

**ICTP/UCSB/TWAS
MINIWORKSHOP ON "FRONTIERS IN MATERIALS SCIENCE"
15 - 18 May 2001**

301/1311-3

*"Structural & Chemical Analysis of Materials by Advanced
Transmission Electron Microscopy Technique"*

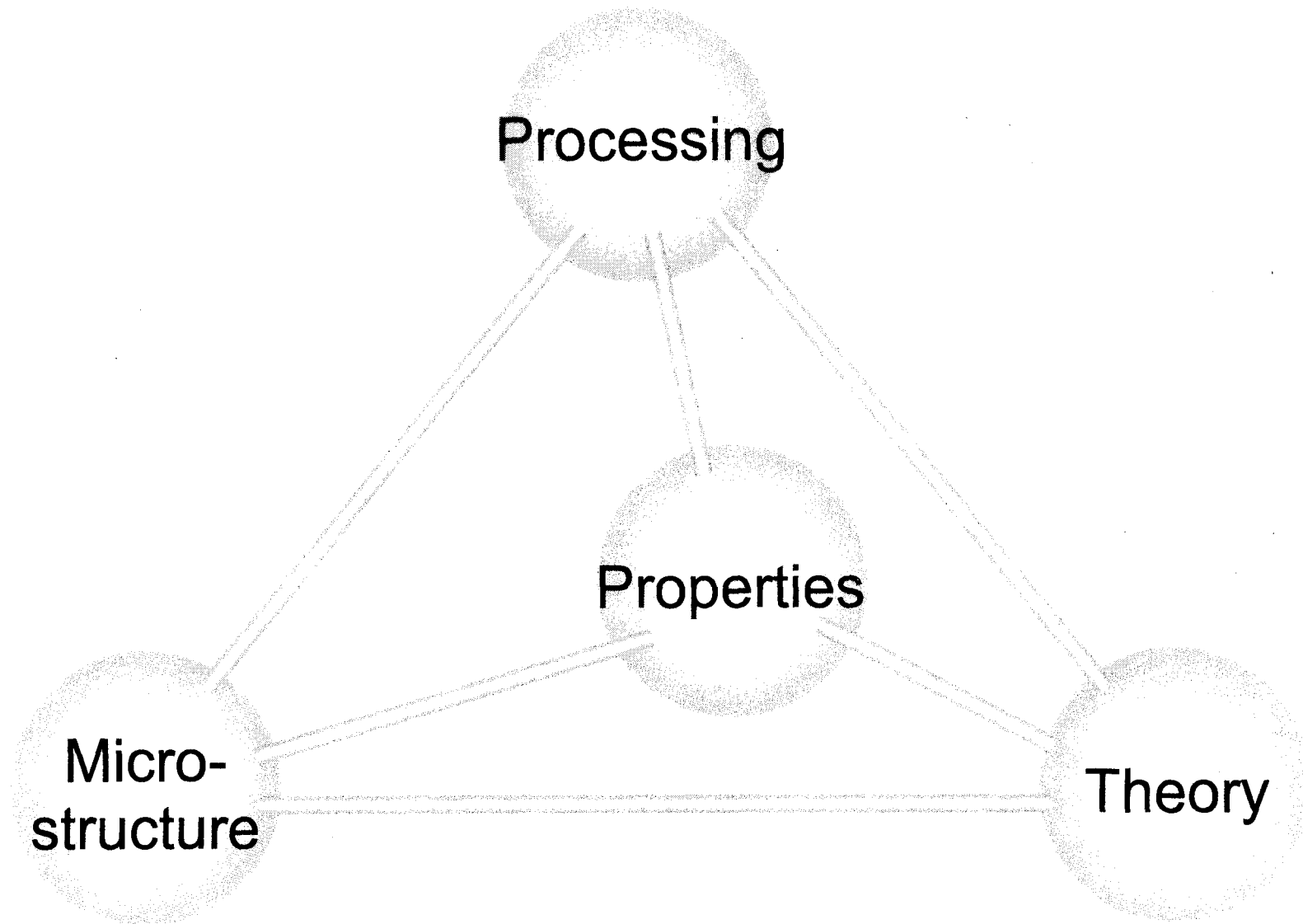
**M. RÜHLE
Max-Planck-Institut für Metallforschung
Stuttgart
Germany**

Please note: These are preliminary notes intended for internal distribution only.

Structural and Chemical Analysis of Materials by Advanced Transmission Electron Microscopy Technique

Manfred Rühle
Max-Planck-Institut für Metallforschung
Seestr. 92, 70714 Stuttgart, Germany

Materials Science



Outline

Introduction: Pathways of microstructural analysis
bridging length scales over many
dimensions

Transmission Electron Microscopy:
Different Techniques
Advantages and Problems

Case Studies

Defects in Metals (Cu, NiAl, ...)
Defects in Ceramics (Al_2O_3 , SrTiO_3 , ...)
Defect at Metal/Ceramic Composites

Nature of Defects

0-dimensional: point defects, point defect cluster
1-dimensional: dislocation
2-dimensional: interfaces
3-dimensional: inclusions

Acknowledgement

K. van Benthem, S. Krämer, C. Scheu, W. Sigle

Specimen preparation: W. Kurtz, T. Wagner

TEM specimen preparation: U. Salzberger, A. Strecker, M.
Sycha

Financial Support

DFG (German Science Foundation)

Volkswagen Foundation

State of Baden-Württemberg

Max Planck Society

Microstructural Parameters

Phases present in materials		X-ray techniques, microprobe
Grain sizes	}	{
Grain size distribution		
Grain shape		
Misorientation of grains ("special" boundaries)		OIM, SAD (in TEM)
Composition of regions close to the GB		Q-AEM
Atomic structure of specific GBs		Q-HREM
Bonding		ELNES: fingerprints and quantitative analysis

Microscopy of Materials:

Many length scales!

Optical Microscopy

- Conventional Optical Microscopy
- Laser-Scanning Microscopy
- Near Field Optical Microscopy
- Video Microscopy

Electron Microscopy

- Scanning Electron Microscopy
 - Imaging
 - Chemical Analysis

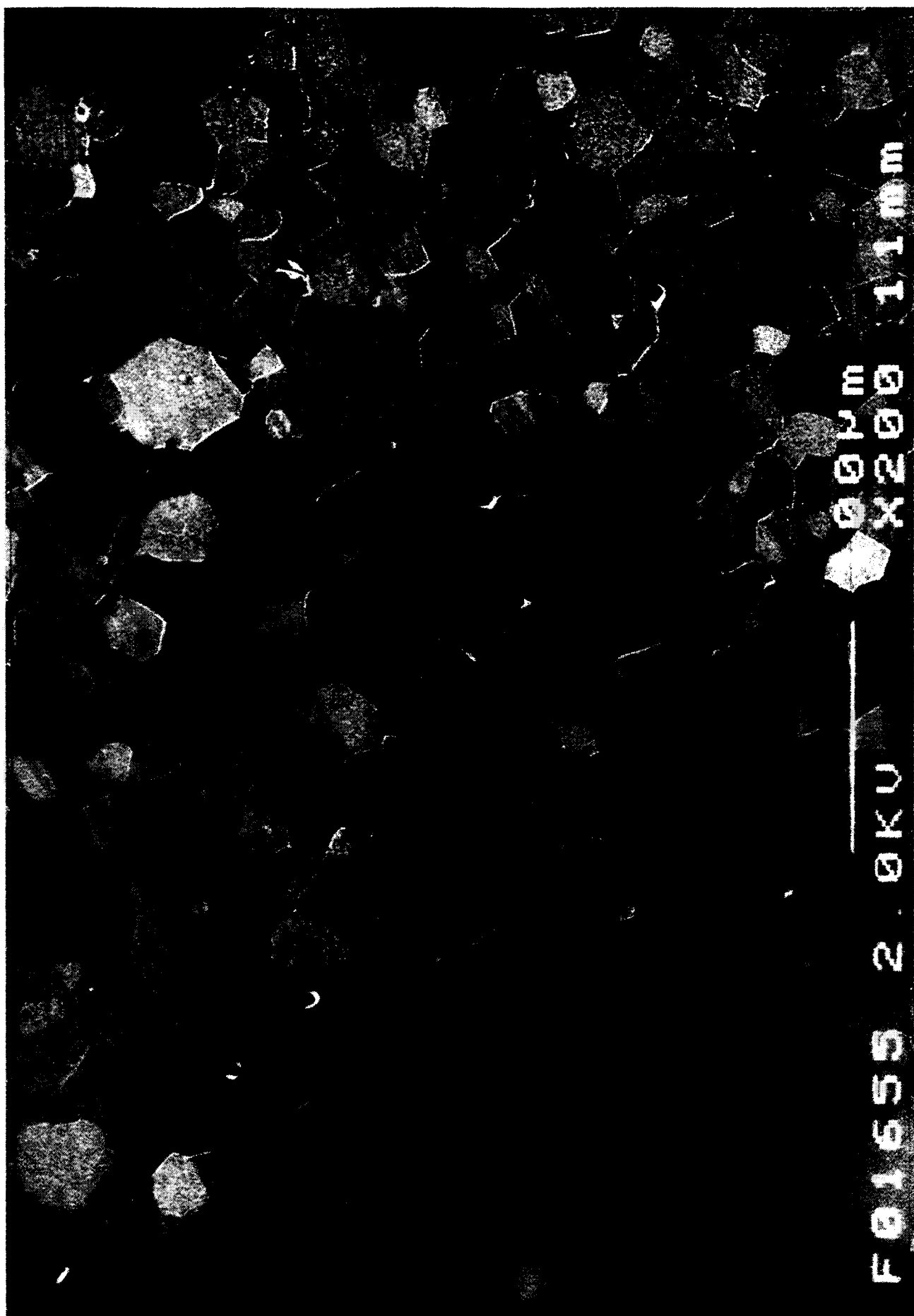
Transmission Electron Microscopy

- Diffraction Studies
- Diffraction Contrast
- Atomic-Resolution Microscopy
- Analytical Studies
 - Energy Dispersive Studies
 - Electron Energy Loss Spectroscopy
 - Electron Loss Near Edge Structure

Atom Probe Field-Ion Microscopy

Probe Techniques:

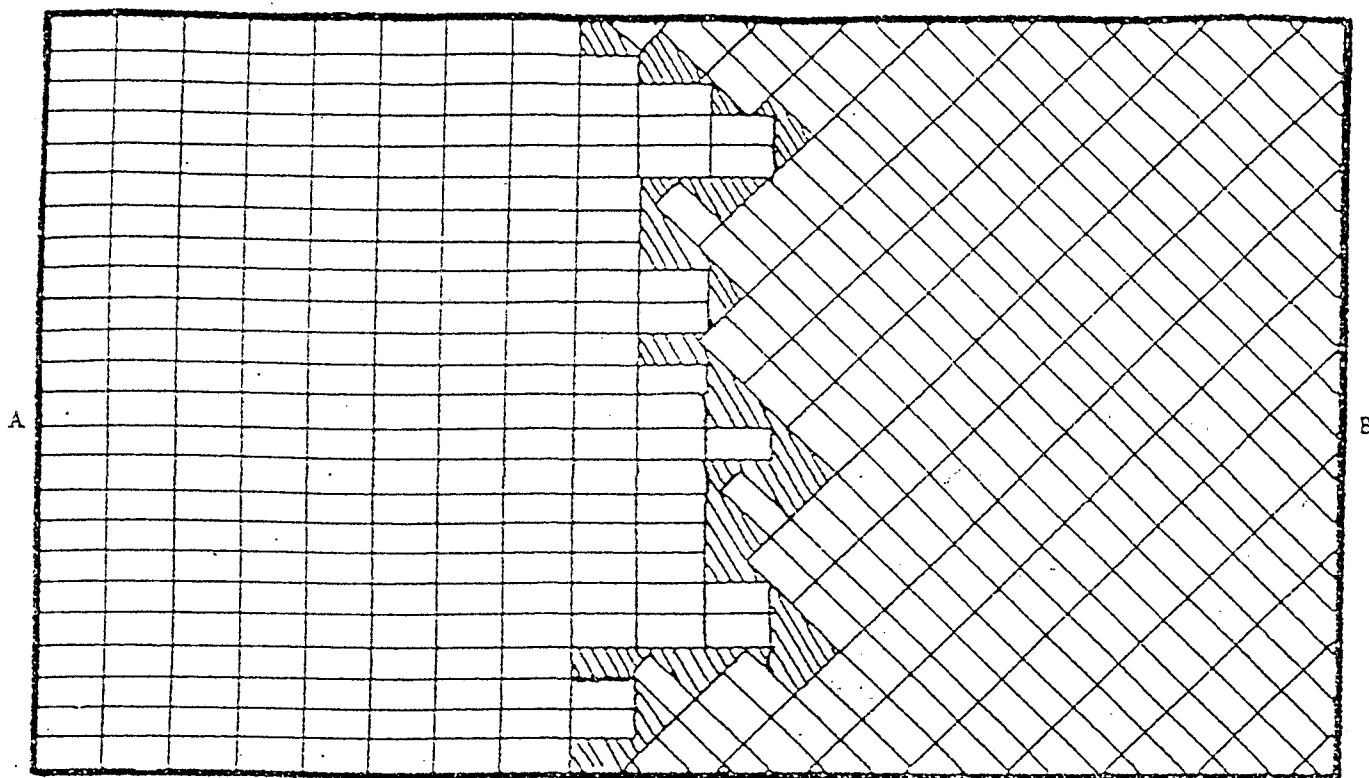
- Scanning Tunneling Microscopy (STM)
 - Scanning Tunneling Spectroscopy
- Atomic Force Microscopy (AFM)



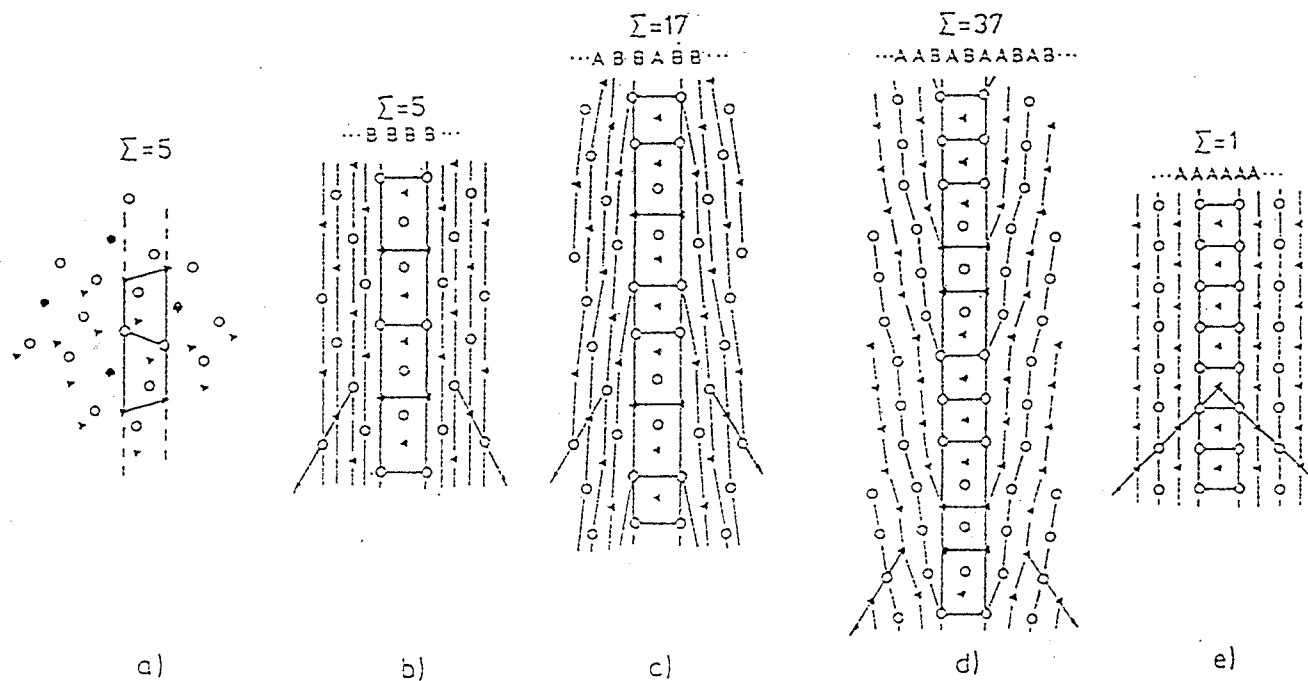
F01655 2 0KV

00200

11mm



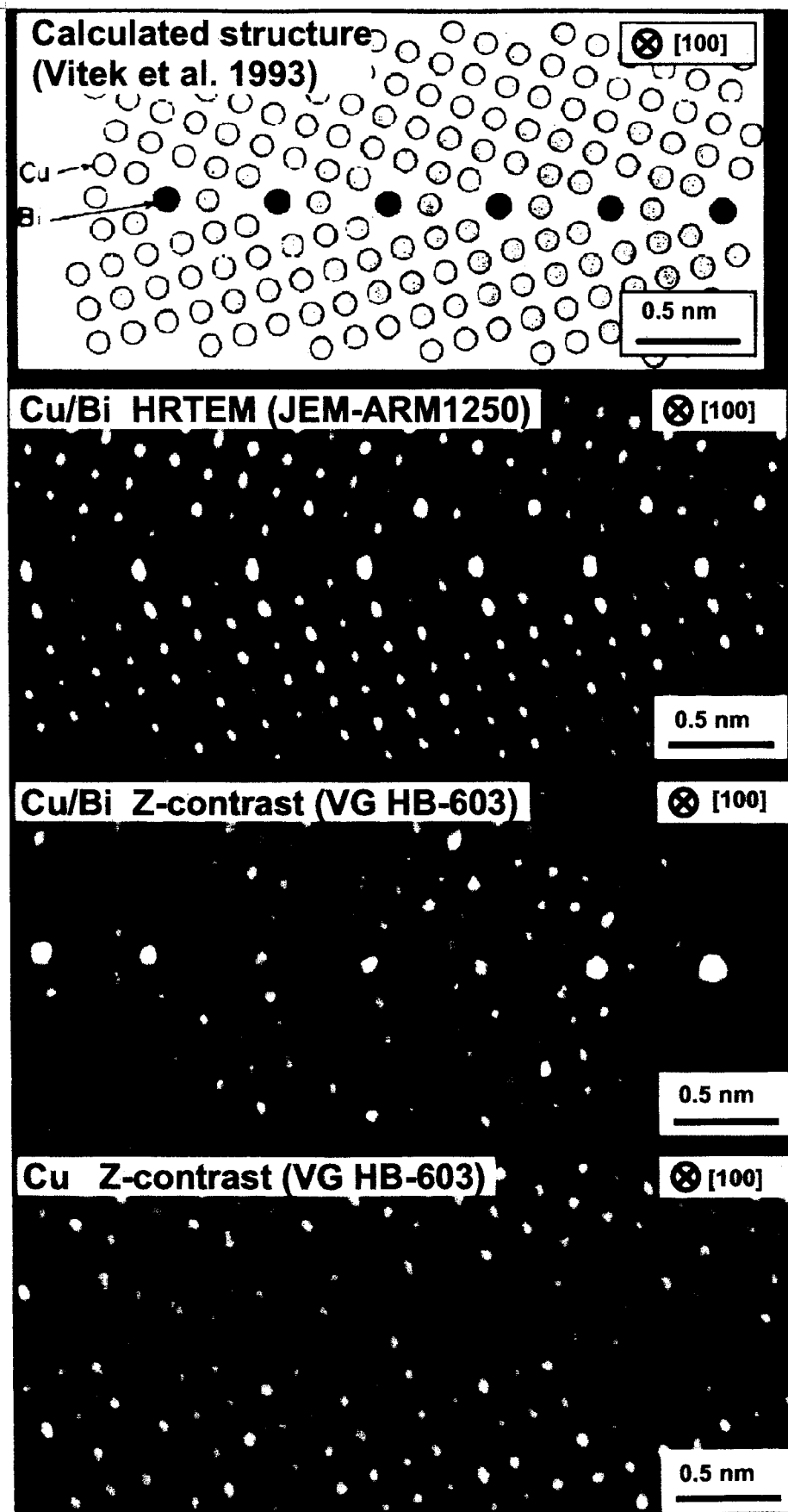
Rosenhain, Ewen. 1912



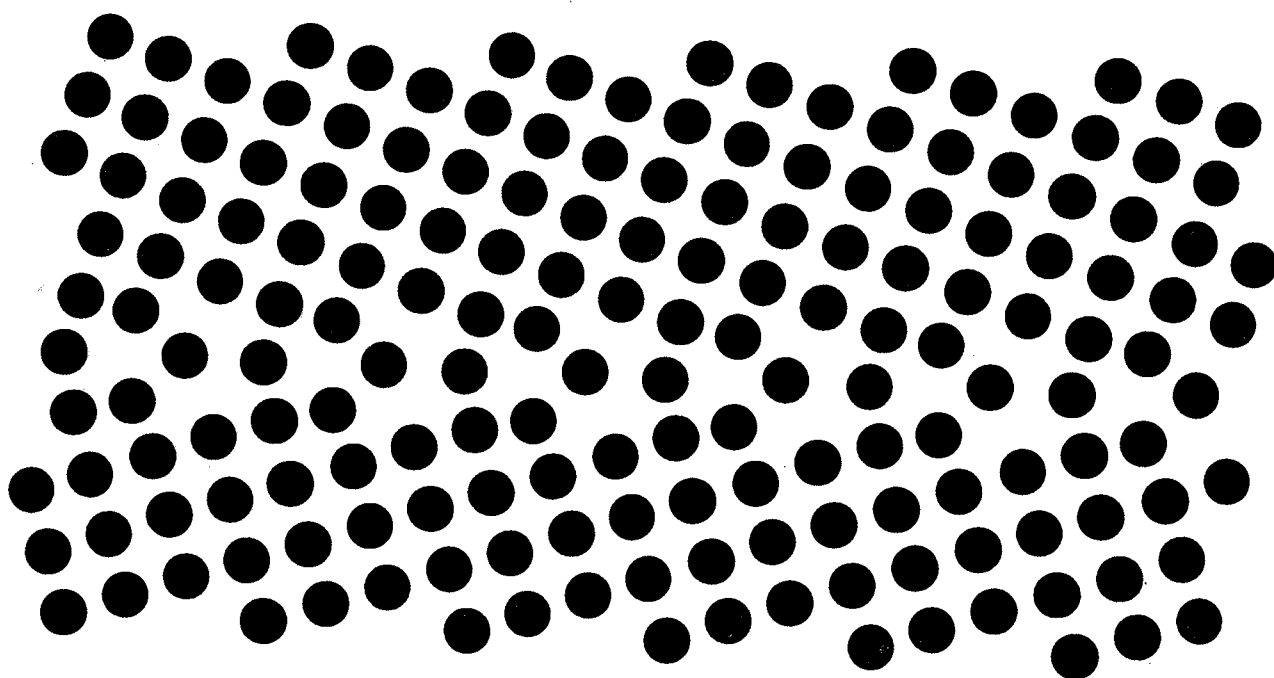
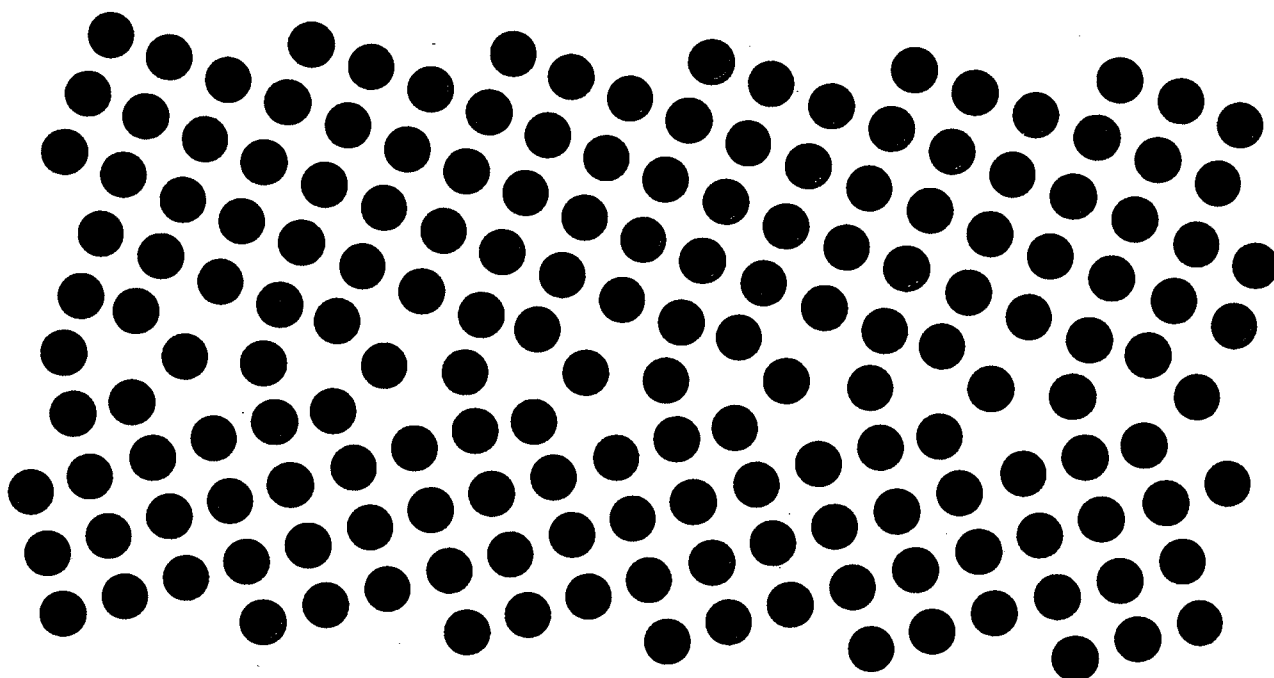
Sutton, Vitek 1983



15KV X2000 9187 1989 GU HPI/M



← (C)



Transmission Electron Microscopy

Conventional TEM

BF/DF/SAD

- Morphology
- Phase Distribution
- Defect Analysis
- *in situ*-Experiments
 - heating
 - cooling
 - deformation

Spectroscopy and CBED

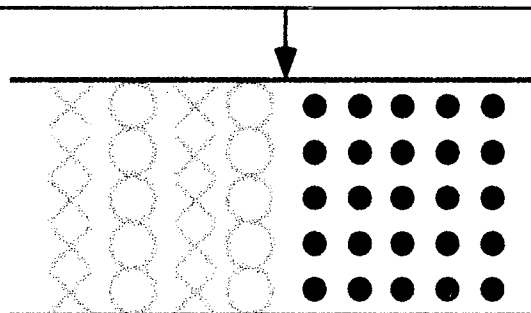
(high spatial resolution)

- EDS
 - compositions
 - gradients
- EELS
(EXELFS, ELNES)
 - compositions
 - gradients
 - electronic states

Lattice Imaging

HREM

- Structure of materials
 - oxides
 - HT superconductors
- Structure of defects
 - interfaces
 - dislocations



Information on chemistry and bonding

Complicated evaluation:
lens aberration of HREM
scattering within specimen

Experimental studies

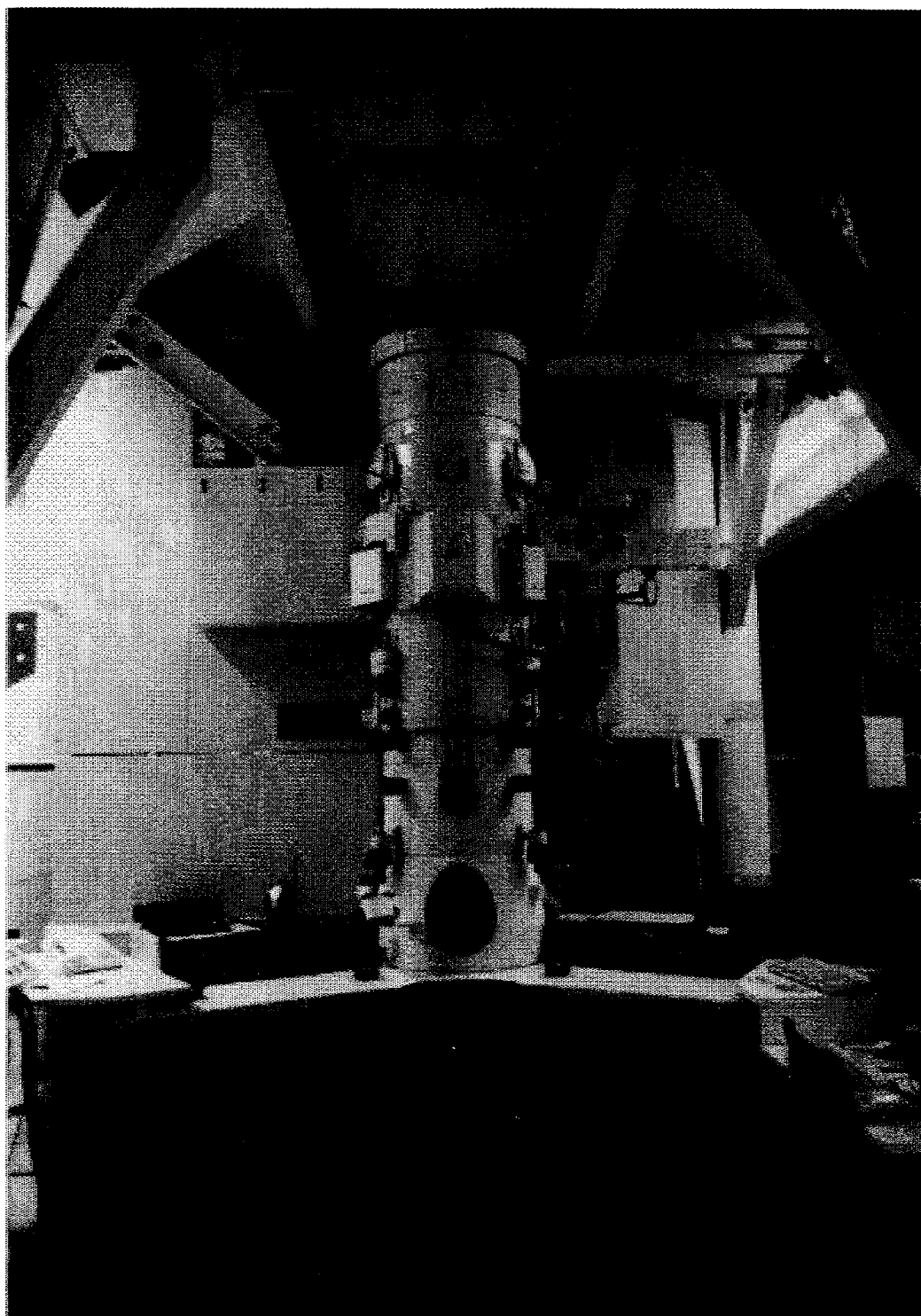


Image simulations

point-to-point 1.0 Å
information limit 0.8 Å } 1250 kV

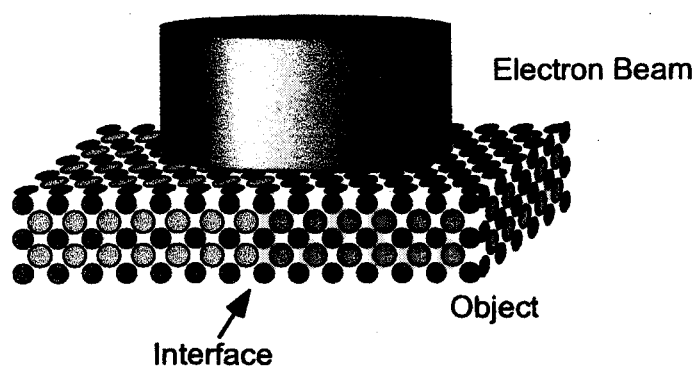
STUTTGART
"ATOMIC RESOLUTION MICROSCOPE"

JEM-ARM1250

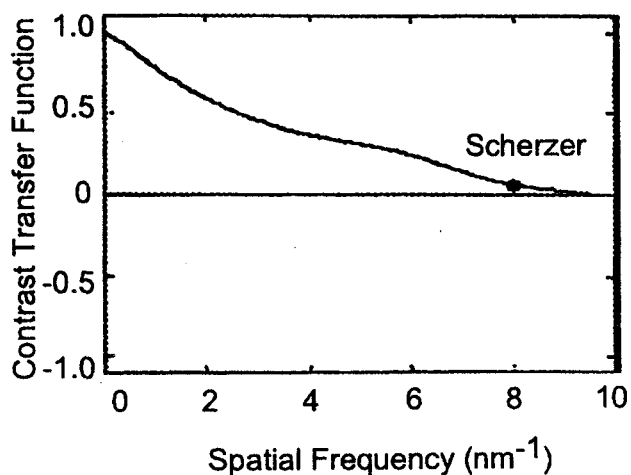
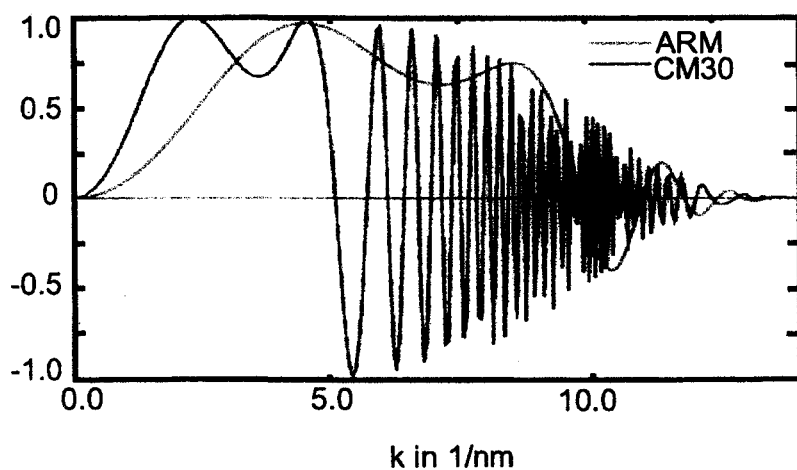
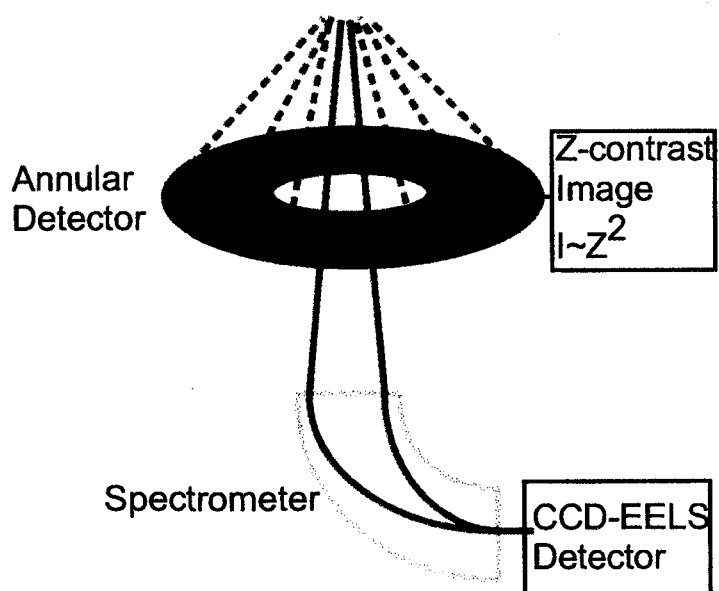
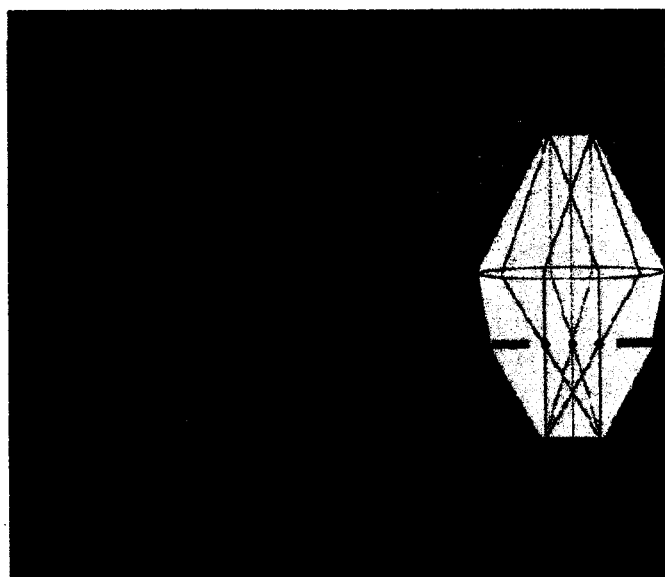
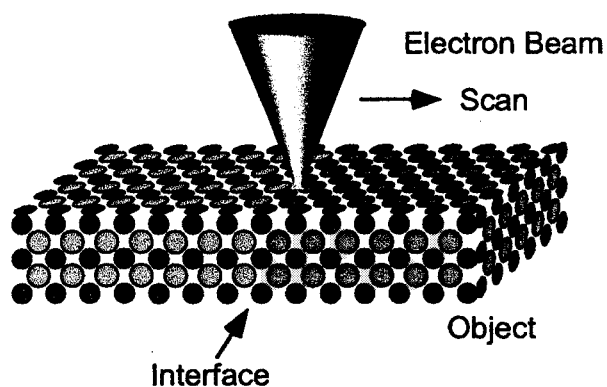


High-Resolution Transmission Electron Microscopy

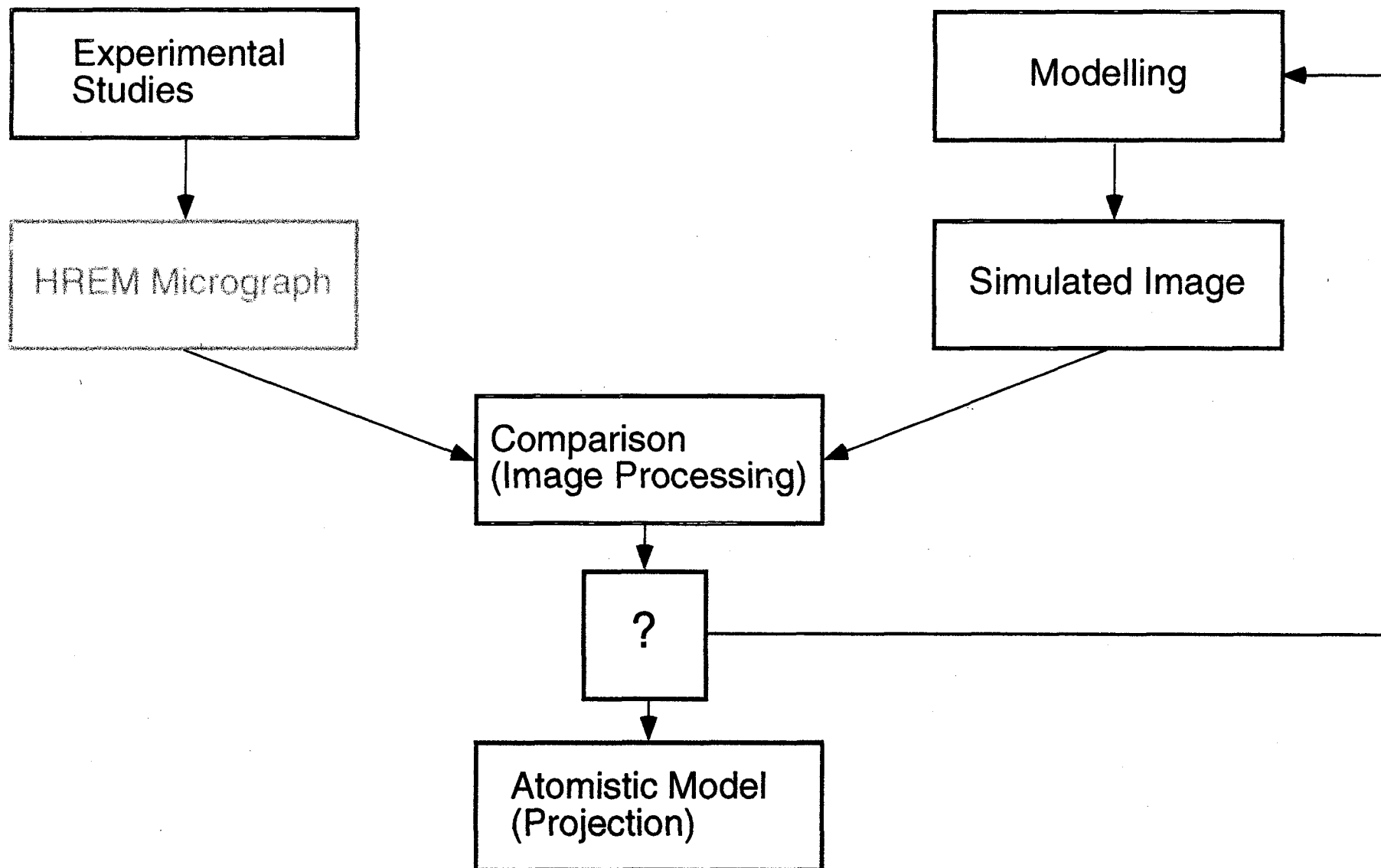
coherent imaging



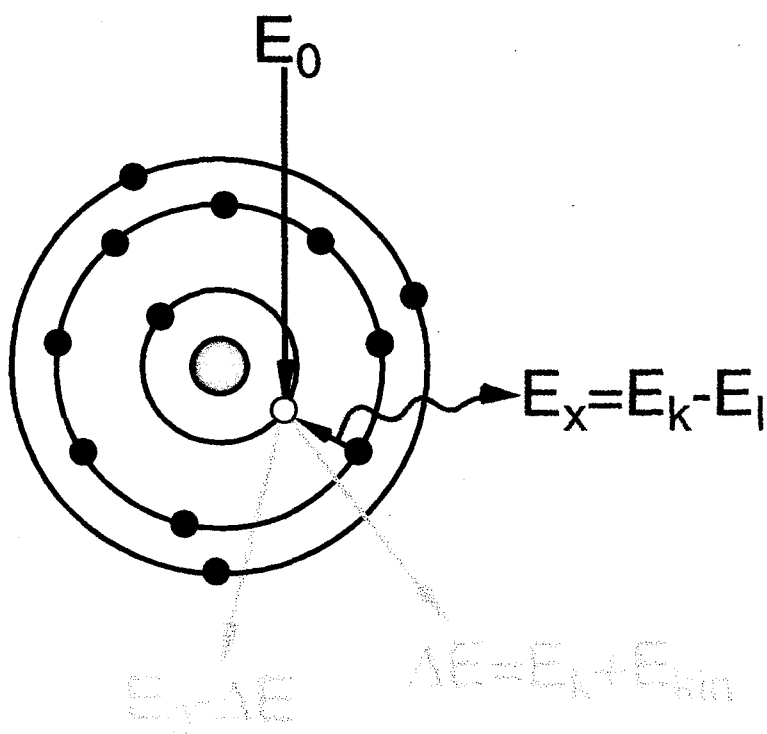
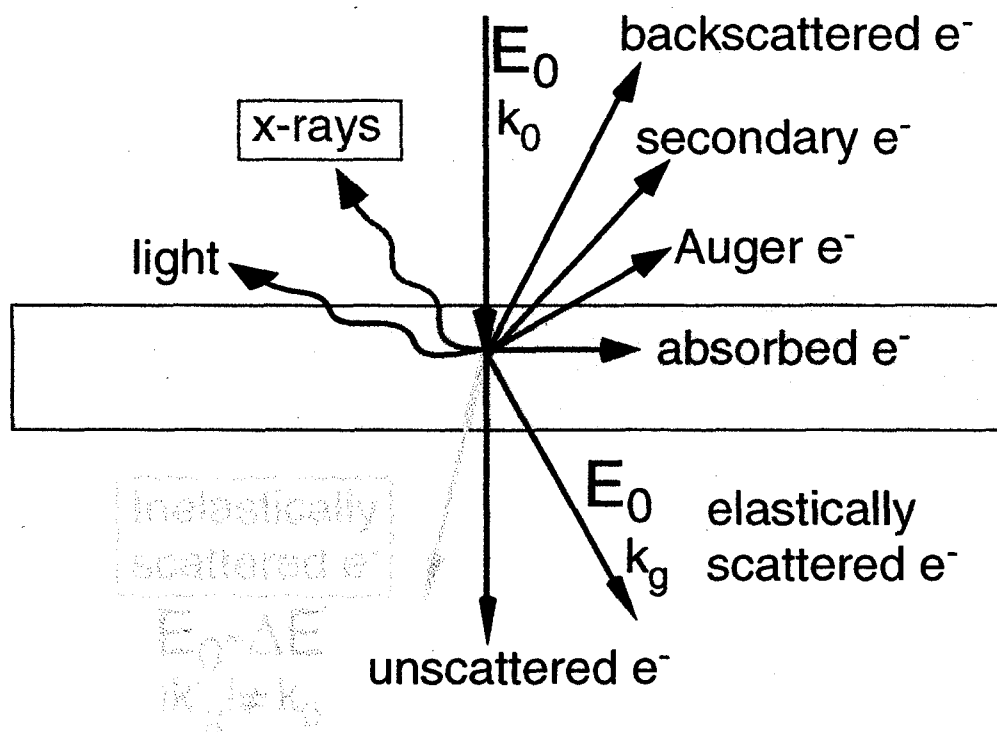
incoherent imaging
(Z-contrast)



Quantitative High-Resolution Electron Microscopy



Elastic and Inelastic Scattering Process



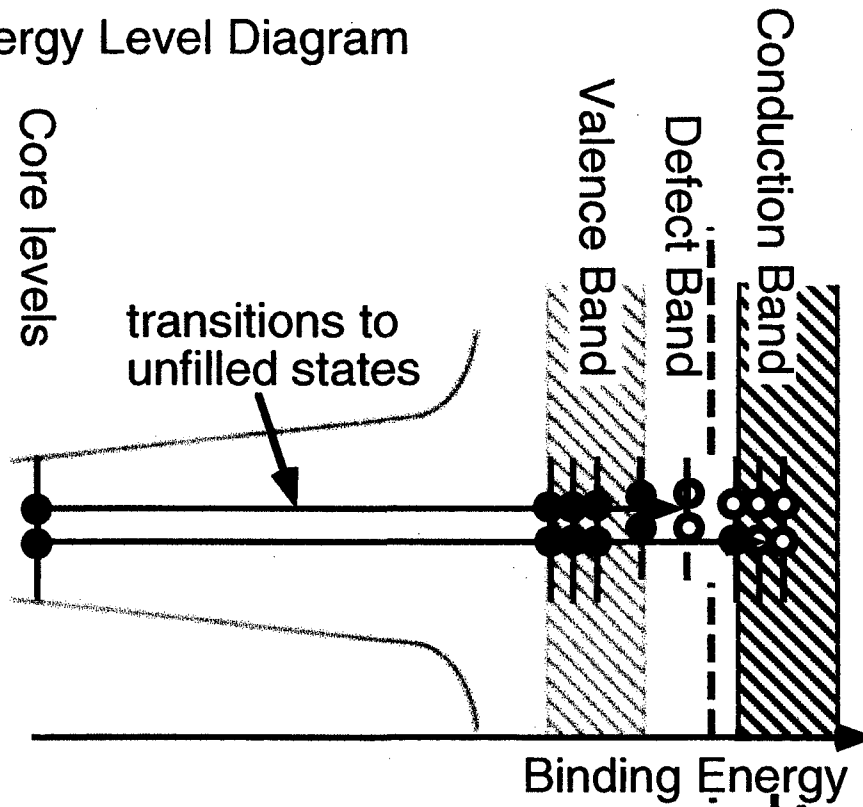
EDS
Energy Dispersive
Spectroscopy

EELS
Electron Energy
Loss Spectroscopy

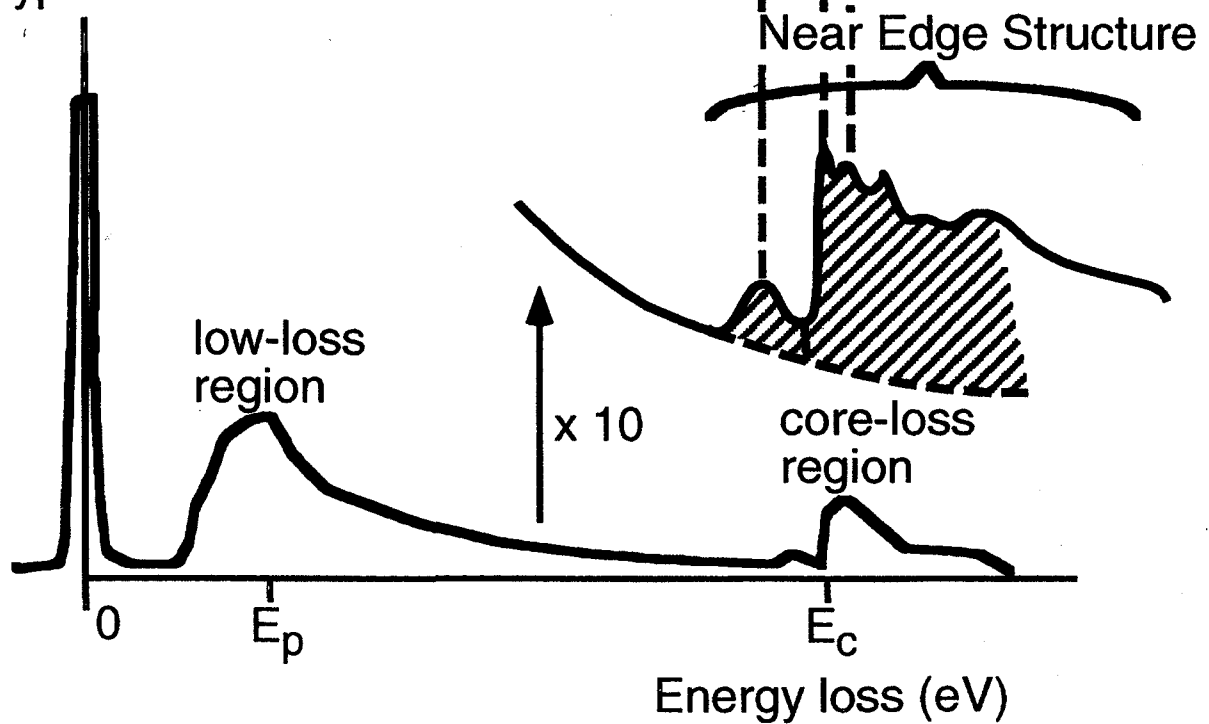
ELNES

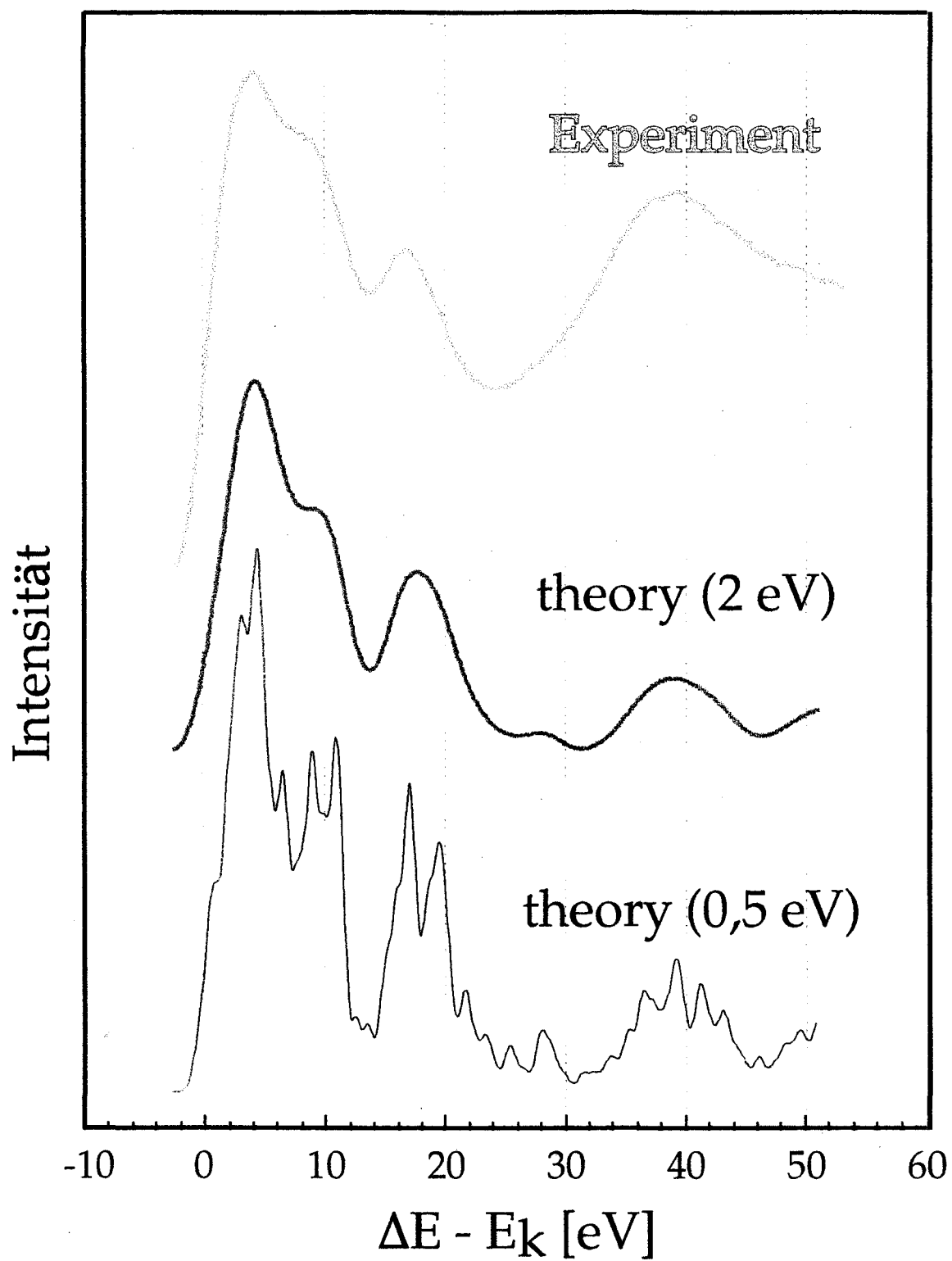
(electron energy-loss near edge structure)

Energy Level Diagram

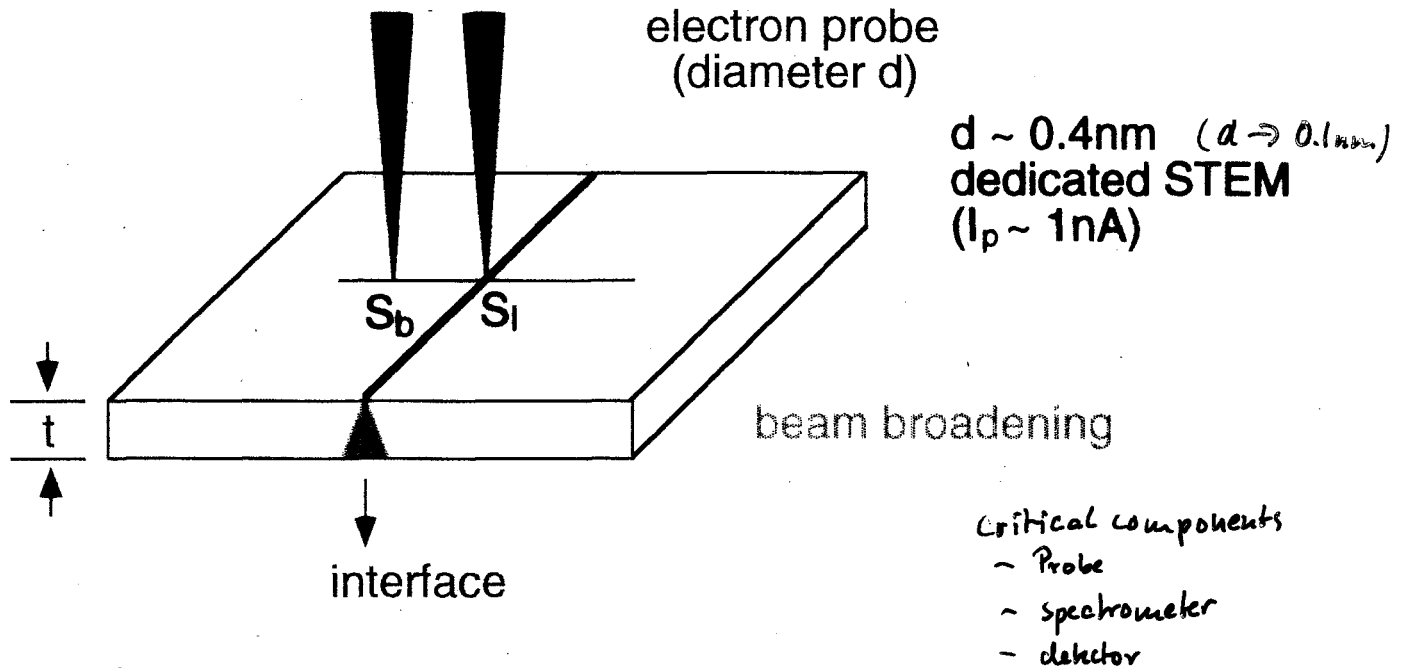


Typical EELS





Analytical Electron Microscopy with High Spatial Resolution



Comparison of S_b and S_l : (EDS and EELS)

- Segregation
- ELNES $\rightarrow$ information on local bandstructures

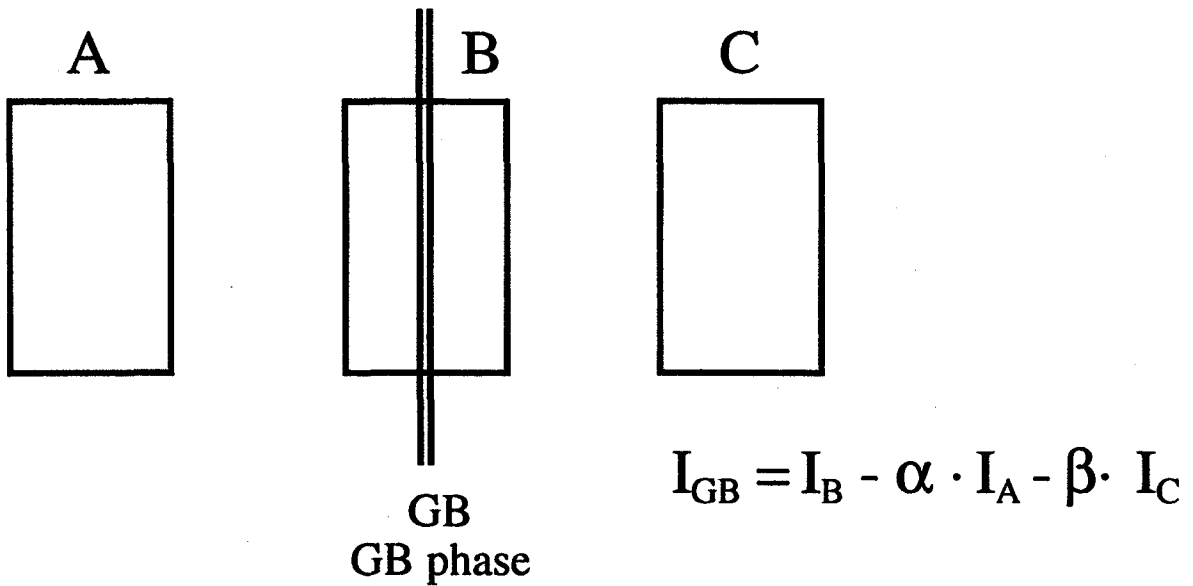
SREELS : Spatially Resolved

Special techniques (with dedicated instruments)

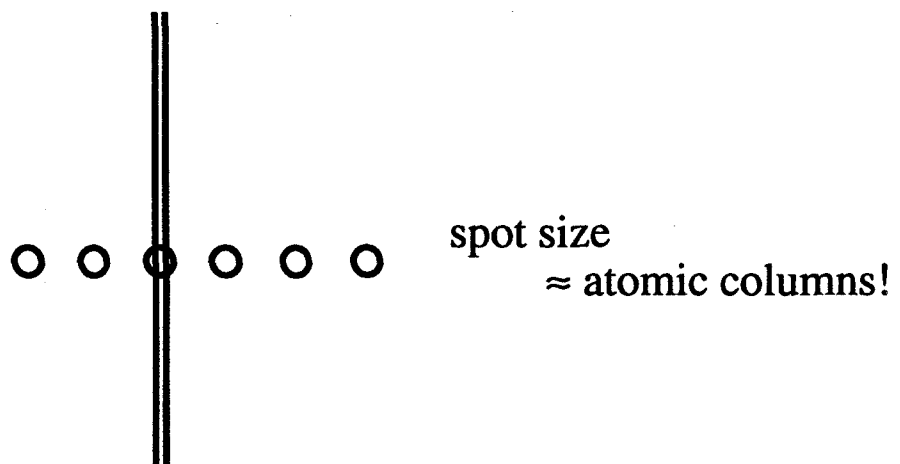
space charges at interfaces

EELS (ELNES) of Grain Boundaries

Spatial difference technique



spot mode



Spatial difference technique

- information on composition
- information on ELNES at interfaces

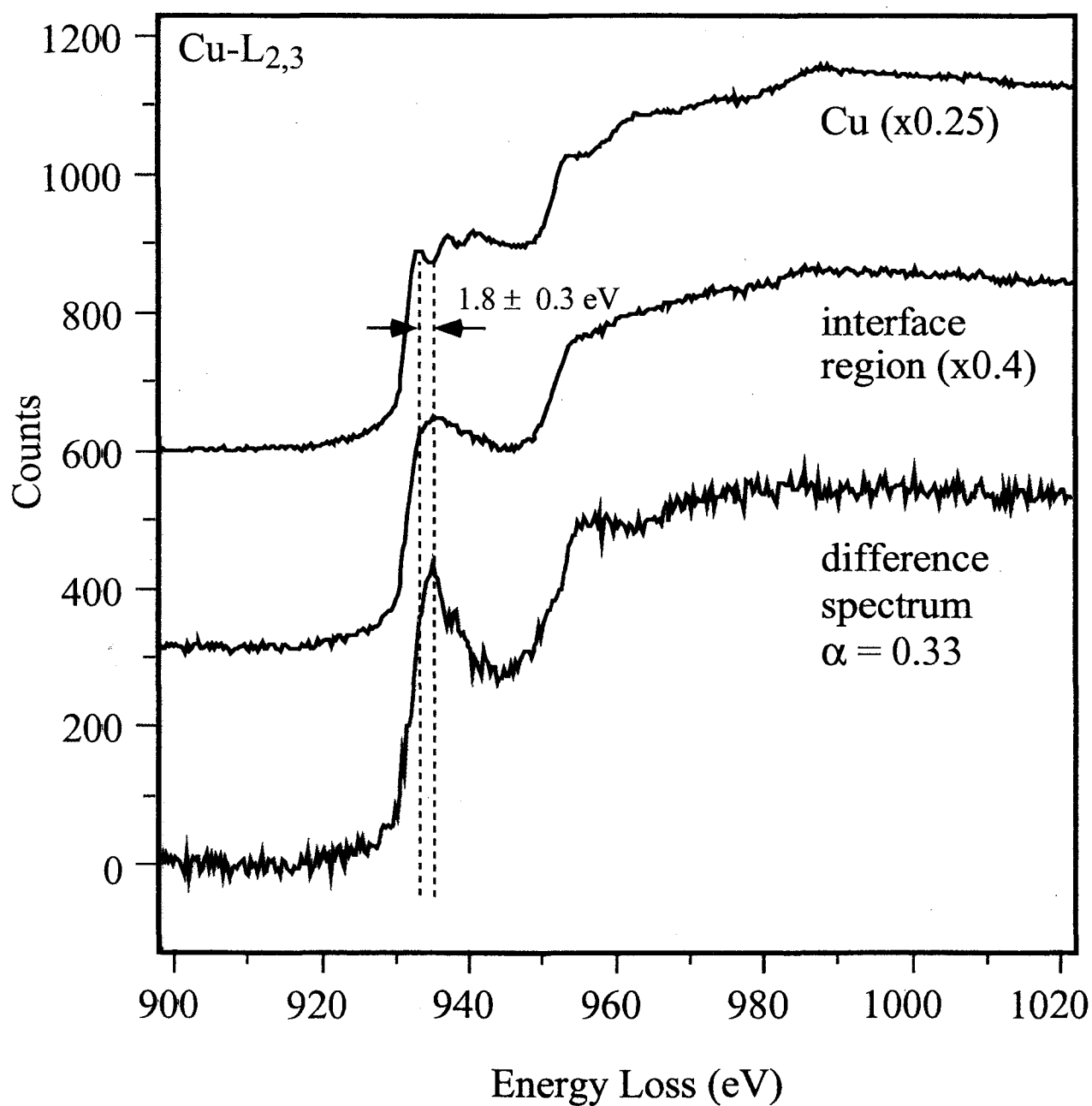
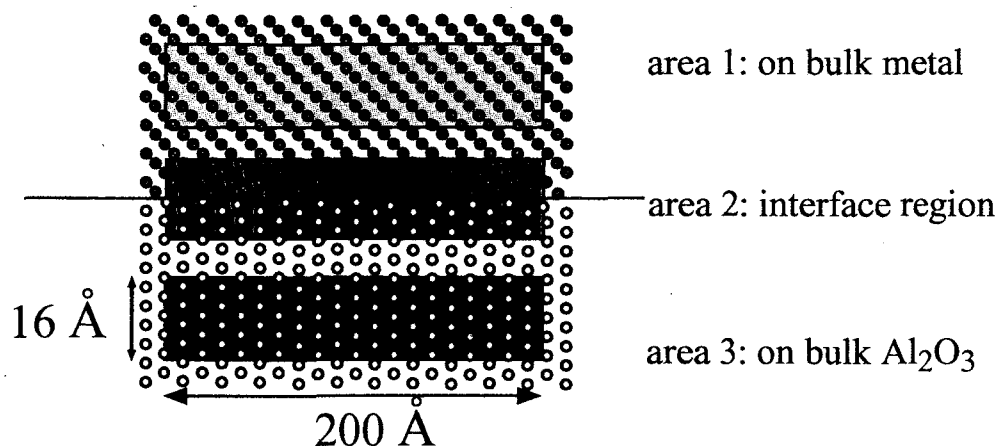
assumption: intensities of incoherent
scattered electrons can be
added

experimental "proof": C. Scheu

theoretical "proof": G. Möbus

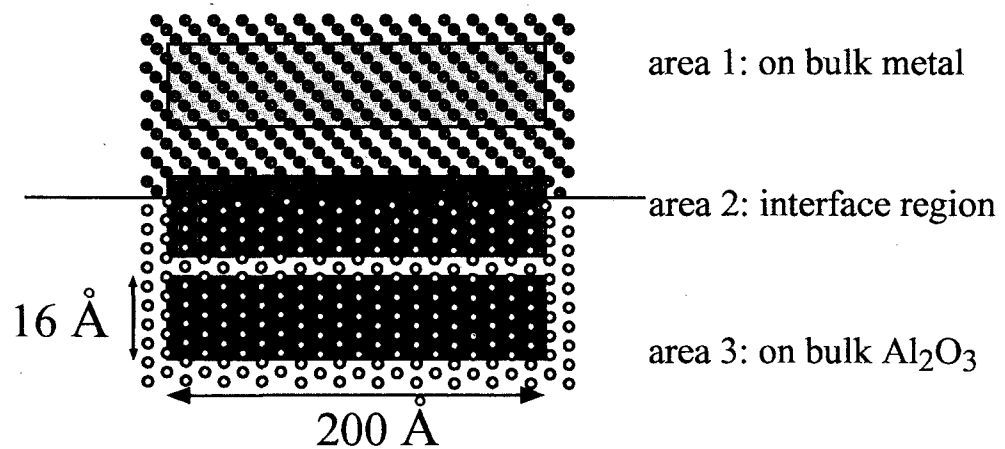
Results Cu/(0001)Al₂O₃ (MBE-Stuttgart T=600°C)

symmetrical case:



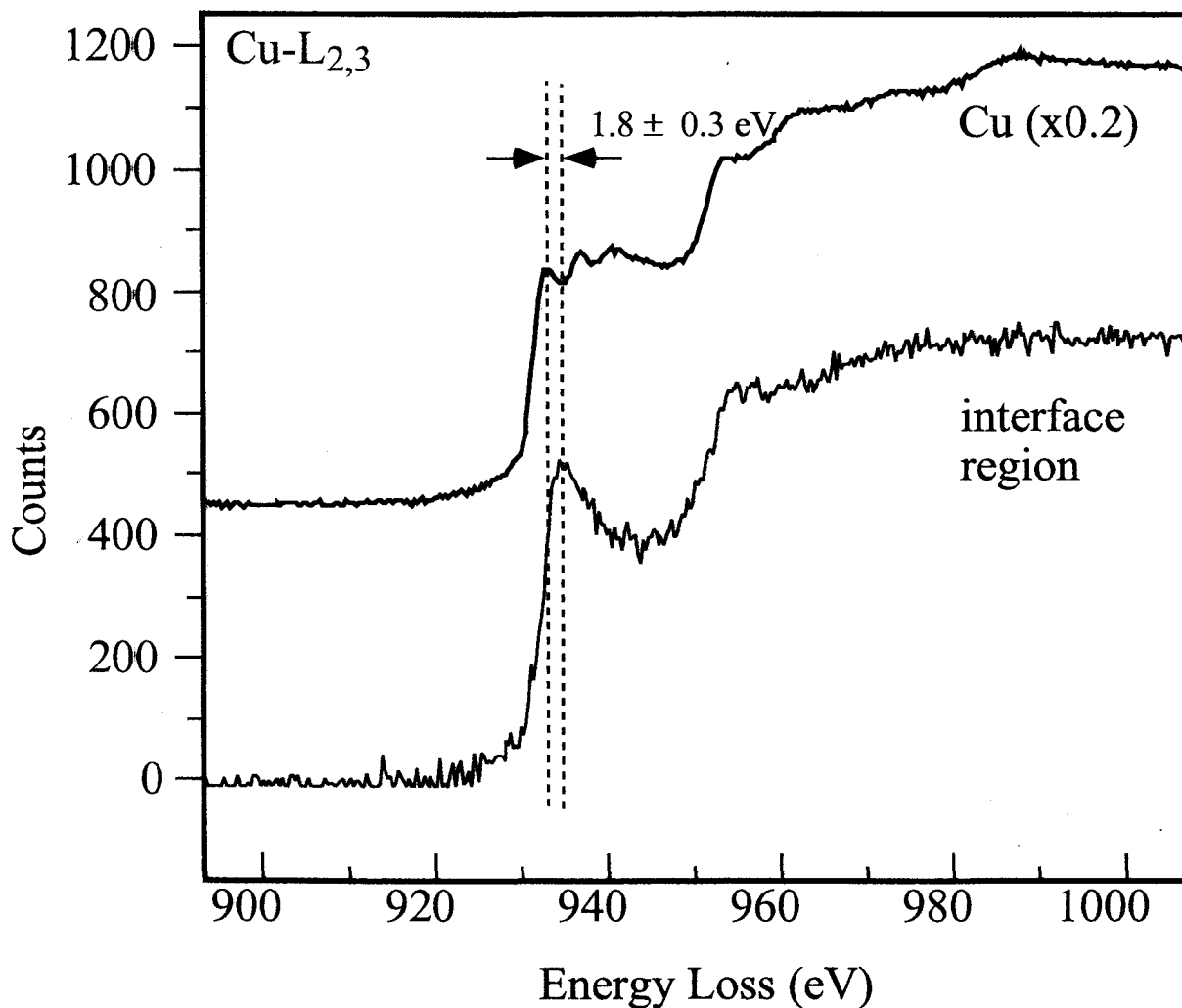
Cu/(0001)Al₂O₃ (T=600°C)

asymmetrical case:

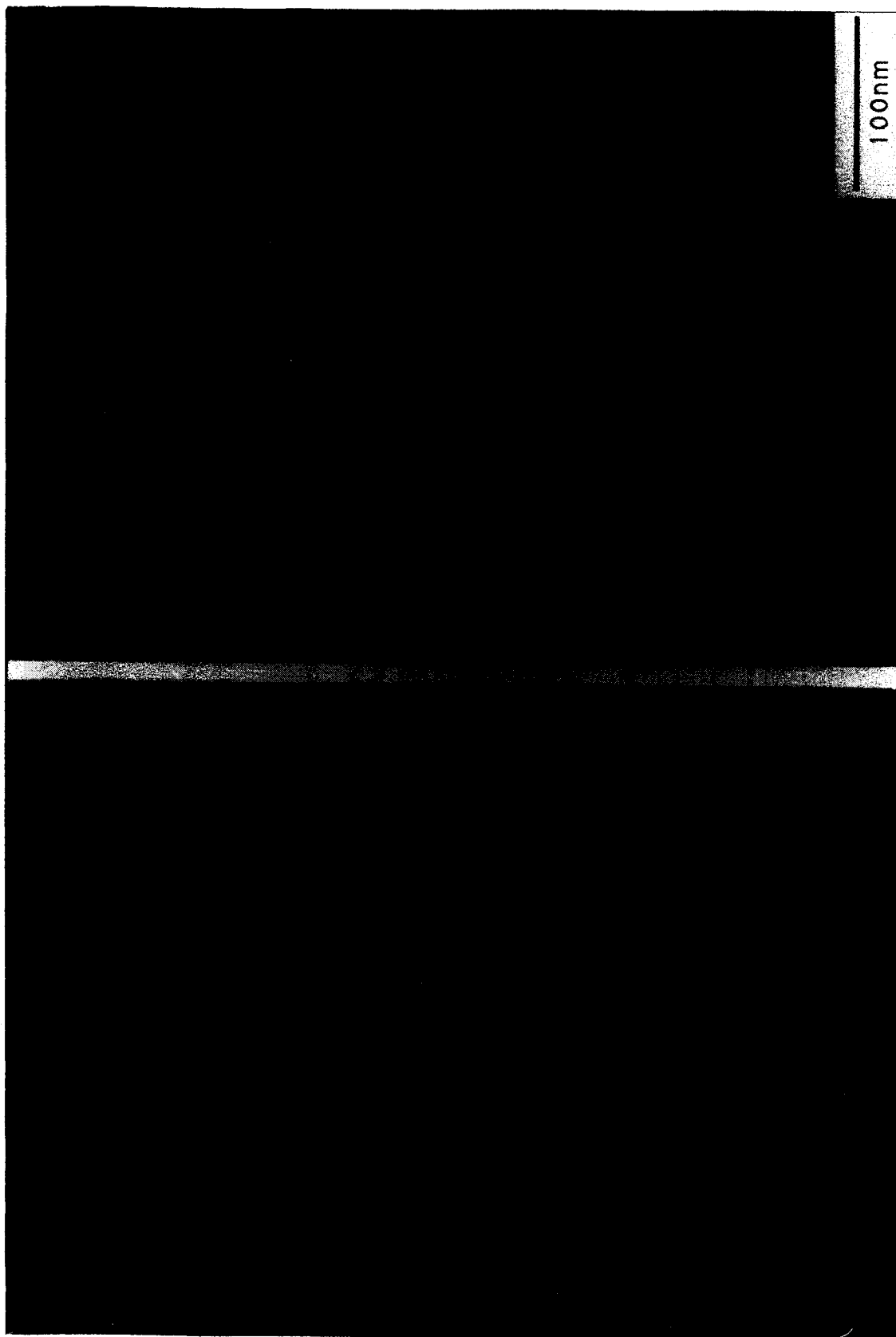


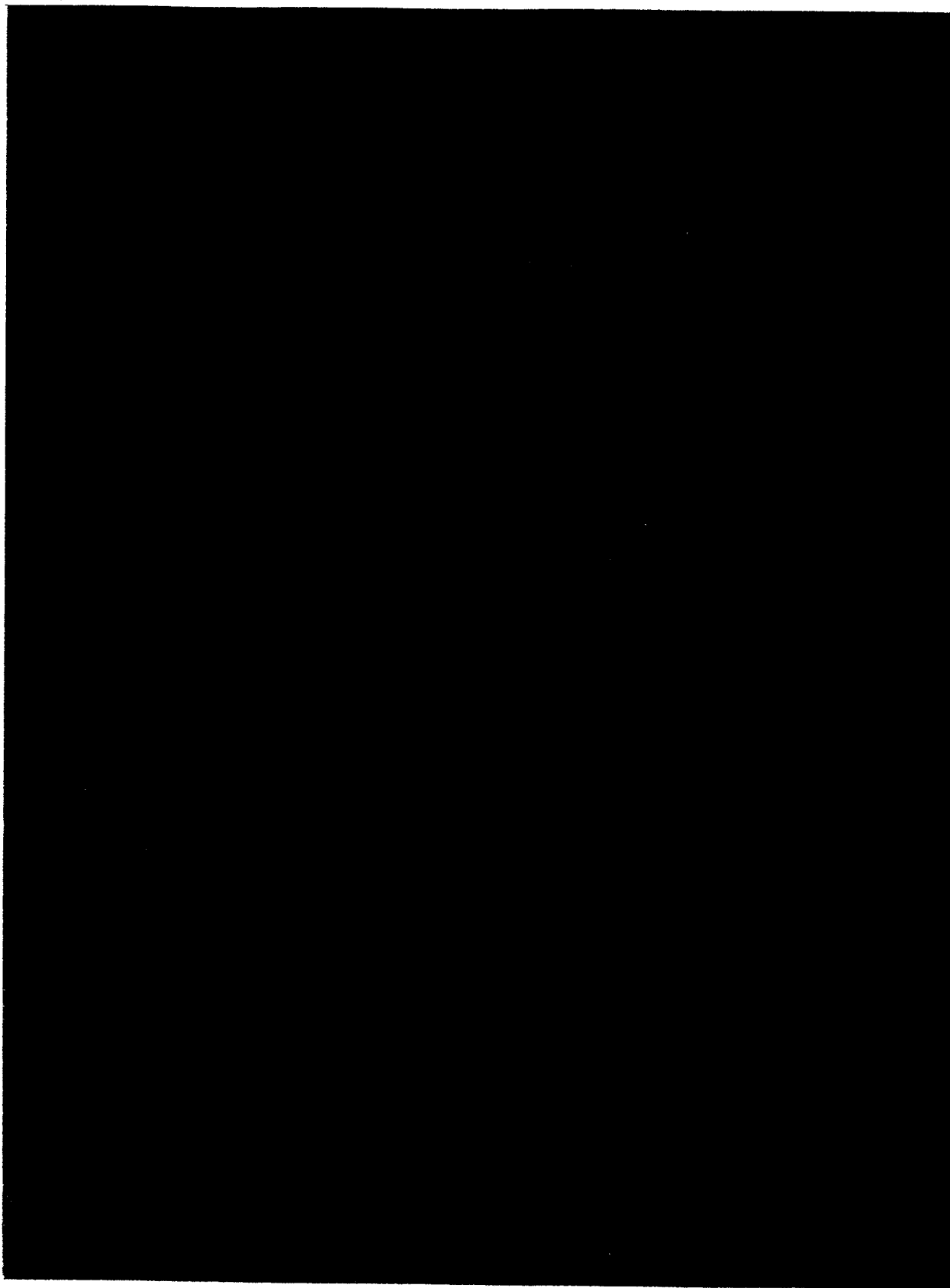
estimation of $\alpha = 0.08$

➔ only $1.28 \text{ \AA}$ of the scanning area contains bulk Cu atoms!



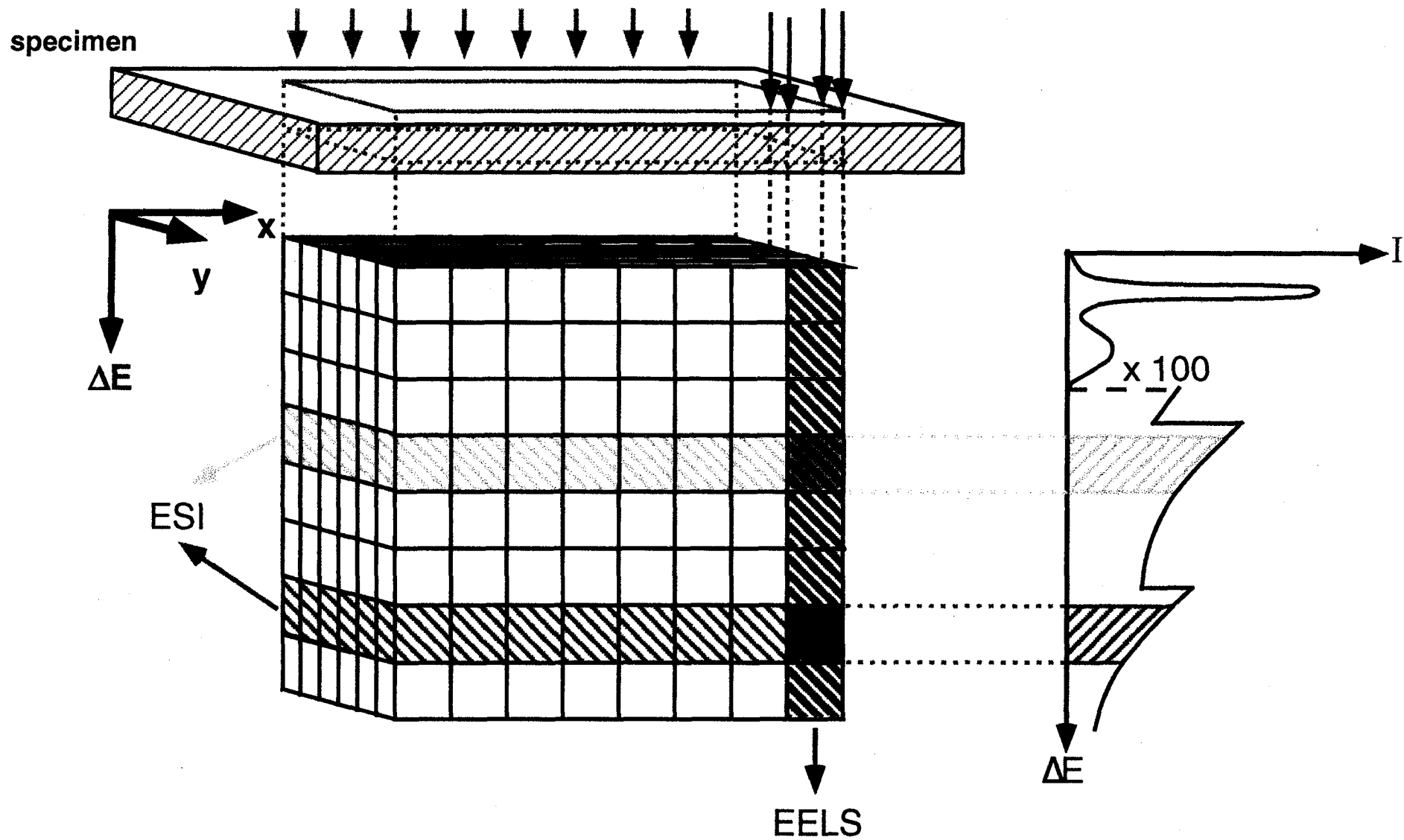
0/1



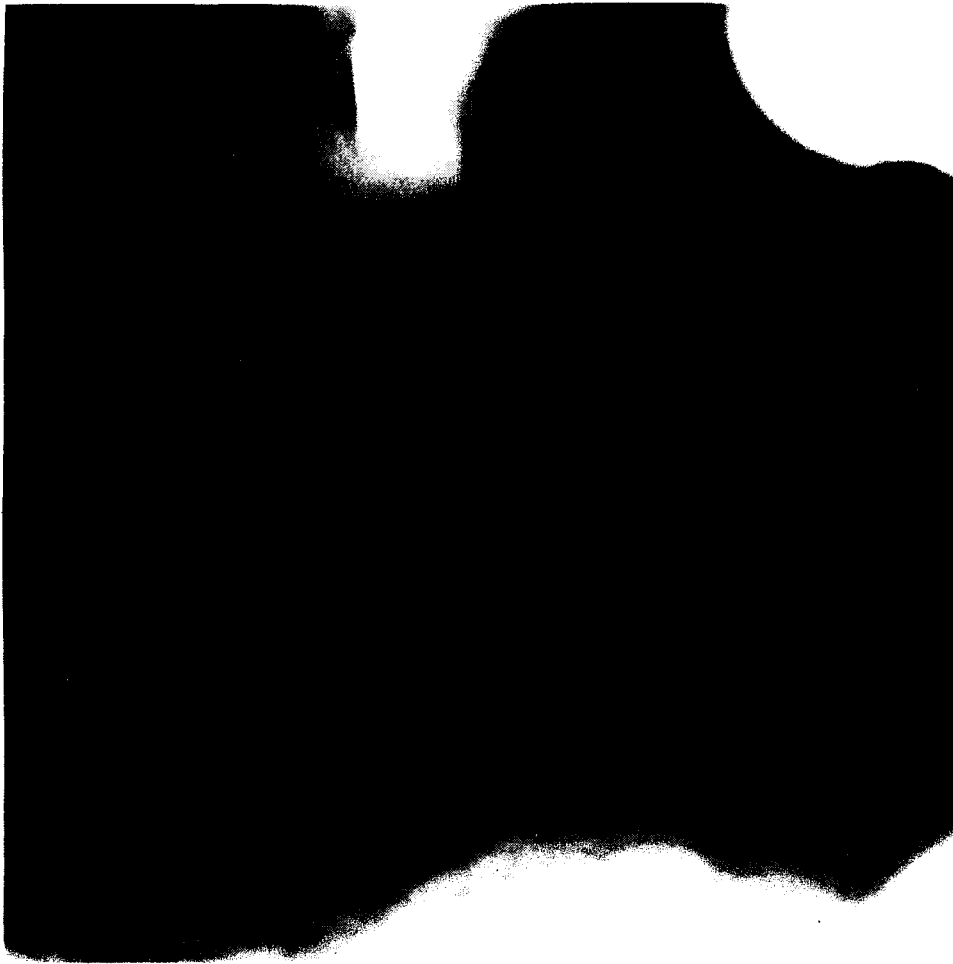


ESI

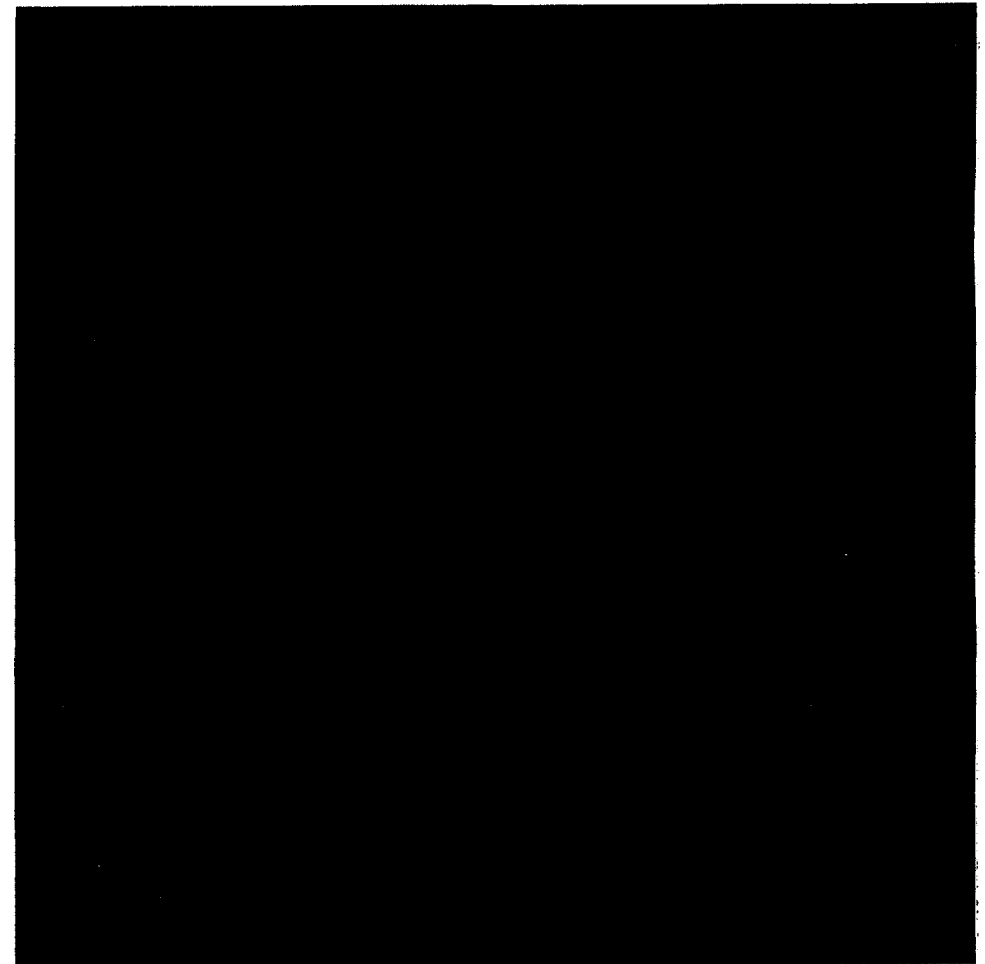
Image mode: $I = I(x, y, \Delta E)$



Sintered polymer derived $\text{Si}_3\text{N}_4/\text{SiC}$ -composite



(bright field image)



red: C (SiC)
green: N (Si_3N_4)
blue: O (am. oxide)

energy resolution
spatial resolution

Revolutionary Developments in Instrumentation

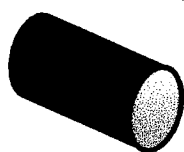
of TEM

- Spherical aberration (C_s) of magnetic lenses can be corrected! (Haider, Rose, Urban)
- Monochromators for electron source will soon be available (Rose)
- Energy filter for electrons with high transmittivity are available (Mandoline filter)
- Advanced detection systems allow collection of each electron (low noise CCD cameras)

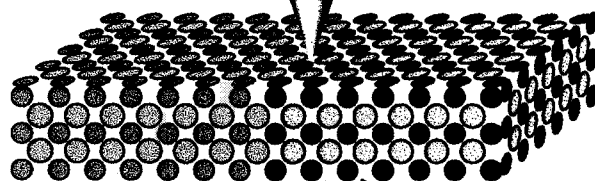
FEG (200 kV)

Monochromator
Energy selecting
slit ($\Delta E \approx 0.2$ eV)

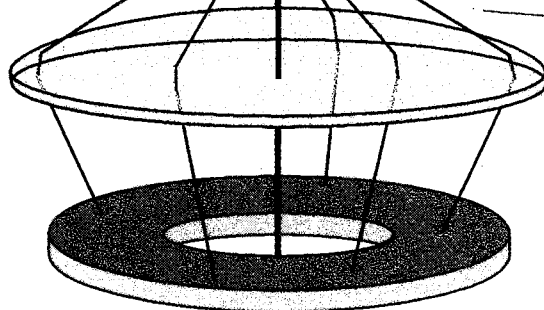
XEDS



Object



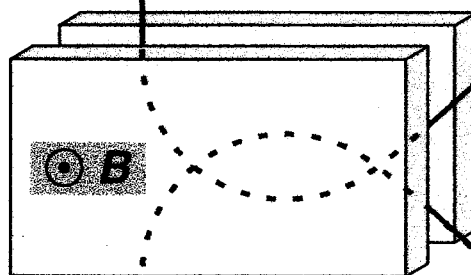
Objective
Lens



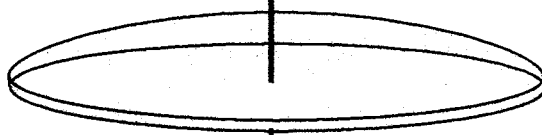
Annular Detector
»Z-Contrast«



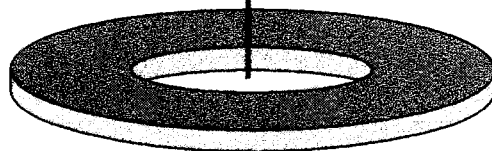
Energy Filter
»Mandoline«



Projector
Lens



Energy filtered
Annular Detector



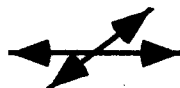
CCD



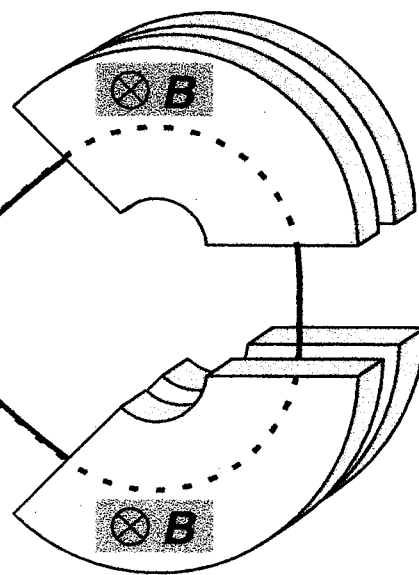
SESAMe

Sub-Evolt-
Sub-Angstrom-
Microscope

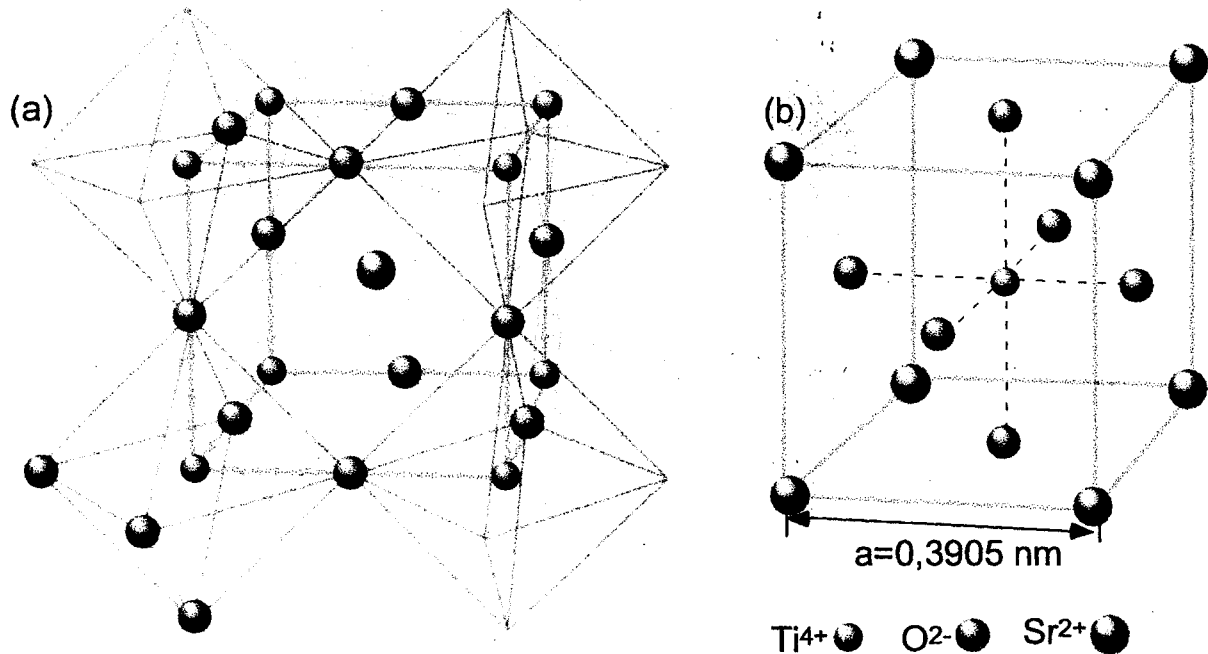
Electron Beam
 $\varnothing < 2$ Å



Large Angle
Scattering



SrTiO₃ - Crystal Properties



Crystal structure of SrTiO₃. Titan is in 6-fold coordination by Oxygen.

Selected properties of SrTiO₃

Crystal structure : Perovskite
cubic → highly symmetrical
complex structure

Lattice constant : $a = 0.3905 \text{ nm}$

Space group : $\text{Pm}\bar{3}\text{m}$

Moos hardness : 6...6.5

Single crystal orientation : (001) and (110)

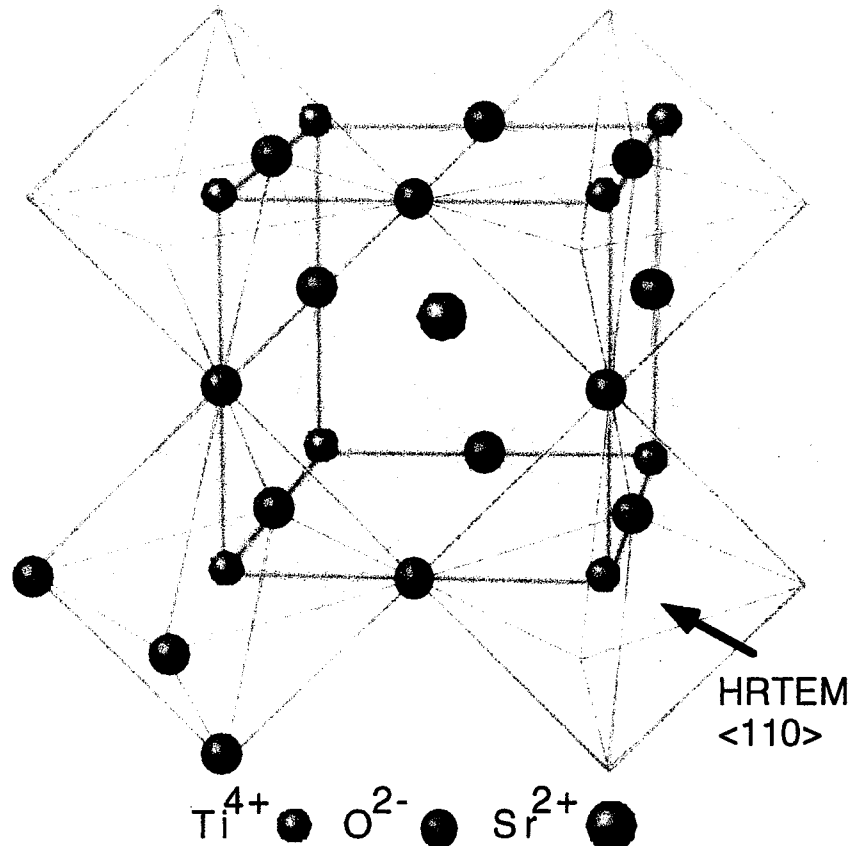
*scattering factors?
charge density distribution?*

crystallography

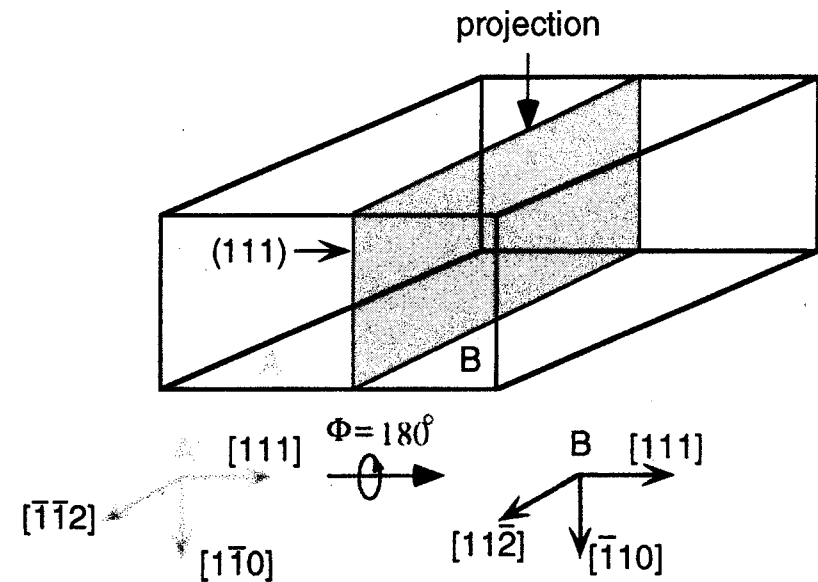
crystal structure of SrTiO_3

Perovskite

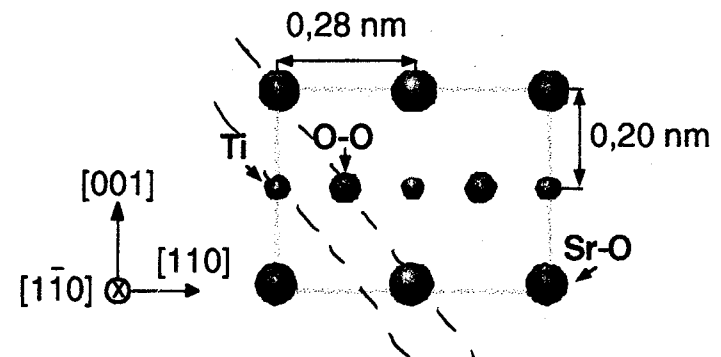
$\text{Pm}\bar{3}\text{m}$; $a = 0,3905 \text{ nm}$



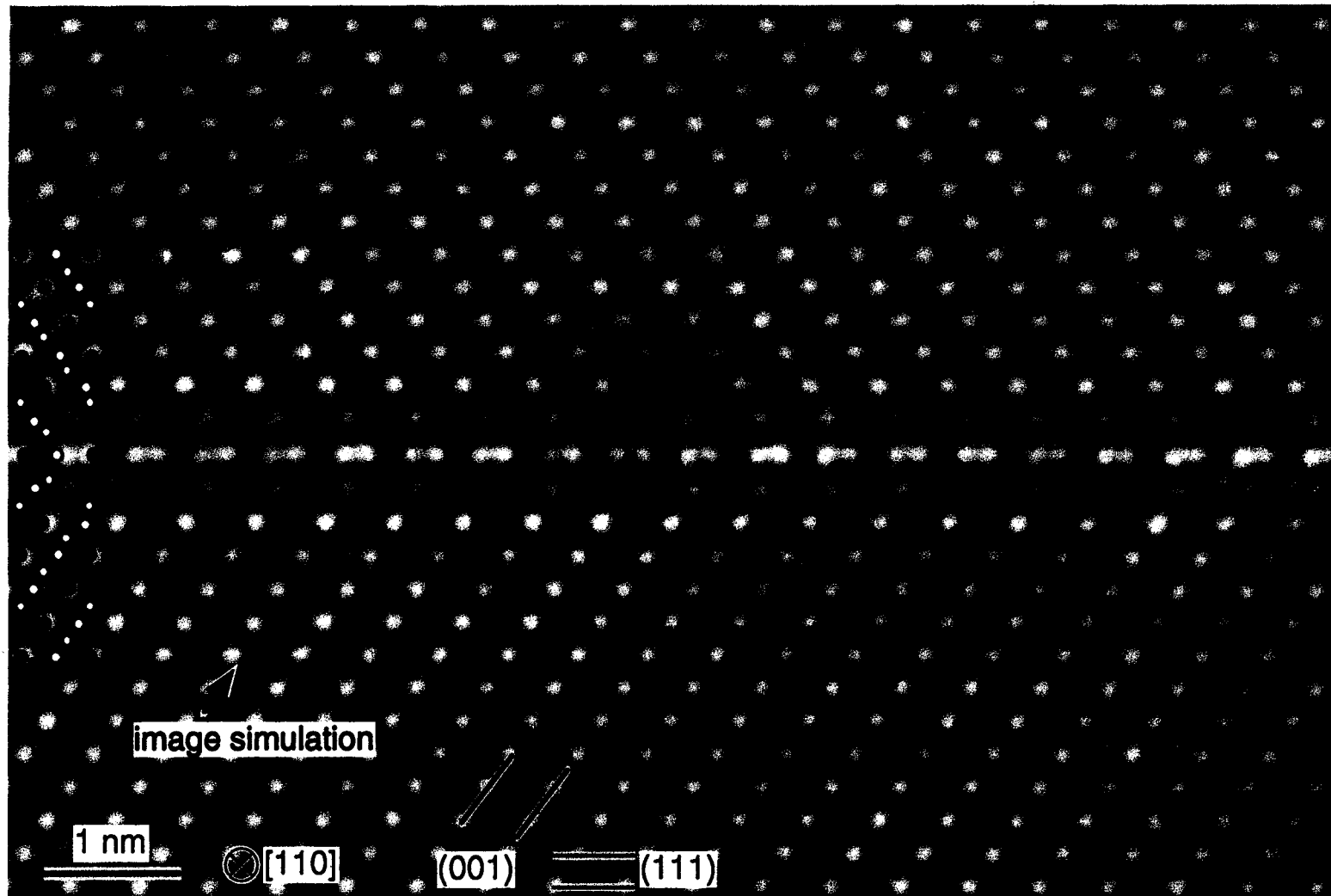
grain boundary crystallography



unit cell in $\langle 110 \rangle$ projection



$\Sigma=3$, (111) grain boundary in $\langle 110 \rangle$ projection



Kienzle

SrTiO₃ $\Sigma 3$ (111) preliminary models

structure models

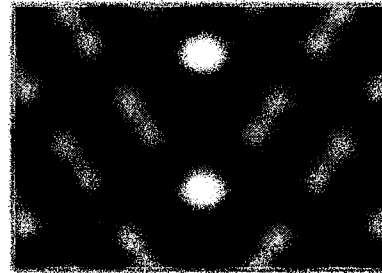
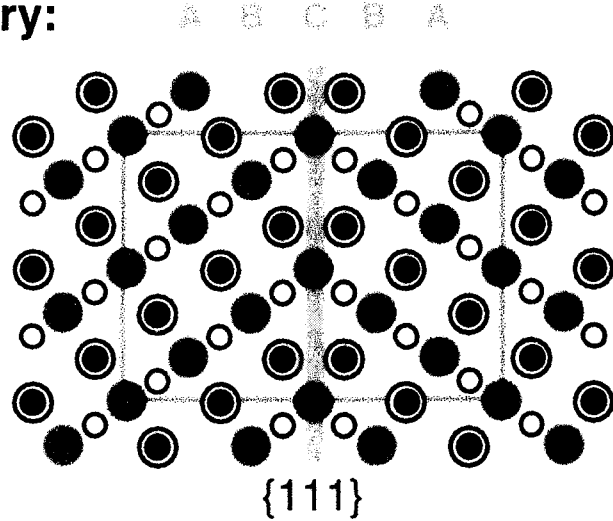
image simulations

$t = 6 \text{ nm}, f = 46 \text{ nm}$

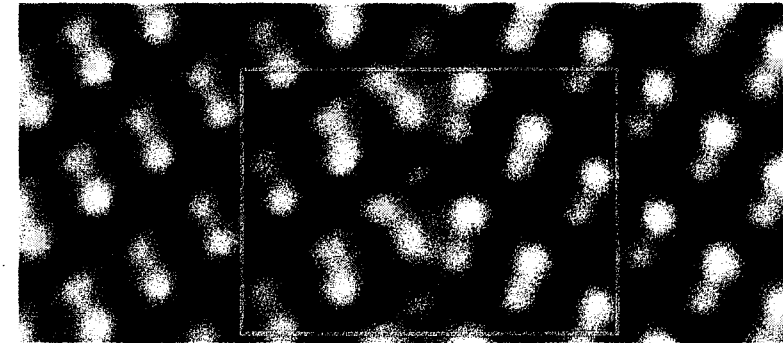
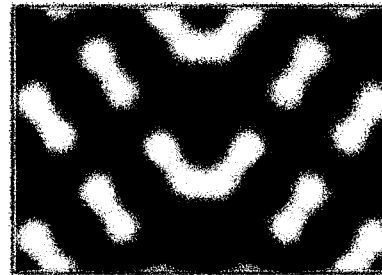
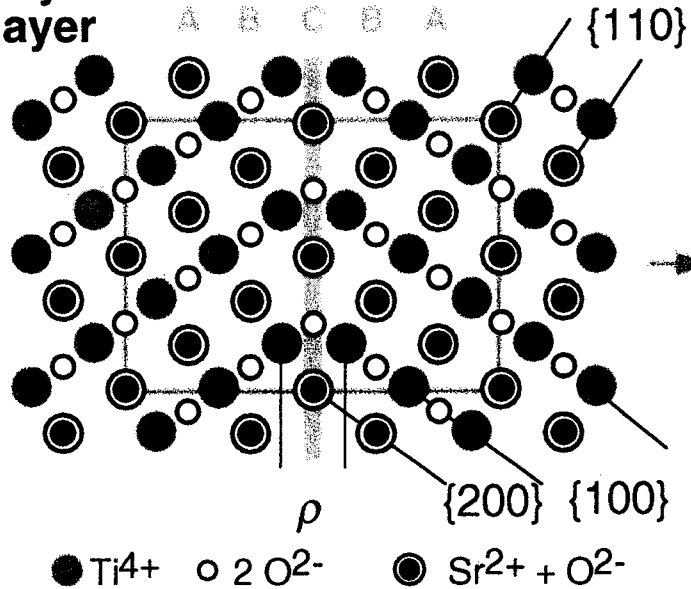
experimental image

$t = 6 \text{ nm}, f = 46 \text{ nm}$

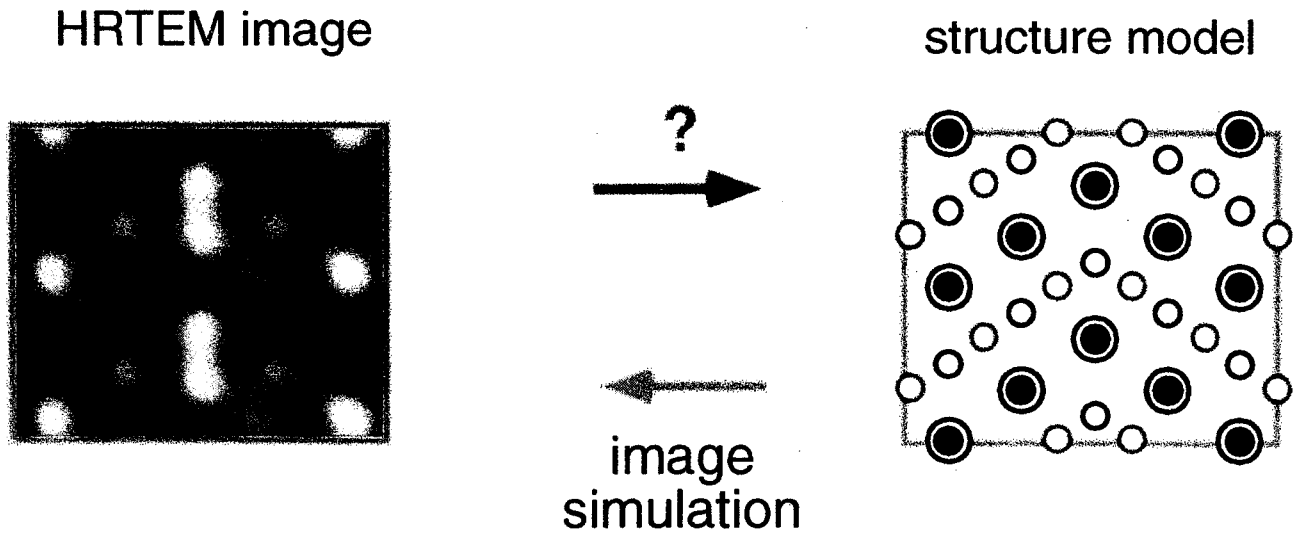
boundary:
Ti layer



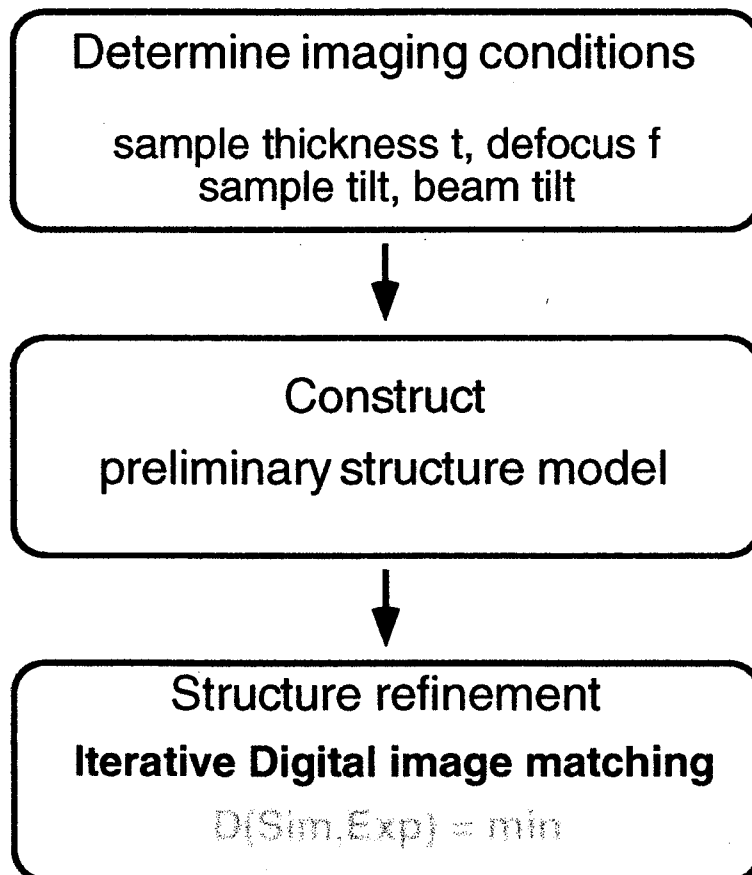
boundary:
Sr-O₃ layer



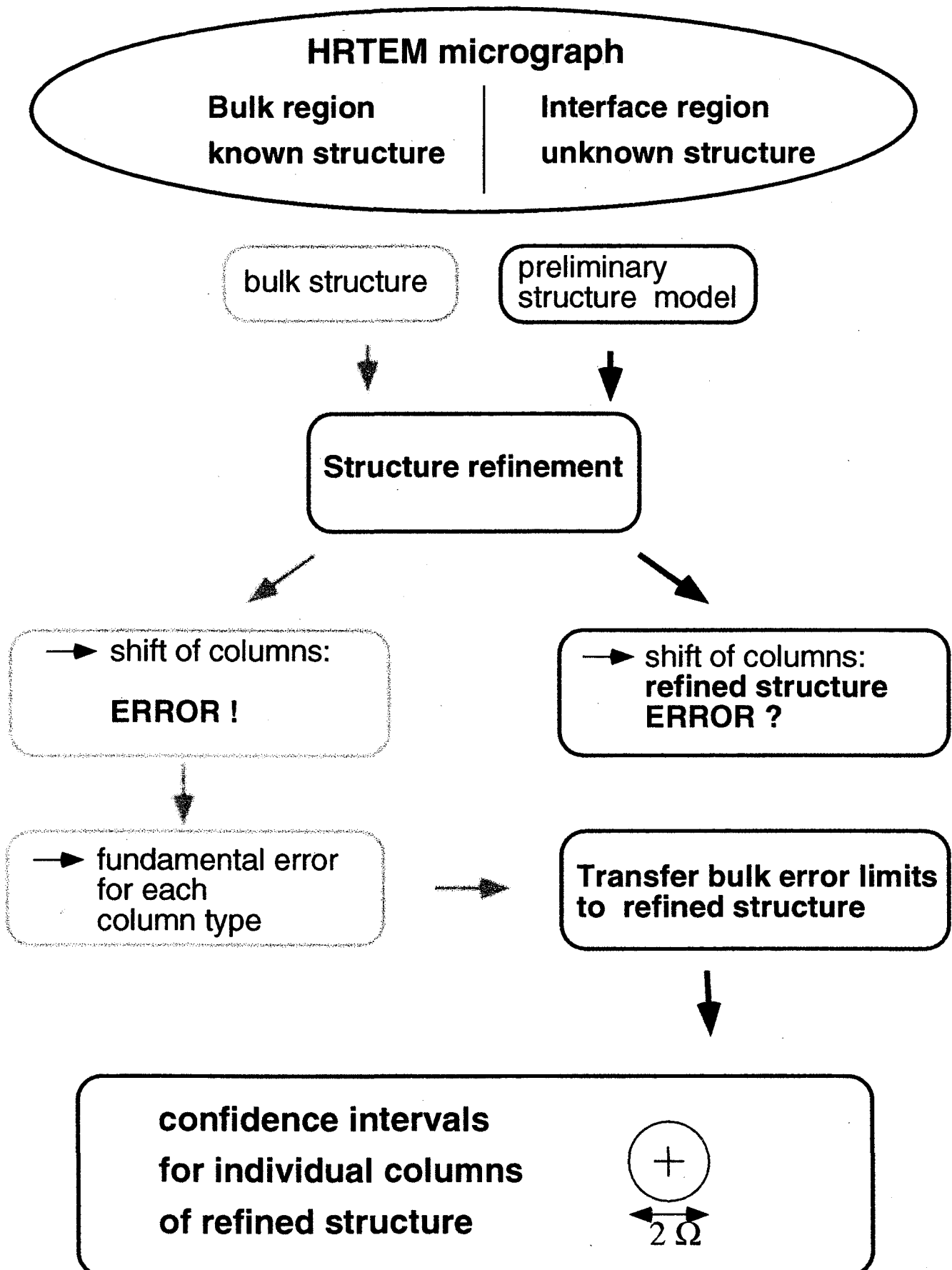
QHRTEM structure retrieval



Structure retrieval



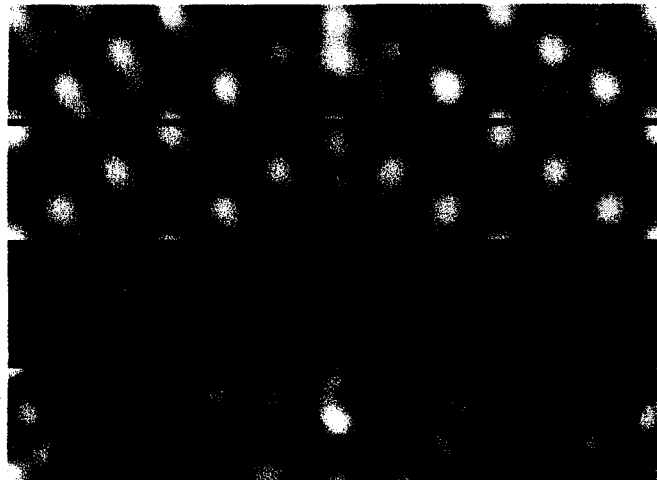
Quantification of error limits



Iterative Digital Image Matching

geometric model

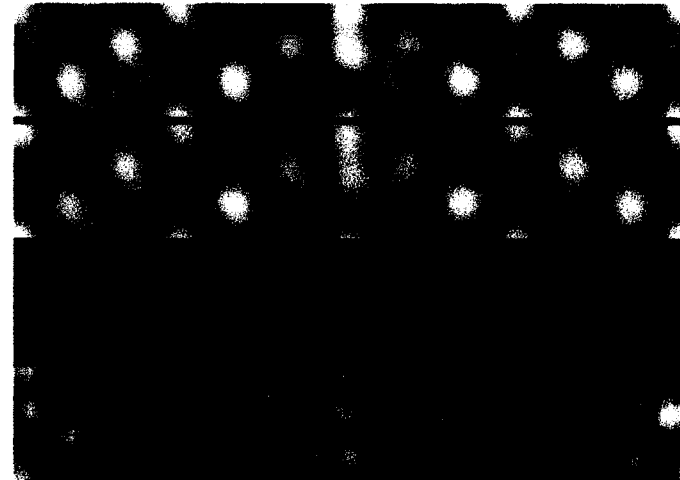
bulk like Interface bulk like



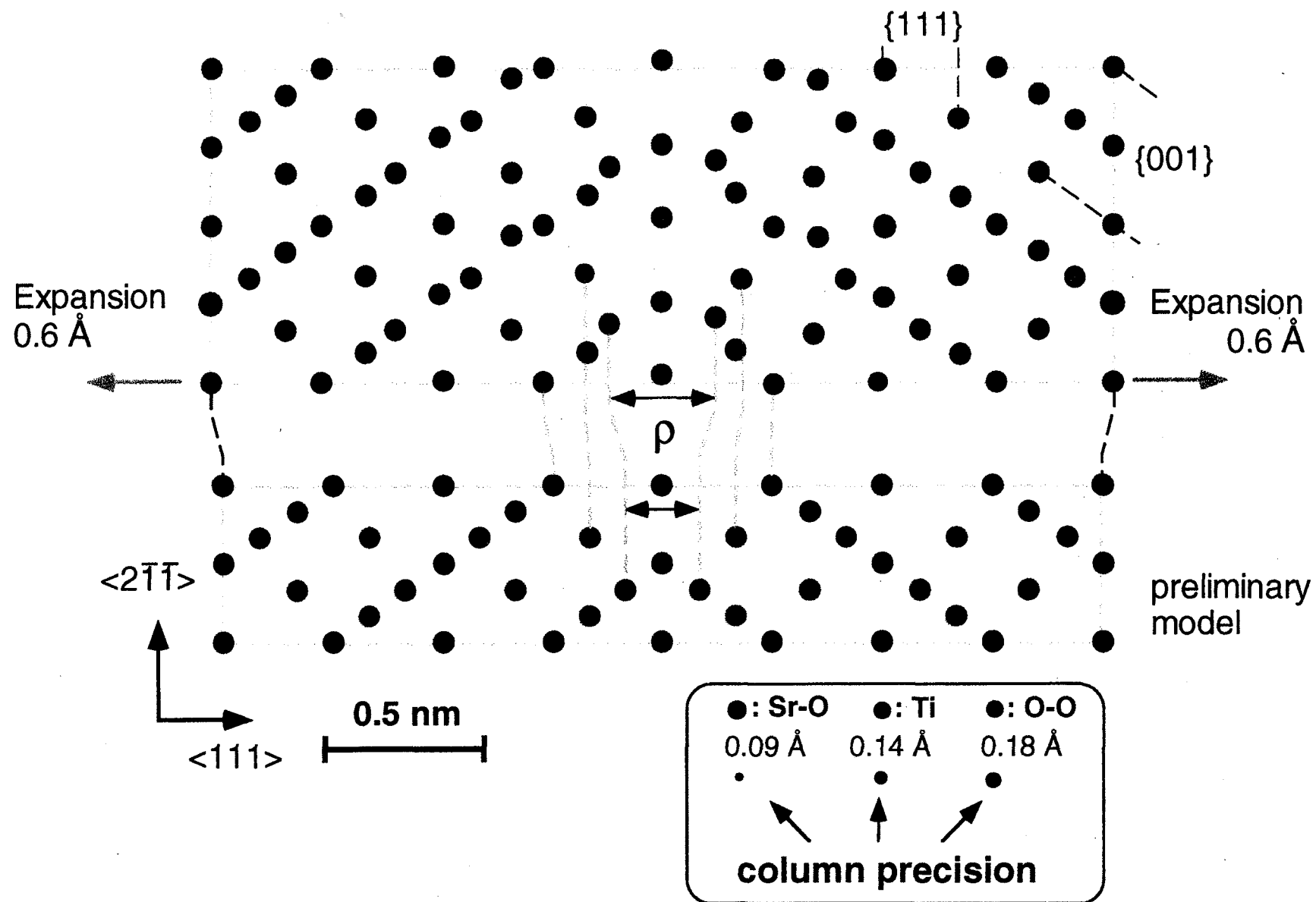
1 nm

refined structure

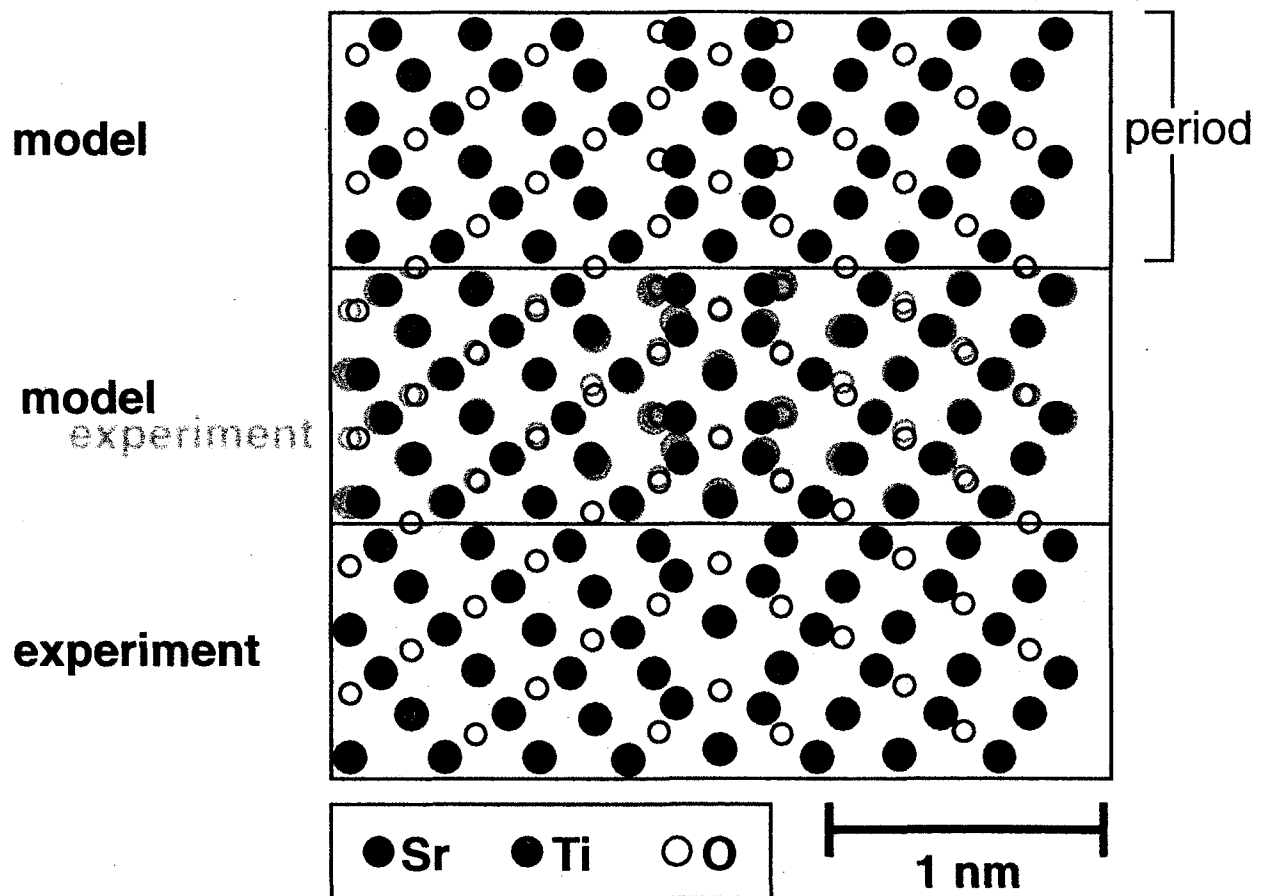
Interface



Refined structure model



computer modelling versus experiment

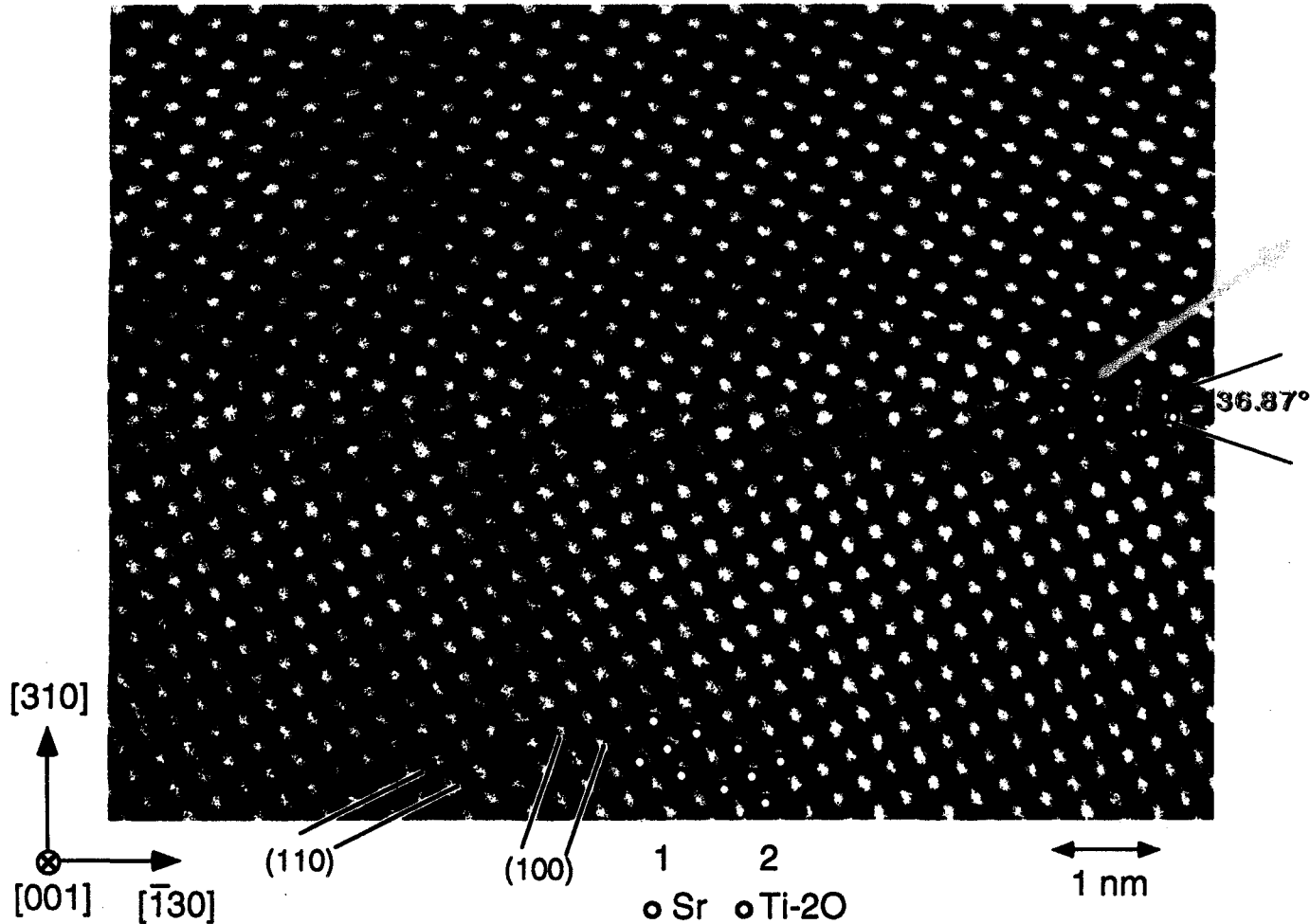


comparision	experiment	modelling
termination	Sr-O	Sr-O
coordination	Ti-O octahedra	Ti-O octahedra
excess volume:	expansion: + 0.6 Å	expansion + 0.3 Å
relaxation at gb	Sr-O inward vs Ti low	Sr-O inward vs Ti large

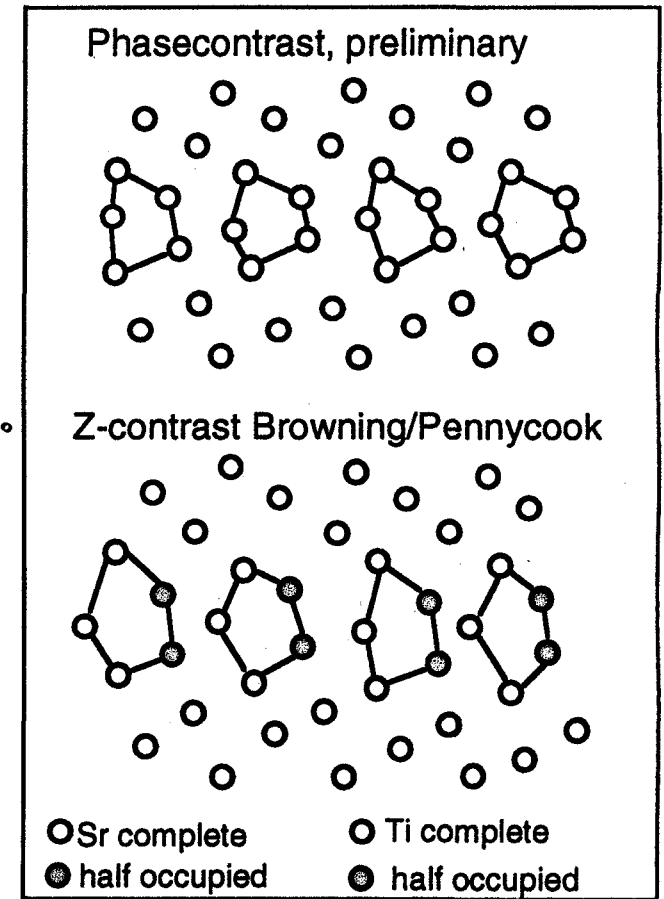
Near $\Sigma 5$ (310) grain boundary / structure

Doping level: Fe/Ti = 0.01 at.%

HRTEM JEM 4000EX

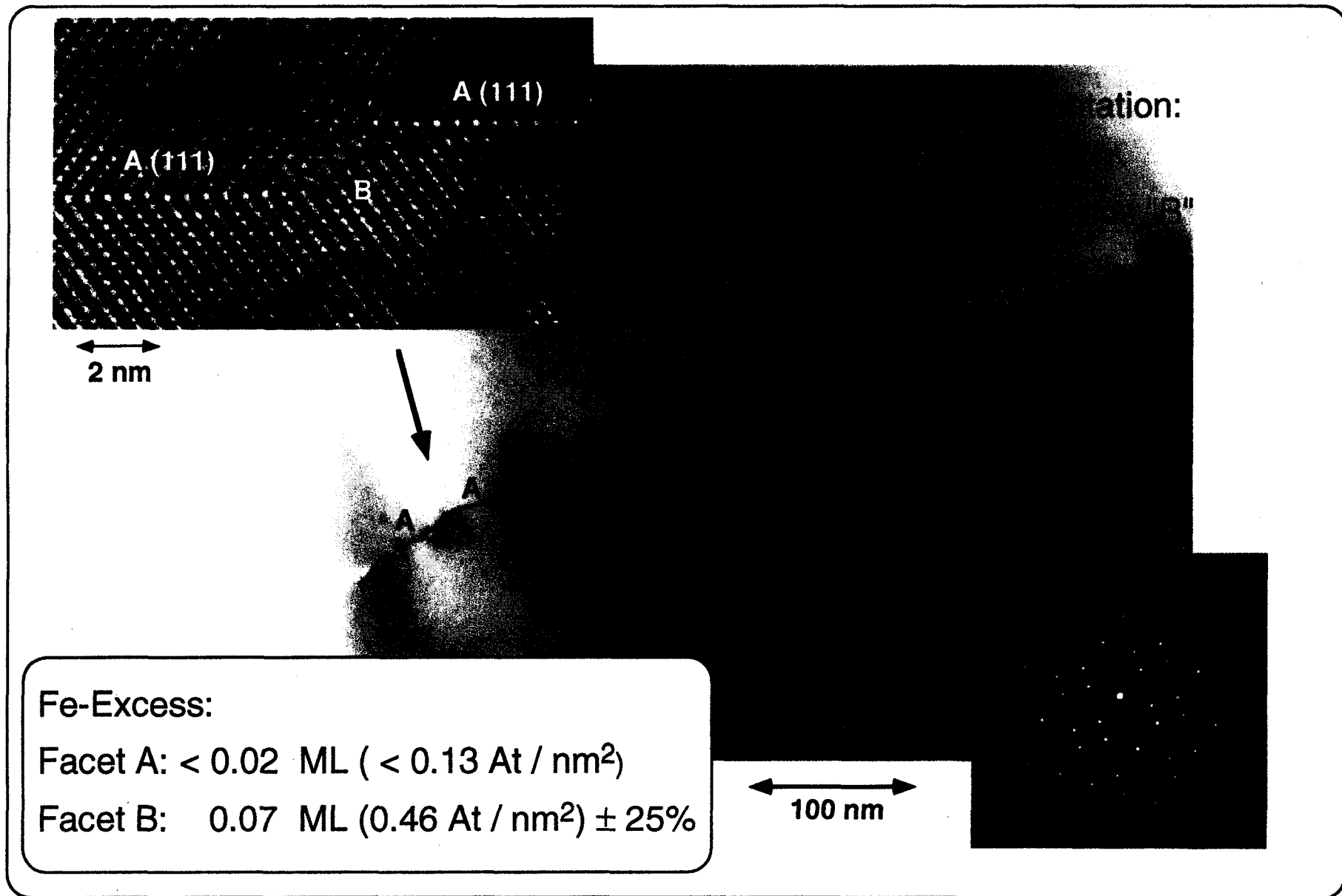


Structure models



Near $\Sigma 3$ (111) grain boundary in polycrystalline SrTiO_3

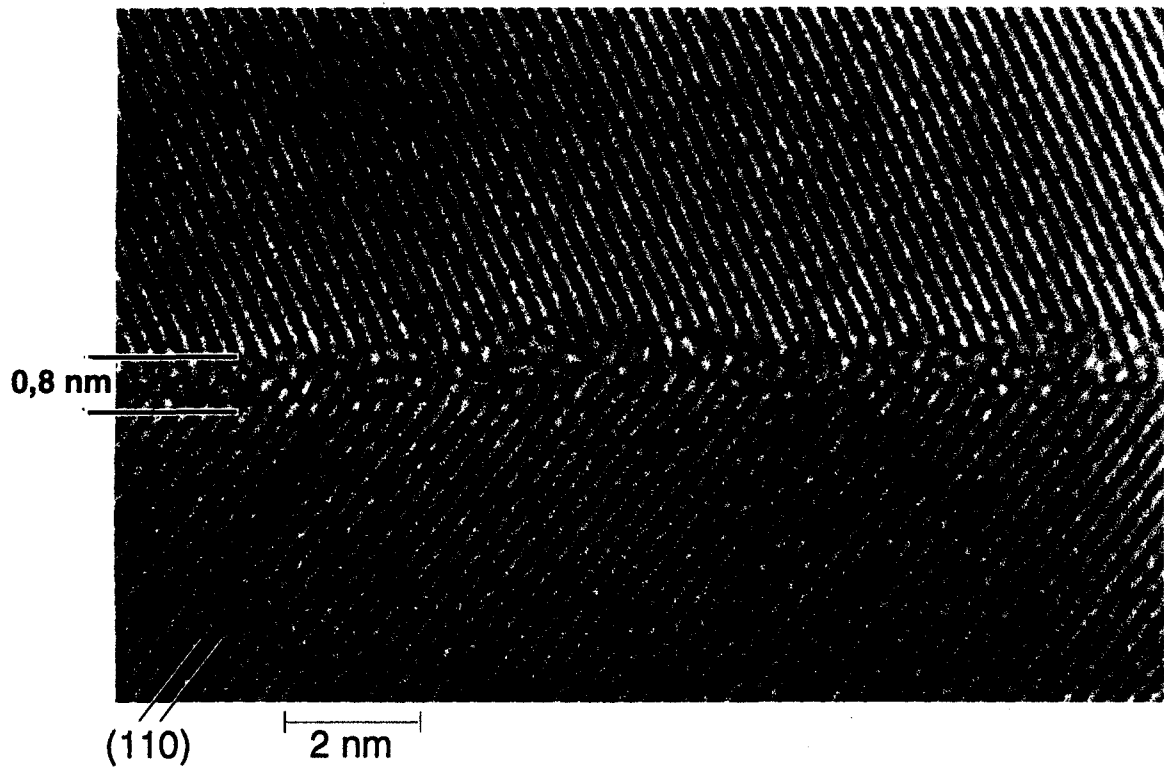
Doping level: Fe / Ti = 0.4 at. %



"general" grain boundaries in polycrystalline SrTiO_3

Doping level: $\text{Fe} / \text{Ti} = 0.4 \text{ at.}\%$

Profiles

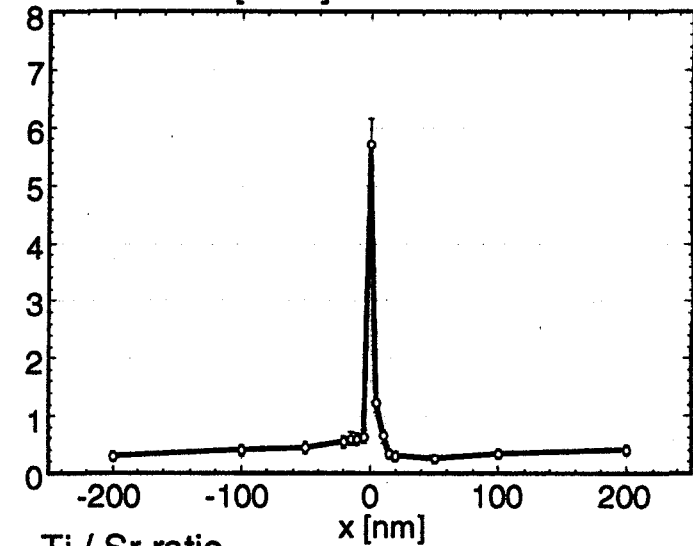


EDS

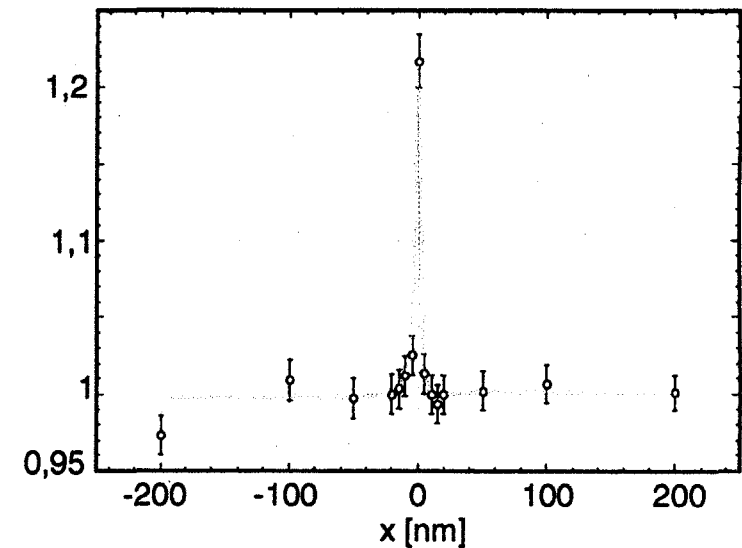
Fe-Excess: $(0.61 \pm 0.04) \text{ ML}$ ($4.0 \text{ At} / \text{nm}^2$)

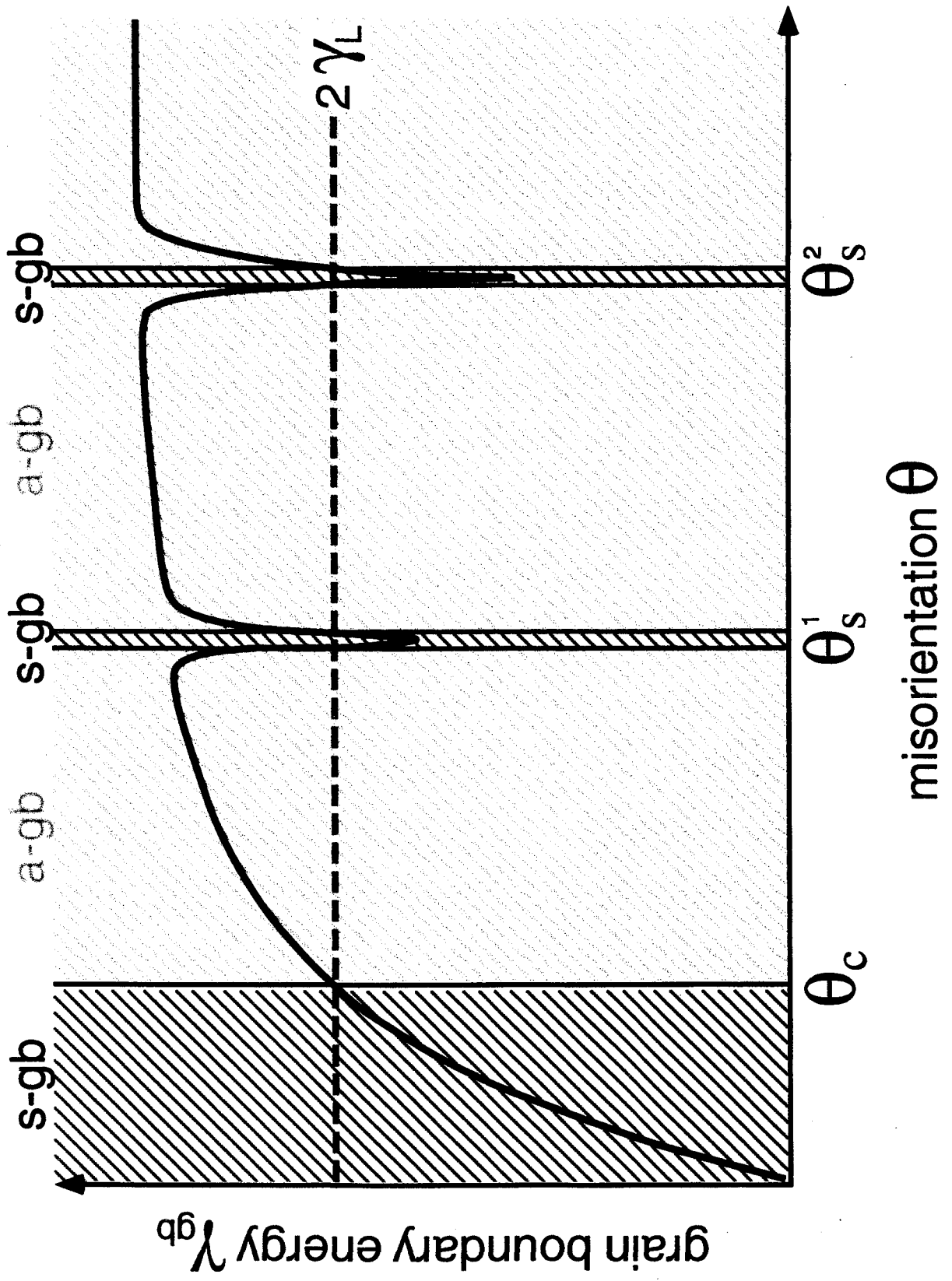
Ti-Excess: $(1.3 \pm 0.14) \text{ ML}$ ($10.5 \text{ At} / \text{nm}^2$)

Fe / Ti ratio [at.%]

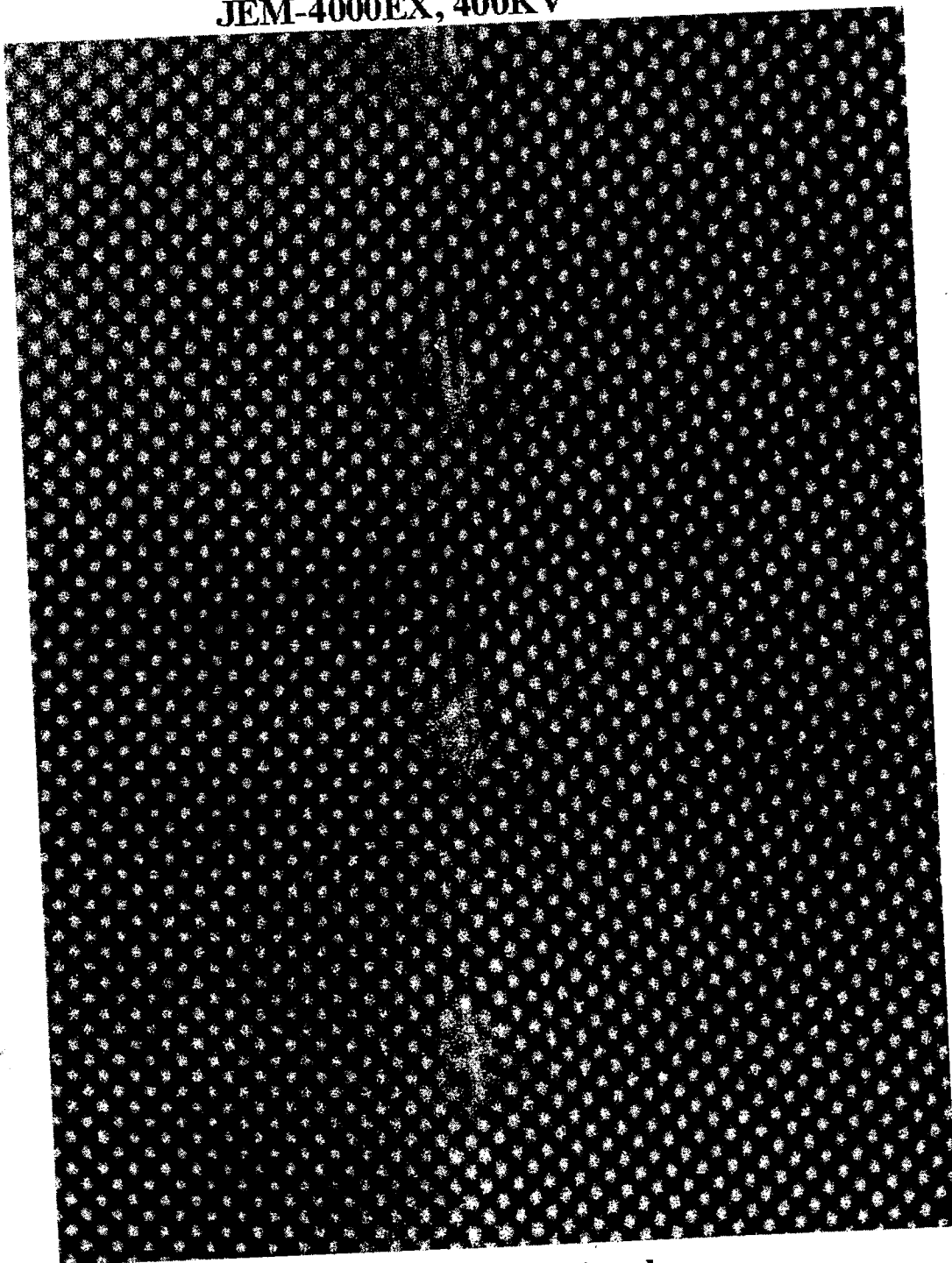


Ti / Sr ratio



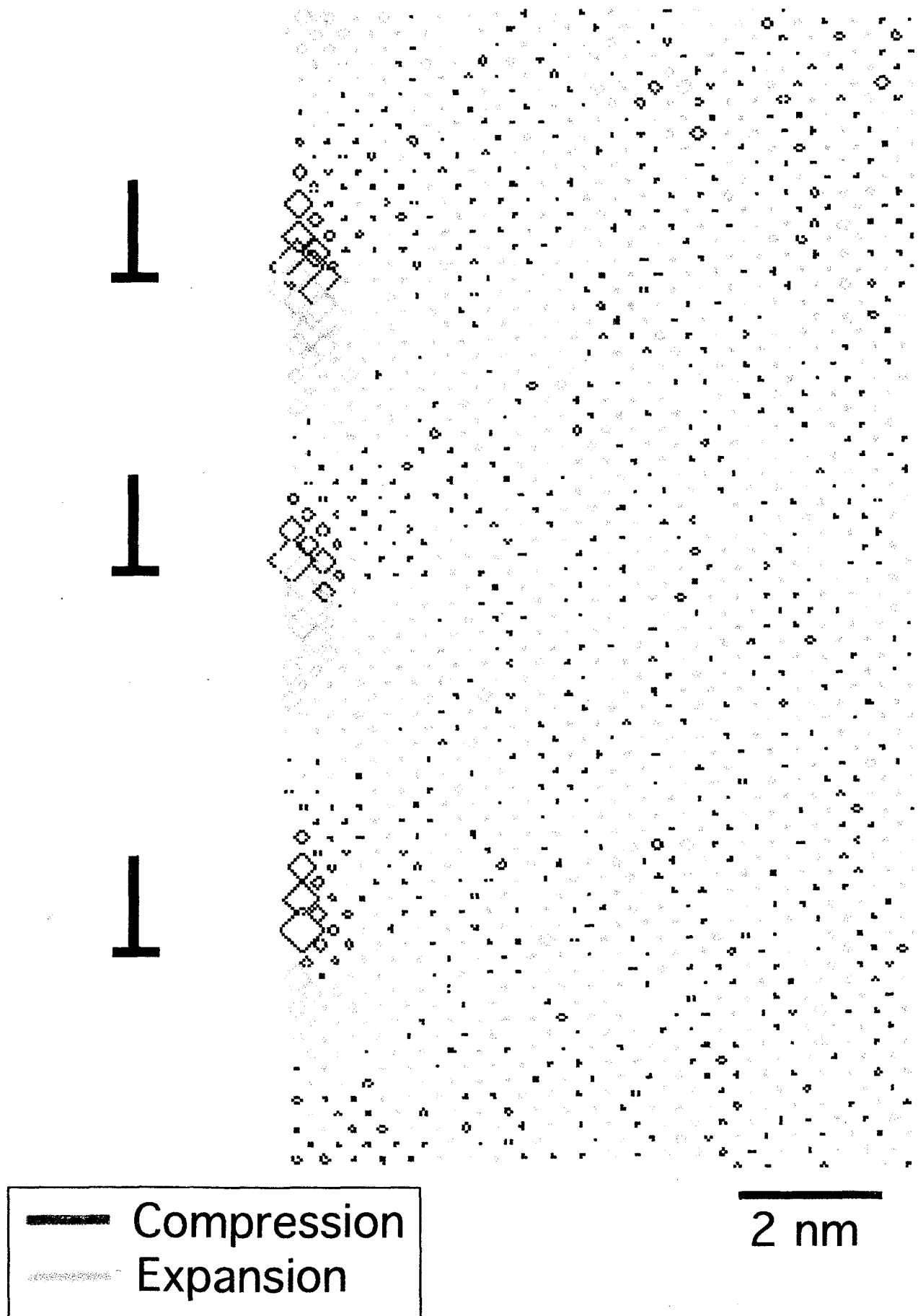


JEM-4000EX, 400KV



Courtesy of Oliver Kienzle

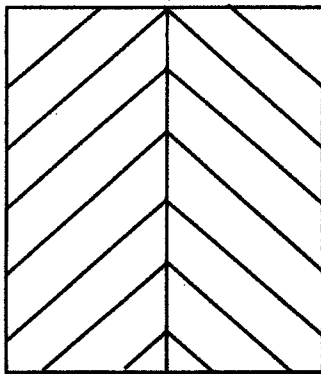
Small-angle grain boundary in SrTiO_3



Summary, conclusion and outlook

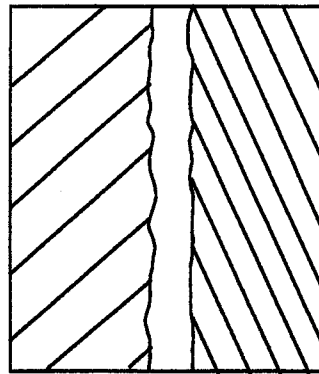
grain boundary structure $\longleftrightarrow$ space charge region

a) special boundaries



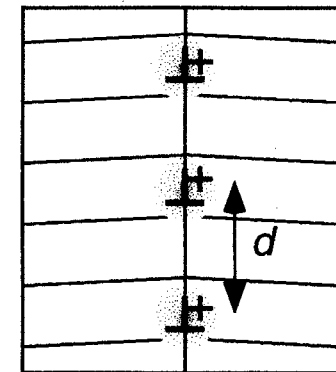
↑
No detectable segregation
 $\Sigma=3$, (111) poly / bicrystal
 $\Sigma=5$, (310) bicrystal
HRTEM: "perfect structures"

b) amorphous film



↑
Sr-deficient
Ti, Fe segregation
 $\text{Ti}_{0.75}\text{Fe}_{0.25}\text{O}_x$
HRTEM, ELNES:
amorphous GB core

c) low angle boundary



outlook

varying d
by misorientation
→ diffusion bonding

Correlation to diffusion across GB, electr. properties

S. Hutt(1999)
O. Kienzle
(J. Maier)

Motivation

Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn
Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd
La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg

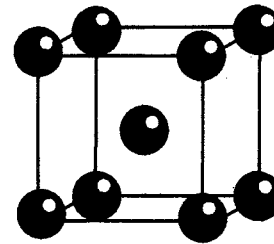
Lattice mismatch:

$\text{Mo} > \text{Ni} > \text{Cr} > \text{Pd}$

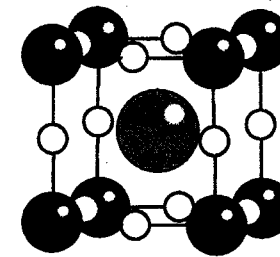
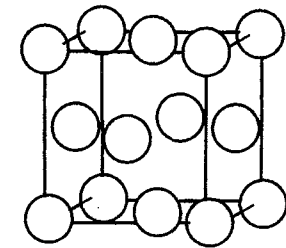
Reactivity:

$\text{Cr} > \text{Mo} > \text{Ni} > \text{Pd}$

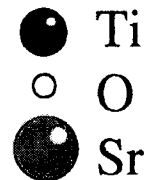
bcc: Cr, Mo



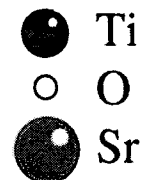
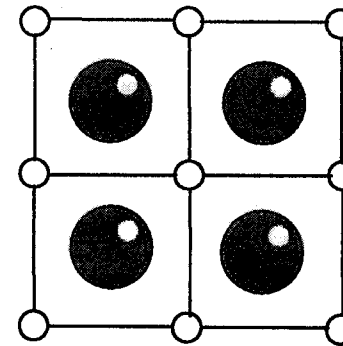
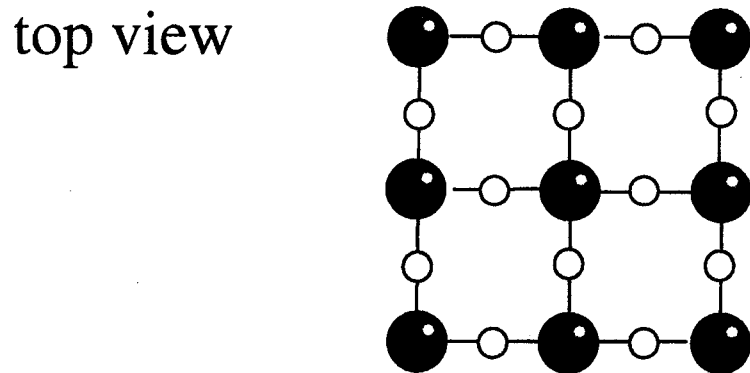
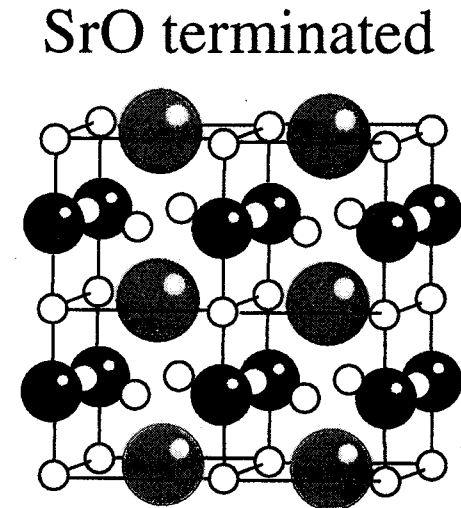
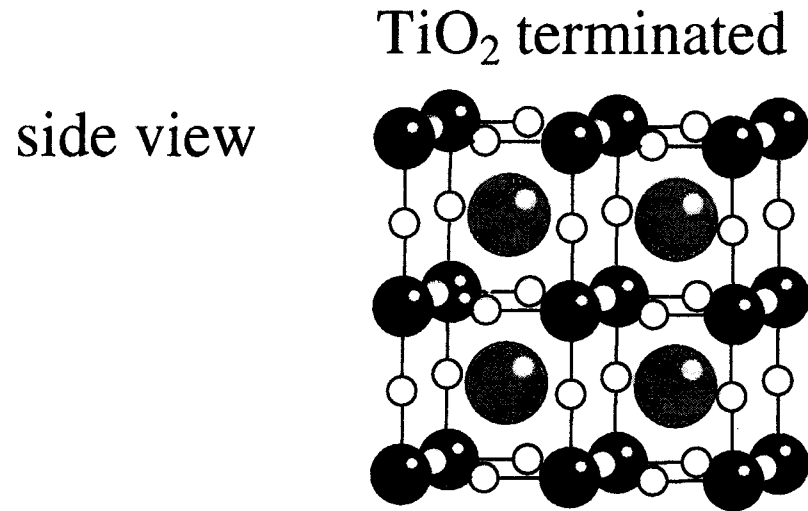
fcc: Ni, Pd



SrTiO_3



Surface Termination



Theory (T. Ochs, T. Classen, C. Elsässer):
metal on top of oxygen is energetically favoured!

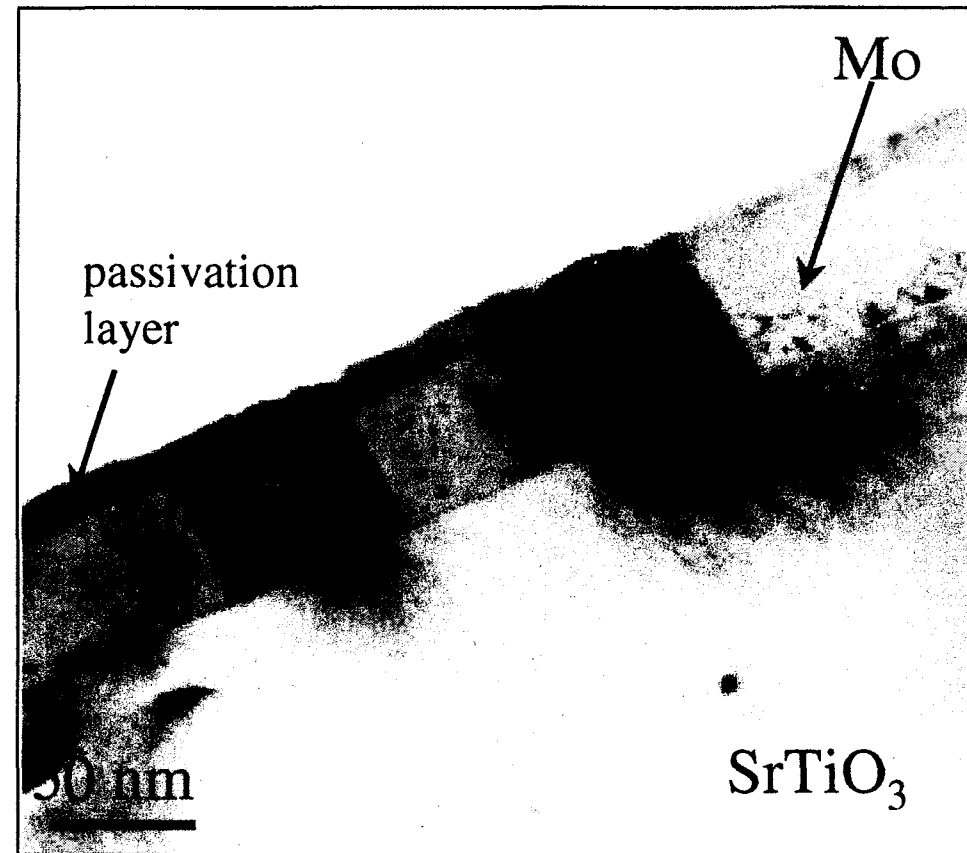
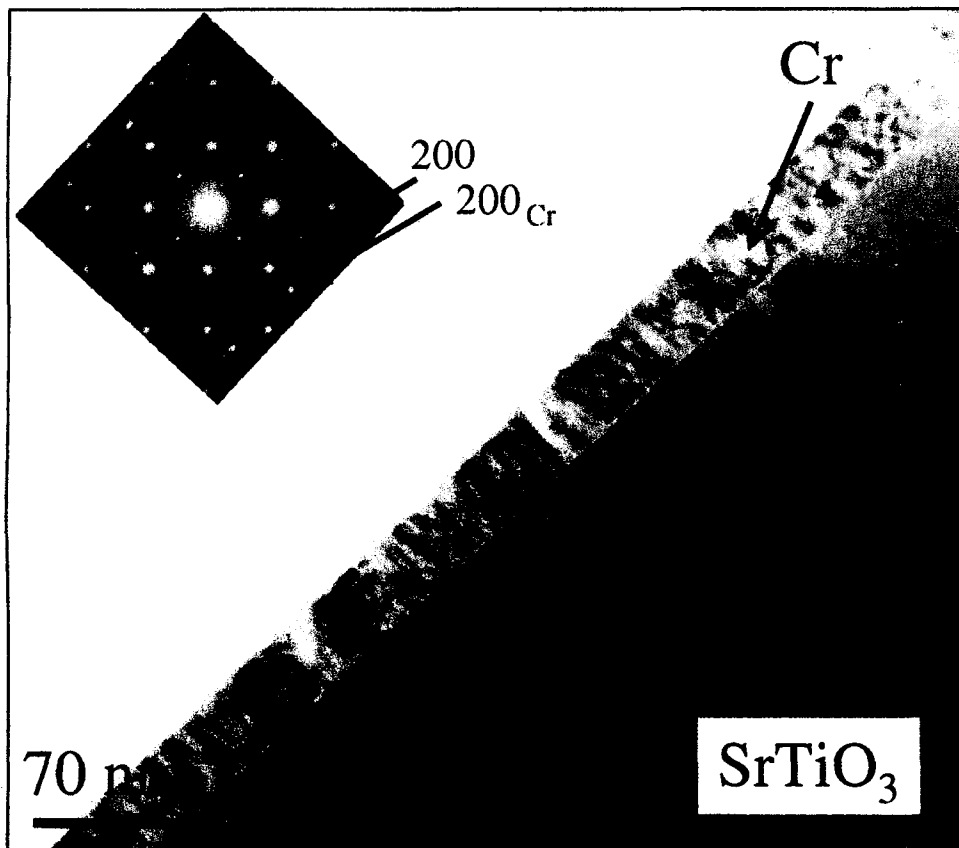
Experimental Details

Specimen fabrication

- nominal 35 - 100 nm thick metal films were grown on (001) SrTiO₃ by MBE in UHV (10^{-8} Pa)
- substrate temperature was varied between 65°C and 650°C for the various metals to obtain epitaxial films

Characterization

- | | |
|------------------------|----------------------------|
| • conventional TEM: | JEOL 2000FX (200 kV) |
| • high-resolution TEM: | JEOL ARM1250 (1250 kV) |
| • analytical TEM: | VG STEM HB 501 UX (100 kV) |

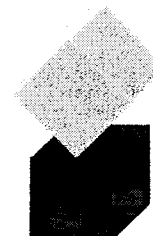


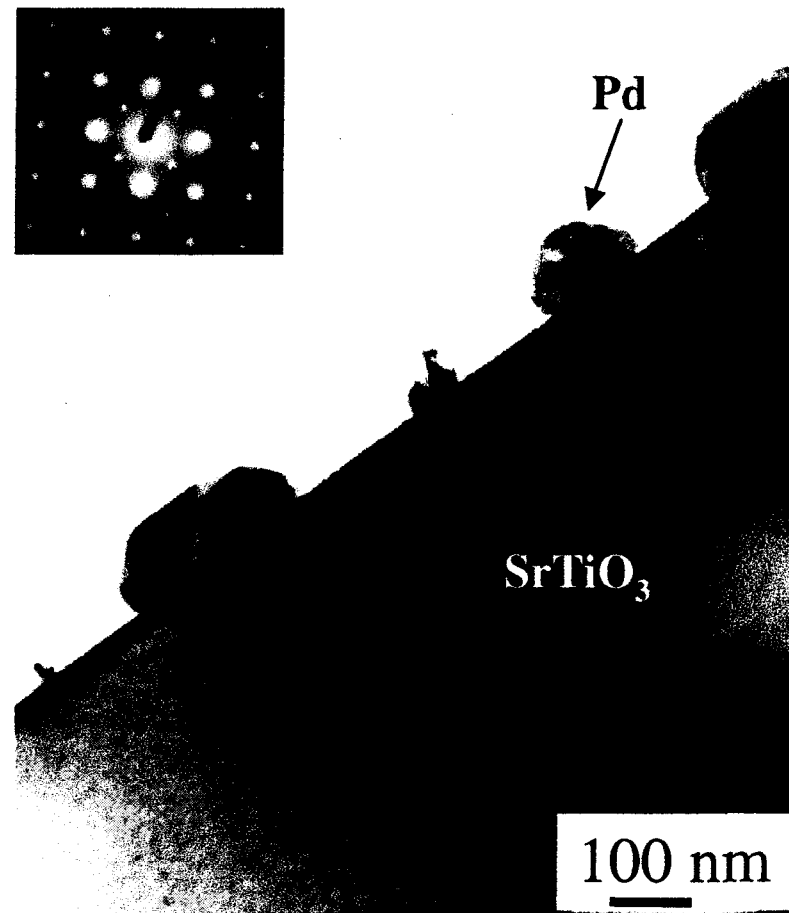
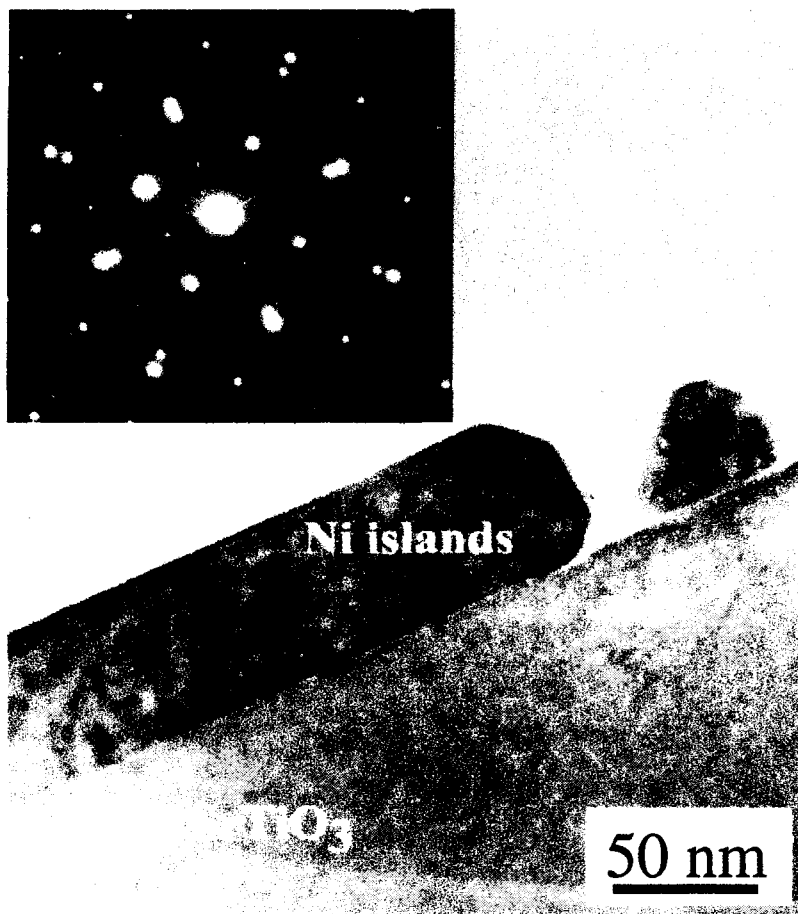
Orientation relationships



$$\begin{aligned} \{100\}_{\text{Cr}} &\parallel \{100\}_{\text{SrTiO}_3} \\ \langle 110 \rangle_{\text{Cr}} &\parallel \langle 100 \rangle_{\text{SrTiO}_3} \end{aligned}$$

$$\begin{aligned} \{110\}_{\text{Mo}} &\parallel \{100\}_{\text{SrTiO}_3} \\ \pm \langle 100 \rangle_{\text{Mo}} &\parallel \langle 100 \rangle_{\text{SrTiO}_3} \end{aligned}$$





Orientation relationship

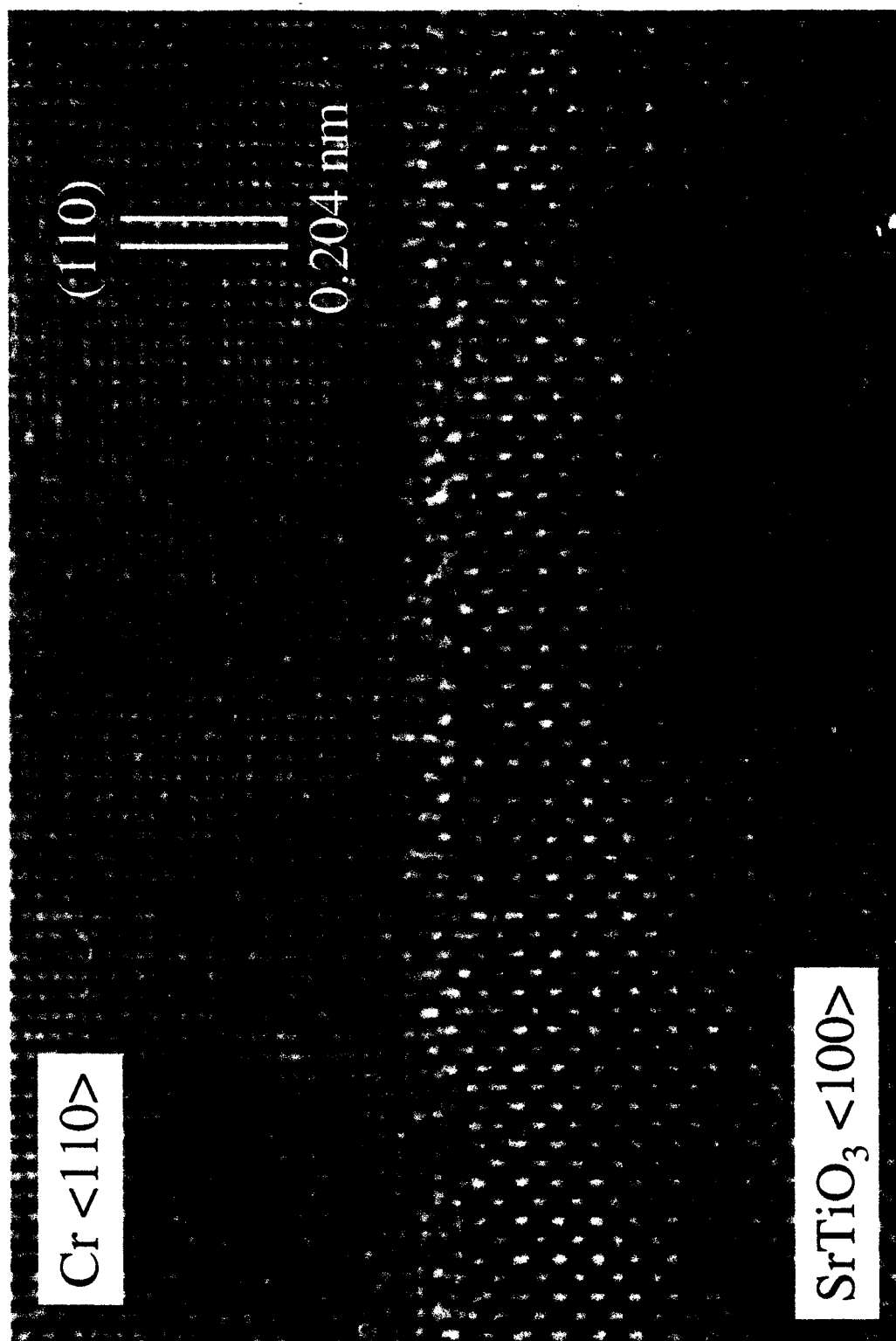
$$\{100\}_{\text{Metal}} \parallel \{100\}_{\text{SrTiO}_3}$$

$$\langle 100 \rangle_{\text{Metal}} \parallel \langle 100 \rangle_{\text{SrTiO}_3}$$

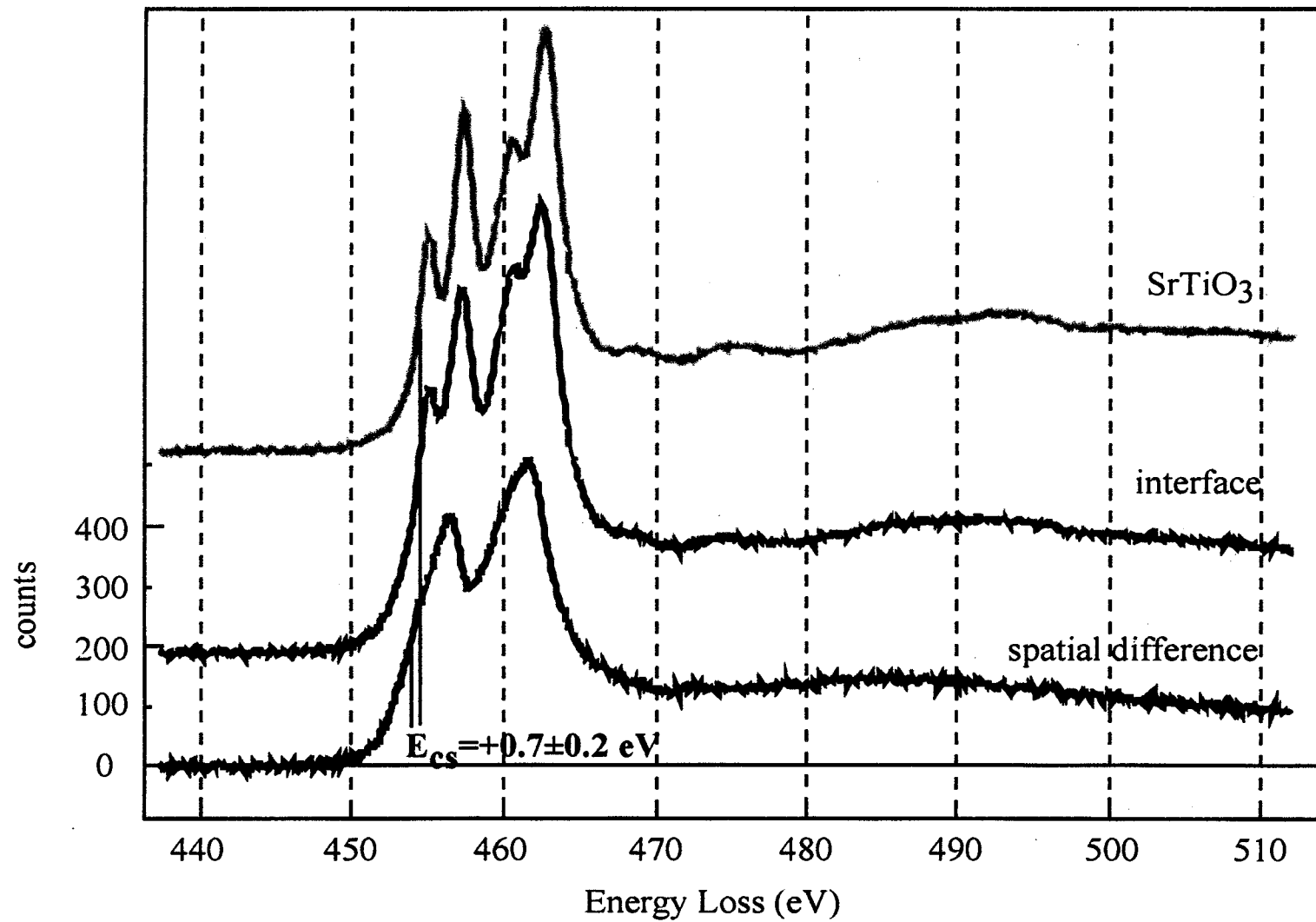


Metal

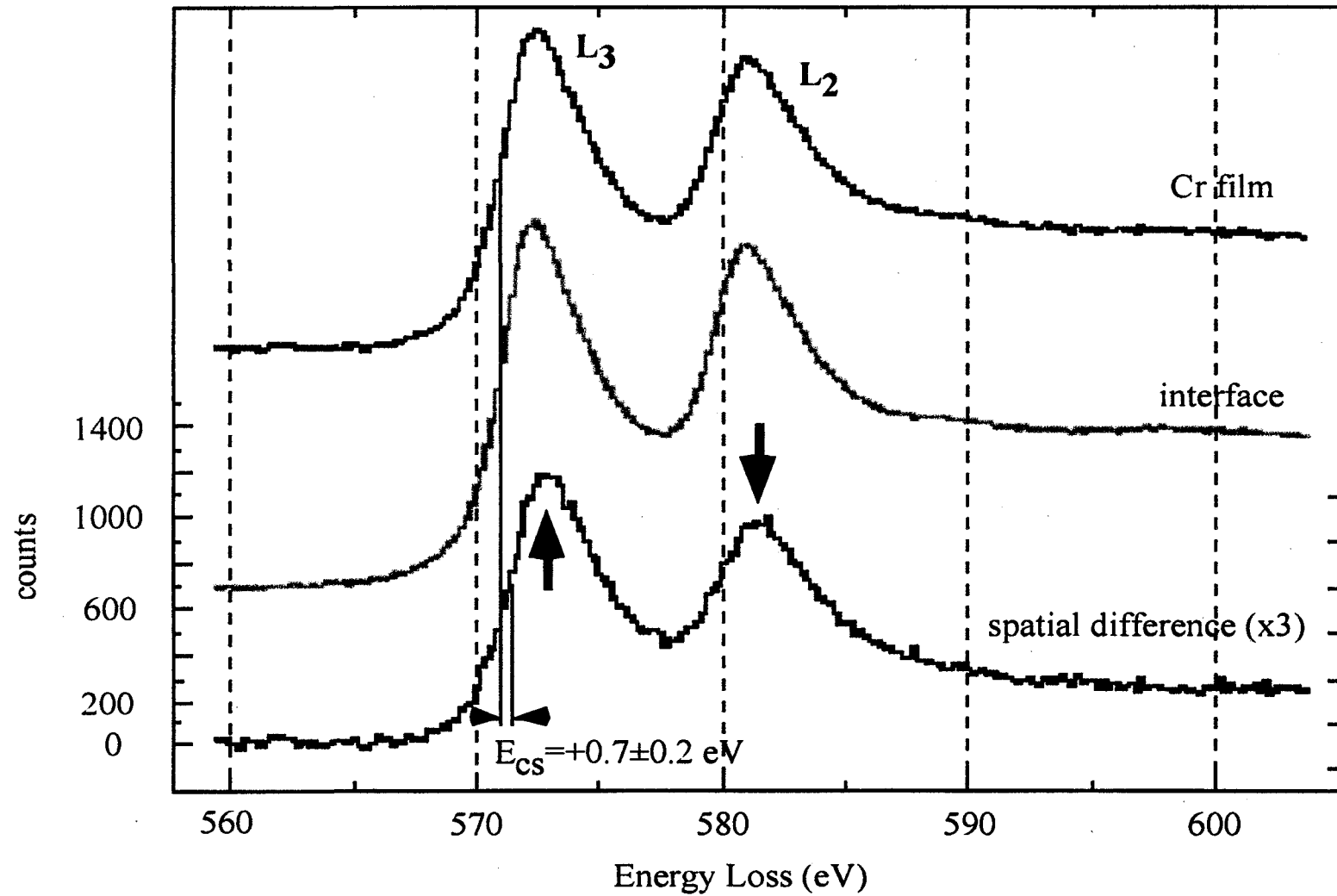
SrTiO₃



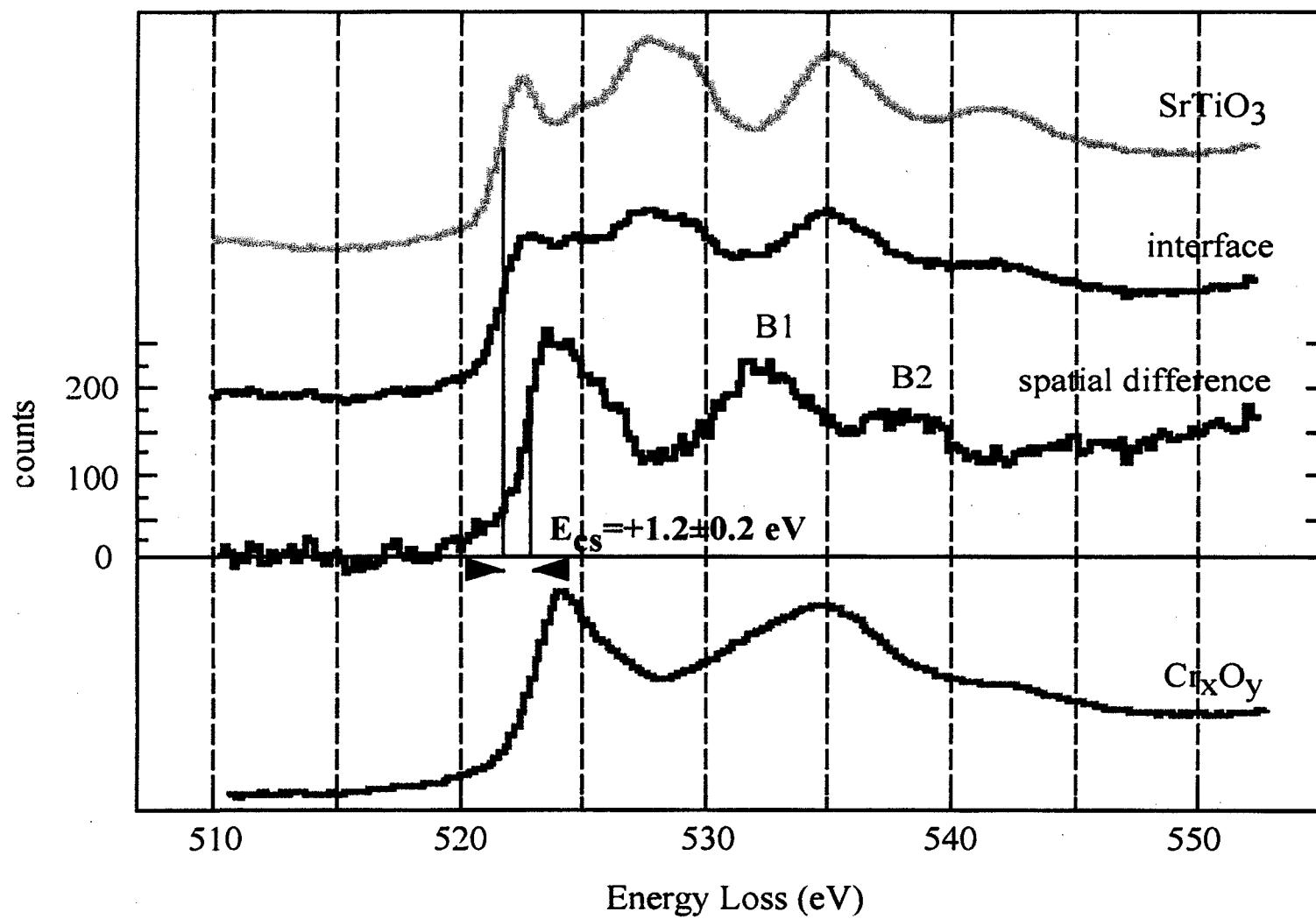
Ti $L_{2,3}$ edge of Cr/SrTiO₃



Cr $L_{2,3}$ edge of Cr/SrTiO₃



O K edge of Cr/SrTiO₃



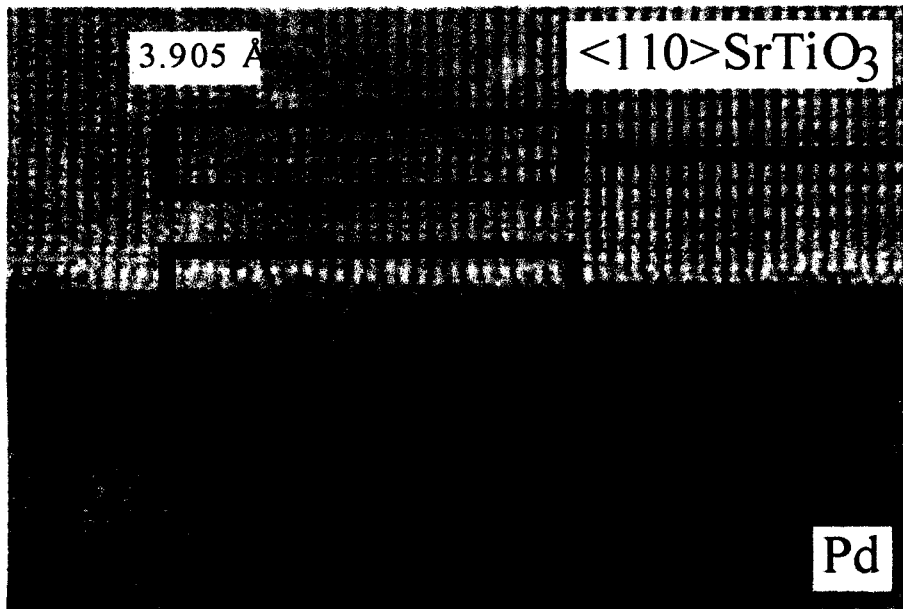
Analysis of the interfacial ELNES of Cr/SrTiO₃

- reduction of Ti at the interface
- reduced crystal field splitting
- O 2p - Cr 3d hybridization
- electron transfer from Cr to O

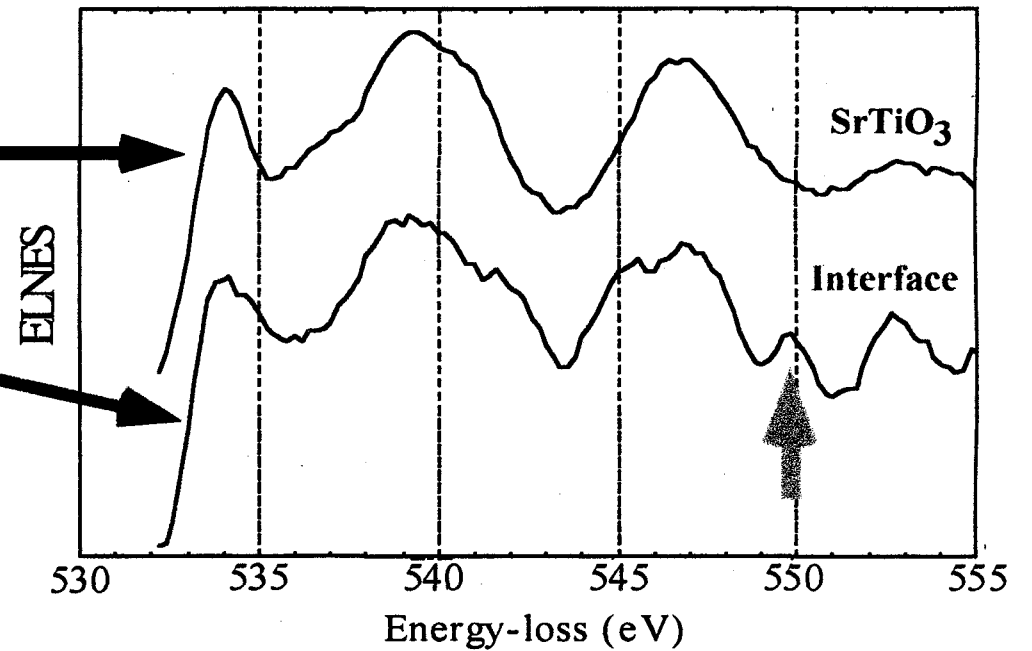
 mixed ionic-covalent bonding between Cr and SrTiO₃

O-K edge of Pd/SrTiO₃

Atomistic structure

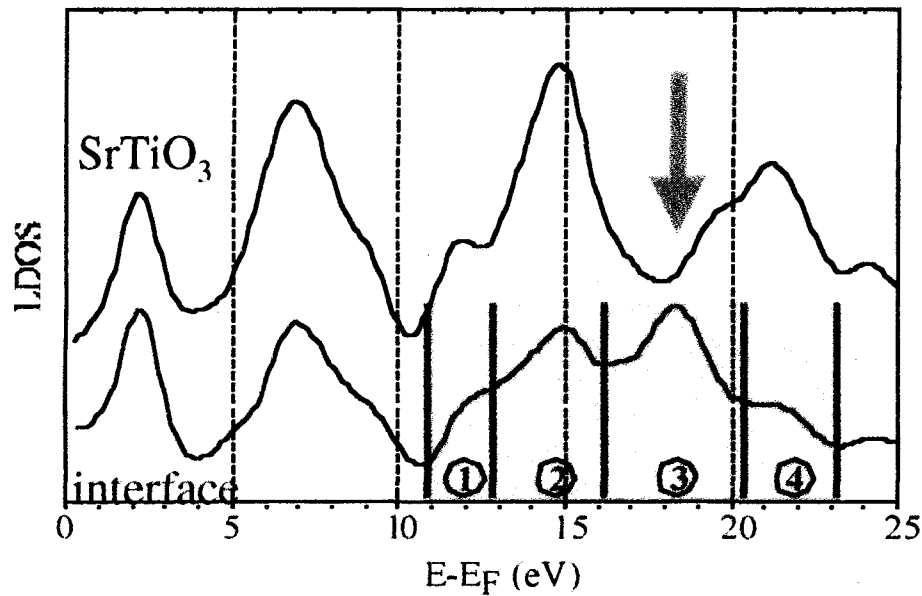


Electronic structure



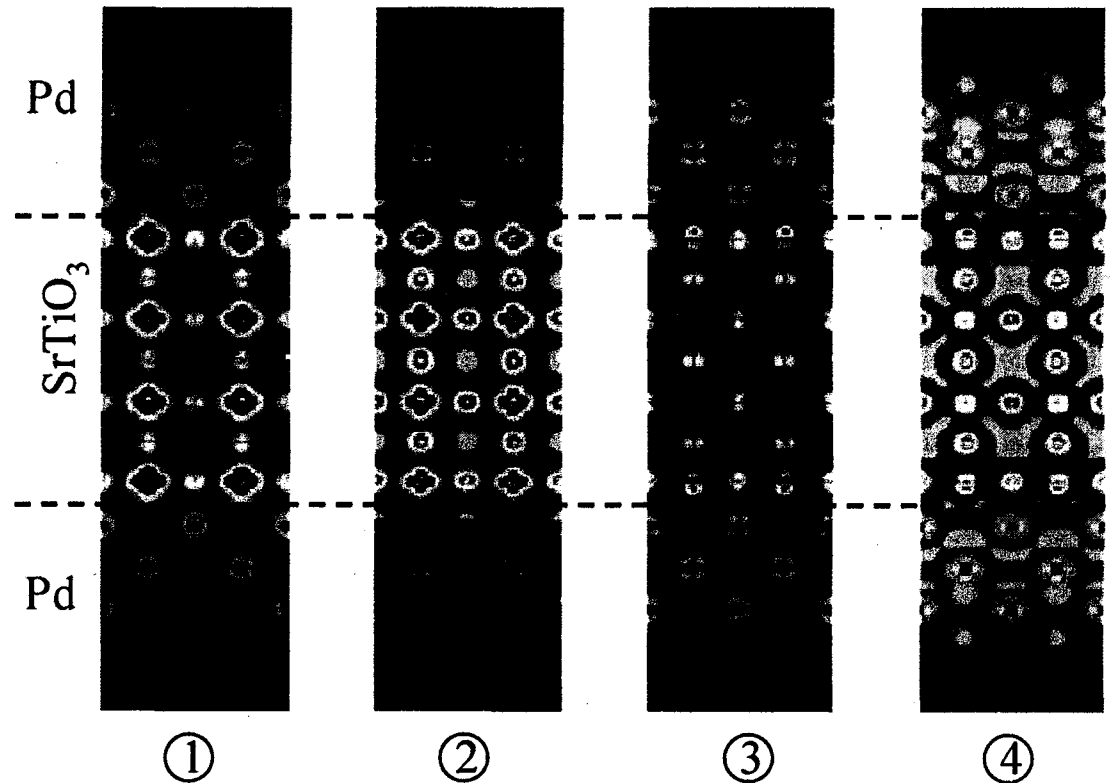
ab-initio calculations of Pd/SrTiO₃

Electronic structure



- Oxygen p-PDOS
- additional peak in region 3
- similar line-shapes in comparison to the experimental data

$|\Psi|^2$ distribution maps



- cut through Ti-O plane (i.e. 010,001-plane)
- appearance of interface specific components in region 3

K. van Benthem and C. Elsässer, unpublished

Analysis of the interfacial ELNES of Pd/SrTiO₃

- experiment and theory provide the same interfacial ELNES



- interaction between Pd and O
- Pd sits above the O of the TiO₂ terminated (001) SrTiO₃
- O 2p - Pd 4d hybridization

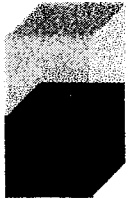


covalent bonding between Pd and SrTiO₃

Summary

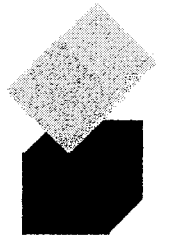
Pd/SrTiO₃ and Ni/SrTiO₃:

- epitaxial growth with cube-on-cube orientation relationship
- island growth
- the electronic structure of **Pd/SrTiO₃** is determined by experiment and theory



Cr/SrTiO₃ and Mo/SrTiO₃:

- epitaxial growth with cube-on-cube but 45° in-plane rotated for **Cr/SrTiO₃** and 45° out-plane rotated for **Mo/SrTiO₃**
- continuous film layers
- the electronic structure of **Cr/SrTiO₃** is determined by experiment



FEG (200 kV)

Monochromator
Energy-selecting
slit ($\Delta E \leq 0.2$ eV)

The Future of Microscopy

21st Century
NTEAM

C_s -Corrector

Electron Beam
 $\varnothing < 1$ Å

Object

Objective
Lens

C_s -Corrector

Energy Filter
»Mandoline«

Projector
Lens

HAADF-Detector

