



**The Abdus Salam  
International Centre for Theoretical Physics**



**2272-4**

**Joint ICTP-IAEA School on Synchrotron Applications in Cultural Heritage and  
Environmental Sciences and Multidisciplinary Aspects of Imaging Techniques**

*21 - 25 November 2011*

**Radiation for Cultural Heritage**

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*IAEA, Vienna  
Austria*

# Radiation for Cultural Heritage

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# Outline

- Scientific methods: Aims of investigations, Probes and techniques
- Comparison of different radiation probes
- X-rays: Portable spectrometers, Laboratory-SR facility
- Charged particle beams
- Neutron beams
- Imaging with UV-Vis-IR radiation
- Laser based techniques
- Dating TL-OSL

# Scientific methods in Art and Archaeology

## Objectives of the scientific examination

- Identification of chemical / biological constituents
- Structural characterization
- Provenance and dating. Authentication
- Manufacture Technology (metallurgy, pottery, coloring)
- Dietary habits (Human, Animal remains)
- Environmental degradation. Conservation

# Scientific methods in CH: Challenges-Requirements

- **Diversity** of materials nature: - Organic, inorganic, biological materials
- **Elemental /Molecular** analysis /**Structural** information
- Analysis at different scales from **sub- m** particles to **cm-** size size samples
- **Quantitative/semi-Q** analysis/ **Qualitative** information
- Material **interactions** for environmental impact
- Optimum analytical **range-sensitivity**
- **In-situ analysis**
- **Non-destructive or even non -invasive analysis!**

# Radiation Probes in Art and Archaeology

## □ X-rays

- X-ray Tubes –unpolarized/polychromatic

- Synchrotron radiation polarized/tunable monoenergetic

Depth: sub-micron to mm, spot size cm to few tens of nm

## □ UV, Vis-IR radiation (conventional, SR- sources)

## □ Laser Induced Techniques (LIF, Raman, LIBS)

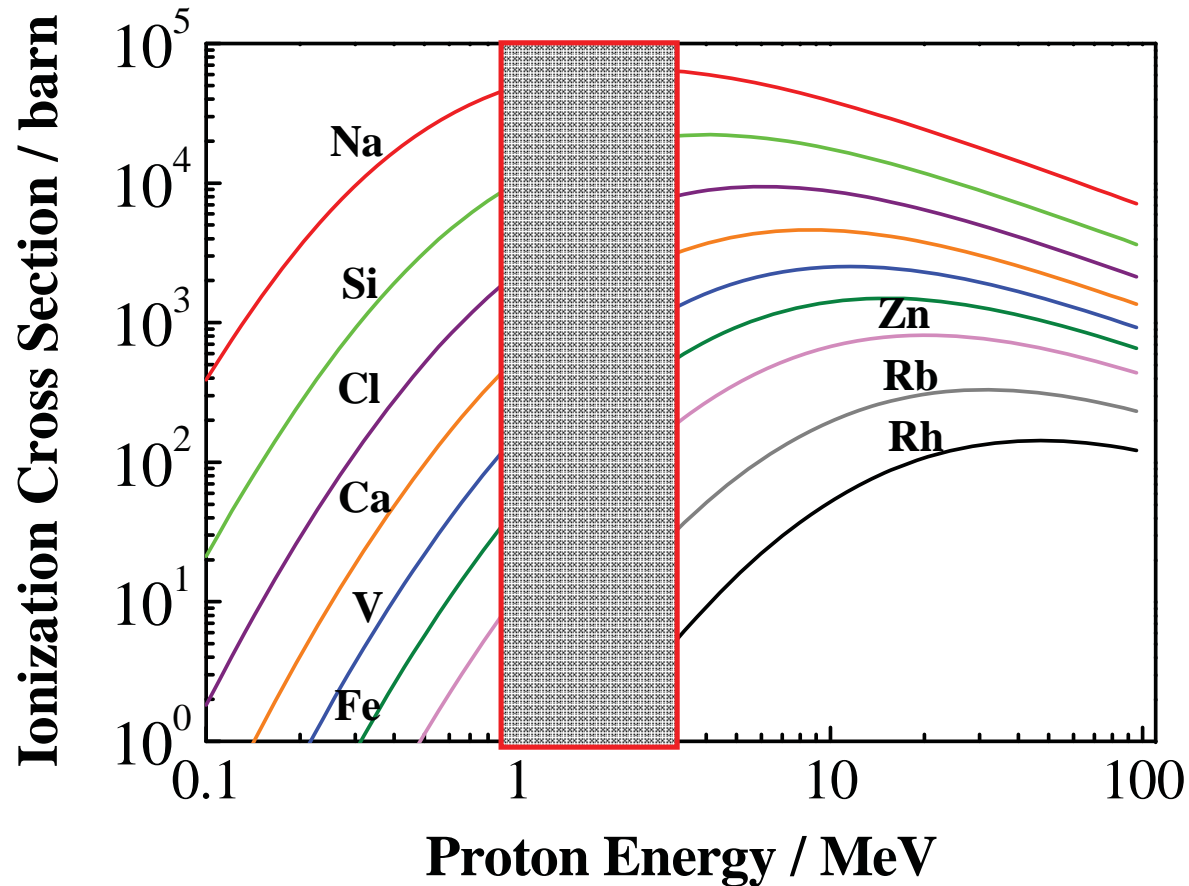
## □ Neutrons : Thermalized/ Fast neutrons ( cm- scale)

## □ Charged particle beams

Depth: sub-micron to mm, spot size mm to micron level

# Comparison of radiation Probes: Protons (1)

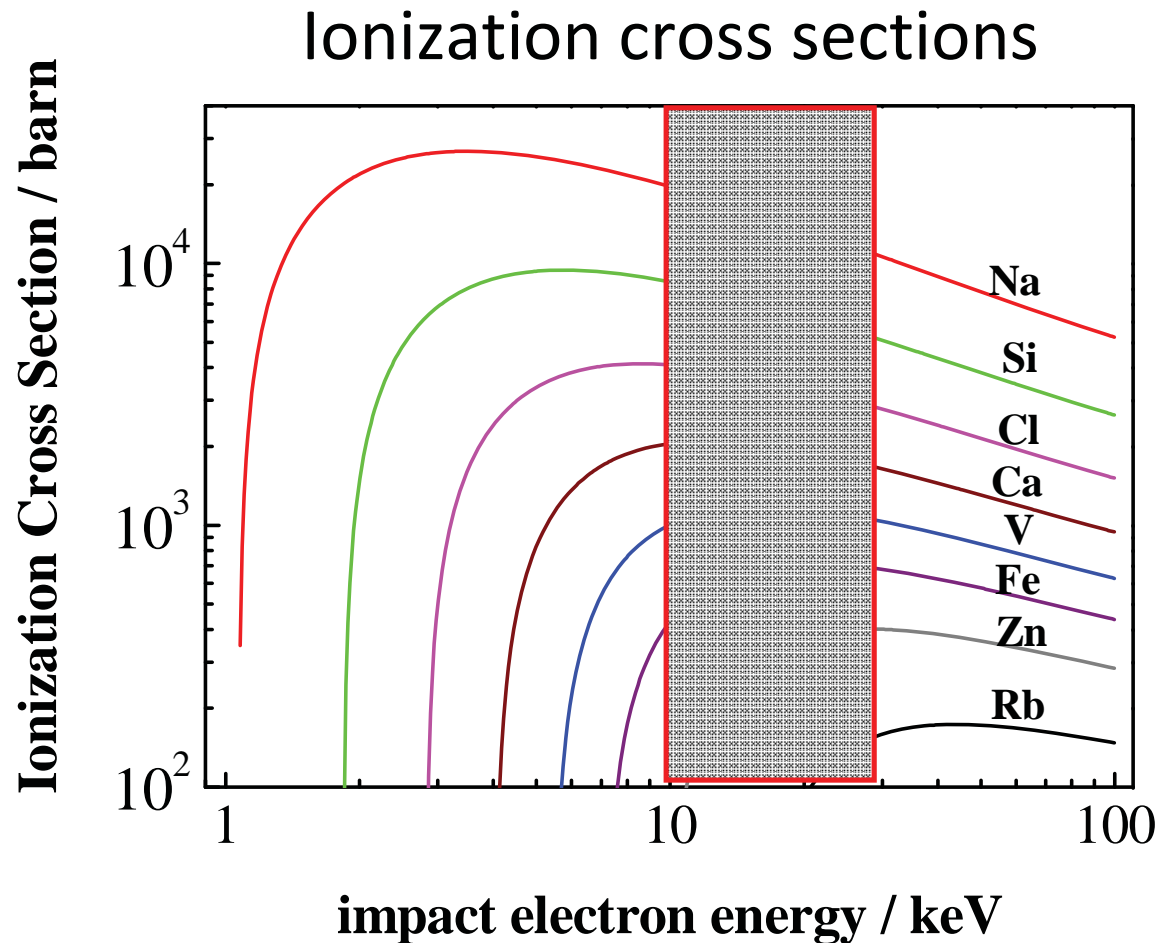
Ionization cross sections



**Protons**  
 **$10$ - $10^5$  barns**

Protons ICS  
through ECPSSR  
theory  
W. Brandt, G.  
Lapicki, *Phys. Rev.*  
A 23 (1981) 1717

# Comparison of radiation Probes: Electrons (2)

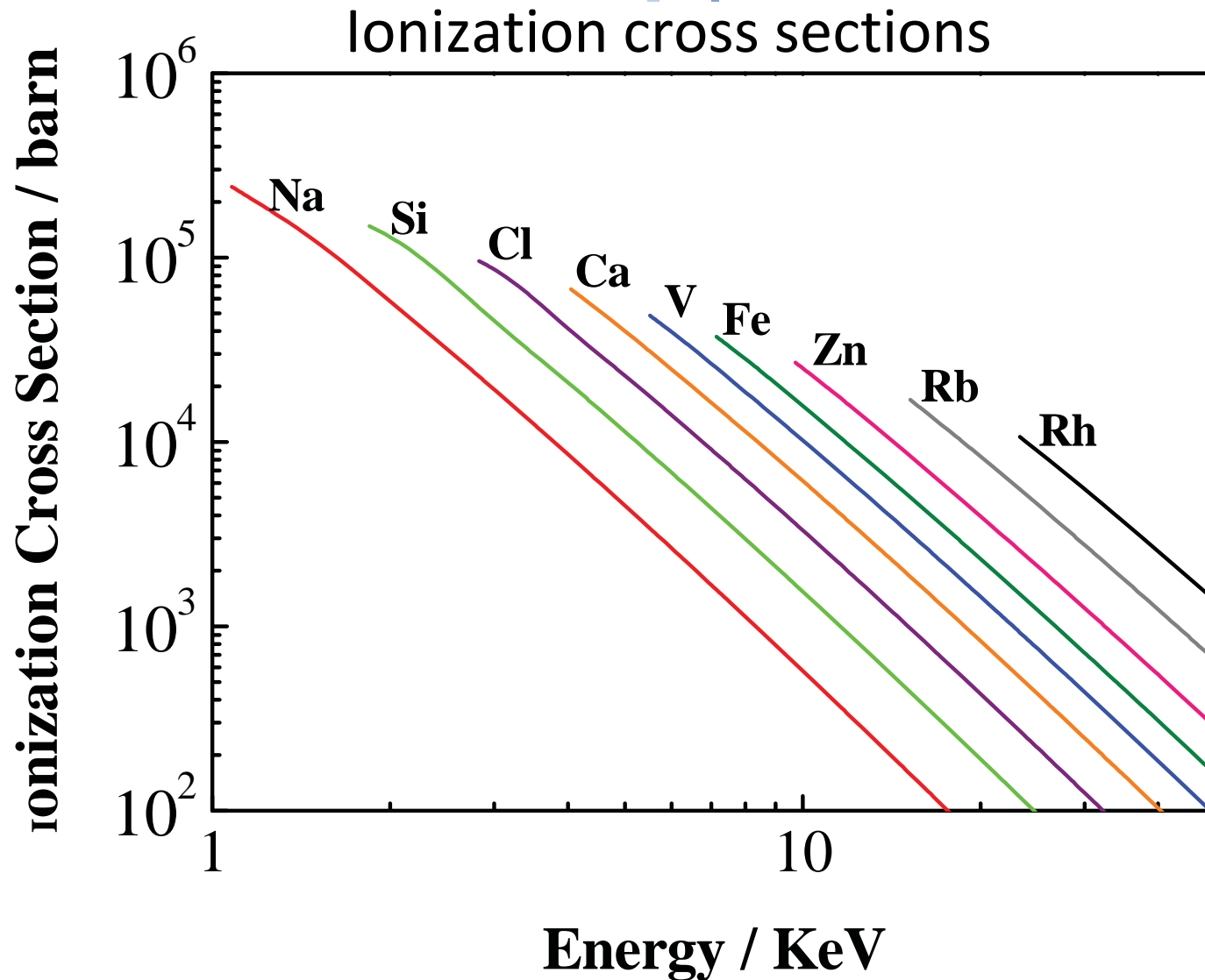


**Electrons**  
 **$10^2$ - $10^4$  barns**

Electrons ICS  
through  
Casnati, E *et. al*, *J  
Phys B At Mol Phys*  
**15**, 155–167.



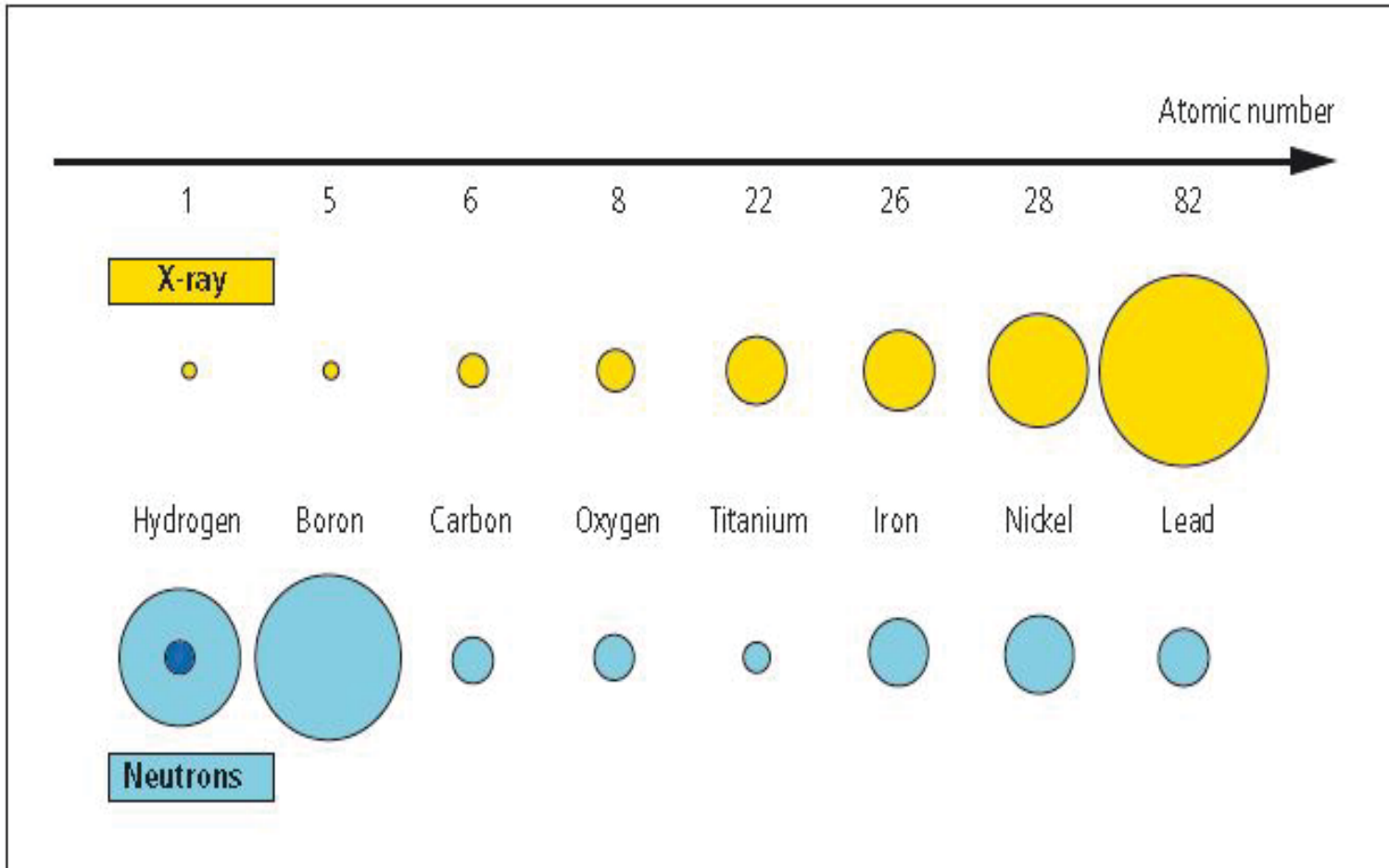
# Comparison of radiation Probes: X-rays (3)



**X-rays**  
 **$10^4$ - $10^5$**   
**barns**

Photon ICS from  
"Elam database"  
Elam W.T. et al.,  
*Radiat. Phys. Chem.*, 63,  
(2002), 121

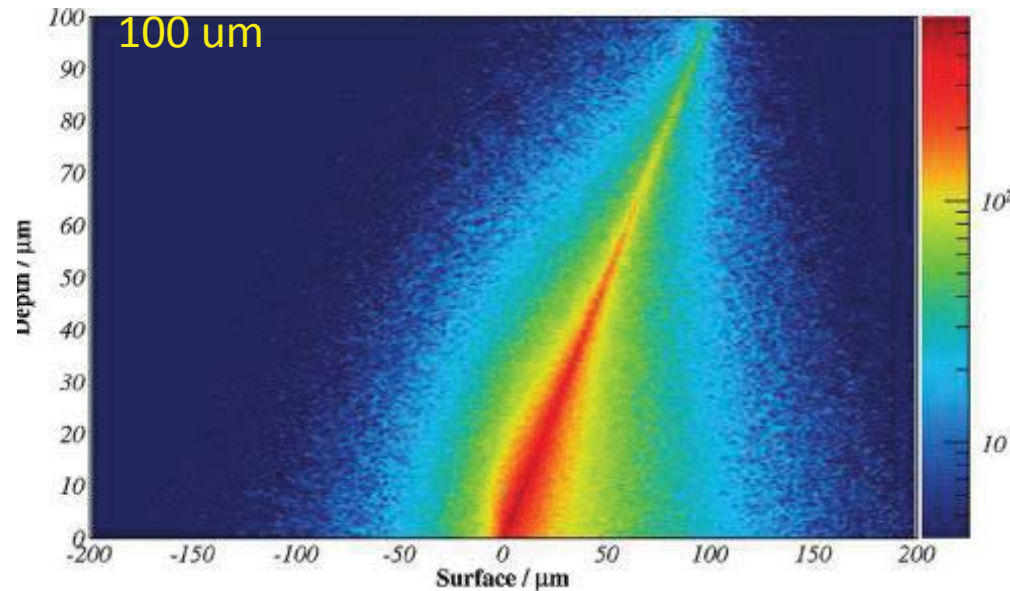
# Thermal Neutrons cross sections



Source: Paul Scherrer Institut, PSI, Switzerland

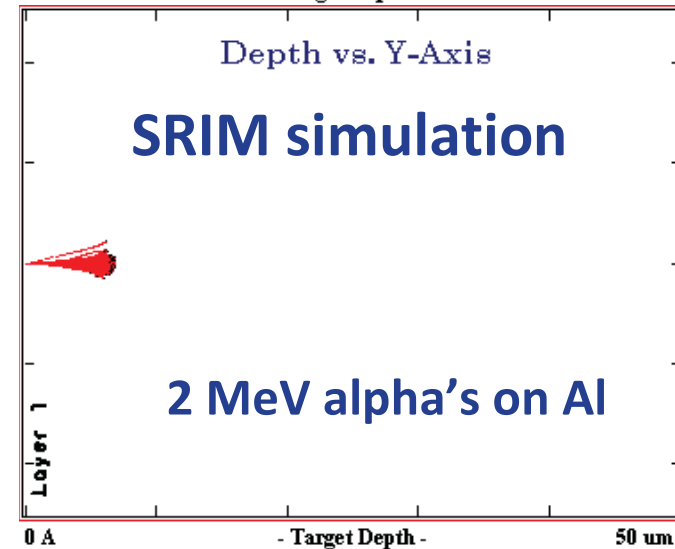
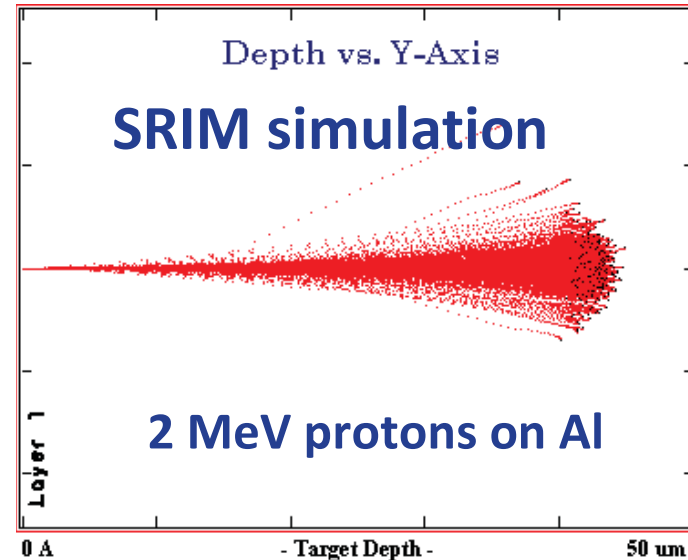
# Information depths by radiation probes:

## Examples:



13 keV, excitation,  
SiO<sub>2</sub> matrix, 5% Cu, 5% Fe  
Topology of secondary fluorescence  
of Fe-K radiation

*Sokaras et al, Anal. Chem. 2009, 81, 4946*



50  $\mu\text{m}$

# Analytical Techniques in Art and Archaeology

## □ Elemental/Isotopic analysis

Sy Micro X-ray Fluorescence analysis (Sy-XRF), Portable XRF  
SEM-EDX, Ion Beam Analysis (IBA): PIXE, RBS, PIGE  
Laser Induced Breakdown Spectroscopy (LIBS)  
Neutron Activation Analysis (NAA, isotope selective)

## □ Structural Information

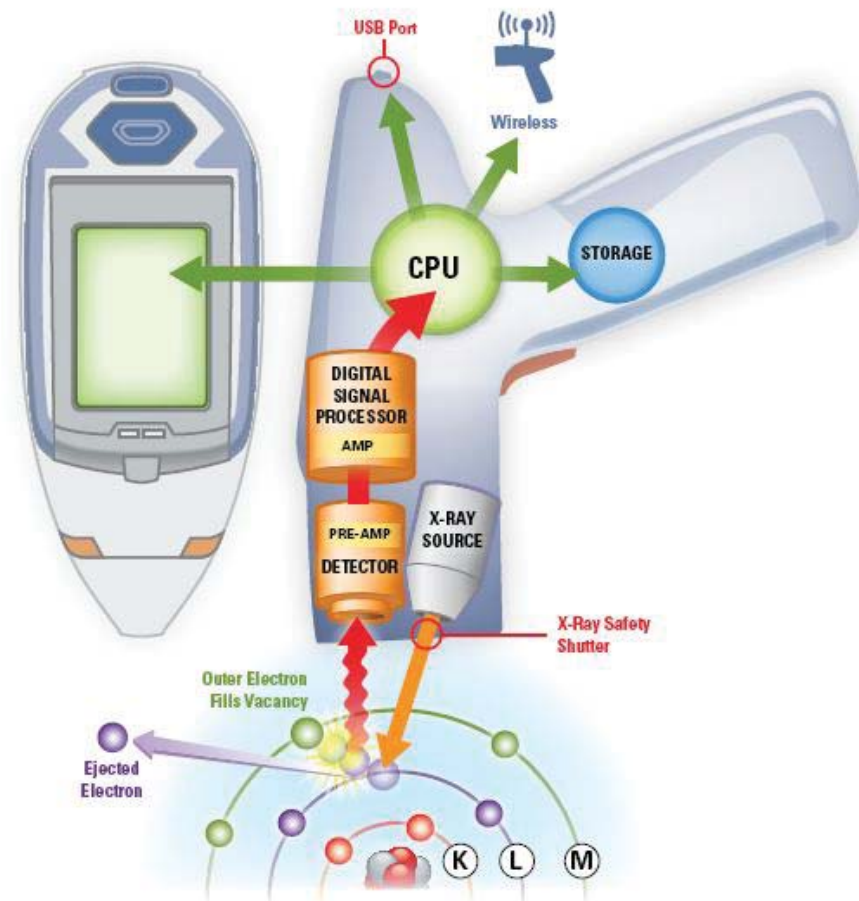
Molecular analysis (Raman, FTIR)  
Mineralogical/Crystalline phase analysis (TOF-ND, XRD)  
Oxidation state, chemical environment, Coordination site:  
XANES - EXAFS

## □ Imaging X-ray and Neutron imaging, UV-Vis-IR

# X-Ray Fluorescence Analysis - XRF

- Typically, from Z=11 or Z=12 upward.
- Sensitivity within the few ppm range (conventional x-ray sources, best excited elements) or few tens of ppb for synchrotron exciting radiation
- Spatial resolution from mm range down to few micrometers
- Portability, Handheld autonomous operation
- Quantitative information for materials presenting good preservation state
- Poor depth resolution: Confocal micro-XRF, micro-PIXE analysis

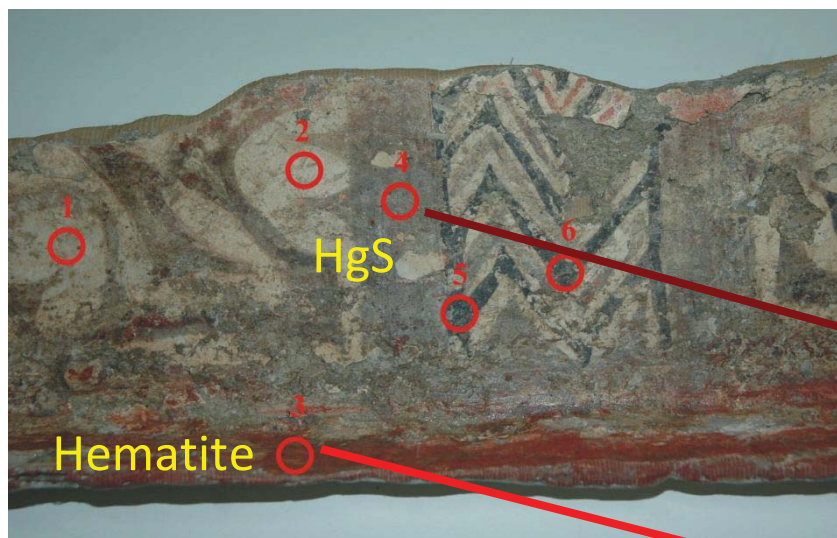
# New Developments in XRF: Hand-held analyzer



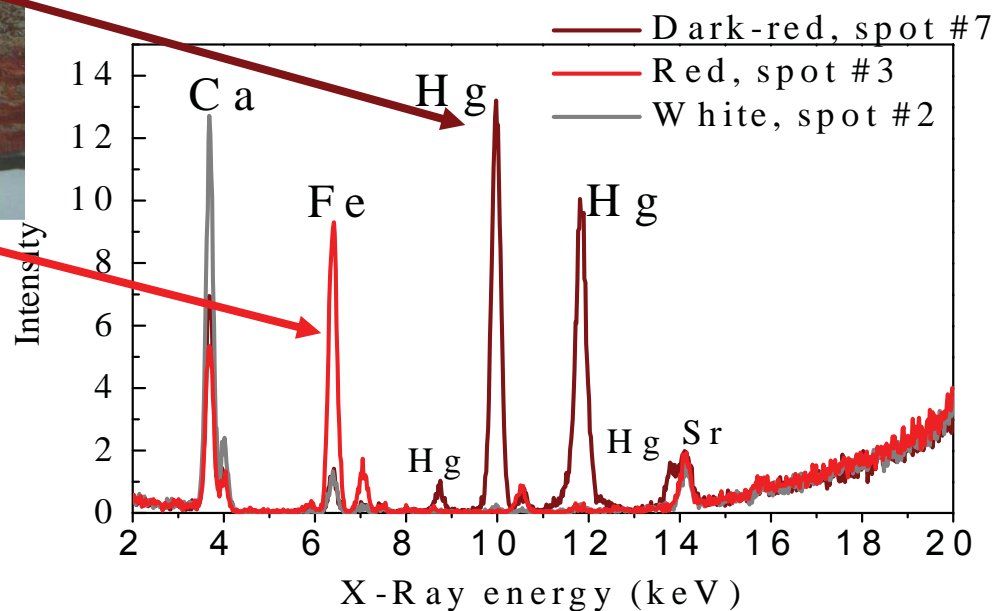
Ankara, Anatolian Civilization  
Museum, 2007



# ÇATALHÖYÜK 7000-8000 B.C. Wall-Painting pigments



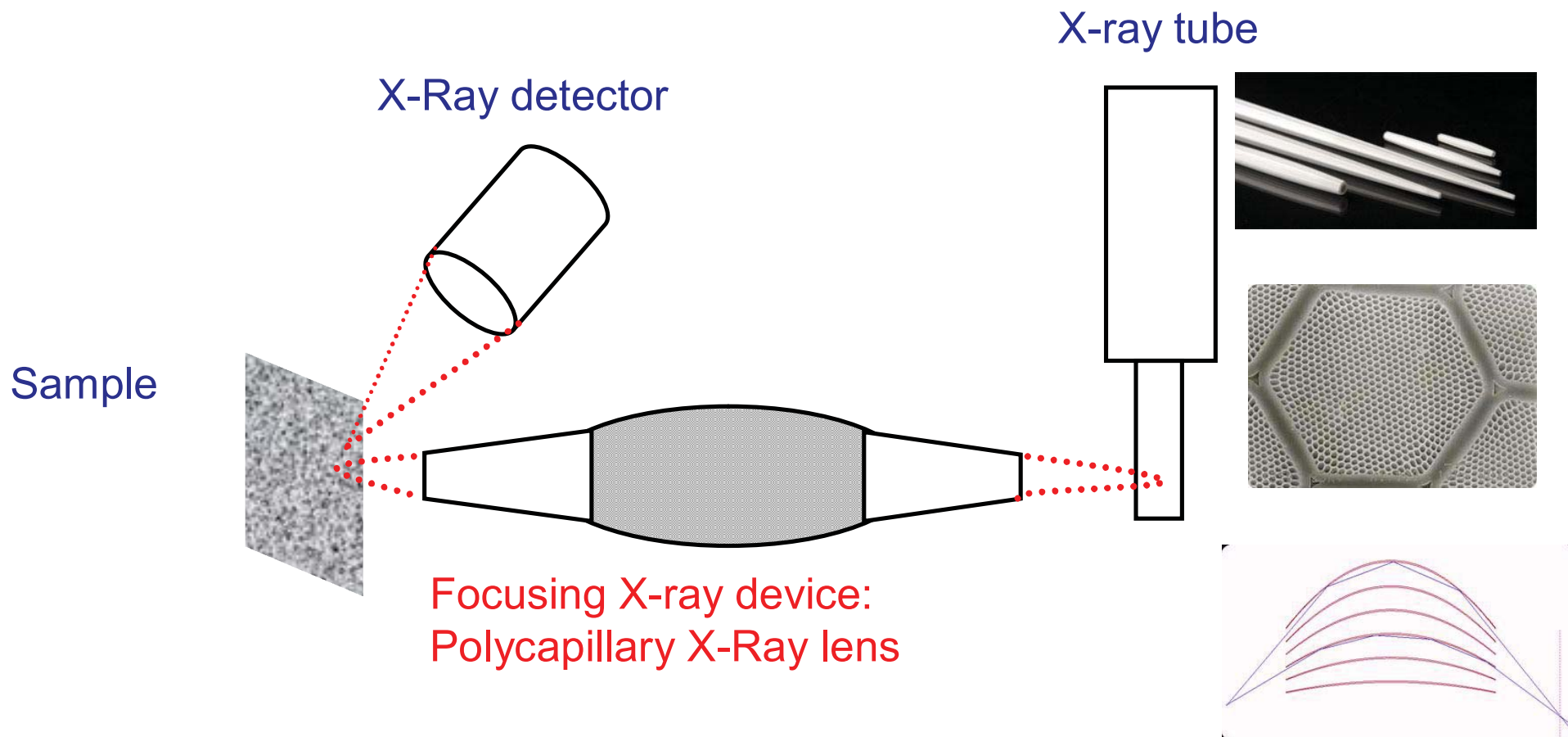
- ✓ Dark Red
- ✓ Red
- ✓ White
- ✓ Black



Hand-held XRF analysis, A. Zararsiz et al. 2008

# Developments in Micro-XRF spectroscopy

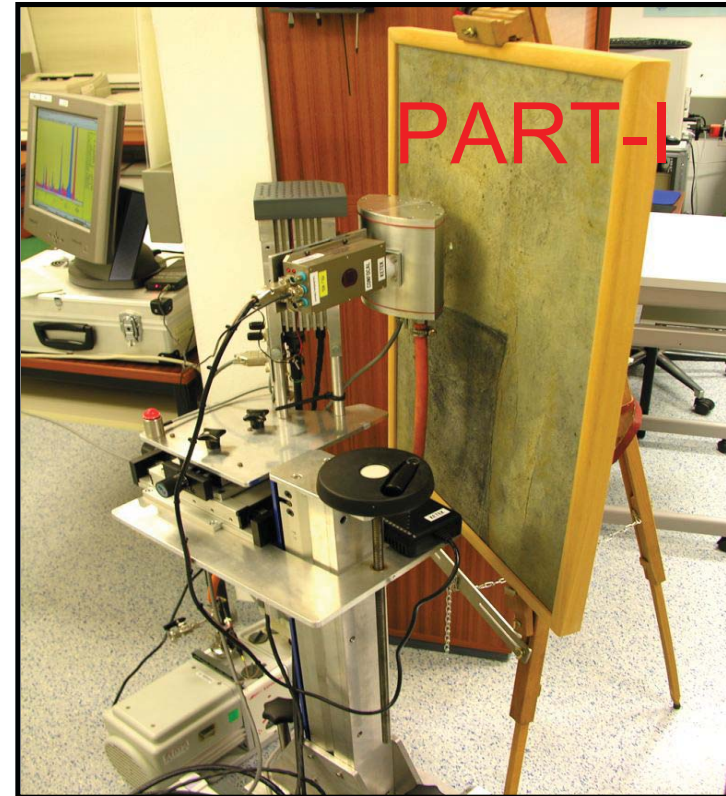
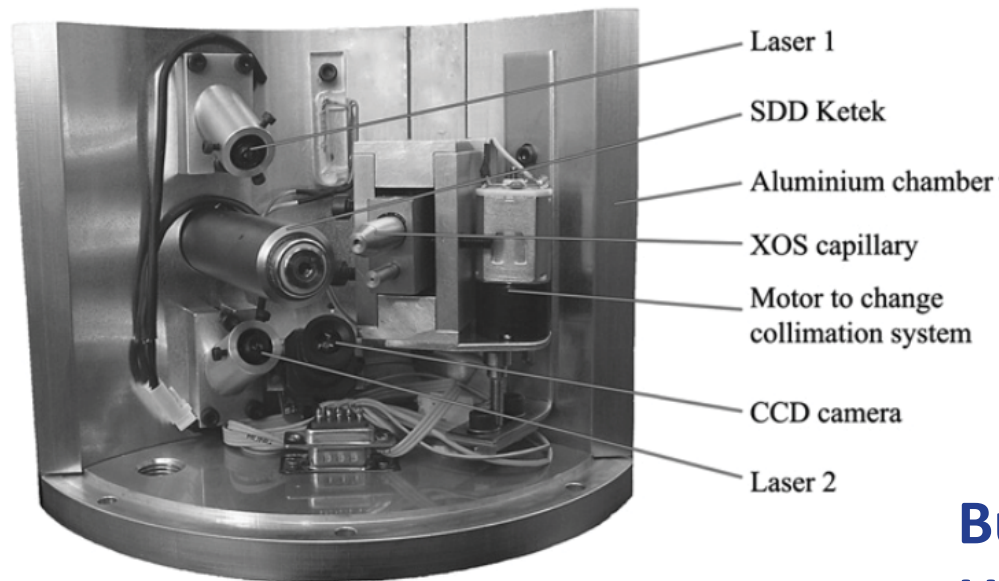
## X-ray lenses





# The Portable IAEA XRF prototype

- ✓ Pd-anode X-ray tube (50W)
- ✓ Pollycapillary lens
- ✓ silicon drift detector
- ✓ 2 laser pointers
- ✓ CMOS camera with
- ✓ Mechanical positioning system



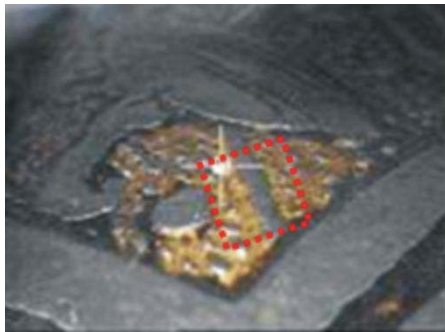
Buzanich et al, SAB 62 (2007) 1252

Uhlir et al, XRS, 37 (2008) 450

# Analytical possibilities: Gilding technique



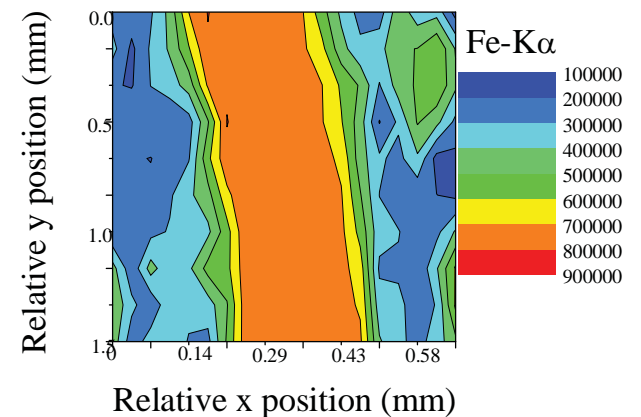
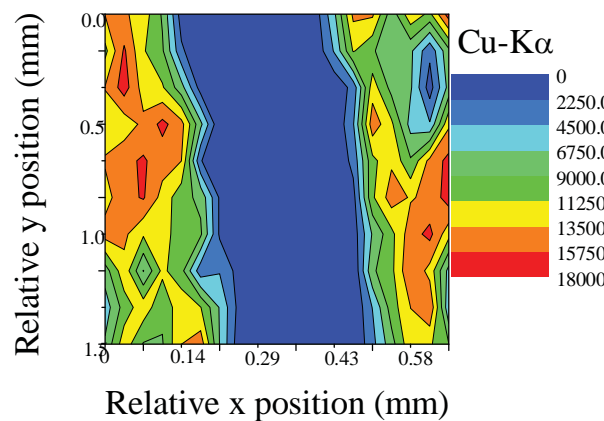
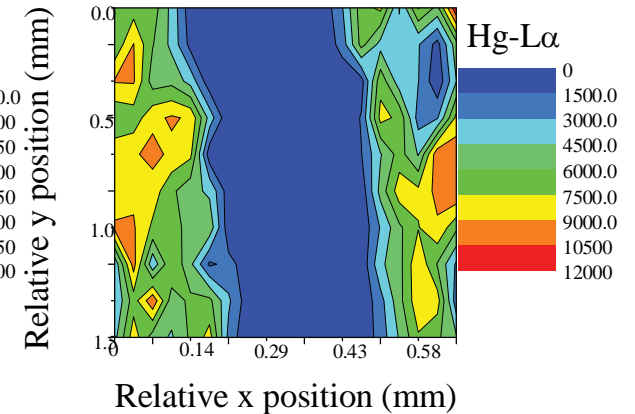
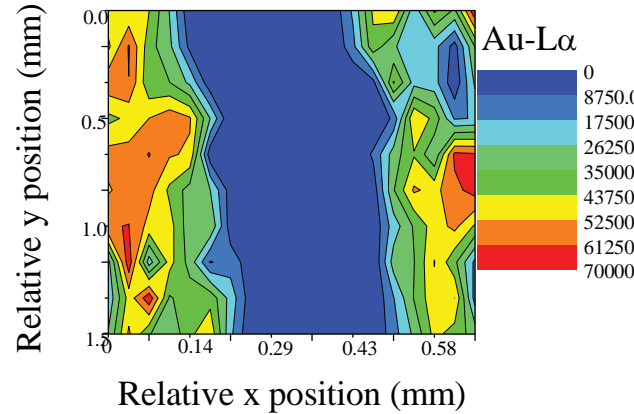
## Palace Armoury, Malta



Area scanned:  $1.9 \times 1.0 \text{ mm}^2$

Step size used: 0.1mm

Time per step: 20 s.



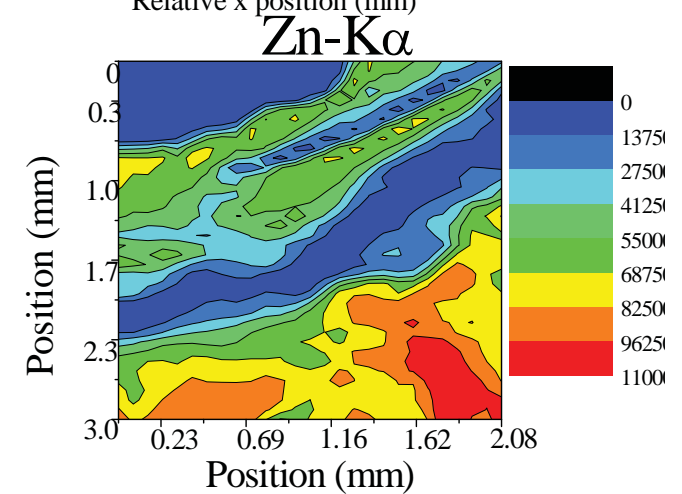
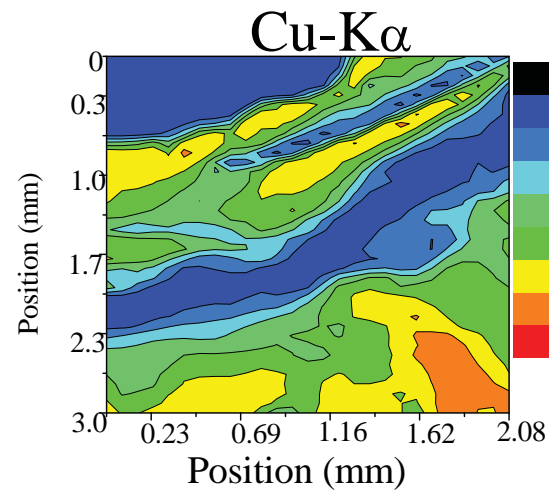
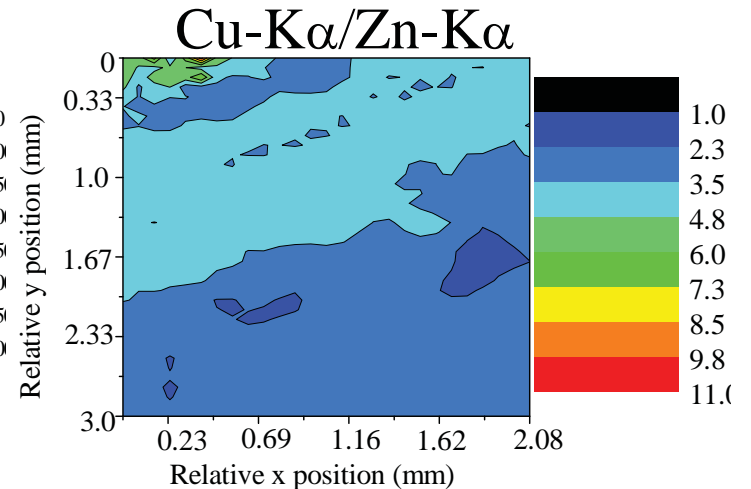
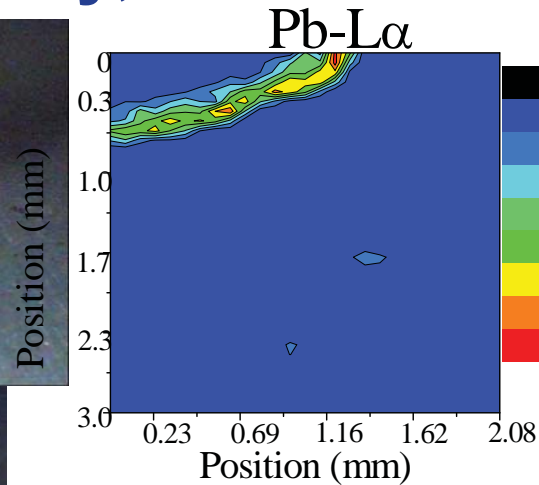
A.G. Karydas, D. Anglos and M.A. Harith, In *Metals and Museums in the Mediterranean: Protecting, Preserving and Interpreting*, ed. Argyropoulos, (2008) 141-177

A.G. Karydas, ICTP-IAEA SR-CH school, 21-11-2011

# Alteration of brass rivets plating



## Palace Armoury, Malta





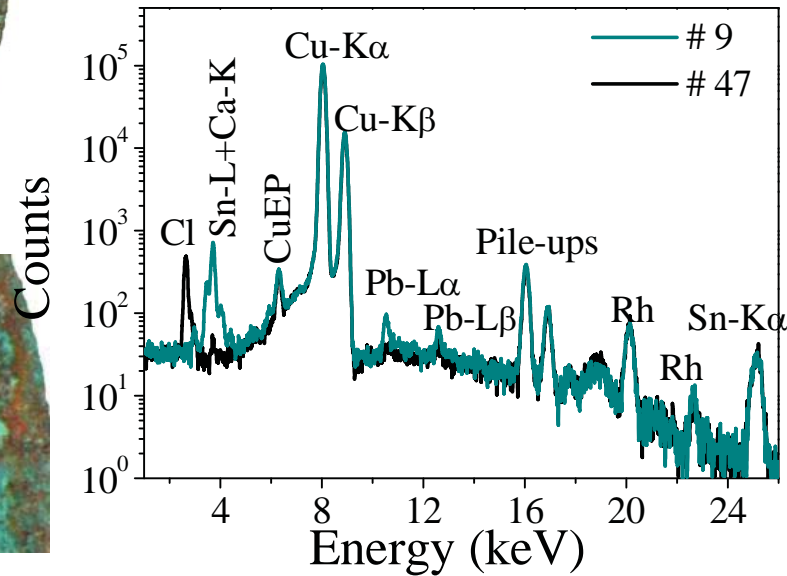
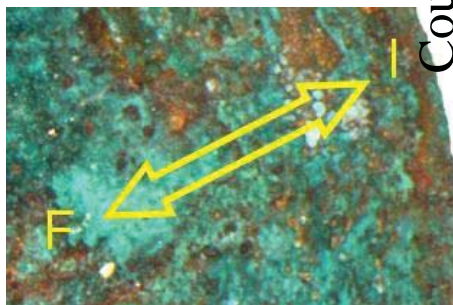
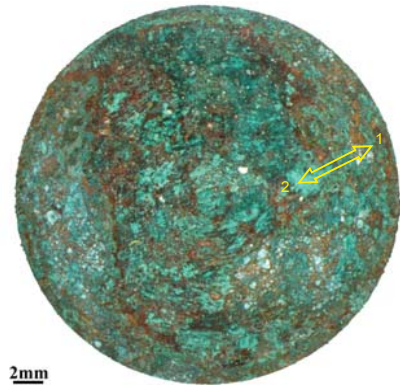
# Analysis of copper coupon corrosion products

Artificially and naturally aged bronze coupon: (Cu: 91.3%, Sn: 7.5%, Pb: 1.0%)



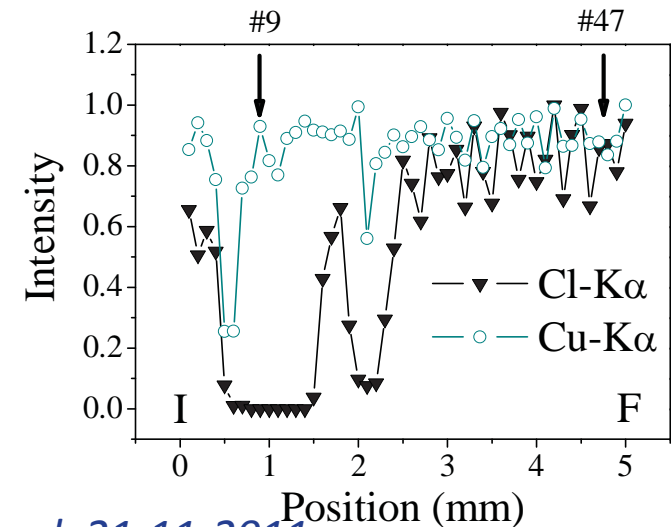
#9 : green area

#47: pale green area



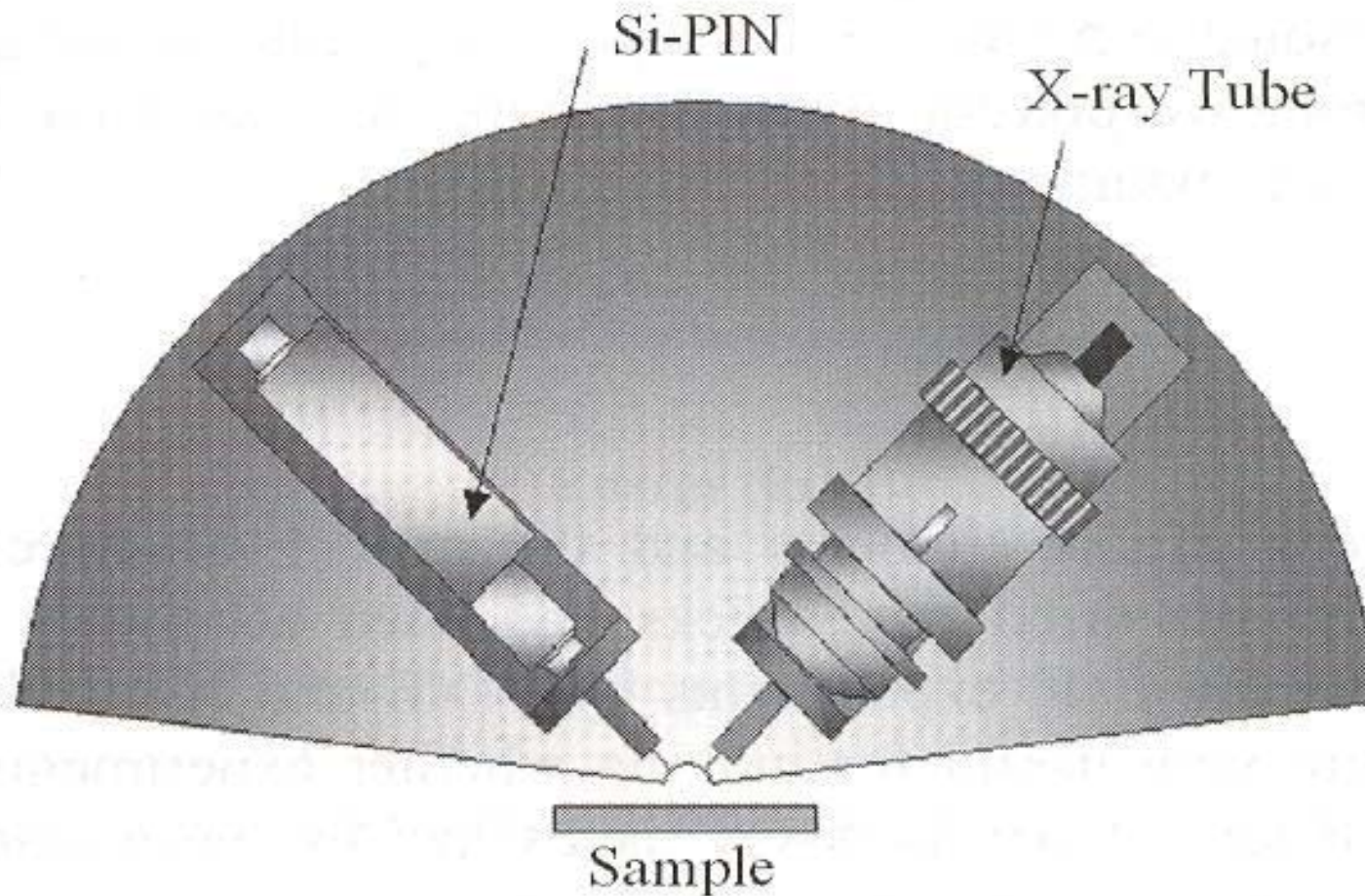
A.G. Karydas, D. Anglos and M.A. Harith, In *Metals and Museums in the Mediterranean: Protecting, Preserving and Interpreting*, ed. Argyropoulos, (2008) 141-177

50kV, 600μA,  
30s/step, 0.1mm/step,  
50 measurements





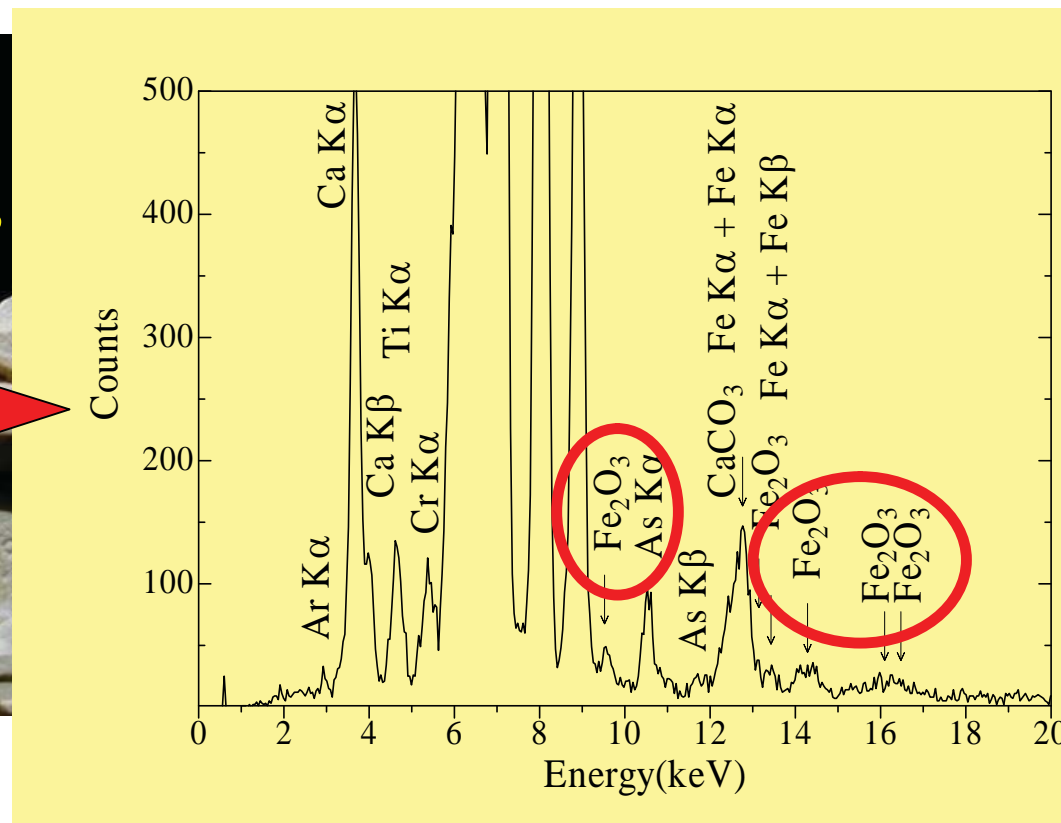
# Combined XRF-XRD Analysis



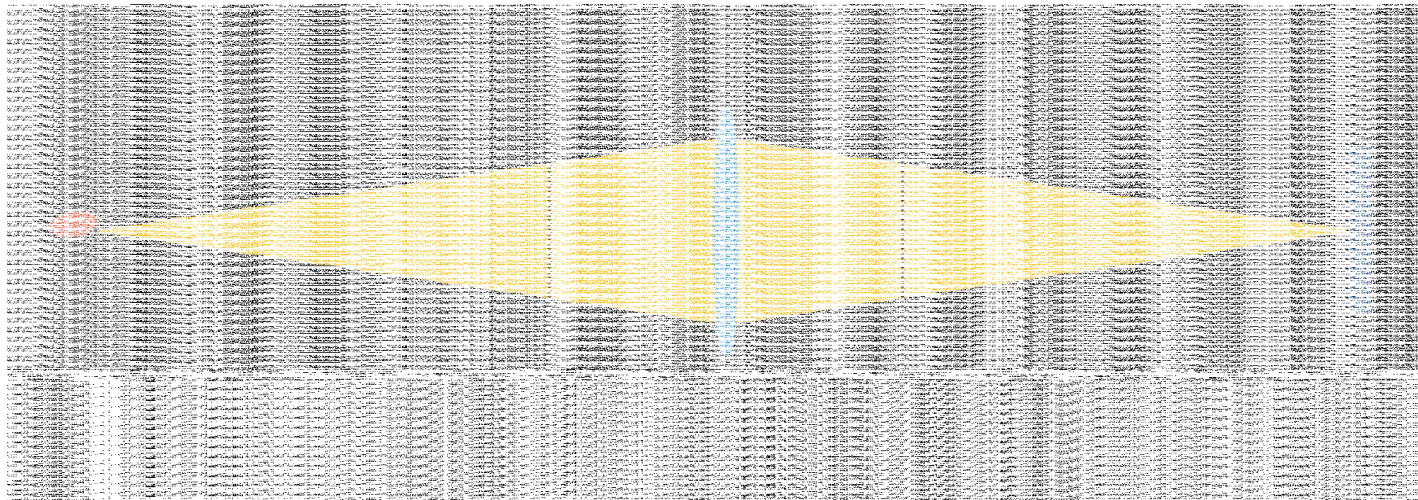
M. Uda, X-rays for Archaeology, 2005



# XRF-XRD, Pigments on Macedonia stele



# Synchrotron radiation features/techniques



- High brilliance
- Wide spectral range, IR- hard X-ray energy region
- Wavelength/energy tunability
- High degree and selection of polarization
- Well-defined time-space structure and coherence



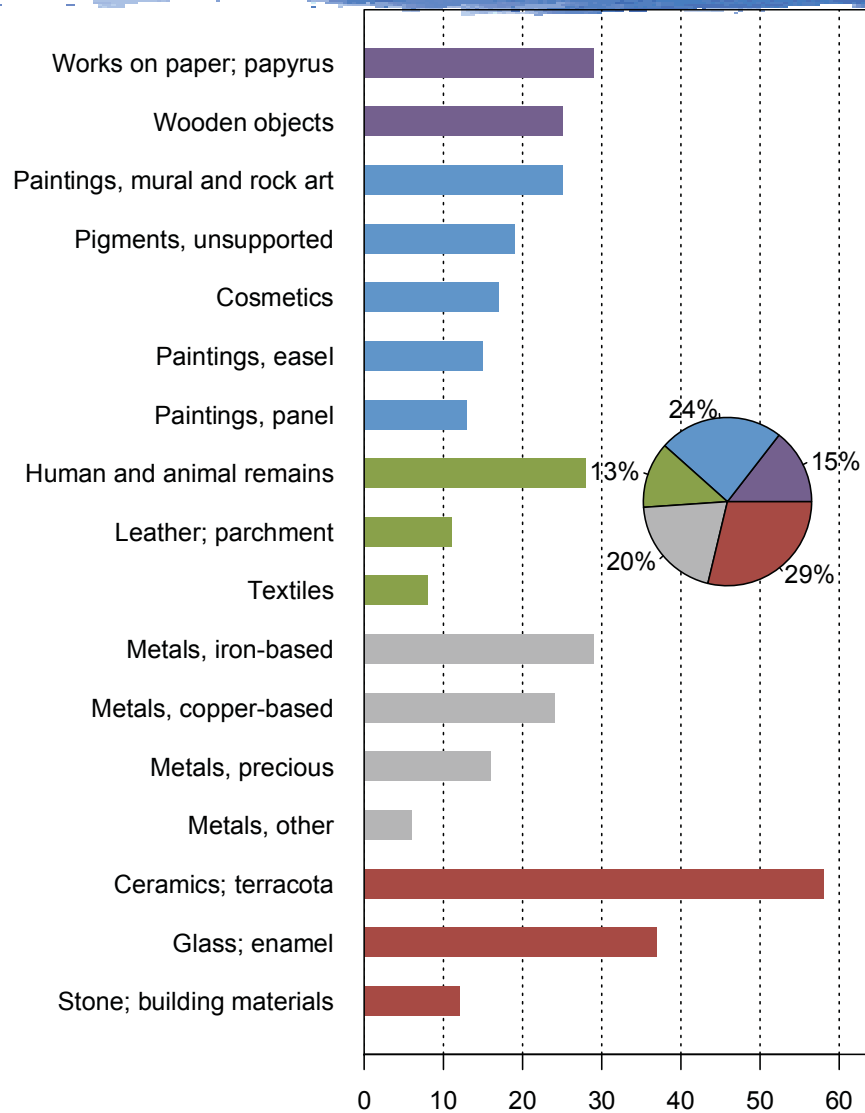
# Advantages of SR techniques

- Low destructiveness allowing protocols involving complementary characterization
- Integrated analytical information (from the atomic to the structural levels), due to the synergistic application of different techniques on the same experimental set-up
- Rapid analysis on a limited quantity of matter;
- The high spatial resolution offered by the beam spot size, in particular for micro-imaging, which allows a selective analysis;
- Some methods are non-invasive and allow for internal inspection of entire objects (hard X-ray, THz);
- The capability to perform real-time in-situ analyses.

# Limitations of SR techniques

- Objects and people need to be transferred;
- Most techniques are invasive (sampling required);
- There is a limited access to beam time, and skills in getting beam time may not always be accessible to new users;
- Communication and knowledge transfer between physical scientists and specialists in the fields of Cultural Heritage, Archaeology and Forensics

# CH materials studied by SR techniques.



Data by L. Bertrand,  
SOLEIL

# Applications in CH of SR techniques

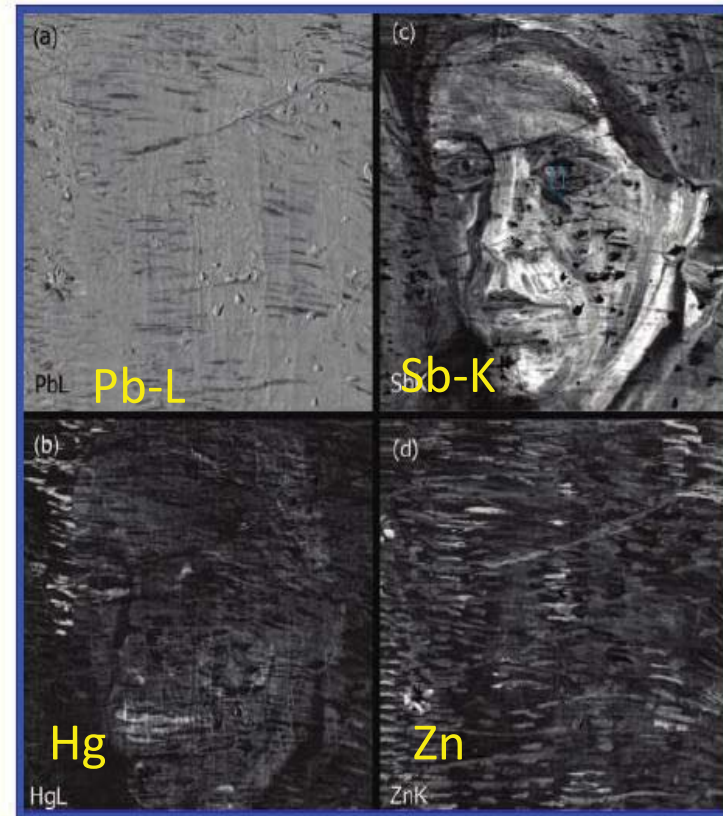
- Provenance
- Ancient technology
- Visualization of hidden/vanished art and historical documents,
- Study of degradation phenomena and products
- Evaluation, monitoring, optimization of consolidation, stabilization and conservation procedures

# Examples : Elemental mapping by Micro-XRF



DORIS at DESY:

38.5 keV, 0.5x0.5 mm<sup>2</sup>, 2s, 17.5x17.5 cm<sup>2</sup>

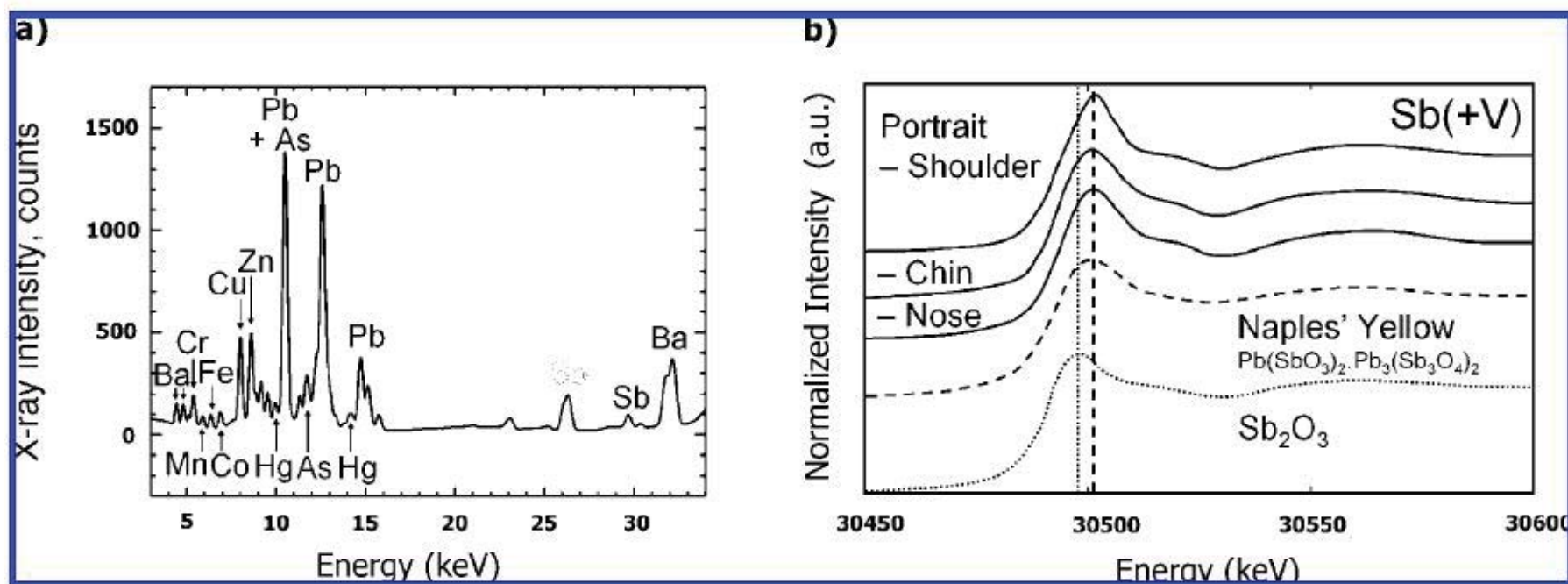


Dik et al., *Anal. Chem.*, **2008**, 80 (16)

Vincent van Gogh, *Patch of Grass*, Paris, Apr-June 1887, oil on canvas, 30 cm × 40 cm, Kroonmuller Museum, Otterlo, The Netherlands



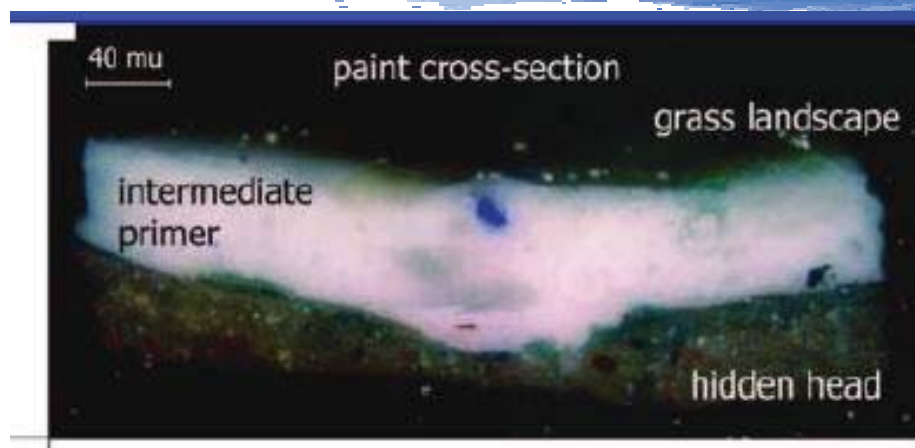
# XANES: Identification of chemical compound



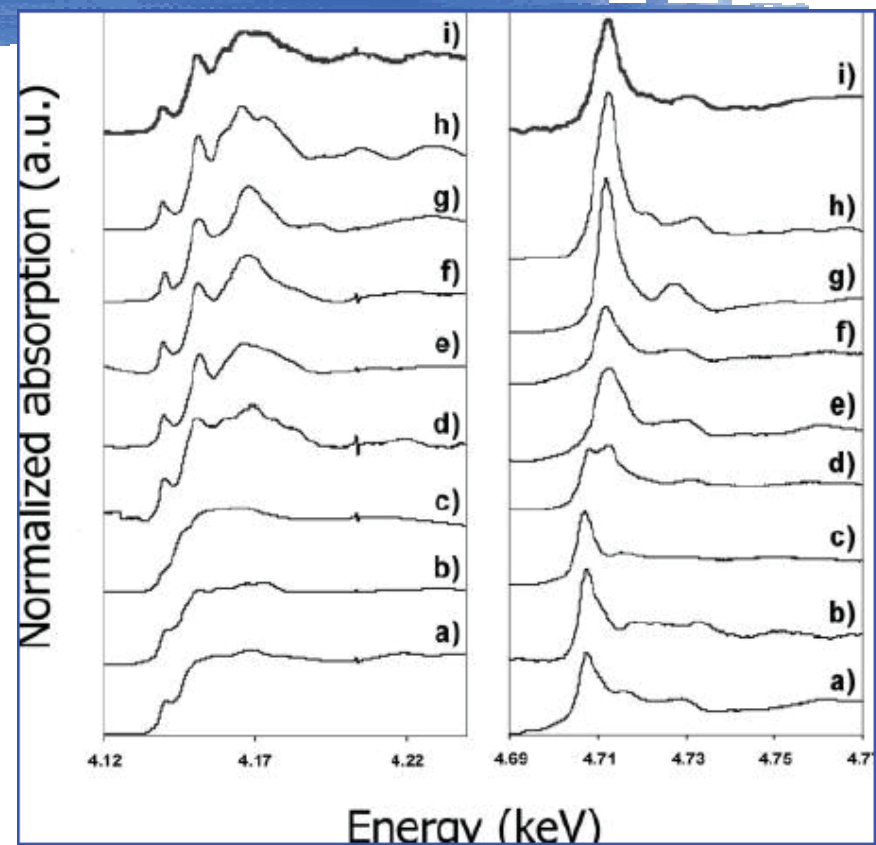
Comparison of Sb K- edge XANES spectra from three positions on the painting to reference XANES spectra of Naples yellow [ $\text{Pb}(\text{SbO}_3)_2 \cdot \text{Pb}_3(\text{Sb}_3\text{O}_4)_2$ ] and antimony white ( $\text{Sb}_2\text{O}_3$ ). All spectra were recorded in the **fluorescent mode**.

Dik et al., Anal. Chem., 2008, 80 (16)

# XANES: Identification of chemical compound

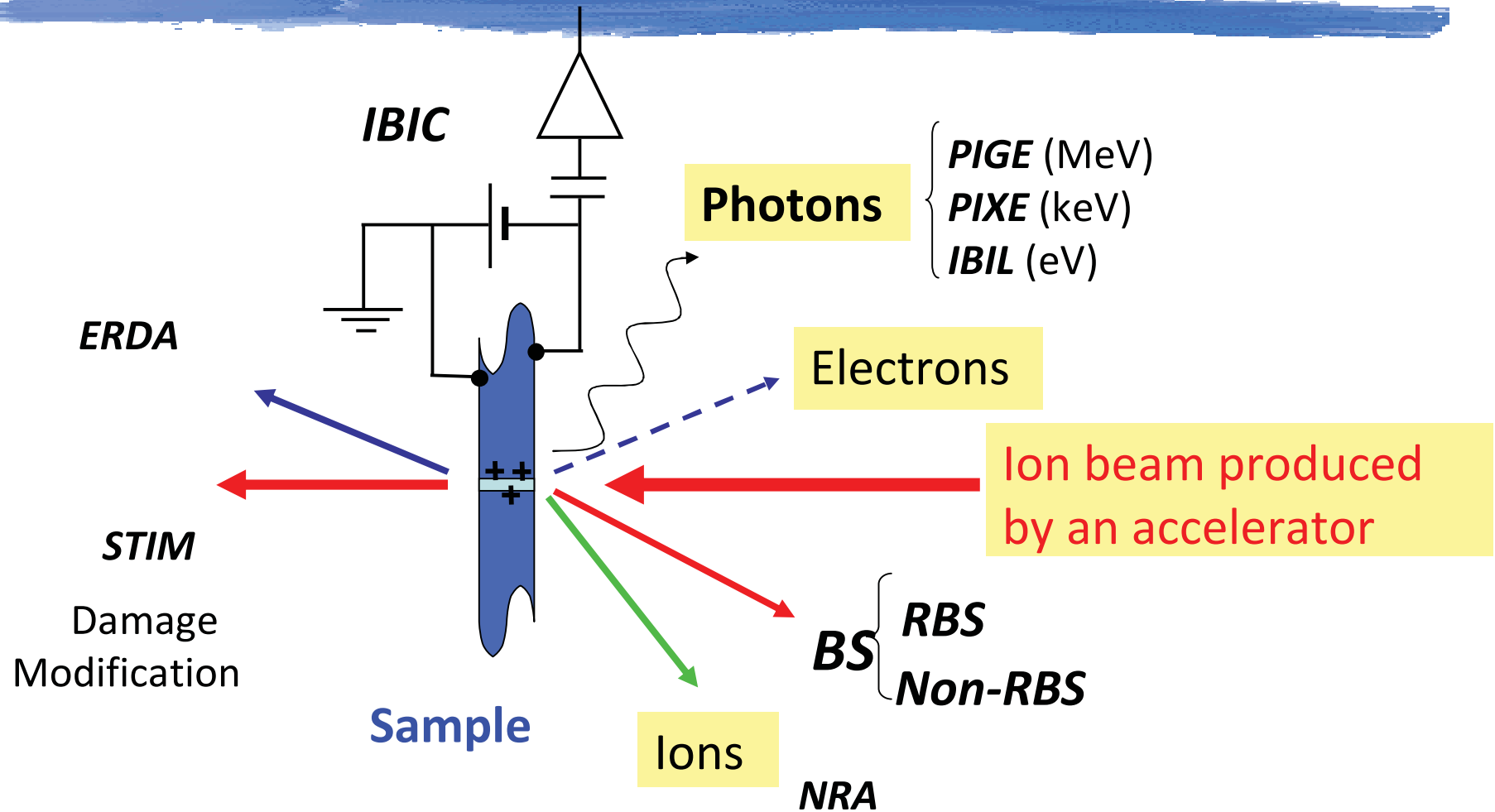


XANES spectra at the Sb-LIII edge and at the Sb-LI edge).



Reference antimony compounds:  $\text{Sb}_2\text{O}_3$  as (a) valentinite and as (b) senarmontite; (c)  $\text{Sb}_2\text{S}_2\text{O}$ , kermesite; (d)  $\text{Sb}_2\text{O}_4$ ; (e)  $\text{Sb}_3\text{O}_6\text{OH}$ , stibiconite; (f)  $\text{KSbO}_3 \cdot 3\text{H}_2\text{O}$ ; (g)  $\text{NaSbO}_3\text{OH} \cdot 3\text{H}_2\text{O}$ ; (h) Naples yellow; and (i) Sb pigment in the cross section of the Van Gogh painting

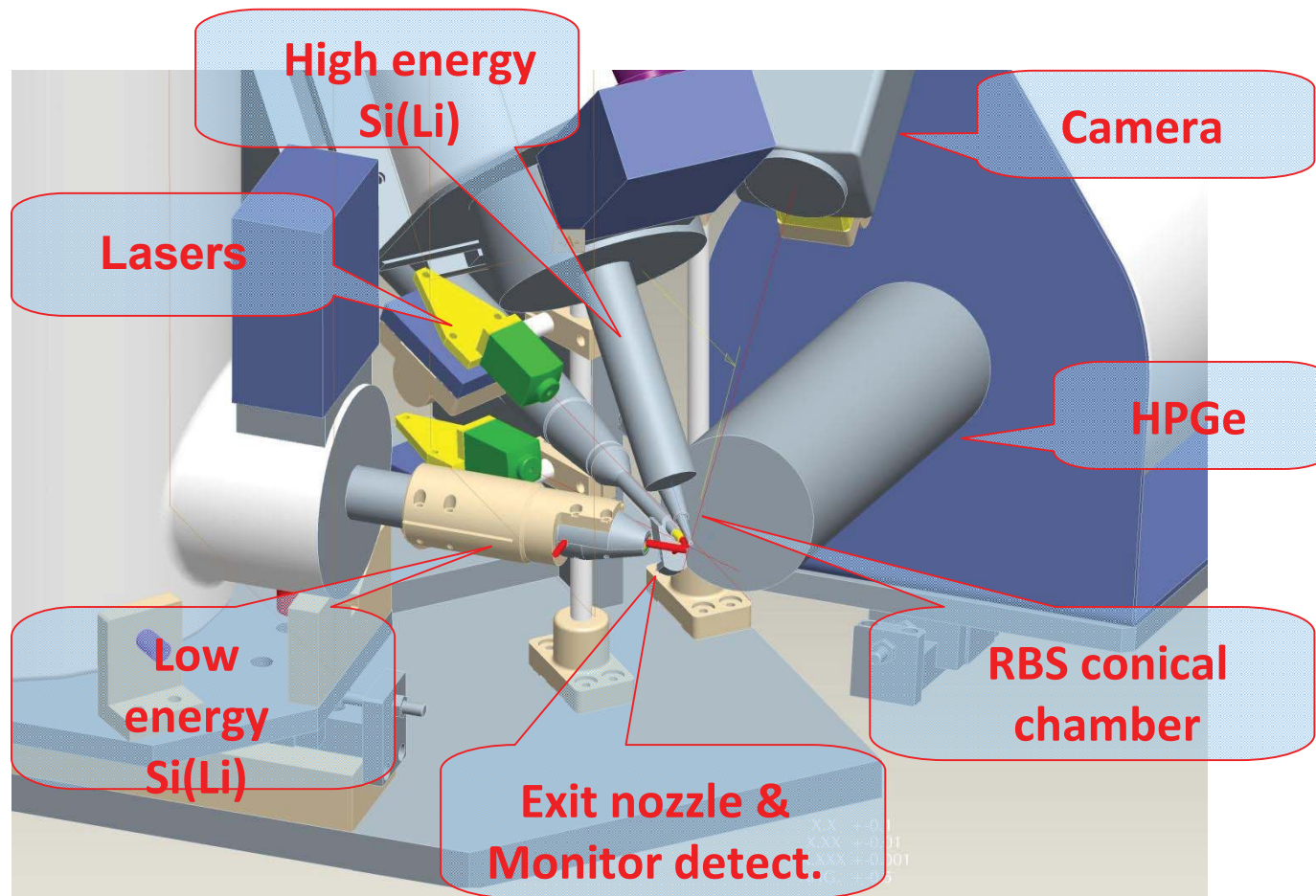
# Ion Beam Analysis



MeV ion beam based techniques constitute a powerful tool for the quantitative determination of the composition and structure of matter



# External Ion-Beam Analysis set-up



Sokaras et al, NIM B, 2011 NCSR “Demokritos”, Athens

# External Ion-Beam Analysis Set-up

Gkikas,  
1921-1922,  
oil-painting



# Synergy of Ion Beam Analysis Techniques

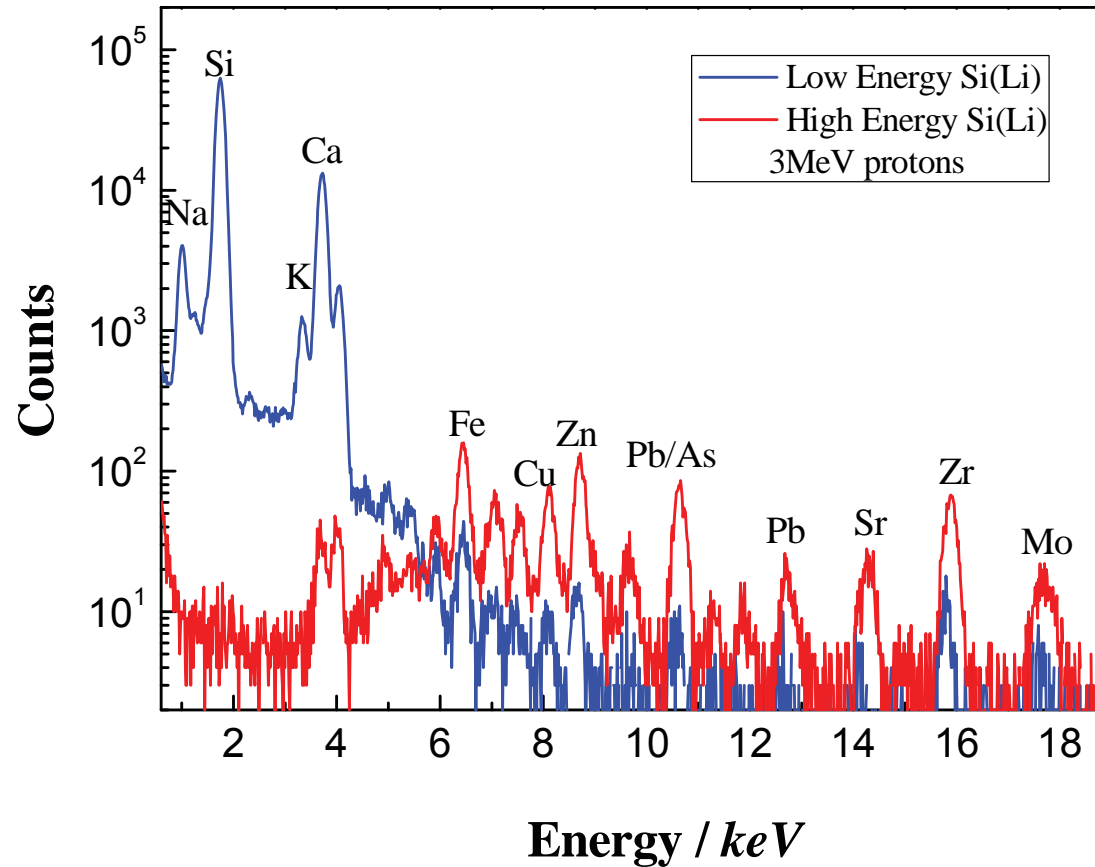
- Multi-elemental and near-surface depth resolved analysis of samples/artifacts.
- Analytical range: Lithium (Z=3) to Uranium (Z=92)
- Analytical sensitivity: From  $\mu\text{g/g}$  (ppm) to % level

## External Ion-Beam

- Non destructive
- No limitations (in general) for the size or shape of the object
- No heating, reduce damage
- No sampling, no charging, no preparation
- Easy sample positioning

# PIXE1: Multi-elemental analysis

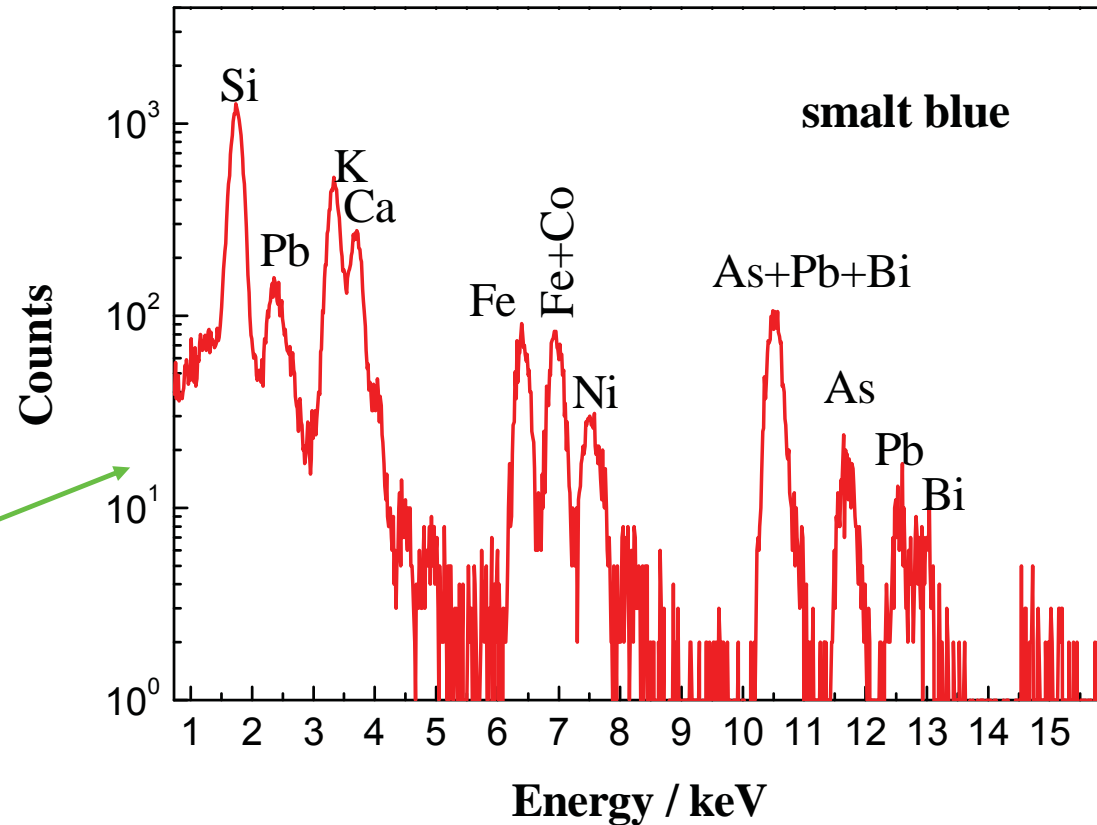
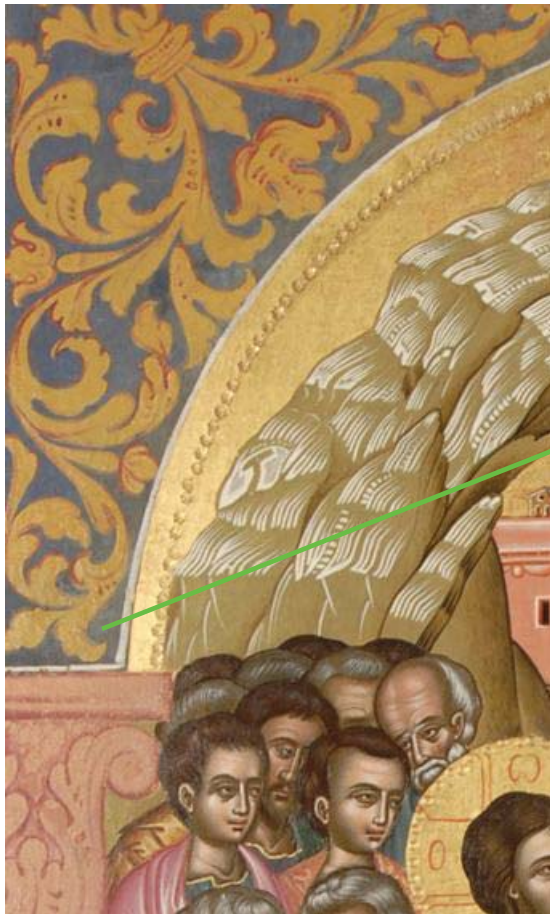
PIXE spectra, SRM Glass BAM-S005B



Sokaras et al, ABC, 2009

# PIXE: Elemental Analysis of smalt

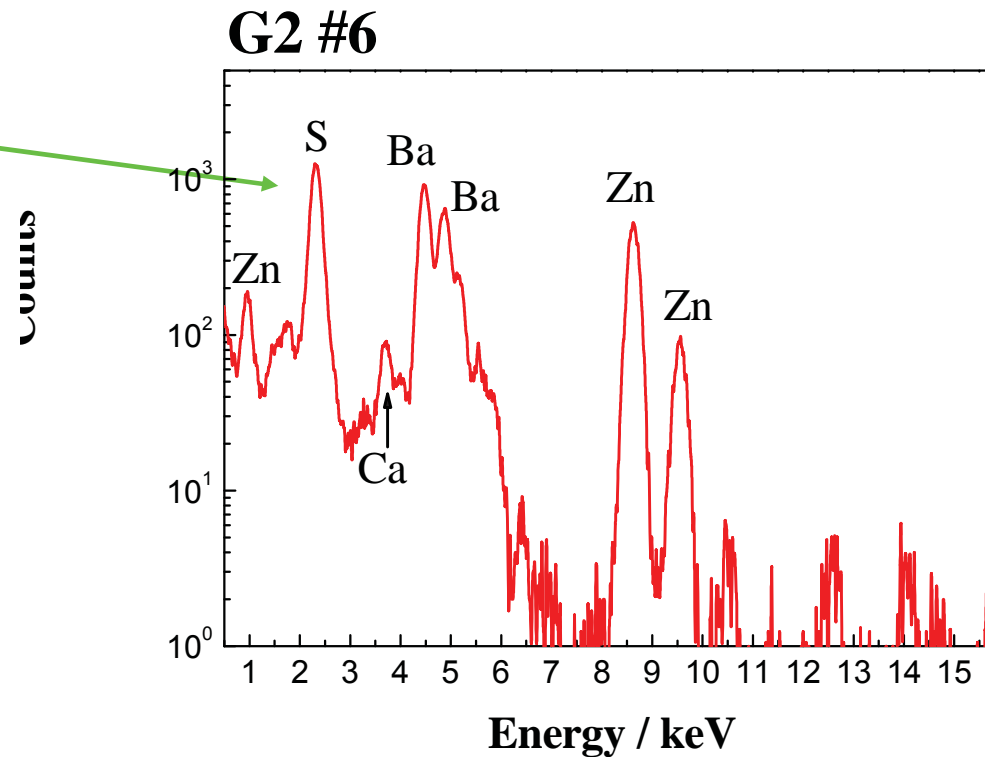
B137 #15



**Smalt Blue: Quartz,  
Smalt minor elements  
Fe, Co, Ni, As Pb, Bi**



# PIXE: Analysis of modern painting pigment



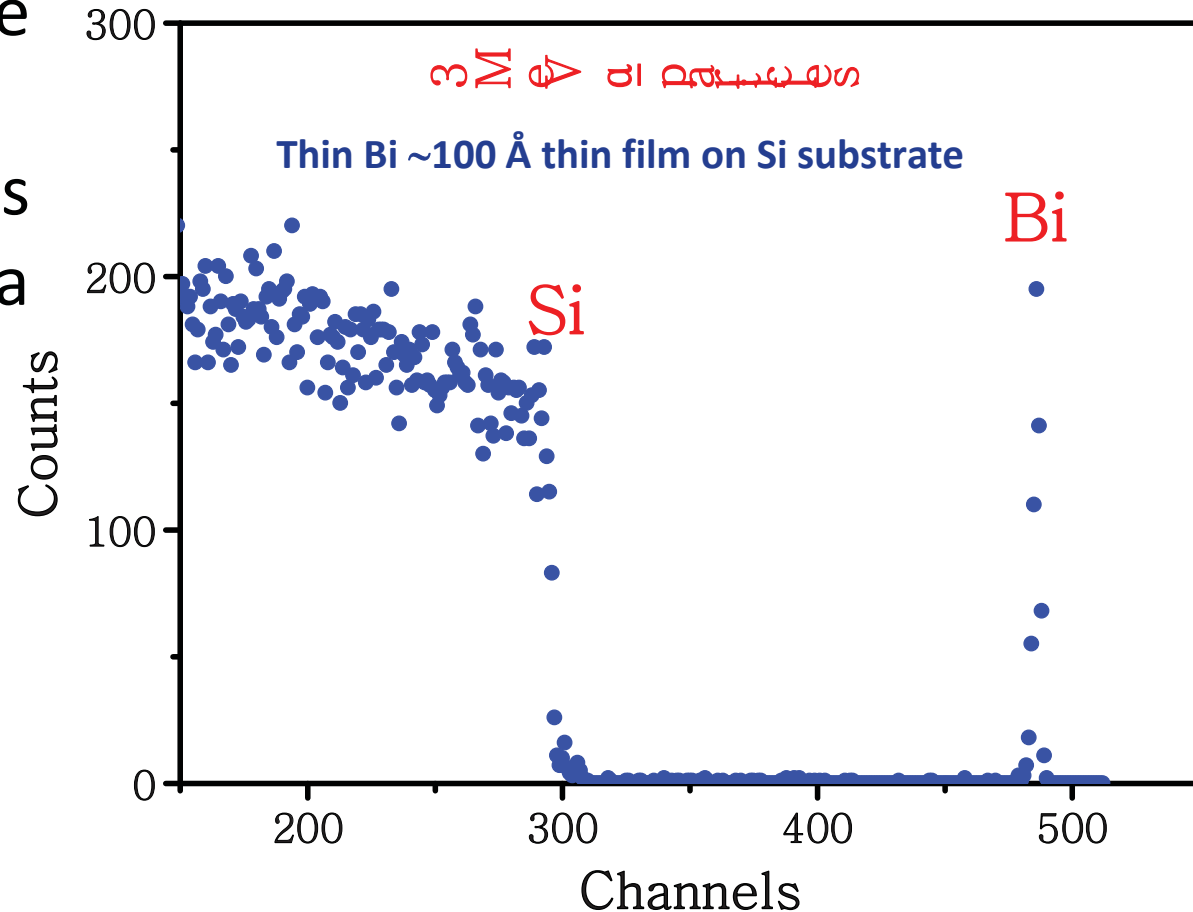
➤ **Lithopone, mixture of ZnS and BaSO<sub>4</sub>**

Greek contemporary painter, Nikos Chatjikyriakos-Gkikas, Benaki museum

# Rutherford-Back Scattering (RBS) Spectroscopy

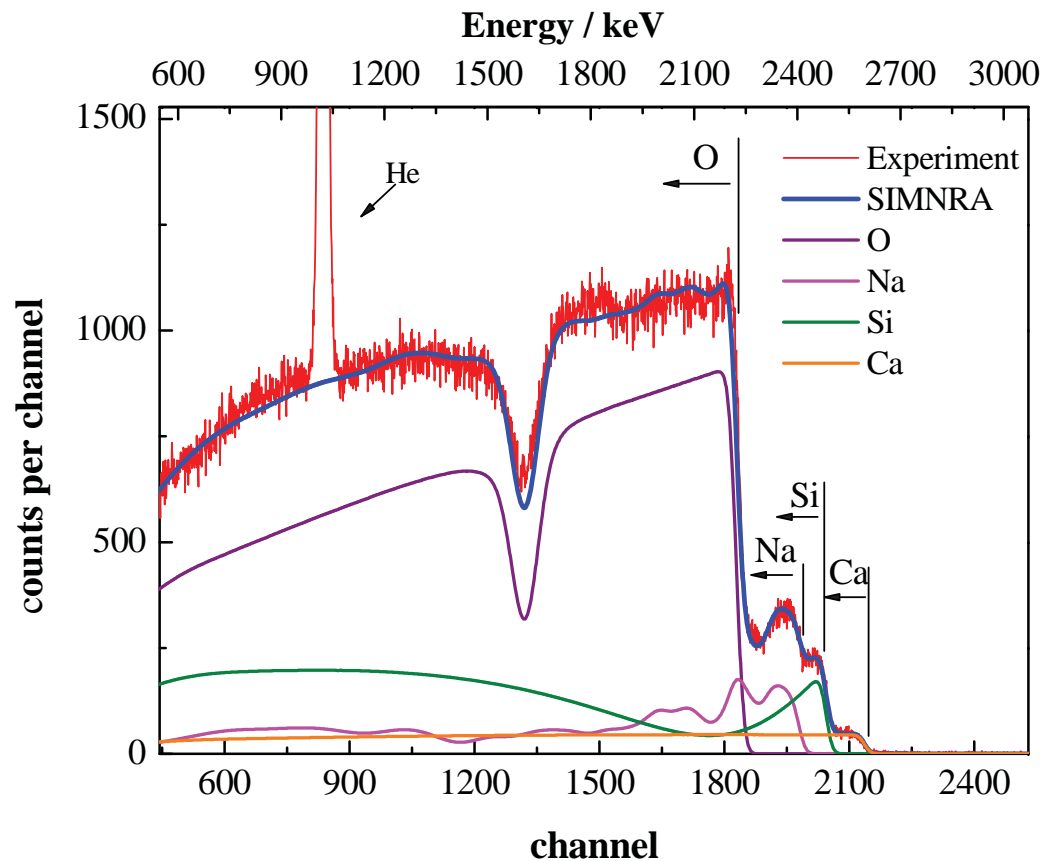
**RBS:** Near-surface depth resolved elemental analysis in the range of a few micrometers.

**Detection:** Energy of the elastically back-scattered charged particles



➤ Concentration profile versus depth is provided by the energy loss of the BS particles

# RBS: Elemental analysis – Homogeneous sample



**NIST Soda Lime glass 620**

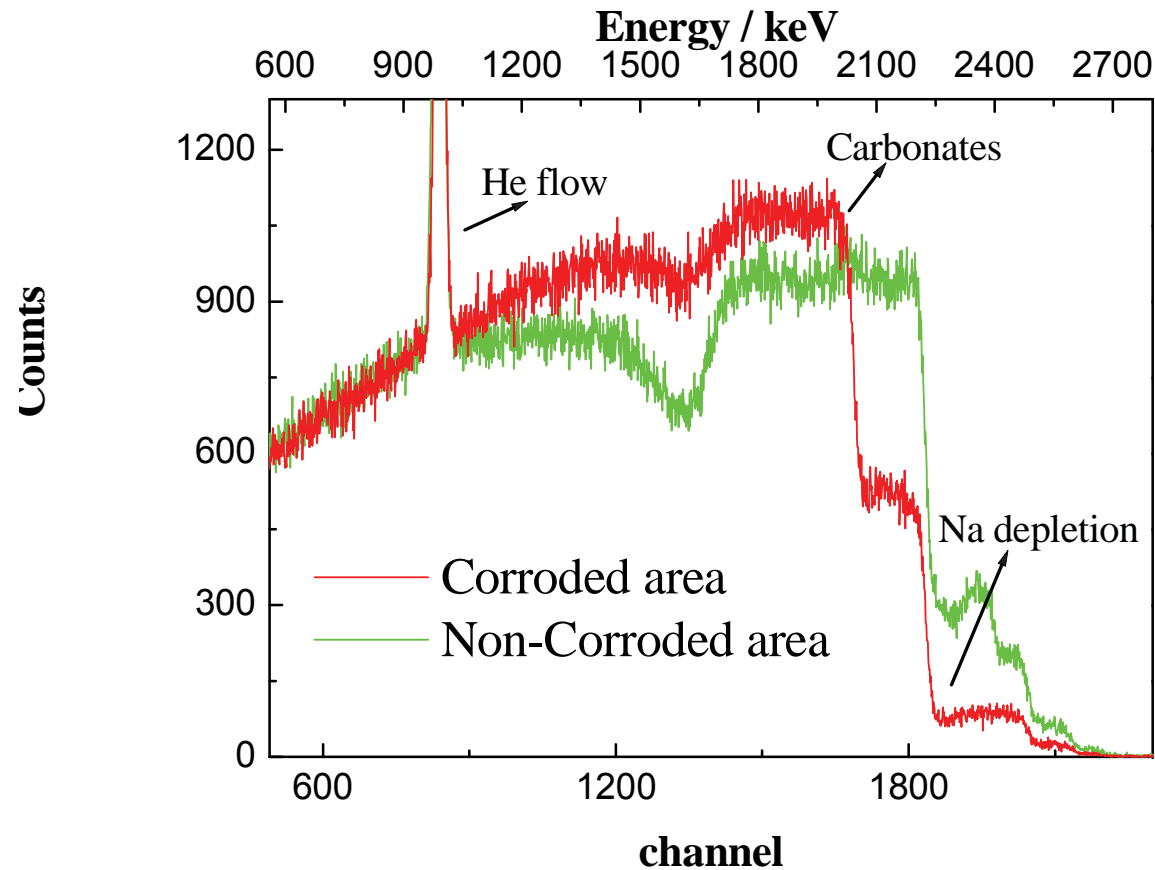
➤ Protons 3MeV

➤ Angle 161°

Element	PIXE	RBS	Certified
<i>O</i>		<b>45.3</b>	<b>47.0</b>
<i>Na</i>	<b>10.6</b>	<b>7.2</b>	<b>10.7</b>
<i>Si</i>	<b>36.5</b>	<b>39.1</b>	<b>33.8</b>
<i>Ca</i>	<b>5.05</b>	<b>6.4</b>	<b>5.08</b>



# RBS: Depth elemental profiling - Corrosion



# PIGE: Particle Induced gamma-ray emission

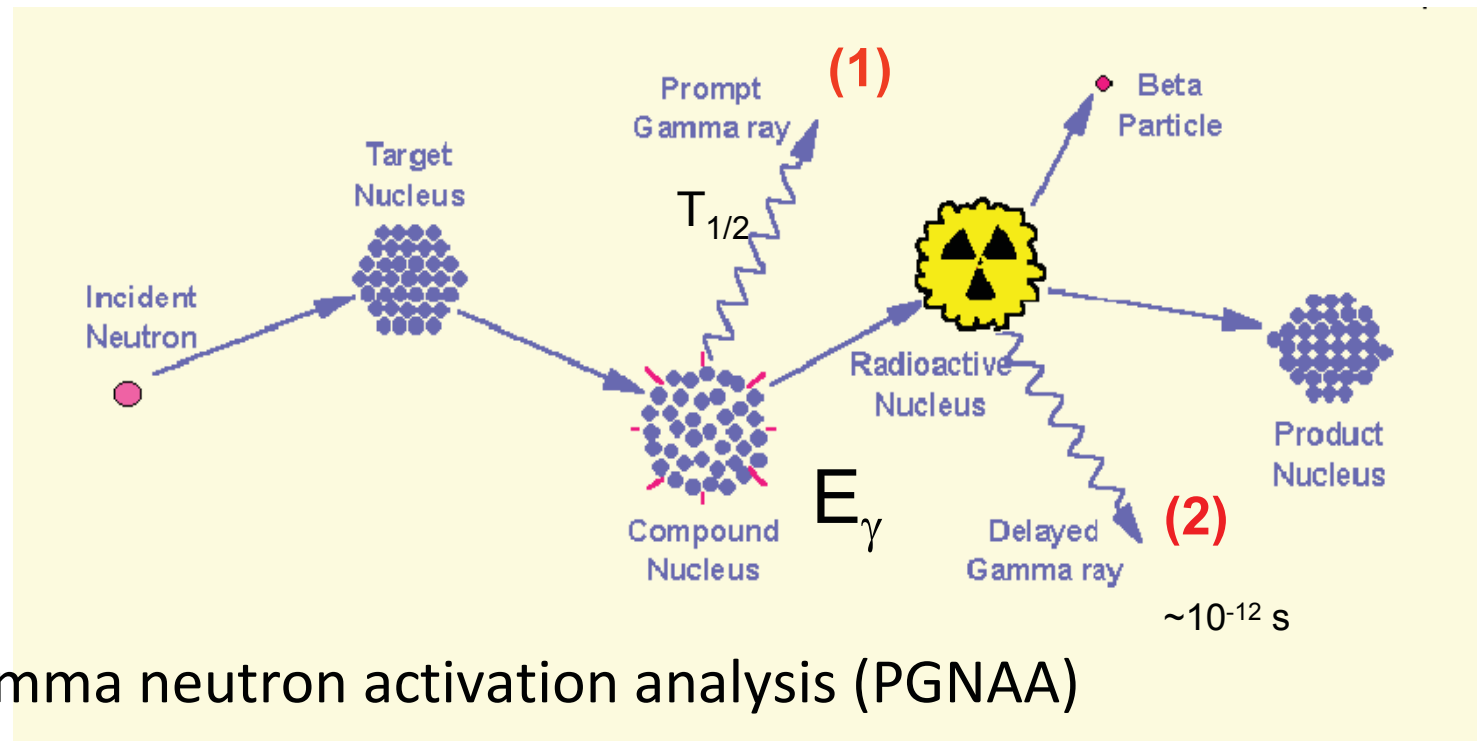
Nuclear Reactions may induce isotope specific characteristic  $\gamma$ -rays emission.

Light elements: Li, B, Na, Mg, Al, Si

Element	$E_p = 1.77 \text{ MeV}$		$E_p = 4.0 \text{ MeV}$	
	$\gamma$ -Ray (keV)	Reaction	$\gamma$ -Ray (keV)	Reaction
Li	478	${}^7\text{Li}(p, p'\gamma){}^7\text{Li}$	478	${}^7\text{Li}(p, p'\gamma){}^7\text{Li}$
B	429	${}^{10}\text{B}(p, \alpha\gamma){}^7\text{Be}$	718	${}^{10}\text{B}(p, p'\gamma){}^{10}\text{B}$
F	6129	${}^{19}\text{F}(p, \alpha\gamma){}^{16}\text{O}$	197	${}^{19}\text{F}(p, p'\gamma){}^{19}\text{F}$
Na	440	${}^{23}\text{Na}(p, p'\gamma){}^{23}\text{Na}$	440	${}^{23}\text{Na}(p, p'\gamma){}^{23}\text{Na}$
Mg	585	${}^{25}\text{Mg}(p, p'\gamma){}^{25}\text{Mg}$	585	${}^{25}\text{Mg}(p, p'\gamma){}^{25}\text{Mg}$
Al	1779	${}^{27}\text{Al}(p, \gamma){}^{28}\text{Si}$	1014	${}^{27}\text{Al}(p, p'\gamma){}^{27}\text{Al}$
Si			1779	${}^{28}\text{Si}(p, p'\gamma){}^{28}\text{Si}$
P			1266	${}^{31}\text{P}(p, p'\gamma){}^{31}\text{P}$

# Neutron Activation Analysis -NAA

Neutron capture by a target nucleus followed by emission of gamma rays



(1) Prompt-gamma neutron activation analysis (PGNAA)

(2) Delayed-gamma neutron activation analysis (NAA)

Instrumental (INAA), *minimum sample preparation*

Radiochemical (RNAA), *chemical analysis of the sample*

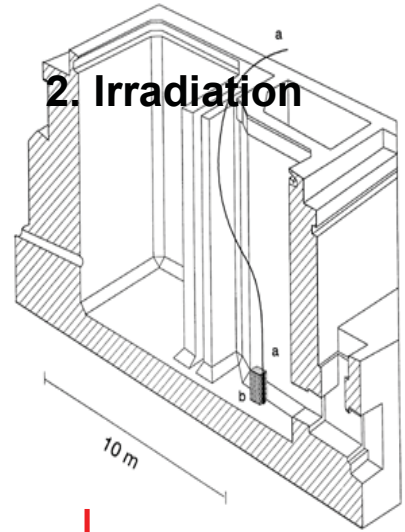
# INAA procedure

## 1. Sample encapsulation

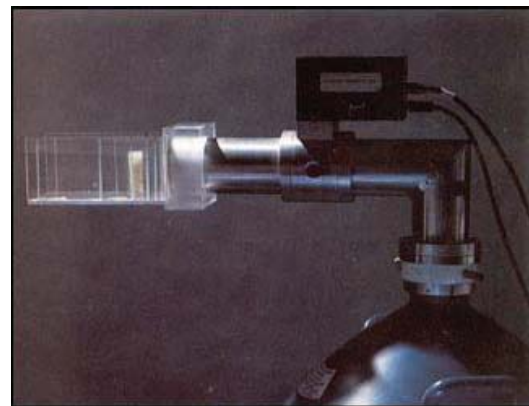
Sample



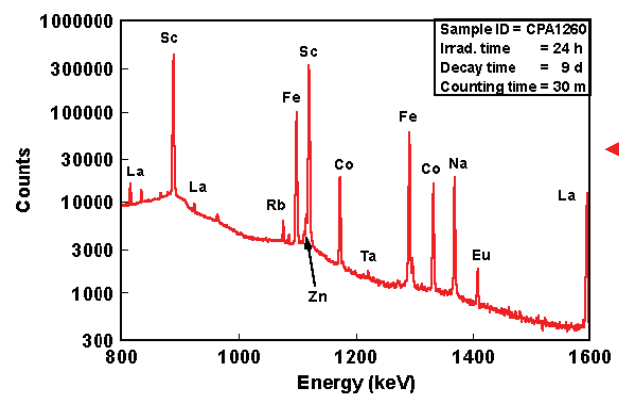
## 2. Irradiation






## 3. Gamma ray counting



## 4. Spectrum analysis



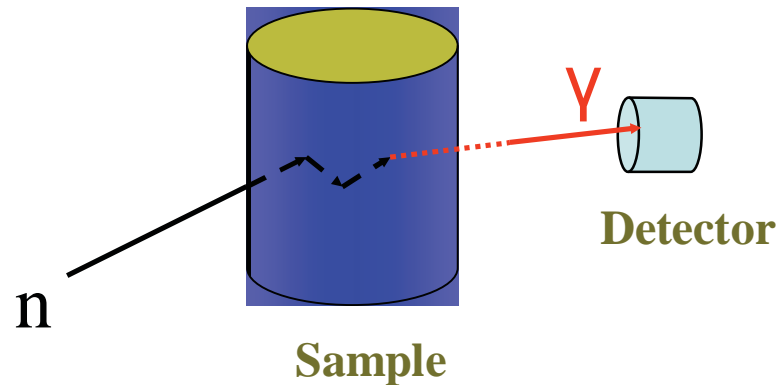
# Analytical features/Advantages of NAA

Legend:  
 ICP-OES  
 NAA  
 ICP-MS

1 1 <b>H</b> 1.0079												13 5 <b>B</b> 10.811	14 6 <b>C</b> 12.011	15 7 <b>N</b> 14.007	15 8 <b>O</b> 15.999	17 9 <b>F</b> 18.993	18 10 <b>Ne</b> 20.180												
3 3 <b>Li</b> 6.941	4 4 <b>Be</b> 9.0122												13 13 <b>Al</b> 26.982	14 14 <b>Si</b> 28.086	15 15 <b>P</b> 30.974	15 16 <b>S</b> 32.065	17 17 <b>Cl</b> 35.453	18 18 <b>Ar</b> 39.948											
11 11 <b>Na</b>	12 12 <b>Mg</b> 24.305	3	4	5	6	7	8	9	10	11	12	13 19 <b>K</b>	14 20 <b>Ca</b>	15 21 <b>Sc</b>	16 22 <b>Ti</b> 47.867	17 23 <b>V</b> 50.942	18 24 <b>Cr</b>	19 25 <b>Mn</b> 54.938	20 26 <b>Fe</b>	21 27 <b>Co</b>	22 28 <b>Ni</b>	23 29 <b>Cu</b> 63.546	24 30 <b>Zn</b>	25 31 <b>Ga</b> 69.723	26 32 <b>Ge</b> 72.64	27 33 <b>As</b>	28 34 <b>Se</b> 78.96	29 35 <b>Br</b> 79.904	30 36 <b>Kr</b> 83.798
37 37 <b>Rb</b>	38 38 <b>Sr</b> 87.62	39 39 <b>Y</b> 88.906	40 40 <b>Zr</b>	41 41 <b>Nb</b> 92.906	42 42 <b>Mo</b> 95.96	43 43 <b>Tc</b> (98)	44 44 <b>Ru</b> 101.07	45 45 <b>Rh</b> 102.91	46 46 <b>Pd</b> 106.42	47 47 <b>Ag</b> 107.87	48 48 <b>Cd</b> 112.41	49 49 <b>In</b> 114.82	50 50 <b>Sn</b> 118.71	51 51 <b>Sb</b>	52 52 <b>Te</b> 127.60	53 53 <b>I</b> 126.90	54 54 <b>Xe</b> 131.29												
55 55 <b>Cs</b>	56 56 <b>Ba</b>	57-71 * #	72 72 <b>Hf</b>	73 73 <b>Ta</b>	74 74 <b>W</b> 183.84	75 75 <b>Re</b> 186.21	76 76 <b>Os</b> 190.23	77 77 <b>Ir</b> 192.22	78 78 <b>Pt</b> 195.08	79 79 <b>Au</b> 196.97	80 80 <b>Hg</b> 200.59	81 81 <b>Tl</b> 204.38	82 82 <b>Pb</b> 207.2	83 83 <b>Bi</b> 208.98	84 84 <b>Po</b> (209)	85 85 <b>At</b> (210)	86 86 <b>Rn</b> (222)												
87 87 <b>Fr</b> (223)	88 88 <b>Ra</b> (226)	89-103 #	104 104 <b>Rf</b> (261)	105 105 <b>Db</b> (262)	106 106 <b>Sg</b> (266)	107 107 <b>Bh</b> (264)	108 108 <b>Hs</b> (270)	109 109 <b>Mt</b> (268)	110 110 <b>Ds</b> (281)	111 111 <b>Rg</b> (272)	112 112 <b>Uub</b> (285)	113 113 <b>Uut</b> (284)	114 114 <b>Uuq</b> (285)	115 115 <b>Uup</b> (288)	116 116 <b>Uuh</b> (291)	118 118 <b>Uuo</b> (294)													
* Lanthanide series			57 57 <b>La</b>	58 58 <b>Ce</b>	59 59 <b>Pr</b> 140.91	60 60 <b>Nd</b>	61 61 <b>Pm</b> (145)	62 62 <b>Sm</b>	63 63 <b>Eu</b>	64 64 <b>Gd</b> 157.25	65 65 <b>Tb</b>	66 66 <b>Dy</b> 162.50	67 67 <b>Ho</b> 164.93	68 68 <b>Er</b> 167.26	69 69 <b>Tm</b> 168.93	70 70 <b>Yb</b>	71 71 <b>Lu</b>												
# Actinide series			89 89 <b>Ac</b> (227)	90 90 <b>Th</b>	91 91 <b>Pa</b> 231.04	92 92 <b>U</b>	93 93 <b>Np</b> (237)	94 94 <b>Pu</b> (244)	95 95 <b>Am</b> (243)	96 96 <b>Cm</b> (247)	97 97 <b>Bk</b> (247)	98 98 <b>Cf</b> (251)	99 99 <b>Es</b> (252)	100 100 <b>Fm</b> (257)	101 101 <b>Md</b> (258)	102 102 <b>No</b> (259)	103 103 <b>Lr</b> (262)												



# Large Sample NAA

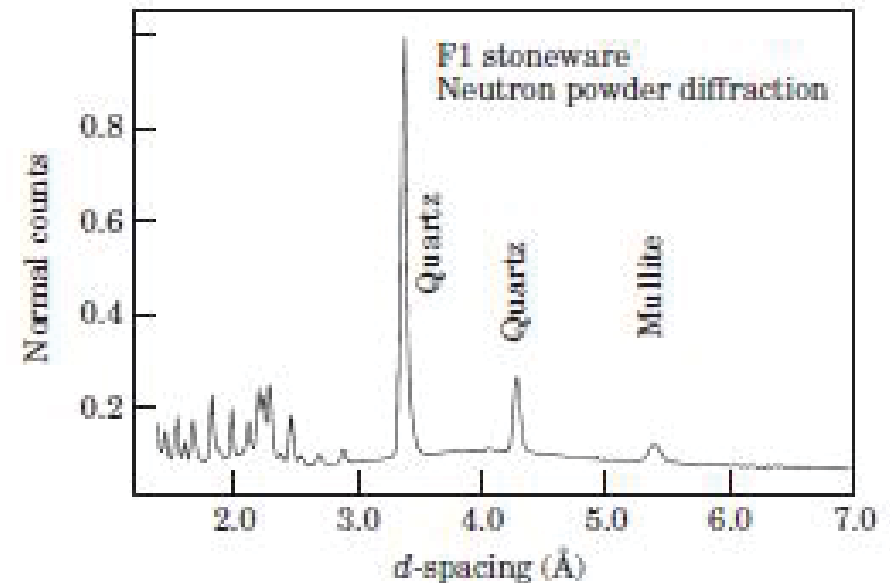
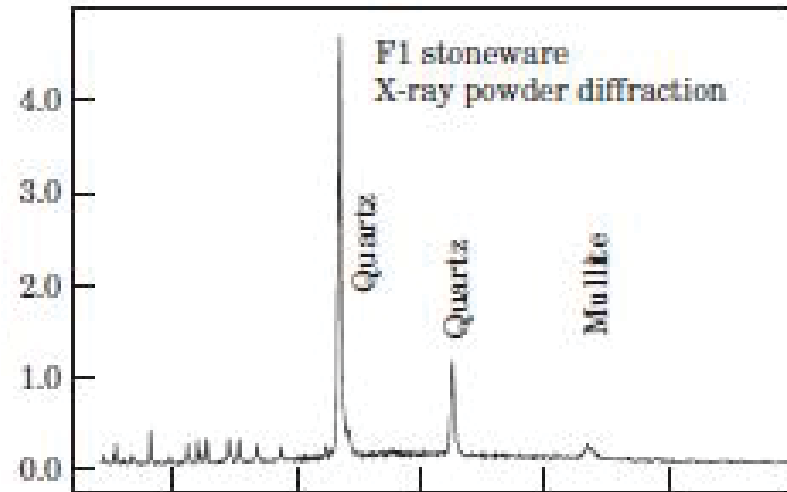
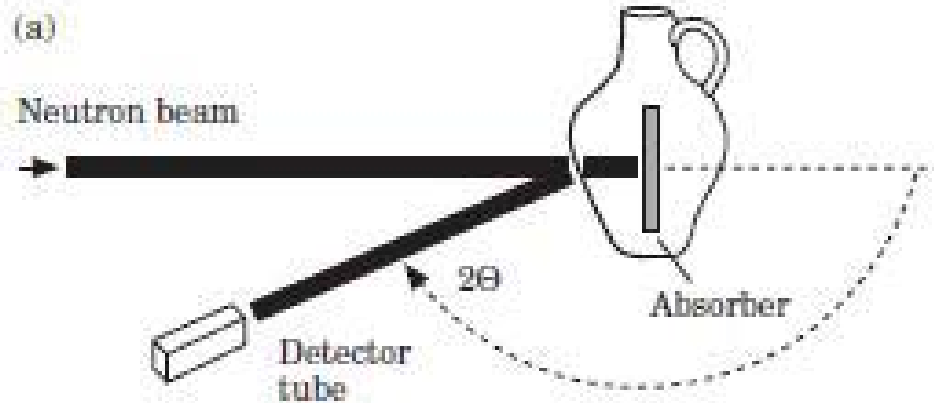


Neutrons and gamma-rays have mean free paths of several centimeters within materials

Neutron Activation an “ideal” tool for Analysis of Large Volume Samples ( $\geq 1$  L)

IAEA CRP on Harmonization of LSNA methodology ( 2009 -), Stamatelatos-Tzika, NCSR “Demokritos”

# Time Of Flight Neutron Diffraction

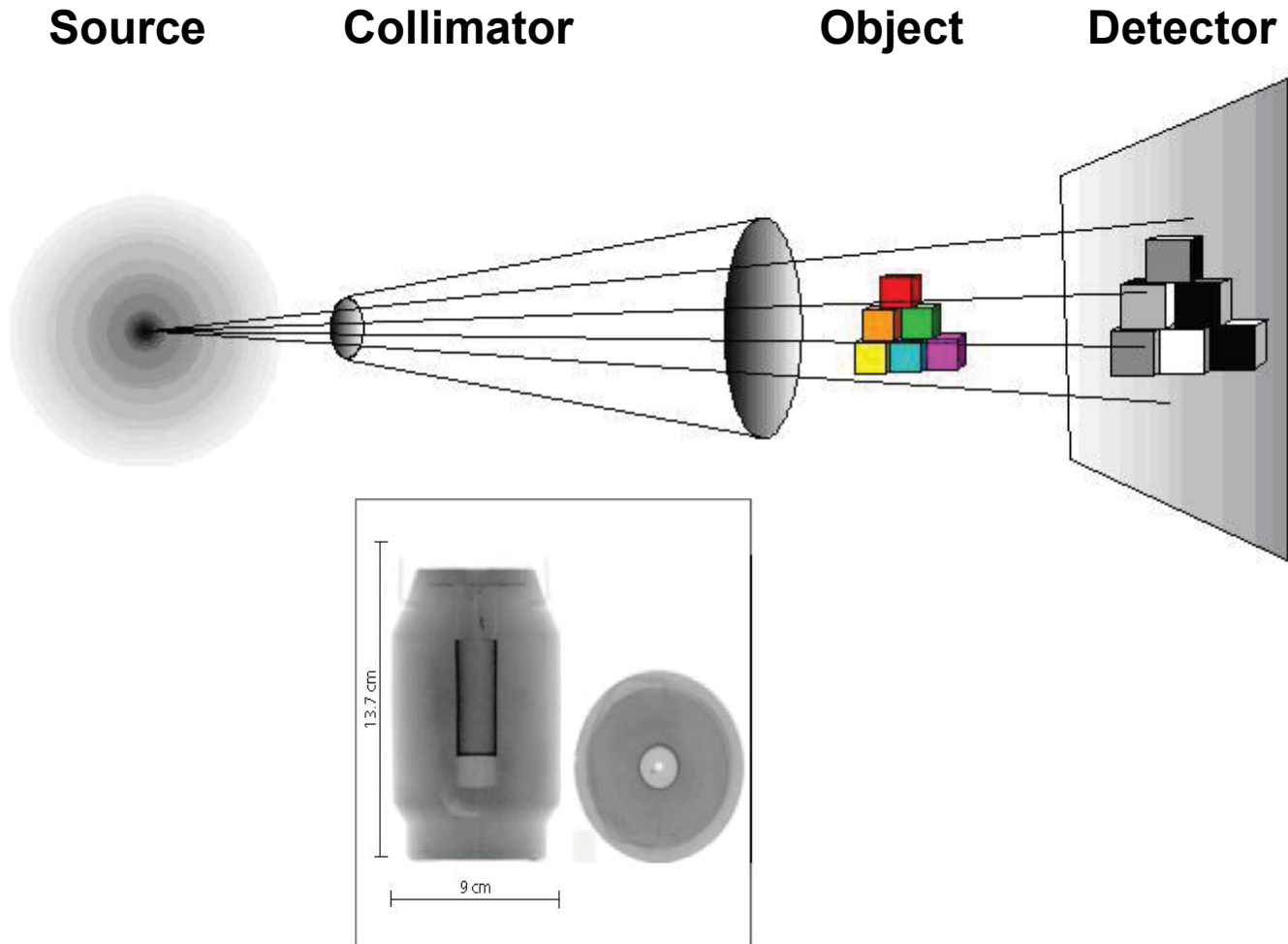


Bragg Equations:

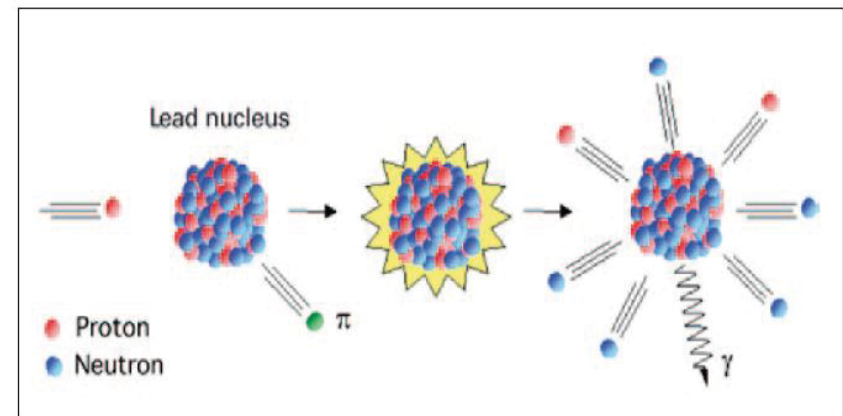
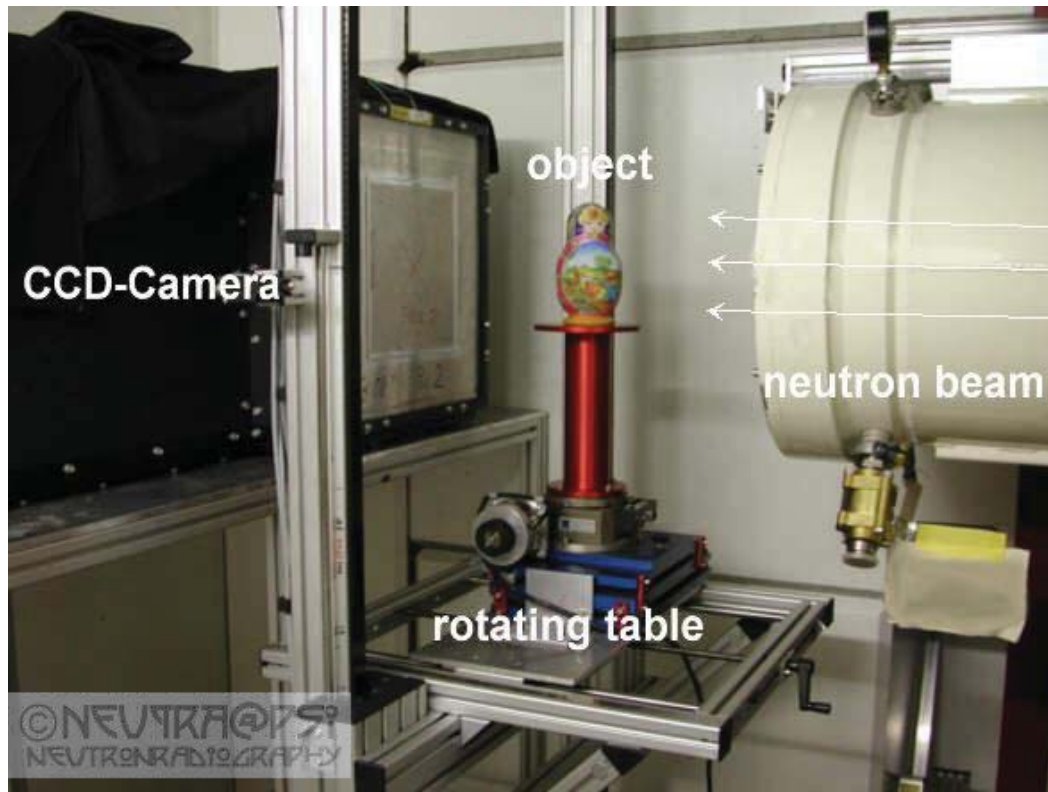
$$\lambda = 2 \cdot d_{hkl} \cdot \sin\Theta$$

$$\text{Time } (\mu\text{s}) = 505.56 \cdot L(\text{meters}) \cdot \frac{1}{d_{hkl}(\text{Angstrom}) \cdot \sin\Theta}$$

# Neutron imaging techniques



# Experimental facility at Paul Scherrer Institut



. **NEUTRA set-up:** Swiss spallation source for research purposes  
Neutrons are produced by a 590 MeV proton beam of 1.3 mA current on Pb target. D<sub>2</sub>O moderator to produce 25 meV thermal neutrons

# Neutrons tomography image

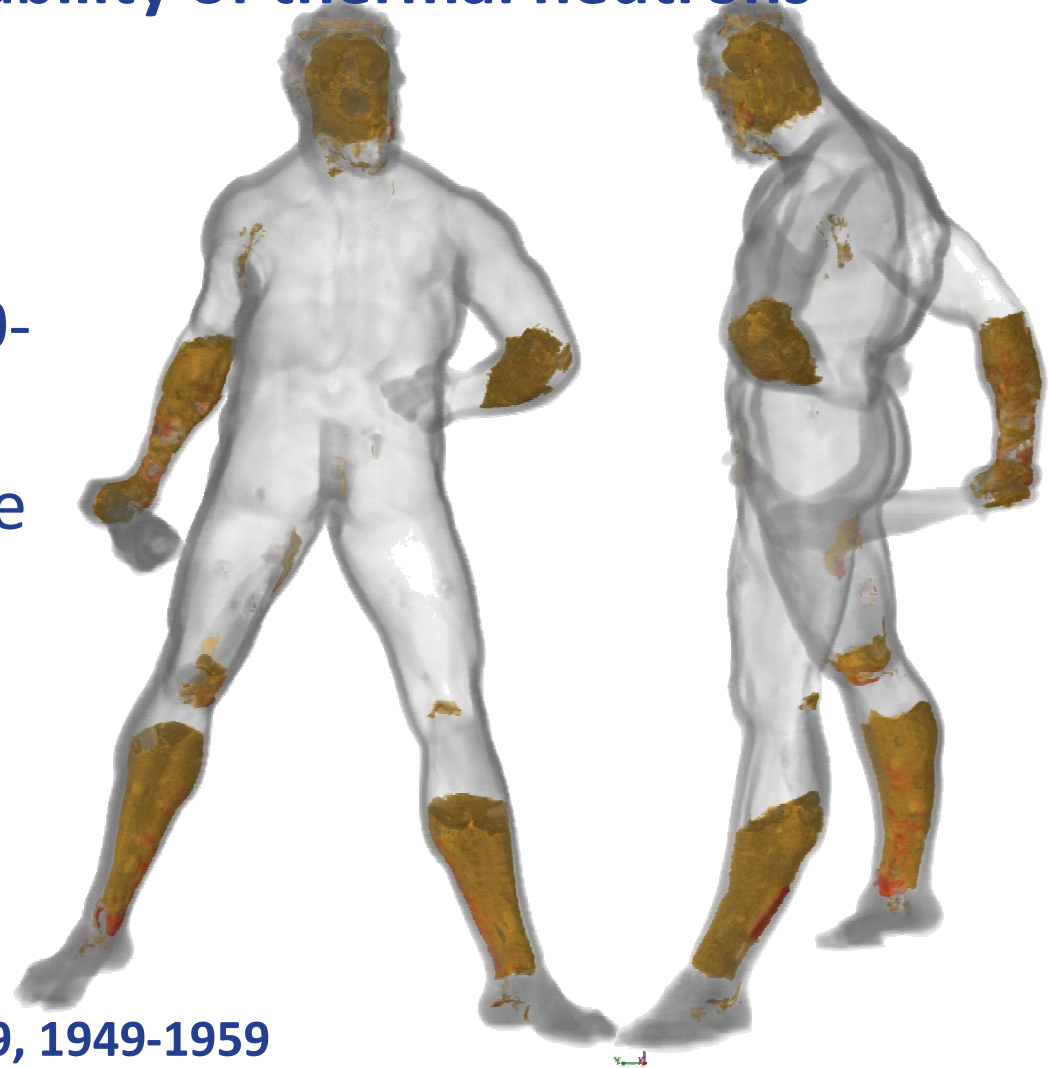
Very high interaction probability of thermal neutrons with H, C and O

Hercules Pomarius,  
Willem van Tetrode's, 1520-1588

Renaissance bronze figurine

Grey color: **Bronze**

Yellow color: Core material  
(**silicate based, clay**)

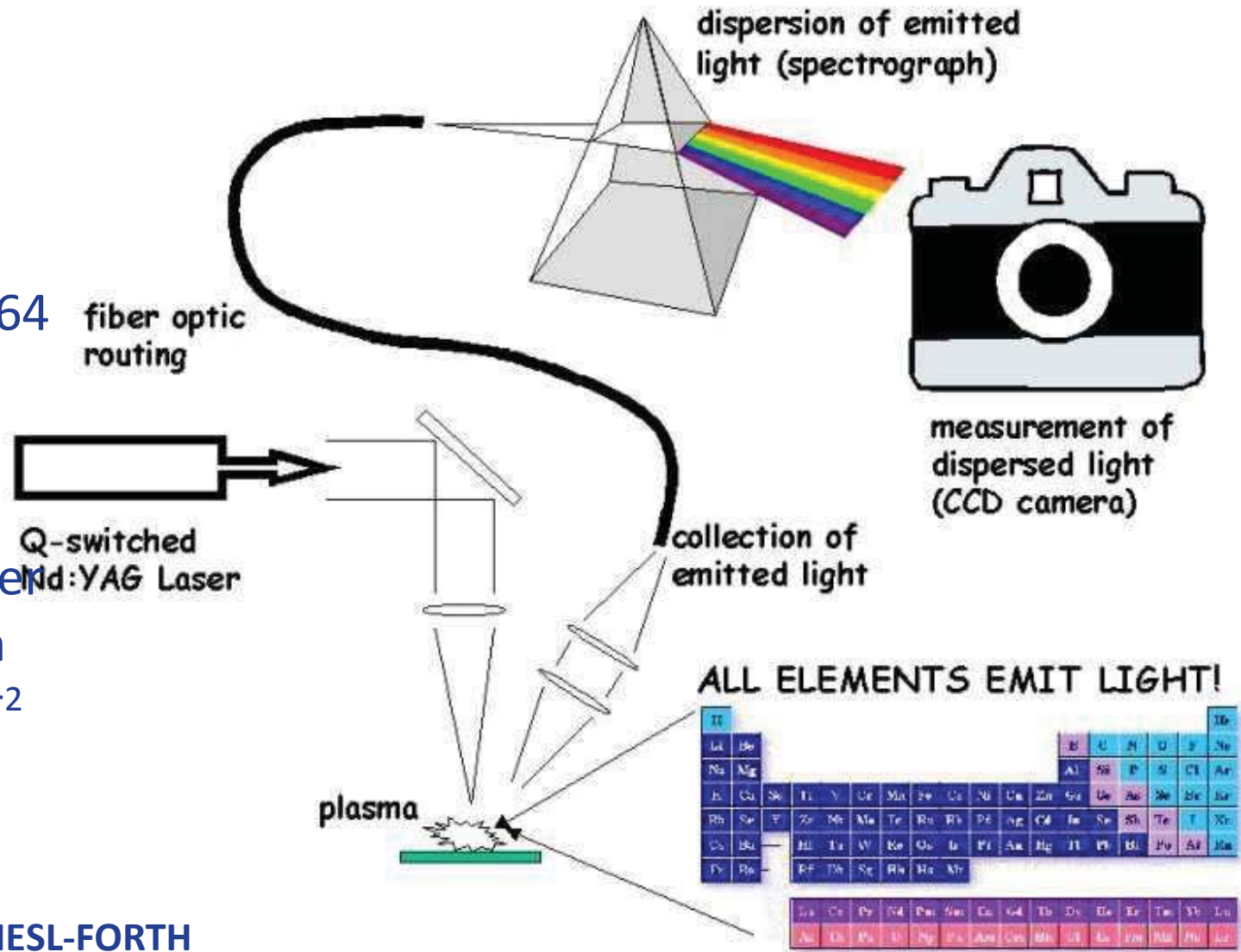


R. Van Lang et al. , ABC, 395, 7, 2009, 1949-1959



# Laser Induced Breakdown Spectroscopy - LIBS

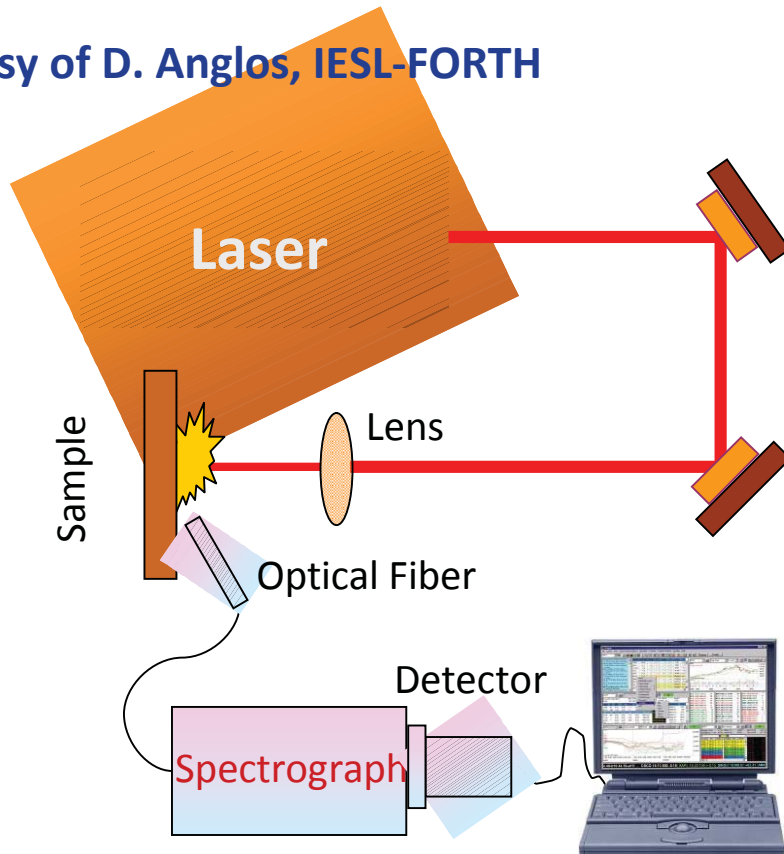
Near infrared region of the electromagnetic spectrum, with a wavelength of 1064 nm. The pulse duration is in the region of 10 ns generating a power density which can exceed  $1 \text{ GW}\cdot\text{cm}^{-2}$  at the focal point



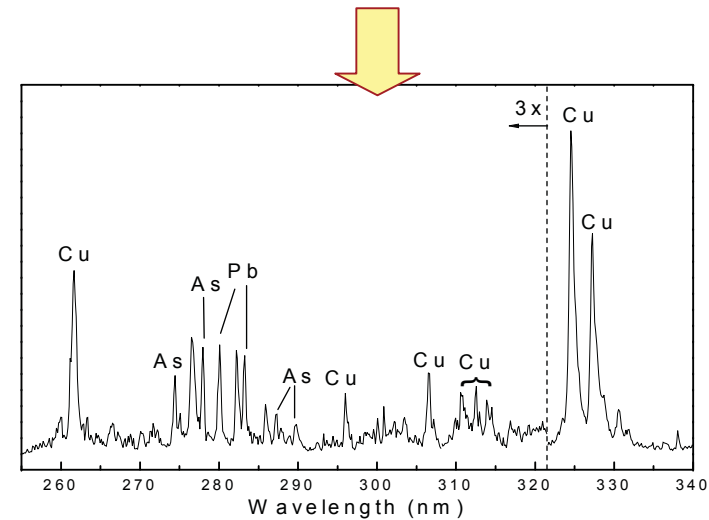
Courtesy of D. Anglos, IESL-FORTH

# Laser Induced Breakdown Spectroscopy - LIBS

Courtesy of D. Anglos, IESL-FORTH



Spectral information from *micro-plasma* generated when a *laser pulse* probes the sample/object surface



Characteristic Spectral Lines ⇒

Which elements?

Line Intensity ⇒

How much of each?

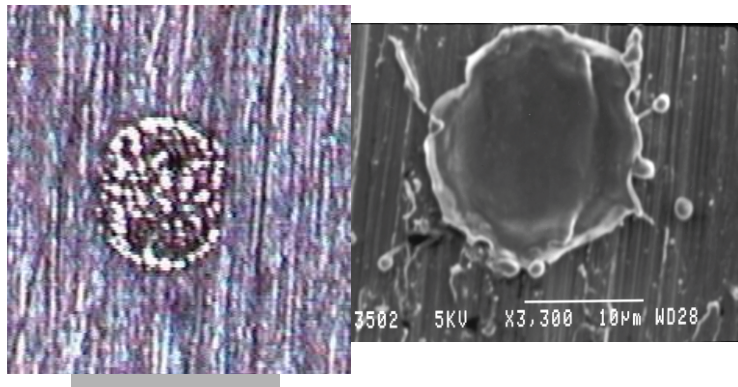
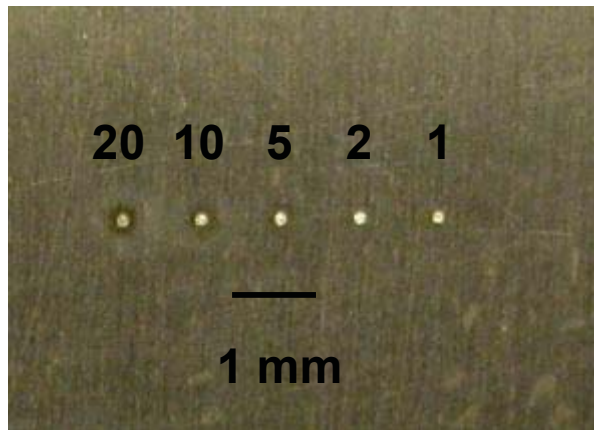
# Laser Induced Breakdown Spectroscopy - LIBS

A periodic table of elements with atomic numbers and symbols. The elements are arranged in rows and columns, with the lanthanide and actinide series shown below the main body of the table. The elements are color-coded: H (1) is red; He (2) is blue; Li (3) is orange; Be (4) is green; B (5) is red; C (6) is blue; N (7) is orange; O (8) is green; F (9) is red; Ne (10) is blue; Na (11) is orange; Mg (12) is green; Al (13) is red; Si (14) is blue; P (15) is orange; S (16) is green; Cl (17) is red; Ar (18) is blue; K (19) is orange; Ca (20) is green; Sc (21) is red; Ti (22) is blue; V (23) is orange; Cr (24) is green; Mn (25) is red; Fe (26) is blue; Co (27) is orange; Ni (28) is green; Cu (29) is red; Zn (30) is blue; Ga (31) is orange; Ge (32) is green; As (33) is red; Se (34) is blue; Br (35) is orange; Kr (36) is green; Rb (37) is red; Sr (38) is blue; Y (39) is orange; Zr (40) is green; Nb (41) is red; Mo (42) is blue; Tc (43) is orange; Ru (44) is green; Rh (45) is red; Pd (46) is blue; Ag (47) is orange; Cd (48) is green; In (49) is red; Sn (50) is blue; Sb (51) is orange; Te (52) is green; I (53) is red; Xe (54) is blue; Cs (55) is orange; Ba (56) is green; La (57) is red; Hf (72) is blue; Ta (73) is orange; W (74) is green; Re (75) is red; Os (76) is blue; Ir (77) is orange; Pt (78) is green; Au (79) is red; Hg (80) is blue; Tl (81) is orange; Pb (82) is green; Bi (83) is red; Po (84) is blue; At (85) is orange; Rn (86) is green; Fr (87) is red; Ra (88) is blue; Ac (89) is orange; Ce (58) is green; Pr (59) is red; Nd (60) is blue; Pm (61) is orange; Sm (62) is green; Eu (63) is red; Gd (64) is blue; Tb (65) is orange; Dy (66) is green; Ho (67) is red; Er (68) is blue; Tm (69) is orange; Yb (70) is green; Lu (71) is red; Th (90) is blue; Pa (91) is orange; U (92) is green; Np (93) is red; Pu (94) is blue; Am (95) is orange; Cm (96) is green; Bk (97) is red; Cf (98) is blue; Es (99) is orange; Fm (100) is green; Md (101) is red; No (102) is blue; Lr (103) is orange.

Courtesy of D. Anglos, IESL-FORTH

# Laser Induced Breakdown Spectroscopy - LIBS

Typical craters on brass



Portable LIBS @  
IESL-FORTH

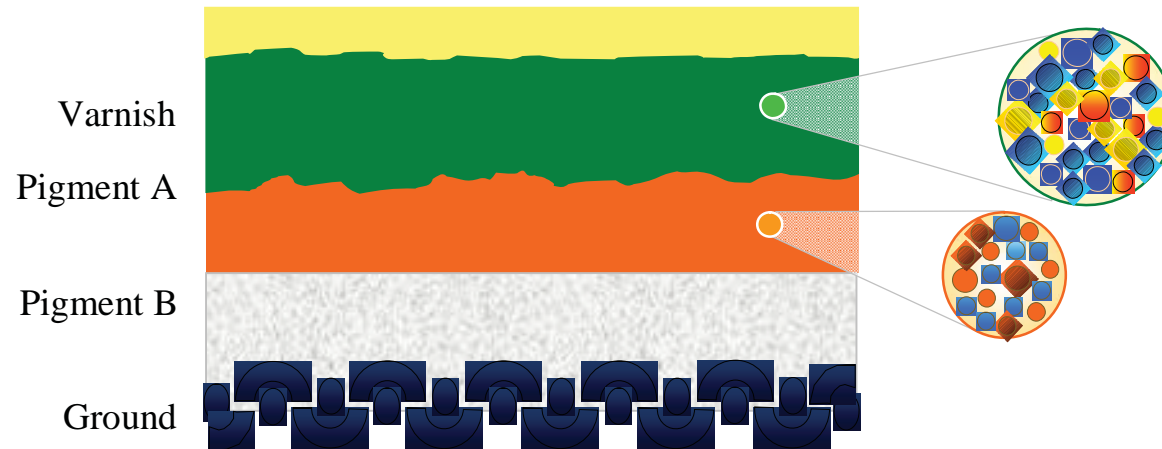


Laser pulse : **10 ns**

Plasma emission: **1 us**

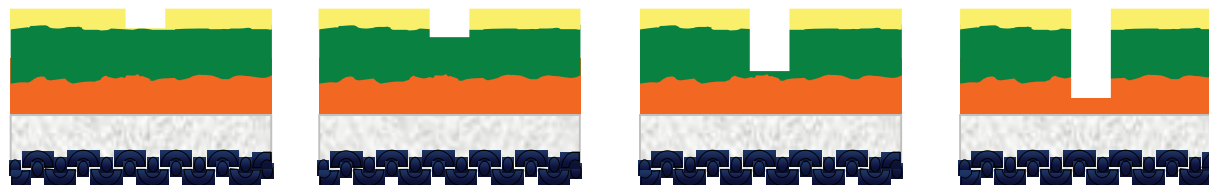
Courtesy of D. Angelos, IESL-FORTH

# Stratigraphic analysis by LIBS



## Stratigraphic analysis

with increasing number of laser pulses



Courtesy of D. Anglos, IESL-FORTH



# Raman spectroscopy



Laser (Visible or near IR) + Matter

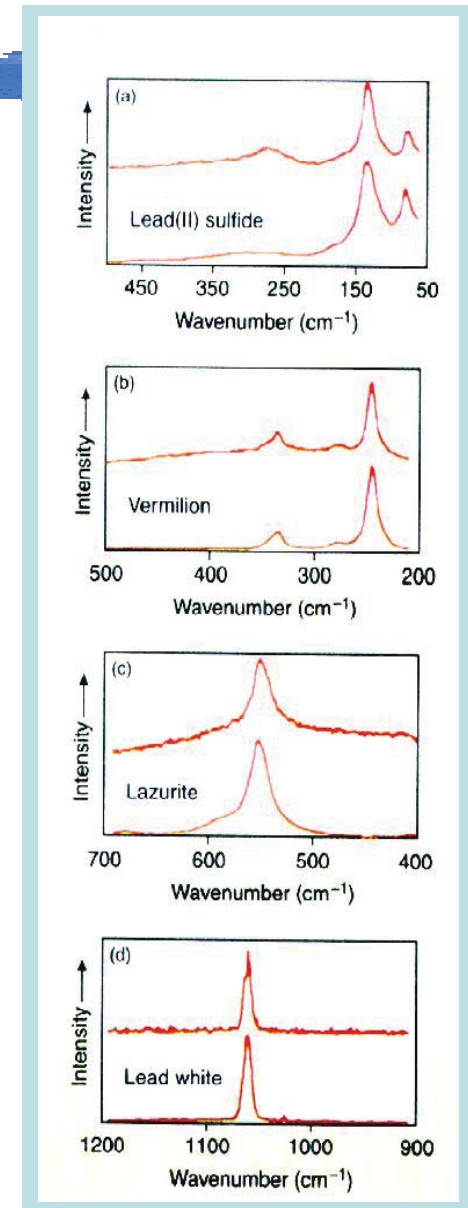
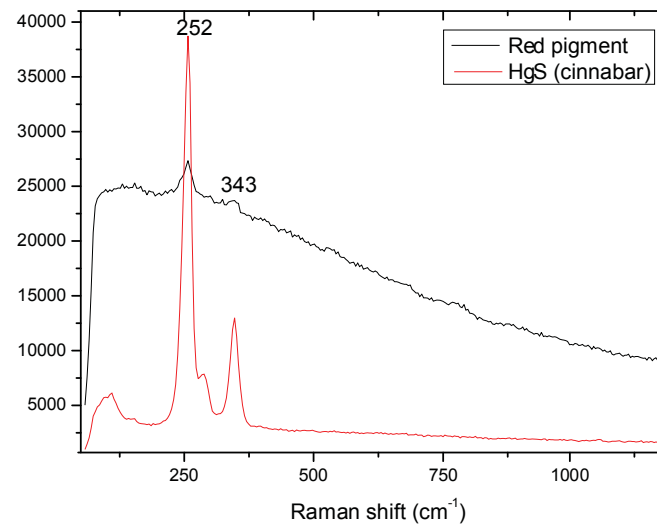
Excitation

Inelastic scattering of light

Raman Spectrum

Chemical information  
(molecular vibrations – chemical bonds)

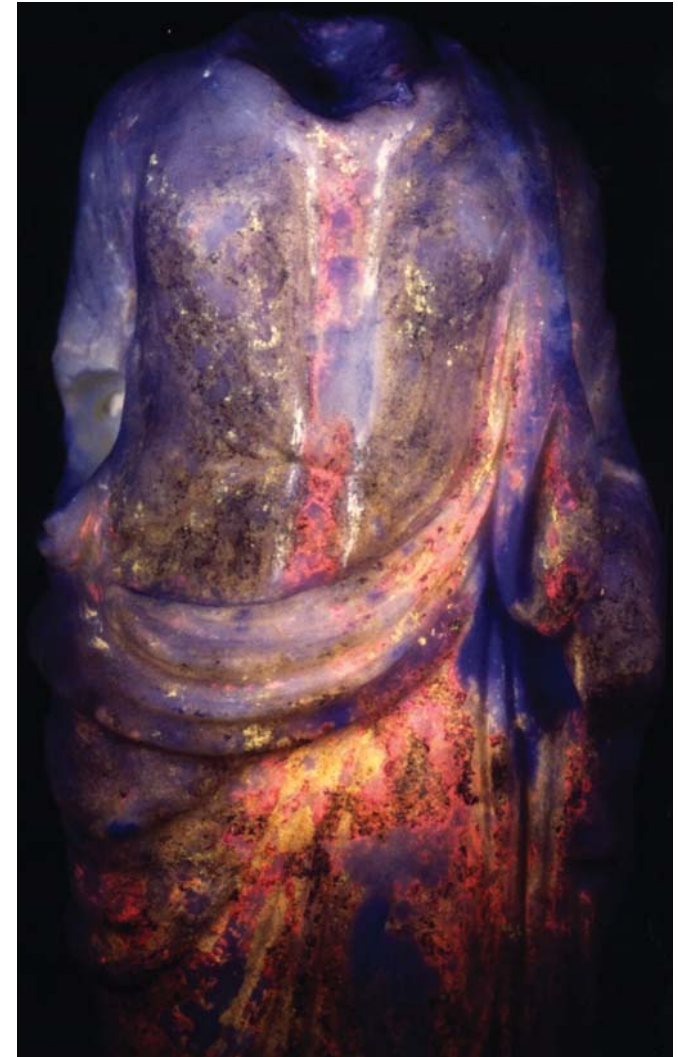
# JY Micro-Raman system



Detecting pigments on Byzantine icon @ **IESL-FORTH**

Courtesy of D. Anglos, IESL-FORTH

# UV Photography, Délos, Musée, Aphrodite





# El-Greco: The Baptism of Christ

- Used techniques before cleaning
  - o CT scan
  - o X-ray radiography
  - o UV, VIS, IR examination with a multi-spectral imaging
  - o Holographic interferometry (1)
  - o Stereoscope
  - o Used XRF (analysis of 30 spots in 4 hours at 15 and 40 KeV)



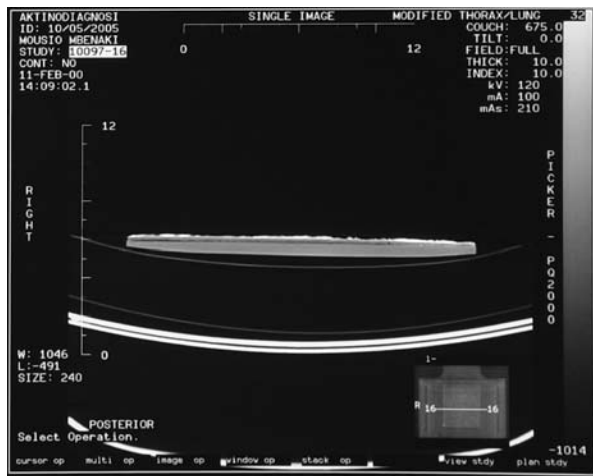
- Used techniques after cleaning
  - o FT n-IR
  - o LIBS
  - o Holographic interferometry (2)

Aloupie et al, Benaki, Journal, 2006



ΕΤΙΣ

# El-Greco: The Baptism of Christ

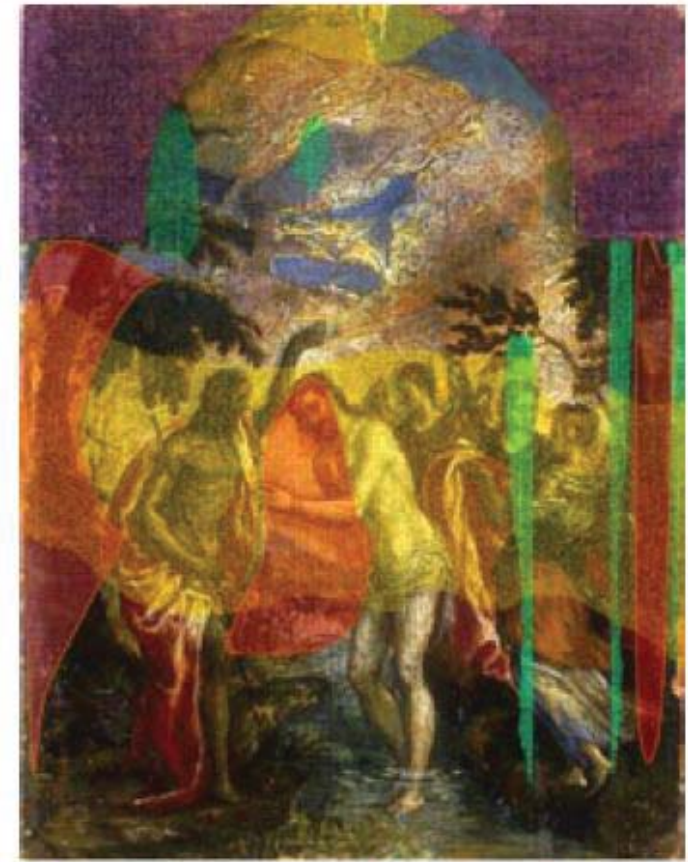


**Computer Aided Tomography (CAT)**

Aloupie et al, Benaki, 2006

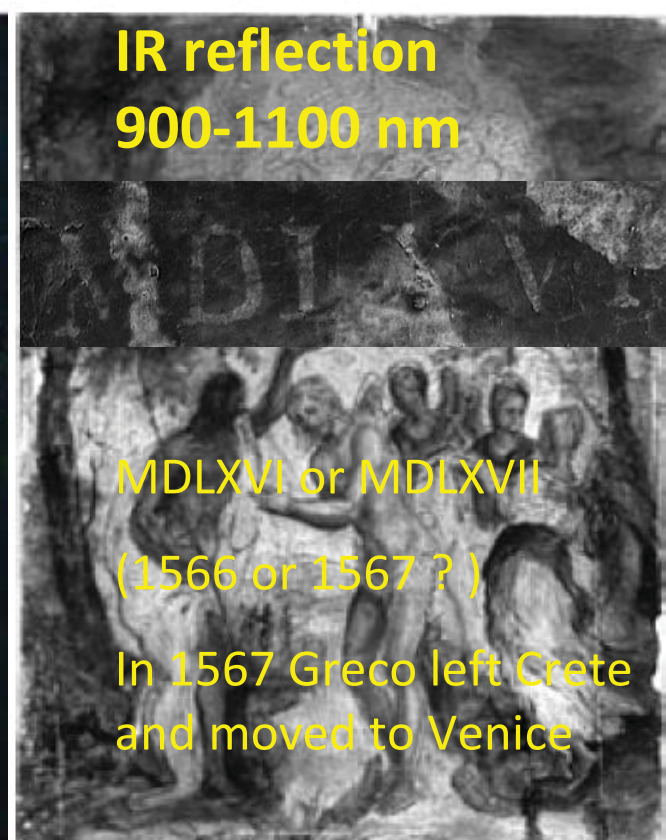


**X-Radiography**



**Digital holographic speckle interferometry by IESL-FORTH**





Practically UV radiation allows the study of the state of preservation of the varnish and the detection/location of interventions on the varnish (cleaning or over-paintings which not fluoresce). Recognition of certain pigments based on their characteristic fluorescence under UV radiation.

Penetrates the varnish and surface paint layers (depending on the pigments). Study of the under-drawing, and changes over-paintings by the painter? Recognition of certain pigments based on their characteristic reflectance in the IR.

# Paintings: UV fluorescence for varnishes

*14th C. Italian panel painting with varying levels of repainting and varnishing, as evident from UV examination using a wood lamp (emission at approximately 360 nm)*



Different varnishes can be discriminated on the basis of their emission spectra even with the same wavelength of excitation

De la Rie, *Studies in Conservation*, 1982 & M. Thoury et al, *Applied Spectroscopy*, 2008

# Conclusions

- ❑ The complexity of cultural heritage related materials due to their heterogeneity at different scale of magnitude and diversity of contained materials requires the optimized synergistic application of different analytical techniques based on radiation probes
- ❑ The big advantage of radiation probes in Ch is that in several cases the analysis and full characterization can be conducted in a fully non-invasive way and some times in-situ



**Thank you for your Attention!!!**