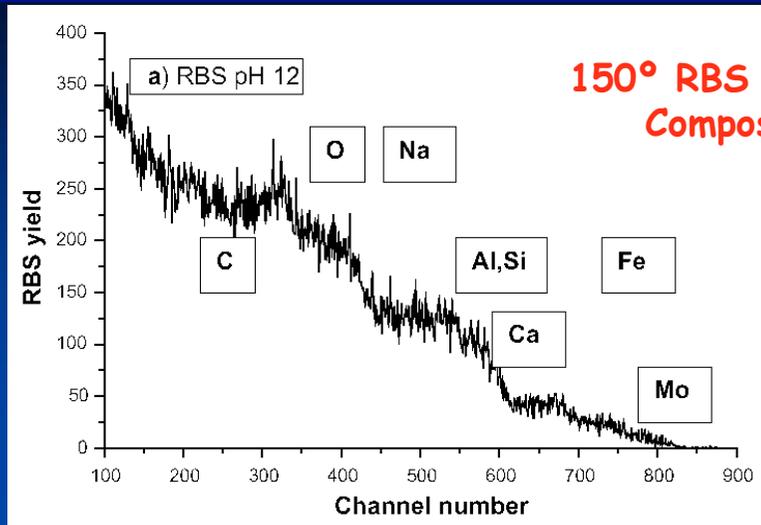
- No, but defect identification and evolution

➤ **X-ray diffraction:**

- Lattice identification, phases



# School of Ion Beam Analysis and Accelerator Applications



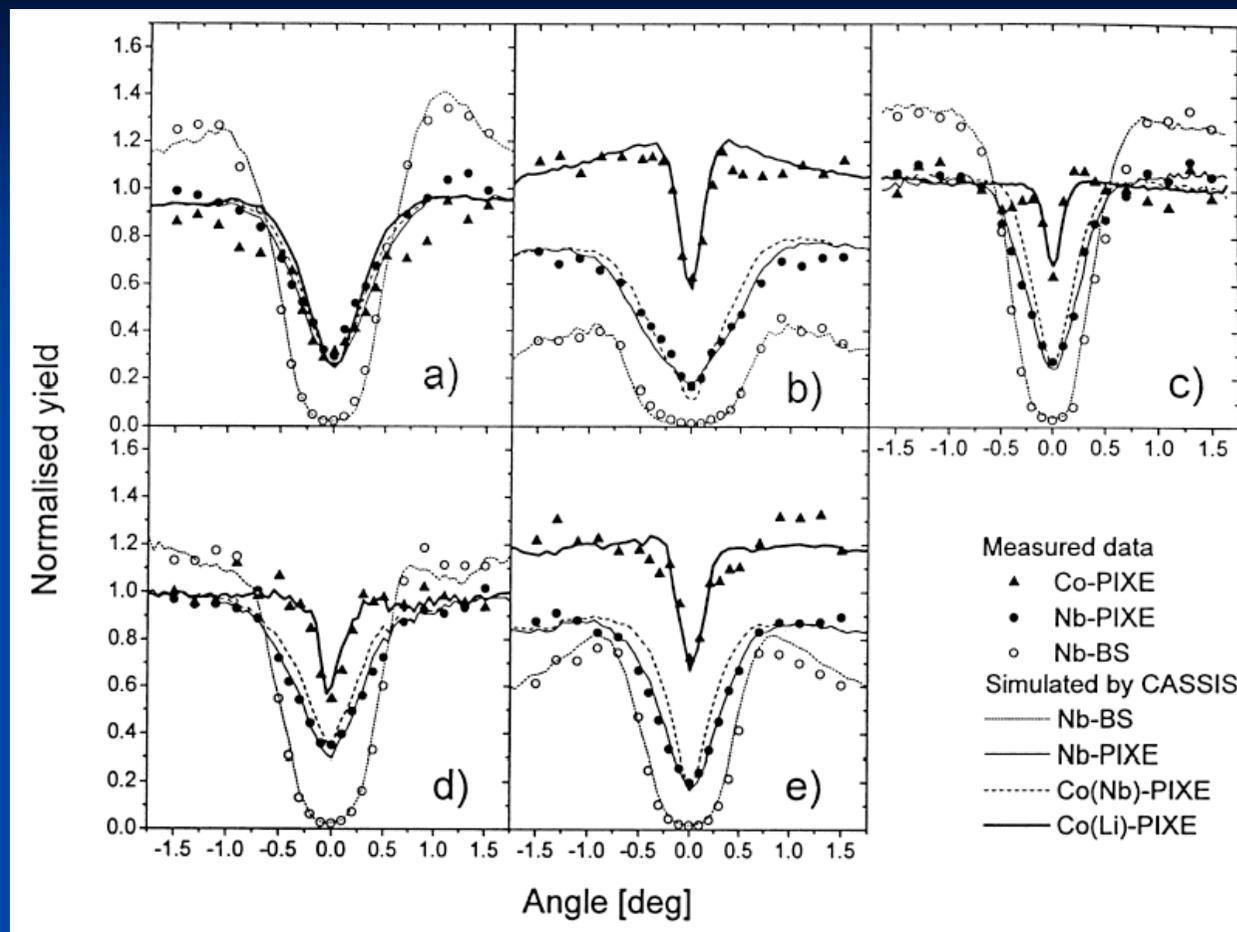
Alkali borosilicate glass samples after leaching at 90°C, 7 days at pH=12. Glass dissolution rate is high at pH 0 and 7, while it is strongly reduced at 12, due to the presence of Fe and Mo containing phases near the surface. Homogeneous hydration of the glass surface.

J.Chêne and P. Trocellier Journal of Non Crystalline Solids 337 (2004) 86-96, Fig. 5



## Structural information Channeling

Experimental angular scans for Nb-BS, Nb-PIXE, Co-PIXE in 0.2 mol% Co doped  $\text{LiNbO}_3$  for the five major axes. Simulations using the CASSIS program for Nb-BS, Nb-PIXE and Co-PIXE (the latter with the assumptions that Co occupies exclusively Li sites or exclusively Nb sites) are also shown.



The normalised yields were determined for: (a)  $\langle 0001 \rangle$  at  $50^\circ$  with respect to the (01-10) plane; (b)  $\langle 02-21 \rangle$  within the (1-102) plane; (c)  $\langle 01-10 \rangle$  at  $15^\circ$  with respect to the (2-1-10) plane; (d)  $\langle 0-441 \rangle$  at  $80^\circ$  with respect to the (2-1-10) plane; and (e)  $\langle 11-20 \rangle$  within the (0001) plane.

The channeling measurements proved that the lattice site of Co in low-doped congruent  $\text{LiNbO}_3$  is a regular Li-site, a crucial finding for interpreting  $^{57}\text{Co}$  Mössbauer effect experiments.

E. Szilágyi et al. Solid State Com. 115 (2000) 535 Fig.3

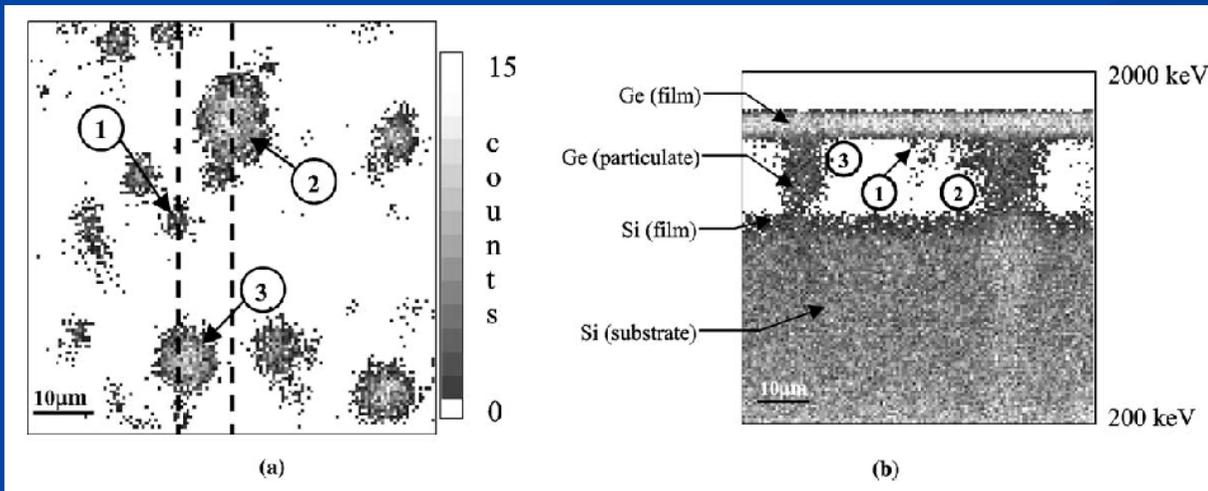
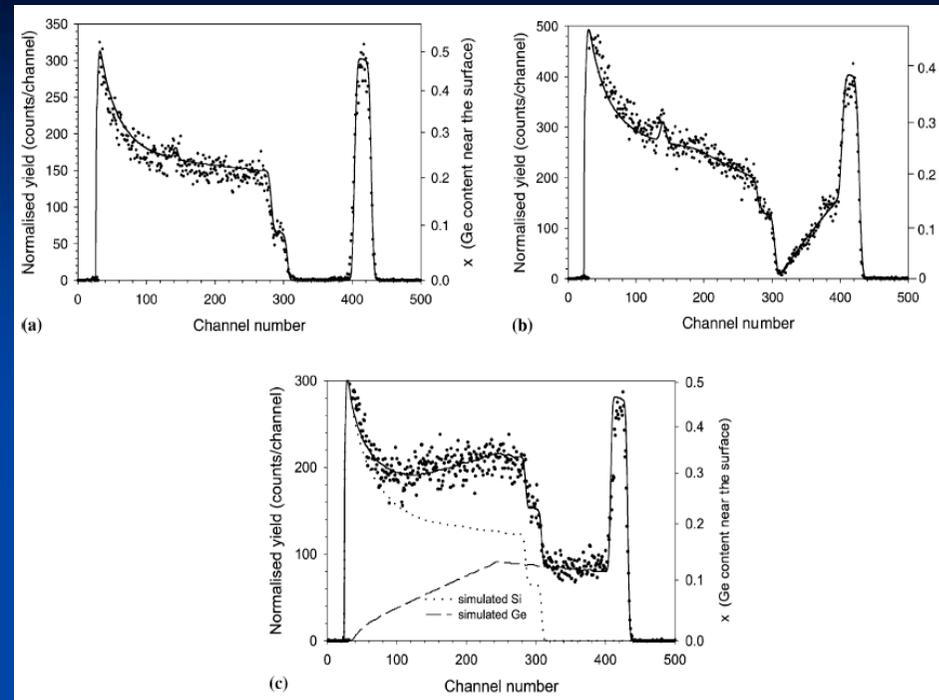


## Micro-RBS mapping of particulates from laser deposited $\text{Si}_{1-x}\text{Ge}_x$ thin films on Si

(a) Ge elemental map

(energy window between ch.320 and 390,  $65 \times 65 \text{ mm}^2$  scanning area,  $128 \times 128$  pixel),

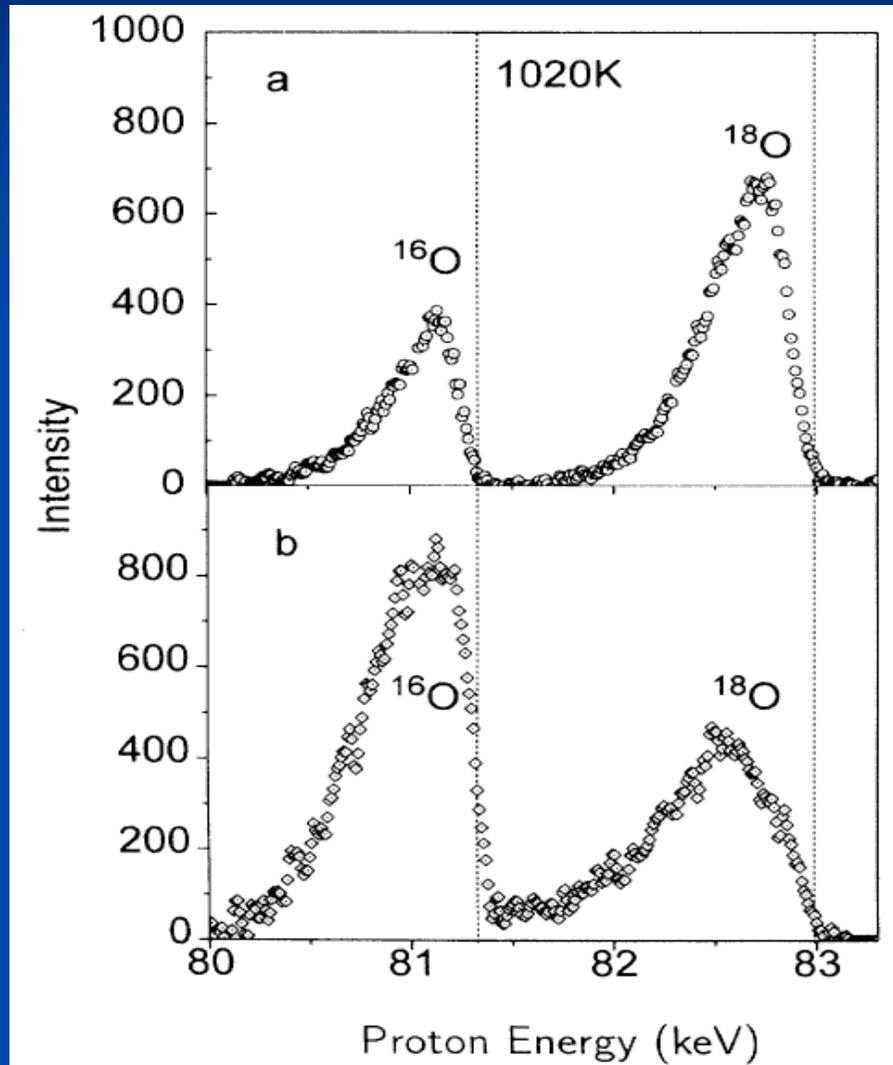
(b) tomographic image created from 15 pixels wide region between the vertical lines.



A.Simon and Z.Kántor, NIMB 190 (2002) 351–356, Figs. 3 and 4



## High depth resolution isotopic tracing by $^{18}\text{O}$ with MEIS (100 keV protons)



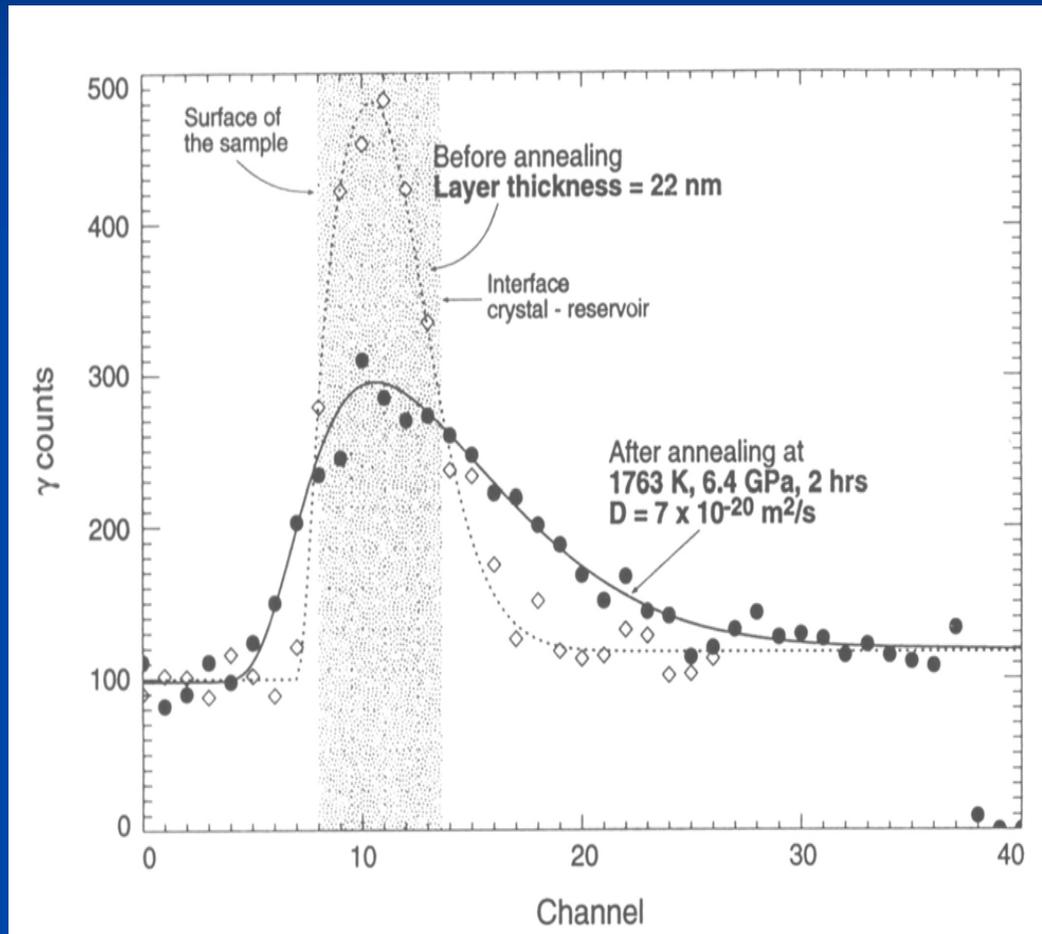
Study of growth mechanisms of ultrathin oxide films on Si, here in the 15-25 Å range. Overlapping isotope depth profiles are found, in contrast to thicker films (>40 Å). After 165 min exposure the growth rate is strongly reduced and  $^{18}\text{O}$  loss is observed. Combination of MEIS with LEIS, XPS and spectroscopic ellipsometry led to a detailed picture of the mechanisms of the first stages of thermal growth of oxide on Si.

MEIS spectra for Si(100) samples oxidized at 1020 K and  $10^{-2}$  Torr. Exposure in  $^{18}\text{O}_2$  were 10 min for both samples. This first oxidation step is followed by oxidation in  $^{16}\text{O}_2$  for (a) 165 min and (b) 2640 min.

E.P. Gusev et al. Phys. Rev. B52 (1995) 1759. Fig.6



## $^{30}\text{Si}$ isotopic tracing study of Si self diffusion in olivine of geophysical interest



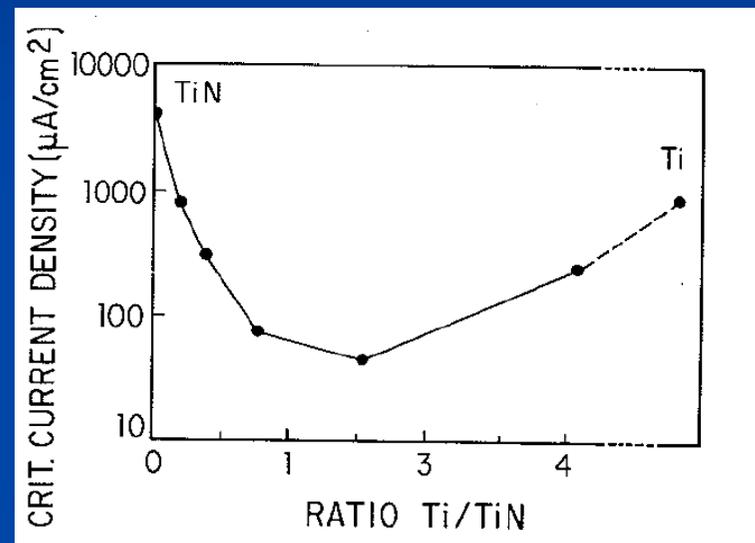
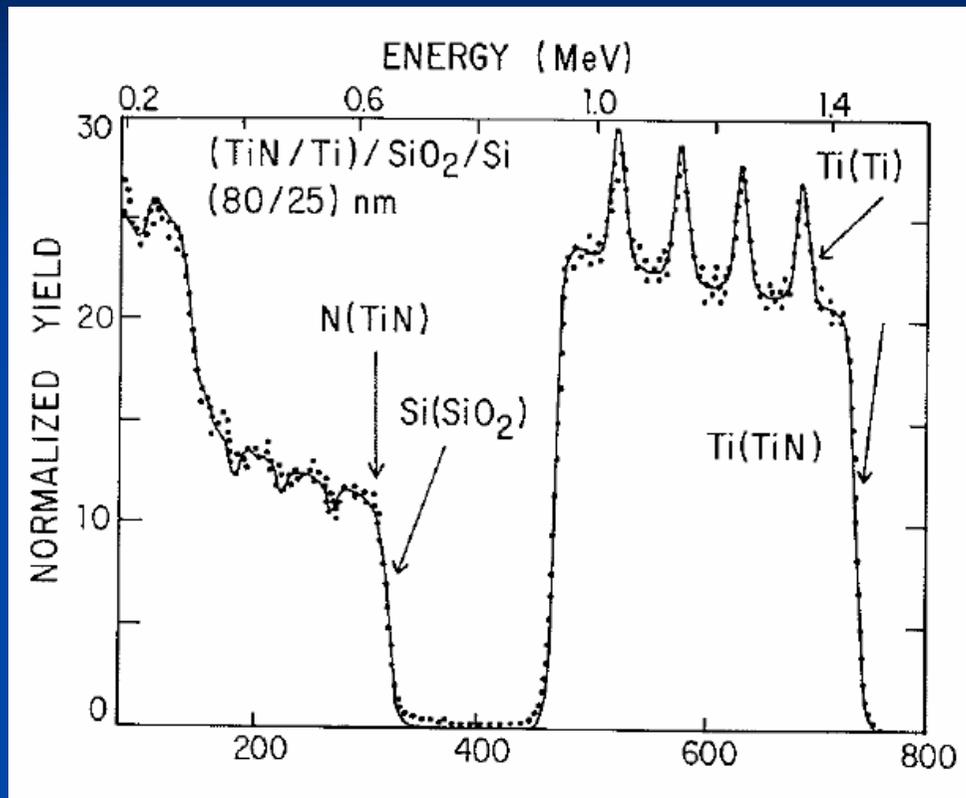
A  $^{30}\text{Si}$  enriched 22 nm thick fosterite layer was deposited by sputtering on the sample and annealed at 1763 K and 9 GPa. The high resolution resonance depth profiling of  $^{30}\text{Si}$  ( $^{30}\text{Si}(p,\gamma)$ ) at 620.4 keV,  $\Gamma=70$  eV) yields the self-diffusion coefficient of Si. From that the viscosity in extreme conditions of olivine, very abundant in the earth mantle, may be calculated.

Fosterite:  $\text{Mg}_2\text{SiO}_4$   
Olivine:  $(\text{Mg,Fe})_2\text{SiO}_4$

F. Bejina et al. Geophysical Research Letters  
24 (1997) 2597, Fig.2



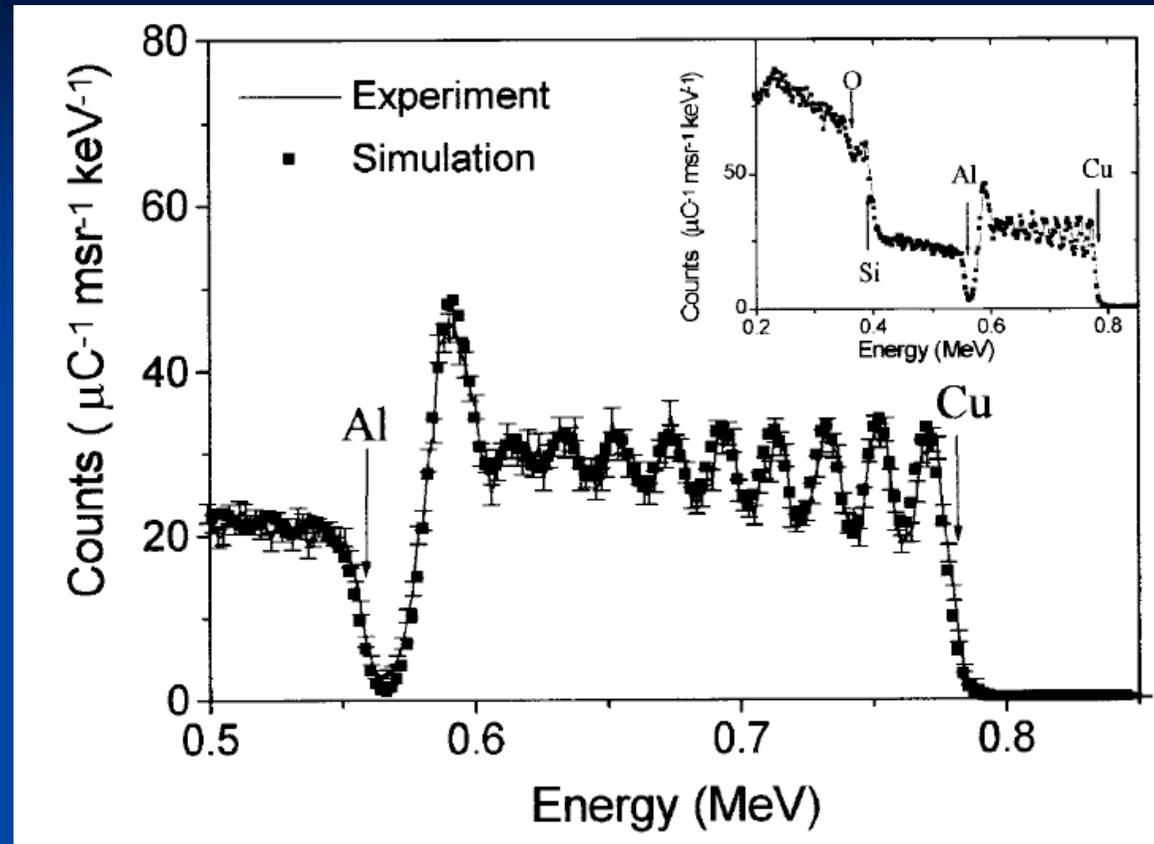
## Multilayer structures



Reactive magnetron deposited multilayered TiN/Ti thin film coating is adherent, stress free, fully dense. 0.5 $\mu\text{m}$  coatings produce same performance as 3 $\mu\text{m}$  single layered ones. Precise knowledge of the compositions and thickness ratios is essential for optimising hardness, elasticity and aqueous corrosion protection. a) 2 MeV  $\alpha$ -RBS spectrum showing the individual layers. b) critical current density vs TiN/Ti ratio for iron dissolution in cyclic voltametric aqueous corrosion tests of coated carbon steels, determined by RBS.  $^{14}\text{N}$  was also measured with  $^{14}\text{N}(d,\alpha)$  reaction.

I.J.R. Baumvol, NIMB, 85 (1994) 230, Fig.3



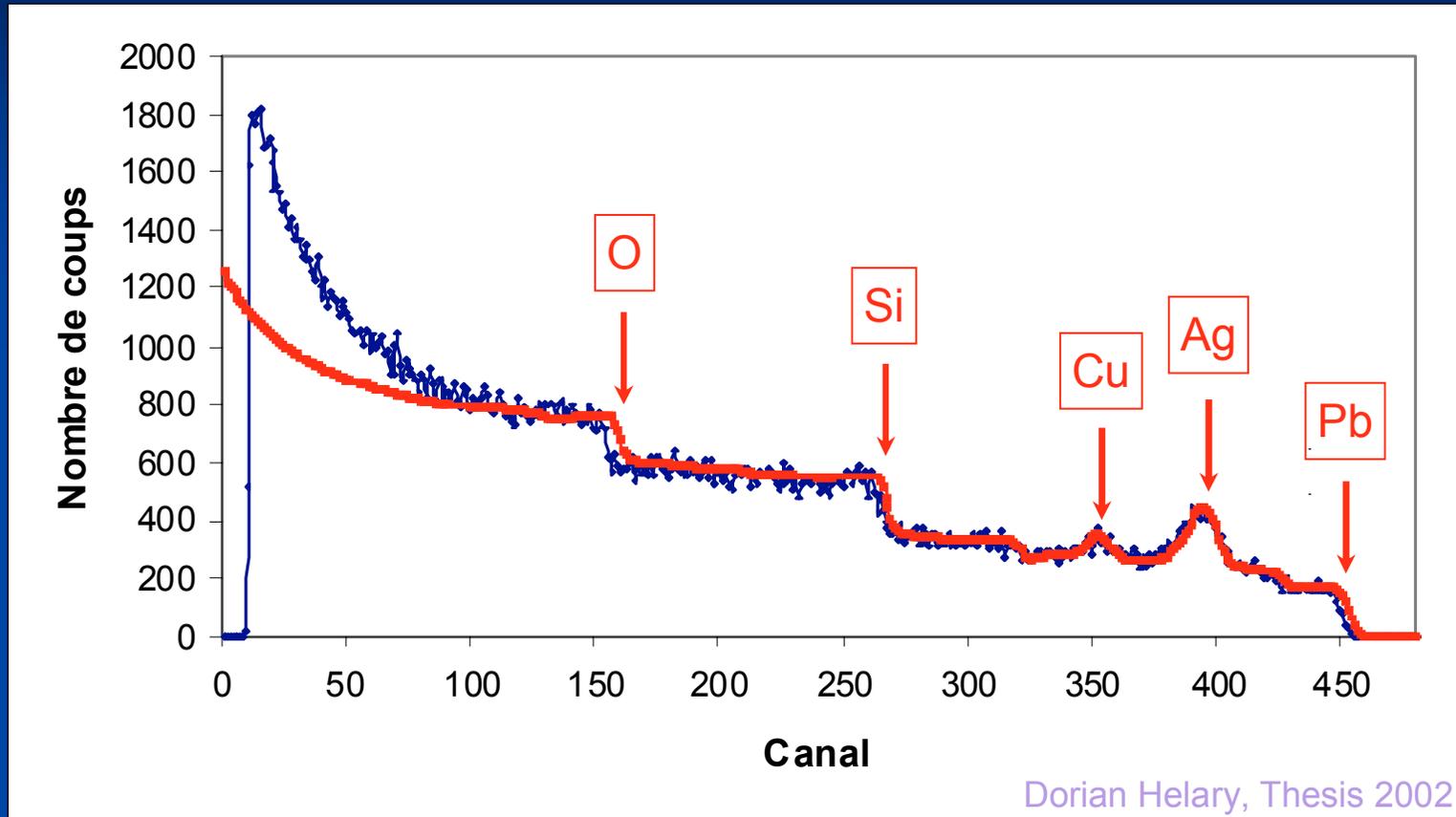


The formation of  $\text{Cu}:\text{Al}_2\text{O}_3$  composite films on  $\text{Si}(001)$  with alternate pulsed laser deposition yielding nucleated Cu nanocrystals with tailored configuration embedded in  $\text{Al}_2\text{O}_3$  was studied in view of their special optical, electrical and magnetic properties. The Cu content of up to 10 successive layers was measured individually with 1 MeV  $\alpha$ -RBS and  $42^\circ$  tilted sample. The first Cu deposit directly on the Si substrate is  $\sim$  twice as rich as the others. In combination with high resolution TEM full information on the growth process was obtained.

R. Serna et al. Appl. Phys. A 71 (2000) 583. Fig. 1



This powerful technique may be extended to many fields of Materials Science not related to Museum research



Experimental (blue) and SIMNRA simulated (red) 3 MeV  $\alpha$ -RBS spectra obtained from a lustered islamic ceramics. Beam diameter 50mm, extracted in He.

Majors (averaged over 6 layers): Pb 0.6%, Ag 3%, Cu 3%, Ca 3%, Si 30%, O 60%.

Overall thickness of the luster 760nm.

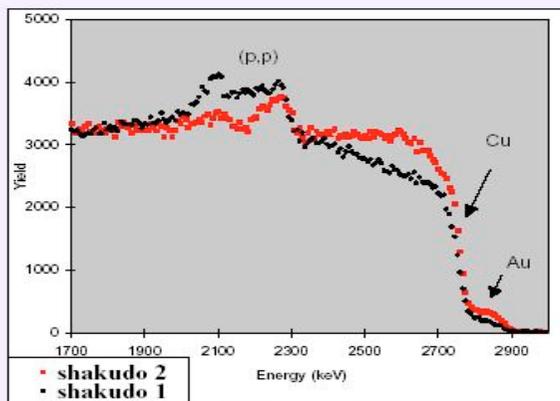
The first layer (40nm) contains no heavy metal and is transparent, a new finding.

Measurements with beams extracted in He - Louvre Laboratory, J. Salomon et al.

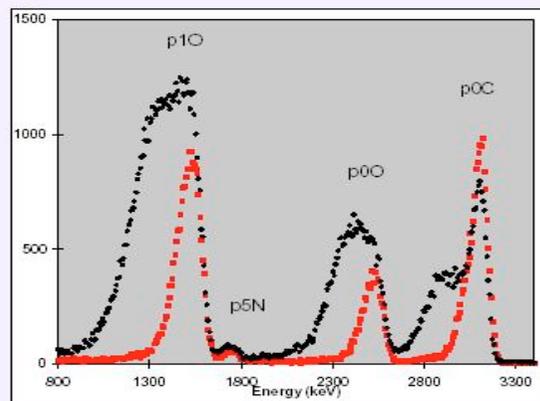


## Tsuba patinas

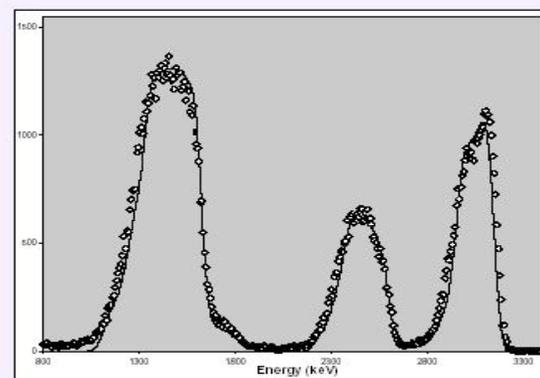
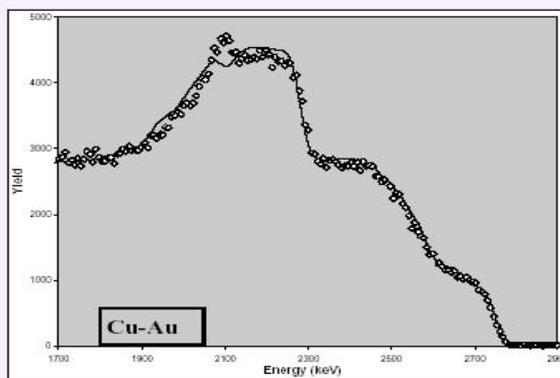
### RBS spectra



### NRA spectra



### Traditional Japanese shakudo



Experiment  
RBS protons at 3 MeV  
NRA deuterons at 2.3 MeV  
(d,p) reactions, C, N, O

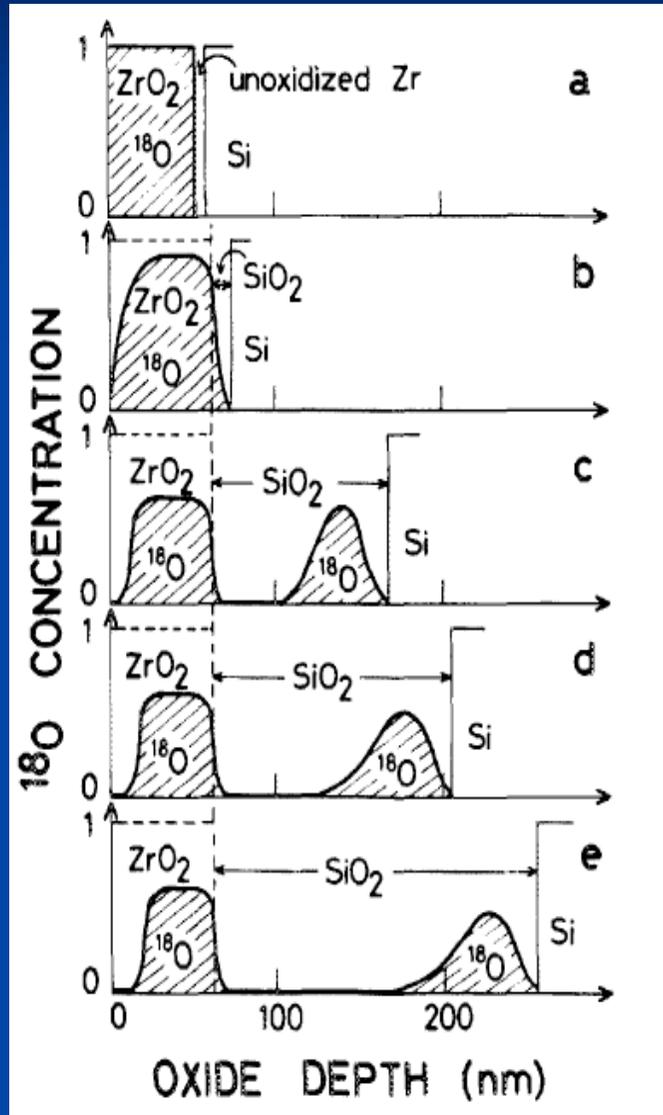
Results  
The combination of the three techniques allows us to identify the composition and tells us the thickness of the patina layer. This is a precious guide for understanding the fabrication techniques.  
The figure shows modern and traditional patinas. Once the fabrication technique is known it can be used as an artist signature for identification (fingerprint).

Tsubas are Japanese sword guards. Shakudo is a Cu-Au alloy (here ~ 4%). The colour of its patina depends on its oxidation state. Red and black dots belong to two different items, with patina thicknesses 1 $\mu$ m and ~ 3 $\mu$ m. The black spectrum shows some bulk carbon.

Louvre Laboratory, Evanthia Ionnidon Thesis 1999



## $^{18}\text{O}$ study of oxygen transport mechanisms during plasma oxidation of Si through an ionic conductor



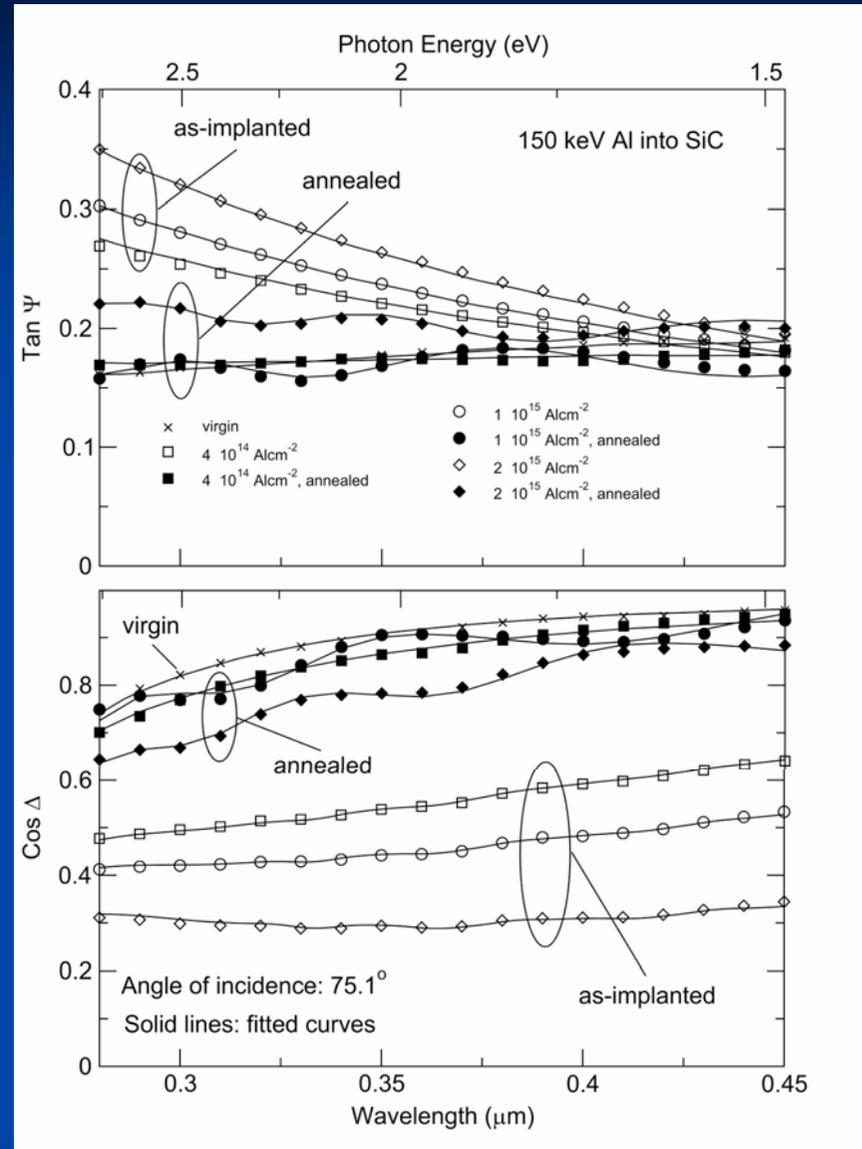
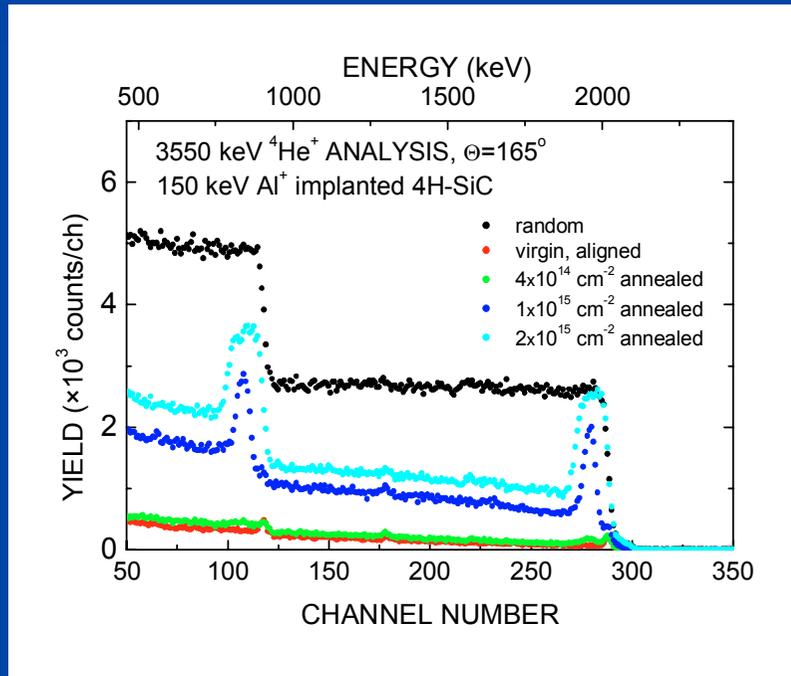
Si does not plasma oxidise directly, but does so through a thin  $\text{ZrO}_2$  layer ( $\sim 60$  nm, 84%  $^{18}\text{O}$  enriched) formed on its surface, active oxygen being provided by this ionic conductor.  $^{18}\text{O}$  tracing with the 629 keV,  $\Gamma=2.1$  keV resonance of  $^{18}\text{O}(p,\alpha)$  showed that the oxygen from the  $\text{ZrO}_2$  penetrates into the silica layer formed, remaining at the oxide/Si interface via a short range migration process, in contrast with thermal oxidation. The process is nearly temperature independent between  $25^\circ\text{C}$  and  $600^\circ\text{C}$ .

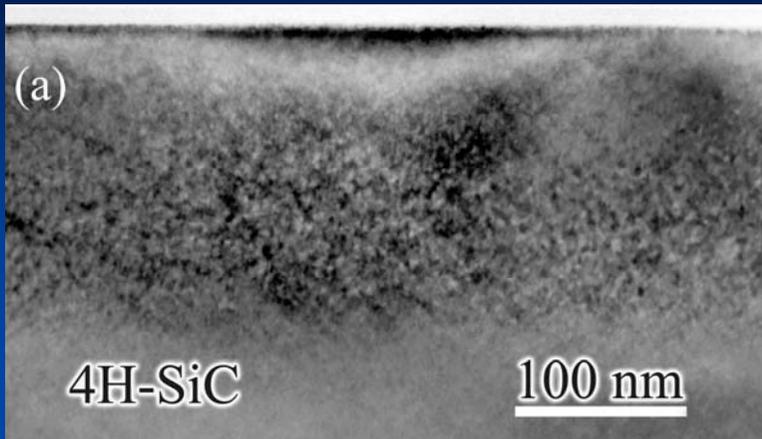
Siejka et al. J.Appl.Phys 56 (1984) 2720 Figs. 8 and 9



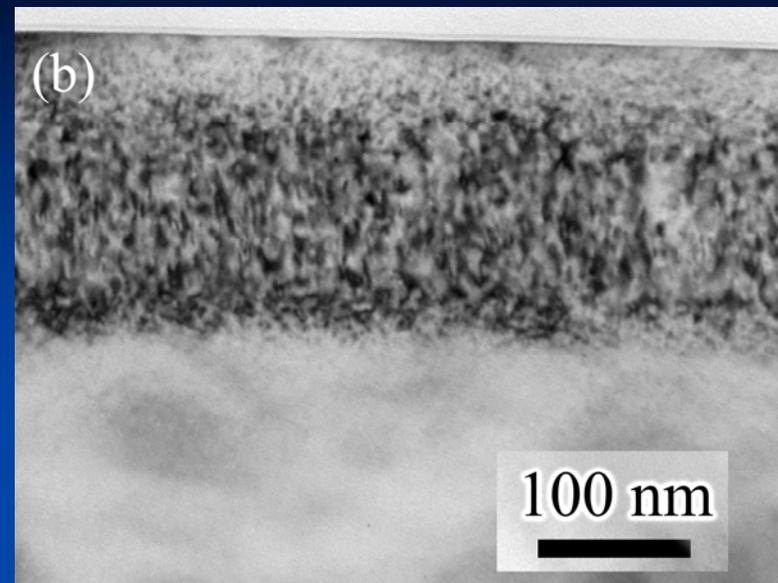
# Ion implantation of 6H-SiC

Only possible technique to locally dope SiC.  
 Radiation damage and annealing behavior was followed by RBS+channeling, Spectroscopic Ellipsometry and high resolution XTEM.

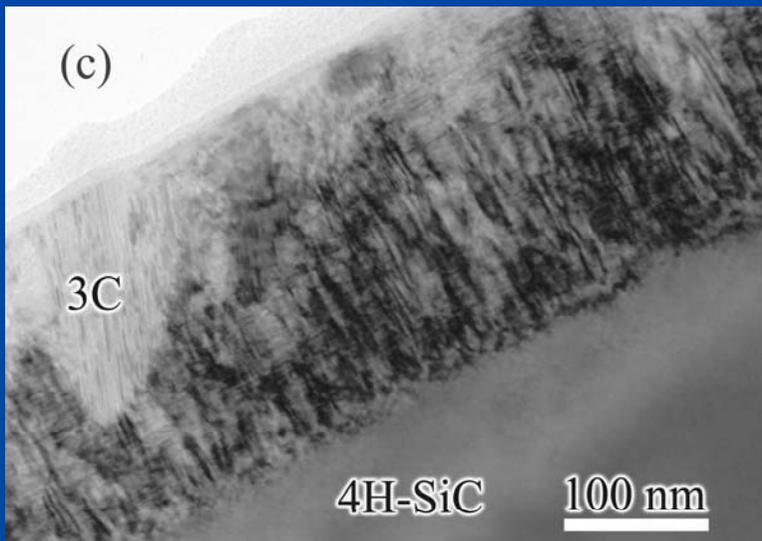




As implanted



Medium dose, annealed



High dose, annealed



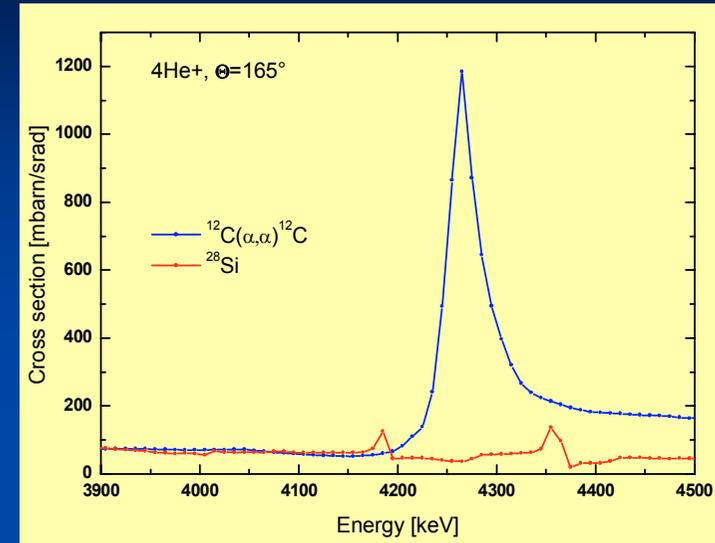
High dose, annealed



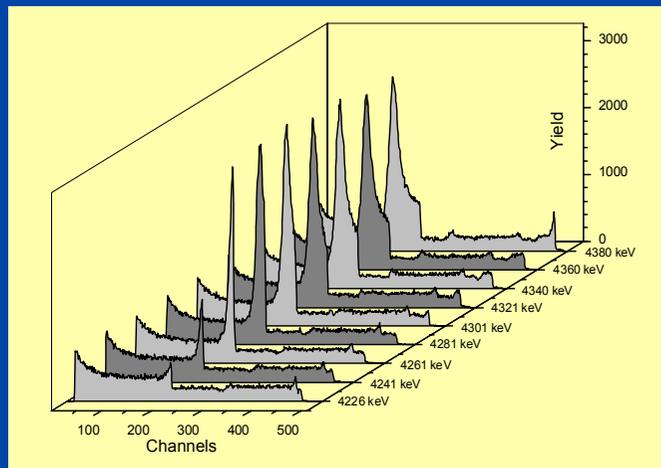
## Ion implantation of 6H-SiC

Resonance in the  $^{12}\text{C}(\alpha,\alpha)^{12}\text{C}$  cross section at 4.26 MeV. By varying the beam energy the C concentration can be mapped in a simple RBS experiment.

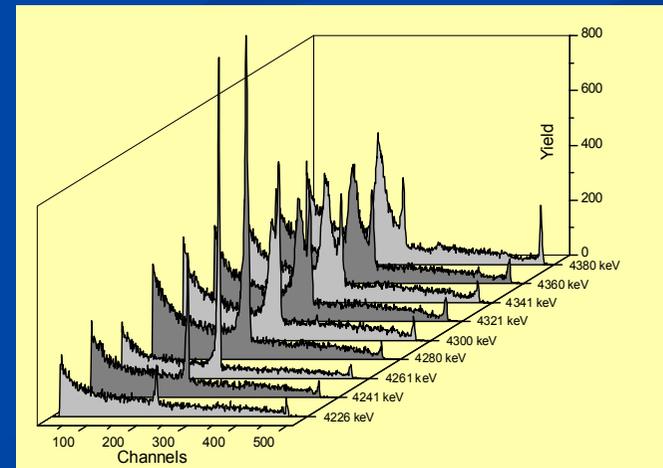
Study of ion implantation induced damage in 6H SiC crystal.

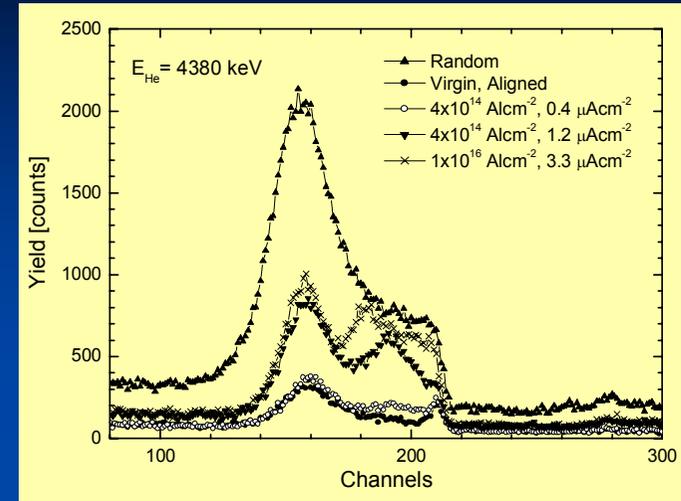
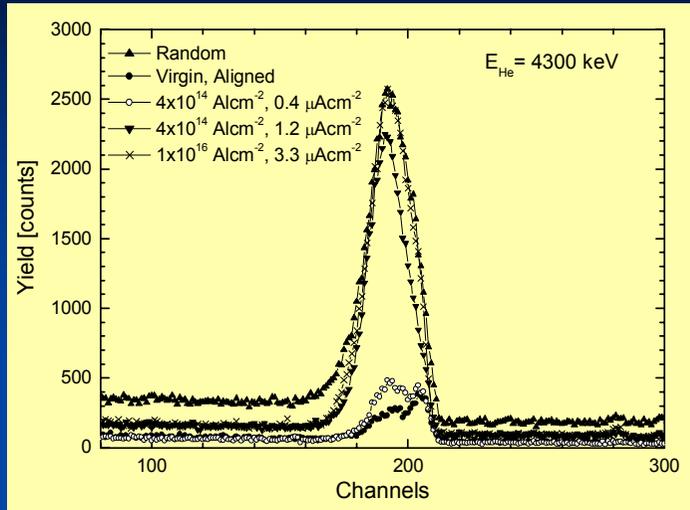


Random direction

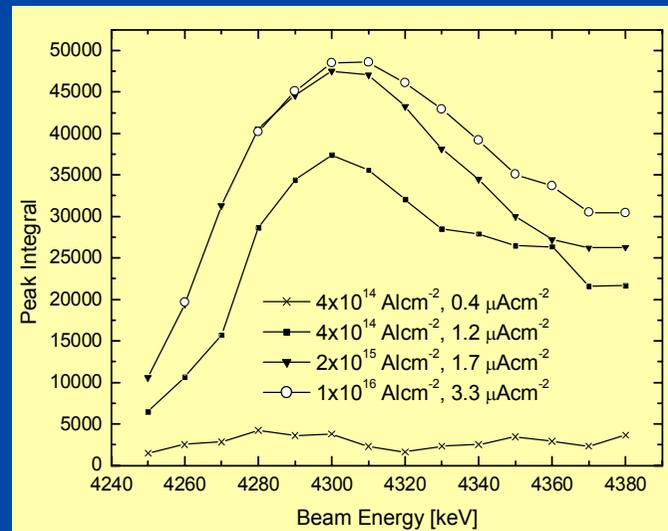


Aligned with C axis





RBS spectra recorded on Al implanted 6H SiC



Deduced damage profile



**Thank you for your attention!**

