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**WINTER COLLEGE ON
HIGH RESOLUTION SPECTROSCOPY**

(8 January - 2 February 1990)

OPTICAL MOLASSES

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OPTICAL MOLASSES, NEW LASER-COOLING MECHANISMS and THE COLDEST GAS IN THE UNIVERSE

National Institute of Standards and
Technology (Formerly NBS)

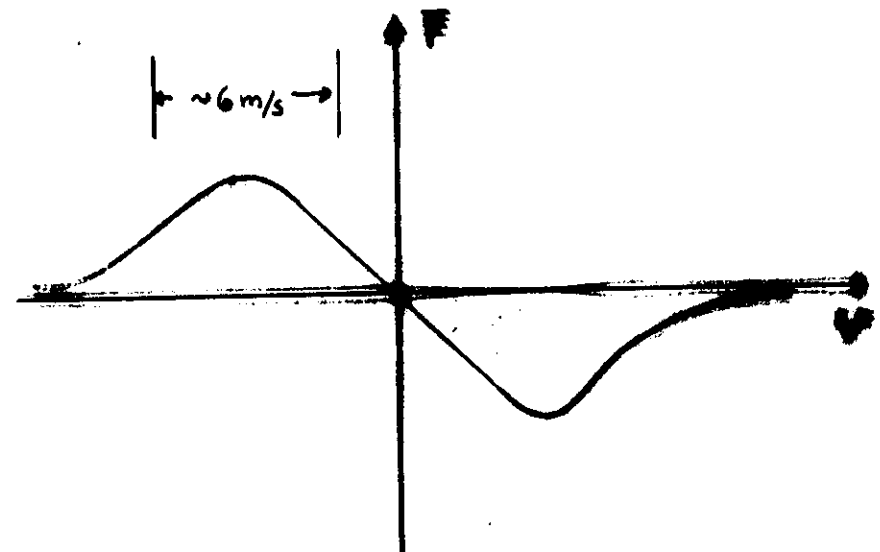
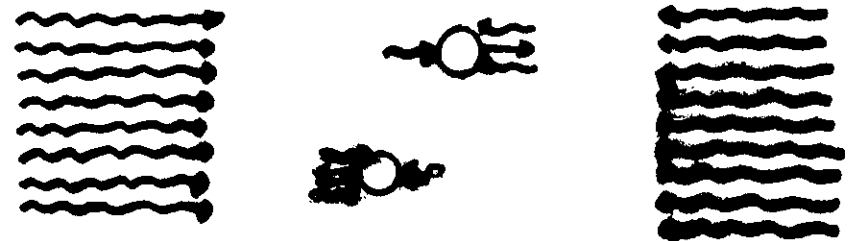
P. Lett
C. Westbrook
S. Rolston
R. Watts
C. Tanner
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H. Metcalf S.U.N.Y.
P. Gould U. Conn.
J. Dalibard B.U.F.

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J. Dalibard ENS
A. Clairon
G. Cohen-Tannoudji

"Balanced" Cooling

(Hansch & Schawlow 1975)



Excitation Lineshape



2-level atom, f fraction in excited state

$$f = \frac{1}{2} \frac{I/I_0}{1 + I/I_0 + \left(\frac{2\Delta}{\Gamma}\right)^2}$$

$$\Delta = \omega - \omega_0$$

$$\Gamma = 1/\tau_n$$

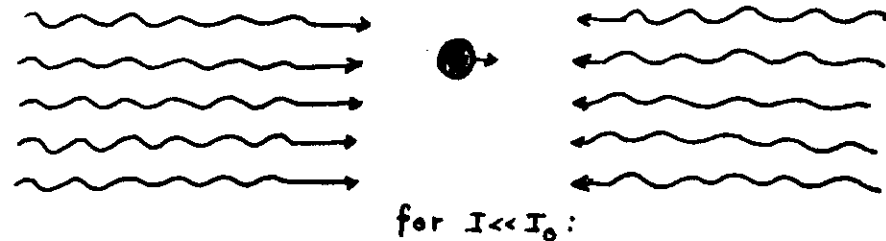
moving atom: $\Delta \rightarrow \Delta - \vec{k} \cdot \vec{v}$

absorption rate $\mathcal{R} = f \cdot \Gamma$

$$I/I_0 \rightarrow 0$$

$$\mathcal{R} = \frac{\Gamma}{2} \frac{I/I_0}{1 + \left[\frac{2(\Delta - \vec{k} \cdot \vec{v})}{\Gamma}\right]^2}$$

OPTICAL MOLASSES 1-D



$$F_L = \frac{\hbar \hbar \Gamma}{2} \frac{I/I_0}{1 + \left[\frac{2(\Delta - \hbar k v)}{\Gamma}\right]^2} \quad F_r = -\frac{\hbar \hbar \Gamma}{2} \frac{I/I_0}{1 + \left[\frac{2(\Delta + \hbar k v)}{\Gamma}\right]^2}$$

for $\hbar k v \ll \Delta, \Gamma$

$$F(v) = F_L + F_r \approx \underbrace{F_L(0) + F_r(0)}_0 + v \left(\frac{dF_L}{dv} + \frac{dF_r}{dv} \right)$$

$$\frac{dF_{L,r}}{dv} = \mp \frac{\hbar \hbar \Gamma}{2} \frac{\left(\frac{I}{I_0}\right) \left[\mp 4 \left(\frac{\Delta \mp \hbar k v}{\Gamma} \right) \frac{\hbar k}{\Gamma} \right]}{\left[1 + \left(\frac{\Delta \mp \hbar k v}{\Gamma} \right)^2 \right]^2} \approx \frac{2 \hbar \hbar^2 \left(\frac{I}{I_0}\right) \cdot \left(\frac{\Delta \mp \hbar k v}{\Gamma}\right)}{\left[1 + \left(\frac{\Delta \mp \hbar k v}{\Gamma} \right)^2 \right]}$$

$$F(v) = \frac{4 \hbar \hbar^2 \left(\frac{\Delta}{\Gamma}\right) \frac{I}{I_0} v}{\left[1 + \left(\frac{\Delta}{\Gamma} \right)^2 \right]^2} = -\alpha v \quad \begin{matrix} \alpha > 0 \text{ for } \\ \Delta < 0 \end{matrix}$$

$$F \cdot v = -\alpha v^2 = \dot{E}_{cool}$$

(cooling rate)

1-D Cooling Limit

heating rate in "mythical" 1-D case ($I/I_0 \ll 1$)

$$\dot{E}_{\text{heat}} = \frac{\dot{p}^2}{2M}$$

Now p , the atomic momentum, random walks
one step = $\hbar k$ for every absorption
and emission.

$$\langle p^2 \rangle = 2N(\hbar k)^2 \quad N = \text{absorbed photons}$$

$$\frac{\dot{p}^2}{2M} = \frac{\hbar^2 k^2}{2M} 2 \mathcal{R}_{\text{abs}}^{(2\text{-beam})} = \frac{\hbar^2 k^2}{M} \Gamma \frac{I/I_0}{1 + (2\Delta/\Gamma)^2} = \dot{E}_{\text{heat}}$$

$$\text{set } \dot{E}_{\text{heat}} - \dot{E}_{\text{cool}} = 0$$

$$\frac{\hbar^2 k^2 \Gamma}{M} \frac{I/I_0}{[]} = \frac{4 \hbar^2 k^2 (\frac{2\Delta}{\Gamma}) v^2 I/I_0}{[]^2}$$

$$\frac{1}{2} M v^2 = \frac{\hbar \Gamma}{8} \frac{1 + (\frac{2\Delta}{\Gamma})^2}{(\frac{2\Delta}{\Gamma})} \approx \frac{1}{2} k_B T$$

$$k_B T_{\text{min}} = \frac{\hbar \Gamma}{2} \quad \Delta = -\Gamma/2 \quad (I/I_0 \ll 1)$$

240 mK for Na

The Doppler Cooling Limit

$$k_B T = -\frac{\hbar \Gamma}{4} \frac{1 + (2\Delta/\Gamma)^2}{2\Delta/\Gamma}$$

$$\text{minimize at } \Delta = -\Gamma/2$$

$$\text{where } k_B T_D = \frac{\hbar \Gamma}{2}$$

$$T_D = 240 \text{ mK for Na} \\ 120 \text{ mK for Cs}$$

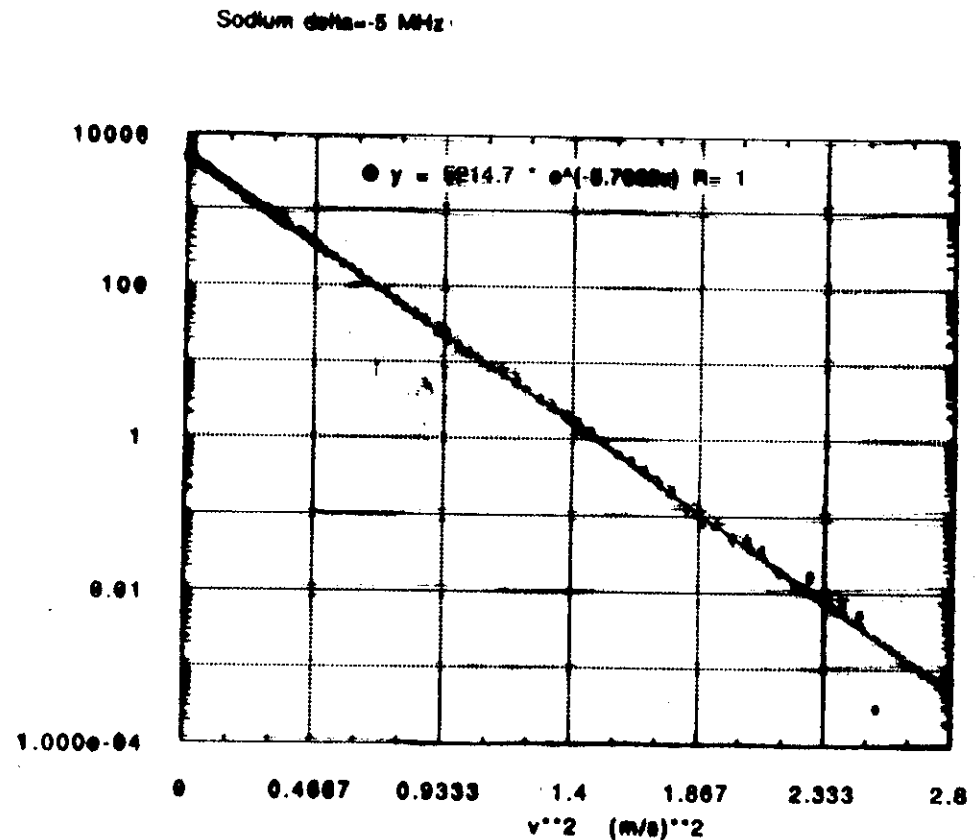
Cooling Limit in 3-D ($I \ll I_0$)

- Consider each component of $\vec{v}$
- Each is damped exactly as in 1-D
- Emissions can go off in all directions (less heating for a given component)
- But other directions contribute to this direction with their emissions
- So: heating is the same.

$$\text{so } \frac{1}{2} m v_x^2 = \frac{k_B \gamma}{4} = \frac{1}{2} k_B T_{\text{min}}$$

$$\frac{1}{2} m |\vec{v}|^2 = \frac{3 k_B \gamma}{4} = \frac{3}{2} k_B T_{\text{min}}$$

$$k_B T_{\text{min}} = \frac{k_B \gamma}{2} \quad \text{in 3-D as in 1-D}$$



OPTICAL MOLASSES

(the "molasses effect")



$$\vec{F} = -\alpha \vec{v}$$

$$\dot{E}_{\text{cool}} = \vec{F} \cdot \vec{v} = -\alpha v^2 \propto E_{\text{KINETIC}}$$

$$\frac{E_{\text{KIN}}}{\dot{E}_{\text{cool}}} \sim 20 \mu\text{s} \quad (\text{optimum cond. for Na})$$

but !

The momentum takes a random walk

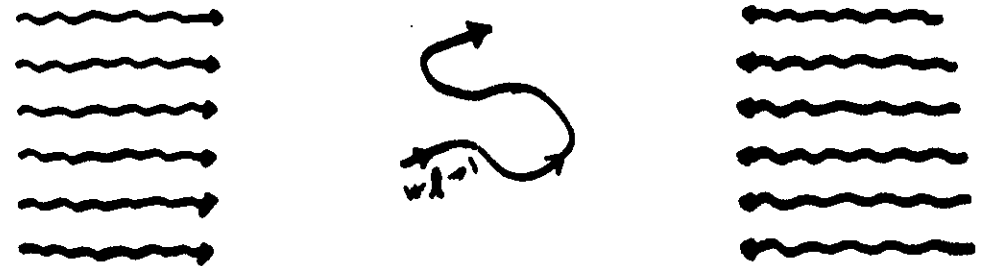
$$\dot{p} \propto \mathcal{R}_{\text{scatt}} \propto \dot{E}_{\text{HEAT}}$$

$$\dot{E}_{\text{HEAT}} = \dot{E}_{\text{COOL}} \Rightarrow \hbar \omega T = \frac{\hbar \delta}{2}$$

$$\langle v^2 \rangle \propto \frac{\mathcal{R}}{\alpha}$$

$$\langle v^2 \rangle^{1/2} \sim 50 \text{ cm/s} \quad (\text{cooling limit})$$

$$T = 240 \text{ nK} \quad (\text{at } T = 1 \text{ mK})$$



$$\text{For Na } \gamma_{\text{sp}} \sim 40 \text{ ns} = \tau$$

$$\text{and } \langle v^2 \rangle^{1/2} \sim 0.5 \text{ m/s}$$

$$l = v \tau \sim 20 \text{ nm}$$

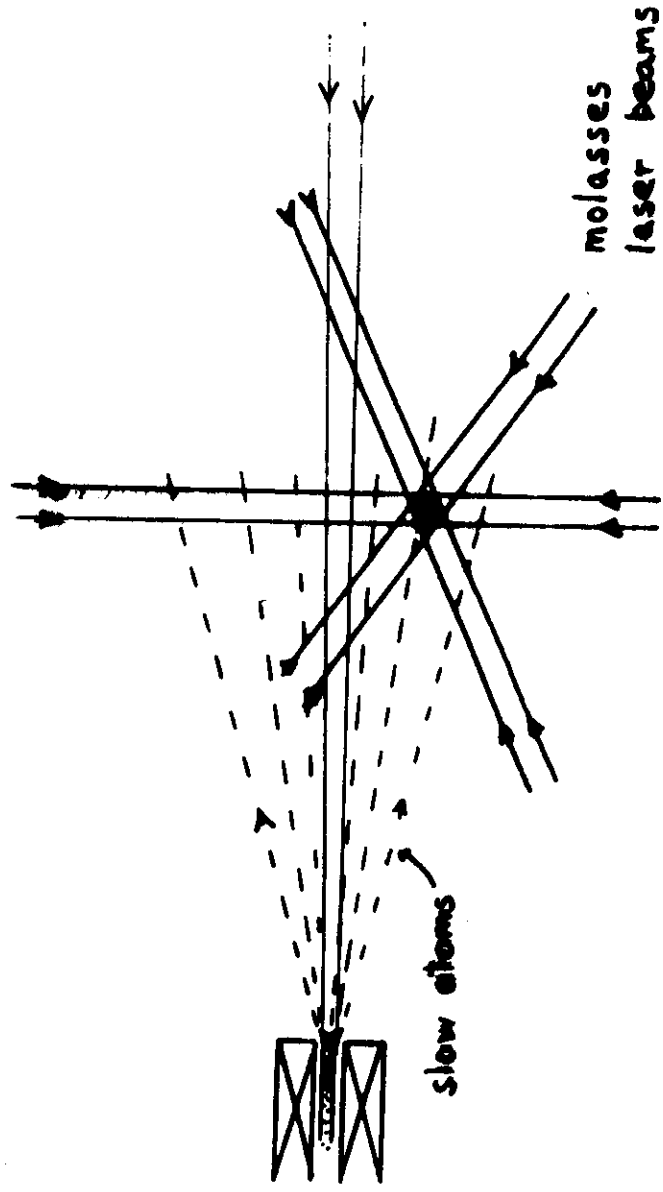
$$\text{random walk: } \langle x^2 \rangle^{1/2} = l \left(\frac{t}{\tau} \right)^{1/2}$$

$$\text{for } t = 1 \text{ sec, } \langle x^2 \rangle^{1/2} \sim 3 \text{ mm}$$

(compare to 500 mm for ballistic motion)

Bell Labs 1985 theory + expt

NBS 1985 theory (Dalibard)



Qubit Molasses: 800 nm 85° (Pulsed)

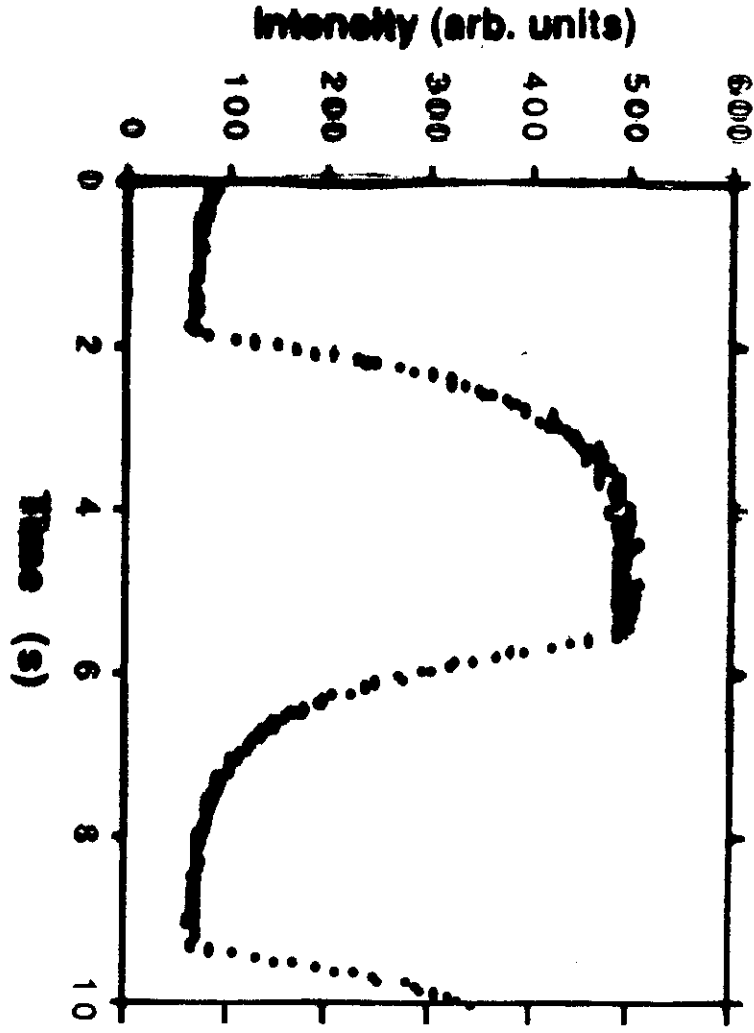
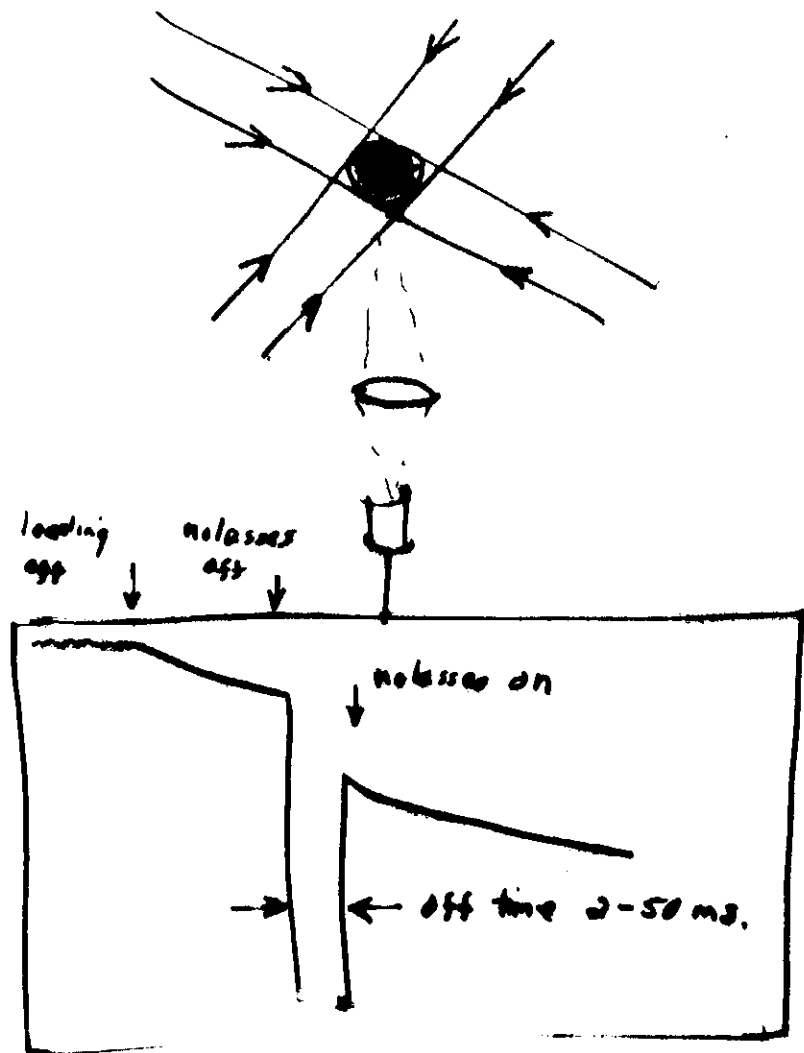


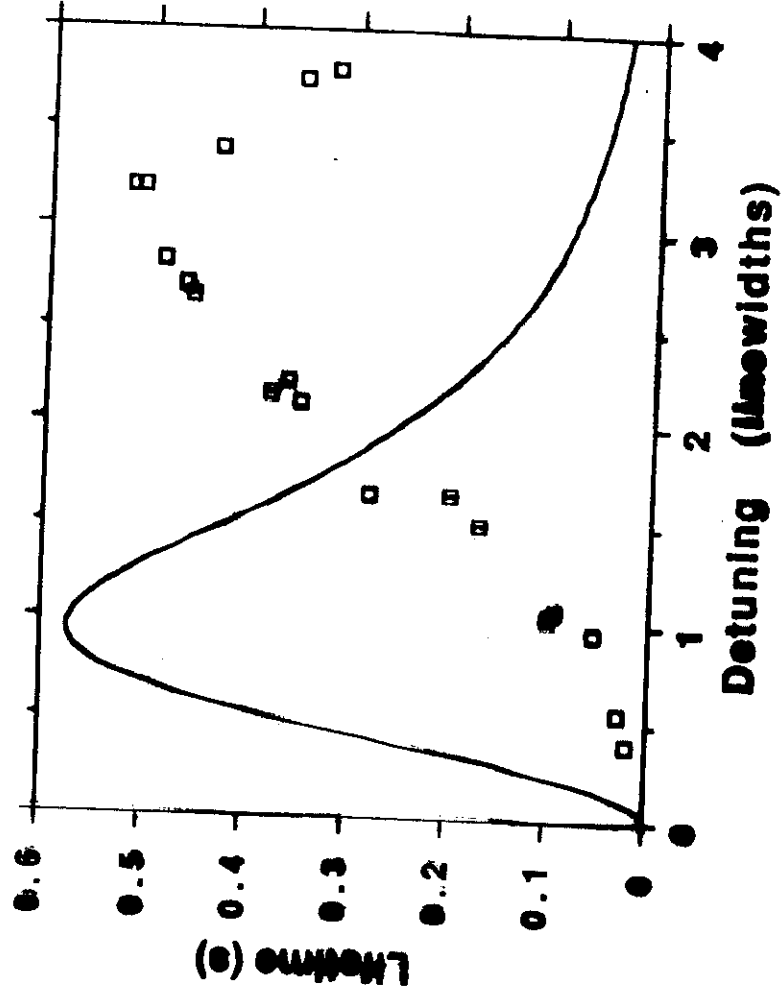
Fig 11

Release + Recapture

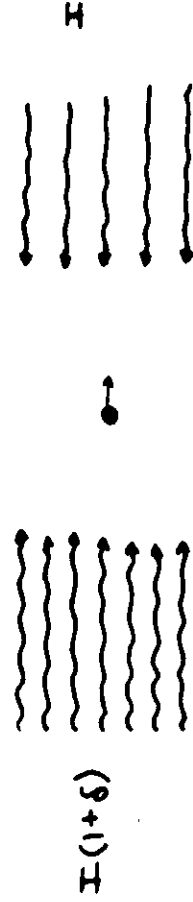


Molasses Temperature

- AT+T Bell Labs 1985
 $T = 240^{+200}_{-60} \mu\text{K}$
- NBS Gaithersburg 1987
 $T \sim 240 \mu\text{K}$
- JILA 1988
(Cesium: $T_{\text{min}} = 125 \mu\text{K}$)
 $T = 100^{+100}_{-30} \mu\text{K}$



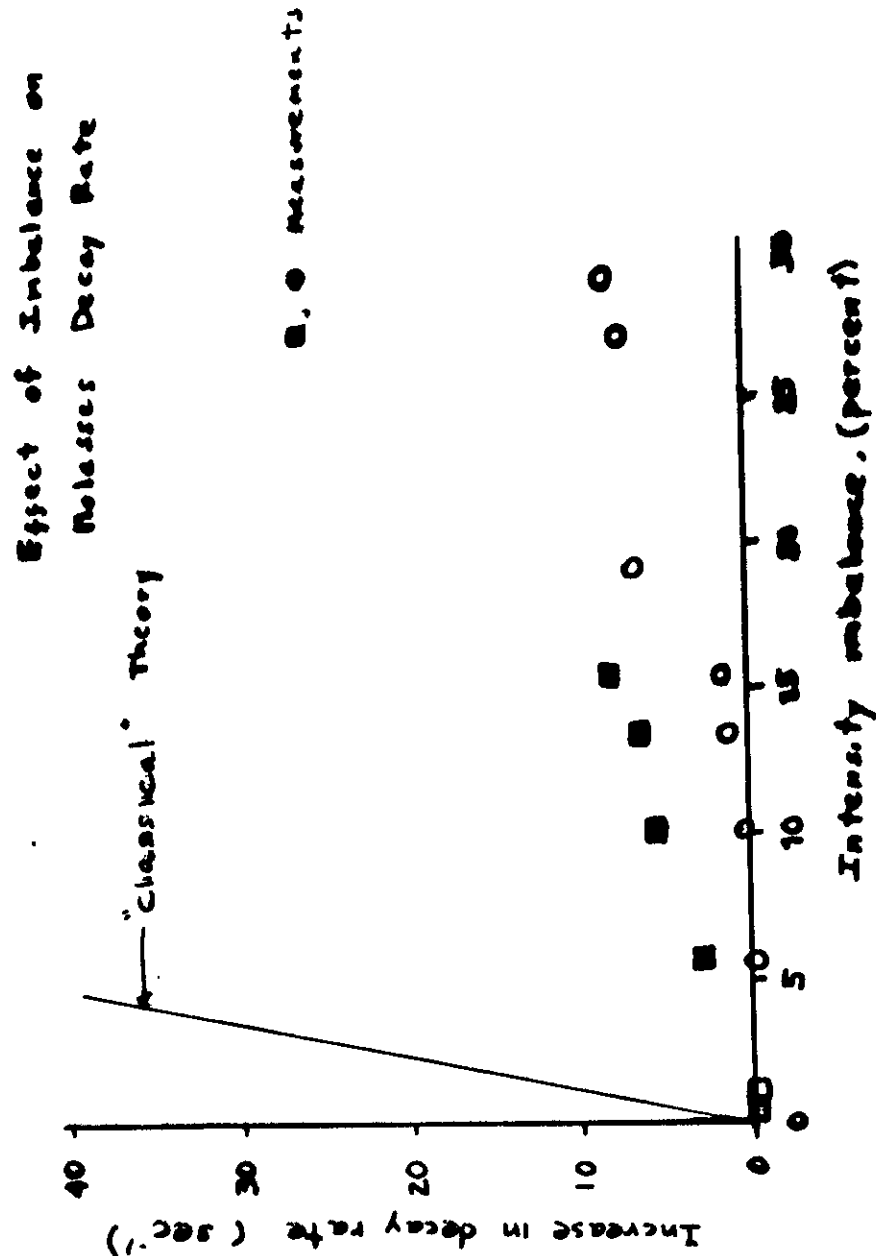
VIBRATIONAL MODES IN Effect of Imbalance on Rotations



$$V_{\text{eff}} = \frac{\delta (\Delta^2 + \frac{1}{4} [1 + \frac{I}{I_0}])}{4\pi\Delta} \approx 35 \text{ cm/s} \quad \text{with } \delta = 0.1$$

and best molecules
($\gamma \approx 15 \text{ ms}$)

Molasses Temperature



• AT+T Bell Labs 1985

$$T = 240 \pm 200 \mu K$$

• NBS Gaithersburg 1987

$$T \sim 240 \mu K$$

• JILA 1988 (Cesium, $T_{\text{lin}} = 125 \mu K$)

$$T = 100 \pm 100 \mu K$$

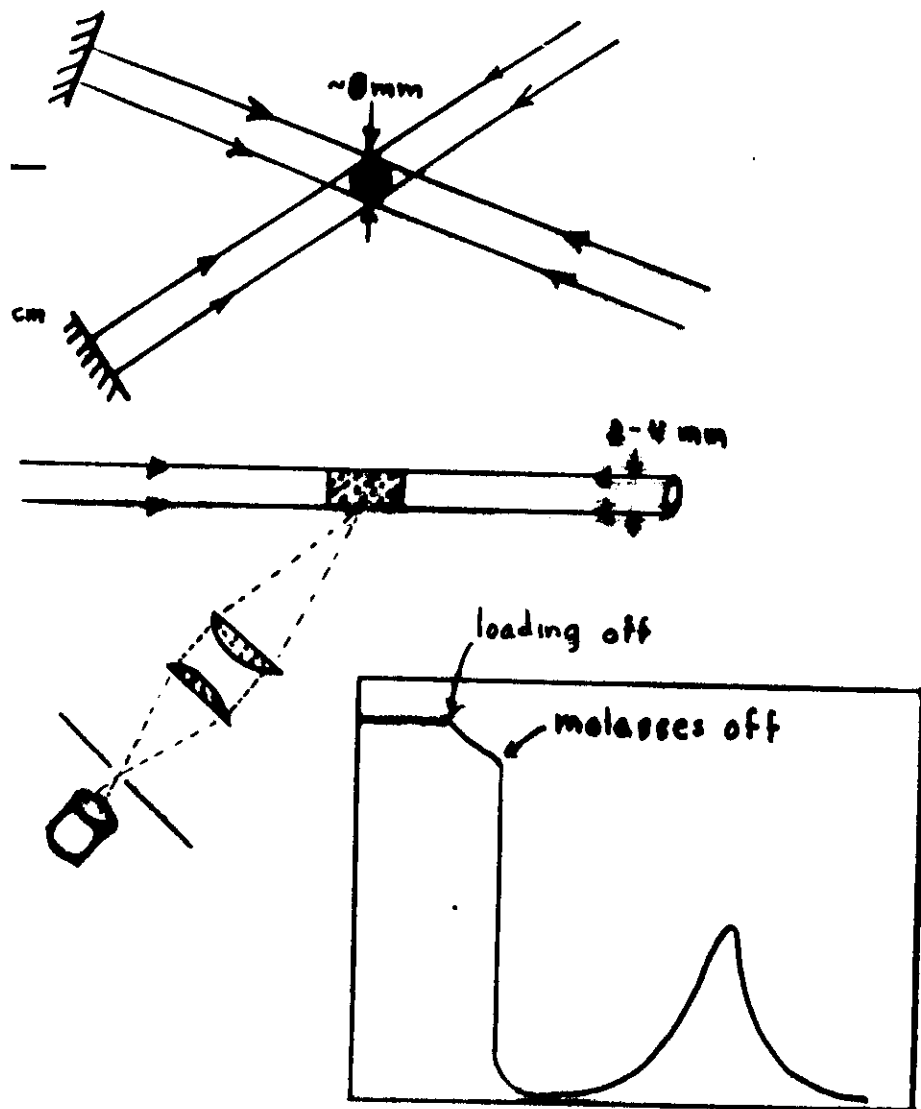
i.e. $k_B T \approx \frac{h^2}{2m}$ in each case

But: $k_B T = \frac{h^2}{2m}$ only for $2 \ll 2\pi k_F$ and $\Delta = -\frac{h^2}{2m}$

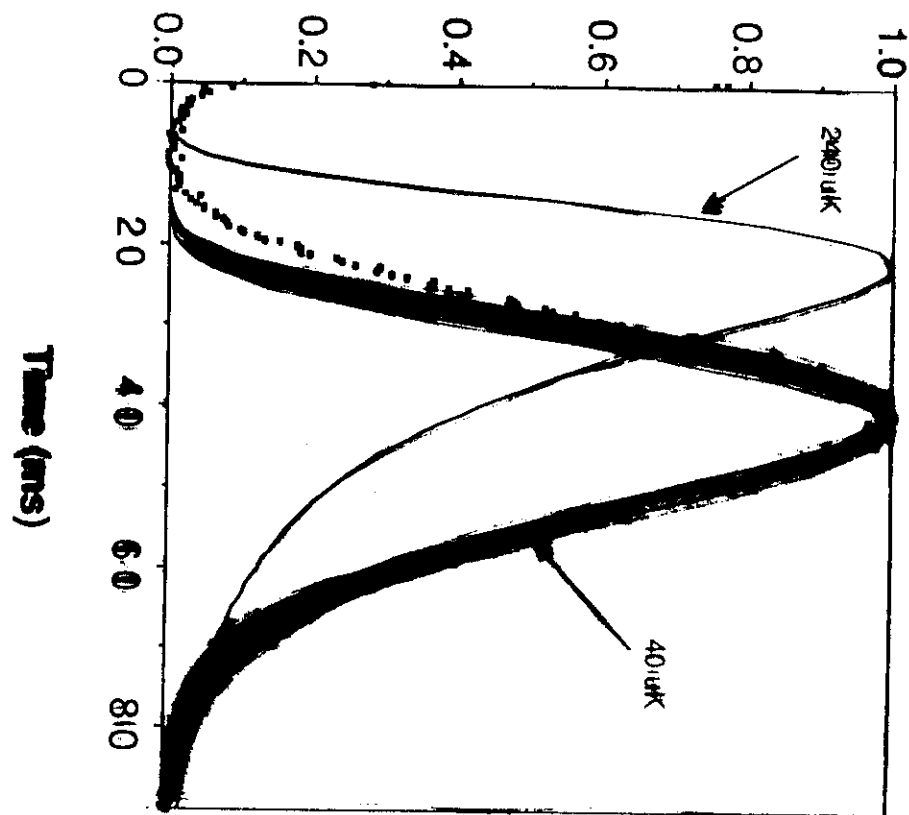
$$k_B T = \frac{\hbar^2}{4m} \cdot \frac{1 + 2\epsilon_0 + \left(\frac{2\Delta}{\epsilon_0}\right)^2}{4\epsilon_0^2}$$

and $2/\epsilon_0 \approx 1$, $2\epsilon_0 > 1$ for NBS expts.

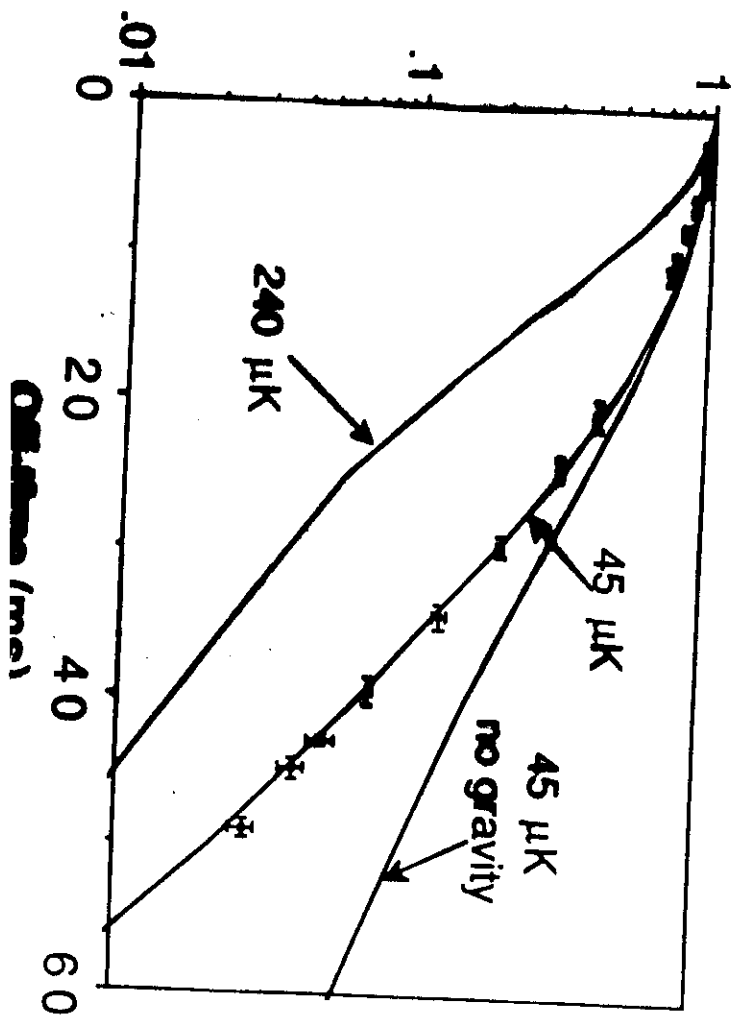
Time of Flight



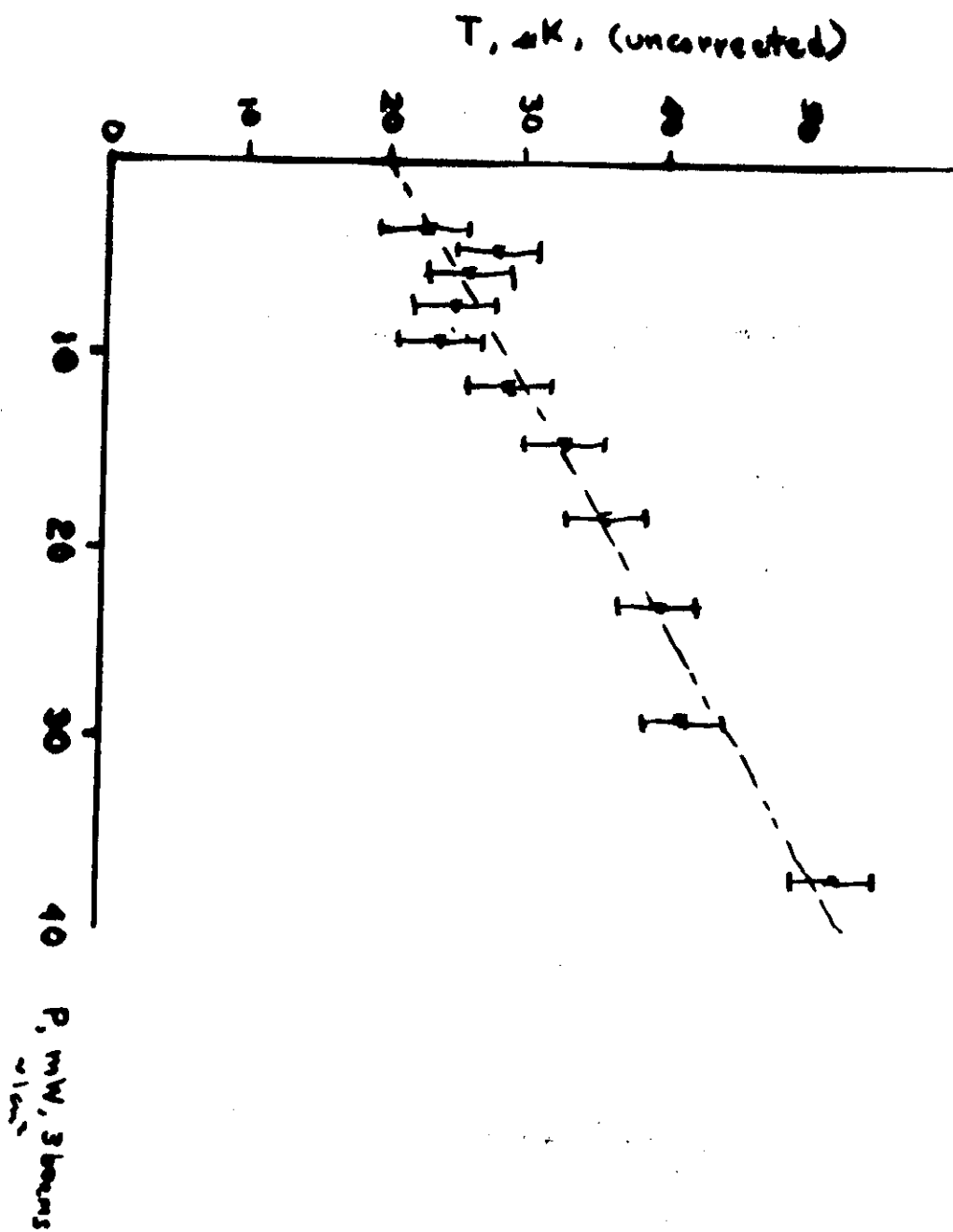
of Atoms



Fraction remaining



Temperature correction



Properties of ultra-cold Molasses

- $T \ll T_D$
- $T \sim 1/I$
- T sensitive to polarization
- T very sensitive to magnetic field

(Na is not a 2-level atom)

(Note that ion traps see T_D even for non-2-level ions)

THE COLDEST GAS IN THE UNIVERSE?

What is the vapor pressure of Na at equilibrium, at $T = 20 \mu\text{K}$?

Classical Thermodynamics, assume constant heat of vaporization:

$$P = P_0 e^{-\frac{\Delta E_v}{k_B T}}$$

for Na $\frac{\Delta E_v}{k_B} = 12400$

so $P(20 \mu\text{K}) = P_0 \cdot e^{-6.2 \times 10^8}$

or $n(20 \mu\text{K}) = 10^{-2.7 \times 10^8} \frac{\text{atoms}}{\text{cubic whatever}}$

$$V_{\text{universe}} \approx 10^{85} \text{ cm}^3$$