

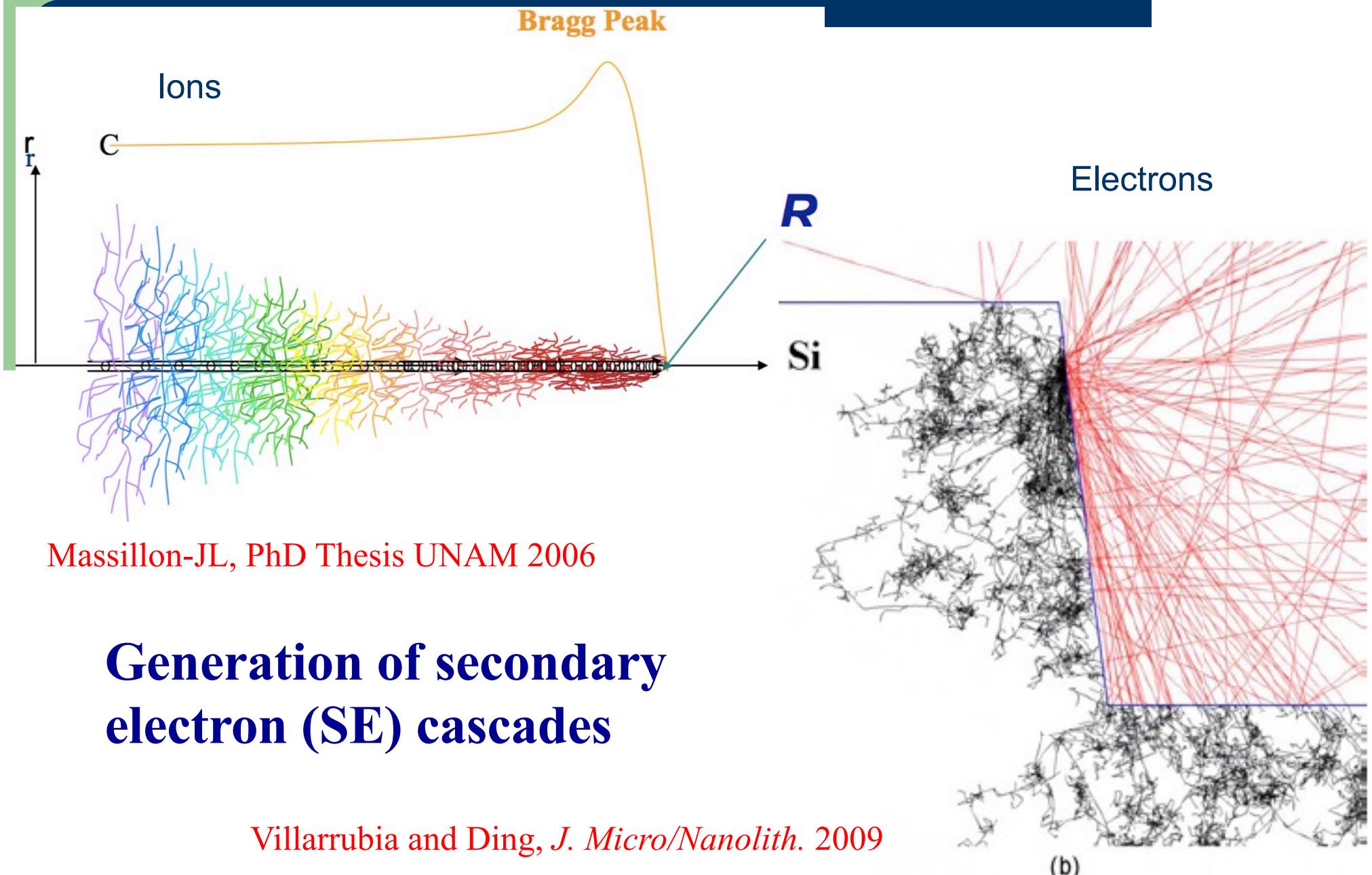
Dosimetric materials: what do we need to know?



Guerda Massillon-JL
massillon@fisica.unam.mx

Instituto de Física, UNAM, México

How can we know the absorbed dose?



Millions of questions ?

Part I

How do these electrons interact with matter?

How many are produced during the interaction?

What is their range?



Irradiated mass?

Part II

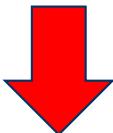
Where are the electrons localized?

Is all the energy transformed into a response?

Part I: Cross sections

Electron beam–solid-state interaction model

Interaction between the charge and spin of an incident particle and those of atomic electrons



Relativistic differential cross section (DCS) for inelastic scattering from quantum theory has two components:

1. Longitudinal excitation: The Coulomb interaction exerts a force parallel to the momentum transfer, q .
2. Transverse excitation: Interaction through virtual photons that are perpendicular to q (negligible at $T < 500$ keV)

Part I: Cross-sections

DCS for the longitudinal excitation in terms of the momentum transfer, q

$$\frac{d^2\sigma_L}{d\omega dq} = \frac{2}{\pi N v^2} \text{Im} \left[\frac{-1}{\epsilon(q,\omega)} \right] \frac{1}{q}, \quad \lambda^{-1} = N\sigma$$

$$\frac{d^2\lambda^{-1}}{d\omega dq} = \frac{2}{\pi v^2} \text{Im} \left[\frac{-1}{\epsilon(q,\omega)} \right] \frac{1}{q}$$

N : number density of scatterers

ω and q : energy and momentum transfer

v : incident electron's speed

$\epsilon(q, \omega)$: momentum and energy-dependent dielectric function

$\text{Im}[-1/\epsilon(q, \omega)]$: energy loss function (ELF)

U Fano, *Ann. Rev. Nucl. Sci.* 13 1 (1963)

Part I: Cross-sections

The inelastic mean free path (IMFP), λ for an electron can be expressed as

$$\lambda^{-1} = \frac{(1 + T'/c^2)^2}{1 + T'/2c^2} \frac{1}{\pi T'} \int_{\omega_{\min}}^{T' - w_{\text{VB}}} \int_{q^-}^{q^+} \text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right] \frac{dq}{q} d\omega$$

$$q^\pm = \sqrt{T'(2 + T'/c^2)} \pm \sqrt{(T' - \omega)(2 + (T' - \omega)/c^2)}$$

w_{VB} : width of the valence band

ω_{\min} : energy required to promote an electron from the top of the valence band to the bottom of the conduction band

Part I: Stopping power

The stopping power can be calculated from the probability for an electron with relativistic kinetic energy, T , to loss energy per unit distance travelled as:

$$SP = \frac{(1+T'/c^2)^2}{1+T'/2c^2} \frac{1}{\pi T'} \int_{\omega_{\min}}^{T' - w_{VB}} \int_{q_-}^{q_+} \text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right] \frac{dq}{q} \omega d\omega,$$

$$q_{\pm} = \sqrt{T'(2 + T'/c^2)} \pm \sqrt{(T' - \omega)(2 + (T' - \omega)/c^2)}$$

$$T' = T - E_g,$$

E_g : the bandgap energy

c : the speed of light,

w_{VB} : the valence band width

$T' - w_{VB}$: to assure that the incident electron preserves enough energy to stay in the conduction band

ω_{\min} : the minimum energy required to promote an electron from the top of the valence band to the bottom of the conduction band

Part I: Full Penn algorithm (NIST)

$$\text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right] = - \int_0^{\infty} d\omega_p G(\omega_p) \text{Im} \left[\frac{-1}{\epsilon_L(q, \omega; \omega_p)} \right],$$

$$G(\omega_p) = -\frac{2}{\pi \omega_p} \text{Im} \left[\frac{-1}{\epsilon(\omega_p)} \right]$$

Due to some limitation on the Lindhard energy loss function, the ELF can be portrayed as a combination of the plasmon pole and the single-electron excitations such as:

$$\text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right] = \text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right]_{pl} + \text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right]_{se}$$

Part I: Full Penn algorithm (NIST)

$$\text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right]_{pl} = G(\omega_0) \frac{\pi}{[\partial \epsilon_L(q, \omega; \omega_p) / \partial \omega_p]_{\omega_p=\omega_0}} \theta(q^-(\omega; \omega_0) - q),$$

$$q^\pm(\omega; \omega_0) = \pm \left[\omega_p \left(\frac{3\pi}{4} \right)^{1/2} \right]^{1/3} + \left[\omega_p^{\frac{2}{3}} \left(\frac{3\pi}{4} \right)^{\frac{1}{3}} + 2\omega \right]^{\frac{1}{2}}$$

$$\begin{aligned} & \text{Im} \left[\frac{-1}{\epsilon(q, \omega)} \right]_{se} \\ &= \int_0^\infty d\omega_p G(\omega_p) \text{Im} \left[\frac{-1}{\epsilon_L(q, \omega; \omega_p)} \right] \theta(q^+(\omega; \omega_p) - q) \theta(q - q^-(\omega; \omega_p)) \end{aligned}$$

JMONSEL vs Experiment

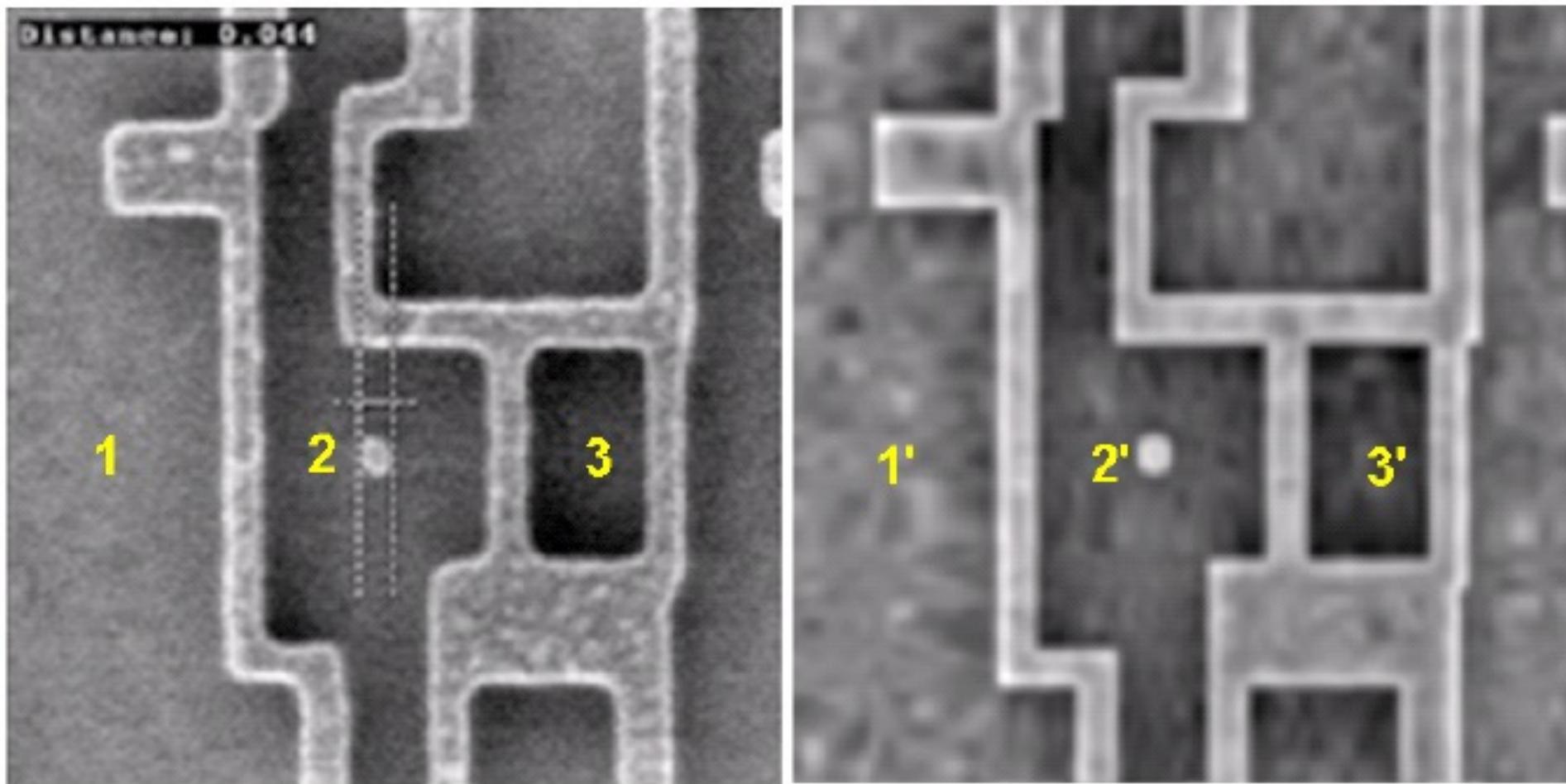


Figure 7. Comparison of measured (left, SEM image, courtesy SEMATECH) and simulated (right) images of an intentional defect array structure. Neighborhoods 1, 2, and 3 ($1'$, $2'$, and $3'$ in the simulation) are progressively more confined and also progressively darker. 1 μm fields of view (Figure reproduced from reference 10).

Consistency in the optical data

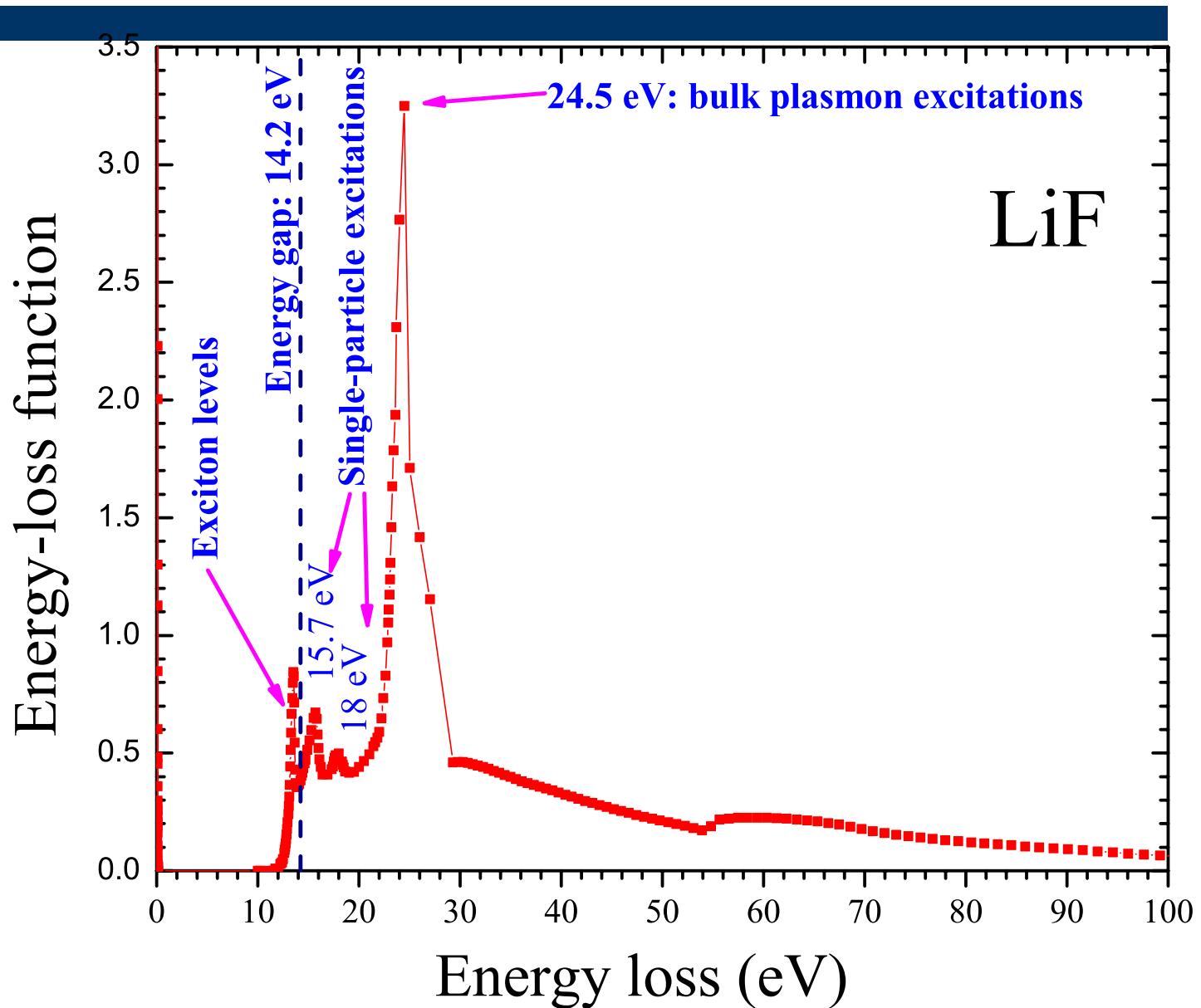
$$P_{\text{eff}} = \frac{2}{\pi} \int_0^{\omega_{\text{max}}} \frac{\text{Im} \left(-\frac{1}{\epsilon(\omega)} \right)}{\omega} d(\omega) + n(0)^{-2}$$

$$Z_{\text{eff}} = -\frac{2m\epsilon_0}{\pi Ne^2} \int_0^{\omega_{\text{max}}} \omega \text{Im} \left[\frac{1}{\epsilon(\omega)} \right] d(\omega),$$

Table 1. KK-sum and f-sum errors.

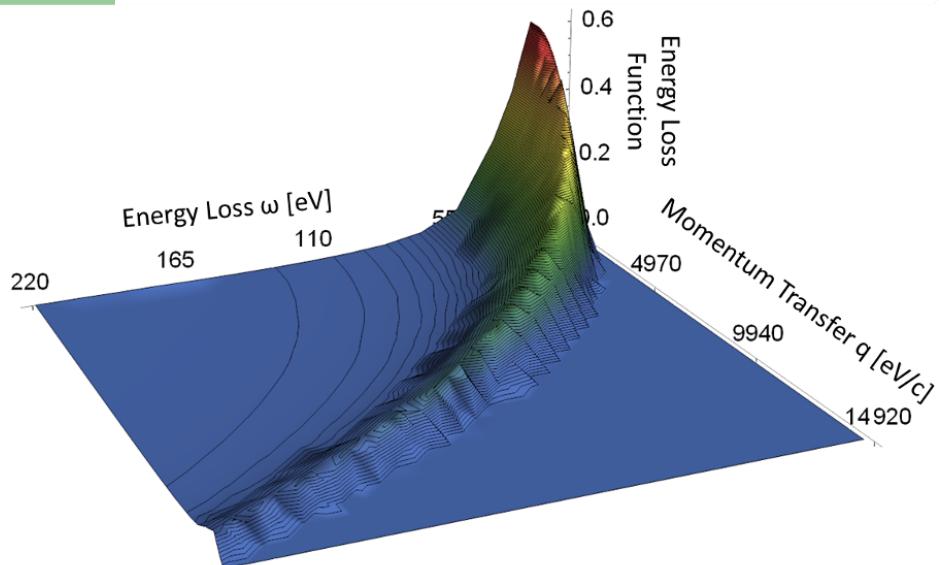
Compound	n(0)	Z	Z _{eff}	f-sum error (%)	P _{eff}	KK-sum error (%)
H ₂ O	8.97	10	10.021	0.21	1.027	2.7
LiF	3	12	13.16	9.69	1.132	13.2
CaF ₂	2.6	38	38.91	2.4	1.058	5.8
Al ₂ O ₃	3.13	50	51.58	3.16	1.038	3.8

Energy-loss function



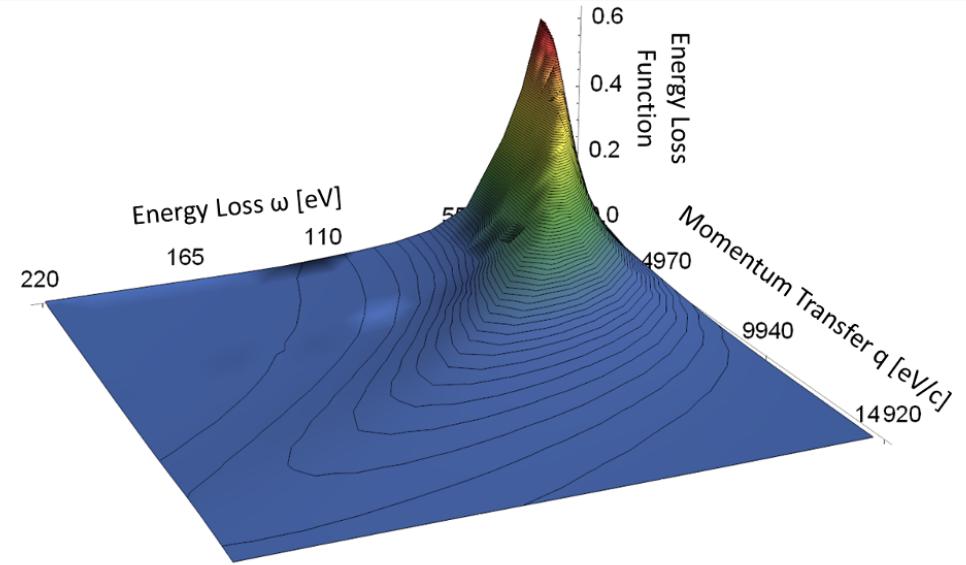
ELF vs momentum vs energy loss

SPA



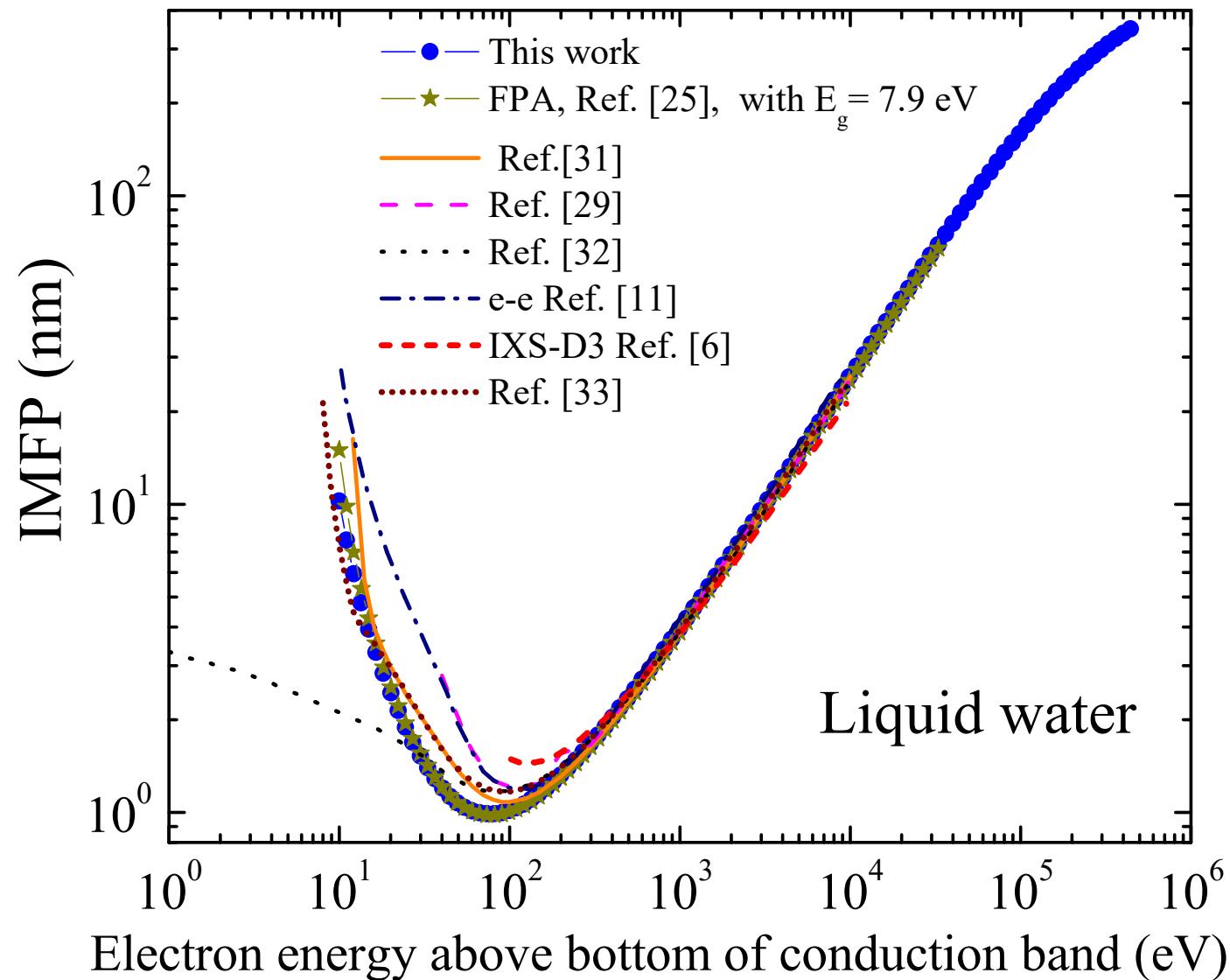
FPA

$q \neq 0$

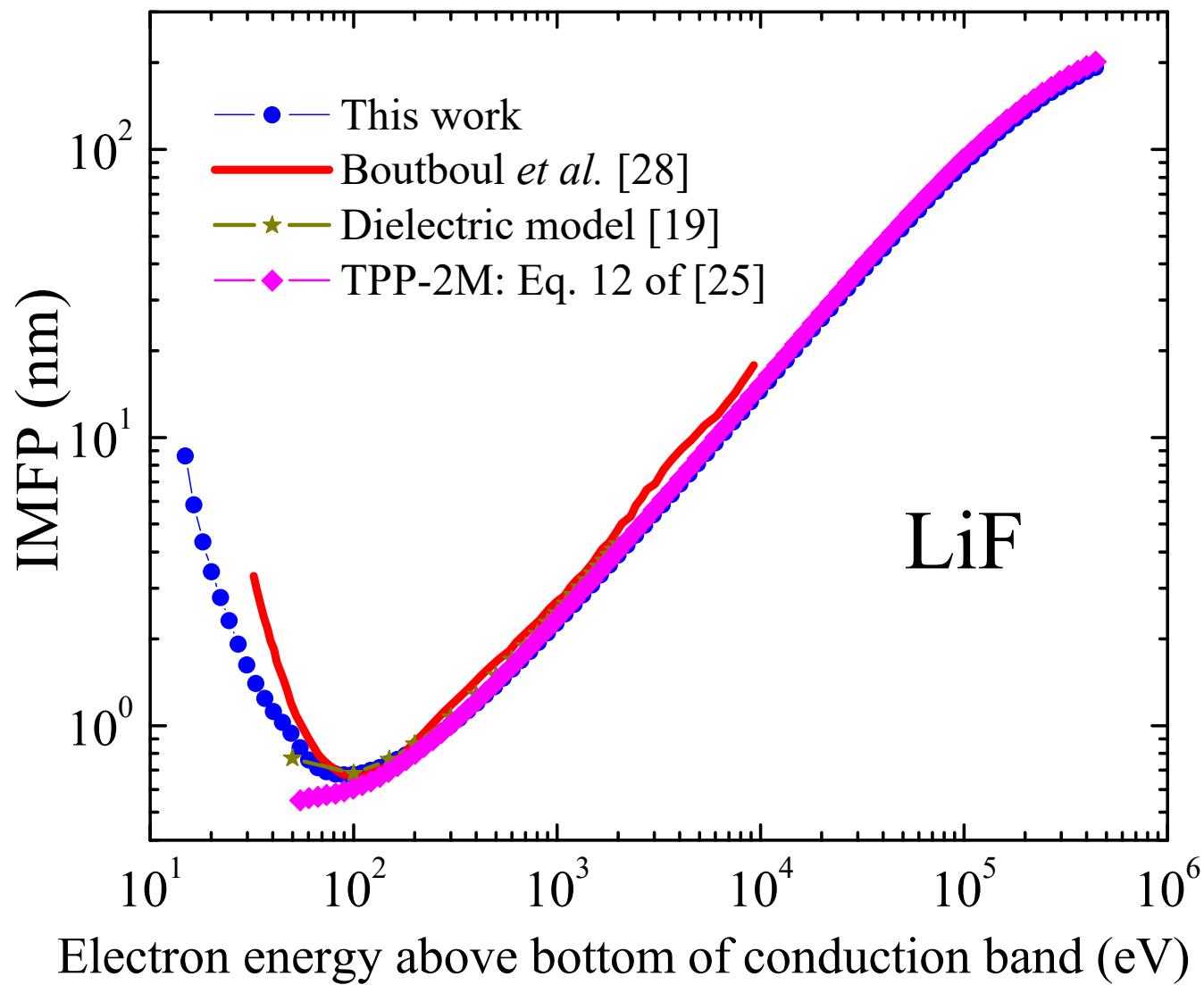


Perspective view of ELF as a function of momentum transfer and energy loss calculated by the FPA for liquid water

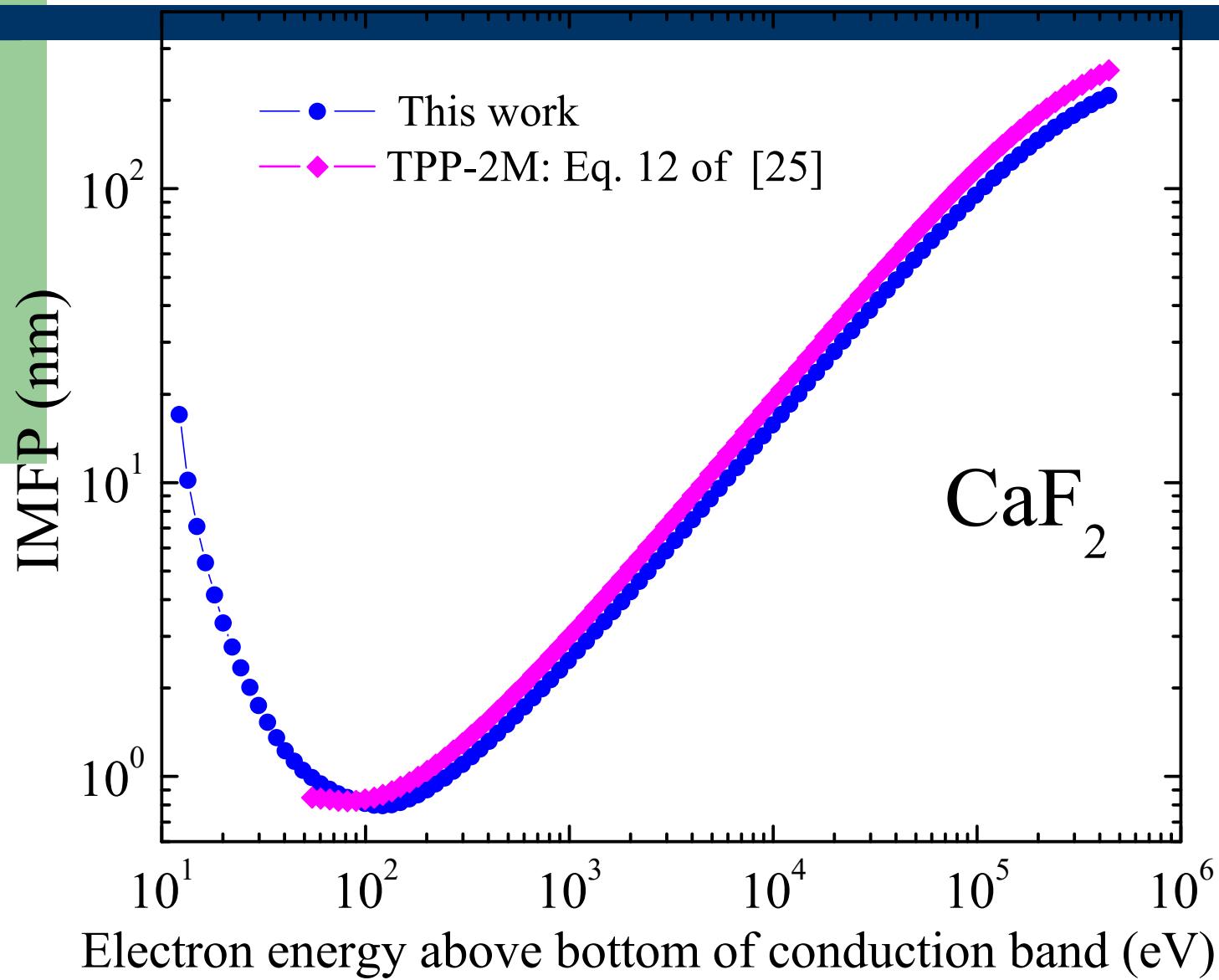
Electron mean free path



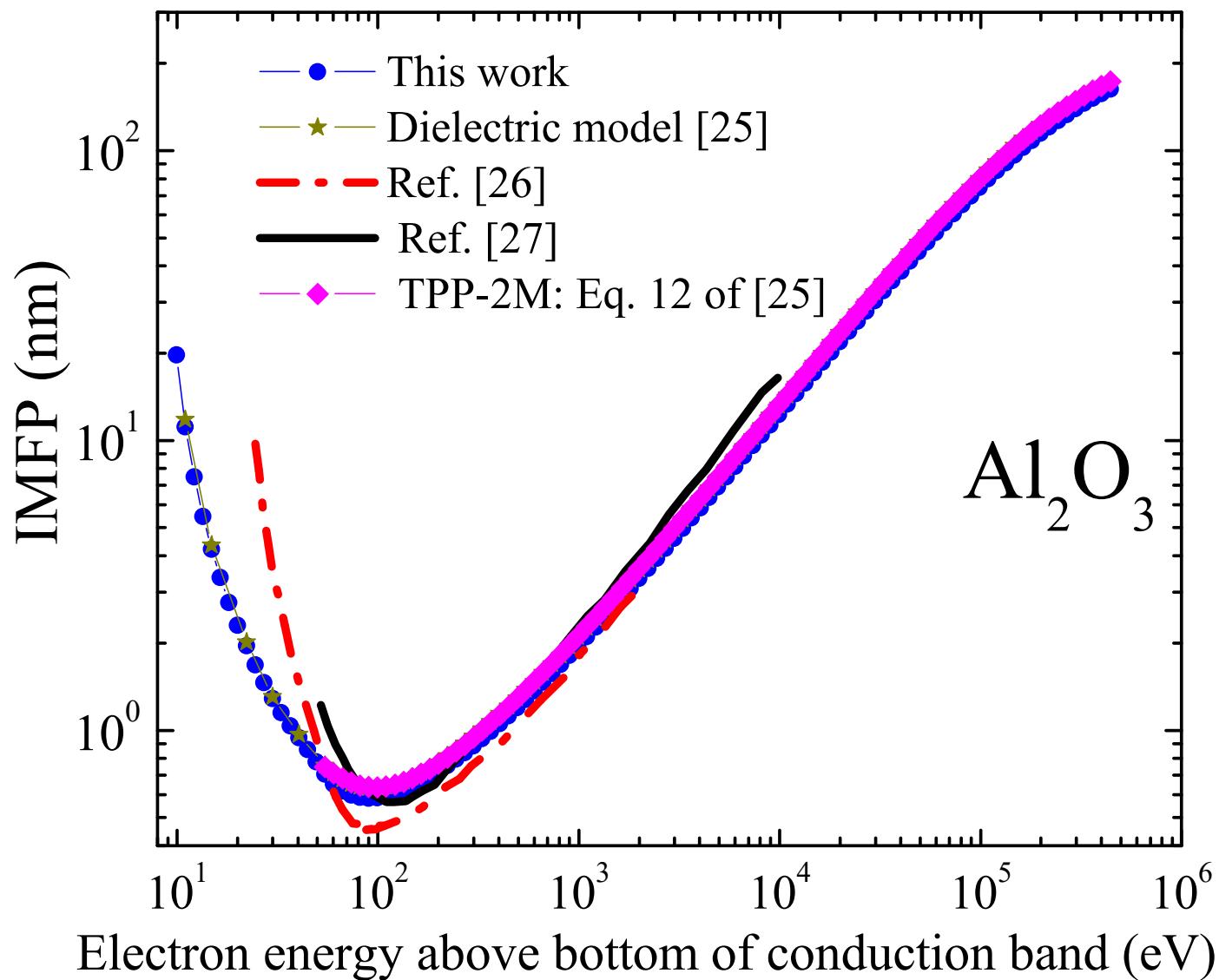
Electron mean free path



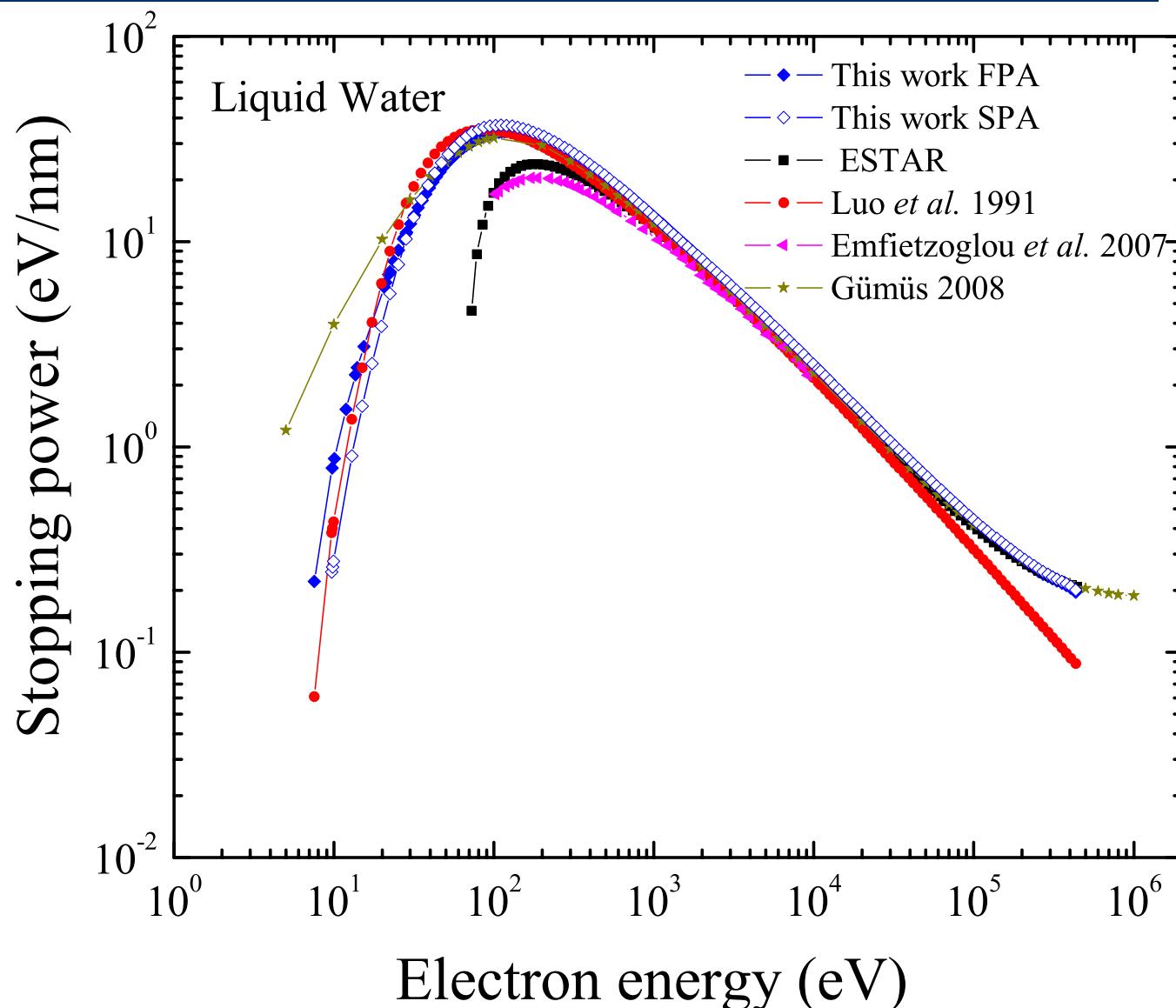
Electron mean free path



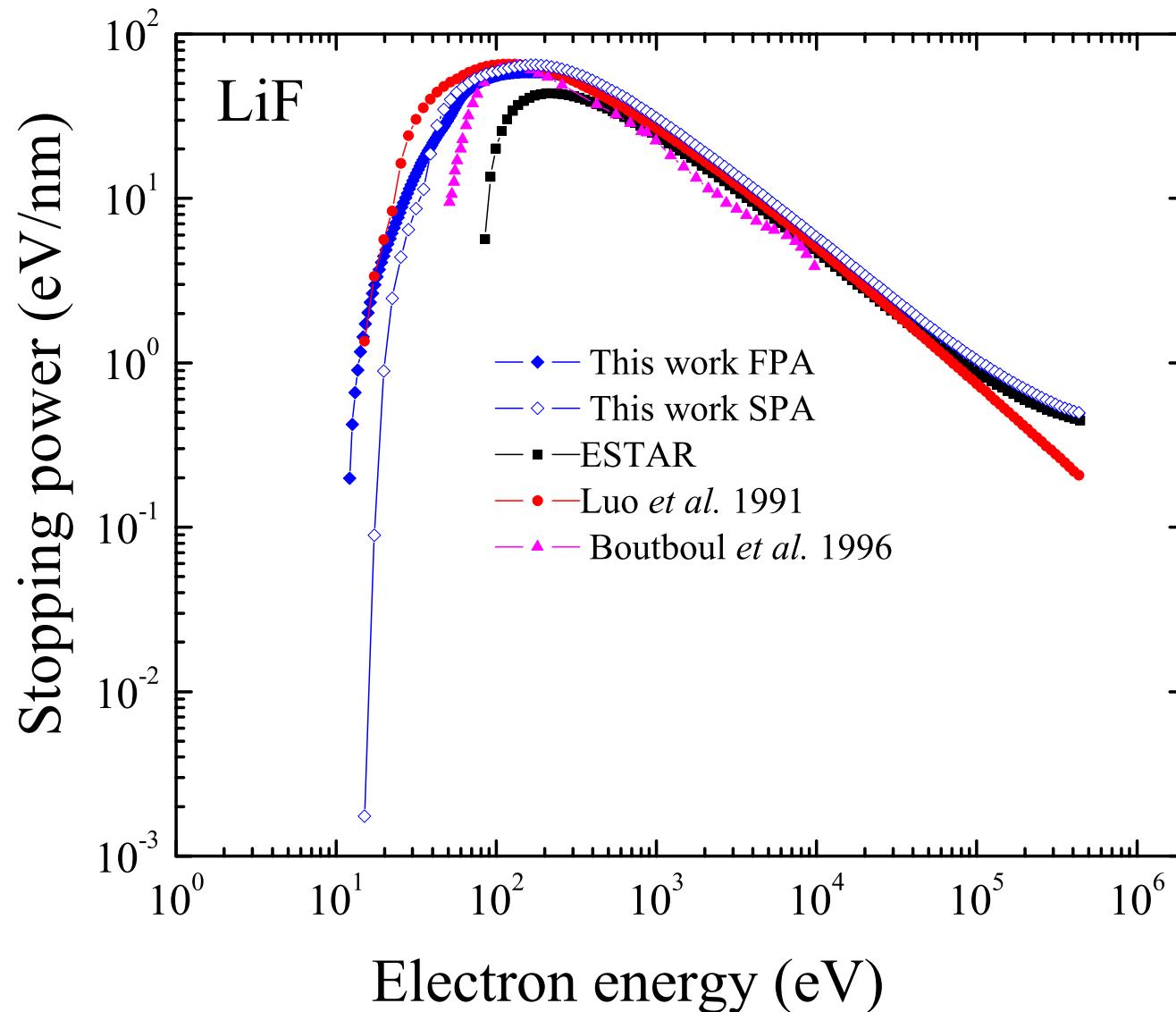
Electron mean free path



Electron Stopping power using linear response: The dielectric model



Electron Stopping power using linear response: The dielectric model



Part II: LiF:Mg,Ti?

During the last few years, density functional theory (DFT) has been considered as a suitable tool to study radiation effects and electronic structures of:

Ti, Mg-doped LiF

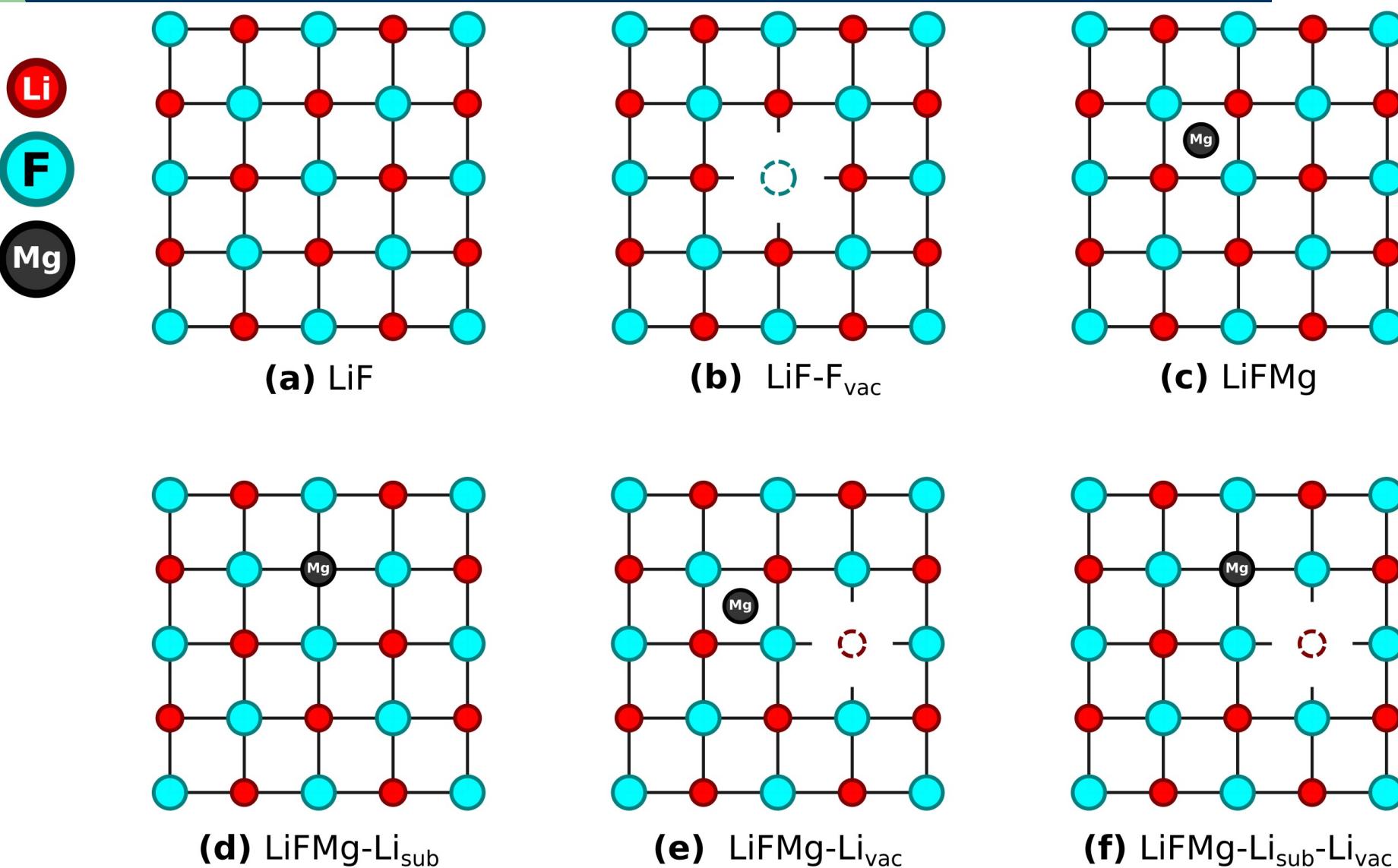


Massillon-JL et al. *J. Phys.:Condens. Matter* **31** (2019) 025502

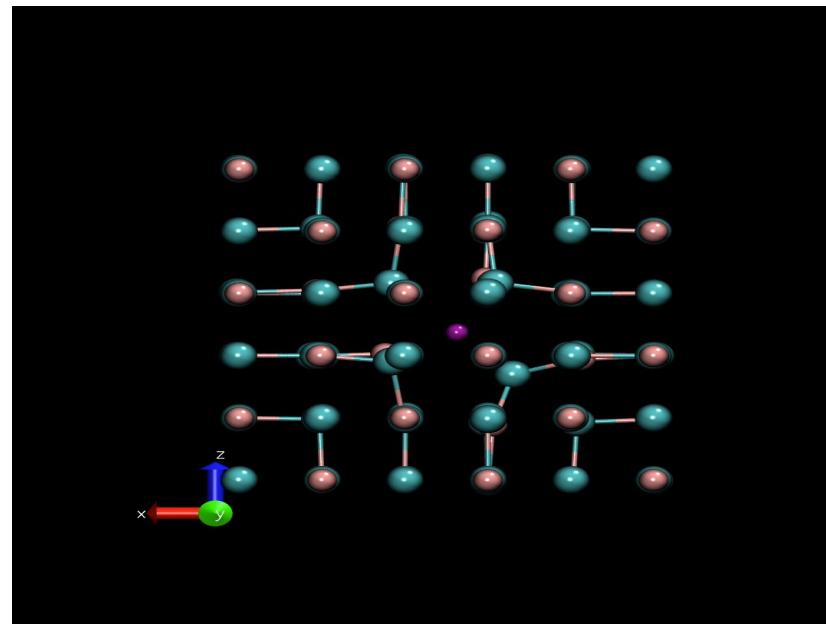
Massillon-JL et al. *Radiat. Meas.* **174** (2024) 107114

Modak and Modak,
Computational Materials Science **202** (2022) 110977

Simplified illustrative structures for the initial configurations

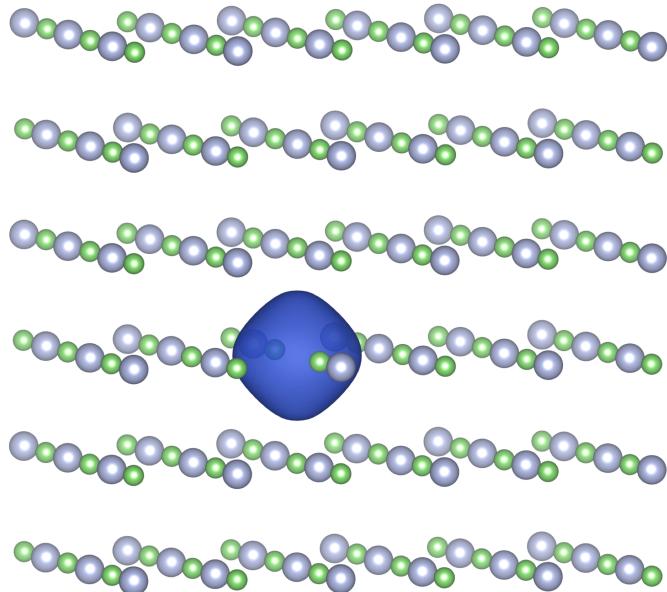


Molecular dynamic simulation of LiF:Mg

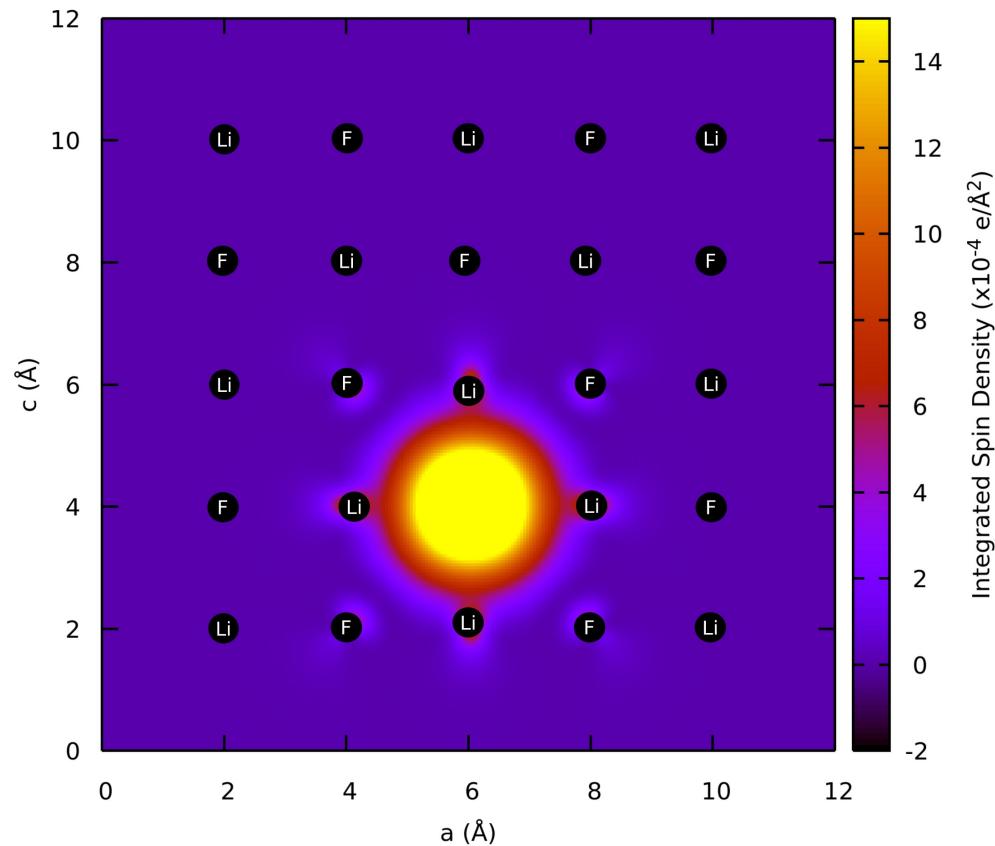


Fluorine vacancy in LiF

Spin density isosurface of the LiF:F vacancy

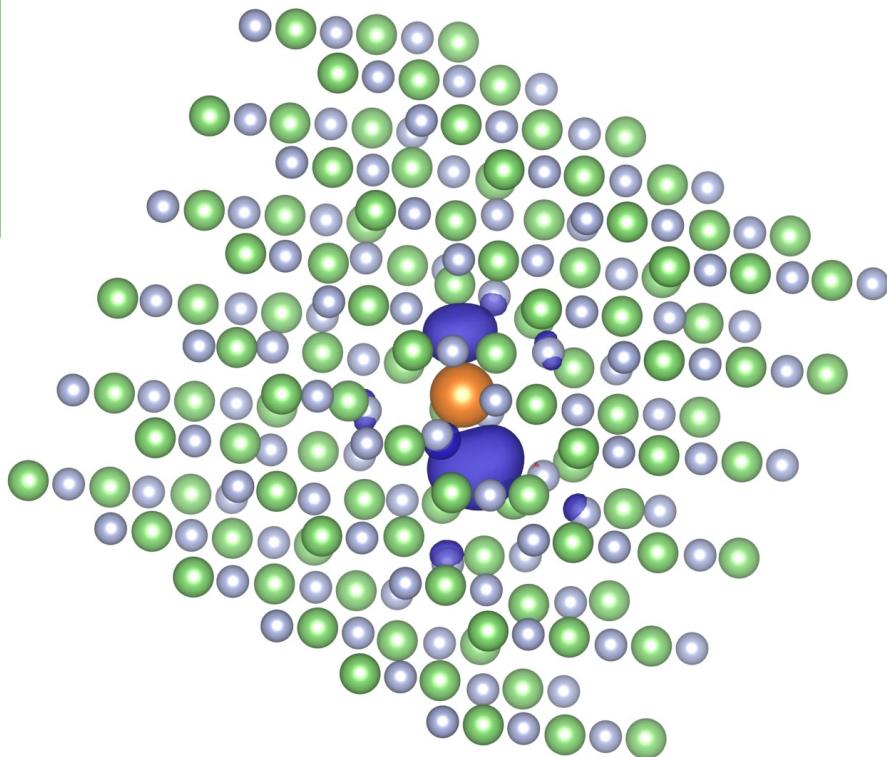


Two-dimensional contour plot of the F vacancy spin density

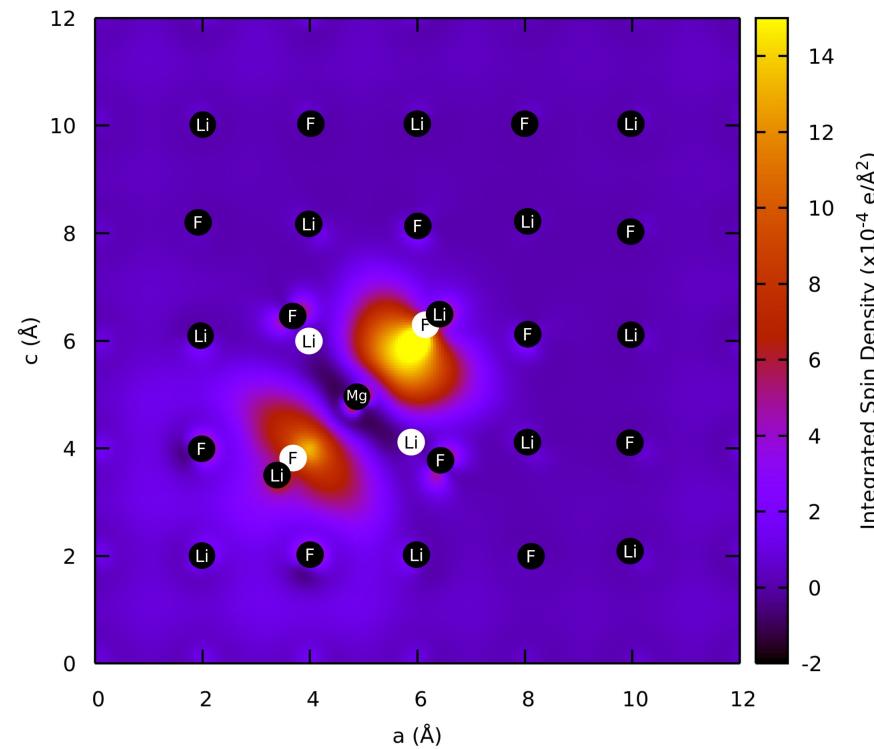


Mg dopant in LiF

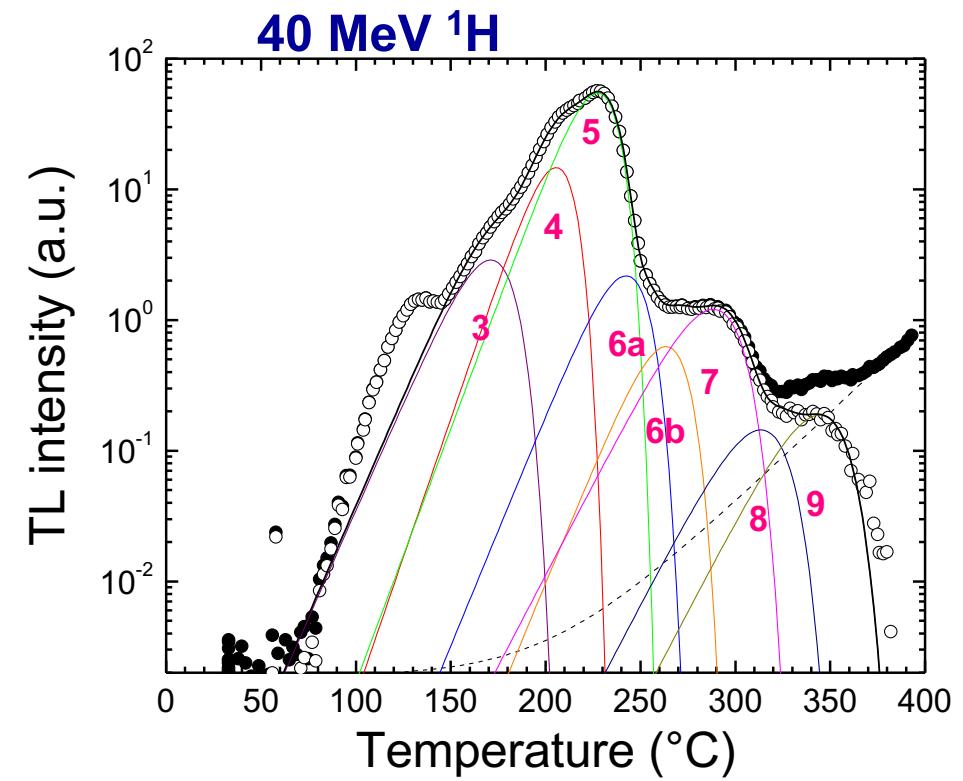
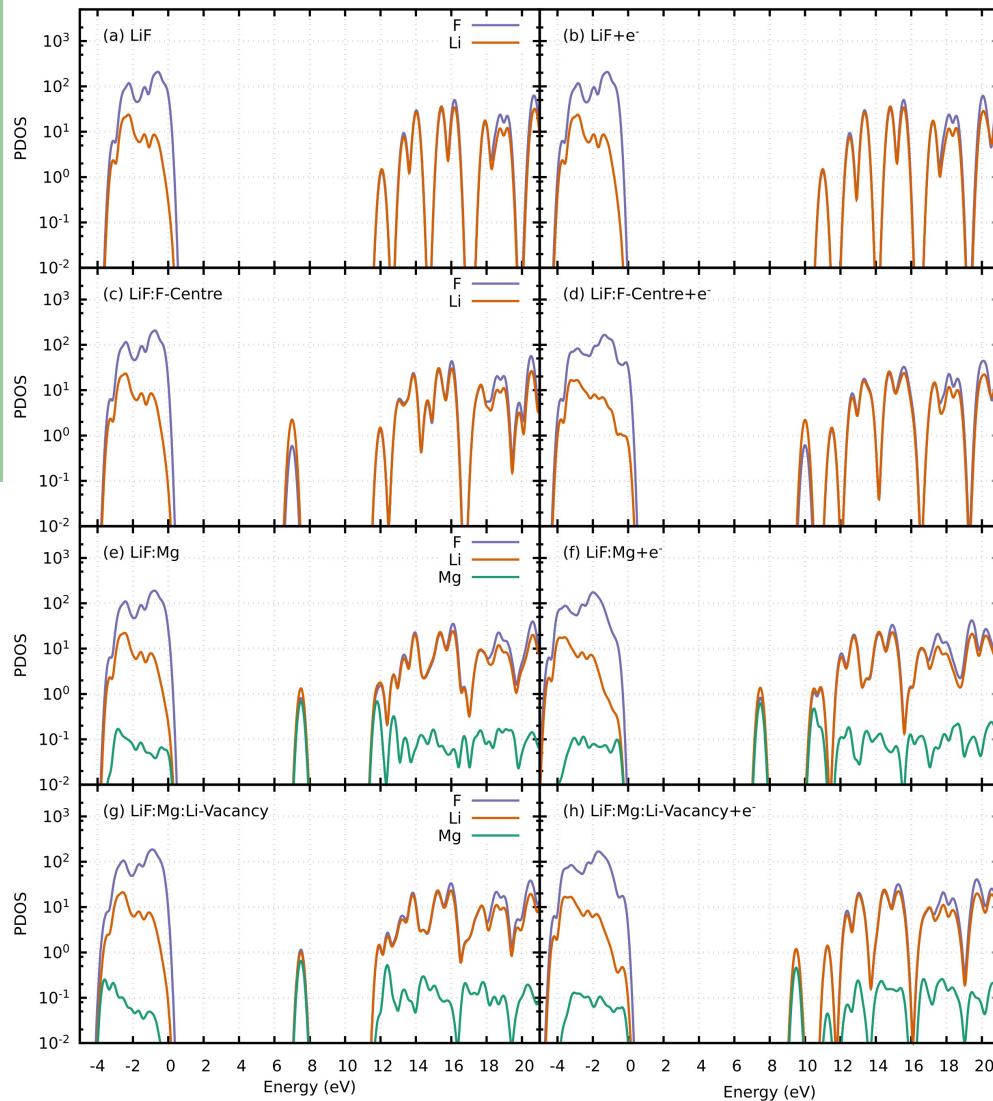
Spin density isosurface of the
LiF:Mg with excess electron



Two-dimensional contour plot of the
LiF:Mg+e⁻ spin density



Density of states for LiF:Mg



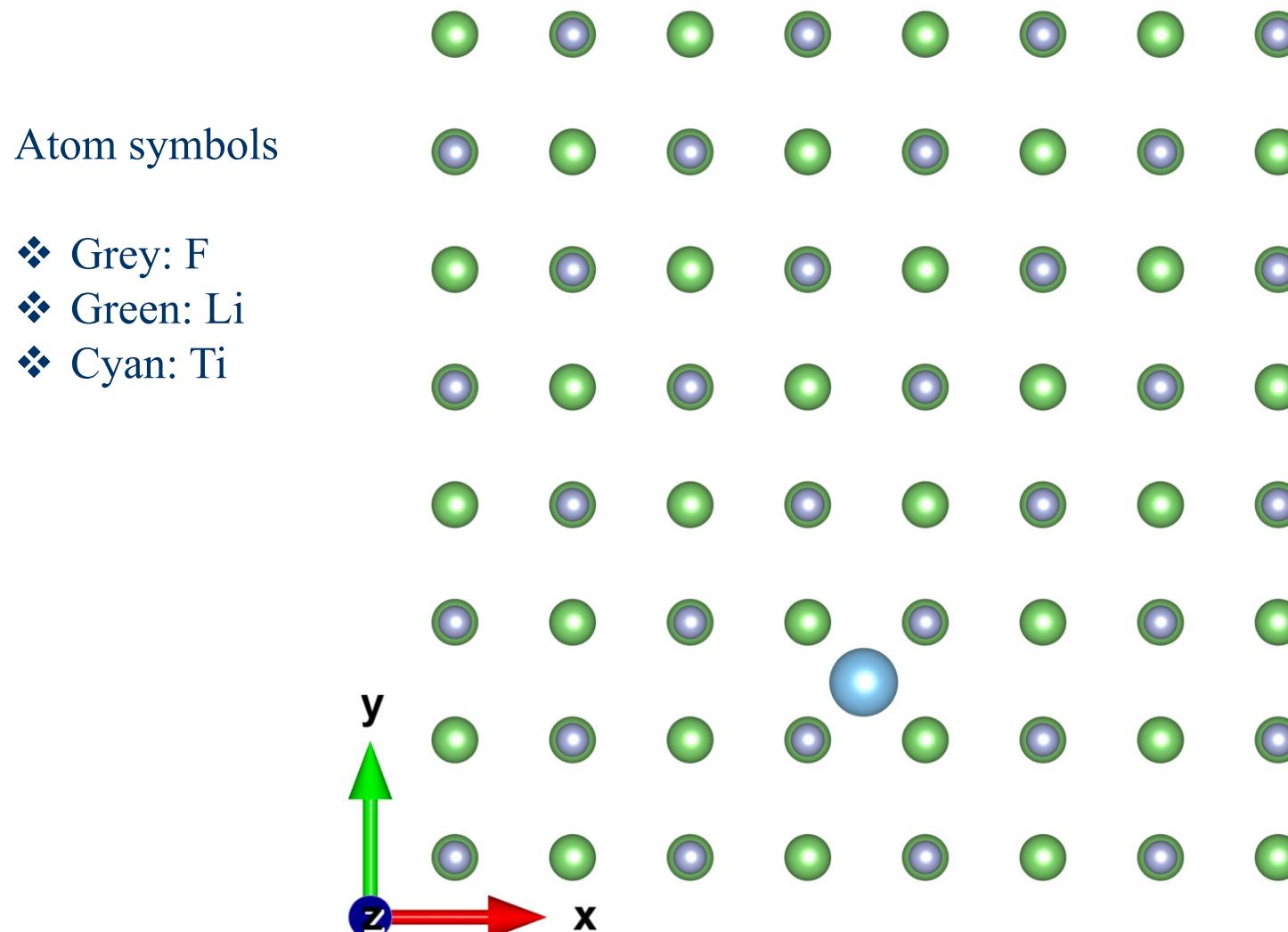
G Massillon-JL, PhD Thesis UNAM 2006

Energy level of the defects in LiF:Mg

Table 1. Kohn–Sham energy levels of the defects, $E_{\text{CBM}} - E_{\text{DS}}$, in eV.

Defect-state	PBE	PBE0	<u>Experiment</u>
LiF-gap	9.06	12.075	<u>14.2 + 2 [44]</u>
F vacancy 1	3.420	<u>5.064</u>	<u>5.061 [11]</u>
F vacancy 2	1.629	<u>1.866</u>	—
LiF:Mg	2.270	<u>4.534</u>	<u>4.428 [10–15]</u>
(LiF:Mg + e^-) ¹	2.268	<u>4.865</u>	—
(LiF:Mg + e^-) ²	—	<u>4.224</u>	<u>4.000 [10–15]</u>
(LiF:Mg + e^-) ³	—	2.688	
(LiF:Mg–Li _{vac}) ¹	2.557	4.487	
(LiF:Mg–Li _{vac}) ²	1.461	1.584	
LiF:Mg–Li _{vac} + e^-	2.514	<u>4.544</u>	
(LiF:Mg–Li _{sub}) ¹	1.092	2.889	
(LiF:Mg–Li _{sub}) ²	0.728	0.919	
LiF:Mg–Li _{sub} + e^-	1.785	<u>3.486</u>	<u>3.263 [10–15]</u>
LiF:Mg–Li _{sub} –Li _{vac}	—	—	
(LiF:Mg–Li _{sub} –Li _{vac} + e^-) ¹	1.095	<u>2.898</u>	
(LiF:Mg–Li _{sub} –Li _{vac} + e^-) ²	<u>0.730</u>	0.922	

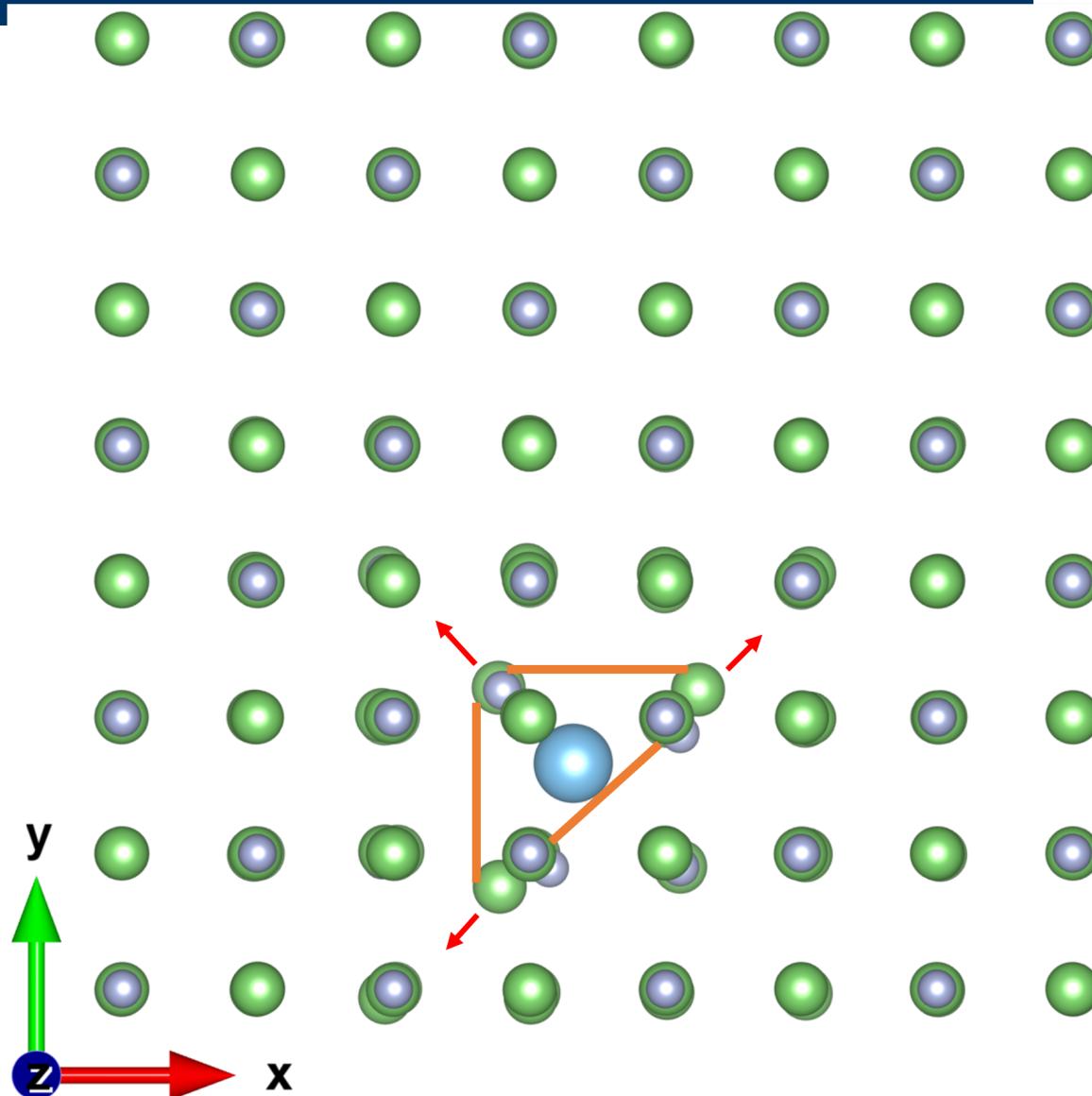
Illustrative structure for the initial configuration of LiF:Ti



Geometry configuration of LiF:Ti after relaxation

Atom symbols

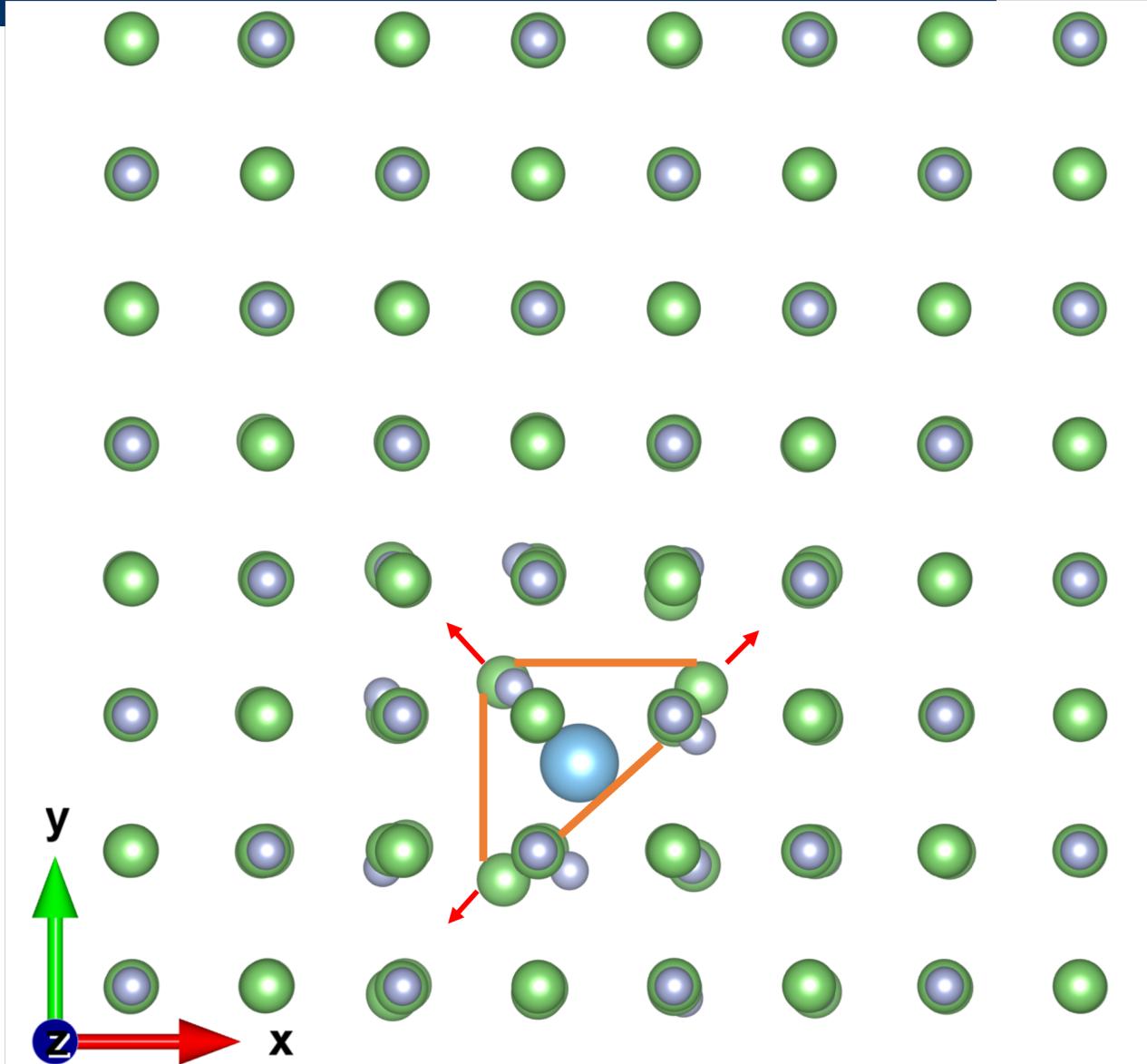
- ❖ Grey: F
- ❖ Green: Li
- ❖ Cyan: Ti



LiF:Ti with an excess electron after relaxation

Atom symbols

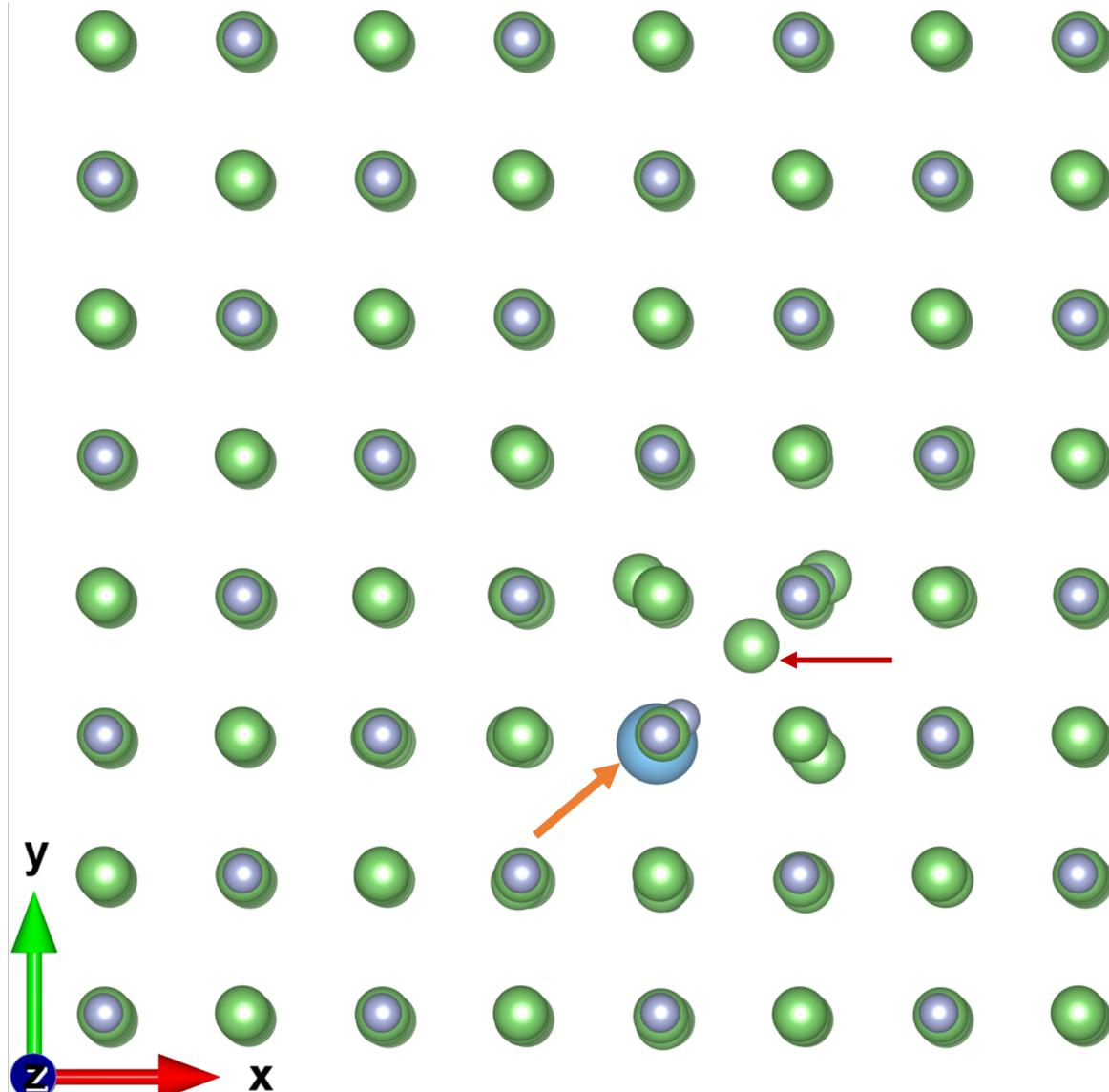
- ❖ Grey: F
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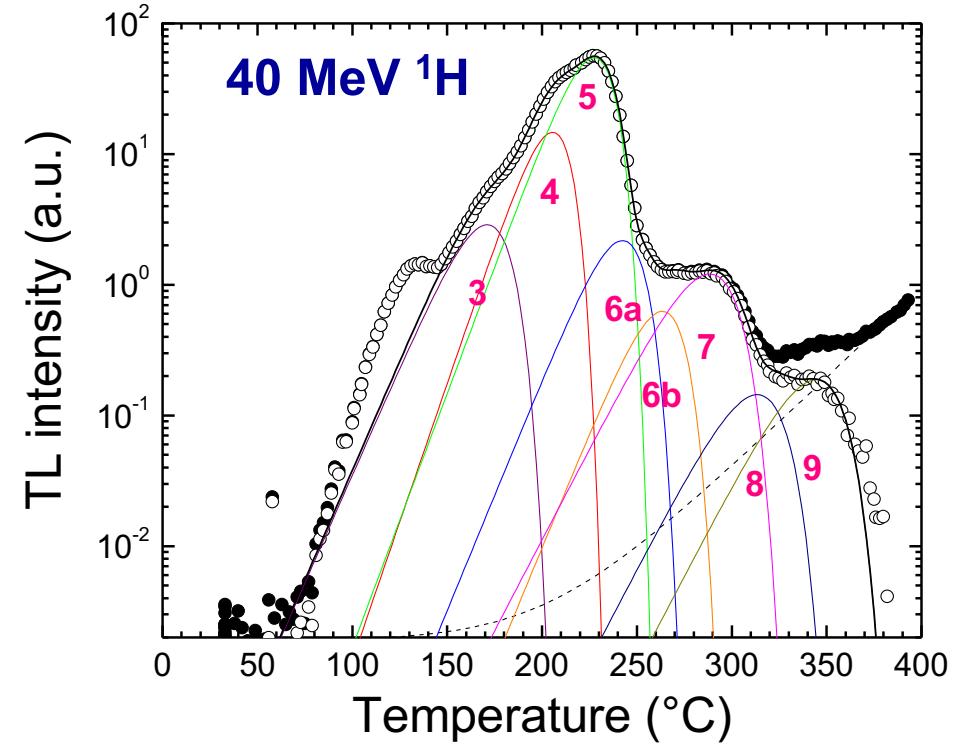
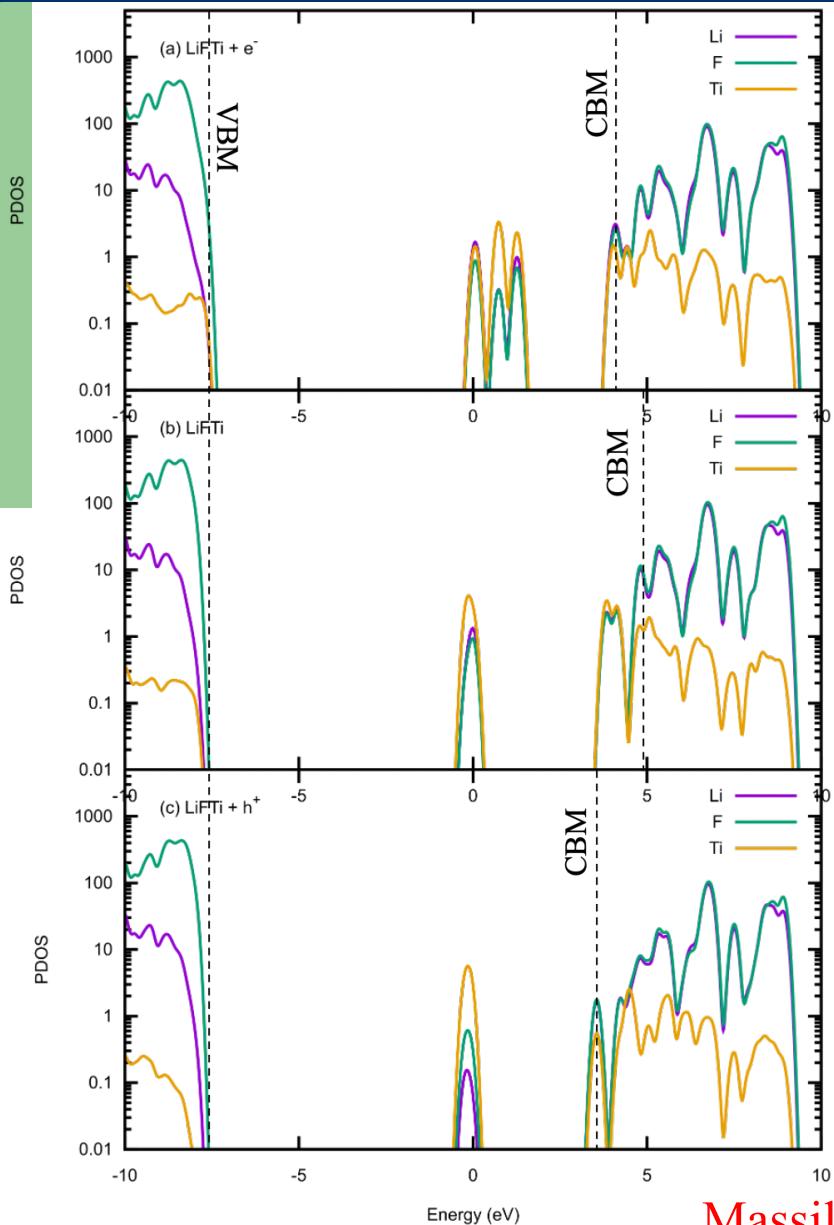
LiF:Ti with a hole after relaxation

Atom symbols

- ❖ Grey: F
- ❖ Green: Li
- ❖ Cyan: Ti



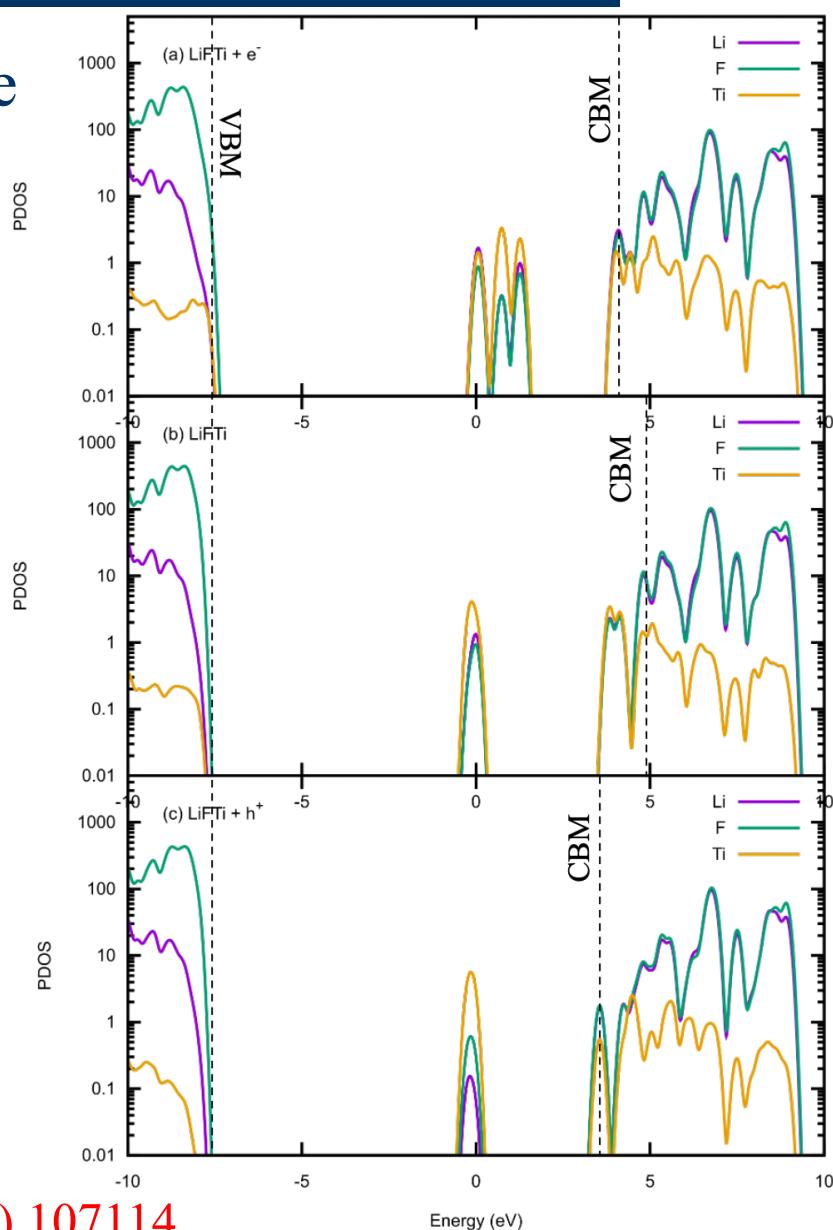
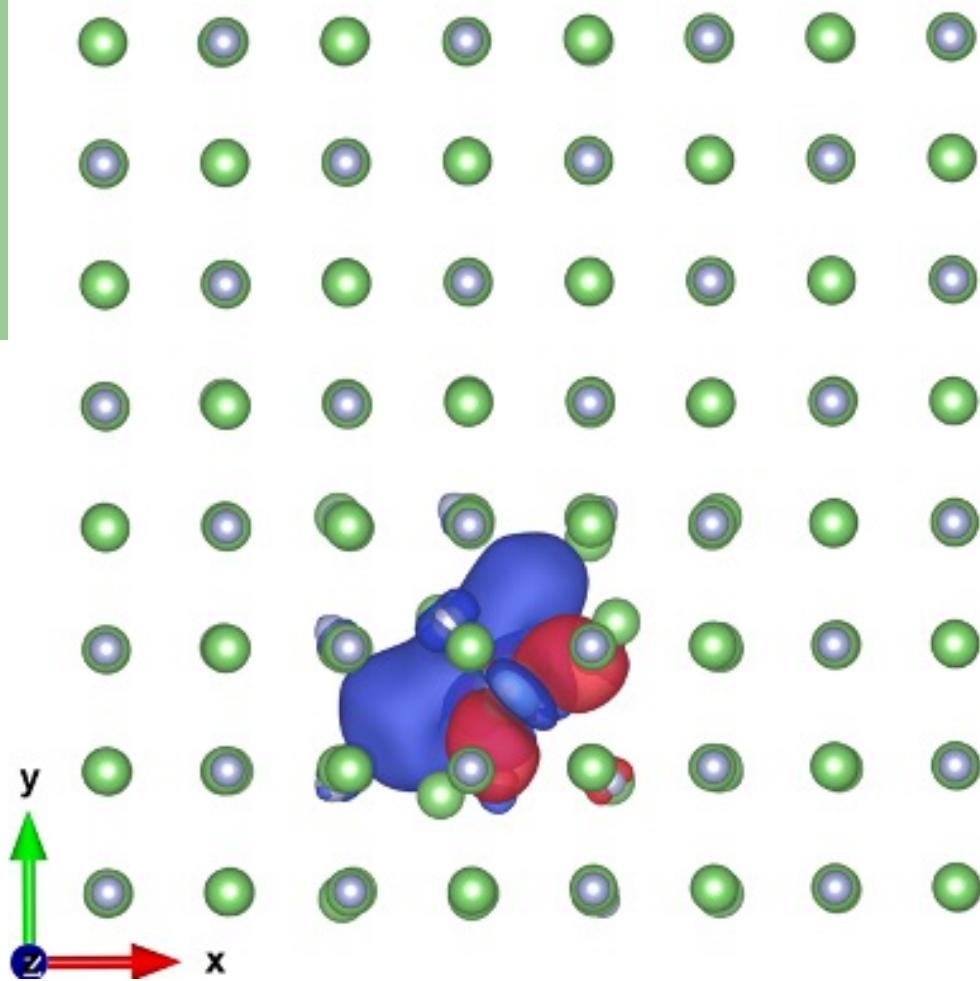
Density of states for LiF:Ti



G Massillon-JL, PhD Thesis UNAM 2006

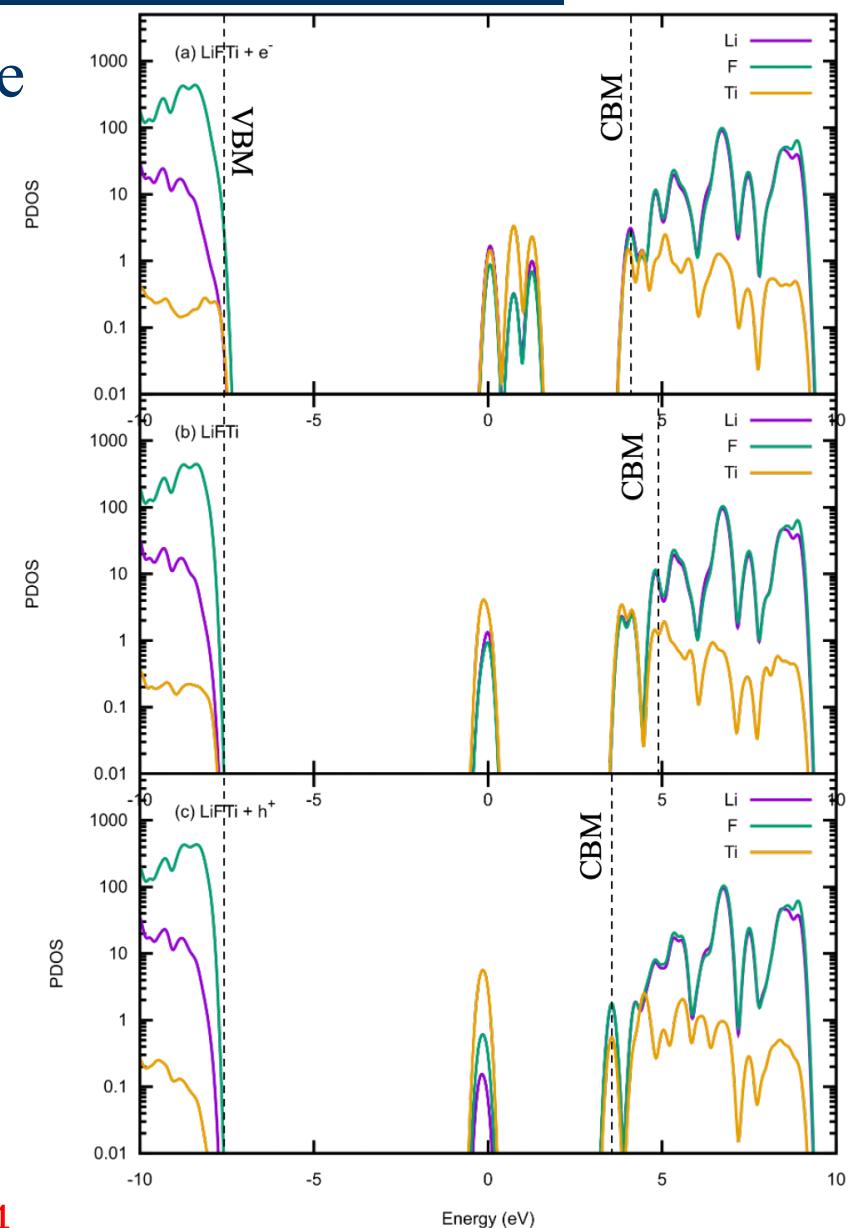
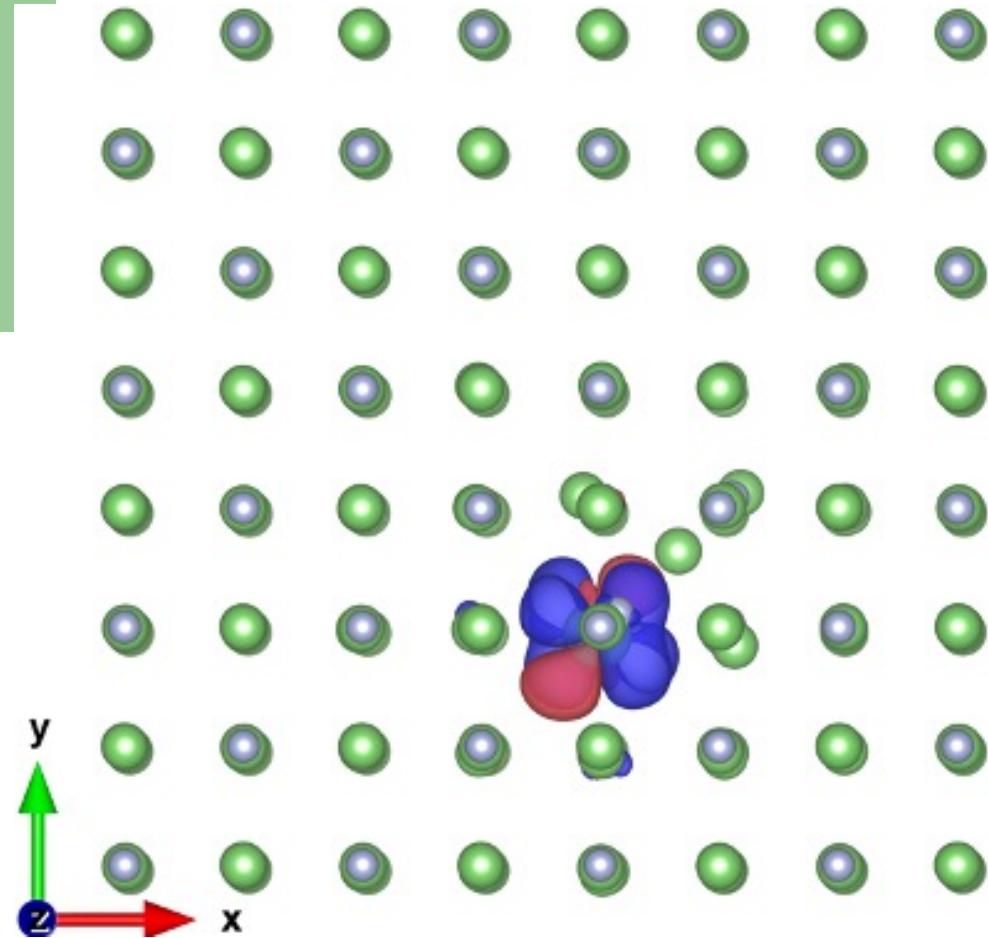
Defect states and electron localization in LiF:Ti

Spin density isosurface: Ti 4p-state



Defect states and hole localization in LiF:Ti

Spin density isosurface: Ti 3d-state



Defect formation energies

Defect formation energy and Thermodynamic transition level, $\varepsilon^{therm}(q_1/q_2)$

$$E_f(X^q) = E_{tot}(X^q) - E_{tot}(bulk) - \sum_i n_i \mu_i + qE_F$$

$$E_f(X^0; E_F = \varepsilon^{therm}(0/-1)) = E_f(X^{-1}; E_F = \varepsilon^{therm}(0/-1))$$

$$\varepsilon^{therm}(0/-1) = E_{tot}(X^{-1}) - E_{tot}(X^0) = [\varepsilon_{i+1}(N) + \varepsilon_{i+1}(N+1)]/2$$

$E_{tot}(X^q)$ and $E_{tot}(bulk)$ are the total energies of the system containing the defect, X , and of the perfect crystal, respectively

$\pm n_i$: number of atoms that have been added (+) or removed (-)

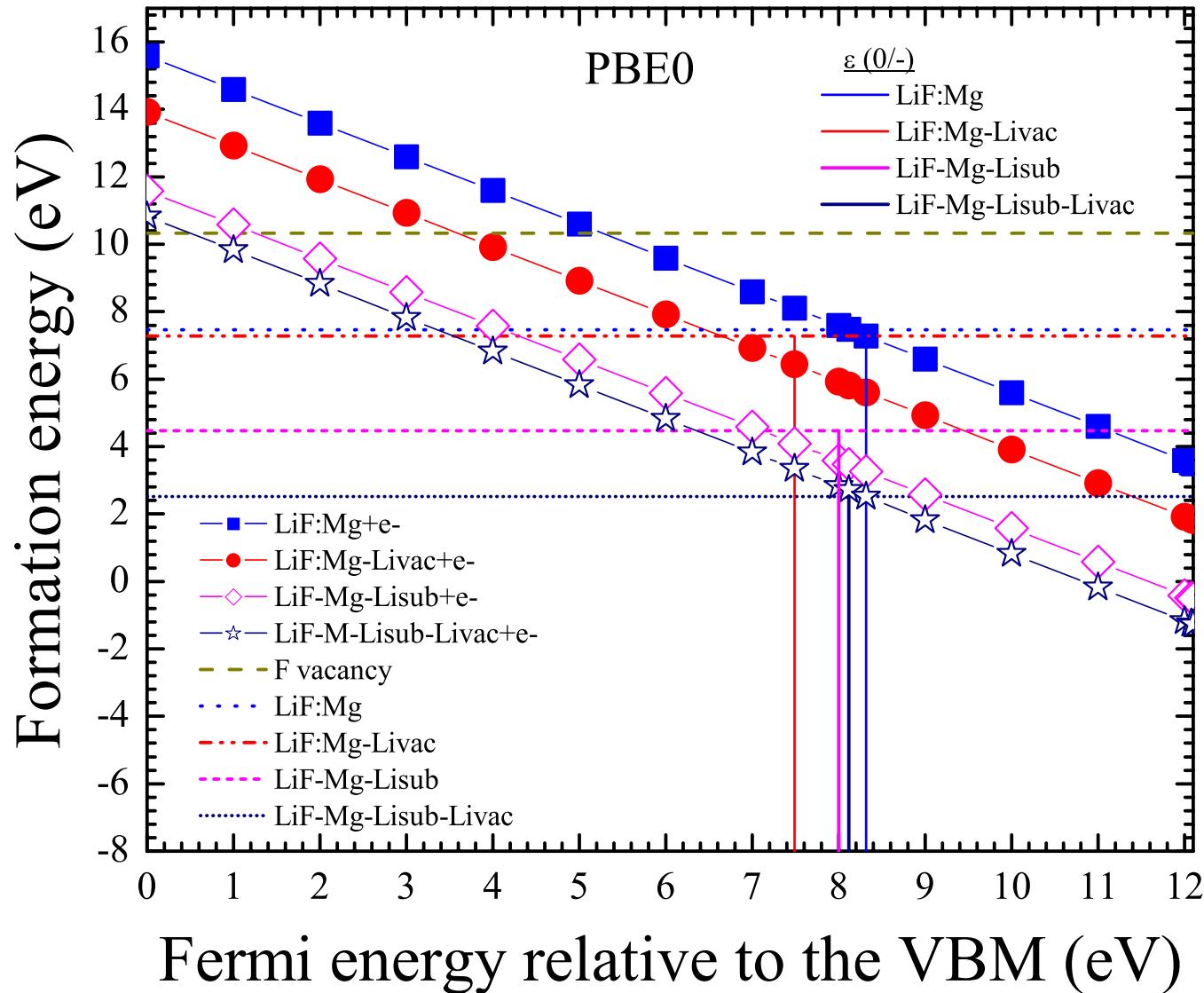
μ_i : chemical potentials

E_F : Fermi energy

E_f : Formation energy

$\varepsilon_{i+1}(N+1)$ eigenvalue of the highest occupied state of the $(N+1)$ -electron system

Defect formation energy



Defect formation energy

Table 2. Defect formation energies, E_f , and thermodynamic transition levels, $\varepsilon^{\text{therm}} (0/-1)$, of various defect-states in LiF at the PBE0 theory level. Energies in eV.

Defect-state	$E_f (E_F = 0)$	$\varepsilon^{\text{therm}} (0/-1)$
LiF		
LiF-F vacancy	10.33	
LiF:Mg	7.463	
LiF:Mg + e^-	15.34	8.32
LiF:Mg-Li _{vac}	7.29	
LiF:Mg-Li _{vac} + e^-	13.92	7.49
LiF-Mg-Li _{sub}	4.48	
LiF-Mg-Li _{sub} + e^-	11.58	8.00
LiF-Mg-Li _{sub} -Li _{vac}	2.53	
LiF-Mg-Li _{sub} -Li _{vac} + e^-	10.83	8.11

LiF:Mg, Summary

- The Mg interstitial creates a void for electron traps and is stable under normal conditions
- All charged defects are created by radiation and stables
- It is possible to identify the electron localization and quantify the energy loss for the defect's formation

Not all the energy deposited into the dosimeter's sensitive volume is transformed to a certain response

LiF:Ti, Summary

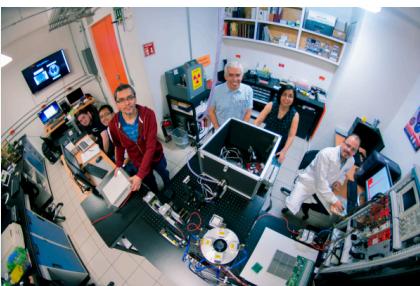
- In the presence of excess electrons, Ti stays in an interstitial position.
- Excess electrons are localized in the Ti 4p-orbitals
- In the presence of holes, Ti becomes a substitutional Li atom and the Li is moved to a tetrahedral interstitial position.
- Holes are localized in the Ti 3d-orbitals which are recombination centers



Ti dopant plays a dual character in LiF when exposed to ionizing radiation

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El egresado de esta maestría tendrá:
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