



INTERNATIONAL ATOMIC ENERGY AGENCY
UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION
INTERNATIONAL CENTRE FOR THEORETICAL PHYSICS
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SMR.550 - 27

SPRING COLLEGE IN MATERIALS SCIENCE ON
"NUCLEATION, GROWTH AND SEGREGATION IN MATERIALS
SCIENCE AND ENGINEERING"
(6 May - 7 June 1991)

TRANSPORT PROCESSES
(INCLUDING RADIATION ENHANCED DIFFUSION)

Part V

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These are preliminary lecture notes, intended only for distribution to participants.

Self-Organization in Solids

by

Anisotropic Transport of Matter

or

Formation of Ordered Defect
Structures in Solids far
away from Thermal Equilibrium

(2)

Structures occur in nature

on all size scales:

(i) gallactical structures

(ii) global structures (see next transparency)

(iii) macroscopic structures (e.g., waves on the ocean, ripples in the sand)

(iv) mesoscopic structures (5 nm ... 500 nm)

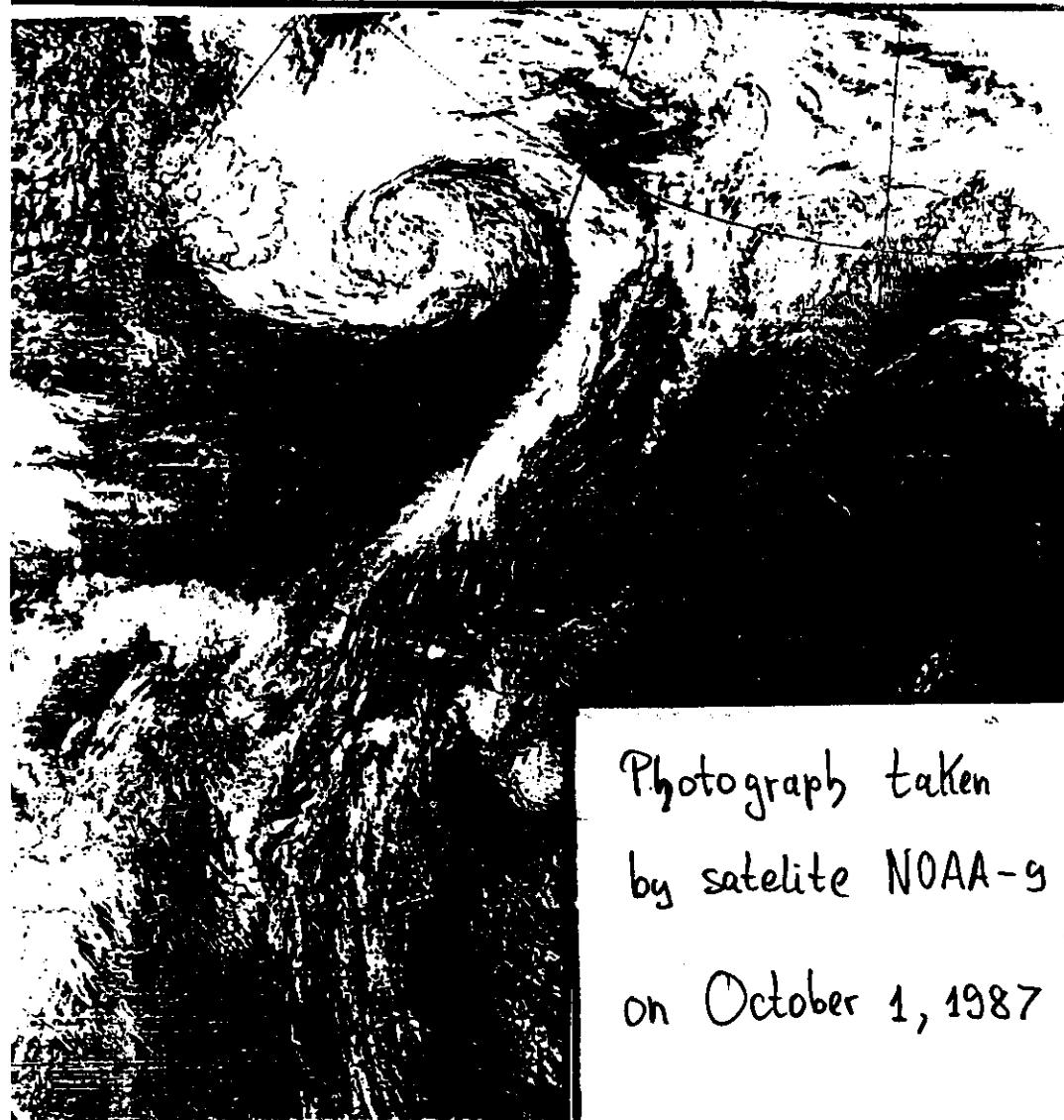
(v) microscopic structures (e.g., crystal lattices)

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(3)

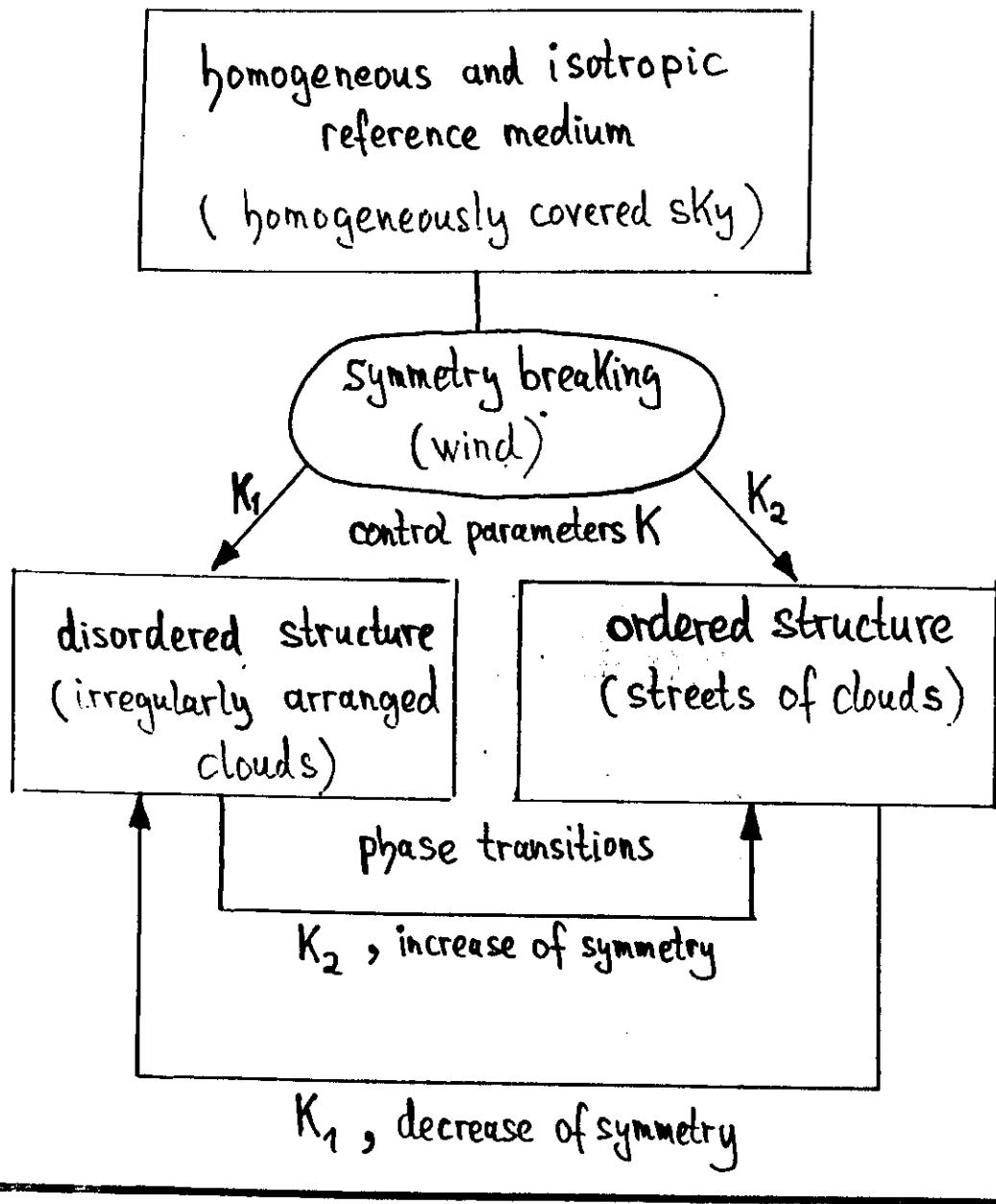
Swirl of clouds (northwest of Scandinavia) —

an example of a structure on the global scale



Photograph taken
by satellite NOAA-9
on October 1, 1987

(4) What is a (spatial) ordered structure?



(5) Ordered Structures in Thermo = dynamic Equilibrium

$$F = U - TS = \text{minimum} \quad T, V = \text{const}$$

↑ ↑

internal energy entropy (disorder)

free energy

How can ordered structures be produced?

High T → high U → high S
gas (molecular disorder)

Low T : $S \rightarrow 0$ (3rd Law)
→ $U = \text{minimum}$

ordered structures

crystalline solids (microscopic order)
water lattices (macroscopic order)

(6) Ordered Structures far from Equilibrium

Prerequisite for formation of order:

$$dS = dS_{in} + dS_{ex} < 0$$

$$dS_{in} > 0 \quad (\text{2}^{\text{nd}} \text{ law})$$

$$\rightarrow dS_{ex} < 0, |dS_{ex}| > dS_{in}$$

dissipative systems

Examples

1) Explosion?

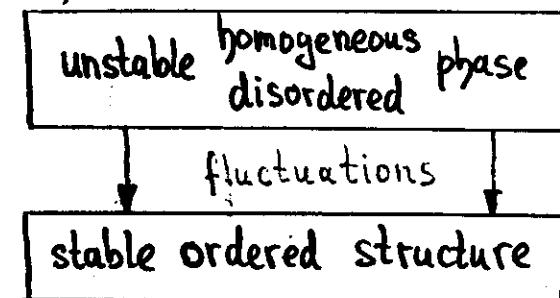
2) Undercooling of a melt

- rapid export of heat
- symmetry breaking, since rate of crystallization depends on crystallographic directions

D1 dentrites (conditionally ordered structure)

(Quasi-) Steady States

- (1) open systems
- (2) far away from equilibrium
- (3) strongly non-linear
- (4) symmetry breaking (by external boundary conditions or internal constraints)
- (5) suitable values of the control parameters (bifurcation):



Examples

1) Laser (Haken)

2) Bénard-type structures in liquids:



Great advantage of structures in solid: metastable

(8) Ordered Structures in Solid Matter

D2 Dislocation pattern in a cyclically deformed Cu single crystal (room temperature, $\dot{\epsilon} = 10^{-4} \dots 10^{-2}$)

D3 Periodically arranged walls of defect clusters
 $\parallel \{100\}$ in p^+ -irradiated Cu ($\leq 100^\circ C$, 3.4 MeV, 0.65 dpa)

D4 Krypton bubble lattice $\parallel \{0001\}$ in Kr $^+$ -implanted α -Zr ($400^\circ C$)

D5 Lattice of stacking-fault tetrahedra in e^- -irradiated Ag (primitive-cubic \parallel fcc host lattice, room temperature, $6.3 \times 10^{27} e^-/m^2$)

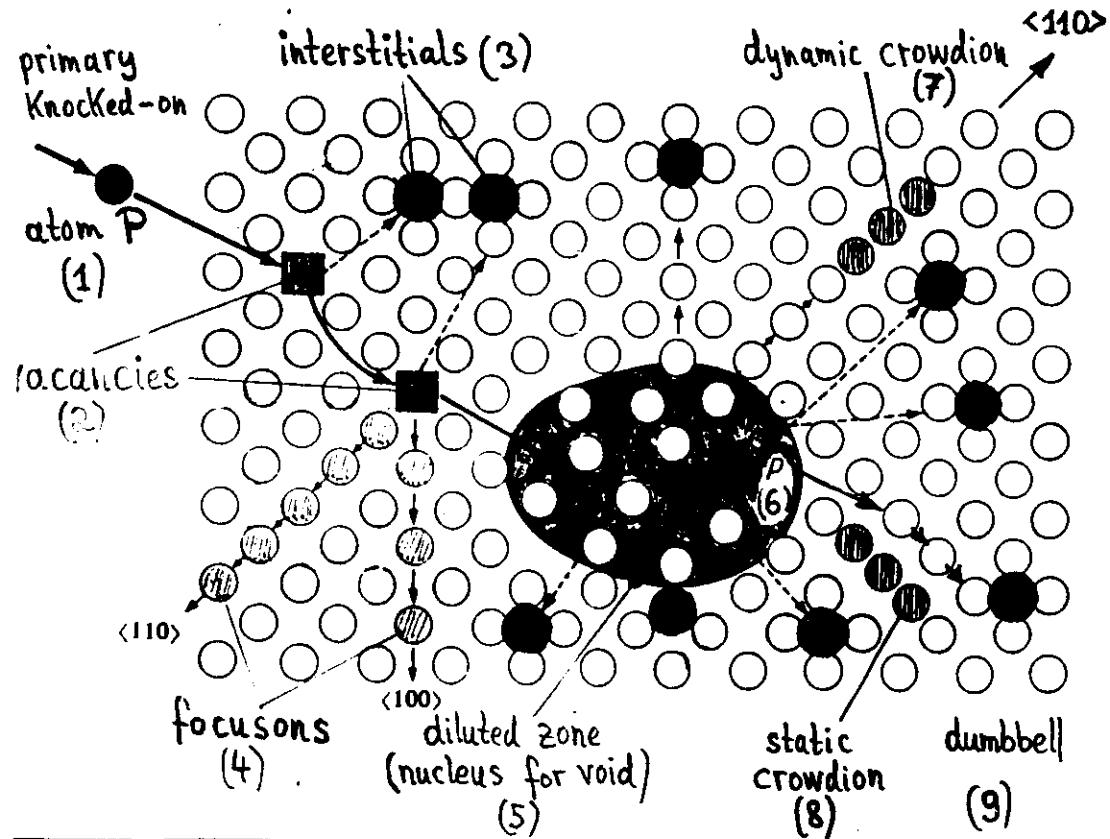
D6 Voids lattice in Nb-1% Zr after V $^+$ implantation (bcc \parallel bcc host lattice, 780°C, 3.1 Mev, 50 dpa)

Void lattices

- high-dose neutron or ion irradiation at about $\frac{1}{3} T_m$
- void lattices \equiv host lattices

(9) Radiation Damage by Fast Neutrons

(fcc metal, (100) plane)

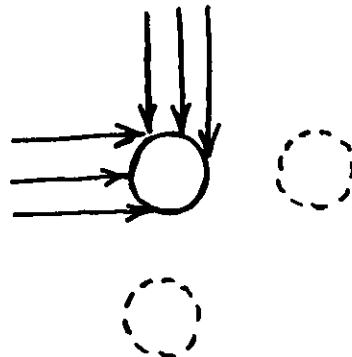


1) "Ballistic" athermal transport: dynamical crowdions

2) Thermally activated diffusion of point defects:

One-interstitial model	two-interstitial model
Vacancies (3d)	Vacancies (3d)
dumbbells (3d)	stable dumbbells (3d) metastable crowdions (1d)

(10) Void-Lattice Formation by Mutual Protection
of Voids against Dynamic Crowdions (Foreman)



Arguments against this mechanism

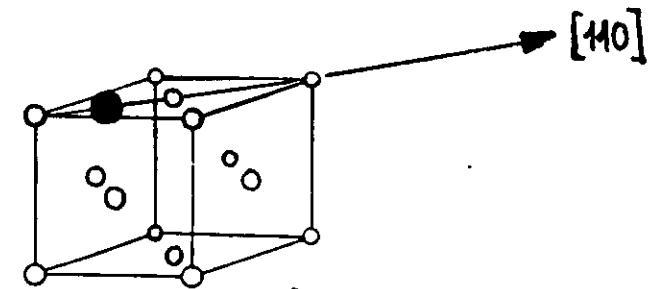
Void hyperlattices in Al (Horsewell & Singh)

(required range of dynamic crowdions $\approx 2A$
 $= 400 \dots 500 \text{ nm}$)

Metal	packing density	dynamic crowdions	void lattices	static crowdions
Au	● ● ●	+++	-	-
Nb	● ● ●	++	+	+
Al	● ● ●	+	++	++

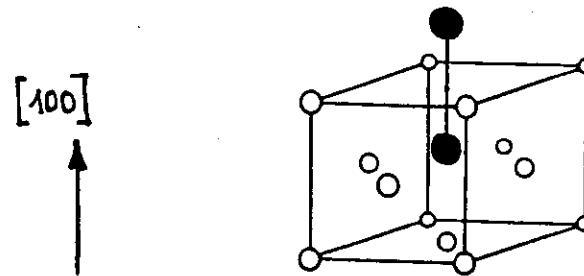
(11) Two - Interstitial Model
(fcc metals)

(static) crowdion (1d)

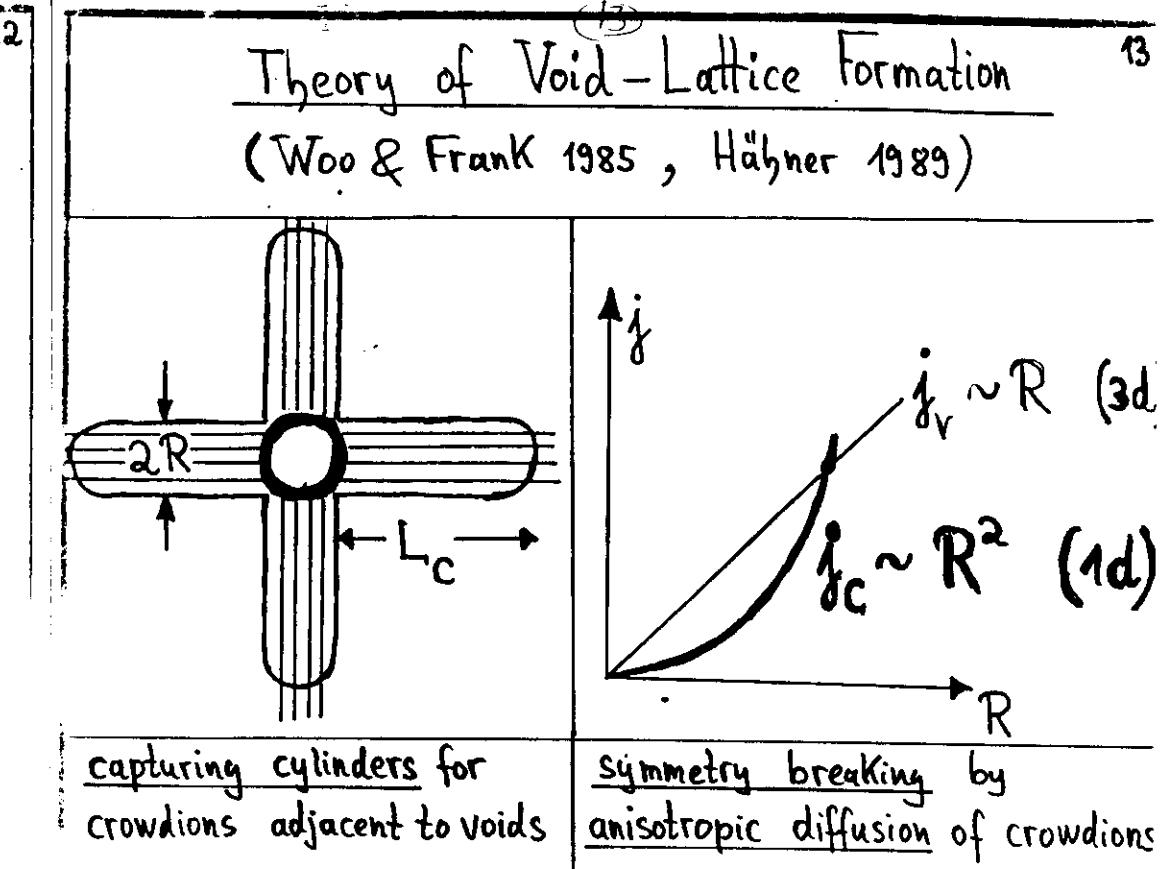
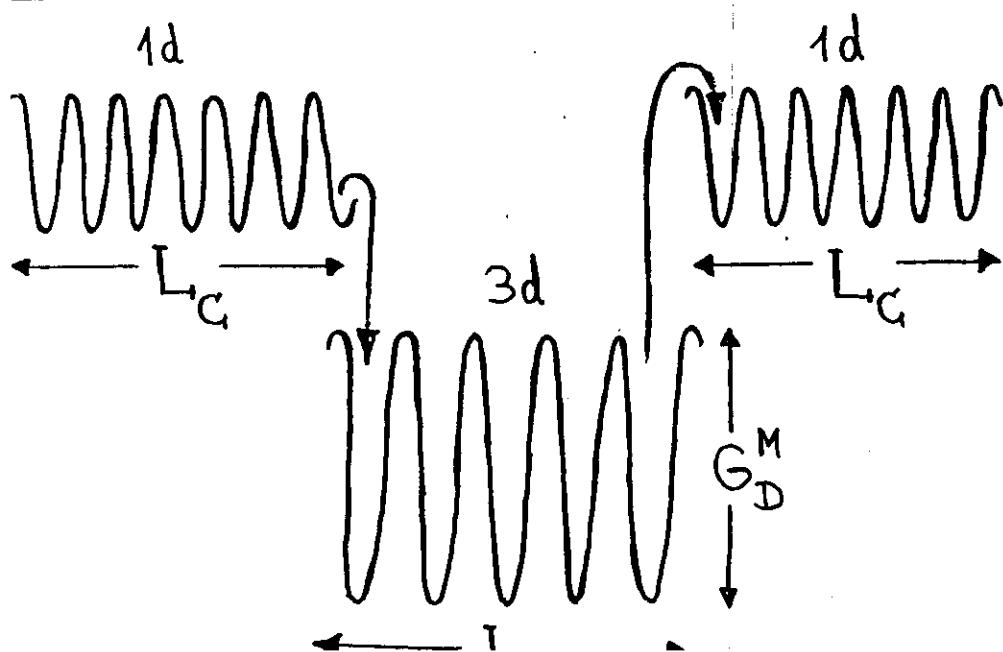
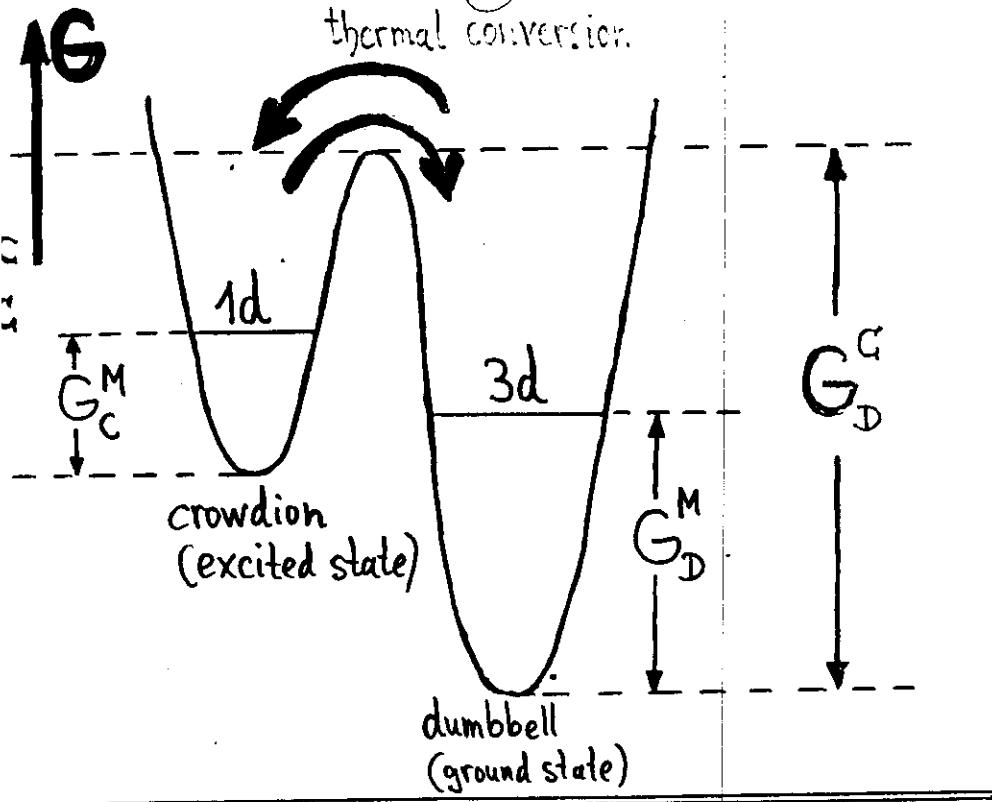


= metastable excited interstitial state

dumbbell (3d)



= ground state of interstitial

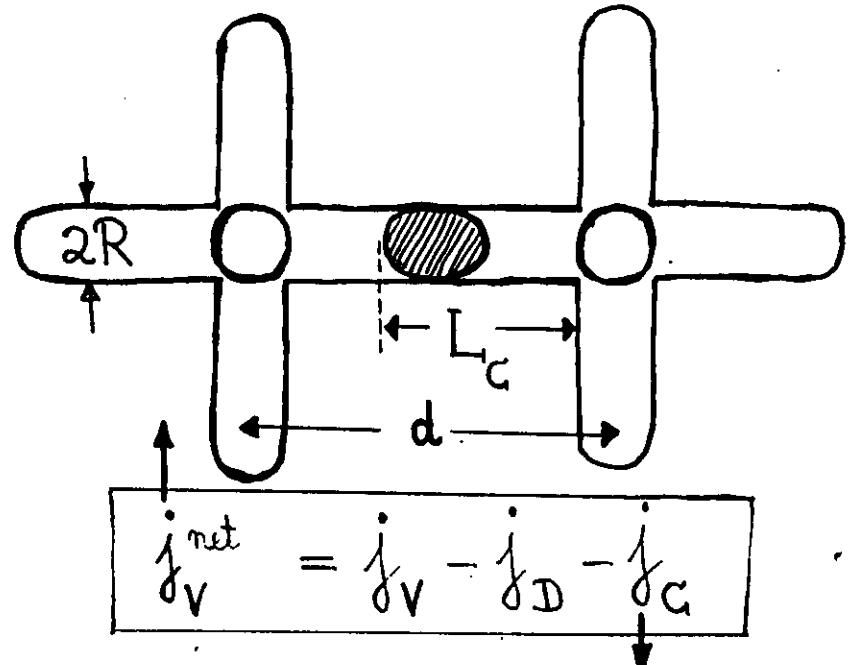


$$j_V^{\text{net}} = j_v - j_D - j_G$$

- voids grow until a final size is reached
- disordered arrangement of voids is conserved
- only true for low void densities (at which capturing cylinders of neighbouring voids do not overlap)

(14) void density — a control parameter!

High void density (capturing cylinders do overlap)



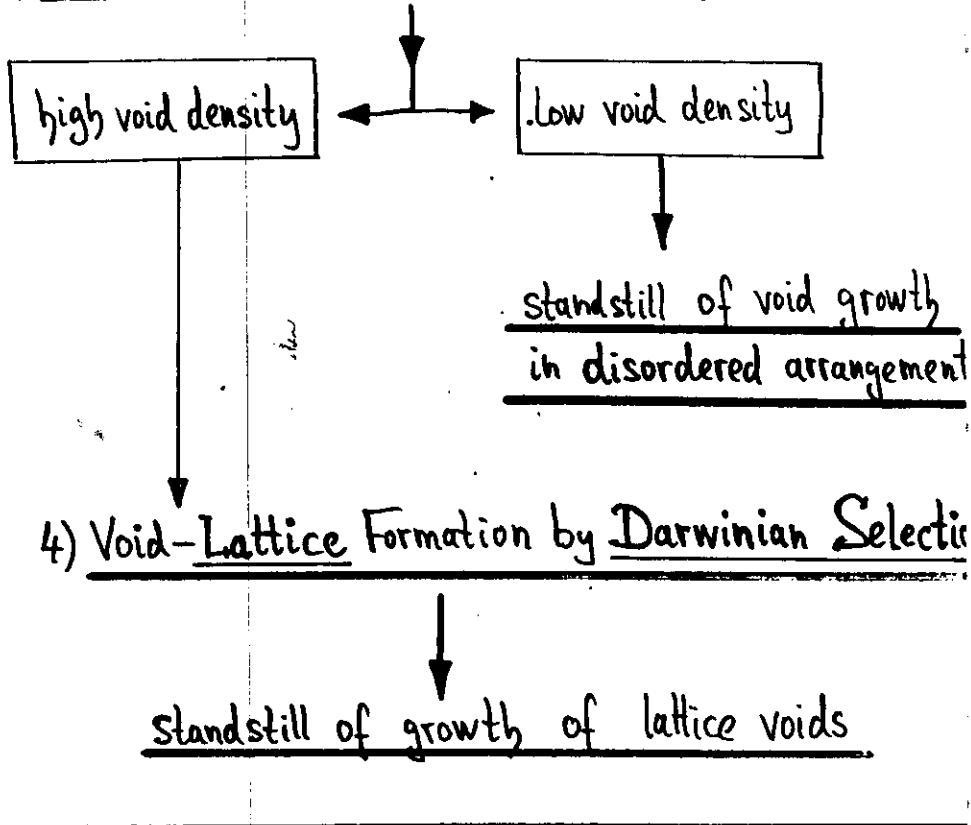
- overlapping leads to enhanced growth rate
- isolated voids shrink away

→ Void-Lattice Formation by Darwinian Selection!

- void lattices and host lattices possess the same structure and orientation
- quantitative result:
undammed mode: $d \approx 2R + 0.7L$

(15) Chronology of Void-Lattice Formation

- Establishing of quasi-steady concentrations of vacancies, dumbbells, and crowdins
- Nucleation of statistically arranged voids at the diluted zones in displacement cascades
- Growth of voids in disordered arrangement



Theory of Stacking-Fault-Tetrahedron Lattices

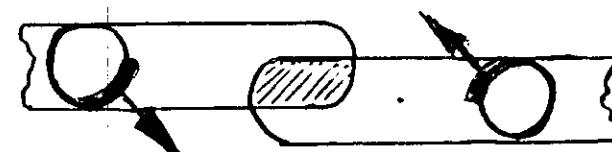
(Zaiser 1991)

- "Low" temperatures, high defect densities, suppression of thermal diffusion, but extensive radiation-induced diffusion
- 1-dimensional transport of matter by dynamic crowdions (symmetry breaking)
- Defocusing of dynamic crowdions in interstitial-rich regions
→ periodic accumulation of interstitials
(distance $\approx L_{\text{G}, \text{dynamic}}$)
- Periodic accumulations of vacancies in the intermediate regions
→ "CsCl" lattice
- Collapse of vacancy-rich regions to stacking-fault tetrahedra arranged in a primitive-cubic lattice (|| fcc host lattice)

Void Lattices Versus Stacking-Fault-Tetrahedron Lattices

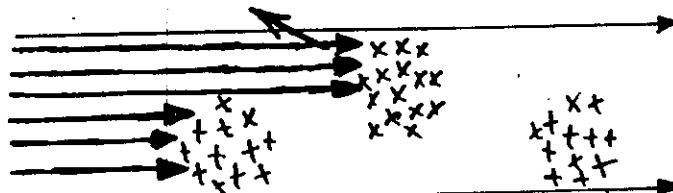
What is the crucial difference between the mechanisms of formation?

Void lattices — selective destruction:



- alignment in close-packed directions
- void lattice \equiv host lattice

Stacking-fault-tetrahedron lattice — selective construction:



- disalignment from close-packed directions
- Stacking-fault-tetrahedron lattice $\not\equiv$ host lattice

