

Foundations of Entropy II

Let's calculate 'em

Lecture series at the
School on Information, Noise, and Physics of Life
Nis 19.-30. September 2022

by Jan Korbel

all slides can be found at: slides.com/jankorbel

Activity II

You have 3 minutes to write down on a piece of paper:

What is the most unexpected/surprising application
of **entropy**
you have seen? (unexpected field, unexpected result)

Entropy-based Adaptive Sampling

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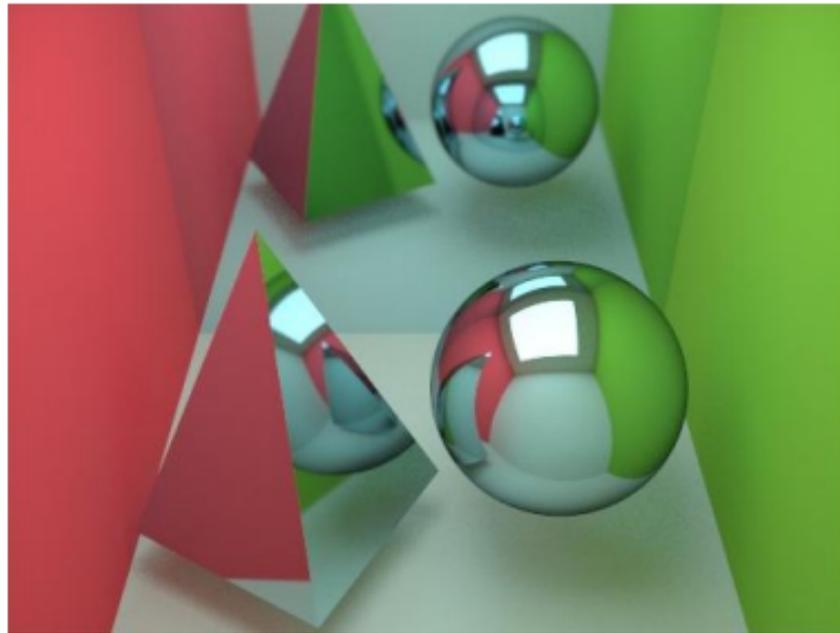


Figure 4: Reference image used in the test in Figure 5 with 1024 rays per pixel.

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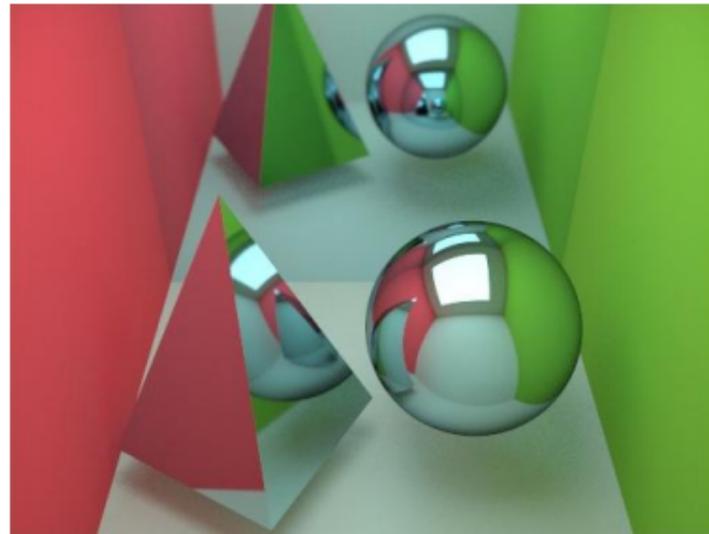


Figure 4: Reference image used in the test in Figure 5 with 1024 rays per pixel.

Using Boltzmann's formula for non-multinomial systems

As we saw in the previous lecture, the multinomial multiplicity

$$W(n_1, \dots, n_k) = \binom{n}{n_1, \dots, n_k}$$

leads to Boltzmann-Gibbs-Shannon entropy.

Are there systems with non-multinomial multiplicity?

What is their entropy?

Ex. I: MB, FD & BE statistics

1. Maxwell-Boltzmann statistics - N distinguishable particles, N_i particles in state ϵ_i

Multiplicity can be calculated as

$$W(N_1, \dots, N_k) = \binom{N}{N_1} \binom{N - N_1}{N_2} \dots \binom{N - \sum_{i=1}^{k-1} N_i}{N_k} = N! \prod_{i=1}^k \frac{1}{N_i!}$$

If ϵ_i has degeneracy g_i , then

$$W(N_1, \dots, N_k) = N! \prod_{i=1}^k \frac{g_i^{N_i}}{N_i!}$$

Then, we get

$$S_{MB} = -N \sum_{i=1}^k p_i \log \frac{p_i}{g_i}$$

Ex. I: MB, FD & BE statistics

2. Bose-Einstein statistics - N indistinguishable particles, N_i particles in state ϵ_i with degeneracy g_i ,

Multiplicity can be calculated as

$$W(N_1, \dots, N_k) = \prod_{i=1}^k \binom{N_i + g_i - 1}{N_i} \quad (|*)$$

Let us introduce $\alpha_i = g_i/N$. Then, we get

$$S_{BE} = N \sum_{i=1}^k [(\alpha_i + p_i) \log(\alpha_i + p_i) - \alpha_i \log \alpha_i - p_i \log p_i]$$

Ex. I: MB, FD & BE statistics

3. Fermi-Dirac statistics - N indistinguishable particles, N_i particles in state ϵ_i with degeneracy g_i , maximally 1 particle per sub-level (thus $N_i \leq g_i$)

Multiplicity can be calculated as

$$W(N_1, \dots, N_k) = \prod_{i=1}^k \binom{g_i}{N_i}$$

Let us introduce $\alpha_i = g_i/N$. Then, we get

$$S_{FD} = N \sum_{i=1}^k [-(\alpha_i - p_i) \log(\alpha_i - p_i) + \alpha_i \log \alpha_i - p_i \log p_i]$$

Ex. II: structure-forming systems

Let us start with a simple example of a coin tossing.

States are:  

But! let's make a small change, we consider *magnetic coins*

The bound (or sticky) state is simply 

State space grows super-exponentially. ($W(n) \sim n^n \sim e^{n \log n}$)

N	W(N)	Configurations
1	2	 
2	5	    
3	14	<div style="display: flex; justify-content: space-between;"> <div style="text-align: center;">      </div> <div style="text-align: center;">      </div> </div> <div style="text-align: right; margin-top: 10px;"> $\left. \begin{array}{l} \text{ } \\ \text{ } \end{array} \right\} 2W(2)$ </div> <div style="text-align: center; margin-top: 10px;">     </div> <div style="text-align: right; margin-top: 10px;"> $\left. \text{ } \right\} 2W(1)$ </div>
4	43	$W(4) = 2W(3) + 3W(2)$

picture taken from: Jensen et al 2018 J. Phys. A: Math. Theor. 51 375002

Multiplicity

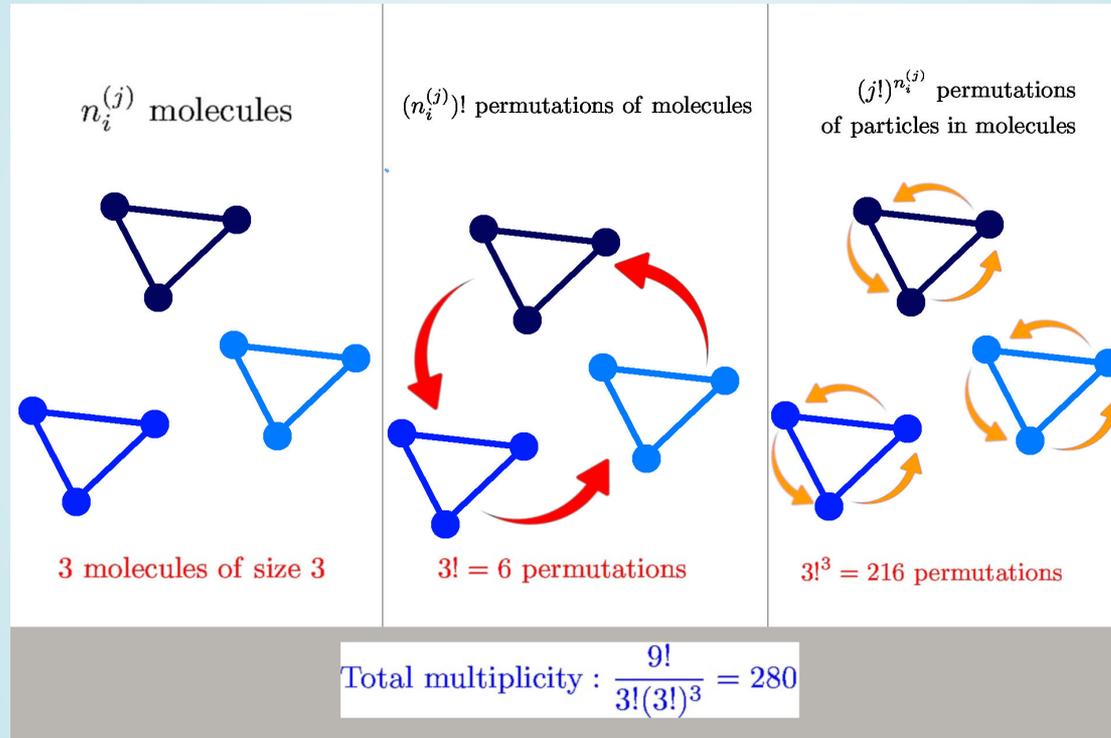
Microstate			Mesostate		Mesostate		
			2 x		1 x		3
			1 x		1 x		3

How to calculate multiplicity

2 x 1 x		1 x 1 x				
	1	1	2	2	3	3
	2	3	1	3	1	2
	3	2	3	1	2	1
	= (1,2,3), (2,1,3)					
	= (1,3,2), (3,1,2)					
	= (2,3,1), (3,2,1)					
	1	1	2	2	3	3
	2	3	1	3	1	2
	3	2	3	1	2	1
	= (1,2,3), (1,3,2)					
	= (2,1,3), (2,3,1)					
	= (3,1,2), (3,2,1)					

Multiplicity

we have $n_i^{(j)}$ molecules of size j in a state $s_i^{(j)}$



General formula:
$$W(n_i^{(j)}) = \frac{n!}{\prod_{ij} n_i^{(j)}! (j!)^{n_i^{(j)}}}$$

Boltzmann's 1884 paper

III. Ueber das Arbeitsquantum, welches bei chemischen Verbindungen gewonnen werden kann; von Ludwig Boltzmann in Graz.

(Aus dem 88. Bde. der Sitzungsber. der k. Akad. der Wiss. zu Wien,
II. Abth. vom 18. Oct. 1883 mitgetheilt vom Hrn. Verf.)

Es seien z. B. A Chlor- und B Wasserstoffatome gegeben. Es wird gefragt, wie wahrscheinlich es ist, dass sich daraus gerade N_1 Chlor-, N_2 Wasserstoff- und N_3 Chlorwasserstoffmolecüle bilden. Hier ist $a_1 = 2$, $b_1 = 0$; $a_2 = 0$, $b_2 = 2$; $a_3 = 1$, $b_3 = 1$; die Anzahl der Chloratome ist $A = 2N_1 + N_3$; die Anzahl der Wasserstoffatome ist $B = 2N_2 + N_3$; die Anzahl der Bildungsweisen:

$$Z = \frac{A! B!}{2^{N_1 + N_2} \cdot N_1! N_2! N_3!}$$

Entropy of structure-forming systems

$$S = \log W \approx n \log n - \sum_{ij} \left(n_i^{(j)} \log n_i^{(j)} - n_i^{(j)} + n_i^{(j)} \log j! \right)$$

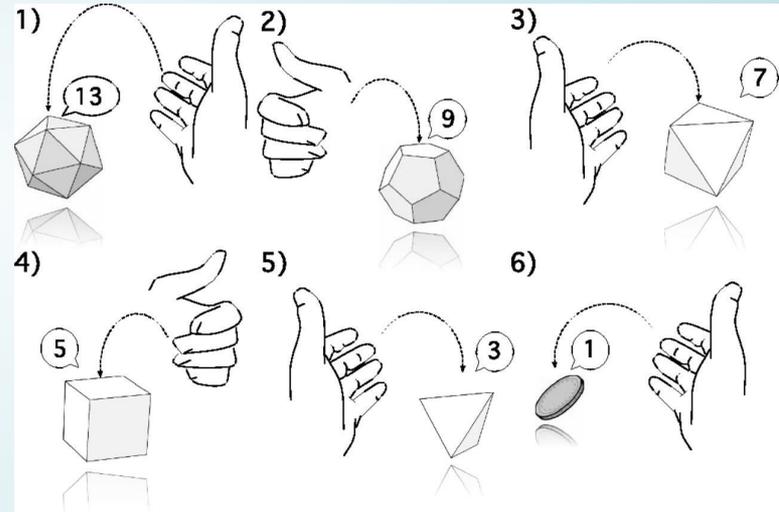
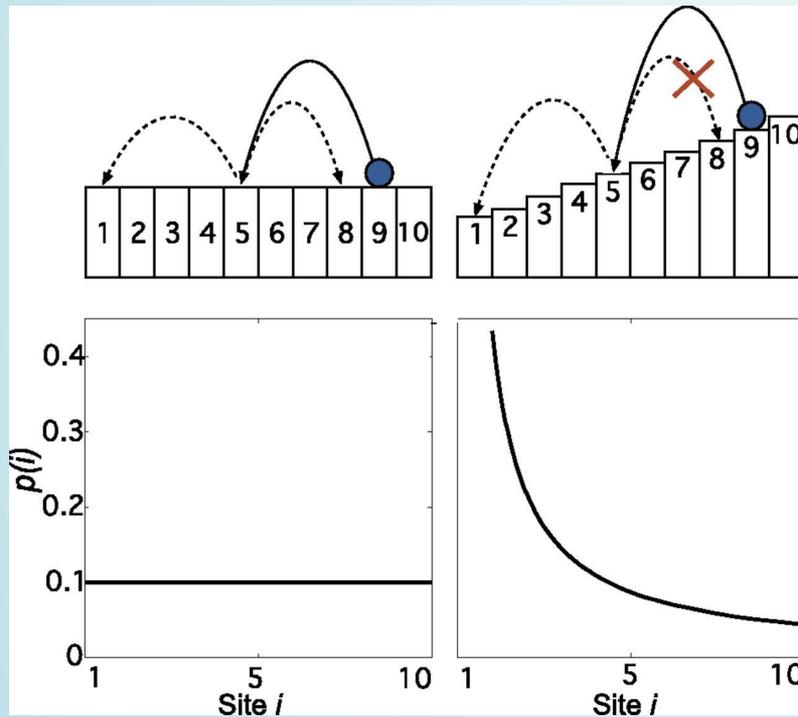
Introduce "probabilities" $\wp_i^{(j)} = n_i^{(j)} / n$

$$S = S/n = - \sum_{ij} \wp_i^{(j)} (\log \wp_i^{(j)} - 1) - \sum_{ij} \wp_i^{(j)} \log \frac{j!}{n^{j-1}}$$

Finite interaction range: concentration $c = n/b$

$$S = S/n = - \sum_{ij} \wp_i^{(j)} (\log \wp_i^{(j)} - 1) - \sum_{ij} \wp_i^{(j)} \log \frac{j!}{c^{j-1}}$$

Ex. III: sample-space reducing processes (SSR)



Multiplicity

The number of states is n . Let us denote the states as $x_n \rightarrow \dots \rightarrow x_1$, where x_1 is the ground state, where the process restarts. Let us sample R relaxation sequences $x = (x_{k_1}, \dots, x_1)$.

The sequences can be visualised as

$r \times i$	W	$W - 1$	$W - 2$	\dots	2	1
1	*	—	—	\dots	*	*
2	—	*	*	\dots	—	*
3	*	—	*	\dots	—	*
\vdots	\vdots	\vdots	\vdots		\vdots	\vdots
$R - 2$	—	*	*	\dots	—	*
$R - 1$	—	*	—	\dots	*	*
R	—	—	*	\dots	—	*
	k_W	k_{W-1}	k_{W-2}	\dots	k_2	k_1

Each run must contain x_1

How many of these sequences contain a state x_j exactly k_j times?

Multiplicity

Number of runs $R \equiv k_1$, number of them containing x_j is k_j

Multiplicity of these sequences: $\binom{k_1}{k_j}$

By multiplying the multiplicity for each state we get

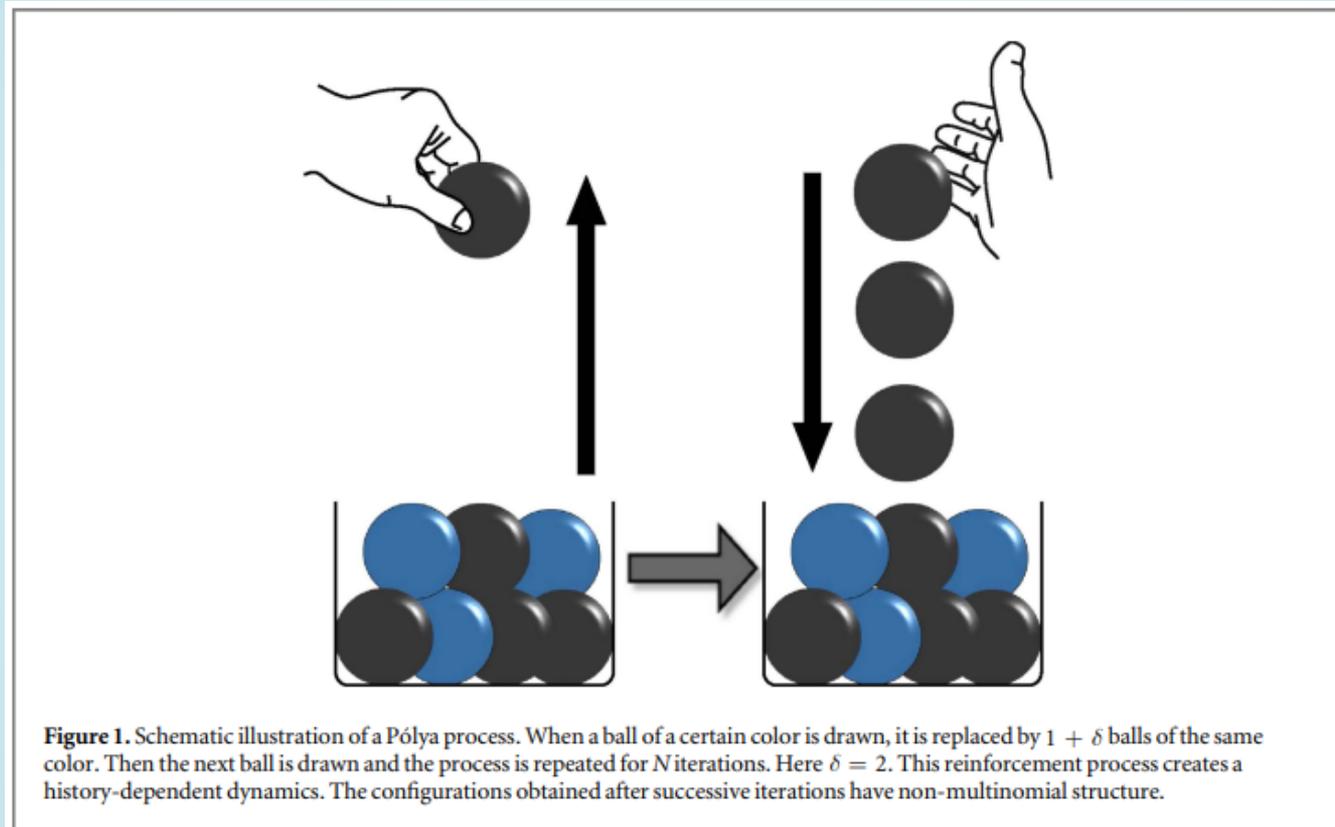
$$W(k_1, \dots, k_n) = \prod_{j=2}^n \binom{k_1}{k_j}$$

$$\begin{aligned} \log W &\approx \sum_{j=2}^n [k_1 \log k_1 - \cancel{k_1} - k_j \log k_j + \cancel{k_j} - (k_1 - k_j) \log(k_1 - k_j) + \cancel{(k_1 - k_j)}] \\ &\approx \sum_{j=2}^n \left[k_1 \log k_1 - k_j \log k_1 - k_j \log \frac{k_j}{k_1} - (k_1 - k_j) \log(k_1 - k_j) \right] \end{aligned}$$

By introducing $p_i = k_i/N$ where N is the total number of steps, we get

$$S_{SSR}(p) = -N \sum_{j=2}^n \left[p_j \log \left(\frac{p_j}{p_1} \right) + (p_1 - p_j) \log \left(1 - \frac{p_j}{p_1} \right) \right]$$

Ex. IV: Pólya urns



Probability of a sequence

We have c colors, initially $n_i(0) \equiv n_i$ balls of color c_i . After a ball is drawn, we return δ balls of the same color to the urn.

After N draws, the number of balls in the urn is

$$n_i(N) = n_i + \delta k_i$$

where k_i is the number of draws of color c_i . The total number of balls is $n(N) = \sum_c n_c(N) = N + \delta N$

The probability of drawing a ball of color c_i in N -th run, is $p_i(N) = n_i(N)/n(N)$. The probability of sequence

$\mathcal{I} = \{i_1, \dots, i_N\}$ is

$$p(\mathcal{I}) = \prod_{j=1}^c \frac{n_j^{(\delta, k_j)}}{n^{(\delta, N)}}$$

where $m^{(\delta, r)} = m(m + \delta) \dots (m + r\delta)$

Probability of a histogram

A histogram $\mathcal{K} = \{k_1, \dots, k_c\}$ is defined as $k_c = \sum_{j=1}^N \delta(i_j, c)$

Thus the probability of observing a histogram is

$$p(\mathcal{K}) = \binom{N}{k_1, \dots, k_c} p(\mathcal{I})$$

$$n_j^{(\delta, k_j)} \approx k_j! \delta^{k_j} (k_j + 1)^{n_j/\delta}$$

$$p(\mathcal{K}) = \frac{N!}{\prod_{j=1}^c k_j!} \frac{\prod_{j=1}^c k_j! \delta^{k_j} (k_j + 1)^{n_j/\delta}}{n^{(\delta, N)}}$$

... technical calculation ...

$$S_{\text{Polya}}(p) = \log p(\mathcal{K}) \approx - \sum_{i=1}^c \log(p_i + 1/N)$$

Ex. IV: q-deformations

This example is rather theoretical, but provides us a useful hint of what happens if there are correlations in the sample space

Motivation: finite versions of \exp and \log

$$\exp(x) = \lim_{n \rightarrow \infty} \left(1 + \frac{x}{n}\right)^n$$

So define

$$\exp_q(x) := (1 + (1 - q)x)^{1/(1-q)}$$

$$\log_q(x) := \frac{x^{1-q} - 1}{1 - q}$$

Let us find an operation s.t.

$$\begin{aligned} \exp_q(x) \otimes_q \exp_q(y) &\equiv \exp_q(x + y) \\ \Rightarrow a \otimes_q b &= [a^{1-q} + b^{1-q} - 1]^{1/(1-q)} \end{aligned}$$

Calculus of q-deformations

In analogy to $n! = 1 \cdot 2 \cdot \dots \cdot n$ introduce $n!_q := 1 \otimes_q 2 \otimes_q \dots \otimes_q n$

It is then easy to show that $\log_q n!_q = \frac{\sum_{k=1}^n k^{1-q} - n}{1-q}$ which can be used for generalized Stirling's approximation $\log_q n!_q \approx \frac{n}{2-q} \log_q n$

Let us now consider a q-deformed multinomial factor

$$\begin{aligned} \binom{n}{n_1, \dots, n_k}_q &:= n!_q \oslash_q (n_1!_q \otimes_q \dots \otimes_q n_k!_q) \\ &= \left[\sum_{l=1}^n l^{1-q} - \sum_{i_1}^{n_1} i_1^{1-q} - \dots - \sum_{i_k}^{n_k} i_k^{1-q} \right]^{1/(1-q)} \end{aligned}$$

Tsallis entropy

Let us consider that the multiplicity is given by a q-multinomial factor $W(n_1, \dots, n_k) = \binom{n}{n_1, \dots, n_k}_q$

In this case, it is more convenient to define entropy as $S = \log_q W$, which gives us:

$$\log_q W = \frac{n^{2-q}}{2-q} \frac{\sum_{i=1}^k p_i^{2-q} - 1}{q-1} = \frac{n^{2-q}}{2-q} S_{2-q}(p)$$

This entropy is known as **Tsallis entropy**

Note that the prefactor is not n but n^{2-q}

(*non-extensivity*) - we will discuss this later

Entropy and energy

Until now, we have been just counting states; let us now discuss the relation with **energy**.

We consider that the states describe the energy of the system (either Hamiltonian or more generalized energy functional)

Therefore, entropy is defined as

$$S(E) := \log W(E)$$

Ensembles

There are a few typical situations:

1. Isolated system = microcanonical ensemble

Let $H(s)$ be the energy of a state s . Multiplicity is then

$$W(E) = \sum_s \delta(H(s) - E)$$

Phenomena like negative "temperature" $T = \frac{dS(E)}{dE} < 0$

2. closed system = canonical ensemble

Total system is composed of the system of interest (S) and the heat reservoir/bath (B). They are weakly coupled i.e.,

$H_{tot}(s, b) = H_S(s) + H_B(b)$ (no interaction energy)

$$W(E_{tot}) = \sum_{s,b} \delta(H_{tot}(s, b) - E_{tot})$$

3. open system = grandcanonical ensemble

Entropy in canonical ensemble

This can be further rewritten as

$$\begin{aligned} &= \int dE_S \sum_s \delta(H_S(s) - E_S) \sum_b (H_B(b) - (E_{tot} - E_S)) \\ &= \int dE_S W_S(E_S) W_B(E_{tot} - E_S) \end{aligned}$$

This is hard to calculate. Typically, the dominant contribution is from the maximal configuration of the integrand, which we obtain from

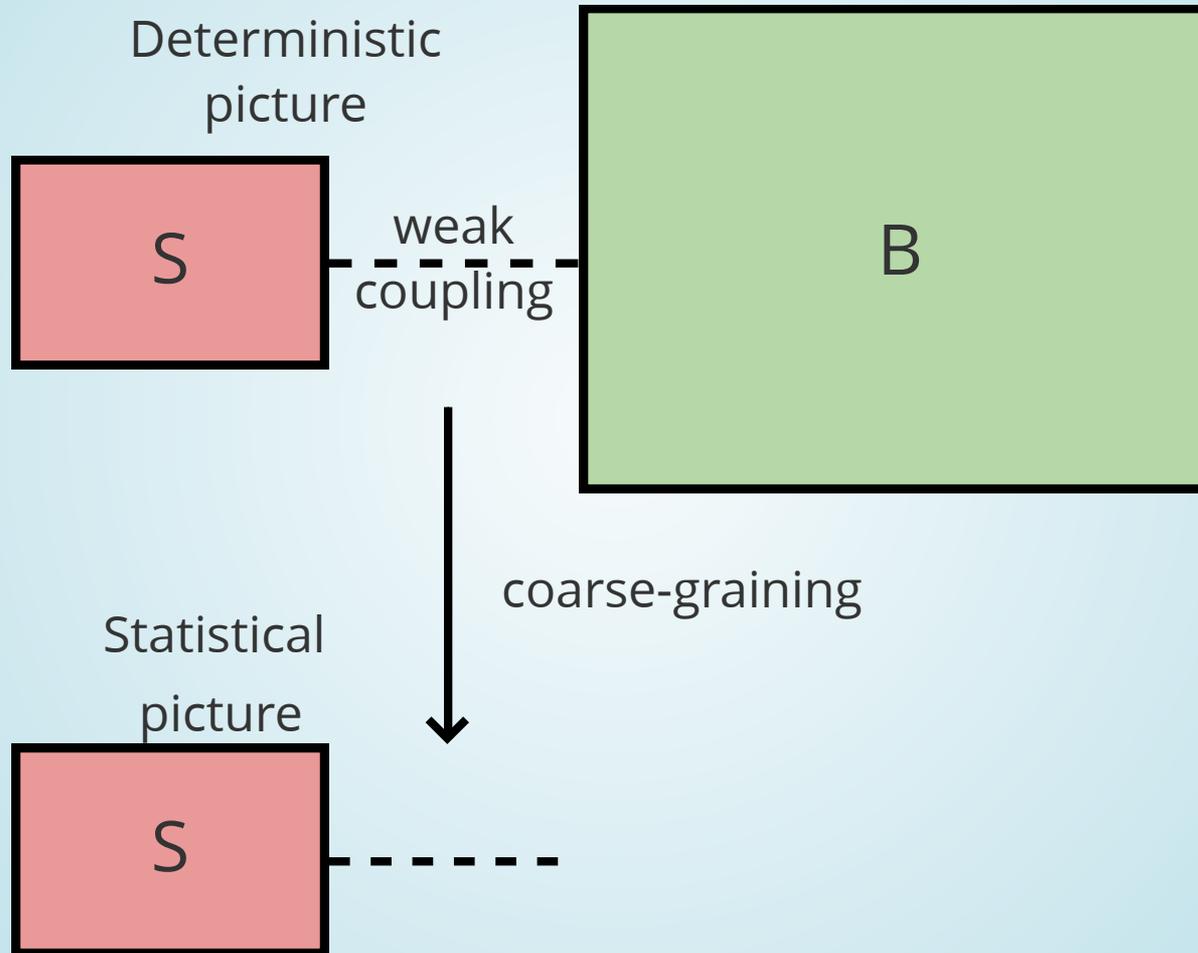
$$\frac{\partial W_S(E_S) W_B(E_{tot} - E_S)}{\partial E_S} \stackrel{!}{=} 0 \Rightarrow \frac{W'_S(E_S)}{W_S(E)} = \frac{W'_B(E_{tot} - E_S)}{W_B(E_{tot} - E_S)}$$

As a consequence $\frac{\partial S_E(E_S)}{\partial E_S} \stackrel{!}{=} \frac{\partial S_B(E_{tot} - E_S)}{\partial E_S} := \frac{1}{k_B T}$

and $\underbrace{S_B(E_{tot} - E_S)}_{\text{free entropy}} = \underbrace{S_B(E_{tot})}_{\text{bath entropy}} - \frac{\partial S}{\partial E_S} E_S + \dots$

This is the emergence of **Maximum entropy principle**

Canonical ensemble & Coarse-graining



Summary