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EXPERIMENTAL WORKSHOP ON
HIGH TEMPERATURE SUPERCONDUCTORS
(11 - 22 April 1988)

NEW HIGH T_c SUPERCONDUCTING OXIDES WITHOUT RARE EARTHS

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These are preliminary lecture notes, intended only for distribution to participants.

New high T_c superconducting oxides
without rare earths

Bi - Sr - Ca - Cu - O superconducting oxides:

based on the Bi-Sr-Cu-O system

(Michel & Raveau) — $\text{Bi}_2\text{Sr}_2\text{Cu}_2\text{O}_{7.5}$

105 K sharp drop in resistivity & 75 K zero resistivity (Maeda)

120 K sharp step and zero resistivity at 80 K (Chu)

103 K diamagnetic susceptibility (Zhao)
and 84 K zero resistance (Xie)

Status:

1. 110 K - 120 K sharp drop resistivity & A.C χ anomalous.
2. 85 K - 75 K zero resistivity & diamagnetic anomalous.
3. Two superconducting phases:
tetragonal (or pseudo-tetragonal)
face centered unit cell with modulation along $\vec{b}$ axes.
— 85 K phase with composition
2212 most single phase.

(1)

tetragonal unit cell without modulation.

— 110 K phase 2212 or 2213
 $c = 30.8 \text{ \AA}$ or $c = 26.8$ (2223)

4. Various structure models related to Aurivillius structure: { Face centered lattice, Body centered lattice
orthorhombic, tetragonal.

What is the 85 K superconducting phase.

Chemical formula	Symmetry	Lattice parameters
$\text{BiSrCaCu}_2\text{O}_{5.5}$ (Maeda)	—	—
$\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_9$ (Hazen & Chu)	A-centered	$5.42 \times 5.44 \times 30.78$
$\text{Bi}_{2.2}\text{Sr}_2\text{Ca}_{0.8}\text{Cu}_2\text{O}_{8+x}$ (Sunshine)	F m m m	$5.42 \times 27.2 \times 30.8$
$\text{Bi}_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_{16}$ (Tarascon)	I 4/m m m	$3.817 \times 3.817 \times 30.6$
$\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+x}$ (Argonne Lab.)	F 4/m m m	$5.41 \times 5.41 \times 30.8$
$\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+x}$ (Institute of Phys) Sics	I 4/m m m	$3.825 \times 3.825 \times 30.82$

(2)

possible impurity phases

1) * Ca_2CuO_3 type :



$$a = 12.234, b = 3.771, c = 3.257 \text{ \AA}$$

2) CaCu_2O_3 type :



$$a = 11.328, b = 12.774, c = 3.88 \text{ \AA}$$

3) SrBi_2O_3 type :



$$a = 13.329, c = 4.257$$

4) * CuO

5) SrCuO_2 : C_{mmm}

$$a = 3.56, b = 16.32, c = 3.92 \text{ \AA}$$

6) * CaCu_2O_3 : P_{mmm}

$$a = 9.85, b = 4.11, c = 3.47$$

7) * Ca_2CuO_3 : I_{mmm}

$$a = 12.23, b = 3.77, c = 3.25$$

③

③

How to prepare :

Starting materials: Bi_2O_3 , CuO , CaCO_3 , SrCO_3 .

composition: $\frac{2212}{1112}, \frac{4334}{1113}$

procedures: ^{in a wide region}

mixing & grinding thoroughly starting materials.

calcining at 850°C ^{-890°C} in air for 8 hrs.

regrinding & pressing into pellet.

sintering at 850°C for 12 hr in air.

quenching down to R.T or cooling in furnace
product: 85K phase.

keeping it at high temperature (very close, but
not above the melting point) for a day.

110K phase amount increase!

50% A.C susceptibility (Tarascon)

over 70% A.C χ (Beijing)

④

melted sample showed sandwich layers: (5)

upper, loose, black, 110 K + 85 K phase

medium, denser, black, 85 K phase

bottom, hard, grey, NO

Maeda's procedure for 1112

1. calcining powders at 800°C for 12 hrs
2. six hours at 820°C
3. six hours at 820°C
4. sintering at 840°C
5. sintering again at 860°C
6. quenching to R. T.

Characterization:

Resistance and A.C. χ { 85 K or lower
110 K or higher

identification (x-ray powder diffraction & electron microscopy)

indexing main reflections.

(5)

Body-centered lattice (I4/mmm) (6)

$$a = a_p \approx 3.8 \text{ \AA}, \quad c = 30.8 \text{ \AA}$$

Face-centered lattice (F4/mmm)

$$a = \sqrt{2} a_p \approx 5.4 \text{ \AA}, \quad c = 30.8 \text{ \AA}$$

For 110 K phase: (orthorhombic or tetragonal)

$$\text{two } c \text{ parameters } \begin{cases} 30.8 \text{ \AA} & (\text{Tarascon}) \\ 36.02 \text{ \AA} & (\text{Zhao}) \end{cases}$$

$$\rightarrow \text{composition? } \begin{cases} 2212 & \text{or } 4424 \\ 4426 & (\text{more copper}) \\ 2223 & (\text{EDK}) \end{cases}$$

Modulation: Incommensurate $\xrightarrow{\text{lock-in}}$ commensurate

incommensurate along b axes for 85 K phase

neither incommensurate nor superlattice

for 110 K phase.

Growth of single crystal: flux method.

1. synthesize the bulk sample (4334 or 1113)
2. melt the powders or bulk sample at 1100°C for 2 hrs

(6)

3. Cool down to 850°C slowly ($10\sim 15^{\circ}\text{C/hr}$) ⑦
4. hold sample at 850°C for 12 hrs or longer.
5. Cool down to R.T. in the furnace.

products: flat superconducting crystal

plate-like ($5\text{ mm} \times 2\text{ mm}$) : black, easily to cleavage along (a,b) plane, like mica.

minor: bar-like crystal: dark green.

flat 'single crystal': anisotropic resistance

behavior above $T_c = 85\text{ K}$

{ metallic in (a,b) plane (semimetal)
semiconductive in $\bar{c}$ direction

Bar-like crystal - Bismuth - poor compound

$a = 11.84 \text{ \AA}$, $b = 12.37$, $c = 19.54 \text{ \AA}$

superconductivity uncertain! Non-superconducting.

Growth of single crystal of Bi-compound: (modified procedure.

1. add the extra Bi_2O_3
1. hold molten sample 24 hr.
3. hold sample at 850°C 3 days
4. the bar-like crystals disappear.

⑦

structure of 85 K phase:

⑧

Common in all models: alternative perovskite + double Bi_2O_2 layers (Aurivillous) modulation along $\bar{b}$ axes or superlattice (?)

Cu-O planes exist, but are separated by Ca cations.

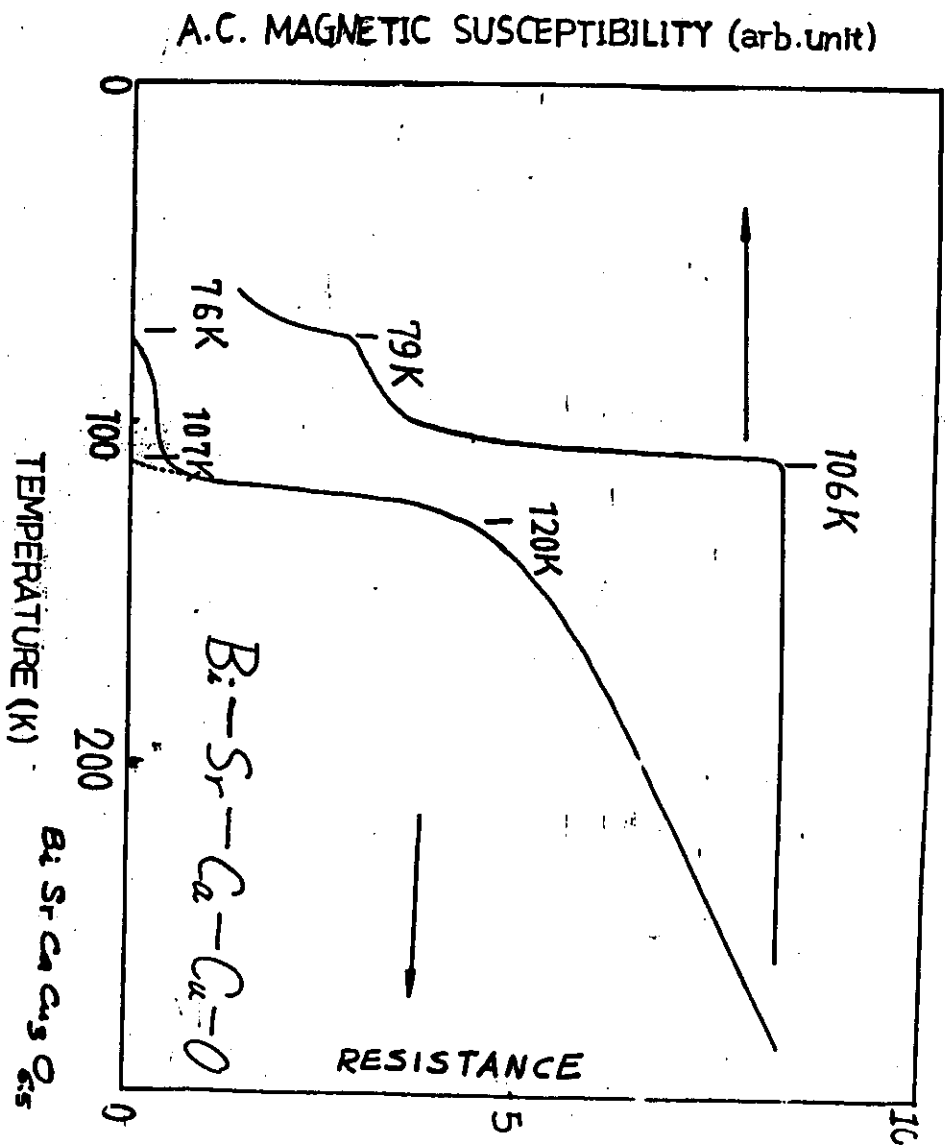
Substitution: V, Ga, Y, Pb replace Bi
→ depress T_c .

In, Cd, Sn replace Bi
→ destroy superconductivity.

Uncertain things:

1. How to isolate pure single phases of 85 K & 110 K.
2. Cu-O chain? Bi-O chain?
3. exact amount & distribution of Oxygen?
4. Valences of Cu & Bi?
mixed valence?
 $\text{Bi}^{3+} \sim 0.96 \text{ \AA}$, $\text{Bi}^{5+} \sim 0.74 \text{ \AA}$
 $\text{Cu}^{2+} \sim 0.72 \text{ \AA}$,
5. disordering of Cu^{2+} & Bi^{5+} ?
5. Ordering or disordering of Ba & Ca ions?
6. Process of molten sample → amorphous
→ Crystalline?

⑧



(10)

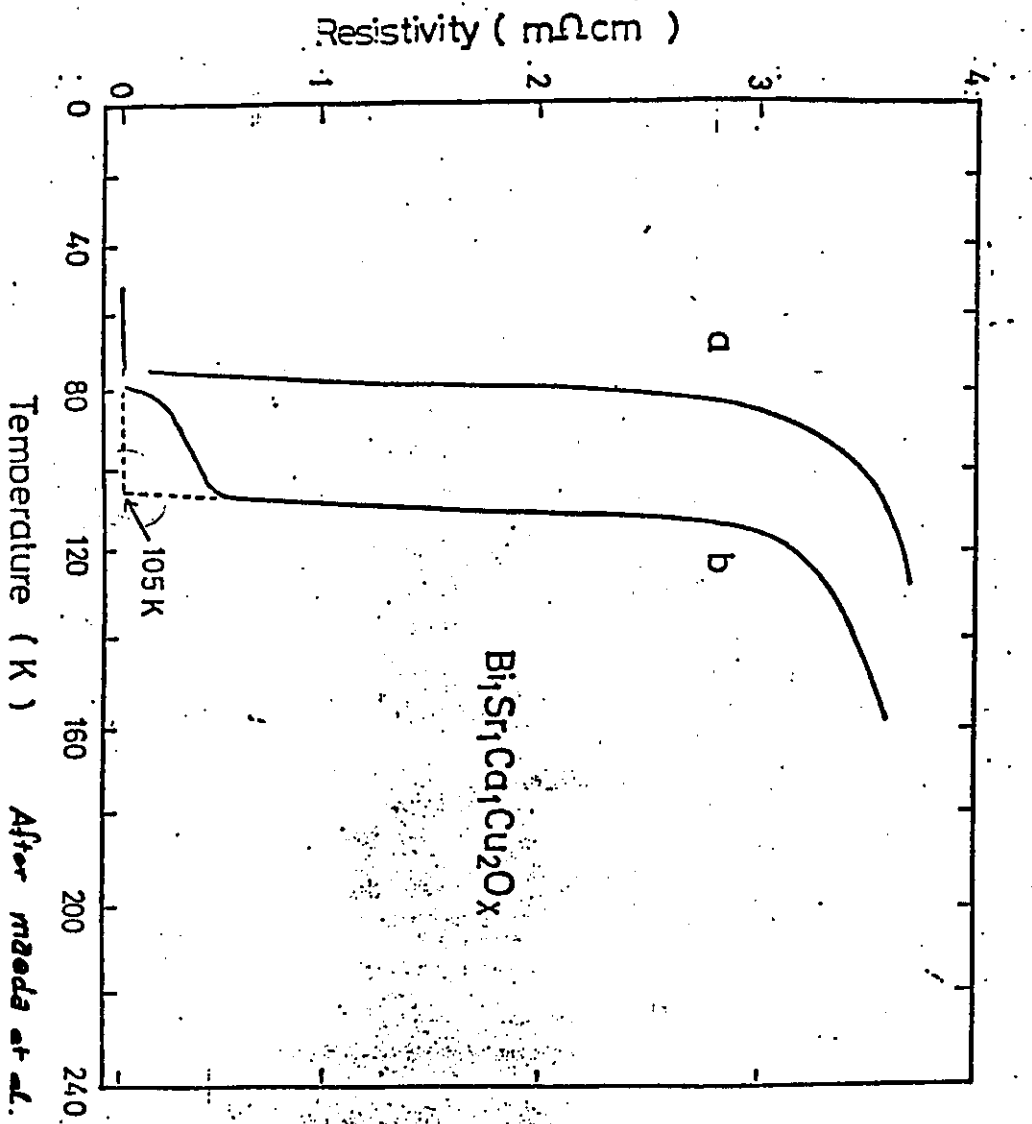
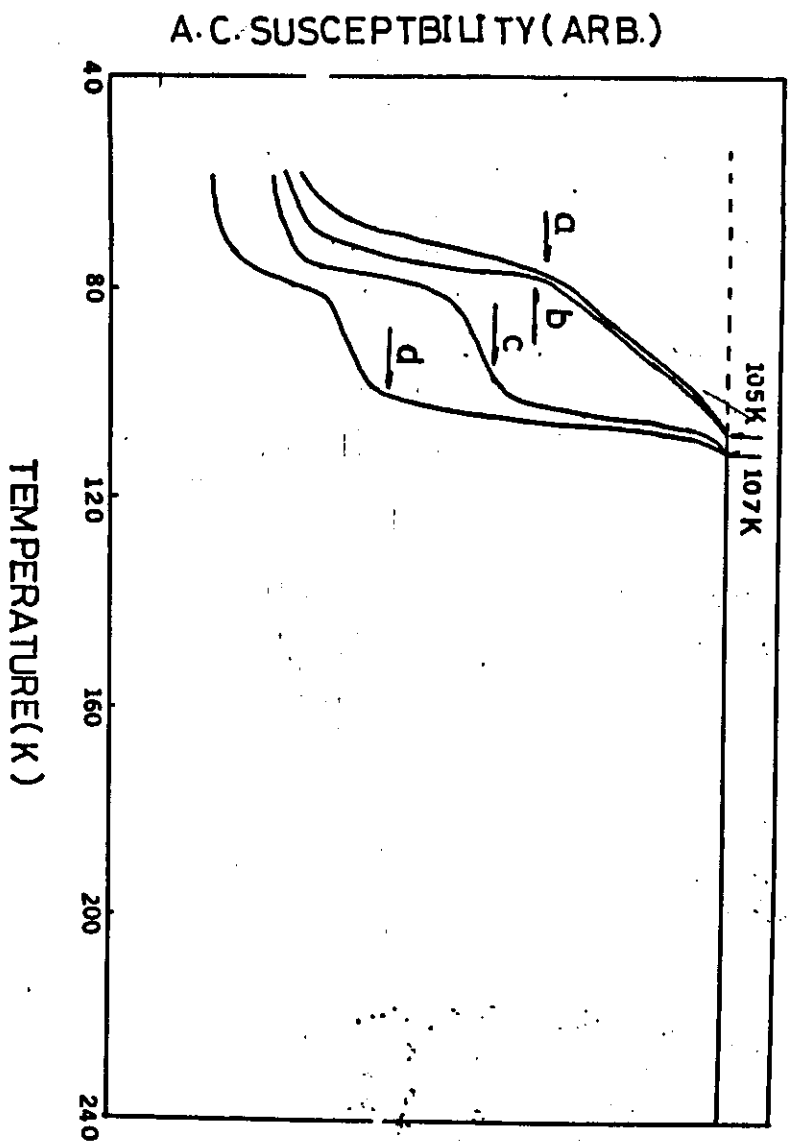


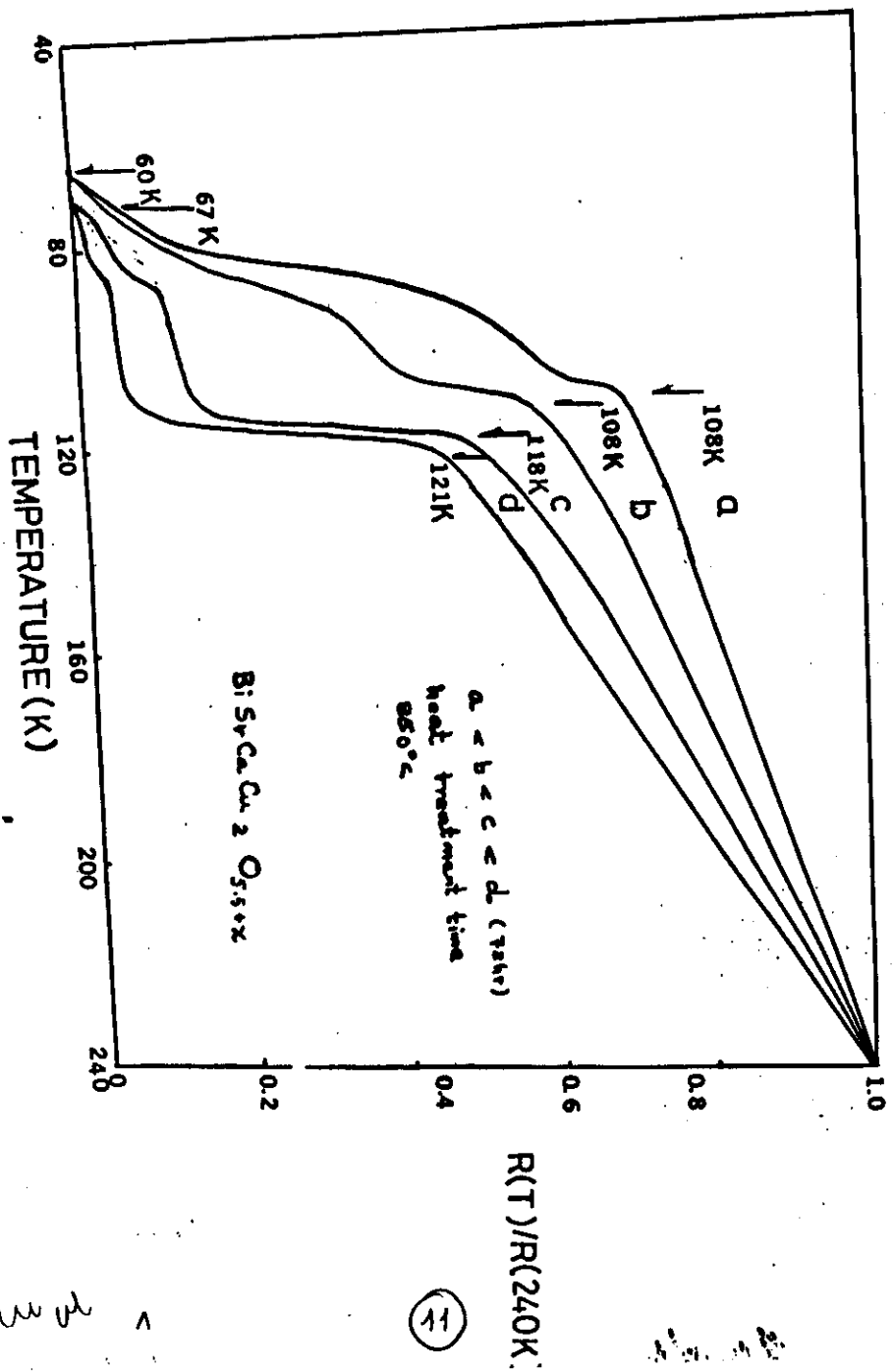
Fig. 1

(9)

After mada at d.

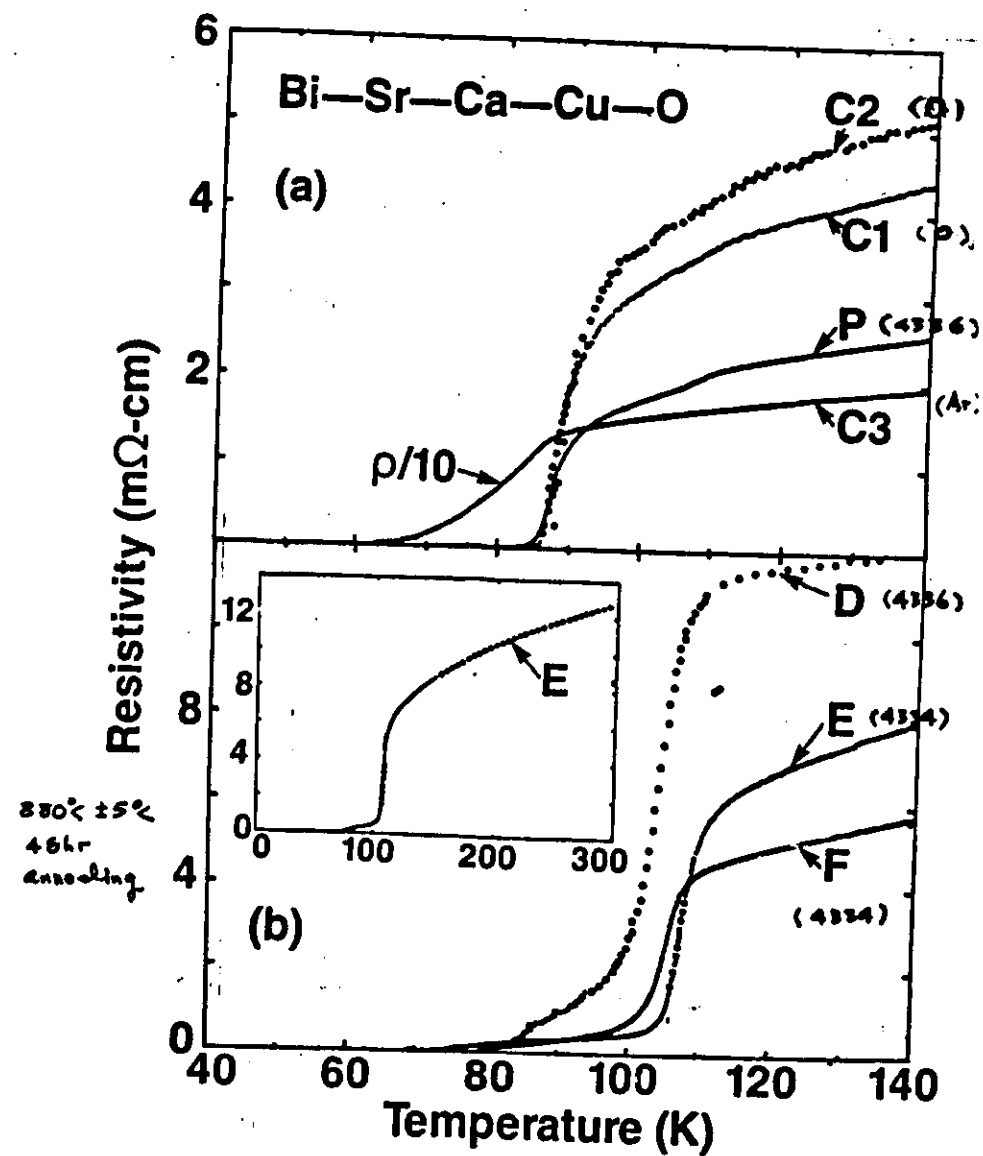


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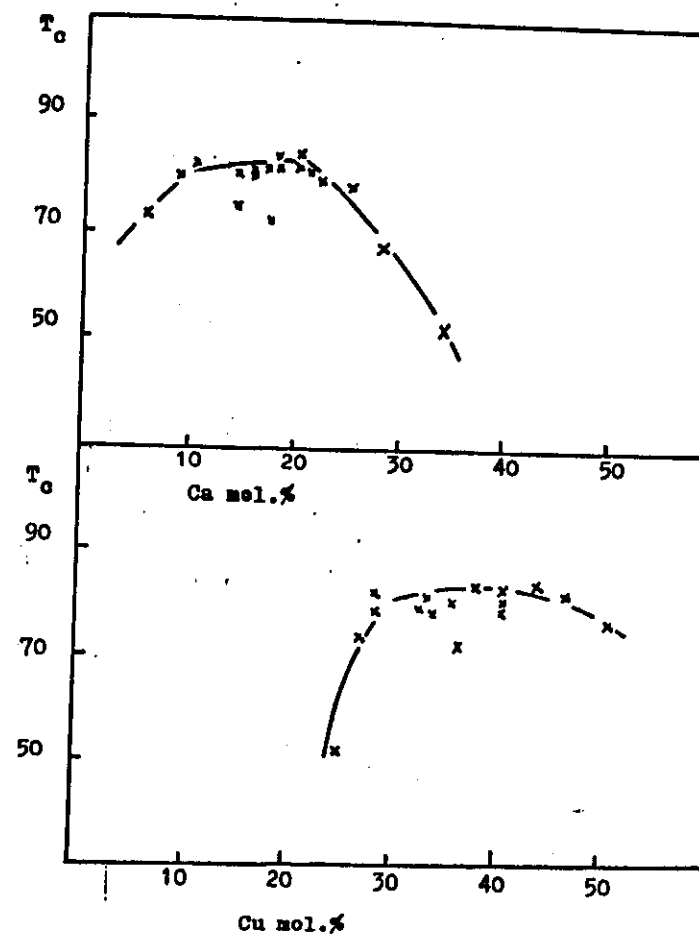
(11)

Fig. 1



After Tarascon

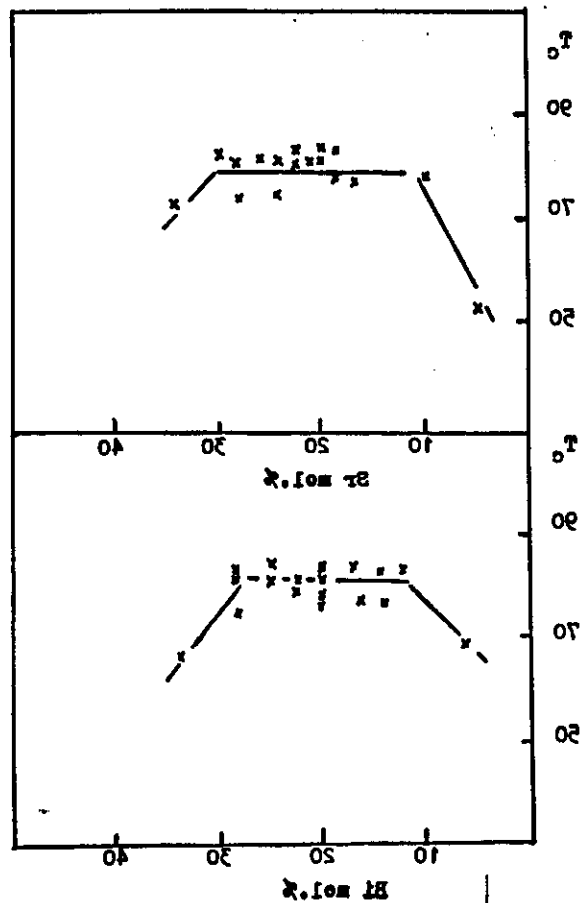
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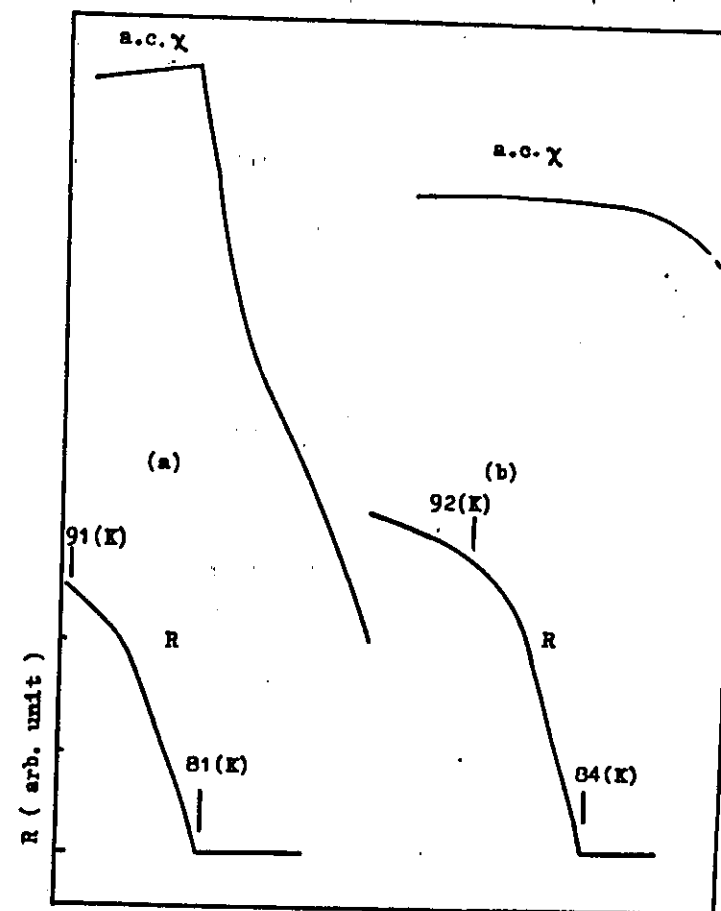
Institute of Physics .
Submitted to J. of Physics. D

(14)

Fig. 5



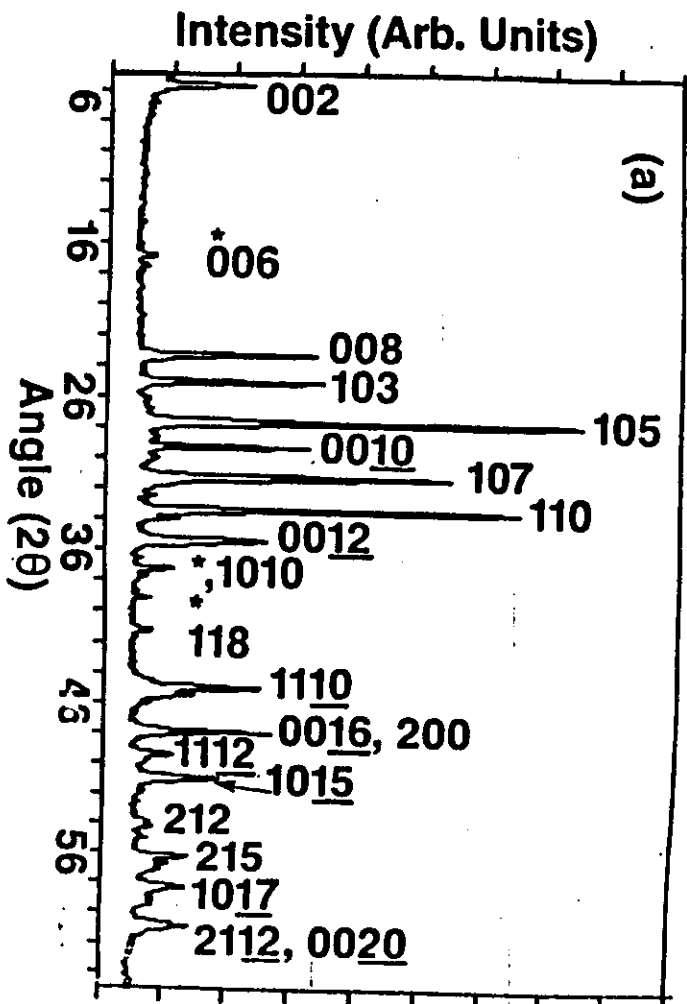
15



a) $\text{Bi}_{6.5}\text{Sr}_{6.5}\text{Ca}_{5.5}\text{Cu}_{10.5}\text{O}_{32.5+\delta}$

b) $\text{Bi}_{5.5}\text{Sr}_{5.5}\text{Ca}_{4.5}\text{Cu}_9\text{O}_{25.5+\delta}$

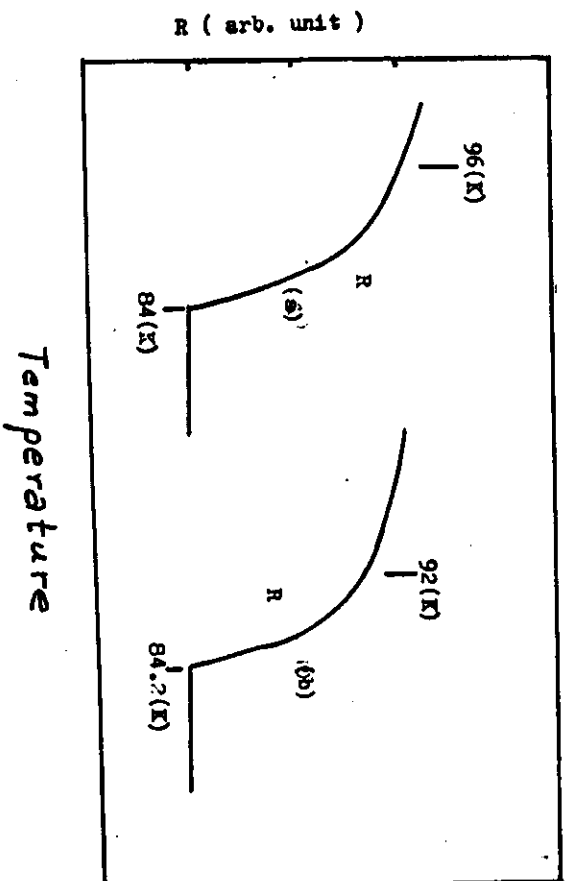
16



Indexing with tetragonal body centered unit
cell $a = 3.817$, $c = 38.6$

- a) $\text{Bi}_{1.7}\text{Sr}_{2.0}\text{Ca}_{3.0}\text{Cu}_{4.3}\text{O}_x$
b) $\text{Bi}_{1.4}\text{Sr}_{1.3}\text{Ca}_{3.0}\text{Cu}_{4.0}\text{O}_{1.8}$

Fig. 7



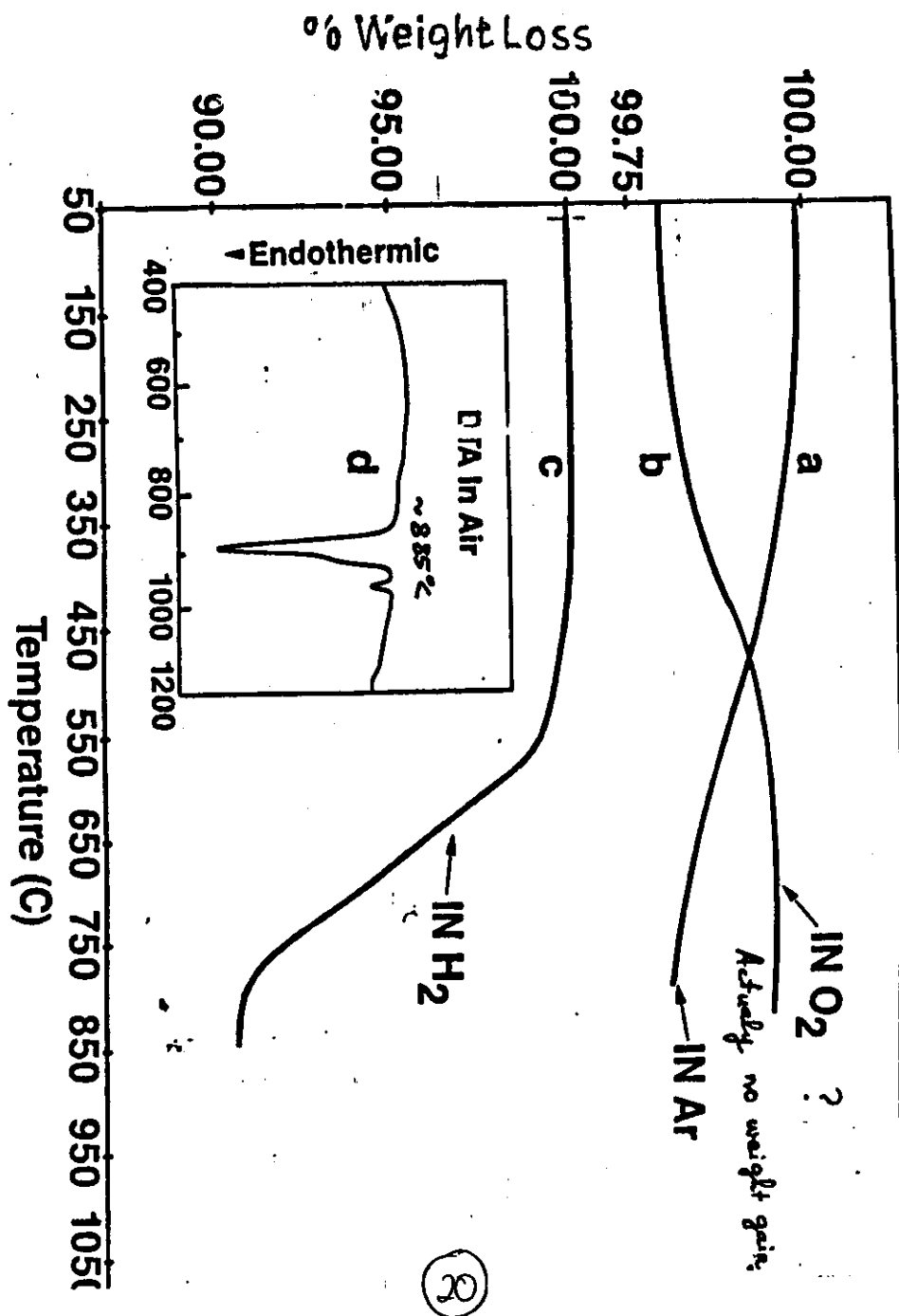
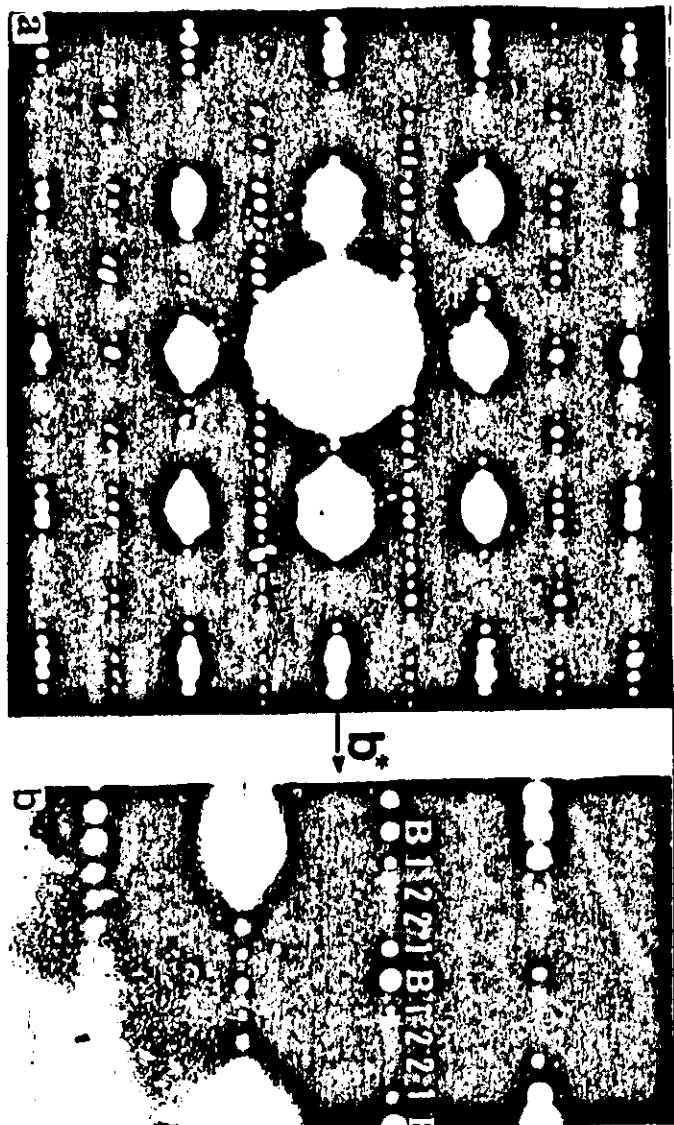


Figure B

X-ray powder diffraction pattern of the 4334 material is shown over the 2θ range from 5 to 65°. The pattern for a polycrystalline powder is shown in (a) with the Miller indices noted above each peak. Shown in (b) is the pattern for highly-c-oriented multicrystalline composite (sample C). Note that only the (00l) reflections with l even are observed. Peaks with asterisks can be indexed in a unit cell of $a=5.4$ and $c=30.6$ Å.

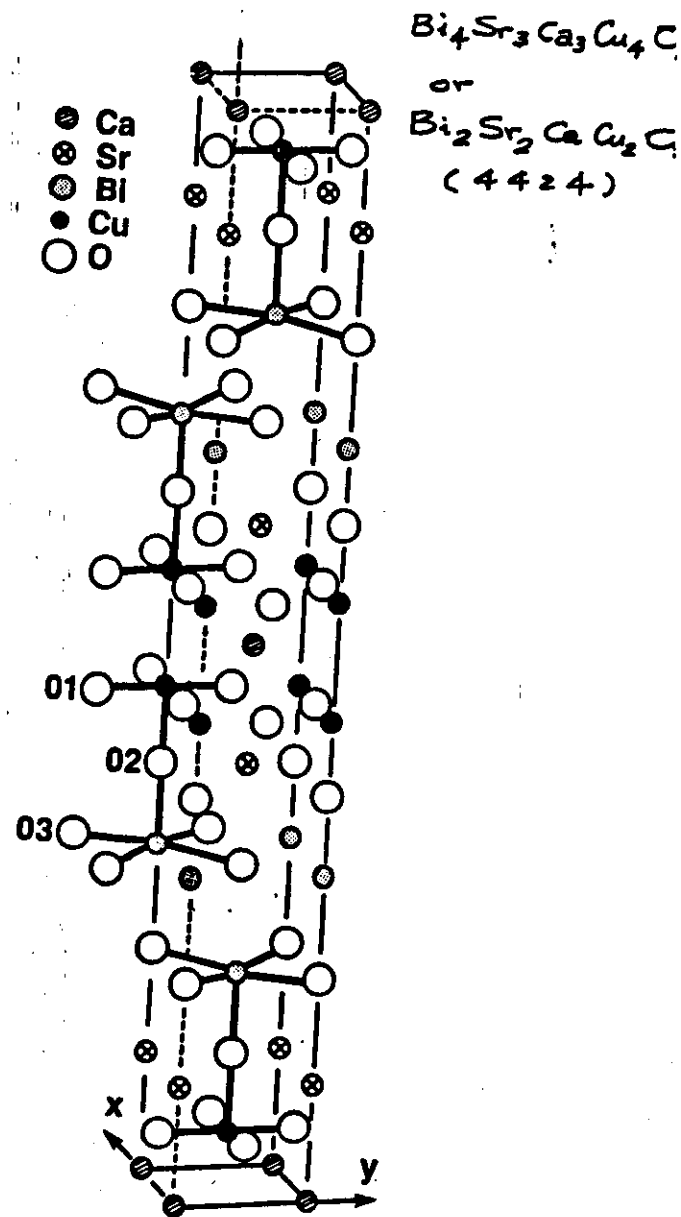
modulation structure along $\vec{b}$ axis
 reported in Inthakun . submitted to J. Phys. C.
 (Ref. Li)



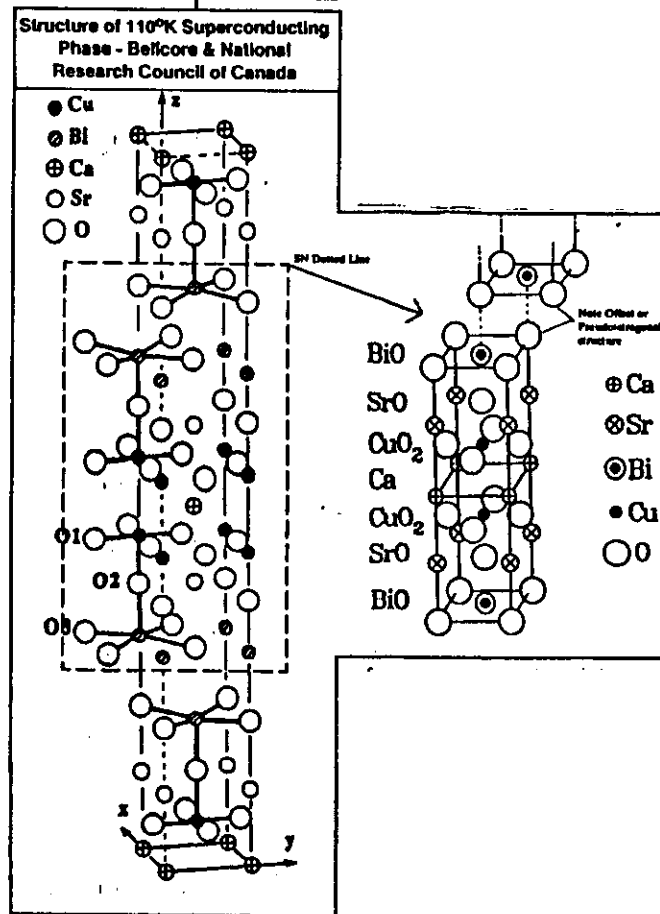
(21)

S.G $I4/mmm$
 $a = 3.817 \text{ \AA}$
 $c = 30.6 \text{ \AA}$
 $R_f = 0.16$
 buckled crystal

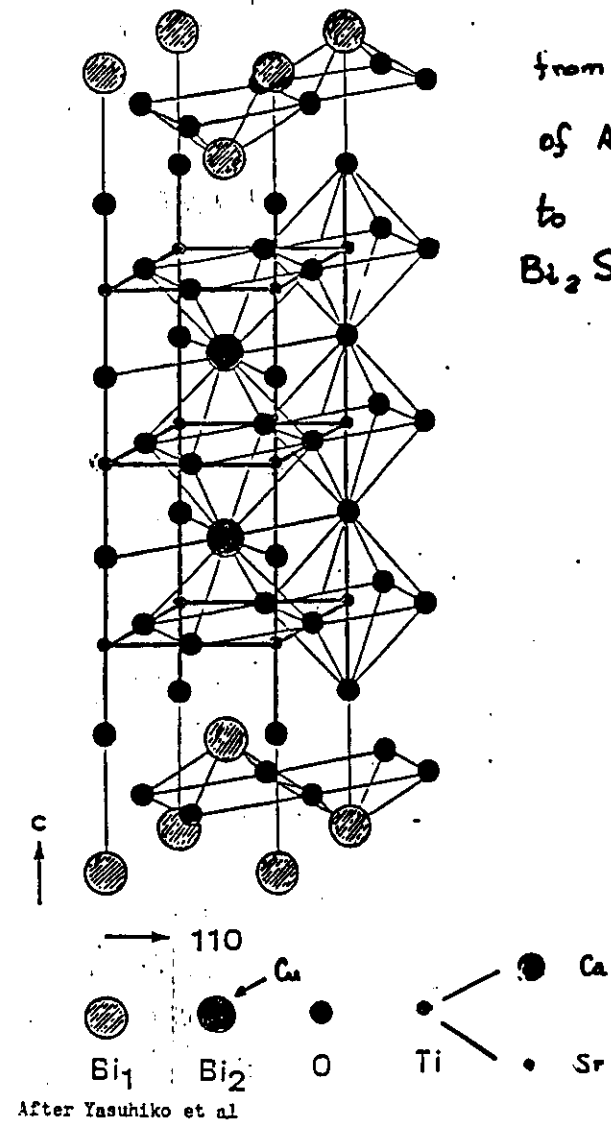
After Tarascon



(22)



23



24



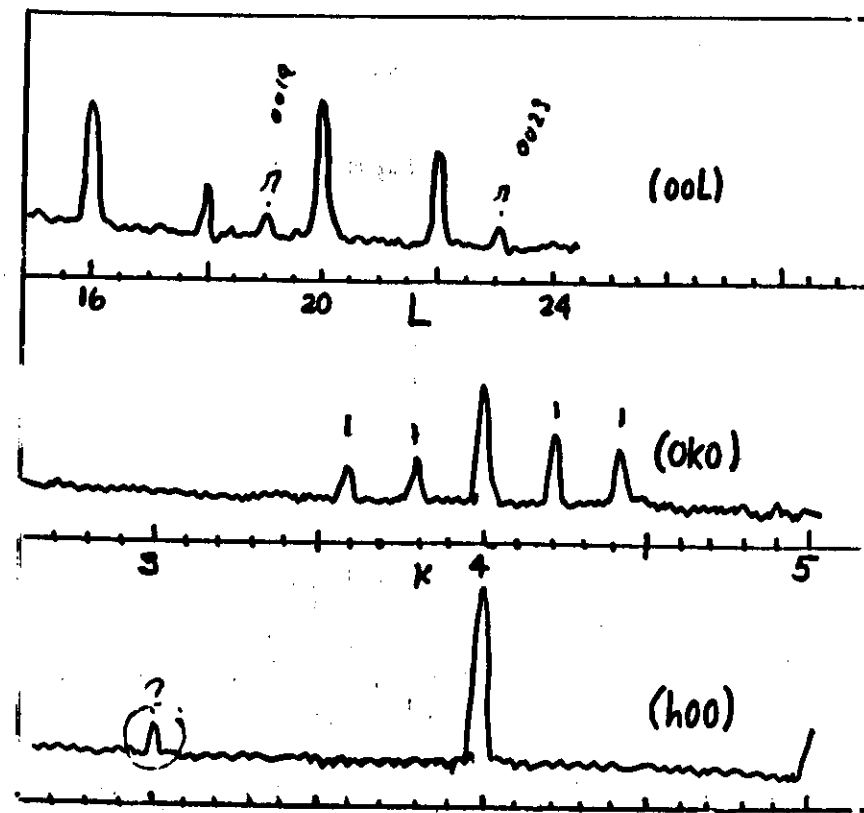
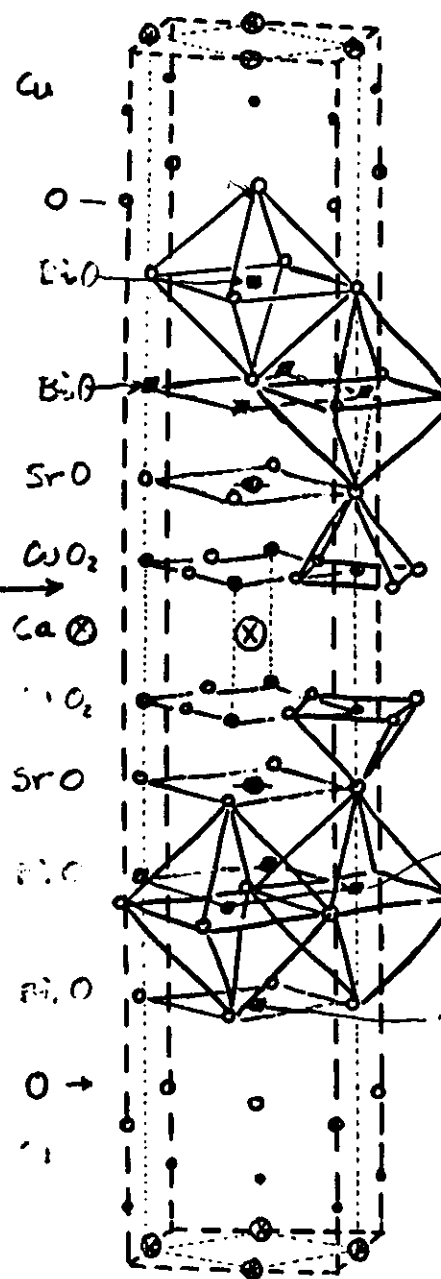
- 1) $5.414 \times 5.488 \times 30.89$
- 2) $Fmmm$, $Z=4$
- 3) Incommensurate $\vec{b}$
- 4) Bi-O octahedra (6)
Cu-O pyramid (5)
- 5) double Bi_2O_7 layer
+ perovskite
- 6) Face centered lattice
(green dot line)
- 7) Body centered Unit cell
(red dot line)

* weak peaks
violate $Fmmm$
symmetry

$R=0.142$

After Sunshine
Bell Lab.

25



Weak peaks that violate $Fmmm$
symmetry.

26

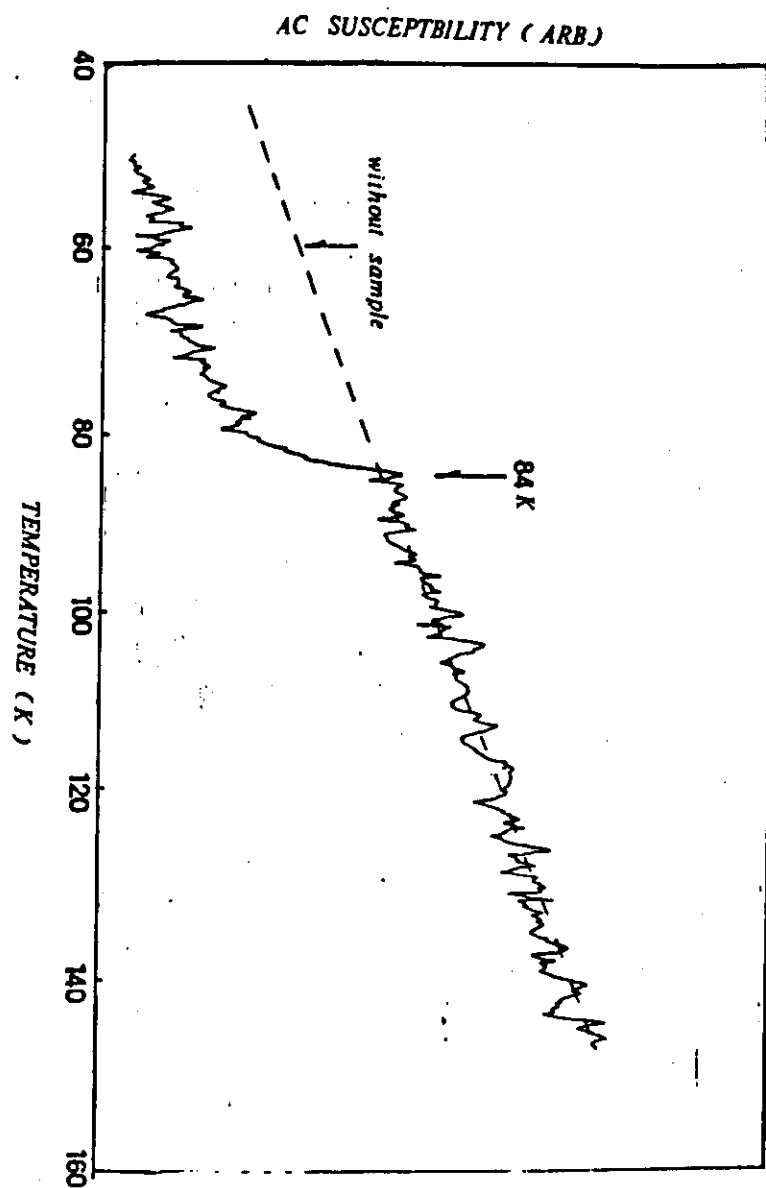
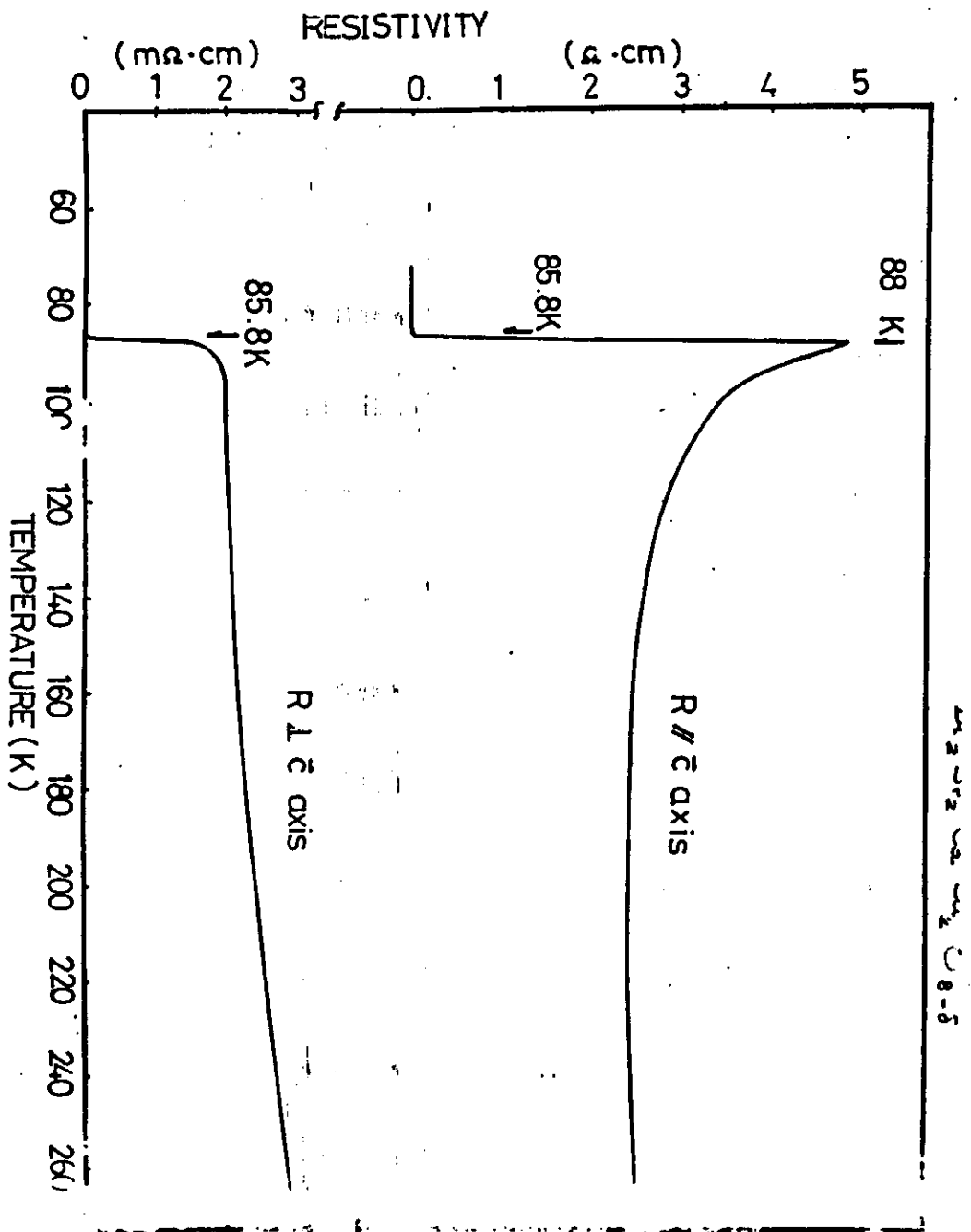
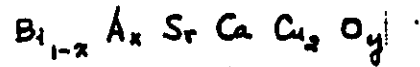
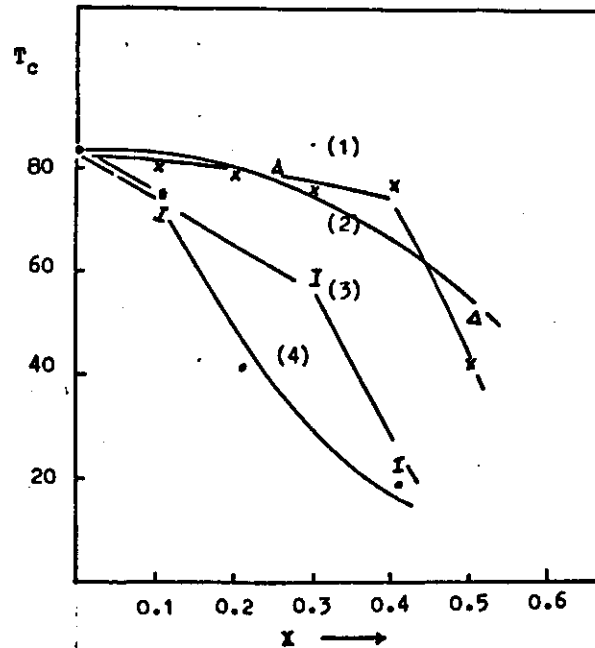


Fig. 4



- (1) V
- (2) Ga
- (3) Y
- (4) Pb



New high T_c superconductors above 120 K
in Tl - Ba - Ca - Cu - O system

Short History : Hermman & Sheng

Onset = 120 K, $T_{zero} = 106$ K

T_c to 114 K (Zhang, Xie)
120 K (Zhang, Xie)
125 K (IBM)

2212 ~ 100 K phase

2213 or 2223 120 K phase.

WARNING !

Thallium - extreme toxic !

M.L.D ~ 12 mg / kg Tl, $\text{Tl}^{3+} \gg \text{Tl}^{+}$

How to prepare :

1. Tl_2O_3 . BaO . CaO . CuO . (1.1.1.3 or 4.3.3.6)
2. mixing and grinding
don't calcine powder !
3. pressing into pellets (diameter 16 mm
(5 T/cm²) thickness 2-3 mm)
4. sintering at 780°C - 800°C for 8 hrs
in air.

5. Cooling to 600°C in the furnace (1.5 hr)

6. quenching to R.T.

polish a little, the surface or thin layer to clean out.

Characterization:

$$T_c = 117-120 \text{ K}$$

strong A.C. χ anomalous.

existence region of 120 K phase.

Ti/Cu ratio more critical.

Ba/(Ca + Ba) 30% - 60%

In high pure alcohol regrounding & resintering push
 J_c 80 A/cm² → 1600 A/cm².

(77 K, 0 field)

Thermal analysis:

DTA no anomalous until decomp.

TG no obvious weight loss

C_p at 120 K phase transformation.

anomalous!

Electron microscopy (ED, HREM) &
x-ray powder diffraction:

$$a = 3.847 \text{ \AA} \text{ (or } \sqrt{2}a) \quad c = 36.02 \text{ \AA}$$

for 120 K phase

$$a = 3.847 \text{ \AA}, \quad c = 29.43 \text{ \AA}$$

for 100 K phase.

Co-existence & intergrowth of two
phases: coherent intergrowth of both phases.

Structure:

1. $\text{Th}_2\text{Ba}_2\text{CaCu}_3\text{O}_9$ related to
Aurivillous.

Similar to $\text{Bi}_4\text{Sr}_3\text{Ca}_3\text{Cu}_4\text{O}_{16}$
by adding extra Cu layers.

Behaviour in the magnetic field $\rightarrow$ weak link between the grains. $\rightarrow$ granulars ?

2. The structure of 100K phase with composition



almost as same as $\text{Bi}_2 \text{Sr}_2 \text{Ca Cu}_2 \text{O}_{8+x}$

(just for the preliminary framework)

Substitution of 3^+ valence ions.

Th is necessary for 120K & 100K phases.

Non successful to eliminate toxic !

for example: In, Cd, Sn, Hf, ...

The rough idea about the relation between the T_c &

c parameters,

considering the number of moleculars or the distance between rare earths layers or Bi (Th) layers;

not only c parameters !

The assumption needs the experiments to support it.

1. pure single phase

2. exact determination of structure: c parameter

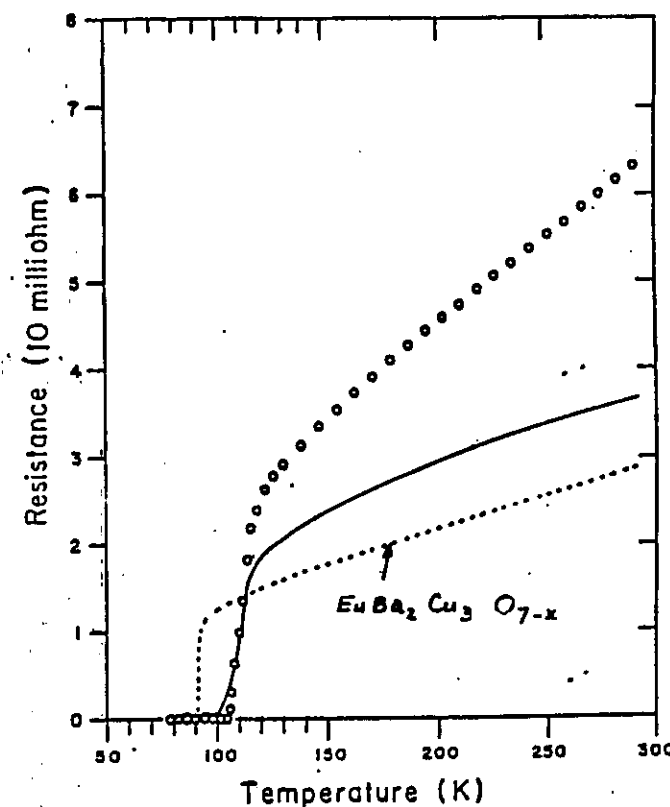
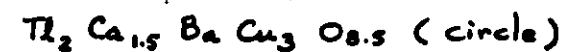
& atomic parameters

3. setup the relation between the T_c & c :

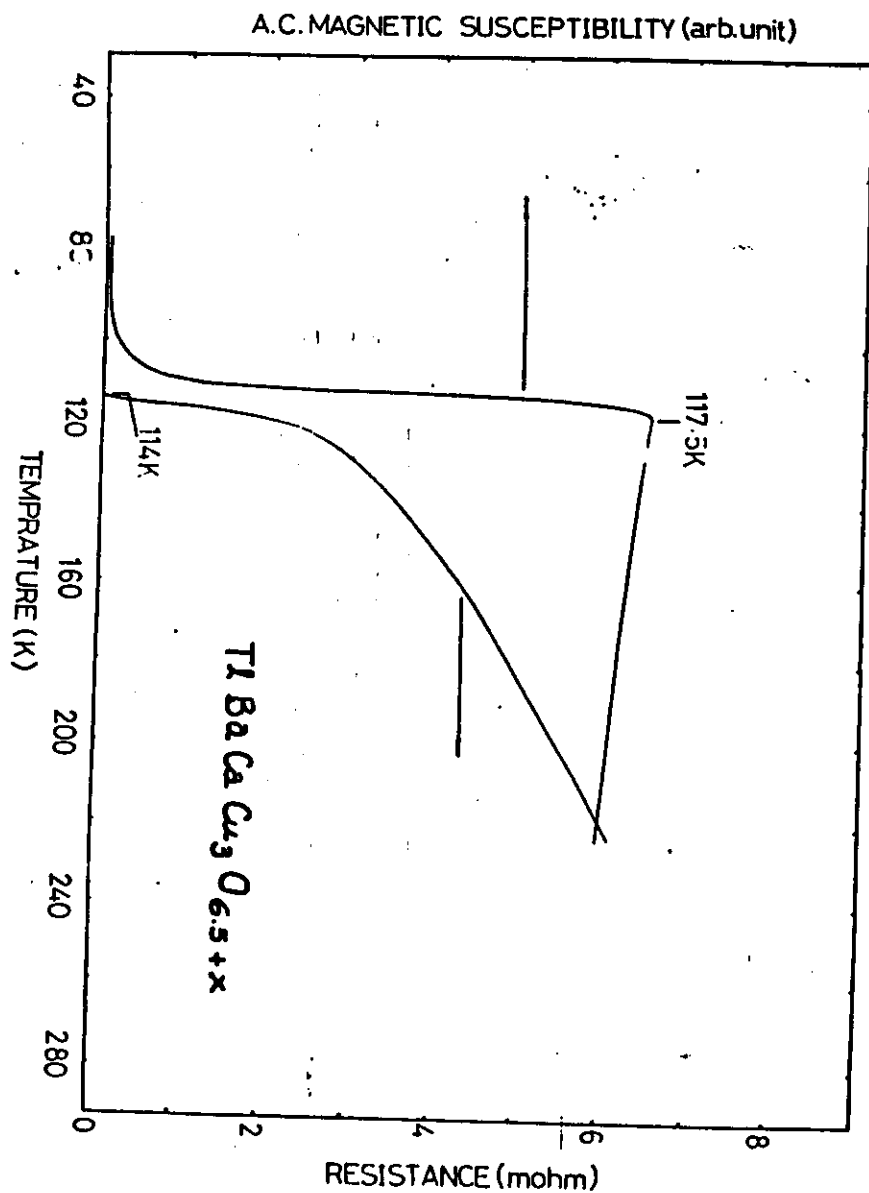
(33)

multi-phases sample

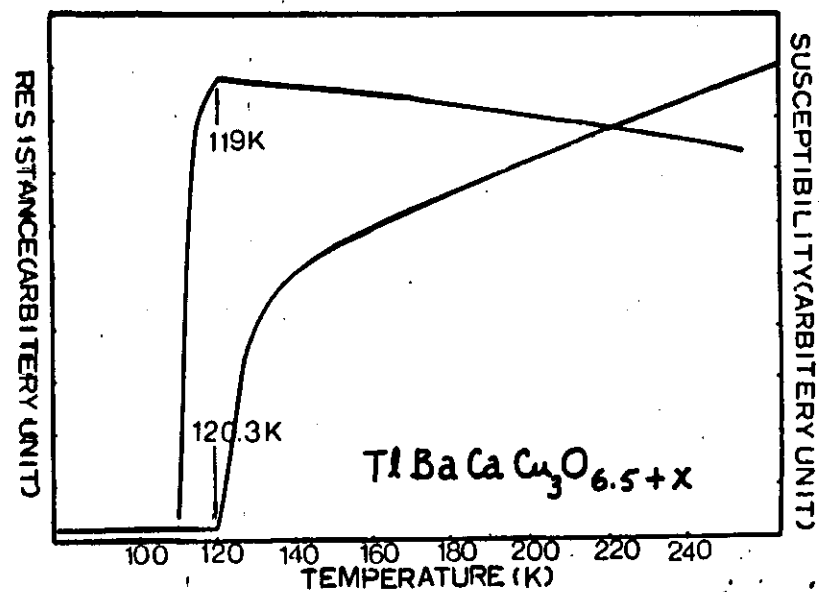
nominal composition,



(34)

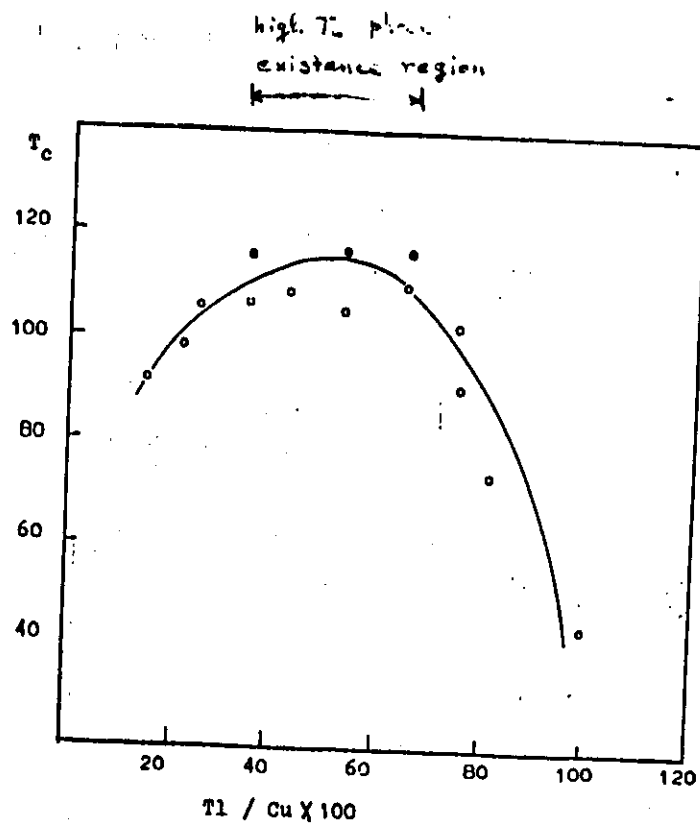


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regrinding + resintering
again!

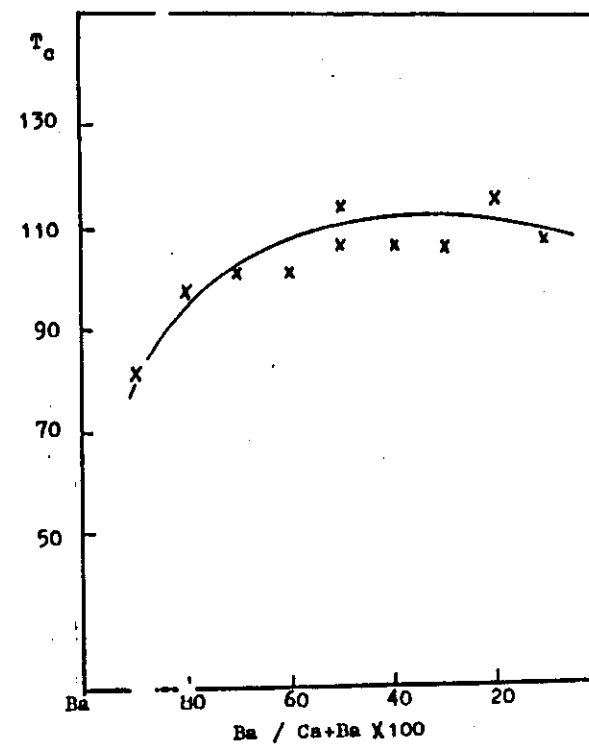
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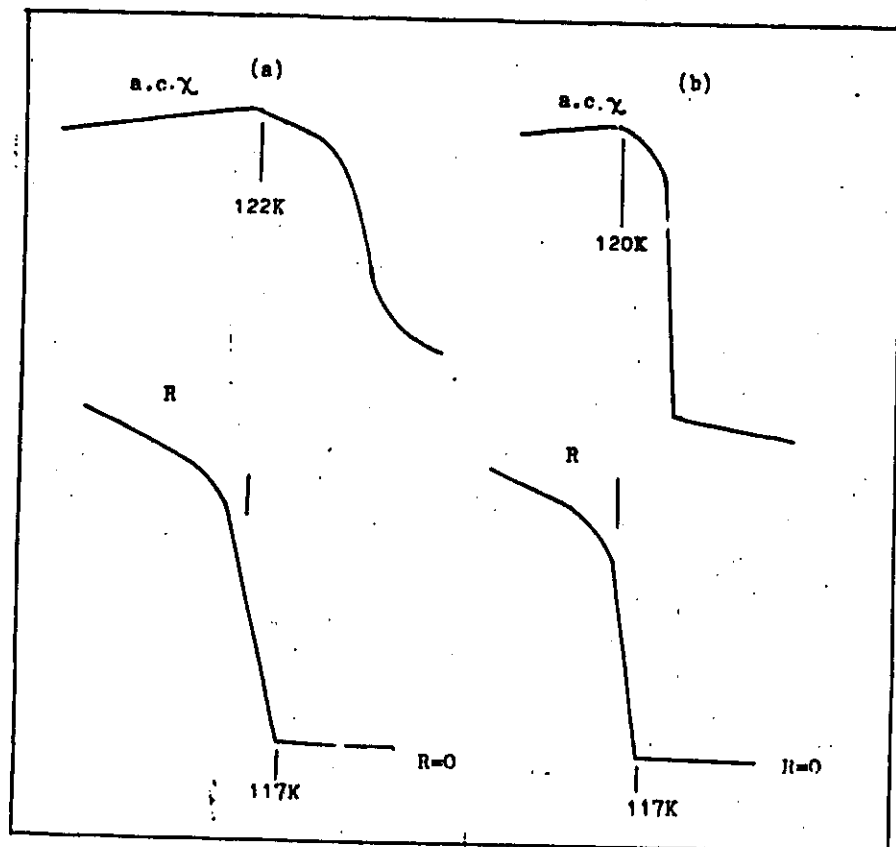
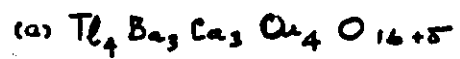
0.33
1:1

0.66
2:1

37

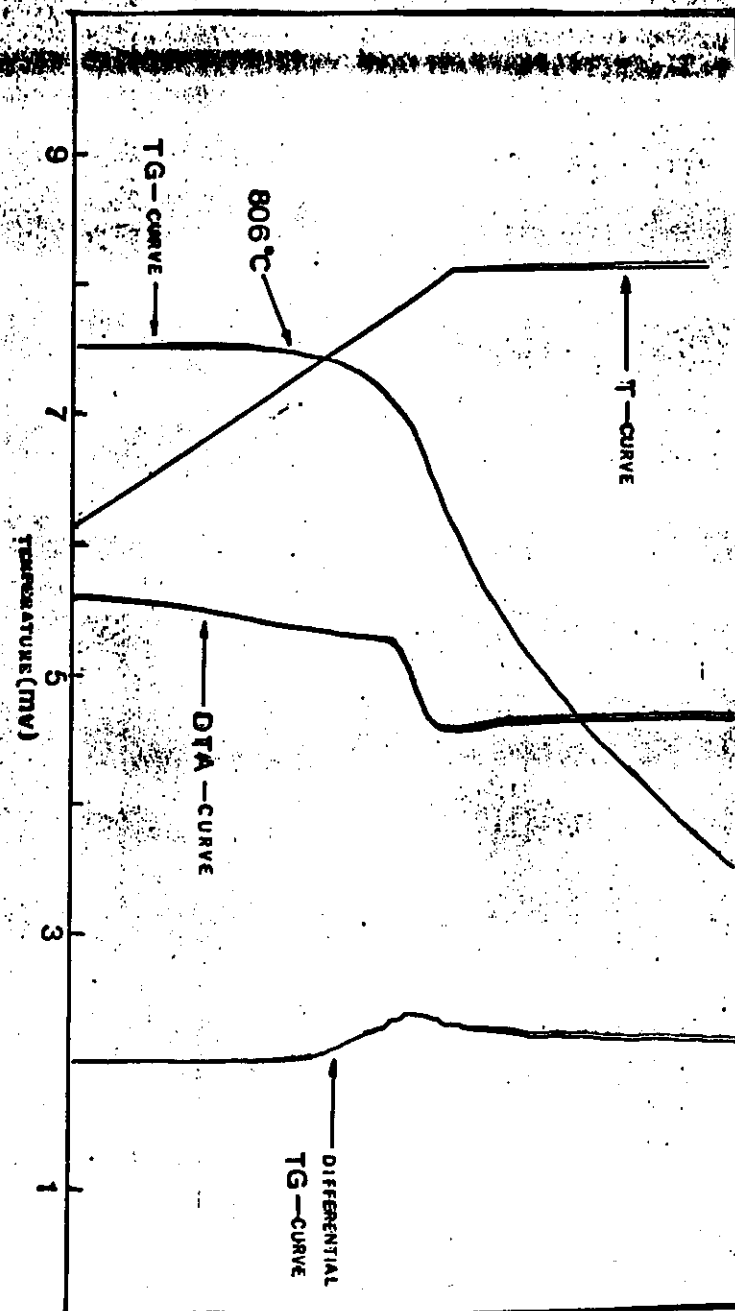


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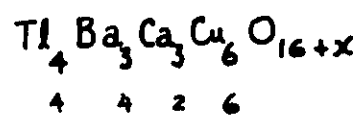
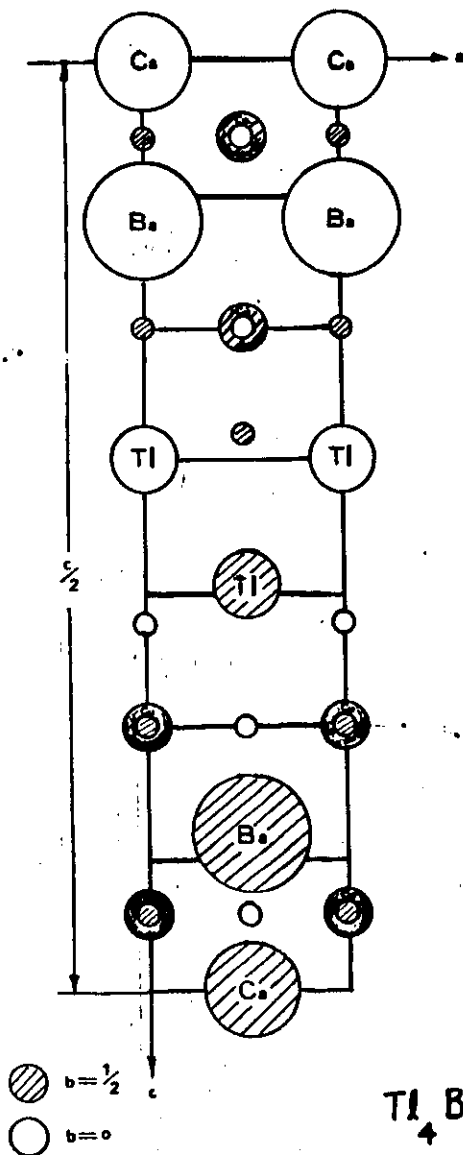
Fig. 6.



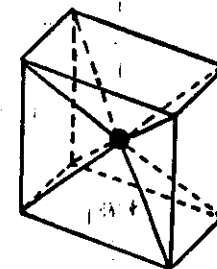
40

a part of thermal analysis in air.
weight loss ~

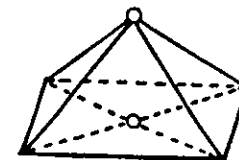
120 K phase



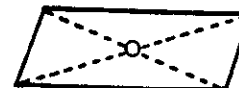
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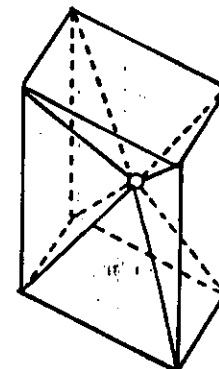
a



d_1



d_2



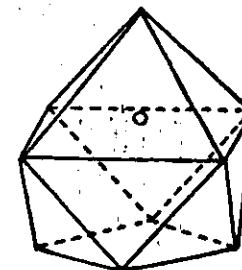
b

a: Cu - O

b: Ba - O

c: Ti - O

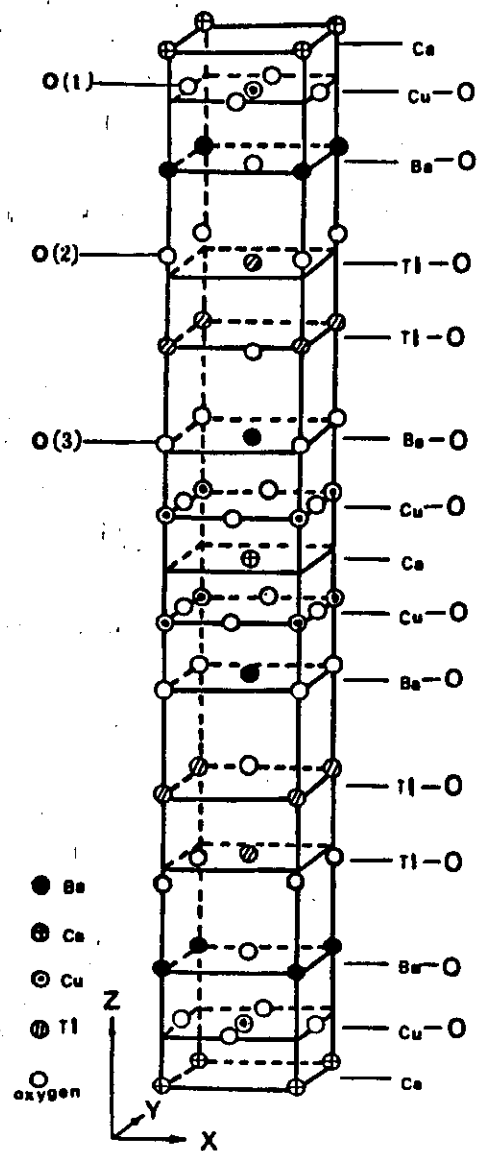
d_1 : Cu - O
 d_2 : Cu - O



c

(42)

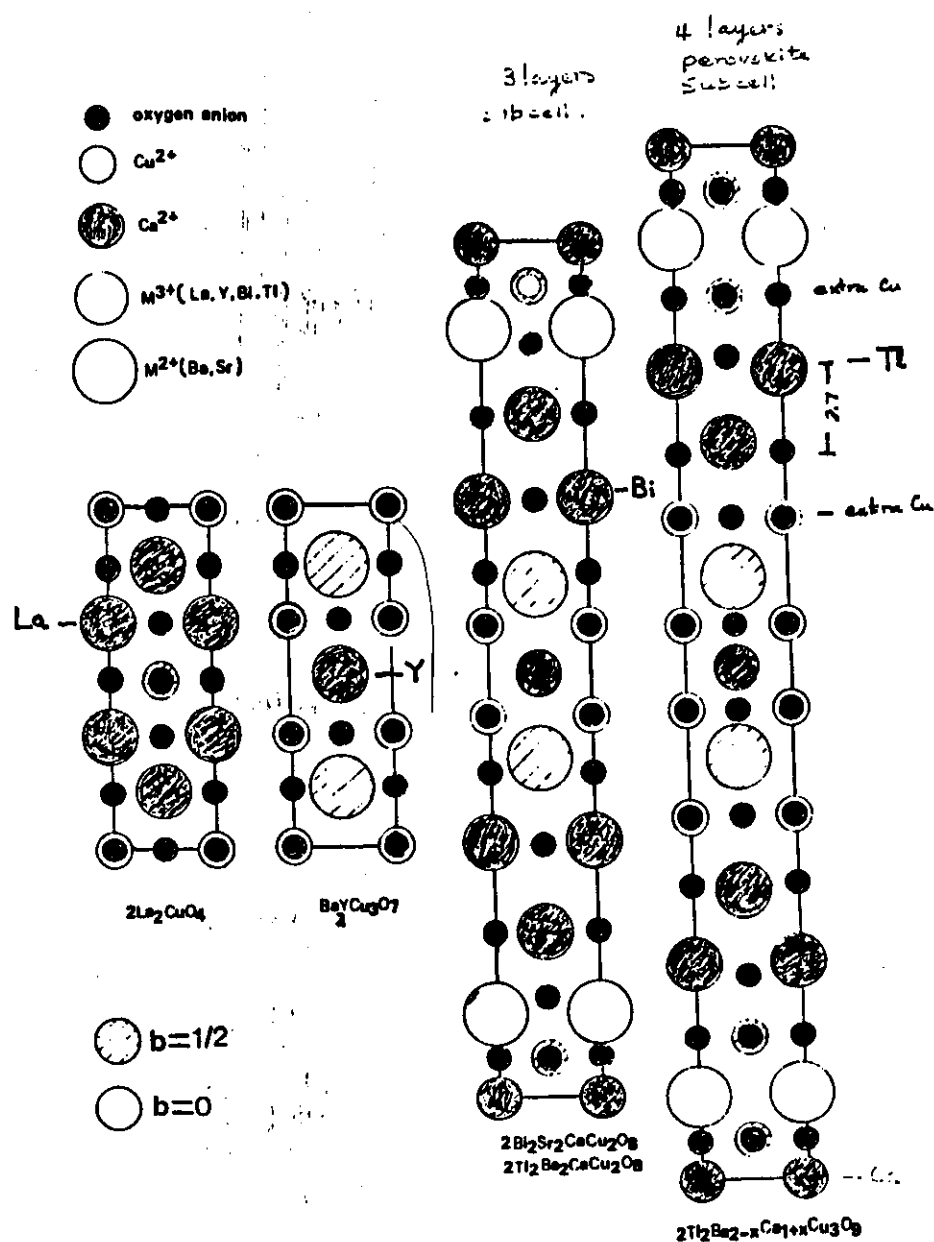
100K phase



(43)



(44)



Ionic radius

Cu^{2+}	0.72 Å	Ba^{2+}	1.34 Å
Sr^{2+}	1.12 Å	Ca^{2+}	0.99 Å
Y^{3+}	0.9 Å	La^{3+}	1.06 Å
Tl^{+1}	1.47 Å	Pb^{2+}	1.19 Å
Tl^{+3}	0.95 Å		

Some stable binary compound

CaCu_2O_3	Pmmm	a=9.85, b=4.11, c=3.47 Å
Ca_2CuO_3	Immm	a=12.23, b=3.77, c=3.25 Å
SrCuO_2	Cmcm	a=3.56, b=16.32, c=3.92 Å

Distances between ions

Cu - O	~ 1.95 Å
Bi - O	~ 2.75 Å
Ca - O	~ 2.45 Å
Sr - O	~ 2.50 Å