



INTERNATIONAL ATOMIC ENERGY AGENCY
UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION
INTERNATIONAL CENTRE FOR THEORETICAL PHYSICS
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SMR/455 - 3

**EXPERIMENTAL WORKSHOP ON HIGH TEMPERATURE
SUPERCONDUCTORS & RELATED MATERIALS
(BASIC ACTIVITIES)**

12 - 30 MARCH 1990

METAL OXIDES: NORMAL-STATE PROPERTIES

LECTURE I

PRELIMINARIES

MAIN-GROUP AND RARE-EARTH OXIDES

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METAL OXIDES: NORMAL-STATE PROPERTIES

LECTURE I

PRELIMINARIES

MAIN-GROUP AND RARE-EARTH OXIDES

SUPERCONDUCTIVITY

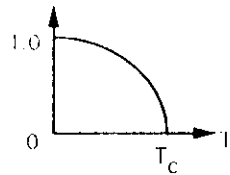
A many-body phenomenon - but it manifests a condensation from a metallic normal state where single-particle theory is applicable

OXIDES

	Insulators	Metals
Single-Valent:	MgO, EuO, MnO	GdO, TiO
	SrO_2	VO_2 PbO_2
Mixed-Valent:	$\text{Mn}^{2+}[\text{Mn}_2^{3+}]\text{O}_4$	
	$\text{Li}[\text{Mn}^{4+}\text{Mn}^{3+}]\text{O}_4$	
	$\text{Fe}^{3+}[\text{Fe}^{3+}\text{Fe}^{2+}]\text{O}_4$	
	$\text{Li}[\text{V}_2^{3.5+}]\text{O}_4$	
		$\text{Li}[\text{Ti}_2^{3.5+}]\text{O}_4$

COMPETING ENERGIES

$$\xi = \Delta / \Delta_0$$



HEAT: $T\Delta S$
Controls solid-state transition
e.g. Superconductor transition

INTRAAATOMIC: U or U' , $V_{LS} = \lambda L \cdot S$
(Present in atom, weakened in solid)
 U = on-site correlation energy
 $\sim (e^2/r_{ij}) \exp(-\gamma r_{ij})$

V_{LS} = multiplet splitting

INTERATOMIC: $b_{ij} = \langle \Psi_i, H' \Psi_j \rangle = \epsilon \langle \Psi_i, \Psi_j \rangle$
 b_{ca} = cation-anion (gives Δ_c, Δ_{nc}),
covalent-mixing parameter $\lambda = b_{ca} / \Delta E$
 b_{cc} = cation-cation $\sim \epsilon \exp(-R/\rho)$
 R = cation-cation separation
 b_{cac} = cation-anion-cation $\sim \epsilon \lambda^2$

ELECTRON-LATTICE: $\Delta_{nc}, \Delta_{SDW}, \Delta_{CDW}, \epsilon_p$

Jahn-Teller Δ_{nc} removes orbital degeneracy
CDW or SDW changes translational period.
(commensurate and incommensurate)
small polaron = "dressed" electron

SINGLE-VALENT COMPOUNDS

(Tight-binding bandwidth: $W \approx 2zb$)

Localized Electrons

Itinerant Electrons

Theory:

crystal-field

(Hubbard)*

Band

n-fold configurations

quasiparticles

electrons/holes

$$U \gg W$$

$$U \approx W$$

$$U \ll W$$

$$4f^n \longleftrightarrow d \longleftrightarrow s, p$$

Properties: μ_A, J_{ij}

magnetic order $T < T_c, T_N$

SDW/CDW

BCS supercond.

χ_m : Curie-Weiss

enhanced

Pauli

Bravais ^{sub} lattice: insulator

s.c./metal

metal

Examples: EuO, MnO

V₂O₃

TiO

* For non-degenerate, half-filled band

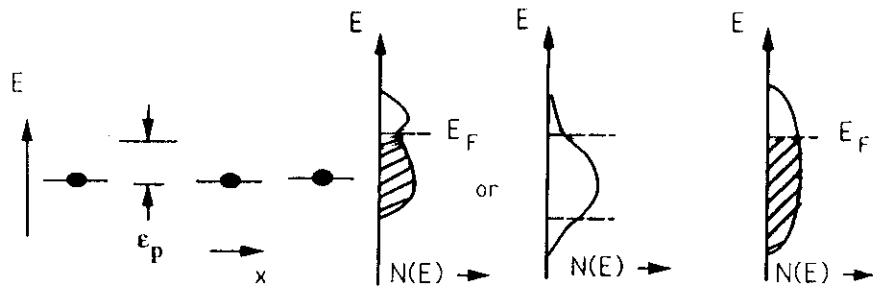
MIXED-VALENT COMPOUNDS

(ω_R^{-1} = Period of trapping lattice vibration)

(τ_h = Interatomic-hopping time)

Small Polarons

$$\tau_h \gg \omega_R^{-1}$$



Itinerant Electrons

$$\tau_h \approx \omega_R^{-1}$$

$$\tau_h \ll \omega_R^{-1}$$

Mobility:

$$\mu = eD/kT$$

$$\mu = eD_0/kT$$

or variable-range hopping

or short-range charge/spin fluctuation

$$\mu = e\tau_s/m^*$$

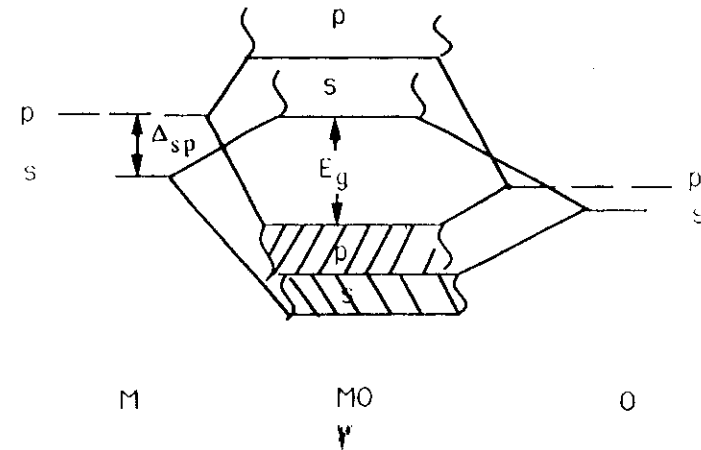
MAIN-GROUP OXIDES

Band Theory Applies

Oxygen: 2s, 2p valence electrons

Metal: only s, p valence electrons outside closed shells.

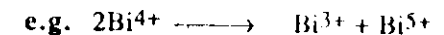
Non-Bravais Lattice (M and O distinguishable atoms)

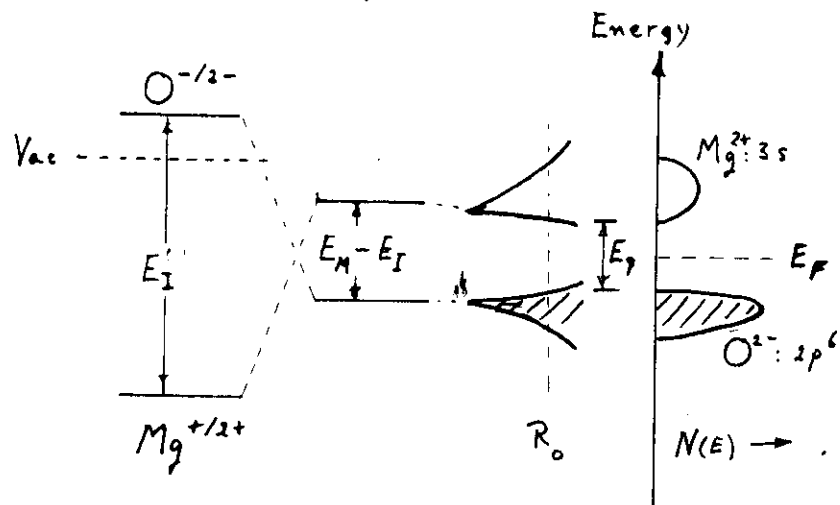
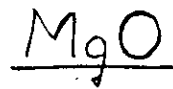


Group-A Metals: Oxides are insulators (large E_g).

Conduction and valence bands inaccessible; introduction of native defects energetically favored.

Group-B Metals: 5s or 6s conduction bands are accessible to electrons (large Δ_{sp}), but disproportionation normally favored over metallic behavior.

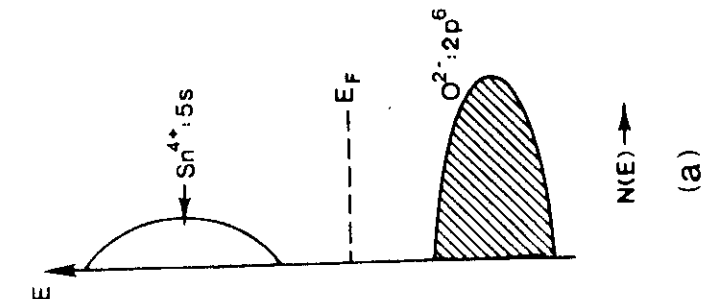
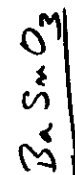
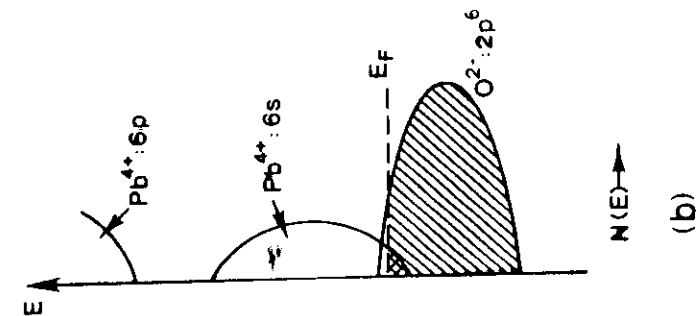
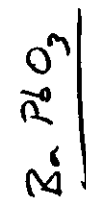
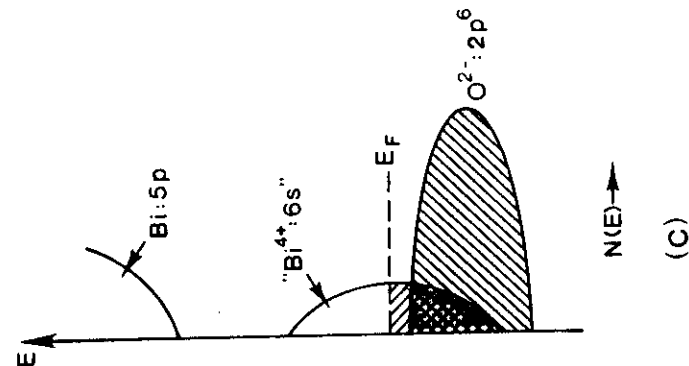
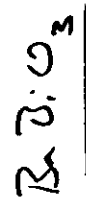


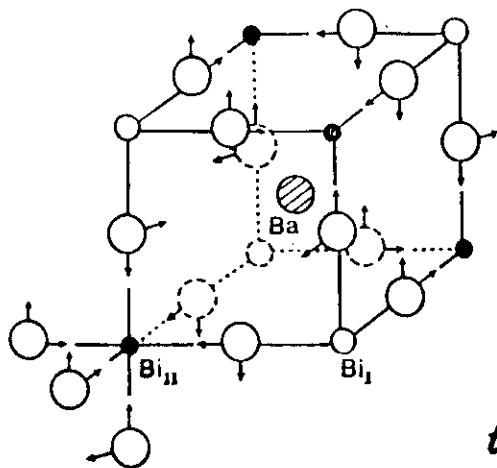


Free Ion Point charge

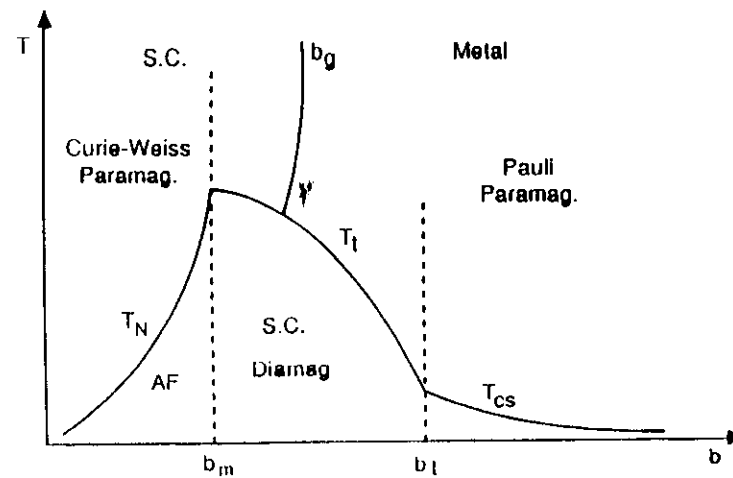
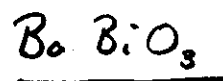
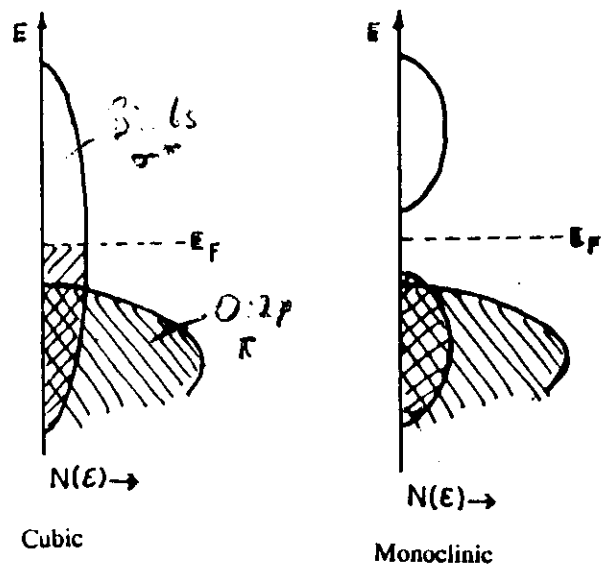
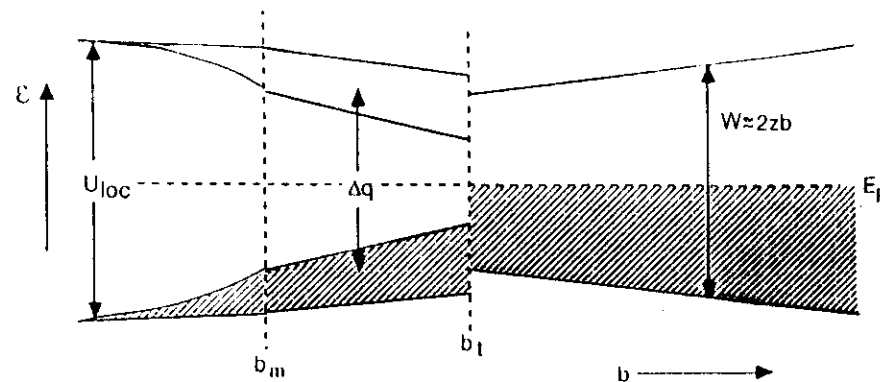
$$E_m = N \sum_i \frac{q_i^2}{r_i^2} = \text{Madelung Energy}$$

E_I - free ion charge transfer energy

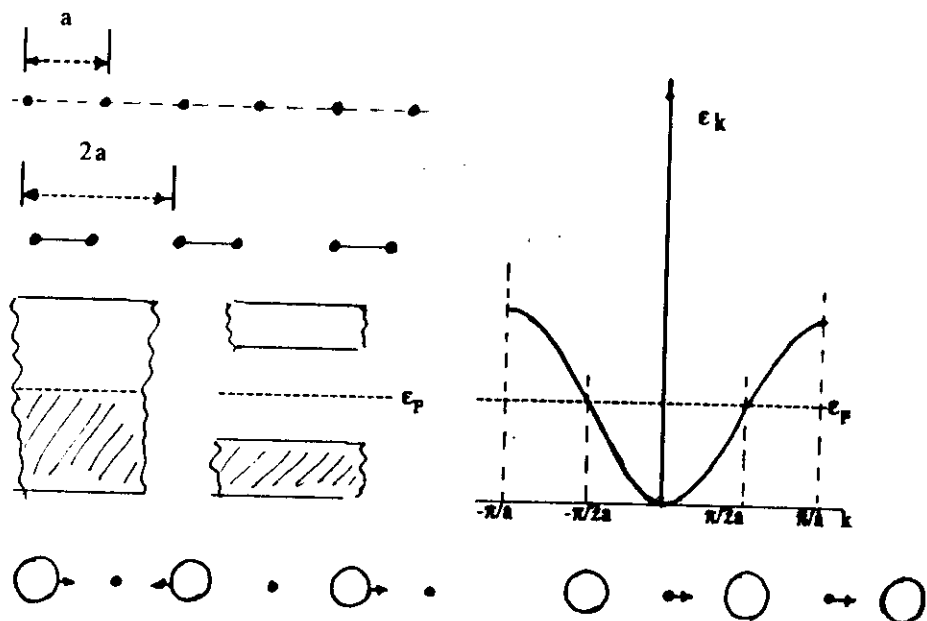




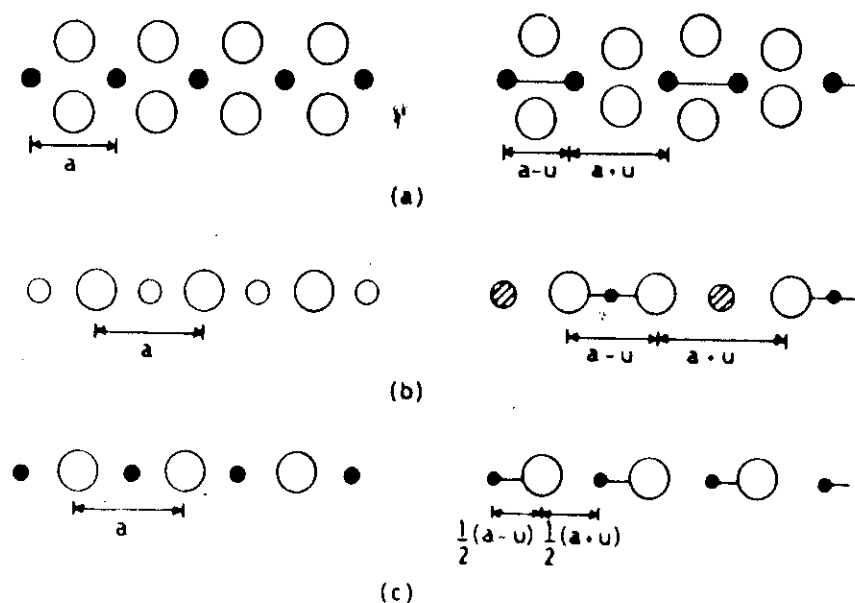
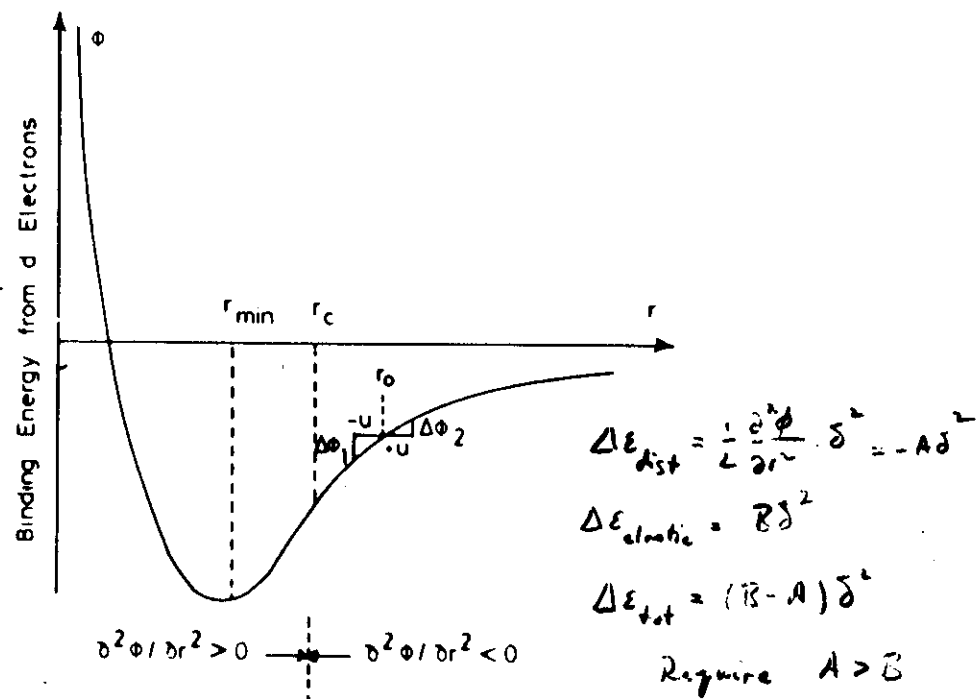
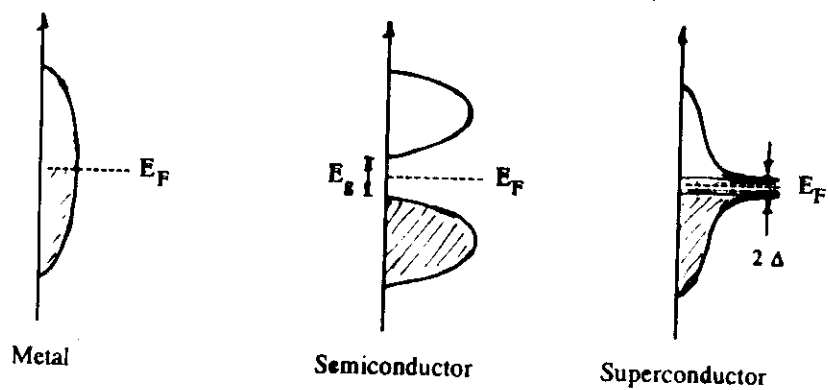
$$t = \frac{r_{Ba} + r_O}{f_2(r_{Bi} + r_O)}$$



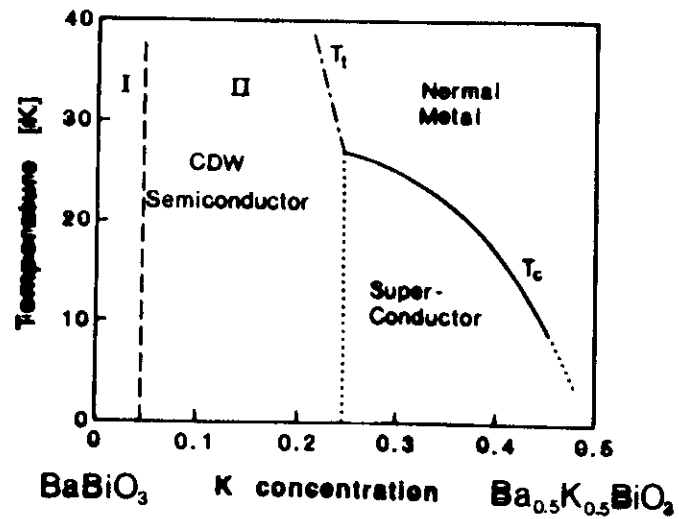
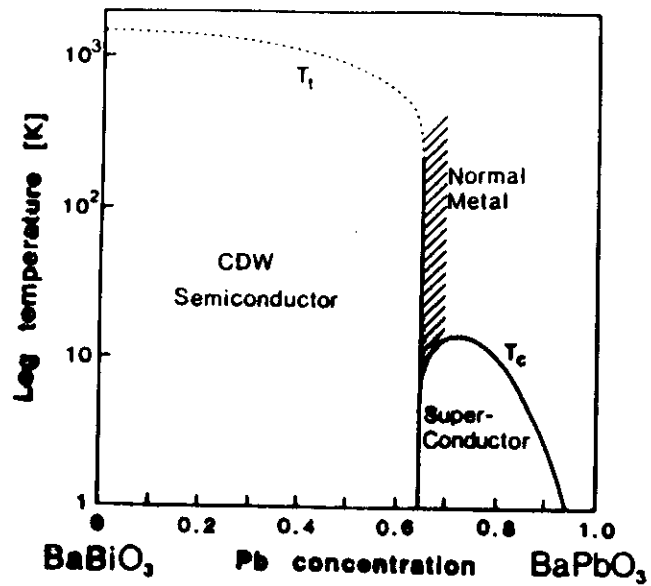
SOFT-MODE INSTABILITIES & SPIN-DENSITY WAVES



Note: May be Commensurate or Incommensurate



RARE-EARTH OXIDES



4fⁿ Hamiltonian

$$H = H_0 + V_{el} + \Delta_{LS} + \Delta_C + H_Z$$

$$\mu_J = gJ\mu_B \text{ with } g = 1 + \frac{J(J+1) + S(S+1) - L(L+1)}{2J(J+1)}$$

$$V_{el} \rightarrow U > E_g$$

γ

