

# Data-driven cartography of materials properties

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Workshop on Crystal Structure Prediction:  
Exploring the Mendeleev Table as a Palette to Design New Materials  
ICTP, Trieste, 13-18 January 2019



# Data-driven materials science: the big picture

NOVEL MATERIALS DISCOVERY

- Design of new materials: preparation, synthesis, and characterization is complex and costly
  - About 240 000 *inorganic* materials are known to exist (Springer Materials)
  - Basic properties determined for very few of them
  - Number of possible materials: practically infinite
- ⇒ New materials with superior properties exist but not yet known
- Data-analytics tools will help to identify trends and anomalies in data and guide discovery of new materials

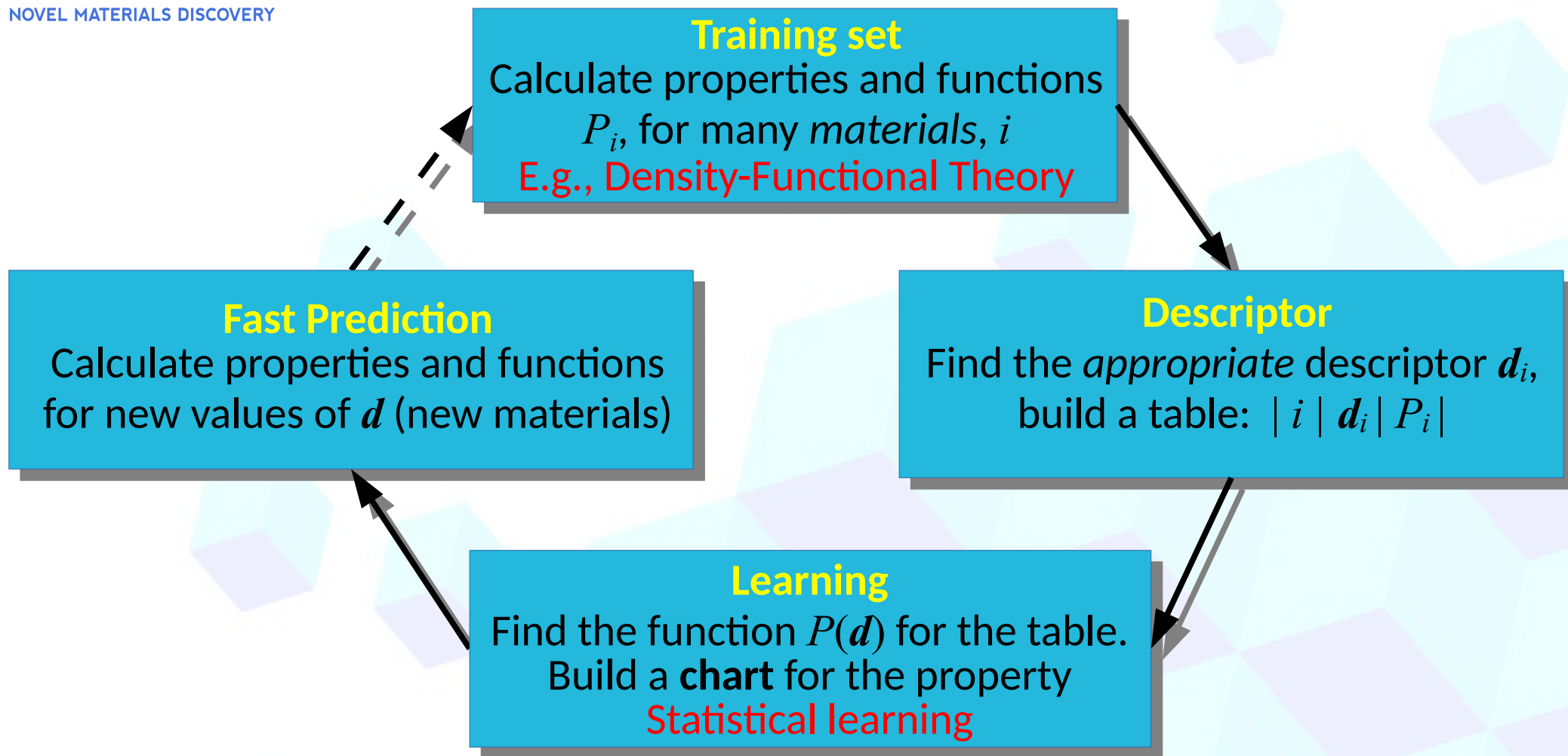
From the **periodic table of the elements** to a **chart of materials**

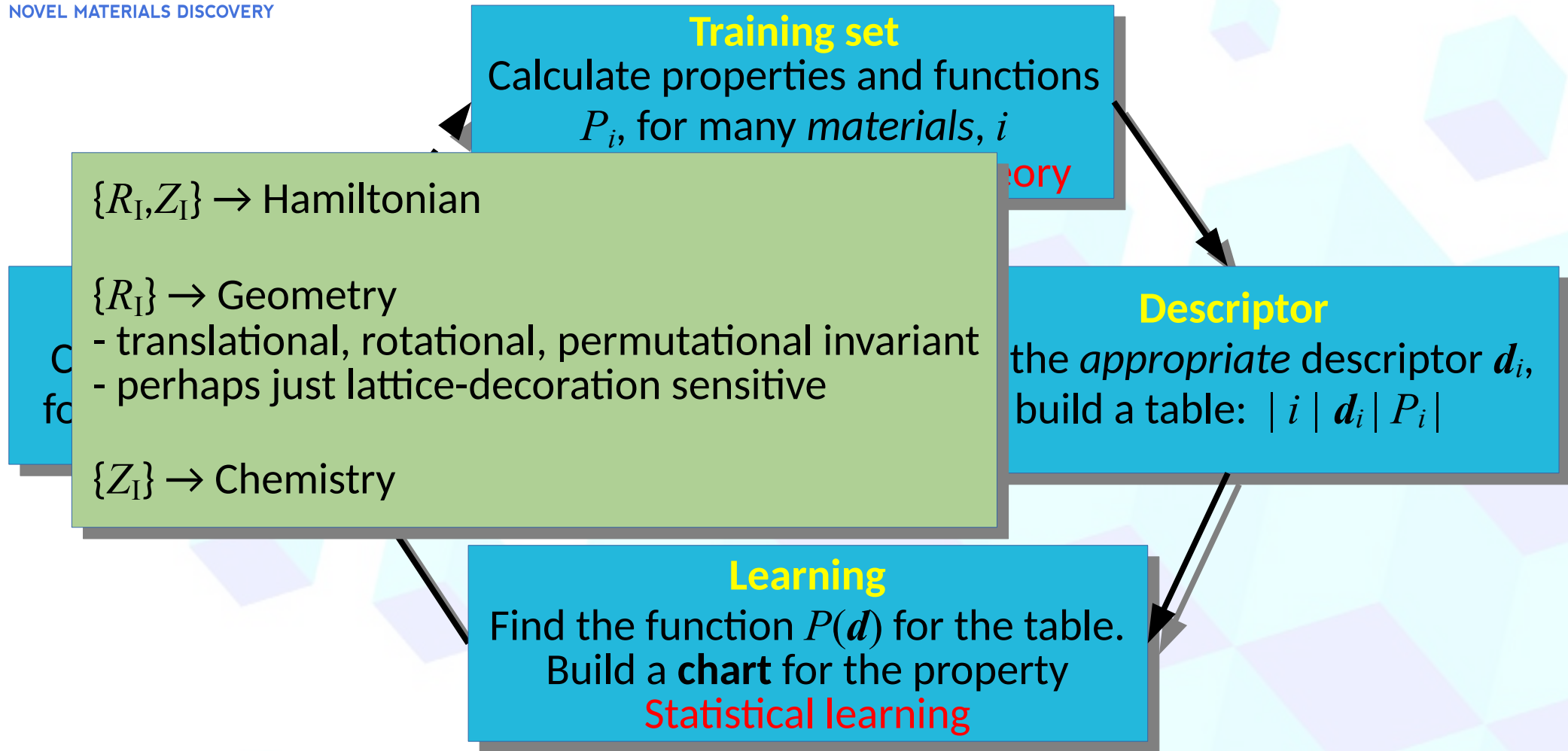
| Reihen | Gruppe I.<br>—<br>R <sup>2</sup> O | Gruppe II.<br>—<br>RO | Gruppe III.<br>—<br>R <sup>2</sup> O <sup>3</sup> | Gruppe IV.<br>RH <sup>4</sup><br>RO <sup>2</sup> | Gruppe V.<br>RH <sup>3</sup><br>R <sup>2</sup> O <sup>5</sup> | Gruppe VI.<br>RH <sup>2</sup><br>RO <sup>3</sup> | Gruppe VII.<br>RH<br>R <sup>2</sup> O <sup>7</sup> | Gruppe VIII.<br>—<br>RO <sup>4</sup> |
|--------|------------------------------------|-----------------------|---------------------------------------------------|--------------------------------------------------|---------------------------------------------------------------|--------------------------------------------------|----------------------------------------------------|--------------------------------------|
| 1      | H=1                                |                       |                                                   |                                                  |                                                               |                                                  |                                                    |                                      |
| 2      | Li=7                               | Be=9.4                | B=11                                              | C=12                                             | N=14                                                          | O=16                                             | F=19                                               |                                      |
| 3      | Na=23                              | Mg=24                 | Al=27.3                                           | Si=28                                            | P=31                                                          | S=32                                             | Cl=35.5                                            |                                      |
| 4      | K=39                               | Ca=40                 | —=44                                              | Ti=48                                            | V=51                                                          | Cr=52                                            | Mn=55                                              | Fe=56, Co=59,<br>Ni=59, Cu=63.       |
| 5      | (Cu=63)                            | Zn=65                 | —=68                                              | —=72                                             | As=75                                                         | Se=78                                            | Br=80                                              |                                      |
| 6      | Rb=85                              | Sr=87                 | —=88                                              | —=90                                             | Nb=94                                                         | Mo=96                                            | —=100                                              | Ru=104, Rh=104,<br>Pd=106, Ag=108.   |
| 7      | (Ag=108)                           | C=112                 | —=120                                             | —=122                                            | —=124                                                         | Te=125                                           | J=127                                              |                                      |
| 8      | Cs=133                             | Ba=137                | ?Di=138                                           | ?Ce=140                                          | —                                                             | —                                                | —                                                  | — — — —                              |
| 9      | (—)                                | —                     | —                                                 | —                                                | —                                                             | —                                                | —                                                  | —                                    |
| 10     | —                                  | —                     | ?Er=178                                           | ?La=180                                          | Ta=182                                                        | W=184                                            | —                                                  | Os=195, Ir=197,<br>Pt=198, Au=199.   |
| 11     | (Au=199)                           | Hg=200                | Tl=204                                            | Pb=207                                           | Bi=208                                                        | —                                                | —                                                  | —                                    |
| 12     | —                                  | —                     | —                                                 | Th=231                                           | —                                                             | U=240                                            | —                                                  | — — — —                              |

Ga=69.7 Ge=72.6

Mendeleev's 1871 periodic table

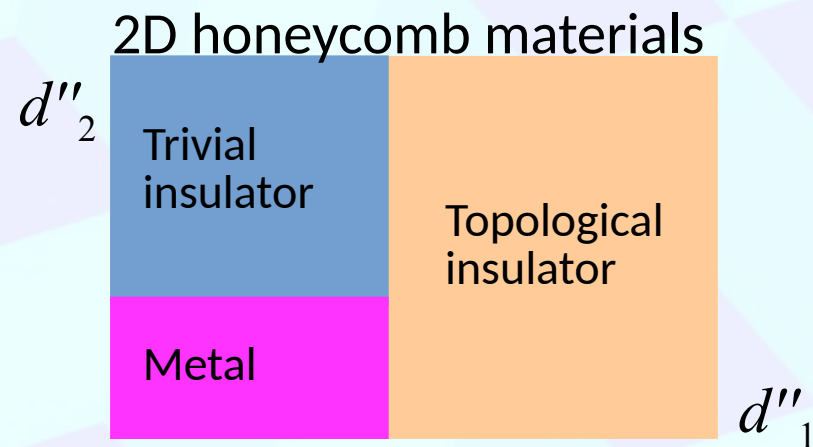
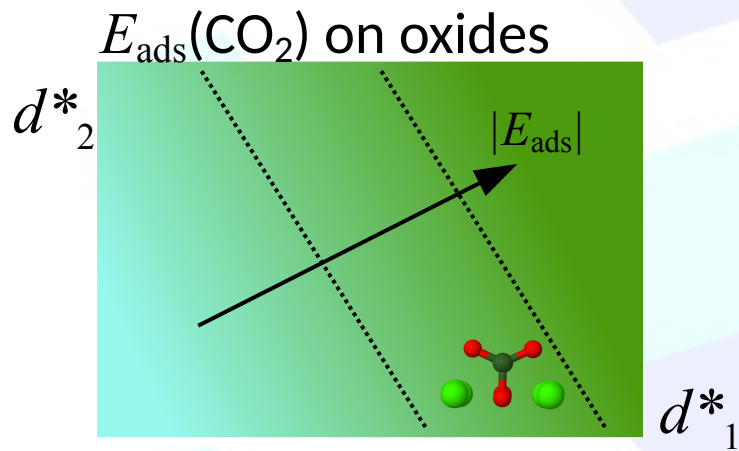
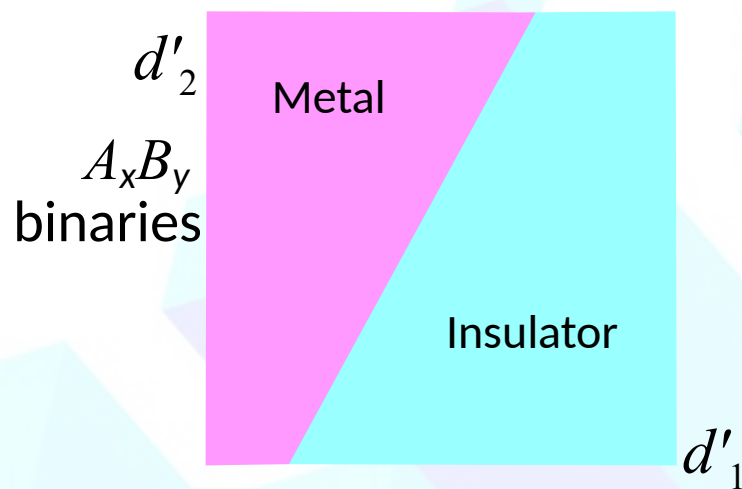
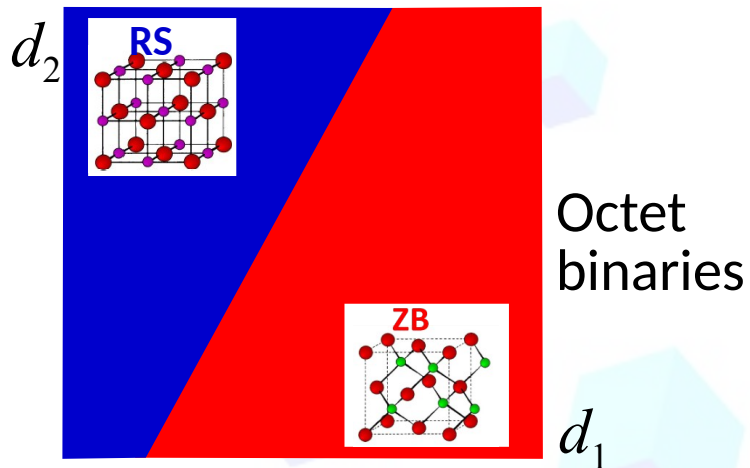
# Data analytics: an ideal flow chart





# Learning/discovering maps of materials properties. A quantum many-body problem

NOVEL MATERIALS DISCOVERY





# Learning/discovering maps of materials properties. A quantum many-body problem

NOVEL MATERIALS DISCOVERY

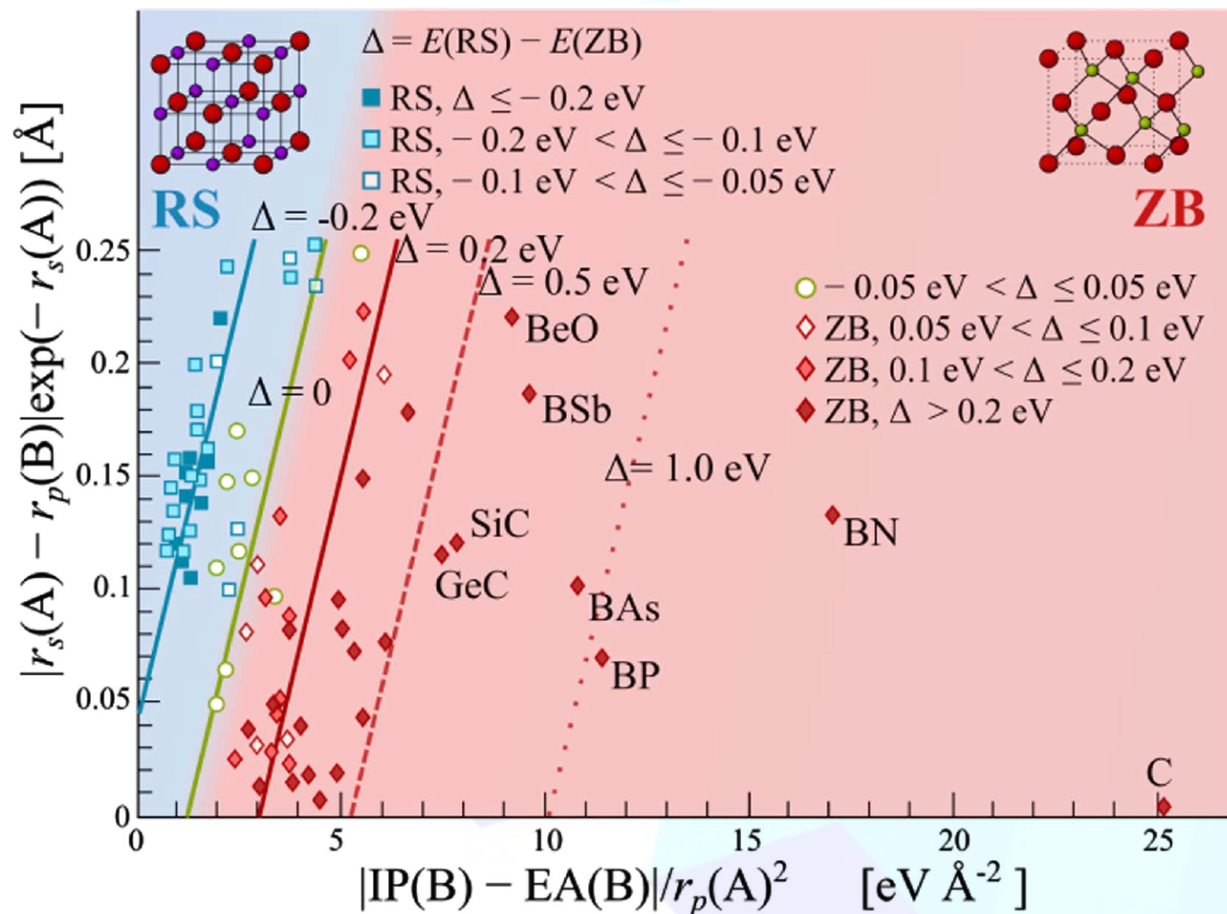
From the Hamiltonian of a physical system, in principle we can derive all properties (observables).

But in practice, the Hamiltonian is often not the starting point.

For instance, given a class of chemical compositions (e.g., via prototype formula, such as  $ABX_3$ ):

- what is the most stable crystal structure of each material in the class?
- which materials are metals / topological insulators / superconductors ?
- which material has the highest melting point?
- which materials has a surface optimal for catalysing some chemical reaction?





Structure map with compressed-sensing algorithm, starting from 7 atomic features



## Compressed sensing

Aim: finding descriptors and learning predictive models

Ansatz:

$$\mathbf{P} = c_1 \mathbf{d}_1 + c_2 \mathbf{d}_2 + \dots + c_n \mathbf{d}_n$$

$\mathbf{P}$ : property of interest

$\mathbf{d}_1, \dots, \mathbf{d}_n$ : features, i.e., (nonlinear) functions of *primary features* (EA, IP, ...)

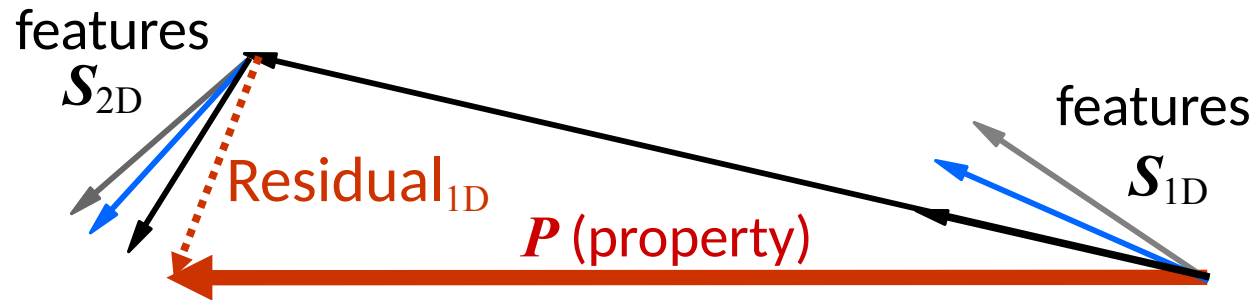
$c_1, \dots, c_n$ : unknown coefficients => as few as possible are *nonzero*

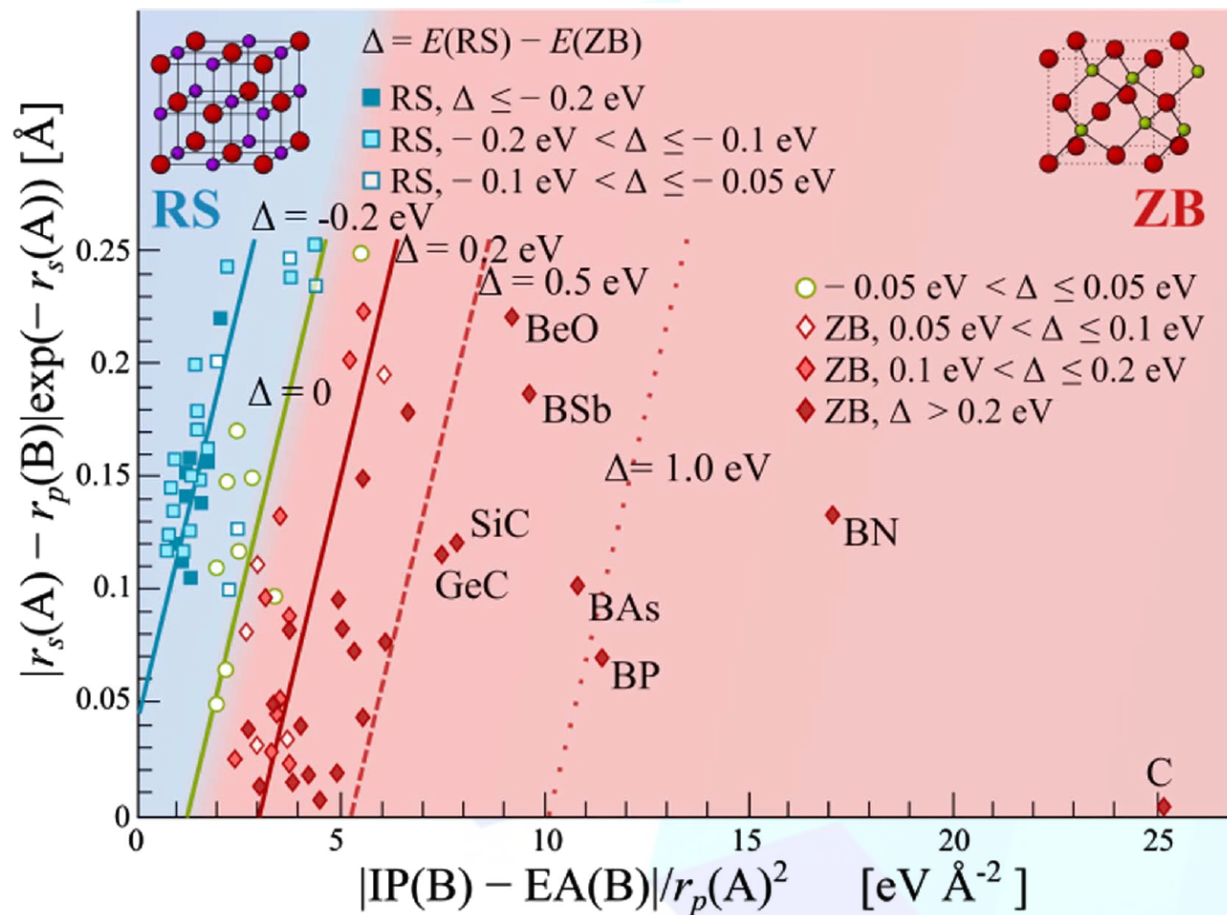
$$\operatorname{argmin}_{\mathbf{c} \in \mathbb{R}^M} \|\mathbf{P} - \mathbf{D}\mathbf{c}\|_2^2 + \lambda \|\mathbf{c}\|_0$$

$\mathbf{d}_i$ : iterative construction with features and operators (+, ×, /, <sup>2</sup>, ...)

$$\frac{\text{IP}(\text{B}) - \text{EA}(\text{B})}{r_p(\text{A})^2}, \frac{|r_s(\text{A}) - r_p(\text{B})|}{\exp(r_s(\text{A}))}$$

$$\operatorname{argmin}_{\mathbf{c} \in \mathbb{R}^M} \|\mathbf{P} - \mathbf{D}\mathbf{c}\|_2^2 + \lambda \|\mathbf{c}\|_0$$





Structure map with compressed-sensing algorithm, starting from 7 atomic features

LMG *et al.*, PRL 2015,  
NJP 2017

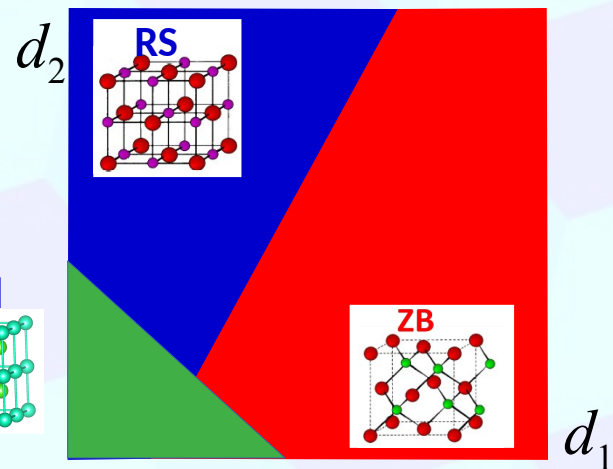
$$\{P^{(1)}, P^{(2)}, \dots, P^{N^T}\} \longrightarrow P^k = \textcircled{d} \cdot c^k$$

$$\arg \min_{\mathbf{c}} (\|\mathbf{P} - \mathbf{D}\mathbf{c}\|_2^2 + \lambda \|\mathbf{c}\|_0)$$

$$\arg \min_{\mathbf{C}} \sum_{k=1}^{N^T} \frac{1}{N_k^M} \|\mathbf{P}^k - \mathbf{D}^k \mathbf{C}^k\|_2^2 + \lambda \|\mathbf{C}\|_0$$

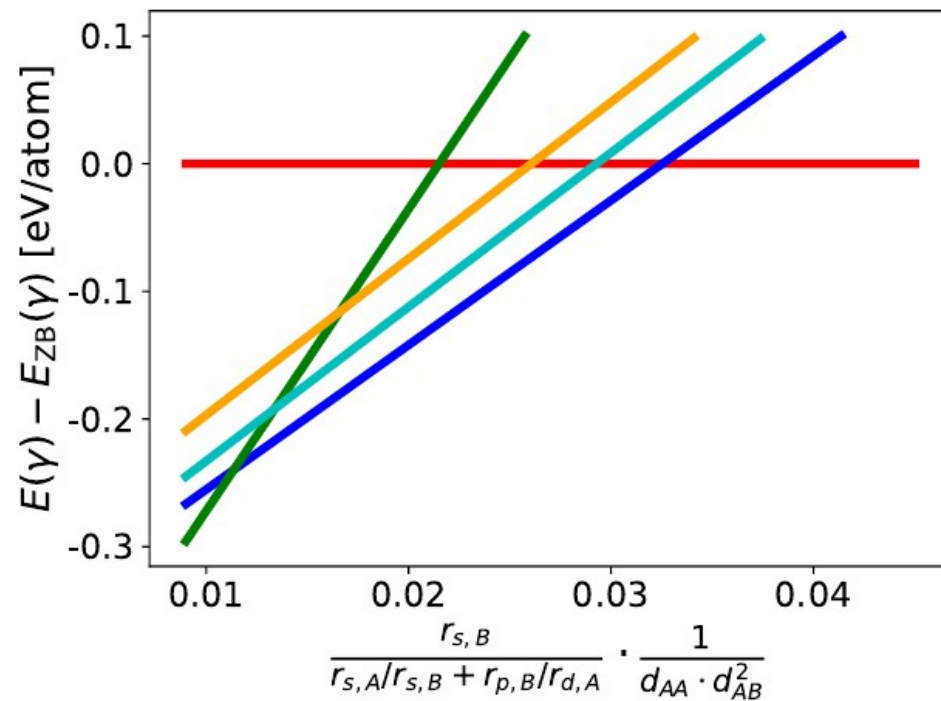
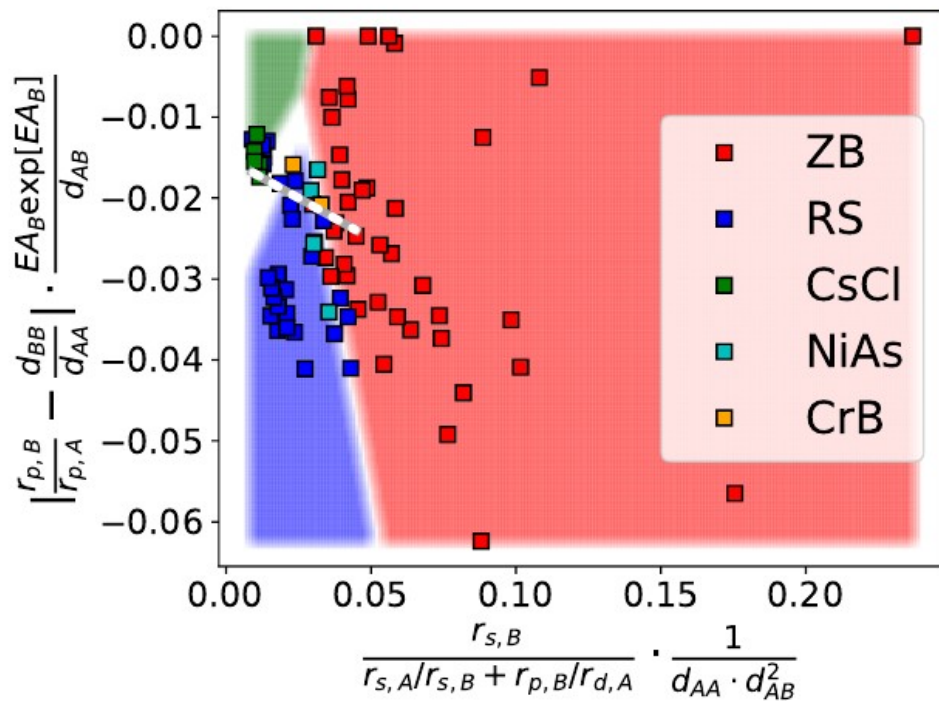
Application: multi-phase stability diagram

Properties: crystal-structure formation energies



# One descriptor to rule them all: Multi-task SISSO Energy differences among 5 crystal structures.

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$$\operatorname{argmin}_{\mathbf{c} \in \mathbb{R}^M} (\underbrace{\|\mathbf{P} - \mathbf{D}\mathbf{c}\|_2^2}_{\text{New cost function to be minimized: overlap of convex domains}} + \lambda \|\mathbf{c}\|_0)$$

New cost function to be minimized:  
overlap of *convex* domains

1. # points in the *convex* overlap domain
2. Area of the domain overlap
3. Distance between domains

Good also for multi-categorical problems  
(see A. F. Bialon *et al.*, Chem. Mater. **28**, 2550 (2016))



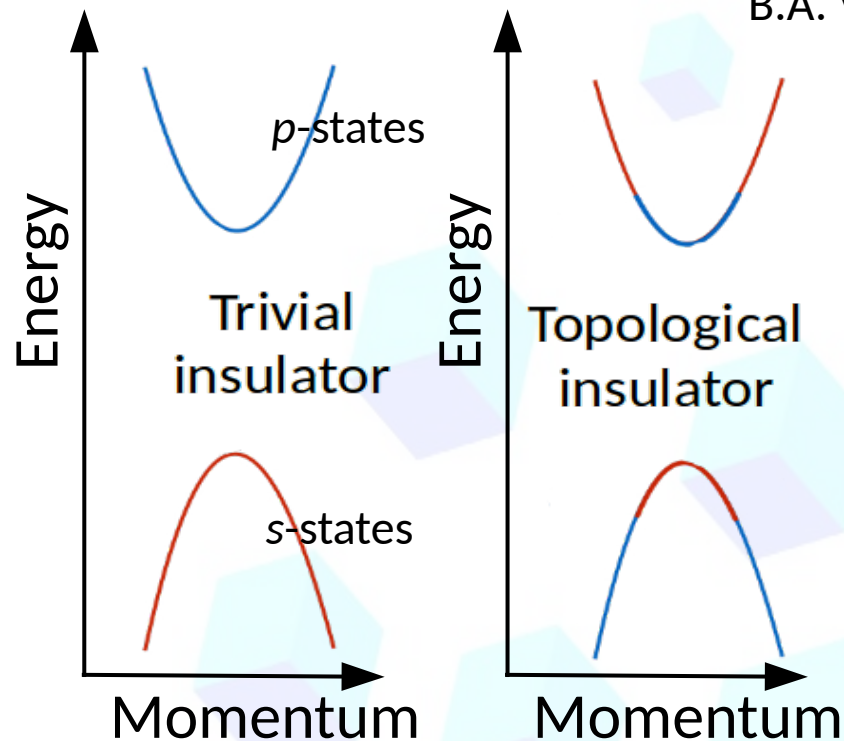


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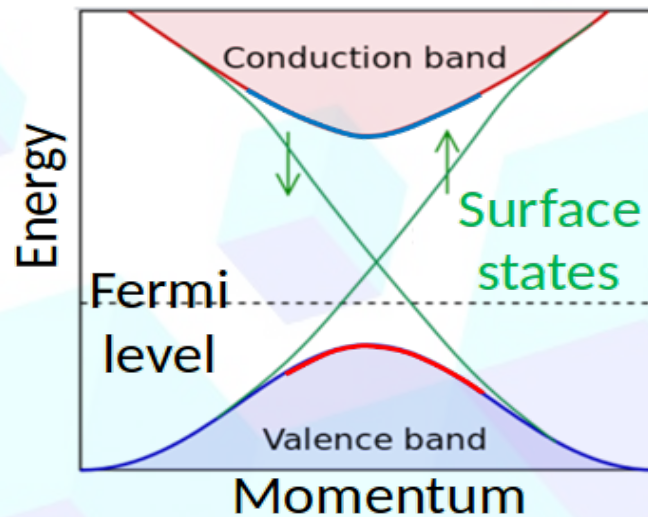
# SISSO: predicting novel honeycomb (~2D) **topological insulators**

NOVEL MATERIALS DISCOVERY

B.A. Volkov, O.A. Pankratov, JETP Lett. **42**, 178 (1985)



Triggered by:  
spin-orbit coupling, alloying,  
strain, electric field

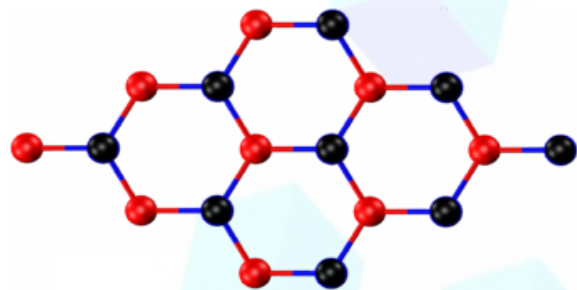
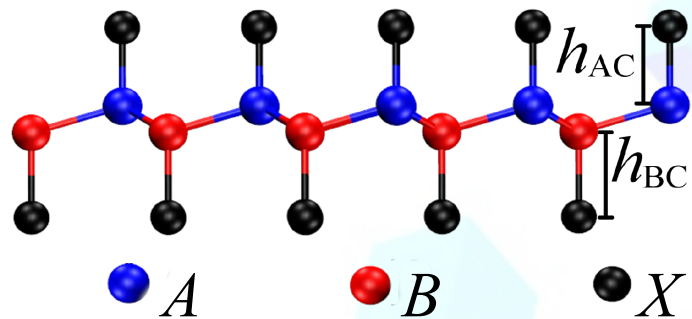


2D topological insulators:  
No backscattering in edge states.  
Characterized by a  $Z_2$  topological index.  
Promising materials in spintronics  
applications, e.g., quantum computing.



# SISSO: predicting novel honeycomb (~2D) topological insulators

NOVEL MATERIALS DISCOVERY

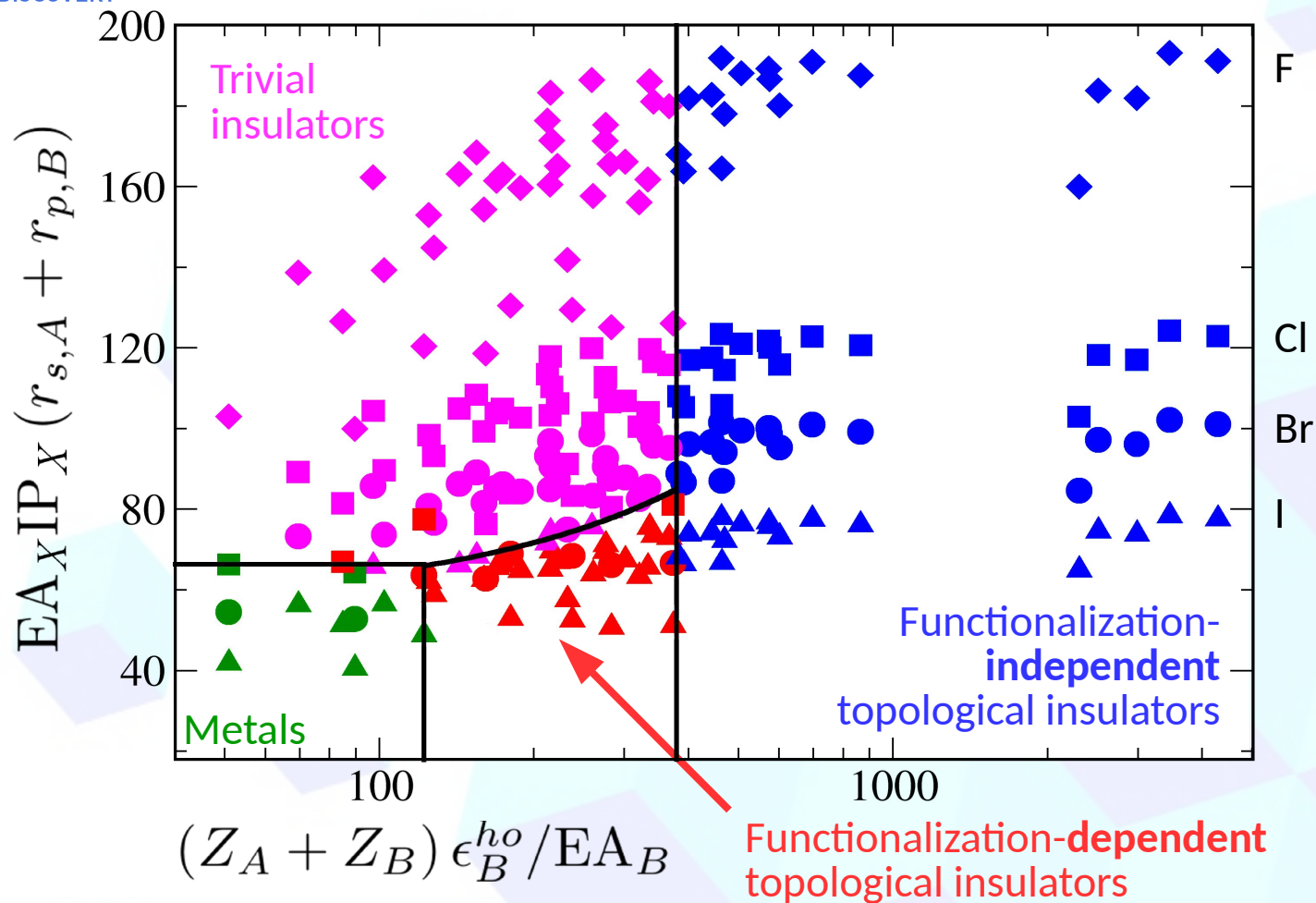


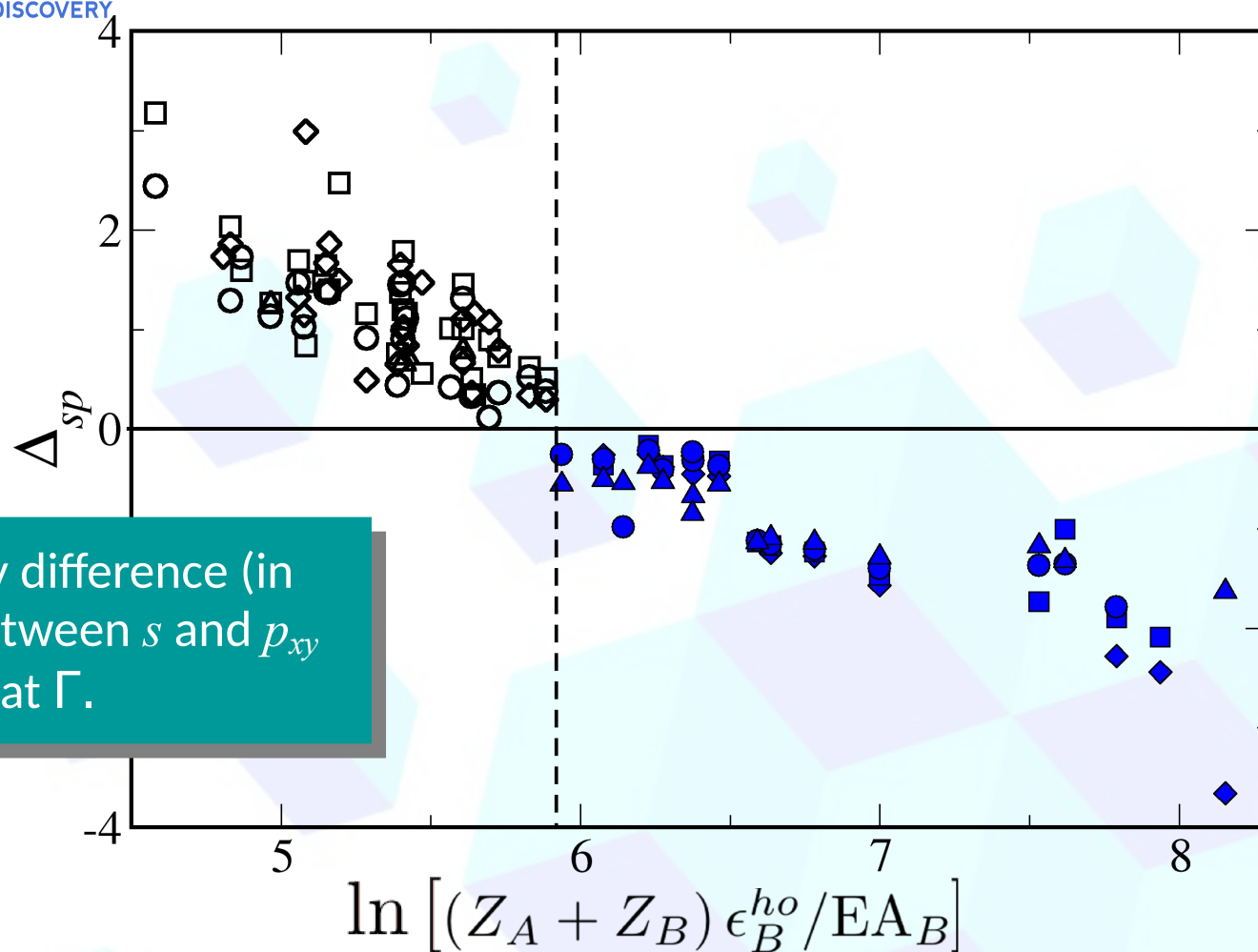
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SISSO: predicting novel honeycomb  
(~2D) **topological insulators**

NOVEL MATERIALS DISCOVERY

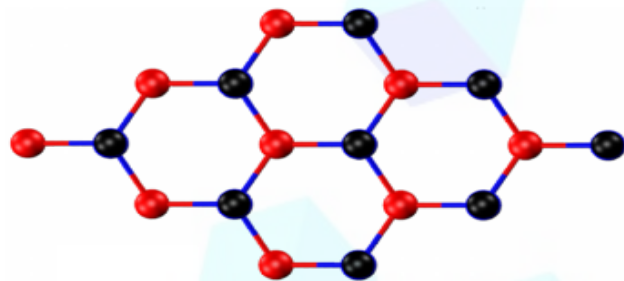
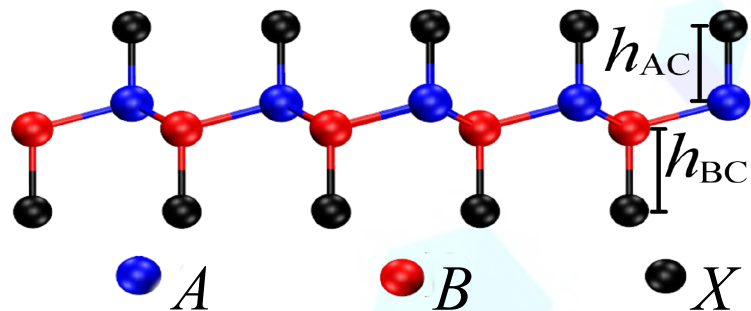




Energy difference (in eV) between  $s$  and  $p_{xy}$  states at  $\Gamma$ .

# SISSO: predicting novel honeycomb (~2D) topological insulators

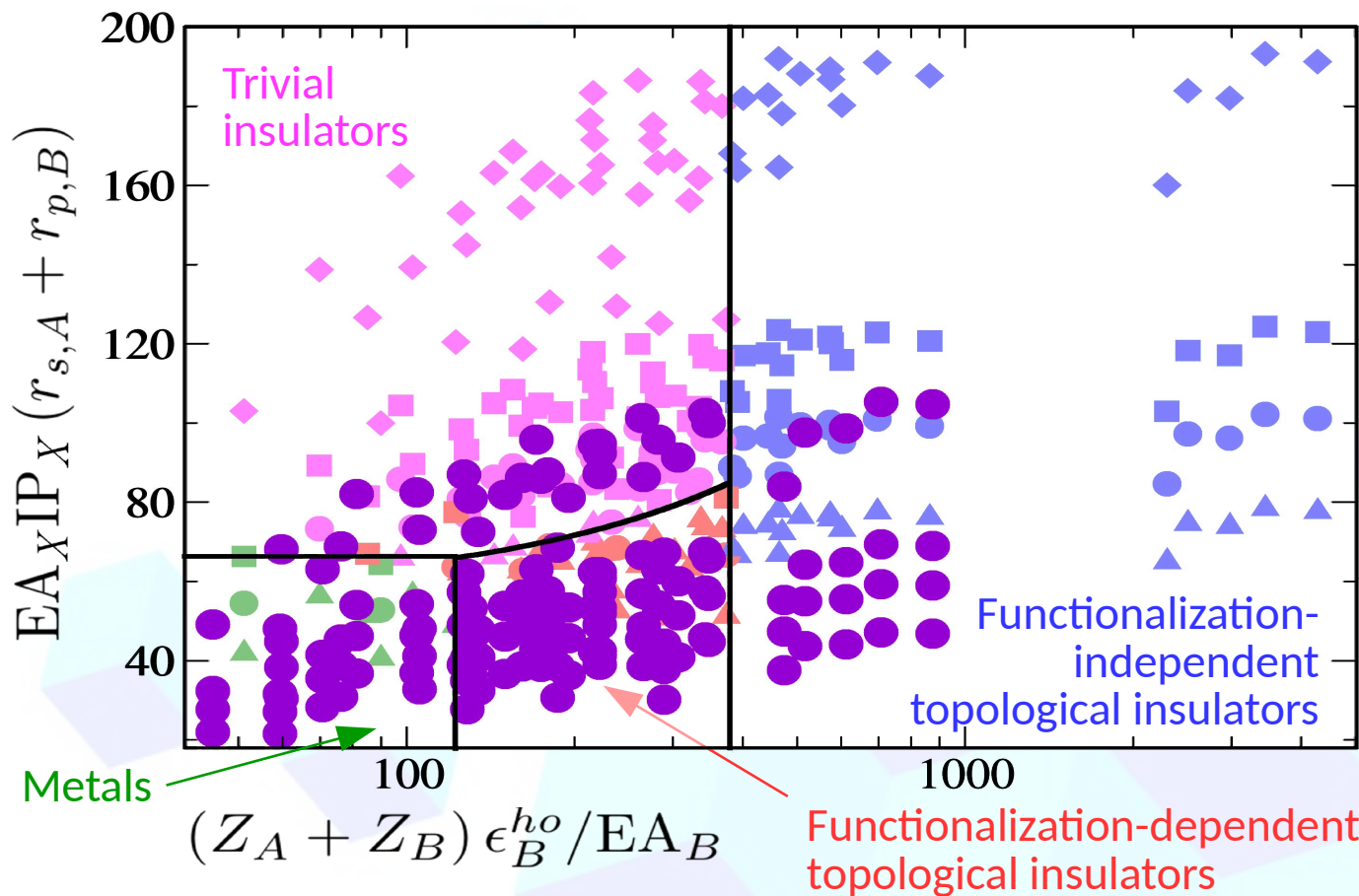
NOVEL MATERIALS DISCOVERY



|   |     |     |     |       |
|---|-----|-----|-----|-------|
| A | II  | III | IV  | IV V  |
| B | VI  | V   | IV  | VI V  |
| X | VII | VII | VII | VI VI |

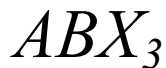
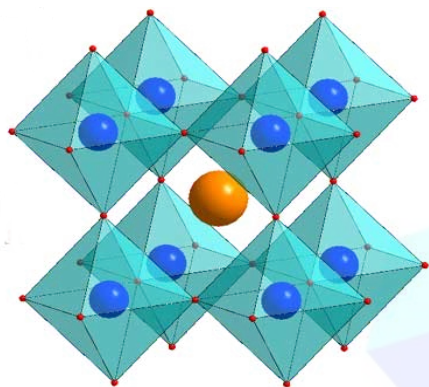
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|  |  |  |  |  |  | IV-VI       |           | V-V        |             | VIA         |             | VIIA       |             | VIII       |              |
|  |  |  |  |  |  | 5           | 6         | 7          | 8           | 9           | 10          | 11         | 12          | 13         | 14           |
|  |  |  |  |  |  | 10.81       | 12.011    | 14.007     | 15.999      | 18.998      | 20.180      | 21.018     | 22.990      | 24.304     | 26.982       |
|  |  |  |  |  |  | B           | C         | N          | O           | F           | Ne          | Na         | Mg          | Al         | Si           |
|  |  |  |  |  |  | BORON       | CARBON    | NITROGEN   | OXYGEN      | FLUORINE    | NEON        | SODIUM     | MAGNESIUM   | ALUMINUM   | SILICON      |
|  |  |  |  |  |  | 13          | 14        | 15         | 16          | 17          | 18          | 19         | 20          | 21         | 22           |
|  |  |  |  |  |  | 26.982      | 28.085    | 30.974     | 32.06       | 35.45       | 39.948      | 39.098     | 40.078      | 44.956     | 47.88        |
|  |  |  |  |  |  | Al          | Si        | P          | S           | Cl          | Ar          | K          | Ca          | Sc         | Ti           |
|  |  |  |  |  |  | ALUMINUM    | SILICON   | PHOSPHORUS | SULPHUR     | CHLORINE    | ARGON       | POTASSIUM  | CALCIUM     | SCANDIUM   | TITANIUM     |
|  |  |  |  |  |  | 31          | 32        | 33         | 34          | 35          | 36          | 37         | 38          | 39         | 40           |
|  |  |  |  |  |  | 69.723      | 72.64     | 74.922     | 78.971      | 79.904      | 83.798      | 85.468     | 87.62       | 88.906     | 91.224       |
|  |  |  |  |  |  | Ga          | Ge        | As         | Se          | Br          | Kr          | Rb         | Sr          | Y          | Zr           |
|  |  |  |  |  |  | GALLIUM     | GERMANIUM | ARSENIC    | SELENIUM    | BROMINE     | KRYPTON     | ROBBIUM    | STRONTIUM   | YTRBIUM    | ZIRCONIUM    |
|  |  |  |  |  |  | 49          | 50        | 51         | 52          | 53          | 54          | 55         | 56          | 57         | 58           |
|  |  |  |  |  |  | 114.82      | 118.71    | 121.76     | 127.60      | 126.90      | 131.29      | 132.905    | 137.327     | 140.908    | 144.913      |
|  |  |  |  |  |  | In          | Sn        | Sb         | Te          | I           | Xe          | Cs         | Ba          | La         | Ce           |
|  |  |  |  |  |  | INDIUM      | TIN       | ANTIMONY   | TELLURIUM   | IODINE      | XENON       | CAESIUM    | BARIUM      | LANTHANUM  | CERIUM       |
|  |  |  |  |  |  | 81          | 82        | 83         | 84          | 85          | 86          | 87         | 88          | 89         | 90           |
|  |  |  |  |  |  | 204.38      | 207.2     | 208.98     | (209)       | (210)       | (222)       | 223        | 226         | 227        | 232.0377     |
|  |  |  |  |  |  | Tl          | Pb        | Bi         | Po          | At          | Rn          | Fr         | Ra          | Ac         | Th           |
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|  |  |  |  |  |  | 112         | 113       | 114        | 115         | 116         | 117         | 118        | 119         | 120        | 121          |
|  |  |  |  |  |  | (285)       | (...)     | (287)      | (...)       | (291)       | (...)       | (...)      | (...)       | (...)      | (...)        |
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Data source: high throughput DFT (FHI-aims, Carlos Mera Acosta)





# Perovskites' stability: Improving on Goldschmidt Tolerance Factor



$$t = \frac{r_A + r_X}{\sqrt{2}(r_B + r_X)}$$

→ Ionic radius

Goldschmidt\* **stable** perovskites:

accuracy 74%

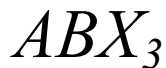
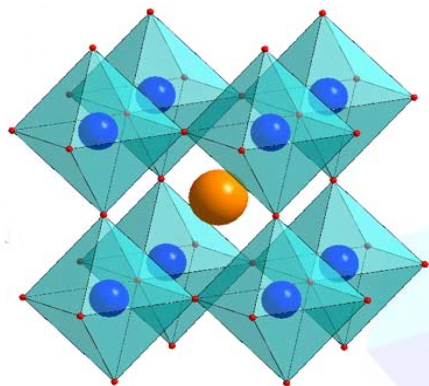
$t = 0.825$

$t = 1.059$

$t$



# Perovskites' stability: Improving on Goldschmidt Tolerance Factor



$$t = \frac{r_A + r_X}{\sqrt{2}(r_B + r_X)}$$

→ Ionic radius

$$\tau = \frac{r_X}{r_B} - n_A \left( n_A - \frac{r_A/r_B}{\ln(r_A/r_B)} \right)$$

→ Oxidation state

→  $1/\mu$  = Octahedral factor

Goldschmidt\* **stable** perovskites:

accuracy 74%

$t = 0.825$

$t = 1.059$

$t$

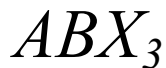
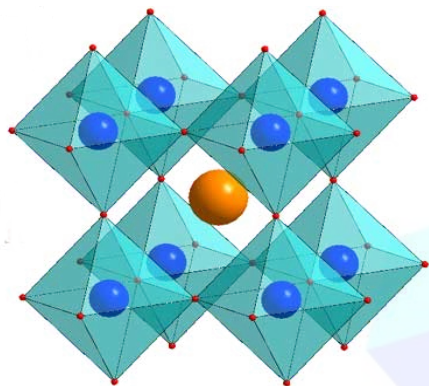
Our **stable** perovskites:

accuracy 92%

$\tau = 4.18$

$\tau$

# Perovskites' stability: Improving on Goldschmidt Tolerance Factor



$$t = \frac{r_A + r_X}{\sqrt{2}(r_B + r_X)}$$

→ Ionic radius

$$\tau = \frac{r_X}{r_B} - n_A \left( n_A - \frac{r_A/r_B}{\ln(r_A/r_B)} \right)$$

→ Oxidation state

→  $1/\mu$  = Octahedral factor

Goldschmidt\* **stable** perovskites:

accuracy 74%

$t = 0.825$

$t = 1.059$

$t$

Our **stable** perovskites:

accuracy 99%

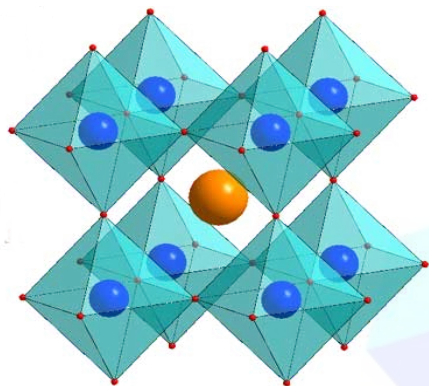
$\tau = 3.31$

$\tau = 4.18$

$\tau = 5.92$

$\tau$

# Perovskites' stability: Improving on Goldschmidt Tolerance Factor



$$t = \frac{r_A + r_X}{\sqrt{2}(r_B + r_X)}$$

Ionic radius

$$\tau = \frac{r_X}{r_B} - n_A \left( n_A - \frac{r_A/r_B}{\ln(r_A/r_B)} \right)$$

Oxidation state

1 /  $\mu$  = Octahedral factor

Goldschmidt\* **stable** perovskites:

accuracy 74%

$t = 0.825$

$t = 1.059$

$t$

Our **stable** perovskites:

accuracy 100%

$\tau = 3.31$

$\tau = 4.18$

$\tau = 5.92$

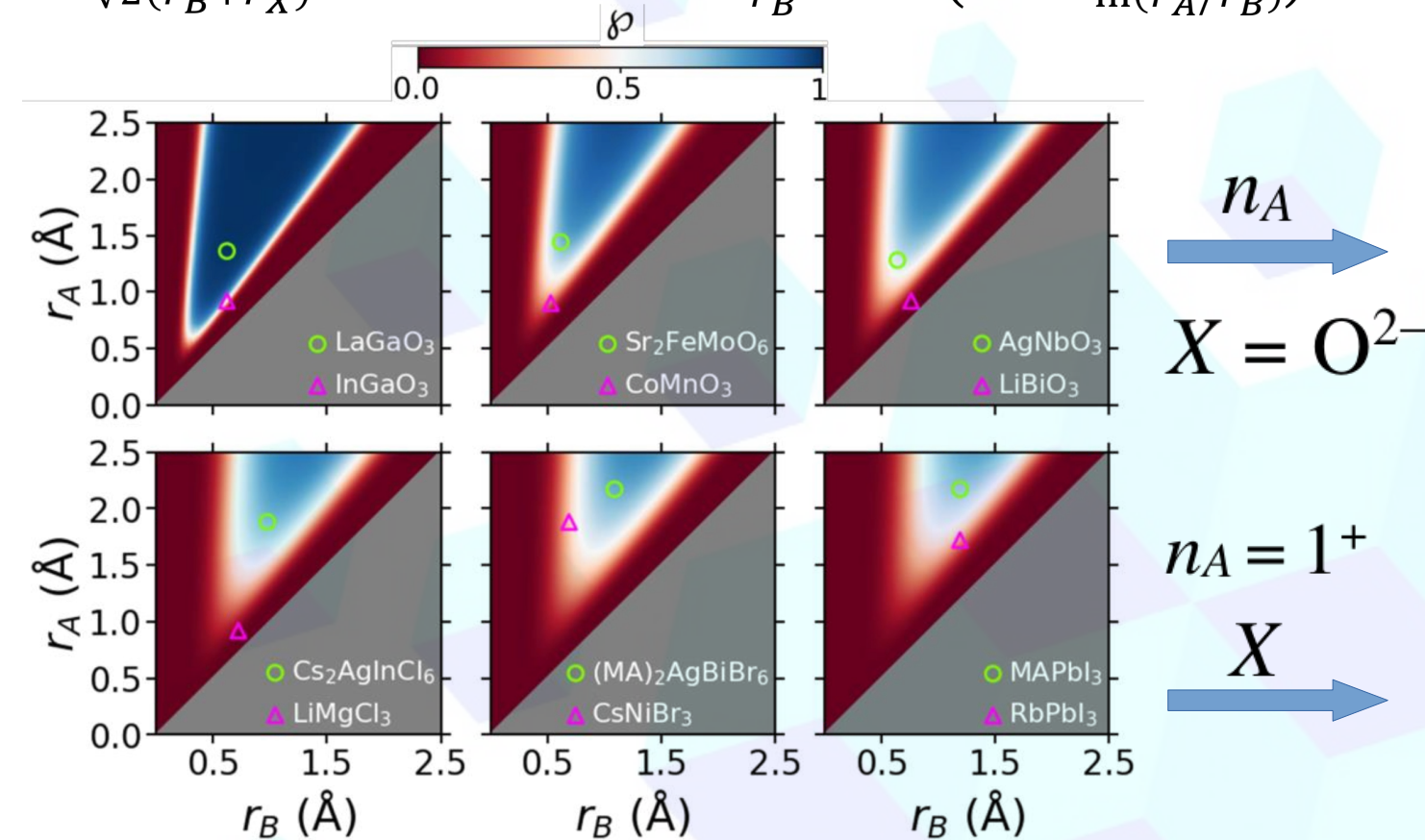
$\tau = 12.08$

$\tau$

# Perovskites' stability: Improving on Goldschmidt Tolerance Factor

NOVEL MATERIALS DISCOVERY

$$t = \frac{r_A + r_X}{\sqrt{2}(r_B + r_X)} \longrightarrow \tau = \frac{r_X}{r_B} - n_A \left( n_A - \frac{r_A/r_B}{\ln(r_A/r_B)} \right)$$



# Perovskites' stability: Improving on Goldschmidt Tolerance Factor

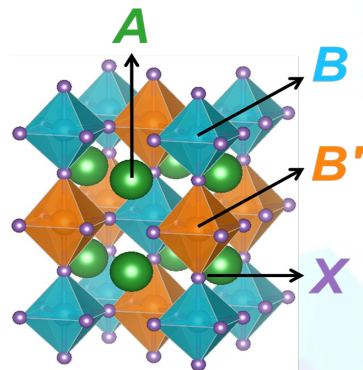
NOVEL MATERIALS DISCOVERY

$$t = \frac{r_A + r_X}{\sqrt{2}(r_B + r_X)}$$

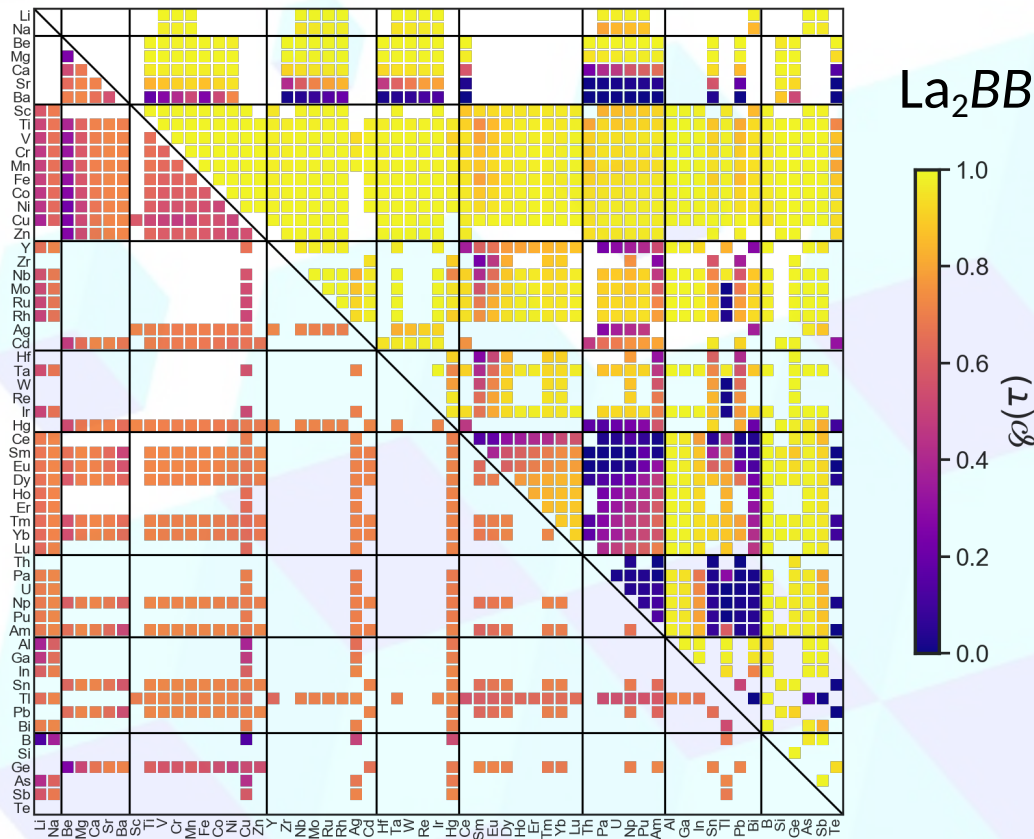


$$\tau = \frac{r_X}{r_B} - n_A \left( n_A - \frac{r_A/r_B}{\ln(r_A/r_B)} \right)$$

Double  
perovskites



$$r_B \rightarrow (r_B + r_{B'})/2$$





## Continuous property

Adsorption energy and OCO angle of adsorbed CO<sub>2</sub> on metal-oxide surfaces

Adsorption energy of O on metal-oxide surfaces

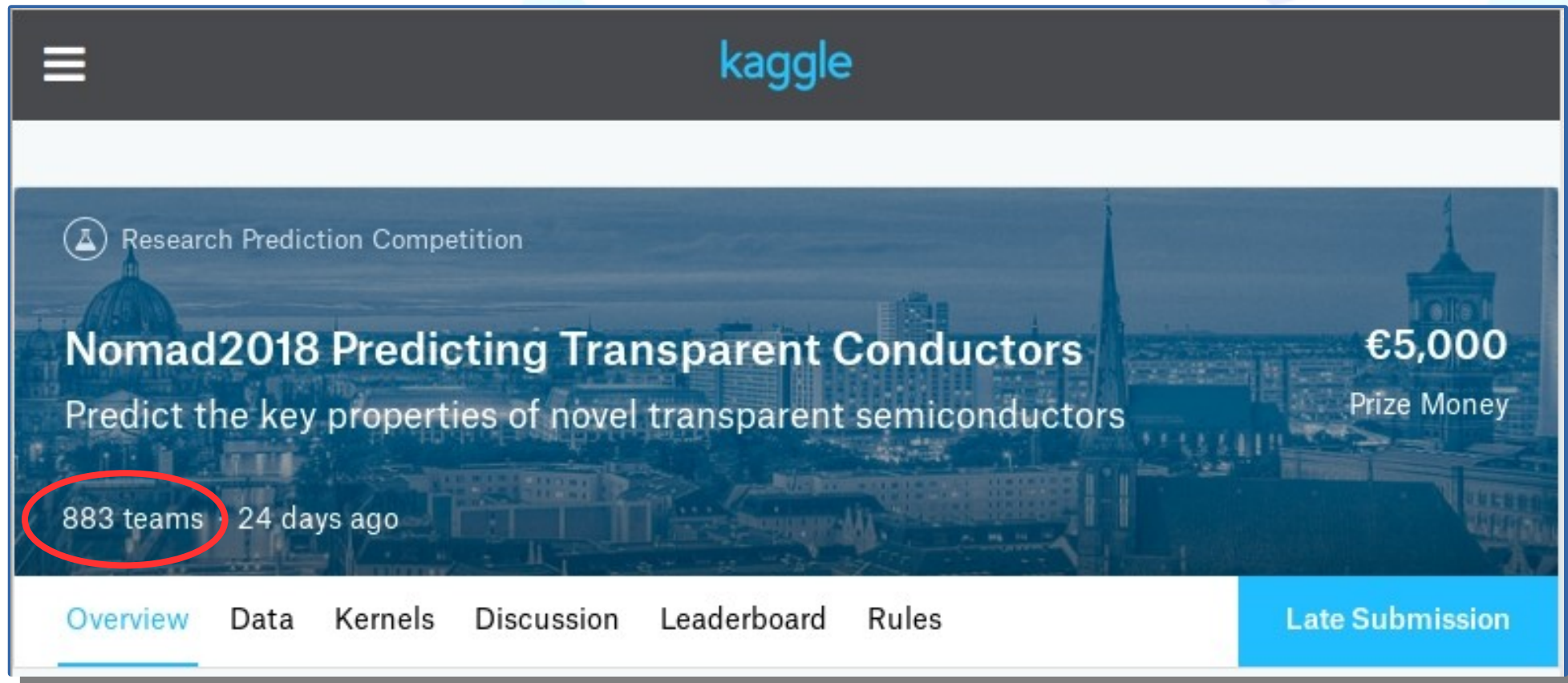
Adsorption energy of metal atoms on metal-alloys surfaces

Features: atoms (of the surface) and pristine surface

## Classification

Tetradymite 5-component 3d topological insulators (vs trivial insulators)





The image is a screenshot of a Kaggle competition page. At the top, there is a dark header with the Kaggle logo in teal. Below the header, the page features a large banner with a cityscape background. On the left side of the banner, there is a circular icon with a flask and the text "Research Prediction Competition". The main title of the competition is "Nomad2018 Predicting Transparent Conductors" in large white font. Below the title, the description "Predict the key properties of novel transparent semiconductors" is written in a smaller white font. To the right of the title, the prize money "€5,000" is displayed in large white font, with "Prize Money" written below it in a smaller font. In the bottom left corner of the banner, the text "883 teams" is circled in red, followed by "24 days ago". At the bottom of the page, there is a navigation bar with links: "Overview", "Data", "Kernels", "Discussion", "Leaderboard", and "Rules". The "Overview" link is underlined. On the far right of the navigation bar, there is a blue button labeled "Late Submission".

Research Prediction Competition

# Nomad2018 Predicting Transparent Conductors

Predict the key properties of novel transparent semiconductors

€5,000  
Prize Money

883 teams · 24 days ago

[Overview](#) [Data](#) [Kernels](#) [Discussion](#) [Leaderboard](#) [Rules](#) [Late Submission](#)



- Materials class:  $\text{Al}_x\text{Ga}_y\text{In}_z\text{O}_3$ ,  $x + y + z = 3$ , in 6 crystal prototypes (C2/m (12), Pna21 (33), R-3c (167), p63/mmc (194), Ia-3 (206), and Fd-3m, up to 96 atoms per unit cell
- Properties: Formation energy and band gap
- **Input: Vegard's-law lattice vectors and coordinates**
- ~3000 data points (2400 training + 600 validation set) calculated via PBE, by FHI-aims + Christopher Sutton





# Crowd sourcing data-driven materials science: [nomad-coe.eu/outreach/nomad-kaggle-2018](http://nomad-coe.eu/outreach/nomad-kaggle-2018)

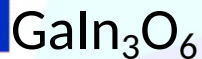
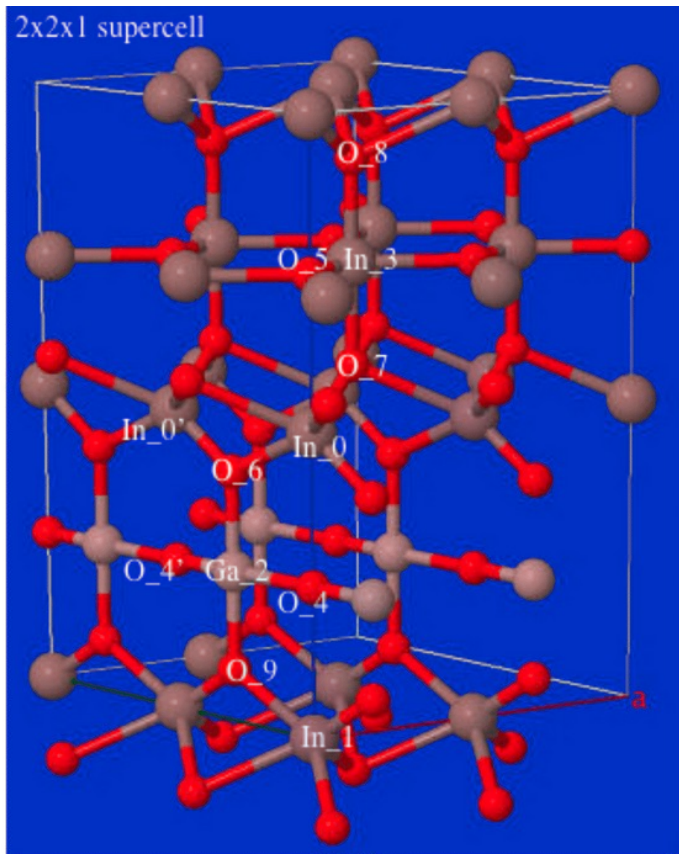
NOVEL MATERIALS DISCOVERY

1st **Takenori Yamamoto (Tony Y.)**  
Descriptor: Crystal-graph  $n$ -grams  
ML method: Kernel ridge regression

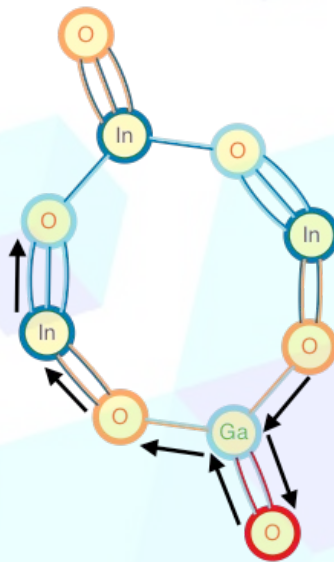
Prediction accuracy  
Formation energy: ~20 meV/cation  
Band gap: ~80 meV

2nd **Dr. Yury Lysogorskiy**  
Descriptor: 7000 average and local-order scalar features  
ML method: Gradient boosting regression trees

3rd **Lars Blumenthal**  
Descriptor: based on Smooth Overlap of Atomic Positions (SOAP)  
ML method: Neural network



**Crystal Graph**

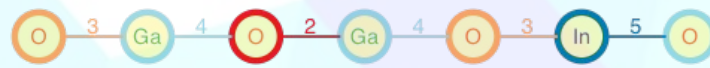


$n$ -grams

**Coordination Number**



**Path Graph**



**Sequence of N-gram Items**

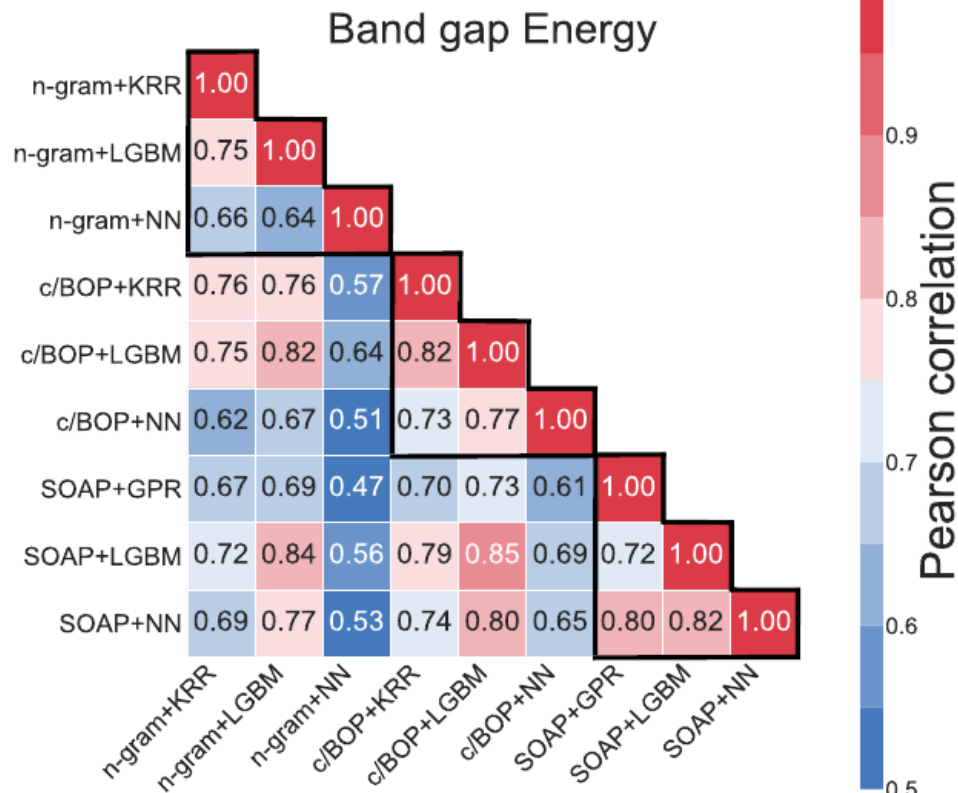
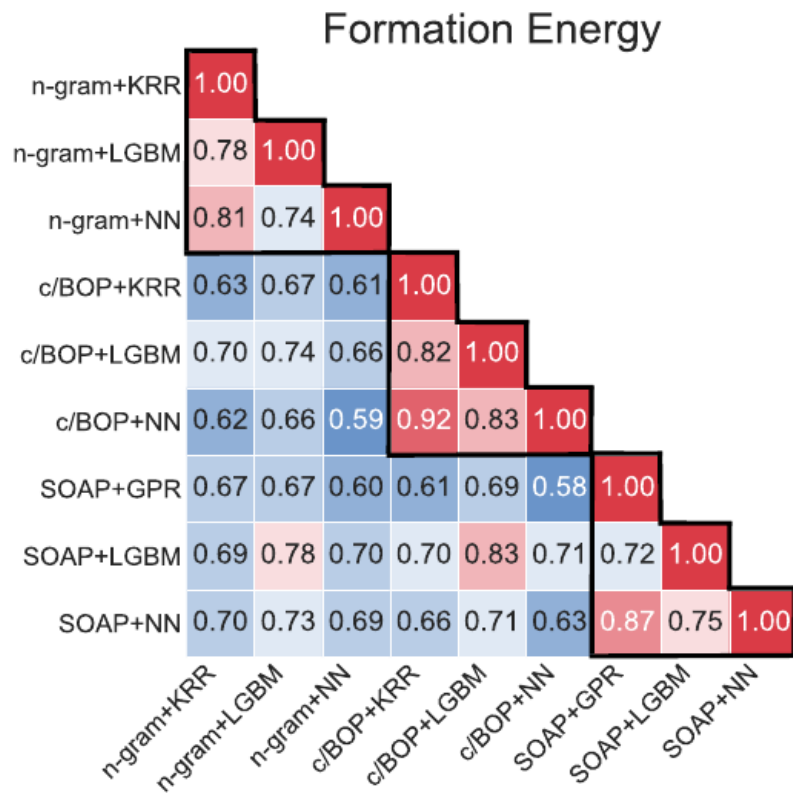






# Crowd sourcing data-driven materials science: The importance of the descriptor

NOVEL MATERIALS DISCOVERY



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NOVEL MATERIALS DISCOVERY

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NOMAD REPOSITORY



THE ARCHIVE



ENCYCLOPEDIA



BIG-DATA ANALYTICS



ADVANCED GRAPHICS



HPC INFRASTRUCTURE



OUTREACH

*The Novel Materials Discovery (NOMAD) Laboratory* maintains the largest Repository, for input and output files of all important computational materials science codes.

From its open-access data, it builds several *Big-Data Services* helping to advance materials science and engineering.

To learn more, click on the buttons above. You can also watch our 3-minute summary on the *NOMAD Laboratory CoE* at [YouTube](#) (or at [YOUKU](#) in China).

## NOMAD Scope and Overview

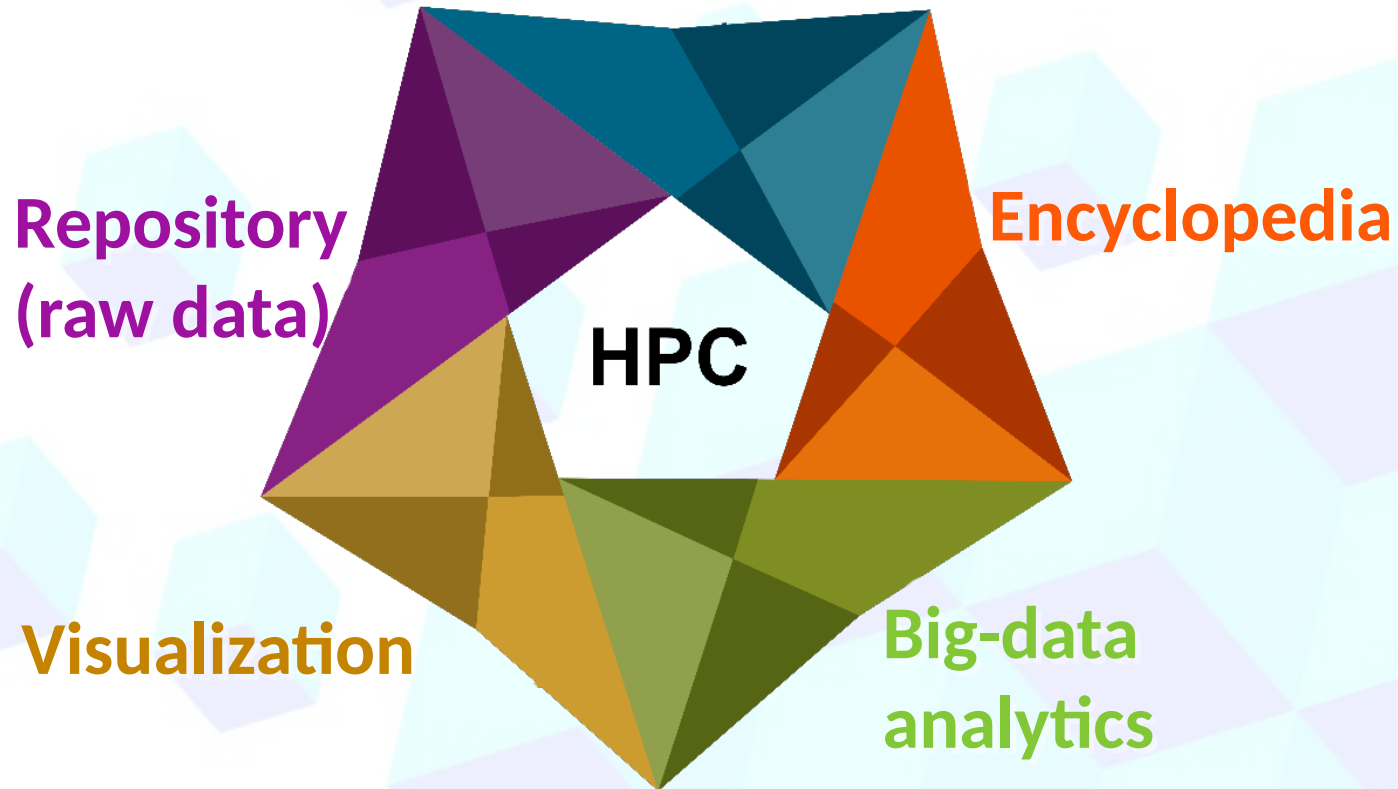
### NOMAD Success Stories

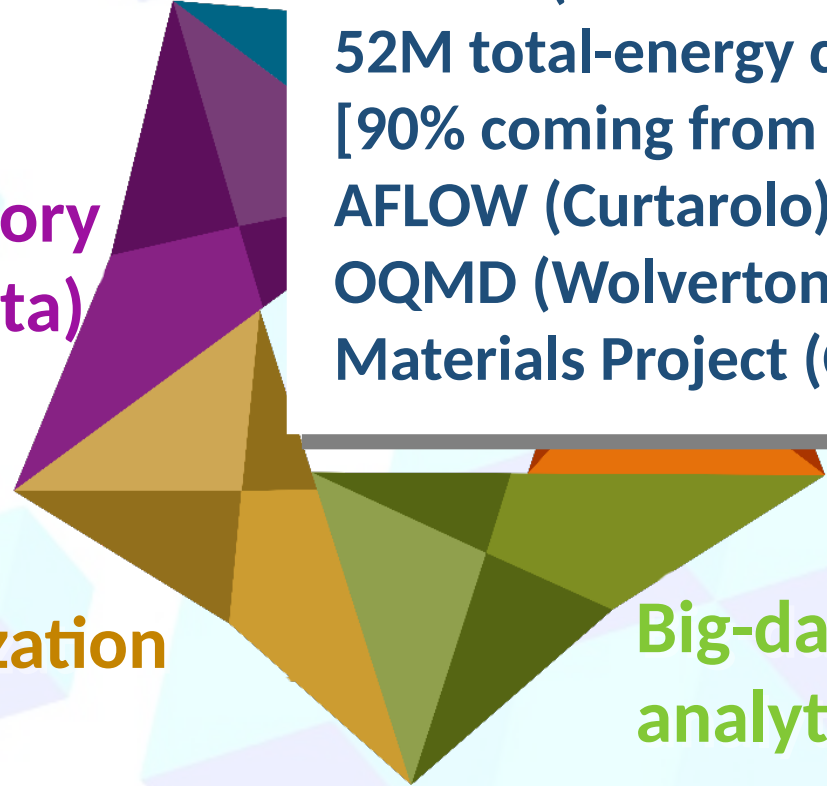
**Finding local patterns and structure**  
in big-data of materials-science  
remains a challenge



**New data mining tools must be developed**  
to help uncover hidden relations  
in materials-science data







Repository  
(raw data)

Archive (code independent)  
52M total-energy calculations  
[90% coming from  
AFLOW (Curtarolo)  
OQMD (Wolverton)  
Materials Project (Ceder)]

Visualization

Big-data  
analytics



*The Novel Materials Discovery (NOMAD) Laboratory* maintains the largest Repository, for input and output files of all important computational materials science codes.

From its open-access data, it builds several **Big-Data Services** helping to advance materials science and engineering.

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## NOMAD Scope and Overview

### NOMAD Success Stories

**Finding local patterns and structure**  
in big-data of materials-science  
remains a challenge



**New data mining tools must be developed**  
to help uncover hidden relations  
in materials-science data

## Materials property prediction

Predicting the metal-insulator classification of elements and binary systems

Prediction of topological quantum phase transitions



## NOMAD Analytics Toolkit



### — Prediction of topological quantum phase transitions

created by: Carlos Mera Acosta<sup>1,2</sup>, Emre Ahmetcik<sup>1</sup>, Adalberto Fazzio<sup>2</sup>, Luca Ghiringhelli<sup>1</sup>, and Matthias Scheffler<sup>1</sup>

<sup>1</sup> Fritz Haber Institute of the Max Planck Society, Faradayweg 4-6, D-14195 Berlin, Germany

<sup>2</sup> Brazilian Nanotechnology National Laboratory, Campinas, SP, Brazil

v1.0.0  
[Last updated: November 9, 2017]

[Back to Analytics Home](#)

[Back to NOMAD CoE Home](#)

**SISSO and Multi-Task SISSO**

Runhai Ouyang, Emre Ahmetcik, Stefano Curtarolo, Christian Carbogno

**Topological Insulators**

Carlos Mera Acosta, Adalberto Fazzio, Runhai Ouyang, Christian Carbogno

**Perovskites**

Christopher J. Bartel, Christopher Sutton, Bryan R. Goldsmith, Runhai Ouyang, Charles B. Musgrave

**NOMAD-Kaggle competition**

Christopher Sutton, Angelo Ziletti, Xiangyue Liu

**NOMAD PIs**

Angel Rubio (MPSD Hamburg), Risto Nieminen (Aalto Univ. Helsinki), Francesc Illas (Univ. Barcelona), Daan Frenkel (Univ. Cambridge), Claudia Draxl (HU Berlin), Alessandro De Vita (Kings College London), Kristian Thygesen (DTU Lyngby); Kimmo Koski (CSC Helsinki), Stefan Heinzel (MPSCD Garching), Jose Maria Cela (BSC Barcelona), Dieter Kranzlmüller (LRZ Munich); Ciaran Clissman (Pintail Dublin)

**All the above**

Matthias Scheffler



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