



2139-34

#### School on Synchrotron and Free-Electron-Laser Sources and their Multidisciplinary Applications

26 April - 7 May, 2010

IR spectroscopy and microscopy in material science Electrodynamics at high pressures of strongly correlated electron systems

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IR spectroscopy and microscopy in material science Electrodynamics at high pressures of strongly correlated electron systems

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# Outline - Selected applications of IRSR in solid state physics

- IR light and IR synchrotron radiation
- Basics of IR spectroscopy in condensed matter physics
  - 1. From Reflectivity to Optical Conductivity
  - 2. The Lorentz-Drude Model
- High-pressure measurements
- Strongly Correlated Electron Systems
  - 1. Pressure-Induced Insulator to Metal Transition in VO<sub>2</sub>
  - 2. The complexity of  $V_2O_3$
  - 3. Pressure vs Alloying effects in NiS2
  - 4. Tuning a CDW instability in RE dichalchogenides

## **Electromagnetic Spectrum**

The "Terahertz gap", Collective Excitations in Macromolecules and exotic electronic materials



#### IR Units: 200 cm<sup>-1</sup>=300 K= 25 meV = 50 $\mu$ m = 7 THz



## **Advantages of IRSR**

Brightness gain

Small samples
Extreme experimental conditions (high pressure, high magnetic field)
Spatial resolution (microscopy)

Flux gain in the THz range ≈ 10 with incoherent IRSR
> 10<sup>4</sup> with coherent IRSR (CSR) lecture by S. Lupi on May 7



Time resolved spectroscopy (≈10 ps)Pump-probe

## Basics of infrared spectroscopy in condensed matter physics

#### **Refractive index and Optical constants**

2



Complex refractive index N = n + ik

Real refractive index (Snell's law, dispersion) Extinction coefficient (absorption)

$$\tilde{N} = \sqrt{\tilde{\varepsilon}(\omega)} \qquad \begin{array}{l} \varepsilon_1 = n^2 - k^2, \\ \varepsilon_2 = 2nk \end{array}$$

$$\tilde{\sigma} = i \frac{\omega}{4\pi} (1 - \tilde{\varepsilon}) \qquad \begin{array}{l} \sigma_1 = \frac{\omega \varepsilon_2}{4\pi}, \\ \sigma_2 = (1 - \varepsilon_1) \frac{\omega}{4\pi} \end{array}$$

Complex dielectric constant

Complex optical conductivity

# Absorbance, Optical Density and the Lambert-Beer's law



### **Optical Reflectivity**

From Maxwell to Fresnel equations

$$\hat{t}_{\perp} = \frac{E_{0t}}{E_{0i}} = \frac{2\mu_1 \cos \psi_i}{\mu_1 \cos \psi_i + \sqrt{\hat{N}^2 - \sin^2 \psi_i}} \qquad \hat{t}_{||} = \frac{E_{0t}}{E_{0i}} = \frac{2\mu_1 \hat{N} \cos \psi_i}{\hat{N}^2 \cos \psi_i + \mu_1 \sqrt{\hat{N}^2 - \sin^2 \psi_i}}$$

$$\hat{r}_{\perp} = \frac{E_{0r}}{E_{0i}} = \frac{\mu_1 \cos \psi_i - \sqrt{\hat{N}^2 - \sin^2 \psi_i}}{\mu_1 \cos \psi_i + \sqrt{\hat{N}^2 - \sin^2 \psi_i}} \qquad \hat{r}_{||} = \frac{E_{0r}}{E_{0i}} = \frac{\hat{N}^2 \cos \psi_i - \mu_1 \sqrt{\hat{N}^2 - \sin^2 \psi_i}}{\hat{N}^2 \cos \psi_i + \mu_1 \sqrt{\hat{N}^2 - \sin^2 \psi_i}}$$



By normal incidence

#### ... and at vacuum interface

## **Kramers-Kronig Transformations**

Following causality principle, the real and imaginary part of any linear response function are univocally determined through Kramers-Kronig Transformations

$$\tilde{r}(\omega) = \sqrt{R(\omega)} \exp[i\vartheta(\omega)] \longrightarrow \chi(\omega) = \ln(\tilde{r}(\omega)) = \ln\sqrt{R(\omega)} + i\vartheta(\omega)$$
$$\vartheta(\omega) = -\frac{2\omega}{\pi} P \int_{0}^{\infty} \frac{\ln\sqrt{R(\omega')}}{\omega'^{2} - \omega^{2}} d\omega'$$

To this aim one should measure  $R(\omega)$  over the broadest energy range: Extrapolation procedures are also used both at high and low frequencies

#### **The Lorentz-Drude model**



## **Diamond Anvil Cell (DAC)**







$$I = I_T / I_0$$
  
Op.D. = -ln(T) =  $\alpha d$ 





0

 $R_{dia/sam} = I_R / I_0$ 

#### **Infrared measurements at high pressure**

Infrared spectroscopy is a bulk, contactless technique which probes both electronic and phononic excitations over a broad energy range



With  $n_d=2.43$  being the diamond refractive index

The full determination of the optical constants can be achieved through:

- Simultaneous measurement of R and T
- Lorentz-Drude fitting
- Kramers-Kronig Transformations

# **Electrodynamics at high pressures of strongly correlated electron systems**

#### **Band insulators**



The band theory is properly defined in the one-electron approximation, i.e. when W is much larger than other electronic energy scales

W >> U, e - ph

#### **Mott-Hubbard insulators**



metal metal Filling-controlled MIT Bandwidth-controlled MIT Bandwidth-controlled MIT

U prevents double on-site occupancy the opening of a gap in the spectra of excitations is induced

$$H = -t \sum_{\langle i,j \rangle,\sigma} c_{i,\sigma}^{\dagger} c_{j,\sigma} + U \sum_{i=1}^{N} n_{i\uparrow} n_{i\downarrow}$$

# **Electron-electron interaction and insulator to metal transition (MIT)**

Electronic correlation: failure of band model





Pressure may increase the W/U ratio inducing a MIT

W: bandwidth or kinetic energy (strongly dependent on atomic distances) U : coulomb repulsion

#### **Strongly correlated electron systems**



5/8

CO

AFI

0.6

doping

PI

7/8

0.8

CAF

Mott insulators (V203, Bechgaard salts, NiS2)



Manganites

3/8

CMR

0.4

FM

Temperature

x=1/8

CO

0

0.2

4/8



## P-Induced Insulator to Metal Transition in Vanadium Dioxide - VO<sub>2</sub>

#### The metal-insulator transition in VO<sub>2</sub>



Strong Interplay between electron-lattice and electron-electron interactions **Driving MIT mechanism Hubbard or Peierls?** High Pressure may disentangle the two mechanisms <u>Chosen techniques</u>: IR and Raman so to monitor independently electronic and structural changes

#### P-dependent reflectance and transmittance of VO<sub>2</sub>

Simultaneous measurements of reflectance and transmittance at 300 K in the M1 monoclinic insulating phase on a thin VO<sub>2</sub> sample



E. Arcangeletti et al, PRL 98, 196406 (2007)

 $VO_{2}$  slab

KBr pellet

#### **Optical conductivity of VO<sub>2</sub> under high pressure**





4500

 $\omega(cm)$ 

#### **Raman spectroscopy on VO<sub>2</sub>**

V-V V-O 19.1 GPa 15.1 GPa 13.1 GPa 13.1 GPa 4.4 GPa 0.2 GPa 200 400 600 800 Raman shift (cift)  M1 phase: 18 Raman active modes
 • R phase: 4 broad band

Raman spectroscopy is a good technique to study structural transition in VO<sub>2</sub>

Monoclinic phase is stable up to 19 GPa (up to 39 GPa from XRD, L. Malavasi et al, unpublished)

> Anomalous behaviour of low frequency modes

Subtle rearrangement of V-V chains New Monoclinic metallic structure?



#### **Raman spectroscopy on VO<sub>2</sub> under high pressure**

Insulator to metal transition and monoclinic to rutile transition are decoupled at high pressure

Metallization process in a new Mx monoclinic structure



Electron correlation is the main localization mechanism (at least at high-P) in VO<sub>2</sub>

E. Arcangeletti et al, PRL98, 196406 (2007) C. Marini et al, PRB 2008

## The complex phase diagram of Vanadium Sesquioxide - V<sub>2</sub>O<sub>3</sub>

## Vanadium Sesquioxide - V<sub>2</sub>O<sub>3</sub>



## Pressure vs Alloying effects in Nickel Disulfide - NiS<sub>2</sub>



Se increases bandwidth while expanding the lattice! Surprising similarities in the T-x, T-P phase diagrams

S.Miyasaka et al, J. Phys. Soc. Jpn. 69, 3166 (2000)

### **Coherent vs Incoherent excitations in NiS<sub>2-x</sub>Se**<sub>x</sub>



#### Mott-Hubbard



#### **Bandwidth vs CT controlled transition**





Two different MITs: 1) under pressure, W / $\Delta$  = const. and W /U decreases, triggering the MIT; 2) Se alloying: W /U = const. whereas W / $\Delta$ increases inducing the MIT;

## Tuning a CDW instability in Rare-Earth Dichalchogenides

#### **Nesting wavevectors and the Peierls transition**



#### Hydrostatic pressure vs. chemical pressure



With increasing P, R( $\omega$ ) increases between 2000 and 5000 cm<sup>-1</sup>

The same kind of behavior is observed by applying chemical pressure (substitution of the Rare Earth ion).

A. Sacchetti et al., PRL 98, 026401 (2007)

## **Optical conductivity**



A. Sacchetti et al., PRL 98, 026401 (2007)

The optical conductivity is recovered through a fitting of the reflectivity data.

It is then possible to give an estimate for the single particle-gap energy.

#### **Single-particle gap**



The high pressure measurement permits to study the evolution of the SP gap as a function of the lattice parameter **by using one single sample**.

#### **Pressure vs. Alloying**

### VO<sub>2</sub>

Cr-doping changes structure within the insulating phase but does not affect electronic properties. Independently on the level of Cr-doping Pressure drives a metallic monoclinic phase

## $V_2O_3$

Besides expanding the lattice, Cr-doping induces phase separation. Pressure drives a more homogeneous metallic phase but does not completely restore the pristine  $V_2O_3$  homogeneous metallic state. Pressure vs doping scaling does not hold

#### NiS<sub>2</sub>

Se expands the lattice contrary to Pressure Se drives metallicity through a reduction of  $\Delta$ : merging of Ni-e<sub>g</sub> and chalchogen pp $\sigma$  bands Pressure increases bandwidth of Ni-e<sub>g</sub>

### CeTe<sub>3</sub>

Pressure and Rare Earth substitution have the same effect on CDW: Reduced nesting because of increased 3-D character



•Review on T- and P-induced MITs on vanadium oxide compounds: Journal of Physics: Condensed Matter 21, 323202 (2009)

•VO<sub>2</sub> and Cr-doped VO<sub>2</sub>: *Phys. Rev. Lett.* **98**, 196406 (2007); *Phys. Rev. B* **77**, 235111 (2008)

• T-dependence of pure V<sub>2</sub>O<sub>3</sub> (Cr-V<sub>2</sub>O<sub>3</sub> still unpublished): *Phys. Rev. Lett.* **75**, *105* (*1995*); *Phys. Rev.* **B** 77, *113107* (*2008*)

•NiS<sub>2</sub> as a function of alloying and pressure *Phys. Rev. B* **80**, 073101 (2009)

•Pressure dependence of RE chalchogenides Phys. Rev. Lett. 98, 026401 (2007); Phys. Rev. B 77, 165132 (2008); Phys. Rev. B 79, 075117 (2009)

#### **Acknowledgements**

The SISSI@ELETTRA group
L. Baldassarre
G. Birarda
M. Kiskinova
S. Lupi
F. Piccirilli
L. Vaccari

P. Calvani and A. Nucara, IRS Group, University of Rome "La Sapienza"

M. Ortolani CNR-IFN U. Schade Synchrotron BESSY II Berlin P. Postorino HP Group, University of Rome "La Sapienza" L. Degiorgi ETH Zurich L. Malavasi Dip. di Chimica, Università di Pavia M. Marsi CNRS-LPSU, France V.A. Sidorov Institute for High Pressure Physics Troitsk, Moscow Region Russia

and many others.....