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**Collapse, adhesion and friction of thermoresponsive polymer brushes probed with
the surface forces apparatus**

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Collapse, Adhesion and Friction of Thermoresponsive Polymer Brushes

Lionel Bureau



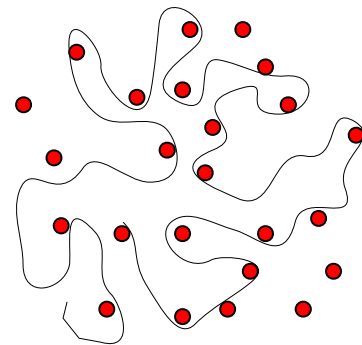
Introduction

- ⇒ Design of *smart surfaces* and interfaces
- ⇒ Alter wetting, adhesion or friction
with a change in environment conditions (T° , pH, ...)
- ⇒ Use of surface-bound *stimuli-responsive polymers*

Introduction

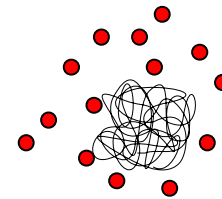
⇒ Thermoresponsive polymer layers:

polymer/solvent systems with a
Lower Critical Solution Temperature (LCST)



$T < \text{LCST}$

collapse
→
←
swelling



$T > \text{LCST}$

Occurs in mixtures with H-bonds between constituents

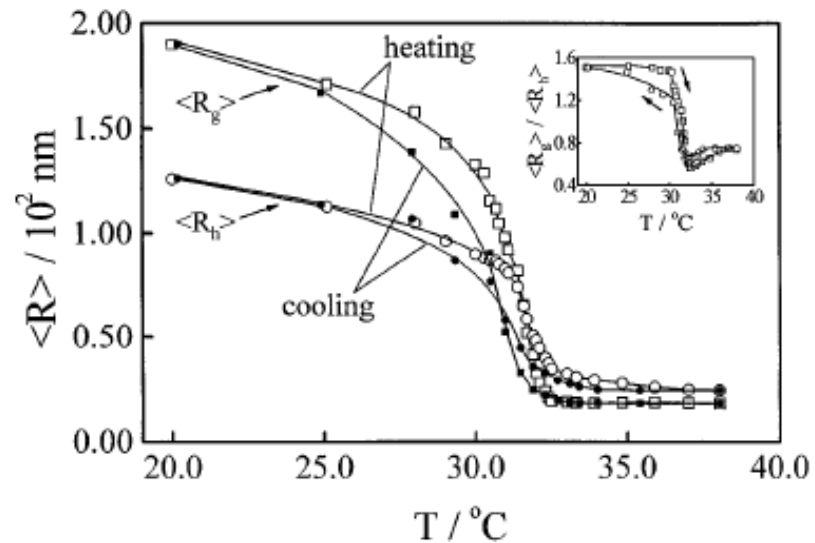
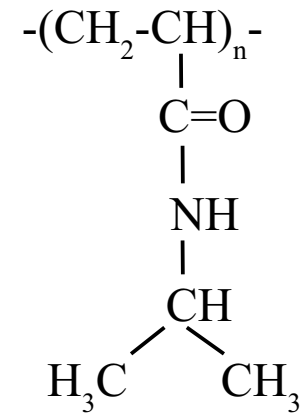
[R. E. Goldstein J. Chem. Phys. **80** (1984)]

Introduction

⇒ poly(N-isopropylacrylamide): PNIPAM

$$T_{\text{lcst}} \sim 32^\circ\text{C}$$

Sharp coil-to-globule transition
Large change in chain size

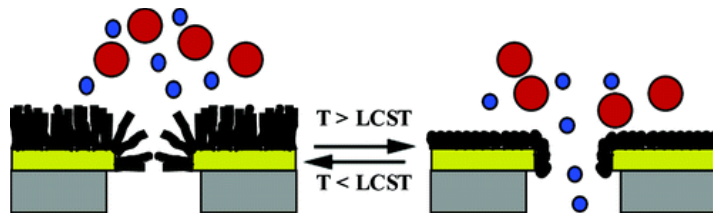


C. Wu and X. Wang, PRL **80** (1998)

Introduction

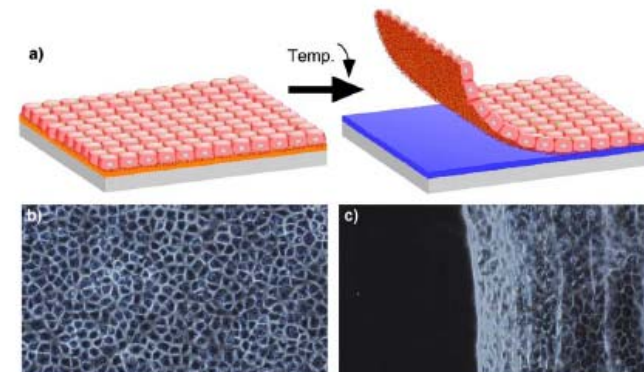
⇒ **PNIPAM grafted surfaces for:**

① actuation and flow control in micro/nanofluidics



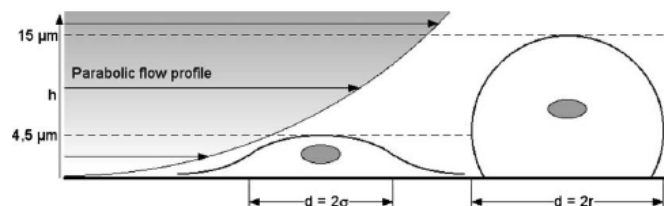
I. Lokuge *et al.* Langmuir **23** (2007)

② thermal switching of adhesion in bioengineering: protein sorting, harvesting of cultured cells



A. Kikuchi and T. Okano, J. Cont. Rel. **101** (2005)

③ cell detachment in microfluidics...



O. Ernst *et al.* Lab Chip **7** (2007)

Introduction

⇒ *Questions:*

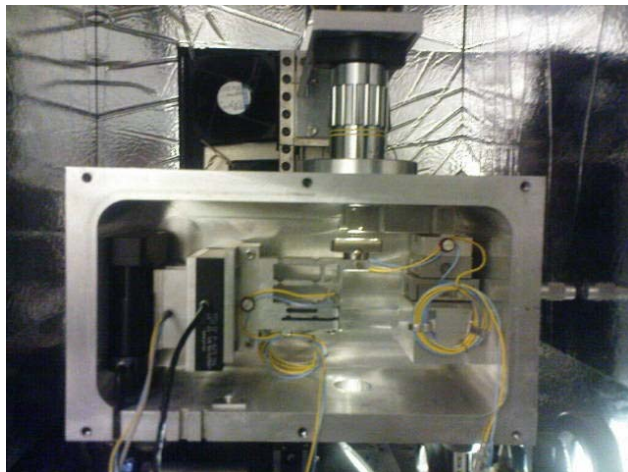
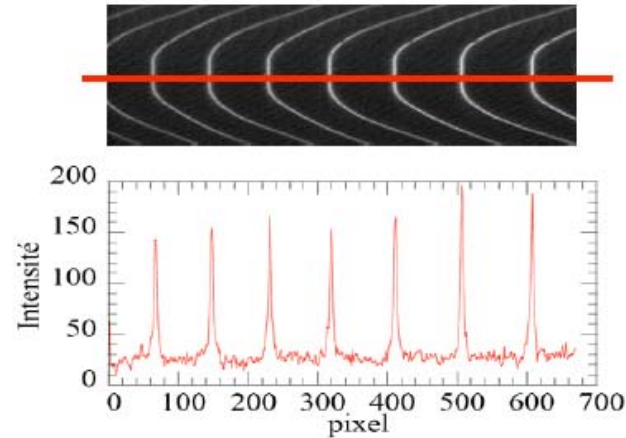
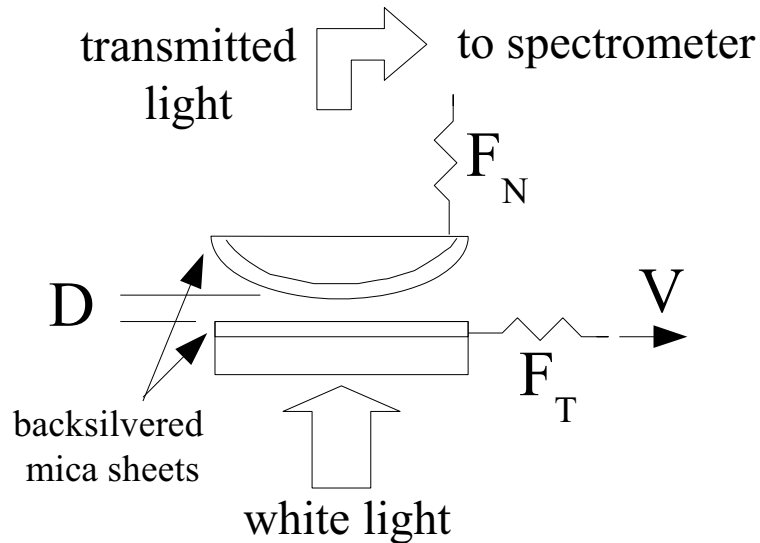
- ❶ effect of chain tethering on coil-globule transition ?
- ❷ influence of molecular structure on the properties of thermoresponsive layers ?

⇒ *Probe the collapse of PNIPAM brushes with the SFA**

* D. Leckband *et al.*, *Langmuir* **23** (2007) – **22** (2006)

I. B. Malham and L. Bureau, submitted to *Langmuir* (2009), preprint <http://fr.arxiv.org/abs/0909.3709>

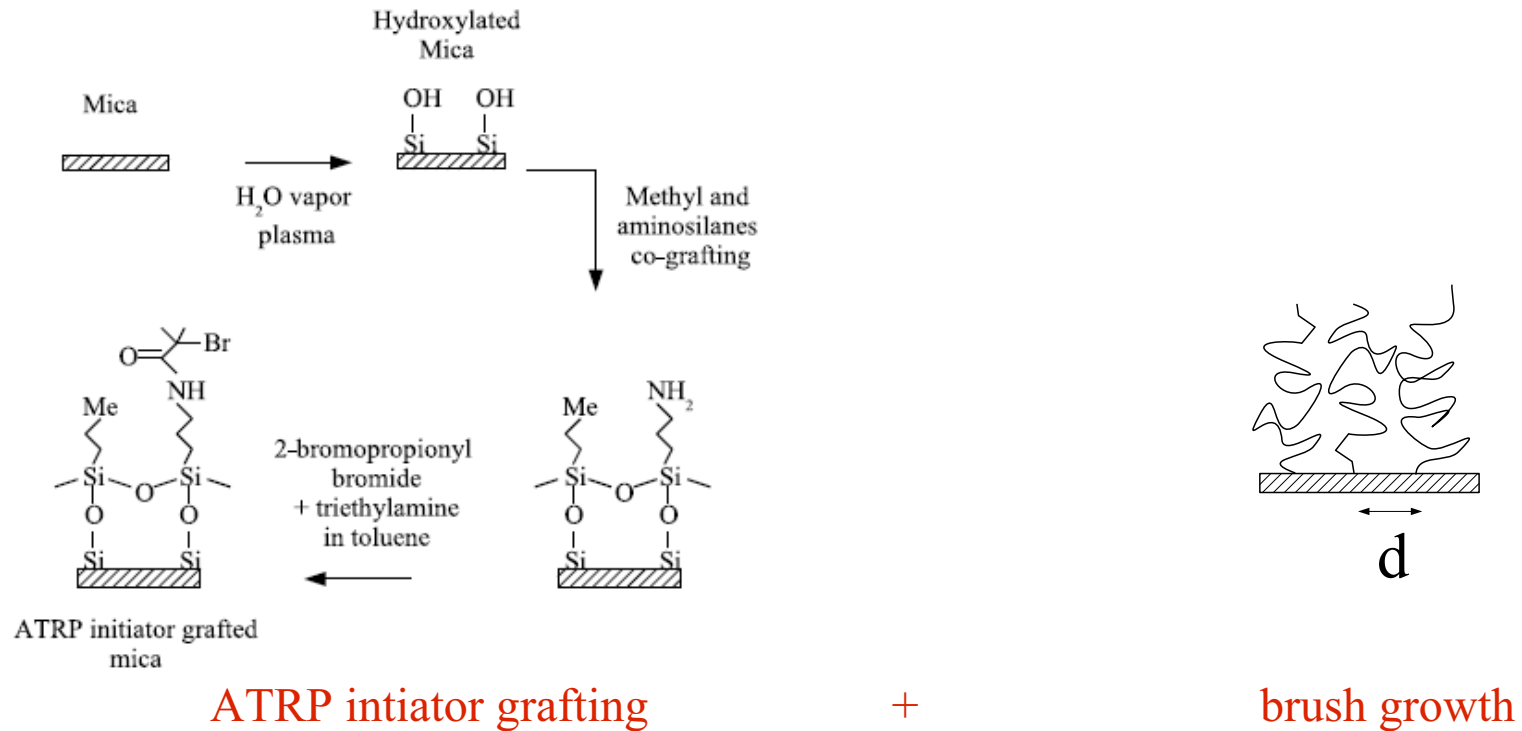
Surface Forces Apparatus



- D measured by MBI @ 30Hz
- F_N measured independently of D
- Shear amplitude up to 500 μm
- Feedback on F_N or D while shearing

Surface Chemistry

Parker et al. J. Phys. Chem. **93** (1989)

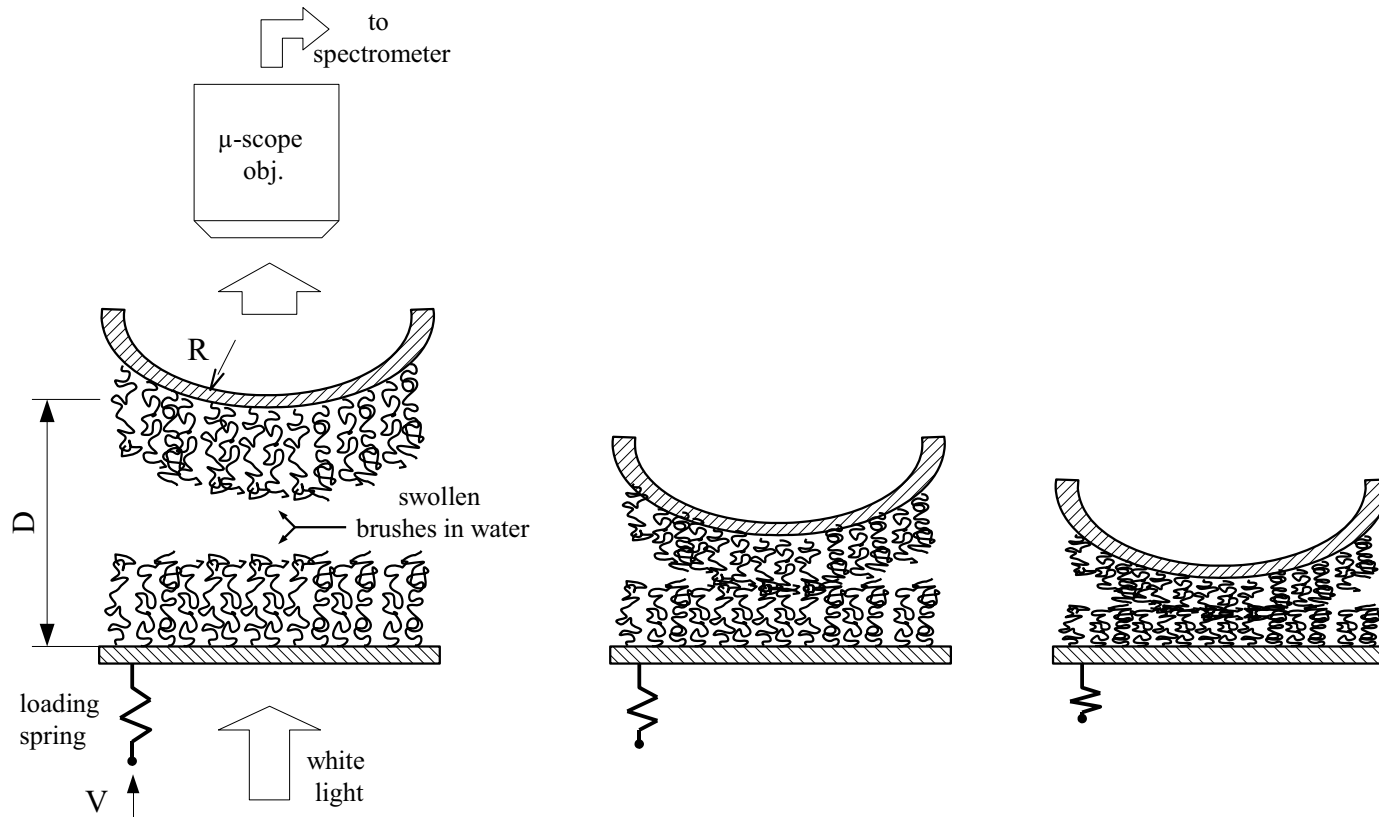


Control parameter: grafting density ($1/d^2$)

$$6 \leq R_F/d \leq 30$$

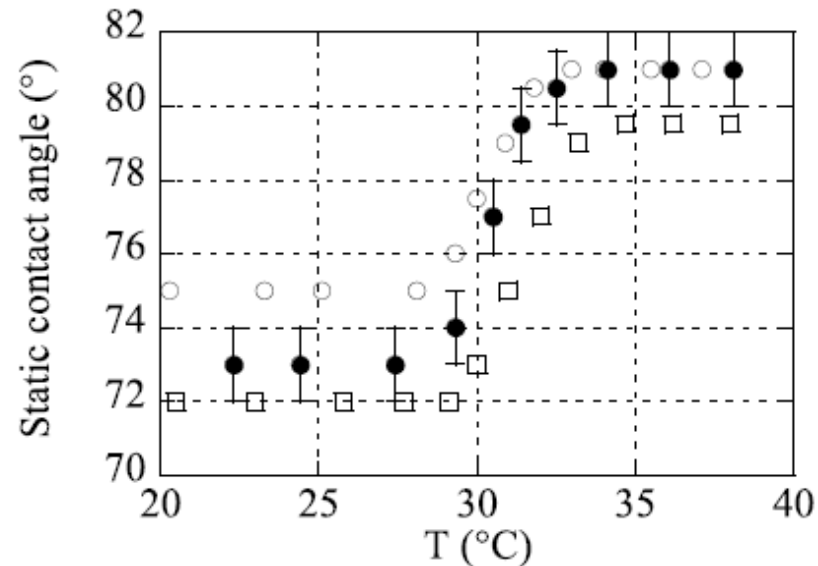
⇒ Stretched chains

Experimental configuration



Compression/adhesion of two identical brushes as a function of temperature

Temperature-dependent wetting

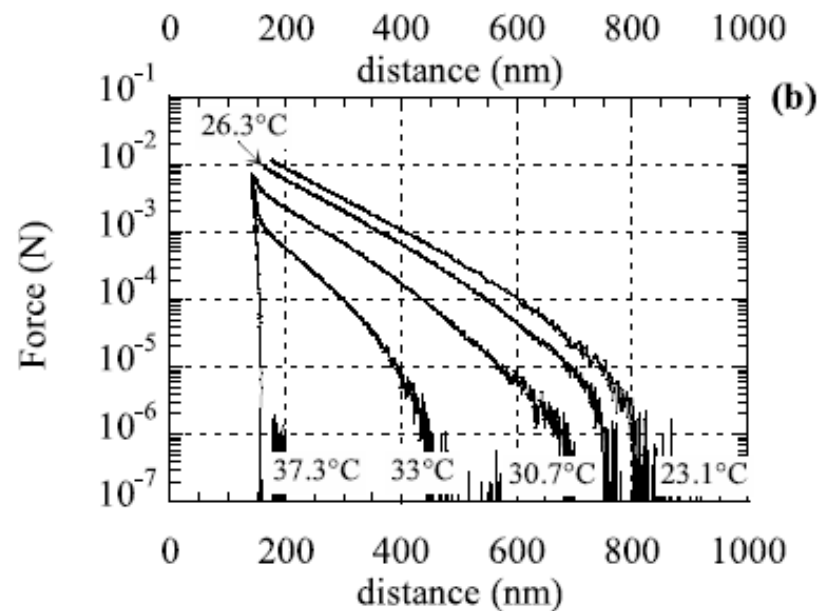
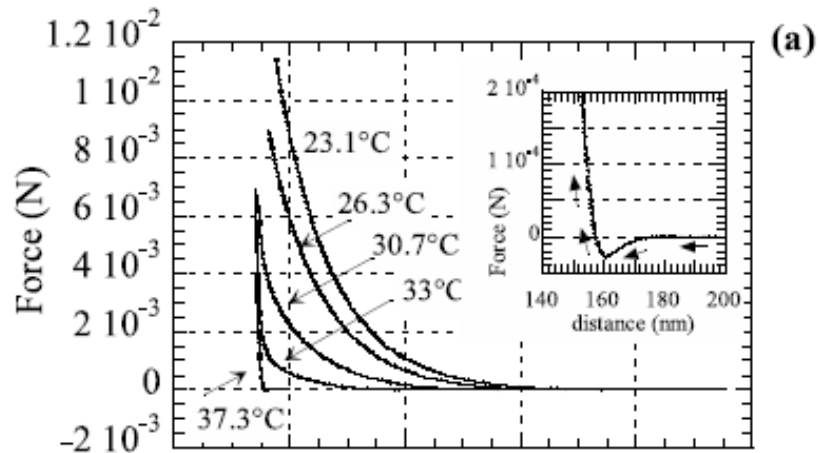


(□) 0.02 ch/nm² - (●) 0.24 ch/nm² - (○) 0.42 ch/nm²

- ① Increase in water contact angle between 30-34°C
- ② Weak sensitivity to grafting density

⇒ Outermost region of brushes exposes CH₃ groups at high T°

Compression

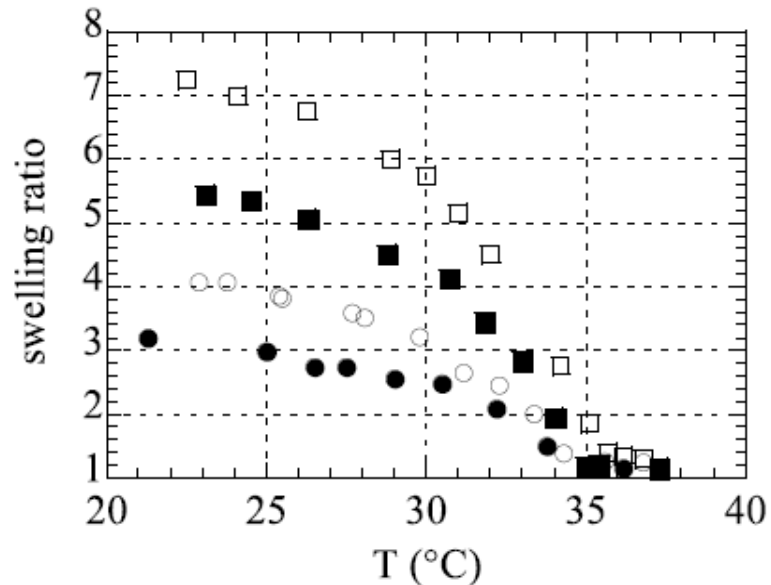


- ① Range of repulsive forces \searrow when $T \nearrow$
 \Rightarrow *collapse of brushes*
- ② At $T > 34^\circ\text{C}$:
attractive interaction upon approach
 \Rightarrow *hydrophobic forces ?*

Density-dependent collapse

Swelling ratio:

$$\alpha = h_{\text{swell}} / h_{\text{dry}}$$



(□) 0.02 ch/nm² - (■) 0.14 ch/nm² - (○) 0.24 ch/nm² - (●) 0.42 ch/nm²

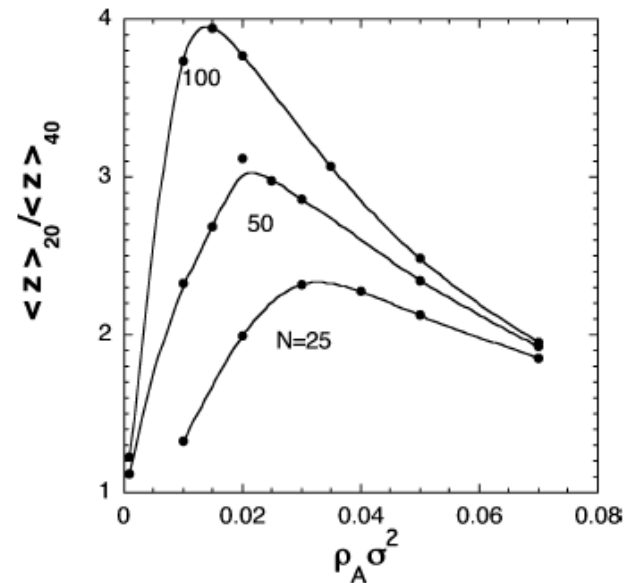
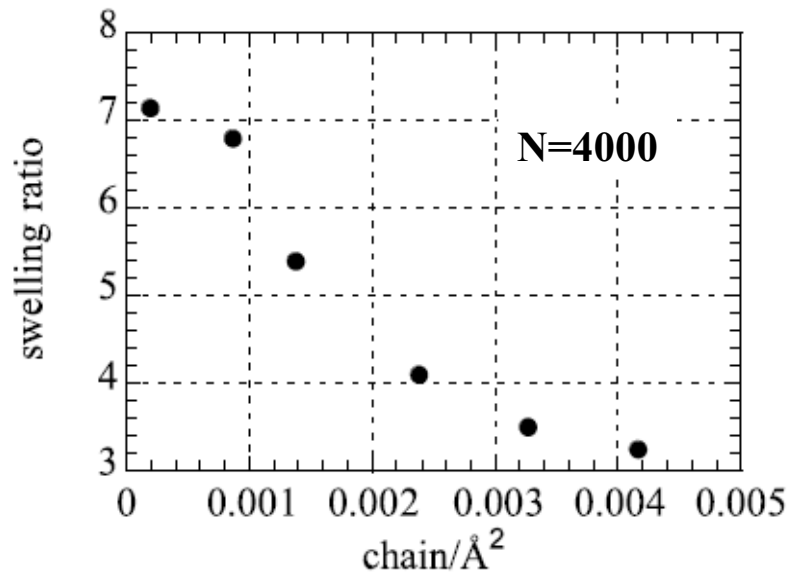
① Collapse of grafted chains broader than free dilute chains

⇒ Agreement with Neutron Reflectivity results [Yim *et al*, *Macromolecules* **39** (2006)]

⇒ Ok with theory by Zhulina *et al.* [*Macromolecules* **24** (1991)]

Density-dependent collapse

② Less dense brushes swell more

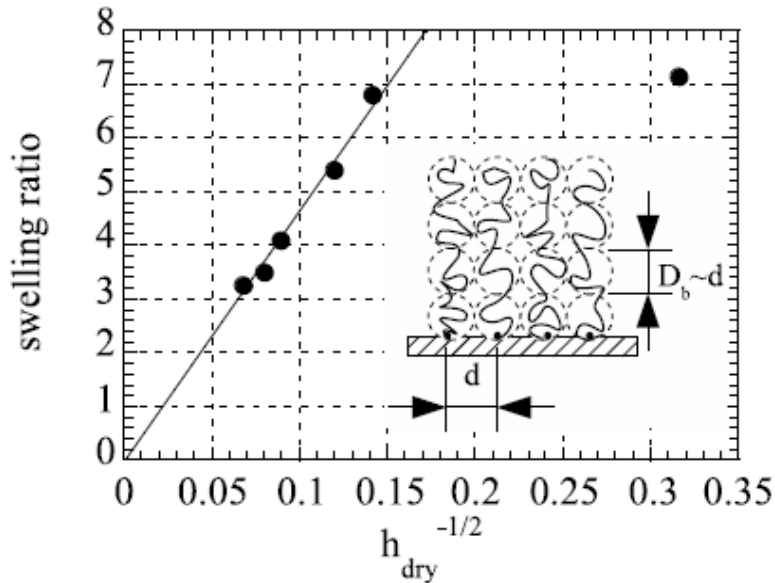


⇒ Qualitative agreement with SCF calculations

[Mendez et al. *Macromolecules* **38** (2005)]

Density-dependent collapse

Alexander-de Gennes scaling ?



$$D_b = g^{\nu} a \quad \nu=3/5 \text{ for semi-dilute blobs}$$

$$h_{\text{swell}} = Na \left(\frac{d}{a} \right)^{1-1/\nu}$$

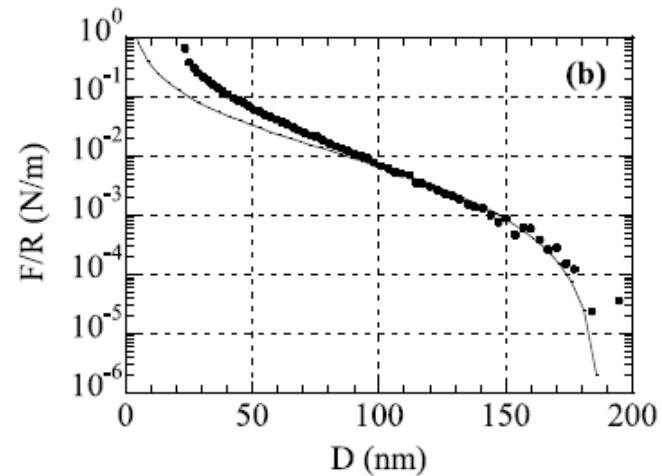
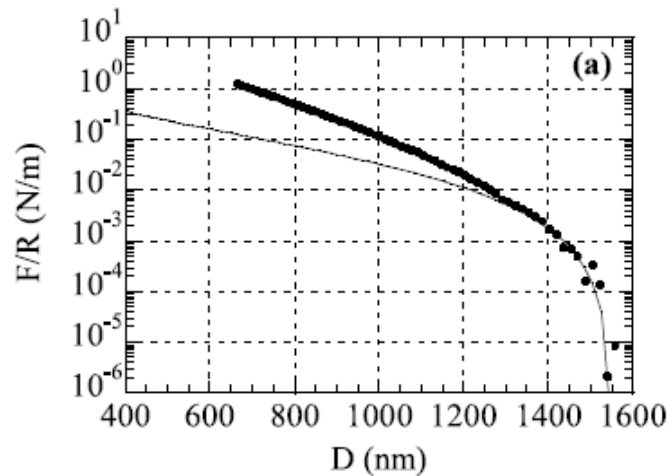
$$h_{\text{dry}} d^2 = Na^3$$

$$\alpha = \frac{h_{\text{swell}}}{h_{\text{dry}}} = \left(\frac{d}{a} \right)^{3-1/\nu}$$

$\nu \approx 1/2 \Rightarrow$ *validity of semi-dilute assumption ?*

... already observed for dense brushes [S. Yamamoto *et al.*, *Macromolecules* **33** (2000)]

Repulsive forces



Alexander-de Gennes prediction
for brush compression:

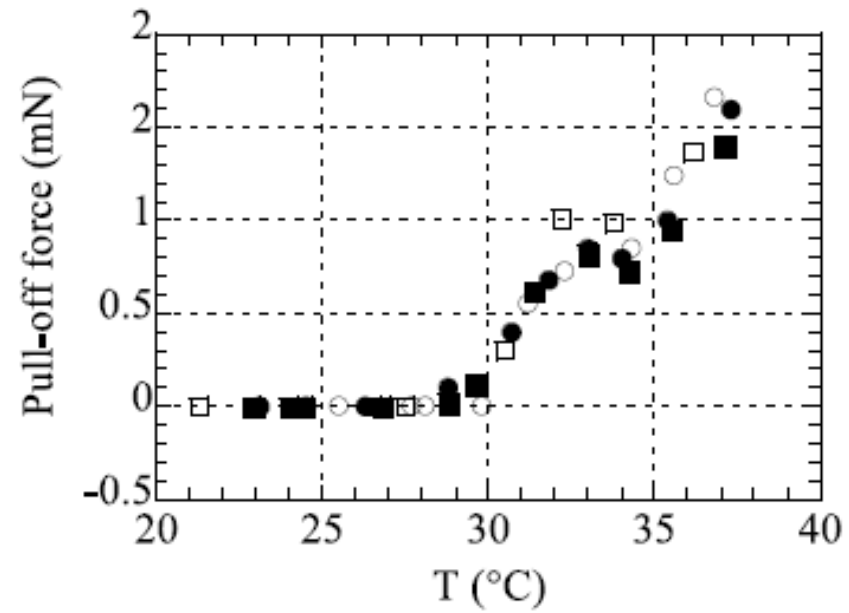
$$\frac{F(D)}{R} = C \left[7 \left(\frac{D}{2L} \right)^{-5/4} + 5 \left(\frac{D}{2L} \right)^{7/4} - 12 \right]$$

Ok at low brush compressions, but...

Poor description under strong compression

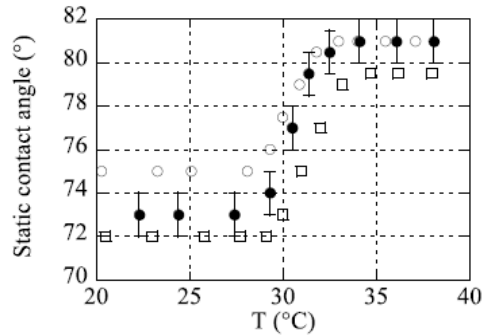
⇒ semi-dilute assumption breaks down when
the monomer volume fraction is ≥ 0.3

Adhesion

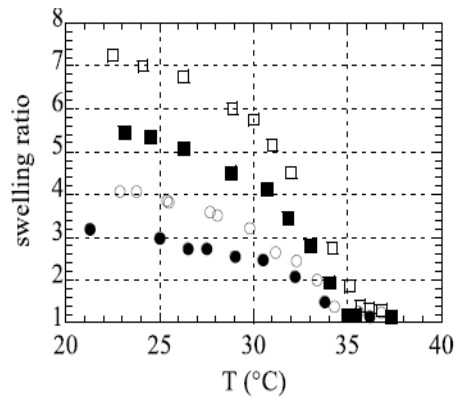


Pull-off force insensitive to grafting density

Adhesion



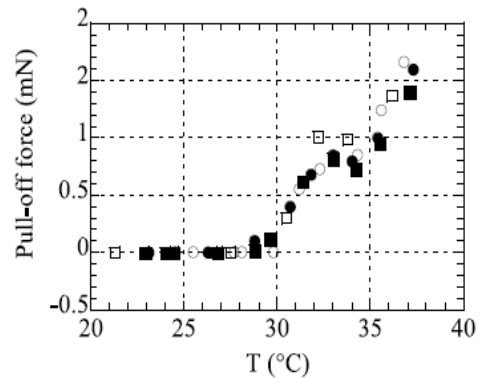
Adhesion increases above 35°C
whereas contact angle plateaus...



① Brush collapse accompanied by exposure of -CH₃ at the polymer/water interface

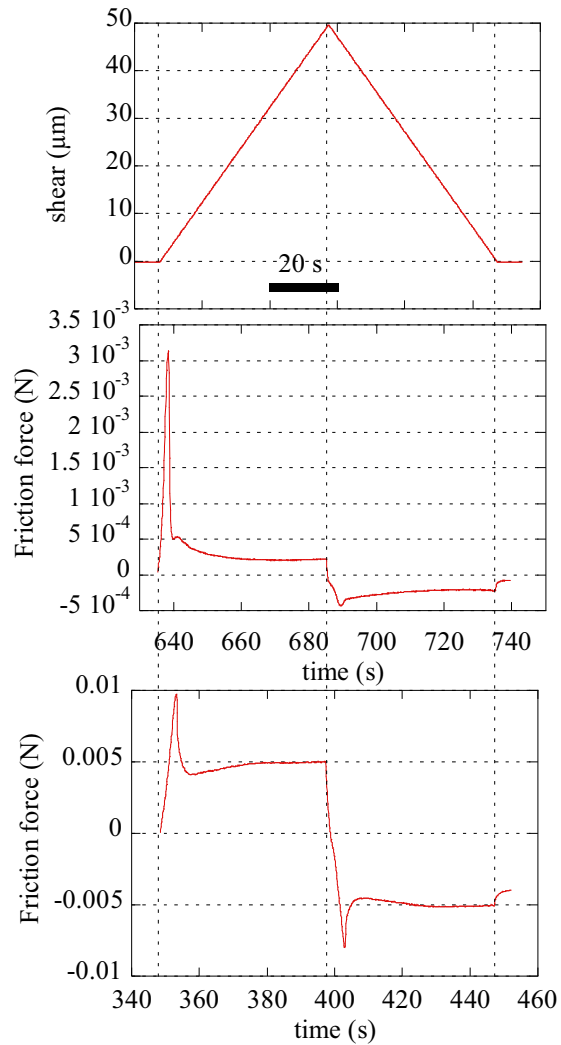
But...

② Another mechanism at play for adhesion:
H-bonds formation ?



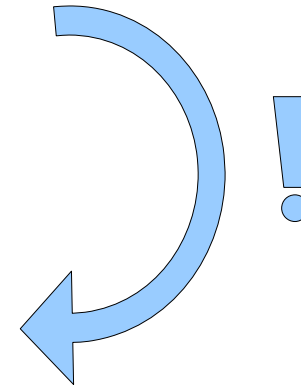
Friction

$$F_N = 3 \cdot 10^{-3} \text{ N}$$



$$T=26^\circ\text{C}: \mathbf{F_{steady} = 2 \cdot 10^{-4} \text{ N}}$$

$$T=35^\circ\text{C}: \mathbf{F_{steady} = 5 \cdot 10^{-3} \text{ N}}$$



Conclusions

- ❶ Dense polymer brushes probed with the SFA
- ❷ Extensive study of the collapse of thermosensitive brushes
⇒ Test of polymer brush theories
- ❸ Intriguing temperature dependence of brush/brush adhesion
⇒ Role in bioadhesion ?
- ❹ Huge effect of temperature on PNIPAM friction