



**The Abdus Salam  
International Centre for Theoretical Physics**



**1936-22**

**Advanced School on Synchrotron and Free Electron Laser Sources  
and their Multidisciplinary Applications**

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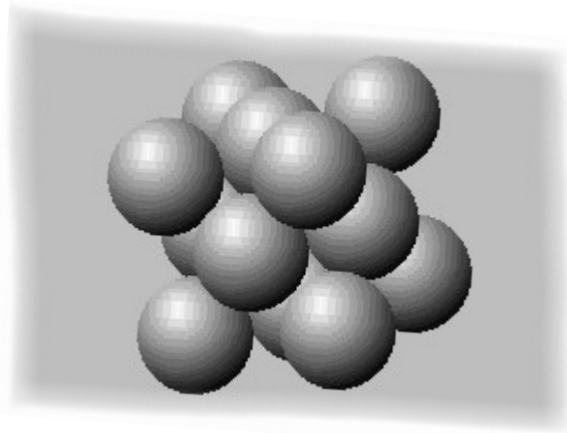
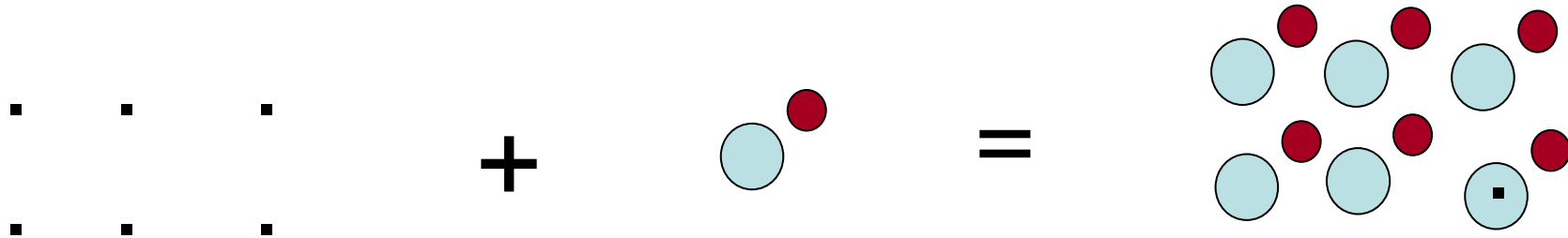
**Single crystal crystallography (Basic Aspects)**

Aldo Craievich  
*University de Sao Paulo  
Brazil*

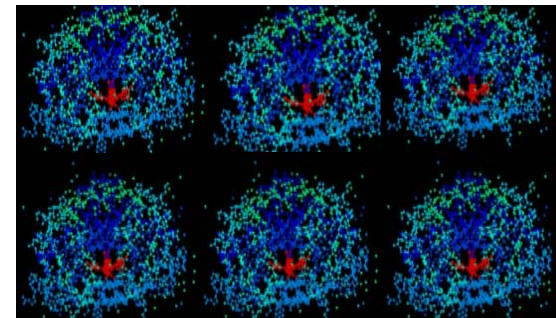
# Basic crystallography for X-ray diffraction

Aldo Craievich  
Institute of Physics  
University of São Paulo Brazil

Point lattice + motif of atoms = Structure

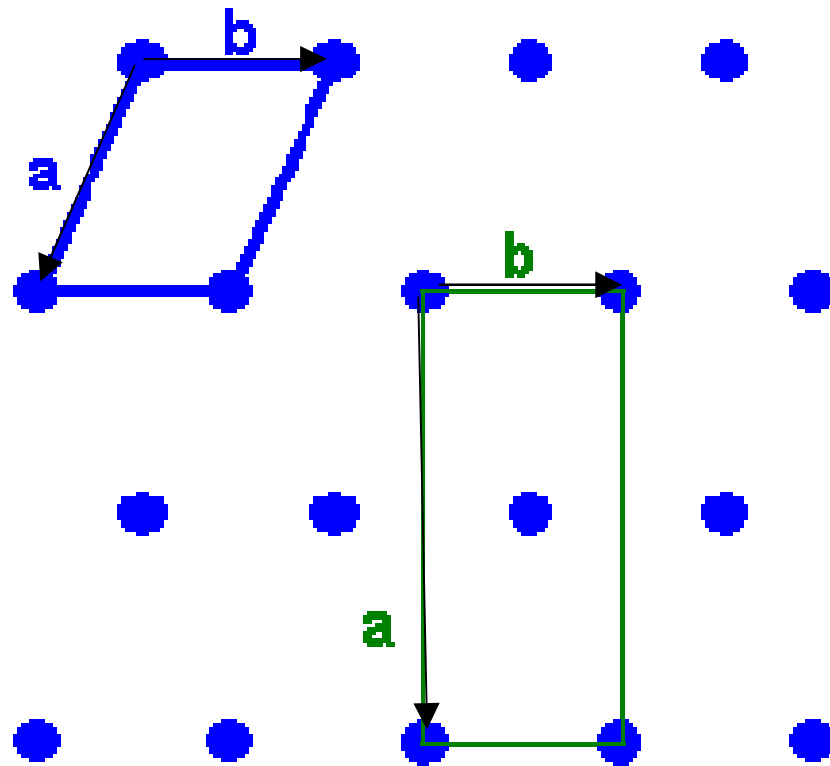


Cu, Au, Ag, ...



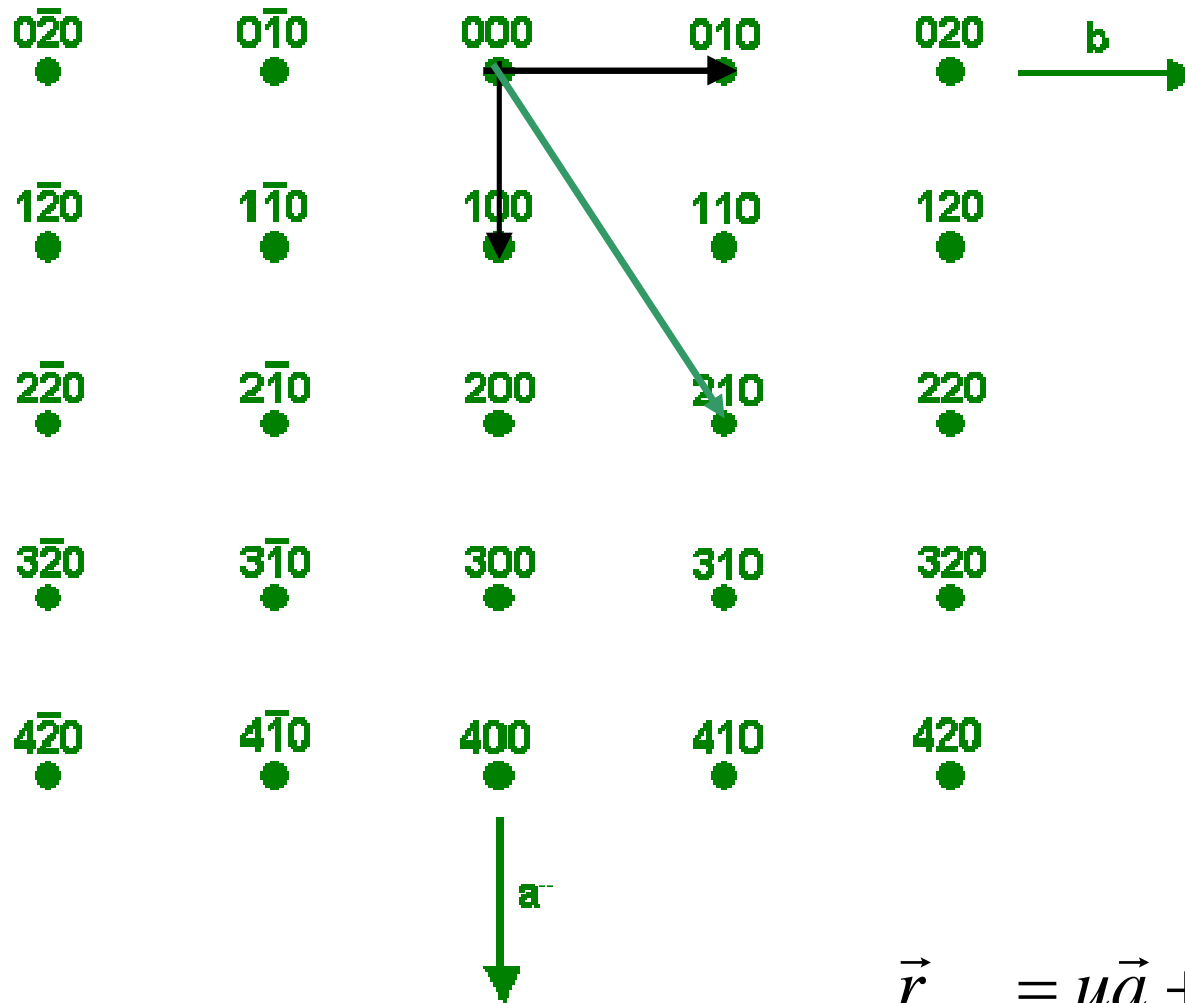
A protein crystal

# Point lattices and unit cells



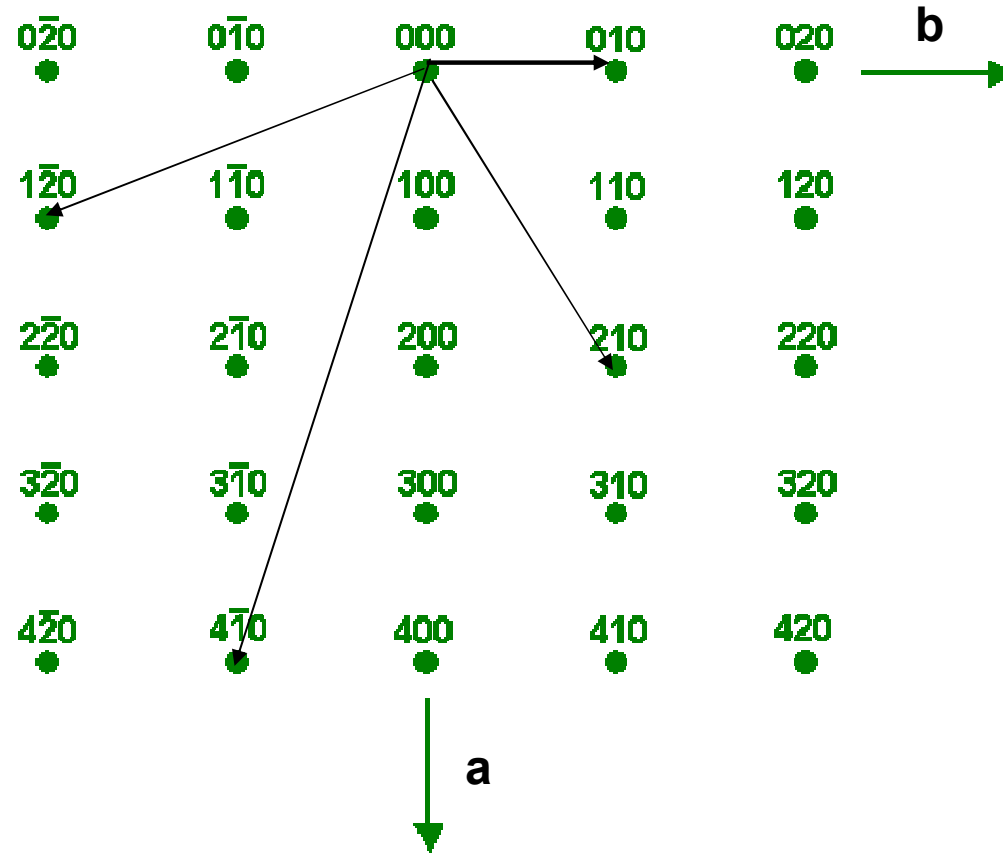
Primitive and non-primitive unit cells

# Lattice points



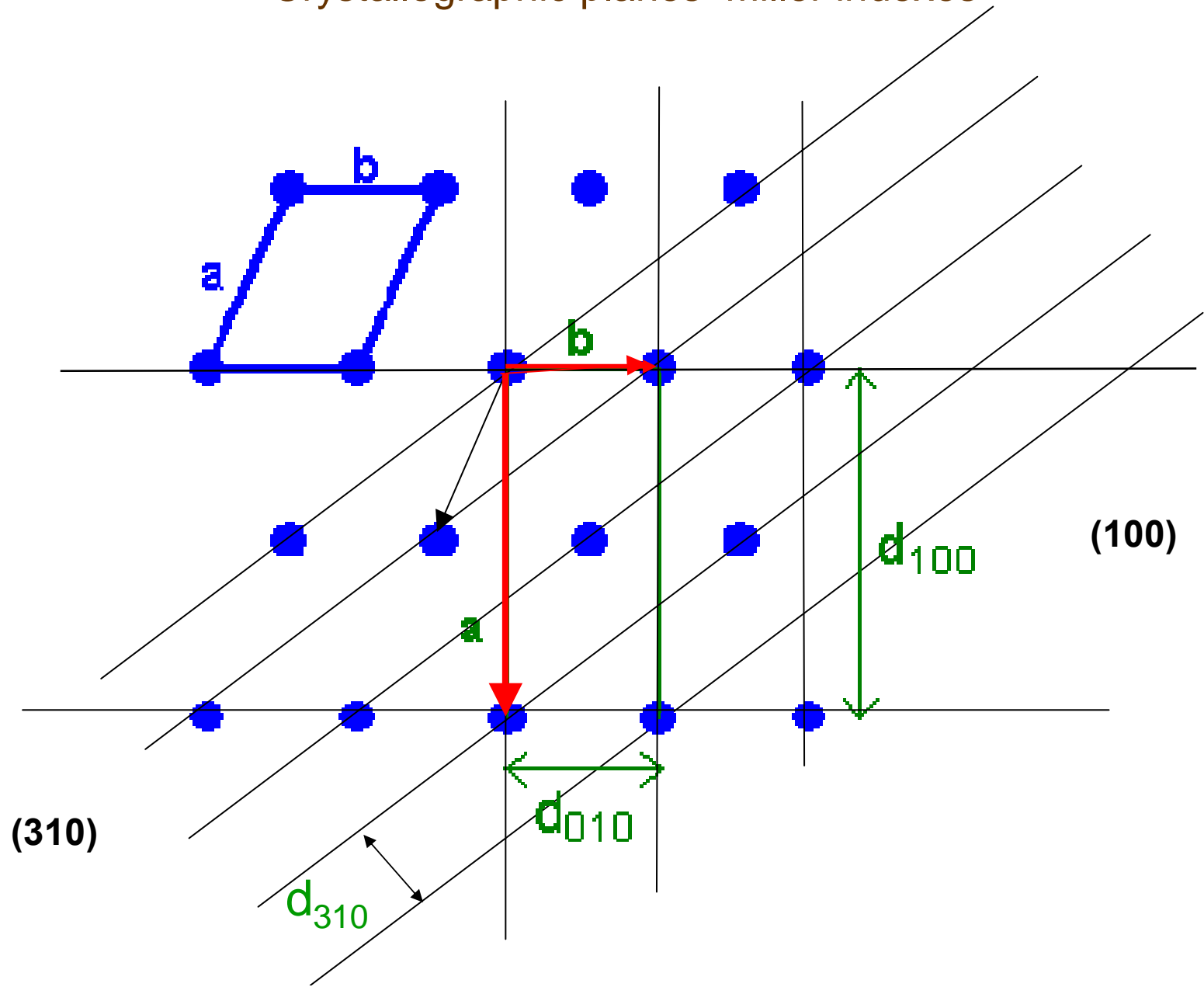
$$\vec{r}_{uvw} = u\vec{a} + v\vec{b} + w\vec{c}$$

# Crystallographic directions

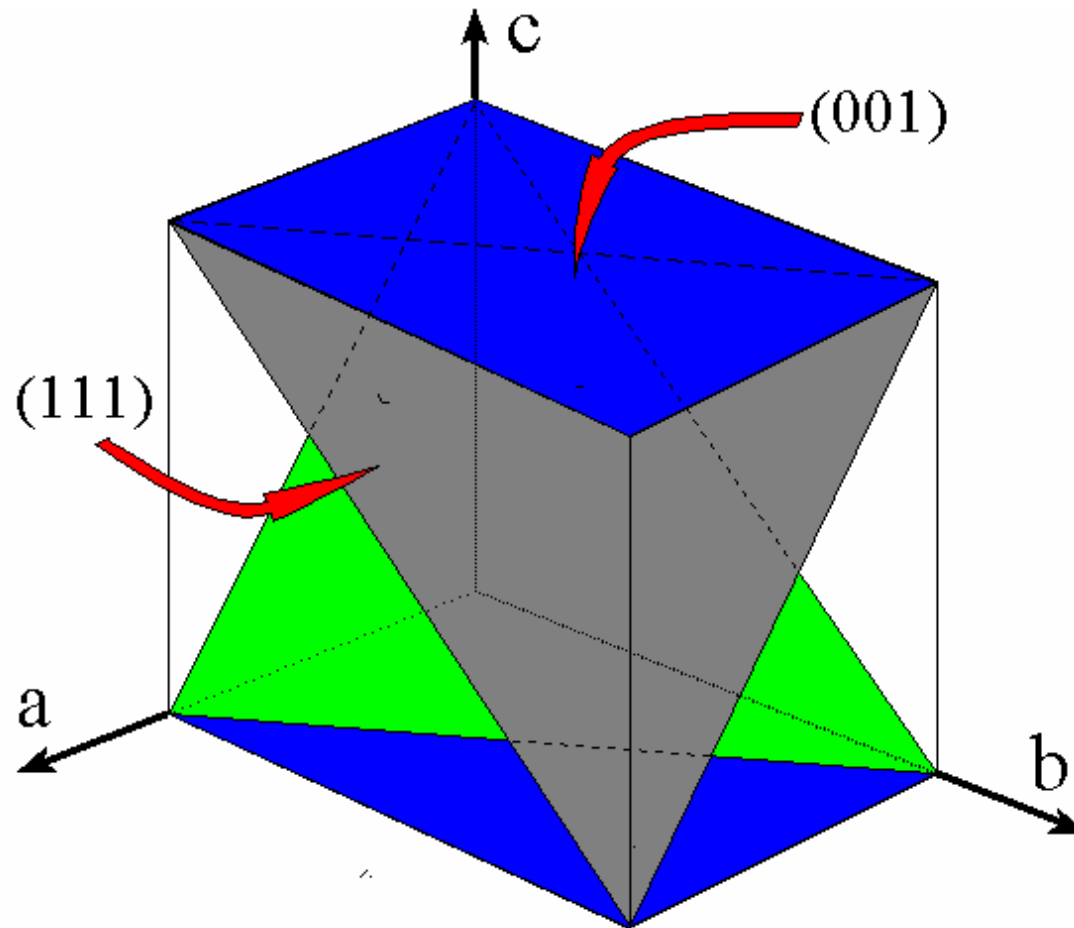


$$\vec{r}_{uvw} = u\vec{a} + v\vec{b} + w\vec{c}$$

# Crystallographic planes Miller indexes



# Miller indexes

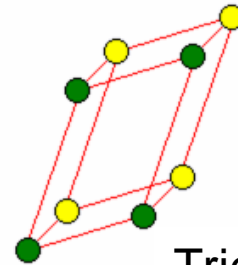




## Crystal systems

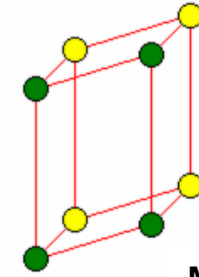
defined by the relationships between unit cell edge length and angles between these edges

- **Triclinic:**  $a \neq b \neq c$   
 $\alpha \neq \beta \neq \gamma$



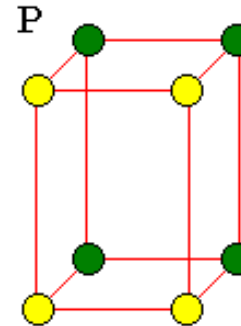
Triclinic

- **Monoclinic:**  $a \neq b \neq c$   
 $\alpha = \gamma = 90^\circ \quad \beta \neq 90^\circ$



Monoclinic

- **Orthorhombic:**  $a \neq b \neq c$   
 $\alpha = \beta = \gamma = 90^\circ$



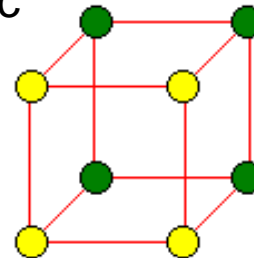
Tetragonal

- **Tetragonal:**  $a = b \neq c$   
 $\alpha = \beta = \gamma = 90^\circ$

- **Hexagonal:**  $a = b \neq c$   
 $\alpha = \beta = 90^\circ \quad \gamma = 120^\circ$

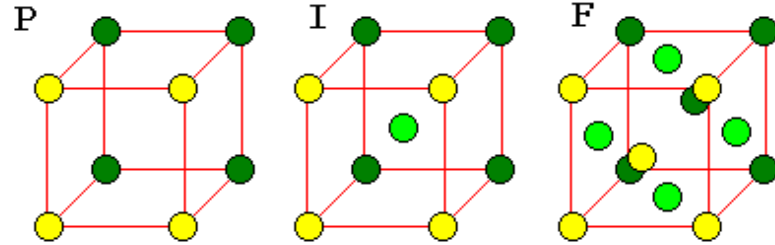
Cubic

- **Cubic:**  $a = b = c$   
 $\alpha = \beta = \gamma = 90^\circ$

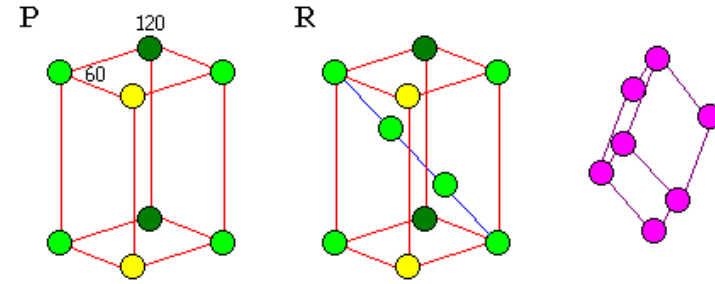


# Bravais lattices

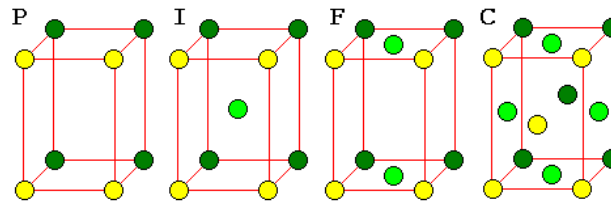
## Cubic (3)



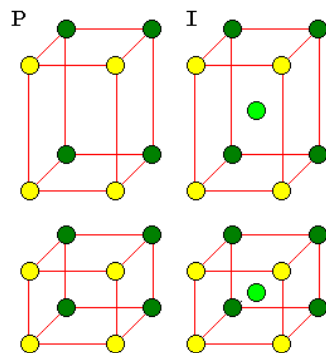
## Hexagonal (2)



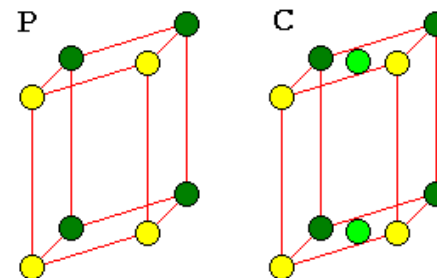
## Tetragonal (4)



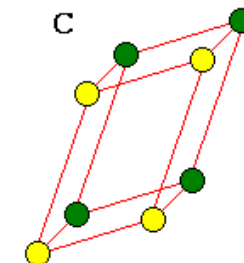
## Orthorhombic (2)



## Monoclinic (2)

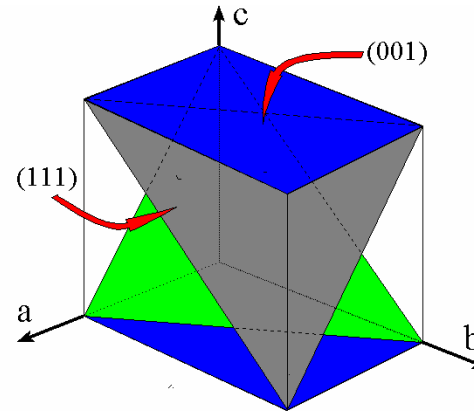


## Triclinic (1)



## Interplanar spacing $d_{hkl}$ for the different crystal systems

Cubic: 
$$d_{hkl} = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$



Orthorhombic: 
$$d_{hkl} = \frac{1}{\sqrt{\frac{h^2}{a^2} + \frac{k^2}{b^2} + \frac{l^2}{c^2}}}$$

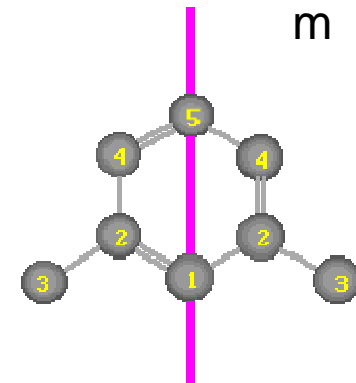
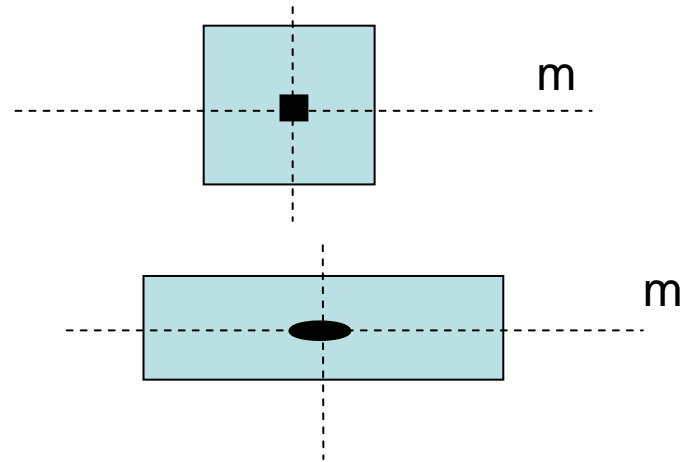
Hexagonal: 
$$d_{hkl} = \frac{1}{\sqrt{\frac{4}{3a^2}(h^2 + hk + l^2) + \frac{l^2}{c^2}}}$$

- etc

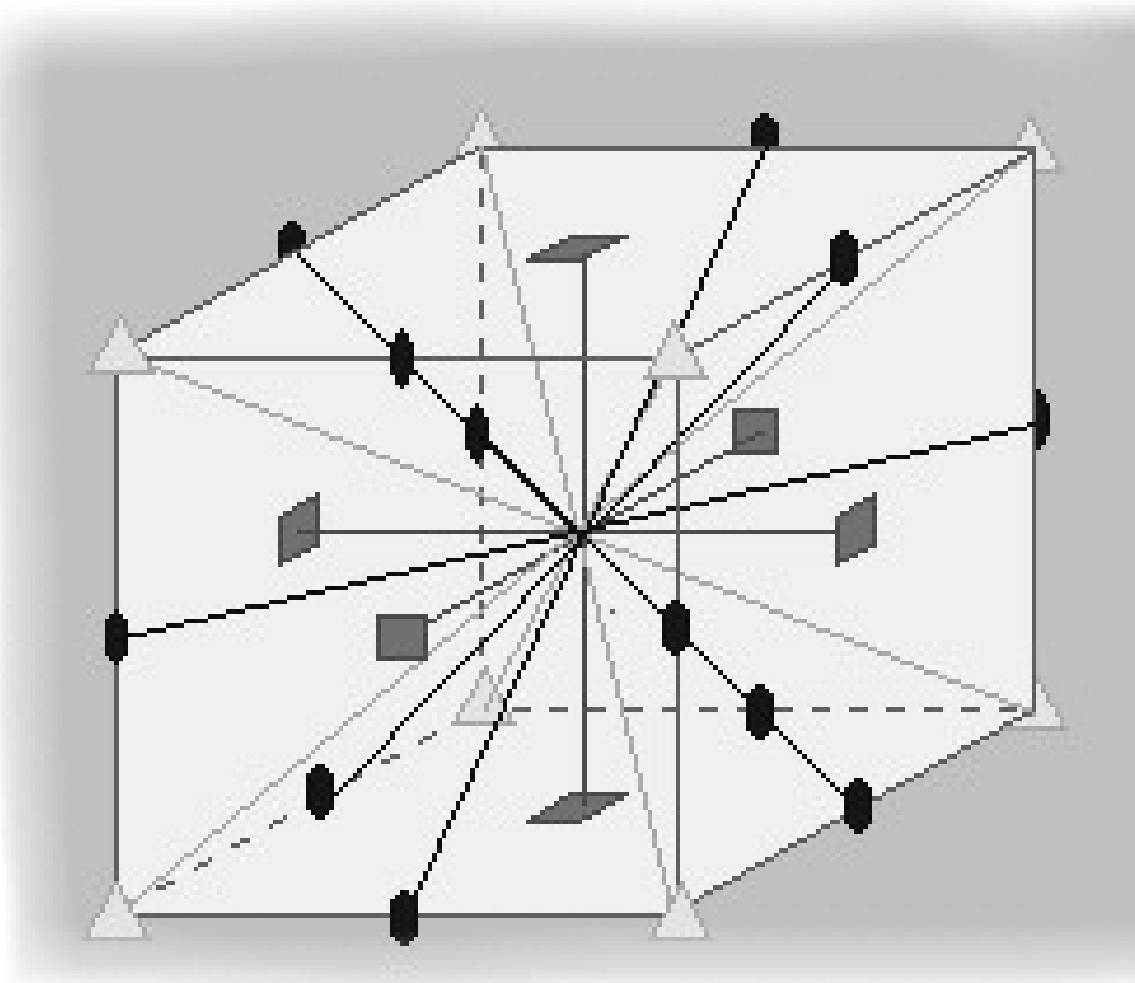
# Symmetry properties

## Point group operations

- Rotations 1, 2, 3, 4, 6
- Mirrors m
- Inversion: i
- Rotation-inversion  $\bar{2}$  ...



# Symmetry operations



9 mirror planes

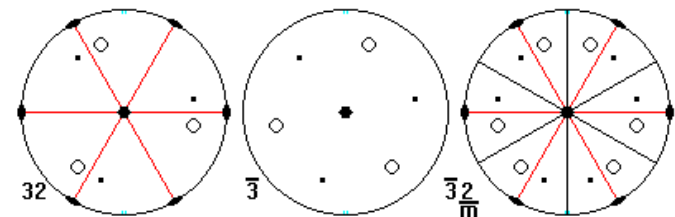
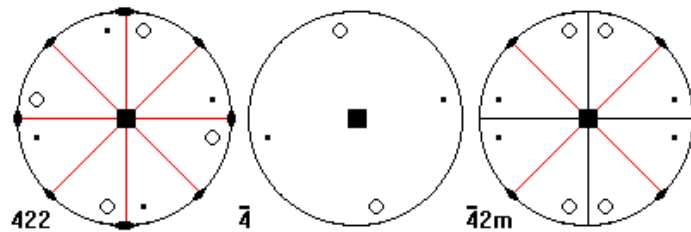
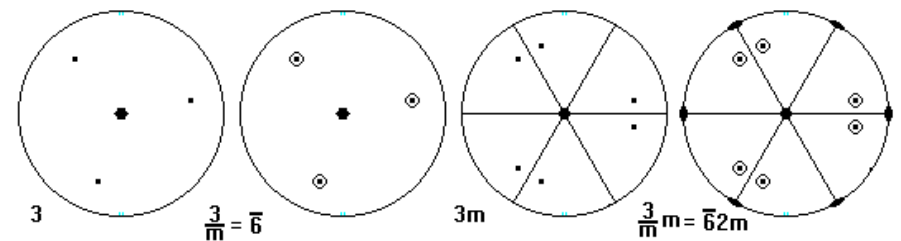
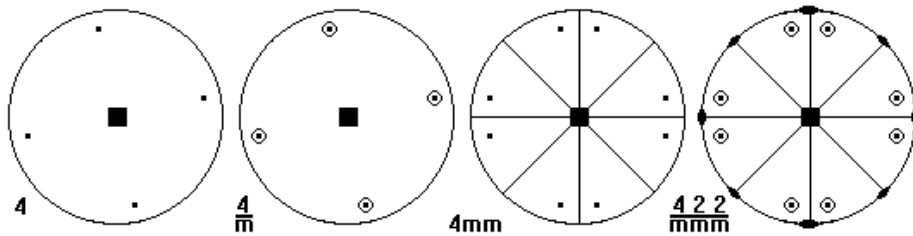
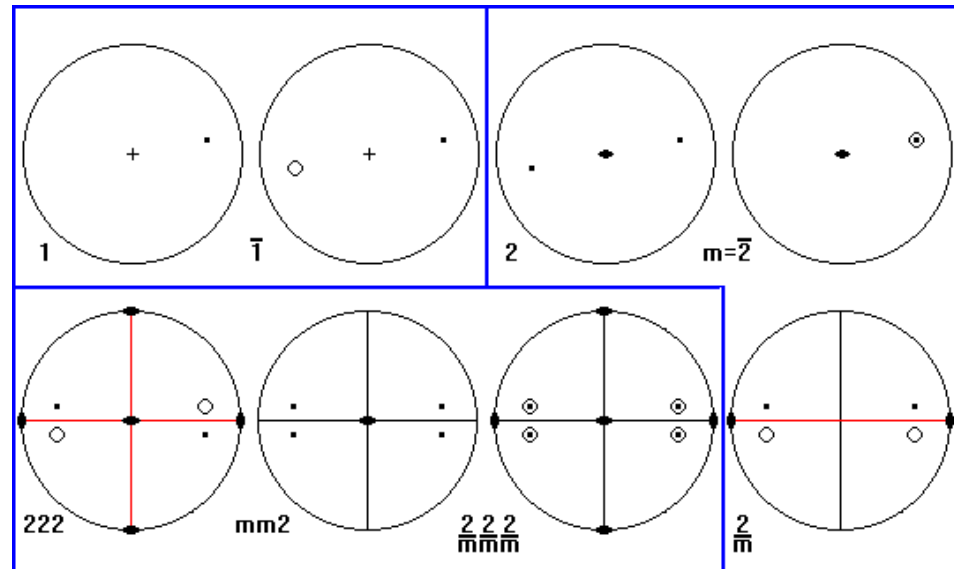
3 tetrad axes

4 triad axes

6 diad axes

# 32 crystallographic point groups

Triclinic,  
Monoclinic and  
orthorombic



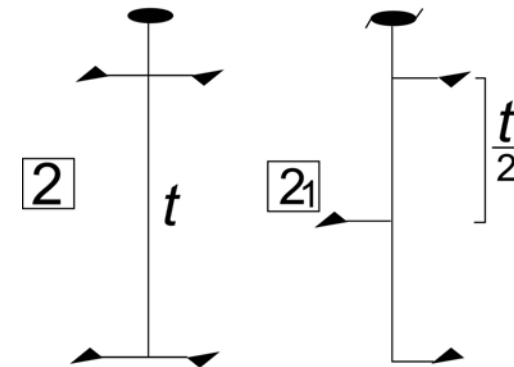
Tetragonal

Hexagonal (Trigonal)

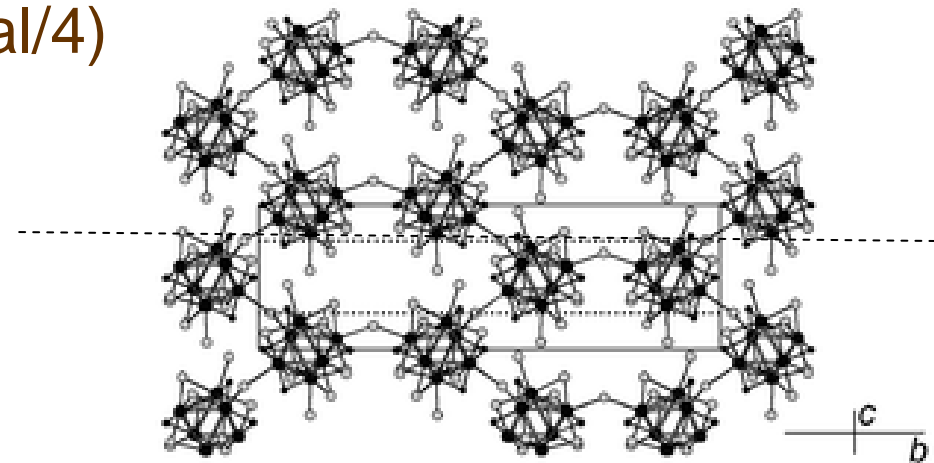
# Translational symmetry operations

- Screw axis
- (Examples:  $2_1$ ,  $4_1$ ,  $4_2$  ...)

2-fold screw axes



- Glide plane  $a/2$ ,  $b/2$ ,  $c/2$ ,  $n$  (2D diagonal/2),  $d$  (3D diagonal/4)



## 230 space groups

- **Triclinic**

1.  $P1$     2.  $P-1$

- **Monoclinic**

3.  $P121$     4.  $P1211$     5.  $C121$

6.  $P1m1$     7.  $P1c1$     8.  $C1m1$

9.  $C1c1$     10.  $P12/m1$     11.  $P121/m1$

12.  $C12/m1$     13.  $P12/c1$

14.  $P121/c1$     15.  $C12/c1$



- **Cubic**

195.  $P 2 3$

198.  $P 21 3$

201.  $P n -3$

204.  $I m -3$

207.  $P 4 3 2$

210.  $F 41 3 2$

213.  $P 41 3 2$

216.  $F -4 3 m$

219.  $F -4 3 c$

222.  $P n -3 n$

225.  $F m -3 m$

228.  $F d -3 c$

196.  $F 2 3$

199.  $I 21 3$

202.  $F m -3$

205.  $P a -3$

208.  $P 42 3 2$

211.  $I 4 3 2$

214.  $I 41 3 2$

217.  $I -4 3 m$

220.  $I -4 3 d$

223.  $P m -3 n$

226.  $F m -3 c$

229.  $I m -3 m$

197.  $I 2 3$

200.  $P m -3$

203.  $F d -3$

206.  $I a -3$

209.  $F 4 3 2$

212.  $P 43 3 2$

215.  $P -4 3 m$

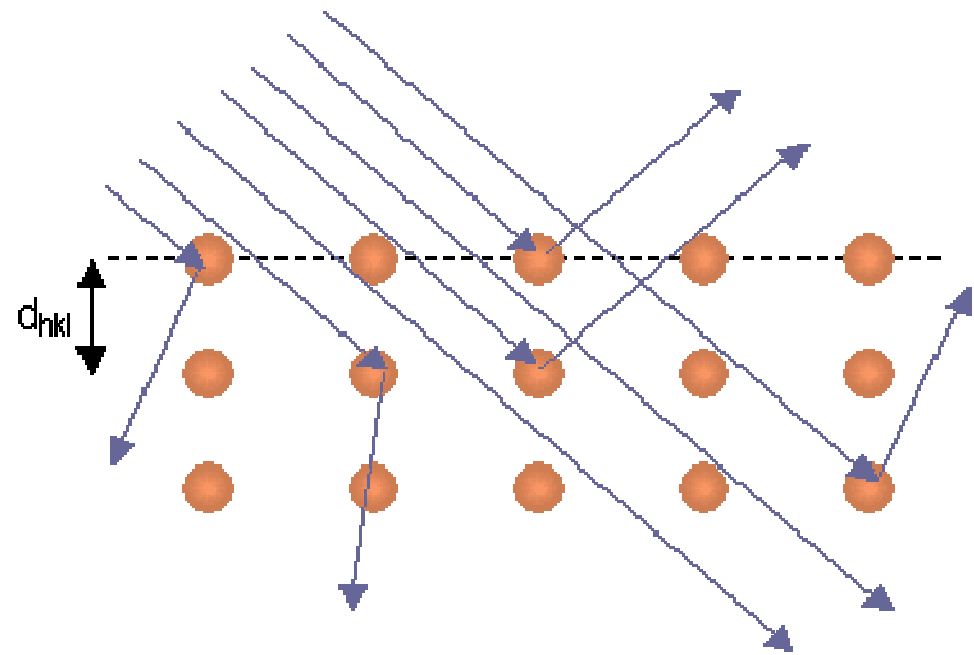
218.  $P -4 3 n$

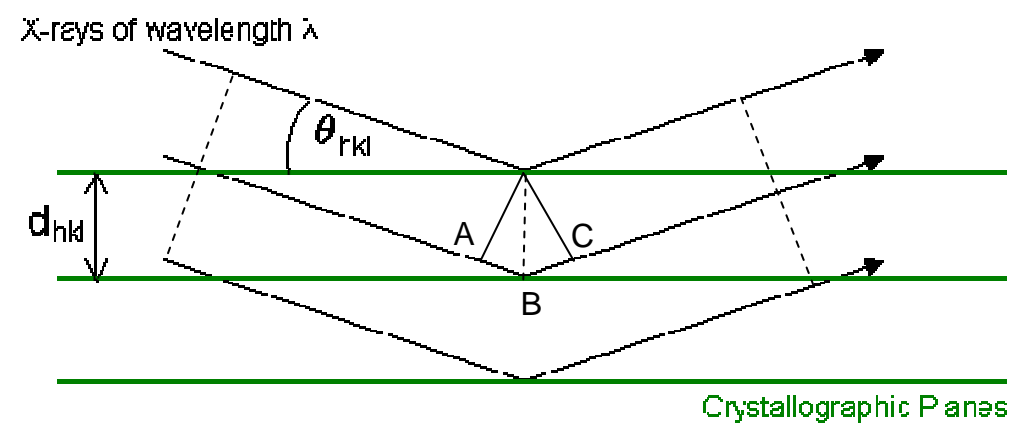
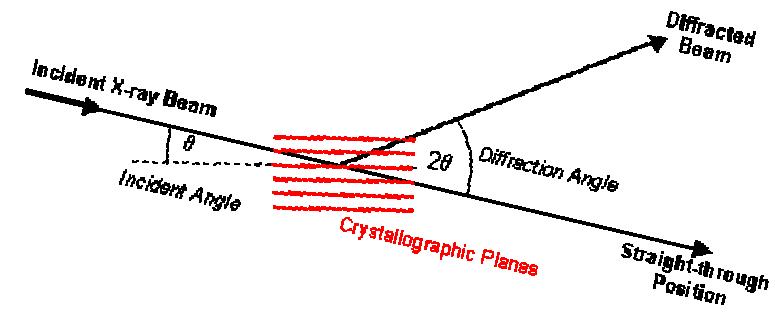
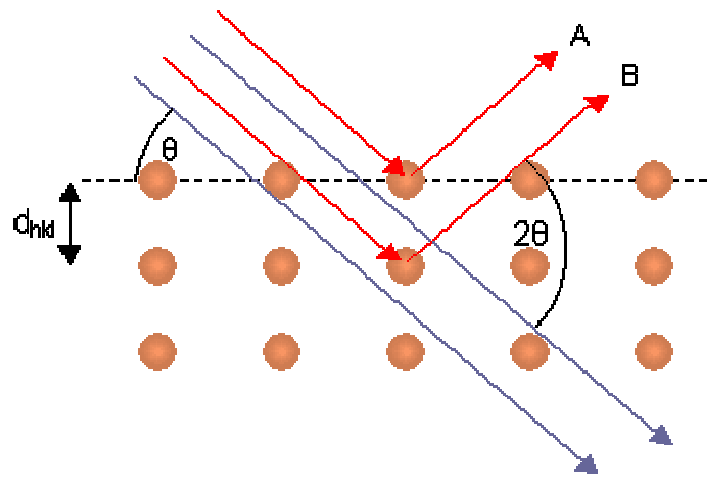
221.  $P m -3 m$

224.  $P n -3 m$

227.  $F d -3 m$

230.  $I a -3 d$





$$\overline{AB} + \overline{BC} = 2d_{hkl} \sin \theta_{hkl} = n\lambda$$

**Bragg law :**

$$2d_{hkl} \sin \theta_{hkl} = n\lambda$$

# Reciprocal lattice

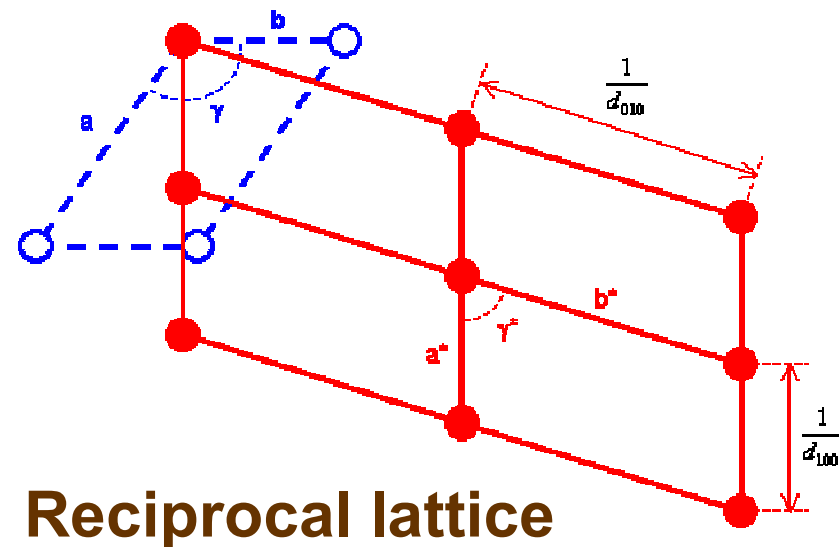
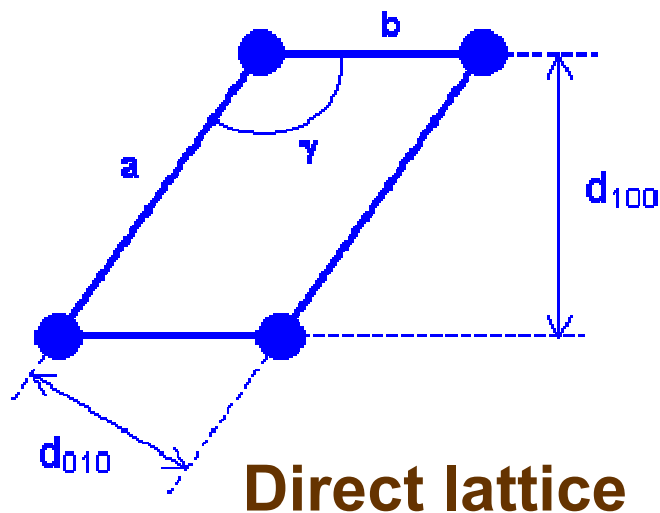
Let be a direct lattice defined by  $\vec{r}_{uvw} = u.\vec{a} + v.\vec{b} + w.\vec{c}$

Definitions of the unit vectors of the reciprocal lattice:

$$\vec{a}^* = \frac{\vec{b} \times \vec{c}}{V_c} \quad \vec{b}^* = \frac{\vec{c} \times \vec{a}}{V_c} \quad \vec{c}^* = \frac{\vec{a} \times \vec{b}}{V_c}$$

Reciprocal space vector:  $\vec{s} = s_x.\vec{a}^* + s_y.\vec{b}^* + s_z.\vec{c}^*$

Reciprocal lattice vector:  $\vec{r}_{hkl}^* = h.\vec{a}^* + k.\vec{b}^* + l.\vec{c}^*$      $h, k, l$  integers



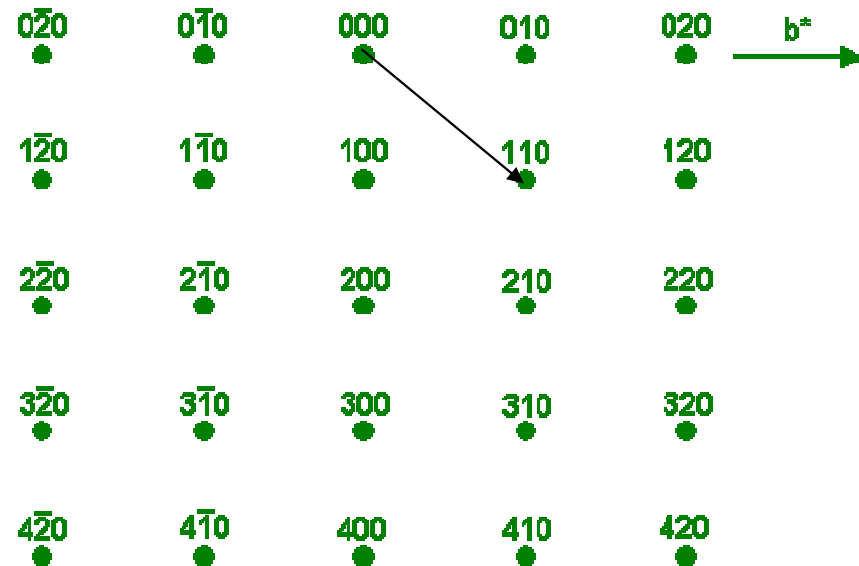
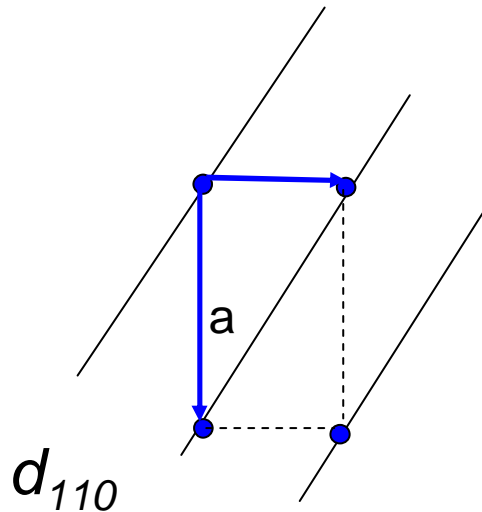
# Properties

i)  $\vec{a} \cdot \vec{a}^* = 1$      $\vec{b}^* \cdot \vec{b} = 1$      $\vec{c}^* \cdot \vec{c} = 1$

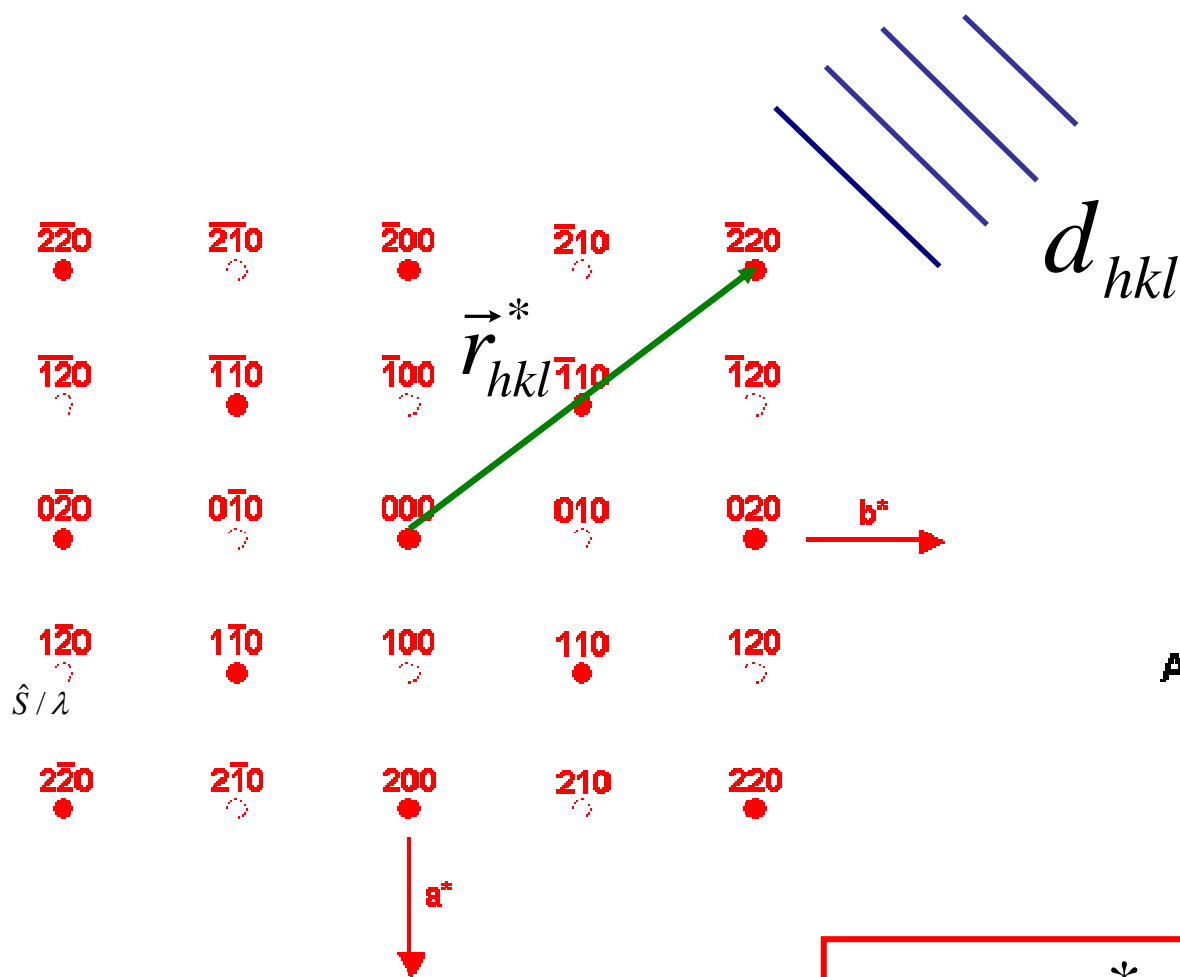
ii)  $\vec{a}^* \cdot \vec{b} = 0$      $\vec{b}^* \cdot \vec{c} = 0$      $\vec{a}^* \cdot \vec{c} = 0$

iii)  $\vec{r}_{hkl}^* = h \cdot \vec{a}^* + k \cdot \vec{b}^* + l \cdot \vec{c}^*$  is perpendicular to the planes (hkl)

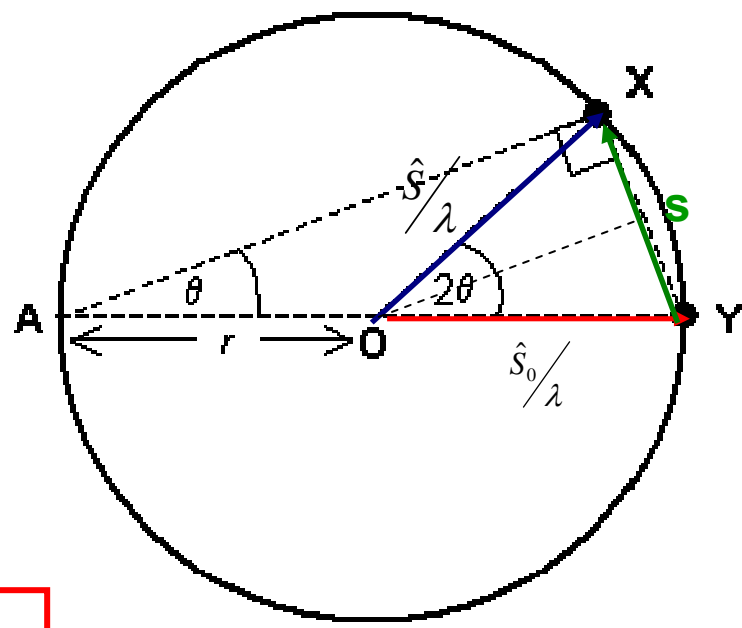
iv)  $d_{hkl} = 1 / |\vec{r}_{hkl}^*|$



$\vec{r}_{110}^* = 1 \cdot \vec{a}^* + 1 \cdot \vec{b}^* + 0 \cdot \vec{c}^*$



### Ewald construction



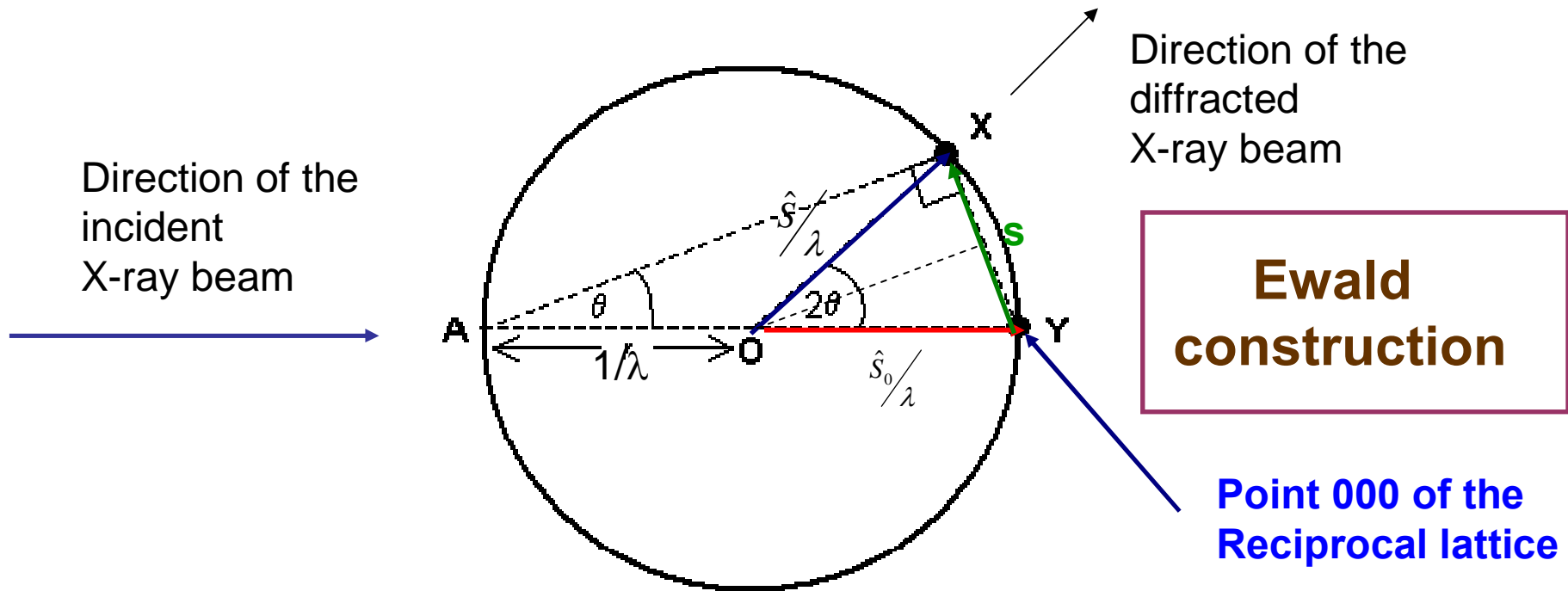
Bragg law in reciprocal space:

$$\vec{S} = \vec{r}_{hkl}^*$$

... is equivalent to:

$$\lambda = 2 \cdot d_{hkl} \cdot \sin \theta_{hkl}$$

(classical Bragg law)



If  $\vec{S} = \vec{r}_{hkl}^*$  (Bragg law in the reciprocal space)

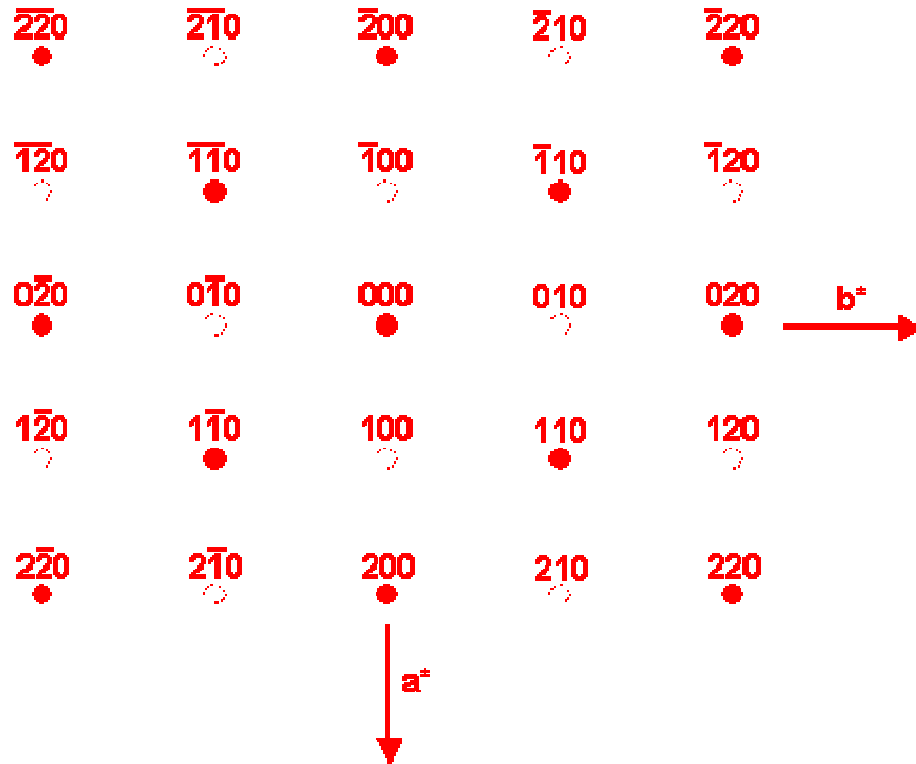
$$2 \cdot \frac{1}{\lambda} \cdot \sin \theta_{hkl} = |\vec{r}_{hkl}^*| = \frac{1}{d_{hkl}}$$

$$2 \cdot d_{hkl} \sin \theta_{hkl} = \lambda$$

(Bragg law in the direct space)

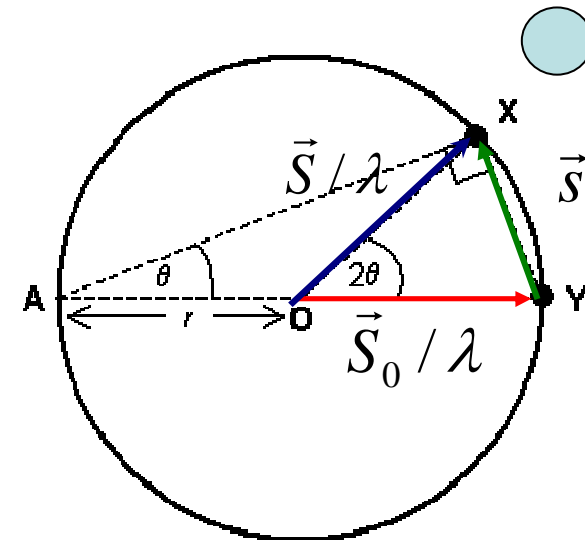
# Application of Bragg law in reciprocal space

(  $\vec{S} = \vec{r}_{hkl}^*$  ) : Ewald construction



Direction of the incident X-ray beam

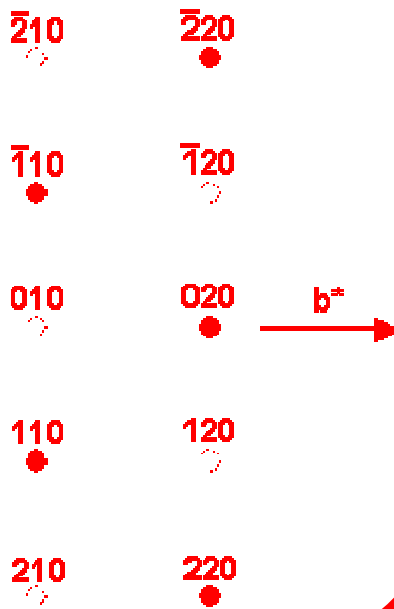
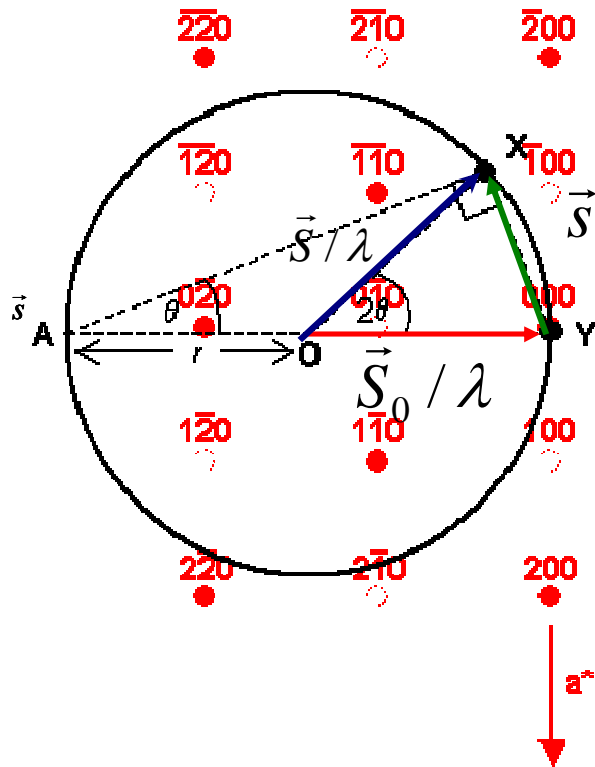
Detector





This example:  $\vec{S} \neq \vec{r}_{hkl}^*$

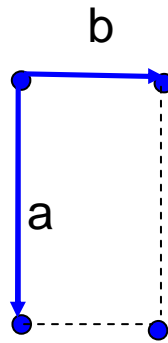
Detector ?



Direction of the incident X-ray beam

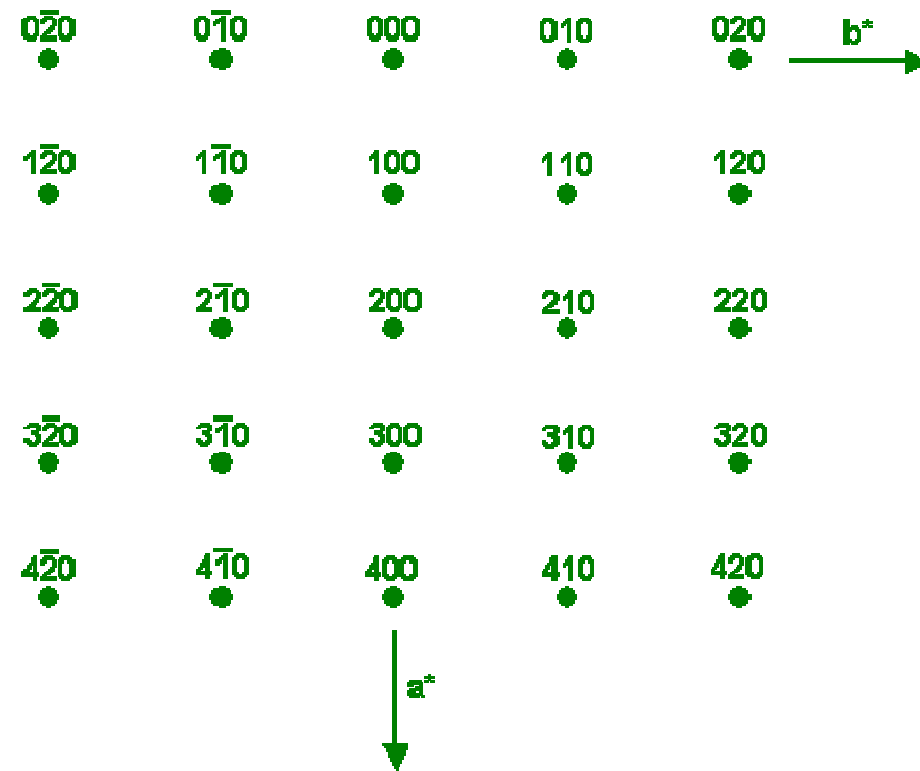
How to proceed in order to determine the right crystal orientation and angular position of the X-ray detector to satisfy Bragg conditions?

## Ewald construction Step 1



Direct unit cell

- Ewald construction Step 2

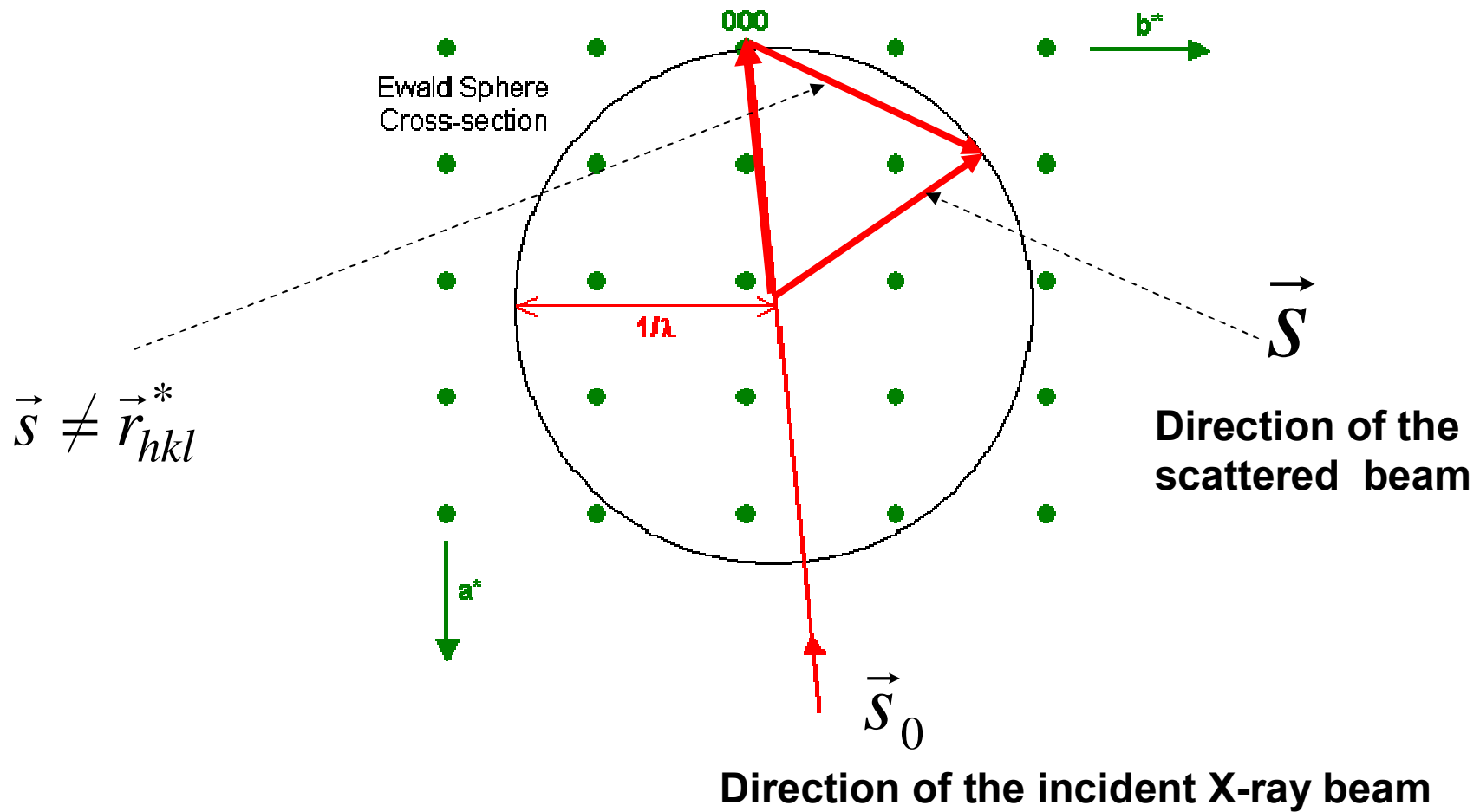


$$\vec{a}^* = \frac{\vec{b} \times \vec{c}}{V_c}$$

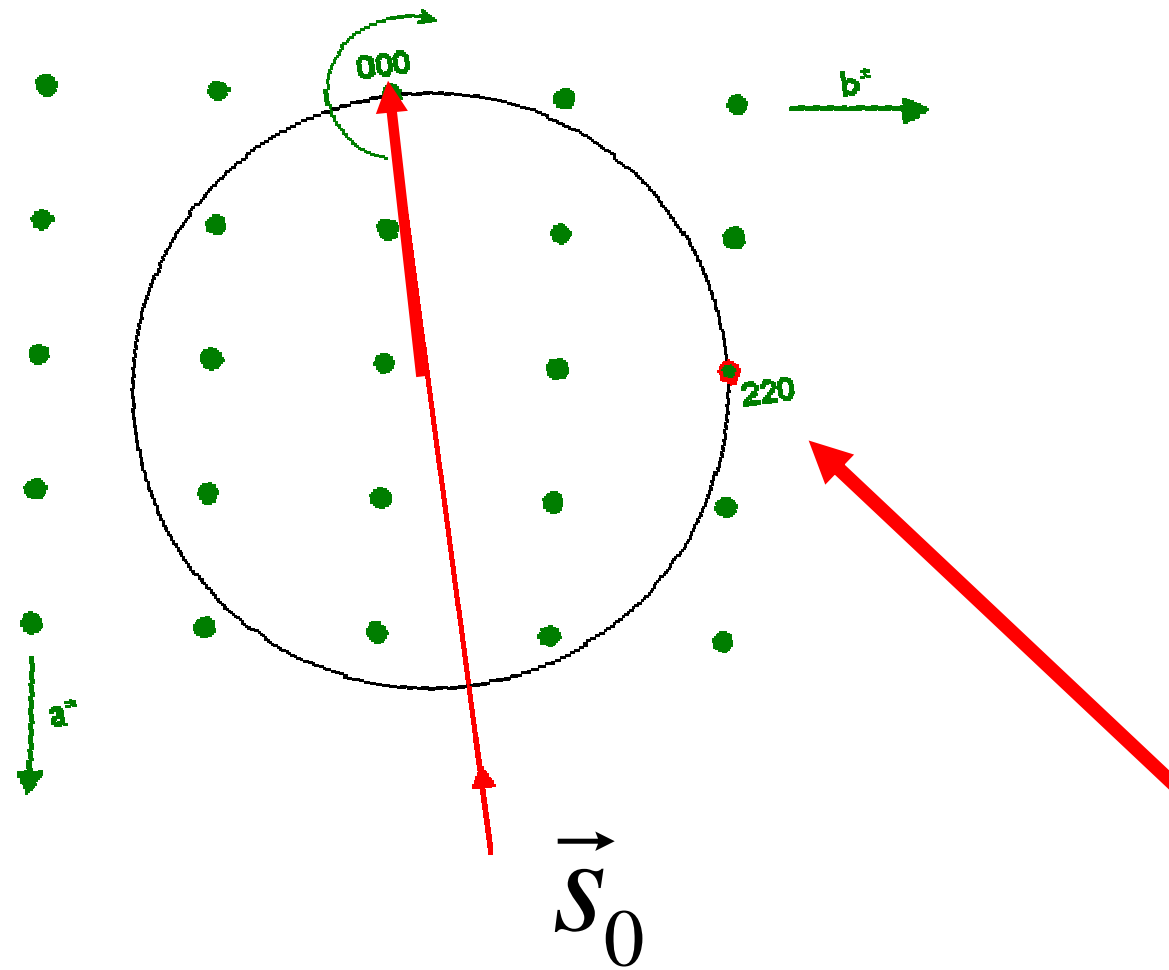
$$\vec{b}^* = \frac{\vec{c} \times \vec{a}}{V_c}$$

$$\vec{c}^* = \frac{\vec{a} \times \vec{b}}{V_c}$$

### Ewald construction Step 3

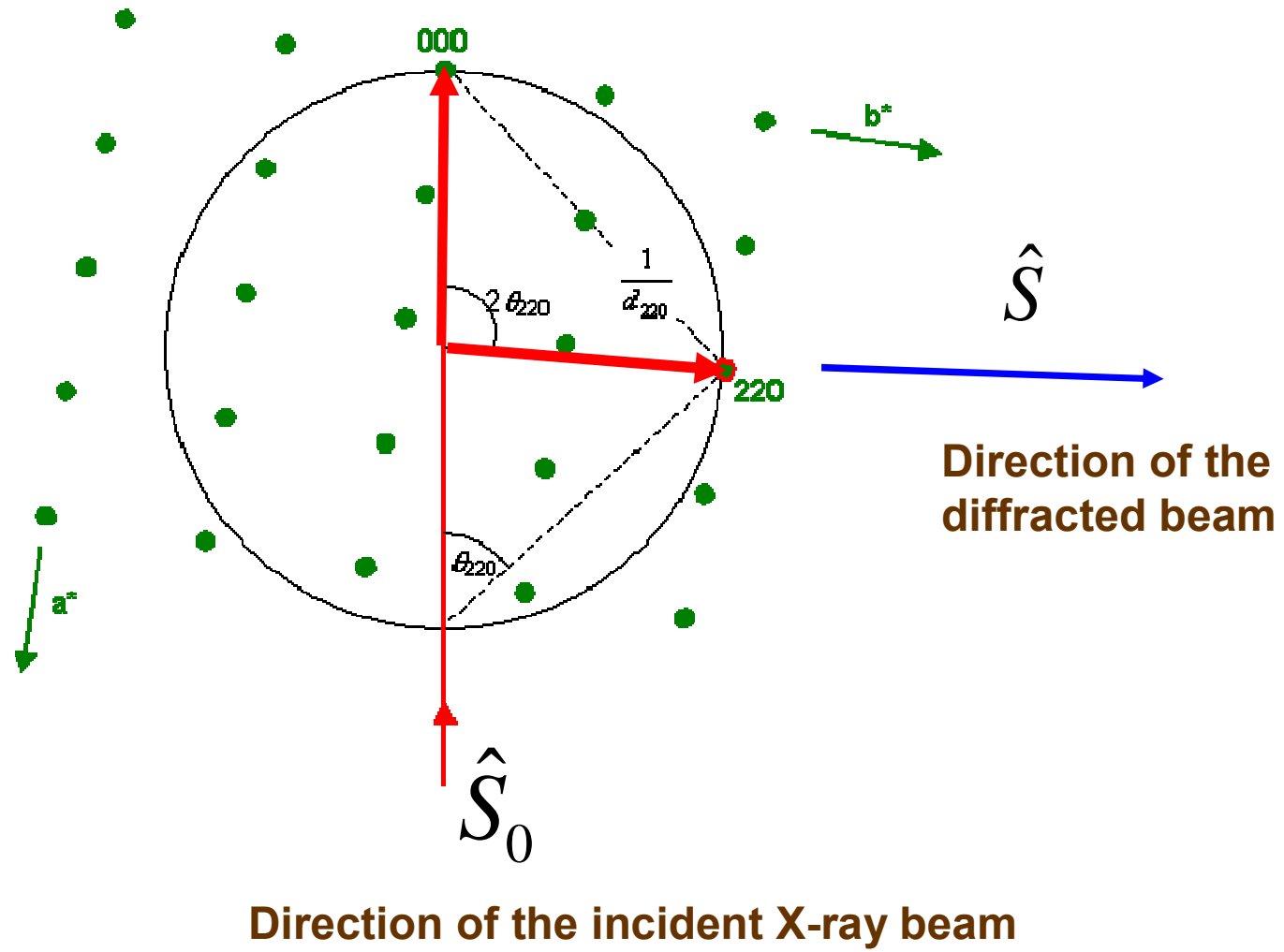


- Ewald construction Step 4

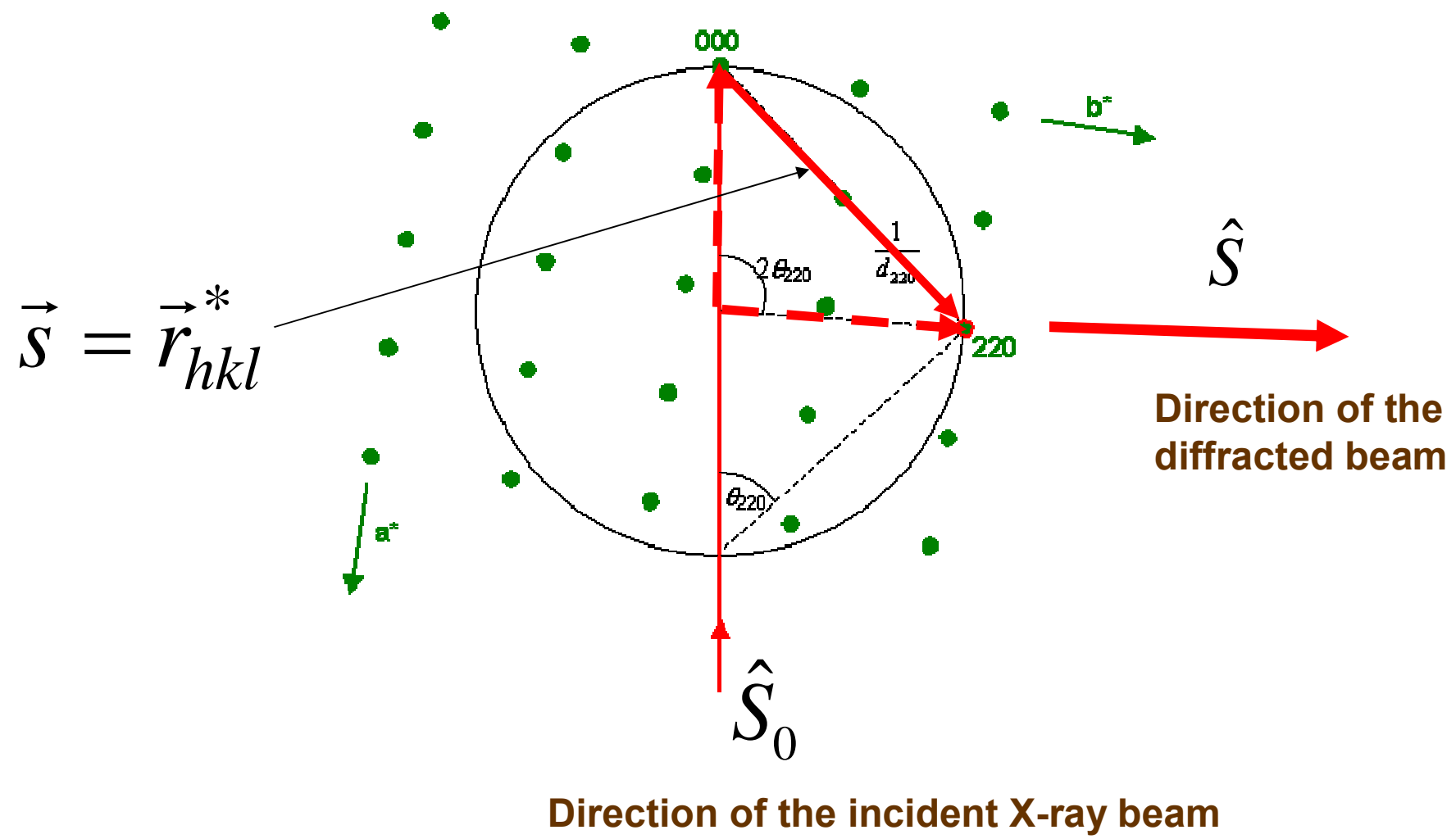


Direction of the incident X-ray beam

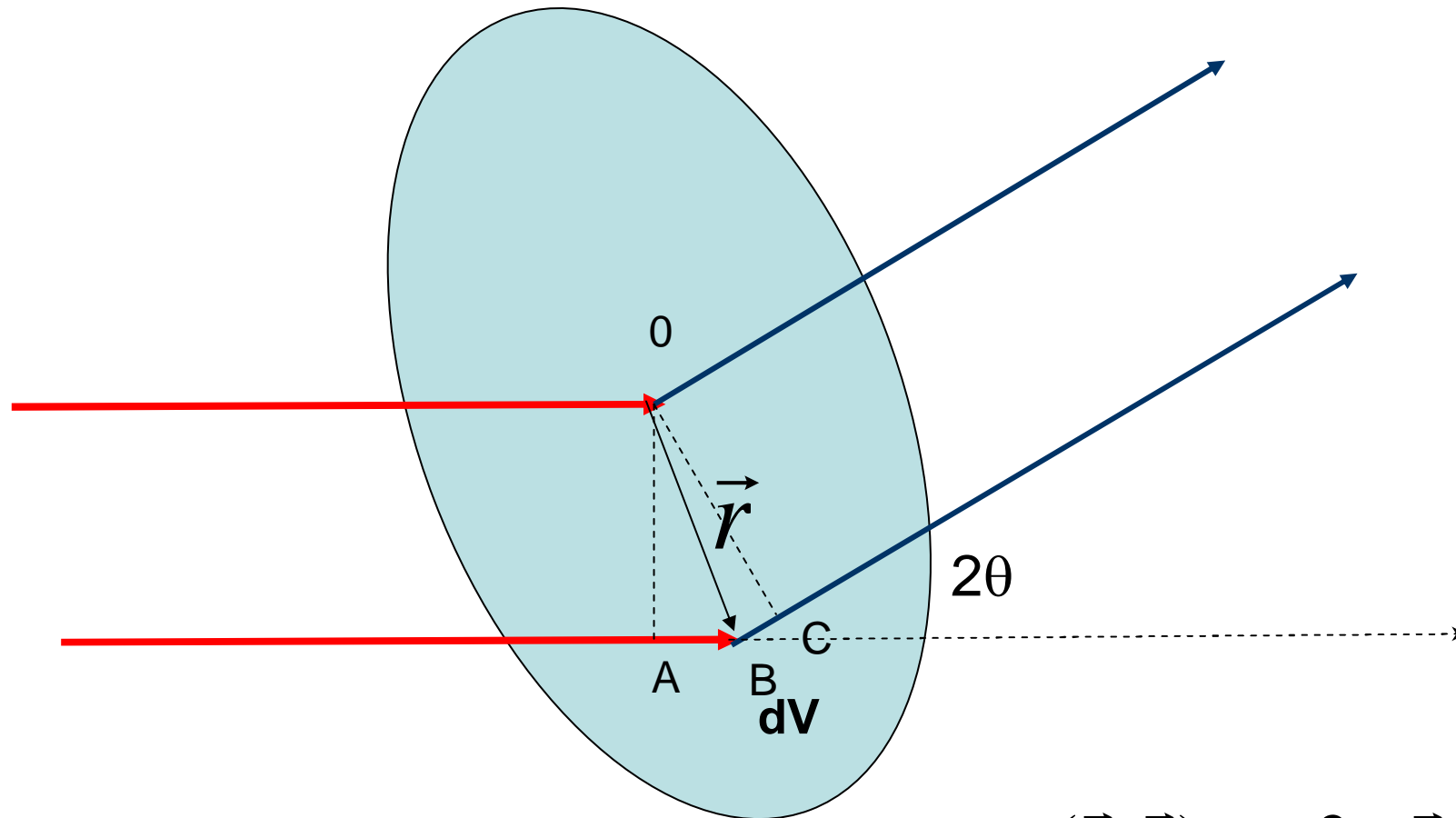
- Ewald construction Step 5



- Ewald construction Step 5



## Amplitude of beams diffracted by materials of arbitrary structure



$$dA(\vec{s}) = A_e \cdot \rho(\vec{r}) \cdot d\vec{r} \cdot e^{i\varphi(\vec{r}, \vec{s})}$$

$$\varphi(\vec{s}, \vec{r}) = -2\pi \cdot \vec{s} \cdot \vec{r}$$

$\rho(\mathbf{r})$   $\longleftrightarrow$   $A(\mathbf{s})$  and Fourier transforms:

$$A(\vec{s})_N = A_e \int \rho(\vec{r}) \cdot e^{-2\pi i \cdot \vec{s} \cdot \vec{r}} d\vec{r}$$

$$\rho(\vec{r}) = (1/A_e) \int A(\vec{s})_N \cdot e^{2\pi i \cdot \vec{s} \cdot \vec{r}} d\vec{s}$$



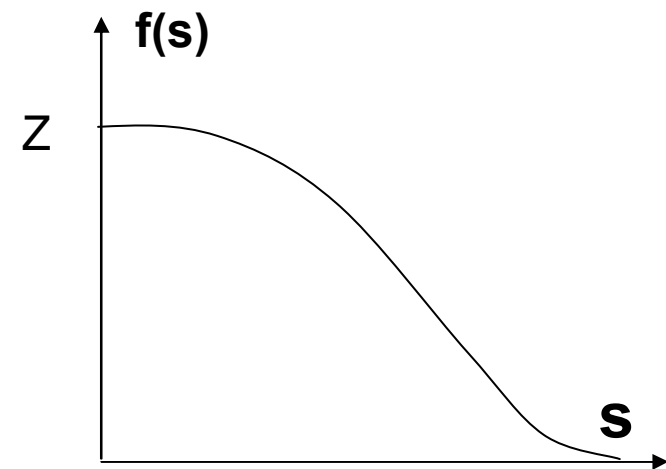
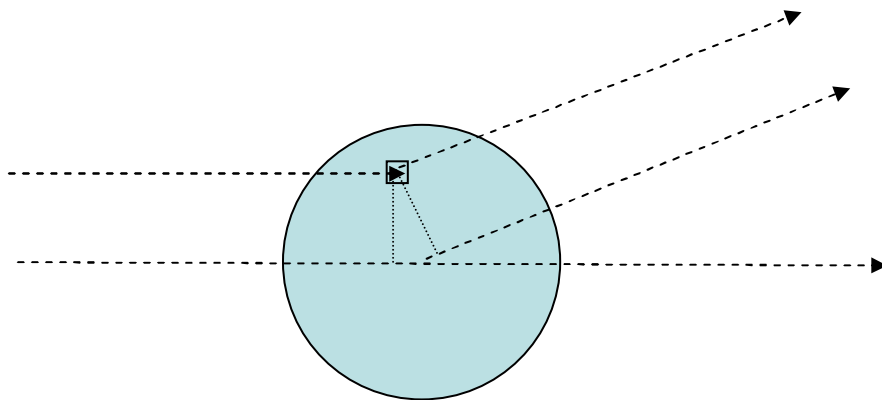
# X-ray diffraction by crystals (3D periodic structures)

## Atomic factor

$$f(\vec{s}) = \frac{\text{Amplitude scattered by one atom}}{\text{Amplitude scattered by one electron}} = \int \rho(\vec{r}) \cdot e^{-2\pi i \vec{s} \cdot \vec{r}} d\vec{r}$$

Atomic factor for  
Isotropic atoms:

$$f(s) = \int 4\pi \cdot r^2 \rho(r) \frac{\sin 2\pi sr}{2\pi sr} dr$$



## Structure factor

$$F(\vec{s} = \vec{r}_{hkl}^*) = \frac{\text{Amplitude .scattered .by.an.unit.cell}}{\text{Amplitude .scattered .by.one.electron}} = \int \rho(\vec{r}).e^{-2\pi i \vec{r}_{hkl}^* \cdot \vec{r}} d\vec{r}$$

$$F(\vec{s}) = \sum_i f_i . e^{-2\pi .i.(\vec{s} \cdot \vec{r}_i)} \quad \text{fi= atomic scattering factor}$$

$$F(\vec{s} = \vec{r}_{hkl}^*) = F_{hkl} = \sum_i f_i e^{-2.\pi.i.(h.x'_i + k.y'_i + l.z'_i)}$$

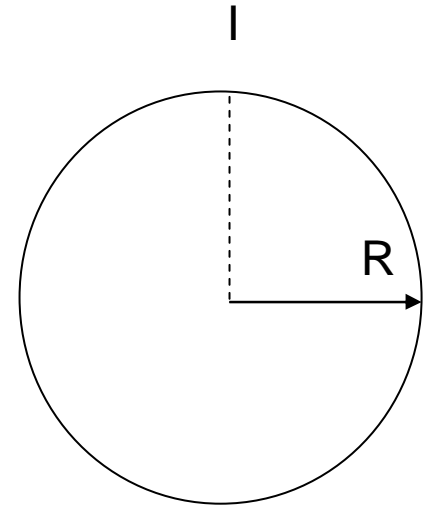
$$[\rho(\vec{r})]_{\text{unit.cell}} = \sum \sum \sum F_{hkl} e^{2\pi .i.\vec{r}_{hkl}^* \cdot \vec{r}}$$

**(F<sub>hkl</sub> are complex numbers)**

$$A_{hkl} = A_e \sum \sum \sum F_{hkl} \cdot e^{i\varphi_{uvw}} = A_e N \cdot F_{hkl}$$

$$\varphi(uvw, hkl) = 2\pi \cdot \vec{r}_{hkl}^* \cdot \vec{r}_{uvw} = 2\pi \cdot n$$

$$A_{hkl} = A_e \sum \sum \sum F_{hkl} \cdot e^{2\pi i (\vec{r}_{hkl}^* \cdot \vec{r}_{uvw})} = A_e N \cdot F_{hkl}$$



$$(I_N)_{hkl} = N^2 \cdot |F_{hkl}|^2$$

$$|F_{hkl}| = (1/N) \cdot \sqrt{I(\vec{r}^* = \vec{r}_{hkl}^*)}$$

$$F_{hkl} = \sum_i f_i e^{-2\pi \cdot i (h \cdot x'_i + k \cdot y'_i + l \cdot z'_i)}$$

$$[\rho(\vec{r})]_{unit\ cell} = \sum \sum \sum F_{hkl} e^{2\pi \cdot i \cdot \vec{r}_{hkl}^* \cdot \vec{r}}$$

$$F_{hkl} = |F_{hkl}| \cdot e^{-i \cdot \varphi_{hkl}} \quad \text{COMPLEX !!!}$$

The phase problem



$$(I_N)_{hkl} = N^2 \cdot |F_{hkl}|^2$$



$$|F_{hkl}| = (1/N) \cdot \sqrt{I(\vec{r}^* = \vec{r}_{hkl}^*)}$$

Experiment

$$F_{hkl} = \sum_i f_i e^{-2\pi \cdot i \cdot (h \cdot x'_i + k \cdot y'_i + l \cdot z'_i)}$$

$$[\rho(\vec{r})]_{unit.cell} = \sum \sum \sum F_{hkl} e^{2\pi \cdot i \cdot \vec{r}_{hkl}^* \cdot \vec{r}}$$

$$F_{hkl} = |F_{hkl}| \cdot e^{-i \cdot \phi_{hkl}} \quad \text{COMPLEX !!!}$$

$$(I_N)_{hkl} = N^2 \cdot |F_{hkl}|^2$$

Experiment

$$|F_{hkl}| = (1/N) \cdot \sqrt{I(\vec{r}^* = \vec{r}_{hkl}^*)}$$

$$F_{hkl} = \sum_i f_i e^{-2\pi \cdot i \cdot (h \cdot x'_i + k \cdot y'_i + l \cdot z'_i)}$$

Structure

## CRYSTALLOGRAPHIC LATTICES

- Point lattices
- Crystallographic directions [uvw]
- Crystallographic planes. Miller indexes (hkl)
- Reciprocal point lattices:
- Relationship between direct and reciprocal lattices

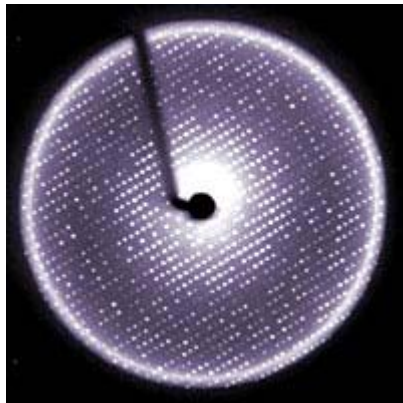
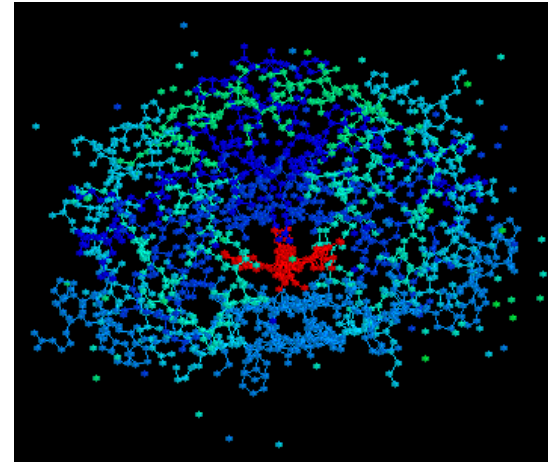
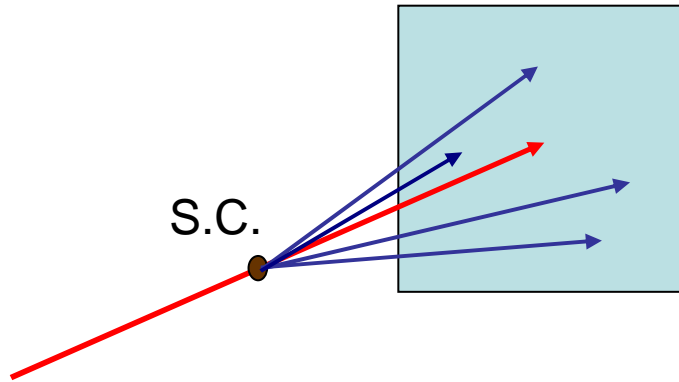
## SYMMETRY

- Crystal systems (6)
- Bravais lattices (14)
- Crystal classes. Point groups (32)
- Space groups (230)

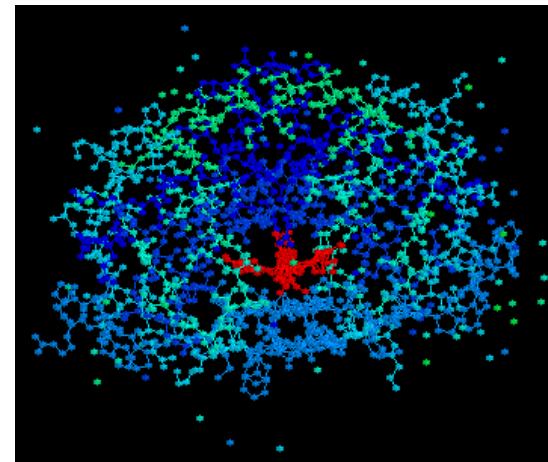
## X-RAY DIFFRACTION

- Ewald construction
- Thompson scattering. (Scattering amplitude by an electron) =  $A_e$
- Atomic scattering factor.  $f(s)$  (Scattering amplitude by an atom)/ $A_e$
- Structure factor:  $F_{hkl}$  (Scattering amplitude by an unit cells)/ $A_e$
  
- The phase problem

# Structure of crystallized proteins; Coordinates of atoms inside the unit cell

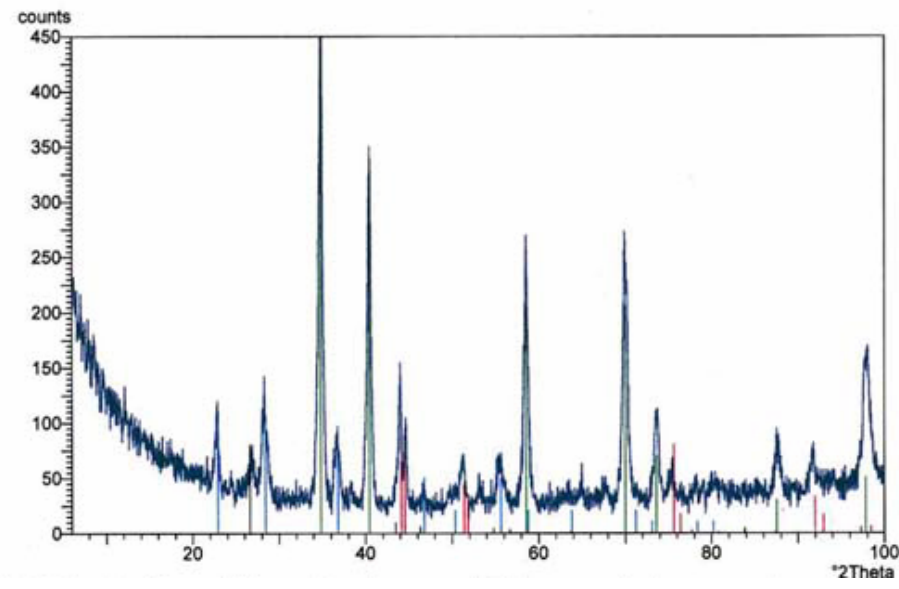
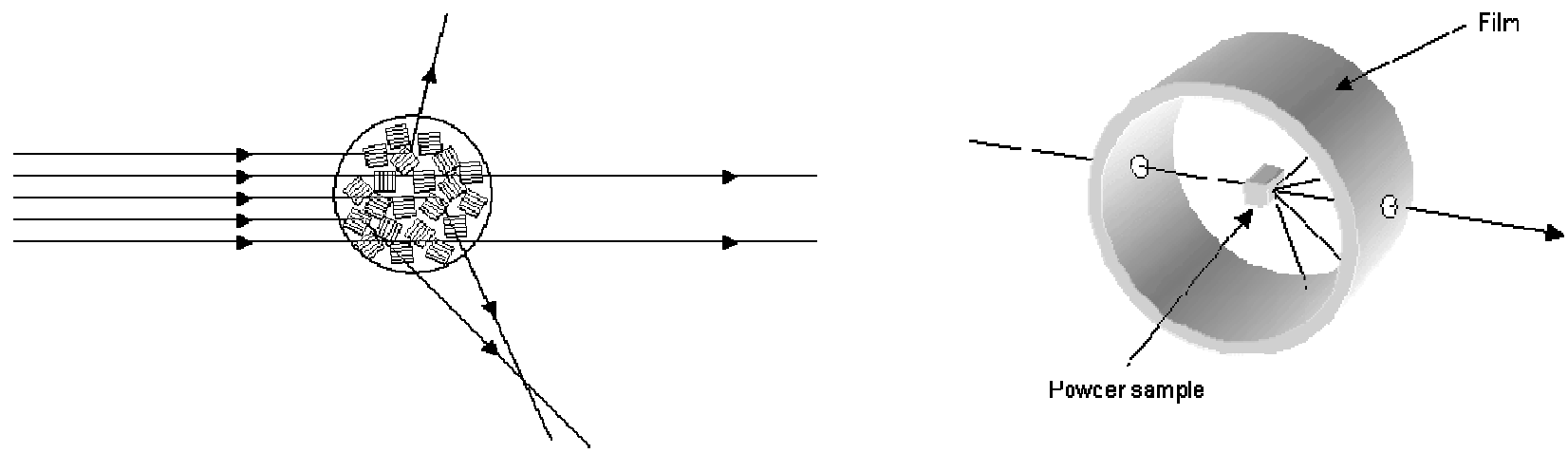


**X-ray diffraction patterns**



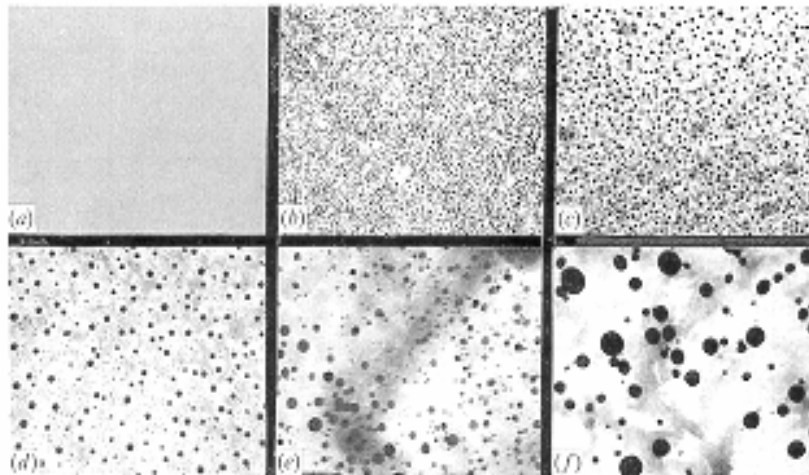
**HIV1**

Powder diffraction: Determination of lattice parameters, lattice thermal expansion, phase analysis, average crystal size and stress measurements, texture determination

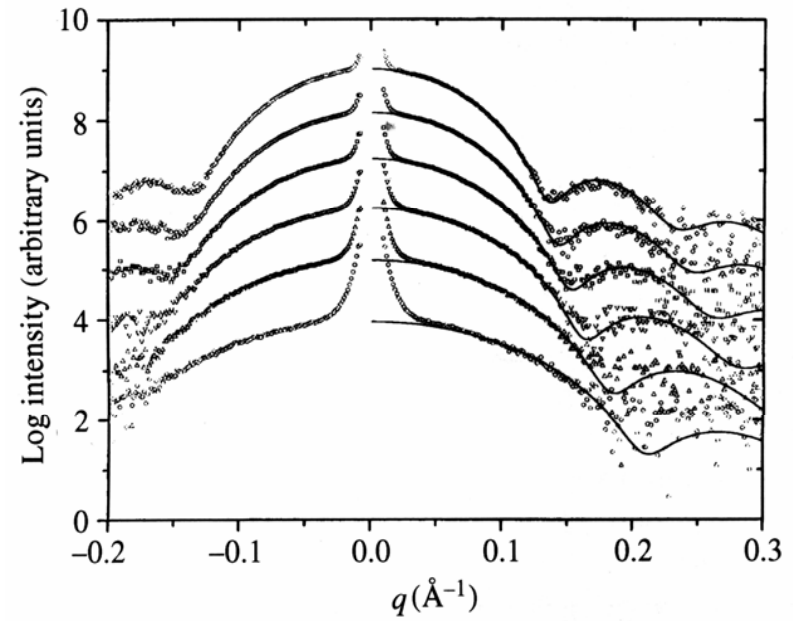




# Small – angle X-ray scattering (SAXS)

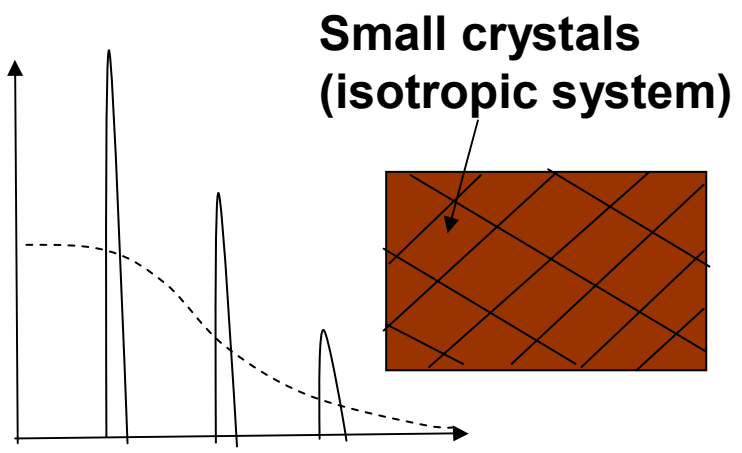
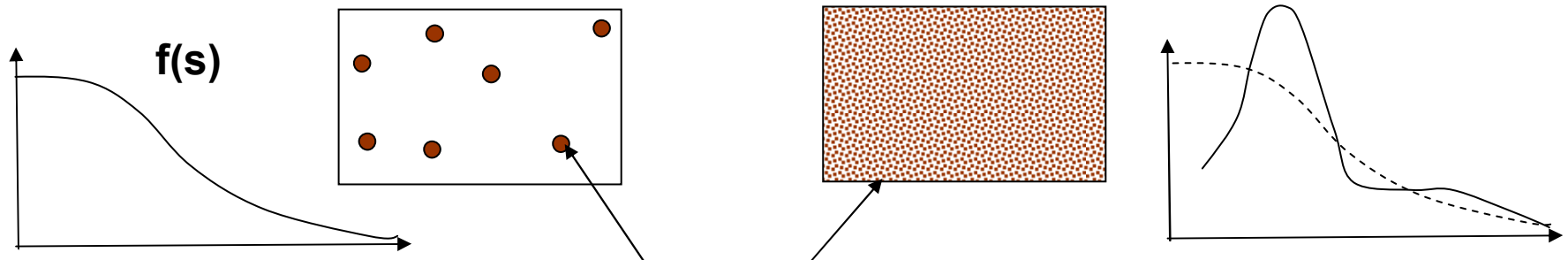


10 nm

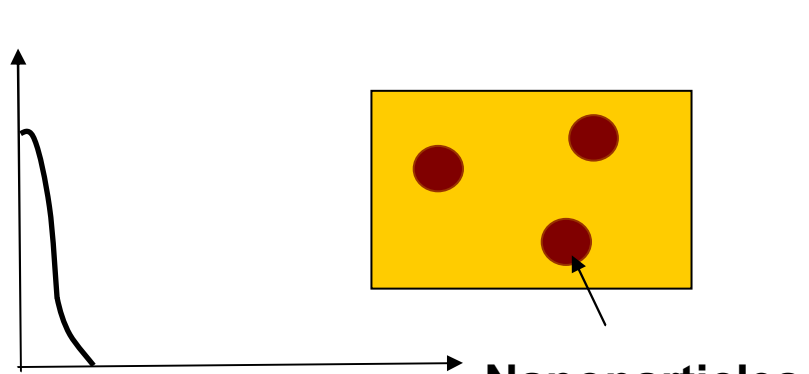
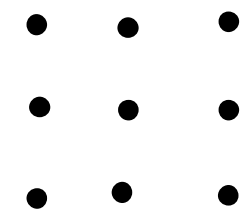
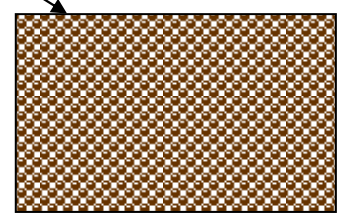


# Lectures and Tutorials on X-ray Scattering

- Single Crystal Crystallog. Basic Aspects (*A. Craievich*)
- Protein Crystallography. Basic Aspects (*D. Lamba*)
- Protein Crystallography. Applications (*M. Polentarutti*)
- Wide-angle scattering. Basic Aspects (*P. Scardi*)
- Powder X-ray Diffraction. Applications (*P. Scardi*)
- Powder X-ray Diffraction. *Tutorial*
- Small-angle X-ray Scattering (SAXS). Basic Aspects (*A. Craievich*).
- SAXS. Applications (*A. Craievich*).
- SAXS. Structure transformations (*A. Craievich*)
- SAXS. Nanomaterials and proteins in solution (*H. Amenitsch*)
- SAXS under extreme conditions (*H. Amenitsch*)
- SAXS. Tutorial (*H. Amenitsch*)
- Inelastic scattering (*G. Monaco*)
- Magnetic X-ray Scattering (*N. Binggeli*)



**Atoms**



**Nanoparticles**

**Proteins in solution**

**Magnetic atoms**

