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X-Ray Emission Techniques for Forensic Applications

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**PORTABLE PORTABLE XRF SPECTROMETERS
FOR IN-SITU MEASUREMENTSMETHODOLOGY
AND APPLICATIONS**

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**PORTABLE XRF SPECTROMETERS
FOR IN-SITU MEASUREMENTS-
METHODOLOGY AND APPLICATIONS**

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- Examples of FPXRF systems
- Parameters of in-situ measurements
- Interfering effects affecting accuracy of XRF analysis
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- Examples of applications
- Method validation
- Conclusions

INTRODUCTION

- Simplicity, speed of operation, flexible requirements for sample preparation
- Immediate analytical results (important for interactive measurement programmes, e.g., study of contaminated sites, remediation activities)
- Non-destructive analysis (e.g., museum samples, works of art, archaeological samples)
- Multi-elemental capabilities
- Good sensitivity
- Local and bulk analysis
- Portability

DESIGN CRITERIA

Excitation sources

Radioisotope	Half-live (years)	X- or γ -ray energy (keV)	Photons per disintegration
^{55}Fe	2.7	MnK x-rays (5.9, 6.5)	0.28
^{109}Cd	1.3	AgK x-rays (22, 25)	1.07
		88	0.04
^{241}Am	433	59.5	0.36

DESIGN CRITERIA

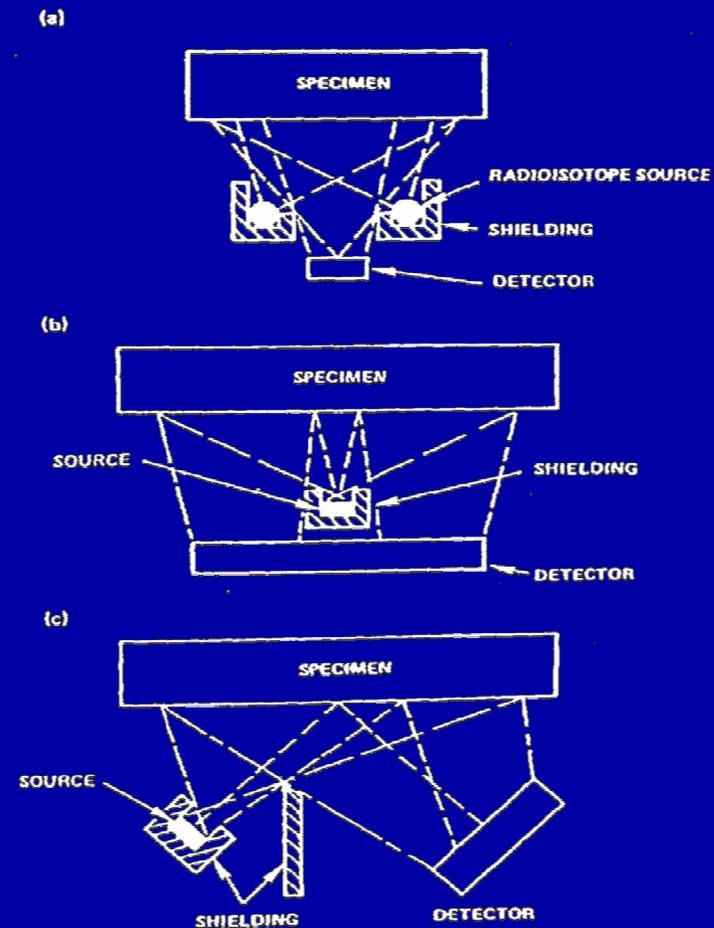
Excitation geometries

Three geometries for
radioisotope-excited X-ray
fluorescence analysis:

a) annular source

b) central source

c) side source



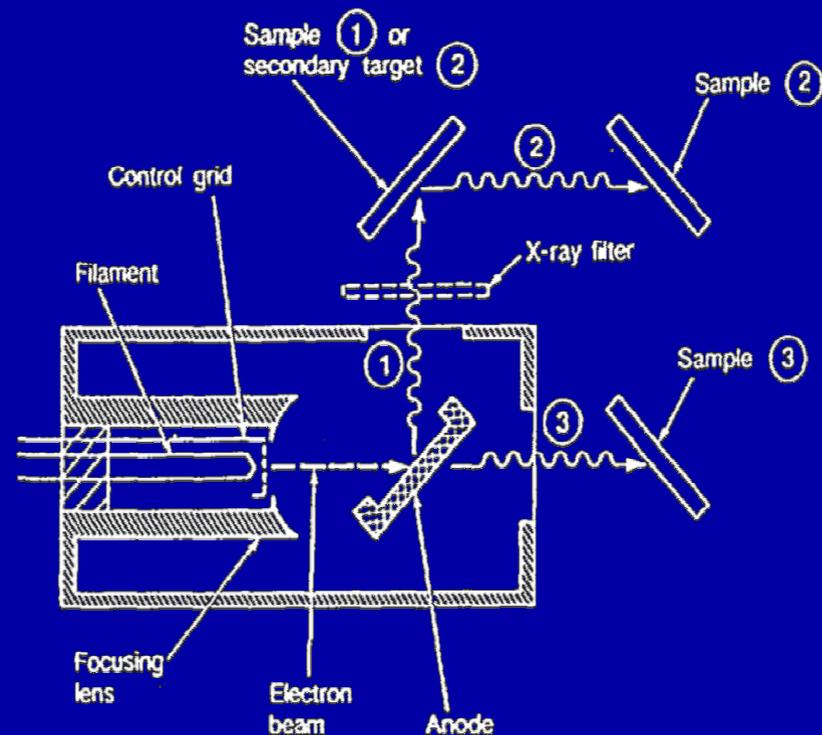
DESIGN CRITERIA

Excitation sources

Typical excitation geometries

for an X-ray tube:

- 1) direct excitation
- 2) secondary target excitation
- 3) transmission geometry



DESIGN CRITERIA

Detectors

Selection of detectors - parameters to be taken into account:

- ▣ question to be answered (single or multi X-ray lines)
- ▣ energy of X-rays to be detected
- ▣ required energy resolution
- ▣ required efficiency
- ▣ portability
- ▣ costs, not only of the detector but also of the related electronics and of the eventual cooling system

DESIGN CRITERIA

Detectors

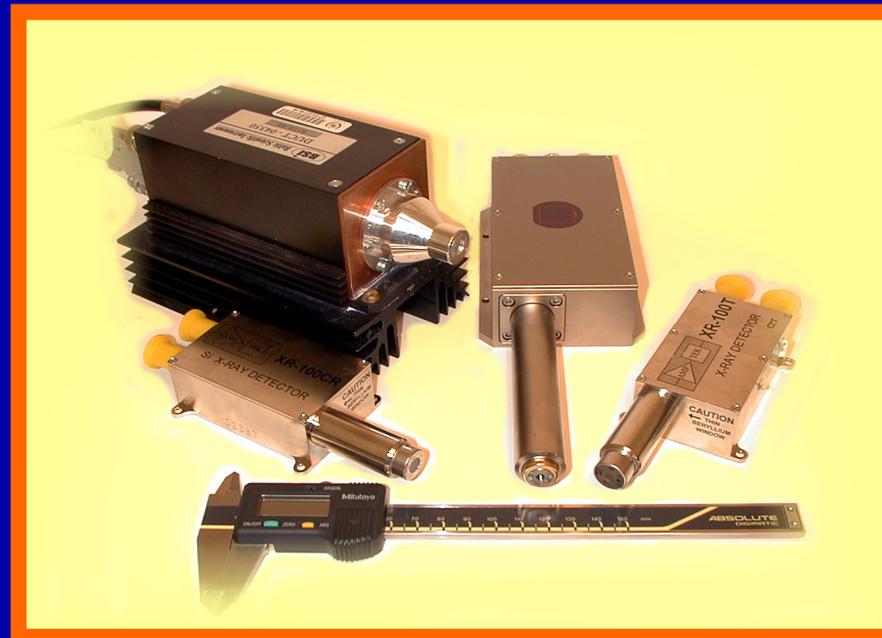
- Room temperature and Peltier-cooled semiconductor detectors:

HgI₂

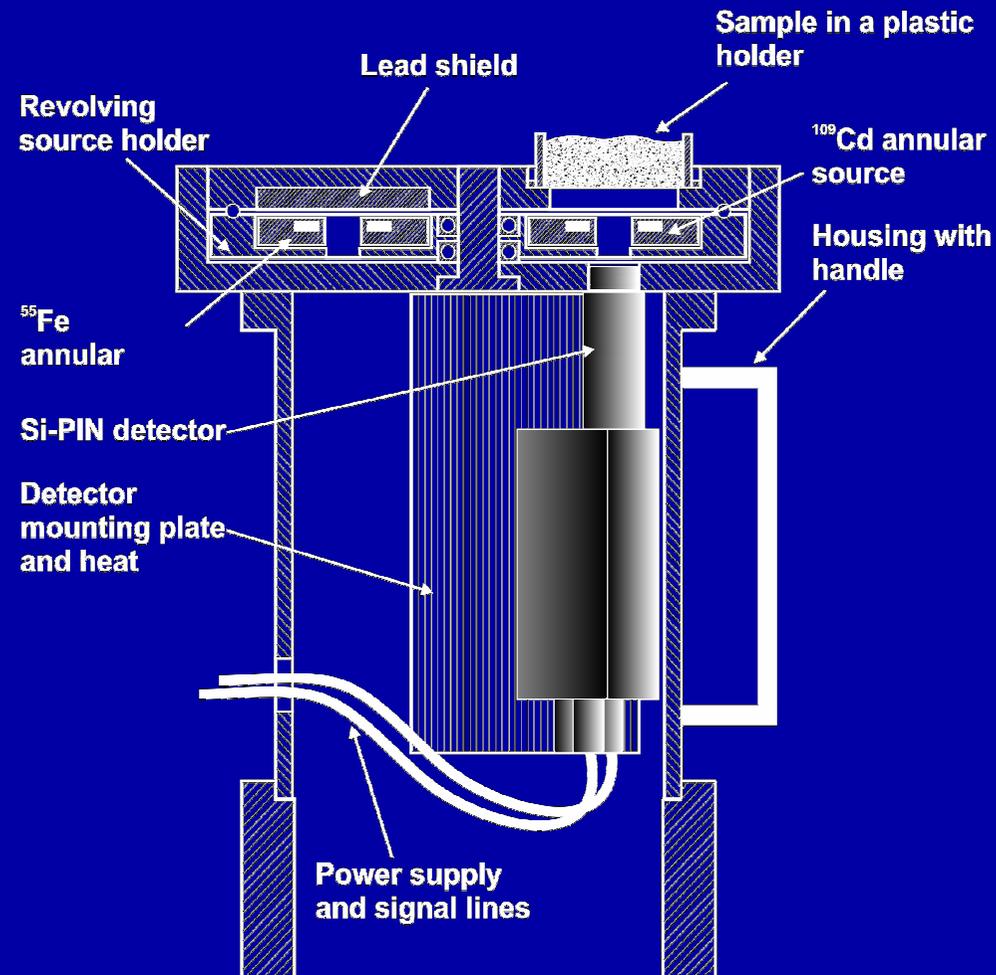
Si-PIN

Si-drift

CdTe



DESIGN CRITERIA Development



EXAMPLES of FPXRF SYSTEMS

- Portable XRF unit with thermoelectrically cooled Si-PIN detector

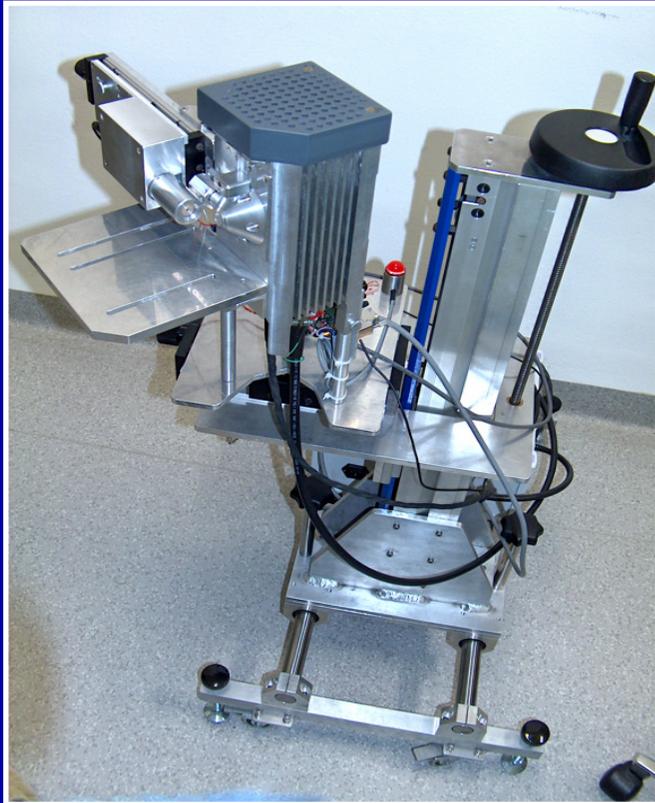


EXAMPLES of FPXRF SYSTEMS

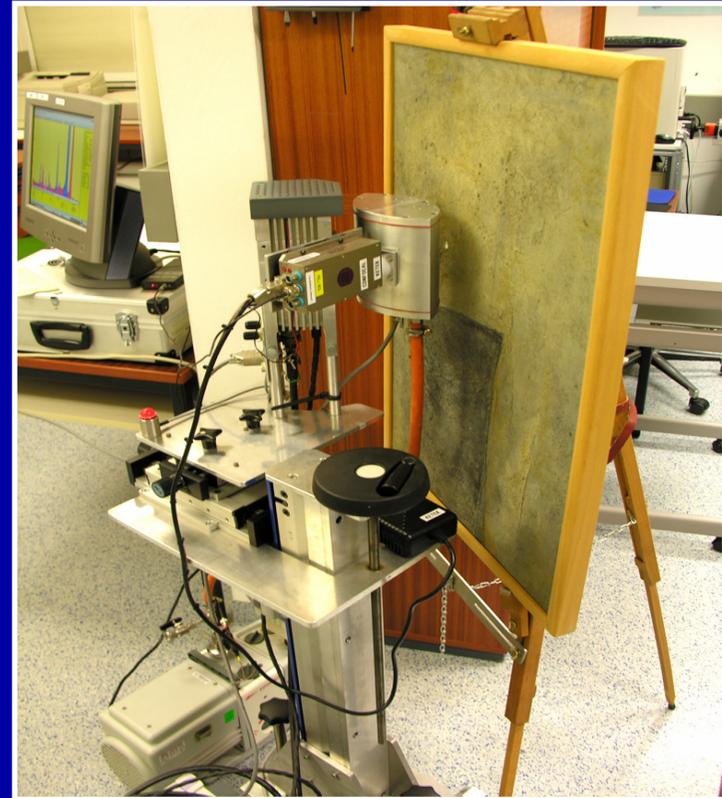
- Portable XRF unit with liquid nitrogen cooled Si(Li) (or HpGe) detector



EXAMPLES of FPXRF SYSTEMS

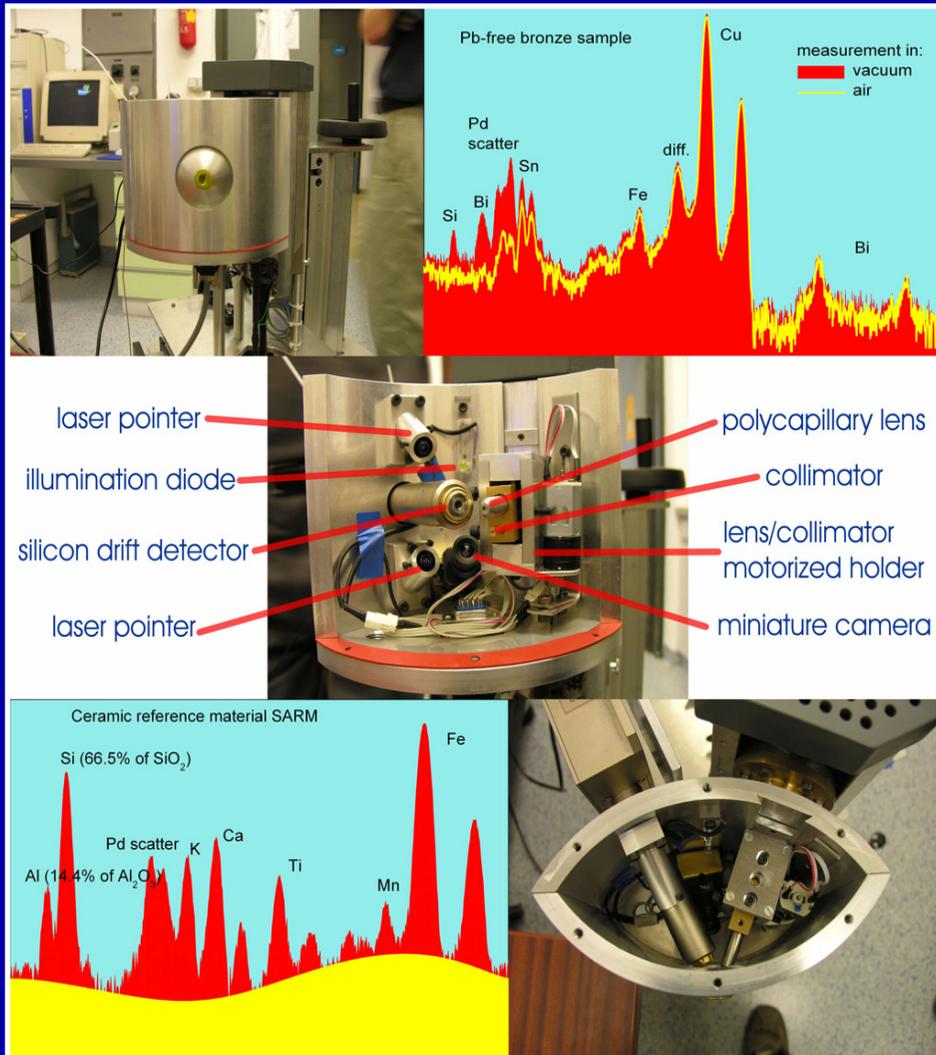


Direct/secondary excitation in air, Si-PIN detector, pinhole collimator



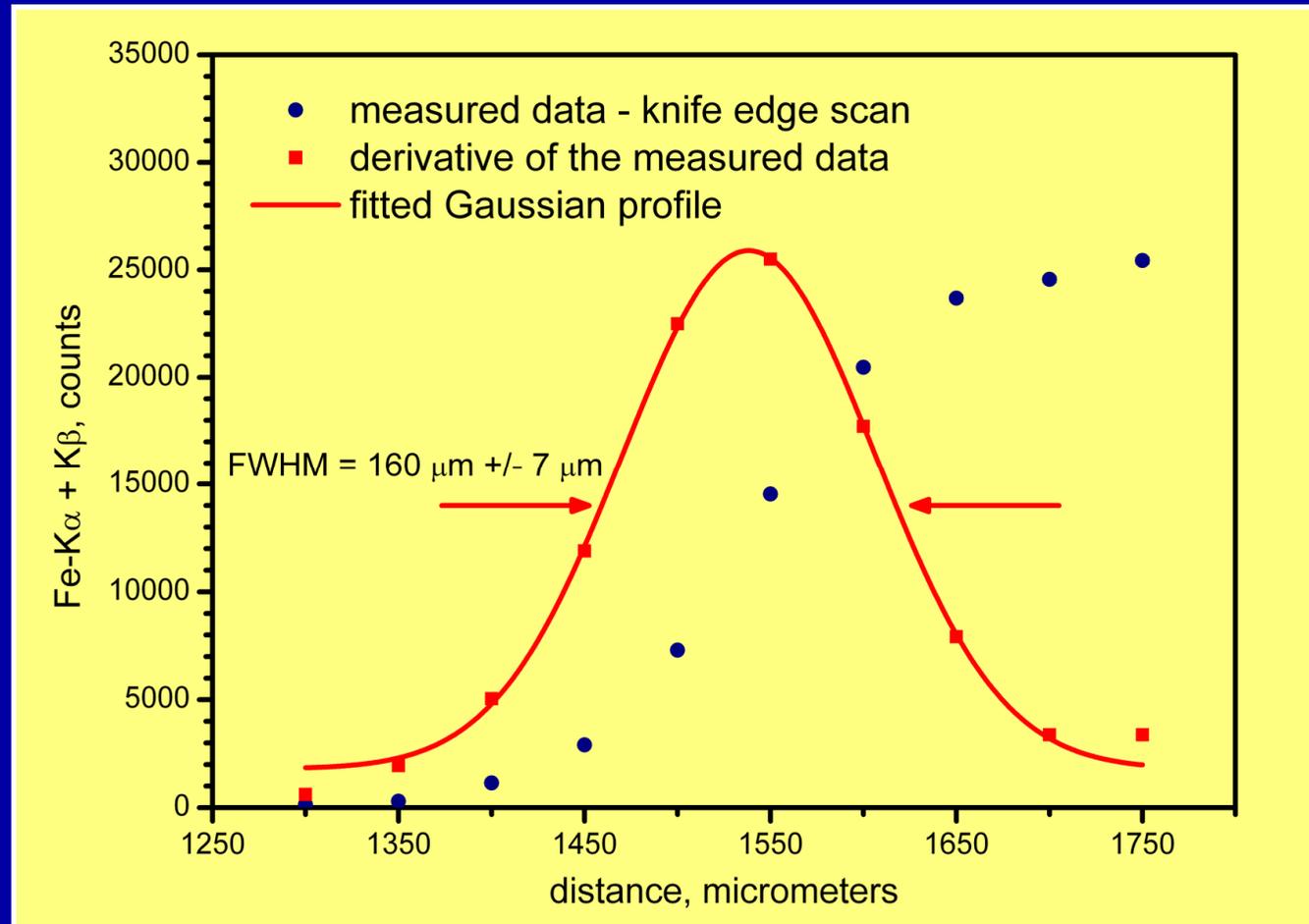
Direct excitation, vacuum attachment, SD detector, polycapillary optics/collimator

EXAMPLES of FPXRF SYSTEMS

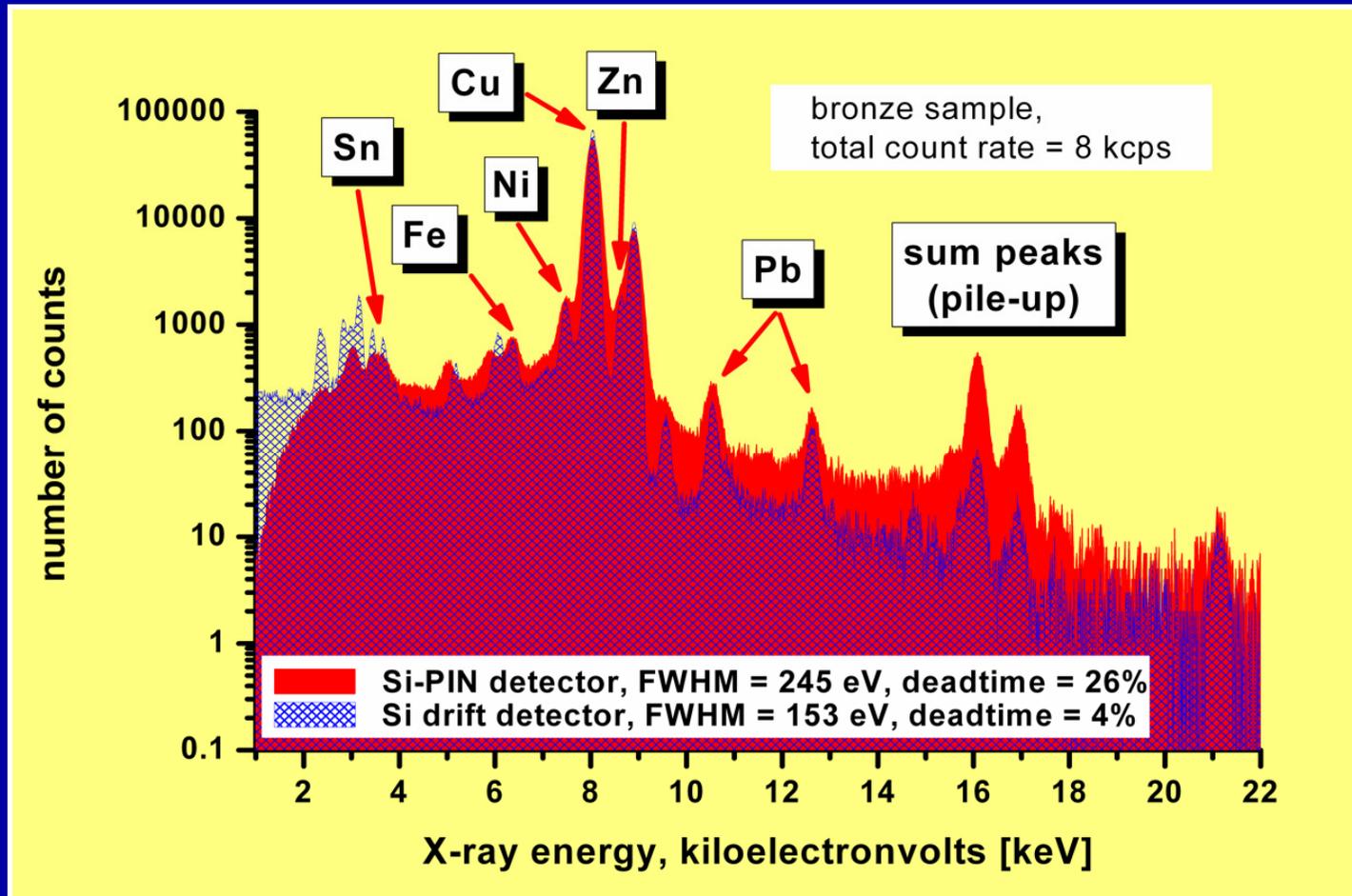


- Pd-anode X-ray tube (50W), direct excitation in vacuum or air
- polycapillary lens and brass collimator,
- silicon drift detector, thermoelectrically cooled, 10 mm² active area, 300 micrometers nominal thickness, 140 eV FWHM at 5.9 keV, 1 microsecond shaping,
- 2 laser pointers,
- CMOS camera with
- mechanical positioning system

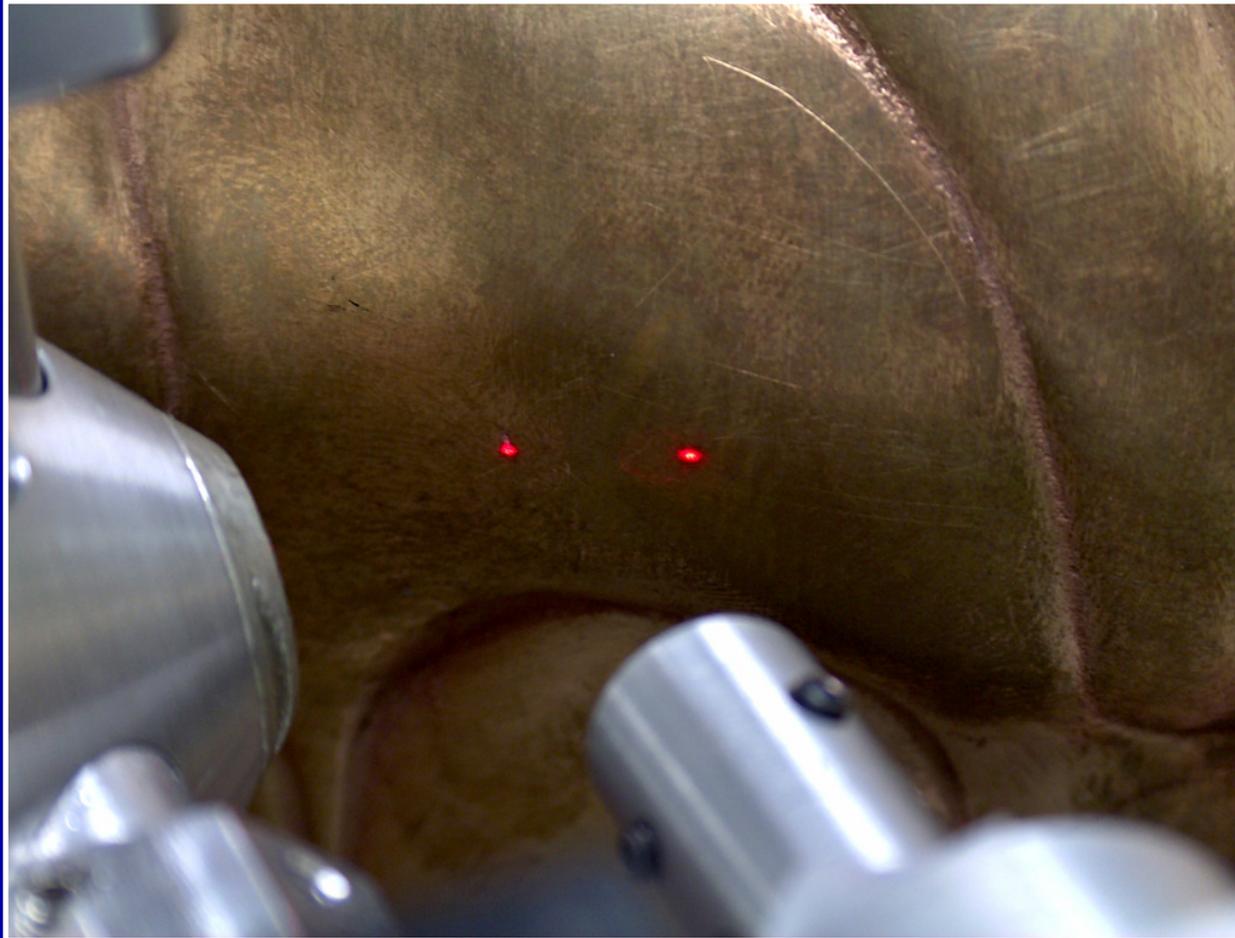
X-ray beam profile, polycapillary



X-ray detector performance

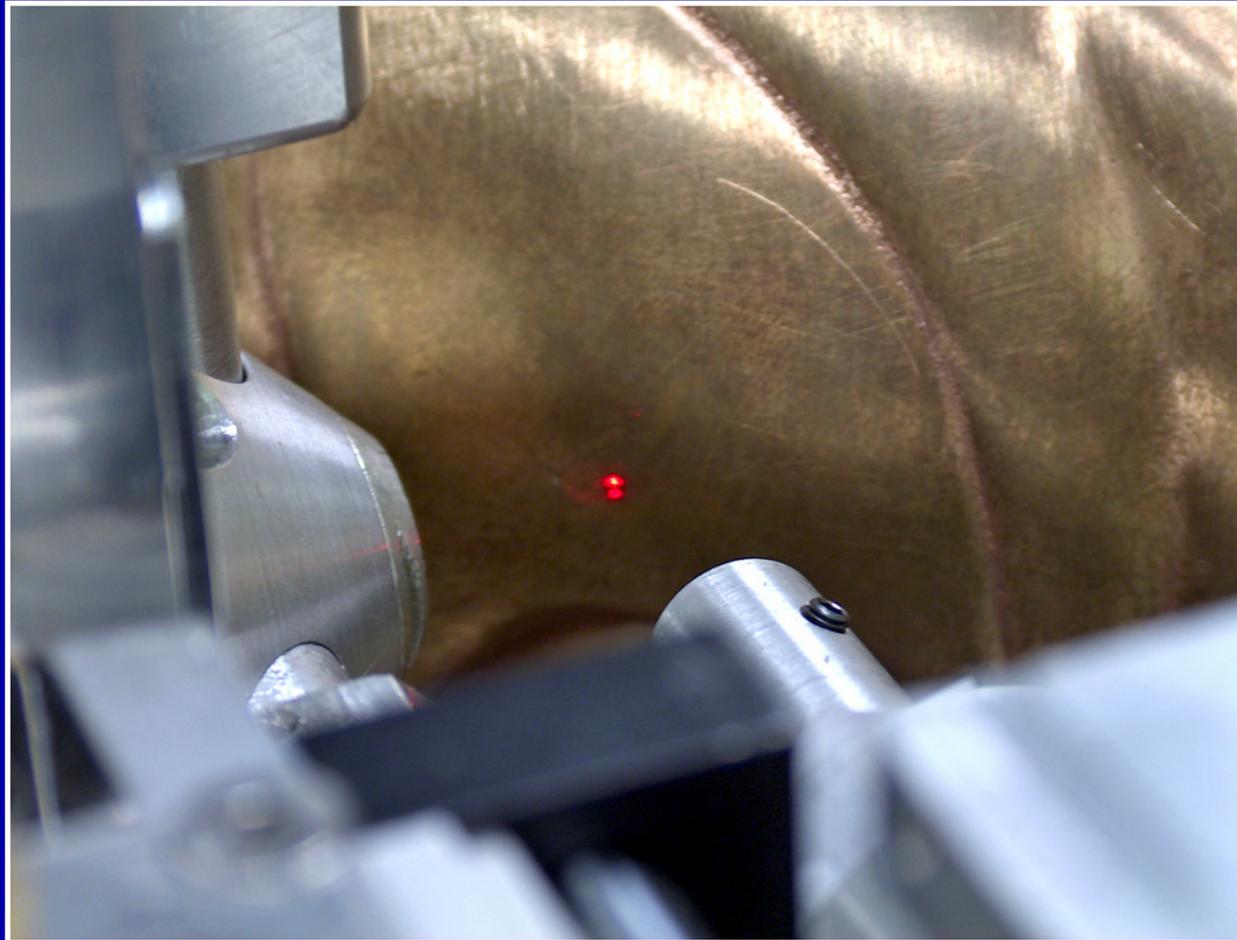


ADJUSTMENTS



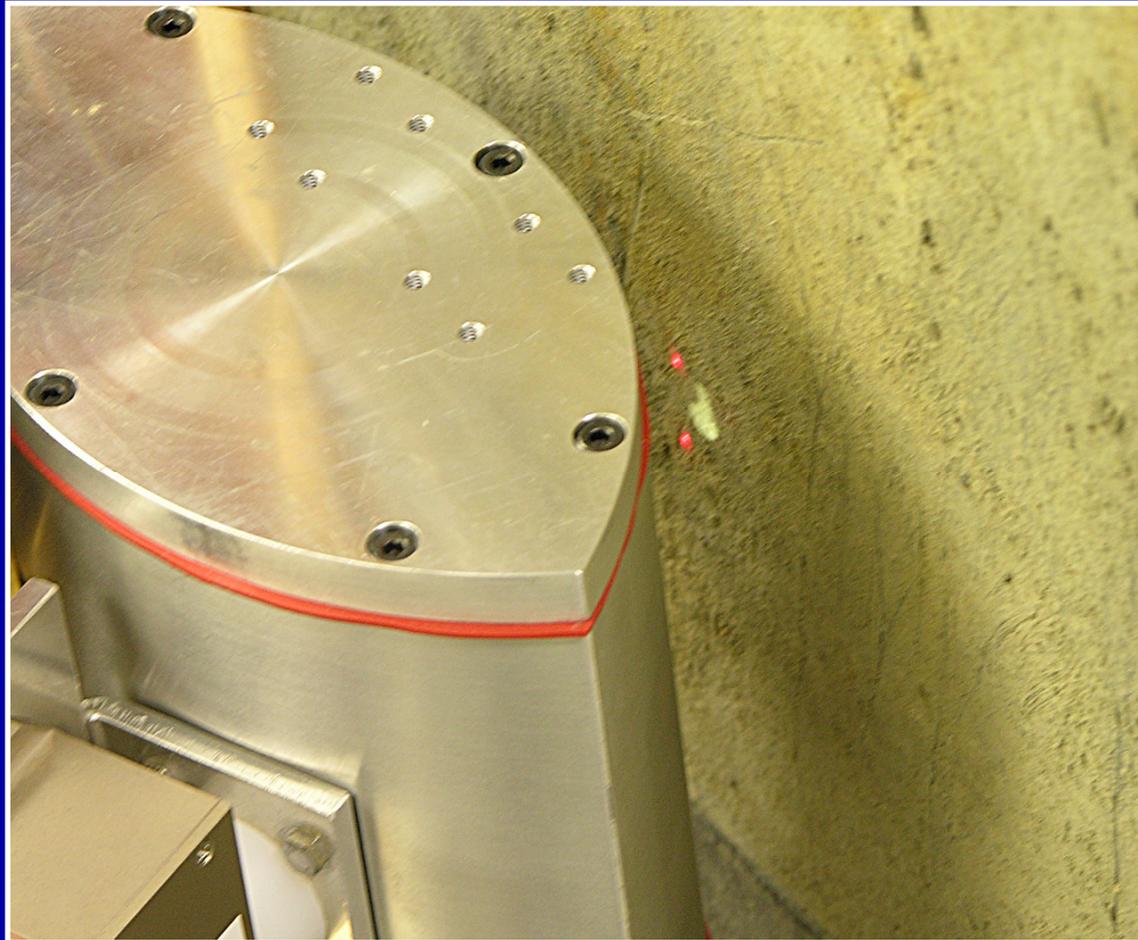
Object not in the optimum measuring position

ADJUSTMENTS

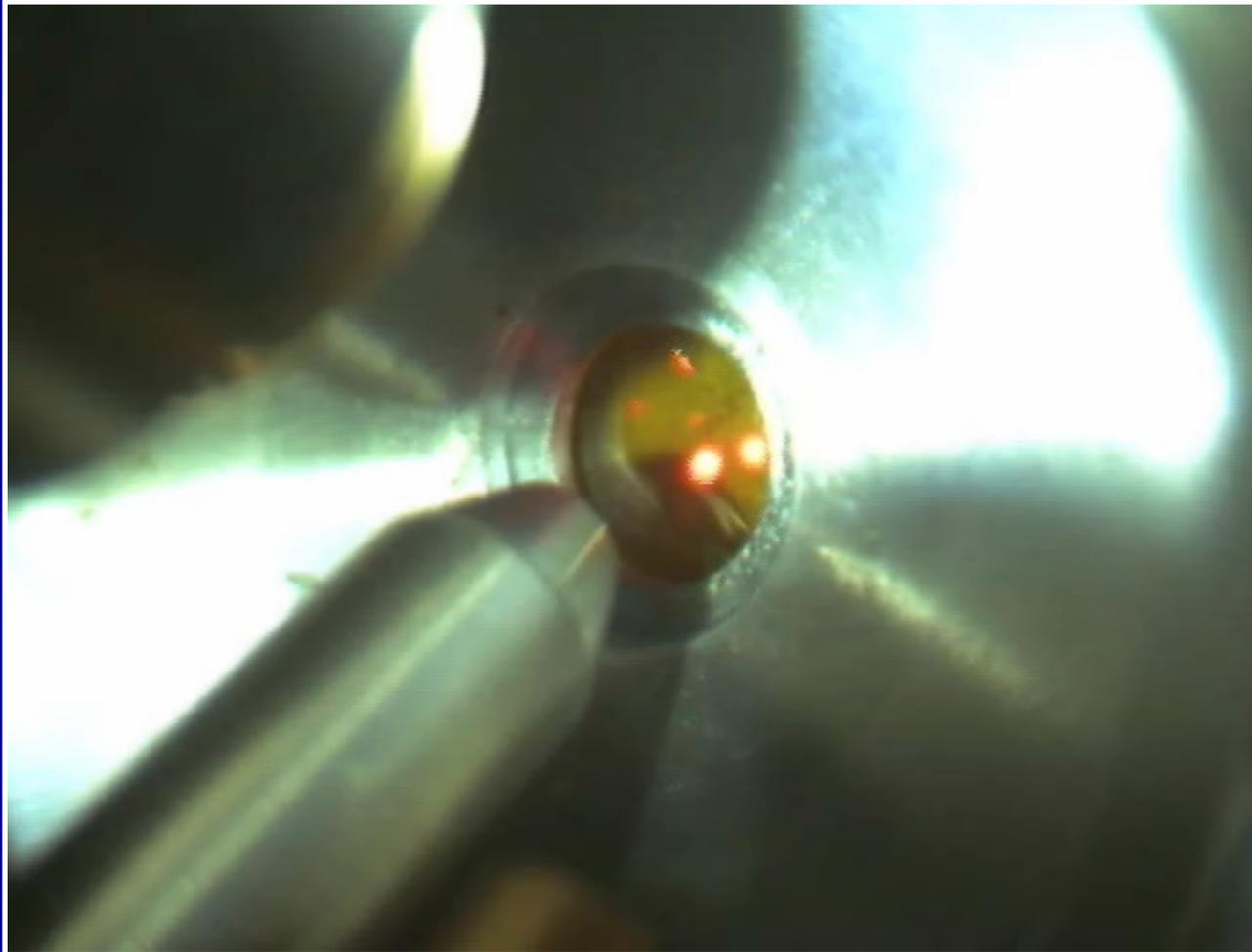


Object in the optimum measuring position

ADJUSTMENTS



ADJUSTMENTS



ANALYTICAL PERFORMANCE

- capable of detecting chemical elements with the atomic number ≥ 11 (sodium)
- detection limits in the PPM range
- beam spot diameter: 0.1 mm, 1 mm
- local and bulk analysis capability

PARAMETERS OF IN-SITU MEASUREMENTS

- Critical penetration depth (information depth)
“Sample thickness from which 99% of a fluorescence signal originates”

$$d_{\text{crit}} = m_{\text{thick}}/\rho = 4.61/(\rho\mu_{\text{tot}})$$

$$\mu_{\text{tot}} = \mu(E_0)\text{csc}\Psi_1 + \mu(E_i)\text{csc}\Psi_2$$

PARAMETERS OF IN-SITU MEASUREMENTS

Element	Information thickness, μm	
	$\bar{Z} = 12.5$	$\bar{Z} = 15$
P	12	79
Ca	50	32
Zn	540	340
Pb	930	570

PARAMETERS OF IN-SITU MEASUREMENTS

- 50 % of d_{crit} \Rightarrow 90% of fluorescence signal
- 35 % of d_{crit} \Rightarrow 80% of fluorescence signal
- 15 % of d_{crit} \Rightarrow 50% of fluorescence signal

Essential in the analysis of heterogeneous samples (mineralogy effects), samples of irregular surface or with surface contamination

PARAMETERS OF IN-SITU MEASUREMENTS

Accuracy

- Analysis of standard reference materials
- Analysis of confirmatory samples by using the reference (confirmatory) laboratory method considered as standard analytical method (minimum 10 % of total number of analyzed samples)

if r^2 between 0.7 and 0.9 – screening level

if $r^2 > 0.9$ PXRF and confirmatory data equivalent

PARAMETERS OF IN-SITU MEASUREMENTS

Detection Limit

Interference-free detection limit not fully adequate for *in-situ* applications

- A. Precision-based DL determined as 3 times st. dev. of the results for replicate analyses of low-concentration samples (5-10 x estimated DL):
 - (i) site specific calibration standards, (ii) appropriate SRMs, and (iii) clean sample matrix spiked with low concentration of target analytes
- B. Field or performance-based DL – analysis of low concentration outliers on data cross plots for log transformed PXRF results vs. log transformed confirmatory results – DL where the linearity disintegrates

PARAMETERS OF IN-SITU MEASUREMENTS

- Blank samples

1. *Instrument blank*

- to verify that no contamination exists in the spectrometer or on the probe window, e.g., Teflon block, quartz block, clean sand, lithium carbonate
- daily analysis before and after measurements

OK if no concentrations > DL

PARAMETERS OF IN-SITU MEASUREMENTS

2. *Method blank*

- to monitor for laboratory-induced (sample preparation) contaminants or interferences e.g., clean silica or lithium carbonate that undergoes the same sample preparation procedure as analyzed samples
- (at least) daily analysis

OK if a method blank doesn't contain any analyte at a concentration above DL

PARAMETERS OF IN-SITU MEASUREMENTS

Total uncertainty

$$\sigma^2_{\text{tot}} = \sigma^2_{\text{sample representation}} + \sigma^2_{\text{sample collection}} + \sigma^2_{\text{sample handling}} + \sigma^2_{\text{sample preparation}} + \sigma^2_{\text{analysis}}$$

Interfering effects affecting accuracy of XRF analysis

- Physical matrix effects

- A. Particle size effects

e.g., Cr in soil: 1 cm² irradiated area,
penetration depth 0.2 mm, DL=200 mg/kg



8.4 μg Cr in the information layer
(single grain of diameter of 135 μm)

Interfering effects affecting accuracy of XRF analysis

Solution: grinding and sieving to a uniform and small particle size fraction

B. Heterogeneity of materials

(preliminary and simple procedure to homogenize the material is recommended – important during validation of XRF results through confirmatory analysis)

Interfering effects affecting accuracy of XRF analysis

C. Surface irregularity effects

- detected X-ray fluorescence intensities are systematically lower than those observed from flat samples
- theoretical model to calculate unevenness factor (source-sample distance, surface peak-valley amplitude, frequency number of convex and concave surfaces)

Interfering effects affecting accuracy of XRF analysis

- D. Mineralogy effects (dependence on mineral assemblage – size, distribution and position of minerals - present in the excited volume (information volume):
- grain size (vs. volume from which the fluorescence signal originates)
 - excitation-detection efficiency

Interfering effects affecting accuracy of XRF analysis

E. Chemical matrix effects

- Absorption
- Enhancement

F. Spectral interferences

- insufficient energy resolution of detector
- overlap of characteristic X-ray lines of two or more elements (first identify all possible spectral interferences and then apply spectrum evaluation software)

Correction procedures for the interfering effects

- Physical matrix effects

- A. Surface irregularity

- 1. Use of Compton and Rayleigh scattered primary radiation*

$$I_{\text{corrected}} = I_{\text{measured}} \times B_{\text{reference}} / B_{\text{measured}}$$

(for silicate rocks effective correction up to about 3 mm only)

Correction procedures for the interfering effects

2. *Use of a special calibration procedure based on relative instrumental calibration factors*

$$R_{ik} = I_i / I_k$$

i – analyte; k – reference element

Correction procedures for the interfering effects

■ B. Mineralogy effects

- From experiments for representative (flat) samples the rel. st.dev. of the average concentration (R) is calculated

$$R = 100(\%) (s/n^{0.5}) / C_{\text{mean}}$$



$$n = [(100 \times s) / (R \times C_{\text{mean}})]^2$$

Correction procedures for the interfering effects

- Chemical matrix effects

A. Methods based on scattered primary radiation
(reduces also surface irregularity and moisture effects)

I_i / I_{Com} vs. concentration

for synthetic calibration samples or site-specific calibration samples analysed by reference analytical method

Correction procedures for the interfering effects

- $I_i / (I_{Com})^S$ vs. concentration

(L.G.Livingstone, X-Ray Spectrometry, 1982, 11, 89)

- For two samples with different matrices a and b, and the same concentration of the analyte, the optimum S from:

$$(I_{i,b} / I_{i,a}) / (I_{sct,b} / I_{sct,a})^S = 1$$

Correction procedures for the interfering effects

- B. Backscatter Fundamental Parameter Method

Q_i , Q_A , and Q_B (element mass per unit area) are unknown:

$$I_i = Q_i K_i(E_0) \frac{1 - \exp\left\{-\left[\sum_j F_j + F_A + F_B\right]\right\}}{\sum_j F_j + F_A + F_B} (1 + F_i^{Enh}) S_i;$$
$$F_j = Q_j \left(\frac{\mu_j(E_0)}{\sin \Psi_1} + \frac{\mu_j(E_i)}{\sin \Psi_2} \right)$$
$$F_{A/B} = Q_{A/B} \left(\frac{\mu_{A/B}(E_0)}{\sin \Psi_1} + \frac{\mu_{A/B}(E_i)}{\sin \Psi_2} \right)$$

EXAMPLES OF APPLICATIONS

(Inorganic pigments)

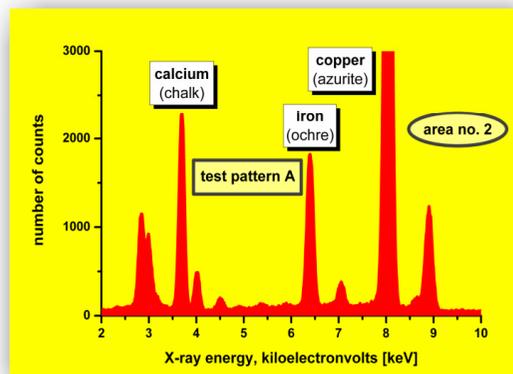
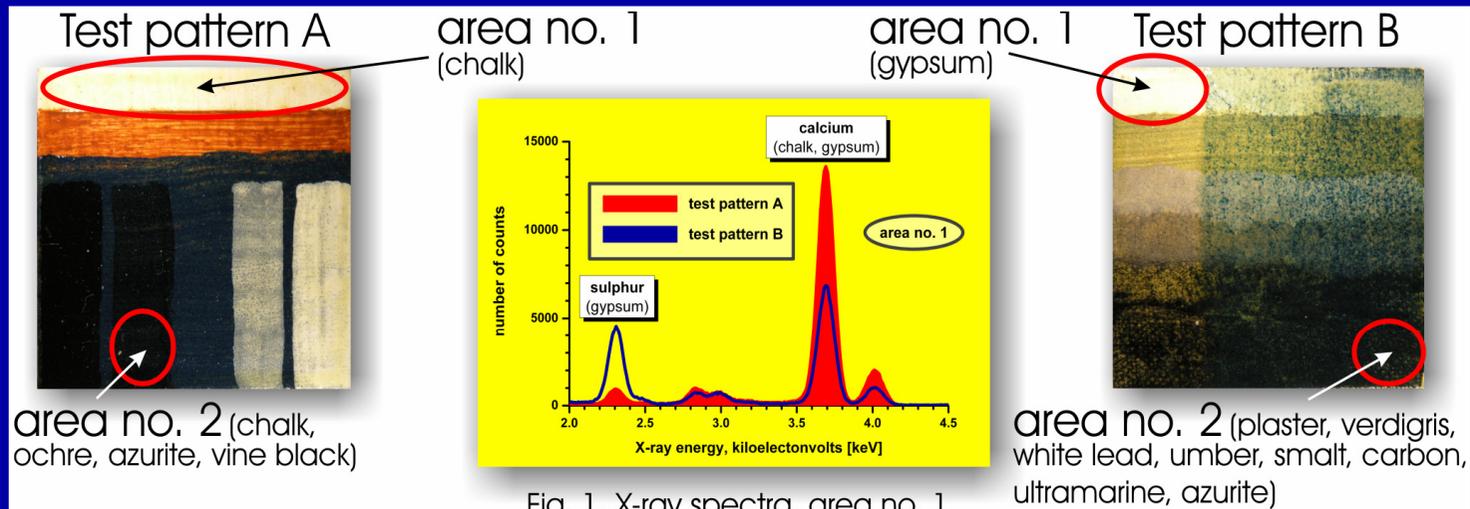


Fig. 2. X-ray spectrum, pattern A, area no. 2

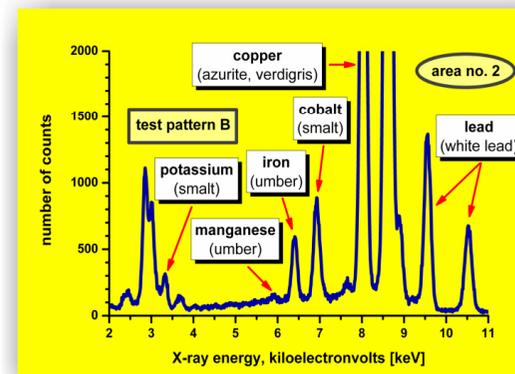


Fig. 3. X-ray spectrum, pattern B, area no. 2

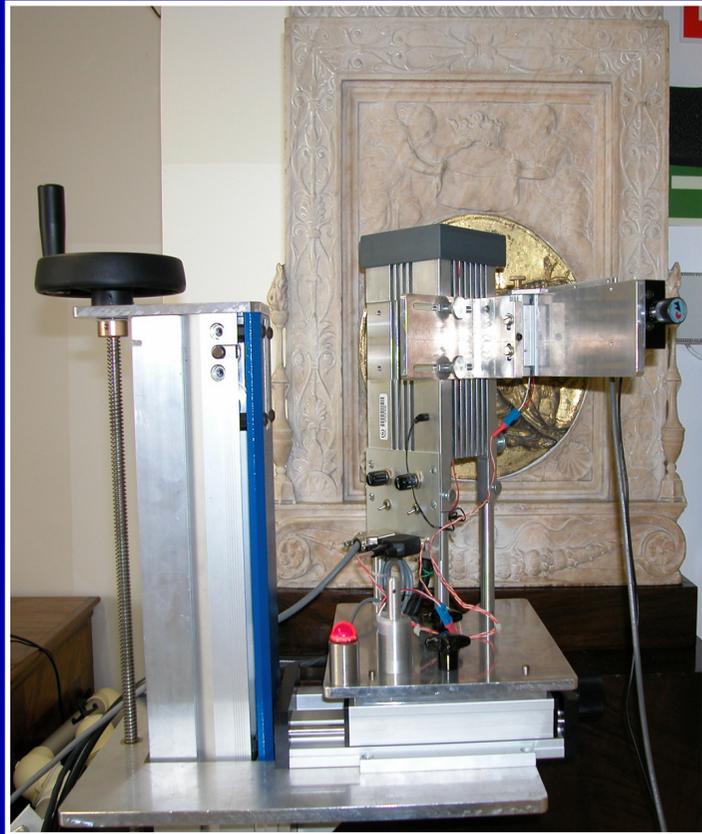
EXAMPLES OF APPLICATIONS

(Investigation of archeological objects from Kunsthistorisches Museum, Vienna)

- Egyptian and Near Eastern Collection: coal pot, small box with lid (galena residue), wooden stele (pigment identification), tweezers (alloy analysis), armlets (alloy analysis)
- Painting gallery: identification of pigments, priming coatings
- Arms and Armour Collection: oriental saddle identification of pigments and metal applications
- Collection of Sculpture and Decorative Arts: bronze tondo, small can, Venice glass (glass composition)
- Museum of Entomology: Benin bronzes (alloy composition)

EXAMPLES OF APPLICATIONS

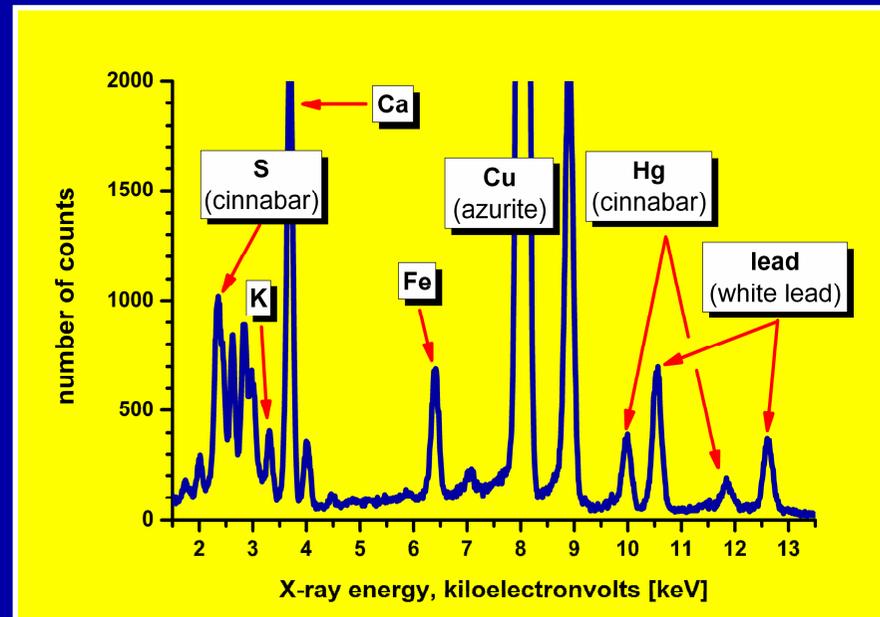
(Analysis of alloys – bronze tondo)



“Virgin and Child” by Donatello (Florence 1386 - 1466) - a bronze tondo from Kunsthistorisches Museum’s (Vienna) collection.
Average composition of the bronze was used for documentation purposes.

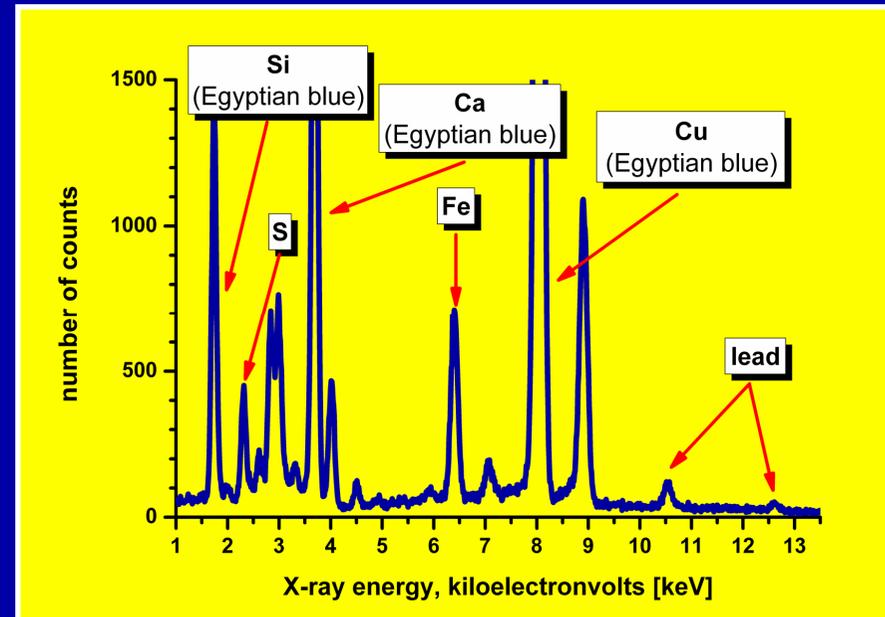
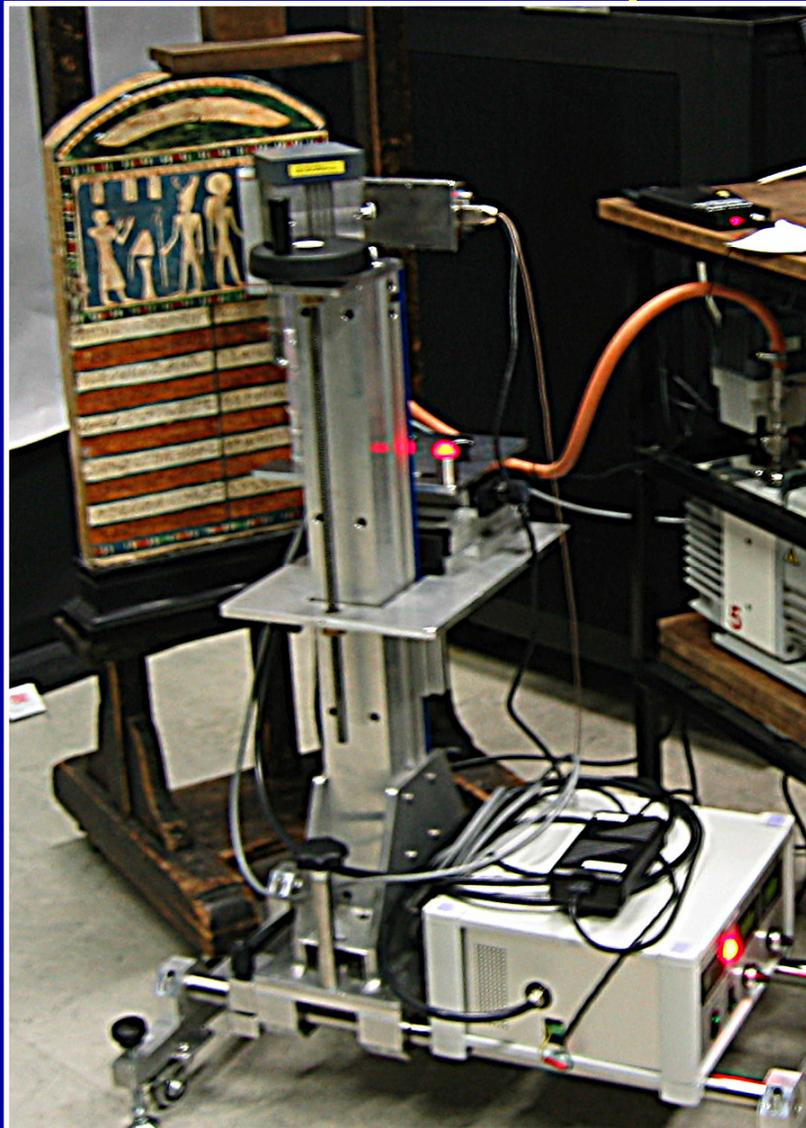
EXAMPLES OF APPLICATIONS

(Oriental saddle)



EXAMPLES OF APPLICATIONS

(Wooden stele)



Identification of Egyptian blue ($\text{CaO}\cdot\text{CuO}\cdot 4\text{SiO}$) pigment in a wooden stele from XXVI Dynasty, Thebes (640 B.C.).

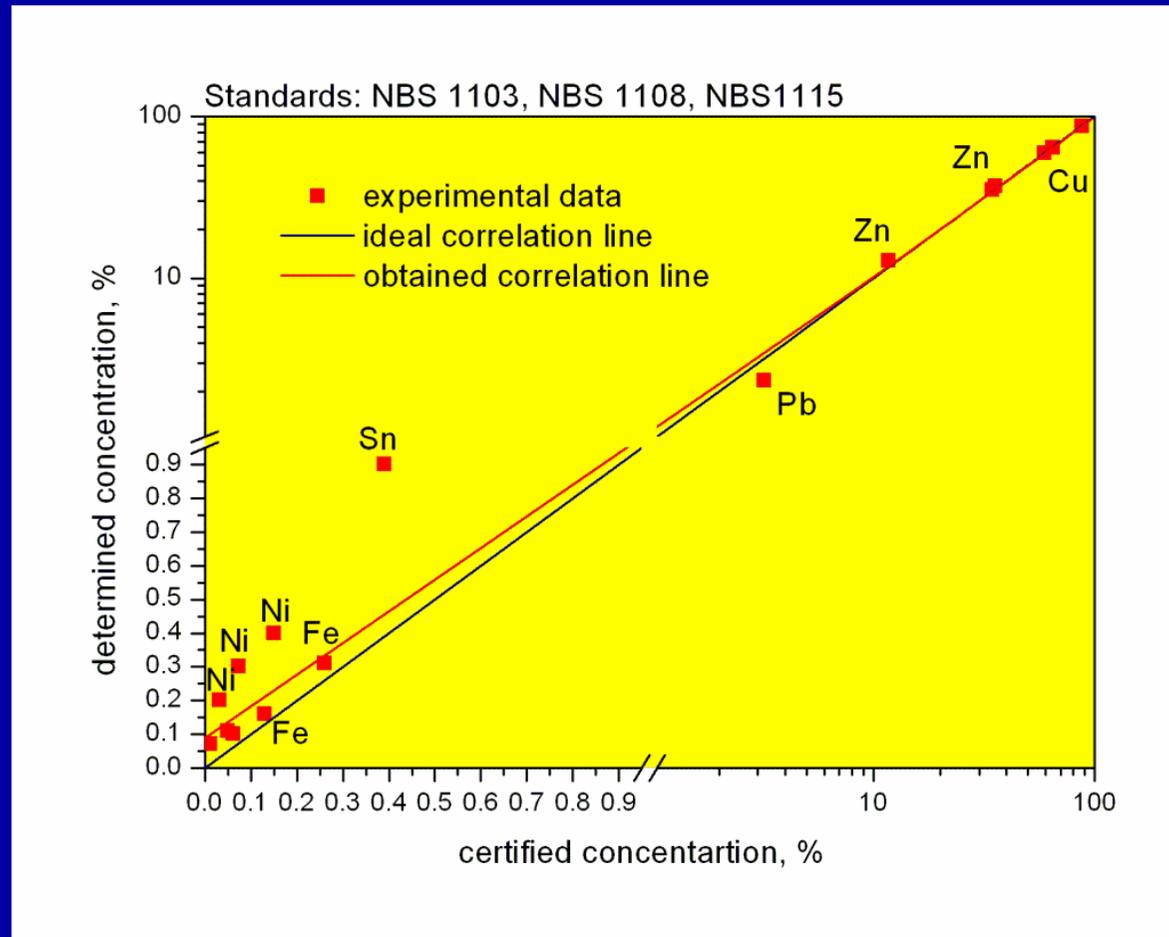
ANALYSIS of SOIL

Concentrations of elements in soil obtained from direct in-situ measurements of the soil surface layer and using the powder soil samples after homogenization in the laboratory. The analyses were performed with portable X-ray spectrometer system utilizing the BFP method. The in-situ concentrations were corrected for the moisture contents.

- a) - less than the detection limit (DL)
 b) – the uncertainties due to counting statistics/range of results

Element	Concentrations of elements determined in two locations, direct in-situ measurements		Concentrations of elements determined in two locations, in-laboratory measurements after processing the soil material	
	Location # 1	Location # 2	Location # 1	Location # 2
mg/g				
K	< DL ^{a)}	< DL	12 ± 2	12 ± 2
Ca	37 ± 5 ^{b)}	47 ± 5	65 ± 2	66 ± 3
Fe	16.1 ± 2	15.5 ± 0.5	22 ± 2	19.8 ± 0.2
µg/g				
Mn	< DL	< DL	400 ± 60	520 ± 40
Zn	54 ± 18	75 ± 30	67 ± 8	68 ± 8
Br	< DL	< DL	4.4 ± 2	4.7 ± 2
Rb	57 ± 10	53 ± 6	84 ± 7	81 ± 3
Sr	72 ± 8	80 ± 10	119 ± 4	129 ± 3
Y	< DL	< DL	23 ± 2	19 ± 3
Zr	80 ± 10	83 ± 6	640 ± 410	490 ± 130
Pb	31 ± 12	44 ± 10	45 ± 5	45 ± 7

Method Validation



Determined versus certified concentration of elements in bronze standards

Conclusions

- *In situ* X-ray fluorescence analysis of works of art and objects of cultural heritage provides very valuable information about inorganic composition of the objects in a non-destructive manner
- Analytical performance of the portable energy dispersive XRF spectrometers is very similar to that of a laboratory XRF system.
- *In situ* measurements require max care and good knowledge of XRF analysis to ensure a proper interpretation of the analytical data
- The application of external vacuum chamber (or helium flash) is essential in the analysis of samples with low-Z elements

Acknowledgment

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Thank you

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