



# DFT and DFT embedding

Ways to approach the mesoscopic world with ab-initio methods  
(without cutting corners)

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Ready to work?

Find this Jupyter Notebook at  
[~/ASESMA2023/Special\\_Topics\\_Seminars/Michele\\_Pavanello](~/ASESMA2023/Special_Topics_Seminars/Michele_Pavanello)





# Acknowledgements



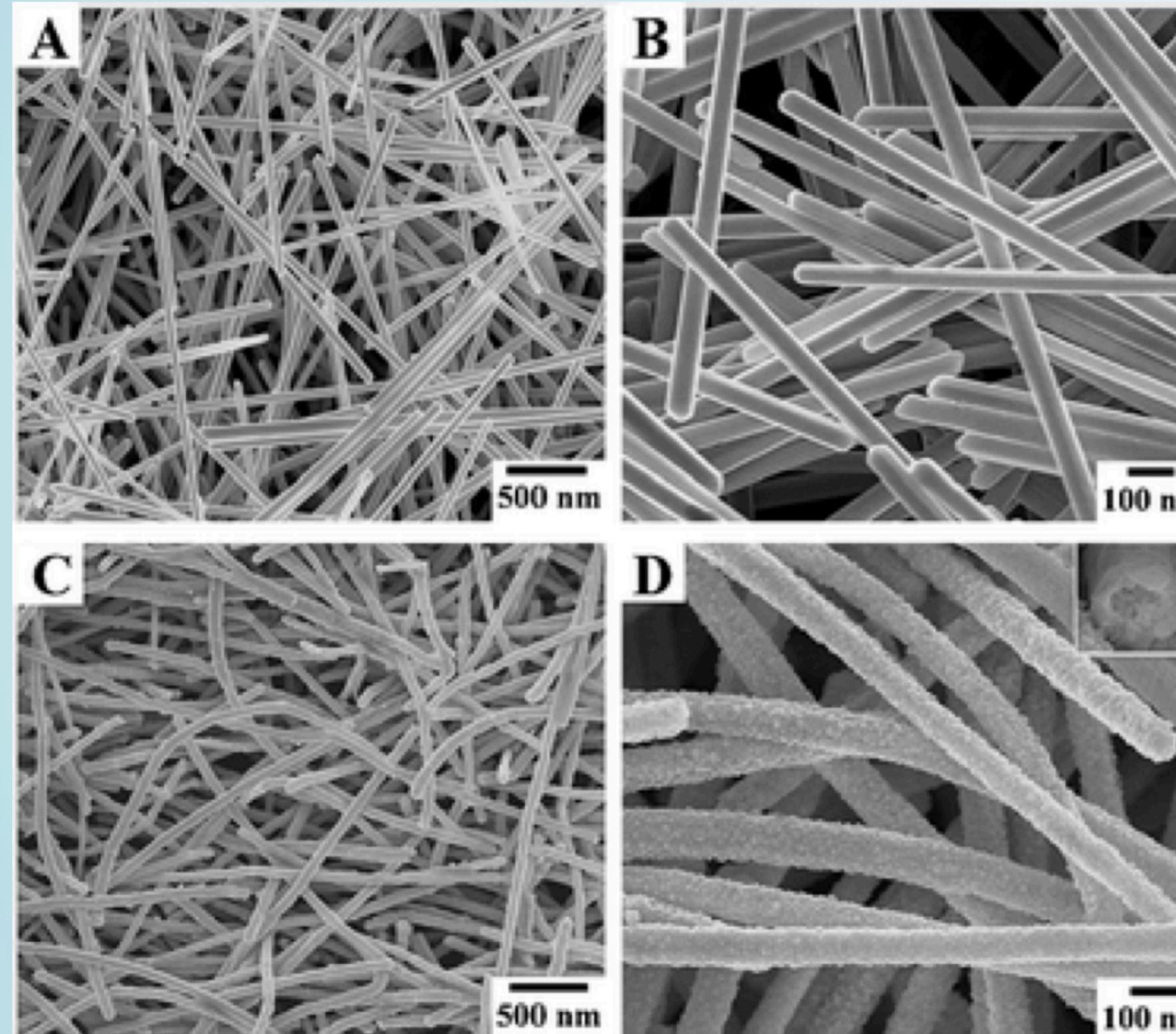
Funding: NSF, DOE, MolSSI, ACS-PRF, Rutgers



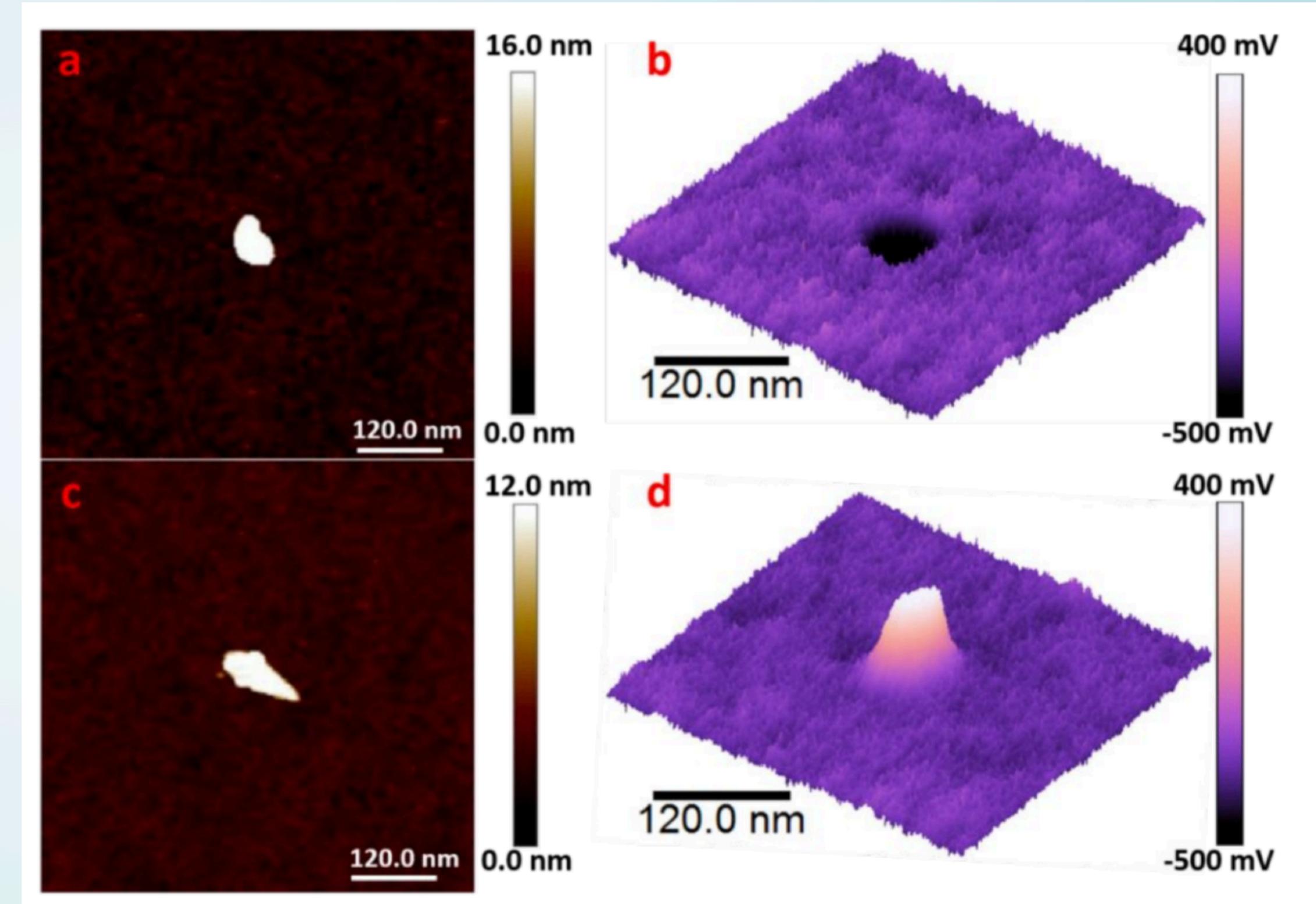


# The Real World

## Photocatalyst



## Catalytic nanoparticles



Chem. Comm., 43, 6551 (2009)

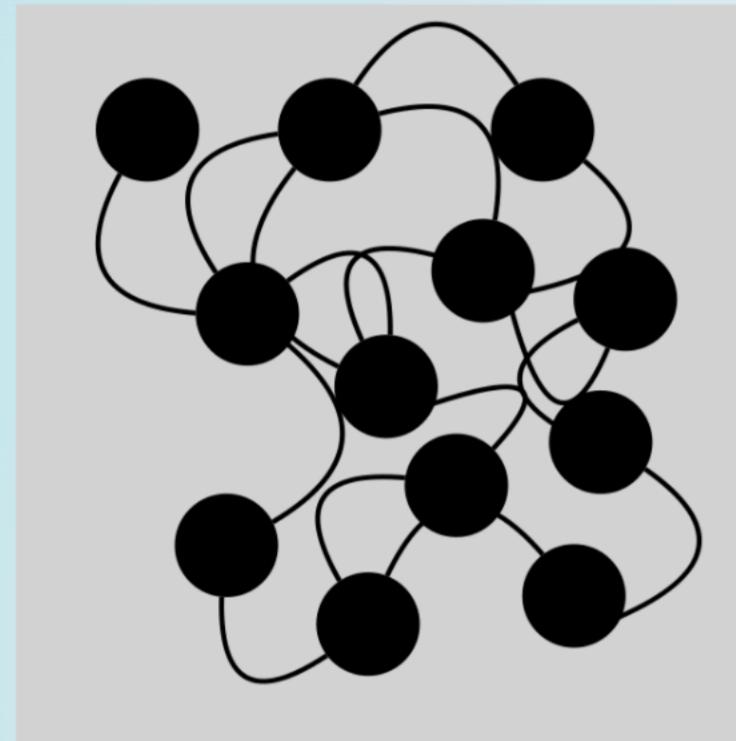
PCCP, 21, 15080 (2019)



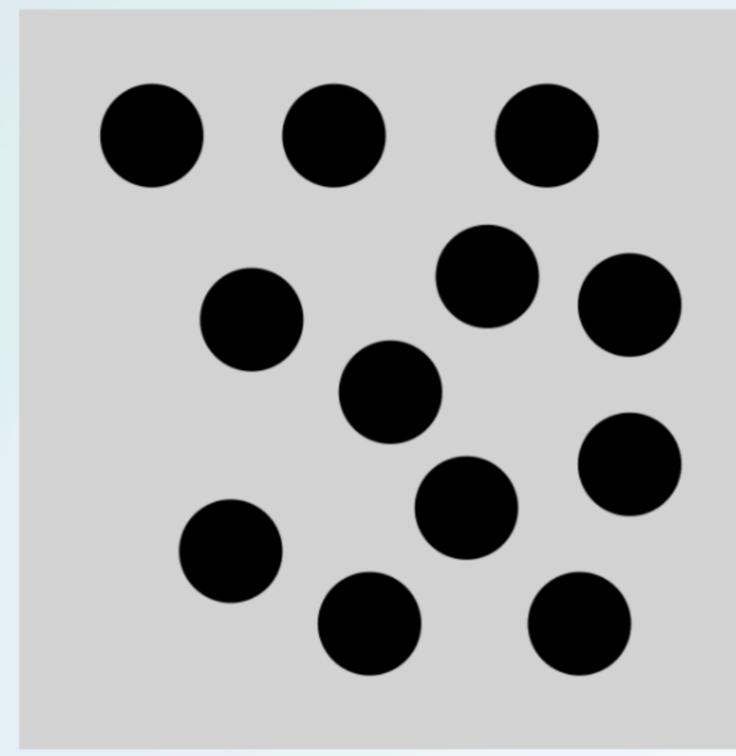


# Available electronic structure methods

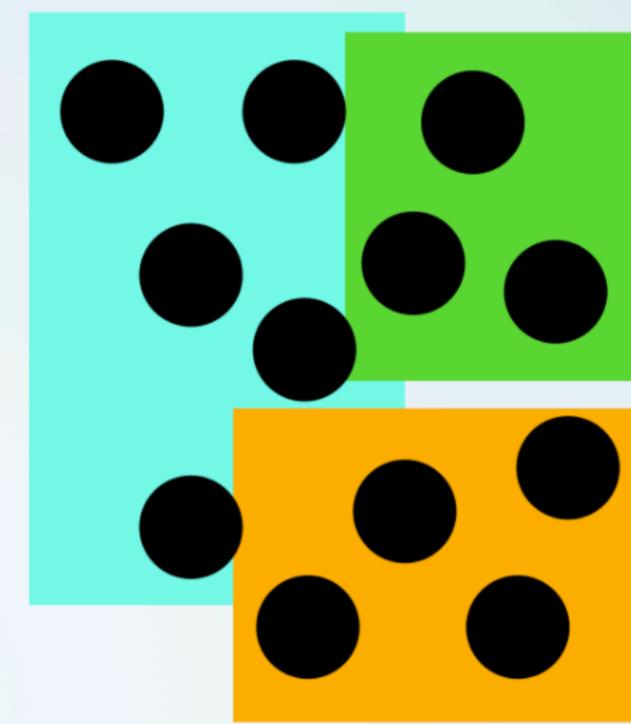
Quantum Chemistry



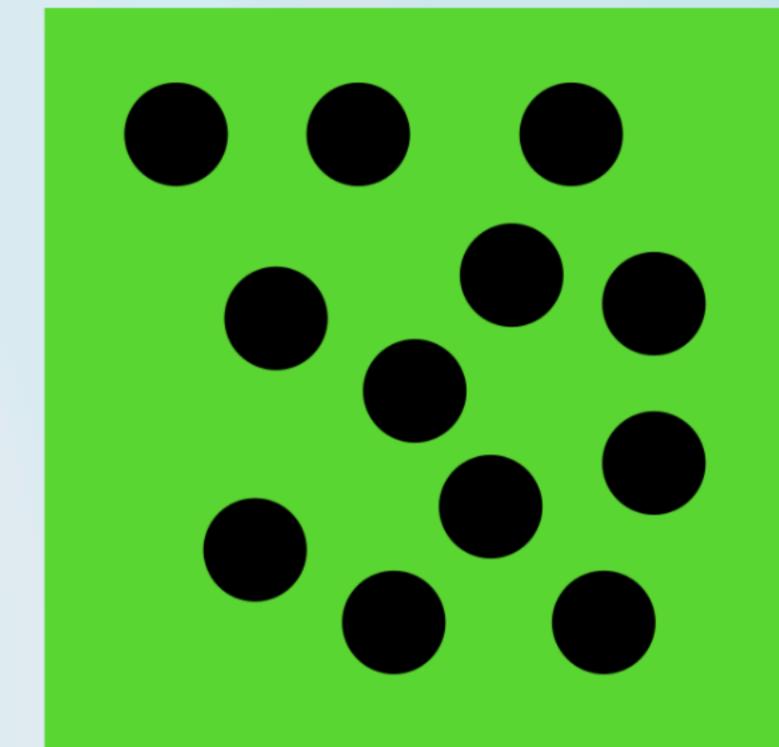
Kohn-Sham DFT



Subsystem DFT



Orbital-free DFT



$$n(\mathbf{r}) = N \int |\Psi(\mathbf{r}, \mathbf{r}_2, \dots, \mathbf{r}_N)|^2 d\mathbf{r}_2 \dots d\mathbf{r}_N$$

$$n(\mathbf{r}) = \sum_i^N |\phi_i(\mathbf{r})|^2$$

$$n(\mathbf{r}) = \sum_I n_I(\mathbf{r})$$

$$n(\mathbf{r}) = N |\psi_1(\mathbf{r})|^2$$

Efficiency

Accuracy





# Non-interacting kinetic energy

The non-interacting kinetic energy is a crucial quantity in DFT:

$$T_s[n] = \sum_i \omega_i \int \phi_i(\mathbf{r}) \nabla^2 \phi_i(\mathbf{r}) d\mathbf{r}$$

where  $\omega_i$  are the occupation numbers.

Note: it is **not** the true kinetic energy, but it allows us to define the KS equations, which give us the KS orbitals which give us the density which give us the energy!

The KS equations are:

$$-\frac{1}{2} \nabla^2 \phi_i(\mathbf{r}) + v_s(\mathbf{r}) \phi_i(\mathbf{r}) = \varepsilon_i \phi_i(\mathbf{r})$$





# We have a nice chain of events:

KS equations  $\rightarrow \{\phi_i\} \rightarrow n(\mathbf{r}) \rightarrow E[n]$ , forces ...

where

$$E[n] = T_s[n] + E_H[n] + E_{xc}[n] + \int n(\mathbf{r})v_{ext}(\mathbf{r})d\mathbf{r}$$

Note  $E_{xc}[n]$ , the exchange-correlation functional





# What is the cost of getting the KS orbitals?

- Must solve KS equations

How expensive is that? Any guesses?

Like solving an eigenvalue problem

Like diagonalizing a matrix

Oh, no! Bad news, it is like  $\mathcal{O}(N^3)$ !

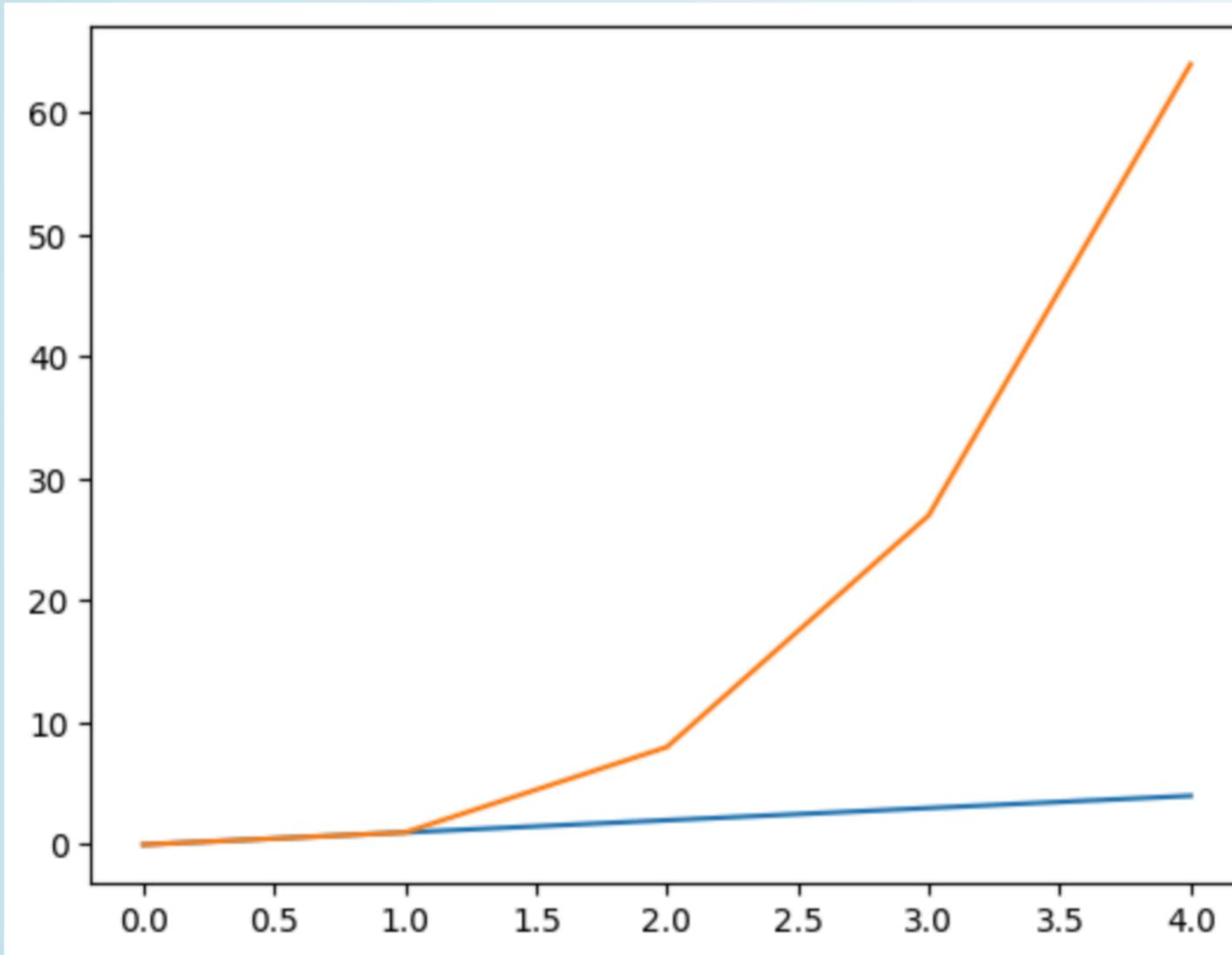




# Why is it bad news?

```
In [3]: import matplotlib.pyplot as plt  
x=np.array([0,1,2,3,4])  
y=x**1  
plt.plot(x,y)  
plt.plot(x,y**3)
```

```
Out[3]: [<matplotlib.lines.Line2D at 0x13fff1870>]
```



We need alternative methods if we wish to approach *realistic* systems





# An option: Orbital-free DFT (OF-DFT)

- The variational function is the electron density only.

$$n(\mathbf{r}) = N|\psi(\mathbf{r})|^2$$

- The energy functional is a **pure** functional of  $n(\mathbf{r})$

$$E[n] = T_s[n] + E_H[n] + E_{xc}[n] + \int n(\mathbf{r})v_{ext}(\mathbf{r})d\mathbf{r}$$

- Ground state is found by direct minimization

$$n_0(\mathbf{r}) = \arg \min_n \left[ E[n] - \mu \left( \int n(\mathbf{r})d\mathbf{r} - N \right) \right]$$

**Opportunities:** (1) No need to diagonalize any matrix! (2) Complexity  $\mathcal{O}[N \ln N]$

**Problem:**  $T_s[n]$  exists, but it is not known...





# Orbital-free vs Kohn-Sham DFT

- KS and OF energy functionals are the same

$$E[n] = T_s[n] + \int n(\mathbf{r})v_{ext}(\mathbf{r})d\mathbf{r} + E_{xc}[n] + E_H[n]$$

- **In practice**, the noninteracting kinetic energy is treated differently

$$T_s[n] \text{ vs } T_s[\{\psi_i[n]\}]$$





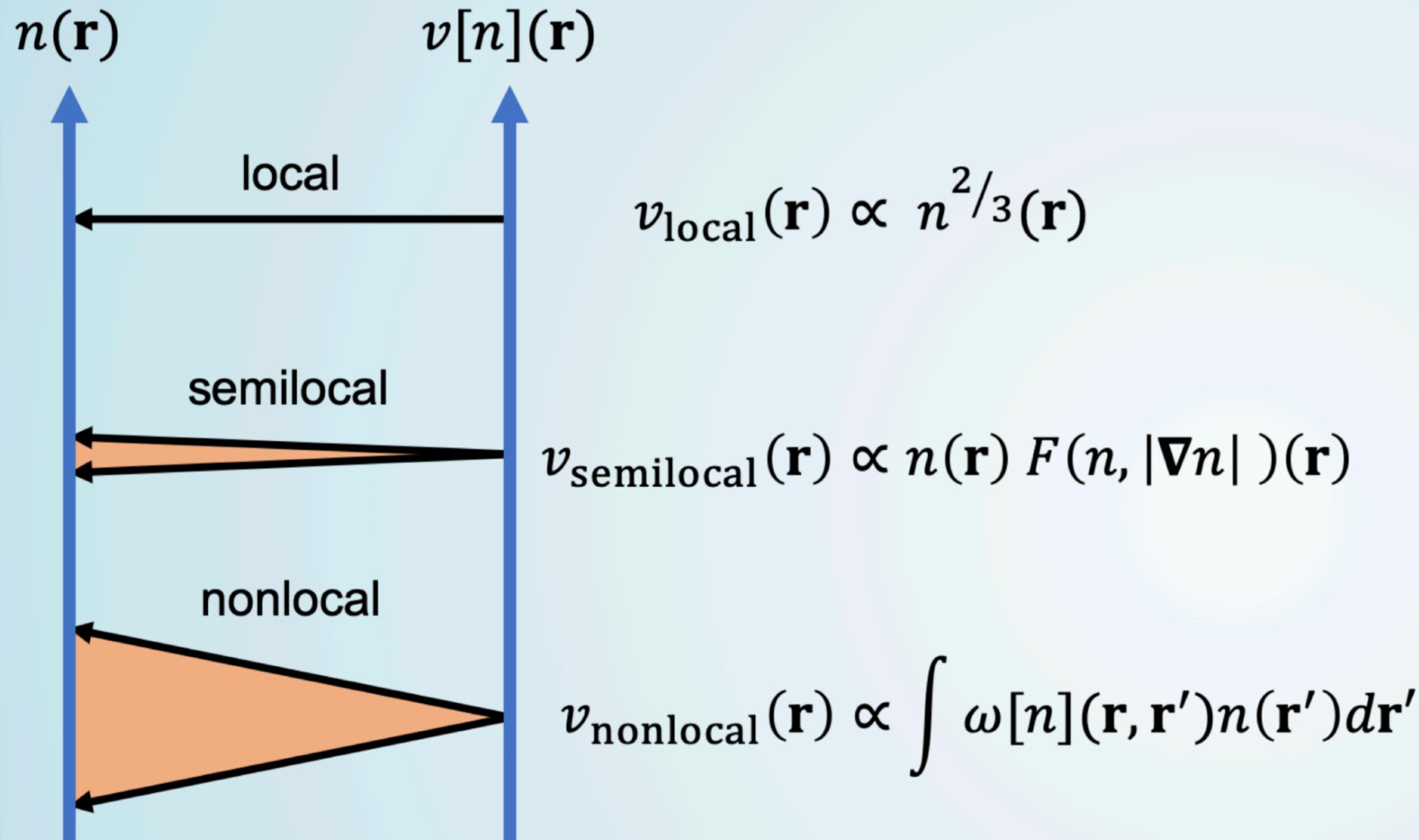
# $T_s[n]$ a long history...

Year	Milestone	Method
1927	Thomas and Fermi present their approximate method that satisfies Pauli principle while avoiding the Schrödinger equation	TF <sup>22,23</sup>
1930	TF is applied to atoms such as Fe yielding qualitative agreement with experimental sequential ionization energies <sup>24</sup>	
1930	Dirac extends TF to include exchange effects	TFD <sup>25</sup>
1935	Slater extends TF to the condensed phase and applies it to metals but no binding is found <sup>26</sup>	
1935	von Weizsäcker proposes the first gradient correction	TFvW, TFvWD <sup>27</sup>
1945	The ENIAC computer is built, patent filed in 1948 <sup>9</sup>	
1949	Reitz implements TFD on ENIAC for computation of electronic properties of atoms <sup>28</sup>	
1949	Feynman, Metropolis, and Teller extend the finite-temperature Thomas-Fermi theory <sup>29</sup>	
1957-1982	Extensions of TFvWD and the Gradient Expansion Approximation	GEA <sup>30-34</sup> and TFλvW <sup>35-38</sup>
1962	Teller proves that TF cannot lead to molecular bonding <sup>39</sup>	
1964	Formal footing of DFT	Hohenberg-Kohn theorems <sup>3</sup>
1965	Mapping to a non-interacting electron system and employment of the exact non-interacting KEDF	KSDFT <sup>1</sup>
1964-1965	Introduction of polarizability in the construction of the KEDF	Non-local KEDF <sup>3,40</sup>
1981-1983	Computational proof that a gradient correction to TF (e.g., TFvWD) leads to molecular binding <sup>41,42</sup>	CAT, <sup>43</sup> WT, <sup>44</sup> WGC, <sup>45</sup> SM, <sup>46</sup> Perrot, <sup>47</sup> Chai-Weeks, <sup>48</sup> XWM <sup>49</sup> KEDFs, real space <sup>50-53</sup> and reciprocal space <sup>54,55</sup>
1985-ongoing	Non-local (two-point) KEDFs for condensed phases	GGA XC <sup>56,57</sup>
1986	Generalized Gradient Approximation for the exchange and correlation energy functional	GGA
1991	Conjointness conjecture <sup>58</sup>	KEDFs <sup>58-63,63-76</sup>
2010-ongoing	Non-local (two-point) KEDF for semiconductors	HC <sup>77</sup> , WGCD, <sup>78</sup> EvW-WGC, <sup>79</sup> KGAP, <sup>80</sup> MGP <sup>81</sup>
2012-ongoing	Machine-learnt KEDFs <sup>82-84</sup>	VT84F, <sup>85</sup> LKT, <sup>86</sup> PGSL <sup>87</sup>
2013-ongoing	General-purpose semi-local (one-point) KEDFs	LMGP, <sup>88,89</sup> LDAK <sup>90</sup> and revHC <sup>91</sup>
2019-ongoing	Non-local (two-point) KEDF for finite systems and subsystem DFT	





## $T_s[n]$ options: (semi)local and nonlocal functionals



Nonlocal with density **independent** kernel

$$T_{NL}[n] = \langle n(\mathbf{r})^\alpha | \omega(\mathbf{r}, \mathbf{r}') | n(\mathbf{r}')^\beta \rangle$$

Wang & Teter, PRB (1992)

Nonlocal with density **dependent** kernel

$$T_{NL}[n] = \langle n(\mathbf{r})^\alpha | \omega[n](\mathbf{r}, \mathbf{r}') | n(\mathbf{r}')^\beta \rangle$$

Huang & Carter, PRB (2010)

Mi & Pavanello, PRB (2019)

Wenhui Mi, MP JCP (2018) • Wenhui Mi, MP PRB (2019) • Xuecheng Shao, WM, MP PRB (2021) Xuecheng Shao, WM, MP JPCL (2021) • Xuecheng Shao, WM, MP JCTC (2021) • Wenhui Mi, MP JPCL (2020)

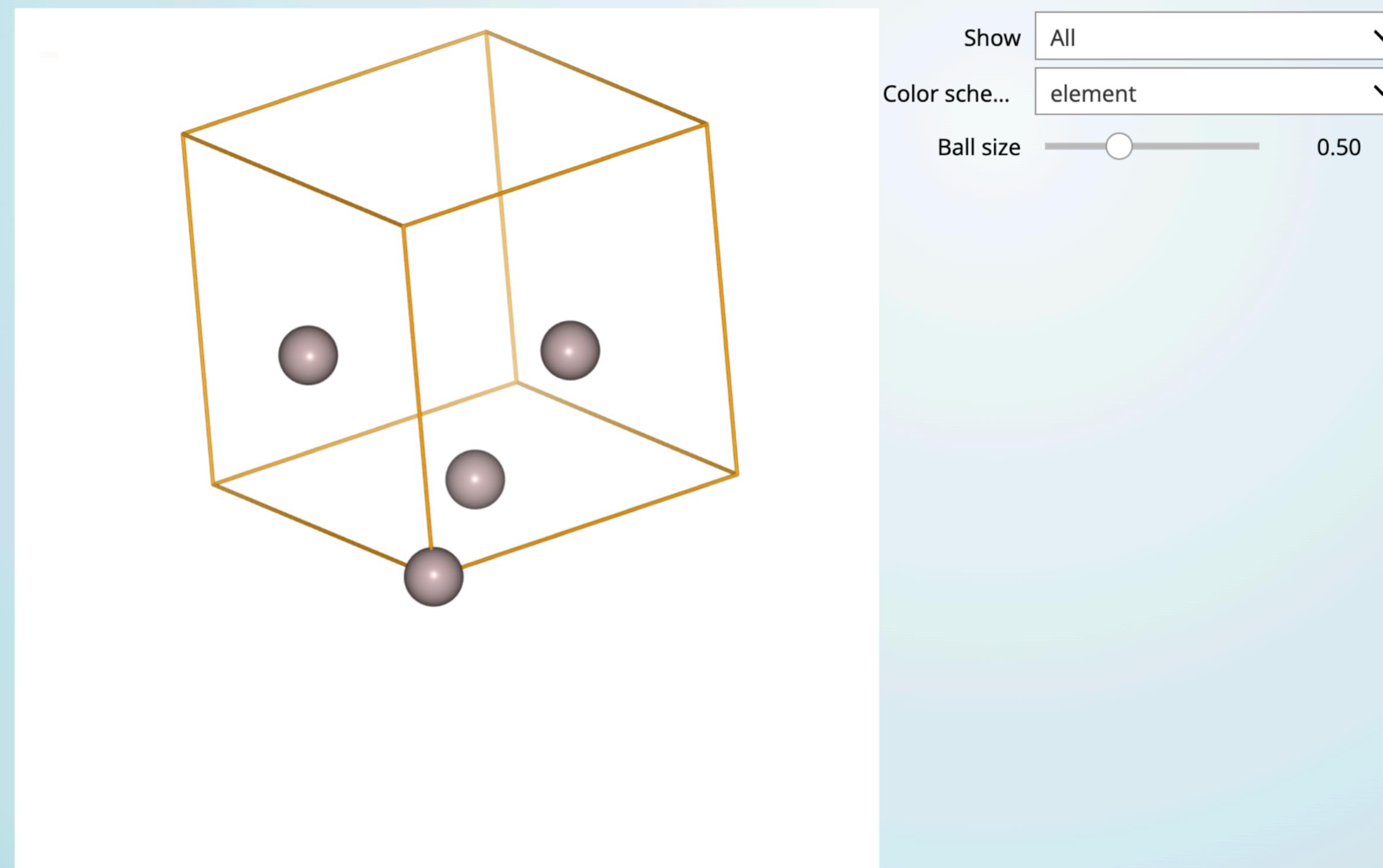




# How hard is it to run OF-DFT simulations?

First generate a model system: bulk Al (cubic cell)

```
In [4]: from ase.build import bulk  
atoms = bulk('Al', 'fcc', a=4.05, cubic=True)  
ions = Ions.from_ase(atoms)  
view(ions, viewer='ngl')
```





## Then run a OF-DFT simulation with DFTpy

```
In [8]: IFrame('http://dftpy.rutgers.edu', width=800, height=300)
```

Out[8]:

The screenshot shows a web browser displaying the DFTpy documentation. The URL is 'http://dftpy.rutgers.edu'. The page has a blue header with the 'DFTpy' logo, which is a stylized red and yellow atom-like symbol. Below the logo is a search bar labeled 'Search docs'. The main content area features the title 'DFTpy: Density Functional Theory with Python' in large, bold, dark font. Below the title is a sub-section titled 'What is DFTpy?'. At the top right of the content area, there are links for 'Index' and 'Source', and a 'Next' button with a circular arrow icon.

```
In [9]: nr = ecut2nr(ecut=35, lattice=ions.cell)  
nr
```

Out[9]: array([20, 20, 20], dtype=int32)

```
In [10]: grid = DirectGrid(lattice=ions.cell, nr=nr)  
PSEUDO = LocalPseudo(grid = grid, ions=ions, PP_list=PP_list)  
rho_ini = DirectField(grid=grid)  
rho_ini[:] = ions.get_ncharges()/ions.cell.volume  
HARTREE = Functional(type='HARTREE')  
XC = Functional(type='XC', name='LDA')
```

setting key: Al -> Al\_lda.oe01.recpot





Let's pick a  $T_s[n]$

- $T_s[n]$  also called KEDF. In this example we use Thomas Fermi + von Weizsacker:

$$T_s[n] = \int C_{TF} n^{5/3}(\mathbf{r}) + \frac{\nabla n^{1/2}(\mathbf{r}) \cdot \nabla n^{1/2}(\mathbf{r})}{2} d\mathbf{r}$$

```
In [5]: KE = Functional(type='KEDF', name='x_TF_y_vW')
```





## Define "total energy" and minimize $E[n]$

```
In [11]: evaluator = TotalFunctional(KE=KE, XC=XC, HARTREE=HARTREE, PSEUDO=PSEUDO)
optimization_options = {'econv' : 1e-6*ions.nat}
opt = Optimization(EnergyEvaluator=evaluator, optimization_options = optimization_options,
                    optimization_method = 'TN')
%timeit -n1 -r1 rho = opt.optimize_rho(guess_rho=rho_ini)
```

Step	Energy(a.u.)	dE	dP	Nd	Nls	Time(s)
0	-8.090977705021E+00	-8.090978E+00	7.877088E-01	1	1	2.329612E-02
1	-8.273226046462E+00	-1.822483E-01	7.033208E-02	2	1	3.579021E-02
2	-8.280858134665E+00	-7.632088E-03	4.803551E-03	7	1	6.396222E-02
3	-8.281101057233E+00	-2.429226E-04	3.640546E-04	5	1	8.519292E-02
4	-8.281135973142E+00	-3.491591E-05	3.590712E-05	6	1	1.159899E-01
5	-8.281138619986E+00	-2.646845E-06	2.500920E-06	5	1	1.379251E-01
6	-8.281138990518E+00	-3.705314E-07	4.568870E-08	8	1	1.676691E-01

#### Density Optimization Converged ####  
Chemical potential (a.u.): 0.30115196527266436  
Chemical potential (eV) : 8.194761597103438  
170 ms ± 0 ns per loop (mean ± std. dev. of 1 run, 1 loop each)

**Opportunities:** Can manipulate rho as you wish - many functionalities are available in DFTpy

...let's try to re-run it with a different  $T_s[n]$ .





## Re-run of OF-DFT with nonlocal $T_s[n]$

```
In [12]: KE = Functional(type='KEDF', name='WT')
evaluator = TotalFunctional(KE=KE, XC=XC, HARTREE=HARTREE, PSEUDO=PSEUDO)
optimization_options = {'econv' : 1e-5*ions.nat}
opt = Optimization(EnergyEvaluator=evaluator, optimization_options = optimization_options,
                    optimization_method = 'TN')
%timeit -n1 -r2 rho = opt.optimize_rho(guess_rho=rho_ini)
```

Step	Energy(a.u.)	dE	dP	Nd	Nls	Time(s)
0	-8.090977705021E+00	-8.090978E+00	7.877088E-01	1	1	6.896973E-03
1	-8.338206651608E+00	-2.472289E-01	5.654301E-02	7	1	5.253196E-02
2	-8.343069740849E+00	-4.863089E-03	3.268387E-03	7	1	9.050608E-02
3	-8.343656792601E+00	-5.870518E-04	3.047555E-04	9	1	1.358750E-01
4	-8.343674713047E+00	-1.792045E-05	2.723036E-05	5	1	1.638479E-01
5	-8.343676820175E+00	-2.107128E-06	1.881566E-06	7	1	1.982689E-01

#### Density Optimization Converged ####

Chemical potential (a.u.): 0.2927374982051514

Chemical potential (eV) : 7.9657923073878125

Step	Energy(a.u.)	dE	dP	Nd	Nls	Time(s)
0	-8.090977705021E+00	-8.090978E+00	7.877088E-01	1	1	6.319046E-03
1	-8.338206651608E+00	-2.472289E-01	5.654301E-02	7	1	4.166794E-02
2	-8.343069740849E+00	-4.863089E-03	3.268387E-03	7	1	7.957697E-02
3	-8.343656792601E+00	-5.870518E-04	3.047555E-04	9	1	1.243918E-01
4	-8.343674713047E+00	-1.792045E-05	2.723036E-05	5	1	1.514030E-01
5	-8.343676820175E+00	-2.107128E-06	1.881566E-06	7	1	1.866379E-01

#### Density Optimization Converged ####

Chemical potential (a.u.): 0.2927374982051514

Chemical potential (eV) : 7.9657923073878125

195 ms ± 5.67 ms per loop (mean ± std. dev. of 2 runs, 1 loop each)

Nonlocal  $T_s[n]$  are somewhat slower than semilocal functionals





# How does OF-DFT compare with KS-DFT?

We will run a Quantum ESPRESSO calculation of the same system using QEpy

```
In [13]: from qepy.driver import Driver  
from qepy.io import QEInput
```

```
In [14]: IFrame('http://qepy.rutgers.edu', width=1000, height=400)
```

Out[14]:

The screenshot shows a Jupyter Notebook interface. In the code cell (In [13]), two imports are shown: `from qepy.driver import Driver` and `from qepy.io import QEInput`. In the next cell (In [14]), an `IFrame` command is used to embed a web page from 'http://qepy.rutgers.edu' with a width of 1000 and a height of 400. The resulting output (Out[14]) is a rendered version of the QEpy documentation website. The page title is 'QEpy - Quantum ESPRESSO in Python'. It features a sidebar with links to 'Contact', 'Installation', 'Tutorials', 'Modules', 'Development', and 'Frequently Asked Questions'. The main content area describes QEpy as turning Quantum ESPRESSO (QE) into a Python DFT engine for nonstandard workflows. Below this is a section titled 'Contributors and funding' with a bulleted list of authors: 'Main author: Xuecheng Shao (Rutgers)' and 'Oliviero Andreussi (UNT), Davide Ceresoli (CNR, Italy), Matthew Truscott'.





## QEpy's input file is a dictionary

```
In [15]: qe_options = {
    '&control': {
        'calculation': "'scf'",
        'pseudo_dir': "./",
    },
    '&system': {
        'ibrav' : 0,
        'degauss': 0.005,
        'ecutwfc': 30,
        'occupations': "'smearing'"
    },
    'atomic_species': ['Al 26.98 Al.pbe-nl-kjpaw_psl.1.0.0.UPF'],
    'k_points gamma': [],
}
```

!wget <http://pseudopotentials.quantum-espresso.org/upf> files/Al.pbe-nl-kjpaw\_psl.1.0.0.UPF

```
In [16]: options = {
    '&electrons': {
        'mixing_beta': 0.5},
    'cell_parameters angstrom':[
        '0.    2.025  2.025',
        '2.025  0.    2.025',
        '2.025  2.025  0.    '],
    'atomic_positions crystal': ['Al    0.0  0.0  0.0'],
    'k_points automatic': ['11 11 11 1 1 1'],
}
```





## Initialize QEpy

```
In [17]: qe_options = QEInput.update_options(options, qe_options=qe_options)
```

```
In [18]: driver = Driver(qe_options=qe_options, logfile=True)
%timeit -n1 -r1 driver.scf()
```

8 s ± 0 ns per loop (mean ± std. dev. of 1 run, 1 loop each)

```
In [19]: driver.get_scf_error()
```

```
Out[19]: 8.733828545114252e-08
```

```
In [20]: driver.stop()
```

KS-DFT is ★ times slower than nonlocal OF-DFT for this small system.





## Poll time

- People say OF-DFT is a "linear-scaling method". Do you agree?

A: Yep! I do!

B: Nah. Not really.

C: Sometimes.



# Can OF-DFT approach surfaces?

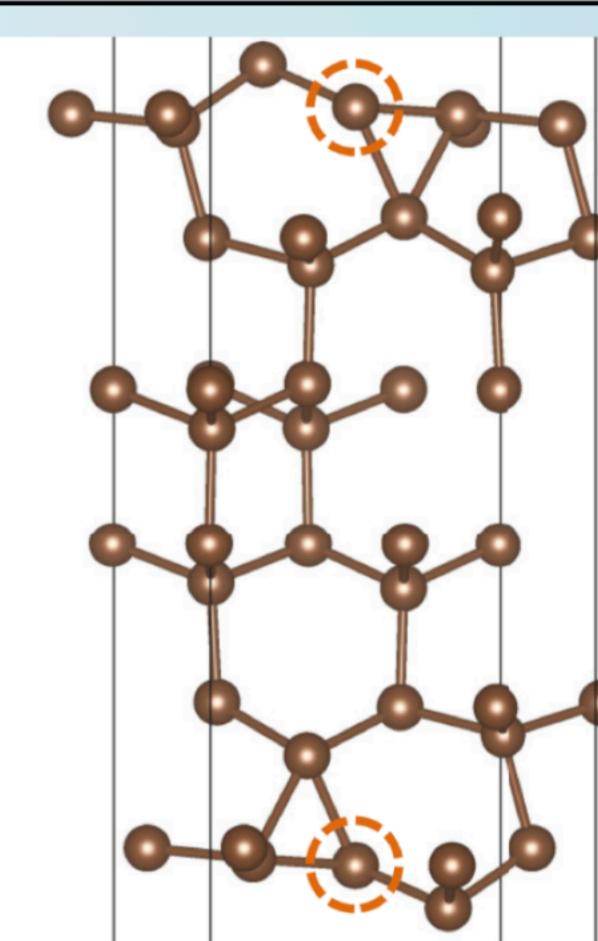
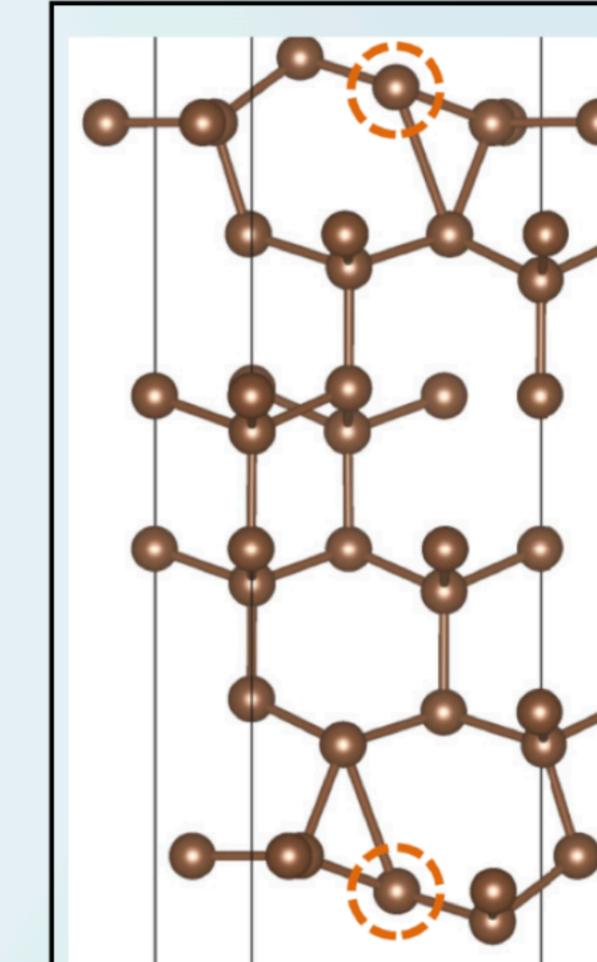
Shao, Mi & Pavanello, Phys. Rev. B **104**, 045118 (2021)

**Surface energies**

System	Surface	Surface energy (eV/Å <sup>2</sup> )			RE
		revHC	KS	KS(US)	
Si	(111)	0.092	0.097	0.113	-5%
	(110)	0.120	0.111	0.127	8%
	(100)	0.159	0.128	0.148	24%
Ge	(111)	0.079	0.078	0.083	1%
	(110)	0.110	0.091	0.096	21%
	(100)	0.114	0.105	0.108	9%

System	Surface energy (eV/Å <sup>2</sup> )			RE
	revHC	KS	KS(US)	
AlP	0.115	0.118	0.100	-3%
AlAs	0.103	0.105	0.087	-2%
AlSb	0.083	0.079	0.069	5%
GaP	0.122	0.114	0.096	7%
GaAs	0.107	0.100	0.081	7%
GaSb	0.089	0.077	0.065	17%
InP	0.096	0.100	0.075	-4%
InAs	0.087	0.090	0.065	-3%
InSb	0.080	0.071	0.054	12%

**Surface reconstruction Si(111)**

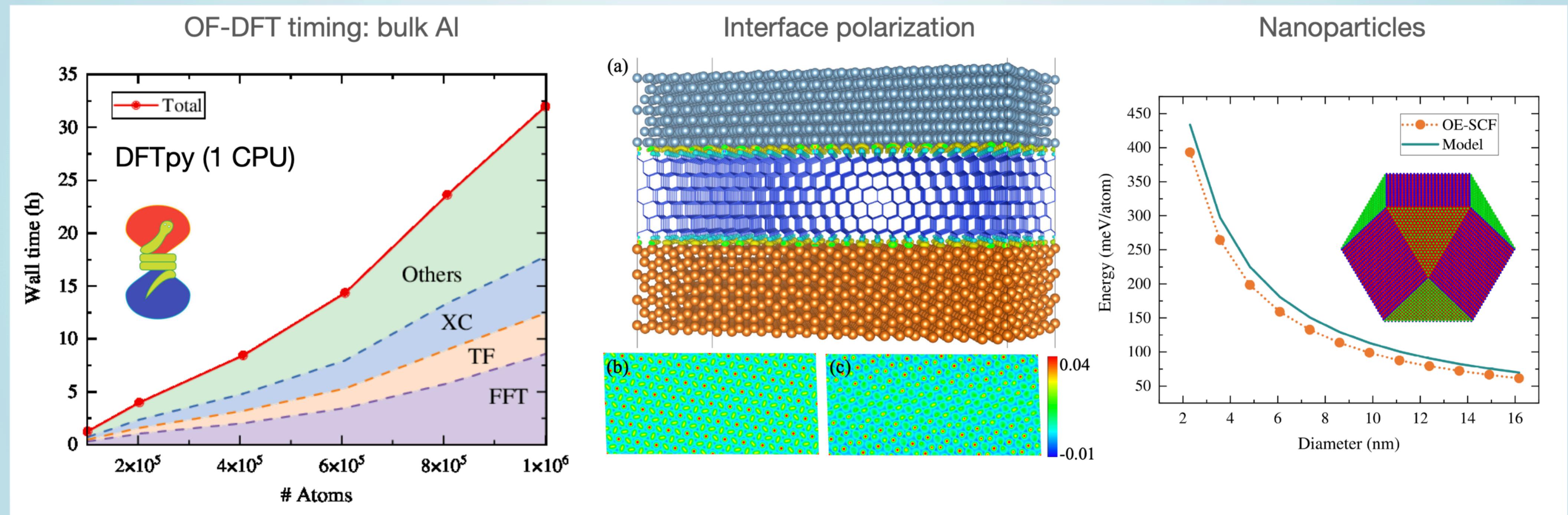


Excellent results for difficult semiconductors and metals!

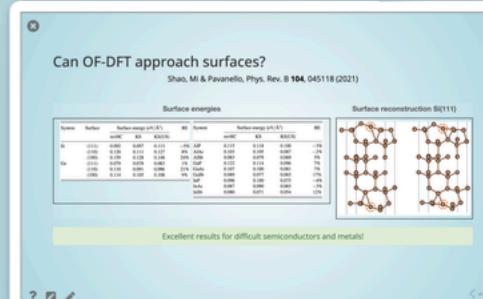


# What else can OF-DFT do?

J. Phys. Chem. Lett. **12**, 4134 (2021) ◆ WIREs: Comp. Mol. Sci. **11**, e1482 (2021)



How about optical spectra and dynamics?





# OF-DFT is limited...



The periodic table displays the following information for each element:

- Symbol:** The element symbol.
- Name:** The element name.
- Atomic Number:** The atomic number (Z).
- Atomic Mass:** The atomic mass (Mw).
- Electron Configuration:** The electron configuration (n s<sup>2</sup>s p<sup>6</sup>p d<sup>10</sup>d f<sup>14</sup>f).

Group	Period	Element	Symbol	Name	Atomic Number	Atomic Mass (Mw)	Electron Configuration
1	1	Hydrogen	H	Hydrogen	1	1.00794	1s <sup>1</sup>
2	1	Lithium	Li	Lithium	3	6.941	1s <sup>2</sup> 2s <sup>1</sup>
3	1	Beryllium	Be	Beryllium	4	9.012182	1s <sup>2</sup> 2s <sup>2</sup>
4	1	Magnesium	Mg	Magnesium	12	24.305	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup>
5	1	Sodium	Na	Sodium	11	22.98976928	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>1</sup>
6	1	Chromium	Cr	Chromium	24	51.9961	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>5</sup>
7	1	Titanium	Ti	Titanium	22	47.867	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>2</sup> 4s <sup>2</sup>
8	1	Vanadium	V	Vanadium	23	50.9415	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>3</sup> 4s <sup>2</sup>
9	1	Nickel	Ni	Nickel	25	54.938045	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>7</sup> 4s <sup>2</sup>
10	1	Cobalt	Co	Cobalt	26	55.845	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>6</sup> 4s <sup>2</sup>
11	1	Iron	Fe	Iron	27	56.93195	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>6</sup> 4s <sup>2</sup>
12	1	Nickel	Ni	Nickel	28	56.6934	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>7</sup> 4s <sup>2</sup>
13	1	Copper	Cu	Copper	29	63.546	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>1</sup>
14	1	Zinc	Zn	Zinc	30	65.38	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup>
15	1	Gallium	Ga	Gallium	31	69.729	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>1</sup>
16	1	Germanium	Ge	Germanium	32	72.64	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>2</sup>
17	1	Arsenic	As	Arsenic	33	74.9216	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>3</sup>
18	1	Selenium	Se	Selenium	34	78.96	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>4</sup>
19	1	Krypton	Br	Krypton	35	83.798	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>6</sup>
20	1	Potassium	Kr	Potassium	36	39.948	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>8</sup>
21	1	Calcium	Ca	Calcium	37	40.078	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>8</sup>
22	1	Scandium	Sc	Scandium	38	44.9559	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>9</sup>
23	1	Titanium	Ti	Titanium	39	47.867	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>10</sup>
24	1	Vanadium	V	Vanadium	40	50.9415	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>11</sup>
25	1	Chromium	Cr	Chromium	41	51.9961	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>12</sup>
26	1	Manganese	Mn	Manganese	42	54.938045	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>13</sup>
27	1	Ferrum	Fe	Ferrum	43	55.845	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>14</sup>
28	1	Cobalt	Co	Cobalt	44	56.93195	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>15</sup>
29	1	Nickel	Ni	Nickel	45	58.6934	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>16</sup>
30	1	Copper	Cu	Copper	46	63.546	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>17</sup>
31	1	Zinc	Zn	Zinc	47	65.38	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>18</sup>
32	1	Gallium	Ga	Gallium	48	69.729	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>19</sup>
33	1	Germanium	Ge	Germanium	49	74.9216	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>20</sup>
34	1	Arsenic	As	Arsenic	50	78.96	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>21</sup>
35	1	Selenium	Se	Selenium	51	83.798	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>22</sup>
36	1	Krypton	Kr	Krypton	52	87.948	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>23</sup>
37	1	Rubidium	Rb	Rubidium	53	88.90585	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>24</sup>
38	1	Strontium	Sr	Strontium	54	89.90585	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>25</sup>
39	1	Yttrium	Y	Yttrium	55	91.224	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>26</sup>
40	1	Zirconium	Zr	Zirconium	56	92.9063	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>27</sup>
41	1	Niobium	Nb	Niobium	57	95.96	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>28</sup>
42	1	Molybdenum	Mo	Molybdenum	58	[98]	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>29</sup>
43	1	Technetium	Tc	Technetium	59	101.07	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>30</sup>
44	1	Ruthenium	Ru	Ruthenium	60	102.9055	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>31</sup>
45	1	Rhodium	Rh	Rhodium	61	106.42	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>32</sup>
46	1	Palladium	Pd	Palladium	62	107.8682	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>33</sup>
47	1	Silver	Ag	Silver	63	112.411	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>34</sup>
48	1	Cadmium	Cd	Cadmium	64	114.818	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>35</sup>
49	1	Indium	In	Indium	65	118.71	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>36</sup>
50	1	Thallium	Tl	Thallium	66	121.76	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>37</sup>
51	1	Lead	Pb	Lead	67	126.90447	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>38</sup>
52	1	Bismuth	Bi	Bismuth	68	131.293	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>39</sup>
53	1	Polonium	Po	Polonium	69	137.327	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>40</sup>
54	1	Astatine	At	Astatine	70	147.948	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>41</sup>
55	1	Radon	Rn	Radon	71	157.948	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>42</sup>
56	1	Barium	Ba	Barium	72	178.49	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>43</sup>
57	1	Lanthanoids	La	Lanthanum	73	180.94788	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>44</sup>
58	1	Hafnium	Ce	Cerium	74	183.84	1s <sup>2</sup> 2s <sup>2</sup> 2p <sup>6</sup> 3s <sup>2</sup> 3p <sup>6</sup> 3d <sup>10</sup> 4s <sup>2</sup> 4p <sup>45</sup>
59	1</td						



# OF-DFT is limited... but why?





# What are nonlocal PPs?





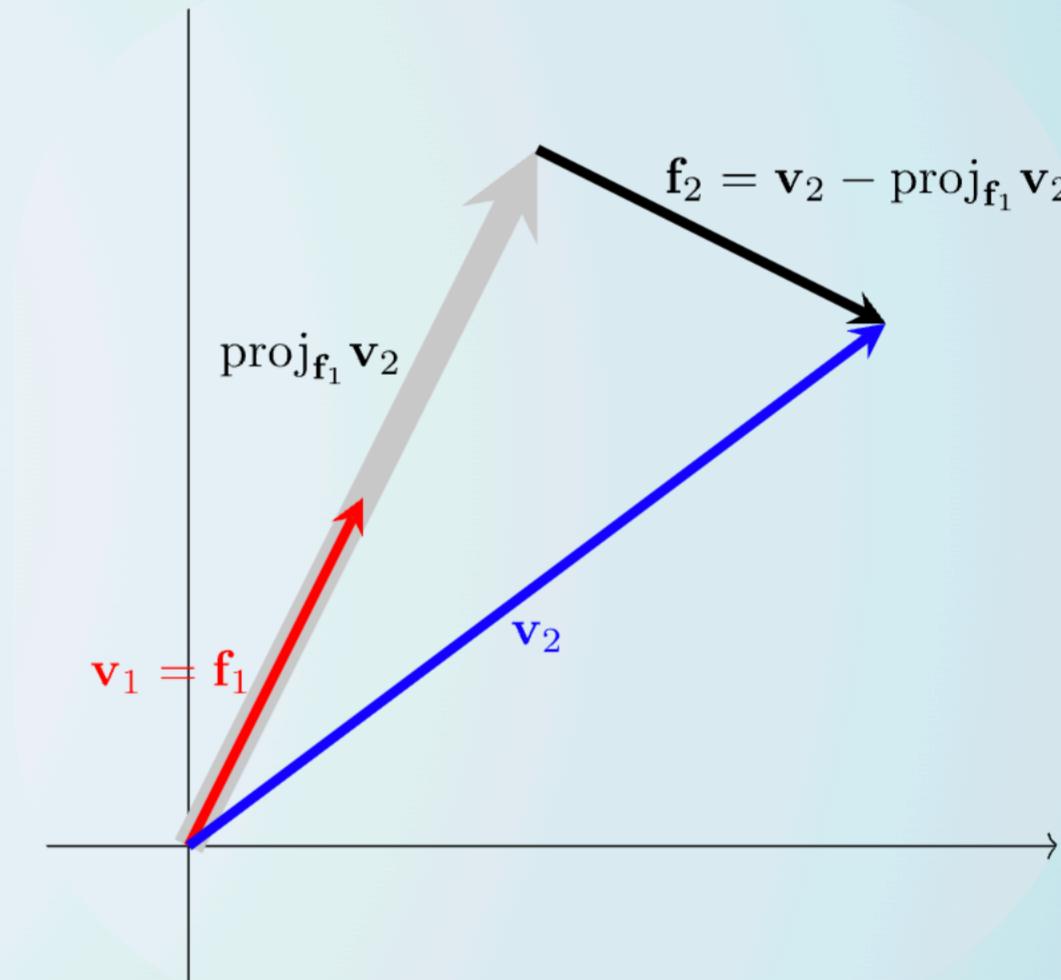
## Core Electrons



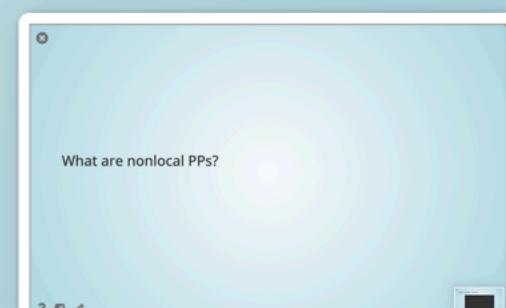
Using pseudo-potentials, core electrons are "hidden"

## Valence Electrons

Need to be orthogonal to the core electrons



The Gram-Schmidt orthogonalization is a good example





## Non-local pseudo-potentials explained

Apply Gram-Schmidt to the set of valence,  $|\phi\rangle$ , and core,  $|\text{core}\rangle$ , electrons:

$$|\phi\rangle \rightarrow |\phi\rangle - |\text{core}\rangle\langle\text{core}|\phi\rangle = \left[1 - |\text{core}\rangle\langle\text{core}|\right]|\phi\rangle$$

$\hat{P}_{nl} = \left[1 - |\text{core}\rangle\langle\text{core}|\right]$  is the **non-local pseudopotential**

In OF-DFT  $|\phi\rangle$  is not available...





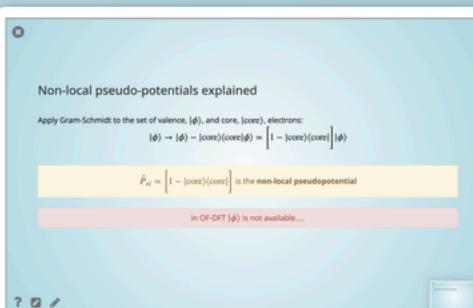
# Poll time / challenge

- How would you fix OF-DFT's impossibility to use nonlocal PPs?

**A:** Use KS-DFT for atoms only, or smaller fragments of the system..and then combine them together with OF-DFT

**B:** Throw OF-DFT away!

**C:** Introduce nonlocal pseudopotentials in OF-DFT somehow





# Subsystem DFT (sDFT) - Theory

## Idea:

- Split system into (smaller) subsystems:  $n(\mathbf{r}) = \sum_I n_I(\mathbf{r})$
- Use KS-DFT **or** OF-DFT for the subsystems
- Energy is split in *additive* and *non-additive* terms:

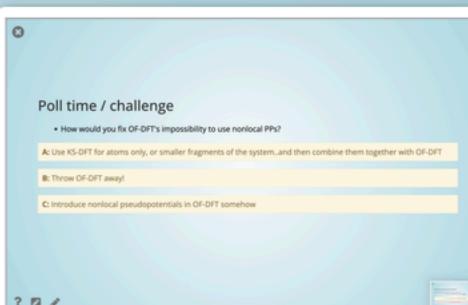
$$E[n] \equiv E[\{n_I\}] = \sum_I E[n_I] + E^{nadd}[\{n_I\}]$$

- The nonadditive energy given by:

$$E^{nadd}[\{n_I\}] = E_{Coul}[n] - \sum_I E_{Coul}[n_I] + E_{xc}[n] - \sum_I E_{xc}[n_I] + T_s[n] - \sum_I T_s[n_I]$$

**Problem:** The non-additive energy contains the non-additive kinetic energy,  $T_s^{nadd}[\{n_I\}]$ , which not known.

**Opportunity:** We can use *fancy non-local*  $T_s^{nadd}[\{n_I\}]$ , previously developed for OF-DFT.





# sDFT implementations

Our codes:

eQE	eDFTpy
2017-2022	2022-present

The competition:

Serenity	CP2k
Neugebauer (Münster)	Luber, Hutter (Zurich)



# Want to learn more about eDFTpy?

```
In [21]: IFrame('http://edftpy.rutgers.edu', width=1000, height=400)
```

Out[21]:

The screenshot shows the homepage of the eDFTpy website. On the left, there is a sidebar with a red logo consisting of two interlocking shapes, a search bar labeled "Search docs", and links for "Contact", "Installation", and "Tutorials". The main content area has a header with a house icon, the text "» eDFTpy: Density Embedding Scheme with Python", and navigation links for "Index" and "Source". Below the header is a "Next" button. The main title "eDFTpy: Density Embedding Scheme with Python" is prominently displayed. A descriptive paragraph explains that eDFTpy combines KS-DFT and OF-DFT, achieving linear scaling with system size. It mentions that the code is based on DFTpy and developed by PRG at Rutgers University-Newark.

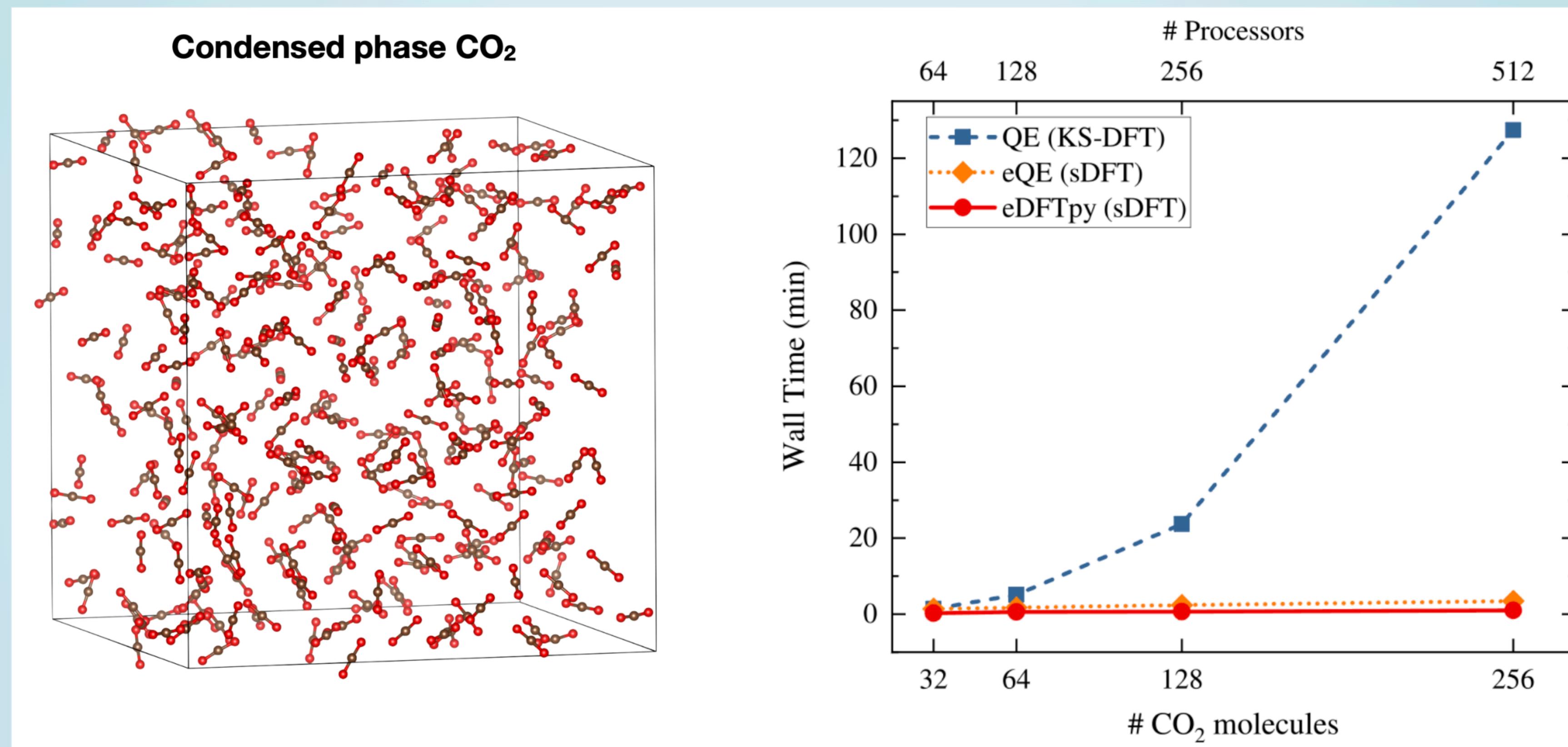
eDFTpy: Density Embedding Scheme with Python

eDFTpy combines Kohn-Sham Density Functional Theory (KS-DFT) and orbital-free DFT (OF-DFT) in an embedding framework that distills the best of both worlds whereby only weak intermolecular interactions are left to OF-DFT, and strong intramolecular interactions are left to KS-DFT. The code achieves near ideal linear-scaling with system size. eDFTpy is based on [DFTpy](#) which is developed by [PRG](#) at [Rutgers University-Newark](#)





# Subsystem DFT (sDFT) - Timings



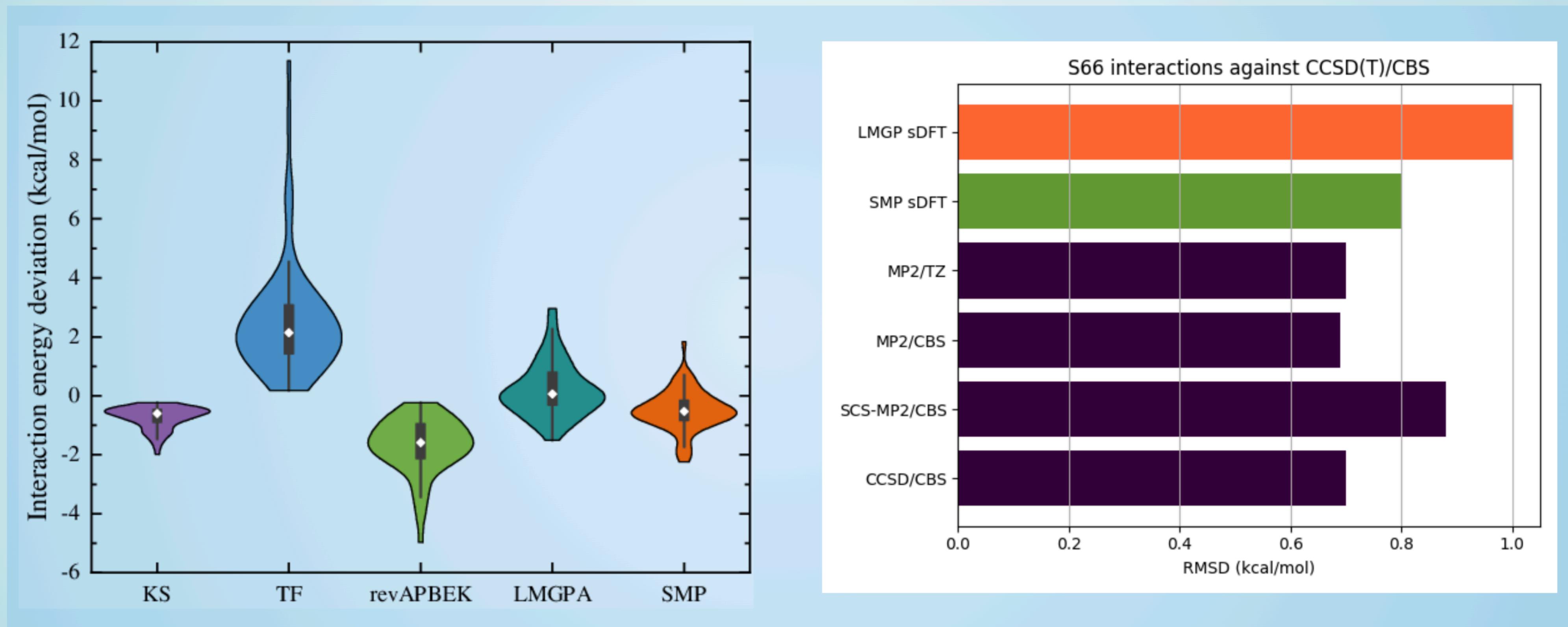
- Comp. Phys. Comm. **269**, 108122 (2021)
- Int. J. Quantum Chem. **117** e25401 (2017)
- J. Phys. Chem. Lett. **10** 7554 (2019)



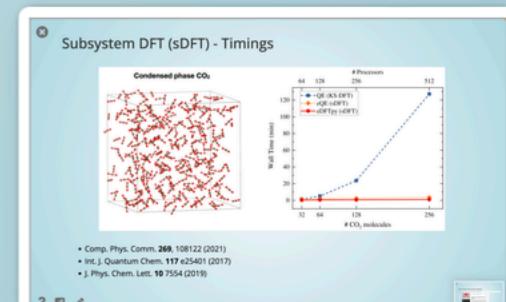


# How accurate is sDFT?

Deviation against CCSD(T) for the S66 test set



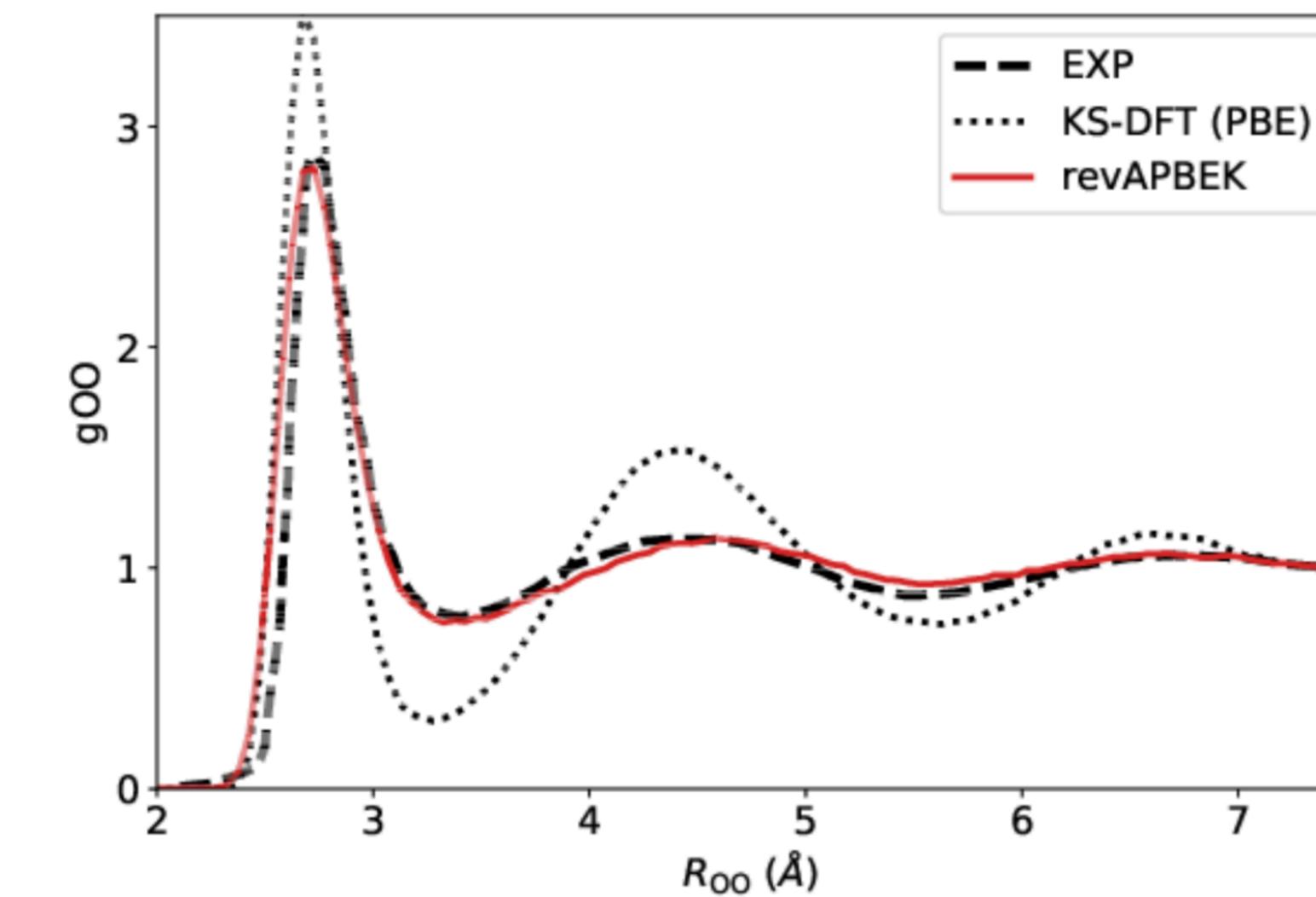
- Comp. Phys. Comm. **269**, 108122 (2021)
- J. Chem. Theory Comput. **17** 3455 (2021)
- J. Phys. Chem. Lett. **11** 272 (2020)
- J. Chem. Theory Comput. **7** 2427 (2011)



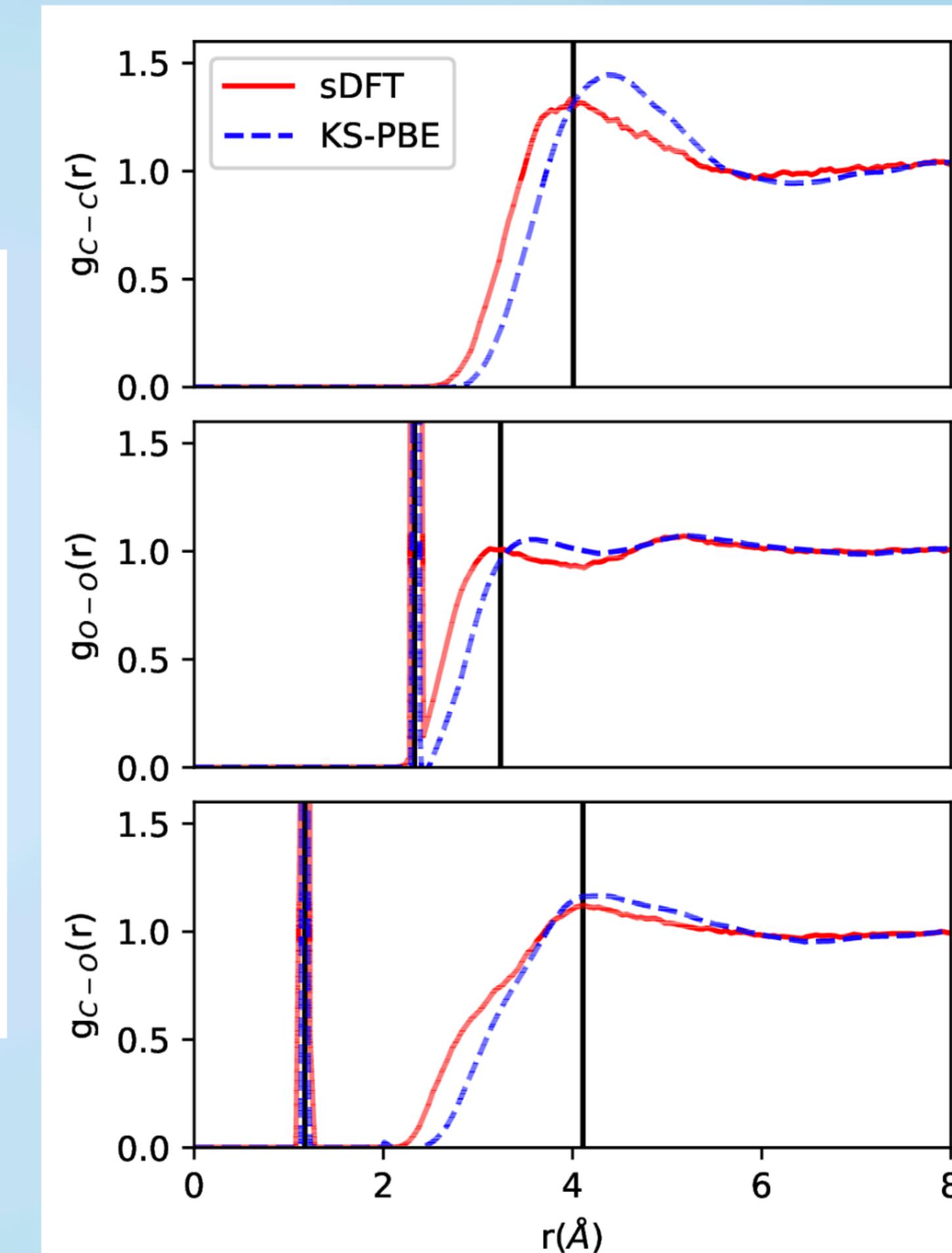


## sDFT success stories

Liquid water



Fluid CO<sub>2</sub>



Structure and dynamics recovered!

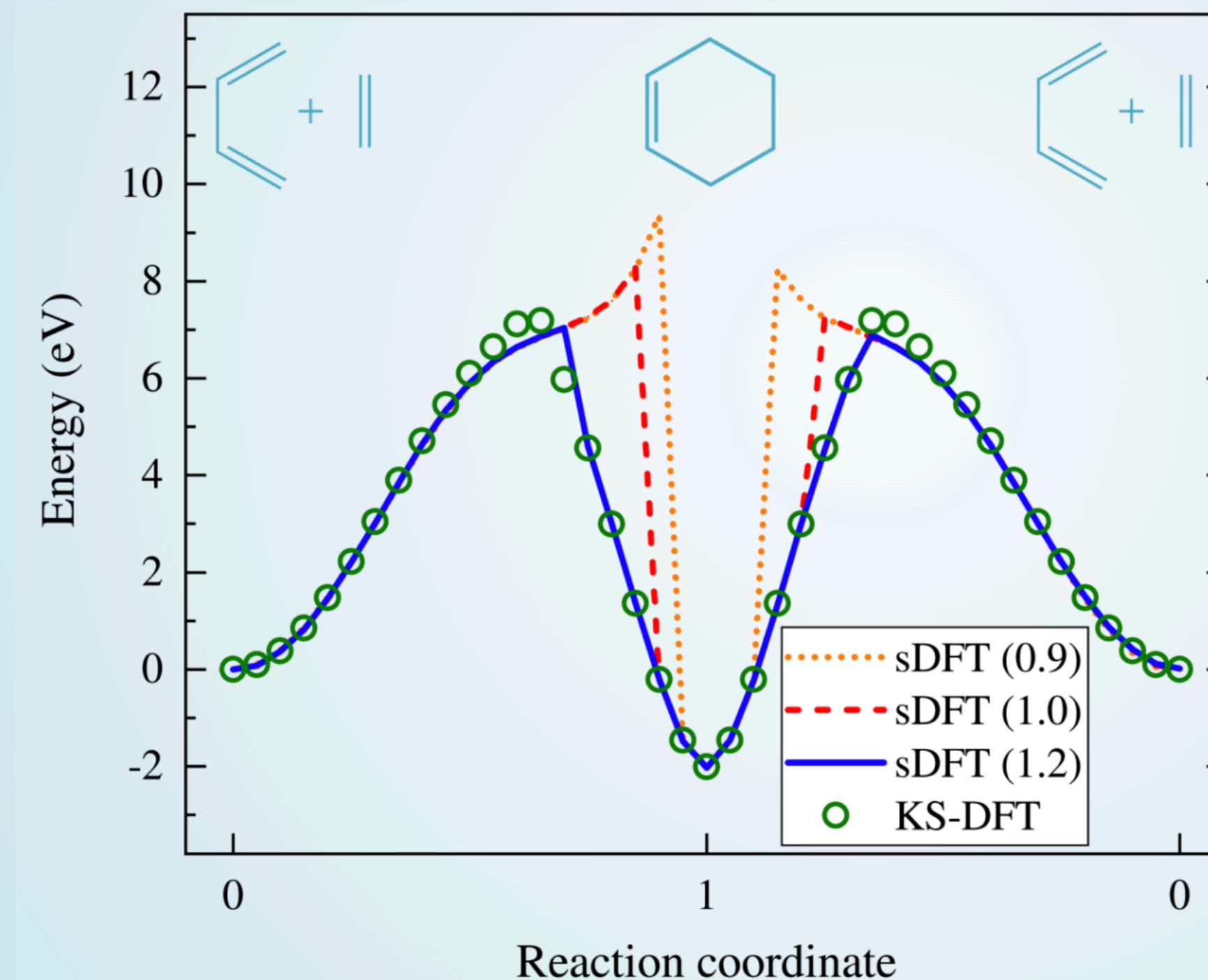




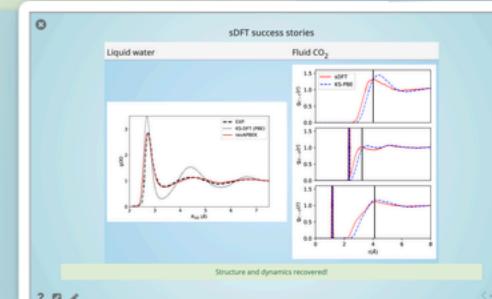
# sDFT is not perfect either...

The nonadditive kinetic energy,  $T_s^{nadd}[n]$ , is not accurate enough when...

- ◆ ...there are covalent bonds between subsystems
- ◆ ...subsystems are too close



Adaptive sDFT has only 1 parameter (rcut) related to the covalent radii.





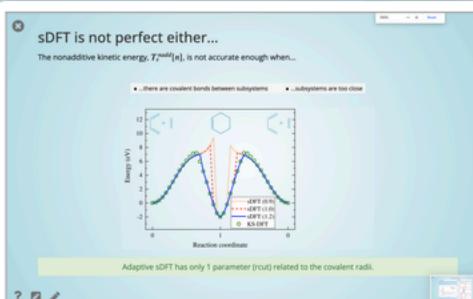
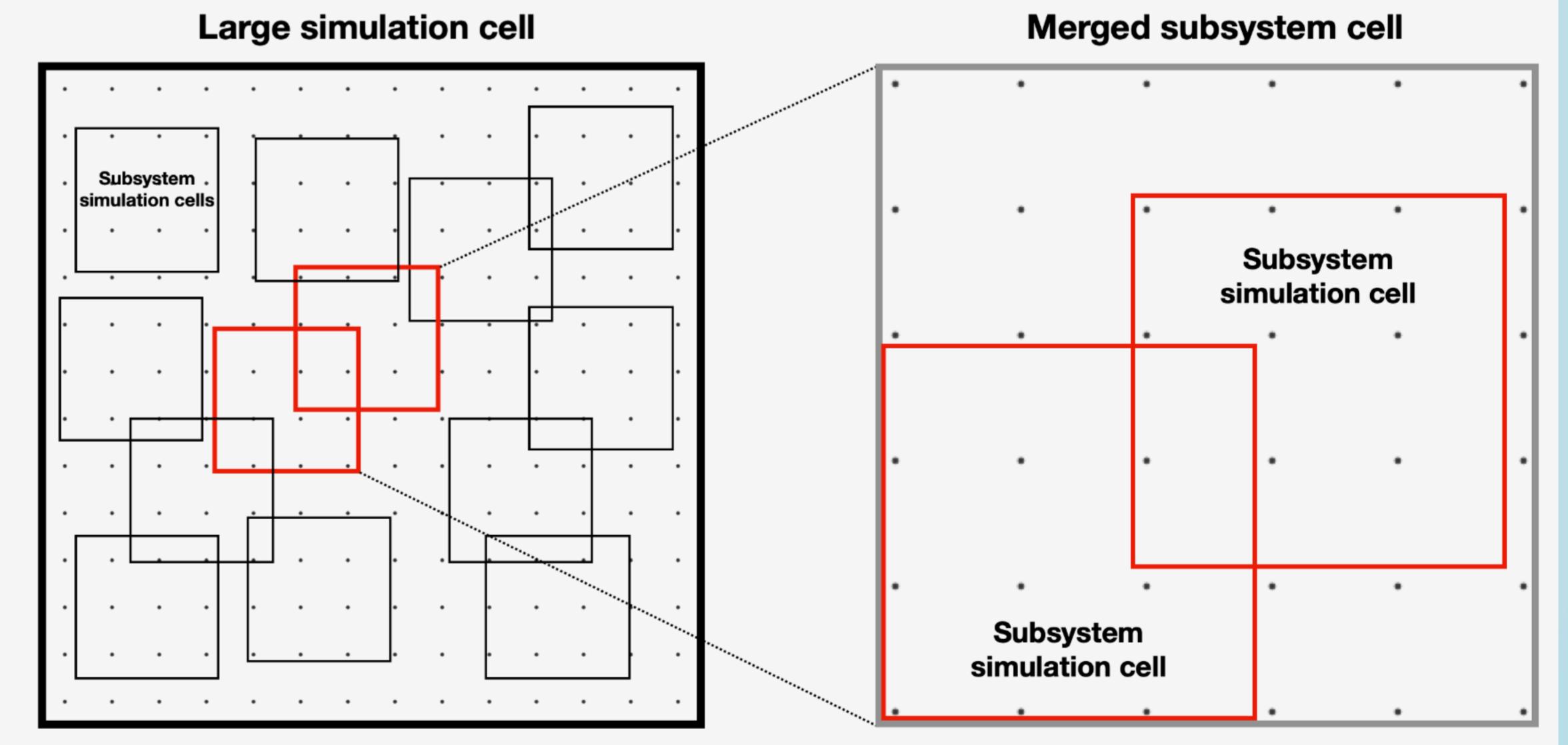
# Adaptive sDFT

To do:

- Must be able to "split" and "merge" subsystems along a dynamics
- Do so *automatically*

We did it... but it was challenging

- Adaptive computational settings
- Adaptive MPI communicators
- Adaptive data allocation



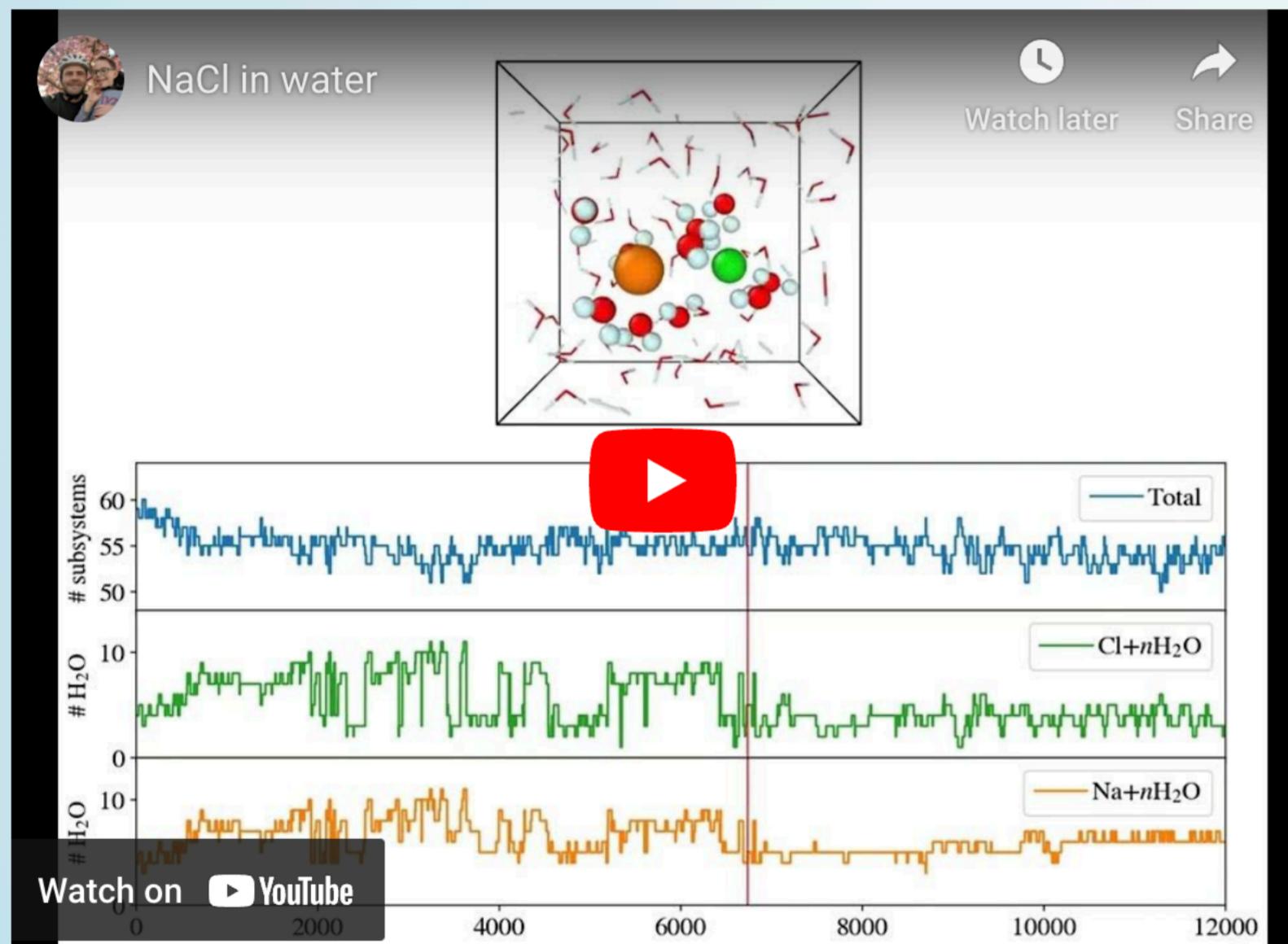


# Adaptive sDFT: dissociation of NaCl in water

<https://www.youtube.com/watch?v=-lmdPHzJfeU>

```
In [22]: YouTubeVideo('-lmdPHzJfeU', width=600, height=450)
```

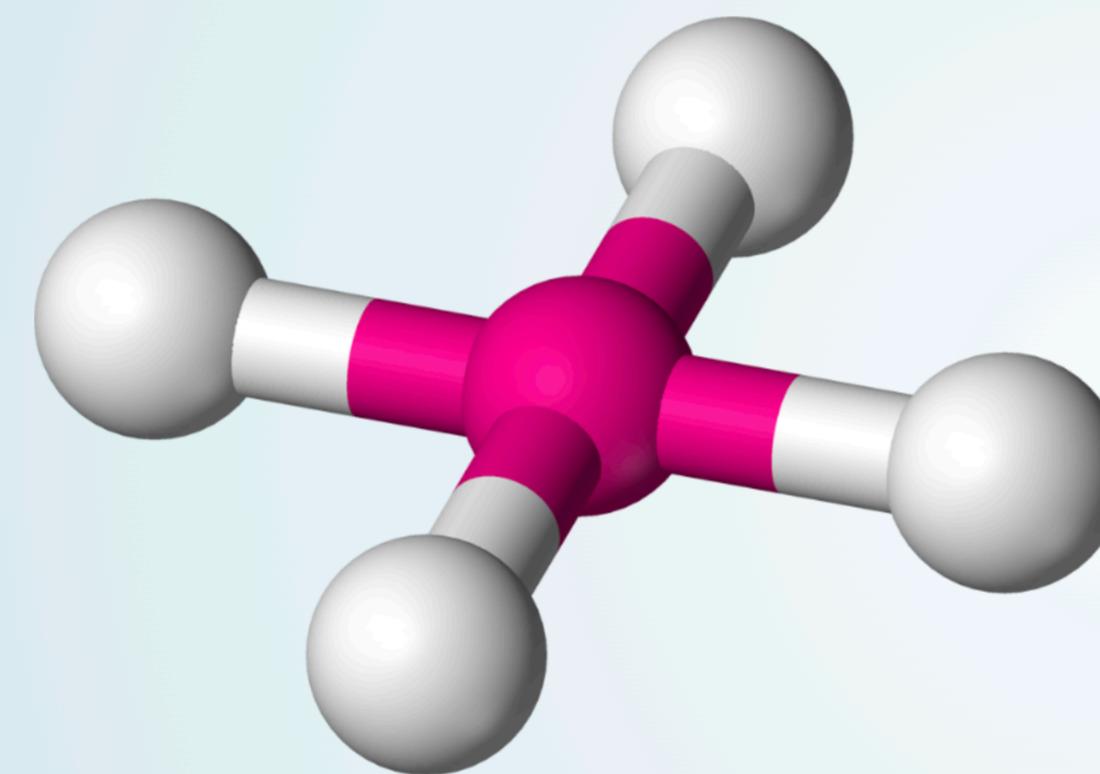
Out[22]:



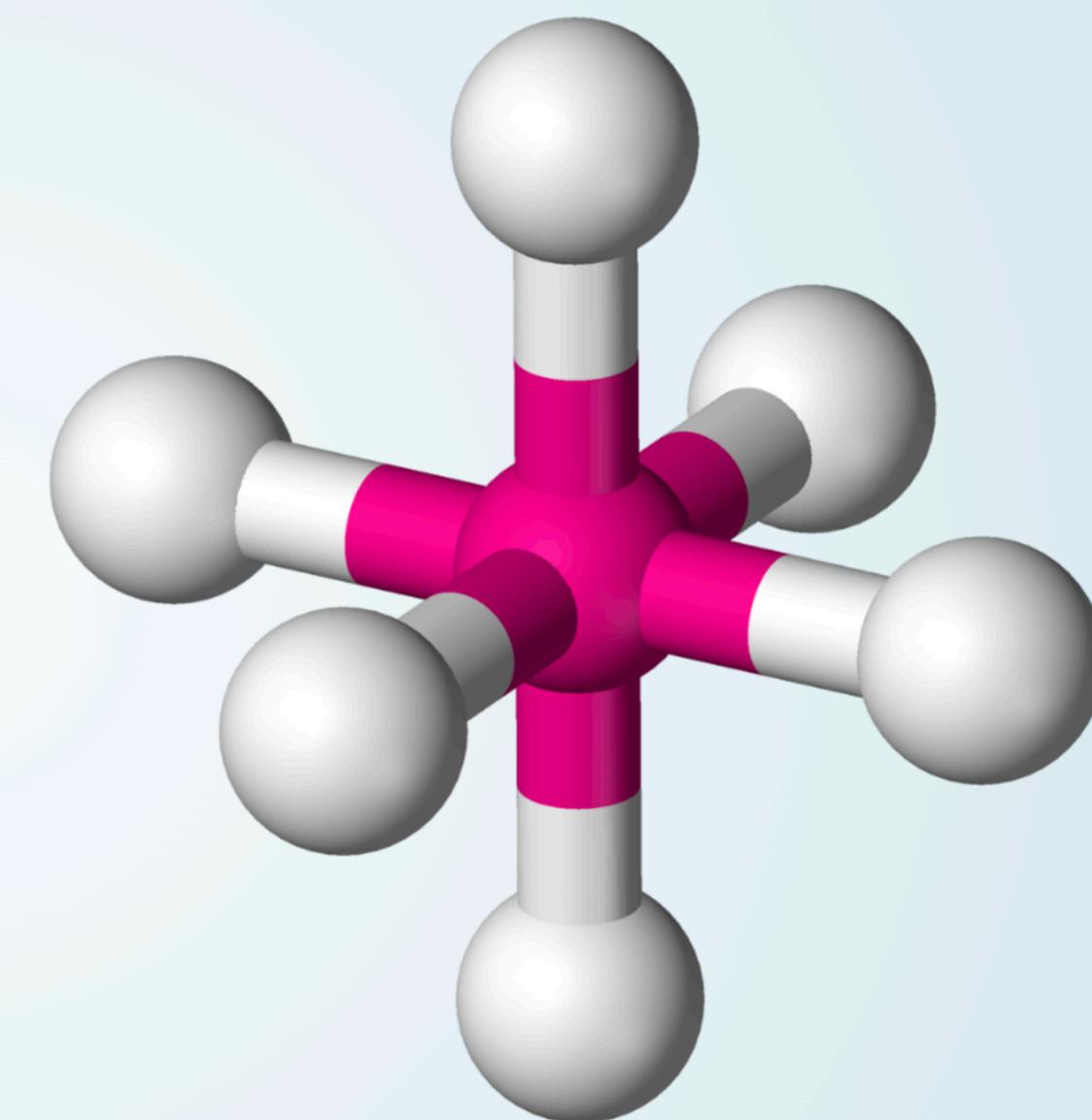


# Adaptive sDFT: hydration of $\text{Ni}^{2+}$ , $\text{Pd}^{2+}$ and $\text{Pt}^{2+}$

Square Planar



Octahedral



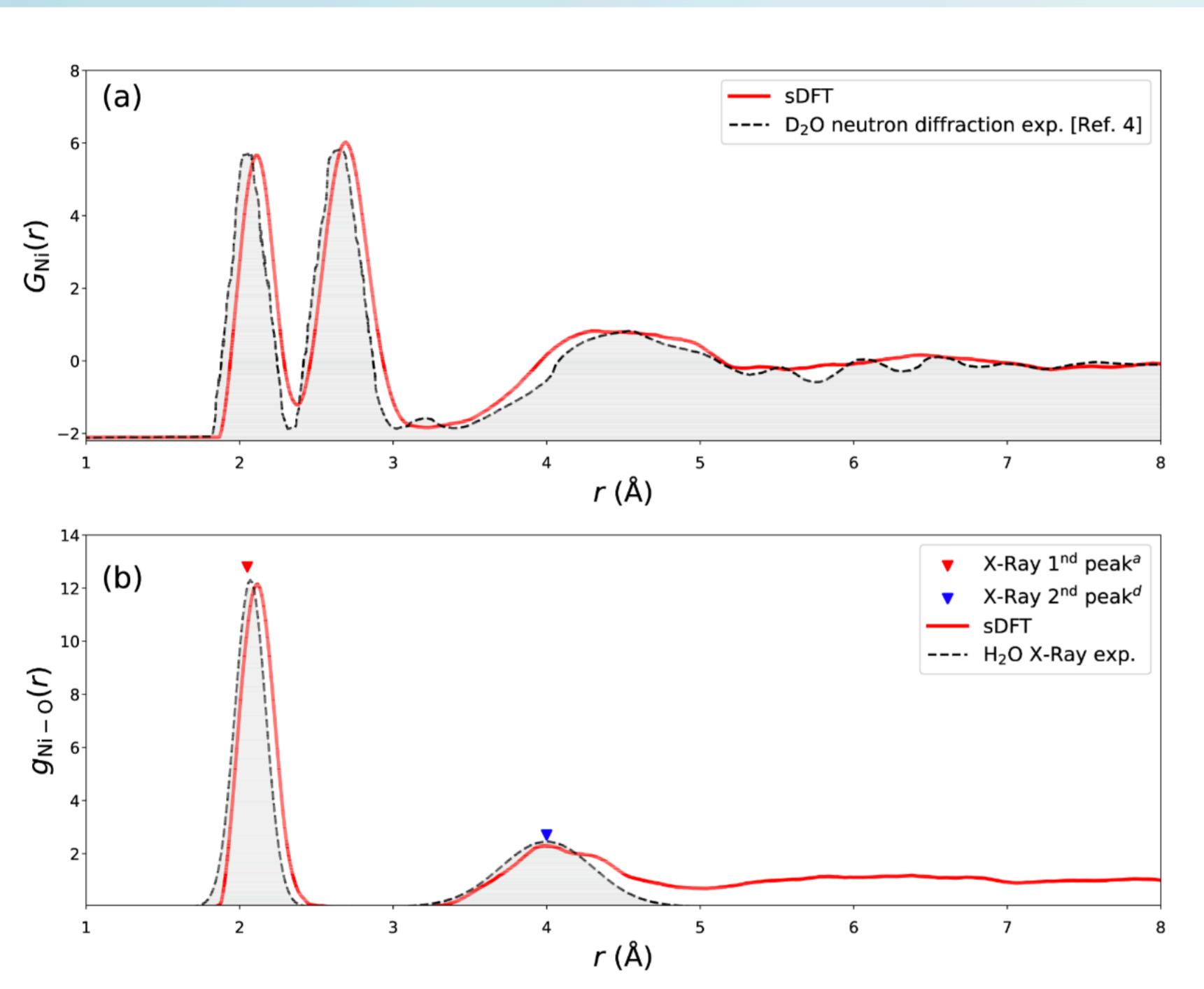
Stay tuned for an upcoming paper!



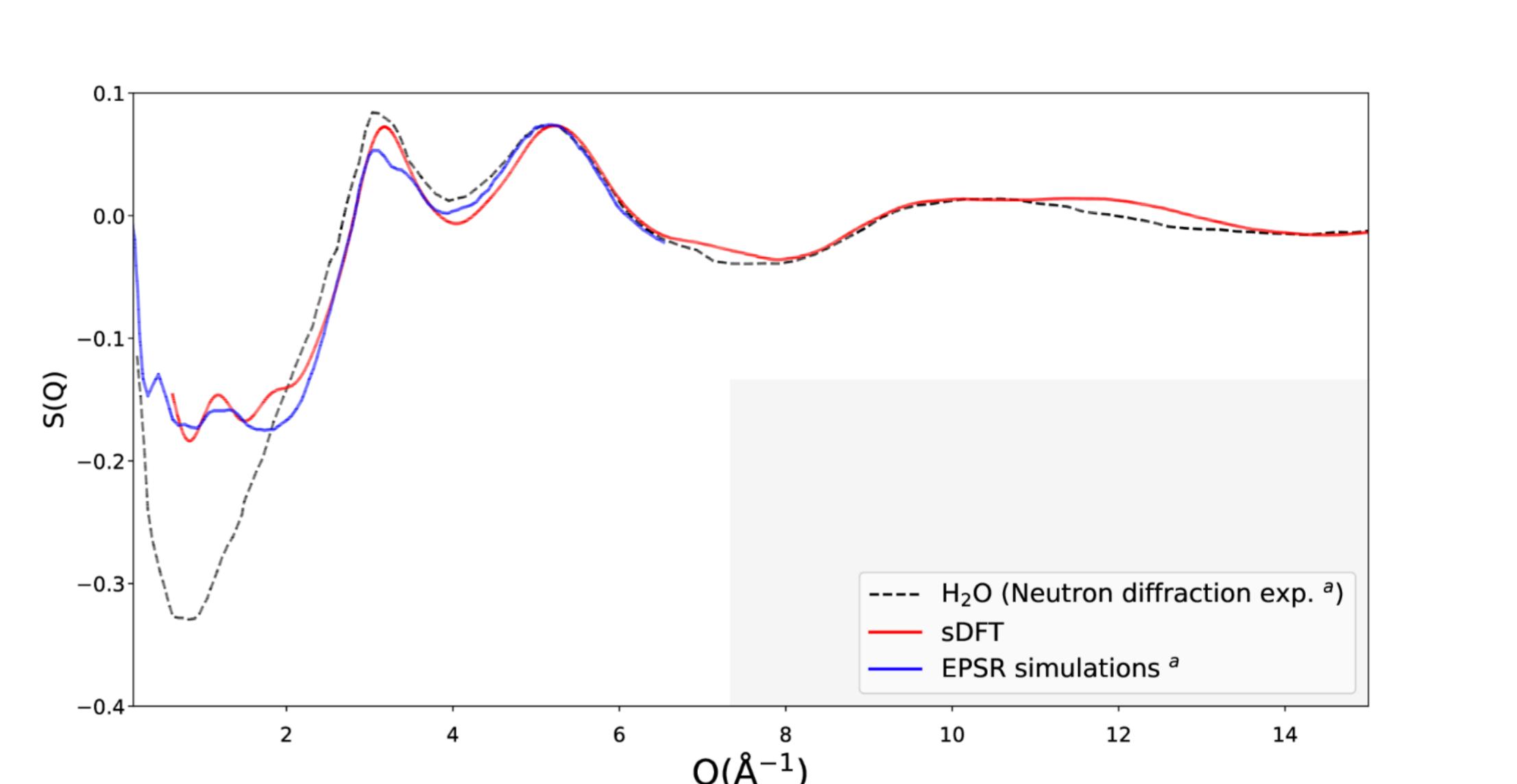


# Comparison with experiments for Ni and Pd

Ni(II) -- Total  $G_{Ni}(r)$  and partial  $G_{Ni-O}(r)$

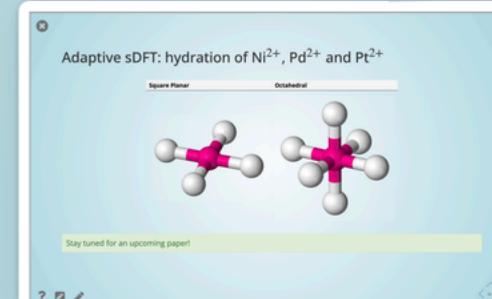


Pd(II) -- Total structure factor  $S(Q)$



- Almost quantitative agreement with experiments
- XRay: Powell et al 1989 J. Phys.: Condens. Matter 1 8721;
- N scattering: Nelson et al J. Phys. C 11, 15 L625 (1978)

- Better agreement with the experiment compared with the EPSR result of Bowron *et al.*





# Despite the speaker, you learned:

## Good things:

- OF-DFT is very fast and linear scaling with system size
- OF-DFT is useful for metals and semiconductors of any shape and size
- Where OF-DFT fails we can use Subsystem DFT. Still linear scaling, and applicable to any inter-subsystem interaction strength
- Therefore, the applicability of KS-DFT is extended to larger, more meaningful systems

## Bad things:

- OF-DFT has limited applicability because of the "nonlocal pseudopotential problem"
- Subsystem DFT is fast but only if subsystems are small

