



INTERNATIONAL ATOMIC ENERGY AGENCY
UNITED NATIONS EDUCATIONAL, SCIENTIFIC AND CULTURAL ORGANIZATION
INTERNATIONAL CENTRE FOR THEORETICAL PHYSICS
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UNITED NATIONS INDUSTRIAL DEVELOPMENT ORGANIZATION



INTERNATIONAL CENTRE FOR SCIENCE AND HIGH TECHNOLOGY

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SMR. 628 - 17

**Research Workshop in Condensed Matter,
Atomic and Molecular Physics
(22 June - 11 September 1992)**

**Working Party on:
'Energy Transfer in Interactions with
Surfaces and Adsorbates'
(31 August - 11 September 1992)**

**"The Role of Adsorbate Degrees of
Freedom in Sticking"
(Part II)**

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Role of adsorbate degrees of freedom in sticking.

Because of large energy separation of rotations and vibrations they can to 1st order be treated separately:

1) Rotations (rotational cooling)

coworkers:

Kasai, Brünner, Häug
Experiments: various

2) Vibrations (vibrational heating)

coworkers:

Kasai, Küchenhoff, Chiba

Experiments:

Kūbiak, Zare $H_2, D_2 / Cu$ (Desorption)
Rendulic " / Cu (Sticking)
Zacharias et al. $H_2, HD, D_2 / Pd$
(Desorption)

χ -distribution NO/Pt; 0.08 eV

J. Segner et al. / Rotational state populations of NO molecules

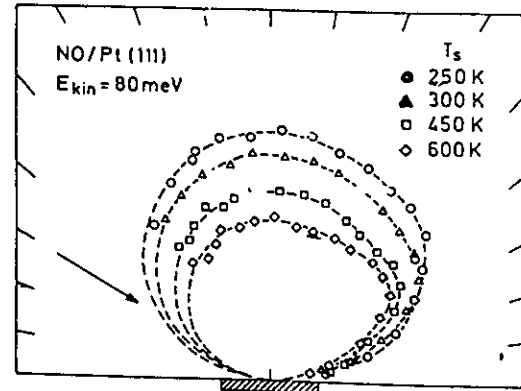


Fig. 3. Angular distributions of NO molecules scattered from a Pt(111) surface at different surface temperatures for $E_{kin} = 80$ meV and an incidence angle of 60° .

ϵ_j -distribution

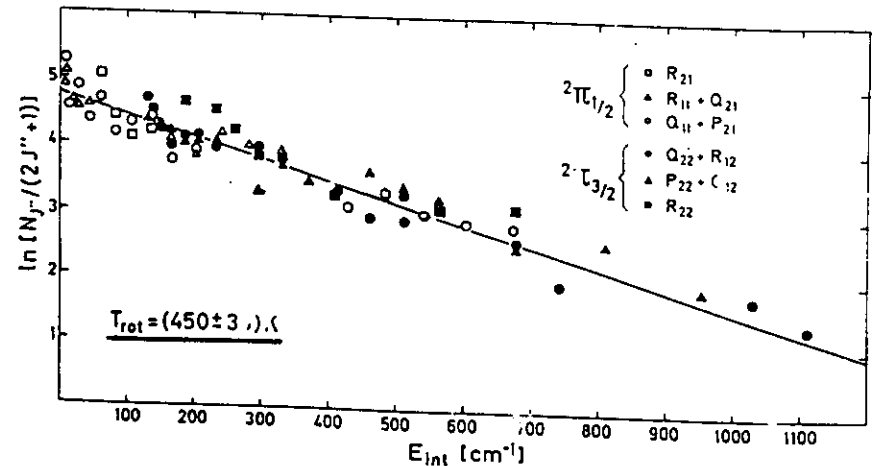


Fig. 4. Rotational state distribution for NO molecules scattered from a Pt(111) surface at 890 K. The straight line represents a Boltzmann distribution fitted to the measured points. Rotational populations of the two electronic ground states are superimposed in the diagram (E_{int} = internal energy).

Segner, ..., Häger, ..., Walther (Sinf. Sei 131 273 ('83))

DAVID S. KING AND RICHARD R. CAVANAGH
Adv. Chem. Phys., **LXXVI**, 45 ('89)

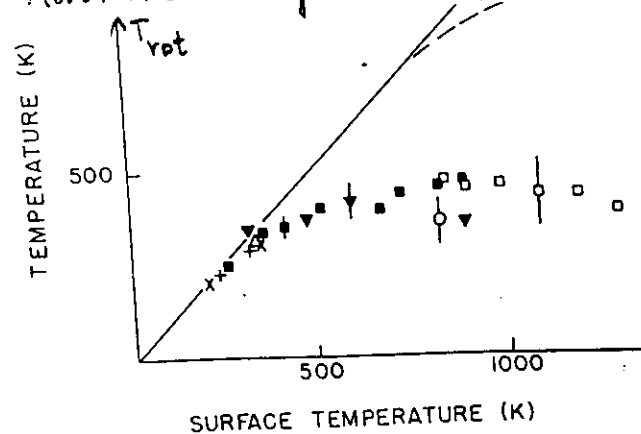
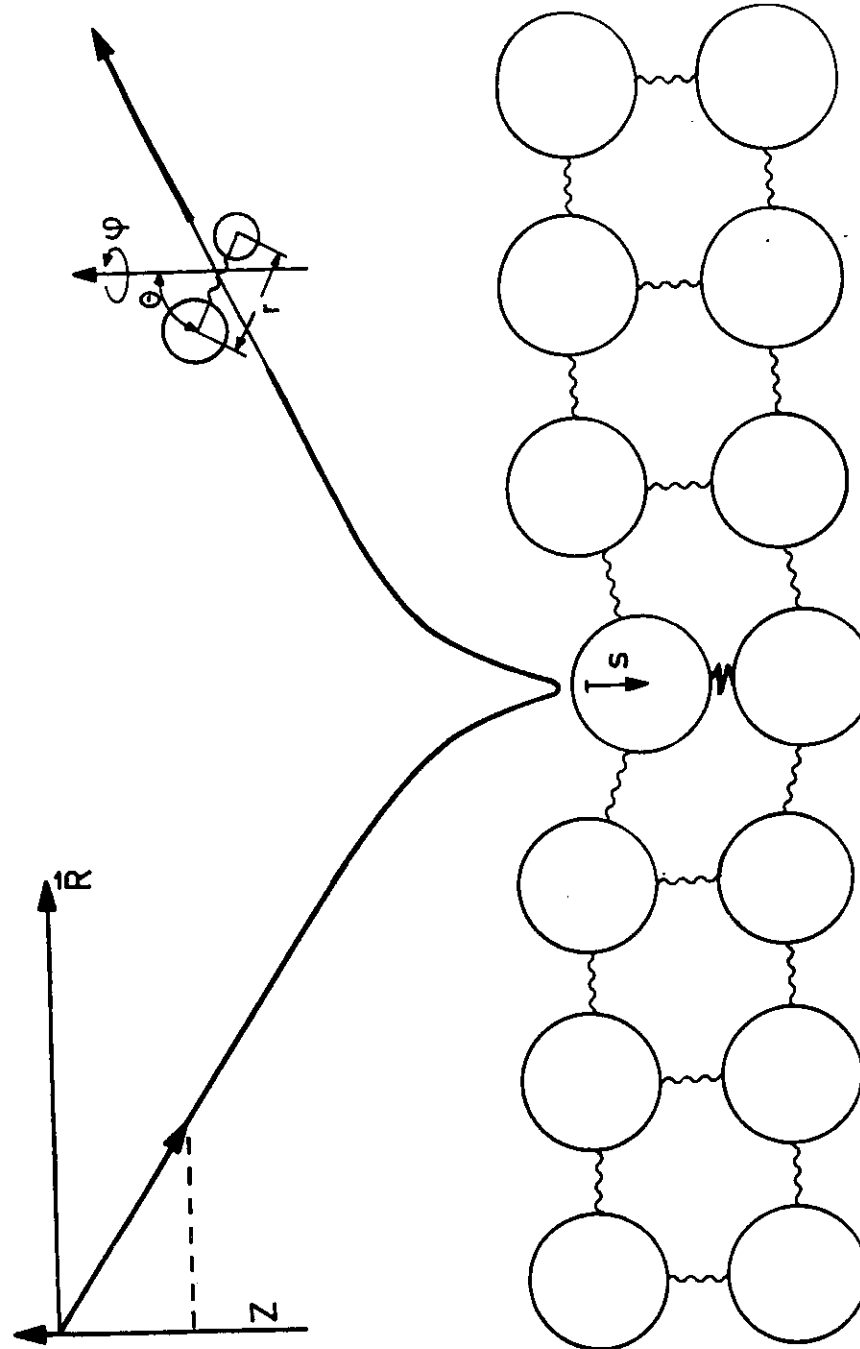
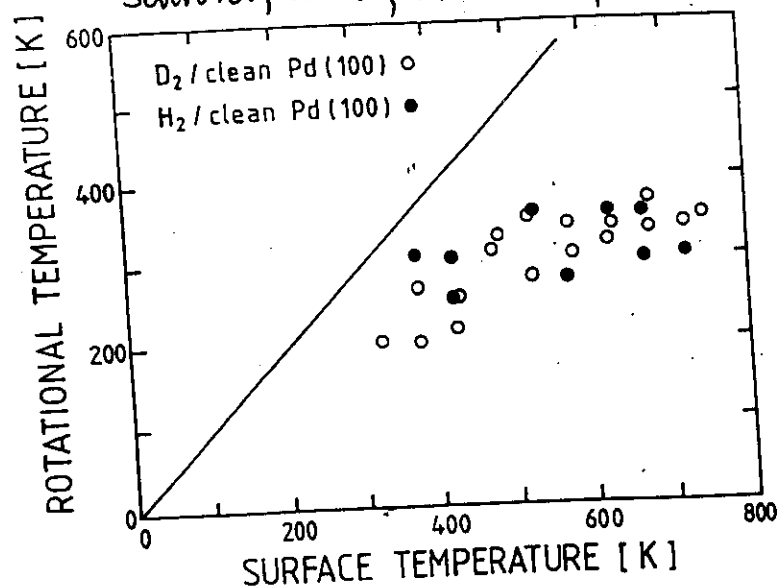


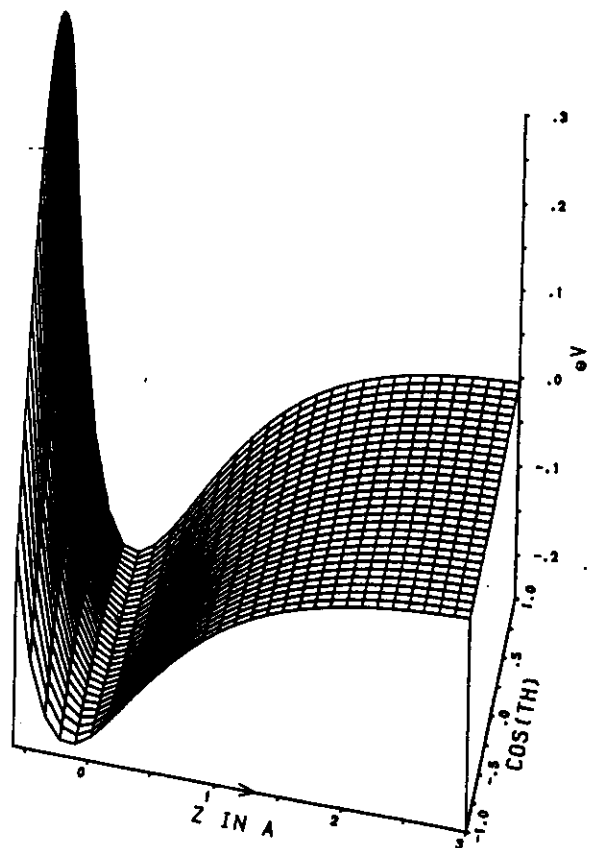
Fig. 5. Rotational temperatures of NO desorbing from Pt(111). The data are representative of data published for (x) neat thermal desorption^{10a}, (+) thermal desorption in the presence of coadsorbed CO^{10b}, (solid squares)²⁴ and (solid triangles)²⁵ trapping/desorption in molecular beam scattering, (open triangle) reaction limited desorption from NO-NH₃ complexes¹⁴, (open circle)²⁶ and (open square)²⁷ NH₃ oxidation reactions. The solid line is for full accommodation. The dashed curve represents results for translational energy measurements in direct inelastic scattering^{18b}.

Schrofer, David, Zadorian; *Surf. Sci* (Goba pnt.)



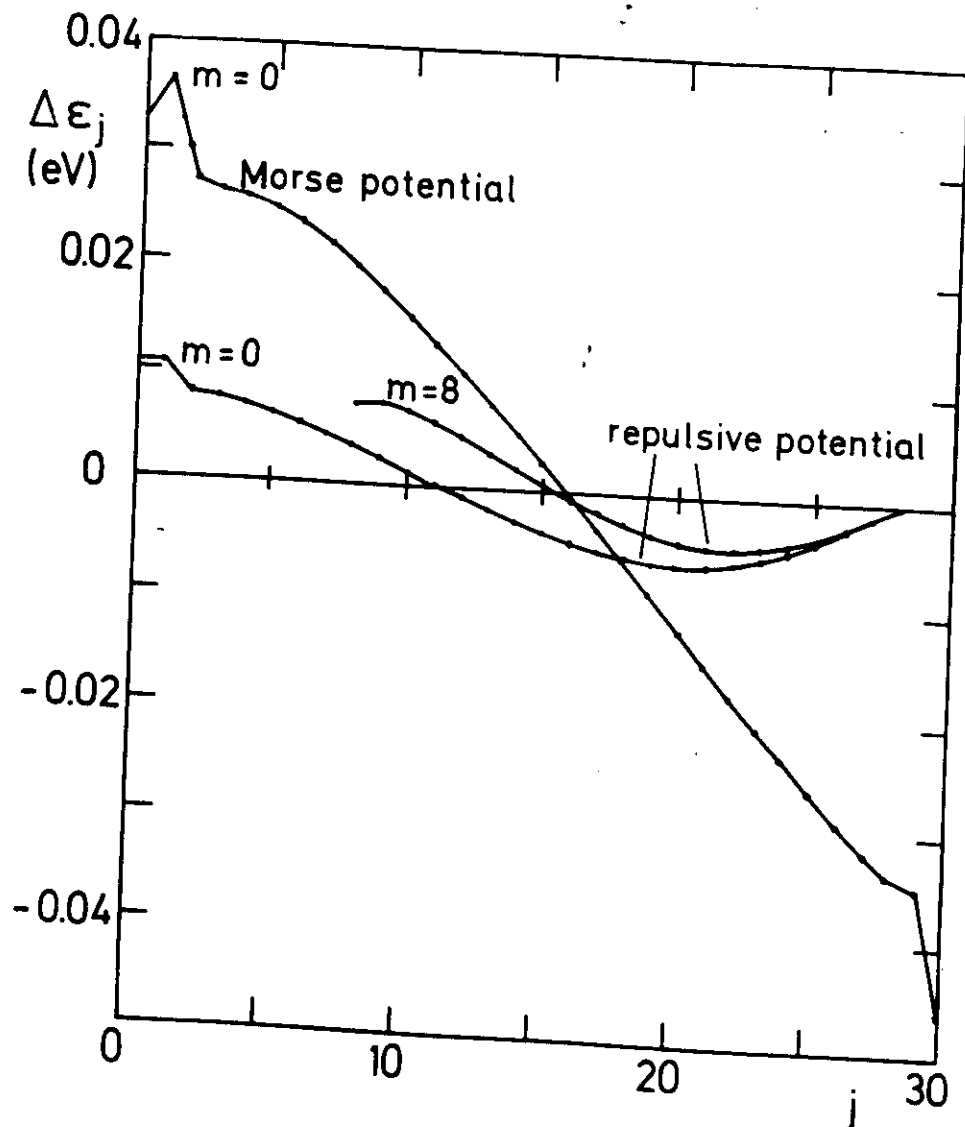
$$NO / Ag \quad 0.25 eV \left[(1 + .25 P_1 - .166 P_2) e^{-2 \times z} - (2 + .3625 P_1 - .155 P_2) e^{-\alpha z} \right]$$

POTENTIAL



PARAMETER

D. E. M. ALPHA	/ V1. V2. V3. V4. V5. V6. V7. V8. V9. V10. V11. V12. V13. V14. V15. V16
.2500E+00	.3000E+00 .3000E+02 .1500E+01
.2500E+00	-.1880E+00 .0000E+00 .0000E+00 .0000E+00 .0000E+00 .1813E+00 -.7750E-01 .0000E+00 .0000E+00 .0000E+00 .0000E+00 .0000E+00 .0000E+00 .0000E+00



Br., Kasai, Müller (Sw4.Sci 161 608 (1985))

Remarks concerning rotational cooling

from kinetic theory (detailed bal.)

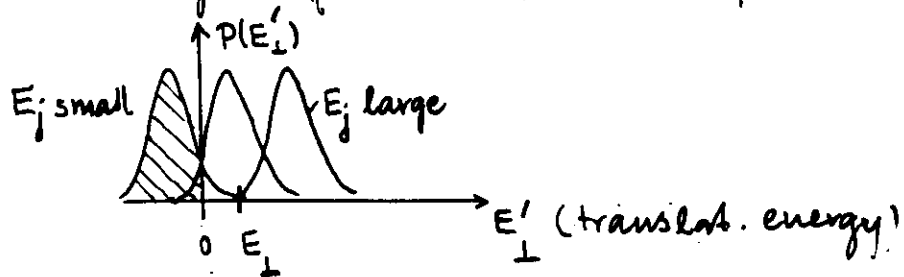
$$P_j^{\text{des}} \propto S(E_j) (2j+1) e^{-E_j/kT_s}$$

sticking coef. $S(E_j) \approx e^{-E_j/E_0}$ (from experim.)

effective temperature

$$T_{\text{eff}} = T_s \frac{E_0}{kT_s + E_0} \quad E_0 \approx 400 \text{ K}$$

Sticking coefficient (qualitatively)

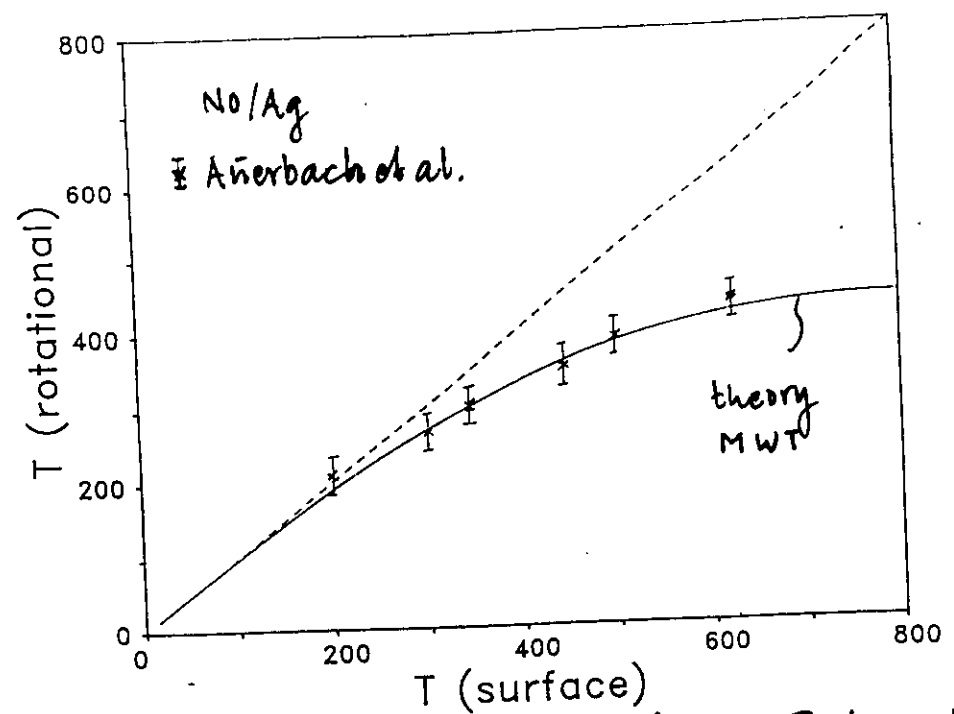


consider $\langle \Delta E_j \rangle = \langle E'_j \rangle - E_j \rightarrow$

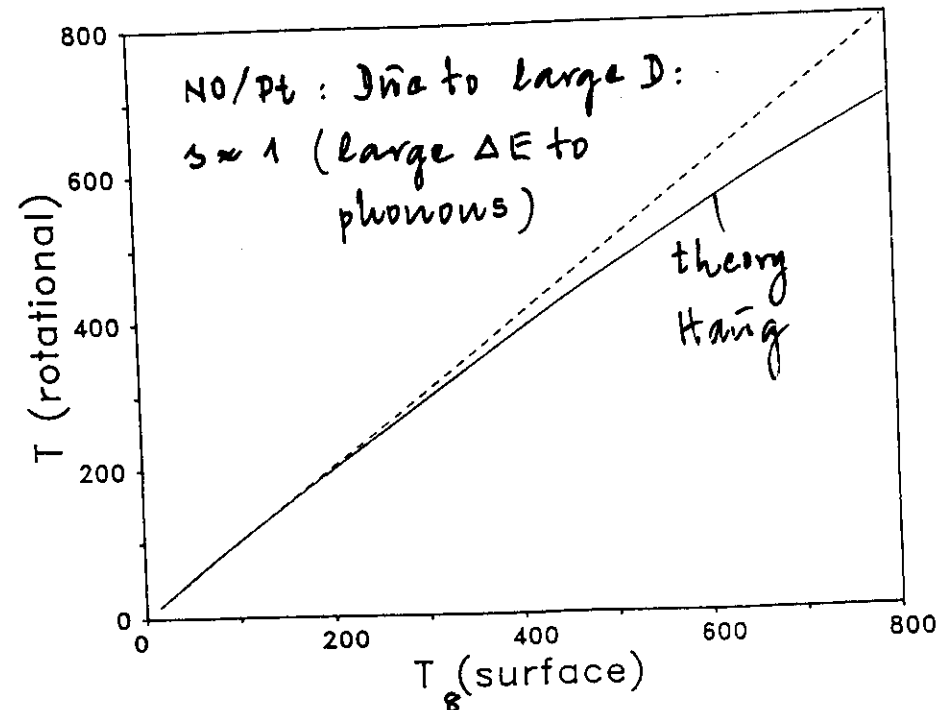
$E_0 \approx$ Energy E_j where

$$\langle \Delta E_j \rangle = 0$$

Question: Phonons?



Theorie: Mülhausen, Williams, Tully; Han



a) Detailed balance may be violated for two or more low lying modes of kinetic equation; in principle possible but difficult to obtain microscopically.

b) Very strong angular dependence of potential in contrast to

α) Direct scattering rotational energy distribution (Jacobs et al.)

β) steric effect of $S(O)$ (Knipfer, Tenno, Koyu, Stolte)

$$\frac{2(S_+ - S_-)}{S_+ + S_-} \approx 0.035 \pm 0.02$$

γ) $A_2 = \langle 3m^2 - j(j+1) \rangle$ (Jacobs et al.)

all can be accounted for by

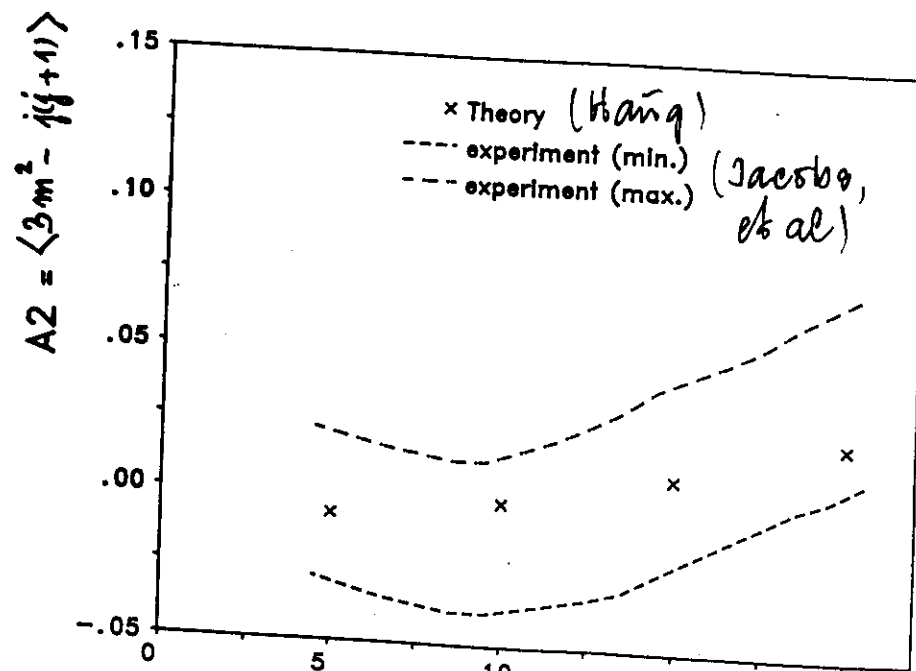
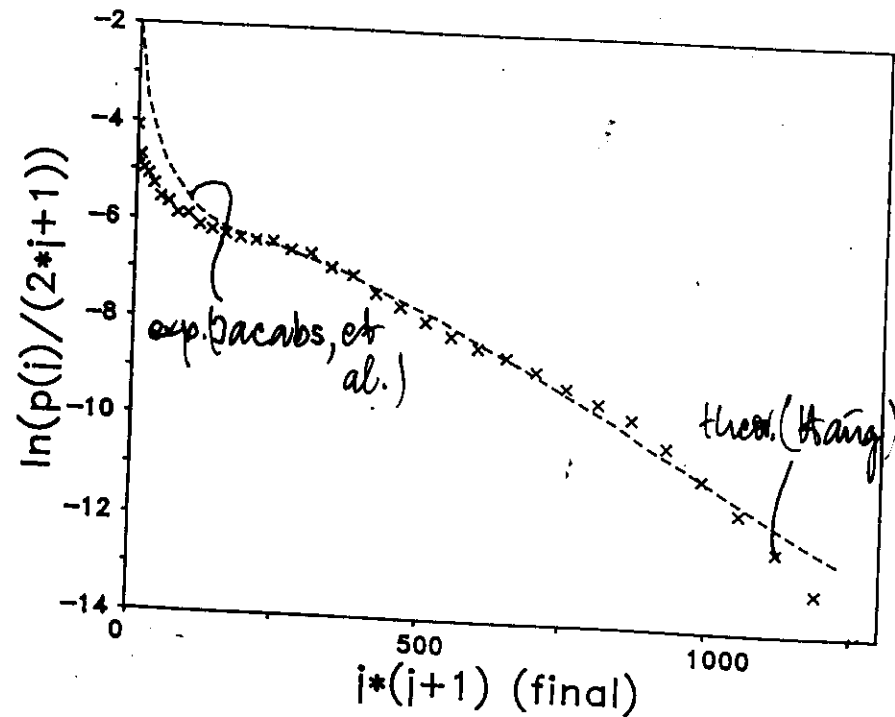
$$Y(\theta) = D \{ R(\theta) e^{-2\alpha z} - A(\theta) e^{-\alpha z} \}$$

$$R(\theta) = 1 + 0.075 P_1(\cos\theta) + 0.075 P_2(\cos\theta)$$

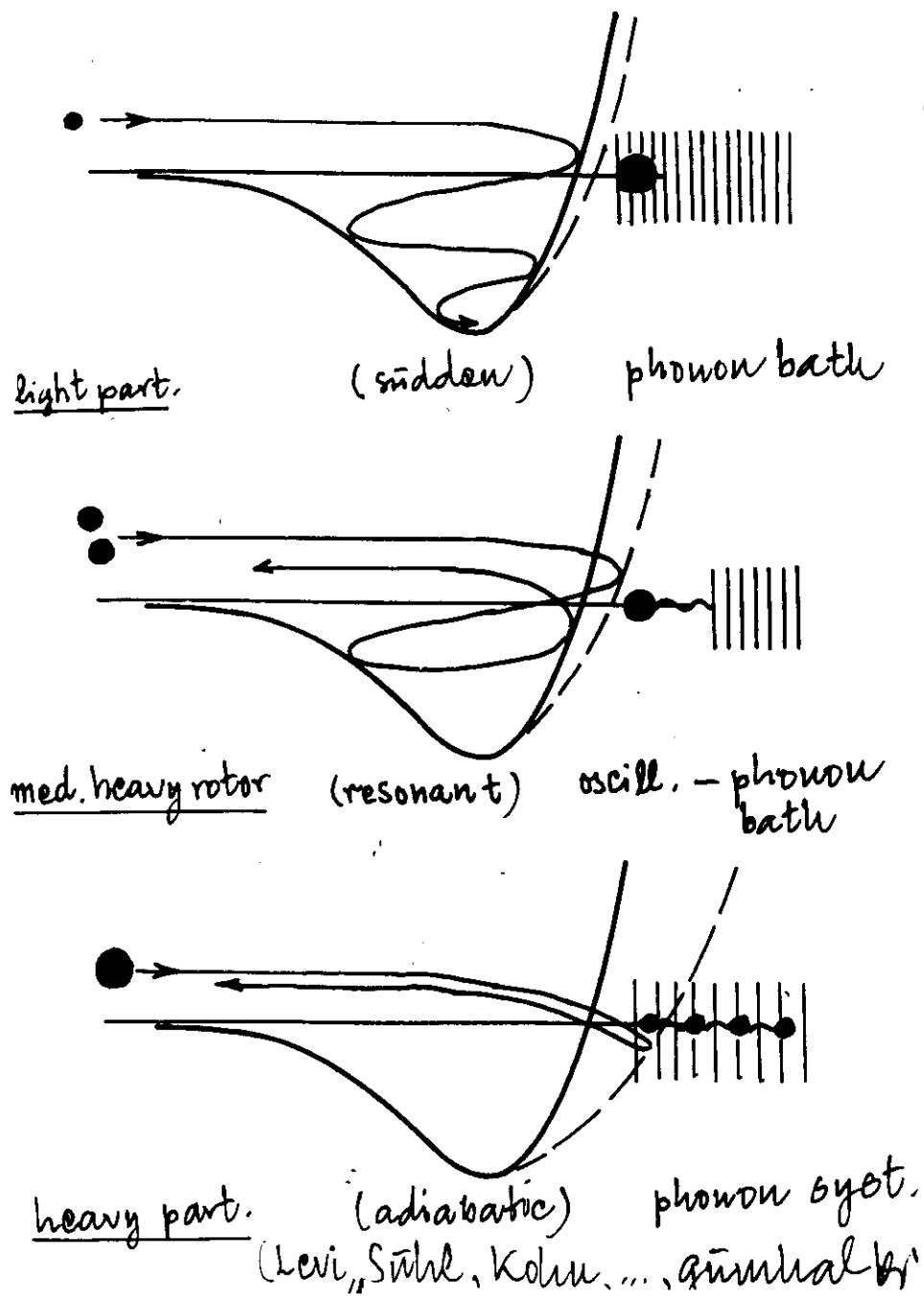
$$A(\theta) = 2(1 + 0.06 P_1(\cos\theta) + 0.075 P_2(\cos\theta))$$

c) Selective Scattering Resonances

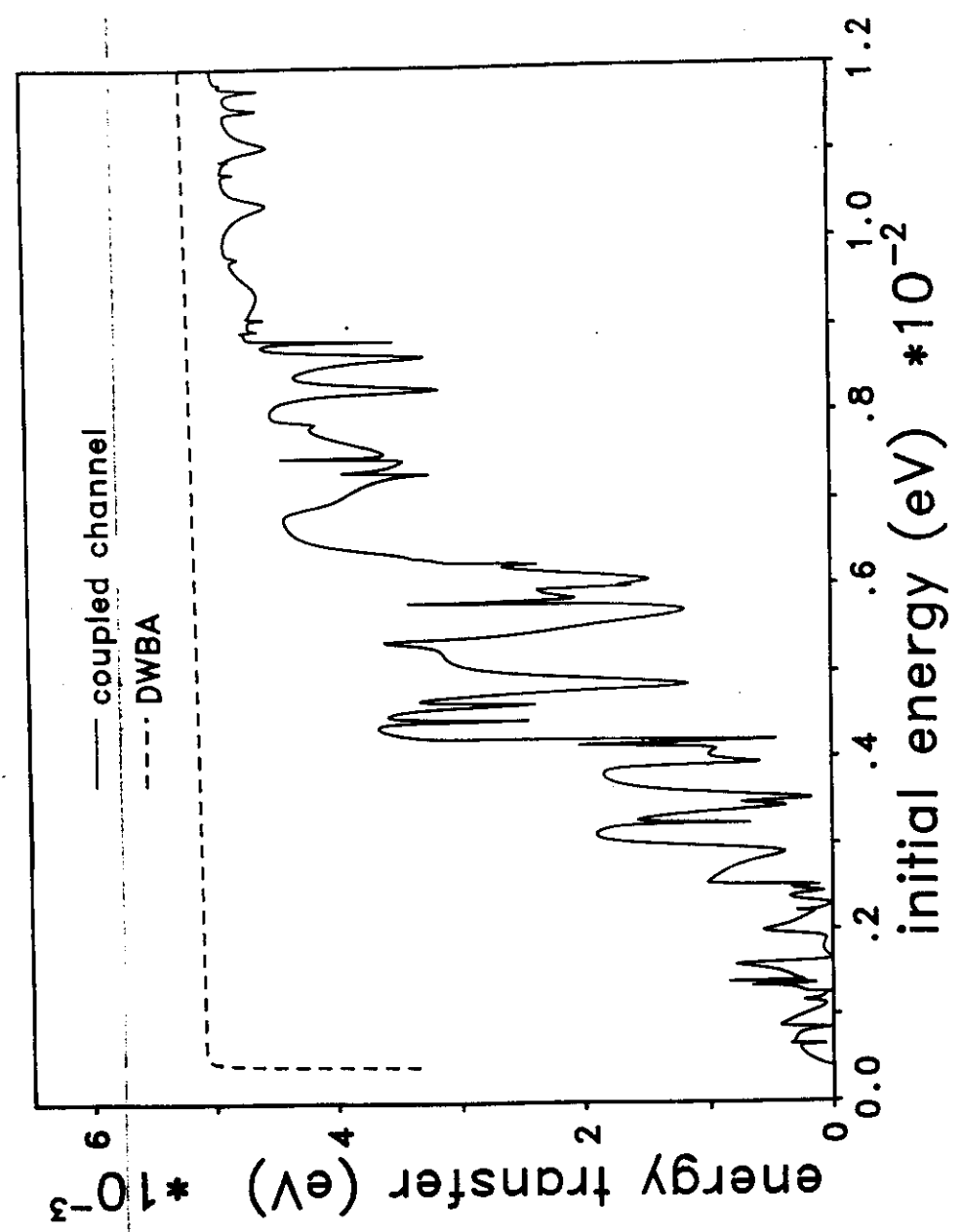
$$S(E_j) \approx \sum P_r \frac{T_{ind}}{T_{ind} + T_{rel}}$$

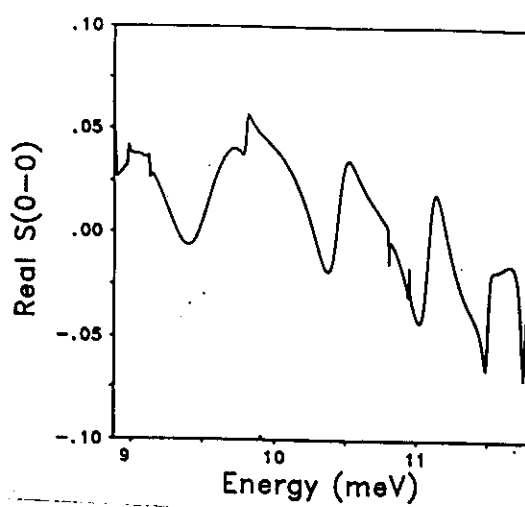


Time scales for sticking (schematic)



selective rotational resonances (Brümmner)

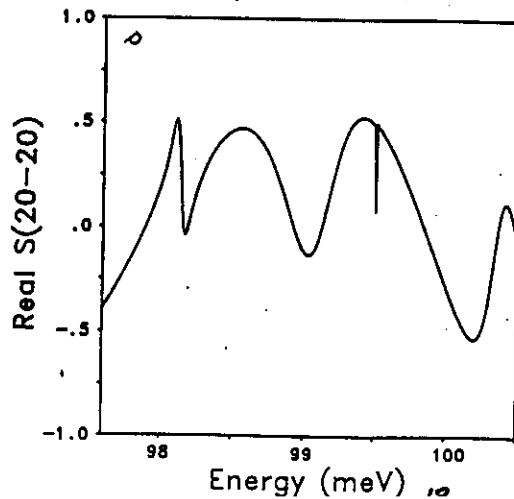
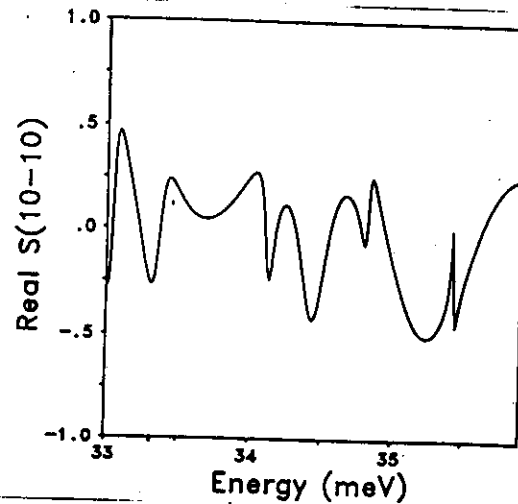




$$j_{im} = 0$$

Increase
of rot. reson.
width with
 j_{im}

$$j_{im} = 10$$



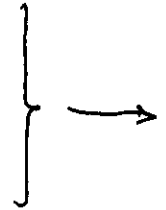
$$j_{im} = 20$$

T. Brügger

Model for vib.-trans. coupling

Neglect:

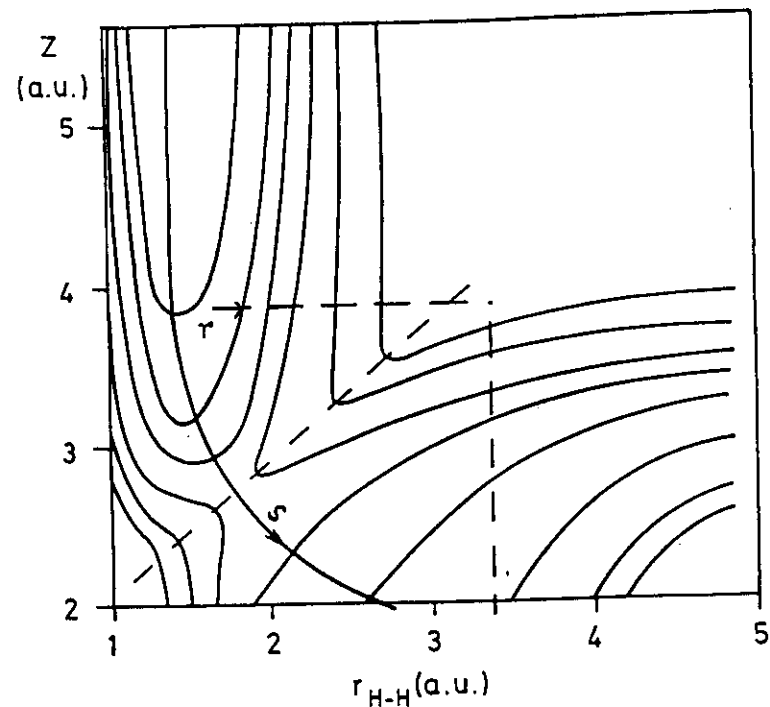
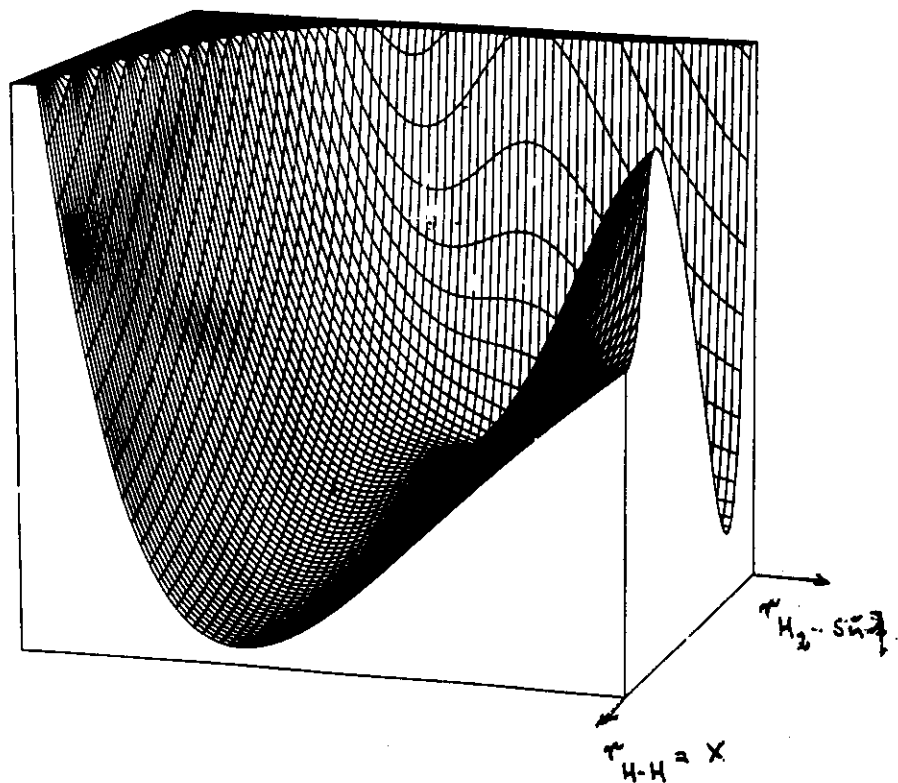
- Rotations
- Corrugation
- ∴ Phonons (NO.?)



2 D-PES, Parameters:

- Barrier $V(s)$
- Curvature $C(s)$
(change in bond length)
- Frequency $\omega(s)$

Potential surface for
H₂/C⁺



Schrödinger equation (harmonic appr. for vibration)

$$\left\{ -\frac{\hbar^2}{2\mu} D_s \eta^{-1} D_s + \eta [V(s) + \hbar \omega(s) n - E] \right\} \psi(r, s) = 0$$

with curvilinear differential operator

$$D_s = \partial_s - \frac{\omega'(s)}{2\omega(s)} r \partial_r$$

coupling term

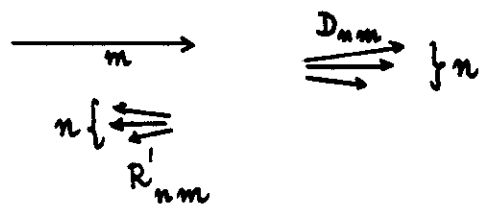
$$\eta(r, s) = 1 - C(s) r / \kappa(s) \quad C(s) = \text{curvature}$$

vibrational wave number

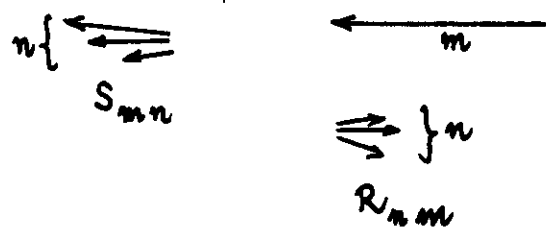
$$\kappa(s) = (\mu \omega(s) / \hbar)^{1/2}$$

Scattering Amplitudes

Desorption (state selective)



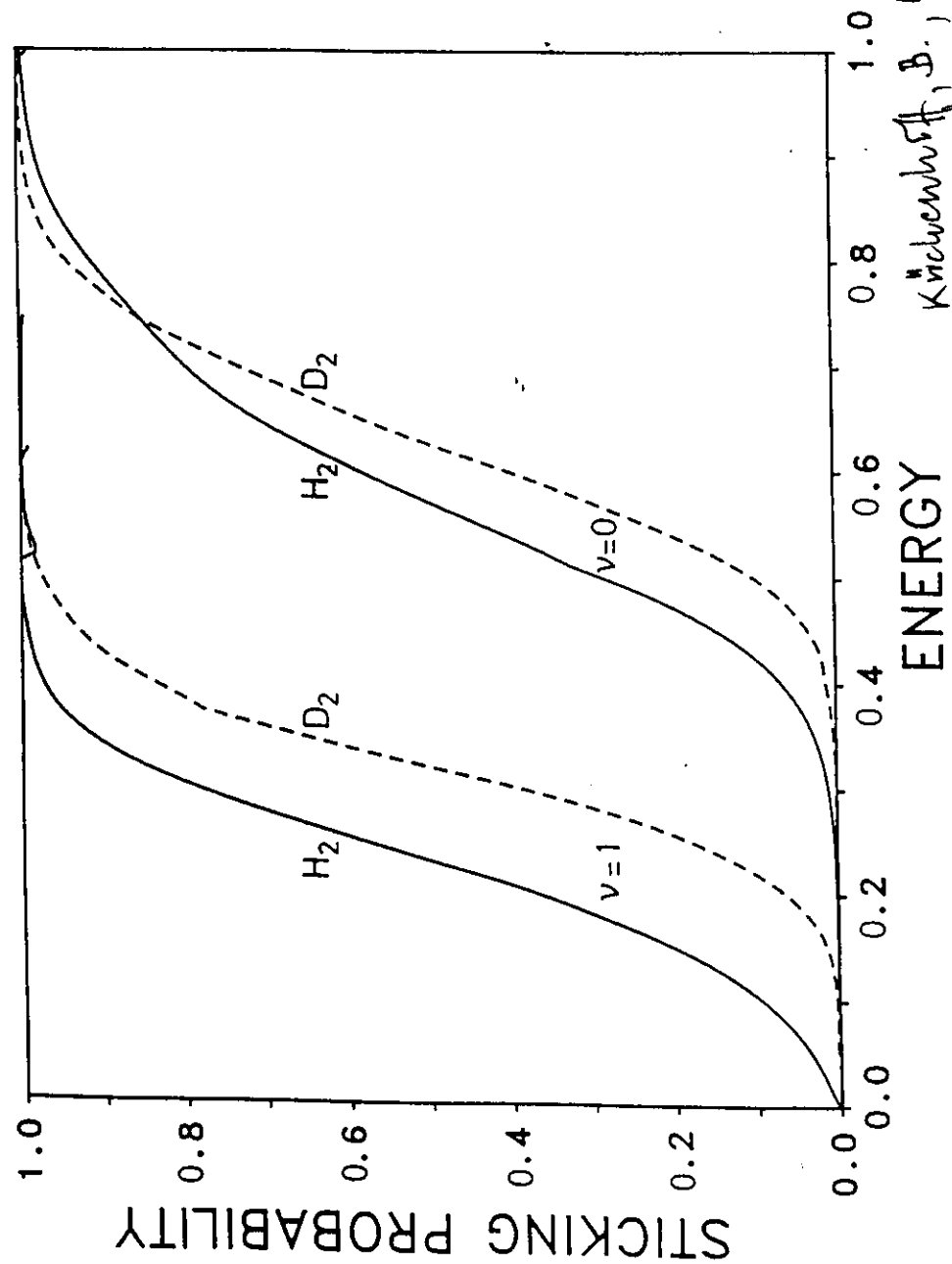
Sticking (state selective)

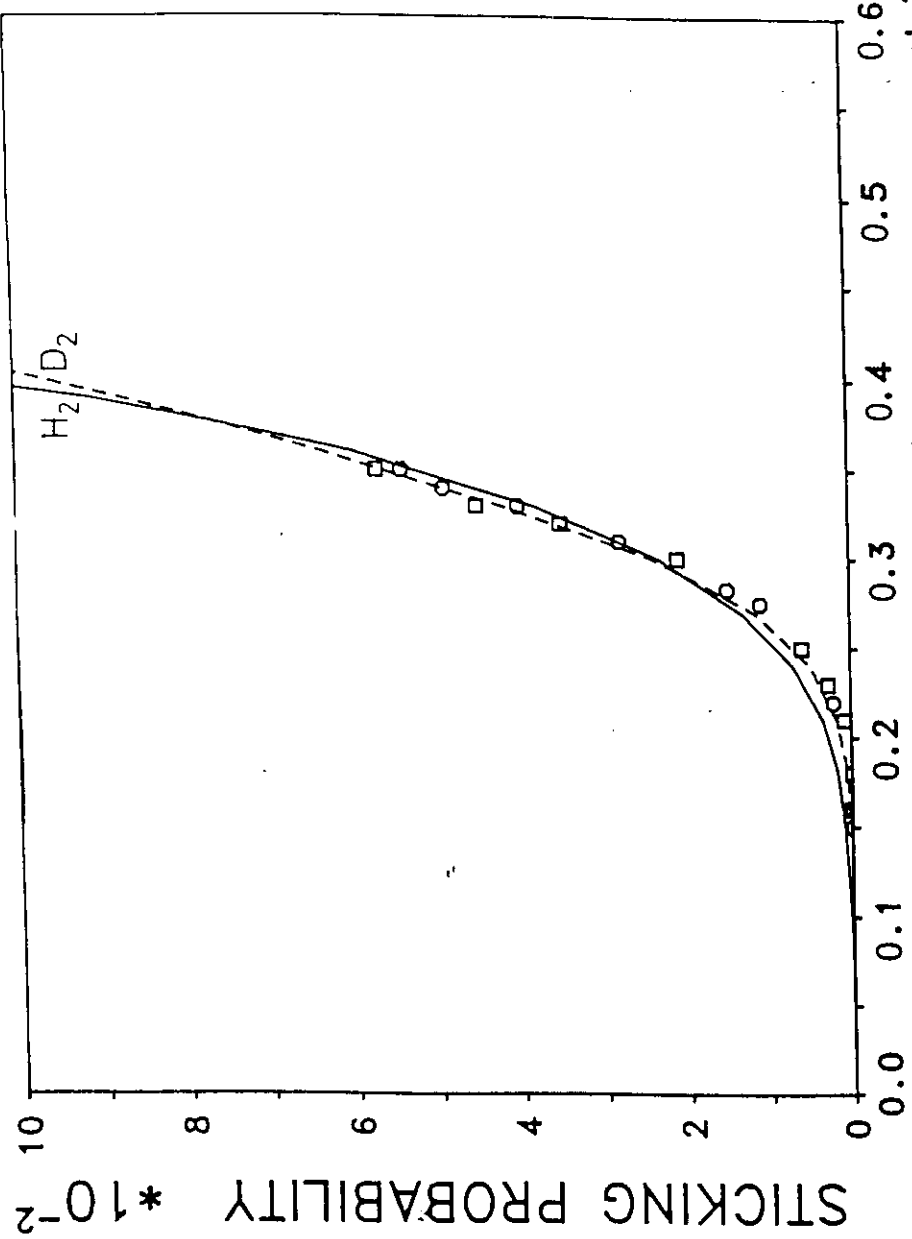


Detailed Balance (time rev. inv.)

$$D_{nm}(E_{tot}) = S_{mn}(E_{tot})$$

Fig. 6



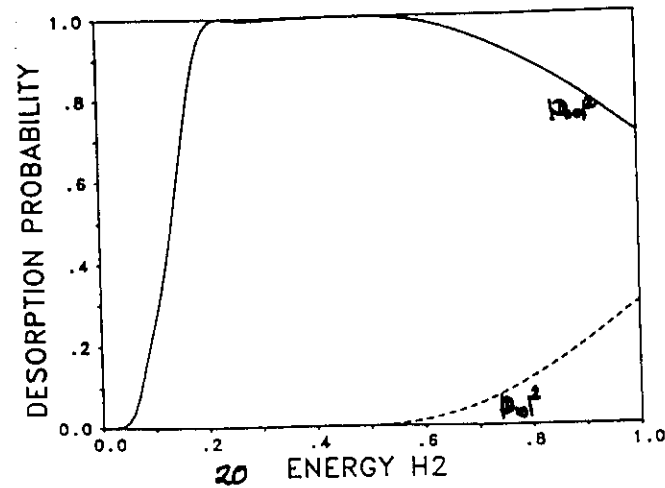
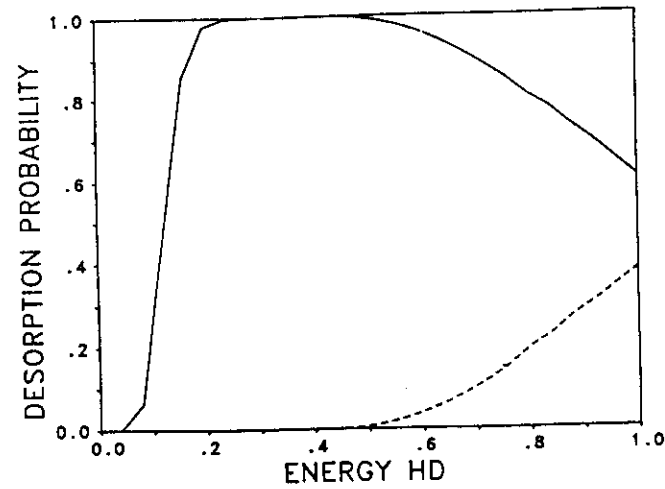
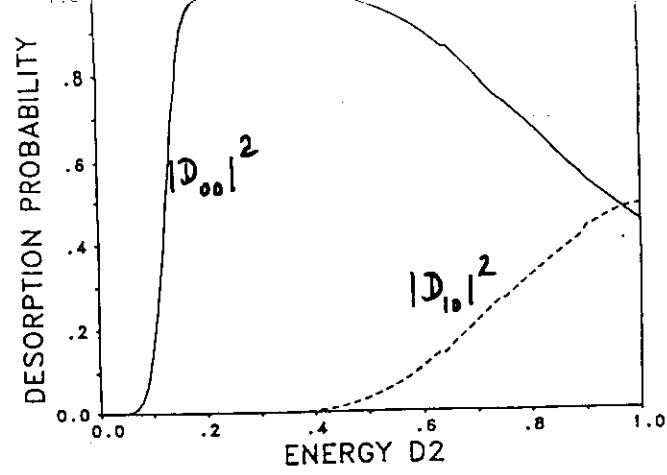


Exper: Rendulic et al. ENERGY μ : Kitchenhoff, B., Chiba

Hydrogen/Pd

Exper: Schröter, David, Zacharias

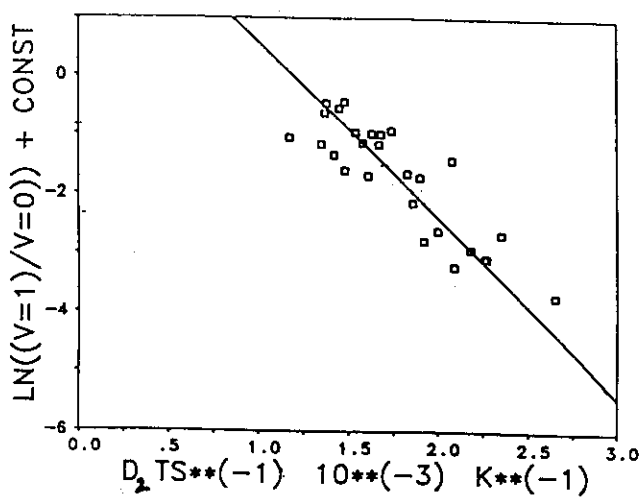
Theor: Kitchenhoff, B., Kasai



Hydrogen/
Pd

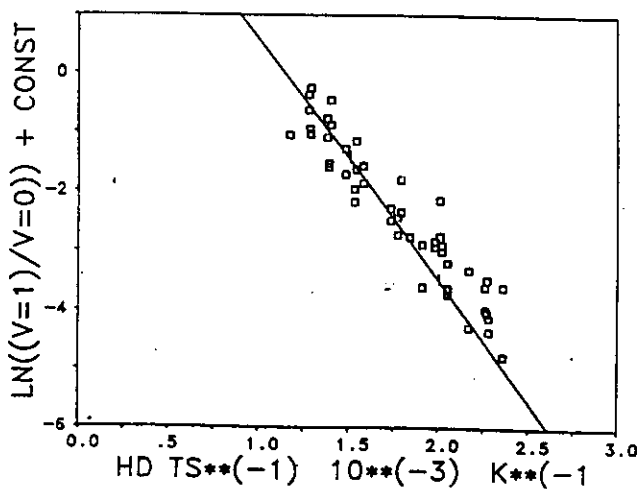
Exper.:
Schroter,
David,
Zacharias

Theor.:
Kitching,
B.

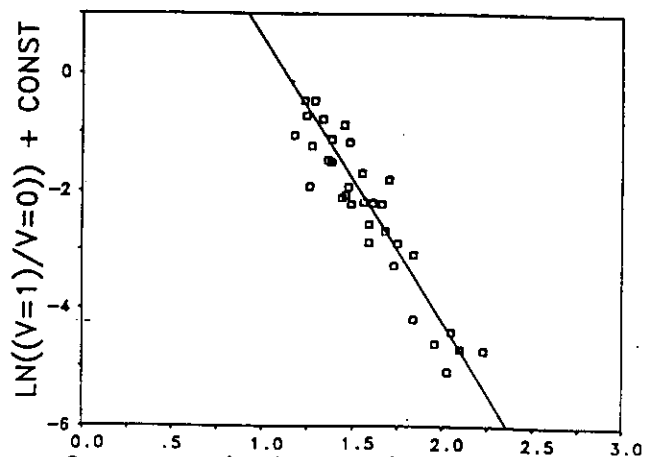


Ea

234 meV
(371)



282 meV
(450)



428 meV
(517)