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Infrared laser-induced desorption after resonant excitation of small molecules condensed on insulator surfaces using the Free Electron Laser "FELIX"

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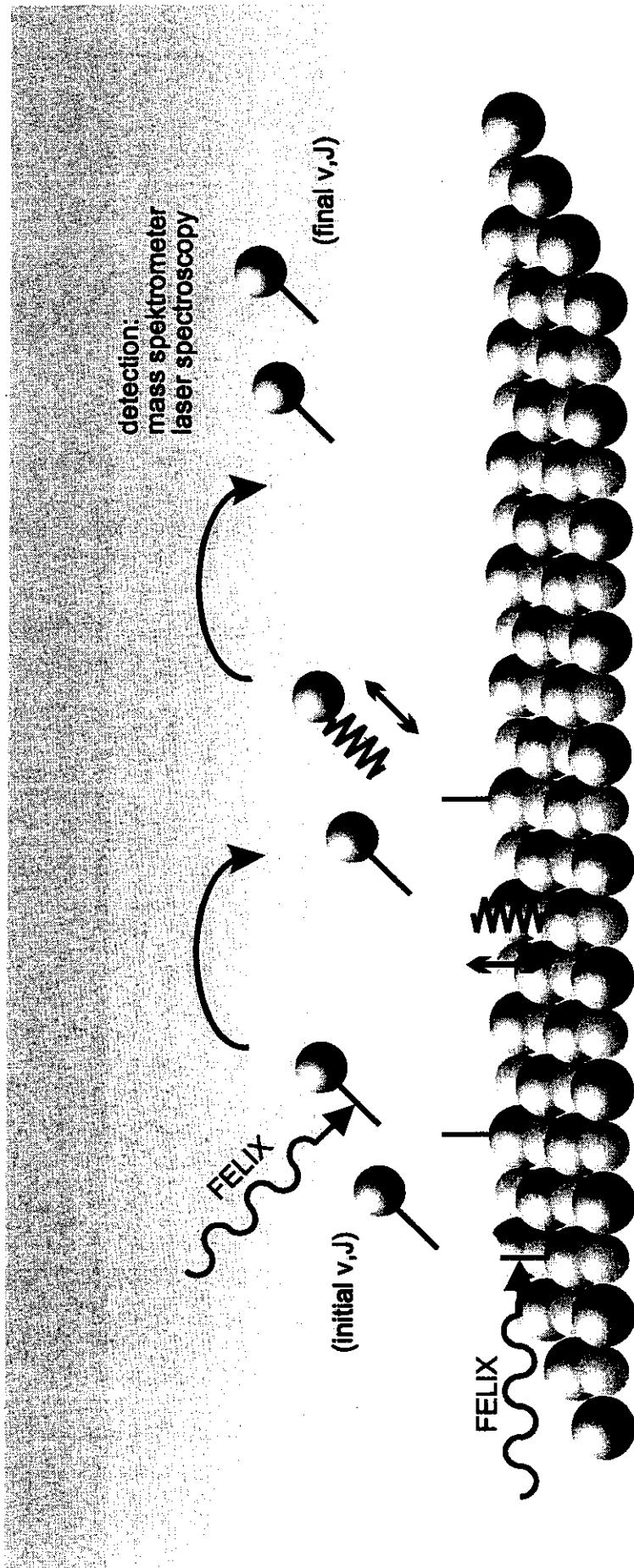
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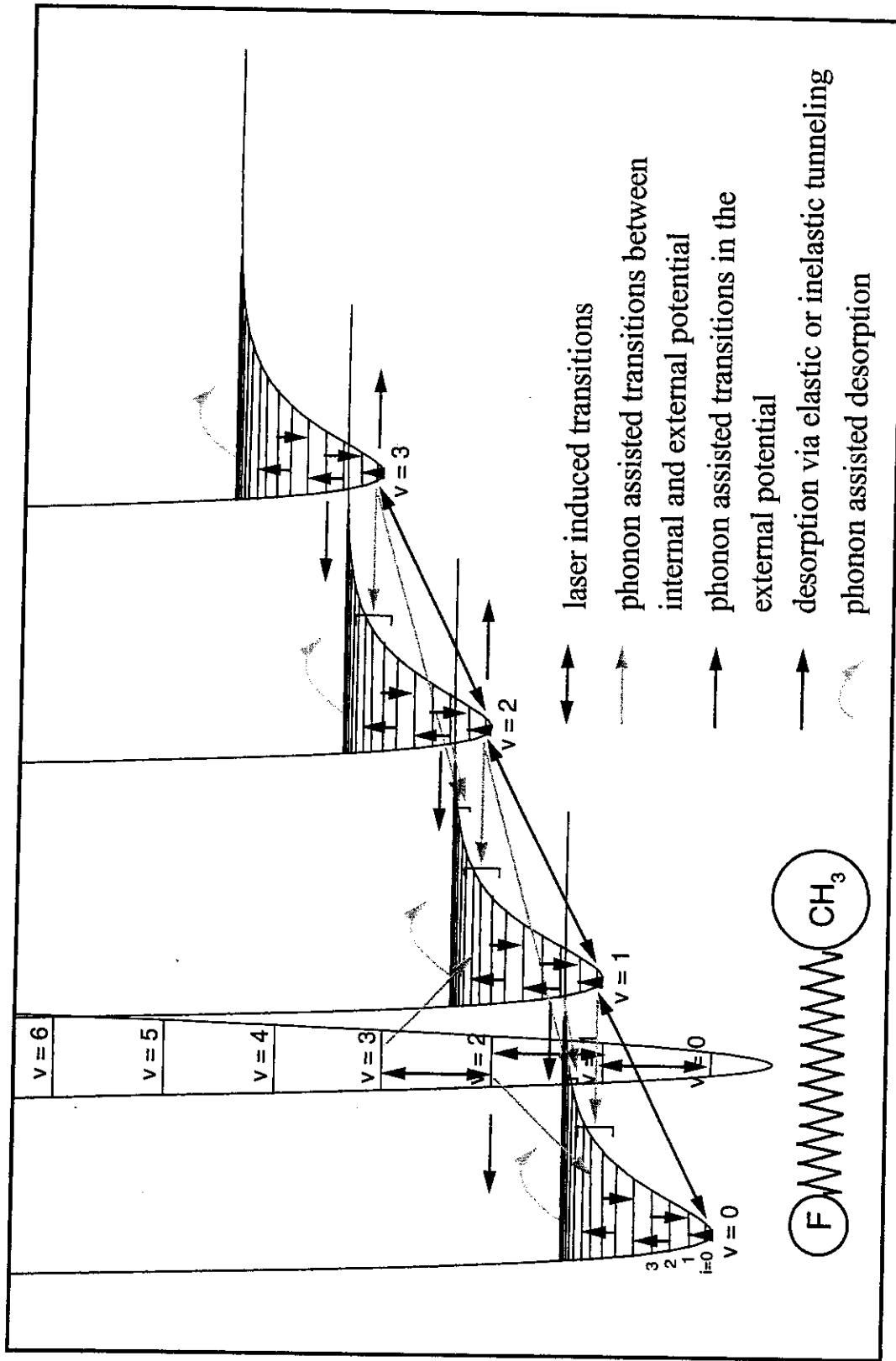
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- Motivation
 - The Free Electron Laser ‘FELIX’
 - Experimental setup
 - Results: CD₃F, methane and N₂O
condensed on NaCl(100)
 - Conclusions
 - Outlook

Motivation

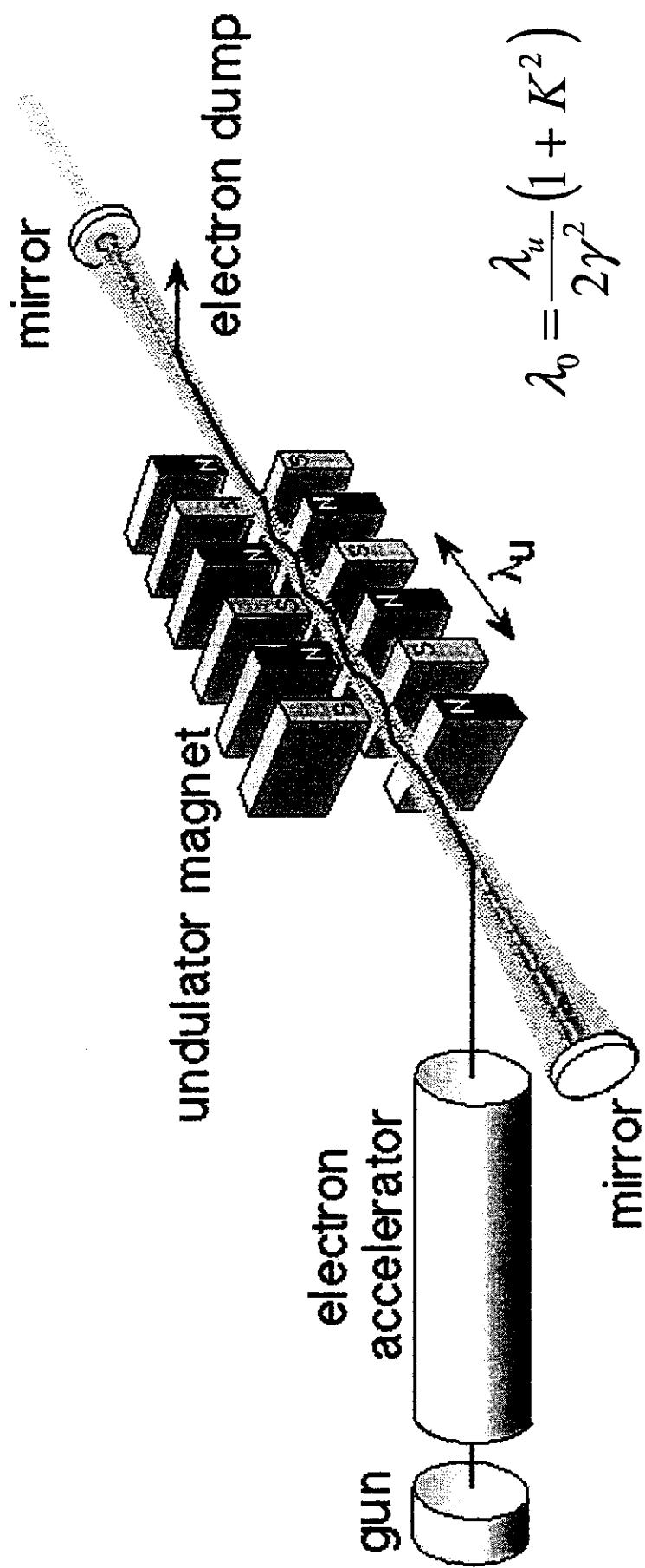


- energy flow in vibrationally excited adsorbates: desorption or relaxation ?
- isotope selective excitation \Rightarrow isotope selective desorption ?

Excitation and relaxation channels



Scheme of the Free Electron Laser



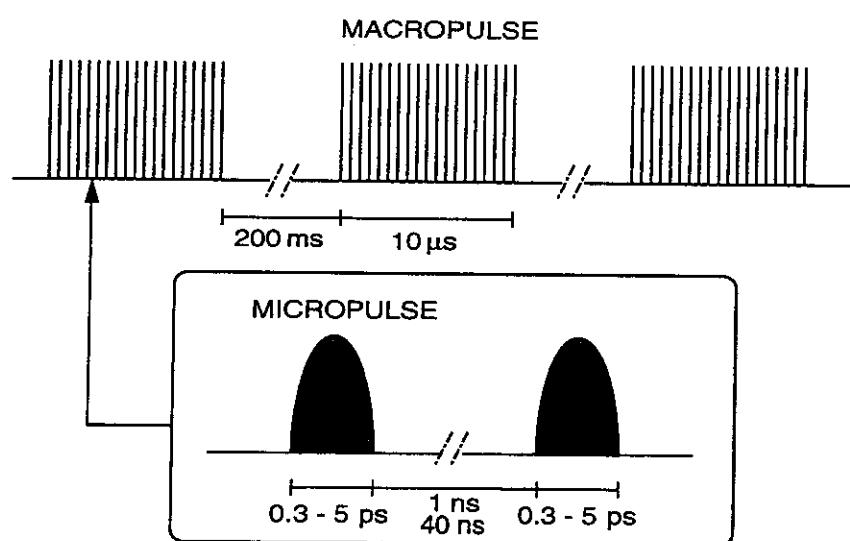
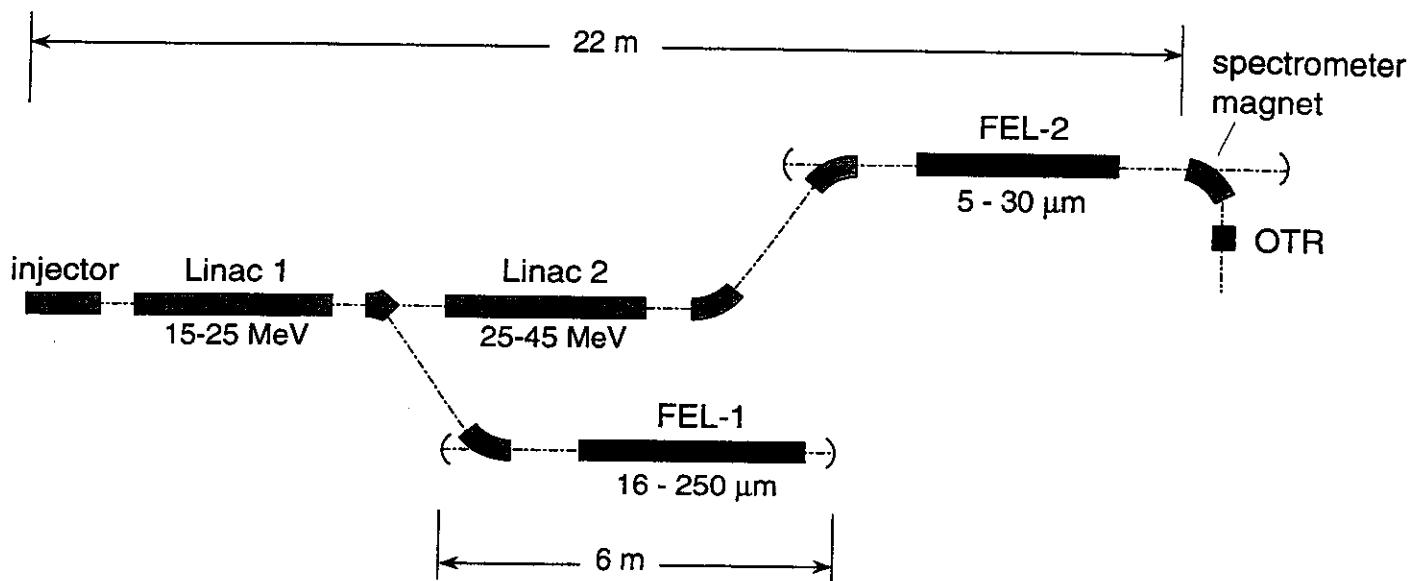
$$\lambda_0 = \frac{\lambda_u}{2\gamma^2} (1 + K^2)$$

λ_0 = wavelength of the radiation λ_u = periodicity of the undulator field

K = dimensionsless constant proportional to the amplitude of the magnetic field

γ = relativistic factor ($\gamma^2 \gg 1$)

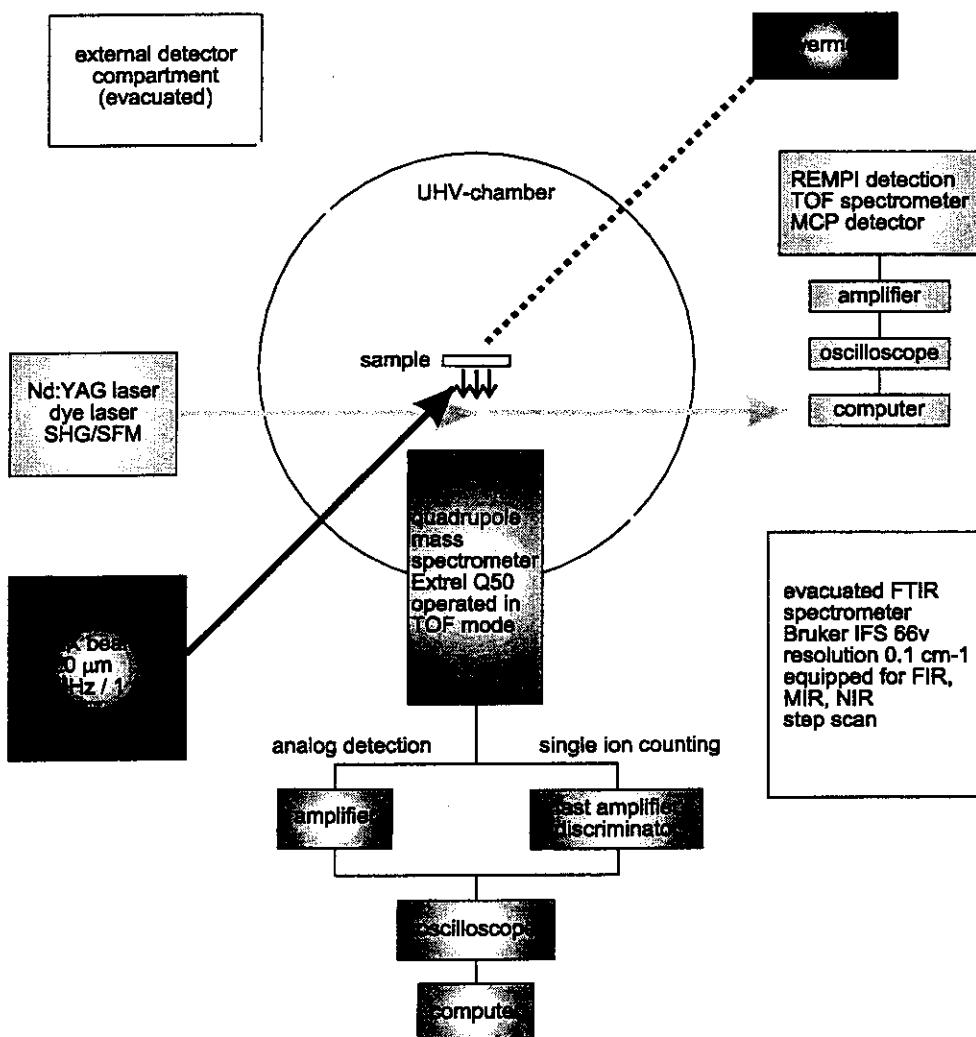
Free Electron Laser for Infrared eXperiments 'FELIX'



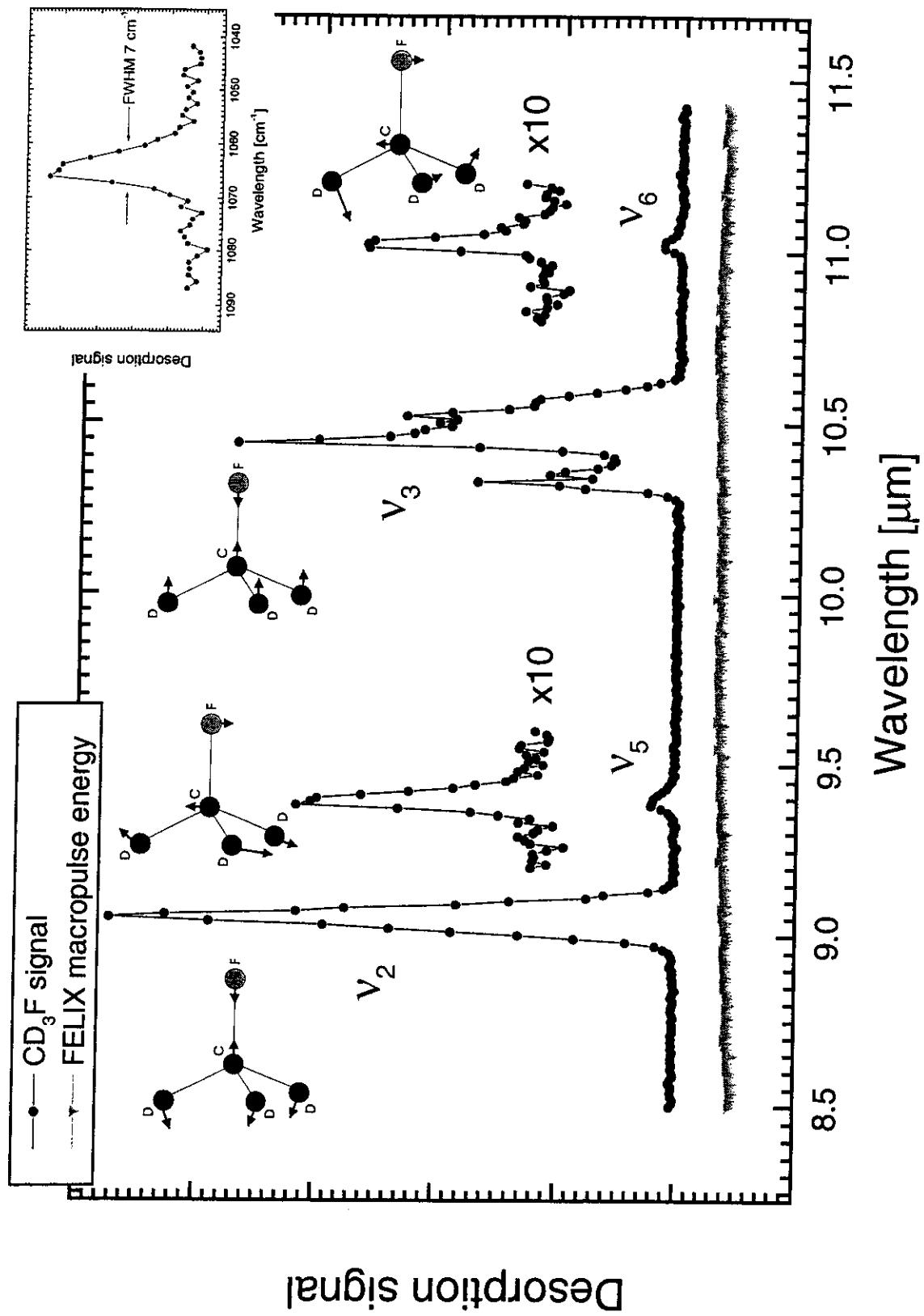
tuning range: $40 - 2000 \text{ cm}^{-1}$
macropulse energy: 20-100 mJ
bandwidth: transform limited

Experimental Setup

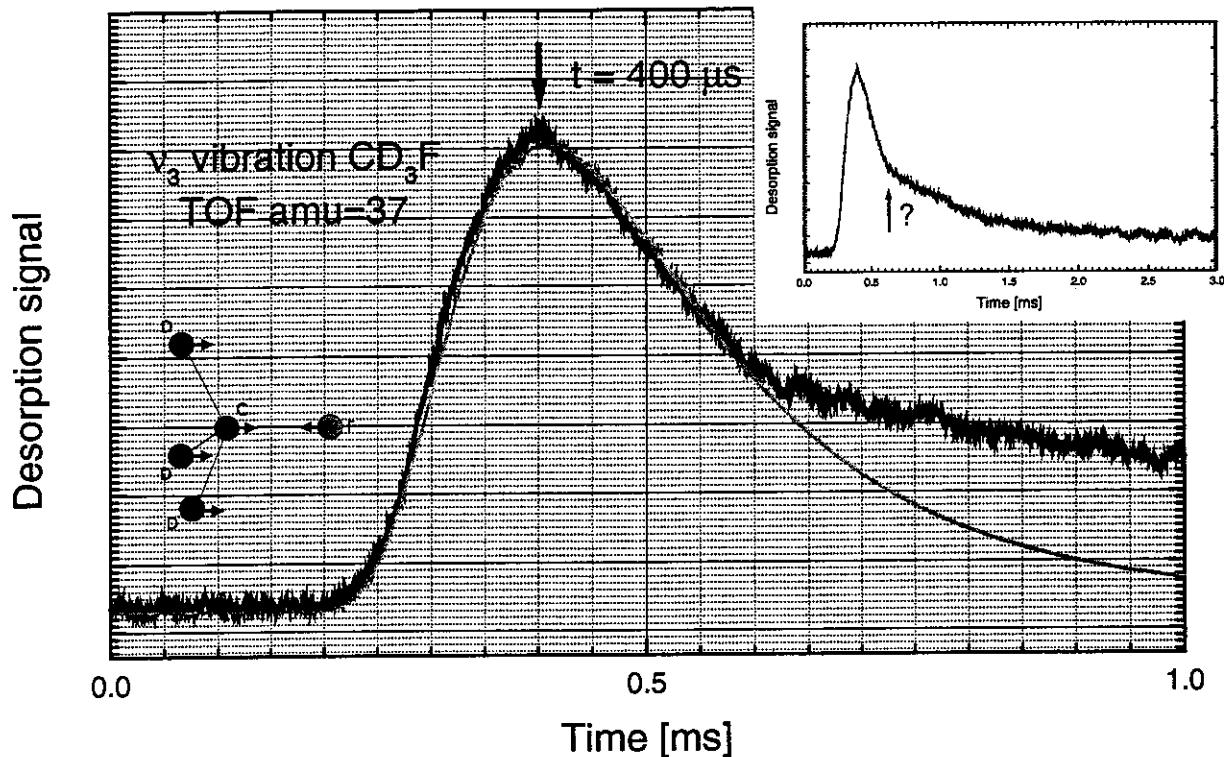
- UHV chamber
- substrate: NaCl(100) single crystal, cleaved under nitrogen atmosphere
- sample coolable down to 20 K, annealing of the sample up to 700 K
- detection of the desorbing species:
 - time resolved with a quadrupole mass spectrometer
(analog or single ion detection)
 - state resolved with a REMPI scheme and TOF detection
- detection and characterization of the adsorbates using FTIR spectroscopy
- condensation of the multilayers at low temperatures, no significant desorption observable



Laser-induced desorption of CD_3F condensed on $\text{NaCl}(100)$



TOF spectra of multilayers CD₃F/NaCl(100)



- estimation of the translational temperature of the desorbing molecules via a Maxwell-Boltzmann distribution

most probable velocity:*

$$v = \sqrt{\frac{2kT}{m}}$$

Translational temperature: 217.4 K

(* after correction for the drift time of $\sim 5 \mu\text{s}$)

- calculated desorption temperatures are well above the measured sample temperature of about 30 K

Determination of translational temperature

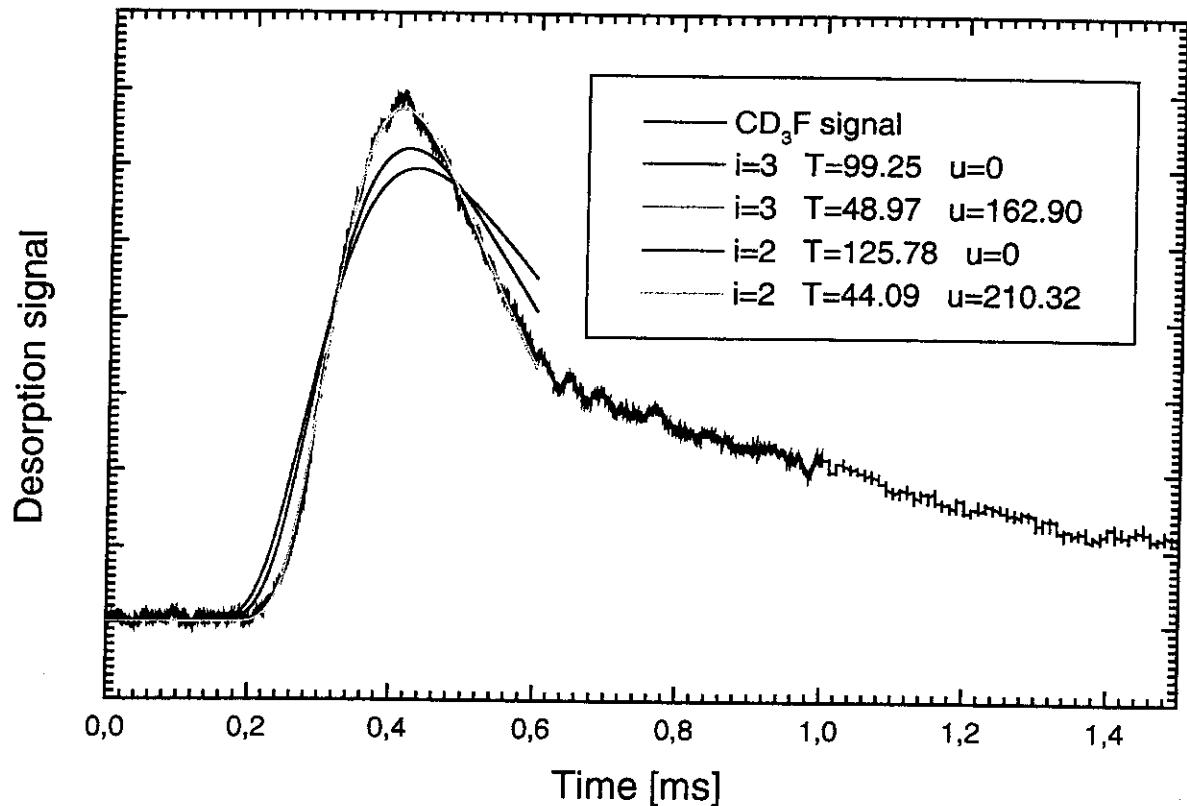
$$f(t) \propto \frac{d^i}{(t - t_{drift})^{i+1}} \cdot e^{-\frac{-m \cdot \left(\frac{d}{(t - t_{drift})} - u\right)^2}{2kT}} dt$$

d distance sample-ionizer

i=2 Maxwell-Boltzmann distribution

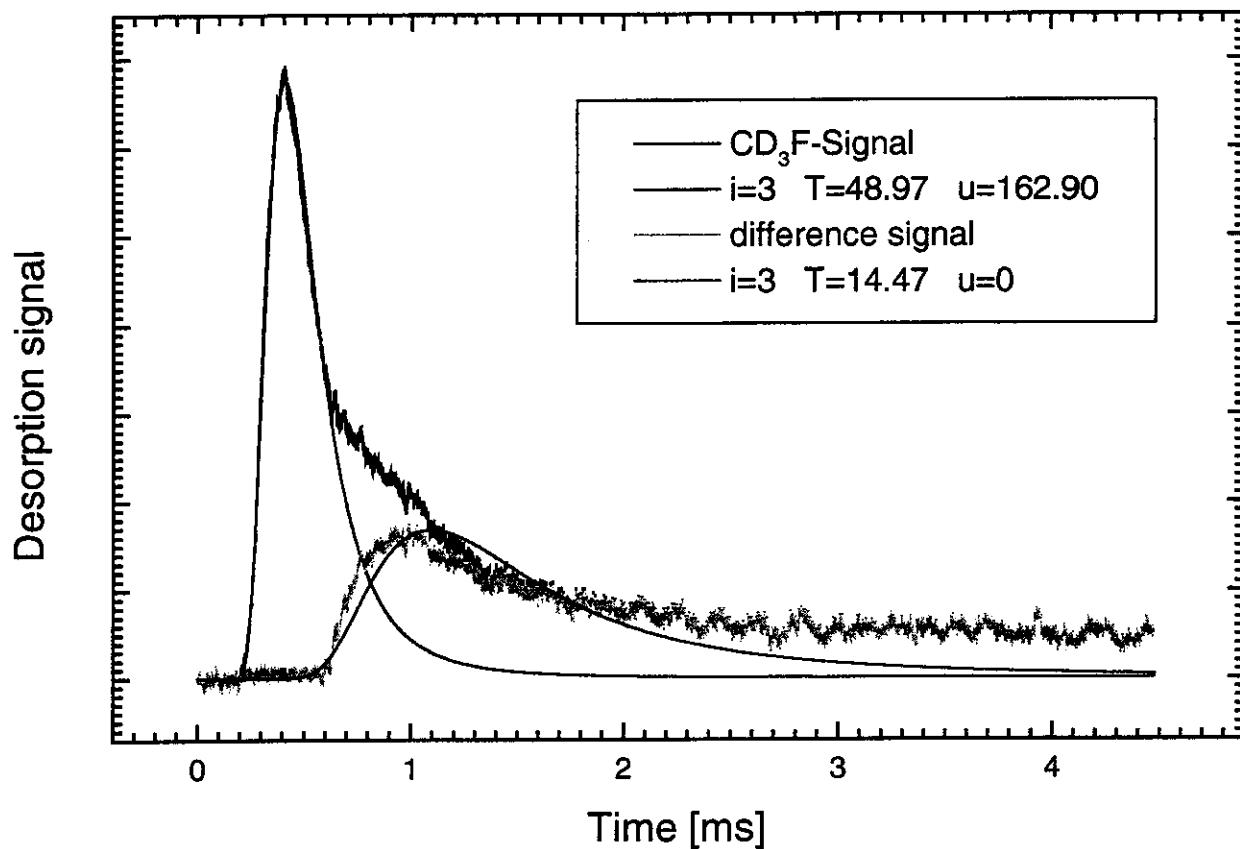
i=3 flux density distribution

t_{drift} flight time ionizer - SEV
u stream velocity



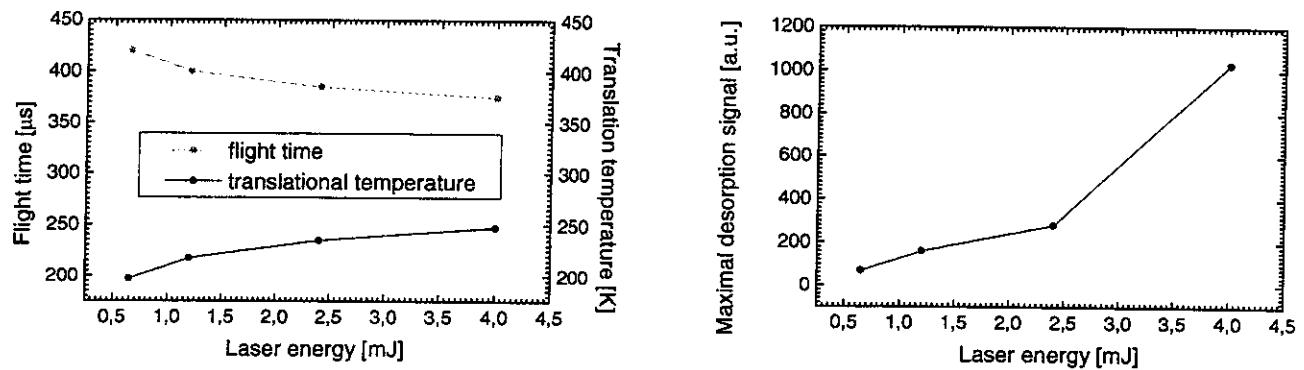
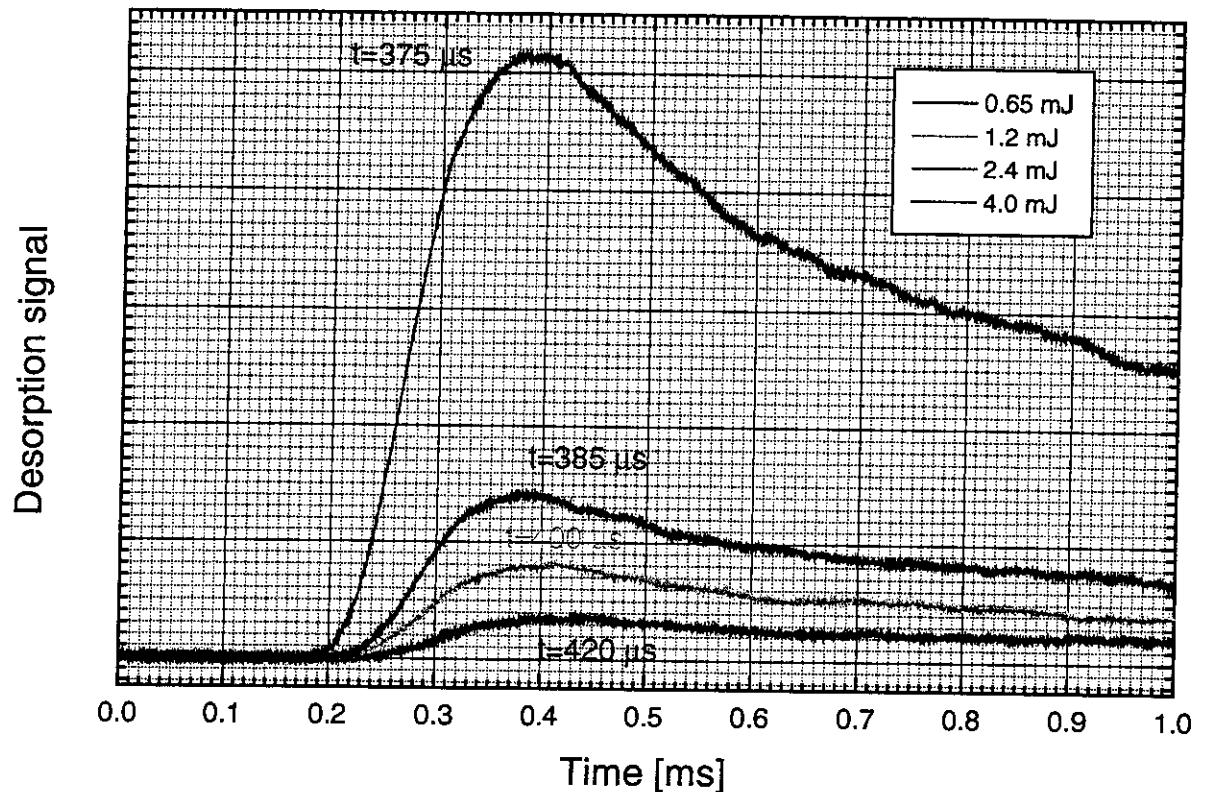
- reasonable fit to the experimental TOF spectra in the first part of the distribution
- better agreement for flux density distribution
- better agreement with overlapping stream velocity

Determination of translational temperature



- bimodal velocity distribution ?
- two desorption channels ?

Dependence of the flight time on the laser energy



- flight time in the TOF spectra depends weakly on the applied laser fluence
- indication for a resonant heating process ?

Desorption mechanism:
Dependence of the desorption yield Y on the laser fluence F

Desorption via a direct process

$$Y \propto F^n$$

$$\ln Y = n \cdot \ln F + \text{const.}$$

Desorption via resonant heating

$$Y = k' \cdot e^{-E_{\text{des}}/k \cdot T}$$

$$k' \propto v_{\text{des}} \cdot \theta$$

$$T = T_0 + \Delta T$$

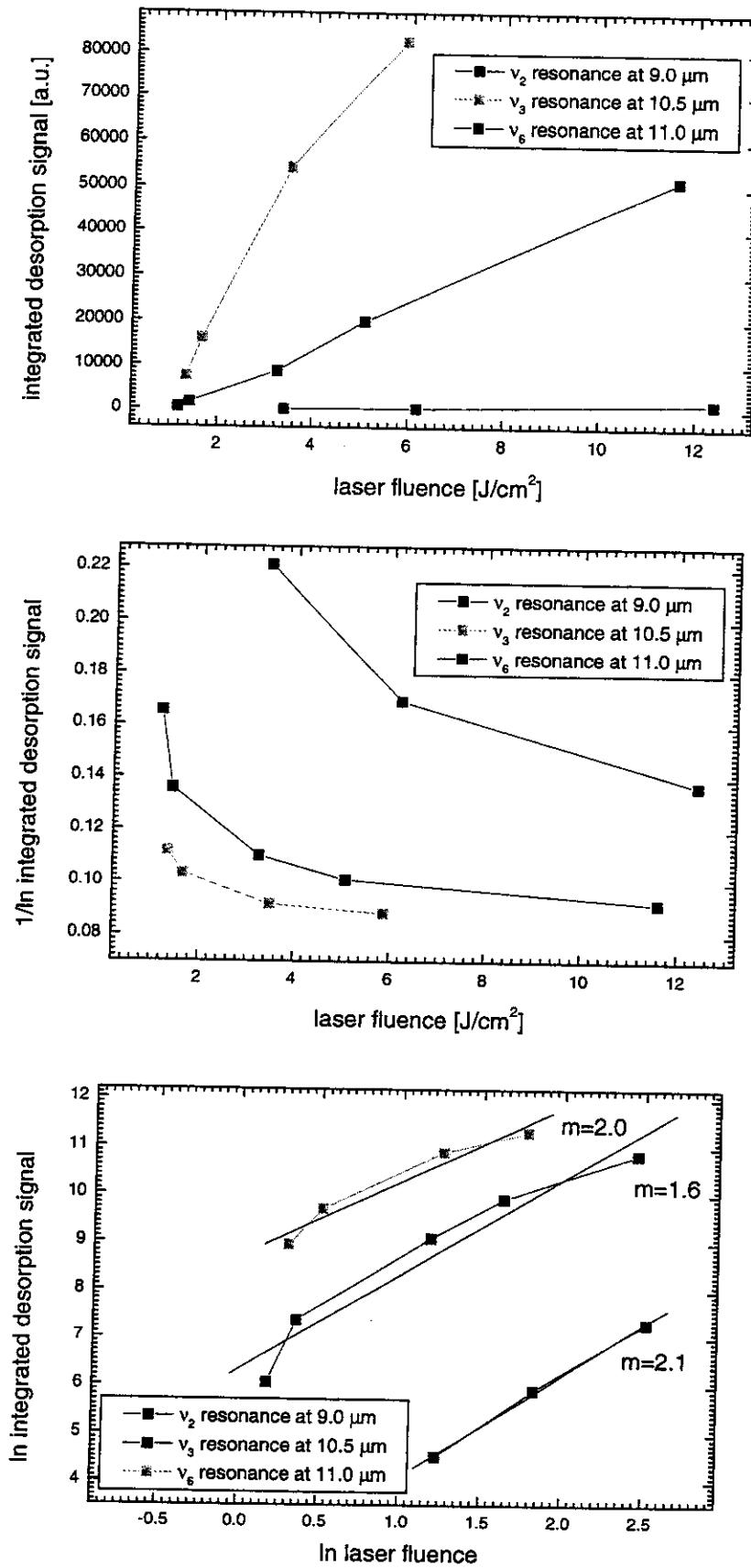
$$\Delta T = c \cdot F$$

$$Y = k' \cdot e^{-E_{\text{des}}/k \cdot (T_0 + c \cdot F)}$$

$$\frac{1}{\ln Y - \ln k'} = -\frac{k}{E_{\text{des}}} \cdot T_0 - \frac{k \cdot c}{E_{\text{des}}} \cdot F$$

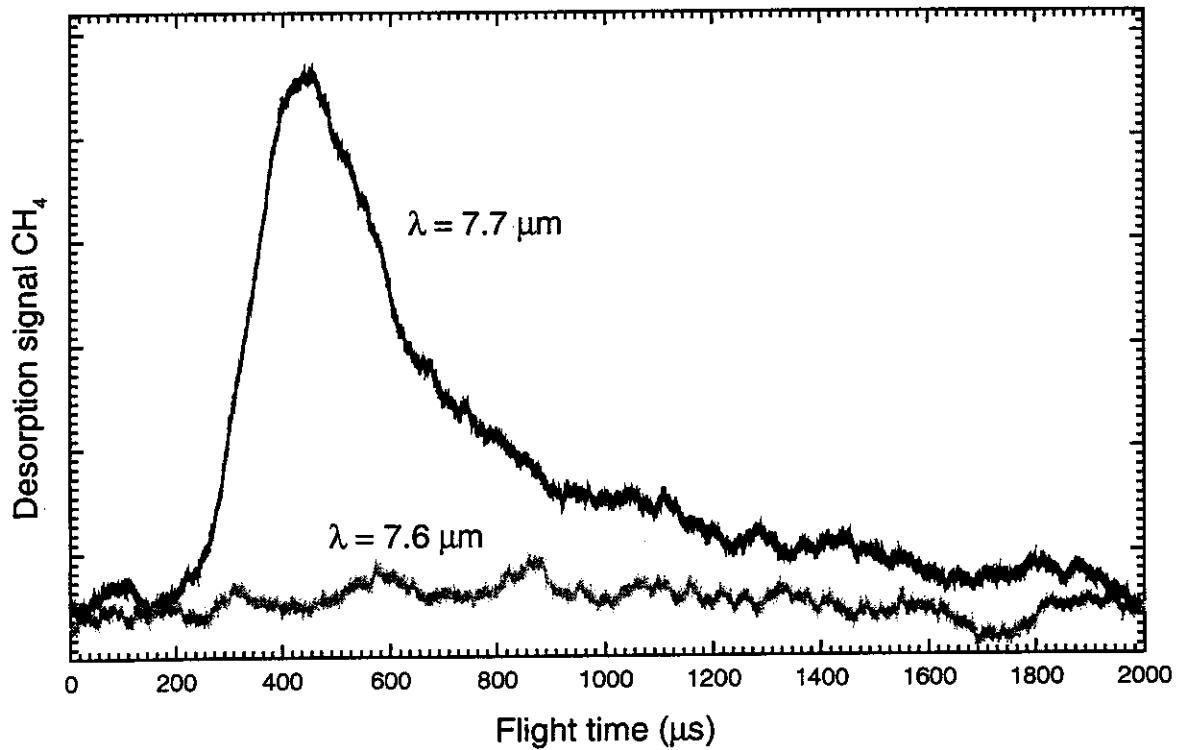
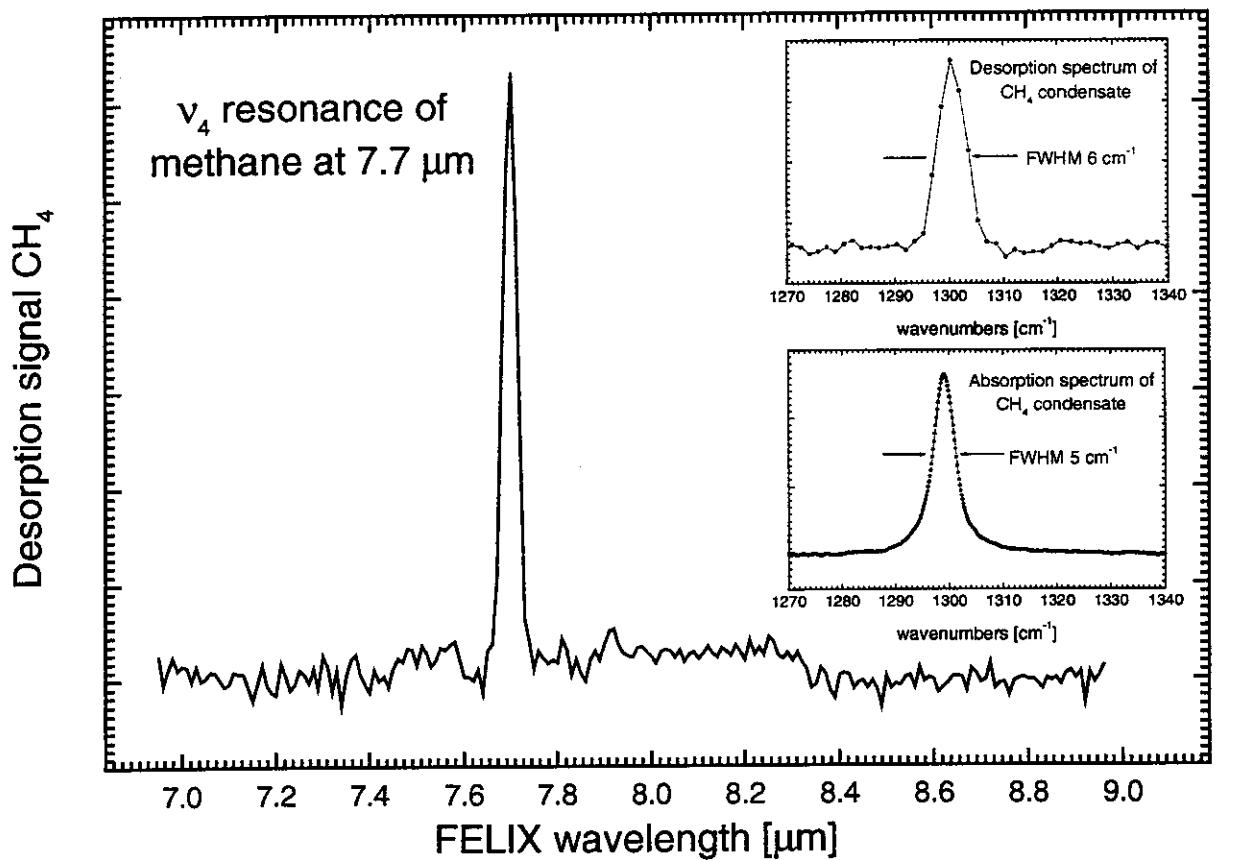
$$\frac{1}{\ln Y - \ln k'} = a + b \cdot F$$

Desorption signal as a function of the laser fluence: Thermal versus direct desorption mechanism

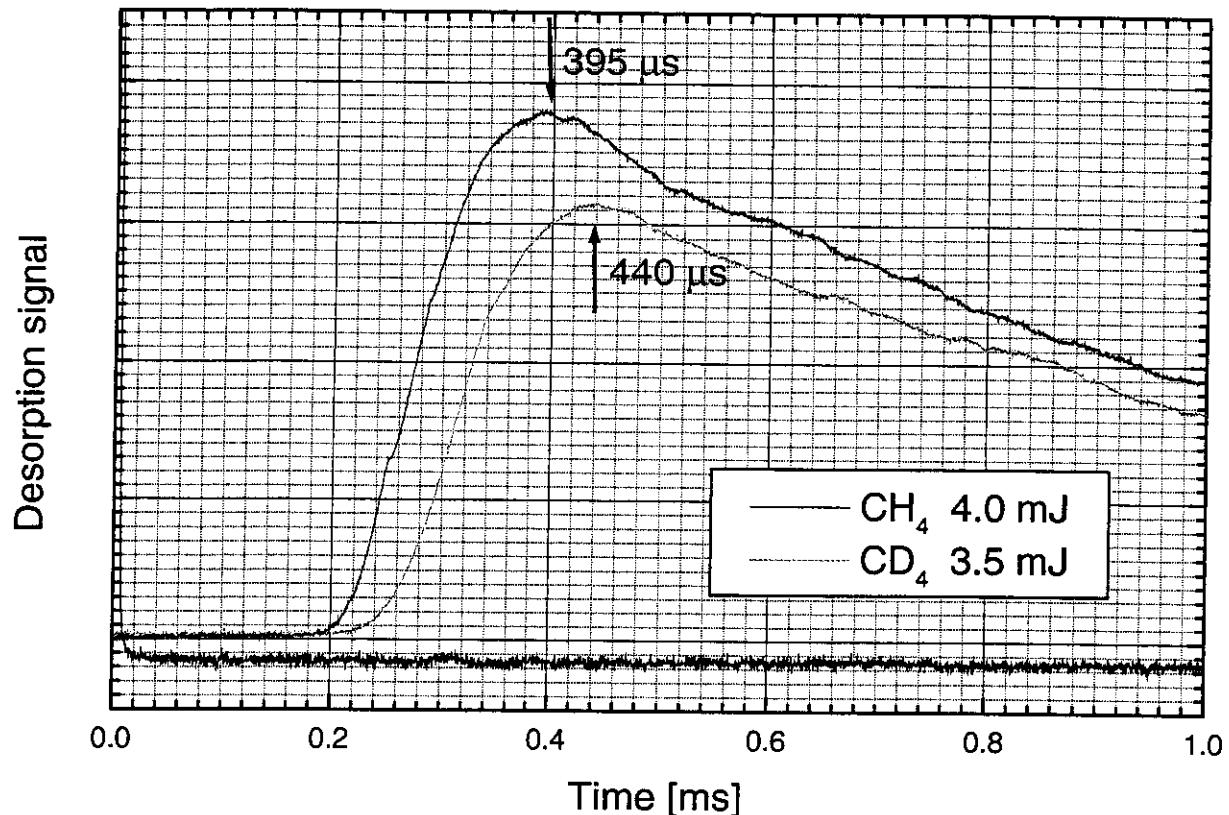


- Indication for a direct desorption mechanism ?

Laser-induced desorption of CH₄ condensed on NaCl(100)



TOF spectra for multilayers CH₄/NaCl(100)



- flight times of CH₄ and CD₄ scale by the square root of the mass
⇒ comparable kinetic energy and desorption mechanism?
- estimation of the translational temperature of the desorbing molecules via a Maxwell-Boltzmann distribution

most probable velocity:*

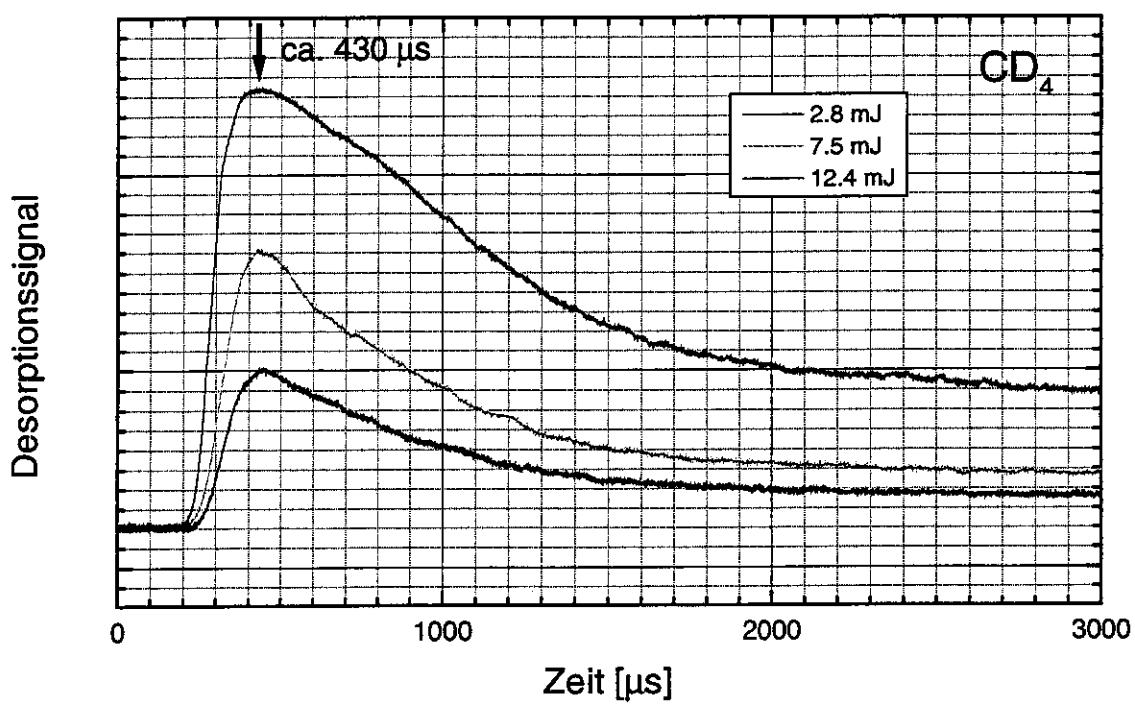
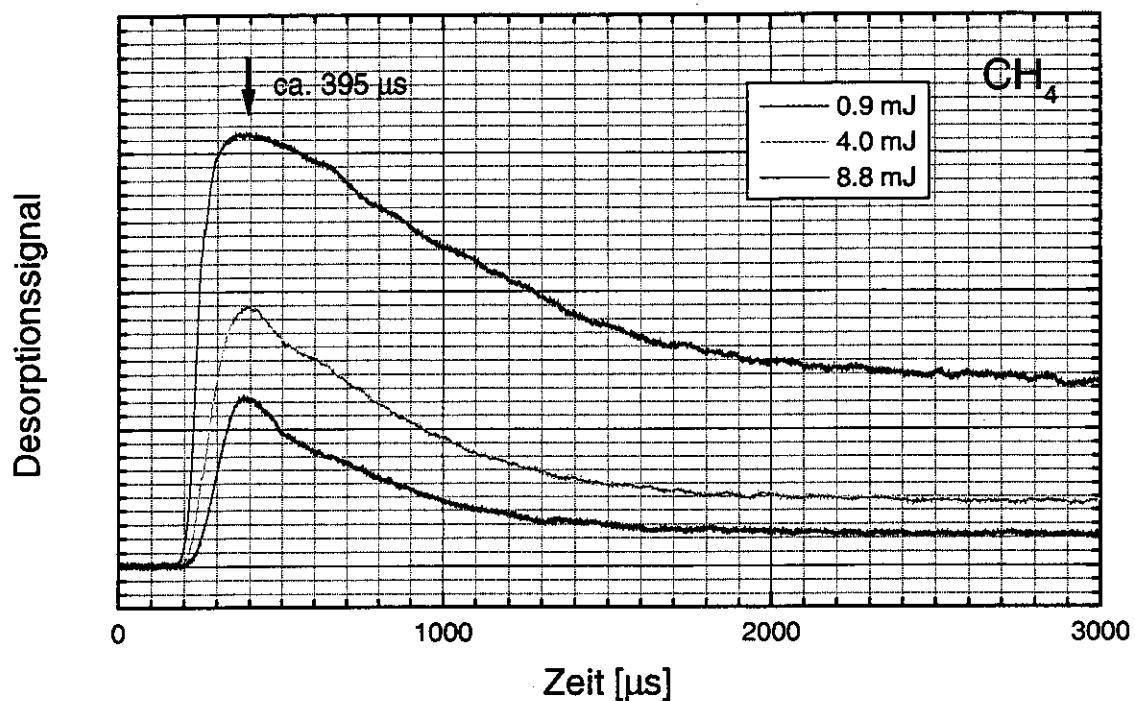
$$v = \sqrt{\frac{2kT}{m}}$$

| | | | |
|----------------------------|-----------------|--------|---------|
| Translational temperature: | CH ₄ | 96.5 K | 99.0 K* |
| | CD ₄ | 97.1 K | 99.3 K* |

(* after correction for the drift time of 5 μ s)

- calculated desorption temperatures well above the measured sample temperature of ~30 K

Dependence of the flight time on the laser energy



- flight time in the desorption spectra of CH_4 and CD_4 is independent on the applied laser energy
- indication for a direct mechanism ?

Determination of translational temperature

$$f(t) \propto \frac{d^i}{(t - t_{drift})^{i+1}} \cdot e^{-\frac{-m \cdot \left(\frac{d}{(t - t_{drift})} - u\right)^2}{2kT}} dt$$

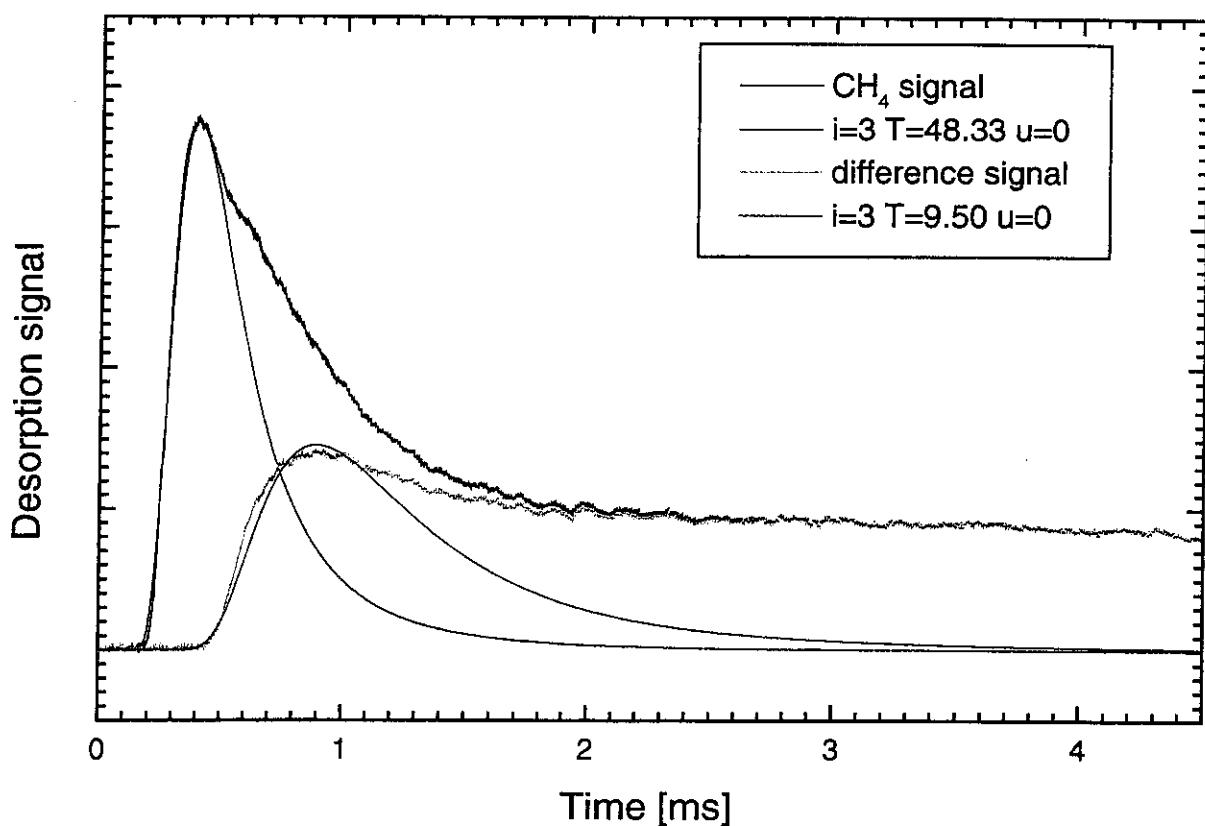
d distance sample-ionizer

t_{drift} flight time ionizer - SEV

i=2 Maxwell-Boltzmann distribution

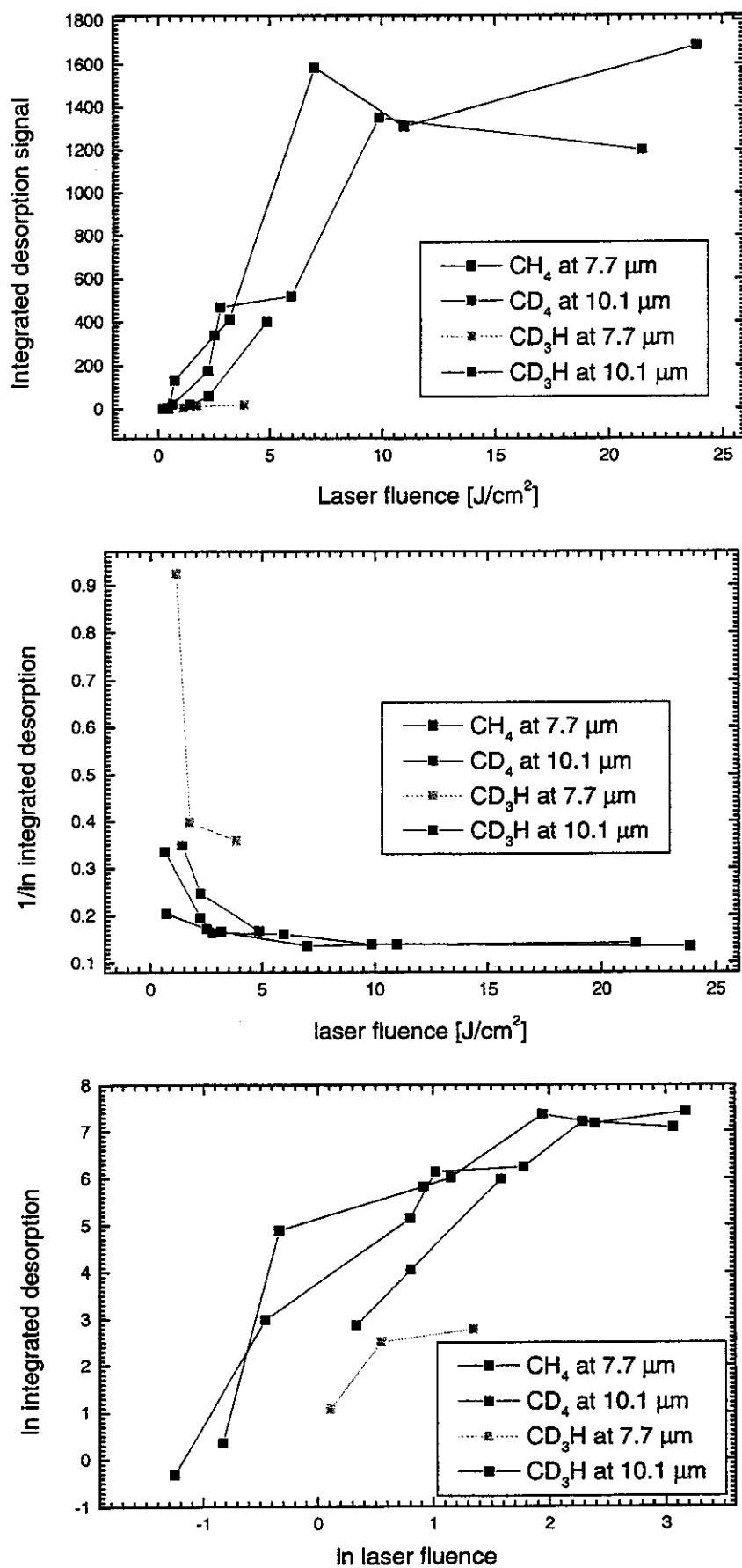
u stream velocity

i=3 flux density distribution



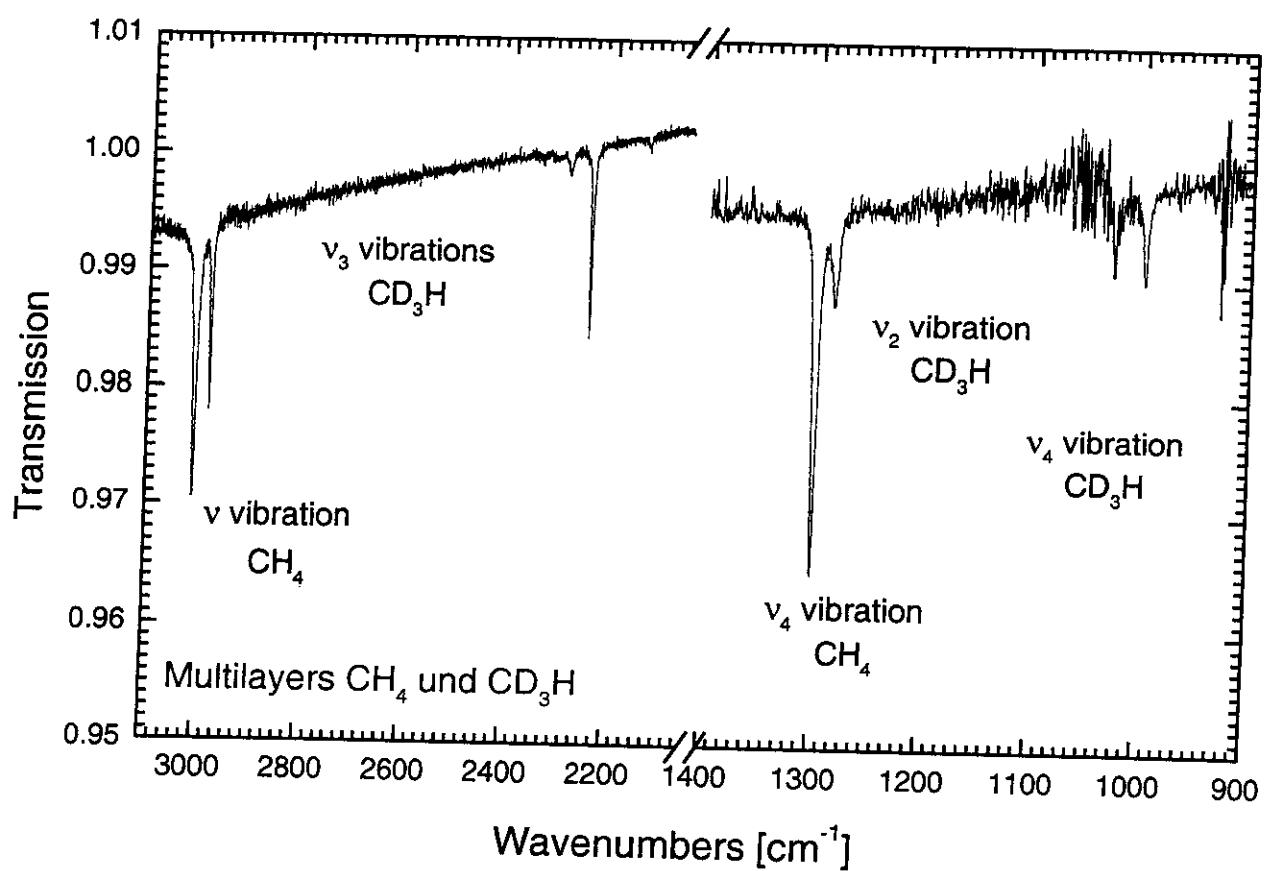
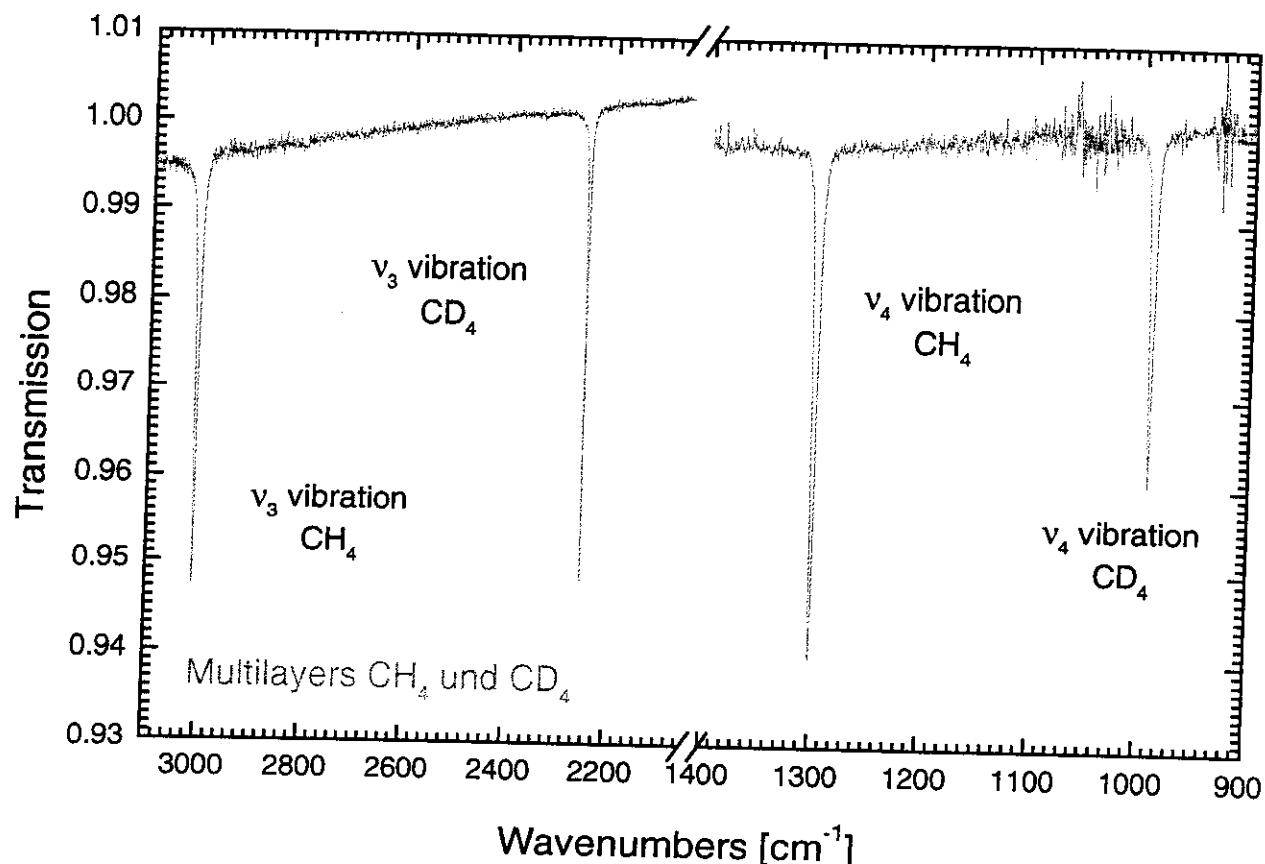
- reasonable fit to the experimental TOF spectra using two functions
- two desorption channels open ?

Desorption signal as a function of the laser fluence



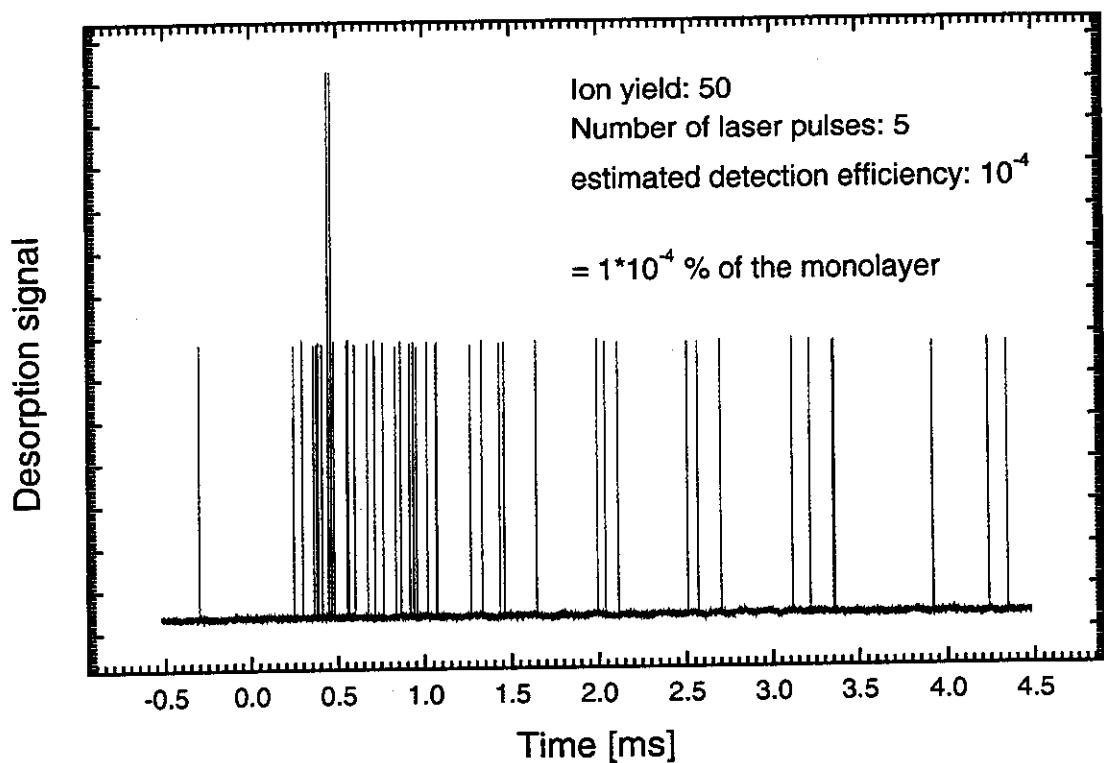
- Indication for a direct desorption mechanism ?

FTIR spectra of the isotope mixtures of methane



Outlook

- resonant monolayer desorption for $\text{CD}_3\text{F}/\text{NaCl}(100)$
detected with single ion detection



- continuation of the studies on multilayer systems
- experiments on monolayers, isotopic mixtures and coadsorbates
- desorption after resonant excitation of external modes of the molecule to the surface
- two color experiments with ultrashort IR laser pulses

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