# In-situ XRF analysis as a diagnostic analytical tool in the conservation field

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## Outline and the PROMET analytical campaigns across the Mediterranean

#### **Outline**

- 1. The PROMET project
- 2. The micro-XRF mobile instrumentation
- 2. Accuracy and pitfalls of micro-XRF analysis
- 3. PROMET campaigns:
- Ancient Messene (2006)
- Malta, Armoury Palace (2006)
- Damascus National Museum (2007)
- Numismatic Museum, Yarmouk University, Irbid, Jordan

### PROMET FP6:2005-2008



#### Aim:

To develop Prototype innovative and advanced analytical methods to survey large collections of metal objects in-situ, making it possible to pinpoint conservation needs without any risk of damaging the artefacts

Efficient, versatile and mobile analytical methodologies: Micro-XRF and Laser Induced Breakdown Spectroscopy

LIBS related tasks were carried out by Prof. D. Anglos FORTH-IESL, Crete

**24 partners**, including Turkey, Syria, Jordan, Morocco, Italy, France, Spain, Czech Republic

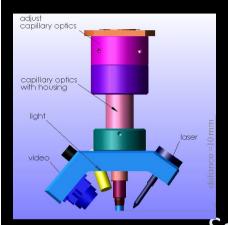


### **Demokritos objectives within PROMET:**

- ✓ To develop, optimize and calibrate the analytical performance of an innovative portable micro-XRF spectrometer
- ✓ To develop and improve analysis procedures, protocols and the standardization of the method
- ✓ To apply the micro-XRF spectrometer for systematic technological and conservation related studies of museum metal collections at the Mediterranean region:
- The study of the manufacture technology of metal alloys
- Non invasive characterization of corrosion products
- Contribution to the assessment of innovative protective coatings

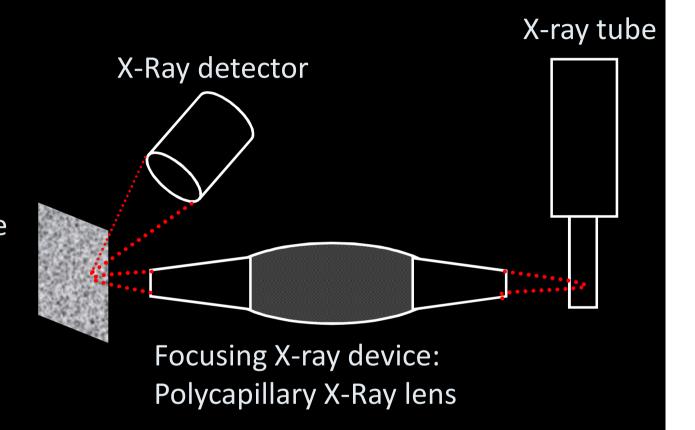


### μ-XRF spectrometer: Principle of operation



Development and Sample application of portable micro-XRF unit

ustomized design of ARTAX by Bruker Nano AXS







### Versatility: In-situ Micro-XRF analyses



Damascus National Mus eum, Syria, October 2007



Laboratory test of TEI coupon

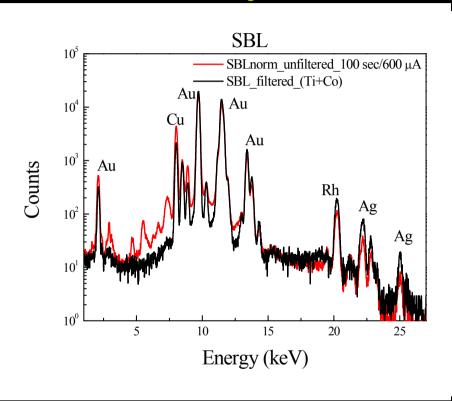


Numismatic Museum of Yarmouk University Irbid, Nov. 2008





### Pitfalls: Interference of XRF signal with diffraction peaks, QC/QA of micro-XRF data



- Diffraction peaks
- ✓ Heterogeneity at the micro-scale
- Definition of the scanning area that represents the alloy bulk composition

#### **Alloys** Filters/ Thickness (µm)

Ti  $(23.6 \pm 0.2)$ Co  $(17.7 \pm 1.3)$  $Pd (11.3 \pm 0.3)$ 

Gold X

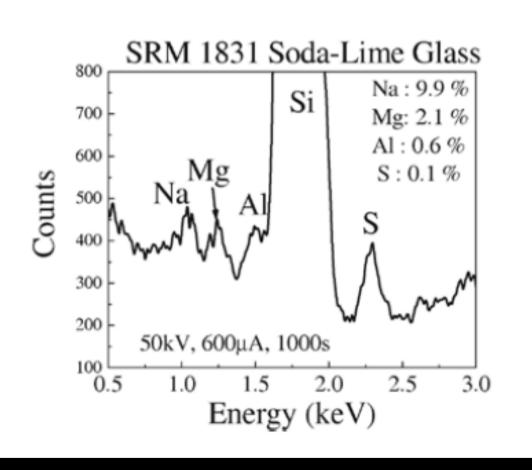






### Analytical range using He atmoshpere

50kV, 600mA, 100s



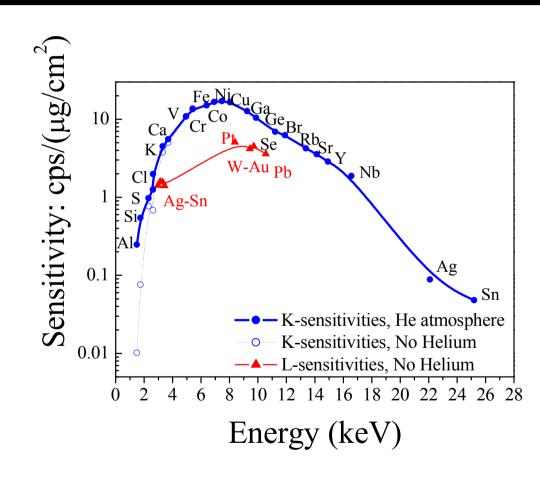
The improvement in the intensity of Al-K and Si-K characteristic X-ray lines is significant, 22 and 7.3 times, respectively.





### **Analytical performance: Elemental sensitivity**



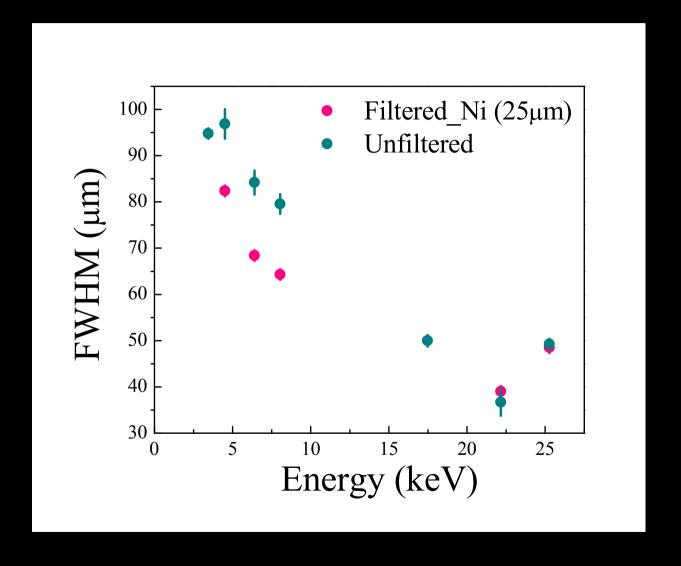


600 μA 50 kV





### **Analytical performance: Spatial resolution**





### **Calibration methodology**

$$I_i(E_k) = G \cdot w_i \cdot \left( \int I(E) \cdot T(E) \cdot \sigma_i(E, E_k) \cdot A_i(E, E_k) \cdot F_i \cdot dE \right) \cdot f_{air}(E_k) \cdot \varepsilon_d(E_k) \cdot \frac{1}{\sin \theta_1}$$

where:

$$A_i(E, E_k) = \frac{1 - \exp[-\mu_{tot}(E, E_k) \cdot \rho X]}{\mu_{tot}(E, E_k)}$$

$$\mu_{tot}(E, E_k) \equiv \sum_{i=1}^{N} w_i \cdot \left( \mu_i(E) / \sin \theta_1 + \mu_i(E_k) / \sin \theta_2 \right)$$

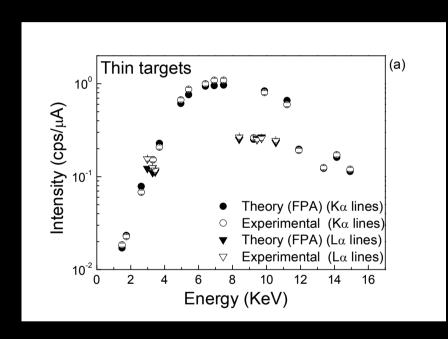
$$F_i = 1 + \sum_{j=1,n} SF_{ij} \left( 1 + \sum_{k=1,n} SF_{jk} \right)$$

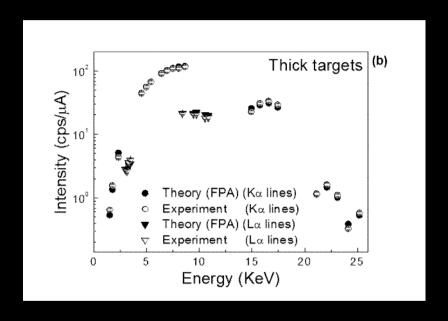
$$T(E) = \left(\sum_{i=0,5} A_i \cdot E^i\right) \cdot \exp(-c \cdot E)$$





# Experimental/simulated pure element thick/thin elemental intensities



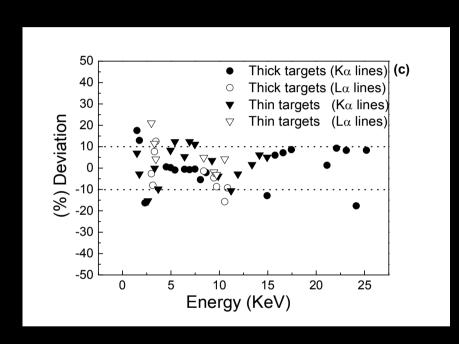


Kantarelou et al., XRS, 2015

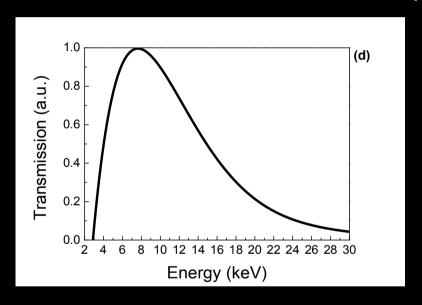




# Results of the fitting procedure



#### Estimated Lens transmission efficiency

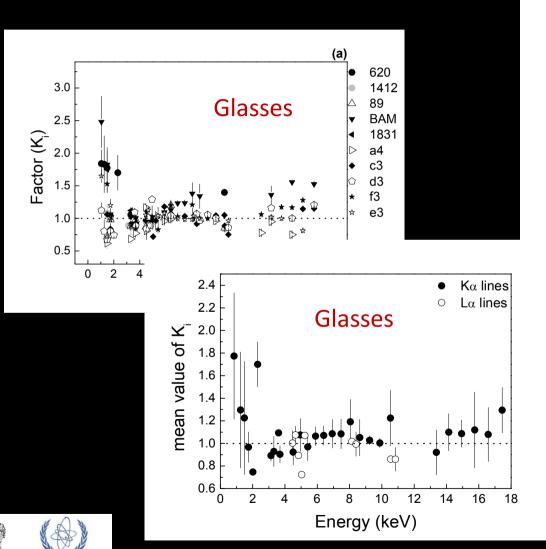


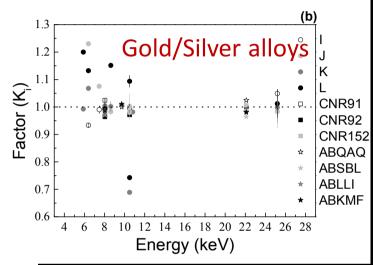
Kantarelou et al., XRS, 2015





# Accuracy/Quantification of CH related materials

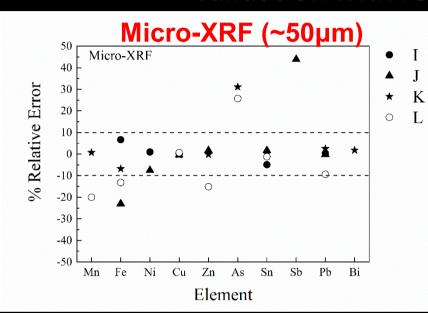


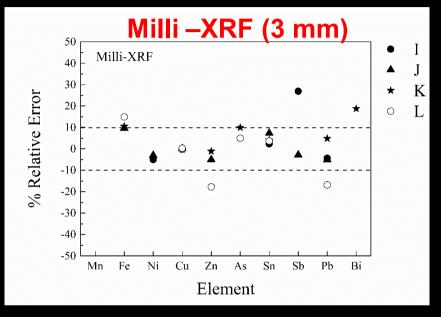




### **Assessment of micro-XRF analysis accuracy**

#### Validation with respect to Cu based RMs





A. Heginbotham et *al.*, An Evaluation of Inter-Laboratory Reproducibility for Quantitative XRF of Historic Copper Alloys, Proceedings of the International Conference on Metal Conservation, METAL 2010, pp 178-188, Edited by Paul Mardikian, Claudia Chemello, Cristopher Watters and Peter Hull, 11-15 October 2010, Charleston, South Carolina, USA





# Semi-QA and Diagnostic Micro-XRF Analysis Methodology:

Variation of the K/L or L/M elemental intensity ratios in single spot, line or area scan measurements

Filtered excitation

Analysis of corroded area vs corrosion free area

Line and area scans to obtain in reasonable measuring time  $(1x1 \text{ mm}^2, 50 \text{ } \mu\text{m} \text{ step}, 10\text{s/step}, ^21.5\text{h})$  intensity maps of the detected characteristic X-ray lines



### Semi-QA and Diagnostic Micro-XRF Analysis

#### **Results obtained:**

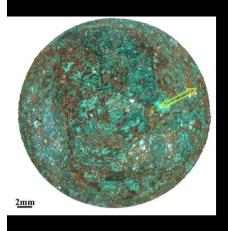
- ➤ Identification of the spatial coexistence of different elements, fingerprints of certain corrosion products or of manufacture techniques.
- Estimation on a semi-quantitative basis of the elements enriched or depleted from the surface
- ightharpoonup Rough estimation of the depths that a certain element is located, namely, on the surface, near surface (~2-10  $\mu$ m) or below ~10  $\mu$ m.
- > Spatial distribution of individual elements
- ► Identification of the presence of certain minor to trace elements that may support provenance and manufacture studies of the metal

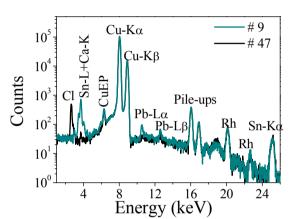
  Andreas Karydas, ICTP, 14th of July 2015



### **Analysis of Copper coupon corrosion**

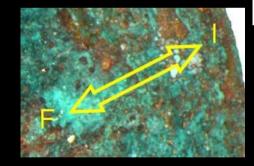
products
Artificially and naturally aged bronze coupon: (Cu: 91.3%, Sn: 7.5%, Pb: 1.0%)



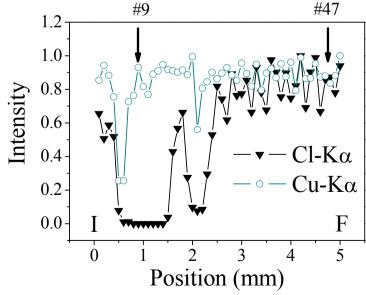


#9 : green area

#47: pale green area



50kV, 600μA, 30s/step,0.1mm/step, 50 measurements



### Analysis of metal corrosion products

Silver coupon (prepared-characterized by Prof. G. M. Ingo,

Polytechnico of Milano)

A – Green A - Paratacamite

B - White B - Chloroargyrite

C - Black C - Silver (oxide)

2 mm

Artificially and naturally

aged silver



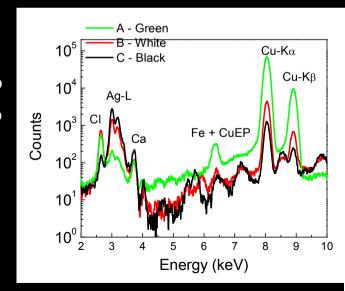
coupon:

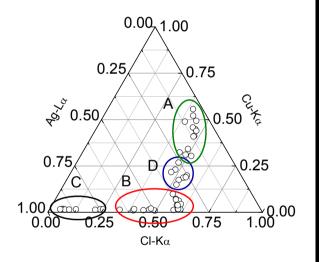
Ag: 92%

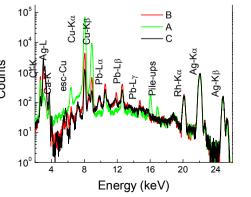
Cu: 6.5%

Pb: 1.5%

50kV, 600μA, 30s/step, 0.1mm/step, 50 measurements







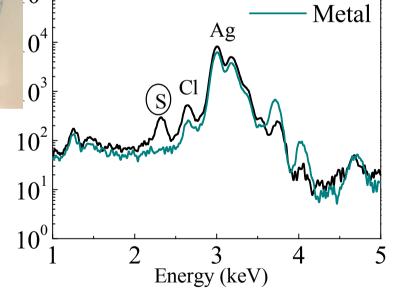
A.G. Karydas et al, PROMET Book, 2008

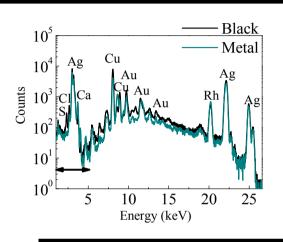
### Au R Cui

# Damascus Archaeological museum: Analysis of silver tarnishing

Silver Bowl 1400 -1300 BC Late Bronze Age Thickness of the layer:  $^{\sim}$  0.5  $\mu m$ 







Tarnish: corrosion mainly caused by the sulfur in the air

Black



### PROMET, Damascus, Syria: **Gilded Bronze figurines**

Late Bronze Age, 1400 B.C. Ugarit site: Issues addressed

- ✓ Manufacture technology (compositional analysis, raw materials)
- ✓ Gilding technique
- ✓ Identification of corrosion products









Kantarelou et al., JAAS, 2015





al Gods





### Compositional analysis of bronze metal

#### **Concentrations (wt. %)**

Fe:  $0.26 \pm 0.03$ 

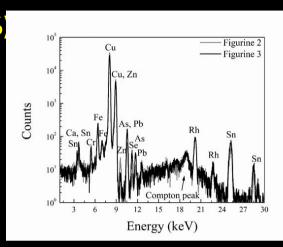
Cu: 92.5 ± 1.0

Zn: 0.60 ± 0.06

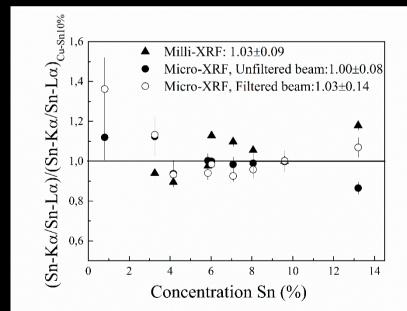
As:  $0.10 \pm 0.01$ 

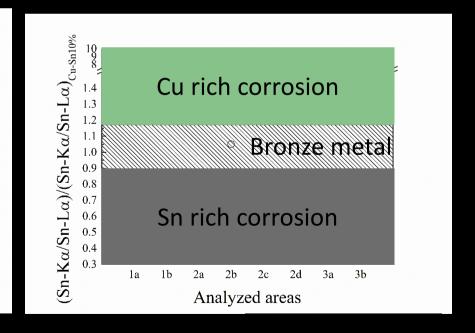
Sn: 6.32 ± 0.30

Pb: 0.040 ± 0.004



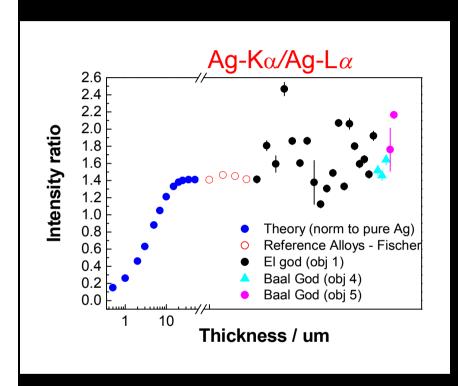
$$G_{S_{R}} = \frac{\left(I_{S_{R}}(K_{\alpha}) / I_{S_{R}}(L_{\alpha})\right)_{S}}{\left(I_{S_{R}}(K_{\alpha}) / I_{S_{R}}(L_{\alpha})\right)_{R}} = \frac{\mu_{R}(E_{1}, E_{K\alpha})}{\mu_{S}(E_{2}, E_{K\alpha})} \times \frac{\mu_{S}(E_{3}, E_{L\alpha})}{\mu_{R}(E_{4}, E_{L\alpha})} \times \frac{(1 + SE_{T})_{R}}{(1 + SE_{T})_{S}}$$

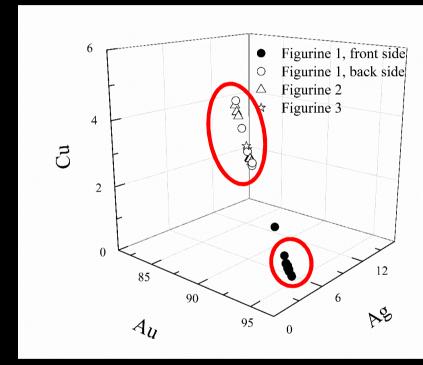






### Compositional analysis of gold foil





Thickness of the gold foil > 10 micrometers

Two (2) main compositional groups:

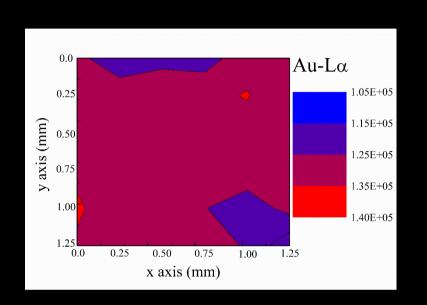
Ag: **14-15%** or **4-5%** 

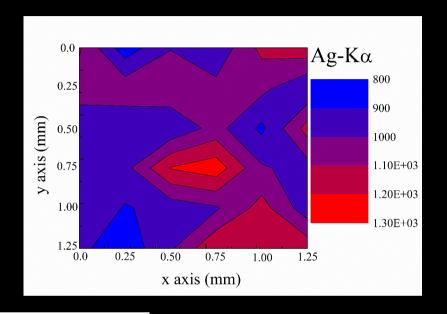
Cu: generally <2-3%

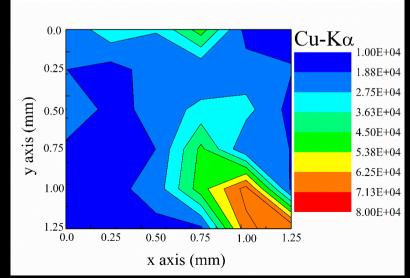




# Imaging of the elements distribution on the gold foil











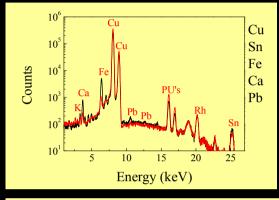
### PROMET, Damascus, Syria: Corrosion products

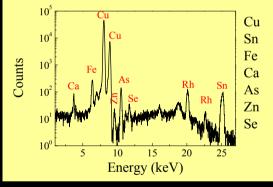
Eail God, Late Bronze Age 1400-1300 B.C.

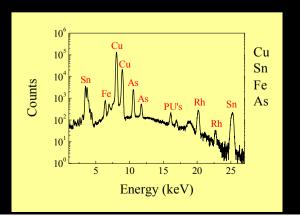
















Cu: 76.08%

### PROMET at Irbid, Jordan

Zn: 21.7%

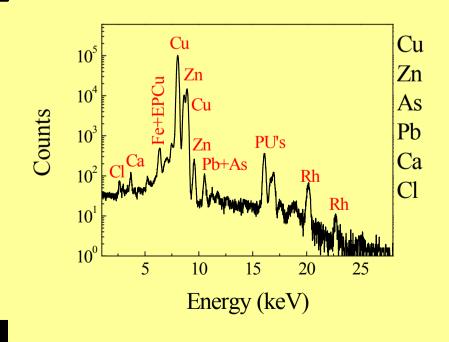
# Ni: 0.80% Sn: 0.43% Pb: 0.43% Fe: 0.39% As: 0.17%

#### **Umm Qais artefacts**

Copper-base bracelet, #216, (Ottoman Period), General + Pitting + Crevice Corrosion, Active corrosion have caused serious crack

A. Arafat et al., Journal of Cultural Heritage, <a href="http://dx.doi.org/10.1016/j.culher.2012.07.003">http://dx.doi.org/10.1016/j.culher.2012.07.003</a>, 14 (3), (2013) 261-269





### **PROMET in Ancient Messene - Objectives:**

Compositional analysis of different typology copper based artifacts through time

Surface characterization of high tin bronze mirrors

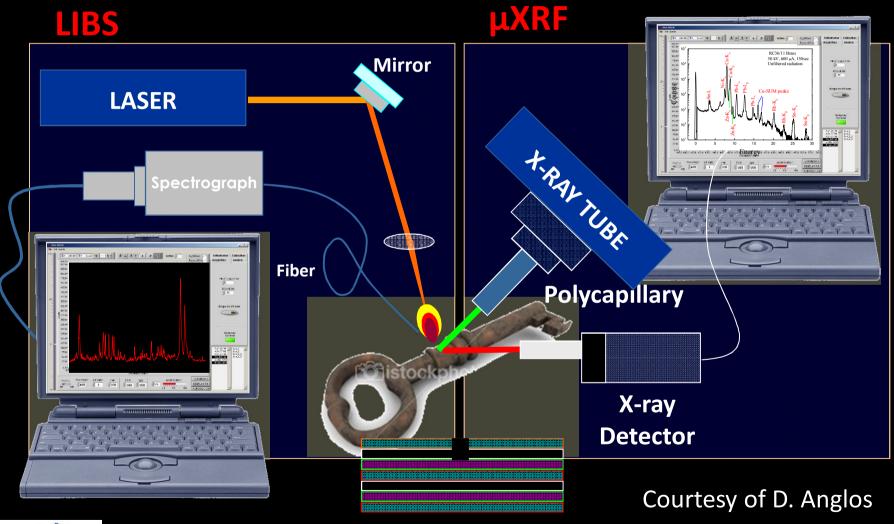
Combined microXRF and LIBS analysis for enhancing in-depth elemental distribution





### **Dynamic combination of micro-XRF and LIBS**

V. Kantarelou et al., METAL-07, Vol. 2, Innovative investigation of metal artifacts, pp. 35-41, (2007)

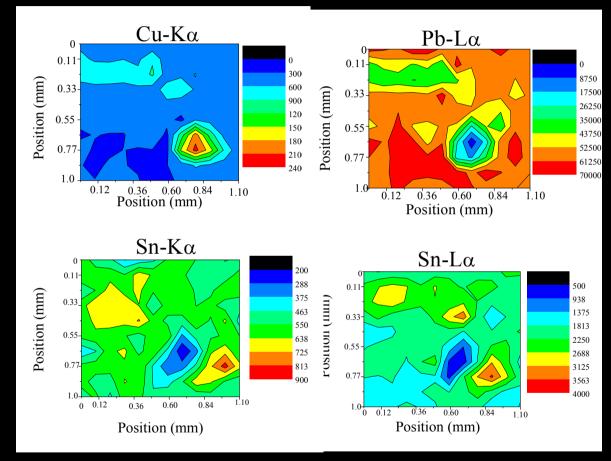




### **PROMET at Ancient Messene**



Micro-XRF elemental mapping of the LIBS ablated area









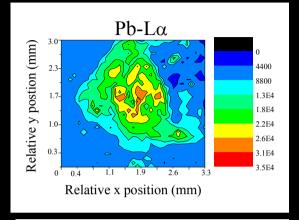


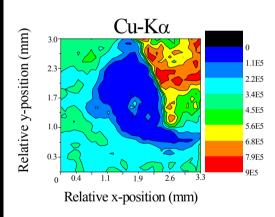
### **PROMET at Ancient Messene**

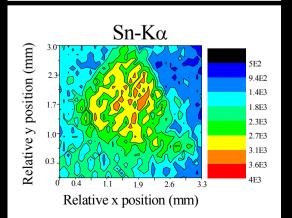
 $\mu$ -XRF analysis of High Tin Bronzes











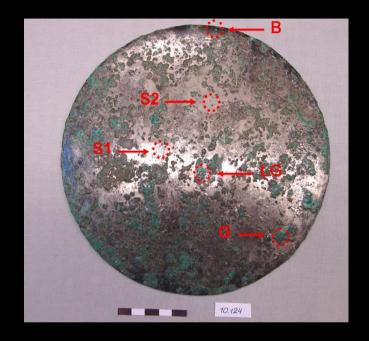


# Ancient Messene, Greece: Manufacture of high tin bronze mirrors





2nd c. BC



Mirror 1 (M1)

(Cu: 70.6%, Sn:26.0%, Pb:3.4 % )
Examined areas : metal, black, silverish

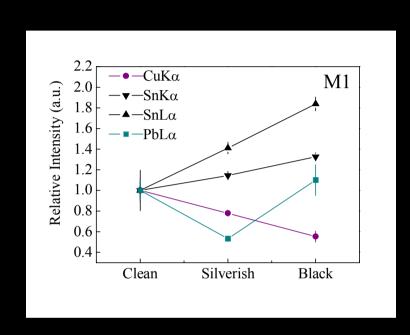
Mirror 2 (M2)

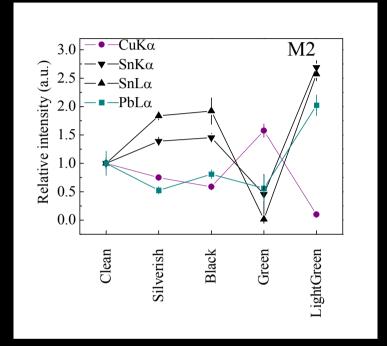
(Cu: 70.2%, Sn:22.9 %, Pb: 6.8 % ) Examined areas : metal, black, silverish, green and light grey



# Ancient Messene, Greece: Surface examination of high tin bronze mirrors







Silverish: Increase of SnL $\alpha$ , decrease of CuKa and PbLa : Sn surface enrichment

Black: Significant increase of  $SnL\alpha$  (Cassiterite?)

Green: Increase of CuKa, Cu corrosion products (malachite?)

Light green: Increase of PbL $\alpha$ , SnL $\alpha$ , decrease of Cu-K, (Lead-white? Cassiterite?)



# Ancient Messene, Greece: Black patina on high tin bronze mirrors



10.010

Sn-Lα

Sn-Lα

Sn-Lα

Sn-Lα

Sn-Lα

Sn-Lα

Sn-Lα

Sn-Lα

Silverish

Layer Thickness (μm)

Silverish  $< 1 \mu m$ 

Black layer with a thickness of very few micror



### PROMET at Armoury Palace, Malta

#### Conservation related issues addressed:

- ✓ Authenticity issues
- ✓ Gilding technology
- ✓ Compositional analysis of armory components (rivets)
- ✓ Identification of surface corrosion products

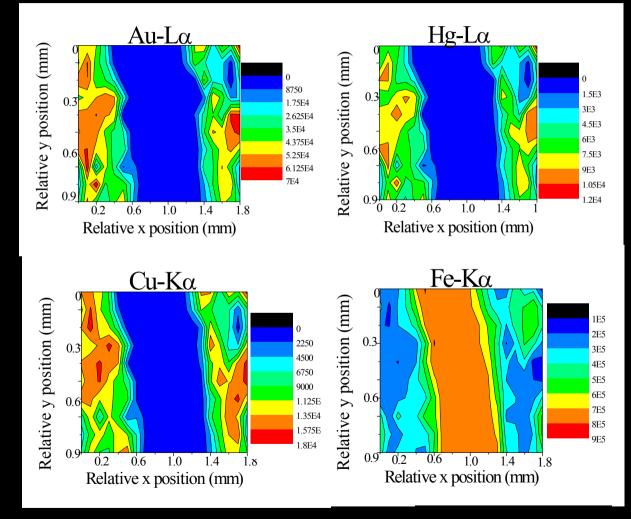




### PROMET at Armoury Palace, Malta

Gilded Iron Alloy Falling Buff  $\mu$ -XRF analysis of gilding areas



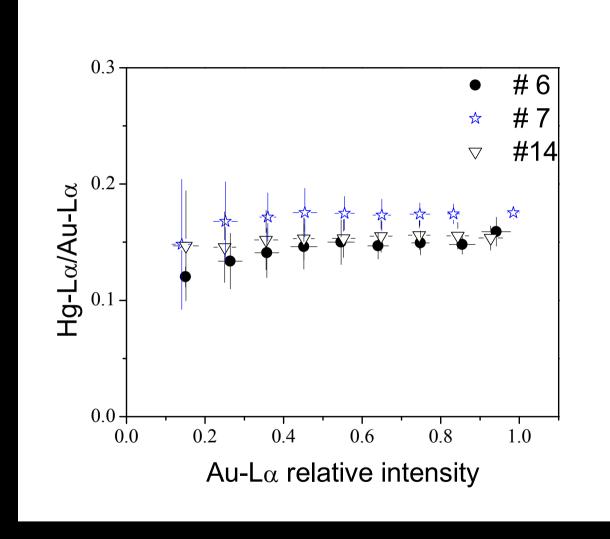




### Fire-Gilding technique: Hg to Au ratio

Palace Armoury, Malta

The Au/Hg mean values were deduced among a subgroup of the area map spots, where the Au intensity varies between 0-10%, 10-20%, 30-40% etc with respect to its maximum value





C. Degrigny et *al.*, Metal-07, pp. 26-34, (2007)

### **Conclusions**

Micro XRF analysis can offer fast elemental distribution maps, contributing thus both towards the identification of surface corrosion products and manufacture techniques as well

The operating conditions of the micro-XRF spectrometer require careful optimization per type of samples analyzed, that it is in many cases not a trivial and straightforward procedure.



### **PROMET Impact**

• The results of the project offer a cost effective approach in identifying the conservation problems and needs of a metals collection using portable diagnostic techniques

• The achieved results can easily be applied to any museum setting world-wide due to the portability of the instruments developed whereas analytical methodologies are easily transferable.



### **Acknowledgements**

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Museum of Ancient Messene, S. Polymenea, Conservator

National Museum at Damascus, Syria

Luda Mahfoud, Abeer Kurdab, Mayada El Saadi and Kasem Yahya

Directorate General of Antiquities and Museums, Mr Maher Azar

Malta Heritage, Dr. C. Degrigny, Dr. S. Golfomitsou, D. Vella

Yarmouk University, Dr Manar Bani Hani, Lina Khrees, Prof. Mohammed S. Shunnaq, Prof. Ziad Al-Saad

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Umm Qais Museum, Mohammad Bashabsheh

