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UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION
INTERNATIONAL CENTRE FOR THEORETICAL PHYSICS
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**WINTER COLLEGE ON
HIGH RESOLUTION SPECTROSCOPY**

(8 January - 2 February 1990)

MULTIPHOTON IONISATION OF ATOMS

II Strong Fields

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Formation of multicharged ions

Experiments

alkaline earths, rare earths

Suran - Zepelacny 1975..... dye laser.
Feldmann et al 1982..... $\sim 10^{10} \text{ W cm}^{-2}$ nsec.
Agostini - Petrone 1985 $\sim 10^{12} \text{ W cm}^{-2}$ psec

noble gases

L'Huillier, Lompré, Mainfray, Manus 1982
1. Atm. 0.53 μ . $\rightarrow 10^{14} \text{ W cm}^{-2}$ psec.
Chin-Yeroual, Lavigne 1985
 CO_2 laser. $\rightarrow 10^{14} \text{ W cm}^{-2}$ nsec.
Luk et al 1983 193 nm $\rightarrow 10^{17} \text{ W cm}^{-2}$ psec.

Perry - Landen 1988

dye laser $\sim 10^{14} \text{ W cm}^{-2}$ psec.
 $\sim 300 \text{ nm.} \rightarrow 10^{14} \text{ W cm}^{-2}$ psec.

Theoretical investigations.

Statistical interpretations

Åberg et al 1983

Crance 1984

Geltman 1985

Quantum calculations

screening of the field Wendum L'Huillier 1986
double ionisation Crance Gyman 1985
Wendum L'Huillier 1987

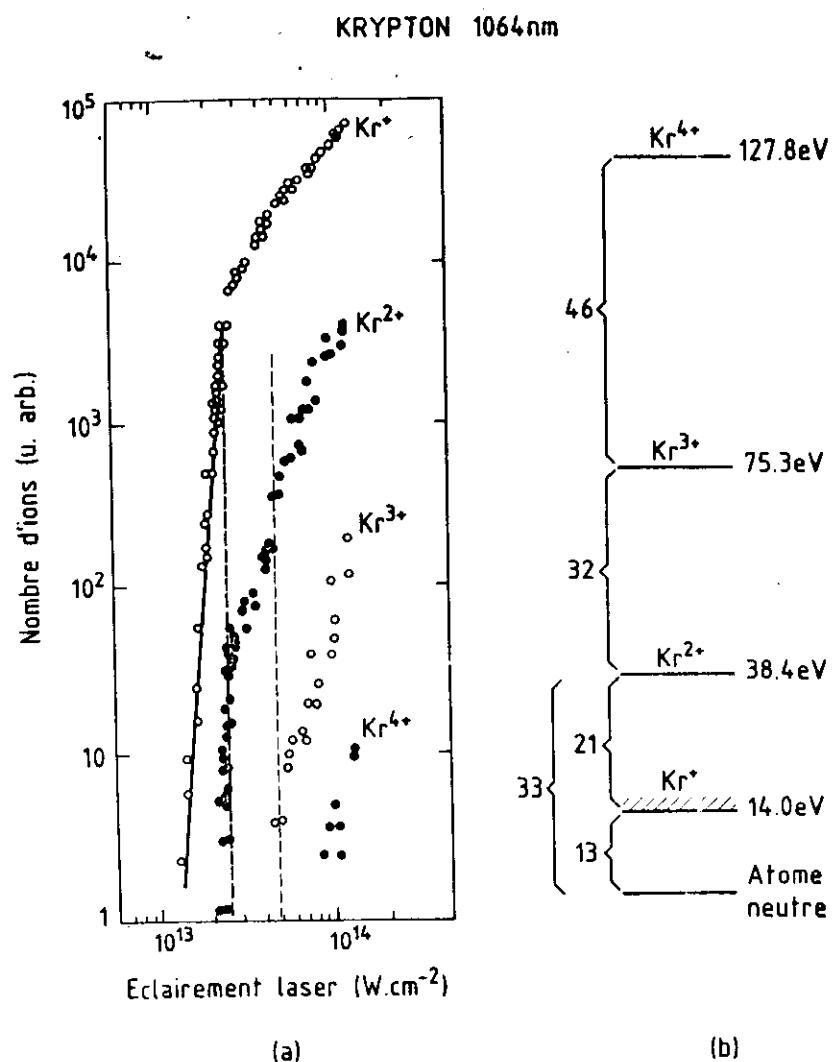


Figure 26- (a) Nombres d'ions Kr^+ à Kr^{4+} en fonction de l'éclairage laser à 1064 nm.
(b) Schéma des énergies d'ionisation du krypton.

Questions

Once single ionisation has been reached, it's easy to obtain multiple ionisation. Why?

Do electrons leave the atom one by one or several together?

Ions are left in ground state or in excited states?

How does the process vary with the laser frequency?

What parameters characterize the process?

How can various experiments be related?

→ simple features -

- time development
- coherence effects
- statistical interpretation

P. Feldmann, J. Trautwein, S.L. Chin, W. von Helden, F.H. Waage
J. Phys. B15 (1982) 1663



Figure 3. As figure 1, 1 = 9395 Å.

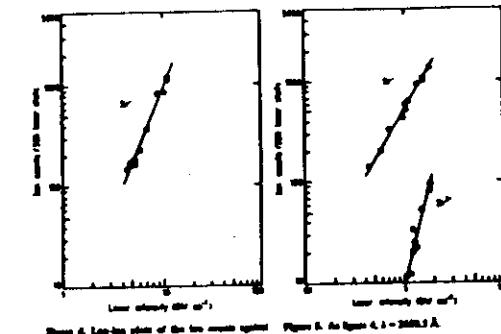


Figure 4. As figure 3, 1 = 9450 Å.

9395cm^{-1}

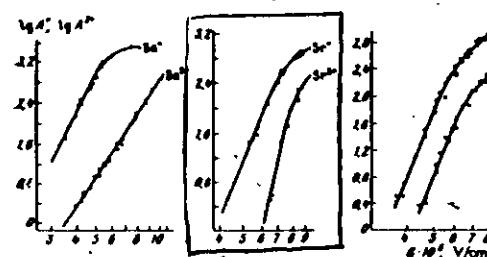
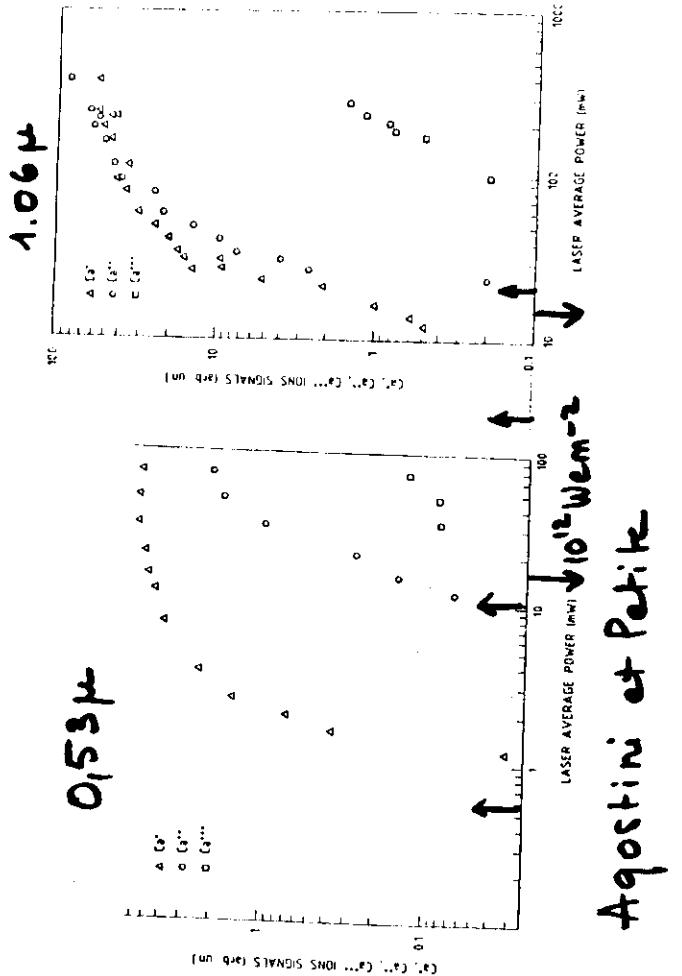
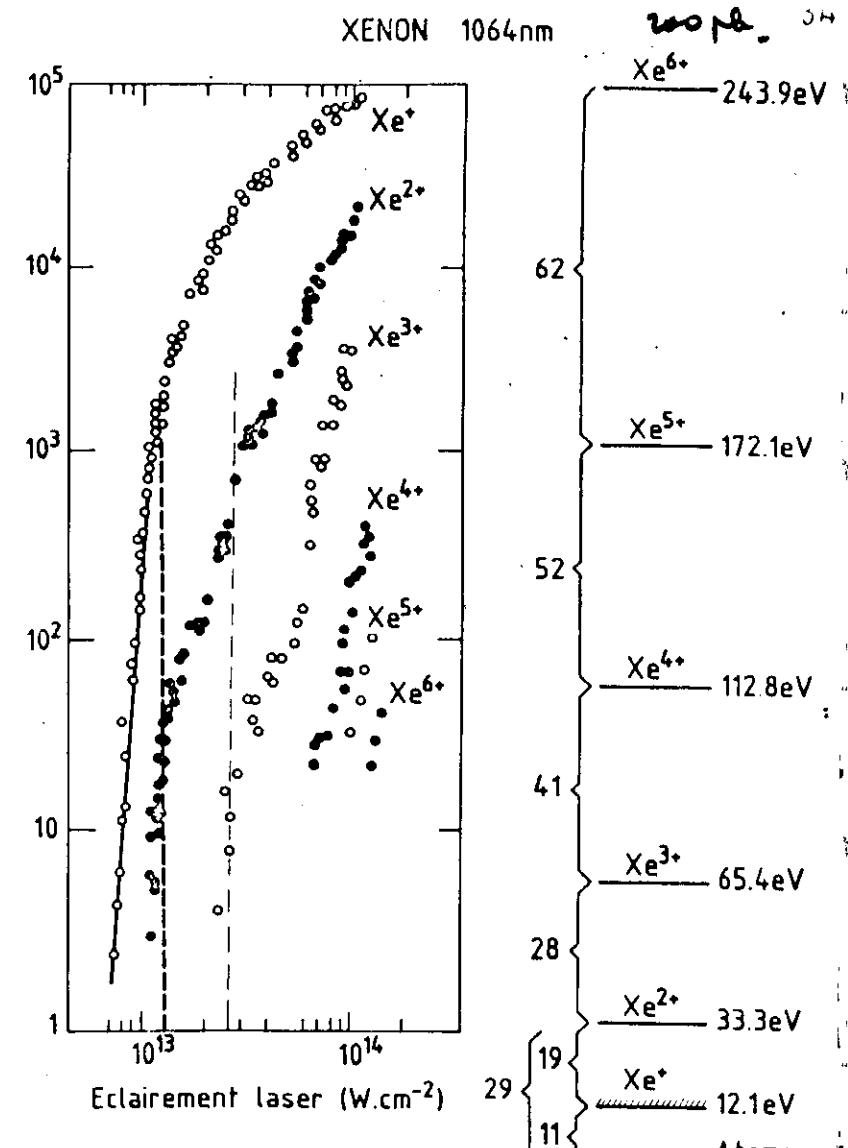


FIG. 2. Amplitudes of the ion signals from singly- and doubly-charged ions of barium, strontium, and europium vs the field strength E .

J. Trautwein, S.L. Chin, W. von Helden, F.H. Waage, J. Phys. B15 (1982) 48
(1979) 1-7

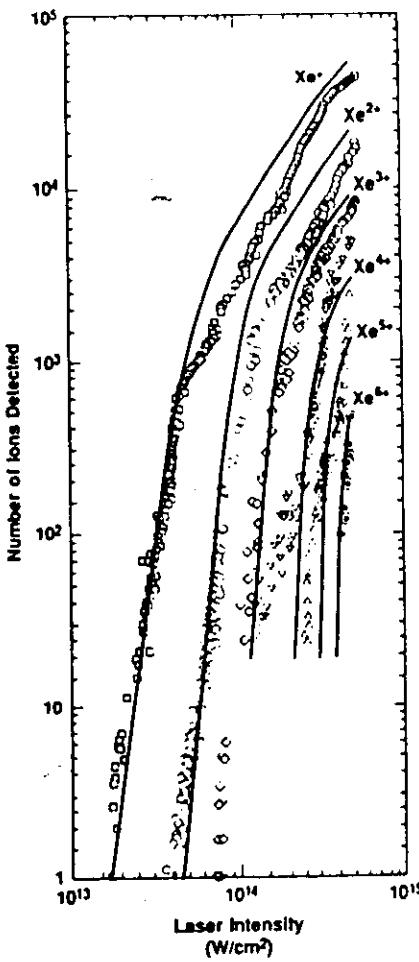


Aqostini et Peltier



L'Huillier Compé Haugray (a)
(b)

Figure 20- (a) Nombres d'ions Xe^+ à Xe^{6+} en fonction de l'éclairage laser sur une échelle doublment logarithmique.
(b) Schéma des énergies d'ionisation du xénon.



Perry et al Phys Rev A 37 747 (88)
5860 Å ~1 ps

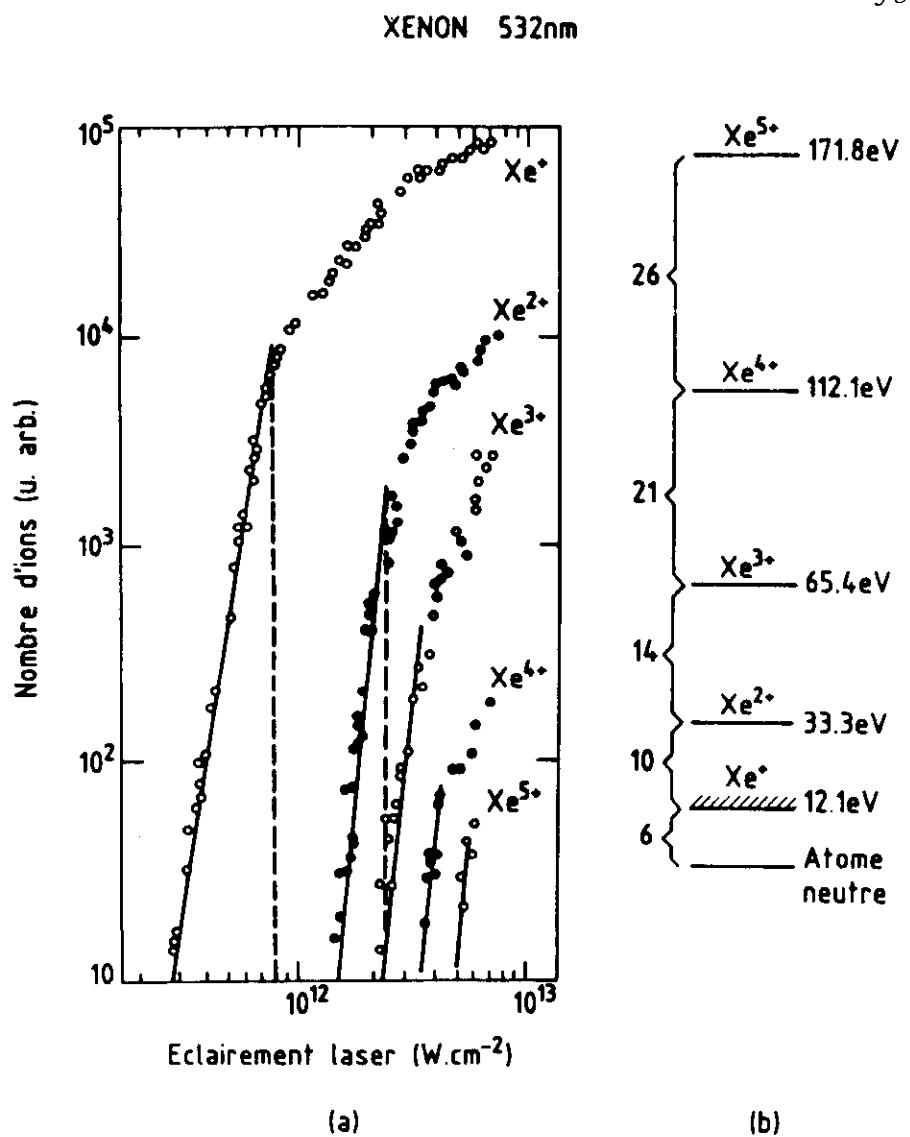


Figure 32 - (a) Nombres d'ions Xe^+ à Xe^{5+} en fonction de l'éclairage laser (seuil de détection élevé d'un facteur 10).
(b) Schéma des énergies d'ionisation du xénon.

L'Huillier et al (1984)

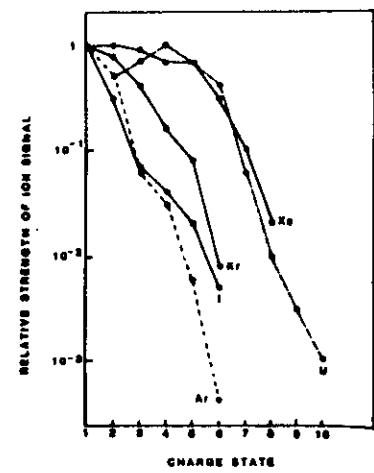


Fig. 4. Relative strength of ion signals for various atoms. No value is given for U^+ because of the interference with UF^+ , UF_2^+ , etc.

143 nm

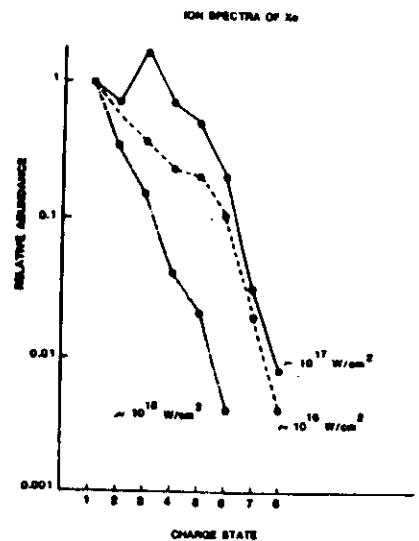
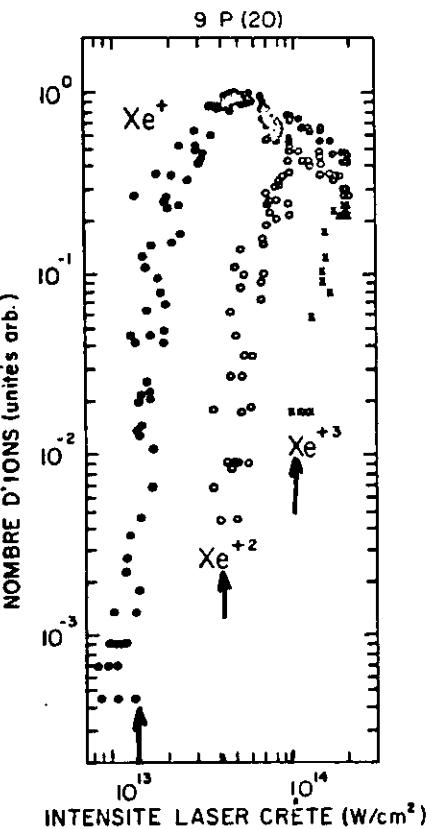


FIG. 5. Relative abundance of charge-state distributions observed in the ion spectra of xenon in the intensity range 10^{16} - 10^{18} W/cm 2 at 193 nm.

T.S. Luk, U.Johann, H.Egger, H.Pummer, C.F.Rhodes Phys. Rev. A32 (1985) 214



Chin Yer geau Louigne

Figure 3.4 - Nombre d'ions de Xe versus Intensité crête en échelle doublement logarithmique avec le laser 1.1 ns à 9.55 μ m.

Double ionisation: direct vs stepwise process

n_0 neutral

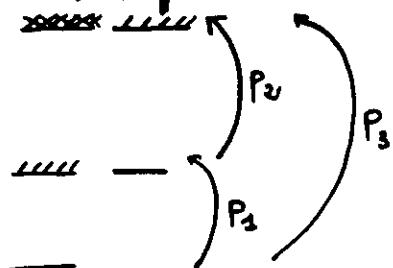
n_1 singly charged ions

n_2 doubly charged ions

Perturbative solution

$$n_0 = 1 \quad n_1 = \Gamma_{01} t \quad n_2 = \Gamma_{02} \pm \frac{\Gamma_{01}\Gamma_{12} t^2}{2}$$

→ different time-development.



Coherence effects

$$\frac{\text{Yield incoherent light}}{\text{Yield coherent light}} = n!$$

Direct process: p_3 photons simultaneously absorbed

Stepwise process: p_1 photons simultaneously absorbed incoherently followed by p_2 photons simultaneously absorbed

chaotic versus coherent enhancement =

$p_3!$ for direct process

$p_1! p_2!$ for stepwise process

Example Xenon 1.06μ . $p_1 = 11$ $p_2 = 19$ $p_3 = 29$

$$p_1! p_2! = 4.8 \cdot 10^{24} \quad p_3! = 8.8 \cdot 10^{30}$$

Statistical interpretation

Noble gases: outer shell $ms^2 mp^6$

comparable orbital size and extraction potential.

\leq

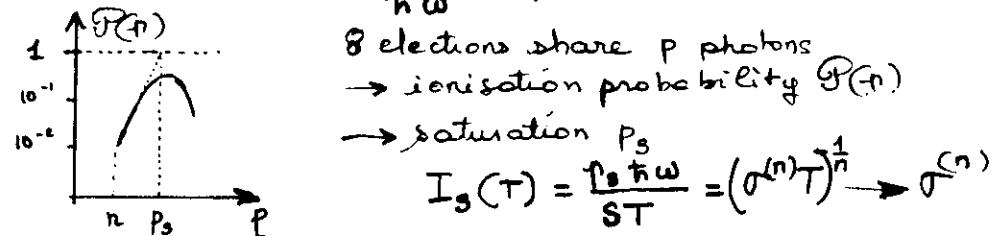
$$S = \pi \langle r \rangle^2$$

E ionisation potential

$$\frac{E}{\hbar \omega} \propto n \quad \text{minimum number of absorptions required}$$

Assumption: an electron is ejected if it absorbs at least n photons in a time $\leq T = \frac{\hbar}{\omega}$

Intensity $I \rightarrow \frac{I ST}{\hbar \omega} = p$ photons



8 electrons share p photons

→ ionisation probability $P(n)$

$$\rightarrow \text{saturation } p_s \quad I_s(T) = \frac{p_s \hbar \omega}{ST} = \left(\rho^n T \right)^{\frac{1}{n}} \rightarrow \rho^n$$

gives an estimate of - ionisation probability

- direct vs stepwise processes
- inground state vs excited states

Saturation intensities

excitation 0.53μ

0.586μ

	Xe	Ne	Xe	Kr	Ar
exp.	$6.8 \cdot 10^6$	$2.6 \cdot 10^{12}$	$4 \cdot 10^{13}$	$5 \cdot 10^{13}$	$6.7 \cdot 10^{13}$
th.	$6.8 \cdot 10^{12}$	$2.6 \cdot 10^{14}$	10^{13}	$2 \cdot 10^{13}$	$4.4 \cdot 10^{13}$

excitation 1.06μ

	Xe	Kr	Ar	Ne
exp.	10^{13}	$2.1 \cdot 10^6$	$2.55 \cdot 10^{13}$	$3.5 \cdot 10^{14}$
th.	$1.24 \cdot 10^{13}$	$2.15 \cdot 10^6$	$4.67 \cdot 10^{13}$	$3 \cdot 10^{14}$

excitation 10μ $\times 10^{13}$

ionisation from	Xe	Xe^+	Xe^{++}	Kr	Kr^+	Ar	Ar^+	Ne	He
exp	1.2	4.8	12	1.9	8.8	2.4	12	6.5	10
th	1.3	4.1	9.7	2.5	7.7	4.5	13.6	19.7	34.5

Saturation intensity

$\sigma^{(n)} I^n$ = probability per unit time

$\sigma^{(n)} I^n t$ = probability per pulse

$\sigma^{(n)} I_s^n t = 1$ defines saturation intensity

$$I_s = (\sigma^{(n)} t)^{-\frac{1}{n}}$$

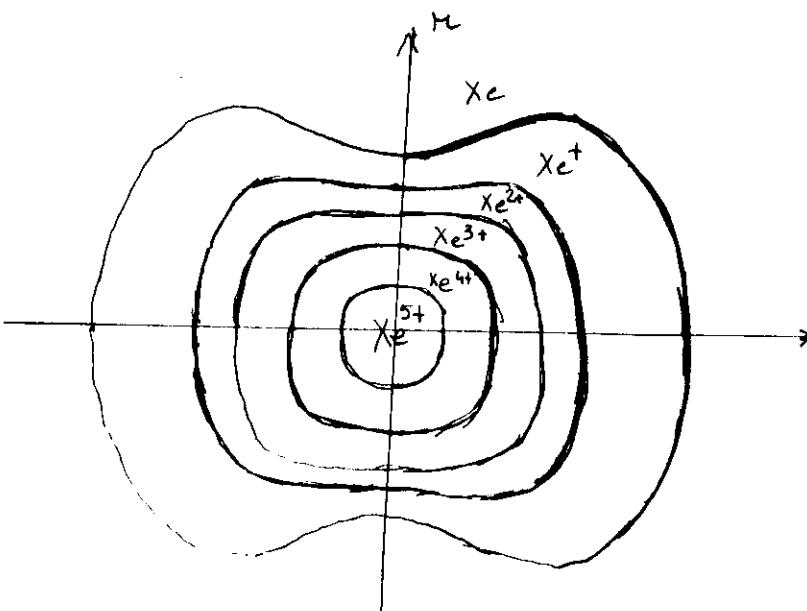
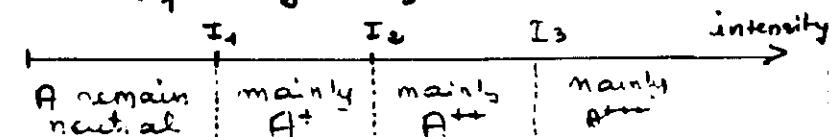
allows one to relate experiments with \neq pulse duration

$A \rightarrow A^+ + e^-$ saturation intensity I_1

$A^+ \rightarrow A^{++} + e^-$ $-$ I_2

$A^{++} \rightarrow A^{+++} + e^-$ $-$ I_3

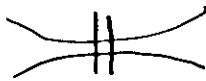
$$I_1 < I_2 < I_3 \dots$$



Application to a gaussian beam

43

$$I(r) = I_0 e^{-r^2}$$



$$\text{Volume where } I > I_0 \propto \log \frac{I_M}{I_0}$$

$A \rightarrow A^+$ saturation I_1

$A^+ \rightarrow A^{++} \quad - \quad I_2$

.....

$$\text{number of ions } A^+ A^{2+} \dots \propto \log \frac{I_M}{I_1}$$

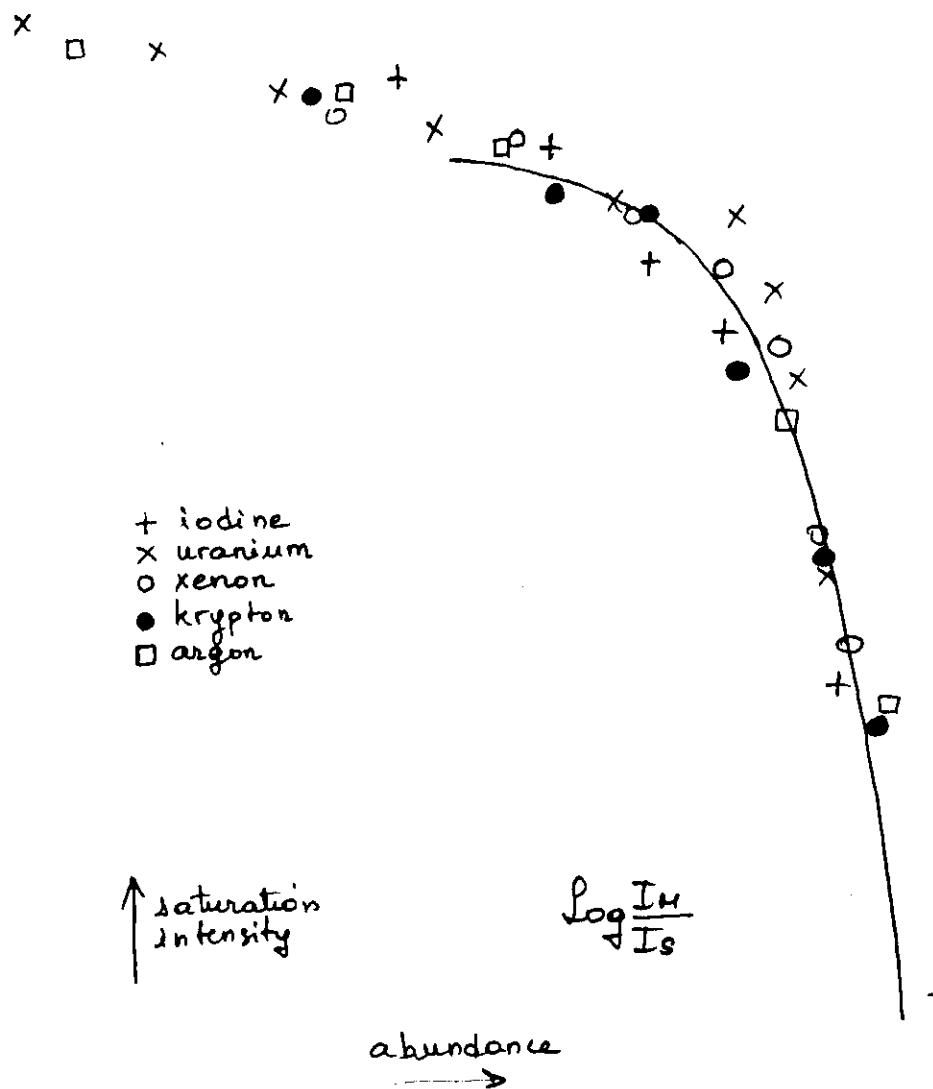
$$\text{number of ions } A^{2+} A^{3+} \dots \propto \log \frac{I_M}{I_2}$$

cumulated abundances should fit $\log \frac{I_M}{I_s}$

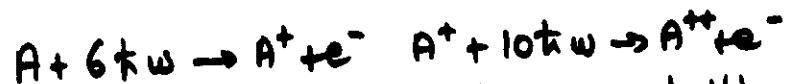
Theoretical ionization intensities

versus experimental abundances (from Bayer et al JOSAB 1 (1984) 1)

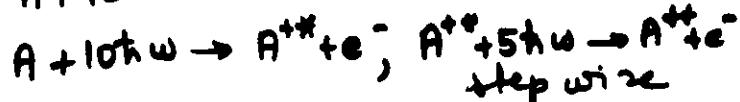
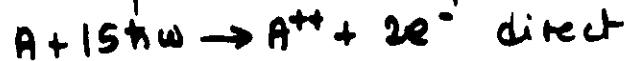
H.M.



Ratio A^{++}/A^+ at saturation
intensity of $A \rightarrow A^+$



The process needs 16 photons: negligible



Two possible processes involving 15 photons

	Xe	Kr	A	Ne
exp.	0.015	0.007	.006	.0015
th. direct	10^{-6}	$7 \cdot 10^{-8}$	$3 \cdot 10^{-9}$	$1.4 \cdot 10^{-15}$
th. stepwise	.026	.005	.01	<u>$4 \cdot 10^{-8}$</u>

= Resonance $Ne^+ + 9\hbar\omega$
detuned by 174 cm^{-1}

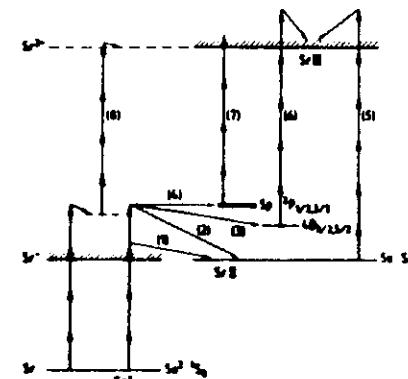
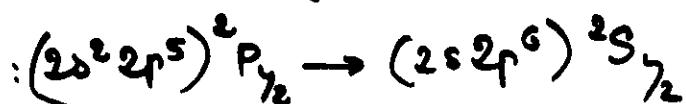


FIG. 1. Energy-level diagram of Sr I and Sr II showing the processes discussed in the text. The process labeled (6) is an example of "direct" or "nonresonant" double ionization.

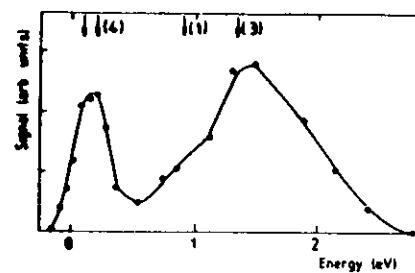


FIG. 2. Typical electron-energy spectrum at 100 GW cm^{-2} . The numbers inside parentheses refer to the processes illustrated in Fig. 1.

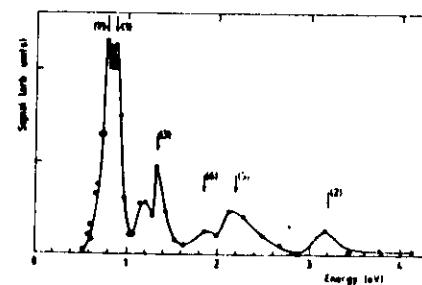


FIG. 4. Typical energy spectrum at 2.8 GW cm^{-2} showing the 2.2-eV component (arrow) corresponding to the second ionization. The numbers in parentheses refer to the processes illustrated in Fig. 1.

Agostini Letite PRA 32 3800 (1985)

Harmonic generation

For an atom

$$\langle \Psi(t) \rangle = a_0 |0\rangle + \sum \alpha_\alpha^{(1)} |\alpha\rangle + \dots + \sum \alpha_\alpha^{(2p)} |\alpha\rangle + \dots + \sum \alpha_\alpha^{(2p+1)} |\alpha\rangle$$

same parity
 as $|0\rangle$
 opposite
 parity

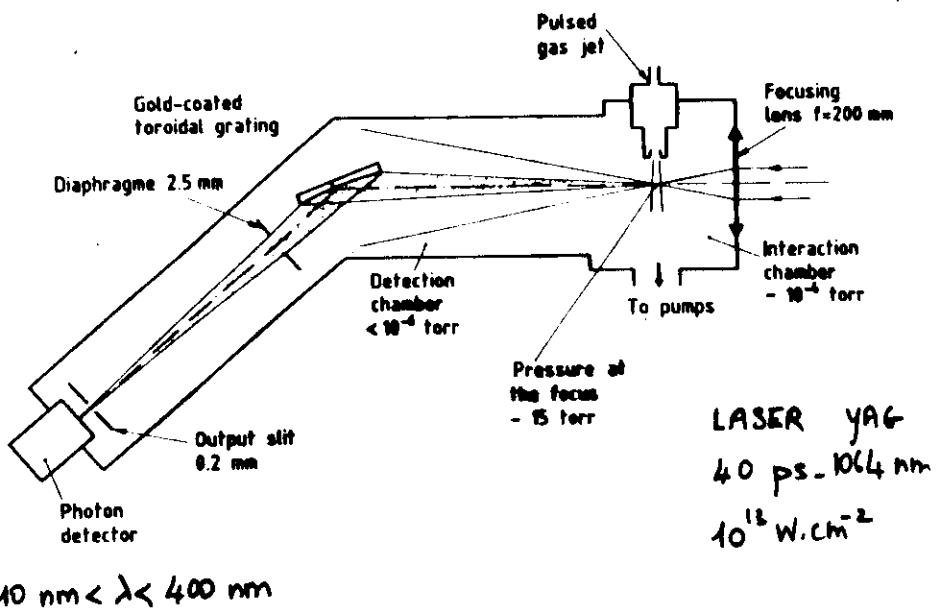
polarisation $\langle \Psi(t) | \vec{z} | \Psi(t) \rangle$

polarisation at frequencies $(2p+1)\omega$
 with leading term $a_0^* a_\alpha^{(2p+1)} \propto \delta^{2p+1}$

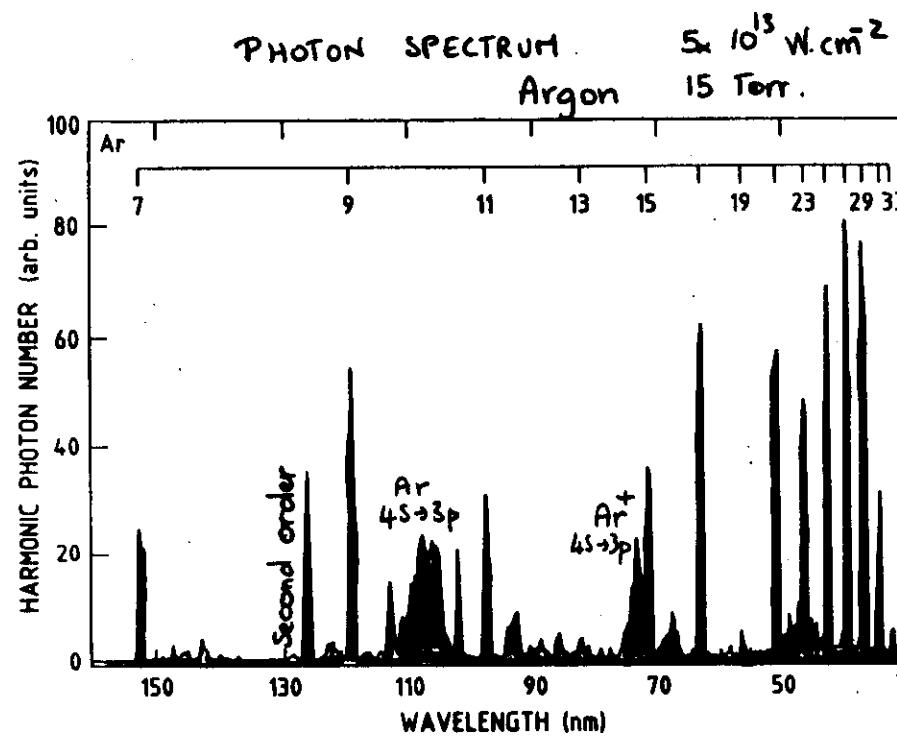
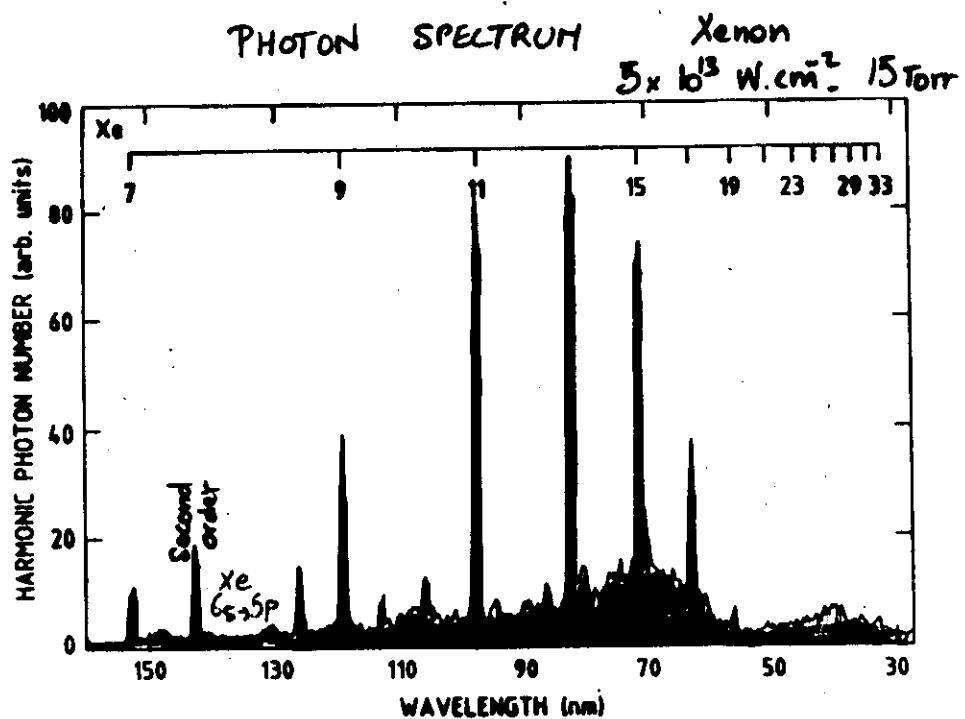
Starting point: Observation of multicharged ions:
 What states are reached? \rightarrow look at fluorescence
 Need to increase the pressure -
 Does the process involve atoms or ions?

Open question: phase matching conditions.

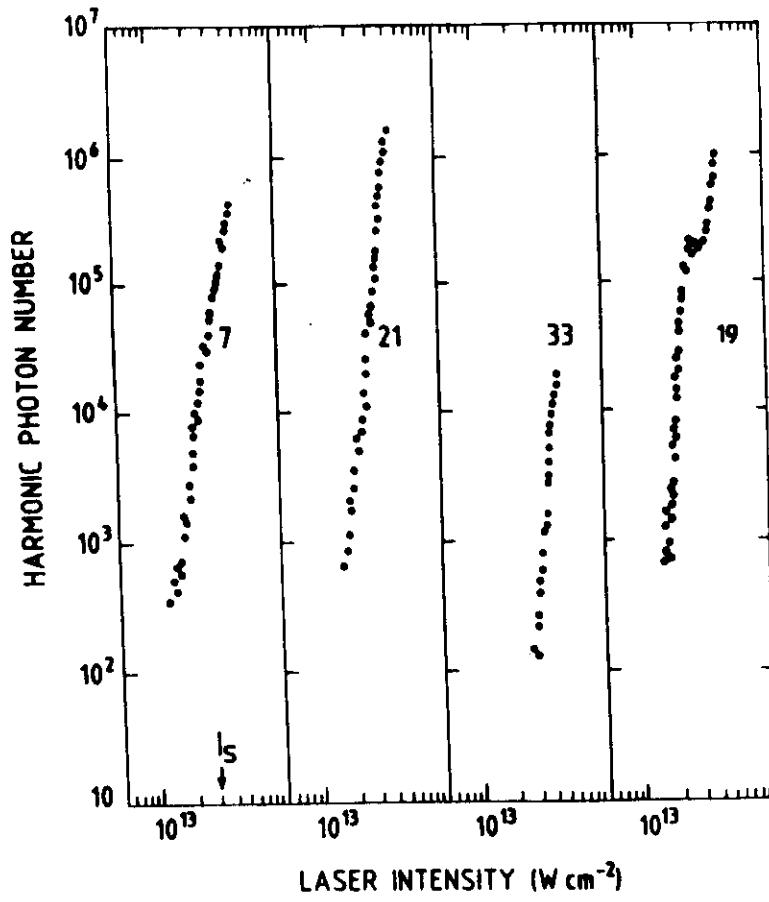
EXPERIMENTAL SET UP



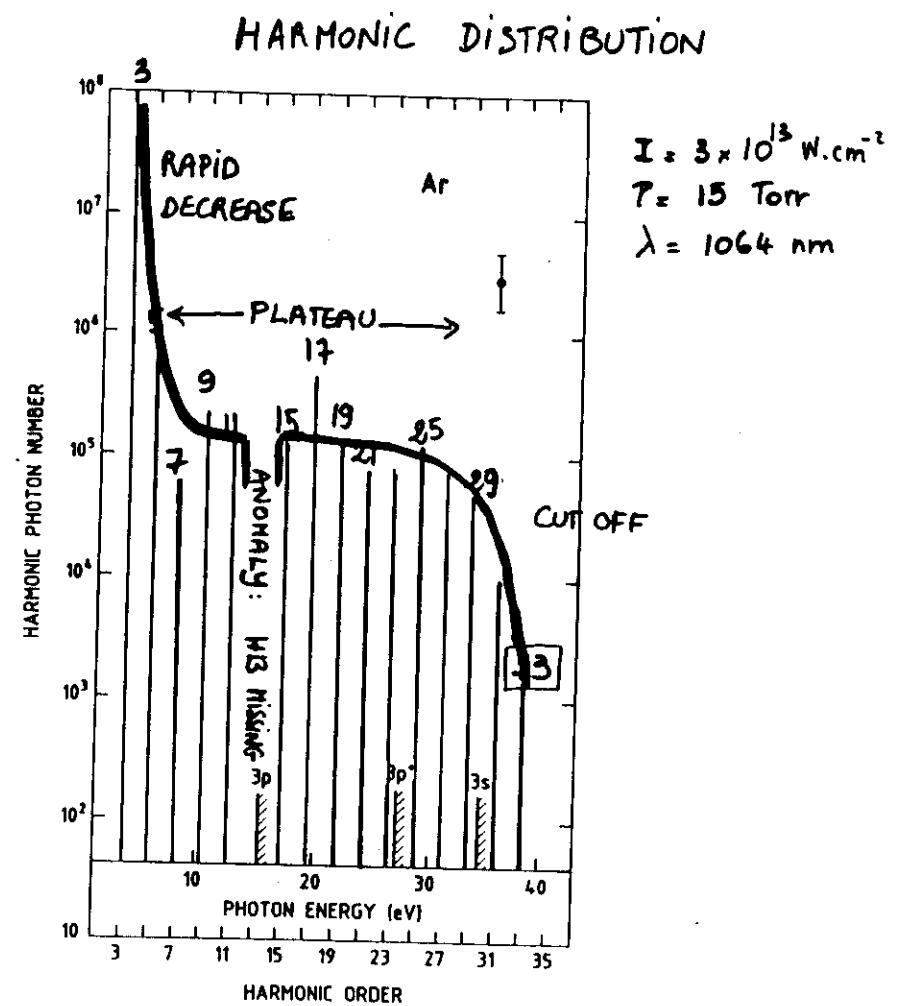
L'Huillier et al 1988

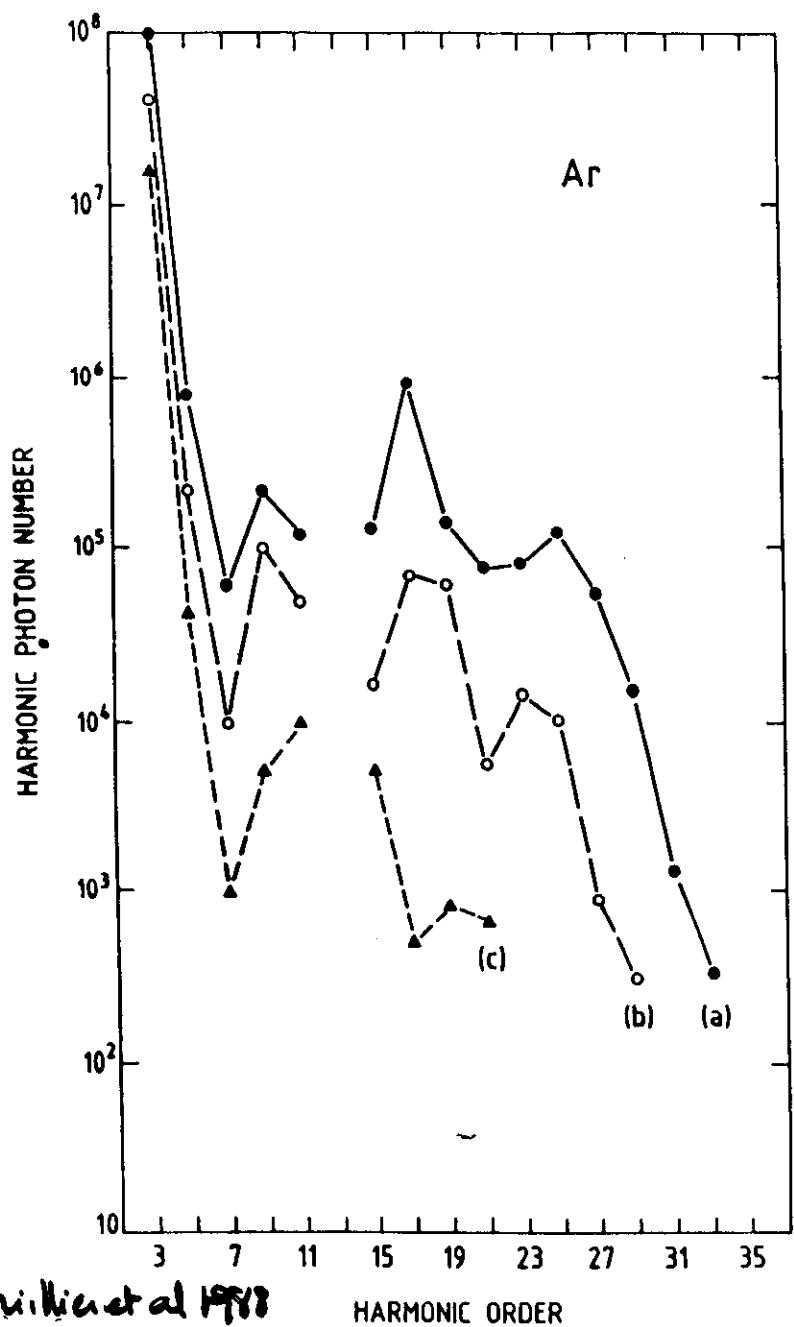


ODD HARMONICS
 FLUORESCENCE (RECOMBINATION)
 CONTINUOUS BACKGROUND



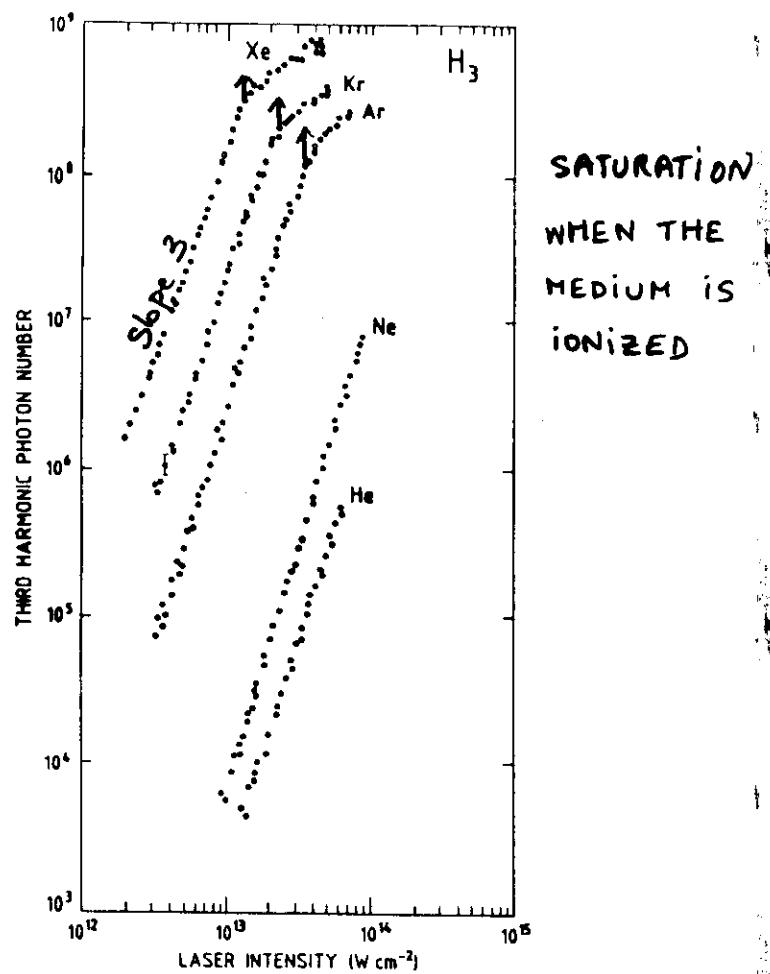
L'Huillier et al 1988





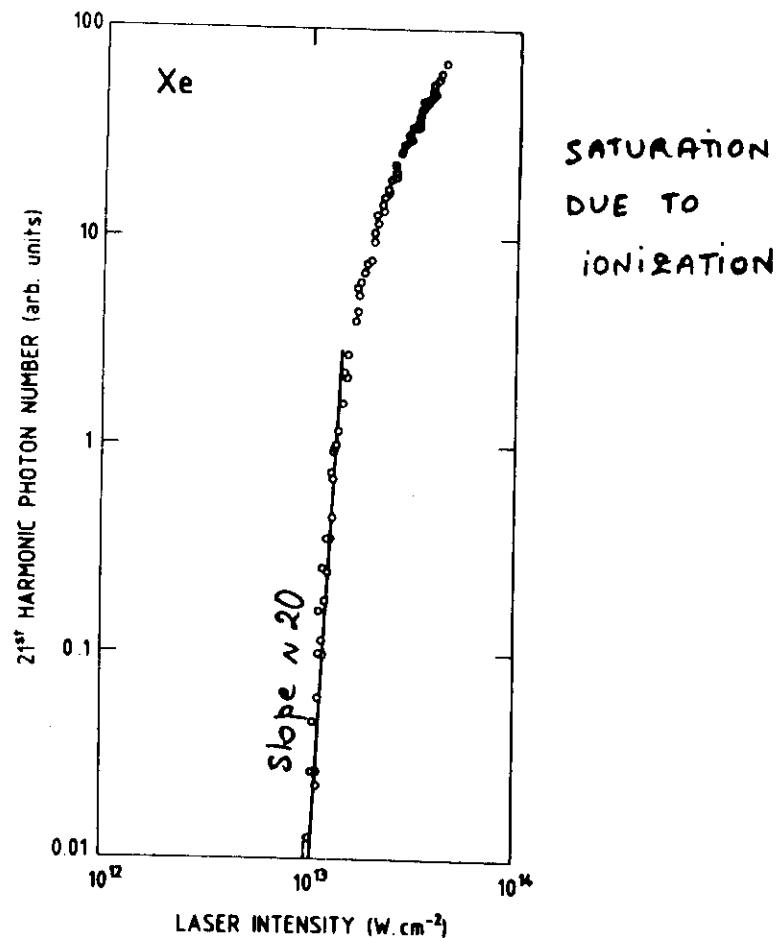
L'Huillier et al 1988

THIRD HARMONIC GENERATION,



L'Huillier et al 1989

HIGH-ORDER HARMONIC GENERATION



L'Huillier et al 1989

