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Workshop on Fluid Mechanics

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Porous Media: Ground Water Flows

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These are preliminary lecture notes, intended only for distribution to participants

- Chemical budget of land and sea
- Dispersion of pollutants
- Diagenetic reactions causing post-depositional mineralisation / alteration of sedimentary deposits.
- Heat budget - hot springs, hydrothermal vent
- Formation of hydrocarbons

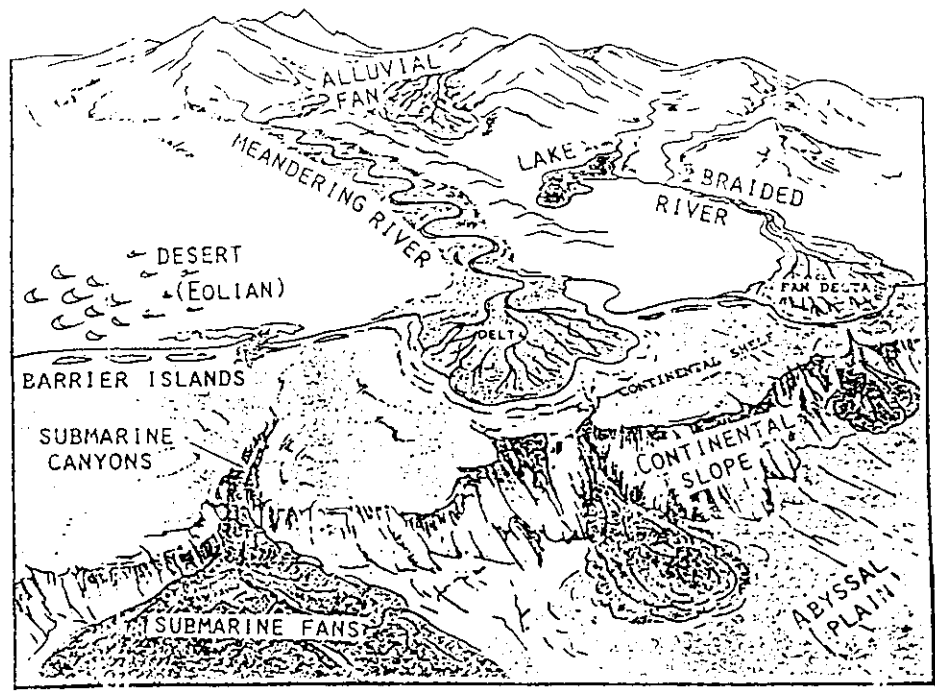


Fig. 5.1. Schematic representation of sedimentary environments on a passive margin

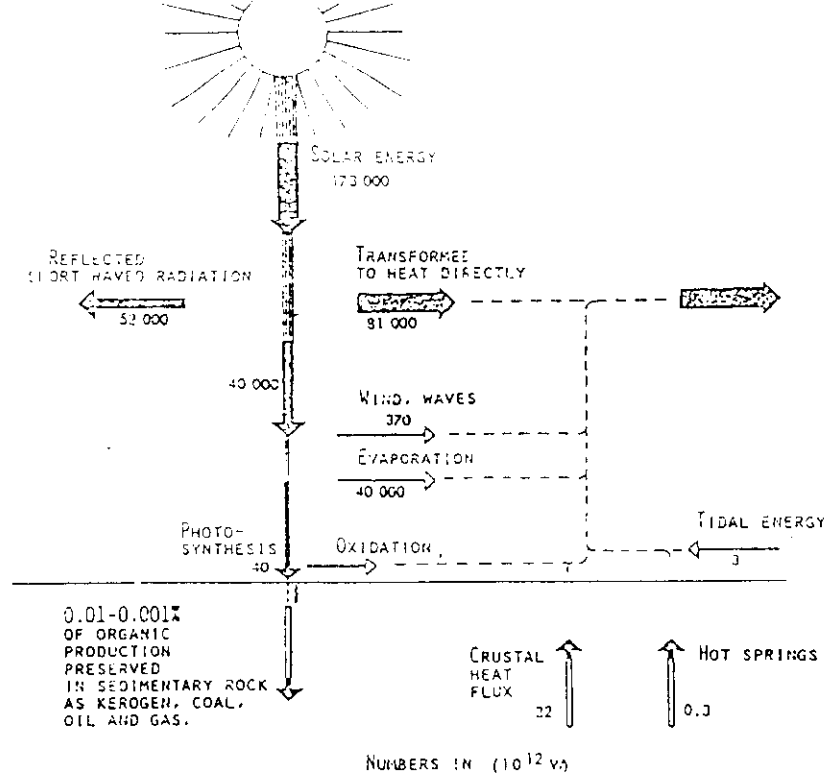


Fig. 13.1. Diagrammatic representation of the energy budget on the earth. We see that only a very small percentage of light energy is used for photosynthesis, and most of this (more than 99%) is broken down, with only a few ppm being converted into fossil energy

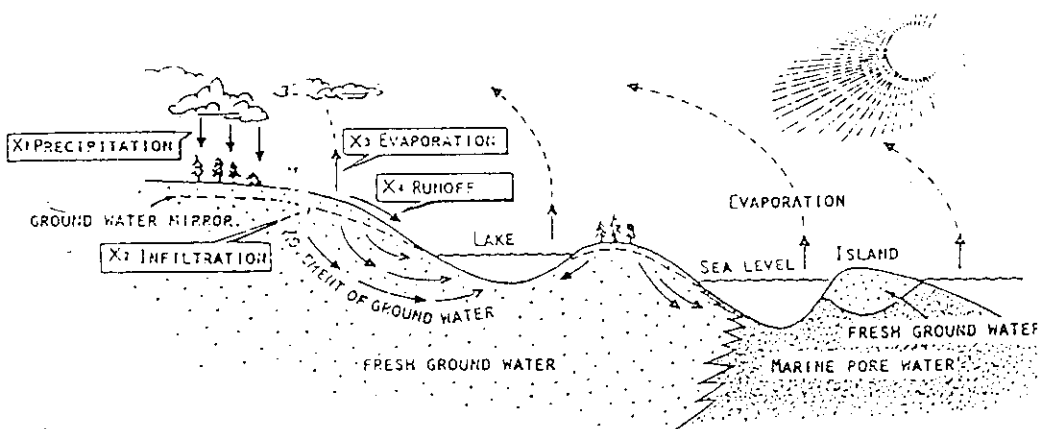


Fig. 5.4. Diagram showing the circulation of groundwater on land and under marine basins

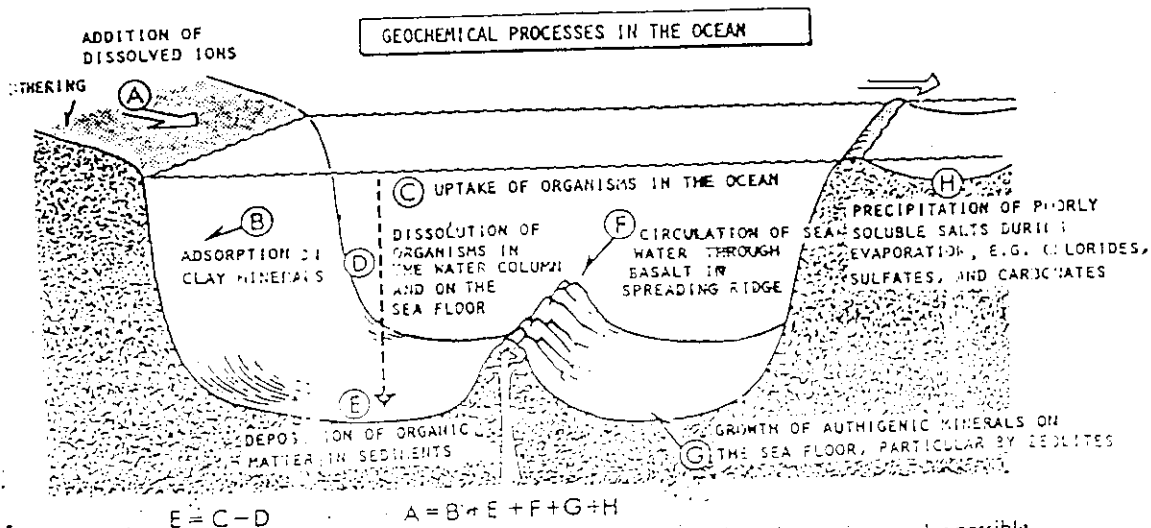


Fig. 7.8. Geochemical processes in the ocean. The addition of ions from the continents, plus possible precipitation of poorly soluble salts during evaporation, is compensated from the sea water by biological



Fluid flows thro' the
pore spaces

if $\phi =$ void fraction

$v =$ velocity of fluid in x -direction

mass flux in x direction through an area A is

give by $\int_A v \phi dA$

If ϕ is uniform, flux per unit area at
any point is $v\phi$

$v\phi$ is called the Darcy velocity, u , a measure
of the local volume flux. $[\nabla \cdot \underline{u} =$

If ϕ is uniform on the scale of the pore spacing,
but varies over longer length scales, we may
use u to describe the flow macroscopically



Figure 2.1. Extremely porous limestone from a Bermuda coral reef, approximately half-scale, containing many interstices with scales of up to a centimeter or so.

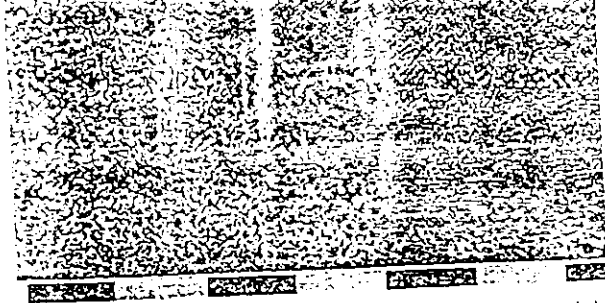


Figure 2.3. Pores in dolomite from the Latemar Massif in northern Italy. The blocks in the scale are 1 cm long. (Photograph courtesy of Dr. E. N. Wilson.)

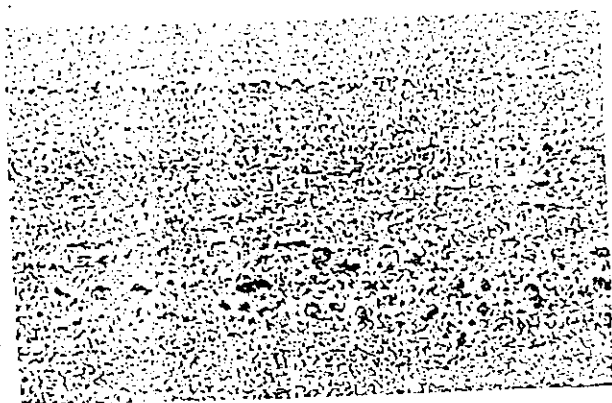


Figure 2.2. A highly porous, partially cemented sandstone, approximately full scale.

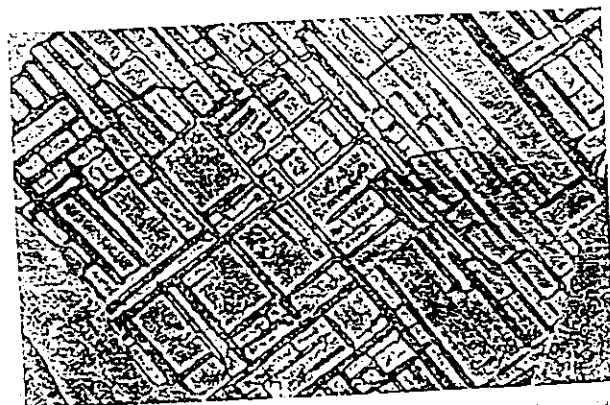


Figure 2.4. A network of cracks, made visible by staining, that provide pathways for flow along a sandstone cleavage plane. Approximately full scale.

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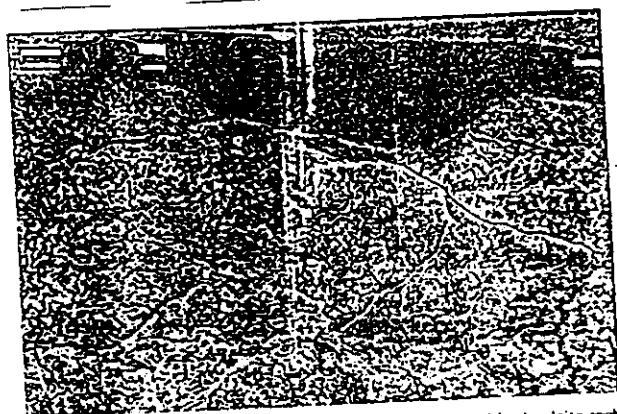


Figure 2.5. The weathered face of a partially dolomitized calcite rock in which veinlets of dolomite produced along fluid pathways stand out in relief in a dense, fine tracery on the left and along a crack on the right. (Photograph courtesy of Dr. E. N. Wilson.)



Figure 2.8. Dolomite of lighter shade has replaced calcite almost completely on the right, except in the cores of some of the grains; the reaction is only partially complete in the center. Magnification $\times 40$. (Photograph courtesy Dr. E. N. Wilson.)

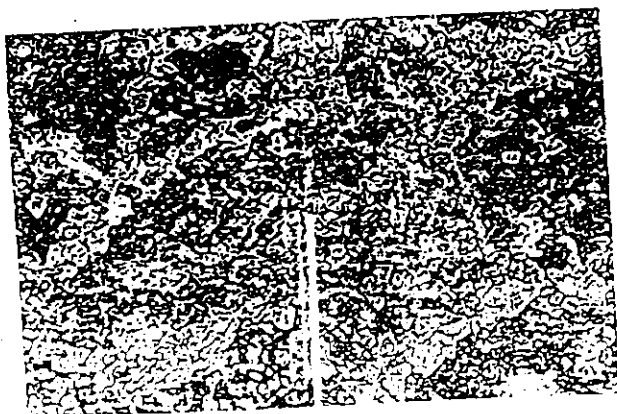


Figure 2.6. Dolomitization along linear pathways (the strings of lighter rhombic crystals) in a calcite matrix. Magnification $\times 20$. (Photograph courtesy of Dr. E. N. Wilson.)

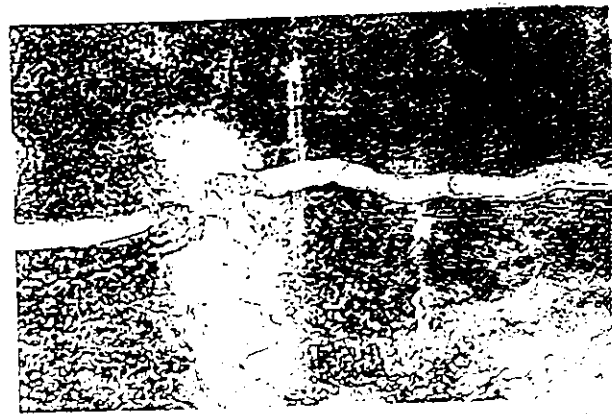


Figure 2.9. A veinlet in which calcite cement has precipitated from solution, intersecting and filling a pore. Magnification $\times 100$. (Photograph courtesy Dr. E. N. Wilson.)

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Darcy's Law

- Pressure gradients drive a flow
- Inertia + friction balance

In each pore, of width δ say, the viscous stresses exert a force

①

$$F_v \propto \frac{\mu v}{\delta^2}$$

μ = dynamic viscosity

δ = pore size

v = local velocity

and the inertia of the fluid generates pressure gradients of magnitude

②

$$\frac{\rho v^2}{\delta}$$

if the path changes direction on the lengthscale δ



$\frac{\delta \rho v}{\mu} \ll 1$
LOW REYNOLDS NUMBER

$$\text{if } \frac{\rho v^2}{\delta} \ll \frac{\mu v}{\delta^2} \Rightarrow v \ll \frac{\mu}{\delta \rho}$$

then the viscous drag dominates the inertia

$$u = v \phi \Rightarrow$$

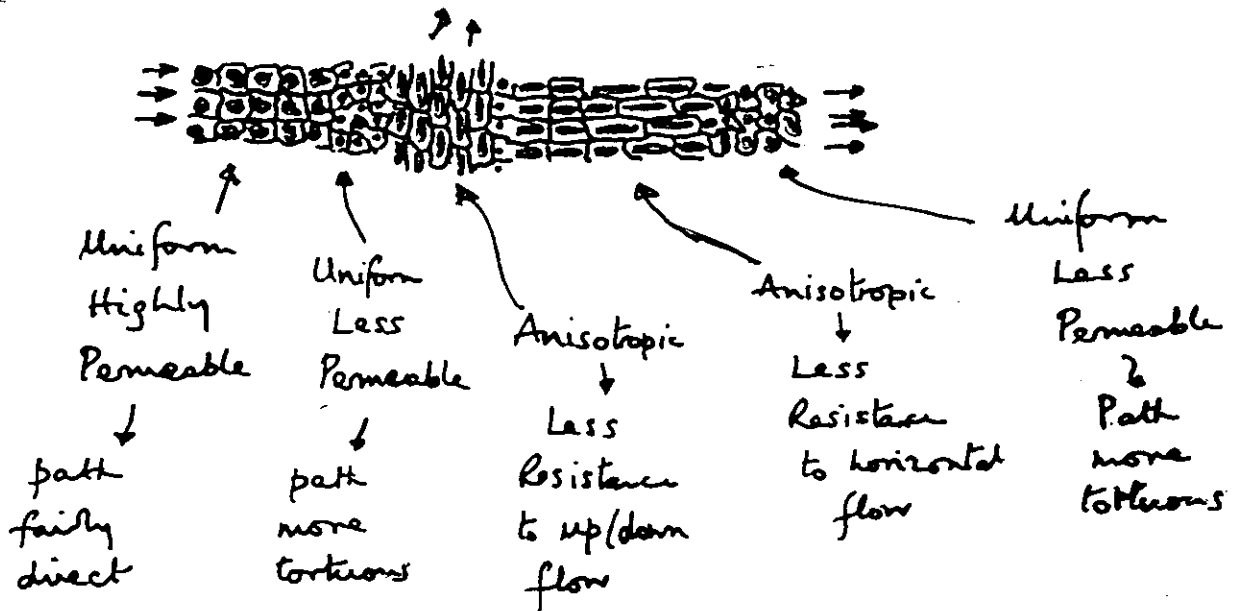
$$u \ll \frac{\mu \phi}{\rho \delta}$$

"Slow" Darcy Velocity
 \Rightarrow Friction dominates

precise form of k is not simple
 in many situations - typically, pore
 spaces are preferentially oriented in one
 direction $\Rightarrow k$ may be anisotropic

$$\text{Tensor } \underline{k} = \begin{pmatrix} k_{11} & k_{12} & k_{13} \\ k_{21} & k_{22} & k_{23} \\ k_{31} & k_{32} & k_{33} \end{pmatrix} \quad \left(\mu \underline{u} = -\underline{k} \cdot \nabla p \right)$$

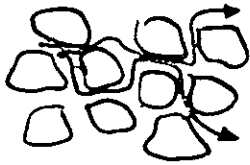
i.e. resistance to motion differs in
 different directions (and may differ
 through space as well!)



However, it is useful to study the case of
 uniform permeability first, and then to address
 anisotropic permeabilities - anisotropies important for
 mixing

Dispersion + Diffusion in Porous Flows

Two marked
fluid elts.
start close
together



Elements move
apart as a
result of dispersion by porous matrix.

Path of fluid from \underline{a} to $\underline{x}(\underline{a}, t)$

$$\text{If } \underline{v}(\underline{a}, t) = \frac{d}{dt} [\underline{x}(\underline{a}, t)] \quad \text{where } \underline{x}(\underline{a}, 0) = \underline{a}$$

$$\text{then } \underline{x}(\underline{a}, t) = \int_0^t \underline{v}(\underline{a}, t') dt' + \underline{a}$$

We seek to describe motion on macroscopic scale,
larger than the pore size l_0 . If the starting point
of our fluid element changes by an amount of
order ϵl_0 (i.e. we start at $\underline{a} + \epsilon \underline{l}_0$) then
the path may also change (blue line). - macroscopic
description \rightarrow we average over these possible starting values
If the mean velocity is $\bar{v}(\underline{a})$, then

$$\text{mean position } \bar{x}(\underline{a}, t) - \underline{a} = \int_0^t \overline{v(\underline{a}, t')} dt' = \bar{v}(\underline{a}) t$$

$$\text{So } \underline{x}(\underline{a}, t) - \bar{x}(\underline{a}, t) = \int_0^t [v(\underline{a}, t') - \bar{v}(\underline{a})] dt'$$

$$\Rightarrow \left[\frac{d}{dt} (\underline{x}(\underline{a}, t) - \bar{x}(\underline{a}, t)) \right] (\underline{x}(\underline{a}, t) - \bar{x}(\underline{a}, t)) = \int_0^t (v(\underline{a}, t) - \bar{v}(\underline{a})) (v(\underline{a}, t') - \bar{v}(\underline{a})) dt'$$

Hence by averaging over a small range of starting ^{various} positions we have

$$\frac{1}{2} \frac{d}{dt} \left\{ \overline{(x(a,t) - \bar{x}(a,t))^2} \right\} = \int_0^t \overline{(v(a,t) - \bar{v}(a)) (v(a,t') - \bar{v}(a))} dt'$$

$\overline{(v(a,t) - \bar{v}(a)) (v(a,t') - \bar{v}(a))}$ correlates the velocity fluctuations at time t and t' . As $(t-t')$ increases, the velocity fluctuations become more decorrelated - individual interstices are convoluted

on the length-scale of the interstices, and hence the velocity fluctuations flow becomes decorrelated when $(t-t') \sim \left(\frac{L_0}{\bar{v}}\right)$

Hence $\overline{(v(a,t) - \bar{v}(a)) (v(a,t') - \bar{v}(a))} = \bar{v}^2 f(t-t')$

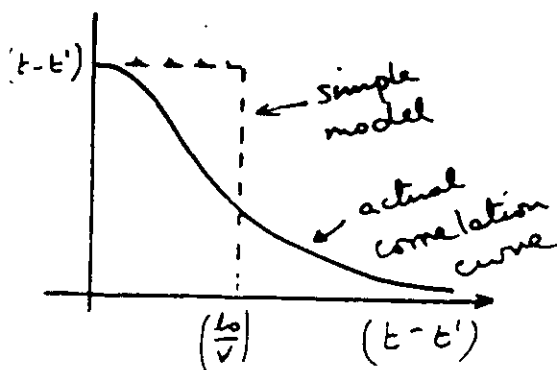
and $\int_0^t d(t-t') \left[\bar{v}^2 f(t-t') \right] \sim \bar{v}^2 \left(\frac{L_0}{\bar{v}}\right)$

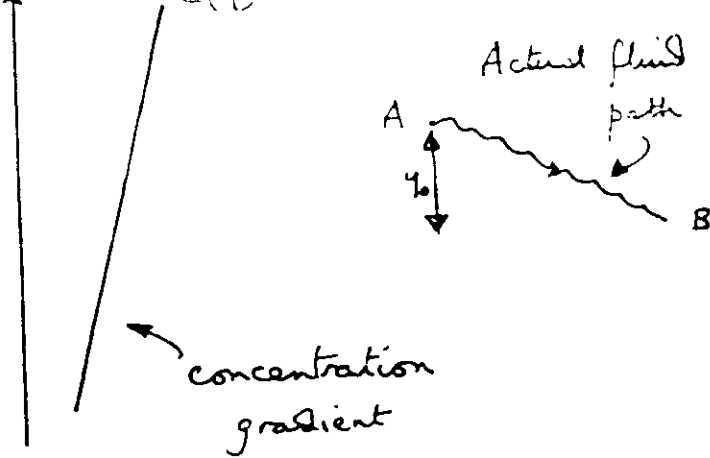
(note, decorrelation time may differ in an anisotropic medium)

Thus $\frac{d}{dt} \left(\overline{(x(a,t) - \bar{x}(a,t))^2} \right) = (2 \bar{v} L_0)$

$\Rightarrow \overline{(x(a,t) - \bar{x}(a,t))^2} = (2 \bar{v} L_0) t$

Hence fluid elements dispersed about mean position by an amount $[(2 \bar{v} L_0) t]^{1/2}$ in time t .





direction of
 mean
 motion
 →
 Vertical deviation of
 the motion of the
 fluid particles can
 cause a dispersive
 concentration flux

Effect of particle moving from A to B

$$C(y_0) = C(0) + y_0 \frac{dc}{dy}$$

← difference in
 concentration
 at A and B

If path deviates by amount y_0 , then

$$-y_0 = \int_{t_0}^{t_f} (v(a, t') - \bar{v}) dt' \quad \left(\begin{array}{l} \text{downward} \\ \text{deviation} \end{array} \right)$$

⇒ Net dispersive flux/arriving at B is $\overline{C(v(a, t_f) - \bar{v})}$ in fluid

averaging over all a , $\overline{(v(a, t_f) - \bar{v})} = 0$

hence $\overline{C(v(a, t_f) - \bar{v})} = \overline{C'(v(a, t_f) - \bar{v})}$

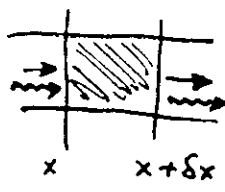
$$\text{and } C' = y_0 \frac{dc}{dy} = - \int_{t_0}^{t_f} (v(a, t') - \bar{v}) dt' \frac{dc}{dy}$$

$$\Rightarrow \overline{C'(v(a, t_f) - \bar{v})} = \frac{dc}{dy} \int_{t_0}^{t_f} \overline{(v(a, t') - \bar{v})(v(a, t') - \bar{v})} dt'$$

$$\Rightarrow \text{Dispersive flux per unit area} = -\phi(\bar{v} l_0) \left(\frac{dc}{dy} \right)$$

⇒ Dispersion enhances
 coefficient
 diffusion by $(\bar{v} l)$

Heat Transfer



Assume rock + liquid are locally in thermal eqm
 $\Rightarrow k \gg \frac{u \delta}{\phi}$ Typically True

→ Advective heat flux into shaded box = $\rho C_p \int_L u T(x)$

→ Advective heat flux out of shaded box = $\rho C_p \int_L u T(x + \delta x)$

→ Diffusive heat flux = $-\bar{k} T_x \Big|_x$ into

= $-\bar{k} T_x \Big|_{x+\delta x}$ out of

Hence $\bar{\rho} C_p \frac{\partial T}{\partial t} = -\rho C_{pL} \nabla \cdot (\underline{u} T) - (-\bar{k} \nabla^2 T)$

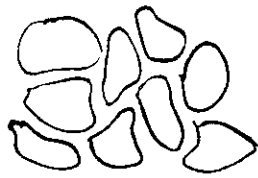
⇒ $\bar{\rho} C_p \left(\frac{\partial T}{\partial t} \right) + \rho C_{pL} (\underline{u} \cdot \nabla) T = \bar{k} \nabla^2 T$
 \bar{x} means the average
 e.g. $(1-\phi)C_{pS} + \dots$

Here we use result that $\nabla \cdot \underline{u} = 0$ since

the net ~~flux~~ flux does not change of liquid



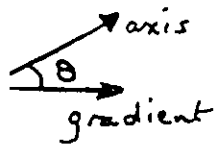
\underline{u} = Darcy velocity
 = flux of liquid per unit area



Molecular Diffusion acts

along pore axes

Molecular flux = $-D_m \frac{\partial c}{\partial s}$ s is in along axis direction



$\frac{\partial c}{\partial s}$ has a component $\cos \theta \frac{\partial c}{\partial s}$ in direction of gradient

So molecular flux = $-D_m \cos \theta \frac{\partial c}{\partial s} = -D_m \cos^2 \theta \frac{\partial c}{\partial y}$

as $ds = dy / \cos \theta$

Net molecular flux, averaged macroscopically is therefore

$$-\phi D_m \overline{\cos^2 \theta} \frac{\partial c}{\partial y} \approx -\phi D_m \overline{\cos^2 \theta} \frac{\partial \bar{c}}{\partial y}$$

(Approximation)

($\overline{\cos^2 \theta}$ called tortuosity)

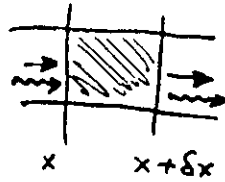
Total diffusive + dispersive transport is

$$-\phi \left\{ D_m \overline{\cos^2 \theta} + \bar{v} l_0 \right\} \frac{\partial \bar{c}}{\partial y} \quad (\bar{v} \text{ is the mean interstitial velocity})$$

or $-\left\{ \phi D_m \overline{\cos^2 \theta} + \bar{u} l_0 \right\} \frac{\partial \bar{c}}{\partial y}$ Diffusion Dominates:

$u = \text{Darcy velocity}$. Typically $\bar{u} \sim 10^{-5} \frac{\text{cm}}{\text{s}}$; $D_m \sim 10^{-5} \text{cm}^2/\text{s} \Rightarrow l_0 \ll D_m / \bar{u}$

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Assume rock + liquid are locally in thermal eqm
 $\Rightarrow k \gg \frac{u \delta}{\phi}$ Typically True

① \rightarrow Advective heat flux into shaded box = $\rho C_p |u| T(x)$

\rightarrow Advective heat flux out of shaded box = $\rho C_p |u| T(x + \delta x)$

② \rightarrow Diffusive heat flux = $-\bar{k} T_x|_x$ into

= $-\bar{k} T_x|_{x+\delta x}$ out of

③ Hence $\bar{\rho} C_p \frac{\partial T}{\partial t} = -\rho C_{pL} \nabla \cdot (\underline{u} T) - (-\bar{k} \nabla^2 T)$

$\Rightarrow \bar{\rho} C_p \left(\frac{\partial T}{\partial t} \right) + \rho C_{pL} (\underline{u} \cdot \nabla) T = \bar{k} \nabla^2 T$

\bar{x} means the average
 e.g. $(1-\phi)C_{pS} + \phi C_{pL}$

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\underline{u} = Darcy velocity
 = flux of liquid per unit area

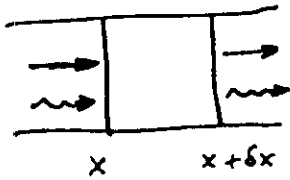
Tracer Transport

- e.g. Dye

Salt

Radio-activity

Consider Non-Reacting System



Let C be the concentration of tracer in the liquid

In the volume between x and $x + \delta x$, the amount of tracer is

$$C \phi \delta x \quad \text{per unit sect } x\text{-fional area}$$

The advective flux of tracer is $-\left\{ uC \Big|_{x+\delta x} - uC \Big|_x \right\}$

$$- (\underline{u} \cdot \underline{\nabla}) C$$

The diffusive flux of tracer is $-\left\{ -\phi D \frac{\partial C}{\partial x} \Big|_{x+\delta x} + \phi D \frac{\partial C}{\partial x} \Big|_x \right\}$

$$\phi D \nabla^2 C \quad \text{remember } D = (D_m \overline{\cos^2 \theta} + \frac{\bar{u} b}{\phi})$$

Conservation of tracer is

$$\phi \frac{\partial C}{\partial t} + (\underline{u} \cdot \underline{\nabla}) C = \phi D \nabla^2 C$$

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Important Result : Tracer travels faster than heat.
anomaly

Physics : { Tracer moves thro' liquid^{phase} only and
is carried by flow

{ Thermal anomaly must heat up rock
as it is swept through the material by
the liquid \rightarrow thermal inertia

Heat :
$$\frac{\partial T}{\partial t} + \left(\frac{\rho C_p \epsilon}{\bar{\rho} C_p} \right) (\underline{u} \cdot \nabla) T = \left(\frac{\bar{k}}{\bar{\rho} C_p} \right) \nabla^2 T$$

Tracer :
$$\frac{\partial c}{\partial t} + \left(\frac{1}{\phi} \right) (\underline{u} \cdot \nabla) c = D \nabla^2 c$$

\Rightarrow Effective advection speed of tracer is $\left(\frac{u}{\phi} \right)$
(i.e. the real velocity)

Effective advection speed of heat is $\left[\left(\frac{\rho C_p \epsilon}{\bar{\rho} C_p} \right) u \right] \approx 1$

Ratio
$$\frac{\text{Tracer Speed}}{\text{Thermal Speed}} \approx \frac{1}{\phi} \gg 1 \quad (\phi \sim 10^{-5})$$

\rightarrow Pump ~~red~~^{green + hot} water into an aquifer : cold ~~red~~^{green} water appears

Consider ~~two~~ dimensional flows :

$$\underline{u} = -\frac{k}{\mu} \underline{\nabla} P$$

In a region of constant, k , permeability,

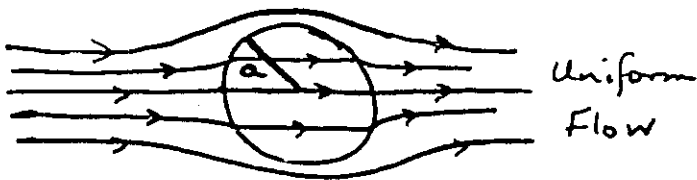
$$\underline{\nabla} \cdot \underline{u} = -\frac{k}{\mu} \nabla^2 P \quad \text{but} \quad \underline{\nabla} \cdot \underline{u} = 0$$

mass conservation

\Rightarrow $\nabla^2 P = 0$ and $\frac{k}{\mu} P$ is a velocity potential for the flow

$$(2-D \Rightarrow \nabla^2 P = \frac{1}{r} \frac{\partial}{\partial r} (r \frac{\partial P}{\partial r}) + \frac{1}{r^2} \frac{\partial^2 P}{\partial \theta^2})$$

e.g. Consider flow round a cylinder of different permeability



Cf. heat transfer around a cylinder of different thermal conductivity

if outside permeability is k_0
 interior " " is k_i

then outside $p \rightarrow -w r \cos \theta$ as $r \rightarrow \infty$
 "($r \cos \theta = x$)"

so expect $p_o = -\left(1 - \frac{B}{r^2}\right) w r \cos \theta$ $r > a$
 Outside

and $p_i = -C w r \cos \theta$ $r < a$
 Inside

with $p_o(z) = p_i(z)$ \leftarrow pressure continuous across boundary

$-\frac{k_0}{\mu} \frac{\partial p_o}{\partial r} \Big|_r=a} = -\frac{k_i}{\mu} \frac{\partial p_i}{\partial r} \Big|_r=a}$ \leftarrow mass flux continuous across boundary

$$\text{Matching} \Rightarrow \left(1 - \frac{B}{a^2}\right) = C$$

$$k_o \left(1 + \frac{B}{a^2}\right) = k_i C$$

$$\Rightarrow \begin{cases} C = \frac{2k_o}{k_o + k_i} \\ B = \left(\frac{k_i - k_o}{k_i + k_o}\right) a^2 \end{cases}$$

Physics:

$k_o \gg k_i \rightarrow$ exterior more permeable than interior of cylinder

\rightarrow flow tries to go round cylinder rather than thro' cylinder

Check:

in far field flow $\underline{u}_o \sim \frac{k_o}{\mu} w$

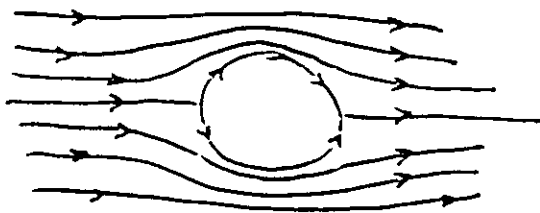
inside cylinder flow is uniform $\underline{u}_i \sim \frac{k_i}{\mu} \left(\frac{2k_o}{k_o + k_i}\right) w$

ratio

$$\frac{\underline{u}_o}{\underline{u}_i} \sim \frac{k_o(k_o + k_i)}{2k_o k_i}$$

If $k_o \gg k_i$, $\frac{\underline{u}_o}{\underline{u}_i} \sim \frac{k_o}{2k_i} \gg 1$

\rightarrow all flow is outside



Hydrodynamic Dispersion - Macro-Scopic

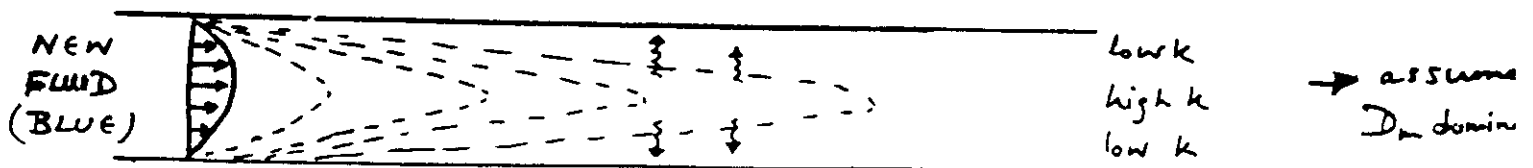
We have seen dispersion acts on the scale of the interstices to mix up the fluid with an effective "local" diffusion coefficient

$$\left\{ D_m \overline{\cos^2 \theta} + \frac{u l_0}{\phi} \right\}$$

However, larger-scale shear in the flow can induce mixing across the flow by intensifying concentration gradients

We may study the larger scale "shear-induced" dispersion as a process of

- (1) Shearing + Intensifying gradients
- (2) Diffusing across gradients

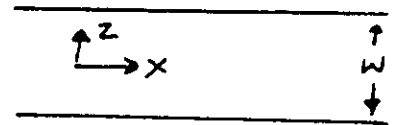


In absence of diffusion cross-flow, a tongue of blue fluid advances into aquifer. - Dispersion causes tracer to mix cross flow \rightarrow first arrival time increases

If we move to the frame of the centre of mass of the new fluid, we expect the cross-stream concentration to equilibrate s.t. the along stream advection balances cross stream diffusion

$$\frac{1}{\phi} \frac{\partial c}{\partial t} + \left(\frac{u}{\phi} \right) \frac{\partial c}{\partial x} = D \left(\frac{\partial^2 c}{\partial x^2} + \frac{\partial^2 c}{\partial z^2} \right)$$

Write $c = \bar{c}(x) + \hat{c}(x, z)$



\bar{c} = mean concentration at location x $\int_0^w dz c(x, z) / w$

\hat{c} = fluctuations about mean, $\hat{c} = c - \bar{c}$

$$\left[\frac{\partial \bar{c}}{\partial t} + \bar{u} \frac{\partial \bar{c}}{\partial x} + \frac{\int \hat{u} \frac{\partial \hat{c}}{\partial x} dz}{w} = D \frac{\partial^2 \bar{c}}{\partial x^2} \right] \text{ underlined term is dispersive flux}$$

Expect $\frac{\partial^2 \hat{c}}{\partial z^2} \sim \left(\frac{\hat{u}}{\phi D} \right) \left(\frac{\partial \bar{c}}{\partial x} \right)$

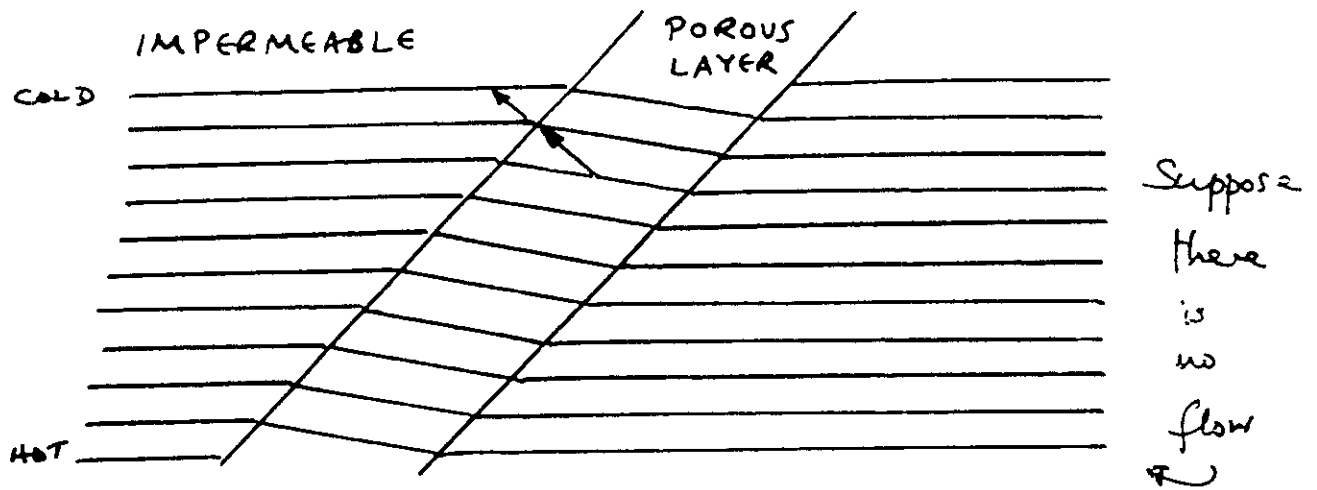
i.e. shear induces cross stream conc'n gradients and these are diffused across stream \hat{u} is magnitude of perturbation of flow from mean.

$$\text{so } \hat{c} \sim \int \int \frac{\hat{u}}{\phi D} \frac{\partial \bar{c}}{\partial x} dz' dz''$$

$$\hat{c} \sim \frac{|\hat{u}|}{\phi D} \frac{\partial \bar{c}}{\partial x} w^2$$

$$\Rightarrow \text{Dispersive flux along the aquifer} = \int_0^w \hat{u} \hat{c} dz \sim \frac{|\hat{u}|^2 w^3}{\phi D} \left(\frac{\partial \bar{c}}{\partial x} \right)$$

Hence in stationary frame $\frac{\partial \bar{c}}{\partial t} \sim -\bar{u} \frac{\partial \bar{c}}{\partial x} + \left(D + \frac{|\hat{u}|^2 w^2}{\phi D} \right) \frac{\partial^2 \bar{c}}{\partial x^2}$



If thermal conductivity in a ~~rock~~ porous layer exceeds that in the rock, then the temperature gradient \perp to the wall is larger in the rock (as shown otherwise it is smaller)

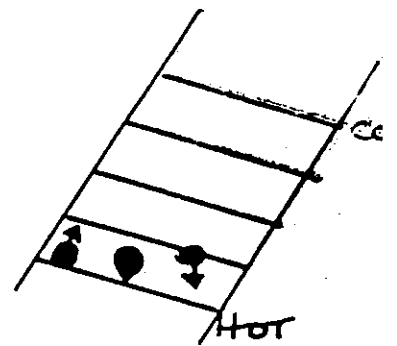
In the rock, the isotherms are horizontal
 \rightarrow In the porous layer they are inclined

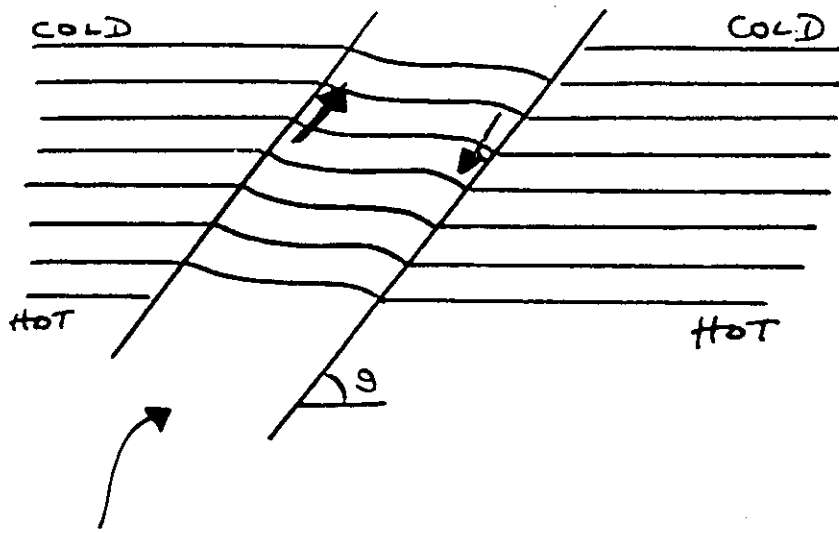
But
$$\rho(T) = \rho(0)(1 - \alpha T)$$

So fluid near upper wall is

hot \rightarrow less dense \rightarrow rises

Fluid near lower wall \leftarrow more cold \rightarrow dense \rightarrow sinks





Flow induced in the porous layer

$$\underline{u} = -\frac{k}{\mu} (\underline{\nabla} p - \rho \underline{g})$$

← Darcy with gravity

$$(\underline{u} \cdot \underline{\nabla}) T = \bar{k} \nabla^2 T$$

← Steady flow HEAT CONS'

So $\underline{\nabla} \wedge \underline{u} = +\left(\frac{k}{\mu}\right) \rho g \alpha (\underline{\nabla} \wedge T \hat{z})$

Vorticity Eqn.

If flow is purely along slot ; $y =$ distance across slot

then

$$u_{yy} = g \frac{k}{\mu} \rho g \alpha T_{yy} \sin \theta$$

← from vorticity eqn.

$$\sin \theta u T_{0z} = \bar{k} T_{yy}$$

← from diffusion eqn.

$$\rightarrow u = \left(\frac{\bar{k} k \rho g \alpha T_{0z}}{\rho g \alpha T_{0z}} \right) (\sin \theta)^{-2} u_{yy}$$

So $u_{212} = \left(\frac{\sin^2 \theta T_{o2} g k \rho(\theta) \alpha}{\mu \bar{k}} \right) u_2$

$\Rightarrow \underline{Ra = \left(\frac{\sin^2 \theta T_{o2} g k \rho(\theta) \alpha}{\mu \bar{k}} \right) d^2}$

Length of flow in cross slot direction

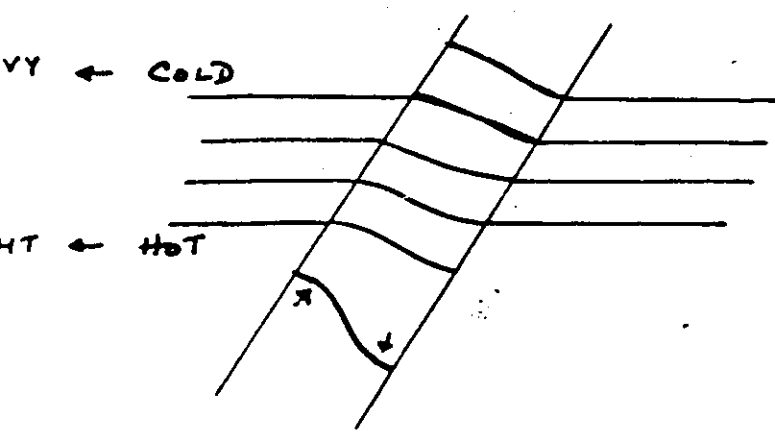
is $\underline{l \sim (Ra)^{-\frac{1}{2}} d}$

Stable $\left\{ \begin{array}{l} \Rightarrow \text{low } Ra, \text{ then flow extends across layer} \\ \text{(diffusion cross slot faster than advection along slot)} \\ \Rightarrow \text{high } Ra, \text{ flow confined to b'dry layer} \end{array} \right.$

$A = \frac{(\Delta \bar{k}) / k_o \cdot \cos \theta}{\cosh(Ra/2)}$

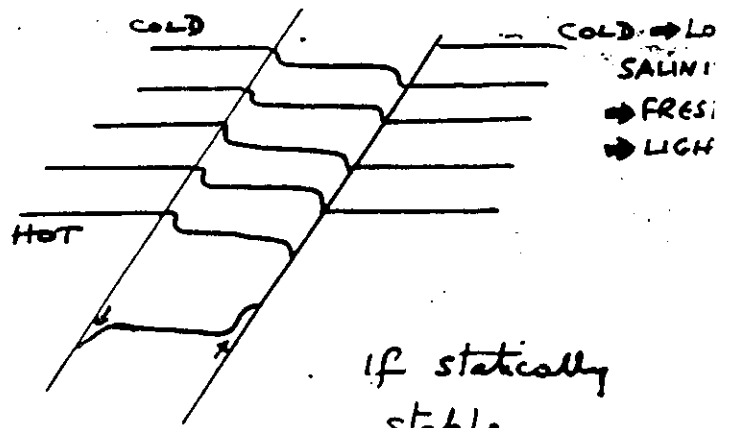
Unstable \rightarrow Flow always extends across layer

$= A \sin\left(\frac{2(Ra)^{1/2}}{d}\right)$



If statically
unstable

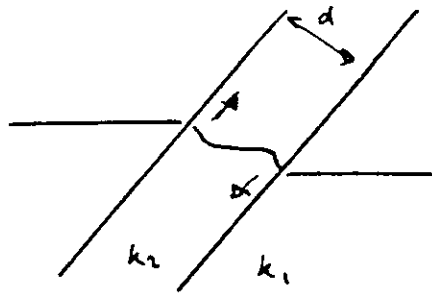
$$\alpha > 0$$



If statically
stable

$$\alpha < 0$$

Process may be important in a number of situations, even with mean zero flow -



Consider flow induced by geothermal temperature gradient and differences in the thermal conductivity (end of lecture notes; set 1).

Flow scales as $\mu \alpha \left((\epsilon - 1) \frac{k_1}{d} \right) \cot \phi g(Ra) \quad \epsilon = \frac{k_1}{k_2}$

(Dimensional argument flow = $f(Ra) \left(\frac{k}{d} \right)$)

↑ Velocity scale of diffusion. If $Ra \gg 1$ then flow becomes confined to boundary layer.

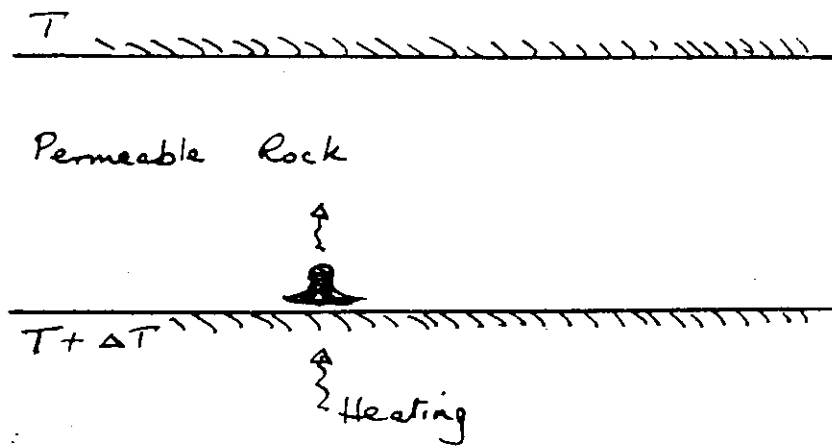
Then dispersion $\sim \left\{ \frac{(\bar{u})^2 d^2}{\phi D} \right\} \sim \left(\frac{k^2}{\phi D} \right) \cot^2 \phi$

So $\frac{\partial \bar{c}}{\partial t} \sim \left(\phi D + \alpha \frac{k^2}{D} \right) \frac{\partial^2 \bar{c}}{\partial z^2} \quad (\alpha = \cot^2 \phi (\epsilon - 1)) \quad (\alpha \sim 1)$

if $k \gg D$, as is typical $\frac{k}{D} \sim 10^2$, then material is dispersed ^{than by diffusion} 10^2 faster $\Rightarrow L_D \sim 10 \text{ km}$

Instabilities in Porous Layers.

1. Rayleigh - Darcy Instability



As a result of heating the floor, thermal instability and hence motion may ensue

However, conduction of heat + friction tend to suppress this motion

Velocity scale $\frac{k}{\mu} (\alpha \Delta T) g \rho_0$ (from Darcy's Law)

Effective conduction velocity scale $\frac{k}{d}$ ← { velocity scale for rate of heat transfer by conduction

if $\frac{\frac{k}{\mu} (\alpha \Delta T) g \rho_0}{(k/d)} > 0(1)$ then

we expect motion

In fact $\left(\frac{k \alpha \Delta T g}{\nu k} \right) d > 4\pi^2$ for motion

Formally use stability theory (c.f. H')

- ① $\frac{\partial p}{\partial z} = -\frac{\mu}{k_v} w - \rho g$ Vertical component of Darcy
- ② $\nabla_h p = -\frac{\mu}{k_h} \underline{u}_h$ horizontal part of Darcy
 $\nabla_h = (\partial_x, \partial_y)$
- ③ $\nabla \cdot \underline{u} = 0$ incompressibility \Rightarrow Boussinesq
- ④ $\Gamma \frac{\partial T}{\partial t} + (\underline{u} \cdot \nabla) T = \kappa \nabla^2 T$ thermal eqn. \Rightarrow conv'n of thermal energy
- ⑤ $\rho = \rho_0 (1 - \alpha T)$

State of Rest is a solution

$$\underline{u}_h = w = 0, \quad p = p_0(z)$$

{ Steady conduction across the layer }

$$\frac{dp_0}{dz} = -\rho g$$

$$T = T_0 + \left(1 - \frac{z}{h}\right) \Delta T$$

Perturb about state of Rest

$$\underline{u} \rightarrow \underline{u}'(x, y, z, t) \quad p \rightarrow p_0(z) + p'(x, y, z, t)$$

$$T \rightarrow T_0 + \Delta T \left(1 - \frac{z}{h}\right) + T'(x, y, z, t)$$

assume ' is a small perturbation \Rightarrow linearise

$$\textcircled{1} \quad \frac{\partial p'}{\partial z} = -\frac{\mu}{k_v} w' + \rho_0 g \alpha T'$$

$$\textcircled{2} \quad \nabla_h p' = -\frac{\mu}{k_h} \underline{u}'_h$$

$$\nabla \cdot \underline{u}' = 0$$

$$\textcircled{3} \quad \Gamma \frac{\partial T'}{\partial t} + \underline{u}' \cdot \nabla T' - w' \frac{\Delta T}{h} = \kappa \nabla^2 T'$$

and seek 2-Dim'l, temporal perturbations of the form

$$\psi = e^{nt} e^{i(kx + mz)}$$

Combining ① + ② from last page, we have

$$0 = -\frac{\mu}{k_v} \psi_{xx} - \frac{\mu}{k_h} \psi_{zz} - \rho_0 g \alpha T'_x \quad (4)$$

$$\textcircled{3} \Rightarrow \Gamma \frac{\partial T'}{\partial t} + \psi_x \frac{\Delta T}{h} = k \nabla^2 T' \quad (5)$$

On the boundaries, upper + lower,

$$T' = 0 \quad \text{and} \quad w = \psi_x = 0$$

hence we set

$$\begin{cases} T' = \hat{T} e^{nt} \sin\left(\frac{\pi m z}{h}\right) e^{ikx/h} & (6) \\ \psi' = \hat{\psi} e^{nt} \sin\left(\frac{\pi m z}{h}\right) e^{ikx/h} & (7) \end{cases}$$

$\left. \begin{array}{l} m \text{ is vertical wave no.} \\ l \text{ is horizontal wave no.} \end{array} \right\}$

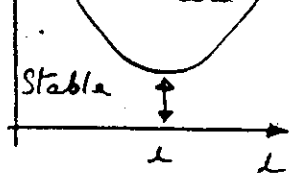
n is growth rate

Combining ④ + ⑤ and substituting ⑥ + ⑦ we have

$$\frac{n k^2 \Gamma}{k} = \frac{l^2}{r l^2 + m^2 \pi^2} Ra - (l^2 + m^2 \pi^2) \quad r = \frac{k_h}{k_v}, Ra = \frac{g \alpha \Delta T}{\nu}$$

$\underline{n < 0}$ we have stability $\Rightarrow Ra \leq \frac{(l^2 + m^2 \pi^2)(r l^2 + m^2 \pi^2)}{l^2}$

$$Ra \leq \frac{(l^2 + m^2 \pi^2)(r^2 l + m \pi)}{l^2}$$



m is integer \rightarrow vertical wave number

fix m , then Ra has a minimum as l varies

Minimum occurs at $l^2 = \frac{l}{\sqrt{r}} m^2 \pi^2$ and so we

require
$$Ra \leq \frac{(1 + \sqrt{r})^2 m^2 \pi^2}{1}$$

This must hold for all $m \rightarrow m=1$ is most unstable
(i.e. 1 cell in vertical).

For total stability, we therefore need

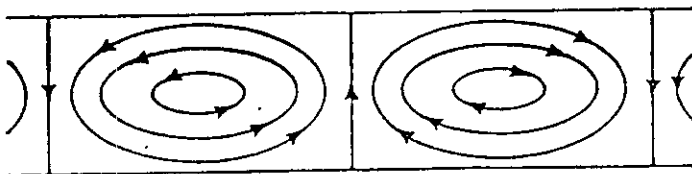
$$Ra \leq (1 + \sqrt{r})^2 \pi^2$$

if $r=1$ then
 $\left(\frac{k_H}{k_V}\right)$

$$Ra = \frac{k \alpha \Delta T g}{k \nu} \leq 4\pi^2$$

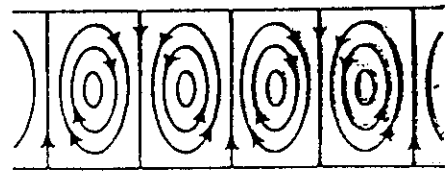
for no motion

More permeable in horizontal
 \rightarrow cells asymmetric (long, squat)



$$k_H = 10k_V$$

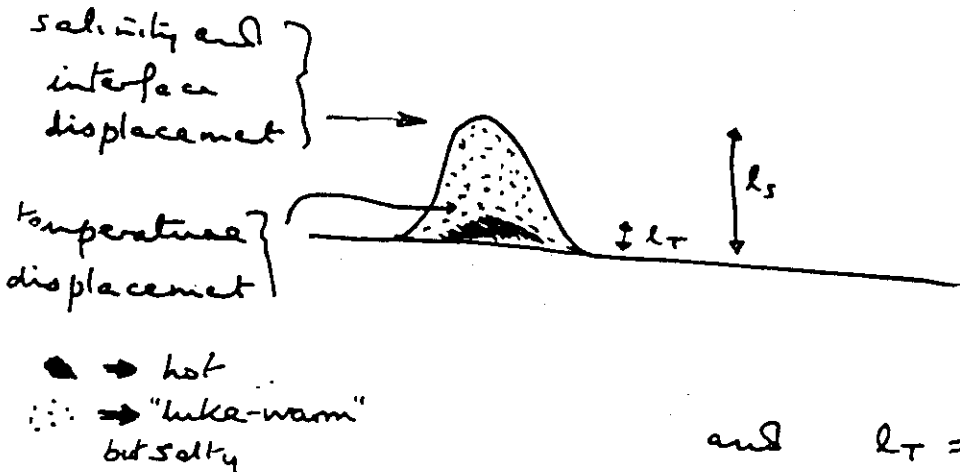
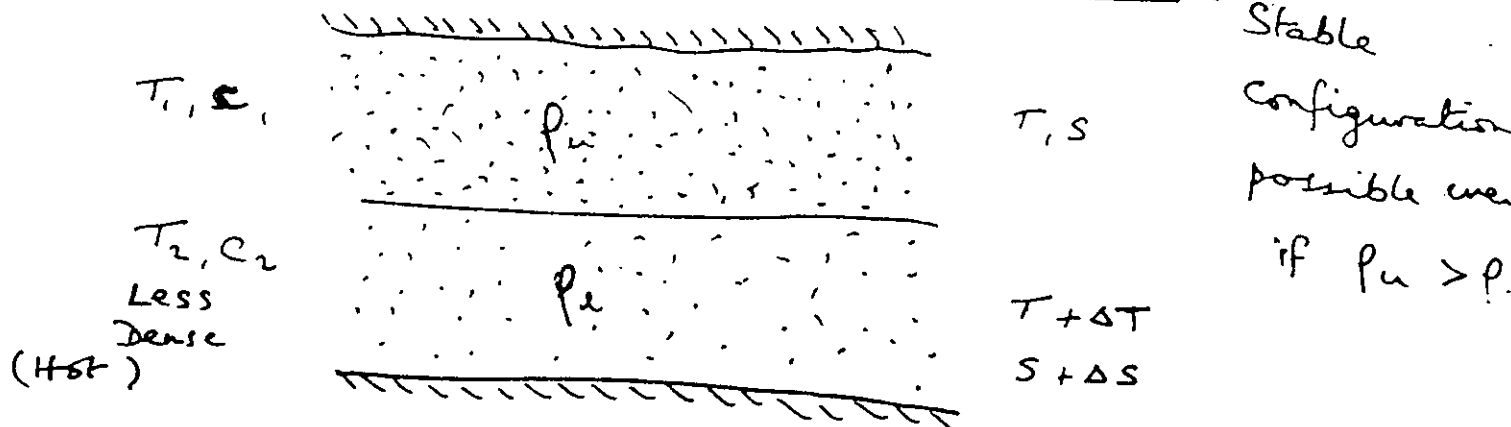
More permeable in vertical
 \rightarrow cells narrow, tall



$$k_H = 0.1k_V$$

Figure 5.3. Cell shapes for the most unstable disturbances for $k_H/k_V = 10$ and 0.1 . When $k_H = k_V$, the cells are square.

Note Hot may be salty + dense - (liquid saturated + in \exists in hot rock
or simply light, owing to thermal expansion



extra salinity in fluid displaced = $l_s \Delta S$

extra temperature = $l_t \Delta T$

and $l_t = l_s \frac{\rho C}{(\rho C)_{total}} \text{ fluid} \Rightarrow \phi l_s$

Hence density excess in displaced fluid = $l_s (\beta \Delta S - \alpha \phi \Delta T)$

Thus lower layer is stable if $\Delta S > (\phi \Delta T) / \beta$

However, density of lower layer fluid is $\beta \Delta S - \alpha \Delta T$ (relative)

and so it is possible that

← $\beta \Delta S - \alpha \Delta T \leq 0$ (apparently unstable, i.e. lower layer less dense)

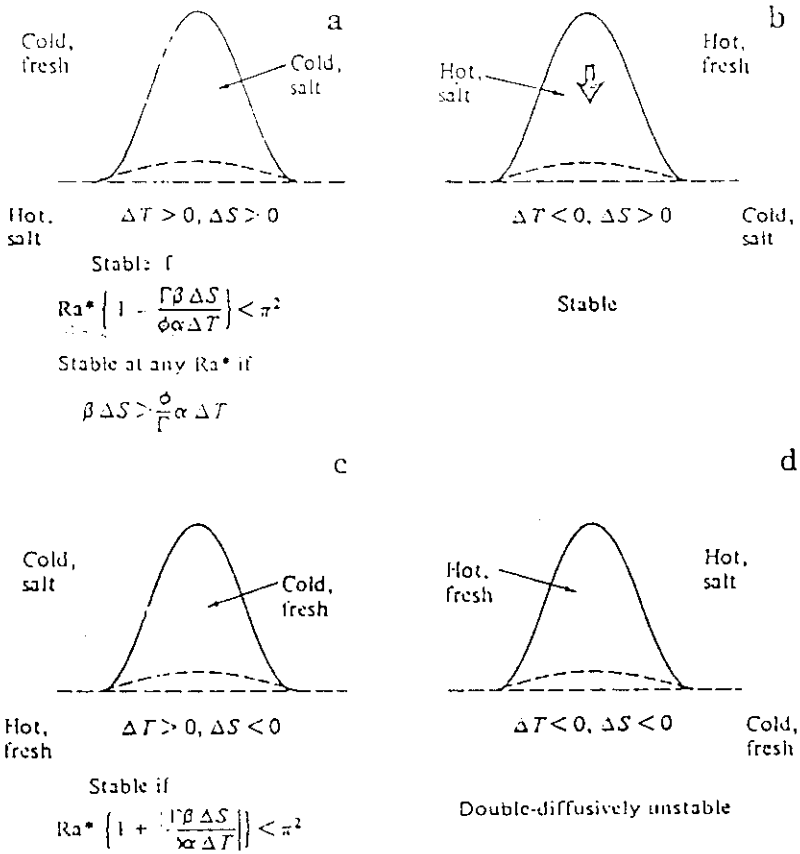
but $\beta \Delta S - \phi \alpha \Delta T \geq 0$ (stable to displacement due to thermal inertia)

so that the medium is stable.

(top diagram inverted)

However, in some situations, the medium can become unstable to double-advective (diffusive) instabilities

\Rightarrow fluid migrates throughout rock leading to variety of reactions



Stable if

$$Ra^* \left(1 - \frac{\beta \Delta S}{\phi \alpha \Delta T} \right) < \pi^2$$

Stable at any Ra^* if

$$\beta \Delta S > \frac{\phi}{\alpha} \Delta T$$

Stable if

$$Ra^* \left(1 + \frac{|\beta \Delta S|}{\alpha \Delta T} \right) < \pi^2$$

Figure 5.5. Isohaline (solid line) and isotherm (dashed line) displacements in diffusive saturated permeable media, with the associated stability characteristics when $\kappa \geq \kappa_c$. In case (c), when $\Delta T \rightarrow 0$, the stability condition given is replaced by one analogous to equation (5.1.15) but with $-\beta \Delta S / \phi \kappa_c$ replacing $\alpha \Delta T / \kappa$.

Salt acts to stabilize the thermal stabilization similar to R-D instability

T, S stabilize
 action if effective λ is large enough to overcome diffusion
 (buoyancy decreases)

Always stable
 T, S both stabilize

Salt fingers
 Buoyancy increases

For example: Dolomites { Permian Basin of West Tex.

Time Scale for Formation Rapid. Magnesium req'd to convert limestone to dolomite for double-diffusive / advective motion

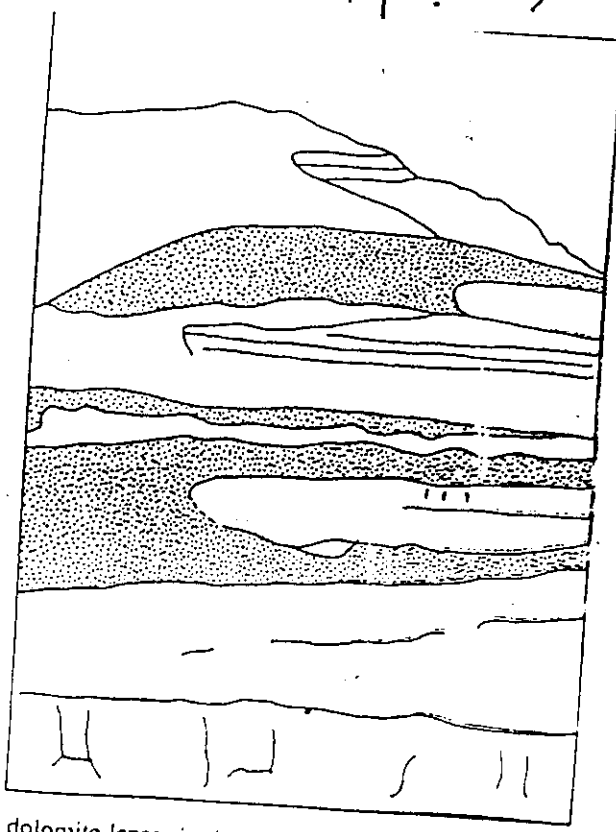
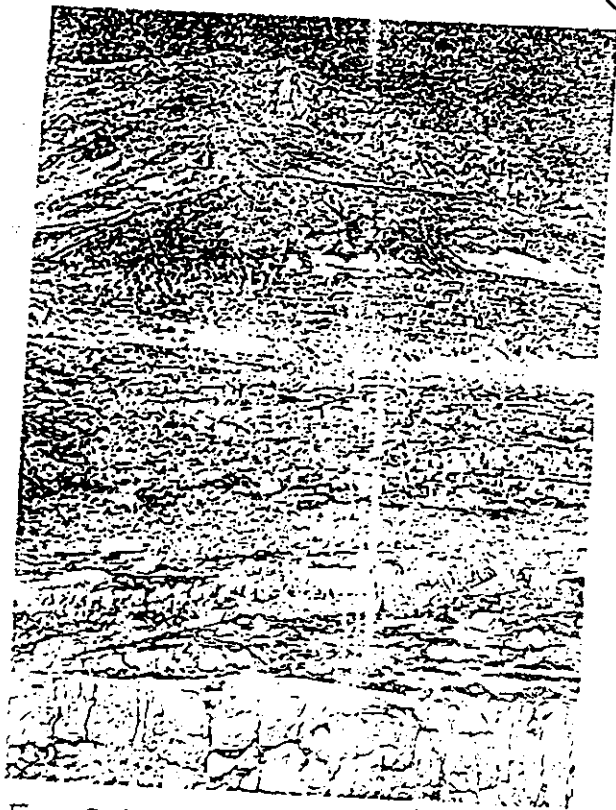
Velocity $w \sim \left(\frac{k_v}{\nu}\right)(g\beta\Delta s)$ c.f. $\left(u = -\frac{k\nabla}{\nu}\right)$ Darcy Law

⇒ Time for flow to migrate through a depth h is

$$T \sim \frac{h}{w} \sim \left(\frac{h\nu}{k_v g\beta\Delta s}\right)$$

if $h \sim 500 \text{ m}$; $\nu \sim 10^{-2} \text{ cm}^2/\text{s}$
 $k_v \sim 10^{-10}$ and $\beta\Delta s \sim 0.05$

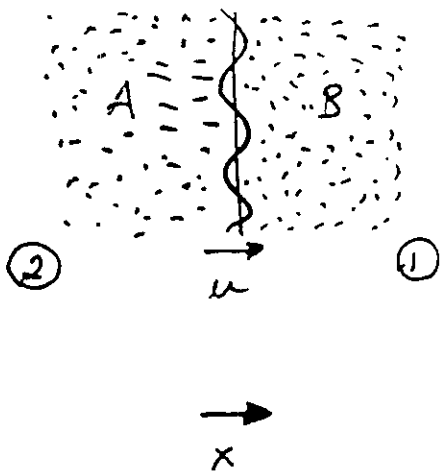
Then $\left\{ \begin{array}{l} T \sim 3000 \text{ yr} \\ w \sim 15 \text{ cm/yr} \end{array} \right\}$ Geologically Short



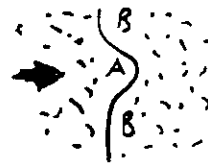
c.f. 10^6 for diff
 $D \sim 10^{-8}$
 $t \sim \frac{h^2}{D}$
 $\sim 10^5 \text{ yr}$

Figure 7.22. A field photograph by Wilson (1989) showing dolomite lenses in the upper, sparsely dolomitized zone (3 in. Figure 7.21). Dolomite is identified in the line drawing by stippling.

Saffman Taylor Instability



if B more viscous than A
then unstable



Finger of A is
less viscous than
surrounding fluid B
 \Rightarrow finger grows.

Suppose interface moves with speed u

\rightarrow Transport (Darcy) velocity is $\frac{\sigma}{\mu} u$.

$$\nabla p_1 = -\mu_1 \frac{\phi u}{k} \rightarrow P_1 = -\mu_1 \frac{\phi u}{k} x + P_0$$

$$\nabla p_2 = -\mu_2 \frac{\phi u}{k} \rightarrow P_2 = -\mu_2 \frac{\phi u}{k} x + P_0$$

Add a perturbation to interface, $u \rightarrow u + \sigma a e^{inx + \sigma t}$

$$\left\{ \begin{array}{l} P_1 \rightarrow P_1 + \mu_1 \frac{\phi}{k} \left(\frac{\sigma a}{k} \right) e^{inx - nx + \sigma t} \\ P_2 \rightarrow P_2 + (-)\mu_2 \frac{\phi}{k} \left(\frac{\sigma a}{k} \right) e^{inx + nx + \sigma t} \end{array} \right\} \begin{array}{l} \text{from } \nabla^2 p = 0 \\ \text{and } p \rightarrow 0 \\ \text{as } x \rightarrow \pm \end{array}$$

But on the moving interface $P_1 = P_2$

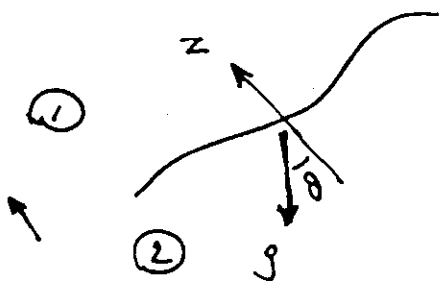
$$\text{hence } -\mu_1 \frac{\phi u}{k} + \mu_1 \frac{\phi \sigma a}{k k} = -\mu_2 \frac{\phi u}{k} - \mu_2 \frac{\phi \sigma a}{k k}$$

$$\Rightarrow (\mu_1 - \mu_2) \frac{\phi u}{k} = \sigma \left(\frac{\phi a}{k k} \right) (\mu_1 + \mu_2)$$

More viscous displaced by less vis.

If the two fluids are of different density, then a gravitational force exists and may act to stabilize the front

If interface has angle θ to the vertical



$$p_1 \rightarrow p_1 - \rho_1 g z \cos \theta$$

$$p_2 \rightarrow p_2 - \rho_2 g z \cos \theta$$

hence dispersion relation becomes

$$(\mu_1 - \mu_2) \frac{\phi u}{k} + (\rho_1 - \rho_2) g \cos \theta = \sigma \left(\frac{a \phi}{nk} \right) (\mu_1 + \mu_2)$$

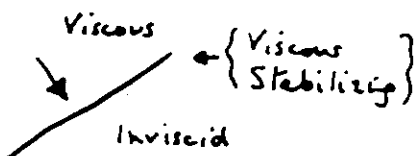
if $\sigma \geq 0$, $u(\mu_1 - \mu_2) > - \left(\frac{k g \cos \theta}{\phi} \right) (\rho_1 - \rho_2)$

Instability requires $u(\mu_1 - \mu_2)$ sufficiently

large if $\rho_1 < \rho_2$ (light
dense)

→ density stabilizes if flow is upward, and heavy fluid displaces light fluid

if $\rho_2 < \rho_1$ → always unstable for upflow



Also note downflow of heavy, viscous fluid into light, inviscid fluid may be stabilised by viscous forces

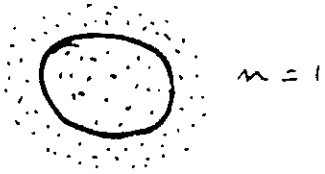
Case of circular geometry :



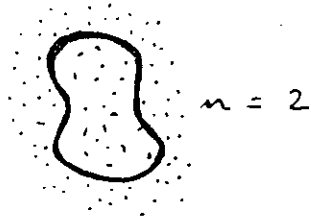
$$u = -\frac{k}{\mu} \nabla P$$

seek modes of the form

$$\eta = \eta_0(t) \exp(in\theta)$$



$n=1$



$n=2$



$n=3$

etc.

Now

$$\nabla^2 P = 0 \Rightarrow$$

\Rightarrow

$$P_{rr} + \frac{1}{r} P_r - \frac{n^2}{r^2} P = 0$$

for $P = P(r) \exp(in\theta)$

Matching velocity and stress at boundary gives

$$[\Delta P] = \sigma \left(\frac{1}{R} - \frac{\eta}{R^2} + \frac{d\eta}{R^2 dt} \right)$$

Surface tension

where the

$$\text{radius of interface} = R + \eta$$

hence

$$\left(\frac{1}{\eta} \right) \frac{d\eta}{dt}$$

growth rate

$$= \frac{Qn}{2\pi R^2} \left(\frac{M_1 - M_2}{M_1 + M_2} \right) - \frac{Q}{2\pi R^2}$$

viscous inst.

$$- \frac{\sigma n(n^2 - 1)}{R^3} \left(\frac{M_1 M_2}{M_1 + M_2} \right)$$

surface tension st.

$$\left(\text{with } M_i = \frac{k}{\mu_i} \right)$$

At each R , most unstable mode is selected $\Rightarrow n(R)$

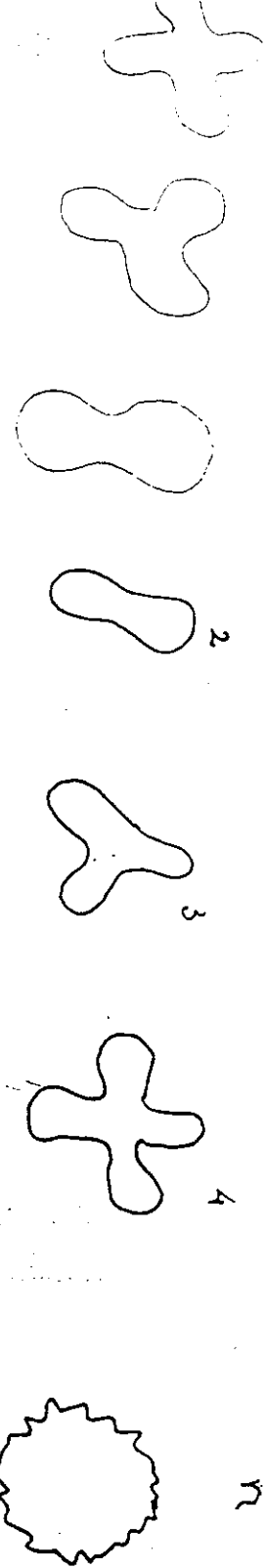
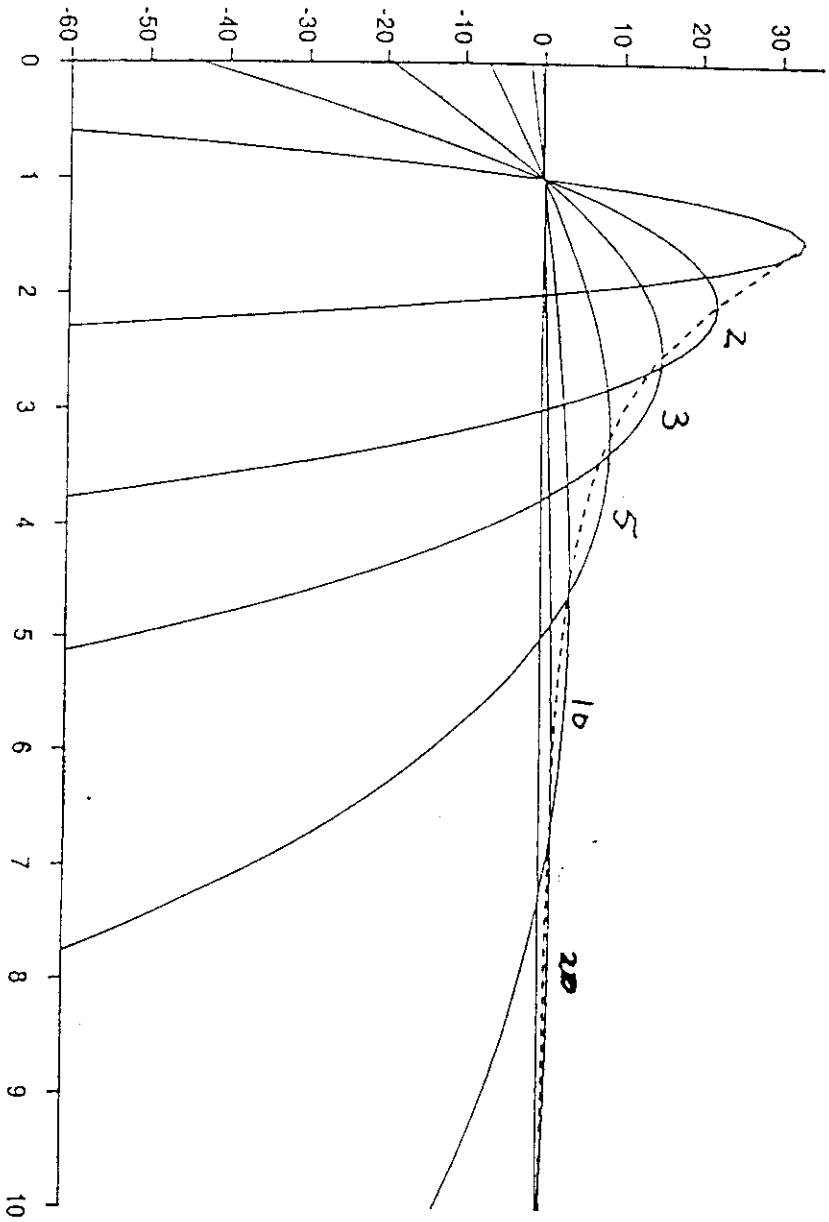
for $R < R_c = \frac{4\sigma M_2 \pi}{Q}$

interface absolutely

stable
($M_1 \gg M_2$)

$\frac{A'}{A}$

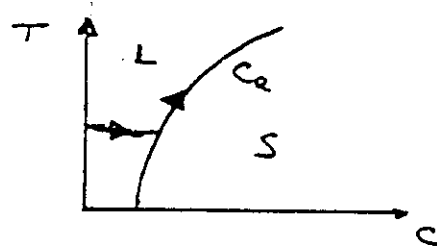
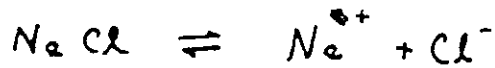
$$\frac{R}{R_0} = 1$$



Reactions: Rock alteration due to fluid flow

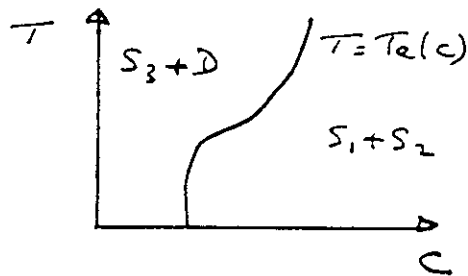
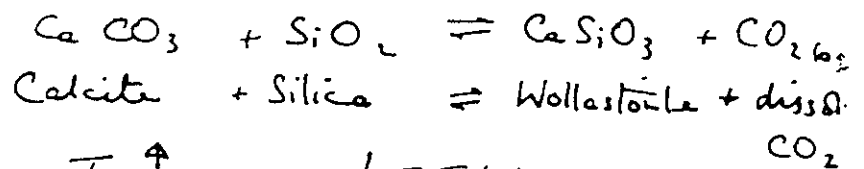
Reaction types

(i) Simple Dissolution / Precipitation

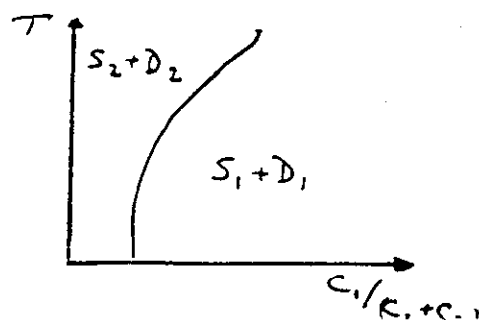
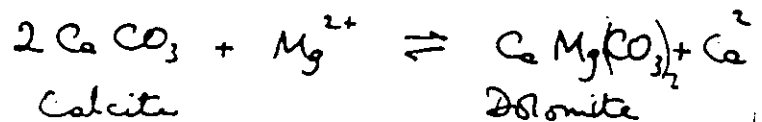


(ii) Dissolution of 2 minerals into a third + a dissolved species

⇒ Wollastonite



(iii) Replacement Reactions Dolomite



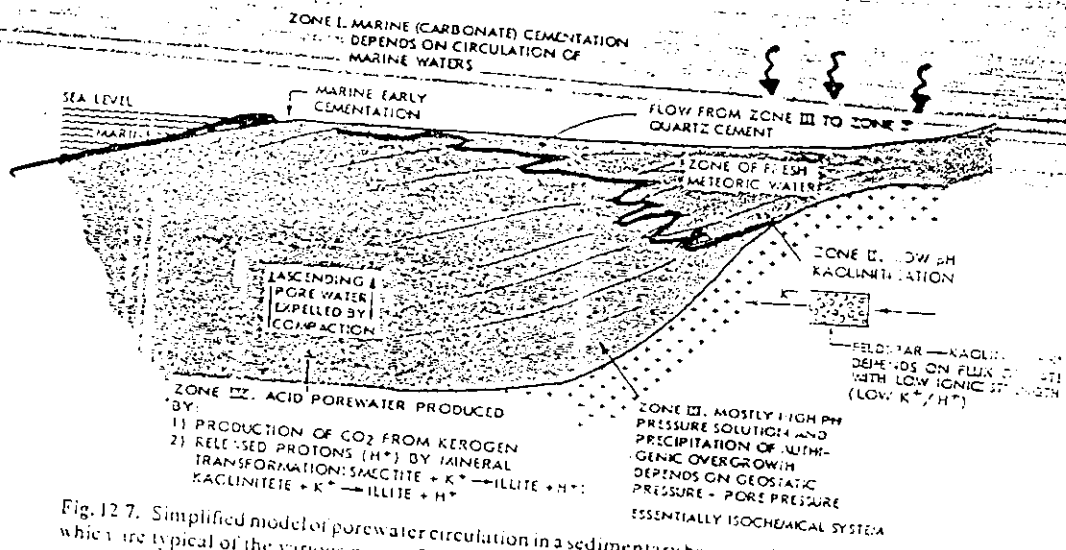


Fig. 12.7. Simplified model of porewater circulation in a sedimentary basin, and the diagenetic reactions which are typical of the various parts of the basin. (Björlykke 1983)

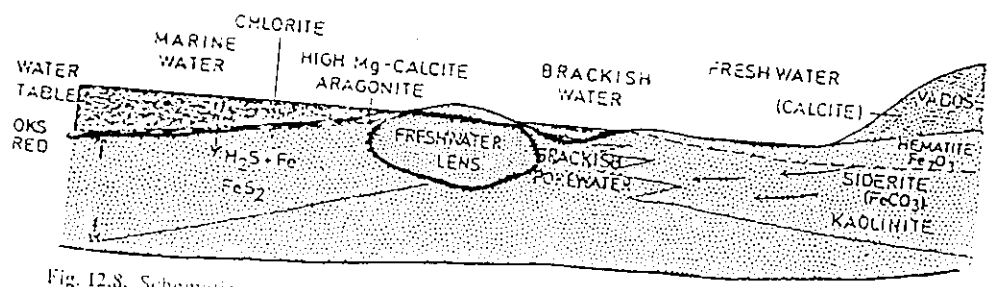


Fig. 12.8. Schematic overview of early diagenetic environment. In the sulphate-reducing zone all iron will precipitate as sulphides. Siderite will only form in the fresh or brackish water environment or below the sulphate-reducing zone

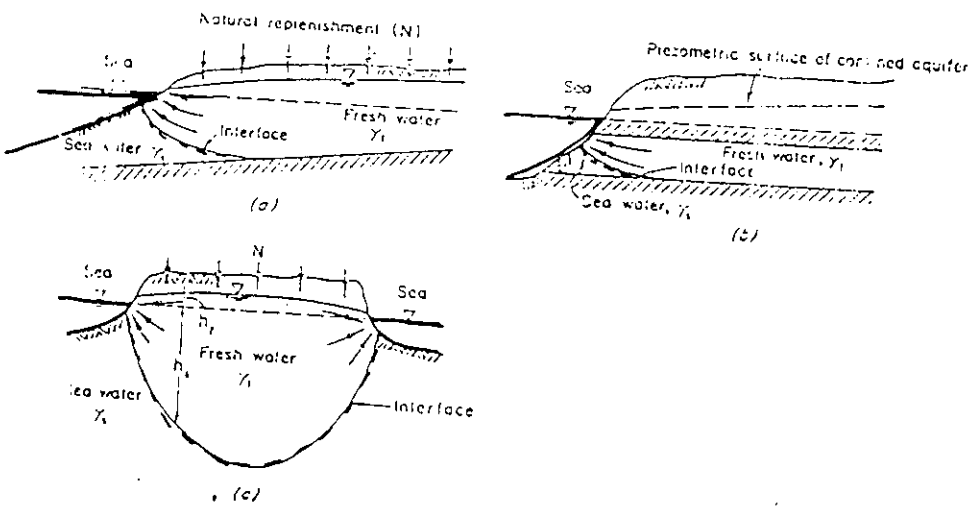
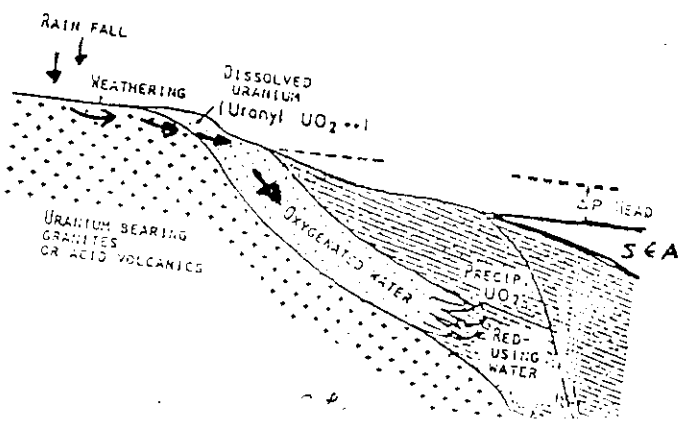
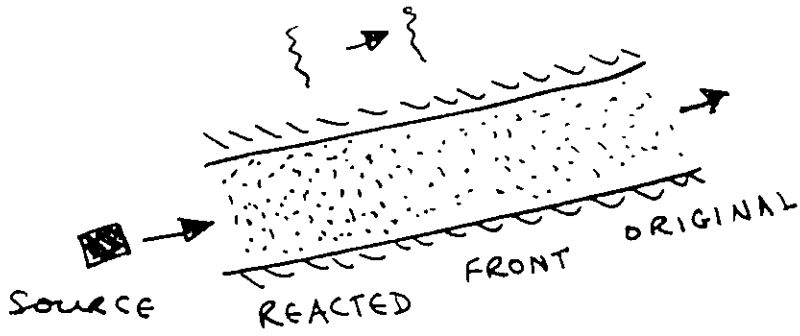


FIG. 9.7.1. Interfaces in coastal aquifers (highly distorted figures).



Types of Reaction driven by Fluid Flow

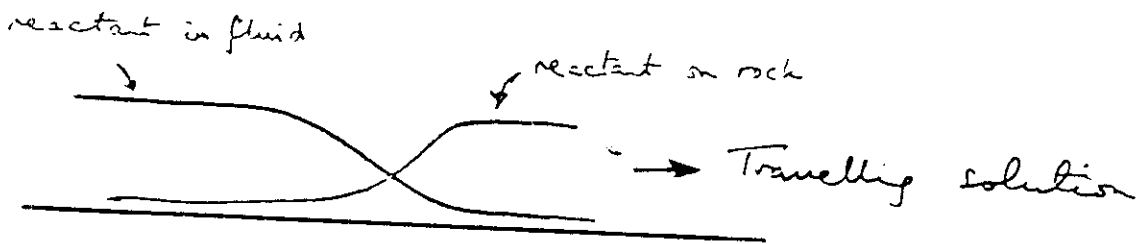


b = reactant on rock
 a = reactant in fluid

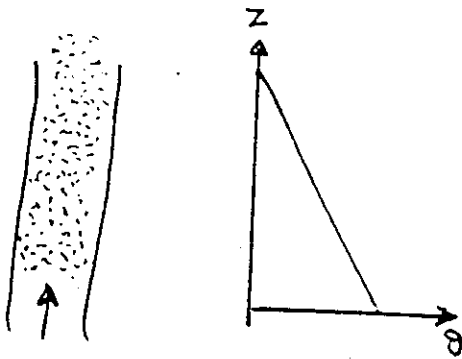
$$\begin{cases} \lambda a_t + u a_x = -a \\ b_t = -a \end{cases}$$

Migrating reaction front

$$(1-a)b \sim \frac{1}{1 - \exp[-kx]}$$



GRADIENT REACTION ZONE



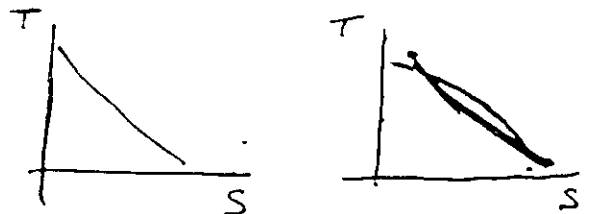
Geothermal Gradient

Fluid ascends and remains in local equilibrium

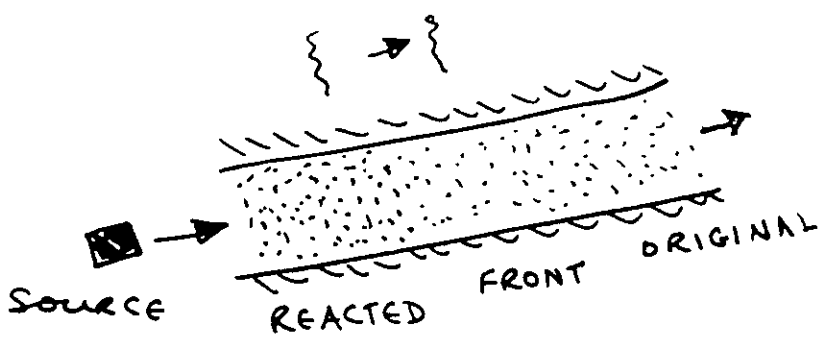
⇒ phase changes occur as consequence of background temperature variation.

→ Flows may be thermally or hydraulically driven.

→ MIXING ZONES



Types of Reaction driven by Fluid Flow

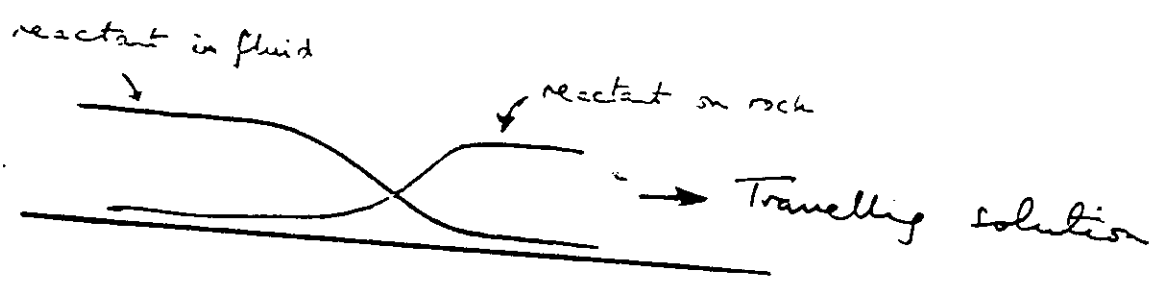


b = reactant on rock
 a = reactant in fluid

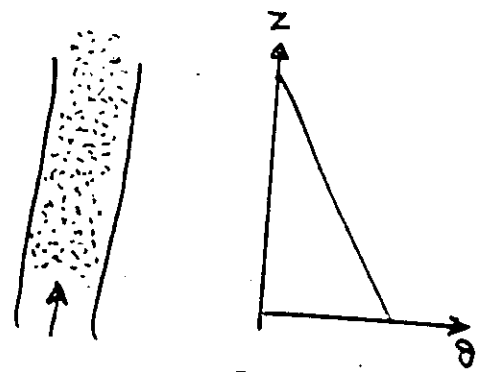
$$\begin{cases} \lambda a_x + u a_x = -a_t \\ b_t = -a_t \end{cases}$$

Migrating reaction front

$$(1-a)b \sim \frac{1}{1 - \exp(-kx)}$$



GRADIENT REACTION ZONE



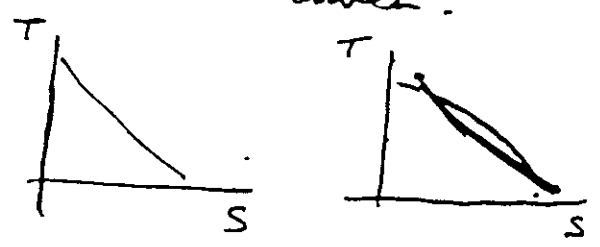
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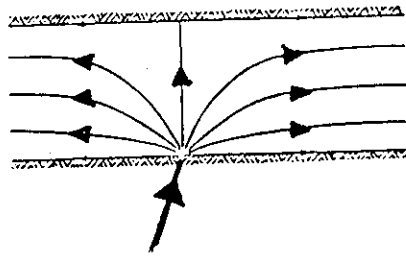


Figure 4.6. Streamlines of flow issuing from a linear fracture into a permeable layer with no superimposed ambient pressure gradient.

permeable
impermeable
permeable

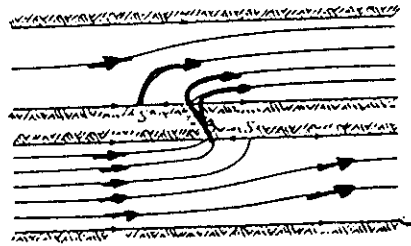
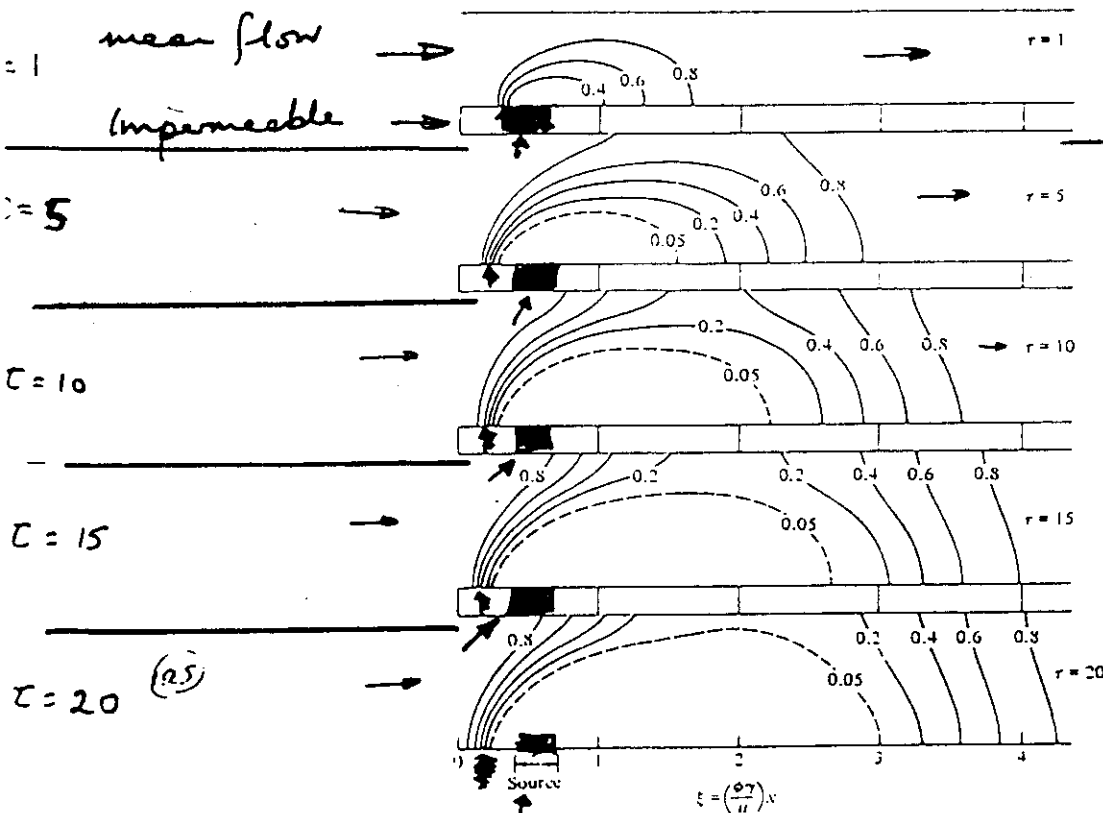


Figure 4.7. Streamlines of flow when fluid moves through a linear fracture from one permeable layer to another.



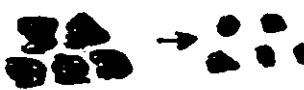
Development
of
Reaction
Zone
following
fracture
thro'
impermeable
layer

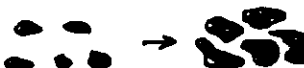
Figure 4.13. Distributions of the unaltered mineral undergoing reaction, $S = s/s_0$. The regions inside the dashed curves indicate zones of virtually complete reaction.

hc

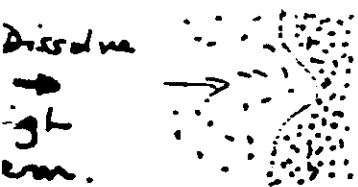
Reaction Front Instabilities

Reaction of interstitial fluid and porous matrix
can change the permeability.

Dissolution \Rightarrow Decrease of matrix blocking flow
 \Rightarrow Increased Permeability

Precipitation \Rightarrow Increase of ~~matrix~~ material blocking
flow between porous matrix
 \Rightarrow Decreased Permeability

As fluid propagates through rock, we expect
the dissolution process to generate fingers of
high permeability \Rightarrow planar interface unstable



finger offers less resistance to flow
 \rightarrow fluid tends to flow into finger
 \rightarrow unstable (c.f. viscous fingering)