



*The Abdus Salam
International Centre for Theoretical Physics*



1867-34

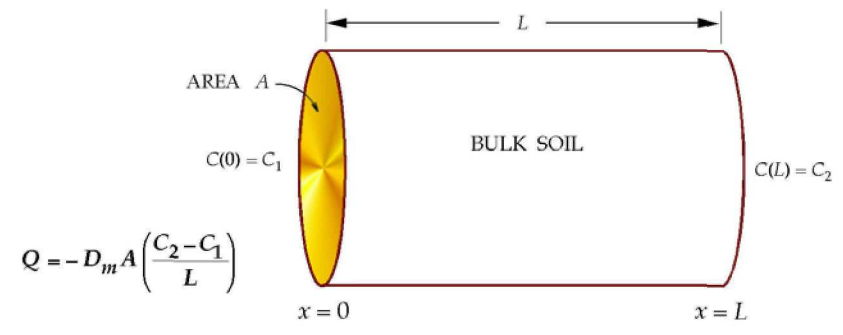
College of Soil Physics

22 October - 9 November, 2007

Transient temperature and water vapor in soils

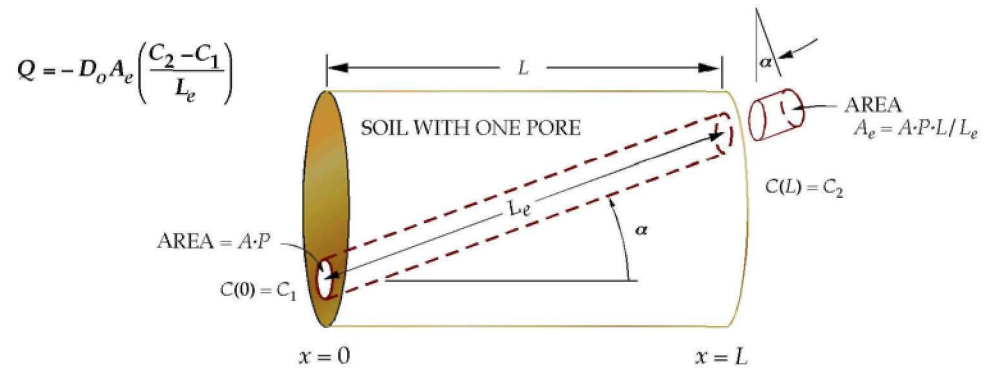
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Davies
USA*

ISOTHERMAL GASEOUS DIFFUSION



$$Q = -D_m A \left(\frac{C_2 - C_1}{L} \right)$$

$$D_m = D_o P \left(\frac{L}{L_e} \right)^2$$



$$Q = -D_o A_e \left(\frac{C_2 - C_1}{L_e} \right)$$

$$D_m = D_o P (L/L_e)^2 \quad (\text{Buckingham, 1904})$$

D_m = DIFFUSION COEFFICIENT
IN BULK SOIL

D_o = DIFFUSION COEFFICIENT
IN AIR

P = AIR-FILLED POROSITY

$$(L/L_e)^2 = \text{TORTUOSITY} \\ = \cos^2 a \approx 0.66 \quad (\text{Penman, 1940})$$

Marshall (1958) used $P(L/L_e)^2 = P^{3/2}$

M & Q (1959) used $P(L/L_e)^2 = P^{4/3}$

A GASEOUS OXYGEN DIFFUSION EXAMPLE

$$q = -D_m \frac{dc}{dx} \quad D_m = 6000 \text{ cm}^2 \cdot \text{day}^{-1}$$



$$q = -6000 \frac{(0 - 2.8 \cdot 10^{-4})}{100 - 0} = 0.0168 \text{ gm} \cdot (\text{cm}^2 \cdot \text{day})^{-1}$$

Note that the oxygen in 60 cm^3 of air can diffuse daily through a 1-cm^2 soil column having a length of 1 m

ISOTHERMAL WATER VAPOR DIFFUSION IN SOILS

$$q_v = -D_m \frac{d\rho_v}{dx}$$

Let us assume that $p_v V = nRT$

$$q_v = -D_m \frac{1}{R_v T} \frac{dp_v}{dx}$$

Know that

$$p_v = p_0 \exp\left(\frac{\phi_{tw}}{RT}\right)$$

p_0 = saturated water vapor pressure

ϕ_{tw} = matric potential + osmotic potential

Neglecting gravity

$$q_v = -D_m \frac{p_v}{(RT)^2} \frac{d\phi_{tw}}{dx}$$

**A robust equation for water vapor diffusion in soil
ignoring temperature gradients.**

$$q_v = -D_m \frac{p_v}{(RT)^2} \frac{d\phi_w}{dx}$$

WHEN SOIL WATER CONTENT IS NOT VERY DRY
(ABOVE PERMANENT WILTING PERCENTAGE)

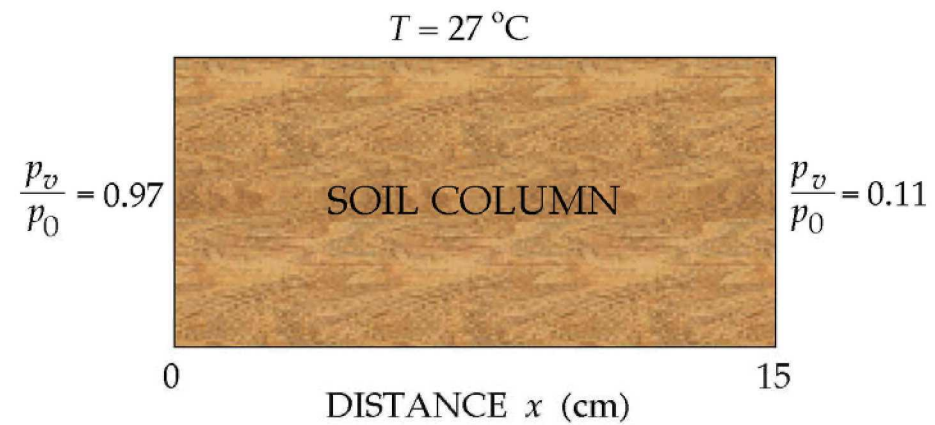
WATER VAPOR PRESSURE $p_v \rightarrow p_0$

RELATIVE HUMIDITY OF SOIL AIR $\rightarrow 1$

THE AMOUNT OF VAPOR THAT DIFFUSES
UNDER ISOTHERMAL CONDITIONS IS VERY SMALL!

WATER VAPOR MOVES MORE READILY THROUGH SOILS
1. IN PRESENCE OF LARGE VAPOR PRESSURE GRADIENTS
2. UNDER A THERMAL GRADIENT IN MOIST SOILS

STEADY STATE WATER VAPOR DIFFUSION



GOAL: MEASURE VAPOR DIFFUSION COEFFICIENT

LIQUID

$$q_{\theta} = -K \frac{dh}{dx}$$

$$q_{\theta} = -K \frac{dh}{d\theta} \frac{d\theta}{dx} = -D_{\theta} \frac{d\theta}{dx}$$

VAPOR

$$q_v = -D_m \frac{d\rho_v}{dx}$$

$$q_v = -D_v \frac{\partial \rho_v}{\partial \theta} \frac{d\theta}{dx}$$

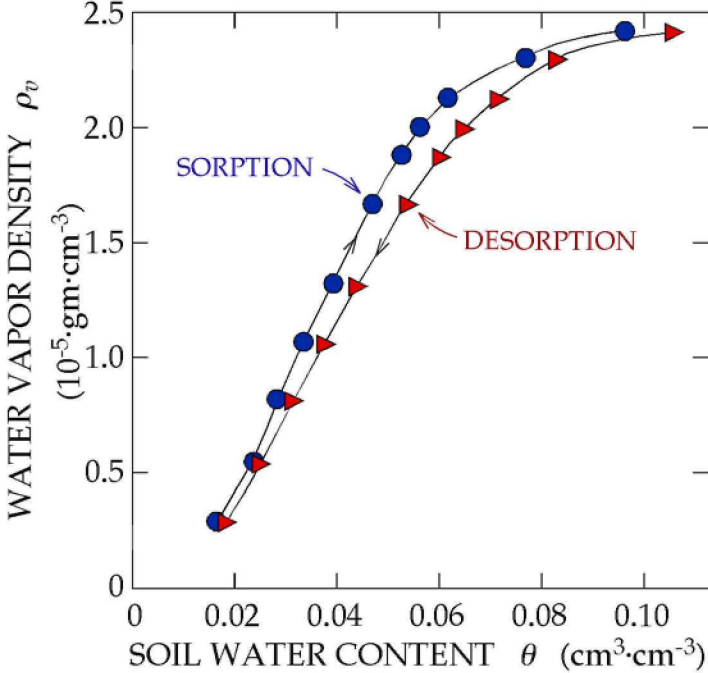
LIQUID & VAPOR

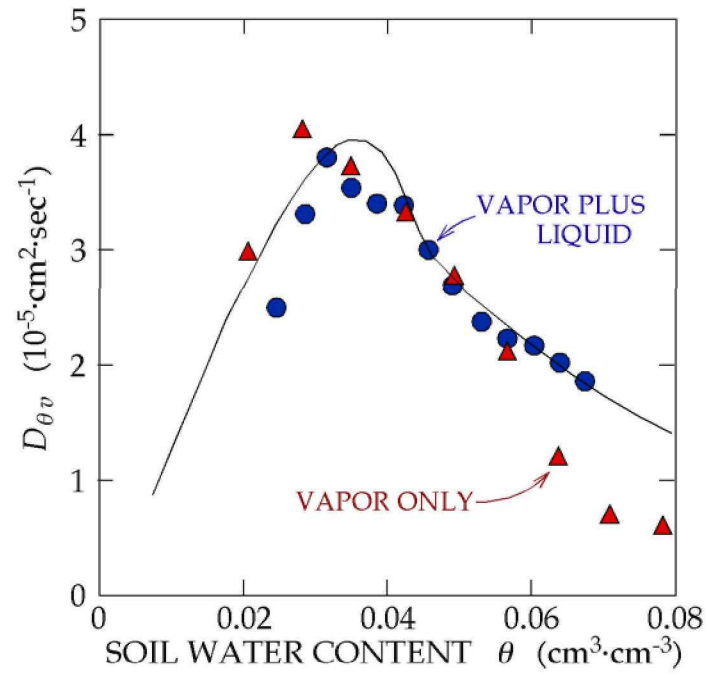
$$q = q_{\theta} + q_v = -\left(D_{\theta} + D_v \frac{d\rho_v}{d\theta}\right) \frac{d\theta}{dx}$$

$$q = -D_{\theta v} \frac{d\theta}{dx}$$

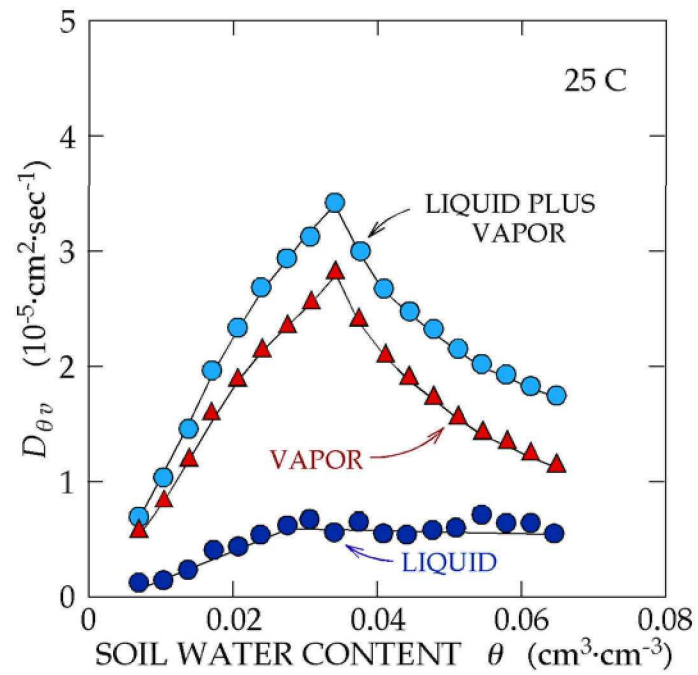
where $D_{\theta v} = D_{\theta} + D_v \frac{d\rho_v}{d\theta}$

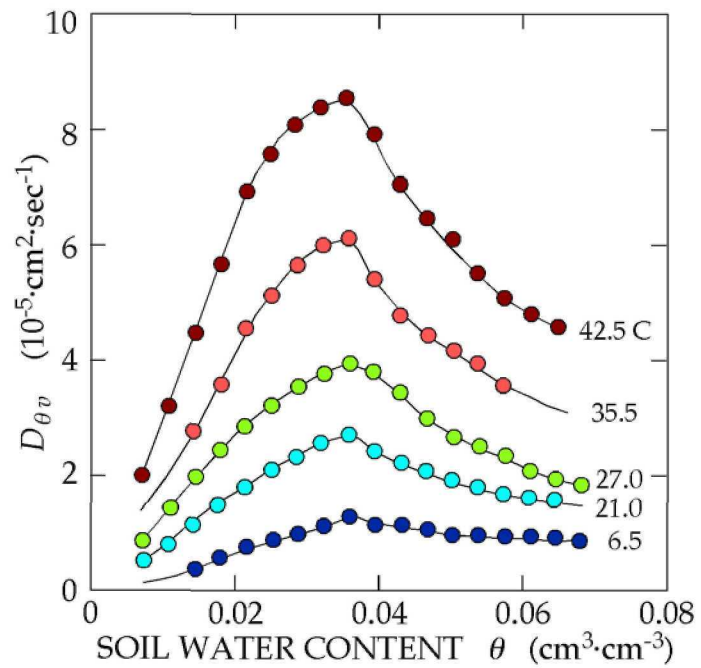
ISOTHERMS



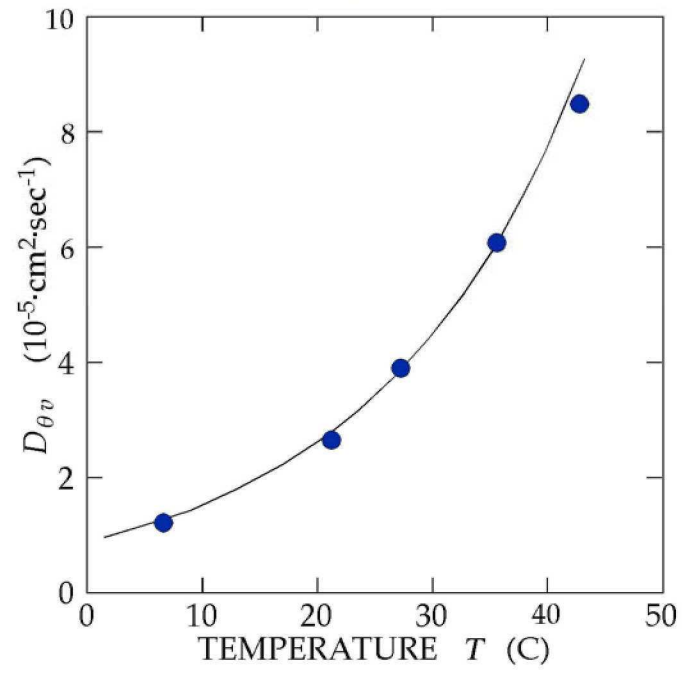


$$\begin{aligned}
 D_v &= D_0(L/L_e)^2(P-\theta) \\
 &= 0.0818 - 0.176\theta
 \end{aligned}$$





$\theta = 0.035 \text{ cm}^3 \cdot \text{cm}^{-3}$



$$\frac{\partial \rho_v}{\partial t} = \frac{\partial}{\partial x} \left(\frac{D_v}{\varepsilon} \frac{\partial \rho_v}{\partial x} \right) - \frac{1}{\varepsilon} \frac{\partial \theta}{\partial t}$$

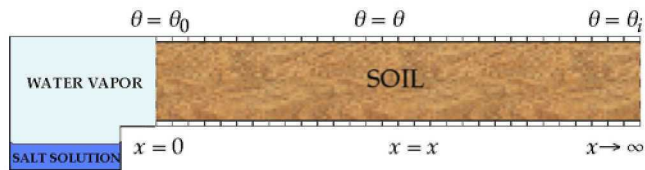
Assume $\frac{\partial \theta}{\partial t} \gg \varepsilon \frac{\partial \rho_v}{\partial t}$

$$\frac{\partial \theta}{\partial t} = \frac{\partial}{\partial x} \left(D_v \frac{\partial \rho_v}{\partial x} \right) \quad \frac{\partial \theta}{\partial t} = \frac{\partial}{\partial x} \left(D_v \frac{\partial \rho_v}{\partial \theta} \frac{\partial \theta}{\partial x} \right)$$

$$\frac{\partial \theta}{\partial t} = \frac{\partial}{\partial x} \left(D_{vap} \frac{\partial \theta}{\partial x} \right) \quad q_\theta = -K \frac{dh}{d\theta} \frac{d\theta}{dx} = -D_\theta \frac{d\theta}{dx}$$

$$\frac{\partial \theta}{\partial t} = \frac{\partial}{\partial x} \left([D_\theta + D_{vap}] \frac{\partial \theta}{\partial x} \right)$$

$$\frac{\partial \theta}{\partial t} = \frac{\partial}{\partial x} \left(D_{\theta v} \frac{\partial \theta}{\partial x} \right)$$



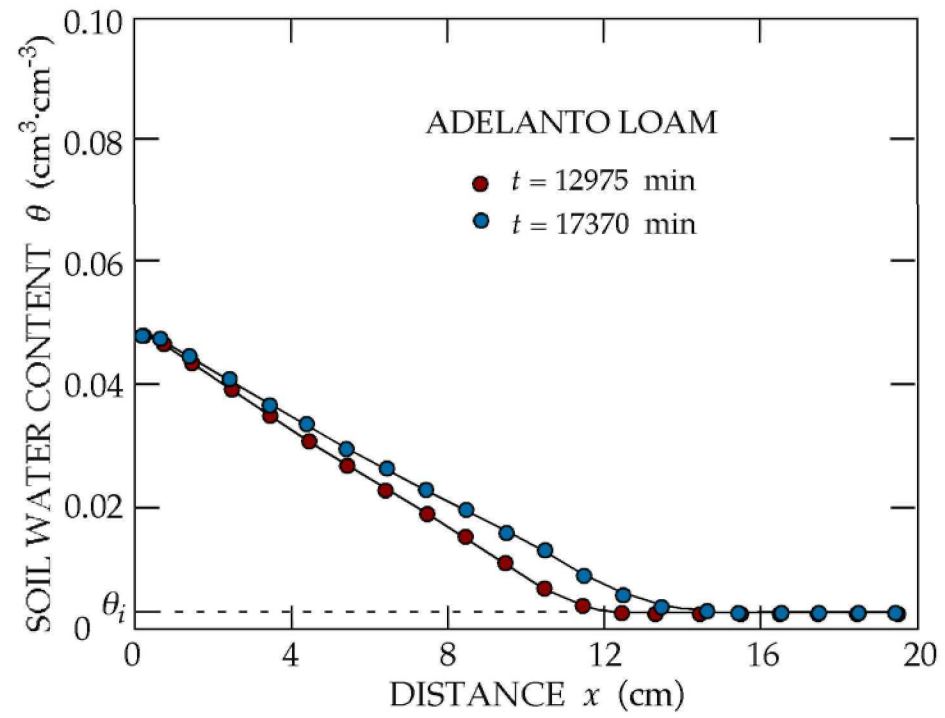
$$\frac{\partial \theta}{\partial t} = \frac{\partial}{\partial x} \left(D_{\theta v} \frac{\partial \theta}{\partial x} \right)$$

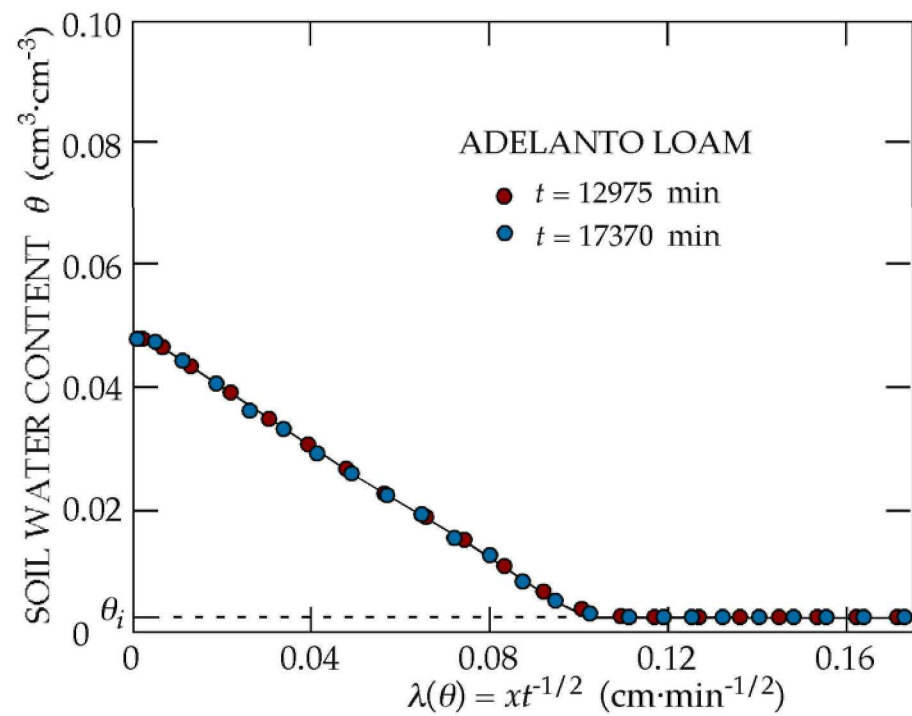
$$\theta = \theta_i \quad x > 0 \quad t = 0$$

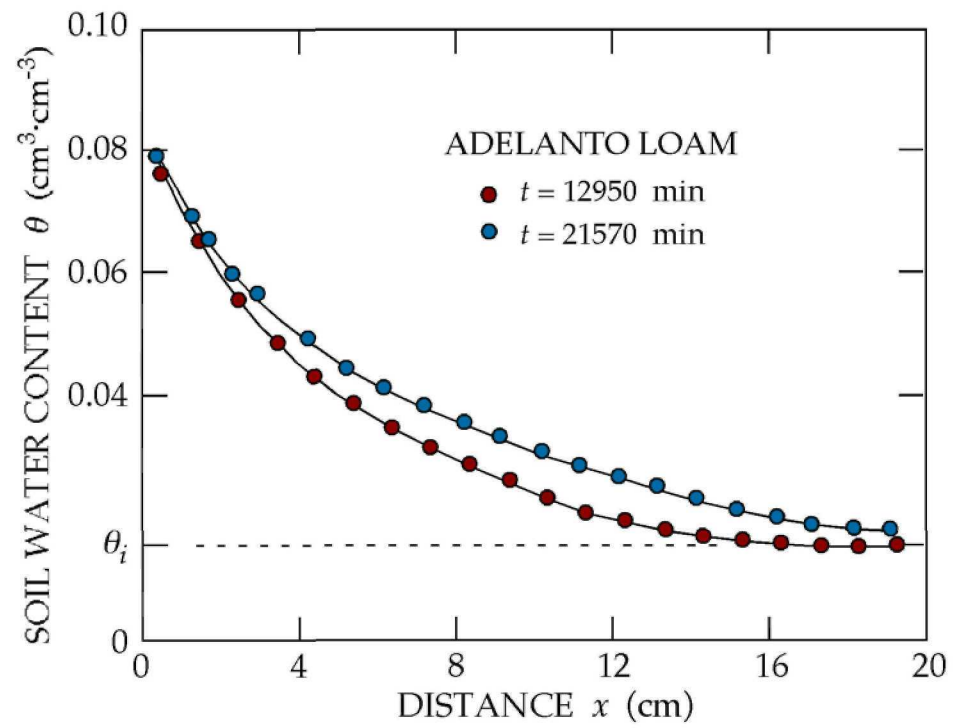
$$\theta = \theta_0 \quad x = 0 \quad t > 0$$

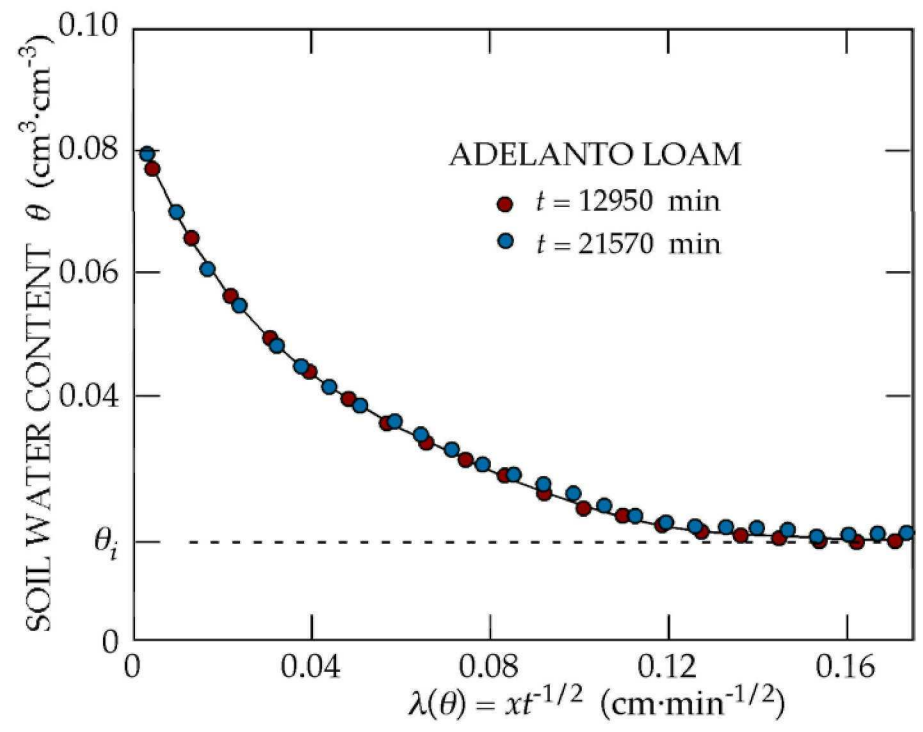
$$\lambda(\theta) = xt^{-1/2}$$

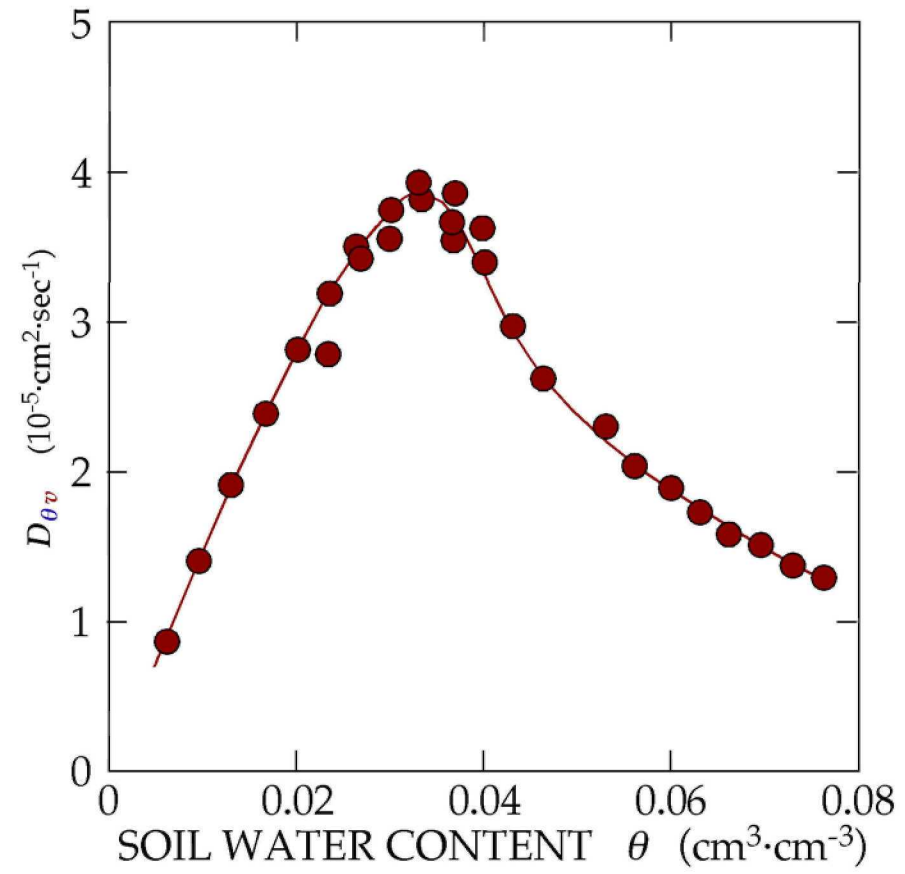
$$D_{\theta v}[\theta(x, t_1)] = -\frac{1}{2t_1} \frac{dx}{d\theta} \int_{\theta_i}^{\theta} x d\theta$$

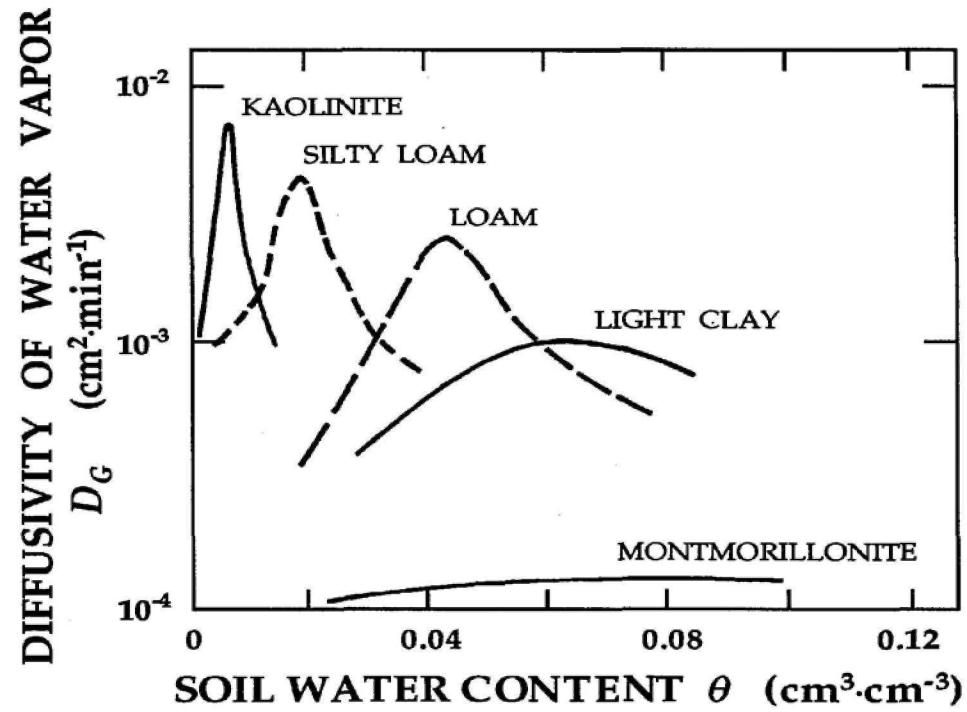












**VALUES $D_{\theta v}$ COULD NOT BE SEPARATED INTO
THEIR INDIVIDUAL COMPONENTS OF
VAPOR AND LIQUID TRANSPORT**

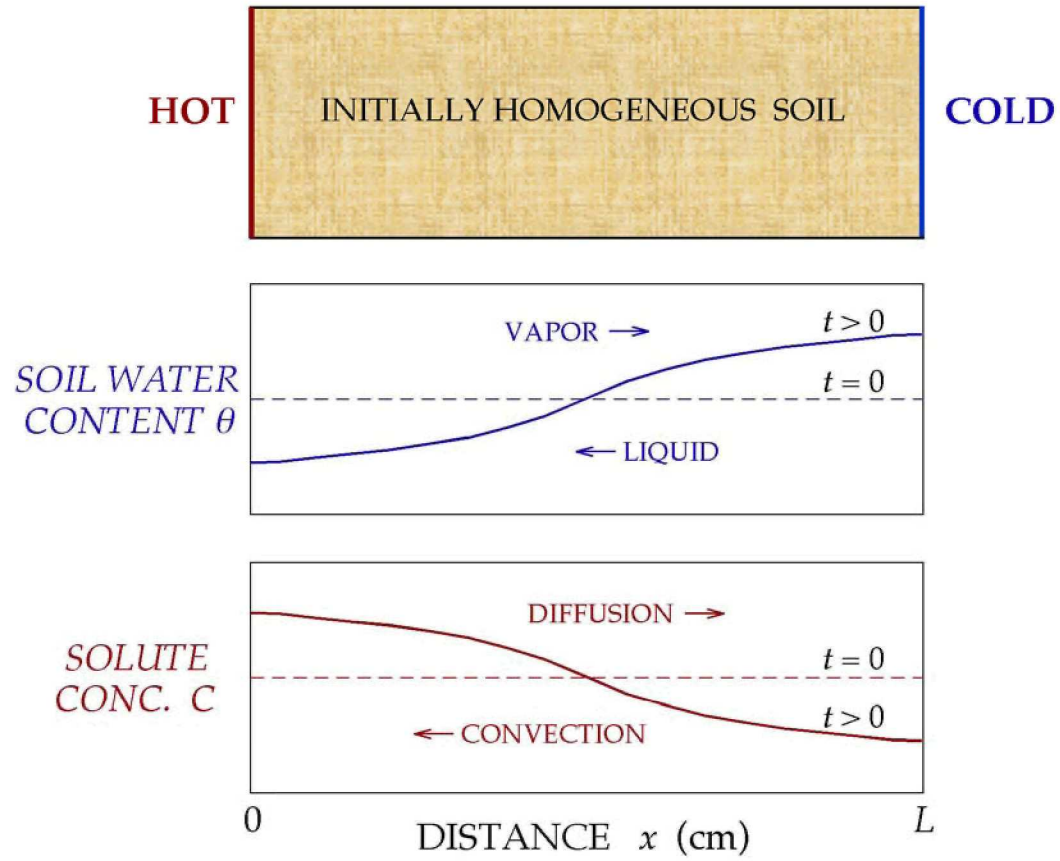
WE LEARNED THAT

**IN DRY SOILS UNDER ISOTHERMAL CONDITIONS,
LARGE GRADIENTS OF VAPOR DENSITY CAUSE
SMALL CHANGES IN LIQUID WATER CONTENT
OVER SEVERAL DAYS AND WEEKS**

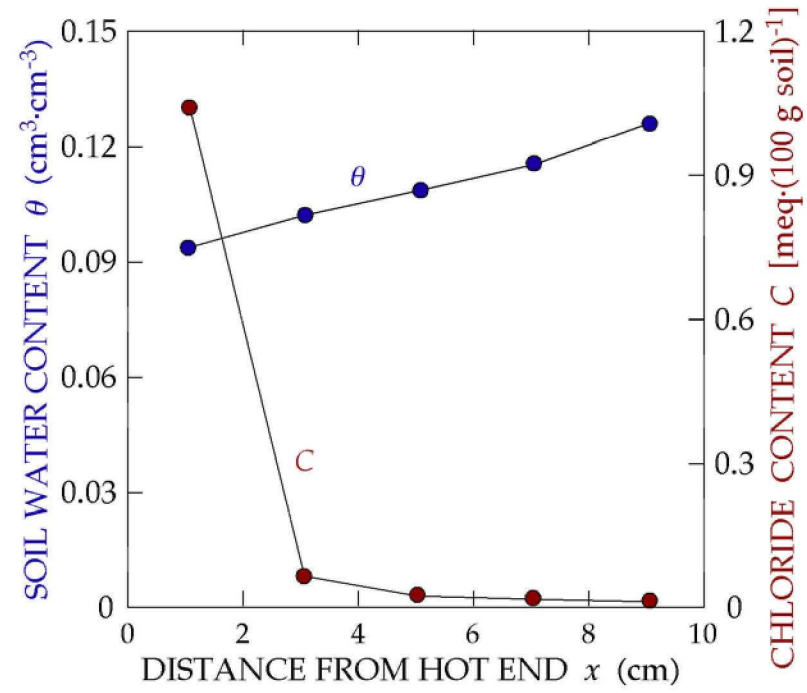
**BUT, AS THE LIQUID SOIL WATER CONTENT INCREASES
VAPOR DENSITY GRADIENTS DRASTICALLY DECREASE
WITH MOST WATER MOVING AS A LIQUID**



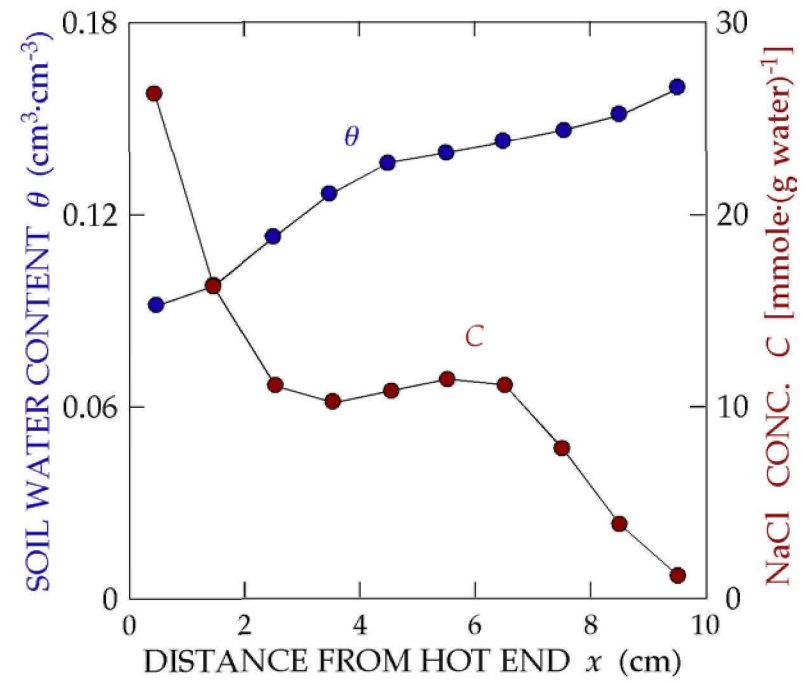
NONISOTHERMAL TRANSPORT PROCESSES



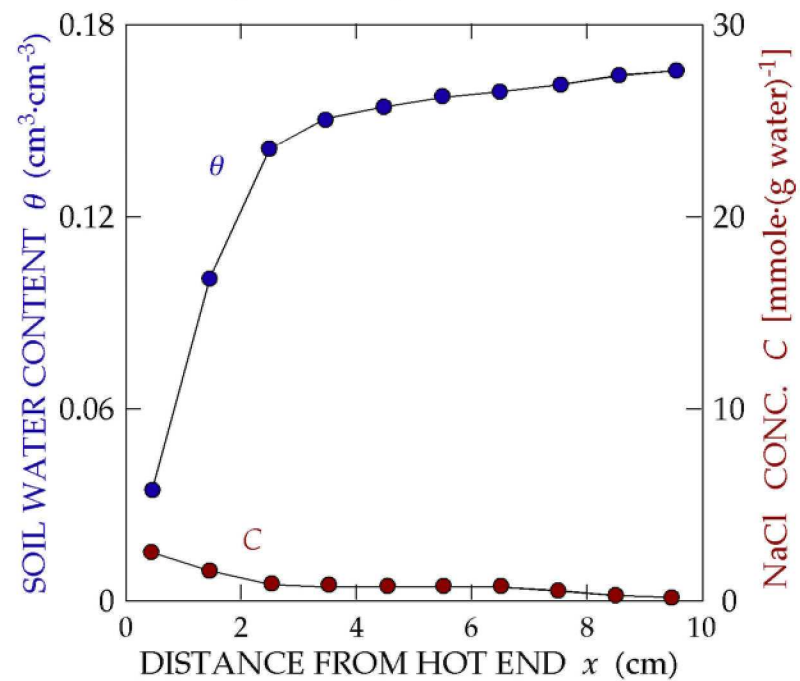
GURR, MARSHALL & HUTTON (1952)



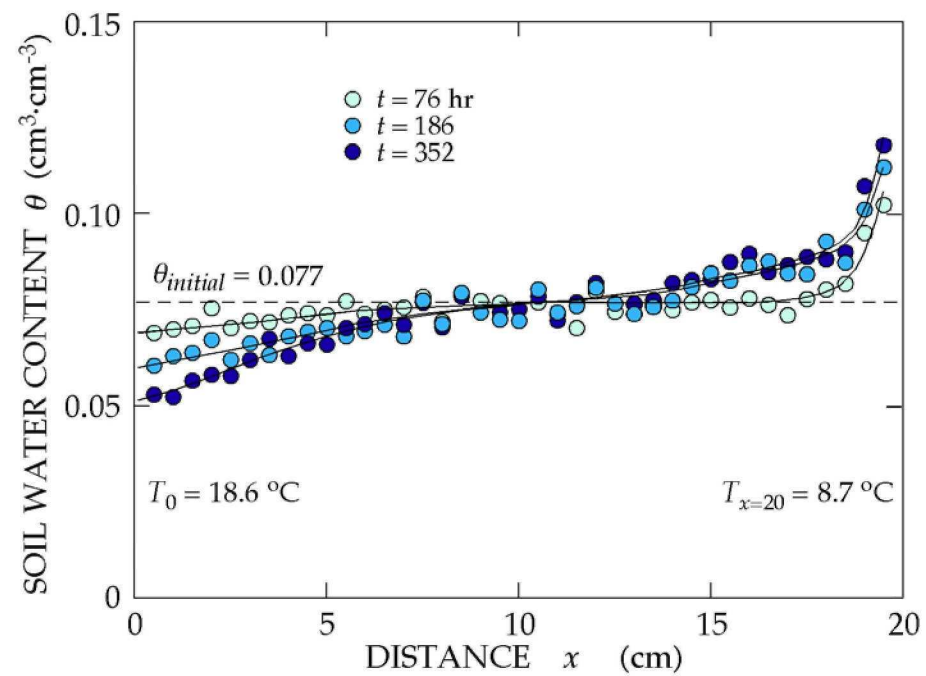
JACKSON, et al. (1965)



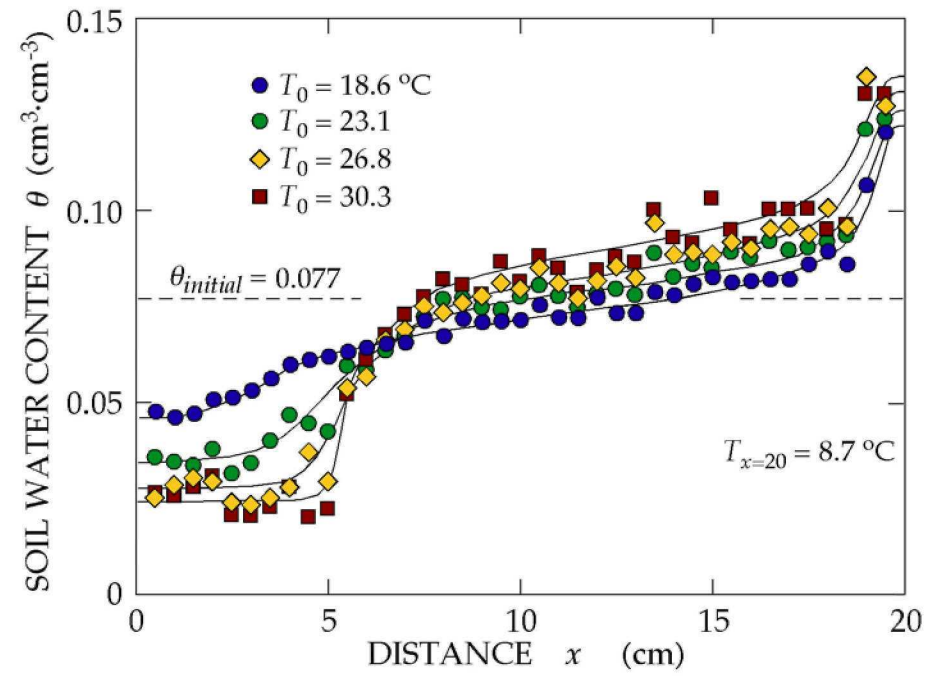
JACKSON, et al. (1965)



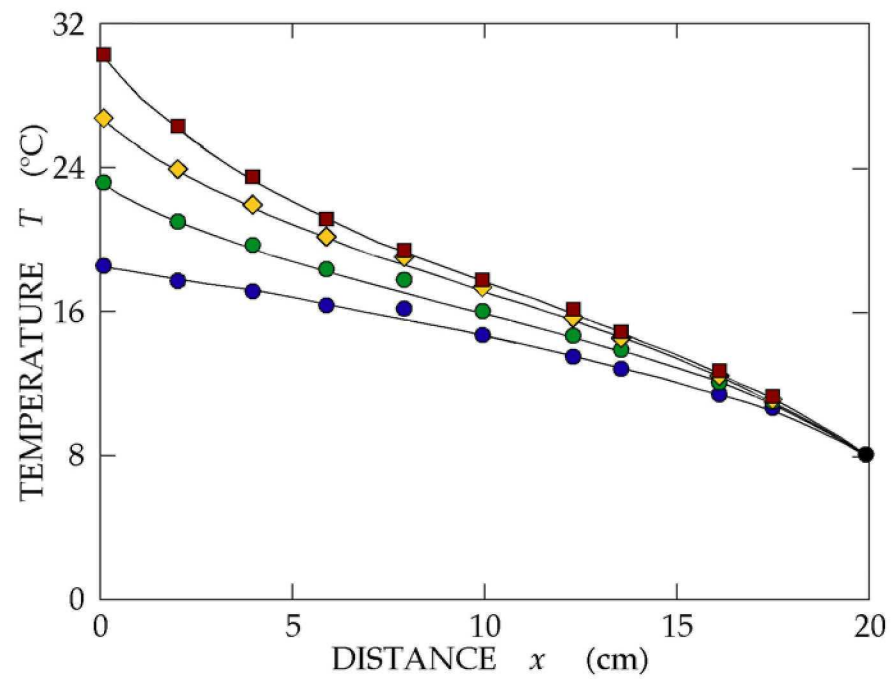
CASSEL (1968)

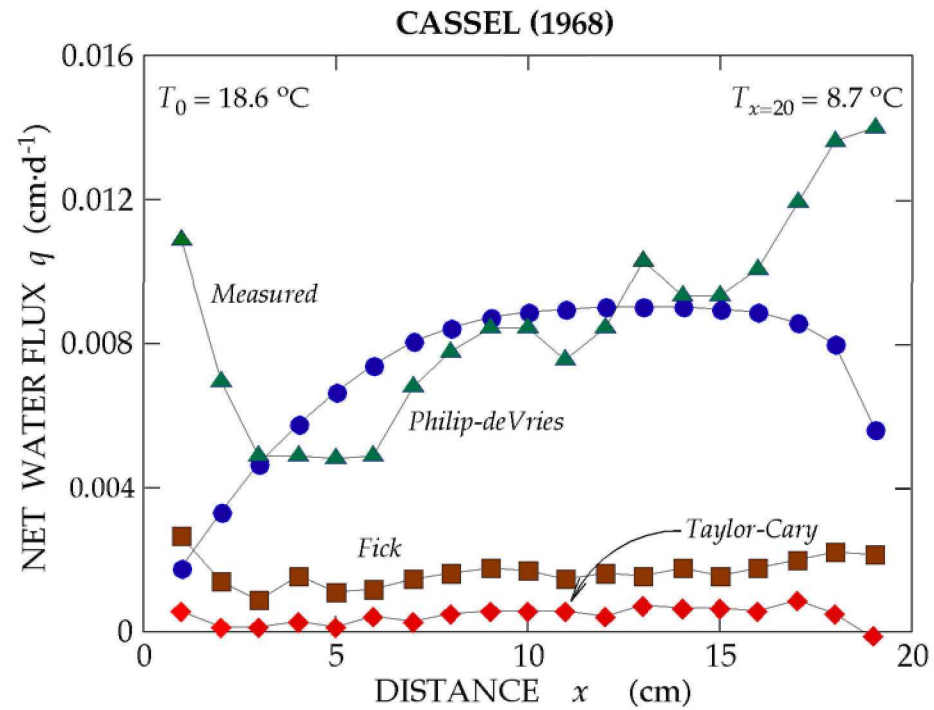


CASSEL (1968)



CASSEL (1968)





MORE RECENT, SIGNIFICANT RESEARCH ON SOIL WATER DIFFUSION WAS PUBLISHED BY NASSAR AND HORTON (1989).

**DAILY FLUCTUATIONS OF WATER AND
SOLUTES IN A TOPSOIL WITHOUT VEGETATION**

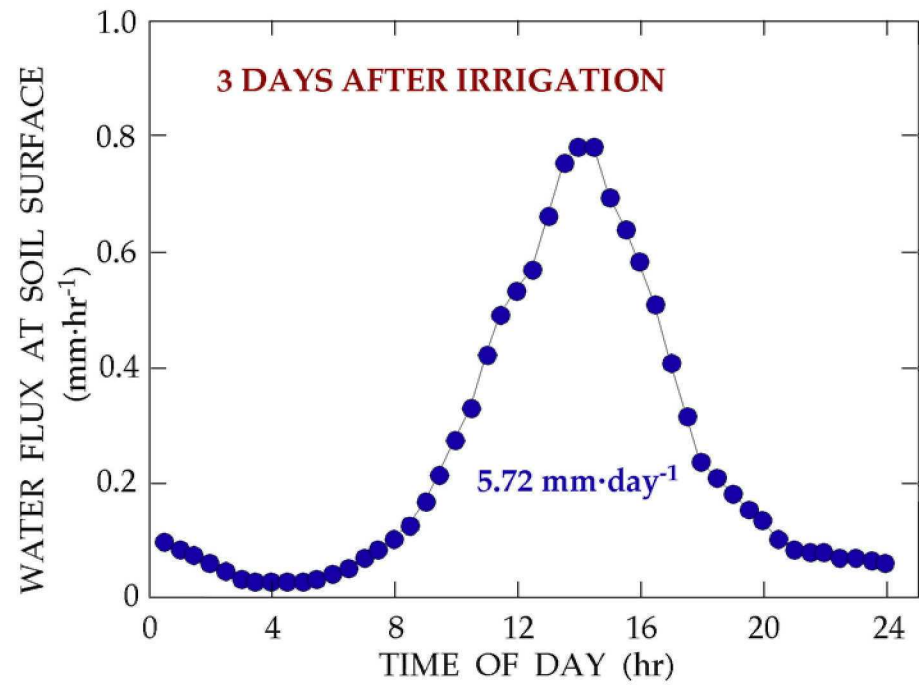
Jackson, Nakayama, Kimball and Reginato (1971)

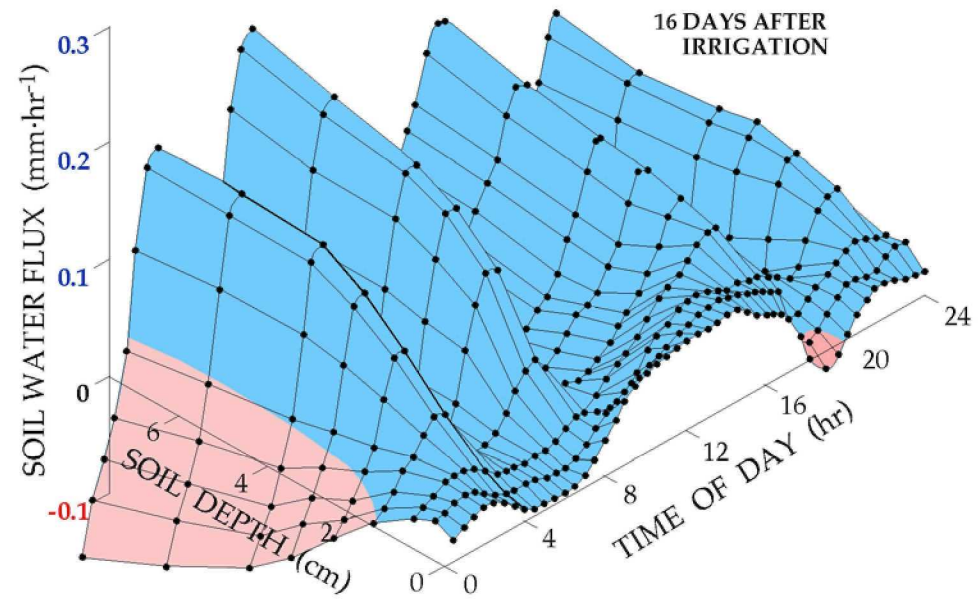
**THOUSANDS OF GRAVIMETRIC MEASUREMENTS
AT 8 SOIL-DEPTH INTERVALS (0 TO 9 cm) 48 TIMES EACH DAY
DURING 16 SELECTED DAYS FOLLOWING A 10-cm IRRIGATION
CONTAINING 12 meq Cl per L.**

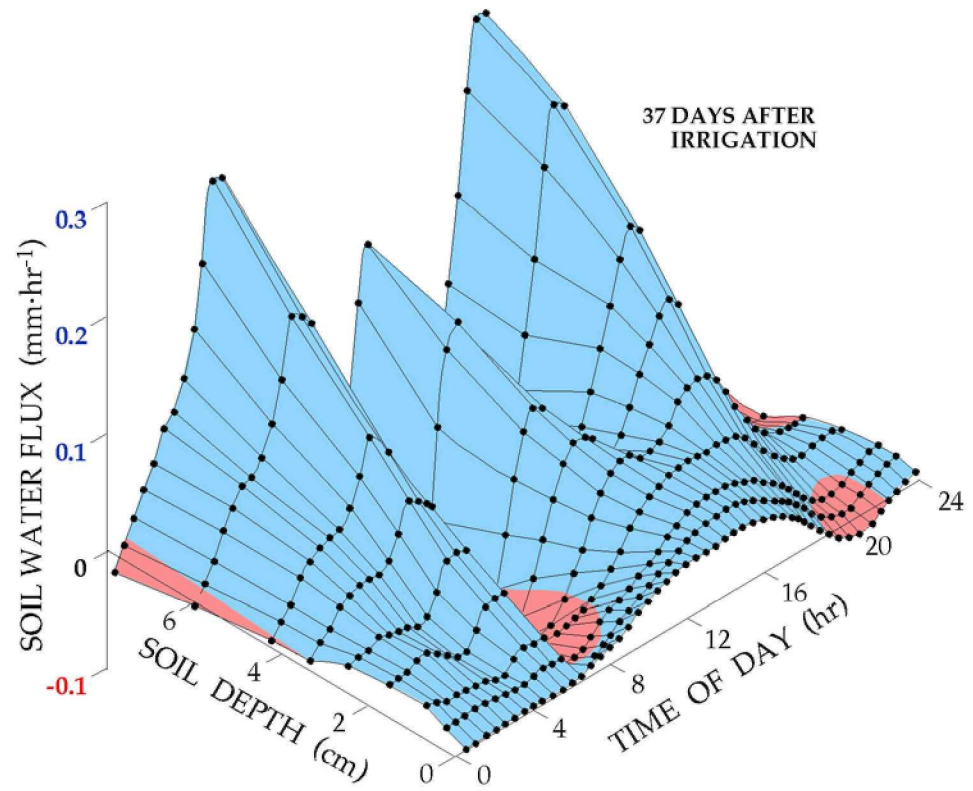
WATER EVAPORATION DATA WERE OBTAINED WITH LYSIMETERS

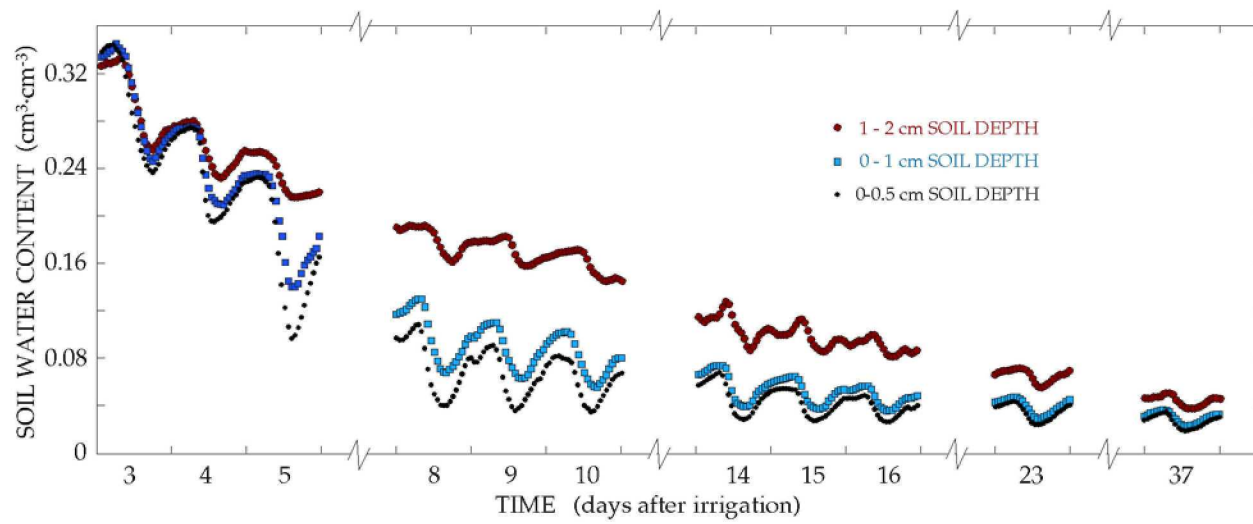
THE EXPERIMENTAL ERROR FOR SOIL WATER CONTENT ($\pm 0.001 \text{ cm}^3 \cdot \text{cm}^{-3}$)

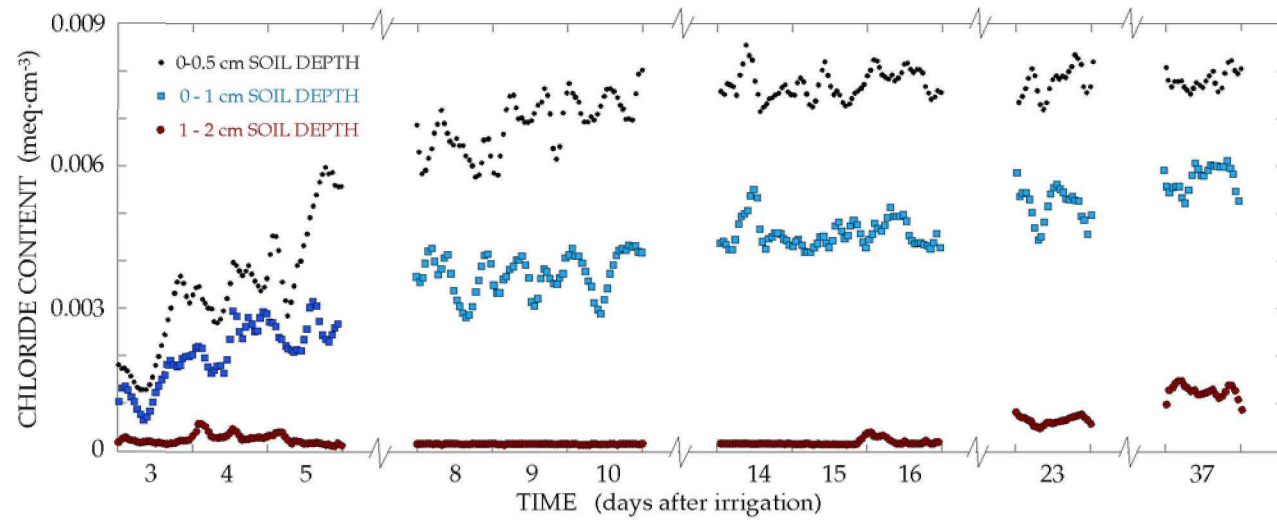
THE EXPERIMENTAL ERROR FOR WATER FLUX DENSITY ($\pm 0.04 \text{ mm} \cdot \text{hr}^{-1}$)

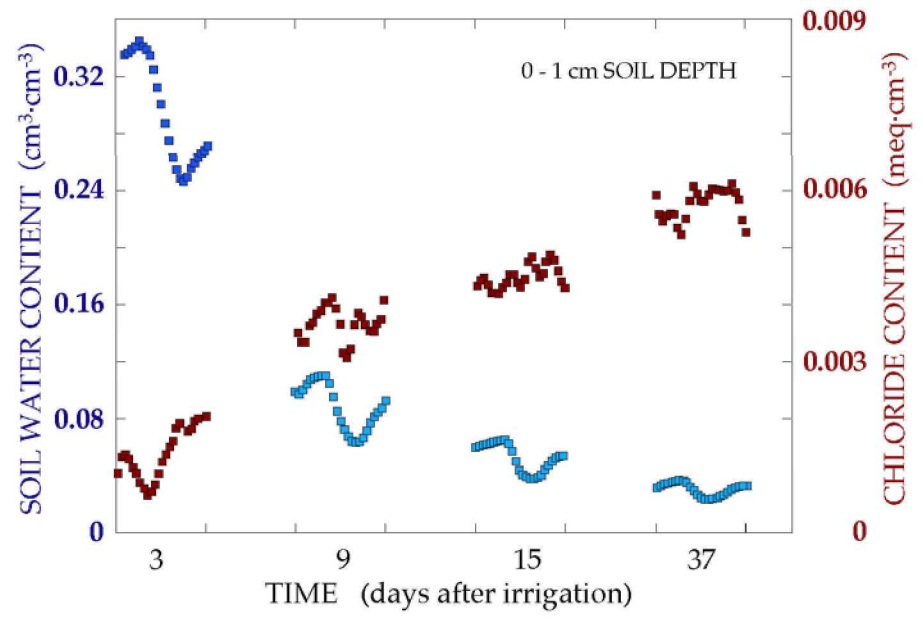


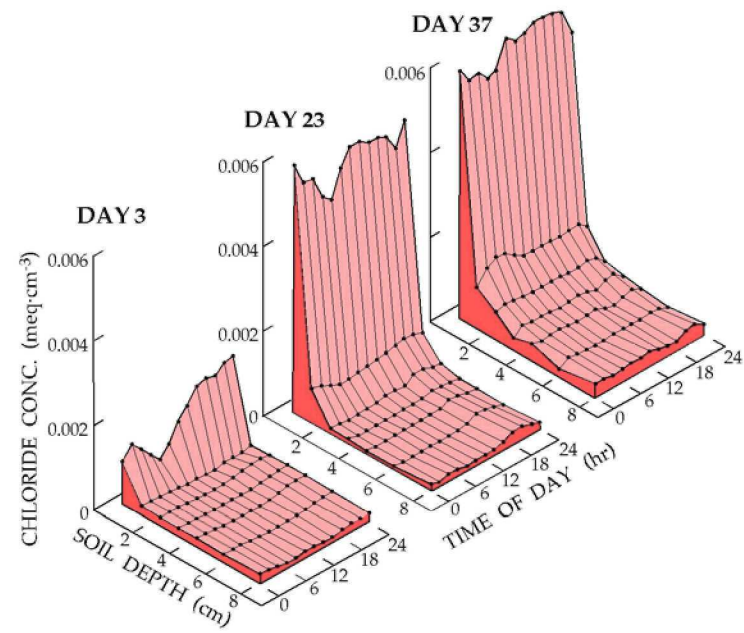












CHLORIDE CONC. INCREASES GREATLY ONLY IN TOP 1 cm

**CHLORIDE STOPPED INCREASING AS θ DECREASED TO
ABOUT $0.04 \text{ cm}^3 \cdot \text{cm}^{-3}$**

**For $\theta < 0.04 \text{ cm}^3 \cdot \text{cm}^{-3}$, liquid water flow is drastically reduced
Cl ions are repelled from negatively charged soil particle surfaces
For $\theta < 0.04 \text{ cm}^3 \cdot \text{cm}^{-3}$, Cl diffusion coefficients approach a value of zero**

**HENCE, SMALL MAGNITUDES OF DOWNWARD Cl DIFFUSION
ARE MATCHED BY SMALL QUANTITIES OF UPWARD Cl CONVECTION**

**Below 1 cm, water vapor and liquid water (with its dissolved salts)
move upward toward the soil surface**

Water leaves the soil surface by evaporation

THE AMOUNT OF SALT LEACHED WITH EACH IRRIGATION OR RAINFALL AND THE AMOUNT OF SALT ACCUMULATED NEAR A SOIL SURFACE IS A DYNAMIC PROCESS

THE FREQUENCY & AMOUNTS OF IRRIGATION TO SALINIZE OR DE-SALINIZE A SOIL DEPEND UPON

- **The physical properties of the soil**
- **The quality of the applied water**
- **The nature and extent of the boundary conditions**
 - **at its surface**
 - **at particular depths within the soil profile**
 - **in the vadose zone below**