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Joint ICTP-IAEA Advanced School on the Role of Nuclear Technology in Hydrogen-Based Energy Systems

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In situ Electrochemical Studies of Fuel Cells by Soft-X-Ray Spectromicroscopy

A. Gianoncelli Elettra Sinctrotrone Trieste Basovizza Italy

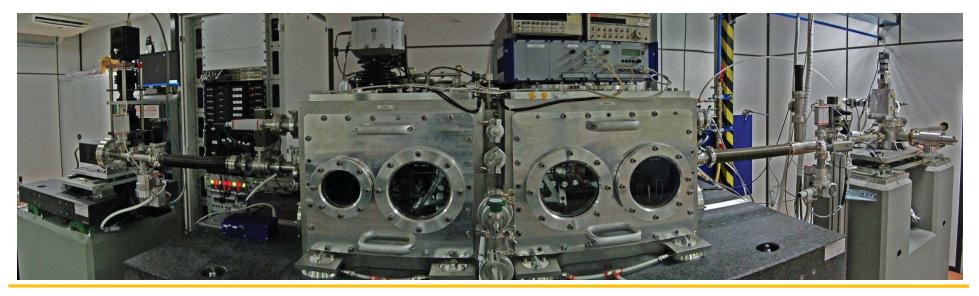




In situ Electrochemical Studies of Fuel Cells by Soft-X-Ray Spectromicroscopy

<u>A. Gianoncelli¹, B. Bozzini², C. Mele², B. Kaulich¹, M. Prasciolu³, M. Kiskinova¹</u>

¹ Elettra - Sinctrotrone Trieste – Basovizza (TS, Italy)
² Dipartimento di Ingegneria dell'Innovazione – Università del Salento – Lecce (Italy)
³ INFM TASC National Lab. – Basovizza (TS, Italy)



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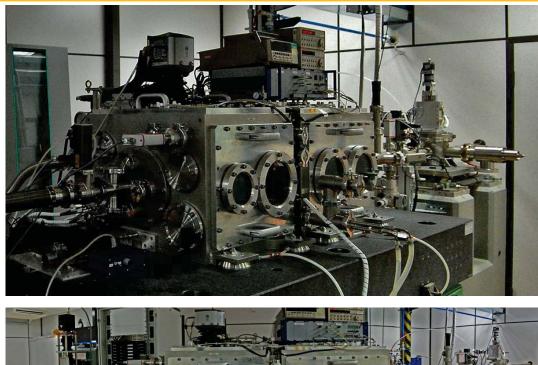


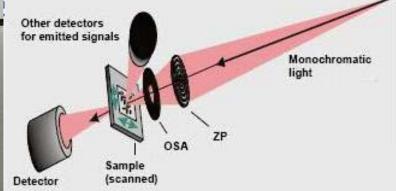
TwinMic

http://www.elettra.trieste.it/twinmic The twin X-ray microscopy station at ELETTRA



Scanning Transmission X-ray Microscope (STXM) at Elettra





- Lateral resolution down to 50 nm
- Absorption and Phase contrast Imaging, XANES, XRF
- Photon energy range: 400÷2200 eV

B. Kaulich et al, TwinMic: A European Twin X-ray Microscopy Station Commissioned at ELETTRA, in Proc. 8th Int. Conf. X-ray microscopy (eds. S. Aoki, Y. Kagoshima, Y. Suzuki), Conf. Proc. Series IPAP 7, pp. 22-25.

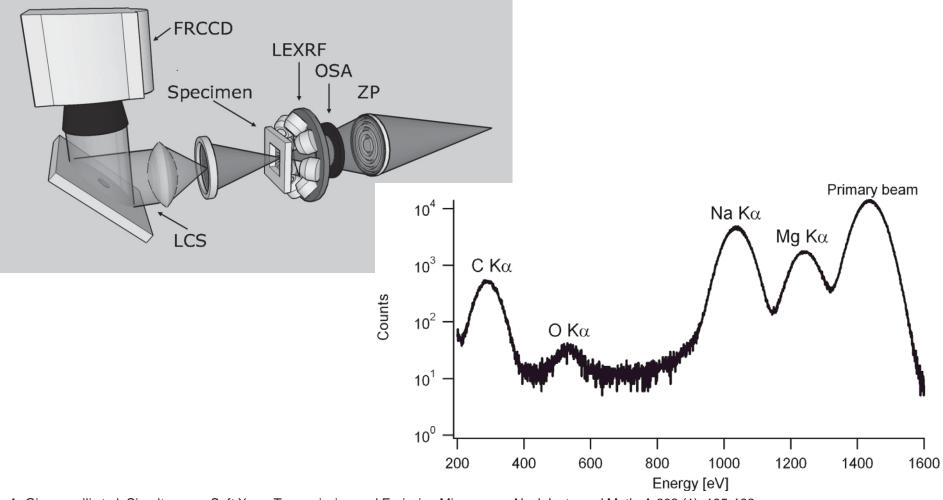
G.R. Morrison et al, A fast-readout CCD system for configured-detector imaging in STXM, in Proc. 8th Int. Conf. X-ray microscopy (eds. S. Aoki, Y. Kagoshima,

- Y. Suzuki), Conf. Proc. Series IPAP 7, 377-379.
- A. Gianoncelli, et al, A fast read-out CCD camera system for scanning X-ray microscopy, Appl. Phys.Lett. 89 (2006), 251117.

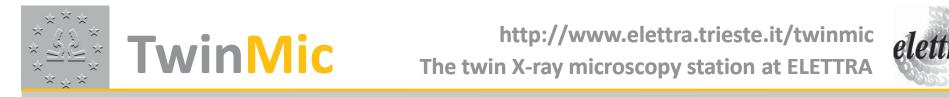




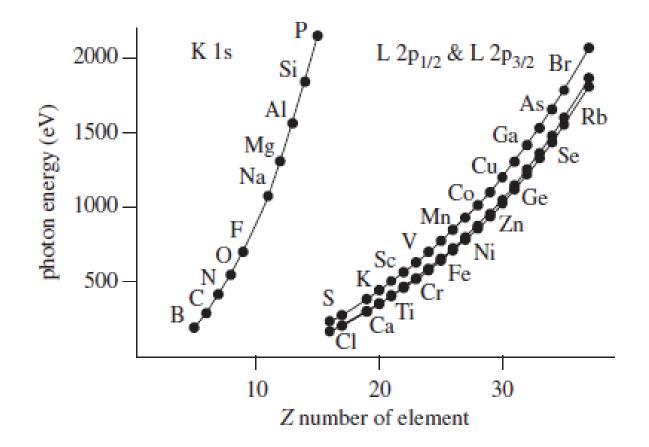
Low Energy X-ray Fluorescence system



A. Gianoncelli et al, Simultaneous Soft X-ray Transmission and Emission Microscopy, Nucl. Instr. and Meth. A 608 (1), 195-198. B. Kaulich et al, Low-energy X-ray fluorescence microscopy opening new opportunities for bio-related research, J. R. Soc. Interface 6 (Suppl 5), S641-S647.



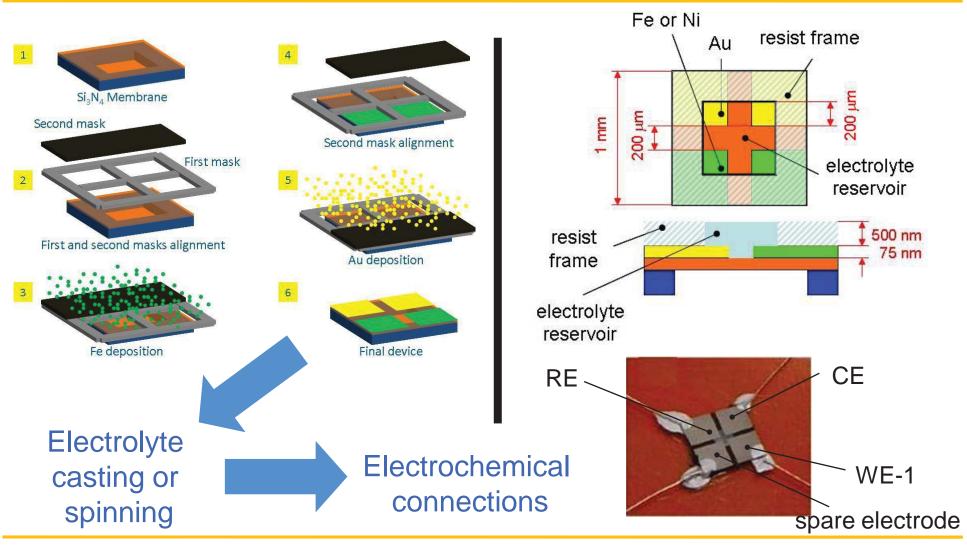
Elements accessible by TwinMic (XANES, XRF, AEI)

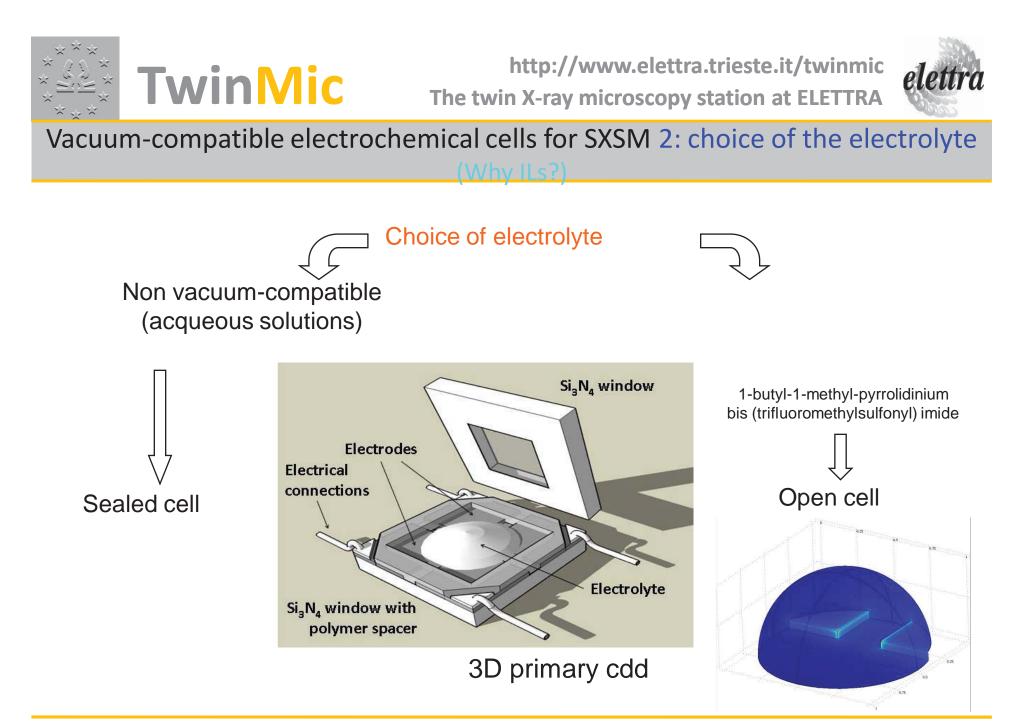


B. Kaulich et al, Low-energy X-ray fluorescence microscopy opening new opportunities for bio-related research, J. R. Soc. Interface 6 (Suppl 5), S641-S647.



Vacuum-compatible electrochemical cells for SXSM : 1: electrodic system





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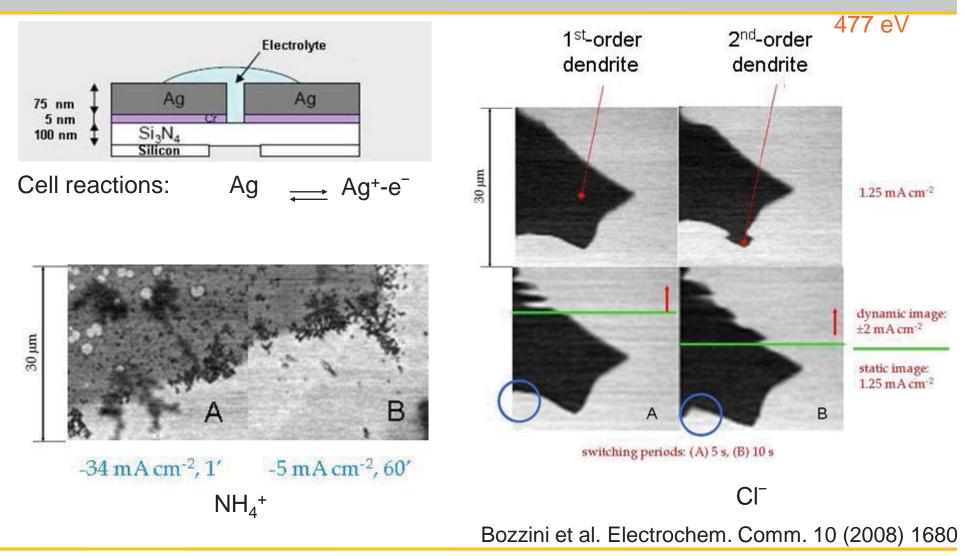
Systems investigated by in situ SXMS: a selection of results

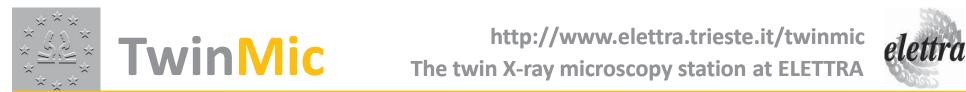
- 1) Feasibility tests with dynamic Ag redox
 - [Bozzini et al. Electrochem. Comm. 10 (2008) 1680]
- 2) Stability of Fe and Ni BPs in aqueous $SO_4^{2-} \pm F^-$:
 - corrosion of BPs as such (R_{Ω} \hat{T} , pinholes, fate of corrosion products: formation of dendrites, shorting paths, mechanical damage)
 - [Bozzini et al. J. Phys. Chem. C 113 (2009) 9783]
- 3) Pickup of Fe and Ni BP corrosion products by Nafion
 - *generation of corrosion products & their fate wrt PEM poisoning* [Bozzini et al. ChemSusChem 7 (2010) 864]
- 4) Stability of Fe and Ni BPs in [BMP][TFSI]
 - corrosion of BPs in novel, ideally hydrating electrolyte, fate of corrosion products
- 5) Anodic (fuel: NaBH₄) and cathodic (oxidiser: O₂ 10⁻⁶ mbar) half-cells with of Fe BPs in galvanic contact with Pt catalyst, el-lyte: Nafion/[BMP][TFSI] composite result of FC operation on corrosion of anodic and cathodic BPs



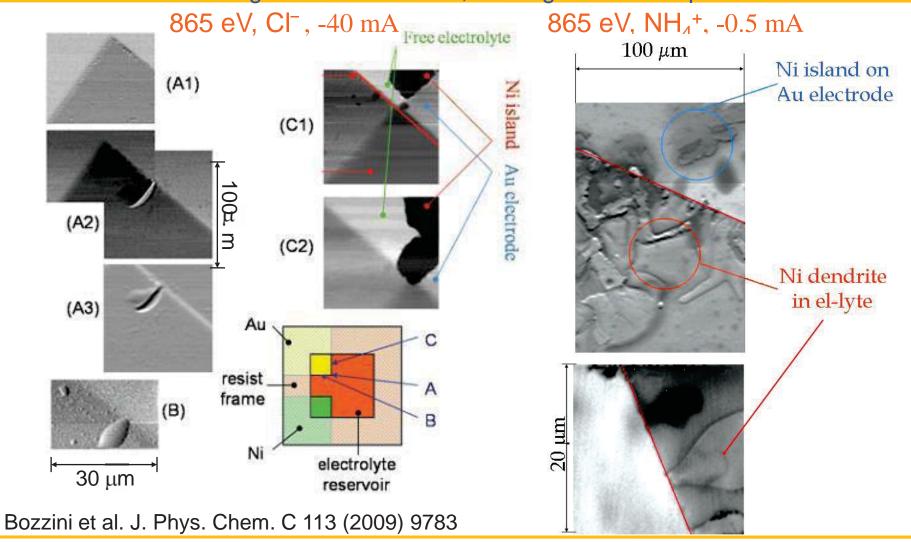
elettra

1- Proving in situ electrochemical STXM with aqueous el-lyte & sealed cell





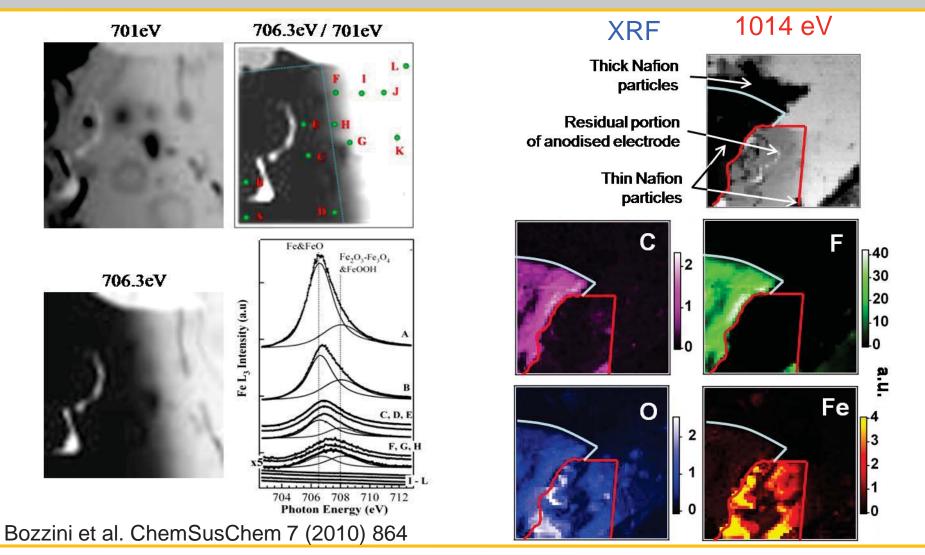
2 - In situ dissolution and redeposition of Ni STXM with aqueous el-lytes Homogeneous dissolution, heterogeneous redeposition

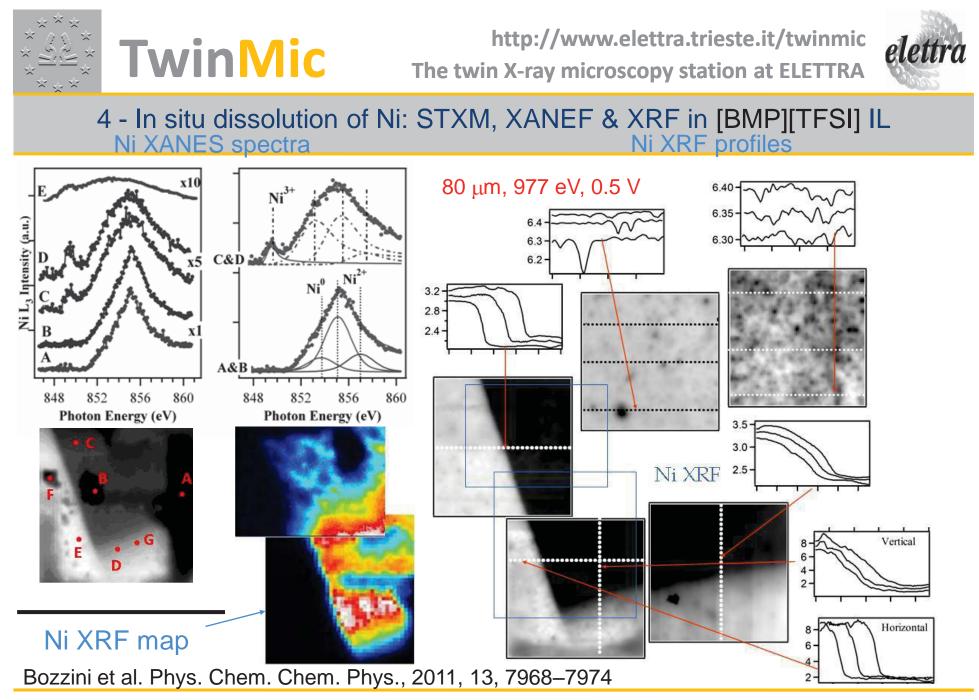






3 - Release of Fe BP corrosion products to aqueous el-lyte and fixation in Nafion

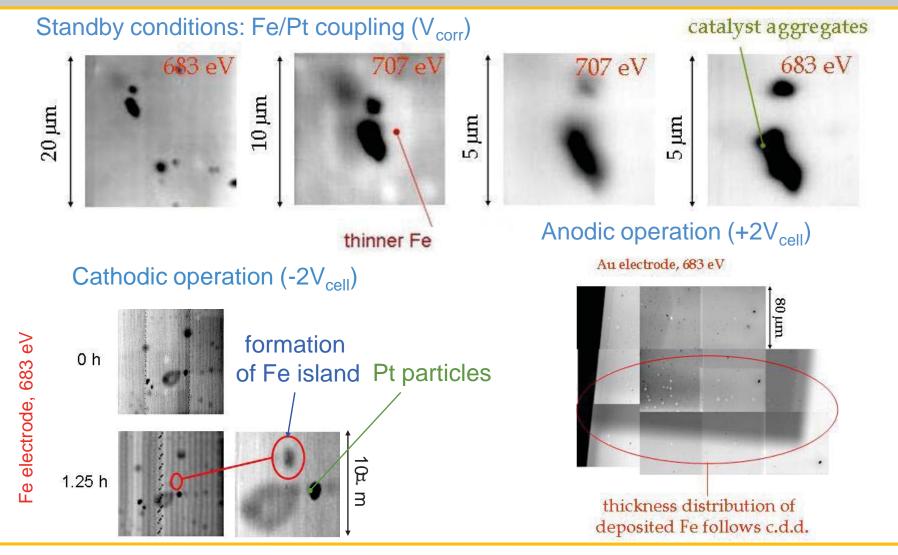








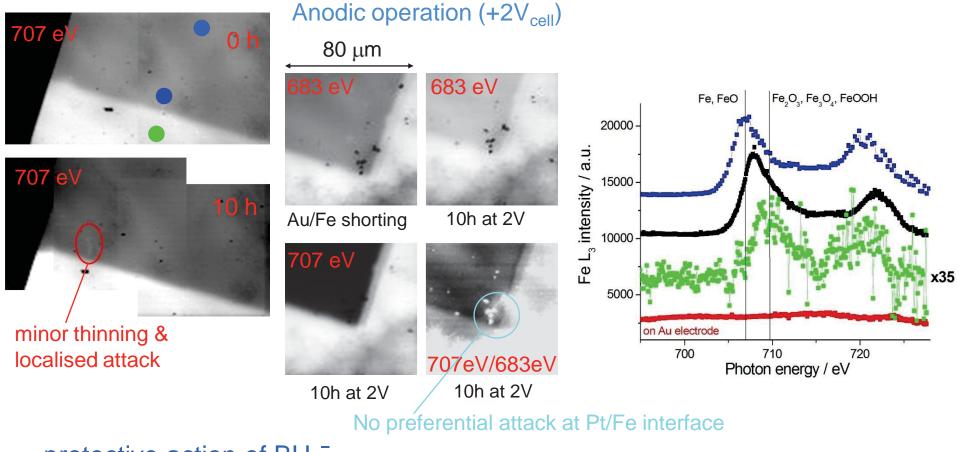
5 - In situ study of the cathodic half-cell (Fe/Pt/O₂)







5 - In situ study of the anodic half-cell (Fe/Pt/NaBH₄)



protective action of BH₄⁻

Standby conditions and cathodic $(-2V_{cell})$ polarisation for 5 h: no measurable effects





Conclusions

In situ electrochemical dynamic spectromicroscopy with resolution 50 nm

Microfabricated vacuum-tight cells: nm thick electrodes and electrolyte

Morphological, chemical and structural effects of electrochemistry

Completed actions: stability of BP matarials under FC conditions in: (i) a range of aqueous systems, (ii) Nafion, (iii) IL, (iv) IL-Nafion composite.

Next action: in situ dynamical analysis of a complete FC combining the described half-cells with a microfluidic approach.

In addition to FC and electrochemical energetics (LIB, supercapacitors) this approach is also expected to impact all fields of electrochemical materials science and bioelectrochemistry