



**The Abdus Salam
International Centre for Theoretical Physics**

The International Union of Geodesy and
Geophysics



2339-12

Workshop on Atmospheric Deposition: Processes and Environmental Impacts

21 - 25 May 2012

Deposition Measurements: Sampling and Analysis

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Deposition Measurements: Sampling and Analysis

ICTP 22 may 2012

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Summary

Deposition sampling:

- Constraints

- Dry deposition

- Wet deposition

Sample analysis:

- Overview of techniques used

- Quality Assurance/ Quality control

Sampling constraints on sites: local positioning

Naturally vegetated, level area is preferred

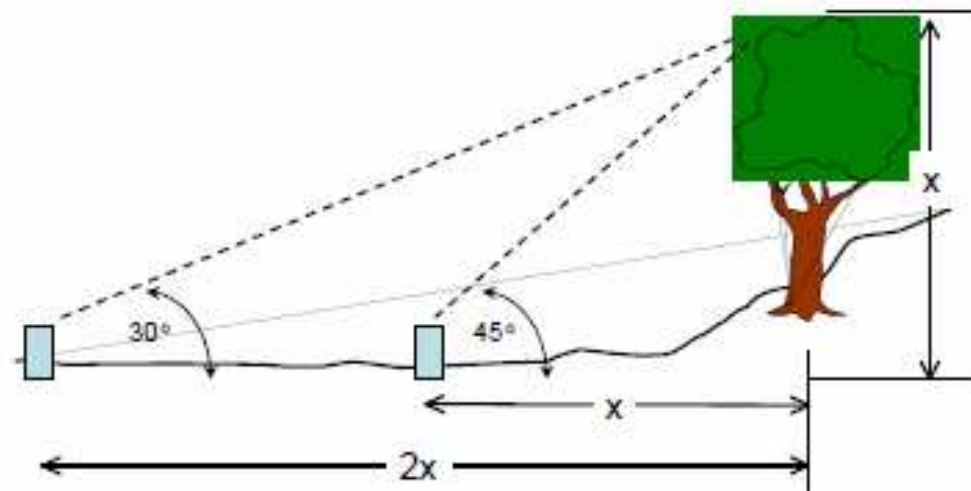
Maintain the height of vegetation less than 0.5 meters

Ground cover should surround the samplers for a distance of 30 meters

Structures projection angles:

30 degrees optimal, 45 degrees maximum

30 metres minimum away of dwellings



Sampling constraints on sites global positioning

Table 2.1: Minimum-Distance Guidelines for GAW Precipitation Chemistry Stations.

Potential Interference	Minimum Distance to Site (km)		Examples, Notes and Local Considerations
	Global	Regional	
SO ₂ or NO _x Point Source >100 tonnes per year >1000 tonnes per year	50 100	20 50	If emission sources (such as power plants, refineries, chemical plants, smelters or other major industrial facilities) are located in the general upwind direction from the collector, then the regional distances indicated should be doubled.
Major Industrial Complex	150	50	
Town, population 1,000-10,000	25	10	Future population growth and associated land development should be considered carefully, especially for towns and villages near a station. If population centres are located in the general upwind direction from the collector, then the regional distances indicated should be doubled.
Town, population 10,000-25,000	50	20	
City, population 25,000-100,000	100	50	
City, population >100,000	200	100	
Major highway, airport, railway, shipping lane, harbour	25	5	Moving sources of pollution, such as air, ground, or water traffic or the medium on which they traverse (e.g., runway, taxiway, road, tracks, or navigable river), should not be within 500 metres of the collector.
Secondary road, heavily travelled	5	1	The local road network around the site is of particular concern. Traffic volume and type as well as road surface will largely determine the impact at the site.
Secondary road, lightly travelled	1	0.5	
Feedlot operations	50	2	Acceptable distances will vary greatly depending on size of the operation. Even small concentrations of animals should be housed no closer than 500 metres. If the feedlot, dairy barn or animal waste pile can be smelled at the collector, it is too close.
Intensive agricultural activities	10	2	Surface storage of agricultural products, fuels, vehicles or other source materials should be kept at least 500 metres from the collector.
Limited agricultural activities	1	0.4	Storage of small amounts of agricultural products, fuels, or other source materials should be kept at least 200 metres from the collector and well sealed.
Parking lot or large paved area	0.5	0.2	On-site parking lots and maintenance yards also need to be kept at least 300 metres from the collector.
Building with fuel combustion	1	0.4	
Sewage treatment plant	20	2	
Active volcano, fumarole, etc.	100	20	Geothermal sites including geysers and springs may have significant emissions and should be avoided.
Natural salt, dust, alkali sources	2	2	Windswept materials from salt and alkali flats as well as sea spray from coastlines can contaminate samples.
Tree line, building	0.05	0.05	

IDAF station : DJOUGOU (BENIN)

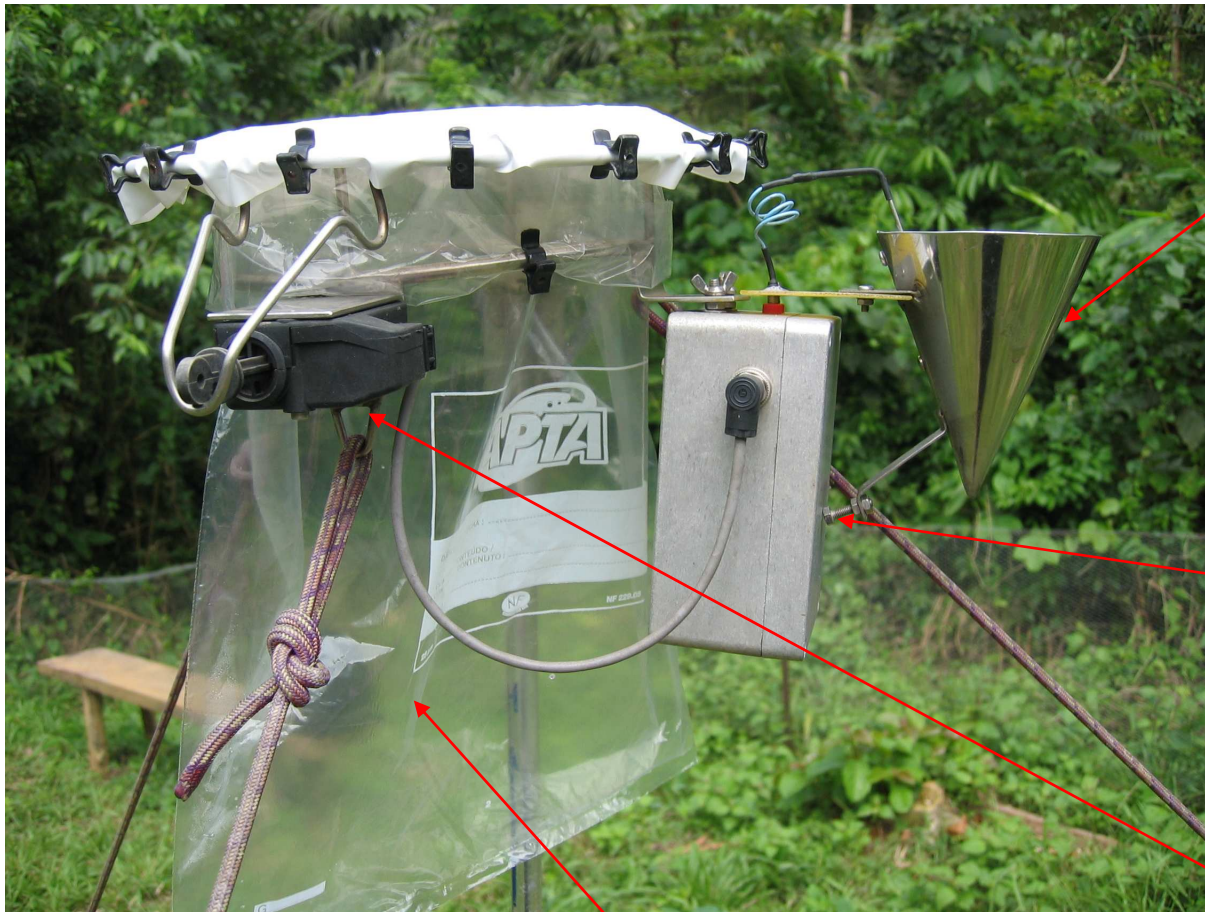
Passive samplers

Rain sampler



Aerosols sampler

Wet deposition: IDAF rain collector



Rain detector

Power supply

Automatic opening
system

Single-use plastic bag

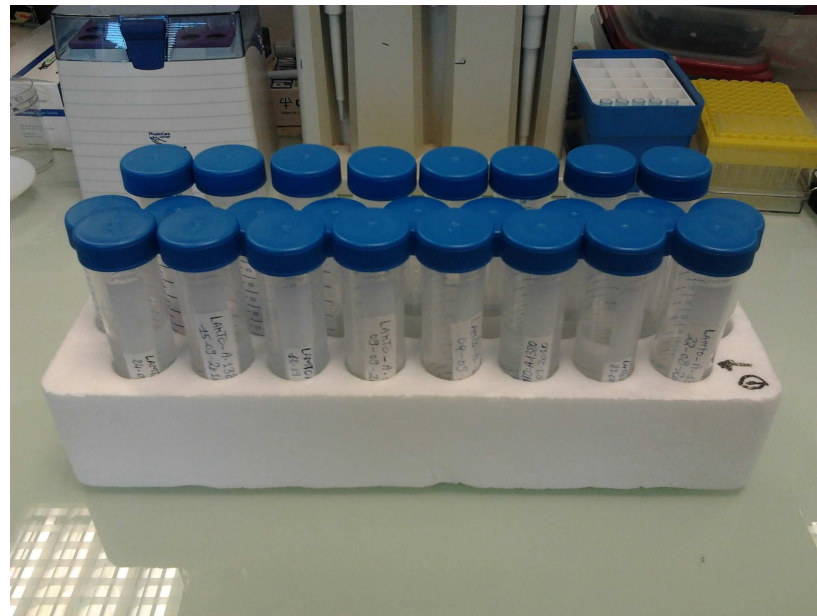
Wet deposition

Rain samples storage: GREINER centrifugation tubes (50 mL) filled and labelled after each rainy event

Rain samples preservation:

Best: freezing in site to -18°C and kept frozen until an alysis

Alternative: adding thymol or formol to avoid bacterial activity



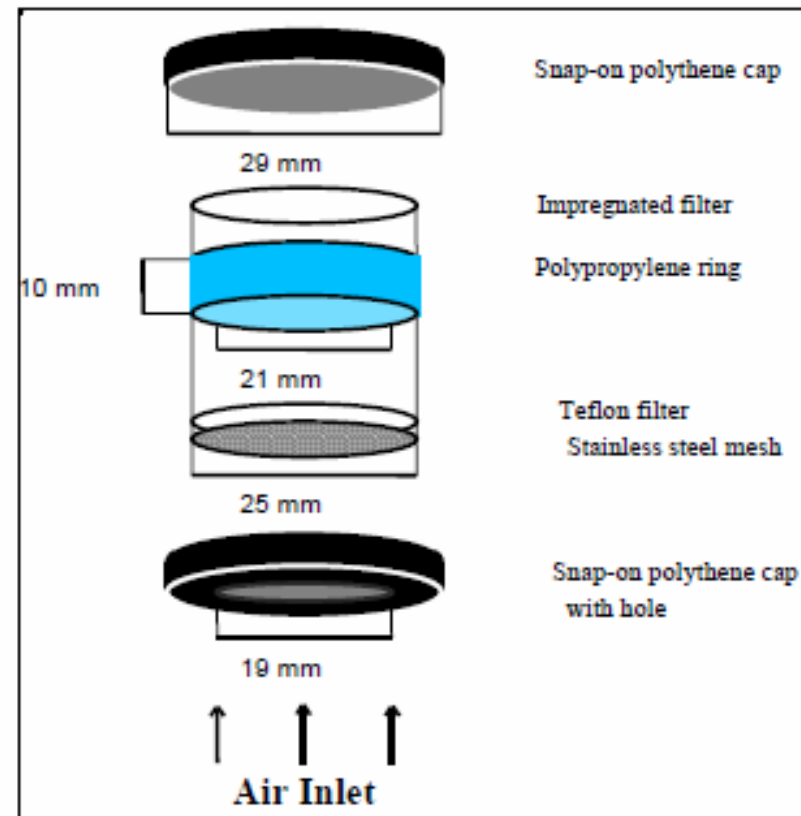
Dry deposition: gases by passive samplers

IDAF Passive samplers holders:
8 samplers exposed together



Dry deposition: gazes by passive samplers

IDAF sampler: developed in
the Laboratoire d'Aérodologie
Based on M. Ferm work
(IVL 1991, 1994)



Passive samplers

2 weeks or one month exposure, the impregnated paper filters react with gazes present in ambient air and form species which can be measured in water by ion chromatography

IDAF samplers are colored to be easily identified ,they are exposed by duplicates to ensure precision and reproducibility

Shipped to the IDAF stations by mail; stored in fridge (4°C before and after exposure)

Gas (Colour of the sampler)	Coating solution	Chemical reaction on the filter
HNO ₃ and SO ₂ (Black)	0.5 g NaOH in 50 mL methanol (pH>12)	$\text{HNO}_3(\text{g}) + \text{OH}^- \rightarrow \text{NO}_3^- + \text{H}_2\text{O}$ $2\text{SO}_2(\text{g}) + 4\text{OH}^- + \text{O}_2 \rightarrow 2\text{H}_2\text{O} + 2\text{SO}_4^{2-}$
NO ₂ (Grey)	0.44 g NaOH + 3.95 g NaI in 50 mL methanol (pH>12)	$2\text{NO}_2(\text{g}) + 3\text{I}^- \rightarrow 2\text{NO}_2^- + \text{I}_3^-$
NH ₃ (White)	1.0 g citric acid in 50 mL methanol	$\text{NH}_3(\text{g}) + \text{H}^+ \rightarrow \text{NH}_4^+$
O ₃ (Grey & black)	0.25 g NaNO ₂ + 0.25 g K ₂ CO ₃ + 0.5 ml redistilled glycerol in 50 ml water	$\text{O}_3(\text{g}) + \text{NO}_2^- \rightarrow \text{NO}_3^- + \text{O}_2$

Passive samplers

Calculation of gases concentration: Fick's first law of diffusion

$$C_{\text{avg}}(\text{ppb}) = [(L/A).X.R.T] / (t.D.P)$$

L/A : depends of the sampler's physical characteristics (m^{-1})

X : amount of gas pollutant trapped on the filter (μmol)

R : ideal gaz constant ($\text{atm K}^{-1} \text{mol}^{-1}$)

t : exposure time (s)

P : mean atmospheric pressure during exposure (atm)

T : mean temperature during exposure (K)

D : diffusion coefficient of the wanted gas in air ($\text{m}^2 \text{s}^{-1}$)

Passive samplers

Easy to use, no electricity supply needed

Good reproducibility (10-20 %)

Mean gas concentrations values during the exposition period

Good agreement active/passive measurements for NO_2 , SO_2 and O_3

Dry Deposition: gazes

measurements by Relaxed Eddy Accumulation (REA)

Air mass selection according to the sign of the vertical speed (+ or -)

Air accumulation in 2 reservoirs (10 min)

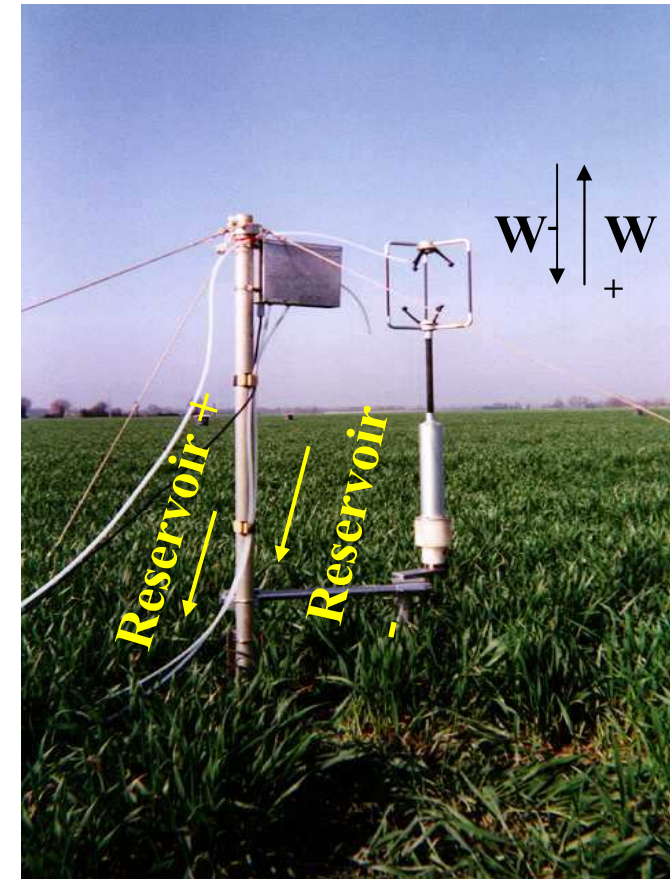
Reservoirs air analysis (2 min)

Reservoirs emptying (3 min)

Flow is proportionnal to the concentration difference between the 2 reservoirs and to the standard deviation of the vertical speed (σ_w)



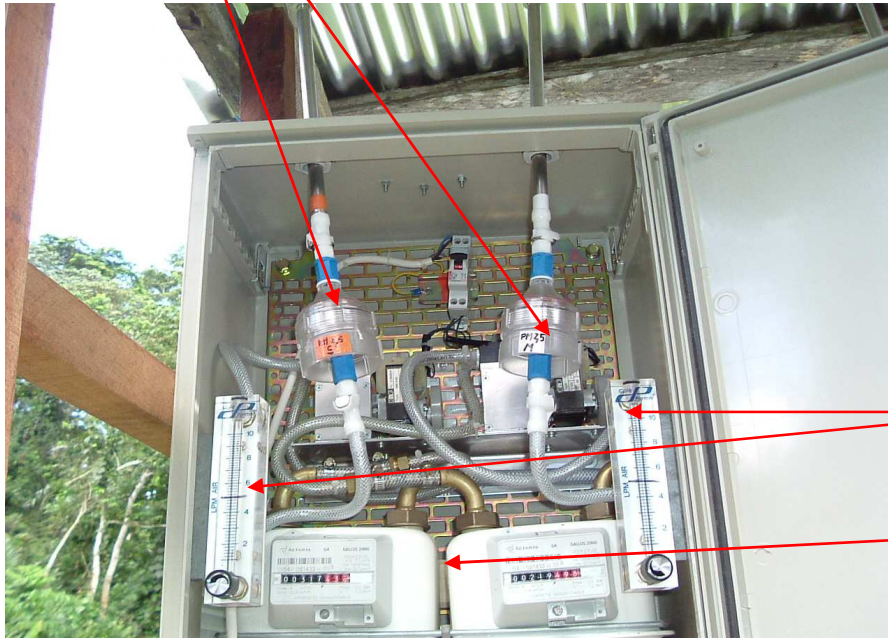
$$F_c = \beta \times \sigma_w \times (c^+ - c^-)$$



Dry deposition: aerosols

IDAF Sampling system

Filters holders (pumps are behind)



PM sampling inlet

Flow meters

Gas counters

Dry deposition: aerosols

Filter holders NILU for 47 mm filters

Inline: size segregation

Open face: total



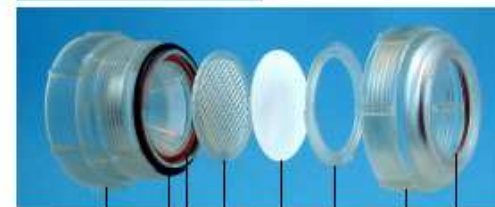
Type: One stage inline 9631
Diameter: 70 mm
Length: 90 mm
(The filter shown is not included)



Type: One stage open face 9633
Diameter: 70 mm
Length: 56 mm
(The filter shown is not included)



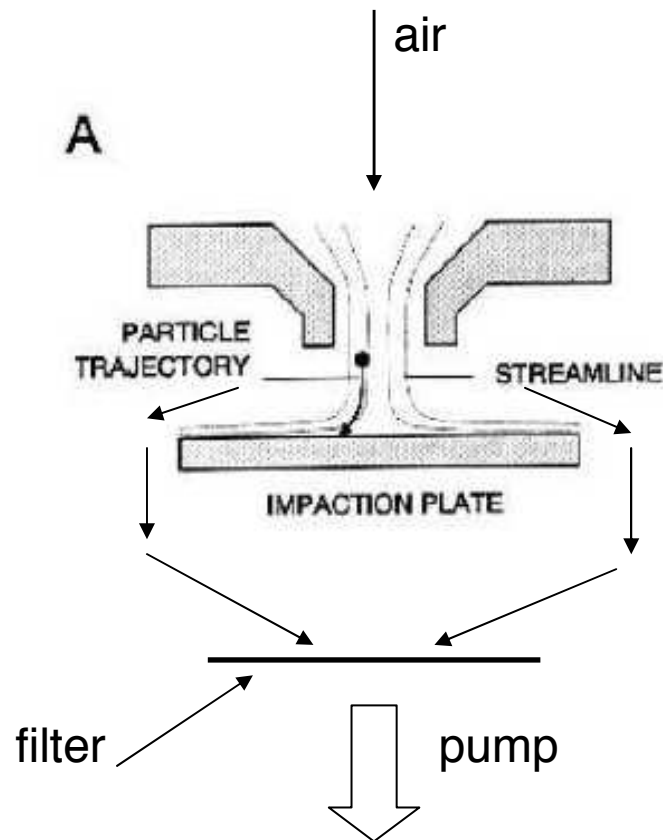
Outlet section 9656
O-ring nitrile 9652
O-ring silicone 9651
Backing 40 mm 9658
O-ring silicone 9651
Inlet inline 9657
Clamping ring 40 mm 9659
Filter 47 mm 9665



Outlet section 9656
O-ring nitrile 9652
O-ring silicone 9651
Backing 40 mm 9658
O-ring silicone 9651
Inlet open face 9655
Clamping ring 40 mm 9659
Filter 47mm 9665

Dry deposition: aerosols

Size segregation: PM 2.5 and PM 10



47 mm filters used in NILU holders

Ions: 0.5 μ m Teflon Zefluor from Pall

Carbon: quartz QMA from Wathman

One week sampling period in sequential mode (15 min/h for example)

4 filters collected together (PM 2,5 and PM 10 ; ions and carbon)

Dry deposition: aerosols

Electrical Low Pressure Impactor: ELPI (DEKATI)

used for intensive measurement periods : real time particle size distribution in the size range 7 nm-10 μm (13 stages cascade impactor)

Cascade impactor



Graphic user interface



Dry deposition: aerosols

13 stages cascade impactor (can be used alone with one pump)

Stage	D50% [μm]
13	10
12	6,8
11	4,4
10	2,5
9	1,6
8	1
7	0,65
6	0,4
5	0,26
4	0,17
3	0,108
2	0,06
1	0,03
Filter stage	0,007



Each impaction stage can be charged with 25 mm filters for chemical analysis

Sample analysis back to lab

- Water system
- pH and conductivity
- Gravimetric analysis
- Ion Chromatography
- Thermal Optical analysis
- Thermal analysis
- ICP/ MS

Water system

Type I ultra pure water:
(18.2 M Ω .cm; TOC < 5 ppb)

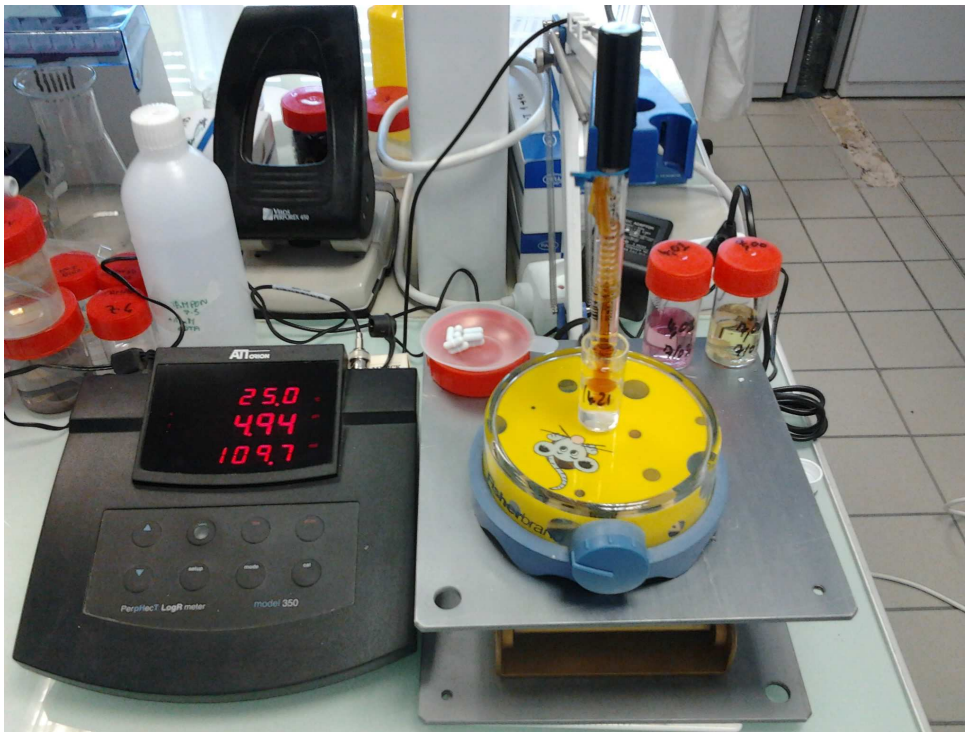
ELGA system

Eluent preparation,
standards dilution,
glassware cleaning and
samples soaking



pH for rain samples

Thermo Orion pH meter and electrode



Temperature compensated (25°C)

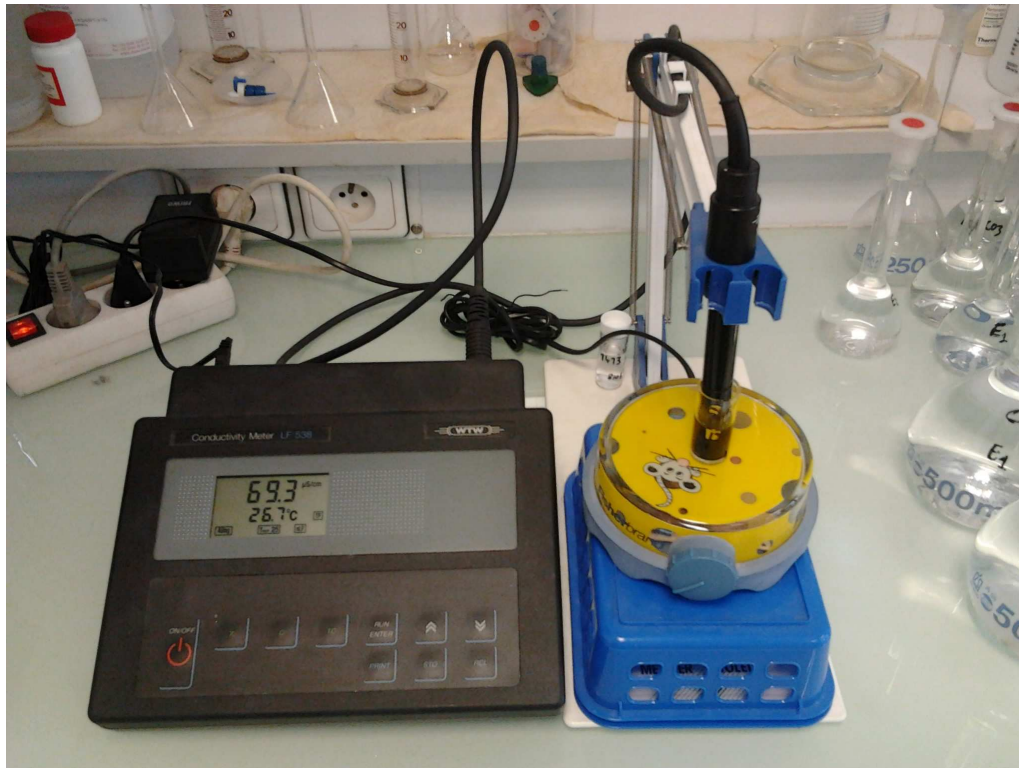
2 points calibration: 4,01 and 7,00

Match with majority of rain samples

Precision: +/- 0,01 pH

Conductivity for rain samples

WTW conductivity meter and sensor



Temperature compensated (25°C)

2 points calibration: 1413 and 23.8 μS

Precision: +/- 1.5%

Gravimetric analysis

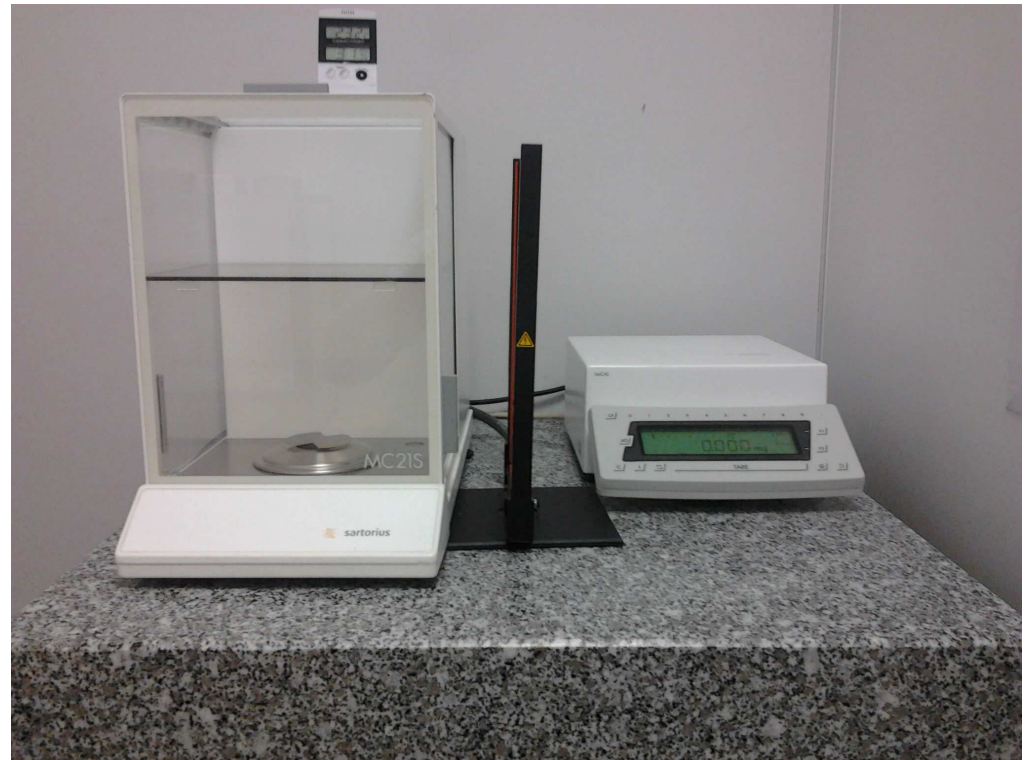
Teflon 47 or 25 mm filters weighted before and after collection

Sartorius MC 21S microbalance

+ Mettler Toledo

Anti-Electrostatic Generator

Precision: $\pm 5 \mu\text{g}$



Ion Chromatography

- Rain samples: direct analysis
- Passive samplers: impregnated filters soaked in ultra pure water
- Aerosols samples: Teflon filters soaked in ultra pure water; 0.2 μm filtration if necessary. Only the soluble part available

Ion Chromatography

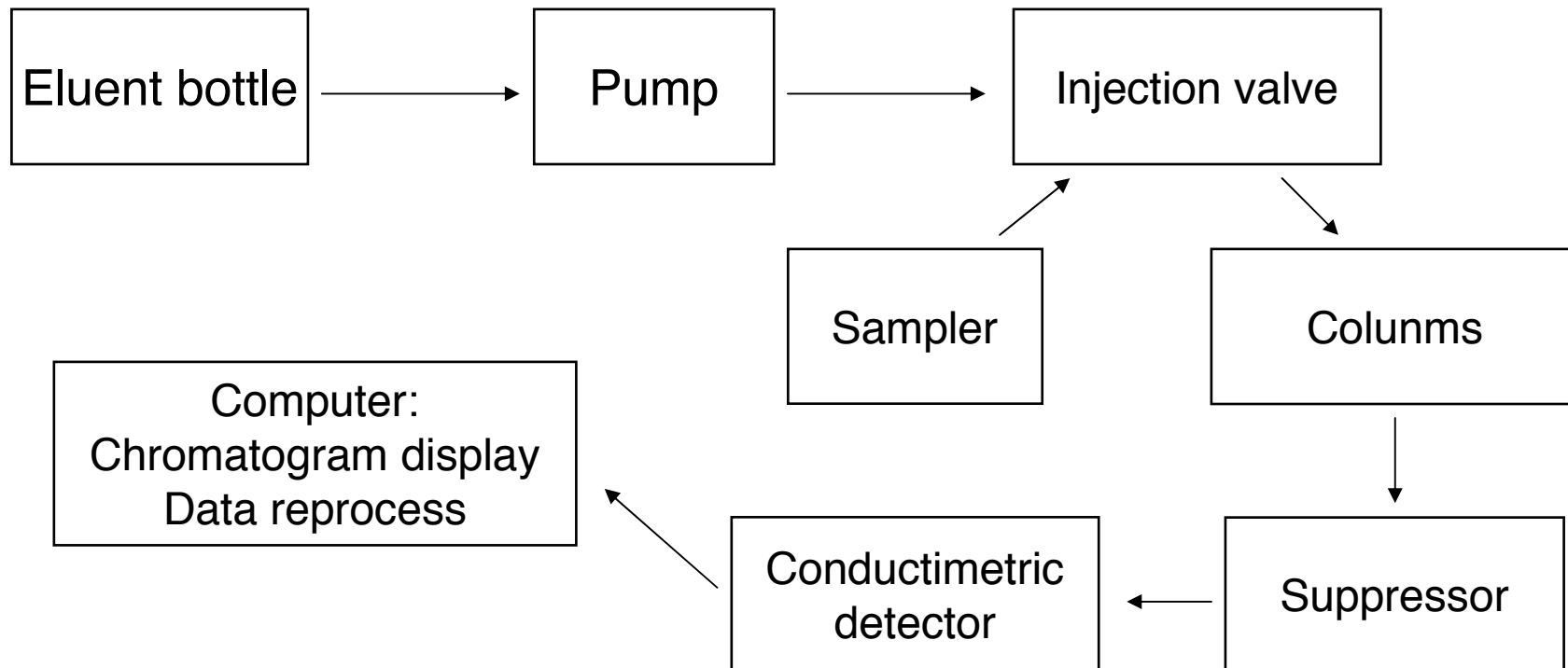
A part of HPLC, our systems from DIONEX

Specificities:

- ion-exchange columns (cations or anions)
- suppressed eluent before detector
- conductivity detector
- calibration by dilution of mono elemental standards
- detection limit: 10 ppb (10 $\mu\text{g/L}$)

Ion Chromatography

Schematic representation:



Ion Chromatography

Cations analysis:
DIONEX ICS 1000
and automated sampler
DIONEX AS50

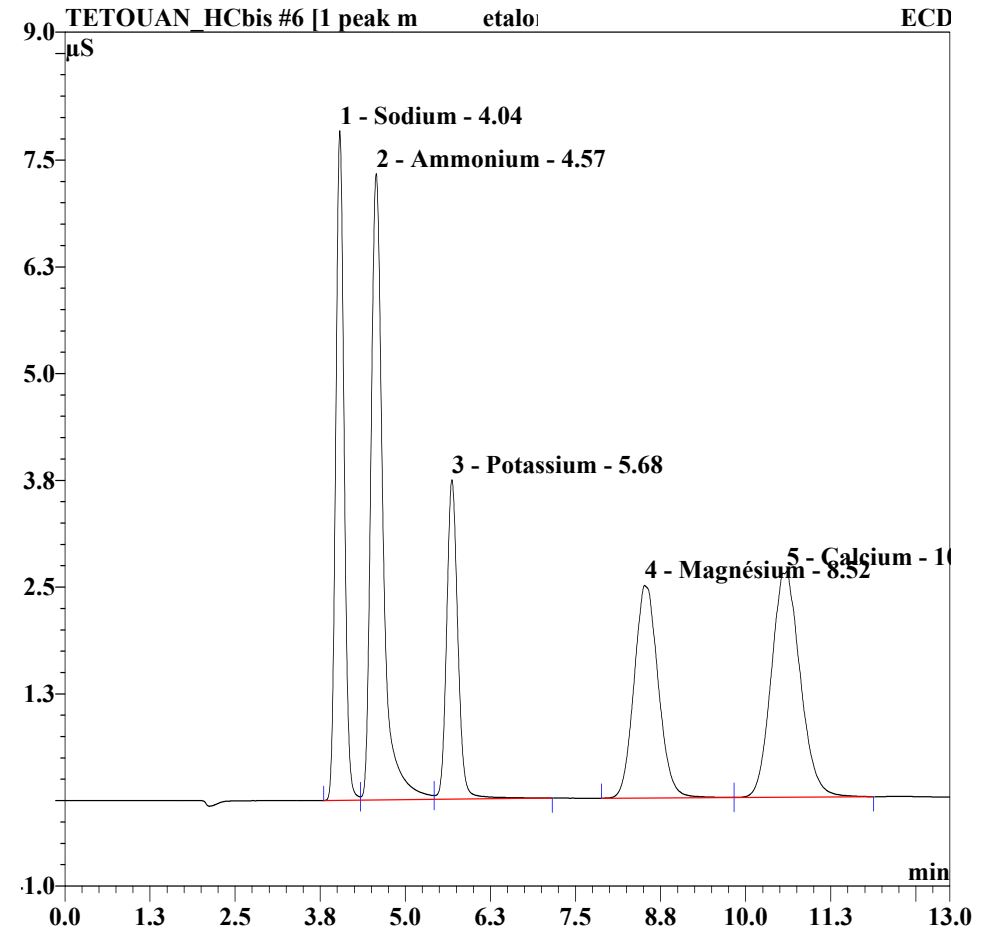


Ion Chromatography

Isocratic mode for majors
inorganic cations:

Columns: CG12A+CS12A

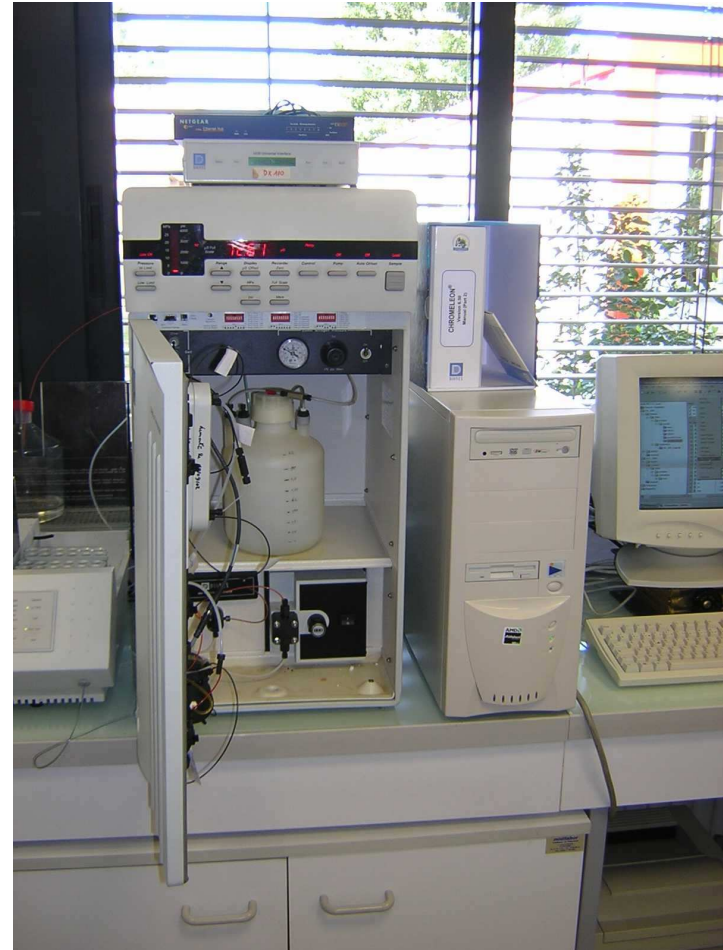
Eluent: 20 mM MSA
(Methanesulfonic acid)



Ion Chromatography

Mineral Anions analysis:
DIONEX DX100
and automated sampler
DIONEX AS 40

Mainly for passive samplers
analysis

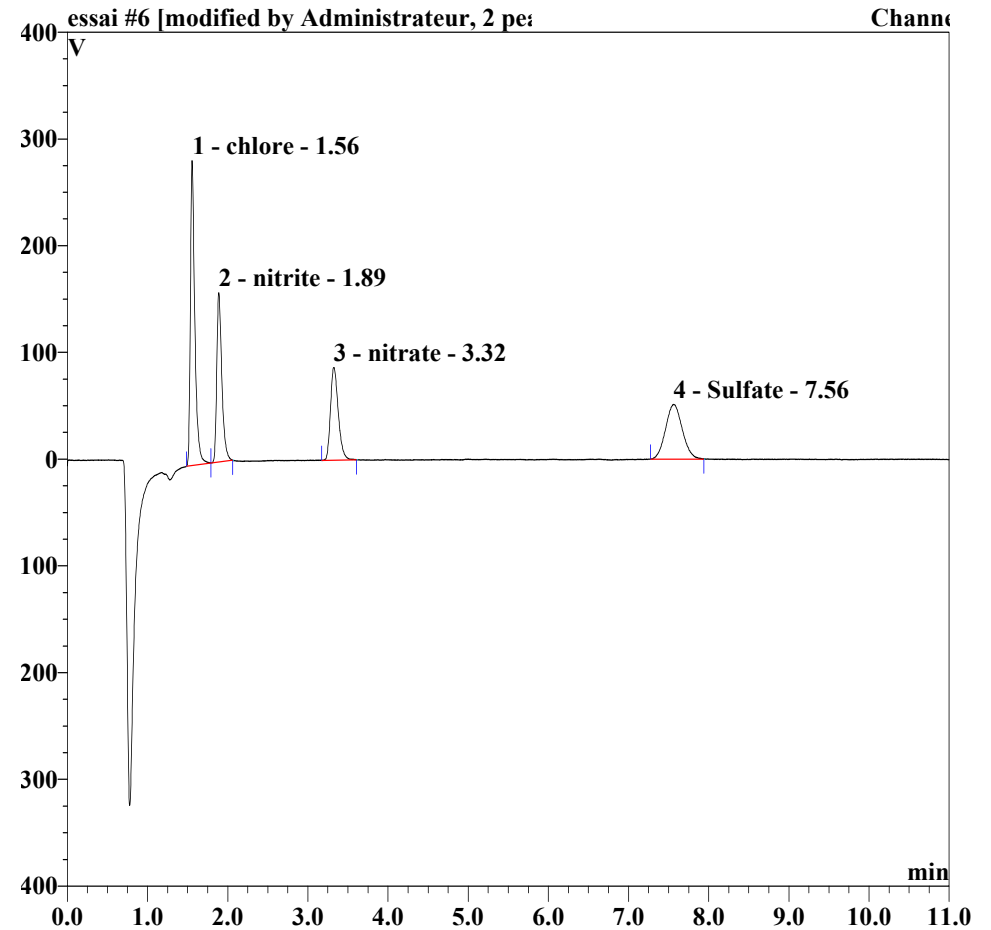


Ion Chromatography

Isocratic mode for
inorganic anions:

Columns: AG4A-SC+
AS4A-SC

Eluent : 1.8mM CO_3^{2-}
/ 1.7 mM HCO_3^-



Ion Chromatography

Anions analysis: DIONEX
DX 500 and automated
sampler AS50

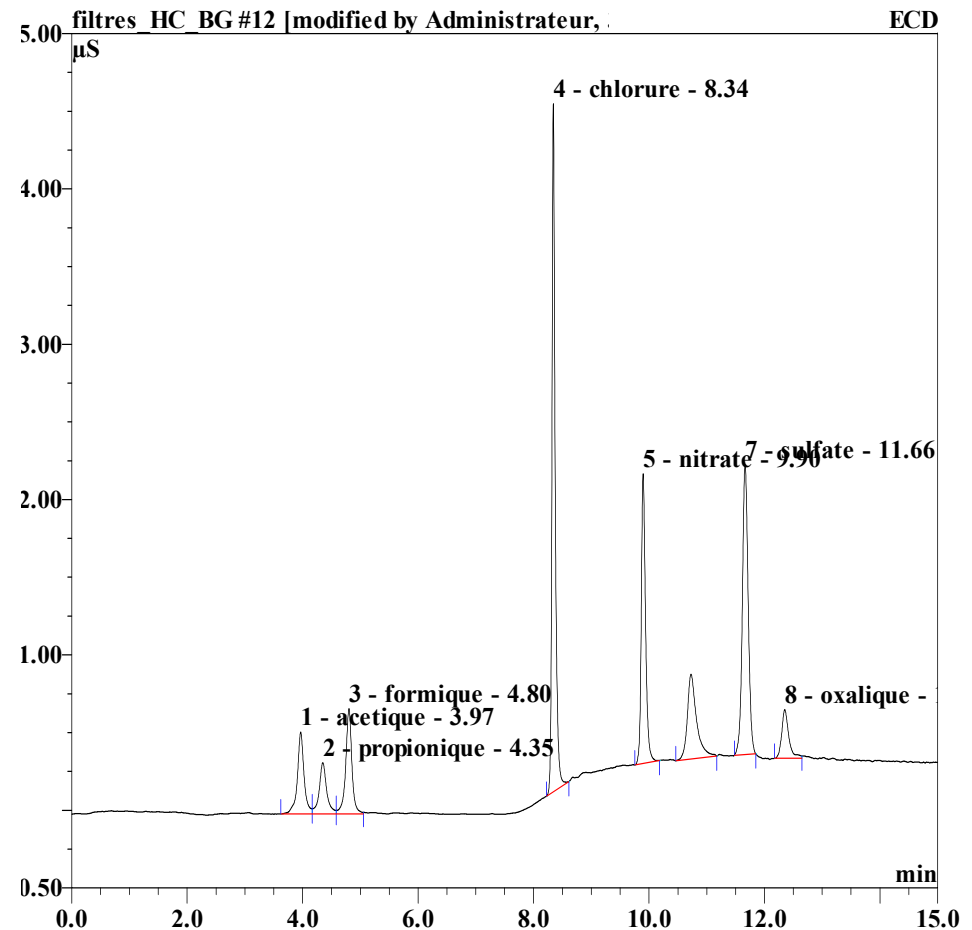


Ion Chromatography

Gradient mode for small organic acids and inorganic anions in the same run

Columns: AG11+AS11

Eluent: 0.4 mM NaOH to 8 mM NaOH



Thermo optical analysis



DRI model 2001 carbon analyser

Quantify in the same run
Organic Carbon and Black
Carbon on an aerosol sample

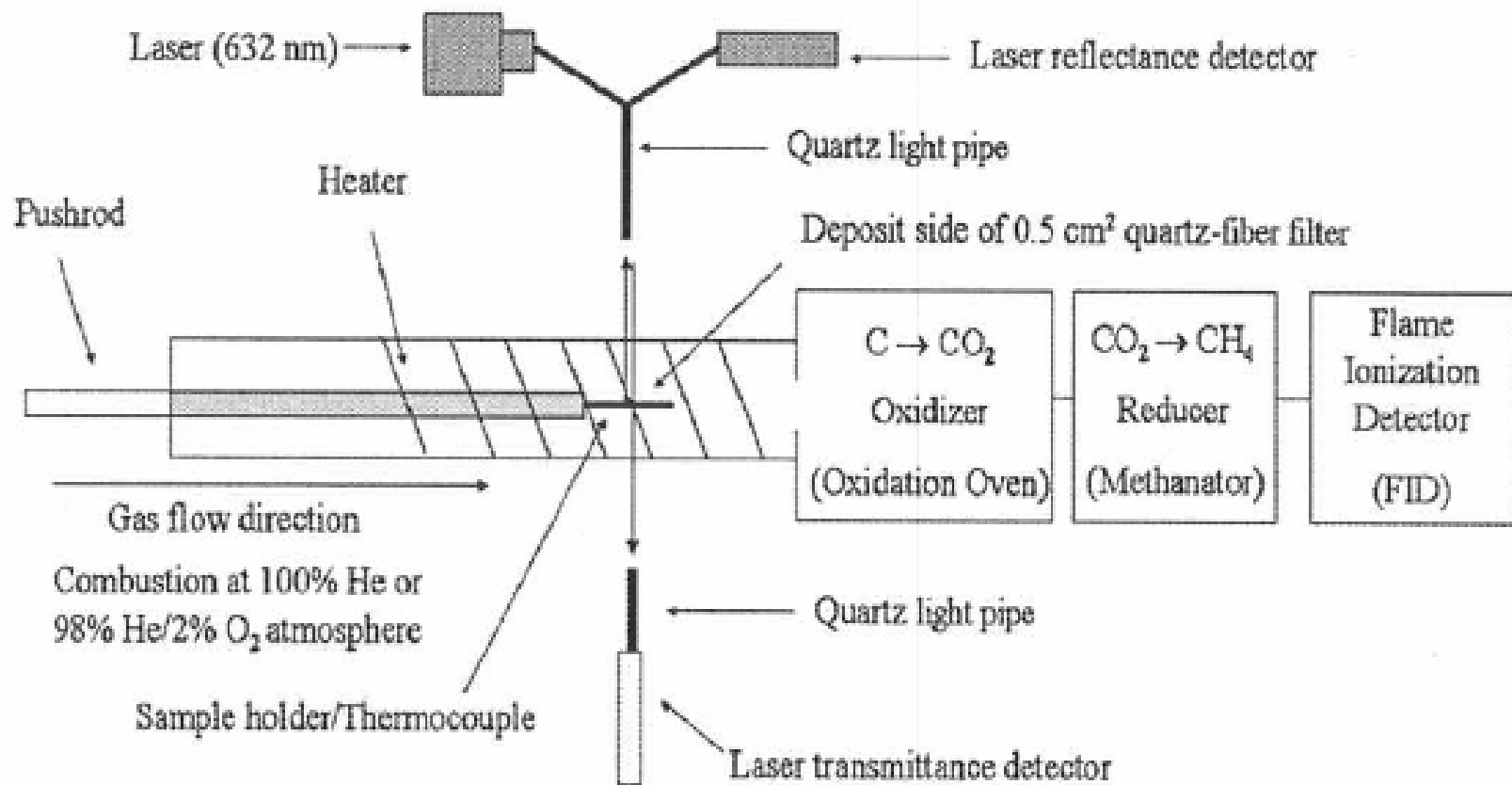
$$TC = OC + BC$$



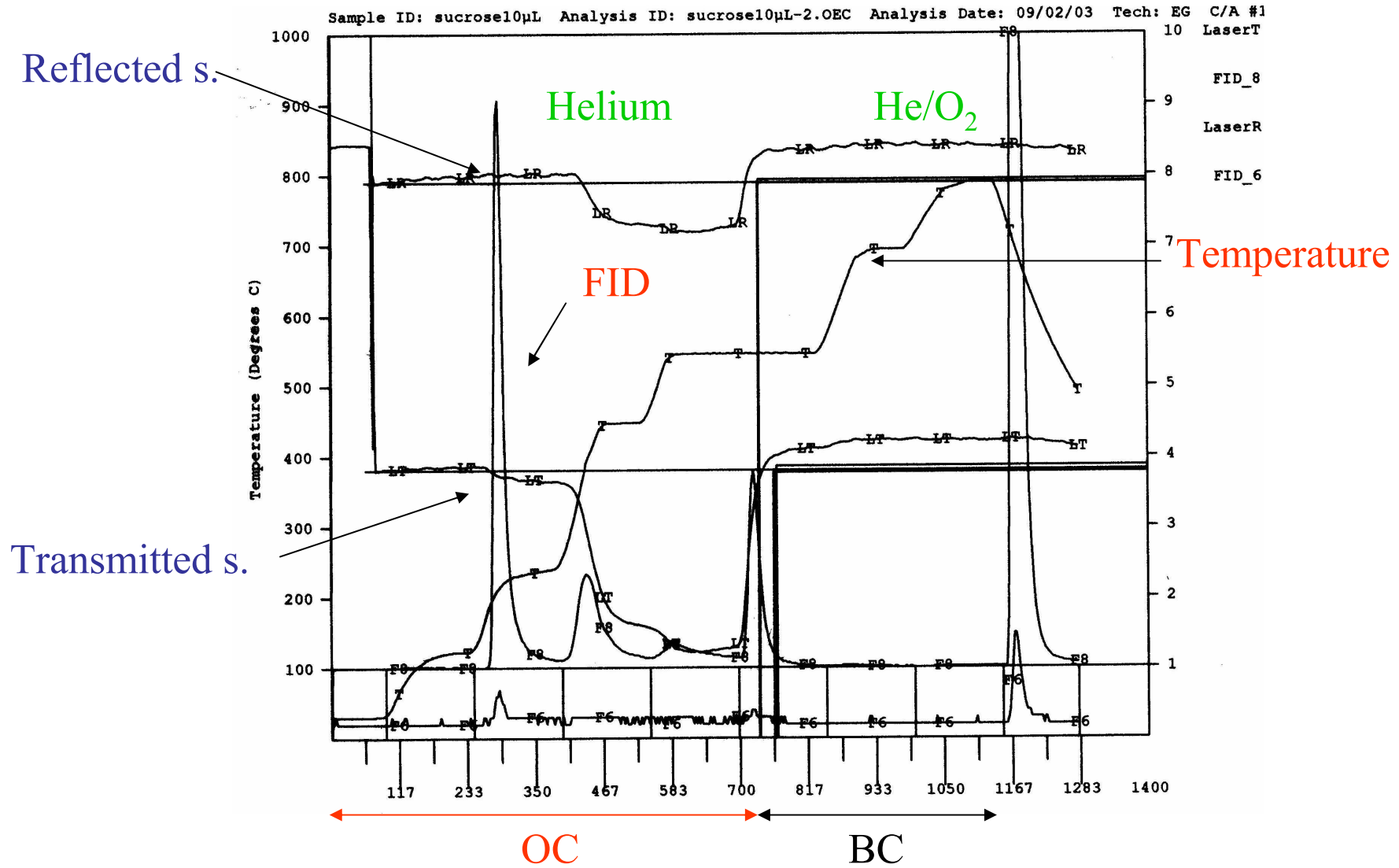
Thermo optical analysis

- For quartz filters samples
- IMPROVE method used (Chow and Watson 2002)
- Calibration with methane and sucrose solutions
- Detection limit: $0.05 \mu\text{g C/ cm}^2$

Schematic representation



Thermogram: IMPROVE method



Thermal analysis

- Used with quartz filters
- Sample is heated to 1000°C and CO₂ evolved is detected by infra-red
- Direct analysis of a sample : TC
- Sample heated 2 hours under oxygen flow to 340°C before analysis : BC
- Then $OC = TC - BC$

Thermal analysis

BRUKER

G4 ICARUS

Carbon Analyser



ICP-MS

Inductively coupled plasma mass spectrometry

Located in GET laboratory

Fast analysis of a lot of elements (80)

Includes isotopic speciation

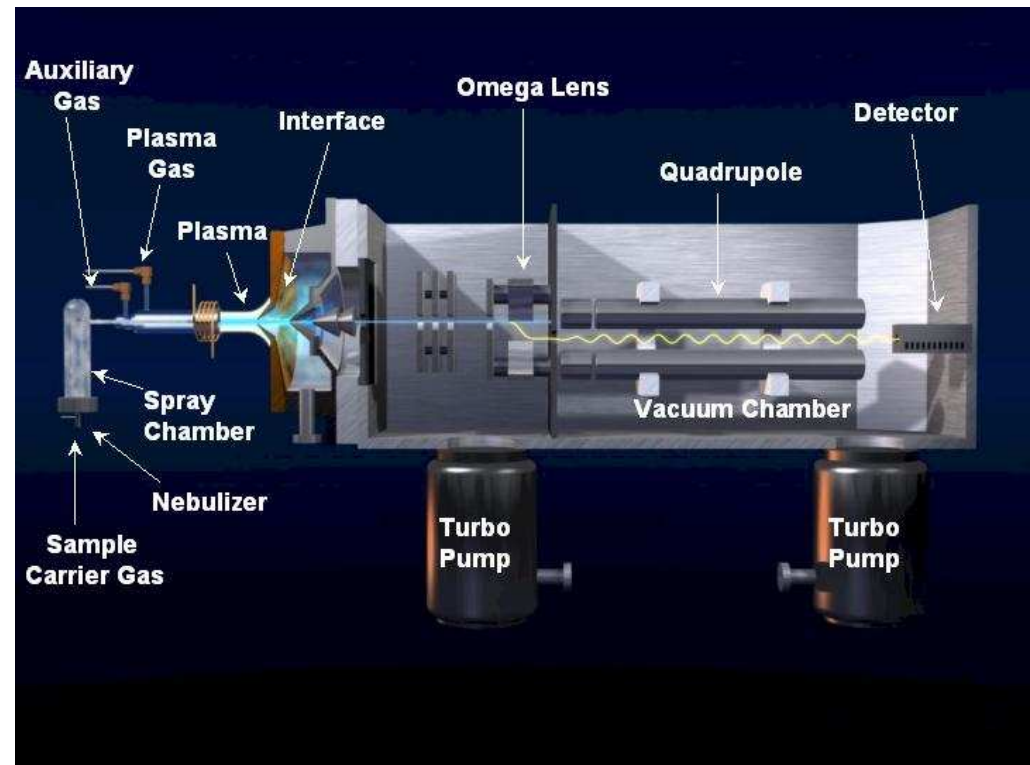
Low detection limit 1 ppt (1 ng/L)

Acid attack on Teflon Filters in clean room

ICP-MS



Thermo Finnigan system

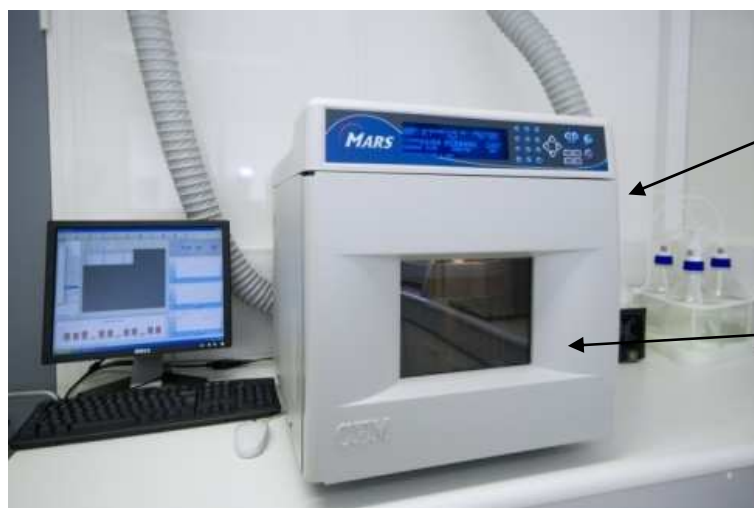


Schematic representation

ICP-MS

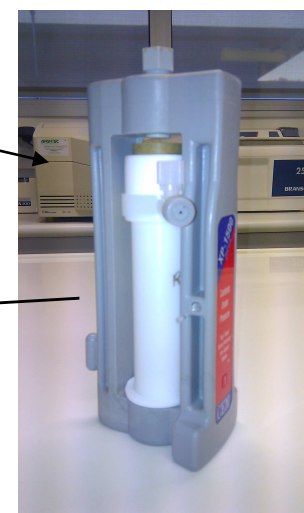
For TEFLON filters, method developed by our students for IDAF samples

- 10 ml nitric acid (HNO_3) bi-distilled + 0,5ml fluorhydric acid (HF)
- ultra-sonication for 15 minutes
- microwave digestion in two steps:
 - 15' ramp until 160°C hold 10 minutes
 - 10' ramp until 180°C hold 30 minutes
- rince with ultra pure water in the liner
- evaporate at 75°C to dry , add 2% bi-distilled nitric ac id before ICP-MS analysis.



Micro-wave oven

Liner



QA/QC

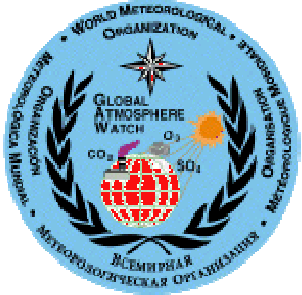
Quality Assurance in the sites:

- Enough sampling material available in the station (prevent lack)
- Procedures for changing of samples included in the field journal if there is changes in the technical staff
- Samples kept frozen just after collection (rain) or refrigerated (aerosols and passive samplers)
- Copies of sampling reporting forms should be sent to the lab (better than originals in case of loosing)
- Regular field blanks are necessary

QA/QC

Quality assurance in the lab

- Daily calibration and check standards
- Standards mother solutions changed every year
- Standard operating procedures for each kind of samples
- Preventive maintenance for instruments
- Regular test of sampling materials :plastic bags, Greiner tubes, plastic glassware used for filters soaking, fresh prepared passive samplers and field blanks



Quality Control

IDAF lab participates twice a year to the WMO/GAW intercomparison study: Analysis of 3 unknown simulated rain samples.

Parameters tested:

pH, conductivity

Na^+ , NH_4^+ , K^+ , Mg^{2+} , Ca^{2+}

Cl^- , NO_3^- , SO_4^{2-}



Quality Control

After WMO publishes statistics results and ring diagrams

GOOD - Green Hexagon

Measurement is within the interquartile range (IQR), defined as the 25th to 75th percentile or middle half (50%) of the measurements.

Examples: sulfate, ammonium, sodium, and potassium.

SATISFACTORY - Blue Trapezoid

Measurement is within the range defined by the median \pm IQR/1.349.

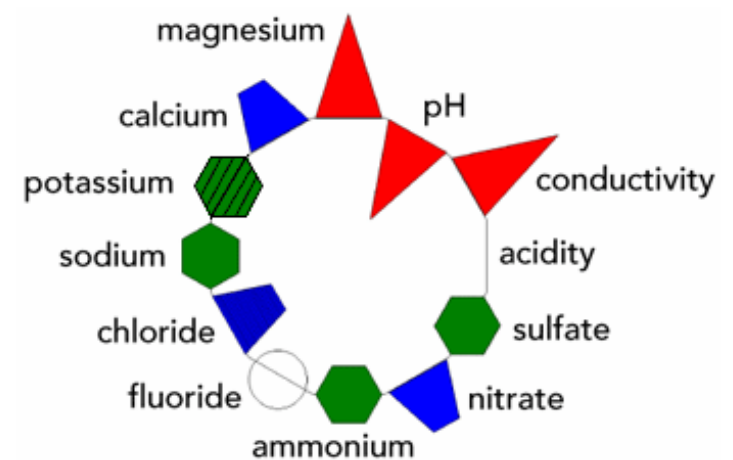
The ratio, IQR/1.349, is the non-parametric estimate of the standard deviation, sometimes called the pseudo-standard deviation.

Examples: nitrate, chloride, and calcium.

UNSATISFACTORY - Red Triangle

Measurement is outside the range defined by the median \pm IQR/1.349.

Examples: pH, conductivity, and magnesium.





Quality Control

Precipitation

Chemistry

Data

Quality

Objectives

Measurement

DQO

pH

± 0.07 units

conductivity

$\pm 7\%$

acidity

$\pm 25\%$

sulfate

$\pm 7\%$

nitrate

$\pm 7\%$

Ammonium

$\pm 7\%$

fluoride

none

chloride

$\pm 10\%$

sodium

$\pm 10\%$

potassium

$\pm 20\%$

calcium

$\pm 15\%$

magnesium

$\pm 10\%$

Sources

WMO/ GAW REPORT No 160 : MANUAL FOR THE GAW PRECIPITATION CHEMISTRY PROGRAMME:Data Quality Objectives and Standard Operating Procedures prepared by the GAW Precipitation Chemistry Science Advisory Group

Ferm and al. 1994. New measurement technique for air pollutants(in swedish), Kemish tidskrift,1,30-32.

Chow, J.C., Watson, J.G., 2002. PM2.5 carbonate concentrations at regionally representative Interagency Monitoring of Protected Visual Environment sites. Journal of Geophysical Research 107 (D21), ICC6-1–ICC6-9. doi:10.1029/2001JD000574.

EMEP manual for sampling and chemical analysis. Norwegian Institute for Air Research

www.aero.obs-mip.fr

<http://qasac-americas.org>

www.dekati.fi

www.get.obs-mip.fr

www.nilu.no/products

www.dionex.com

<http://idaf.sedoo.fr>

www.thermoscientific.com

www.dri.edu