



INTERNATIONAL ATOMIC ENERGY AGENCY
UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION



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WORKING PARTY ON "FRACTURE PHYSICS"
(29 May - 16 June 1989)

OVERVIEW
FUNDAMENTALS OF FRACTURE

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These are preliminary lecture notes, intended only for distribution to participants.

Overview

FUNDAMENTALS OF FRACTURE.

Economic Implications

POLARITY OF MECHANICAL RESPONSE

Dislocations + Cracks.

Elastic Aspects

Atomic Aspects

Qm Force Laws.

Mechanical Failure

Structural materials

Economic value/loss.

Bridge failures

Aircraft failures H: Tech.

Gas + oil pipelines

Nuclear pressure vessels H: Tech.

Space Appls.

H: Tech.

E). Devices

High Tech :

Tradeoff of performance/Risk

Requires precise assessment.

Practice

1. Initial design requirements -
Matl nominal prop.
2. Processing Quality Control
Flaw control (NDE)
3. In-use Monitoring (NDE)

NDE

Engineering Practice

1. Flaw Detection + characterization
Flow size / shape (crack/void)
critical size: $1\mu\text{m} - \frac{1}{2}\text{m}$.

2. Must know characteristics
of flaw growth.

Critical size/stress
Chemical environment

Growth under cyclic stress

FRACTURE MECHANICS

Specification of mats in terms
of critical toughness tests

J-integral / COD / K_{Ic}

What Role for Fundamental Knowledge?

NEW

What makes mats tough?
Role of external Chem. Environ.
Ultimate Limits -

MECHANICAL POLARITY

Fully Ductile

Ag, Ag, Cu, Al ...



1. Low strength.
2. High energy absorption to failure
(High toughness)
3. Pure deformation: dislocations
No cracks.

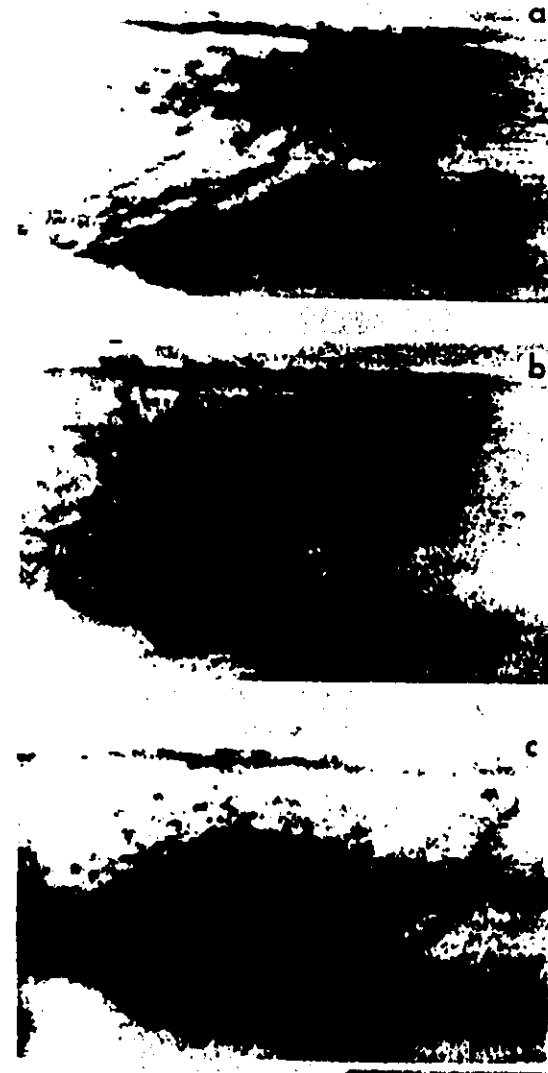
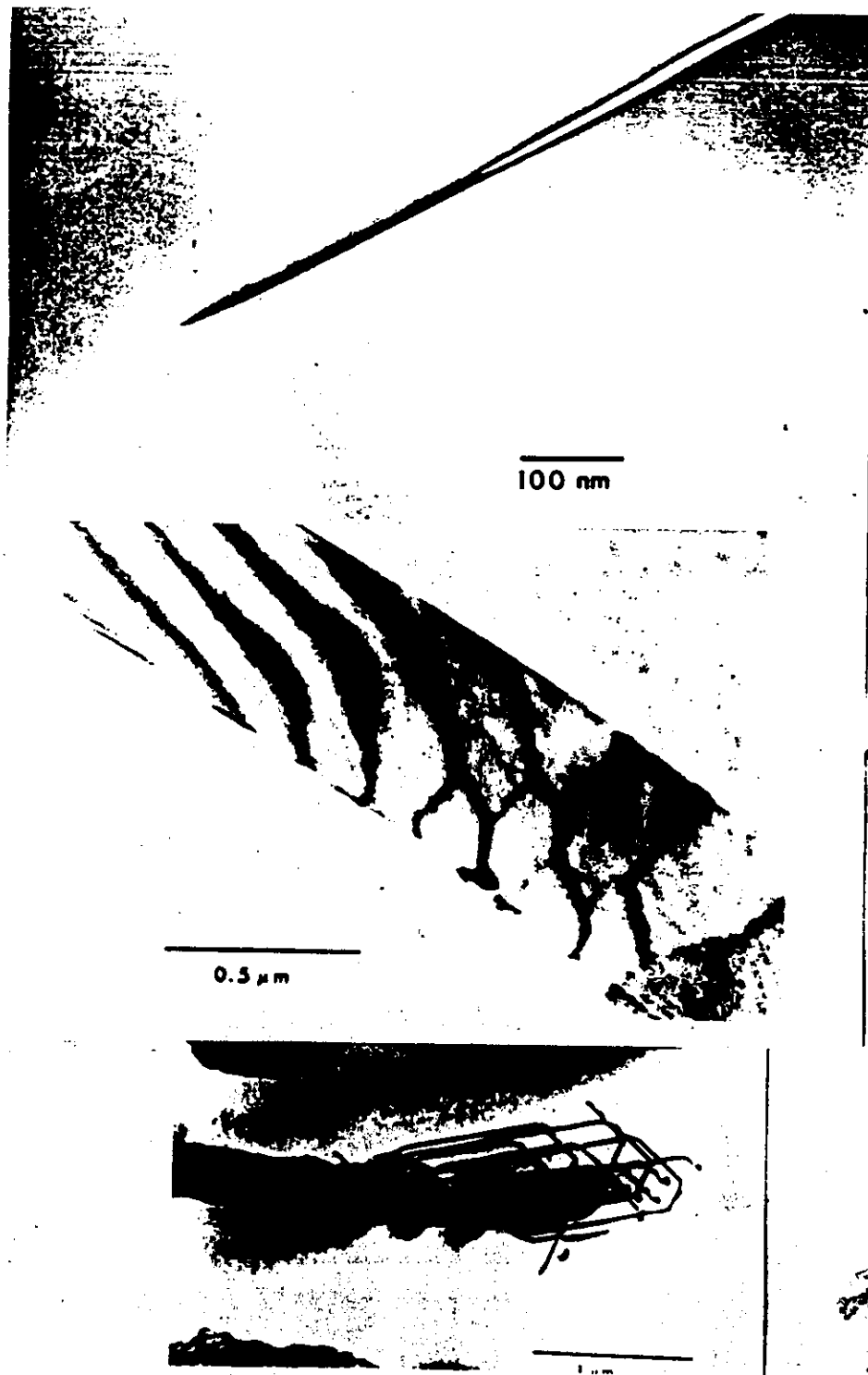
Fully Brittle

C, Si, Mica,
SiO₂ ...

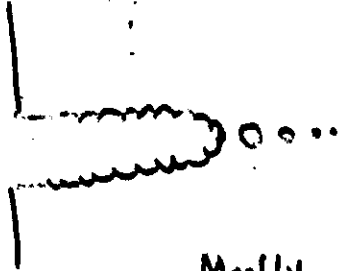


1. Hi Strength (near theoretical bond strength)
2. ≈ 0 toughness $\approx$ No energy to fail.
3. Pure Crack propagation -
No deformation.

Technological mats intermediate.

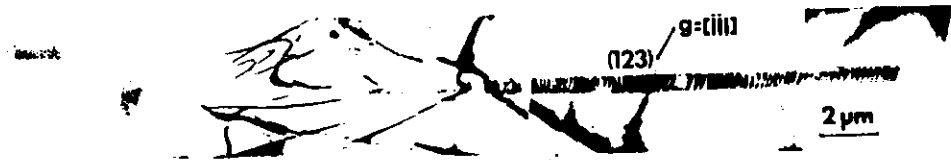


"Fracture" of a Techn. Matl. (Fe)



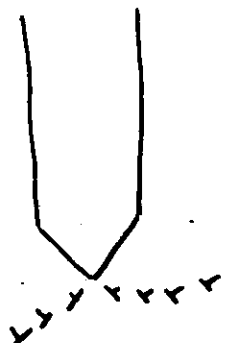
Mostly a ductile process.

1. Initiation of holes/voids
2. Growth of voids by pure deformation shear processes. (shear bands, shear cracks ???)





a 100 μm



100 μm

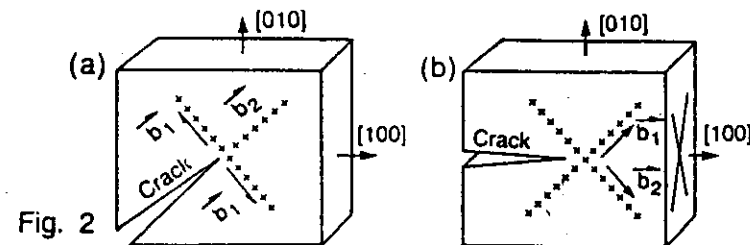


Fig. 2

Figure 2. A schematic representation of (a) $(\bar{1}10)$ plane crack with (110) and $(\bar{1}10)$ plane edge dislocations, (b) (010) plane crack with (110) and $(\bar{1}10)$ plane edge dislocations. The more common screw dislocations are lightly shown on the (100) face.



Figure 3. A photomicrograph of crack tip deformation in 3(b). The horizontal line is the crack. The cross are edge dislocations, some screw dislocations are also seen parallel to the crack.

Also Fu.
Zn

III
ke k_c T
ke k_c k_c

Ohr
ke k_c T

Burns, Scripta Met. 20 489 (1980)

Very small k_e based on measured
shielding (?) (dipoles?)



Pugh, See NATO
Atomistics of Fracture

"Pure" cases exist, but as a rule,
 Cracks AND dislocations coexist + interact
 Relative toughness depends on crack/dislocation
 interactions

Direct dislocation/crack interact.

Crack as source of Disl.

External Chemicals modify crack/disl. interact.

Tools

Elasticity (plasticity?) (ties to macroscopic)

Atomic scale theories / ideas

Bonding of Defective Solids (force laws)

Microstructural characterization

cl. microscopy.

exper. study of crack structure Difficult

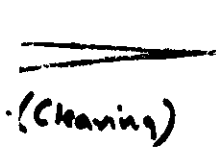
Disciplines

fract. Mech., Matls. Science, Physics
 Chemistry

A Hint

Classification Scheme for Cracks

Atomic Crack



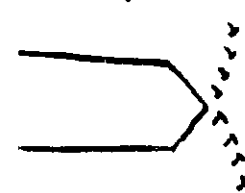
{ hi mobility,
 Lo resistance

$$\sigma \propto \frac{K}{\sqrt{r}} \quad f \propto K^2$$

Shielded by disl.

$$K = K_0 + K_D$$

Emitting Crack



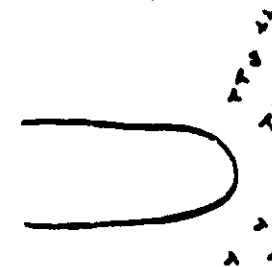
"zero" mobility

Zero force

moves only
 by emission

$$\sigma \propto \frac{1}{\sqrt{r}} \quad r \gg \frac{1}{2}$$

Absorbing Crack



"Zero" mobility

Zero force

moves by
 Plastic opening

σ not singular
 (but stress
 concentration)

Again

No pure cases — understand how
 one relates to another.