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Some unexpected X-ray interactions with matter

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Some unexpected X-ray interactions with matter

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Outline

• Glass ceramics

• Polymers

• Protein solutions

7.3.3 ALS BM26A ESRF



Energy range

- Most SR experiments used for structure determination 2 < E < 30 keV photons
- Electronic structure studies E < 2 keV

 I will be dealing with structure determination, i.e. photons around 20 keV



X-ray beam

- At modern sources size 20 500 micron
- On special microfocus lines < 2 micron
- Number of photons in beam 10¹⁰ 10¹⁴/sec
- absorption in sample most times ≈ 60%
- This means hardly any heating due to beam (10⁻⁴ K/sec)

This is only relevant at cryogenic temperatures



Radiation interaction

- Elastic/ inelastic scattering
 - All/most photon energy 'leaves' the sample
- Absorption
 - Mostly converted to photo electrons
 - Maximum electron energy is photon energy minus ionisation energy (i.e. E – tiny bit)
 - 30 keV photo electrons do not have sufficient energy to displace atoms in amorphous/crystalline state
 - In liquids/polymers free radicals can be created



Kinetic energy not enough for direct structural change



Confession time!

I know hardly anything about radiation damage

Well, there is this guy Andre....

Why talk here then?

You mean THE Andre???....Say no more.....!!!

I'll stick to phenemological descriptions Sensible! I think this Andre guy owes you a pint....



Do we really deposit energy in samples?

- The beams are intense but not extreme
- The photon energies are not enormous
- They have much less energy than compared with electron microscopy or ion bombardments

• So what is the big deal?



Protein crystallography

- Very popular at the moment
- In principle simple experiment
- Radiation damage issues widely studied
- Can be well controlled when sample are cryo-cooled (≈ 80 Kelvin)
- This freezes the free radicals
- This avoids major structural damage



Estimates of TB burden





TB Crystals **Rv3628 Rv1908c Rv2428 Rv2991 Rv2610** 0 Rv2438c Rv0014c₃₃₁ Rv0014c₂₇₉ Rv0813c Rv0018c **Rv1846c Rv0877** 0 T **Rv2461c Rv2276** Rv2883c ML2640 Rv2667 Rv0733 **Rv**

Protein crystallography

- One of the main applications of SR
- 25% of beam lines at the ESRF are dedicated to it



The Nobel Prize in Chemistry 2009

"for studies of the structure and function of the ribosome"





Thomas A. Steitz

Credits: Micheline Pelletier/Corbis

Ada E. Yonath



Molecular Biology Venkatraman

Ramakrishnan

Protein crystallography







Protein crystal warmed up after experiment



temperature



Ed Mitchel/Sean McSweeney



What happened?

Free radicals become mobile





Not a single pac man but many





Pacman(s) didn't only eat Blinky, Pinky, Inky and Clyde but they eat parts of the maze as well





Glass ceramics



Composite material

 $\begin{array}{c} \text{Li}_2\text{O} \ge \text{Al}_2\text{O}_3 \ge \text{nSiO}_2\\ \text{MgO} \ge \text{Al}_2\text{O}_3 \ge \text{nSiO}_2\\ \text{ZnO} \ge \text{Al}_2\text{O}_3 \ge \text{nSiO}_2 \end{array}$













How to make them?





Lithium disilicate

LiSi₂O₅



Figure 1.2: Layer structure of lithium disilicate crystal phase(Li₂Si₂O₅)[17]. Note: Silicon atoms are hidden in the tetrahedral structure, an example is marked above. Reprinted with the permission of The American Ceramic Society, www.ceramics.org, [copyright 2002]. All rights reserved.



Accidentally developed when looking for materials for missile nose cones





More benign use





S. Donald Stookey 1953

Early version of Corningware



Glass devitrification experiments



LiSi₂O₅



Experiment on 200 micron thick platelet

temperature



time



$I(q) = S(q)^* |F(q)|^2$



Form factor peaks (up to 5th order)

 $\Delta R/R \sim 0.04$

SAXS





Time-resolved SAXS and WAXS

- SAXS
 - Size of particles
 - Crystalline volume fraction
 - Diffusion or reaction limited process
- WAXS
 - Which phases
 - Crystallisation kinetics
 - Etc. etc. etc.

But that is not today's story



Post mortem optical microscopy



Partially crystallised

Real beam size

Fully crystallised







Post mortem powder diffraction





Sample should only be partially crystallised according to recipe Amorphous halo absent



Amorphous halo still present



In irradiated region the sample has crystallised faster then in the non-irradiated region



SEM partially crystallised (not irradiated)





- Sample is in principle bulk crystallising
- There clearly is a textured crystalline layer on the surface
 - Not really all that uncommon
- The degree of crystallinity in surface layers is higher than in the bulk


SEM fully crystallised sample



Irradiated by X-rays

Not irradiated by X-rays



Powder diffraction



Irradiated

Not irradiated



- X-rays on during whole process
 - Fine morphology
 - No surface layer
 - Faster crystallisation
- X-rays on only during crystallisation
 - Texture increased
 - Surface layer less prominent
- X-rays off
 - Coarse morphology
 - (Textured) surface layer
 - Slower crystallisation



- The X-rays influence the crystallisation process
- The X-rays induce crystallisation
- Strongest effect during the thermal nucleation treatment
- Flux 10¹¹ photons/sec in 0.3 x 2 mm²

But:

- The X-rays influence the crystallisation in a larger area than the direct beam
- This only occurs in the vertical direction



What is happening?

- Local heating? Not sufficient energy deposited to influence kT dependent processes (10⁻⁴ K/s)
- Most likely due to electrons liberated in sample (photo electric effect)
- One way or another these electrons help to create nucleation sites





Mechanism of electron-irradiation-induced

recrystallization in Si

Frantz, J. Tarus, K. Nordlund, and J. Keinonen PHYSICAL REVIEW B, VOLUME 64, 125313

For electron irradiation, it has been recently shown that even quite low energy (25 keV) electron bombardment can produce recrystallization of amorphous pockets in silicon, germanium, and gallium arsenide.

Doses low enough to avoid sample heating due to electron beam





By geometric rearrangement we mean that after a bond between atoms *i* and *j* breaks, the atom *i* can reform the bond with some other atom *k* in the local neighbourhood and similarly with *j* and some other atom *I*). Although this is quite unlikely to occur in the crystalline state, in an amorphous state produced by irradiation there are weak bonds which are relatively

easy to break.

Frantz, J. Tarus, K. Nordlund, and J. Keinonen PHYSICAL REVIEW B, VOLUME 64, 125313



But...

- This would not explain the vertical extension of the affected region
- This is around 200 micron (either vertical side of beam)
- The path length of scattered electrons is at most microns



But...

- 10 keV photons can travel around 200 micron in this sample
- The synchrotron photons are polarized
- They will be scattered in the vertical direction (both elastic as well as inelastic)



Beam direction view

Side view sample Max photon penetration depth WAXS SAXS



What is most likely:

- Part of photons in direct beam get absorbed and release electrons
- Part of photons scatter inside sample
- These ultimately get absorbed and release electrons









Earlier work

- On-line lysozome crystallisation
- Radiation damage in beam spot
- 8 keV photons
- No crystallisation in region of 500 mm above/below beam
- Hydrolysis products don't 'travel' that far (2 micron)



Quantum dots





Just for aesthetics a second picture





Quantum dots

- Interesting for their optical properties
- Small size (≈ 1 nm)
- Preferably monodisperse, i.e. all the same size
- Can be made with thermal treatments of glass
- PbS in silica glass matrix is well studied system



Real time SAXS



That looks ok!

- Particles grow
- Finite size
- Very monodisperse

• Everything is ok, isn't it?

WAXS/powder diffraction

The difference between exposed-not exposed

The conclusion:

- PbS quantum dots grow
- Scattering pattern dominated by metallic lead particles!!
- Only small volume fraction but large contrast
- Metallic Pb is not found outside of the irradiated spot.
- Metallic Pb R = 40 nm, PbS R = 2 nm

Polymer processing

Semi crystalline polymers

Polyethylene, polypropylene, nylon, PVC,
i.e. 95% of the world polymer consumption

polyethylene

Molten it looks like spaghetti

In solid state semi-crystalline - amorphous parts - crystalline parts

The ratio amorphous/crystalline and the spatial orientation determine the macroscopic properties DUBBL

Heating/cooling cycle on HDPE

SAXS (lamellar spacing)

WAXS (crystalline order)

Fundamental question

- What is the crystallisation process?
 - Nucleation and growth
 - Spinodal decomposition like
- Nucleation and growth
 - Lamellae and crystallinity develop at same time
 - SAXS and WAXS signals increase simultaneous
- 'Spinodal'
 - Long range density fluctuations first
 - SAXS signal increases before WAXS

On-line WAXD during quenching

The hidden mesophases

30 °C/s

100 °C/s

But what happens when one is not careful.....

$$\overline{M}_{w} = x$$

Problem:

- One finds this back in the DSC data
- This means that instead of being a local 'problem' the total affected volume is much larger then the irradiated area

- Microtubule (rigid rod) polymer solutions
- Beam spot 300 micron vertical
- Radiation damage around 1 mm vertical direction

Fibre diffraction pattern











Worrying detail

- Calculations show only 0.1 weight% of material is damaged
- In most cases this would not change macroscopic morphology of sample
- Here it definitely does



Further results in the literature

- Au particles formation and Ostwald ripening
- Surface crystallisation at cryogenic temperatures
- Water evaporation due to lowered surface tension
- And probably more that I'm not aware of...



Prof. Rontgen says:

Don't underestimate the power of my rays!

Thanks for your attention





So:

- Even in hard condensed matter samples 10 keV photons can screw up a time-resolved experiment
- Not only direct footprint area of beam affected but area perpendicular in photon polarization direction as well









