Impacts of Marine Heatwaves on Ocean Biogeochemistry

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What will you learn in this lecture?



Biological activity and circulation set the mean distribution of chemicals in the ocean

atmosph	nere	
	Air-sea CO ₂ exchange	[carbon]
ocean	🍈 🚥 🔗	
	$\begin{array}{c} \text{CO}_2 + \text{H}_3\text{O} \leftrightarrow \text{H}_3\text{CO}_3\\ \\ \text{Carbonate}\\ \text{chemistry} \\ \text{H}^+ & \text{HCO}_3\\ \\ \text{H}^+ & \text{CO}_3^-\end{array}$	Vertical gradient

Gases and chemistry in the ocean

- Solubility
- Carbonate chemistry



Marine heatwaves impact multiple ocean biogeochemical processes

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Breakout question (in groups of 4 - 5 people)

This map shows the annualmean surface ocean nitrate concentration $[NO_3^{-1}]$.

- Why are the tropical Pacific and subpolar regions characterized by such high nitrate concentrations?
- 2. Why are the subtropics characterized by near-zero nitrate concentrations?



The role of biology



<u>Photosynthesis</u> 106 CO₂ + 16 HNO₃ + H₃PO₄ →organic matter + 150 O₂

 $\frac{\text{Remineralization}}{\text{organic matter} + 150 \text{ O}_2}$ $\rightarrow 106 \text{ CO}_2 + 16 \text{ HNO}_3 + \text{H}_3\text{PO}_4$

The role of biology





The role of biology



Nutrient upwelling







Sarmiento and Gruber (2006)

26 24 22

Liang et al. (2017)

The cruise you would never want to go on



Breakout question

This figure shows the nitrate concentration $[NO_3^{-1}]$ on the cruise you would never want to go on.

Why is $[NO_3^{-1}]$ so elevated in the deep North Pacific?



Signatures of remineralization in the thermohaline circulation



Sarmiento and Gruber (2006); Emerson and Hedges (2008)



The North Pacific subsurface is characterized by the highest nutrient and carbon concentrations and the lowest oxygen concentrations in the global ocean.



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	Carbonate chemistry H HCO ₃	gradient
	↓ ● H* CO320	

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Breakout question



- If you open a can of soda and leave it out, it will eventually go "flat". Why does this happen?
- 2. If you wanted to prevent it from going flat, would you put it on the warm countertop or in the cold refrigerator?

Gas solubility is a function of temperature



Carbonate chemistry



Dissolved Inorganic Carbon



Dissolved Inorganic Carbon (DIC) DIC is the sum of the concentrations of all the inorganic carbon species in the ocean

 $\mathsf{DIC} = [\mathsf{H}_{2}\mathsf{CO}_{3}] + [\mathsf{HCO}_{3}^{-}] + [\mathsf{CO}_{3}^{2-}]$

Breakout question



Dissolved Inorganic Carbon (DIC) DIC is the sum of the concentrations of all the inorganic carbon species in the ocean

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If the ocean absorbs anthropogenic CO₂ from the atmosphere,

- Does DIC increase or decrease?
- Does [H⁺] increase or decrease?

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Schematic illustration of possible MHW impact on BGC















An example: The Blob



Mogen et al. (2022)

An example: The Blob





Mogen et al. (2022)

An example: The Blob





Mogen et al. (2022)



Compound extremes

correlation of SST and [H+]





Burger et al. (2022)

Breakout question



This afternoon, we will do a 'hands on' activity wherein we quantify the changes in biogeochemistry associated with 'Blob 2.0'.

Make a prediction How will Blob 2.0 impact...

- Surface ocean oxygen
- Surface ocean carbon (DIC)
- Phytoplankton biomass (chlorophyll)

??

What did you learn in this lecture?



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