

Neutron irradiation effects – point defect balance equations, swelling, Reactor pressure vessel

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1. *Materials selection strategie*
2. *Defect balance equations*
3. *Observed microstructural defects in fcc & bcc metals*
4. *Void swelling*
5. *Reactor pressure vessel*

References:

1. *Chapter 17 & 18 of D. R. Olander, Fundamental Aspects of Nuclear Reactor Fuel Elements, NTIS (1976)*

Engineering material selection for (nuclear) engineering application

Traditional engineering materials selection

*P.Hosemann, M. Fratoni,
D. Frazer, M.F. Ashby
Scripta Mat 2018*

Requirement defined by the engineers

Materials Property

Stiffness

Elastic Modulus

Temperature

Melting point, creep, etc

Stress

Yield stress

Lifetime

Creep,

Load cycles

Cycles to failure

Environment

Corrosion rates

Accident scenario

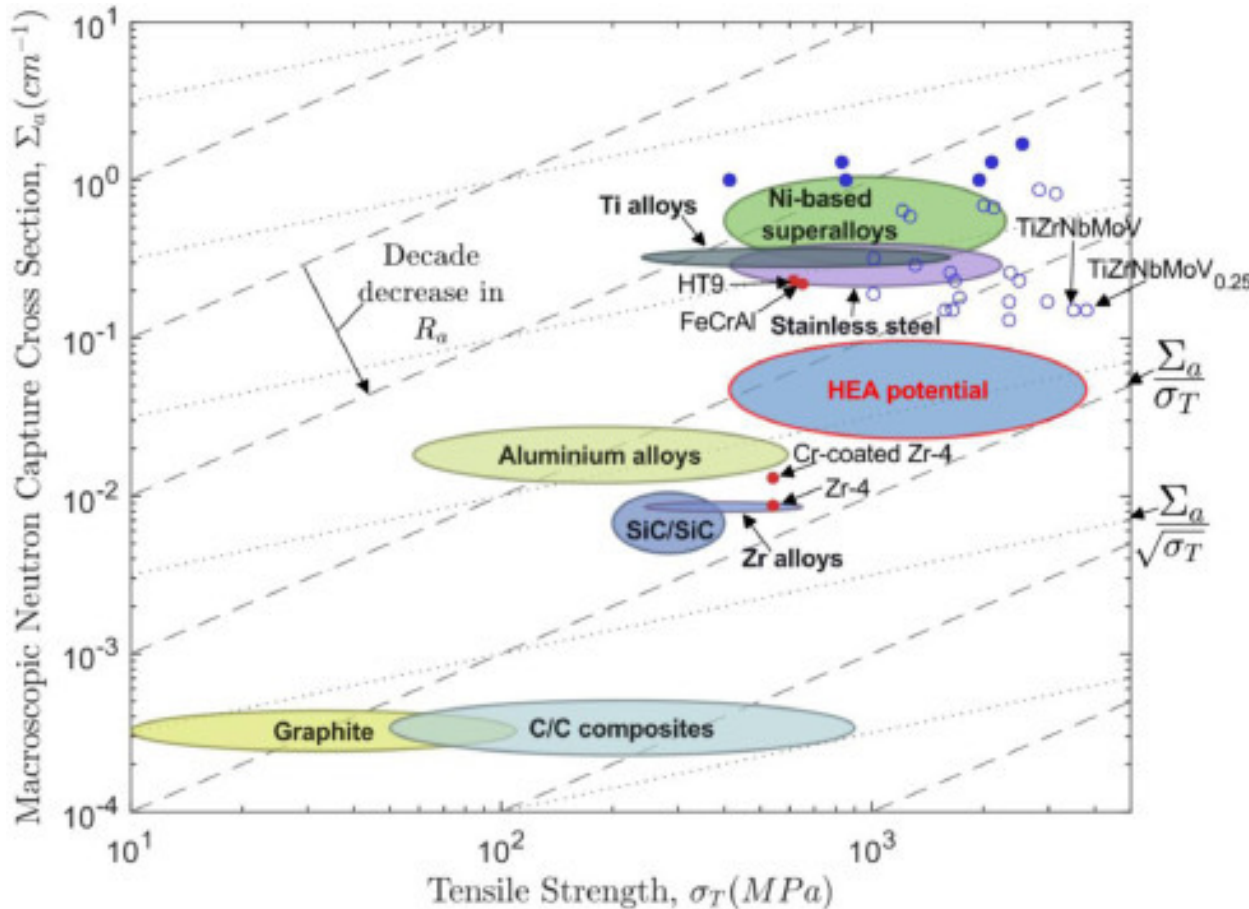
*Elongation,
fracture toughness*

Light weight, X-ray absorption

Density

Materials properties considering neutron capture

Michael Moschetti, Patrick A. Burr, Edward Obbard, Jamie J. Kruzic, Peter Hosemann, Bernd Gludovatz



Design considerations for high entropy alloys in advanced nuclear applications

JNM 2022

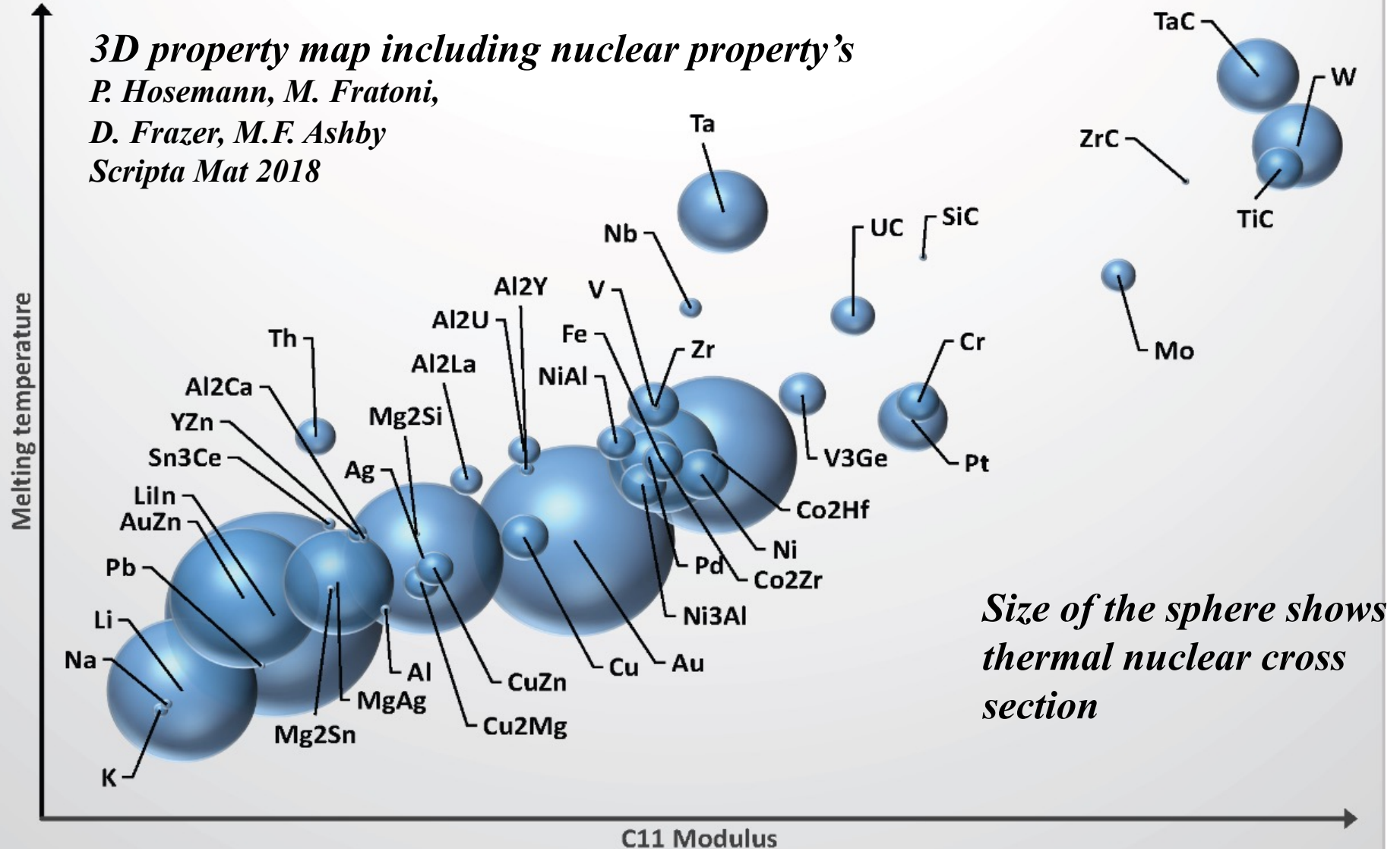
Materials selection for nuclear applications

3D property map including nuclear property's

P. Hosemann, M. Fratoni,

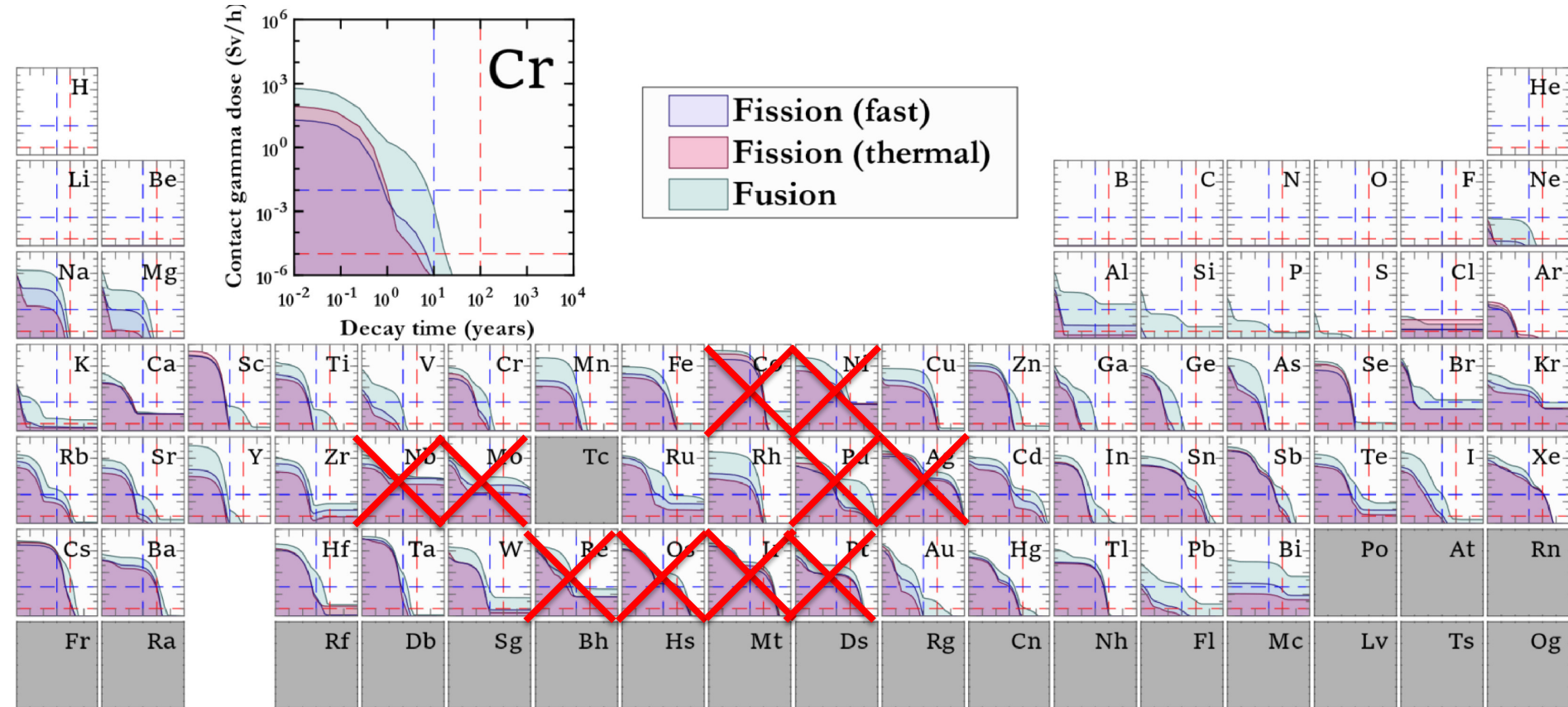
D. Frazer, M.F. Ashby

Scripta Mat 2018

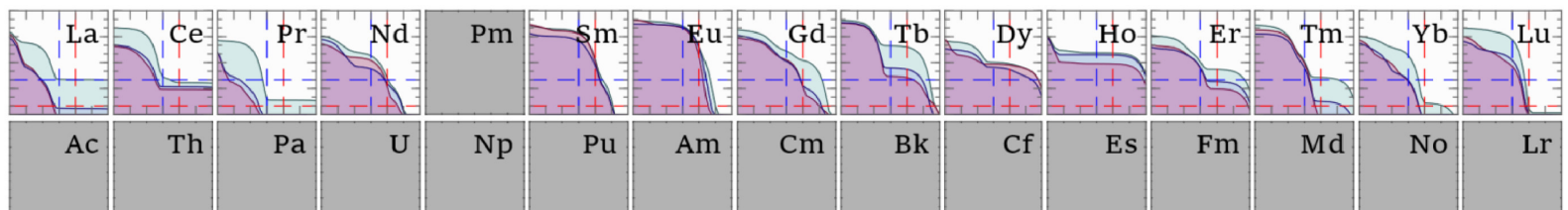


Selection of elements for LA materials. Decay time after service

M. Moschetti, et al. JNM 2022

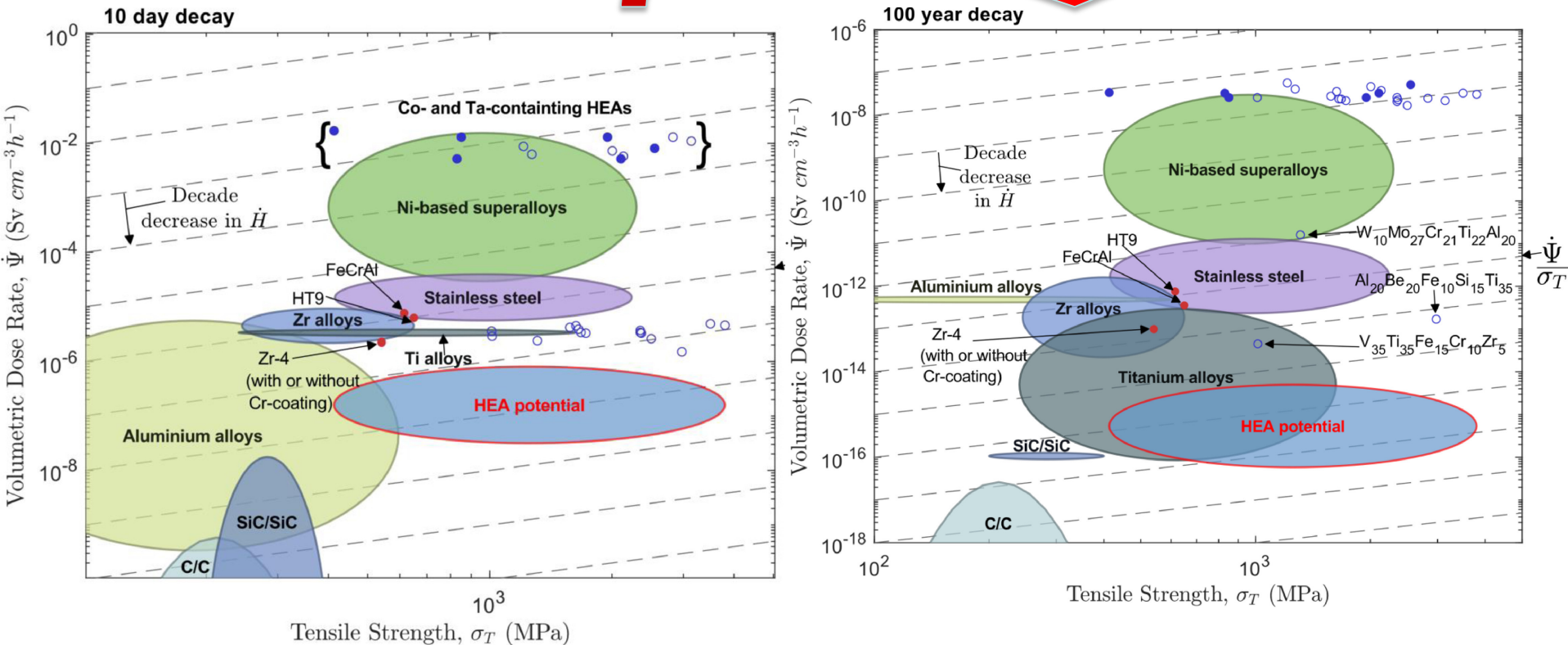


PWR and SFR, fusion first-wall (5 years of service)



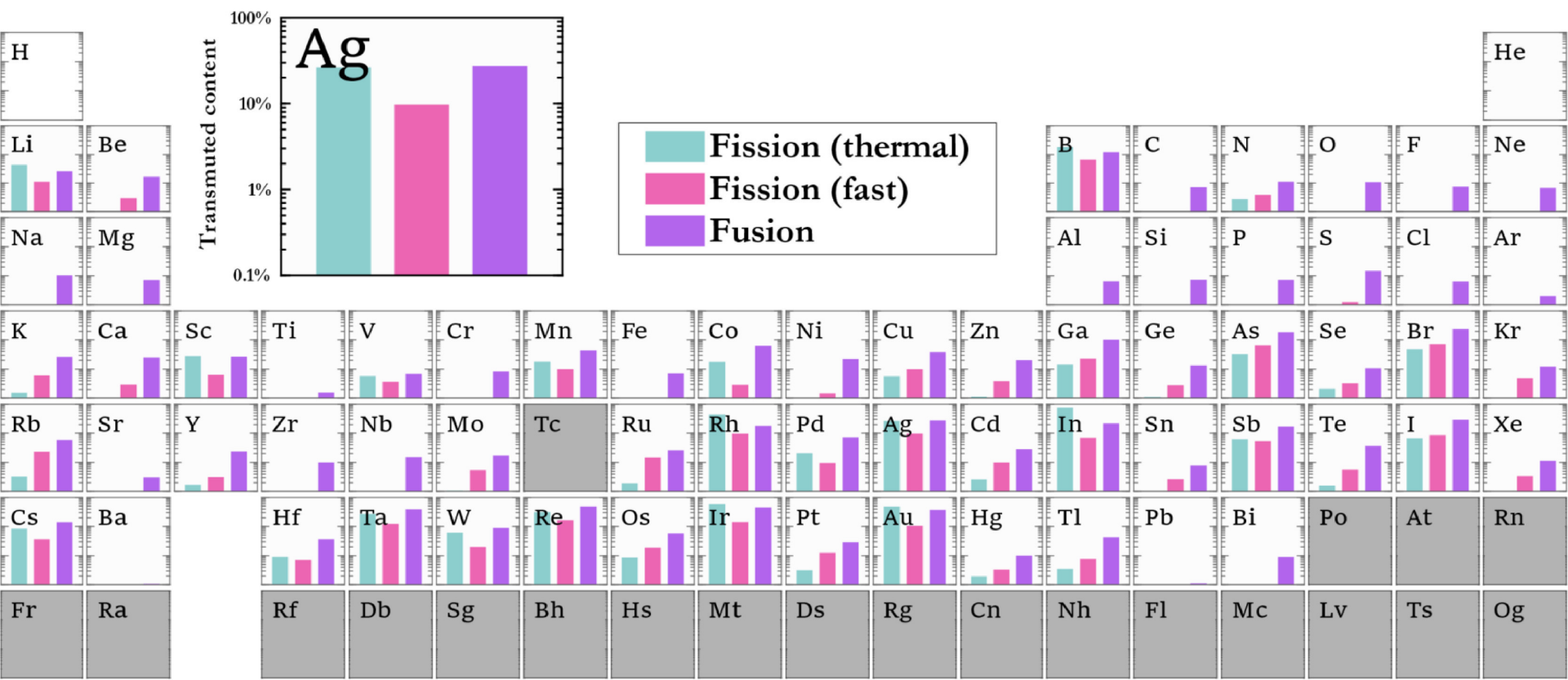
Volumetric dose rate (irradiation under thermal fission conditions)

M. Moschetti, et al. JNM 2022

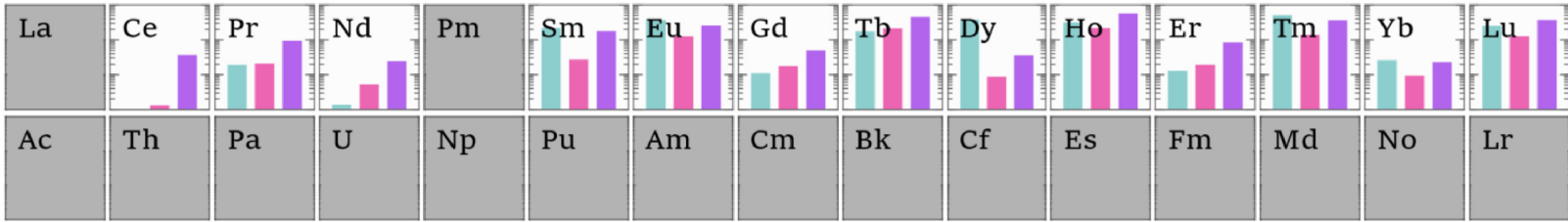


Compositional stability of irradiated materials

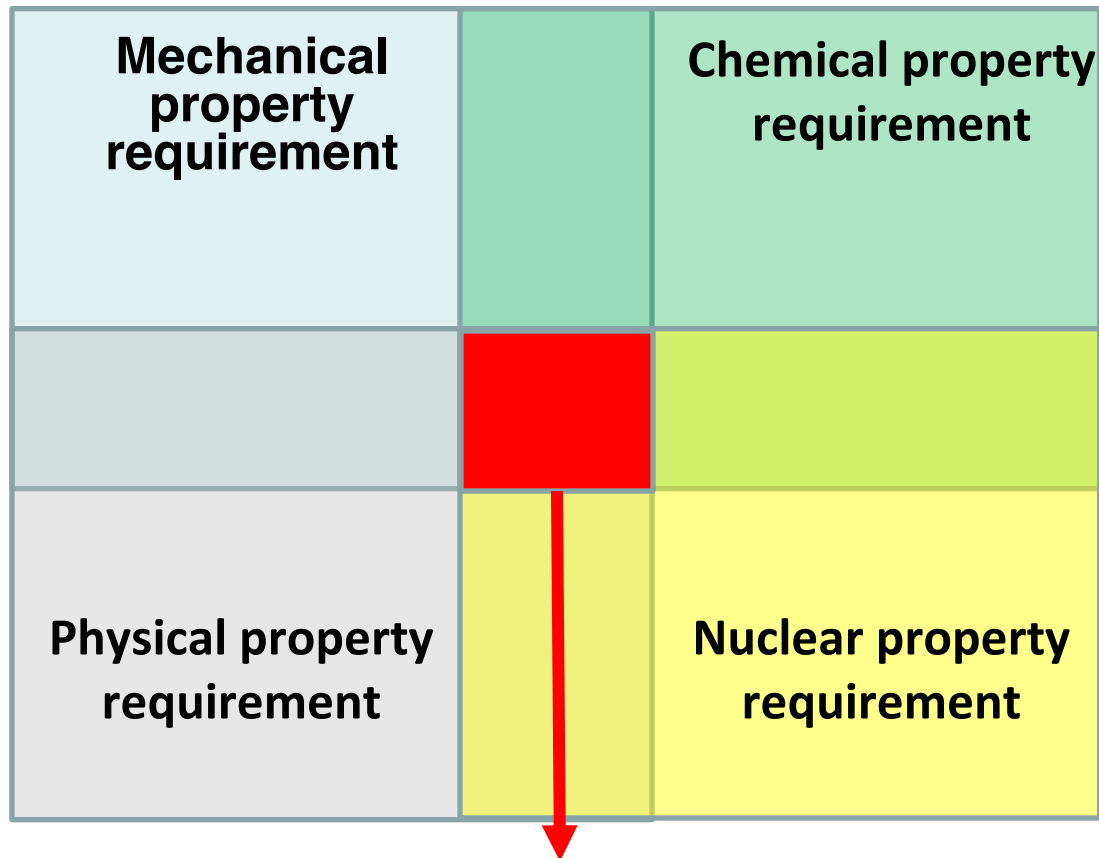
M. Moschetti, et al. JNM 2022



PWR and SFR, fusion first-wall (5 years of service)



Materials selection for nuclear applications



Limited design space!

Crystals and defects why are they important?

"Crystals are like people: it is the defects in them that make them interesting",

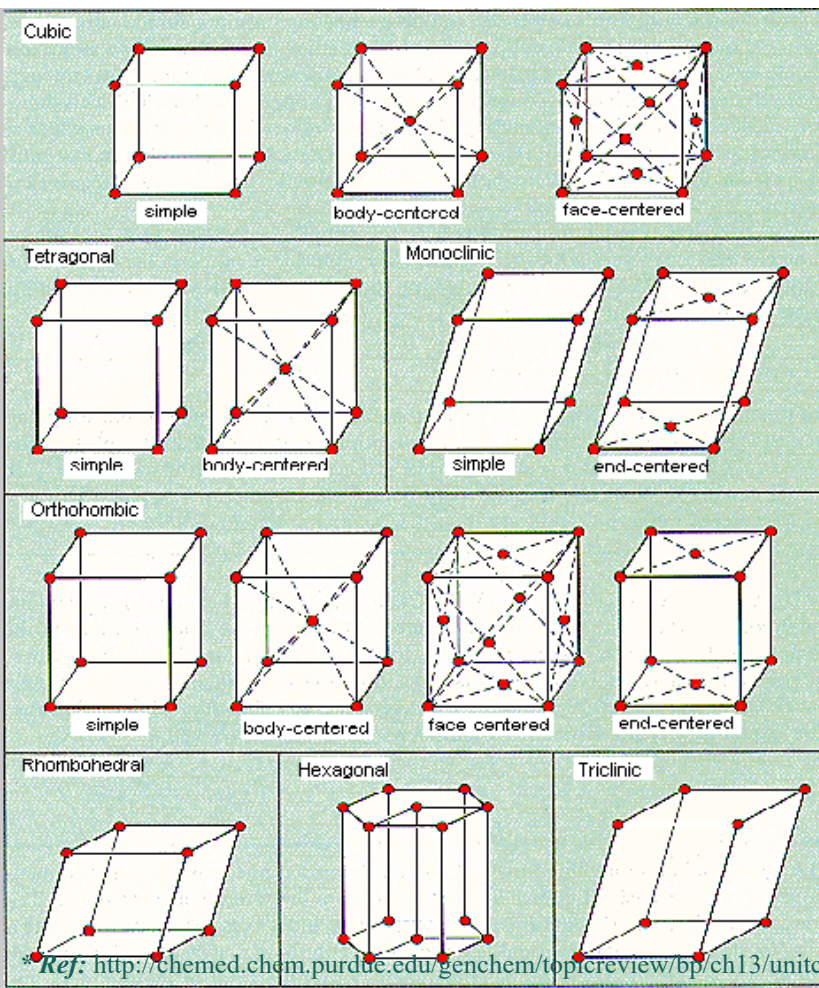
Colin Humphreys Introduction to Analytical electron microscopy

Definition of crystal structures

Unit Cell:

- Smallest unit translation of “space lattice” that reproduces a macroscopic crystal

- Crystalline solids: Many types of crystal arrangements, but all reduce to 14 unique lattice structures (Bravais lattice)
- Most common structural materials are of



Atomic Mac

The Atomic Mac™

Registered to:
GREGORY TETRAULT
For use only on a single machine

 Hexagonal

 Cubic body centered

 Rhombohedral

 Cubic

 Cubic face centered

 Monoclinic

 Orthorhombic

 Tetragonal

H	He																	He
Li	Be																	Ne
Na	Mg																	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba			Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra			Rf	Db	Sg	Bh	Hs	Mt	Uun	Uuu	Uub	Uut	Uuq	Uup	Uuh	Uus	Uuo
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu				
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr				

Defects in crystals

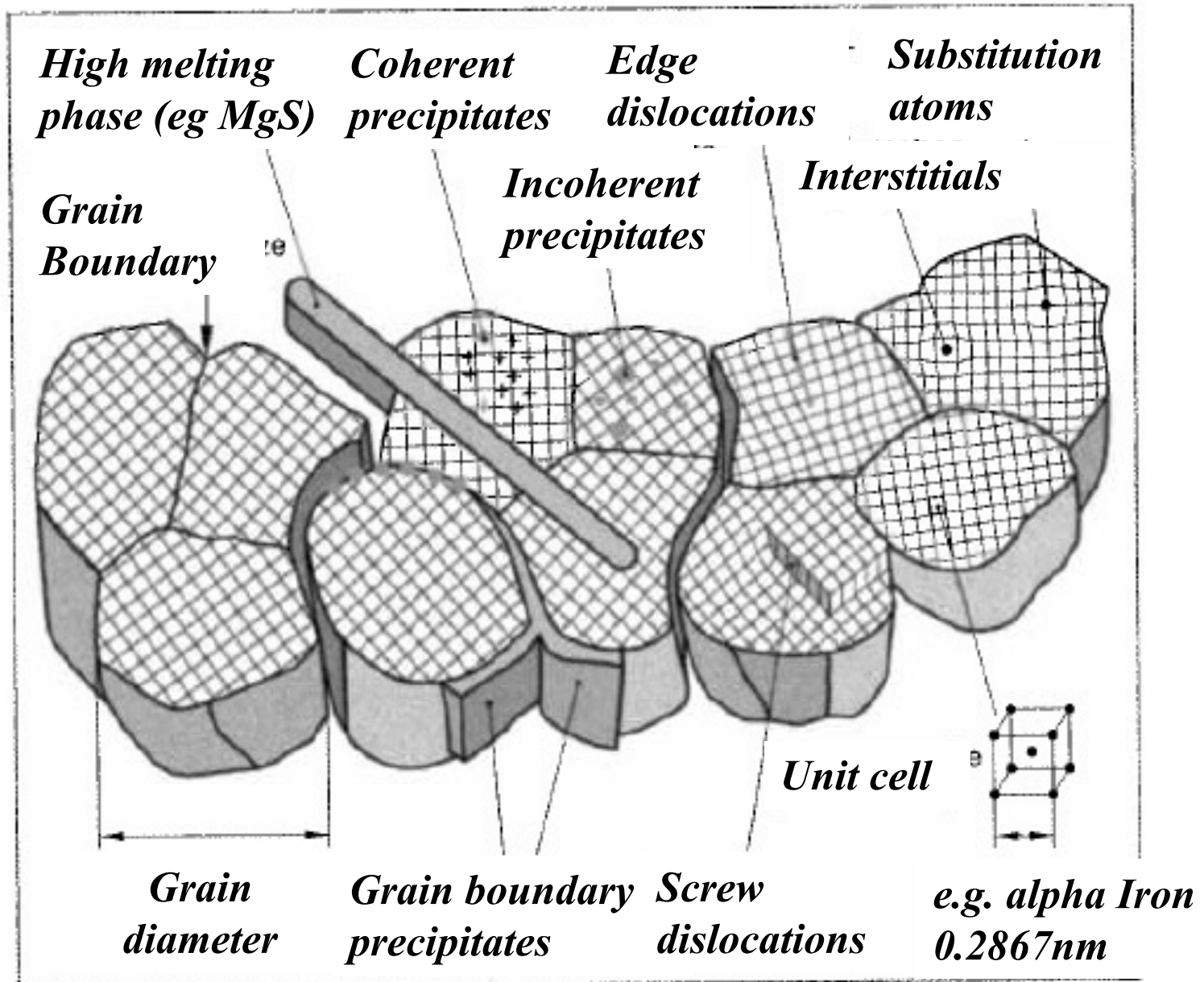
0-dimensional defects:

1 dimensional defects:

2 dimensional defects:

3 dimensional defect:

Real materials always have defects.

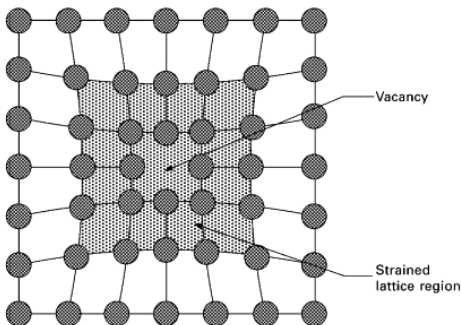
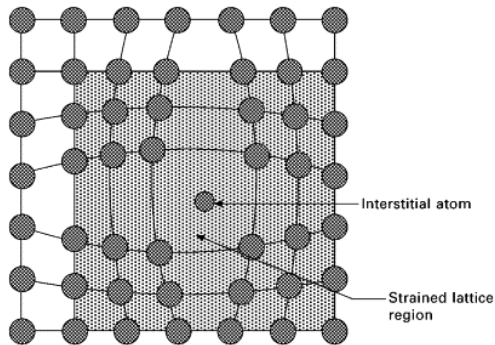


Important note for later understanding!

Each defect has a property

Significant higher strain field
→ *Strong attraction towards loops*

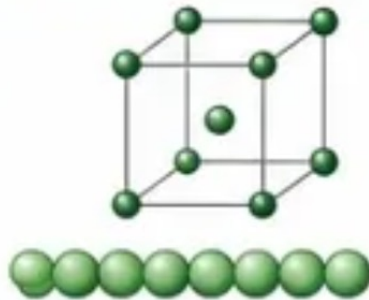
Much faster diffusion



Relative lower strain field
→ *weaker attraction towards loops*

Much slower diffusion

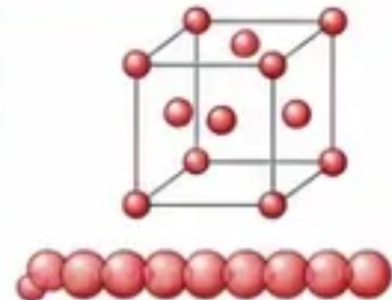
B) BODY-CENTERED CUBIC (BCC)



- Atoms at corners + 1 at center
- Atoms per unit cell = 2
- Coordination Number = 8
- Relation: $\sqrt{3}a = 4r$
- Packing Efficiency:

= 68%

C) FACE-CENTERED CUBIC (FCC)



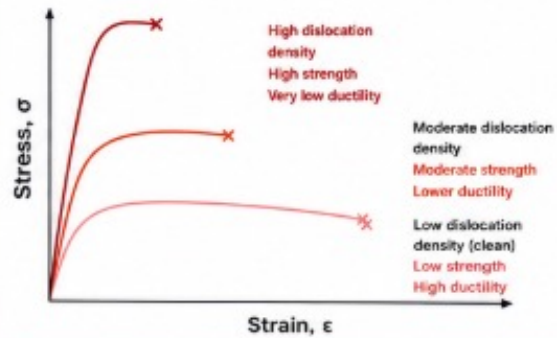
- Atoms at corners + face centers
- Atoms per unit cell = 4
- Coordination Number = 12
- Relation: $\sqrt{2}a = 4r$
- Packing Efficiency:

= 74%

How do defects change mechanical properties?

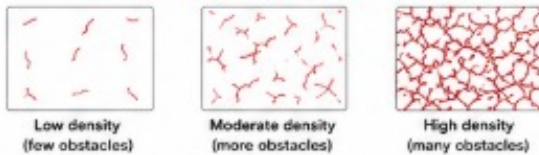
A. DISLOCATION DENSITY

More dislocations → higher strength, less ductility



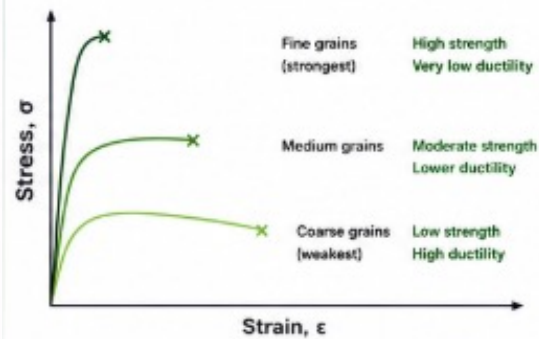
TREND: $\sigma_y \uparrow$ $\sigma_{UTS} \uparrow$ Slope (E) ~ same $\epsilon_f \downarrow$

MICROSTRUCTURE VIEW (dislocations)



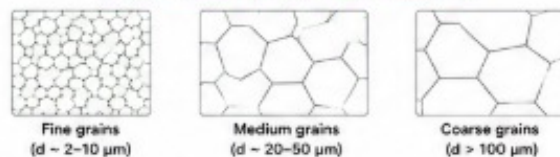
B. GRAIN SIZE (HALL-PETCH)

Smaller grains → higher strength, less ductility



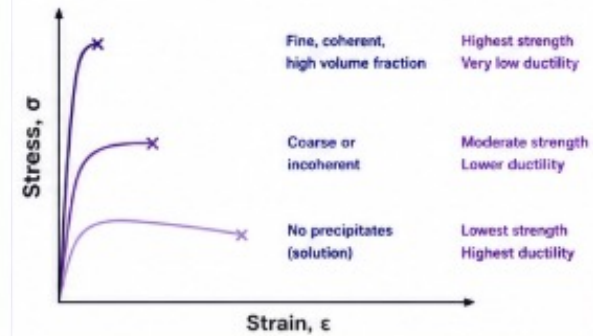
TREND: $\sigma_y \uparrow$ $\sigma_{UTS} \uparrow$ Slope (E) $\uparrow$ slightly $\epsilon_f \downarrow$

MICROSTRUCTURE VIEW (grain structure)



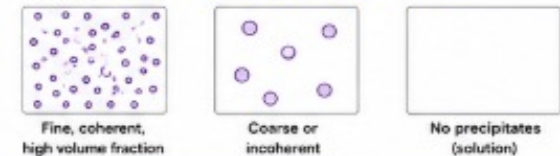
C. PRECIPITATES

More/stronger precipitates → higher strength, less ductility

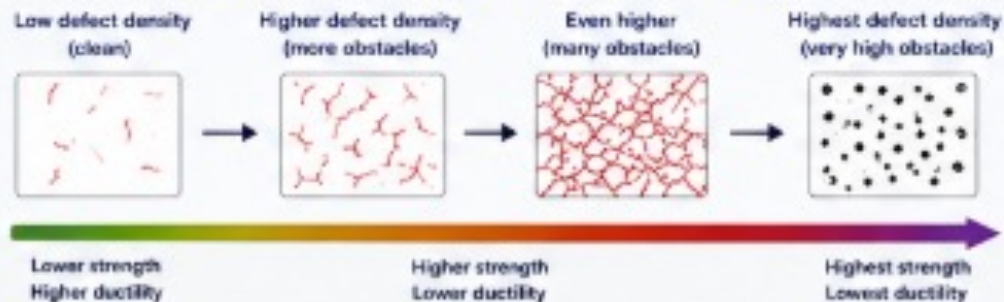


TREND: $\sigma_y \uparrow$ $\sigma_{UTS} \uparrow$ Slope (E) ~ same $\epsilon_f \downarrow$

MICROSTRUCTURE VIEW (precipitates)



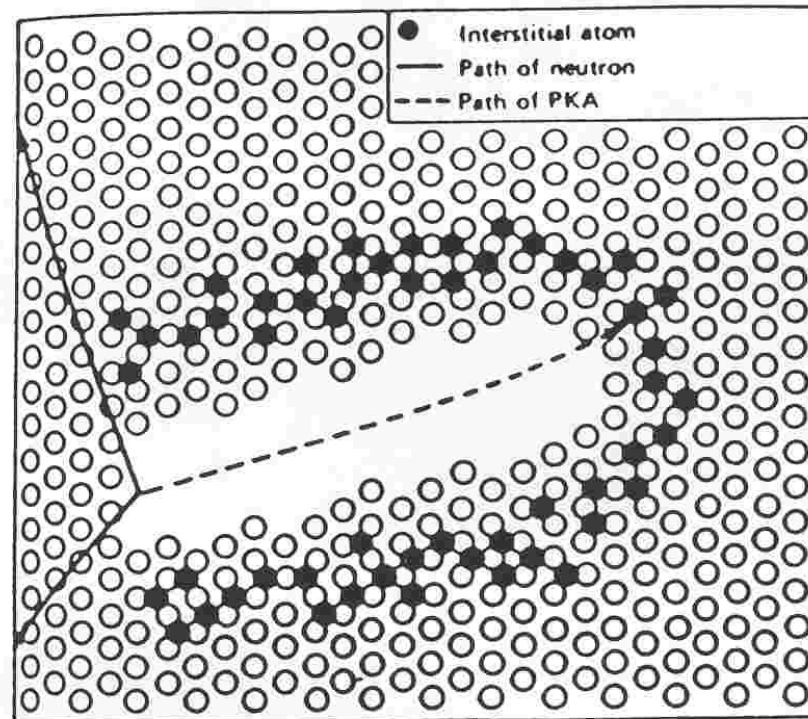
OVERALL EFFECT OF INCREASING DEFECT DENSITY



How does the radiation damage turn into these defects?

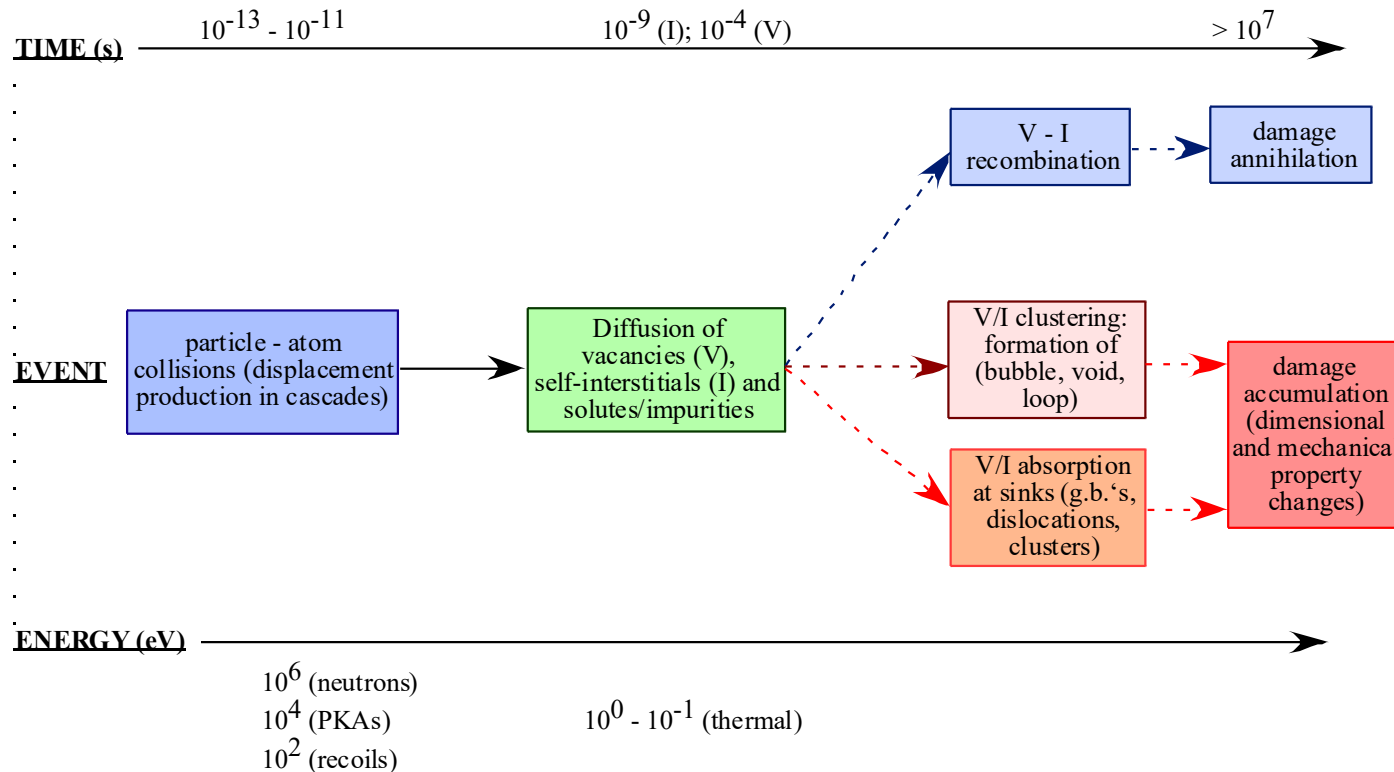
V-I interaction rate theory

Radiation damage creates point defects

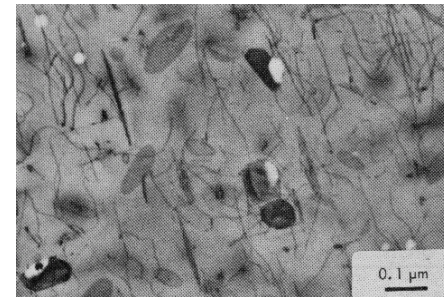


**Original model of the displacement spike,
J.A. Brinkman, *Amer. J. Phys* 24 (1956) 24.**

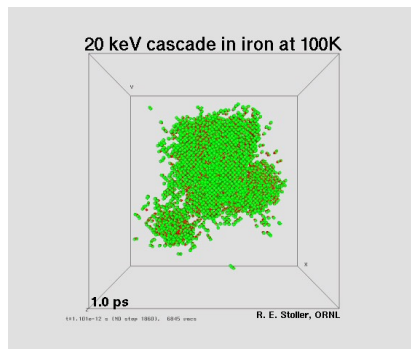
Radiation effects are governed by the ultimate fate of point defects (& transmutants)



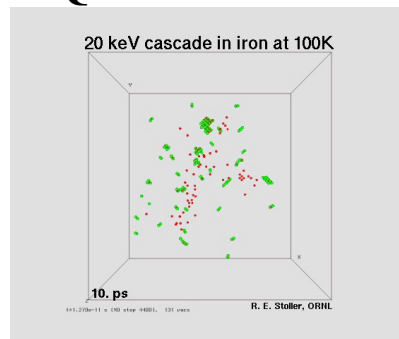
Voids, I-loops
& precipitates
in stainless steel



Nascent cascade



“Quenched” cascade



Number of defects:

Nascent cascade: $N_i = N_v = v(T)$

“Quenched” cascade: $N_i' = N_v' < v(T)$

Point Defect Balance Equations

Objective: determine the concentration of vacancies and self-interstitials in an irradiated solid

Assumes infinite medium (no long-range spatial gradients of C_v or C_i)

Assumes that only V and I are mobile (no mobility of self-interstitial or vac-clusters)

$$\frac{\partial C_{i/v}}{\partial t} = \text{Rate of change of } C_{i/v} = \text{Production} - \text{Loss} \quad (9.1)$$

Point Defect Balance Equations

Loss Mechanisms:

Recombination ($V + \text{SIA} = \text{null}$)

$$\frac{\text{recombination rate}}{\text{unit volume}} = k_{iv} C_i C_v \quad (9.4)$$

$$k_{iv} \cong \frac{500\Omega}{a_o^2} (D_i + D_v) \approx \frac{500\Omega}{a_o^2} D_i \quad (9.5)$$

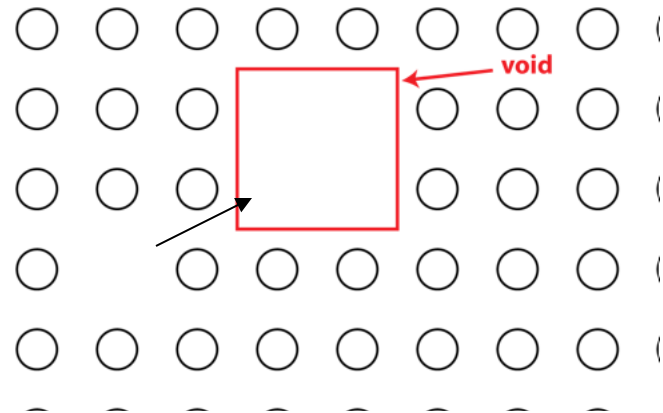
k_{iv} = recombination rate constant

500... combinatorial number describing the number of sites the recombination can occur (~250 atoms/nm³ in bcc Fe)

a₀ = lattice constant

interstitials are mobile

Point Defect Balance Equations



Absorption at voids for vacancies

Rate of absorption # vacancies (or self-interstitials) per unit time

$$J_v^C = 4\pi R D_v (C_{vb} - C_R) \quad (9.6)$$

Volumetric absorption rate = $N_v \cdot J_v^C$:

$$= D_v \underbrace{[4\pi R N_v]}_{S_v^C} (C_{vb} - C_R) \quad (9.7a)$$

$$S_v^C = 4\pi R N_v \quad (\text{Cavity sink strength}) \quad (9.7b)$$

$$\text{FYI: } J_v^{\text{void}} = 4\pi R^2 D_v \left. \frac{\partial C}{\partial r} \right|_{r=R} = 4\pi R D_v [C_v^{\text{bulk}} - C_v^{\text{eq}}]$$

C_R = concentration of solute species at cavity surface

C_b = concentration of solute species (vacancies) in solid far from cavity in the bulk

R = cavity radius

Point Defect Balance Equations

Absorption at voids

Self-interstitials:

$$C_{ib} = C_i$$

$$C_R \approx C_i^{eq} \approx 0$$

$$\text{Volumetric absorption rate} = D_i S_i^c C_i$$

*No interstitials present
in equilibrium*

Vacancies:

$$C_{vb} = C_v$$

$$C_R \approx C_v^{eq} * \exp((\Omega/kT(2\gamma/R-p)))$$

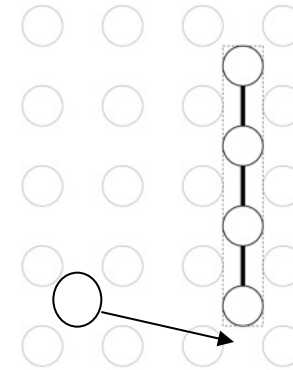
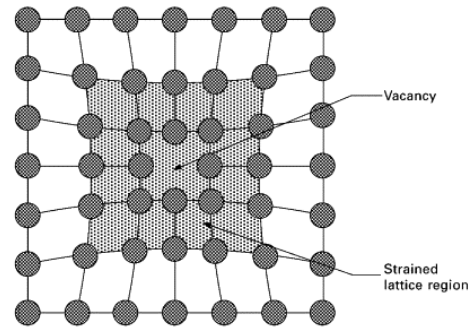
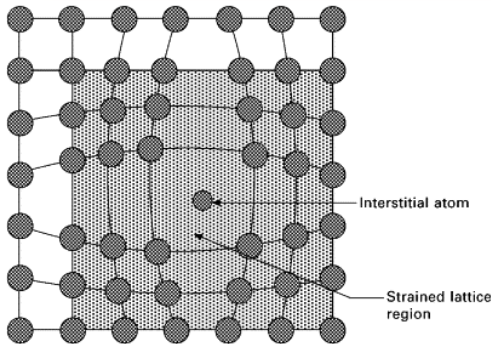
*Vacancies present in
equilibrium*

Volumetric absorption rate

$$= D_v S_v^c C_v - D_v S_v^c C_v^{eq} \exp[(\Omega/kT(2\gamma/R-p))]$$

where, **2nd term represents the production of vacancies by thermal emission** from cavities

Point Defect Balance Equations



Loss Mechanisms:

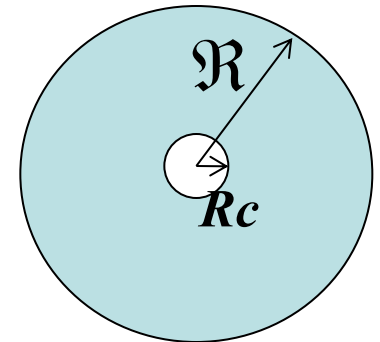
Absorption at dislocations

$$\text{Volumetric absorption rate} = D_{iv} S_{iv}^{\perp} C_{iv} \quad (9.8)$$

$\rho = \text{dislocation density}$

$$S_i^{\perp} = \frac{2\pi\rho}{\ln(\mathfrak{R} / R_{ci})} = Z_i\rho \quad S_v^{\perp} = \frac{2\pi\rho}{\ln(\mathfrak{R} / R_{cv})} = Z_v\rho \quad (9.9)$$

$$\underline{R_{cv} < R_{ci}} \quad (\text{strain field interactions}) \quad \therefore \quad \underline{S_v^{\perp} < S_i^{\perp}}$$



Dislocations are biased sink for self-interstitials!!

Summary equation

Point Defect Balance interstitials:

$$\frac{dC_i}{dt} = \underbrace{Q_i}_{\text{production}} - \underbrace{k_{iv}C_iC_v}_{\text{recombination}} - \underbrace{D_iC_iS_i^t}_{\text{absorption}} \quad S_i^t = S_i^c + S_i^\perp$$

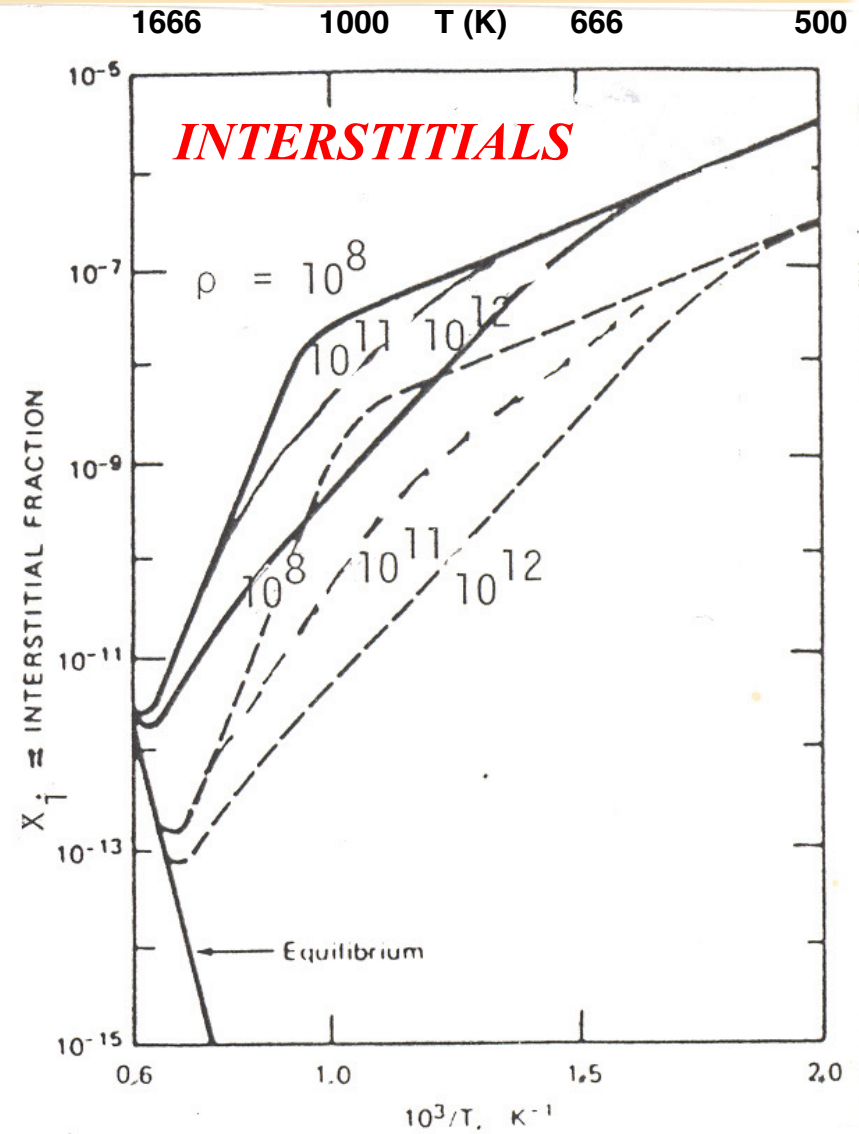
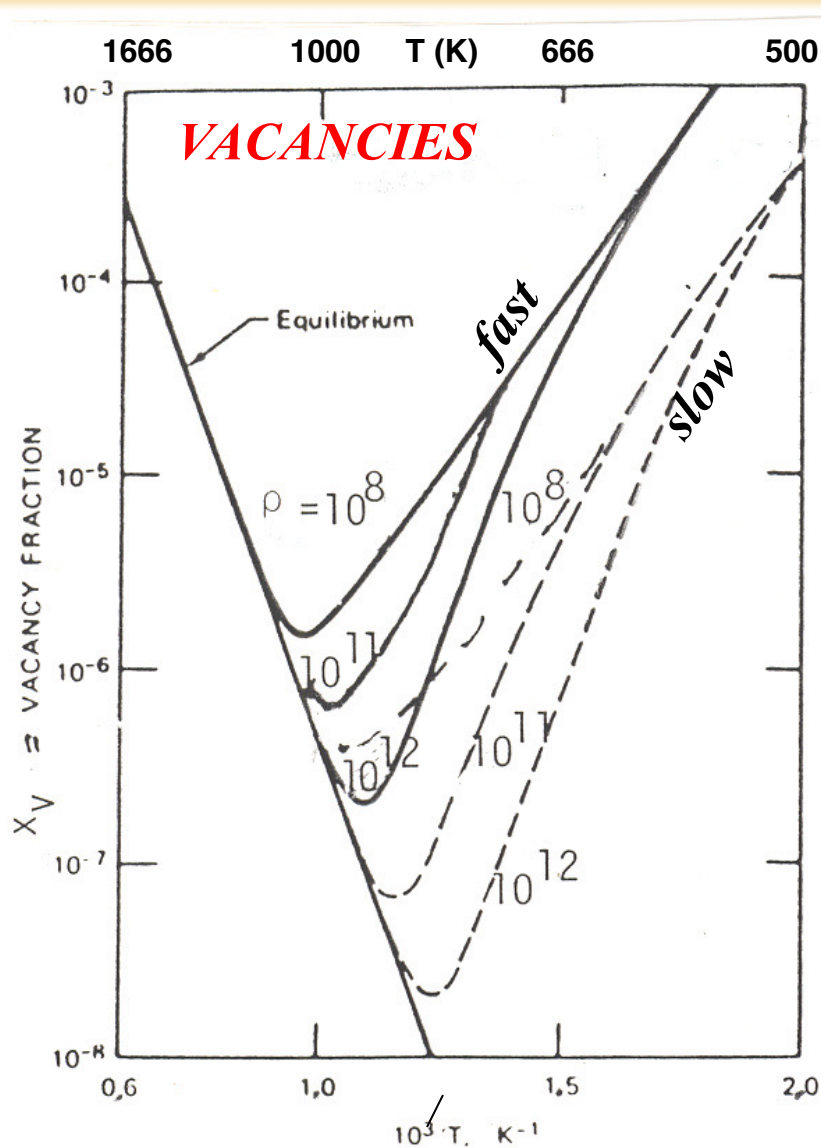
Point Defect Balance vacancies

$$\frac{dC_v}{dt} = Q_v - k_{iv}C_iC_v - D_vC_vS_v^t + Q_v^{th} \quad S_v^t = S_v^c + S_v^\perp$$

Thermal vacancies

$$Q_v^{th} = D_vS_v^cC_v^{eq} \exp\left[\frac{\Omega}{kT}\left(\frac{2\gamma}{R} - p\right)\right]$$

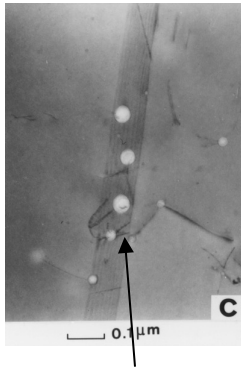
Solutions to the Point Defect Balance Equations



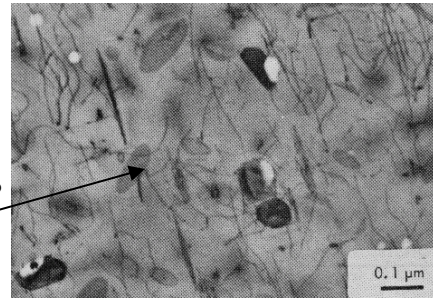
$K = 10^{-6}$ dpa/s (dashed) and 10^{-4} dpa/s (solid)

Observed microstructural changes as a result

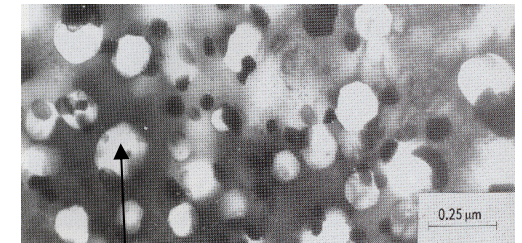
- Cause: long-range diffusion and clustering of displacement damage (defects) produces microstructural changes in the form of dislocation loops, stacking fault tetrahedra, precipitates, bubbles, voids and solute segregation



Voids, I-loops & precipitates in SS

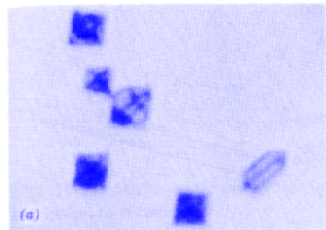


GB solute segregation in SS

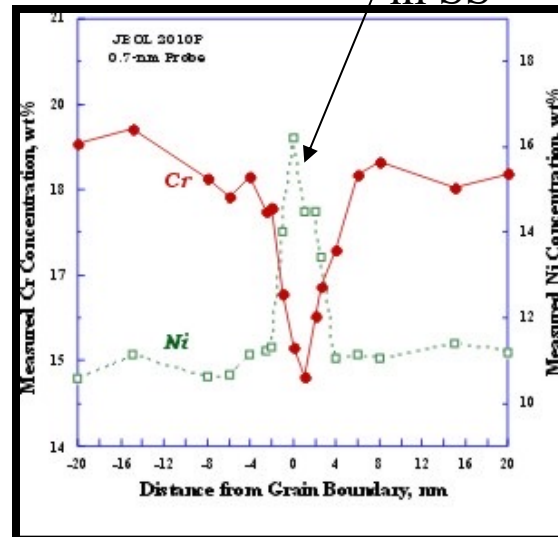
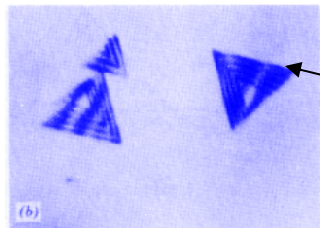


Void swelling in SS

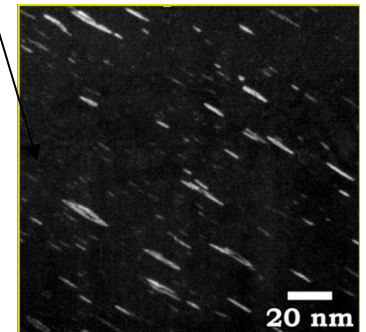
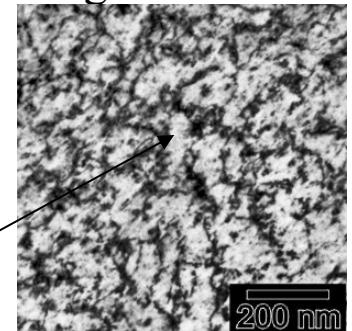
GB He bubbles in SS



SFT in Au



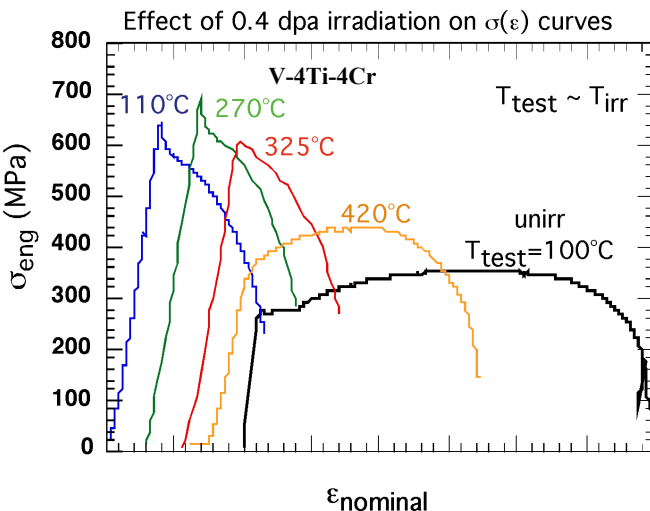
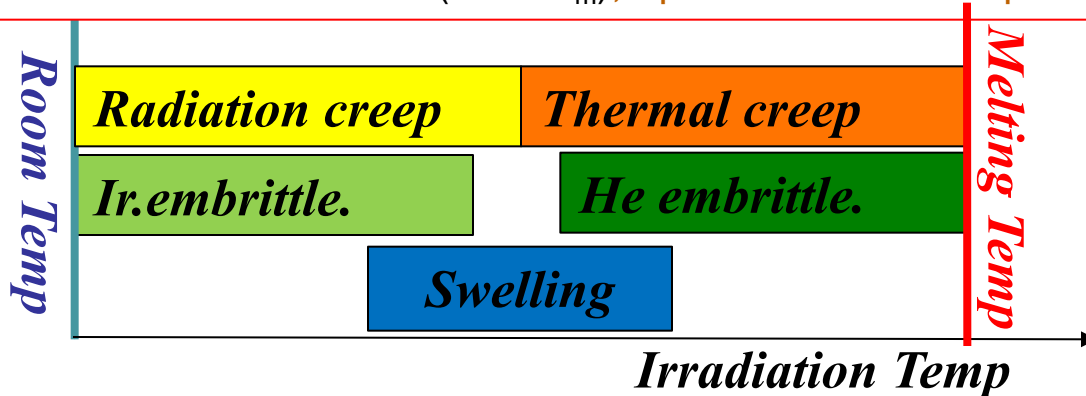
Dislocation-loops & network dislocations



Irradiation effects on structural materials

- Exposure to neutrons degrades the mechanical performance of structural materials and impacts the economics and safety of current & future fission power plants:

- **Irradiation hardening** and embrittlement/decreased uniform elongation ($< 0.4 T_m$)
- Irradiation ($< 0.45 T_m$) and thermal ($> \sim 0.45 T_m$) **creep**
- Volumetric swelling ($0.3 - 0.6 T_m$)
- High temperature **He embrittlement** ($> 0.5 T_m$); Specific to fusion & spallation accelerators



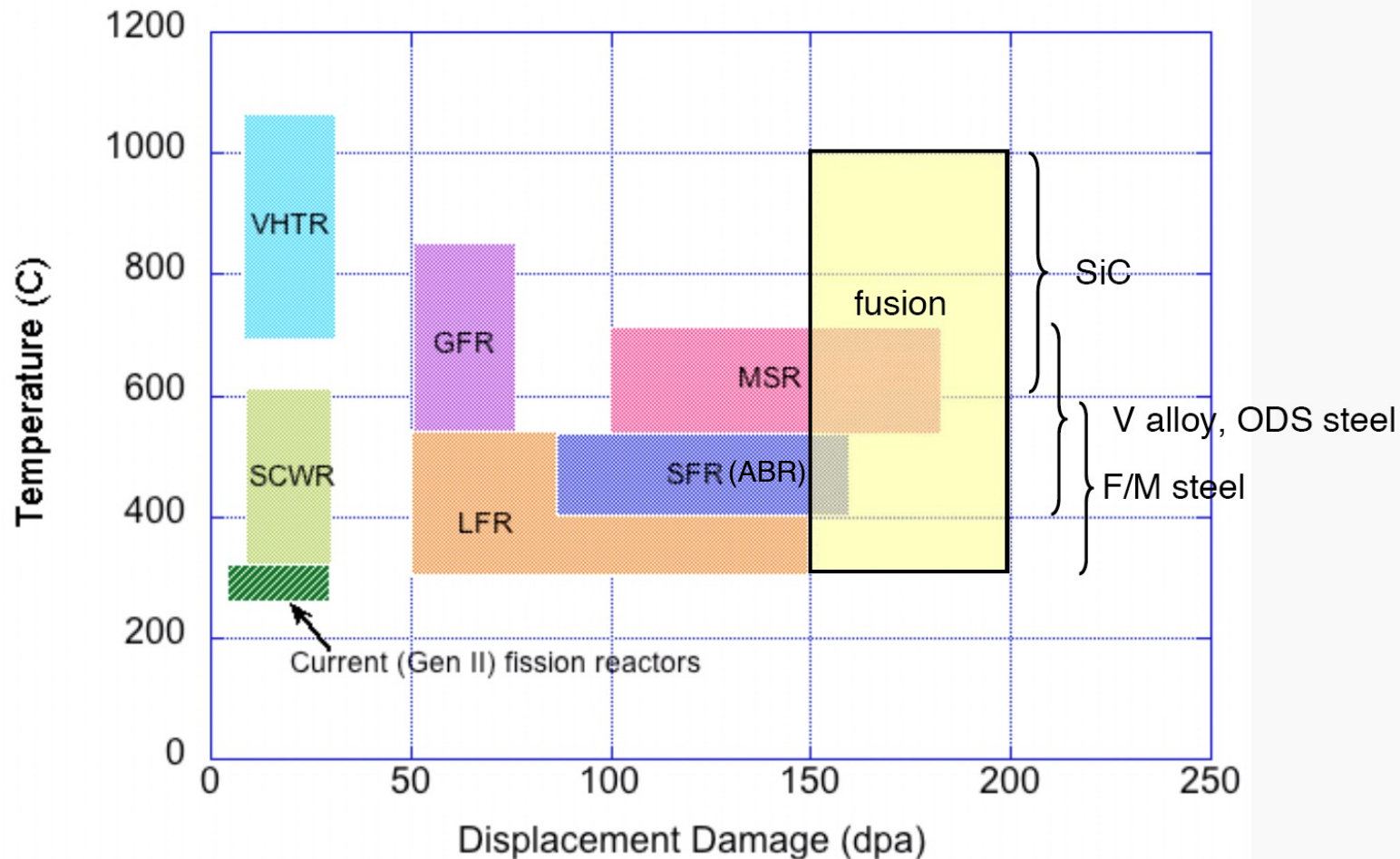
Variables

- Materials (Fe-based steels, Vanadium and Ni-based alloys, Refractory metals & alloys, SiC, ODS alloys)
- Initial microstructure (cold-worked, annealed)
- Irradiation temperature
- Chemical environment & thermal- mechanical loading
- Neutron flux, fluence and energy spectrum
 - materials test reactor irradiations typically at accelerations of $10^2 - 10^4$

Synergistic Interactions between all of them!

Extreme environments in Advanced Nuclear Energy Systems

*



Very High Temperature Reactor
Super-Critical Water Reactor
Gas-cooled Fast Reactor

Lead-cooled Fast Reactor
Sodium-cooled Fast Reactor
Molten Salt Reactor

Why is it so complicated to pick the right materials?

So many parameters!

Dose

Dose rate

Temperature

Environment (corrosion)

Operation time

cycling of temperature or pressure

Elements in the material

Radiation effects in materials:

Swelling

Design for Swelling resistance

What makes a materials Swelling resistant?

- **Low interaction with neutrons (low X-section)**
 - **Choose your alloying elements wisely.**
- **bcc metal (less bias)**
- **Lots of interfaces as defect sinks**
 - **Nanoparticles (Oxide Dispersed Strengthened)**
 - **High dislocations density -> cold work, martensite**
 - **High interface density → martensite**

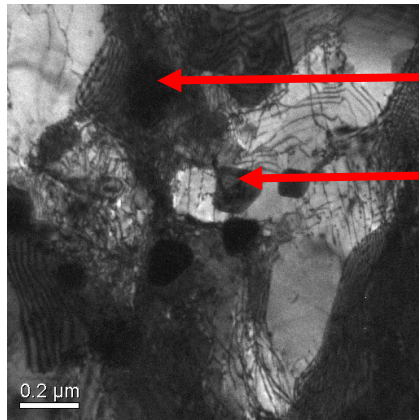
*Ferritic/
Martensitic
Materials*

*ODS
Materials*

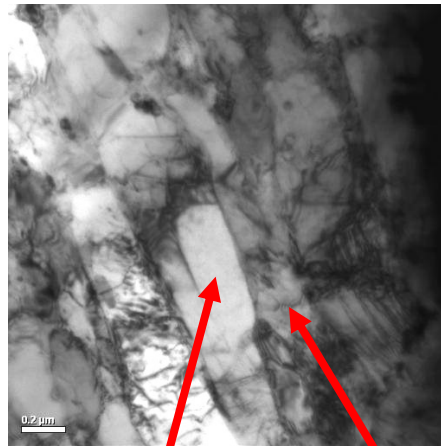
*Ferritic-martensitic steels: typical microstructure (T91)**

Fe	Cr	Mo	Mn	V	Nb	Ni	Si	Cu	C	P	Al	S	N
Bal	8.37	0.90	0.45	0.216	0.076	0.21	0.28	0.17	0.10	0.009	0.022	0.003	0.048

Solution Anneal: 1066 ° C: 46mins air cool, Tempering: 790 ° C: 42mins air cool

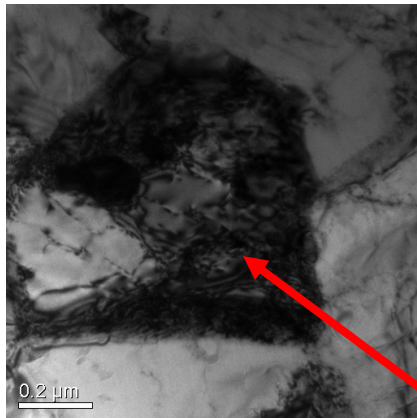


Dislocation tangles
precipitate



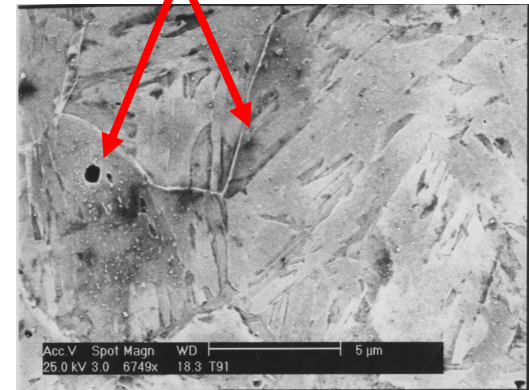
Lath

Subgrain

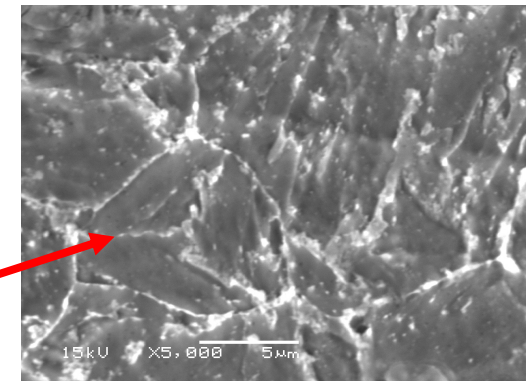


Dislocation cell

Prior austenite grain boundary (PAGB)



Packet boundary



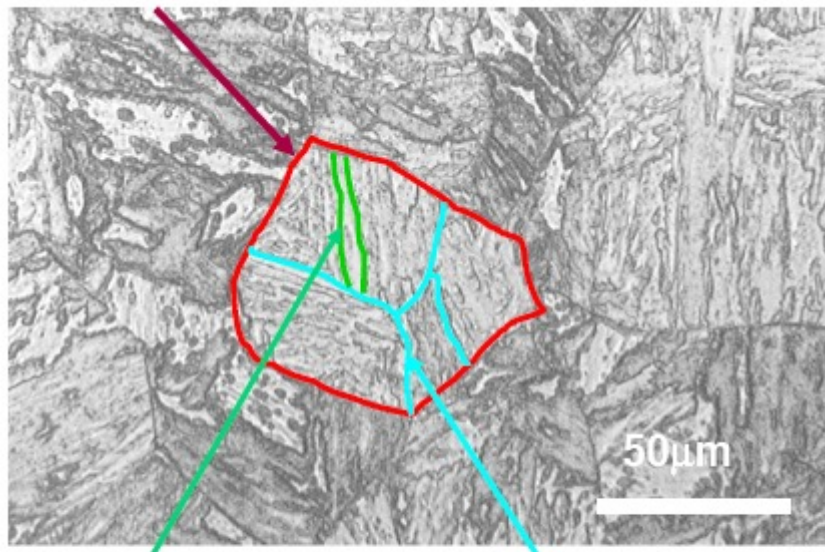
*Ferritic-martensitic steels: typical microstructure (F82H, Eurofer97)**

Fe – **8** ~ **9** Cr - **1~2** W - V, Ta (F82H, JLF-1, EUROFER 97, etc.)

Based on Mod9Cr (T91: Fe 9Cr-1Mo V,Nb)

Mo replaced by **W**, Nb replaced by **Ta**, and reduced N, to reduce activation

Prior Austenitic Grain boundary

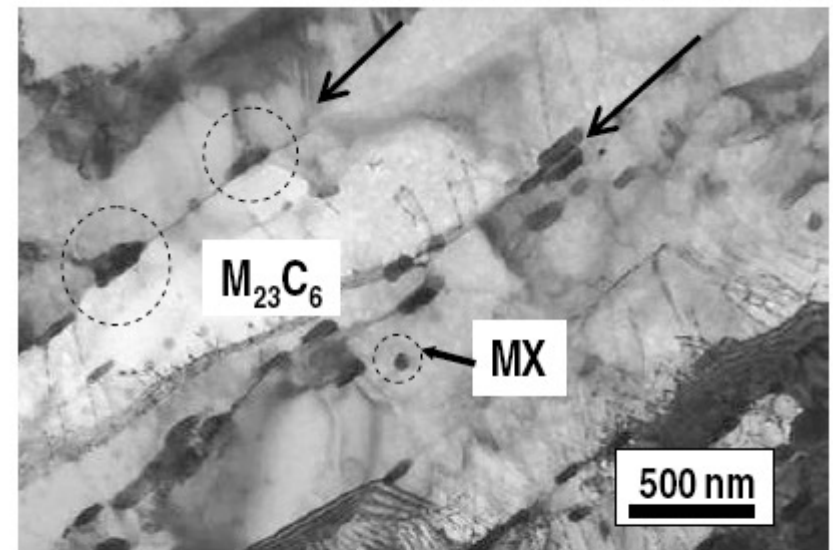


Packet boundary

Block boundary

- Fully tempered Martensitic steels
- Fine boundaries and precipitates make the steel highly irradiation and heat resistant.
- Ferrite phase formation should be avoided to have good performance.

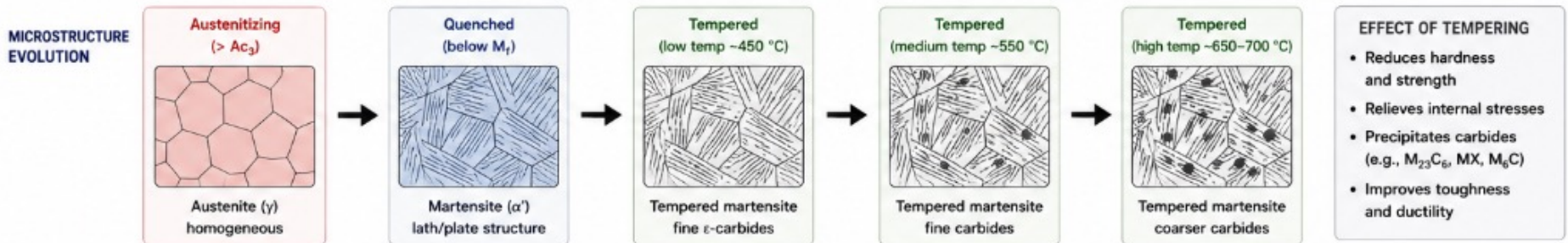
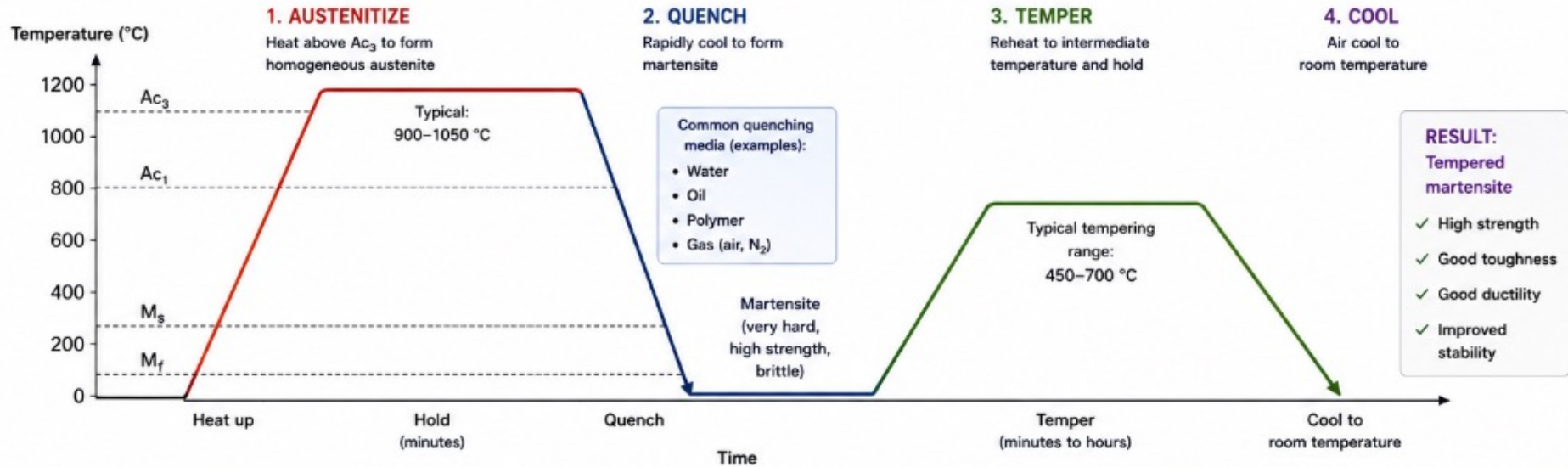
Martensite lath boundary



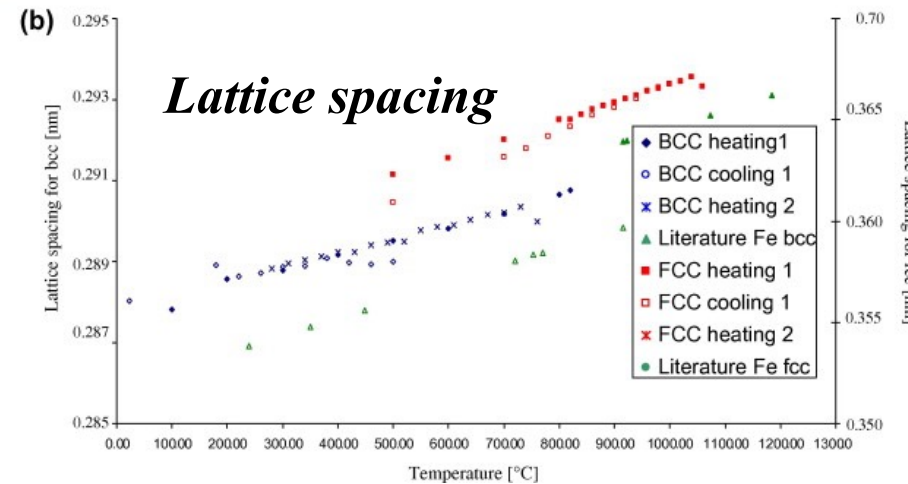
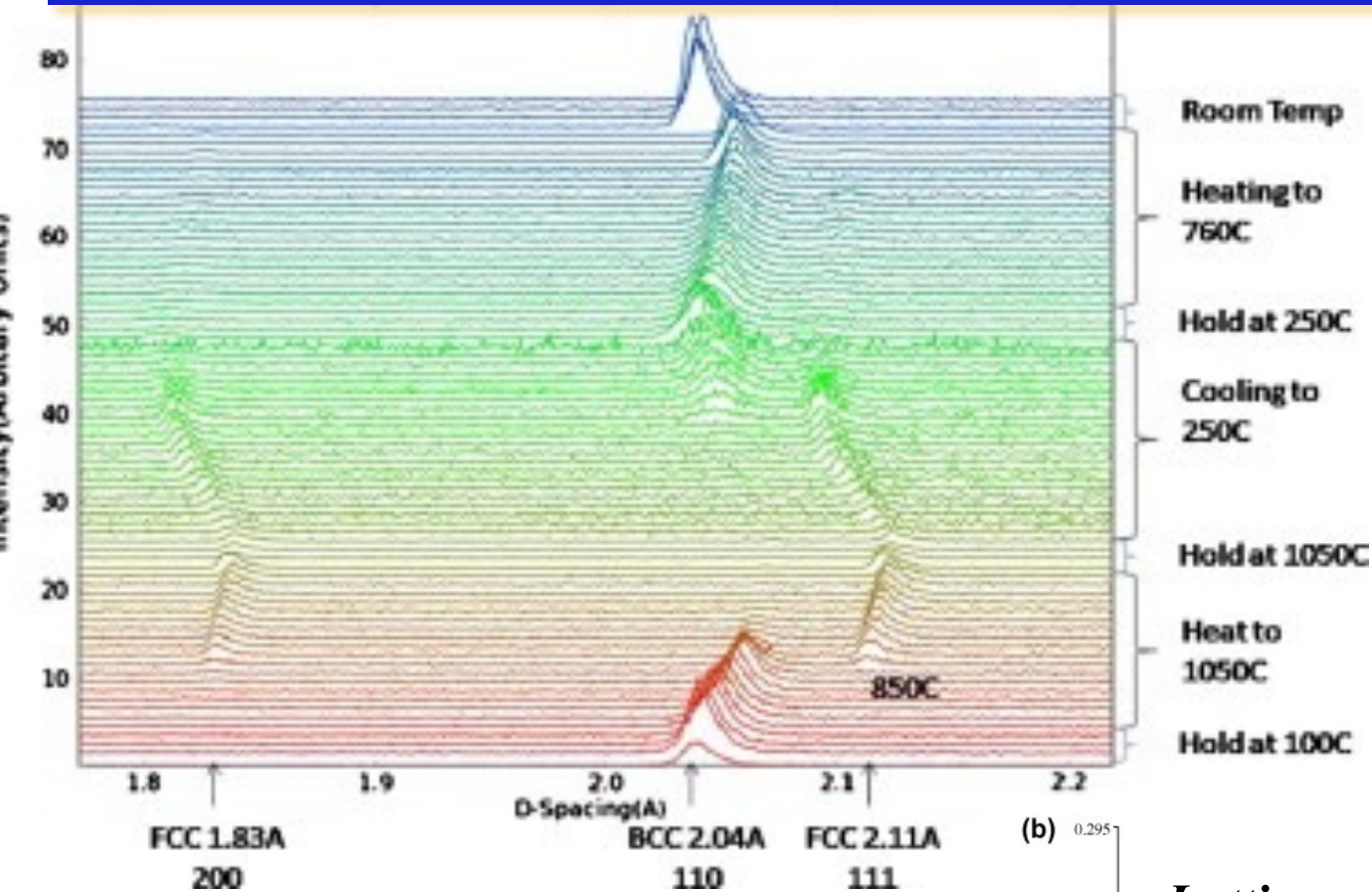
$M_{23}C_6$: Major precipitate, Cr rich.

MX : Ta, V rich precipitates

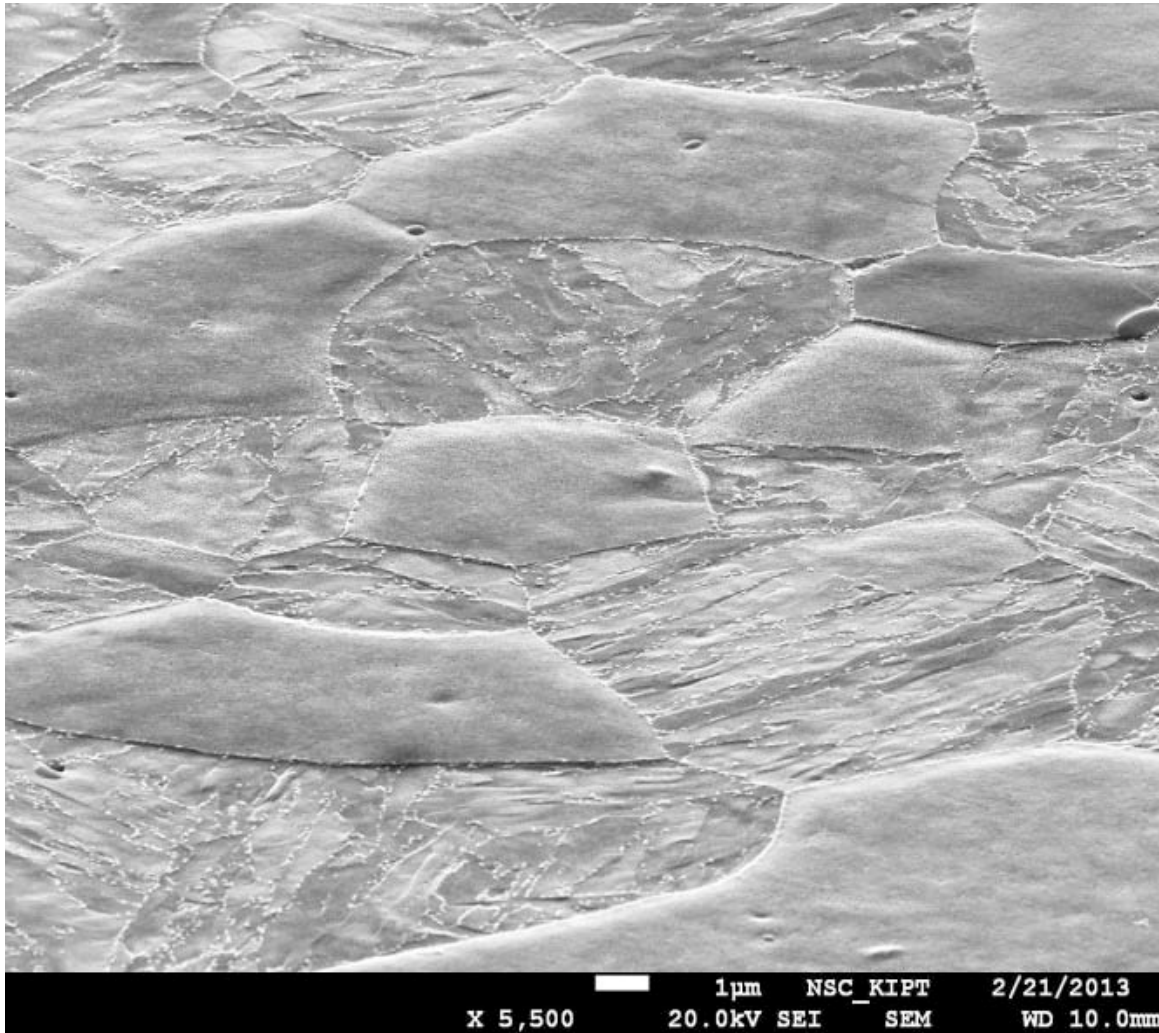
Ferritic-martensitic steels how is it formed?



In-situ diffraction experiment



Irradiation induced swelling effect of microstructure

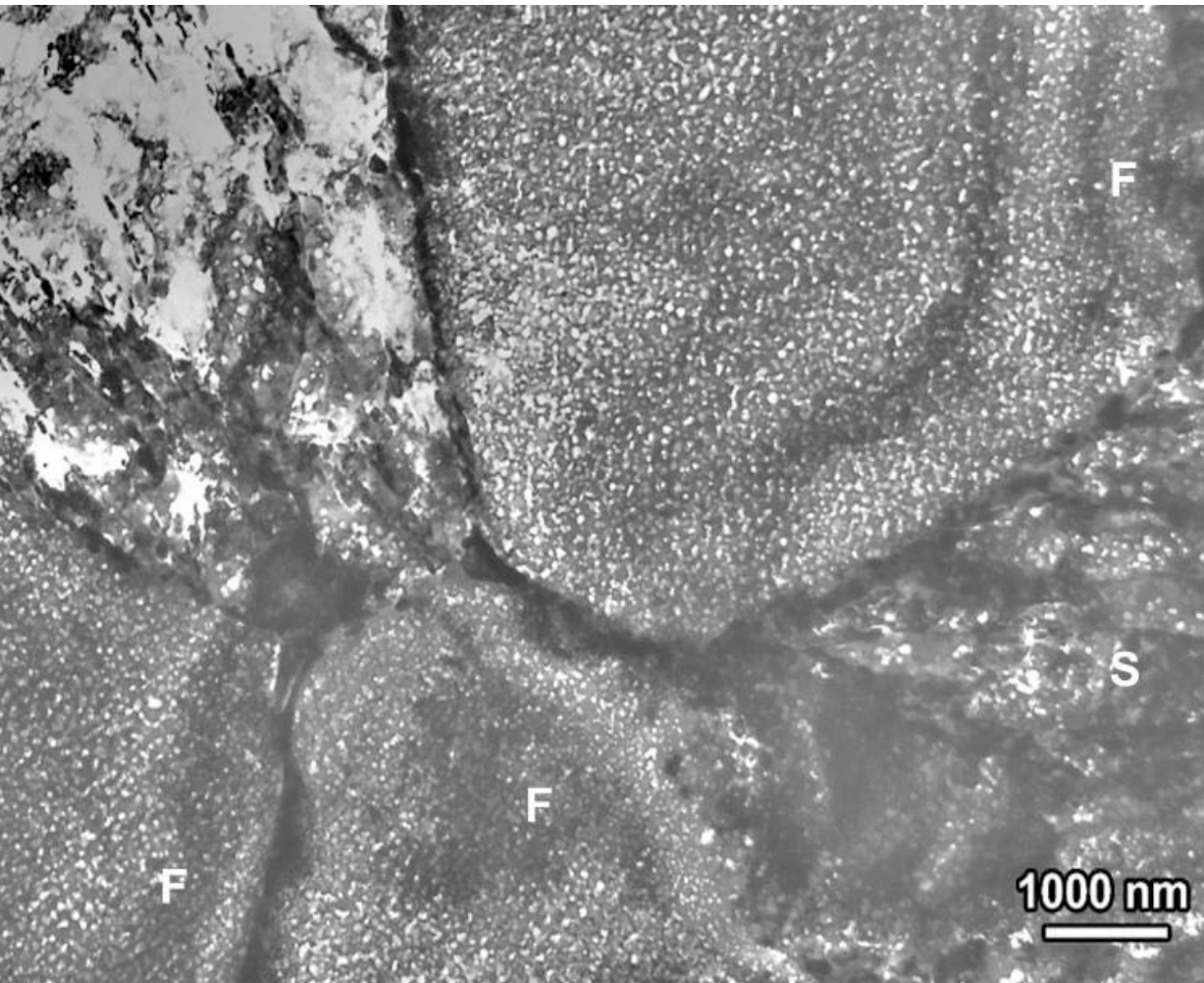


F. Garner 2015:

***Ferritic grains swell (grow)
more than martensitic
grains!***

***Ferritic (bcc) have a smaller
capture radius for vacancies
See NE220***

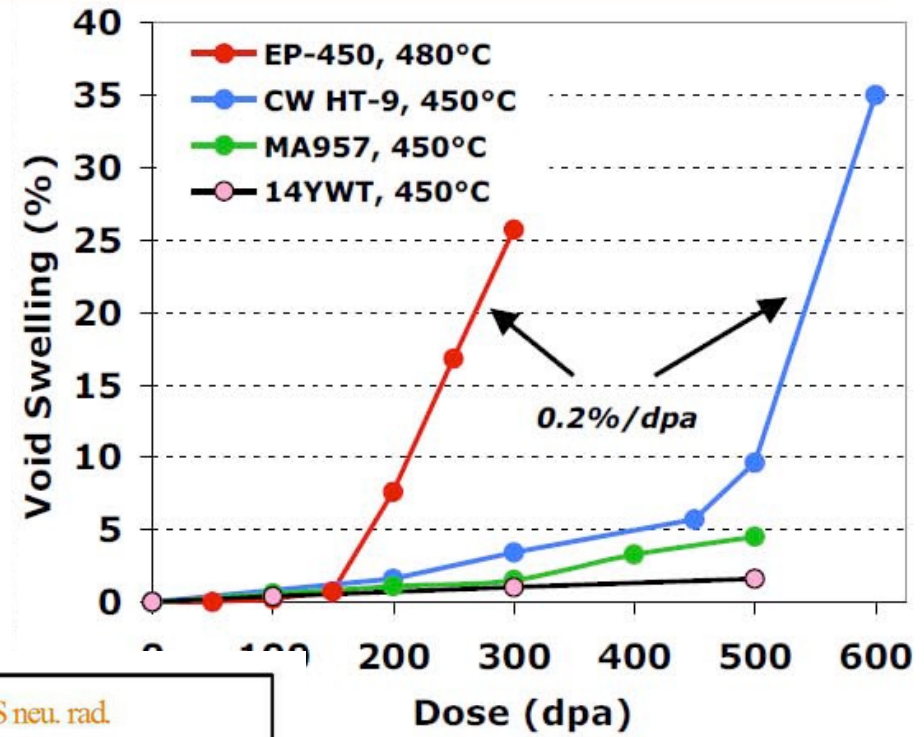
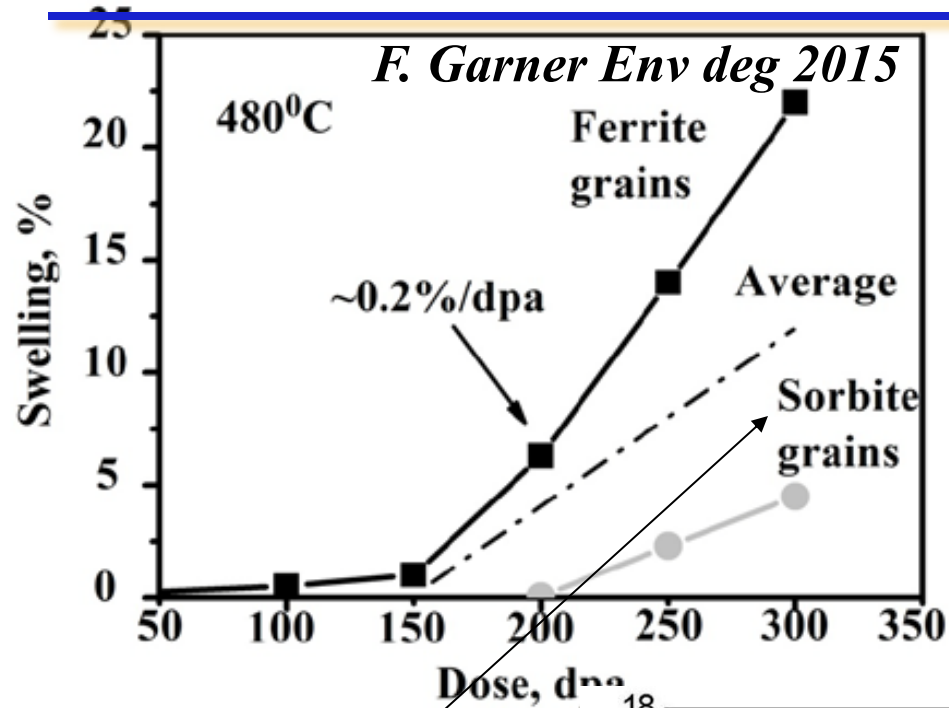
Irradiation induced swelling effect of microstructure



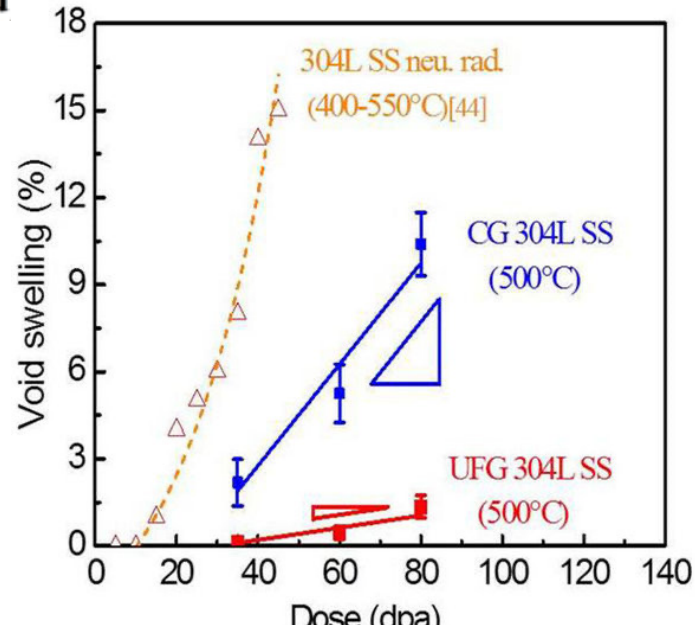
F. Garner 2015:

*Ferritic grains swell
(grow) more than
martensitic grains!*

Irradiation induced swelling effect of microstructure



*Tempered
Martensite*

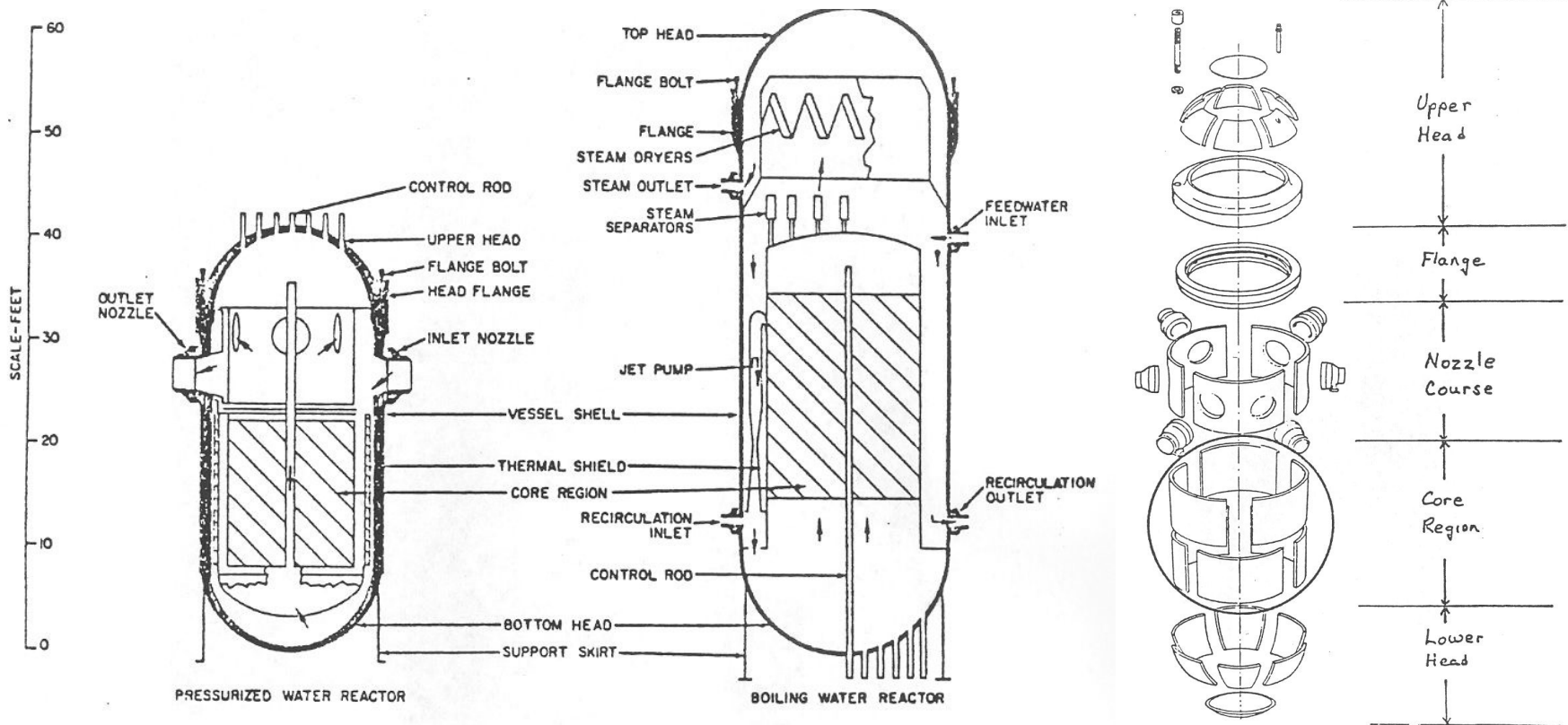


C. Sun Nature reports 2015

Rector pressure vessel

Reactor Pressure Vessel (RPV) embrittlement

- The RPV is perhaps the **most critical safety component** in a nuclear power plant. RPVs are fabricated from low alloy carbon steel (typically A302, A533B and A508-type). Early technology: plates rolled into appropriate shapes and welded together into an integral vessel. The head is bolted to the vessel and can be periodically removed (e.g., for re-fueling). More recently, rings have been fabricated from single ingot forgings, which eliminates the longitudinal weld.



Forging of large vessels

Older RPV's have been extensively welded.

However, today forging is preferred.



Forging of large vessels

Reactor pressure vessel components are among the largest forged components made today!

Plants today: Japan, China, France Russia Korea

The largest forgings can be 650t with a 14000t press.



Second largest press 13.000t at Seah Besteel.

RPV steels and alloys

Steel Identification	Composition (wt.%) ^a								
	C max.	Mn	P max.	S ^u max.	Si	Mo	Ni	Cu	Other
PLATE									
SA302 Grade B Class 1	0.25	1.10/1.55	0.035	0.040	0.13/0.32	0.41/0.64	-		
SA533 Grade B Class 1	0.25	1.15/1.50	0.035	0.040	0.15/0.30	0.45/0.60	0.40/0.70		
SA533 Grade B Class 1 (Low Copper)	(Same but for P, Cu)		0.017 ^b 0.012 ^c					0.12 ^b 0.10 ^c	
SA537 Grade B Class 1	0.24	0.95/1.65	0.035	0.040	0.13/0.55	0.08	0.25 ^d	0.35	Cr-0.25
FORGING									
SA508 Class 2	0.27	0.50/0.90	0.025	0.025	0.15/0.35	0.45/0.60	0.50/0.90		Cr-0.25/ 0.45 V-0.05 max.
WELD - MANUAL									
E-7018 A1	0.12	0.90	0.03	0.04	0.80		0.40/0.65	-	
E-8018 C3	0.12	0.45/1.50	0.03	0.03	0.80	0.35	0.80/1.10	-	V-0.05
WELD SUBMERGED ARC									
High MnMoNiC	0.13	2.25	0.020	0.020	0.10	0.40	0.55	-	

^a Product analysis unless noted; maximum allowable analysis unless range given.

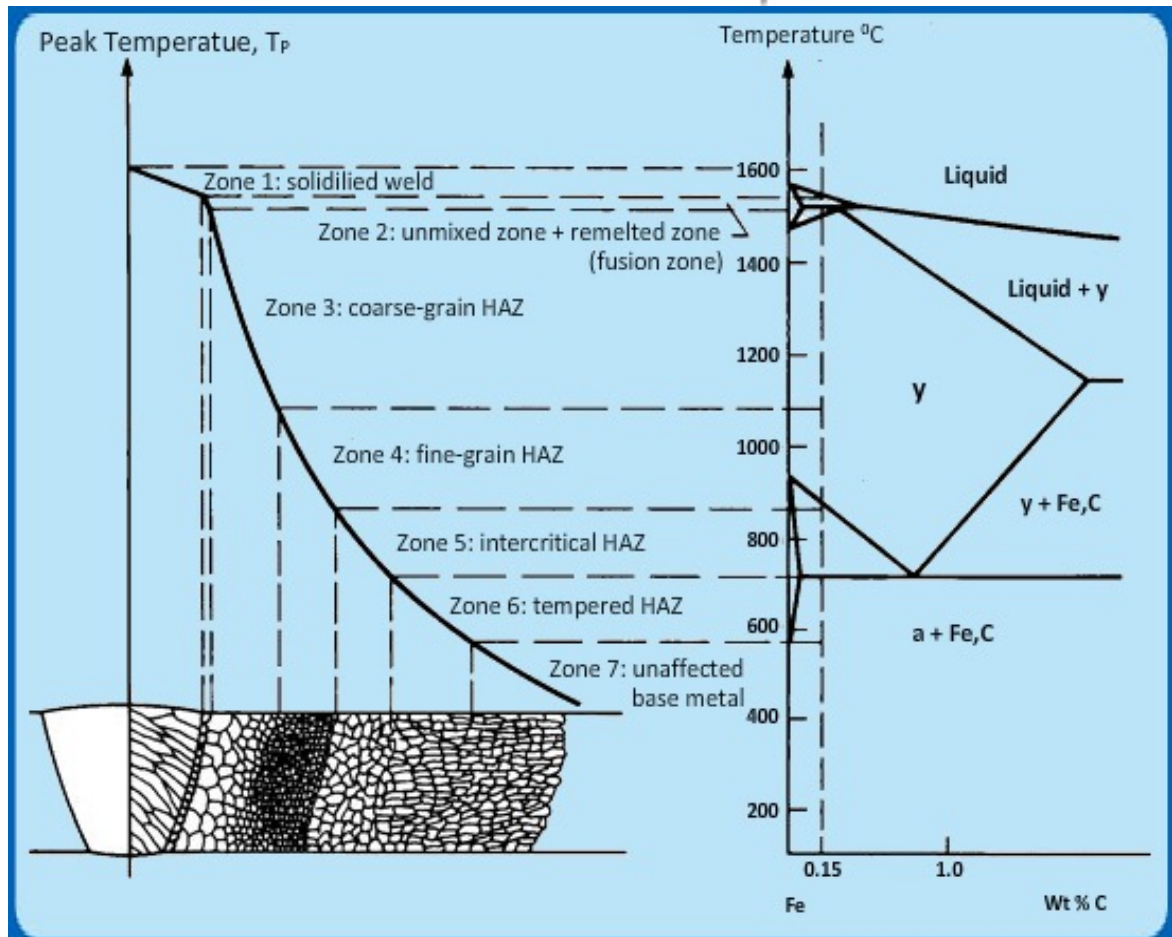
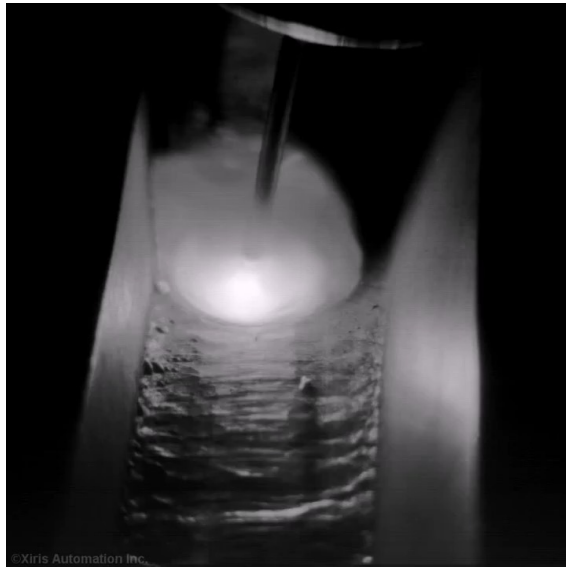
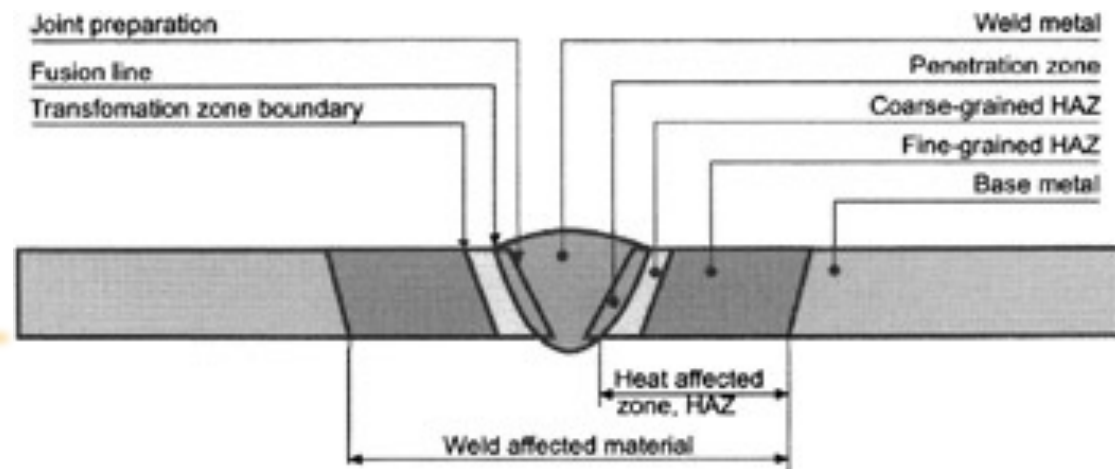
^b Maximum allowable product analysis.

^c Maximum allowable heat analysis.

^d Typical filler analysis, usually specified by vessel fabricator.

These steels have specified yield strengths in the range of 220 - 350 MPa, ultimate tensile strengths in the range of 350 - 450 MPa and elongations in the range of 18-25%.

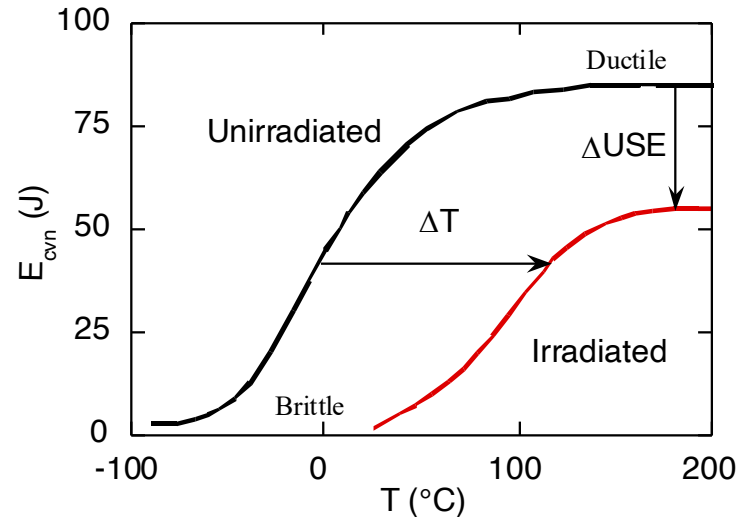
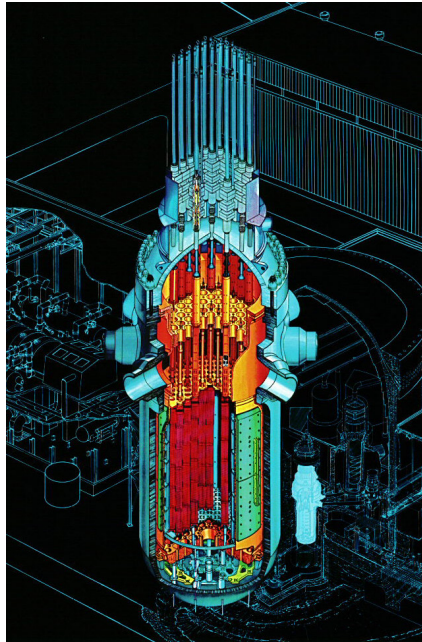
Welding an important aspect



RPV embrittlement

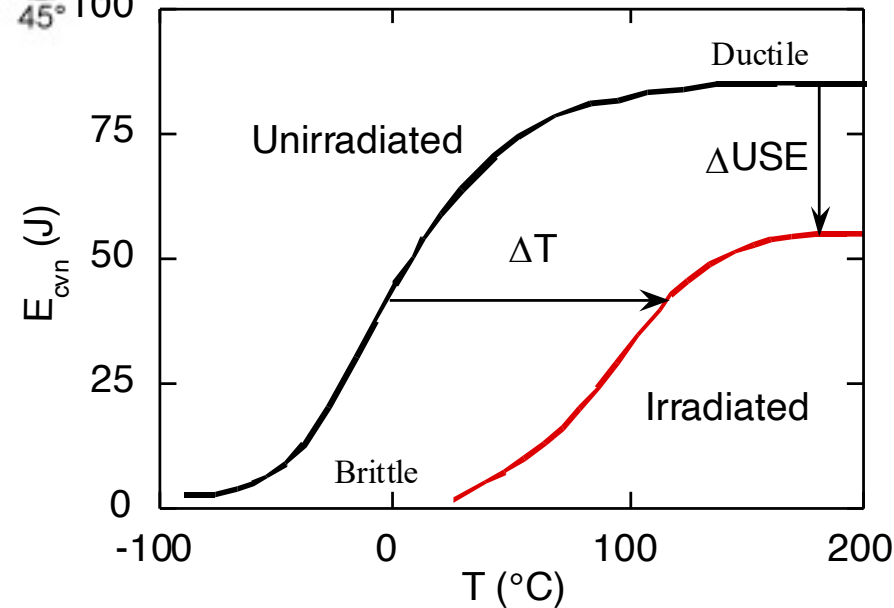
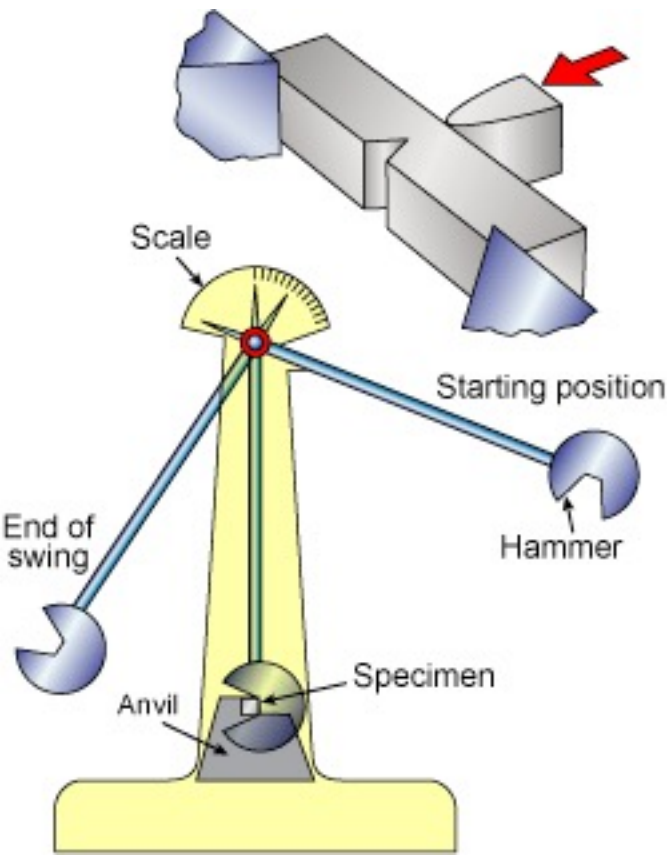
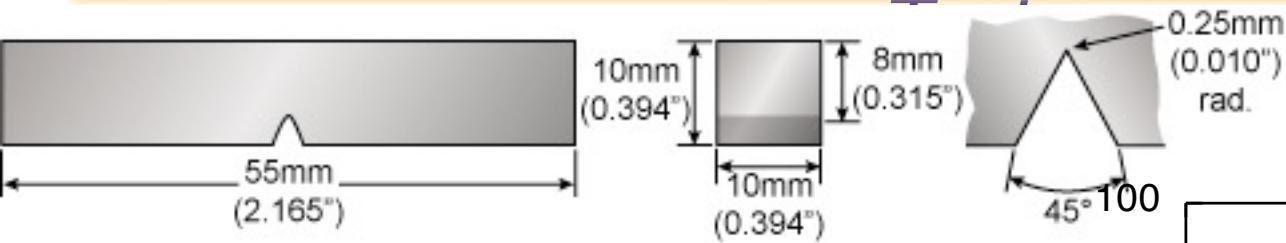
RPV embrittlement

- Reactor pressure vessel (RPV) embrittled by exposure to neutrons, manifested by transition temperature increases (ΔT) and upper shelf energy decreases (ΔUSE)



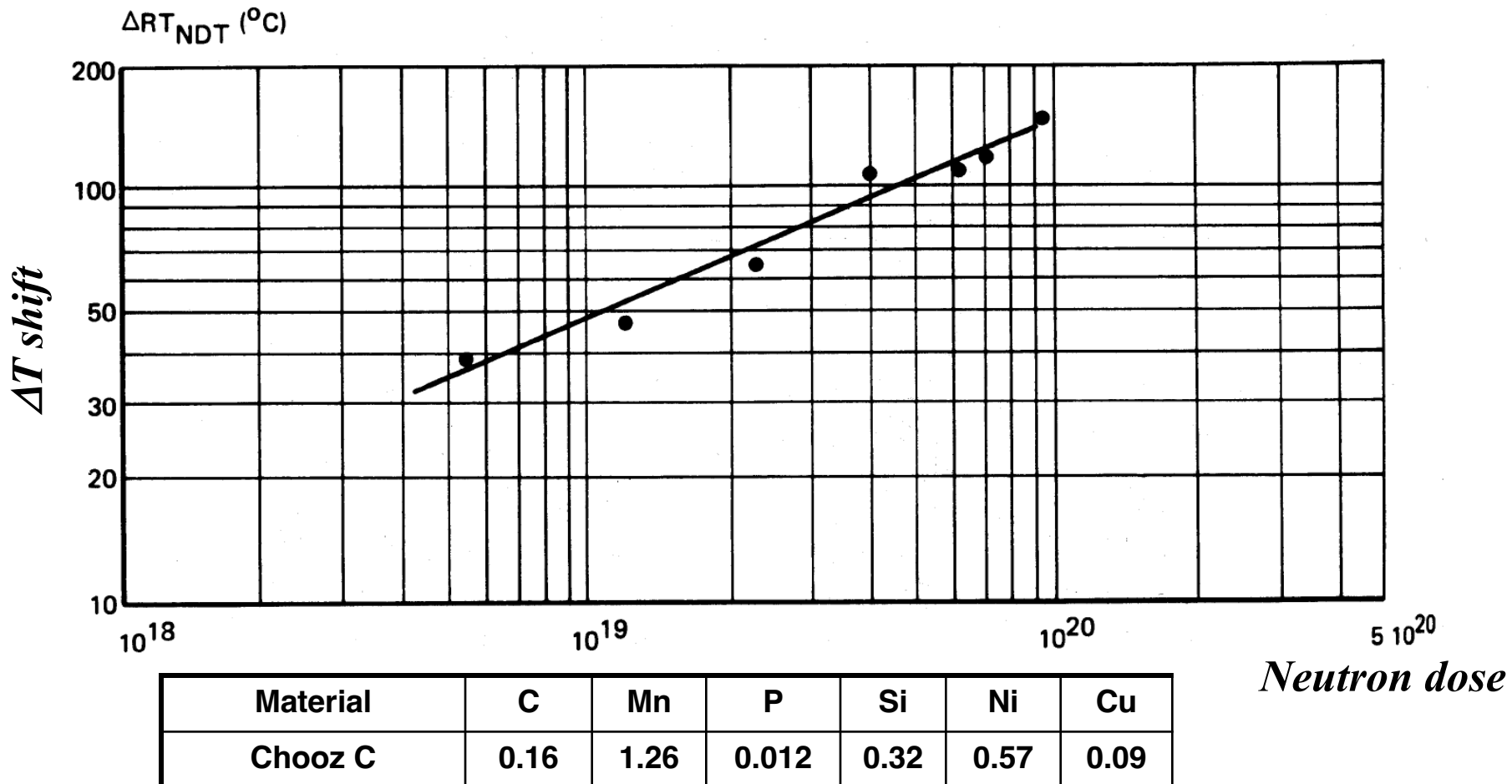
- Early recognition of possible material degradation by regulators and industry led to RPV surveillance programs. Most reactors include capsules with representative RPV steels located on inside of RPV (higher dose rate) to provide early embrittlement estimates and database.

How to test for embrittlement: Charpy V-Notch



Measured embrittlement

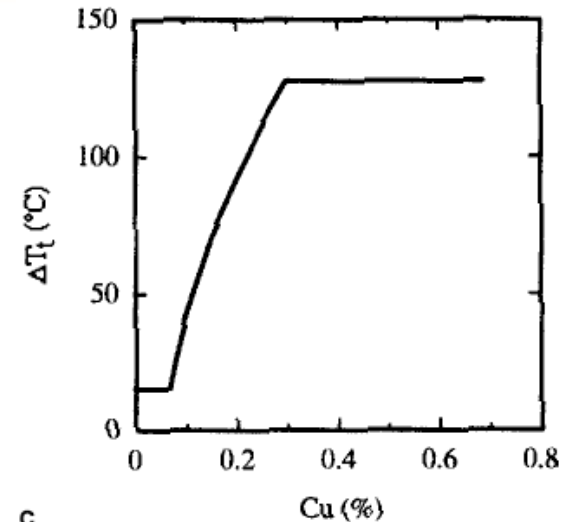
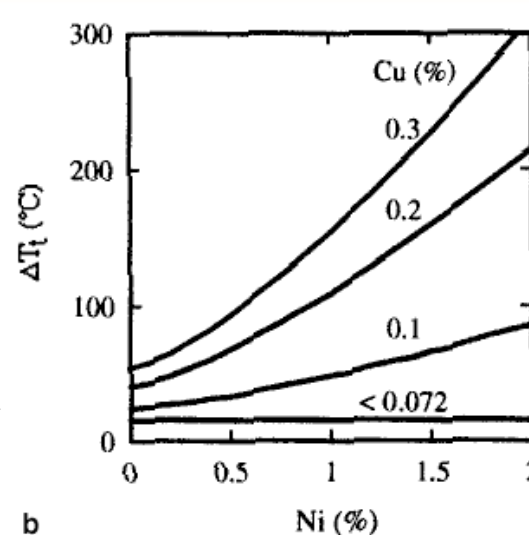
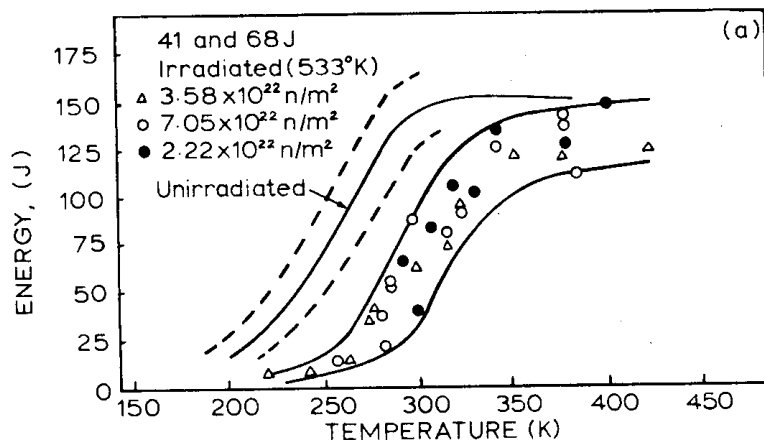
Surveillance data for RPV embrittlement of the CHOOZ A PWR in France –
Surveillance material is from the Chooz pressure vessel C material*



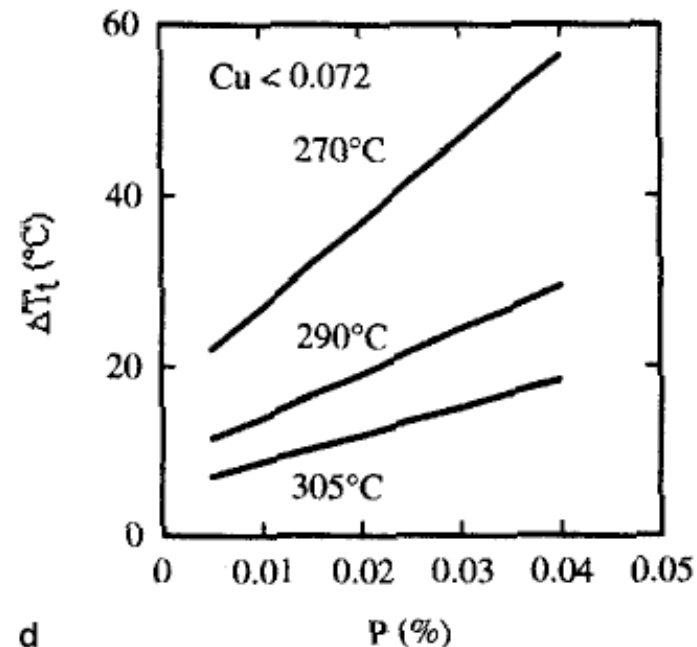
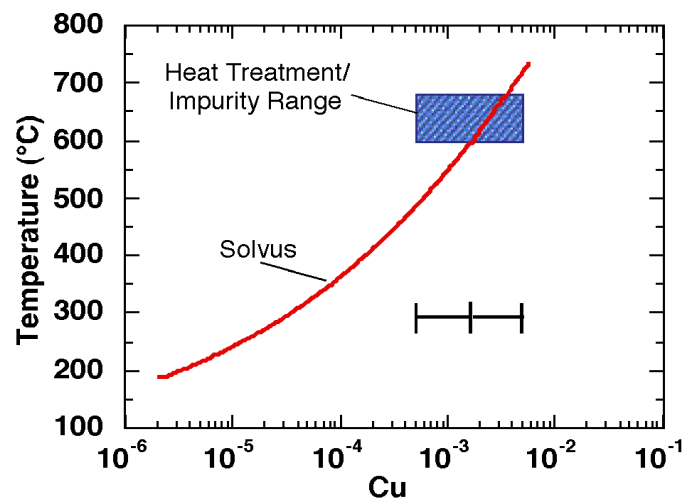
What is the underlying cause? What will happen at higher fluence?

* J.C. Van Duysen, J. Bourgoin, C. Janot, and J.M. Pennisson, **ASTM STP 1125** (1992) 117-130.

Embrittlement Variable Trends



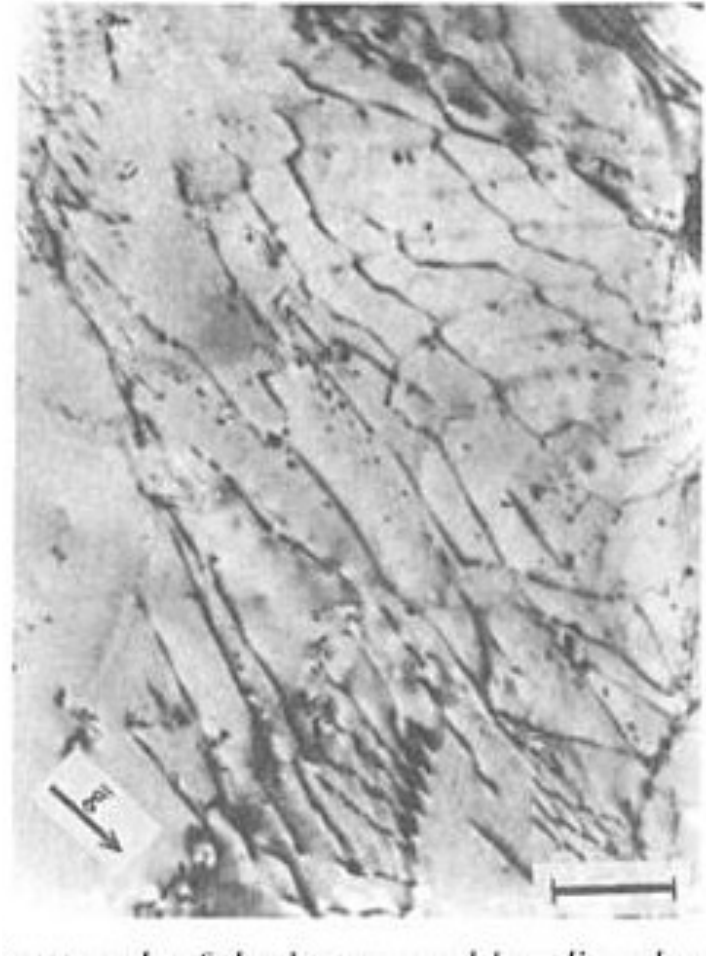
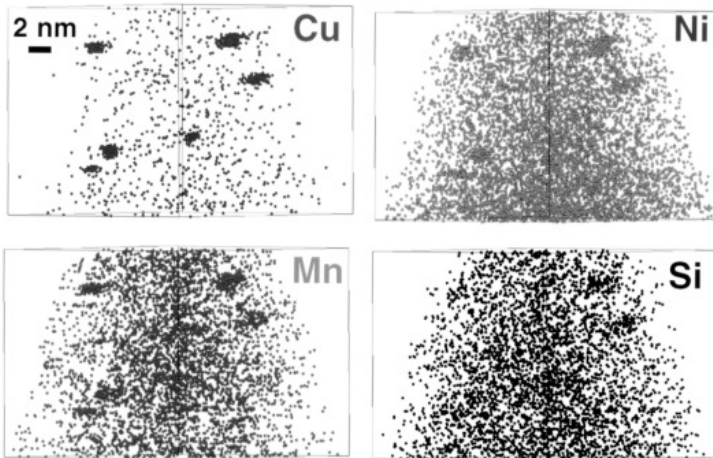
Minor alloying impurities have a large effect on the DBTT



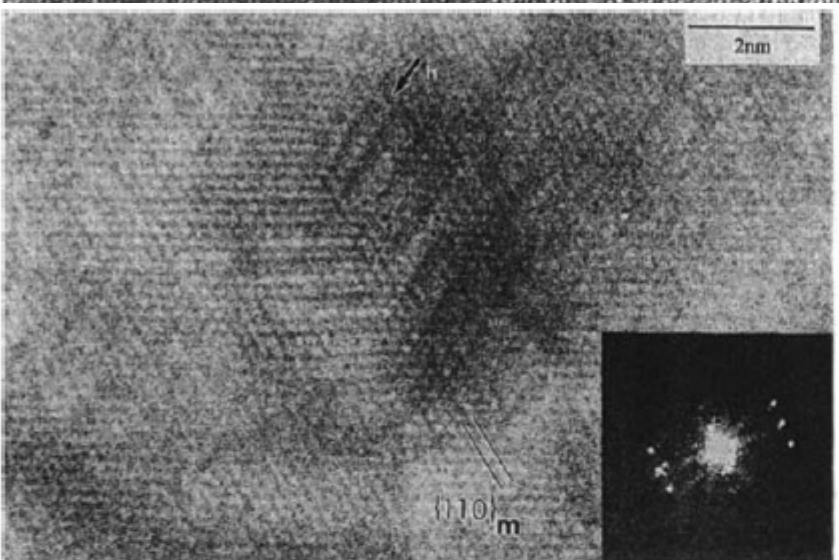
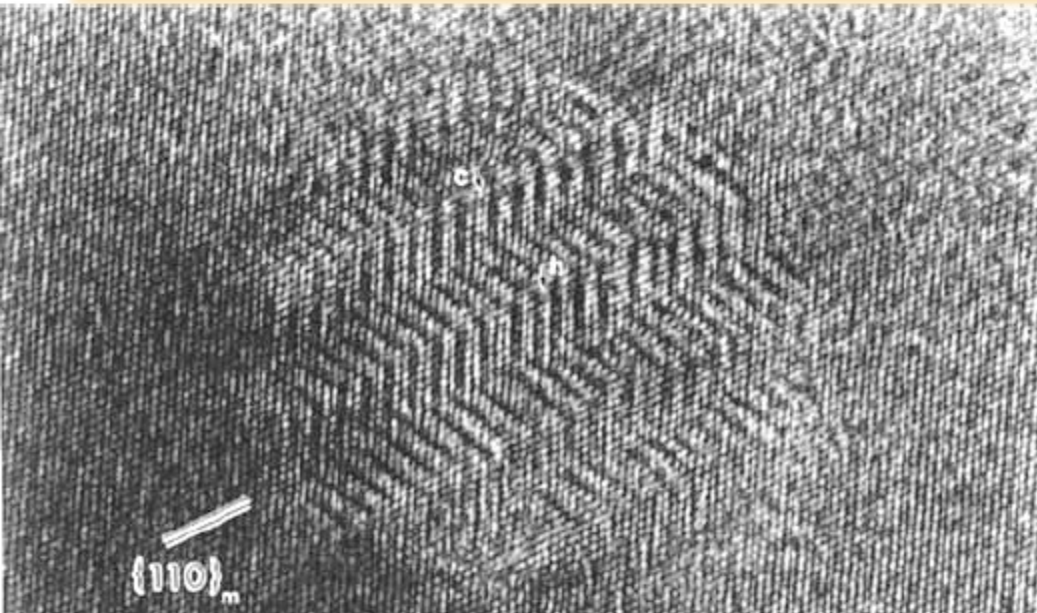
Defects in RPV materials

Material	C	Mn	P	Si	Ni	Cu
Chooz C	0.16	1.26	0.012	0.32	0.57	0.09

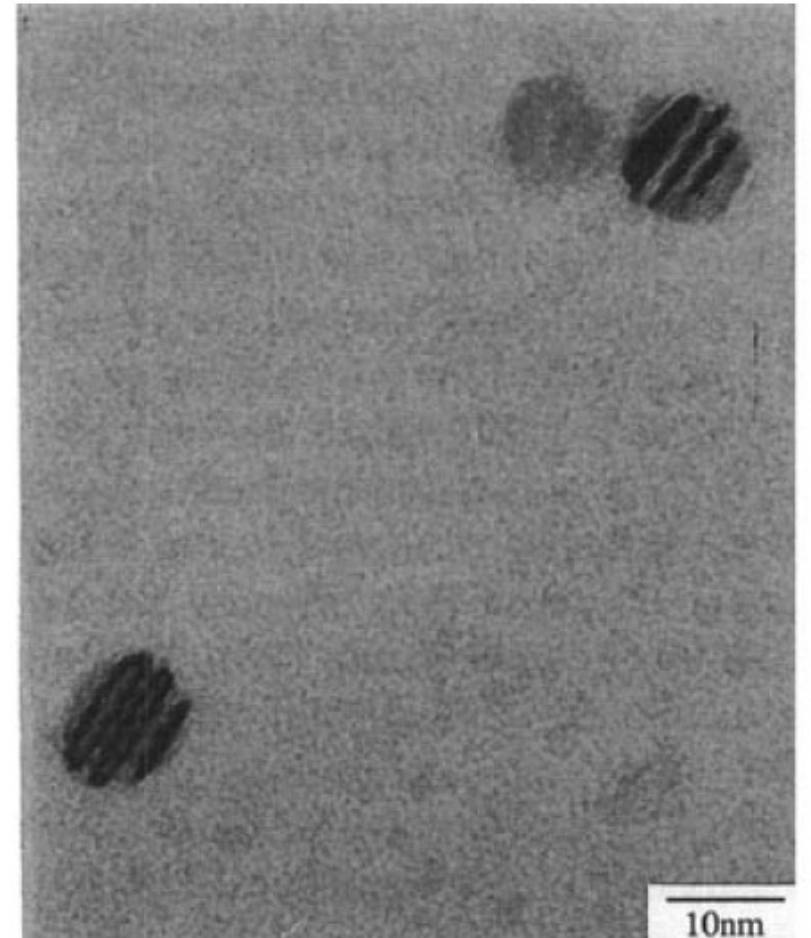
irradiated microstructure observed in the TEM*



HR images of Cu in Fe



(a)



P. J. Othen, M. L. Jenkins & G. D. W. Smith (1994): High-resolution electron microscopy studies of the structure of Cu precipitates in α -Fe, Philosophical Magazine A, 70:1, 1-24

Lifetime of RPV assessment

Pressurized thermal shock

- Brittle fracture is not a problem during normal operation because the temperature is high ($\sim 300^\circ \text{C}$) and the main stress on the vessel is due only to residual stresses and the membrane (hoop) stress due to the internal pressure in the vessel.
- If cold water from the emergency core cooling system (ECCS) should be injected into the system (because of a LOCA) near the end of the vessel lifetime when K_{Ic} is lowest due to irradiation embrittlement, brittle failure is possible.
- The rapid reduction in the inner wall temperature during ECCS creates tensile thermal stresses which add to existing stresses.
- Small flaws (cracks) on the inner wall of the vessel may have grown throughout the reactor lifetime due to fatigue during shutdown and startups.
- If the increased crack size and the corresponding increase in the applied stress intensity factor (K_I) exceeds K_{Ic} , the crack will grow rapidly.
- The crack may not penetrate the entire vessel wall because temperature, stress and irradiation dose change through the wall thickness. The crack will stop when the applied K_I reaches the crack arrest toughness K_{Ia} .

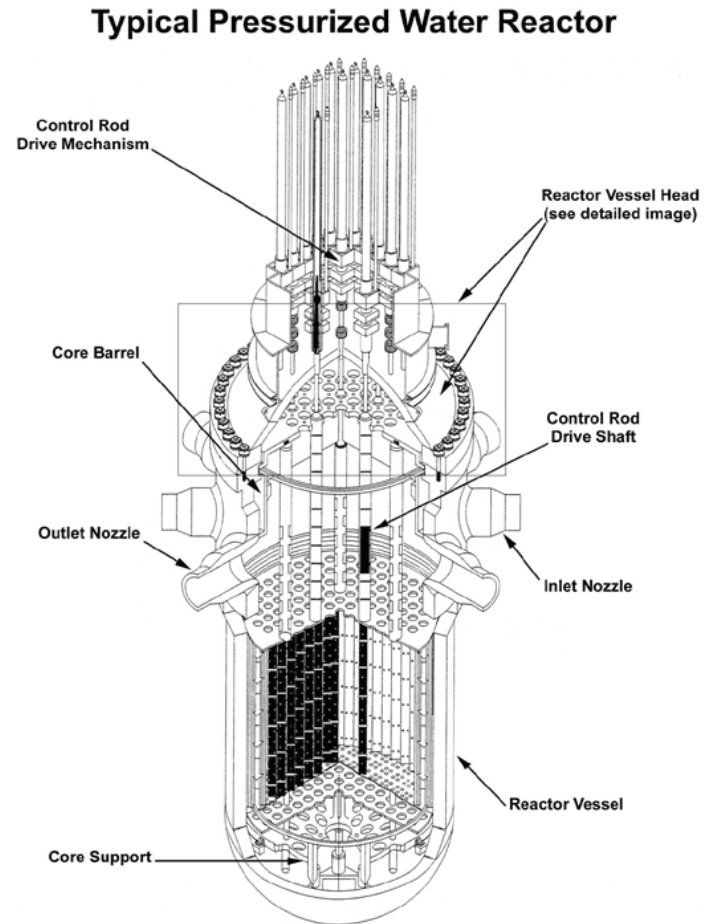
Remedies for RPV embrittlement and PTS

- “**New**” **pressure vessel**: eliminate impurities (Cu, P, high Ni/Mn) which enhance embrittlement, especially in weld material
- **Old vessels**:
 - a) **anneal critical zones** to remove radiation embrittlement; nozzle welds and beltline welds - annealing at $\sim 450^{\circ}$ C for 1 week (extensively in Russia, not in US)
 - b) **change fuel management** strategy (go to low leakage core):
 - c) replace steel reflector elements at core periphery with materials which can more effectively absorb fast neutrons (tungsten and titanium hydride)
 - d) heat the ECCS water to avoid/reduce thermal shock
 - e) demonstrate conservatism in the LEFM analysis

Davis Besse incident

Reactor vessel head corrosion

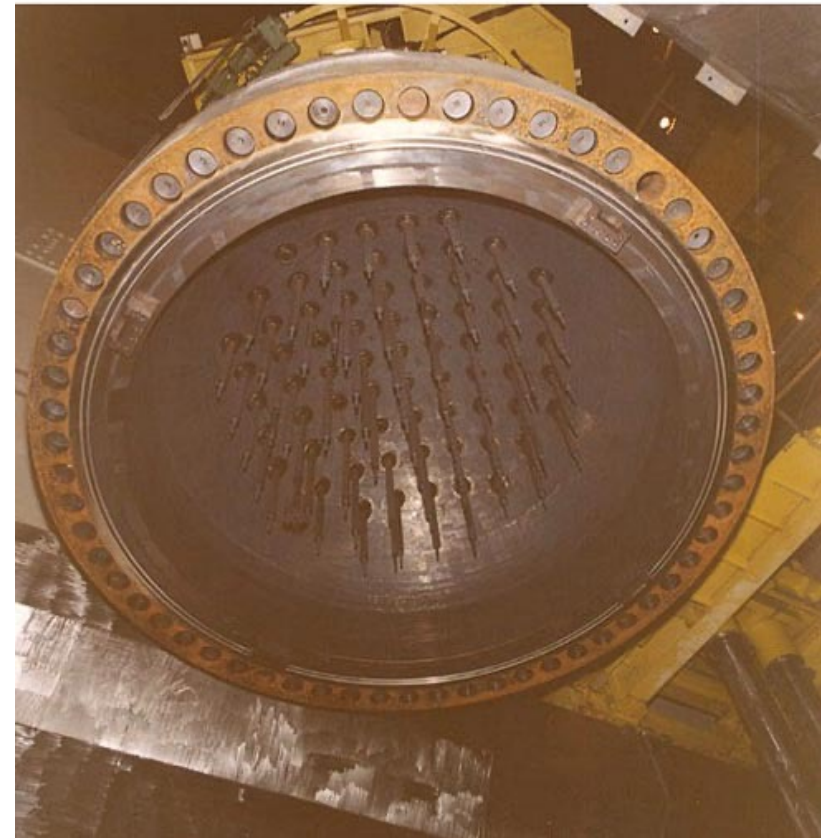
- **Based on experience from French reactors, it was learned that axial cracks could form (likely stress corrosion cracking) in Control Rod Drive Mechanism penetration at the top (& bottom) of RPV**
- **U.S. NRC issued Bulletin 2001-01 “Circumferential cracking of reactor pressure vessel head penetration nozzles” (Aug 2001).**
 - **Inspections in fall 2001 revealed vessel head penetration nozzle cracks at Three Mile Island Unit 1, Crystal River Unit 3, Nort Anna Unit 1, and Oconee Unit 3.**
- **August 2002, the NRC issued Bulletin 2002-02, “Reactor Pressure Vessel Head and Vessel Head Penetration Nozzle Inspection Programs”**
- **Inspections at several PWRs in 2002, including Davis-Besse, found leakage and cracks that have required repairs or replacement**
- **NRC Order on Feb 11, 2003 requires specific inspections**



Reactor vessel head corrosion (Davis-Besse)

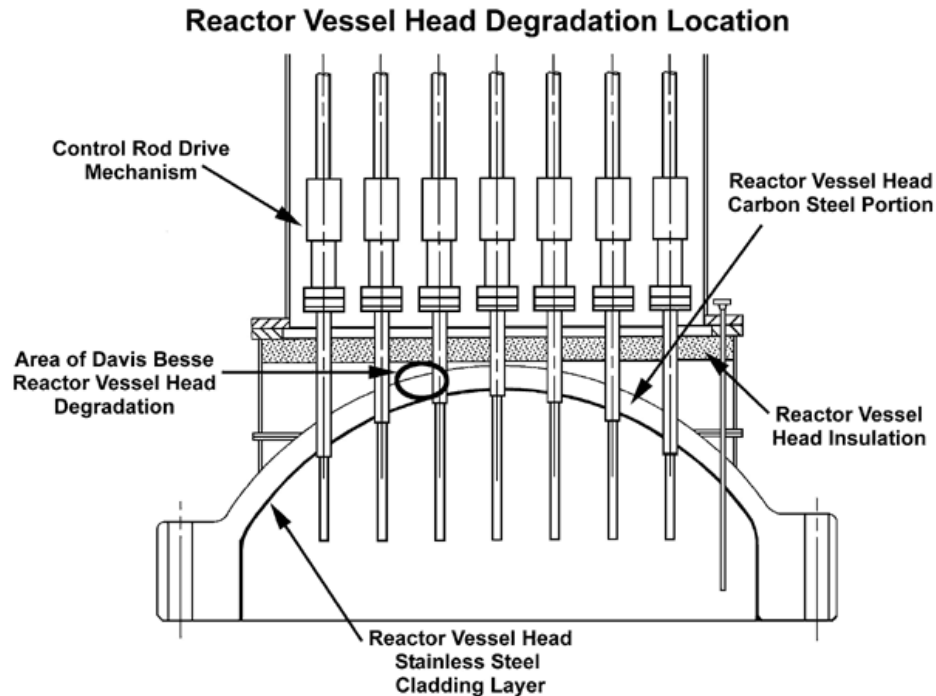
- **Leaking of primary coolant and evidence of boric acid (H_3BO_3) outside Control Rod Drive Mechanisms (Reactor Pressure Vessel Head) at Davis-Besse indicated a cracking problem as early as fall 2001**
- **Davis-Besse management (First Energy Nuclear Corporation) continued to operate until forced by NRC order in Feb 2002 to specifically investigate the vessel head and associated penetrations.**

Davis-Besse Reactor Vessel Head

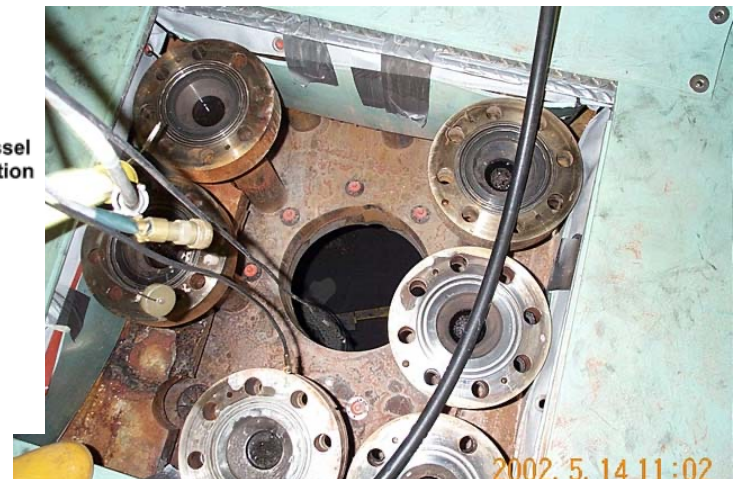


Reactor vessel head corrosion (Davis-Besse, 3/02)

- **Leaking of primary coolant and evidence of boric acid outside Control Rod Drive Mechanisms (Reactor Pressure Vessel Head) at Davis-Besse indicated a cracking problem as early as fall 2001**
- **Davis-Besse management (First Energy Nuclear Corporation) continued to operate until forced by NRC order in Feb 2002 to specifically investigate the vessel head and associated penetrations.**
- **Inspection at Davis-Besse revealed cracking in CRDM Nozzles 1, 2 and 3 - and, upon further investigation - significant corrosion of RPV head**



Top-view of Vessel Head, region around CRDM 3 (cutout)

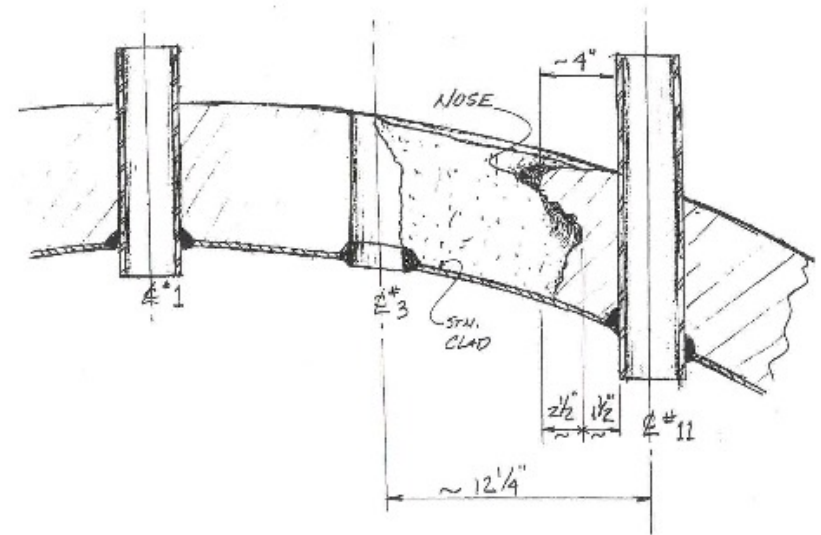


Reactor vessel head corrosion (Davis-Besse, 3/02)

“The wastage area was found to extend approximately 5 inches downhill on the RPV head from the penetration for CRDM nozzle 3 and was approximately 4 to 5 inches at its widest part. The minimum remaining thickness of the RPV head in the wastage area was found to be approximately 3/8 inch. This thickness was attributed to the thickness of the stainless steel cladding on the inside surface of the RPV head, which is nominally 3/8 inch thick.”



Davis Besse Reactor Vessel Head Degradation Head Cutaway View



The above figure shows the Davis Besse reactor vessel head degradation between nozzle #3 and nozzle #11. This sketch was provided to the NRC by the Licensee.

Reactor vessel head corrosion (Davis-Besse, 3/02)

*Further, the stainless steel cladding had significantly bowed upward and cracked, indicating substantial plastic deformation - **the cladding, not designed as a pressure containment structure is all that prevented a severe loss of coolant accident***



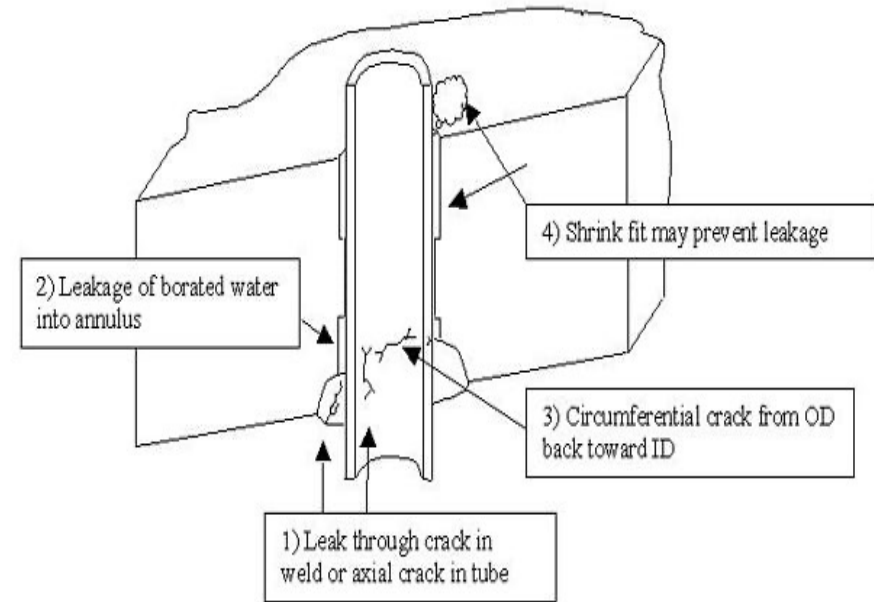
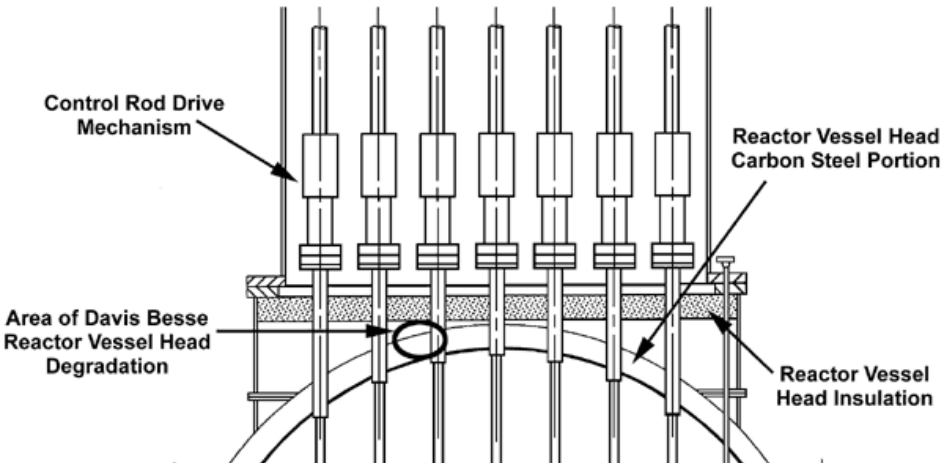
“The investigation of the causative conditions surrounding the degradation of the RPV head at Davis-Besse is continuing. Boric acid or other contaminants could be contributing factors. Other factors contributing to the degradation might include the environment of the RPV head during both operating and shutdown conditions (e.g., wet/dry), the duration for which the RPV head is exposed to boric acid, and the source of the boric acid (e.g., leakage from the CRDM nozzle or from sources above the RPV head such as CRDM flanges).”



What seems to have happened?

Typical Pressurized Water Reactor

Reactor Vessel Head Degradation Location





Top View - Nozzle #3 Removed

Davis Besse
Control Rod Drive Mechanism
Access Hatch Open



Nozzle #3
Removed

Reactor
Vessel
Head
Nozzle
to CRDM
Flange

Control Rod
Drive
Mechanism
in Place





Outcome:

First Energy Corporation pay fines of \$23.7 million, with an additional \$4.3 million to be contributed to environmental agencies.

Andrew Siemaszko 5 years in prison and 250.000\$ fine. Additional worker was convicted with three years probation