



INTERNATIONAL ATOMIC ENERGY AGENCY
UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION
INTERNATIONAL CENTRE FOR THEORETICAL PHYSICS
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SMR.550 - 26

SPRING COLLEGE IN MATERIALS SCIENCE ON
"NUCLEATION, GROWTH AND SEGREGATION IN MATERIALS
SCIENCE AND ENGINEERING"
(6 May - 7 June 1991)

TRANSPORT PROCESSES
(INCLUDING RADIATION ENHANCED DIFFUSION)

Part IV

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These are preliminary lecture notes, intended only for distribution to participants.

Diffusion in Amorphous Metallic Alloys

Motivation

- Example of diffusion in disordered solid media
- Metastability of amorphous state
- Crystallization is influenced by diffusion or even diffusion-controlled
(practical significance for thermal stability of amorphous materials)

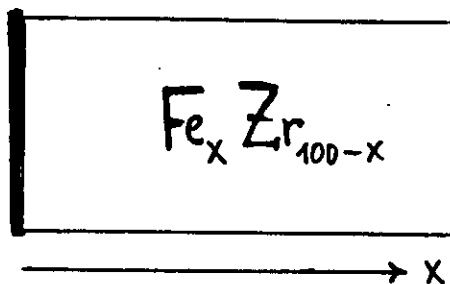
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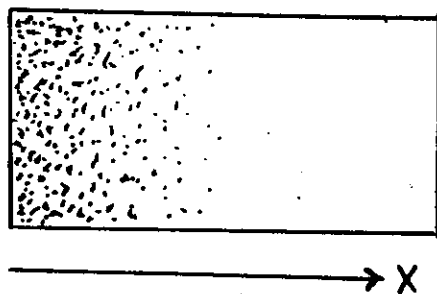
Diffusion experiments with radioactive tracer atoms

(J. Horváth)

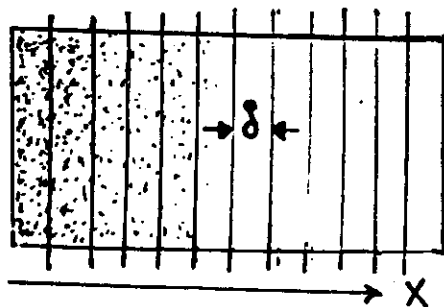
^{59}Fe
or
 ^{95}Zr



electro-chemical deposition of an approximately 1 nm thin tracer layer



annealing in a furnace at a temperature T for a duration t



sectioning and measurement of radioactivity of the individual sections

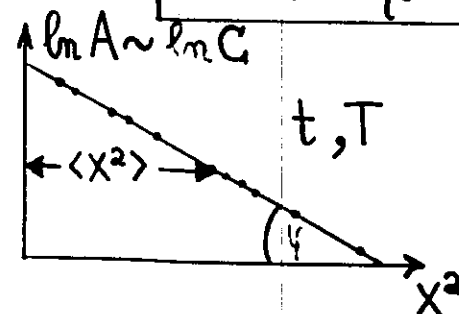
This talk is confined to
1) self-diffusion,

Determination of diffusion coefficients from penetration profiles

Diffusion equation: $\frac{\partial C}{\partial t} = D \nabla^2 C$

Thin-film solution ($d_T \ll \langle x \rangle$):

$$C(x,t) = \eta_0 (\pi Dt)^{-1/2} \exp(-x^2/4Dt)$$



$$|\tan \varphi| = 1/4Dt$$

- Cross-check: Data points must lie on straight line
- Procedure remains valid if D depends on t :

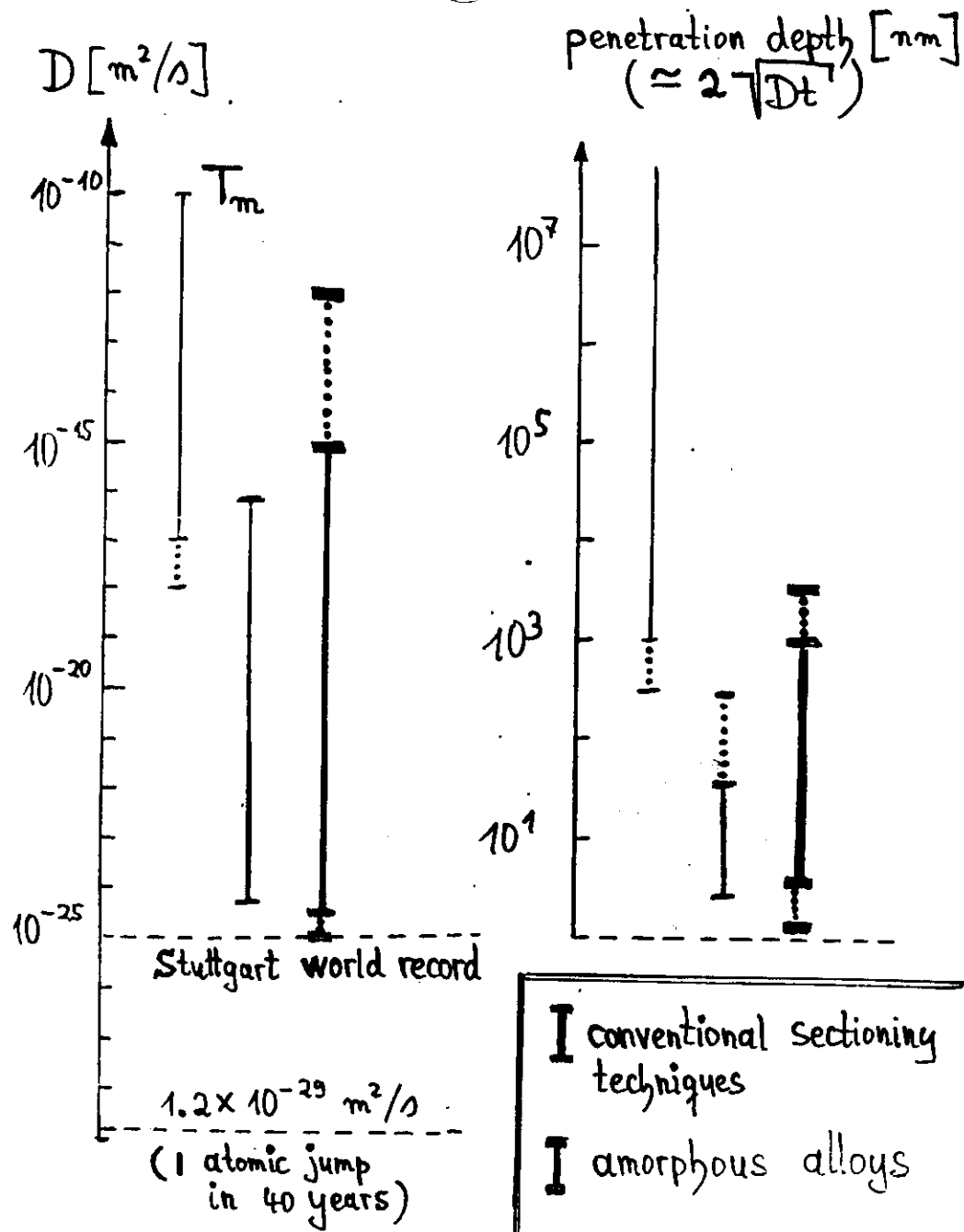
$$\langle D(t) \rangle \equiv t^{-1} \int_{t=0}^t D(t') dt'$$

$$D(t) = \langle D(t) \rangle + t \frac{d\langle D(t) \rangle}{dt}$$

- Mean penetration depth: $\langle x \rangle \approx \sqrt{\langle x^2 \rangle} = \sqrt{2 \langle t \rangle}$

- ⇒
- Measurement of small diffusion coefficients requires measurement of small penetration depths and/or long diffusion annealing times
 - Duration of annealing is limited by the

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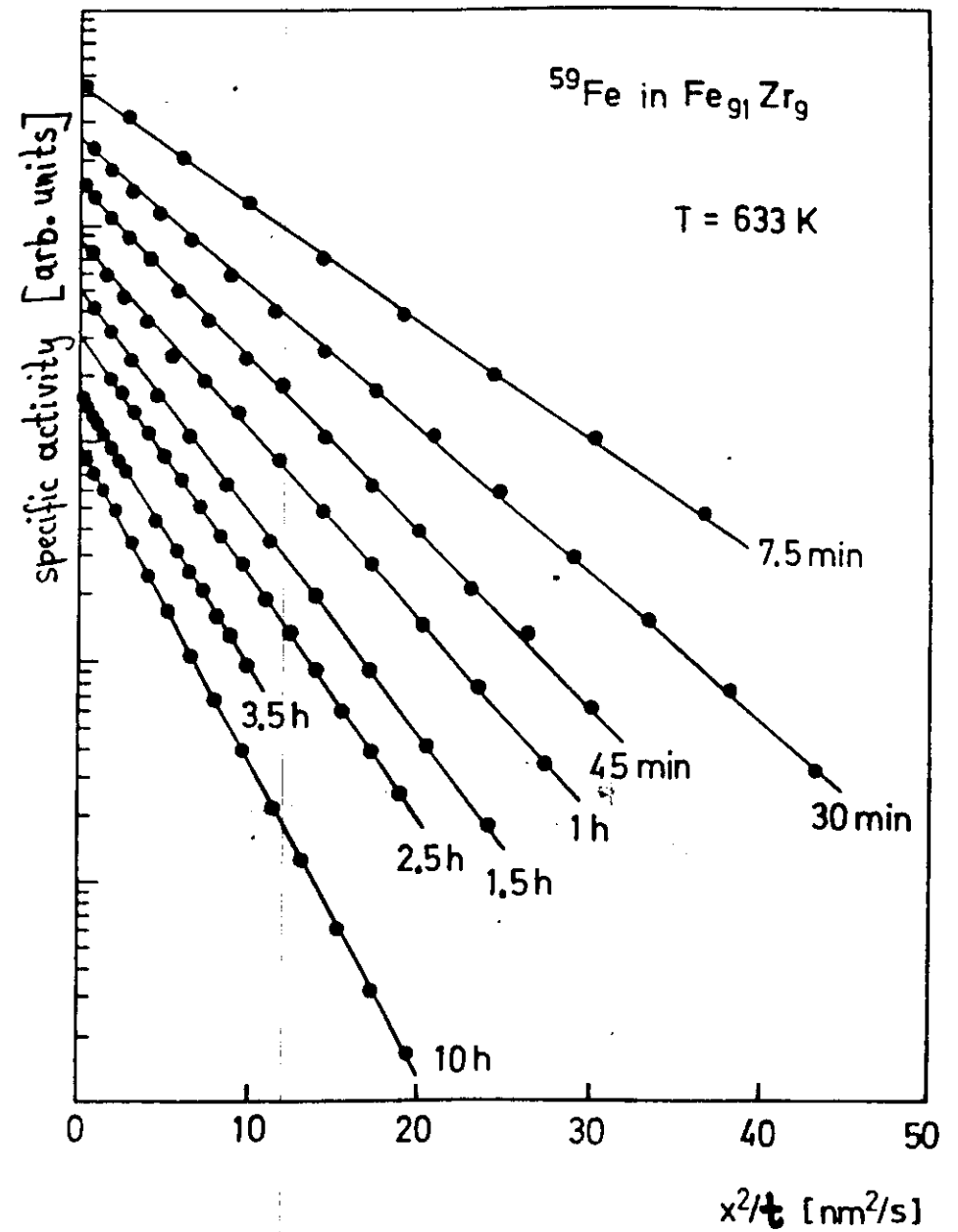
I conventional sectioning techniques

I amorphous alloys

I ion-beam sputtering
 (K. Maier)

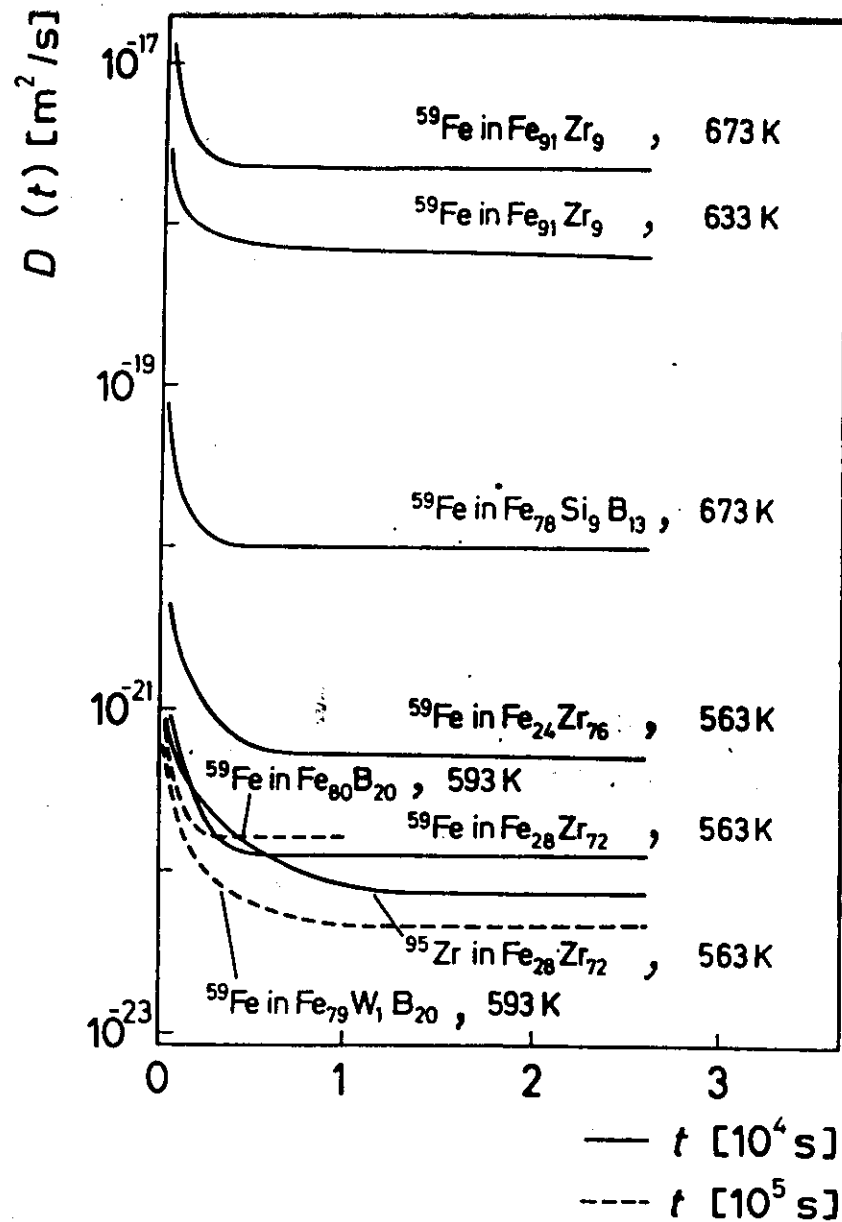
Diffusion profile

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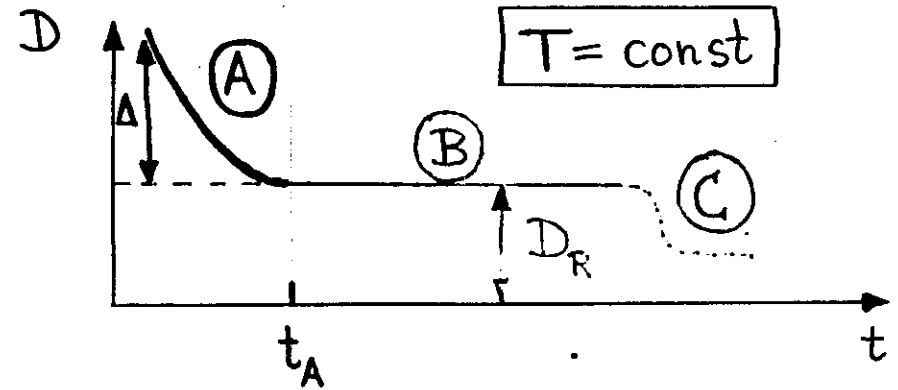


- Data points lie nicely on straight lines
- Diffusion coefficient decreases in the

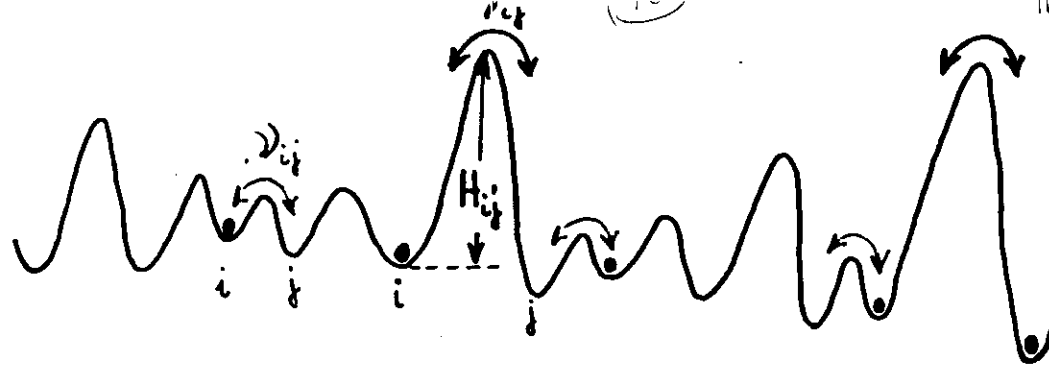
Isothermal decrease of diffusion coefficients



Schematic illustration and description of the isothermal change of the diffusivities in amorphous alloys



- ① Initial decrease (Δ, t_A) depends on prehistory of specimen; may be eliminated by pre-annealing
- ② Plateau value D_R is independent of prehistory; depends on diffusion temperature in a unique way
 - ➔ relaxed amorphous state
 - ≡ thermodynamically (fairly stable) metastable state
- ③ Decrease due to crystallization (crystallized state ≡ stable equilibrium)



Short t (low T):

- low barriers $H_{ij} \rightarrow$ high jump frequencies ν_{ij}
- local, independent processes: $\bar{\nu} \equiv \langle \nu_{ij} \rangle$
- effective activation enthalpy:

$$\boxed{Q^s \equiv -\frac{d}{d(1/kT)} \ln D^s = H_0 - H_m^2/kT}$$

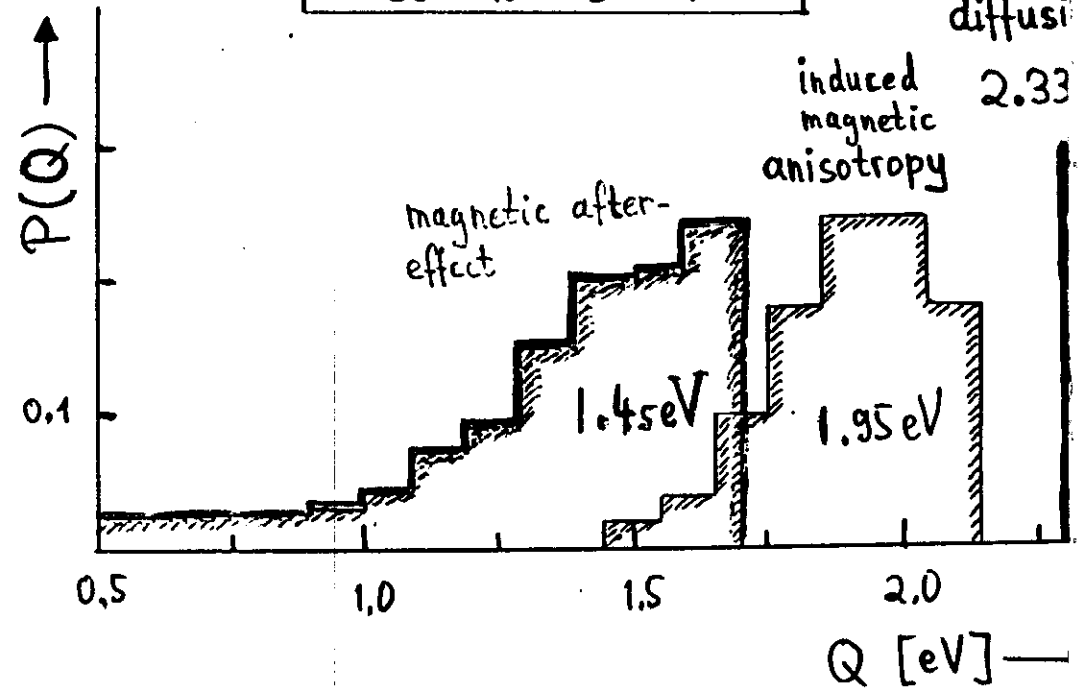
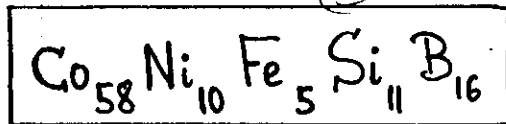
- two-level systems, (magnetic after-effect)

Long t (high T):

- high barriers $H_{ij} \rightarrow$ low jump frequencies ν_{ij}
- series of dependent processes: $1/\bar{\nu} \equiv \langle 1/\nu_{ij} \rangle$
- effective activation enthalpy:

$$\boxed{Q^l \equiv -\frac{d}{d(1/kT)} \ln D^l = H_0 + H_m^2/kT}$$

- (long-range) Diffusion



magnetic after-effect:

$$t = 180 \text{ s}, T \approx 400 \text{ K}$$

$$Q^s \equiv \langle Q \rangle \approx 1.45 \text{ eV}$$

induced magnetic anisotropy:

$$t \approx 10^5 \text{ s}, T \approx 500 \text{ K}$$

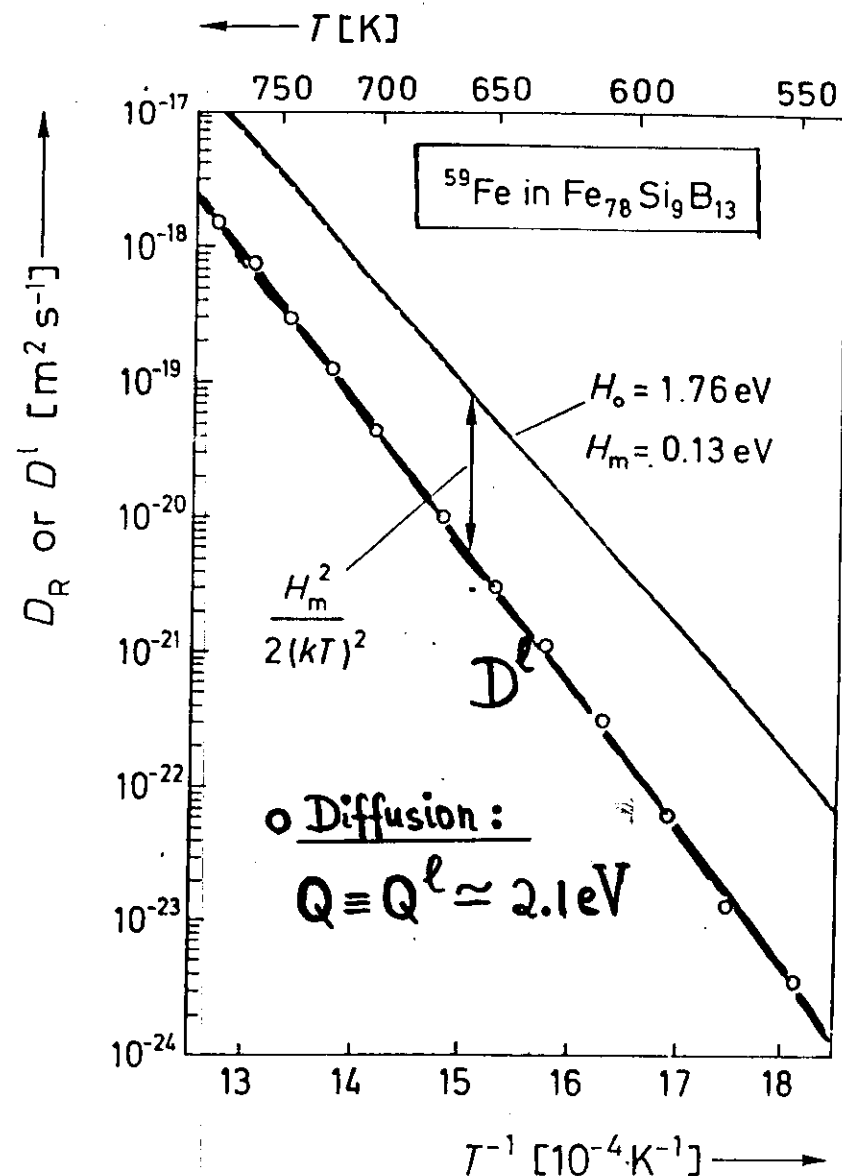
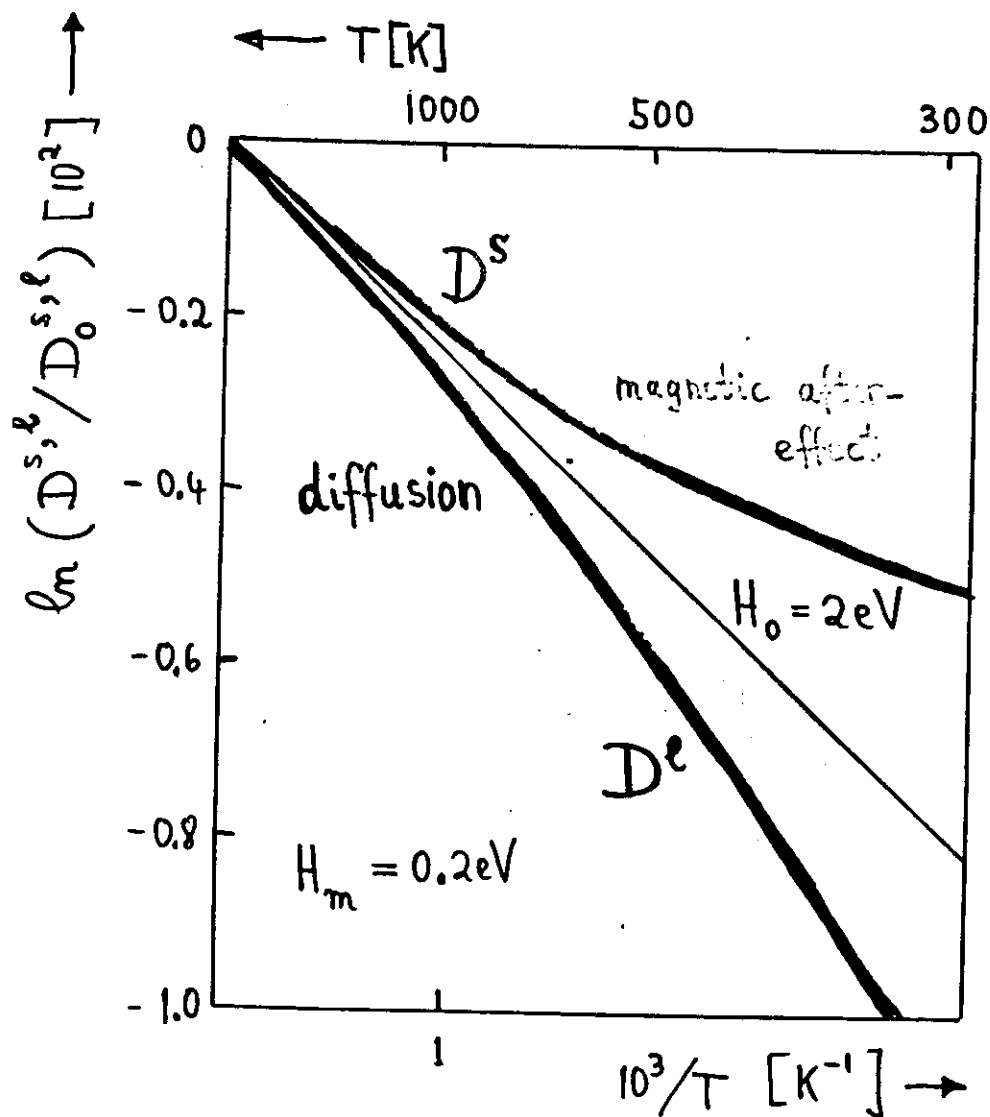
$$\langle Q \rangle \approx 1.95 \text{ eV}$$

^{59}Fe diffusion

$$t \approx 10^5 \text{ s}, 690 \text{ K} \leq T \leq 730 \text{ K}$$

$$Q^l \approx 2.33 \text{ eV}$$

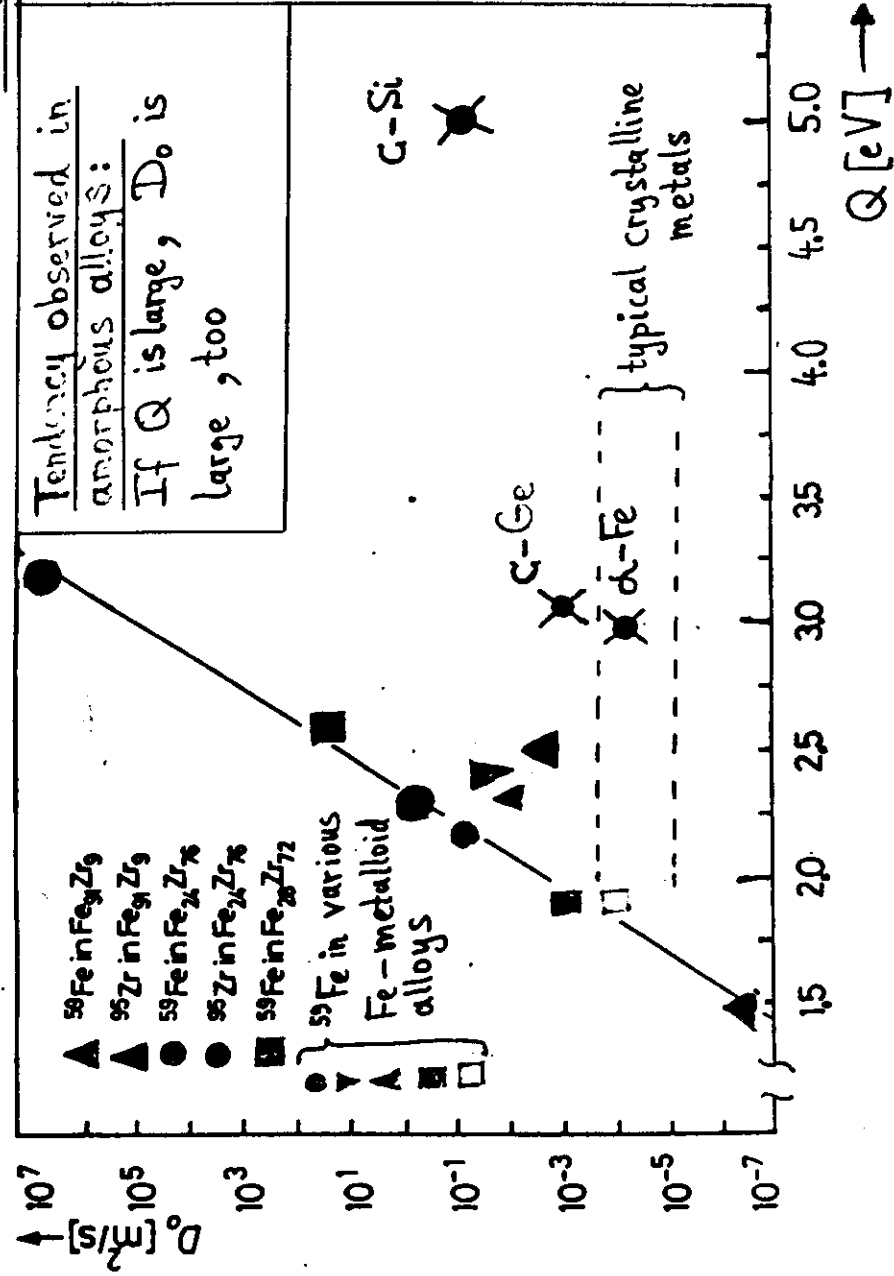
Quasi-Arrhenius Laws



⁵⁹Fe diffusion in amorphous Fe₇₈Si₉B₁₃ obeys a quasi-Arrhenius law with $Q \approx 2.1 \text{ eV}$ (T. Lort - H. = 1.76 eV and $H_m = 0.13 \text{ eV}$)

Correlation between Q and D_0 -

a key for understanding diffusion in solid and amorphous state



Self-diffusion mechanisms

in
relaxed
amorphous alloys

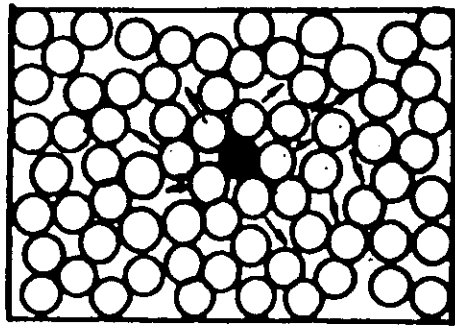
Example: Zr-Fe alloys

Limiting cases:

^{95}Zr in Zr-rich alloy ($\bullet$):
 Q & D_0 very large

^{58}Fe in Fe-rich alloy ($\blacktriangle$):
 Q & D_0 very small

⁹⁵Zr in Zr-rich matrix, e.g., $Zr_{76}Fe_{24}$ ⁽¹⁶⁾



⁹⁵Zr large atom, Zr-rich matrix densely packed

⇒ Motion of ⁹⁵Zr requires simultaneous motion of the neighbouring atoms:

(i) large migration enthalpy, $H_T^M \approx 3.2 \text{ eV} \equiv Q$

(ii) large "diffusion efficiency", $D_0 \approx 7 \times 10^{+6} \frac{\text{m}^2}{\text{s}}$

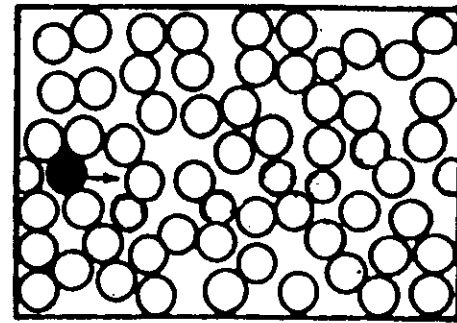
simultaneous motion of many atoms;

large migration entropy, $S_T^M \approx 29 k \equiv S$

(delocalization of residual free volume)

⇒ collective motion of many atoms

⁵⁹Fe in Fe-rich matrix, e.g., $Fe_{91}Zr_9$ ⁽¹¹⁾



⁵⁹Fe small atom; Fe-rich matrix loosely packed

⇒ ⁵⁹Fe can jump on vacant neighbouring sites formed by fluctuations:

(i) small migration enthalpy, $H_T^M \approx 1.5 \text{ eV} \equiv Q$

(ii) small "diffusion efficiency", $D_0 \approx 3 \times 10^{-7} \frac{\text{m}^2}{\text{s}}$,

only 1 atom participates in jump;

negative migration entropy, $S_T^M = -1.7 k \equiv S$

(Contraction of residual free volume on a neighbouring site of ⁵⁹Fe)

⇒ collective preparation of the jump of a single atom

Summary

- (i) The extremely small self-diffusion coefficients in amorphous alloys became measurable by the radiotracer technique in combination with ion-beam sputtering.
- (ii) In unrelaxed amorphous alloys enhanced diffusion arises from the presence of excess free volume (indirect diffusion via quasi-vacancies).
- (iii) Relaxed amorphous alloys are in a metastable equilibrium state. In this state, self-diffusion occurs via direct collective mechanisms, which may be more or less efficient than the diffusion mechanisms in crystals.
- (iv) The quasi-Arrhenius laws for D_R correspond to half-widths of the Q-spectra of less than 0.3 eV.