

2339-9

Workshop on Atmospheric Deposition: Processesand Environmental Impacts

21 - 25 May 2012

Links Between Air Quality and Atmospheric Deposition

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Links Between Air Quality and Atmospheric Deposition

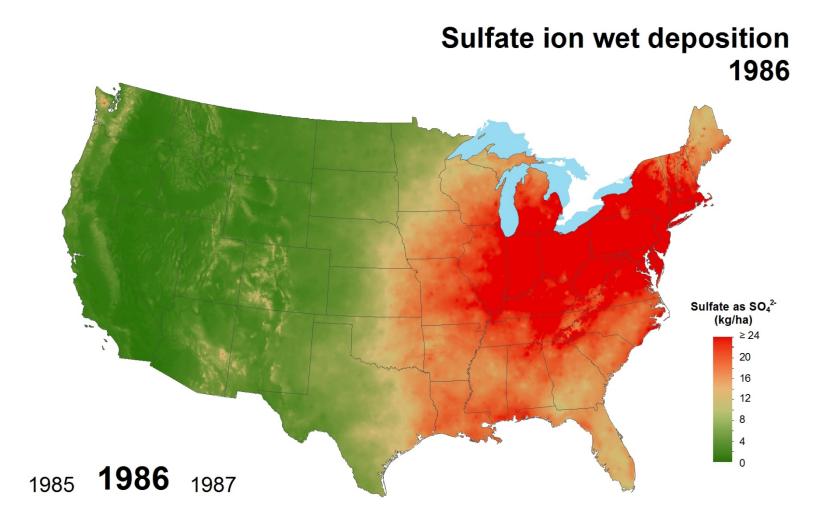
Christopher Lehmann National Atmospheric Deposition Program Illinois State Water Survey - Prairie Research Institute University of Illinois, Urbana-Champaign

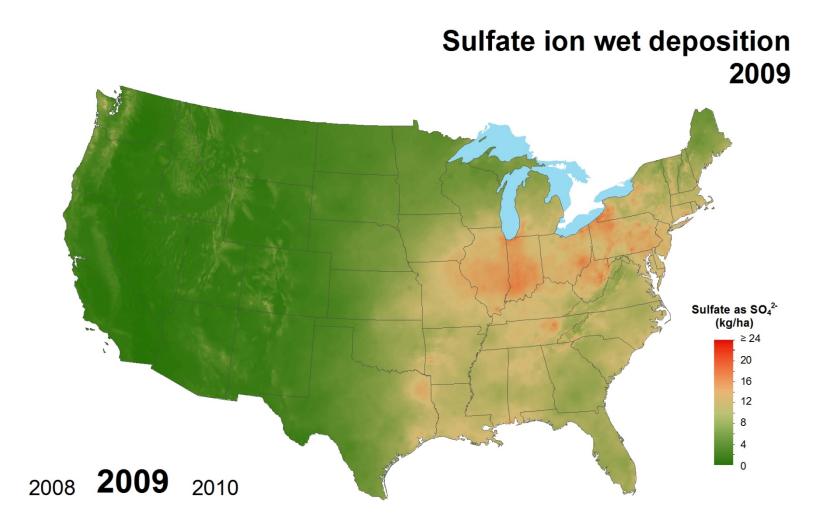


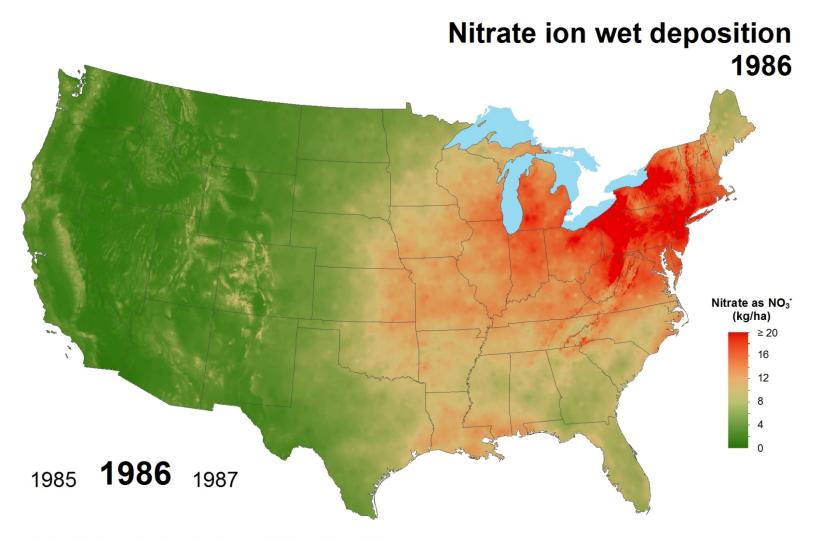


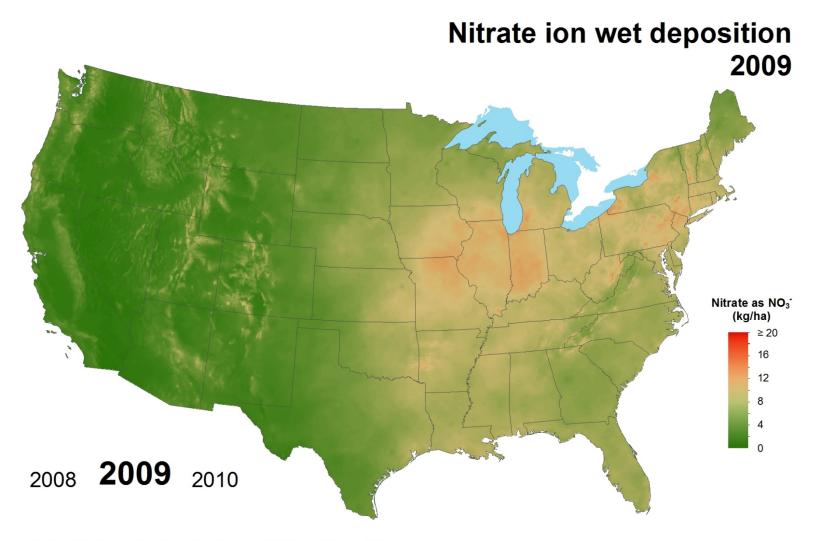
Outline

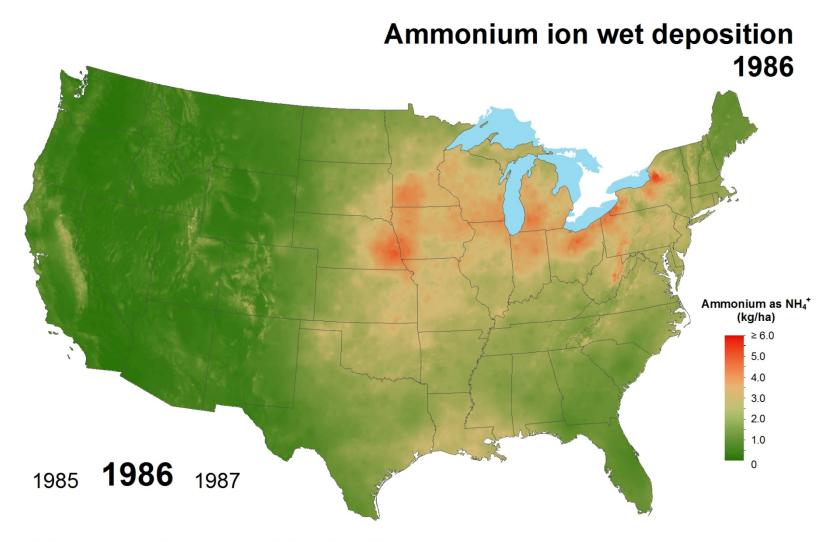
- 1. Sulfur and nitrogen interactions in the atmosphere
- 2. Evaluating trends in emissions, air quality, and wet/dry deposition
- 3. Remarks about passive air samplers

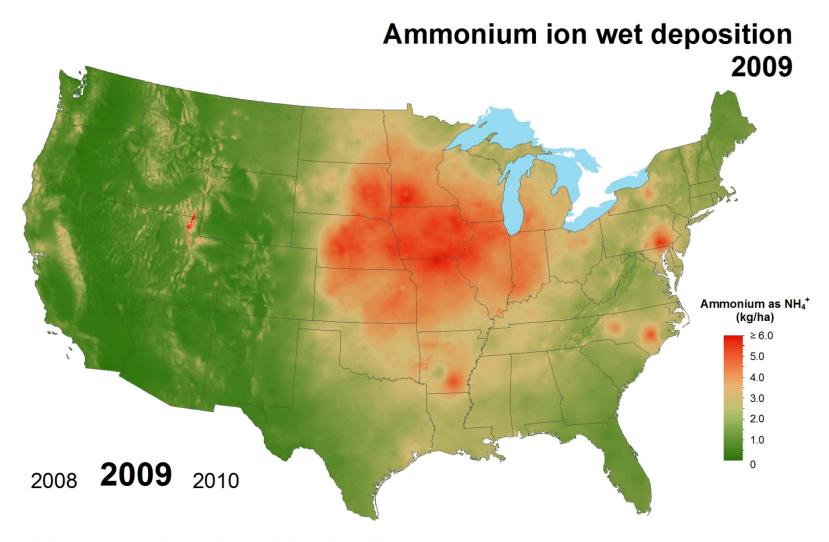












Aqueous Phase Chemistry: Sulfate-Ammonia

Considering a system containing only sulfate and ammonia, the following species are formed in transition:

- 1. Sulfuric Acid: H₂SO₄
- 2. Ammonium Bisulfate:(NH₄)HSO₄
- 3. Letovicite: $(NH_4)_3H(SO_4)_2$
- 4. Ammonium Sulfate: (NH₄)₂SO₄
- 5. "Ammonia Rich"

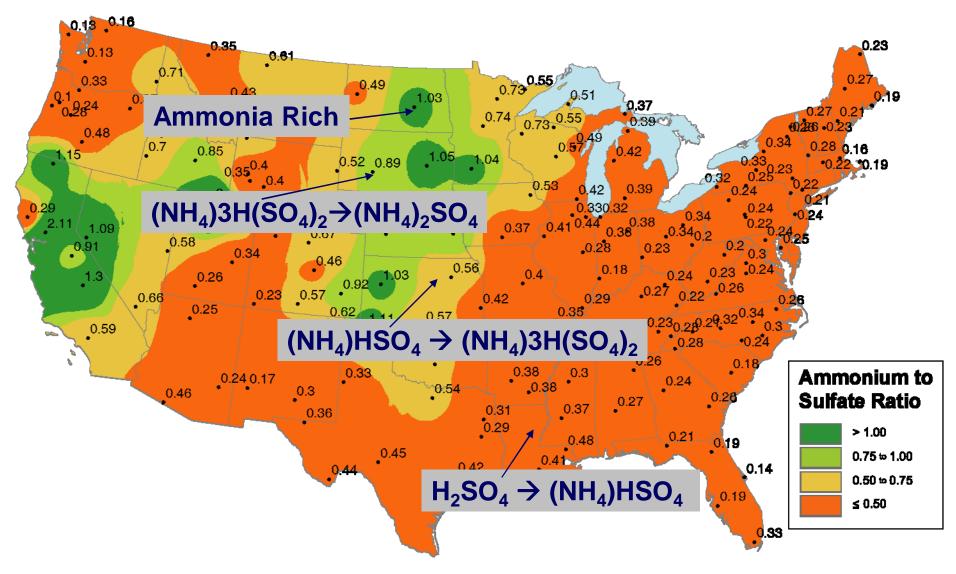
 $NH_4/SO_4 = 0.00$ $NH_4/SO_4 = 0.50$ $NH_4/SO_4 = 0.75$ $NH_4/SO_4 = 1.00$ $NH_4/SO_4 > 1.00$

These species have different physical properties that affect long-range transport in the atmosphere

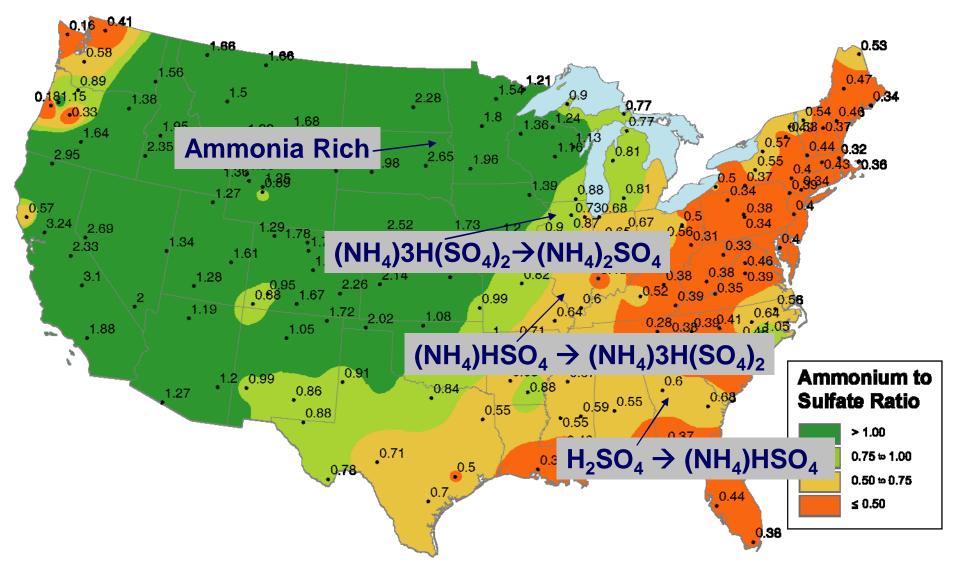
Impacts of Ammonia-Rich Environment

- (NH₄)₂SO₄ is most stable form of ammonia and can transport long distances
 - $-SO_4^{2-}$ will react with all available NH_4^+ .
 - NH_4^+ in excess of that required to neutralize SO_4^{2-} is available to react with other species, such as NO_3^- to form NH_4NO_3 .
 - NH₄NO₃ is a labile species that will partition between gas and aqueous phases and can deposit locally
 - NH₃/NH₄⁺ in excess of that required to neutralize acidic species is soluble and can deposit locally

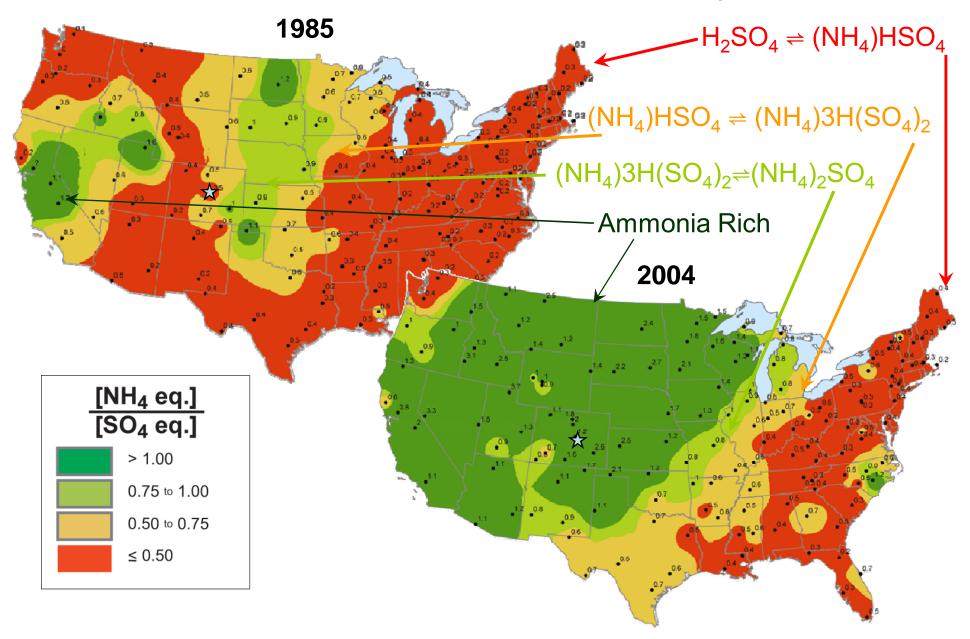
Ammonium/Sulfate Ratio, 1984-1986



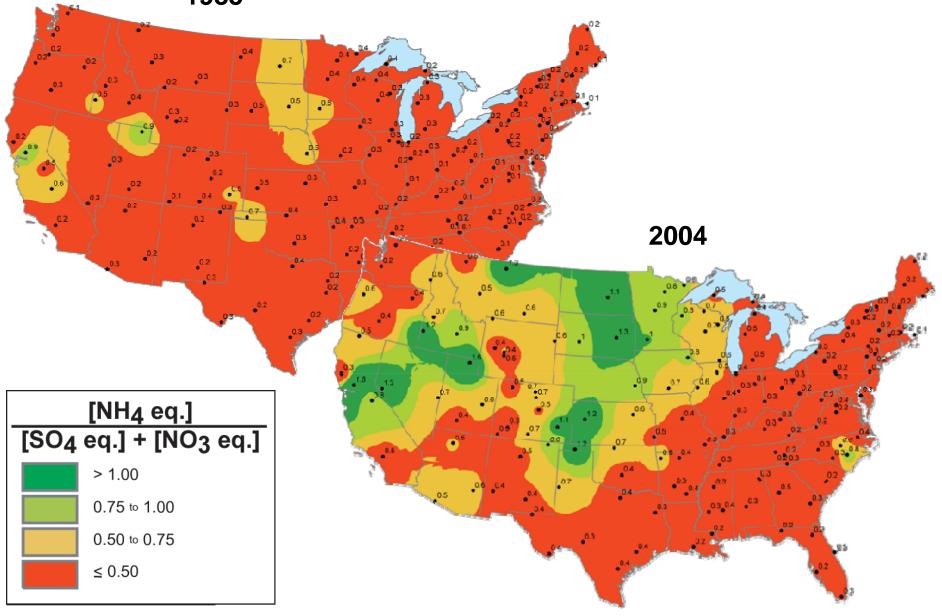
Ammonium/Sulfate Ratio, 2002-2004

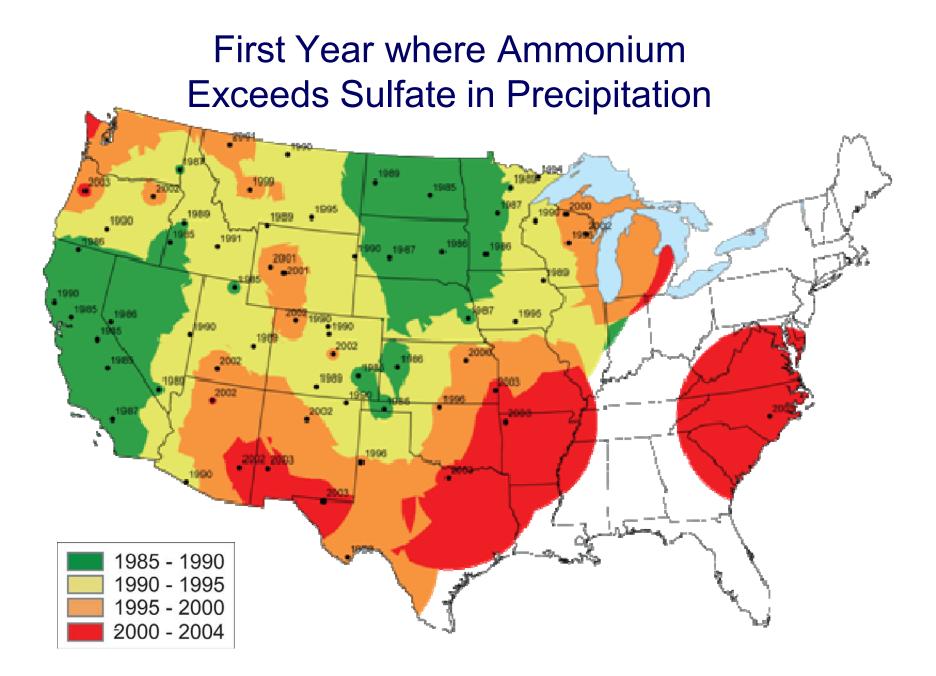


Ammonium to Sulfate Ratio in Precipitation



Ammonium to Sulfate plus Nitrate Ratio in Precipitation





Sulfate, Ammonium & Nitrate Reactions

$$\frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} < 1.0$$

$$\frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} < 1.0$$

$$\frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} = 0.50$$

$$\frac{[NH_{4}^{+} + NH_{4}^{+} + NSO_{4}]}{[NH_{4}^{+} + NH_{4}^{+} + HSO_{4}]} = \frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} = 0.75$$

$$\frac{[NH_{4}^{+} + NH_{4}^{+} + NH_{4}^{+} + NH_{4}^{+} + NH_{4}^{+} + HSO_{4}]}{[SO_{4}^{2^{\circ}} eq.]} = \frac{2}{[NH_{4}^{+} + NH_{4}^{+} + SO_{4}^{-}]} = \frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} = 1.00$$

$$\frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} > 1.0 \quad \frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} < 1.0$$

$$\frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} > 1.0 \quad \frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} > 1.0 \quad \frac{[NH_{4}^{+} eq.]}{[SO_{4}^{2^{\circ}} eq.]} > 1.0$$

Transport Distance

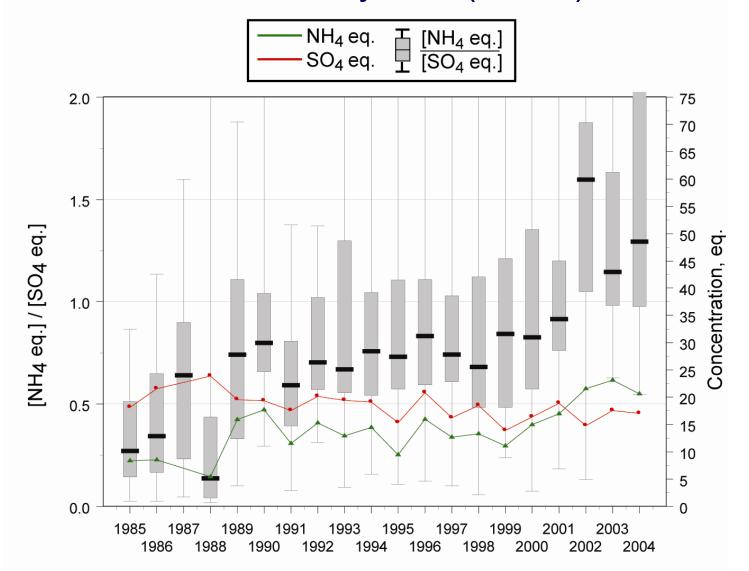
Sulfate, Ammonium & Nitrate Reactions

$$\frac{[\text{NH}_{4}^{+} \text{eq.}]}{[\text{SO}_{4}^{2^{\circ}} \text{eq.}]} < 1.0$$

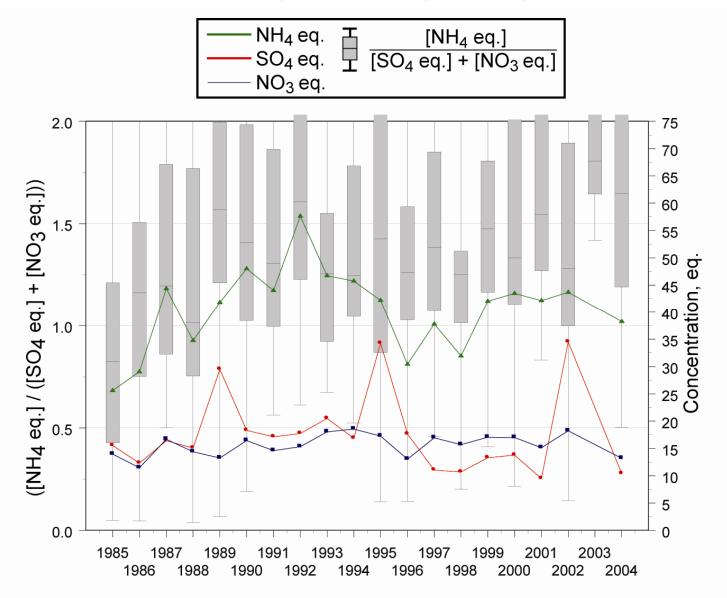
$$\frac{[\text{NH}_{4}^{+}] + (\text{NH}_{4}^{+}] +$$

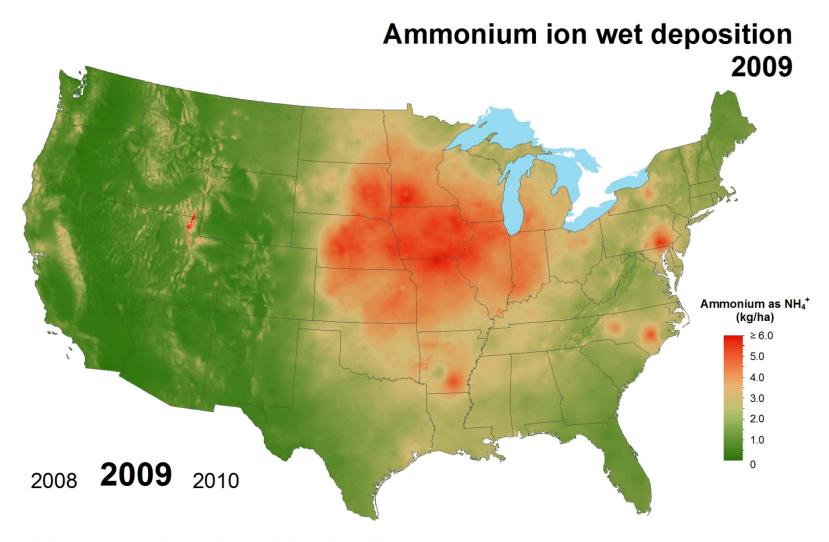
Finalayson-Pitts & Pitts, 2000

Ammonium / Sulfate Ratio in Precipitation Teller County, CO (CO21)



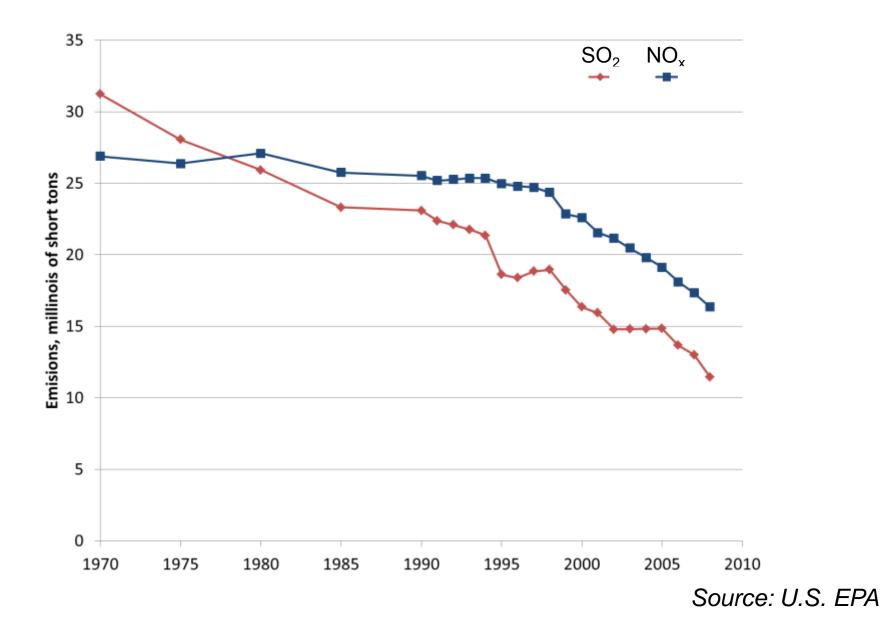
Ammonium / (Sulfate + Nitrate) Ratio in Precipitation Logan, UT (UT01)





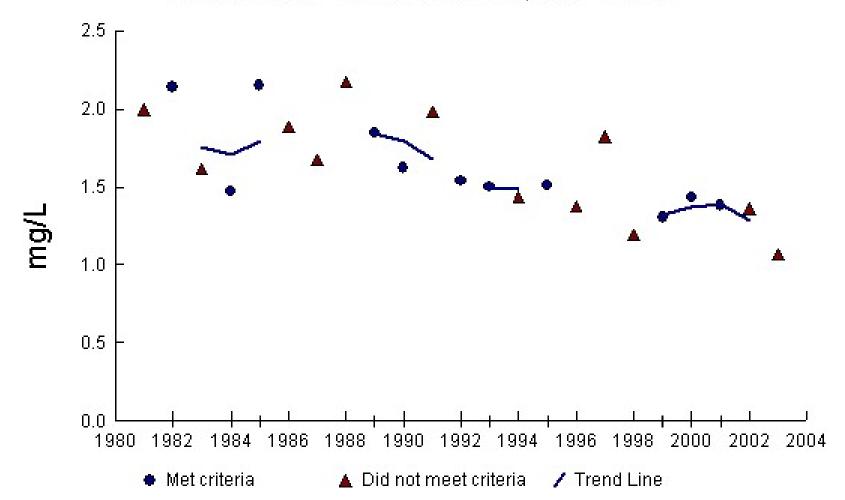
How do trends in wet deposition compare to trends in emissions?

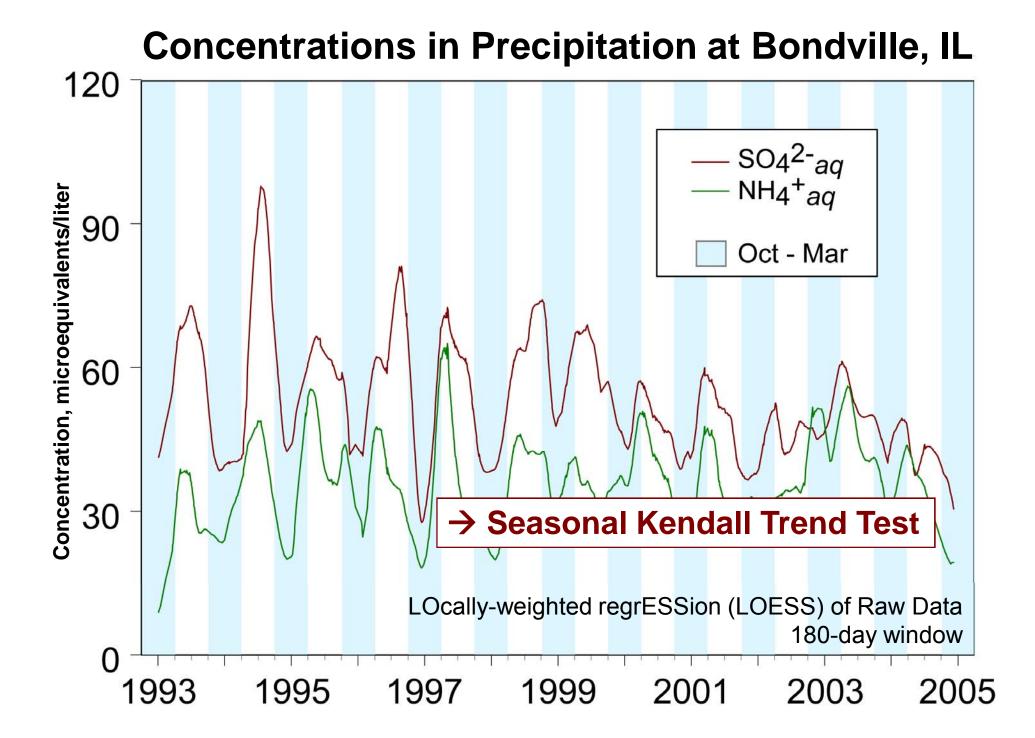
Trends in US Emissions



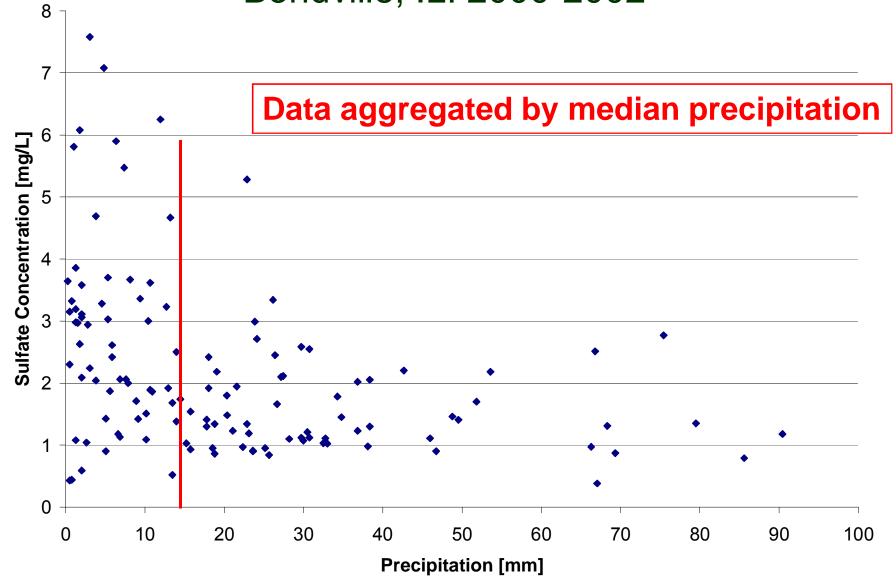
Evaluating Trends in Real Data

NADP/NTN Site MA01 Annual SO4 concentrations, 1981-2003



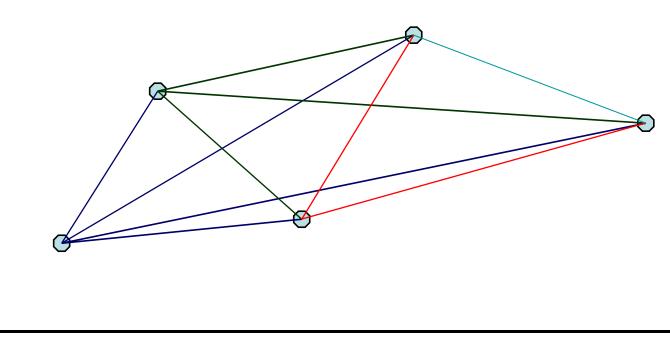


Sulfate Concentration vs. Precipitation Amount at Bondville, IL: 2000-2002



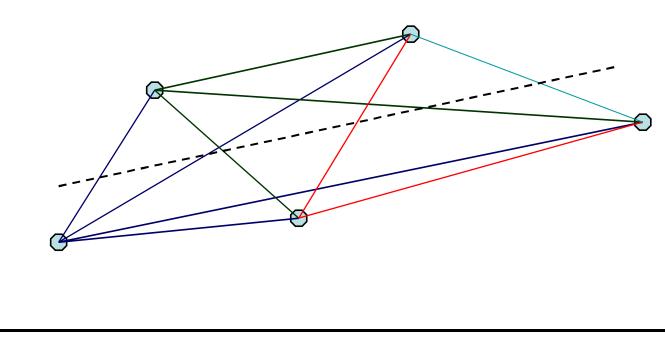
Kendall Trend Test

- Non-parametric test of monotonic trend
- Stepwise evaluation increasing/decreasing trend
- Insensitive to missing values, outliers



Sen's Median Estimator

- Non-parametric estimate of trend magnitude
- Median of slopes between all points
- Insensitive to missing values, outliers



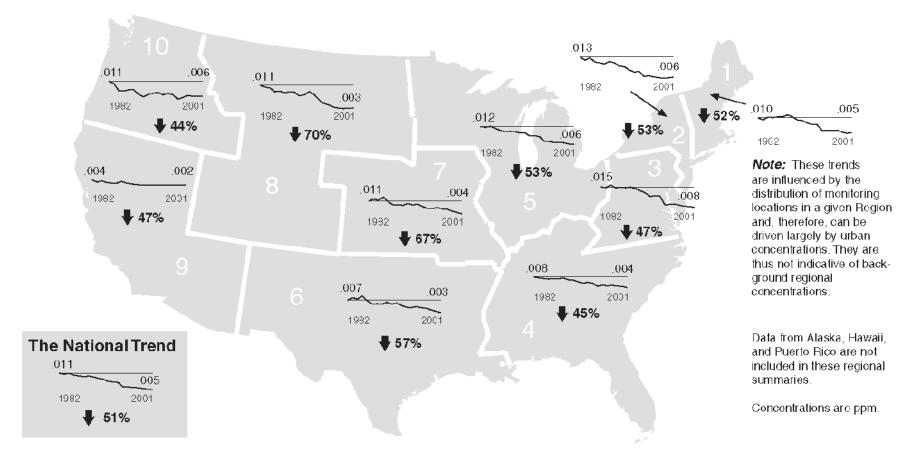


Computer Program for the Kendall Family of Trend Tests

Scientific Investigations Report 2005–5275

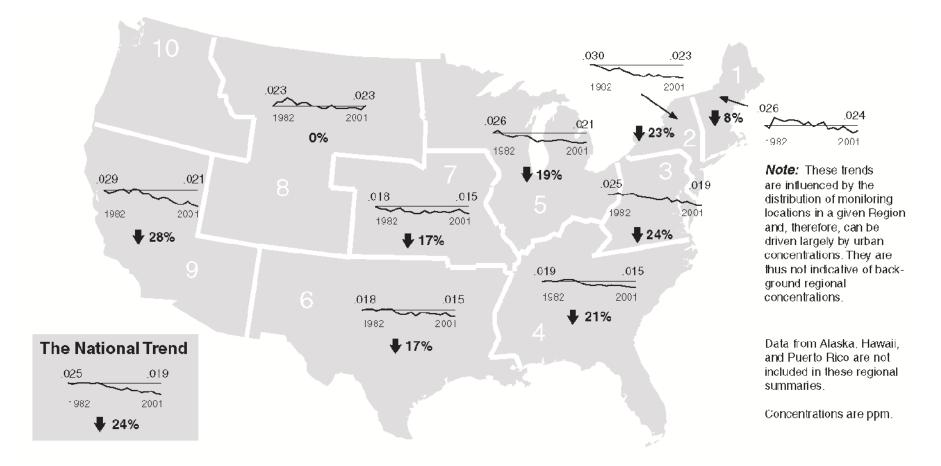
U.S. Department of the Interior U.S. Geological Survey

Trends in Sulfur Dioxide



U.S. EPA, National Air Quality and Emissions Trends Report, 2003

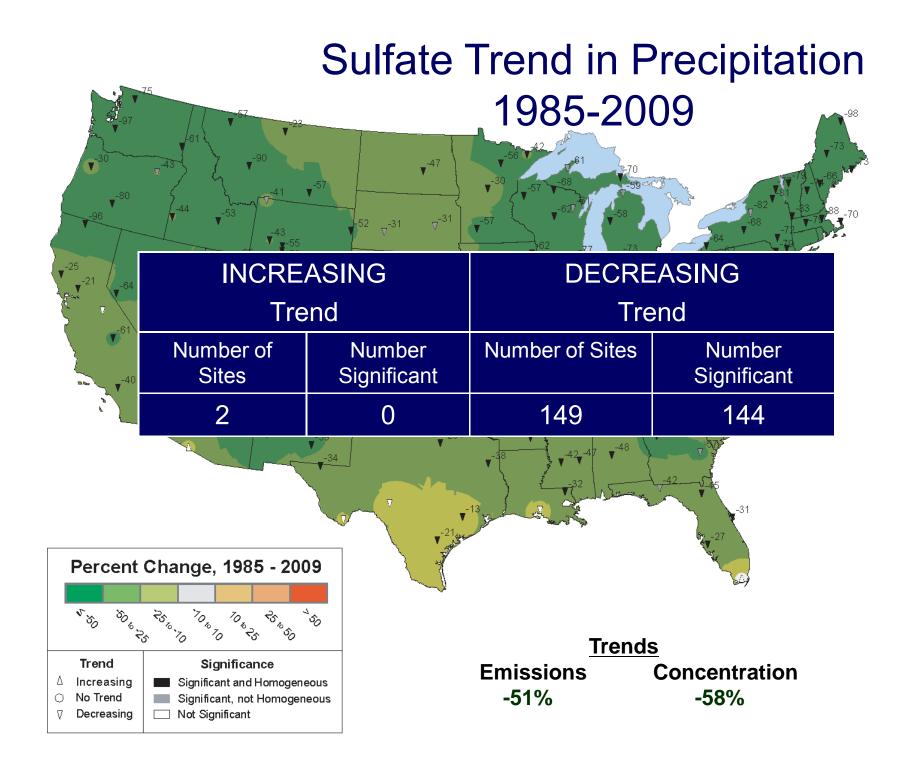
Trends in Nitrogen Dioxide

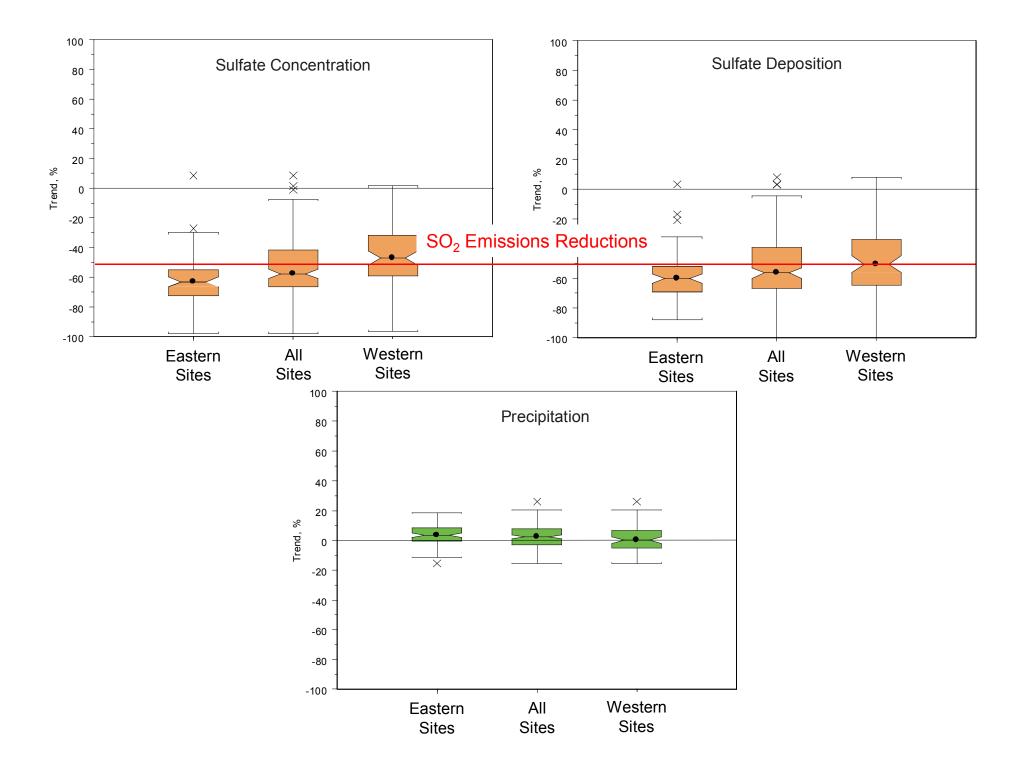


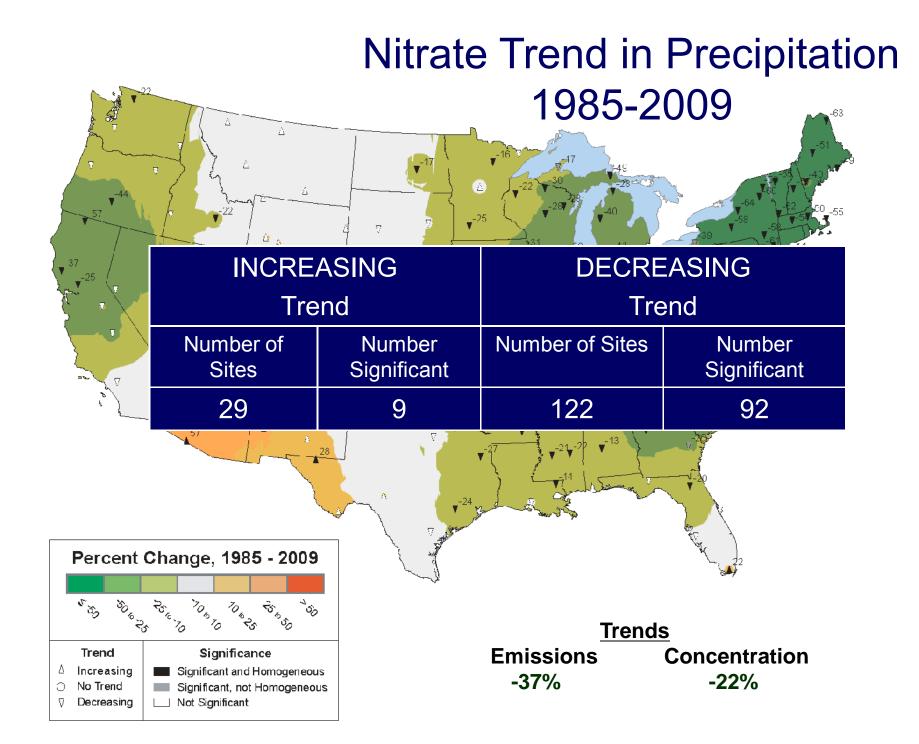
U.S. EPA, National Air Quality and Emissions Trends Report, 2003

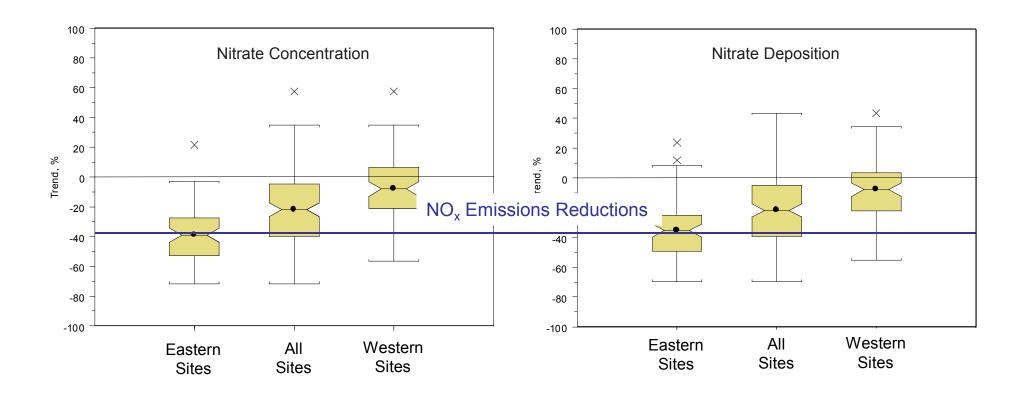
Evaluating Trends in NADP-NTN Data

- Data from 151 sites, operational between 1985-2009 (~210,000 weekly data sets)
- Precipitation-weighted mean seasonal averages
- Seasonal Kendall Trend Test
 - Null Hypotheses:
 - Trend is zero (no trend)
 - Trends are homogeneous (same in every season)
 - Significance Level
 - $p \le 0.1$ for trend significance
 - p > 0.1 for homogeneity
 - Trend magnitude by Sen's Median Estimator



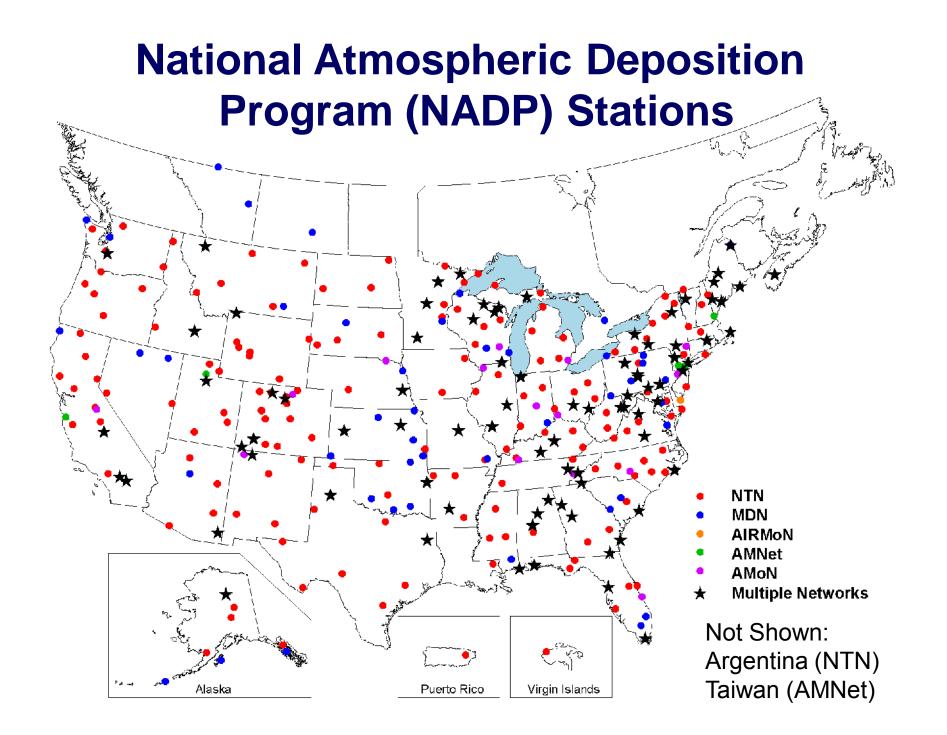




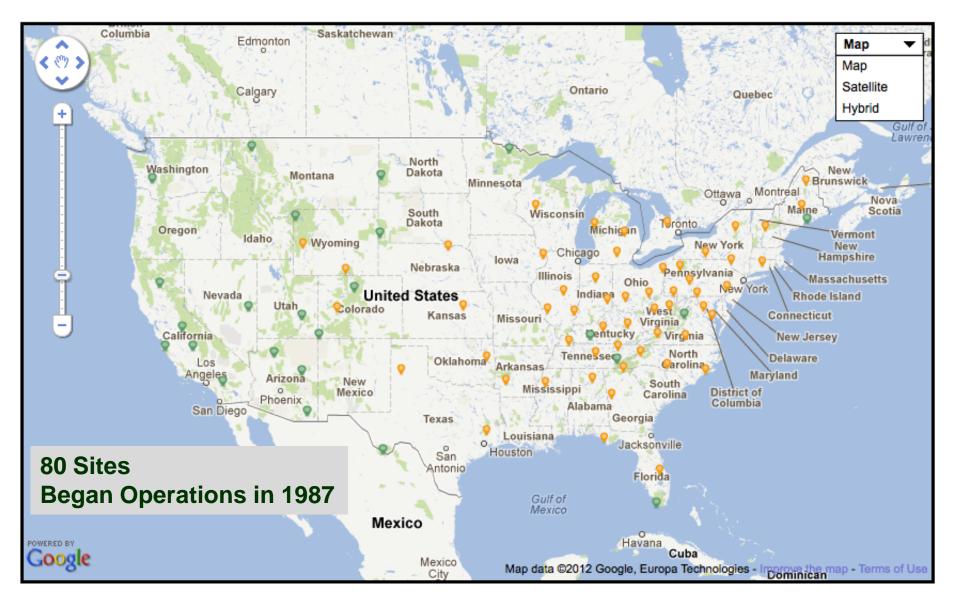


For trend analysis method, details see:

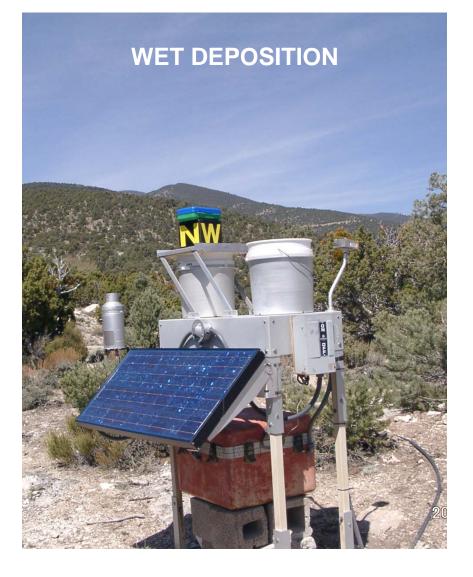
http://nadp.isws.illinois.edu/dl/ clehmann/trends/ How do trends in emissions compare to trends in air quality and wet/dry deposition?



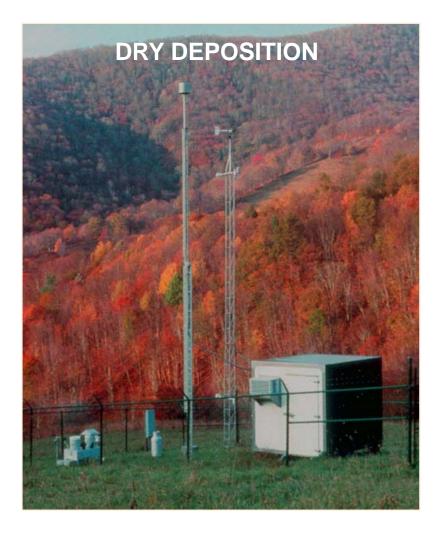
Clean Air Status and Trends Network (CASTNET)

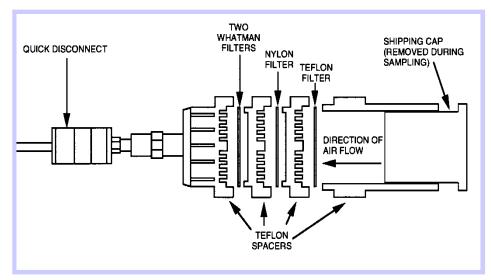


NTN NV05 Great Basin National Park



CASTNET CTH110 Connecticut Hill, NY





CASTNET Filter Pack



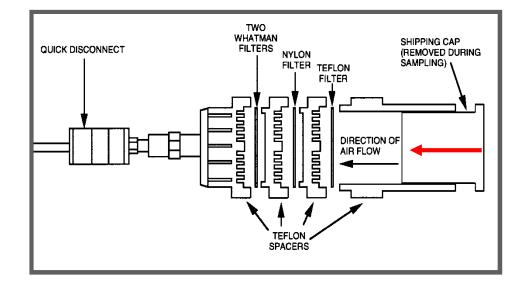


Images provided by U.S. EPA's Clean Air Status and Trends Network

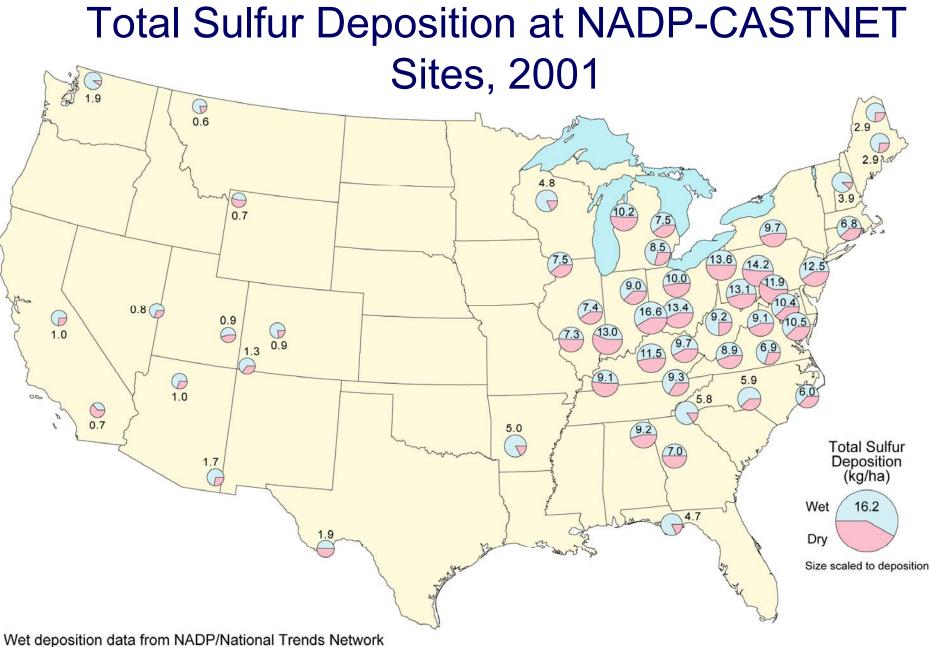
Clean Air Status & Trends Network (CASTNET)

- Weekly 3-stage filter pack samples
 - Teflon filter: particulate sulfate and nitrate
 - Nylon filter: nitric acid, sulfur dioxide
 - Impregnated cellulose: sulfur dioxide



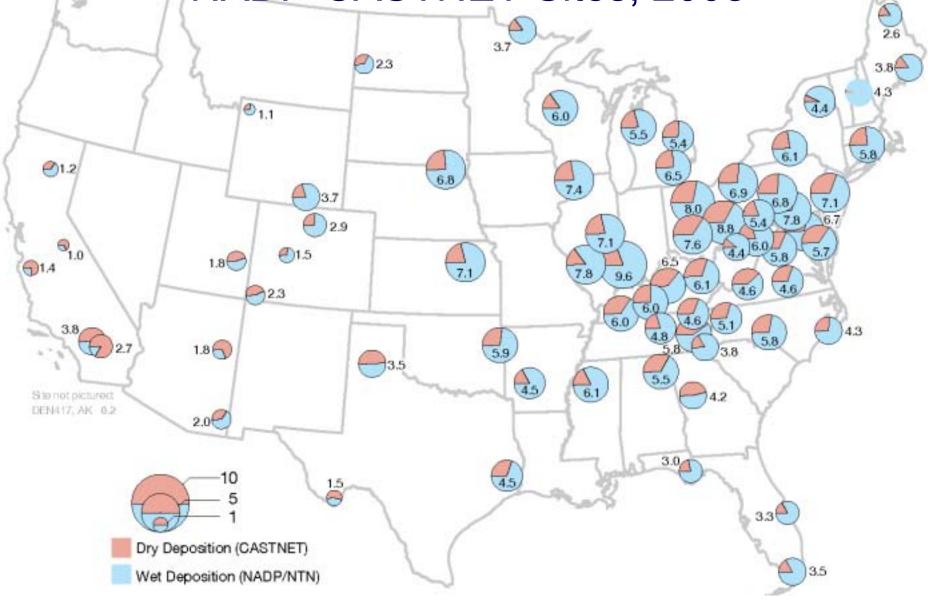


Images provided by Clean Air Status and Trends Network (U.S. EPA)

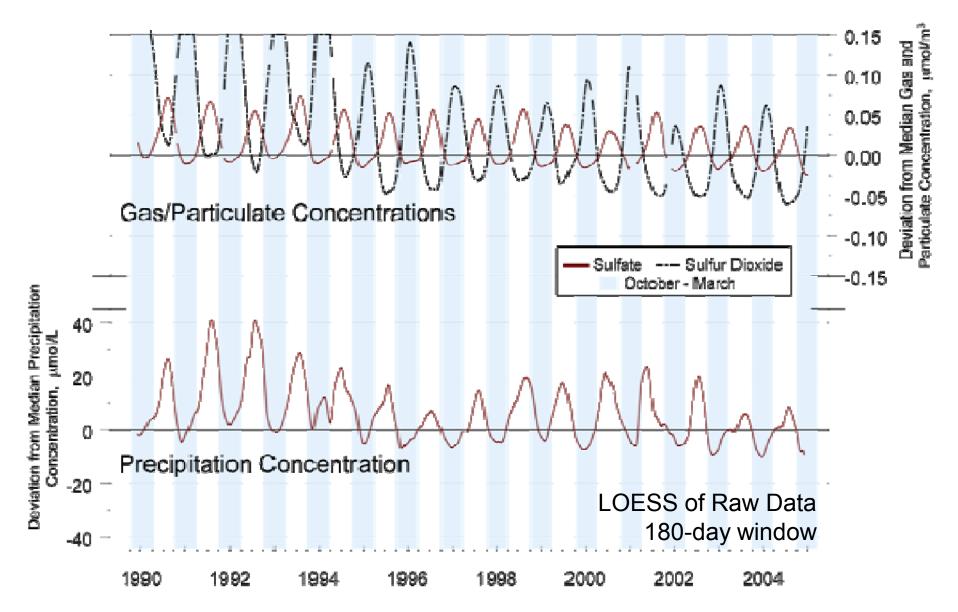


Wet deposition data from NADP/National Trends Network Dry deposition data from Clean Air Status and Trends Network

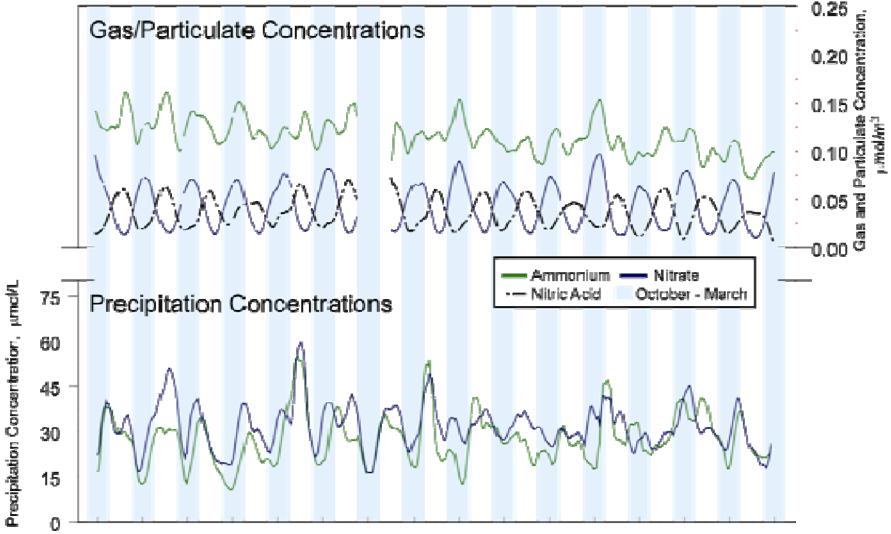
Total Inorganic Nitrogen Deposition at NADP-CASTNET Sites, 2008



Sulfur Dioxide, Particulate Sulfate, and Precipitation Sulfate Tucker County, WV (WV18)

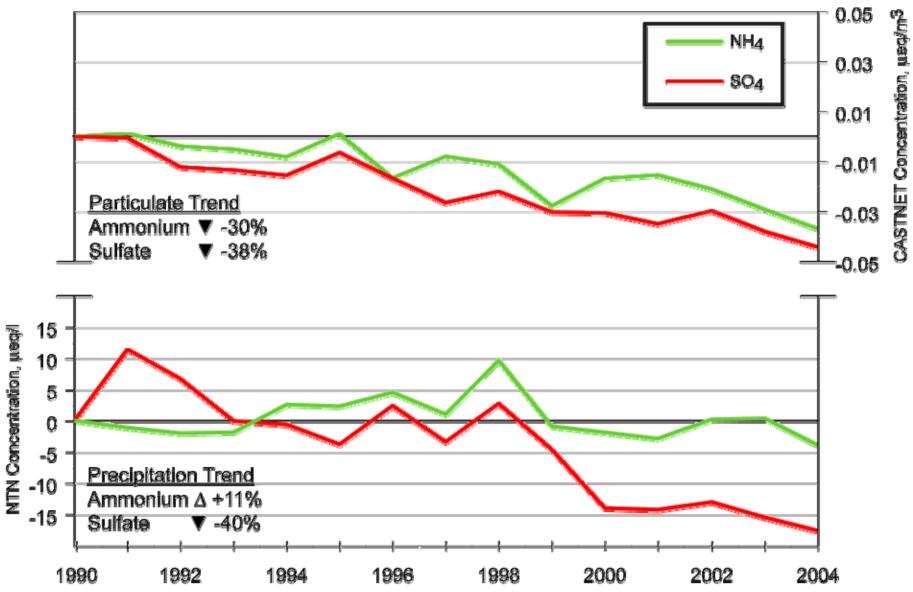


Ammonium, Nitrate, and Nitric Acid Concentrations Bondville, IL (IL11)

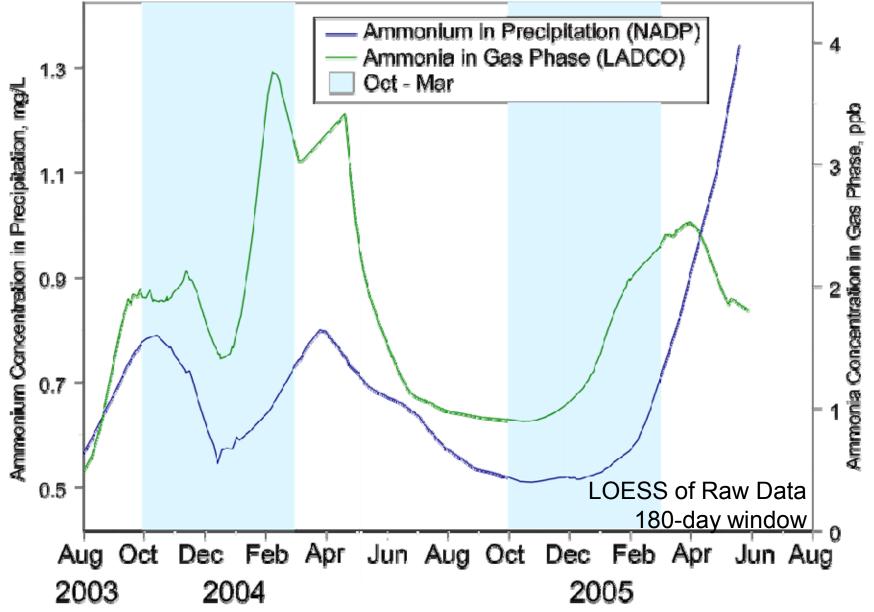


LOESS of Raw Data 180-day window

Ammonium vs. Sulfate Trends Bondville, IL



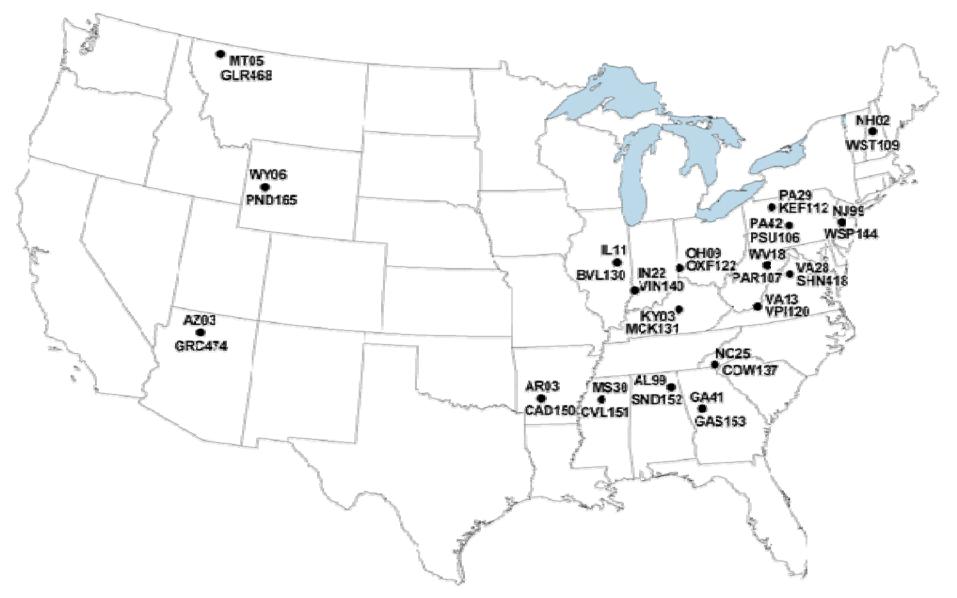
Ammonia vs. Ammonium Concentrations Bondville, IL

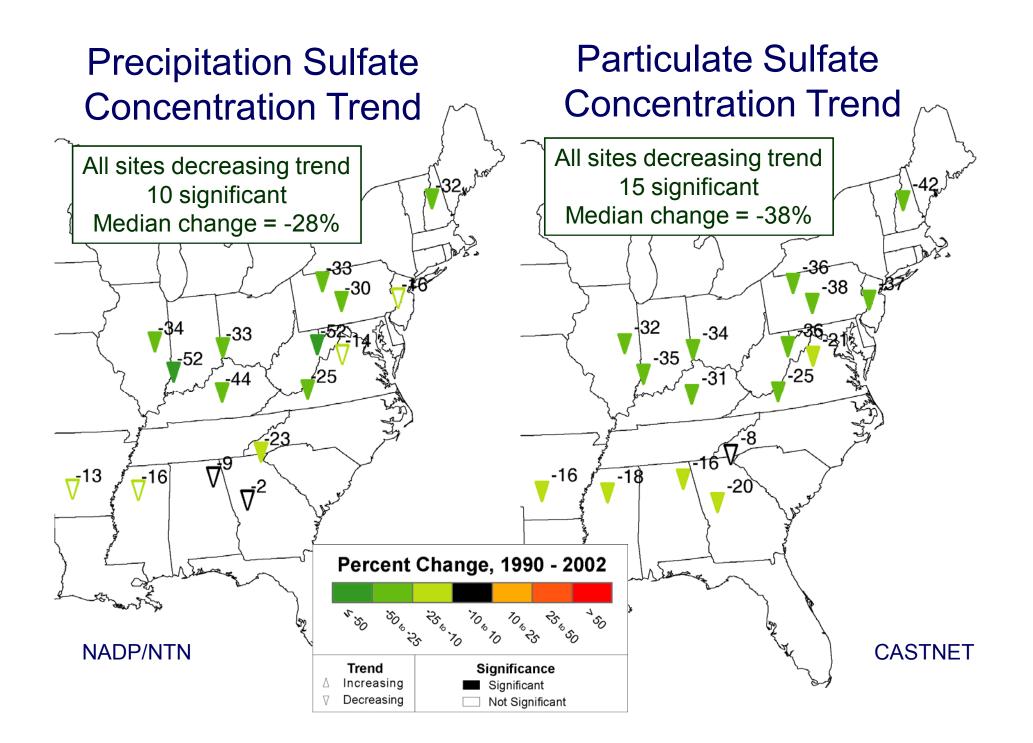


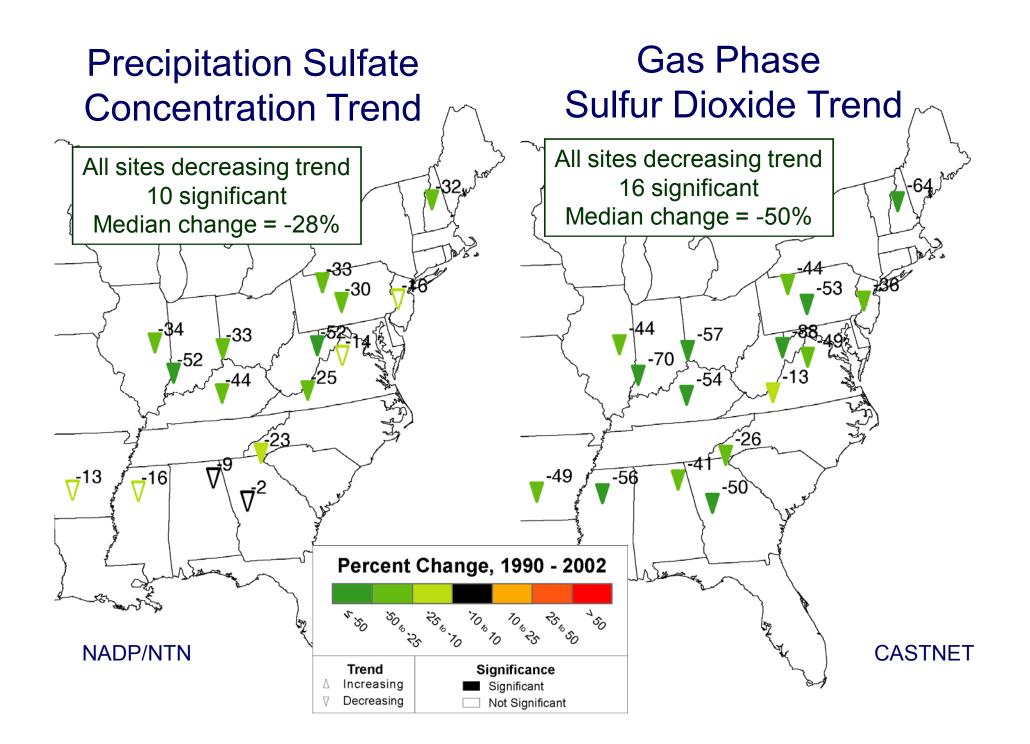
Comparing Trends in Wet and Dry Deposition

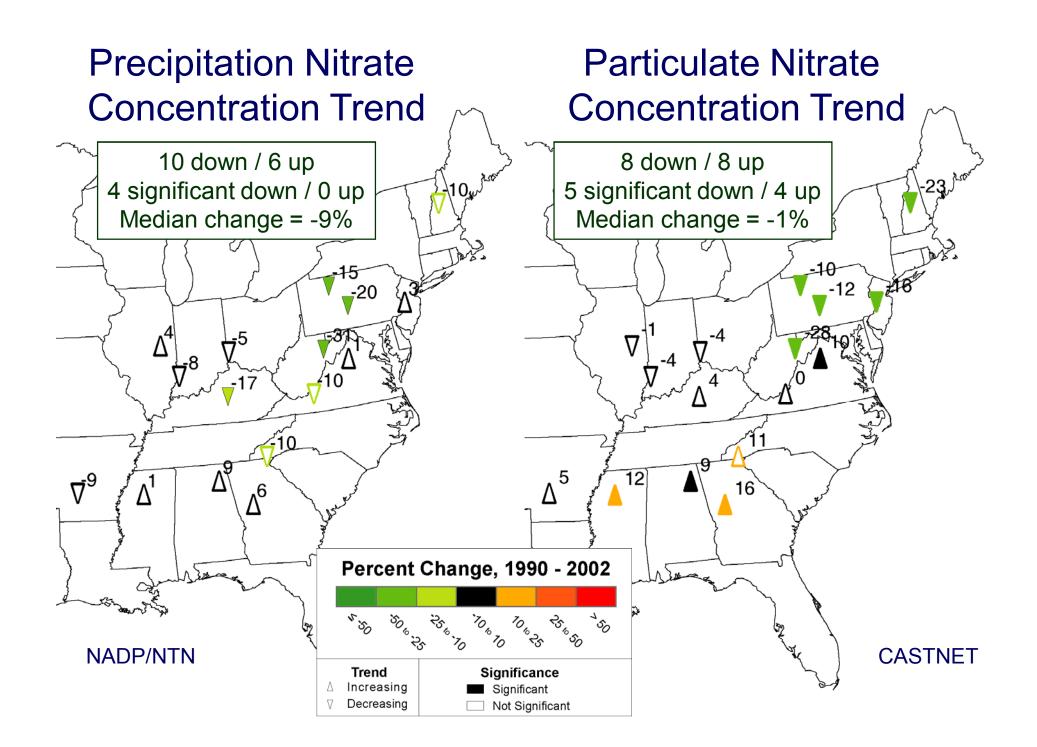
- Data from 19 collocated (< 10 km) wet and dry deposition monitoring sites, 1990 to 2004
- Wet deposition from precipitation concentrations (NADP/NTN)
 - Ammonium, sulfate, nitrate
- Dry deposition from gas and particulate concentrations (CASTNET)
 - Particulate ammonium, sulfate, and nitrate
 - Gas phase sulfur dioxide and nitric acid

Collocated Sites Evaluated



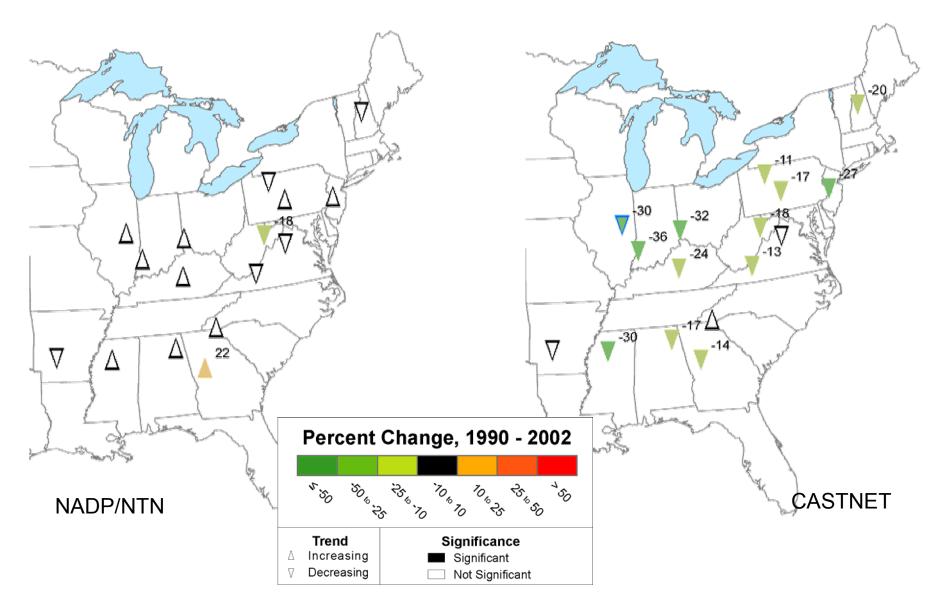






Precipitation Ammonium Concentration Trend

Particulate Ammonium Concentration Trend



Why Wet-Only Deposition?



BULK DEPOSITION COLLECTOR



Can't we just measure bulk deposition to measure total deposition?

TOTAL = WET + DRY



UNKNOWS:

- Capture efficiency of surrogate surface?
- Reemission/volatilization of pollutants?



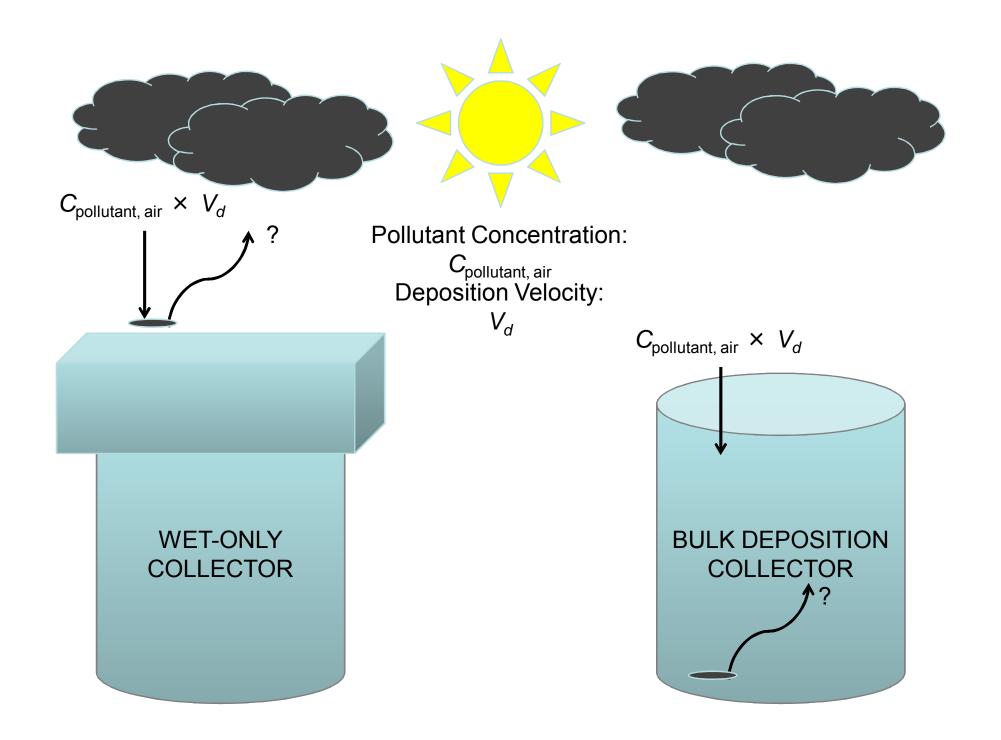


<u>UNKNOWS:</u>

- Capture efficiency of surrogate surface?
- Reemission/volatilizati on of pollutants?
- Gas capture and partitioning in water surface?
- Evaporation?



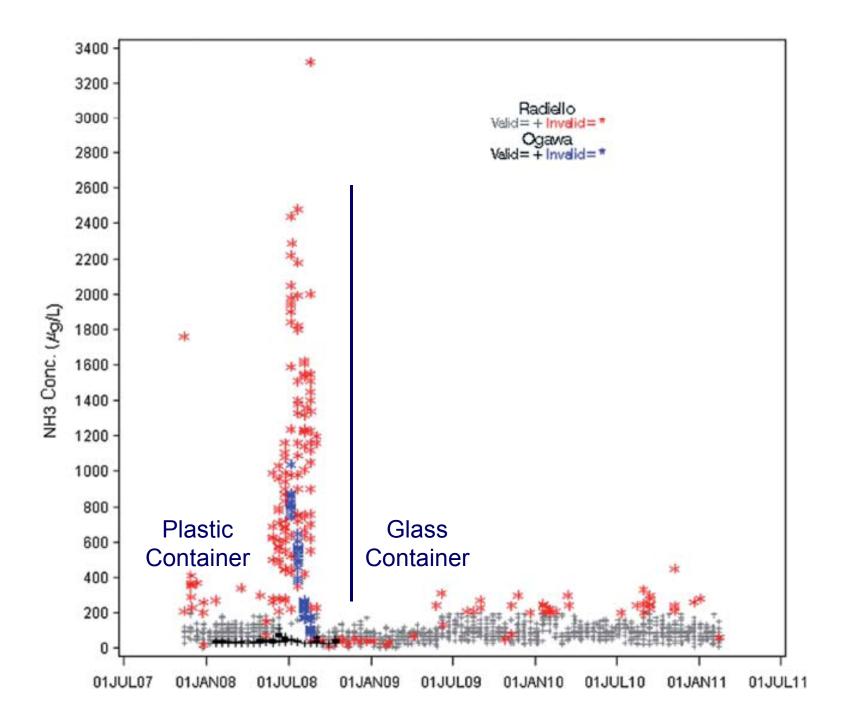


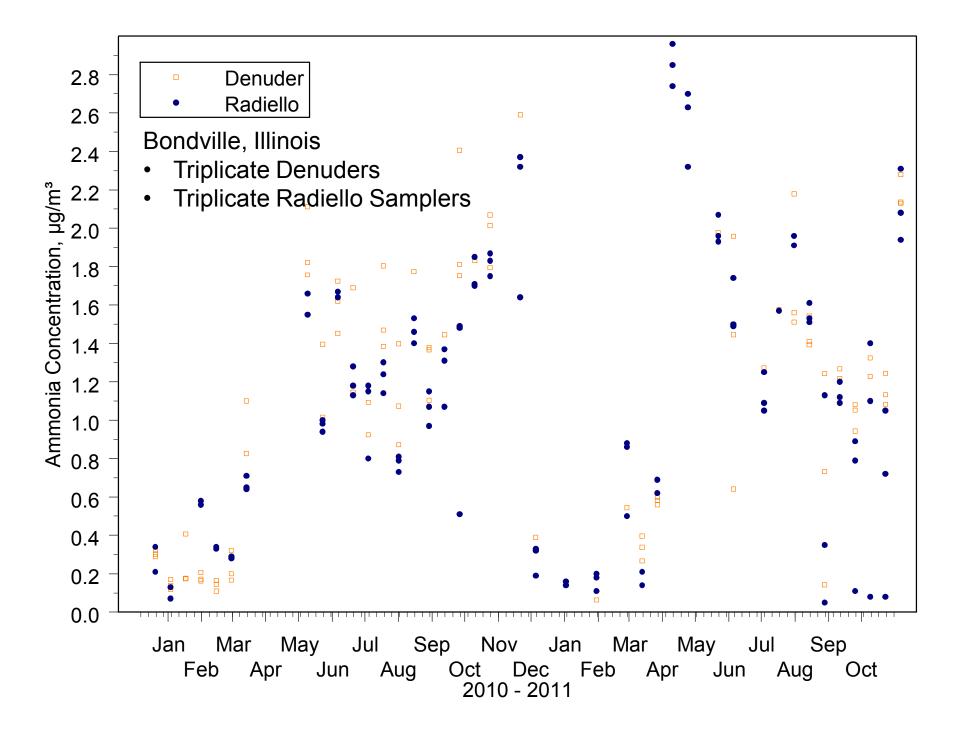


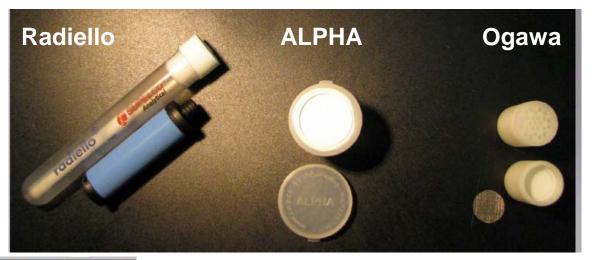
Some Remarks about Passive Air Samplers

Good Practices for Passive Air Samplers

- Must run blanks (laboratory and travel)
- Triplicate samples are best to assess variability
- Should compare passive measurements with a reference method (e.g., denuder, or a continuous analyzer)









CHAMBER STUDY PRELIMINARY DATA (Please do not cite)

	ALPHA	Ogawa	Blue Radiello
Mean Concentration (µg/m³)	15.49	15.54	11.87
Bias (µg/m³)	+1.88	+1.92	-1.75
Standard Deviation (µg/m³)	1.36	1.44	0.32
Travel Blank (µg/m³)	0.40	0.45	0.22

Test Conditions

- Concentration = 13.62 µg/m³
- 99th percentile concentration of AMoN

Action Research.ILLINOIS

College of Fine and Applied Arts | University of Illinois at Urbana-Champaign



- Action Research.Illinois
- ESLARP
- Metro East Citizens Air Project

Contact Information

Resources

Criteria Pollutants and Your Health

Toxics and Your Health

Events and News

Illinois Global Action Research

Outreach Weekend Registration

Check out the Action Research.Illinois Blogspot!

Spotlight . . .

Martin Wolske receives 2011 Library Journal Teaching Award Enriching young minds through community engagement

Metro East Citizens Air Quality Project

The Metro East Citizens Air Quality Project aims to promote community-based efforts to address air pollution by empowering community members as citizen scientists with the knowledge and skills to advocate for improved air quality in their neighborhoods.

This community oriented project will integrate citizen engagement and education programs, monitoring, and research over the next three years. We will focus on drawing awareness at the neighborhood level to the health risks associated with air pollution, while providing citizens with the tools to address local air quality and public health concerns.

Long Term Project Goals

At the end of three years, our project aims to:

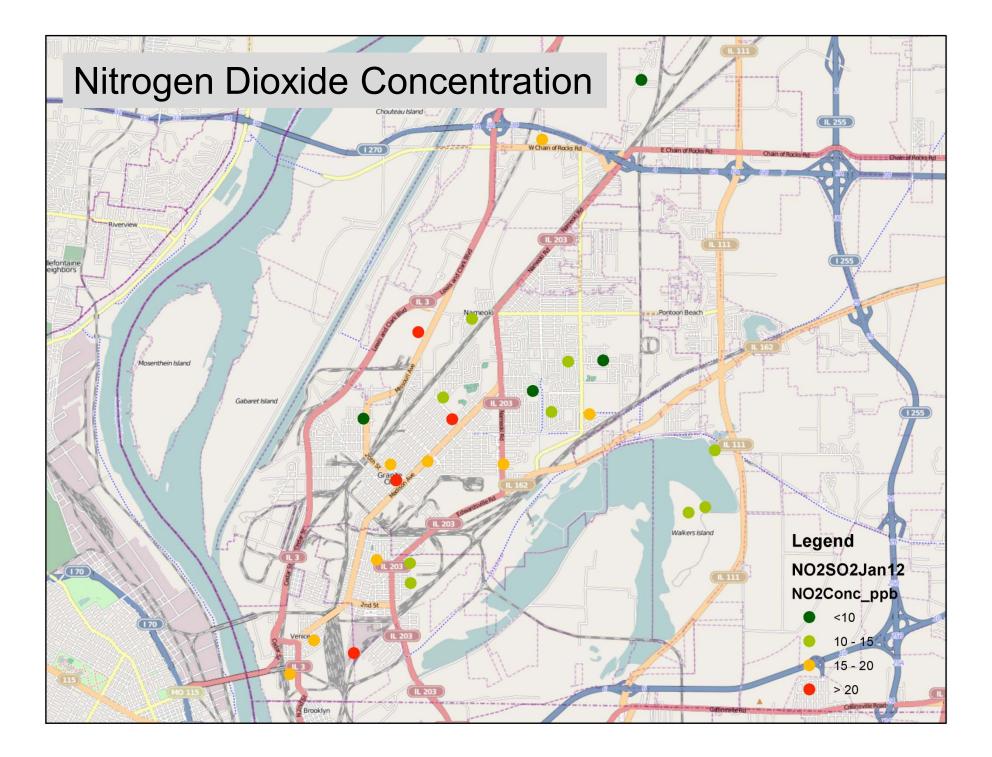
- Establish a solid foundation for citizen activism that will sustain the project as a community led 501(c)3 organization.
- Foster working relationships with citizens and government representatives working on air pollution issues in the Metro East.
- Contribute to joint community/state initiatives to address air quality and public health.
- · Contribute to reductions in local impacts to those at risk populations.

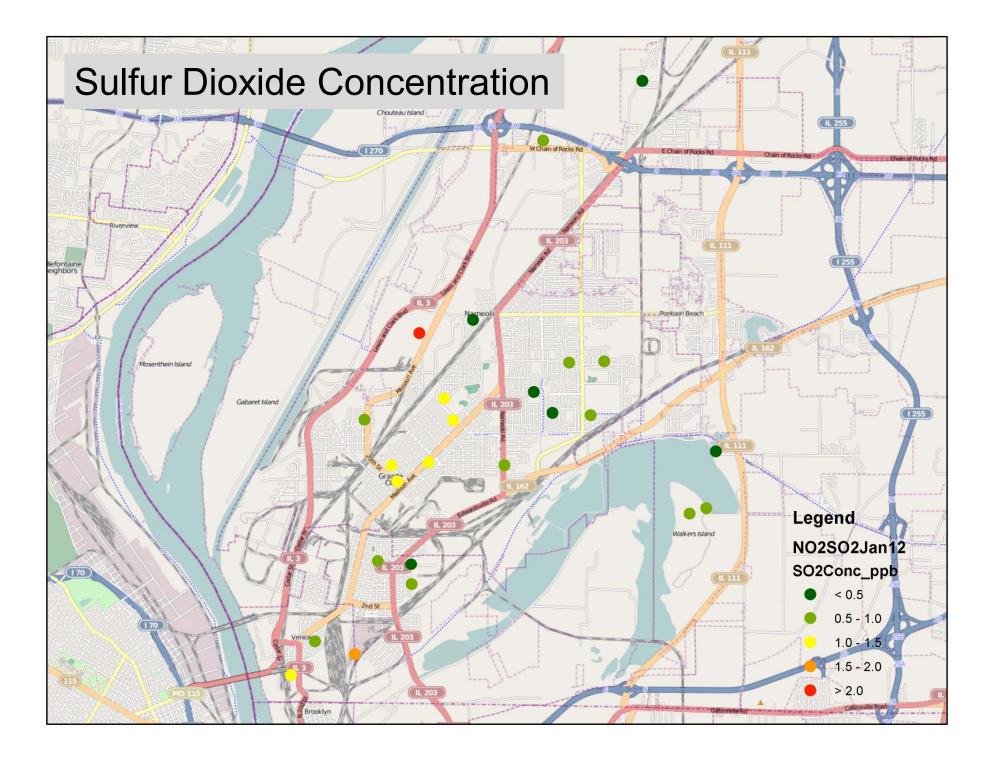
Sign up for daily air quality alerts!

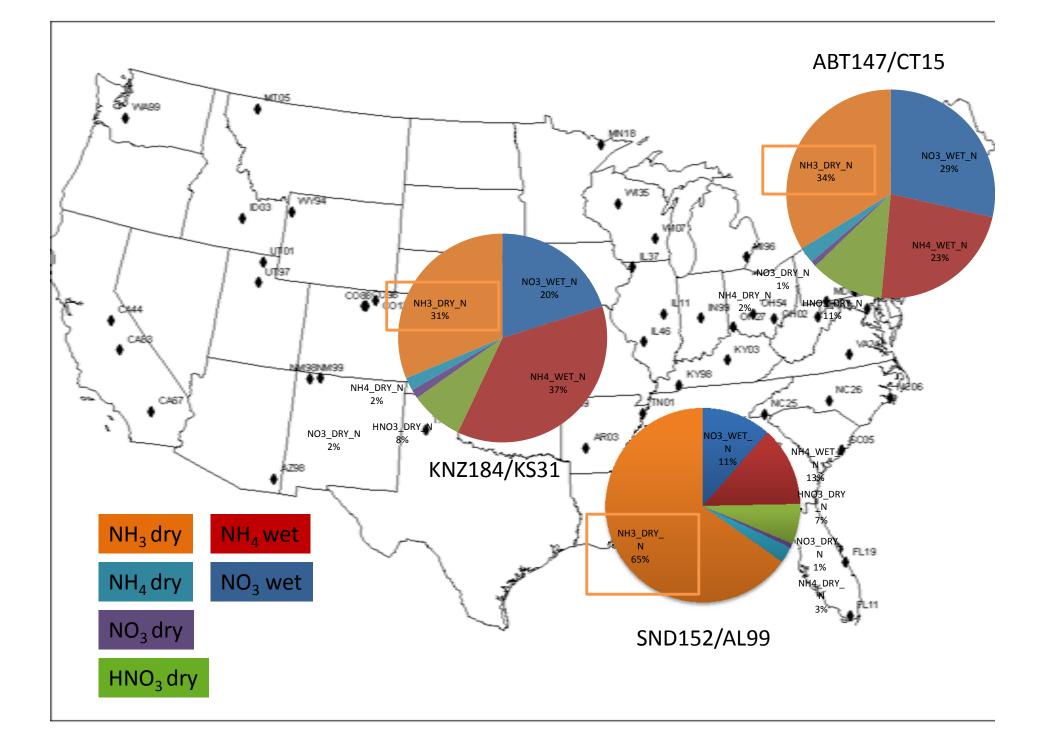


EnviroiTlash provides air quality information such as forecasts and action day notifications via email for your area of interest. <u>Storn-Up</u>

Click here to be added to a listserve to receive AirNews relevant to the Metro East Area

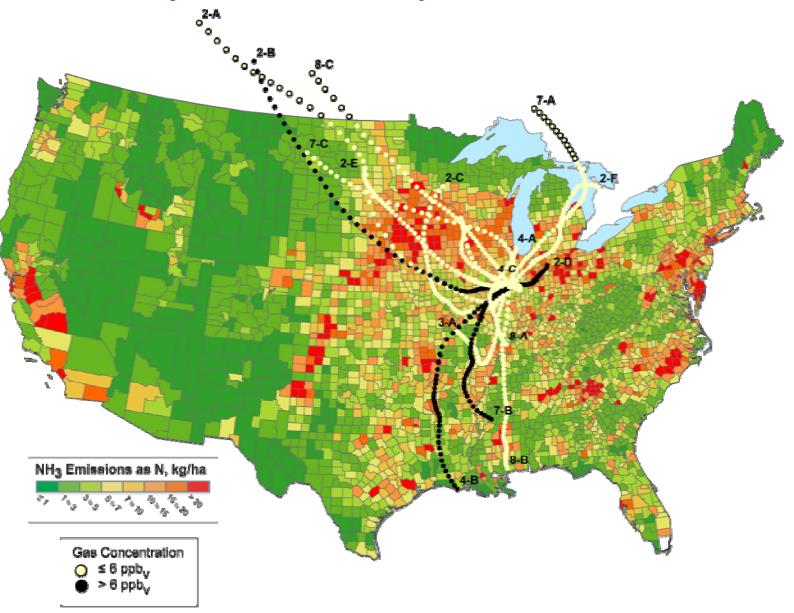




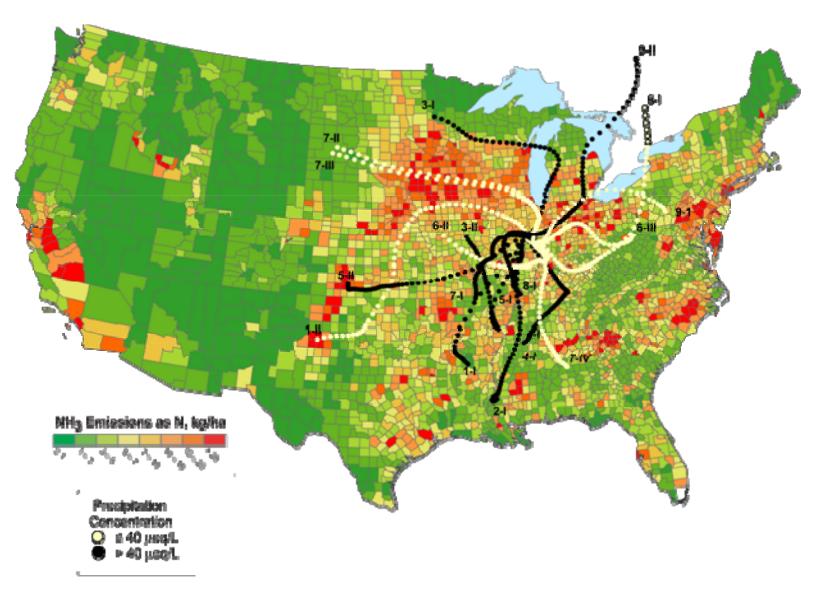


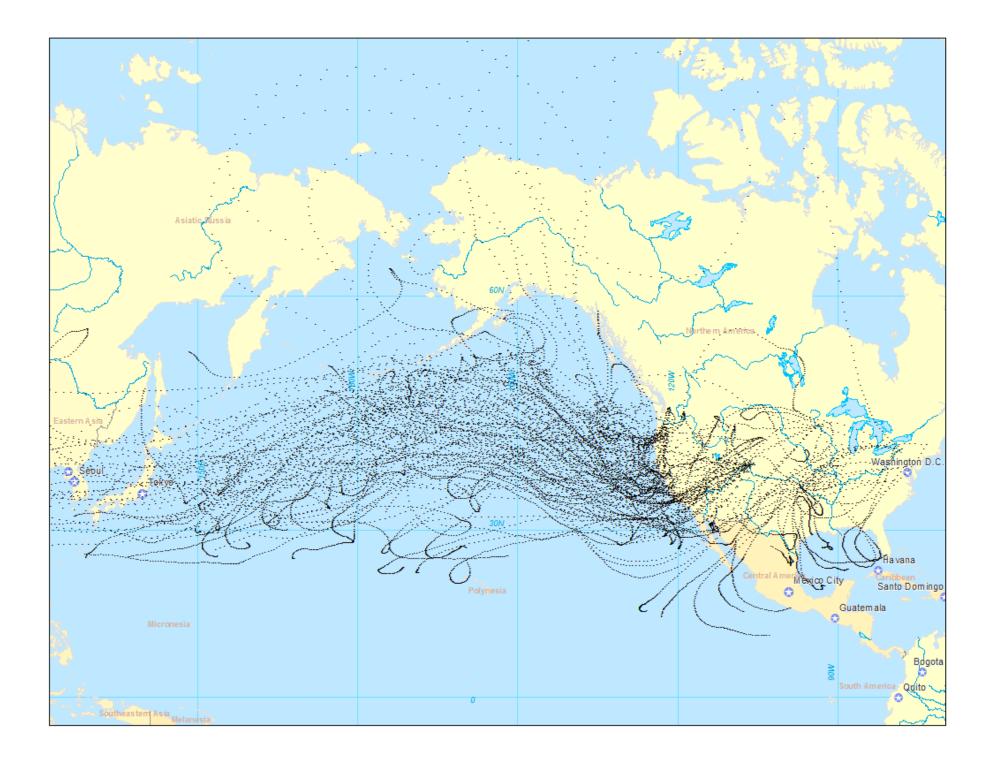
Trajectory Analysis Examples

Back Trajectories – Daily Ammonia Gas Events

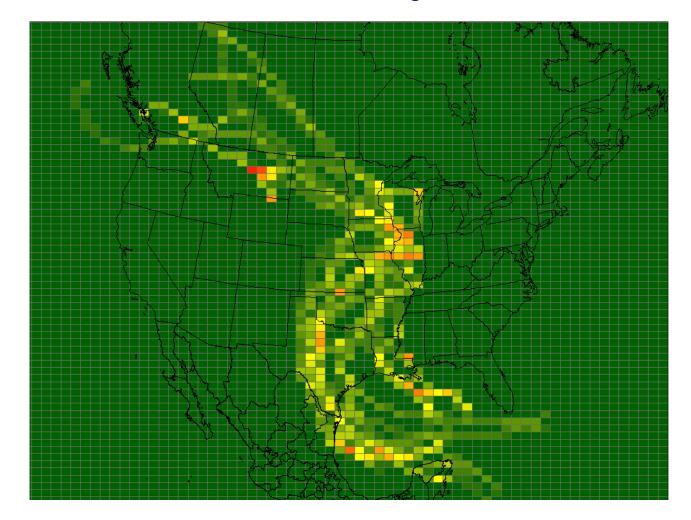


Back Trajectories – Precipitation Events





GIS Analysis



Air Resources Laboratory

Conducting research and development in the fields of air quality, atmospheric dispersion, climate, and boundary layer

Enter search term(s)

Go

ARL site only
 All NOAA
 Advanced Search

ARL Home

HYSPLIT Model

READY

- IN READY News IN Transport & Dispersion
- IN Get/Run HYSPLIT >> IN Volcanic Ash
- Transfer Coefficient Matrix for Fukushima Daiichi
- Short-Range Ensemble Dispersion Forecasts
- Gaussian Plume Model
 Balloon Flight
 Forecasting Tools
- Current & Forecast Meteorology
- North America

HYSPLIT - Hybrid Single Particle Lagrangian Integrated Trajectory Model

The HYSPLIT model can be run interactively on the READY web site or installed on a PC (Mac) and run using a graphical user interface (GUI).

HYSPLIT-WEB (Internet-based)

- Run HYSPLIT Trajectory Model
- Run HYSPLIT Dispersion Model
- Run HYSPLIT for Volcanic Ash
- Spain HYSPLIT
- HYSPLIT for National Weather Service Forecast Offices (NOAA employees only)

PC Windows-based HYSPLIT

- Download Public (unregistered) Version
- Download Registered Version (registration required)
- Graphical Utilities
- Meteorological Data Conversion Utilities
- HYSPLIT-compatible Meteorological Data

2012 PC HYSPLIT Workshop

A 3 day HYSPLIT workshop will be given June 26 - 28, 2012 in Silver Spring, Maryland. The workshop will focus on the use of the recently updated version 4.9 of the model and its PC Graphical User Interface (GUI). Details and registration instructions can be found at: http://www.ertcorp.com/HYSPLIT/ I.

For more information....

http://nadp.isws.illinois.edu clehmann@illinois.edu

