



**The Abdus Salam  
International Centre for Theoretical Physics**



**1938-2**

**Workshop on Nanoscience for Solar Energy Conversion**

***27 - 29 October 2008***

**Quantum Dot Solar Cells: Semiconductor Nanocrystals As Light Harvesters**

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Notre Dame IN 46556-0579  
USA*

# Meeting the Energy Demand



**14 TW Challenge**

# Quantum Dot Solar Cells. Semiconductor Nanocrystals as Light Harvesters

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Radiation Laboratory**

**Dept. of Chemical & Biomolecular Engineering  
University of Notre Dame, Notre Dame, Indiana 46556-0579**

**Support: US DOE (BES)**

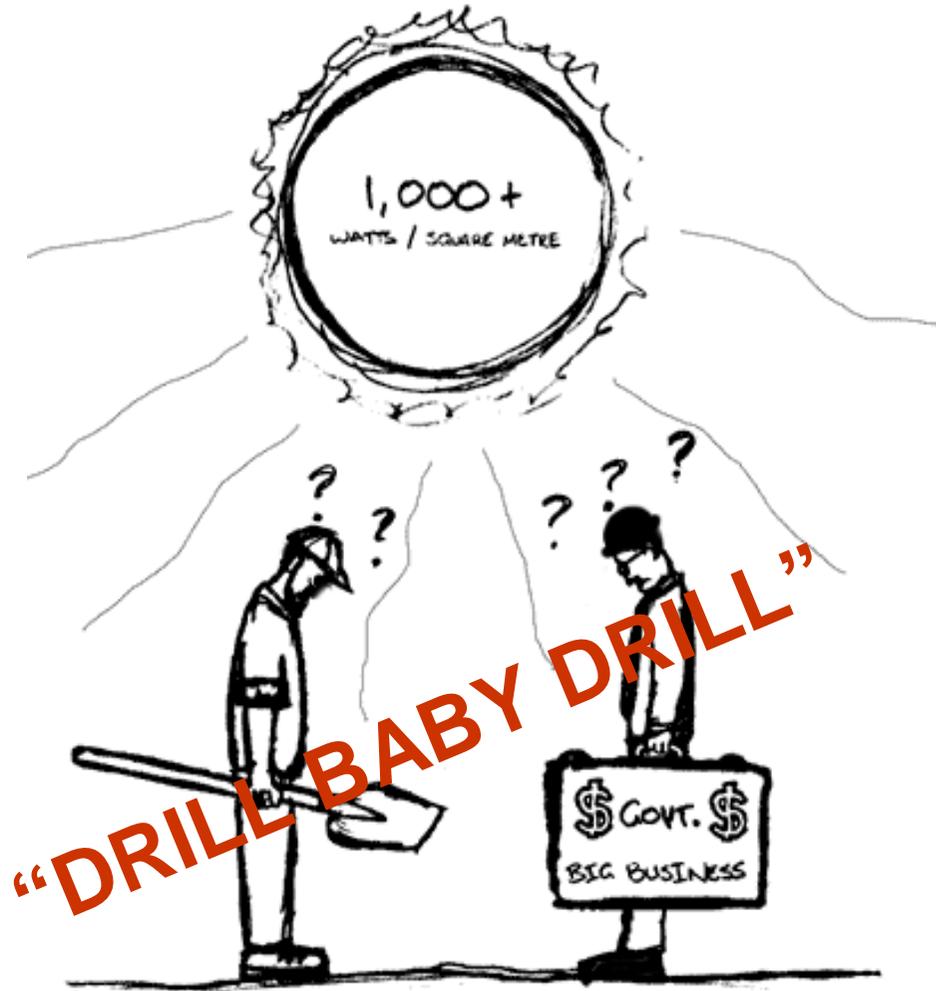
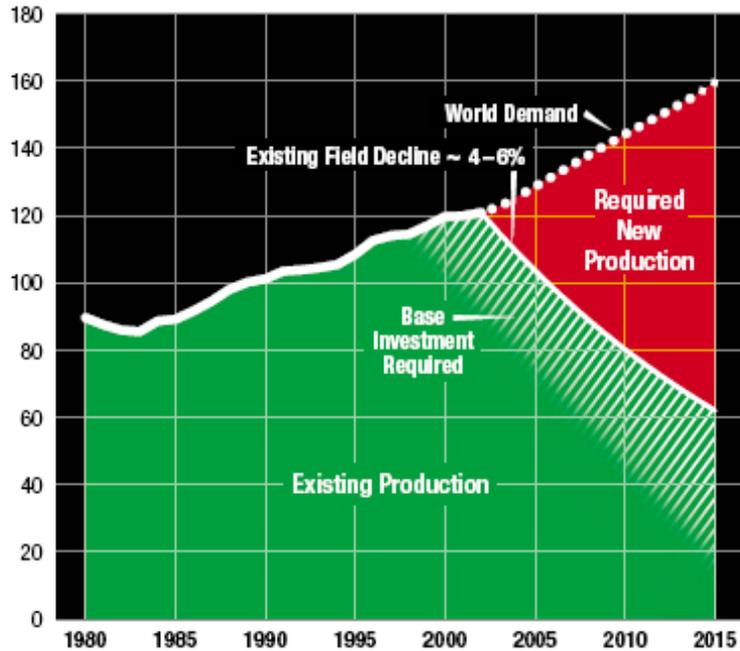
<http://www.nd.edu/~pkamat>



# Meeting the Energy Demand

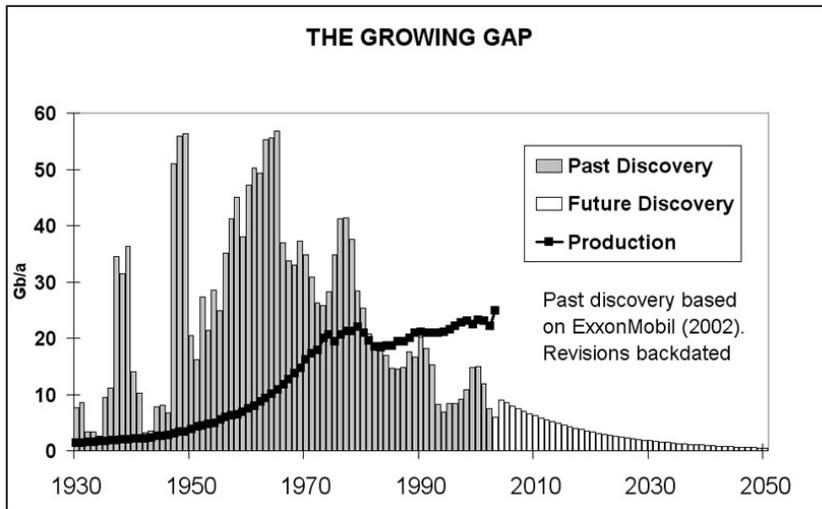
## Supplying Oil and Gas Demand Will Require Major Investment

Millions of Barrels per Day of Oil Equivalent (MBOE)



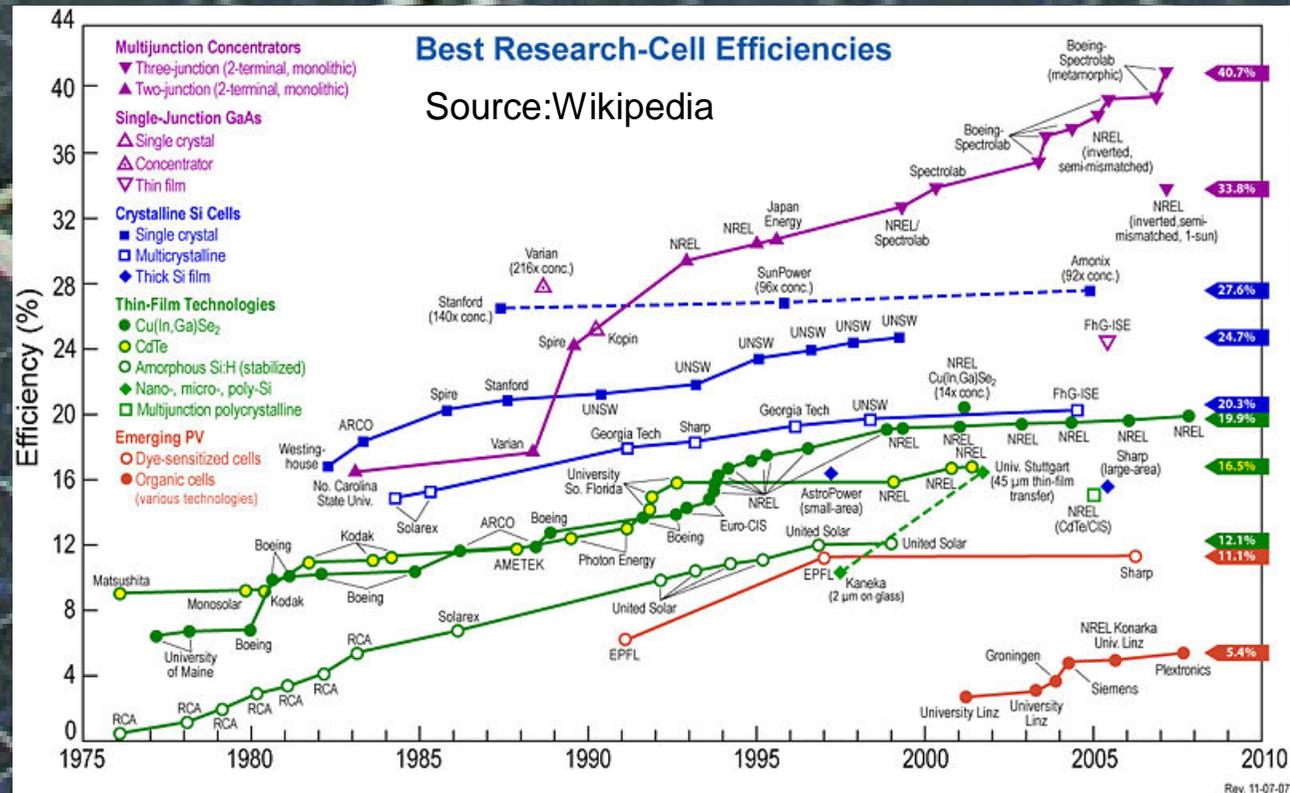
IF WE NEED A NEW ENERGY SOURCE FOR THE FUTURE  
I SUPPOSE WE COULD TRY DIGGING HERE!

[www.solarsales.com.au](http://www.solarsales.com.au)

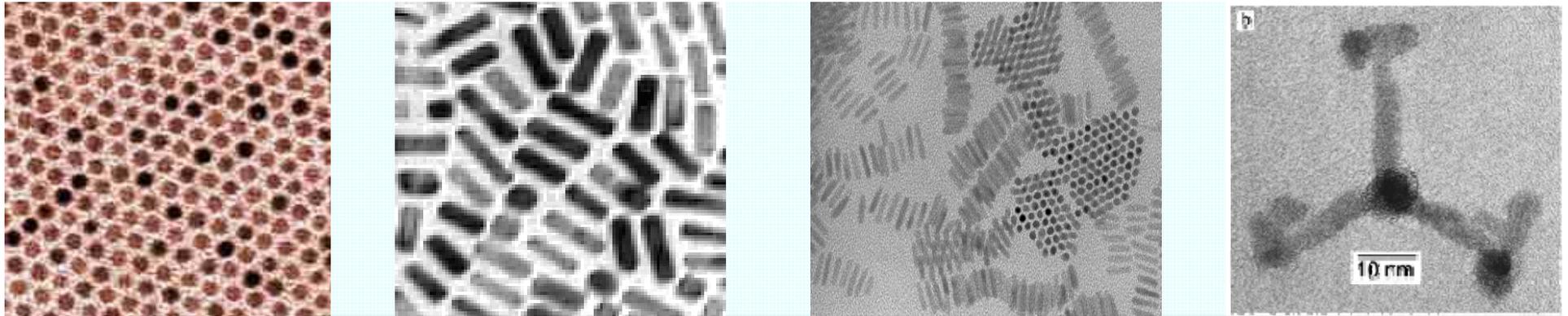


World Oil Production vs. Discovery  
Source: Dr. C.J. Campbell

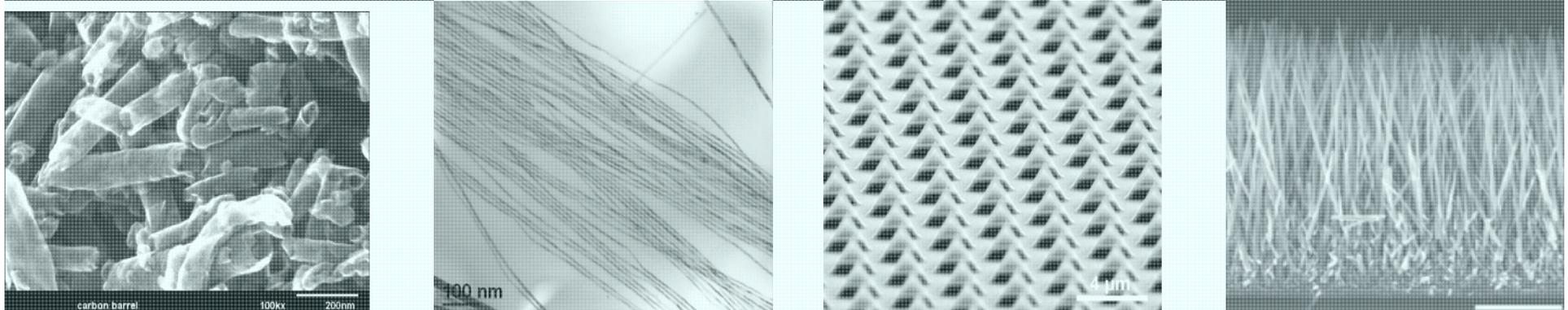
# Solar Revolution



What if all you needed to capture energy was a new coat of paint?



# Can we address the clean energy challenge with Nanotechnology?



# Our Research Focus

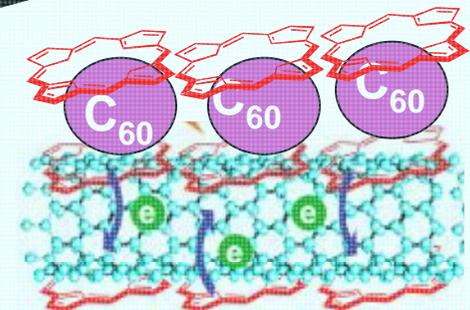
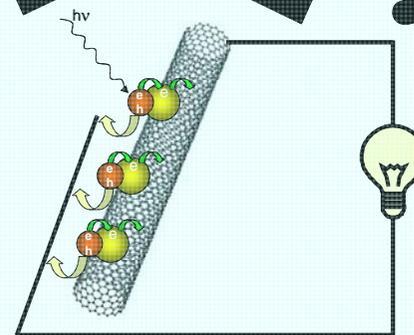
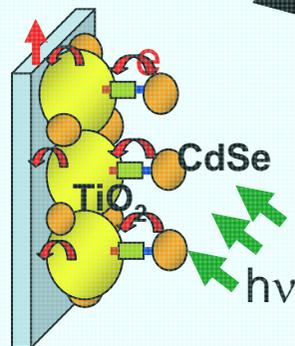
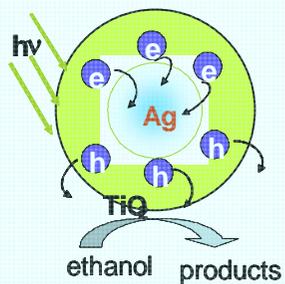
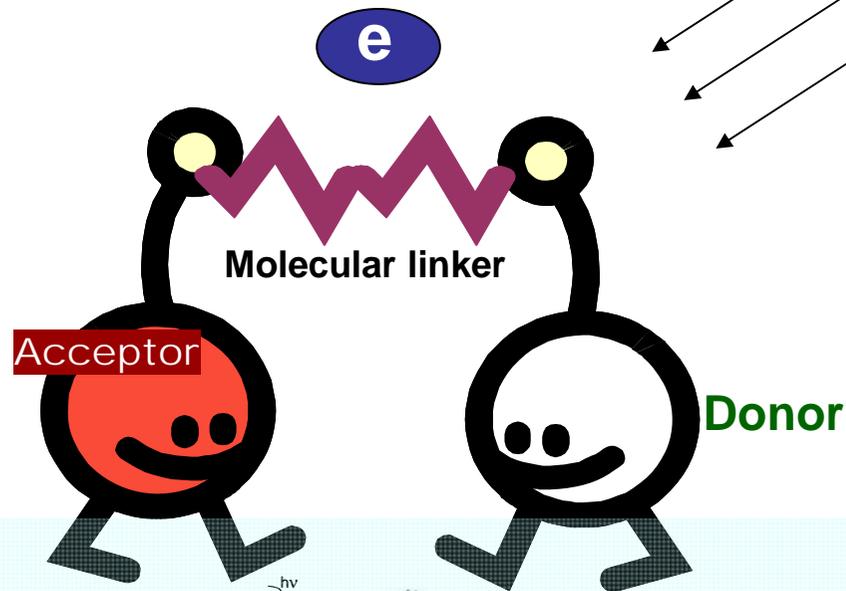
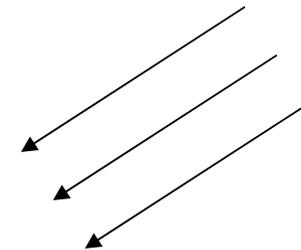
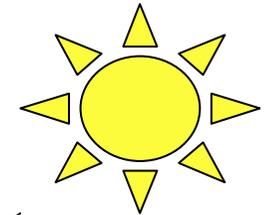
Photoinduced electron transfer in light harvesting systems

Donor- Acceptor

Semiconductor-Sensitizer

Semiconductor-Semiconductor

Semiconductor-Metal



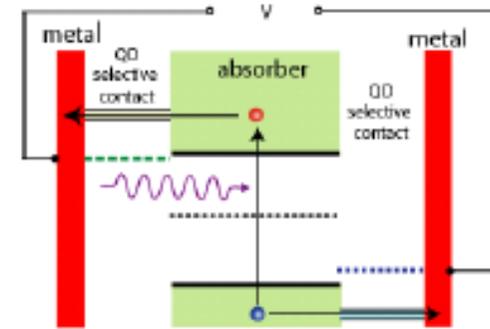
# Quantum Dot Solar Cells

Tunable band edge

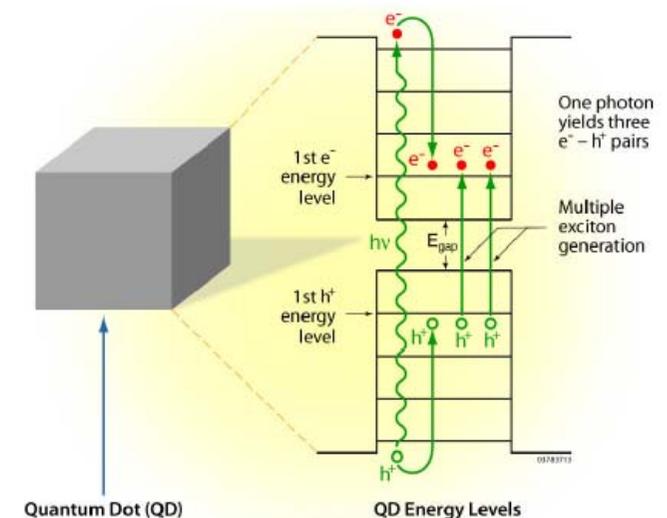
Offers the possibility to harvest light energy over a wide range of visible-ir light with selectivity

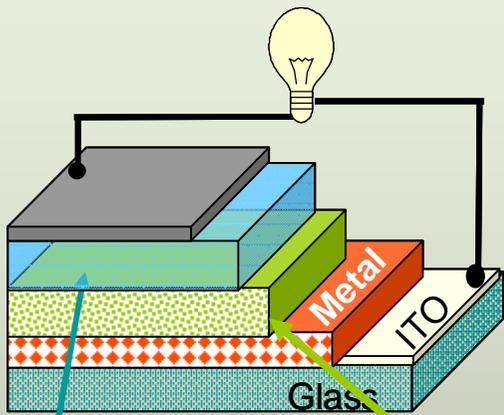


Hot carrier injection from higher excited state  
(minimizing energy loss during thermalization of excited state)



Multiple carrier generation solar cells.  
Utilization of high energy photon to multiple electron-hole pairs

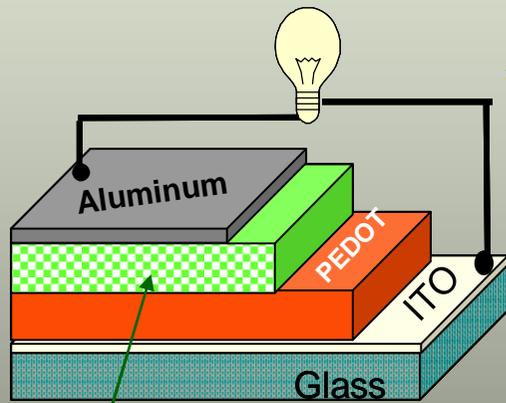




Redox or Hole Transport Layer

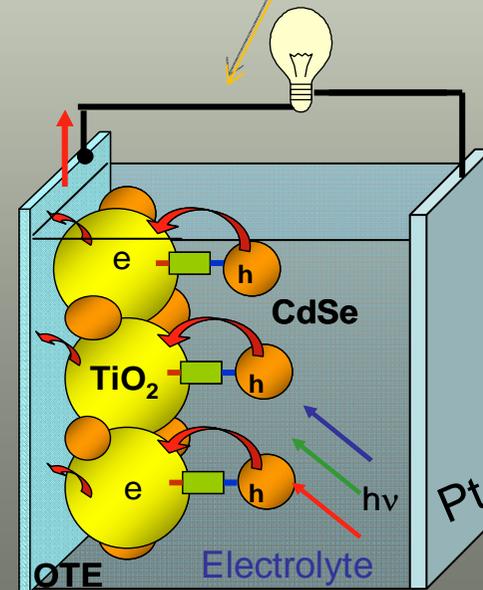
Semiconductor Nanocrystals

**Metal Junction Solar Cell**

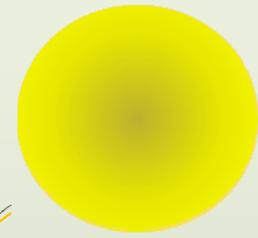


Active Polymer/Semiconductor Layer

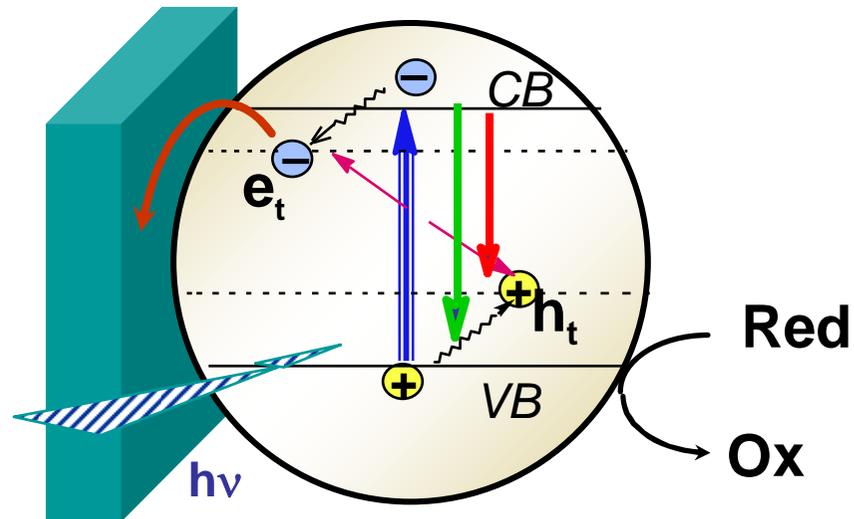
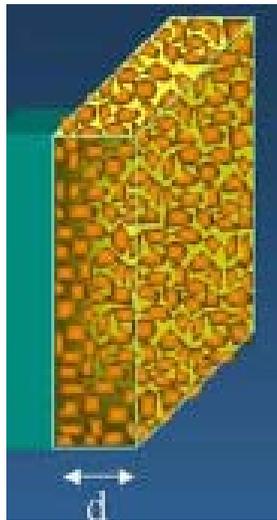
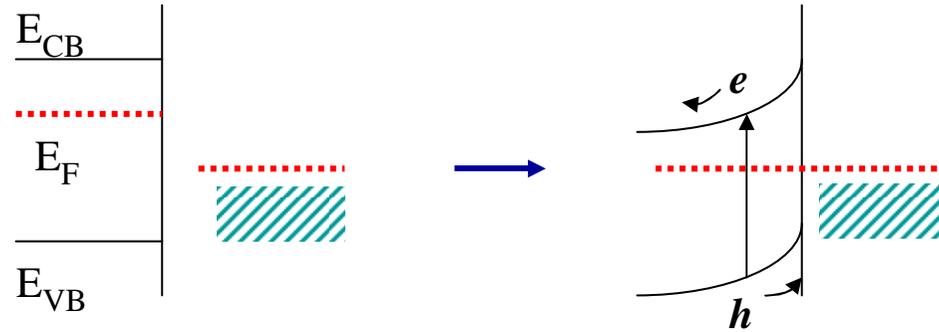
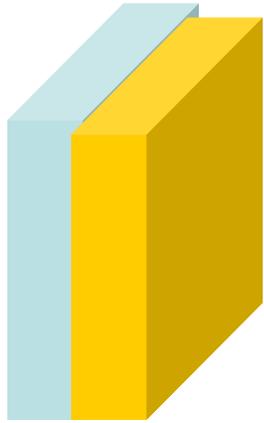
**Polymer Solar Cell**



**Quantum Dot Sensitized Solar Cell**

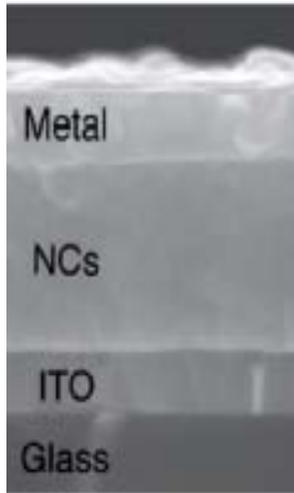


# 1. Semiconductor/metal Interface



Because of smaller dimensions charge separation and transport issues in nanostructure films need to be tackled

# Metal/PbSeNC/Metal Sandwich Photovoltaic Cell

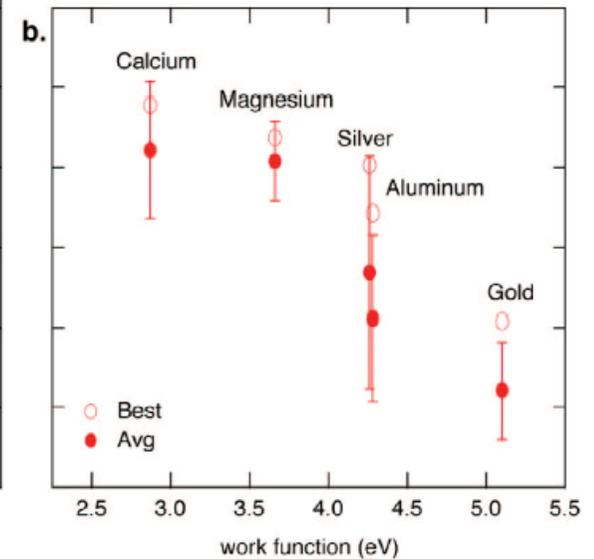
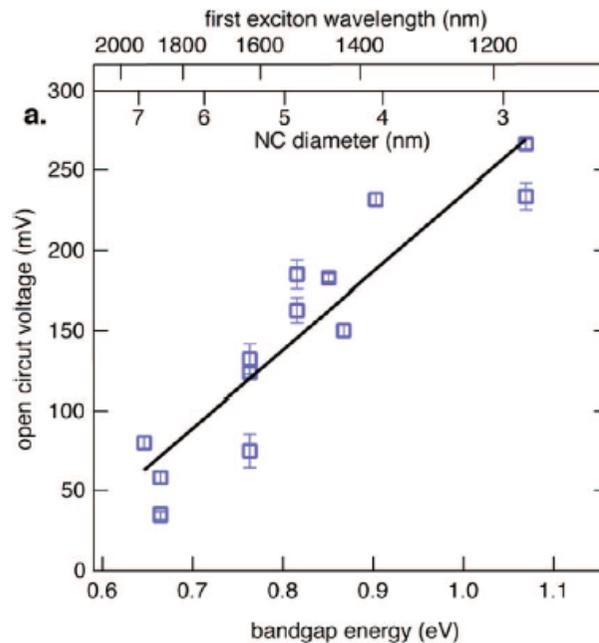
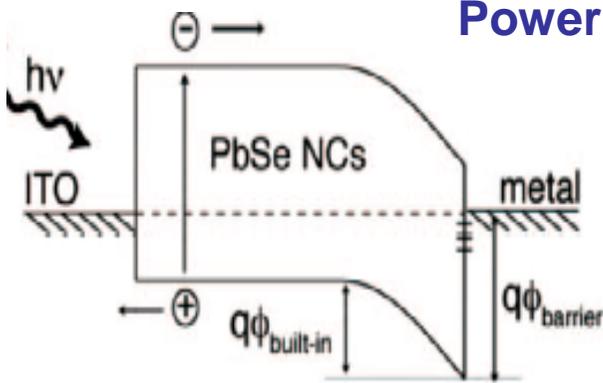
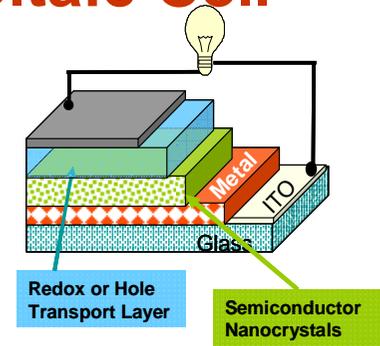


PbSe NC film, deposited *via* layer-by-layer dip coating

Short-circuit photocurrent ( $>21 \text{ mA cm}^{-2}$ ) by way of a Schottky junction

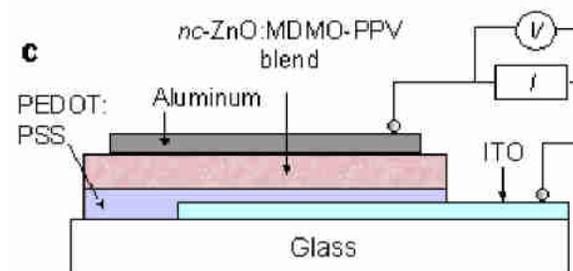
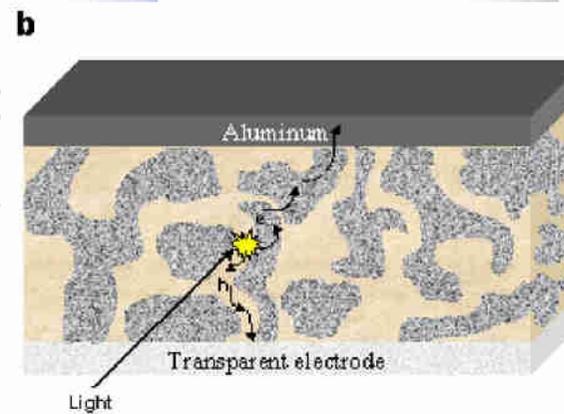
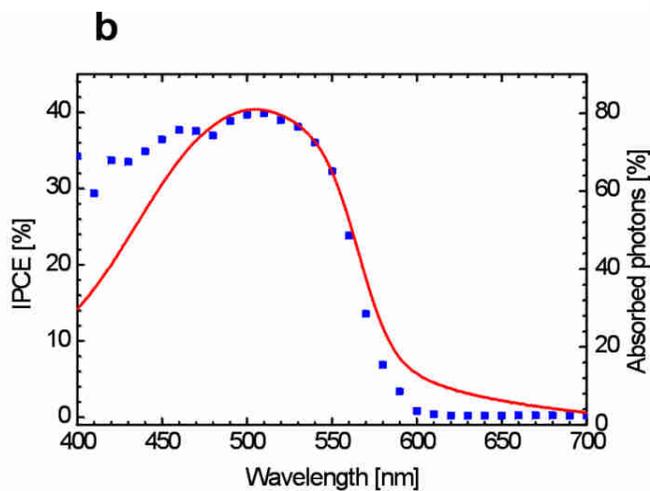
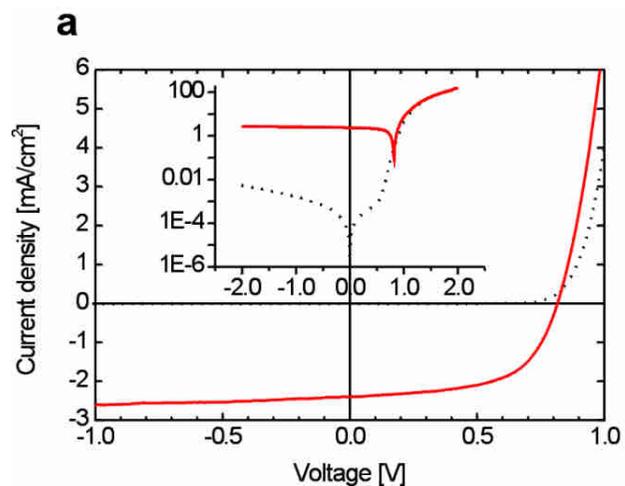
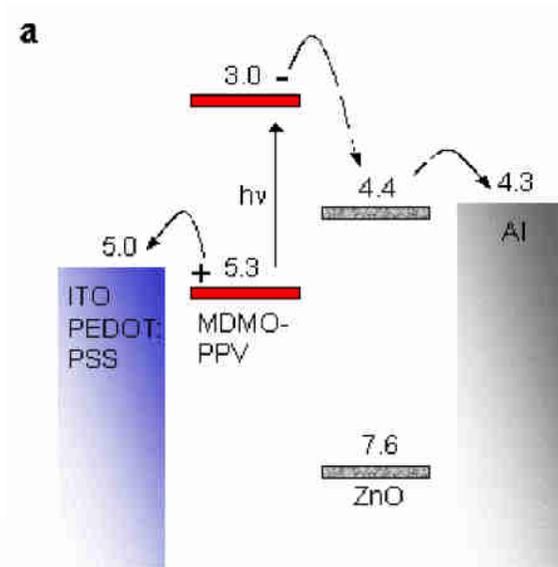
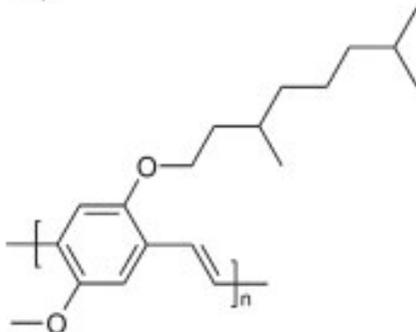
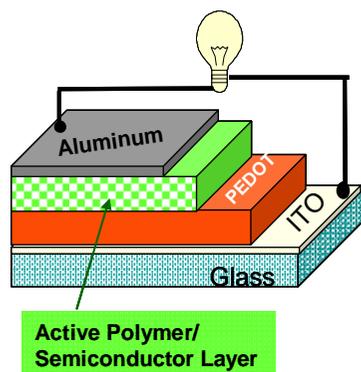
EQE of 55-65% in the visible and up to 25% in the infrared region

Power conversion efficiency of 2.1%.

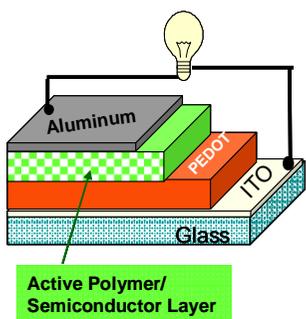


Luther et al *Nano Lett.*,  
Vol 8, 2008, 3488

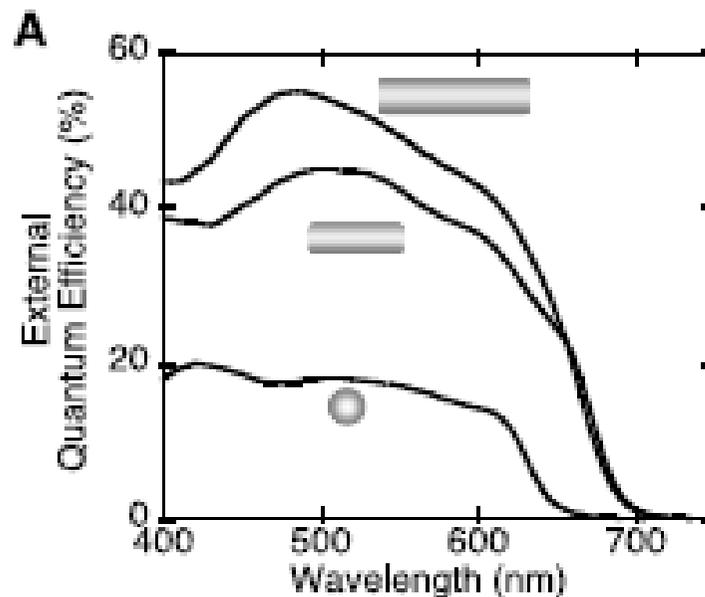
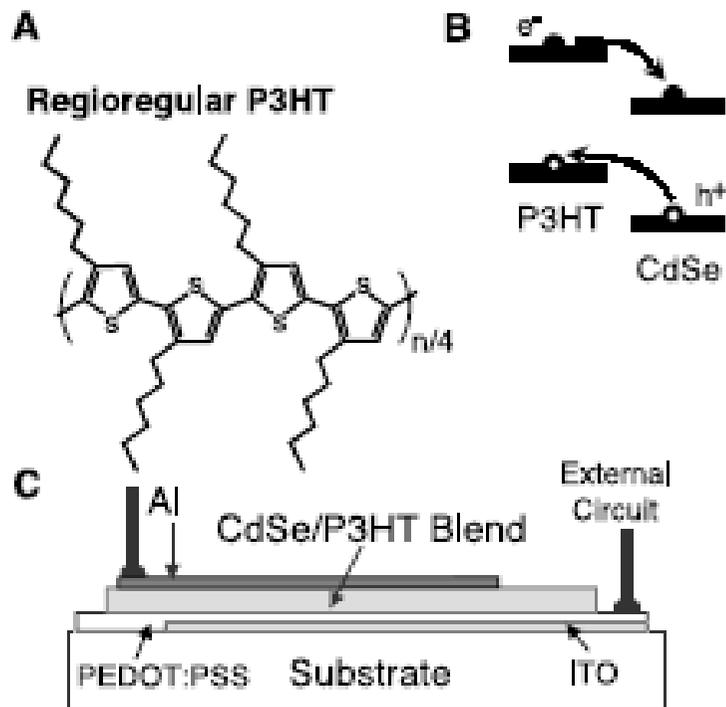
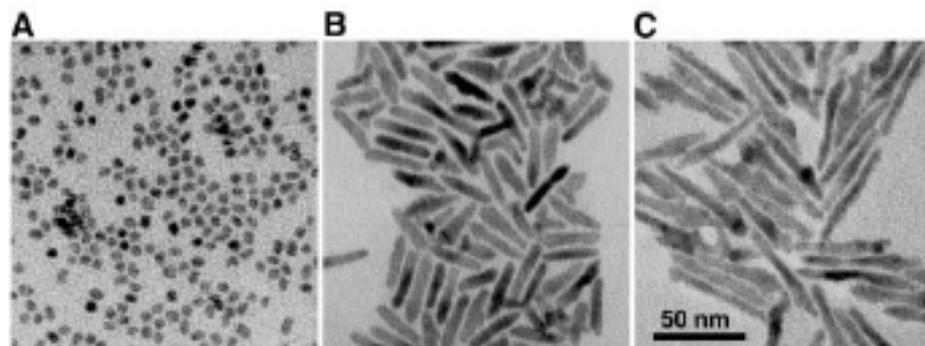
## 2. Polymer –Semiconductor Nanocrystal Hybrids



Adv. Mater, 2004, 16, 1009-1013

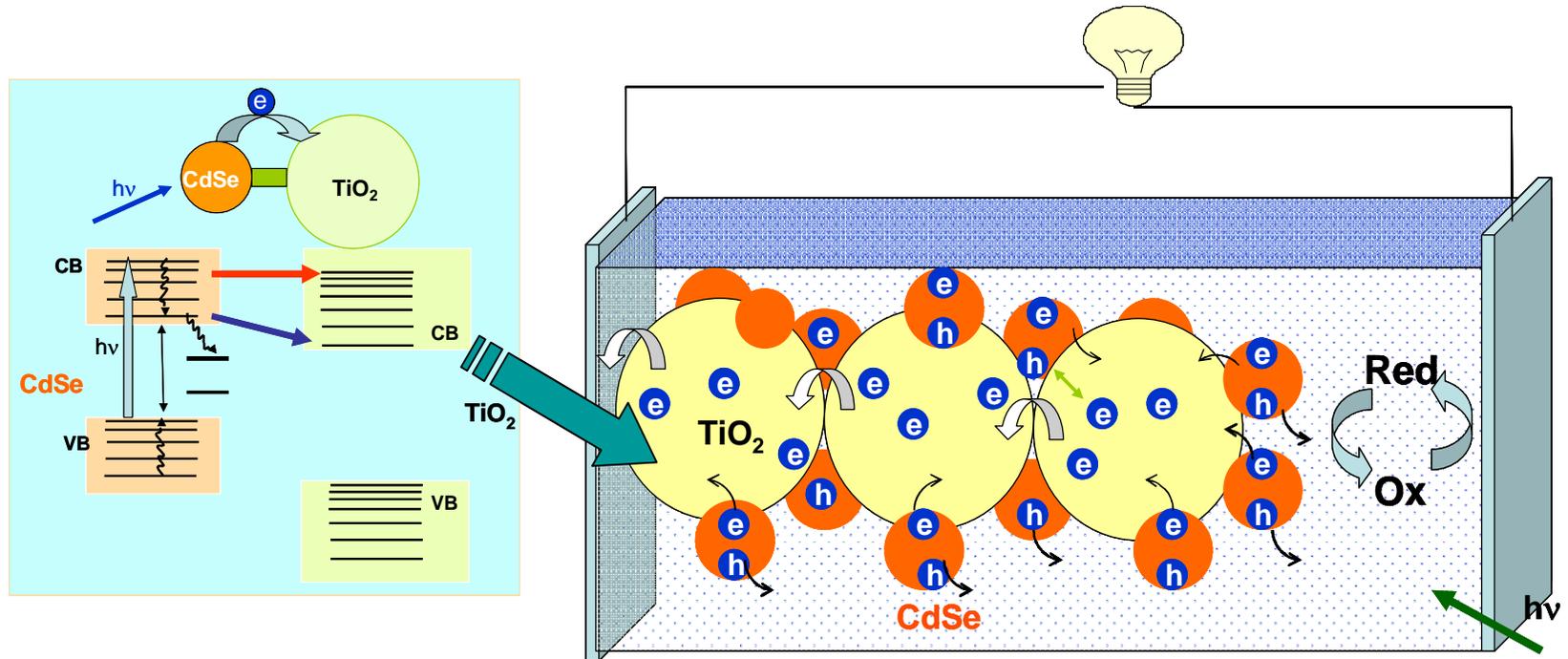


# Hybrid Nanorod-Polymer Solar Cells

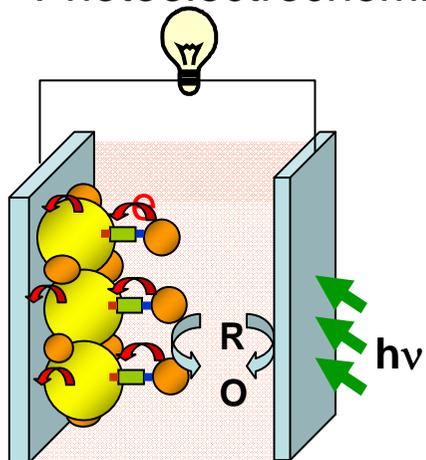


SCIENCE VOL 295, 2002 2425

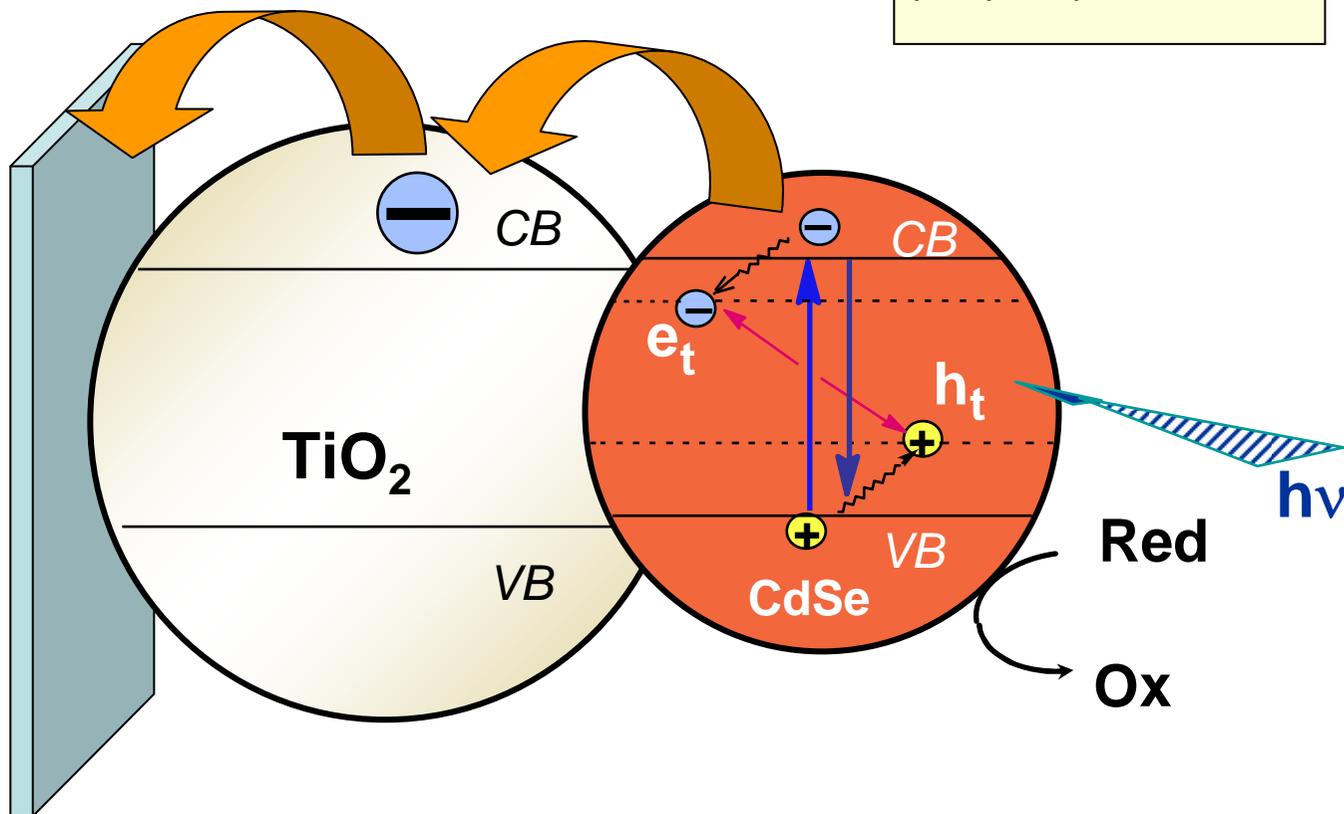
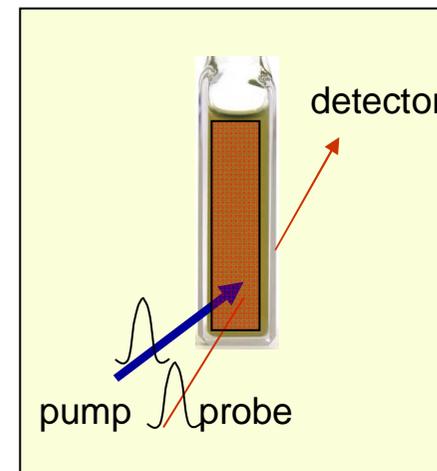
### 3. Quantum Dot Sensitized Solar Cell (QDSSC)



# Photoelectrochemistry

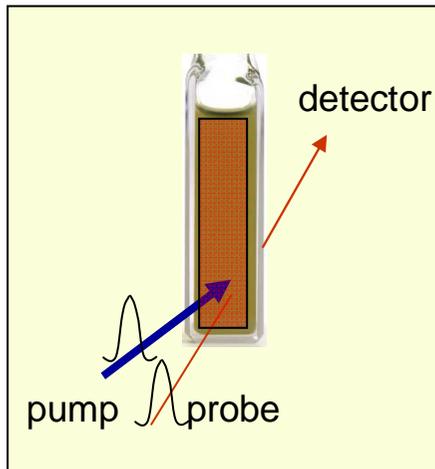


# Spectroscopy

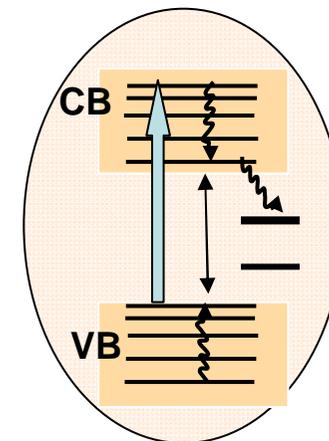


GERISCHER H, LUBKE M

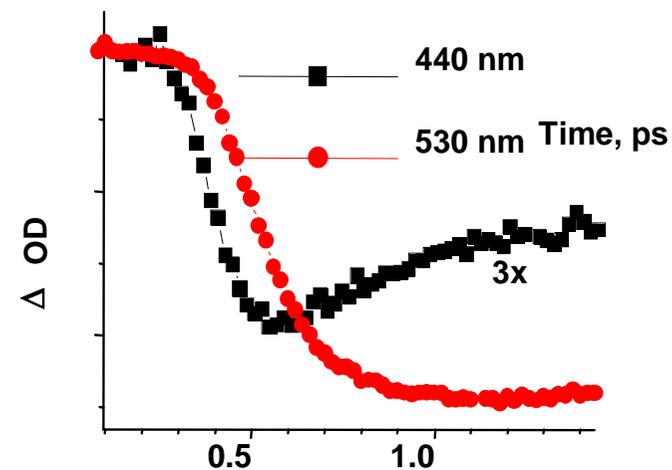
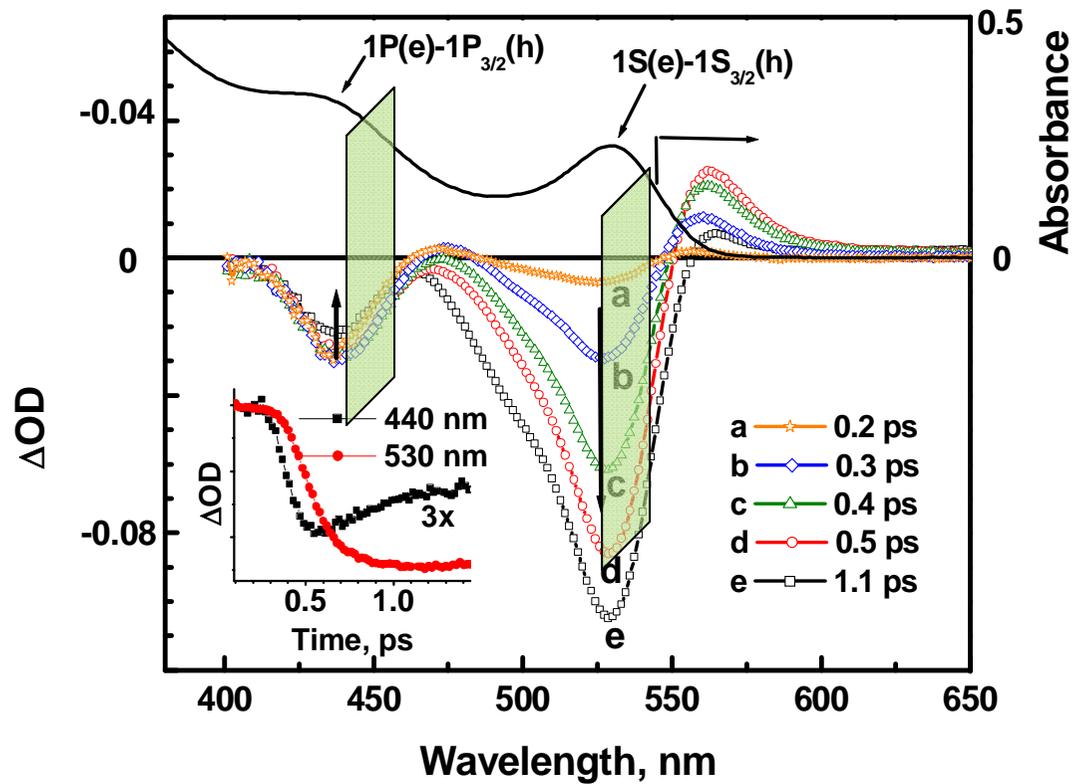
A PARTICLE-SIZE EFFECT IN THE SENSITIZATION OF TiO<sub>2</sub> ELECTRODES BY A CDS DEPOSIT  
 JOURNAL OF ELECTROANALYTICAL CHEMISTRY 204 (1-2): 225-227 1986



## Photoexcitation of CdSe Quantum Dots



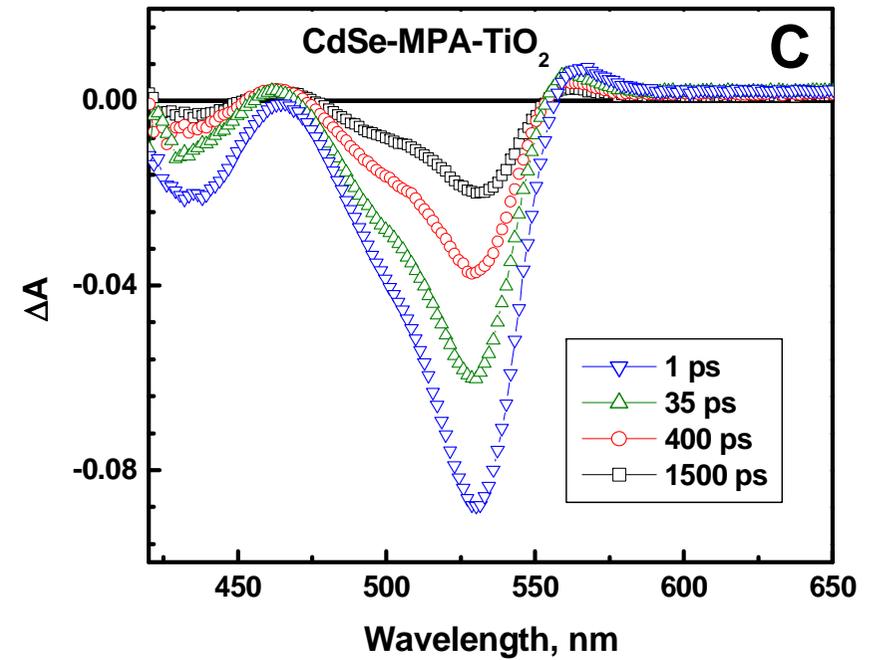
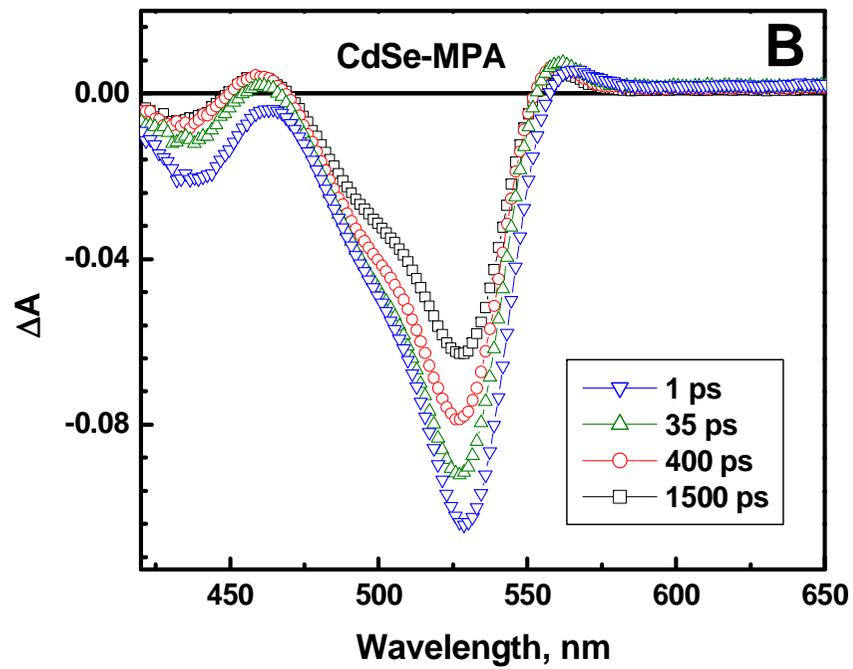
CdSe



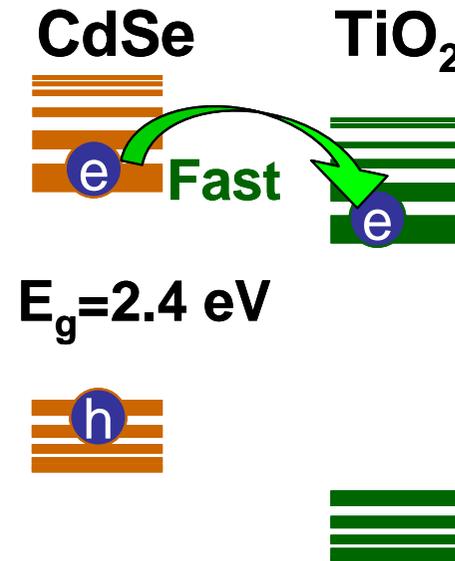
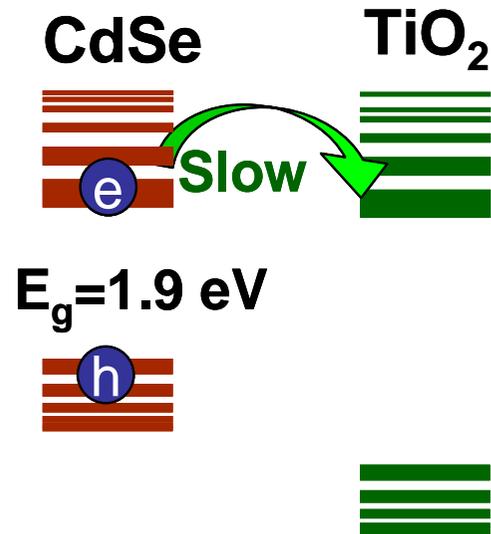
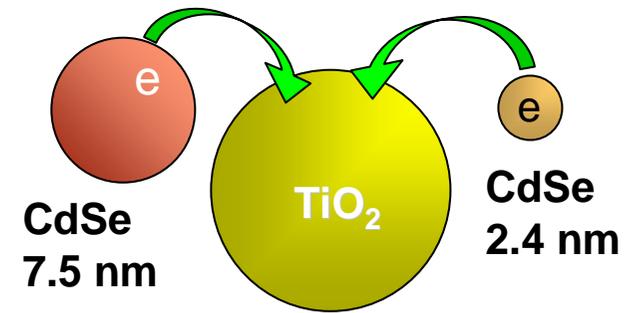
# Charge Separation in TiO<sub>2</sub>/CdSe

## Transient Bleaching Recovery of 3 nm CdSe Quantum Dots

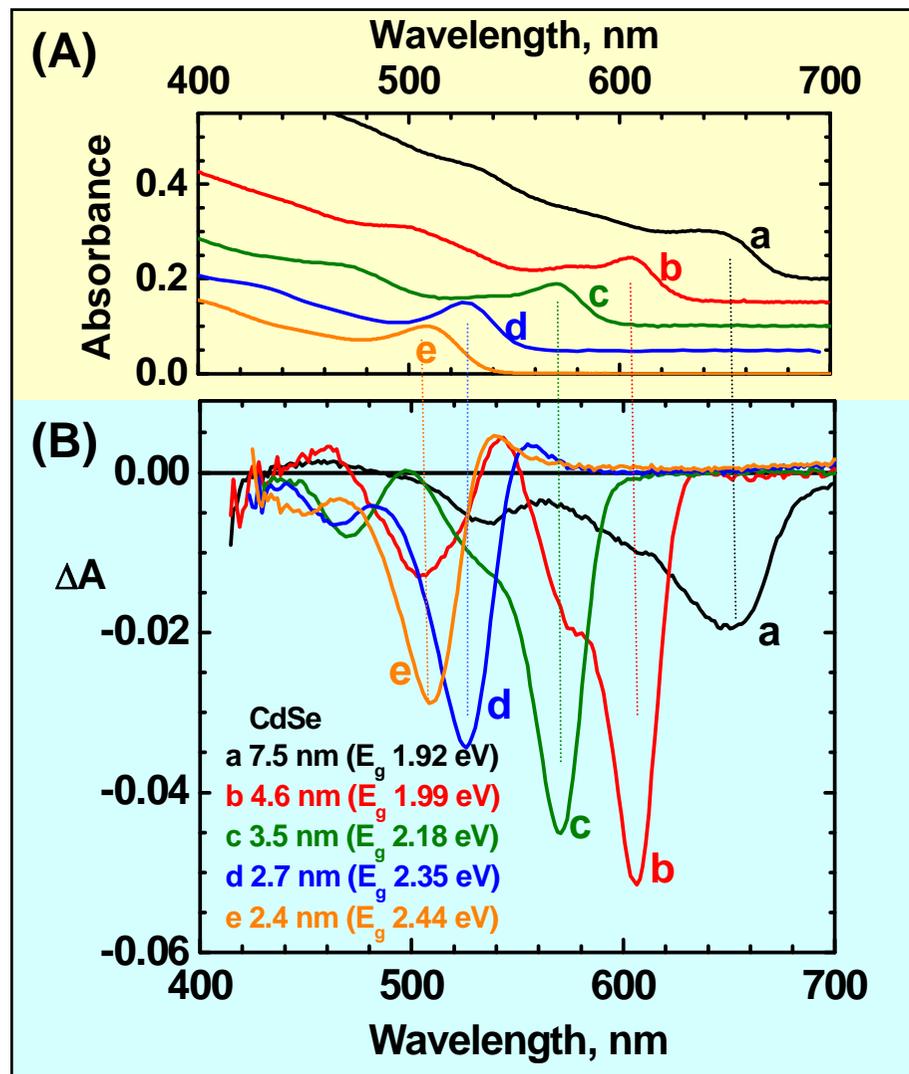
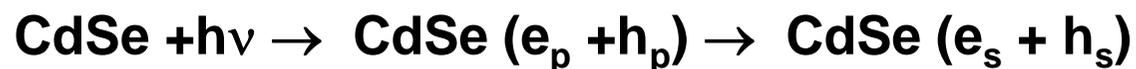
Ex. 387 nm



# Modulation of the charge injection process by controlling the particle size?



## Photoexcitation of CdSe Quantum Dots

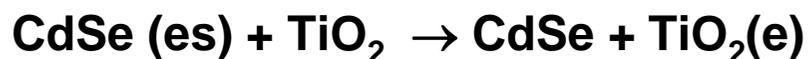


CdSe quantum dots of size 2.4 nm to 7.5 nm were excited with 387 nm laser pulse (130 fs)

As the particle size decreases from 7.5 nm to 2.4 nm, the first ( $^1S_{3/2}^1S_e$ ) excitonic peak shifts from 645 nm (1.92 eV) to 509 nm (2.44 eV).

Transient bleach corresponds to the first excitonic bleach

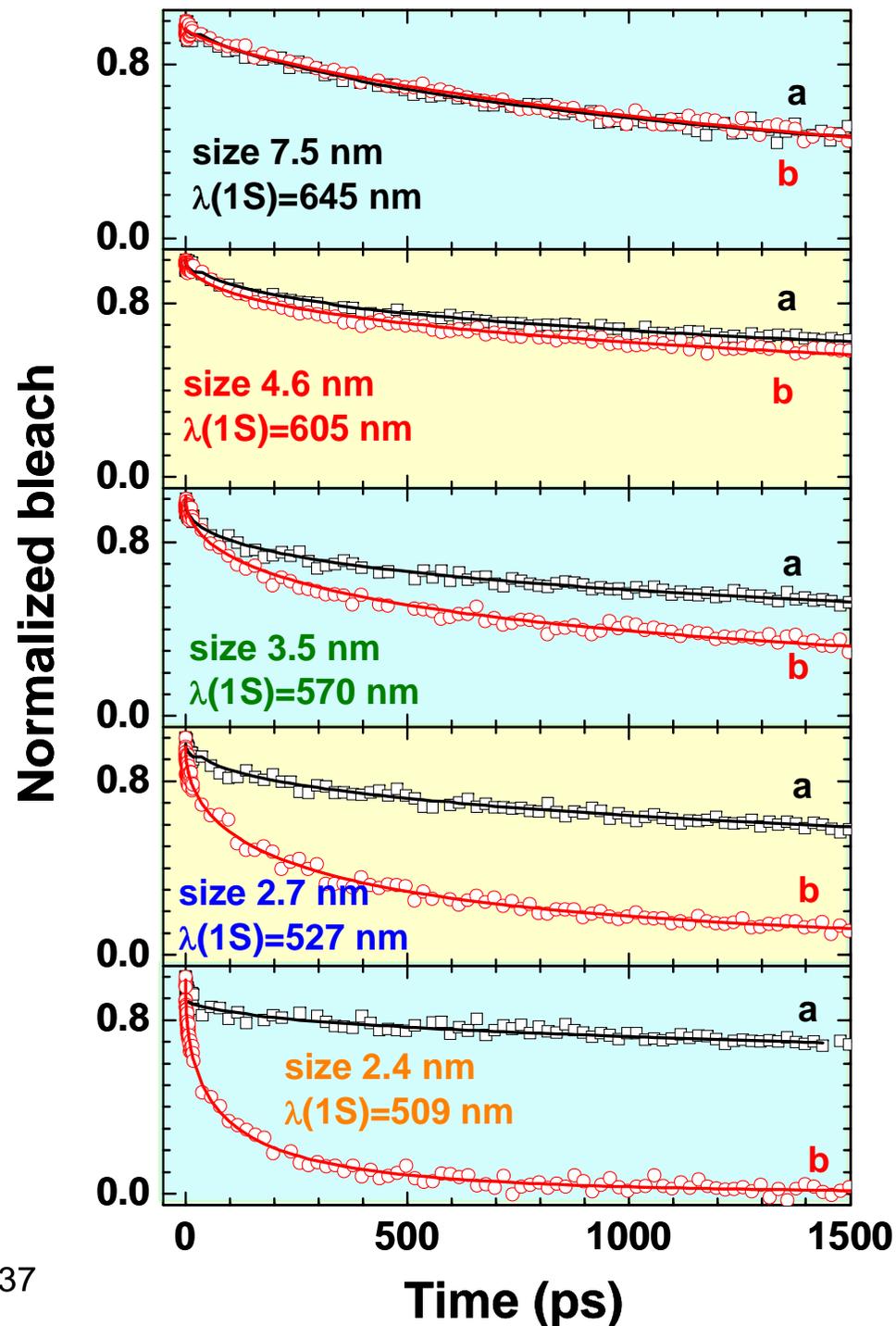
# Electron transfer between CdSe and TiO<sub>2</sub>



Analysis of Bleaching Recovery

$$\Delta A(t) = \Delta A(0) \times \exp[-(t/\tau)\beta]$$

- where  $\tau$  is the peak value of the characteristic lifetime



## Size Dependent Quenching Phenomenon

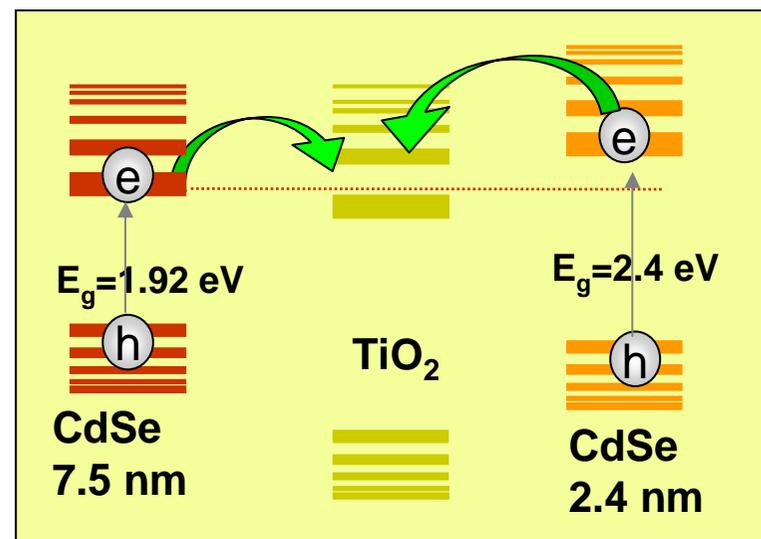
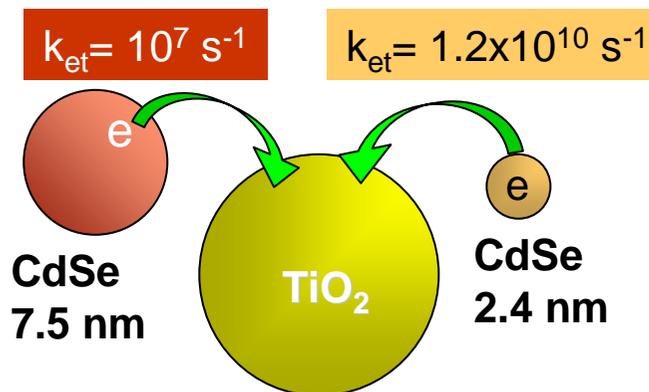
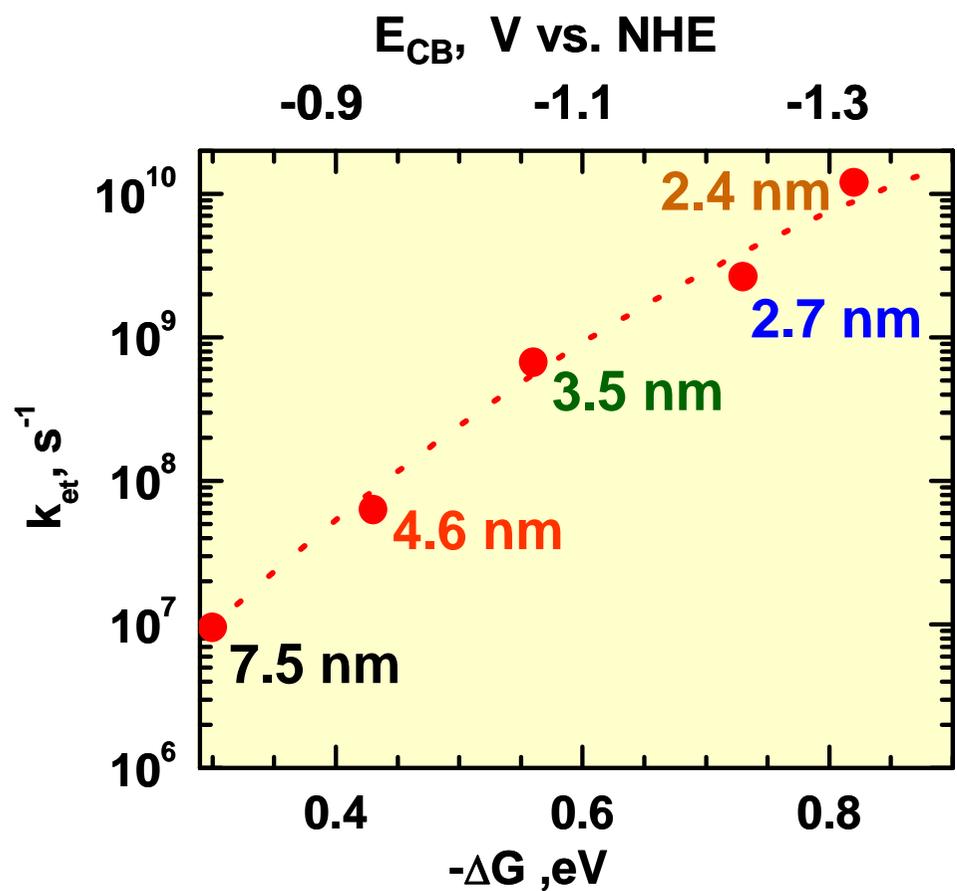
Diameter	$E_g$	$\tau_{\text{CdSe}}$	$\beta_{\text{CdSe}}$	$\tau_{\text{CdSe-TiO}_2}$	$\beta_{\text{CdSe-TiO}_2}$	$k_{\text{et}}$
[nm]	[eV]	[ps]		[ps]		[s <sup>-1</sup> ]
7.5	1.92	2332	0.697	2281	0.755	$9.58 \times 10^6$
4.6	1.99	7224	0.474	4961	0.446	$6.3 \times 10^7$
3.5	2.18	4420	0.417	1117	0.475	$6.7 \times 10^8$
2.7	2.35	6739	0.457	357	0.505	$2.65 \times 10^9$
2.3	2.44	23119	0.51	83	0.493	$1.2 \times 10^{10}$

$$\Delta A(t) = \Delta A(0) \times \exp[-(t/\tau)\beta]$$

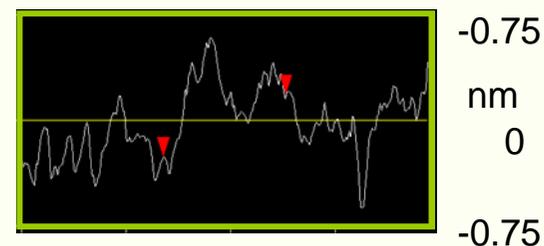
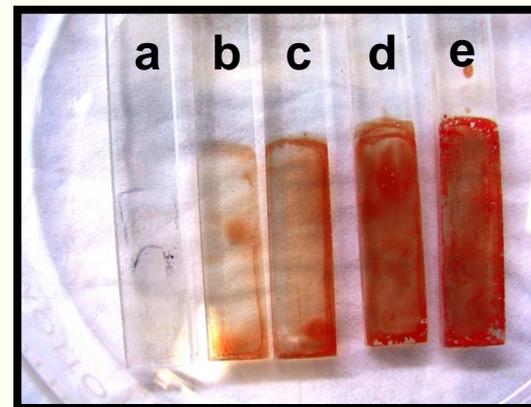
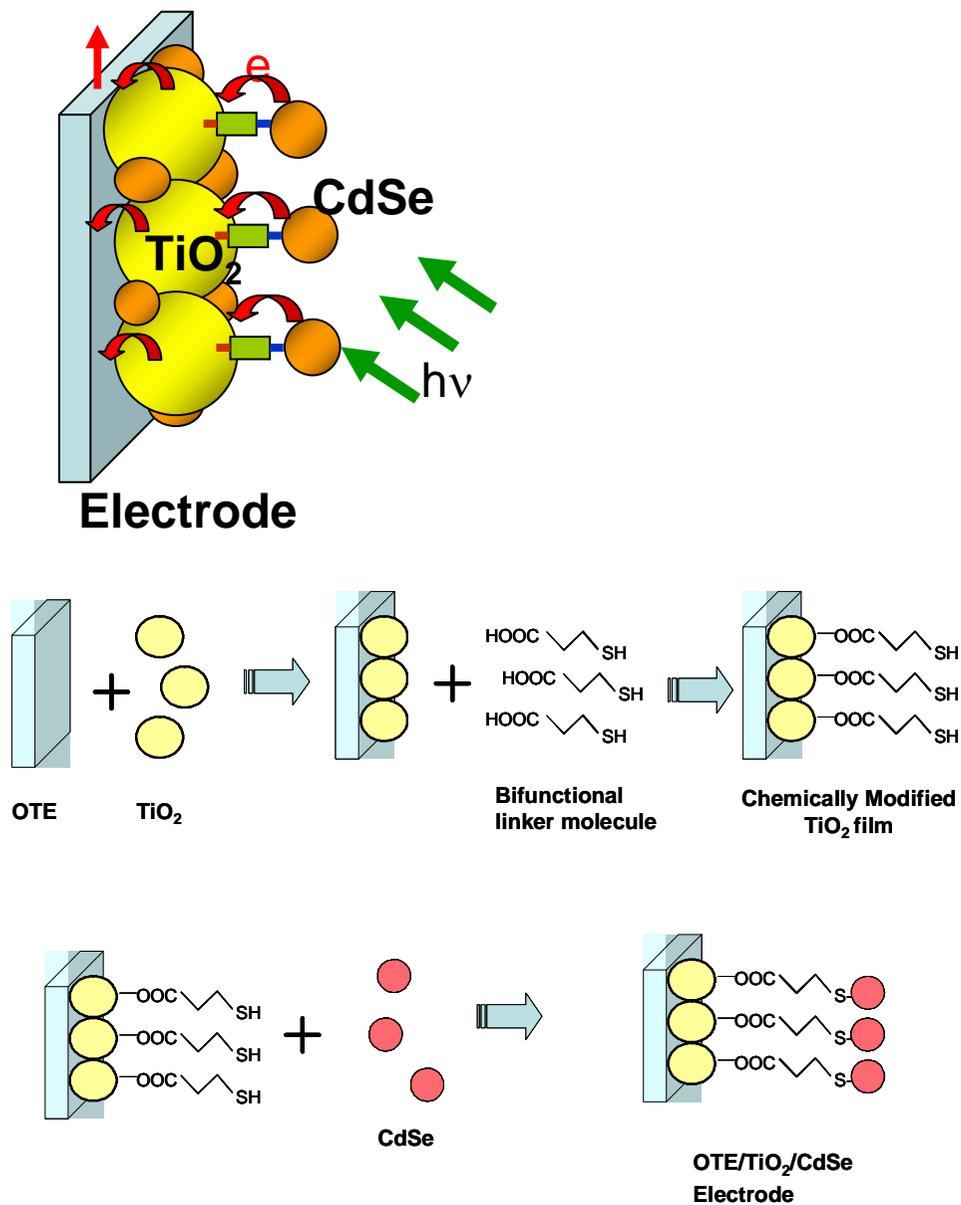
- where  $\tau$  is the peak value of the characteristic lifetime

$$1/\tau' - 1/\tau = k_{\text{et}}$$

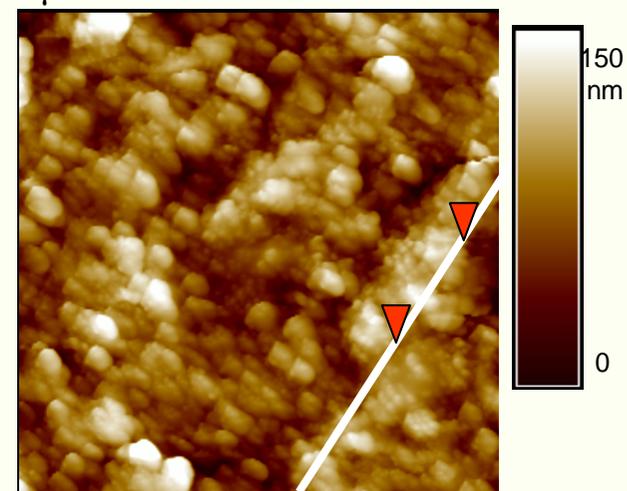
# Size Dependent Electron transfer between CdSe and TiO<sub>2</sub>



# Linking Q-CdSe to TiO<sub>2</sub> particles



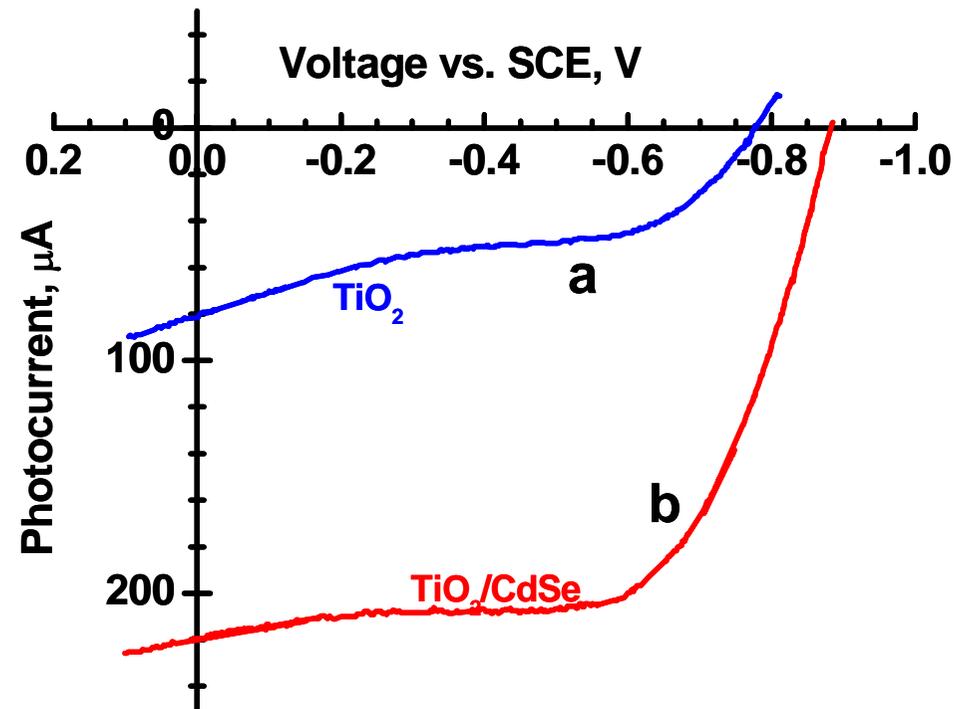
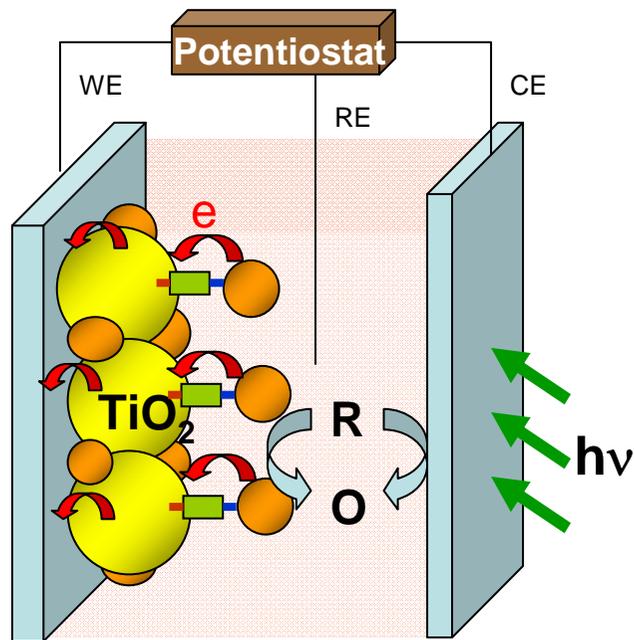
2 μm



0

2 μm

# Photoelectrochemical behavior of Q-CdSe-TiO<sub>2</sub> films



I-V characteristics of (a) OTE/TiO<sub>2</sub> and (b) OTE/TiO<sub>2</sub>/MPA/CdSe films. Electrolyte 0.1 M Na<sub>2</sub>S. The filtered lights allowed excitation of TiO<sub>2</sub> and CdSe films at wavelengths **greater than 300 and 400 nm** respectively

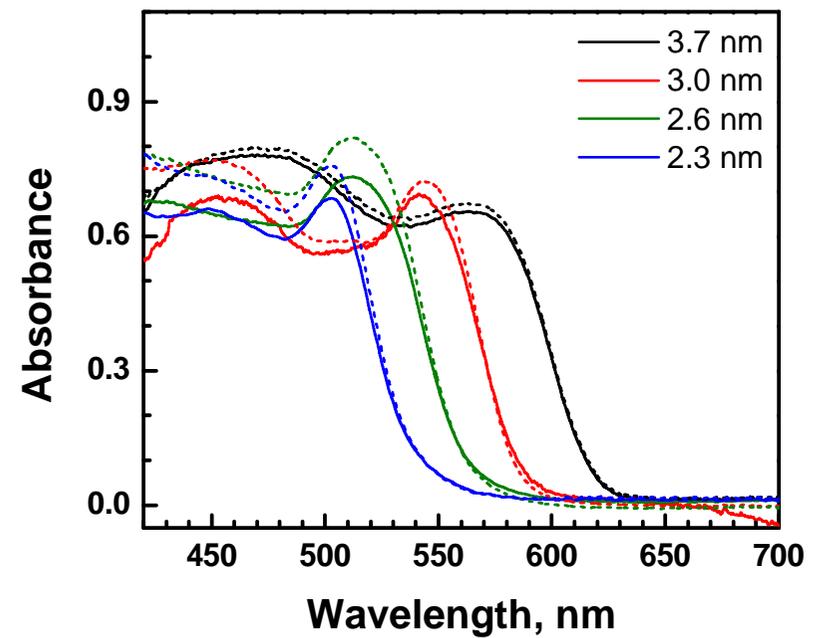
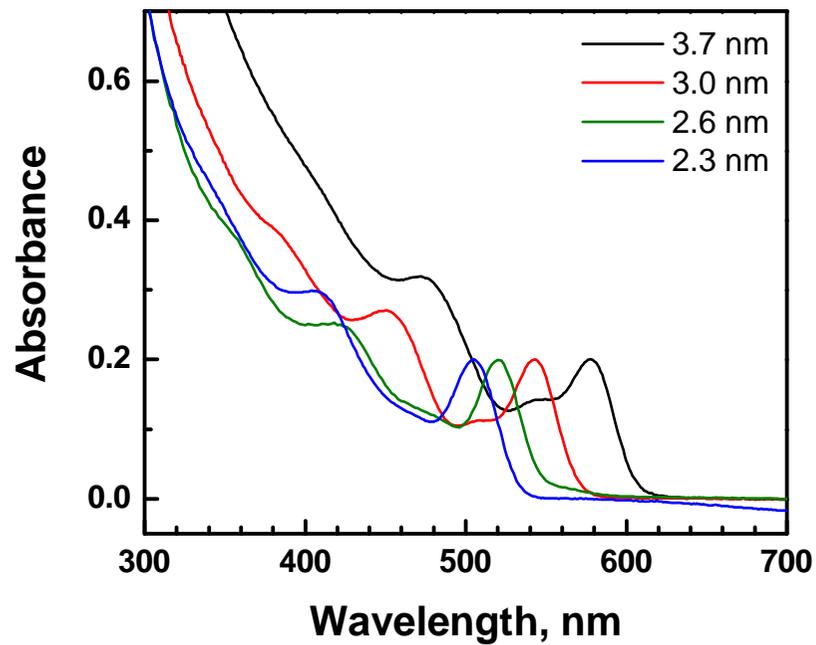
# Modification of TiO<sub>2</sub> Films with Different Size CdSe Particles



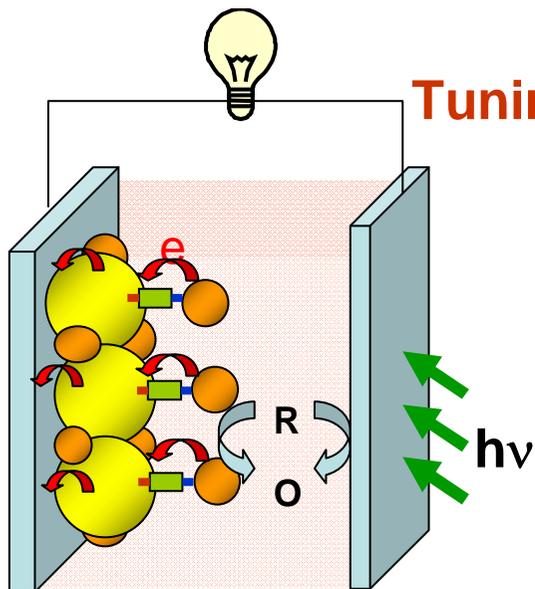
2.3nm      3.0nm  
2.6nm      3.7nm



2.3nm      3.0nm  
2.6nm      3.7nm

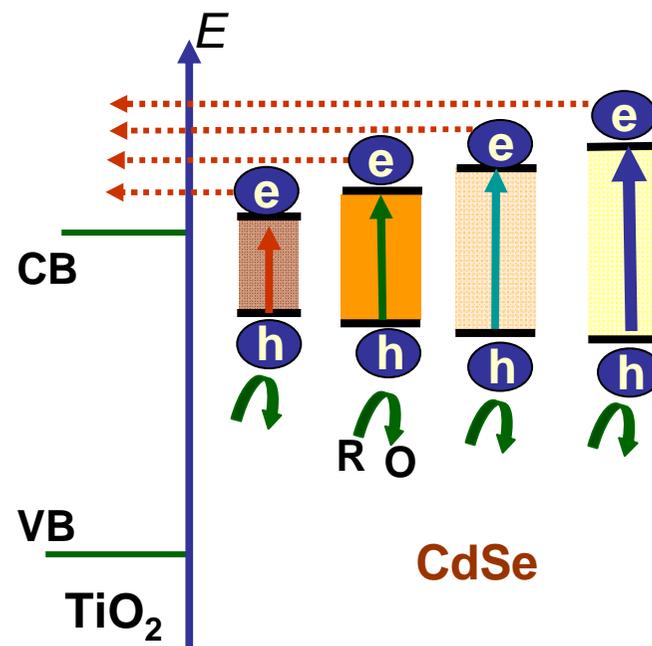
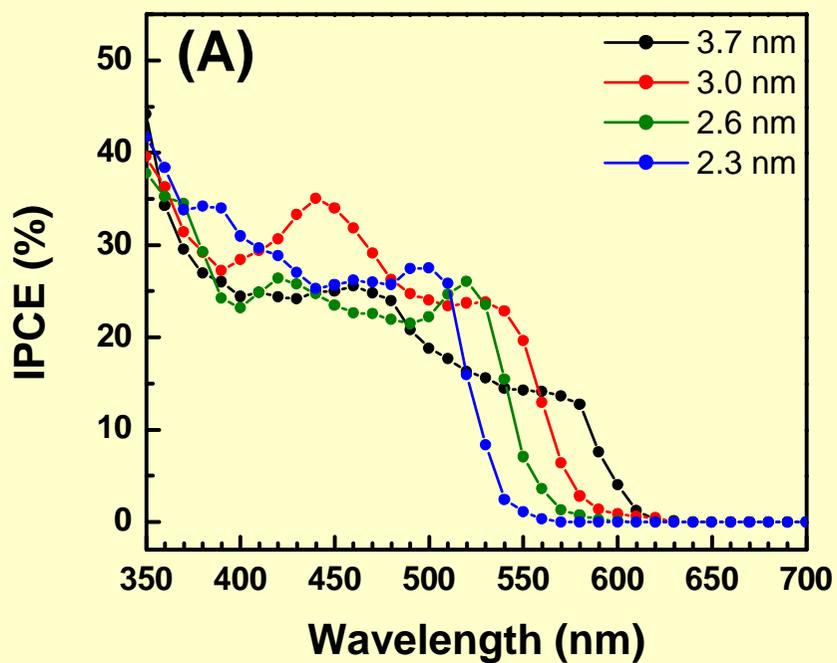


# Tuning the Photoresponse of Quantum Dot Solar Cells



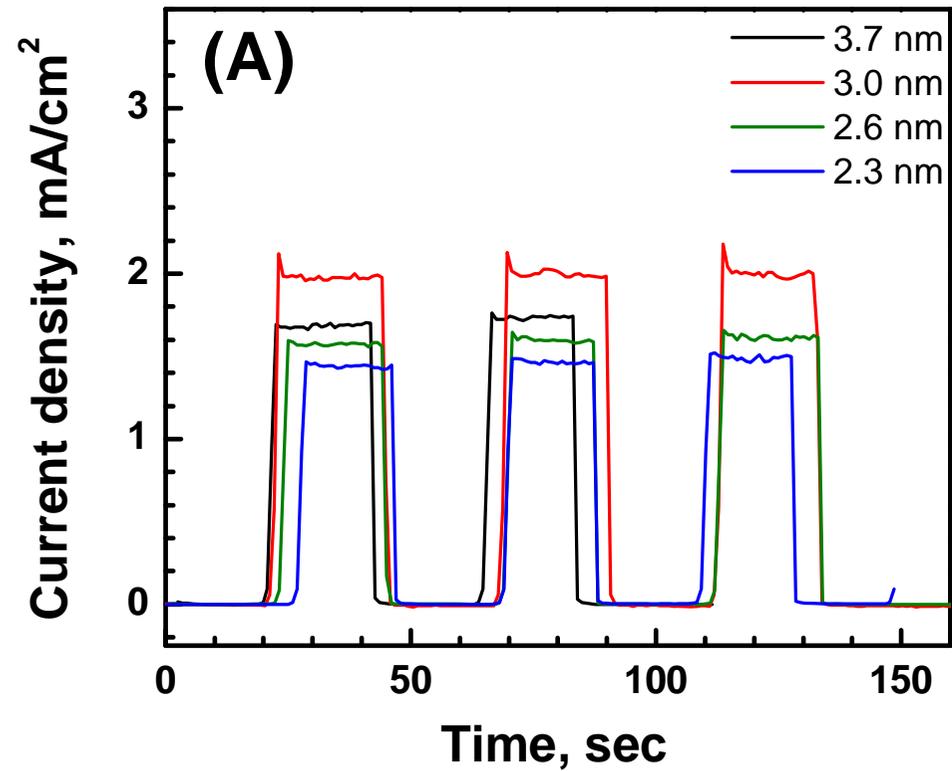
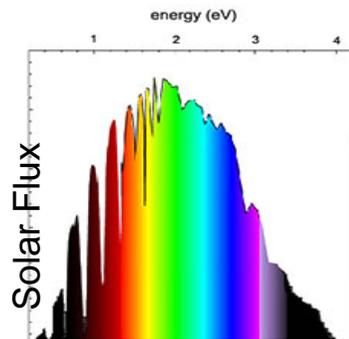
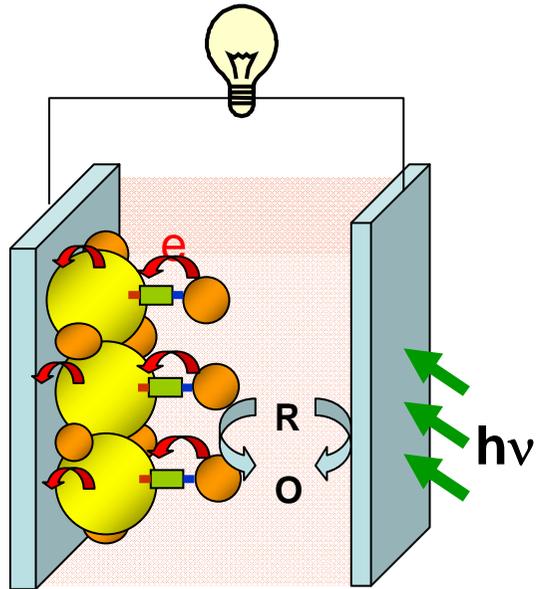
IPCE or Ext. Quantum Eff.

$$= (1240/\lambda) \times (I_{sc}/I_{inc}) \times 100$$

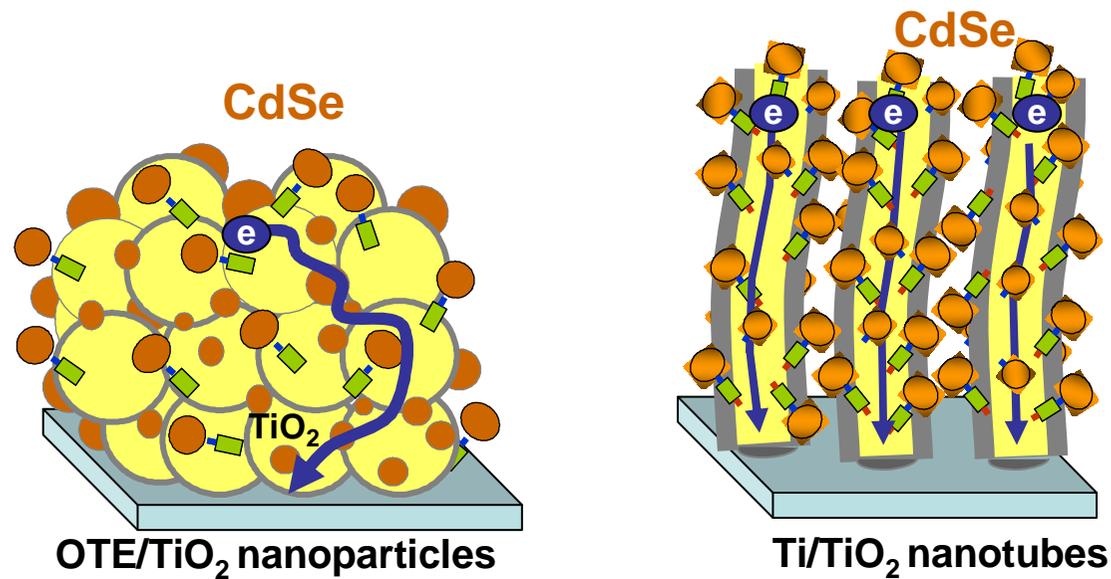


# Photocurrent Response

Efficiency of Charge Injection vs. Light absorption



# Can we employ the nanowire/nanorod architecture to improve the performance of quantum dot solar cells?



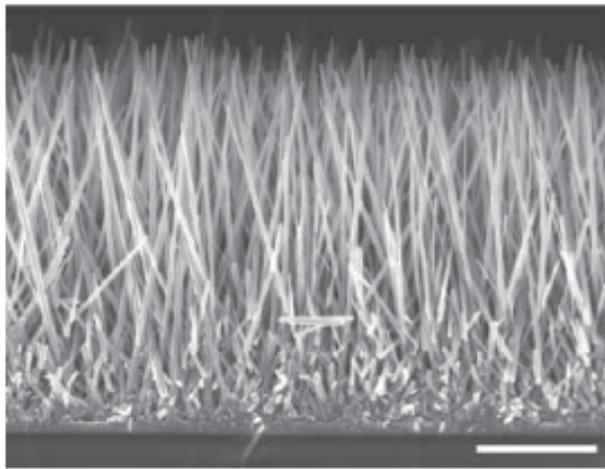
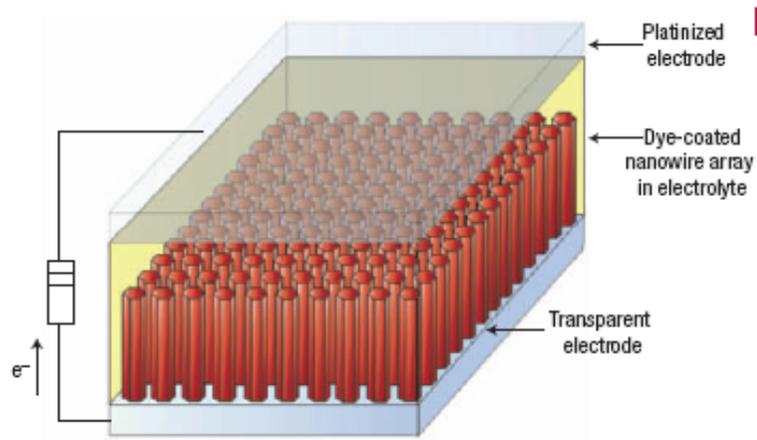
(a)

(b)

## Recent advances

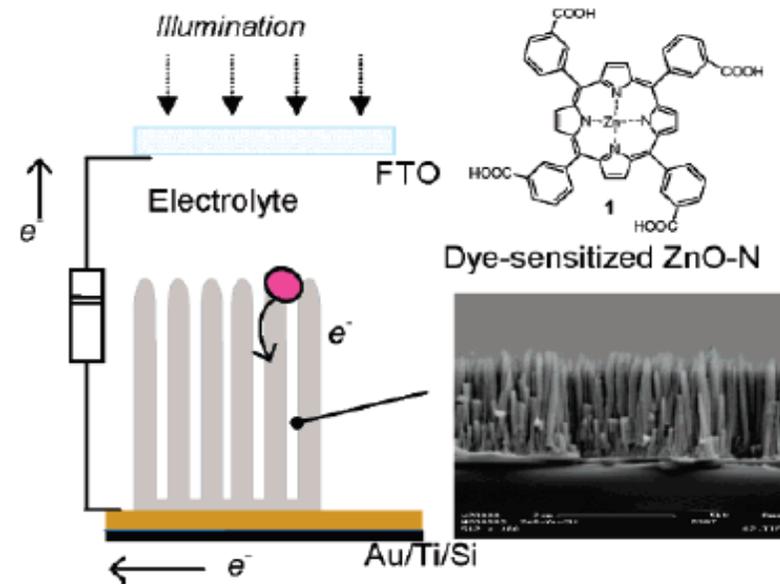
### Nanowire dye-sensitized solar cells

LAW, GREENE, JOHNSON, SAYKALLY, YANG *Nature Materials* 4 , 455, 2005



### Fast Electron Transport in Metal Organic Vapor Deposition Grown Dye-sensitized ZnO Nanorod Solar Cells

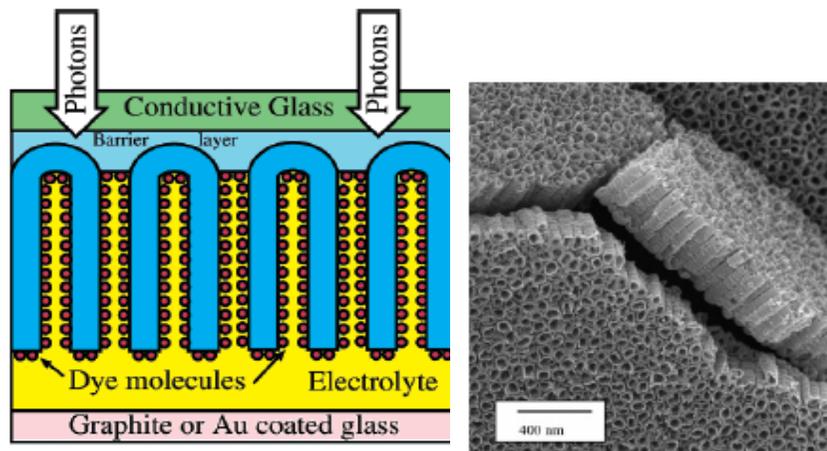
Galoppini, Rochford, Chen, Saraf, Lu, Hagfeldt, and Boschloo *J. Phys. Chem. B*; **2006**; *110* 16159



*Electron transport in solar cells with ZnO-nanorod electrodes was about 2 orders of magnitude faster ( $30\mu\text{s}$ ) than ZnO-colloid electrodes*

Mor, G. K. et al *Use of highly-ordered TiO<sub>2</sub> nanotube arrays in dye-sensitized solar cells.*

Nano Lett., **2006.** 6, 215-218.

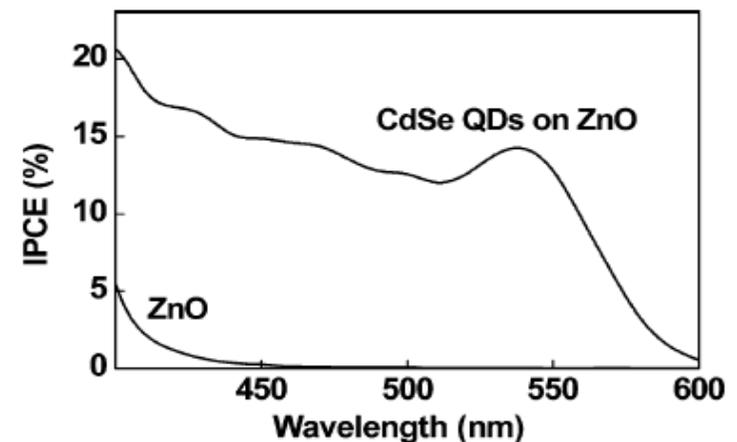
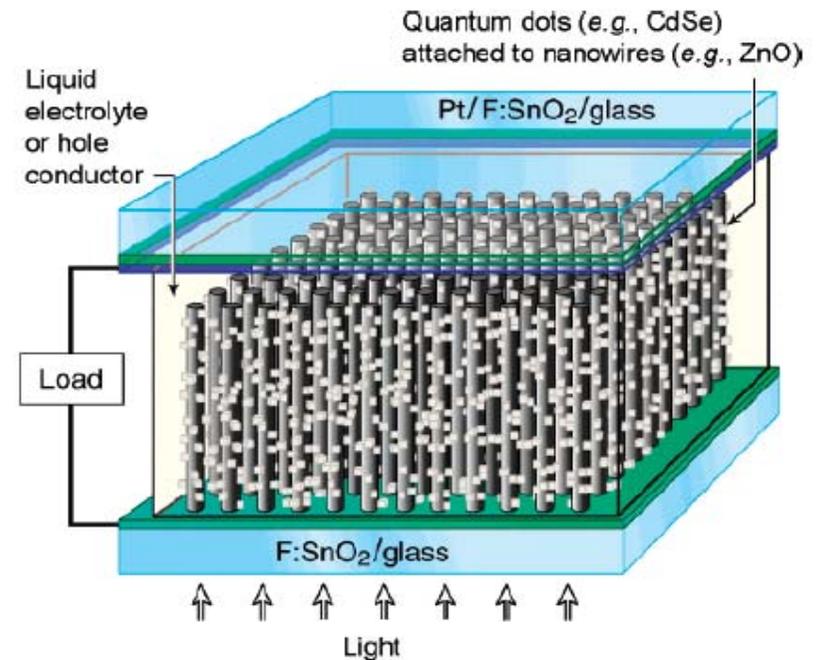


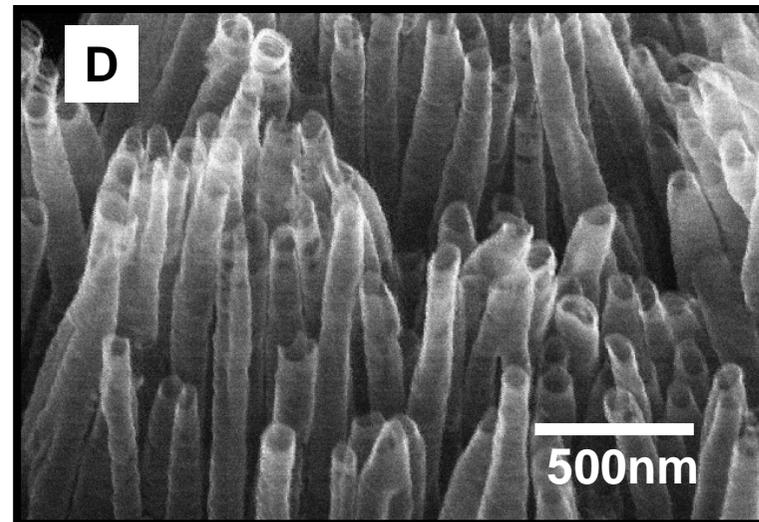
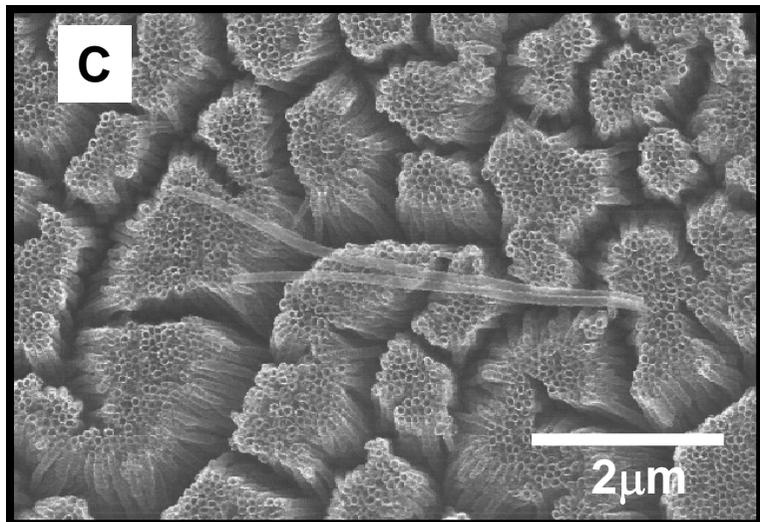
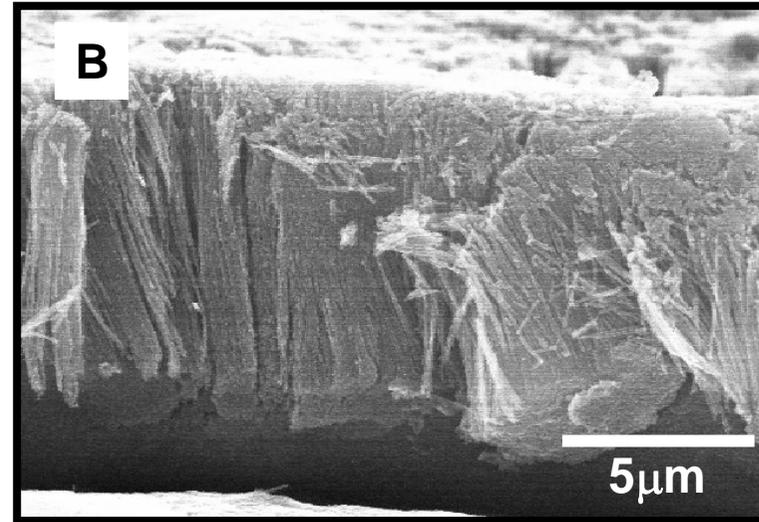
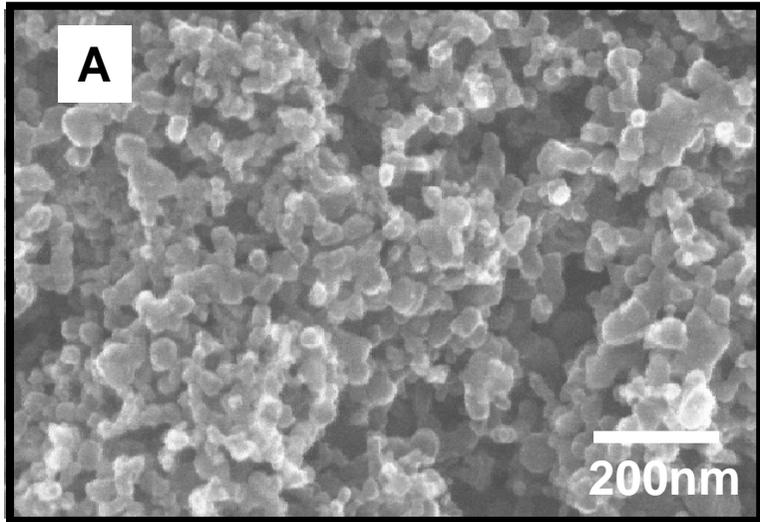
Martinson, A. B. F. et al., *ZnO nanotube based dye-sensitized solar cells ZnO nanotube based dye-sensitized solar cells.*

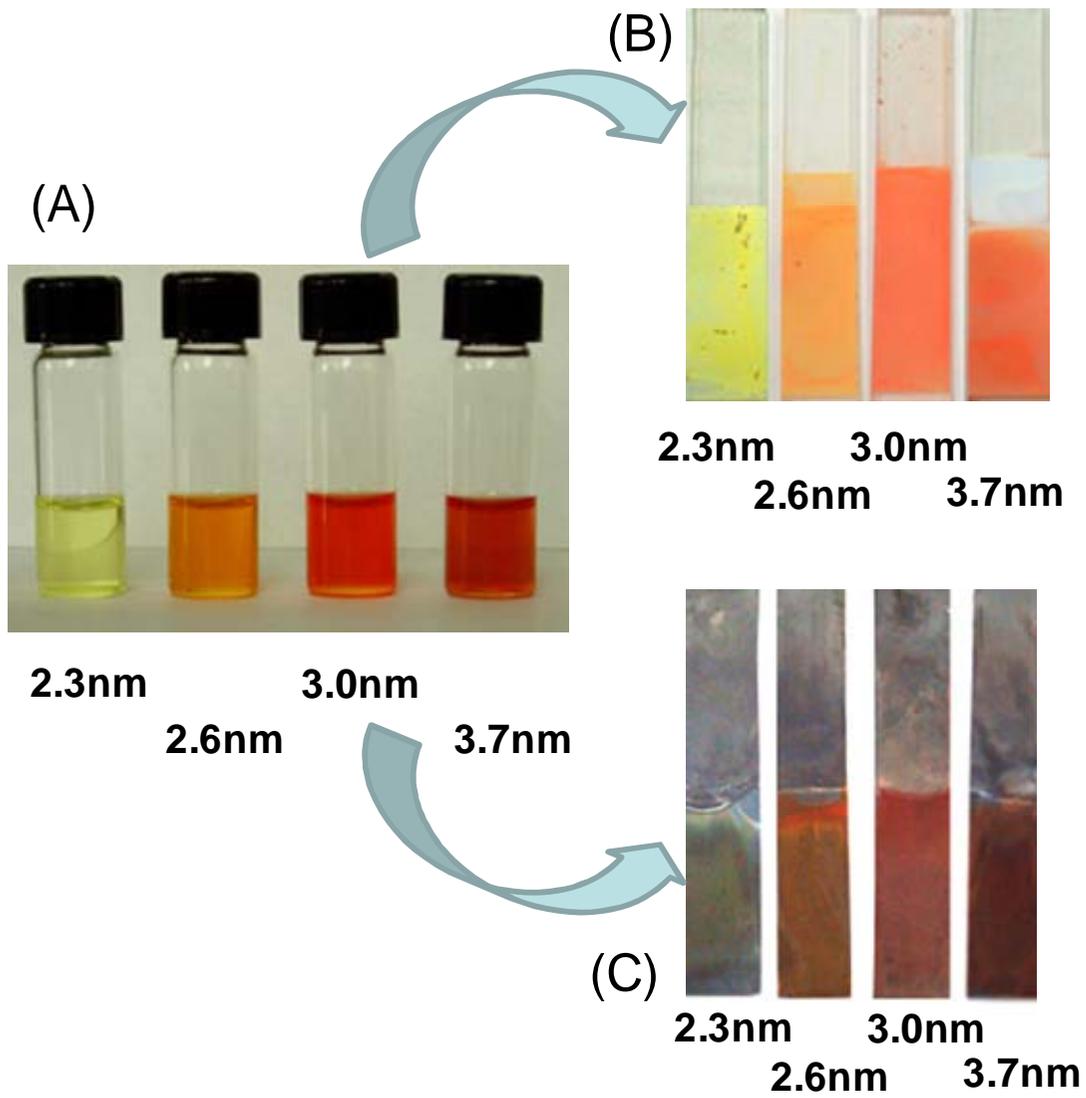
Nano Lett., **2007.** 7, 2183-2187.

Leschkies, K. S et al *Photosensitization of ZnO nanowires with CdSe quantum dots for photovoltaic devices.*

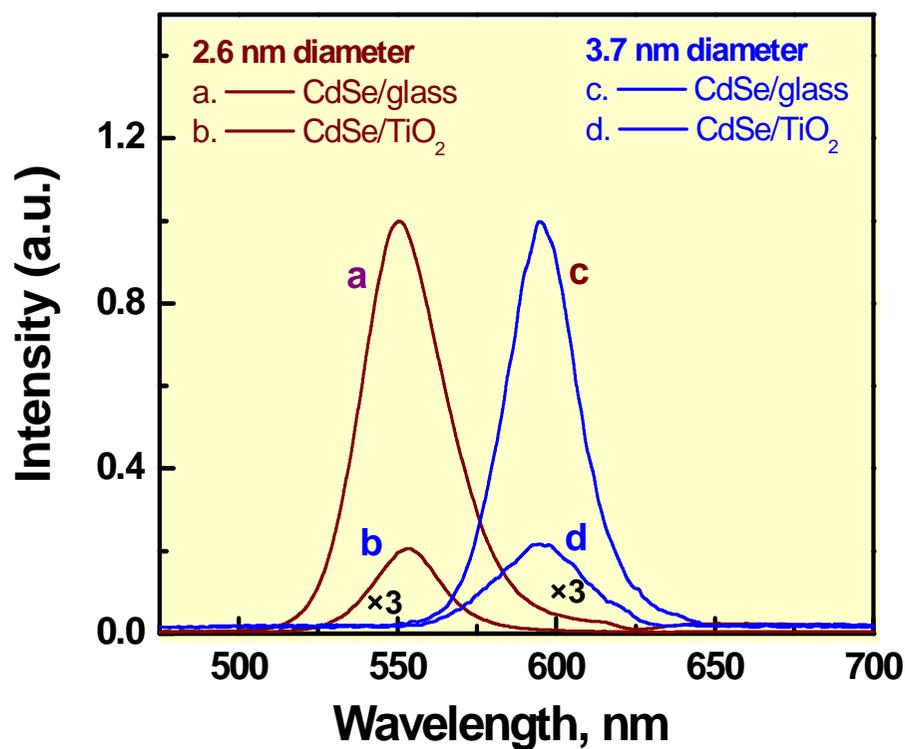
Nano Lett., **2007.** 7, 1793-1798.



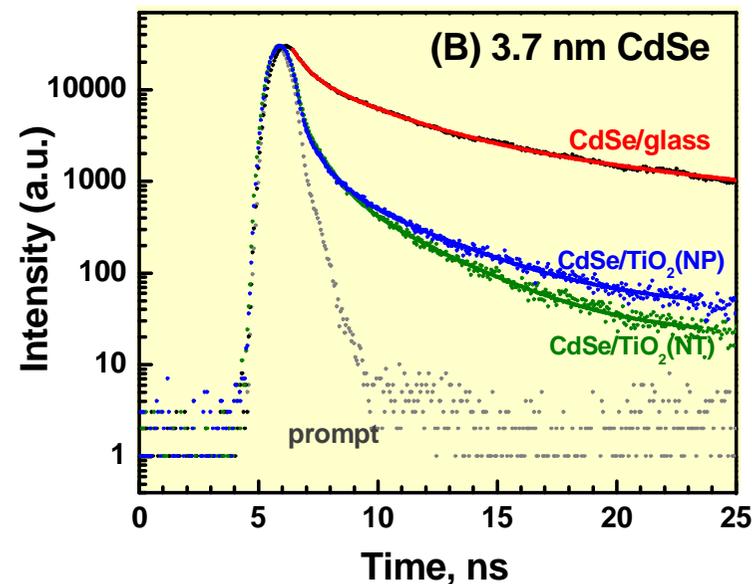




## Photoemission of CdSe/glass and CdSe/TNP

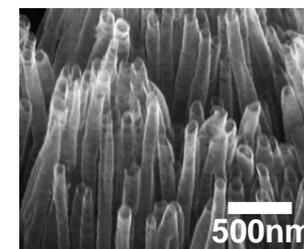
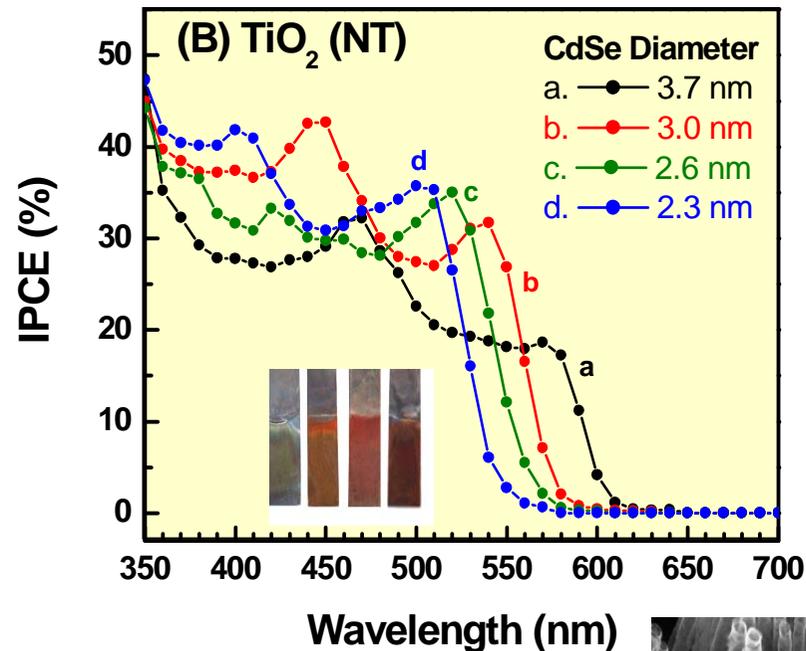
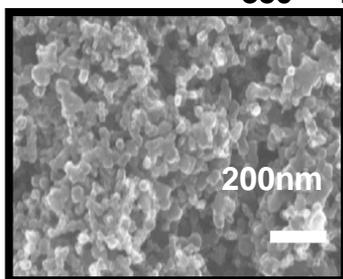
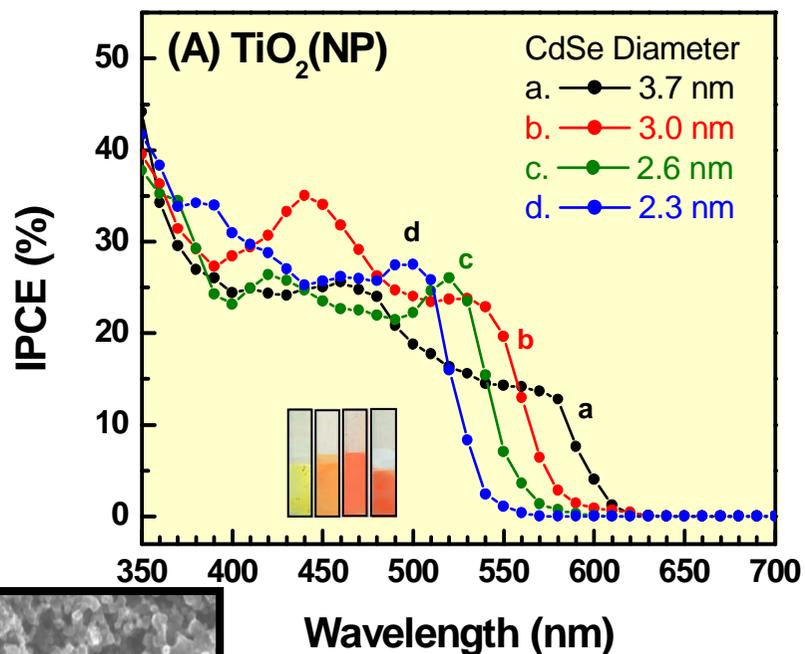


Emission spectra of CdSe QDs (a, c) on glass and (b, d) chemically bound to TiO<sub>2</sub> nanoparticle films at 2 different sizes of QDs (2.7 and 3.7 nm). Excitation was at 480 nm.



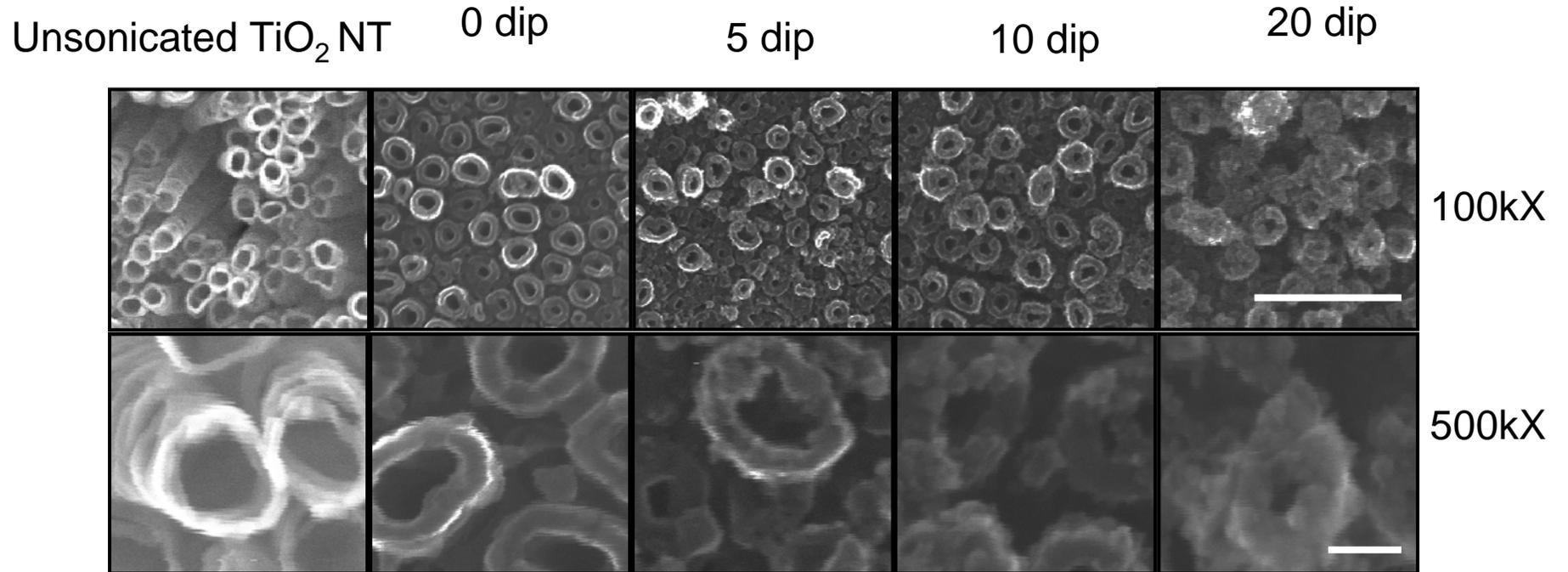
		$\langle t \rangle$ , ns	$k_{et}$ , sec <sup>-1</sup>
CdSe/ glass	2.6nm	4.1	
	3.7nm	7.9	
CdSe/ tnp	2.6nm	0.4	2.5E+09
	3.7nm	1.3	6.3E+08
CdSe/ tnt	2.6nm	0.4	2.2E+09
	3.7nm	1.5	5.5E+08

## Quantum Dot Solar Cells – Particle versus Tube Architecture

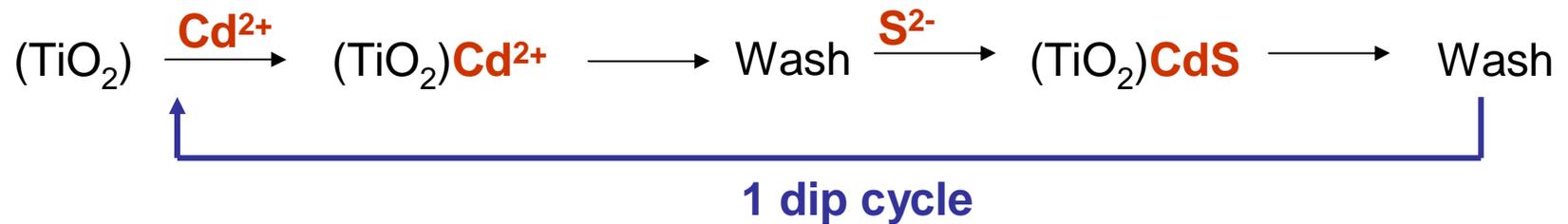


	$I_{sc}$ mA/cm <sup>2</sup>	$V_{oc}$ V	$P_{max}$ mW/cm <sup>2</sup>	FF
CdSe-TNP	1.64	0.591	0.25	0.26
CdSe-TNT	1.95	0.582	0.29	0.26

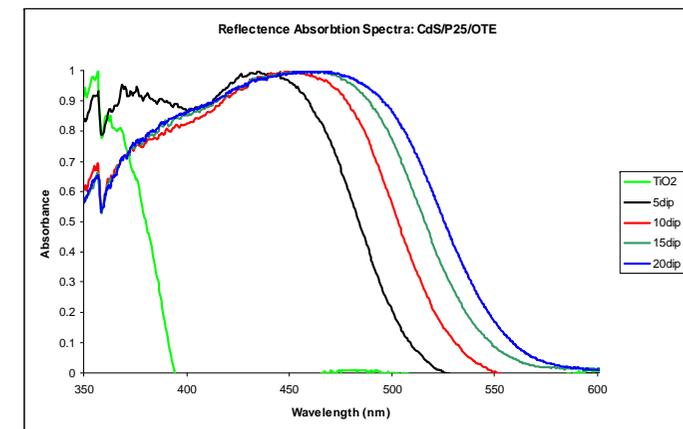
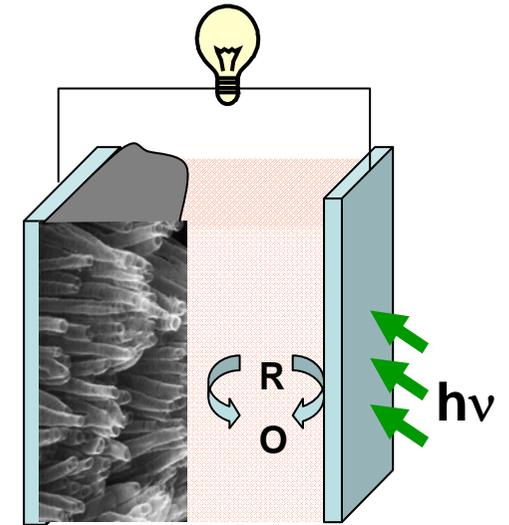
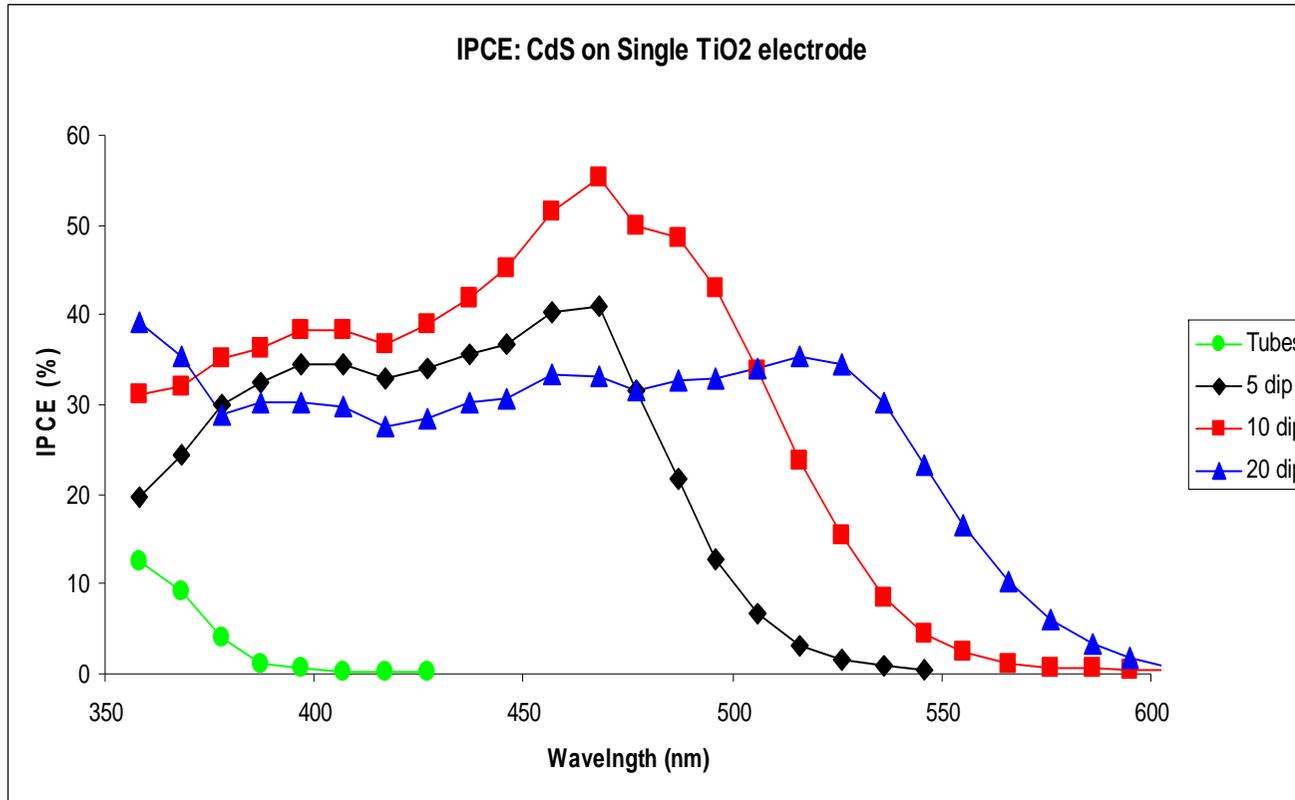
## Depositing CdS quantum dots on TiO<sub>2</sub> nanotubes



Scale bars: Top 500nm, bottom 50nm



# Photocurrent Response of TiO<sub>2</sub> (nanotube)CdS Films



# Carbon nanostructures as conduits to transport charge carriers

## Advantages

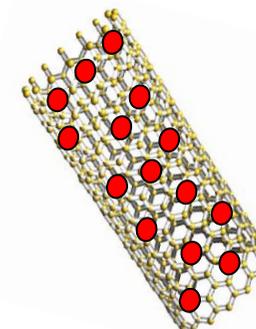
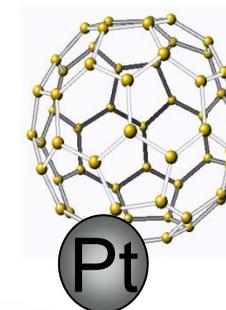
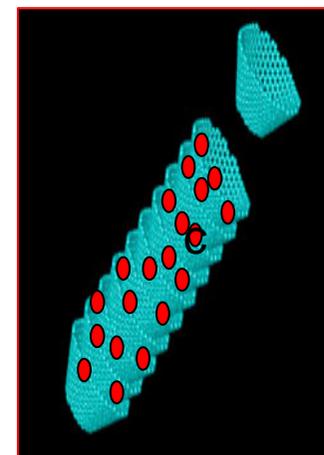
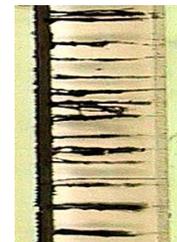
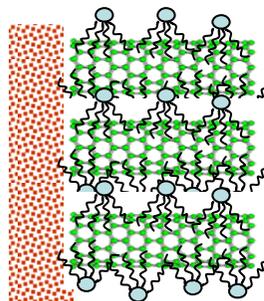
- High surface area
- Good electronic conductivity, excellent chemical and electrochemical stability
- Good mechanical strength

## Goal

Effective utilization of carbon nanostructures for improving the performance of energy conversion devices

- To develop electrode assembly with CNT supports
- Improve the performance of light harvesting assemblies
- Facilitate charge collection and transport in nanostructured assemblies

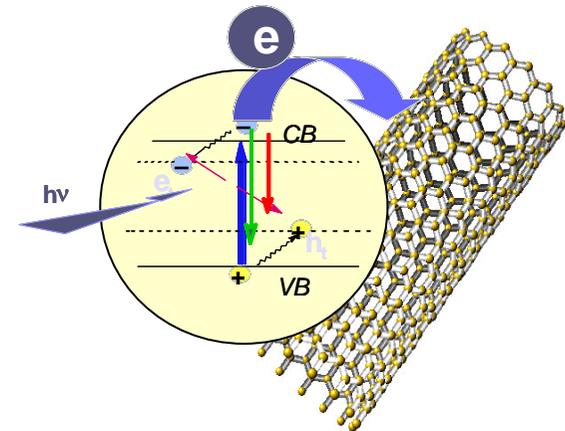
.....towards achieving ordered assemblies on electrode surface



Pt

# SWCNT- TiO<sub>2</sub> composite films

- Mesoscopic TiO<sub>2</sub> films are extensively used in Dye-Sensitized Solar Cells
- A carbon nanotube support architecture can disperse the TiO<sub>2</sub> particles and facilitate charge collection and charge transport within the film.
- The first step is to design the SWCNT-TiO<sub>2</sub> network and test the feasibility of the composite system in solar cells

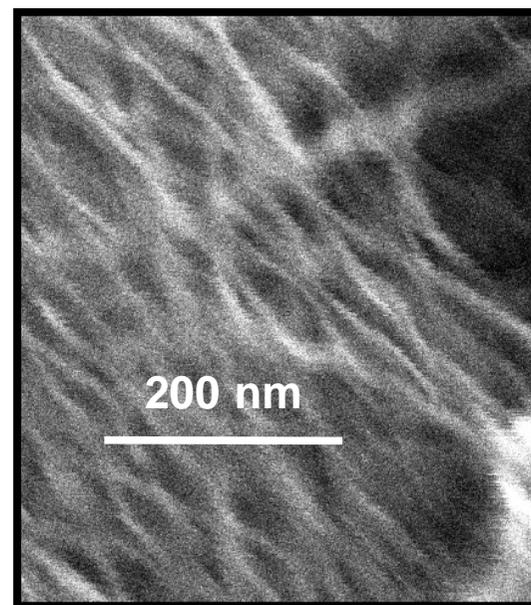
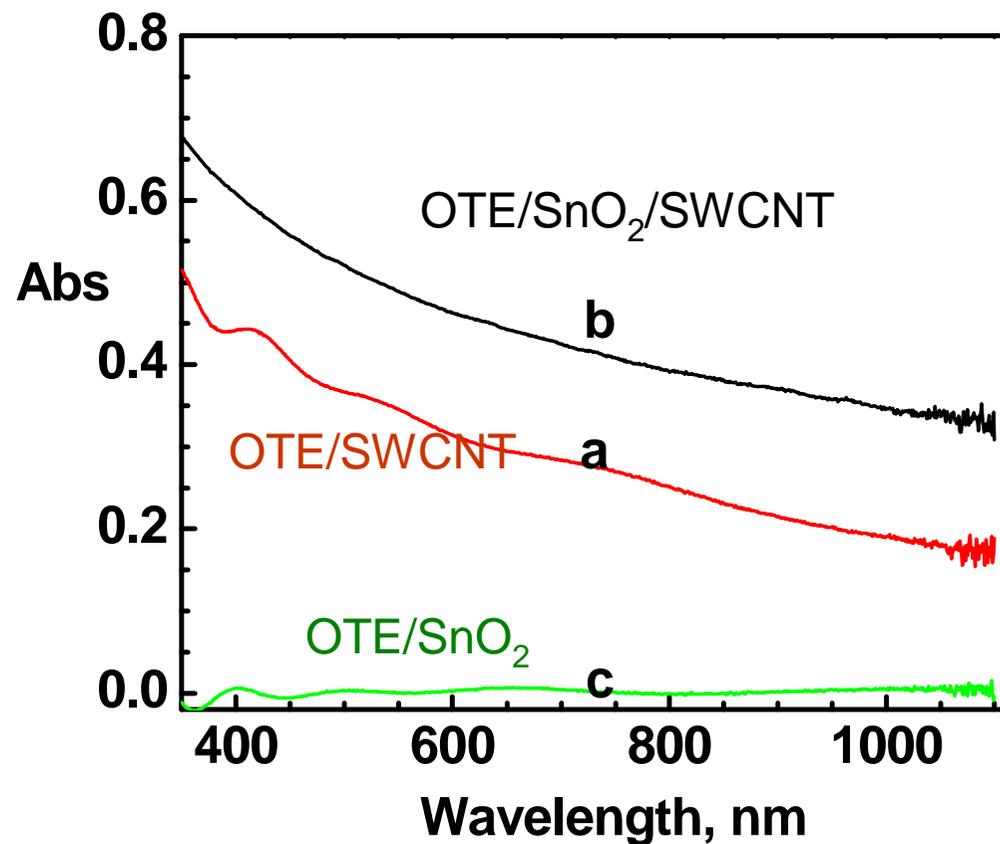
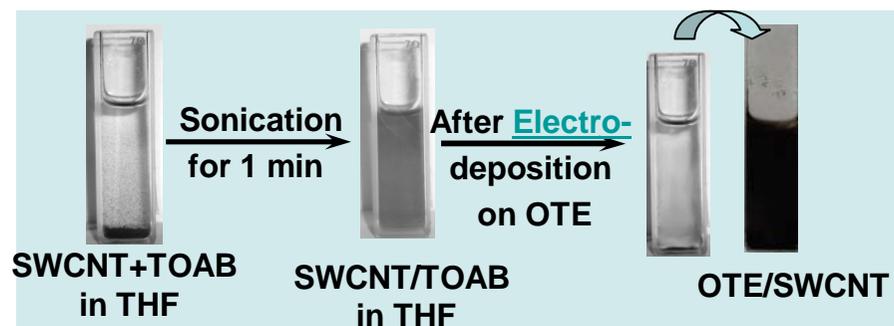


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Kongkanand, A.; Domínguez, R.M.; Kamat, P.V., *Single Wall Carbon Nanotube Scaffolds for Photoelectrochemical Solar Cells. Capture and Transport of Photogenerated Electrons. Nano Lett.*, **2007**, 7, 676-680.

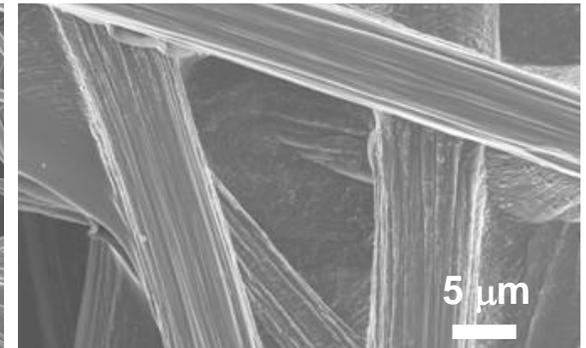
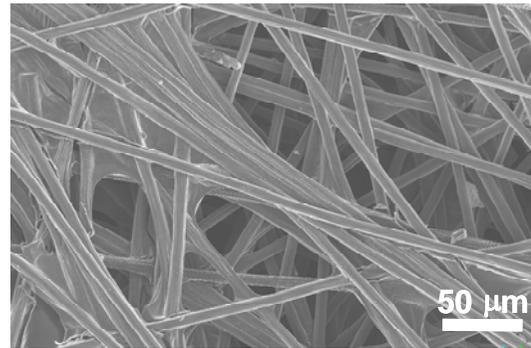
Vietmeyer, F.; Seger, B.; Kamat, P.V., *Anchoring ZnO Particles on Functionalized Single Wall Carbon Nanotubes. Excited State Interactions and Charge Collection. Adv. Mater.*, **2007**, 19: 2935-2940

## Electrophoretic Deposition of SWCNT on Electrode Surfaces



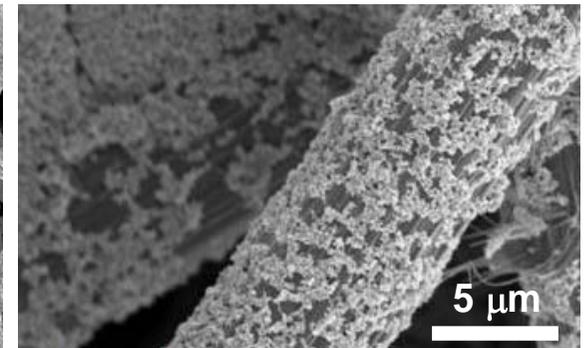
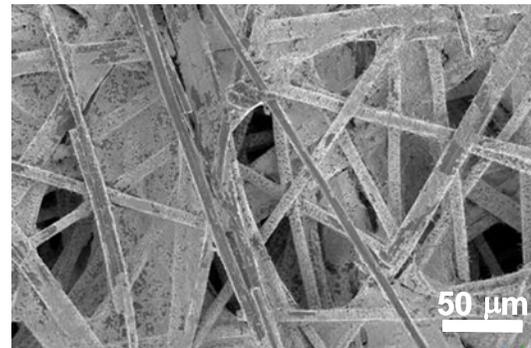
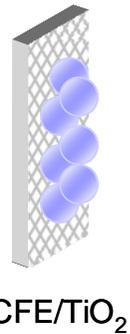
**Carbon Fiber Paper  
(CFE)**

**A**



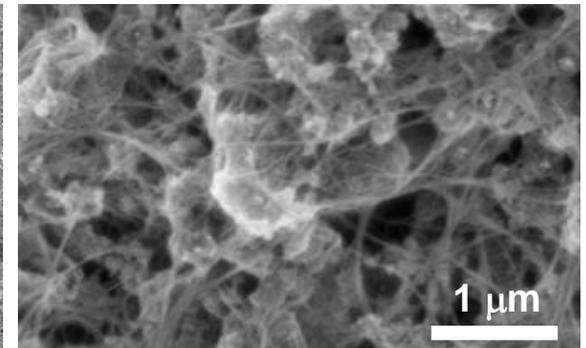
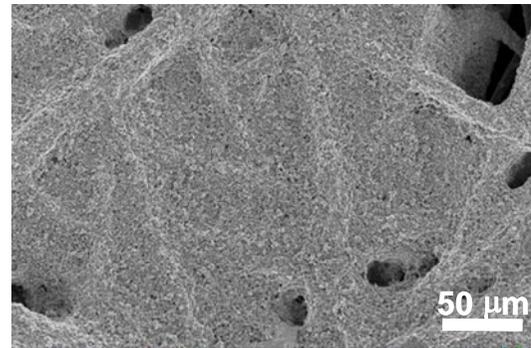
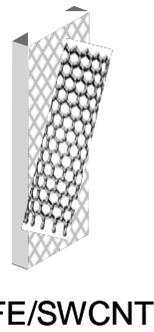
**TiO<sub>2</sub> Deposition  
On CFE**

**B**



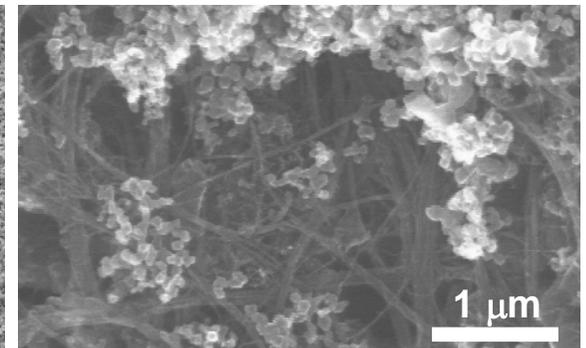
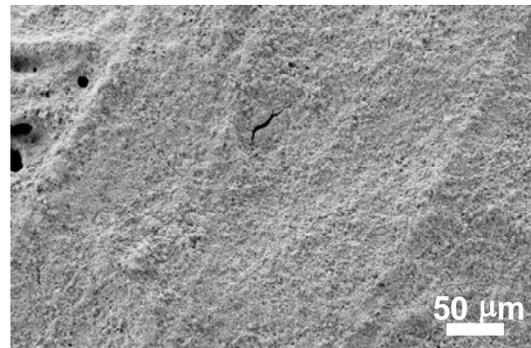
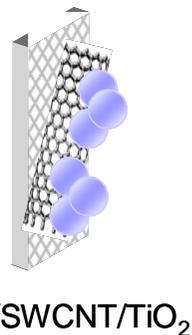
**SWCNT –Deposition  
On CFE**

**C**



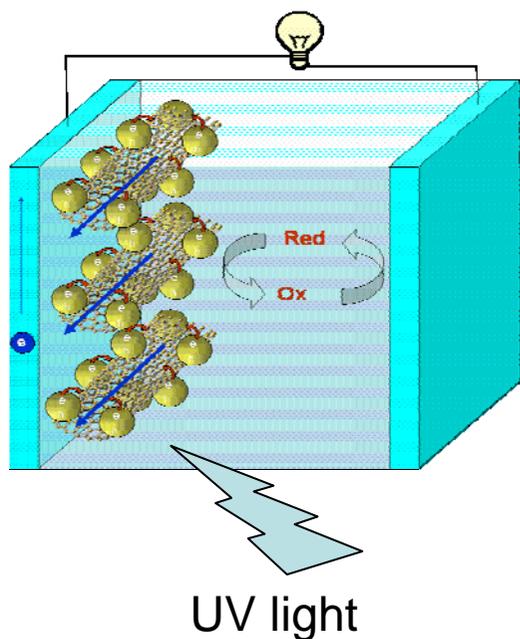
**SWCNT-TiO<sub>2</sub>  
on CFE**

**D**



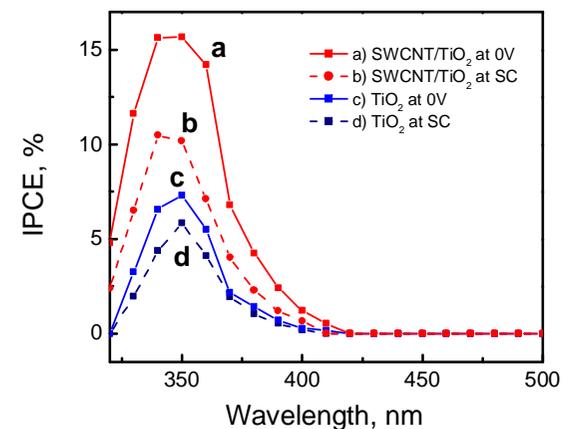
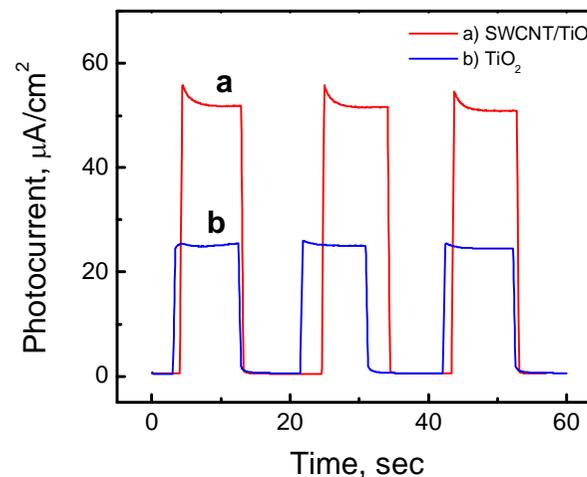
# Photocurrent Generation

## CFE/TiO<sub>2</sub> versus CFE/SWCNT -TiO<sub>2</sub>

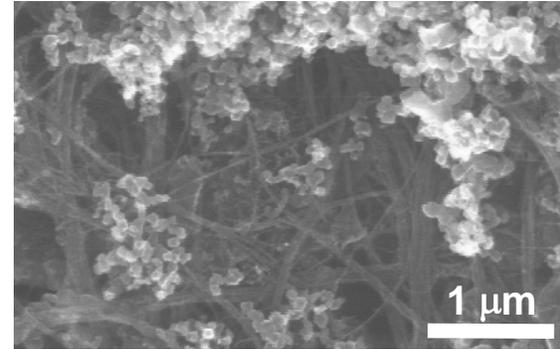
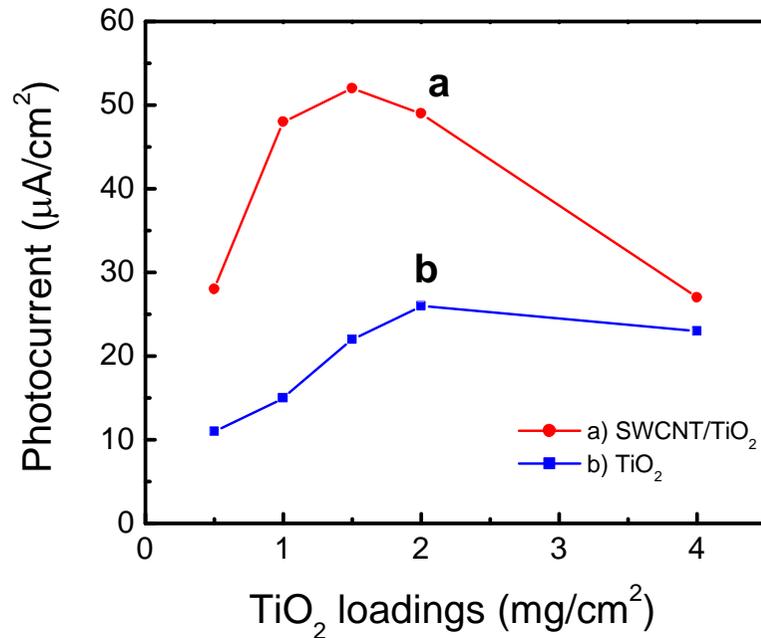


**Higher IPCE (increase of factor ~2) was observed for mesoscopic CFE/SWCNT-TiO<sub>2</sub> films**

**The results are indicative of better charge collection and transport provided by the SWCNT -Network**



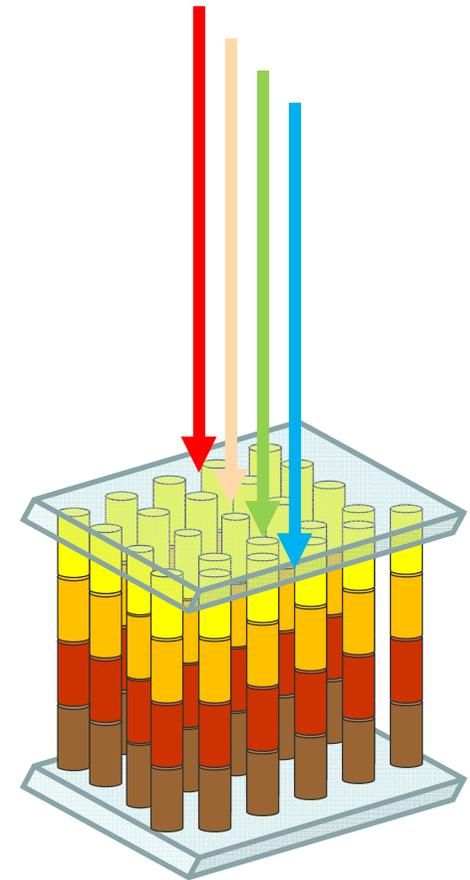
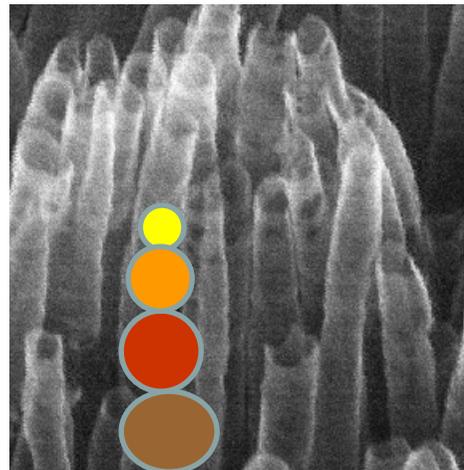
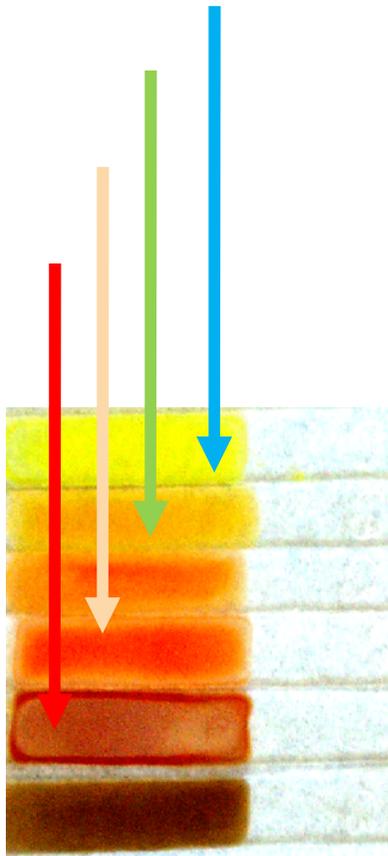
## Dependence of TiO<sub>2</sub>/SWCNT Ratio on the Photocurrent Generation



Increasing the TiO<sub>2</sub> concentration results in enhanced photocurrent as they are dispersed on SWCNT network.

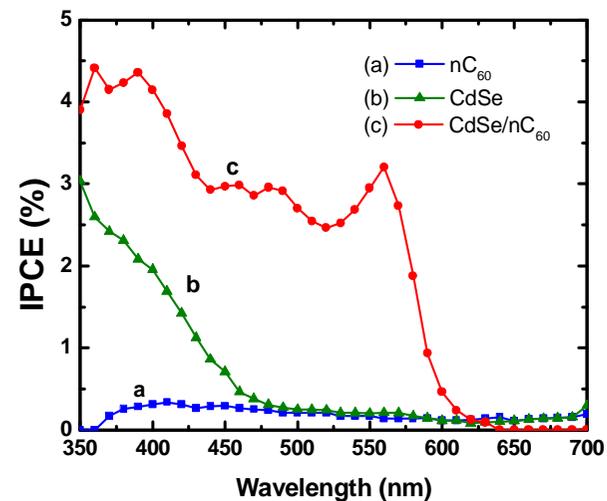
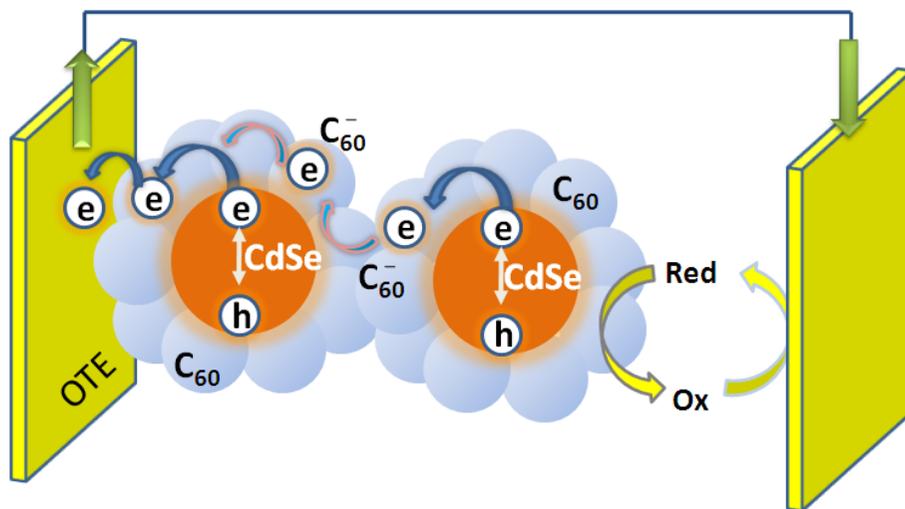
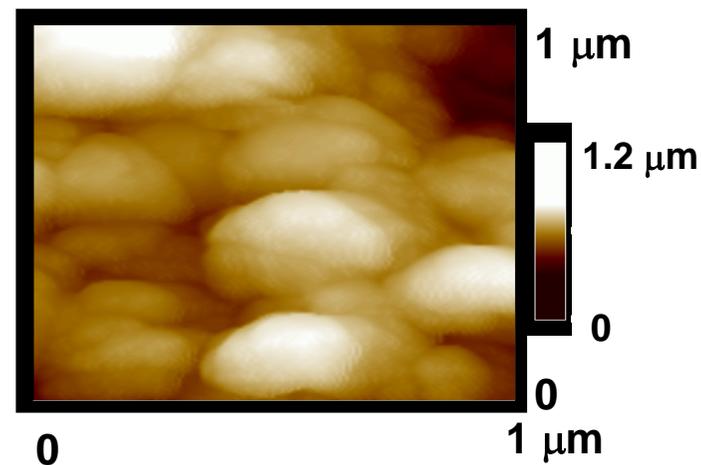
At concentrations greater than 2 mg/cm<sup>2</sup> the beneficial effect of SWCNT disappears. Under these conditions, TiO<sub>2</sub> particles aggregate and the charge recombination dominates

Where do we go from here?

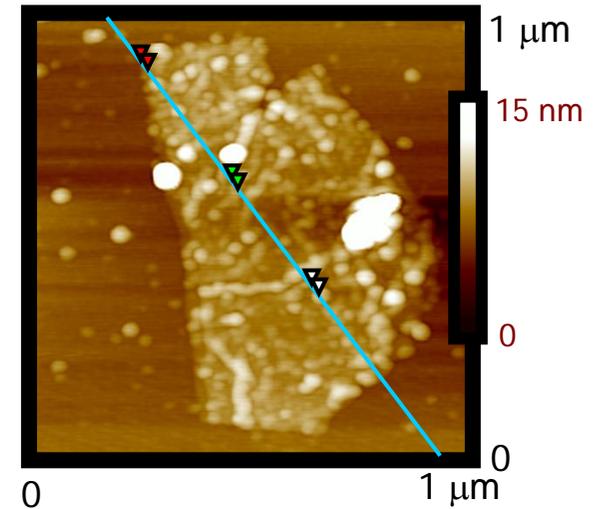
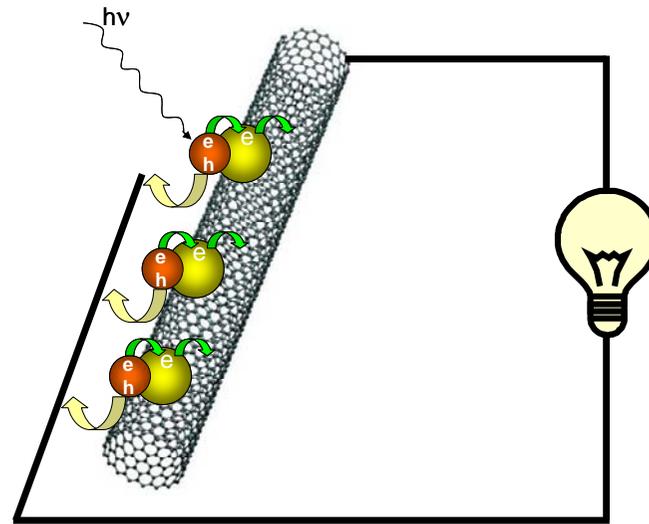


## Capping CdSe with an Electron Acceptor Shell

Electrophoretic deposition of Cluster films

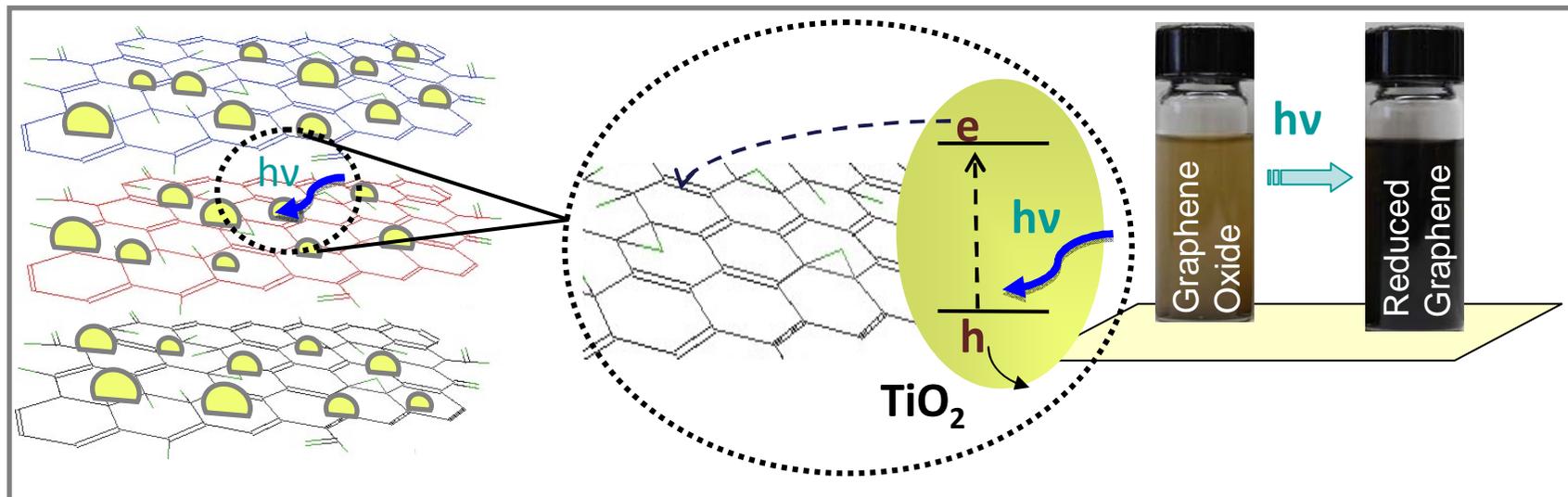


# Organized light harvesting assembly using carbon nanostructures



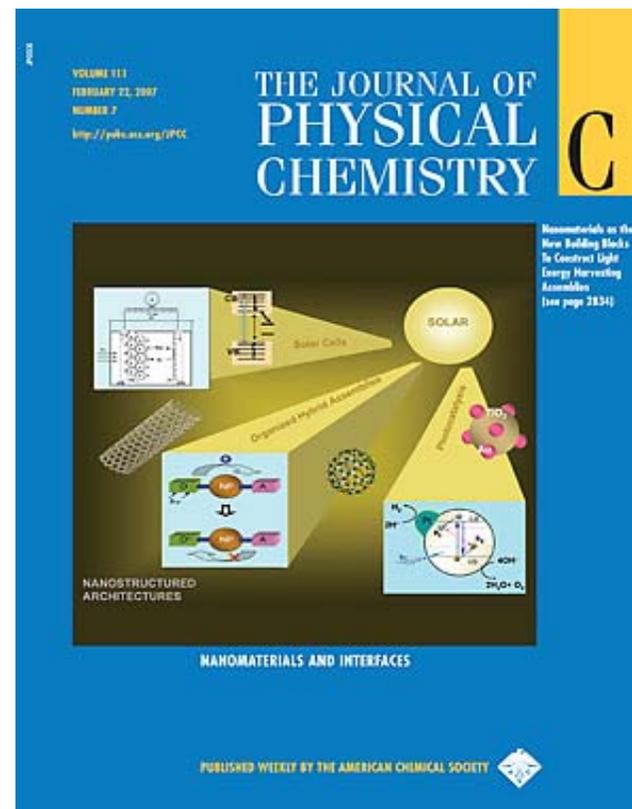
## Graphene-Semiconductor Nanocomposites

*ACS Nano, 2008, 2, 1487-1491*



# Summary

- Unique properties of quantum dots offer new opportunities to develop low-cost and high efficiency solar cells
- 1-D architectures are useful for designing next generation solar cells.
- Opportunities exist for carbon nanostructures to facilitate capture and transport of electrons in nanostructure semiconductor based solar cells.



Kamat, P. V. *Meeting the Clean Energy Demand: Nanostructure Architectures for Solar Energy Conversion (Review)*

J. Phys. Chem. C, 2007. **111** 2834 - 2860.

*Quantum Dot Solar Cells. Semiconductor Nanocrystals as Light Harvesters (Centennial Feature)* J. Phys. Chem. C 2008, **112**, in press

# Researchers/Collaborators

## Graduate students

Brian Seger (Chem. Eng.)  
David Baker (Chem. Eng.)  
Kevin Tvrdy (Chemistry)  
Clifton Harris (Chemistry)  
Matt Baker (Physics)  
Ian Lightcap (Chemistry)  
Philix Vietmeyer (Chemistry)  
Yanghai Yu (Chem. Eng.)  
Istvan Robel (Physics)

## Collaborators

Dr. K. G. Thomas (India)  
Prof. Fukuzumi (Osaka U.)  
Prof Ken Kuno (UND)  
Prof. K. Vinodgopal (IUN)

## Post-Docs/Visiting Scientists

Jin Ho Bang

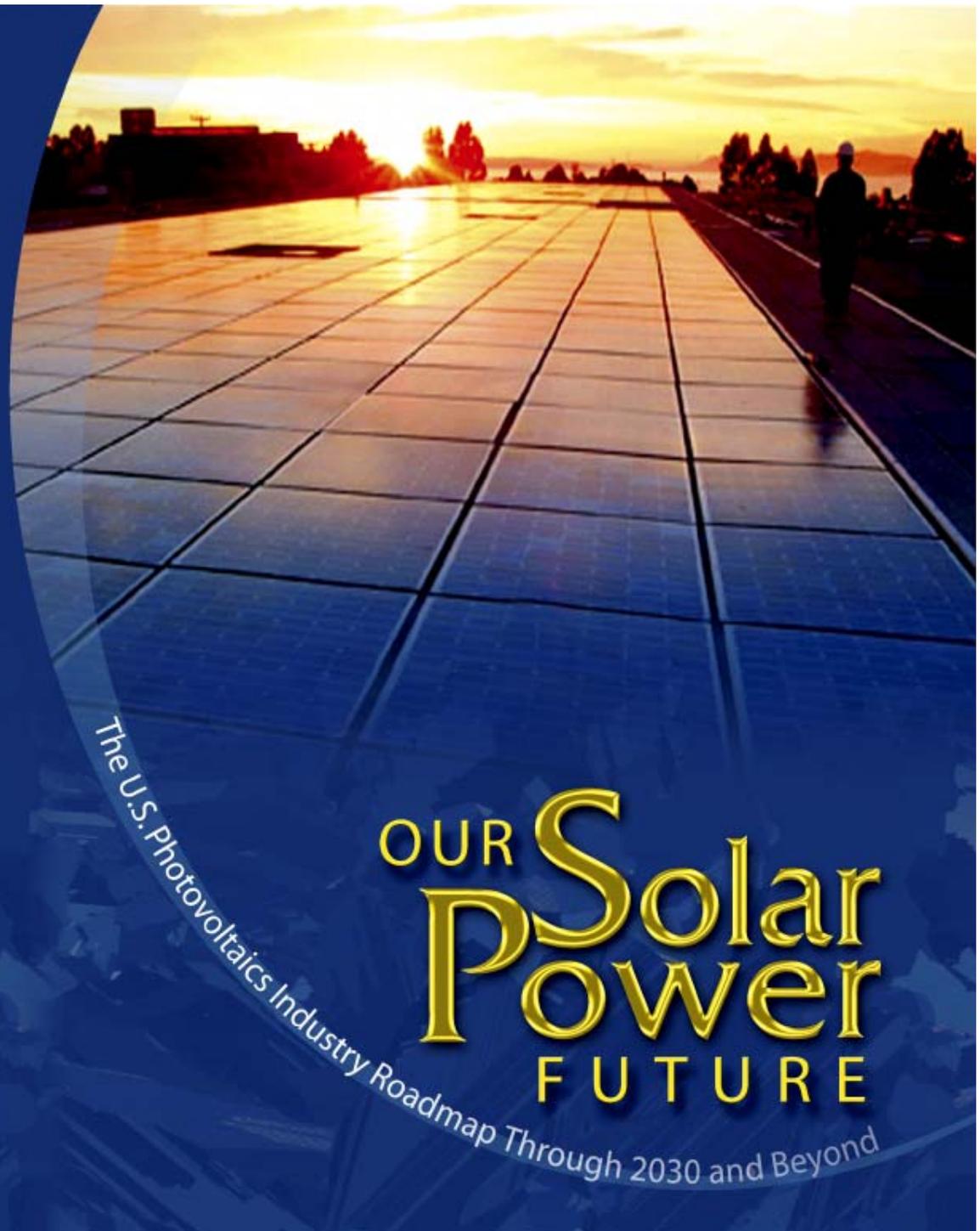
## Undergraduate students

Pat Brown  
Chris Rodriguez  
David Riehm  
Rachel Staran



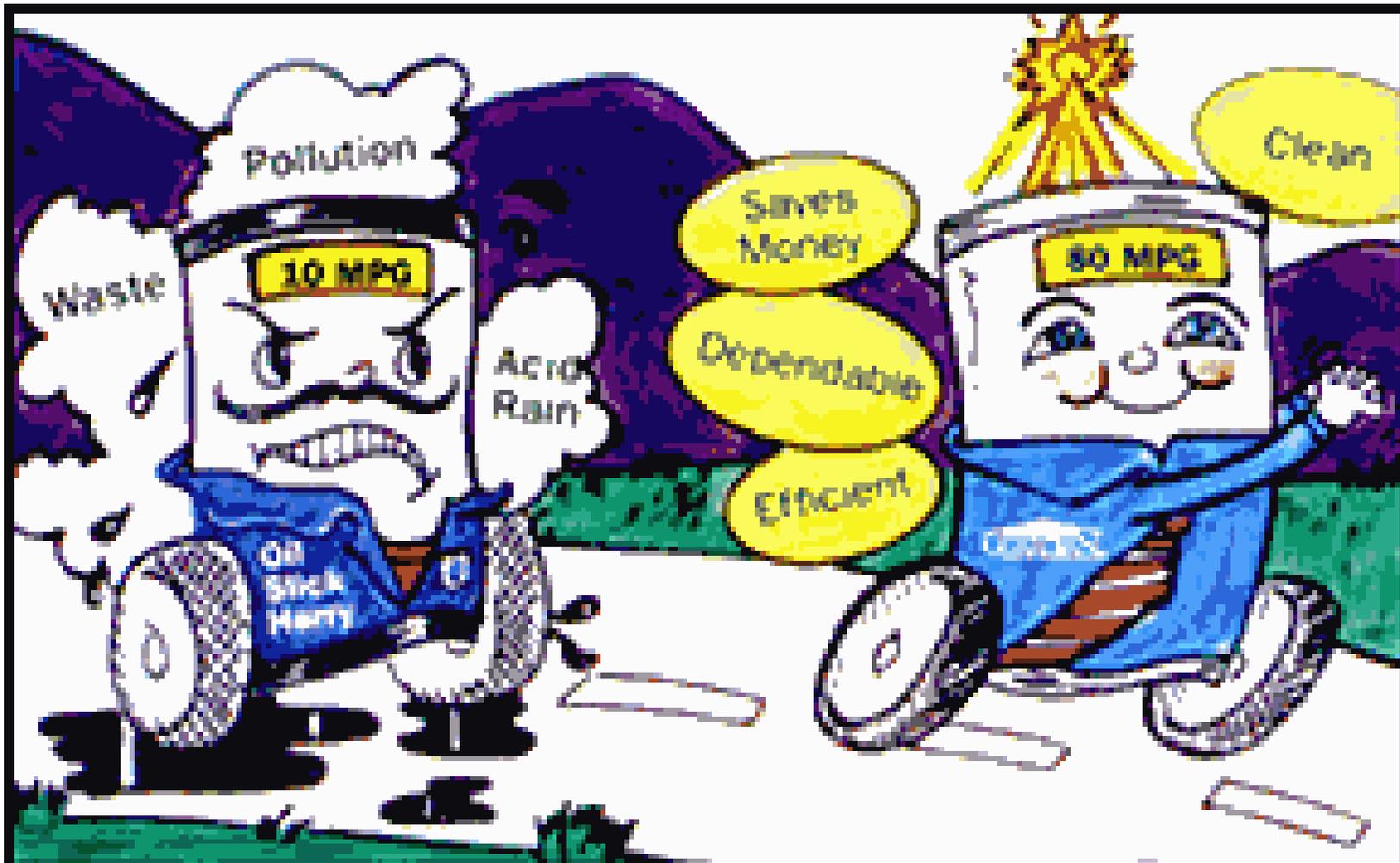
## What will the future hold?

Over the last twenty years, the per-kWh price of photovoltaics has dropped from about \$500 to nearly \$5; think of what the next twenty years will bring.



The U.S. Photovoltaics Industry Roadmap Through 2030 and Beyond

OUR Solar  
Power  
FUTURE



"Move over Oil Slick Harry. Your energy guzzling days are over. It's time for a cleaner more efficient heater like me!"

[http://www.theleveredge.com/images/isw\\_cartoon.gif](http://www.theleveredge.com/images/isw_cartoon.gif)