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SPRING COLLEGE IN MATERIALS SCIENCE

ON

"METALLIC MATERIALS"

(11 May - 19 June 1987)

THEORETICAL MODELING OF DEFECTS AND EMBRITTLEMENT

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THEORETICAL MODELING OF DEFECTS AND EMBRITTLEMENT

1. Aims and methods of theory. The macroscopic theory of embrittlement; thermodynamics and fracture mechanics. What determines the strength of a grain boundary? Fundamental questions: the role of interfacial energy, the importance of precipitates and dislocations, leading to a discussion of elementary processes on the atomic scale.

2. The solid state physicist's contribution. The electron theory of cohesion and bonding in solids. The tight-binding description. Bond energies and interatomic forces.

3. Simpler models for atomistic simulation of defects and their interactions. The second moment approximation. The embedded atom method. The scope and limitations of these models will be described with some examples, including hydrogen embrittlement.

Suggested references and further reading :

D.G.Pettifor, in Theoretical Metallurgy, eds. R.W.Cahn and P.Haasen, North Holland, 1985.

Jong K.Lee, ed., Interatomic potentials and crystalline defects, The Metallurgical Society of AIME, 1981.

R.H.Latensien and J.R.Pickens, eds. Atomistics of fracture, Plenum 1985.

M.H.Finnis, Interatomic forces and the simulation of cracks, Invited lecture at the NATO Advanced Research Workshop on the Chemistry and Physics of Fracture, Bad Reichenhall, 1986, available as report no. TP.1196 from the Main Library, B465, Harwell Laboratory, Oxon OX11 0RA, UK. This contains several other references to papers dealing with the topics of these lectures.

3 LECTURES

by M.W. FINNIS

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①

Theoretical modelling of defects and embrittlement

MIKE FINNIS (HARWELL)

LECTURE I

1. Introduction to embrittlement
2. The variety of parameters involved ($\gamma, d, \sigma_e, \sigma_0$)
3. The importance of a proper understanding of BOND ENERGY

②

Classes of Embrittlement

Solute segregation

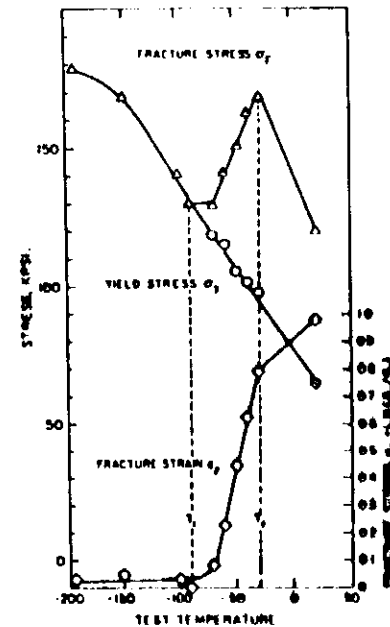
- Reduced high T creep properties of ferritic and Ni based alloys
- Cracking of welds
- RTE : (slow cooling through 300-600C)
 - Fe embrittled by S
 - Ni-Cr steels embrittled by P, Sn, Sb, As (inhibited by Mo)

LME

H embrittlement

He embrittlement

Molybdenum (Passmore, 1964)

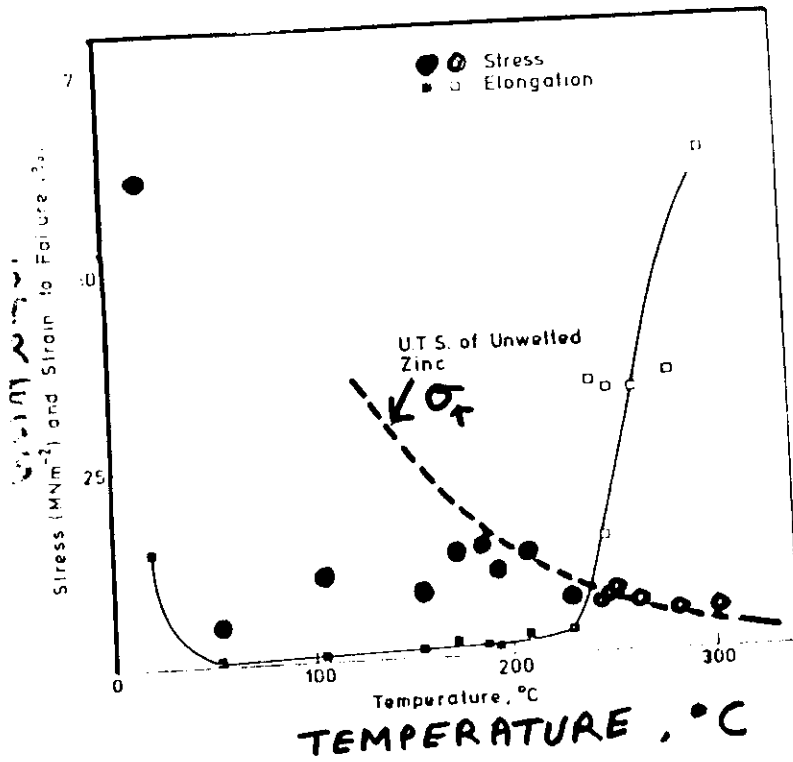


Effect of test temperature on the tensile properties of annealed, fine-grained molybdenum.

Example of a bcc metal

EMBRITTLEMENT OF Zn BY Ga

(4)

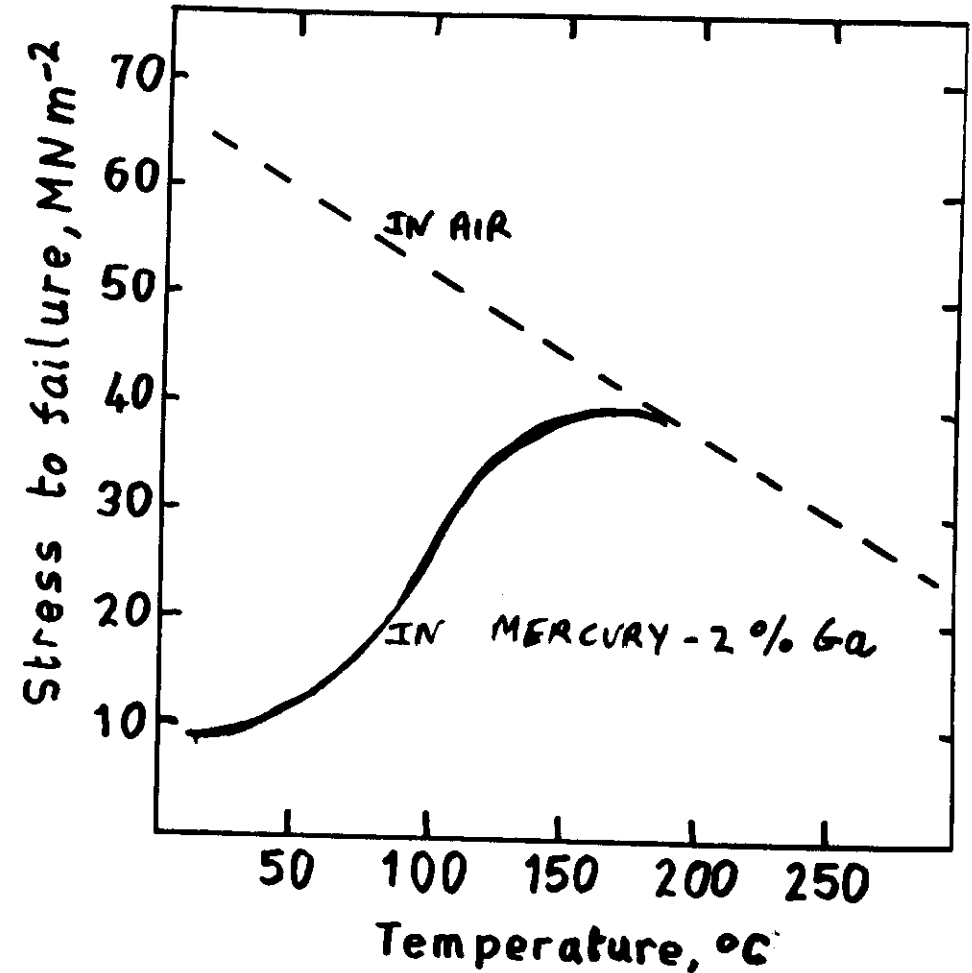


AFTER OLD & TREVENA

Example of an bcc metal

(5)

EMBRITTLEMENT OF Al by Hg-2%Ga

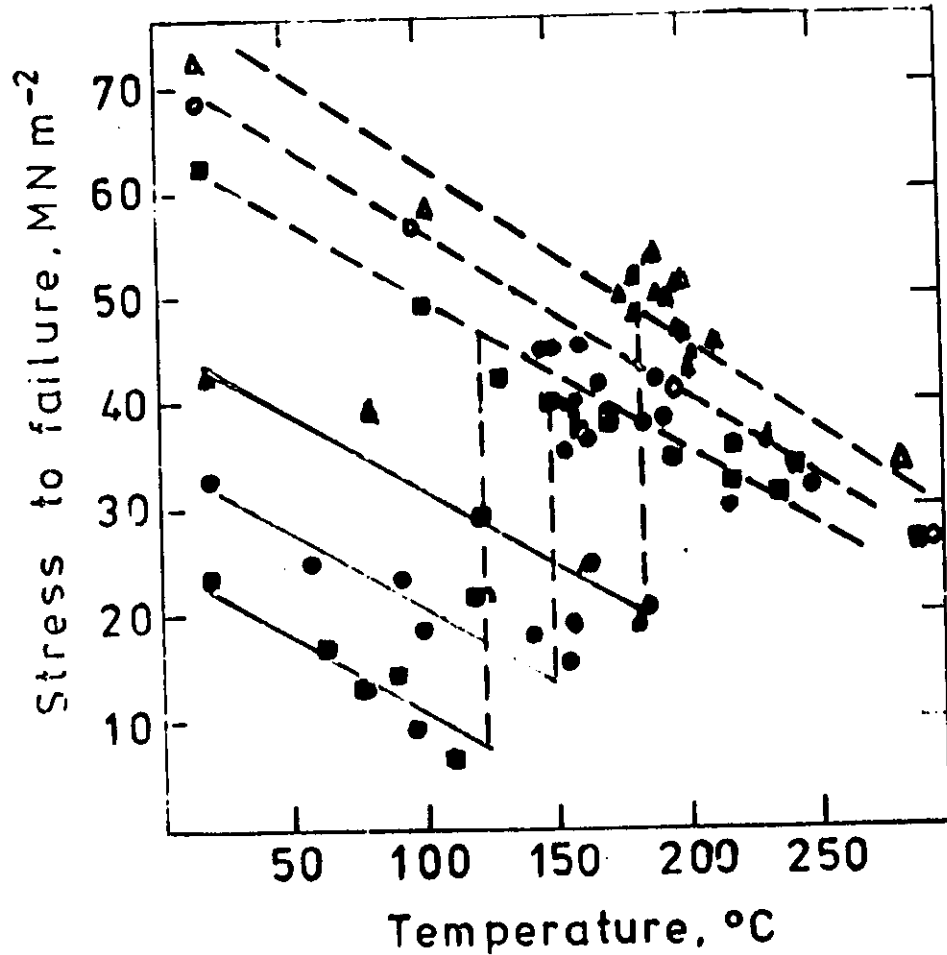


SCHEMATIC (AFTER PREECE & WESTWOOD)

A less abrupt transition

③

EMBRIITLEMENT OF AL by Hg-3%Zn

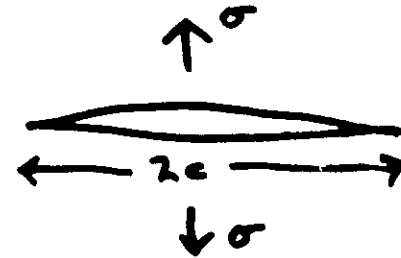


AFTER ICHINOSE

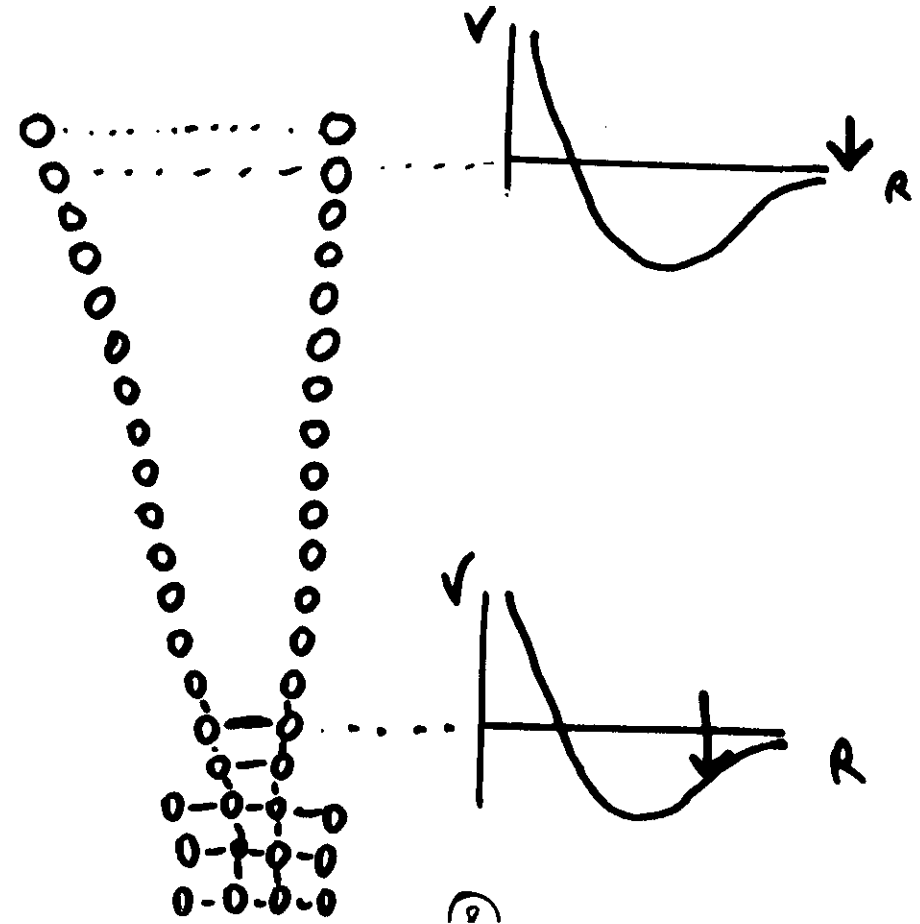
At 150°C get BRITTLE $\rightarrow$ DUCTILE
as d is increased (normal)

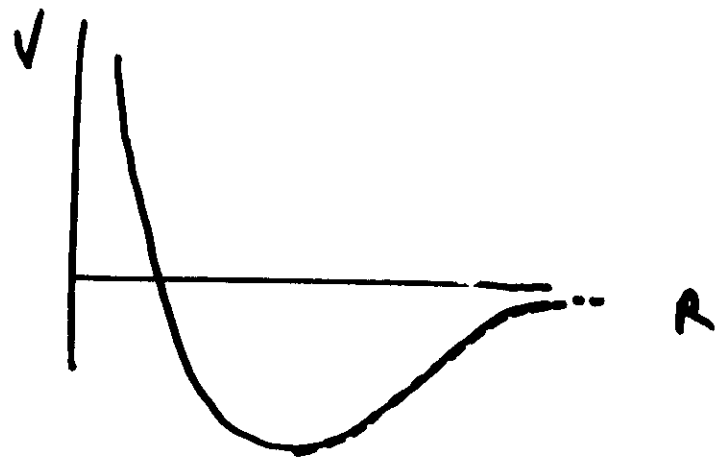
1. 2. 3. 4. 5. 6. 7. 8. 9. 10.

CRACK PROPAGATION



$$\sigma = \sqrt{\frac{2E}{c}}$$

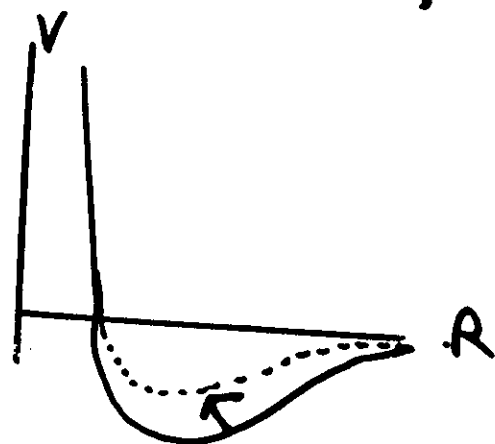




Integrated force
 → well depth = cohesion
 • γ

Grain boundaries: $\gamma \equiv 2\gamma_s - \gamma_{gb}$

Environment:



(a)

Figure 1. The Pile-up of Dislocations

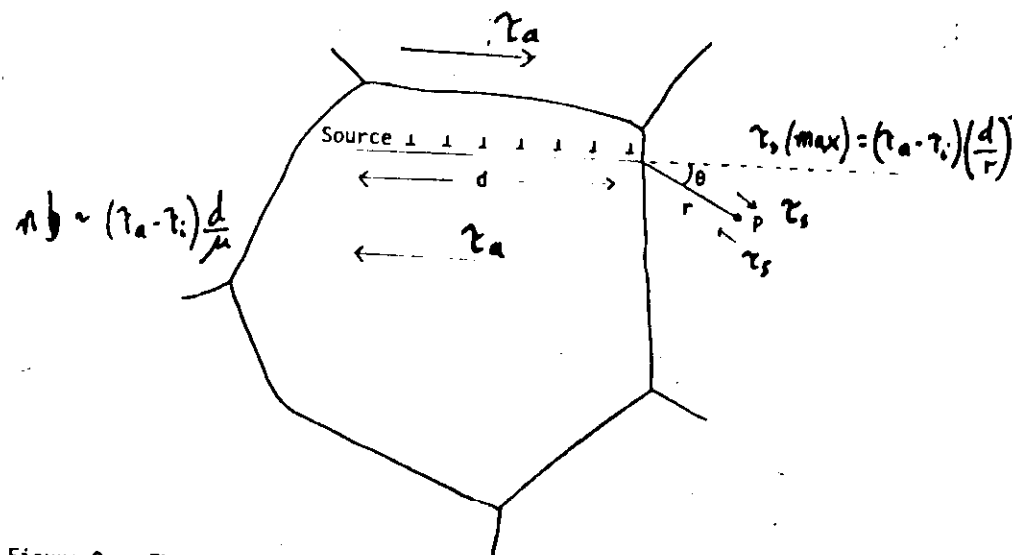
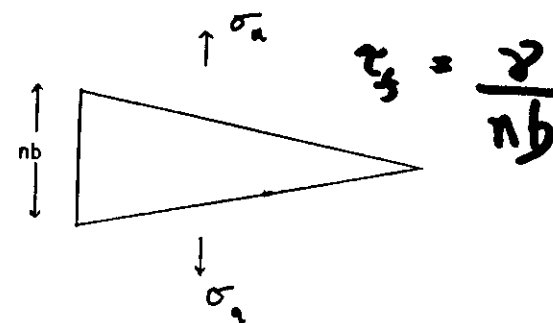
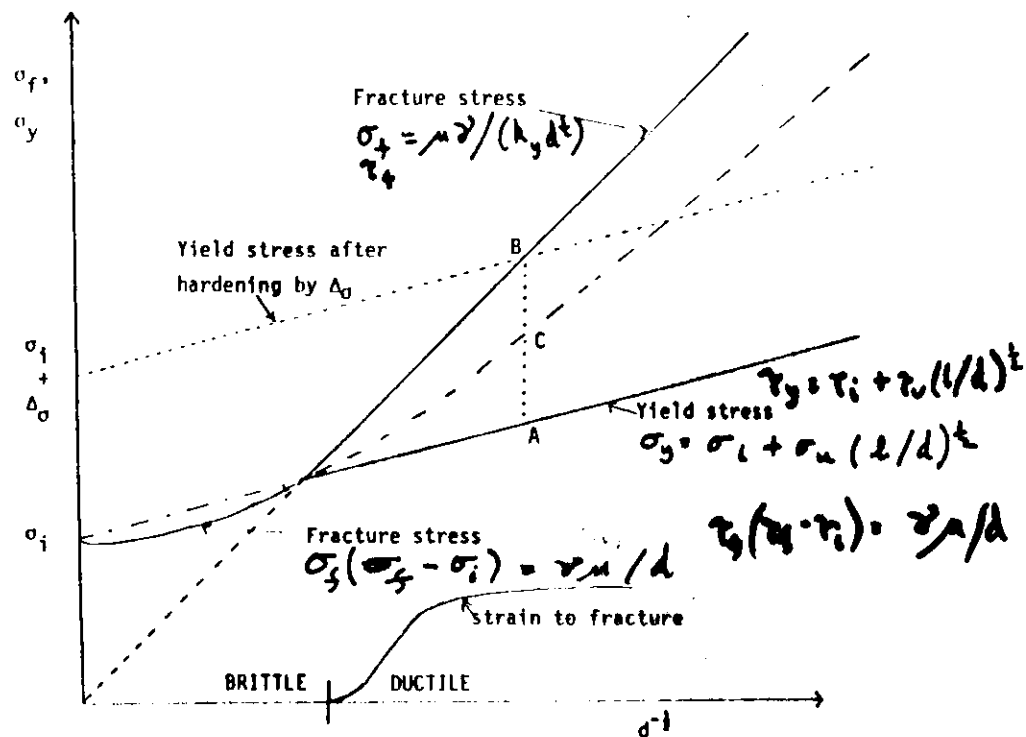


Figure 2. The Wedge Shaped Crack



The Cottrell Theory
 (Crack or Yield?)

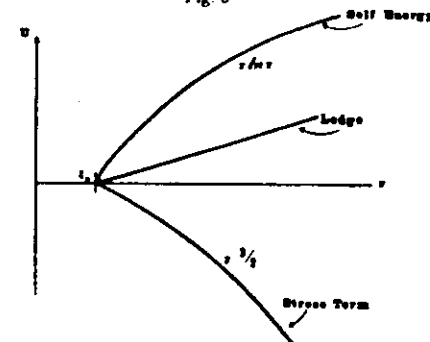
(b)



The Cottrell Theory
(Crack or yield?)

(11)

Fig. 6



Schematic variation of the three terms in eqn. (17) as a function of the radius.

Table 3. Three-dimensional results

Crystal	Activation energy (electron volts)	Radius of activated loop (in units of b)
Pb		
Au		
Cu		
Ag		
Al		
Ni		
Spontaneous emission—No activated state for $r > l_0$		
Na	0.03	1.5
Fe	2.2	5.1
W	329	80.7
Fe*	19	17
LiF	58	32
NaCl	{ 62 } (240)	{ 33 } (98)
MgO	206	37
Al ₂ O ₃	858	30
Bi	111	30
Ge	280	42
C	261	27
Be	180	22
Zn	107	21.2

Cleave or blunt?
Not thermally activated
except Na and Fe

(12)

2. REASONS FOR ABANDONING PAIR POTENTIALS

(1) Experimental E_v / E_{coh}

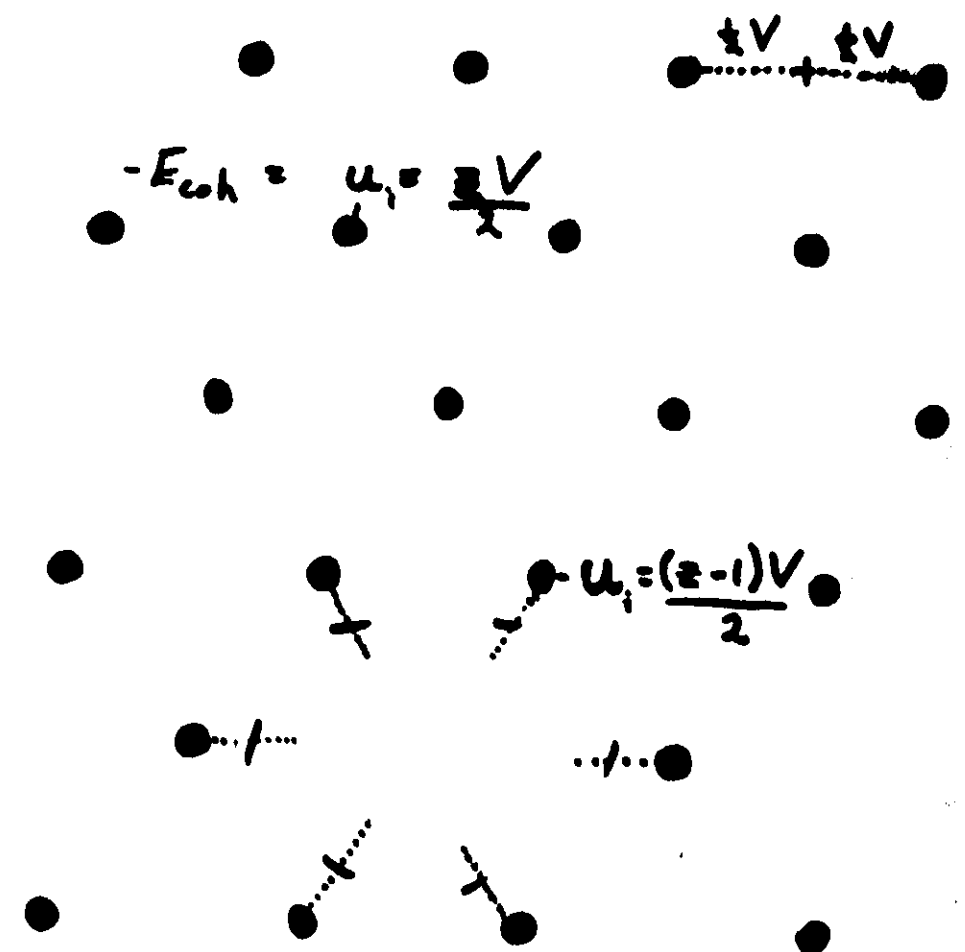
Nb	2.04 / 7.57
Ta	2.9 / 8.10
Cr	1.2 / 4.10
Mo	2.24 / 6.82
W	3.15 / 8.90
Fe	1.77 / 4.28

(2) Cauchy relation

$C_{12} \neq C_{44}$ in real cubic metals
(equilibrium pair potentials
predict $C_{12} = C_{44}$)

(13)

In the old pairwise models, e.g. Lennard Jones
Vacancy formation energy $\approx$ cohesive energy
Energy per atom u_i



$$E_v = z \times \left\{ \frac{(z-1)V}{2} - \frac{zV}{2} \right\}$$

$$= -\frac{zV}{2} = E_{coh}$$

Theoretical Modelling of defects and embrittlement

MIKE FINNIS (HARWELL)

LECTURE II

AIM : TO DERIVE A BOND ENERGY
CONCEPT FOR METALS
BASED ON ELECTRON THEORY
(QUANTUM MECHANICAL)

→ THE TIGHT-BINDING BOND MODEL

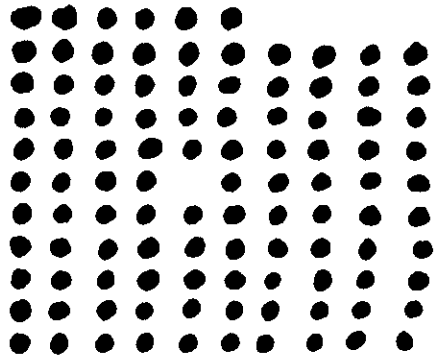
= A SOLID STATE PHYSICIST'S VERSION
OF COULSON'S 'BOND ORDER' CONCEPT
APPROPRIATE FOR 10^{23} ATOMS.

Some additional material is included
here in the copied transparencies.

OUTLINE

1. RECAP - $E_v^{\frac{1}{2}}$ and E_{coh}
2. SUMMARY OF ATOMISTIC SIMULATION WITH TB MODEL
3. SOLID STATE PHYSICS APPROACH TO TOTAL ENERGY - LDF THEORY
4. HARRIS - FOULKES APPROXIMATION
 $\rho = \rho^{\frac{1}{2}}$
5. THE DENSITY OF STATES and LOCAL DENSITY OF STATES
6. TB MODEL → TBB MODEL
7. MOMENTS OF LOCAL D.O.S.
8. EXAMPLES FOR D-BAND METALS
 - BOND ENERGY IN FCC AND HCP
 - VACANCY
9. ADVANTAGES AND DISADVANTAGES OF TBB - DISCUSSION.

VACANCY FORMATION



$$-E_v^* = \frac{1}{2} z V = E_{coh}$$

N atoms

ENERGY

$$N E_{coh} + E_v^* + N^{\frac{1}{2}} \gamma$$

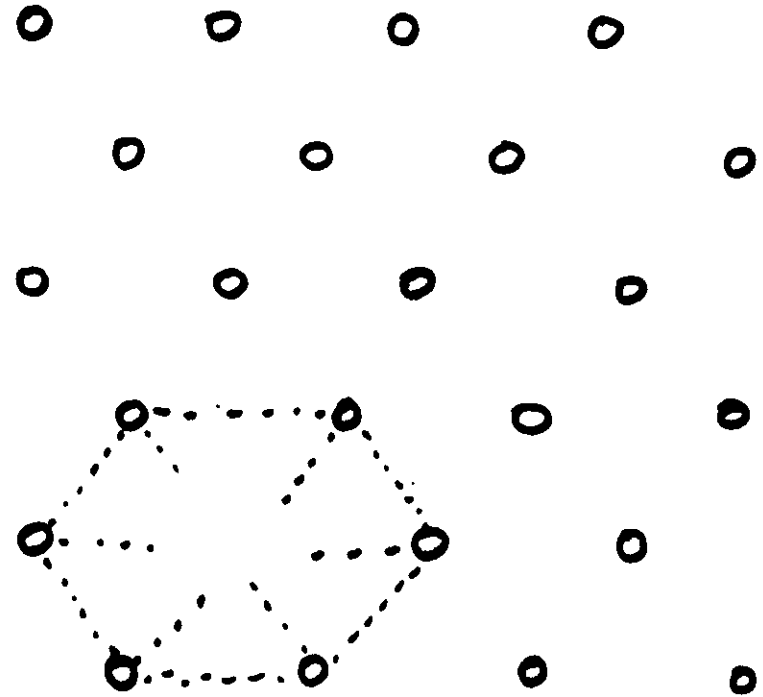
N+1 atoms (perfect)

ENERGY

$$(N+1) E_{coh} + N^{\frac{1}{2}} \gamma$$

Volume change? $\Delta E \approx 0 V \left(\frac{\delta V}{V} \right)^2 \sim 0 \frac{1}{N^2} \sim O\left(\frac{1}{N}\right)$

(14)



$$+ E_{coh} + (z \phi)^{\frac{1}{2}} = u$$

$$\begin{aligned} E_v^* &= (z u' - z u) \\ &= z [(z-1)\phi)^{\frac{1}{2}} - (z\phi)^{\frac{1}{2}}] \\ &= -z [z^{\frac{1}{2}} (1 - \frac{1}{z}) \phi^{\frac{1}{2}} - z^{\frac{1}{2}} \phi^{\frac{1}{2}}] \\ &= \frac{1}{2} (z\phi)^{\frac{1}{2}} \end{aligned}$$

$$E_v^* = \frac{1}{2} E_{coh}$$

VIBES!

TB model for atomistic simulation

- Vacancy : Allan, Lannoo 1970 (metal)
- Surfaces and interfaces
semiconductor, sp^3 : Chadi 1979
metal : W, Mo : Spaanjaard
Terakura, Terakura
and Hamada 1981
Masuda-Tsindo,
Hamada and Terakura
1984
- Dislocations
bcc metals, μ_6 : Masuda, Yamamoto
and Joyama 1983
hcp Ti, ' μ_6 ' : Legrand 1985

$$E = \sum_i \int_{E_i}^{\infty} n_i(E) dE + E_{PAIR}$$

Numerical procedure of previous TB approaches
for $\mu_n, n > 4$

$$\Delta E = \int \epsilon n_i^{(1)} d\epsilon - \int \epsilon n_i^{(2)} d\epsilon$$

For force on atom i ,
 $n_j(\epsilon)$ must be evaluated for
all j to obtain $\sum \Delta n_j$.

e.g. Legrand evaluated Δn_i
to μ_6 , other Δn_j to μ_2 .

Direct calculation of force? $\rightarrow$ TBB
Tight Binding Bond
model.

TBB model applications:

Pettifor and Padloucky (1985)

Separation of structures of
metal-metalloid AB compounds
of types:

NaCl

CsCl

NiAs

MnP

FeB

CrB

Ohta, Finnis, Sutton, Pettifor (to be published)

Grain boundary structures, vacancy.

Paxton et al, grain boundary in Si.
(to be published)

TOTAL ENERGY

1) Superimpose atomic charges g_i
 $\rightarrow g^{\dagger}$

Kohn-Sham effective potential

$$V_{\text{eff}} = v(r) + v_H^{\dagger}(r) + \mu_{xc}^{\dagger}(r)$$

2) Solve single-particle Hamiltonian

$$\tilde{H} = -\frac{1}{2} \nabla^2 + V_{\text{eff}} ; \quad \tilde{H} |n\rangle = \tilde{\epsilon}_n |n\rangle$$

3) Construct output charge density
 g^{out}

4) Iterate until $g \rightarrow g^{\text{sc}}$

$$E_{\text{Total}}^{\text{sc}} = \sum_n a_n \epsilon_n^{\text{sc}} - \int g^{\text{sc}} \left(\frac{1}{2} v_H^{\text{sc}} + \mu_{xc}^{\text{sc}} \right) \\ + E_{xc}[g^{\text{sc}}] + E_{\text{el}}$$

FOULKES AND HARRIS APPROXIMATION

(J. Harris, PRB 31 1770
(1985))

$$E_{TOTAL}^{Harris} = \sum_n a_n \tilde{\epsilon}_n - \int g^{\dagger} (\frac{1}{2} v_H^{\dagger} + \mu_{xc}^{\dagger}) \\ + E_{xc}[g^{\dagger}] + E_{ii} \\ + O[g^{sc} - g^{\dagger}]^2$$

N.B.: $E_{TOTAL}^{Harris} \neq E[g^{\dagger}]$

so we cannot say $E_{TOTAL}^{Harris} > E_{TOTAL}^{sc}$

Spectroscopic constants of homonuclear
dimers

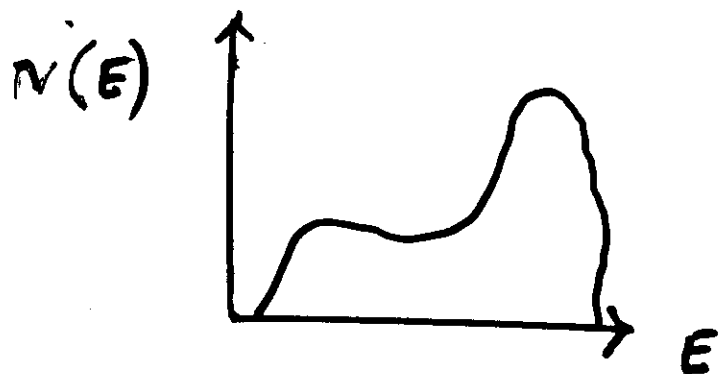


J. Harris PRB 31 1770 (1985)

		Harris	Painter & Averill	Expt.
Be ₂	E _{coh}	0.49	0.50	~0.1
	R	4.50	4.63	4.66
	ω	45	45	28
C ₂	E _{coh}	8.7	7.19	6.2
	R	2.20	2.36	2.35
	ω	246	232	230
N ₂	E _{coh}	10.7	11.34	9.91
	R	2.03	2.08	2.07
	ω	346	296	292
F ₂	E _{coh}	3.7	3.32	1.65
	R	2.71	2.62	2.68
	ω	120	133	111
Cu ₂	E _{coh}	2.9	2.65	2.03
	R	4.10	4.10	4.10
	ω	35	41	33

THE DENSITY OF STATES

S.S.P. vol 35



$$N(E) = \sum_n \delta(E - \epsilon_n)$$

where

$$H \psi_n(r) = \epsilon_n \psi_n(r)$$

LOCAL D.O.S. :

$$N_i(E) = \sum_n \delta(E - \epsilon_n) |a_n^i|^2$$

$$a_n^i = \int d^3r \phi_i^*(r) \psi_n(r)$$

$\phi_i(r)$ is an atomic orbital on site i (index i includes site & L)

Notation

$$H |n\rangle = \epsilon_n |n\rangle$$

$$\psi_n(r) = \langle r | n \rangle$$

$$\begin{aligned} a_n^i &= \int d^3r \phi_i^*(r) \psi_n(r) \\ &= \int d^3r \langle i | r \rangle \langle r | n \rangle \end{aligned}$$

$$\left. \begin{aligned} \int d^3r |r\rangle \langle r| &= 1 \\ \sum_n |n\rangle \langle n| &= 1 \end{aligned} \right\} \text{COMPLETENESS}$$

$$a_n^i = \langle i | n \rangle$$

$$\begin{aligned} N_i(E) &= \sum_n \delta(E - \epsilon_n) \langle i | n \rangle \langle n | i \rangle \\ &= -\frac{1}{\pi} \text{Im} (E^* - H)^{-1} \langle i | n \rangle \langle n | i \rangle \\ &= -\frac{1}{\pi} \text{Im} \langle i | n \rangle \langle n | (E^* - H)^{-1} | n \rangle \langle n | i \rangle \end{aligned}$$

$$N_i(E) = -\frac{1}{\pi} \text{Im} \langle i | (E^* - H)^{-1} | i \rangle$$

$$(E-H)^{-1} \equiv G$$

i.e. $EG = HG + 1$

$$\text{Im} \langle i | EG | j \rangle = \text{Im} \langle i | HG | j \rangle$$

Cohesive energy

Convenient notation results if we introduce the output density operator

$$g^{\text{out}} = \sum_n a_n |n\rangle \langle n|$$

where

$$\tilde{H} |n\rangle = \tilde{\epsilon}_n |n\rangle.$$

so that

$$\sum_n a_n \tilde{\epsilon}_n \rightarrow \text{Tr } g^{\text{out}} \tilde{H}$$

We then have the basis independent form:

$$E_{\text{total}}^{\text{min}} = \text{Tr } g^{\text{out}} \tilde{H} - \text{Tr } g^{\dagger} (\epsilon_H^{\dagger} + \mu_H^{\dagger}) + E_{\text{el}}(g^{\dagger}) + E_H$$

Cohesive energy

Subtract from E_{TOTAL} the energy of separated atoms $E_{\text{ATOMS}}^{\text{sc}} =$

$$\sum_i \left\{ \text{Tr } g_i (T + \frac{1}{2} v_{\text{Hi}} + v_i) + E_{\text{xc}}[g_i] \right\}$$

to get

$$E_{\text{COH}}^{\text{Harris}} = E_{\text{TOTAL}}^{\text{Harris}} - E_{\text{ATOMS}}^{\text{sc}}$$

$$E_{\text{COH}}^{\text{Harris}} = \text{Tr} (g^{\text{out}} - g^{\text{f}}) \tilde{H} + \Delta E_{\text{es}}[g^{\text{f}}] + \Delta E_{\text{xc}}[g^{\text{f}}]$$

The ' Δ ' represent changes on formation of the solid.

E_{es} represents all electrostatic terms.

So far have not even made LDA - above is exact

(7.9)

BOND ENERGY

Recall

$$E_{\text{COH}}^{\text{Harris}} = \text{Tr} (g^{\text{out}} - g^{\text{f}}) \tilde{H} + \Delta E_{\text{es}}[g^{\text{f}}] + \Delta E_{\text{xc}}[g^{\text{f}}]$$

The first term will be

the chemical bonding,

the last terms are a pair potential.

Work with a basis of atomic orbitals.

Orbitals on atom i are $|\phi_{i\alpha}\rangle$.

Atomic charge density

$$g_i(r) = \sum_{\alpha} \langle r | \phi_{i\alpha} \rangle g_i^{\text{val}} \langle \phi_{i\alpha} | r \rangle$$

$$\text{and } g^{\text{f}} = \sum_i g_i$$

$$= \sum_{i\alpha} \langle r | \phi_{i\alpha} \rangle (g^{\text{f}})^{\text{val}} \langle \phi_{i\alpha} | r \rangle$$

(7.10)

Bond energy

$$\begin{aligned} \text{Tr} (g^{\text{out}} - g^{\dagger}) \tilde{H} \\ = \sum_{i, \alpha} \sum_{j, \beta} \left((g^{\text{out}})^{i\alpha j\beta} - (g^{\dagger})^{i\alpha j\beta} \delta_{i\alpha}^{j\beta} \right) \tilde{H}_{i\alpha j\beta} \\ + \sum_{i, \alpha} \sum_{j \neq i, \beta} (g^{\text{out}})^{i\alpha j\beta} \tilde{H}_{i\alpha j\beta} \end{aligned}$$

First term is site-diagonal part
= energy associated with change in
occupancy of atomic orbitals

$$\text{e.g. } 2\tilde{\epsilon}_s + 2\tilde{\epsilon}_p \rightarrow \tilde{\epsilon}_s + 3\tilde{\epsilon}_p$$

Second term = BOND ENERGY

$$\begin{aligned} E_{\text{BOND}} &= \sum_{i \neq j, \alpha, \beta} (g^{\text{out}})^{i\alpha j\beta} \tilde{H}_{i\alpha j\beta} \\ &\equiv \sum_{i, \alpha} \int^{\epsilon_f} n_{i, \alpha}(\epsilon) (E - \tilde{H}_{i, \alpha i, \alpha}) d\epsilon \end{aligned}$$

(21)

The pair potential

$$\begin{aligned} \Delta E_{\text{es}} &= \sum_{i \neq j} \text{Tr} g_i (\pm v_{ij} + v_j) + E_{ii} \\ &= \text{pairwise} \end{aligned}$$

$$\Delta E_c = \frac{1}{2} \sum_{i \neq j} \iint g_i(r) \frac{h_c(g(r), r, r')}{|r - r'|} g_j(r')$$

(Lindholm and Lundqvist suggest that
 $h_c \sim \text{constant}$)

So $\Delta E_c \approx \text{pairwise}$.

From exchange energy:

$$E_x = -\frac{1}{4} \sum_{i, \alpha} \sum_{j, \beta} \sum_{l, \mu} \sum_{m, \nu} g^{i\alpha j\beta} g^{l\mu m\nu} (i\alpha m\nu | l\mu j\beta)$$

where

$$(i\alpha m\nu | l\mu j\beta) = \iint d\mathbf{r} d\mathbf{r}' \phi_{i\alpha}^*(\mathbf{r}) \phi_{m\nu}(\mathbf{r}) \phi_{l\mu}^*(\mathbf{r}') \phi_{j\beta}(\mathbf{r}') \frac{1}{|\mathbf{r} - \mathbf{r}'|}$$

$$\begin{aligned} \Delta E_x &= -\frac{1}{4} \sum_{i, \alpha} \sum_{j, \beta} (g^{\dagger})^{i\alpha i\alpha} (g^{\dagger})^{j\beta j\beta} (i\alpha j\beta | j\beta i\alpha) \\ &= \text{pairwise} \end{aligned}$$

(22)

Interatomic Forces

$$E_{\text{coh}} = \text{Tr} (g^{\text{out}} - g^{\text{t}}) \tilde{H} + E_{\text{PAIR}}$$

$$\frac{\partial E_{\text{coh}}}{\partial x_n} = \text{Tr} (g^{\text{out}} - g^{\text{t}}) \frac{\partial \tilde{H}}{\partial x_n} + \frac{\partial E_{\text{PAIR}}}{\partial x_n}$$

$$+ \text{Tr} \left(\frac{\partial g^{\text{out}}}{\partial x_n} - \frac{\partial g^{\text{t}}}{\partial x_n} \right) \tilde{H}$$

$\nwarrow$
 $\sum_n \frac{\partial a_n}{\partial x_n} \tilde{E}_n$
 $\equiv 0$

$\searrow$
 $\sum_{ia} \frac{\partial (g^{\text{t}})^{iaia}}{\partial x_n} \tilde{H}_{iaia}$
 $\equiv 0$

So

$$\frac{\partial E_{\text{coh}}}{\partial x_n} = \sum_{ia} \sum_{jp} (g^{\text{out}})^{iajp} \frac{\partial \tilde{H}^{jpia}}{\partial x_n} - \sum_{ia} (g^{\text{t}})^{iaia} \frac{\partial \tilde{H}_{iaia}}{\partial x_n} + \frac{\partial}{\partial x_n} E_{\text{PAIR}}$$

- i) $S = 1$ (diagonal overlap matrix)
- ii) 2 centre H
- iii) Neutral atoms (perfect screening)
- iv) $\tilde{H}_{iaip} = 0$ if $a \neq \beta$

$$\frac{\partial E_{\text{coh}}}{\partial x_n} = 2 \sum_{ia} \sum_{jp} (g^{\text{out}})^{iajp} \frac{\partial \tilde{H}^{jpia}}{\partial x_n}$$

$$(g^{\text{out}})^{iajp} = -\frac{\lambda}{\pi} \int \frac{E_g}{2m} \tilde{G}^{iajp}(E+10) dE$$

The following 2 pages are a briefer 'derivation' of the above TBB model from the TB model.

The TBB model TIGHT BINDING MODEL

Band picture:

$$U = \int^{\epsilon_F} \epsilon n(\epsilon) d\epsilon + U_{rep} \quad \swarrow \text{Pairwise}$$

$$U_{rep} = - \int d\epsilon n(\epsilon) \left(\frac{1}{2} \phi(r) + \mu_{xc}(\epsilon) \right) + E_{xc}[n(\epsilon)] + E_{ii}$$

Kohn-Sham 1965

Harris 1985, PRB 31, 1770

(Justifies pairwise description of U_{rep})

$$U_{band} = \int^{\epsilon_F} \epsilon n(\epsilon) d\epsilon \\ \equiv -\frac{1}{\pi} \int^{\epsilon_F} \epsilon \text{Im Tr } G$$

Definition of G :

$$(\epsilon - H)G = 1$$

(35)

$$U_{band} = -\frac{1}{\pi} \int^{\epsilon_F} \epsilon \text{Im Tr } G d\epsilon \\ = -\frac{1}{\pi} \int^{\epsilon_F} \text{Im Tr } H G d\epsilon \\ = -\frac{1}{\pi} \int^{\epsilon_F} \sum_{ij} \text{Im } G_{ij} H_{ji} d\epsilon \\ = \frac{\sum_{ij} q_{ij} H_{ij}}{\dots}$$

$$= \underbrace{\delta_{ii}^{ii}}_{\text{'charge' or 'potential'}} + \underbrace{H_{ii}^{ii}}_{\text{'charge' or 'potential'}}$$

$$U_{coh} = \sum_i \sum_j q_{ij} H_{ij} + U_{rep}$$

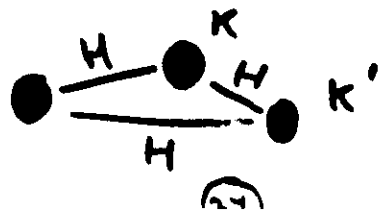
Forces:

$$\delta U = \sum_i \sum_j q_{ij} \delta H_{ij} + \delta U_{rep}$$

(36)

Moments of local DOS

$$\begin{aligned}
 \mu_N &= \int_{-\infty}^{\infty} E^N \rho_i(E) dE \\
 &= \int E^N \sum_{\mathbf{k}} \delta(E - \epsilon_{\mathbf{k}}) \langle i | \mathbf{k} \rangle \langle \mathbf{k} | i \rangle \\
 &= \sum_{\mathbf{k}} \langle i | \mathbf{k} \rangle E^N \langle \mathbf{k} | i \rangle \\
 &= \sum_{\mathbf{k}} \langle i | \mathbf{k} \rangle \langle \mathbf{k} | H^N | \mathbf{k} \rangle \langle \mathbf{k} | i \rangle \\
 &= \sum_{\mathbf{k}, \mathbf{k}'} \langle i | \mathbf{k} \rangle \langle \mathbf{k} | H^N | \mathbf{k}' \rangle \langle \mathbf{k}' | i \rangle \\
 &= \langle i | H^N | i \rangle \\
 &= \sum_{\mathbf{k}, \mathbf{k}', \mathbf{k}'' \dots} \langle i | H | \mathbf{k} \rangle \langle \mathbf{k} | H | \mathbf{k}' \rangle \langle \mathbf{k}' | H | \mathbf{k}'' \rangle \dots \langle \mathbf{k}'' | H | i \rangle
 \end{aligned}$$



CANONICAL d-BAND PARAMETERS

$$\begin{aligned}
 H_{11} = H_{44} &= dd\sigma(R) = -2\left(\frac{5}{R}\right) \\
 H_{22} = H_{33} &= dd\pi(R) = +\left(\frac{5}{R}\right) \\
 H_{55} &= dd\sigma(R) = -\left(\frac{5}{R}\right)
 \end{aligned}$$

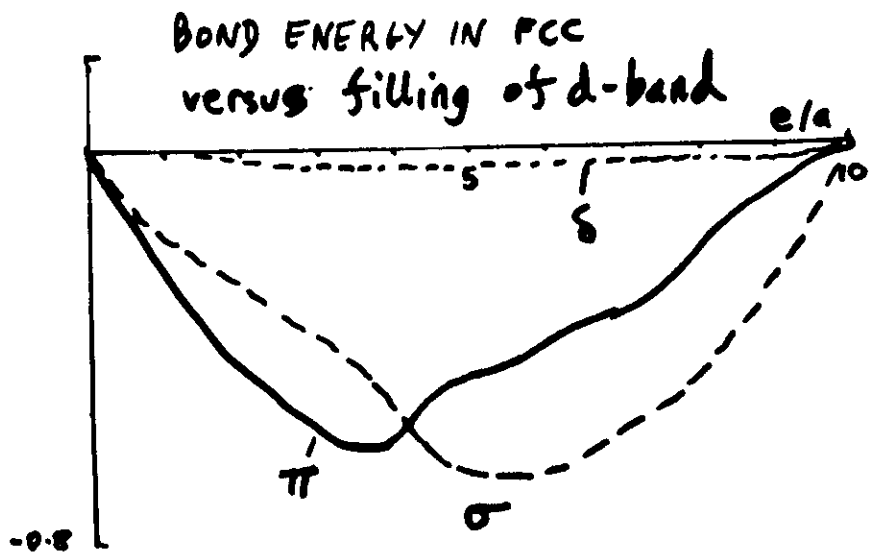
Energy of bond

$$U_{bond}^{ii} = dd\sigma \times g_{55} + dd\pi \times (g_{22} + g_{33}) + dd\sigma \times (g_{11} + g_{44})$$

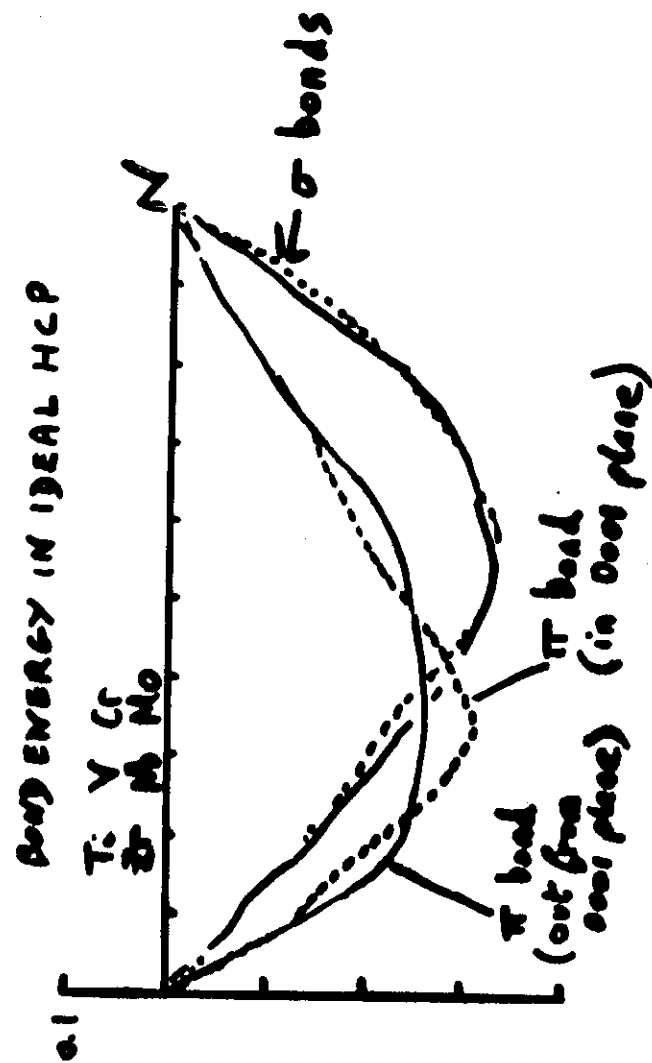
Radial force

$$F_{bond}^{ii} = -\frac{5}{R} \times U_{bond}^{ii}$$

But generally also transverse F.



(39)



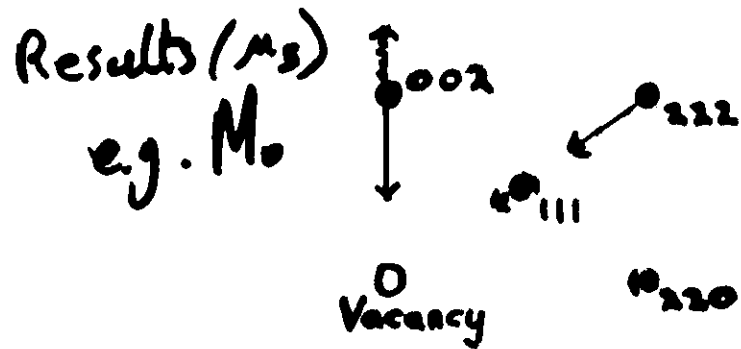
(40)

$$\frac{c}{a} \rightarrow < 1.6 \text{ in } \text{Ti, Ti}$$

(21)

Vacancy in FCC TRANSITION METALS

Ohta, Finnis, Pettifor & Sutton
to be published



.... $\rightarrow$ = Force in perfect lattice
 $\rightarrow$ = Force due to vacancy

Due to vacancy:
Bond strengthening enhances defect forces

Ohta, Finnis, Pettifor and Sutton, to be published

.... $\rightarrow$ Forces on perfect lattice due to atom at origin; only 1st + 2nd neighbour occur

$\rightarrow$ Defect forces. In a pair-potential model these are equal and opposite to $\rightarrow$.

TBB MODEL: DISCUSSION

ADVANTAGES (with recursion method)

1. Physical description of charge redistribution between bonds
2. Controlled introduction of local environment (μ_n)
3. Simplest physical model of structural energy differences
4. CPU time $\propto N$ (c.f. N^3 for matrix inversion)
so large problems should be suitable

DISADVANTAGES

1. Limited basis. Need to investigate transferability of M_{ij} . How good are the simplest assumptions, e.g. d orbitals, orthogonality?
2. Still costly ($10^2 - 10^3 \times$ RAM and V_{ij})
3. $F \approx \nabla E$

Theoretical Modelling of defects and embrittlement

MIKE FINNIS (HARWELL)

LECTURE III

AIM : TO EXPLAIN A SIMPLE
MODEL OF BOND ENERGY

→ THE EMBEDDED ATOM MODEL
(N-BODY POTENTIAL)

INTERATOMIC FORCES

1. PAIR POTENTIALS:

$$U_i = \frac{1}{2} \sum_j V_{ij} (+ F(\text{volume}))$$

2. BOND ANGLE FORCES

$$U_i = \sum_{j,k} f(R_{ij}, R_{ik})$$

3. EMBEDDED ATOM MODELS

$$U_i = f(\rho_i) + \frac{1}{2} \sum_j V_{ij}$$

4. TIGHT-BINDING MODELS

$$U_i = \sum_n \epsilon_n w_{ni} + \frac{1}{2} \sum_j V_{ij} \equiv$$

TBB

$$U_i = \sum_{j,i} g_{ij} H_{ij} + \frac{1}{2} \sum_j V_{ij}$$

5. DENSITY FUNCTIONAL

$$\sum_i U_i = \int^{\epsilon_F} \epsilon n(\epsilon) d\epsilon - \int n(r) (\frac{1}{2} \phi(r) + \mu_{xc}(r)) \\ + E_{xc}[n(r)] + E_{ii}$$

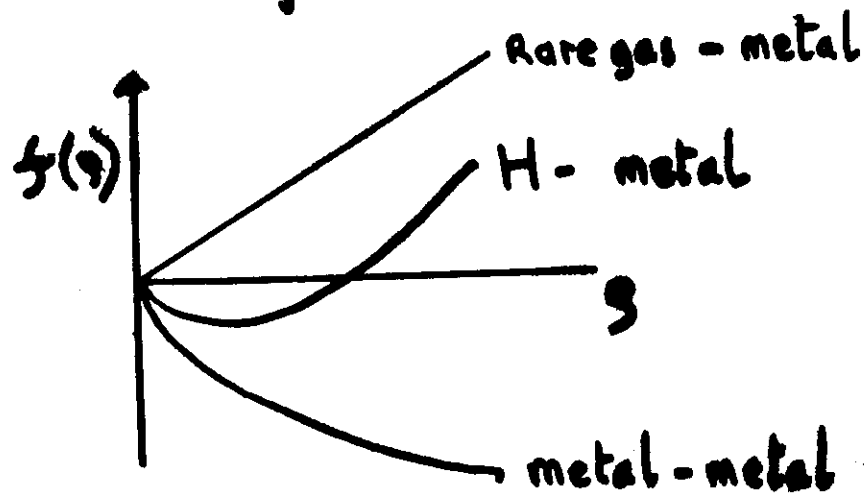
Empirical N-body potentials - EAM

Stolt & Zaremba
Norshov
Daw & Baskes

Finnis & Sinclair
Ackland et al
Bacon et al
Ercolessi et al

$$E_i = f(\rho_i) + \frac{1}{2} \sum_j V(R_{ij})$$

$$\rho_i = \sum_j \phi(R_{ij})$$



'EFFECTIVE' PAIR INTERACTION

$$f(\rho_i) = f(\rho_i^0) + f' \cdot \delta \rho_i + \frac{1}{2} f'' (\delta \rho_i)^2 +$$

$$\delta \rho_i = \sum_j \frac{\partial \phi}{\partial R_{ij}} (R_{ij}^0) \cdot (\delta R_i - \delta R_j) + \frac{1}{2} \sum_j \frac{\partial^2 \phi}{\partial R_{ij}^2} (\delta R_i - \delta R_j)^2 +$$

$$\text{So: } f(\rho_i) = f(\rho_i^0) + f' \cdot \sum_j \delta \phi(R_{ij}) + \frac{1}{2} f'' \left(\sum_j \delta \phi(R_{ij}) \right)^2 +$$

$$\underline{V_{eff}(R) = V(R) + 2 f'(\rho_0) \phi(R)}$$

$$\text{bcc: } \Omega C_{44} = \frac{8}{9} V'_{eff}(R_1) + \frac{4}{9} V''_{eff}(R_1) R_1^2 + V'_{eff}(R_2) R_2$$

$$\Omega (C_{11} - C_{12}) = \frac{4}{3} V'_{eff}(R_1) R_1 + V''_{eff}(R_2) R_2$$

$$\Omega (C_{12} - C_{44}) = \underline{f''(\rho_0) \left(\frac{8}{3} \phi'(R_1) R_1 + 2 \phi'(R_2) R_2 \right)}$$

$$H \psi = E \psi$$

$$U_{\text{band}} = \sum_i \int_0^{E_F} \epsilon n_i(\epsilon) d\epsilon \quad \leftarrow \begin{array}{l} \text{energy} \\ \text{density of states} \\ \text{on atom } i \end{array}$$

Moments

$$\mu_n = \int_{-\infty}^{\infty} \epsilon^n n_i(\epsilon) d\epsilon$$

The densities of states n_i are characterised by calculating their lowest n moments.

Simplest is μ_2 approximation.

$$\mu_2 = \sum_j H_{ij}^2 \propto \text{no. of neighbours}$$

$$\Rightarrow n(\epsilon) = \frac{10}{\sqrt{2\pi\mu_2}} \exp\left(-\frac{\epsilon^2}{2\mu_2}\right)$$

$$\Rightarrow \int_0^{E_F} \epsilon n(\epsilon) d\epsilon = -10 \sqrt{\frac{\mu_2}{2\pi}} \exp\left(-\frac{E_F^2}{2\mu_2}\right) \propto \sqrt{E}$$

Finnis and Sinclair : bcc transition metals

$$E_i = -\sqrt{g_i} + \frac{1}{2} \sum_j V(R_{ij})$$

$$g_i = \sum_j \phi(R_{ij})$$

$$\phi(R_{ij}) = A^2 (R_{ij} - d)^2 \quad R_{ij} < d$$

$$V(R_{ij}) = (R_{ij} - c)^2 (c_0 + c_1 r + c_2 r^2) \quad R_{ij} < c$$

cutoffs c and d between 2nd and 3rd neighbours.

PARAMETERS
 c_0, c_1, c_2, A are determined by fitting to $E_{\text{coh}}, \rho, B, c_{12}-c_{44}$.

Various cutoffs then to fit C_{44} (non-linear) and ν_V^0 .

c and d .

Vacancy formation energies and volumes

Matthai and Bacon 1985

Harder and Bacon 1986

$$V_{ir}^{\dagger} = - \frac{\Omega}{B} \frac{dE_{ir}^{\dagger}}{d\Omega} \quad (\text{Finnis \& Sachdev 1976})$$

	$E_{ir}^{\dagger}$ (eV)		$V_{ir}^{\dagger}$ (Ω)
	Calc.	Expt.	
V	1.83	2.1 ± 0.2	0.74
Nb	2.48	2.6 ± 0.3	0.96
Ta	2.89	2.9 ± 0.6	0.83
Cr	1.91	-	0.84
Mo	2.84	3.0 ± 0.2	0.93
W	3.62	4.0 ± 0.3	0.68
Fe	1.81	1.6 ± 0.2	0.85

Vacancy migration energies

Harder and Bacon 1986 Phil Mag A 54 65

	E_{ir}^m (eV)	
	calc.	expt.
V	0.76	1.3 ± 0.3
Nb	0.91	1.4 ± 0.3
Ta	1.22	1.4 ± 0.6
Cr	0.71	-
Mo	1.32	1.5 ± 0.2
W	1.49	1.5 ± 0.3
Fe	0.91	1.3 ± 0.3

Divacancy binding energies

	1st neighbour	2nd neighbour
V	0.20	0.18
Nb	0.28	0.23
Ta	0.28	0.28
Cr	0.22	0.24
Mo	0.30	0.26
W	0.43	0.40
Fe	0.16	0.20

Harder and Bacon 1985

Interstitial Formation Energies

Ackland + Thetford 1986

	$\langle 100 \rangle$	$\langle 110 \rangle$	$\langle 111 \rangle$	Crowdion
V	4.94	4.16	4.61	4.60
Nb	4.82	4.49	4.80	4.86
Ta	8.07	6.85	7.16	7.16
Mo	7.21	7.04 (6.98)	7.22	7.19
W	9.82	9.64	8.92	8.88

Without added core:

Ta	4.10	3.97	3.93	3.81
Mo	7.20	7.05	7.25	7.21

Miller (1981) pair potential

Mo	6.91	4.81	5.85	5.75
----	------	------	------	------

Harder
+ Bacon
1986

All calculations used a $7 \times 7 \times 7$ block. Estimated maximum error: ± 0.04 eV.

NB.: Bent interstitial $\approx \langle 530 \rangle$ in Mo.

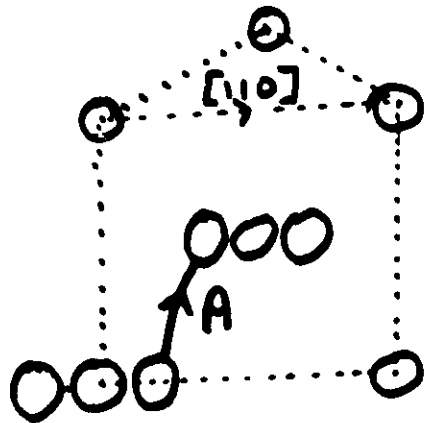
: Crowdion in W.

Interstitial Migration

Mo (Harder + Bacon 1986)

3 Paths for $\langle 110 \rangle$ dumbbell migration.

	E_i^m		
	N-body	2-body	Expt.
A: 2-d	0.17	1.03	$\sim 0.05 \text{ eV}$ (recovery)
B: 3-d	0.16	0.63	
C: double 2-d	0.24	0.92	



Interstitial migration

W (Ackland + Thetford)

Migration of crowdion

Barrier to rotation $\langle 111 \rangle$ to $\langle 11\bar{1} \rangle$
via $\langle 110 \rangle = 0.72 \text{ eV}$

Barrier to 1-d migration
via $\langle 111 \rangle$ dumbbell = 0.03 eV

CONCLUSIONS

1. Atomistic simulation with empirical N-body (embedded atom) potentials is more realistic than with pairwise potentials and not significantly more expensive BvT
2. For structures and strain fields more quantum mechanical models are required - 1000x more computation. For 0.1 eV accuracy, we need fully self-consistent calculations (LDF).
3. Better ask for mechanisms and plausible structures rather than precise numbers.

Theoretical Modelling of Defects and Embrittlement

M. W. Finnis

REFERENCES

LECTURE 1

1. M G Nicholas and C F Old, J. Mater. Sci. 14 (1979), 1-18 - This is a major review of liquid metal embrittlement (LME) with 171 references.
2. M W Finnis, in 'Defects and Fracture', eds. Sih and Zorski, Martinus Nijhoff, The Hague, (1982). - A theoretical approach to LME.
3. B L Eyre, B C Edwards and J M Titchmarsh, AERE - R 10343 (1982). The Physical Metallurgy of Reversible Temper Embrittlement (79 references).
4. Cottrell A H, Transactions of the Metallurgical Society of AIME 212 (1958) 192-203. A classic paper on the brittle-ductile transition; the approach is to compare the criteria for yield with the criteria for propagation of a wedge crack ∇ .
5. J. R. Rice and R Thomson, Phil. Mag. 29 (1974) 73. Another classic paper, discussing the competition between cleavage and blunting of a crack. See also the article by Sinclair and Finnis & in 'Atomistics of Fracture' referred to on my abstract.

LECTURE II

See also refs. on my abstract.

1. M. Hashimoto, Y. Ishida, R. Yamamoto and M. Doyama, *Acta Metall.* 32 (1984) 1.
2. M. Hashimoto, Y. Ishida, S. Wakayama, R. Yamamoto, M. Doyama and T. Fujiwara, *Acta Metall.* 32 (1984) 13.
In the above articles the TB model is used to discuss electronic changes at an impurity (P, B) in a grain boundary in Fe, of relevance to embrittlement. The atomic positions were relaxed with Morse potentials.
3. V. Heine, R. Haydock, D. W. Bullett and M. J. Kelly, *Solid State Physics* 35 (1980) eds Ehrenreich, Seitz and Turnbull, Academic Press, NY. Contains articles on the TB model and the recursion method.
4. D. J. Chadi, *J. Vac. Sci. Technol.* 16 (1979) 1290.
5. D. J. Chadi, *Phys. Rev. B* 19 (1979) 2074.
6. K. Maeda and A. Sato, *Phil. Mag. A* 44 (1981) 799.
7. K. Maeda, R. Yamamoto and M. Doyama, *J. Phys. F*, 13 (1983) 1407.
8. B. Legrand, *Phil. Mag. A* 52, (1985), 83.
9. G. Treglia, M. C. Desjonqueres and D. Spanjaard, *J. Phys. C* 16 (1983) 2407.
10. G. Treglia, F. Ducastelle and D. Spanjaard, *J. Physique* 41 (1980) 281.
11. I. Terakura, K. Terakura and N. Hamada, *Surf. Sci.* 111 (1981) 479.
12. K. Maeda-Jindo, N. Hamada and K. Terakura, *J. Phys. C* 17 (1984) 1271.
13. G. Moraitis, B. Stuppel and F. Gautier, *Phil. Mag. B* 52 (1985), 971.
14. M. W. Finnis and D. G. Pettifor, in 'The Recursion Method and its Applications', eds D. G. Pettifor and D. L. Weaire, Springer Verlag, Berlin (1985).
15. D. G. Pettifor and R. Podloucky, *J. Phys. C* 19 (1986), 315.
16. J. C. Slater and G. F. Koster, *Phys. Rev.* 94 (1954) 149.
17. C. A. Coulson, *Proc. Roy. Soc.* 169 A, (1939), 413.
18. W. G. Penney, *Proc. Roy. Soc.* 158 A (1937), 306.
19. G. Allan, *Ann. Phys.* 5 (1970) 169.
20. J. E. Lennard-Jones, *Proc. Roy. Soc.* 158 A, (1937), 28.
21. J. Harris, *Phys. Rev. B* 31 (1985) 1770.
22. A. P. Sutton, M. W. Finnis, D. G. Pettifor and Y. Ohta
 ↳ The TBB model and the vacancy problem. Preprints on request.
23. Y. Ohta, M. W. Finnis, A. P. Sutton and D. G. Pettifor

LECTURE III

1. The article by Heine in Lecture II ref 3 gives an account of the $\sqrt{z}$ dependence of site energy in the μ_2 approximation.
 2. C M Sayers, Phil Mag B 50, (1984), 635.
This gives an analytic model in the μ_2 approximation which explains the effect of hydrogen on bonds.
The bond energy varies as $\sqrt{z}/z = z^{-1/2}$, so is weakened by the addition of a metal-H bond.
 3. M S Daw and M I Baskes, Phys. Rev. Lett. 50 (1983) 1285.
 4. M S Daw and M I Baskes, Phys. Rev. B. 29 (1984) 6443.
 5. S M Foiles, Phys. Rev. B 32 (1985) 7685
 6. M W Finnis and J E Sinclair, Phil. Mag. A 50 (1984) 45.
- Erratum, Phil. Mag. A 53 (1986) 161.
 7. C L Matthai and D J Bacon, Phil. Mag A 52, (1985), 1.
- Refs. 3 - 7 are versions of the embedded atom method with applications.

SUPERCOMPUTERS

Notes on an ad hoc lecture
by M. W. FINNIS

SUPERCOMPUTERS

$10^2 - 10^3$ MFLOPS

CRAY-2 — 1,200 MFLOPS
256 Mwords (4 PROCESSORS)

Machines:

USA: CRAY, ETA
(IBM 3090 & CRAY-1)

JAPAN: NEC (Osaka)
FUJITSU (Nagoya)
HITACHI (Tokyo)

Fastest single processor = NEC

CRAY-2 noted for its very large memory, 256MB

Farid Abraham, IBM:

500 processors

To achieve more than another $\times 10$ in speed
will probably require parallel processing, which
is developing rapidly.

(61)

HEIRARCHY OF MODELS

1. ENGINEERING MODELS (e.g. hot pressing)

$$\dot{\epsilon} = f(\sigma, T)$$

↓↑

finite elements
- 3D calculations need
more computer power

2. DERIVATION OF CONSTITUTIVE LAWS

- MICROSTRUCTURE MODELS

$$\dot{\epsilon} = f(\rho, T, d, \sigma, c)$$

↓↑

3. ATOMISTIC MODELS (e.g. molecular dynamics,
monte carlo)

$$E^s, E^m, \sigma_u,$$

↓↑

cascade effects, diffusion mechanisms

4. ELECTRONIC MODELS (e.g. LDF model)

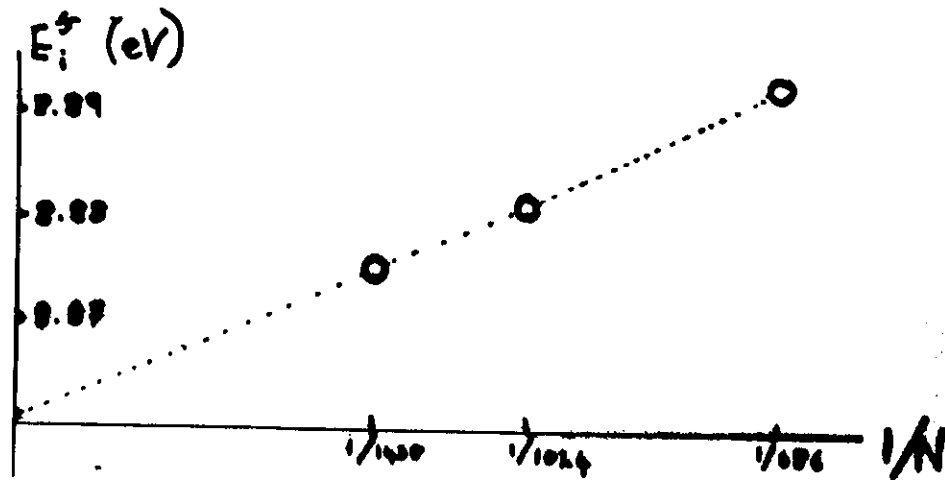
$$E \approx F[\rho]$$

$$\text{Solve } H\psi = \epsilon_n \psi$$

PROGRESS LIMITED BY : 1) MODEL
(BOTTLENECKS) 2) ALGORITHMS
3) COMPUTERS

(62)

E_i^* versus Number of relaxed atoms
 (W potential, FS + core
 Ackland + Thetford, TR1194, 1986)



1024 atoms = $8a \times 8a \times 8a$

This slide, which I had no time to show, indicates the size of block needed in a molecular dynamics calculation to get a converged interstitial formation energy. The block has flexible periodic boundary conditions. (63)

Bob Averbach & Roy Benedek (Argonne)

CASCADE SIMULATION

200,000 atoms ($36a \times 36a \times 36a$)

Cu, Gibson-II

2000 timesteps $\equiv 10^{-11}$ sec

5 hours on 1 processor of CRAY-2
 (cost at Harwell prices £ 1.5k)
 40 Mwords.

Results

A few interstitials, ~~with~~
 f.c.s.

Vacancies left as spike
 solidifies epitaxially.

These are my notes from a talk
 given by Averbach in Feb 1987.

(64)

