

SMR 1302 - 16

WINTER SCHOOL ON LASER SPECTROSCOPY AND APPLICATIONS

19 February - 2 March 2001

***Elementary Introduction
to
Subdoppler Laser Spectroscopy
Part I and II***

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Largo Enrico Fermi, 2 - Firenze, Italy

These are preliminary lecture notes, intended only for distribution to participants.

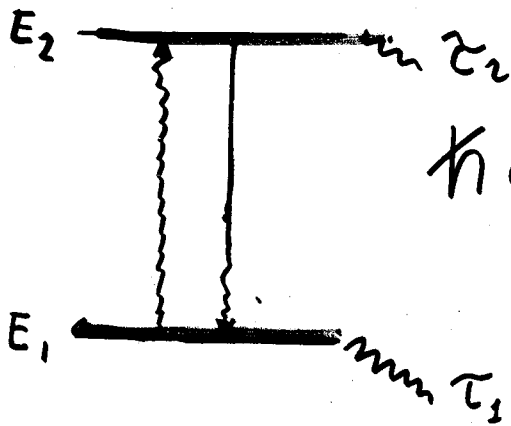
Elementary introduction
to

Subdoppler Laser Spectroscopy

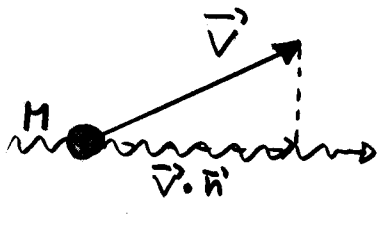
M. Inguscio # 1, 2

ALLARGAMENTO DOPPLER DI RIGHE SPETTRALI (1° ordine)

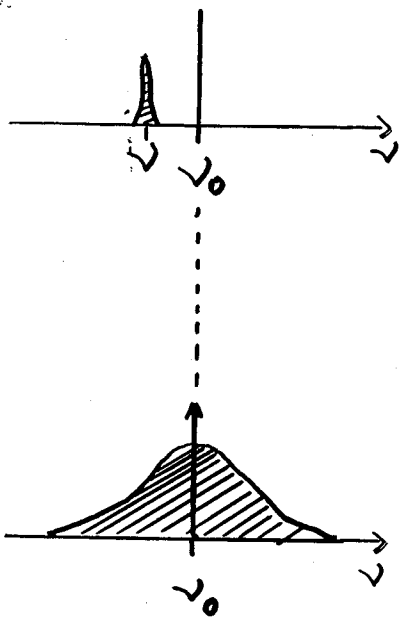
DOPPLER BROAD. (1st ORDER)



$$\hbar \omega_0 = E_2 - E_1$$



$$\nu = \nu_0 \left(1 + \frac{\vec{V} \cdot \vec{n}}{c} \right)$$



NOTE! $\frac{\Delta \nu_D}{\nu}$ independent

from \vec{V}

At room temp

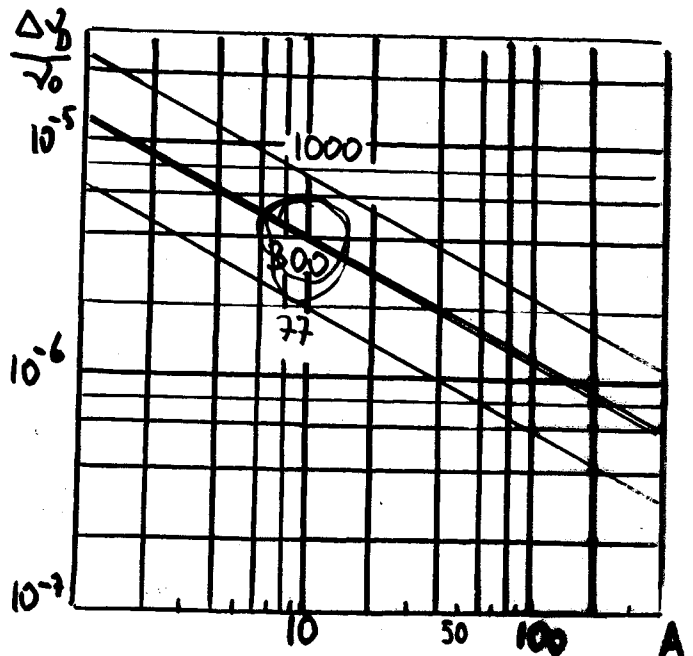
$$\frac{\Delta \nu}{\nu} \sim 2 \cdot 10^{-6}$$

$$I = I_0 \exp \left[-\frac{M c^2}{2 k T} \left(\frac{\nu - \nu_0}{\nu_0} \right)^2 \right]$$

$$\Delta \nu_D = 7.163 \times 10^{-7} \omega_0 \sqrt{\frac{T}{A}}$$

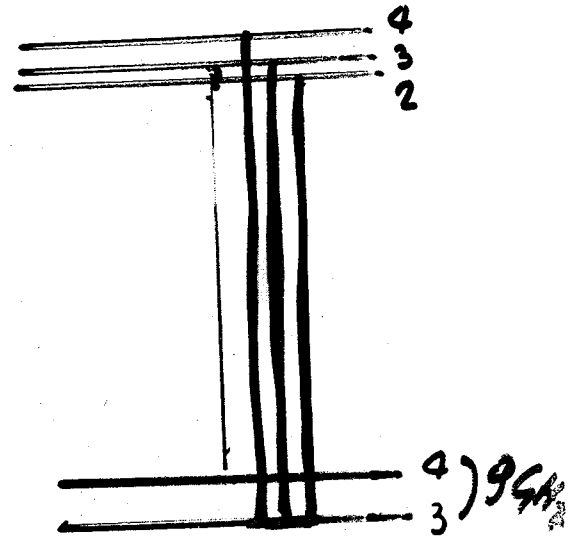
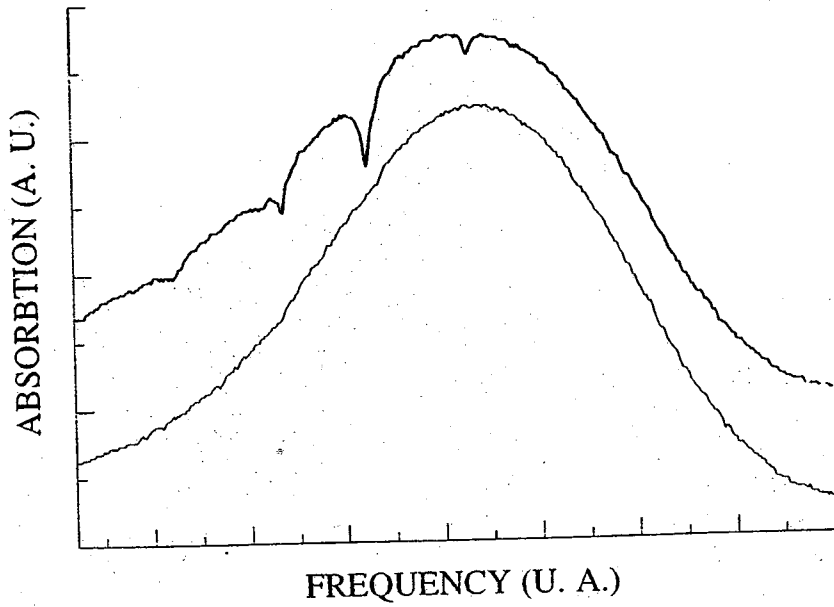
$$10^8 - 10^{10} \text{ Hz}$$

$$\Delta \nu_D \approx 10^2 \Delta \nu_{\text{NAT}} \text{ in the VIS.}$$



SATURATION SPECTROSCOPY

Cs



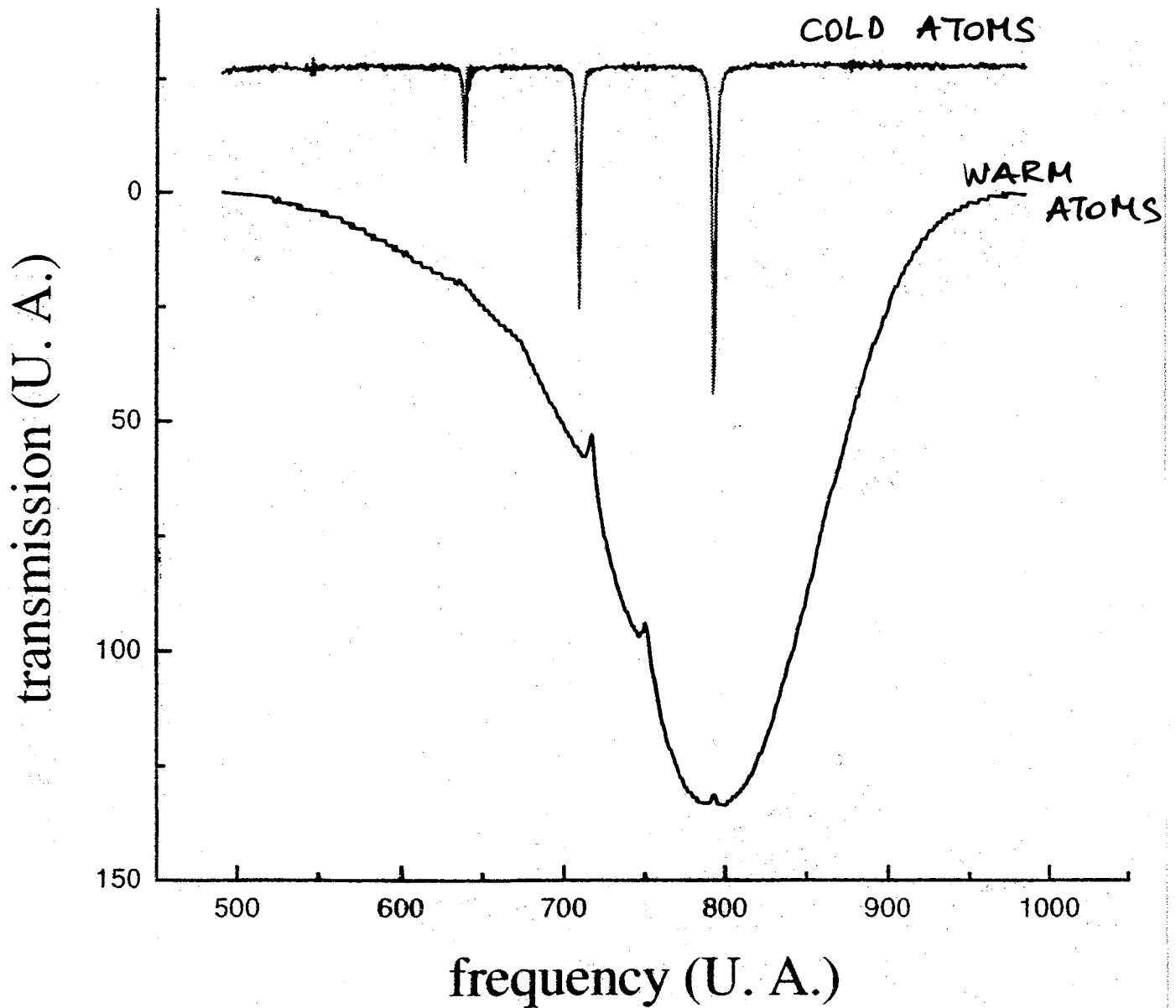
Different transitions masked by Doppler width.

... as if above, at least

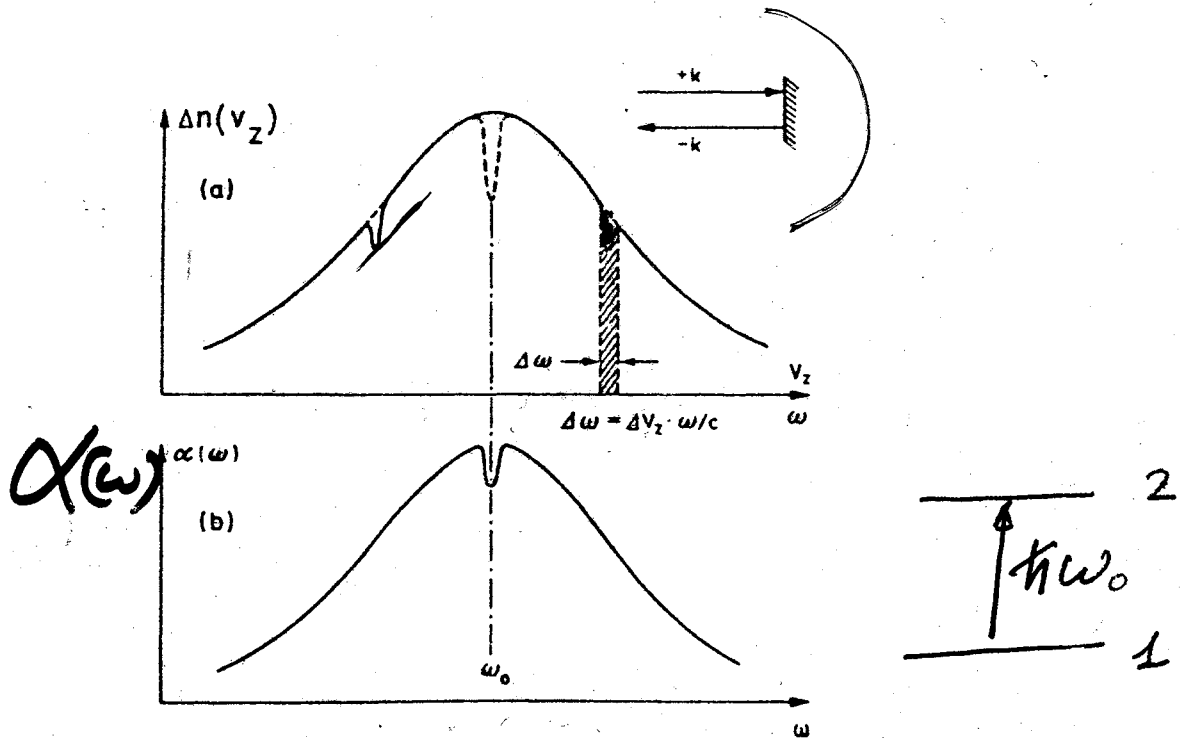
in one direction, were

not moving

NOW THEY ARE AT
" REST ,,



SUBDOPPLER SPECTROSCOPY



EFFETTO DOPPLER:

ATOMO CON VELOCITA' \vec{v} SI MUOVE IN UN CAMPO LASER DI FREQUENZA ω_L E VETTORE D'ONDA \vec{k}

NEL SISTEMA DI RIFERIMENTO DELL'ATOMO

ω_L E' VISTA COME $\sim \omega' = \omega_L - \vec{k} \cdot \vec{v}$

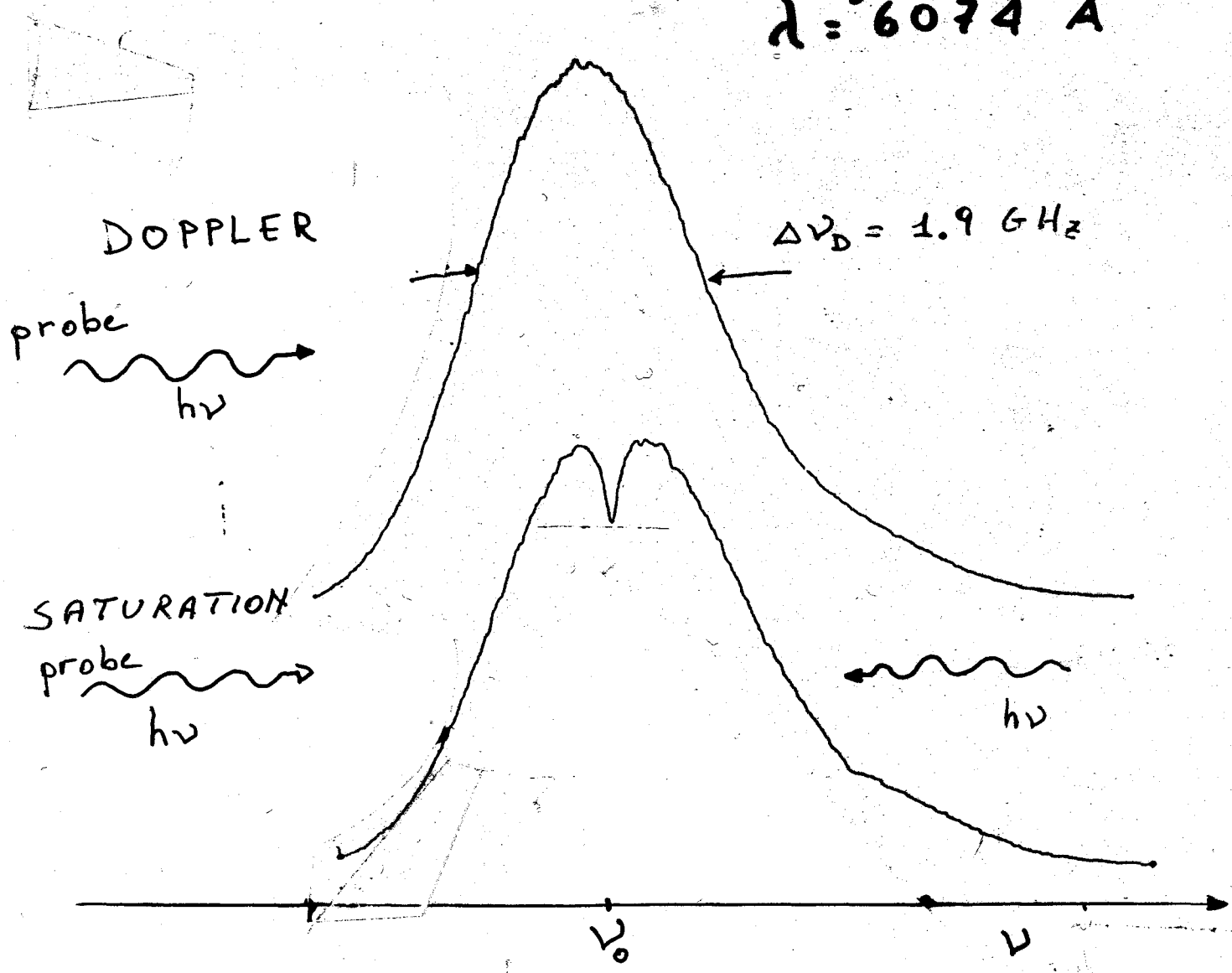
CONDIZIONI DI RISONANZA:

$$\omega_L = \vec{k} \cdot \vec{v} + \omega_0$$

OPTOGALVANIC SATURATION SPECTROSCOPY

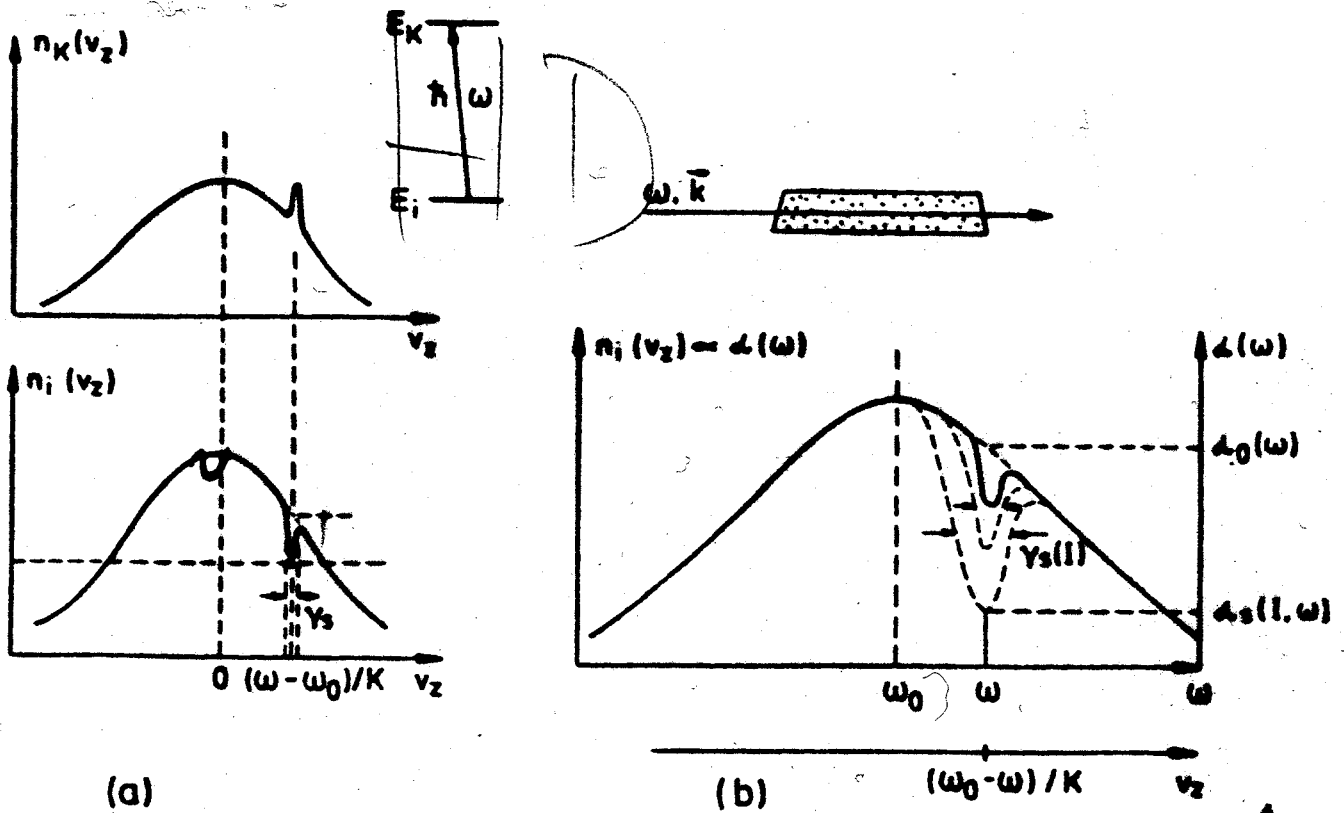
an example : Neon transition

FROM METASTABLE $1s_4 \rightarrow 2p_3$
 $\lambda = 6074 \text{ \AA}$

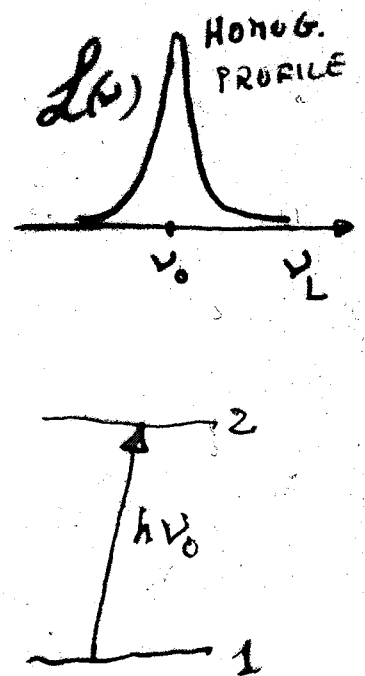
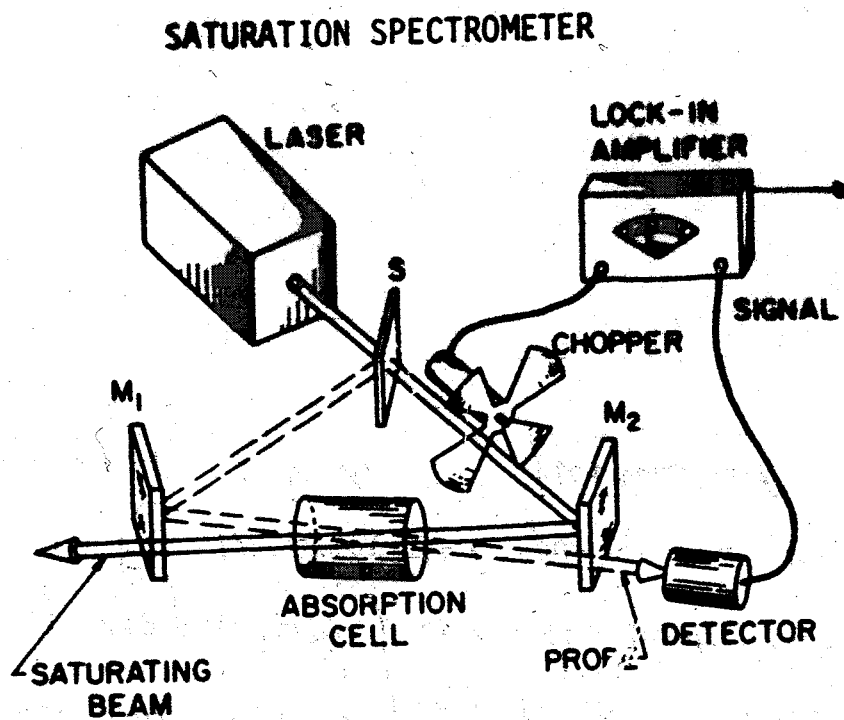


HOW REMOVE THE DOPPLER
BACKGROUND ?

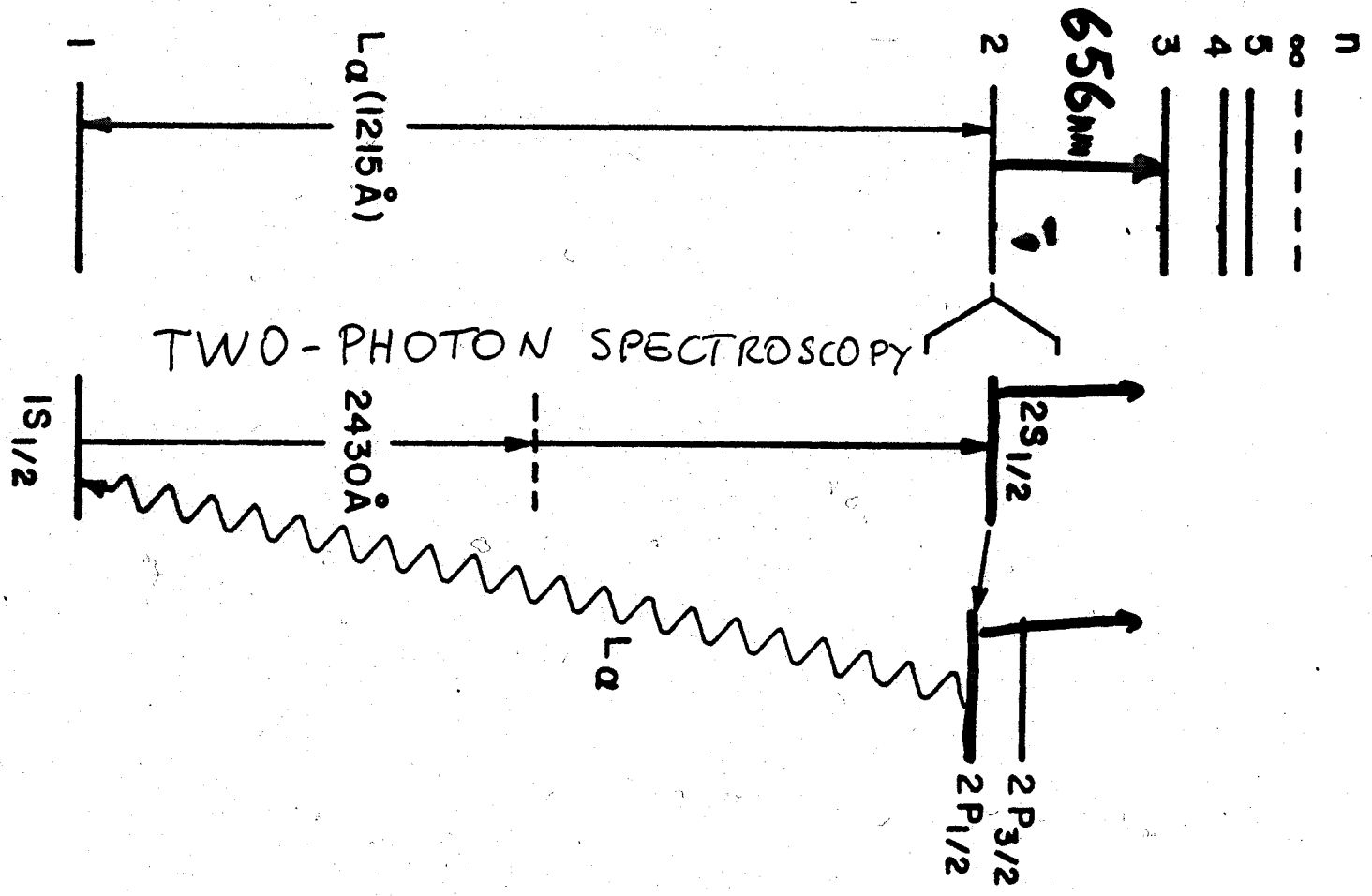
SATURATION SPECTROSCOPY



Hänsch SCHEME



SATURATION SPECTROSCOPY



-8-

2.4.

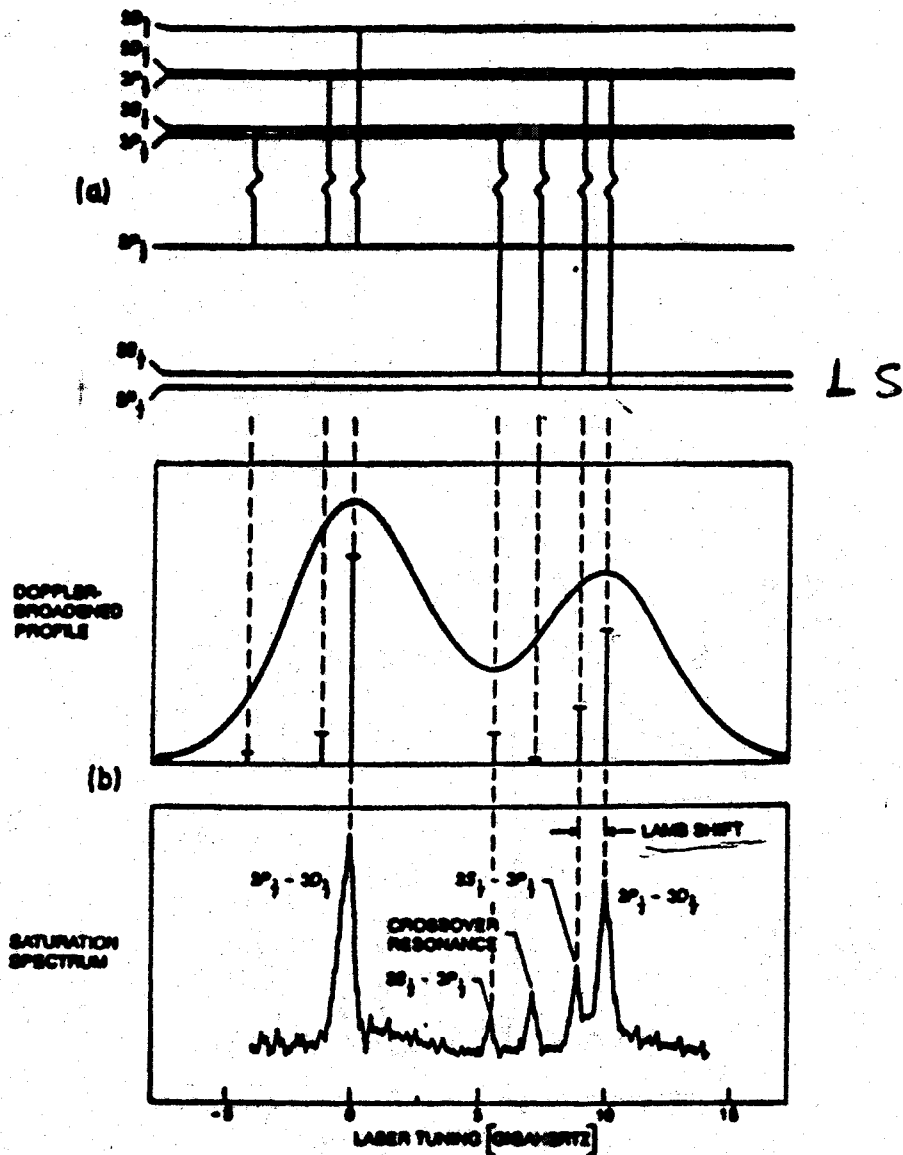
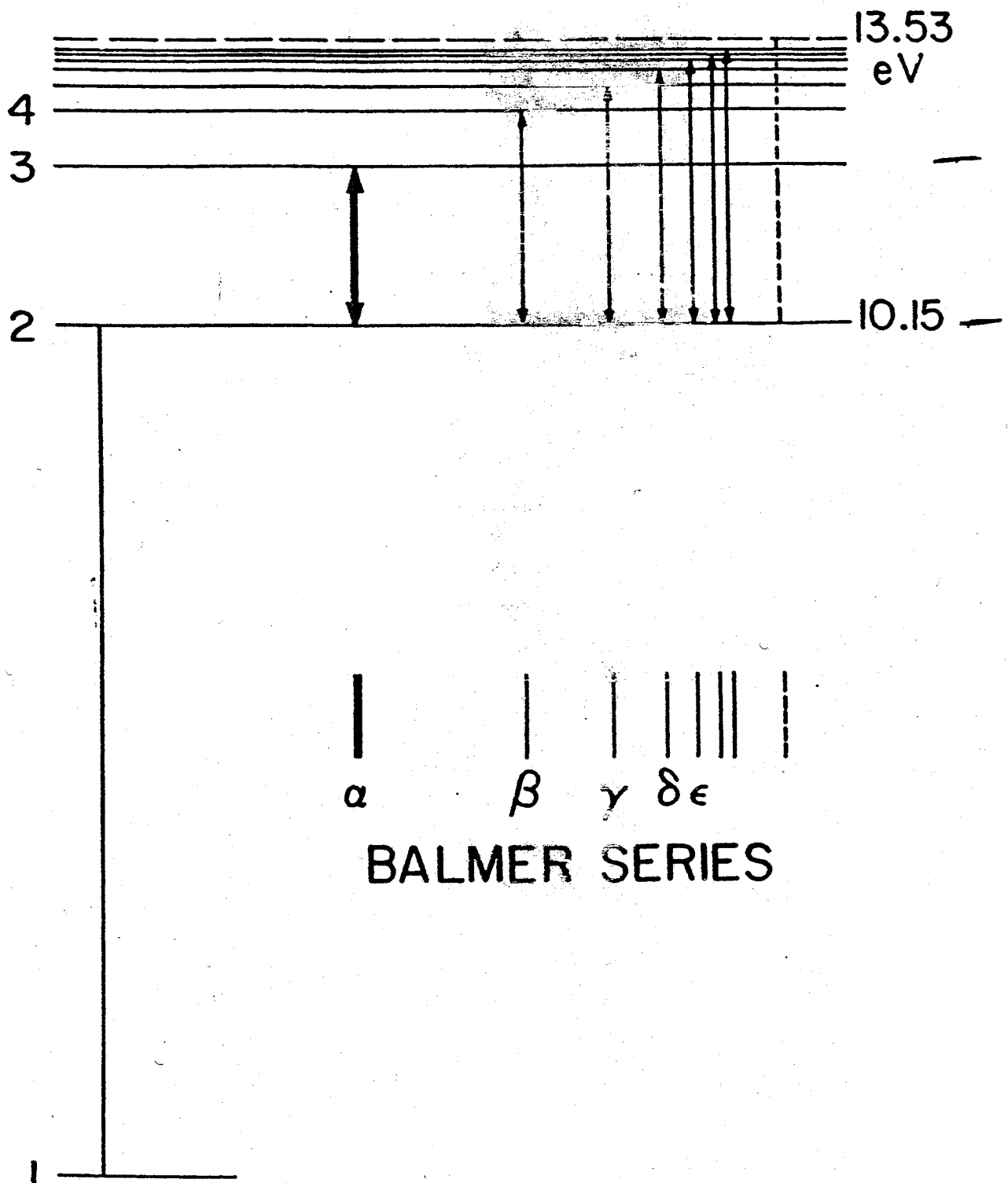


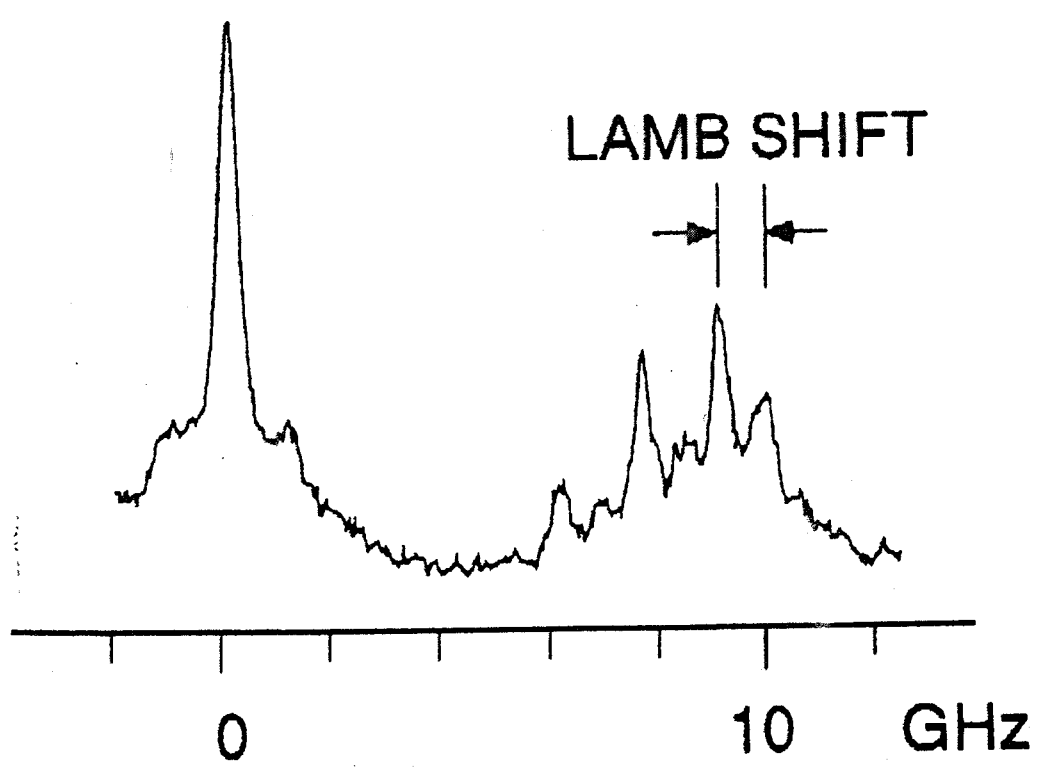
Fig. 10.21a,b. Measurement of the Rydberg constant by saturation spectroscopy of the Hydrogen Balmer α transition. (a) Level scheme, (b) Doppler profiles and saturation spectrum of the Balmer α line in a hydrogen discharge [10.31a]

SUBDOPPLER RESOLUTION OF
HYDROGEN BALMER α , WITH
OBSERVATION OF FINE STRUCTURE
AND LAMB SHIFT.

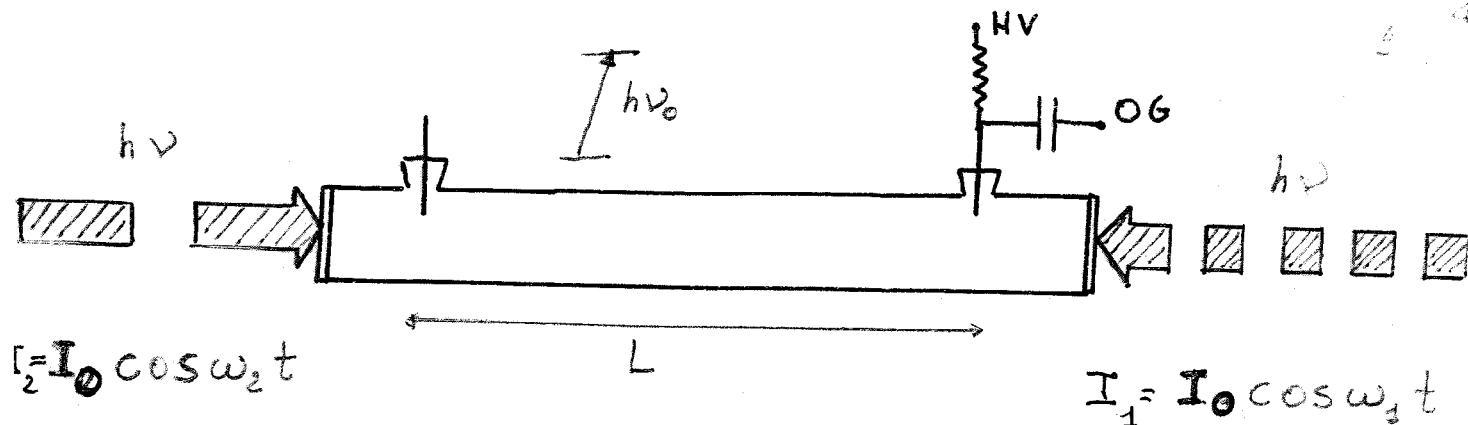
HYDROGEN TERMS



EVEN WITH A SMALL DIODE LASER.



INTERMODULATED OPTOGALVANIC SPECTR.



THE OG SIGNAL INDUCED BY TWO COUNTERPROP. BEAMS, MODULATED AT FREQUENCIES

$$\omega_1 = 2\pi f_1 \quad ; \quad \omega_2 = 2\pi f_2$$

is given by

$$I = I_1 + I_2$$

$$S_{OG} = K I \left[1 - e^{-\alpha(\omega)L} \right] \quad k = \text{const.}$$

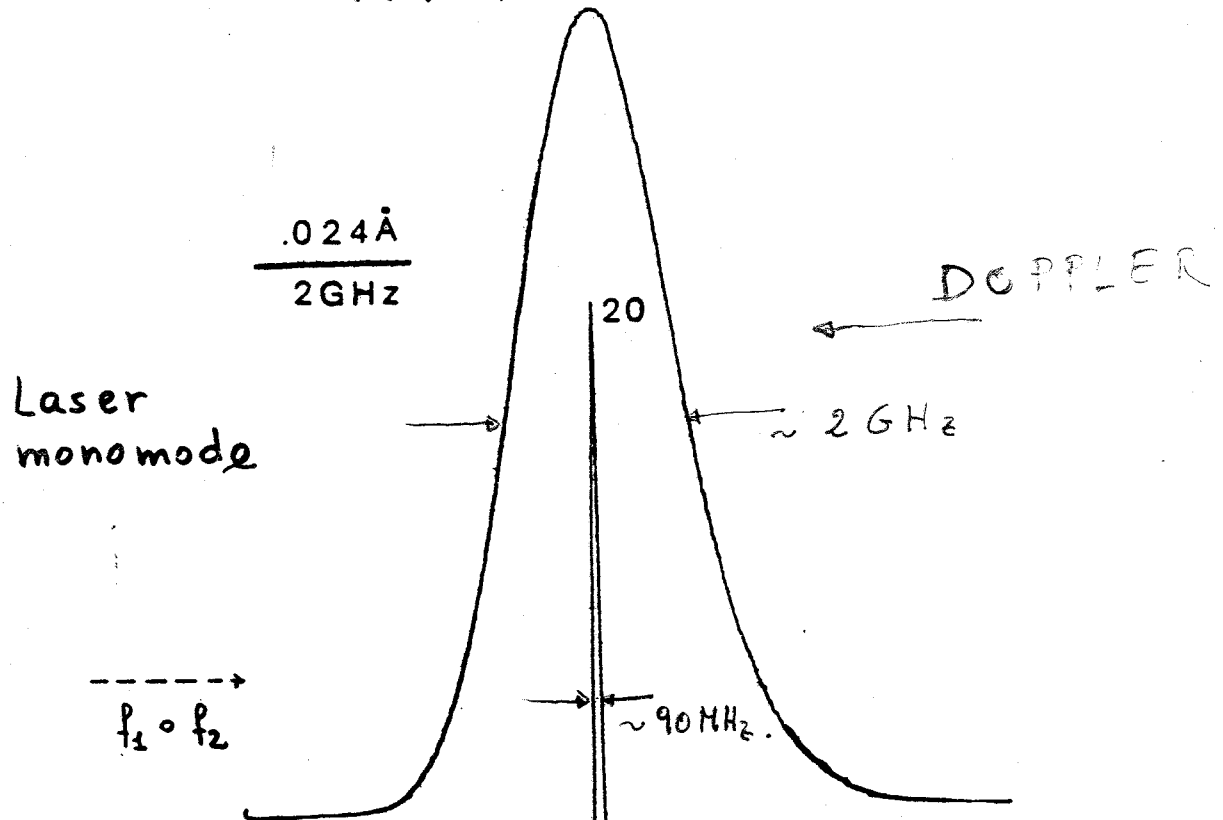
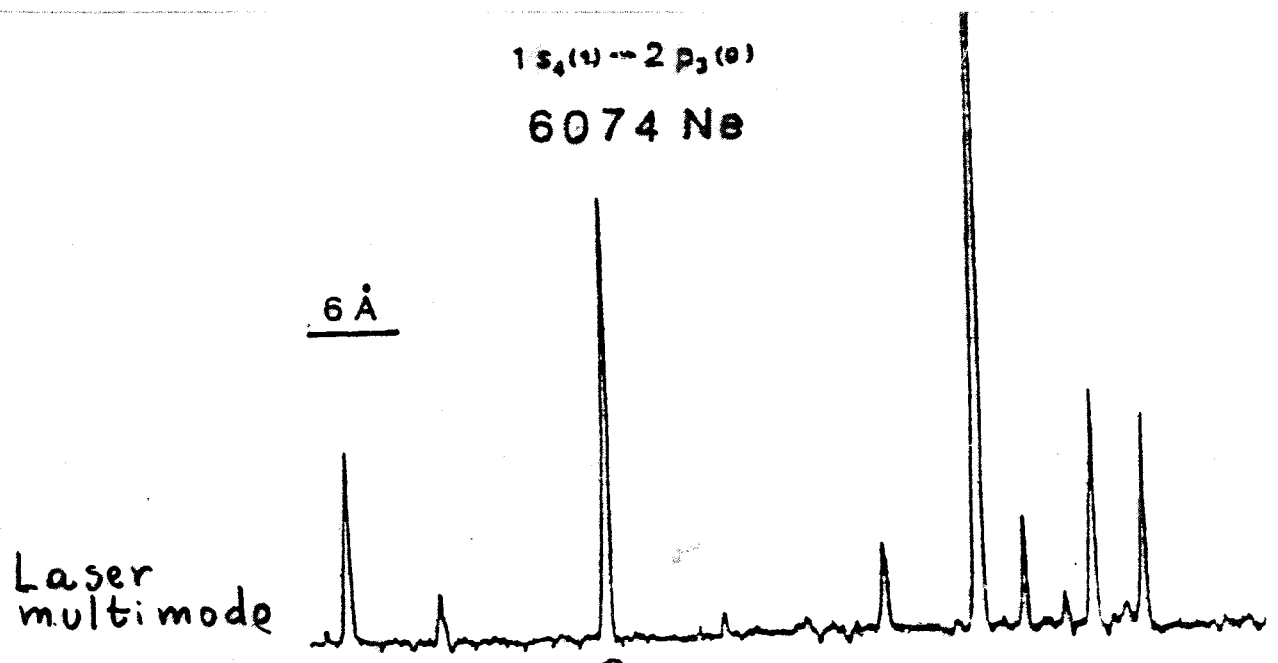
where

$$\alpha(\omega) = \alpha_0(\omega) \left[1 - \frac{S_0}{2} \frac{(\gamma_s/2)^2}{(\omega - \omega_0)^2 + (\gamma_s/2)^2} \right]$$

If $\alpha L \ll 1$ ("optically" thin sample)

$$S_{OG} = KL \alpha_0(\omega) \left[I - \frac{I^2}{2I_{SAT}} \frac{(\gamma_s/2)^2}{(\omega - \omega_0)^2 + (\gamma_s/2)^2} \right]$$

$1s_4(1) \rightarrow 2p_3(0)$
6074 Ne



$f_1 = f_2$

$\sim 90 \text{ MHz}$

INTERM.

^{20}Ne 90.92%

^{22}Ne 8.823%

$\rightarrow f_1$
 $\leftarrow f_2$
 $f_1 + f_2$

22

$\Delta\nu_H = 0.09 \text{ GHz}$
 $(\Delta\nu_{\text{rad.}} = 21 \text{ MHz})$

1.7 GHz (I.S.)

Being :

$$I = I_1 + I_2 = I_0 \sin \omega_1 t + I_0 \sin \omega_2 t$$
$$= I_0 (\sin \omega_1 t + \sin \omega_2 t)$$

$$S_{OG} = KL d_0(\omega) \Big|_{\omega_1} + \leftarrow \text{DOPPLER}$$

$$KL d_0(\omega) \Big|_{\omega_2} + \leftarrow \text{DOPPLER}$$

$$KL d_0(\omega) \Big|_{\substack{(\omega_1 + \omega_2) \\ \text{or } (\omega_1 - \omega_2)}} \times \frac{(\gamma_s/2)^2}{(\omega - \omega_0)^2 + (\gamma_s/2)^2} \leftarrow \text{SUB-DOPPLER}$$

FINALLY

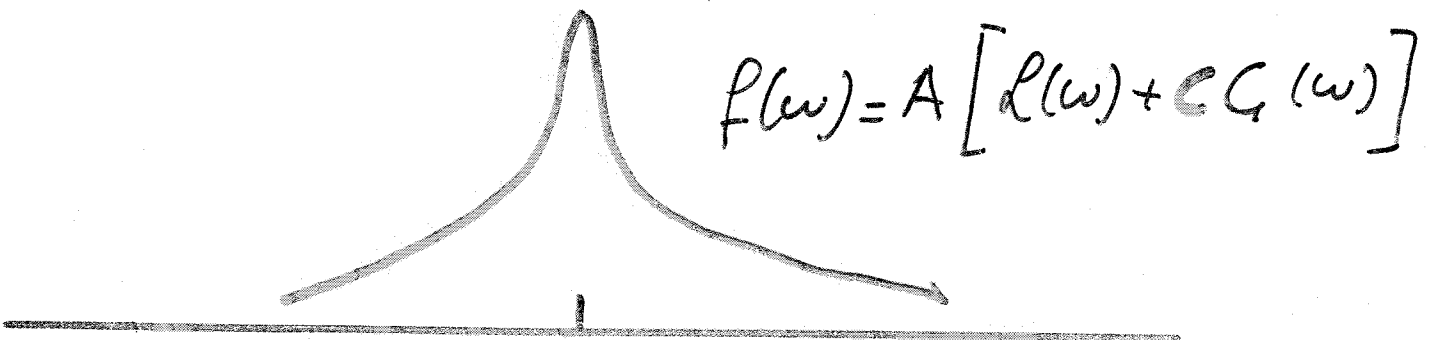
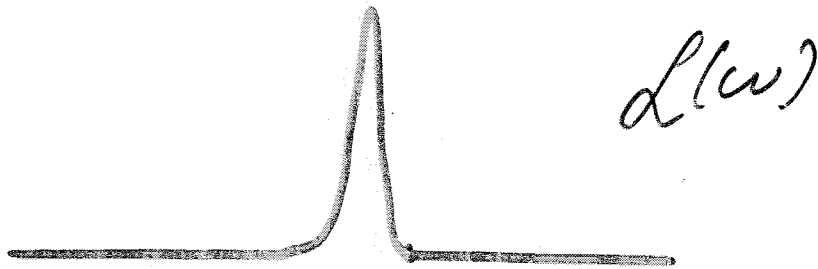
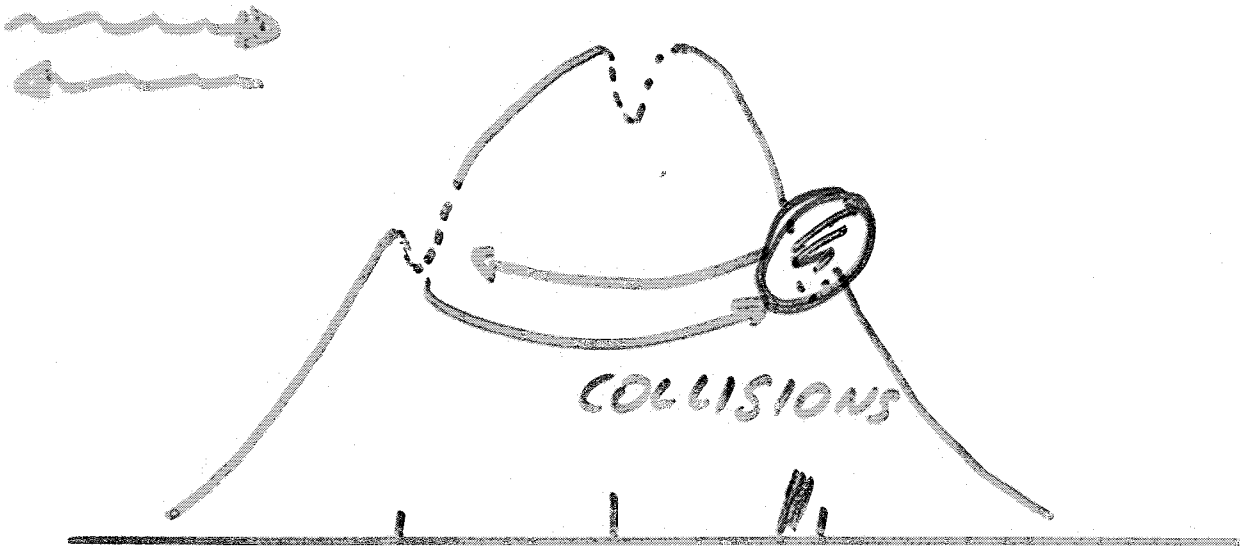
$$S_{OG}^{(\omega_1 + \omega_2)} = K \cdot \ell \left[\frac{(\omega - \omega_0)}{\Delta \omega_D} \right]^2 \cdot \frac{(\gamma_s/2)^2}{(\omega - \omega_0)^2 + (\gamma_s/2)^2}$$

If $\gamma_s \ll \Delta \omega_D$

$$S_{OG}^{(\omega_1 + \omega_2)} = K \frac{(\gamma_s/2)^2}{(\omega - \omega_0)^2 + (\gamma_s/2)^2} \quad \text{HOMOGENEOUS LINESHAPE}$$

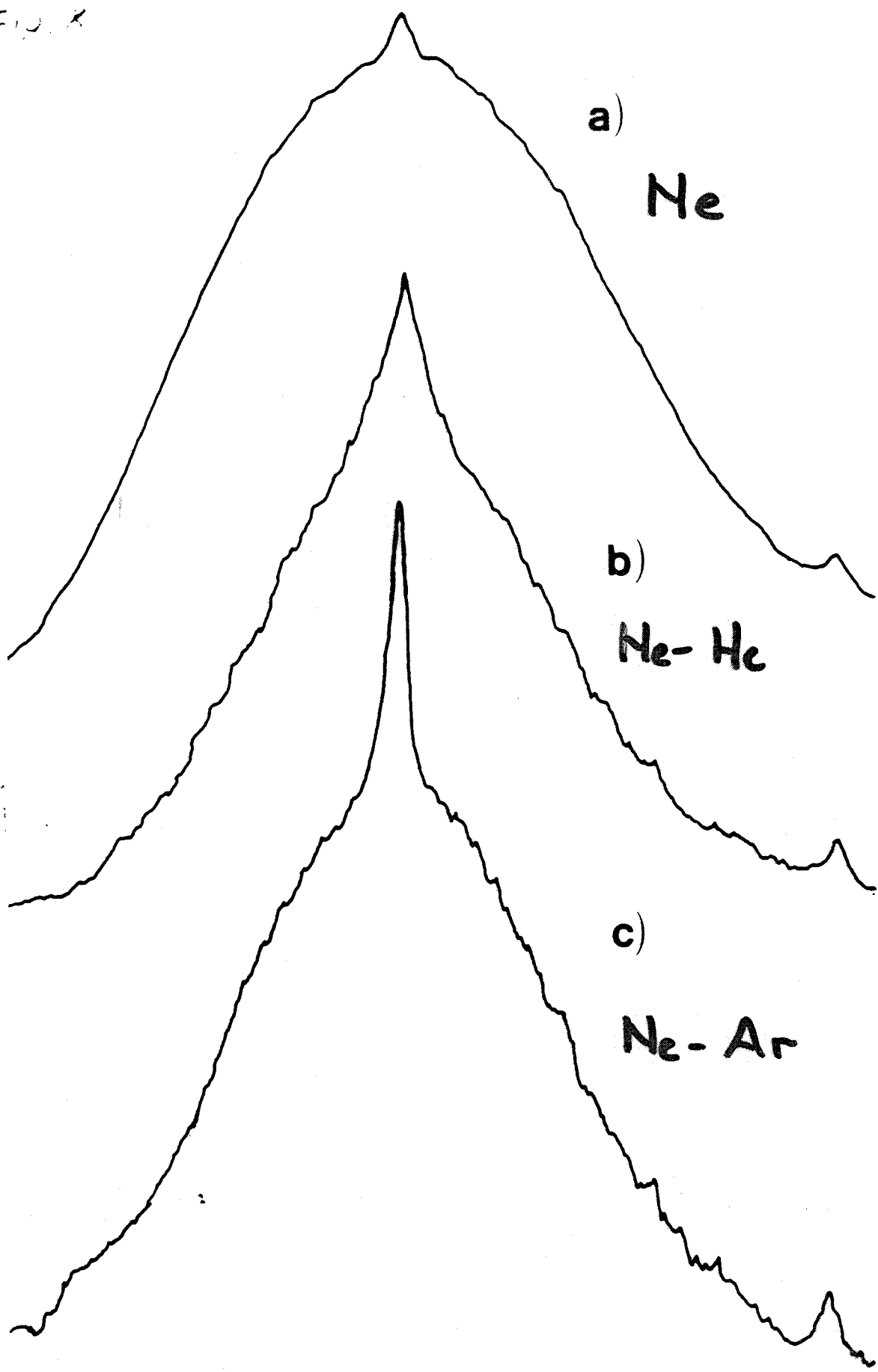
VELOCITY CHANGING COLLISIONS

16



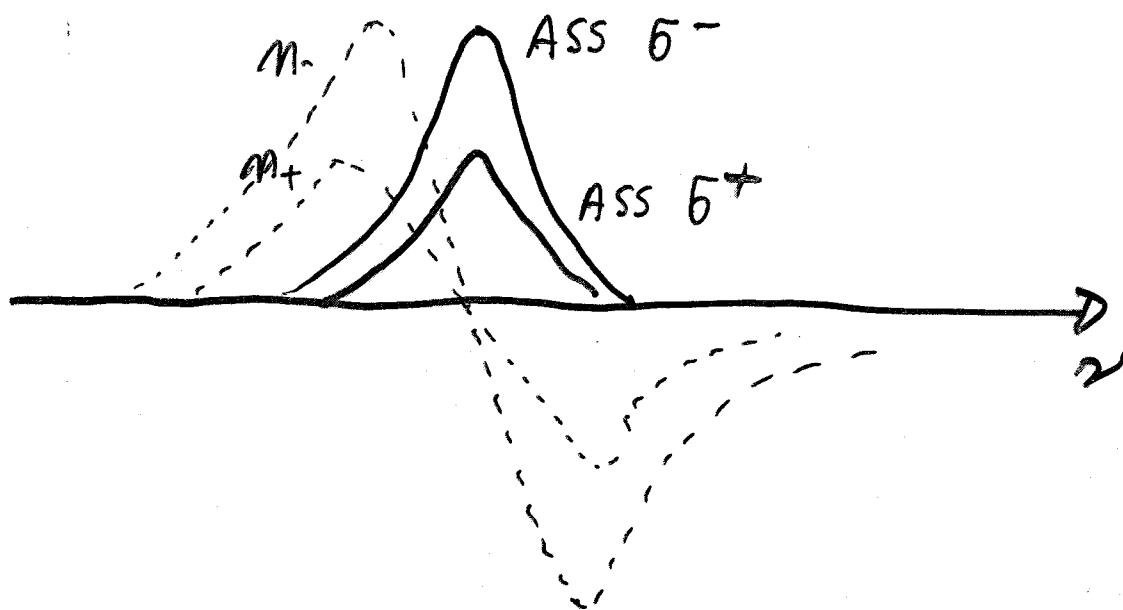
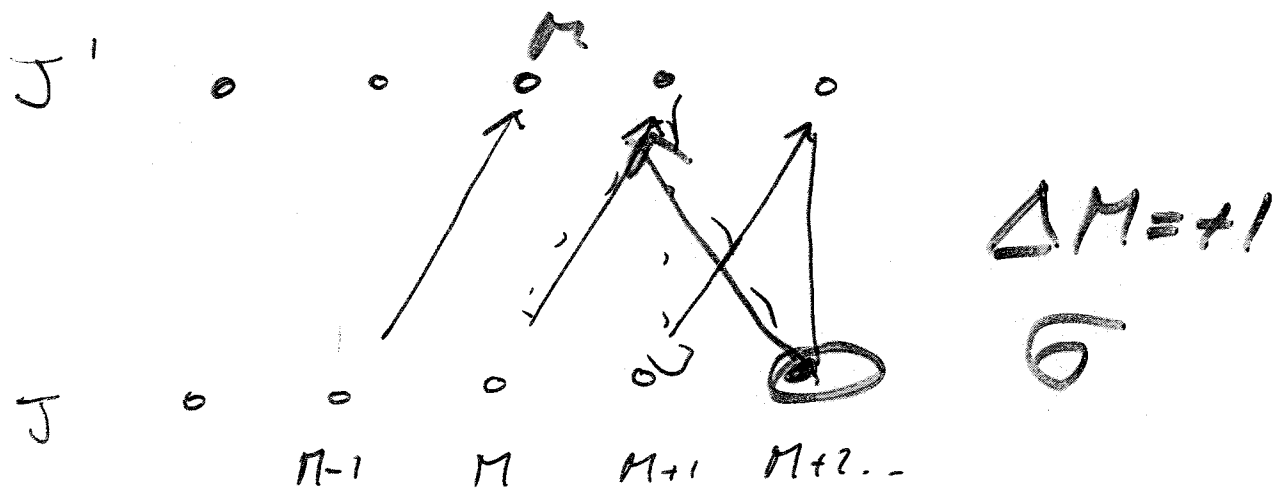
$$c \sim 2\sqrt{\pi} \lg 2 \frac{\tau_1 \gamma}{\tau_2 \Delta \nu_D}$$

FIG. 8

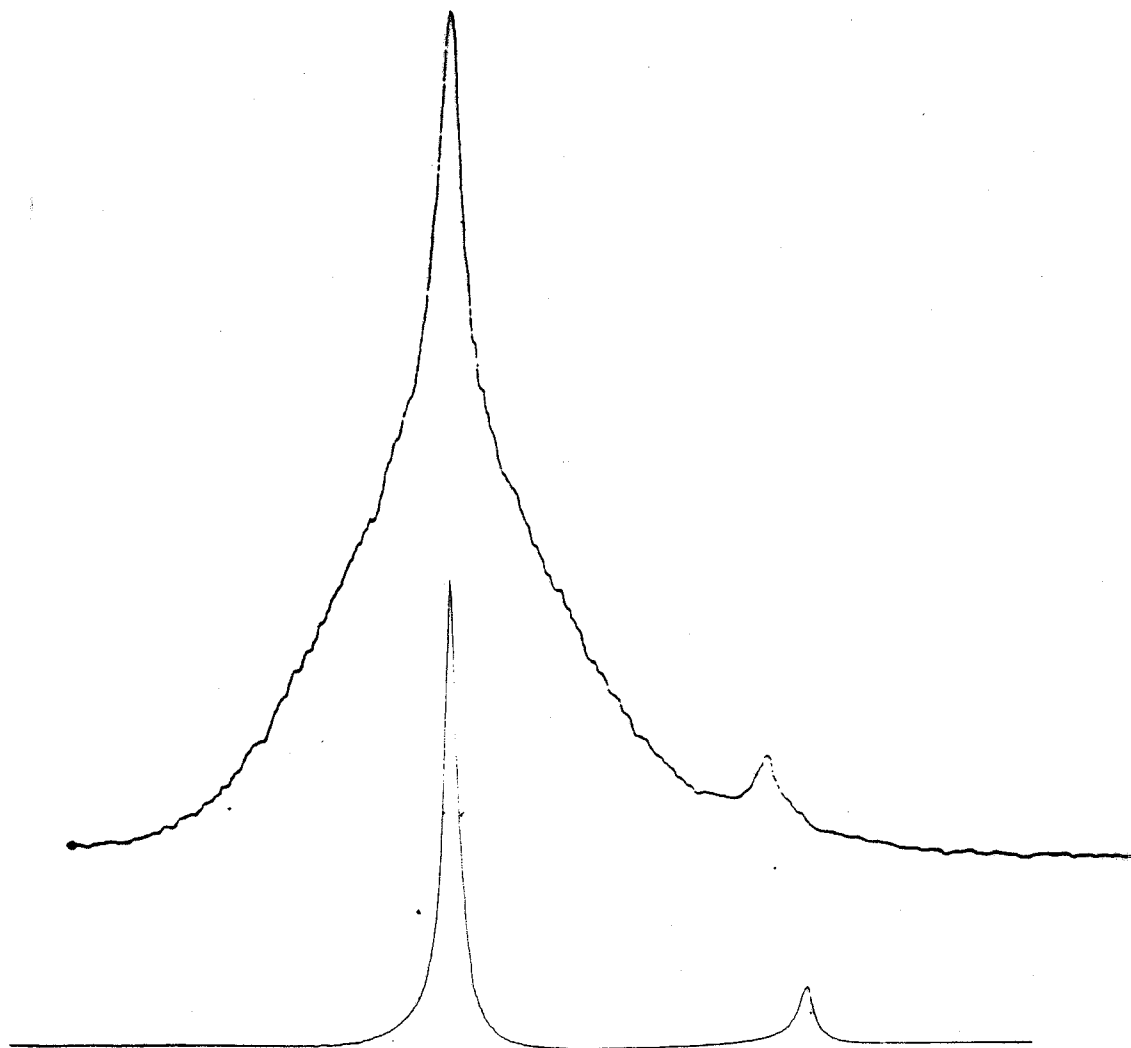
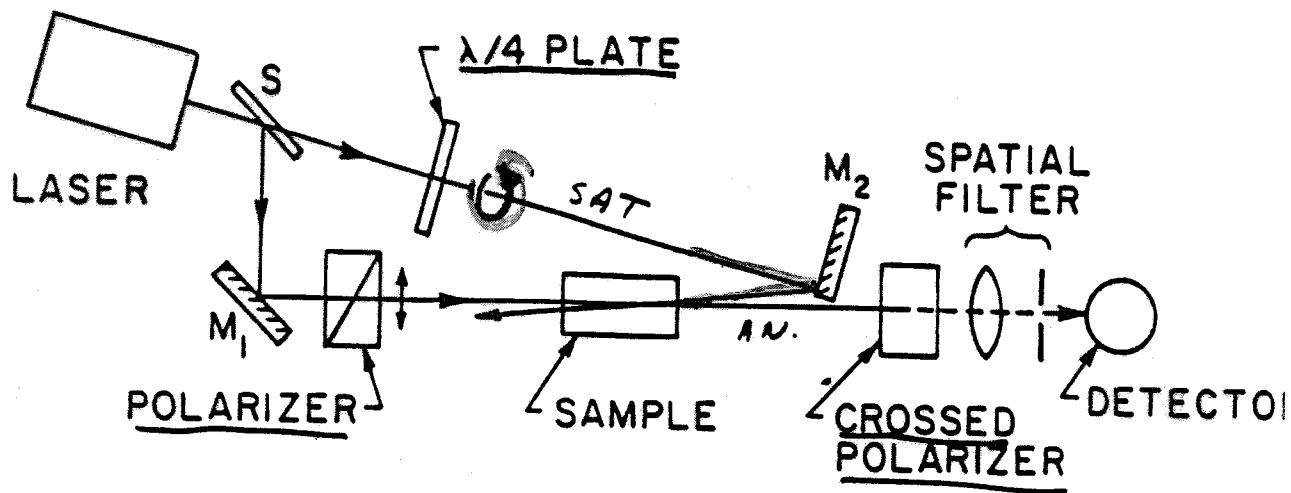


ν laser \rightarrow

CON LUCE POL. CIRCOLARE
 PROVOCO UN'ORIENTAZIONE
 (SAT. SELETTIVA IN M)



CIRCULARLY POLARIZED LIGHT
 CREATES AN ORIENTATION
 (SATURATION SELECTIVE IN M)

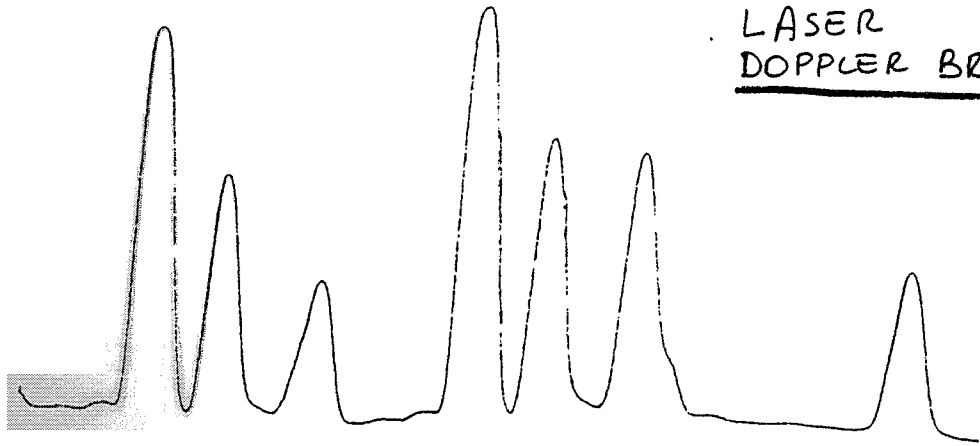
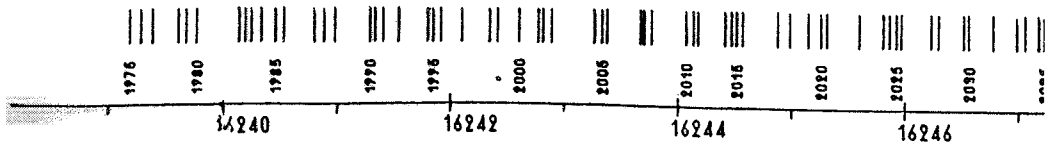


I₂ SPECTRUM (GOOD FOR λ CALIBRATION)

9



FOURIER
TRANSFORM



LASER
DOPPLER BROAD.

1939	16232.7893	2.1	24	1971	16238.4971	1.1	32	2003	16242.8907	.7	51	2035	16247.2425	.7	42
1940	16232.9659	.6	32	1972	16238.5940	.6	31	2004	16243.2738	.6	23	2036	16247.4690	1.7	17
1941	16233.0872	.7	33	1973	16238.7526	.5	38	2005	16243.3428	.9	17	2037	16247.6540	.5	44
1942	16233.1332	.4	41	1974	16239.1854	.6	26	2006	16243.3862	.5	38	2038	16247.8451	6.6	17
1943	16233.2791	2.4	15	1975	16239.1964	.6	25	2007	16243.6739	.7	39	2039	16247.8973	.7	38
1944	16233.6513	1.8	28	1976	16239.2974	.5	47	2008	16243.7898	.8	38	2040	16248.2979	.6	43
1945	16233.7256	2.3	45	1977	16239.4006	1.8	16	2009	16243.7774	1.0	21	2041	16248.4641	.6	27
1946	16233.7472	.9	48	1978	16239.6225	.4	53	2010	16244.0828	1.0	24	2042	16248.5285	.5	46
1947	16233.8414	.7	25	1979	16239.6932	.7	31	2011	16244.1465	1.3	46	2043	16248.6747	.8	21
1948	16234.1210	.5	48	1980	16239.7828	.7	27	2012	16244.1902	.7	18	2044	16248.9210	.6	37
1949	16234.4843	.8	31	1981	16240.1529	.5	49	2013	16244.4252	.6	42	2045	16249.1452	.5	33
1950	16234.5782	.6	30	1982	16240.2074	.8	21	2014	16244.4804	1.1	18	2046	16249.3674	.9	26

$\pm .002 \text{ cm}^{-1}$
 $\pm 6.0 \text{ MHz}$

Fig. 15 Registrazione in fluorescenza stimolata di una regione dello spettro della molecola di I₂ e confronto con le righe tabulate.

DOPPLER + HYPERFINE

10
36

15 comp J even ~ EQUAL INTENSITY
21 comp J odd FOR $J > 30$

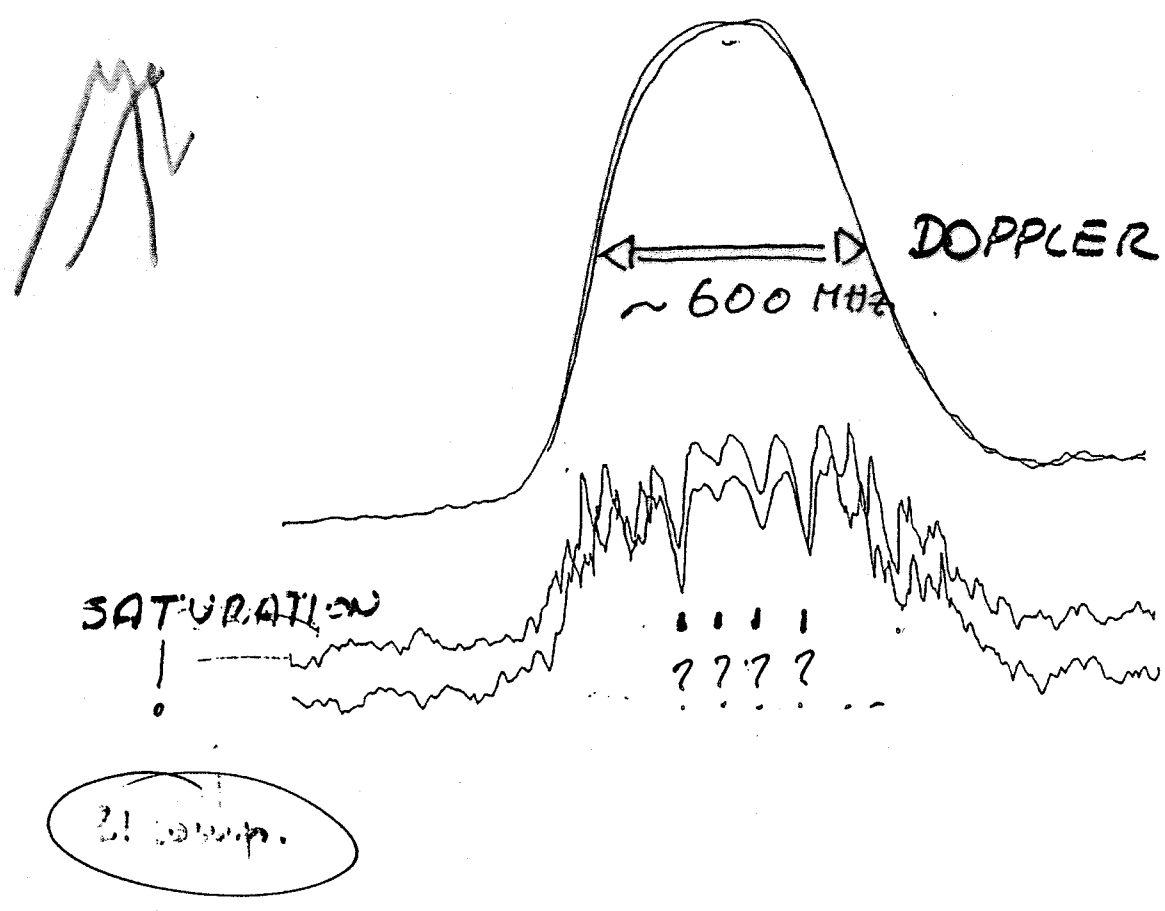
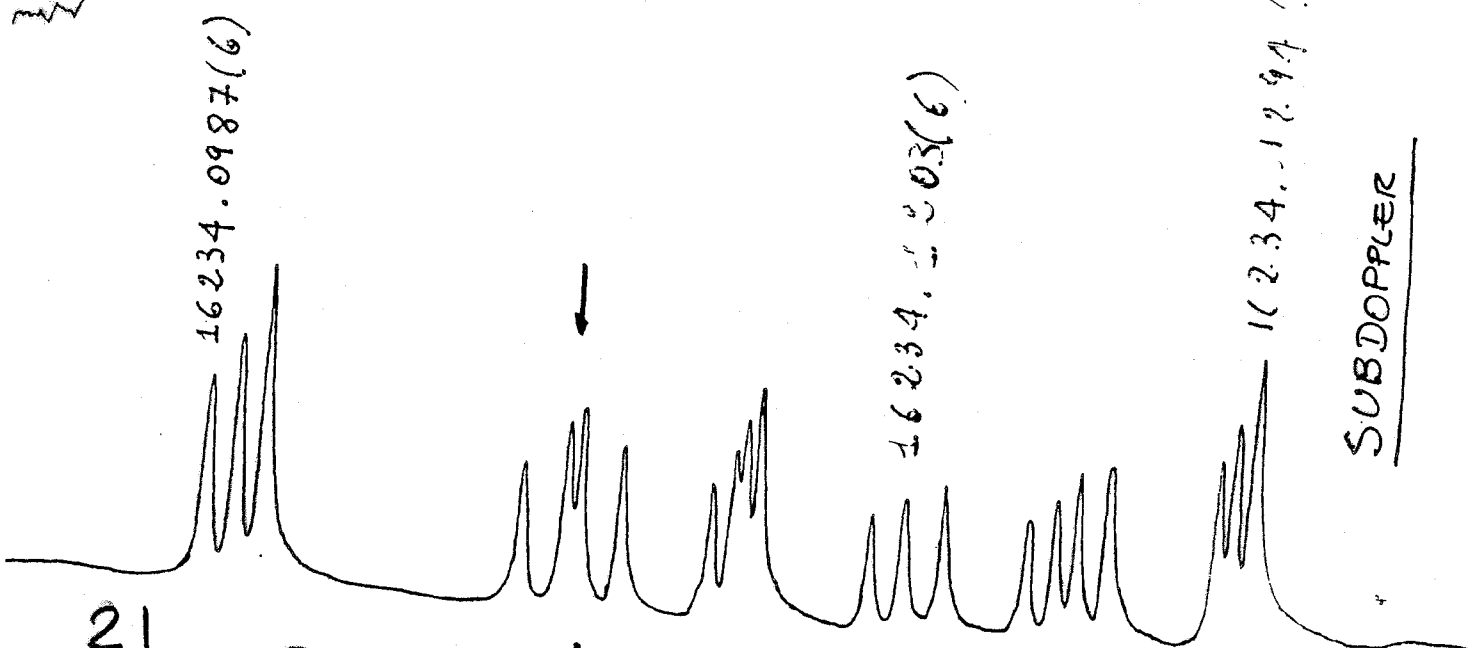
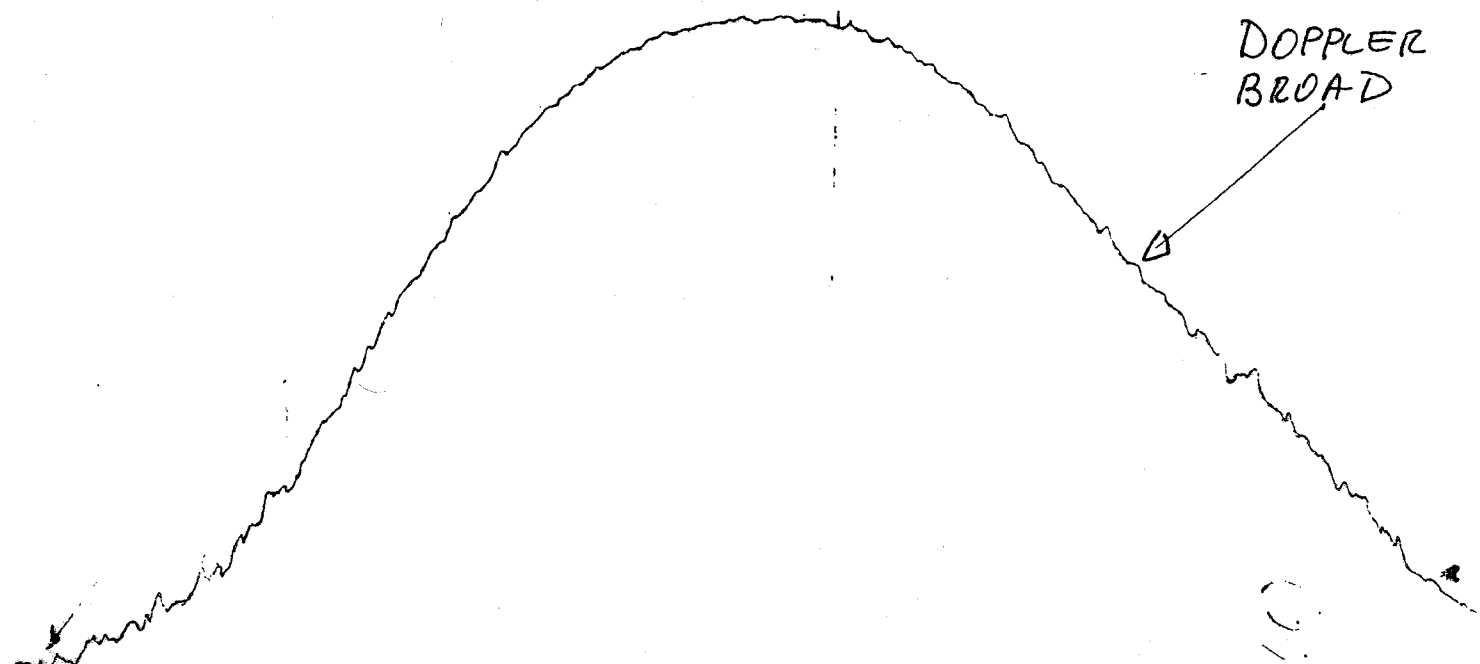
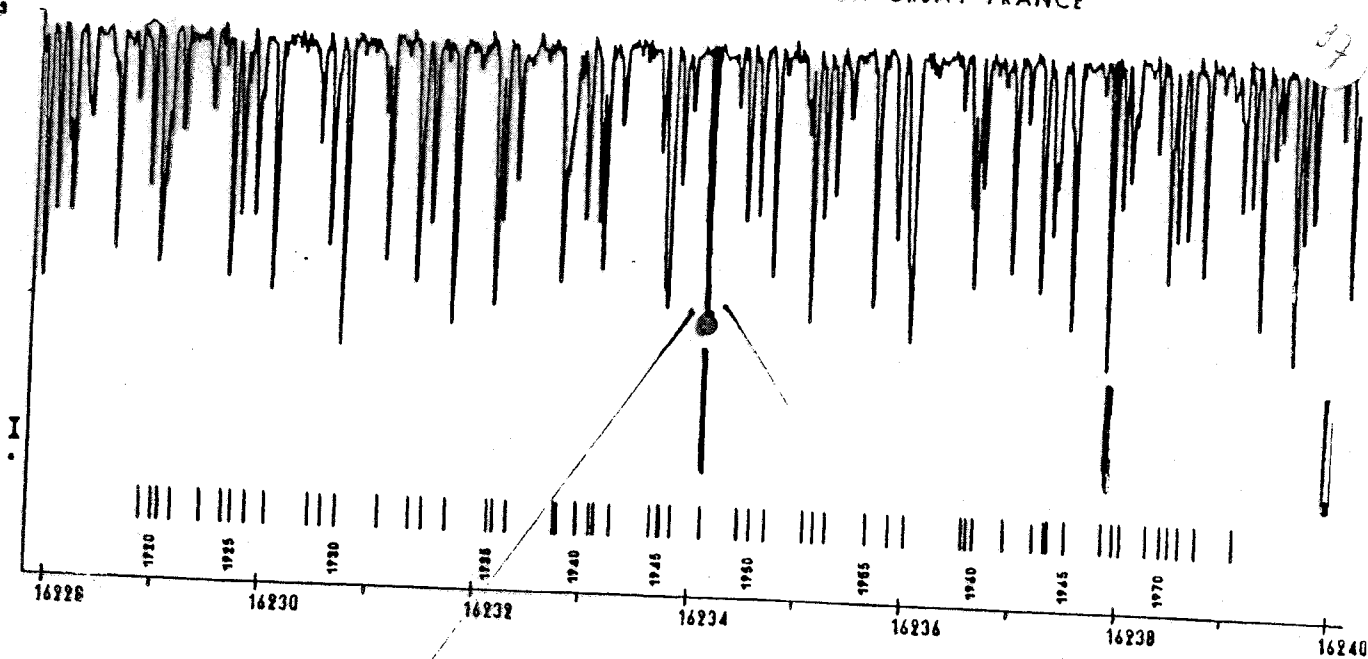


Fig. 22 Struttura iperfina della molecola di Iodio .
 Profilo di assorbimento con fascio opposto saturante .

.7033

37



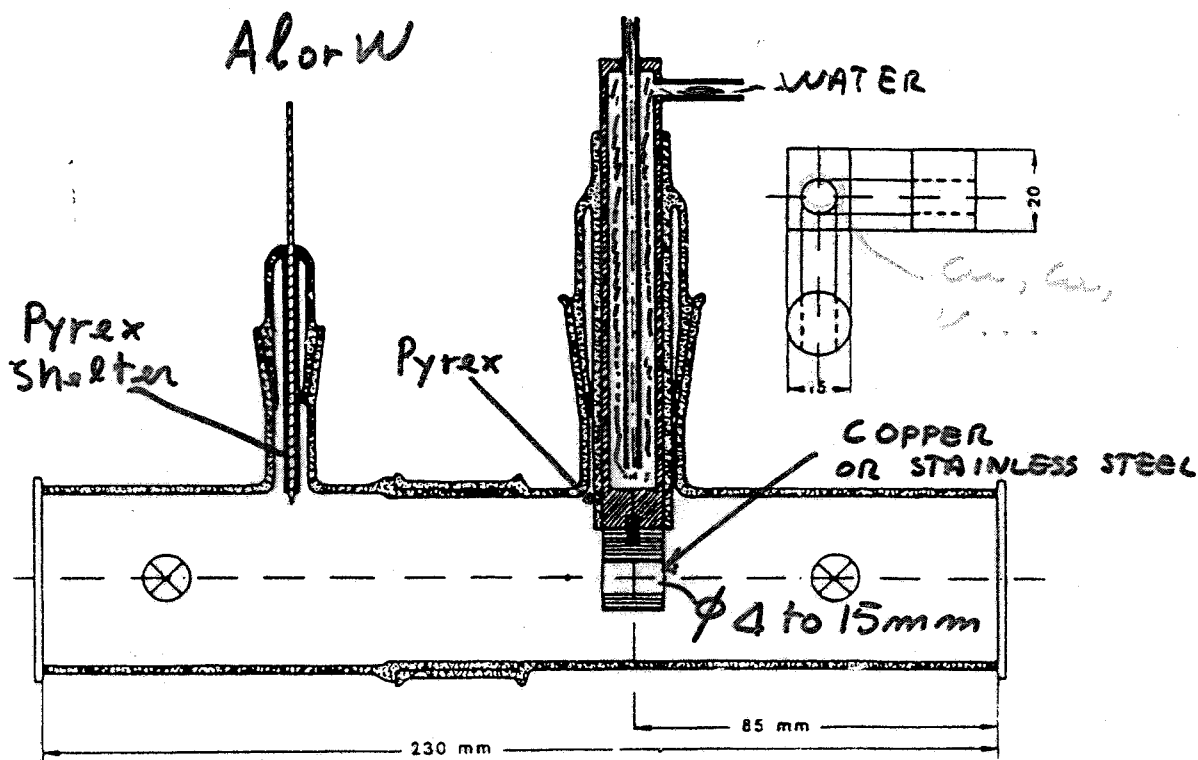
21
COMPON. Polarization spectra of hpf component

OG INVESTIGATION OF REFRACTORY ELEMENTS

$$I = 2.5 \text{ C mA}$$

f. $(Ar \text{ or } Ne)_2, C, 2 Torr$

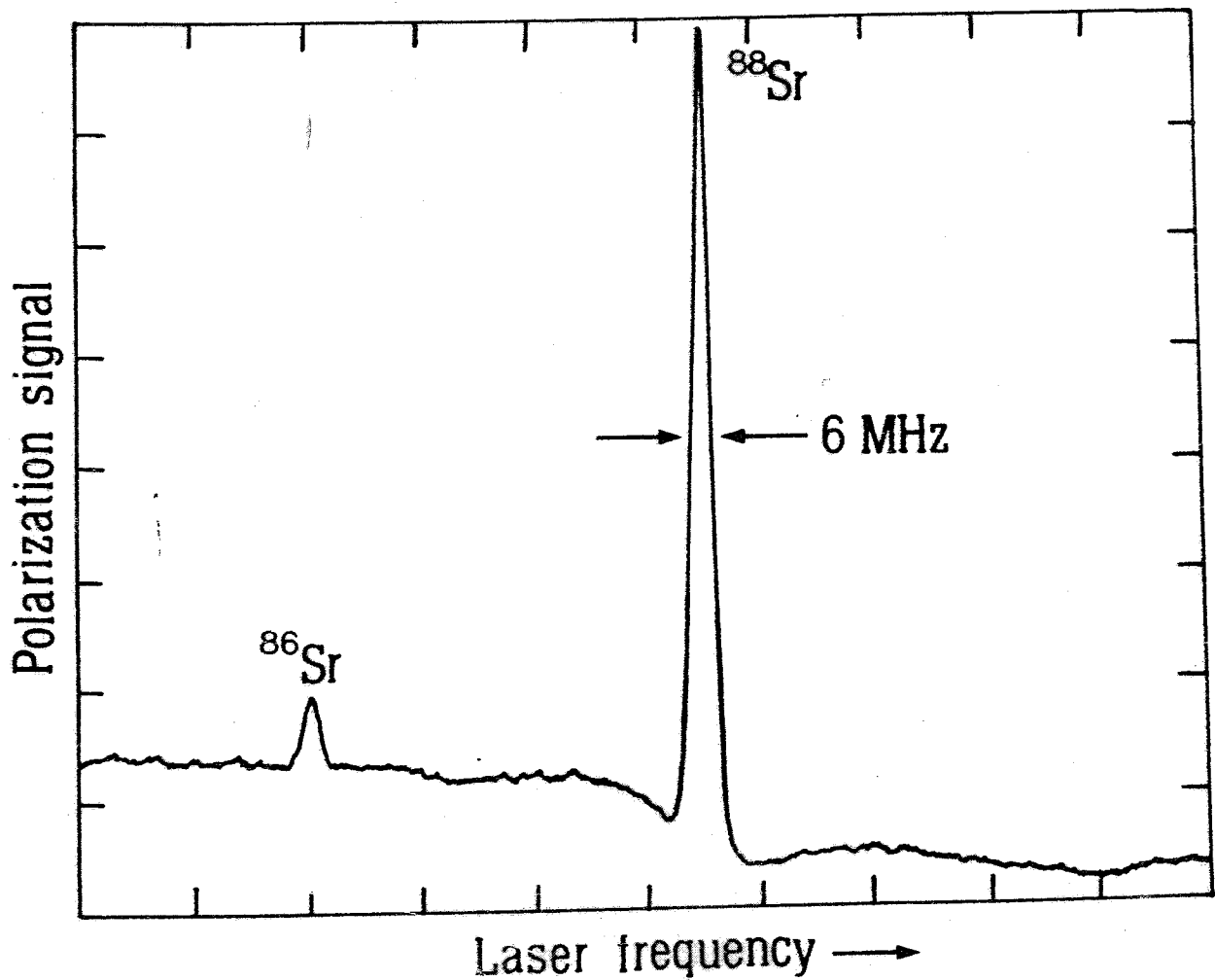
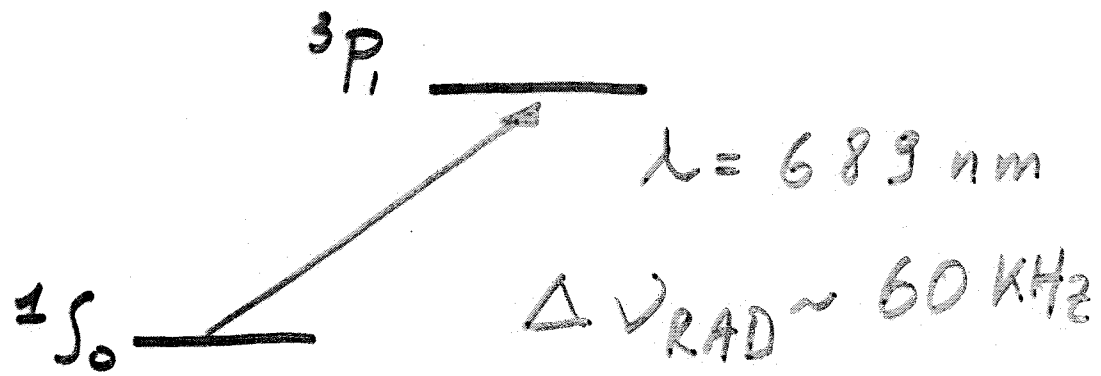
(Sealed off)
OR
FLOWING INTO THE
LASER



M. Inguscio. "High Resolution Intermodulated and Double Resonance Atomic Spectroscopy in a Hollow Cathode" J. Phys. (Paris) 44 C (1983)

Hollow Cathode Discharge

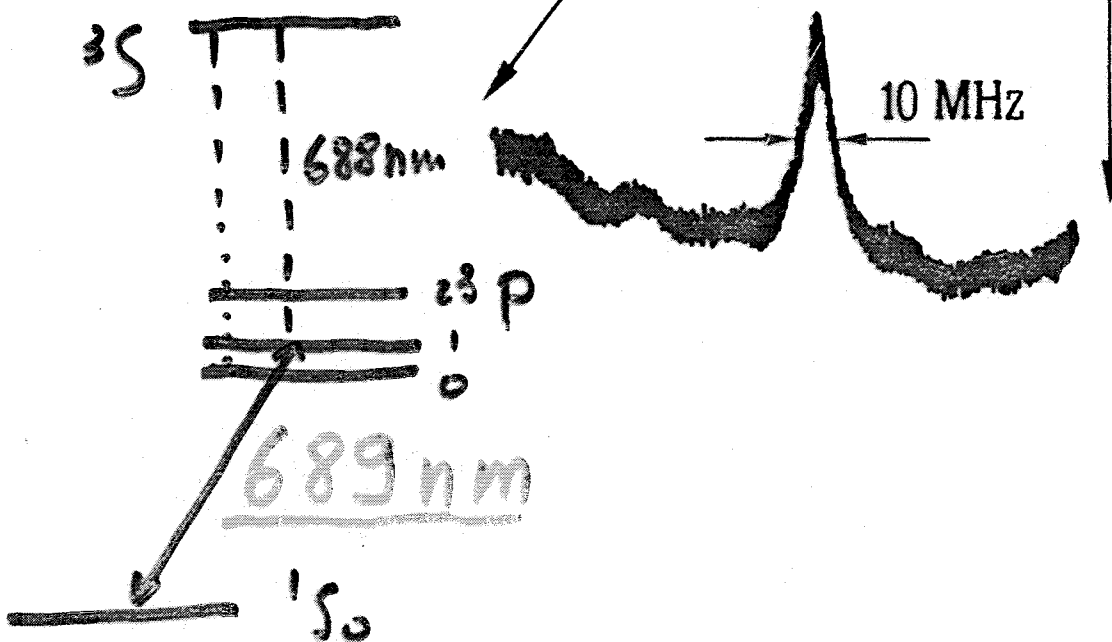
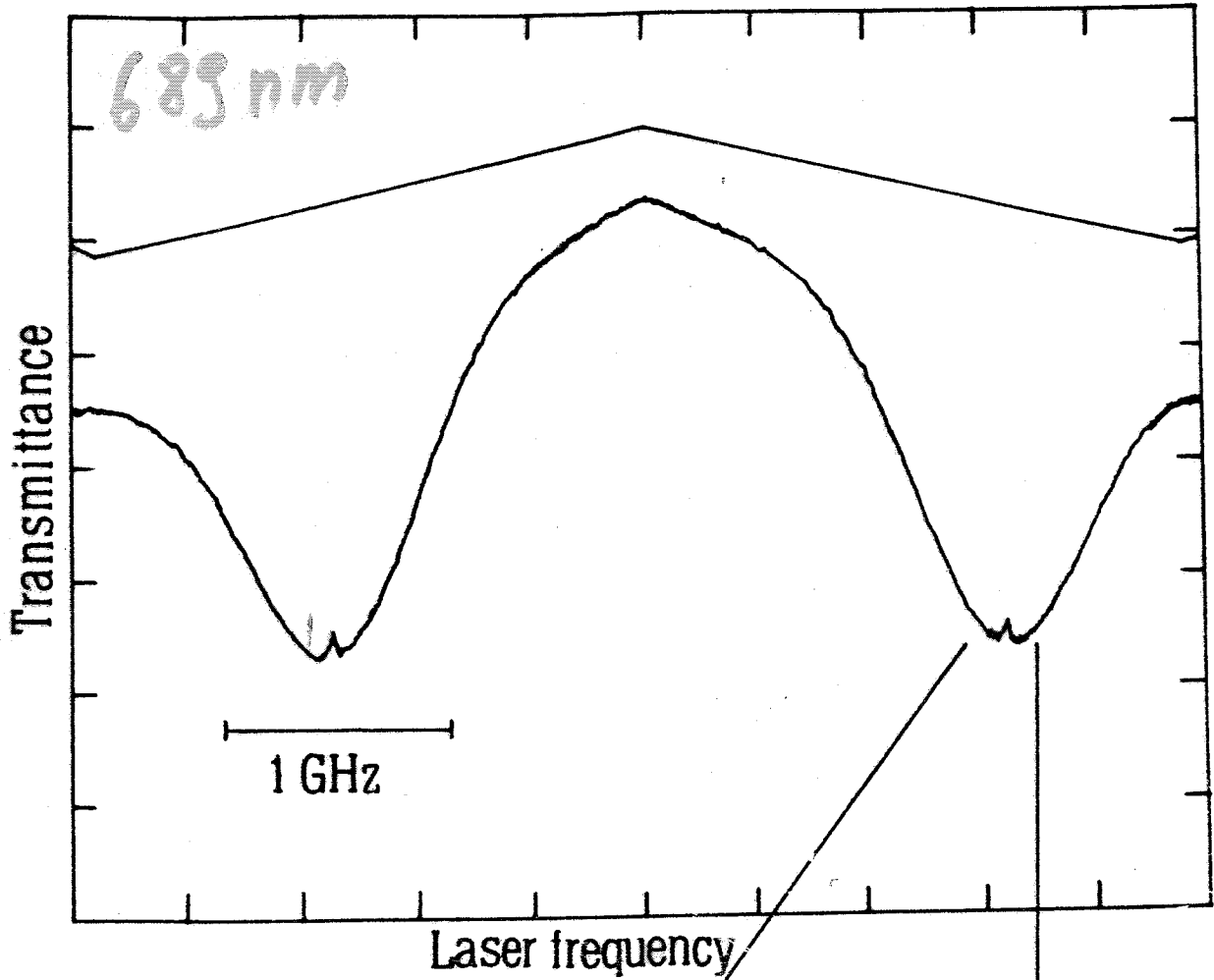
Sr intercombination line ⁸³ ~~83~~



POLARIZATION SPECTROSCOPY

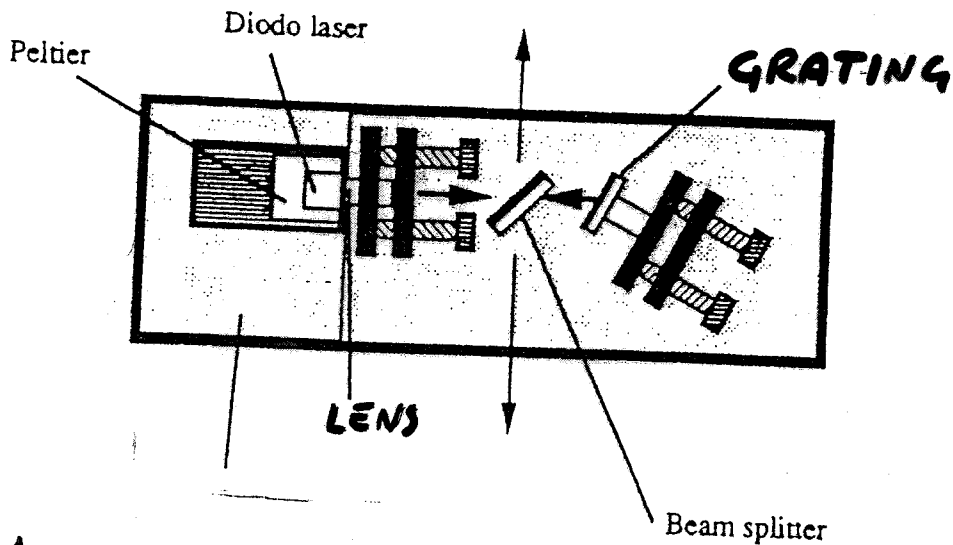
(HOLLOW CATHODE DISCHARGE)

ABSORPTION FROM S_F IN A HOLLOW CATHODE

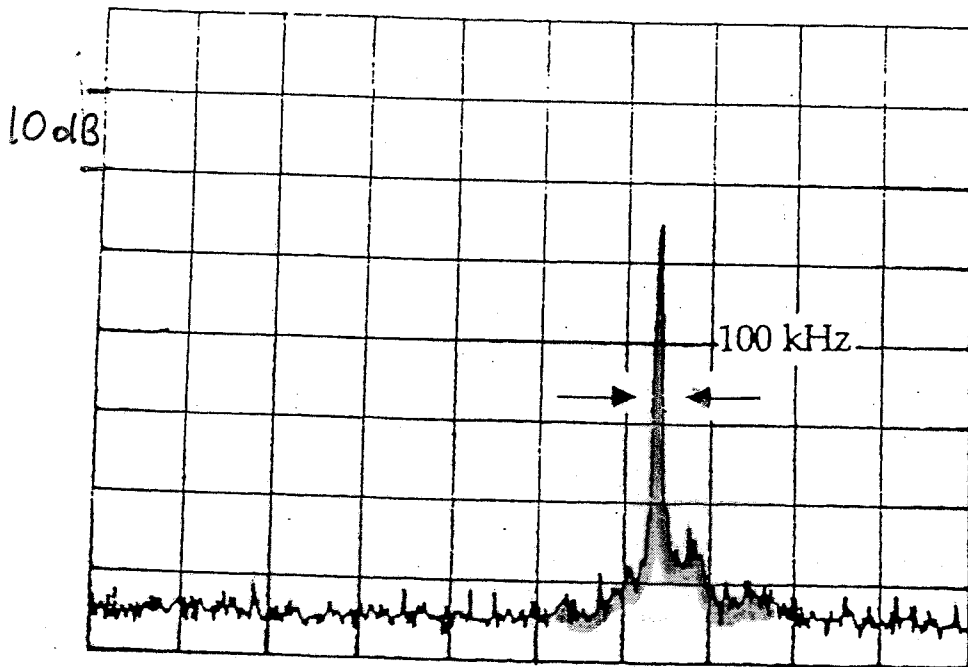


GRATING FEEDBACK NARROW BAND SEMI CONDUCTOR DIODE LASERS

(1)



$$\frac{\Delta \nu}{\nu} \lesssim 10^{-11} \text{ (fast)}$$



BEATING 2 INDEPENDENT SDL

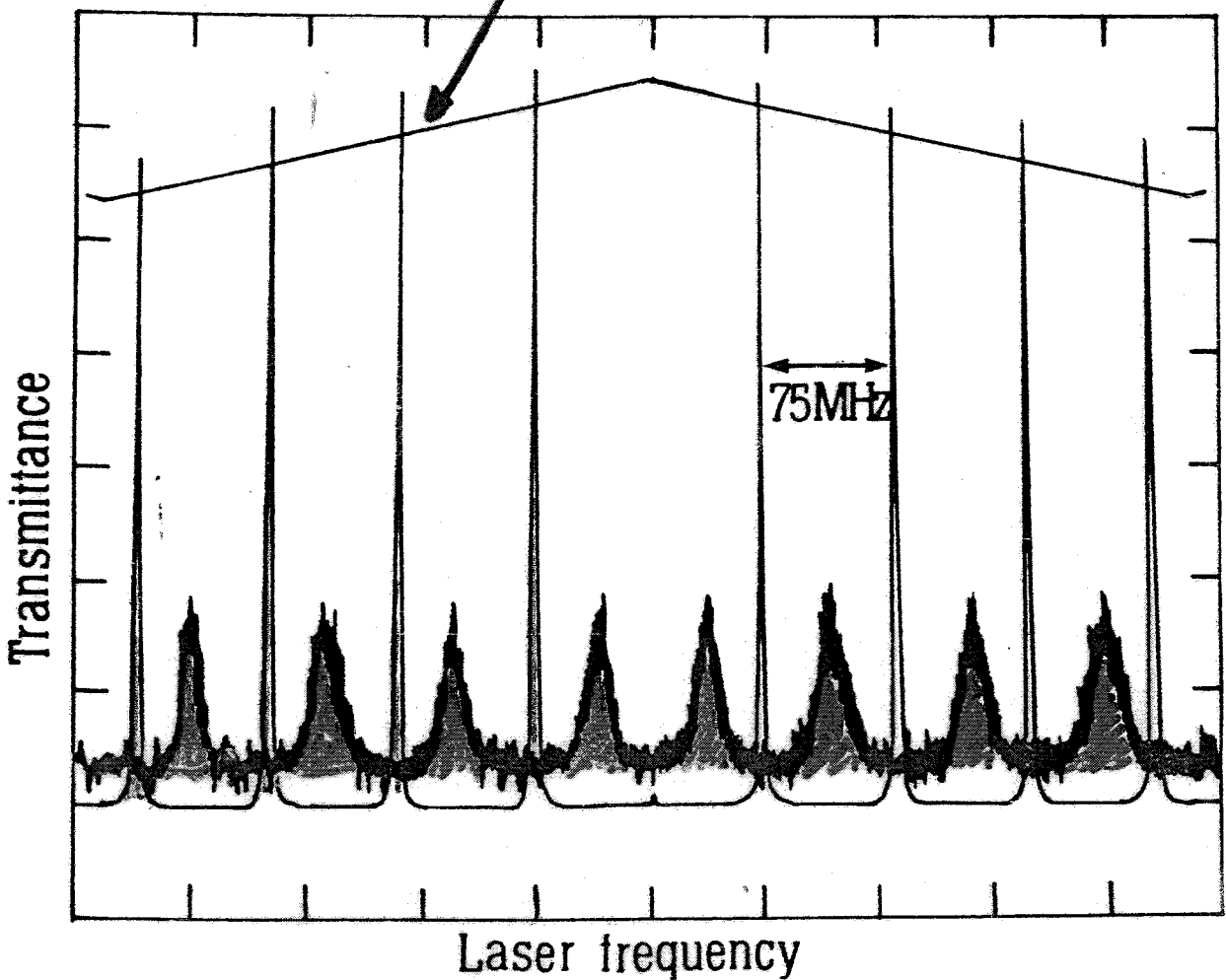
2 GHz apart

50 msec SWEEP

InGaP laser at 690 nm

LASER → ()

tuning current

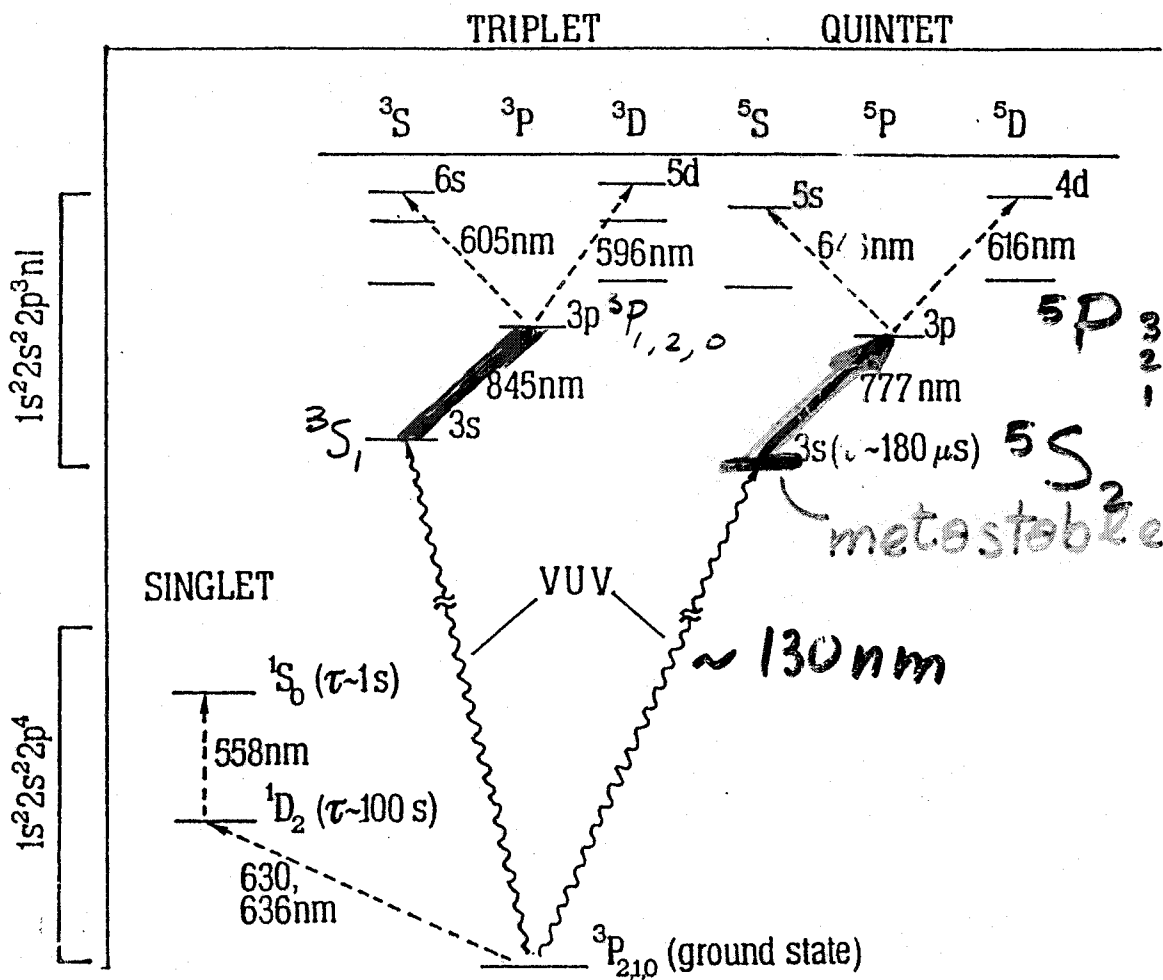


G. TINO, M. de Angelis, L. Gianfrani,
M. Barsanti, M. I.

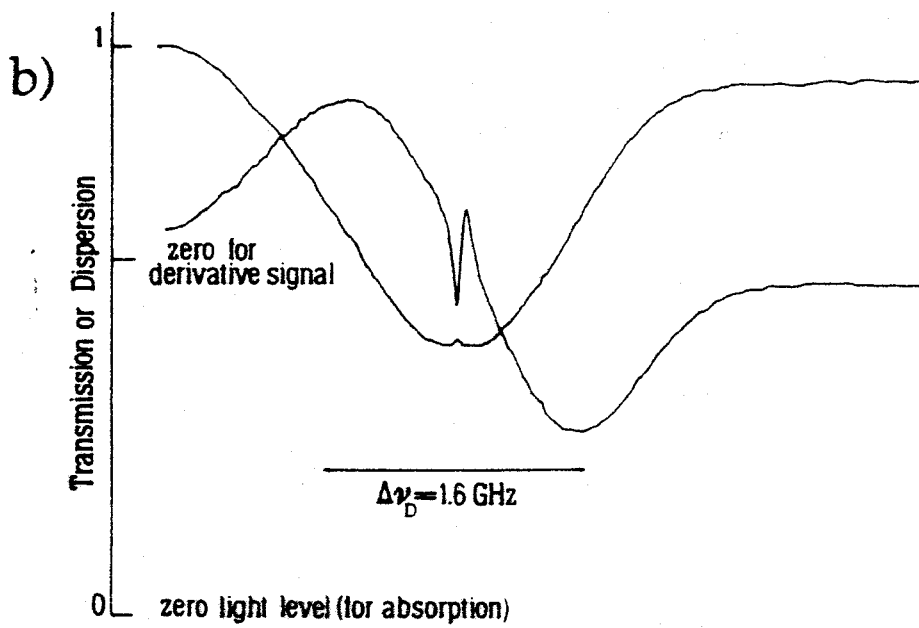
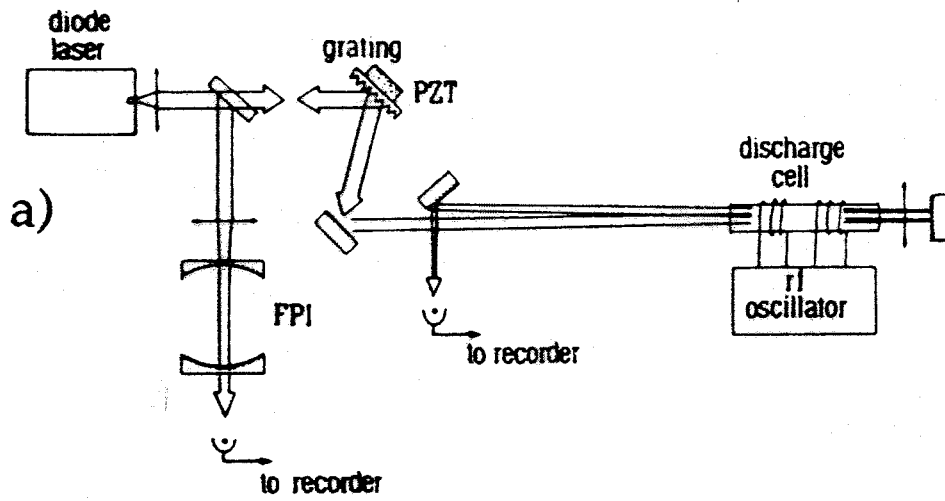
Appl. Phys. B (1992) (NAPOLI + FIRENZE)

high resolution spectroscopy and OXYGEN-ATOM

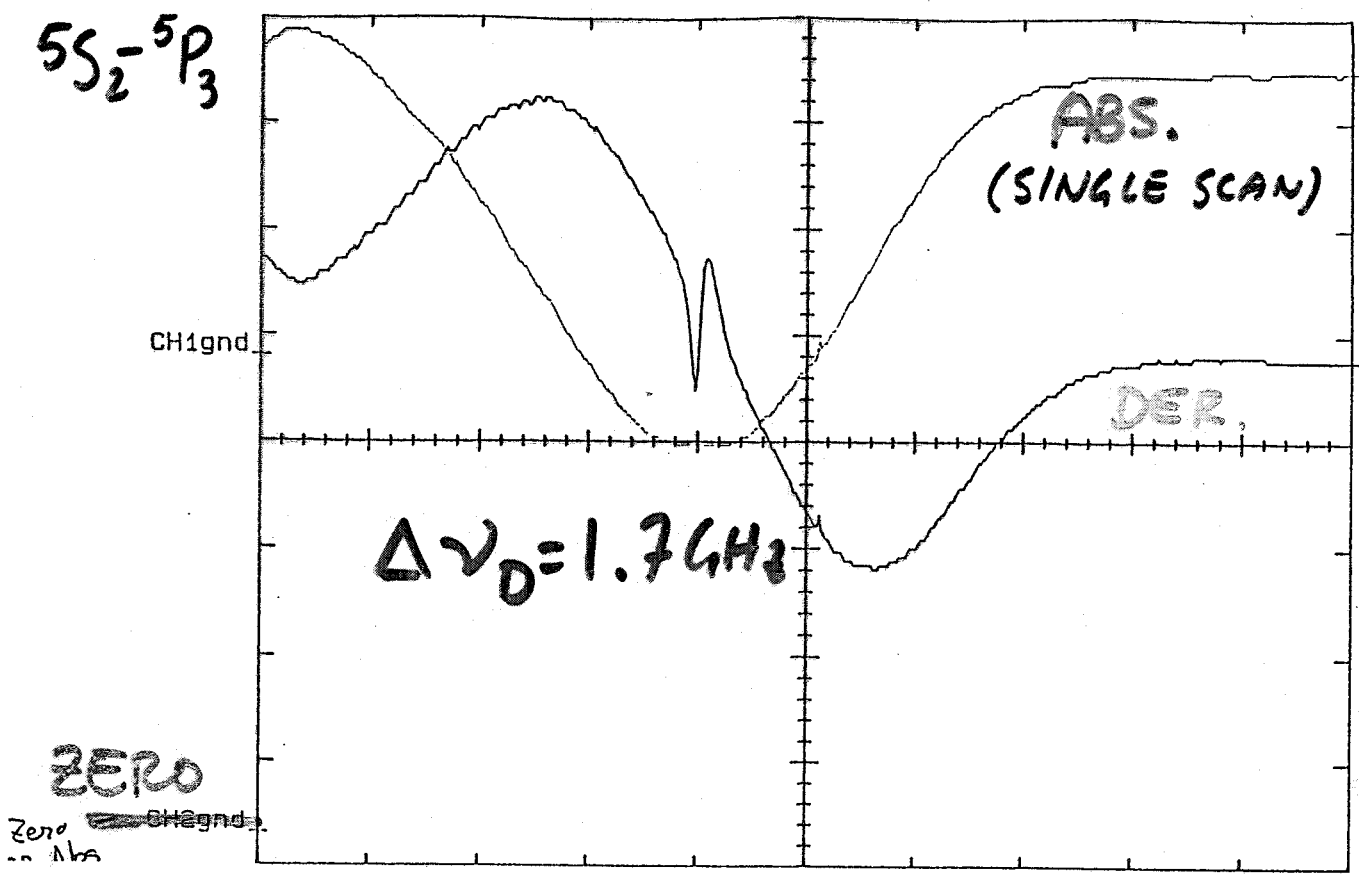
- Isotope and hyperfine structures
- Cooling



main difficulty is the level scheme of the atom itself



$5S_2 - 5P_3$



note the low noise level!

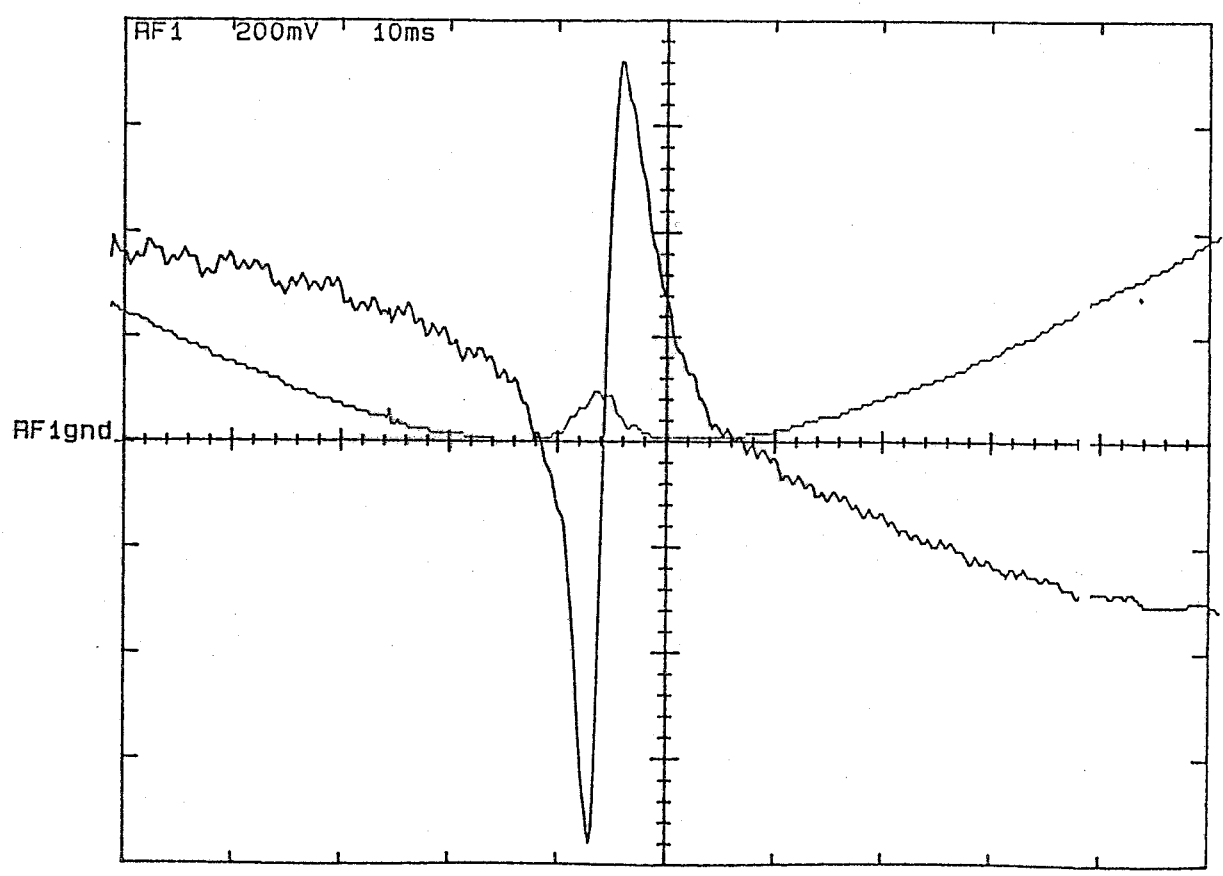
9/2

SATURATION SPECTROSCOPY

IN A 10MS -1.1/MV EX11

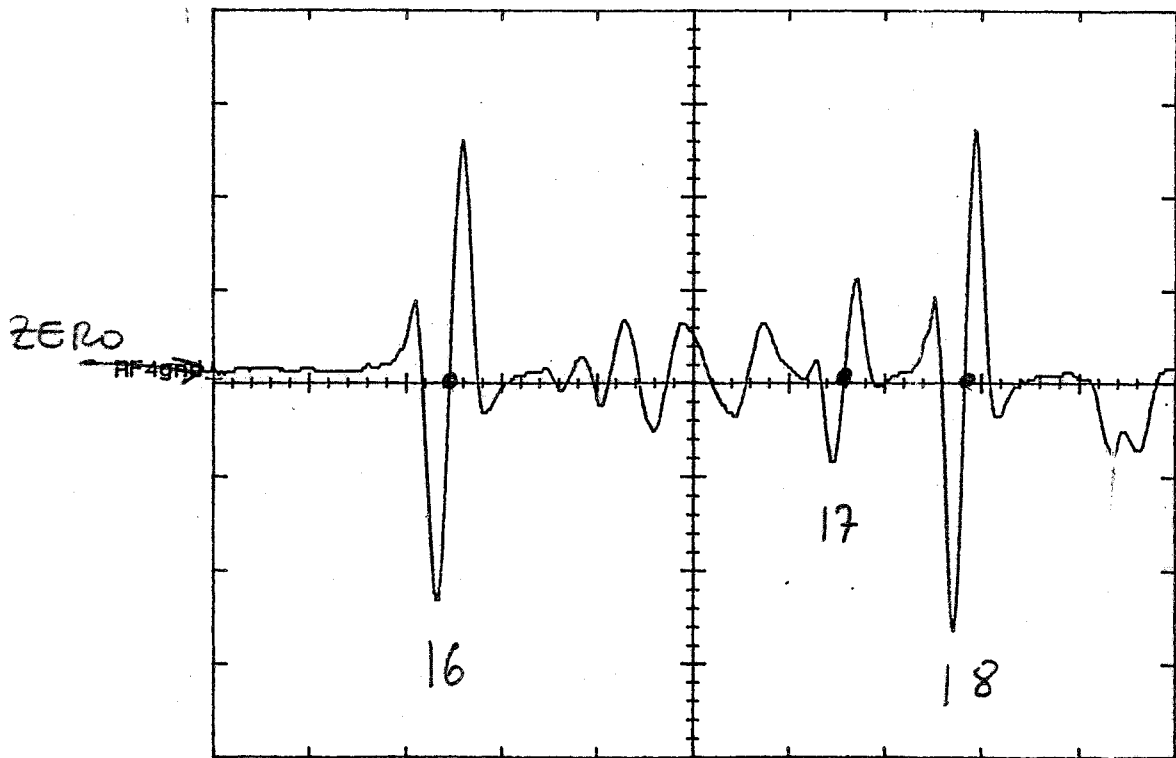
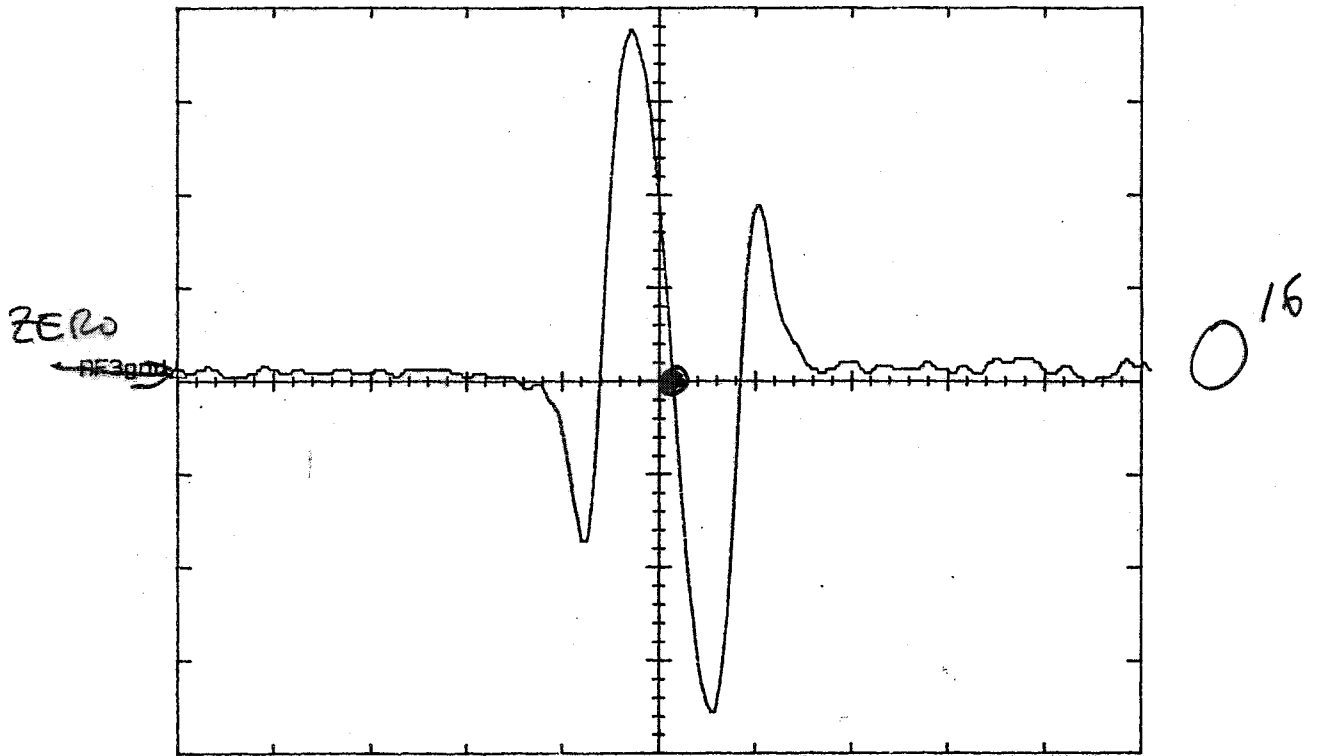
Al Ga As
777 nm

RF2 500mV 10ms

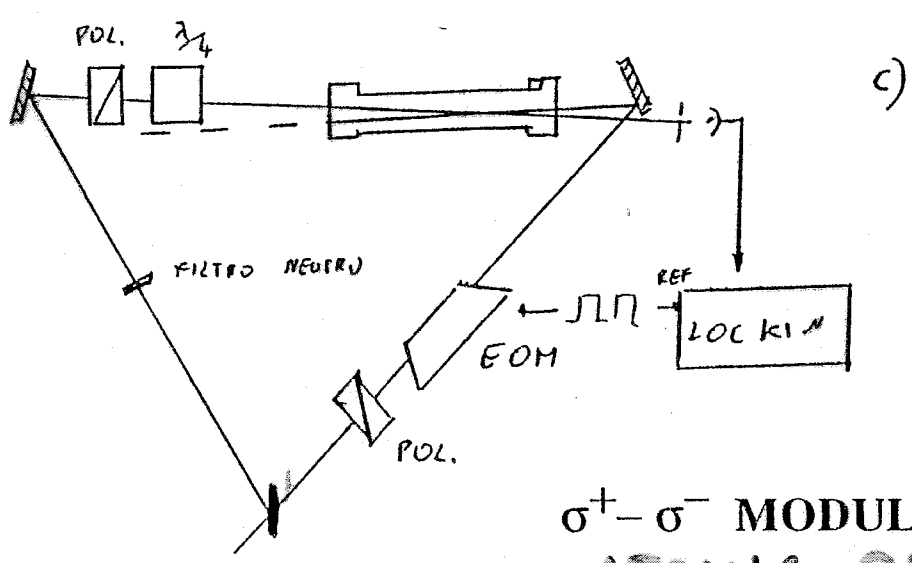


P.R. LETTERS 64 2999 (1990) G.N. Tino et al.

3rd DERIVATIVE (NO BASELINE)



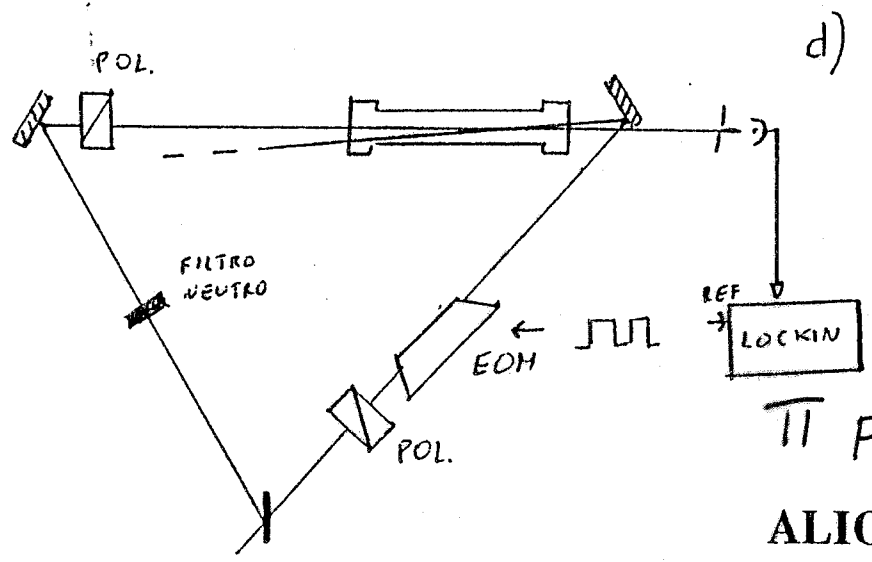
25/9/90 F. MARIN, P. DENATALE, M. I., M. PREVEDELLI,
 L. R. ZINK and G. TINO OPTICS LETT. ~~1992~~
 15 JAN. 1992



6 pump polarization

$\sigma^+ - \sigma^-$ MODULATION
 ATOMIC ORIENTATION

$$O_{SS} = \sum_{m_J}^{+J} n_{m_J} m_J$$

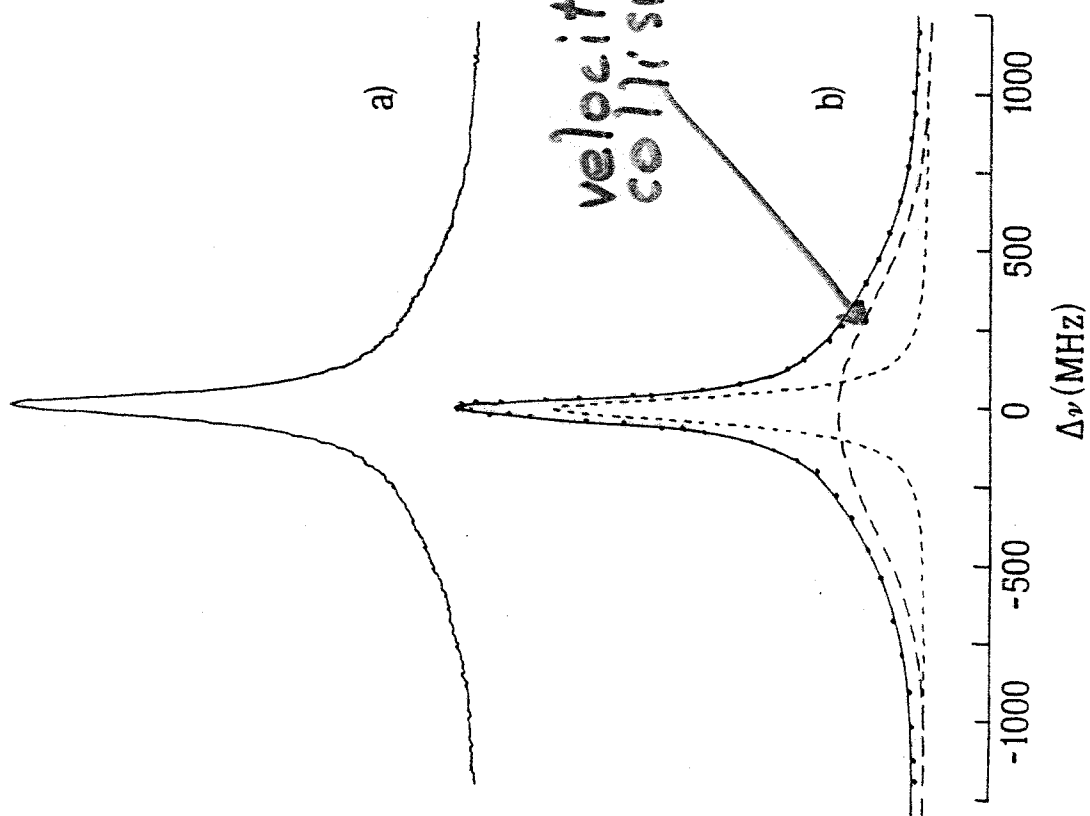


Π pump polarization

ALIGNMENT

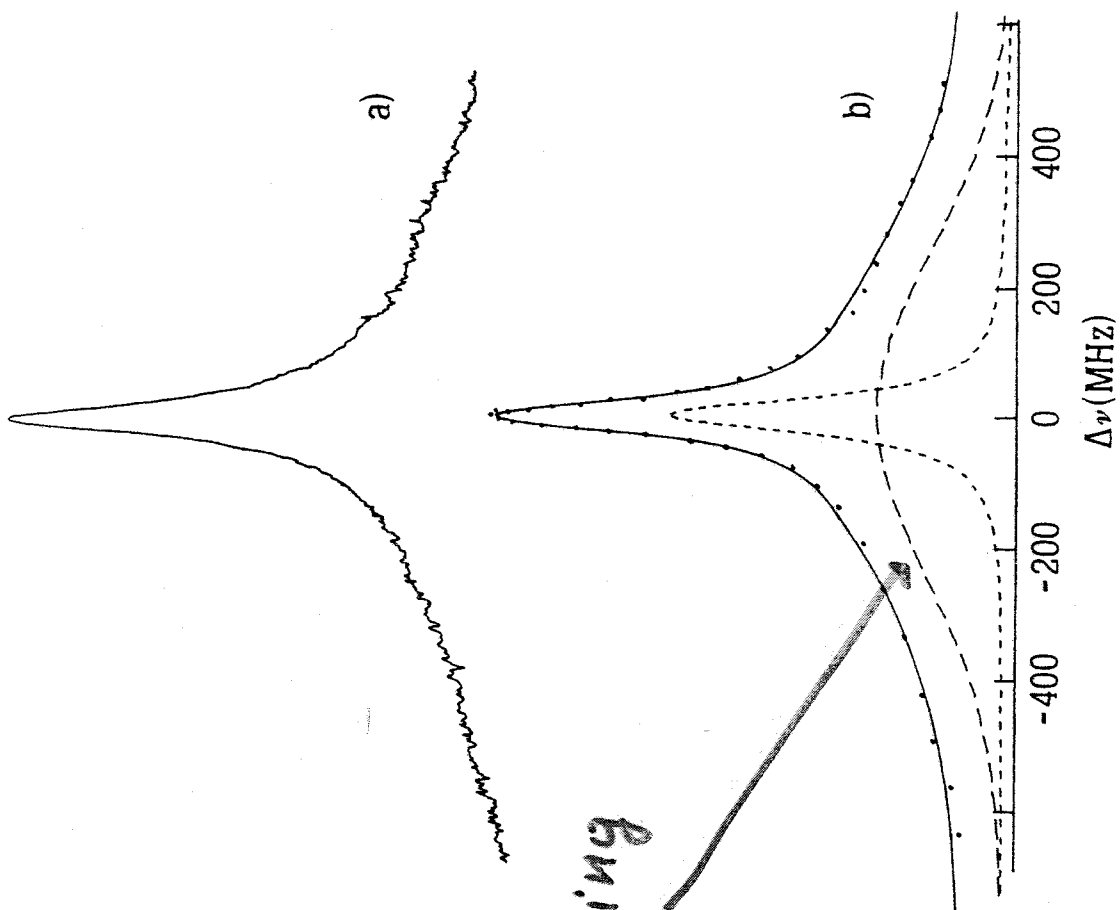
ATOMIC ALIGNMENT

$$O_{SS} = \sum_{m_J}^{+J} n_{m_J} m_J^2$$

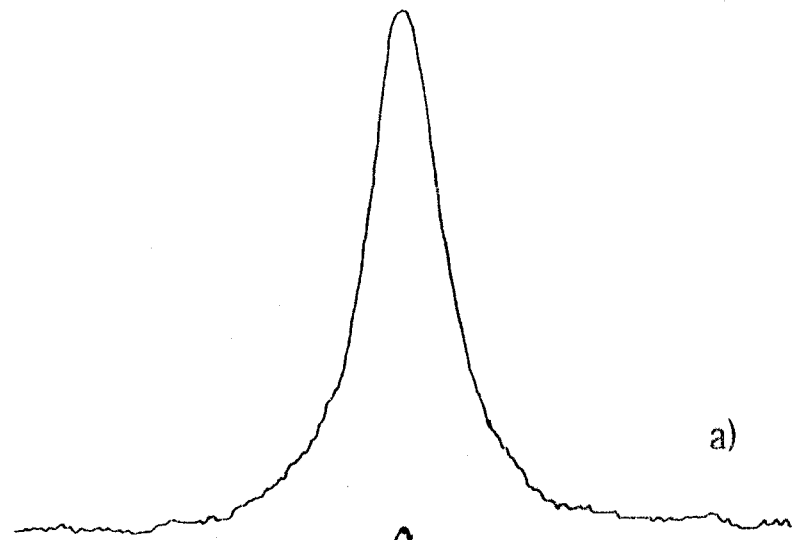


SATURATION (POPULATION)

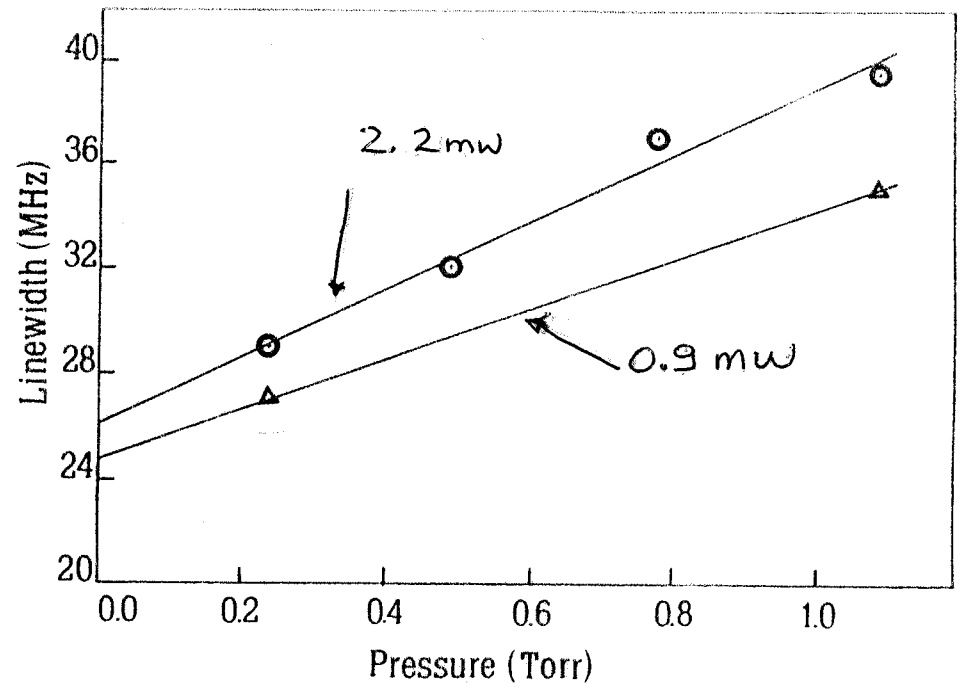
velocity-changing collisions.



σ^+ / σ^- (ORIENTATION)

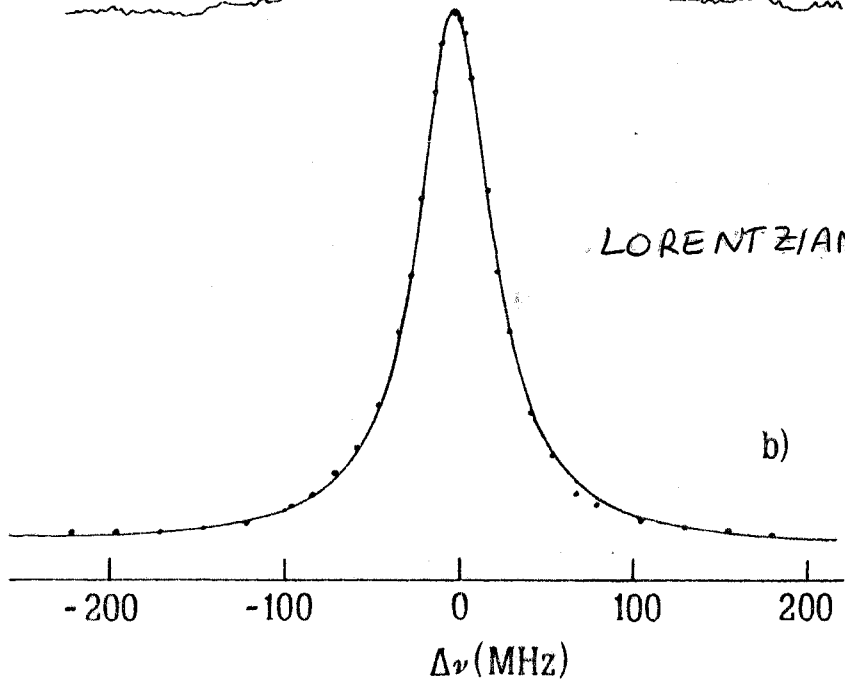


a)



$$\tau(^5P_3) = 7 \text{ nsec}$$

LORENTZIAN



b)

π (resonance)

RADIATIVE LINEWIDTH LIMITED RESOLUTION

M. deAngelis et al. Phys. Rev. A 44 5811 (91)

High sensitivity molecular
spectroscopy with
semiconductor diode
Lasers (an introduction)

M. Inguscio # 3

MOLECULAR SPECTRA

ELECTRONIC - UV

VIBRATIONAL - IR

ROTATIONAL - MW → FIR

~ NO VISIBLE - NEAR IR

Sources

Detectors

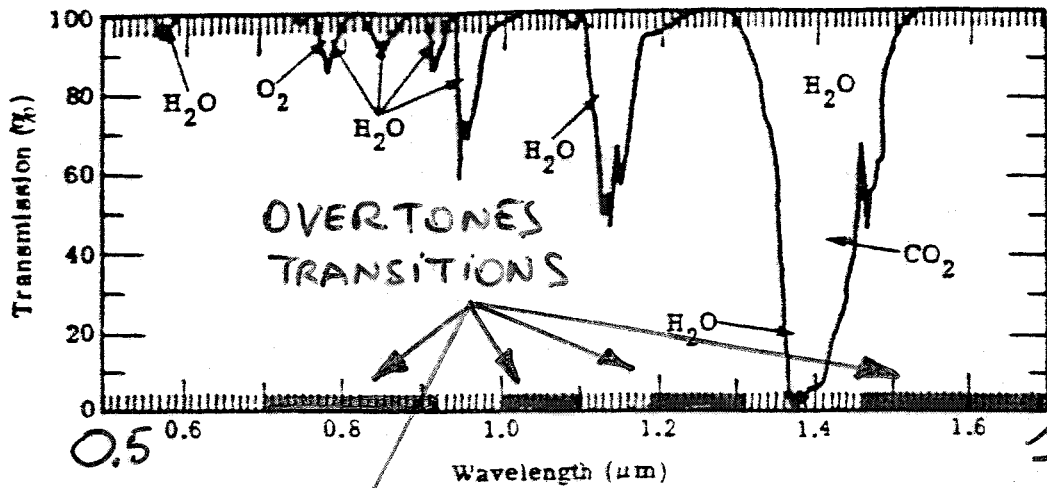
Transmission (Atmospheric or environment windows)

TRACE GAS SENSING

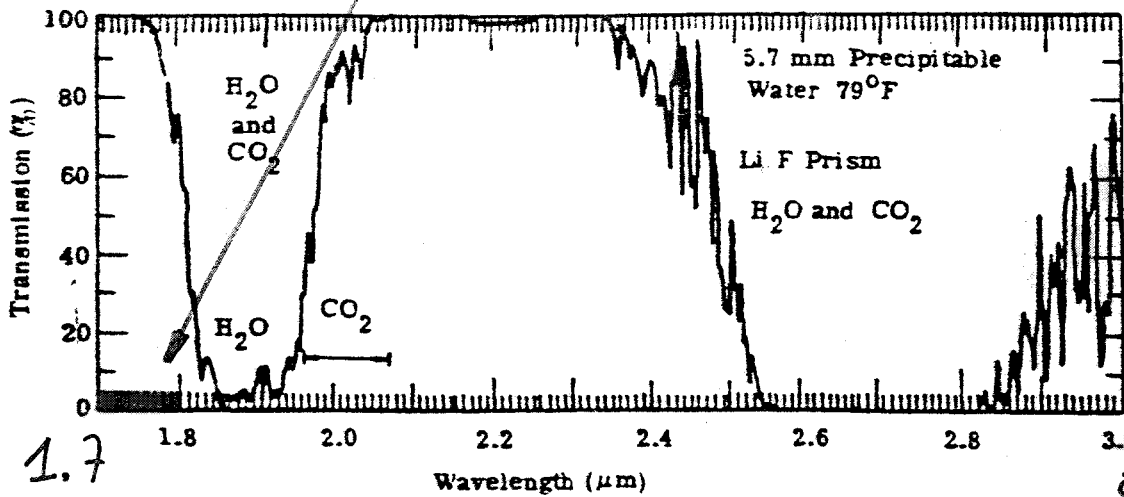
SEA LEVEL
100 meter path length

(3)

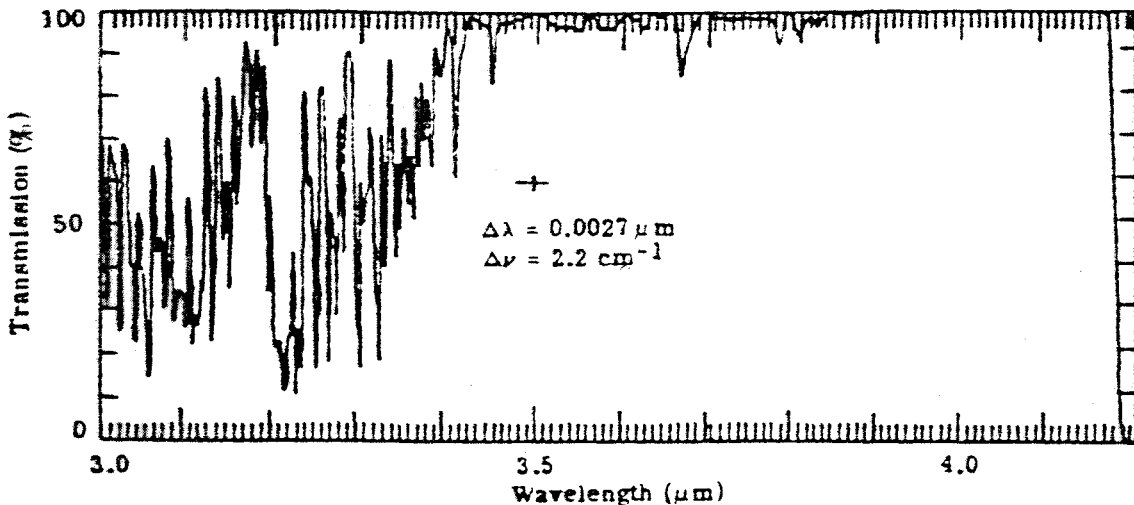
ATMOSPHERIC ABSORPTION



(a) 0.5 to 1.7 μm .



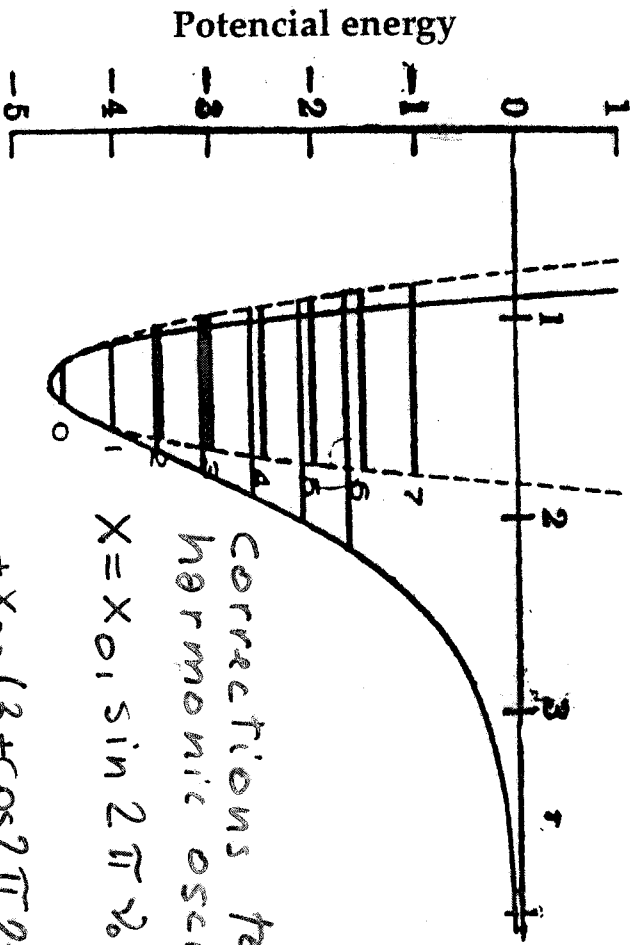
(b) 1.7 to 3.0 μm .



(c) 3.0 to 4.2 μm .

THE OVERTONE TRANSITIONS

(Intensity depends on the anharmonicity)

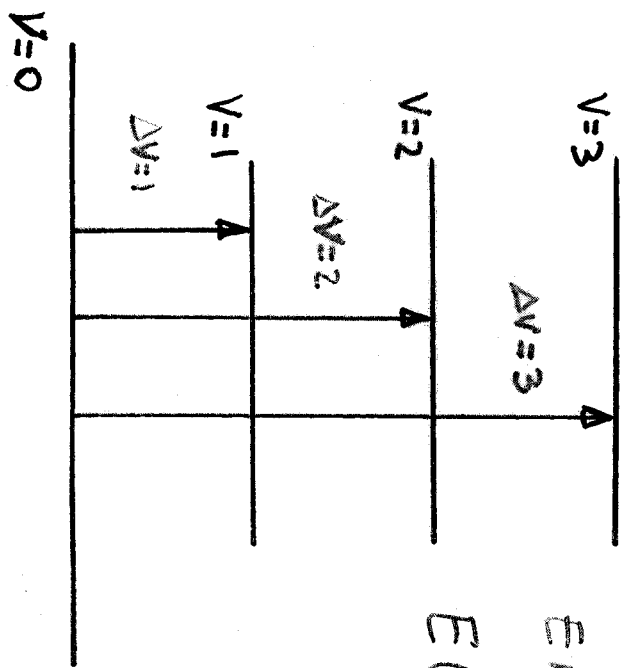


Corrections to the harmonic oscillation

$$\begin{aligned}
 X &= X_{01} \sin 2\pi \nu_0 t + \\
 &+ X_{02} (3 + \cos 2\pi 2\nu_0 t) + \\
 &+ X_{03} \sin 2\pi 3\nu_0 t + \dots
 \end{aligned}$$

ENERGIES

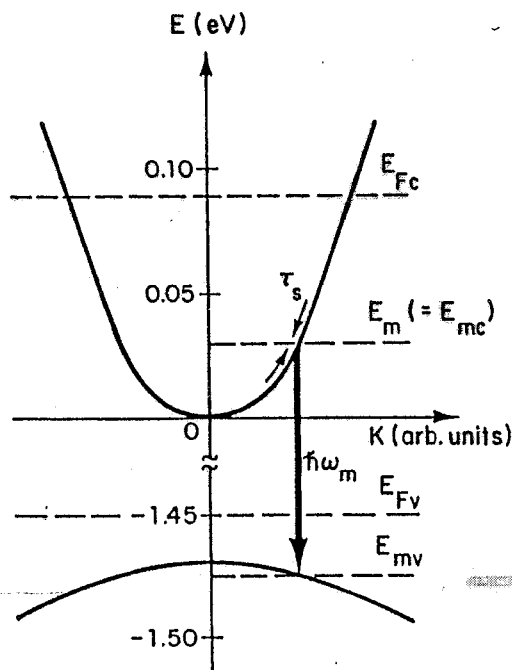
$$\begin{aligned}
 E(v) &= h\nu_0 \left(v + \frac{1}{2}\right) - h\nu_0 X_0 \left(v + \frac{1}{2}\right)^2 \\
 &+ h\nu_0 y_0 \left(v + \frac{1}{2}\right)^3 \dots
 \end{aligned}$$



SEMICONDUCTOR DIODE LASER

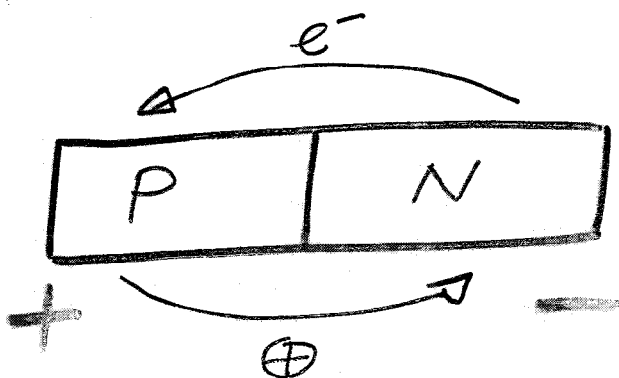
25 (7)

DIRECT INTERBAND LASER TRANSITION (GaAl)As



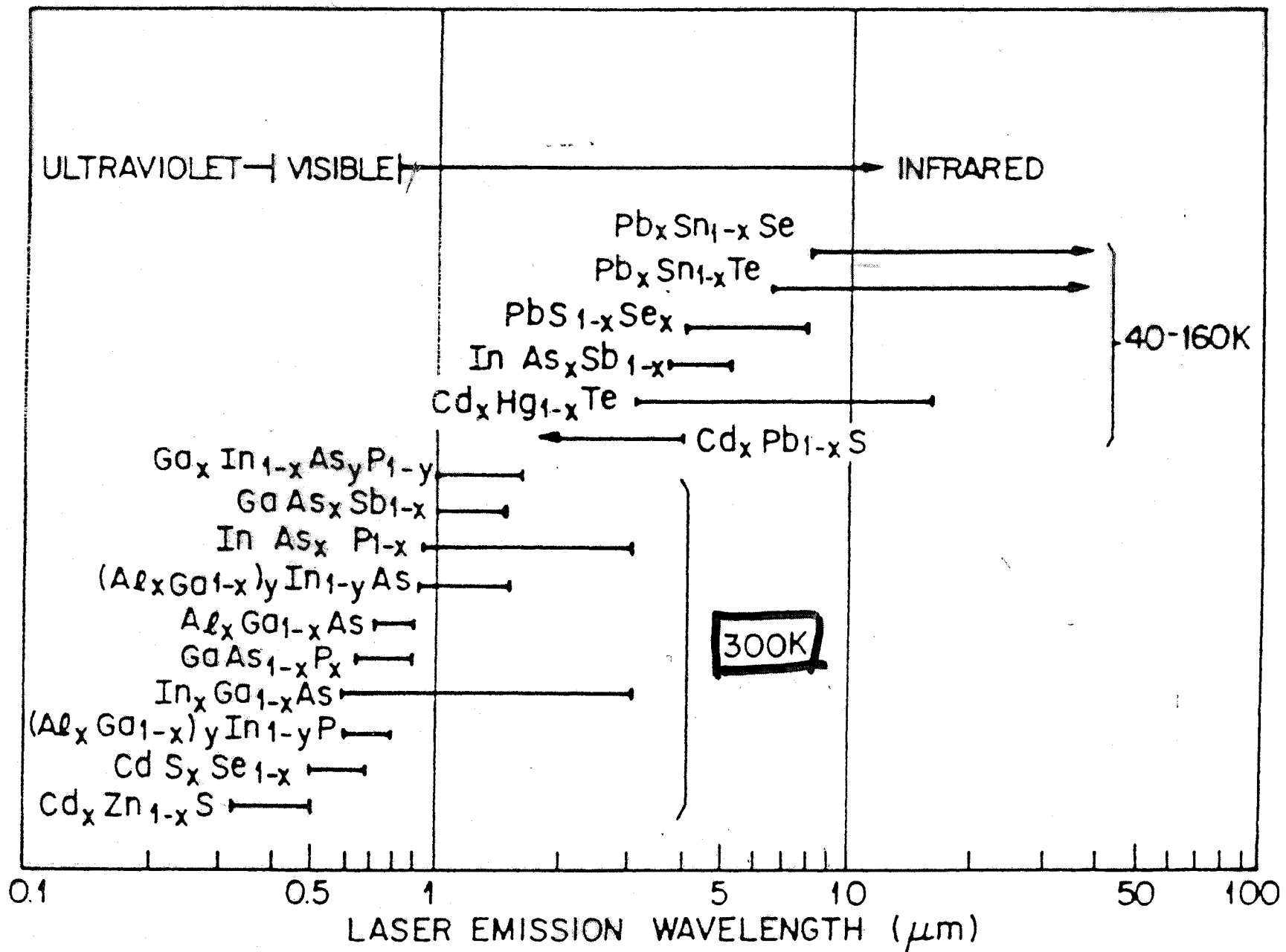
BAND GAP

ig. 8 Schematic diagram of the simplified band structure of (GaAl)As showing the direct interband laser transition.



Also InGaP
InGaAsP

40-



Emission wavelenghts of direct semiconductors

(17)

SHOT NOISE $\propto \sqrt{N_{\text{phot}}}$ (fluctuations in the photon distrib.)

800 nm, 5 mW $\sim 10^{-8}$

here we are a factor

10-20 db worse, so

we expect

$$\frac{\Delta I}{I} \sim 10^{-6}$$

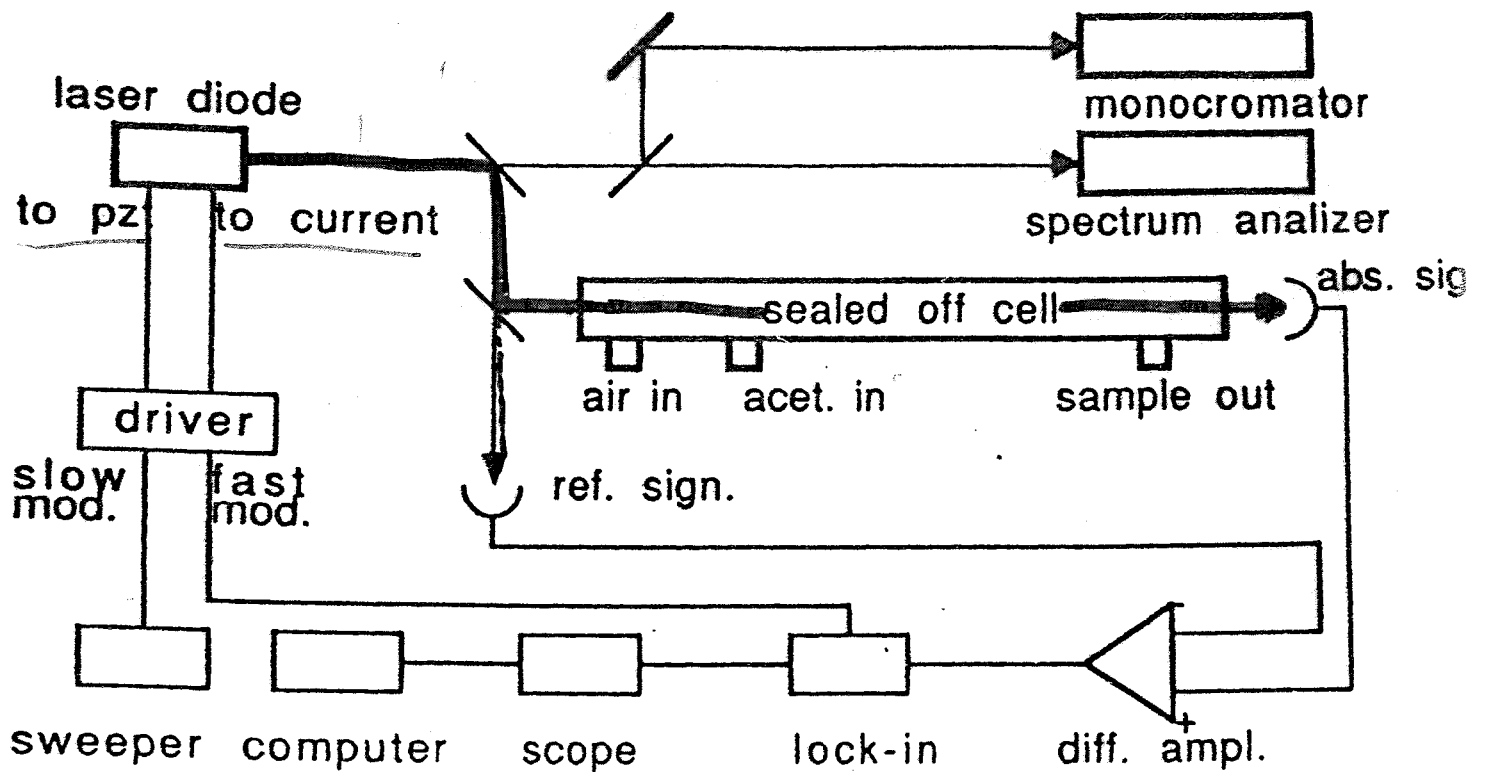


HIGH SENSITIVITY IN PURE
ABSORPTION EVEN FOR VERY
WEAK TRANSITIONS

~~more or less~~ general ...

19

EXPERIMENTAL APPARATUS (SIMPLEST) FOR OVERTONE SPECTROSCOPY



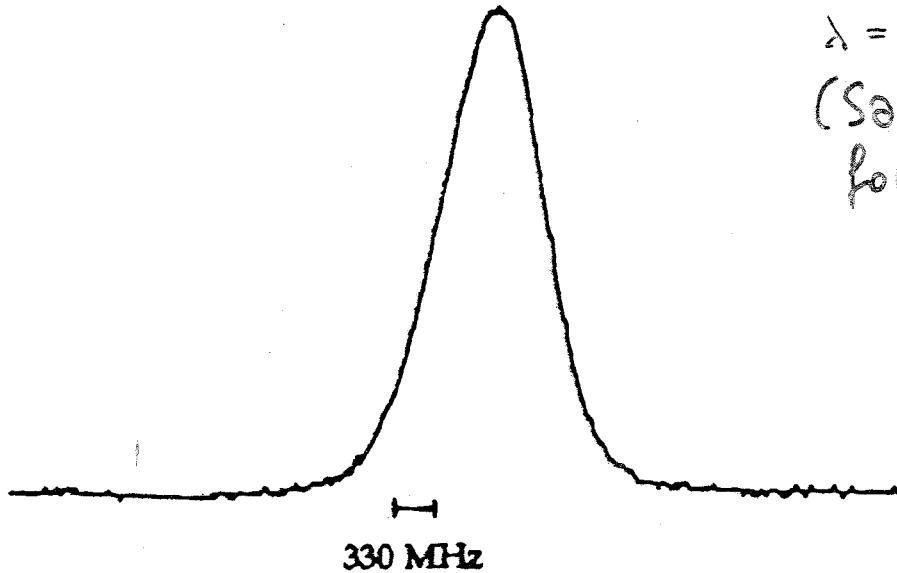
Simple source, simple detector,
simple electronics...

PURE ABSORPTION SIGNAL FOR THE R(9) COMPONENT (C₂H₂)

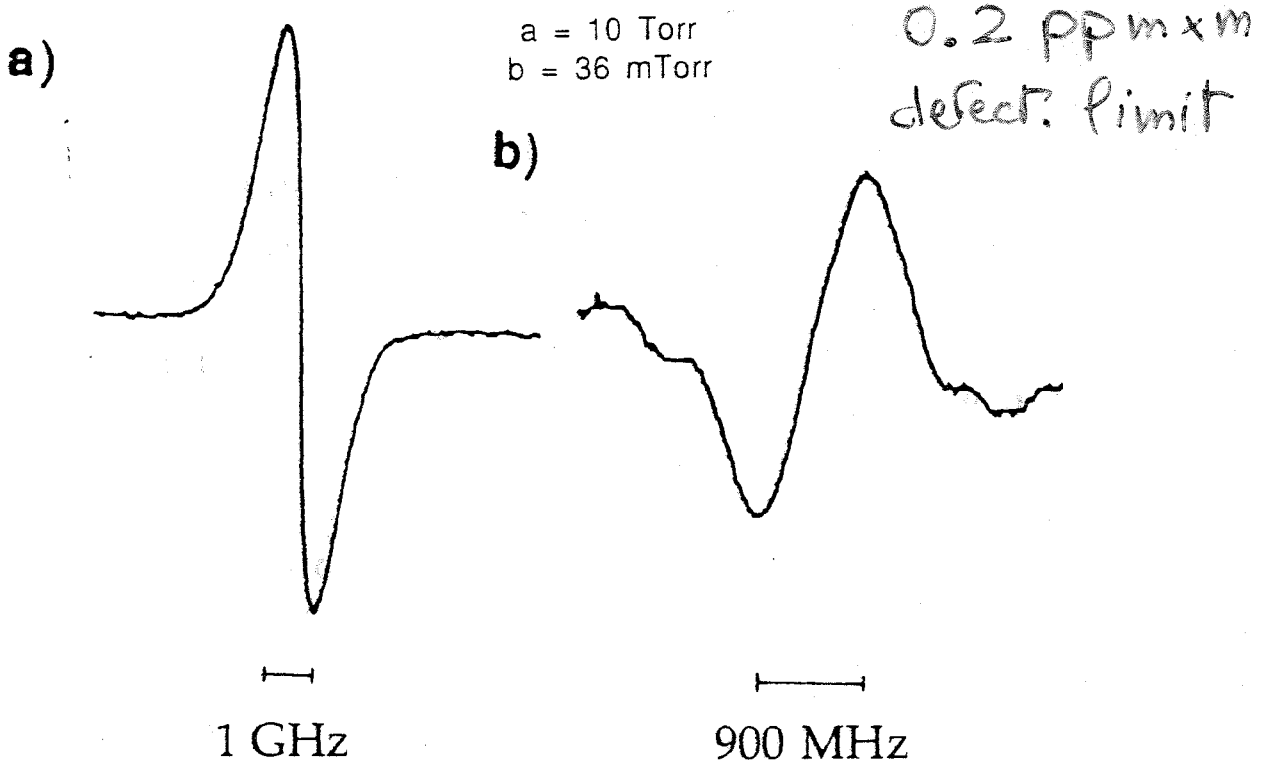
20

pressure = 10 Torr
 pathlength = 1.5 meter
 absorption = a few percent

3rd OVERTONE
 $\lambda = 788 \text{ nm}$
 (Same laser for R(9))



DERIVATIVE SIGNAL



Limit of detection in absorption : 10^{-6} , 1Hz detect. bandw.

Last limit using FM techniques : up to 10^{-8}

Air pressure detection limit : 0.2 ppm

IMPROVING SENSITIVITY...

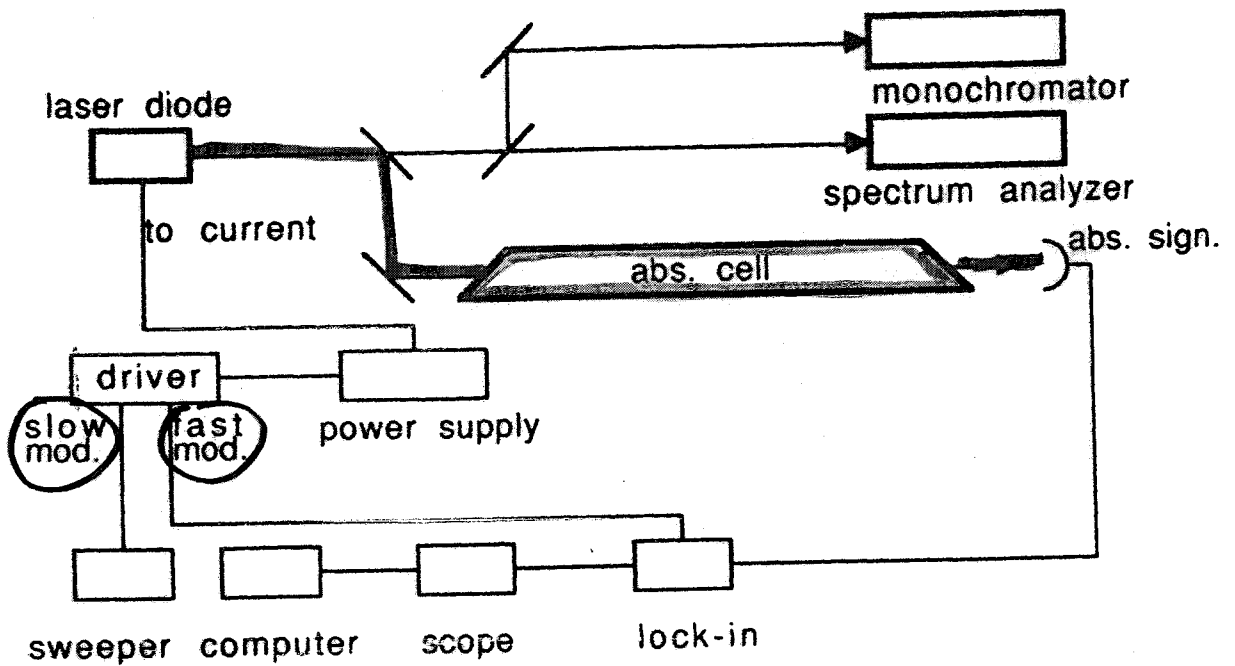
(21)

1st METHOD :

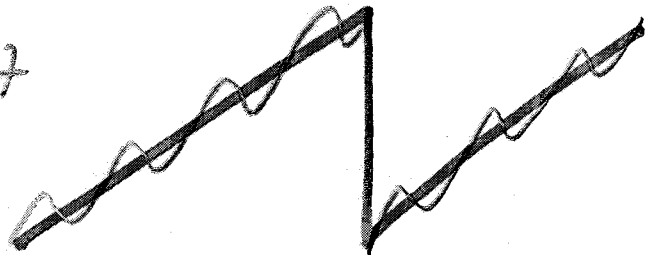
LOW WAVELENGTH MODULATION

MODULATION FREQUENCY 1 KHz

(SMALLER THAN MOLECULAR LINEWIDTH AND THE ABSORPTION IS PROBED BY MANY SIDE BANDS)

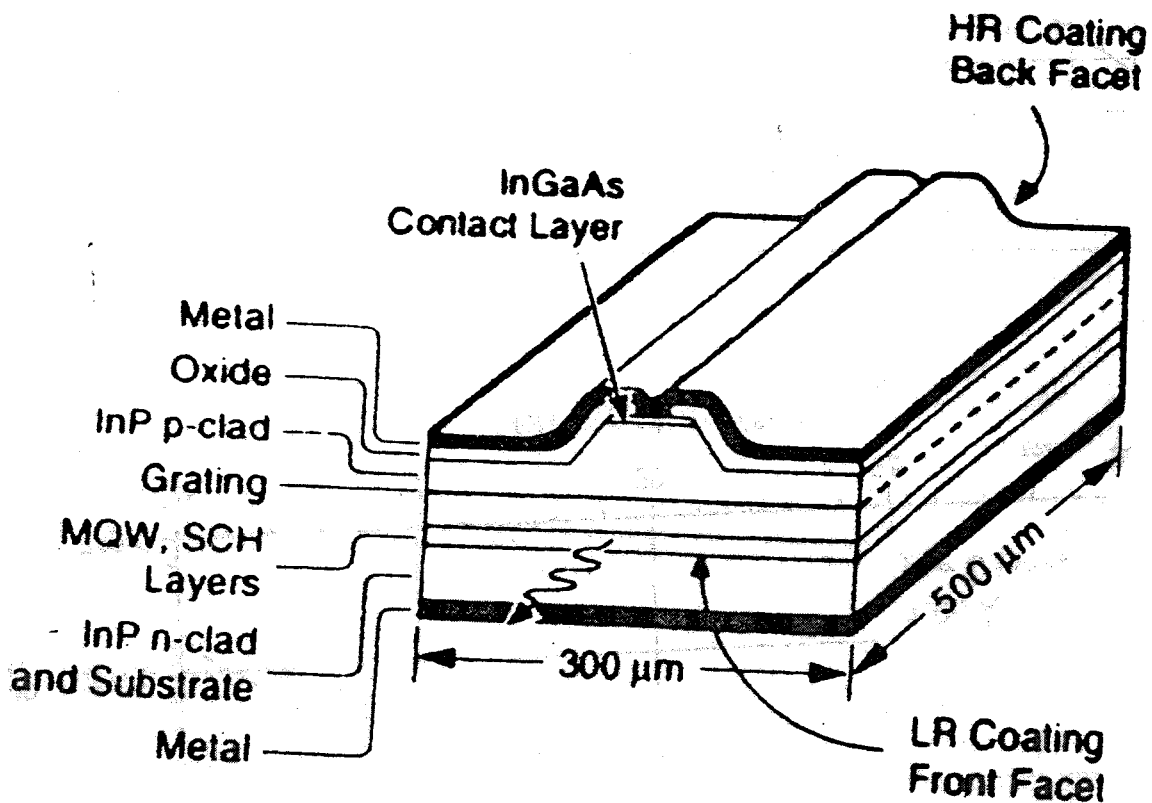
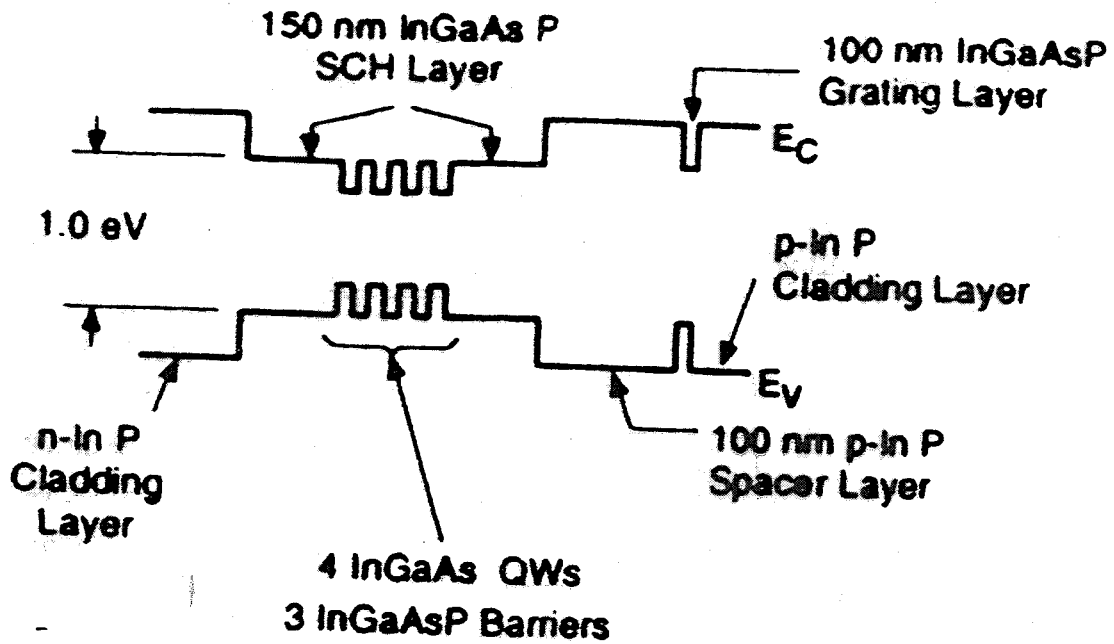


$$\alpha_{\min} = 4.5 \cdot 10^{-7}$$



DFB LASER

30

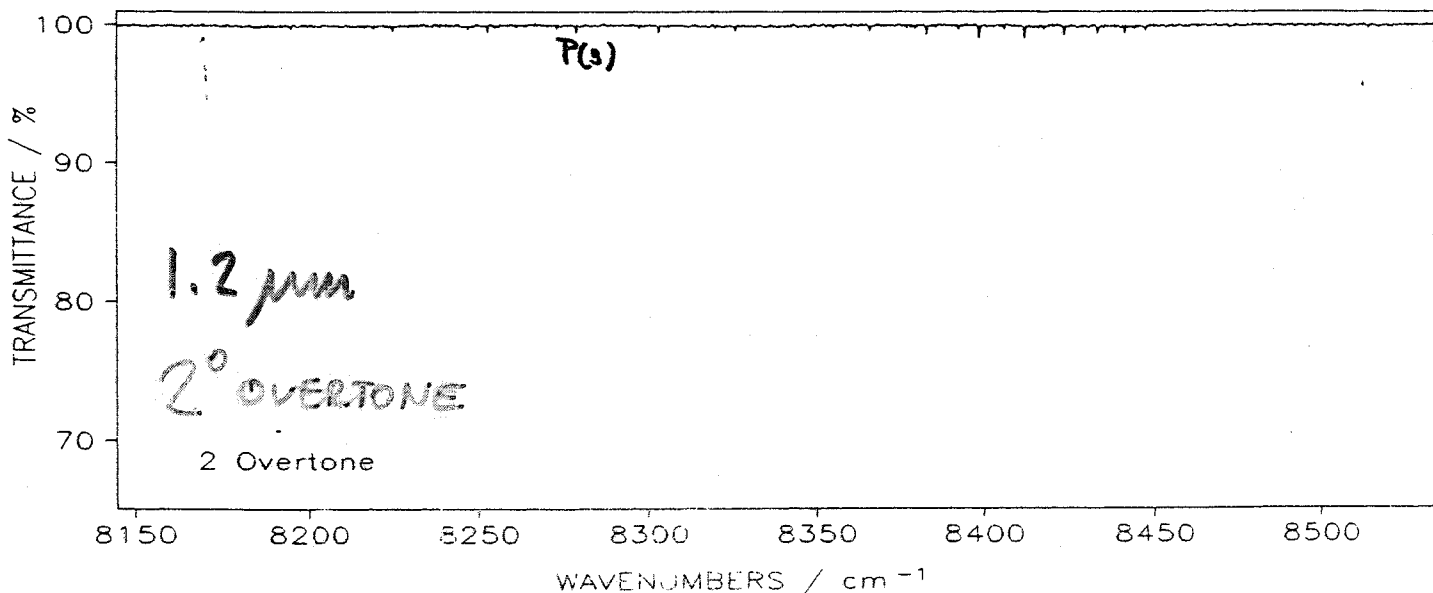
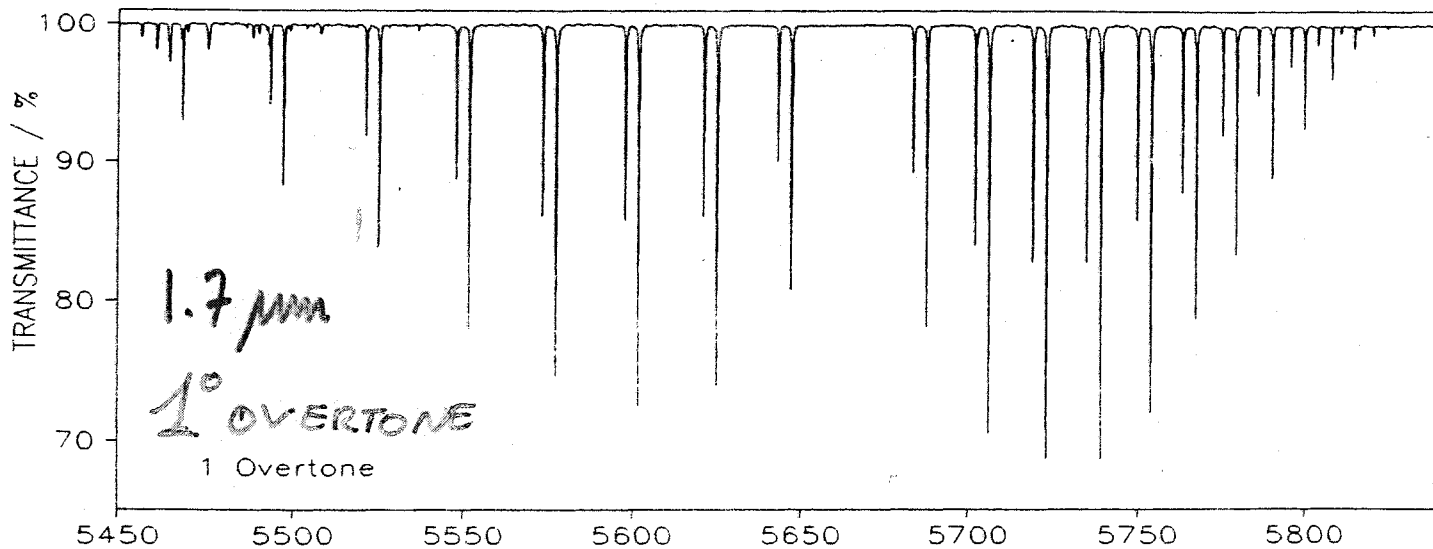


Physical structure and energy band diagram of near-infrared DFB lasers.

REF: D.E. COOPER, R.U. MARTINELLI,
LASER FOCUS WORLD, NOVEMBER 1992

HCl Fourier Transform Spectra

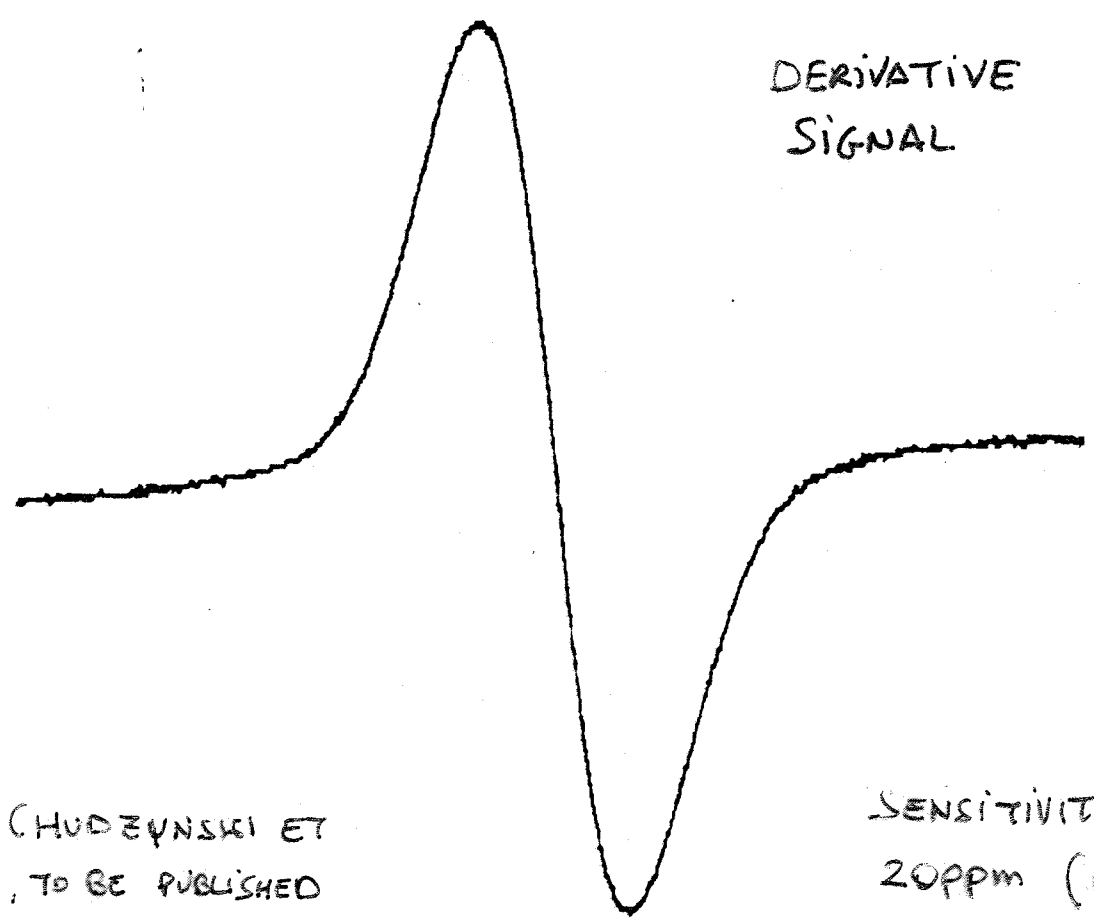
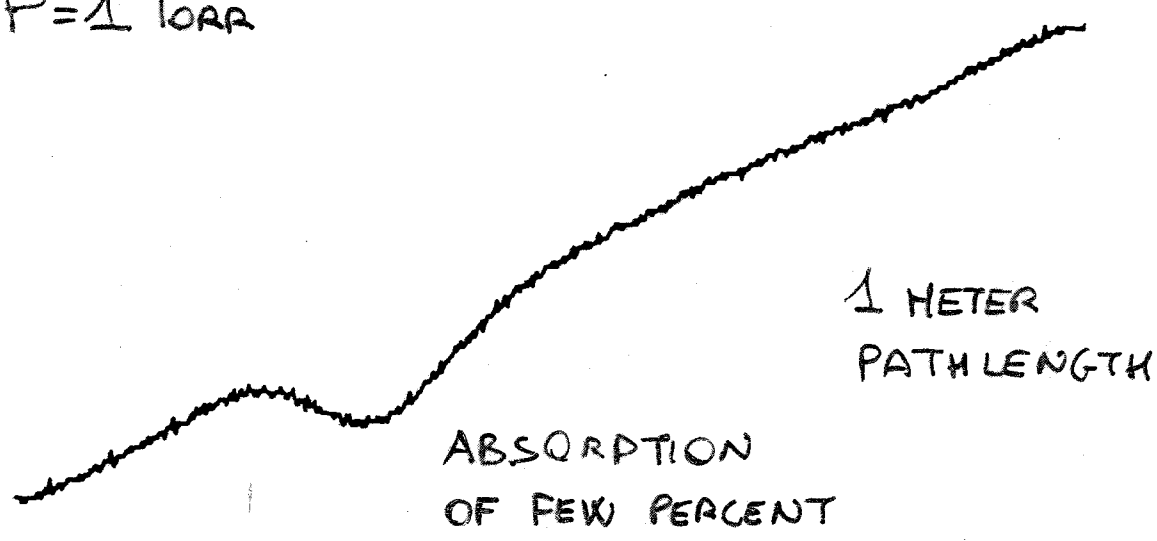
HCl



HCl
2nd OVERTONE
AT 1.2 μm

P(3) COMPONENT

P=1 TORR



S. CHUDZYNSKI ET AL., TO BE PUBLISHED

SENSITIVITY REACHED:
20PPM (PURE HCl)

500 MHz

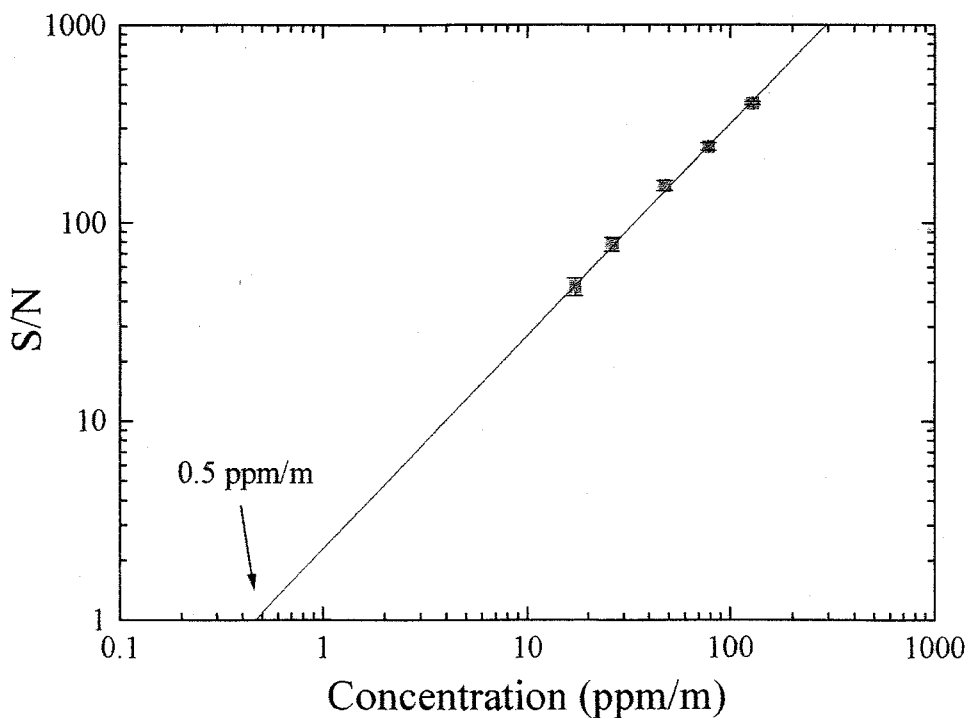
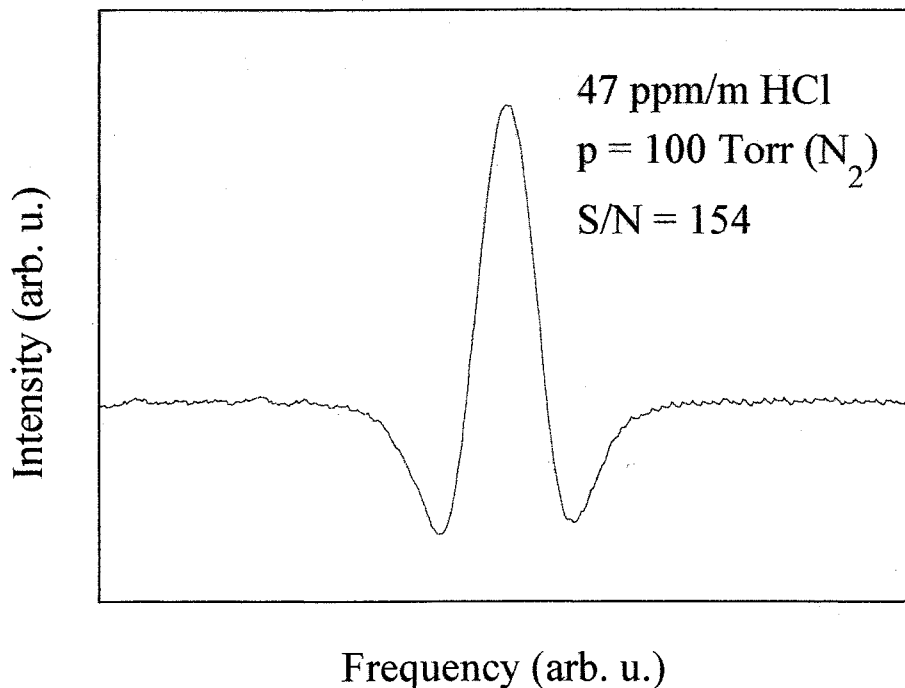
Detection of HCl on the first and second overtones at 1.7 μm and 1.2 μm using semiconductor diode lasers

C. Corsi¹, S. Czudzynsky², F. D'Amato³, M. De Rosa⁴, K. Ernst²

LENS-INFM

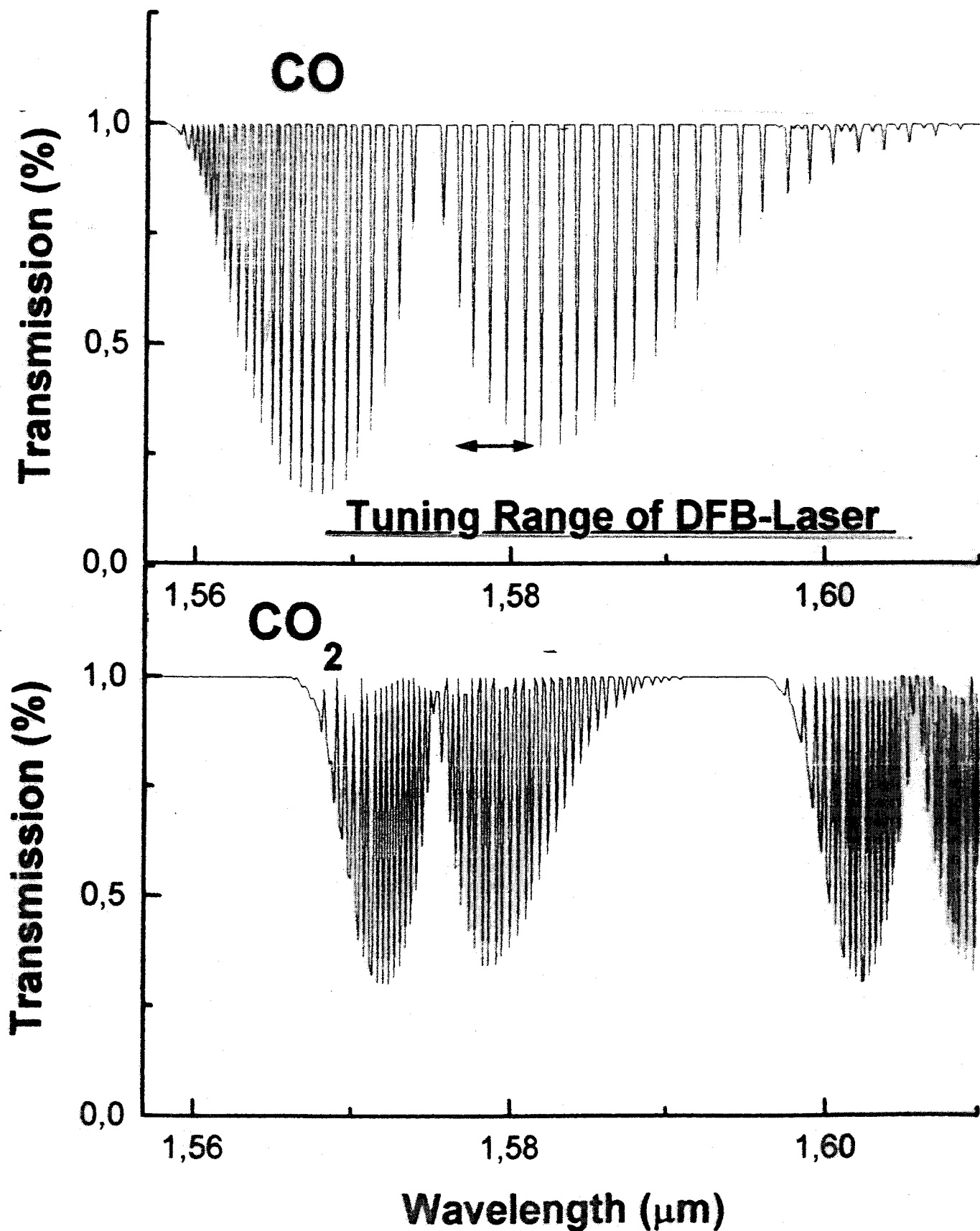
¹ Dip. di Neuroscienze, Univ. Firenze; ² Dip. di Fisica, Univ. Warsaw;

³ ENEA, Frascati; ⁴ SIT, Firenze



FOURIER TRANSFORM

Overtone Spectra of CO and CO₂ in the 1.5 μ m region



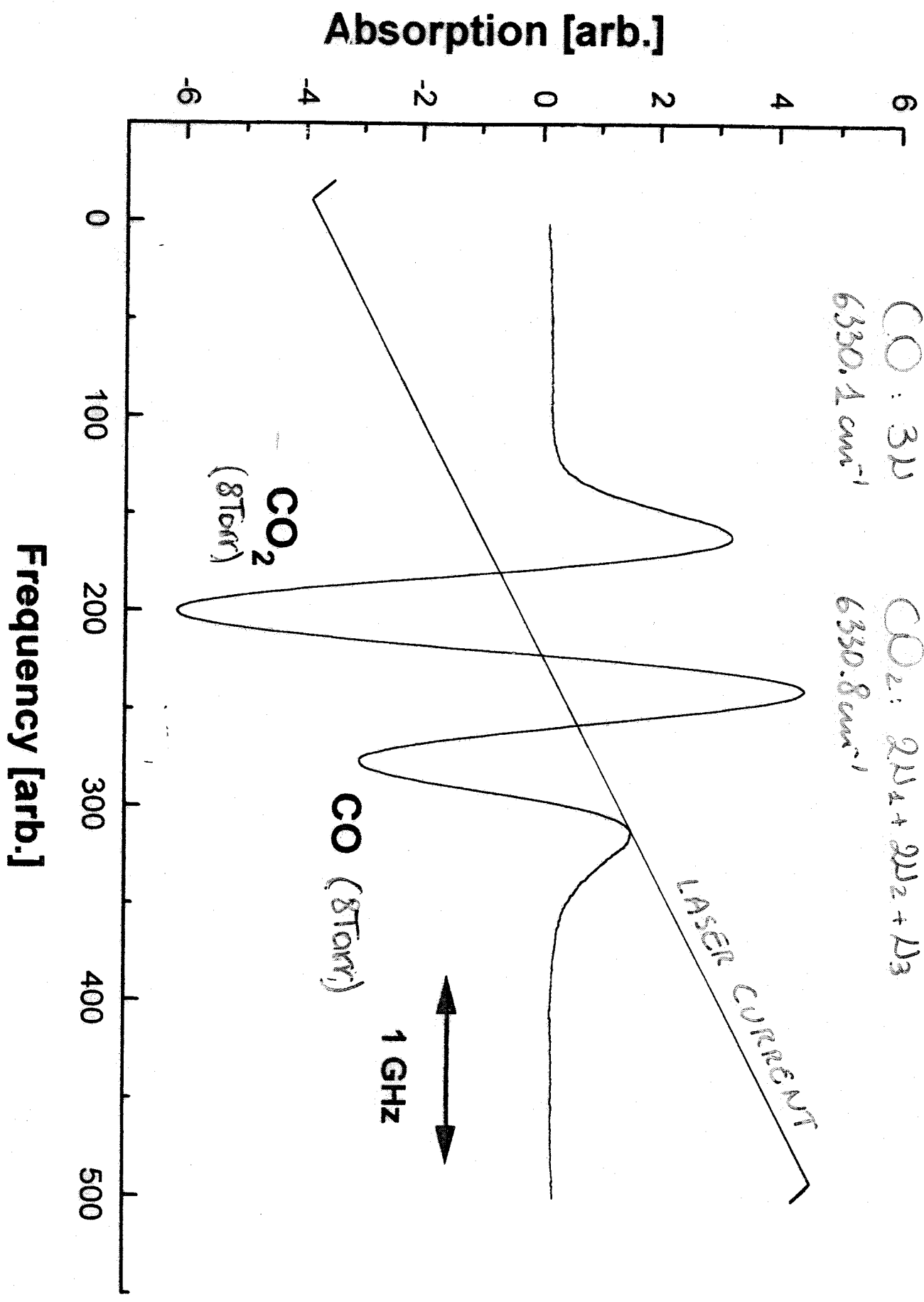
SIMULTANEOUS DETECTION BY ONE SINGLE SCAN

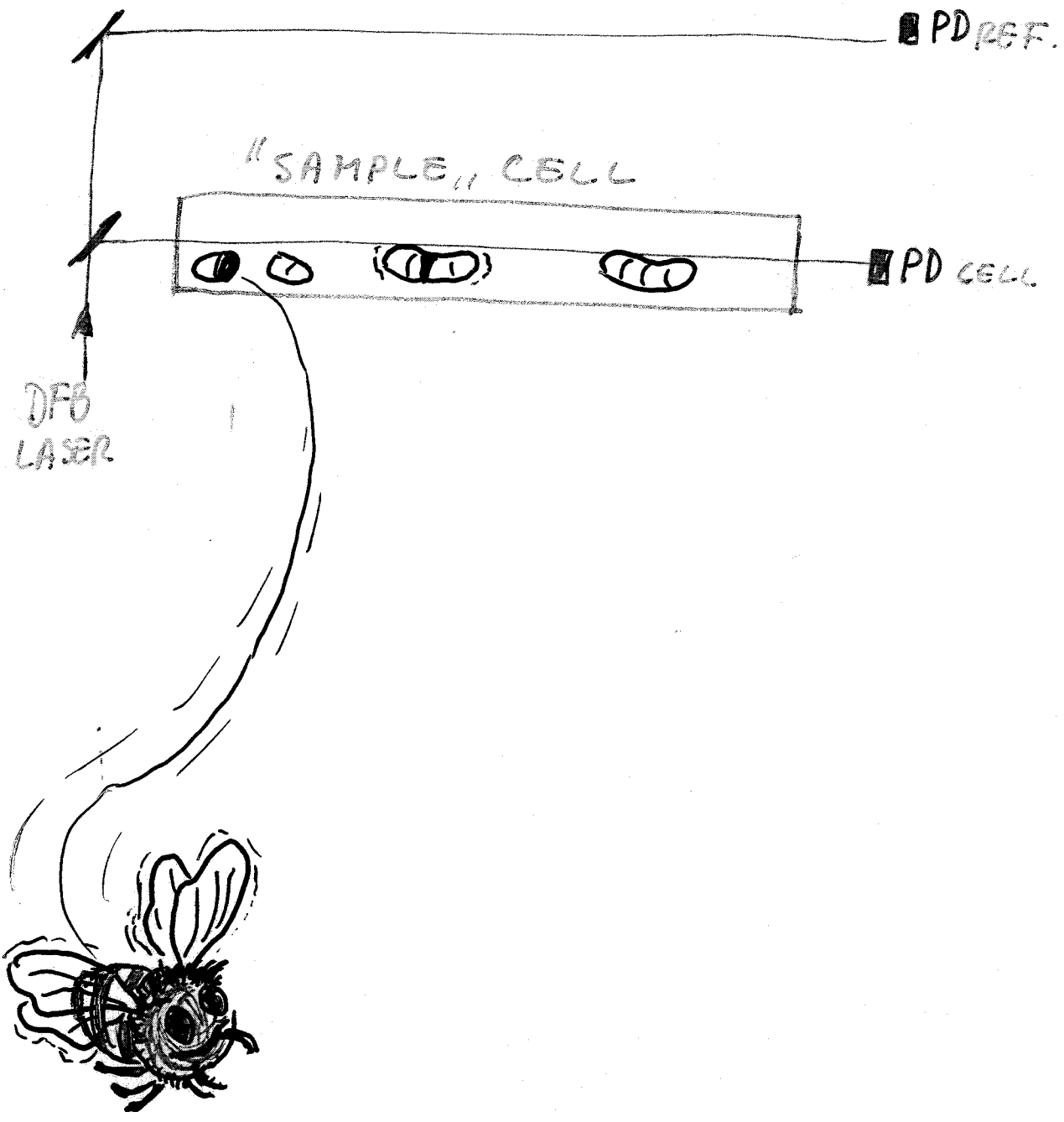
1 mt absorption cell

1578 mm

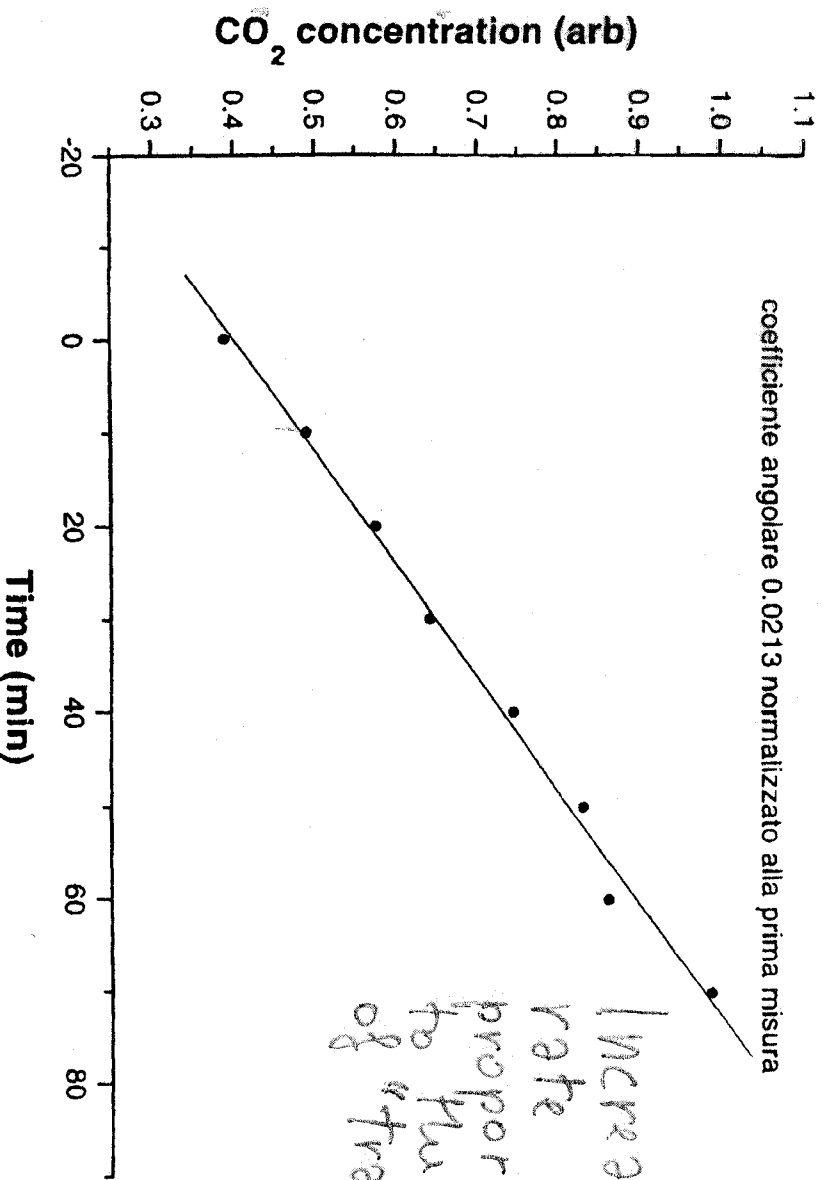
LGJ - FIDELITY

CO : 3D
6330.1 cm⁻¹
CO₂ : 2J₁ + 2J₂ + J₃
6330.8 cm⁻¹

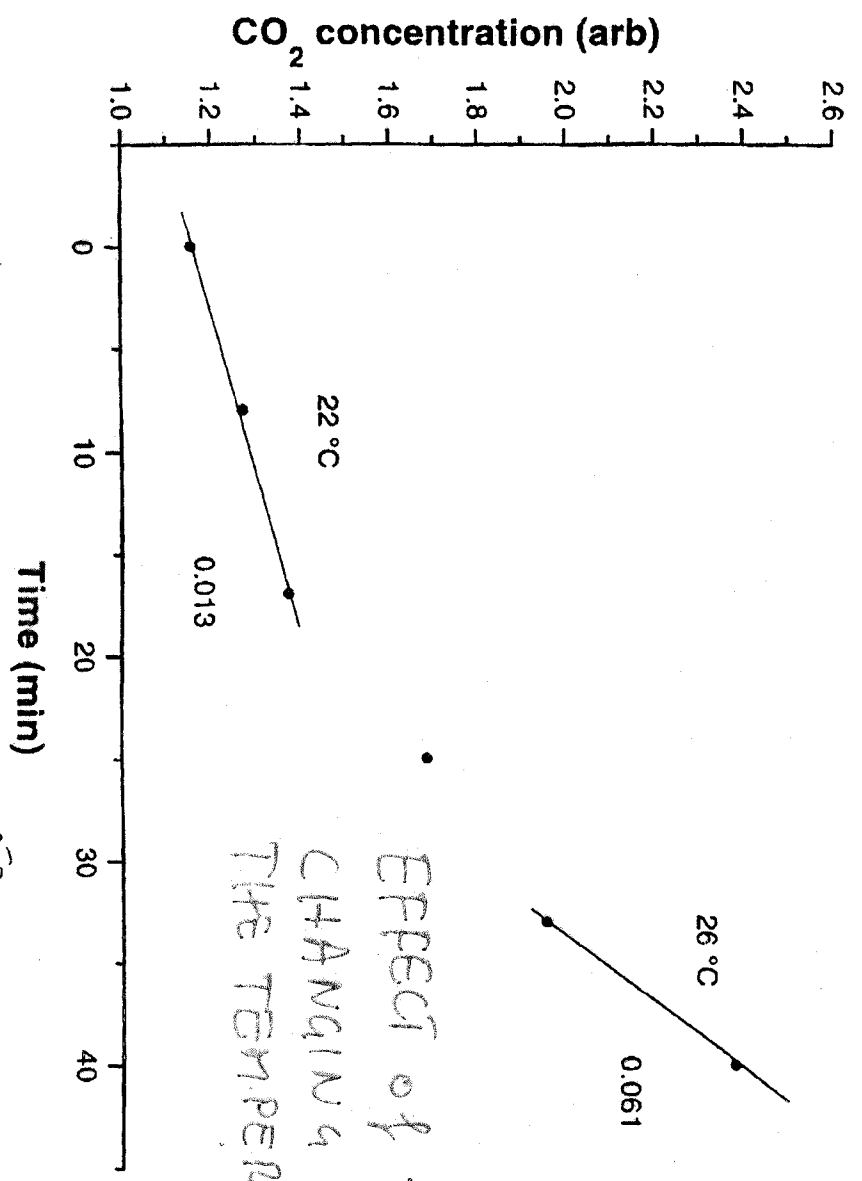




(4)



(5)



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Why using diode lasers for monitoring gases?

- fast response time (ms)

→ DETECTION OF HAZARDOUS GASES

- selectivity

- non-intrusiveness

→ HOSTILE ENVIRONMENTAL CONDITIONS

- sensitivity

→ DETECTION AT PPM LEVEL

- low energy consumption

- low cost

- small-size

- remote sensing

→ FIBER OPTICS

IDEAL FOR
"IN SITU"
MEASUREMENTS

Applications

Industrial Applications

-Control of combustion processes

O₂, CO, HCl

-Detection of hazardous gases

CO, H₂S, CH₄

-Monitoring of exhaust gases

CO, CO₂

-Monitoring of emission gases in power plants

NH₃

Medical Applications

-Non invasive diagnostic of human breath

¹²CO₂, ¹³CO₂, NH₃, C₂H₆

Geophysical Applications

-Monitoring of volcanic gases

H₂S, H₂O, CO₂