



INTERNATIONAL ATOMIC ENERGY AGENCY
UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION
INTERNATIONAL CENTRE FOR THEORETICAL PHYSICS
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H4-SMR 471/14

COLLEGE ON MEDICAL PHYSICS

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REFERENCES ON ELECTRETS

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References on Electrets

(XII)

- 1 - Electrets, Topics in Applied Physics, Vol 33, ed. G.M. Sessler, Springer-Verlag, 1980
- 2 - Bauser, H and Rungge W., Health Physics 34 (1978) 97
- 3 - Marcacenas S. and R.L. Zimmerman, in Xth Ann. Conference of the Ass. Med. Phys. Brazil S. Paulo, ed. S. Watanabe 1979, p. 488-491
- 4 - CAMERON J. & Marcacenas S., Nucl. Instr. and Methods 175 (1980) 117
- 5 - Crivinel P. and Marcacenas S., Physica Medica, 3 (1988) 207
- 6 - Crivinel P, Marcacenas S. and Cameron J., Nucl. Instr. and Methods A287 (1990) 580
- 7 - Several Ph.D. and Ma. Sci. Thesis from the Institute of Phys. & Chemistry, USP, SCARLOS, SP, Brazil, 13560 under orientation of S. Marcacenas.

DOSIMETRY with electrets I

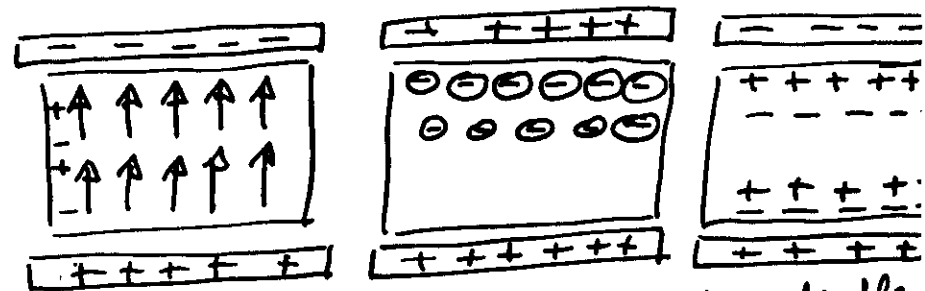
1- WHAT IS AN ELECTRET?

AN INSULATOR (DIELECTRIC) WITH A LONG TERM POLARIZATION \vec{P} (FROM MINUTES TO CENTURIES).

2- SOURCES OF \vec{P} :

- dipoles (OH^- , OH_2 , dipolar molecules)
- ions (H^+ , Na^+ , K^+ etc)
- electrons or holes (e^- or e^+)
- defects (M center, vacancies, interstitials)

3- Distribution of sources for \vec{P}



Dipoles

injected space charge

ion-double layer

4- CHARGING METHODS

(II)

THERMO ELECTRET

CORONA ELECTRET

e^- injection ELECTRET

FOTO ELECTRET

RADIO-ELECTRET

TRIBO ELECTRET

ELECTROLYSIS-ELECTRET

NATURAL ELECTRETS (PHASE-
-TRANSITION OR COSTA RIBEIRO effect,

5- MATERIALS

POLYMERS (TEFLON, P.E., PVC etc)

IONIC CRYSTALS (LiF, NaCl, CaSO₄ etc)

glasses

ice

CERAMICS

HIGH-INSULATION SEMICONDUCTORS
(SiO₂ and others)

6 - BIOELECTRETS

BIOPOLYMERS (PROTEINS, DNA, RNA,
POLYSACCHARIDES)

TISSUES (BONES, TEETH, HAIR etc)

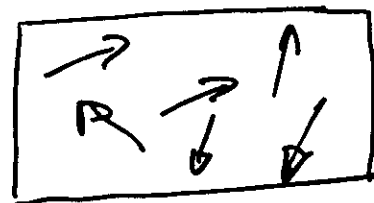
BOUND-WATER + AMINO ACIDS

7- EXPERIMENTAL TECHNIQUES

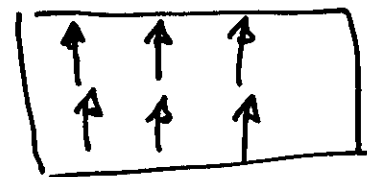
(III)

FOR MEASUREMENTS

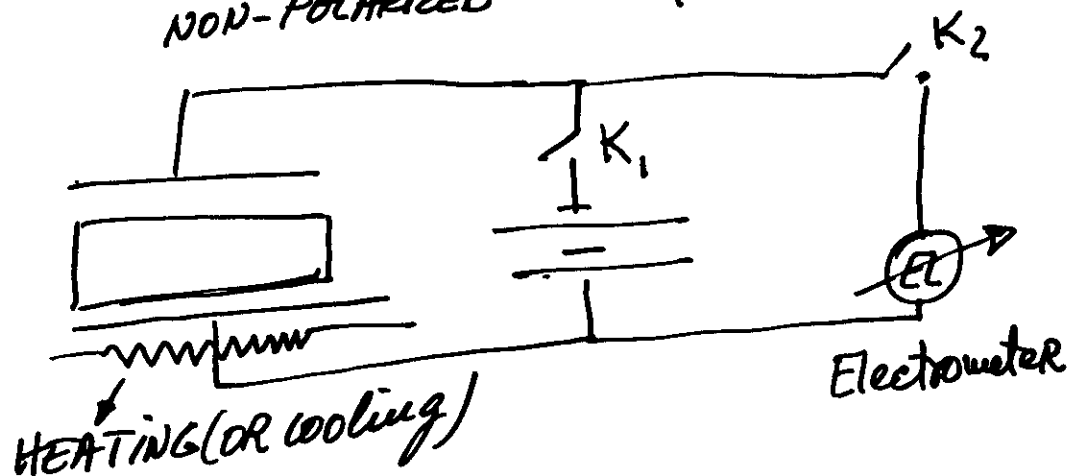
Thermal Stimulated Polarization (TSP)
Thermal Stimulated Depolarization (TSD)



NON-POLARIZED



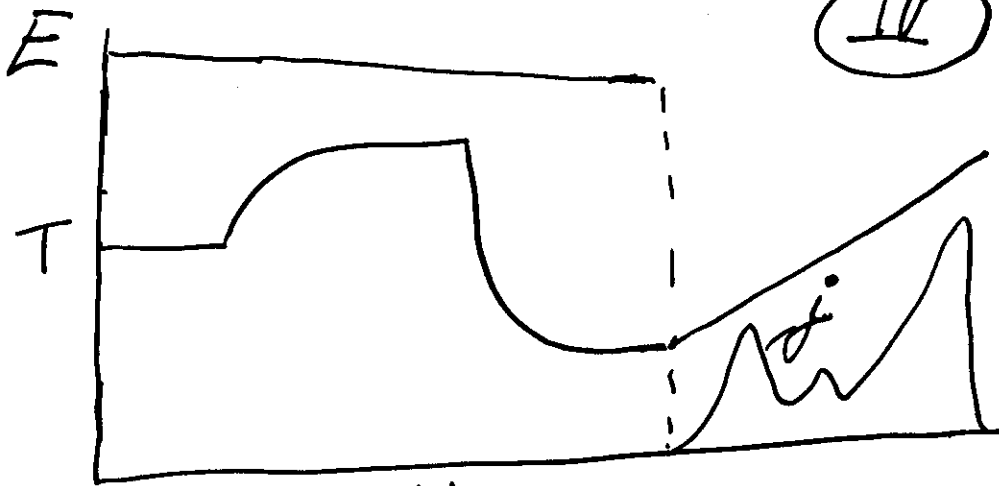
POLARIZED



Polarization - K₁ ON K₂ off

Depolarization - K₁ off K₂ on

(IV)

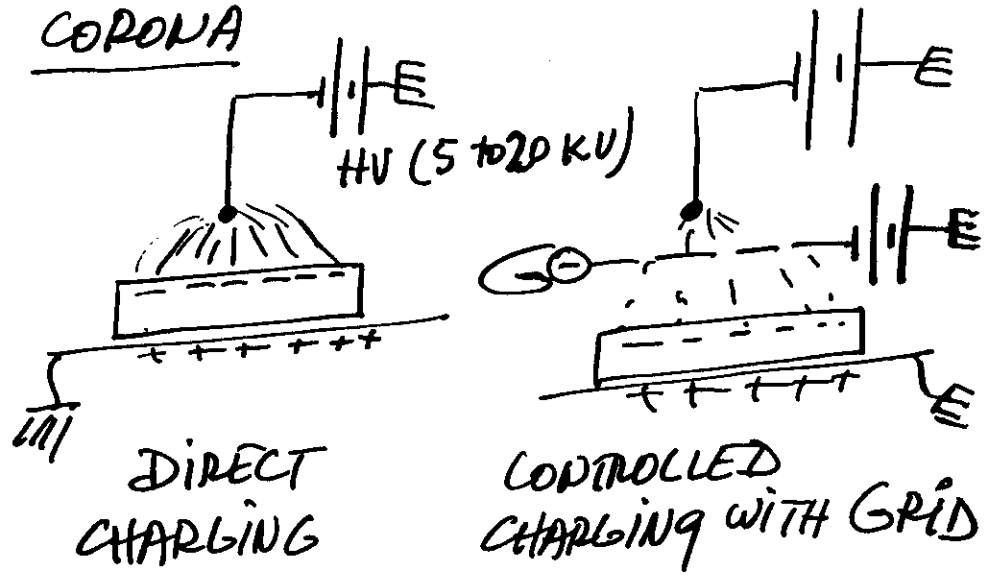


→ TIME

TYPICAL CYCLE FOR TSP AND TSD

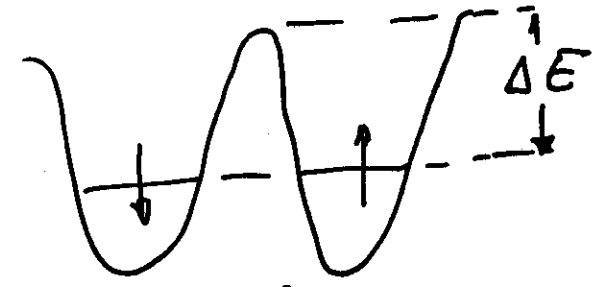
$j = -\frac{d\vec{P}}{dt}$ depolarization current-density

CORONA



8- RELAXATION TIMES FOR (V)
ELECTRET DISCHARGE

$\tau = \tau_0 \exp + \Delta E / kT$



$\tau_0 \sim 10^{13}$ sec

$\Delta E \sim 0.1$ to 2 eV

For teflon for ex. 10-20 years at R, depending on purity and environment.

9- Applications:

- condenser microphones
- ion-chamber dosimeters
- air and pollution filters
- biophysics & medical physics
- and others.

9 - Dosimetry with Electrets (VI)

a - using electret ionization - chambers

b - using TSD curves

c - using electret-transducers

10 - Electret Radiation Dosimeter (ERD) ion-chamber

The electret is used as:

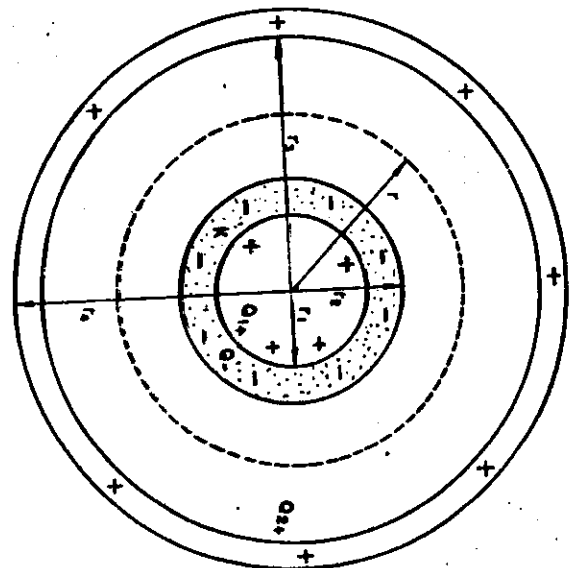
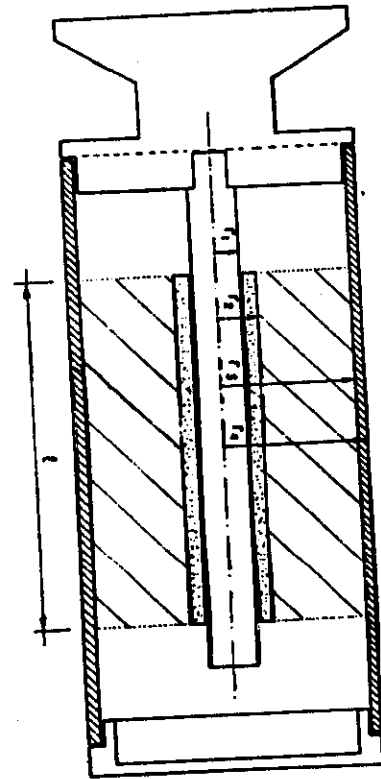
- electric field source

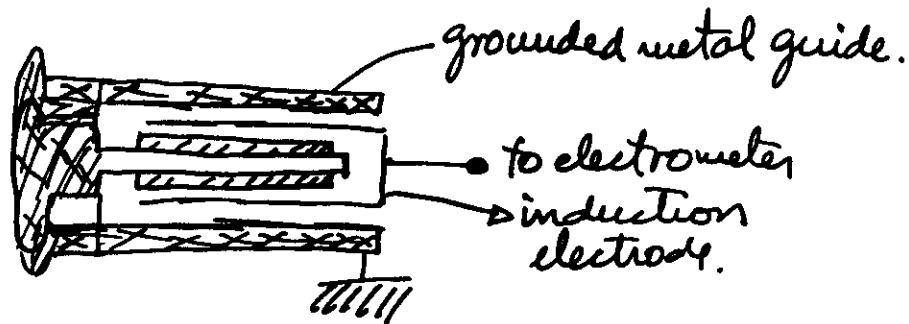
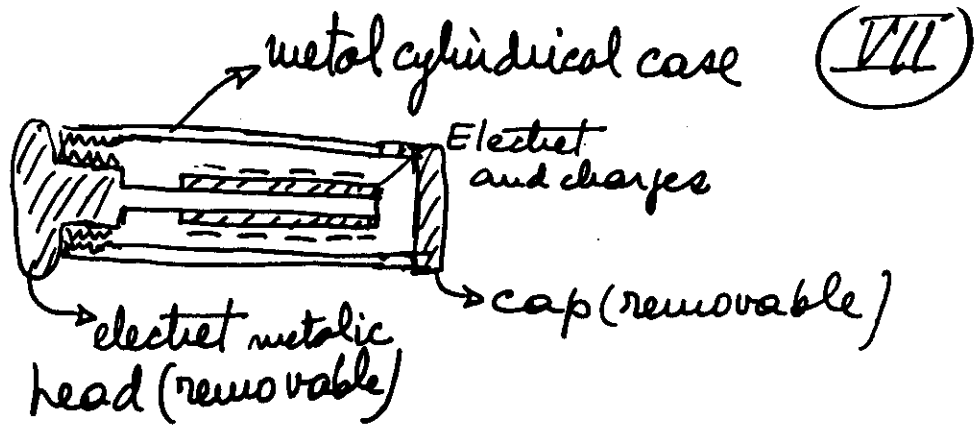
- charge detector

Simultaneously.

Marcarenhas and coll. ~ 1979
introduced the cylindrical ERD.

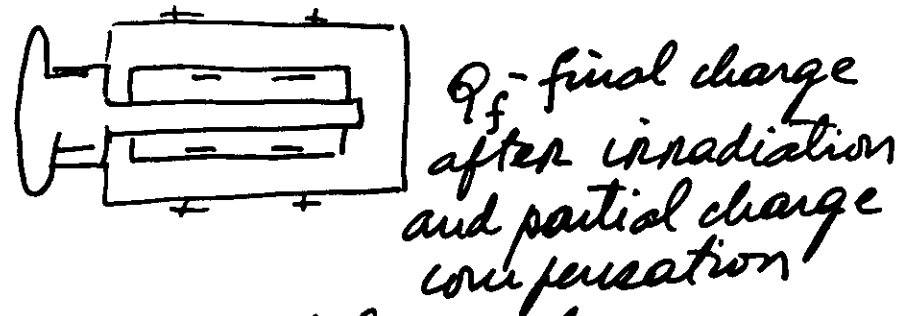
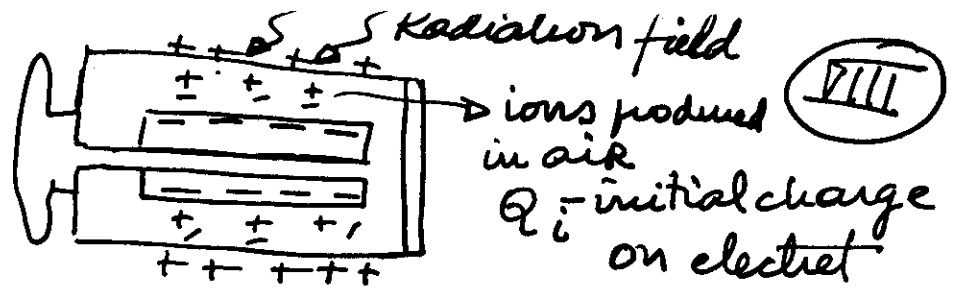
Figure 1 - Electret dosimeter with a cylindrical active volume
 $r_1 = 1,2 \text{ mm}$; $r_2 = 1,45 \text{ mm}$; $r_3 = 7,5 \text{ mm}$;
 $r_4 = 8,0 \text{ mm}$; $l = 30,0 \text{ mm}$





Typical quantities and dimensions

- electret thickness \rightarrow 0.1 to 2 mm
- electret length \rightarrow 4 to 10 cm
- internal radius \rightarrow 2 - 4 mm
- external radius \rightarrow 3 - 5 mm
- electret charging potential (teflon) \rightarrow -100V to -1000V
- electret charge \rightarrow $\sim 10^{-6} - 10^{-9}$ coul/cm²
- ion chamber volume \rightarrow $\sim 10 - 100$ cm³
- typical range \rightarrow 0.1 - 10 R



Fundamental equation:

$$\underline{Dose} \sim Q_i - Q_f$$

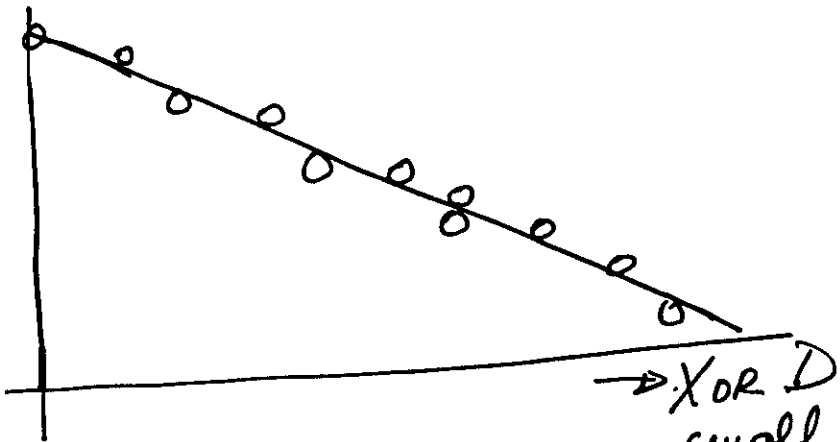
$$\text{or } D = K(Q_i - Q_f)$$

$K \rightarrow$ depends on material, geometry, radiation quality just like for any ion-chamber; also:

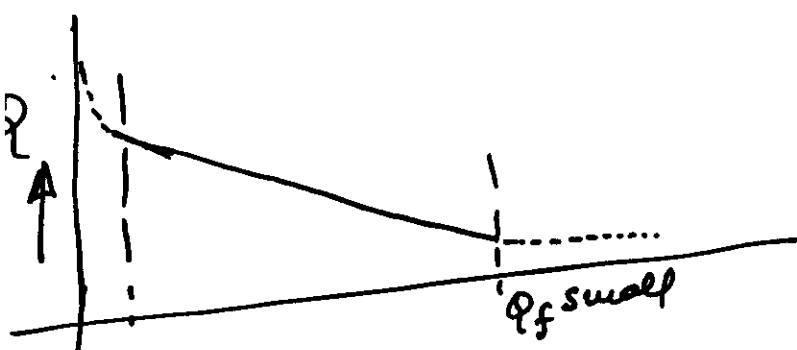
If sufficient charges are maintained for chamber electric field to maintain proper value (no-recombination) and initial charge is not above critical field (ion multiplication).

Typical calibration curves for ERD

(IX)

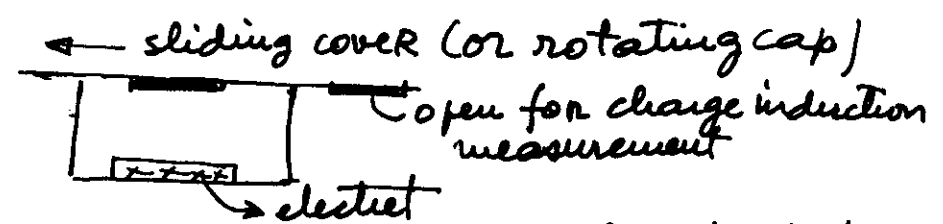


If Q_i very large or Q_f very small



typical values $K = \frac{50 \text{ mGy}}{2 \cdot 10^{-9} \text{ C}} \approx 25 \text{ mGy/C}$
 for typical pocket pen-dosimeter
 electret chamber minimum dose detectable
 of the order of $\pm 5 \times 10^{-5} \text{ Gy}$ (5 mRads)

For higher doses (.1 - 1 Gy) (\bar{X})
 a planar ERD was built
 with high mechanical precision
 and small volumes and special
 charge compensation measurement
 device.



guard-rings were used on electret detector
 for field homogeneity and precision of meas.
 Other results found with ERD and
 other ideas:

- neutron ERD (fast and thermal) using proper walls
- electron ERD using thin mylar walls
- α -ERD using source inside dosimeter
- X and γ -ray energy measurements using tandem ERD

MANY new ideas to explore

Advantages of ERD

(VI)

- Reading does not erase the dosimeter
- Re-usable
- simple to operate and read
- inexpensive
- good stability and small fading ($< 1\%$ in 30 days and $< 0.1\%$ in 30 days for special purposes)
- all the advantages of ion-chamber (well known theory, flexibility of design and use of wall materials)
- fast reading
- low cost reader
- portable or computerized reader

Disadvantages:

- still needs to be tested intensively in personnel and ambient dosimetry
- has to compete with investments made in traditional techniques.