

# Foundations of Entropy II

## Let's calculate 'em

Lecture series at the  
School on Information, Noise, and Physics of Life  
Nis 19.-30. September 2022

by Jan Korbel  
all slides can be found at: **[slides.com/jankorbel](https://slides.com/jankorbel)**

# Activity II

You have 3 minutes to write down on a piece of paper:

What is the most unexpected/surprising application  
of **entropy**  
you have seen? (unexpected field, unexpected result)

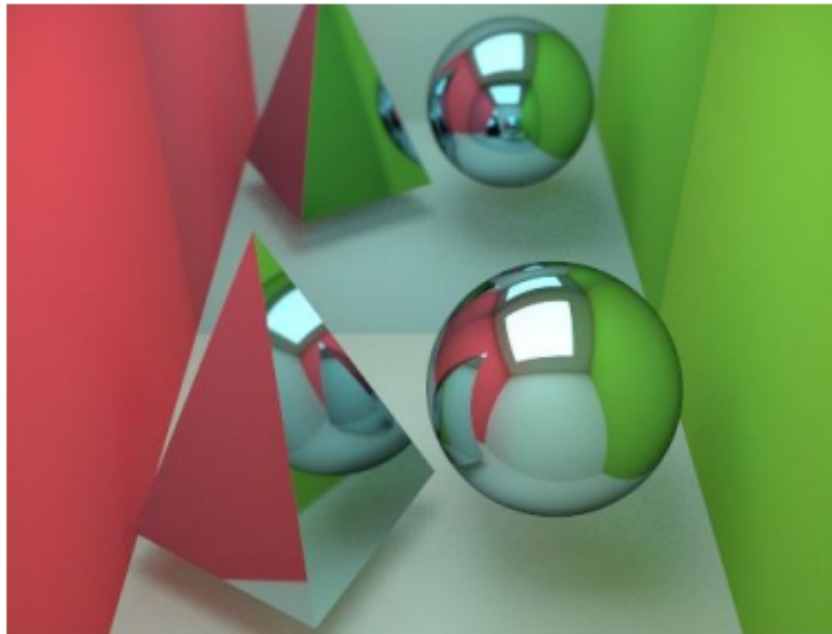
# Entropy-based Adaptive Sampling

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*Figure 4: Reference image used in the test in Figure 5 with 1024 rays per pixel.*

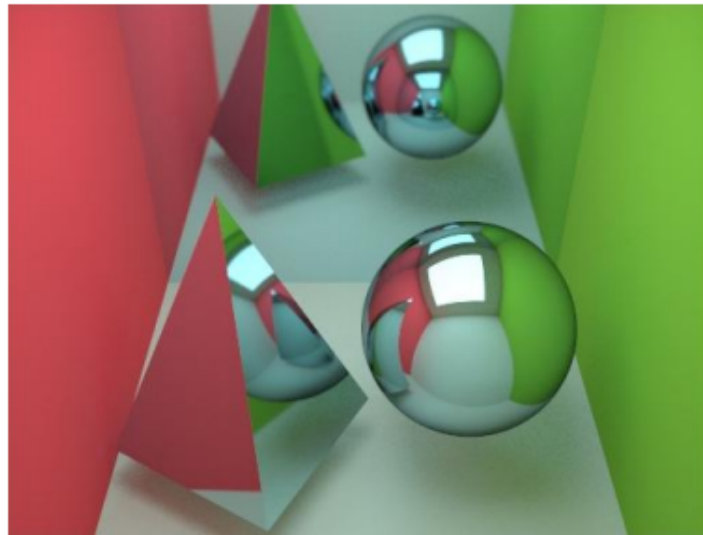
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# Using Boltzmann's formula for non-multinomial systems

As we saw in the previous lecture, the multinomial multiplicity

$$W(n_1, \dots, n_k) = \binom{n}{n_1, \dots, n_k}$$

leads to Boltzmann-Gibbs-Shannon entropy.

Are there systems with non-multinomial multiplicity?

What is their entropy?

# Ex. I: MB, FD & BE statistics

**1. Maxwell-Boltzmann statistics** -  $N$  distinguishable particles,  $N_i$  particles in state  $\epsilon_i$

Multiplicity can be calculated as

$$W(N_1, \dots, N_k) = \binom{N}{N_1} \binom{N - N_1}{N_2} \dots \binom{N - \sum_{i=1}^{k-1} N_i}{N_k} = N! \prod_{i=1}^k \frac{1}{N_i!}$$

If  $\epsilon_i$  has degeneracy  $g_i$ , then

$$W(N_1, \dots, N_k) = N! \prod_{i=1}^k \frac{g_i^{N_i}}{N_i!}$$

Then, we get

$$S_{MB} = -N \sum_{i=1}^k p_i \log \frac{p_i}{g_i}$$

# Ex. I: MB, FD & BE statistics

**2. Bose-Einstein statistics** -  $N$  indistinguishable particles,  $N_i$  particles in state  $\epsilon_i$  with degeneracy  $g_i$ ,

Multiplicity can be calculated as

$$W(N_1, \dots, N_k) = \prod_{i=1}^k \binom{N_i + g_i - 1}{N_i} \quad (|*)$$

Let us introduce  $\alpha_i = g_i/N$ . Then, we get

$$S_{BE} = N \sum_{i=1}^k [(\alpha_i + p_i) \log(\alpha_i + p_i) - \alpha_i \log \alpha_i - p_i \log p_i]$$

# Ex. I: MB, FD & BE statistics

**3. Fermi-Dirac statistics** -  $N$  indistinguishable particles,  $N_i$  particles in state  $\epsilon_i$  with degeneracy  $g_i$ , maximally 1 particle per sub-level (thus  $N_i \leq g_i$ )

Multiplicity can be calculated as

$$W(N_1, \dots, N_k) = \prod_{i=1}^k \binom{g_i}{N_i}$$

Let us introduce  $\alpha_i = g_i/N$ . Then, we get

$$S_{FD} = N \sum_{i=1}^k [-(\alpha_i - p_i) \log(\alpha_i - p_i) + \alpha_i \log \alpha_i - p_i \log p_i]$$




# Ex. II: structure-forming systems






















Let us start with a simple example of a coin tossing.

States are:  

**But!** let's make a small change, we consider *magnetic coins*


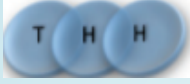




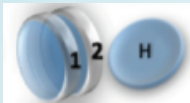

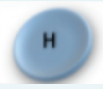

The bound (or sticky) state is simply 

State space grows super-exponentially. ( $W(n) \sim n^n \sim e^{n \log n}$ )

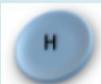

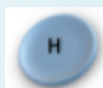

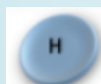




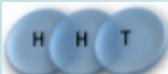





N	W(N)	Configurations
1	2	 
2	5	    
3	14	<div>      </div> <div>      </div> <div>     </div>
4	43	$W(4) = 2 W(3) + 3 W(2)$

picture taken from: Jensen et al 2018 J. Phys. A: Math. Theor. 51 375002

# Multiplicity

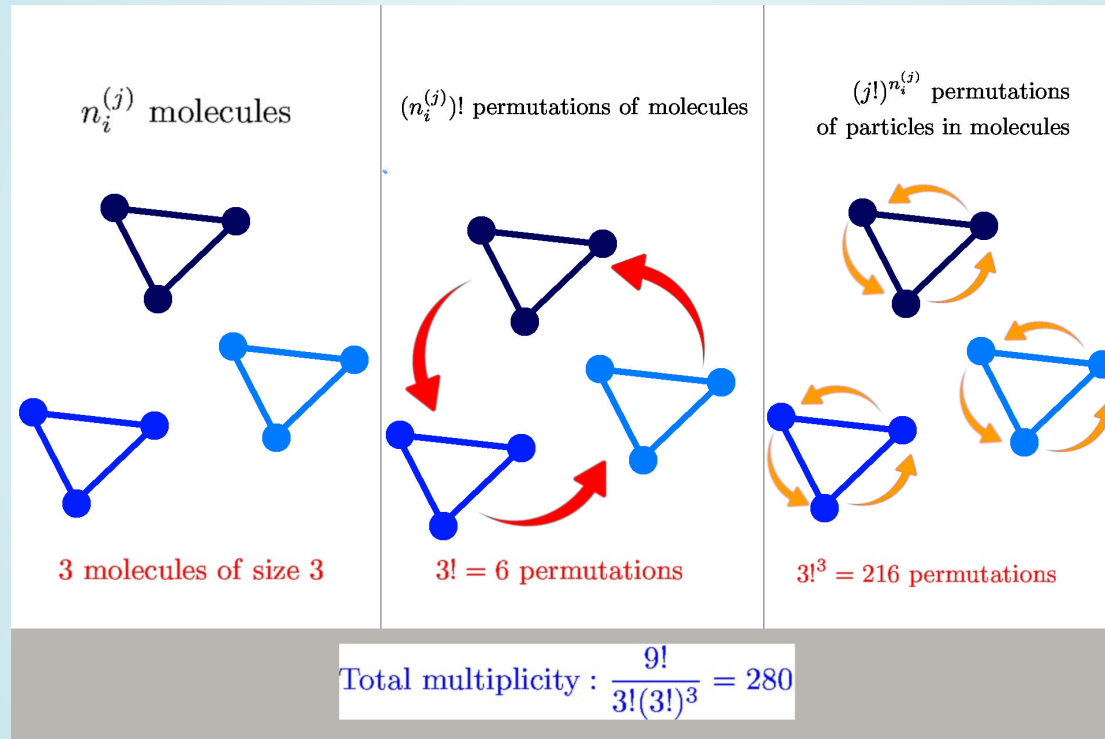
Microstate	Mesostate	Mesostate
  	$2 \times$  $1 \times$ 	3
  	$1 \times$  $1 \times$ 	3

## How to calculate multiplicity

2 x 	1x 							1 x 	1x 																																		
	1	1	2	2	3	3								1	1	2	2	3	3																								
	2	3	1	3	1	2								2	3	1	3	1	2																								
	3	2	3	1	2	1								3	2	3	1	2	1																								
	(1,2,3)			(2,1,3)																			(1,2,3)			(1,3,2)																	
	(1,3,2)			(3,1,2)																			(2,1,3)			(2,3,1)																	
	(2,3,1)			(3,2,1)																			(3,1,2)			(3,2,1)																	

# Multiplicity

we have  $n_i^{(j)}$  molecules of size  $j$  in a state  $s_i^{(j)}$



General formula:  $W(n_i^{(j)}) = \frac{n!}{\prod_{ij} n_i^{(j)}! (j!)^{n_i^{(j)}}}$

# Boltzmann's 1884 paper

## III. *Ueber das Arbeitsquantum, welches bei chemischen Verbindungen gewonnen werden kann; von Ludwig Boltzmann in Graz.*

(Aus dem 88. Bde. der Sitzungsber. der k. Akad. der Wiss. zu Wien, II. Abth. vom 18. Oct. 1883 mitgetheilt vom Hrn. Verf.)

Es seien z. B.  $A$  Chlor- und  $B$  Wasserstoffatome gegeben. Es wird gefragt, wie wahrscheinlich es ist, dass sich daraus gerade  $N_1$  Chlor-,  $N_2$  Wasserstoff- und  $N_3$  Chlorwasserstoffmoleküle bilden. Hier ist  $a_1 = 2$ ,  $b_1 = 0$ ;  $a_2 = 0$ ,  $b_2 = 2$ ;  $a_3 = 1$ ,  $b_3 = 1$ ; die Anzahl der Chloratome ist  $A = 2 N_1 + N_3$ ; die Anzahl der Wasserstoffatome ist  $B = 2 N_2 + N_3$ ; die Anzahl der Bildungsweisen:

$$Z = \frac{A! B!}{2^{N_1 + N_2} \cdot N_1! N_2! N_3!}$$

# Entropy of structure-forming systems

$$S = \log W \approx n \log n - \sum_{ij} \left( n_i^{(j)} \log n_i^{(j)} - n_i^{(j)} + n_i^{(j)} \log j! \right)$$

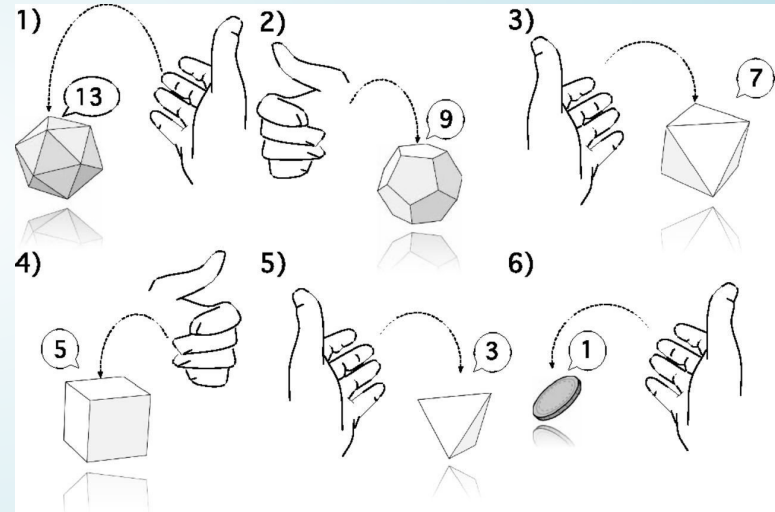
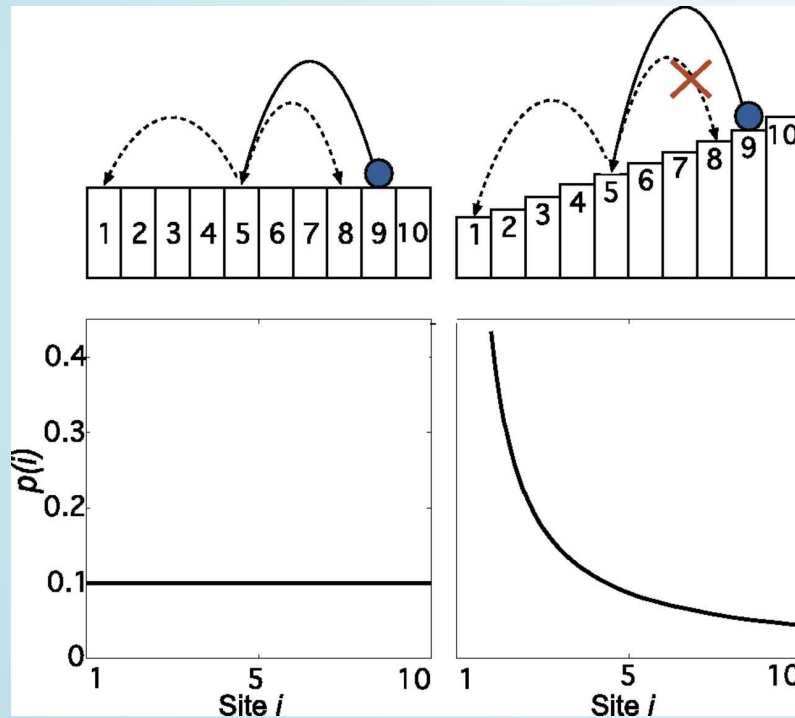
Introduce "probabilities"  $\wp_i^{(j)} = n_i^{(j)} / n$

$$\mathcal{S} = S/n = - \sum_{ij} \wp_i^{(j)} (\log \wp_i^{(j)} - 1) - \sum_{ij} \wp_i^{(j)} \log \frac{j!}{n^{j-1}}$$

Finite interaction range: concentration  $c = n/b$

$$\mathcal{S} = S/n = - \sum_{ij} \wp_i^{(j)} (\log \wp_i^{(j)} - 1) - \sum_{ij} \wp_i^{(j)} \log \frac{j!}{c^{j-1}}$$

# Ex. III: sample-space reducing processes (SSR)



# Multiplicity

The number of states is  $n$ . Let us denote the states as  $x_n \rightarrow \cdots \rightarrow x_1$ , where  $x_1$  is the ground state, where the process restarts. Let us sample  $R$  relaxation sequences  $x = (x_{k_1}, \dots, x_1)$ .

The sequences can be visualised as

$r \times i$	$W$	$W - 1$	$W - 2$	$\dots$	$2$	$1$
1	*	—	—	$\dots$	*	*
2	—	*	*	$\dots$	—	*
3	*	—	*	$\dots$	—	*
$\vdots$	$\vdots$	$\vdots$	$\vdots$		$\vdots$	$\vdots$
$R - 2$	—	*	*	$\dots$	—	*
$R - 1$	—	*	—	$\dots$	*	*
$R$	—	—	*	$\dots$	—	*
	$k_W$	$k_{W-1}$	$k_{W-2}$	$\dots$	$k_2$	$k_1$

Each run must contain  $x_1$

How many of these sequences contain a state  $x_j$  exactly  $k_j$  times?



# Multiplicity

Number of runs  $R \equiv k_1$ , number of them containing  $x_j$  is  $k_j$

Multiplicity of these sequences:  $\binom{k_1}{k_j}$

By multiplying the multiplicity for each state we get

$$W(k_1, \dots, k_n) = \prod_{j=2}^n \binom{k_1}{k_j}$$

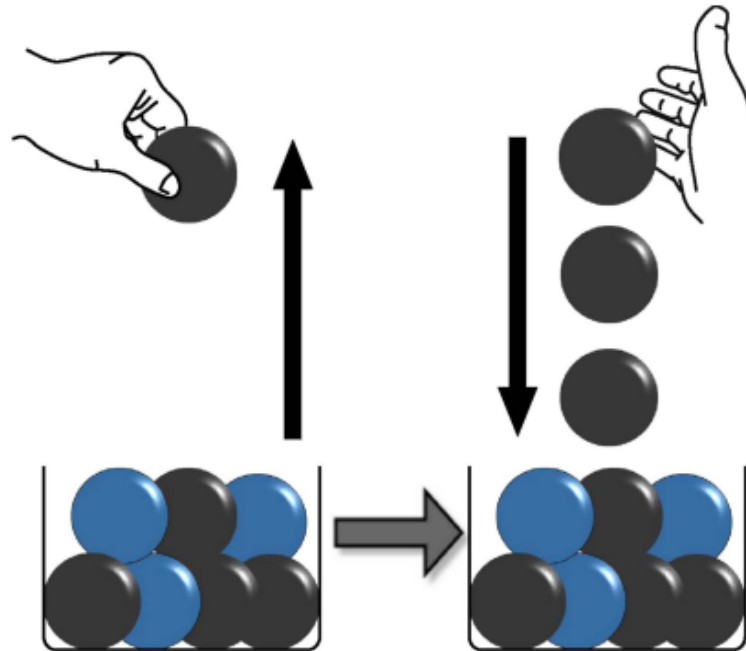
$$\begin{aligned} \log W &\approx \sum_{j=2}^n [k_1 \log k_1 - \cancel{k_1} - k_j \log k_j + \cancel{k_j} - (k_1 - k_j) \log(k_1 - k_j) + \cancel{(k_1 - k_j)}] \\ &\approx \sum_{j=2}^n \left[ k_1 \log k_1 - \textcolor{red}{k_j} \log \textcolor{red}{k_1} - k_j \log \frac{k_j}{\textcolor{red}{k_1}} - (k_1 - k_j) \log(k_1 - k_j) \right] \end{aligned}$$

By introducing  $p_i = k_i/N$  where  $N$  is the total number of steps, we get

$$S_{SSR}(p) = -N \sum_{j=2}^n \left[ p_i \log \left( \frac{p_i}{p_1} \right) + (p_1 - p_i) \log \left( 1 - \frac{p_i}{p_1} \right) \right]$$



# Ex. IV: Pólya urns



**Figure 1.** Schematic illustration of a Pólya process. When a ball of a certain color is drawn, it is replaced by  $1 + \delta$  balls of the same color. Then the next ball is drawn and the process is repeated for  $N$  iterations. Here  $\delta = 2$ . This reinforcement process creates a history-dependent dynamics. The configurations obtained after successive iterations have non-multinomial structure.

# Probability of a sequence

We have  $c$  colors, initially  $n_i(0) \equiv n_i$  balls of color  $c_i$ . After a ball is drawn, we return  $\delta$  balls of the same color to the urn. After  $N$  draws, the number of balls in the urn is

$$n_i(N) = n_i + \delta k_i$$

where  $k_i$  is the number of draws of color  $c_i$ . The total number of balls is  $n(N) = \sum_c n_c(N) = N + \delta N$

The probability of drawing a ball of color  $c_i$  in  $N$ -th run, is  $p_i(N) = n_i(N)/n(N)$ . The probability of sequence

$\mathcal{I} = \{i_1, \dots, i_N\}$  is

$$p(\mathcal{I}) = \prod_{j=1}^c \frac{n_j^{(\delta, k_j)}}{n^{(\delta, N)}}$$

where  $m^{(\delta, r)} = m(m + \delta) \dots (m + r\delta)$

# Probability of a histogram

A histogram  $\mathcal{K} = \{k_1, \dots, k_c\}$  is defined as  $k_c = \sum_{j=1}^N \delta(i_j, c)$

Thus the probability of observing a histogram is

$$p(\mathcal{K}) = \binom{N}{k_1, \dots, k_c} p(\mathcal{I})$$

$$n_j^{(\delta, k_j)} \approx k_j! \delta^{k_j} (k_i + 1)^{n_i/\delta}$$

$$p(\mathcal{K}) = \frac{N!}{\prod_{j=1}^c k_j!} \frac{\prod_{j=1}^c k_j! \delta^{k_j} (k_j + 1)^{n_j/\delta}}{n^{(\delta, N)}}$$

... technical calculation ...

$$S_{Polya}(p) = \log p(\mathcal{K}) \approx - \sum_{i=1}^c \log(p_i + 1/N)$$

# Ex. IV: q-deformations

This example is rather theoretical, but provides us a useful hint of what happens if there are correlations in the sample space

Motivation: finite versions of  $\exp$  and  $\log$

$$\exp(x) = \lim_{n \rightarrow \infty} \left(1 + \frac{x}{n}\right)^n$$

So define

$$\exp_q(x) := (1 + (1 - q)x)^{1/(1-q)}$$

$$\log_q(x) := \frac{x^{1-q} - 1}{1 - q}$$

Let us find an operation s.t.

$$\begin{aligned} \exp_q(x) \otimes_q \exp_q(y) &\equiv \exp_q(x + y) \\ \Rightarrow a \otimes_q b &= [a^{1-q} + b^{1-q} - 1]^{1/(1-q)} \end{aligned}$$

# Calculus of q-deformations

In analogy to  $n! = 1 \cdot 2 \cdot \dots \cdot n$  introduce  $n!_q := 1 \otimes_q 2 \otimes_q \dots \otimes_q n$

It is then easy to show that  $\log_q n!_q = \frac{\sum_{k=1}^n k^{1-q} - n}{1-q}$  which can be used for generalized Stirling's approximation  $\log_q n!_q \approx \frac{n}{2-q} \log_q n$

Let us now consider a q-deformed multinomial factor

$$\begin{aligned} \binom{n}{n_1, \dots, n_k}_q &:= n!_q \oslash_q (n_1!_q \otimes_q \dots \otimes_q n_k!_q) \\ &= \left[ \sum_{l=1}^n l^{1-q} - \sum_{i_1}^{n_1} i_1^{1-q} - \dots - \sum_{i_k}^{n_k} i_k^{1-q} \right]^{1/(1-q)} \end{aligned}$$

# Tsallis entropy

Let us consider that the multiplicity is given by a q-multinomial factor  $W(n_1, \dots, n_k) = \binom{n}{n_1, \dots, n_k}_q$

In this case, it is more convenient to define entropy as  $S = \log_q W$ , which gives us:

$$\log_q W = \frac{n^{2-q}}{2-q} \frac{\sum_{i=1}^k p_i^{2-q} - 1}{q-1} = \frac{n^{2-q}}{2-q} S_{2-q}(p)$$

This entropy is known as **Tsallis entropy**

Note that the prefactor is not  $n$  but  $n^{2-q}$

(*non-extensivity*) - we will discuss this later

# Entropy and energy

Until now, we have been just counting states; let us now discuss the relation with **energy**.

We consider that the states describe the energy of the system (either Hamiltonian or more generalized energy functional)

Therefore, entropy is defined as

$$S(E) := \log W(E)$$

# Ensembles

There are a few typical situations:

## 1. Isolated system = microcanonical ensemble

Let  $H(s)$  be the energy of a state  $s$ . Multiplicity is then

$$W(E) = \sum_s \delta(H(s) - E)$$

Phenomena like negative "temperature"  $T = \frac{dS(E)}{dE} < 0$

## 2. closed system = canonical ensemble

Total system is composed of the system of interest (S) and the heat reservoir/bath (B). They are weakly coupled i.e.,

$H_{tot}(s, b) = H_S(s) + H_B(b)$  (no interaction energy)

$$W(E_{tot}) = \sum_{s,b} \delta(H_{tot}(s, b) - E_{tot})$$

## 3. open system = grandcanonical ensemble



# Entropy in canonical ensemble

This can be further rewritten as

$$\begin{aligned} &= \int dE_S \sum_s \delta(H_S(s) - E_S) \sum_b (H_B(b) - (E_{tot} - E_S)) \\ &= \int dE_S W_S(E_S) W_B(E_{tot} - E_S) \end{aligned}$$

This is hard to calculate. Typically, the dominant contribution is from the maximal configuration of the integrand, which we obtain from

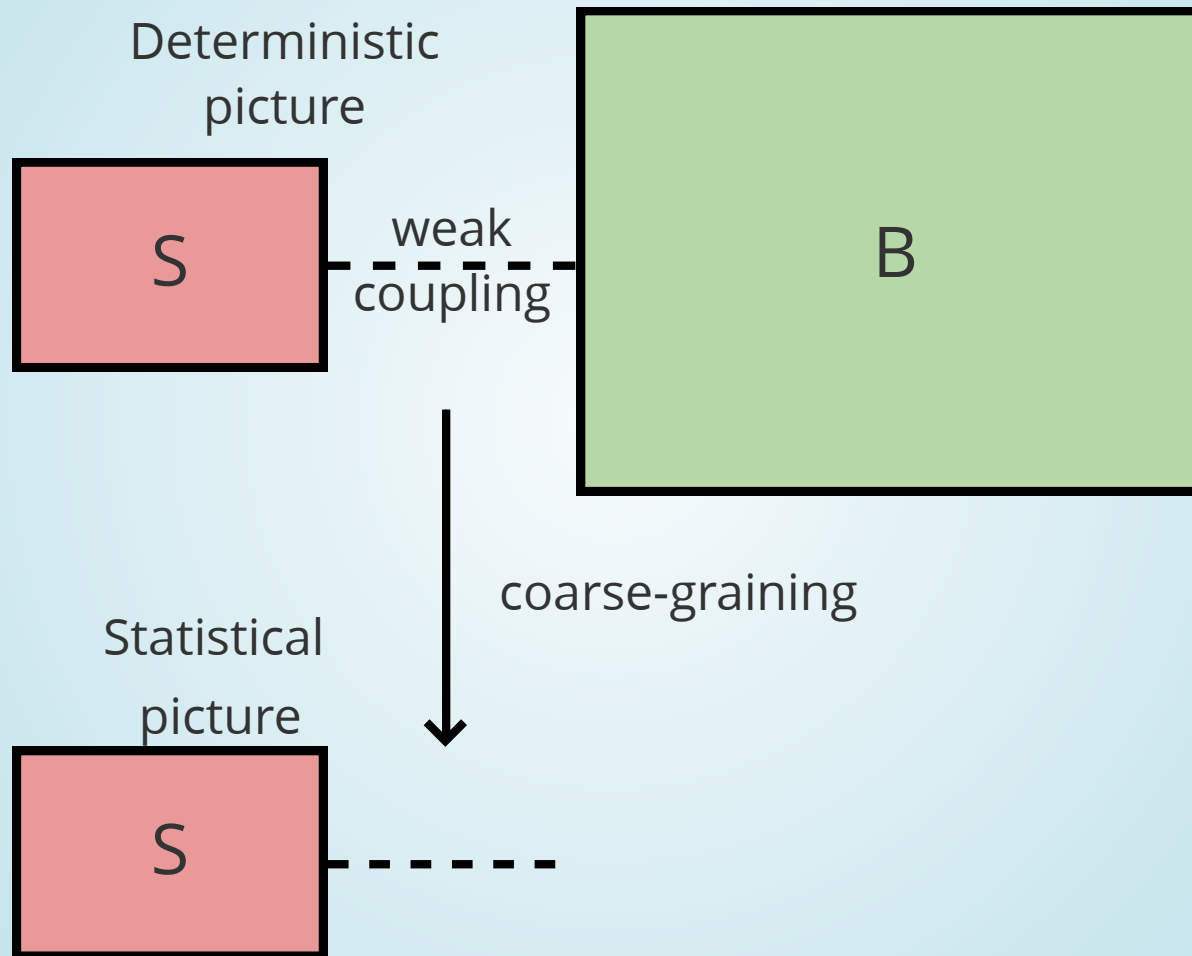
$$\frac{\partial W_S(E_S) W_B(E_{tot} - E_S)}{\partial E_S} \stackrel{!}{=} 0 \Rightarrow \frac{W'_S(E_S)}{W_S(E)} = \frac{W'_B(E_{tot} - E_S)}{W_B(E_{tot} - E_S)}$$

$$\text{As a consequence } \frac{\partial S_E(E_S)}{\partial E_S} \stackrel{!}{=} \frac{\partial S_B(E_{tot} - E_S)}{\partial E_S} := \frac{1}{k_B T}$$

$$\text{and } \underbrace{S_B(E_{tot} - E_S)}_{\text{free entropy}} = \underbrace{S_B(E_{tot})}_{\text{bath entropy}} - \frac{\partial S}{\partial E_S} E_S + \dots$$

This is the emergence of **Maximum entropy principle**

# Canonical ensemble & Coarse-graining



# Summary