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Friction measurements in the single-molecule limit

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Friction measurements in the single-molecule limit

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Co-workers:

- Cambridge (Expt/simulation): J Ellis, AP Jardine, H Hedgeland, AR Alderwick, D Ward, B Lechner, F Tuddenham.
- Tel Aviv (Expt): G Alexandrowicz
- Grenoble (Neutron spin-echo): P Fouquet
- Milano (Calculation): GP Brivio, G Fratesi (Alkali adsorption)
- Genova (Theory): R Ferrando (CO/Pt(111))
- Rutgers (Expt): BJ Hinch (Cp⁻/Cu(111))

Friction from adsorbate dynamics



Motion induced by thermal excitation

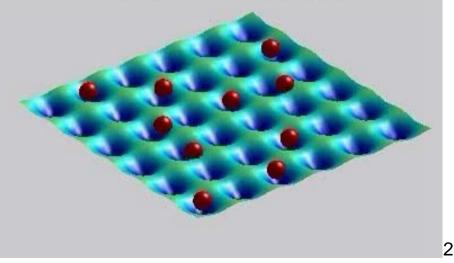
- Modelled as "Adsorbate plus heat bath"
- Langevin equation:

$$m\ddot{\mathbf{r}}_{i} = -\nabla V(x, y) - \eta m\dot{\mathbf{r}}_{i} + \xi_{i}(t) + \sum_{i \neq j} \mathbf{F}_{ij}$$

Three aspects to the motion:

- Energy Landscape
 - STATIC FRICTION.
- Atomic scale friction
 - KINETIC FRICTION.
- Interactions
 - Co-operative effects.

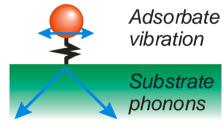
Na/Cu(001), 200K, t =0 ps



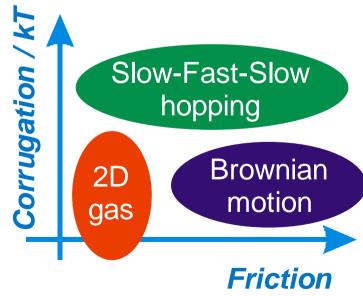


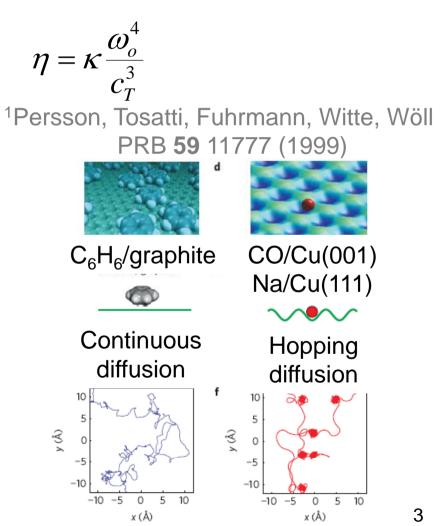
Connection through frustrated translation.

Vibrational damping¹.



Dynamic regimes:





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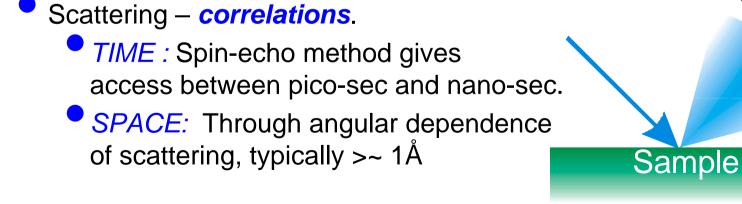
Measurements

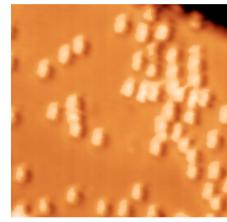
Response

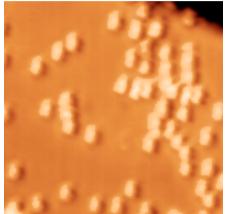
Microscopy – *snapshots*.

Typically long time-scales $> \sim 1 \text{ msec}$

CO/Cu(111) Bartels et al. JCP 123 201102 (2005)







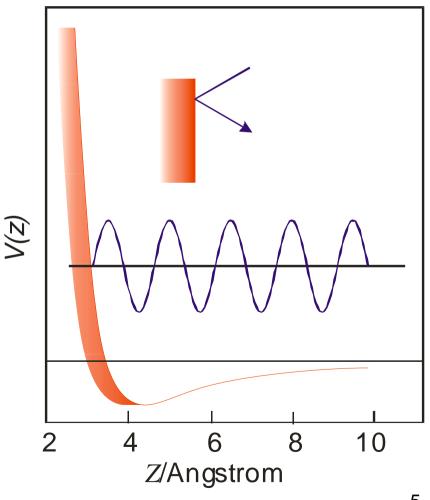


 ΔK



Helium-surface interaction

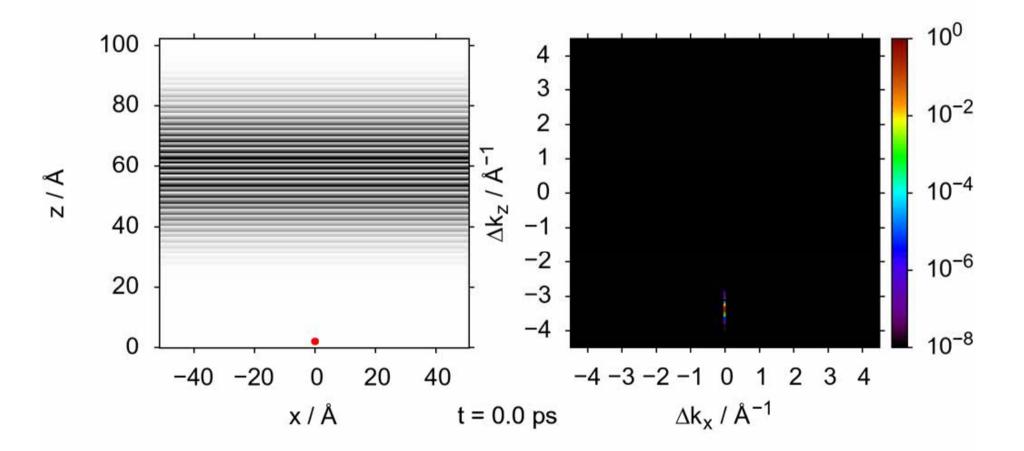
- Classic *van der Waals* form
- Well-depth <10meV
- Incident energy <50meV
- Collision time
 - ~ a few pico-seconds



Elastic scattering



Static target with a single adsorbate

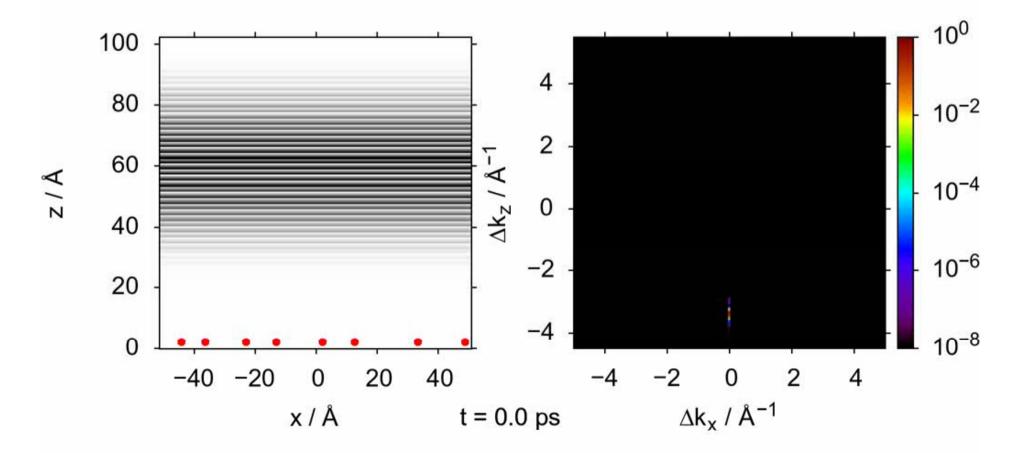


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Quasi-elastic scattering

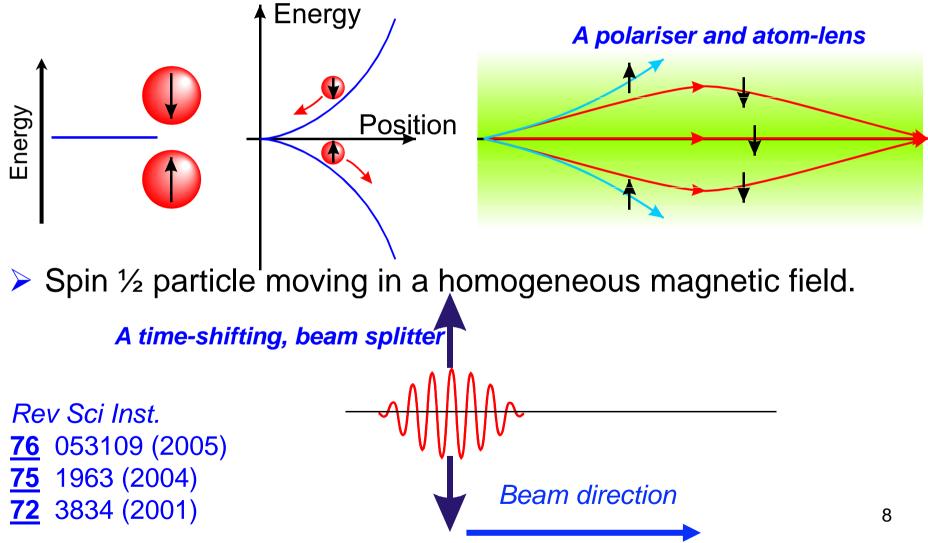


Surface with several moving atoms





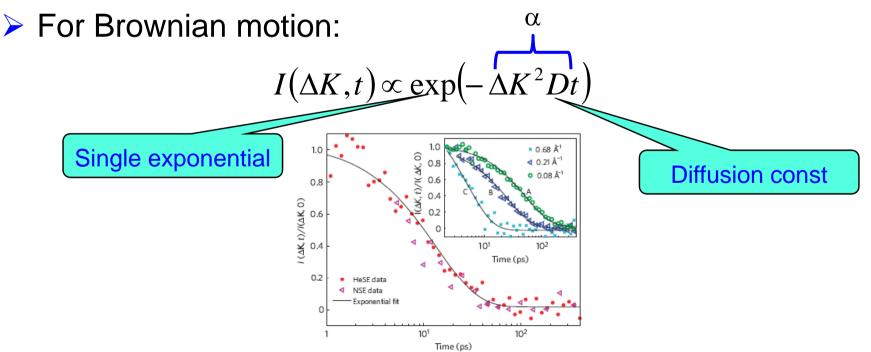
> Spin $\frac{1}{2}$ particle in an inhomogenous magnetic field.



Intermediate scattering function



- The spin-echo method gives the ISF
 - It is a measure of decay, with time, of the structural correlation at the surface.
 - Temporal correlation is measured directly
 - Spatial correlation measured (indirectly) through the dependence on ΔK .



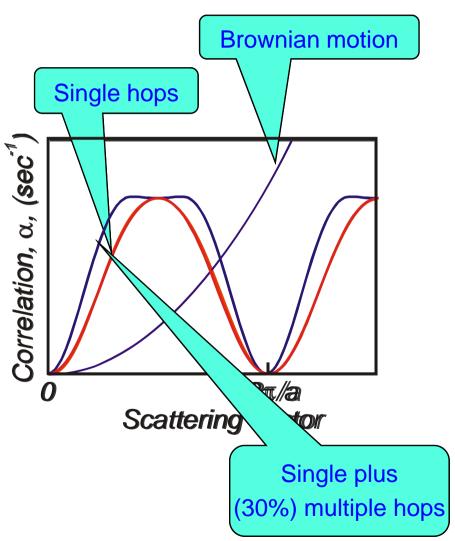


Analytic models - jump dynamics

Decay rate of the correlation has an analytic form for simple dynamics.

$$\alpha(\mathbf{K}) \propto v \sum P_j \sin^2 \left(\frac{\mathbf{K} \cdot \mathbf{r}_j}{2} \right)$$

- Such behaviour is only observed for independent particles (typically at low coverage).
- The "shape" changes drastically, with coverage, if cooperative motion occurs (eg: repulsive, pairwise forces)



Case studies

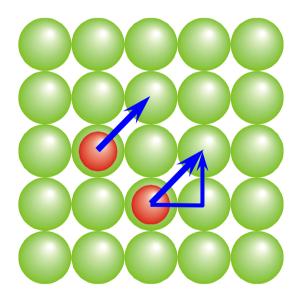


- Landscape (static friction)
 - CO/Cu(001) at low coverage.
- Intermolecular forces
 - Do repulsive interactions inevitably result in pairwise forces?
 - Comparison of Na/Cu(001) and CO/Pt(111).
- Kinetic friction
 - Friction in weak and strongly chemisorbed molecular systems;
 - Comparison of C_6H_6 /Graphite and C_5H_5 /Cu(111).



> CO/Cu(001):

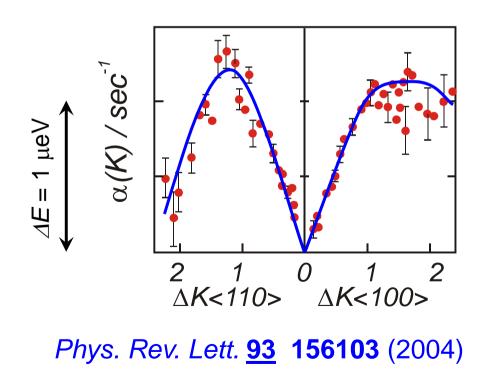
an early measurement to determine the energy landscape

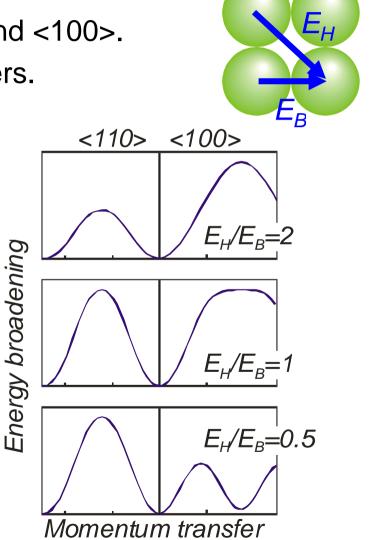


Science 304 1790 (2004)

Energy Landscape – CO / Cu(001)

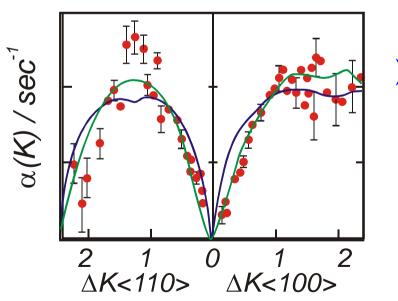
- Similar broadening along <110> and <100>.
- Analytic picture gives similar barriers.
- Confirmed by full MD-simulation.







- Full MD simulations confirm the landscape
- > Friction:
 - Best "fit" **tollowippijnfoiotate**n h**as/txin**getijc=fr(it/t2)nps⁻¹ (Scaling nyeedled) (1/13) ps⁻¹.



- Fewer long jumps observed.
- Either:
 - Friction is position dependent,
 - Molecular degrees of freedom are involved in the dynamics.

 η = (1/8) ps⁻¹ from the T-mode linewidth. (Graham et al JCP <u>104</u> (1996) 5311)

Adsorbate interactions



Pairwise forces

Na/Cu(001)

Heat of adsorption very strongly coverage dependent.

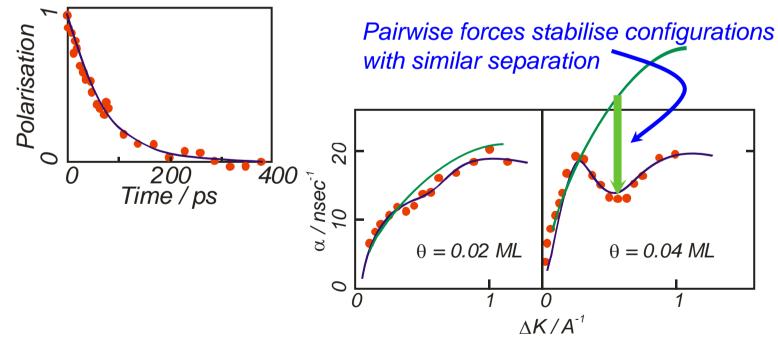
Strong dipole repulsion.

CO/Pt(111)

- Heat of adsorption decreases strongly with coverage.
- Attributed to pairwise forces (Consensus on the origin or magnitude of these forces is absent)

Alkali metal dynamics: Na/Cu(001)

- Low coverage data gives the landscape, as before.
- Medium coverage data shows the effect of interactions.
- Typically the polarisation follows an exponential decay at low and moderate coverage (but not at high coverage)

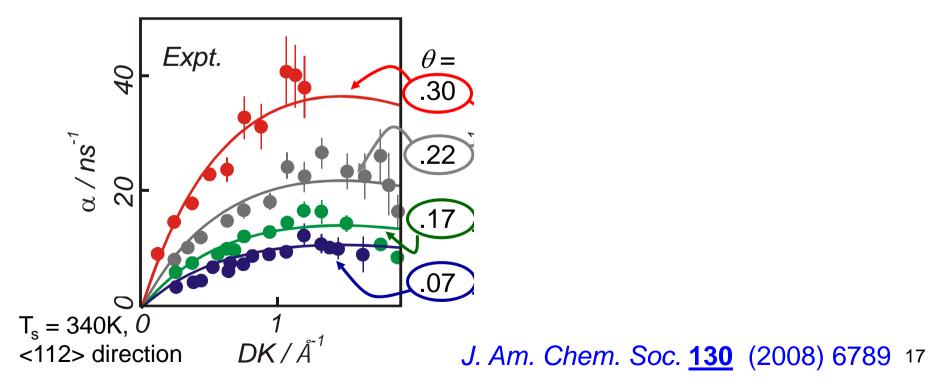


Phys. Rev. Lett. 97 156103 (2006)

CO interactions: CO/Pt(111)



- Uncorrelated motion with strongly interacting adsorbates !
 - Hopping rate increases with coverage, but with fixed K-dependence.
- No force law (eg ~1/r³) can simultaneously explain heat of adsorption and the dynamics. See simulation
 - Pairwise interactions ~1/10 of the expected magnitude.







> Kinetic friction in molecular systems.

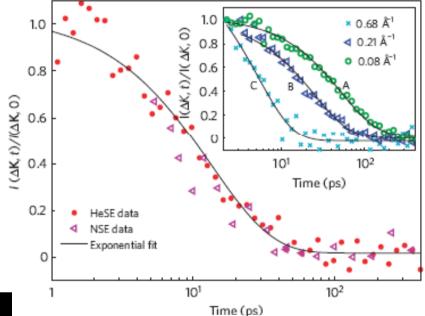
> What factors determine the kinetic friction?

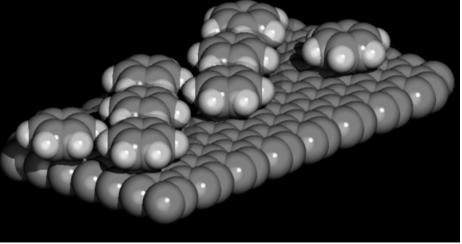
- Strength of the adsorbate-substrate bond.
- Magnitude of the corrugation in the energy landscape (static friction)

Nano-scale friction



- Single-molecule friction
 - Benzene on graphite
- Special features
 - Negligible corrugation
 Arrenhius plot
 - Strong friction
 - Brownian motion



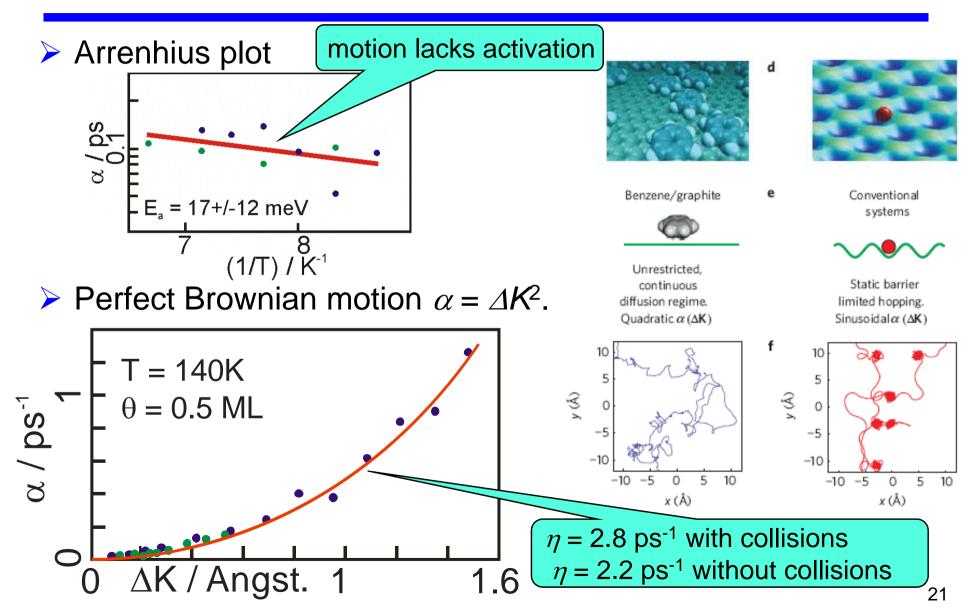


Coverage = 0.5ML Temperature 140K

> Nature. Phys. (2009) doi: 10.1038/NPHYS1335 20

Benzene on graphite





Summary



> The helium spin-echo method:

- Uniquely sensitive to surface motion for pico-sec to nano-sec regime.
- Static and dynamic effects in simple, well characterised systems can be measured precisely.
- Pairwise forces between adsorbates are not necessarily obvious.
- Strong friction seen in an weakly adsorbed, un-corrugated system:
 C₆H₆/graphite.
- Similar friction in a strongly adsorbed, corrugated system:
 - C₅H₅-/Cu(111).

Recent perspective article: PCCP. <u>11</u> (2009) 3355

> Open questions:

- Molecular origin of friction white noise approximation
- Role of internal, molecular, degrees of freedom (CO/Cu(001))

etc.....



THE END