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abdus salam
international centre for theoretical physics

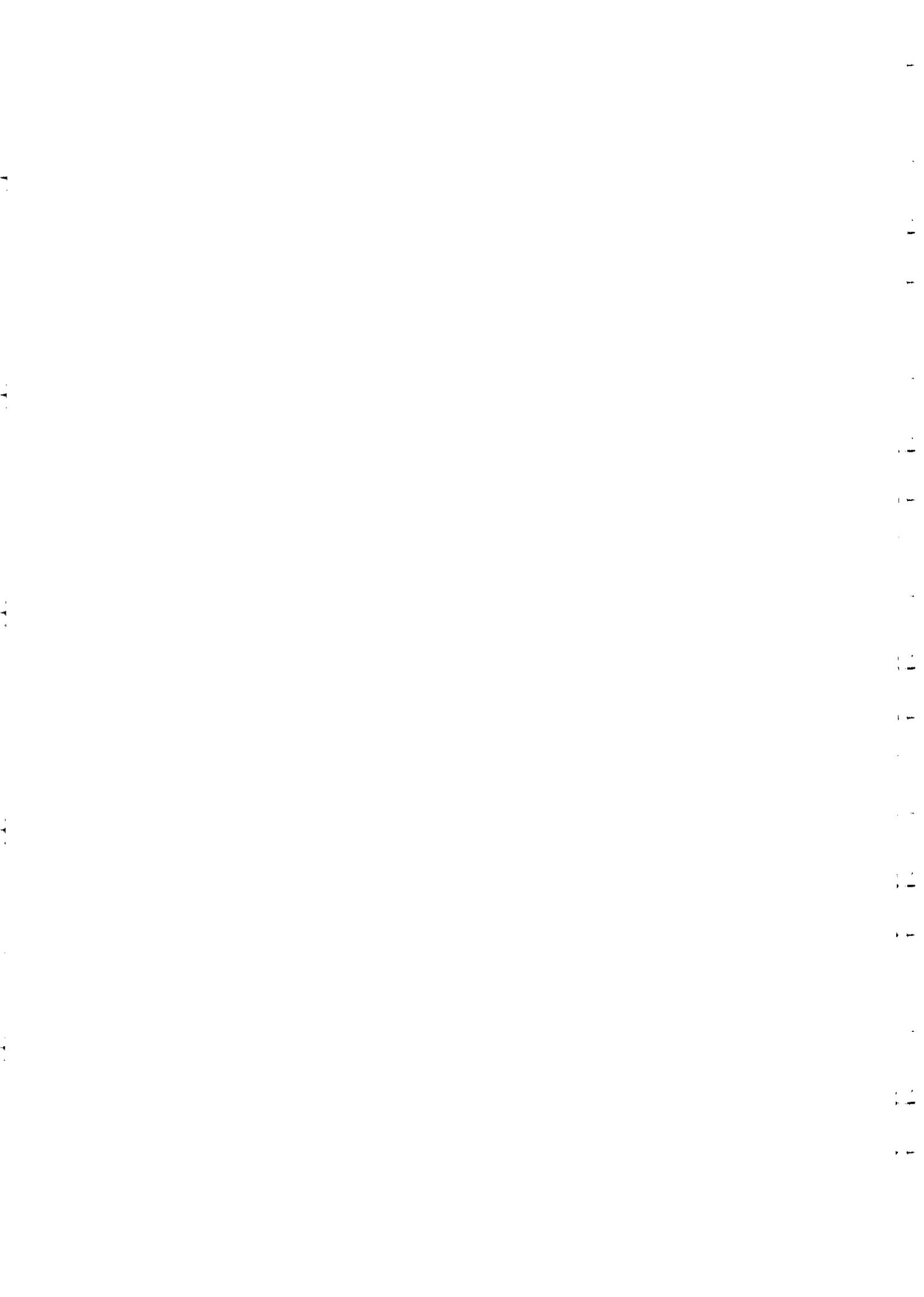
SMR.1135 - 7

**THIRD WORKSHOP ON
THIN FILMS PHYSICS AND TECHNOLOGY
(8 - 24 MARCH 1999)
including
TOPICAL CONFERENCE ON
MICROSTRUCTURE AND SURFACE MORPHOLOGY
EVOLUTION IN THIN FILMS
(24 - 26 MARCH 1999)**

**"Carbon Nitride Thin Films -
Growth and Properties"**

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These are preliminary lecture notes. intended only for distribution to participants.



**Third Workshop on Microstructure and
Surface Morphology Evolution in Thin Films
Trieste, Italy, 8-26 March 1999**

**Carbon Nitride Thin Films
- Growth and Properties**

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<http://www.ifm.liu.se/Thinfilm/BCN>



LINKÖPINGS UNIVERSITET

Carbon Nitride Thin Films - Growth and Properties

Niklas Hellgren



LINKÖPINGS UNIVERSITET



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Dr. Hans Sjöström

Prof. Jan-Eric Sundgren

Prof. Lars Hultman

et al....

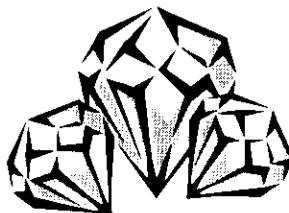


Outline

- **Background**
 - **Hard Materials**
 - **Carbon Nitrides, β -C₃N₄**
- **Thin Film Deposition - Magnetron Sputtering**
- **Structure & Mechanical Properties of CN_x Thin Films**
- **Influence of Deposition Conditions on the Properties of CN_x**
 - **Role of Nitrogen**
 - **Influence of Plasma Parameters**
- **Multilayers & Nanostructured CN_x films**
- **Applications of CN_x**
- **Summary**

Background - Hard Materials

Hardness = A materials ability to resist plastic and elastic deformation when exposed to an external load



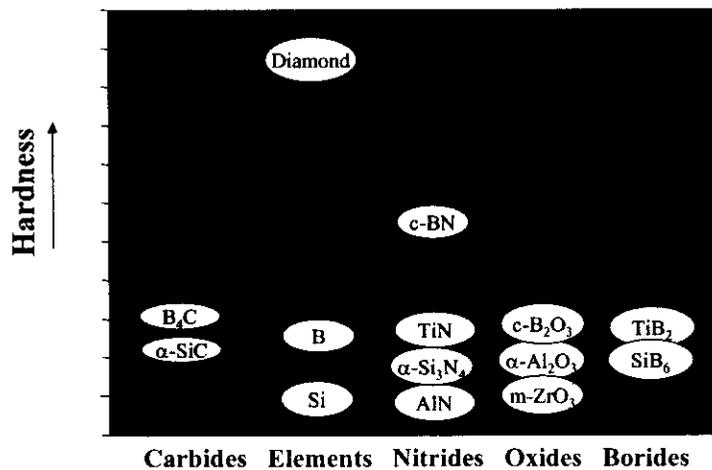
Hard Materials

Requirements for forming a hard solid:

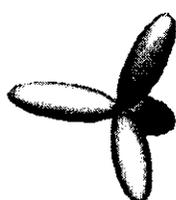
- 1 Small molar volume => Small atoms, closely packed
- 2 Strong bonds => Covalent or ionic bondings
- 3 Crystal structure => 3-dimensional network

Suitable materials: B, C, N, ... e.g. Diamond & c-BN

Hard Materials

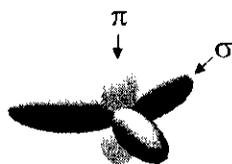


Carbon Bond Hybridization



sp^3

**Diamond
(3D)**



sp^2

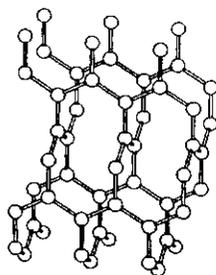
**Graphite
(2D)**



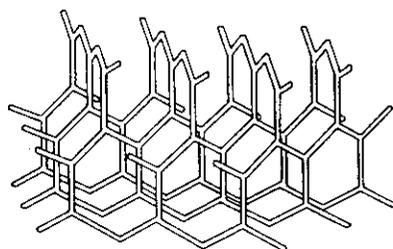
sp^1

**Organic
compounds
(1D)**

sp^2 -Carbon Networks



Bct-4

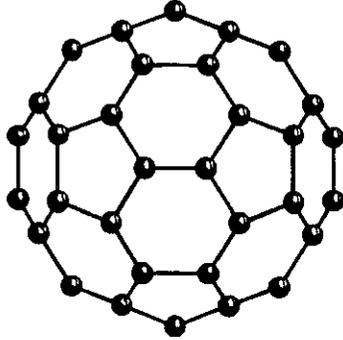


H-6

Predicted hardness comparable to c-BN

Fullerenes

C_{60}



Estimated bulk modulus
of a single C_{60} molecule:
843 GPa

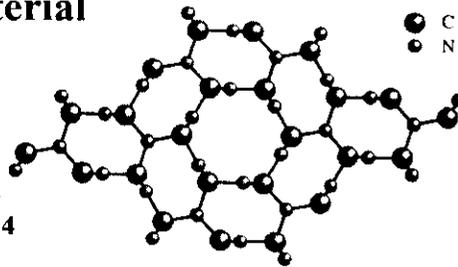
(diamond: 443 GPa)

However, crystalline
 C_{60} -films have weak van
der Waals-bonds
between molecules =>
weak solid

$\beta-C_3N_4$

New predicted material

$\beta-C_3N_4$ has the
hexagonal
structure of $\beta-Si_3N_4$



Calculated bulk-modulus
427 GPa

Liu & Cohen, Science 245 (1989) 841

Bulk Modulus

Semi-empirical relationship:

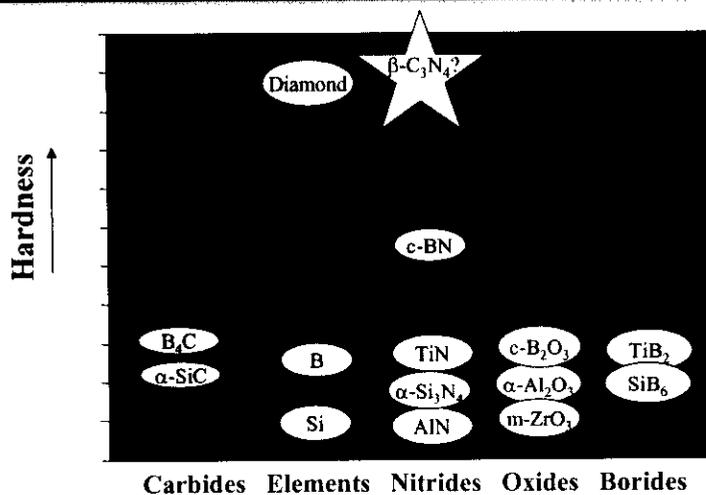
$$B = \frac{(19.71 - 2.20\lambda)}{d^{3.5}}$$

λ = ionicity (empirical parameter ranging from 0 to 2)
 d = bond length (Å)

Compound	Calculated B (GPa)	Measured B (GPa)
Diamond	444	435
c-BN	367	369
β -Si ₃ N ₄	265	256
β -C ₃ N ₄	427	???

M.L. Cohen, J. Hard Materials, vol. 2, No. 1-2 (1991) 13-27

Hard Materials



Techniques used for CN deposition

- **RF/DC Magnetron Sputtering**
- **Ion beam assisted deposition (IBAD)**
- **Arc evaporation**
- **Laser ablation**
- **Nitrogen implantation**
- **Chemical vapor deposition (CVD)**
- **Wet chemistry**
-

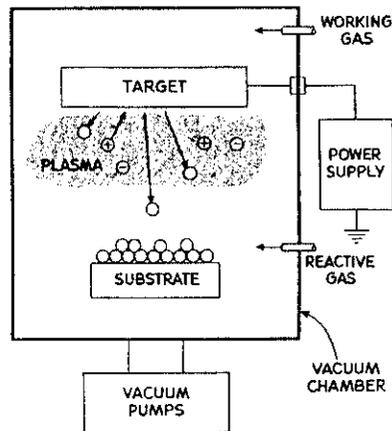
Difficult to synthesize β -C₃N₄

Thin Film Growth

- **Growth occurs as a result of condensation of vapor (atoms, clusters, ions, molecules, radicals, etc.) on a surface**
- **The growth conditions (energy input, chemical reactions etc.) defines the microstructure of the film**
- **The microstructure defines the properties of the film and hence the performance of the component**

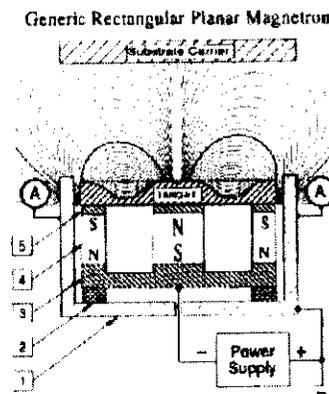
Sputtering

- Ions are accelerated to the negatively biased target
- If sufficient energy => target atoms are knocked out - Sputtering
- Atoms are transported through the vapor and condense on the substrate - Film growth



Magnetron sputtering

- Electrons are trapped in the discharge by means of a magnetic field
- Increased electron path length => electrons undergo more ionizing collisions => higher degree of ionization => Higher sputtering yield

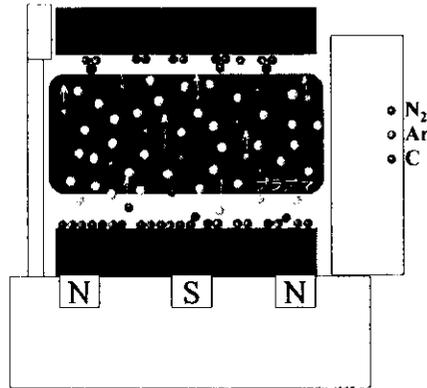


Magnetron sputtering of CN_x

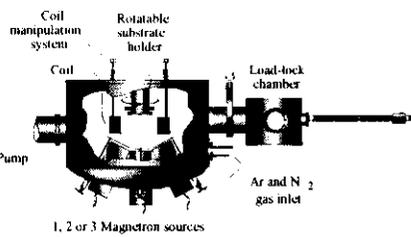
- Carbon nitride films can be grown by reactive sputtering from a graphite target in N_2 or N_2/Ar discharge

Film structure influenced by:

- Pressure and gas mixture
- Energy and flux of particles impinging on the growth surface
- Substrate temperature T_s
- ...



UHV deposition system



Film Characterization

Growth & Plasma
 Surface Profilometry
 Langmuir Probe
 Mass-spectrometry

Composition:
 RBS
 NRA

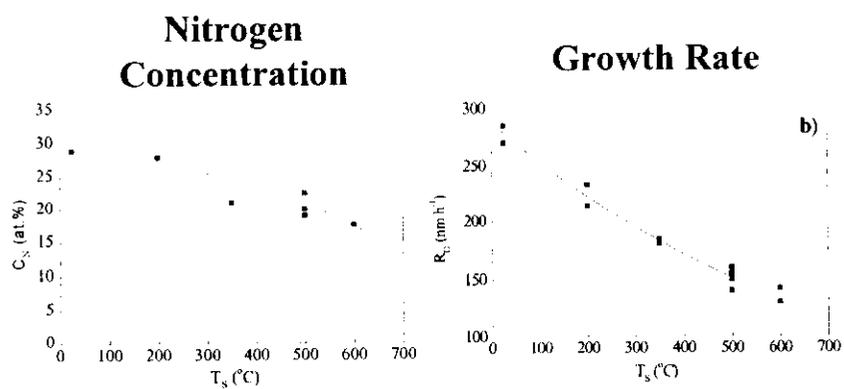
Bondings:
 XPS
 FTIR
 Raman
 EELS
 NEXAFS
 XAS

Structure & Morphology:
 SEM
 TEM
 AFM
 STM

Mechanical:
 Nanoindentation
 Pin-on-disc
 Surface Acoustic Wave
 Stress

Optical & Electrical:
 Ellipsometry
 Four-point probe

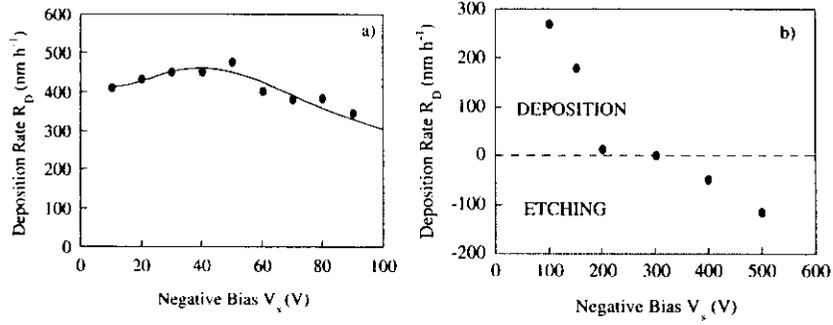
Effect of Substrate Temperature



Max obtained Nitrogen concentration ~ 35 at.%
57 at.% N required for C_3N_4

Effect of Substrate Bias

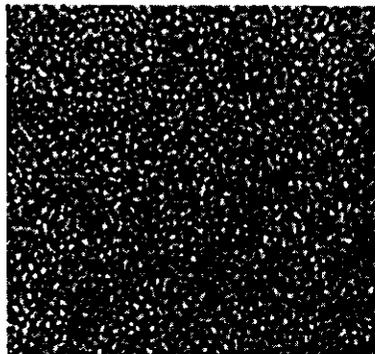
Growth Rate



No film growth for $V_b > 200$ V

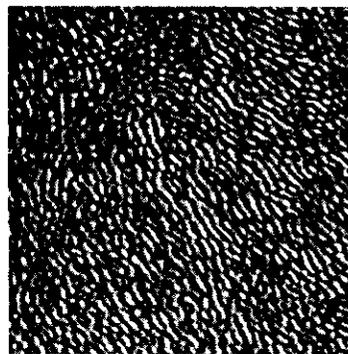
Microstructure of CN_x ($x < 1.33$)

$T_s < 150$ °C



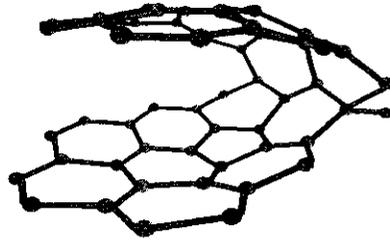
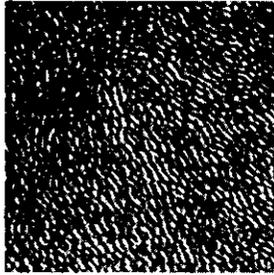
Amorphous

$T_s > 200$ °C



"Fullerene-like"

CN_x - Proposed Microstructure

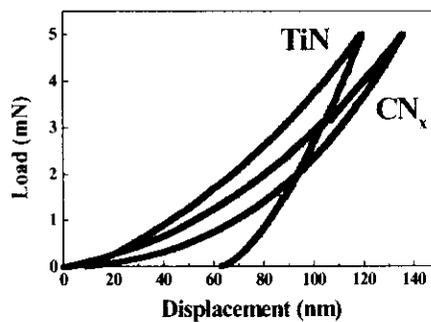


Bent and intersecting basal-planes!

The bending might be caused by the incorporation of pentagons in the otherwise hexagonal basal planes, just like in the case of fullerenes => "Fullerene-like" microstructure

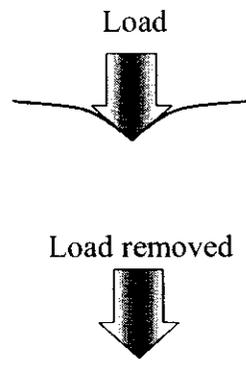
Sjöström *et al.*, Phys. Rev. Lett. 75 (1995) 1336

Nanoindentation

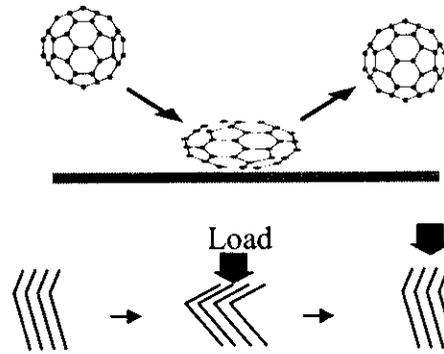


H = 40 - 60 GPa

E = ~ 200 GPa



Deformation Mechanism



Bond angle deformation - Not bond breaking

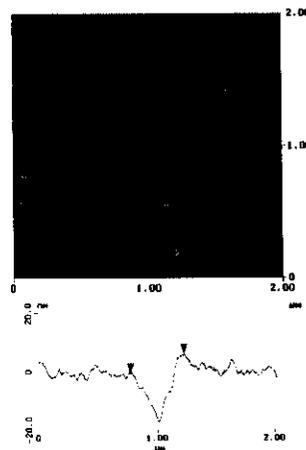
Deformation Mechanisms

Before



Surface structure conserved in the indent!

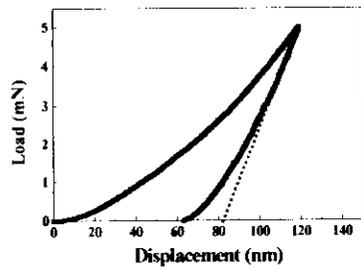
After



Niklas Hellgren, TMR Meeting, March 99

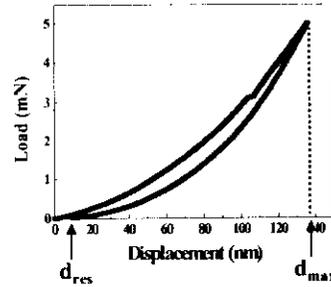
Nanoindentation

"Normal"
Elasto-plastic deformation



H & E may be deduced!

CN_x
Plastic deformation

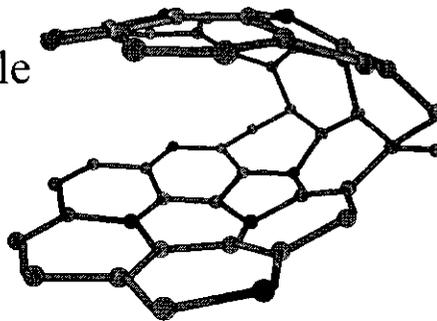


H & E = ??
Instead: d_{\max} & R

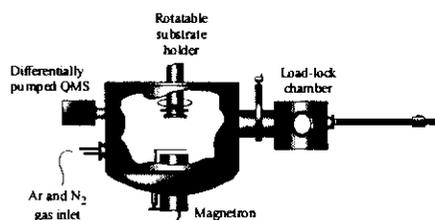
The Role of N in Hard & Elastic CN_x

Objectives:

To study which role nitrogen plays in promoting the fullerene-like microstructure to form



Film Growth



$P_b = 1 \times 10^{-9}$ Torr
 $P_{tot} = 3.0$ mTorr (Ar+N₂)
 $I_t = 0.3$ A
 $E_i \sim 10 - 15$ eV
Thickness = 0.5 μ m

**Unbalanced
d.c. magnetron
sputtering from
graphite target**

$T_s = 100, 350, 550$ °C
N₂-fraction: 0 - 100%

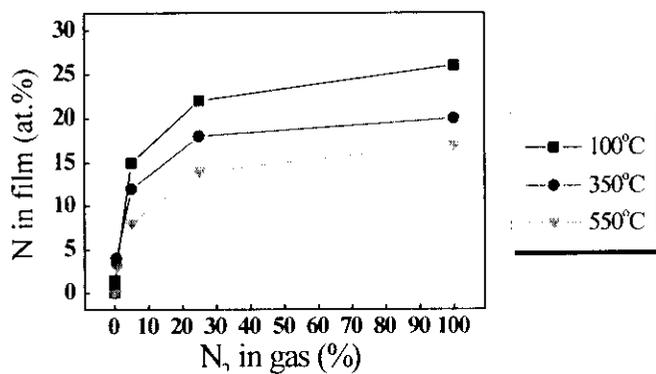
Film Analysis

Composition: RBS

Structure: XPS
REELS
Raman spectroscopy
TEM
SEM

Mechanical: Nanoindentation

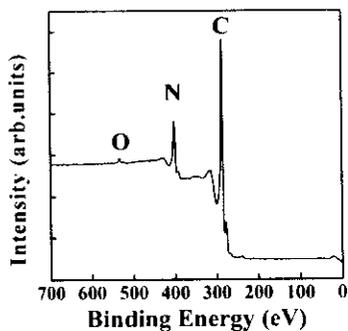
Composition



Saturation of C_N Higher $T_S \Rightarrow$ Lower C_N

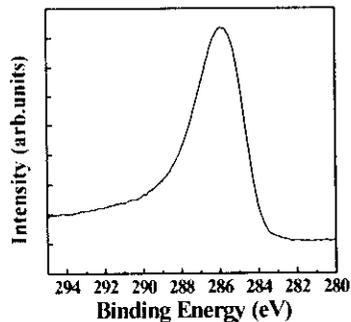
XPS

Overview



< 2 at% Oxygen

C 1s

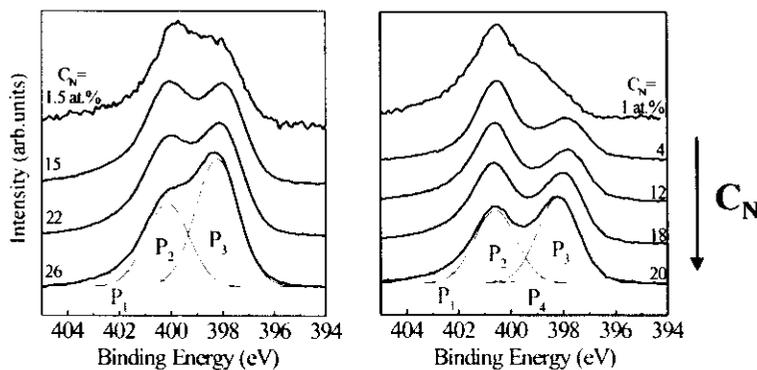


Deconvolution difficult

XPS N1s

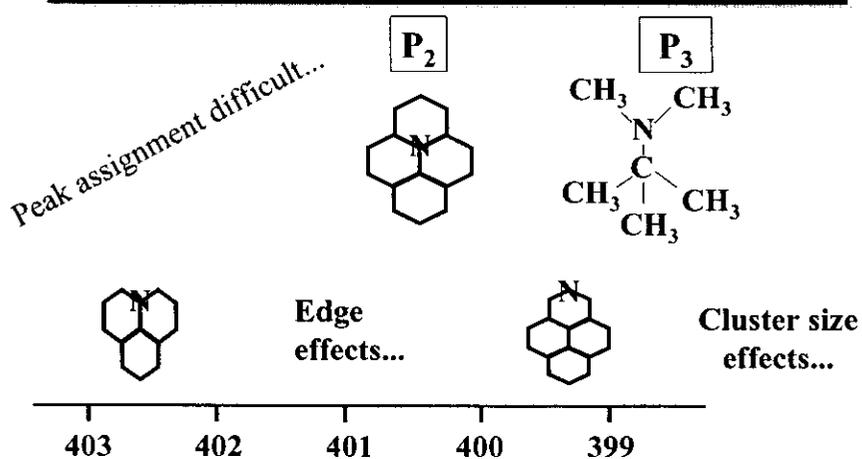
$T_s = 100\text{ }^\circ\text{C}$

$T_s = 350\text{ }^\circ\text{C}$



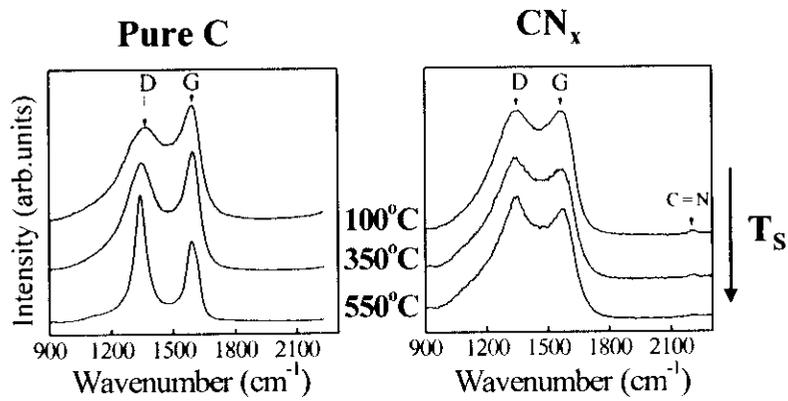
Assignments: P_2 N - sp^2 C P_3 N - sp^3 C

Calculated N1s binding energies



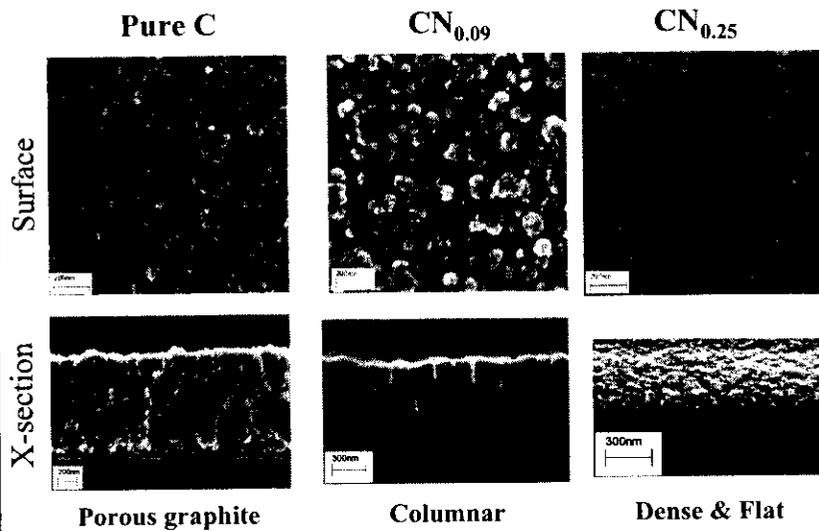
Å. Johansson & S. Stafström, *to be published*

Raman Spectroscopy



Higher T_s & Lower C_N ⇒ More ordered structure

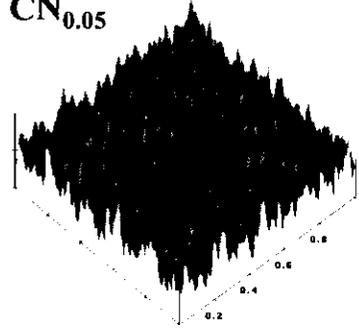
SEM



AFM Surface roughness

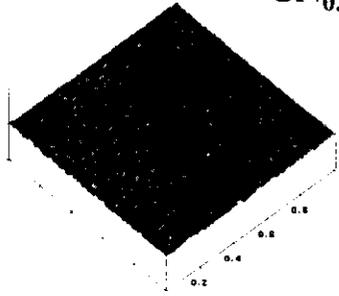
$T_s = 350\text{ }^\circ\text{C}$

CN_{0.05}



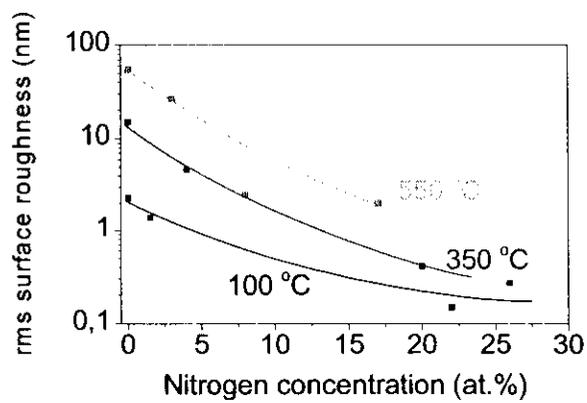
Rms roughness 4.7 nm

CN_{0.25}



0.41 nm

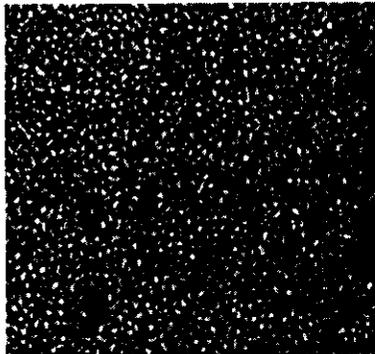
AFM Surface roughness



Lower T_s & More Nitrogen \Rightarrow Smoother film

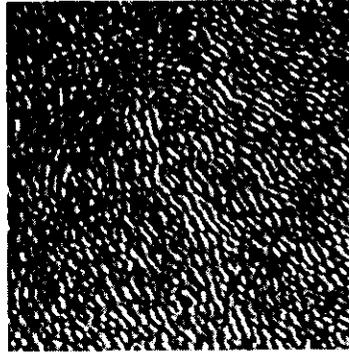
TEM Plan-views (CN_x)

100 °C



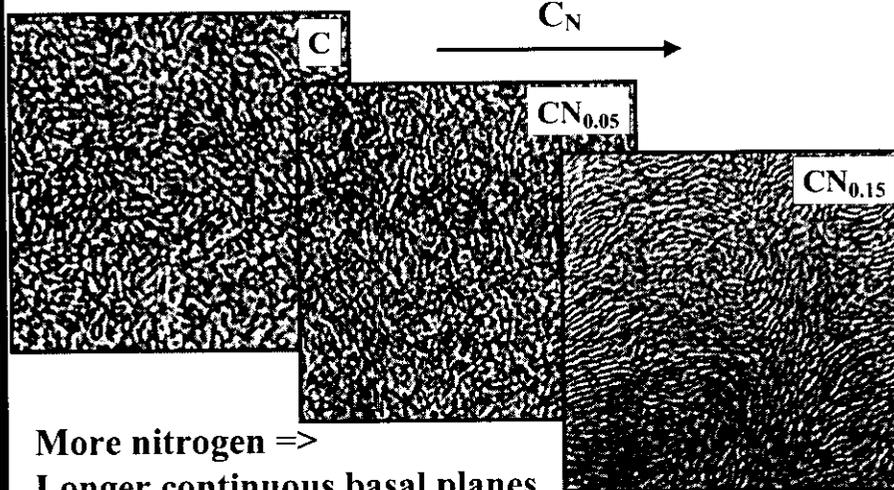
Amorphous

350 °C



"Fullerene-like"

TEM Plan-views, $T_S > 200$ °C



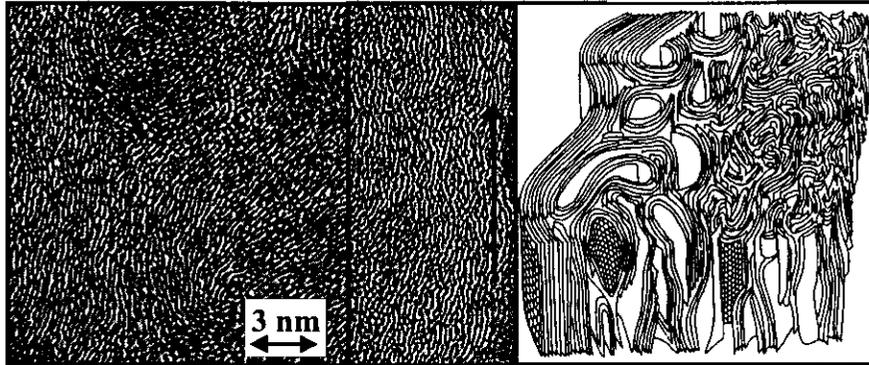
More nitrogen =>
Longer continuous basal planes

Structure of CN_x ($0.15 < x < 0.3$), $T_s > 200$ °C

Plan-View

X-View

Schematics

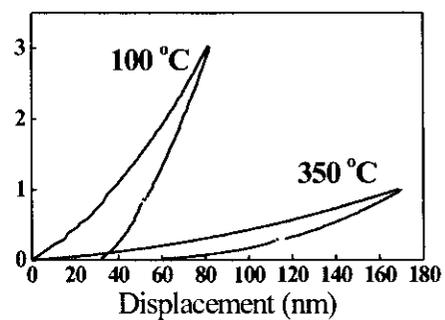
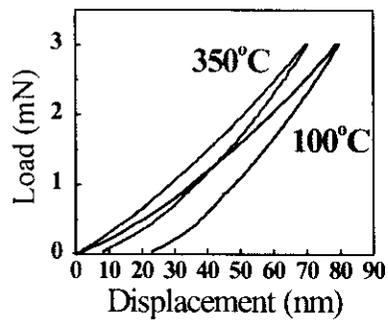


Schematic image of turbostratic carbon from S.C. Bennet and D.J. Johnson, Carbon 17 (1979) 25

Nanoindentation

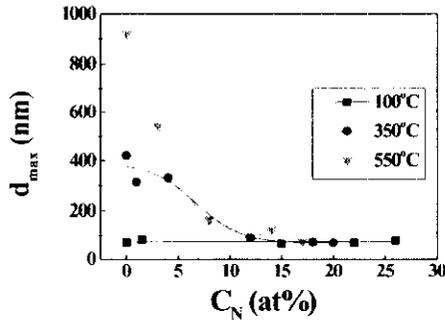
CN_x

Pure C

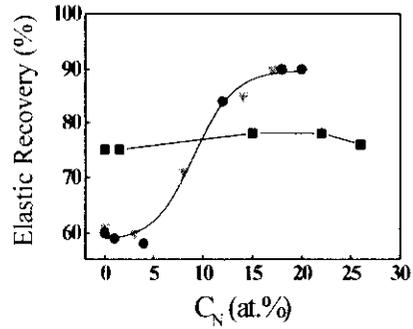


Nanoindentation

Displacement @ 3 mN

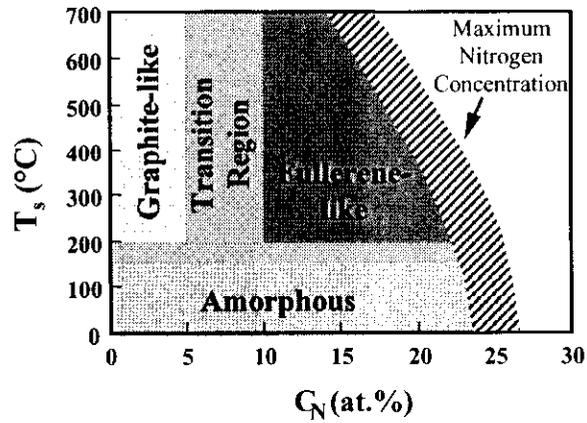


Elastic Recovery



3 characteristic regions

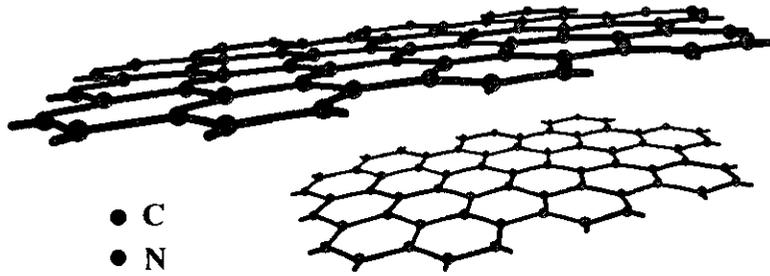
"Phase"-diagram



Structures ($T_s > 200\text{ }^\circ\text{C}$)

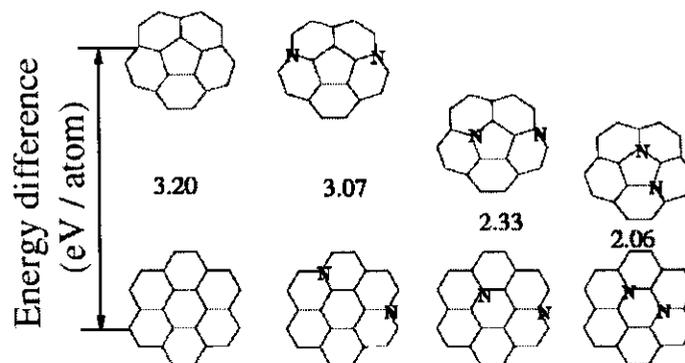
$C_N < 5\text{ at.}\%$

Graphite-like domains



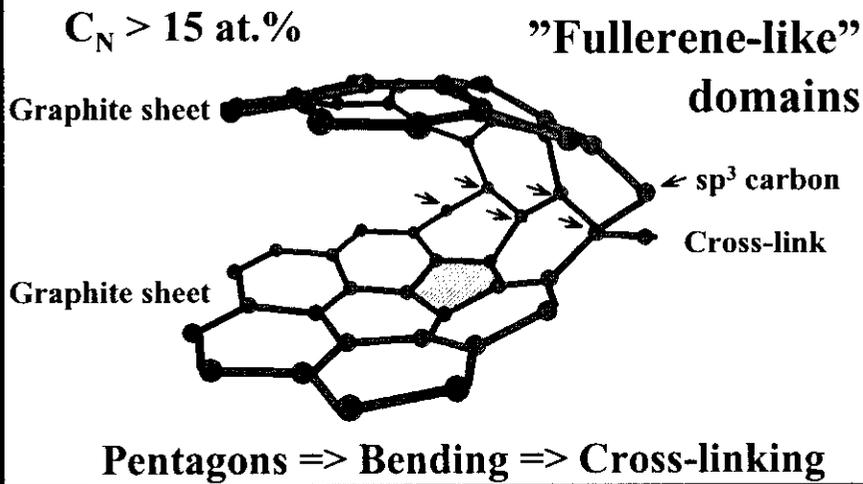
Nitrogen in substitutional graphite sites

Energy to form Pentagons

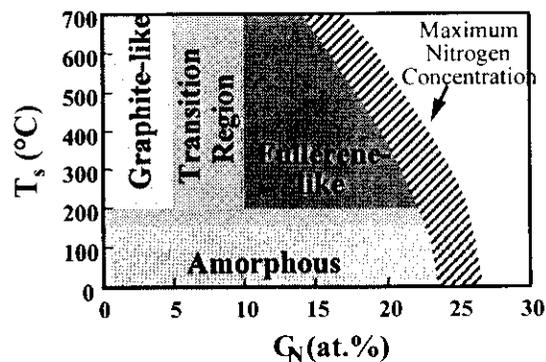


Sjöström *et al.*, Phys. Rev. Lett. **75** (1995) 1336

Structures ($T_s > 200\text{ }^\circ\text{C}$)



Conclusions



Nitrogen does play an important role in the formation of hard and elastic CN_x thin films !!!

Influence of plasma parameters on the growth and properties of magnetron sputtered CN_x thin films

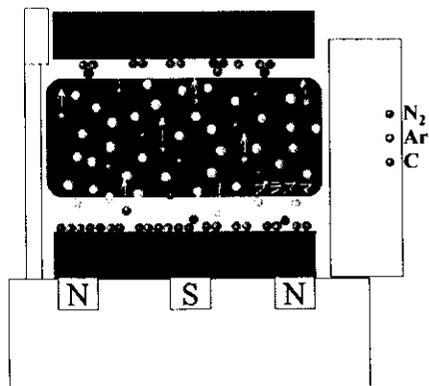
Objectives:

- Properties sensitive to process conditions => Diverging reports in the literature => Systematic study required
- Ion-assisted deposition may help to lower the critical T_s for forming the hard and elastic “fullerene-like” phase

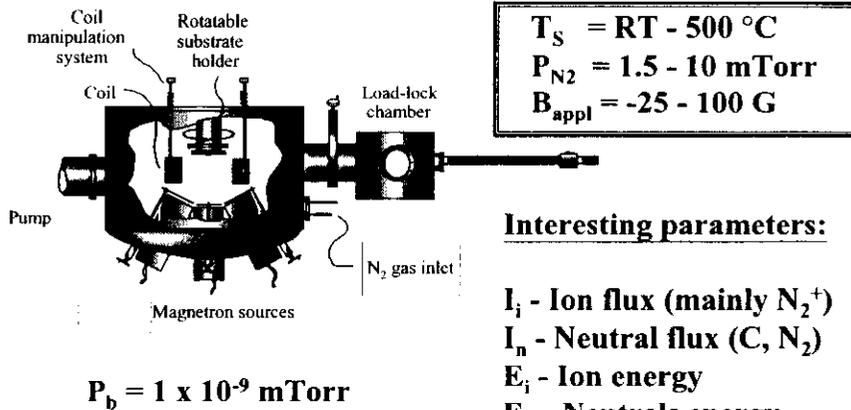
Magnetron Sputtering

Interesting parameters:

- I_i - Ion flux (mainly N_2^+)
- I_n - Neutral flux (C, N_2)
- E_i - Ion energy
- E_n - Neutrals energy
- T_s - Substrate temperature



Sputtering conditions



$T_s = RT - 500\text{ }^\circ\text{C}$
 $P_{N_2} = 1.5 - 10\text{ mTorr}$
 $B_{\text{appl}} = -25 - 100\text{ G}$

$P_b = 1 \times 10^{-9}\text{ mTorr}$

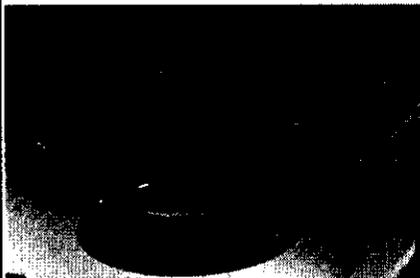
Interesting parameters:

- I_i - Ion flux (mainly N_2^+)
- I_n - Neutral flux (C, N_2)
- E_i - Ion energy
- E_n - Neutrals energy
- T_s - Substrate temperature



Niklas Hellgren, MRS 1998

The Coil



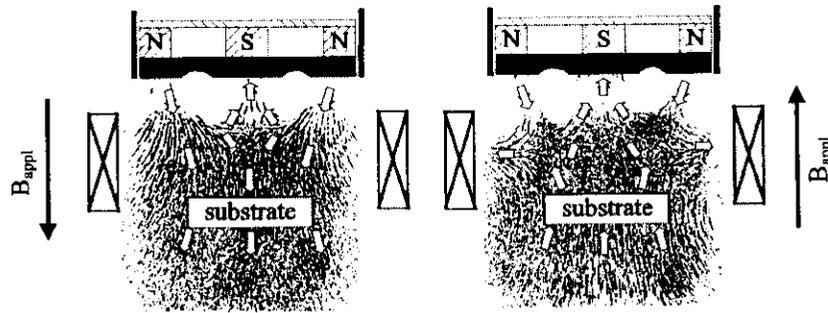
Variable magnetic field =>

Controllably-unbalanced magnetron sputtering



Niklas Hellgren, MRS 1998

Distribution of the Magnetic Field



The applied field B_{appl} supports the outer poles.

The applied field B_{appl} supports the center pole.

Controllably-Unbalanced Magnetron Sputtering



Niklas Hellgren, MRS 1998

From I. Ivanov et al., J. Vac. Sci. Technol. A 12 (1994) 314

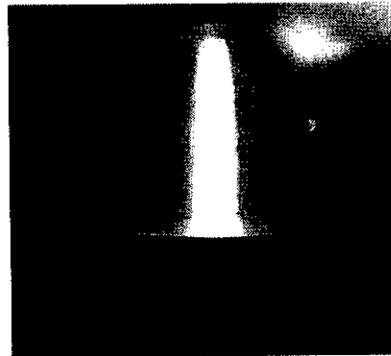
Effect on the plasma

Repulsive field: - 25 G



Weak diffuse plasma

Attractive field: +100 G

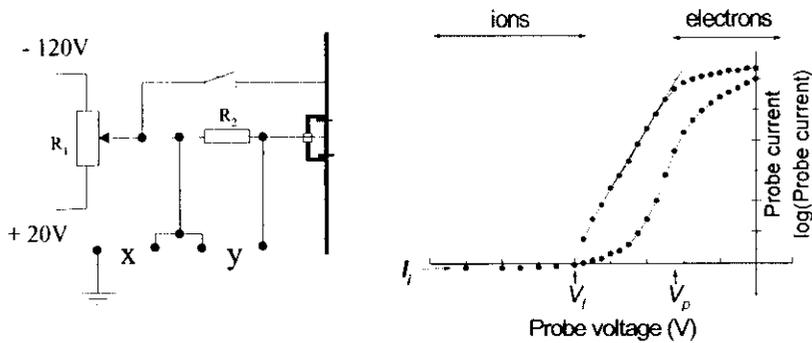


Strong, focused plasma



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Probe Measurements



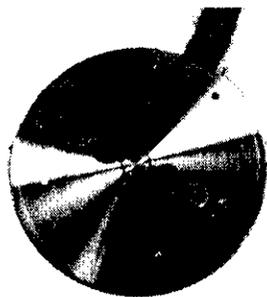
By measuring the current through a probe in the substrate position, as a function of bias voltage, the ion flux and energy can be deduced

I. Petrov *et al.*, Contrib. Plasma Phys. 28 (1988) 157



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Probe measurements



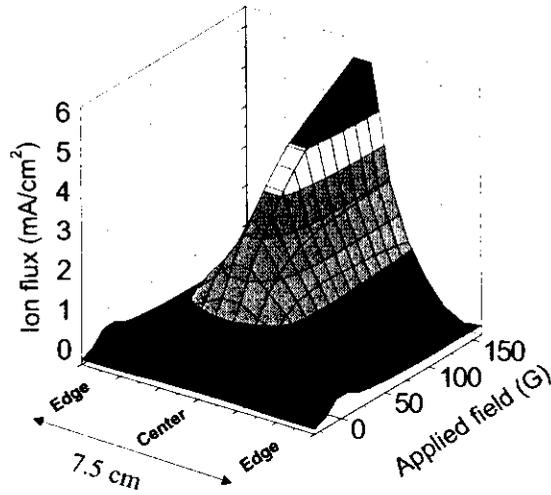
"7-point probe"

Since the conditions are not always homogenous over the substrate area, it is important to know the particle flux and energies at all positions.



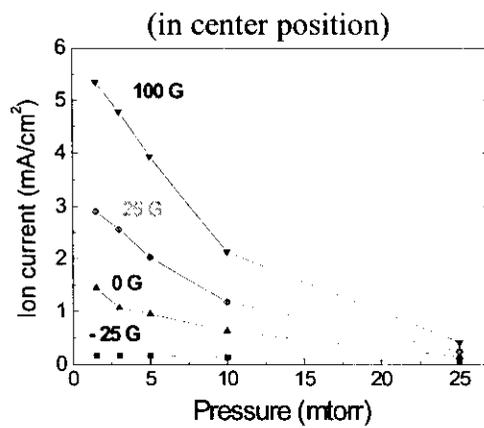
Niklas Hellgren, MRS 1998

Ion flux vs. Applied magnetic field & position



Niklas Hellgren, MRS 1998

Ion flux vs. Pressure



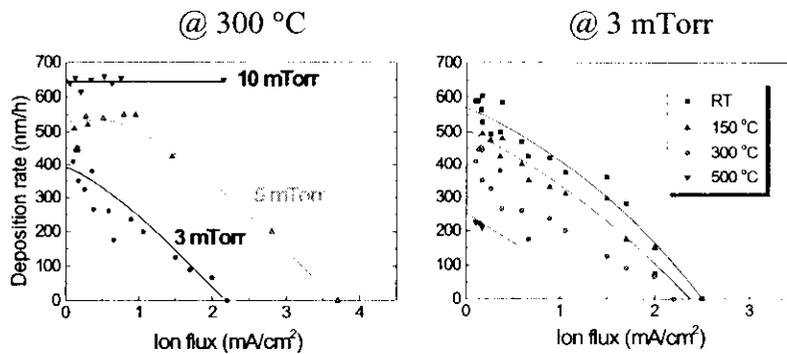
Higher pressure => Lower ion flux, due to gas scattering



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Growth rate vs. Ion flux

Ion energy $E_i = 20-25$ eV

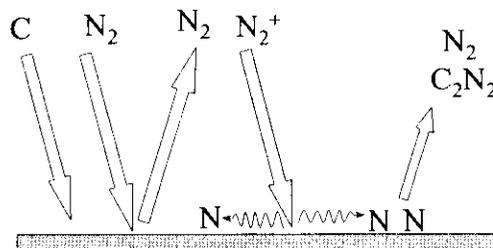


High flux, High T_s & Low pressure
 \Rightarrow More etching (+ densification)



Niklas Hellgren, MRS 1998

Suppression of growth and N-incorporation rates

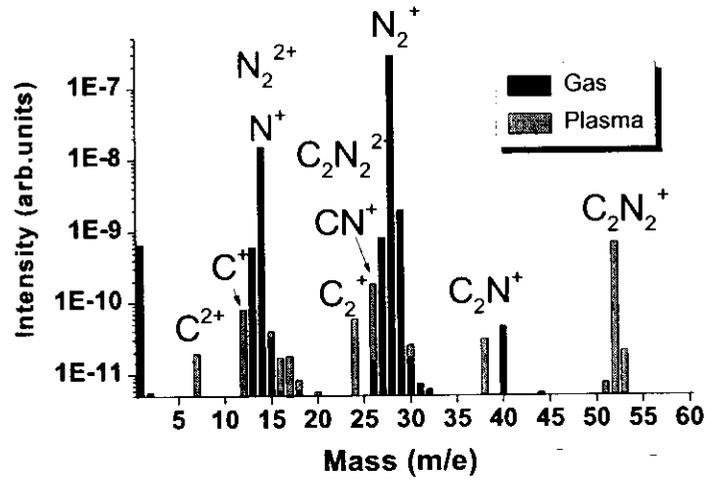


Thermal and ion-stimulated
 desorption of N₂ and CN-species



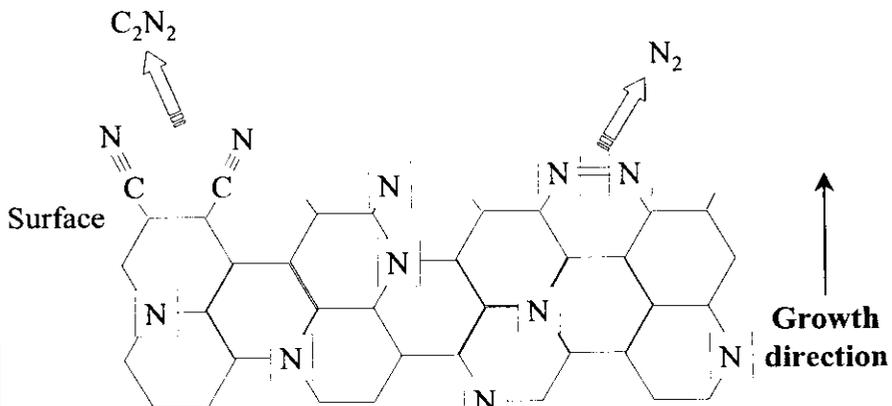
Niklas Hellgren, MRS 1998

Mass spectra



Niklas Helgren, MRS 1998

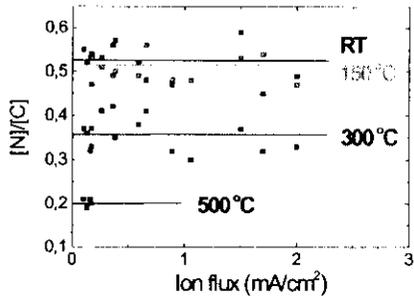
Possible mechanism for Desorption of N_2 and C_2N_2



Niklas Helgren, MRS 1998

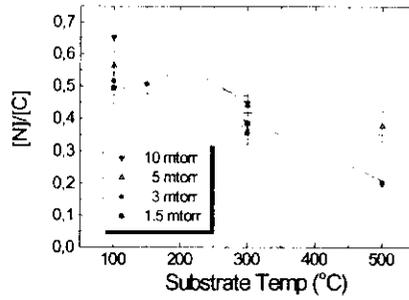
Nitrogen incorporation

vs. Ion flux
(@ 3 mTorr)



No clear effect of ion flux

vs. Temperature
(Average over all fluxes)



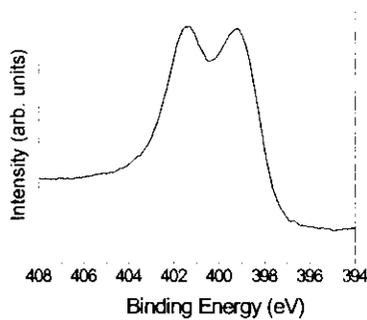
High pressure & Low T_S
=> More Nitrogen



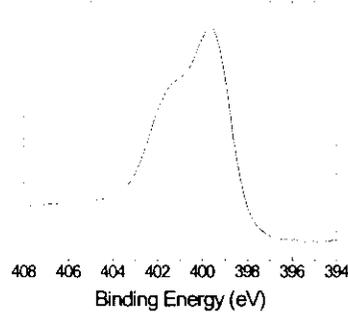
Niklas Hellgren, MRS 1998

XPS - N1s

High T_S & High flux



Low T_S & Low flux



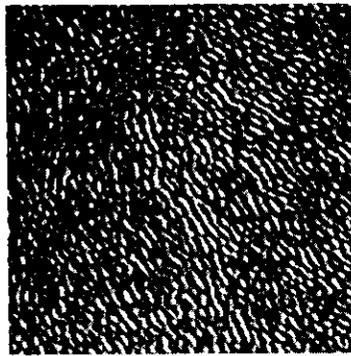
Fingerprint from "Fullerene-like" and Amorphous microstructures



Niklas Hellgren, MRS 1998

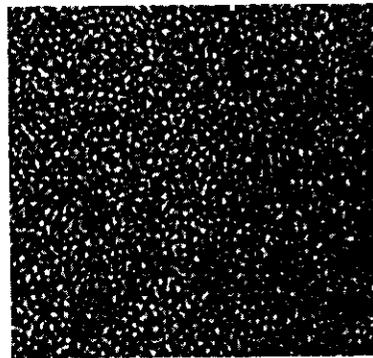
Microstructure

"Fullerene-like"



Hard & Elastic

Amorphous



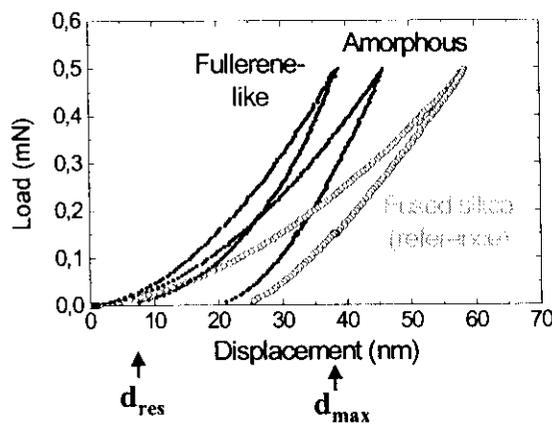
Fairly hard, but less elastic



Niklas Hellgren, MRS 1998

Nanoindentation

Load-displacement curves



High elasticity =>
Difficult to calculate
E & H

Instead compare
max displacement d_{max}
and elastic recovery

$$R = (d_{max} - d_{res}) / d_{max}$$

High ion flux & ~ 300 °C => High elasticity

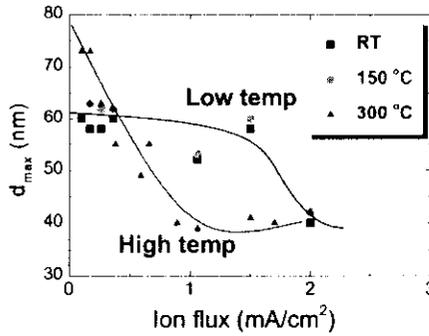


Niklas Hellgren, MRS 1998

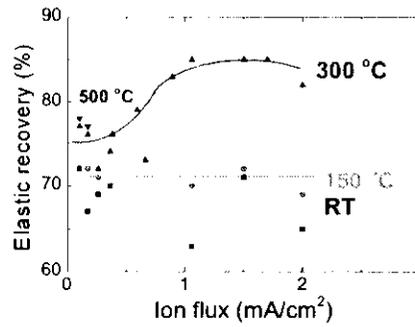
Nanoindentation

Elastic recovery @ 0.5 mN

Displacement @ 0.5 mN



Elastic recovery

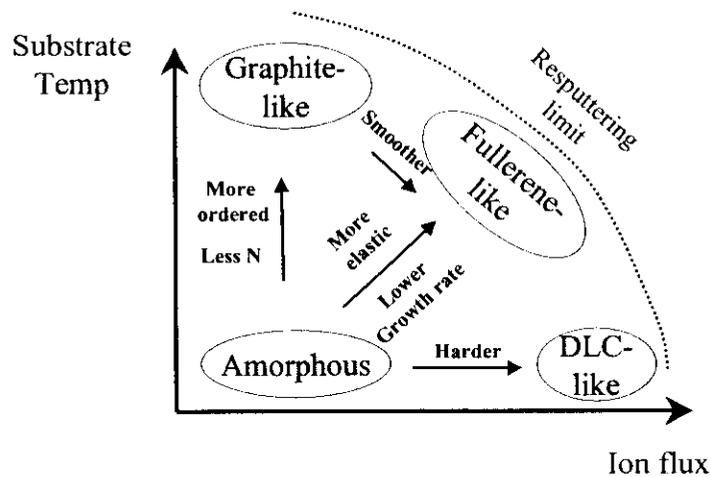


Low pressure, high T_s and high ion flux
 \Rightarrow Hardest and most elastic films



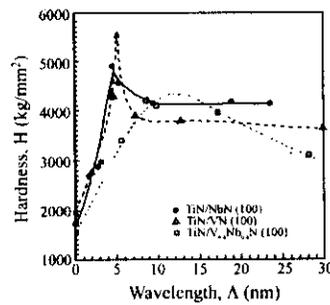
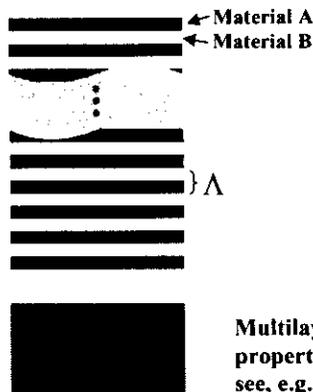
Niklas Hellgren, MRS 1998

Summary - Growth of CN_x thin films



Niklas Hellgren, MRS 1998

Other ways to grow hard materials - Multilayers



Multilayered structures may have improved mechanical properties compared to the single layered films. see, e.g., Shinn *et al.*, J. Mater. Res. 7 (1992) 901

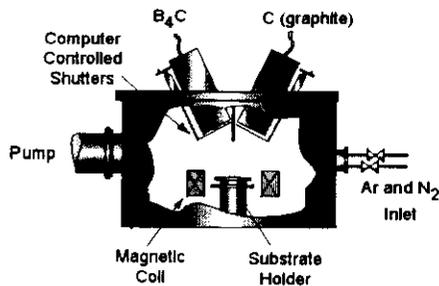
Also internal stresses may be reduced => better adhesion

CN_x/BN:C Multilayers

Objectives:

- Grow multilayers of hard and elastic materials (fullerene-like CN_x and BN:C)
- Find proper growth conditions at which CN_x and BN:C can be sequentially deposited
- Study the structure and mechanical response of as-grown structures

Film growth & characterization



Growth conditions:

C and B₄C targets

$P_b = 1 \times 10^{-7}$ Torr

$P_{tot} = 3.0$ mTorr

(Ar+N₂, 60+40%)

$I_i = 0.2$ A

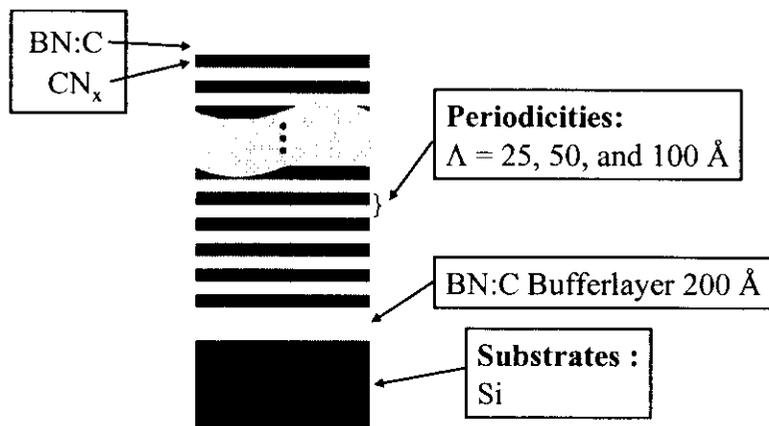
$j_i/j_n \sim 3$, $E_i \sim 20$ eV

Thickness = 0.5 μ m

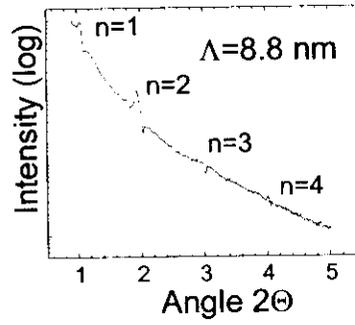
$T_s = 250$ °C

Characterization: RBS, XRD, TEM, Nanoindentation

As-grown Structures



XRD

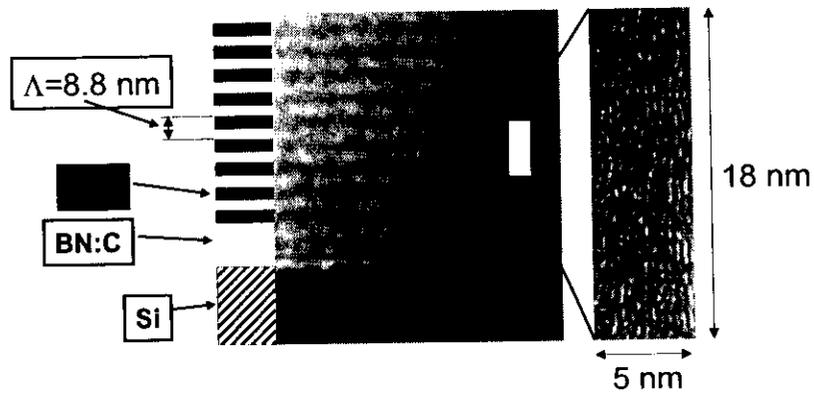


Linear regression of $\sin^2\theta$ versus n^2

$$\text{slope} = \left(\frac{\lambda}{2 \times \Lambda}\right)^2$$

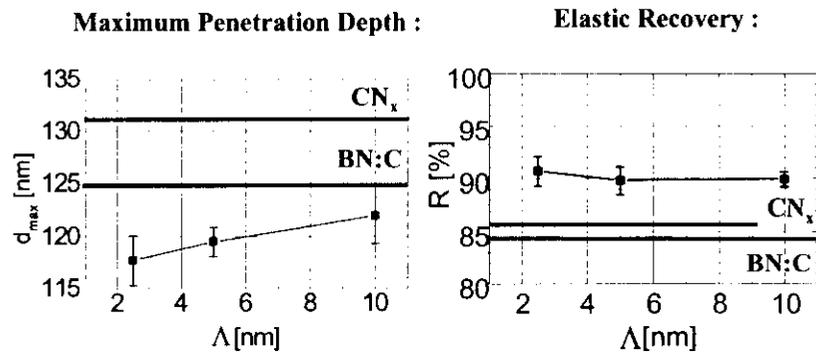
Well defined layer-interfaces

Microstructure



Well defined layer-interfaces, BUT continuous microstructure

Nanoindentation

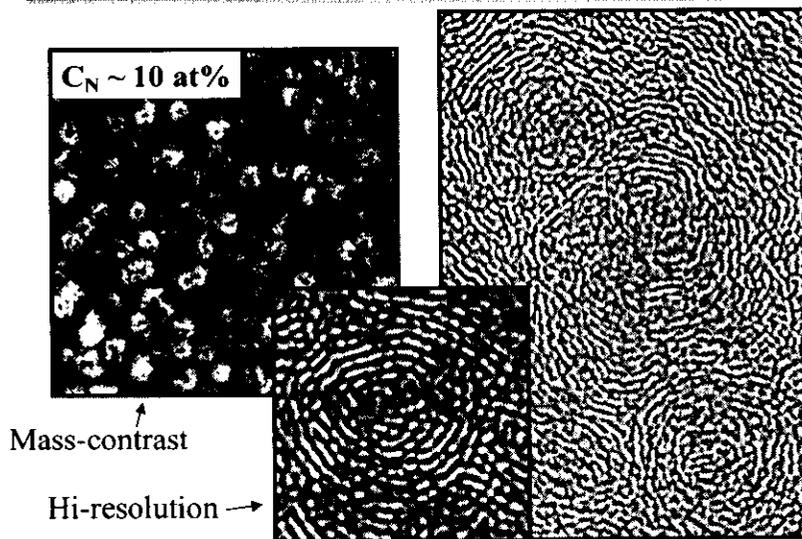


Multilayers => Slightly improved mechanical properties

Conclusions - BN/CN Multilayers

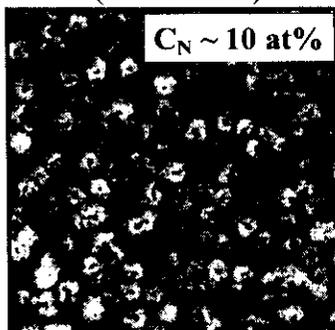
- Possible to grow CN_x /BN:C Multilayers
- Well defined layer interfaces
- Continuous microstructure over the layer interfaces
- Slightly improved mechanical properties

CN Nanotubes - HREM

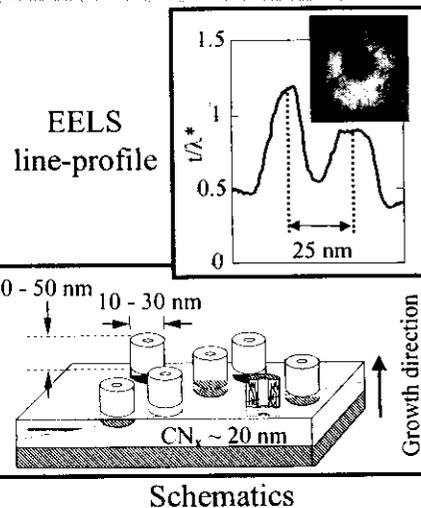


Carbon nitride "nanotubes"

Annular Dark Field Image
(Z - contrast)



Densely packed tubules
 $\sim 1 \times 10^4 \mu\text{m}^{-2}$



K. Suenaga, *et al.*, "Carbon Nitride Nanotubulite" Chem. Phys. Lett. 300 (1998) 695

Summary - magnetron sputtered CN_x

Growth and structural evolution of CN_x thin films is strongly dependent on CHEMICAL ETCHING during growth

Mainly influenced by:

- Substrate Temperature
- Fluxes & energies of deposited particles
- Nitrogen concentration

Difficult to grow homogenous films

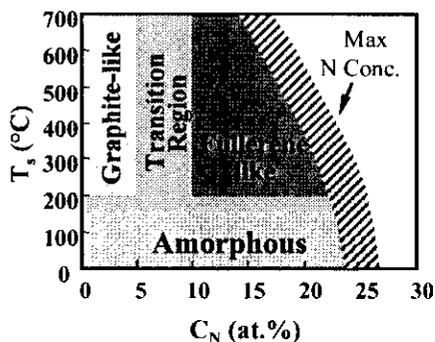
=> be careful with interpretations



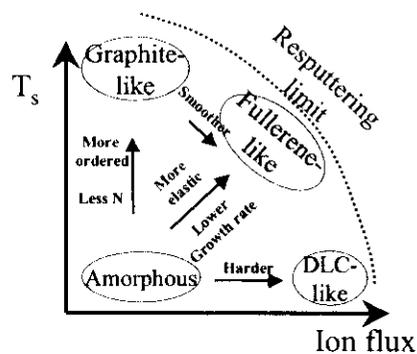
Niklas Hellgren, TMR Meeting, March 99

Summary - magnetron sputtered CN_x

Role of Nitrogen



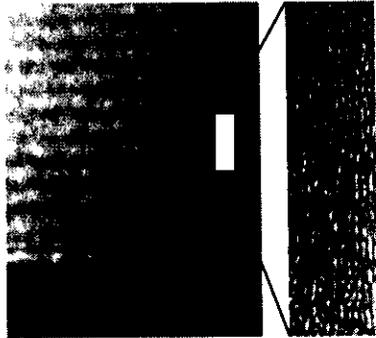
Role of Ion Flux



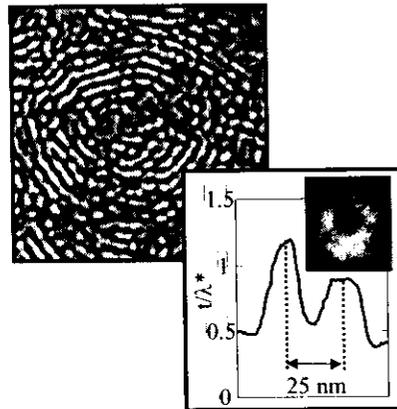
Niklas Hellgren, TMR Meeting, March 99

Summary - magnetron sputtered CN_x

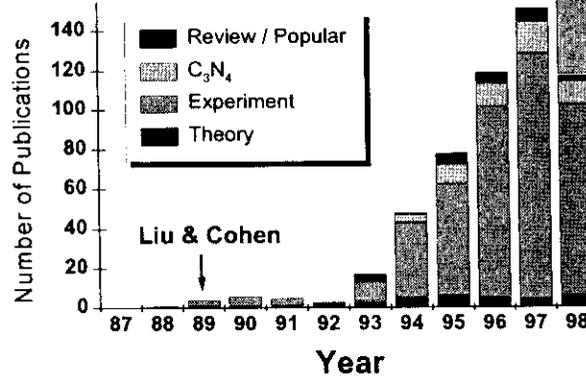
$CN_x/BN:C$ Multilayers



CN nanostructures



CN publications



Liu & Cohen started a world-wide effort to try to synthesize crystalline C_3N_4 films. However, few and incomplete reports claiming success

Difficulties in C_3N_4 synthesis

- Most efforts have resulted in nitrogen deficient (< 40 at.%) and/or amorphous films.
- Higher T_S required to grow crystalline films, BUT higher $T_S \Rightarrow$ lower nitrogen concentration.
- Thus C_3N_4 growth very difficult (impossible?) by PVD techniques.
- Many claims of small C_3N_4 crystals in an amorphous, nitrogen deficient matrix. Very unlikely!!
- Most promising results from hot filament chemical vapor deposition, HFCVD.
See, e.g., Chen et al., J. Crystal Growth 179 (1997) 525
- No " C_3N_4 " films have shown to be superhard \Rightarrow Grain boundaries and other structural defects also important for the overall performance.
- BUT CN_x ($x < 1.33$) has very interesting properties for technological applications!

Applications of hard and elastic CN_x Films

Wear Protective coatings on e.g.:

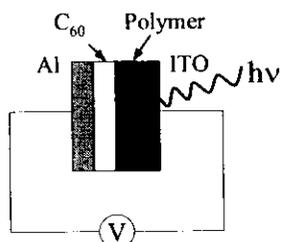
- Magnetic recording media and recording heads
(already used by IBM)
- Bearings
- Ortopaedic Implants
(e.g. hip joints)
- Scalpels, razor blades,...

However, problem with adhesion to metals. Can be solved by bufferlayers, graded interfaces, multilayers,...



Applications of Nanostructured CN_x Films

Solar Cells/ Photovoltaic Devices

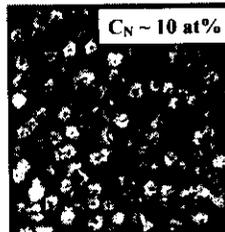


Evaporated C_{60} may be replaced by nanostructured sputtered CN_x

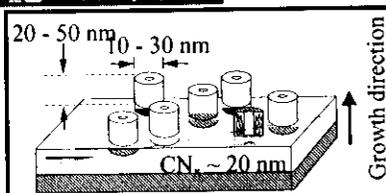


Niklas Hellgren, TMR Meeting, March 29

Electron Field Emitters



Out-standing "CN nano-tubes" may be good field emitters



Conclusions

- New predicted hard material: $\beta-C_3N_4$
- Difficult to produce $\beta-C_3N_4$
- Non-crystalline CN_x ($x < 1.33$) has many interesting properties
- Many possible applications of CN_x
- Deposition conditions need to be controlled!