Terahertz and Water

Winter College on Optics: Terahertz Optics and Photonics ICTP, Trieste, 6 February 2023

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Terahertz and Water: outline

1. Terahertz time-domain spectroscopy

- 1. Examples of non-linear responses
- 2. Linear and non-linear spectroscopy
- 3. Optical rectification (OR) and electro-optical sampling (EOS)
- 4. Models to obtain the optical properties (Fresnel equations)

2. Water and solvation

- 1. Infrared absorption modes of liquid water
- 2. Introduction to solvation science
- 3. Study ions in water with THz: rattling, hydration layers
- 4. Gold nanoparticles
- 5. Long-range protein effects on water
- 6. Non-linear terahertz response of water

A non-linear response you can touch

Hooke's law, linear response

The elongation, x, of a rubber band is proportional to the applied force, F

1. $z = k \cdot F$





A non-linear response you can touch

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The elongation, x, of a rubber band is proportional to the applied force, F

1. $z = k \cdot F$

...only up to a certain value of F !

beyond that, we can Taylor-expand k(F)

2. $k = k(F) = k_1 + k_2 \cdot F + k_3 \cdot F^2 + ...$

and we have a non-linear response of the elongation w.r.t. the applied force

3. $z = k_1 \cdot F + k_2 \cdot F^2 + k_3 \cdot F^3 + ...$

Typically, the non-linear terms are small and important only for large applied force



A non-linear response you can hear



Apply an AC voltage, the membrane deforms accordingly, oscillating. This modulates the air density, which we hear.

0

2

Low input voltage

2 -500 Hz -2 2 8 0 4 6 ms

Membrane deformation

4

ms

8

6

A non-linear response you can hear



A non-linear response you can hear



500 Hz, low distortion



500 Hz, high distortion



<u>https://www.youtube.com/watch?v=7dLArMd-y64</u> Julian Krause <u>https://www.youtube.com/@JulianKrause</u>

A nonlinear response you could see



A nonlinear response you could see



A nonlinear response you could see



https://solarsystem.nasa.gov/solar-system/sun/overview/

Nonlinear optics allows to *make new* colors! Example: SHG



P.A. Franken, et al, Physical Review Letters 7, p. 118 (1961)

http://www.vincentkemlin.website/?page_id=140 https://en.wikipedia.org/wiki/Second-harmonic_generation

Linear spectroscopy, elastic response: output frequency = input frequency



Linear optics

- In general one expects that when an electric field E is applied on a material, it excites dipoles and induces a polarization density P, i.e. P = f(E)
- For linear optics in a material that is <u>dispersionless</u>, <u>lossless</u>, <u>isotropic</u>, and <u>without free charges</u>:

$$\boldsymbol{P}(t) = \varepsilon_0 \chi^{(1)} \boldsymbol{E}(t)$$

- $\varepsilon_0 \sim 8.854 \cdot 10^{-12} F/m$ vacuum electric permittivity
- $\chi^{(1)}$ is the first-order electric susceptibility (scalar!)

Linear optics: Lorentz model

- The Lorentz model allows to find an expression for the susceptibility $\chi^{(1)}$

An atom can be thought of as positively a charged nucleus that is surrounded by an negatively charged "electron cloud". Let x(t) be equal to the distance between the nucleus and the "centre of mass" of the electron cloud. Then if the charge is denoted by q, the atomic dipole moment is simply qx(t).



Linear optics: Lorentz model

A classical damped and driven harmonic oscillator



Optical Properties of Solids, Wooten, Ch.1, <u>https://doi.org/10.1016/C2013-0-07656-6</u> https://en.wikipedia.org/wiki/Harmonic_oscillator

Linear optics: Lorentz model

• If $E(t) = E(\omega)e^{-i\omega t}$ and $x(t) = x(\omega)e^{-i\omega t}$, write $x(\omega)$ as a function of $E(\omega)$

$$\ddot{x}(t) + 2\gamma \dot{x}(t) + \omega_o^2 x(t) = \frac{qE(t)}{m}$$
$$\frac{d^2 x}{dt^2} = (-i\omega)^2 x(\omega) e^{-i\omega t} = -\omega^2 x(\omega) e^{-i\omega t}$$
$$\omega^2 x(\omega) e^{-i\omega t} + 2\gamma(-i\omega) x(\omega) e^{-i\omega t} + \omega_0^2 x(\omega) e^{-i\omega t} = \frac{qE(\omega)e^{-i\omega t}}{m}$$
$$x(\omega) = \frac{q}{m} \cdot \frac{E(\omega)}{\omega_0^2 - \omega^2 - i2\gamma\omega}$$

Linear optics: Susceptibility

- For $E(t) = E(\omega)e^{-i\omega t}$ and $P(t) = P(\omega)e^{-i\omega t}$, we apply a field and the medium acquires a polarization $P(\omega) = \varepsilon_0 \chi^{(1)} E(\omega)$
- This equals the average dipole moment per unit volume at ω . Assuming N independent oscillators:

$$P(\omega) = Np(\omega) = Nqx(\omega)$$

$$\chi^{(1)} = \frac{Nq^2}{\varepsilon_0 m} \frac{1}{\omega_0^2 - \omega^2 - i2\gamma\omega}$$
 displacement by field
- electron cloud

$$E(\omega) \neq 0$$

$$E(\omega) = qx(\omega)$$

$$E(\omega) = qx(\omega)$$

$$E(\omega) = 0$$

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Linear optics: Susceptibility

•
$$\chi^{(1)}(\omega) = Nqx(\omega) = \frac{Nq^2}{\varepsilon_0 m} \frac{1}{\omega_0^2 - \omega^2 - i2\gamma\omega}$$

• Calculate the real and imaginary parts of $\chi^{(1)}(\omega)$ and draw the trends around the resonance ω_0

$$\begin{pmatrix} \frac{c}{a+ib} \end{pmatrix} = \left(\frac{c}{a+ib}\right) \left(1 = \frac{a-ib}{a-ib}\right) = \left(\frac{ac}{a^2+b^2}\right) - i\left(\frac{bc}{a^2+b^2}\right)$$

$$Re(\chi^{(1)}) = \frac{Nq^2}{\varepsilon_0 m} \frac{\omega_0^2 - \omega^2}{(\omega_0^2 - \omega^2)^2 + (2\gamma\omega)^2}$$

$$Im(\chi^{(1)}) = \frac{Nq^2}{\varepsilon_0 m} \frac{2\gamma\omega}{(\omega_0^2 - \omega^2)^2 + (2\gamma\omega)^2}$$

Frequency [arb.]

A lot of optical functions!

• $1 + \chi = \bar{n}^2 =$	ε (complex functions!)
Conductivity	$\sigma=\sigma_1+i\;\sigma_2$
Dielectric constant	$\varepsilon = \varepsilon_1 + i \ \varepsilon_2 \varepsilon = 1 + \frac{4\pi i \sigma}{\omega} \varepsilon_1 = 1 - \frac{4\pi \sigma_2}{\omega} \varepsilon_2 = \frac{4\pi \sigma_1}{\omega}$
Refractive index	$ar{n}=n+i\;k$ $arepsilon=ar{n}^2$ $arepsilon_1=n^2-k^2$ $arepsilon_2=2nk$
	$n^2=(arepsilon_1+\sqrt{arepsilon_1^2+arepsilon_2^2})/2$ $k^2=(-arepsilon_1+\sqrt{arepsilon_1^2+arepsilon_2^2})/2$
Absorption coefficient	$lpha(\omega)=2\omegak/c\equiv\omegaarepsilon_2/nc$
Classical skin depth	$\delta = c/\omega k$
Surface impedance	$Z=4\pi$ $/car{n}$
Reflectivity	$R = \frac{(n-1)^2 + k^2}{(n+1)^2 + k^2} \text{(at normal incidence)}$
• The real parts often (not σ !) relate to <u>refraction</u> , the	
imaginary parts to <u>absorption</u> . What about R?	
Solid State Physics, G. Grosso and G. Pastori Parravicini, Academic Press (2000), Ch.11	
	https://doi.org/10.1016/B978-0-12-304460-0.X5000-2
	<u>https://doi.org/10.1016/B978-012304460-0/50011-6</u>

Nonlinear spectroscopy: output frequency ≠ input frequency

$$\chi \quad \text{wave eq.} \\ E \implies P \implies E_{out}$$

$$\chi = \chi(E) = \chi^{(1)} + \chi^{(2)}E + \chi^{(3)}E^2 + \dots P = P^{(1)} + P^{(2)} + P^{(3)} + \dots$$

Linear optics
$$P^{(1)} \equiv \epsilon_o \chi^{(1)} E^1$$

Second-order response
$$P^{(2)} \equiv \epsilon_o \chi^{(2)} E^2$$

Third-order response $P^{(3)} \equiv \epsilon_o \chi^{(3)} E^3$

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Nonlinear spectroscopy: output frequency ≠ input frequency

Nonlinear spectroscopy directly measures the anharmonicity! \rightarrow Invaluable tool to study matter

$$|\chi^{(1)}(\omega)| \approx 1$$
 $|\chi^{(2)}(2\omega)| \approx 10^{-12} \frac{m}{V}$ $|\chi^{(3)}(3\omega)| \approx 10^{-23} \frac{m^2}{V^2}$



$$P(t) = \varepsilon_{0}\chi E(t)$$

$$\chi(E) = \chi^{(1)} + \chi^{(2)}E + \chi^{(3)}E^{2} + \dots$$

$$P = P^{(1)} + P^{(2)} + P^{(3)} + \dots$$
Linear optics Second-order response Third-order response
$$P^{(1)} \equiv \epsilon_{o}\chi^{(1)}E^{1} \qquad P^{(2)} \equiv \epsilon_{o}\chi^{(2)}E^{2} \qquad P^{(3)} \equiv \epsilon_{o}\chi^{(3)}E^{3}$$

$$If E(t) = E_{o}\cos(\omega t) \text{ find } P^{(1)}, P^{(2)}, \text{ and } P^{(3)}.$$
Mixing one beam via $\chi^{(1)}, \chi^{(2)}, \text{ and } \chi^{(3)}$
"Elastic" response $\omega_{\text{input}} = \omega \rightarrow \omega_{\text{output}} = \omega$

$$P^{(1)} = \varepsilon_{0}\chi^{(1)}E_{0}\cos(\omega t)$$

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$$\cos^{2}\theta = \frac{1 + \cos(2\theta)}{2}$$

• $P^{(2)} = \varepsilon_0 \chi^{(2)} E_0^2 \cos^2(\omega t) =$

$$P(t) = \varepsilon_{0}\chi E(t)$$

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• $P^{(3)} = \varepsilon_{0}\chi^{(3)}E_{0}^{3}\cos(\omega t) = \varepsilon_{0}\chi^{(3)}E_{0}^{3}\frac{3\cos(\omega t)+\cos(3\omega t)}{4}$

THz generation: Optical rectification (OR)



THz generation: Optical rectification (OR)

What changes if a pulsed laser source is used?



- OR can be described as DFG of 'tails' (bw): $\omega_0 \omega_{0-bw} = \omega_{THz} \ll \omega_0$
- EOS: $\chi^{(2)}$ effect; $\omega_0 \omega_{THz} \approx \omega_0$. "Some sort of inverse OR".
- EOS is based on Pockels' / linear electro-optic effect. Here, a (low frequency) electric field induces birefringence in a medium with $\chi_{bulk}^{(2)} \neq 0$. The amount of ellipticity (polarization rotation) scales with THz field!

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THz detection: electro-optical sampling (EOS)

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THz generation: charge-current methods

• When we apply an electric field E_{in} to a material, it develops a polarization P which can act as the source of the emitted signal, E_s .



• This is the **only way** *E_s* can be emitted, if material has no free changes

Optics Communications 256, 400 (2005)

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- This is the **only way** *E_s* can be emitted, if material has no free changes
- "Whenever a free charge is accelerated, it emits radiation"
- If the material excited by E_{in} has free charges, the field can create a current, J, and the current can be the source of the emitted signal.



• This is the way an antenna works, and also synchrotron radiation - which is, at simplest, an antenna with relativistic "fast" charges.

Optics Communications 256, 400 (2005) https://en.wikipedia.org/wiki/Synchrotron_radiation

THz generation/detection by current: photoconductive antenna methods

- A photoconductive (PC) antenna consists of two metal electrodes (often gold) that are coated on a semiconductor substrate (GaAs), with a gap in between the electrodes. The "Auston switch" has an H-like structure with a gap in the order of 5 μm. Bias voltage is applied across the electrodes (10s V, 10s KHz AC).
- THz generation is obtained as follows:
 - 1. We excite the gap with an ultra-short pulse of light (~10-100 fs)
 - 2. Light generate free carriers and, hence, a current J in the material
 - 3. The bias field across the gap accelerates the free carriers
 - 4. The accelerated charges emit THz radiation (polarized || bias)



THz generation by charge-current(?): Plasma



THz-TDS with OR and EOS

- 1 THz = 33.36 cm⁻¹ = 4.14 meV (λ =300 μ m)
- Covers the 3÷100 cm⁻¹ spectral region
- Pulsed lasers \rightarrow THz time-domain spectroscopy



THz-TDS measures linear response, $\chi^{(1)}$

1) Detect THz time-shape with and without the sample



Hoffmann Ph.D. 2006

THz-TDS measures linear response, $\chi^{(1)}$

1) Detect THz time-shape with and without the sample



2)Fourier transform (magnitude shown)



Hoffmann Ph.D. 2006

THz-TDS measures linear response, $\chi^{(1)}$

1) Detect THz time-shape with and without the sample







3) Model to obtain optical parameters

$$\frac{E_{\rm sam}}{E_{\rm ref}}(\omega) = R(\omega)e^{i\phi(\omega)}$$

 $R(\omega)$: amplitude ratio $\phi(\omega)$: phase difference

Index of refraction
$$n_r(\omega) = 1 + rac{c}{\omega d} \phi$$

Absorption coefficient $\alpha(\omega) = -\frac{2}{d} \ln \left(\frac{(n_r(\omega) + 1)^2}{4n_r(\omega)} R \right)$

> *c*: speed of light *d*: sample thickness

THz-TDS setup



Ti:Sa 1 kHz 800 nm 50 fs 1 W

generation

Fresnel equations

The **Fresnel equations** (or **Fresnel coefficients**) describe the **amount of** reflected and transmitted electromagnetic radiation (including light) when incident on an interface between different optical media.



Partial transmission and reflection of a pulse travelling from a low to a high refractive index medium.

https://en.wikipedia.org/wiki/Fresnel_equations

Fresnel equations

The **Fresnel equations** (or **Fresnel coefficients**) describe the **amount of** reflected and transmitted electromagnetic radiation (including light) when incident on an interface between different optical media.



The s polarization refers to polarization of a wave's electric field *normal* to the plane of incidence (inside the plane of the interface). The p polarization refers to polarization of the electric field *in* the plane of incidence

Partial transmission and reflection of a pulse travelling from a low to a high refractive index medium.



https://en.wikipedia.org/wiki/Fresnel_equations

Fresnel equations

There are two sets of Fresnel coefficients for the two different linear polarization components of the incident wave (s and p).

Please note that these coefficients quantify amplitude and phase, not intensity!



https://en.wikipedia.org/wiki/Fresnel_equations





Field transmitted by the sample, for an "optically thick" sample (2) on an "optically thick" substrate (3)

$$\widetilde{E}_{S}(\omega) = E_{IN}\widetilde{t}_{12}(\omega)\widetilde{t}_{23}(\omega)e^{i\frac{\omega d}{c}\widetilde{n}_{2}(\omega)}\widetilde{t}_{34}(\omega)e^{i\frac{\omega D}{c}\widetilde{n}_{3}(\omega)}$$

Get the reference by removing the sample (medium 1 = medium 2) $\widetilde{E}_R(\omega) = E_{IN} \widetilde{t}_{13}(\omega) e^{i\frac{\omega d}{c}\widetilde{n}_1(\omega)} \widetilde{t}_{34}(\omega) e^{i\frac{\omega D}{c}\widetilde{n}_3(\omega)}$

$$E_{IN} \xrightarrow{\text{medium 1}} \underbrace{\text{medium 2}}_{i_{1}} \underbrace{\text{medium 3}}_{j_{1}} \xrightarrow{\text{medium 3}} \underbrace{\text{medium 4}}_{j_{1}} \underbrace{\text{spolarization}} \widetilde{r}_{i_{j}}(\omega) = \frac{\widetilde{n}_{i}(\omega)\cos(\vartheta_{i}) - \widetilde{n}_{j}(\omega)\cos(\vartheta_{j})}{\widetilde{n}_{i}(\omega)\cos(\vartheta_{i}) + \widetilde{n}_{j}(\omega)\cos(\vartheta_{j})}$$

$$\widetilde{r}_{i_{1}}(\omega) = \frac{\widetilde{n}_{i}(\omega)\cos(\vartheta_{i}) + \widetilde{n}_{j}(\omega)\cos(\vartheta_{j})}{\widetilde{n}_{i}(\omega)\cos(\vartheta_{i}) + \widetilde{n}_{j}(\omega)\cos(\vartheta_{j})}$$

$$\widetilde{p} = e^{i\frac{\omega d}{c}\widetilde{n}_{2}(\omega)} \mathbf{q} = e^{i\frac{\omega D}{c}\widetilde{n}_{3}(\omega)} \underbrace{\mathbf{q}^{*}\mathbf{r}_{23}^{*}\mathbf{p}^{*}\mathbf{t}_{12}^{*}\mathbf{t}_{34}}_{i_{1}} \underbrace{\widetilde{n}}_{i_{1}}(\omega) = \frac{2\widetilde{n}_{i}(\omega)\cos(\vartheta_{i}) + \widetilde{n}_{j}(\omega)\cos(\vartheta_{j})}{\widetilde{n}_{i}(\omega)\cos(\vartheta_{i}) + \widetilde{n}_{j}(\omega)\cos(\vartheta_{j})}$$

$$\widetilde{E}_{R}(\omega) = \frac{\widetilde{t}_{12}(\omega)\widetilde{t}_{23}(\omega)}{\widetilde{t}_{13}(\omega)} e^{i\frac{\omega d}{c}}(\widetilde{n}_{2}(\omega) - \widetilde{n}_{1}(\omega))$$
For normal incidence and in air:
$$= \frac{2}{1 + \widetilde{n}_{2}(\omega)}\frac{2\widetilde{n}_{2}(\omega)}{\widetilde{n}_{2}(\omega) + \widetilde{n}_{3}(\omega)}\frac{1 + \widetilde{n}_{3}(\omega)}{2}e^{-\frac{\omega d}{c}k_{2}(\omega)}e^{i\frac{\omega d}{c}(n_{2}(\omega) - 1)}$$
If low absorbing/high refracting:
$$= \frac{2}{1 + n_{2}(\omega)}\frac{2n_{2}(\omega)}{n_{2}(\omega) + n_{3}(\omega)}\frac{1 + n_{3}(\omega)}{2}e^{-\frac{\omega d}{c}k_{2}(\omega)}e^{i\frac{\omega d}{c}(n_{2}(\omega) - 1)}$$

and $\tilde{n}_3(\omega)$ in eq.4.47 can be taken as real. Writing in this case the ratio $\frac{\tilde{E}_S(\omega)}{\tilde{E}_R(\omega)}$ as amplitude multiplied phase, $Ae^{i\Delta\phi}$, we get $\Delta\phi \approx \frac{\omega d}{c}(n_2(\omega)-1)$ and $A \approx \frac{2}{1+n_2(\omega)} \frac{2n_2(\omega)}{n_2(\omega)+n_3(\omega)} \frac{1+n_3(\omega)}{2} e^{-\frac{\omega d}{c}k_2(\omega)}$, then

$$n_2(\omega) \approx 1 + \frac{c\Delta\phi}{\omega d} \tag{4.48}$$

$$k_2(\omega) \approx -\frac{c}{\omega d} \ln \frac{A(n_2(\omega) + 1)(n_2(\omega) + n_3(\omega))}{2n_2(\omega)(1 + n_3(\omega))}.$$
 (4.49)

In this last case when the substrate is replaced by air/vacuum nothing changes in the expression for $n_2(\omega)$, while $k_2(\omega) \approx -\frac{c}{\omega d} \ln \frac{A(n_2(\omega)+1)^2}{4n_2(\omega)}$.

http://hdl.handle.net/10077/8588

In general, may consider multiple reflections



Note: some of the experiments I will show you later have been measured with an FTIR...



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https://en.wikipedia.org/wiki/The_Blue_Marble







https://en.wikipedia.org/wiki/Pale_Blue_Dot https://en.wikipedia.org/wiki/The_Blue_Marble



Liquid water: network of hydrogen-bonded molecules that fluctuates (~ps)







https://en.wikipedia.org/wiki/Pale_Blue_Dot https://en.wikipedia.org/wiki/The_Blue_Marble

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MISSING LINK

"No one really understands water." P. Ball, Nature 452, 291

https://water.lsbu.ac.uk/water/index.html https://doi.org/10.1002/9781119300762.wsts0002

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Water is "black" in the terahertz range

- Hydrogen bond lifetime is ~1 ps
 >1/1ps = 1 THz: <u>THz radiation probes collective, intermolecular water modes</u>
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(again, the full IR absorption of liquid H_2O)



J.E. Bertie, Z. Lan, Applied Spectroscopy 50, 1047 (1996)

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THz spectroscopy on liquid water

- THz radiation probes intermolecular water modes
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What happens when we add a solute?

• When a solute is added to water, the fluctuating hydrogen-bonded network rearranges, and both the solvent-solvent and solvent-solute entropy end enthalpy change non-trivially



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- "The difficulty in understanding all reactions in aqueous media begins with the media." Chem. Rev. 105 335 2005
- The structures of proteins, nucleic acids, and cell membranes, as well as many other biological molecules, strictly depend upon water being the solvent.

Solvation

 Solvation science studies "how the solvent is involved in the control, mediation and regulation of chemical reactions" RESOLV cluster of excellence, <u>https://www.solvation.de/</u>

- Solvation science studies "how the solvent is involved in the control, mediation and regulation of chemical reactions" RESOLV cluster of excellence, <u>https://www.solvation.de/</u>
- A solvation process involves many steps (not necessarily in this order)



1. Pure liquid solvent

2. Creation of cavity

3. Insertion of solute

https://en.wikipedia.org/wiki/Solvation

1.



Pure liquid solvent • Liquids have rapidly changing molecular order with high cohesive interactions between molecules. Enthalpy (binding interactions between molecules) and entropy (large because molecules can move) determine the free energy and hence the system's structure.

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- For solvation to happen, the solvent must make space for a solute. This is both entropically (less configurations) and enthalpically (less bonds) unfavorable for the solvent.
- Enthalpy cost for the solute: it has to dissolve, leading to fewer solute-solute interactions. When the solute occupies the cavity: enthalpically favorable solvent-solute interactions; favorable entropy of mixing (solute crystal and pure solvent taken separately are more ordered than the comixture of solvent and solute).

- The enthalpy and entropy of solvent-solute interactions give the energy gained upon solvation. All these contributors taken together constitute the **solvation energy**.
- A solvation process involves many steps (not necessarily in this order)



1. Pure liquid solvent

2. Creation of cavity

3. Insertion of solute

https://en.wikipedia.org/wiki/Solvation

Angew. Chem. Int. Ed. 2019, 58, 3000-3013

FTIR measurements on aqueous salt solutions: ions have short-range effects on water



Figure 2. Schematic representation of ion hydration. Left: bulk-like water (blue), water in the vicinity of ions (purple and dark green), and hydration water (pink and light green) contribute to the total absorption of the solution. Subtraction of a concentration-weighted bulk-water spectrum (middle) yields the effective absorption of the solvated ion or ion pair (right). The resulting spectrum is a superposition of a negative part arising from the difference in the absorption between hydration and bulk water and the (positive) absorption arising from the hydrated anion/cation or hydrated ion pair.

 Over a century ago Hofmeister observed that salts can precipitate (salt out) or dissolve/denature (salt in) egg white and serum proteins depending on the constituent ions. It was assumed that the Hofmeister series reflects the long-range structuring of water by specific ions ("structure makers vs. breakers").

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J. Am. Chem.Soc. 2012, 134, 1030–1035

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- FTIR shows that anions and cations have specific absorption modes, and MD calculations explain them as the "rattling" of the ions in the hydration "cage" of water molecules



- Over a century ago Hofmeister observed that salts can precipitate (salt out) or dissolve/denature (salt in) egg white and serum proteins depending on the constituent ions. It was assumed that the Hofmeister series reflects the long-range structuring of water by specific ions ("structure makers vs. breakers").
- Also possible to quantify the "size" of the hydration layers for different ions (how many water molecules are affected)



FTIR measurements on gold nanoparticles



FTIR measurements on gold nanoparticles



- The libration is enhanced by positive Au NP (suppressed by negative charge)
- The stretch is suppressed by positive Au NP (enhanced by negative charge)

J. Phys. Chem. B 2019, 123, 6521-6528



- The OH stretch blue-shifts with positive Au NP / weaker H-bonds
- (The OH stretch red-shifts with negative Au NP / stronger H-bonds)

J. Phys. Chem. B 2019, 123, 6521–6528

FTIR measurements on gold nanoparticles

When we add a positive charge, e.g., Au³⁺ vs. Au⁰:

- In the O–H stretch (MIR) region, we observed an overall blue shift for the hydration water around NPs.
- Both intermolecular modes in the THz range are found to be red-shifted.
- These observations indicate that the intermolecular hydrogen bond is weakened for all gold NPs (more so in Au3+) when compared to bulk water



How good is THz-TDS to measure liquid water?



- Source: amplified Ti:Sa, 0.4 mJ/pulse, 790 nm, 1 kHz, 100 fs
- "tilted-front" <u>OR generation</u>: 12x12x12 mm MgO:sLN 62deg
- <u>EOS detection</u>: 0.5 mm thick
 ZnTe <001> (Eksma)
- Energy conversion eff. η~2·10⁻⁴; resulting in a peak electric field of ~30 kV/cm
- Measurements are performed by detecting the field transmitted by a reference and by the sample, then FT:
 - The magnitude ratio ~ α
 - The phase difference ~ n

Optics Letters 41, 5801 (2016)

How good is THz-TDS to measure liquid water?

Transmission of 0.5 mm thick water (10x penetration depth!) vs. temperature



Optics Letters 41, 5801 (2016)

How good is THz-TDS to measure liquid water?



THz-TDS can measure $\chi^{(1)}$ from water with high resolution: $\Delta n \sim 0.001$ $\Delta \alpha \sim 0.2 \text{ cm}^{-1}$

THz-thermometer:

- Non-contact
- Time-resolved
- ≤0.5 C resolution

Optics Letters 41, 5801 (2016)

THz-TDS of water solution with biomolecules

 Average absorption coefficient of an aqueous solution with an added concentration c of biomolecules



- Water strongly absorbs THz
- Typically, any solute will absorb less
- When the concentration of the solute increases, the THz absorption decreases
- The "water removal effect" is calculated by assuming that an amount of water equal to the amount of solute is removed from the solution (red line)
- The difference between the measurement (blue) and the water removal (red) is the absorption by solute + hydration layer.

J. Am. Chem. Soc. 2014, 136, 12800–12807

Proteins can have long-range effects on water



Fig. 1. Difference in the THz absorption coefficient at 2.25 THz relative to bulk water plotted against concentration to 3 mM at 15°C, 20°C, and 22°C (more extensive averaging was done at 22°C because of the slightly smaller effect). The absorbance depends nonlinearly on concentration in this region.

- "We observe an unexpected trend in the measured terahertz absorbance of the five helix bundle protein λ^*_{6-85} as a function of the protein-to-water molar ratio."
- "The trend can be explained by overlapping solvation layers around the proteins." Theory and experiment suggest an influence on the correlated water network motion beyond <u>20 Å</u>, greater than the pure structural correlation length usually observed (compare with the diameter of 1x H₂O, ~3 Å).

PNAS | December 26, 2007 | vol. 104 | no. 52 | 20749–20752

Protein hydration and molecular recognition

- Example of the correlation between protein function and hydration dynamics: antifreeze proteins (AFPs).
- AFPs enable the survival of organisms in sub-freezing environments by binding to nano-sized ice crystals and preventing the growth: tough recognition problems
- By combined THz spectroscopy and MD simulations, long-range hydration (20 A) dynamics seems to contribute to the recognition
- Hydration dynamics in AFPs have a gradient of retardation toward the "ice binding plane"
- This gradient, which we will call a " hydration funnel ", seems to assist molecular recognition: an entropically favorable docking of a nano-sized ice



J. Am. Chem. Soc. 2014, 136, 12800-12807

THz-TDS can distinguish aggregation

• We studied homogeneous aqueous solutions of native human lysozyme and lysozyme fibrils in water



TEM of insulin fibrils, 5 μ m scale

R. Liu et al./ Biochemical and Biophysical Research Communications 391 (2010) 862-867



HEW lysozyme, size ~2 nm

J. Am. Chem. Soc. 2011, 133, 8942–8947

• Protein aggregation related to some diseases (Alzheimer, Type-II diabetes, ...)

Probing native and fibrils of human lysozyme



- First time > 10 kV/cm peak THz used to study bio-macromolecules in water
- Can measure >0.5 mm thick water-based samples:
 - sensitive to low concentrations
 - obtain full dielectric response

Probing native and fibrils of human lysozyme



What happens with fibrils?

Take two solutions containing the same amount of proteins, the first in native form, the latter with the same amount of proteins in aggregated fibrils

Native





What happens with fibrils?

Take two solutions containing the same amount of proteins, the first in native form, the latter with the same amount of proteins in aggregated fibrils: **double the concentration**

Native



Aggregated



Probing native and fibrils of human lysozyme

THz-TDS is able to look at the bulk-averaged dipole of a solute+hydration unit and:

- Detect perturbations in the water network induced by proteins up to large distances from the protein surface (for native human lysozyme, up to ~42 Å away for a THz probe with frequency <1 THz)
- 2. Distinguish between solutions of proteins in different aggregation states





Liquid-liquid phase separation

The key role of solvent in condensation: Mapping water in liquid-liquid phase-separated FUS

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"Membrane-less compartmentalization via phase separation in living cells has been linked to the formation of pathological protein aggregates found in neurodegenerative diseases"

"Two thermodynamic driving forces have been proposed: proteinprotein and protein-water interactions (mostly enthalpic), as well as the release of preorganized hydration water into the bulk (mostly entropic)"

Liquid-liquid phase separation



FIGURE 4 Scheme showing the proposed water-mediated contribution of FUS to LLPS. Formation of phase-separated droplets is supported by an increase in tetrahedral coordination of water molecules (bound water; *thick black*) and minimization of less favorable water interactions (wrap water; *red*). To see this figure in color, go online.

Biophysical Journal 120, 1266–1275, April 6, 2021

Terahertz and Water: outline

1. Terahertz time-domain spectroscopy

- 1. Examples of non-linear responses
- 2. Linear and non-linear spectroscopy
- 3. Optical rectification (OR) and electro-optical sampling (EOS)
- 4. Models to obtain the optical properties (Fresnel equations)

2. Water and solvation

- 1. Infrared absorption modes of liquid water
- 2. Introduction to solvation science
- 3. Study ions in water with THz: rattling, hydration layers
- 4. Gold nanoparticles
- 5. Long-range protein effects on water
- 6. Non-linear terahertz response of water
Liquid water is "black" in the THz range

- THz radiation probes intermolecular water modes
- Water strongly absorbs THz: sensitive spectroscopic probe



How do microwaves heat water?



The simplest nonlinear experiment

- At 1 THz we have intense THz sources
- Measure transmission vs. position w.r.t. focus: Z-scan



Away from focus: Low peak power, Linear optical functions (α, n)

Close to focus: High peak power, Nonlinear optical effects (α^{NL} , n^{NL})

M. Sheik-Bahae et al., IEEE Journal of Quantum Electronics 26, 760, 1990

Intensity detector





Tcypkin *et al.*, Opt. Express 27, 10419, 2019 Novelli *et al.*, Appl. Sci. 10, 5290, 2020

A few groups found similar transient signals ~1 THz... different explanations



Tcypkin *et al.*, Opt. Express 27, 10419, 2019 Tcypkin *et al.*, Phys. Rev. Appl. 15, 054009, 2021 Ghalgaoui *et al.*, J. Phys. Chem. Lett. 11, 7717, 2020; Phys. Rev. Lett. 126, 097401, 2021

Pump-probe on liquid water at 12.3 THz

 The microscopic motion associated to the absorption between about 10 and 20 THz is better understood

Wagging

• Free-electron lasers are bright in this range









Natural dynamics of the damped dipole reorientations: 30-120 fs



Natural dynamics of the damped dipole reorientations: 30-120 fs

Resonant field: driven water dipole reorientations, transient anisotropy

S(ω) [arb. units]

Summary: NL THz and pure water



Novelli *et al.*, Materials 13, 1311, 2020; Novelli *et al.*, J. Phys. Chem. B 124, 4989, 2020; Novelli *et al.*, Appl. Sci. 10, 5290, 2020; Novelli *et al.*, Phys. Chem. Chem. Phys. 24, 653, 2022; ..., Novelli, AIP Advances 12, 115319, 2022; Novelli *et al.*, in preparation

IT IS VERY IMPORTANT TO LOOK AT THE RIGHT FREQUENCY (for the intermolecular modes of the water network, this is terahertz range!)

https://www.thepoke.co.uk/2017/07/20/gif-frames-per-second-match-leg-speed/



IT IS VERY IMPORTANT TO LOOK AT THE RIGHT FREQUENCY (**for the intermolecular modes of the water** THANK YOU !!! Fabio.novelli@rub.de

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