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international centre for theoretical physics

SMR.1135 - 10

**THIRD WORKSHOP ON
THIN FILMS PHYSICS AND TECHNOLOGY
(8 - 24 MARCH 1999)**
including
**TOPICAL CONFERENCE ON
MICROSTRUCTURE AND SURFACE MORPHOLOGY
EVOLUTION IN THIN FILMS
(24 - 26 MARCH 1999)**

**"H-mediation, surface segregation, ultra-high doping
and their effects on Si and $Si_{1-x}Ge_x$ growth kinetics
during GS-MBE and UHV-CVD"**

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**3rd ICTP/IUVSTA Workshop on Thin Film Physics
Trieste, Italy**

**H-Mediation, Surface Segregation, Ultra-High Doping
and their Effects on Si and $\text{Si}_{1-x}\text{Ge}_x$ Growth Kinetics
during GS-MBE and UHV-CVD**

Joe Greene
University of Illinois

Monday, March 22, 8:30 to 10:30

Outline:

1. ALE Si
 - a. Mechanisms
 - b. Modeling
2. GS-MBE/UHV-CVD Si
 - a. Mechanisms
 - b. Modeling
3. B doping
 - a. Incorporation mechanism
 - b. Electronic properties
4. Ultra-high B doping
 - a. Incorporation
 - b. Effect on R_{Si}
 - c. Modeling
 - d. Electronic properties
5. GS-MBE/UHV-CVD $\text{Si}_{1-x}\text{Ge}_x$
 - a. Mechanism
 - b. Modeling
 - c. B doping and electronic properties

Atomic-level Control during Film Growth under Highly Kinetically Constrained Conditions

- * **$\text{Si}_{1-x}\text{Ge}_x$ atomic-layer epitaxy (ALE)**

- Low-temperature growth
 - δ -doping

- * **Effects of H coverage during GS-MBE/UHV-CVD**

- Ultra-high doping vs film growth kinetics
 - Minimize Ge segregation during $\text{Si}_{1-x}\text{Ge}_x$ growth

- * **Fundamental limits of epitaxy**

- $T_{\text{epi}} \rightarrow t_{\text{epi}}$
 - Use of hyperthermal beams
 - Metastable phases ($\text{Ge}_{1-x}\text{Sn}_x$, $\text{Ti}_{1-x}\text{Al}_x\text{N}$,)

- * **Evolution of surface morphology**

- Kinetic roughening
 - Strain-induced roughening/smoothening

- * **Evolution of microstructure and texture**

- Hyperthermal beams
 - Texture inheritance

Atomic-Layer Epitaxy (ALE)

ALE is a cyclic vapor-phase epitaxial film growth technique which incorporates *self-limiting* kinetic processes which result in the deposition of θ ML per growth cycle where $\theta \leq 1$ ML.

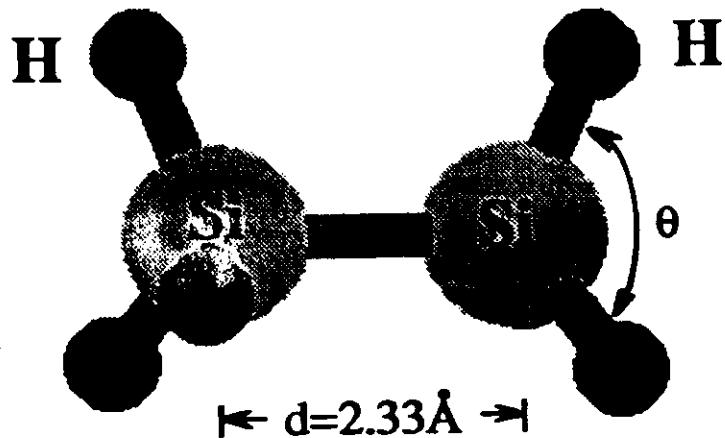
$$\rightarrow R_{\text{cycle}} = \theta \text{ (ML/cycle)}$$

Advantages:

- $R \neq f(T_s, P_{\text{gas}})$; wide process windows
- \therefore overall $E_a = 0$
- \therefore low T_s possible
- Excellent control over film thickness:
 $t = R_{\text{cycle}}(\# \text{cycles})$
- Selective area deposition
- δ -doping
- QW heterostructures
-
-

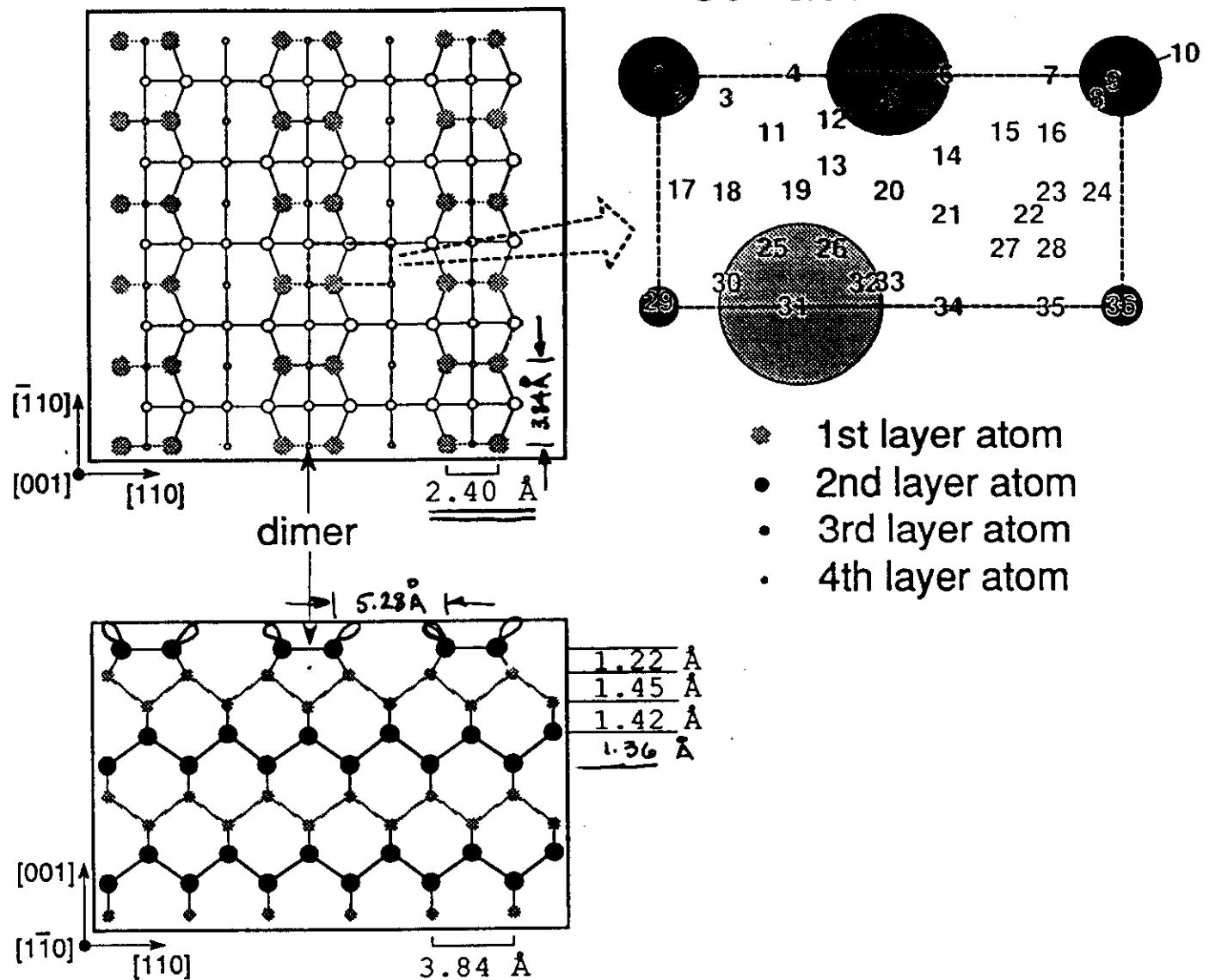
Precursor for Si ALE: Si_2H_6

- High sticking probability on Si & Ge
 $\sigma(\text{Si}_2\text{H}_6) > 10^4 \times \sigma(\text{SiH}_4)$
- Gas phase dimer $\approx \text{Si}(001)2\times1$

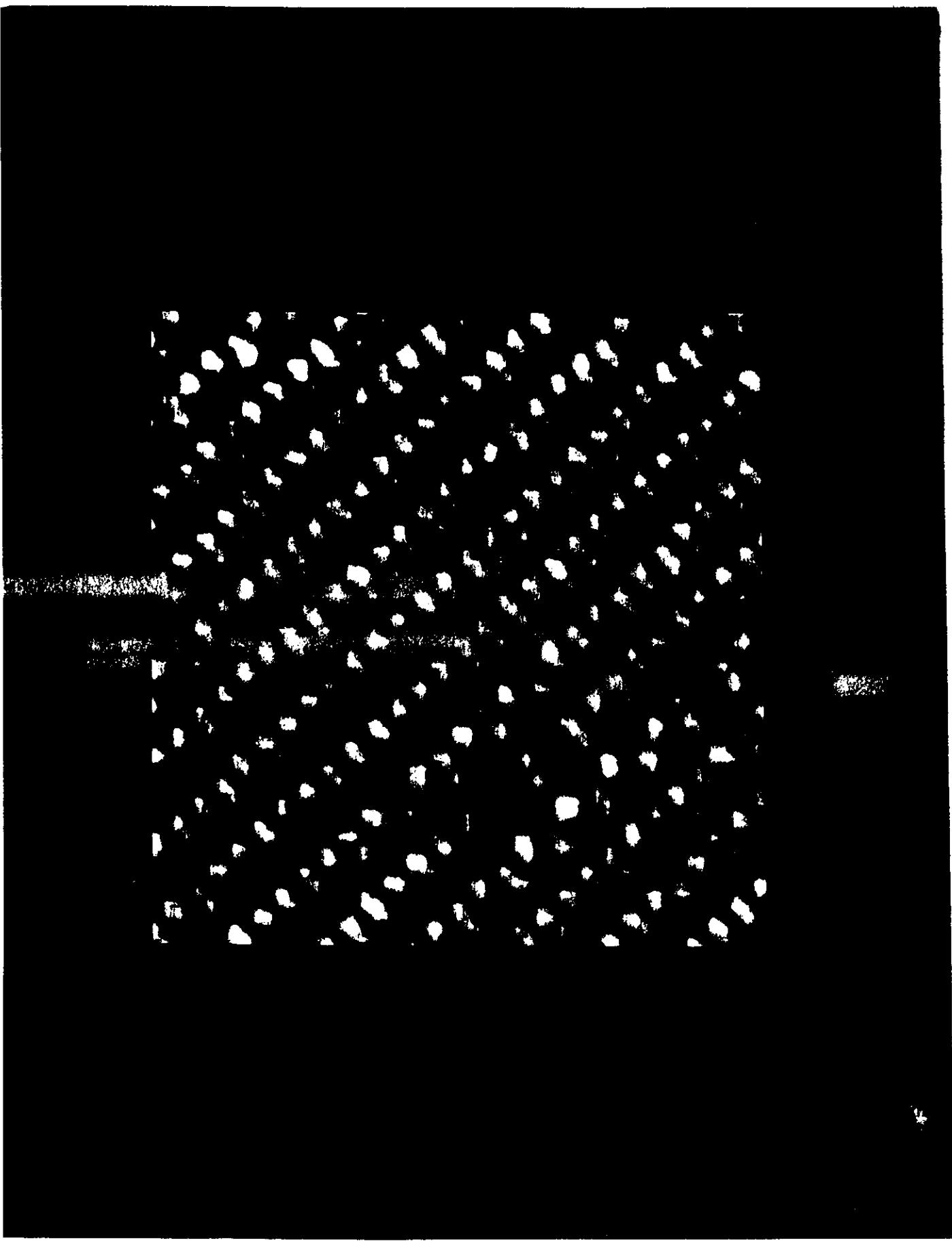


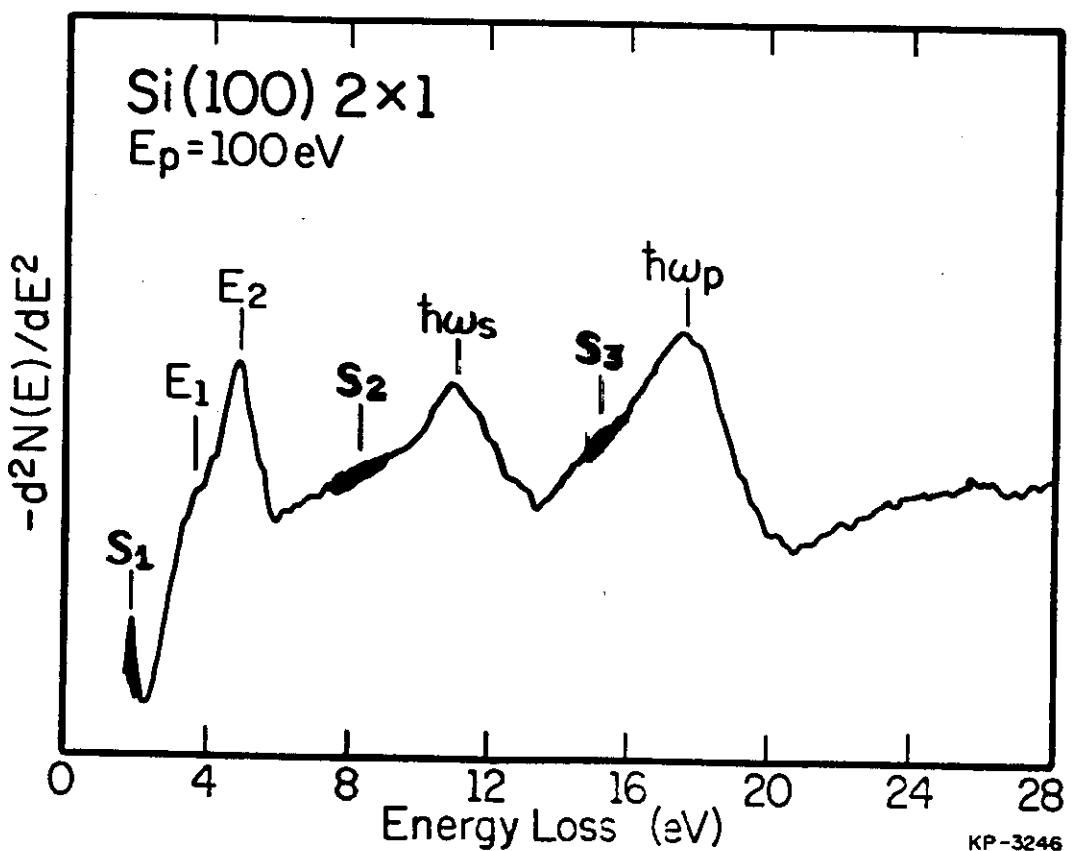
- $E_b(\text{Si-Si}) = 3.3 \text{ eV}$
- $E_b(\text{Si-H}) = 3.8 \text{ eV}$
- $d(\text{Si-Si}) = 2.33 \text{ \AA}$
- $d(\text{Si-H}) = 1.49 \text{ \AA}$
- $\theta(\text{Si-Si-H}) = 110.3^\circ$

Surface unit cell

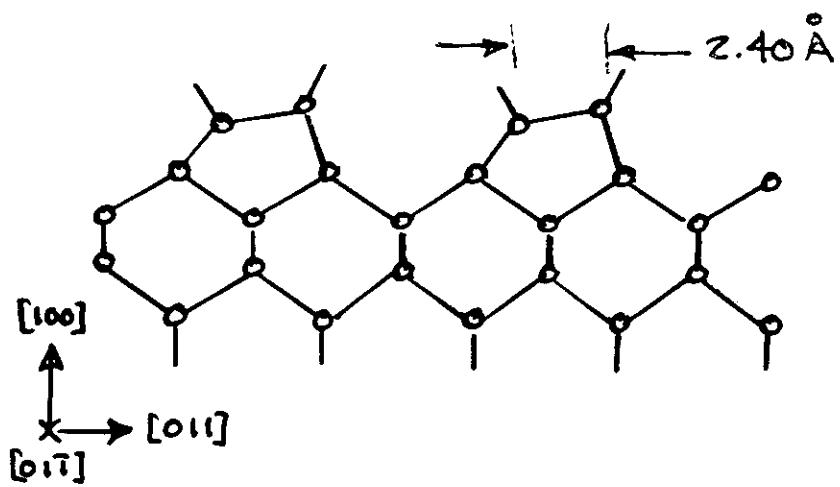


M. Kitabatake and J.E. Greene, J. Appl. Phys.
73, 3183 ('93).

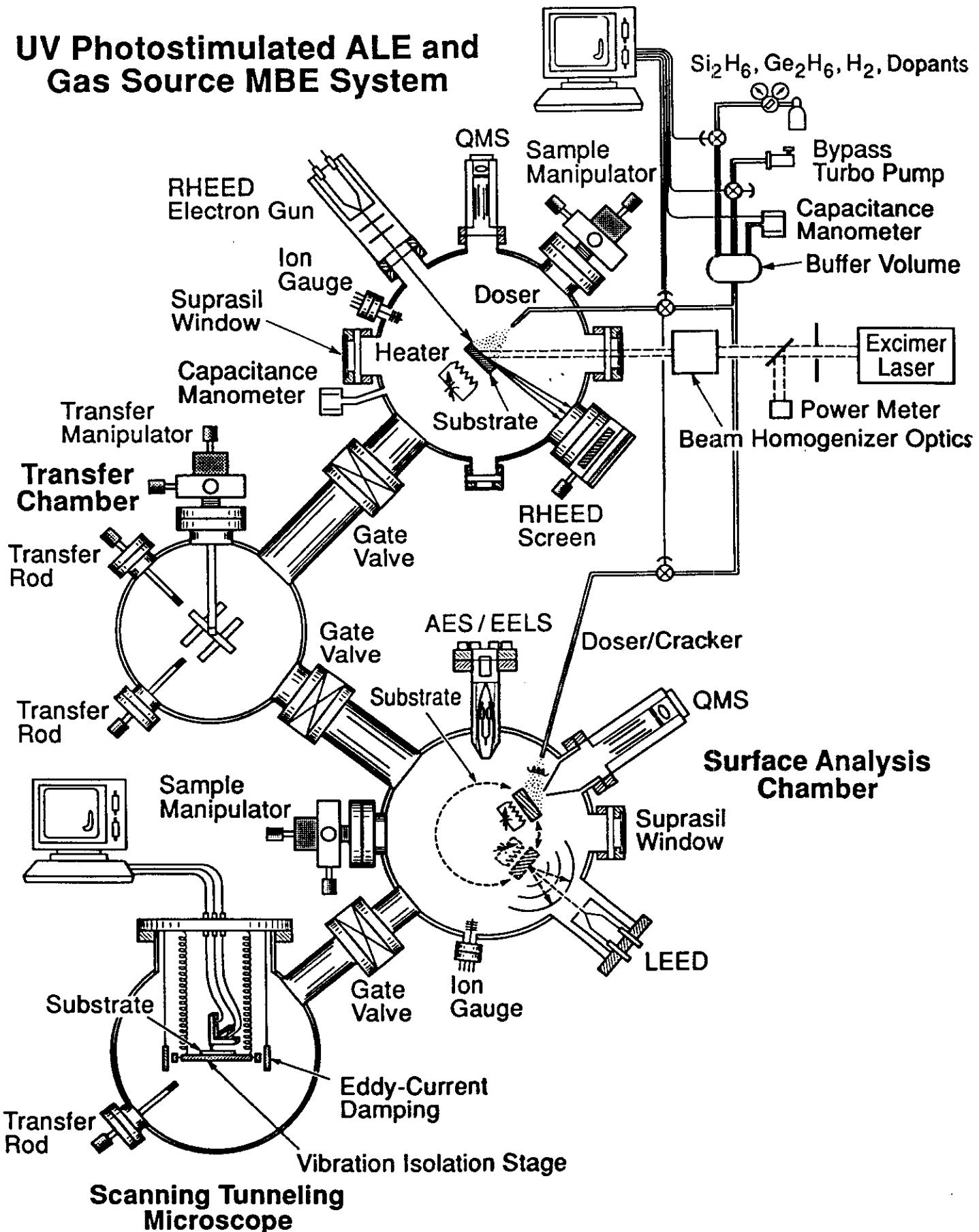




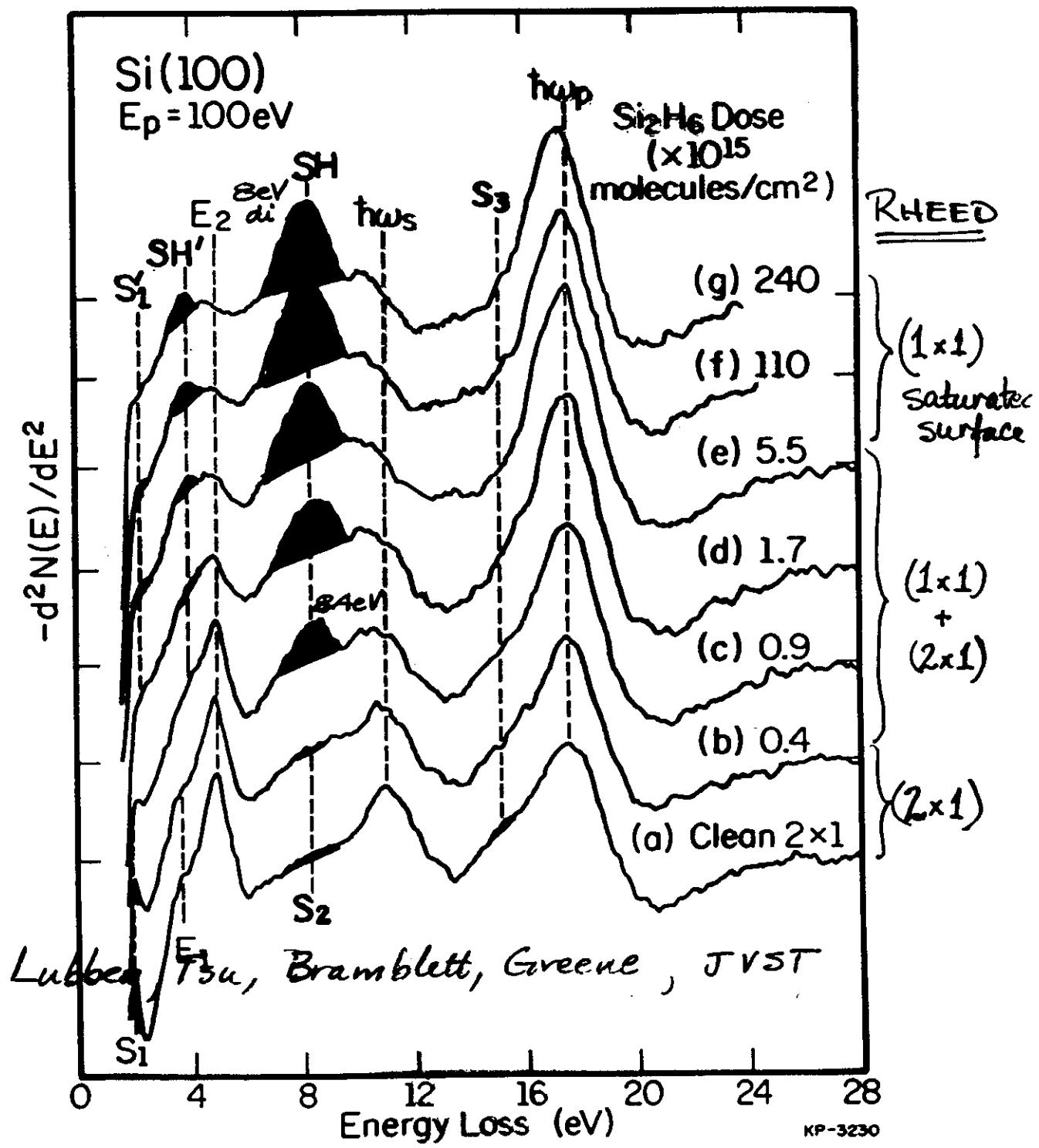
S_1 = surface dangling bond state
 S_2, S_3 = backbond states



UV Photostimulated ALE and Gas Source MBE System

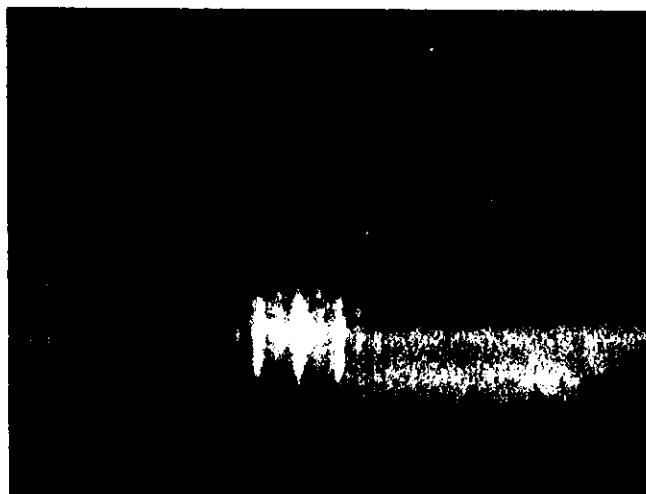


$T_s = 500 \text{ K}$

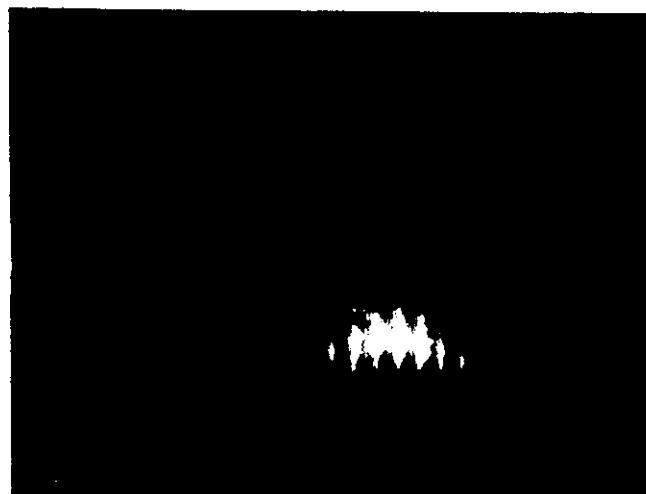




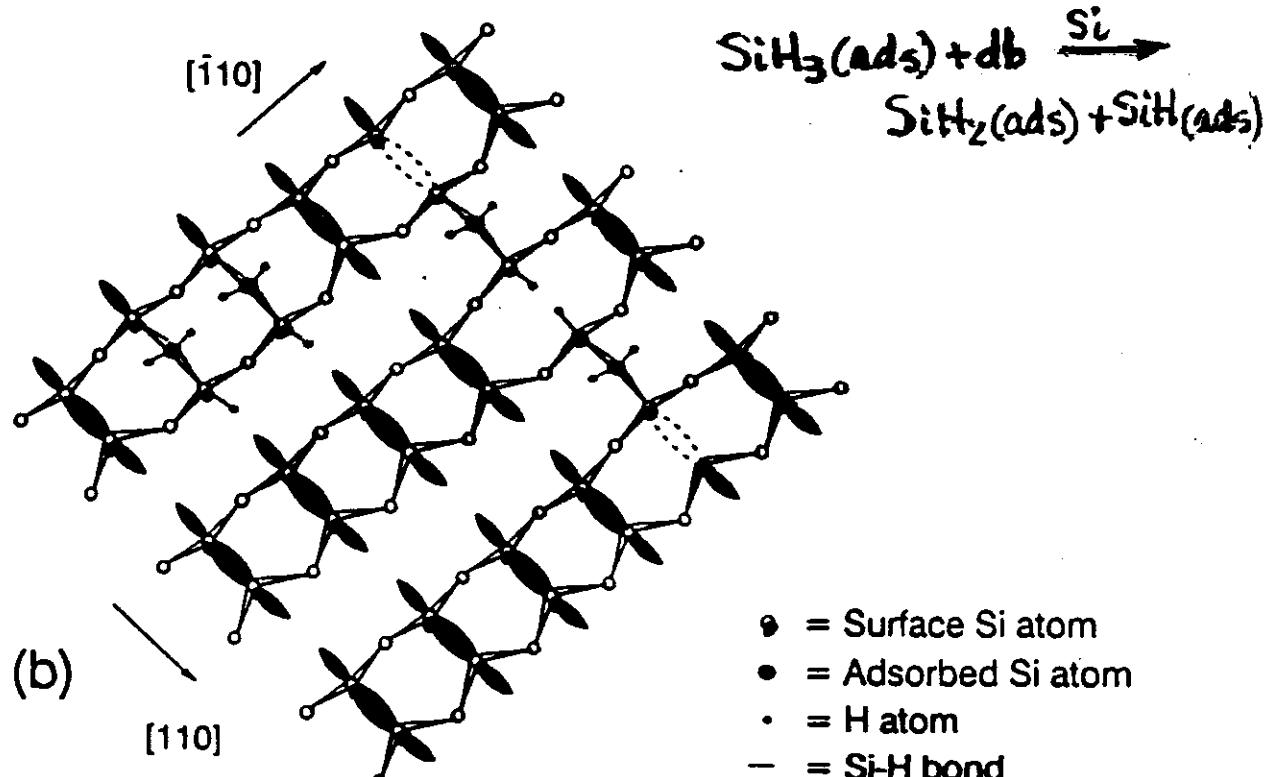
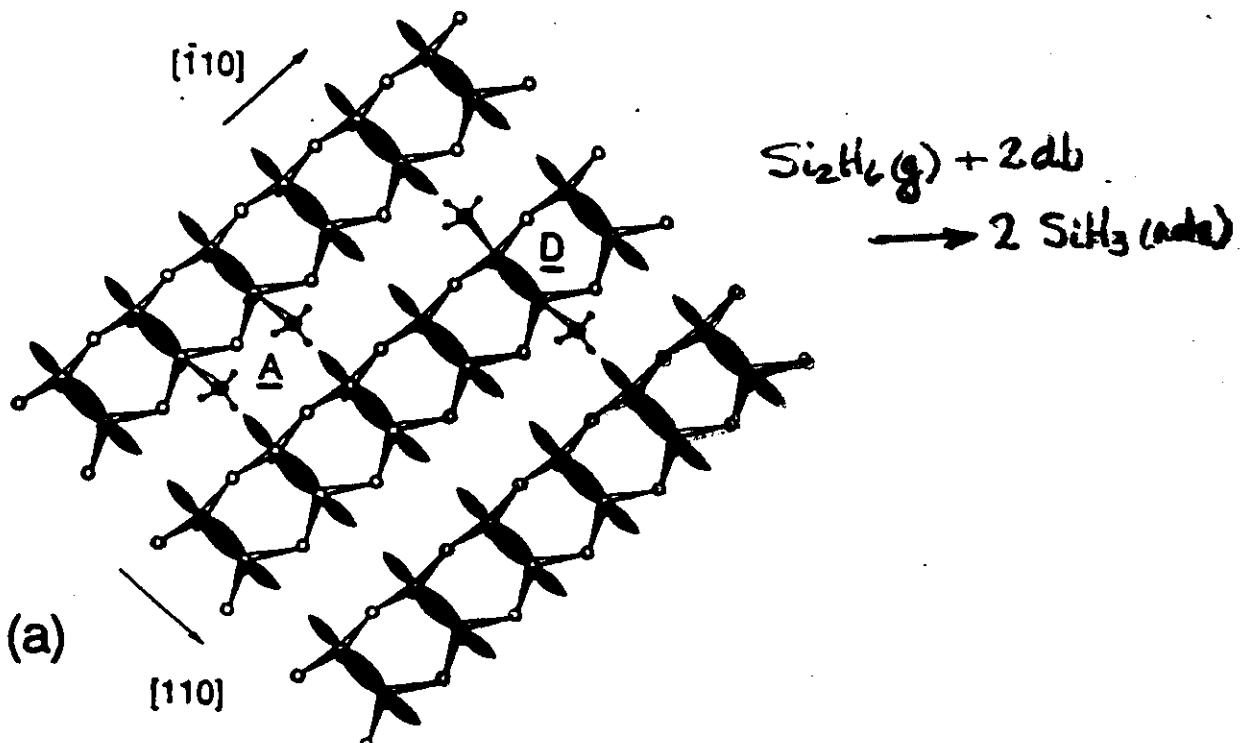
1×1



2×1
+
 1×1



2×1



- = Surface Si atom
- = Adsorbed Si atom
- = H atom
- = Si-H bond
- = Dangling bond
- = Si-Si bond
- — = Dimer bond

7. Schematic illustration of the Si(001) surface after (a) dissociative chemisorption of Si_2H_6 onto a single dimer (D) and onto adjacent dimers (A), and (b) dissociation of the adsorbed SiH_3 radicals to form SiH_2 and H.

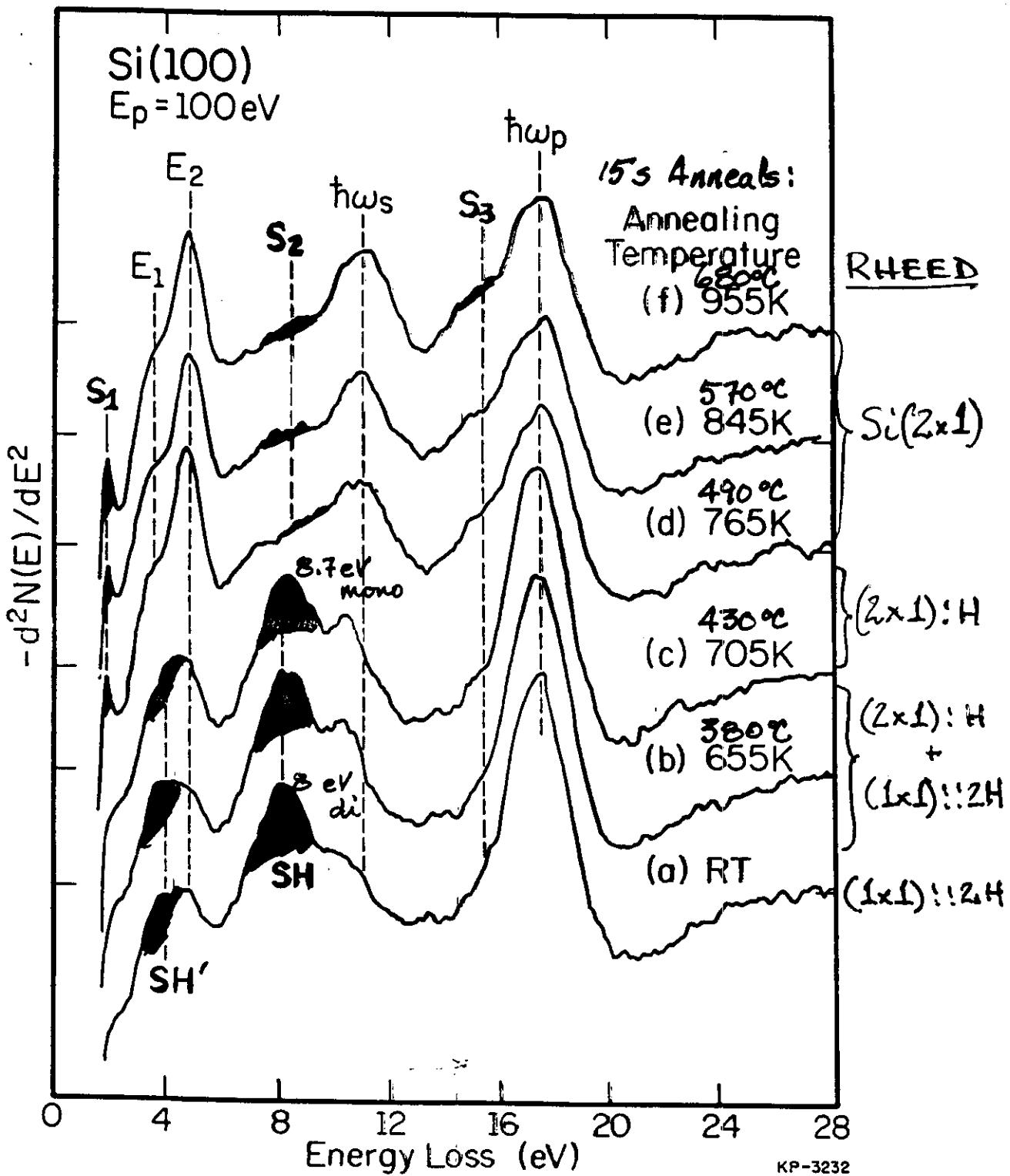
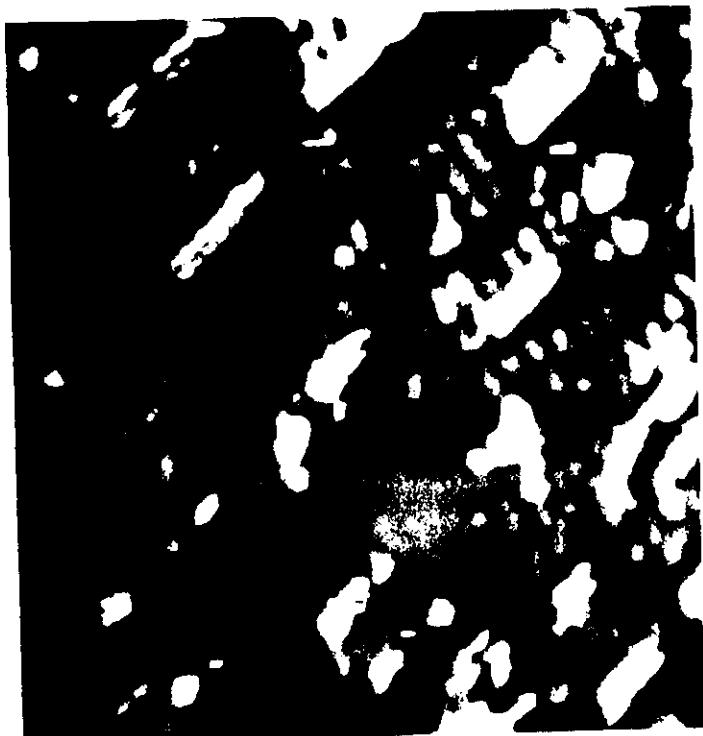


Figure 4. EELS spectra, obtained using a primary electron energy E_p of 100 eV, from a Si_2H_6 -saturated Si(100) (1x1)::2H dihydride surface following a 15 s anneal at (a) 300 K, (b) 655 K, (c) 705 K, (d) 765 K, (e) 845 K, and (f) 955 K.

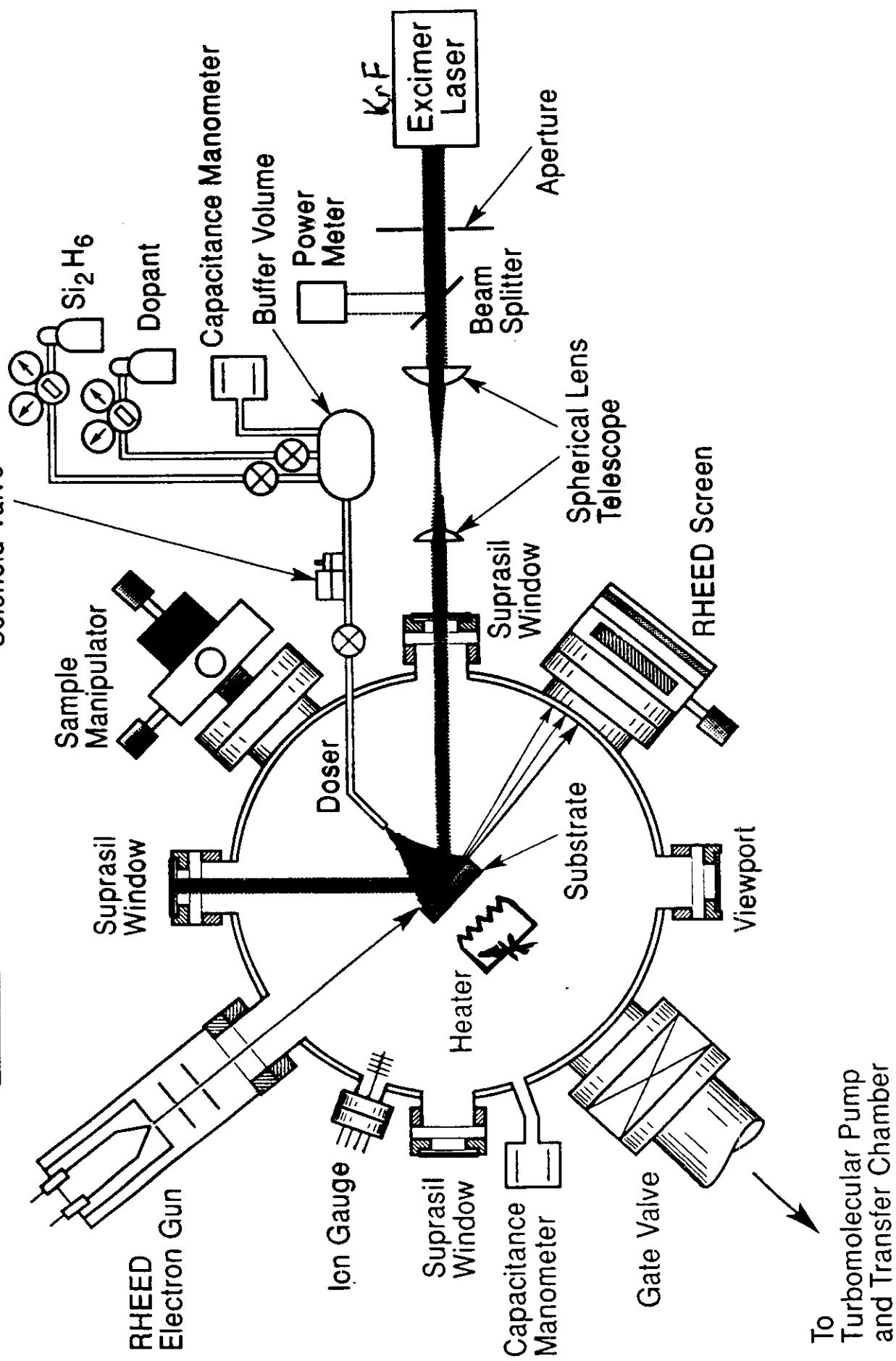
Si_2H_6 dose = $3.2 \times 10^{14} \text{ cm}^{-2}$
 $T_s = 380^\circ\text{C}$
200 Å × 200 Å

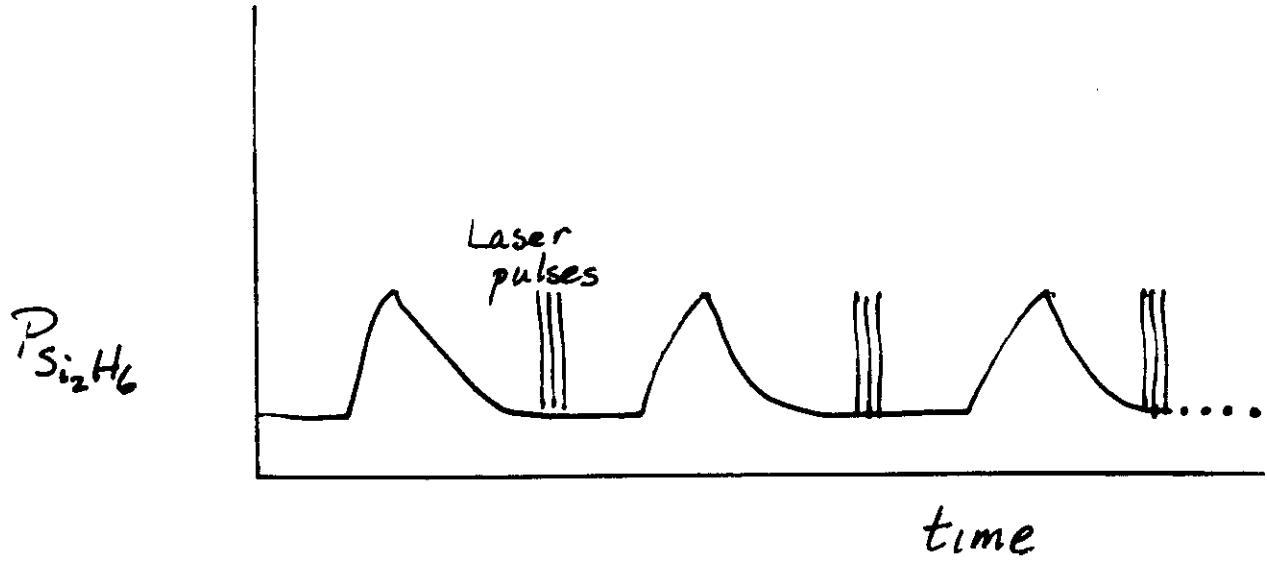


Lin, Hinchorn, Chuang, Tsu, Lubben, Greene
Phys. Rev. B

Si ALE System

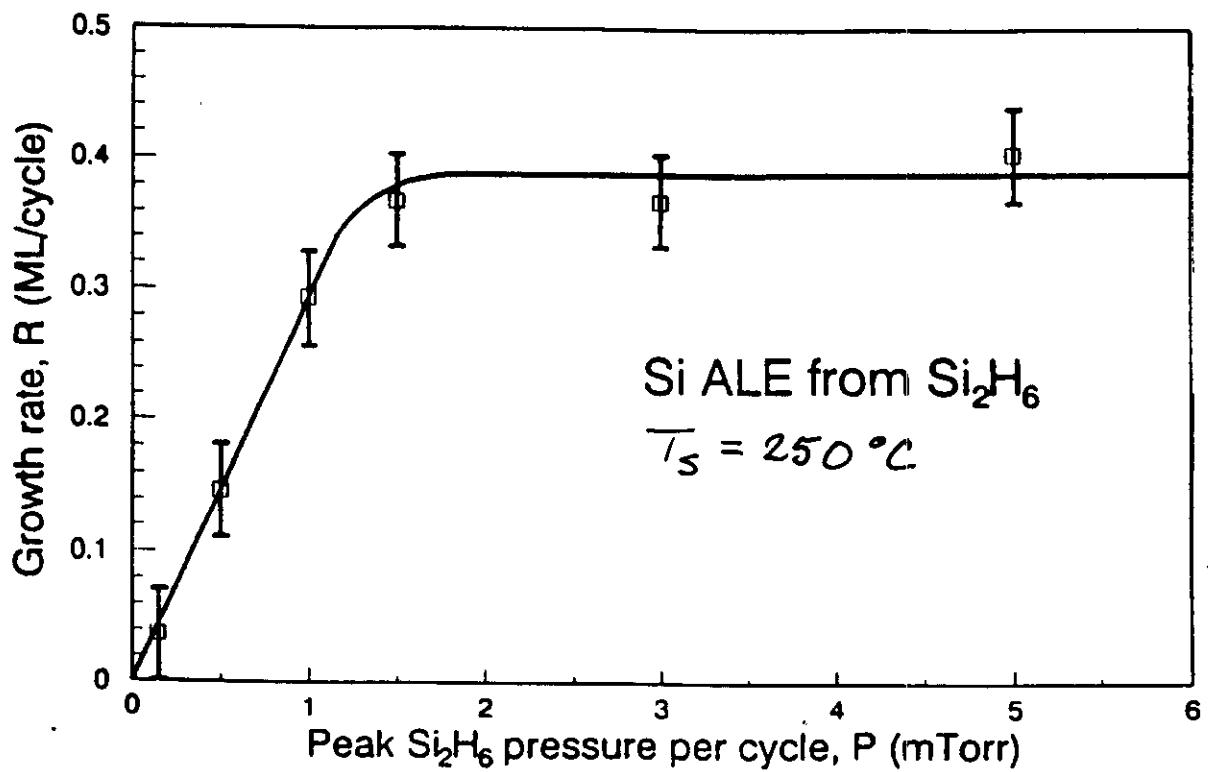
Flow Meter and Solenoid Valve

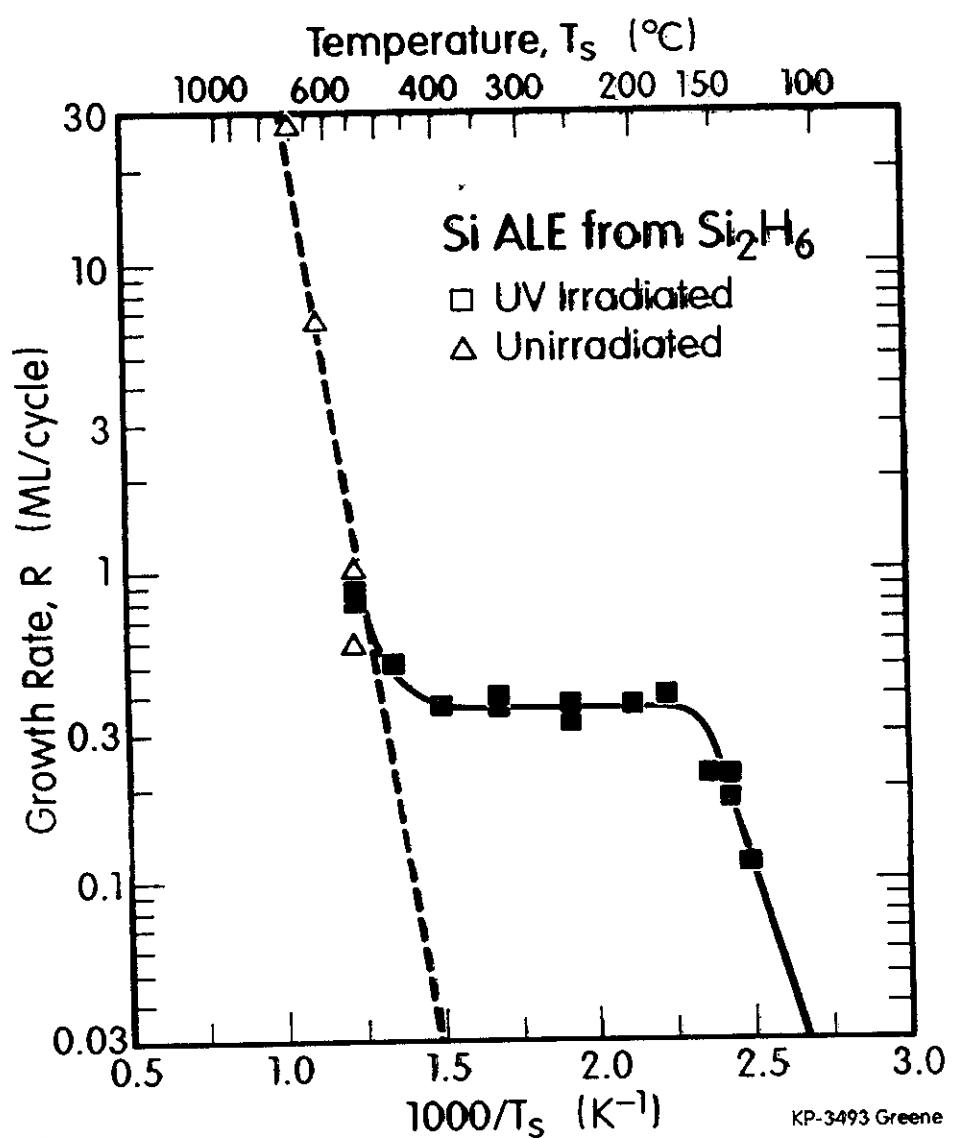




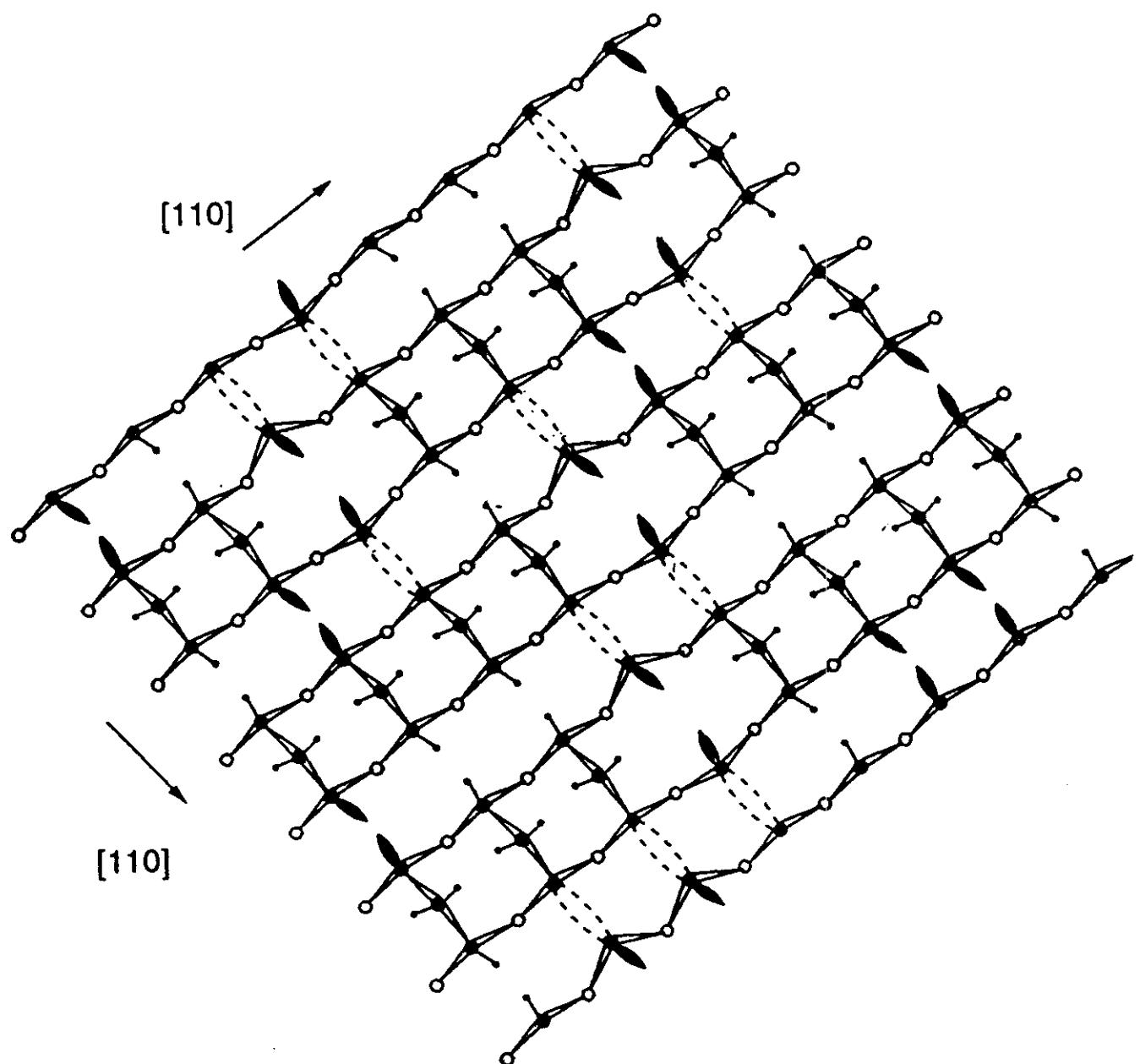
$$180^\circ\text{C} \leq T_s \leq 400^\circ\text{C}$$

$$P_{\max} \geq 1.5 \text{ mTorr}$$



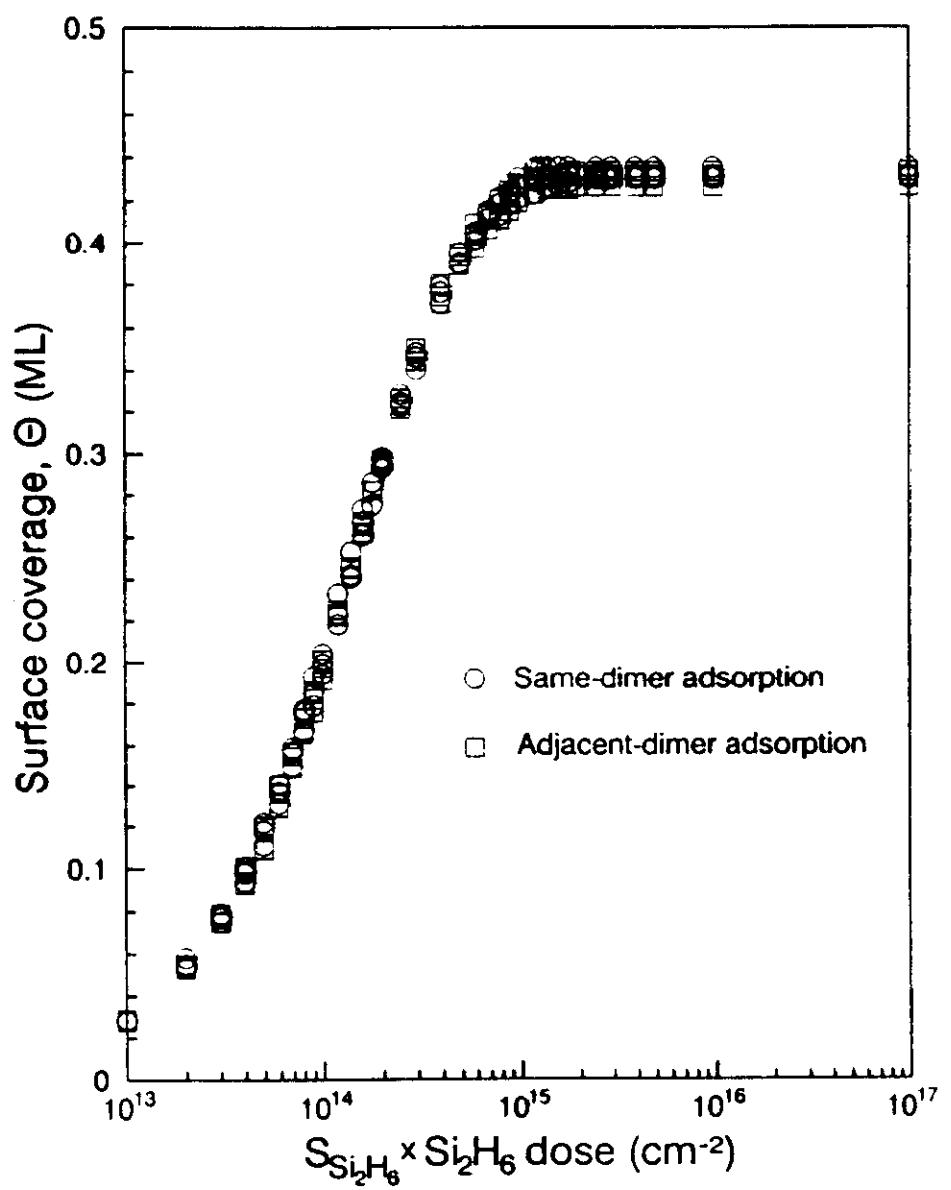


Lubben, Tsu, Bramblett, Greene, JVST A



- = Surface Si atom
- = Adsorbed Si atom
- = H atom
- = Dangling bond
- = Si-Si bond

D.S. Liu, T.-C. Chiang, R. Tsu, J.E. Greene,
Phys. Rev. B, in press.



Analytical Transition-State Model

Si (001) 2x1 ALE

$$\frac{dn_{SiH_3}}{dt} = 2 J_{Si_2H_6} S_{Si_2H_6}(T_s) [1 - N_{oc} - N_{bl}] - n_{SiH_3} k_{d, SiH_3} \\ - (n_{SiH_3})^2 k_{a, Si_2H_6} - n_{SiH_3} k_{dis, SiH_3}$$

$$\frac{dn_{SiH_2}}{dt} = \underline{n_{SiH_3} k_{dis, SiH_3}} - n_{SiH_2} k_{d, SiH_2} - n_{SiH_2} k_{dis, SiH_2}$$

$$k_{dis, SiH_3} = \nu e^{(-E_{dis, SiH_3}/kT_s)}$$

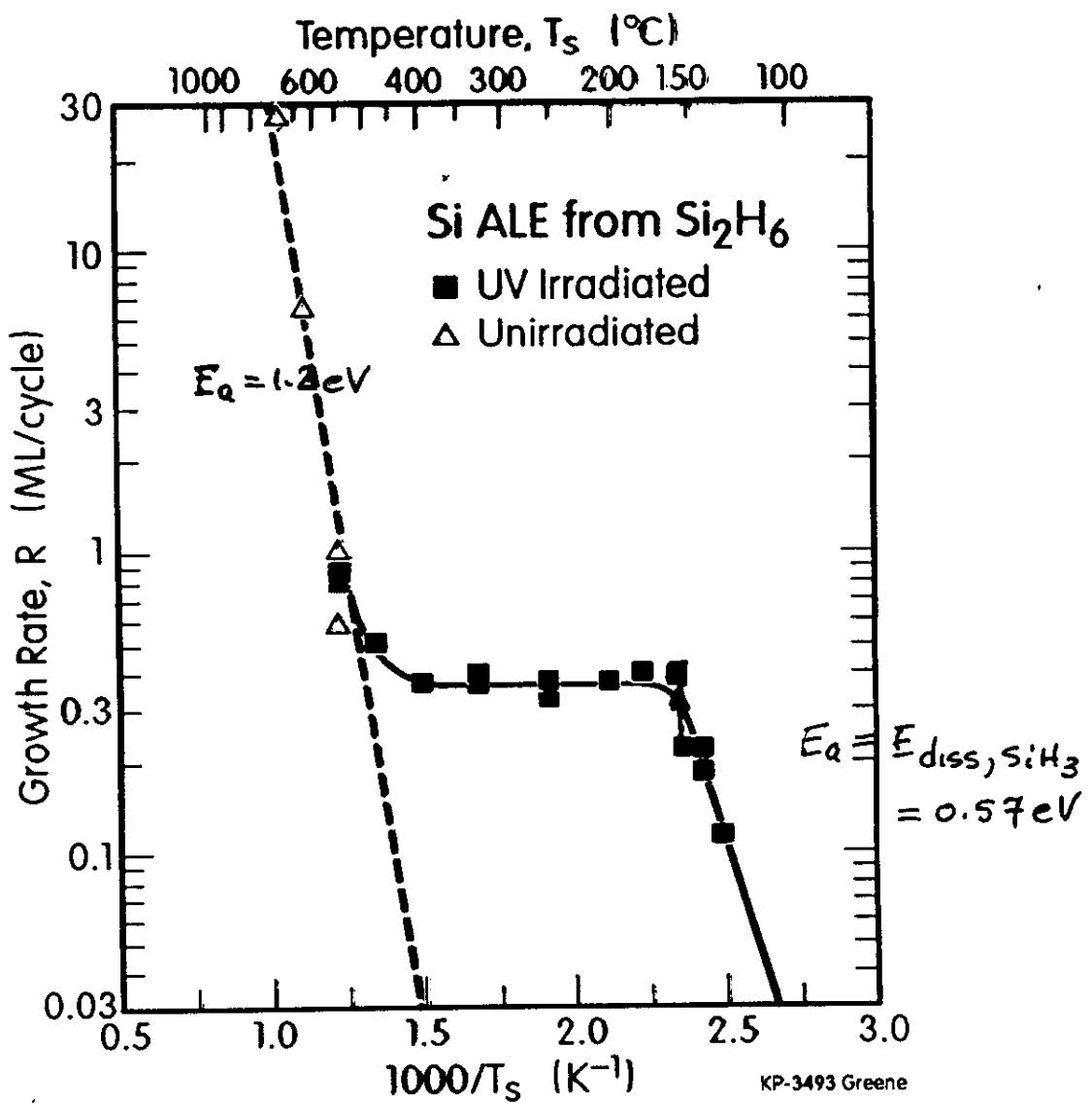
$$N_{un} + N_{oc} + N_{bl} = 1$$

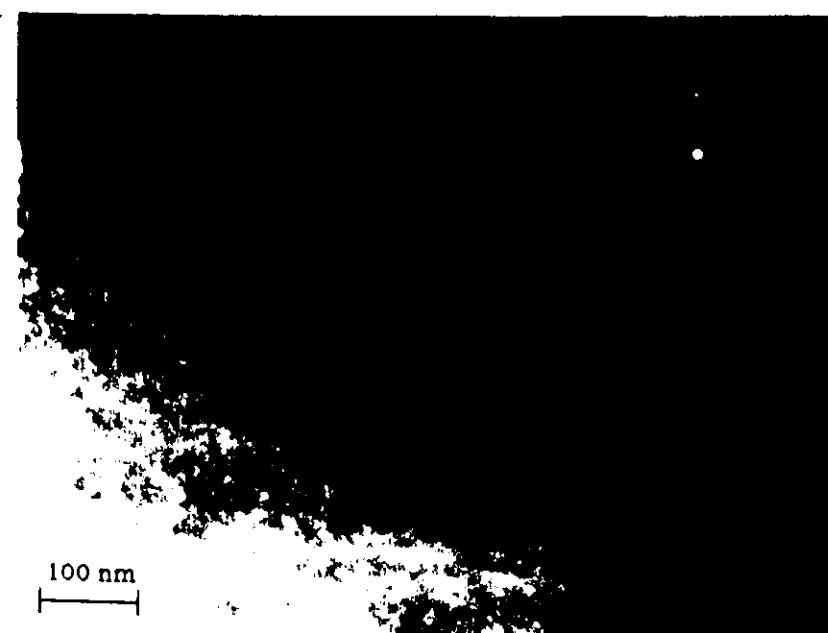
Monte Carlo simulation $\rightarrow N_{bl} = f(N_{oc})$

$$n_{SiH_3}^* = 0.4 n_s - n_{SiH_2}^*$$

$$n_{SiH_2}^* = 0.4 n_s \underline{\Delta t} \nu e^{(-E_{dis, SiH_3}/kT_s)}$$

$$\therefore \text{ALE : } R = n_{Si}^* = n_{SiH_2}^* \approx 0.4 \text{ ML/cycle}$$

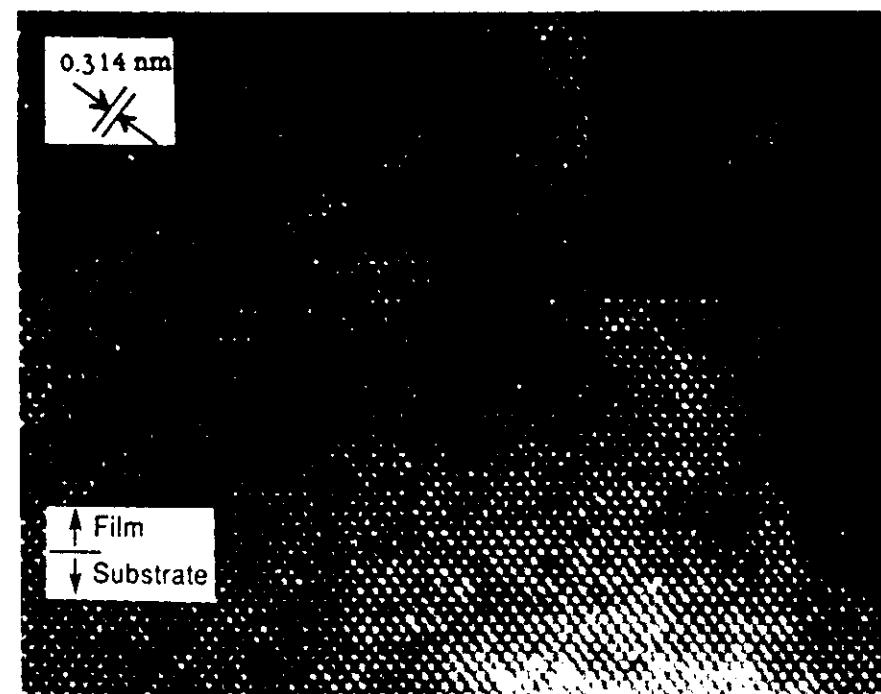




Plan-view

[110]
[001] X [110]

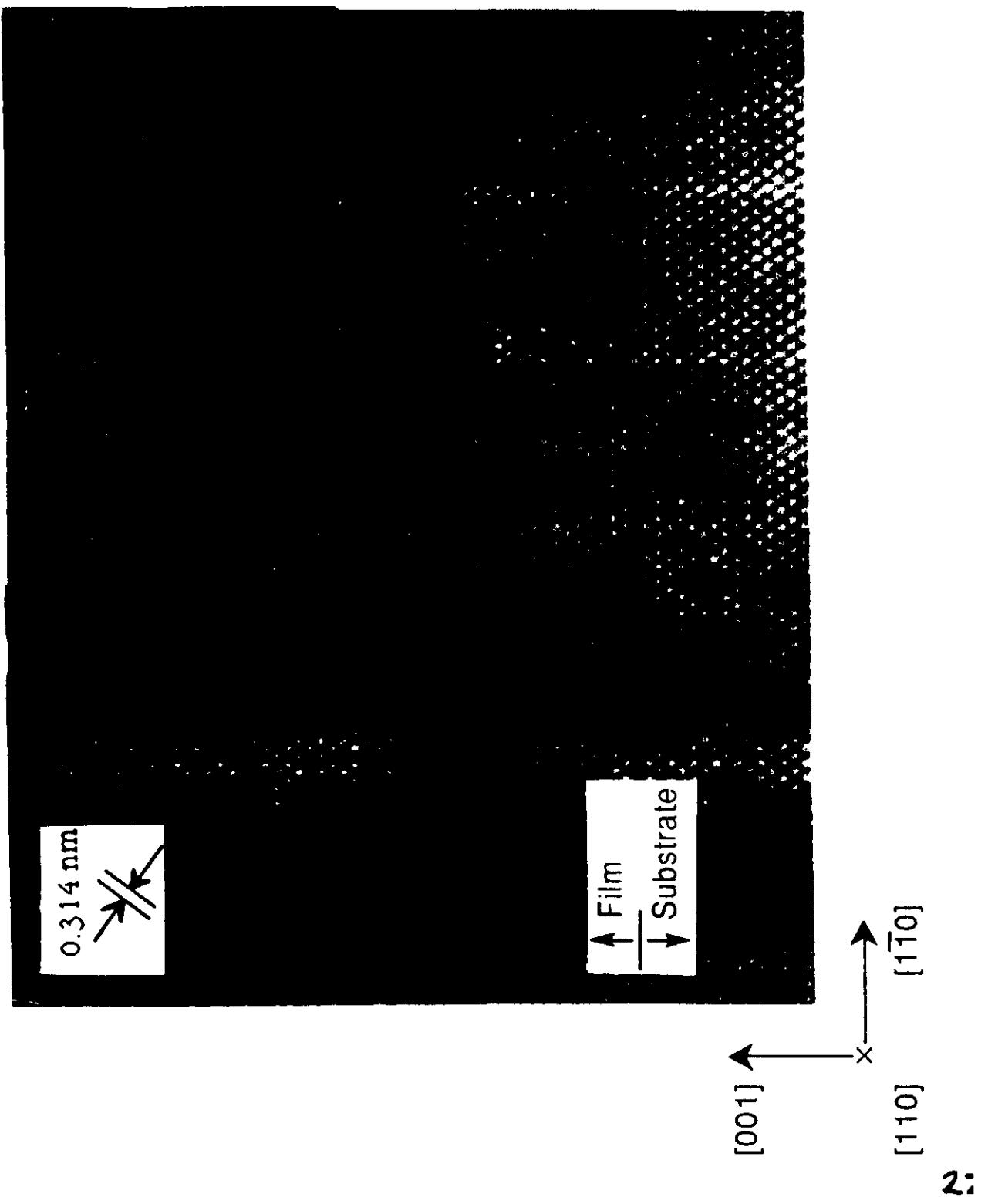
100 nm



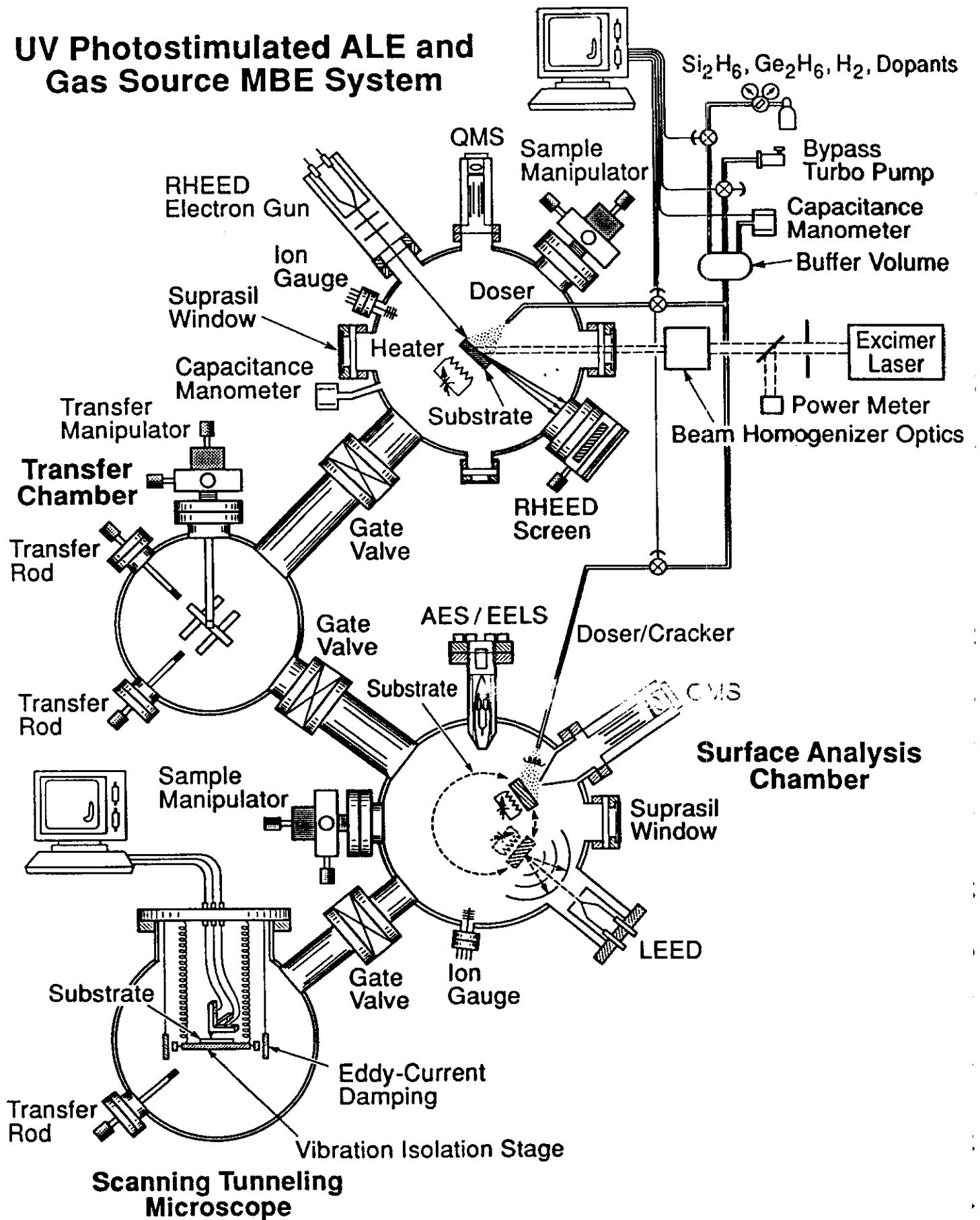
Cross-section

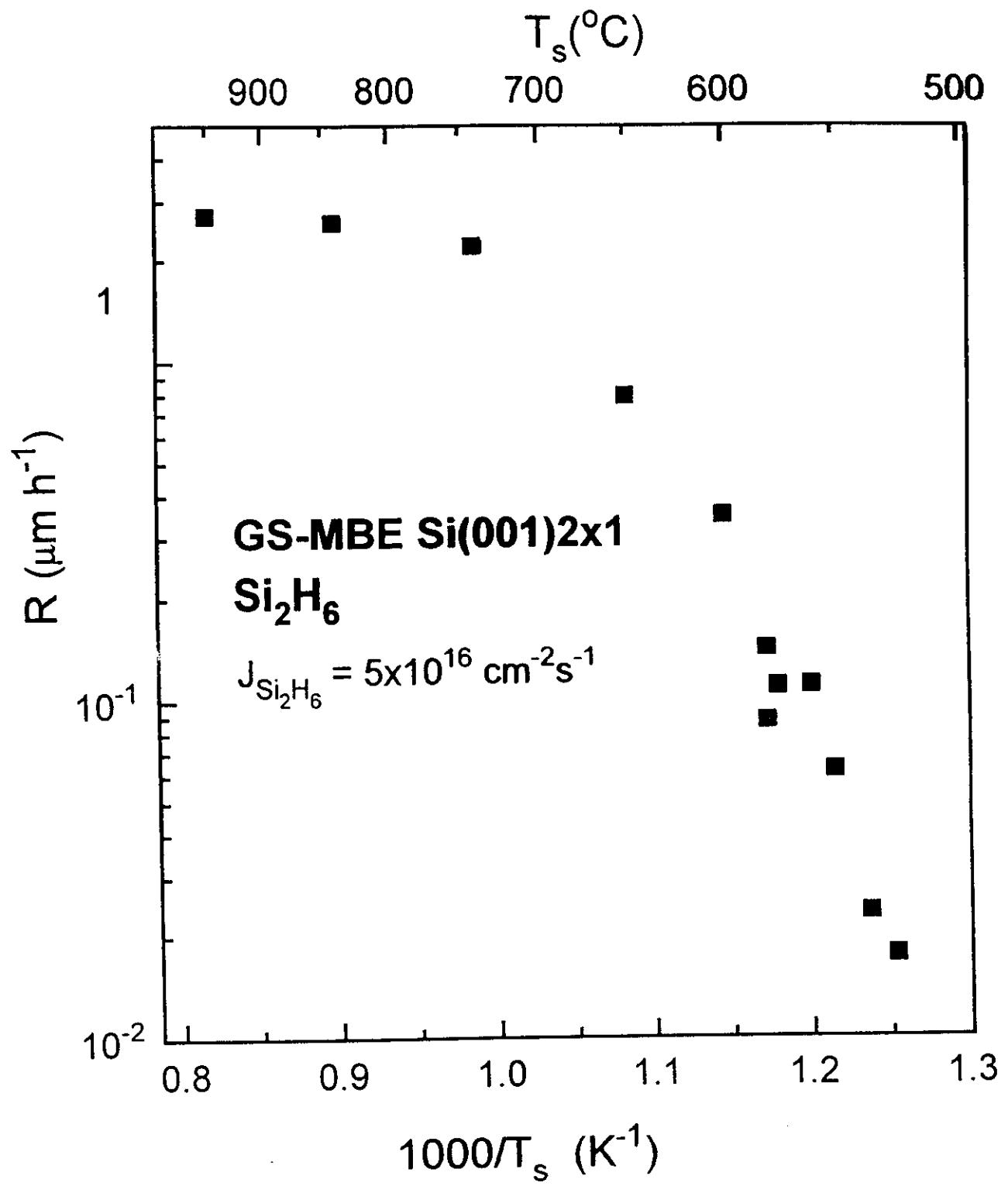
[001] ↑
[110] X [110]

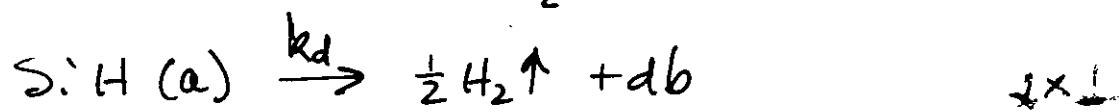
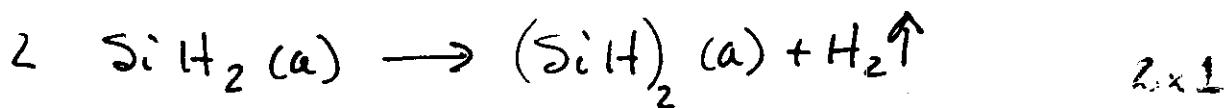
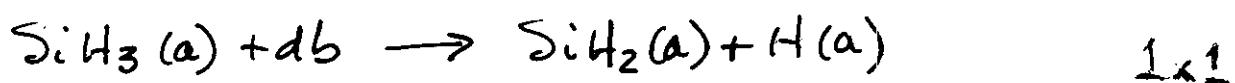
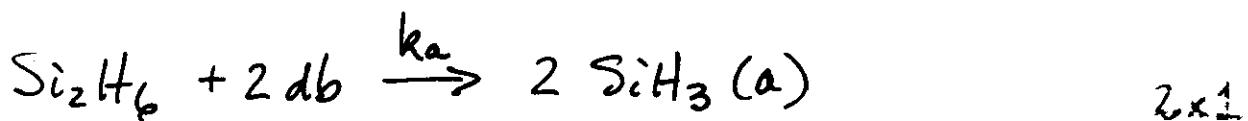
Film
Substrate



UV Photostimulated ALE and Gas Source MBE System







$$\frac{d\theta_{\text{db}}}{dt} = -2k_a J_{\text{Si}_2\text{H}_6} \theta_{\text{db}}^2 + k_d (1 - \theta_{\text{db}})$$

∴ at steady state:

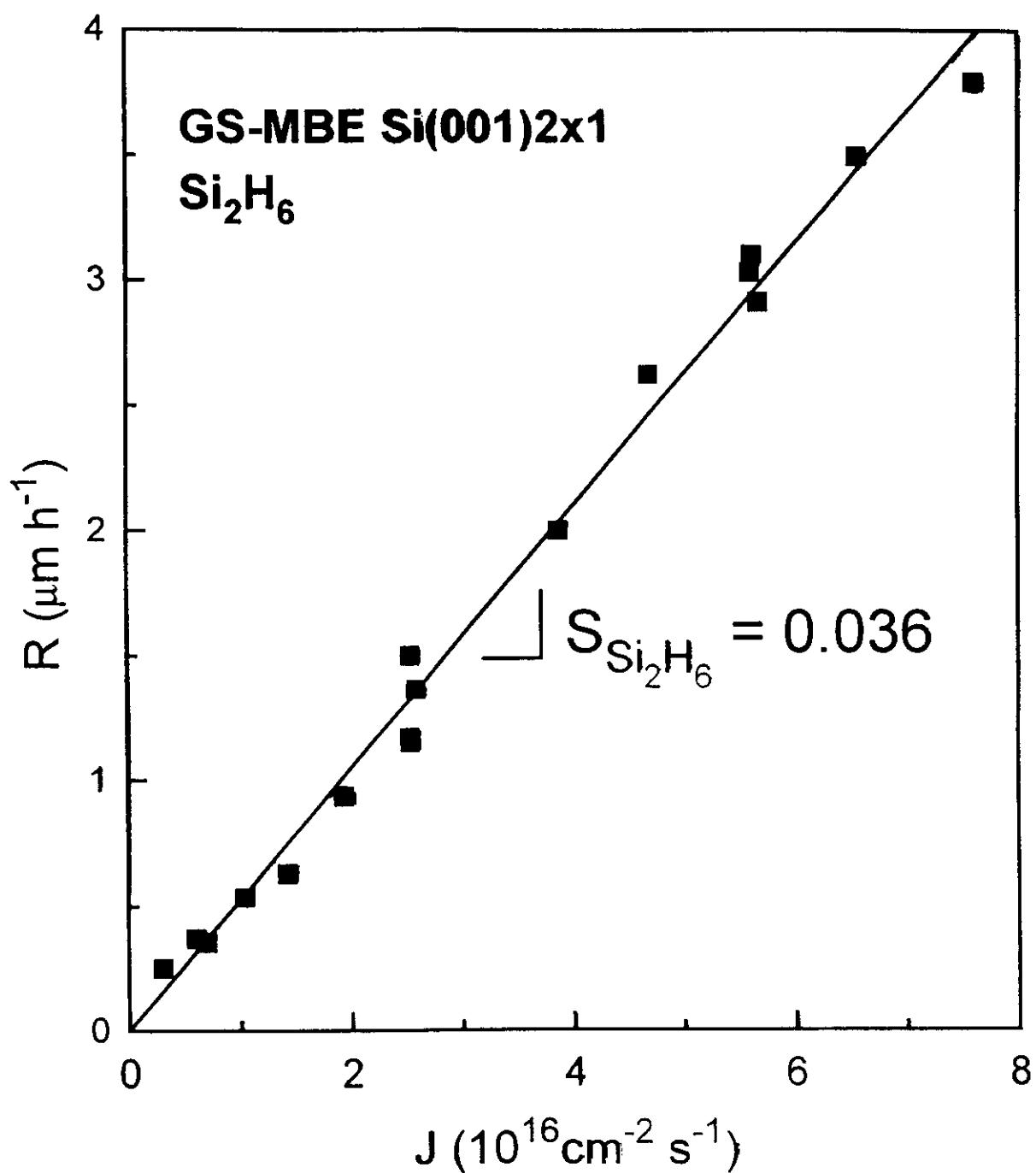
$$\theta_{\text{db}} = \frac{k_d}{4k_a J_{\text{Si}_2\text{H}_6}} \left[\left(1 + \frac{8k_a J_{\text{Si}_2\text{H}_6}}{k_d} \right)^{1/2} - 1 \right]$$

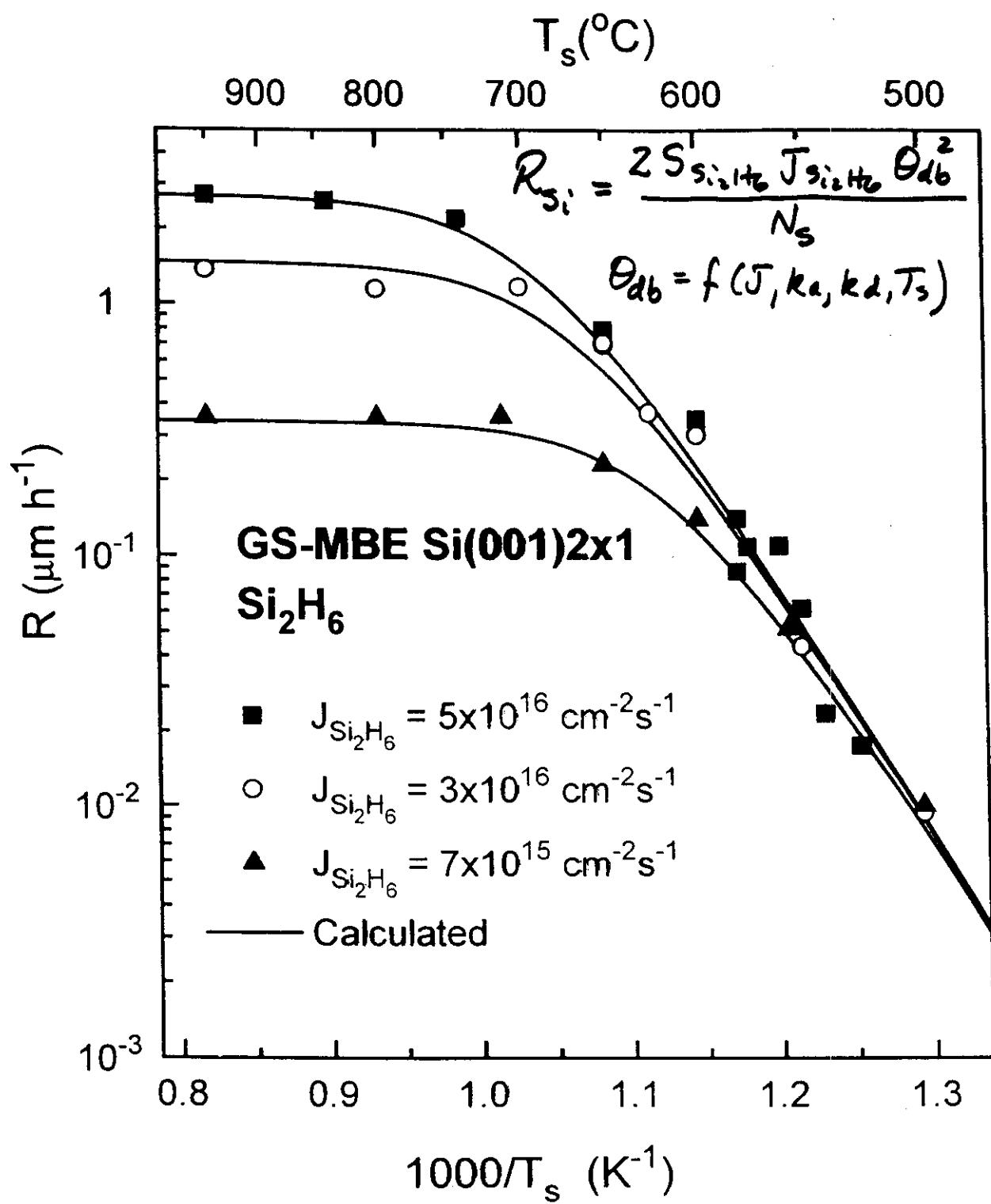
$$R = \frac{2N_s k_a J_{\text{Si}_2\text{H}_6} \theta_{\text{db}}^2}{N}$$

where $k_a = S_{\text{Si}_2\text{H}_6}/N_s$ (measured)

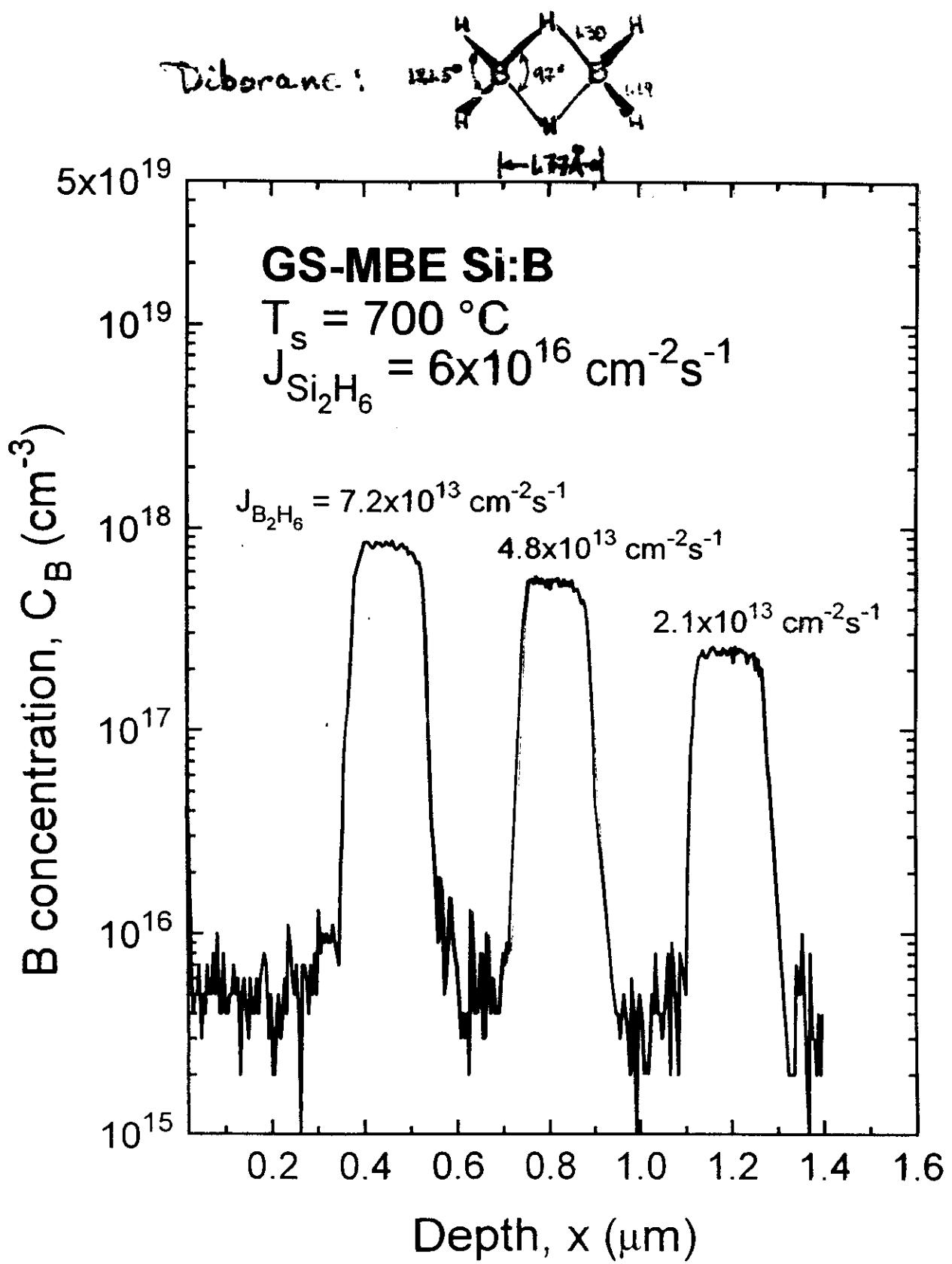
$$k_d = 7.9 \times 10^{-6} \exp\left(\frac{-2.04 \text{ eV}}{k T_s}\right) \text{ s}^{-1}$$

(Yates et. al., JCP
92, 5700 (1990)).

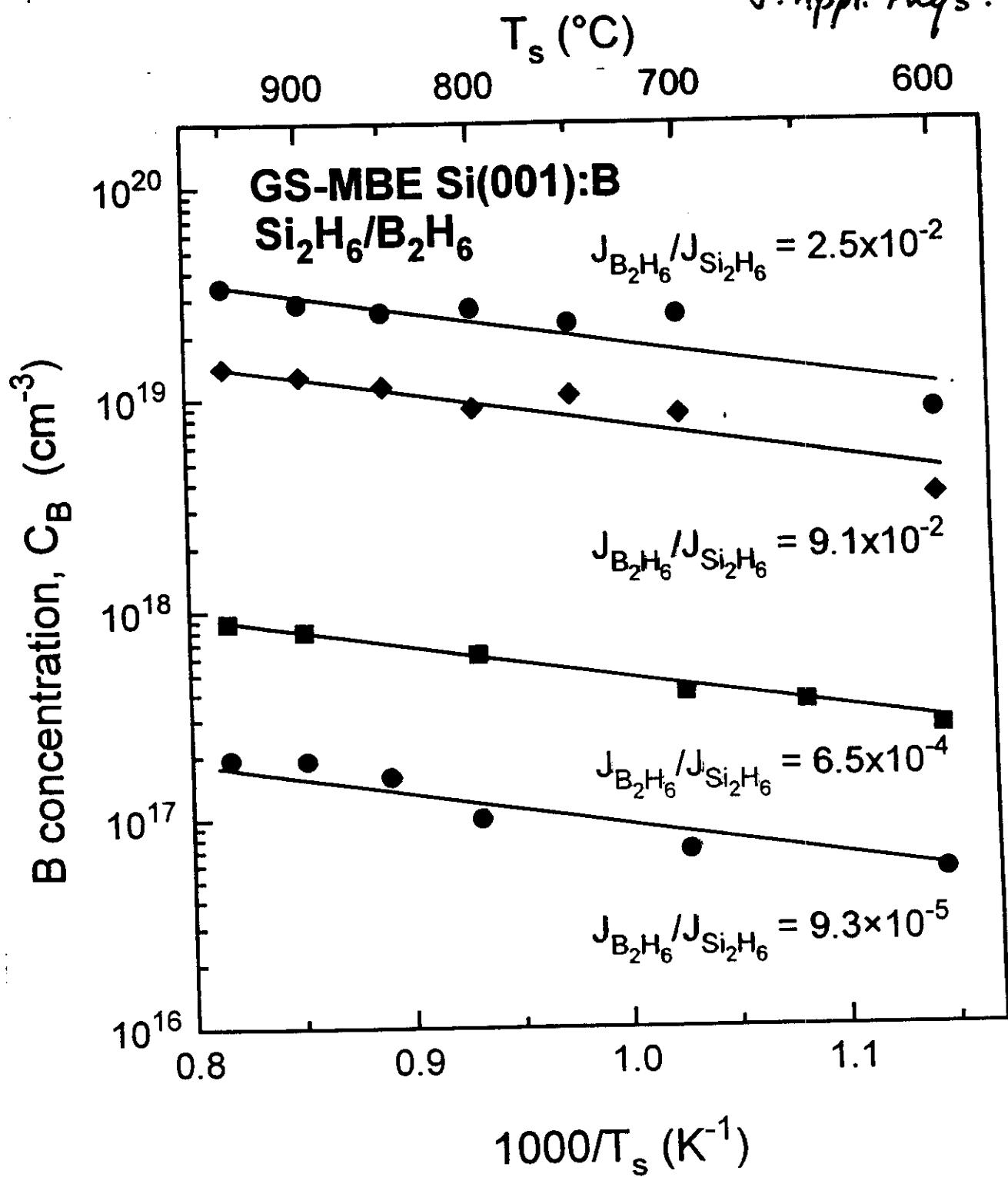




T.R. Bramblett, Q. Lu, T. Karazawa, M-A. Hasan,
S.K. Jo, and J.E. Greene, J. Appl. Phys.

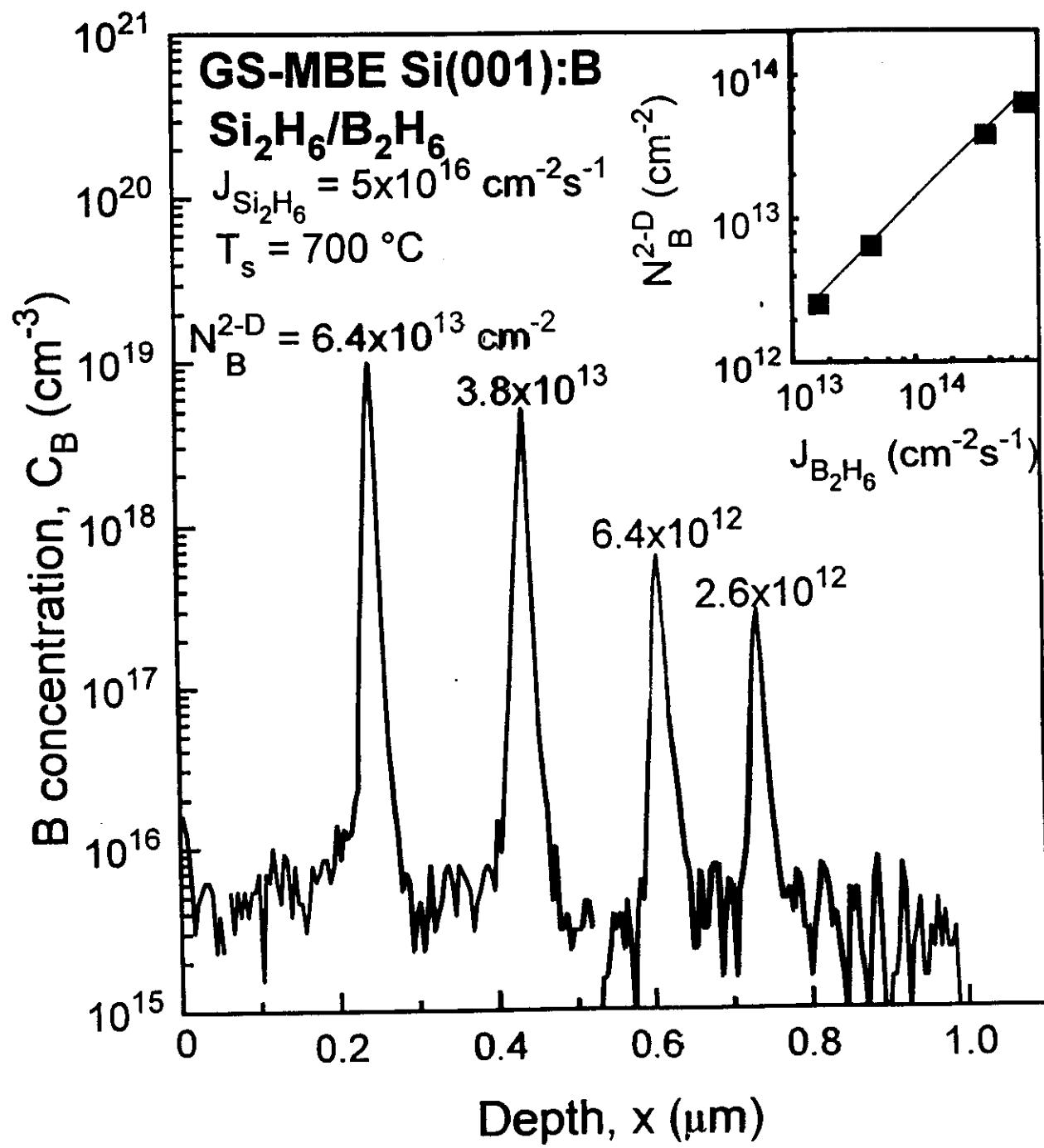


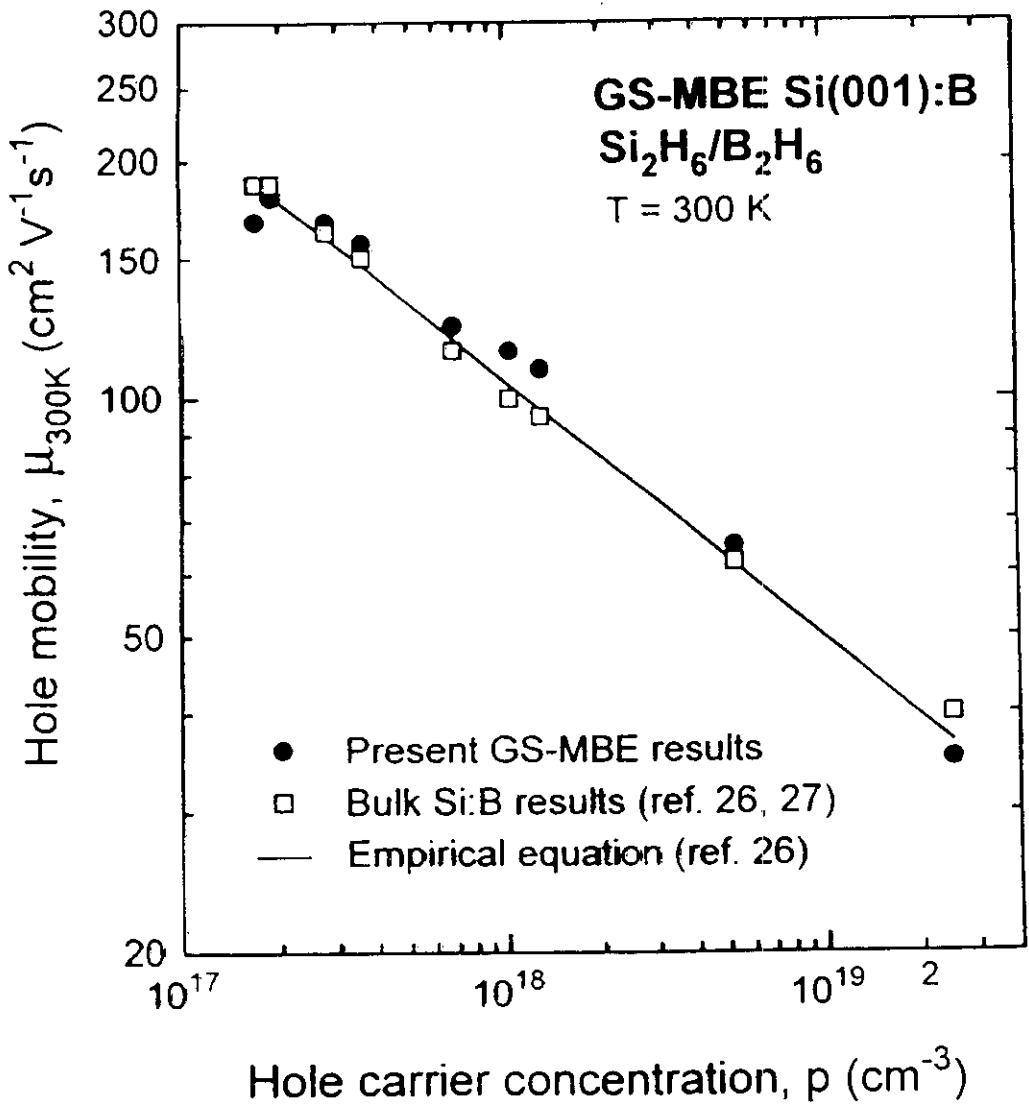
Lu, Bramblett, Lee, Hasan, Kawasawa, Greene
J. Appl. Phys.

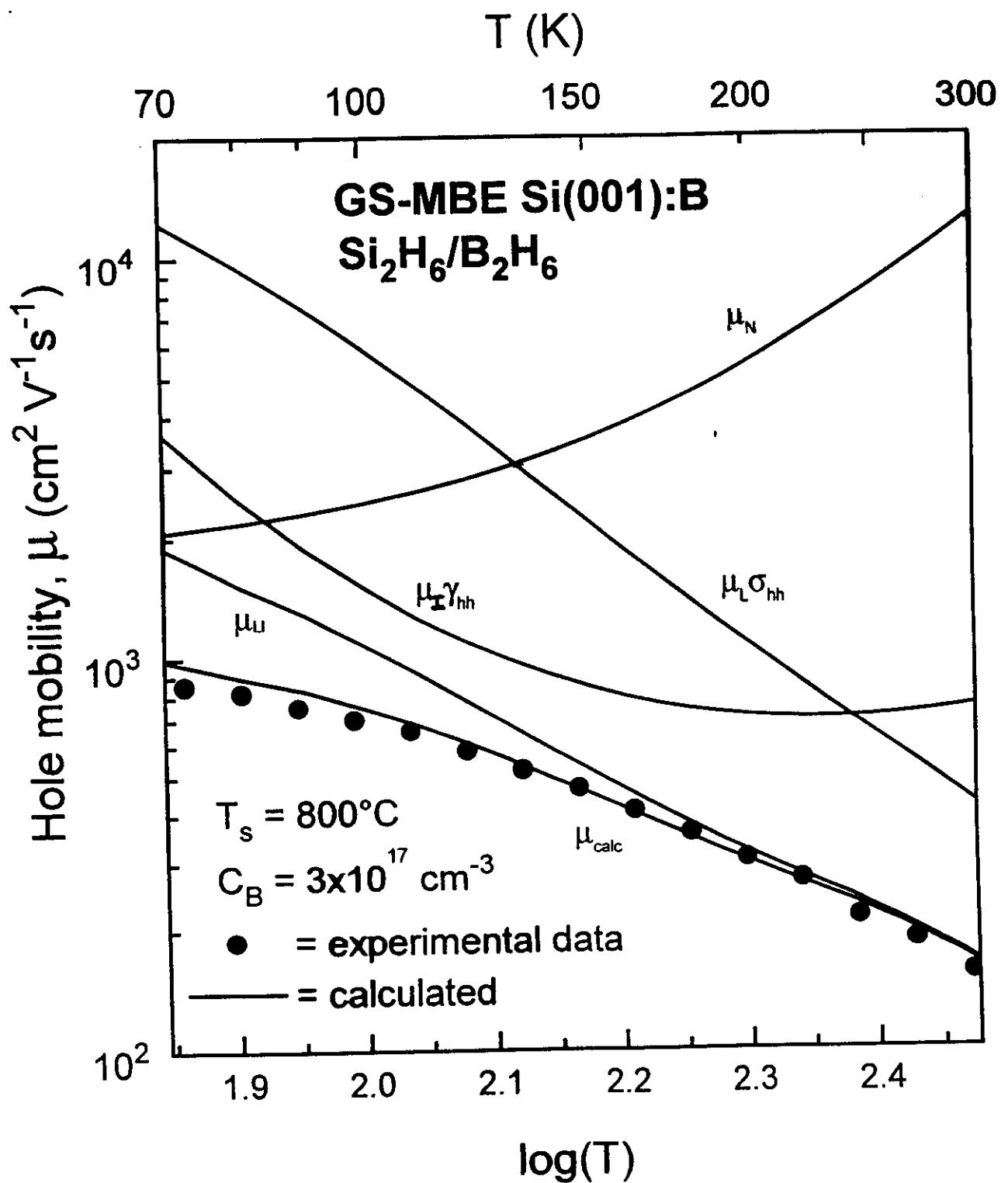


$$S_{\text{B}_2\text{H}_6}/S_{\text{Si}(001)} = \frac{6.4 \times 10^{-4}}{1.4 \times 10^{-3}} \quad @ 600^{\circ}\text{C}$$

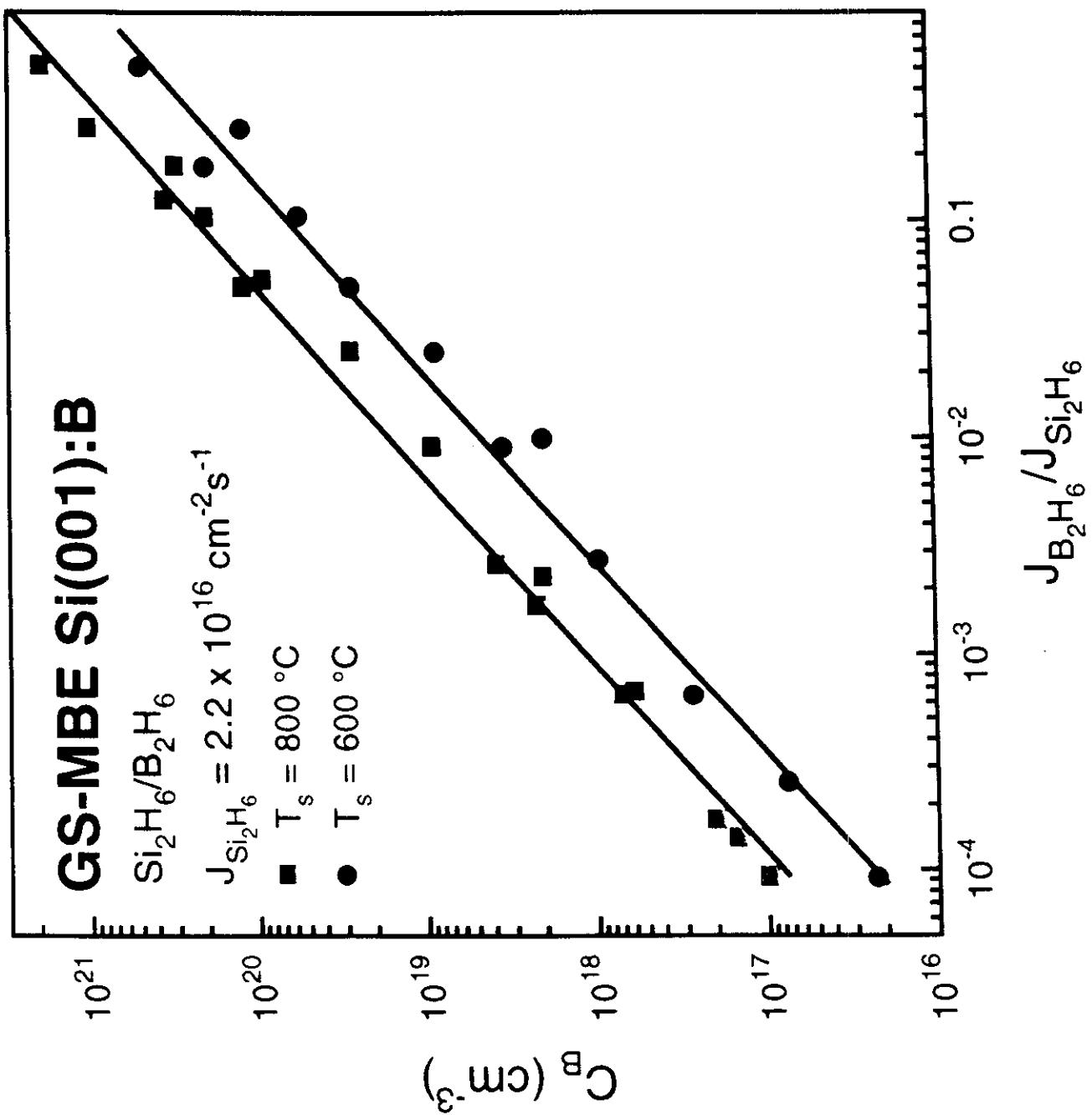
$$9.5 \quad @ 950^{\circ}\text{C}$$

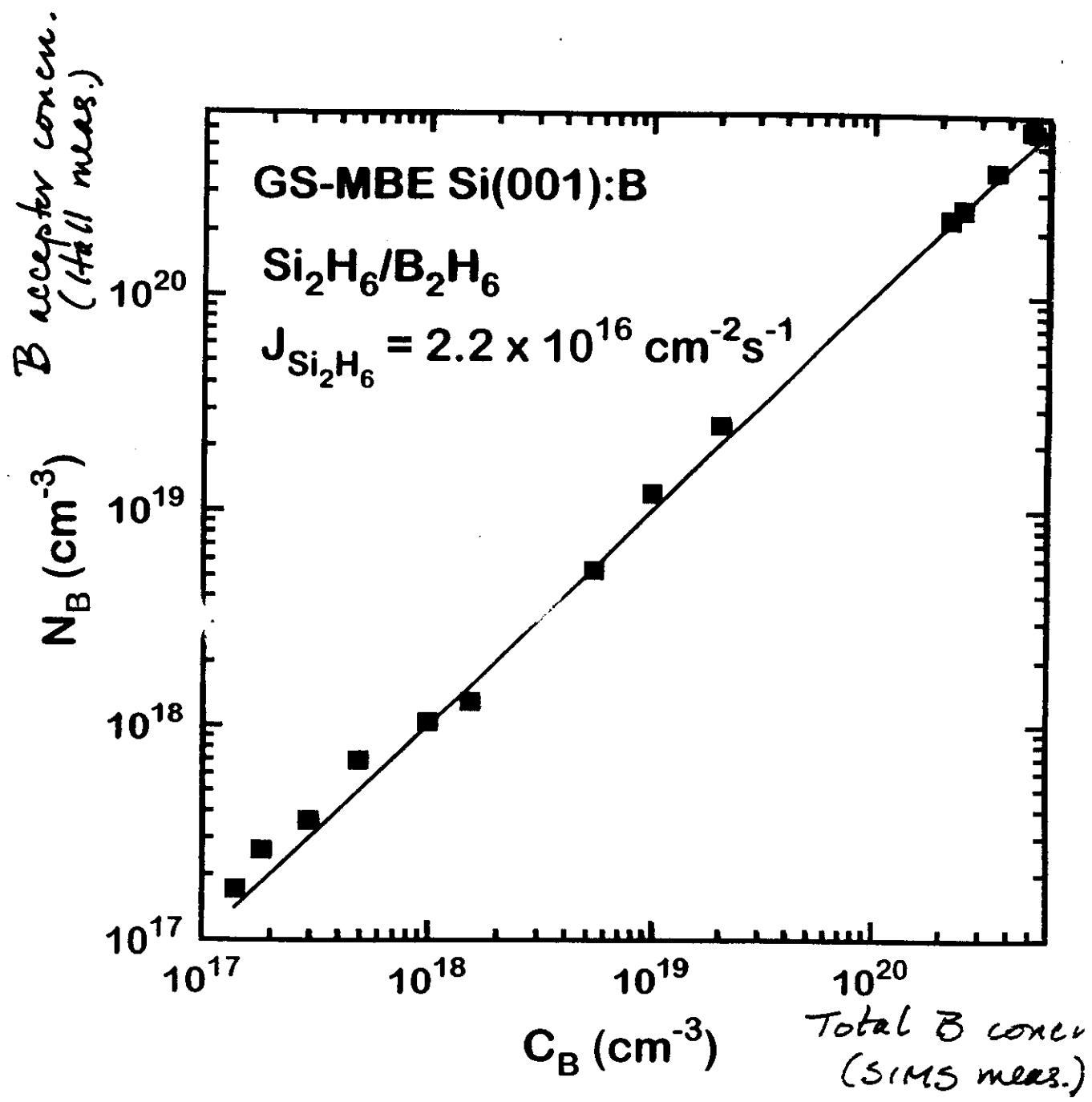


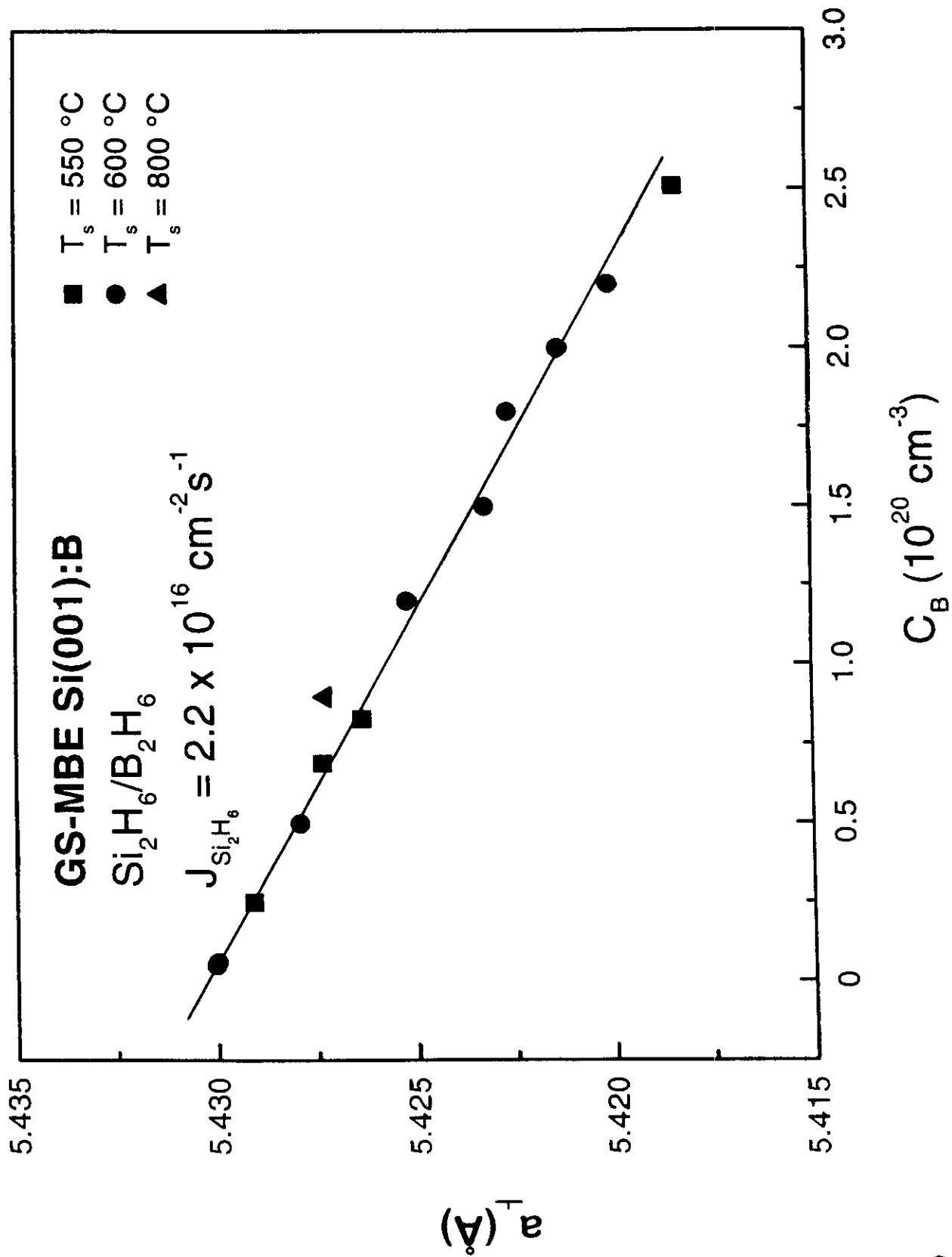




Lu, Bramblett, Lee, Itasan, Karasawa, Greene
J. Appl. Phys. 72, 3067 (1995)





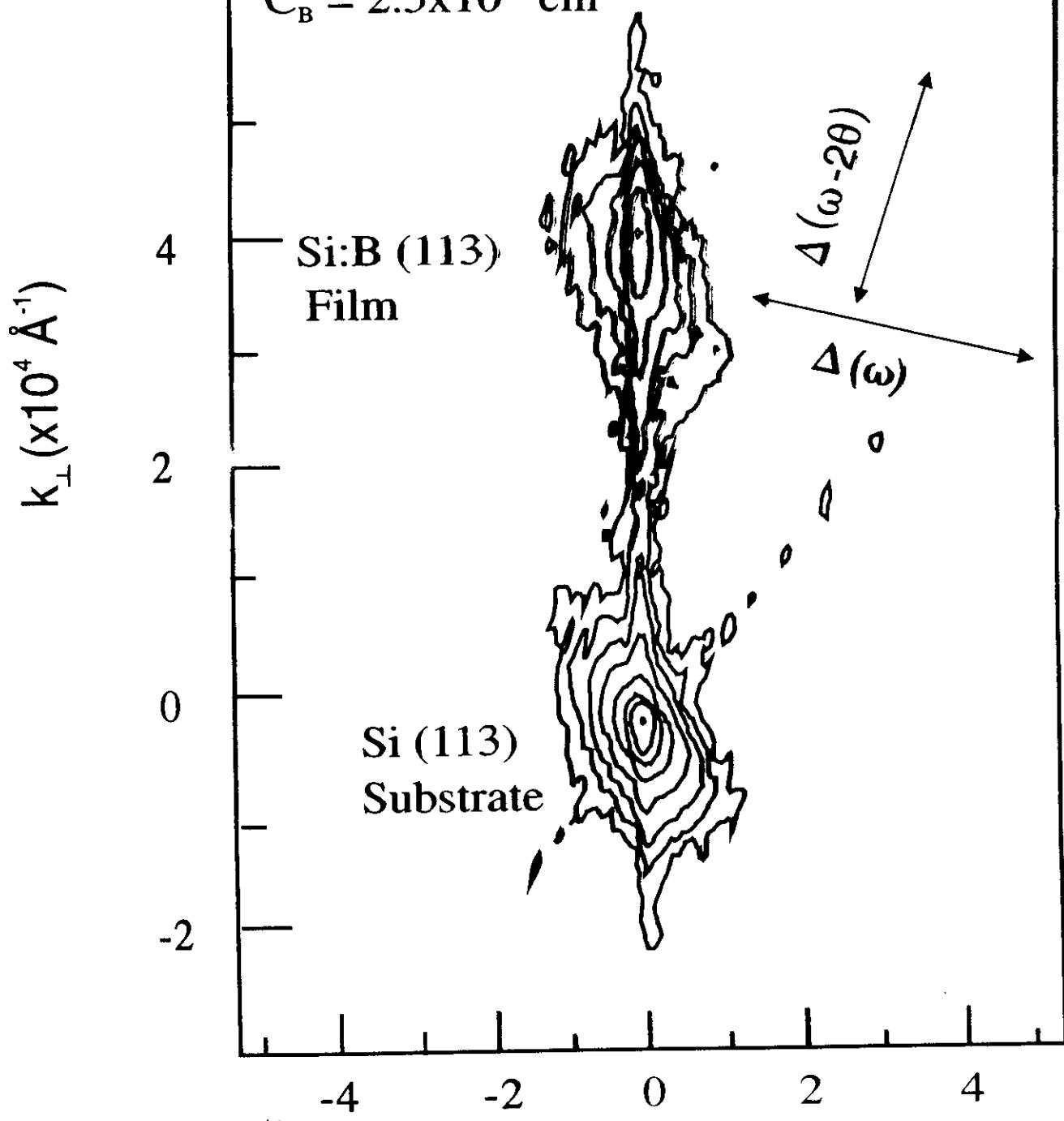


GS-MBE Si(001):B

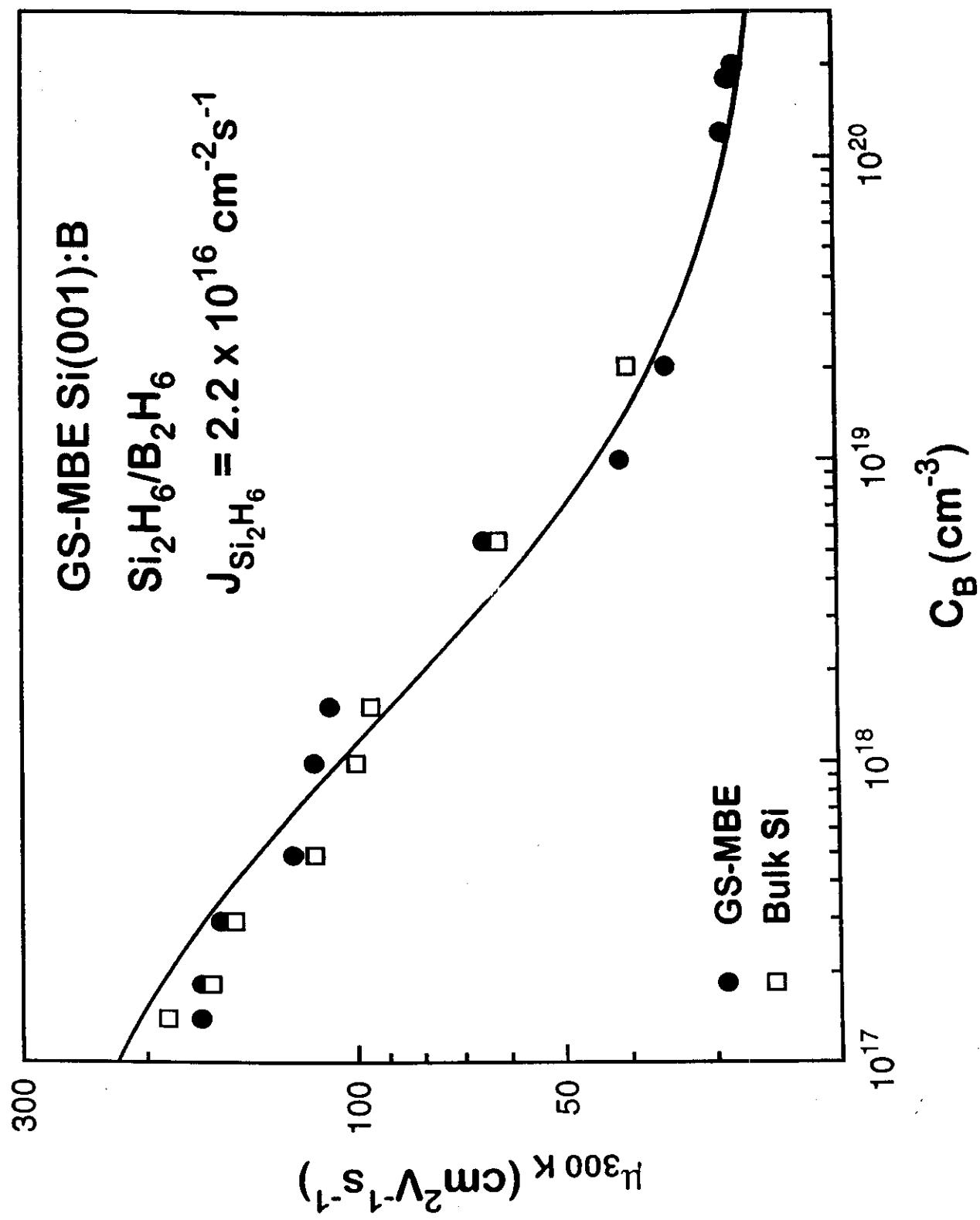
$\text{Si}_2\text{H}_6/\text{B}_2\text{H}_6$

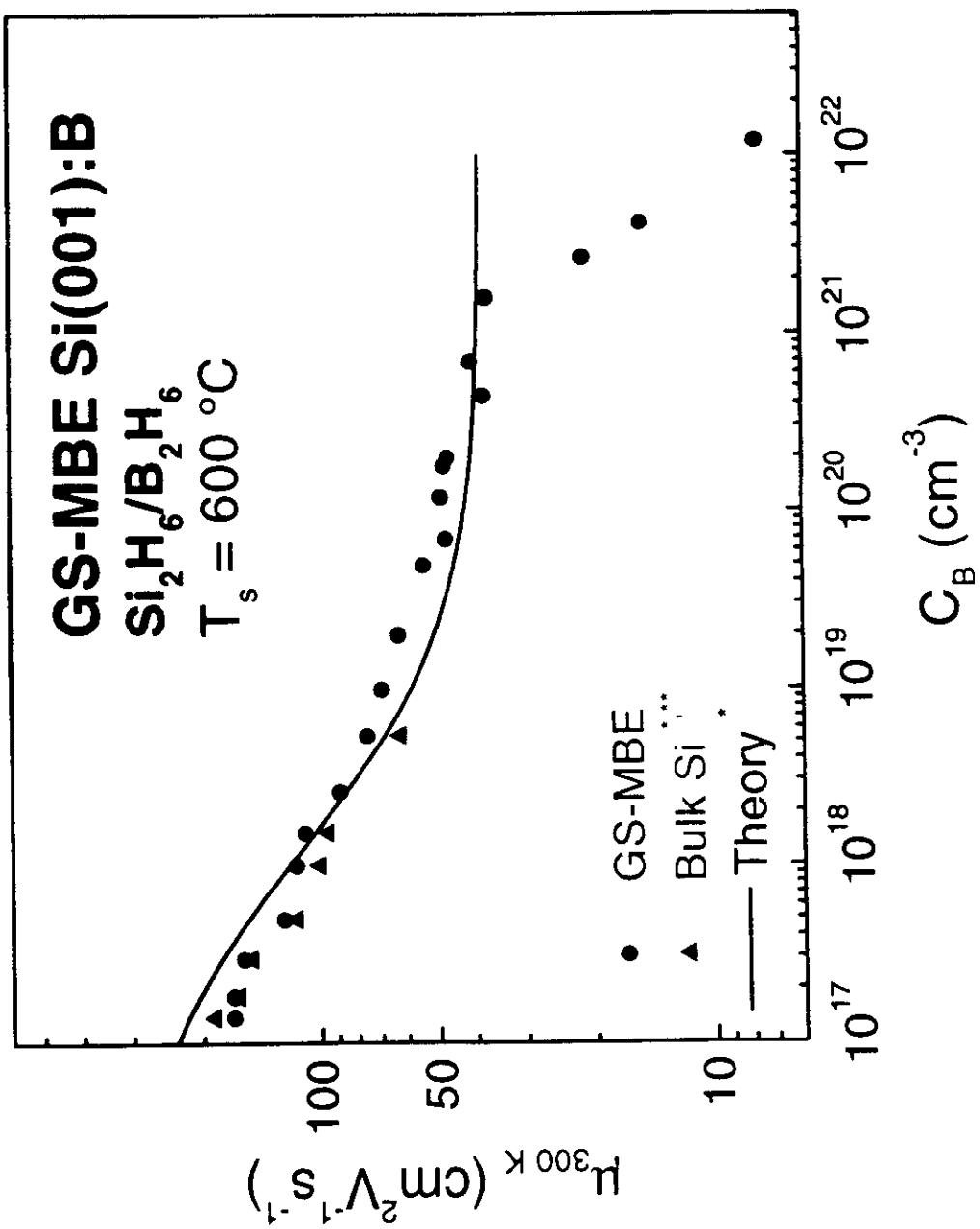
$T_s = 600^\circ \text{C}$

$C_B = 2.5 \times 10^{20} \text{ cm}^{-3}$



$$\frac{\Delta a}{a_{\text{Si}}} \leq 4 \times 10^{-5} !!$$



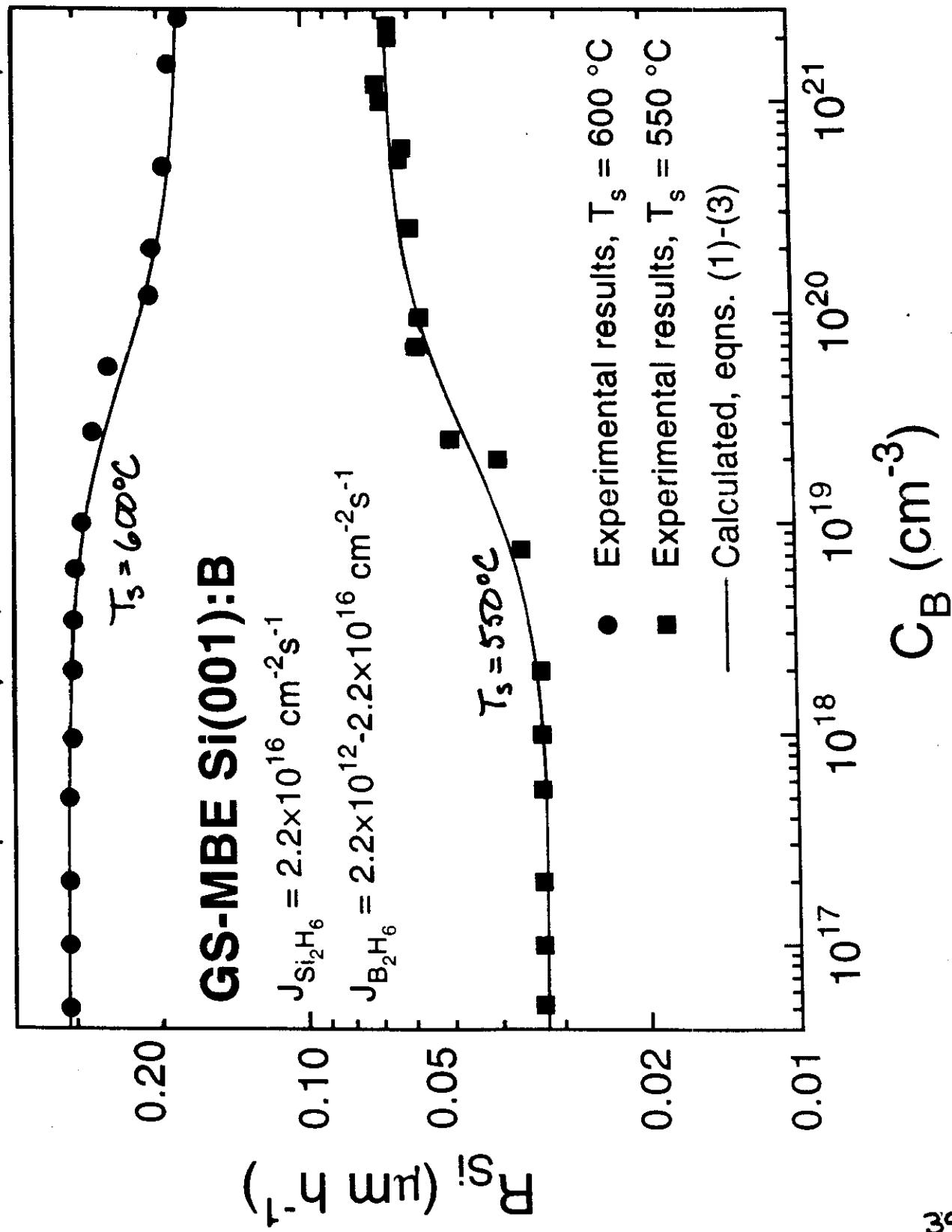


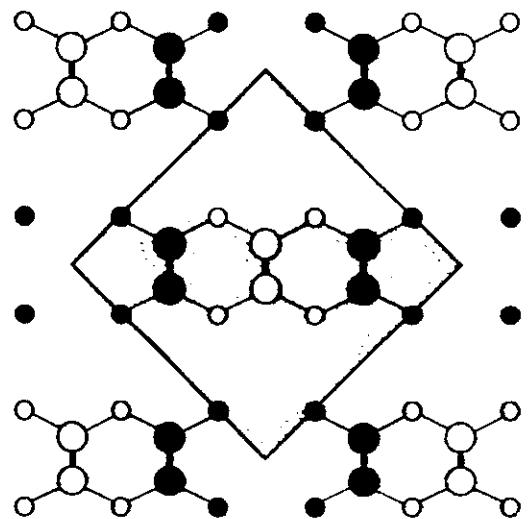
*D. Caughey, R. Thomas, Proc. IEEE 55 2192 (1967).

G. A. Slack, and M. A. Hussain J. Appl. Phys., 70, 2694 (1991).

**S. Wagner, J. Electrochem. Soc. 119 1570 (1972).

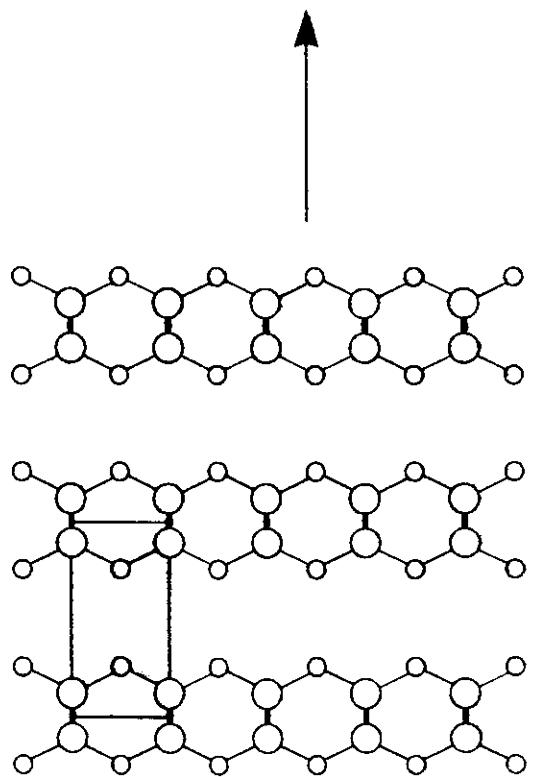
Glass, Igin, Sardela, Lu, Carlsson, Abelson, Greene, Surf. Sci. 392,
463 (1997).



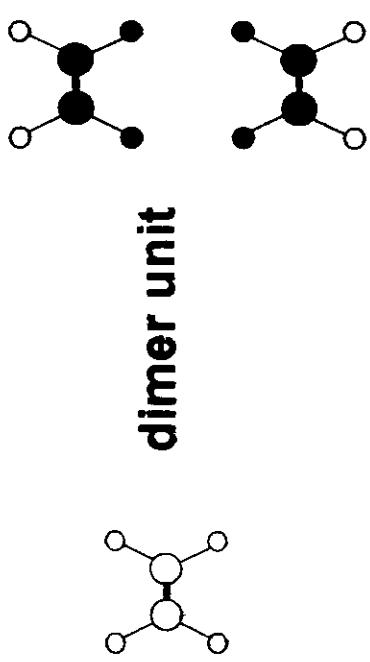


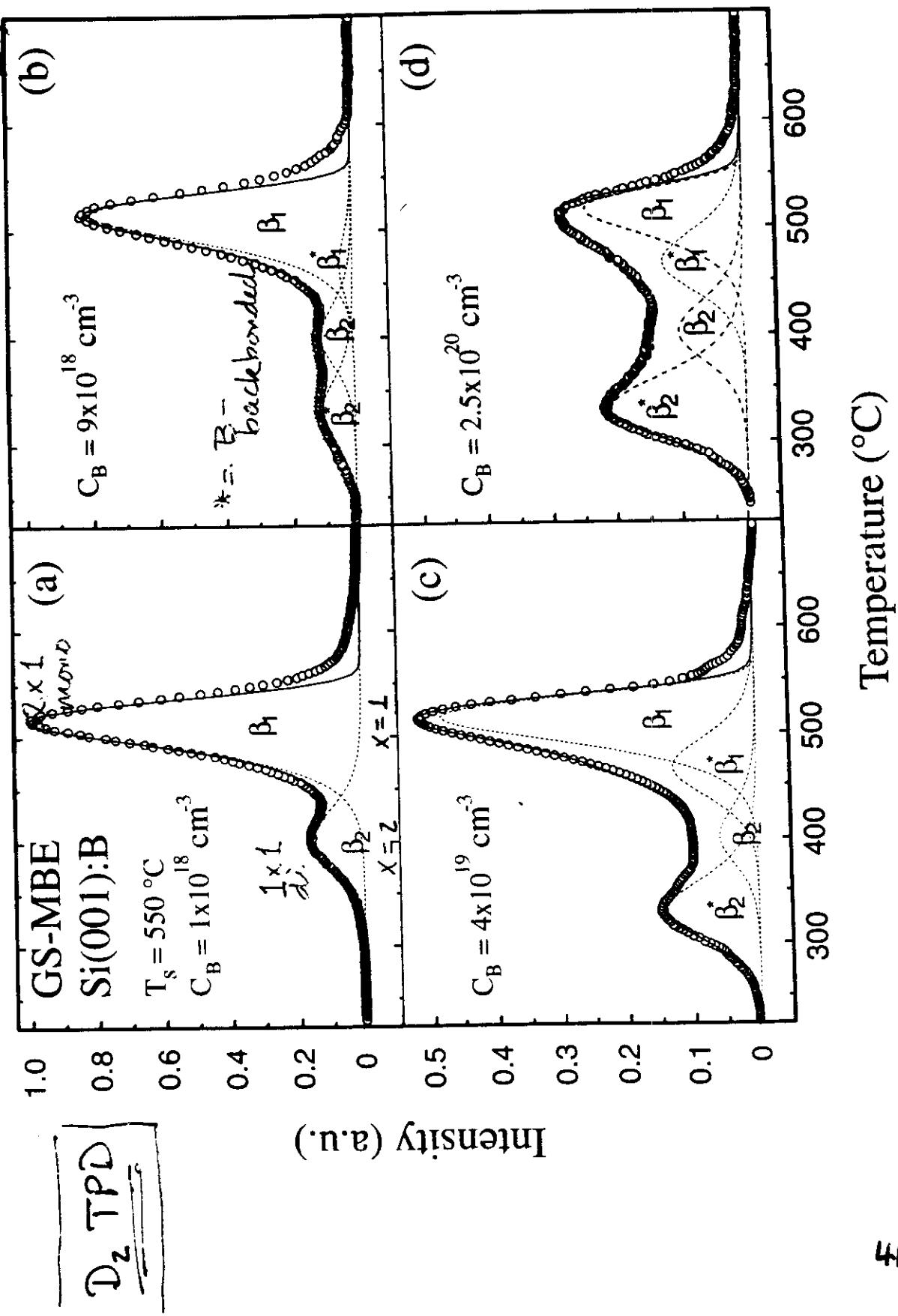
c(4×4) local structure

B-backbonded dimer unit ○ ● top layer Si atoms
2nd layer B atoms ● 2nd layer B atoms
2nd layer Si atoms ○ 2nd layer Si atoms



2×1 reconstruction





GS-MBE Si(001):B

$T_s = 550^\circ\text{C}$

θ_B (ML)

GS-MBE Si(001):B

$$\frac{\theta_B}{C_B} = \frac{1 - \theta_B}{1 - C_B} \exp\left(-\frac{\Delta H_s}{kT_s}\right)$$

$$\begin{aligned} \theta_{sat,B} &= 0.5 \text{ ML} \\ \Delta H_s &= -0.53 \text{ eV} \end{aligned}$$

■ Experimental results

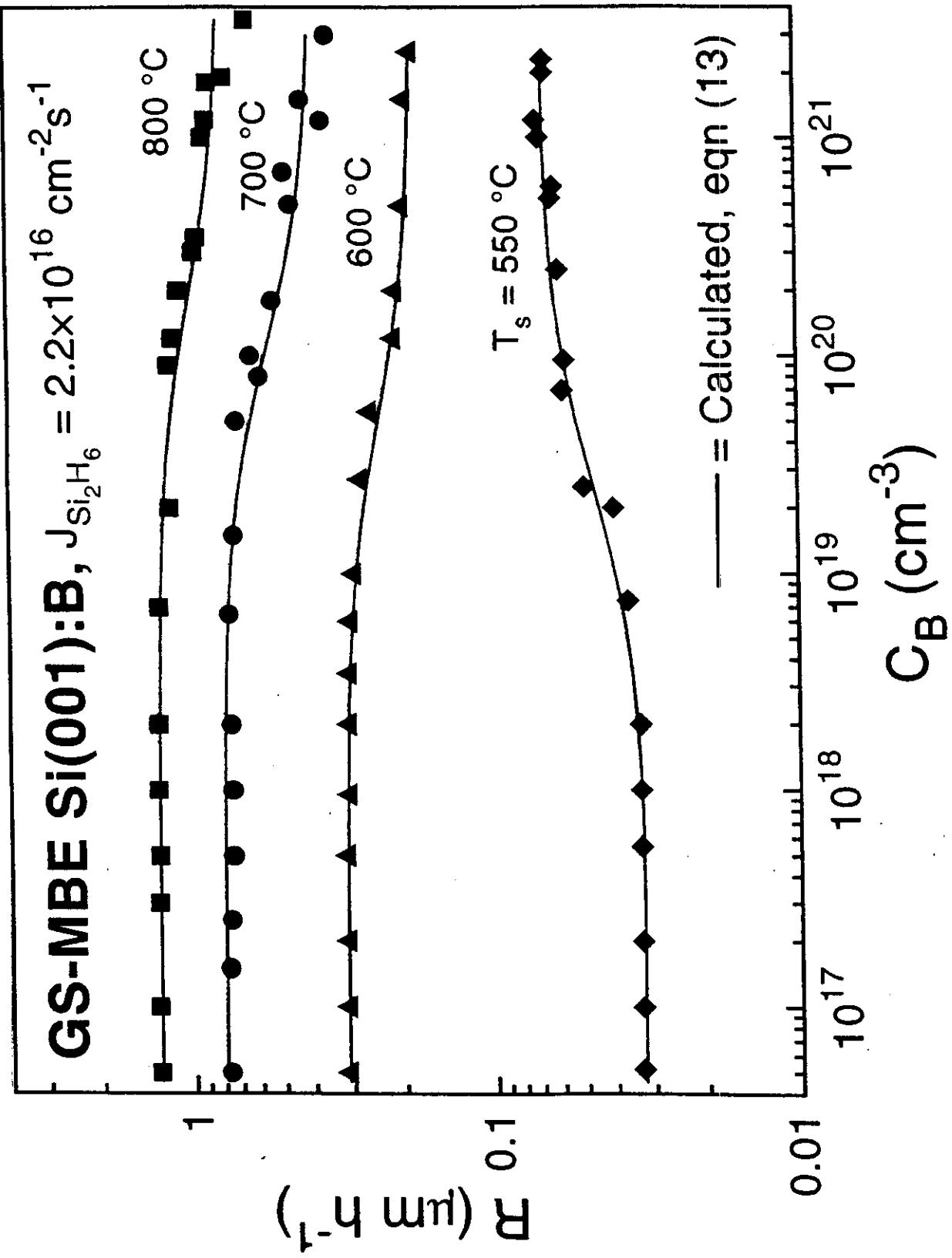
— Calculated, eqn. (8)

$C_B (\text{cm}^{-3})$

10¹⁸ 10¹⁹ 10²⁰ 10²¹

Kurt Wieser, Günter Tröger, Peter Aßelmann, Greene, T., Appl. Phys. 1981, 32, 271

$$R_{Si} = \frac{2 J_{Si_2H_6} T_{Si_2H_6}}{n_{Si}} \Theta_{dB}^2 \quad \text{where: } \Theta_{dB} = f(f^*, C_B, \Theta_S)$$



GS-MBE/UHV-CVD Si(001) : B
 Growth Kinetics ($\text{Si}_2\text{H}_6/\text{B}_2\text{H}_6$)

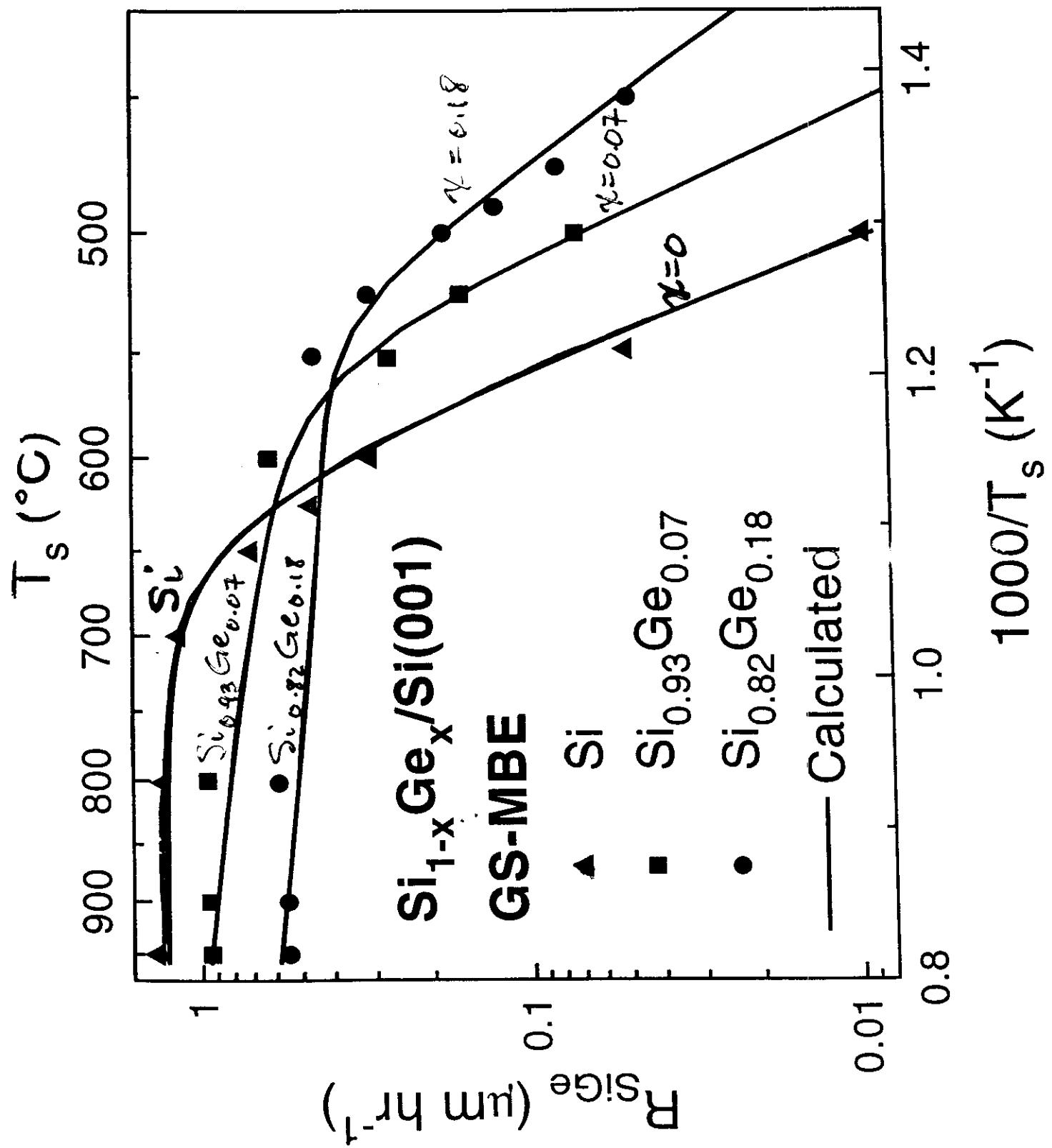
$$R_{\text{Si}} = \frac{2 J_{\text{Si}_2\text{H}_6} S_{\text{Si}_2\text{H}_6} \Theta_{\text{db}}^2}{N_{\text{Si}}} ; \boxed{R_{\text{Si}} \propto \Theta_{\text{db}}^2}$$

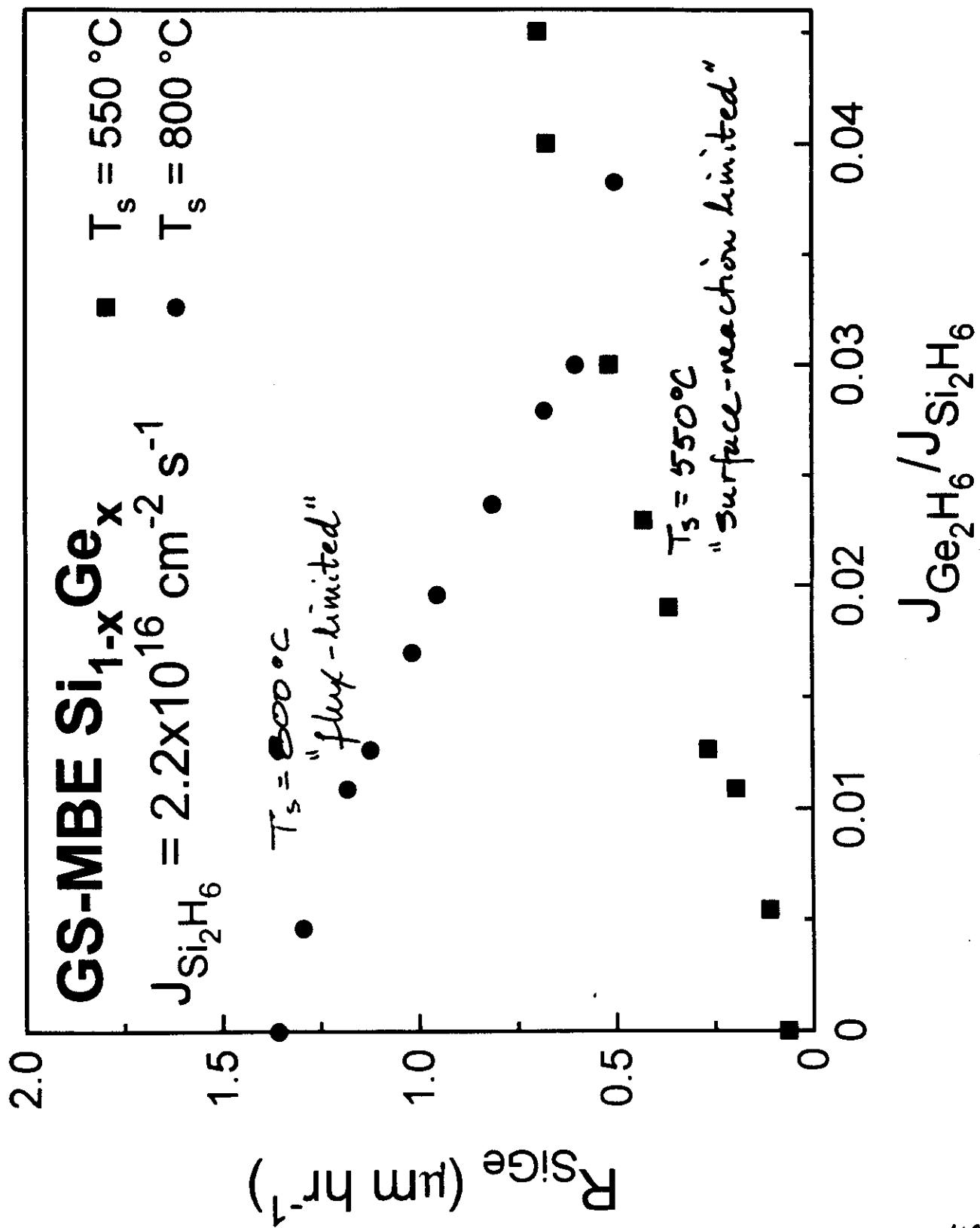
At $C_B \geq 1 \times 10^{19} \text{ cm}^{-3}$:

- $\Theta_{\text{db}} \uparrow$ due to
 - * Si dimer vacancies.
 - * deactivation of Si_{db}^* ($\approx 1 \text{ db/Si}_2^*$).
- But, $E_{\text{d},\text{H}_2} \uparrow$ due to
 - * weakening of H-Si $^+$ bonds caused by Si $^{+2}$ -B backdonor charge transfer and increased strain.

Thus:

- In the flux-limited ($hi T_s$) growth mode
 - * $\Theta_{\text{H}} \rightarrow 0$
 - * $\therefore R_{\text{Si}} \uparrow$ since $\Theta_{\text{db}} \uparrow$
- In the surface-reaction ($lo T_s$) growth mode
 - * Θ_{H} is large
 - * $\therefore R_{\text{Si}} \uparrow$ since net $\Theta_{\text{db}} \uparrow$





Calculating gas-source $\text{Si}_{1-x}\text{Ge}_x$
growth rates requires:

$$\left. \begin{array}{l} S_{\text{Si}_{1-x}\text{Ge}_x}(\theta_{\text{Ge}}, T_s) \\ S_{\text{Ge}_{1-x}\text{Si}_x}(\theta_{\text{Ge}}, T_s) \end{array} \right\} \left. \begin{array}{l} \text{Si: } S_{\text{Si}_{1-x}\text{Ge}_x}, S_{\text{Si}_{1-x}\text{Ge}_x}^{\text{Ge}} \\ \text{Ge: } S_{\text{Ge}_{1-x}\text{Si}_x}, S_{\text{Ge}_{1-x}\text{Si}_x}^{\text{Ge}} \end{array} \right.$$

$$\theta_{\text{Ge}}(x, T_s) > \Delta H_{\text{Si}, \text{Ge}} [\theta_{\text{H}}(x, T_s)]$$

$$E_{\text{Si}}^H(\theta_{\text{Ge}})$$

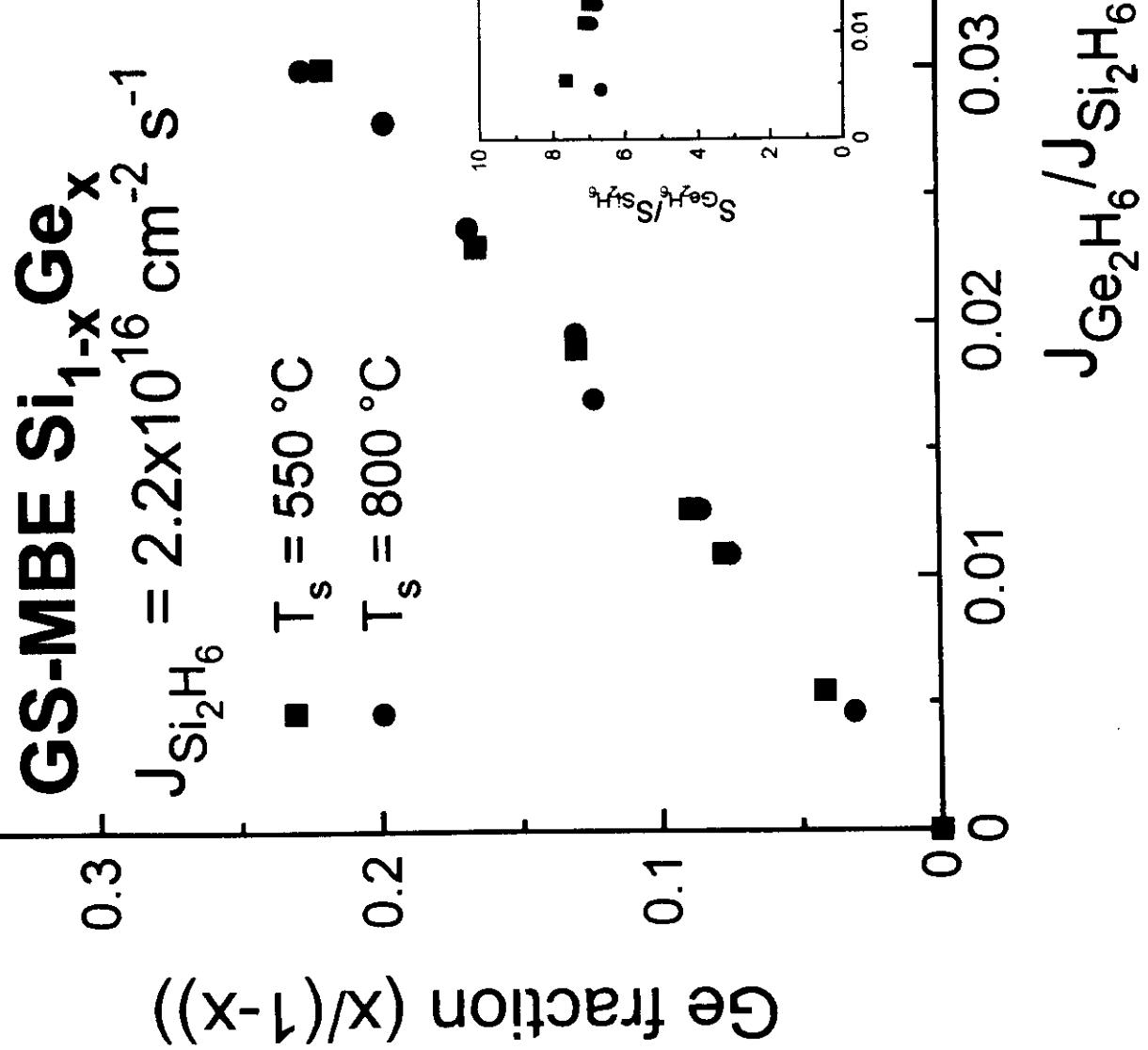
$$E_{\text{Ge}}^H(\theta_{\text{Ge}})$$

$$f_{\text{Si}, \text{db}}(x, \theta_{\text{Ge}}, T_s)$$

$$f_{\text{Ge}, \text{db}}(x, \theta_{\text{Ge}}, T_s)$$

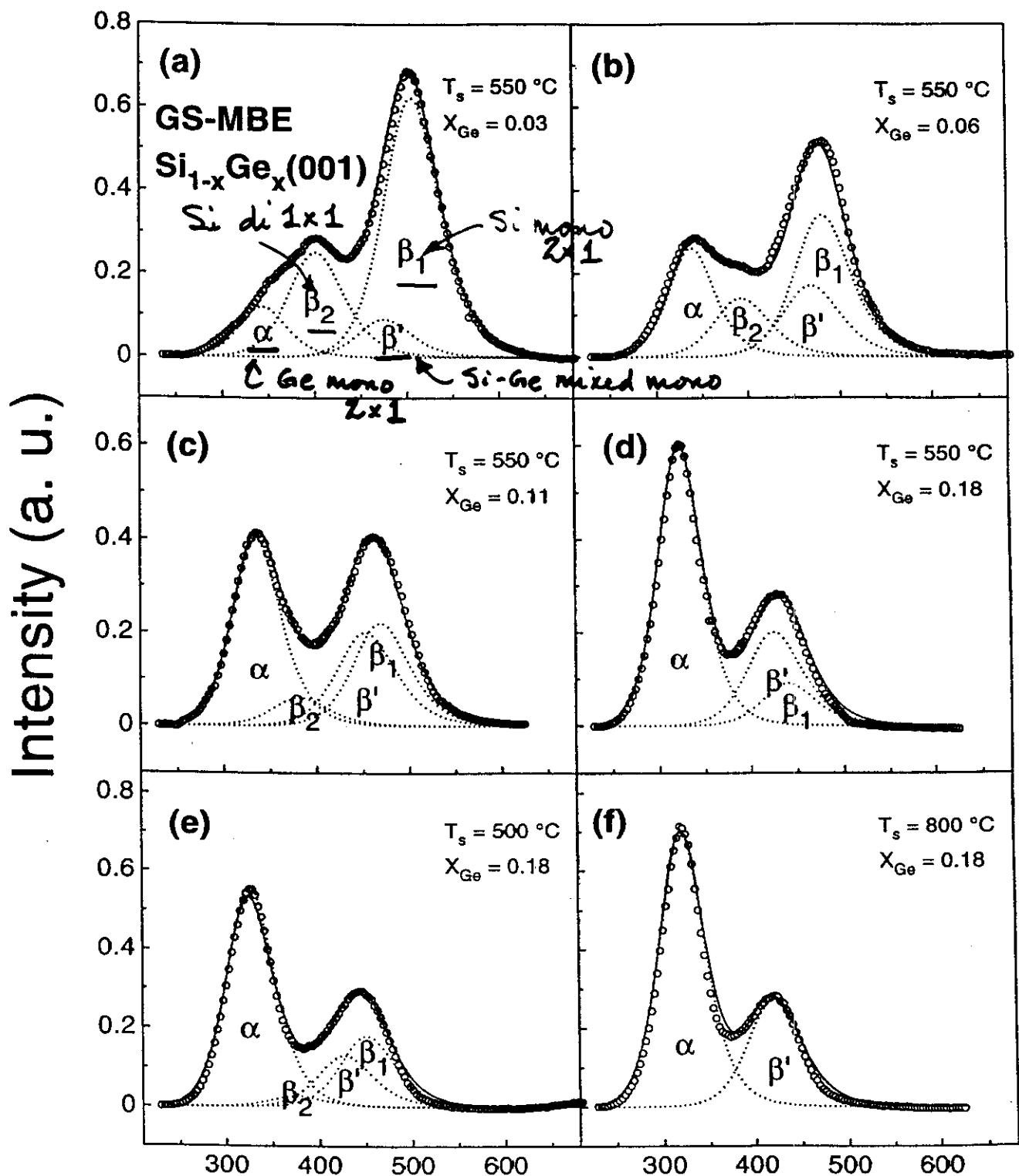
Knowledge of H desorption pathway from Si:Ge

Kinetic order of all adsorption, surface decomposition, and desorption steps



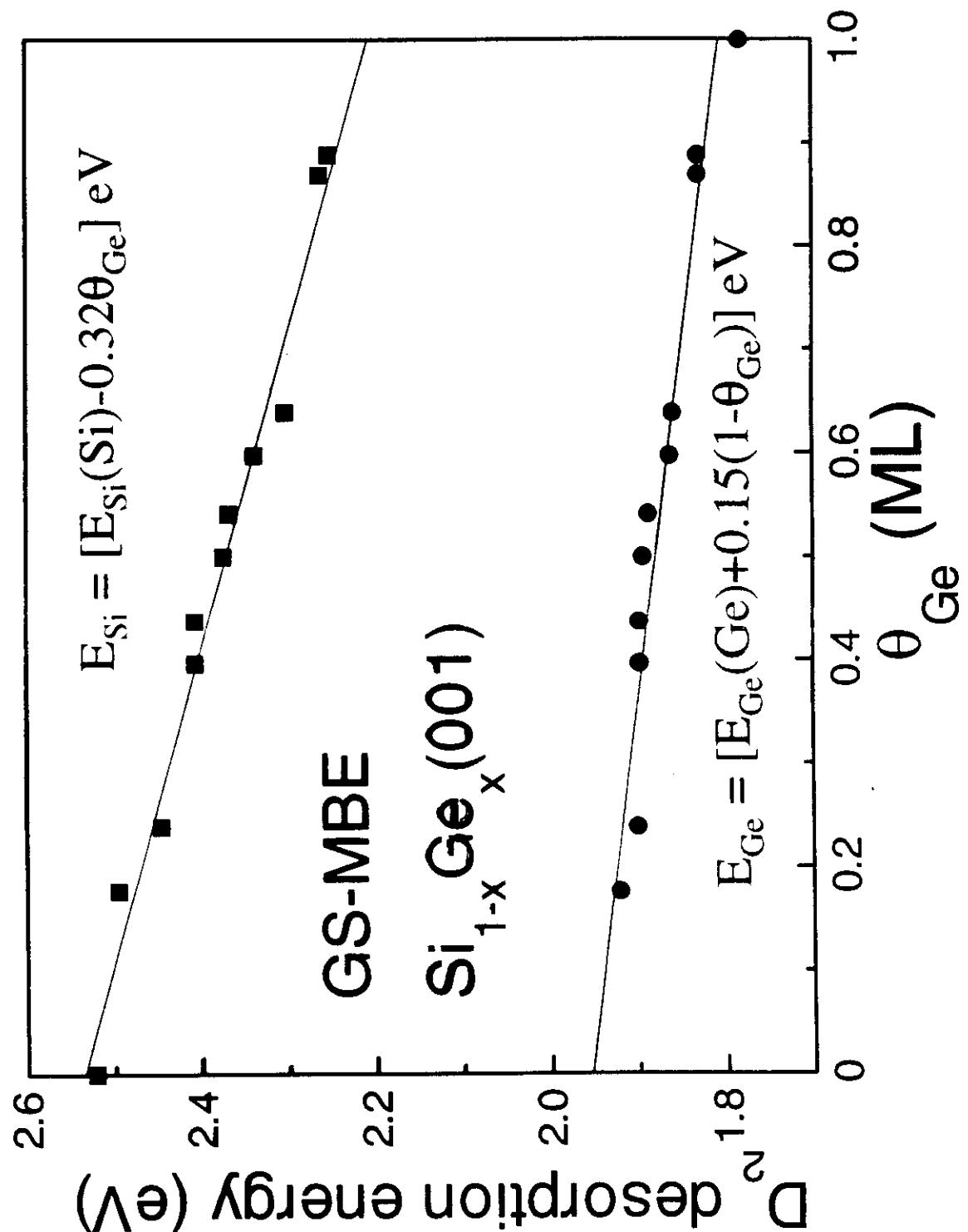
β_1 = Si monodeuteride, 2×1
 β_2 = Si dideuteride, 1×1

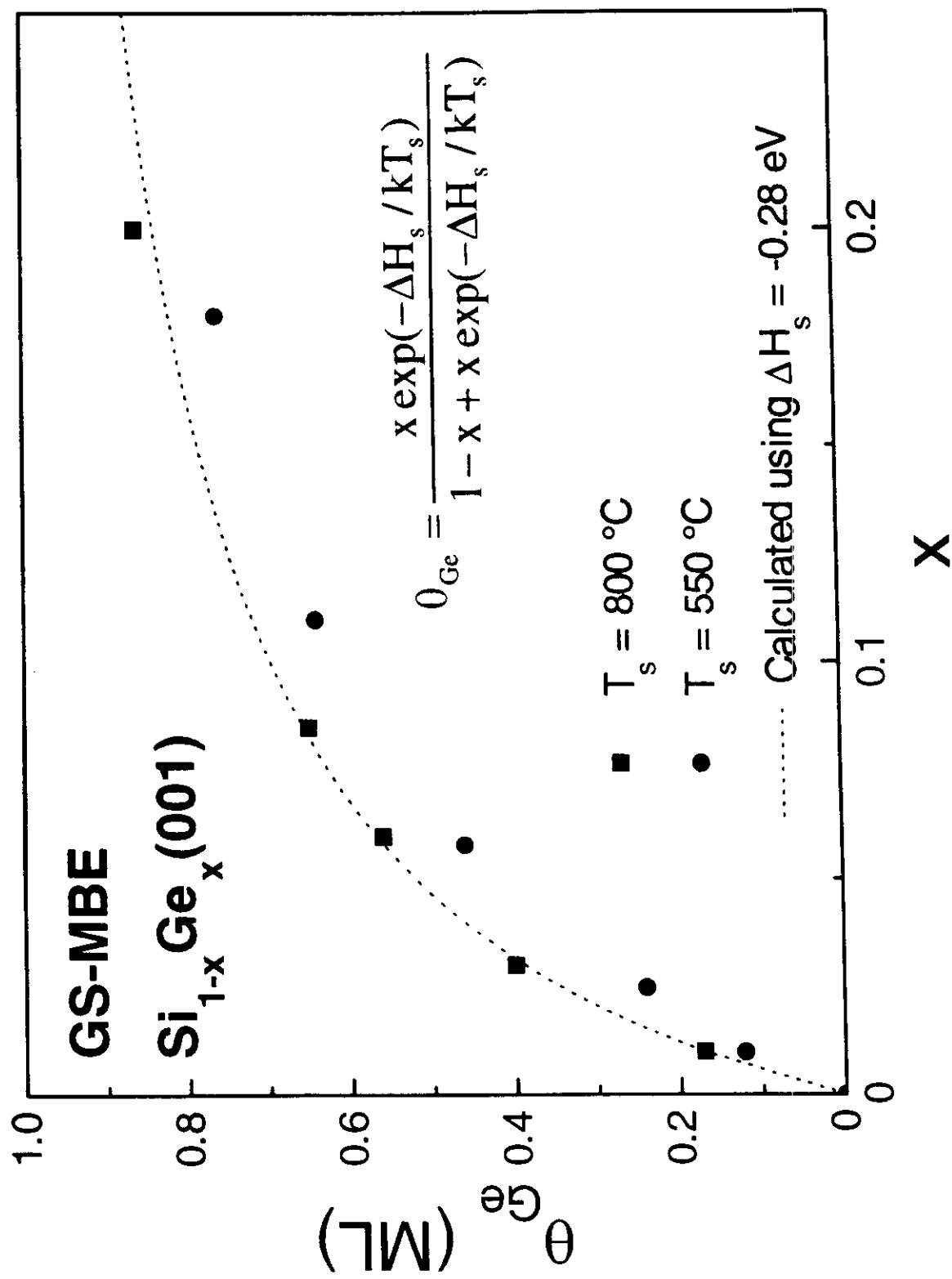
β' = Ge-Si monodeuteride
 α = Ge monodeuteride

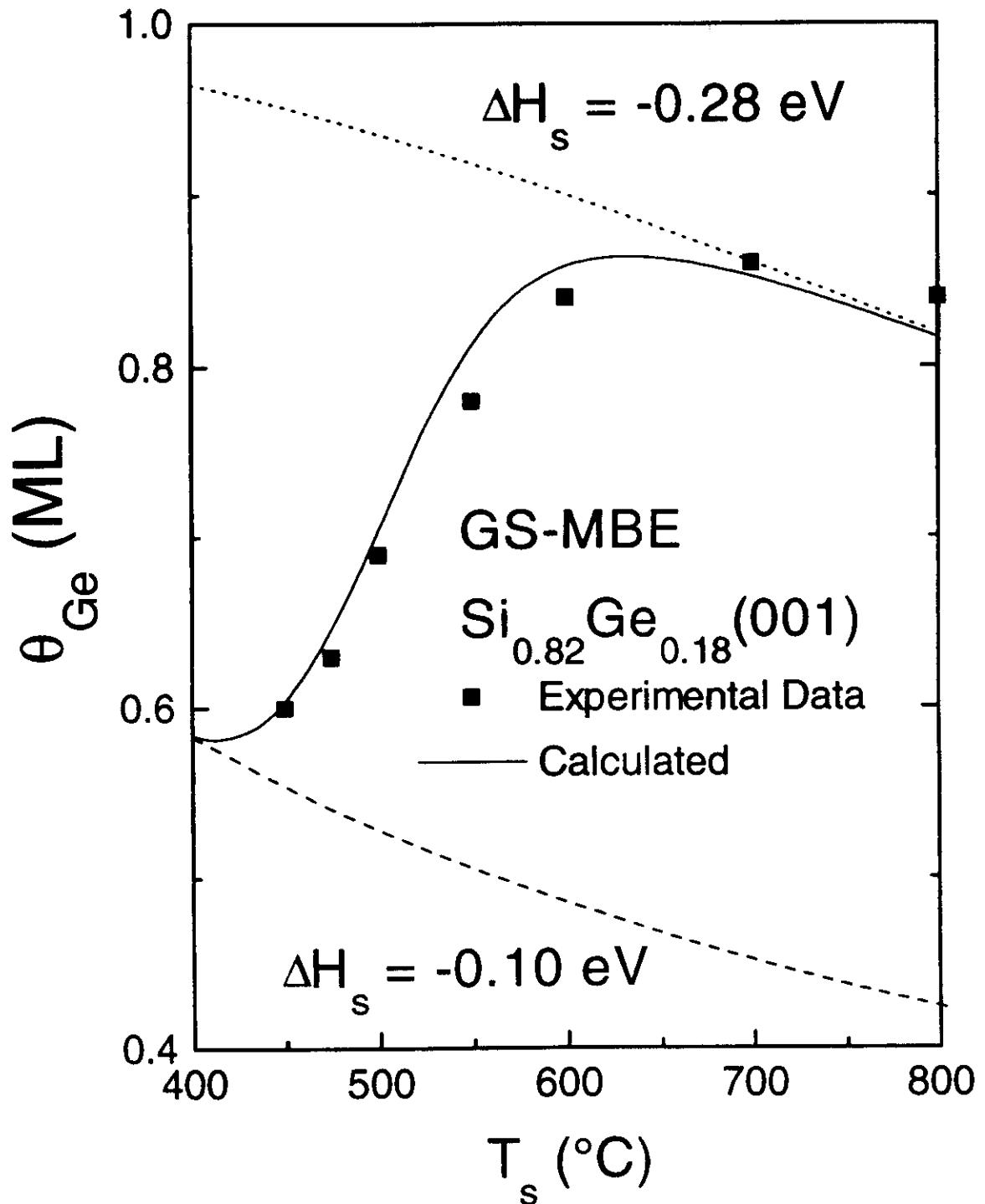


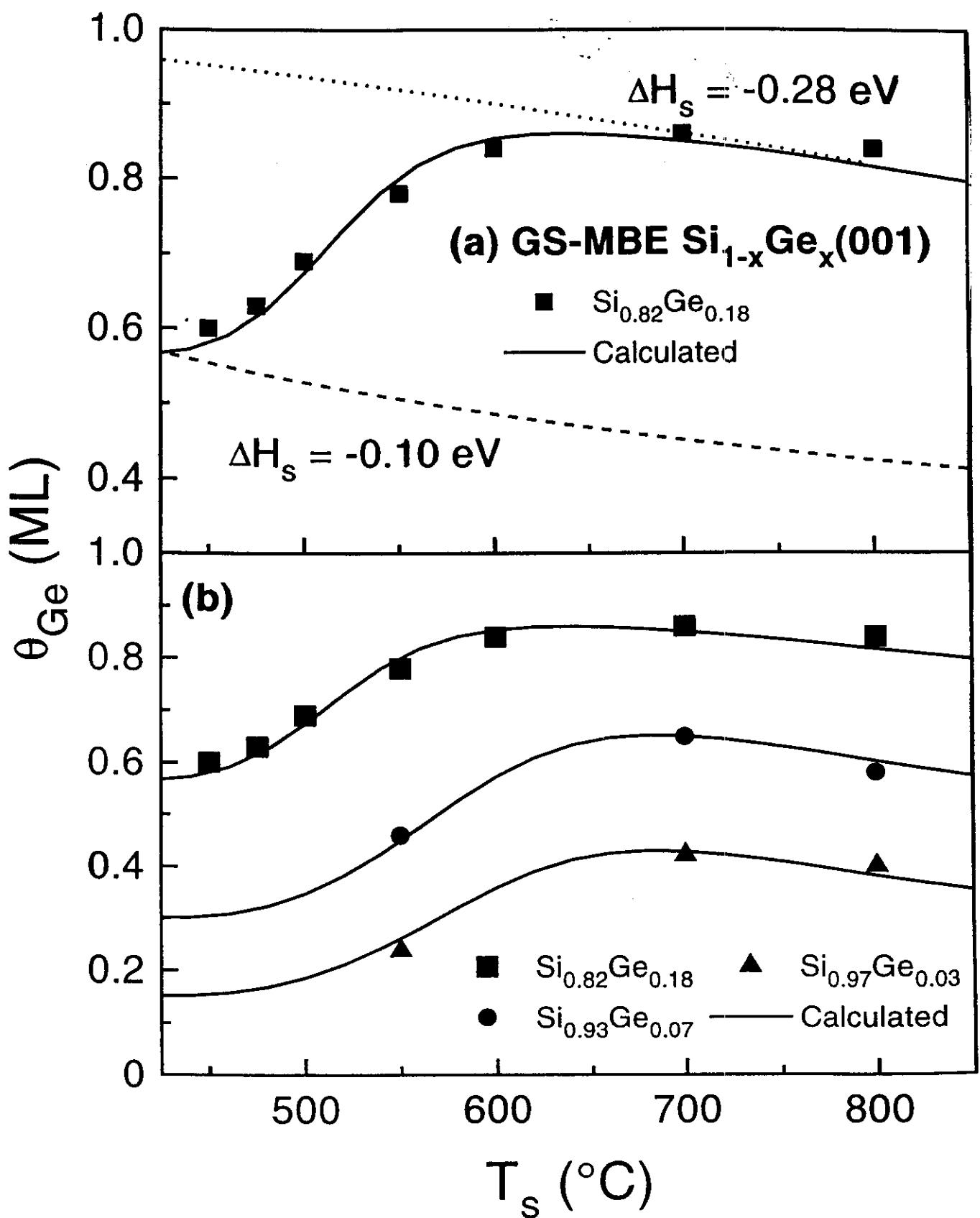
$$I_{\beta_1} + I_{\beta'_1} + I_\alpha = \text{constant} \Rightarrow n_{\text{Si}} \quad (\text{i.e. fully-strained})$$

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Gas-source Si_{1-x}Ge_x growth rate

$$R_{Si,Ge} = \frac{2\theta_{Si} f_{Si,db}^2}{N_{Si,Ge}} \left(J_{Si_2H_6} S_{Si_2H_6}^{Si} + J_{Ge_2H_6} S_{Ge_2H_6}^{Si} \right) \\ + \frac{2\theta_{Ge} f_{Ge,db}^2}{N_{Si,Ge}} \left(J_{Si_2H_6} S_{Si_2H_6}^{Ge} + J_{Ge_2H_6} S_{Ge_2H_6}^{Ge} \right)$$

$$\theta_{Ge} = f_{Si,db} \left(\frac{x \exp(-\Delta H_s^1/kT_s)}{1 - x + x \exp(-\Delta H_s^1/kT_s)} \right) \\ + (1 - f_{Si,db}) \left(\frac{x \exp(-\Delta H_s^2/kT_s)}{1 - x + x \exp(-\Delta H_s^2/kT_s)} \right)$$

where:
 $\Delta H_s^1 = -0.28 \text{ eV}$
 $\Delta H_s^2 = -0.10 \text{ eV}$

$$\theta_{Ge} + \theta_{Si} = 1$$

$$f_{Si,db} = \left[1 + \sqrt{\frac{2J_{Si_2H_6} S_{Si_2H_6}}{N_{Si,Ge} k_b \exp(-E_{Si}^H/kT_s)}} \right]$$

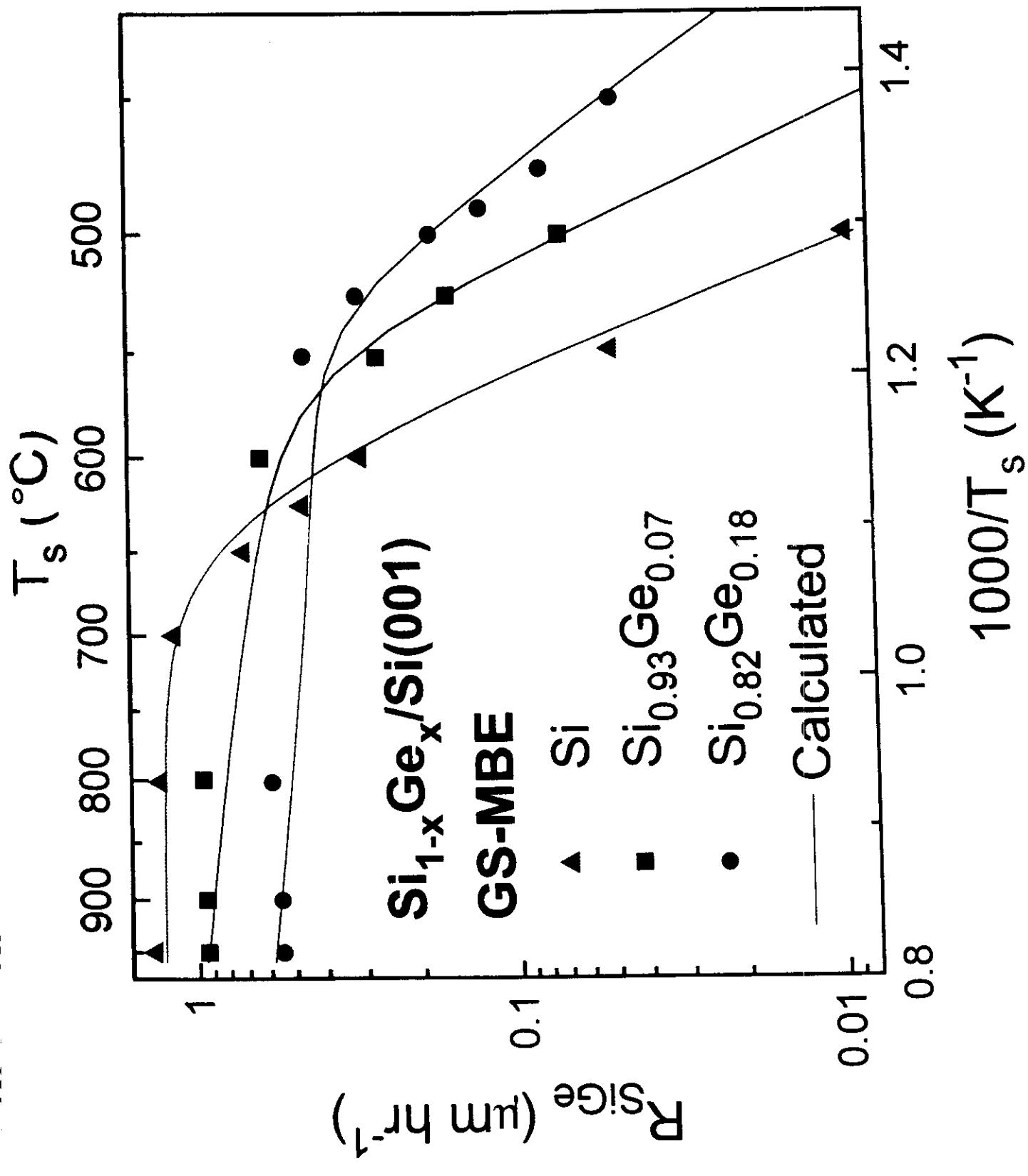
$$f_{Ge,db} = \left[1 + \sqrt{\frac{2J_{Ge_2H_6} S_{Ge_2H_6}}{N_{Si,Ge} k_b \exp(-E_{Ge}^H/kT_s)}} \right]$$

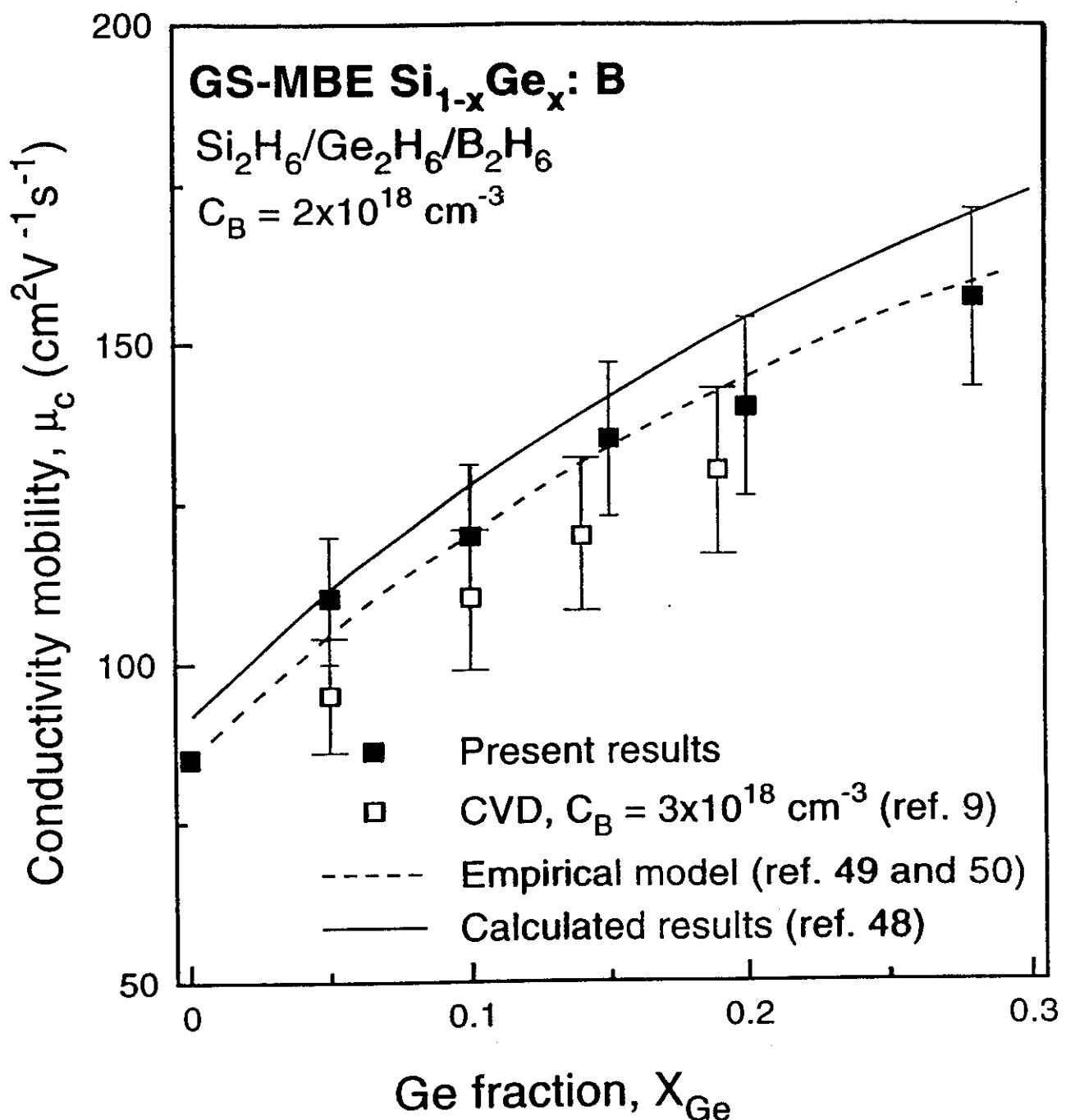
$$\bar{E}_{Si}^H = [2.52 - 0.32 \theta_{Ge}]$$

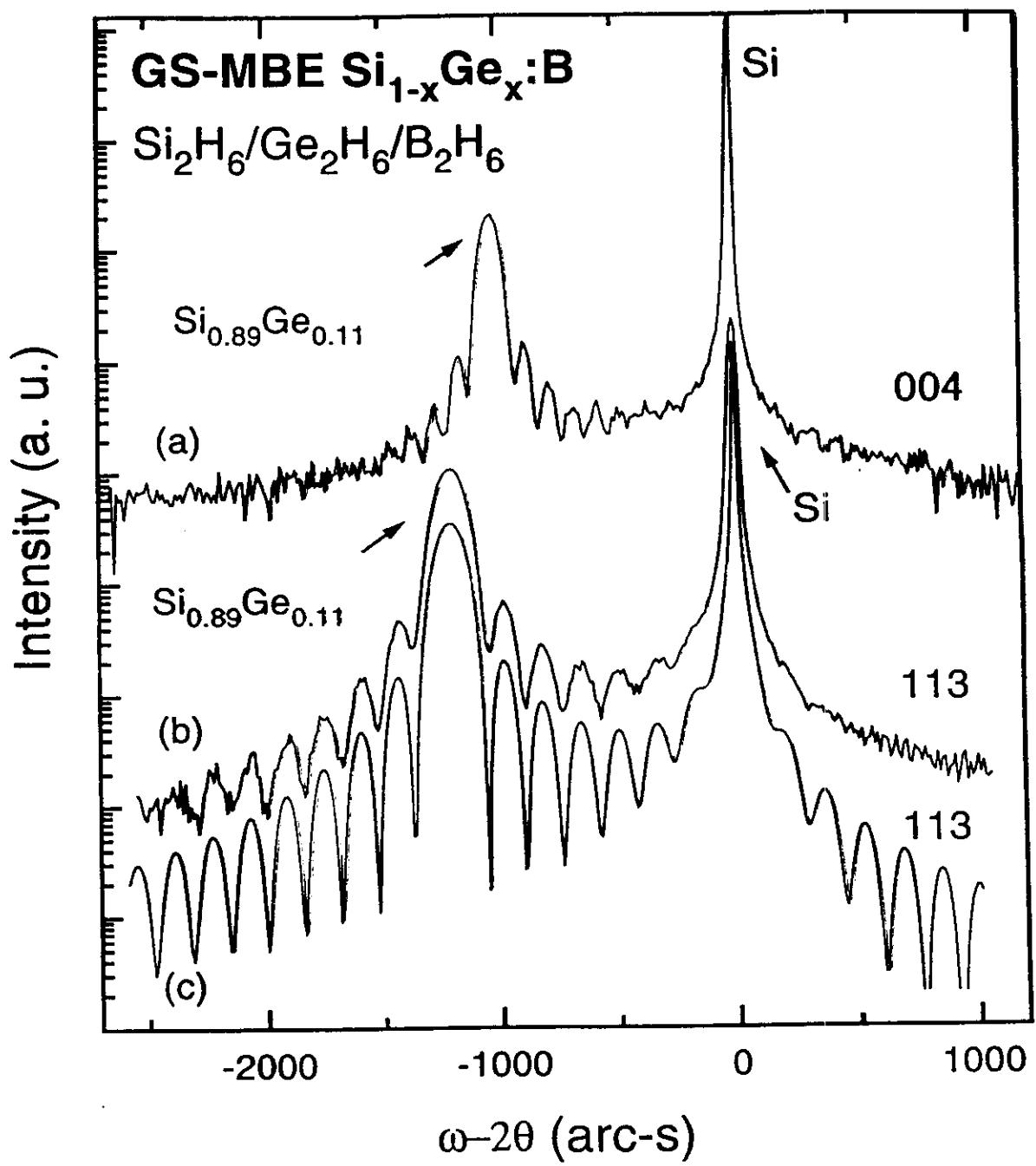
$$\bar{E}_{Ge}^H = [1.80 + 0.15 (1 - \theta_{Ge})]$$

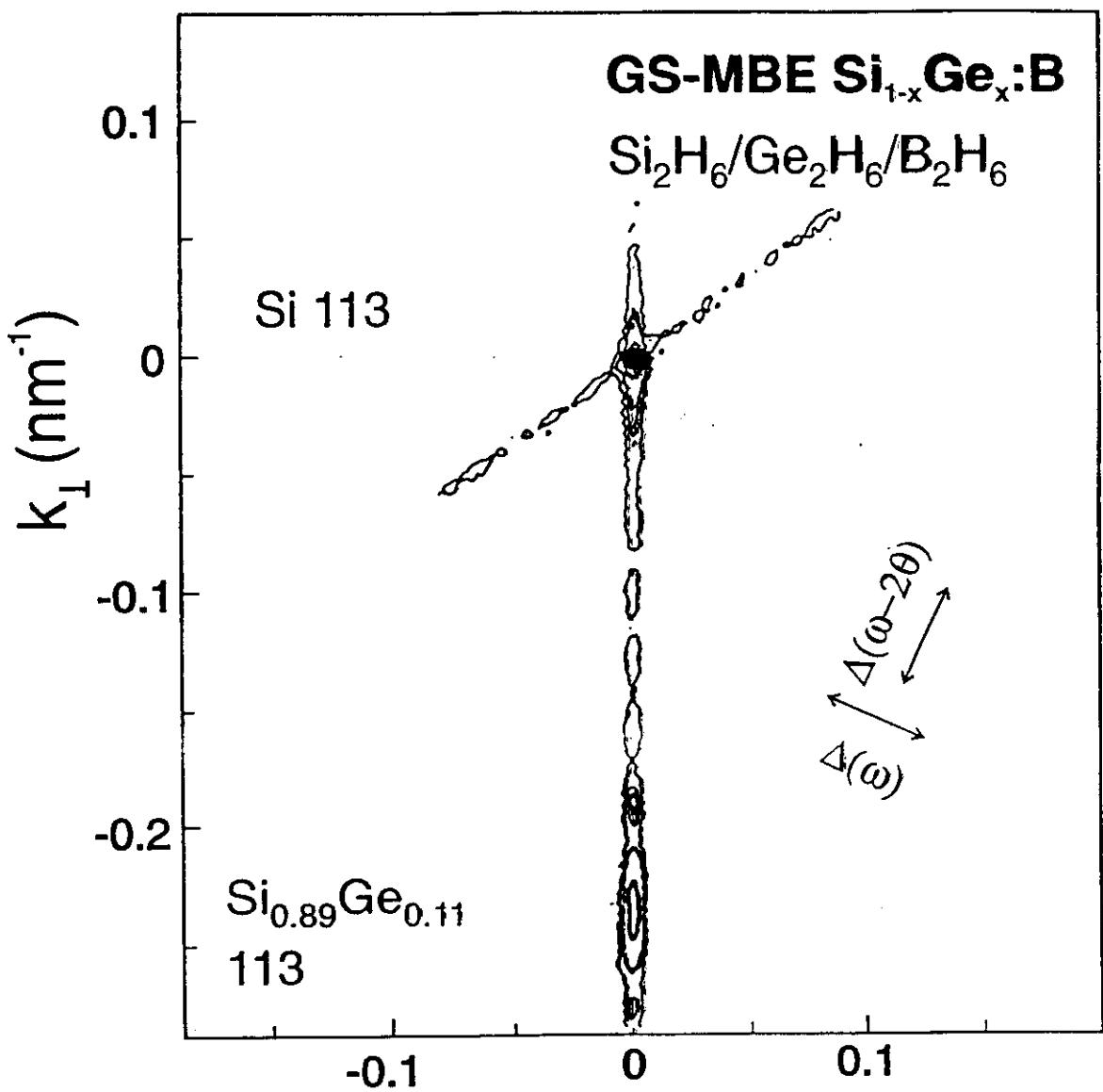
$$S_{Si_2H_6} = \theta_{Si} S_{Si_2H_6}^{Si} + \theta_{Ge} S_{Si_2H_6}^{Ge}$$

$$S_{Ge_2H_6} = \theta_{Ge} S_{Ge_2H_6}^{Ge} + \theta_{Si} S_{Ge_2H_6}^{Si}$$





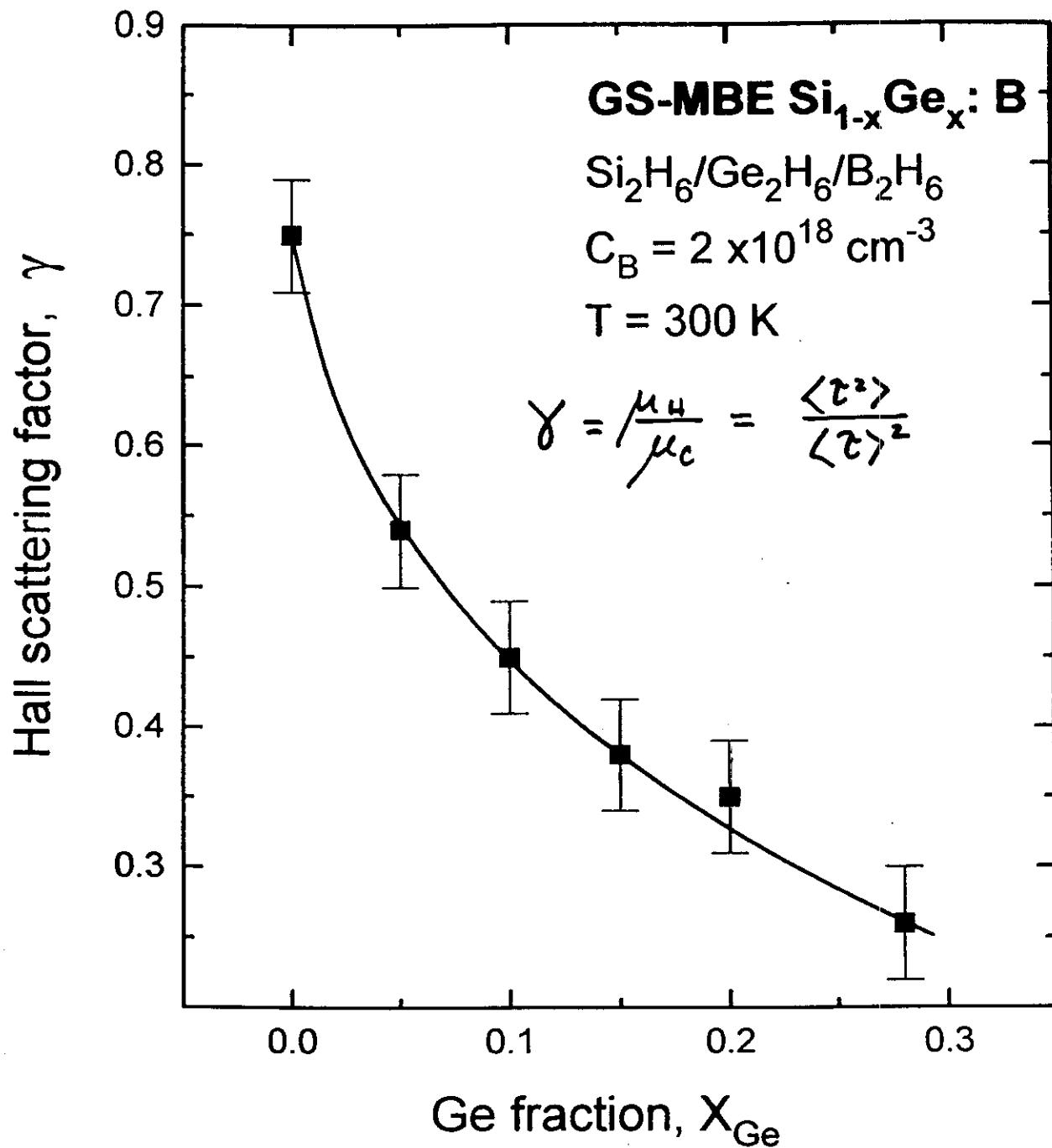


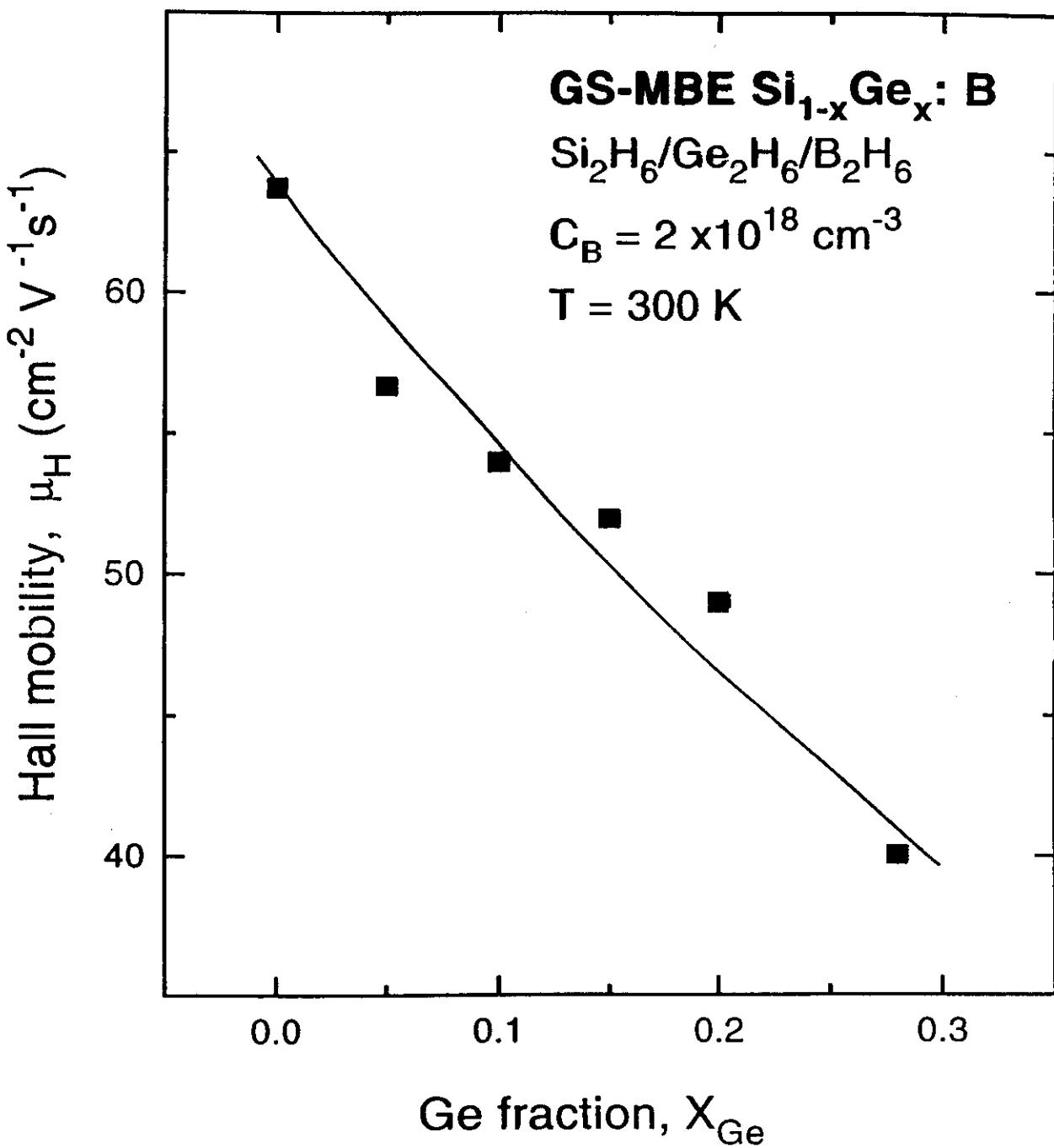


$$a_{\parallel, \text{Si:Ge}} = 0.54312 \pm 0.00002 \text{ nm}$$

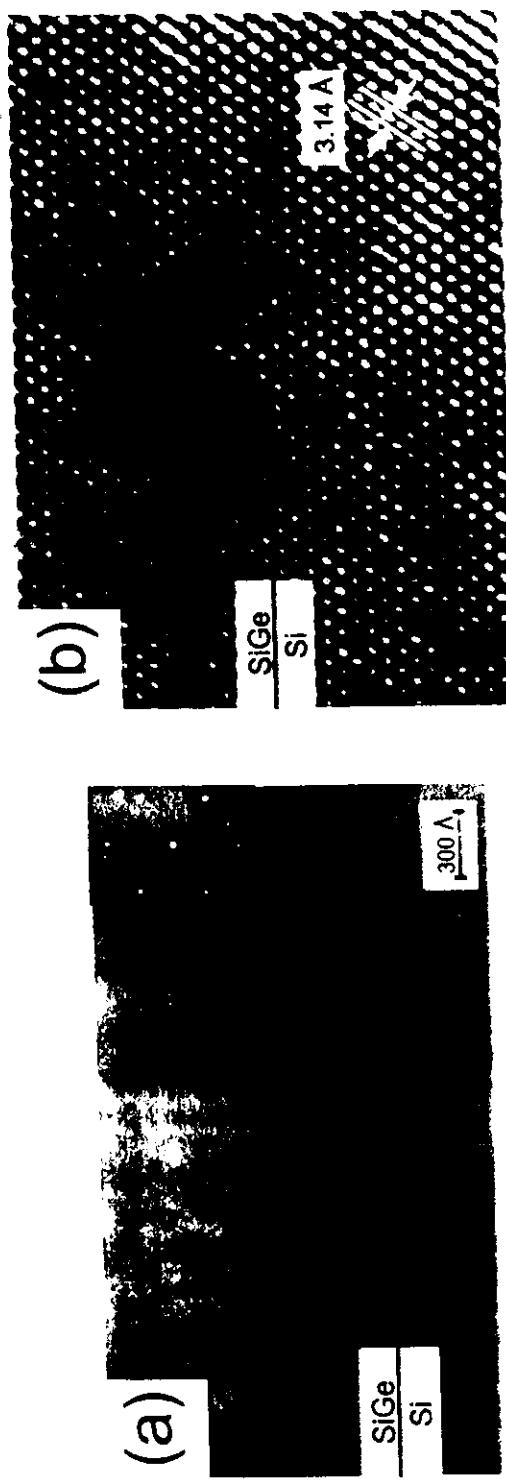
$$\therefore \frac{\Delta a}{a_{\text{Si}}} \leq 4 \times 10^{-5} !$$

$$\Gamma_{\parallel} = 24.5 \text{ arc-s}, \Gamma_{\omega} = 24.9 \text{ arc-s}, \Gamma_{\omega-2\theta} = 24.7 \text{ arc-s}$$





GS-MBE $Si_{1-x}Ge_x/Si(001)$



$X = 0.18, T_s = 500^{\circ}\text{C}$