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Scanning photoemission microscope: A powerful tool for studies on Fuel cells and catalysis supporting material

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Scanning photoemission microscope: A powerful tool for studies on Fuel cells and catalysis supporting material

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The Microscopy Section at Elettra

Trieste

Elettra Synchrotron Light Source laboratory

•ESCAmicroscopy: 400-1200 eV XPS spectro-microscopy •SPECTROmicroscopy: 20-90 eV XPS, ARUPS spectro-microscopy •NANOspectroscopy: LEEM, X-PEEM spectro-microscopy

•TwinMic: X-ray transmission and absorption spectro-microscopy

•SISSY: infrared spectro-microscopy





SPEM layout and performance







Energy resolution









SPEM experiments: main topics



Nanostructures/devices characterization

- MWCNTs mass transport and reactivity
- e-noses
- Size dependent electronic properties of semiconductors

Electrochemistry/SOFC

- Electrochemical stability of materials
- Challenging experiments: high temps, biasing, low concentrations

Catalysis

- 'Material' gap: from model crystalline materials to metal nano-particles on metal oxide.
- In situ PLD particle deposition



A. Barinov et al. Phys. Rev. Lett. 99, 046803 (2007)

Gas phase oxidation of MCNT





Effects on size and surface chemistry on the conductivity of MBEgrown GaAs nanowires (in collaboration with S. Rubini – TASC Laboratory - Italy)





•Increasing surface-to-volume ratio

•Size dependence of the depletion width, band gap widths, recombination barriers, etc.

•i.e. Debye length comparable to the radius of NW



Chemical and electronic characterization of nanosensors

(in collaboration with A. Kolmakov – Souther Illinois Uni. - USA)

•Chemical & electronic characterization under working conditions \circ SnO₂, VO_x, ...

•Sensing properties vs oxygen, hydrogen, ...







) | 1993



•Addressing the electron transport in a workin device (temperature, close biasing, etc.) •Surface stoichiometry, coordination, oxidation state, etc.







A. Kolmakov et al. ACSNano 2 (2008) 1993

elettra Spectromicroscopy and photoluminescence analysis of ZnO nanostructures





M. Kazemian Abyaneh et al. J Nanopart Res 13(2010)1311



Ti6Al4V-SiC_f composite produced by HIP for aeronautical applications



(in collaboration with S. Kaciulis – ISMN-CNR, Rome, Italy)



S. Kaciulis et al. Surf. Interface Anal. 2010, 42, 707–711

Compositional and electronic study of TCO nano and microtubes (in collaboration with A. Cremades - UCM - Spain)

•Catalyst free growth of TCO structures (Sn_xO_y/In_xO_y/In_xN_y,Ge_xO_y,etc.)

•SPEM characterization of morphological complex structures difficult with other PEM



SPEM images



D. A. Magdas et al. Superlat. and Microstr. 45 (2009) 429-434 D. Maestre et al. Journ. of Appl. Phys. 103, 093531 (2008) D. Maestre et al. J. Phys. Chem. C, 2010, 114 (27), pp 11748–11752

Electronic behaviour of a single structure

•Charging due to differences in the electronic structure •Mapping of the charging with the multichannel acquisition





Local chemical composition of the structures

•Heterogeneous elemental distribution locally defined •Fine chemical analysis



Degradation of light emitting diodes: a SPEM analysis

(in collaboration with P. Melpignano CRP, R. Zamboni CNR-ISMN)





Operating SOFC: mass transport

(in collaboration with M. Backhaus- Corning Inc. - USA)



Strongly constraining experimental setup



•Real samples
•High T = 650-700℃
•pO₂=1x10⁻⁶ mbar
•Applied potentials

-2V<U<+2V

•Surface sensitive

technique
•High lateral resolution

Elemental distribution at electrolyte/LSM interface



Surface composition change with bias



1-Mn 3p USM YSZ YSZ USM

bias in V

Observation and explanation of electrochemical cathode activation

Strong current increase under negative bias when Mn spreads on electrolyte

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- •Mn2+ electrolyte surface enrichment→electrolyte surface conductivity → direct oxygen incorporation into electrolyte
- Oxygen incorporation extends under bias from TPB to the entire electrolyte surface
- M. Backhaus et al. Solid State Ionics 179 (2008) 891-895, M. Backhaus et al. Advances in Solid Oxide Fuel Cells III 28 (4), 2007





B. Bozzini et al. PCCP in press



'Material' gap: from model crystalline materials to metal nanoparticles on metal oxide. In situ PLD particle deposition



•Reducing rate: Micro-part.> Nano-crystalline film > Nano-particles

•Micro-particles of similar sizes show variation in the reactive properties: different structure, local environ.

P. Dudin et al. J. Phys. Chem. C 2008, 112, 9040-9044



M. Dalmiglio et al. J. Phys. Chem. C http://dx.doi.org/10.1021/jp910370r



M. Dalmiglio et al. J. Phys. Chem. C, 114(2010) 16885





Thank You



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