



**The Abdus Salam
International Centre for Theoretical Physics**



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**Advanced School on Synchrotron and Free Electron Laser Sources
and their Multidisciplinary Applications**

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**Scanning Photoemission Microscopy:
Applications and Examples**

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Scanning Photoemission Microscopy: Applications and Examples

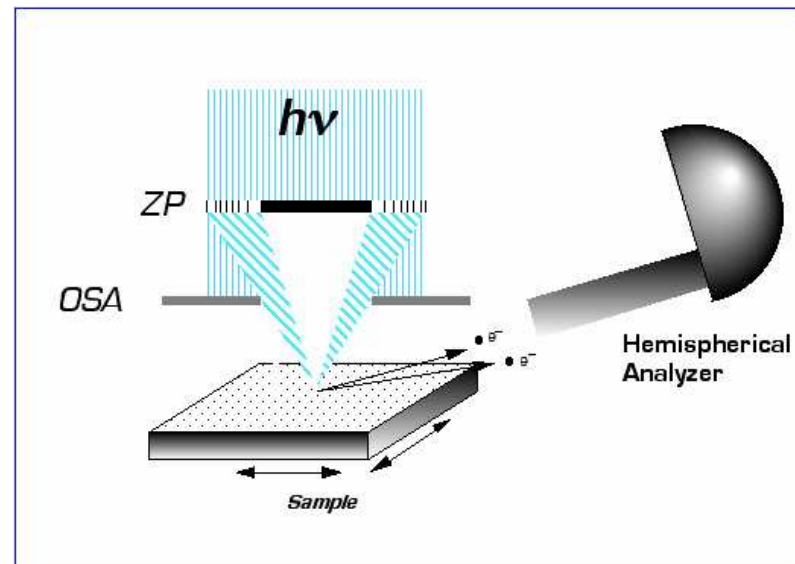
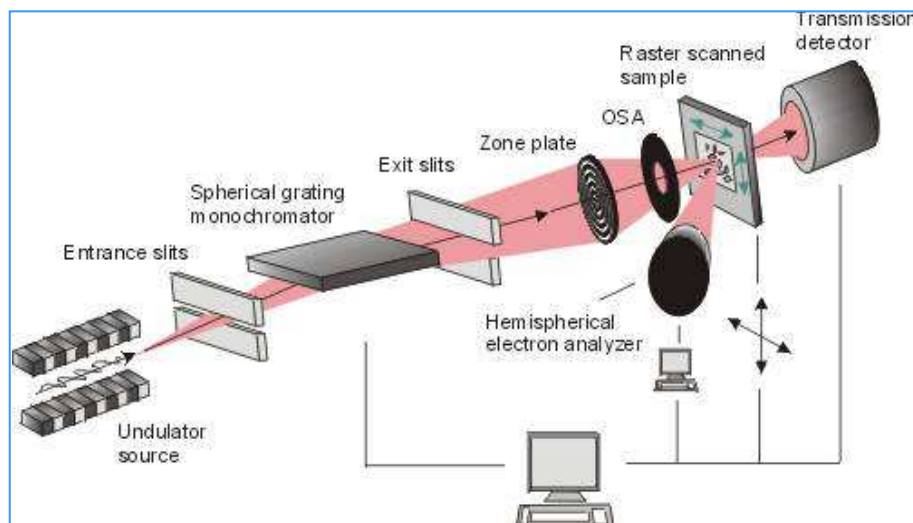
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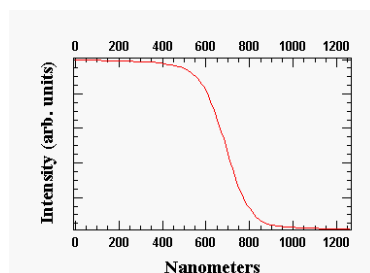
X-ray microscopy: method characteristics

Beamline Layout and SPEM setup



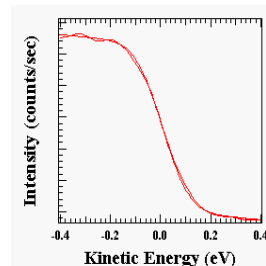
SPEM actual performances

Spatial resolution



- **Best resolution:**
~150nm (ZonePlates ltd)
- **Best transmission:**
8% (ZP ltd.)

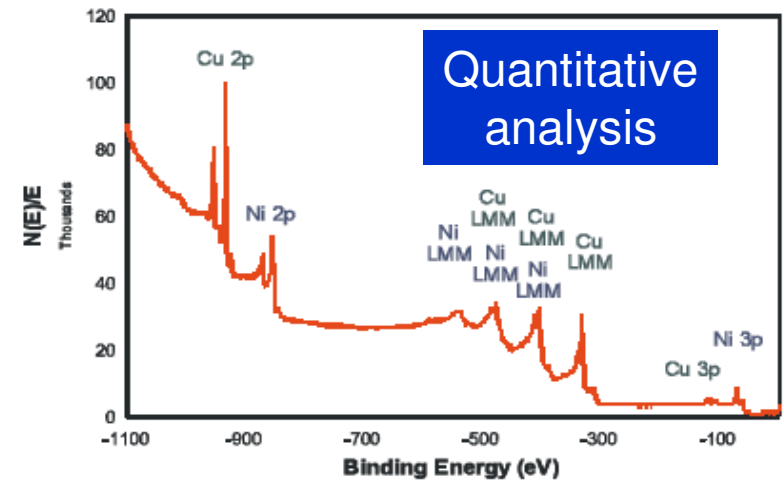
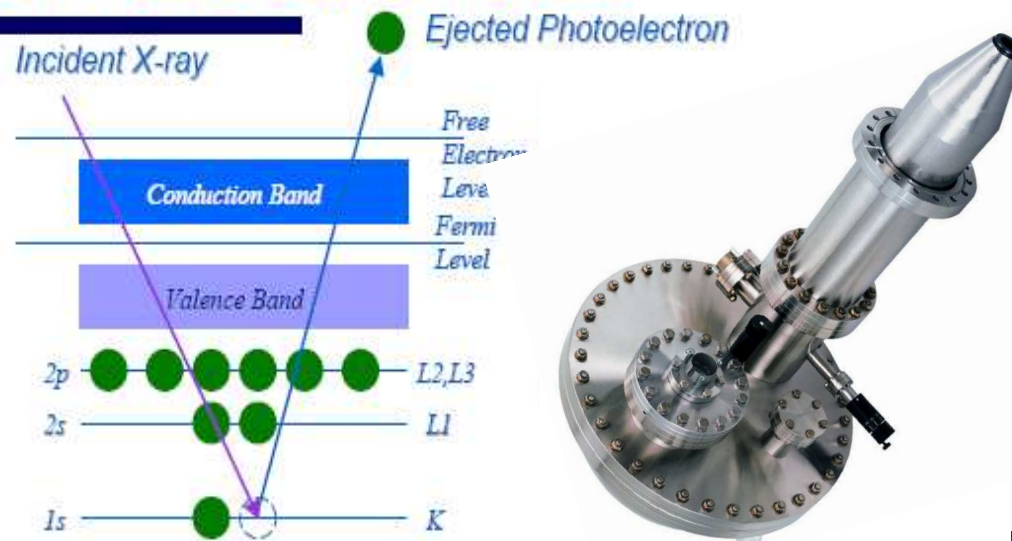
Overall energy resolution



- **Energy resolution:** ~180meV
- **Standard conditions**
- **Room Temperature**
- **Photon Energy:** 500 eV

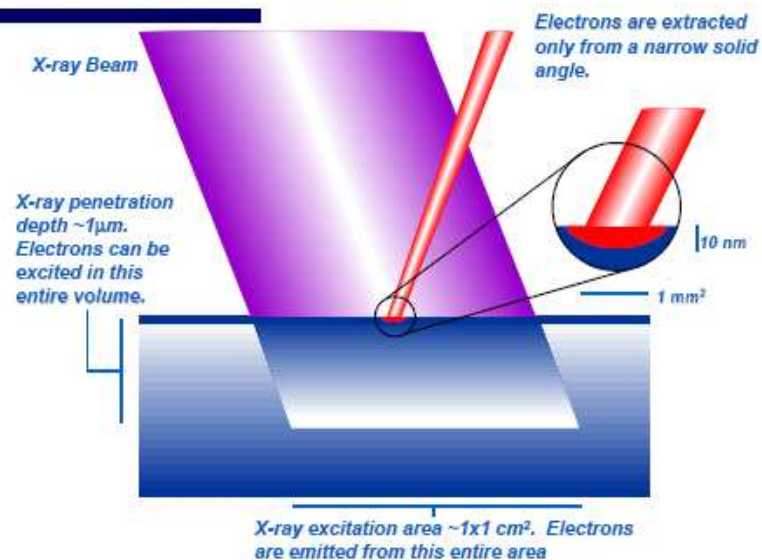
Photon energy range: 350 eV (min) – 900 eV (actual, undulator transmission)

The Photoelectric Process

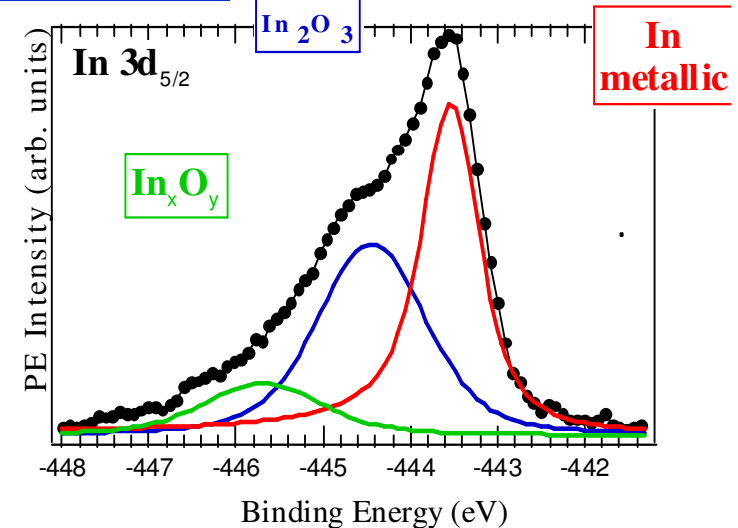


X-ray Photoelectron Spectroscopy

Small Area Detection



Fine Chemical analysis



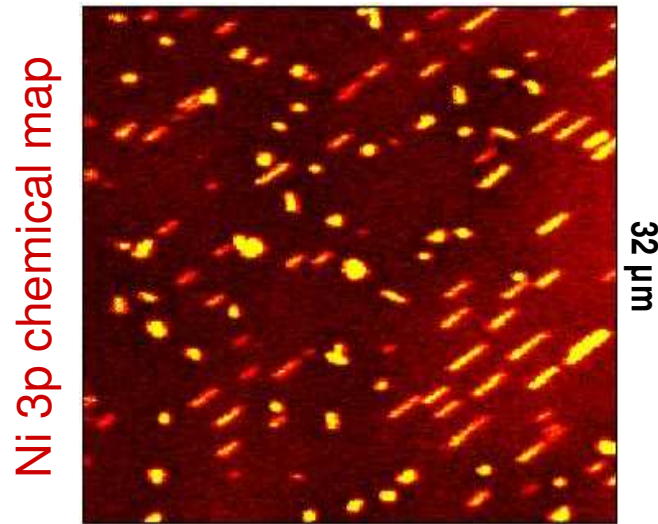
SPEM experiments: main topics

- Bulk-adlayer interfaces: metal/metal metal/semiconductor
 - Silicide formation (Ni, Co, Pd, Pt, Ag, Au)
 - Surface alloying and alloying (Ni/Pd, Au/Rh, Rh/Au)
- Catalysts&catalysis
 - Size gap
 - Model reactions (Rh, Pt, Ru)
- Nano and micro clusters properties
- Material characterization
- Organic and inorganic NT and nanostructures

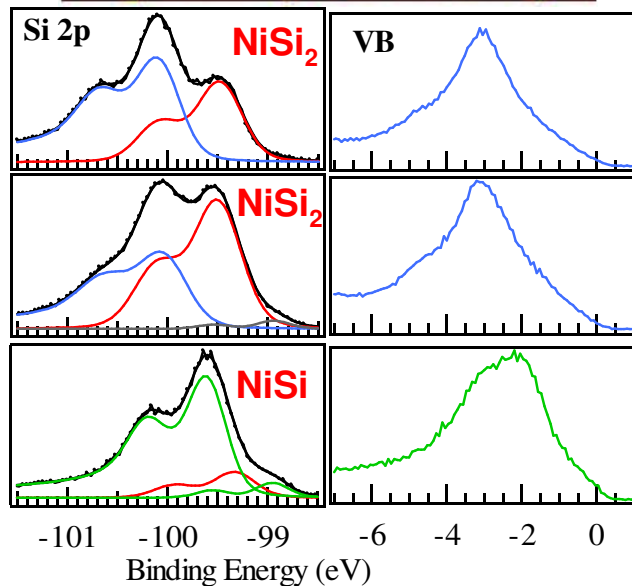
Ni silicides on Si(111): nucleation of 2D and 3D heterogeneous phases

- Deposition of 2ML of Ni on Si(111)-7x7 and thermal activation of the silicide formation

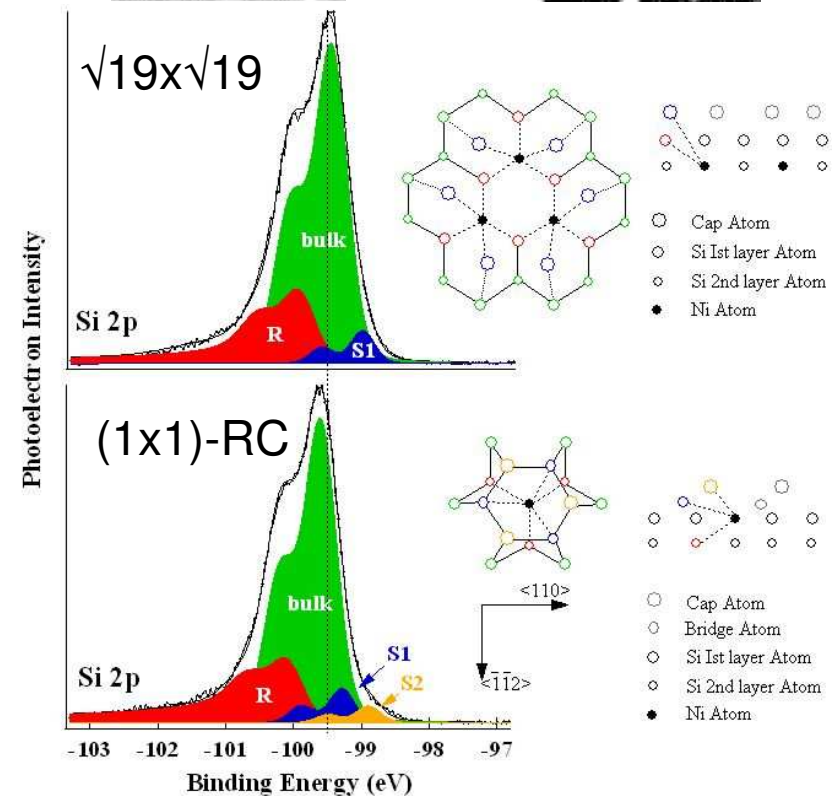
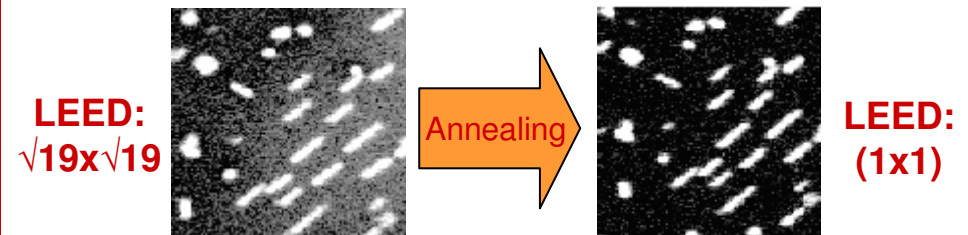
3D phases



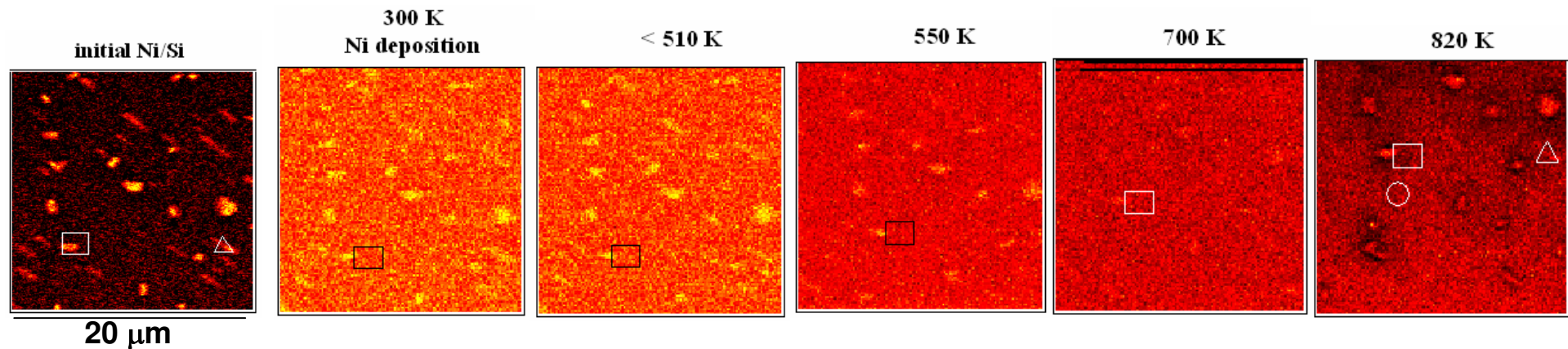
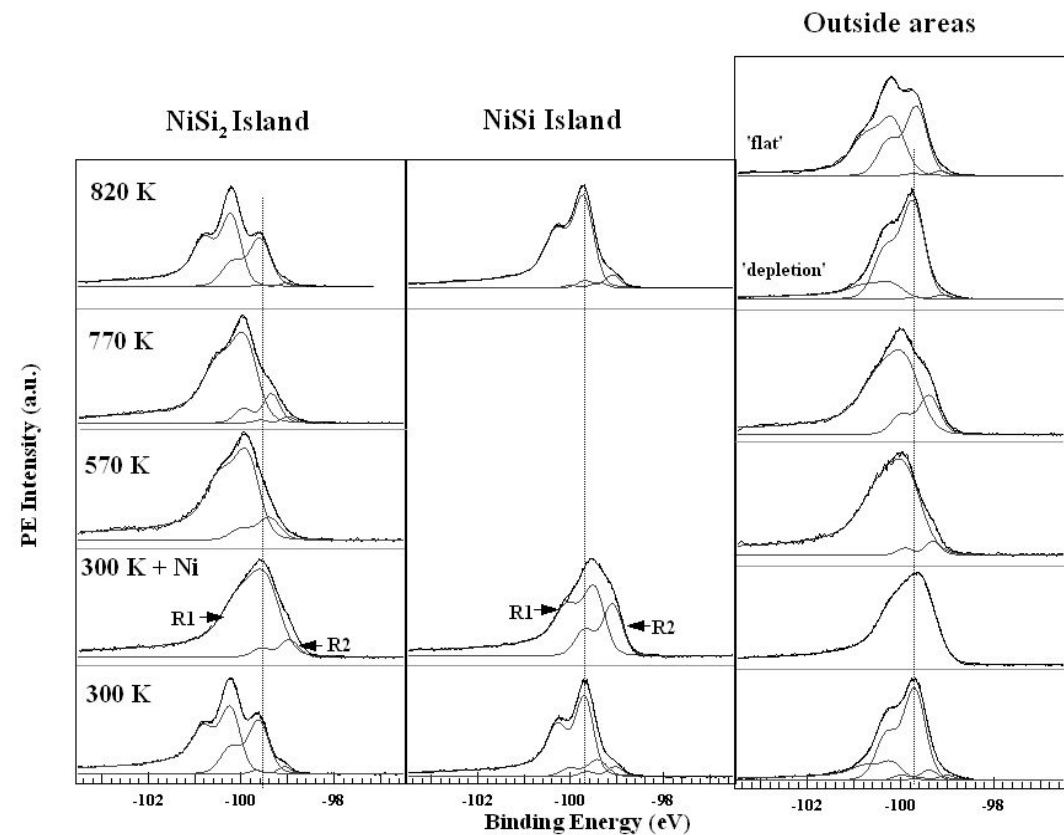
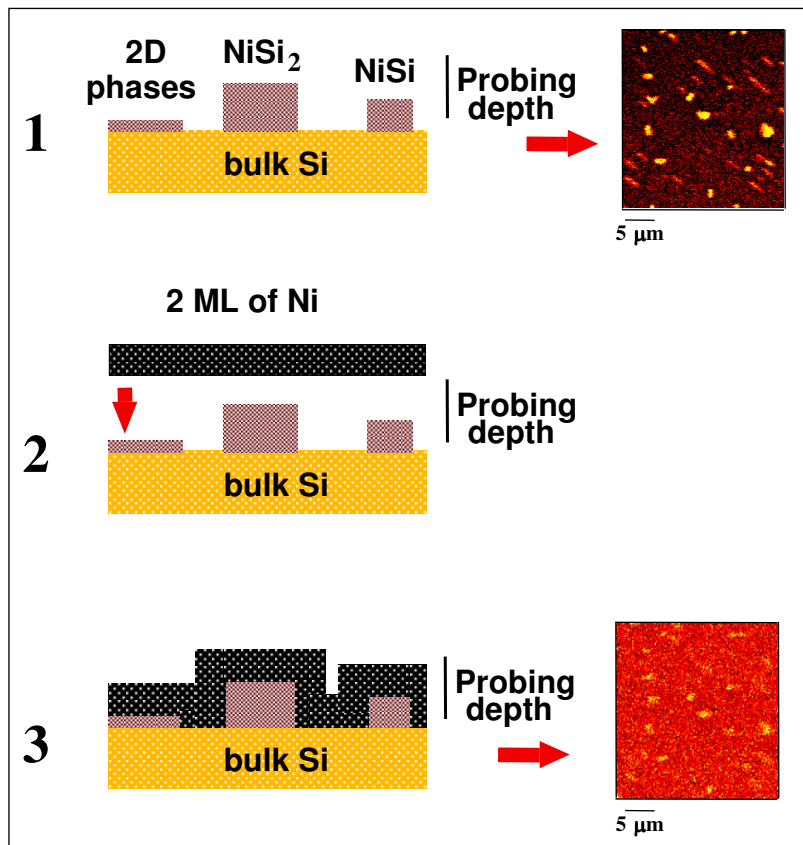
Si 2p and VB spectra
on single island



2D phases



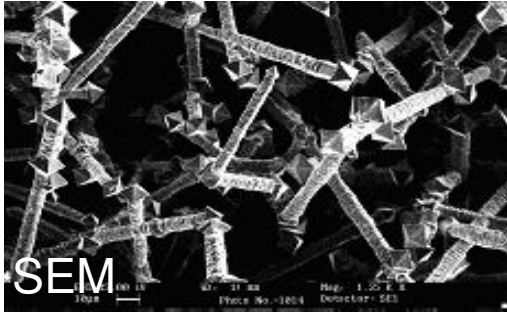
Ni silicides on Si(111): intermediate phases



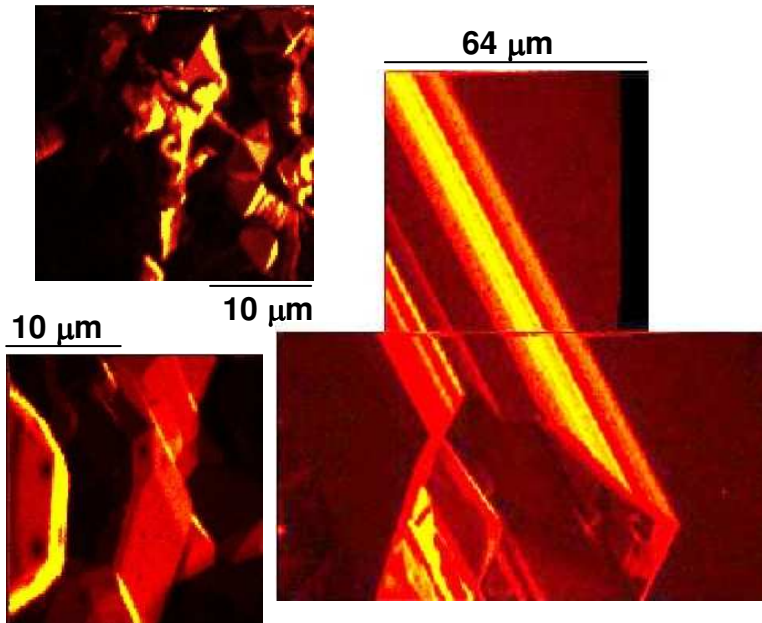
by Photoelectron Microscopy (in collaboration with A. Cremades UCM)

- Catalyst free growth of TCO structures ($\text{Sn}_x\text{O}_y/\text{In}_x\text{O}_y/\text{In}_x\text{N}_y, \text{Ge}_x\text{O}_y$, etc.)

D. Alina Magdas et al. APL 88, 113107 (2006)

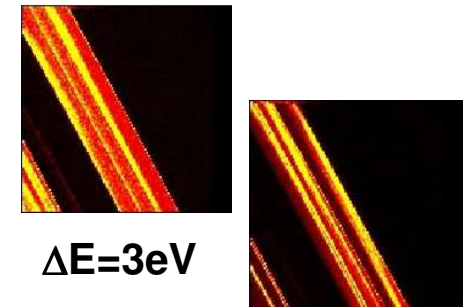
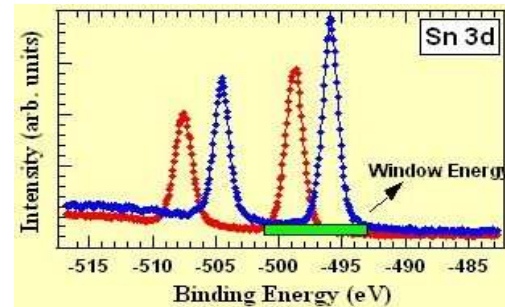


SPEM images



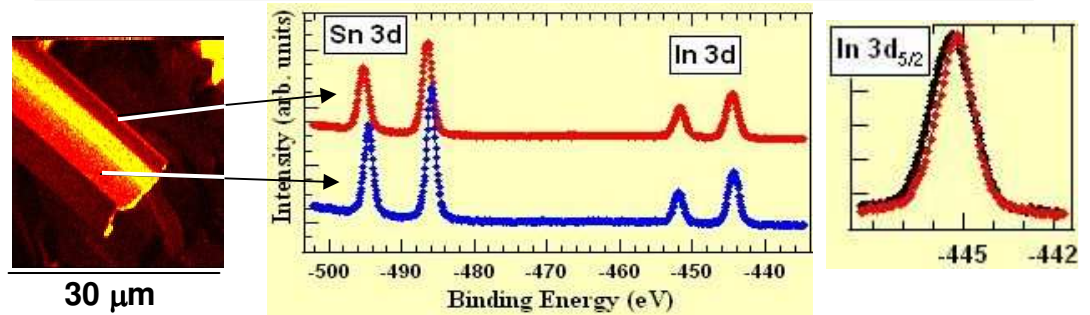
Electronic behaviour of a single structure

- Charging due to differences in the electronic structure
- Mapping of the charging with the multichannel acquisition



Local chemical composition of the structures

- **Heterogeneous elemental distribution locally defined**
- **Fine chemical analysis**



Degradation of light emitting diodes: a SPEM analysis

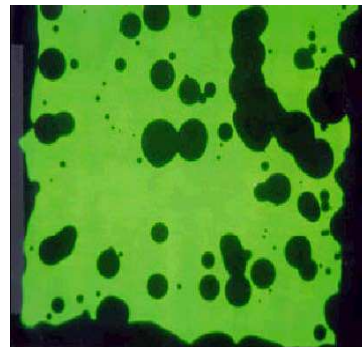
(in collaboration with P. Melpignano CRP, R. Zamboni CNR-ISMN)



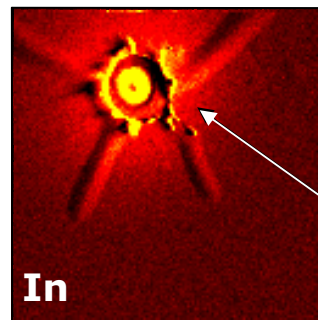
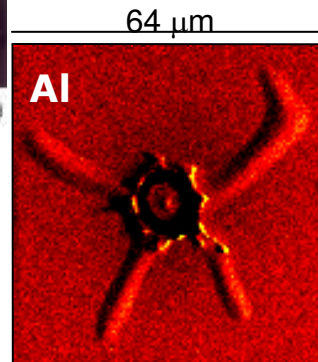
OLED Display Screen (from Universal Display Corp.)

**OLED exposed to atmospheric moisture:
failure due to light emission**

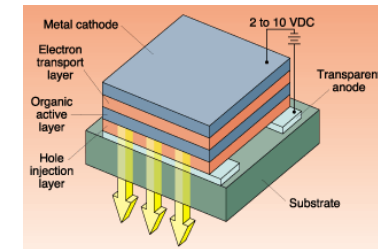
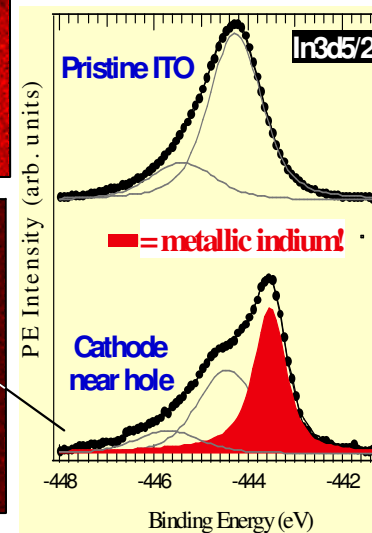
“Clean” experiment: OLED growth and operated in the SPEM (UHV ambient) : failure due to light emission



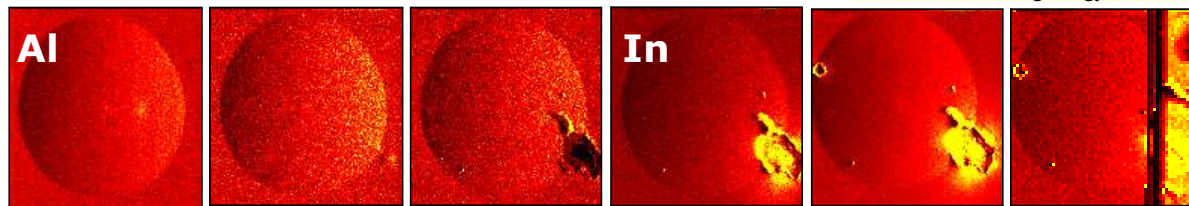
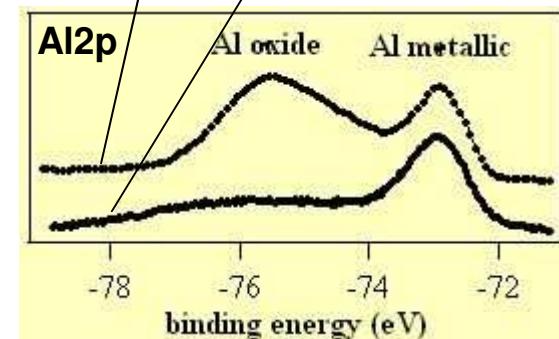
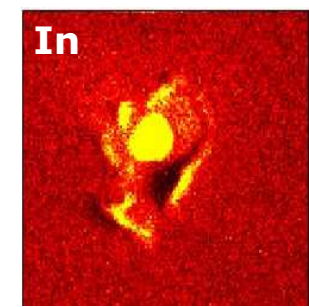
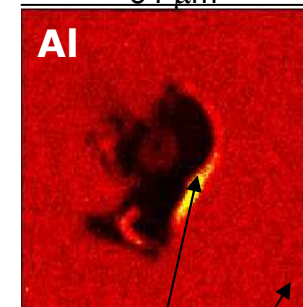
Dark spot in OLED



•Decomposition of ITO

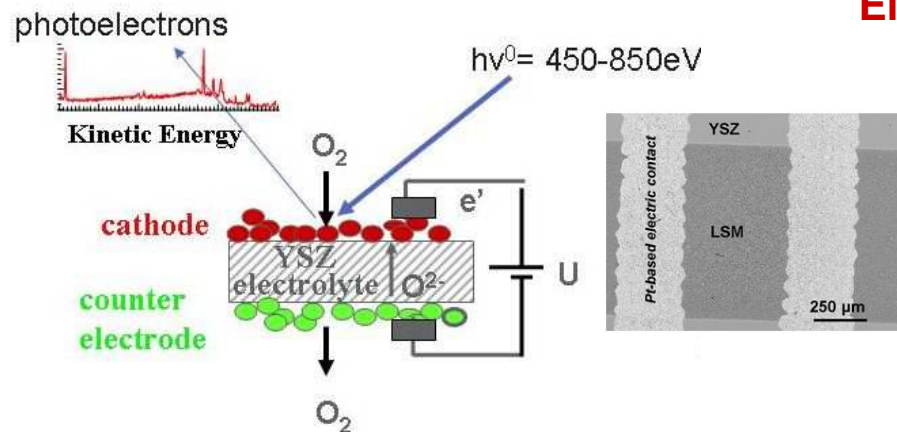


64 μ m



increasing voltage and operating time

Photoelectron microscopy study of operating LSM-YSZ cathode surfaces of SOFC *(in collaboration with M. Backhaus Corning Inc.)*

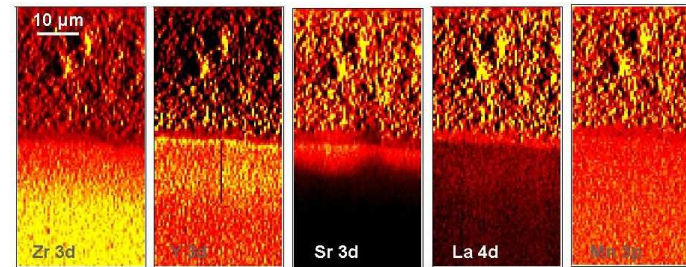


Strongly constraining experimental setup

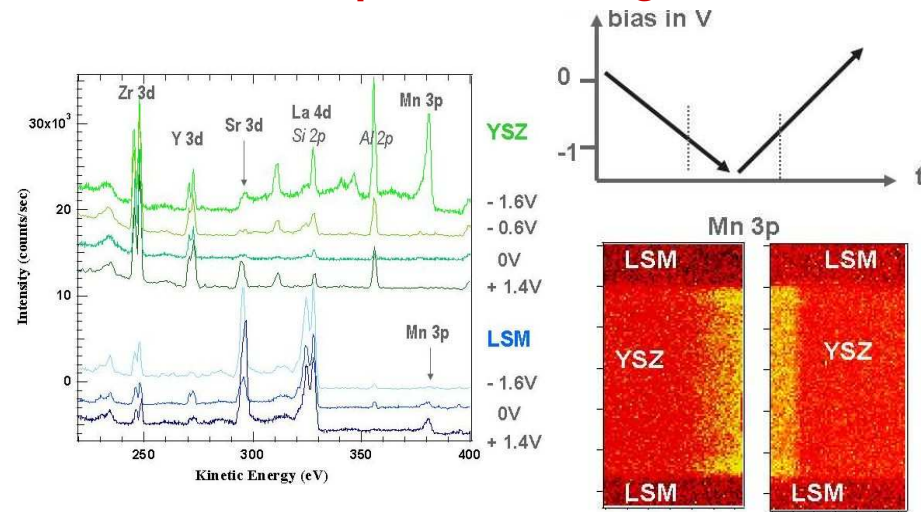


- Real samples
- High $T = 650-700^\circ\text{C}$
- $p\text{O}_2 = 1 \times 10^{-6}$ mbar
- Applied potentials $-2\text{V} < U < +2\text{V}$
- Surface sensitive technique
- High lateral resolution

Elemental distribution at electrolyte/LSM interface



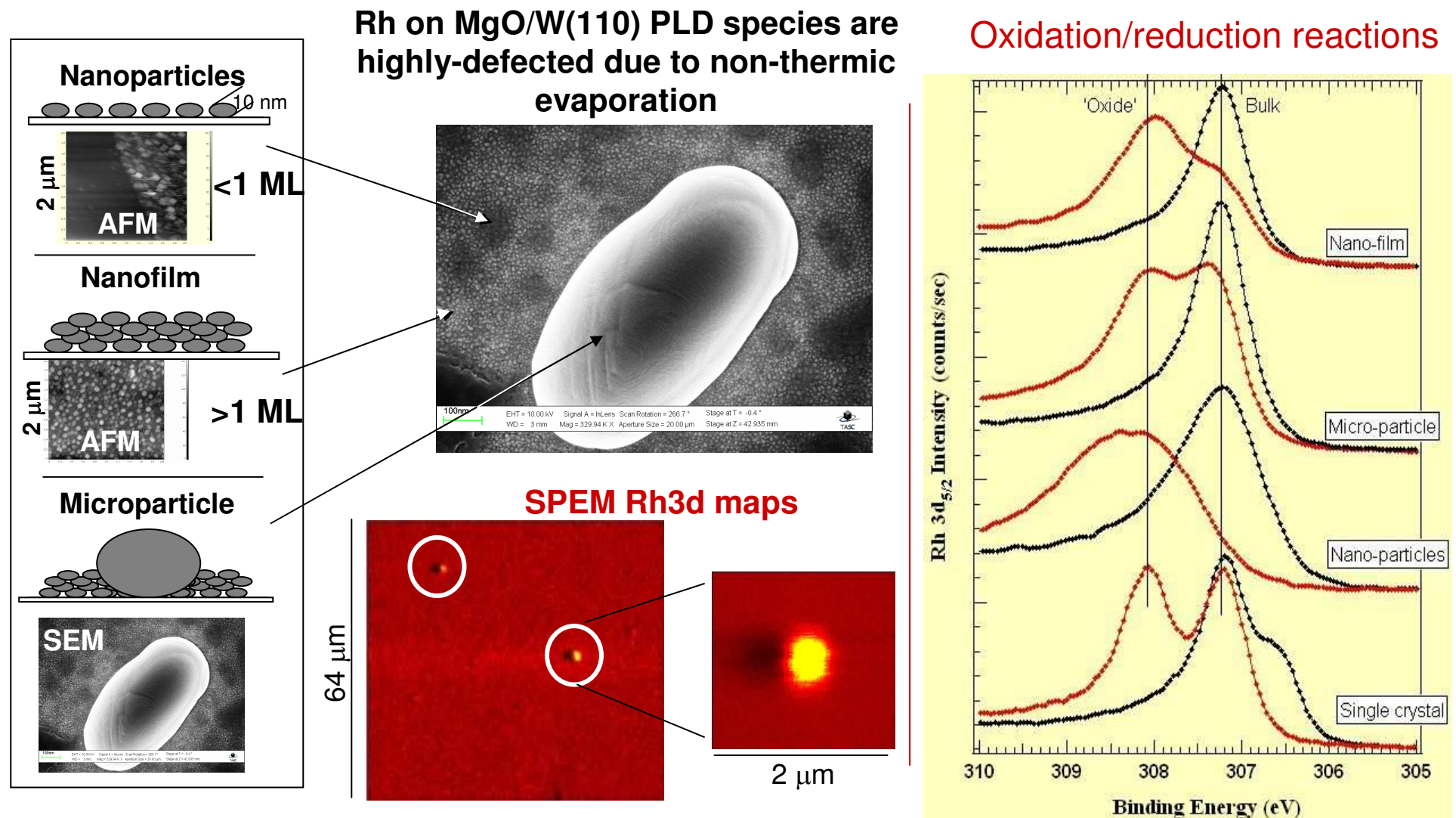
Surface composition change with bias



Observation and explanation of electrochemical cathode activation

- Strong current increase under negative bias when Mn spreads on electrolyte
- Mn^{2+} electrolyte surface enrichment $\rightarrow$ electrolyte surface conductivity $\rightarrow$ direct oxygen incorporation into electrolyte
- Oxygen incorporation extends under bias from TPB to the entire electrolyte surface

'Material' gap: from model crystalline materials to metal nanoparticles on metal oxide. In situ PLD particle deposition



- The nanoparticles/nanofilm possess different oxidation/reduction ability than the microparticle
- Reducing rate: Micro-part. > Nano-crystalline film > Nano-particles
- Micro-particles of similar sizes show variation in the reactive properties: different structure, local environ.

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