

Neural Network Potentials to explore the Crystal Structure Landscape

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Credits:

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PANNA

Properties from Artificial Neural Network Architectures

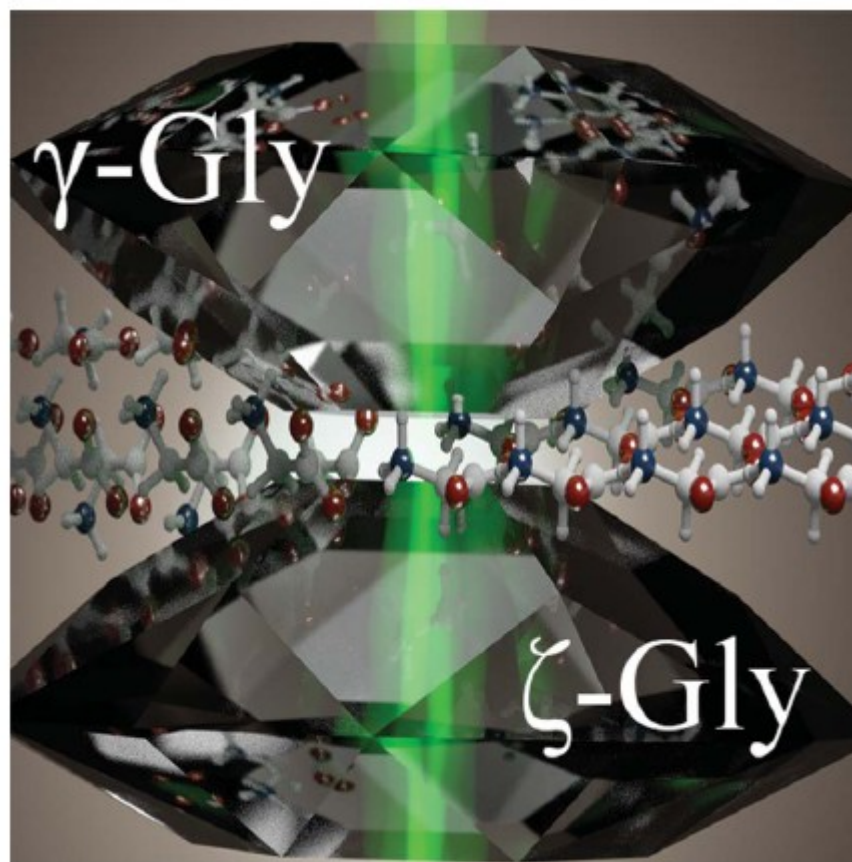
<https://gitlab.com/pannadevs/panna>



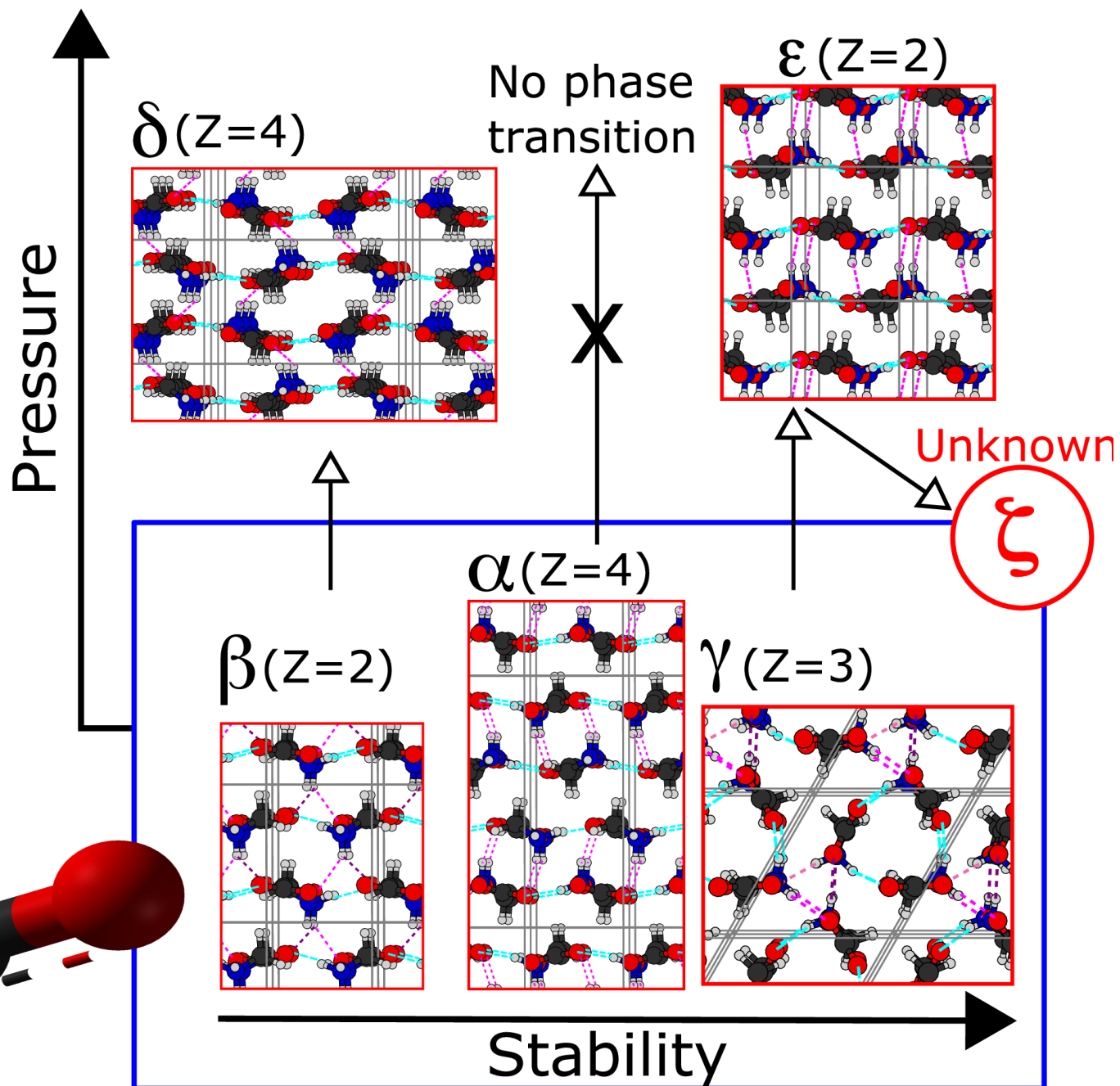
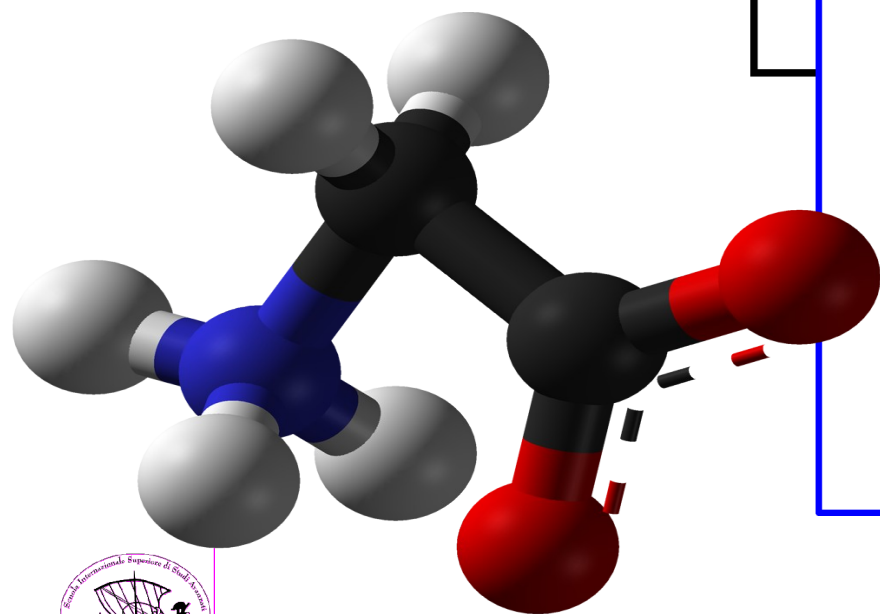
CSP is a formidable task

- CSP problem: Name a chemical or stoichiometric formula; find the (local) minima of the free energy landscape under given thermodynamic conditions (often at certain T,P)
- “What is the most stable structure of glycine at ambient conditions?” “What is the carbon structure that is stable at very high pressures”
- Challenges:
 - A very vast space of possibilities.
 - Free energy landscape is very expensive to obtain accurately

ζ -Glycine: Insight into the mechanism of a polymorphic phase transition



Glycine



How to tackle CSP?

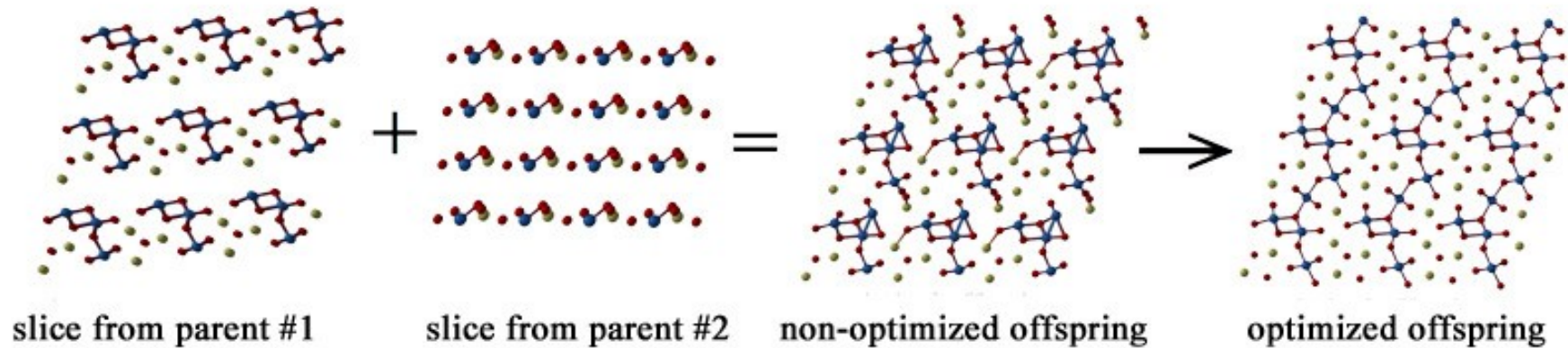
Explore: Use smart algorithms to explore as much of the landscape as possible

Molecular dynamics / Monte Carlo walkers

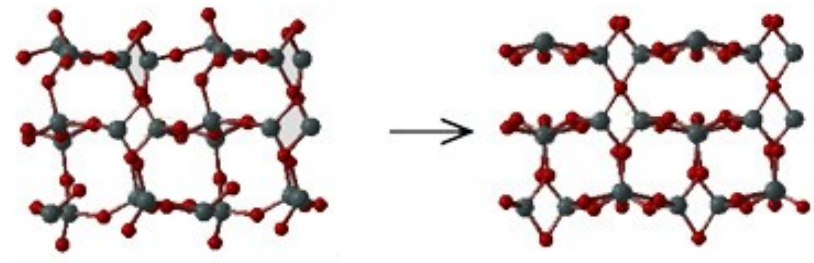
- Simulated annealing
 - Metadynamics
 - Basin hopping
 - Minima hopping
 - Genetic algorithm
-

Genetic algorithm

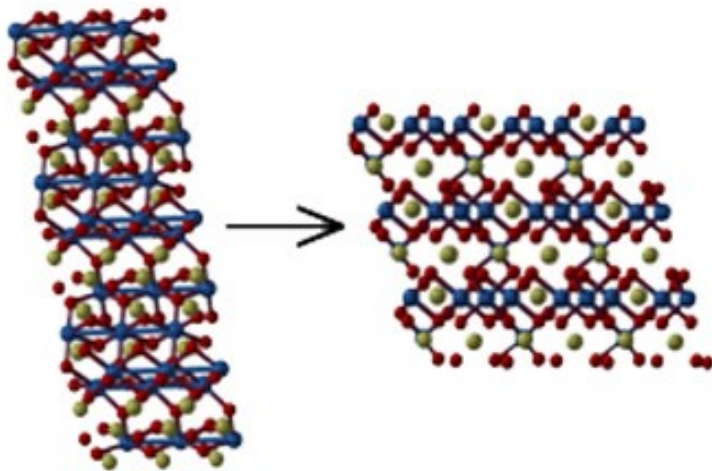
(a) heredity



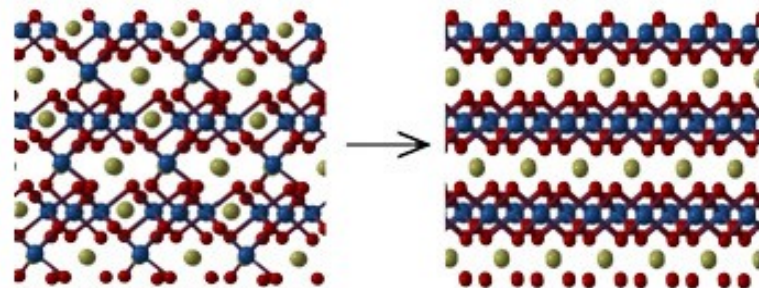
(c) softmode mutation



(b) lattice mutation



(d) permutation



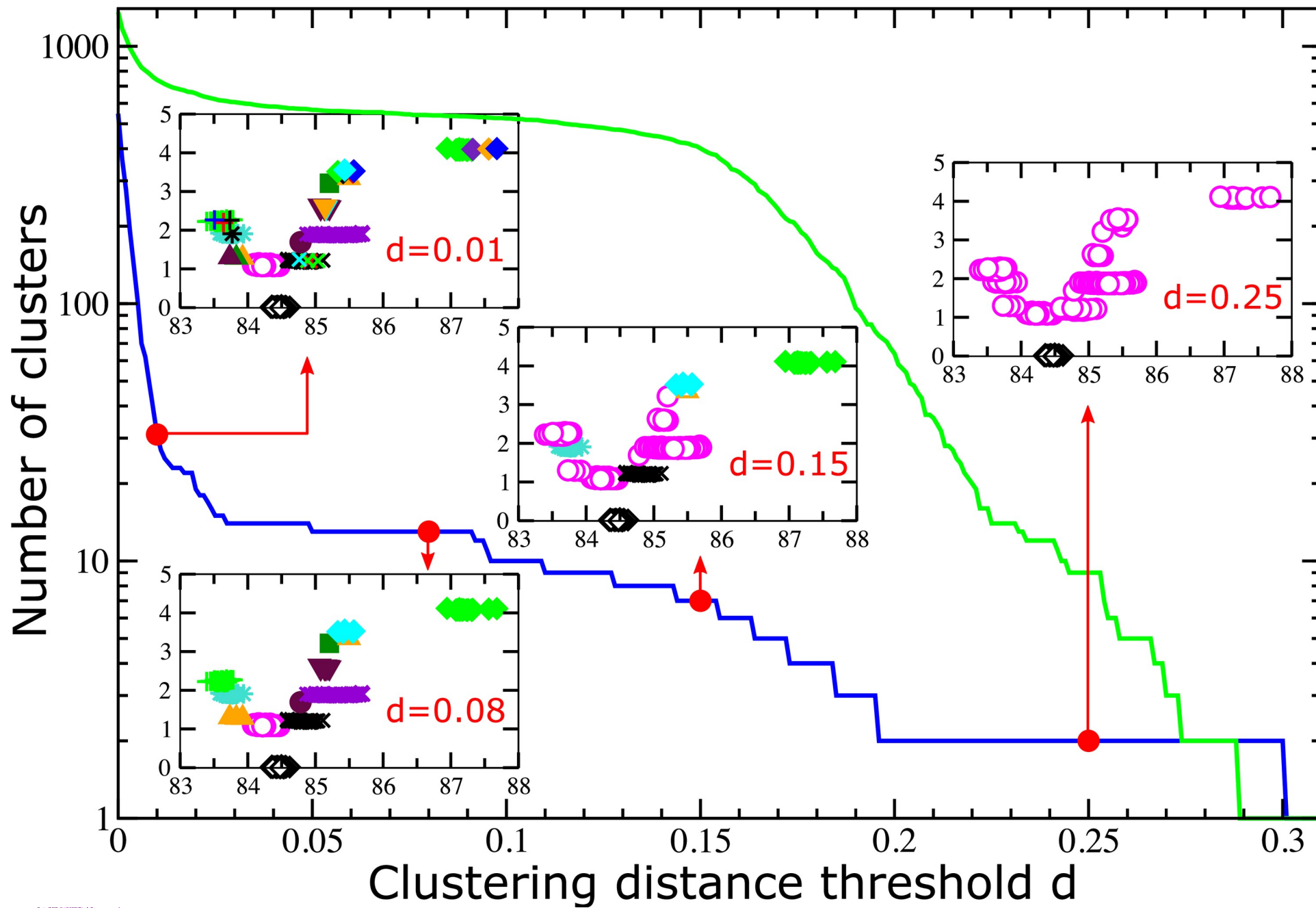
USPEX operations

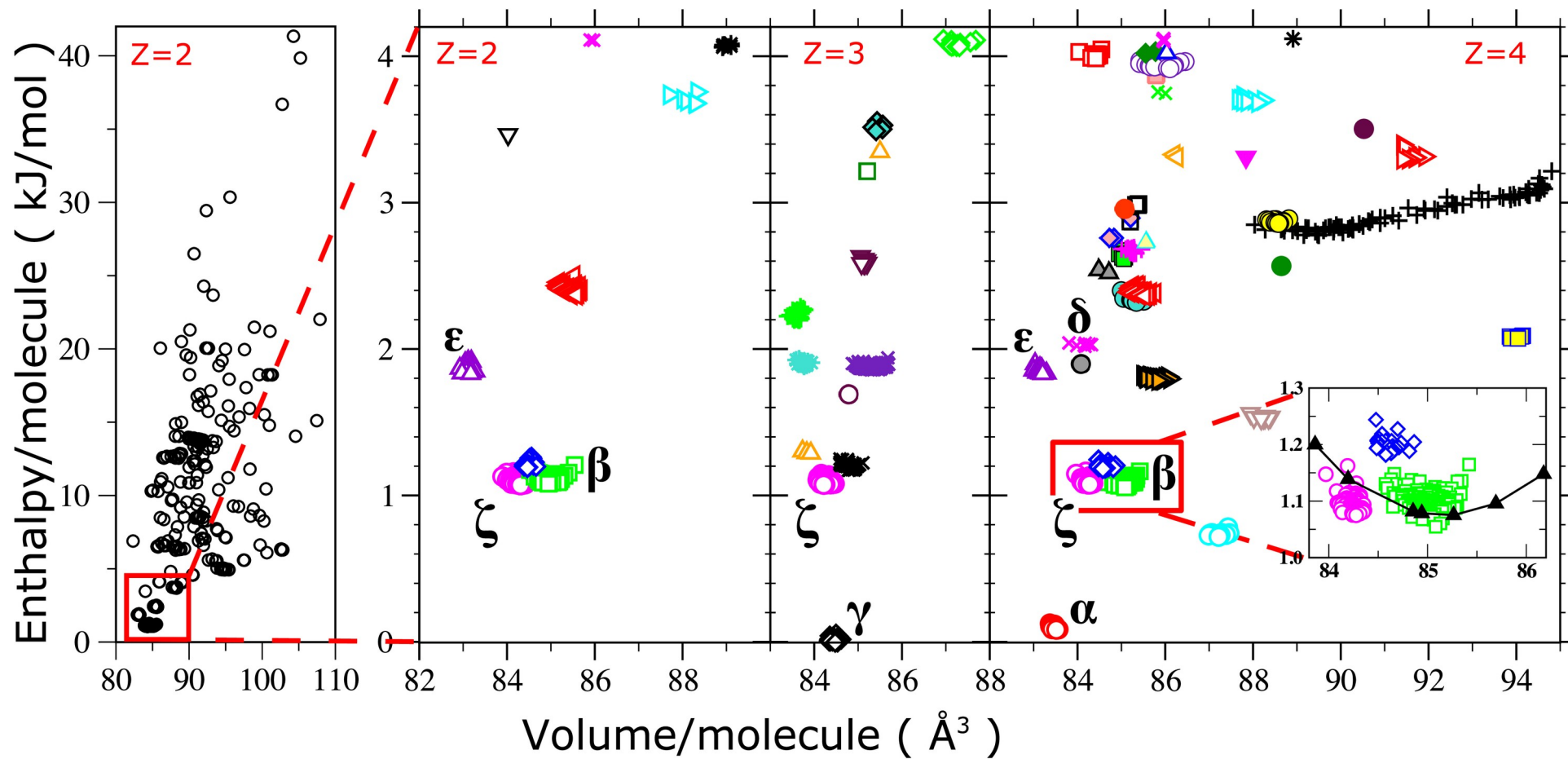


+



+ vdWDF + clustering





ζ -phase

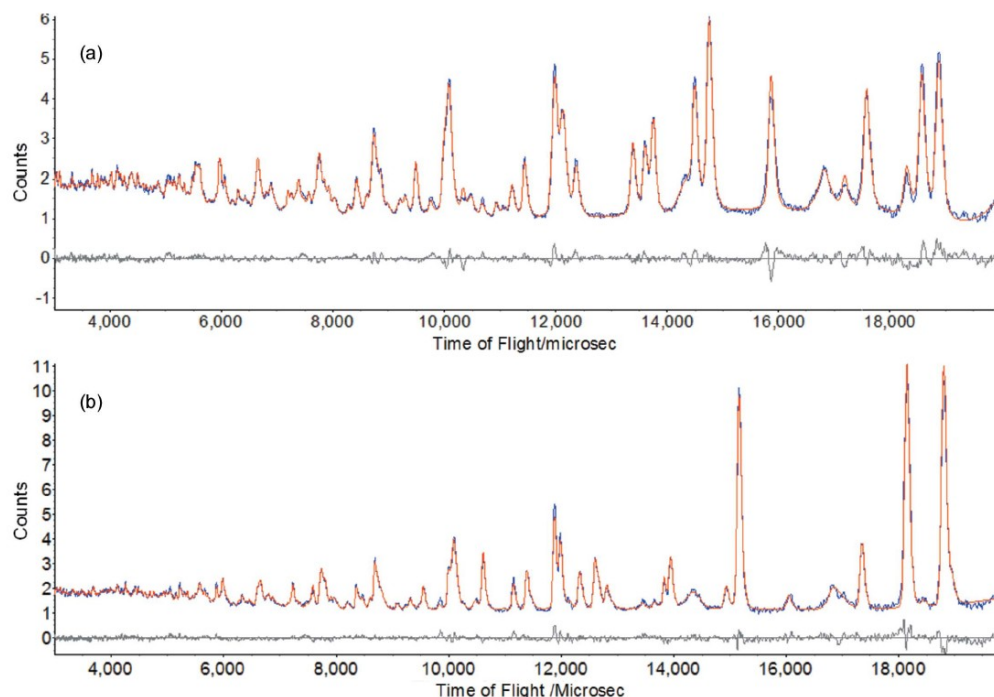


Figure 2
(a) Rietveld fit of the neutron powder diffraction pattern of ζ -glycine at 100 K (blue = observed, red = calculated). In addition to the peaks ζ -glycine, the pattern also shows the presence of residual ε - and a trace of γ -glycine. Other peaks arise from the sample environment, namely the pressure marker and the Al_2O_3 and ZrO_2 components of the anvils of the pressure cell. (b) Rietveld fit of the neutron powder diffraction pattern of β -glycine (contaminated with ζ - and a trace of γ -glycine) at 290 K. A 1 \AA d spacing approximates to 4837 \mu s in time-of-flight.

E Kucukbenli, CH Pham, SdG,
C Bull, G Flowitt-Hill, HY Playford, M Tucker, S Parsons
Int Union Crist J 4, 569–574 (2017)

Exploring the phase space for larger molecules (ex. CLR)
 requires fast and accurate energetics

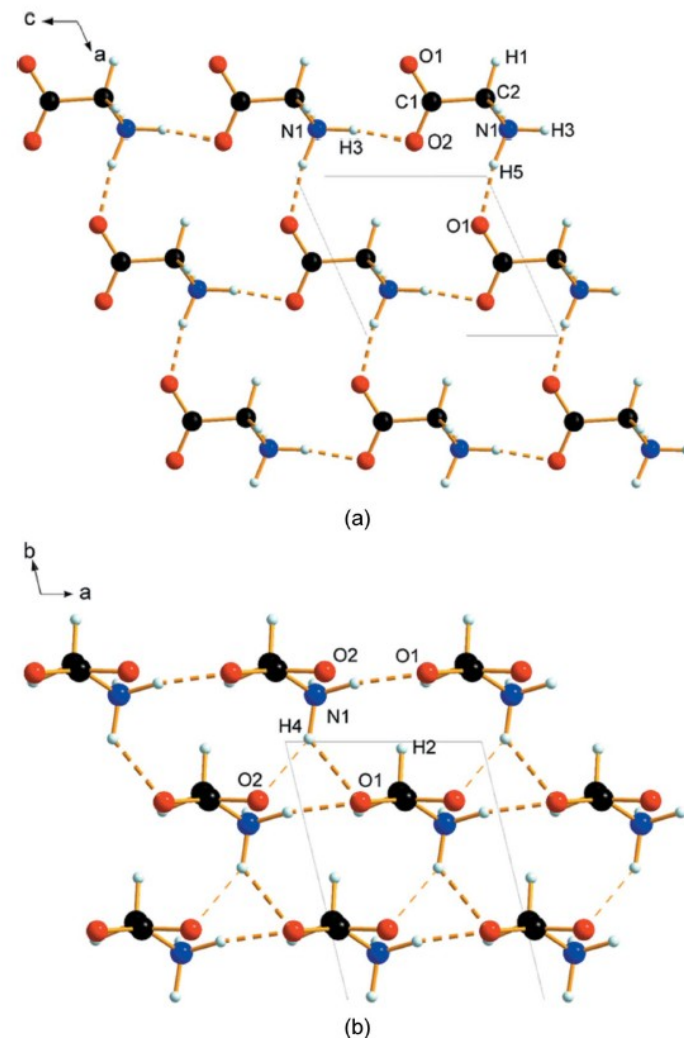
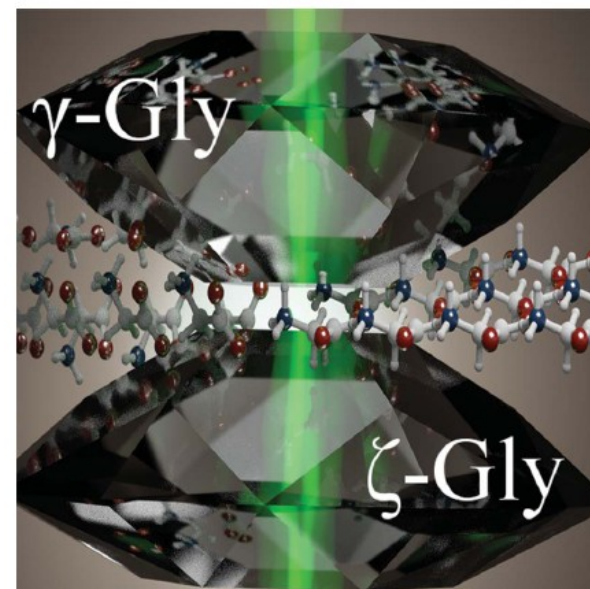


Figure 3
Intermolecular interactions in ζ -glycine. (a) Layers formed in the ac plane, viewed along b . (b) Stacking of the layers, viewed along c .

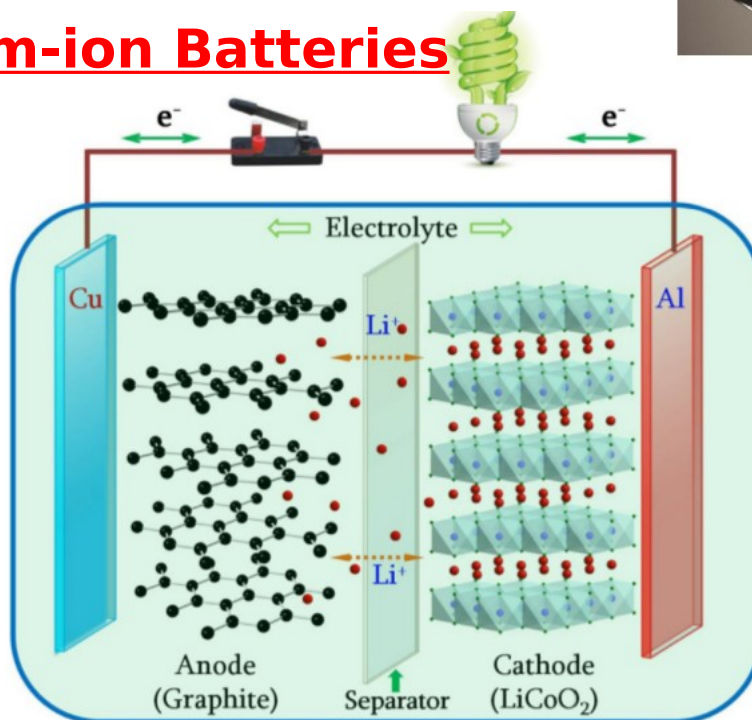
Complete ^{13}C Chemical Shift Assignment for Cholesterol Crystal



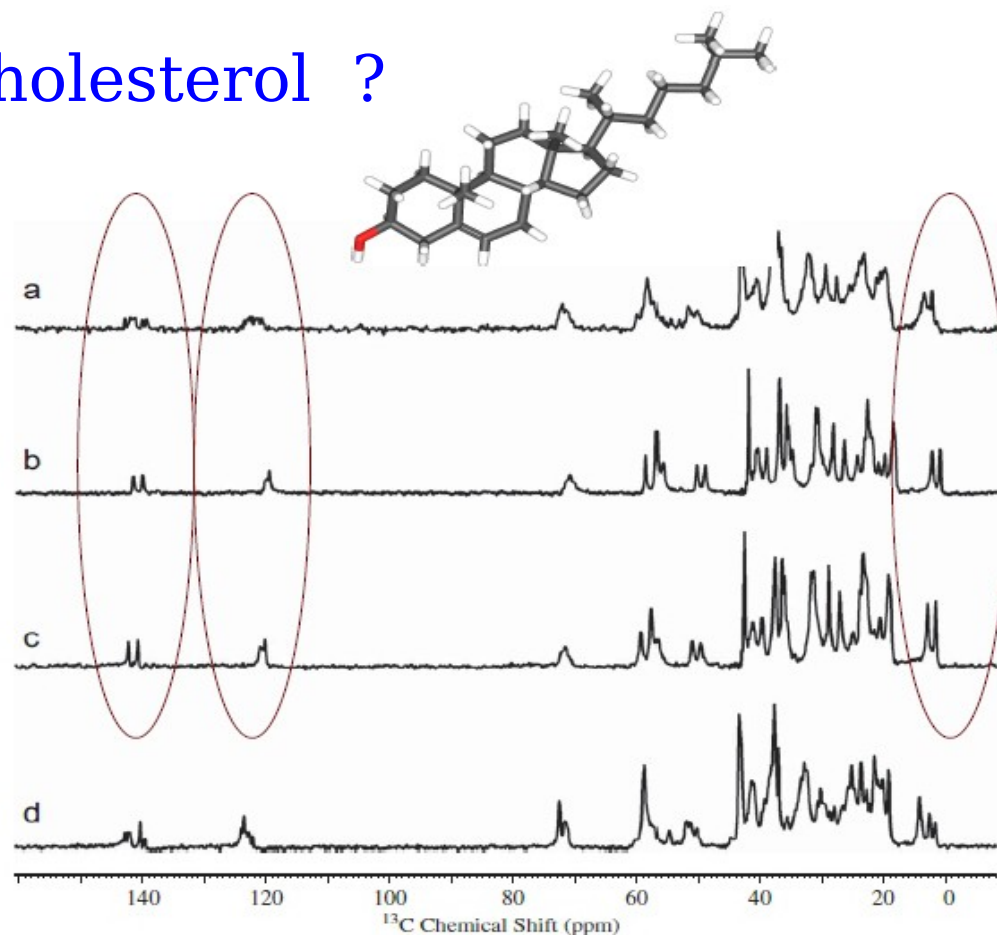
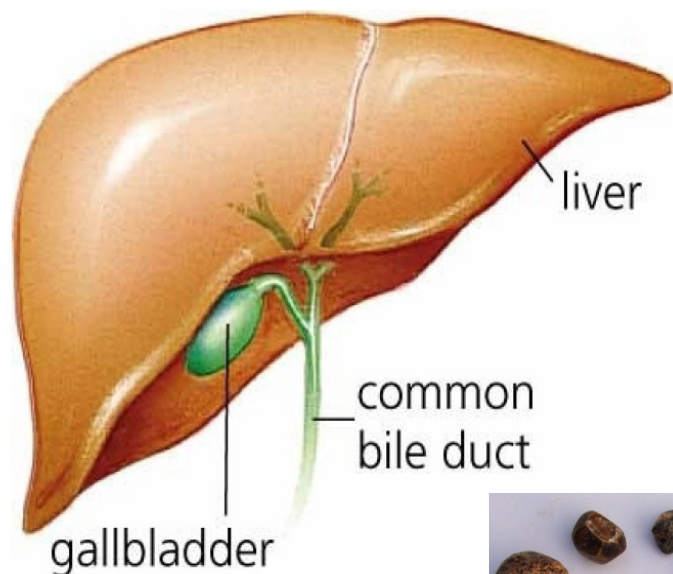
ζ -Glycine: Insight into the mechanism of a polymorphic phase transition



Lithium-ion Batteries



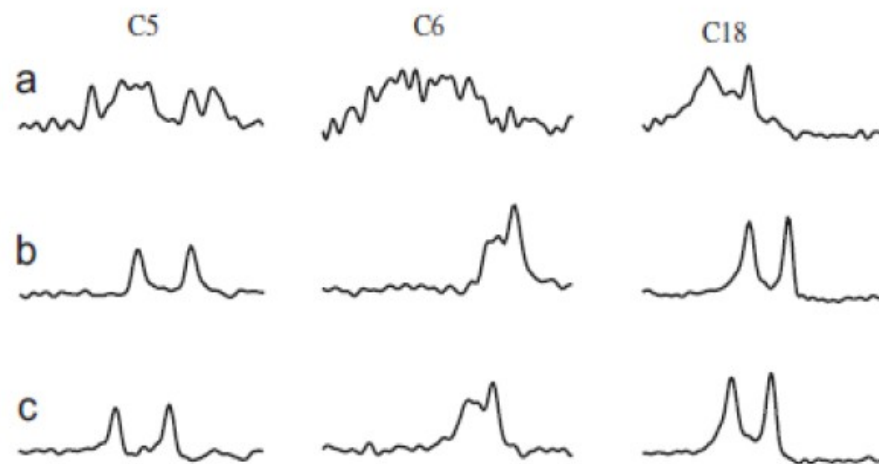
Why Cholesterol ?

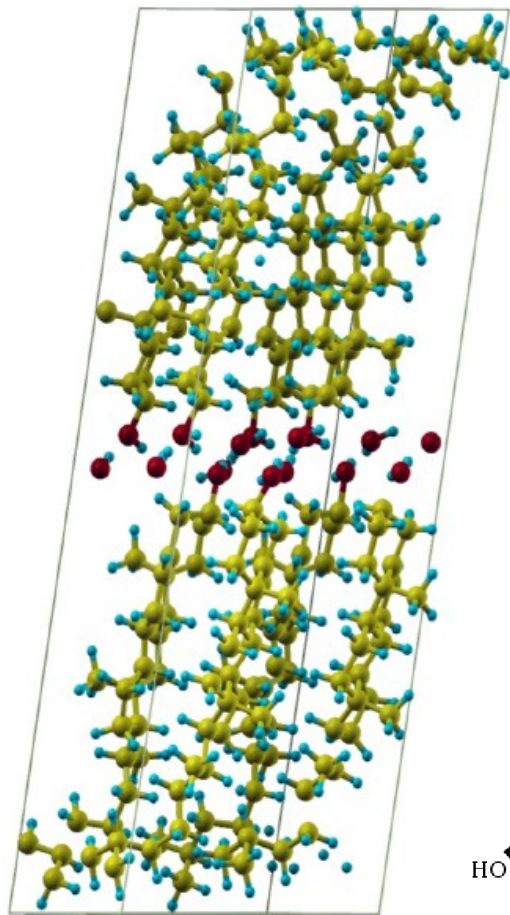


a) Gallbladder Cancer

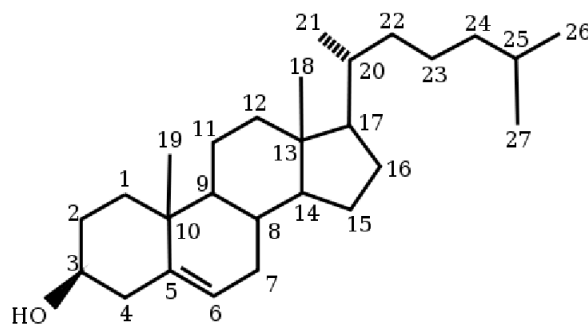
b) Chronic Cholecystitis

c) Xantho-Granulomatous Cholecystitis

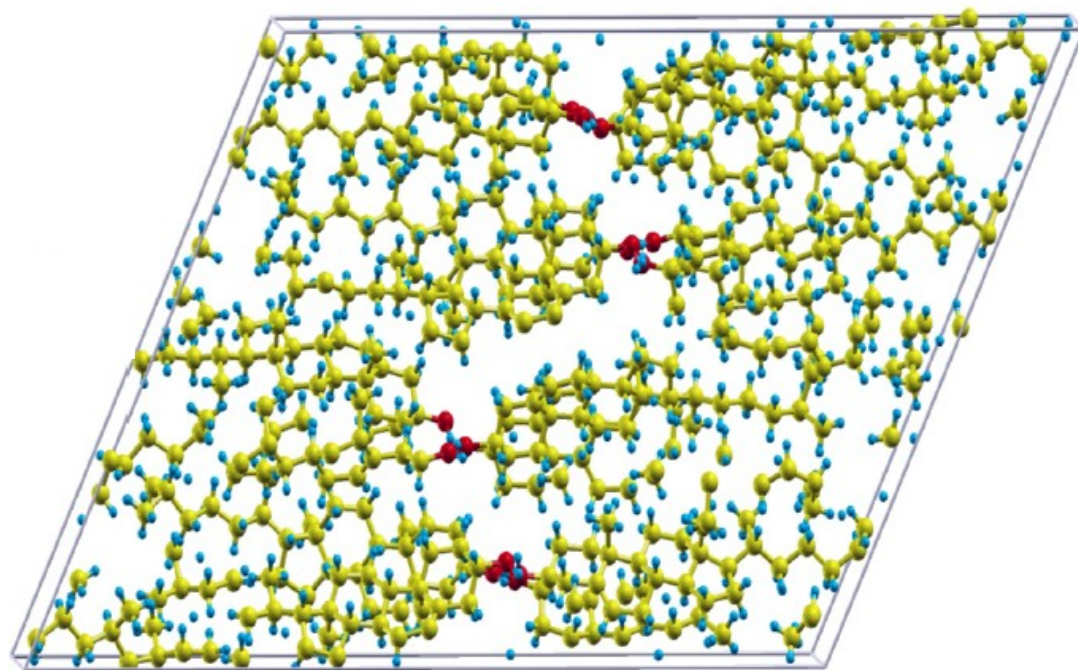




Monohydrate Cholesterol (ChM)
8 CLR + 8 w molecules - 616 atoms



High temperature Anhydrous Cholesterol (ChAh)
16 CLR mol - 1184 atoms



Low temperature Anhydrous Cholesterol (ChAl)
8 CLR mol - 592 atoms

(not shown)

Lithium Interaction with Graphene-like Materials



~10Wh

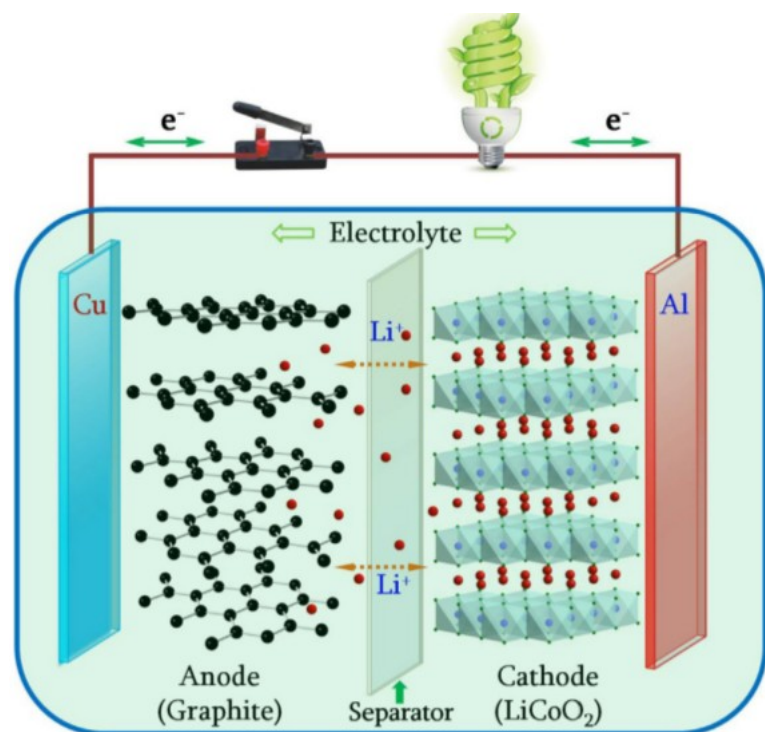


~100Wh



~10,000Wh

Lithium ion batteries



- Cathode: Source of lithium
- Electrolyte: Ionic conductivity
- Anode: Lithium holder
- Current collectors

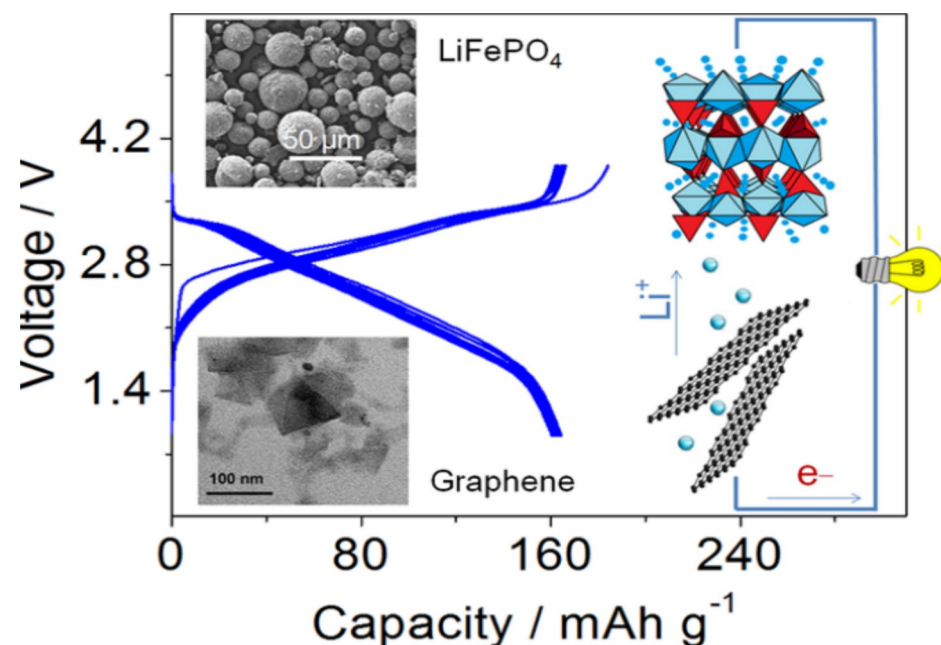
Materials Today 19(2):109-123, 2016

Capacity: The amount of Li absorbed by anode

- Stoichiometry of Li adsorbed graphite is LiC_6

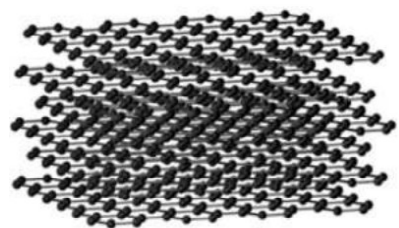
Alternative anode materials :

Graphene due to its large surface to mass ratio and good electrical conductivity.

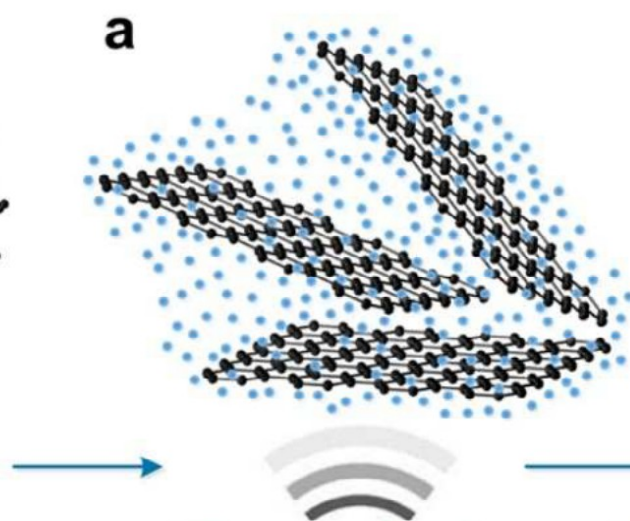


- graphene nanoflakes as alternative anode
- Flakes ~30-100 nm lateral dimension
- **Very high Li uptake: LiC_2**

Hassoun et al. Nano Lett. 2014, 14, 4901-4906



Graphite



Ultrasonication



Ultracentrifugation



Graphene
ink

Traditionally model potentials construction requires a lot of physical intuition and are strongly dependent on the available experimental information.

Not transferable to experimentally unexplored regions.


Limited accuracy due to rigid functional form.

DFT is a viable option to gather accurate information but requires a systematic approach to build a potential that can incorporate its features.



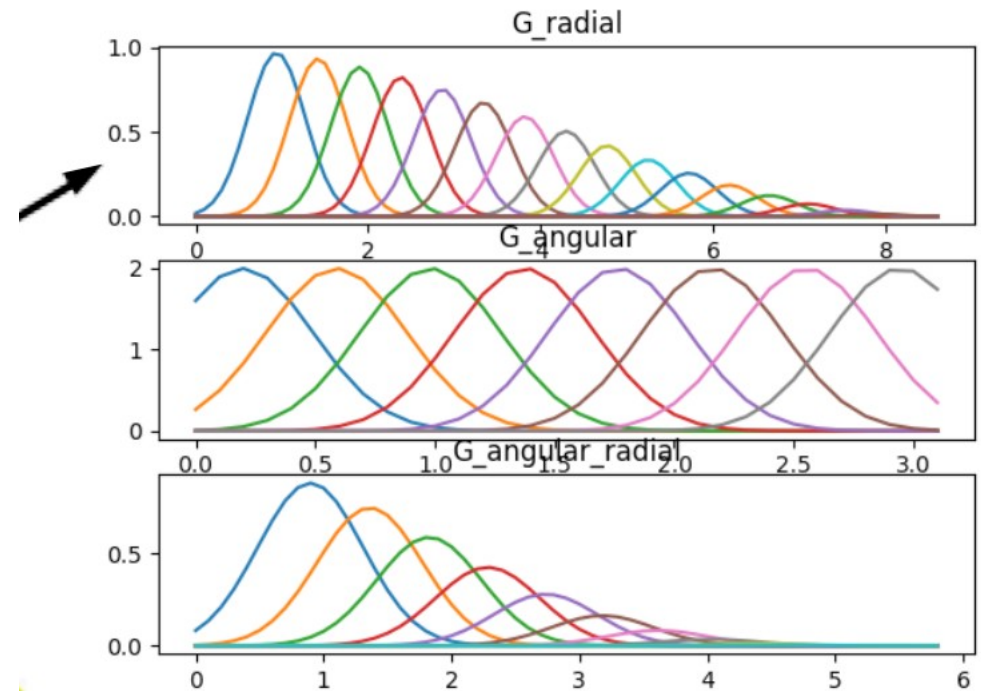
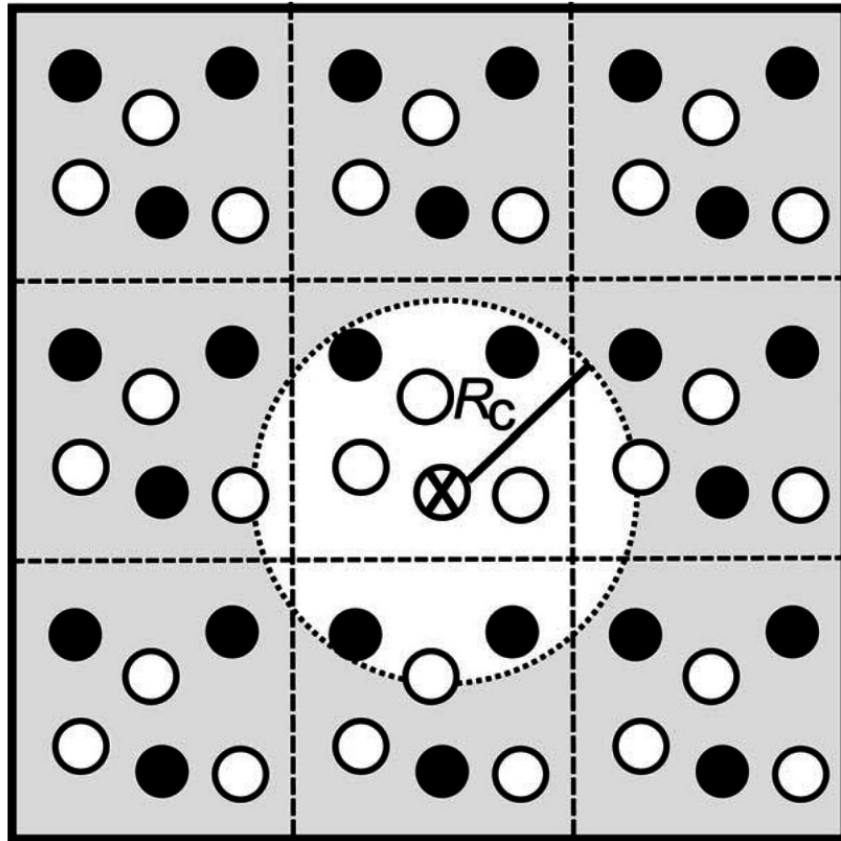
**Artificial
Intelligence**
is better than non

Replace the expensive DFT total energy calculations (or other accurate methods) with an interatomic potentials built to reproduce DFT data in a variety of environments

$$E(c) = \sum_{\alpha} \sum_{i \in \alpha} \varepsilon_{\alpha}(\mathbf{d}_i) + \text{long range contrib}$$


- Kernel Ridge Regression (and Gaussian Processes)
- Neural Networks
- local environment descriptors

- Modified Behler-Parrinello descriptor



Symmetry Functions

The radial part

$$G^R = \sum_{j \neq i} e^{-\eta(R_{ij}-R_s)^2} f_c(R_{ij})$$

The angular part

$$G^A = 2^{1-\xi} \sum_{jk \neq i} (1 + \cos(\theta_{ijk} - \theta_s))^\xi e^{-\eta(\frac{R_{ij}+R_{ik}}{2}-R_s)^2} f_c(R_{ij}) f_c(R_{ik})$$

$$f_c(R) = \frac{1}{2} \left(1 + \cos\left(\frac{\pi R}{R_c}\right) \right)$$

J. Behler and M. Parrinello, Phys. Rev. Lett. **98**, 146401 (2007)

J.S.Smith, O.Isayev and A.E.Roitberg, Chem. Sci., 2017, **8**, 3192-3203

Representation

$$G_{m,s;i}^R = \sum_{i \neq j}^{\text{All atoms kind s}} e^{-\eta(r_{ij}-R_m)^2} f_c(r_{ij})$$

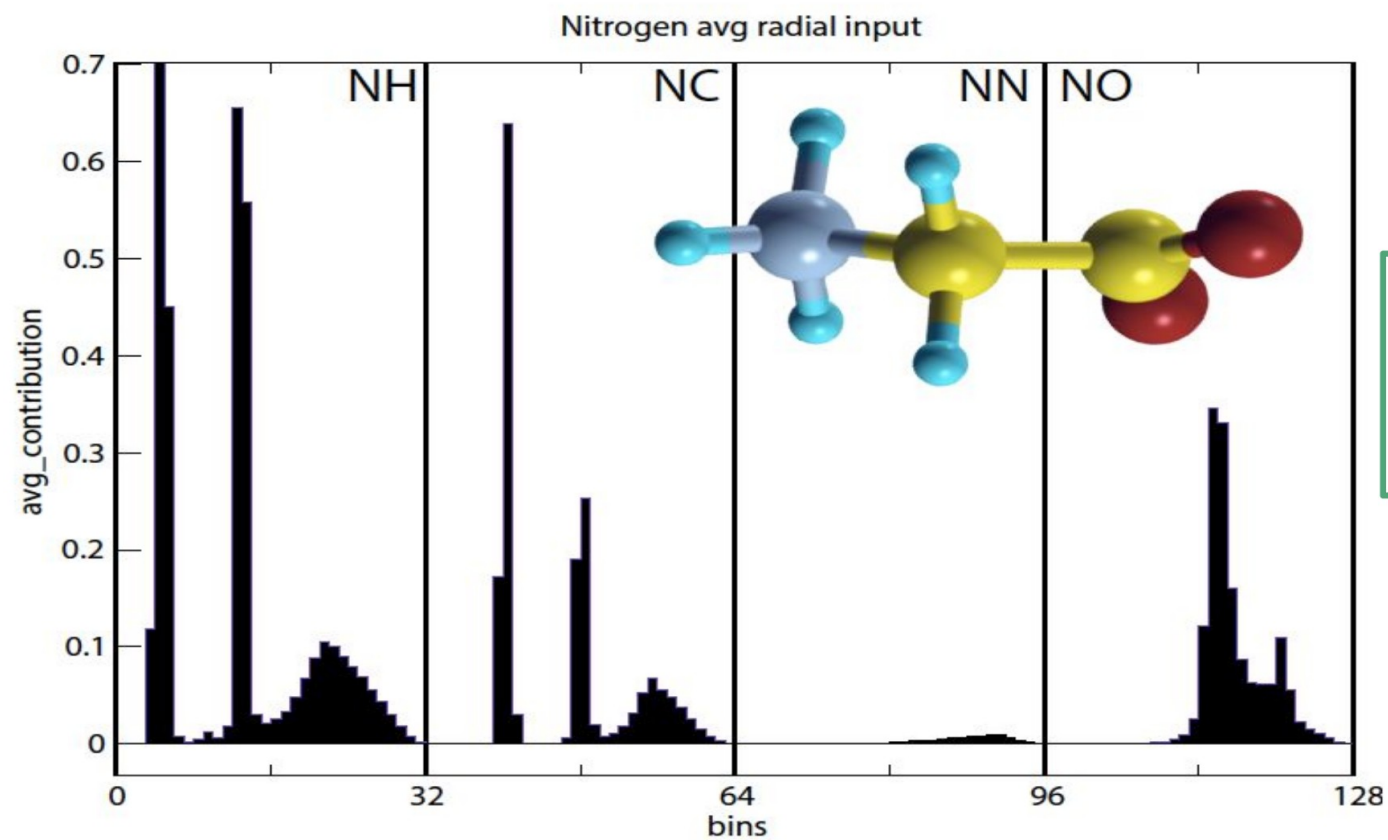
$$f_c(r_{ij}) = \begin{cases} 0.5 \left[\cos \left(\frac{\pi r_{ij}}{R_c} \right) + 1 \right] & \text{if } r_{ij} \leq R_c \\ 0 & \text{if } r_{ij} \geq R_c \end{cases}$$

$R_0=0.5\text{\AA}$, $R_c= 4.6\text{\AA}$
32 bins per pair:
 $32 \times 4 = 128$ parameters

J. Behler and M. Parrinello, PRL, 98.14 (2007).

Smith et al, Chem Sci 8 3192 (2017)
DOI: 10.1039/c6sc05720a

Average G-radial for N in GLY



$R_0=0.5\text{\AA}$, $R_c= 4.6\text{\AA}$
32 bins per pair:
 $32 \times 4 = 128$ parameters

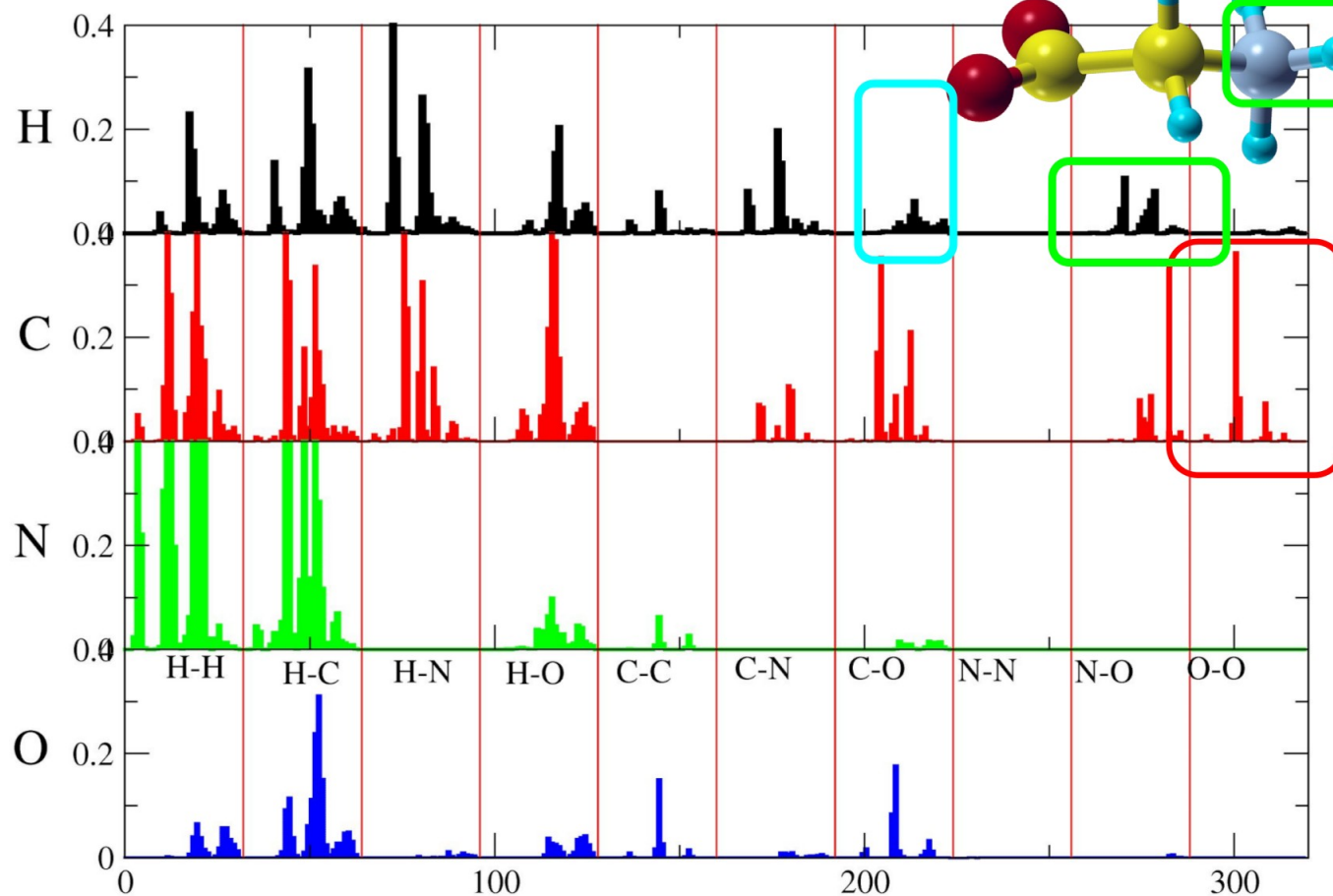
Representation

$$G_{n,m,s;i}^A = 2^{1-\xi} \sum_{\substack{\text{All atom of kind s} \\ j,k \neq i}} (1 + \lambda \cos(\Theta_{ijk} - \Theta_n))^\xi e^{-\eta \left(\frac{r_{ij} + r_{ik}}{2} - R_m \right)^2} f_c(r_{ij}) f_c(r_{ik})$$

Smith et al, Chem Sci (2016)
DOI: 10.1039/c6sc05720a

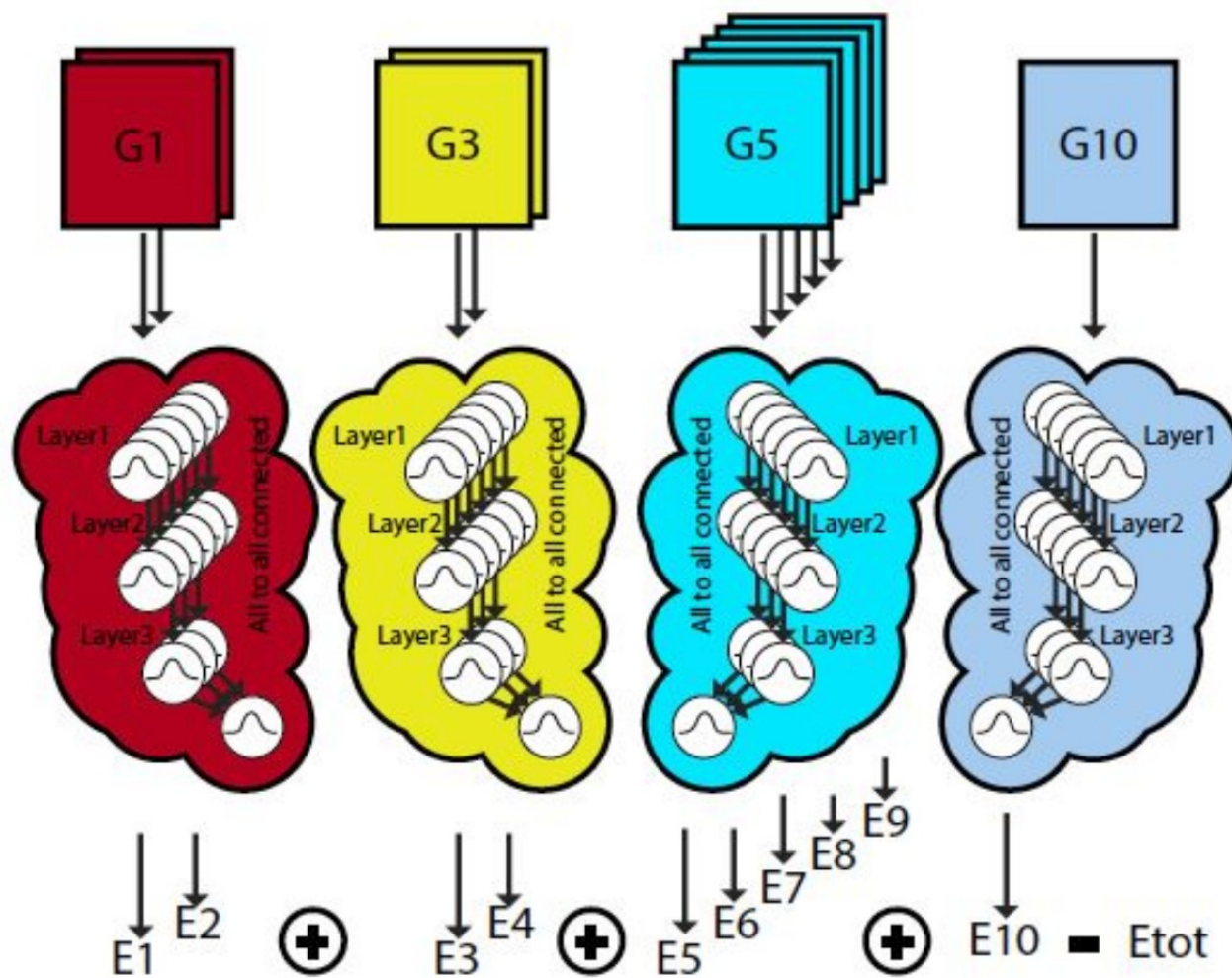
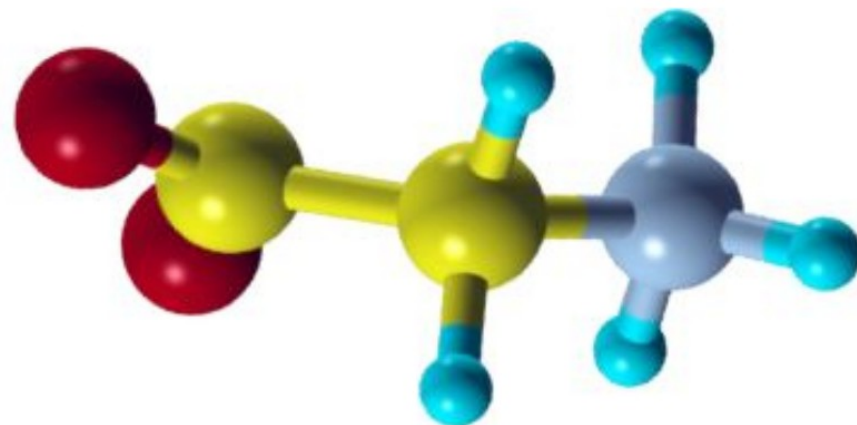
$R_0=0.5\text{\AA}$, $R_c= 3.1\text{\AA}$
8 angular bin for each 8 radial bin
64 bins per trimer:
 $64 \times 10 = 640$ parameters

Average G-angular in GLY

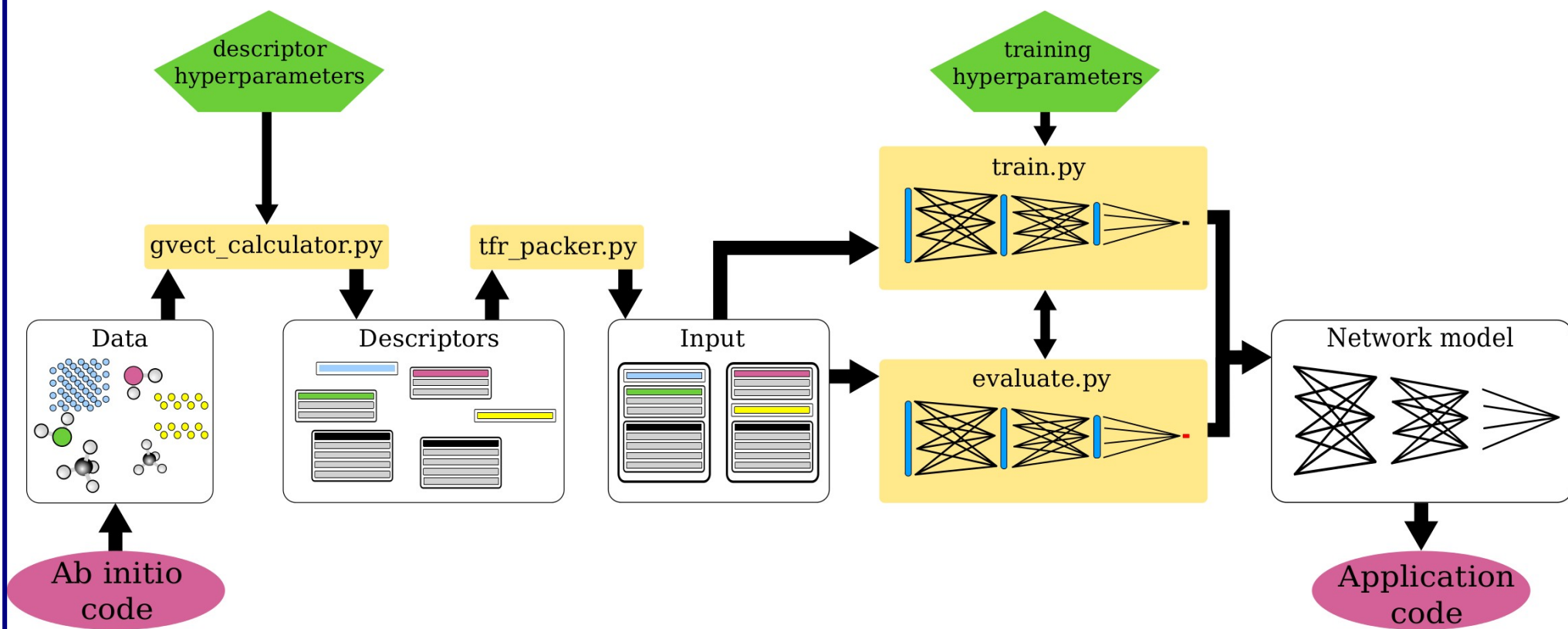


Halved version:
 8 angular for 4 radial
 32 per trimer
 $32 \times 10 = 320$ parameters

GLY



PANNA workflow

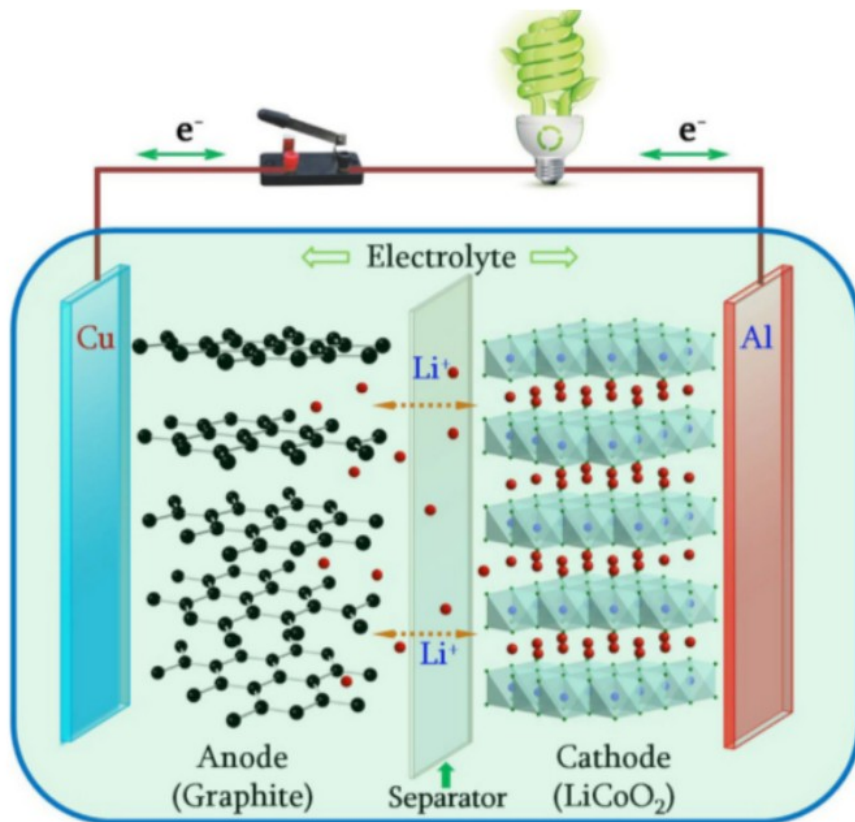


R Lot, F Pellegrini, Y Shaidu, E Kucukbenli, *arXiv:1907.03055*

<https://gitlab.com/pannadevs/panna>

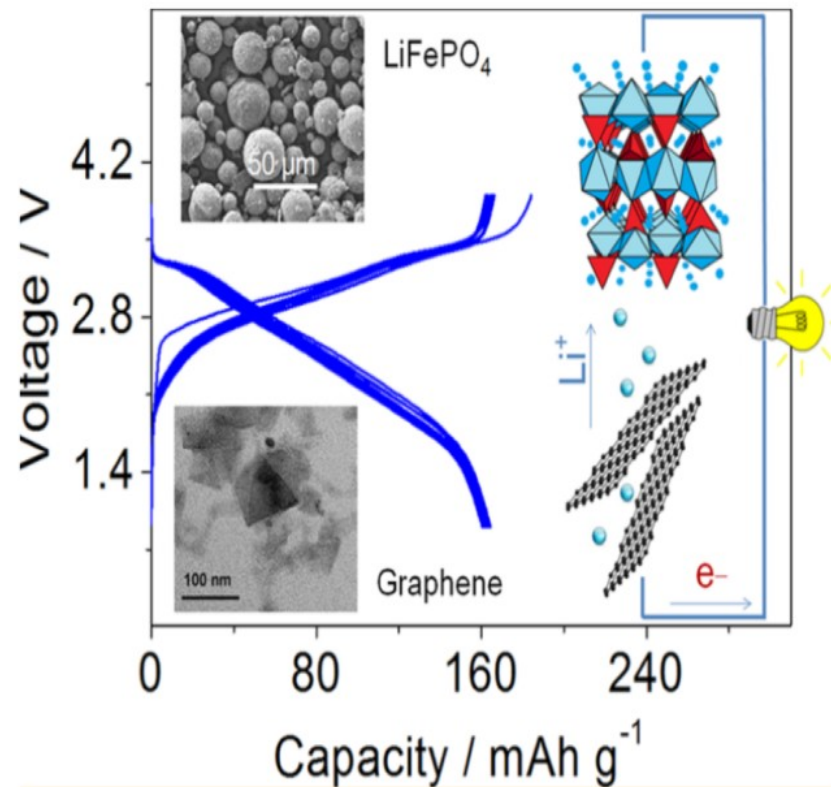


Lithium ion batteries



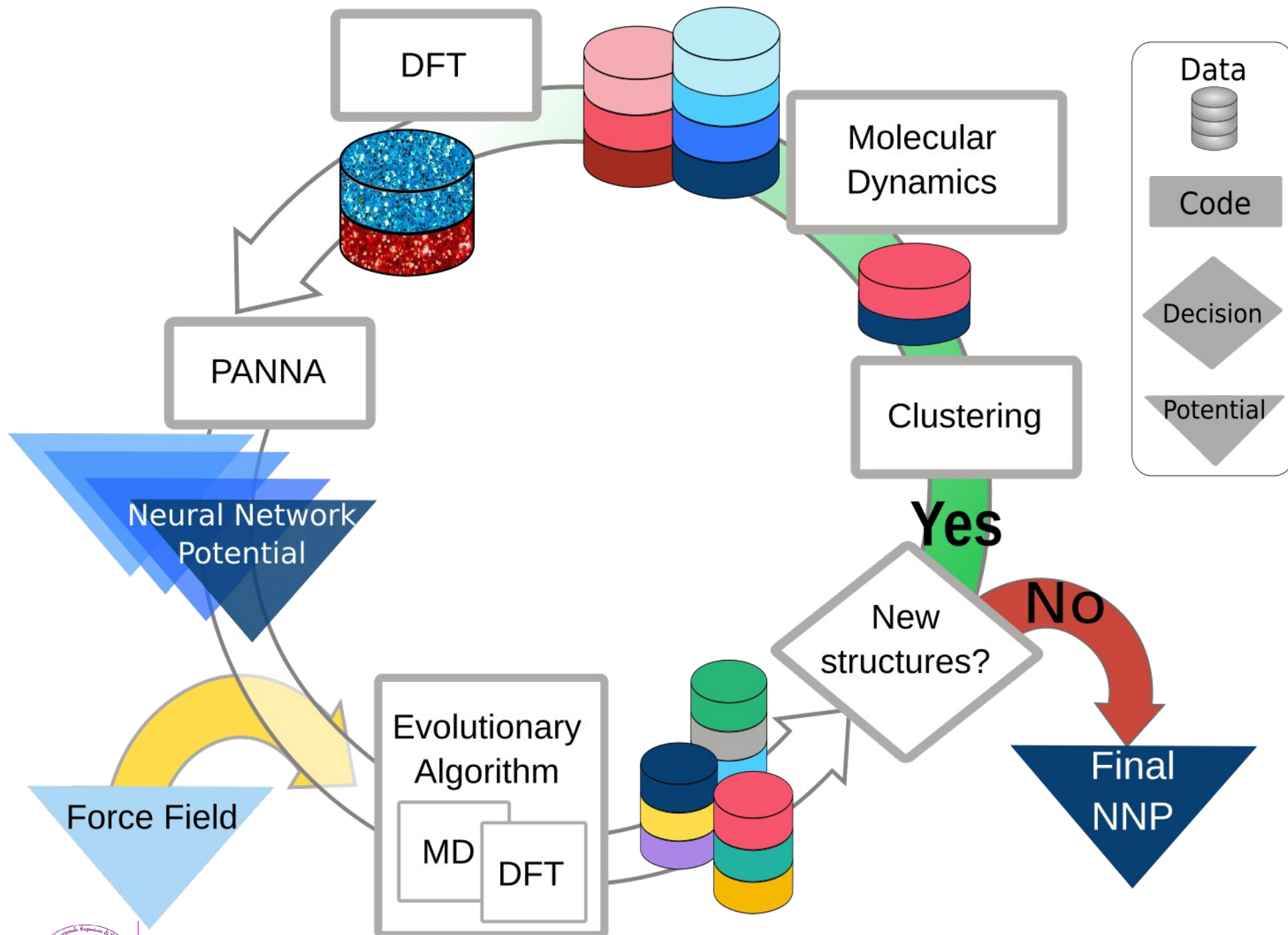
- graphite: LiC₆

Today 19(2):109-123, 2016

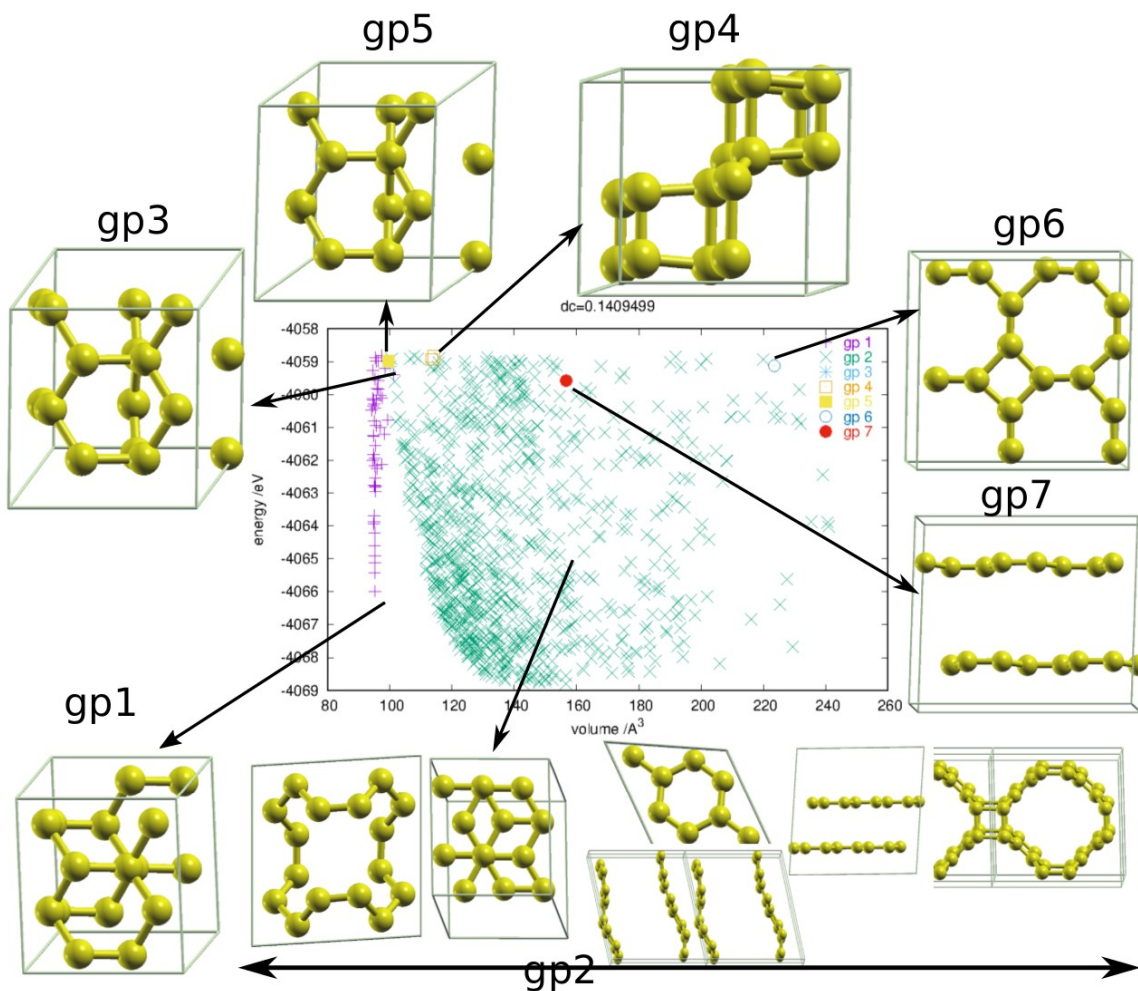


- graphene: LiC₂

Hassoun et al. Nano Lett. 2014, 14, 4901-4906



Carbon Systems



$$D = \frac{1}{2} \left(1 - \frac{\mathbf{F}_1 * \mathbf{F}_2}{|\mathbf{F}_1| |\mathbf{F}_2|} \right)$$

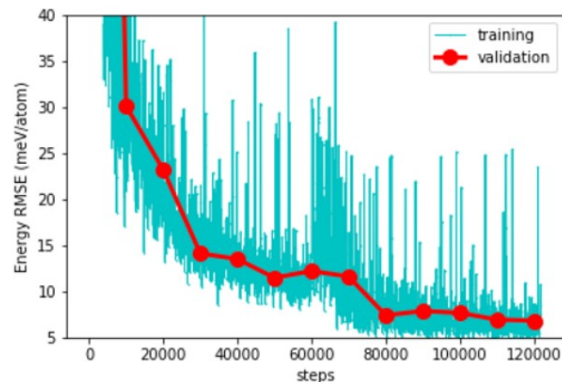
Training Parameters

- Architectures: 144:64:32:1
- Activation function: gaussian:gaussian:linear
- minimized quantity:

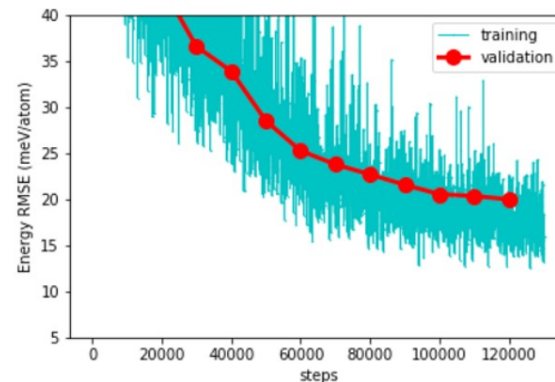
$$Loss = E_{Loss} + \beta F_{Loss}$$

Carbon Systems: training and validation

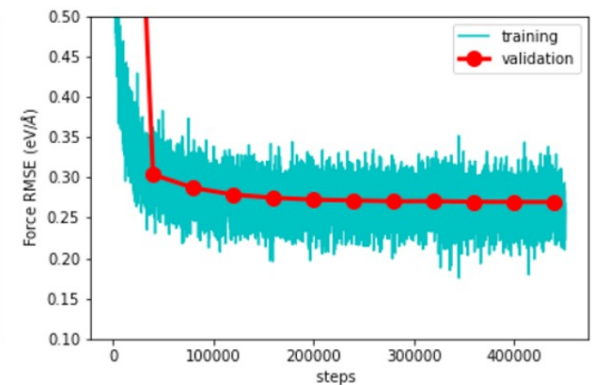
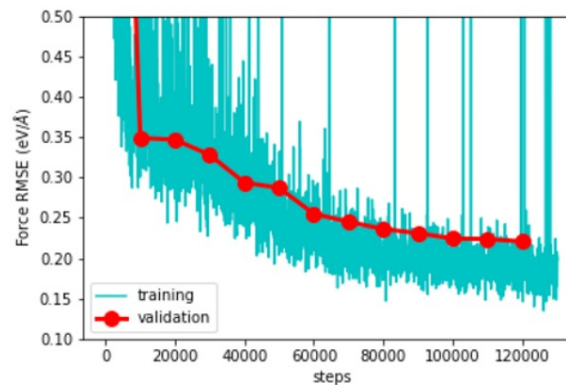
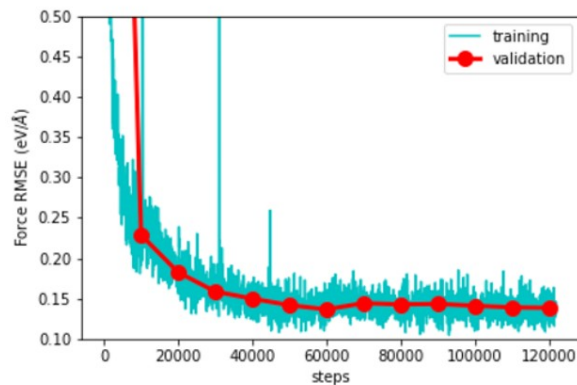
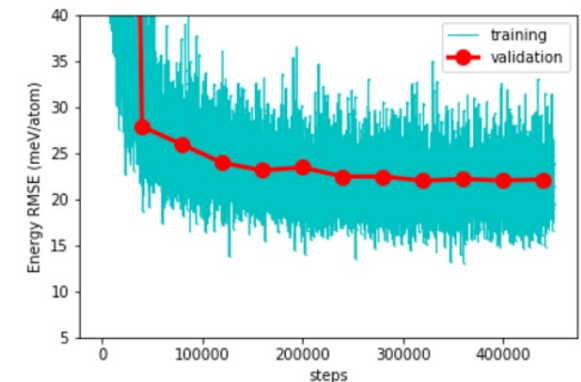
1st iteration



2nd iteration



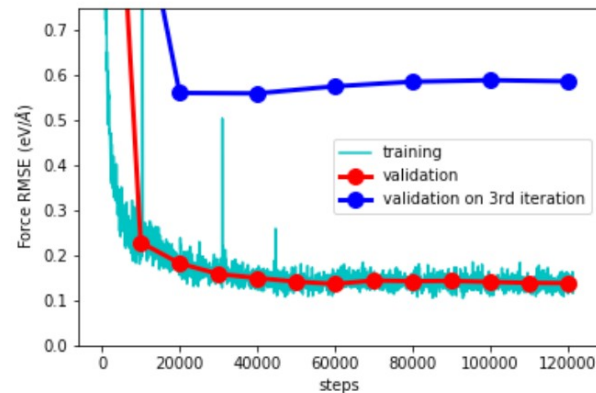
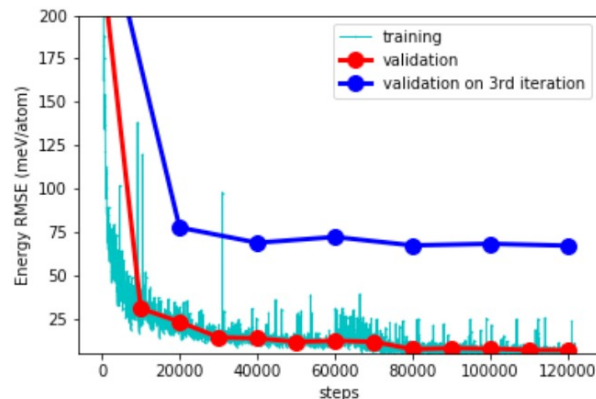
3rd iteration



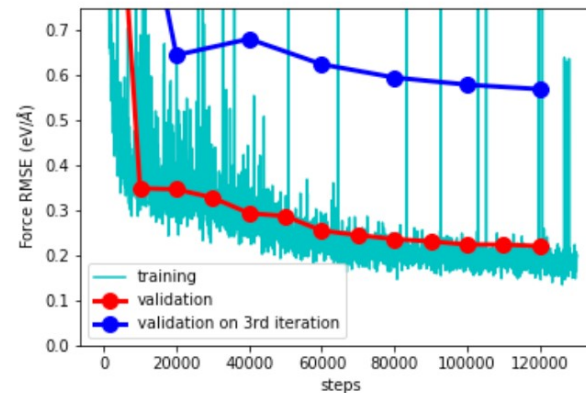
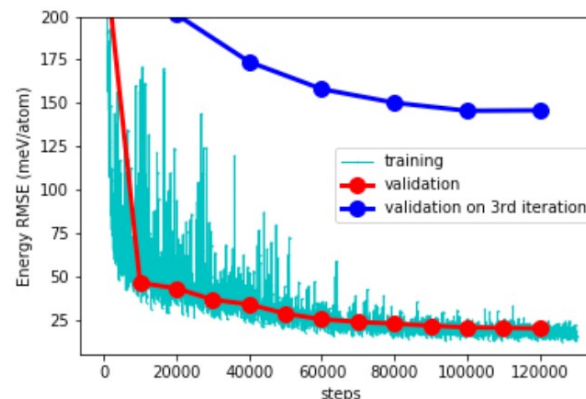
- 20 % of the data set is set aside for validation

Carbon Systems: training and validation

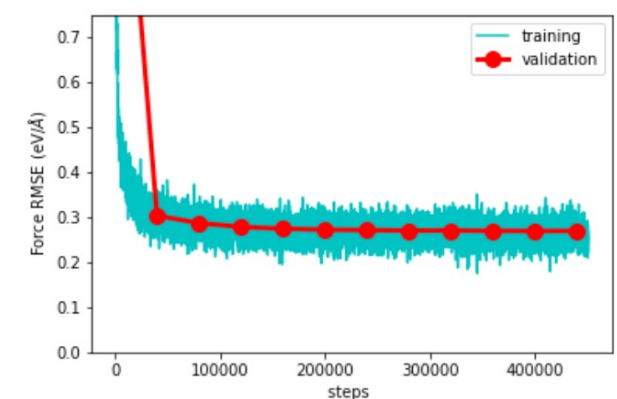
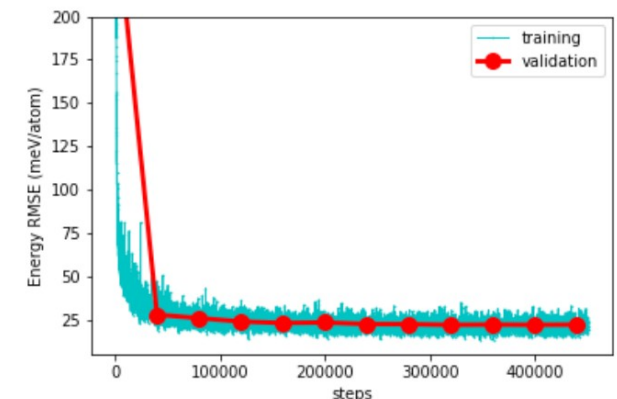
1st iteration



2nd iteration



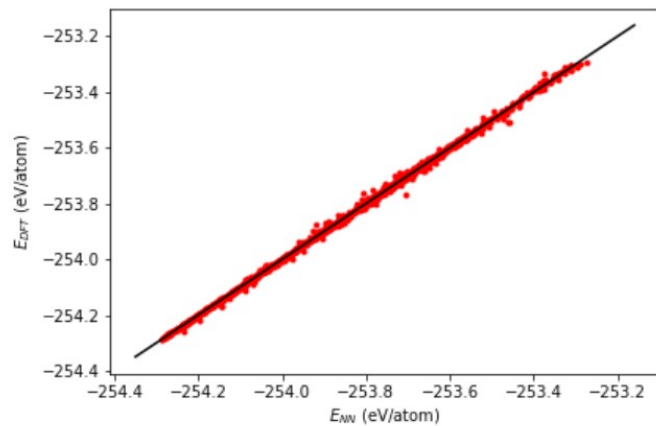
3rd iteration



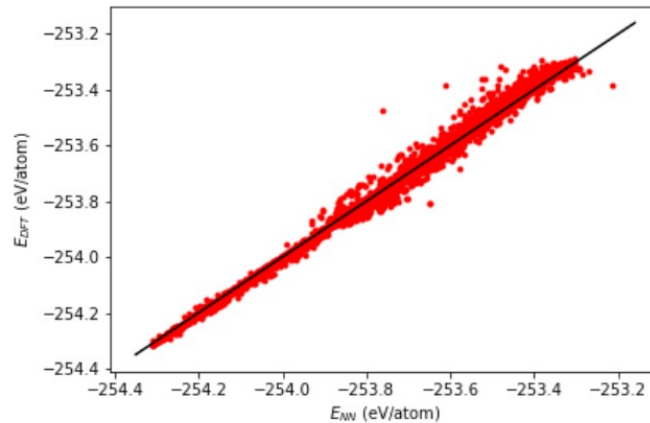
- 20 % of the data set is set aside for validation

Error distribution: energies

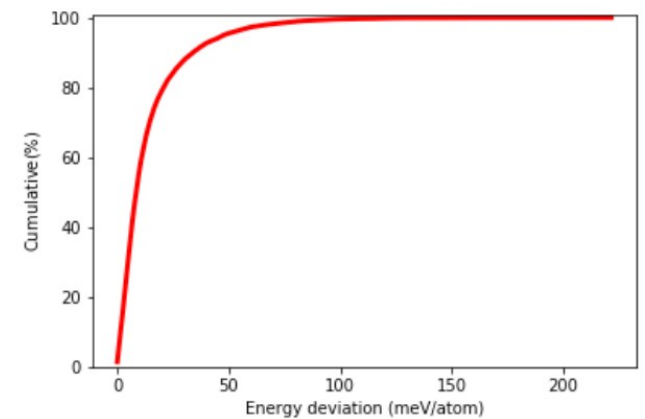
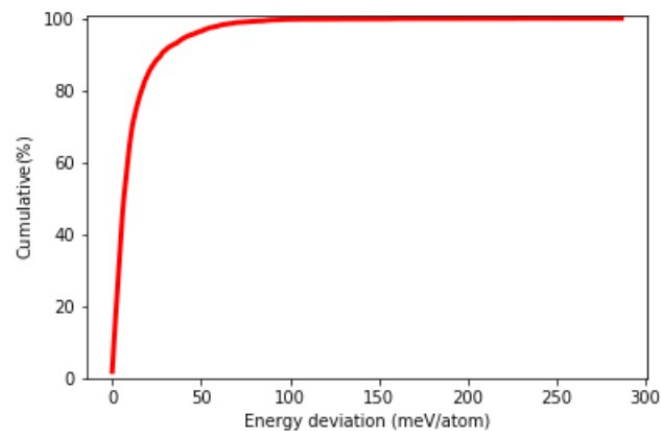
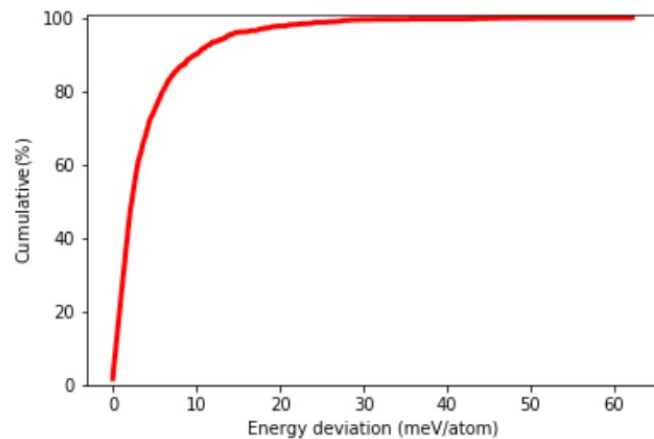
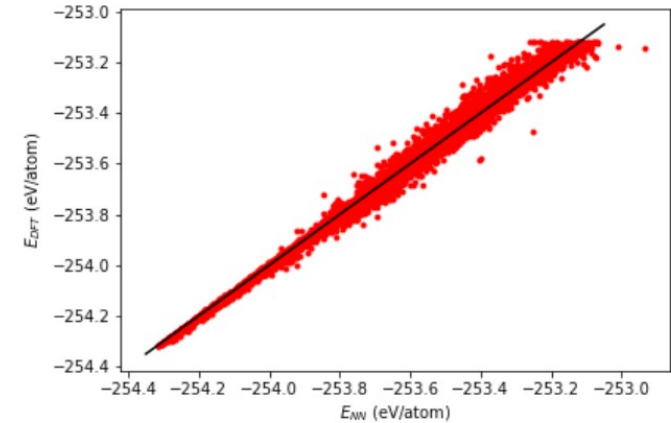
1st iteration



2nd iteration

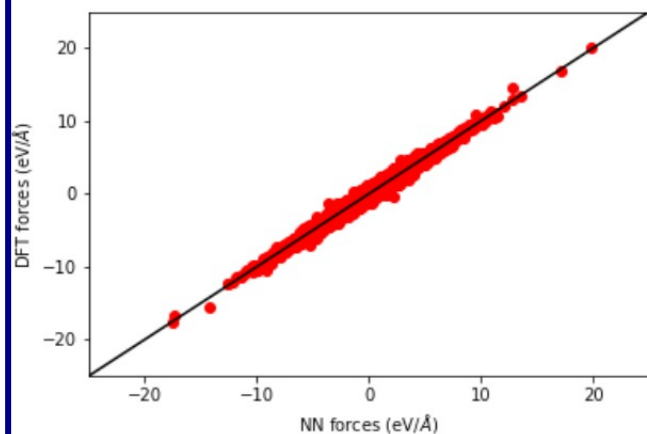


3rd iteration

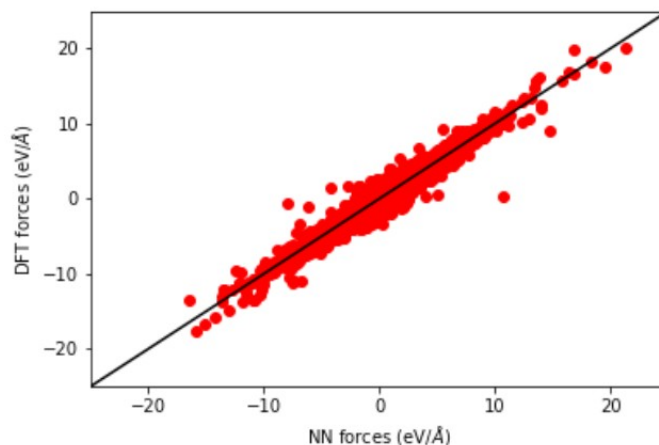


Error distribution: forces

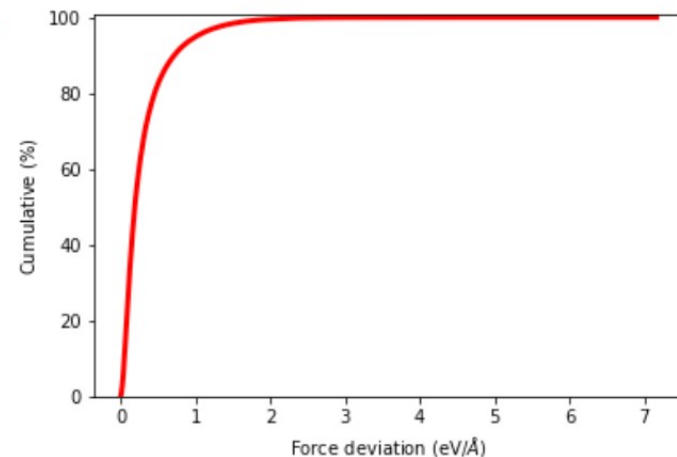
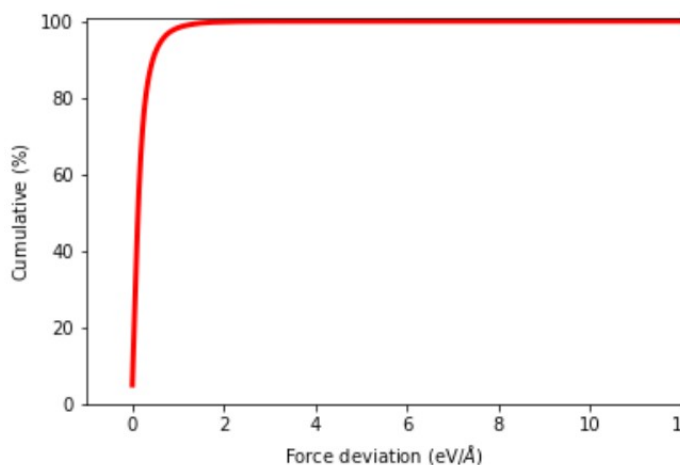
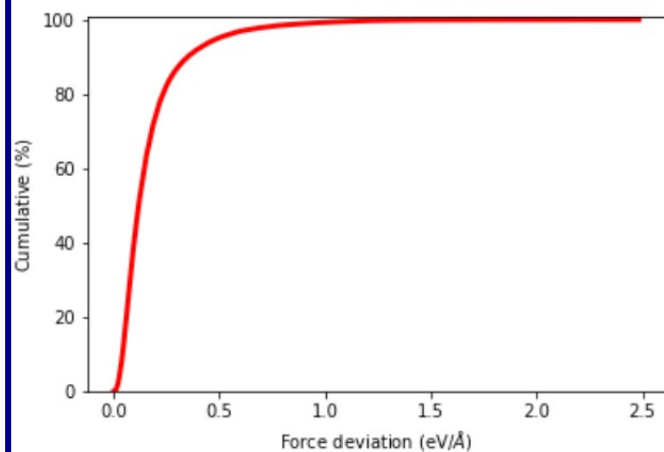
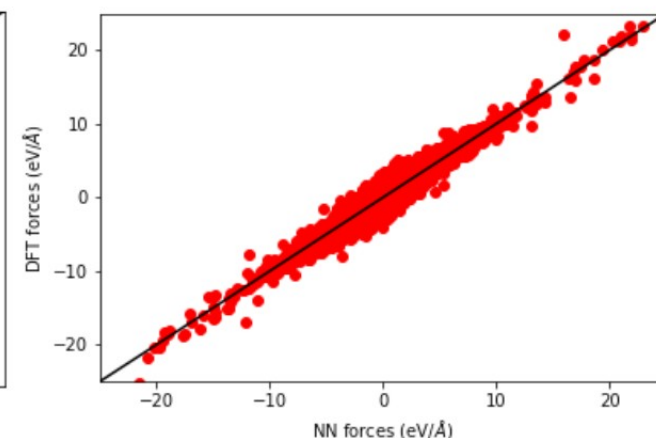
1st iteration



2nd iteration



3rd iteration



Effect of diversity on training and validation RMSE

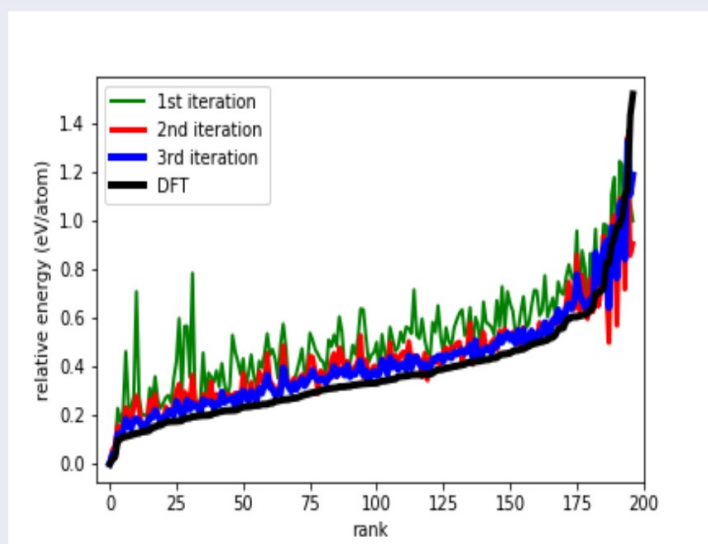
Train \ Validate	T error	All D	$D < 0.15$	$D < 0.10$	$D < 0.05$	$D_{12} < 0.05$
All D	22.070	22.131	20.938	15.161	7.659	7.302
$D < 0.15$	18.066	80.424	18.422	13.342	5.949	17.567
$D < 0.10$	8.563	162.327	52.178	9.369	4.391	76.260
$D < 0.05$	2.633	879.207	452.598	89.022	2.585	650.075
$D_{12} < 0.05$	2.574	174.257	88.311	51.972	2.739	2.627

Train \ Validate	All D	$D < 0.15$	$D < 0.10$	$D < 0.05$	$D_{12} < 0.05$
All D	0.2696	0.2717	0.1974	0.0829	0.0785
$D < 0.15$	0.5969	0.2523	0.1789	0.0766	0.1873
$D < 0.15$	0.9571	0.4410	0.1472	0.0617	0.3112
$D < 0.15$	3.2641	2.0028	0.7243	0.0529	0.8699
$D_{12} < 0.05$	0.9641	0.8934	0.5440	0.0529	0.0504

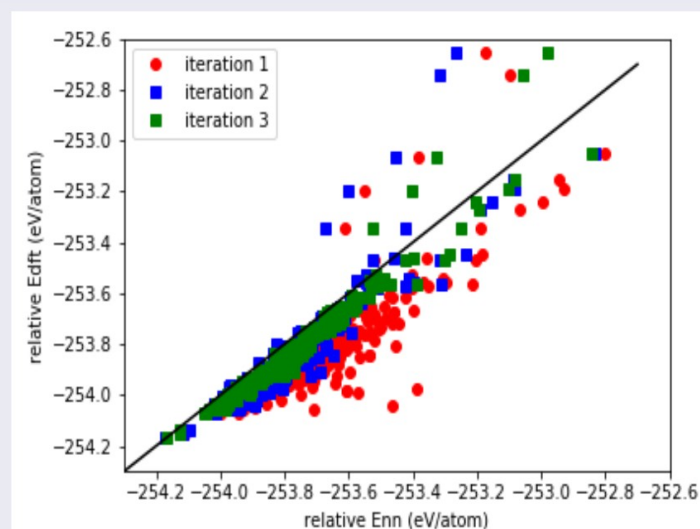
Energy ordering of test structures

- 197 different sp^3 C structures³

Ordering



energy correlation

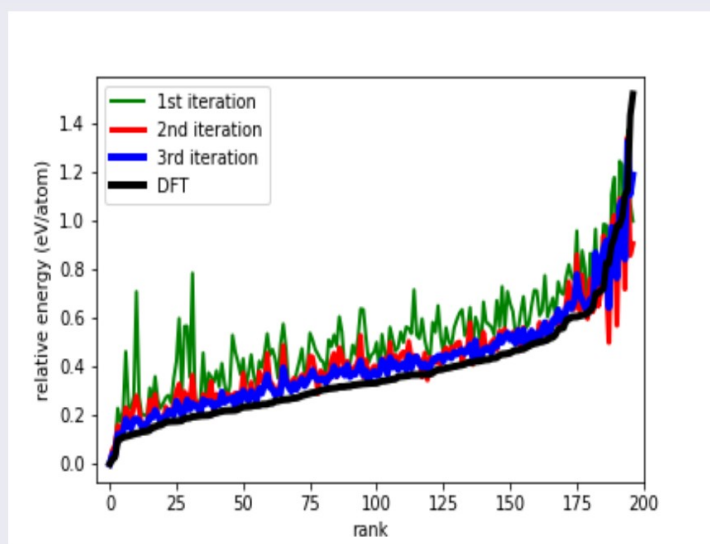


³V.L. Deringer, G. Csanyi and D.M. Proserpio, Chem. Phys. Chem. 2017, **18**, 873–877

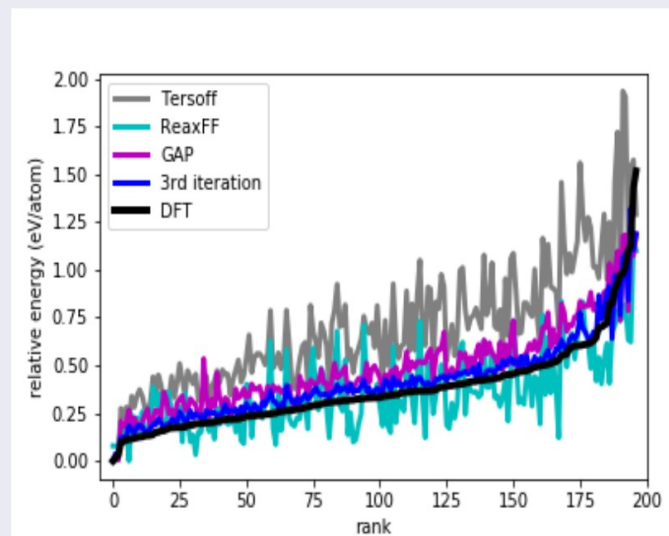
Energy ordering of test structures

- 197 different sp^3 C structures⁴

Ordering



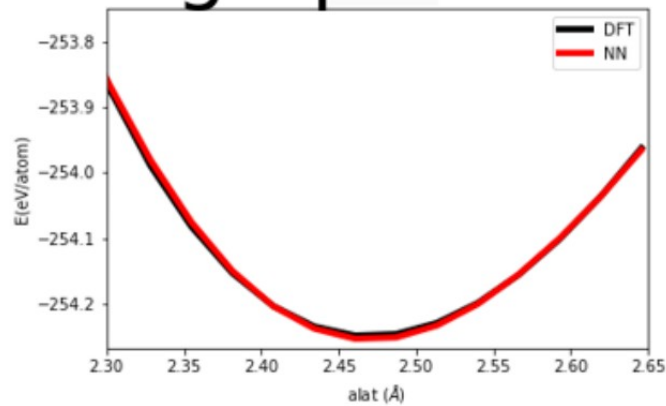
NN vs force fields



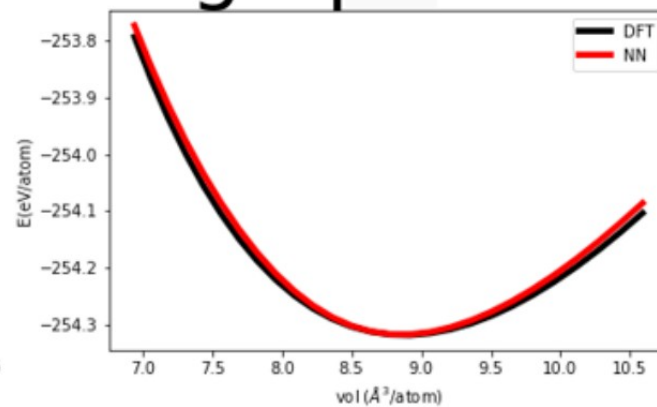
⁴V.L. Deringer, G. Csanyi and D.M. Proserpio, Chem. Phys. Chem. 2017, **18**, 873–877

Equation of State

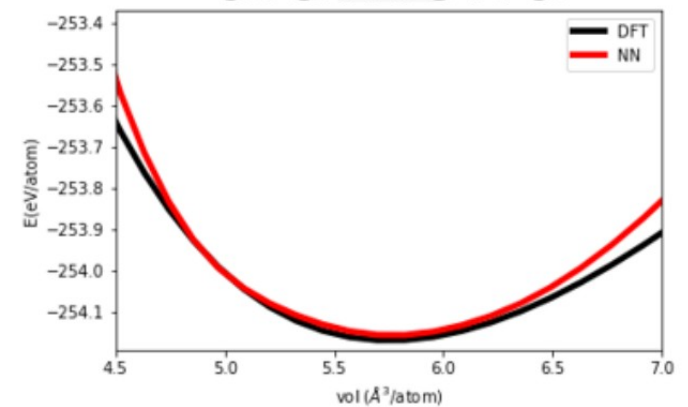
graphene



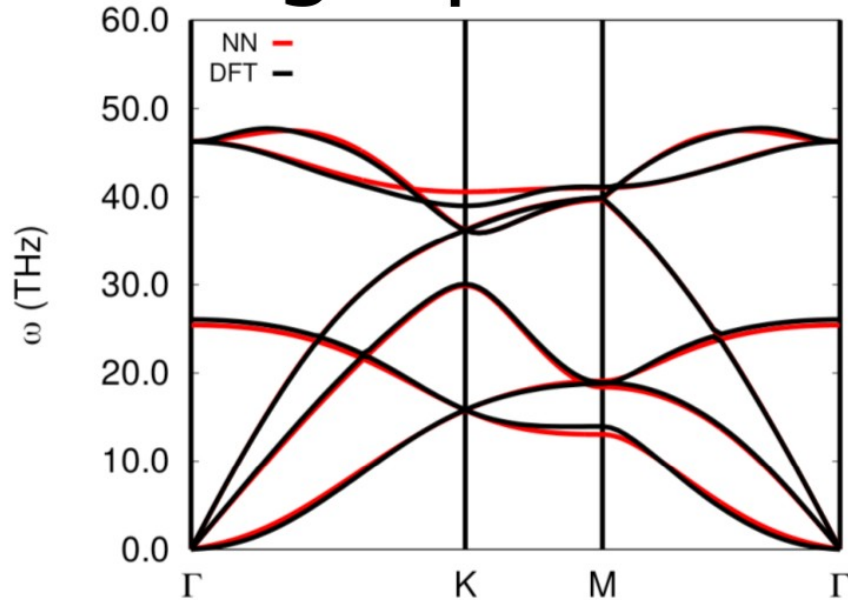
graphite



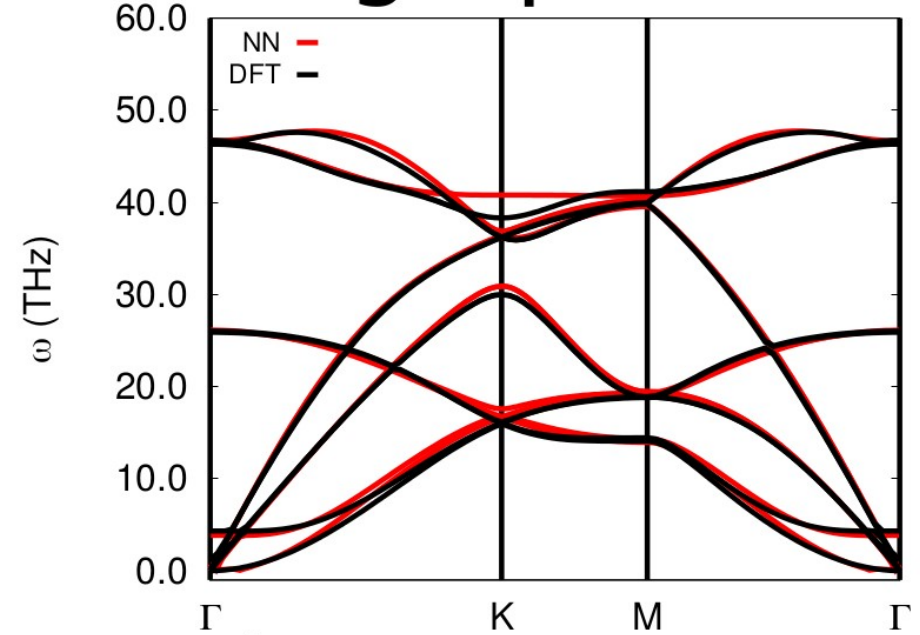
diamond



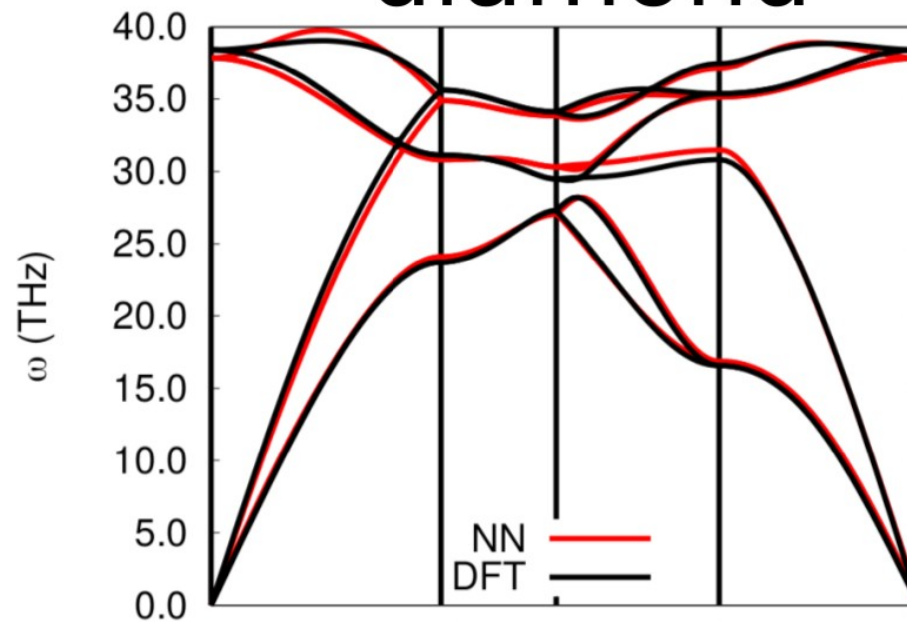
graphene



graphite

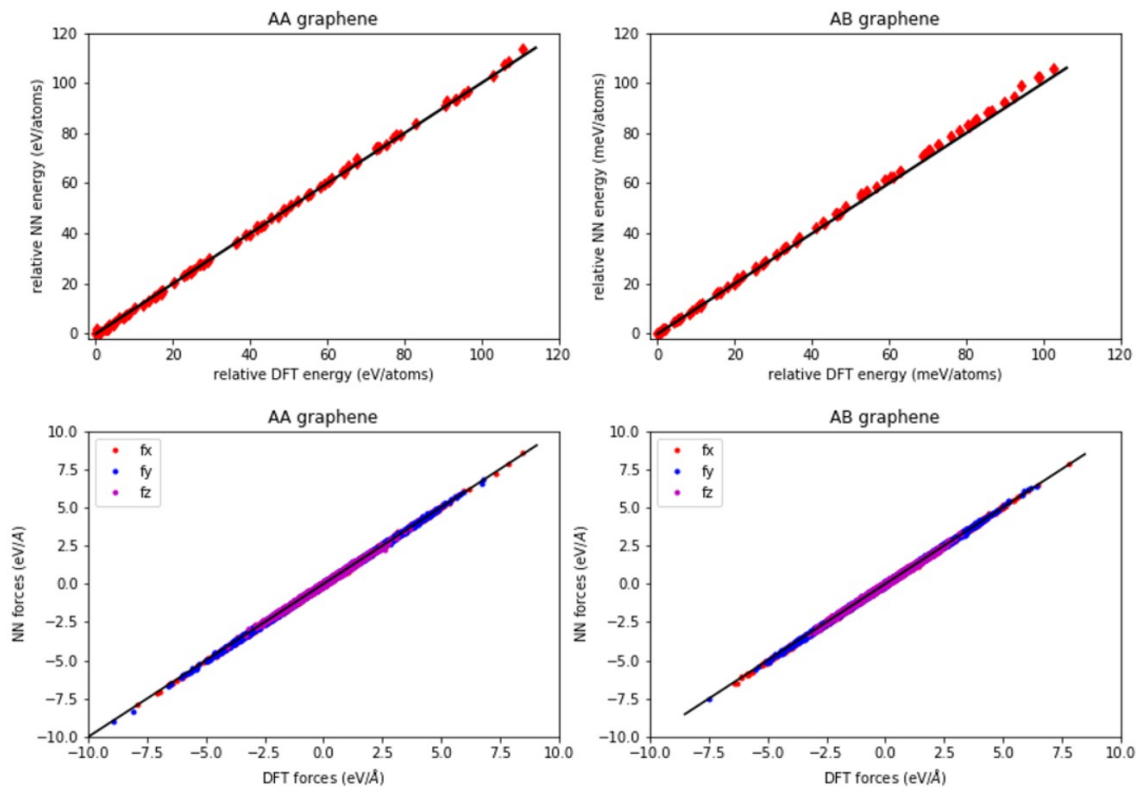


diamond



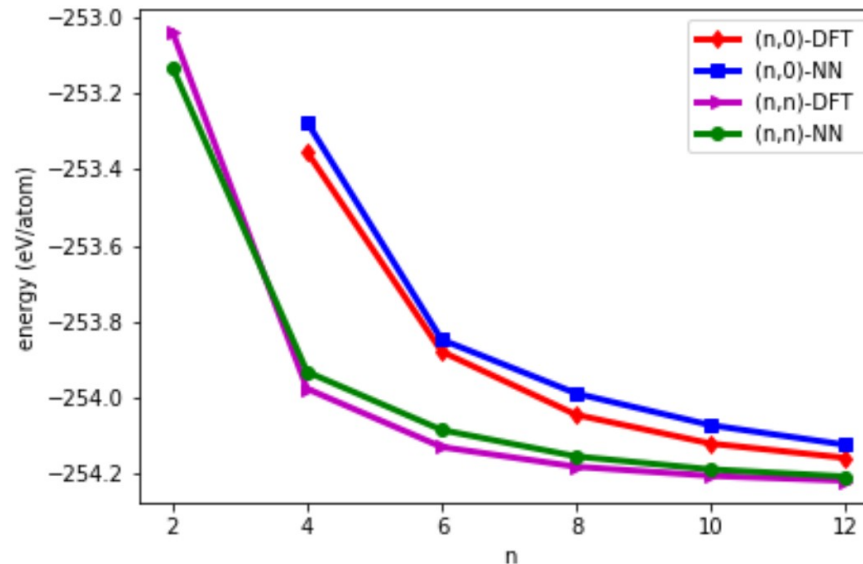
Bilayer Graphene

- configurations generated potential via NVT MD in which the system was heated up from 300 K to 1000K using Nose-Hoover thermostat chain over a period of 1ns.
- Excellent agreement with DFT



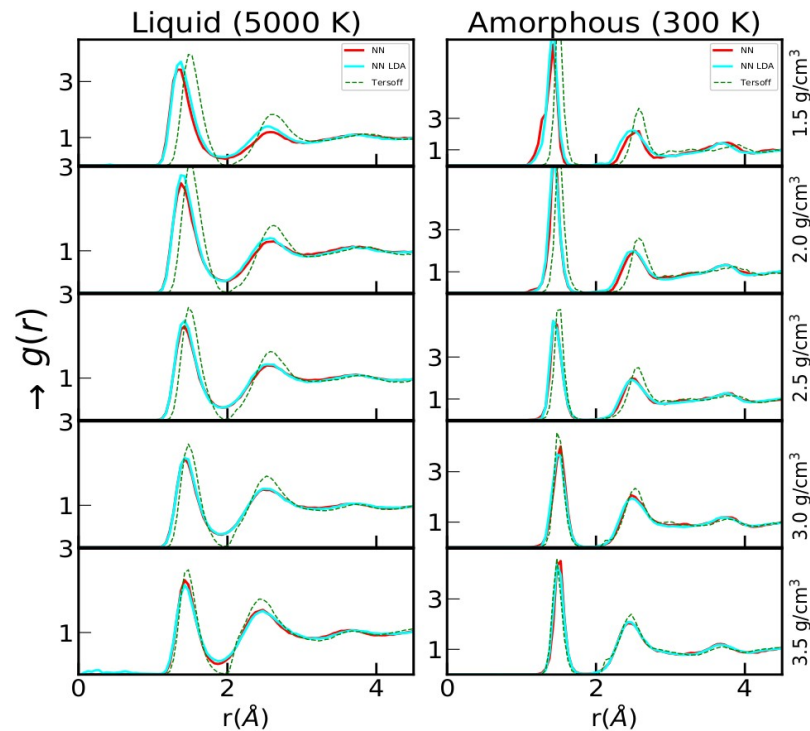
Carbon Nanotubes

- Zigzag nanotube designated by $(n,0)$ and Armchair nanotube designated by (n,n) . n specifies the diameter of the tube as
$$d(n, m) = \frac{a}{\pi} \sqrt{n^2 + nm + m^2}$$
, a is the lattice parameter
- Trends excellently captured
- Good agreement with DFT

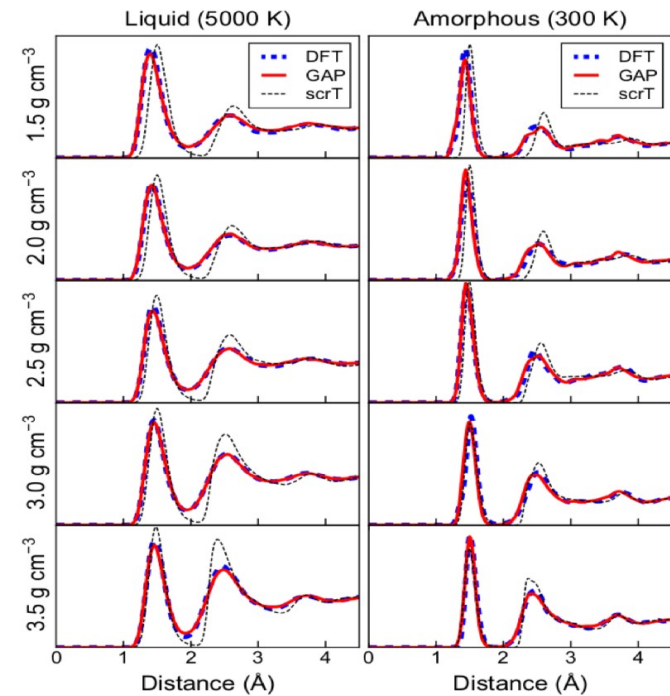


Amorphous Carbon: radial distribution function

a) Our result

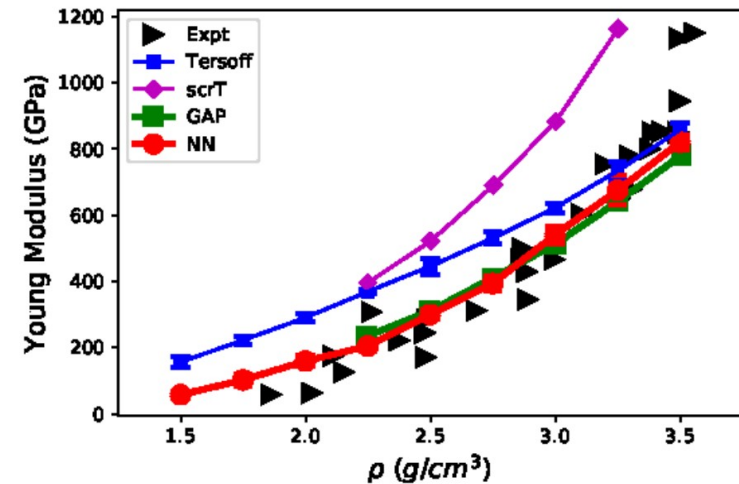
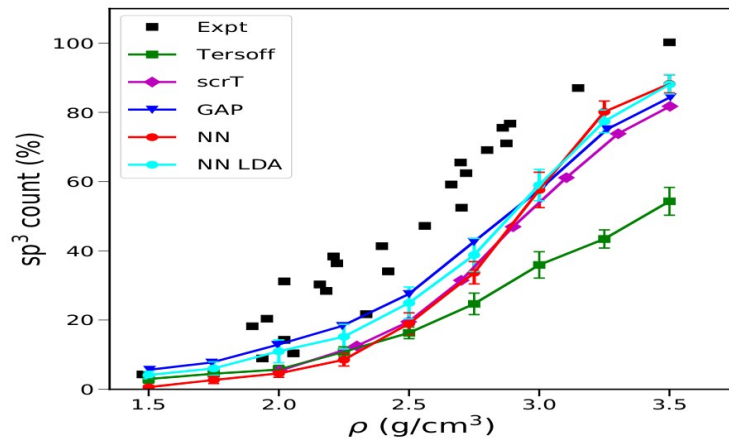


b) GAP



V.L. Deringer and G.Csányi, PRB 95, 094203, (2017)

Amorphous Carbon: sp^3 fraction and Young Modulus

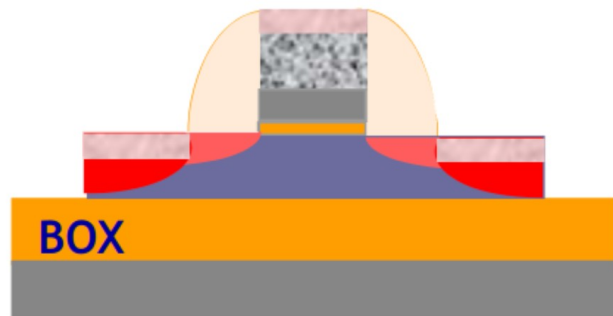


- Fallon et al. PRB 48, 4777 (1993).
- J. Schwan et al. Journal of Applied Physics 79, 1416 (1996)
- V.L. Deringer and G.Csányi, PRB 95, 094203, (2017)

- B.Schultrich et al. Diamond and Related Materials 5 (1996) 914-918
- B.Schultrich et al. Surface and Coatings Technology 98 (1998) 1097-1101

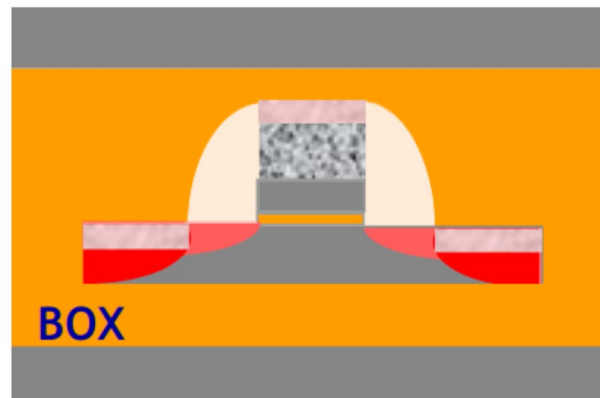
Solid Phase Epitaxy for Silicon

Bottom transistor



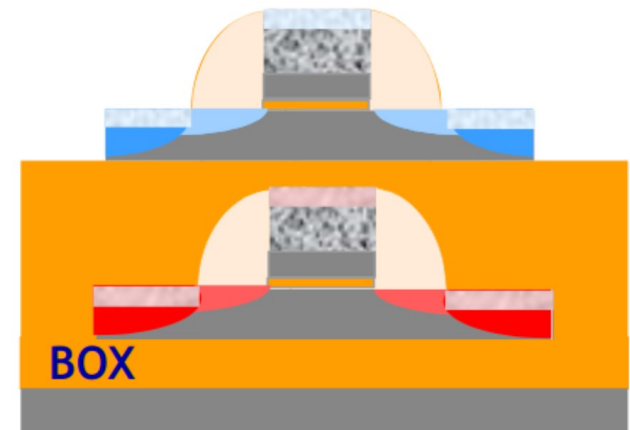
⚡ Salicide stability with top FET TB

Top film realization



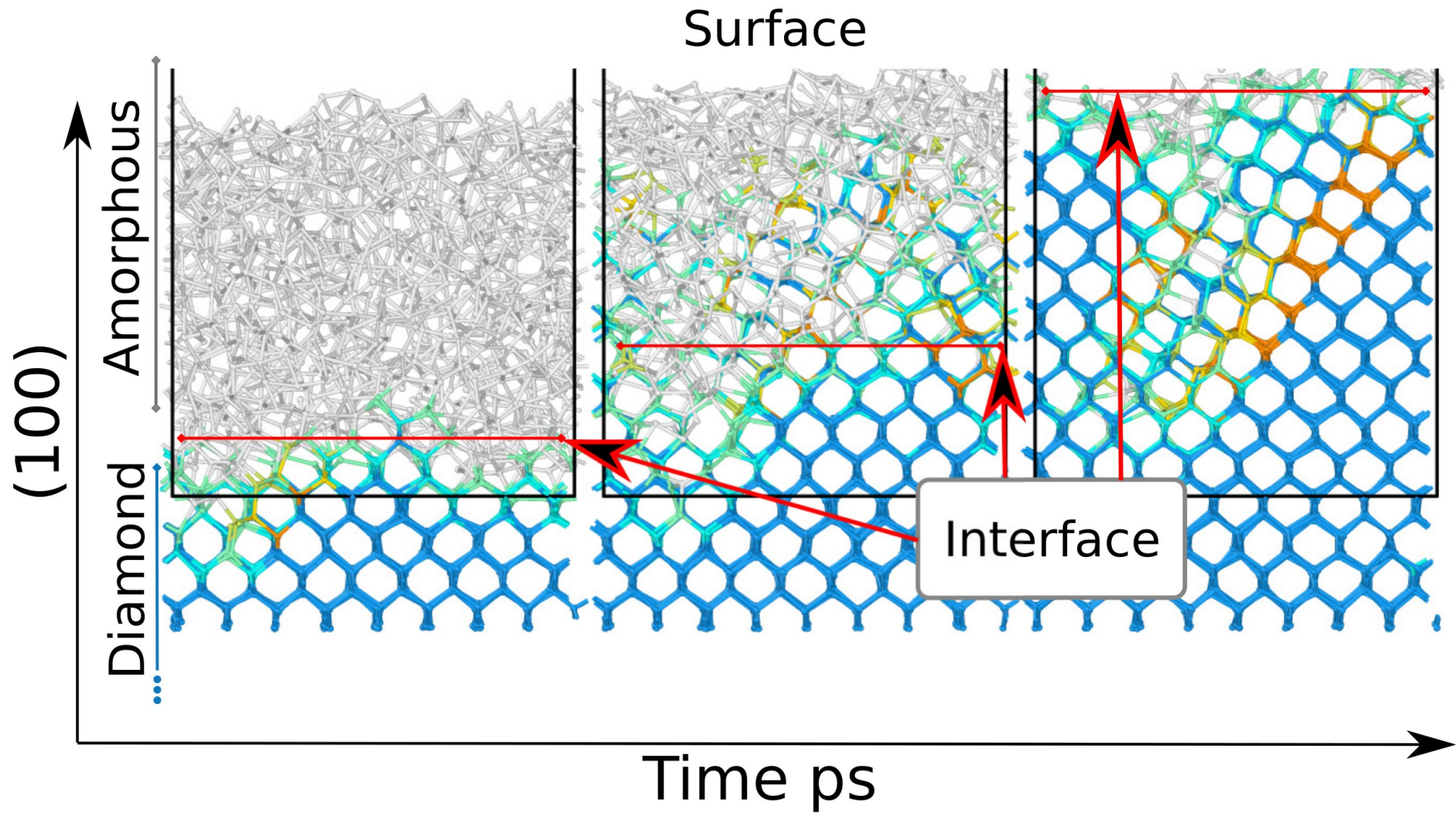
⚡ Cristalline quality/ thickness control

Top transistor



⚡ Reach high perf. with 650°C TB

Problem statement

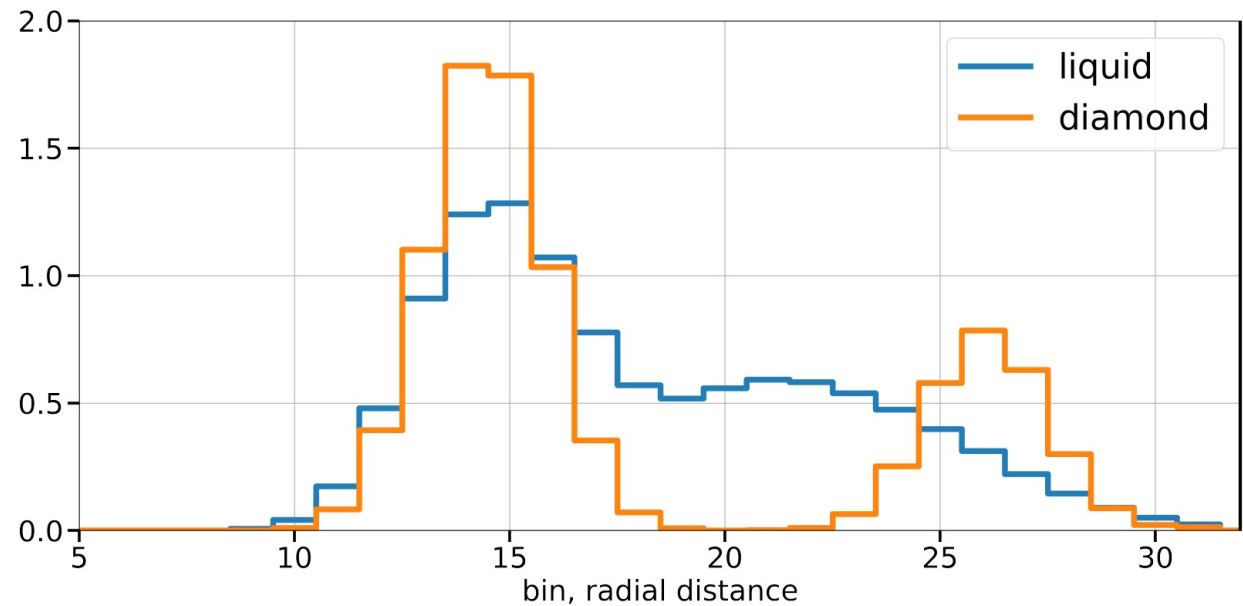
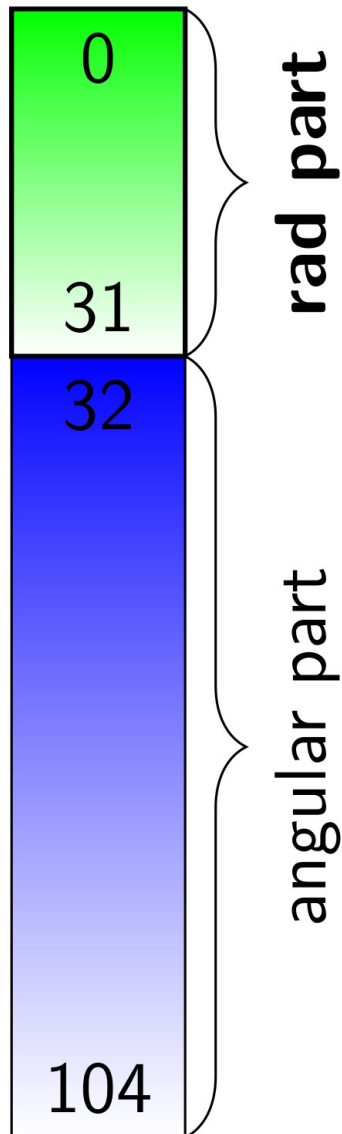


Olson and Roth 1988

The input: $\mathbf{G}_i(\{\mathbf{x}_j\} | d(\mathbf{x}_i, \mathbf{x}_j) < r_{cut})$

The radial part

$$G_{m,s;i}^R = \sum_{\substack{\text{All atoms kind s} \\ i \neq j}} e^{-\eta(r_{ij} - R_m)^2} f_c(r_{ij})$$

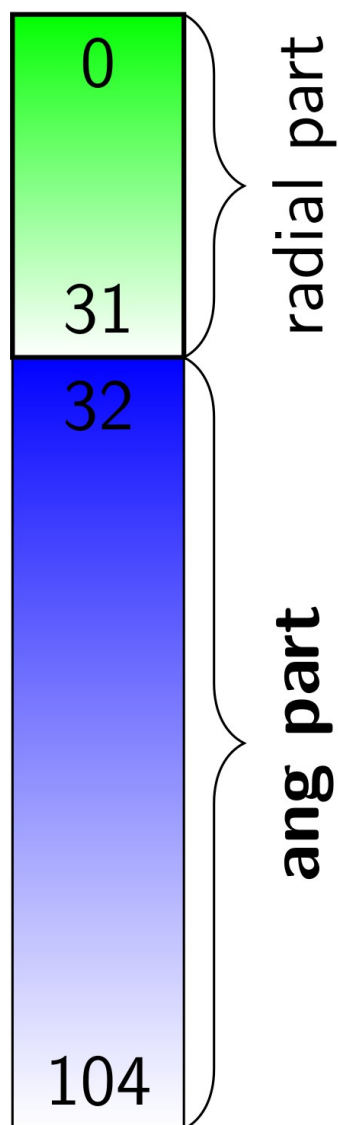


with thermal vibrations

Behler and Parrinello 2007



The input: Descriptor

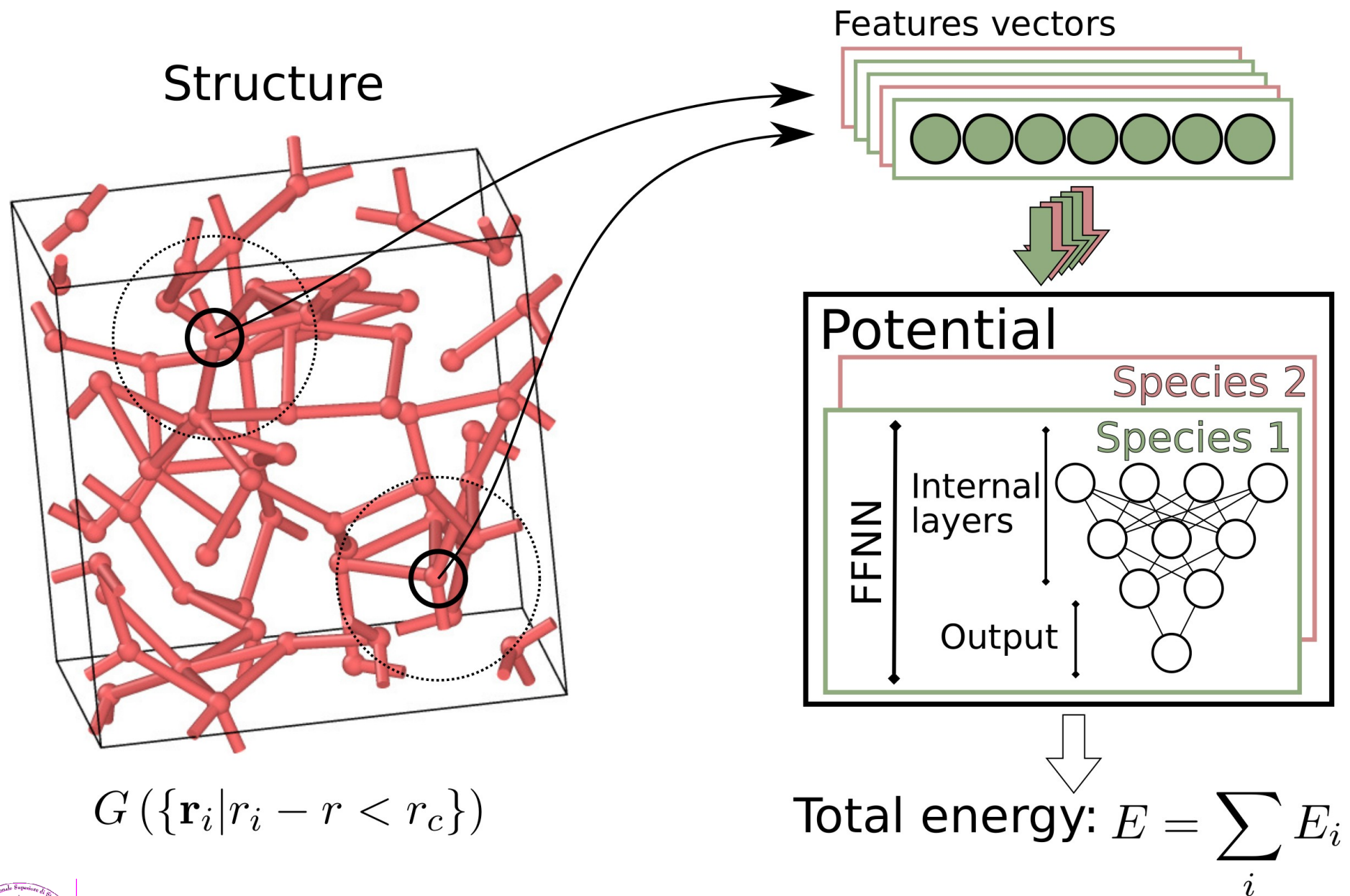


The angular part

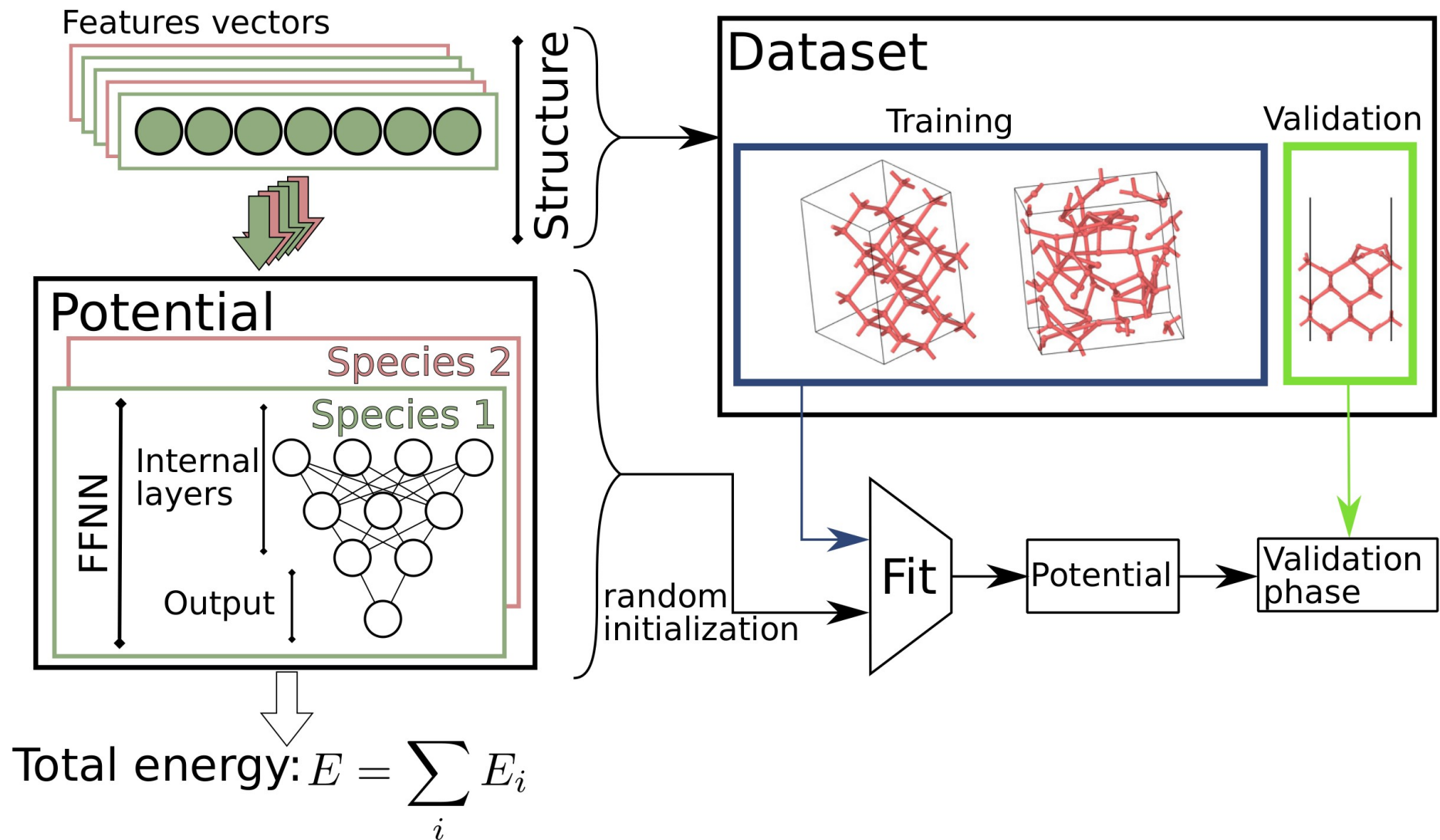
$$G_{n,m,s;i}^A = 2^{1-\xi} \sum_{j,k \neq i}^{\text{All atom of kind s}} (1 + \cos(\Theta_{ijk} - \Theta_n))^\xi e^{-\eta \left(\frac{r_{ij} + r_{ik}}{2} - R_m \right)^2} f_c(r_{ij}) f_c(r_{ik})$$

Lot et al. 2020; Smith, Isayev, and Roitberg 2017

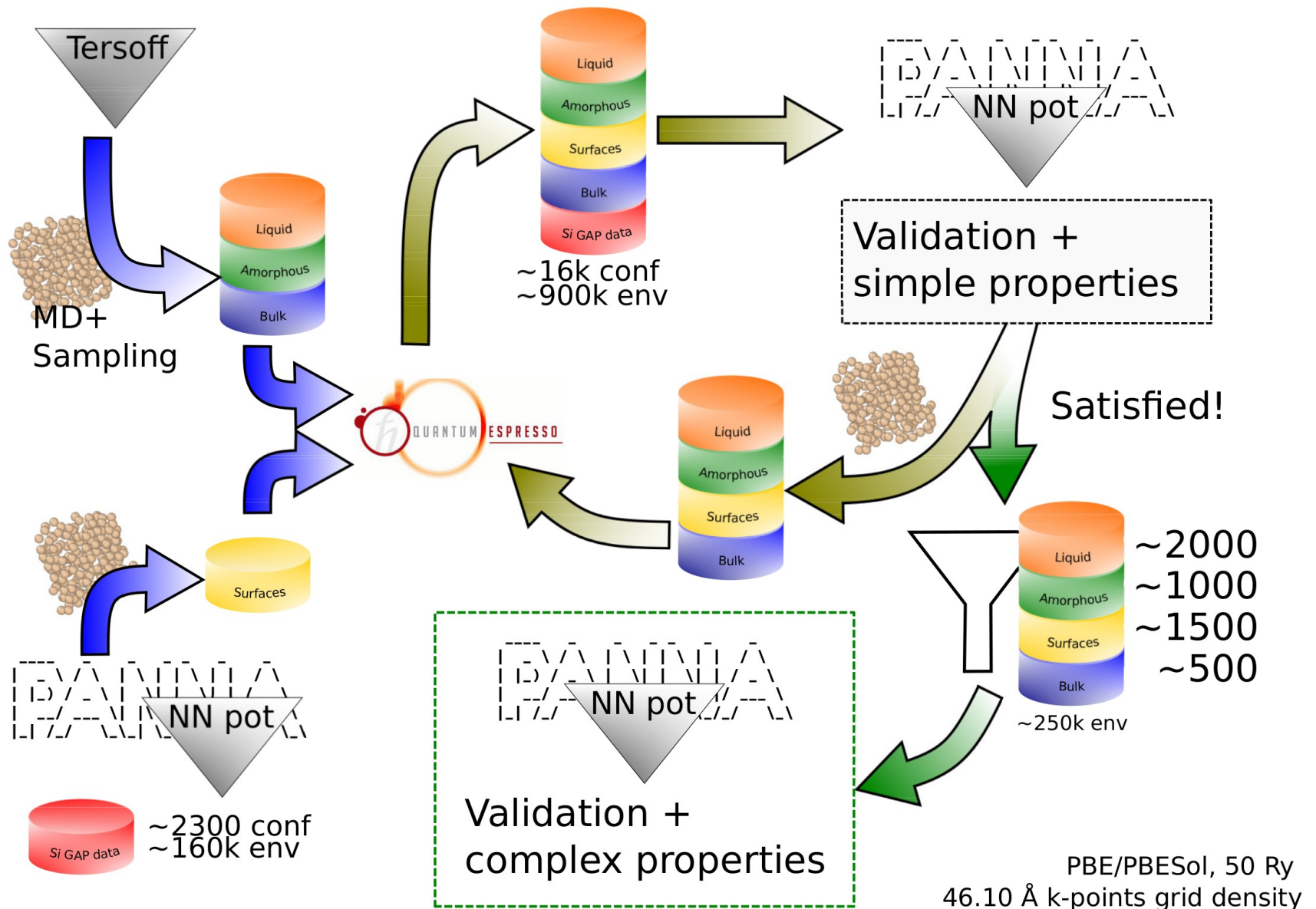
From a FFNN to a potential



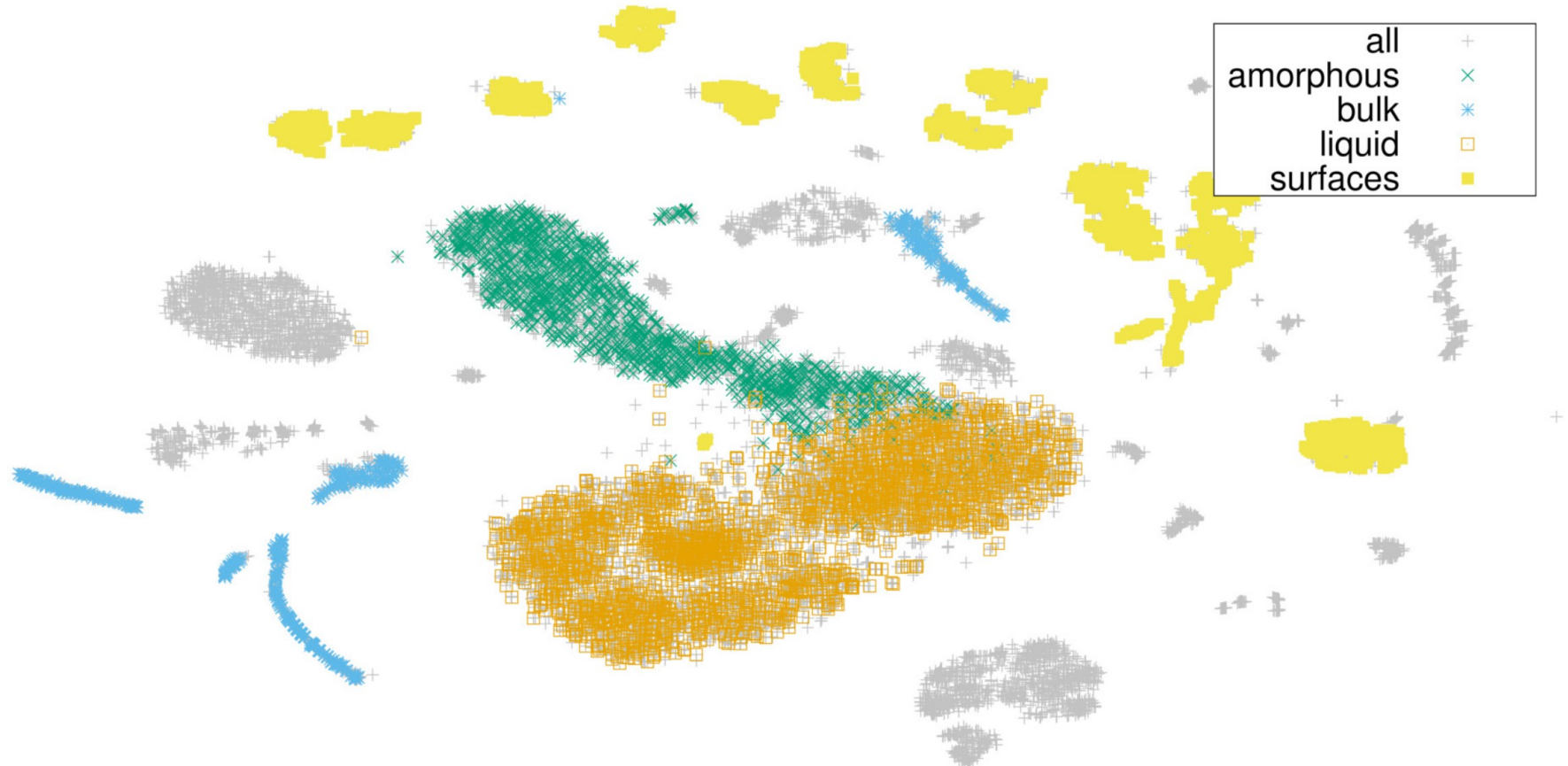
The fitting phase



Work workflow

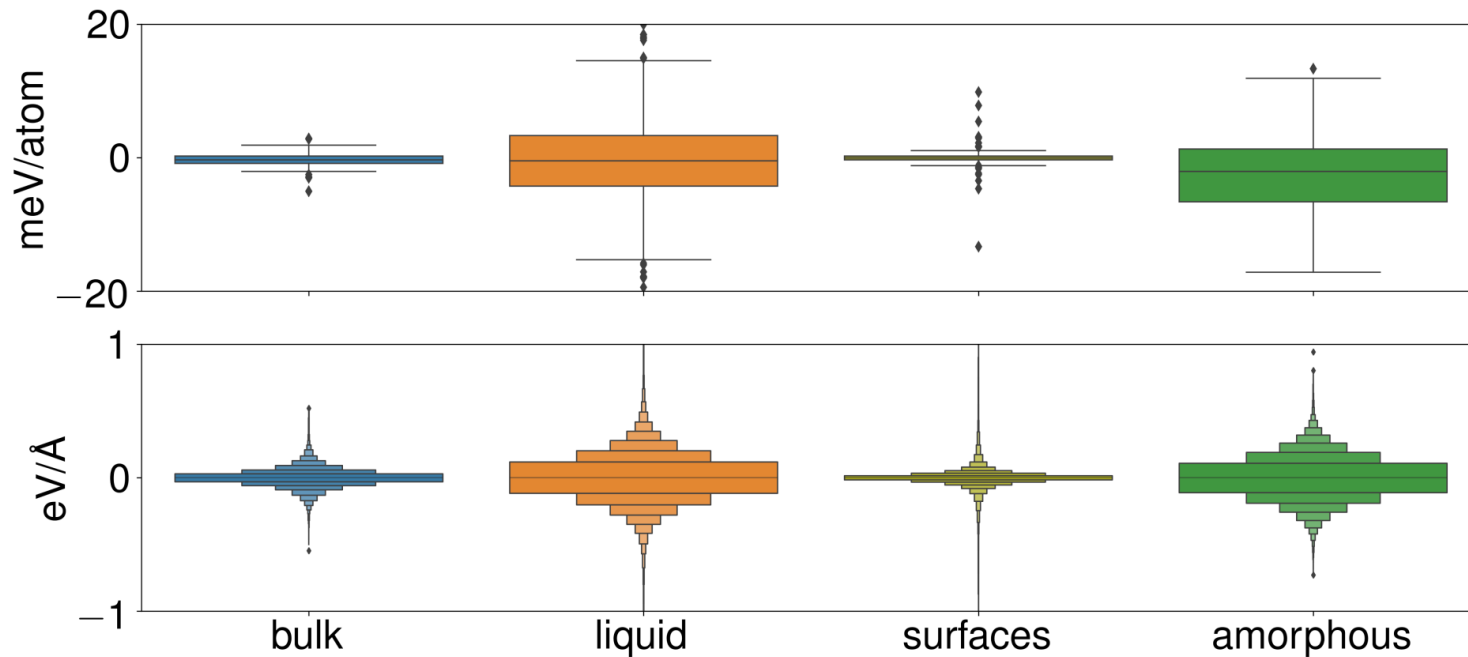


Dataset and validation



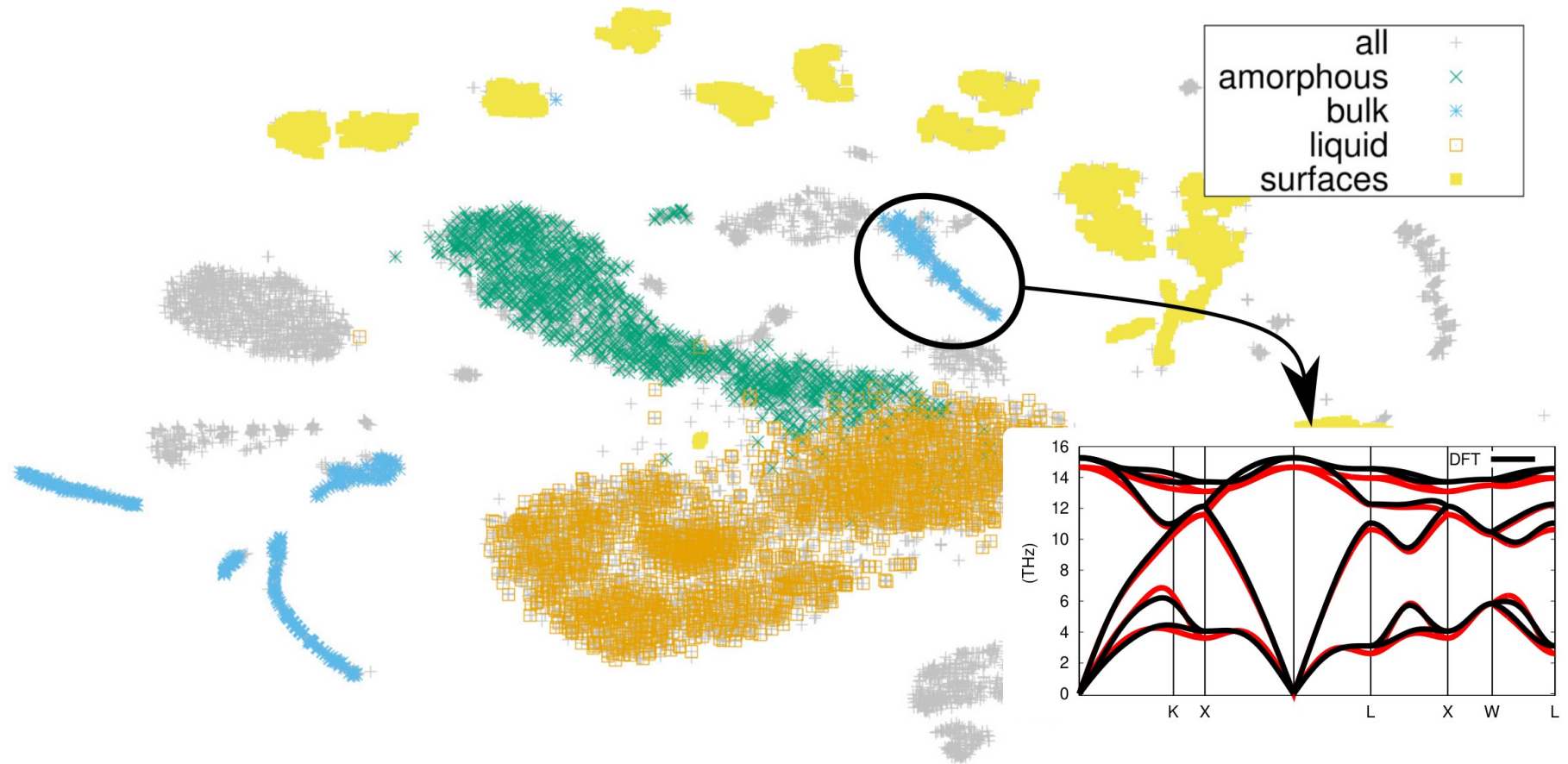
t-sne, ≈ 5000 points for $\approx 250k$ environments

Validation on the dataset

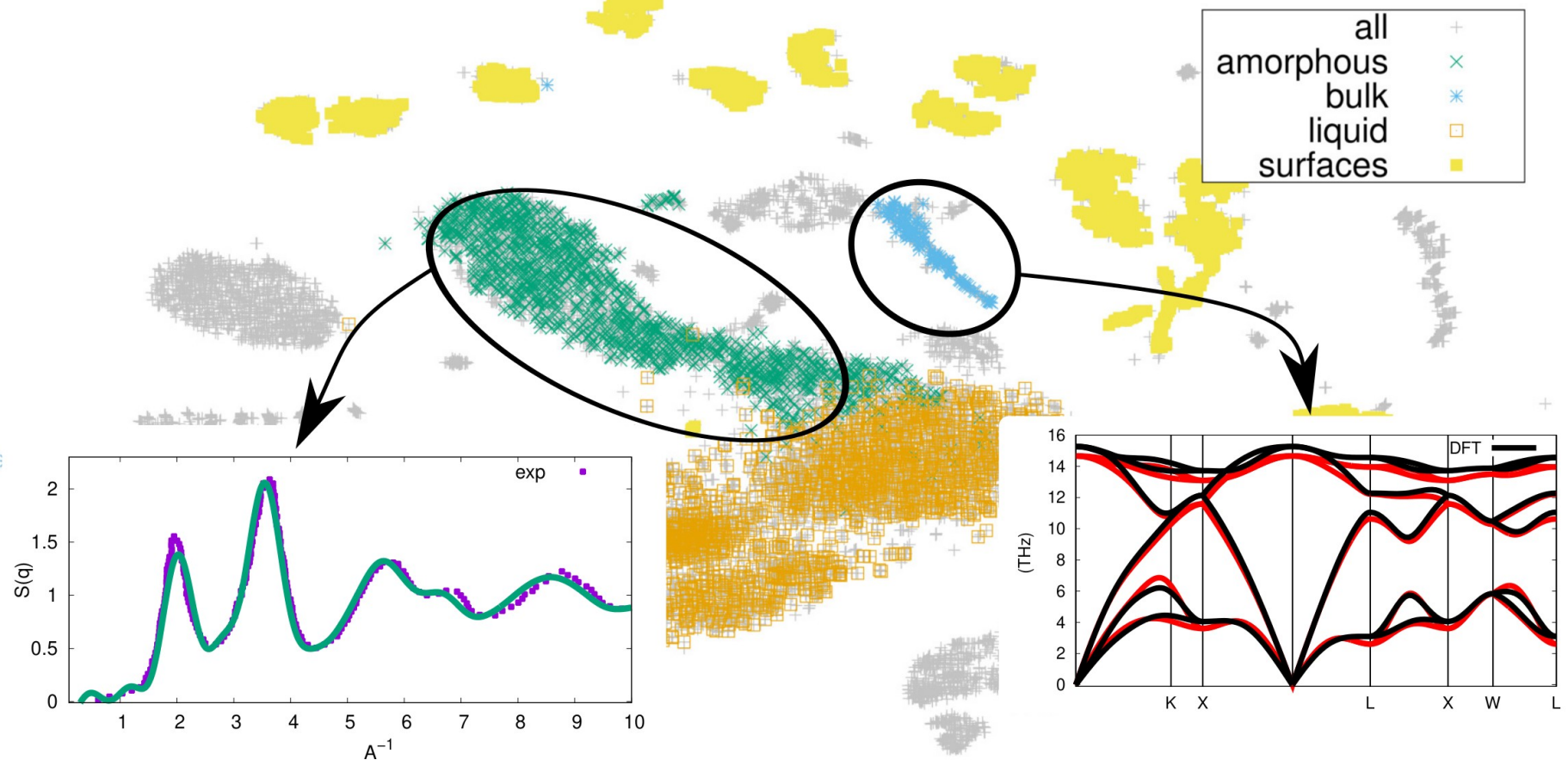


	bulk	liquid	surfaces	amorphous
Energy [meV/atom]	0.9	7.5	3.4	5.8
Forces [meV/Å]	66	196	103	170

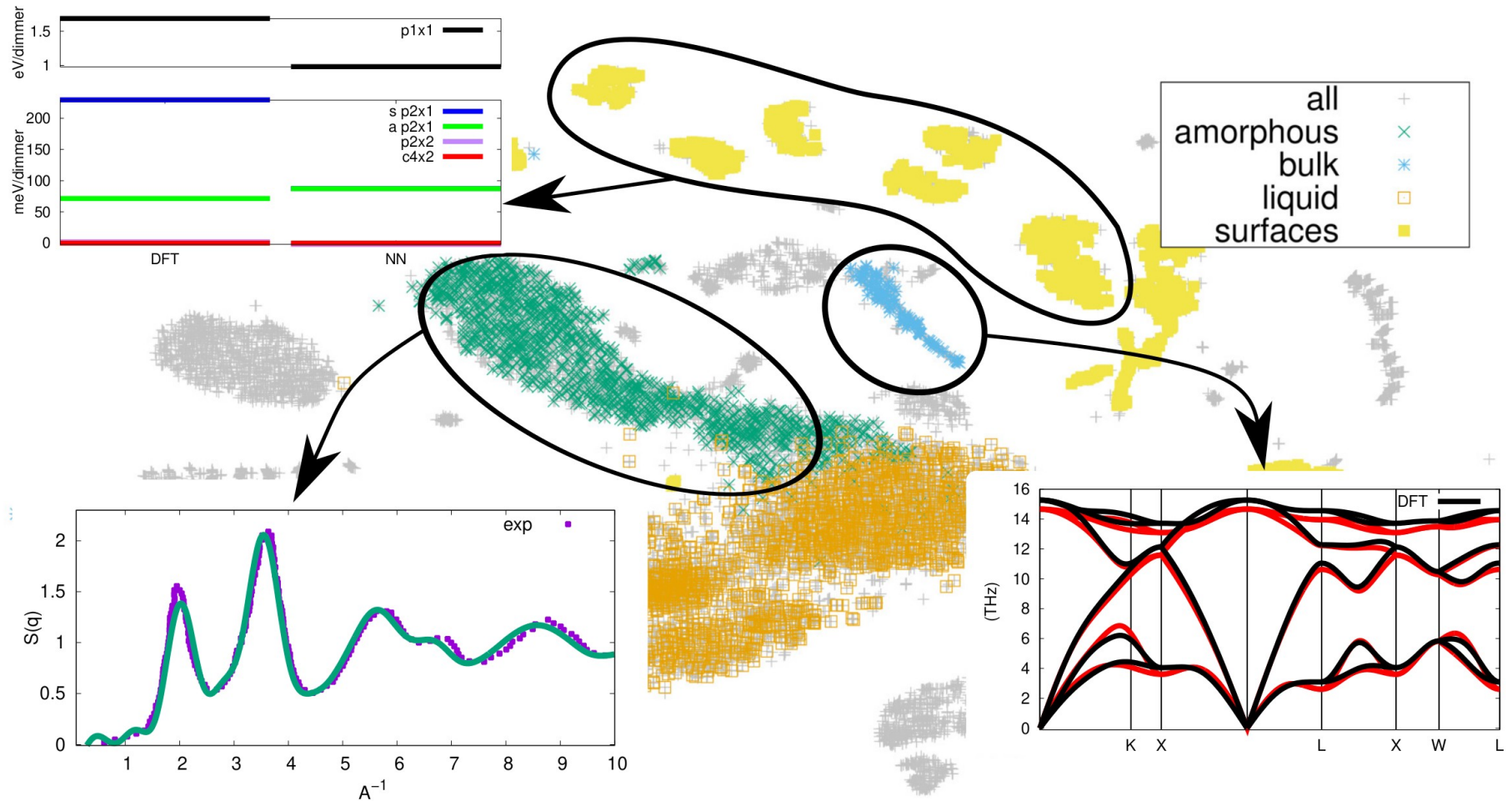
Validation on physical properties



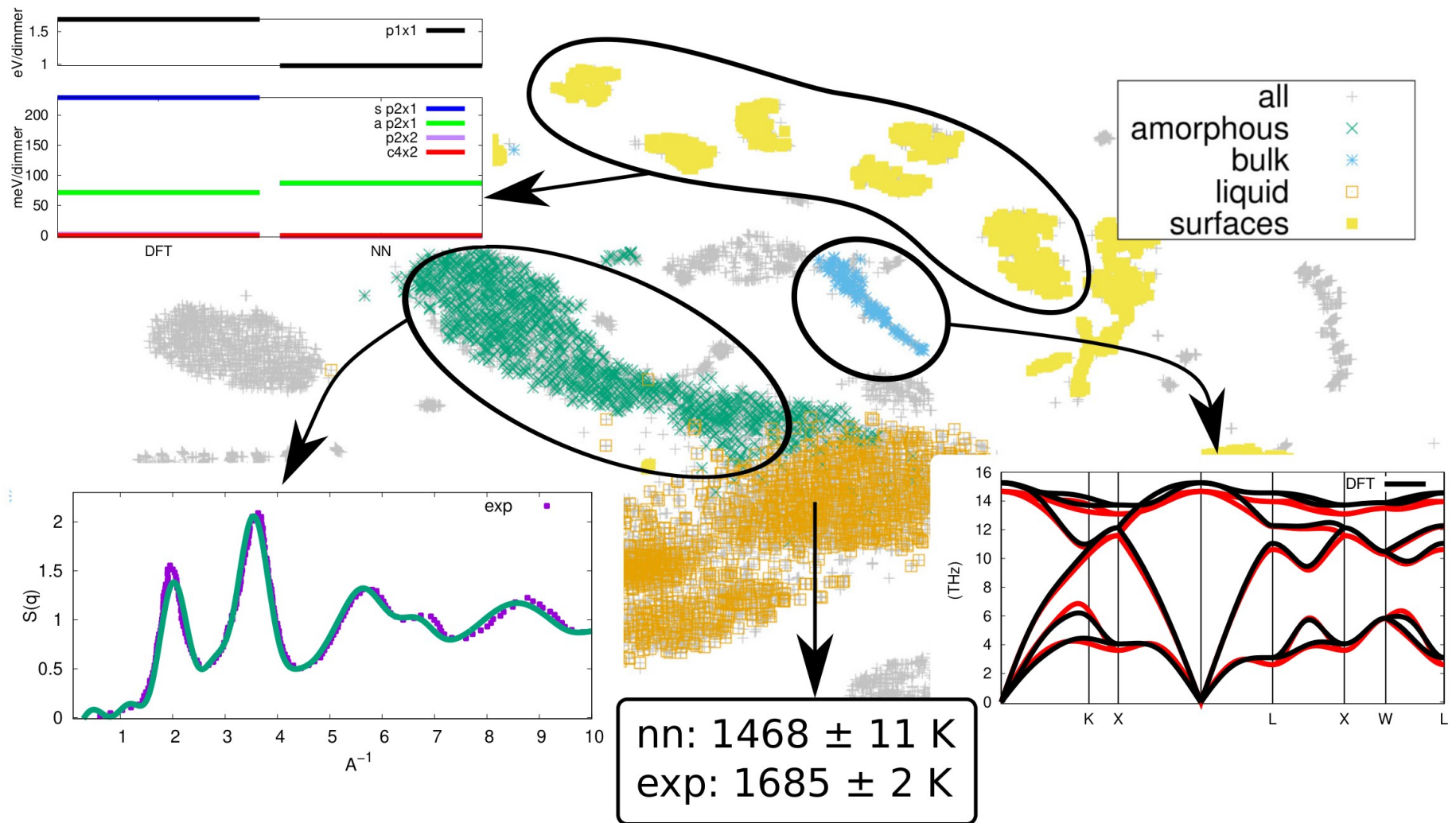
Validation on physical properties



Validation on physical properties

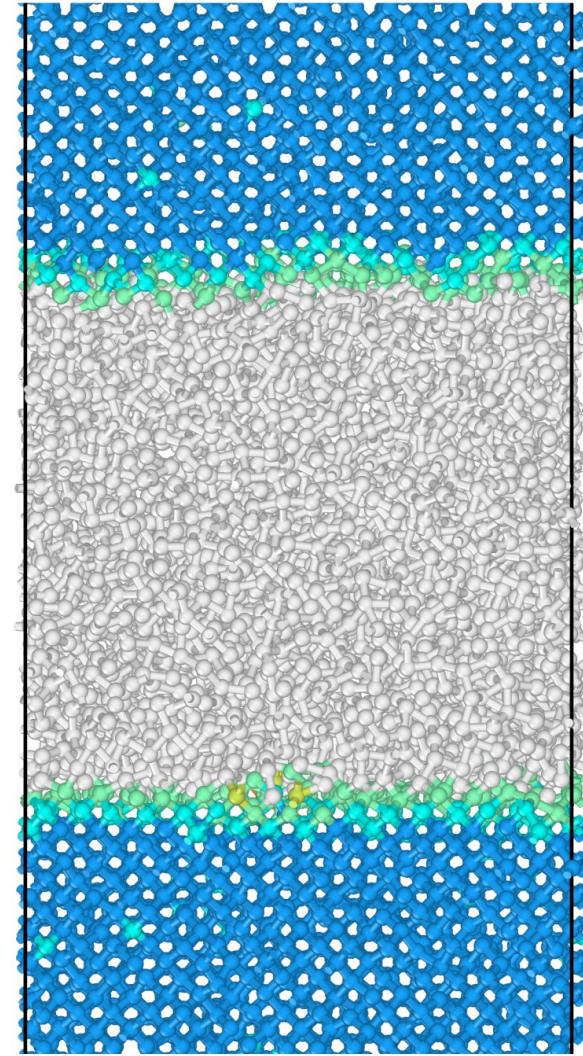


Validation on physical properties



Melting temperature

NN pbe	$1468 \pm 11\text{K}$
DFT pbe	$1540 \pm 50\text{K}$
NN PBESol	$1194 \pm 28\text{K}$
GAP PBESol	$1213 \pm 21\text{K}$
Experiments	$1685 \pm 2\text{K}$

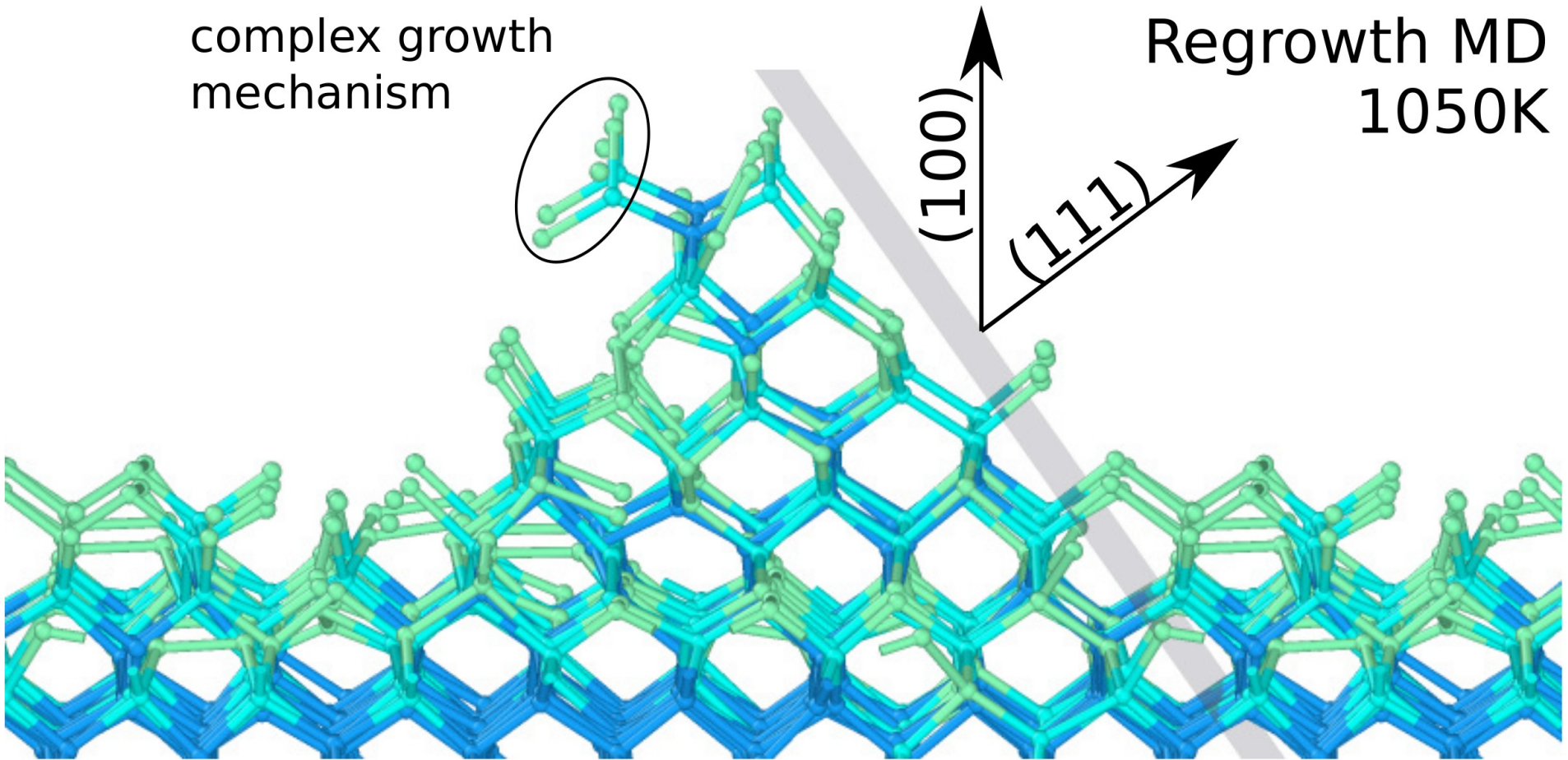


Yoo, Xantheas, and Zeng 2009; Jinnouchi, Karsai, and Kresse 2019

SPE, What do we see?

complex growth
mechanism

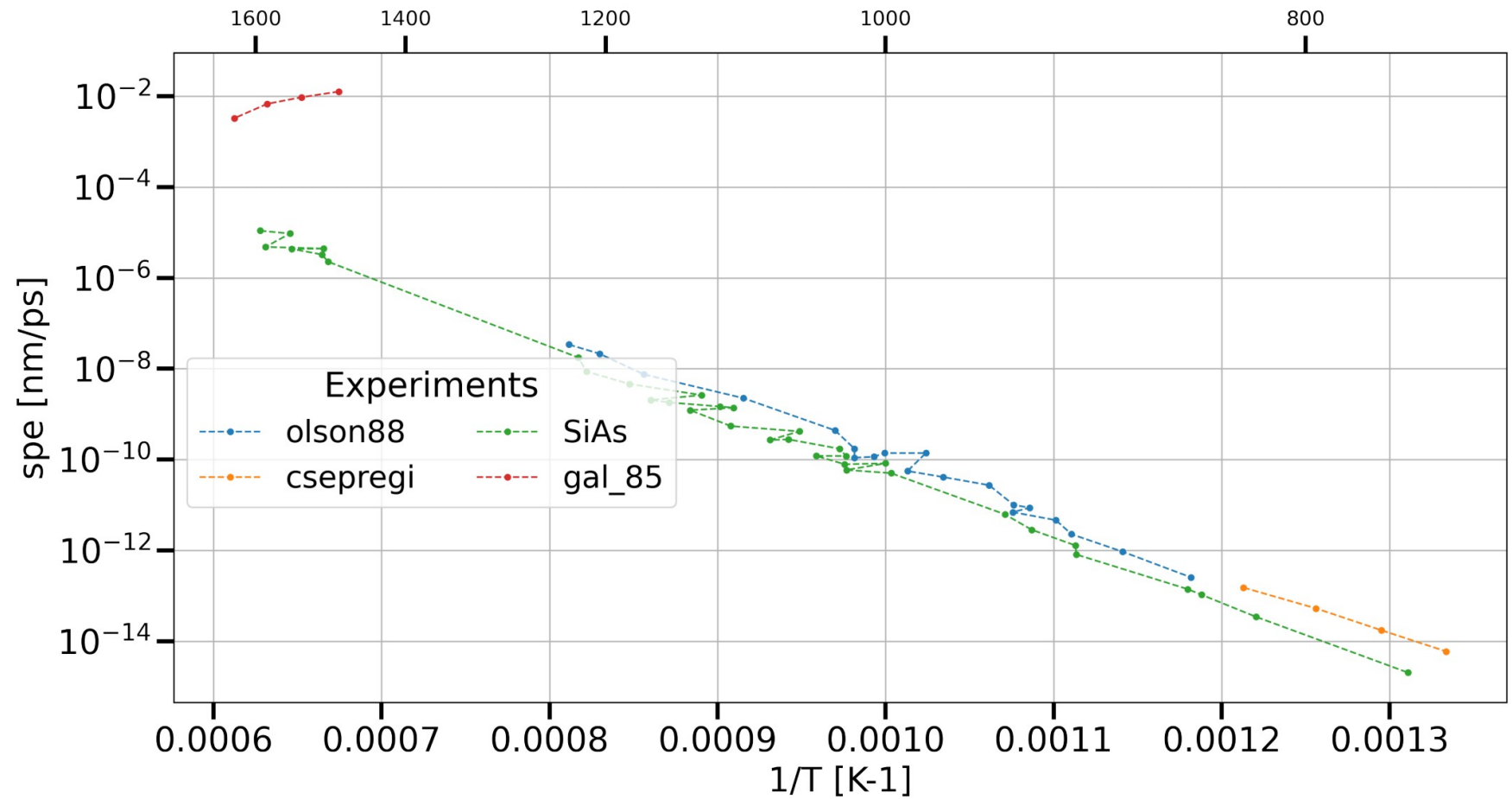
Regrowth MD
1050K



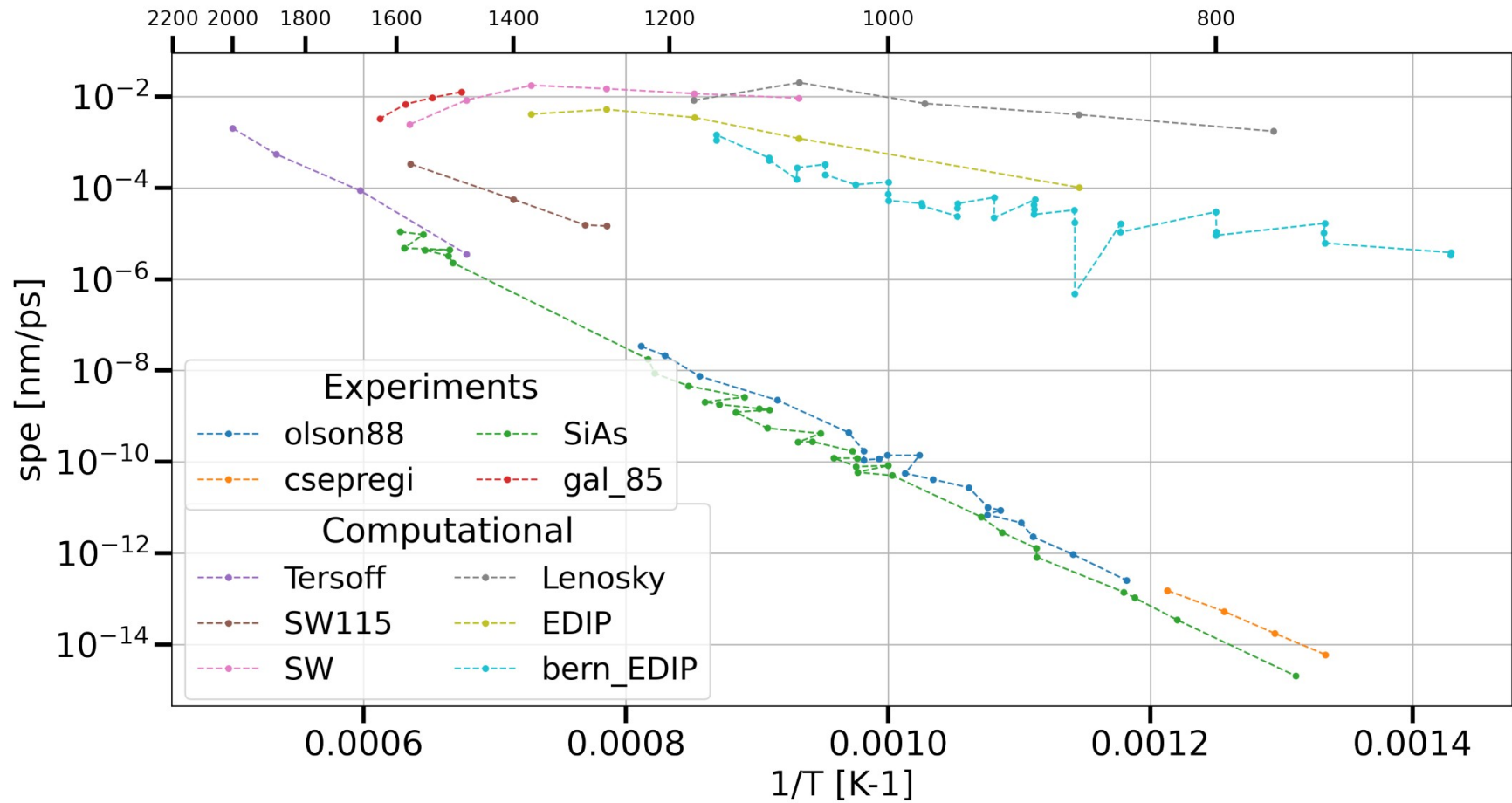
Thermally activated process:

$$v = v_0 \exp\left(-\frac{\Delta E}{k_b T}\right)$$

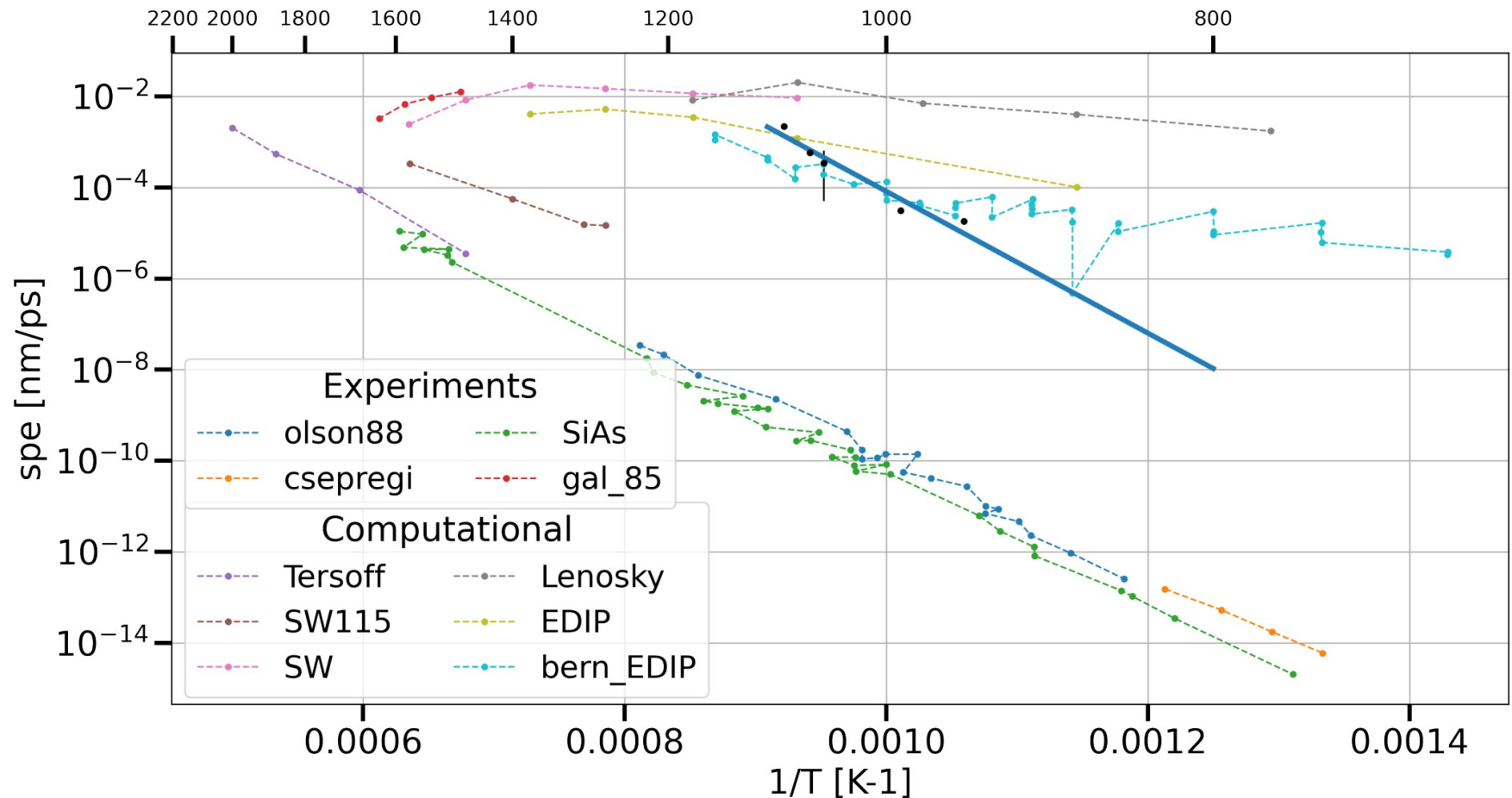
SPE, What do we see?



SPE, What do we see?



SPE, What do we see?



panna: 3.15 eV, experiments¹: 2.73 eV



Olson and Roth 1988.

Conclusion and outlook

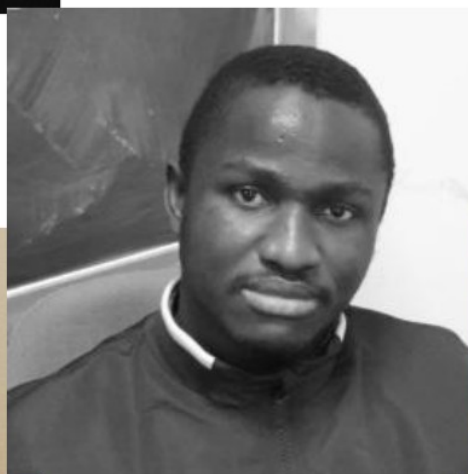
- NN potentials are a valid way to model physical phenomena at the atomistic level
- As a byproduct, PBE-sol xc-functional is not suitable to study thermal related phenomena in silicon
- We are obtaining a correct energy barrier for SPE with pure ab-initio data.
- Improve the amorphous quality
- Isolate the main events for SPE
- Develop a better KMC model.



Emine Kucukbenli



Ruggero Lot



Yusuf Shaidu



Franco Pellegrini



Stefano de Gironcoli