

 COLUMBIA CLIMATE SCHOOL
LAMONT-DOHERTY EARTH OBSERVATORY

The spectral roots of hydrologic sensitivity:
why warmer atmospheres rain more

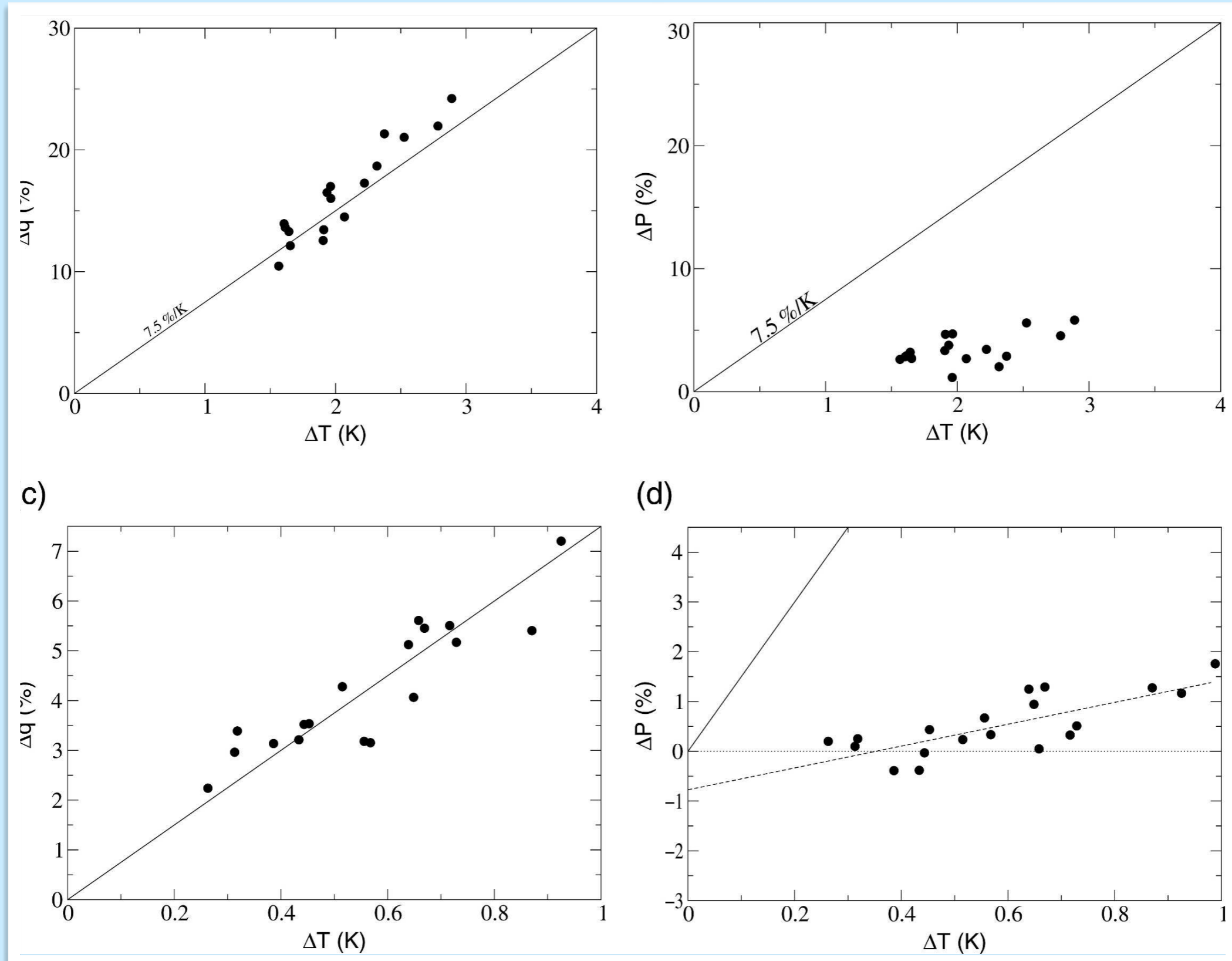
Robert Pincus
Lamont-Doherty Earth Observatory

Warmer atmospheres rain more

TABLE 3. CHANGES DUE TO DOUBLING CO₂ AND ENHANCING SEA SURFACE TEMPERATURES (GLOBAL 2-YEAR MEAN)

	Tropospheric temperature (K)	Atmospheric moisture content (%)	Surface temperature		Precipitation (%)
			Globe (K)	Land (K)	
11 LM	3.08	20	2.26	3.05	5.6
5 LM	3.02	18	2.21	2.86	4.9

Warmer atmospheres rain more



Energy conservation and global rainfall

Water balance within the atmosphere requires

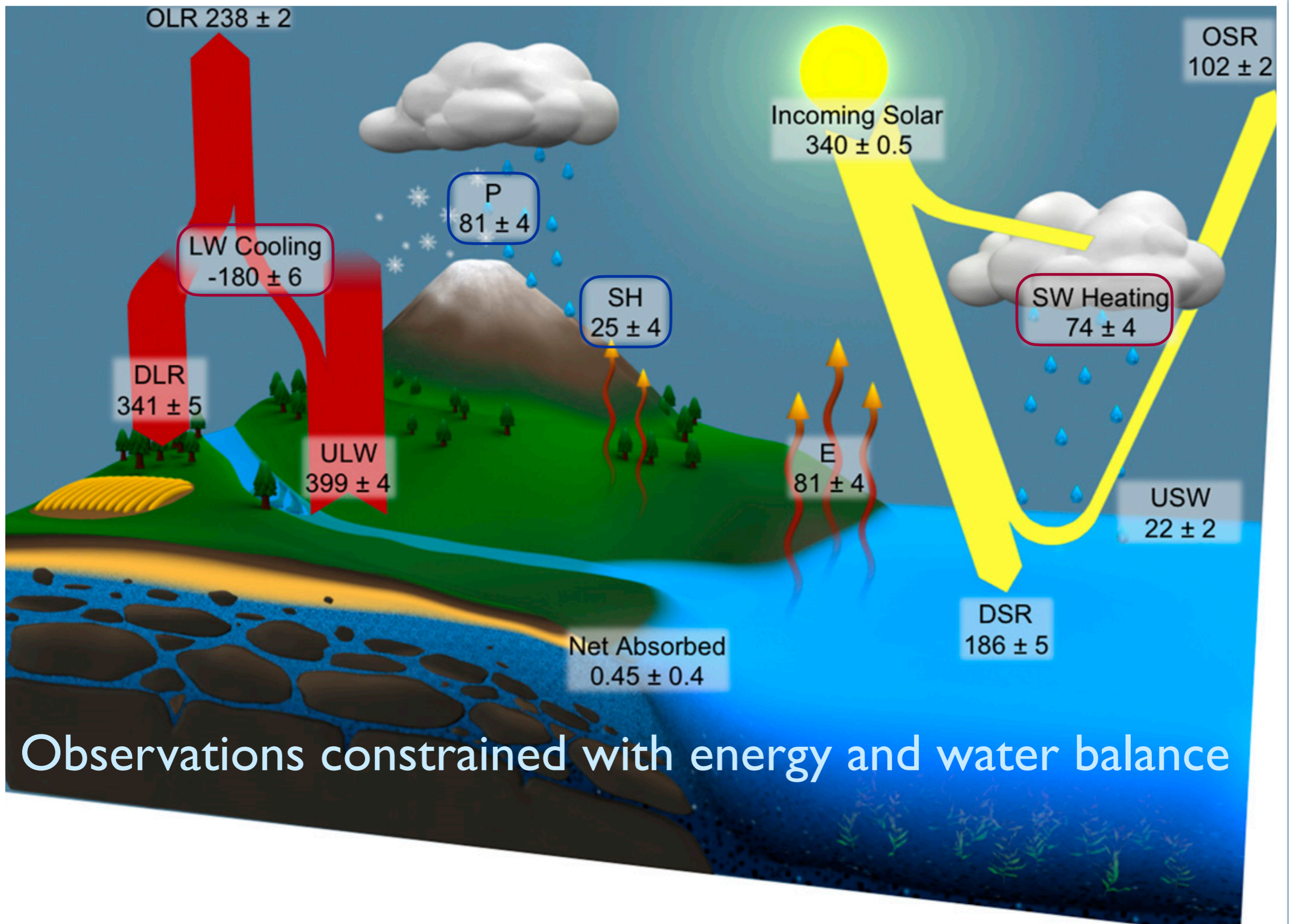
$$\text{Surface evaporation} - \text{surface precipitation} = 0$$

Energy balance within the atmosphere requires

Net atmospheric radiative heating + latent heat flux + sensible heat flux = 0
or, in energy units,

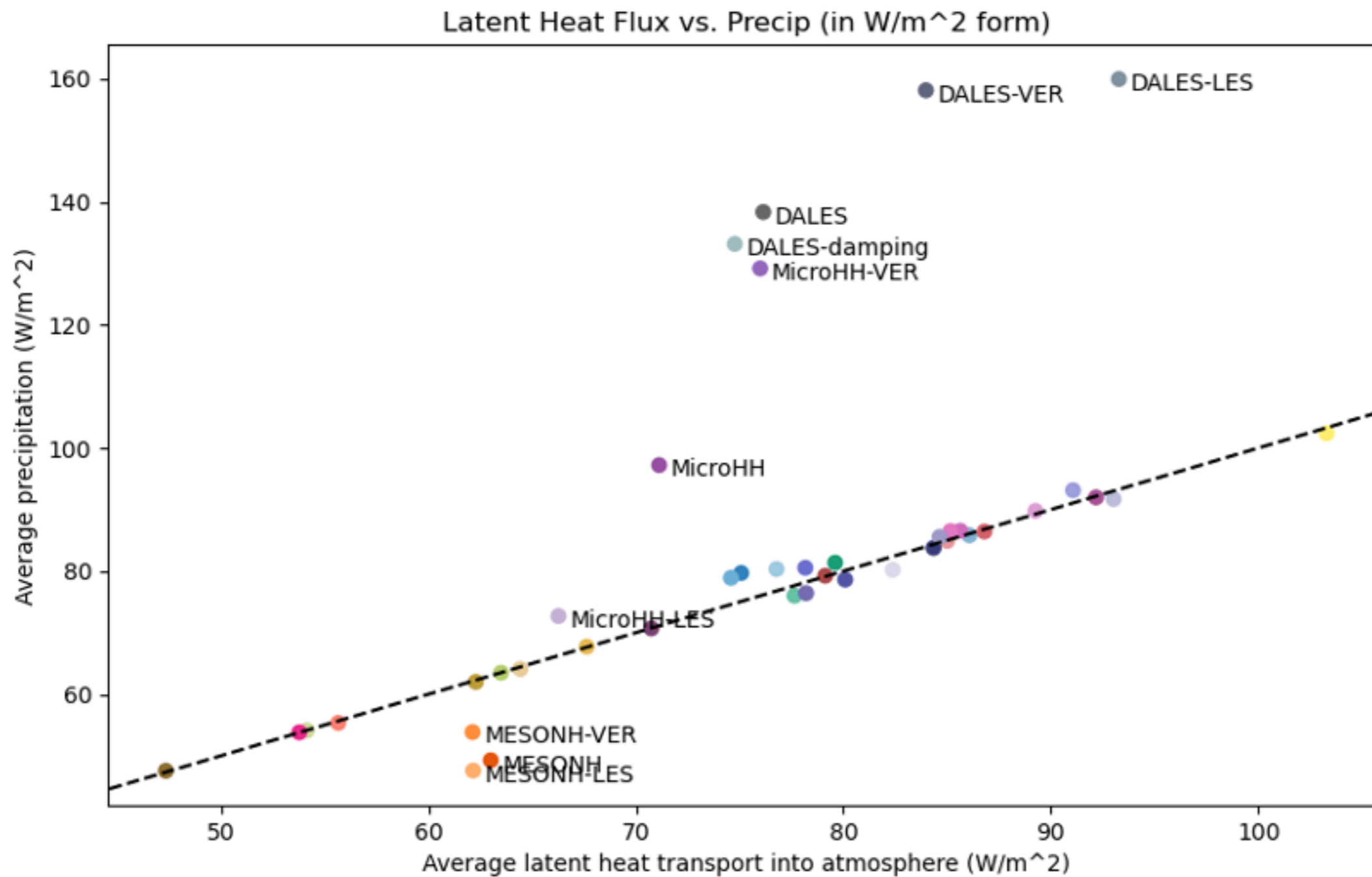
$$H_{SW} - Q_{LW} + L_v P + SHF = 0$$

In Earth's atmosphere $H_{SW} \ll |Q_{LW}|$ and $L_v P \ll SHF$



Observations constrained with energy and water balance

(Results from RCEMIP - beware diagnostics)



Radiation change constrains precipitation change with temperature

How does precipitation with temperature (what is the “hydrologic sensitivity”)?

$$L_v \frac{dP}{dT_S} = \frac{dQ_{LW}}{dT_S} - \frac{dH}{dT_S} - \frac{dSHF}{dT_S}$$

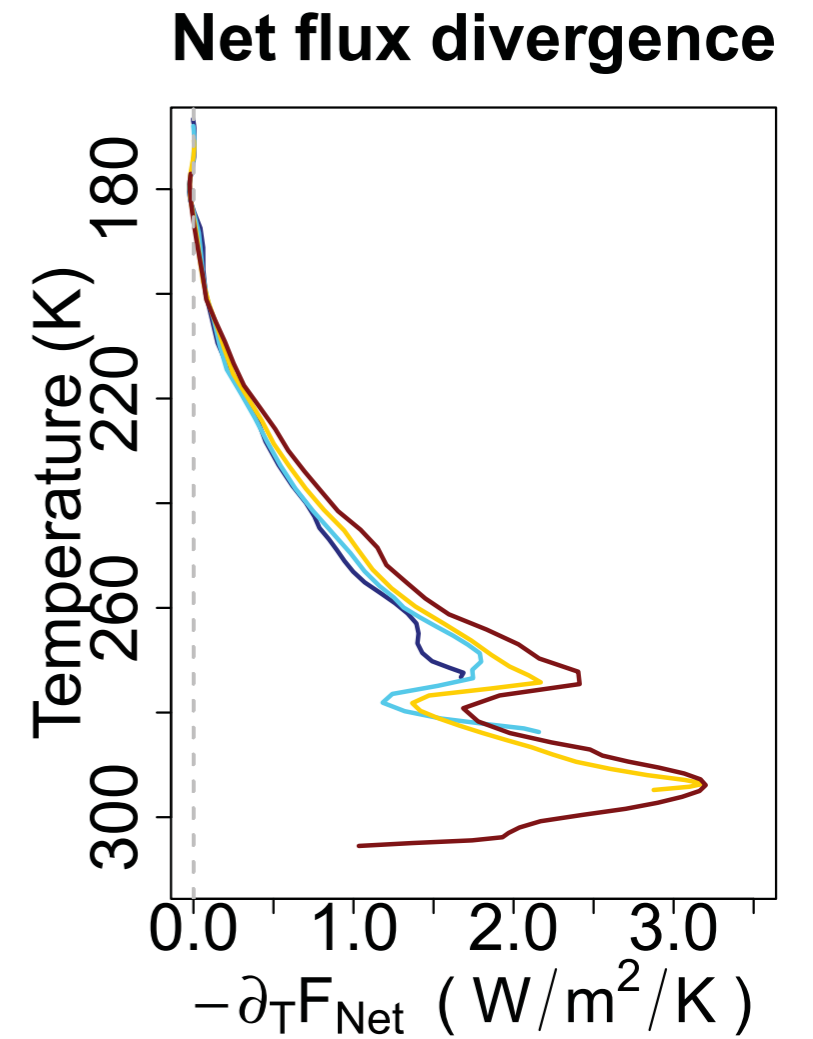
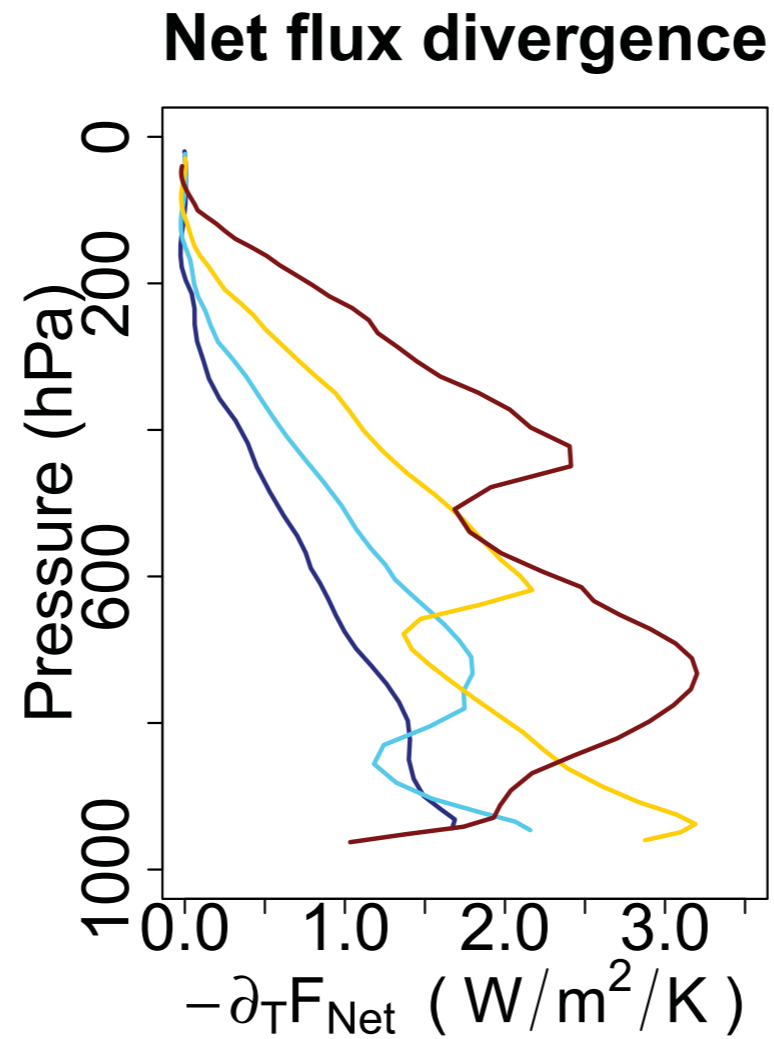
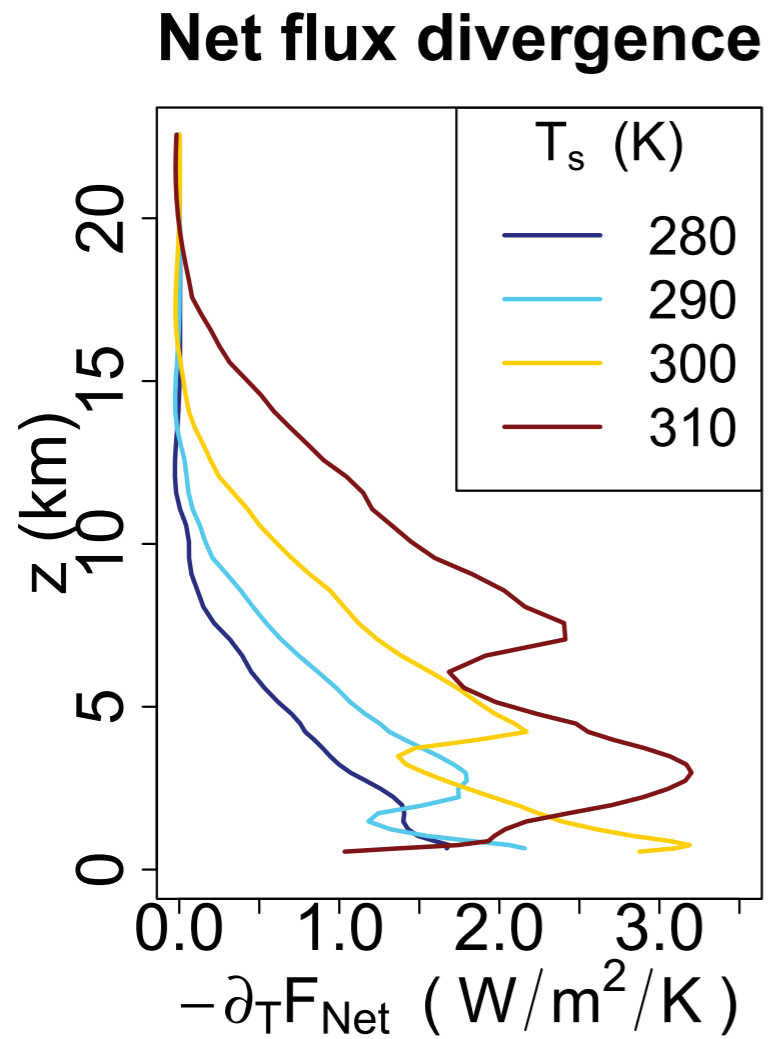
In climate change simulations latent heat fluxes often decrease a little with temperature but to first order

$$L_v \frac{dP}{dT_S} \approx \boxed{\frac{dQ_{LW}}{dT_S}} - \frac{dH}{dT_S} \sim \frac{dQ_{LW}}{dT_S}$$

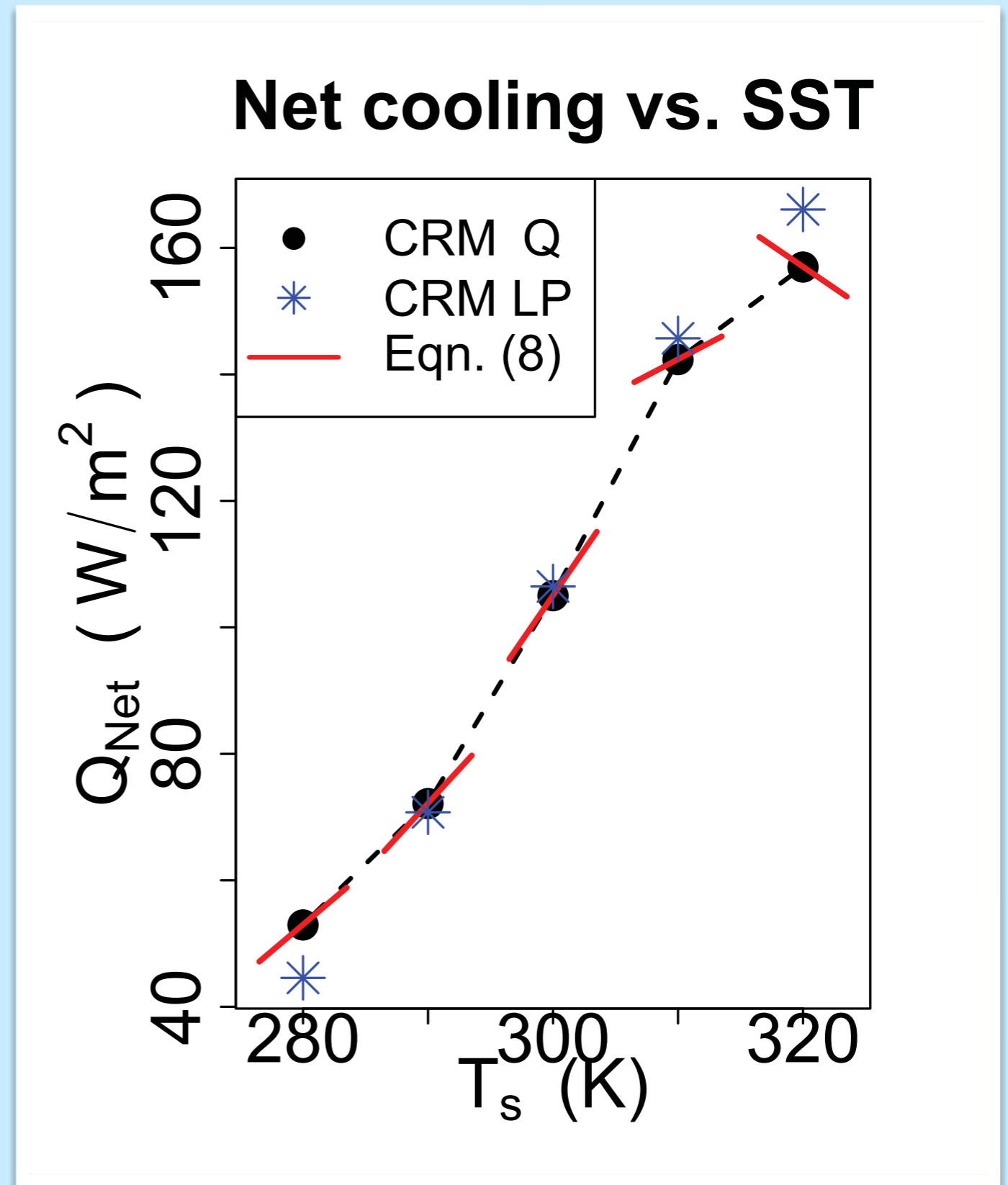
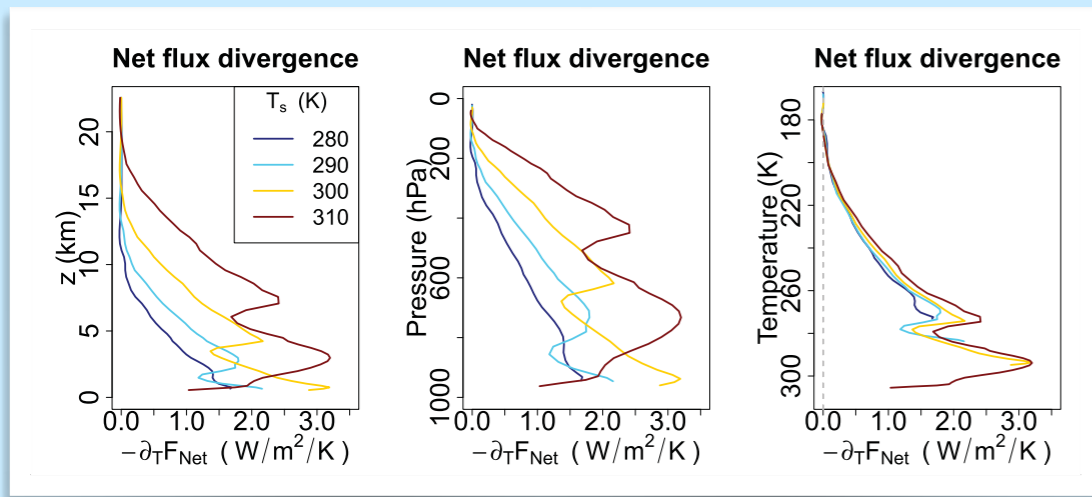
Can we understand $\frac{dQ_{LW}}{dT_S}$?

(Spoiler alert: everything depends on radiative transfer and water vapor thermodynamics)

“The atmosphere deepens in temperature coordinates”



“The atmosphere deepens in temperature coordinates”



An explanation: Simpson's "law"

To the extent that

The absorption coefficient of water vapor is constant with T, p and

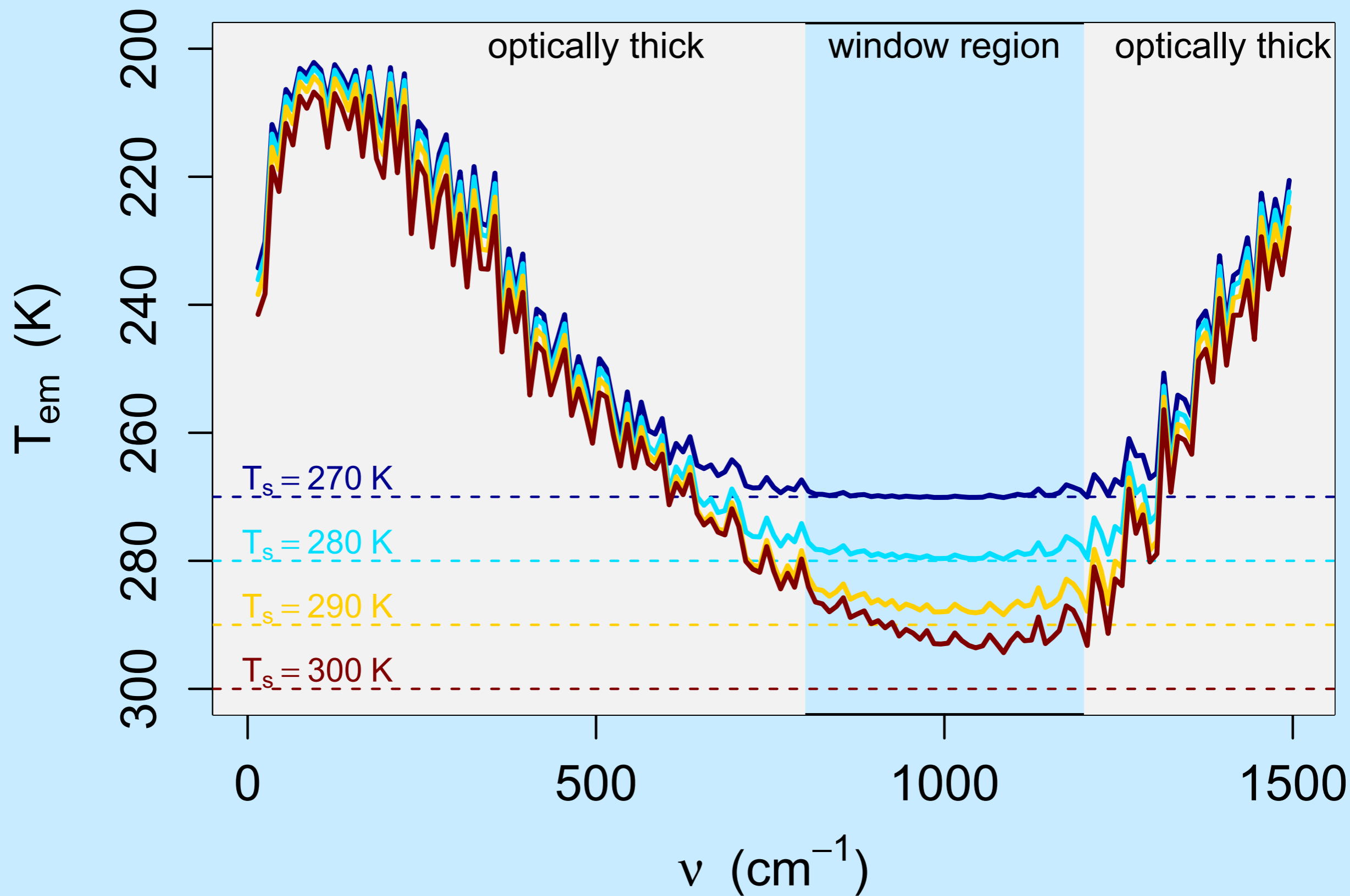
Water vapor path depends only on local temperature

Then

Water vapor optical depth is a function of temperature so

Cooling by water vapor is independent of temperature in optically thick regions

An explanation: Simpson's "law"



The atmosphere cools to space and warms from the surface

The monochromatic flux divergence across the whole atmosphere is

$$Q_\nu = \int_0^{\tau_\nu^*} \pi B_\nu(T(\tau_\nu)) e^{-\tau_\nu} d\tau_\nu - \int_0^{\tau_\nu^*} \pi (B_\nu(T_s) - B_\nu(T(\tau_\nu))) e^{-(\tau_\nu^* - \tau_\nu)} d\tau_\nu$$

i.e. by **cooling to space** and **heating from the surface** at each τ

[You can get here by using the solutions to Schwartchild's equation for up and down flux to compute net flux

$$Q_\nu = (F_\nu^-(0) - F_\nu^+(0)) - (F_\nu^-(\tau^*) - F_\nu^+(\tau^*))]$$

“Hydrologic sensitivity” is dominated by changes in cooling to space

The change in total flux divergence with surface temperature is

$$\frac{dQ_{\nu}}{dT_s} = \frac{d}{dT_s} \int_0^{\tau_{\nu}^*} \pi B_{\nu}(T(\tau_{\nu})) e^{-\tau_{\nu}} d\tau_{\nu} - \frac{d}{dT_s} \int_0^{\tau_{\nu}^*} \pi (B_{\nu}(T_s) - B_{\nu}(T(\tau_{\nu}))) e^{-(\tau_{\nu}^* - \tau_{\nu})} d\tau_{\nu}$$

because surface exchanges are small and get smaller at higher T_s

$$\approx \pi B_{\nu}(T(\tau_{\nu})) e^{-\tau_{\nu}} \frac{d\tau_{\nu}}{dT_s} + \pi \int_0^{\tau_{\nu}^*} \frac{d}{dT_s} B_{\nu}(T(\tau_{\nu})) e^{-\tau_{\nu}} d\tau_{\nu}$$

because cooling is dominated by water vapor and “Simpson’s law”

$$\approx \pi B_{\nu}(T(\tau_{\nu})) \frac{d\epsilon}{dT_s}$$

Remember emissivity is bounded by $0 \leq \epsilon \leq 1$

... in spectral regions where opacity changes with temperature

Emission from the atmosphere is dominated by water vapor and carbon dioxide

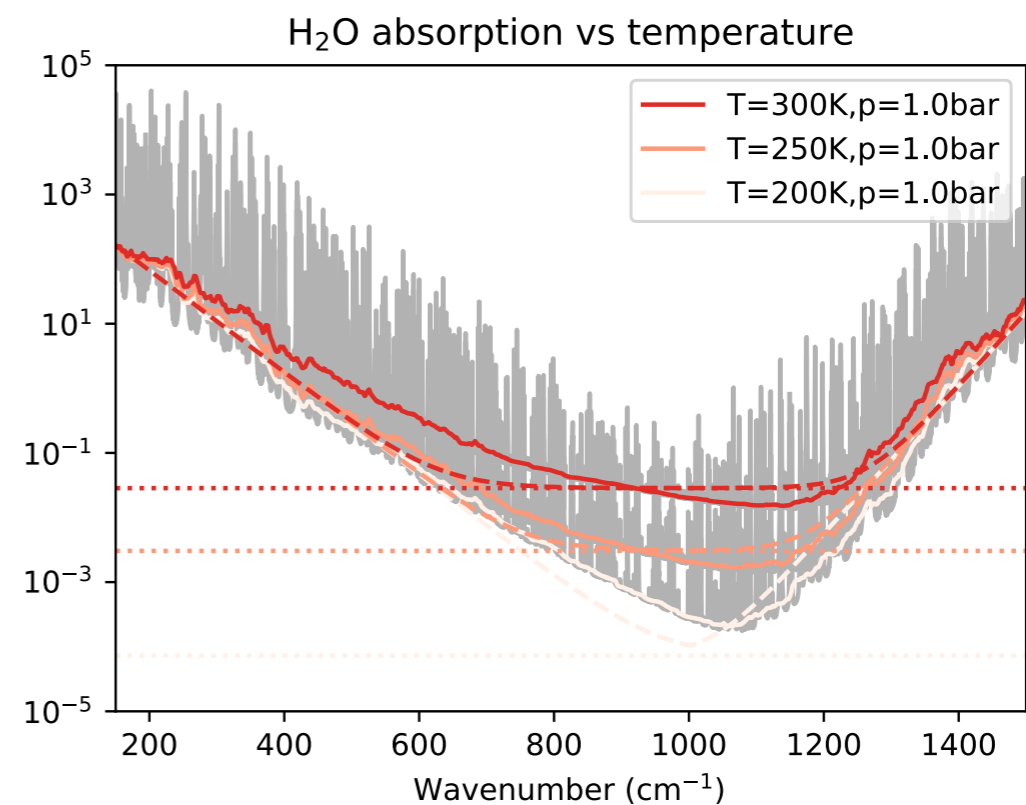
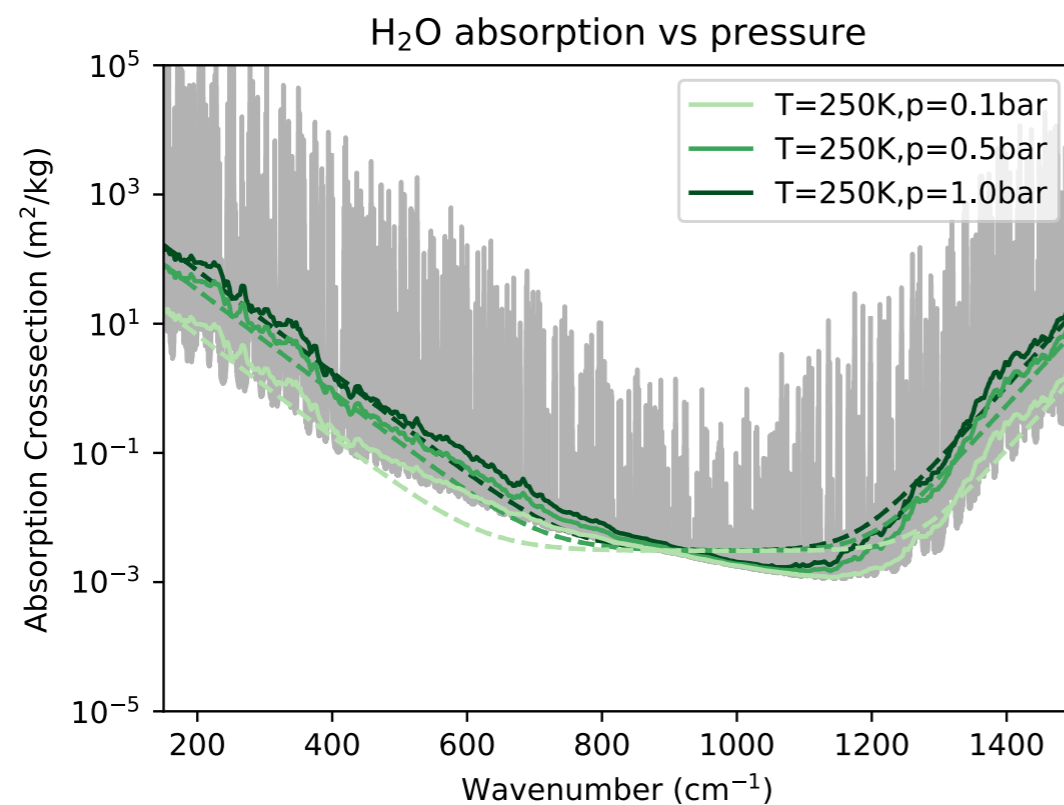
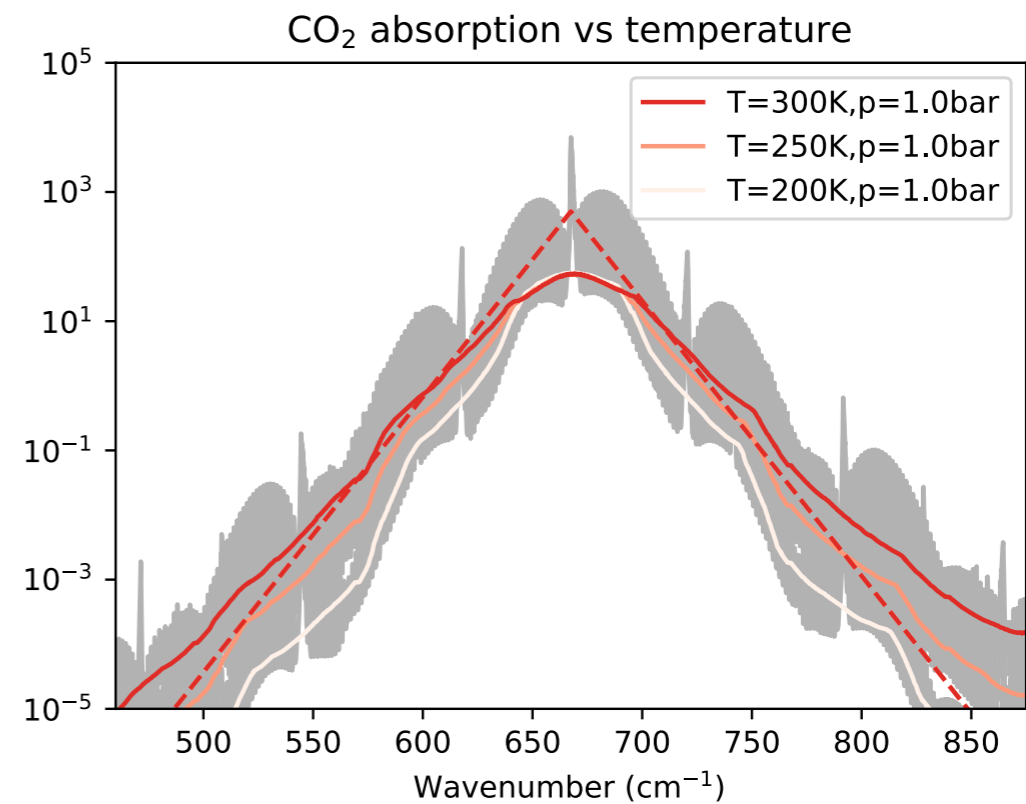
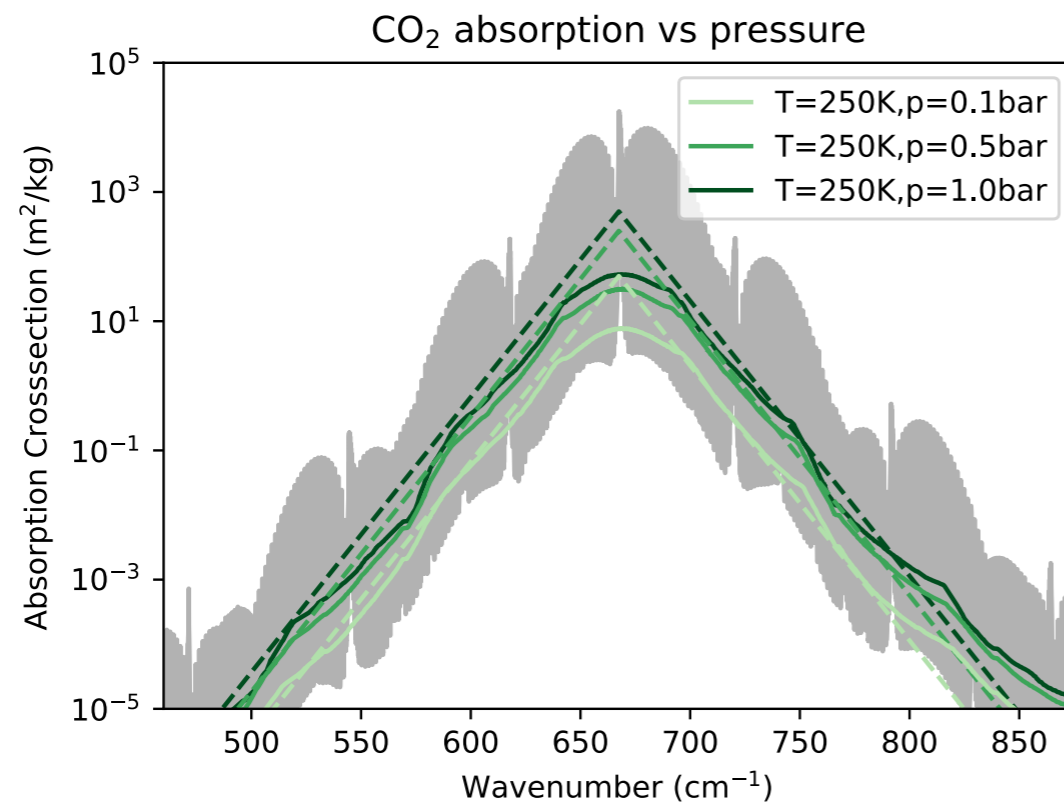
$$\tau_\nu = \tau_{\nu,\text{CO}_2} + \tau_{\nu,\text{H}_2\text{O}} = \tau_{\nu,\text{CO}_2} + Dk_{\nu,\text{H}_2\text{O}}WVP$$

neglecting **pressure broadening** and **self-continuum**

So that

$$\frac{dQ_\nu}{dT_s} = e^{-\tau_{\nu,\text{CO}_2}} \left(\pi B_\nu(T_s) \frac{d \ln WVP}{dT_s} \tau_{\nu,\text{H}_2\text{O}} e^{-\tau_{\nu,\text{H}_2\text{O}}} \right)$$

Complicated spectroscopy can be usefully idealized



... in spectral regions where opacity changes with temperature

Emission from the atmosphere is dominated by water vapor and carbon dioxide

$$\tau_\nu = \tau_{\nu,\text{CO}_2} + \tau_{\nu,\text{H}_2\text{O}} = \tau_{\nu,\text{CO}_2} + Dk_{\nu,\text{H}_2\text{O}}WVP$$

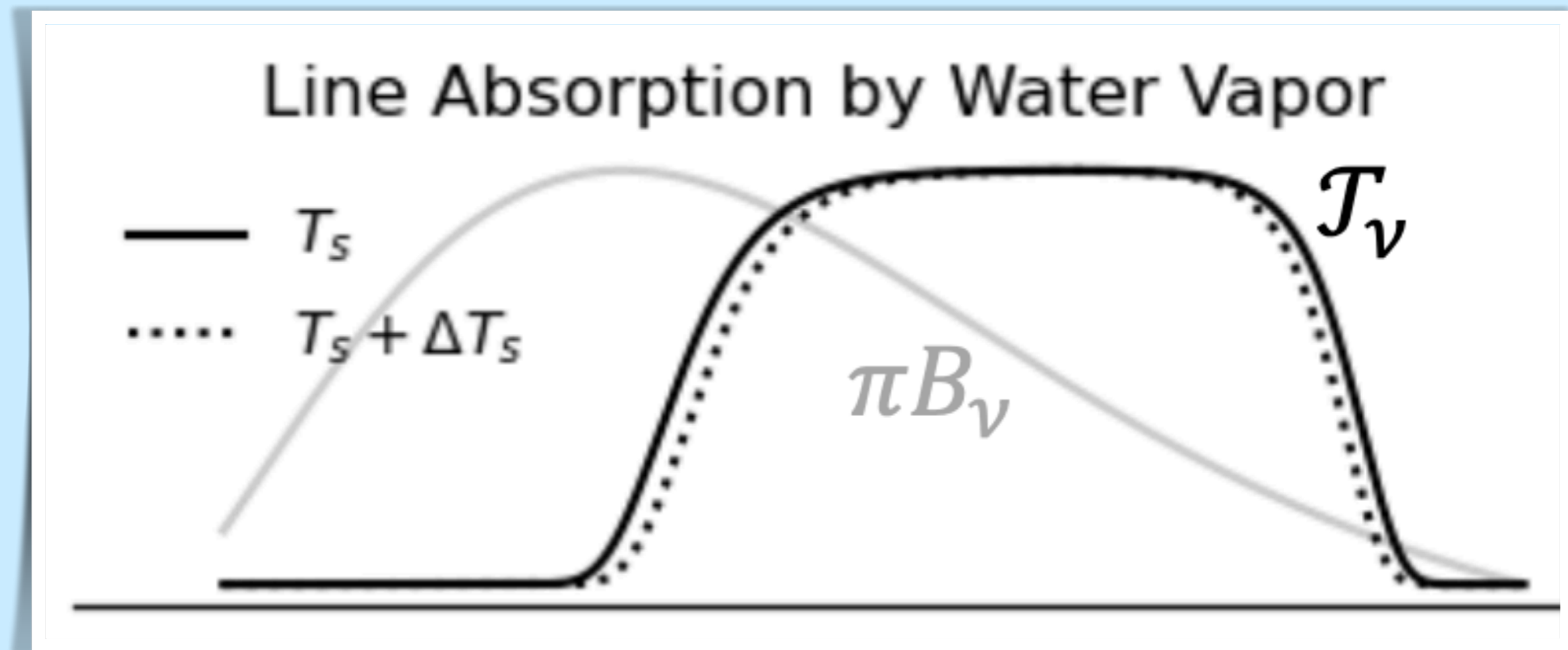
neglecting **pressure broadening** and **self-continuum**

So that

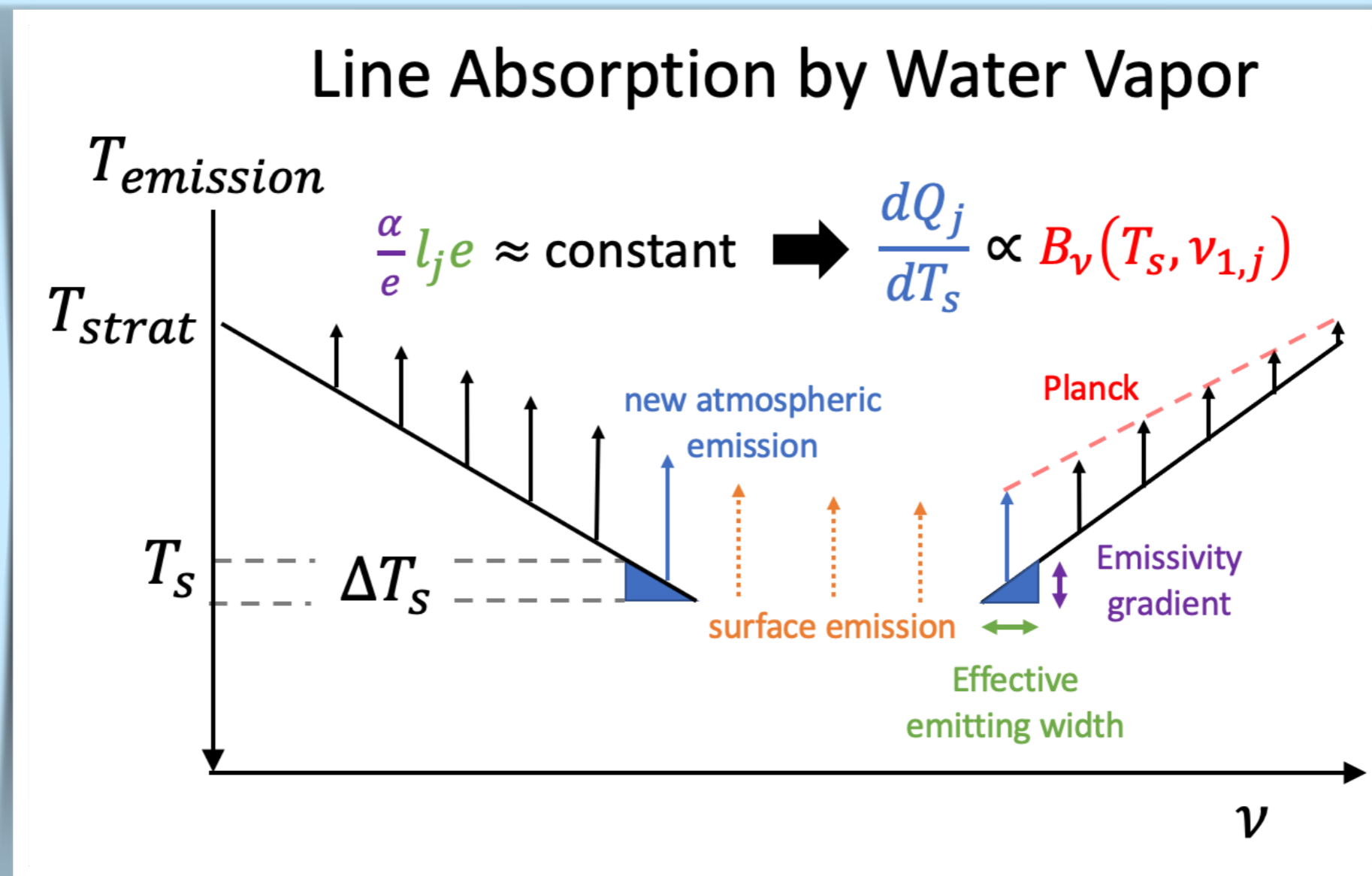
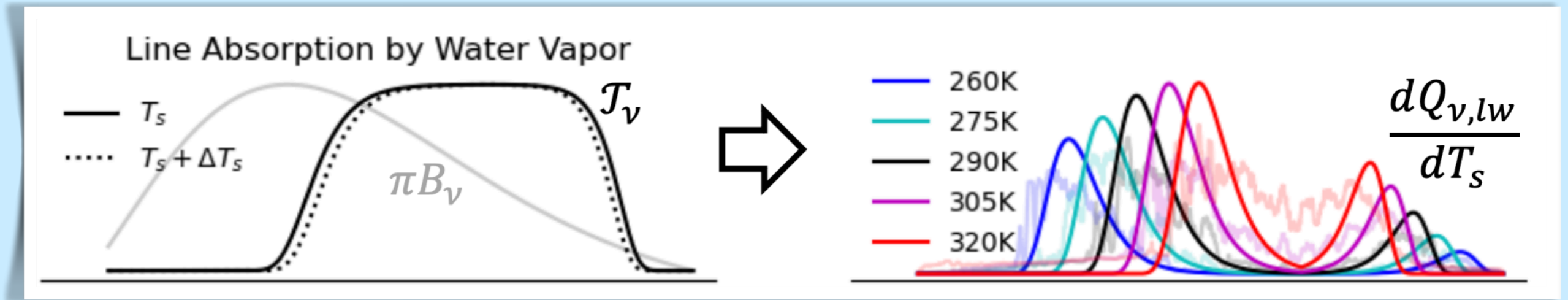
$$\frac{dQ_\nu}{dT_s} = e^{-\tau_{\nu,\text{CO}_2}} \left(\pi B_\nu(T_s) \frac{d \ln WVP}{dT_s} \tau_{\nu,\text{H}_2\text{O}} e^{-\tau_{\nu,\text{H}_2\text{O}}} \right)$$

Each source of absorption - water vapor lines, water vapor continuum, carbon dioxide - **adds interesting physics**

Hydrologic sensitivity results from the closing water vapor window

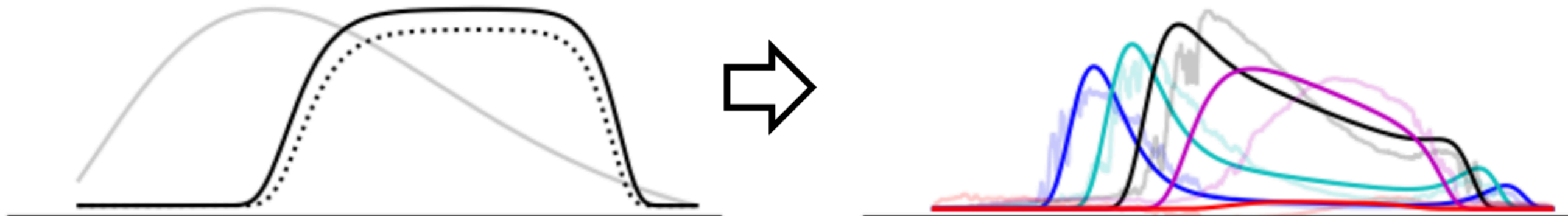


Hydrologic sensitivity results from the closing water vapor window

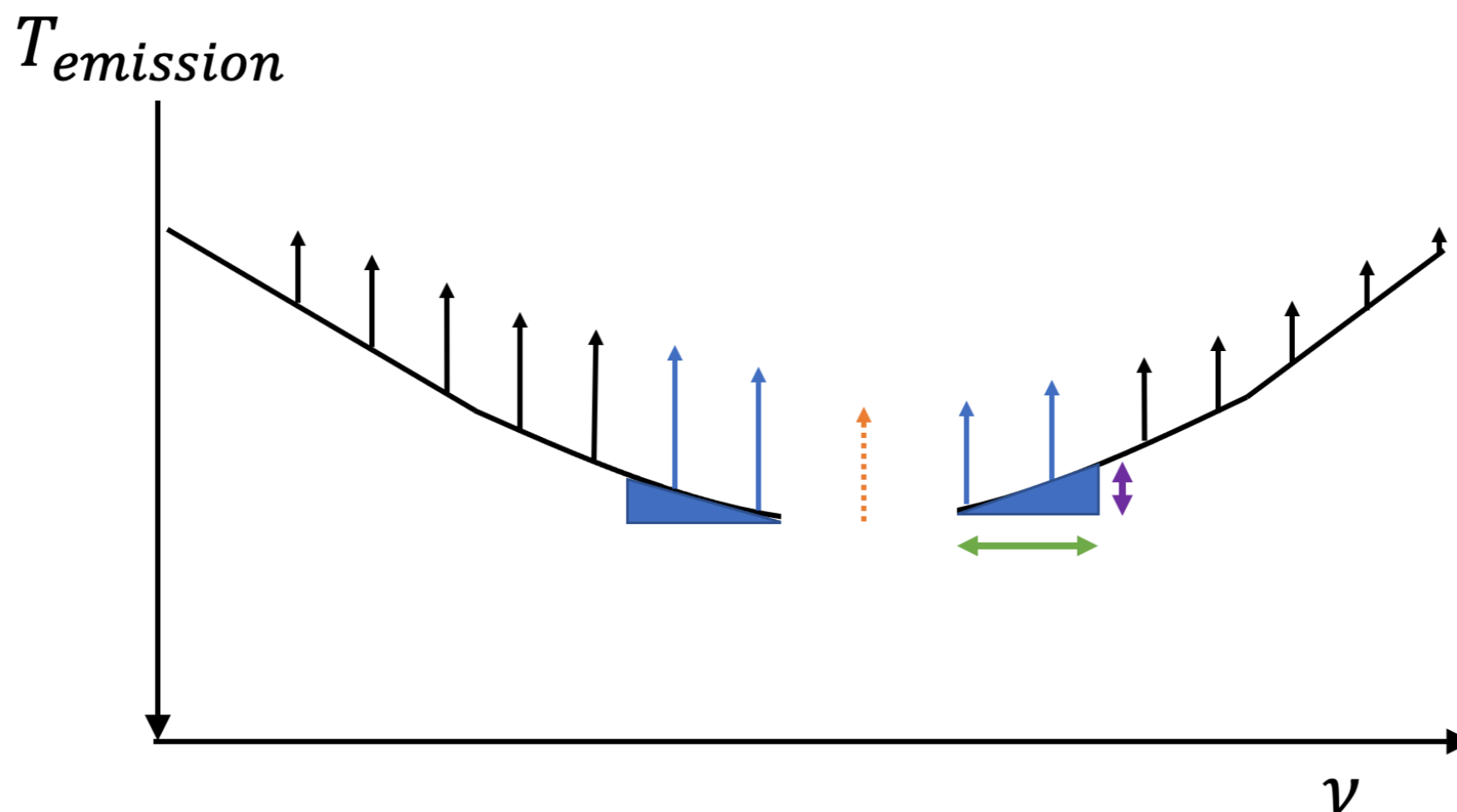


The window darkens as it closes

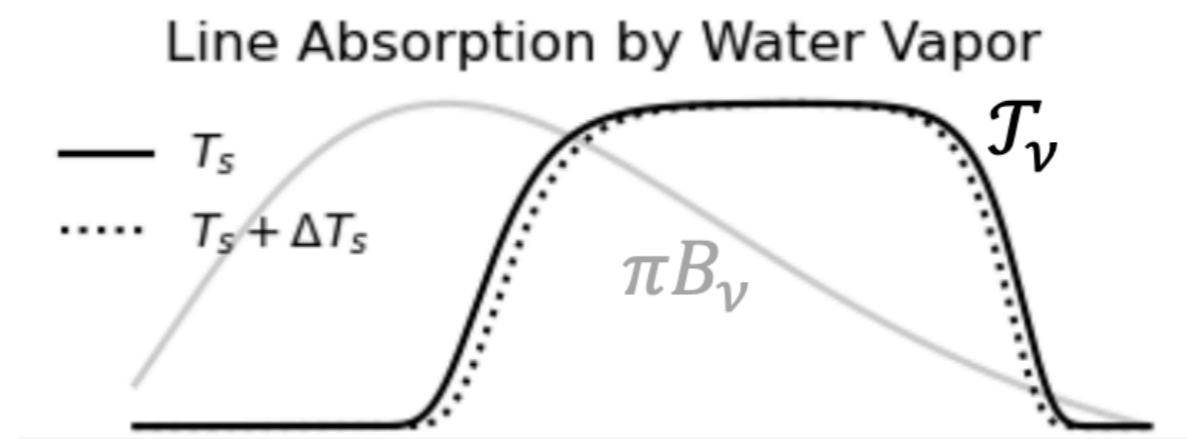
Water Vapor Atmosphere With Continuum



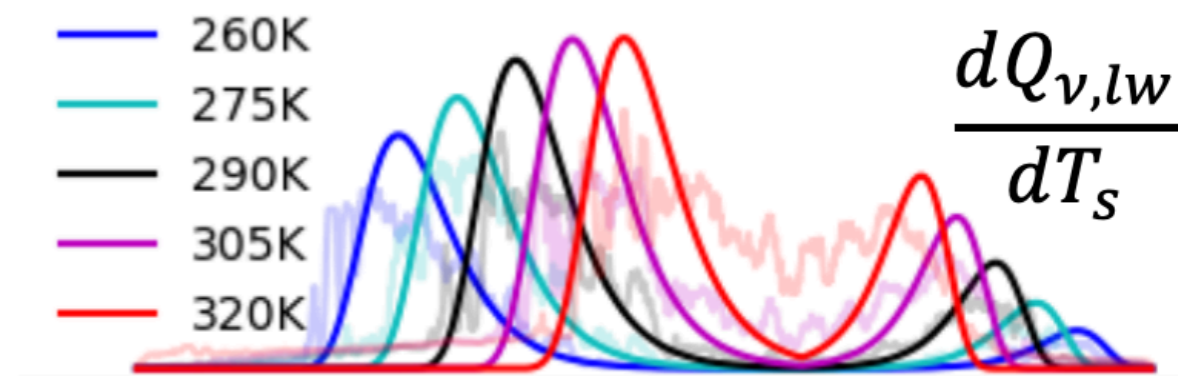
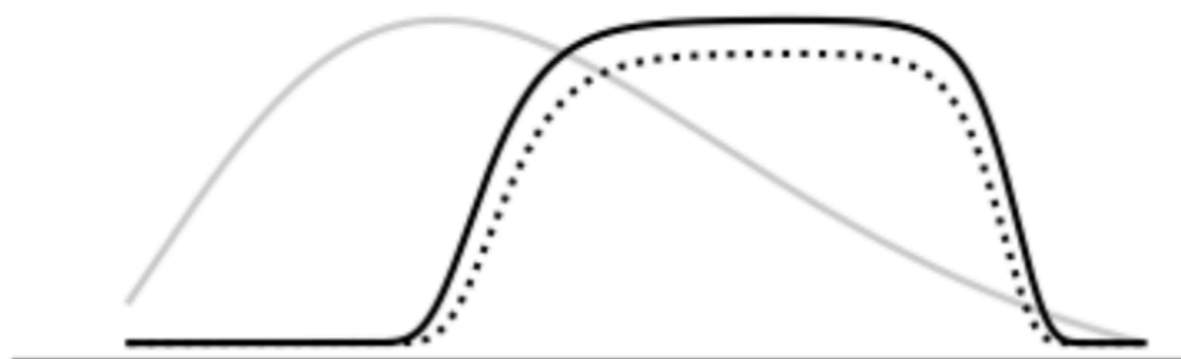
Total Absorption by Water Vapor



The window darkens as it closes



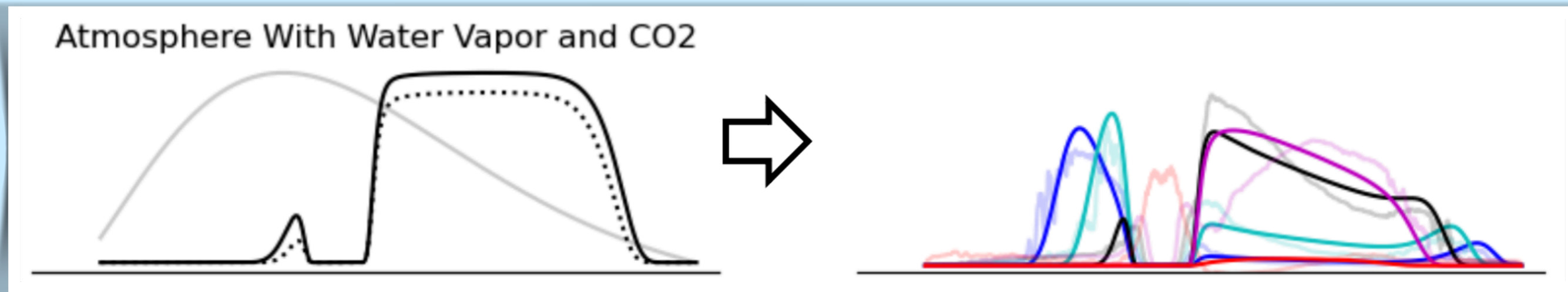
Water Vapor Atmosphere With Continuum



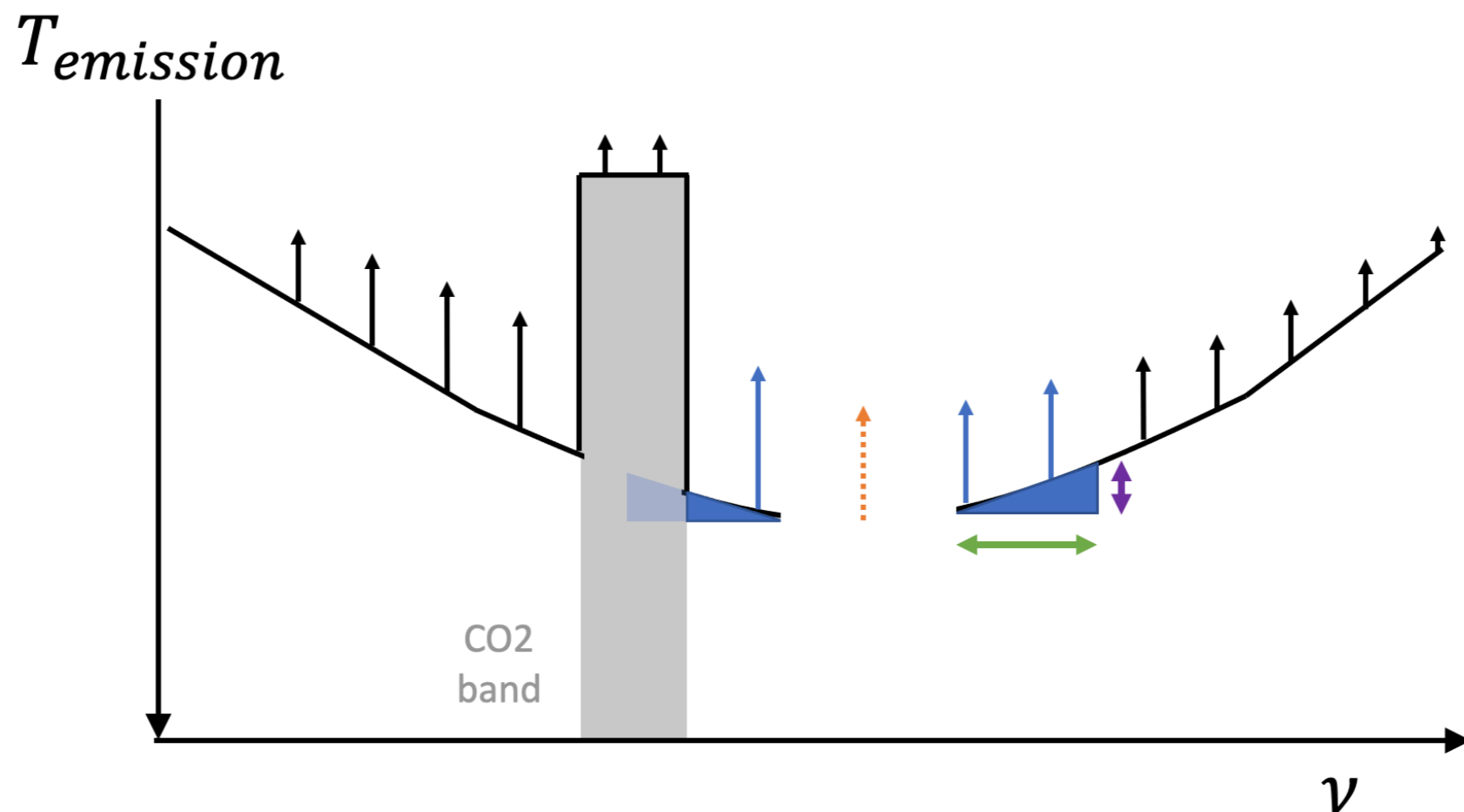
The window darkens from the edges, then the whole window goes opaque at once, introducing stronger temperature dependence

Once the whole window is opaque it's out of play

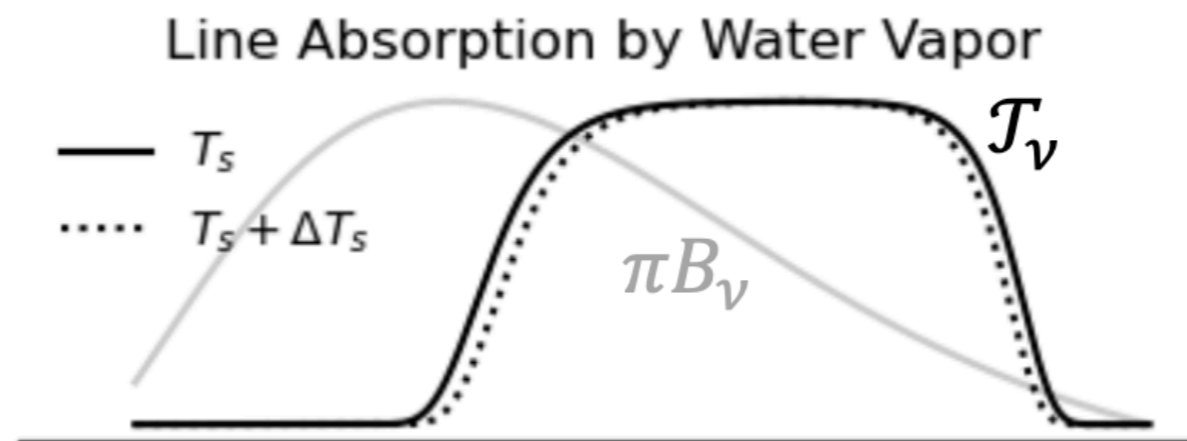
Carbon dioxide masks, with another temperature dependence



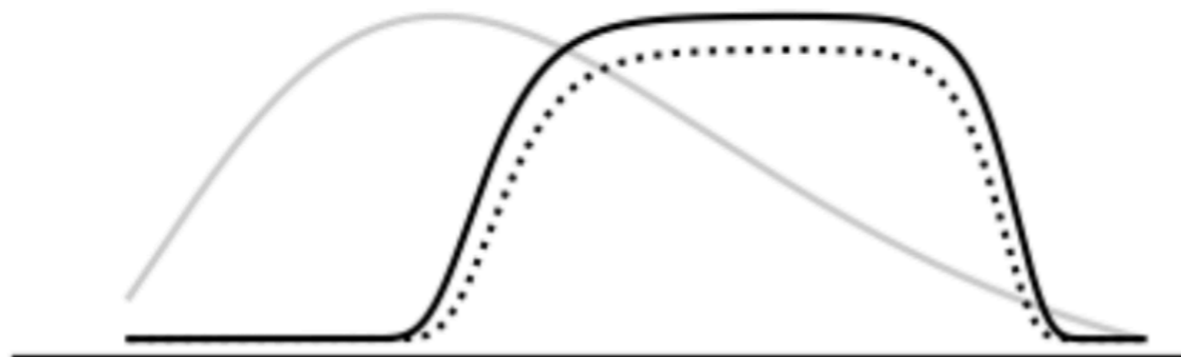
Absorption by Water Vapor and CO2



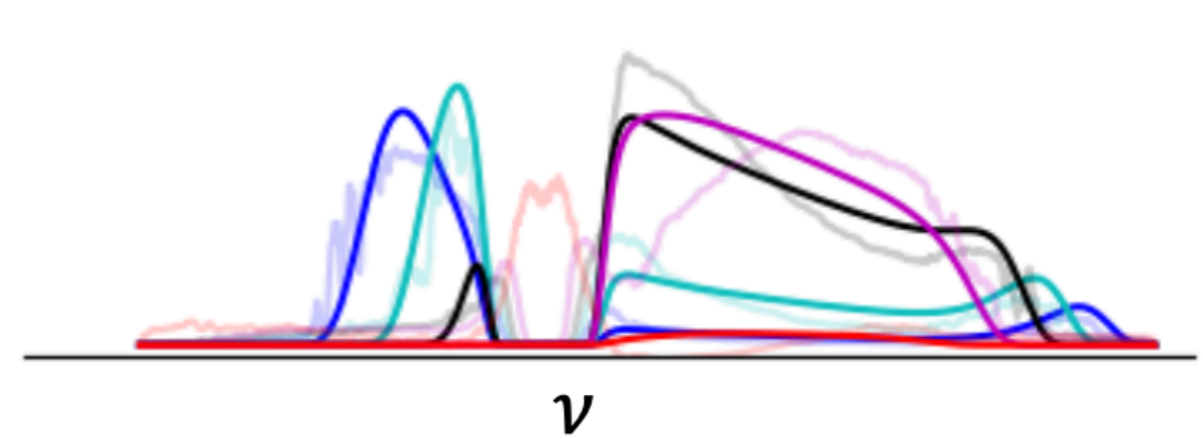
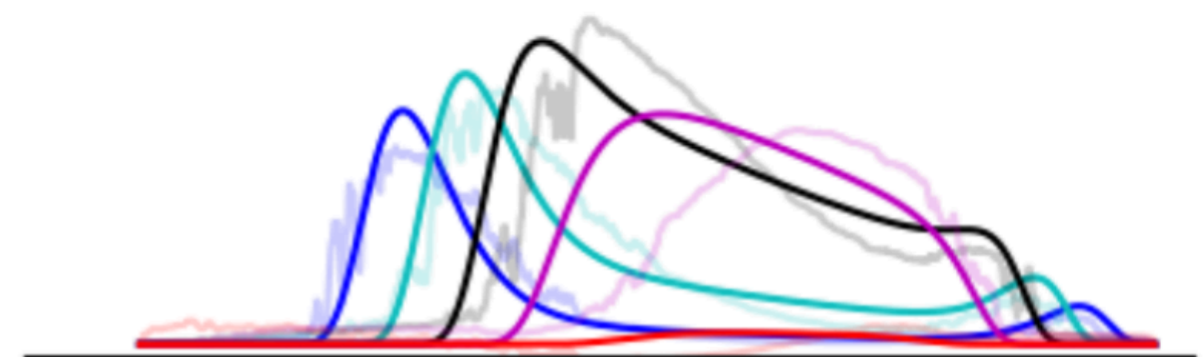
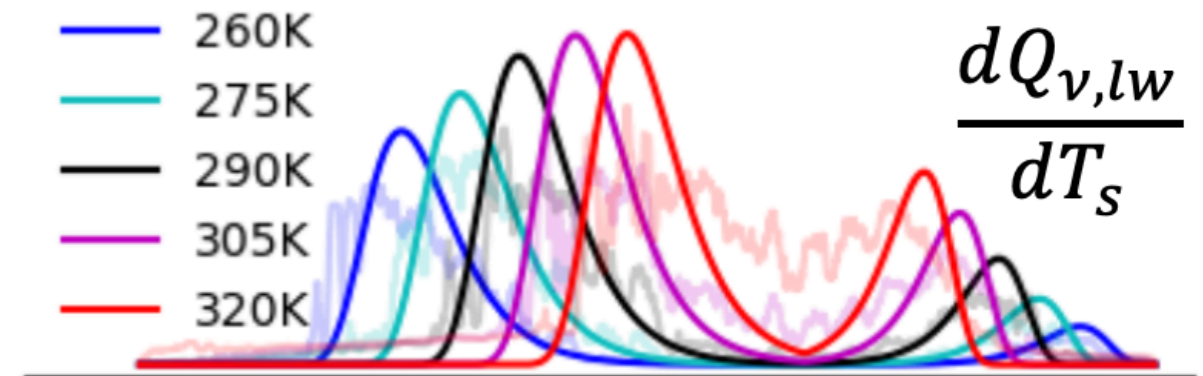
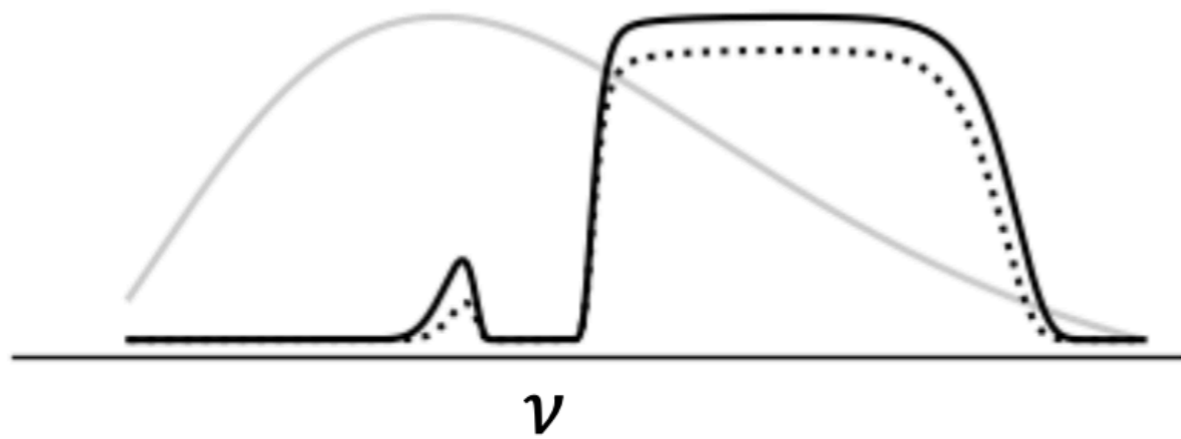
Carbon dioxide masks, with another temperature dependence



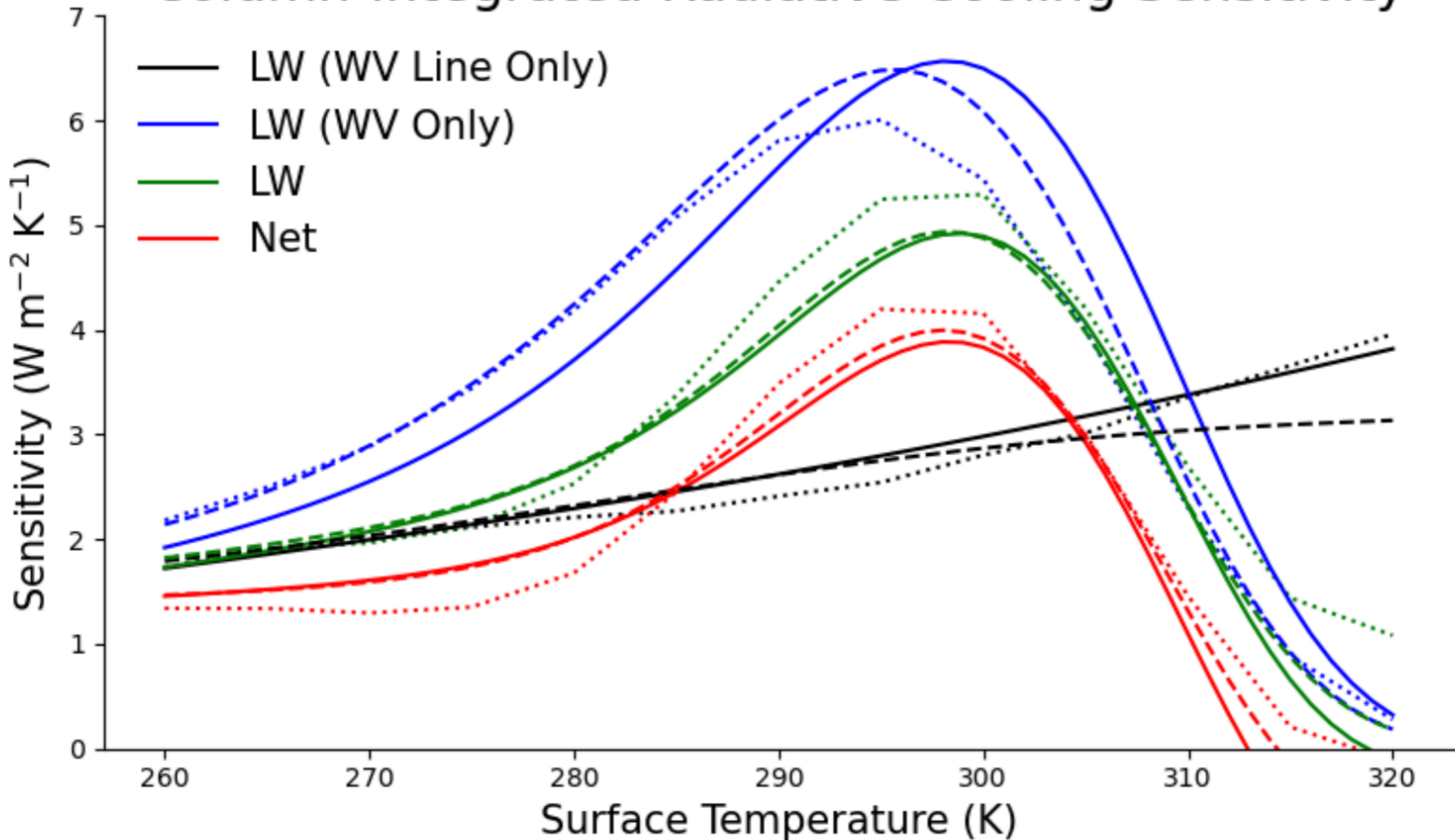
Water Vapor Atmosphere With Continuum



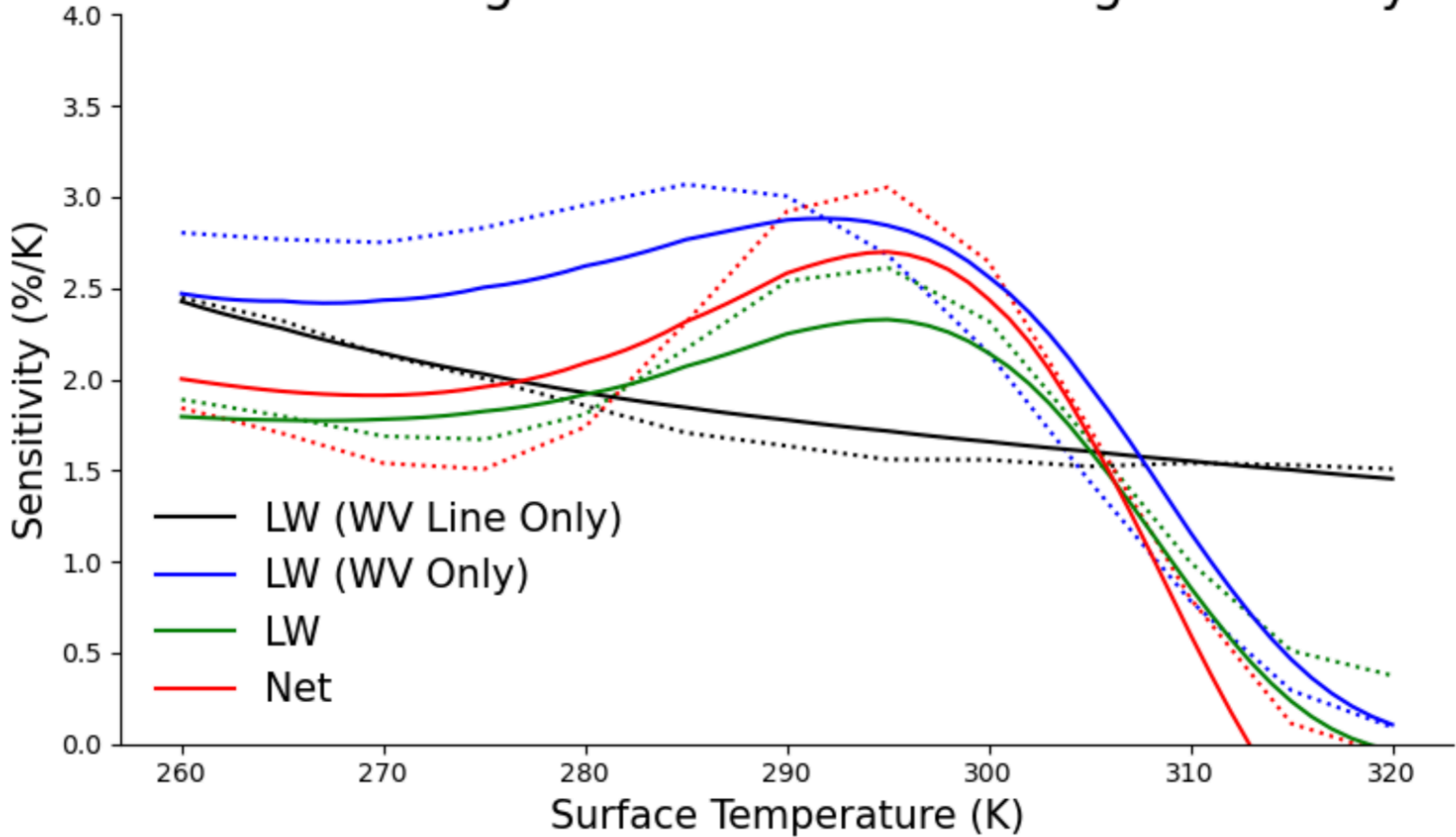
Atmosphere With Water Vapor and CO2



Column-Integrated Radiative Cooling Sensitivity



Column-Integrated Radiative Cooling Sensitivity



Hydrologic sensitivity is a consequence of spectroscopy

The atmosphere rains more with surface temperature not because it warms but because it warms and moistens simultaneously

The scale of hydrologic sensitivity is **set** by the surface Planck function and **shaped** by the spectroscopy of water vapor and carbon dioxide

Hydrologic sensitivity peaks at $\sim 298\text{K}$ and drops off quickly at higher temperatures

From theory to models

Shortwave heating also increases with warming, damping sensitivity

Changes in sensible heat fluxes modify hydrological sensitivity

Partly addressed by considering cooling in the free atmosphere

Clouds can mask changes in atmospheric emission and/or lower the apparent surface temperature

CO₂ forcing damps these already-small estimates

It's unclear how to integrate across varying temperatures

... but hydrologic sensitivity atmospheric component of the climate feedback and models can be understood in the same framework