



# Overview of China Nuclear Waste Vitrification

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# About Wuhan and WHUT



## Wuhan city:

- 1) At the intersection of Yangtze River.
- 2) Largest city in Central China.
- 3) 3500-year history.

## WHUT:

- 1) Student: ~50,000;  
faculty: ~3,300.
- 2) Ranking in the world: ~400  
(THE Ranking).
- 3) Best University in glass  
and ceramics (in China).

## Bird view of WHUT

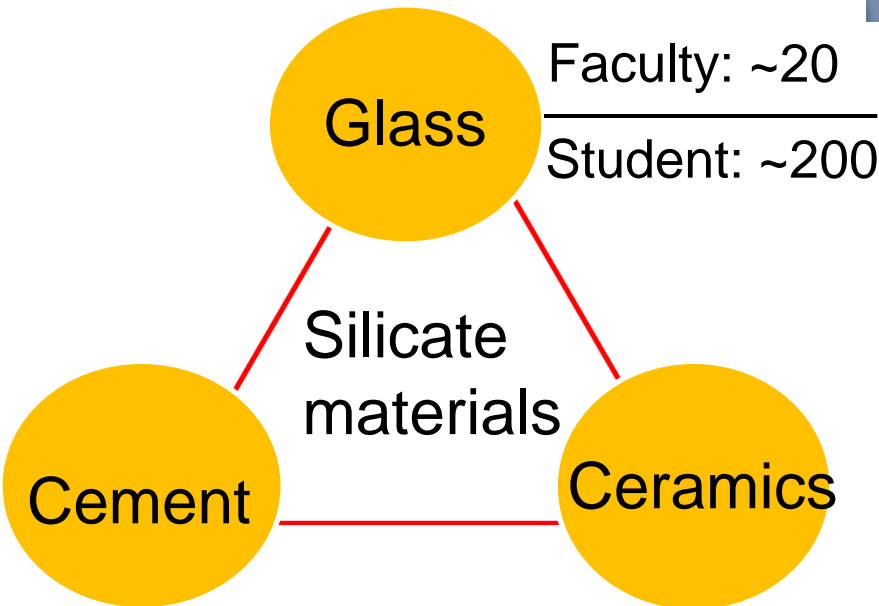


# About SMART Lab

## Faculty and student:

- Faculty: ~65.
- Post-doc.: ~10.
- Student: ~500,  
Intl. stud.: 15.

## Research areas:



Aalborg, Denmark



Prof. Yue



Prof. Zhao

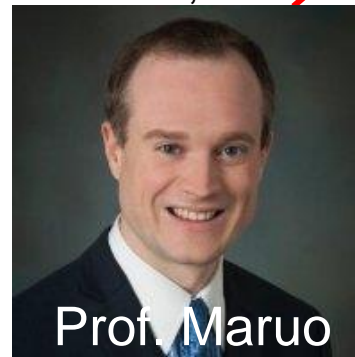
Ex-president of  
ICG



Prof. Peng

**World-renowned  
glass scientist  
@ SMART**

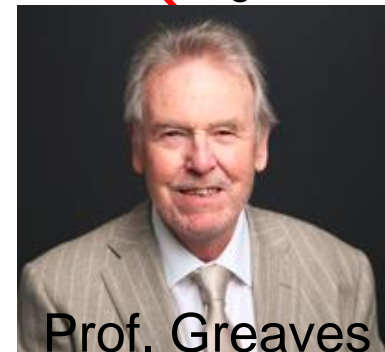
PSU, US



Prof. Maruo

Adjunct  
prof.

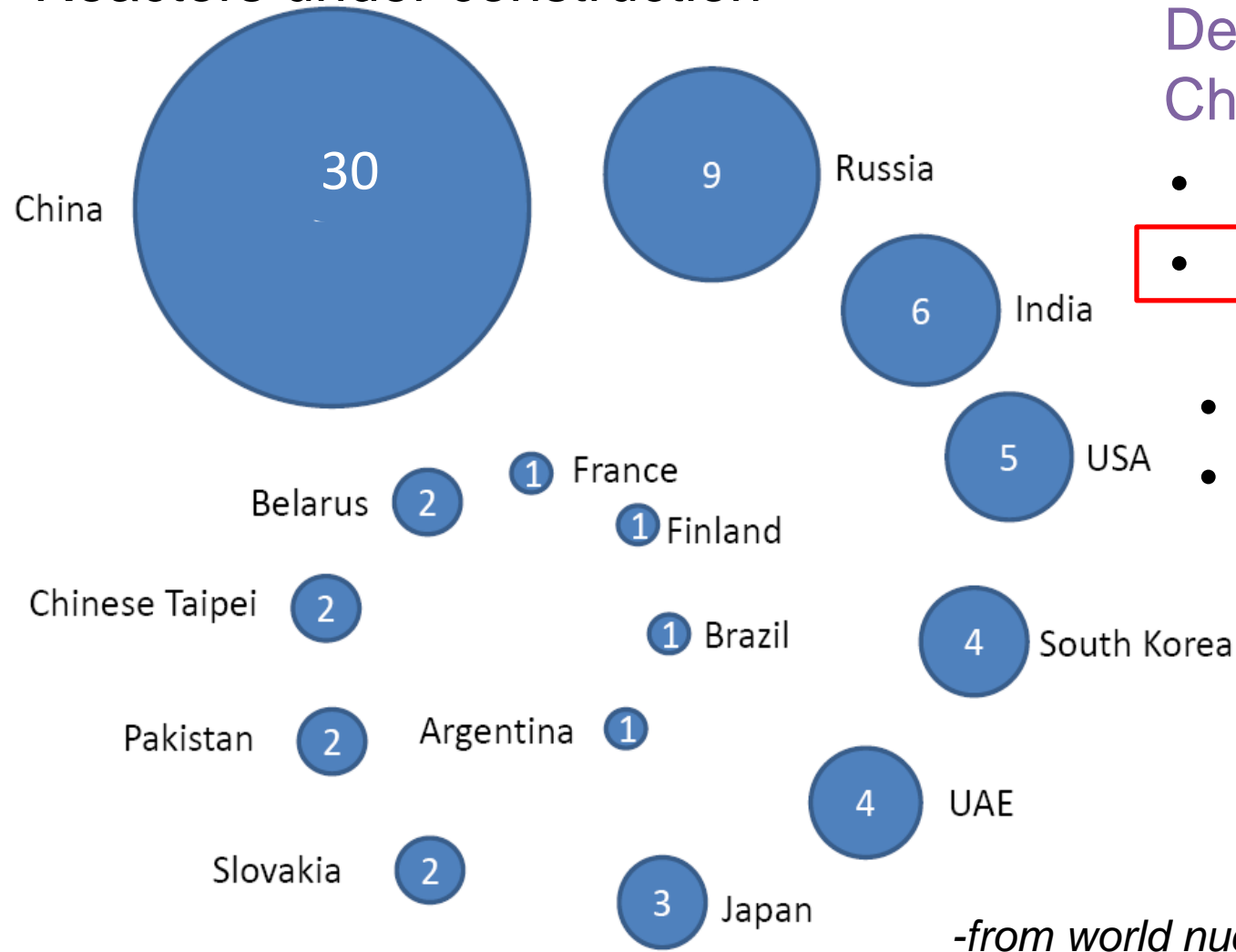
Cambridge, UK



Prof. Greaves

# Background – growing of China nuclear power

## Reactors under construction



## Development plan of China nuclear power:

- 2020: 58 GW;
- 2035: 150 GW.

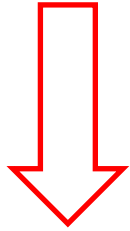
- ~10% of the total;
- 2<sup>nd</sup> world largest NNPs country.

*-from world nuclear association (2015)*

**Fastest-growing in the world.**

# Background – China nuclear waste management

High-level waste



Glass

Intermediate/low-level waste

Optional



Cement

# A typical China Nuclear Waste

- High Fe, Na, S, RE.

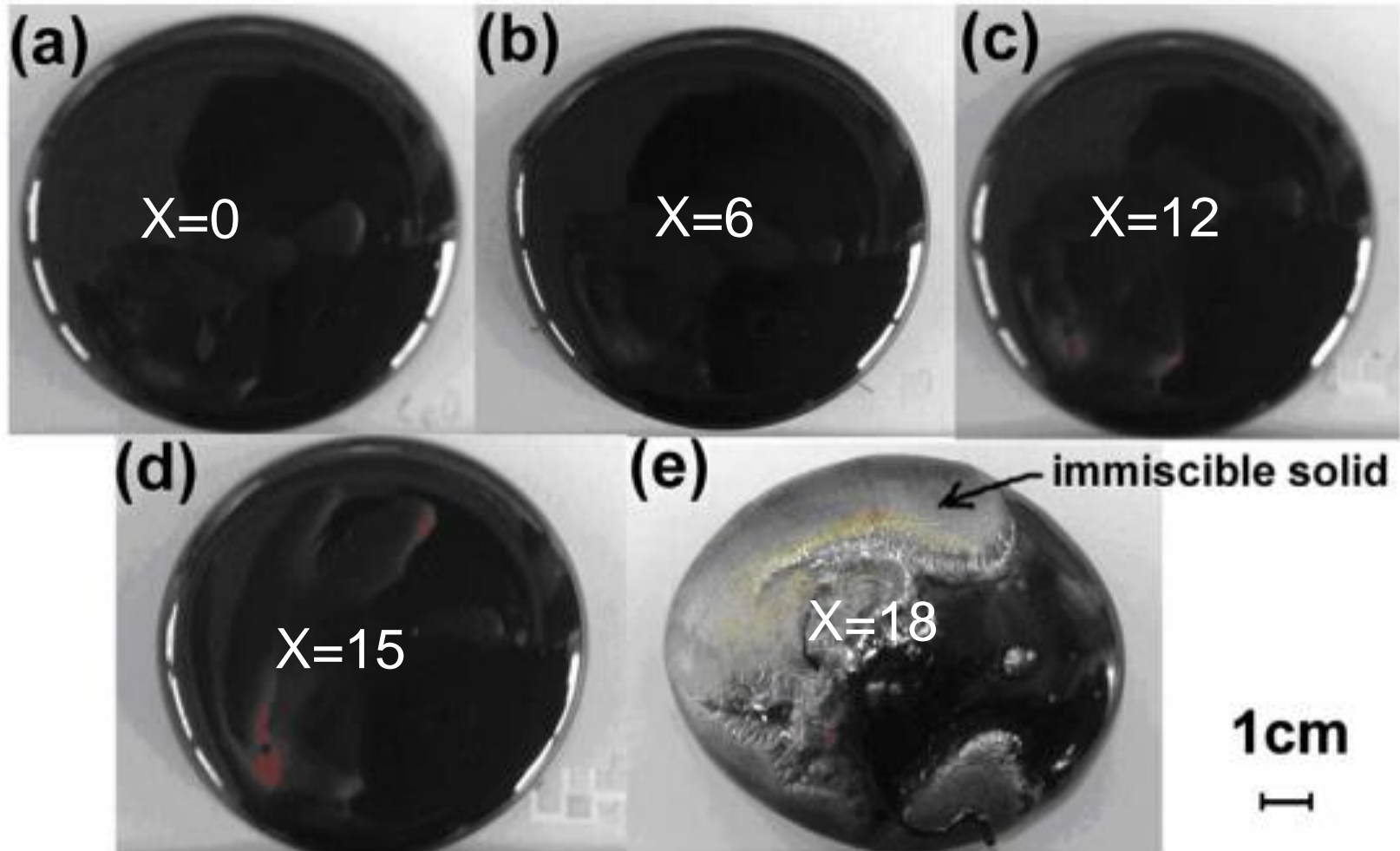
(Oxides)	(Total oxides in VPC simulant)/(g • L <sup>-1</sup> )	(Oxides)	(Total oxides in VPC simulant)/(g • L <sup>-1</sup> )
Al <sub>2</sub> O <sub>3</sub>	9, 116	MnO <sub>2</sub>	0, 105
BaO	0, 010	MoO <sub>3</sub>	1, 154
Cr <sub>2</sub> O <sub>3</sub>	1, 625	Na <sub>2</sub> O	46, 870
Cs <sub>2</sub> O	0, 695	NiO	3, 604
Fe <sub>2</sub> O <sub>3</sub>	20, 513	P <sub>2</sub> O <sub>5</sub>	0, 349
K <sub>2</sub> O	1, 293	SO <sub>3</sub>	4, 418
La <sub>2</sub> O <sub>3</sub>	15, 177	SrO	0, 180
		TiO <sub>2</sub>	n d. <sup>4)</sup>
		V <sub>2</sub> O <sub>5</sub>	4, 612
		Y <sub>2</sub> O <sub>3</sub>	0, 072

- Waste loading in BSi glass: 16%.



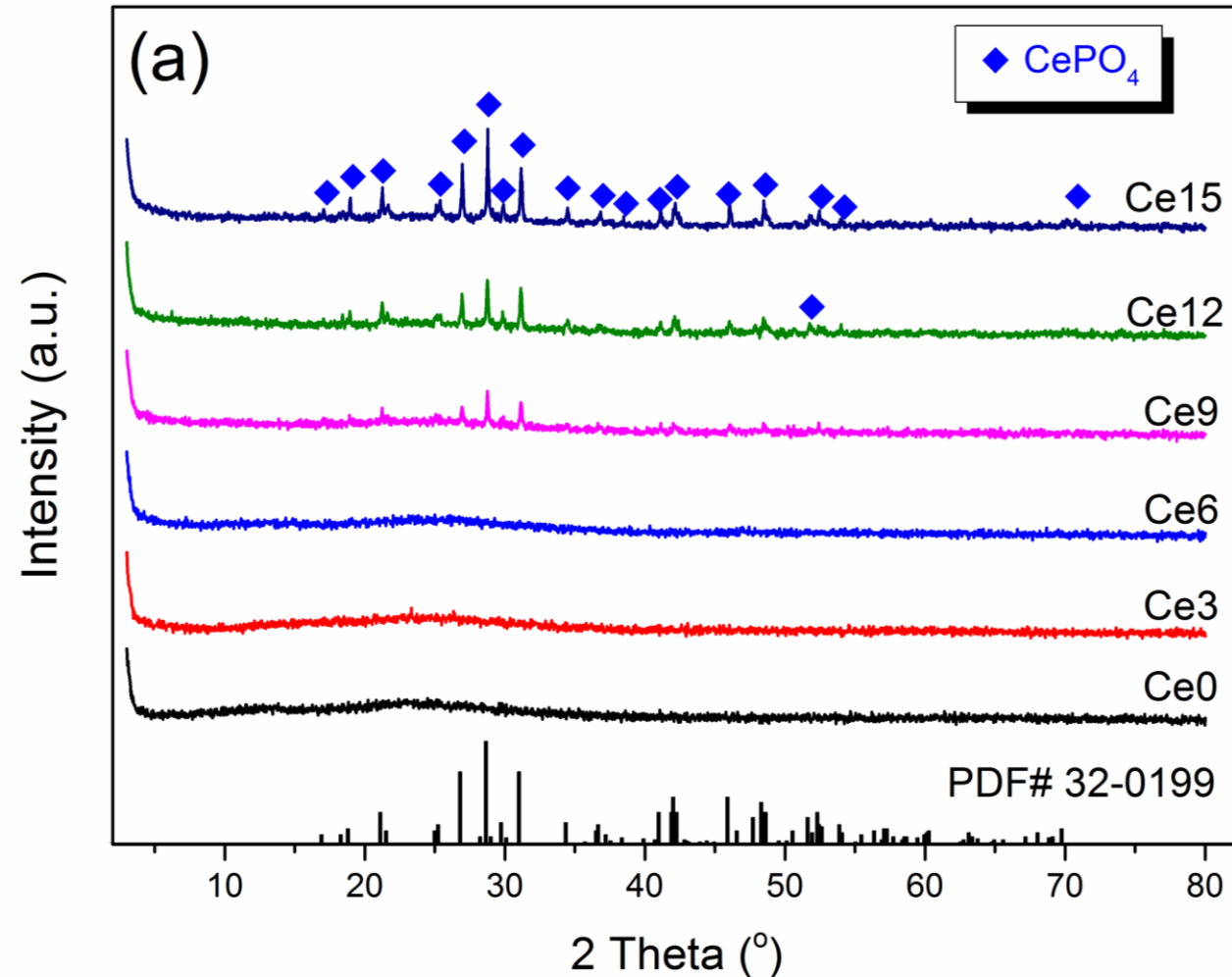
# FeP glass immobilizing Re

$x\text{CeO}_2-(100-x)(36\text{Fe}_2\text{O}_3-10\text{B}_2\text{O}_3-54\text{P}_2\text{O}_5)$  in mol%



Photos of FeP glasses containing different amounts of Ce

# FeP glass immobilizing Re



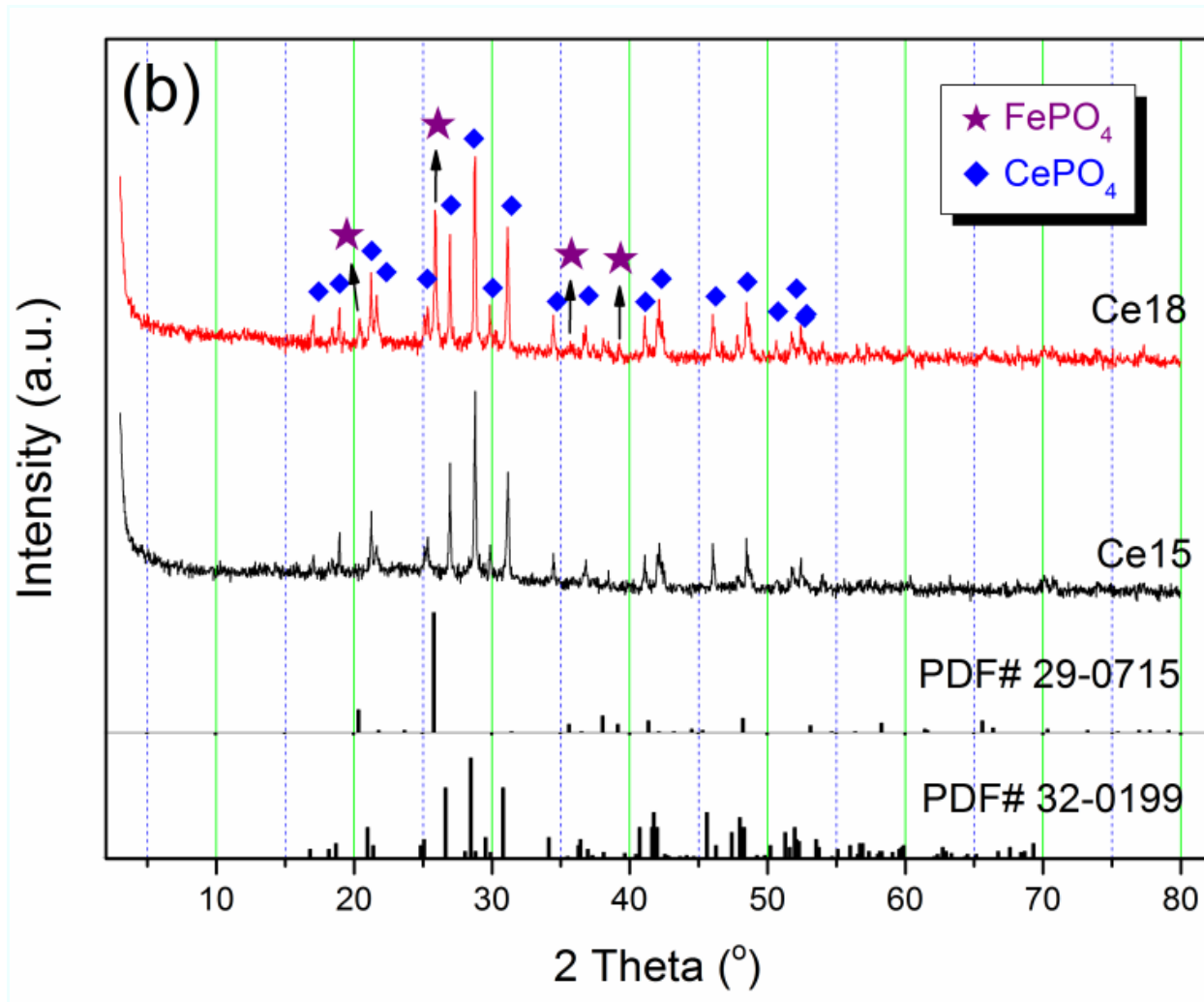
- Monazite ( $\text{CePO}_4$ ) formed, when  $X \geq 9$ .
- Monazite is a durable phase.

XRD patterns of FeP glasses

Fu Wang, *et al.* *J. of Non-Cryst. Solids*,  
409 (2015) 76-82 .



# FeP glass immobilizing Re



- FePO<sub>4</sub> formed, when  $X \geq 18$ .
- FePO<sub>4</sub> is an undurable phase.

XRD patterns of FeP glasses

Fu Wang, *et al.* *J. of Non-Cryst. Solids*, 409 (2015) 76-82 .

# Glass-ceramics

## Motivation

**Barium borosilicate glass**

(improve sulfate solubility)

**Zirconolite, titanite phases**

(improve TRUs solubility)

**Melting-  
thermal  
treatment**

1200°C-3h

$T_g < T < T_c$  -4h

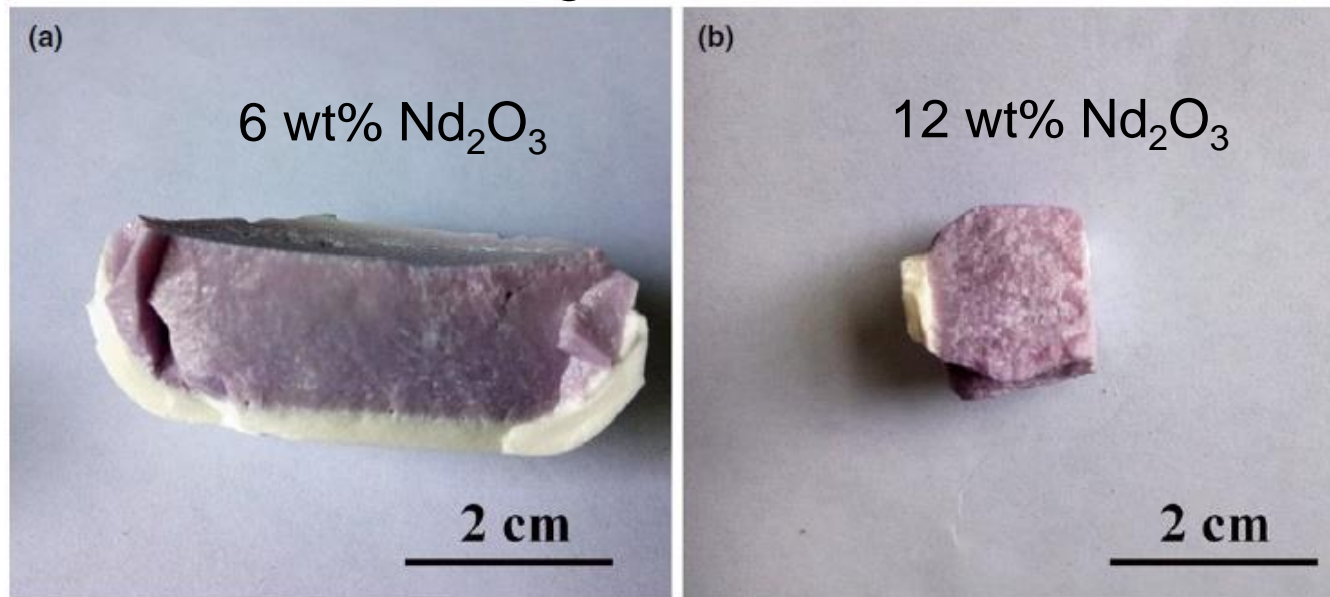
**Barium borosilicate glass-ceramics containing  
zirconolite, titanite phases**

# BaBSi glass-ceramics (Nd effect)

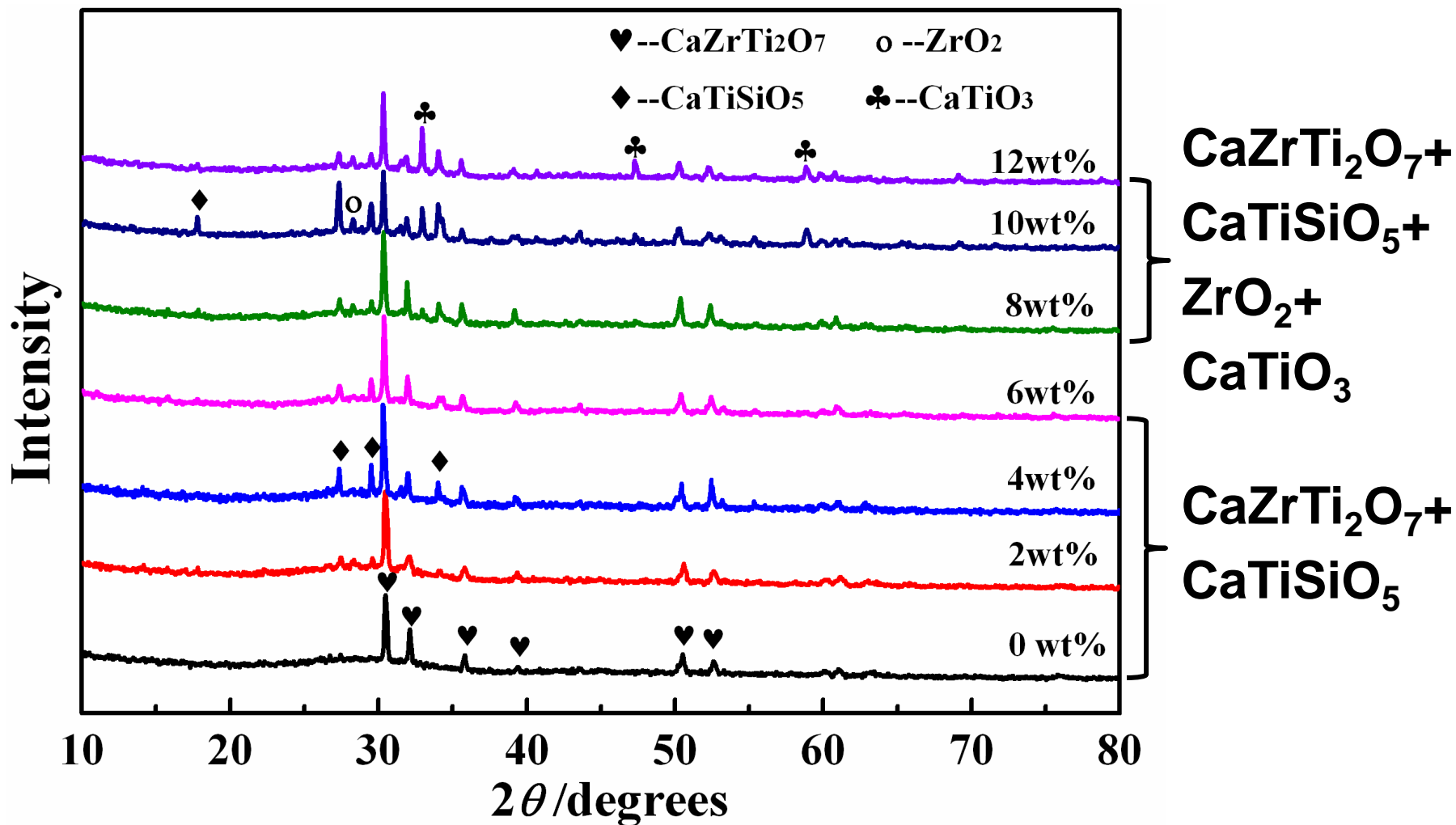
## Composition of glass-ceramics (wt%)

Samples	SiO <sub>2</sub>	B <sub>2</sub> O <sub>3</sub>	Na <sub>2</sub> O	BaO	CaO	TiO <sub>2</sub>	ZrO <sub>2</sub>	Nd <sub>2</sub> O <sub>3</sub>
Nd-0	27.50	11.00	5.50	11.00	12.77	18.19	14.03	0
Nd-2	26.50	10.60	5.30	10.60	12.77	18.19	14.03	2
Nd-4	25.50	10.20	5.10	10.20	12.77	18.19	14.03	4
Nd-6	24.50	9.80	4.90	9.80	12.77	18.19	14.03	6
Nd-8	23.50	9.40	4.70	9.40	12.77	18.19	14.03	8
Nd-10	22.50	9.00	4.50	9.00	12.77	18.19	14.03	10
Nd-12	21.50	8.60	4.30	8.60	12.77	18.19	14.03	12

## Photos of glass-ceramics



# BaBSi glass-ceramics

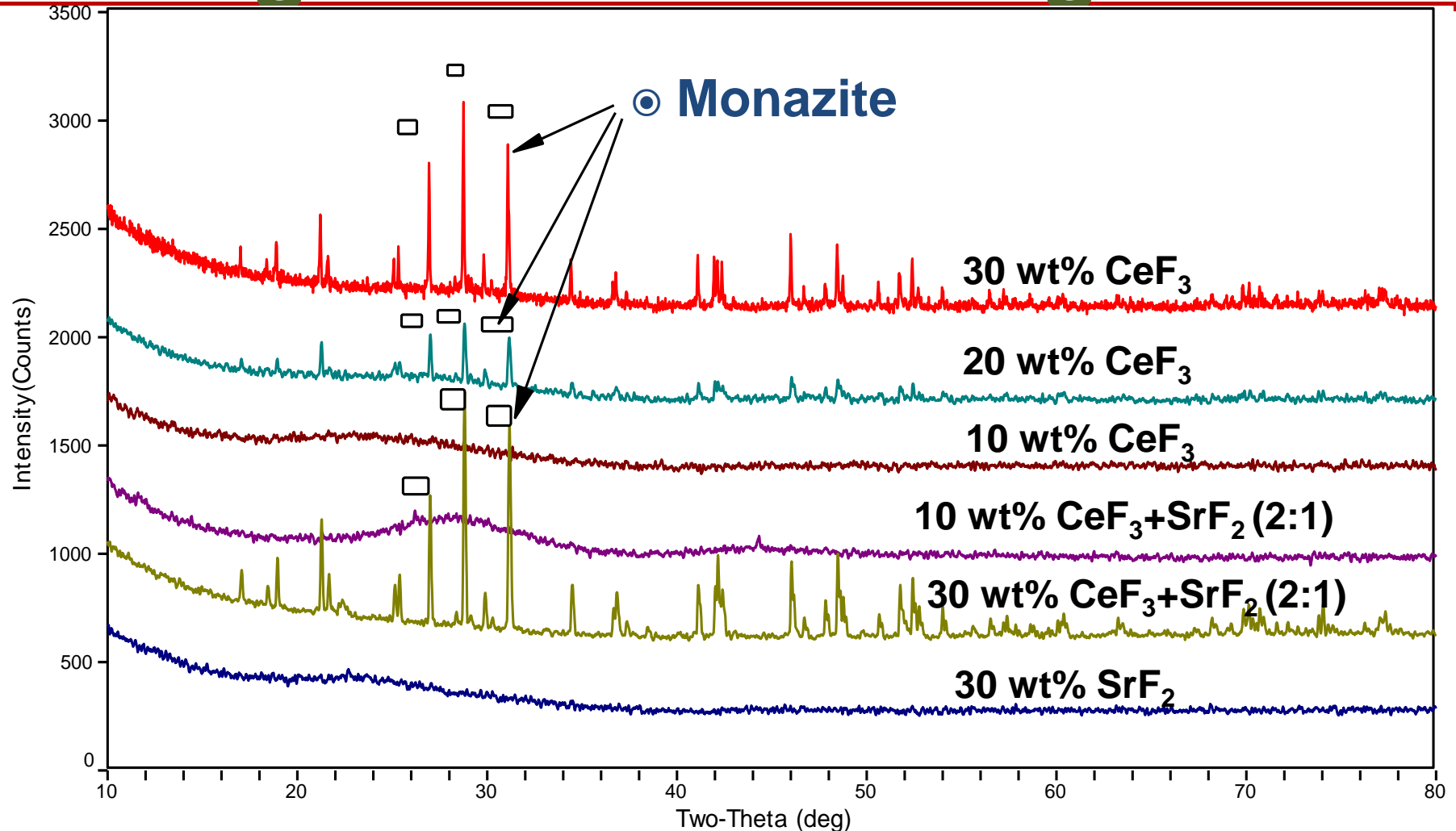


XRD patterns with different contents of  $\text{Nd}_2\text{O}_3$

# Fluoride wastes from molten salt reactors

- MSR utilizes liquid molten fluoride salts as coolant, or even the fuel in the molten salt mixture.
- Reprocessing includes fluorination, distillation to separate uranium and other FPs from fluoride salts.
- Typical simulated fluoride wastes (mol%)
  - 18.8LiF-23.8NaF-0.1MgF<sub>2</sub>-57KF-0.3PF(SrF<sub>2</sub>-SrF-CeF<sub>3</sub>)
  - 8.4CsF-8.3SrF<sub>2</sub>-37.8SmF<sub>3</sub>-29.7ZrF<sub>4</sub>-15.8CeF<sub>4</sub>

# FeBP glass immobilizing fluorides



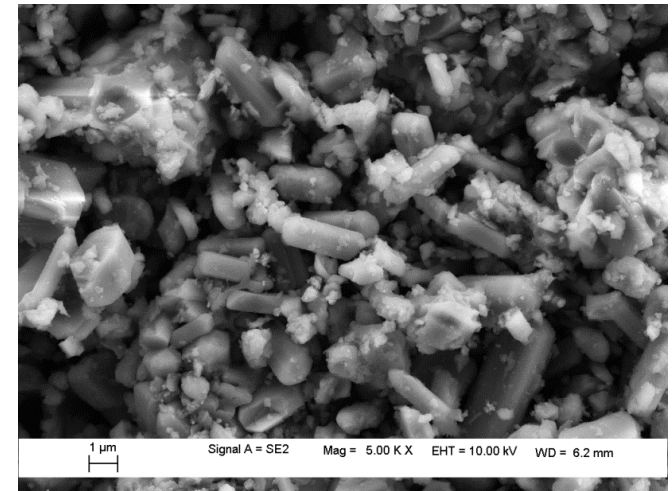
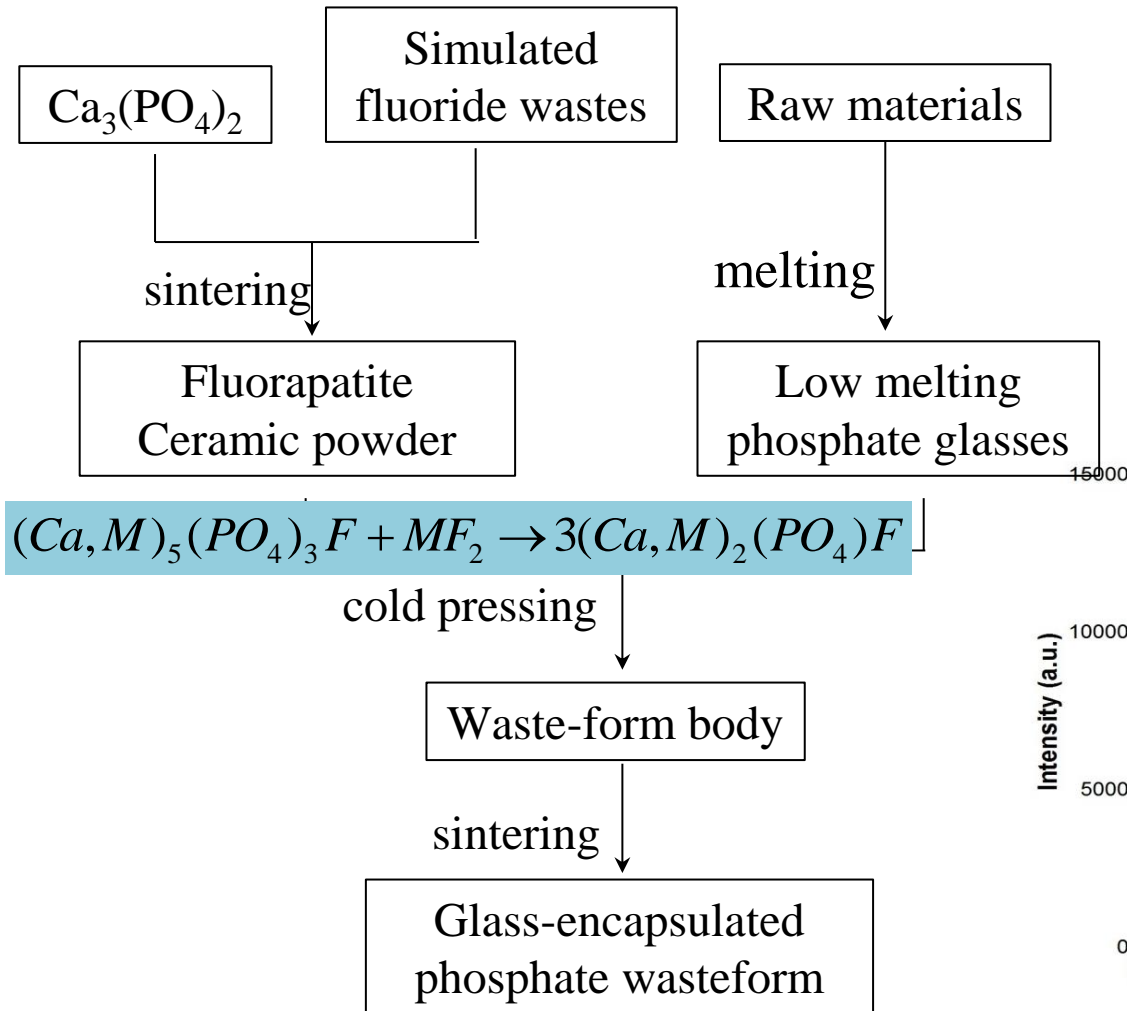
XRD patterns of FeBP glasses

- Monazite formed when  $\text{CeF}_3 \geq 20$  wt%.
- No crystallization when  $\text{SrF}_2 = 30$  wt%.

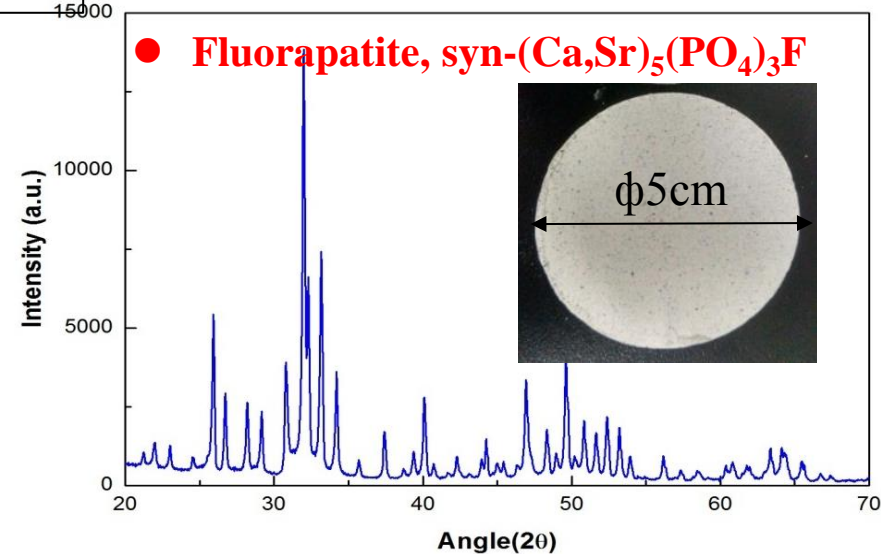


# Phosphate glass encapsulated waste-form

## Experimental



Fluorapatite crystals with hexagonal



XRD pattern and wasteform photo  
(SrF<sub>2</sub> as simulant)

# Works doing @ WHUT - ISG corrosion study

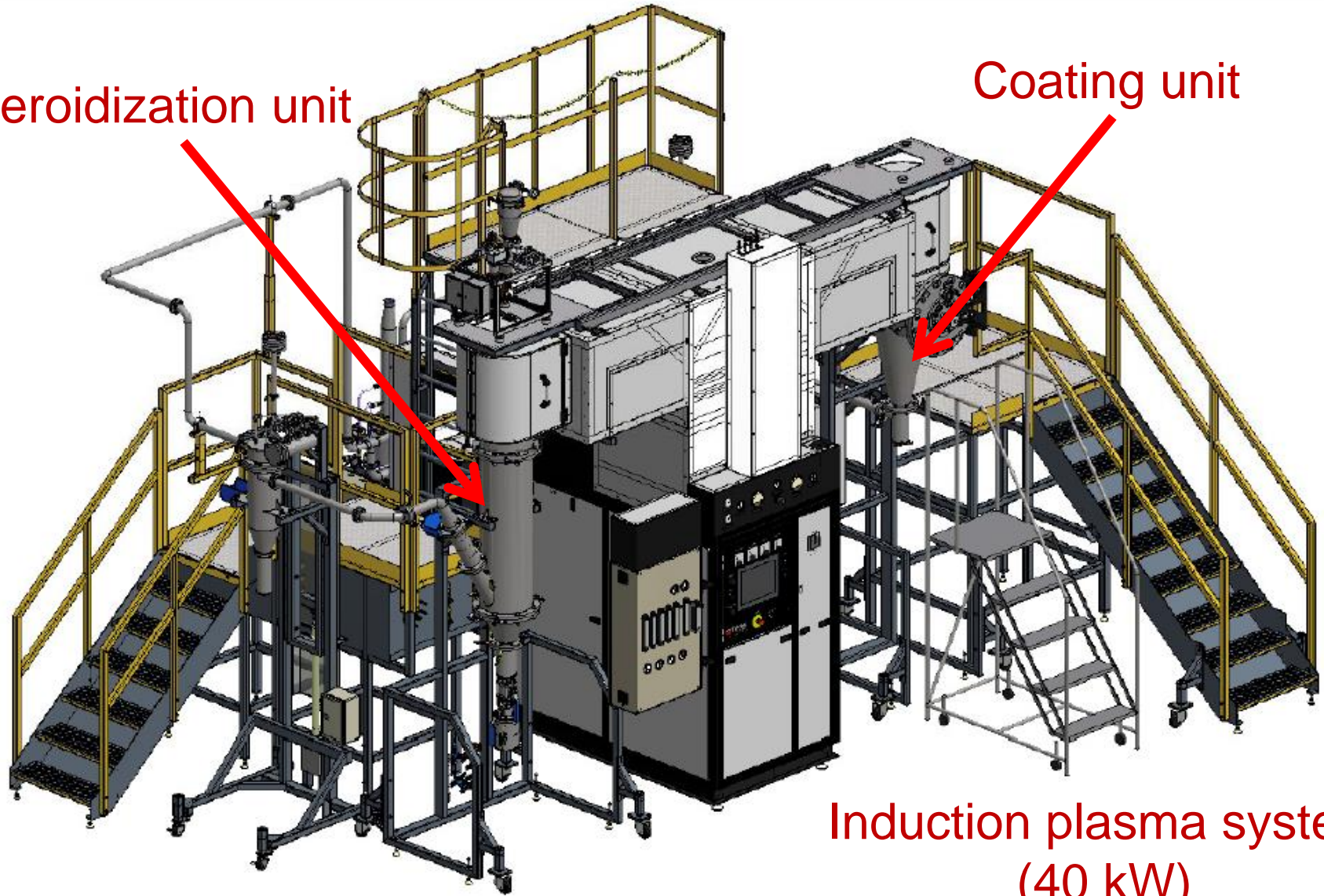


# Works doing @ WHUT - ISG corrosion study

Assessment of PCT (surface area)

Spheroidization unit

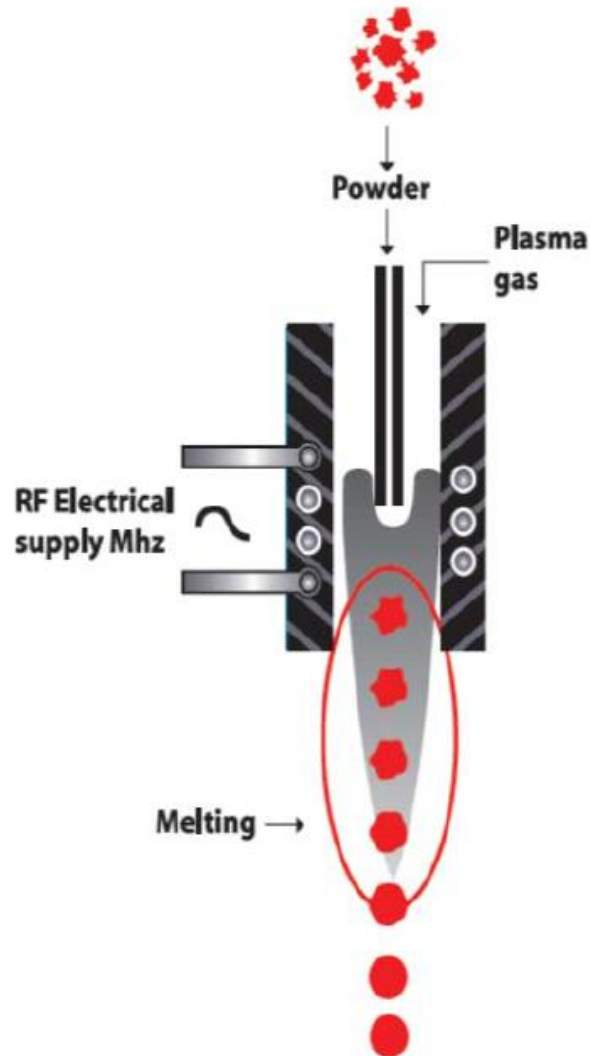
Coating unit



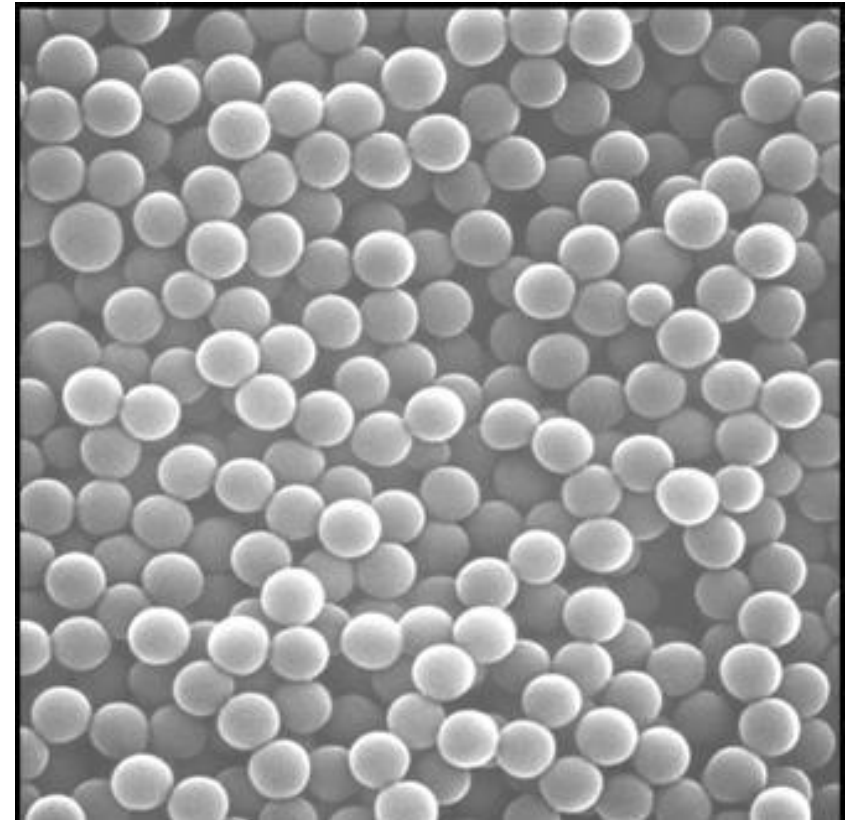


# Works doing @ WHUT - ISG corrosion study

Assessment of PCT (surface area) – glass spheroidization



Powder spheroidization



Uniform size of glass beads  
with smooth surface

# Summary and outlooks

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- ❖ HLW generated from defense program is **urgent** to be vitrified, and **a large amount of HLW** will be produced from reprocessing of spent nuclear fuel soon.
- ❖ R&D of China nuclear waste vitrification is **in the preliminary stage**, and **is very needed**, in order to construct our own vitrification facility.



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# Thanks for your attention and comments!

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