



INTERNATIONAL ATOMIC ENERGY AGENCY
UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION



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SMR 115 - 53

(21 January - 22 March 1985)

SPIN-POLARIZED HYDROGEN

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ICTP, Trieste, March '85
lecture #1

SPIN-POLARIZED HYDROGEN

(by D. Kleppner - M.I.T.)

- or, a study in laser-free atomic physics
(and, hopefully, molecule-free atomic physics)

"To understand hydrogen
is to understand everything"
attributed to Victor F. Weisskopf
by Gerhardt Herzberg

"I never said it -
but I wish I had!"
- V. F. Weisskopf

A "Thumbnail" History of Hydrogen

Paracelsus ~ 1510 probably saw H_2 .

Cavendish ~ 1774 isolated & identified H_2 .

Lavoisier ~ 1776 $H_2 + O \rightarrow H_2O$: gave name "hydrogen".

Fraunhofer ~ 1818 solar absorption - G line = $H\alpha$.

Angstrom ~ 1860's lab. measurement of H spectrum.

Huggins ~ 1870's - astron. " " " "

Balmer ~ 1883 $\frac{1}{\lambda} = R \left(\frac{1}{2^2} - \frac{1}{n^2} \right)$ $n=3,4,5$

Bohr ~ 1912 H "cherry" \Rightarrow birth of quant. mech.

Einstein ~ 1926 relativistic e theory

Rabi ~ 1946 hyperfine structure - problems?

Lamb ~ 1948 Lamb shift \Rightarrow Q.E.D.

Purcell & Field, van Hulst ~ 1950 21cm line in space.

Metyuen & Hoagland ~ 1965 recomb. lines $n=101 \rightarrow 100$

1970's & 80's: Lamb shift measured - proton structure limit; 1s-2s 2-photon spectroscopy.

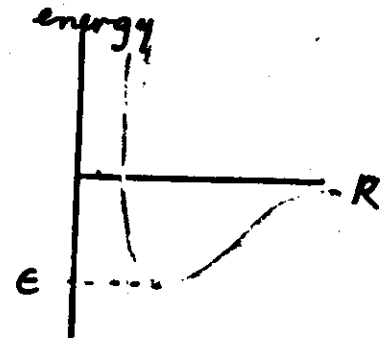
saturation spectroscopy of Balmer series,

QED of 1-photon "hydrogen", etc....

1980 - Spin-polarized H created by Silver & Waltraud, et al M.I.T. **WHY?**

Solid, Liquid or Gas?

The state of matter is determined by a competition between energy order due to attractive potential ϵ , and disorder due to kinetic energy $E \sim kT$.



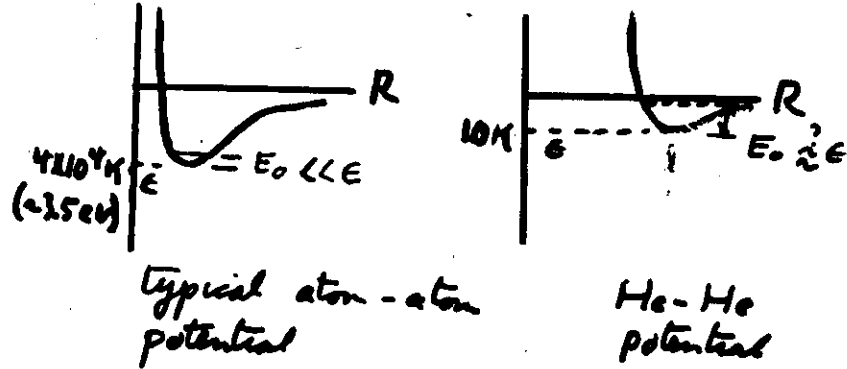
$E \gg \epsilon$ kinetic energy wins: gas

$E \ll \epsilon$ potential energy wins: solid

$E \approx \epsilon$ intermediate state: liquid

$E \sim kT$ As $T \rightarrow 0$, all matter becomes solid, with one (or two) exceptions.

Exception - He remains liquid as $T \rightarrow 0$.



In He, the 0-point energy E_0 is so large that as $T \rightarrow 0$, $E \rightarrow E_0 \sim E$. System forms a "loose" liquid, but not a solid.

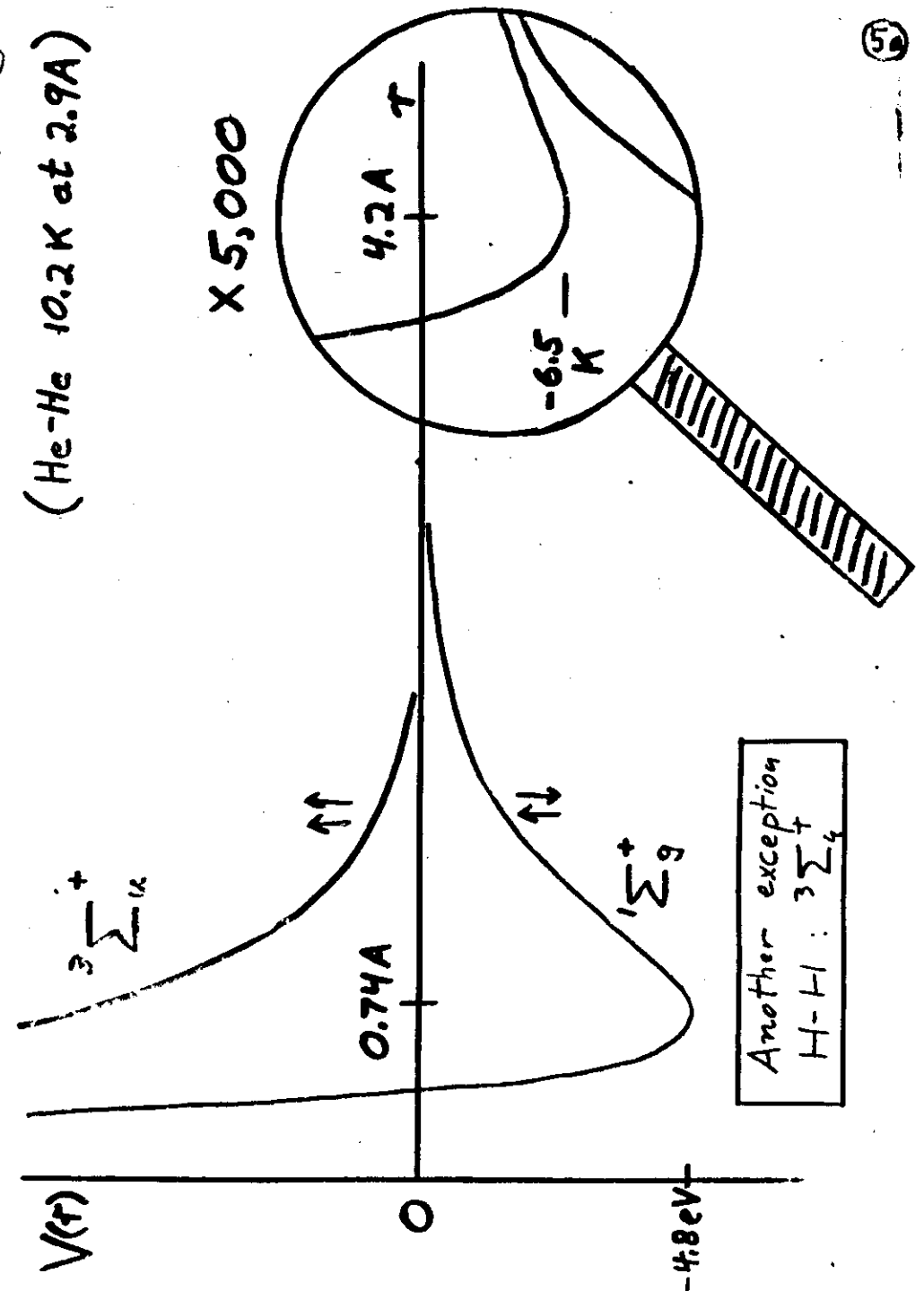


E_0 neglected
no K.E.
 $S = S_{max}$



E_0 included
much K.E.
 $S(\text{actual}) \sim \frac{1}{4} S_{max}$

(He-He 10.2 K at 2.9 Å)



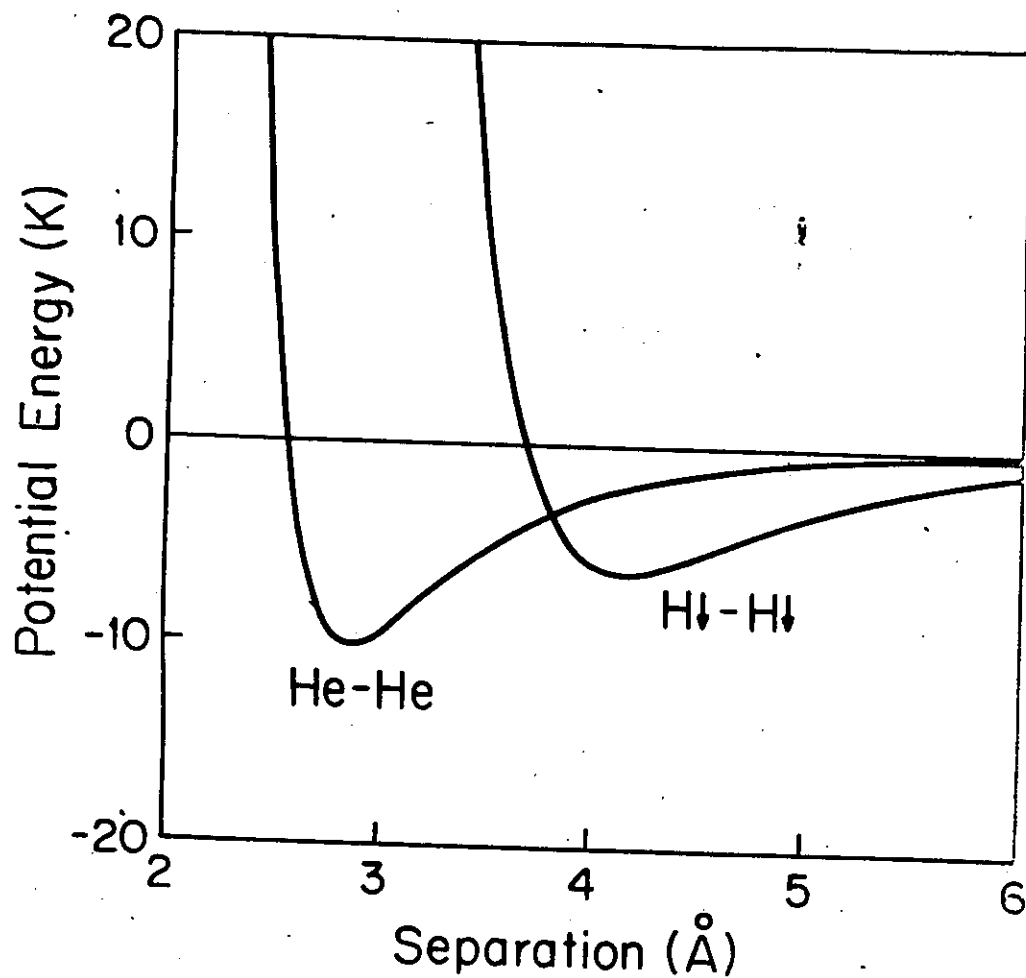


Fig. 5.1

Effect of zero-point energy on equation of state

- can be predicted with a universal scaling law.

Diagram illustrating the potential well $V(r)$ and the zero-point energy level $-E_0 = -\frac{1}{2} \hbar \omega_0$. The potential is approximated as $V = -E + \frac{1}{2} U''|_{r_0} (r-r_0)^2$. The zero-point energy level is shown at $-E_0 = -\frac{1}{2} \hbar \omega_0$. The equation $U''|_{r_0} = \text{const} \times \frac{E}{r_0^2}$ is shown, with a note that it depends on the form of U .

$$\omega_0 = \sqrt{\frac{U''}{M}} = \sqrt{\frac{\text{const} \cdot E}{M r_0^2}}$$

$$\frac{\text{O-point energy}}{\text{Binding energy}} = \frac{\frac{1}{2} \hbar \omega_0}{E} = \text{const} \sqrt{\frac{\hbar^2}{M E r_0^2}}$$

$$\eta = \frac{\hbar^2}{M E r_0^2} = \text{"quantum parameter"}$$

Law of corresponding states

reduced variables: $E^* = \frac{\text{energy}}{E}$;

$T^* = \frac{kT}{E}$; $r^* = \frac{\text{distance}}{r_0}$; $M^* = \frac{\text{mass}}{M}$.

All systems having same form for potential have same thermodynamic behavior in terms of reduced variables. Behavior depends only on η .

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Why is spin-polarized hydrogen
(H \downarrow) interesting?

- because it is predicted to remain a gas as $T \rightarrow 0$. This prediction is very firm, based on the law of corresponding states and other studies. So far, it has been confirmed to 60 mK.
- because it offers the opportunity to study matter in a new quantum regime, the regime where the translational wave packets overlap ($\lambda_{\text{deBroglie}} > \text{distance between atoms}$)
- because it has created a new technology for producing H at high density and low temperature with many potential applications to spectroscopy, atomic physics, particle & nuclear physics, etc.

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Plan of lectures

I. Introduction -

H \downarrow and Bose condensation

- I } strategies for making H \downarrow
- II } what we have discovered
- how we fooled Mother Nature
- how Mother Nature fooled us
- prospects

IV applications - including spectroscopy

Reference:

Lectures on Spin-Polarized Hydrogen
T.J. Greytak & D. Kleppner

in New Trends in Atomic Physics, Vol II

G. Grynberg & A. Stora, ed.

North Holland, 1984

Copies will be distributed in a week or two

H₂ and the Bose-Einstein Transition

A real gas, at low density, is accurately described by classical statistical mechanics. For instance, momentum described by Maxwell-Boltzmann statistics.

classical view



low density: $r_a \ll a$

quantum view



low density:
 $\Lambda_d \ll a$

For most atoms, $\Lambda_d = \frac{h}{Mv_{thermal}} \ll r_a$.

So - if gas is at low density classically, it is also at low density quantum-mechanically.

Quant. mech. effects important when

$$a \sim \Lambda = \frac{h}{p_{thermal}} = \frac{h}{\sqrt{2\pi kTM}}$$

$$\text{density } n = \frac{1}{a^3} \text{ atoms cm}^{-3}$$

$$n_0 = (\text{const}) \times \left(\frac{2\pi kTM}{h^2} \right)^{3/2}$$

For Bose particles, phase transition occurs when

$$n_0 = (2.612...) \left(\frac{2\pi kTM}{h^2} \right)^{3/2}$$

A very quick summary of Bose-Einstein transition.

Note: H₂ obeys Bose statistics. To interchange 2 H atoms, first interchange electrons, then protons. Each interchange reverses sign of wavefunction. $(-1)/(-1) = +1$. This is true so long as the atoms are not electronically excited. (In such a case, one would have 4 fermions, instead of 2 bosons.) The situation is analogous to the deuteron, which is composed of 2 fermions - a proton and a neutron - but behaves like a simple boson at low energy.

(summary of Bose transition - continuous).

Mean occupation number, \bar{N}_p

- non interacting particles
- momentum \vec{p}
- energy ϵ_p .

$$\bar{N}_p = \begin{cases} \frac{1}{\exp \beta(\epsilon_p - \mu) - 1} & \text{Bose particles } (\beta = \frac{1}{kT}) \\ \frac{1}{\exp \beta(\epsilon_p - \mu) + 1} & \text{Fermi particles} \\ \frac{1}{\exp \beta(\epsilon_p - \mu)} & \text{Boltzmann law - classical} \end{cases}$$

$$\sum \bar{N}_p = N \quad (\text{total \# particles})$$

of states: box quantization $V = L^3$

$$p_x = \frac{2\pi\hbar}{L} j_x; \quad p_y = \frac{2\pi\hbar}{L} j_y; \quad p_z = \frac{2\pi\hbar}{L} j_z$$

j_x, j_y, j_z integers

$$\sum_{j_x, j_y, j_z} \rightarrow \left(\frac{L}{2\pi\hbar}\right)^3 \iiint dp_x dp_y dp_z$$

Change from integral over momenta to integral over energy.

$$\epsilon_p = \frac{p^2}{2M} \quad \iiint dp_x dp_y dp_z \rightarrow 4\pi \int p^2 dp$$

$$\rightarrow 4\pi (2M)^{3/2} \int \sqrt{\epsilon} d\epsilon.$$

summarizing:

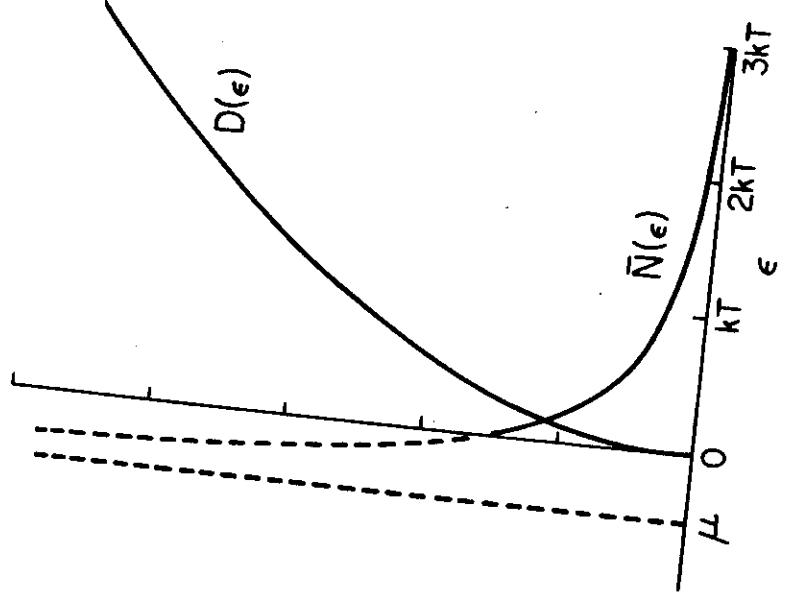
$$\sum_{j_x, j_y, j_z} \rightarrow \frac{V}{(2\pi)^3} \left(\frac{2M}{\hbar^2}\right)^{3/2} \int \sqrt{\epsilon} d\epsilon = \int D(\epsilon) d\epsilon.$$

$$D(\epsilon) = \frac{V}{(2\pi)^3} \left(\frac{2M}{\hbar^2}\right)^{3/2} \sqrt{\epsilon} = \text{density of states in energy.}$$

Require:

$$\int \bar{N}(\epsilon) D(\epsilon) d\epsilon = N.$$

$$\bar{N}(\epsilon) = \frac{1}{\exp \beta(\epsilon - \mu) - 1} \quad \mu \text{ is adjustable parameter (chemical potential).}$$



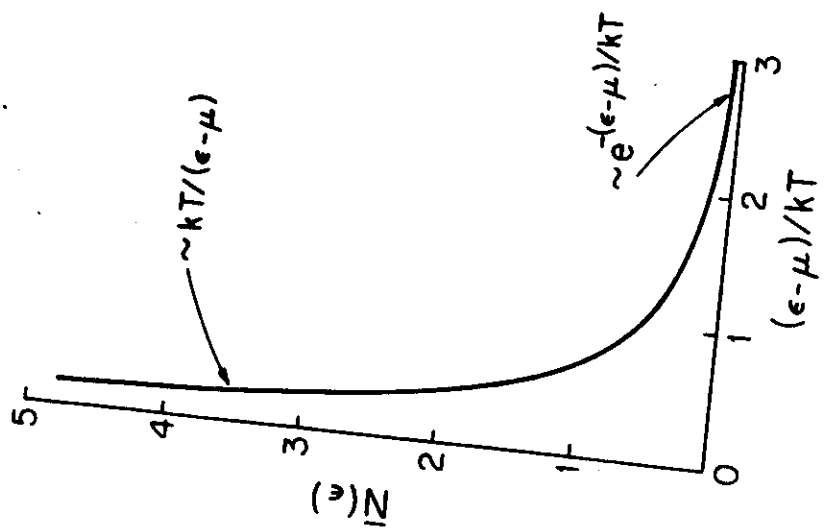
As T decreases, μ increases toward 0 so that $N = \int \bar{N}(\epsilon) D(\epsilon) d\epsilon$ is satisfied.
Maximum temperature: $\mu = 0$, $T = T_0$

$$N = V \left(\frac{M k T_0}{2 \pi \hbar^2} \right) \frac{2}{\sqrt{\pi}} \int_0^\infty \frac{\sqrt{x} dx}{e^x - 1}$$

$$= V \frac{1}{\Lambda(T_0)^3} \times 2.612 \dots$$

$$n = \left(\frac{M k T_0}{2 \pi \hbar^2} \right)^{3/2} \times 2.612 \dots$$

Fig. 1.1



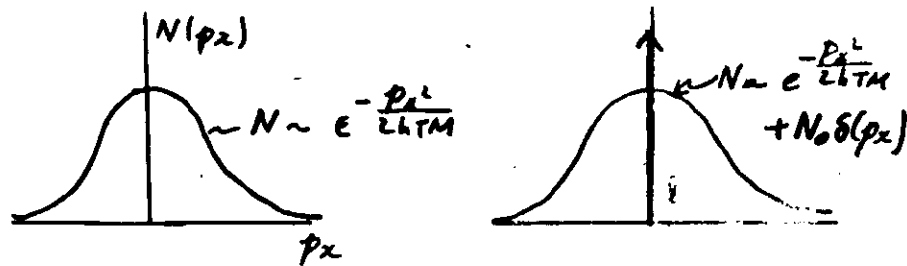
What happens if $T < T_0$?
Finite number of particles drop into the 0-energy state.
(This state was discarded in converting sum over states into integral over energy.)
Can show that the number of particles in 0-energy state is

$$\bar{N}_0 = N \left(1 - T/T_0 \right)^{3/2}$$

$T < T_0$
 $T > T_0$

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Evidence for Bose-Einstein condensation



Most systems: solid before T_0 is reached.

Liquid ^4He : $n = 2.20 \times 10^{22} \text{ atoms cm}^{-3}$
(liquid density)

$$T_0 = 3.15 \text{ K}$$

Close to superfluid temp. 2.17 K

Is superfluid ^4He a

Bose condensate?

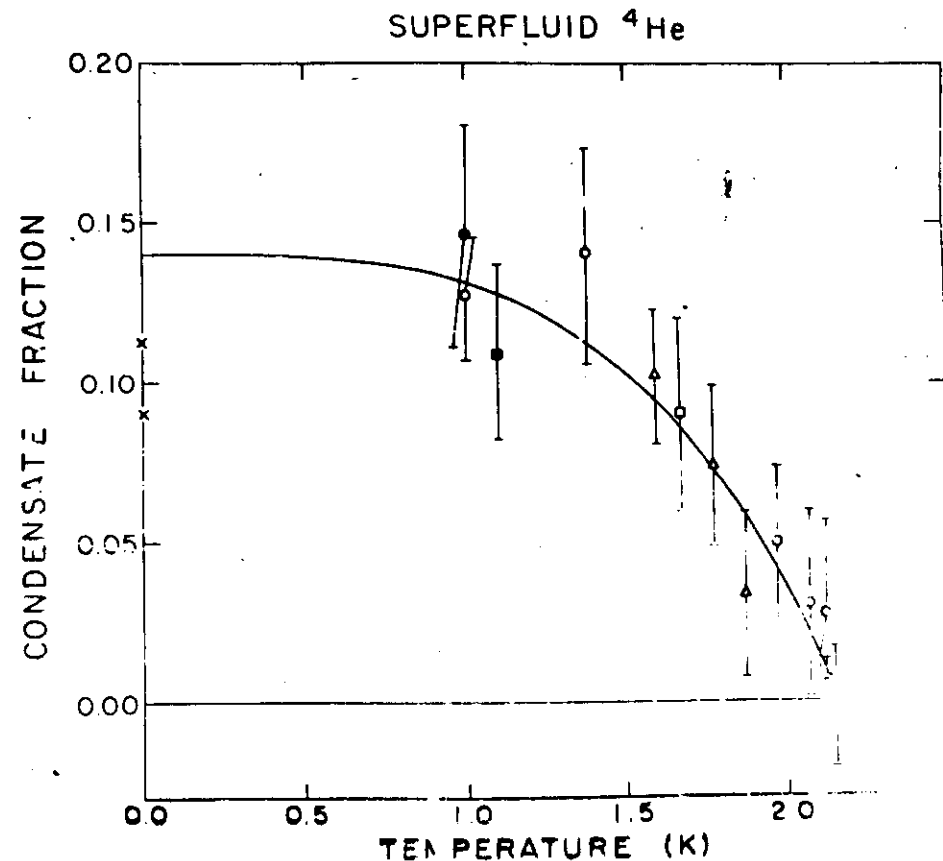
$$T_0 = \frac{2\pi\hbar^2}{3m} \left(\frac{n}{2.612} \right)^{2/3}$$

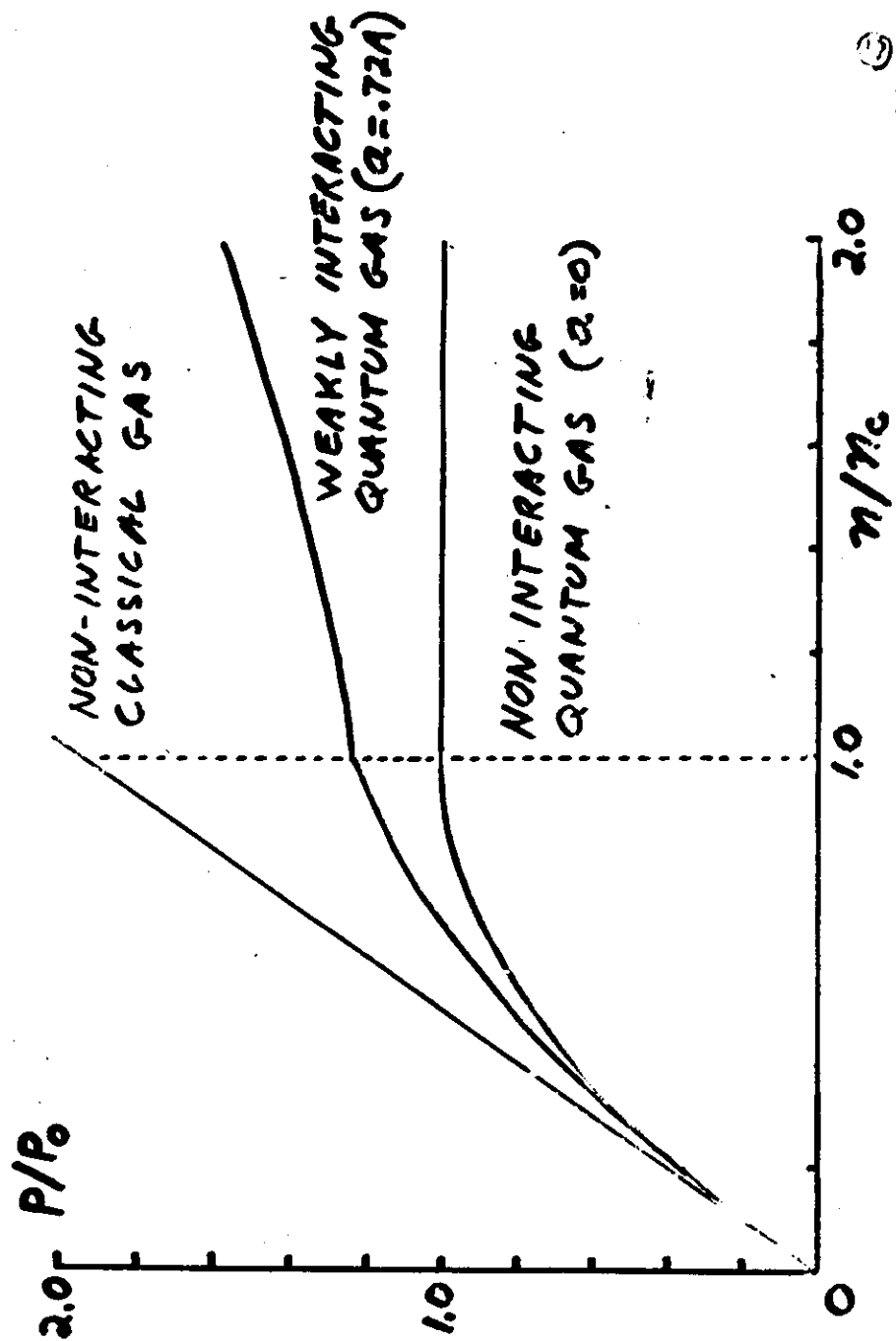
require minimum M :

Hydrogen: for $T_0 = 0.3 \text{ K}$, $n = 6 \times 10^{21} \text{ cm}^{-3}$

Can B-E condensation be achieved?

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Quest to observe B-E transition in H

Required: atomic hydrogen

- at temperature $< 1\text{K}$
(previous lowest temperature, 78K)
- at density $> 10^{19}\text{cm}^{-3}$
(previous maximum density for H (without H_2) $\sim 10^{12}\text{cm}^{-3}$)
- highly spin polarized
(a few percent of wrong spin state would rapidly destroy system)
- stabilized against any sort of spin relaxation

